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Foreword

The purpose of this techno-economic analysis is to compare a set of biofuel conversion technologies selected for their promise and near-term technical viability. Every effort is made to make this comparison on an equivalent basis using common assumptions. The process design and parameter value choices underlying this analysis are based on public domain literature only. For these reasons, these results are not indicative of potential performance, but are meant to represent the most likely performance given the current state of public knowledge.

List of Acronyms

AGR	acid gas removal
ASU	air separation unit
BTL	biomass to liquids
CFB	circulating fluidized bed
DCFROR	discounted cash flow rate of return
DME	dimethyl-ether
FCI	fixed capital investment
FT	Fischer-Tropsch
GGE	gallon of gasoline equivalent
HRSG	heat recovery steam generator
HT	high temperature
IC	indirect costs
IGCC	integrated gasification combined cycle
IRR	internal rate of return
ISU	Iowa State University
LHV	lower heating value
LT	low temperature
MEA	monoethanolamine
MJ	megajoule
MM	million
MTG	methanol to gasoline
MW	megawatt
Nm ³	normal cubic meter
NREL	National Renewable Energy Laboratory
PSA	pressure swing adsorption
PV	product value
Sasol	South African Coal, Oil, and Gas Corporation
SPR	slurry phase reactor
SMR	steam methane reforming
SWGS	sour water-gas-shift
TCI	total capital investment
TDIC	total direct and indirect cost
TIC	total installed cost
tpd	tons per day
TPEC	total purchased equipment cost
WGS	water-gas-shift

Executive Summary

This study compares capital and production costs of two biomass-to-liquid production plants based on gasification. The goal is to produce liquid transportation fuels via Fischer-Tropsch synthesis with electricity as a co-product. The biorefineries are fed by 2,000 metric tons per day of corn stover. The first biorefinery scenario is an oxygen-fed, low-temperature (870°C), non-slugging, fluidized bed gasifier. The second scenario is an oxygen-fed, high-temperature (1,300°C), slugging, entrained flow gasifier. Both are followed by catalytic Fischer-Tropsch synthesis and hydroprocessing to naphtha-range (gasoline blend stock) and distillate-range (diesel blend stock) liquid fractions. (Hydroprocessing is a set of refinery processes that removes impurities and breaks down large molecules to fractions suitable for use in commercial formulations.)

Process modeling software (Aspen Plus) is utilized to organize the mass and energy streams and cost estimation software is used to generate equipment costs. Economic analysis is performed to estimate the capital investment and operating costs. A 20-year discounted cash flow rate of return analysis is developed to estimate a fuel product value (PV) at a net present value of zero with 10% internal rate of return. All costs are adjusted to the year 2007. The technology is limited to commercial technology available for implementation in the next 5–8 years, and as a result, the process design is restricted to available rather than projected data.

Results show that the total capital investment required for n^{th} plant scenarios is \$610 million and \$500 million for high-temperature and low-temperature scenarios, respectively. PV for the high-temperature and low-temperature scenarios is estimated to be \$4.30 and \$4.80 per gallon of gasoline equivalent (GGE), respectively, based on a feedstock cost of \$75 per dry short ton. The main reason for a difference in PV between the scenarios is because of a higher carbon efficiency and subsequent higher fuel yield for the high-temperature scenario. Sensitivity analysis is also performed on process and economic parameters. This analysis shows that total capital investment and feedstock cost are among the most influential parameters affecting the PV, while least influential parameters include per-pass Fischer-Tropsch-reaction-conversion extent, inlet feedstock moisture, and catalyst cost.

In order to estimate the cost of a pioneer plant (first of its kind), an analysis is performed that inflates total capital investment and deflates the plant output for the first several years of operation. Base case results of this analysis estimate a pioneer plant investment to be \$1.4 billion and \$1.1 billion for high-temperature and low-temperature scenarios, respectively. Resulting PVs are estimated to be \$7.60/GGE and \$8.10/GGE for high-temperature and low-temperature pioneer plants, respectively.

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Introduction

This study investigates the economic feasibility of the thermochemical pathway of gasification of biomass to renewable transportation fuels. The objective is to compare capital investment costs and production costs for n^{th} plant biorefinery scenarios based on gasification. The selected scenarios are high-temperature (slagging) gasification and low-temperature (dry-ash) gasification, each followed by Fischer-Tropsch synthesis and hydroprocessing. They are designed to produce liquid hydrocarbon fuels from 2,000 dry metric tons (2,205 dry short tons) per day of agricultural residue, namely corn stover. Corn stover is chosen as a feedstock in order to facilitate comparisons with biochemical and pyrolysis biofuels scenarios [1, 2].

The two scenarios were chosen from many options according to the following criteria:

1. The technology under consideration should be commercially ready in the next 5–8 years.
2. The size of the biorefinery should be feasible with current agricultural productivity and within a realistic feedstock collection area.
3. In addition, the desired end product should be compatible with the present fuel infrastructure, i.e., gasoline and/or diesel.

The high-temperature gasification scenario is based on a steam/oxygen-fed entrained flow, slagging gasifier similar to that described in Frey and Akunuri [3]. The low-temperature gasification scenario is based on a pressurized, steam/oxygen-fed fluidized bed gasifier developed by the Gas Technology Institute and reported by Bain [4]. The main areas of operation are feedstock preprocessing, gasification, syngas cleaning, syngas conditioning/upgrading, fuel synthesis, power generation, and air separation (for oxygen production), as shown in Figure 1. Process modeling software is utilized to organize the mass and energy streams, and cost estimation software is used to generate equipment costs. Economic analysis is performed to estimate the capital investment and operating costs. A 20-year discounted cash flow rate of return (DCFROR) analysis is developed to estimate a fuel product value (PV) at a net present value of zero with 10% internal rate of return. All costs are adjusted to the year 2007.

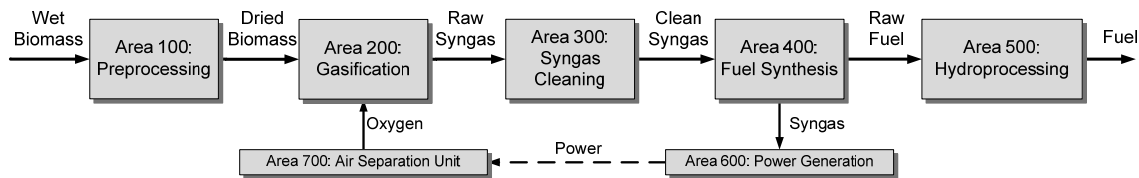


Figure 1. Overall process flow diagram for both scenarios

Background

Biorenewable Resources

The world population has long utilized materials that were in close proximity. The nearest resource available to the human population is the organic matter in the environment around it. This organic matter is present for a limited amount of time due to its decomposable nature. Brown [5] defines this material, or biorenewable resources, as organic material of recent biological origin. It is a renewable resource if the rate of consumption is equal to the regeneration or growth and therefore must be used only if preserving biodiversity [6]. As a result, these resources have been important contributors to the world economy, providing foodstuffs, transportation, energy, and construction materials, as well as serving many other functions.

Biorenewable resources for generating energy can be classified as woody biomass, energy crops, residues, and municipal waste [6]. The first two are primary resources while the remaining are secondary resources, meaning that their primary use has already occurred. Woody biomass includes logging products and energy crops include short-rotation trees such as poplar and fast-growing grasses such as switchgrass. Residues can come from logging processing or agricultural processing (e.g., corn stover). According to Perlack et al. [7], the energy crop and agricultural residue potential in the United States is 1.4 billion annual tons. According to the U.S. Department of Energy's "Roadmap for Agriculture Biomass Feedstock Supply in the U.S.," there is potential for 2 billion annual tons, including municipal waste and biosolids such as manure.

Many end products can be produced from these resources. Aside from the conventional use of biomass for human food consumption, livestock feed, and building materials, there are many new pathways to provide renewable alternatives to our transportation, infrastructure, and energy. Combustion of biomass offers a way to provide heat and power and displaces coal and fuel oil. Fermentation of carbohydrates and liquefaction of biomass through fast pyrolysis yield liquid products with the potential to displace petrochemicals. Gasification of biomass allows for chemical and liquid fuel synthesis, which is the focus of this study.

Developing an economy that involves biorenewable resources, especially biofuels, has many benefits. According to Greene et al. [8], biofuel production has the potential to provide a new source of revenue for farmers by generating \$5 billion per year. Additionally, toxic and greenhouse gas emissions can be reduced by the use of biofuels. In the same study, Greene et al. report that 22% of the United States's total greenhouse gas emissions could be reduced if biofuels were developed to replace half of the petroleum consumption. Arguably, the most important benefit of biofuel production is the potential for closing the carbon cycle.

Gasification

Gasification is a high-temperature and catalytic pathway for producing biofuels. It is defined as the partial oxidation of solid, carbonaceous material with air, steam, or oxygen into a flammable gas mixture called producer gas or synthesis gas [5]. The synthesis gas contains mostly carbon monoxide and hydrogen with various amounts of carbon dioxide, water vapor, and methane. Typical volumetric energy content of synthesis gas is 4–18 MJ/Nm³ [9]. Comparatively, natural gas (composed of mostly methane) energy content is 36 MJ/Nm³ [9]. Much of the energy content

of the biomass is retained in the gas mixture by partial oxidation rather than full oxidation of the biomass, which would result in the release of mostly thermal energy. Historically, gasification of coal and wood produced “town gas,” which was subsequently burned in street lamps [10]. Additionally, during the World Wars, vehicles were adapted to operate with gasification reactors [10]. During this same time period, Germany developed the catalytic synthesis of transportation fuels from synthesis gas [11]. The same concept is still in use today by the South African Coal, Oil, and Gas Corporation (Sasol) to produce motor fuels and liquid by-products using coal [11].

Reactions

Four stages occur during gasification of carbonaceous material: drying, devolatilization, combustion, and reduction [9]. First, the moisture within the material is heated and removed through a drying process. Second, continued heating devolatilizes the material where volatile matter exits the particle and comes into contact with the oxygen. Third, combustion occurs, where carbon dioxide and carbon monoxide are formed from carbon and oxygen. The combustion stage is very exothermic and provides enough heat for the last stage, the reduction reactions, to occur. The last stage includes water gas reaction, Boudouard reaction, water-gas-shift reaction, and methanation reaction (Table 1). As all these stages progress, solid fixed carbon remains present. The amount of fixed carbon varies depending on the equivalence ratio.

Table 1. Reactions Occurring within the Reduction Stage of Gasification

Name	Reaction
Water Gas	$C + H_2O \rightarrow CO + H_2$
Boudouard	$C + CO_2 \rightarrow 2CO$
Water-Gas-Shift	$CO + H_2O \rightarrow CO_2 + H_2$
Methanation	$CO + 3H_2 \rightarrow CH_4 + H_2O$

When the equivalence ratio (defined as the actual air/fuel ratio all divided by the stoichiometric air/fuel ratio) increases, solid fixed carbon (i.e., char) decreases until enough oxidizer is available for complete conversion (Figure 2). This point of complete conversion occurs at an equivalence ratio of approximately 0.25. At nearly the same point, the maximum synthesis gas energy content (without accounting for sensible energy) is reached.

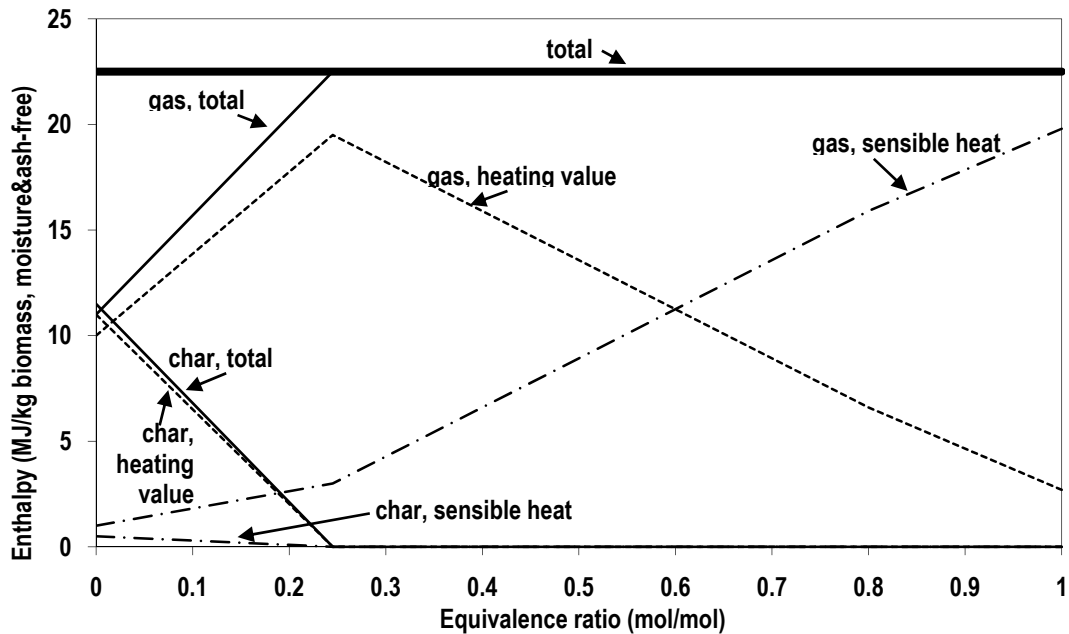


Figure 2. Typical energy content of the products of gasification of wood using air varied by equivalence ratio [12]

Gasifier Types

Three types of reactors are used for gasification: fixed bed, fluidized bed, and entrained flow [13]. Fixed beds are fed with biomass from the top of the reactor and form a bed, which gasifies as air moves through the bed (Figure 3). As the material releases volatile components, the char and ash exit through a grate at the bottom. Typical operating temperature range is 750°–900°C. The two main types of fixed-bed gasifiers are updraft and downdraft. The advantage of fixed-bed technology is its simplicity, but the technology is limited in scale-up and has low heat mixing due to high channeling potential within the reactor [14].

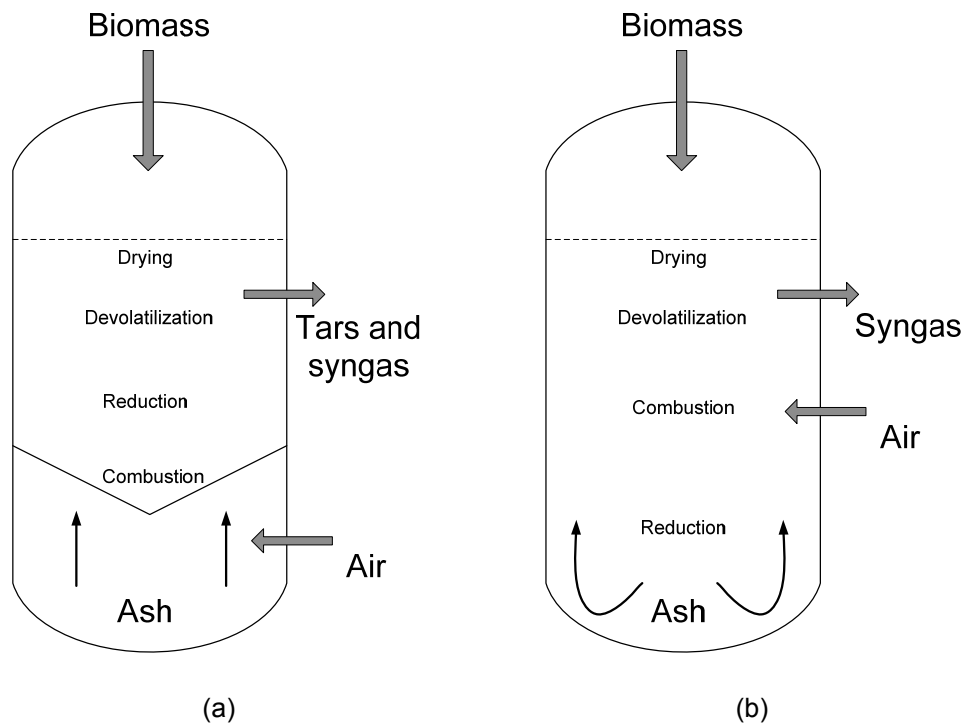


Figure 3. Design of fixed-bed (a) updraft and (b) downdraft gasifiers showing reaction zones [13]

When the volumetric gas flow is increased through the grate, the fixed bed becomes a fluidized bed. Fluidized-bed gasifiers are so named because of the inert bed material that is fluidized by oxidizing gas creating turbulence through the bed material (Figure 4). Biomass enters just above the top of the bed and mixes with hot, inert material, creating very high heat and mass transfer. Operating temperature range is the same as for the fixed bed. Advantages of the fluidized bed include flexible feeds, uniform temperature distribution across the bed, and large volumetric flow capability [15]. The main types of fluidized-bed gasifiers are circulating fluidized bed (CFB) and bubbling fluidized bed. Bubbling-bed gasifiers are directly heated from the combustion reactions occurring in the bed. They produce gas and the ash and char fall out the bottom or the side. The CFB recycles the char through a cyclone while the product leaves out the top of the cyclone. There are also indirectly heated fluidized beds that use a hot material such as sand to provide the heat needed for gasification, as shown in Figure 4. Fluidized beds have high carbon conversion efficiencies and can scale up easily [14].

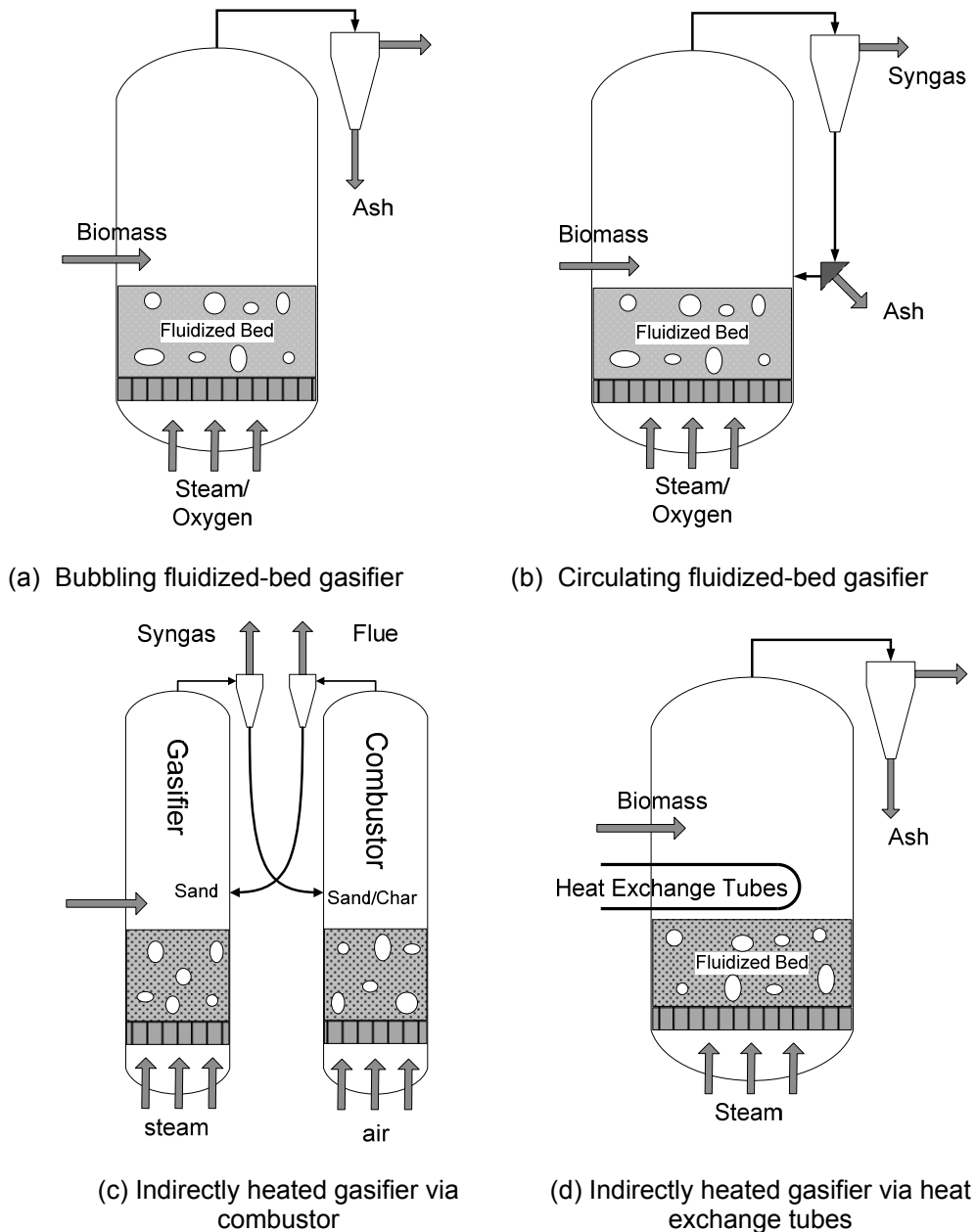


Figure 4. Fluidized bed gasifier designs of (a) and (b) directly heated type and (c) and (d) indirectly heated type [16]

Another type of gasifier is the entrained-flow gasifier (Figure 5). Normally operated at elevated pressures (up to 50 bar), it requires very fine fuel particles gasified at high temperatures to ensure complete gasification during the short residence times in the reactor. The Energy Research Centre of the Netherlands has investigated this type of gasification and reported promise with biomass, as long as the biomass is pretreated to certain requirements [17]. To keep the residence time at approximately the time for a particle to fall the length of the reaction zone, fuel particles smaller than 1 mm and high temperatures (1100°–1500°C) are necessary for successful operation.

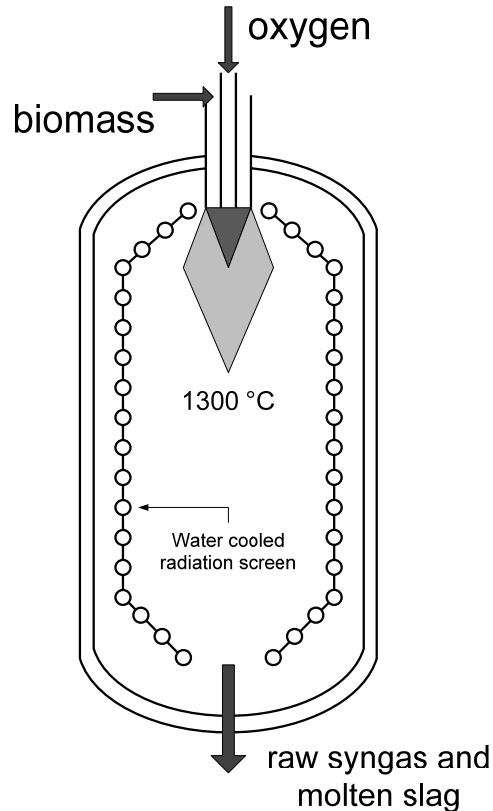


Figure 5. Entrained-flow gasifier [18]

Entrained-flow gasification mixes the fuel with a steam/oxygen stream to form a turbulent flow within the gasifier. Ash-forming components melt in the gasifier and form a liquid slag on the inside wall of the gasifier, effectively protecting the wall itself. The liquid flows down and is collected at the bottom. To form the slag, limestone can be added as a fluxing material. For herbaceous biomass, such as switchgrass or corn stover, which is high in alkali content, there may be sufficient inherent fluxing material present [18]. Advantages of entrained-flow gasification are that tar and methane content are negligible and high carbon conversion occurs due to more complete gasification of the char. Syngas cleanup is simplified because slag is removed at the bottom of the gasifier, negating the need for cyclones and tar removal [19]. The disadvantages are that very high temperatures need to be maintained and the design and operation is more complex. An entrained-flow gasifier co-firing up to 25% biomass with coal has been developed by Shell in Buggenum, Netherlands. Another gasifier developed by Future Energy in Freiburg, Germany, uses waste oil and sludges. Both are operating at commercial scale [17].

Biomass Preprocessing

A degree of biomass processing is required before gasification can occur. Most gasifiers require smaller size feedstock than is typically collected during harvest. Therefore, a significant degree of size reduction is necessary. A typical setup for size reduction is using a two-step process in which a chipper accomplishes the primary reduction and a hammer mill is used for the secondary reduction [20]. In addition, the maximum moisture content for biomass gasification is 20%–30%

(wet basis), and for normal operation it is less than 15% (wet basis) [9]. Therefore, a drying process is required to prepare the feedstock for gasification.

The main benefit of drying biomass is to avoid using energy within the gasifier to heat and dry the feedstock [21]. Drier biomass also makes for more stable temperature control within the gasifier. Rotary dryers typically operate utilizing hot flue gas from a downstream process as the drying medium. They have high capacity but require long residence times. In addition, rotary dryers have a high fire hazard when using flue gas [21]. To avoid using flue gas, rotary dryers can use superheated steam, essentially an inert gas, when a combined-cycle heat and power system is used downstream. That system has significant steam available for use because of the steam produced in the steam cycle. An advantage of using steam for drying is better heat transfer and therefore shorter residence time.

Pretreatment options for entrained-flow gasification include torrefaction followed by grinding to 0.1-mm particles, grinding to 1-mm particles, pyrolysis to produce bio-oil/char slurry (bioslurry), and initial fluidized-bed gasification of larger particles coupled to an entrained-flow gasifier. Torrefaction, essentially an oxygen-free roasting process, causes the biomass particles to be brittle, which makes for easy grinding but releases up to 15% of the energy in the biomass via volatile compounds [17]. The fluidized/entrained coupled option is attractive because of an overall energy efficiency of 80%–85%, but it is expensive because two gasifiers are used in series.

The bioslurry option is illustrated in Figure 6. Basically, a flash-pyrolysis process yields bio-oil and char and is followed by a slagging, entrained-flow gasifier. Because this process utilizes an entrained-flow gasifier, the feed must be pressurized. Fortunately, the pyrolysis slurry, already in an emulsified liquid state, can be pressurized easily. Technology for slurry feeding is state of the art as a result of experience with coal slurries [17]. The bioslurry still contains 90% of the energy contained in the original biomass [22]. Another advantage is that no inert gas is needed for solids pressurization, avoiding dilution of the feed, which would dilute the syngas. In the search for cost-effective methods for production of syngas, this option has potential, but it isn't as developed as other technologies such as fluidized-bed gasification. The biggest challenge is constructing and operating a large-scale pyrolysis process, because large-scale systems have not been demonstrated [17].

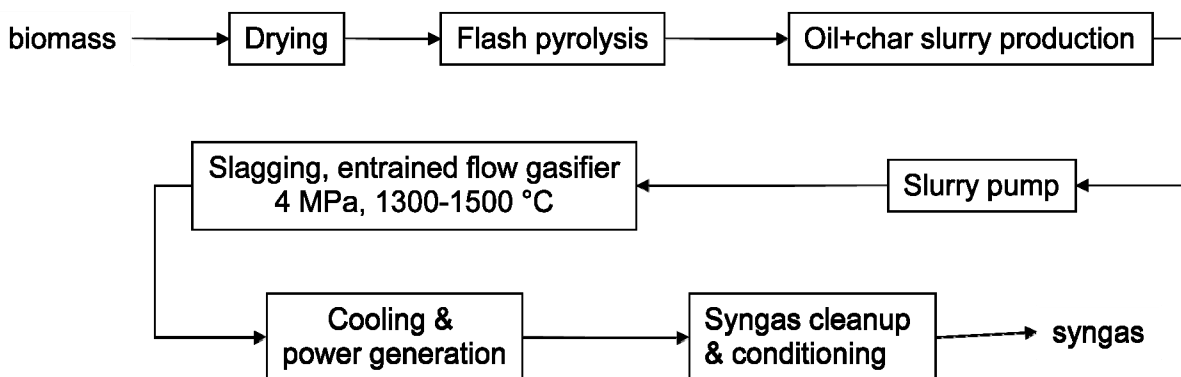


Figure 6. Schematic of a biomass pretreatment via fast pyrolysis followed by an entrained-flow gasifier [17]

Syngas Cleaning

Because the raw syngas leaving the gasifier contains particulates, tars, alkali compounds, sulfur compounds, nitrogen compounds, and other contaminants, those components must be removed or reduced significantly. Particulates and tars have the potential for clogging downstream processes. Sulfur and nitrogen have the potential to poison downstream processes, especially catalysts used in fuel synthesis applications. Moreover, another motivation for cleaning syngas is meeting environmental emissions limits.

The syngas must be cooled before conventional gas cleanup is utilized. This can be accomplished in two ways: direct quench by injection of water and indirect quench via a heat exchanger. Direct quench is less expensive but dilutes the syngas. Direct quenching can also be used to clean up the gas by removing alkali species, particulates, and tars [23].

Particulates are defined as inorganic mineral material, ash, and unconverted biomass or char [24]. In addition, bed material from the gasifier is included in the particulates. Switchgrass feedstock typically has 10% inorganic material in the form of minerals. Many gasifiers operate with a 98%–99% carbon conversion efficiency where 1%–2% of the solid carbon is in the form of char [24].

Particulates are primarily removed by physical systems such as cyclones, in which the heavy particles fall down the center while the gases rise up and out of the cyclone. The initial step for particulate removal is usually a cyclone. Importantly, particulates should be removed before the gas is cooled down for cold gas cleaning. If removed after gas cooling, tars can condense onto the particulates and potentially plug equipment. Alternatively, barrier filters, which operate above tar condensation temperatures, use metal or ceramic screens or filters to remove particulates while allowing the gas to remain hot. Barrier filters, however, have presented problems in sintering and breaking [24].

Even more critical to downstream syngas applications is tar removal. Tars are defined as higher-weight organics and oxygenated aromatics heavier than benzene (78 g/mol). They are produced from volatilized material after polymerization [24]. A review by Milne et al. [25] of tars produced during gasification covers different removal methods. One method uses physical removal via wet gas scrubbing of tars in a scrubbing tower for the “heavy tars” followed by a venturi scrubber for lighter tars. This setup is similar to the direct quench cooling mentioned previously, as cooling occurs as well. Tar concentration is reported to be lower than 10 ppm by volume at the exit of this setup [23]. The disadvantage of this setup is that wastewater treatment is required to dispose of the tar and can be expensive.

The other method for tar removal is catalytic or thermal conversion to non-condensable gas. This is also known as hot gas cleaning, as it occurs at temperatures at or above gasification temperatures. Catalytic conversion can occur at temperatures as low as 800°C, and thermal conversion can occur at temperatures up to 1200°C. The energy required for thermal tar cracking may not be cost-competitive because of the temperature rise required from the gasification temperature to crack the high refractory tars [24].

Alkali compounds such as calcium oxide and potassium oxide are present in biomass. When gasified, both either vaporize or concentrate in the ash. Condensation of these compounds begins at 650°C, and they can deposit on cool surfaces causing equipment clogging, equipment corrosion, and catalyst deactivation [26]. According to Stevens [26], research on alkali adsorption filters using bauxite has been promising but not demonstrated on a large scale. Stevens concludes that the best current method for alkali removal is using proven syngas cooling followed by wet scrubbing, where the addition of water cools the syngas and physically removes small particles and liquid droplets.

Wet scrubbing also removes ammonia that forms during gasification from the nitrogen in the biomass. Without proper removal, ammonia can deactivate catalysts as well. Complete ammonia removal can be accomplished through wet scrubbing [27]. For gasifiers coupled to a catalytic or thermal tar reformer, most of the ammonia can be reformed to hydrogen and nitrogen [27]. Sulfur in the biomass mostly forms hydrogen sulfide (H₂S) with small amounts of carbonyl sulfide (COS). Hydrogen sulfide is removed by three main ways: chemical solvents, physical solvents, and catalytic sorbents. For chemical removal, amine-based solvents are typically utilized to chemically bond with H₂S. Physical removal takes advantage of the high solubility of H₂S using an organic solvent. Typical setups of both chemical and physical removal involve an absorber unit followed by a solvent regenerator unit, known as a stripper. Operation usually occurs at temperatures lower than 100°C and medium to high pressures (150–500 psi) [27]. Sulfur leaving these two systems is around 1–4 ppm and can require further removal, especially for fuel synthesis. In that case, a syngas polishing step using a fixed-bed zinc oxide-activated carbon catalyst removes H₂S and COS to achieve the parts per billion levels necessary for fuel synthesis. Halides, present in trace amounts in the biomass, can also be removed with the zinc oxide catalyst [27]. The water streams with the scrubbed-out impurities are sent to wastewater treatment.

End-Use Product

After syngas is cleaned of particulates, impurities, and contaminants, there is sufficient energy content for producing a higher valued product. There are three main large-scale biomass gasification pathways that have been researched and suggested for producing higher valued products: power generation, liquid-fuel synthesis, and chemical synthesis. According to Wender [28], the three largest commercial uses for syngas are ammonia production from hydrogen, methanol synthesis, and hydrocarbon synthesis via the Fischer-Tropsch process.

Power Generation

Power is most effectively generated using gasification by combusting the syngas in a gas turbine to provide mechanical work for a generator. Steam can then be generated by recovering heat from the hot syngas, and the steam in turn is used for mechanical work via a steam turbine. This gasifier plus gas turbine and steam turbine setup is known as integrated gasification combined cycle (IGCC) power generation. The level of particulates, alkali metals, and tar can decrease the performance of the gas turbine. Consonni and Larson [29] found that particulates can cause turbine blade erosion; 99% of 10-micron or smaller particles should be removed. In addition, they also report that alkali metals corrode the turbine blades and tars condense on the turbine blades, both hindering operation and escalating turbine failure. Fortunately, nearly all alkali metals and tars can be removed using proven wet scrubbing techniques.

Using the IGCC approach to generate power, Bridgwater et al. [30] and Craig and Mann [23] expect biomass to produce power with efficiencies in the range of 35%–40% with large scale systems (greater than 100 MW net output) at the high end of the range. Moreover, Craig and Mann suggest that future advanced turbine systems could reach 50% biomass-to-power efficiency.

Synthetic Fuels and Chemicals

Instead of converting the energy content of the syngas to power, the energy content can be condensed into a liquid energy carrier, or fuel. The conversion of syngas to fuels can only occur in the presence of proper catalysts [31]. The catalytic reactions basically build up the small molecules in the syngas (i.e., carbon monoxide and hydrogen) into larger compounds that are more easily stored and transported. A summary of many catalytic pathways to fuels and chemicals is shown in Figure 7. In most catalytic synthesis reactions, syngas cleanliness requirements are very high. Most impurities and contaminants are removed to low parts-per-million and even parts-per-billion concentrations. This means that significant cost must be directed toward syngas cleaning.

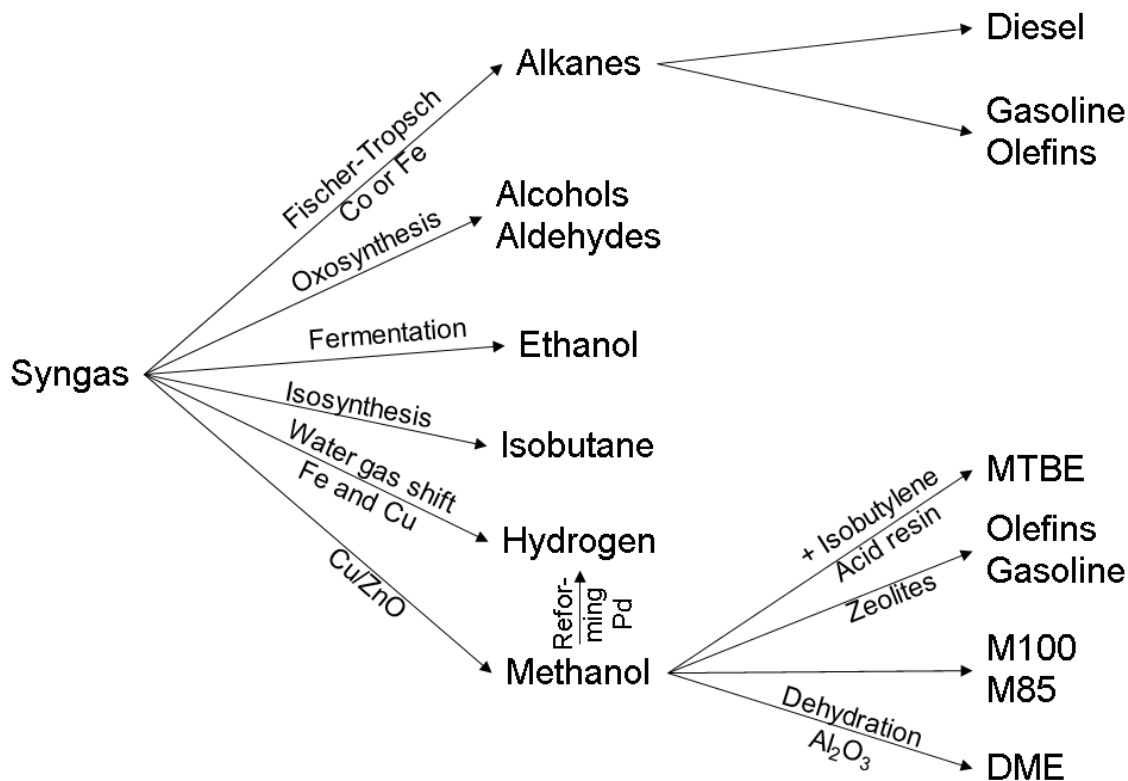


Figure 7. Main syngas conversion pathways [32]

Methanol to Gasoline

Methanol is one of the top chemicals produced in the world [32]. Most commercially produced methanol is synthesized via steam methane reforming and autothermal reforming. The synthesis of methanol from syngas is highly exothermic (equation 1). The reaction occurs over a

Cu/ZnO/Al₂O₃ catalyst at temperatures of 220°–275°C and pressures of 50–100 bar with a catalyst lifetime of 2–5 years [31]. Wender [28] reports syngas-to-methanol conversion efficiency can reach 99% with recycle, but per-pass efficiency is only about 25%.



Although methanol can be used directly as a liquid fuel, it can also be converted into the conventional transportation fuel range. This process is known as the methanol-to-gasoline (MTG) process and was developed by the Mobil Oil Corporation [31]. In that process, methanol is heated to 300°C and dehydrated over alumina catalyst at 27 atm, yielding methanol, dimethyl ether (DME), and water. The exiting mixture reacts with a zeolite catalyst at 350°C and 20 atm to produce 56% water and 44% hydrocarbons by weight. Of the hydrocarbon product, 85% is in the gasoline range, with 40% of the gasoline-range product being aromatic. However, limitations on the aromatic content of gasoline have been proposed in legislation [31]. The thermal efficiency of the MTG process is 70% [11]. The overall MTG process usually includes multiple MTG reactors in parallel to perform periodic catalyst regeneration by burning off coke deposits [11]. Mobil operated a commercial plant producing 14,500 barrels per day in New Zealand during the 1980s [32]. Alternatively, the reaction process could be stopped directly after the methanol synthesis to focus on producing DME, because DME can be used as a diesel fuel because it has a high cetane number. DME is formed from the dehydration reaction of methanol over an acid catalyst γ -alumina. Per-pass efficiency can be as high as 50%. Overall, syngas-to-DME production efficiency is higher than that for syngas to methanol [31]. However, DME is in gaseous form at atmospheric conditions and needs to be pressurized for use in diesel engines [33]. Therefore, engine modification is required; this is the main disadvantage for DME use as transportation fuel.

Fischer-Tropsch

Fischer-Tropsch (FT) catalytic synthesis is a highly exothermic reaction producing a wide variety of alkanes (equation 2).



For gasoline-range products, higher temperatures (300°–350°C) and iron catalysts are typically used. For diesel-range and wax products, lower temperatures (200°–240°C) and cobalt catalysts are typically used [34]. Operating pressures are in the range of 10–40 bar. Product distribution can be estimated using the Anderson-Schulz-Flory chain-growth-probability model, in which longer hydrocarbon chains form as the temperature decreases. At high temperatures, selectivity favors methane and light gases. This is a disadvantage if liquid fuel production is the focus. At low temperatures, selectivity favors long-carbon-chain wax products requiring further hydrocracking to the diesel range in a separate unit, which adds more construction cost but is necessary for liquid fuel production.

Because of the highly exothermic reaction, the heat must be removed or the catalyst can be deactivated. Two main types of reactors have been designed: a fixed-bed tubular reactor and a

slurry-phase reactor (Figure 8). Heat removal is crucial to the process and has been the focus of reactor design development [31]. The fixed-bed reactor has many catalyst tubes with heat removal achieved by steam generation on the outside of the tubes [35]. The fixed-bed reactor is simple to operate and is well suited for wax production due to simple liquid/wax removal. However, it is more expensive to build because of the many tubes and has a high pressure drop across the reactor [36]. The slurry-phase reactor (SPR) operates by suspending catalyst in a liquid and bubbling the syngas through from the bottom. Disadvantages of the SPR are more complex operation and difficult wax removal. However, SPRs cost approximately 40% less to build than fixed-bed reactors [36].

FT diesel is very low in sulfur, low in aromatic content, and has a high cetane number, making it very attractive as a conventional fuel alternative. Emissions across the board decrease when using FT diesel. A South Africa-based company, Sasol, has been producing transportation fuel since 1955 using the FT process and supplies 41% of South Africa's transportation fuel requirements [31].

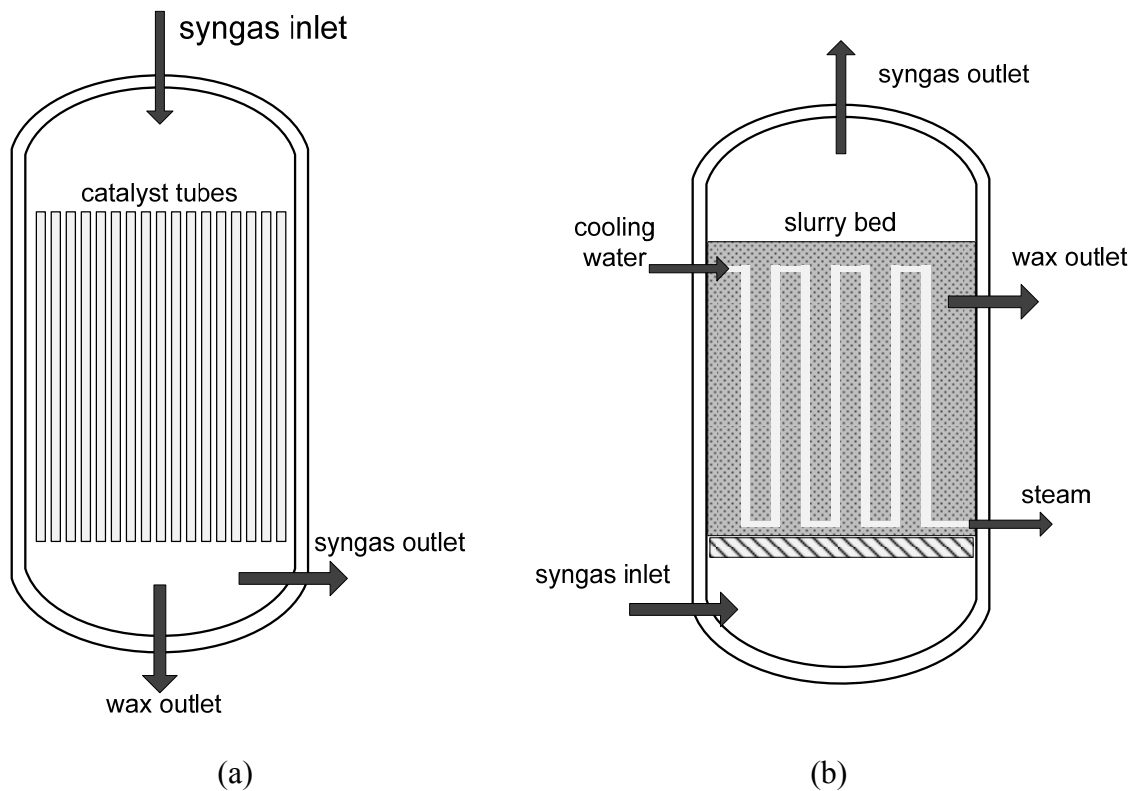


Figure 8. Fischer-Tropsch reactor types (a) multi-tubular fixed bed and (b) slurry bed [31]

Techno-Economic Analysis

For biofuels technologies to be utilized in commercial applications, the economic feasibility must be determined. A feasibility analysis is also called a techno-economic analysis, in which the technical aspects of a project are coupled to the economic aspects. First, the basic theoretical configuration is developed and a mass and energy balance is performed. Second, cost estimation

allows the investment and production cost of a biorefinery to be determined. With rising interest in biorenewable resources, many techno-economic studies have been performed on power generation and biofuel scenarios. These studies assist in understanding how the physical process relates to the cost of producing renewable alternatives. Accuracy of results from these studies is usually $\pm 30\%$ of the actual cost [5].

Previous studies of gasification-based biomass-to-liquid production plants estimate the cost of transportation fuels to range from \$12/GJ to \$16/GJ (\$1.60–\$2.00 per gallon of gasoline equivalent) [16, 37-40]. The same studies estimate total capital investment in the range of \$191 million for 2,000 dry metric ton per day (tpd) input [39] to \$541 million for 4,500 dry metric tpd input [38].

A 1,650 dry metric tpd biomass-to-methanol plant based on gasification, with a production cost of \$15/GJ (\$0.90/gal of methanol), is reported by Williams et al. [16] in 1991\$ for \$45/dry metric ton of biomass. Williams et al. also shows the production cost of methanol-derived natural gas to be \$10/GJ (\$0.60/gal of methanol). However, that study concludes that if a carbon tax system was developed for lifecycle carbon emissions, then renewable methanol could become competitive with natural-gas-derived methanol at a tax of approximately \$90 per metric ton of carbon. A more recent study by Larson et al. of switchgrass-to-hydrocarbons production in 2009 reports a production cost of \$15.3/GJ (\$1.90/gal of gasoline) in 2003\$ for a 4,540 dry metric tpd (5,000 dry short tpd) plant based on gasification [38].

Table 2 compares four biofuel production studies based on gasification. A range of cost year, plant size, and feedstock cost show the diversity of characteristics and assumptions that techno-economic studies use. In addition, resulting capital investment costs of the studies have a large range. For example, the capital investment costs of the Phillips et al. [39] and Tijmensen et al. [40] studies are \$191 million and \$387 million, respectively, at similar plant sizes. Reasons for such a significant difference are choice of technologies and level of technology development. The Phillips et al. study is a target study, meaning that it assumes future technology improvement and therefore lower costs. Direct comparison is difficult because of the varying assumptions used by each study.

Table 2. Previous Techno-Economic Studies of Biomass-Gasification Biofuel Production Plants

	Williams et al. [16]	Phillips et al. [39]	Tijmensen et al. [40]	Larson et al. [38]
Cost Year	1991	2005	2000	2003
Plant Size (dry metric ton per day)	1,650	2,000	1,741	4,540
Feedstock	Generic biomass	Poplar	Poplar	Switchgrass
Fuel Output	Methanol	Ethanol	FT liquids	Diesel, gasoline
Feedstock Cost (\$/dry short ton)	41	35	33	46
Capital Investment (\$MM)	N/A	191	387	541
Product Value (\$/GJ)	15	12	16	15
Product Value (\$/GGE)	1.90	1.60	2.00	1.85

Methodology

The following steps were undertaken to perform the analysis in this study:

- Collect performance information on relevant technologies for systems under evaluation
- Perform down-selection process with developed criteria to identify most appropriate scenarios
- Design process models using Aspen Plus process engineering software
- Size and cost equipment using Aspen Icarus Process Evaluator software, literature references, and experimental data
- Determine capital investments and perform discounted cash flow analysis
- Perform sensitivity analysis on process and economic parameters
- Perform pioneer plant cost growth and performance analysis.

Down-Selection Process

A number of process configurations for the gasification-based, biomass-to-liquids (BTL) route were initially considered. These configurations are listed in Table 3 and discussed in the following sections.

Table 3. Process Configurations Considered in Down Selection Process

Gasifier Block	Entrained flow, slagging gasifier
	Fluid bed, dry ash gasifier
	Transport gasifier, dry ash (e.g., Kellog, Brown, and Root)
	Indirect gasifier, dry ash (e.g., Battelle-Columbus Labs)
Syngas Cleaning	Water scrubbing
	Catalytic tar conversion/reduction
	Thermal tar conversion/reduction
	Amine-based acid gas removal
	Physical sorbent-based acid gas removal (e.g., Selexol, Rectisol)
Fuel Synthesis	Fischer-Tropsch
	Mixed alcohols
	Methanol to gasoline (MTG)
	Dimethyl ether
	Syngas fermentation

Preliminary Criteria

The initial technology configuration options are reviewed and screened in accordance with three criteria.

First, the technology under consideration should be commercially ready in the next 5–8 years and preferably with high technology development. High technology development increases the likelihood that a configuration will perform at the scale defined in this study. For example, coal gasification has been demonstrated commercially at large scales [11]. While similar scale biomass gasifiers have not been proven commercially, the technology development of coal is assumed to apply for biomass in 5–8 years.

Second, the size of biorefinery should be feasible with typical agricultural productivity and within a realistic collection area. For example, if one-third of total land use surrounding the biorefinery is for stover collection and each acre provides conservatively 1 short dry ton per year, then the required collection radius is 35 miles and the amount of biomass transported to the biorefinery is approximately 2,300 short tons (2,090 metric tons) per day. The collection area with a 35-mile radius is assumed to be realistic. In addition, previous studies by Tijmensen et al., Phillips et al., and Lau et al. have used similar plant sizes [39-41].

Third, the desired product should be compatible with the present transportation fuel infrastructure, i.e., gasoline and diesel-range hydrocarbons.

Scenario Selection

For the gasification area, two gasifiers are selected for modeling. First, an entrained-flow, slagging gasifier is chosen because of its commercial application with coal (GE, Siemens, Shell, and ConocoPhillips) and its potential for use with biomass. Moreover, process modeling of this gasifier is simple because it can be closely approximated at thermodynamic equilibrium [3]. Second, a fluidized-bed, dry-ash gasifier is chosen because of experience at the Gas Technology Institute and because of data availability. Also, a report by Bain [4] at the National Renewable Energy Laboratory contains collected and analyzed data for fluidized-bed gasification. In addition, Iowa State University is currently operating an atmospheric-pressure fluidized-bed gasifier as either air or oxygen/steam fed.

The syngas cleaning area is chosen to include configurations that have less technological complexity than previous studies. Phillips et al. [39] and Larson et al. [38] both employ an external catalytic tar-reforming process for dry-ash gasification. Because of low technological development in tar conversion and its inherent complexity, a direct-contact syngas quenching and scrubbing system is instead chosen for this study. In the case of the slagging gasifier, high temperatures inhibit tar formation, yet quenching and particulate and ammonia removal are still required. An amine-based chemical absorber/stripper configuration is chosen for removal of hydrogen sulfide and carbon dioxide. This configuration is chosen due to data availability, as compared to proprietary physical gas cleaning process such as Rectisol and Selexol.

Two fuel synthesis configurations under consideration produce liquid hydrocarbons: Fischer-Tropsch (FT) synthesis and MTG. FT synthesis has been proven in operation at commercial scale for many years by Sasol [11]. Because of more accessible data and long industrial experience, FT synthesis is the only fuel synthesis option chosen. Because of this selection, a

post-synthesis fuel upgrading area is also necessary, as FT products need to be separated and hydroprocessed.

Scenarios Not Selected

The indirect, dry-ash gasifier and the mixed alcohol synthesis configurations are not considered due to previous work by Phillips et al. [39]. The transport gasifier design, though a promising technology, is not considered because of reactor complexity, unproven commercial-scale operation, and lack of public domain data. Tar conversion via external thermal or catalytic cracking is not considered due to lack of public domain data and commercial scale experience. Acid gas removal using proprietary technology (e.g., Rectisol or Selexol) is not considered because of a lack of public operational data. MTG, including methanol synthesis, is not considered because of time constraints and limited operational data. DME and syngas fermentation are not considered because of limited commercial scale experience and incompatibility with present fuel infrastructure.

Project Assumptions

The main project assumptions for process and economic analysis are listed in Table 4. A more extensive list can be found in Appendix A. The process design is assumed to incorporate an “nth plant” level of implementation experience. This is a significant assumption for a process concept that has yet to undergo detailed engineering development and see its first commercial application. The design and operation of an nth plant is likely to diverge from the design and operation considered in this study. However, the nth plant evaluation is chosen to provide an analysis similar to studies completed by other groups, including NREL and Pacific Northwest National Laboratory.

Table 4. Main Assumptions Used in nth Plant Scenarios

Main Assumptions
The plant is modeled as n th plant
Plant capacity is 2,000 dry metric ton/day
Feedstock is corn stover at 25% moisture
Feedstock ash content at 6%
Feedstock is purchased at plant gate for \$75/dry short ton
All financial values are adjusted to 2007 cost year
Plant is 100% equity financed
Fuel PV is evaluated at 10% internal rate of return
Plant initiates operation in 5–8 year time frame
Plant life is 20 years
Plant availability is 310 days per year (85%)

Process Description

High-Temperature Scenario Overview

The high-temperature (HT) or slagging scenario is a 2,000 dry metric ton (2,205 dry short ton) per day corn stover-fed gasification biorefinery that produces naphtha-range and distillate-range liquid fractions to be used as blend stock for gasoline and diesel, respectively, as well as electricity for export. It is based on pressurized, oxygen-blown, entrained-flow gasification. The HT scenario is an nth plant design, meaning significant design, engineering, and operating experience has been achieved.

Figure 9 shows seven main areas of operation for the HT scenario. Feedstock preprocessing (Area 100) is where the stover is chopped, dried, and ground to 1-mm size and 10% moisture. Gasification (Area 200) contains the stover pressurization for solids feeding, gasification, and slag removal. Synthesis gas cleaning (Area 300) contains cold gas cleaning technologies where the syngas is quenched and scrubbed of particulates, ammonia, hydrogen sulfide, and carbon dioxide. Area 300 also contains the water-gas-shift reaction, which occurs before the hydrogen sulfide and carbon dioxide removal. This adjusts the ratio of hydrogen to carbon monoxide for optimal fuel synthesis. Fuel synthesis (Area 400) contains syngas boost pressurization, contaminant polishing via zinc oxide guard beds, Fischer-Tropsch reactor, and hydrocarbon gas/liquid separation. Hydroprocessing (Area 500) produces the final fuel blend and is treated as a black box utilizing published data. Power generation (Area 600) contains gas and steam turbines along with a heat recovery steam generator. Area 700 contains the air separation unit (ASU) where oxygen is separated from air and pressurized for use in the gasifier. For cost analysis uses only, a balance-of-plant area includes the cooling tower area, cooling water system, waste solids and liquids handling area, and feed water system. Detailed process flow diagrams can be found in Appendix D and detailed stream data can be found in Appendix E.

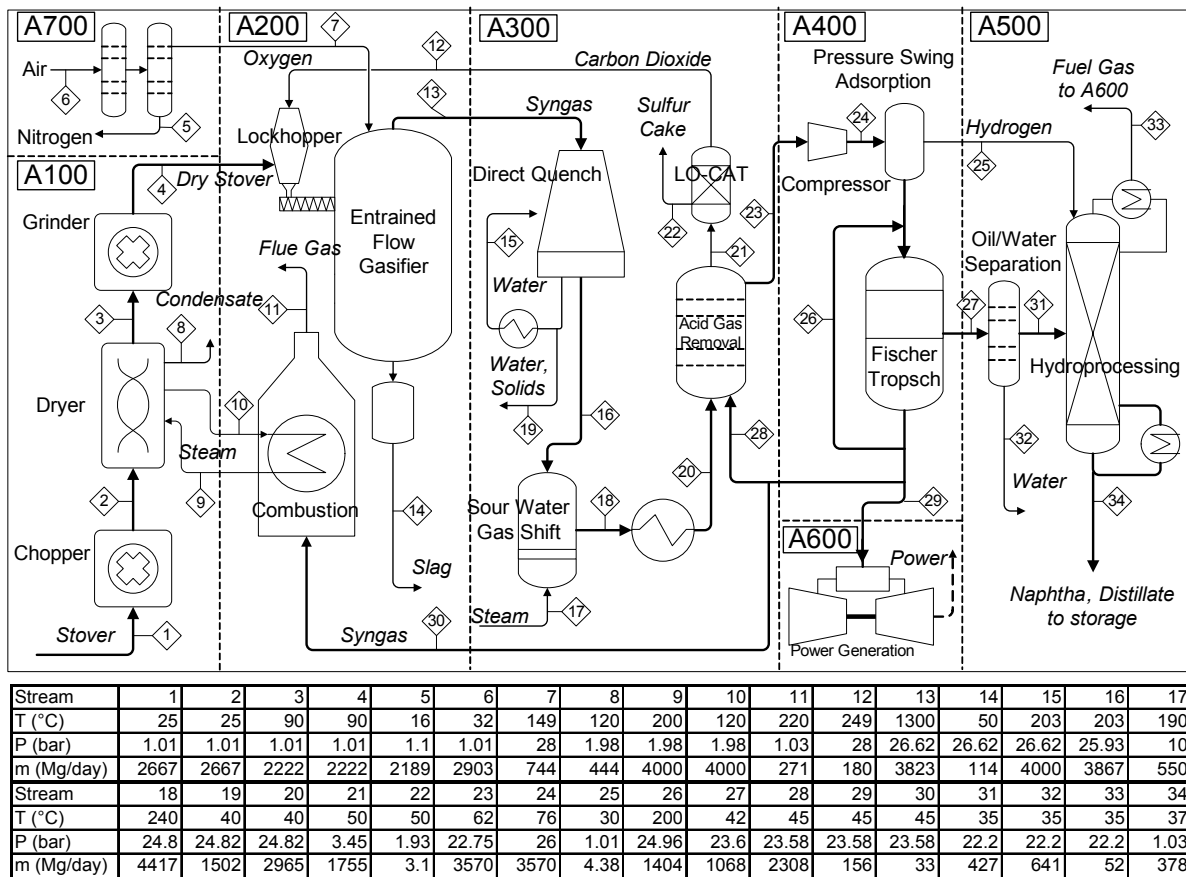


Figure 9. Overall process flow diagram for HT scenario (parallelograms enclosing numbers in the diagram designate individual process streams, which are detailed in the accompanying table)

Recycle streams are utilized to provide better syngas-to-FT-products conversion. Unconverted syngas in the fuel synthesis area is recycled to the syngas cleaning area to remove carbon dioxide and allow for further conversion in the Fischer-Tropsch reactor. A small portion of unconverted syngas is sent to a steam boiler to raise steam required for drying the biomass. The balance of unconverted syngas is combusted in a gas turbine and waste heat is recovered in a steam generator for steam turbine power. Power generated is used throughout the plant and excess is sold.

Some of the largest consumers of power are the ASU and hydroprocessing area at 11.6 MW and 2.2 MW, respectively. Another consumer of power is the hammer mill for grinding the dried biomass in Area 100; it requires 3.0 MW. The amine/water solution recirculation pump in Area 300 requires approximately 0.9 MW. Syngas compressors throughout the plant require a significant amount of power as well. Gross plant power production is 48.6 MW and net electricity for export is 13.8 MW.

Low-Temperature Scenario Overview

The low-temperature (LT) or dry-ash scenario is a 2,000 dry metric ton (2,205 dry short ton) per day corn stover-fed gasification biorefinery that produces naphtha-range and distillate-range

liquid fractions to be used as blend stock for gasoline and diesel, respectively, as well as electricity for export. It is based on a pressurized, oxygen/steam-blown fluidized-bed gasifier developed by Gas Technology Institute. The LT scenario is an nth plant design, meaning significant design, engineering, and operating experience has been achieved.

Figure 10 shows the seven main areas of operation for the LT scenario. Feedstock preprocessing (Area 100) is where the stover is chopped, dried, and ground to 6-mm size and 10% moisture. Gasification (Area 200) contains the stover pressurization for solids feeding, gasification, and char and ash removal. Synthesis gas cleaning (Area 300) contains cold gas cleaning technologies where the syngas is quenched and scrubbed of particulates, ammonia, hydrogen sulfide, and carbon dioxide. Fuel synthesis (Area 400) contains syngas boost pressurization, contaminant polishing via zinc oxide beds, Fischer-Tropsch reactor, and hydrocarbon gas/liquid separation. Also included within Area 400 is the steam methane reformer (SMR) to reduce methane content and water-gas-shift (WGS) to adjust the ratio of hydrogen and carbon monoxide.

Hydroprocessing (Area 500) produces the final fuel blend and is treated as a black box utilizing published data. Power generation (Area 600) contains gas and steam turbines along with a heat recovery steam generator. Area 700 contains the ASU that separates oxygen from air and pressurizes it for use in the gasifier. Detailed process flow diagrams can be found in Appendix D and detailed stream data can be found in Appendix E.

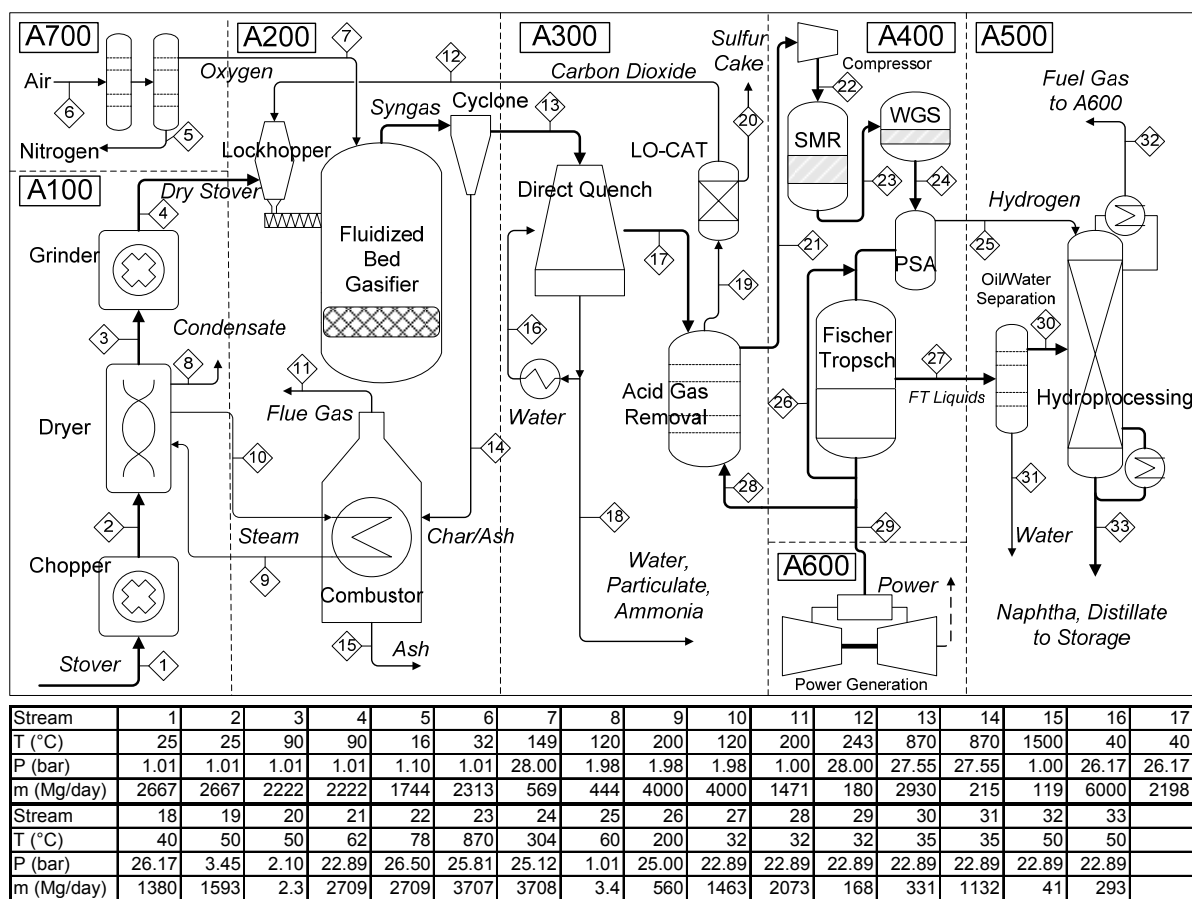


Figure 10. Overall process flow diagram for LT scenario (parallelograms enclosing numbers in the diagram designate individual process streams, which are detailed in the accompanying table)

Recycle streams are utilized to provide better FT products conversion. Unconverted syngas in the fuel synthesis area is recycled to the syngas cleaning area to remove carbon dioxide and allow for further conversion in the Fischer-Tropsch reactor. The balance of unconverted syngas is combusted in a gas turbine and waste heat is recovered in a steam generator for steam turbine power. Power generated is used throughout the plant and excess is sold. Unconverted carbon within the gasifier in the form of char is collected and combusted in a furnace to produce heat, thereby generating steam for the drying of the biomass.

Some of the largest consumers of power are the ASU and hydroprocessing area at 9.1 MW and 1.7 MW, respectively. Another consumer of power is the hammer mill for grinding the dried biomass in Area 100; it requires 1.1 MW. The amine/water solution recirculation pump in Area 300 requires approximately 0.7 MW. Syngas compressors throughout the plant require a significant amount of power as well. Gross plant power production is 40.7 MW and net electricity for export is 16.3 MW.

Area 100 Preprocessing

The preprocessing area contains all the unit operations required for preparing the biomass for feeding into the gasifier. Biomass enters the plant gate at 25 wt% moisture on wet basis in bales. The corn stover composition is shown below in Table 5. Ash content is assumed to be 6% by

weight. The composition of char formed in the gasifier is also shown in Table 5. Forklifts transport the bales to conveyors where the stover is separated from any metal in a magnetic separator. The first modeled operational area is a primary biomass chopper to complete the initial size reduction step and prepare stover for drying.

Table 5. Stover and Char Elemental Composition (wt %)

Element	Stover	Char
Ash	6.00	0
Carbon	47.28	68.05
Hydrogen	5.06	3.16
Nitrogen	0.80	0.29
Chlorine	0	0
Sulfur	0.22	0.15
Oxygen	40.63	28.34

The next area of operation is the direct-contact steam drying, which is modeled as a rotary steam dryer with exiting biomass moisture of 10% on wet basis. For steam dryers, Amos [21] suggests a 9:1 ratio of steam to evaporated moisture. Therefore, 4,000 metric tpd steam is utilized in a loop and heated to 200°C from the hot combustion flue gases exiting the syngas- or char-fired combustor in Area 200. Steam mixes with 25°C biomass and enters the dryer. At the exit, steam at 120°C returns to the combustor for reheating, and dried biomass exits at 90°C and is conveyed to the grinding area.

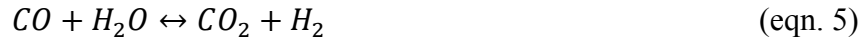
The grinding area has the same configuration as the chopping area except that the grinder requires significantly more power because of the larger size reduction. The grinder reduces the size of the biomass to 1-mm and 6-mm particles for the HT and LT scenarios, respectively. The power requirements of the grinder for the HT and LT scenarios are 3,000 kW and 1,100 kW, respectively. Energy requirements for grinding are determined using the correlations for specific energy (kWh per short ton), which is adapted from Mani et al. [42].

Area 200 Gasification

The gasification area of the plant produces synthesis gas using pressurized gasifiers. Also in this area, slag, char, and ash are removed. This area also includes lockhoppers for biomass pressurization and a fired combustor that provides heat to raise steam for drying the stover.

Dried and ground stover enters the area and is immediately conveyed to a lockhopper system for pressurized feeding. Carbon dioxide from the syngas cleaning area is used as pressurization gas. According to Lau et al. [41], a lockhopper system is the best setup for pressurized feeding of solids, despite higher operating costs due to high inert gas usage. A proven track record with biomass is the main reason for their recommendation. The power requirement of a lockhopper system using biomass is 0.082 kW/metric tpd, resulting in a 180-kW system. Higman and van der Burgt [43] report inert gas usage as 0.09 kg/kg for 25-bar applications. This results in a 180 metric ton per day carbon dioxide addition into the hopper. It is assumed that only 5% of the inert gas leaks into the gasifier while the rest is vented by the lockhopper.

Pressurized biomass is then conveyed into the gasifier. Oxygen at 95% purity is produced from the air separation unit. A fixed 0.35 mass ratio of oxygen to biomass is used for the entrained-flow gasifier as reported by Henrich [18]. Steam addition to the gasifier is set at 0.48 mass ratio of steam to biomass in accordance with Probst and Hicks [11] and explained further in Appendix C. This gasifier operates at a temperature of 1,300°C, meaning that equilibrium can be modeled according to Frey and Akunuri [3]. The reactions shown in equations 3–9 are modeled using equilibrium constants.



The LT scenario gasifier uses a 0.26 mass ratio of oxygen to biomass at a gasification temperature of 870°C. This ratio is developed from the data found in an IGT gasifier study by Bain [4]. In that study, Bain develops mass balances for an IGT gasifier operating with woody biomass. Steam addition to the gasifier is calculated using a 40/60 steam to oxygen mass ratio, consistent with experiments performed at Iowa State University using corn stover feedstock and a steam/oxygen-blown fluidized-bed gasifier. Low-temperature gasification cannot be modeled at equilibrium with or without approach temperatures for reactions. Instead, an elemental mass balance calculation and adjustment is performed to ensure all inlet and outlet streams are accounted for across the gasifier. For details on the LT gasifier mass balance calculation, see Appendix C.

Yield from each gasifier is different. As Table 6 shows, hydrocarbons and tars are not produced in the high-temperature gasifier because of near equilibrium conditions. Also, more hydrogen formation occurs in the high-temperature gasifier as a result of the water-gas-shift reaction (equation 5) and because thermodynamically, nearly no methane, ethane, and ethylene are produced. The low-temperature gasifier, on the other hand, produces a significant amount of methane, ethane, and ethylene in the syngas, requiring downstream reforming. In the HT scenario, slag forms from the ash when the ash melts and flows on the inside walls. It is then collected at the bottom and removed for storage and subsequent waste removal. In accordance with Frey and Akunuri [3], it is assumed that 95% of the ash in the stover becomes slag while the rest becomes fly ash.

Table 6. Syngas Composition (Mole Basis) Leaving Gasifier for Gasification Scenarios Evaluated

Component	High Temperature (mole fraction)	Low Temperature (mole fraction)
Carbon Monoxide	0.264	0.240
Hydrogen	0.310	0.200
Carbon Dioxide	0.137	0.274
Water	0.280	0.194
Nitrogen	0.002	0
Methane	6×10^{-6}	0.055
Ethane	0	0.0061
Ethylene	0	0.013
Ammonia	3.1×10^{-5}	0.0094
Hydrogen Sulfide	6.72×10^{-4}	0.00112
Carbonyl Sulfide	2.6×10^{-5}	0
Tar (Anthracene)	0	5.00×10^{-4}
Oxygen	0	0
Argon	0.006	0.006

Directly after the low-temperature gasifier, initial syngas cleaning by cyclones captures char and ash. The cyclones are split into two trains because of high volumetric gas flow. Each train contains a medium efficiency cyclone, followed by high efficiency cyclones to capture particulates. Overall particulate removal efficiency for the cyclone area is 99%. Nearly particulate-free syngas travels to the more rigorous syngas cleaning in Area 300. Captured char in the LT scenario is collected and combusted in a fluidized-bed combustor providing energy for heating low-pressure steam, which is used for drying the stover. Syngas produced in the HT scenario contains fly ash, which is subsequently removed in a direct-water-quench unit. The combustion area in the HT scenario receives unconverted syngas from the fuel synthesis area, as char is not produced. For both scenarios, the combustor is assumed to operate adiabatically, resulting in an exit flue gas temperature of approximately 1,800°C. Hot flue gas heats 120°C steam to 200°C and loops to the stover drying area.

Area 300 Syngas Cleaning

After the initial particulate removal accomplished by the cyclones, the syngas still contains some particulates and all of the ammonia, hydrogen sulfide, and other contaminants. Area 300 removes these species using a cold-gas-cleaning approach, which is presently proven in many commercial configurations. Hydrogen sulfide and carbon dioxide, collectively known as acid gas, are absorbed via amine scrubbing. The LO-CAT hydrogen sulfide oxidation process is used to separate carbon dioxide from hydrogen sulfide with subsequent recovery of solid sulfur. In addition, the HT scenario contains a sour water-gas-shift (SWGS) process (sour because of the presence of sulfur), whereas the LT scenario situates the water-gas-shift (WGS) process directly upstream from the Fischer-Tropsch reactor.

Because of a less than optimal hydrogen to carbon monoxide ratio from the gasifier, a WGS reaction is necessary at some point in the process to adjust to the optimum Fischer-Tropsch ratio of 2.1. Therefore, a significant WGS activity is required, so a sizable amount of carbon dioxide is produced. To keep that carbon dioxide from building up in downstream processes, the SWGS

reactor is located before the acid gas removal area. This SWGS unit operation is the most significant difference between the HT and LT scenarios in this area.

In the HT scenario, the syngas arriving from the gasifier is cooled by direct-contact water quench to the operating temperature of the SWGS unit. In addition to cooling, the direct water quench removes all of the fly ash, sludge, and black water to prevent downstream plugging. At this point, a portion of the syngas is diverted to the SWGS unit, which is modeled at equilibrium conditions and has an exit gas temperature of 300°C. A ratio of 3:1 water to carbon monoxide is reached by addition of steam to the SWGS reactor. After the syngas is combined, the gas is further cooled to prepare for the acid gas removal. In the LT scenario, the direct quench unit condenses the syngas, removing approximately 90% of ammonia and 99% of solids. Tar is condensed in this unit and can be recycled back into the gasifier using a slurry pump, but this configuration is not modeled. A water treatment facility for the direct-quench effluent is not modeled but is accounted for in a balance-of-plant cost.

The next step for cleanup is the removal of acid gas (carbon dioxide and hydrogen sulfide) through the use of an amine-based solvent in a chemical gas absorption system. At this point in the cleaning process, hydrogen sulfide and carbon dioxide content is approximately 900 ppm and 30% on a molar basis, respectively. Sulfur must be removed to at least 0.2 ppm for Fischer-Tropsch synthesis [31]. According to the Gas Processors Suppliers Association *Engineering Data Book* [44], amine-based systems are only capable of removing sulfur down to 4 ppm. Therefore, a zinc oxide guard bed is required to remove the difference. In this study, 20% concentrated monoethanolamine (MEA), capable of absorbing 0.4 mole acid gas per mole amine, is used as the absorbent. The process setup is based on a report by Nexant Inc. [27]. Hydrogen sulfide leaves the top of the absorber at 4 ppm and CO₂ leaves at 2%, which is 99% and 90% removal, respectively. The clean syngas is now ready for polishing to final cleanliness requirements. A stripper is utilized to desorb the acid gas and regenerate the amine solution. Before the acid gas and amine solution enters the stripper, a heat exchanger raises the temperature to 90°C.

Acid gas is brought to the LO-CAT sulfur recovery system to isolate hydrogen sulfide and convert it to solid sulfur. The LO-CAT system sold and owned by Gas Technology Products uses oxygen and a liquid solution of ferric iron to oxidize hydrogen sulfide to elemental solid sulfur [45]. This system is suitable for a range of 150 lb to 20 tons per day sulfur recovery and 100 ppm to 10% H₂S concentration in sour gas, as reported by Nexant Inc. [27]. The sulfur production in this model is approximately 3 metric tpd with H₂S concentration of approximately 150 ppm, which is within the reported ranges. First, the H₂S is absorbed/oxidized, forming solid sulfur and water, while the ferric iron converts to ferrous iron. The second vessel oxidizes the ferrous iron back to ferric iron and the sulfur cake is removed while the iron solution is recycled back into the absorber [46]. The carbon dioxide gas stream from the absorber is split, and a portion is compressed and used in biomass pressurization while the rest is vented to the atmosphere.

Area 400 Fuel Synthesis

Conversion from syngas to liquid fuel occurs in Area 400, the fuel synthesis area. The major operations in this area are zinc oxide/activated carbon gas polishing, steam methane reforming (SMR) (only in the LT scenario), water-gas-shift (only in the LT scenario), Fischer-Tropsch (FT) synthesis, hydrogen separation via pressure swing adsorption (PSA), FT products separation, and unconverted syngas distribution. Another major difference between the LT and HT scenarios is in this area. Area 400 in the LT scenario contains the water-gas-shift reaction and steam methane reformer, as recycle streams contain a high enough content of methane and ethylene to significantly accumulate and cause dilution.

A compressor is the first operation in Area 400, boosting the pressure to 25 bar for FT synthesis. Then the syngas is heated to 200°C and passes through zinc oxide/activated carbon fixed-bed sorbent. This polishing guard bed acts as a barrier to any upstream non-normal contaminant concentrations and provides sulfur removal down to synthesis requirements. To limit downstream catalyst poisoning, the syngas stream must be cleaned of these components. Removal to 50 ppb sulfur is possible with zinc oxide sorbent [27]. To comply with reported requirements, the sorbent removes sulfur to approximately 200 ppb. In addition to sulfur, halides are removed by the sorbent. Syngas contaminant level requirements for Fischer-Tropsch synthesis are shown in Table 7.

Table 7. Fischer-Tropsch Gas Cleanliness Requirements [31]

Contaminant	Tolerance Level
Sulfur	0.2 ppm (200 ppb)
Ammonia	10 ppm
HCN	10 ppb
Halides	10 ppb

Methane, nitrogen, and carbon dioxide act as inert gases in the FT synthesis. At this point in the LT scenario, an SMR step is utilized. Syngas is heated to 870°C through a fired heater and passed through a reformer nickel-based catalyst to reduce methane, ethylene, and ethane content. It is assumed that the SMR can be modeled to operate at equilibrium. Steam is added to bring the steam to methane ratio to approximately 6.0, which at 870°C and 26 bar results in about 1.5% equilibrium methane content in the exit stream [47]. For the HT scenario, the SMR step is not necessary. The WGS reaction is now employed for the LT scenario to increase the H₂:CO ratio. A portion of the gas is diverted through the fixed catalyst bed while the rest bypasses the reactor, similar to the SWGS unit in the HT scenario.

The exiting H₂:CO ratio after WGS is slightly above 2.1 in order for the excess hydrogen to be separated and used in the hydroprocessing area. A PSA process is employed to isolate a stream of hydrogen. Because only a small amount of hydrogen needs to be separated from the syngas stream for downstream use, a small percentage of the syngas is directed to the PSA unit. Hydrogen removal efficiency within the PSA unit is assumed to be 85%, and the unit produces pure hydrogen [41]. After the PSA, the syngas rejoins the main gas line and enters the FT reactor.

The Fischer-Tropsch synthesis reactor operates at 200°C and 25 bar using a cobalt catalyst according to equation 10. Per-pass carbon monoxide conversion in the reactor is set at 40%. The product distribution follows the Anderson-Schulz-Flory alpha distribution where chain growth factor, α , depends on partial pressures of H₂ and CO and the temperature of the reactor, as reported by Song et al. [48] for cobalt catalyst and shown in equation 11, where y is the molar fraction of carbon monoxide or hydrogen and $Temp$ is the reactor operating temperature in kelvin. The reactor is based on a fixed-bed type reactor and that choice is reflected by the low per-pass CO conversion.



$$\alpha = \left[0.2332 \cdot \frac{y_{CO}}{y_{CO} + y_{H_2}} + 0.6330 \right] \cdot [1 - 0.0039(Temp - 533)] \quad (\text{eqn. 11})$$

To ensure that the hydrocarbon product distribution leans toward the production of diesel fuel, the value of alpha should be at least 0.85 and preferably greater than 0.9, as shown in Figure 11. The reactor operating temperature needed to achieve a chain growth value of 0.9 is approximately 200°C. This produces 30 wt% wax in the FT products, requiring hydrocracking before addition to the final fuel blend. All exiting effluent is cooled to 35°C and the liquid water and hydrocarbons are separated in a gas/liquid knock-out separator. Unconverted syngas is split into four streams: direct recycle to the FT reactor, recycle to the acid gas removal area, purge to the combustor in Area 200, and to the gas turbine in the power generation area. The LT scenario does not include sending a syngas stream to the combustor in Area 200 because char is used. Overall CO conversion is 66% due to recycling syngas. The recycle ratio is approximately 1.95 for both scenarios.

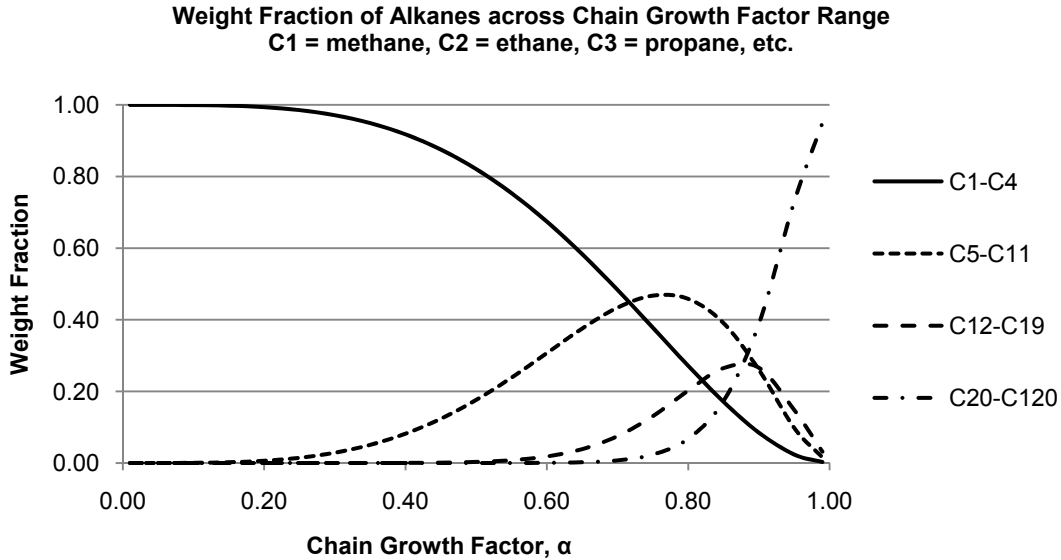


Figure 11. Fischer-Tropsch product distribution as a function of chain growth factor (α) using equation 11 [48]

Area 500 Hydroprocessing

FT products from the fuel synthesis area contain significant amounts of high-molecular-weight wax. Hydrogen is required to crack these high-molecular-weight paraffins to low-molecular-weight hydrocarbons. A product distribution is specified in Table 8, as detailed in Shah et al. [49]. It is assumed that the hydroprocessing area contains a hydrocracker for converting the wax fraction and a distillation section for separating naphtha, diesel, and lighter-molecular-weight hydrocarbons. Also, hydrogen is assumed to be recycled within this area as needed. Methane and propane are separated and used to fuel the gas turbine in the power generation area. The hydroprocessing area is modeled as a “black box.”

Table 8. Hydroprocessing Product Distribution [49]

Component	Mass Fraction
Methane	0.0346
LPG (propane)	0.0877
Gasoline (octane)	0.2610
Diesel (hexadecane)	0.6167

Area 600 Power Generation

A gas turbine and steam turbine provide the means to produce the power that is required throughout the plant and the excess power that is generated for export. Unconverted syngas from Fisher-Tropsch synthesis and fuel gas from hydroprocessing are combusted in a gas turbine, producing hot flue gas and shaft work. The flue gas exchanges heat with water in a heat-recovery steam generator to produce steam for the steam turbines, which subsequently produce more shaft work. Electric generators attached to both the gas turbine and the steam turbine produce electricity from the shaft work.

Area 700 Air Separation

Because 95% purity oxygen is used for both scenarios, a cryogenic ASU is employed rather than purchasing oxygen. A two-column cryogenic oxygen/nitrogen separation system is employed with subsequent oxygen compression and nitrogen vent. Air pre-cooling is accomplished by exchanging heat with exiting nitrogen. This area requires a significant amount of power, as explained in the results section, which is provided by the power generation area.

Methodology for Economic Analysis

Capital investment and PV of each scenario are determined by finding all equipment costs and operating costs for the construction and operation of a plant for 20 years. Total capital investment is based on the total equipment cost plus additional installation costs and indirect costs (such as engineering, construction, and contingency costs). Annual operating costs are determined and a discounted cash flow rate of return analysis is developed. PV per unit volume of fuel is determined at a net present value of zero and a 10% internal rate of return. The major economic assumptions used in this analysis are listed in Table 9. A detailed list of assumptions can be found in Appendix A.

Table 9. Main Economic Assumptions for nth Plant Scenarios

Parameter	Assumption
Financing	100% equity
Internal rate of return (after taxes)	10%
General plant depreciation period	7 years (all areas except Area 600)
Steam plant depreciation period	20 years (Area 600 only)
Construction period	2.5 years with total capital investment spent at 8%, 60%, and 32% per year during years before operation
Startup time	0.5 years, where during that time revenues, variable operating costs, and fixed operating costs are 50%, 75%, and 100% of normal, respectively.
Income tax rate	39%
Contingency	20% of fixed capital investment
Electricity cost	5.4 cents/kWh
Working capital	15% of fixed capital investment
Land purchase	6% of total purchased equipment cost
Plant availability	310 days per year (85%)

Unit operations from the scenarios are sized and costs estimated using Aspen Icarus Process Evaluator software based on the Aspen Plus simulation data. Unique equipment costs for such equipment as the gasifier and Fischer-Tropsch synthesis reactor are estimated externally using literature references. Additionally, some equipment, such as the biomass dryer and lockhoppers, require literature references to determine the sizing, whereby their costs are subsequently

estimated using Aspen Icarus. The hydroprocessing plant area is modeled as a “black box” and therefore its costs are estimated as an overall scaled area cost from literature.

The costs of each equipment item and process area are scaled based on a scaling stream and scaling size factor (n) according to equation 12, where the size factor is between 0.6 and 1.0, depending on the equipment type.

$$Cost_{new} = Cost_0 * \left[\frac{Stream\ Size_{new}}{Stream\ Size_0} \right]^n \quad (\text{eqn. 12})$$

All purchased equipment costs determined via Aspen Icarus contain an installation factor that accounts for piping, electrical, and other costs required for installation. However, this installation factor tends to be significantly lower than metrics suggested by Peters et al. [50]. Therefore, rather than using the software-derived installation factors, an overall installation factor is applied to most equipment. A 3.02 overall installation factor is used, as suggested by Peters et al. for solid-liquid plants. Basically, the purchased equipment cost of a piece of equipment is multiplied by the installation cost to determine its installed cost. For the gasification unit, a 2.35 installation factor is used according to a National Energy Technology Laboratory study by Reed et al. [51]. It is assumed that all gas compressors receive a 1.2 installation factor, which is consistent with Aspen Icarus. The Chemical Engineering Plant Cost Index is used to bring the cost to 2007 U.S. dollars whenever a source for an estimated cost is from a previous year [52]. For multiple unit operations that operate in parallel or in trains, a train cost factor is applied. The reason for the factor, as reported by Larson et al. [38], is because those units share some piping, electrical, and other installation costs. It is applied as shown in equation 13, where n is the number of units in the train and m is the train factor with a value of 0.9.

$$Cost_{train} = Cost_{unit} * n^m \quad (\text{eqn. 13})$$

Table 10 explains the methodology used to estimate capital investment. After total purchased equipment cost (TPEC) and total installed cost (TIC) are determined, indirect costs are applied. Indirect costs (IC) include engineering and supervision, construction expenses, and legal and contractors’ fees at 32%, 34%, and 23% of TPEC, respectively [50]. Project contingency is added as 20% of total direct and indirect cost (TDIC). TDIC is set as the sum of IC and total installed costs (TIC). Adding project contingency to TDIC gives the fixed capital investment. Total capital investment (TCI) is determined by adding working capital to fixed capital investment and thereby represents the overall investment required for each scenario.

Table 10. Methodology for Capital Cost Estimation for nth Plant Scenarios

Parameter	Method
Total Purchased Equipment Cost (TPEC)	Aspen Icarus Process Evaluator, references
Total Installed Cost (TIC)	TPEC Installation Factor
Indirect Cost (IC)	89% of TPEC ^a
Total Direct and Indirect Costs (TDIC)	TIC + IC
Contingency	20% of TDIC
Fixed Capital Investment (FCI)	TDIC + Contingency
Working Capital (WC)	15% of FCI
Total Capital Investment	FCI + WC

^a Indirect costs are broken down into engineering and supervision, construction expenses, and legal and contractors' fees at 32%, 34%, and 23%, respectively, for a total of 89% of TPEC.

Raw material costs are inflated to 2007 U.S. dollars using the Industrial Inorganic Chemical Index also used by Phillips et al. [39]. Annual variable operating costs are determined from material stream flows. Variable operating costs and their respective cost methods are shown in Table 11. Natural gas for use in the gas turbine to produce power during startup and backup periods is assumed to be employed 5% of the annual operating time. Solids disposal costs are for the handling and removal of ash in the LT scenario and slag in the HT scenario. Wastewater disposal cost is applied to the sludge and black water produced during direct syngas quench. Catalyst costs are not calculated on an annual basis because the catalysts for all reactors are assumed to be replaced every 3 years. Instead they are accounted for in the discounted cash flow analysis.

Table 11. Variable Operating Cost Parameters Adjusted to \$2007

Variable Operating Costs	Cost information
Feedstock	\$75/dry short ton
LO-CAT Chemicals	\$176/metric ton of sulfur produced as reported in Phillips et al. [39]
Amine Makeup	\$1.09/lb as reported in Phillips et al. and set as 0.01% of the circulating rate [39]
Process Steam	\$8.20/ton (Peters et al.) [50]
Cooling Water	\$0.31/ton (Peters et al.)
Hydroprocessing	\$4.00/barrel produced as reported by Robinson and Dolbear [53]
Natural Gas (for backup)	\$6.40/thousand standard cubic feet as the average wellhead price for 2007 [54]
Ash/Char Disposal	\$23.52/ton [39]
Wastewater Disposal	\$3.30/hundred cubic feet [39]
Electricity	\$0.054/kWh ^a
Sulfur	\$40.00/ton [39]
Fischer-Tropsch Catalyst (cobalt)	\$15/lb and 64 lb/ft ³ density; applied on first operation year and then every three years ^a
Water-Gas-Shift Catalyst (copper-zinc)	\$8/lb and 900 kg/m ³ ; applied on first operation year and then every three years. Sour shift and normal WGS are assumed to operate with same catalyst ^a
Steam Methane Reforming Catalyst (nickel-aluminum)	\$15/lb and 70 lb/ft ³ ; applied on first operation year and then every three years ^a
Pressure Swing Adsorption	\$2/lb ^a

^a Assumed.

Fixed operating costs include employee salaries, overhead and maintenance, and insurance and taxes. Salaries are calculated similarly to Phillips et al. [39], where employees include a plant manager, shift supervisors, lab technicians, maintenance technicians, shift operators, yard workers, and office clerks. The labor index developed by the Bureau of Labor Statistics [55] is used to adjust the labor cost to \$2007. Overhead is calculated as 60% of total salaries; maintenance cost and taxes/insurance cost are both 2% of total installed equipment cost in accordance with Aden et al. [56].

For the DCFROR analysis, the capital investment is spent over a 2.5 year construction period, with 8% in the first half-year, followed by 60% and 32% for the second and third years. Working capital is applied in the year before operation and recovered at the end of the plant life. A standard modified accelerated cost recovery system (MACRS) is used, with the steam plant depreciating over 20 years and the rest of the plant over a 7-year period consistent with IRS

allowances. The project life is 20 years. Plant availability of 310 days per year (85%) is assumed and affects raw materials purchase as well as fuel production. The PV per gallon of gasoline equivalent is calculated for a set net present value of zero including a 10% internal rate of return.

Methodology for Major Equipment Costs

The software used for determining equipment costs is not capable of estimating every unit in this study. Some units such as the gasifiers and Fischer-Tropsch reactors are unique pieces of equipment that are underestimated if estimated as a simple vertical pressure vessel. Therefore, literature sources are used to help estimate sizes and costs of many units. The following section details a few of the more important units.

The biomass dryer costs are estimated by determining the drying contact area. According to Couper [57], typical rotary dryers have a diameter of 6 feet and solids holdup of 8%. Assuming a bulk density of 100 kg/m³ for ground stover and 1,000 kg/m³ for moisture in the stover, the resulting total surface area required for drying is 1,880 m². The surface area provides enough information for estimating the costs because rotary dryer costs are estimated based on surface area in Aspen Icarus. Details on dryer sizing can be found in the Detailed Calculations section of Appendix C.

The lockhopper system sizes are estimated by referring to a U.S. Department of Energy report completed by Combustion Engineering, Inc. [58], in which residence times and operating pressures are given. The biomass receiving bin, lockhopper, and feed bin costs are then estimated with Aspen Icarus. Details on lockhopper sizing can be found in the Detailed Calculations section of Appendix C.

The high-temperature gasifier cost is estimated from Reed et al. [51]. The total bare erected cost (installed cost) of a train of eight high-temperature E-Gas gasifiers (2,500 metric tpd coal) including syngas cooling costs is \$638 million (\$2006). It is assumed that the syngas cooling accounts for 20% of that cost, and therefore the estimated installed cost in millions of \$2006 for a 2,000 metric tpd high-temperature gasifier follows the formula in equation 14, resulting in \$57 million installed cost.

$$Cost_{HTgasifier} = \frac{(638 \cdot 80\%)}{8} \cdot \left[\frac{2000MT}{2500MT} \right]^{0.7} \quad (\text{eqn. 14})$$

Installed cost for a fluidized-bed gasifier is described in Larson et al. [38] and is calculated as shown in equation 15, where $Cost_{0_gasifier}$ is \$6.41 million (\$2003), $Stream Size_0$ is 41.7 metric ton per hour, and n is 0.7. The gasifier is evaluated at 300 short tpd because that appears to be the highest proven capacity for a GTI gasifier. Therefore, seven fluidized-bed gasifiers are used in parallel. It is assumed that the gasifier train follows the train cost formula (equation 13), resulting in \$19 million installed cost.

$$Cost_{LTgasifier} = Cost_{0_gasifier} * \left[\frac{Stream Size}{Stream Size_0} \right]^n \quad (\text{eqn. 15})$$

In a similar manner, the FT reactor is estimated as described in Larson et al. [38], where base installed cost is \$10.5 million (\$2003), base sizing value is 2.52 million scf/h of synthesis gas flow, and sizing exponent is 0.72. An installation factor of 3.6 is assumed for the FT reactor as found in Peters et al. [50] for liquid production plants. This allows the purchased cost of the unit to be back-calculated.

The acid gas removal (AGR) area cost is evaluated using information from Phillips et al. [39], following equation 12 where the base stream size is 4,000 short tpd and the base cost is \$5.45 million. The stream size is the mass flow of the synthesis gas entering the AGR as the sum of fresh syngas from gas scrubbing and unconverted syngas from the fuel synthesis area.

Capital investment for the hydroprocessing area is found in Robinson et al. [53]. That study reports a volumetric unit cost of \$4,000 per barrel per standard day. Assuming the typical hydroprocessing refinery produces 25,000 barrels per day, the base cost, C_0 , is \$100 million. Assuming a scaling exponent of 0.65, the cost of Area 500 is found using equation 12. The cost details for both of the gasifiers, the AGR area, the FT reactor, and the hydroprocessing area can be found in the Detailed Calculations section of Appendix C.

Methodology for Sensitivity Analysis

Sensitivity parameters are chosen to reflect the change in PV. The parameters are either economic or process parameters. The sensitivity bounds are chosen based on what is expected to be observed in the construction and operation of a biomass-to-liquids production plant. The chosen favorable, baseline, and unfavorable sensitivity variables are shown in Table 12.

Table 12. Sensitivity Parameters for nth Plant Scenarios

Parameter	Favorable	Baseline	Unfavorable
Availability (hours/year)	8,000	7,446	7,000
Balance of Plant (% of TPEC) ^a	8	12	16
Catalyst Cost (%) ^b	50	100	200
Catalyst Lifetime (years)	5	3	1
CO Conversion in FT Reactor (%)	30	40	50
Compressor Install Factor	1.0	1.2	3.0
Contingency (% of TDIC) ^c	10	20	30
Feedstock Cost (\$/dry short ton)	50	75	100
Feedstock Moisture (% wet)	20	25	30
Price of Electricity (¢/kWh)	7.0	5.4	3.0
Total Capital Investment (% of baseline)	70	100	130

^a TPEC=total purchased equipment cost.

^b All catalyst costs are varied over this range.

^c TDIC=total direct and indirect cost.

Methodology for Pioneer Plant Analysis

Economic analysis is based on an n^{th} plant design, and before a project is undertaken the pioneer (1^{st}) plant cost is important to estimate. We use a method begun by the RAND Corporation that estimates pioneer plant costs and plant performance. Using this methodology, two main areas of the n^{th} plant economic analysis are adjusted: capital investment and plant performance. Through a series of parameters, a cost inflation factor is generated to inflate the capital investment. In addition, a plant performance factor is calculated that reduces the fuel sales, feedstock purchase, and variable operating costs for the first several years that the plant is in operation. Each year, the plant performance factor is increased until full performance is attained. For the purpose of determining a range of pioneer plant costs, baseline, optimistic, and pessimistic values are chosen. The details of the RAND methodology can be found in Merrow et al. [59]. The following section explains the reasoning behind the parameters chosen for the scenarios.

Cost growth and plant performance factors are calculated as shown in equations 16 and 17 in accordance with Merrow et al. [59].

$$\begin{aligned} \text{Cost Growth} = & 1.1219 - 0.00297 * PCTNEW - 0.02125 * IMPURITIES \\ & - 0.01137 * COMPLEXITY + 0.00111 * INCLUSIVENESS \\ & - 0.06351 * PROJECT DEFINITION \end{aligned} \quad (\text{eqn. 16})$$

$$\begin{aligned} \text{Plant Perf.} = & 85.77 - 9.69 * NEWSTEPS + 0.33 * BALEQS - 4.12 \\ & * WASTE - 17.91 * SOLIDS \end{aligned} \quad (\text{eqn. 17})$$

The variables used in determining pioneer plant performance (equation 17) are defined as follows:

NEWSTEPS (0+): The feedstock handling area is chosen as a new step because of the large scale, which has not been demonstrated with biomass. The gasifier and solids feeding are also included as a new step because a pressurized biomass feeding system has not been demonstrated at a commercial scale except for in limited campaigns.

BALEQS (0 to 100): The mass and energy balances cannot be validated with current plant data, so a value of zero is chosen.

WASTE (0 to 5): Waste streams for gasification include scrubber sludge, black water, gasifier slag, fly ash, and sulfur. The scrubber sludge and black water require chemical treatment and the sulfur requires special handling. A mid-range value of 2.5 is chosen.

SOLIDS (0 or 1): Solids are present; therefore a value of 1 is used.

Variables used in determining pioneer plant cost growth (equation 16) are defined as follows:

PCTNEW (0 to 100%): The percentage cost of the gasifier, solids pressurizing, and solids feeding out of the total purchased equipment cost.

IMPURITIES (0 to 5): There are two major recycle streams in the gasification process, and there is the possibility of inert component buildup. There is also a potential for equipment corrosion due to sulfur components, hydrogen chloride, and hydrogen, so a value of 4 is assigned.

COMPLEXITY (0+): There are nine continuously linked steps in the gasification process. These include feedstock handling, solids feeding, gasification, amine scrubbing, sour water-gas-shift, pressure swing adsorption, Fischer-Tropsch synthesis, hydroprocessing, and air separation.

INCLUSIVENESS (0 to 100): Land costs and startup costs are considered in the TCI; however, they have not been rigorously investigated. A value of 33% is used.

PROJECT DEFINITION (2 to 8): The gasification platform is considered to be in the study design stage so a value of 7 is assigned.

The factors are applied to the capital investment and plant performance as shown in equations 18 and 19. Expenses and revenues affected by the plant performance factor are fuel sales, feedstock purchase, co-product credits, and variable operating costs.

$$TCI_{pioneer} = \frac{TCI_{nth}}{Cost\ Growth} \quad (\text{eqn. 18})$$

The *Cost Growth* factor causes the TCI of the pioneer plant to increase from the n^{th} plant cost.

$$Cost_{pioneer}(t) = Cost_{nth}(t) * \frac{(Plant\ Perf. + 20 * (t - 1))}{100} \quad (\text{eqn. 19})$$

$Cost_{nth}(t)$ is the n^{th} plant expense or revenue at year t . The plant performance factor is applied at year 1 and increases by 20% each year until 100% performance is reached. The chosen parameters and calculated factors for baseline, optimistic, and pessimistic scenarios are shown in Table 13.

Table 13. Pioneer Plant Analysis Parameters and Factors

Parameter	Baseline	Optimistic	Pessimistic	Range
NEWSTEPS	2	1	3	0+
BALEQS	0	0	0	0–100
WASTE	4	3	5	0–5
SOLIDS	1	1	1	0 or 1
Plant Performance	38.18	49.93	22.31	0–100
PCTNEW	19 (9) ^a	10 (5) ^a	25 (20) ^a	0–100
IMPURITIES	4	3	5	0–5
COMPLEX	9	6	12	0+
INCLUSIV.	33	50	0	0–100
PROJ. DEF.	7	6	8	2–8
Cost Growth (HT)	0.47	0.63	0.30	0–1
Cost Growth (LT)	0.50	0.65	0.31	0–1

^a Value in parentheses is value chosen for LT scenario.

Results and Discussion

Process Results

With a higher carbon conversion efficiency (see mass and energy balance discussions below), the HT scenario produces 61.0 GGE of fuels per dry metric ton of feedstock compared to 47.2 GGE/metric ton for the LT scenario. Resulting annual fuel outputs are 41.7 million GGE/yr for the HT scenario and 32.3 million GGE/yr for the LT scenario.

Along with lower fuel yield, the LT scenario consumes less power (Table 14). The LT scenario and HT scenario total power usage are 15 and 22 MW, respectively. Major contributions to this result are a lower grinder power due to less strict biomass size requirement, lower pressurized oxygen consumption in the gasifier, and generally lower downstream mass flow rates throughout the plant for the LT scenario. Lower syngas yield also means that there is less unconverted syngas and fuel gas from the hydroprocessing area available for the gas turbine. Therefore, the LT scenario generates 31 MW compared to 36 MW generated by the HT scenario. Due to unoptimized flow rates of the recycle streams, the LT scenario actually generates a net 16 MW of power, which is more than the 14 MW produced in the HT scenario. Reducing the net power generation is achievable by increasing the recycle ratio and thereby increasing conversion, but a consequence is higher flow rates and therefore larger and more expensive equipment. The focus of this study is to produce liquid fuels. However, procedures to optimize recycle ratios, equipment sizes, and fuel production rates are not within the scope of this study and are not undertaken.

Table 14. Power Generation and Usage

Power (MW)	HT Scenario	LT Scenario
USAGE		
Chopper	0.50	0.50
Grinder	2.96	1.10
Lockhopper System	0.18	0.18
Lean Amine Solution Pump	0.86	0.69
Syngas Booster Compressor	1.25	0.96
PSA Compressor	0.15	0.11
Recycle Compressor	0.39	0.29
Hydroprocessing Area	2.24	1.73
Oxygen Compressor (ASU)	3.61	2.80
Air Compressor (ASU)	7.94	6.31
Sour Gas Shift Steam Compressor	1.59	0
CO ₂ Compressor	0.39	0.39
Total Usage	22.06	15.06
GENERATION		
Gas Turbine	26.25	21.02
Steam Turbine	9.63	10.40
Total Generated	35.88	31.42
Net Export	13.82	16.36

An energy balance of the scenarios shows that the biomass-to-fuels efficiency for the LT and HT scenarios is 39% and 50% on a lower-heating-value (LHV) basis, respectively (Table 15). When the net electricity is added, the efficiencies are 43% and 53% on an LHV basis, respectively. The LT scenario is expected to be lower because mass and energy loss occurs in the production and removal of char and tar. Char and tar energy loss sums to 7.5% of the energy in the biomass. In this scenario, char is combusted in a fluidized-bed combustor to provide heat for biomass drying. Biomass drying in the HT scenario is accomplished by a syngas purge. However, the most significant energy loss in the LT scenario, about 25%, occurs across the gasifier. One reason for higher energy loss in the LT scenario is because conversion efficiency increases with increasing operating temperature. A second reason is the loss of energy during the cooling of the syngas after the gasifier. More effective capture of the energy in the hot syngas would increase the overall energy efficiency.

High exothermicity of the FT reaction causes a significant portion of the chemical energy in the syngas to leave as thermal energy in both scenarios. A higher loss across the FT reactor is observed in the HT scenario due to higher flow rates. Energy closure, as shown in Table 15, is approximately 90% for both scenarios. It is assumed that the last 10% is due mostly to heat loss from the cooling of the syngas by direct quench rather than capturing the heat and raising steam.

Table 15. Overall Energy Balance on LHV Basis

	HT Scenario	LT Scenario
IN		
Biomass	100.0	100.0
OUT		
Fuel	49.7	38.5
Net Electricity	3.5	4.2
Power Gen Losses	4.2	3.1
FT Reactor Losses	16.2	12.5
Gasifier Losses	12.1	24.9
Char	-	6.3
Tar	-	1.2
Syngas Purge	1.8	-
Total ^a	87.5	90.7

^aThe balance of energy is assumed to come from various heating and cooling losses.

A carbon balance analysis shows that 26% and 34% of the carbon in the biomass is passed on to the fuels for the LT and HT scenarios, respectively (Table 16). Approximately 99% of the carbon is accounted for. Major carbon losses include carbon dioxide flue gases, LO-CAT venting, and lockhopper venting. Char leaving the LT scenario is accounted for in the A200 flue gas because the char is combusted for process heat. Also, because the LT scenario produces low-molecular-weight hydrocarbons in the gasification process, a small fraction is dissolved in the liquid effluent of the wet scrubber. Carbon dioxide also dissolves in the wet scrubber effluent stream. Another carbon loss comes from the hydrocarbons that dissolve in the acid gas removal area.

Table 16. Overall Carbon Balance

	HT Scenario		LT Scenario	
	kmol/h	%	kmol/h	%
IN				
Biomass	3,281	100	3,281	100
OUT				
Fuel	1,111	34	862	26
A300 CO ₂ Vent	1,458	45	1,294	39
A600 Flue Gas	334	10	302	9
A200 Flue Gas	39	1	227	7
Lockhopper Vent	159	5	162	5
Wet Scrubber Effluent	154	5	318	10
Tar	0	-	35	1
Dissolved Hydrocarbons	0	-	46	1
Total	3,257	99	3,245	99

Throughout both scenarios, steam and cooling water are required as utilities. Because a pinch analysis (a method to optimize heat exchange) is not undertaken for this study, integration of the heat streams is not optimized. Therefore, it is assumed that the resulting heating and cooling requirements within the model represent steam and cooling water utilities, whereby they are recycled at a ratio of 9:1. In other words, fresh steam and cooling water utility input to the scenarios are calculated at 10% of the required circulating rate.

Cost Estimating Results

Capital and Operating Costs for nth Plant

The breakdown of costs by area and resulting total capital investment is shown in Table 17. Total capital investment amounts for the HT and LT scenarios are \$606 million and \$498 million, respectively. Major areas of investment are the gasification area in the HT scenario and the fuel synthesis area in the LT scenario. Moreover, these two areas contain significant differences in capital investment between the scenarios. The installed cost of the entrained-flow gasifier is significantly higher than the cost of the fluidized-bed gasifiers, even when seven of the latter are used in parallel. Area 400 costs for the LT scenario are higher than for the HT scenario due to the steam methane reformer and additional heat-exchange equipment required for the high operational temperature. A significant portion of the capital cost is due to gas compression, such as the air compressor in the air separation unit and the syngas booster compressor. Due to high purchase costs, compressors make up approximately 18% of the TPEC for each scenario. Detailed accounting of equipment found in each process area can be found in Appendix B.

Table 17. Capital Investment Breakdown for nth Plant Scenarios

Area	HT Scenario		LT Scenario	
	Installed Cost		Installed Cost	
	\$MM	%	\$MM	%
A100: Preprocessing	23	7	23	9
A200: Gasification	68	22	28	11
A300: Syngas Cleaning	34	11	29	12
A400: Fuel Synthesis	49	16	59	23
A500: Hydroprocessing	33	11	30	12
A600: Power Generation	46	15	39	15
A700: Air Separation Unit	24	8	20	8
Balance of Plant	33	11	27	11
Total Installed Cost	309		254	
Indirect Cost	130		107	
Total Direct and Indirect Cost	439		361	
Contingency	88		72	
Fixed Capital Investment	527		433	
Working Capital	79		65	
Total Capital Investment	606		498	

Annualized costs for operation of the plant are shown in Table 18. The percentage displayed also represents percentage of PV. The largest annual incurred costs for both scenarios are the average return on investment and feedstock purchase. Utilities such as steam and cooling water are higher for the LT scenario due to heating and cooling of the syngas before and after the SMR and steam input to the SMR. Waste disposal costs are equal because equal amounts of ash or slag are by-products of both plants. Annual hydroprocessing area costs and income taxes are higher for the HT scenario because of its higher fuel production rate. Fixed costs and capital depreciation are also higher due to higher TCI.

Catalyst costs are not determined on an annual basis because catalysts are assumed to be replaced only every 3 years. Table 19 contains catalyst replacement costs. The catalyst cost for the ZnO guard bed and PSA unit are equal across the scenarios because the volumes of the units are assumed to be the same. The FT catalyst for the HT scenario is significantly more expensive because of a higher gas flow rate and hence, higher catalyst use. Using a DCFROR analysis, the PVs at a net present value of zero for the LT and HT scenarios are \$4.83/GGE and \$4.27/GGE, respectively. Further detail of the yearly cash flow of the life of the plant can be found in Appendix B.

Table 18. Annual Operating Cost Breakdown for nth Plant Scenarios

	HT Scenario		LT Scenario	
	\$MM	%	\$MM	%
Average Return on Investment	58.2	32.7	48.3	31.0
Feedstock	51.3	28.9	51.3	32.9
Capital Depreciation	26.3	14.8	21.7	13.9
Average Income Tax	21.9	12.3	18.0	11.6
Fixed Costs	14.3	8.1	12.4	8.0
Hydroprocessing	4.4	2.5	3.0	2.0
Steam	2.7	1.5	3.5	2.2
Cooling Water	2.3	1.3	3.5	1.6
Waste Disposal	1.5	0.3	1.5	0.3
Other Raw Material Costs	1.4	0.8	1.3	0.8
Co-product Credits	(5.6)	-3.1	(6.6)	-4.2

Table 19. Catalyst Replacement Costs for Both Scenarios (3-Year Replacement Period)

Catalyst	HT scenario	LT scenario
Water-Gas-Shift (copper-zinc)	\$115,000	\$105,000
Steam Reforming (nickel-aluminum)	N/A	\$103,000
ZnO Guard Bed	\$424,000	\$424,000
PSA Packing	\$497,000	\$497,000
Fischer-Tropsch (cobalt)	\$7,687,000	\$6,128,000

Sensitivity Results for nth Plant

The results of sensitivity analysis are summarized in Figure 12 and Figure 13 for the HT and LT scenarios, respectively. Total capital investment and feedstock purchase cost have the highest effect on the PV at approximately $\pm\$0.80$ and $\pm\$0.40$ per GGE, respectively, for both scenarios. Due to the high percentage of equipment cost for compressors, the compressor installation factor has a very high effect on PV as well. When the compressor installation factor is increased to 3.0, which is the usual installation factor for most of the equipment, the PV increases by \$0.71 and \$0.78 per GGE for the LT and HT scenarios, respectively. Parameters with a lesser, but still significant, effect are the contingency factor (as percentage of total direct and indirect costs) and plant availability, both with an effect of approximately $\pm\$0.20$ per GGE. Parameters with the least effect are generally characteristic of the process rather than of the economics. For example, catalyst life, feedstock moisture, and carbon monoxide conversion in the FT reactor affect the PV less than $\pm\$0.15$ per GGE.

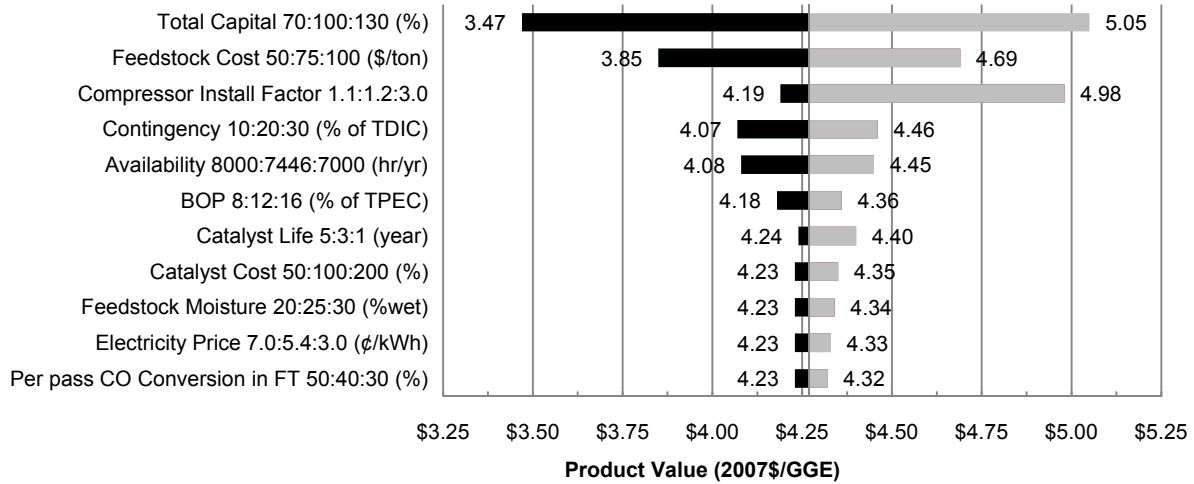


Figure 12. Sensitivity results for HT nth plant scenario

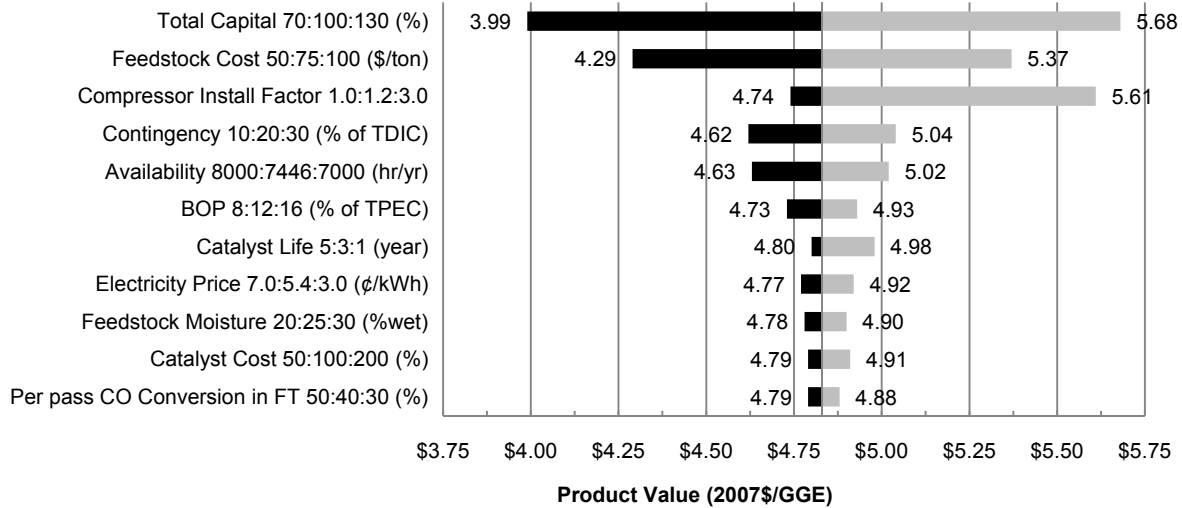


Figure 13. Sensitivity results for LT nth plant scenario

Additionally, the size of the plants can be varied by feedstock input rate. The effects of plant size on PV and TCI are shown in Figure 14 and Figure 15, respectively. When the plant size is reduced to 500 metric ton per day, the two scenarios approach equal PV. Also, as the plant size is reduced from the baseline, the difference in capital investment decreases. As the plant size increases past the baseline, the slope of the PV levels out, suggesting that the benefits of lower PV may not be worth the significant increase in capital cost (Figure 14).

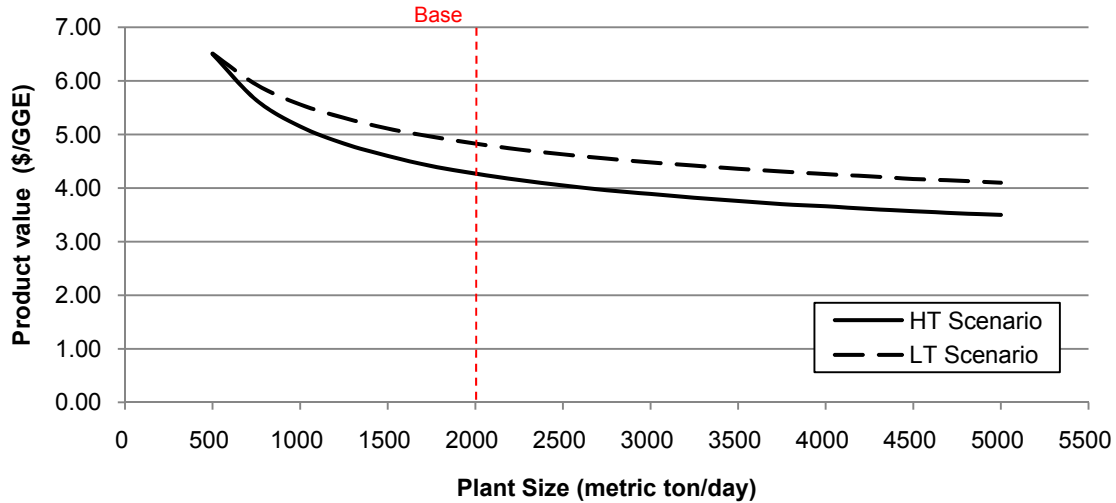


Figure 14. The effect of plant size on product value (per gallon of gasoline equivalent) for nth plant scenarios

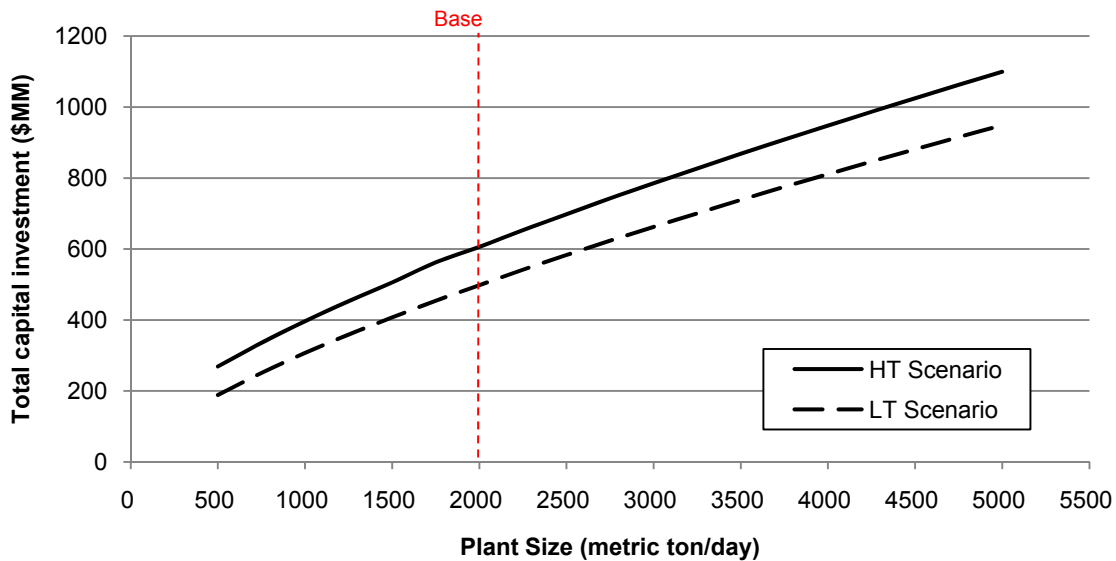


Figure 15. The effect of plant size on total capital investment for nth plant scenarios

Pioneer Plant Analysis Results

The total capital investment for a base-case pioneer plant is expected to double from the nth plant scenario, as detailed in Table 20. PV for a base case pioneer plant is estimated to increase to \$7.60 and \$8.10 per GGE for the LT and HT scenarios, respectively. Table 20 also shows estimates for the optimistic and pessimistic cases. An important observation is that the pioneer-plant PV for the LT scenario is actually lower than for the HT scenario. The reason behind this inverted result is because of the higher capital cost inflation (cost growth factor) in the HT scenario due to higher gasification area capital costs.

Table 20. Pioneer Plant Analysis Results

Analysis	HT Scenario		LT Scenario	
	TCI (\$MM)	PV (\$/GGE)	TCI (\$MM)	PV (\$/GGE)
n th Plant	606	4.27	498	4.83
1 st Plant Base	1,400	8.10	1,080	7.60
1 st Plant Optimistic	1,030	6.30	830	6.30
1 st Plant Pessimistic	2,200	12.60	1,740	11.50

Comparison with Previous Techno-Economic Studies

Two previous BTL studies that specifically use biomass feedstock, low-temperature gasification, and Fischer-Tropsch synthesis technology are Tijmensen et al. [40] and Larson et al. [38]. In order to compare with these studies, major economic and process parameters from the present nth plant LT scenario are adjusted to reflect similar values of the previous studies. First, the plant size of the present study is adjusted to increase equipment costs and raw materials purchases. As a result, the annual biomass input and TCI are affected. Second, availability in hours per year, rate of return, cost year, and feedstock cost are adjusted. The combined effect of all adjusted parameters causes the present study's product value to reflect the comparison study.

A comparison to the IGT-R scenario (which employs a low-temperature IGT gasifier and a steam methane reformer) in Tijmensen et al. shows that fuel product value is higher in the present study, as summarized in Table 21. Of all the scenarios developed by Tijmensen et al., the IGT-R scenario is most similar to the present study because of the reformer. The IGT-R scenario has a TCI of \$387 million, a feedstock cost of \$33 per short ton, and a product value of \$1.90/GGE. An important characteristic of the Tijmensen et al. study is that it does not include a hydroprocessing area. Therefore, it might be expected that the TCI would be higher for the present study, which includes hydroprocessing. However, that is not the case, as the TCI of the present study using Tijmensen et al. parameters is \$339 million, which is lower than the reported \$387 million of the earlier study. Another important observation is that the annual fuel production for the present study, with adjusted parameters, is 30.2 million gallons per year, compared to 39.8 million gallons per year of FT products reported by Tijmensen et al. One reason for lower annual fuel production in the present study is because of a loss during hydroprocessing. Therefore, due to lower annual fuel production and hence lower fuel revenue, the present study has a higher product value compared to Tijmensen et al.

Table 21. Comparison of nth Plant LT Scenario to Tijmensen et al. Study IGT-R Scenario [40]

Parameter	Tijmensen et al. Study (IGT-R Scenario)	Present Study n th Plant LT Scenario	Present Study w/ Tijmensen et al. Parameters
Plant Size (dry tons/day)	1,920	2,205	1,920
Annual Biomass Input (tons)	640,000	684,100	640,000
Total Capital Investment (\$MM)	387	498	339
Availability (hour/year)	8,000	7,446	8,000
Rate of Return (%)	10	10	10
Cost Year	2000	2007	2000
Feedstock Cost (\$/short ton)	33.00	75.00	33.00
Efficiency (% LHV, incl. elec.)	50.1	42.7	42.7
Fuel Yield (MMGGE/yr)	39.8	32.3	30.2
Product Value (\$/GJ)	16.50	39.80	25.17
Product Value (\$/GGE)	2.00	4.83	3.05

A comparison to the FT-OT-VENT scenario (which is low-temperature gasification with carbon dioxide vent and once through FT synthesis) reported by Larson et al. is summarized in Table 22. In a similar fashion to the previous comparison, the parameters are adjusted to approximate the comparison study. Some important observations are made from this comparison. First, the TCI of the present study with adjusted parameters is significantly higher. Second, the net electricity is significantly lower for the present study. Third, the PV is significantly higher for the present study. Essentially, the Larson et al. study generates more revenue from selling electricity and recovers the capital investment in less time. In addition, annual operating costs for the Larson et al. study are lower than for the present study. Therefore, because of lower revenue from electricity and higher annual operating costs, the present study has a higher fuel product value when compared on a similar basis to Larson et al.

Table 22. Comparison of nth Plant LT Scenario to Larson et al. Study FT-OT-VENT Scenario [38]

Parameter	Larson et al. Study (FT-OT-VENT Scenario)	Present Study n th Plant LT Scenario	Present Study w/ Larson et al. Parameters
Plant Size (dry tons/day)	5,000	2,205	5,001
Annual Biomass Input (tons)	1,458,000	684,000	1,459,000
Total Capital Investment (\$MM)	541	498	678
Availability (hour/year)	7,000	7,446	7,000
Debt/Equity (% Equity)	60	100	60
Rate of Return (%)	12	10	12
Cost Year	2003	2007	2003
Electricity Price (cents/kWh)	4.0	5.4	4.0
Net Electricity (MW)	207	16.3	37.1
Feedstock Cost (\$/short ton)	46.00	75.00	46.00
Plant Yield (MMGGE/yr)	63.3	32.3	68.9
Product Value (\$/GJ)	15.25	39.80	26.80
Product Value (\$/GGE)	1.85	4.83	3.25

Summary of nth Plant Scenarios

The HT scenario requires more power and capital investment, yields more fuel per ton of feedstock, and subsequently produces more fuel per year compared to the LT scenario. The total capital investment requirements for the LT and HT scenarios are \$498 million and \$606 million, respectively. Despite higher capital investment for the HT scenario, the product value (PV) is lower. PVs for the LT and HT scenarios are \$4.83 and \$4.27 per gallon of gasoline equivalent, respectively. The main reason for a lower PV is because of increased fuel revenue. The main nth plant scenario results are shown in Table 23. A detailed summary of costs can be found in Appendix B.

Table 23. Main Scenario nth Plant Results

Scenario	TCI (\$MM)	TPEC (\$MM)	Fuel Yield (GGE/metric ton)	Annual Fuel Output (MMGGE/yr)	Net Electricity Export (MW)	PV (\$/GGE)
High Temperature	605.9	145.7	61.0	41.7	13.8	4.27
Low Temperature	498.3	120.4	47.2	32.3	16.4	4.83

Conclusions

This analysis compares capital and operating cost for two biomass-to-liquids scenarios: high-temperature (HT) gasification and low-temperature (LT) gasification. The selection of these scenarios allows for direct comparison between two modes of gasification: slagging and non-slagging. The slagging, entrained-flow gasifier employed for the HT scenario results in higher plant costs (about 20%) than the LT scenario, which employs a fluidized bed gasifier. On the other hand, the higher carbon conversions for the HT gasifier result in a lower PV compared to the LT scenario. A biomass-to-liquids plant is expected to produce fuels costing in the range of \$4–\$5 per gallon gasoline equivalent using present gasification and Fischer-Tropsch synthesis technology. The factors chiefly responsible for this relatively high PV are feedstock costs and investment return on the capital to build a \$500 million to \$650 million plant to process 2,000 metric tons of biomass per day. A pioneer plant analysis estimates that the total capital investment for a pioneer plant would double and PV would increase by approximately 60% from the n^{th} plant scale. This uncertainty suggests that economics are still a major challenge for biomass-to-liquids production plants.

Parameters with the most sensitive effect on PV are total capital cost, feedstock purchase cost, and compressor installation factor, affecting the PV \pm \$0.40–\$0.80 per gallon. Less expensive biomass feedstock that is lower in ash content than that used in the present study will have higher fuel yield and have the potential to significantly decrease PV. Gas compression is a major portion of capital investment, and sensitivity analysis shows that installation costs of compressors have a high effect on PV. Factors with little effect on the PV are mostly related to the process, such as carbon monoxide conversion in the FT reactor, feedstock inlet moisture, and catalyst lifetime.

Because of time and resource constraints, the techno-economic study presented includes a few shortcomings. The process configuration is not fully optimized for heat integration. While some recycle streams are included, a complete heat exchange network for heat recovery is not conceptualized. In addition, some areas such as FT product separation and hydroprocessing are not modeled rigorously and can be improved with detailed mass and energy flows. Further studies can benefit by optimizing heat exchange networks and modeling in detail areas such as FT product separation and hydroprocessing. More detailed modeling will likely result in changes to economic components such as equipment cost, installation factors, and contingency. However, because increasing model detail may result in tradeoffs between the different economic components, the net effect of increased model detail on PV is difficult to predict.

This analysis tracks PV based on commercial technology for which sufficient public domain data existed in 2007. In order to evaluate the economics of biofuels produced from gasification based on future scenarios, particular attention will be needed on the most sensitive parameters—feedstock cost and capital cost. Attention should also focus on commercial biomass-to-fuels gasification plants that come online. As these plants do come online, cost growth and plant performance factors will improve, thereby decreasing the pioneer plant PV.

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Appendix A. Techno-Economic Model Assumptions

Financial Assumptions

- Capital investment
 - Equity: 100%
 - Working capital (% of FCI): 15%
- Depreciation model
 - Zero Salvage Value for both general plant and steam/ power plant
 - Type of depreciation: Double-Declining-Balance Depreciation Method (DDB) as per IRS Modified Accelerated Cost Recovery System (MACRS) guidelines
 - Depreciation period (years):
 - General plant: 7
 - Steam/power system: 20
- Construction & startup:
 - Construction period (years): 2.5
 - % spent in year “-3”: 8%
 - % spent in year “-2”: 60%
 - % spent in year “-1”: 32%
 - Startup time (years): 0.5
 - Revenues (% of normal): 50%
 - Variable costs (% of normal): 75%
 - Fixed cost (% of normal): 100%
- Other
 - Internal rate of return: 10%
 - Income tax rate: 39%
 - Operating hours per year: 8,406

Capital Costs

- Cost year for analysis: 2007; cost escalation is applied using the Chemical Engineering Plant Cost Index
- The plant is designed based on the state of the technology, at the n^{th} plant level of experience
- Most equipment installation factors are applied using Peters et al. for solid-fluid plants (i.e., 3.02 installation factor)

- Materials of construction are carbon steel, stainless steel, alloys, and refractory where necessary
- Sensitivity parameters involving changes in equipment size or capacity use scaling exponents available in literature

Operating Costs

- Working capital is assumed to be 15% of the total capital investment
- Annual maintenance materials are 2% of the total installed equipment cost
- Boiler feedwater and wastewater treatment costs are derived from prior NREL work
- Fresh cooling water and steam costs are calculated at 10% of the required circulation rate, meaning a 9:1 ratio of water recycling
- Employee salary estimation is same as that chosen by Phillips et al.
- Employee salaries are indexed to the year of 2007 following the data of the Bureau of Labor Statistics

Feedstock, Products, and By-Products

- Feedstock is corn stover (comprising stalks, leaves, cobs, and husks)
 - Moisture content in the feedstock is 25%
- Feed rate is 2,000 dry metric ton per day
 - The feedstock delivery logistics are not considered
 - The feedstock is delivered to the feed handling area of the plant
- Feed cost is assumed to be \$75/dry short ton at the gate
- Gasoline and diesel products are sold for over-the-fence prices
- Gasoline energy content is 115,000 BTU/gallon
- Fly ash and slag incur a solids waste disposal cost
- Solid sulfur and electricity are sold as by-products

Process Assumptions

For both scenarios, most of the process was modeled with the aid of Aspen Plus software. The process was divided by logical process areas, which are named below.

Area 100 - Preprocessing

- Biomass is dried down to 10%
 - Steam raised from hot flue gas is used to dry the feedstock
 - Steam to moisture removal ratio is set at 9:1 in accordance with Amos
 - Heat is provided by combusting char and unreacted syngas
- Grinder reduces biomass to 6-mm or less

- The energy required for grinding is calculated separately using literature correlations by Mani et al.

Area 200 - Gasification

- Scenario 1: Entrained-flow gasifier is modeled using thermodynamic equilibrium
- Scenario 2: Fluidized-bed gasifier is modeled using a mass balance calculation
- 95% purity oxygen produced from air separation unit provides oxidizer
- Carbon dioxide is used as solids pressurization gas
- All char produced in LT scenario is combusted for process heat

Area 300 - Syngas Cleaning

- Particulates, tar, and partial ammonia removal via wet scrubbing
 - Scrubbing water is recycled at 90% rate
 - Particulate handling (not modeled)
 - High-temperature gasifier: particulate decant slurry is sent back into slagging gasifier
 - Low-temperature gasifier: particulate decant slurry is piled and landfilled; excess water is sent to aerobic water treatment (not modeled)
 - Makeup water compensates for water lost via particulate slurry
 - Process water condensate is used as makeup water
- Sour water-gas-shift occurs at equilibrium and is modeled as such
- Carbon dioxide, hydrogen sulfide, and excess ammonia removal via amine scrubbing acid gas removal (AGR) at pressure
 - 99% of sulfur is removed and 90% of carbon dioxide
 - Monoethanolamine (MEA) is the scrubbing solvent
 - Carbon dioxide is vented following LO-CAT removal of H₂S
- Hydrogen sulfide is converted to solid sulfur via LO-CAT oxidation (99% conversion)
- Ammonia can be disposed of by decomposition (not modeled) in
 - Gasifier burner (slagging gasifier)
 - Char and syngas combustor (fluidized-bed gasifier)
- Zinc oxide and activated carbon guard bed polishing assumed (not modeled in detail)

Area 400 - Fuel Synthesis

- Water-gas-shift occurs at equilibrium and is modeled as such
- Pressure swing adsorption (PSA) is employed to remove excess H₂ at an efficiency of 85% and 99% purity
 - The PSA system employs two trains with six reactors each to account for all stages of pressurization, depressurization, purging, etc.
 - PSA adsorbers are filled 2/3 with activated carbon and 1/3 with molecular sieve
- Syngas is catalytically converted to fuels by one-step Fischer-Tropsch synthesis followed by wax hydrocracking and fuel separation
 - FT synthesis employs cobalt catalyst
 - 40% syngas conversion to fuels
 - Part of the unconverted syngas is recycled
 - A fraction of the recycle is sent to the AGR to prevent CO₂ buildup
 - The overall recycle ratio is about 1.9
- A syngas purge is used as fuel in the combustor side of the biomass dryer (only in HT scenario)
- Excess syngas is sent to a gas turbine for power production

Areas 500, 600, 700

- Hydroprocessing and product distillation costs are estimated as a “black box” based on literature capital cost and operating cost information from Robinson et al.
 - Literature yield data is used for estimating the relative yields of gasoline and diesel

Miscellaneous

- Combustion occurs with 20% excess oxygen

Appendix B. Detailed Costs

Cost Summaries

HT Biomass-to-Liquids Scenario Summary

2,000 Dry Metric Tonnes Biomass per Day
 High Temperature Entrained Flow Gasifier, Sulfur Removal, Fischer-Tropsch Synthesis, Hydroprocessing, Combined Cycle Power
 All Currency in 2007\$ and Volume in Gallons Gasoline Equivalent (GGE)

Product Value (\$/gal) \$4.26

Total Production at Operating Capacity (MM gal / year) 41.7
 Product Yield (gal / Dry US Ton Feedstock) 61.0
 Delivered Feedstock Cost \$/Dry US Ton \$75
 Internal Rate of Return (After-Tax) 10%
 Equity Percent of Total Investment 100%

Capital Costs			Operating Costs (cents/gal product)		
Area 100: Pretreatment	\$22,700,000	7%	Feedstock	123.0	28.9%
Area 200: Gasification	\$67,800,000	22%	Steam	6.4	1.5%
Area 300: Syngas Cleaning	\$33,500,000	11%	Cooling Water	5.5	1.3%
Area 400: Fuel Synthesis	\$49,400,000	16%	Other Raw Materials	3.4	0.8%
Area 500: Hydrocracking/Hydrotreating	\$33,000,000	11%	Waste Disposal	1.3	0.3%
Area 600: Power Generation	\$45,600,000	15%	Hydroprocessing	10.6	2.5%
Area 700: Air Separation	\$24,300,000	8%	Fixed Costs	34.4	8.1%
Balance of Plant	\$33,100,000	11%	Co-product credits	-13.3	-3.1%
Total Installed Equipment Cost	\$309,400,000		Capital Depreciation	63.0	14.8%
Indirect Costs	129,700,000		Average Income Tax	52.4	12.3%
(% of TPI)	21.4%		Average Return on Investment	139.5	32.7%
Project Contingency	79,000,000		Operating Costs (\$/yr)		
Total Project Investment (TPI)	\$605,900,000		Feedstock	\$51,300,000	
Installed Equipment Cost per Annual Gallon	\$7.42		Steam	\$2,700,000	
Total Project Investment per Annual Gallon	\$14.52		Cooling Water	\$2,300,000	
Loan Rate	N/A		Other Raw Matl. Costs	\$1,400,000	
Term (years)	N/A		Waste Disposal	\$1,500,000	
Capital Charge Factor	0.176		Hydroprocessing	\$4,400,000	
Gasifier Efficiency -HHV %	82.1		Fixed Costs	\$14,300,000	
Gasifier Efficiency -LHV %	87.9		Co-product credits	-\$5,600,000	
Overall Plant Efficiency (incl. electricity) -HHV %	52.7		Capital Depreciation	\$26,300,000	
Overall Plant Efficiency -LHV %	53.0		Average Income Tax	\$21,900,000	
Availability (%)	85.0%		Average Return on Investment	\$58,200,000	
Plant Hours per year	7446		Total Plant Electricity Usage (KW)	22,065	
			Electricity Produced Onsite (KW)	35,880	
			Electricity Purchased from Grid (KW)	0	
			Electricity Sold to Grid (KW)	13,815	
			Plant Electricity Use (KWh/gal product)	6.1	

Figure B-1. Economic analysis summary for HT scenario

LT Biomass-to-Liquids Process Engineering Analysis

2,000 Dry Metric Tonnes Biomass per Day
 Low Temperature Fluidized Gasifier, Sulfur Removal, Fischer-Tropsch Synthesis, Hydroprocessing, Combined Cycle Power
 All Currency in 2007\$ and Volume in Gallons Gasoline Equivalent (GGE)

Product Value (\$/gal) \$4.83

Total Production at Operating Capacity (MM gal / year) 32.3
 Product Yield (gal / Dry US Ton Feedstock) 47.2
 Delivered Feedstock Cost \$/Dry US Ton \$75
 Internal Rate of Return (After-Tax) 10%
 Equity Percent of Total Investment 100%

Capital Costs			Operating Costs (cents/gal product)		
Area 100: Pretreatment	\$22,700,000	9%	Feedstock	158.9	32.9%
Area 200: Gasification	\$28,200,000	11%	Steam	10.9	2.2%
Area 300: Syngas Cleaning	\$29,300,000	12%	Cooling Water	7.8	1.6%
Area 400: Fuel Synthesis	\$58,700,000	23%	Other Raw Materials	4.1	0.8%
Area 500: Hydrocracking/Hydrotreating	\$29,500,000	12%	Waste Disposal	1.5	0.3%
Area 600: Power Generation	\$38,900,000	15%	Hydroprocessing	9.4	2.0%
Area 700: Air Separation	\$19,500,000	8%	Fixed Costs	38.4	8.0%
Balance of Plant	\$27,200,000	11%	Co-product credits	-20.4	-4.2%
			Capital Depreciation	67.2	13.9%
Total Installed Equipment Cost	\$253,900,000		Average Income Tax	55.9	11.6%
			Average Return on Investment	149.5	31.0%
Indirect Costs	107,200,000				
(% of TPI)	21.5%		Operating Costs (\$/yr)		
Project Contingency	65,000,000		Feedstock	\$51,300,000	
			Steam	\$3,500,000	
Total Project Investment (TPI)	\$498,300,000		Cooling Water	\$3,500,000	
			Other Raw Matl. Costs	\$1,300,000	
Installed Equipment Cost per Annual Gallon	\$7.86		Waste Disposal	\$1,500,000	
Total Project Investment per Annual Gallon	\$15.43		Hydroprocessing	\$3,000,000	
			Fixed Costs	\$12,400,000	
Loan Rate	N/A		Co-product credits	-\$6,600,000	
Term (years)	N/A		Capital Depreciation	\$21,700,000	
Capital Charge Factor	0.177		Average Income Tax	\$18,000,000	
			Average Return on Investment	\$48,300,000	
Gasifier Efficiency - HHV %	64.3		Total Plant Electricity Usage (KW)	15,044	
Gasifier Efficiency - LHV %	68.8		Electricity Produced Onsite (KW)	31,420	
Overall Plant Efficiency - HHV %	43.0		Electricity Purchased from Grid (KW)	0	
Overall Plant Efficiency - LHV %	43.3		Electricity Sold to Grid (KW)	16,376	
Availability (%)	85.0%				
Plant Hours per year	7446		Plant Electricity Use (KWh/gal product)	5.4	

Figure B-2. Economic analysis summary for LT scenario

Detailed Equipment Lists

High-Temperature Scenario Equipment Lists

Table B-1. Detailed Equipment List for Areas 100 and 200 of HT Scenario

Equipment Number	Number Required	Number Spares	Equipment Name	Original Equip Cost (per unit)	Base Year	Total Original Equip Cost (Req'd & Spare) in Base		Scaled Uninstalled Cost in 2007\$	Installed Cost Base Year	Installed Cost in 2007\$	Cost Source
						Year	Year				
A100.CONV1	2		Bale Transport Conveyor	\$400,000	2000		\$800,000	\$1,066,531	\$1,296,000	\$1,727,781	Aden et al. 2002
A100.CONV2	2		Bale Unwrapping Conveyor	\$150,000	2000		\$300,000	\$399,949	\$357,000	\$475,940	Aden et al. 2002
A100.CONV3	1		Belt Press Discharge Conveyor	\$50,000	2000		\$50,000	\$66,658	\$94,500	\$125,984	Aden et al. 2002
A100.SCALE	2		Truck Scales	\$34,000	2000		\$68,000	\$90,655	\$167,960	\$223,918	Aden et al. 2002
A100.FORK1	4	1	Truck Unloading Forklift	\$18,000	2000		\$90,000	\$119,985	\$90,000	\$119,985	Aden et al. 2002
A100.FORK2	4		Bale Moving Forklift	\$18,000	2000		\$72,000	\$95,988	\$72,000	\$95,988	Aden et al. 2002
A100.SLAB	1		Concrete Feedstock-Storage Slab	\$450,655	2000		\$450,655	\$600,797	\$991,441	\$1,321,754	Aden et al. 2002
A100.MAGSEP	1		Magnetic Separator	\$13,863	1998		\$13,863	\$18,700	\$18,022	\$24,310	Aden et al. 2002
A100.A100CHOP.CHGRIN01	4		Chopper	\$105,100	2007		\$420,400	\$420,400	\$1,105,258	\$1,105,258	Process Evaluator software Aspen Icarus
A100.A100CHOP.CHMIX01	1		Chopper Conveyor	\$61,400	2007		\$61,400	\$61,400	\$185,428	\$185,428	Aspen Icarus
A100.A100CHOP.CHSEP01	1		Chopper Screen with Recycle Conveyor	\$20,800	2007		\$20,800	\$20,800	\$62,816	\$62,816	Aspen Icarus
A100.A100DRY.DRDRY01	10		Dryer	\$633,700	2007		\$6,337,000	\$6,337,000	\$15,201,647	\$15,201,647	Aspen Icarus
A100.A100GRIN.GRGRIN01	4		Grinder	\$167,100	2007		\$668,400	\$668,400	\$1,757,266	\$1,757,266	Aspen Icarus
A100.A100GRIN.GRMIX01	1		Grinder Conveyor	\$61,400	2007		\$61,400	\$61,400	\$185,428	\$185,428	Aspen Icarus
A100.A100GRIN.GRSEP01	1		Grinder Screen with Recycle Conveyor	\$20,800	2007		\$20,800	\$20,800	\$62,816	\$62,816	Aspen Icarus
A100					Subtotal		\$9,434,718	\$10,049,464	\$21,647,582	\$22,676,317	
A200.A200COMB.CBREAC01	1		Combustor - Steam Boiler	\$1,450,500	2007		\$1,450,500	\$1,450,500	\$4,380,510	\$4,380,510	Aspen Icarus
A200.A200SLAG.SLREAC01	1		Entrained Flow, Slagging Gasifier	\$23,234,043	2006		\$23,234,043	\$24,433,879	\$54,600,000	\$57,419,616	Reed et al. 2007
A200.A200SLAG.SLSEP01	1		Slag collector/separator	\$35,100	2007		\$35,100	\$35,100	\$106,002	\$106,002	Aspen Icarus
A200.A200SLAG.SLSEP03	3		Direct Quench Syngas Cooler	\$396,200	2007		\$1,188,600	\$1,188,600	\$3,589,572	\$3,589,572	Aspen Icarus
A200.GSHOP01	1		Biomass Receiving Hopper	\$151,400	2007		\$297,900	\$297,900	\$899,658	\$899,658	Aspen Icarus
A200.GSTANK01	1		Lockhopper	\$229,100	2007		\$229,100	\$229,100	\$691,882	\$691,882	Aspen Icarus
A200.GSTANK02	1		Biomass Feeding Bin	\$228,900	2007		\$228,900	\$228,900	\$691,278	\$691,278	Aspen Icarus
A200					Subtotal		\$26,664,143	\$27,863,979	\$64,958,902	\$67,778,518	

Table B-2. Detailed Equipment List for Areas 300, 400, and 500 of HT Scenario

Equipment Number	Number Required	Number Spares	Equipment Name	Original Equip Cost (per unit)	Base Year	Total Original Equip Cost (Req'd & Spare) in Base Year	Scaled Uninstalled Cost in 2007\$	Installed Cost Base Year	Installed Cost in 2007\$	Cost Source
A300.A300AGR.AGRarea	1		High Pressure Amine System	\$6,949,800	2005	\$6,949,800	\$7,798,857	\$20,988,396	\$23,552,549	Phillips et al. 2007
A300.A300SGS.SGCOMP01	2		Sour Water Gas Shift Steam Compressor	\$1,381,900	2007	\$2,763,800	\$2,763,800	\$3,316,560	\$3,316,560	Aspen Icarus
A300.A300SGS.SGREAC01	1		Sour Water Gas Shift Reactor	\$66,600	2007	\$66,600	\$66,600	\$201,132	\$201,132	Aspen Icarus
A300.A300SUL.SUCOL01	1		LO-CAT Absorber	\$23,800	2007	\$23,800	\$23,800	\$71,876	\$71,876	Aspen Icarus
A300.A300SUL.SUREAC01	1		LO-CAT Oxidizer Vessel	\$1,000,000	2007	\$1,000,000	\$1,000,000	\$3,020,000	\$3,020,000	Phillips et al. 2007
A300.A300SUL.SUSEP01	1		Sulfur Separator	\$15,900	2007	\$15,900	\$15,900	\$48,018	\$48,018	Aspen Icarus
A300.CLCOMP01	2		Carbon Dioxide Compressor	\$1,181,200	2007	\$2,362,400	\$2,362,400	\$2,834,880	\$2,834,880	Aspen Icarus
A300.CLDROM01	1		Liquid Collection Tank	\$29,600	2007	\$29,600	\$29,600	\$89,392	\$89,392	Aspen Icarus
A300.CLHEAT03	1		Direct Quench Syngas Cooler	\$91,500	2007	\$91,500	\$91,500	\$276,330	\$276,330	Aspen Icarus
A300.CLMIX01	1		Venturi Scrubber	\$27,100	2007	\$27,100	\$27,100	\$81,842	\$81,842	Aspen Icarus
A300					Subtotal	\$13,330,500	\$14,179,557	\$30,928,426	\$33,492,579	
A400.FSCOMP01	2		Booster Syngas Compressor	\$1,007,100	2007	\$2,014,200	\$2,014,200	\$2,417,040	\$2,417,040	Aspen Icarus
A400.FSCOMP02	1		Recycle Syngas Booster Compressor	\$748,400	2007	\$748,400	\$748,400	\$898,080	\$898,080	Aspen Icarus
A400.FSCOMP03	1		PSA Booster Compressor	\$1,461,700	2007	\$1,461,700	\$1,461,700	\$1,754,040	\$1,754,040	Aspen Icarus
A400.FSHEAT01	1		Syngas Heater	\$73,400	2007	\$73,400	\$73,400	\$221,668	\$221,668	Aspen Icarus
A400.FSHEAT03	1		Syngas Cooler	\$137,400	2007	\$137,400	\$137,400	\$414,948	\$414,948	Aspen Icarus
A400.FSHEAT04	1		Recycle Syngas Pre-heater	\$21,500	2007	\$21,500	\$21,500	\$64,930	\$64,930	Aspen Icarus
A400.FSREAC01	1		Fischer-Tropsch Reactor	\$8,888,889	2003	\$8,888,889	\$11,617,468	\$32,000,000	\$41,822,886	Larson et al. 2005
A400.FSSEP01	2		ZnO Sulfur Removal Beds	\$61,000	2007	\$122,000	\$122,000	\$368,440	\$368,440	Aspen Icarus
A400.FSSEP02	12		Pressure Swing Absorption Unit	\$33,300	2007	\$399,600	\$399,600	\$1,206,792	\$1,206,792	Aspen Icarus
A400.FSSEP03	1		FT knock-out Column	\$39,600	2007	\$39,600	\$39,600	\$119,592	\$119,592	Aspen Icarus
A400.FSSEP04	1		Water Separator	\$47,900	2007	\$47,900	\$47,900	\$144,658	\$144,658	Aspen Icarus
A400					Subtotal	\$13,954,589	\$16,683,168	\$39,610,188	\$49,433,074	
A500.HYREAC01	1		Hydroprocessing Unit	\$9,377,483	2007	\$9,377,483	\$9,377,483	\$28,320,000	\$28,320,000	Robinson & Dolbear 2007
A500.HYTANK01	1		Diesel 30-day Storage Tank	\$1,167,600	2007	\$1,167,600	\$1,167,600	\$3,526,152	\$3,526,152	Aspen Icarus
A500.HYTANK02	1		Gasoline 30-day Storage Tank	\$371,900	2007	\$371,900	\$371,900	\$1,123,138	\$1,123,138	Aspen Icarus
A500					Subtotal	\$10,916,983	\$10,916,983	\$32,969,290	\$32,969,290	

Table B-3. Detailed Equipment List for Areas 600 and 700 of HT Scenario

Equipment Number	Number Required	Number Spares	Equipment Name	Original Equip Cost (per unit)	Base Year	Total Original Equip Cost (Req'd & Spare) in Base		Scaled Uninstalled Cost in 2007\$	Installed Cost Base Year	Installed Cost in 2007\$	Cost Source
						Year	Year				
A600.COMBB	1		Combustion Turbine - Electric Generator	\$22,404,000	2007	\$22,404,000	\$22,404,000	\$22,404,000	\$26,884,800	\$26,884,800	Aspen Icarus
A600.CWPUMP	1	1	Cooling Water Pump	\$5,900	2007	\$11,800	\$11,800	\$11,800	\$35,636	\$35,636	Aspen Icarus
A600.ECON1_HRSG	1		Heat Recovery Steam Generator	\$202,200	2007	\$202,200	\$202,200	\$202,200	\$610,644	\$610,644	Aspen Icarus
A600.HPPUMP	1	1	High Pressure Steam Pump	\$266,700	2007	\$533,400	\$533,400	\$533,400	\$1,610,868	\$1,610,868	Aspen Icarus
A600.HPSEP	1		High Pressure Steam/Water Separation	\$107,400	2007	\$107,400	\$107,400	\$107,400	\$324,348	\$324,348	Aspen Icarus
A600.LPEXP_ELECGEN	1		Combined Steam Turbine - Electric Gen.	\$4,709,600	2007	\$4,709,600	\$4,709,600	\$4,709,600	\$5,651,520	\$5,651,520	Aspen Icarus
A600.LPSEP	1		Low Pressure Water/Steam Separation	\$108,800	2007	\$108,800	\$108,800	\$108,800	\$328,576	\$328,576	Aspen Icarus
A600.O2COMP	1		Air Compressor	\$8,431,900	2007	\$8,431,900	\$8,431,900	\$8,431,900	\$10,118,280	\$10,118,280	Aspen Icarus
A600					Subtotal	\$36,509,100	\$36,509,100	\$36,509,100	\$45,564,672	\$45,564,672	
A700.COMP1	2		Air Compressor	\$3,346,500	2007	\$6,693,000	\$6,693,000	\$6,693,000	\$8,031,600	\$8,031,600	Aspen Icarus
A700.COOLER	1		Air Cooler	\$27,200	2007	\$27,200	\$27,200	\$27,200	\$82,144	\$82,144	Aspen Icarus
A700.GOXCLR-1	1		Oxygen Compressor Cooler	\$23,300	2007	\$23,300	\$23,300	\$23,300	\$70,366	\$70,366	Aspen Icarus
A700.GOXCLR-2	1		Oxygen Compressor Cooler	\$23,000	2007	\$23,000	\$23,000	\$23,000	\$69,460	\$69,460	Aspen Icarus
A700.GOXCMP-1	2		Oxygen Compressor	\$1,489,600	2007	\$2,979,200	\$2,979,200	\$2,979,200	\$3,575,040	\$3,575,040	Aspen Icarus
A700.HIGH-P.cond	1		High Pressure Column Condenser	\$20,300	2007	\$20,300	\$20,300	\$20,300	\$61,306	\$61,306	Aspen Icarus
A700.HIGH-P.cond acc	1		High Pressure Column Condenser Accumulator	\$40,500	2007	\$40,500	\$40,500	\$40,500	\$122,310	\$122,310	Aspen Icarus
A700.HIGH-P.reflux pump	1	1	High Pressure Column Reflux Pump	\$14,300	2007	\$28,600	\$28,600	\$28,600	\$86,372	\$86,372	Aspen Icarus
A700.HIGH-P.tower	1		High Pressure Column Tower	\$314,300	2007	\$314,300	\$314,300	\$314,300	\$949,186	\$949,186	Aspen Icarus
A700.INTRC1	1		Air Compressor Intercooler	\$338,300	2007	\$338,300	\$338,300	\$338,300	\$1,021,666	\$1,021,666	Aspen Icarus
A700.INTRC2	1		Air Compressor Intercooler	\$304,500	2007	\$304,500	\$304,500	\$304,500	\$919,590	\$919,590	Aspen Icarus
A700.INTRC3	1		Air Compressor Intercooler	\$222,500	2007	\$222,500	\$222,500	\$222,500	\$671,950	\$671,950	Aspen Icarus
A700.LOW-P.reb	1		Low Pressure Column Reboiler	\$19,600	2007	\$19,600	\$19,600	\$19,600	\$59,192	\$59,192	Aspen Icarus
A700.LOW-P.tower	1		Low Pressure Column Tower	\$2,581,600	2007	\$2,581,600	\$2,581,600	\$2,581,600	\$7,796,432	\$7,796,432	Aspen Icarus
A700.TSA	1		Water Knock-out Drum	\$35,900	2007	\$35,900	\$35,900	\$35,900	\$108,418	\$108,418	Aspen Icarus
A700.TURB-1	2		Gas Expander	\$86,100	2007	\$172,200	\$172,200	\$172,200	\$520,044	\$520,044	Aspen Icarus
A700.WK01	1		Water Knock-out Drum	\$57,700	2007	\$57,700	\$57,700	\$57,700	\$174,254	\$174,254	Aspen Icarus
A700					Subtotal	\$13,881,700	\$13,881,700	\$13,881,700	\$24,319,330	\$24,319,330	
Total						\$124,691,733	\$130,083,951	\$130,083,951	\$259,998,390	\$276,233,779	
Total (with BOP)						\$139,654,741	\$145,694,026	\$145,694,026	\$291,198,196	\$309,381,833	

Low-Temperature Scenario Equipment Lists

Table B-4. Detailed Equipment List for Areas 100 and 200 of LT Scenario

Equipment Number	Number Required	Number Spares	Equipment Name	Original Equip Cost (per unit) in Base Year	Base Year	Total Original Equip Cost (Req'd & Spare) in Base Year	Scaled Uninstalled Cost in 2007\$	Installed Cost Base Year	Installed Cost in 2007\$	Cost Source
A100.CONV1	2		Bale Transport Conveyor	\$400,000	2000	\$800,000	\$1,066,531	\$1,296,000	\$1,727,781	Aden et al. 2002
A100.CONV2	2		Bale Unwrapping Conveyor	\$150,000	2000	\$300,000	\$399,949	\$357,000	\$475,940	Aden et al. 2002
A100.CONV3	1		Belt Press Discharge Conveyor	\$50,000	2000	\$50,000	\$66,658	\$94,500	\$125,984	Aden et al. 2002
A100.SCALE	2		Truck Scales	\$34,000	2000	\$68,000	\$90,655	\$167,960	\$223,918	Aden et al. 2002
A100.FORK1	4	1	Truck Unloading Forklift	\$18,000	2000	\$90,000	\$119,985	\$90,000	\$119,985	Aden et al. 2002
A100.FORK2	4		Bale Moving Forklift	\$18,000	2000	\$72,000	\$95,988	\$72,000	\$95,988	Aden et al. 2002
A100.SLAB	1		Concrete Feedstock-Storage Slab	\$450,655	2000	\$450,655	\$600,797	\$991,441	\$1,321,754	Aden et al. 2002
A100.MAGSEP	1		Magnetic Separator	\$13,863	1998	\$13,863	\$18,700	\$18,022	\$24,310	Aden et al. 2002
A100.A100CHOP.CHGRIN01	4		Chopper	\$105,100	2007	\$420,400	\$420,400	\$1,105,258	\$1,105,258	Aspen Icarus
A100.A100CHOP.CHMIX01	1		Chopper Conveyor	\$61,400	2007	\$61,400	\$61,400	\$185,428	\$185,428	Aspen Icarus
A100.A100CHOP.CHSEP01	1		Chopper Screen with Recycle Conveyor	\$20,800	2007	\$20,800	\$20,800	\$62,816	\$62,816	Aspen Icarus
A100.A100DRY.DRDY01	10		Dryer	\$633,700	2007	\$6,337,000	\$6,337,000	\$15,201,647	\$15,201,647	Aspen Icarus
A100.A100GRIN.GRGRIN01	4		Grinder	\$167,100	2007	\$668,400	\$668,400	\$1,757,266	\$1,757,266	Aspen Icarus
A100.A100GRIN.GRMIX01	1		Grinder Conveyor	\$61,400	2007	\$61,400	\$61,400	\$185,428	\$185,428	Aspen Icarus
A100.A100GRIN.GRSEP01	1		Grinder Screen with Recycle Conveyor	\$20,800	2007	\$20,800	\$20,800	\$62,816	\$62,816	Aspen Icarus
A100					Subtotal	\$9,434,718	\$10,049,464	\$21,647,582	\$22,676,317	
A200.A200COMB.CBCYC01	3		Combustor Cyclone (medium efficiency)	\$35,400	2007	\$106,200	\$106,200	\$320,724	\$320,724	Aspen Icarus
A200.A200COMB.CBCYC02	3		Combustor Cyclone (high efficiency)	\$6,700	2007	\$20,100	\$20,100	\$60,702	\$60,702	Aspen Icarus
A200.A200COMB.CBMIX01	1		Ash Storage Vessel	\$142,800	2007	\$142,800	\$142,800	\$431,256	\$431,256	Aspen Icarus
A200.A200COMB.CBREAC01	1		Combustor - Steam Boiler	\$1,450,500	2007	\$1,450,500	\$1,450,500	\$4,380,510	\$4,380,510	Aspen Icarus
A200.A200CYC.CYC01	2		1st train, medium efficiency cyclone	\$20,300	2007	\$40,600	\$40,600	\$122,612	\$122,612	Aspen Icarus
A200.A200CYC.CYC02	4		1st train, high efficiency cyclone	\$24,900	2007	\$99,600	\$99,600	\$300,792	\$300,792	Aspen Icarus
A200.A200CYC.CYMIX02	1		Char Collector and conveyor	\$84,400	2007	\$84,400	\$84,400	\$254,888	\$254,888	Aspen Icarus
A200.GSREAC01	7		Fluidized Bed Gasifier (Pressurized)	\$1,096,170	2003	\$7,673,191	\$10,028,594	\$14,843,424	\$19,399,838	Larson et al. 2005
A200.GSTANK01	7		Biomass Receiving Hopper	\$71,700	2007	\$501,900	\$501,900	\$1,247,712	\$1,247,712	Aspen Icarus
A200.GSTANK02	7		Lockhopper	\$47,700	2007	\$333,900	\$333,900	\$830,068	\$830,068	Aspen Icarus
A200.GSTANK03	7		Biomass Feeding Bin	\$47,700	2007	\$333,900	\$333,900	\$830,068	\$830,068	Aspen Icarus
A200					Subtotal	\$10,787,091	\$13,142,494	\$23,622,756	\$28,179,170	

Table B-5. Detailed Equipment List for Areas 300, 400, and 500 of LT Scenario

Equipment Number	Number Required	Number Spares	Equipment Name	Original Equip Cost (per unit) in Base Year	Base Year	Total Original Equip Cost (Req'd & Spare) in Base Year	Scaled Uninstalled Cost in 2007\$	Installed Cost Base Year	Installed Cost in 2007\$	Cost Source
A300.A300AGR.AGRarea	1		High Pressure Amine System	\$6,050,000	2005	\$6,050,000	\$6,789,129	\$18,271,000	\$20,503,168	Phillips et al. 2007
A300.A300SUL.SUCOLO1	1		LO-CAT Absorber	\$16,200	2007	\$16,200	\$16,200	\$48,924	\$48,924	Aspen Icarus
A300.A300SUL.SUREAC01	1		LO-CAT Oxidizer Vessel	\$1,000,000	2007	\$1,000,000	\$1,000,000	\$3,020,000	\$3,020,000	Phillips et al. 2007
A300.A300SUL.SUSEP01	1		Sulfur Separator	\$16,200	2007	\$16,200	\$16,200	\$48,924	\$48,924	Aspen Icarus
A300.CLCMP01	2		Carbon Dioxide Compressor	\$1,176,900	2007	\$2,353,800	\$2,353,800	\$2,824,560	\$2,824,560	Aspen Icarus
A300.CLHEAT01	2		Direct Quench Recycle Cooling	\$188,800	2007	\$377,600	\$377,600	\$1,140,352	\$1,140,352	Aspen Icarus
A300.CLHEAT02	1		Venturi Recycle Cooling	\$91,500	2007	\$91,500	\$91,500	\$276,330	\$276,330	Aspen Icarus
A300.CLMIX01	1		Venturi Scrubber	\$26,800	2007	\$26,800	\$26,800	\$80,936	\$80,936	Aspen Icarus
A300.CLSEP03	2		Direct Quench Syngas Cooler	\$188,800	2007	\$377,600	\$377,600	\$1,140,352	\$1,140,352	Aspen Icarus
A300.CLSEP04	1		Venturi Liquid Collection Tank	\$74,500	2007	\$74,500	\$74,500	\$224,990	\$224,990	Aspen Icarus
A300					Subtotal	\$10,384,200	\$11,123,329	\$27,076,368	\$29,308,536	
A400.A400COND.CDHEAT01	1		Syngas Heater	\$60,500	2007	\$60,500	\$60,500	\$182,710	\$182,710	Aspen Icarus
A400.A400COND.CDHEAT02	1		Syngas Pre-heater Furnace	\$1,949,500	2007	\$1,949,500	\$1,949,500	\$5,887,490	\$5,887,490	Aspen Icarus
A400.A400COND.CDHEAT03	1		Reformed Syngas Waste Heat Boiler	\$396,600	2007	\$396,600	\$396,600	\$1,197,732	\$1,197,732	Aspen Icarus
A400.A400COND.CDHEAT04	1		Syngas Cooler #2	\$41,200	2007	\$41,200	\$41,200	\$124,424	\$124,424	Aspen Icarus
A400.A400COND.CDREAC01	1		Steam Methane Reformer	\$1,650,800	2007	\$1,650,800	\$1,650,800	\$4,985,416	\$4,985,416	Aspen Icarus
A400.A400COND.CDREAC02	1		Water Gas Shift Reactor	\$136,600	2007	\$136,600	\$136,600	\$412,532	\$412,532	Aspen Icarus
A400.A400COND.CDSEP01	2		ZnO Sulfur Removal Beds	\$46,400	2007	\$92,800	\$92,800	\$280,256	\$280,256	Aspen Icarus
A400.FSCOMP01	2		Booster Syngas Compressor	\$921,600	2007	\$1,843,200	\$1,843,200	\$2,211,840	\$2,211,840	Aspen Icarus
A400.FSCOMP02	1		Recycle Syngas Booster Compressor	\$725,400	2007	\$725,400	\$725,400	\$870,480	\$870,480	Aspen Icarus
A400.FSCOMP03	1		PSA Booster Compressor	\$1,482,100	2007	\$1,482,100	\$1,482,100	\$1,778,520	\$1,778,520	Aspen Icarus
A400.FSDRUM01	1		PSA Knock-out	\$1,482,100	2007	\$1,482,100	\$1,482,100	\$4,475,942	\$4,475,942	Aspen Icarus
A400.FSHEAT03	1		Syngas Cooler	\$165,200	2007	\$165,200	\$165,200	\$498,904	\$498,904	Aspen Icarus
A400.FSHEAT04	1		Recycle Syngas Pre-heater	\$24,300	2007	\$24,300	\$24,300	\$73,386	\$73,386	Aspen Icarus
A400.FSREAC01	1		Fischer-Tropsch Reactor	\$7,303,889	2003	\$7,303,889	\$9,545,928	\$26,294,000	\$34,365,342	Larson et al. 2005
A400.FSSEP02	12		Pressure Swing Absorption Unit	\$30,500	2007	\$366,000	\$366,000	\$1,105,320	\$1,105,320	Aspen Icarus
A400.FSSEP03	1		FT knock-out Column	\$72,100	2007	\$72,100	\$72,100	\$217,742	\$217,742	Aspen Icarus
A400.FSSEP04	1		Water Separator	\$39,200	2007	\$39,200	\$39,200	\$118,384	\$118,384	Aspen Icarus
A400					Subtotal	\$17,792,289	\$20,034,328	\$50,596,694	\$58,668,036	
A500.HYREAC01	1		Hydrocracking/Hydrotreating Unit	\$7,927,152	2007	\$7,927,152	\$7,927,152	\$23,940,000	\$23,940,000	Robinson & Dolbear 2007
A500.HYTANK01	1		Gasoline 30-day Storage Tank	\$646,300	2007	\$646,300	\$646,300	\$1,951,826	\$1,951,826	Aspen Icarus
A500.HYTANK02	1		Diesel 30-day Storage Tank	\$1,200,700	2007	\$1,200,700	\$1,200,700	\$3,626,114	\$3,626,114	Aspen Icarus
A500					Subtotal	\$9,774,152	\$9,774,152	\$29,517,940	\$29,517,940	

Table B-6. Detailed Equipment List for Areas 600 and 700 of LT Scenario

Equipment Number	Number Required	Number Spares	Equipment Name	Original Equip Cost (per unit) in Base Year	Base Year	Total Original Equip Cost (Req'd & Spare) in Base		Scaled Uninstalled Cost in 2007\$	Installed Cost Base Year	Installed Cost in 2007\$	Cost Source
						Year	Year				
A600.COMBB	1		Combustion Turbine - Electric Generator	\$18,607,700	2007	\$18,607,700	\$18,607,700	\$18,607,700	\$22,329,240	\$22,329,240	Aspen Icarus
A600.CWPUMP	1	1	Cooling Water Pump	\$5,900	2007	\$11,800	\$11,800	\$11,800	\$35,636	\$35,636	Aspen Icarus
A600.ECON1_HRSG	1		Heat Recovery Steam Generator	\$202,200	2007	\$202,200	\$202,200	\$202,200	\$610,644	\$610,644	Aspen Icarus
A600.HPPUMP	1	1	High Pressure Steam Pump	\$266,700	2007	\$533,400	\$533,400	\$533,400	\$1,610,868	\$1,610,868	Aspen Icarus
A600.HPSEP	1		High Pressure Steam/Water Separation	\$107,400	2007	\$107,400	\$107,400	\$107,400	\$324,348	\$324,348	Aspen Icarus
A600.LPEXP_ELECGEN	1		Combined Steam Turbine - Electric Gen.	\$5,056,300	2007	\$5,056,300	\$5,056,300	\$5,056,300	\$6,067,560	\$6,067,560	Aspen Icarus
A600.LPSEP	1		Low Pressure Water/Steam Separation	\$108,800	2007	\$108,800	\$108,800	\$108,800	\$328,576	\$328,576	Aspen Icarus
A600.O2COMP	1		Air Compressor	\$6,331,200	2007	\$6,331,200	\$6,331,200	\$6,331,200	\$7,597,440	\$7,597,440	Aspen Icarus
A600					Subtotal	\$30,958,800	\$30,958,800	\$30,958,800	\$38,904,312	\$38,904,312	
A700.COMP1	2		Air Compressor	\$3,119,600	2007	\$6,239,200	\$6,239,200	\$6,239,200	\$7,487,040	\$7,487,040	Aspen Icarus
A700.COOLER	1		Air Cooler	\$24,300	2007	\$24,300	\$24,300	\$24,300	\$73,386	\$73,386	Aspen Icarus
A700.GOXCLR-1	1		Oxygen Compressor Cooler	\$23,300	2007	\$23,300	\$23,300	\$23,300	\$70,366	\$70,366	Aspen Icarus
A700.GOXCLR-2	1		Oxygen Compressor Cooler	\$23,000	2007	\$23,000	\$23,000	\$23,000	\$69,460	\$69,460	Aspen Icarus
A700.GOXCMP-1	2		Oxygen Compressor	\$1,514,700	2007	\$3,029,400	\$3,029,400	\$3,029,400	\$3,635,280	\$3,635,280	Aspen Icarus
A700.HIGH-P.cond	1		High Pressure Column Condenser	\$20,300	2007	\$20,300	\$20,300	\$20,300	\$61,306	\$61,306	Aspen Icarus
A700.HIGH-P.cond acc	1		High Pressure Column Condenser Accumulator	\$36,300	2007	\$36,300	\$36,300	\$36,300	\$109,626	\$109,626	Aspen Icarus
A700.HIGH-P.reflux pump	1	1	High Pressure Column Reflux Pump	\$14,300	2007	\$28,600	\$28,600	\$28,600	\$34,320	\$34,320	Aspen Icarus
A700.HIGH-P.tower	1		High Pressure Column Tower	\$279,900	2007	\$279,900	\$279,900	\$279,900	\$335,880	\$335,880	Aspen Icarus
A700.INTRC1	1		Air Compressor Intercooler	\$338,300	2007	\$338,300	\$338,300	\$338,300	\$405,960	\$405,960	Aspen Icarus
A700.INTRC2	1		Air Compressor Intercooler	\$304,500	2007	\$304,500	\$304,500	\$304,500	\$919,590	\$919,590	Aspen Icarus
A700.INTRC3	1		Air Compressor Intercooler	\$222,500	2007	\$222,500	\$222,500	\$222,500	\$671,950	\$671,950	Aspen Icarus
A700.LOW-P.reb	2		Low Pressure Column Reboiler	\$19,600	2007	\$39,200	\$39,200	\$39,200	\$118,384	\$118,384	Aspen Icarus
A700.LOW-P.tower	1		Low Pressure Column Tower	\$1,538,900	2007	\$1,538,900	\$1,538,900	\$1,538,900	\$4,647,478	\$4,647,478	Aspen Icarus
A700.TSA	1		Water Knock-out Drum	\$30,100	2007	\$30,100	\$30,100	\$30,100	\$90,902	\$90,902	Aspen Icarus
A700.TURB-1	2		Gas Expander	\$89,200	2007	\$178,400	\$178,400	\$178,400	\$538,768	\$538,768	Aspen Icarus
A700.WK01	1		Water Knock-out Drum	\$64,800	2007	\$64,800	\$64,800	\$64,800	\$195,696	\$195,696	Aspen Icarus
A700					Subtotal	\$12,421,000	\$12,421,000	\$12,421,000	\$19,465,392	\$19,465,392	
Total						\$101,552,251	\$107,503,567	\$107,503,567	\$210,831,043	\$226,719,704	
Total (with BOP)						\$113,738,521	\$120,403,995	\$120,403,995	\$236,130,768	\$253,926,068	

Discounted Cash Flow

High-Temperature Scenario

Table B-7. Discounted Cash Flow Sheet for Construction Period and Years 1-8 of HT Scenario

DCFROR Worksheet											
Year	-2	-1	0	1	2	3	4	5	6	7	8
Fixed Capital Investment	\$50,890,395	\$316,115,651	\$168,595,014								
Working Capital			\$79,028,913								
Loan Payment				\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0
Loan Interest Payment	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0
Loan Principal	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0
GGE (Gallon of Gasoline Equiv.) Sales			\$133,364,635	\$177,819,513	\$177,819,513	\$177,819,513	\$177,819,513	\$177,819,513	\$177,819,513	\$177,819,513	\$177,819,513
Diesel Sales			\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0
By-Product Credit			\$4,173,208	\$5,564,277	\$5,564,277	\$5,564,277	\$5,564,277	\$5,564,277	\$5,564,277	\$5,564,277	\$5,564,277
Plant Performance			1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00
Total Annual Sales			\$137,537,843	\$183,383,791	\$183,383,791	\$183,383,791	\$183,383,791	\$183,383,791	\$183,383,791	\$183,383,791	\$183,383,791
Annual Manufacturing Cost											
Raw Materials			\$44,894,145	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594
SWGS catalysts			\$114,621	\$0	\$0	\$114,621	\$0	\$0	\$114,621	\$0	\$0
Steam reforming catalysts			\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0
ZnO			\$424,410	\$0	\$0	\$424,410	\$0	\$0	\$424,410	\$0	\$0
Pressure Swing Adsorption Packing			\$497,135	\$0	\$0	\$497,135	\$0	\$0	\$497,135	\$0	\$0
FT catalysts			\$7,686,720	\$0	\$0	\$7,686,720	\$0	\$0	\$7,686,720	\$0	\$0
Other Variable Costs			\$11,727,856	\$13,403,264	\$13,403,264	\$13,403,264	\$13,403,264	\$13,403,264	\$13,403,264	\$13,403,264	\$13,403,264
Fixed Operating Costs			\$14,345,785	\$14,345,785	\$14,345,785	\$14,345,785	\$14,345,785	\$14,345,785	\$14,345,785	\$14,345,785	\$14,345,785
Total Product Cost			\$79,690,672	\$79,056,643	\$79,056,643	\$87,779,529	\$79,056,643	\$79,056,643	\$87,779,529	\$79,056,643	\$87,779,529
Annual Depreciation											
General Plant											
DDB			\$128,361,546	\$91,686,818	\$65,490,585	\$46,778,989	\$33,413,564	\$23,866,831	\$17,047,737		
SL			\$64,180,773	\$53,483,977	\$45,843,409	\$40,931,615	\$38,982,491	\$38,982,491	\$38,982,491		
Remaining Value			\$320,903,864	\$229,217,046	\$163,726,461	\$116,947,472	\$83,533,909	\$59,667,078	\$42,619,341		
Actual			\$128,361,546	\$91,686,818	\$65,490,585	\$46,778,989	\$38,982,491	\$38,982,491	\$38,982,491		
Steam Plant											
DDB			\$5,819,551	\$5,383,084	\$4,979,353	\$4,605,902	\$4,260,459	\$3,940,925	\$3,645,355	\$3,371,954	
SL			\$3,879,700	\$3,777,603	\$3,688,410	\$3,612,472	\$3,550,382	\$3,503,044	\$3,471,767	\$3,458,414	
Remaining Value			\$71,774,458	\$66,391,374	\$61,412,021	\$56,806,119	\$52,545,660	\$48,604,736	\$44,959,380	\$41,587,427	
Actual			\$5,819,551	\$5,383,084	\$4,979,353	\$4,605,902	\$4,260,459	\$3,940,925	\$3,645,355	\$3,458,414	
Net Revenue			(\$76,333,925)	\$7,257,245	\$33,857,210	\$44,219,371	\$61,084,198	\$61,403,732	\$52,976,416	\$100,868,733	
Losses Forward				(\$76,333,925)	(\$69,076,681)	(\$35,219,471)	\$0	\$0	\$0	\$0	
Taxable Income			(\$76,333,925)	(\$69,076,681)	(\$35,219,471)	\$8,999,900	\$61,084,198	\$61,403,732	\$52,976,416	\$100,868,733	
Income Tax			\$0	\$0	\$0	\$3,509,961	\$23,822,837	\$23,947,455	\$20,660,802	\$39,338,806	
Annual Cash Income			\$57,847,171	\$104,327,147	\$104,327,147	\$92,094,301	\$80,504,310	\$80,379,692	\$74,943,459	\$64,988,341	
Discount Factor		1.21	1.1	1	0.909090909	0.826446281	0.751314801	0.683013455	0.620921323	0.56447393	
Annual Present Value	\$645,181,377		\$52,588,337	\$86,220,783	\$78,382,530	\$62,901,646	\$49,986,843	\$45,372,241	\$38,457,845	\$30,317,541	
Total Capital Investment + Interest		\$61,577,378	\$347,727,216	\$247,623,927							
Net Present Worth			\$0								

Table B-8. Discounted Cash Flow Sheet for Years 9-20 of HT Scenario

DCFROR Worksheet	9	10	11	12	13	14	15	16	17	18	19	20
Year												
Fixed Capital Investment												
Working Capital												(\$79,028,913)
Loan Payment	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0
Loan Interest Payment	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0
Loan Principal	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0
GGE (Gallon of Gasoline Equiv.) Sales	\$177,819,513	\$177,819,513	\$177,819,513	\$177,819,513	\$177,819,513	\$177,819,513	\$177,819,513	\$177,819,513	\$177,819,513	\$177,819,513	\$177,819,513	\$177,819,513
Diesel Sales	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0
By-Product Credit	\$5,564,277	\$5,564,277	\$5,564,277	\$5,564,277	\$5,564,277	\$5,564,277	\$5,564,277	\$5,564,277	\$5,564,277	\$5,564,277	\$5,564,277	\$5,564,277
Plant Performance	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00
Total Annual Sales	\$183,383,791	\$183,383,791	\$183,383,791	\$183,383,791	\$183,383,791	\$183,383,791	\$183,383,791	\$183,383,791	\$183,383,791	\$183,383,791	\$183,383,791	\$183,383,791
Annual Manufacturing Cost												
Raw Materials	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594
SWGS catalysts	\$0	\$114,621	\$0	\$0	\$114,621	\$0	\$0	\$114,621	\$0	\$0	\$114,621	\$0
Steam reforming catalysts	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0
ZnO	\$0	\$424,410	\$0	\$0	\$424,410	\$0	\$0	\$424,410	\$0	\$0	\$424,410	\$0
Pressure Swing Adsorption Packing	\$0	\$497,135	\$0	\$0	\$497,135	\$0	\$0	\$497,135	\$0	\$0	\$497,135	\$0
FT catalysts	\$0	\$7,686,720	\$0	\$0	\$7,686,720	\$0	\$0	\$7,686,720	\$0	\$0	\$7,686,720	\$0
Other Variable Costs	\$13,403,264	\$13,403,264	\$13,403,264	\$13,403,264	\$13,403,264	\$13,403,264	\$13,403,264	\$13,403,264	\$13,403,264	\$13,403,264	\$13,403,264	\$13,403,264
Fixed Operating Costs	\$14,345,785	\$14,345,785	\$14,345,785	\$14,345,785	\$14,345,785	\$14,345,785	\$14,345,785	\$14,345,785	\$14,345,785	\$14,345,785	\$14,345,785	\$14,345,785
Total Product Cost	\$79,056,643	\$87,779,529	\$79,056,643	\$79,056,643	\$87,779,529	\$79,056,643	\$79,056,643	\$87,779,529	\$79,056,643	\$79,056,643	\$87,779,529	\$79,056,643
Annual Depreciation												
General Plant												
DDB												
SL												
Remaining Value												
Actual												
Steam Plant												
DDB	\$3,119,057	\$2,885,128	\$2,668,743	\$2,468,587	\$2,283,443	\$2,112,185	\$1,953,771	\$1,807,238	\$1,671,696	\$1,546,318	\$1,430,344	\$1,323,069
SL	\$3,458,414	\$3,458,414	\$3,458,414	\$3,458,414	\$3,458,414	\$3,458,414	\$3,458,414	\$3,458,414	\$3,458,414	\$3,458,414	\$3,458,414	\$3,458,414
Remaining Value	\$38,468,370	\$35,583,242	\$32,914,499	\$30,445,912	\$28,162,468	\$26,050,283	\$24,096,512	\$22,289,273	\$20,617,578	\$19,071,260	\$17,640,915	\$16,317,847
Actual	\$3,458,414	\$3,458,414	\$3,458,414	\$3,458,414	\$3,458,414	\$3,458,414	\$3,458,414	\$3,458,414	\$3,458,414	\$3,458,414	\$3,458,414	\$3,458,414
Net Revenue	\$100,868,733	\$92,145,848	\$100,868,733	\$100,868,733	\$92,145,848	\$100,868,733	\$100,868,733	\$92,145,848	\$100,868,733	\$100,868,733	\$92,145,848	\$100,868,733
Losses Forward	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0
Taxable Income	\$100,868,733	\$92,145,848	\$100,868,733	\$100,868,733	\$92,145,848	\$100,868,733	\$100,868,733	\$92,145,848	\$100,868,733	\$100,868,733	\$92,145,848	\$100,868,733
Income Tax	\$39,338,806	\$35,936,881	\$39,338,806	\$39,338,806	\$35,936,881	\$39,338,806	\$39,338,806	\$35,936,881	\$39,338,806	\$39,338,806	\$35,936,881	\$39,338,806
Annual Cash Income	\$64,988,341	\$59,667,381	\$64,988,341	\$64,988,341	\$59,667,381	\$64,988,341	\$64,988,341	\$59,667,381	\$64,988,341	\$64,988,341	\$59,667,381	\$64,988,341
Discount Factor	0.424097618	0.385543289	0.350493899	0.318630818	0.28966438	0.263331254	0.239392049	0.217629136	0.197844669	0.17985879	0.163507991	0.148643628
Annual Present Value	\$27,561,401	\$23,004,358	\$22,778,017	\$20,707,288	\$17,283,515	\$17,113,461	\$15,557,692	\$12,985,361	\$12,857,597	\$11,688,724	\$9,756,094	\$9,660,103
Total Capital Investment + Interest												(\$11,747,144.32)
Net Present Worth												

Low-Temperature Scenario

Table B-9. Discounted Cash Flow Sheet for Construction Period and Years 1-8 of LT Scenario

DCFROR Worksheet													
Year		-2	-1	0	1	2	3	4	5	6	7	8	
Fixed Capital Investment		\$41,888,460	\$259,981,649	\$138,656,880									
Working Capital				\$64,995,412									
Loan Payment				\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0
Loan Interest Payment		\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0
Loan Principal		\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0
GGE (Gallon of Gasoline Equiv.) Sales				\$117,025,289	\$156,033,719	\$156,033,719	\$156,033,719	\$156,033,719	\$156,033,719	\$156,033,719	\$156,033,719	\$156,033,719	\$156,033,719
Diesel Sales				\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0
By-Product Credit				\$4,945,498	\$6,593,998	\$6,593,998	\$6,593,998	\$6,593,998	\$6,593,998	\$6,593,998	\$6,593,998	\$6,593,998	\$6,593,998
Plant Performance				1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00
Total Annual Sales				\$121,970,788	\$162,627,717	\$162,627,717	\$162,627,717	\$162,627,717	\$162,627,717	\$162,627,717	\$162,627,717	\$162,627,717	\$162,627,717
Annual Manufacturing Cost													
Raw Materials				\$44,894,145	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594
WGS catalysts				\$104,732	\$0	\$0	\$104,732	\$0	\$0	\$104,732	\$0	\$0	\$0
Steam reforming catalysts				\$103,412	\$0	\$0	\$103,412	\$0	\$0	\$103,412	\$0	\$0	\$0
ZnO				\$424,410	\$0	\$0	\$424,410	\$0	\$0	\$424,410	\$0	\$0	\$0
Pressure Swing Adsorption Packing				\$497,135	\$0	\$0	\$497,135	\$0	\$0	\$497,135	\$0	\$0	\$0
FT catalysts				\$6,127,680	\$0	\$0	\$6,127,680	\$0	\$0	\$6,127,680	\$0	\$0	\$0
Other Variable Costs				\$11,238,097	\$12,843,539	\$12,843,539	\$12,843,539	\$12,843,539	\$12,843,539	\$12,843,539	\$12,843,539	\$12,843,539	\$12,843,539
Fixed Operating Costs				\$12,404,834	\$12,404,834	\$12,404,834	\$12,404,834	\$12,404,834	\$12,404,834	\$12,404,834	\$12,404,834	\$12,404,834	\$12,404,834
Total Product Cost				\$75,794,444	\$76,555,967	\$76,555,967	\$83,813,336	\$76,555,967	\$76,555,967	\$76,555,967	\$83,813,336	\$76,555,967	\$76,555,967
Annual Depreciation													
General Plant													
DDB				\$104,833,121	\$74,880,801	\$53,486,286	\$38,204,490	\$27,288,922	\$19,492,087	\$13,922,919			
SL				\$52,416,561	\$43,680,467	\$37,440,400	\$33,428,929	\$31,837,075	\$31,837,075	\$31,837,075			
Remaining Value				\$262,082,803	\$187,202,002	\$133,715,716	\$95,511,226	\$68,222,304	\$48,730,217	\$34,807,298			
Actual				\$104,833,121	\$74,880,801	\$53,486,286	\$38,204,490	\$31,837,075	\$31,837,075	\$31,837,075			
Steam Plant													
DDB				\$4,979,012	\$4,605,586	\$4,260,167	\$3,940,654	\$3,645,105	\$3,371,722	\$3,118,843	\$2,884,930		
SL				\$3,319,341	\$3,231,990	\$3,155,679	\$3,090,709	\$3,037,588	\$2,997,087	\$2,970,327	\$2,958,903		
Remaining Value				\$61,407,813	\$56,802,227	\$52,542,060	\$48,601,405	\$44,956,300	\$41,584,577	\$38,465,734	\$35,580,804		
Actual				\$4,979,012	\$4,605,586	\$4,260,167	\$3,940,654	\$3,645,105	\$3,371,722	\$3,118,843	\$2,958,903		
Net Revenue				(\$63,635,790)	\$6,585,363	\$28,325,297	\$36,669,236	\$50,589,570	\$50,862,953	\$43,858,462	\$83,112,848		
Losses Forward					(\$63,635,790)	(\$57,050,426)	(\$28,725,129)	\$0	\$0	\$0	\$0		
Taxable Income				(\$63,635,790)	(\$57,050,426)	(\$28,725,129)	\$7,944,107	\$50,589,570	\$50,862,953	\$43,858,462	\$83,112,848		
Income Tax				\$0	\$0	\$0	\$3,098,202	\$19,729,932	\$19,836,551	\$17,104,800	\$32,414,011		
Annual Cash Income				\$46,176,343	\$86,071,750	\$86,071,750	\$75,716,179	\$66,341,818	\$66,235,199	\$61,709,581	\$53,657,740		
Discount Factor				1.21	1.1	1	0.909090909	0.826446281	0.751314801	0.683013455	0.620921323	0.56447393	0.513158118
Annual Present Value	\$530,655,988			\$41,978,494	\$71,133,678	\$64,666,980	\$51,715,169	\$41,193,049	\$37,388,043	\$31,666,772	\$25,031,732		
Total Capital Investment + Interest		\$50,685,036	\$285,979,814	\$203,652,292									
Net Present Worth				\$0									

Table B-10. Discounted Cash Flow Sheet for Years 9-20 of LT Scenario

DCFROR Worksheet	9	10	11	12	13	14	15	16	17	18	19	20
Year												
Fixed Capital Investment												
Working Capital												(\$64,995,412)
Loan Payment	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0
Loan Interest Payment	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0
Loan Principal	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0
GGE (Gallon of Gasoline Equiv.) Sales	\$156,033,719	\$156,033,719	\$156,033,719	\$156,033,719	\$156,033,719	\$156,033,719	\$156,033,719	\$156,033,719	\$156,033,719	\$156,033,719	\$156,033,719	\$156,033,719
Diesel Sales	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0
By-Product Credit	\$6,593,998	\$6,593,998	\$6,593,998	\$6,593,998	\$6,593,998	\$6,593,998	\$6,593,998	\$6,593,998	\$6,593,998	\$6,593,998	\$6,593,998	\$6,593,998
Plant Performance	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00
Total Annual Sales	\$162,627,717	\$162,627,717	\$162,627,717	\$162,627,717	\$162,627,717	\$162,627,717	\$162,627,717	\$162,627,717	\$162,627,717	\$162,627,717	\$162,627,717	\$162,627,717
Annual Manufacturing Cost												
Raw Materials	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594	\$51,307,594
WGS catalysts	\$0	\$104,732	\$0	\$0	\$104,732	\$0	\$0	\$104,732	\$0	\$0	\$104,732	\$0
Steam reforming catalysts	\$0	\$103,412	\$0	\$0	\$103,412	\$0	\$0	\$103,412	\$0	\$0	\$103,412	\$0
ZnO	\$0	\$424,410	\$0	\$0	\$424,410	\$0	\$0	\$424,410	\$0	\$0	\$424,410	\$0
Pressure Swing Adsorption Packing	\$0	\$497,135	\$0	\$0	\$497,135	\$0	\$0	\$497,135	\$0	\$0	\$497,135	\$0
FT catalysts	\$0	\$6,127,680	\$0	\$0	\$6,127,680	\$0	\$0	\$6,127,680	\$0	\$0	\$6,127,680	\$0
Other Variable Costs	\$12,843,539	\$12,843,539	\$12,843,539	\$12,843,539	\$12,843,539	\$12,843,539	\$12,843,539	\$12,843,539	\$12,843,539	\$12,843,539	\$12,843,539	\$12,843,539
Fixed Operating Costs	\$12,404,834	\$12,404,834	\$12,404,834	\$12,404,834	\$12,404,834	\$12,404,834	\$12,404,834	\$12,404,834	\$12,404,834	\$12,404,834	\$12,404,834	\$12,404,834
Total Product Cost	\$76,555,967	\$83,813,336	\$76,555,967	\$76,555,967	\$83,813,336	\$76,555,967	\$76,555,967	\$83,813,336	\$76,555,967	\$76,555,967	\$83,813,336	\$76,555,967
Annual Depreciation												
General Plant												
DDB												
SL												
Remaining Value												
Actual												
Steam Plant												
DDB	\$2,668,560	\$2,468,418	\$2,283,287	\$2,112,040	\$1,953,637	\$1,807,115	\$1,671,581	\$1,546,212	\$1,430,246	\$1,322,978	\$1,223,755	\$1,131,973
SL	\$2,958,903	\$2,958,903	\$2,958,903	\$2,958,903	\$2,958,903	\$2,958,903	\$2,958,903	\$2,958,903	\$2,958,903	\$2,958,903	\$2,958,903	\$2,958,903
Remaining Value	\$32,912,244	\$30,443,825	\$28,160,538	\$26,048,498	\$24,094,861	\$22,287,746	\$20,616,165	\$19,069,953	\$17,639,706	\$16,316,728	\$15,092,974	\$13,961,001
Actual	\$2,958,903	\$2,958,903	\$2,958,903	\$2,958,903	\$2,958,903	\$2,958,903	\$2,958,903	\$2,958,903	\$2,958,903	\$2,958,903	\$2,958,903	\$2,958,903
Net Revenue	\$83,112,848	\$75,855,478	\$83,112,848	\$83,112,848	\$75,855,478	\$83,112,848	\$83,112,848	\$75,855,478	\$83,112,848	\$83,112,848	\$75,855,478	\$83,112,848
Losses Forward	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0	\$0
Taxable Income	\$83,112,848	\$75,855,478	\$83,112,848	\$83,112,848	\$75,855,478	\$83,112,848	\$83,112,848	\$75,855,478	\$83,112,848	\$83,112,848	\$75,855,478	\$83,112,848
Income Tax	\$32,414,011	\$29,583,637	\$32,414,011	\$32,414,011	\$29,583,637	\$32,414,011	\$32,414,011	\$29,583,637	\$32,414,011	\$32,414,011	\$29,583,637	\$32,414,011
Annual Cash Income	\$53,657,740	\$49,230,744	\$53,657,740	\$53,657,740	\$49,230,744	\$53,657,740	\$53,657,740	\$49,230,744	\$53,657,740	\$53,657,740	\$49,230,744	\$53,657,740
Discount Factor	0.424097618	0.385543289	0.350493899	0.318630818	0.28966438	0.263331254	0.239392049	0.217629136	0.197844669	0.17985879	0.163507991	0.148643628
Annual Present Value	\$22,756,120	\$18,980,583	\$18,806,710	\$17,097,009	\$14,260,393	\$14,129,760	\$12,845,236	\$10,714,044	\$10,615,898	\$9,650,816	\$8,049,620	\$7,975,881
Total Capital Investment + Interest												(\$9,661,153.89)
Net Present Worth												

Appendix C. Scenario Modeling Details

Property Method

The model setup includes a particle size distribution in order to better estimate the solids simulation in the grinding and cyclone operations. It operates globally with the Redlich-Kwong-Soave with Boston-Mathias modification (RKS-BM) property method, which is recommended for medium-temperature refining and gas processing operations including combustion and gasification. During acid-gas absorption and stripping, another property method, ELECNRTL, is used for more accurate simulation. For solids handling, such as in the pretreatment area and cyclones, the SOLIDS property method is used.

Stream/Block Nomenclature

All streams and blocks within the model follow a specific alphanumeric notation with the purpose of clarity and consistency across scenarios and across platforms. Each area within the model (e.g., Area 200 gasification) has a two letter abbreviation (e.g., gasification is GS). These abbreviations are used for naming streams as well as blocks. In addition to serving purposes mentioned above, the notation is descriptive (e.g., the notation REAC describes a block as a reactor). Another example is SGAS, which describes a stream that contains syngas. Aspen Plus software limits block and stream names to be eight characters.

Figure C-1 shows the pattern of notation for a syngas stream in the gasification area.

Area		Number		Description			
G	S	0	1	S	G	A	S

Figure C-1. Stream nomenclature used in model

Similarly, the notation for the first reactor block in the gasification area is shown in Figure C-2.

Area		Description				Number	
G	S	R	E	A	C	0	1

Figure C-2. Block nomenclature used in model

Table C-1 contains the abbreviations for areas, unit operation block descriptions, and stream descriptions.

Table C-1. Detailed Description of Stream and Block Nomenclature

Area	Description	Name	Block	Name	Stream	Name
Plant	All Areas	PL	Reactor	REAC	Biomass	BMAS
A100	Pretreatment	PR	Mixer	MIX	Steam	STM
A100CHOP	Chopping	CH	Heat Mixer	QMX	Flue Gas	FLUE
A100DRY	Drying	DR	Work Mixer	WMX	Syngas	SGAS
A100GRIN	Grinding	GR	Splitter	SPL	Ash	ASH
A200	Gasification	GS,SL	Separator	SEP	Carbon Dioxide	CO₂
A200CYC	Cyclones	CY	Cyclones	CYC	Air	AIR
A200COMB	Combustion	CB	Flash Drum	DRUM	Hydrogen	HYD
A300	Syngas Cleaning	CL	Column	COL	FT Products	FT
A300AGR	Acid Gas Removal	AG	Distillation	DIST	Water	WAT
A300SUL	Sulfur Recovery	SU	Grinder	GRIN	Oxygen	OX
A400	Fuel Synthesis	FS	Dryer	DRY	Sulfur	SUL
A400COND	Syngas Conditioning	CD	Heater	HEAT	Fuel	FUEL
A400MTG	Methanol to Gasoline	MG	Heat Exchanger	HX	Tar	TAR
A500	Hydrocracking	HY	Tank/Hopper	TANK	Char	CHAR
A500	Fuel Separation	SE	Pump	PMP	Acid Gas	AG
A600	Power Generation	PG	Compressor	COMP	Lean MEA Soln.	MEAL
A700	Air Separation Unit		Turbine	TURB	Rich MEA Soln.	MEAR
					Light Gases	LGAS
					Nitrogen	NTGN

A special notation is used for heat and work streams. In the case that the first reactor in the gasification area includes a heat stream leaving the unit, it follows the nomenclature shown in Figure C-3.

Q or W		Area		Block Description			Number
Q	-	G	S	R	E	A	1

Figure C-3. Heat and work stream nomenclature used in model

The Q or W sets the stream apart as a heat or work stream. The block description is limited to three characters and the number is limited to one character.

Aspen Plus Calculator Block Descriptions

High-Temperature Scenario

AIRCOMB

This block calculates the nitrogen that accompanies the oxygen in the air inlet for the combustion of unconverted syngas. Molar nitrogen flow (in kmol/h) is calculated as follows:

$$\dot{M}_{N_2} = \left(\frac{0.79}{0.21}\right) \cdot \dot{M}_{O_2} \quad (\text{eqn. 20})$$

where \dot{M}_{O_2} is molar flow of oxygen in kmol/h.

AMINE

This block calculates the molar flow of monoethanolamine (MEA) needed for the required acid gas removal (CO₂ and H₂S) arriving from syngas quench and FT unconverted syngas recycle stream. The MEA is able to capture 0.35 moles acid gas per mole MEA. Additionally, the MEA is diluted as explained in DILUTH2O.

Molar MEA flow (in kmol/h) is calculated by

$$\dot{M}_{MEA} = (\dot{M}_{CO_2, syn} + \dot{M}_{CO_2, rec} + \dot{M}_{H_2S, syn})/0.35 \quad (\text{eqn. 21})$$

where $\dot{M}_{CO_2, syn}$ is molar flow of CO₂ from the syngas after the syngas quench, $\dot{M}_{CO_2, rec}$ is the molar flow of CO₂ from the unconverted syngas recycle after the FT synthesis, and $\dot{M}_{H_2S, syn}$ is the molar flow rate of H₂S from the syngas quench.

Because the MEA solution in the amine absorption unit is to be 20 wt% concentrated with water, the flow of water must be calculated. Mole flow of water is calculated as

$$\dot{M}_{H_2O} = \frac{\dot{M}_{MEA} * MW_{MEA}/0.20}{MW_{H_2O}} \quad (\text{eqn. 22})$$

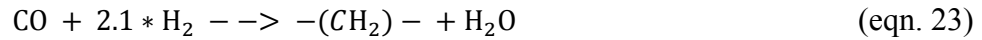
BIOELEM

Because the high-temperature gasifier is modeled at equilibrium, the simulation software requires that all components in the input are located in the conventional stream. Therefore, this block splits the biomass into the following compounds based on its ultimate analysis: carbon, hydrogen, oxygen, sulfur, nitrogen, and ash. Water in the biomass is not affected because it is already a conventional component. Biomass in the exit stream is set to zero.

FTDISTR

This calculator block calculates an alpha chain growth parameter using the equation by Song et al. (2004) for cobalt catalyst. Inlet and outlet streams are defined and calculated. FT products include paraffins from C1 through C20. FT waxes are paraffins at C30.

FT reaction is as follows:



Section 100 sets the CO conversion.

Section 200 calculates the reaction extent (in lbmol) based on an alpha value of 0.9.

-----Section 100-----

Percent conversion of CO is calculated as follows, and then the molar amount of converted CO (COCONV) is calculated knowing the molar amount of CO entering (COIN).

$$\text{PERCEN} = 40$$

$$\text{CONV} = \text{PERCEN} / 100.0$$

$$\text{COCONV} = \text{COIN} * \text{CONV}$$

-----Section 200-----

R1, R2, R3, etc. represent the molar reaction extent (lbmol/h) that is utilized in the FT reactor for each reaction (i.e., $\text{CO} + 3 * \text{H}_2 \rightarrow \text{CH}_4 + \text{H}_2\text{O}$, $2 * \text{CO} + 5 * \text{H}_2 \rightarrow \text{C}_2\text{H}_6 + 2 * \text{H}_2\text{O}$, etc.). The coefficients of each reaction extent are calculated by solving a set of 21 equations shown below in Table C-2 and as described in the Detailed Calculations section of this Appendix.

Table C-2. Reaction Extent Equations for Each Alkane Hydrocarbon

Alkane Component	Equation
C1	$R1 = COCONV * 0.01$
C2	$R2 = COCONV * 0.018/2$
C3	$R3 = COCONV * 0.0243/3$
C4	$R4 = COCONV * 0.02916/4$
C5	$R5 = COCONV * 0.0328/5$
C6	$R6 = COCONV * 0.03543/6$
C7	$R7 = COCONV * 0.0372/7$
C8	$R8 = COCONV * 0.03826/8$
C9	$R9 = COCONV * 0.03874/9$
C10	$R10 = COCONV * 0.03874/10$
C11	$R11 = COCONV * 0.03835/11$
C12	$R12 = COCONV * 0.03766/12$
C13	$R13 = COCONV * 0.03672/13$
C14	$R14 = COCONV * 0.03559/14$
C15	$R15 = COCONV * 0.03432/15$
C16	$R16 = COCONV * 0.03294/16$
C17	$R17 = COCONV * 0.0315/17$
C18	$R18 = COCONV * 0.03002/18$
C19	$R19 = COCONV * 0.02852/19$
C20	$R20 = COCONV * 0.02702/20$
C30	$R21 = COCONV * 0.36473/30$

GRIND

This block calculates the power requirement (in kW) for grinding the biomass from the chop size of 15 mm to the final size of 1 mm. This power requirement data is found in Mani et al. and is for 12% exiting moisture. The correlation was changed from a polynomial (quadratic) regression, which Mani et al. used, to a power regression because the power regression more accurately matched the data. S_{cut} is the final grind size in the units of millimeters.

$$P_{grind} = (28.76 \cdot S_{cut}^{-0.81}) \cdot \dot{m}_{biomass} \quad (\text{eqn. 24})$$

HRSG

This calc block totals the heat transfer areas of all the heat exchangers in A600 Power Generation for use in Aspen Icarus software costing of a heat recovery steam generator, which is estimated as a waste heat boiler.

HUMIDITY

This block sets humidity of the air entering the air separation unit.

HV-101, HV-203, HV-445

This block calculates the lower and higher heating values of the following streams: biomass, syngas, and FT products.

LOCKHOP

This block calculates the CO₂ required for pressurizing the lockhopper. Higman et al. reports 0.09 kg of pressurization gas is required per kg of biomass.

$$XCO2 = 0.09 * BIOMAS \quad (\text{eqn. 25})$$

MEATEMP

This block sets the temperature of the incoming monoethanolamine solution entering the absorber column in the AGR area.

MOISTURE

This block sets the moisture content of the entering biomass to the preprocessing area and sets the biomass moisture content exiting the biomass dryer. Also, the steam loop flow rate for drying the biomass is set at 9 times the amount of moisture removed during the drying process.

Moisture content (% wet basis) of entering biomass feed, XMOIS1 = 25. Inlet mass flow of moisture, WATERI, is computed.

$$WATERI = FEED * XMOIS1/100/(1 - XMOIS1/100) \quad (\text{eqn. 26})$$

Moisture content (% wet basis) of biomass exiting the dryer, XMOIS2 = 10. Mass flow of moisture, WATERO, is computed.

$$WATERO = FEED * XMOIS2/100/(1 - XMOIS2/100) \quad (\text{eqn. 27})$$

Specify steam required to remove moisture, STEAMI.

$$STEAMI = 9 * (WATERI - WATERO) \quad (\text{eqn. 28})$$

O2COMB

Oxygen is required to combust the char and syngas that provides the energy necessary for drying the biomass. A system of stoichiometric combustion reactions is set up to sum all the oxygen required to fully combust the unconverted syngas purge from the FT synthesis outlet. The reactions are shown in Table C-3.

Table C-3. Combustion Reactions to Determine Required Oxygen

Component	Reaction
CO	$CO + 0.5 \cdot O_2 \rightarrow CO_2$
H ₂	$H_2 + 0.5 \cdot O_2 \rightarrow H_2O$
CH ₄	$CH_4 + 2O_2 \rightarrow 2H_2O + CO_2$
C ₂ H ₆	$C_2H_6 + 3.5 \cdot O_2 \rightarrow 3H_2O + 2CO_2$
C ₂ H ₄	$C_2H_4 + 3O_2 \rightarrow 2H_2O + 2CO_2$
C ₂ H ₂	$C_2H_2 + 2.5O_2 \rightarrow H_2O + CO_2$
C ₃ H ₈	$C_3H_8 + 5O_2 \rightarrow 4H_2O + 3CO_2$
C ₄ H ₁₀	$C_4H_{10} + 6.5O_2 \rightarrow 5H_2O + 4CO_2$
C ₅ H ₁₂	$C_5H_{12} + 8O_2 \rightarrow 6H_2O + 5CO_2$
C ₆ H ₁₄	$C_6H_{14} + 9.5O_2 \rightarrow 7H_2O + 6CO_2$
C ₇ H ₁₆	$C_7H_{16} + 11O_2 \rightarrow 8H_2O + 7CO_2$
C ₈ H ₁₈	$C_8H_{18} + 12.5O_2 \rightarrow 9H_2O + 8CO_2$
C ₉ H ₂₀	$C_9H_{20} + 14O_2 \rightarrow 10H_2O + 9CO_2$
Tar	$C_{14}H_{10}(tar) + 16.5O_2 \rightarrow 5H_2O + 14CO_2$
H ₂ S	$H_2S + 1.5O_2 \rightarrow H_2O + SO_2$
NH ₃	$NH_3 + 1.75O_2 \rightarrow 1.5H_2O + NO_2$

The molar flow rate of oxygen entering the combustor is summed and multiplied by a factor of 1.25 in order to combust with 25% excess air as shown in equation 29 below.

$$\begin{aligned} \dot{M}_{O_2,in} = 1.25 \cdot (\dot{M}_{CHAR,in} + 0.5\dot{M}_{CO,in} + 0.5\dot{M}_{H_2,in} + 2\dot{M}_{CH_4,in} + 3.5\dot{M}_{C_2H_6,in} + \\ 3\dot{M}_{C_2H_4,in} + 2.5\dot{M}_{C_2H_2,in} + 5\dot{M}_{C_3H_8,in} + 6.5\dot{M}_{C_4H_{10},in} + 8\dot{M}_{C_5H_{12},in} + \\ 9.5\dot{M}_{C_6H_{14},in} + 11\dot{M}_{C_7H_{16},in} + 12.5\dot{M}_{C_8H_{18},in} + 14\dot{M}_{C_9H_{20},in} + \\ 16.5\dot{M}_{TAR,in} + 1.5\dot{M}_{H_2S,in} + 1.75\dot{M}_{NH_3,in}) \end{aligned} \quad (\text{eqn. 29})$$

O2TURB

This block calculates the molar flow rate of air (oxygen and nitrogen) required to combust syngas obtained from FT synthesis and the fuel gas obtained from Area 500 in the gas turbine of Area 600. A value of 25% excess air is assumed. The calculations are similar to the methodology in *O2COMB*.

OXYSET

This block sets the entering oxygen at 0.35 lb oxygen per lb dry biomass into the gasifier.

$$\dot{m}_{O_2,gas} = 0.35/100 \cdot \dot{m}_{biomass} \quad (\text{eqn. 30})$$

SWGSSTM

This block sets the steam flow into the sour water-gas-shift reactor to be at a ratio of 3:1 water to carbon monoxide. This ratio ensures enough water-gas-shift activity occurs within the reactor.

$$\dot{m}_{STM,addition} = 3.0 \cdot \dot{m}_{CO} - \dot{m}_{H2O} \quad (\text{eqn. 31})$$

Low-Temperature Scenario

AMINE

This block calculates the molar flow of monoethanolamine (MEA) needed for the required acid gas removal (CO₂ and H₂S) arriving from syngas quench and FT unconverted syngas recycle stream. The MEA is able to capture 0.35 moles acid gas per mole MEA. Additionally, the MEA is diluted as explained in *DILUTH2O*.

Molar MEA flow (in kmol/h) is calculated by

$$\dot{M}_{MEA} = (\dot{M}_{CO_2,syn} + \dot{M}_{CO_2,rec} + \dot{M}_{H_2S,syn})/0.35 \quad (\text{eqn. 32})$$

where $\dot{M}_{CO_2,syn}$ is molar flow of CO₂ from the syngas after the syngas quench, $\dot{M}_{CO_2,rec}$ is the molar flow of CO₂ from the unconverted syngas recycle after the FT synthesis, and $\dot{M}_{H_2S,syn}$ is the molar flow rate of H₂S from the syngas quench.

Because the MEA solution in the amine absorption unit is to be 20 wt% concentrated with water, the flow of water must be calculated.

Mole flow of water is calculated as

$$\dot{M}_{H_2O} = \frac{\dot{M}_{MEA} * MW_{MEA}/0.20}{MW_{H_2O}} \quad (\text{eqn. 33})$$

BIOELEM

Same as for the HT scenario.

DILUTH2O

This block sets the MEA solution to be 20% concentrated with water.

FTDISTR

Same as for the high-temperature scenario.

GASYIELD

The following model describes how the fluidized-bed gasifier keeps an elemental mass balance. Experiments performed at Iowa State University provide the initial gasifier product distribution,

and the model adjusts the yields of those experiments in order to balance carbon, hydrogen, sulfur, nitrogen, oxygen, and ash.

The approach taken to balance each element across the gasifier is by “floating” a component of each element. The “floating” component for carbon is the char. All sulfur and nitrogen not found in the char is assumed to form hydrogen sulfide and ammonia, respectively. Therefore, sulfur and nitrogen are balanced. Next, elemental hydrogen is adjusted in the model by either converting diatomic hydrogen to steam or decomposing steam to diatomic hydrogen. Oxygen balance is more complex. Because gasification operates at fuel-rich conditions, diatomic oxygen should not be present in the syngas leaving the gasifier. Therefore, diatomic oxygen cannot be the “floating” component. Instead, oxygen is balanced by adjusting the carbon monoxide or carbon dioxide in the exiting syngas. Because there is one oxygen difference between those two components, the oxygen can be adjusted to help close the balance.

Carbon balance follows the flow chart shown in Figure C-4. If there is less gaseous carbon out than total carbon in, then the difference is made up of char carbon, CCARB. Char is assumed to be composed of 68% carbon with the rest as H, O, N, and S. Ash is considered apart from the char and is considered inert in the model. Because the char is now fixed, the only pathway for sulfur and nitrogen to take is to form hydrogen sulfide and ammonia. Therefore, the sulfur and nitrogen are balanced.

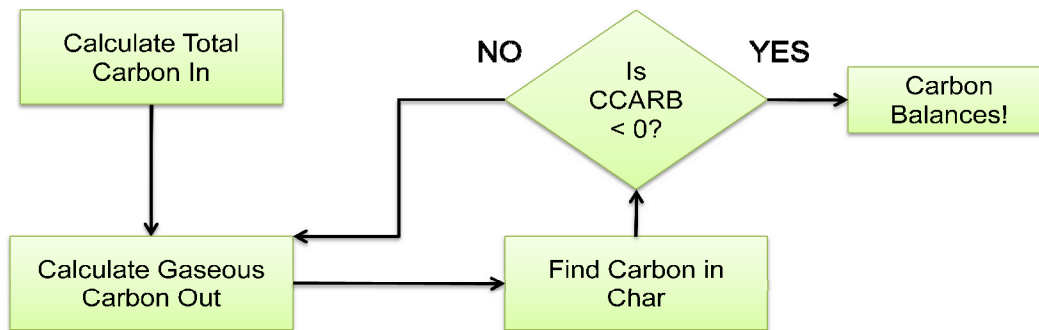


Figure C-4. Decision diagram for carbon balance

Next, as shown in Figure C-5, hydrogen is balanced. Knowing hydrogen in the char and in gaseous products, the hydrogen required (HREQD) is calculated as the sum of those two components. If the hydrogen required is less than hydrogen available (HAVAIL), made up of hydrogen in steam, biomass moisture, and in the biomass itself (THYD), then there is enough hydrogen available to balance. To balance hydrogen, the product yield swings toward either steam or diatomic hydrogen.

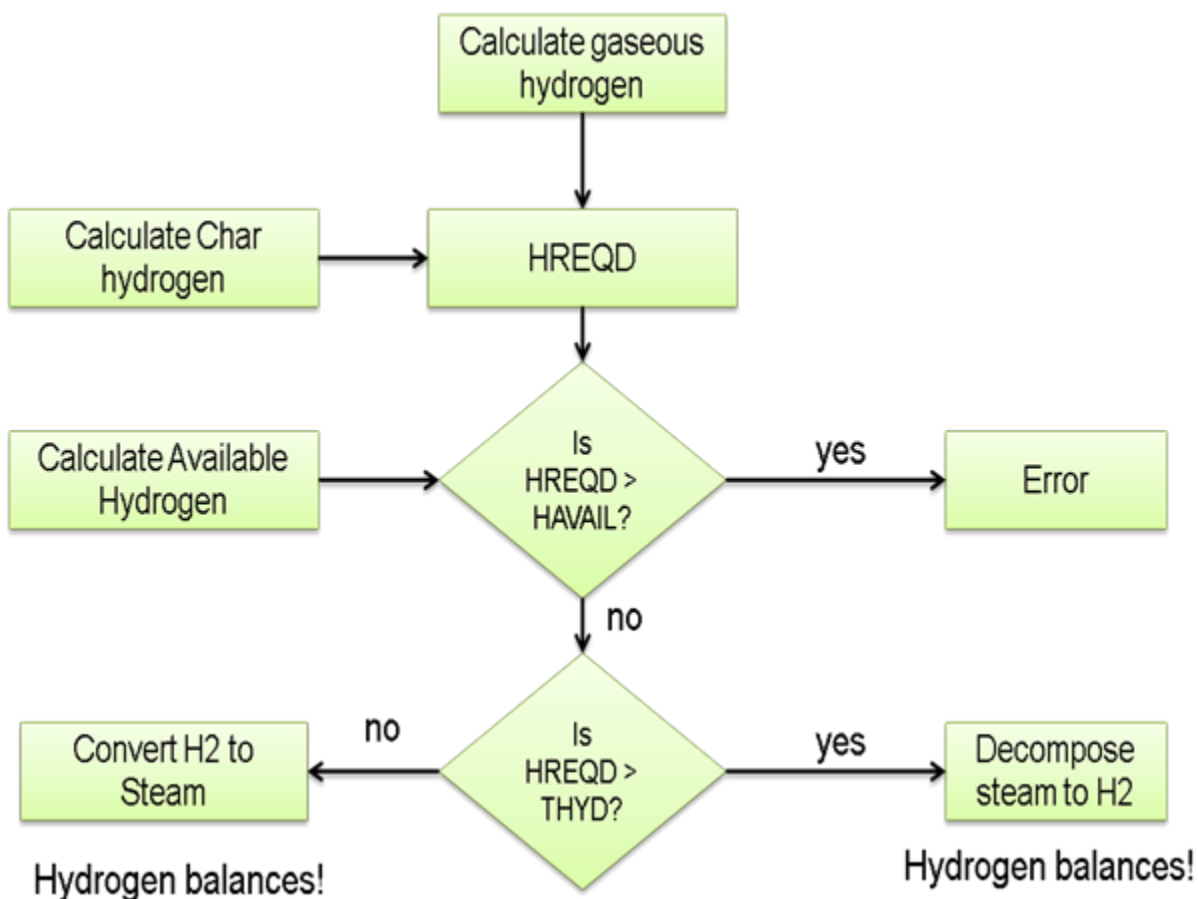


Figure C-5. Decision diagram for hydrogen balance

The only element left to balance is oxygen, which is accomplished by forcing creation of carbon monoxide or creation of carbon dioxide as shown in Figure C-6. The required oxygen (OREQD), made up of oxygen in char and oxygen in syngas, is checked against the available oxygen found in the entering oxygen, steam, and biomass. If there is more oxygen available than required, then the option is to move the excess oxygen to CO₂ by decreasing CO. If there is still oxygen present when CO is decreased to zero, then the yields need to be adjusted because excess oxygen is still present. If there is an oxygen deficit (OREQD > OAVAIL), then CO is increased and CO₂ is decreased. After that, if there is still an oxygen deficit, then insufficient oxygen is present and yields need to be adjusted. When all these steps are completed and no errors are generated, there is an elemental mass balance across the gasifier.

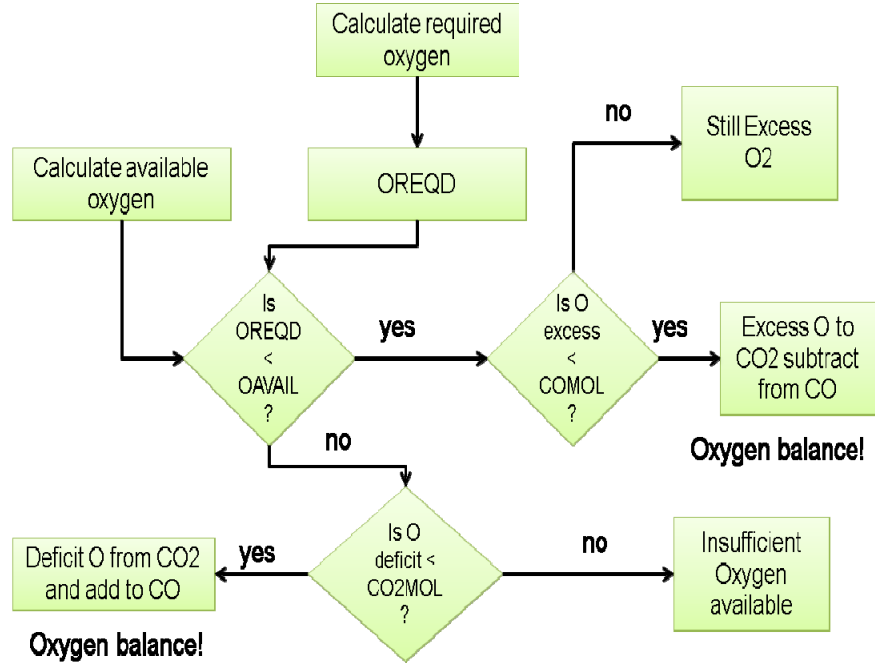


Figure C-6. Decision diagram for oxygen balance

GRIND

This block calculates the power requirement (in kW) for grinding the biomass from the chop size of 12 mm to the final size of 6 mm. This power requirement data is found in Mani et al. and is for 12% exiting moisture. The correlation has been changed from a polynomial regression (which Mani et al. used) to a power regression because the power regression fit the data better. S_{cut} is in millimeters.

$$P_{grind} = (28.76 \cdot S_{cut}^{-0.81}) \cdot \dot{m}_{biomass} \quad (\text{eqn. 34})$$

HUMIDITY

This block sets humidity of the air entering the air separation unit.

HV-101, HV-203, HV-445

This block calculates the lower and higher heating values of the following streams: biomass, syngas, and FT products.

MOISTURE

This block is the same as found in the HT scenario.

O2COMB

This block is the same as found in the HT scenario.

O2TURB

This block is the same as found in the HT scenario.

OXYSET

This block sets the entering oxidizing agents, oxygen and steam, into the gasifier. A linear correlation with temperature, T_{gas} (in Fahrenheit), adapted from Bain for oxygen, is used because as oxygen increases in the gasifier, the temperature increases. Mass flow of oxygen, $\dot{m}_{O_2,gas}$, is in percentage of dry feedstock.

$$\dot{m}_{O_2,gas} = (-11.567 + 0.02375 \cdot T_{gas})/100 \cdot \dot{m}_{biomass} \quad (\text{eqn. 35})$$

The steam feed rate is set at 0.66 lb steam per lb oxygen.

$$\dot{m}_{steam,gas} = 0.66 \cdot \dot{m}_{O_2,gas} \quad (\text{eqn. 36})$$

Because 95% purity oxygen is produced in the air separation unit, argon mass flow is set at 5% of molar oxygen flow.

$$\dot{m}_{argon} = 0.05 \cdot \left(\frac{\dot{m}_{O_2,gas}}{MW_{O_2}} \right) * MW_{Ar} \quad (\text{eqn. 37})$$

Aspen Plus Model Design Specifications

High-Temperature Scenario

DS-1

The exiting temperature of air in the heat exchanger used to pre-cool the air entering the cryogenic distillation column is varied until a net duty of zero is observed.

FSSPL02

This design specification varies the fraction of unconverted syngas that is piped to Area 200 for the combustion of syngas. The syngas, in turn, provides the heat required to dry the biomass.

H2SPLIT

This design spec calculates the required hydrogen that needs to be reserved by the PSA unit for use in Area 500: Hydrocracking. A typical yield from hydrocracking is shown in Table C-4. Because the FT products are hydrogen deficient relative to the final blend, makeup hydrogen is required. The syngas purge amount going to the pressure swing adsorption (PSA) unit is varied so that the calculated delivered hydrogen matches the required hydrogen to Area 500. Without showing the detailed calculations, the basic steps are first calculating the carbon and hydrogen content in the FT product stream. The carbon mass flow is the same as that of the final blend stream flow. Using the blend fractions in Table C-4, the amount of hydrogen is calculated in the final blend and the difference in hydrogen is determined. The difference is multiplied by 1.1 to obtain the delivered hydrogen mass flow rate to hydrocracking area.

Table C-4. Hydroprocessing Product Blend

Component	Mass Fraction
Fuel Gas (methane)	0.034
LPG (propane)	0.088
Gasoline (n-octane)	0.261
Diesel (n-hexadecane)	0.617

O2-101, O2-203, O2-445

These design specifications vary the amount of oxygen inlet to the heating value blocks (HV-101, HV-203, HV-445) so as to be stoichiometric in the combustion of the duplicate stream.

O2-SULF

This design specification varies the amount of oxygen into the LO-CAT oxidizer unit to fully oxidize the H₂S into solid sulfur.

SGSTEMP

The temperature of operation in the sour water-gas-shift reactor is varied until the exiting equilibrium molar ratio of H₂/CO is just above the optimal FT ratio (2.1). A small amount of hydrogen is captured in the PSA unit bringing that ratio down to the optimum for FT synthesis.

Low-Temperature Scenario

DS-1

This design specification is the same as for the HT scenario.

H2SPLIT

This design specification is the same as for the HT scenario.

O2-101, O2-203, O2-445

These design specifications are the same as in the HT scenario.

O2-SULF

This design specification is the same as for the HT scenario.

STMRECOV

Heat can be recovered from the combustion of syngas and char. This specification varies the steam flow rate (stream 280) to bring the combustion flue gas (stream 252) down to 200°C via heat exchanging.

WGSTEMP

The temperature of operation in the water-gas-shift reactor is varied until the exiting equilibrium molar ratio of H₂/CO is just above the optimal FT ratio (2.1). A small amount of hydrogen is captured in the PSA unit, bringing that ratio down to the optimum for FT synthesis.

Detailed Calculations

Aspen Plus Model Calculations and Notes

Outline

Defining Units

Plant Input	$\text{MJ} := 10^6 \text{J}$	$\text{MMcf} := 10^6 \text{ft}^3$
Plant Output		
Carbon Efficiency to Fuels	$\text{kPa} := 10^3 \cdot \text{Pa}$	$\text{Cp} := 100 \text{poise}$
Energy Content		
FT Reaction Conversion Solver	$\rho_{\text{water}} := 1000 \frac{\text{kg}}{\text{m}^3}$	$\text{MW}_{\text{H}_2\text{O}} := 18.02 \frac{\text{gm}}{\text{mol}}$
Equipment Sizing		
Dryer	$\text{kmol} := 1000 \text{mol}$	$\text{MMBTU} := 10^6 \text{BTU}$
Lockhoppers	$\text{lbmol} := \frac{\text{kmol}}{2.2}$	$\text{HHV}_{\text{stover}} := 17.65 \frac{\text{MJ}}{\text{kg}}$
Slag/Char Collection		$\text{HHV}_{\text{stover}} = 7.588 \times 10^3 \frac{\text{BTU}}{\text{lb}}$
PSA Unit	$\text{bbl} := 42 \text{gal}$	
Fuel Storage		
LT Gasifier Cost	$\rho_{\text{gas}} := 737.22 \frac{\text{kg}}{\text{m}^3}$	$100 \frac{\text{kg}}{\text{m}^3} = 6.243 \frac{\text{lb}}{\text{ft}^3}$
FT Reactor Cost		
Acid Gas Removal Area Cost	$\rho_{\text{diesel}} := 840 \frac{\text{kg}}{\text{m}^3}$	$\text{therm} := 100000 \text{BTU}$
A500 Hydroprocessing Area Cost		
Reactors and Catalysts		
Natural Gas Utility Usage	$\text{MMgal} := 10^6 \text{gal}$	$\text{dekatherm} := 10 \text{therm}$
	$\text{kJ} := 1000 \text{J}$	$P_{\text{ref}} \equiv 1 \text{atm}$
	$\text{bpsd} := \frac{42 \text{gal}}{\text{day}}$	$T_{\text{ref}} \equiv 298 \text{K}$
	$\text{PJ} := 10^{15} \text{J}$	$\text{GJ} := 10^9 \text{J}$

Plant Input

Biomass

$$m_{\text{dot_biomass}} := 2000 \frac{\text{tonne}}{\text{day}}$$

$$\text{Availability} := 310 \text{ day}$$

$$\text{Load} := 7446 \text{ hr}$$

Elemental Composition

$$\text{Carbon} \quad \text{Frac}_{\text{C_biomass}} \equiv 0.4728 \quad \text{MW}_{\text{C}} := 12.01 \frac{\text{gm}}{\text{mol}}$$

$$\text{Oxygen} \quad \text{Frac}_{\text{O_biomass}} \equiv 0.4063 \quad \text{MW}_{\text{O}} := 16. \frac{\text{gm}}{\text{mol}}$$

$$\text{Hydrogen} \quad \text{Frac}_{\text{H_biomass}} \equiv 0.0506 \quad \text{MW}_{\text{H}} := 1.01 \frac{\text{gm}}{\text{mol}}$$

$$\text{Sulfur} \quad \text{Frac}_{\text{S_biomass}} \equiv 0.0022 \quad \text{MW}_{\text{S}} := 32.07 \frac{\text{gm}}{\text{mol}}$$

$$\text{Nitrogen} \quad \text{Frac}_{\text{N_biomass}} \equiv 0.008 \quad \text{MW}_{\text{N}} := 14.01 \frac{\text{gm}}{\text{mol}}$$

$$\text{Ash} \quad \text{Frac}_{\text{A_biomass}} \equiv 0.0600$$

Elemental Mass Flow

$$m_{\text{dot_C_in}} := m_{\text{dot_biomass}} \cdot \text{Frac}_{\text{C_biomass}} \quad m_{\text{dot_C_in}} = 945.6 \frac{\text{tonne}}{\text{day}}$$

$$m_{\text{dot_O_in}} := m_{\text{dot_biomass}} \cdot \text{Frac}_{\text{O_biomass}} \quad m_{\text{dot_O_in}} = 812.6 \frac{\text{tonne}}{\text{day}}$$

$$m_{\text{dot_H_in}} := m_{\text{dot_biomass}} \cdot \text{Frac}_{\text{H_biomass}} \quad m_{\text{dot_H_in}} = 101.2 \frac{\text{tonne}}{\text{day}}$$

$$m_{\text{dot_S_in}} := m_{\text{dot_biomass}} \cdot \text{Frac}_{\text{S_biomass}} \quad m_{\text{dot_S_in}} = 4.4 \frac{\text{tonne}}{\text{day}}$$

$$m_{\text{dot_N_in}} := m_{\text{dot_biomass}} \cdot \text{Frac}_{\text{N_biomass}} \quad m_{\text{dot_N_in}} = 16 \frac{\text{tonne}}{\text{day}}$$

$$m_{\text{dot_A_in}} := m_{\text{dot_biomass}} \cdot \text{Frac}_{\text{A_biomass}} \quad m_{\text{dot_A_in}} = 120 \frac{\text{tonne}}{\text{day}}$$

Elemental Mole Flow

$$n_{\text{dot_C_in}} := \frac{m_{\text{dot_C_in}}}{MW_C}$$

$$n_{\text{dot_C_in}} = 911.278 \frac{\text{mol}}{\text{s}}$$

$$n_{\text{dot_O_in}} := \frac{m_{\text{dot_O_in}}}{MW_O}$$

$$n_{\text{dot_O_in}} = 587.818 \frac{\text{mol}}{\text{s}}$$

$$n_{\text{dot_H_in}} := \frac{m_{\text{dot_H_in}}}{MW_H}$$

$$n_{\text{dot_H_in}} = 1160 \frac{\text{mol}}{\text{s}}$$

$$n_{\text{dot_S_in}} := \frac{m_{\text{dot_S_in}}}{MW_S}$$

$$n_{\text{dot_S_in}} = 1.588 \frac{\text{mol}}{\text{s}}$$

$$n_{\text{dot_N_in}} := \frac{m_{\text{dot_N_in}}}{MW_N}$$

$$n_{\text{dot_N_in}} = 13.218 \frac{\text{mol}}{\text{s}}$$

Biomass Moisture

$$\text{moist}_{\text{in}} := 0.25 \quad \text{moist}_{\text{dried}} := 0.10$$

$$m_{\text{dot_moist_in}} := \frac{\text{moist}_{\text{in}} \cdot m_{\text{dot_biomass}}}{1 - \text{moist}_{\text{in}}}$$

$$m_{\text{dot_moist_in}} = 666.667 \frac{\text{tonne}}{\text{day}}$$

$$m_{\text{dot_moist_dried}} := \frac{\text{moist}_{\text{dried}} \cdot m_{\text{dot_biomass}}}{1 - \text{moist}_{\text{dried}}}$$

$$m_{\text{dot_moist_dried}} = 222.222 \frac{\text{tonne}}{\text{day}}$$

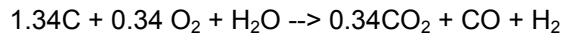
$$\rho_{\text{bulk_stover}} := 100 \frac{\text{kg}}{\text{m}^3}$$

Source: *Kaliyan and Morey, 2005* for 0.66-0.8 mm sized particles

HT Gasifier Steam/Oxygen Addition

Source: *Probstein and Hicks, 2006*

Stoichiometric/thermo-neutral requirement for synthesis gas according to following equation:



Oxygen to Carbon: 0.25

Steam to Carbon: 0.75

$$m_{\dot{O}_2_in} := 0.35 \cdot m_{\dot{biomass}} \quad m_{\dot{O}_2_in} = 700 \frac{\text{tonne}}{\text{day}}$$

$$n_{\dot{O}_2_in} := \frac{m_{\dot{O}_2_in}}{2 \cdot MW_O} \quad n_{\dot{O}_2_in} = 253.183 \frac{\text{mol}}{\text{s}}$$

$$\text{Ratio}_{O_2_to_C} := \frac{n_{\dot{O}_2_in}}{n_{\dot{C}_in}} \quad \text{Ratio}_{O_2_to_C} = 0.278$$

Steam addition ratio is then three times that of oxygen minus the moisture in the biomass

$$n_{\dot{H}_2O_in} := 3 \cdot \text{Ratio}_{O_2_to_C} \cdot n_{\dot{C}_in} - \frac{m_{\dot{moist_dried}}}{MW_{H_2O}} \quad n_{\dot{H}_2O_in} = 616.817 \frac{\text{mol}}{\text{s}}$$

$$m_{\dot{H}_2O_in} := n_{\dot{H}_2O_in} \cdot MW_{H_2O} \quad m_{\dot{H}_2O_in} = 960 \frac{\text{tonne}}{\text{day}}$$

Plant Output

HT Fuel Production

$$\dot{m}_{\text{gasHT}} := 112.78 \frac{\text{tonne}}{\text{day}}$$

$$\dot{m}_{\text{dieselHT}} := 266.5 \frac{\text{tonne}}{\text{day}}$$

$$\dot{v}_{\text{gasHT}} := \frac{\dot{m}_{\text{gasHT}}}{\rho_{\text{gas}}}$$

$$\dot{v}_{\text{dieselHT}} := \frac{\dot{m}_{\text{dieselHT}}}{\rho_{\text{diesel}}}$$

$$\dot{v}_{\text{gasHT}} = 40413 \frac{\text{gal}}{\text{day}}$$

$$\dot{v}_{\text{dieselHT}} = 83812 \frac{\text{gal}}{\text{day}}$$

$$\dot{v}_{\text{gasHT}} = 962 \frac{\text{bbl}}{\text{day}}$$

$$\dot{v}_{\text{dieselHT}} = 1996 \frac{\text{bbl}}{\text{day}}$$

$$\dot{v}_{\text{gasoline_per_year}} := \dot{v}_{\text{gasHT}} \cdot \text{Load}$$

$$\dot{v}_{\text{gasoline_per_year}} = 12.538 \text{ MMgal}$$

$$\dot{v}_{\text{diesel_per_year}} := \dot{v}_{\text{dieselHT}} \cdot \text{Load}$$

$$\dot{v}_{\text{diesel_per_year}} = 26.003 \text{ MMgal}$$

LT Fuel Production

$$\dot{m}_{\text{gasLT}} := 87.12 \frac{\text{tonne}}{\text{day}}$$

$$\dot{m}_{\text{dieselLT}} := 205.86 \frac{\text{tonne}}{\text{day}}$$

$$\dot{v}_{\text{gasLT}} := \frac{\dot{m}_{\text{gasLT}}}{\rho_{\text{gas}}}$$

$$\dot{v}_{\text{dieselLT}} := \frac{\dot{m}_{\text{dieselLT}}}{\rho_{\text{diesel}}}$$

$$\dot{v}_{\text{gasLT}} = 31218 \frac{\text{gal}}{\text{day}}$$

$$\dot{v}_{\text{dieselLT}} = 64741 \frac{\text{gal}}{\text{day}}$$

$$\dot{v}_{\text{gasLT}} = 743 \frac{\text{bbl}}{\text{day}}$$

$$\dot{v}_{\text{dieselLT}} = 1541 \frac{\text{bbl}}{\text{day}}$$

$$\dot{v}_{\text{gasoline_per_yearLT}} := \dot{v}_{\text{gasLT}} \cdot \text{Load}$$

$$\dot{v}_{\text{gasoline_per_yearLT}} = 9.685 \text{ MMgal}$$

$$\dot{v}_{\text{diesel_per_yearLT}} := \dot{v}_{\text{dieselLT}} \cdot \text{Load}$$

$$\dot{v}_{\text{diesel_per_yearLT}} = 20.086 \text{ MMgal}$$

Carbon Efficiency to Fuels

HT Scenario

Gasoline Carbon

$$\text{Frac}_{C_gasoline} := \frac{8 \cdot 12.01}{8 \cdot 12.01 + 18 \cdot 1.01} \quad \text{Frac}_{C_gasoline} = 0.841$$

$$\dot{m}_{dot_gasHT} = 112.78 \frac{\text{tonne}}{\text{day}}$$

$$\dot{m}_{dot_C_gasHT} := \text{Frac}_{C_gasoline} \cdot \dot{m}_{dot_gasHT} \quad \dot{m}_{dot_C_gasHT} = 94.835 \frac{\text{tonne}}{\text{day}}$$

Diesel Carbon

$$\dot{m}_{dot_dieselHT} = 266.5 \frac{\text{tonne}}{\text{day}}$$

$$\text{Frac}_{C_diesel} := \frac{16 \cdot 12.01}{16 \cdot 12.01 + 34 \cdot 1.01} \quad \text{Frac}_{C_diesel} = 0.848$$

$$\dot{m}_{dot_C_dieselHT} := \text{Frac}_{C_diesel} \cdot \dot{m}_{dot_dieselHT} \quad \dot{m}_{dot_C_dieselHT} = 226.096 \frac{\text{tonne}}{\text{day}}$$

$$\dot{m}_{dot_C_outHT} := \dot{m}_{dot_C_gasHT} + \dot{m}_{dot_C_dieselHT}$$

$$\text{C_effHT} = 0.339$$

$$\dot{m}_{dot_C_outHT} = 320.931 \frac{\text{tonne}}{\text{day}} \quad \text{C_effHT} := \frac{\dot{m}_{dot_C_outHT}}{\dot{m}_{dot_C_in}}$$

LT Scenario

Gasoline Carbon

$$m_{\text{dot_gasLT}} = 87.12 \frac{\text{tonne}}{\text{day}}$$

$$m_{\text{dot_C_gasLT}} := \text{Frac}_{\text{C_gasoline}} \cdot m_{\text{dot_gasLT}}$$

$$m_{\text{dot_C_gasLT}} = 73.258 \frac{\text{tonne}}{\text{day}}$$

Diesel Carbon

$$m_{\text{dot_dieselLT}} = 205.86 \frac{\text{tonne}}{\text{day}}$$

$$m_{\text{dot_C_dieselLT}} := \text{Frac}_{\text{C_diesel}} \cdot m_{\text{dot_dieselLT}}$$

$$m_{\text{dot_C_dieselLT}} = 174.649 \frac{\text{tonne}}{\text{day}}$$

$$m_{\text{dot_C_outLT}} := m_{\text{dot_C_gasLT}} + m_{\text{dot_C_dieselLT}}$$

$$m_{\text{dot_C_outLT}} = 247.908 \frac{\text{tonne}}{\text{day}}$$

$$C_{\text{effLT}} := \frac{m_{\text{dot_C_outLT}}}{m_{\text{dot_C_in}}}$$

$$C_{\text{effLT}} = 0.262$$

Energy Content

This section acquires the energy content (on a LHV basis) from the Aspen Plus data and converts to megawatts for use in developing an energy balance

Biomass

$$E_{\text{biomass}} := 1400313 \frac{\text{MJ}}{\text{hr}} \quad E_{\text{biomass}} = 388.976 \text{ MW}$$

Fuel

$$E_{\text{fuelHT}} := 695598 \frac{\text{MJ}}{\text{hr}} \quad E_{\text{fuelHT}} = 193.222 \text{ MW}$$

$$E_{\text{fuelLT}} := 539292 \frac{\text{MJ}}{\text{hr}} \quad E_{\text{fuelLT}} = 149.803 \text{ MW}$$

Char/Tar

$$E_{\text{char_LT}} := 87792 \frac{\text{MJ}}{\text{hr}} \quad E_{\text{char_LT}} = 24.387 \text{ MW}$$

$$E_{\text{tar_LT}} := 16980 \frac{\text{MJ}}{\text{hr}} \quad E_{\text{tar_LT}} = 4.717 \text{ MW}$$

Raw Syngas

$$E_{\text{rawsyngas_HT}} := 1230712 \frac{\text{MJ}}{\text{hr}} \quad E_{\text{rawsyngas_HT}} = 341.864 \text{ MW}$$

$$E_{\text{rawsyngas_LT}} := 964054 \frac{\text{MJ}}{\text{hr}} \quad E_{\text{rawsyngas_LT}} = 267.793 \text{ MW}$$

Energy Loss across the Gasifier

Energy lost across the gasifier is calculated as difference in energy in the biomass and energy in the raw syngas and char (only in LT scenario)

$$E_{\text{gasifierloss_HT}} := E_{\text{biomass}} - E_{\text{rawsyngas_HT}} \quad E_{\text{gasifierloss_HT}} = 47.111 \text{ MW}$$

$$E_{\text{gasifierloss_LT}} := E_{\text{biomass}} - E_{\text{rawsyngas_LT}} - E_{\text{char_LT}} \quad E_{\text{gasifierloss_LT}} = 96.796 \text{ MW}$$

Unconverted Syngas used in A600 Power Generation

$$E_{\text{syngasA600_HT}} := 129332 \frac{\text{MJ}}{\text{hr}} \quad E_{\text{syngasA600_HT}} = 35.926 \text{ MW}$$

$$E_{\text{syngasA600_LT}} := 109708 \frac{\text{MJ}}{\text{hr}} \quad E_{\text{syngasA600_LT}} = 30.474 \text{ MW}$$

Fuel Gas from A500 used in A600 Power Generation

$$E_{\text{fuelgas_HT}} := 104114 \frac{\text{MJ}}{\text{hr}} \quad E_{\text{fuelgas_HT}} = 28.921 \text{ MW}$$

$$E_{\text{fuelgas_LT}} := 80718 \frac{\text{MJ}}{\text{hr}} \quad E_{\text{fuelgas_LT}} = 22.422 \text{ MW}$$

Fischer-Tropsch Product

$$E_{\text{FTliquids_HT}} := 782894 \frac{\text{MJ}}{\text{hr}} \quad E_{\text{FTliquids_HT}} = 217.471 \text{ MW}$$

$$E_{\text{FTliquids_LT}} := 606801 \frac{\text{MJ}}{\text{hr}} \quad E_{\text{FTliquids_LT}} = 168.556 \text{ MW}$$

Electricity Generated

$$E_{\text{elecgenOUT_HT}} := 48.55 \text{ MW} \quad E_{\text{elecgenOUT_LT}} := 40.73 \text{ MW}$$

Net Electricity (exported)

$$E_{\text{elecnet_HT}} := 13.8 \text{ MW} \quad E_{\text{elecnet_LT}} := 16.3 \text{ MW}$$

Power Generation Loss

The loss is the difference between electric generation out and the gas energy in

$$E_{\text{A600losses_HT}} := E_{\text{syngasA600_HT}} + E_{\text{fuelgas_HT}} - E_{\text{elecgenOUT_HT}}$$

$$E_{\text{A600losses_HT}} = 16.296 \text{ MW}$$

$$E_{\text{A600losses_LT}} := E_{\text{syngasA600_LT}} + E_{\text{fuelgas_LT}} - E_{\text{elecgenOUT_LT}}$$

$$E_{\text{A600losses_LT}} = 12.166 \text{ MW}$$

Loss across FT Reactor

$$E_{\text{FTreactorlosses_HT}} := 226737 \frac{\text{MJ}}{\text{hr}} \quad E_{\text{FTreactorlosses_HT}} = 62.983 \text{ MW}$$

$$E_{\text{FTreactorlosses_LT}} := 175128 \frac{\text{MJ}}{\text{hr}} \quad E_{\text{FTreactorlosses_LT}} = 48.647 \text{ MW}$$

Unconverted Syngas used for Biomass Drying

Only in HT scenario

$$E_{\text{biomass_drying_HT}} := 24663 \frac{\text{MJ}}{\text{hr}} \quad E_{\text{biomass_drying_HT}} = 6.851 \text{ MW}$$

Fischer-Tropsch Reaction Conversion Solver

This section solves for the reaction fractional conversion for each reaction in the Fischer-Tropsch reactor. A set of equations is developed and solved. The resulting ϵ values ($\epsilon_1 - \epsilon_{30}$) are used directly in the Aspen Plus conversion reactor block. The reactions in the reactor block are defined as molar extent.

Depending on the alpha chain growth probability, the reactor forms different product composition.

Step 1: choose the expected alpha chain growth value

$$\alpha_{FT} := 0.9$$

Step 2: using the α_{FT} chain growth, the mole fraction of each hydrocarbon in the FT product is calculated.

$M1 := \alpha_{FT}^{1-1} \cdot (1 - \alpha_{FT})$	$M1 = 0.1$	$M11 := \alpha_{FT}^{11-1} \cdot (1 - \alpha_{FT})$	$M11 = 0.035$
$M2 := \alpha_{FT}^{2-1} \cdot (1 - \alpha_{FT})$	$M2 = 0.09$	$M12 := \alpha_{FT}^{12-1} \cdot (1 - \alpha_{FT})$	$M12 = 0.031$
$M3 := \alpha_{FT}^{3-1} \cdot (1 - \alpha_{FT})$	$M3 = 0.081$	$M13 := \alpha_{FT}^{13-1} \cdot (1 - \alpha_{FT})$	$M13 = 0.028$
$M4 := \alpha_{FT}^{4-1} \cdot (1 - \alpha_{FT})$	$M4 = 0.073$	$M14 := \alpha_{FT}^{14-1} \cdot (1 - \alpha_{FT})$	$M14 = 0.025$
$M5 := \alpha_{FT}^{5-1} \cdot (1 - \alpha_{FT})$	$M5 = 0.066$	$M15 := \alpha_{FT}^{15-1} \cdot (1 - \alpha_{FT})$	$M15 = 0.023$
$M6 := \alpha_{FT}^{6-1} \cdot (1 - \alpha_{FT})$	$M6 = 0.059$	$M16 := \alpha_{FT}^{16-1} \cdot (1 - \alpha_{FT})$	$M16 = 0.021$
$M7 := \alpha_{FT}^{7-1} \cdot (1 - \alpha_{FT})$	$M7 = 0.053$	$M17 := \alpha_{FT}^{17-1} \cdot (1 - \alpha_{FT})$	$M17 = 0.019$
$M8 := \alpha_{FT}^{8-1} \cdot (1 - \alpha_{FT})$	$M8 = 0.048$	$M18 := \alpha_{FT}^{18-1} \cdot (1 - \alpha_{FT})$	$M18 = 0.017$
$M9 := \alpha_{FT}^{9-1} \cdot (1 - \alpha_{FT})$	$M9 = 0.043$	$M19 := \alpha_{FT}^{19-1} \cdot (1 - \alpha_{FT})$	$M19 = 0.015$
$M10 := \alpha_{FT}^{10-1} \cdot (1 - \alpha_{FT})$	$M10 = 0.039$	$M20 := \alpha_{FT}^{20-1} \cdot (1 - \alpha_{FT})$	$M20 = 0.014$

All hydrocarbons greater than C20 make up the balance and are modeled using C30.

$$M30 := 1 - \left(M1 + M2 + M3 + M4 + M5 + M6 + M7 + M8 + M9 + M10 \dots \right. \\ \left. + M11 + M12 + M13 + M14 + M15 + M16 + M17 + M18 + M19 + M20 \right)$$

$$M30 = 0.122$$

Step 3: Set up a series of equations to solve along with guess values (required for Mathcad)

For a nominal 1000 moles of CO input, the expected CO output is 600 moles because 40% is converted.

Known $CO_{out} := 600$ $CO_{in} := 1000$ <----- 40% conversion of CO

Guess $\epsilon_1 := 20$ $\epsilon_2 := 20$ $\epsilon_3 := 20$ $\epsilon_4 := 20$ $\epsilon_5 := 20$
 $\epsilon_6 := 20$ $\epsilon_7 := 20$ $\epsilon_8 := 20$ $\epsilon_9 := 20$ $\epsilon_{10} := 20$
 $\epsilon_{11} := 20$ $\epsilon_{12} := 20$ $\epsilon_{13} := 20$ $\epsilon_{14} := 20$ $\epsilon_{15} := 20$
 $\epsilon_{16} := 20$ $\epsilon_{17} := 20$ $\epsilon_{18} := 20$ $\epsilon_{19} := 20$ $\epsilon_{20} := 20$
 $\epsilon_{30} := 20$ $D := 0.1$ <----- This value to be varied until COconv is equal to desired.

A nominal 400 moles of CO are converted in the FT reactor. The sum of the exiting amount of moles in the FT product distribution will not be 400, because moles are not conserved. Mass is conserved, however. Therefore, the variable "D" represents a factor that adjusts all the conversions (ϵ_1 , ϵ_2 , etc.).

The resulting value of D is 0.1 meaning that 40 moles of FT products exit the reactor.

Given

$$D = \left(\begin{array}{l} \epsilon_1 + \frac{1}{2}\epsilon_2 + \frac{1}{3}\epsilon_3 + \frac{1}{4}\epsilon_4 + \frac{1}{5}\epsilon_5 + \frac{1}{6}\epsilon_6 + \frac{1}{7}\epsilon_7 + \frac{1}{8}\epsilon_8 + \frac{1}{9}\epsilon_9 + \frac{1}{10}\epsilon_{10} + \frac{1}{11}\epsilon_{11} + \frac{1}{12}\epsilon_{12} \dots \\ + \frac{1}{13}\epsilon_{13} + \frac{1}{14}\epsilon_{14} + \frac{1}{15}\epsilon_{15} + \frac{1}{16}\epsilon_{16} + \frac{1}{17}\epsilon_{17} + \frac{1}{18}\epsilon_{18} + \frac{1}{19}\epsilon_{19} + \frac{1}{20}\epsilon_{20} + \frac{1}{30}\epsilon_{30} \end{array} \right)$$

$$M1 = \frac{\epsilon_1}{D} \quad M2 = \frac{\frac{1}{2}\epsilon_2}{D} \quad M3 = \frac{\frac{1}{3}\epsilon_3}{D} \quad M4 = \frac{\frac{1}{4}\epsilon_4}{D} \quad M5 = \frac{\frac{1}{5}\epsilon_5}{D} \quad M6 = \frac{\frac{1}{6}\epsilon_6}{D}$$

$$M7 = \frac{\frac{1}{7}\epsilon_7}{D} \quad M8 = \frac{\frac{1}{8}\epsilon_8}{D} \quad M9 = \frac{\frac{1}{9}\epsilon_9}{D} \quad M10 = \frac{\frac{1}{10}\epsilon_{10}}{D} \quad M11 = \frac{\frac{1}{11}\epsilon_{11}}{D} \quad M12 = \frac{\frac{1}{12}\epsilon_{12}}{D}$$

$$M13 = \frac{\frac{1}{13}\epsilon_{13}}{D} \quad M14 = \frac{\frac{1}{14}\epsilon_{14}}{D} \quad M15 = \frac{\frac{1}{15}\epsilon_{15}}{D} \quad M16 = \frac{\frac{1}{16}\epsilon_{16}}{D} \quad M17 = \frac{\frac{1}{17}\epsilon_{17}}{D} \quad M18 = \frac{\frac{1}{18}\epsilon_{18}}{D}$$

$$M19 = \frac{\frac{1}{19}\epsilon_{19}}{D} \quad M20 = \frac{\frac{1}{20}\epsilon_{20}}{D} \quad M30 = \frac{\frac{1}{30}\epsilon_{30}}{D}$$

Solve := Find($\epsilon_1, \epsilon_2, \epsilon_3, \epsilon_4, \epsilon_5, \epsilon_6, \epsilon_7, \epsilon_8, \epsilon_9, \epsilon_{10}, \epsilon_{11}, \epsilon_{12}, \epsilon_{13}, \epsilon_{14}, \epsilon_{15}, \epsilon_{16}, \epsilon_{17}, \epsilon_{18}, \epsilon_{19}, \epsilon_{20}, \epsilon_{30}$)

Solve =

	0
0	0.01
1	0.018

e1 := Solve ₀	e1 = 0.01	e11 := Solve ₁₀	e11 = 0.03835	D = 0.1
e2 := Solve ₁	e2 = 0.018	e12 := Solve ₁₁	e12 = 0.03766	
e3 := Solve ₂	e3 = 0.0243	e13 := Solve ₁₂	e13 = 0.03672	
e4 := Solve ₃	e4 = 0.02916	e14 := Solve ₁₃	e14 = 0.03559	
e5 := Solve ₄	e5 = 0.0328	e15 := Solve ₁₄	e15 = 0.03432	
e6 := Solve ₅	e6 = 0.03543	e16 := Solve ₁₅	e16 = 0.03294	
e7 := Solve ₆	e7 = 0.0372	e17 := Solve ₁₆	e17 = 0.0315	
e8 := Solve ₇	e8 = 0.03826	e18 := Solve ₁₇	e18 = 0.03002	
e9 := Solve ₈	e9 = 0.03874	e19 := Solve ₁₈	e19 = 0.02852	
e10 := Solve ₉	e10 = 0.03874	e20 := Solve ₁₉	e20 = 0.02702	
		e30 := Solve ₂₀	e30 = 0.36473	

Step 4: The guess value of D is varied until the sum of all reaction conversions (ϵ_1, ϵ_2 , etc.) sum to 1.0 as seen below. This means that all 400 moles of CO are converted as expected.

$$\text{CO}_{\text{conv}} := \epsilon_1 + \epsilon_2 + \epsilon_3 + \epsilon_4 + \epsilon_5 + \epsilon_6 + \epsilon_7 + \epsilon_8 + \epsilon_9 + \epsilon_{10} + \epsilon_{11} + \epsilon_{12} + \epsilon_{13} + \epsilon_{14} + \epsilon_{15} + \epsilon_{16} \dots \\ + \epsilon_{17} + \epsilon_{18} + \epsilon_{19} + \epsilon_{20} + \epsilon_{30}$$

$$\text{CO}_{\text{conv}} = 1$$

Step 5: Each value for ϵ is imported into Aspen Plus

Equipment Sizing

Rotary Dryer Source: *Process Engineering Economics* by James Couper, 2003

Typical rpm of rotary dryers $\text{rpm}_{\text{dryer}} := 4$

Typical product of $\text{rpm} \cdot \text{diameter}(\text{feet})$ equals 15-25. Assume value of 25 for larger end

$$D_{\text{dryer}} := \frac{25 \text{ ft}}{\text{rpm}_{\text{dryer}}} \quad D_{\text{dryer}} = 6.25 \text{ ft}$$

Typical residence times are 5-90 minutes and holdup of solids is 7%-8%. Assume 5 minutes and 8%.

$$t_{\text{res}} := 5 \text{ min} \quad \text{holdup} := 0.08$$

Typical exit gas temperature is 10°-20°C above the entering solids.

Feed rate into plant is 2000 ton/day with bulk density of stover equal to 100kg/m³. Water density is accounted for as well.

$$m_{\text{dot_feed}} := 2000 \frac{\text{tonne}}{\text{day}} \quad m_{\text{dot_moist_in}} = 666.667 \frac{\text{tonne}}{\text{day}}$$

$$\rho_{\text{bulk_stover}} = 100 \frac{\text{kg}}{\text{m}^3} \quad \rho_{\text{water}} = 1000 \frac{\text{kg}}{\text{m}^3}$$

$$\text{Volume of solids in dryer} \quad V_{\text{solids}} := \left(\frac{m_{\text{dot_feed}}}{\rho_{\text{bulk_stover}}} + \frac{m_{\text{dot_moist_in}}}{\rho_{\text{water}}} \right) \cdot t_{\text{res}} \quad V_{\text{solids}} = 71.759 \text{ m}^3$$

$$\text{Volume of solids and steam} \quad V_{\text{dryer_total}} := \frac{V_{\text{solids}}}{\text{holdup}} \quad V_{\text{dryer_total}} = 896.991 \text{ m}^3$$

$$\text{Length of theoretical dryer} \quad \text{length}_{\text{dryer}} := \frac{V_{\text{dryer_total}}}{D_{\text{dryer}} \cdot \frac{2 \cdot \pi}{4}} \quad \text{length}_{\text{dryer}} = 314.708 \text{ m}$$

$$\text{Surface area of theoretical dryer} \quad A_{\text{surf_dryer}} := \text{length}_{\text{dryer}} \cdot \pi \cdot D_{\text{dryer}} \quad A_{\text{surf_dryer}} = 1883.4 \text{ m}^2$$

Max surface area as reported by Aspen Icarus is 185 m²; therefore approximately 10 dryers are required.

$$\text{Feed throughput in each dryer (used for Icarus input)} \quad \frac{m_{\text{dot_feed}} + m_{\text{dot_moist_in}}}{10} = 24495.8 \frac{\text{lb}}{\text{hr}}$$

Lockhopper System

Source: *CE IGCC Repowering Project Bins and Lockhoppers*, Combustion Eng. 1993

Note: this report's feedstock is coal

Assumptions from report

- A receiving bin is situated before the lockhopper with a 40 minute residence time
- Design pressure is for 50 psia.
- Cycle time for lockhopper system is designed for 10 minutes resulting in approximately 50,000 cycles per year
- Storage volume for lockhopper and feed bin is assumed to be 10 minutes
- Approximate lockhopper and feed bin vessel thickness is 1.5 inches and design pressure is for 450 psia
- Volume is theoretical + 33%

Residence Time

$$\text{biomass receiving bin} \quad t_{\text{res_rbin}} := 40 \text{ min} \quad \epsilon_{\text{void}} := 25\%$$

$$\text{biomass lockhopper} \quad t_{\text{res_lock}} := 10 \text{ min}$$

$$\text{biomass feed bin} \quad t_{\text{res_fbin}} := 10 \text{ min}$$

$$m_{\text{dot_feed_lock}} := m_{\text{dot_feed}} + m_{\text{dot_moist_dried}}$$

$$m_{\text{dot_feed_lock}} = 2222 \frac{\text{tonne}}{\text{day}}$$

$$\text{Density of feed} \quad \rho_{\text{stover_10\%moist}} := \frac{\rho_{\text{bulk_stover}} \cdot 2000 + \rho_{\text{water}} \cdot 222}{2222}$$

$$\rho_{\text{stover_10\%moist}} = 189.919 \frac{\text{kg}}{\text{m}^3}$$

HT Scenario Lockhopper System (1 train)

$$\text{Volume of biomass receiving bin} \quad V_{\text{r_bin}} := \frac{t_{\text{res_rbin}} \cdot m_{\text{dot_feed_lock}}}{\rho_{\text{stover_10\%moist}}} \cdot \frac{1}{1 - \epsilon_{\text{void}}}$$

$$V_{\text{r_bin}} = 433 \text{ m}^3$$

$$V_{\text{lock}} = 108 \text{ m}^3$$

$$\text{Volume of biomass lockhopper} \quad V_{\text{lock}} := \frac{t_{\text{res_lock}} \cdot m_{\text{dot_feed_lock}}}{\rho_{\text{stover_10\%moist}}} \cdot \frac{1}{1 - \epsilon_{\text{void}}}$$

$$V_{\text{f_bin}} = 108 \text{ m}^3$$

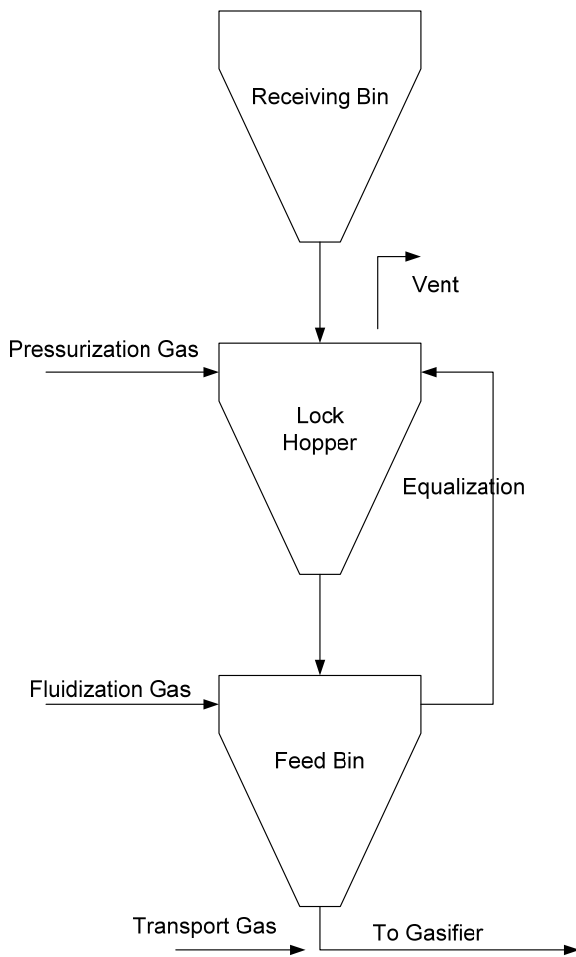
$$\text{Volume of biomass feed bin} \quad V_{\text{f_bin}} := \frac{t_{\text{res_fbin}} \cdot m_{\text{dot_feed_lock}}}{\rho_{\text{stover_10\%moist}}} \cdot \frac{1}{1 - \epsilon_{\text{void}}}$$

LT Scenario Lockhopper System (7 trains)

Volume of biomass receiving bin $V_{r_binLT} := \frac{V_{r_bin}}{7}$ $V_{r_binLT} = 61.909m^3$

Volume of biomass lockhopper $V_{lockLT} := \frac{V_{lock}}{7}$ $V_{lockLT} = 15.477m^3$

Volume of biomass feed bin $V_{f_binLT} := \frac{V_{f_bin}}{7}$ $V_{f_binLT} = 15.477m^3$



Source: Combustion Engineering 1993

Lockhopper Power Consumption

Source: *Techno-Economic Analysis of Hydrogen Production by Gasification of Biomass* by Lau et al. [2002]

Specific power of lockhopper, kW/tonne/day $SP_{lock} := 0.082 \frac{\text{kW}}{\frac{\text{tonne}}{\text{day}}}$

Biomass inlet to gasifier $m_{dot_gasifier} := m_{dot_biomass} + m_{dot_moist_dried}$

Power_{lock} := $SP_{lock} \cdot m_{dot_gasifier}$ Power_{lock} = 182.222 kW

Fly Ash Collection Storage Tank (assume 7 days storage)

$\rho_{ash} := 700 \frac{\text{kg}}{\text{m}^3}$ (assumed) $m_{dot_ash} := 5.88 \frac{\text{tonne}}{\text{day}}$

$V_{tank} := \frac{m_{dot_ash}}{\rho_{ash}} \cdot 7\text{day}$ $V_{tank} = 58.8\text{m}^3$ $V_{tank} = 2.077 \times 10^3 \text{ft}^3$

Slag Separation Drum (5 minute residence time, 20% volume)

$\rho_{slag} := 2700 \frac{\text{kg}}{\text{m}^3}$ $m_{dot_slag} := 114 \frac{\text{tonne}}{\text{day}}$ $\epsilon_{void_slag} := 0.8$

$V_{drum} := \frac{m_{dot_slag}}{\rho_{slag}} \cdot 5\text{min} \cdot \frac{1}{1 - \epsilon_{void_slag}}$ $V_{drum} = 0.733\text{m}^3$

Slag Collection Storage Tank (7 days storage)

$V_{slag_storage} := \frac{m_{dot_slag}}{\rho_{slag}} \cdot 7\text{day}$ $V_{slag_storage} = 295.6\text{m}^3$

Char Collection Storage Bin (1-day residence time, 80% volume)

$\rho_{char} := 2700 \frac{\text{kg}}{\text{m}^3}$ $m_{dot_char} := 214 \frac{\text{tonne}}{\text{day}}$ $\epsilon_{void_char} := 0.2$ $V_{chardrum} = 99.074\text{m}^3$
assume 20% voidage

$V_{chardrum} := \frac{m_{dot_char}}{\rho_{char}} \cdot 1\text{day} \cdot \frac{1}{1 - \epsilon_{void_char}}$ Note: the resulting volumes are used to assist in costing using Aspen Icarus

Pressure Swing Adsorption Unit Sizing

$$P_i := 3.1415 \quad \text{nm} := 10^{-9} \cdot \text{m}$$

References in parentheses are given at the end of this section.

The adsorption unit is 1/3 molsieve and 2/3 activated carbon

	(a)	(b)	(b)
Molsieve 13X	$\text{BulkDens} := 43 \cdot \frac{\text{lb}}{\text{ft}^3}$	$\text{SA} := 1320 \cdot \frac{\text{m}^2}{\text{gm}}$	$\text{PoreVol} := 0.51 \cdot \frac{\text{cm}^3}{\text{gm}}$

Determine dry volumetric flow rate of the syngas stream at atmospheric pressure and 25 deg C

$$\text{VolFlowRate} := (167 - 1) \cdot \frac{\text{kmol}}{\text{hr}} \cdot 22.414 \frac{\text{m}^3}{\text{kmol}} \cdot \frac{14.696 \text{ psi}}{400 \text{ psi}} \cdot \frac{(273.15 + 25) \cdot \text{K}}{273.15 \cdot \text{K}}$$

$$\text{VolFlowRate} = 149.211 \frac{\text{m}^3}{\text{hr}}$$

Mole fraction of components that are adsorbed

$$\text{CO} := 23 \quad \text{CO}_2 := 1 \quad \text{CH}_4 := 1$$

Actual flow rate of components adsorbed

$$\text{FlowRateAds} := \text{VolFlowRate} \cdot \frac{\text{CO} + \text{CO}_2 + \text{CH}_4}{100}$$

$$\text{FlowRateAds} = 37.303 \frac{\text{m}^3}{\text{hr}}$$

Adsorbent capacity

$$\text{AdsCap} := 0.34 \cdot \frac{\text{ft}^3}{\text{lb}} \cdot \frac{14.696 \text{ psi}}{400 \text{ psi}} \cdot \frac{(273.15 + 25) \cdot \text{K}}{273.15 \cdot \text{K}} \quad \text{(d) SCF/lb corrected for P and T to actual cm}^3\text{/gm; PSA occurs at ambient temperature}$$

$$\text{AdsCap} = 0.851 \frac{\text{cm}^3}{\text{gm}}$$

Mass of molsieve required

$$\text{CycleTime} := 5 \cdot \text{min}$$

$$\text{MolSieveMass} := \frac{\text{FlowRateAds} \cdot \text{CycleTime}}{\text{AdsCap}} \quad \text{MolSieveMass} = 3.652 \times 10^3 \text{ kg}$$

Determine volume and length of molsieve bed and activated carbon bed

$$\text{BedVolume} := \frac{\text{MolSieveMass}}{\text{BulkDens}} \quad \text{BedVolume} = 5.302 \text{ m}^3$$

$$\text{Diam} := 4 \text{ ft} \quad \text{Diam} = 1.219 \text{ m} \quad (\text{assumed})$$

$$\text{Length} := \frac{\text{BedVolume}}{\text{Pi} \cdot \text{Diam}^2} \times 4 \quad \text{Length} = 3.725 \text{ ft} \quad (\text{Just molsieve bed})$$

$$\text{RxtrLength} := 3 \cdot \text{Length} \quad (\text{bed is } 1/3 \text{ molsieve, } 2/3 \text{ activated carbon})$$

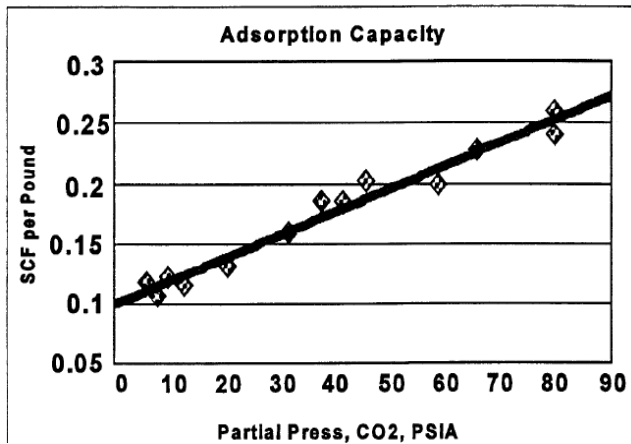
$$\text{RxtrLength} = 11.175 \text{ ft} \quad \text{RxtrLength} = 3.406 \text{ m}$$

$$\text{RxtrVolume} := \text{RxtrLength} \cdot \text{Diam}^2 \cdot 0.25 \cdot \pi \quad \text{RxtrVolume} = 3.977 \text{ m}^3$$

(a) http://www.sigmaaldrich.com/Brands/Aldrich/Tech_Bulletins/AL_143/Molecular_Sieves.html

(b) US Pat 6117810

(d) WO/1998/058726 BULK SEPARATION OF CARBON DIOXIDE FROM METHANE USING NATURAL CLINOPTILOLITE --extrapolate to partial pressure of $\text{CO}_2 + \text{CH}_4 + \text{N}_2 + \text{CO} = 32.6\% \cdot 400 \text{ psi}$



HT Scenario Fuel Storage

Gasoline Storage Tank (30 days storage)

$$m_{\text{dot_gasHT}} = 112.78 \frac{\text{tonne}}{\text{day}}$$

$$v_{\text{dot_gasHT}} = 4.041 \times 10^4 \frac{\text{gal}}{\text{day}}$$

$$V_{\text{gas_tankHT}} := v_{\text{dot_gasHT}} \cdot 30\text{day}$$

$$V_{\text{gas_tankHT}} = 4589\text{m}^3$$

Diesel Storage Tank (30 days storage)

$$m_{\text{dot_dieselHT}} = 266.5 \frac{\text{tonne}}{\text{day}}$$

$$v_{\text{dot_dieselHT}} = 8.381 \times 10^4 \frac{\text{gal}}{\text{day}}$$

$$V_{\text{diesel_tankHT}} := v_{\text{dot_dieselHT}} \cdot 30\text{day}$$

$$V_{\text{diesel_tankHT}} = 9518\text{m}^3$$

Note: the resulting volumes are used to assist in costing using Aspen Icarus software

LT Scenario Fuel Storage

Gasoline Storage Tank (30 days storage)

$$m_{\text{dot_gasLT}} = 87.12 \frac{\text{tonne}}{\text{day}}$$

$$v_{\text{dot_gasLT}} = 3.122 \times 10^4 \frac{\text{gal}}{\text{day}}$$

$$V_{\text{gas_tankLT}} := v_{\text{dot_gasLT}} \cdot 30\text{day}$$

$$V_{\text{gas_tankLT}} = 3545 \text{ m}^3$$

Diesel Storage Tank (30 days storage)

$$m_{\text{dot_dieselLT}} = 205.86 \frac{\text{tonne}}{\text{day}}$$

$$v_{\text{dot_dieselLT}} = 6.474 \times 10^4 \frac{\text{gal}}{\text{day}}$$

$$V_{\text{diesel_tankLT}} := v_{\text{dot_dieselLT}} \cdot 30\text{day}$$

$$V_{\text{diesel_tankLT}} = 7352 \text{ m}^3$$

Note: the resulting volumes are used to assist in costing using Aspen Icarus

LT Gasifier Cost

Source: *Larson et al. 2005* in 2003\$

$$C_{0_gasifier} := 6.41 \cdot 10^6 \text{ \$MM} \quad S_{0_gasifier} := 41.7 \frac{\text{tonne}}{\text{hr}} \quad S_{\max} := 120 \frac{\text{tonne}}{\text{hr}} \quad f := 0.7$$

Biomass throughput of 300 tpd

$$S_{\text{gasifierLT}} := 300 \frac{\text{ton}}{\text{day}} \quad S_{\text{gasifierLT}} = 11.34 \frac{\text{tonne}}{\text{hr}}$$

The cost (\$MM) of one train at 300 ton per day

$$C_{\text{gasifierLT}} := C_{0_gasifier} \cdot \left(\frac{S_{\text{gasifierLT}}}{\frac{\text{tonne}}{\text{hr}}} \cdot \frac{1}{\frac{S_{0_gasifier}}{\frac{\text{tonne}}{\text{hr}}}} \right)^f$$

$$C_{\text{gasifierLT}} = 2.576 \times 10^6 \text{ \$MM}$$

Because 2,205 ton/day we need 7 gasifiers but we can apply the multiple train scaling exponent

$$m_{\text{train}} := 0.9$$

$$C_{\text{gasifierLTtrain}} := C_{\text{gasifierLT}} \cdot 7^{m_{\text{train}}} \quad C_{\text{gasifierLTtrain}} = 1.484 \times 10^7 \text{ \$MM}$$

FT Reactor Costing

Source: *Larson et al. 2005* in 2003\$

$$C_{FT_base} := 10.5 \text{ \$MM} \quad f_{FT2} := 0.72 \quad S_{FT_base} := 2.52 \frac{\text{MMcf}}{\text{hr}}$$

HT Scenario

$$M_{\text{dot_FTHT}} := 13829 \frac{\text{kmol}}{\text{hr}} \quad V_{\text{standard_FTHT}} := M_{\text{dot_FTHT}} \cdot 22.4 \frac{\text{L}}{\text{mol}}$$

$$V_{\text{standard_FTHT}} = 10.939 \frac{\text{MMcf}}{\text{hr}}$$

$$C_{\text{FTHT_reac}} := C_{\text{FT_base}} \cdot \left(\frac{V_{\text{standard_FTHT}}}{S_{\text{FT_base}}} \right)^{f_{\text{FT2}}}$$

$$C_{\text{FTHT_reac}} = 30.217$$

Installed cost \$MM (assume 3.6 install factor consistent with Peters et al.)

LT Scenario

$$M_{\text{dot_FTLT}} := 11400 \frac{\text{kmol}}{\text{hr}} \quad V_{\text{standard_FTLT}} := M_{\text{dot_FTLT}} \cdot 22.4 \frac{\text{L}}{\text{mol}}$$

$$V_{\text{standard_FTLT}} = 9.018 \frac{\text{MMcf}}{\text{hr}}$$

$$C_{\text{FTLT_reac}} := C_{\text{FT_base}} \cdot \left(\frac{V_{\text{standard_FTLT}}}{S_{\text{FT_base}}} \right)^{f_{\text{FT2}}}$$

$$C_{\text{FTLT_reac}} = 26.294$$

Installed cost \$MM (assume 3.6 install factor consistent with Peters et al.)

Acid-Gas Removal Area Cost

Source: Phillips et al. 2007 in 2005\$

Calculated by adding the input syngas streams to the absorber column

$$S_{\text{AGR_base}} := 332910 \frac{\text{lb}}{\text{hr}} \quad f_{\text{AGR}} := 0.65 \quad C_{\text{AGR_base}} := 5446503$$

HT Scenario

$$S_{\text{AGR_HT}} := (2965 + 2308) \frac{\text{tonne}}{\text{day}} \quad S_{\text{AGR_HT}} = 484374 \frac{\text{lb}}{\text{hr}}$$

$$C_{\text{AGR_HT}} := C_{\text{AGR_base}} \cdot \left(\frac{S_{\text{AGR_HT}}}{\frac{\text{lb}}{\text{hr}}} \cdot \frac{1}{\frac{S_{\text{AGR_base}}}{\frac{\text{lb}}{\text{hr}}}} \right)^{f_{\text{AGR}}} \quad C_{\text{AGR_HT}} = 6949808$$

LT Scenario

$$S_{\text{AGR_LT}} := (2070 + 2190) \frac{\text{tonne}}{\text{day}} \quad S_{\text{AGR_LT}} = 391321 \frac{\text{lb}}{\text{hr}}$$

$$C_{\text{AGR_LT}} := C_{\text{AGR_base}} \cdot \left(\frac{S_{\text{AGR_LT}}}{\frac{\text{lb}}{\text{hr}}} \cdot \frac{1}{\frac{S_{\text{AGR_base}}}{\frac{\text{lb}}{\text{hr}}}} \right)^{f_{\text{AGR}}} \quad C_{\text{AGR_LT}} = 6049946$$

A500 Hydroprocessing Area Cost

Source: Robinson et al. 2007 in 2007\$

Note: "bpsd" is barrels per standard day

$$\text{AreaCost}_0 := \frac{4000}{\text{bpsd}} \quad S_{0_HY} := 25000 \text{ bpsd} \quad f_{HY} := 0.65 \text{ (assumed)}$$

$$C_{0_HY} := \text{AreaCost}_0 \cdot S_{0_HY} \quad C_{0_HY} = 100000000$$

HT Scenario

$$m_{\text{dot_FTL_HT}} := 428 \frac{\text{tonne}}{\text{day}} \quad \rho_{\text{FTL}} := 750 \frac{\text{kg}}{\text{m}^3} \quad \text{(from Aspen Plus model)}$$

$$v_{\text{dot_FTL_HT}} := \frac{m_{\text{dot_FTL_HT}}}{\rho_{\text{FTL}}} \quad v_{\text{dot_FTL_HT}} = 3.589 \times 10^3 \text{ bpsd}$$

$$C_{HY_HT} := C_{0_HY} \left(\frac{v_{\text{dot_FTL_HT}}}{\text{bpsd}} \cdot \frac{1}{\frac{S_{0_HY}}{\text{bpsd}}} \right)^{f_{HY}} \quad C_{HY_HT} = 2.832 \times 10^7$$

Power required for A500

$$\text{Power}_{\text{per_bpsd}} := \frac{15 \text{ kW} \cdot \text{hr}}{\text{bpsd} \cdot \text{day}}$$

$$\text{Power}_{\text{areaHT}} := \text{Power}_{\text{per_bpsd}} \cdot v_{\text{dot_FTL_HT}} \quad \text{Power}_{\text{areaHT}} = 2.243 \text{ MW}$$

LT Scenario

$$m_{\text{dot_FTL_LT}} := 330.42 \frac{\text{tonne}}{\text{day}} \quad \text{(from Aspen Plus model)}$$

$$v_{\text{dot_FTL_LT}} := \frac{m_{\text{dot_FTL_LT}}}{\rho_{\text{FTL}}} \quad v_{\text{dot_FTL_LT}} = 2.771 \times 10^3 \text{ bpsd}$$

$$C_{HY_LT} := C_{0_HY} \left(\frac{v_{\text{dot_FTL_LT}}}{\text{bpsd}} \cdot \frac{1}{\frac{S_{0_HY}}{\text{bpsd}}} \right)^{f_{HY}} \quad C_{HY_LT} = 2.394 \times 10^7$$

Power required for A500

$$\text{Power}_{\text{areaLT}} := \text{Power}_{\text{per_bpsd}} \cdot v_{\text{dot_FTL_LT}} \quad \text{Power}_{\text{areaLT}} = 1.732 \text{ MW}$$

Reactors and Catalysts

Fischer-Tropsch Reactor and Cobalt Catalyst

FT reactor volume

Using gas hourly space velocity and actual volumetric flow rate, the volume of the reactor is determined

$$\text{GHSV} = \frac{v_0}{V}$$

$$\text{GHSV}_{\text{FT}} := 100 \cdot \text{hr}^{-1} \quad (\text{assumed})$$

$$v_{\text{rate_actHT}} := 6.298 \frac{\text{m}^3}{\text{s}} \quad (\text{from Aspen Plus model})$$

$$V_{\text{FTHT}} := \frac{v_{\text{rate_actHT}}}{\text{GHSV}_{\text{FT}}} \quad V_{\text{FTHT}} = 226.728 \text{m}^3 \quad V_{\text{FTHT}} = 8.007 \times 10^3 \text{ft}^3$$

$$v_{\text{rate_actLT}} := 5.021 \frac{\text{m}^3}{\text{s}}$$

$$V_{\text{FTLT}} := \frac{v_{\text{rate_actLT}}}{\text{GHSV}_{\text{FT}}} \quad V_{\text{FTLT}} = 180.756 \text{m}^3 \quad V_{\text{FTLT}} = 6.383 \times 10^3 \text{ft}^3$$

FT catalyst cost

$$C_{\text{cost}} := \frac{15}{\text{lb}} \quad (\text{assumed})$$

$$\rho_{\text{Co}} := 64 \frac{\text{lb}}{\text{ft}^3} \quad (\text{assumed})$$

$$C_{\text{vol_cost}} := C_{\text{cost}} \cdot \rho_{\text{Co}}$$

$$C_{\text{vol_cost}} = 960 \frac{1}{\text{ft}^3}$$

Replacement cost of cobalt catalyst

$$C_{\text{total_costHT}} := C_{\text{vol_cost}} \cdot V_{\text{FTHT}} \quad C_{\text{total_costHT}} = 7.687 \times 10^6$$

$$C_{\text{total_costLT}} := C_{\text{vol_cost}} \cdot V_{\text{FTLT}} \quad C_{\text{total_costLT}} = 6.128 \times 10^6$$

Water Gas Shift Reactor and Catalyst

Sour WGS reactor volume (HT scenario)

Using gas hourly space velocity and actual volumetric flow rate, the volume of the reactor is determined

$$\text{GHSV} = \frac{v_0}{V}$$

$$\text{GHSV}_{\text{WGS}} := 1000 \cdot \text{hr}^{-1} \quad (\text{assumed})$$

$$v_{\text{rate_actSWGS}} := 2.008 \frac{\text{m}^3}{\text{s}} \quad (\text{from Aspen Plus})$$

$$V_{\text{SWGS}} := \frac{v_{\text{rate_actSWGS}}}{\text{GHSV}_{\text{WGS}}} \quad V_{\text{SWGS}} = 7.229 \text{m}^3 \quad V_{\text{SWGS}} = 255.283 \text{ft}^3$$

WGS reactor volume (LT scenario)

$$v_{\text{rate_actWGS}} := 1.834 \frac{\text{m}^3}{\text{s}} \quad (\text{from Aspen Plus})$$

$$V_{\text{WGS}} := \frac{v_{\text{rate_actWGS}}}{\text{GHSV}_{\text{WGS}}} \quad V_{\text{WGS}} = 6.602 \text{m}^3 \quad V_{\text{WGS}} = 233.162 \text{ft}^3$$

WGS and SWGS Catalyst Cost

$$\text{CatCost}_{\text{WGS}} := \frac{8}{\text{lb}} \quad \rho_{\text{cat_WGS}} := 56 \frac{\text{lb}}{\text{ft}^3} \quad \rho_{\text{cat_WGS}} = 897.034 \frac{\text{kg}}{\text{m}^3}$$

$$\text{CatCost}_{\text{vol_WGS}} := \text{CatCost}_{\text{WGS}} \cdot \rho_{\text{cat_WGS}} \quad \text{CatCost}_{\text{vol_WGS}} = 448 \frac{1}{\text{ft}^3}$$

Replacement cost of WGS catalyst

$$\text{TotalCatCost}_{\text{SWGS}} := \text{CatCost}_{\text{vol_WGS}} \cdot V_{\text{SWGS}} \quad \text{TotalCatCost}_{\text{SWGS}} = 114367$$

$$\text{TotalCatCost}_{\text{WGS}} := \text{CatCost}_{\text{vol_WGS}} \cdot V_{\text{WGS}} \quad \text{TotalCatCost}_{\text{WGS}} = 104456$$

Steam Methane Reformer Reactor and Catalyst (LT Scenario)

$$\text{GHSV}_{\text{SMR}} := 2600 \text{hr}^{-1} \quad (\text{assumed})$$

$$v_{\text{rate_actSMR}} := 7.082 \frac{\text{m}^3}{\text{s}} \quad (\text{from Aspen Plus model})$$

$$V_{\text{SMR}} := \frac{v_{\text{rate_actSMR}}}{\text{GHSV}_{\text{SMR}}} \quad V_{\text{SMR}} = 9.806 \text{m}^3 \quad V_{\text{SMR}} = 346.29 \text{ft}^3$$

SMR Catalyst Cost

$$\text{CatCost}_{\text{SMR}} := \frac{4.67}{\text{lb}} \quad \rho_{\text{cat_SMR}} := 64 \frac{\text{lb}}{\text{ft}^3} \quad \rho_{\text{cat_SMR}} = 1.025 \times 10^3 \frac{\text{kg}}{\text{m}^3}$$

$$\text{CatCost}_{\text{vol_SMR}} := \text{CatCost}_{\text{SMR}} \cdot \rho_{\text{cat_SMR}} \quad \text{CatCost}_{\text{vol_SMR}} = 298.88 \frac{1}{\text{ft}^3}$$

Replacement cost of SMR catalyst

$$\text{TotalCatCost}_{\text{SMR}} := \text{CatCost}_{\text{vol_SMR}} \cdot V_{\text{SMR}} \quad \text{TotalCatCost}_{\text{SMR}} = 103499$$

Natural Gas Utility Consumption

Annual natural gas requirement at 5% of yearly operating hours

Natural gas properties

Price (Source: Energy Information Administration)

$$\text{HHV}_{\text{ng}} := 54 \frac{\text{MJ}}{\text{kg}} \quad \text{MW}_{\text{ng}} := 16.04 \frac{\text{gm}}{\text{mol}}$$

$$\rho_{\text{ng}} := 22.4 \frac{\text{L}}{\text{mol}}$$

$$\text{Cost}_{\text{ng}} := \frac{6.4}{1000 \text{ ft}^3}$$

HT Scenario

$$\text{Cost}_{\text{ng}} \cdot \frac{\rho_{\text{ng}}}{\text{MW}_{\text{ng}}} = 286.335 \frac{1}{\text{ton}}$$

$$P_{\text{required_plantHT}} := 32.813 \text{ MW} \quad (\text{from Aspen Plus model, includes power required for gas turbine air compressor})$$

$$\text{Eff}_{\text{ng_to_power}} := 0.35 \quad (\text{assumed})$$

$$m_{\text{dot_ngHT}} := \frac{P_{\text{required_plantHT}}}{\frac{\text{Eff}_{\text{ng_to_power}}}{\text{HHV}_{\text{ng}}}} \quad m_{\text{dot_ngHT}} = 1.378 \times 10^4 \frac{\text{lb}}{\text{hr}}$$

Annual natural gas requirement

$$M_{\text{ngHT}} := m_{\text{dot_ngHT}} \cdot \text{Availability} \cdot 0.05 \quad \boxed{M_{\text{ngHT}} = 2563 \text{ ton}}$$

Average flow rate of natural gas

$$m_{\text{dot_ng_5\%HT}} := \frac{M_{\text{ngHT}}}{8760 \text{ hr}} \quad m_{\text{dot_ng_5\%HT}} = 585.14 \frac{\text{lb}}{\text{hr}}$$

LT Scenario

$P_{\text{required_plantLT}} := 24.3\text{MW}$ (from Aspen Plus model, includes power required for gas turbine air compressor)

$$m_{\text{dot_ngLT}} := \frac{\frac{P_{\text{required_plantLT}}}{\text{Eff}_{\text{ng_to_power}}}}{\text{HHV}_{\text{ng}}} \quad m_{\text{dot_ngLT}} = 1.02 \times 10^4 \frac{\text{lb}}{\text{hr}}$$

Annual natural gas requirement

$$M_{\text{ngLT}} := m_{\text{dot_ngLT}} \cdot \text{Availability} \cdot 0.05 \quad \boxed{M_{\text{ngLT}} = 1898 \text{ ton}}$$

Average flow rate of natural gas

$$m_{\text{dot_ng_5\%LT}} := \frac{M_{\text{ngLT}}}{8760 \text{ hr}} \quad m_{\text{dot_ng_5\%LT}} = 433.331 \frac{\text{lb}}{\text{hr}}$$

Appendix D. Process Flow Diagrams

High-Temperature Scenario

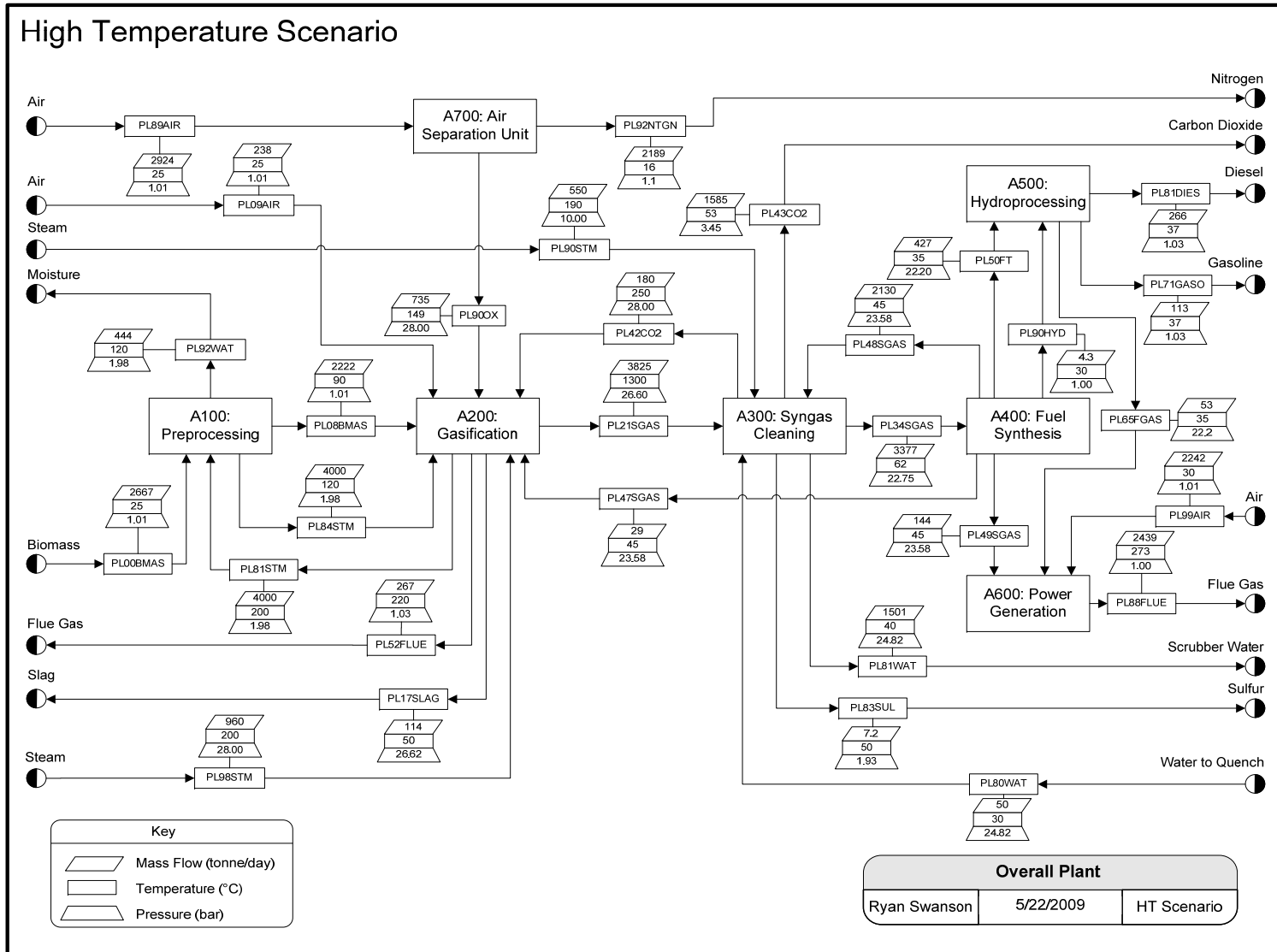


Figure D-1. Overall plant area process flow diagram for HT scenario

High Temperature Scenario: Area 100

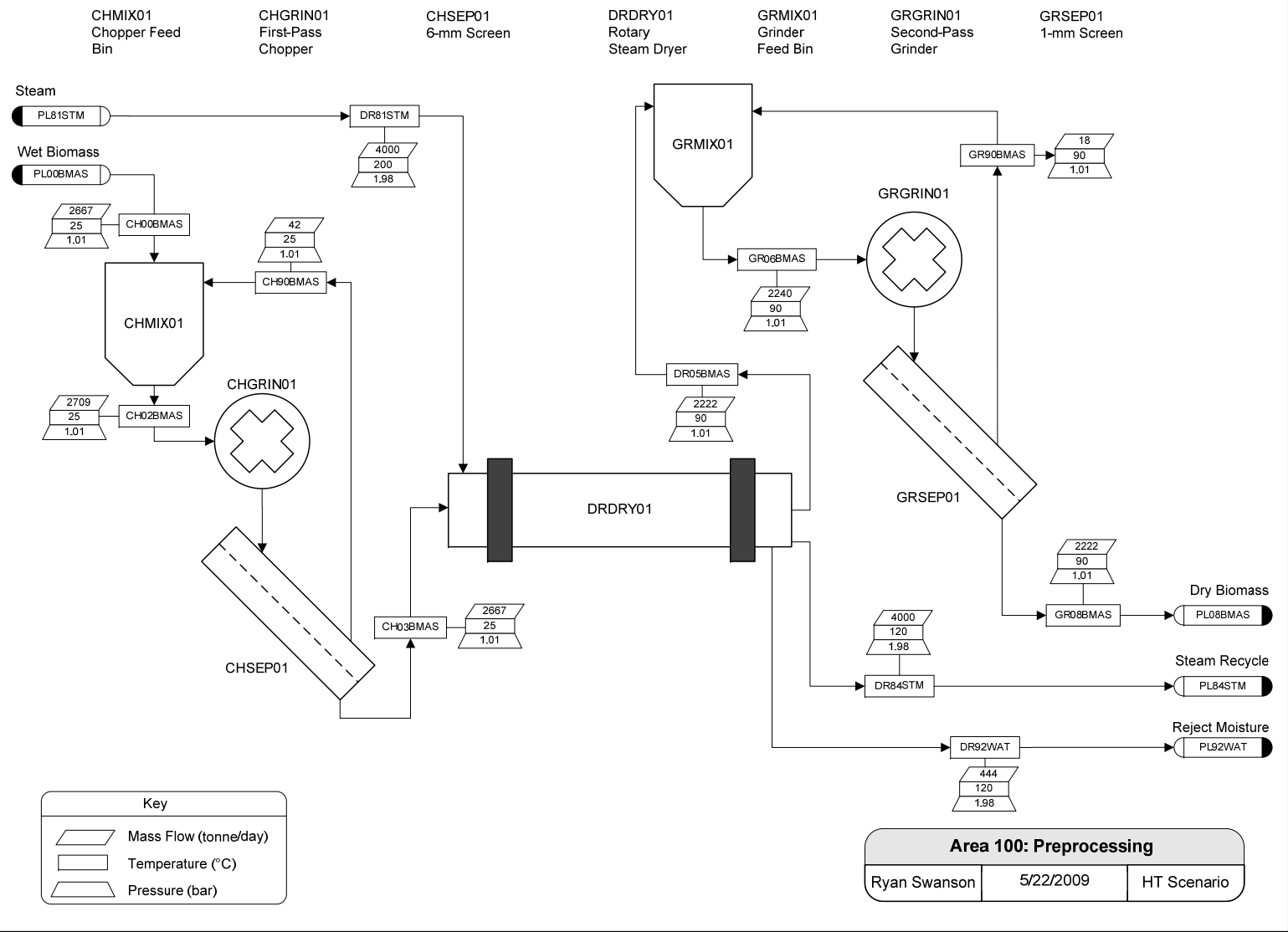


Figure D-2. Preprocessing area process flow diagram for HT scenario

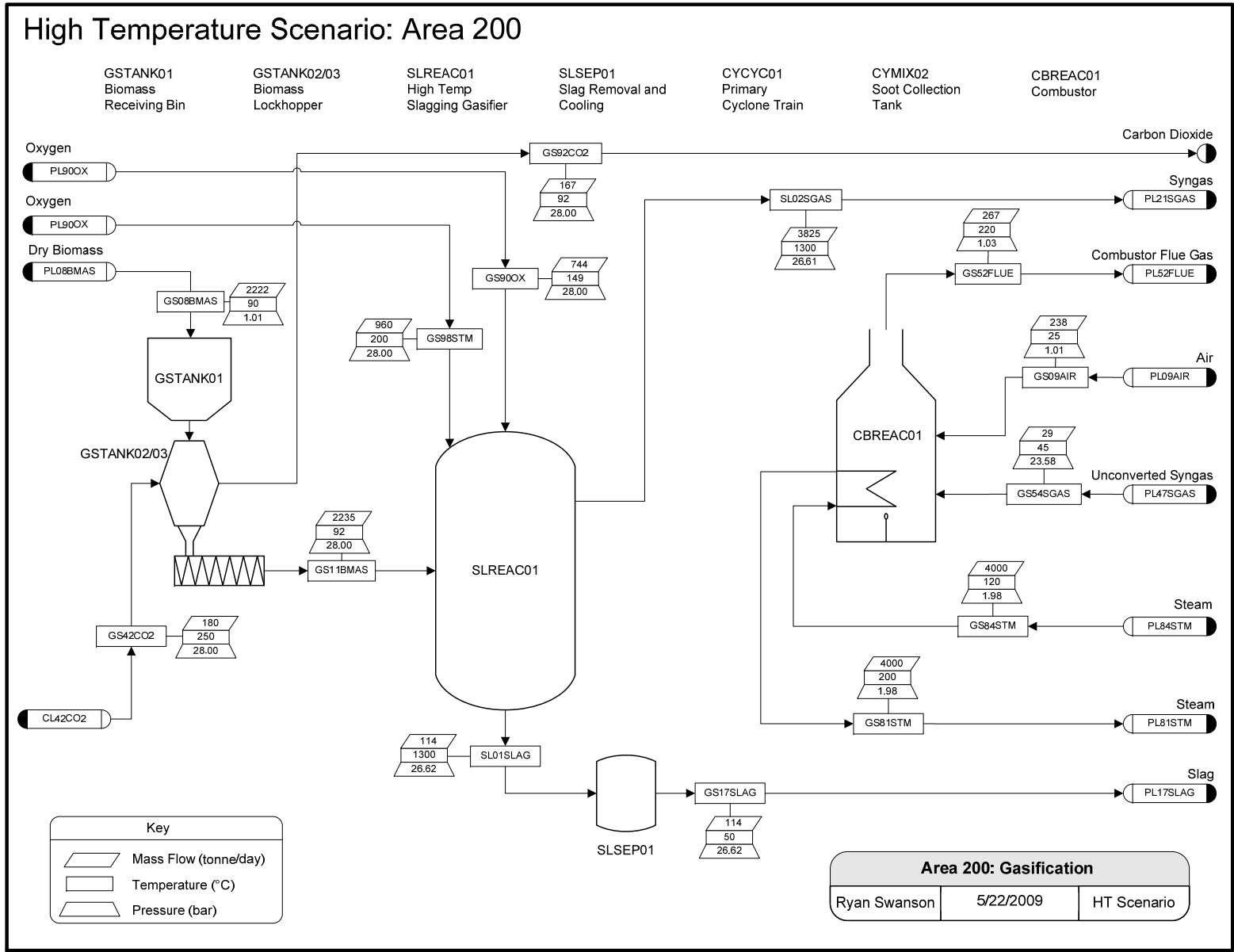


Figure D-3. Gasification area process flow diagram for HT scenario

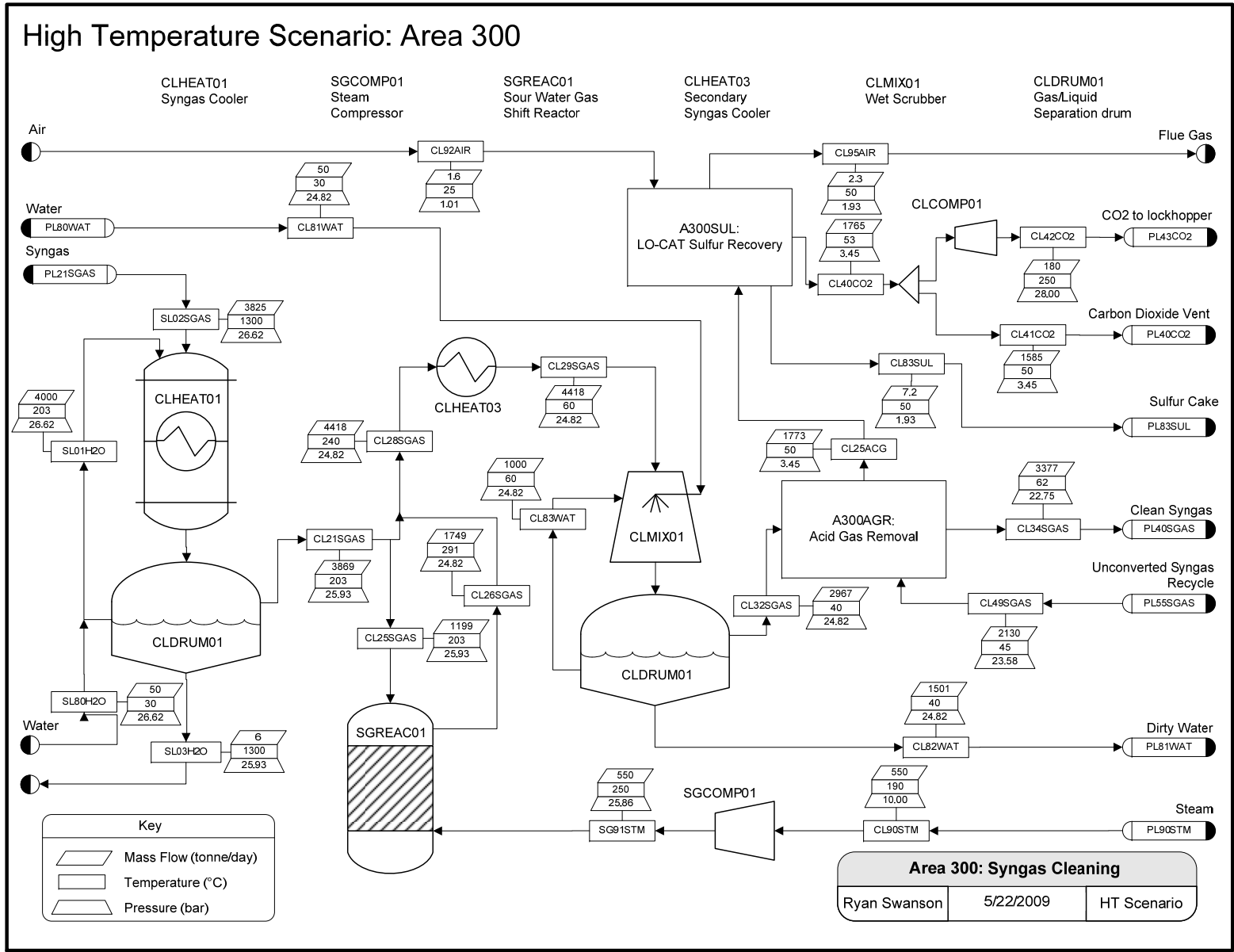


Figure D-4. Syngas cleaning area process flow diagram for HT scenario

High Temperature Scenario: Area 300AGR

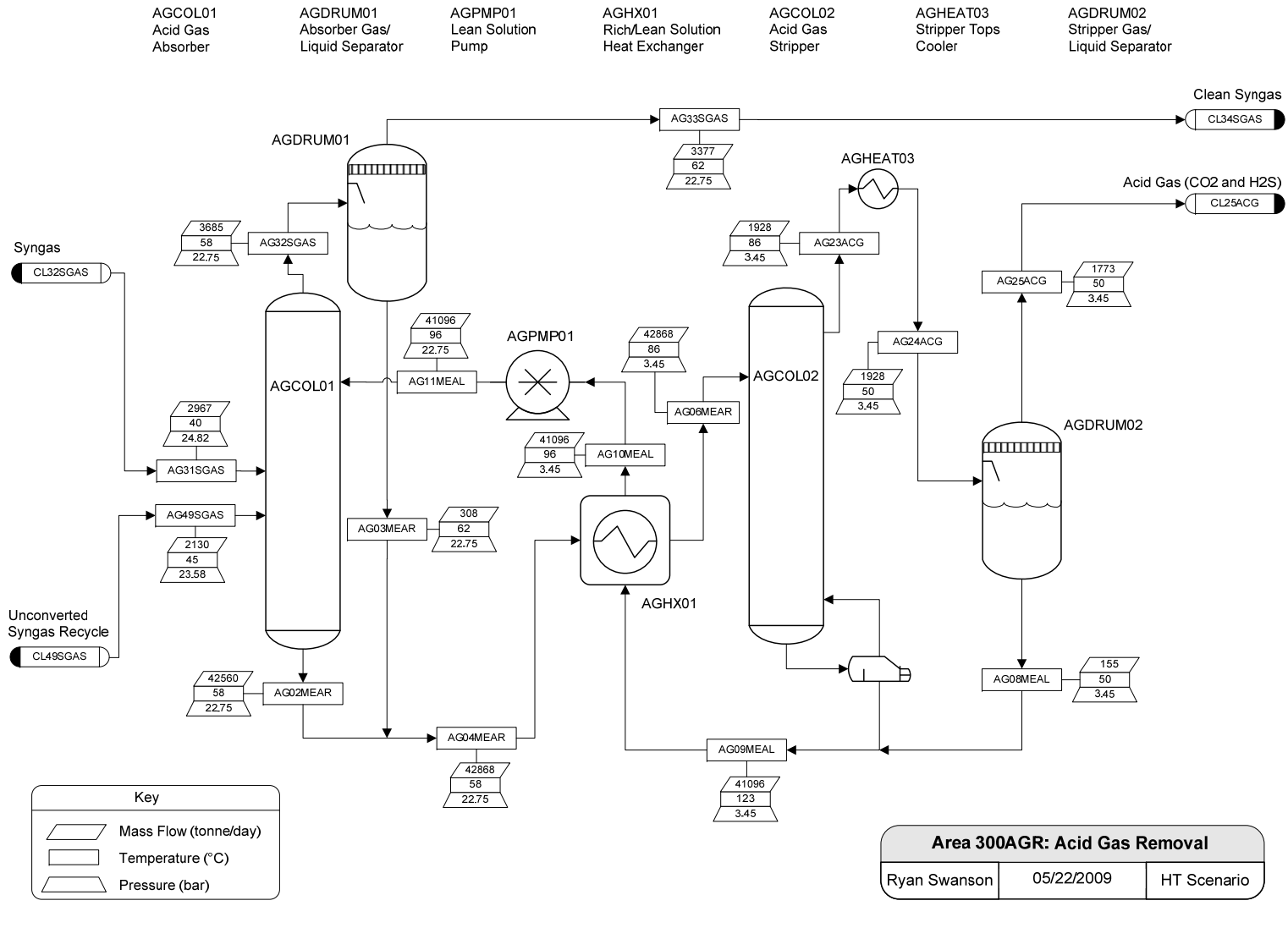


Figure D-5. Acid gas removal area process flow diagram for HT scenario

High Temperature Scenario: Area 300SUL

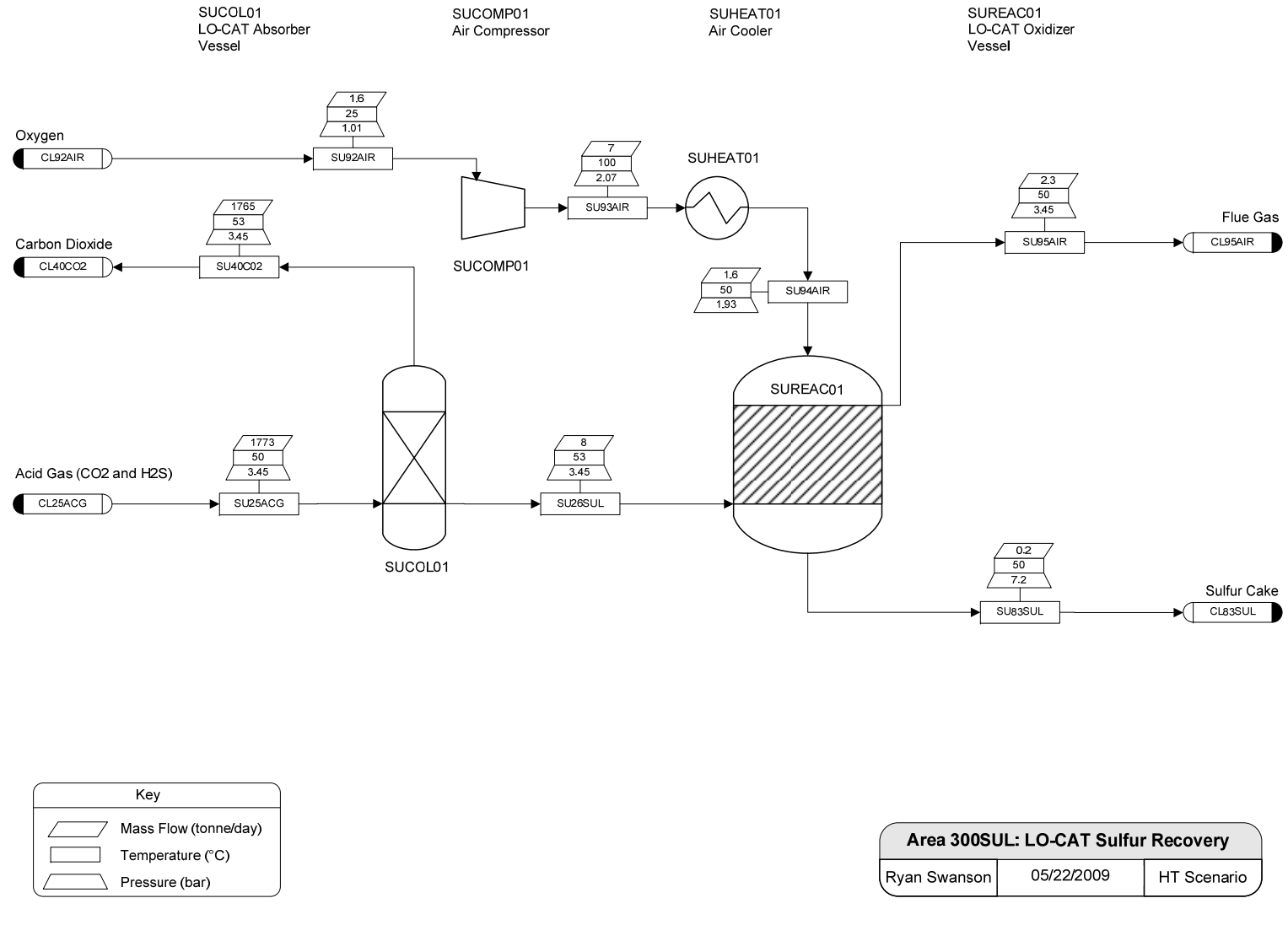


Figure D-6. Sulfur recovery area process flow diagram for HT scenario

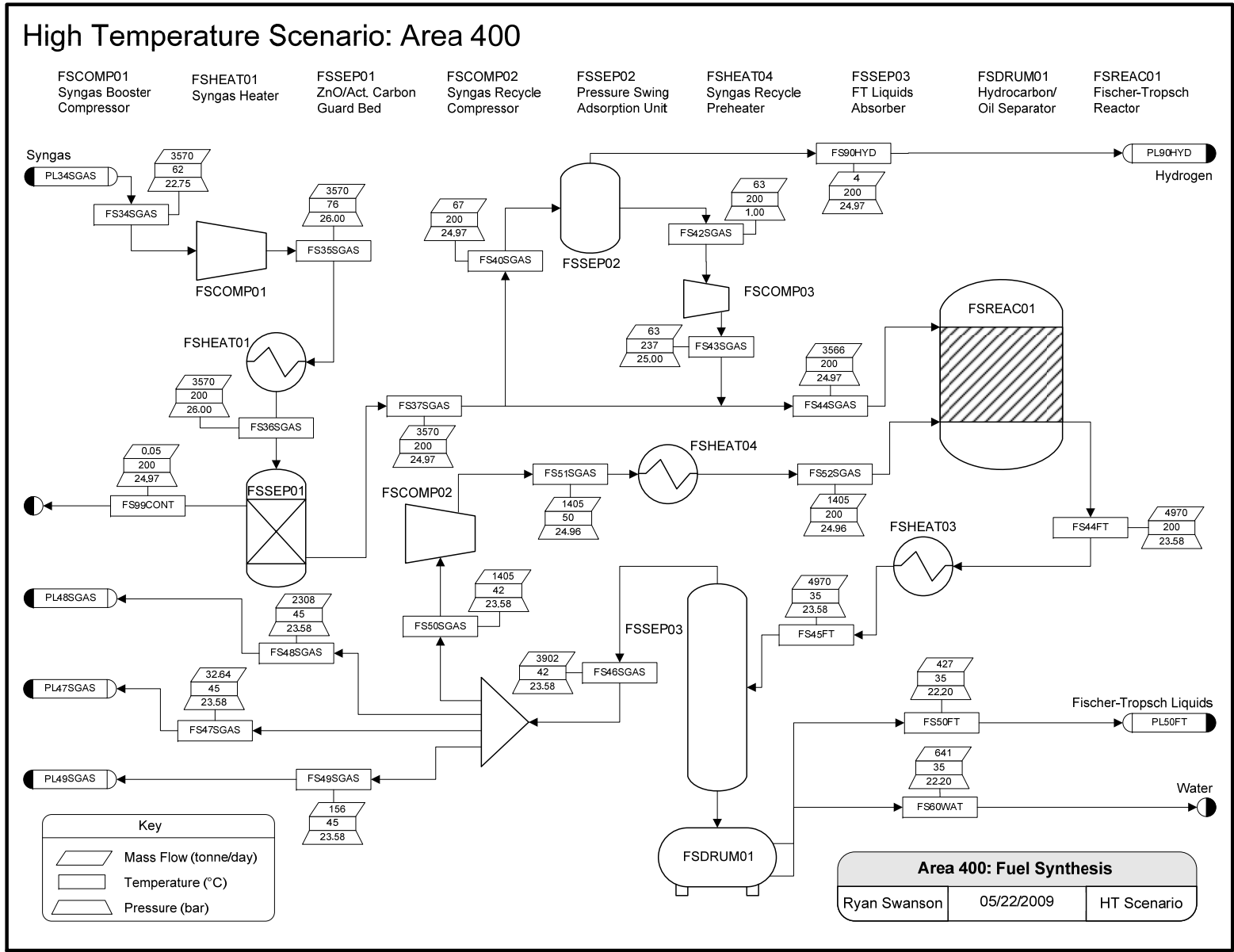


Figure D-7. Fuel synthesis area process flow diagram for HT scenario

High Temperature Scenario: Area 500

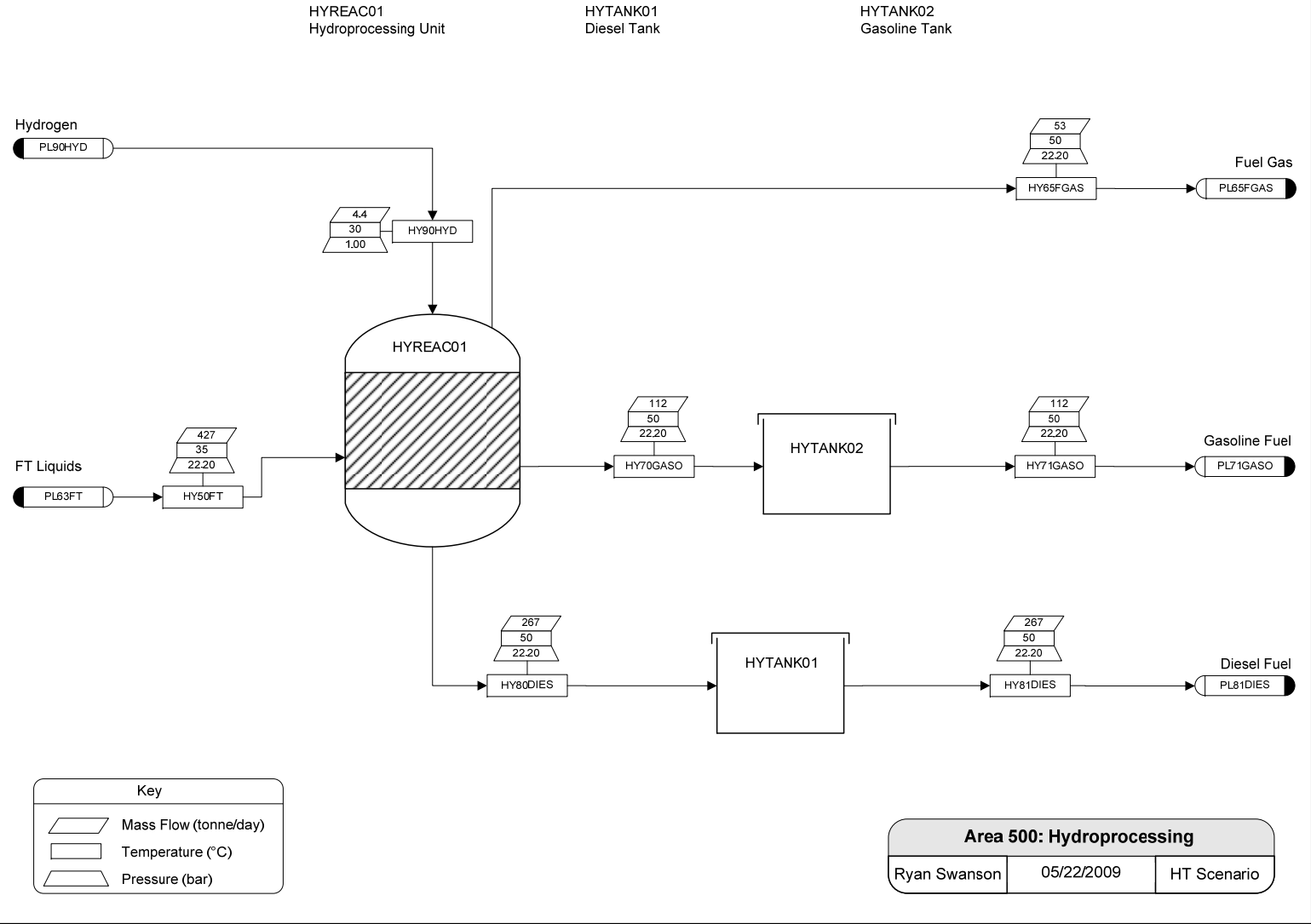


Figure D-8. Hydroprocessing area process flow diagram for HT scenario

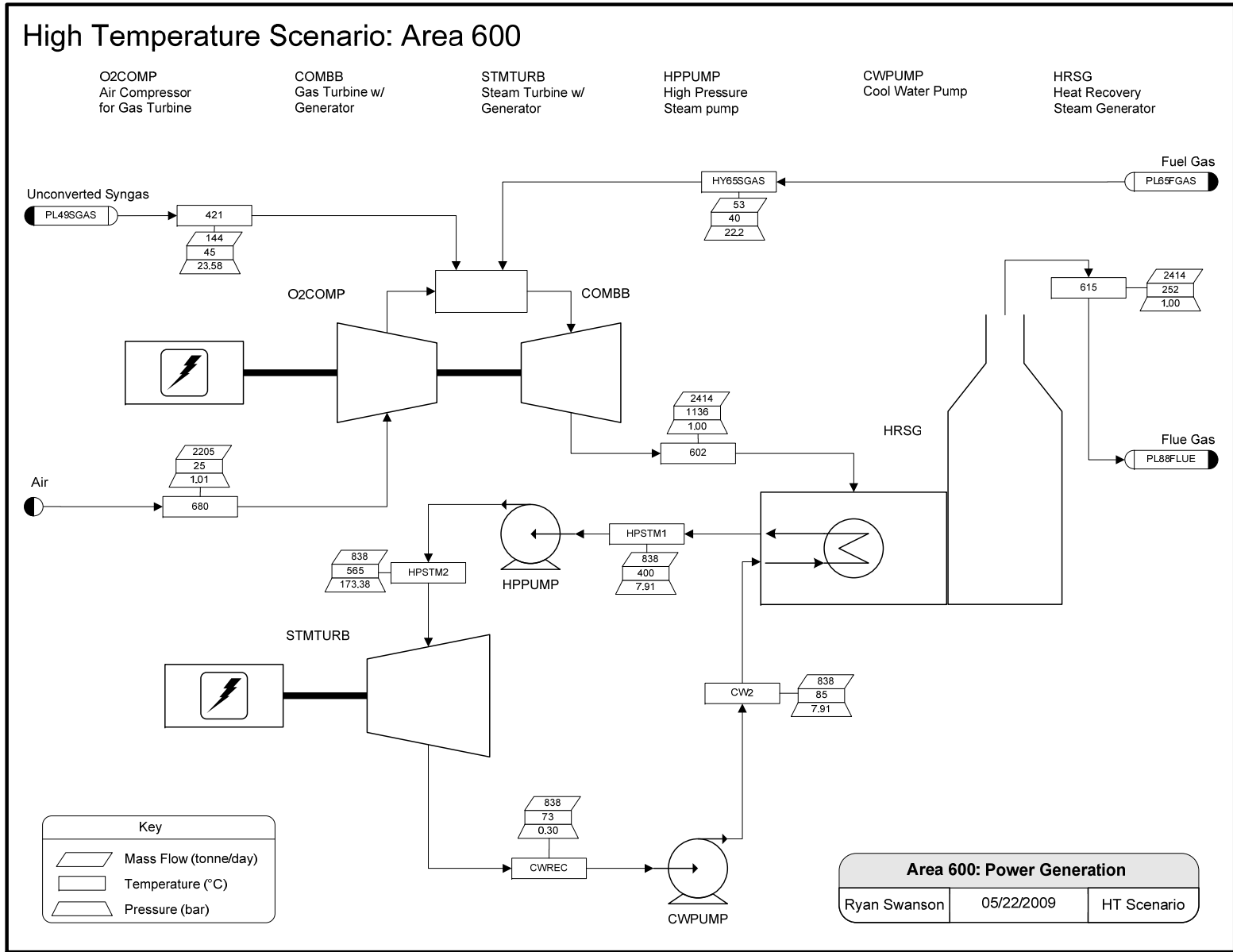


Figure D-9. Power generation area process flow diagram for HT scenario

High Temperature Scenario: Area 700

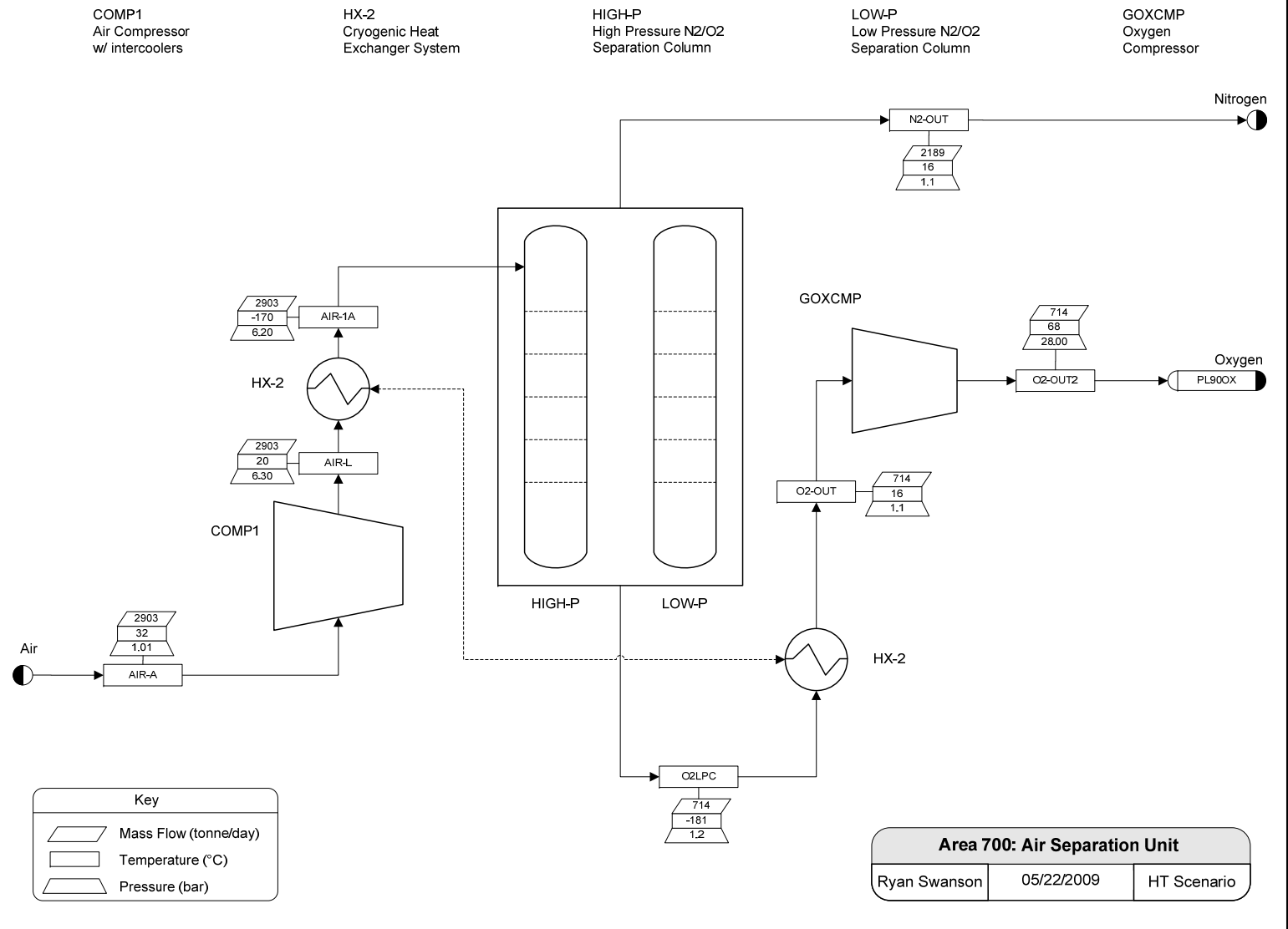


Figure D-10. Air separation unit process flow diagram for HT scenario

Low-Temperature Scenario

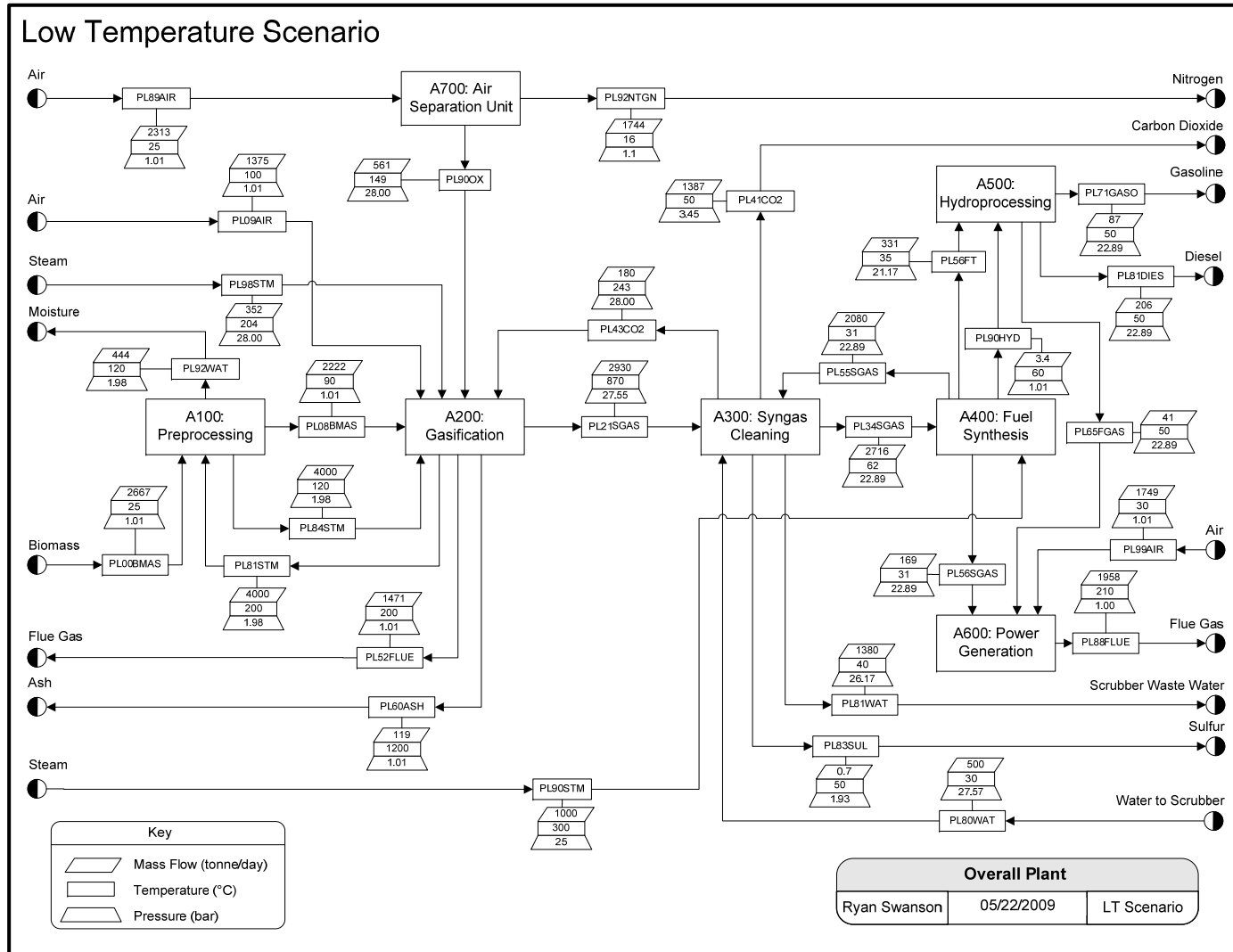


Figure D-11. Overall plant area process flow diagram for LT scenario

Low Temperature Scenario: Area 100

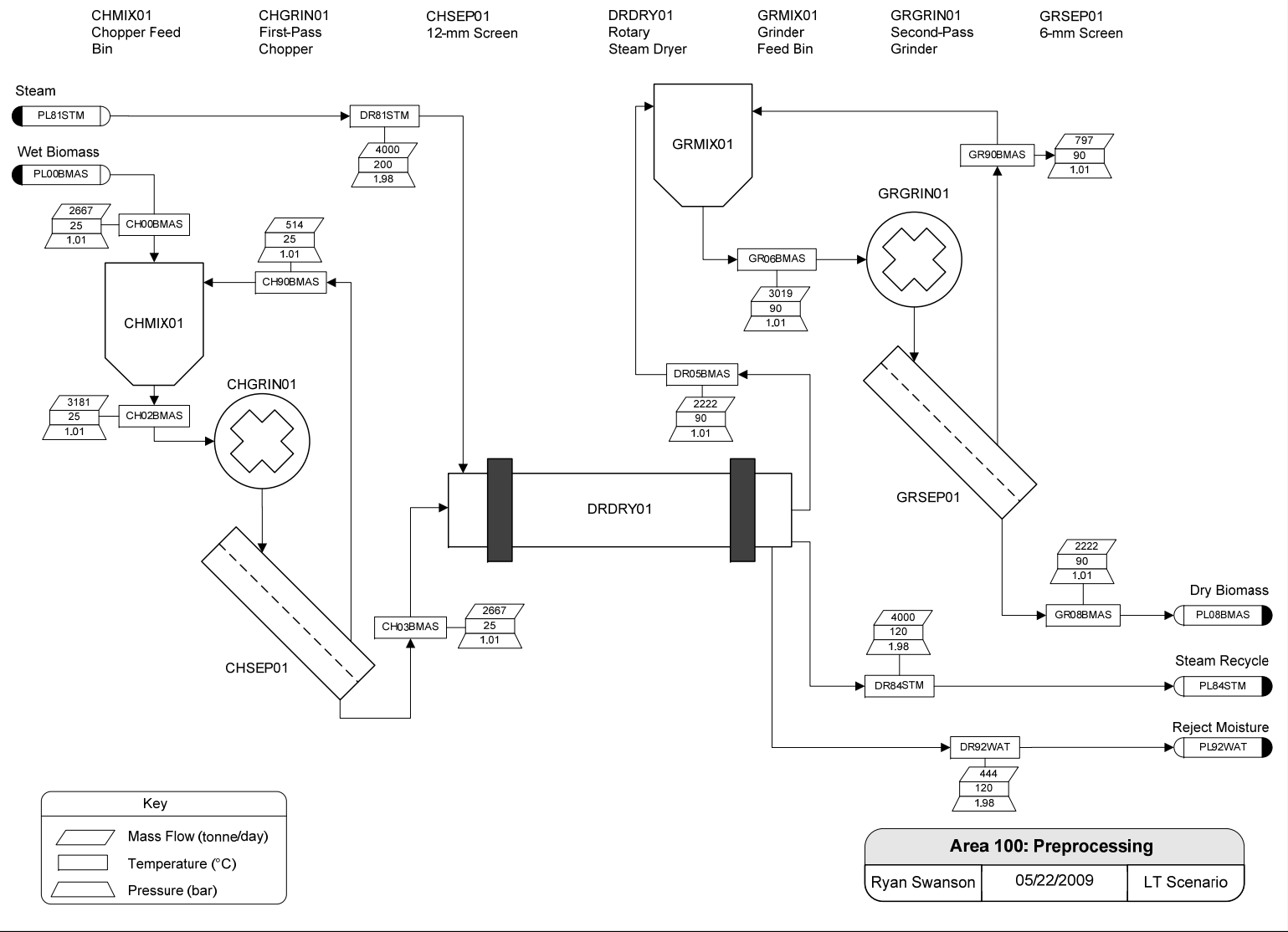


Figure D-12. Preprocessing area process flow diagram for LT scenario

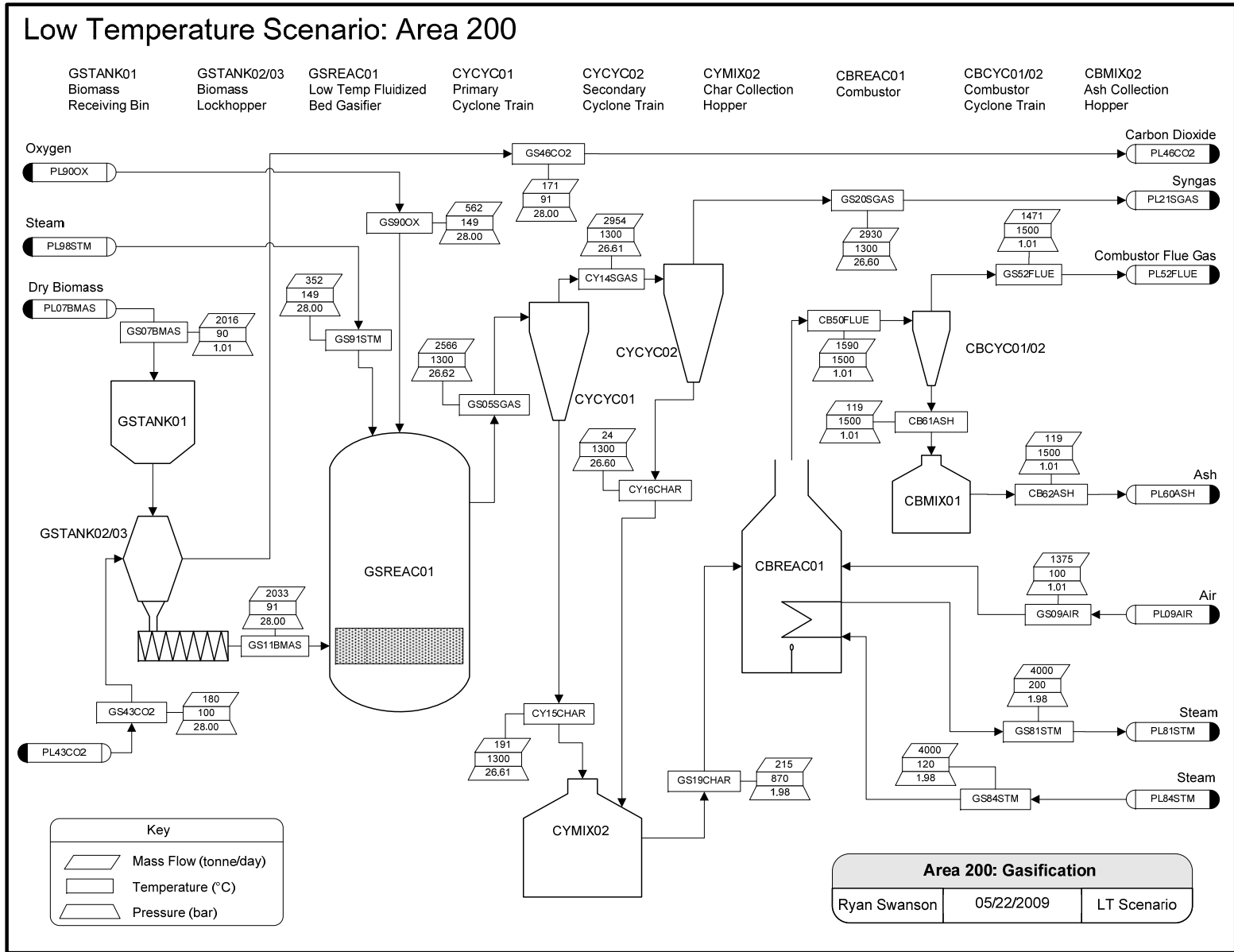


Figure D-13. Gasification area process flow diagram for LT scenario

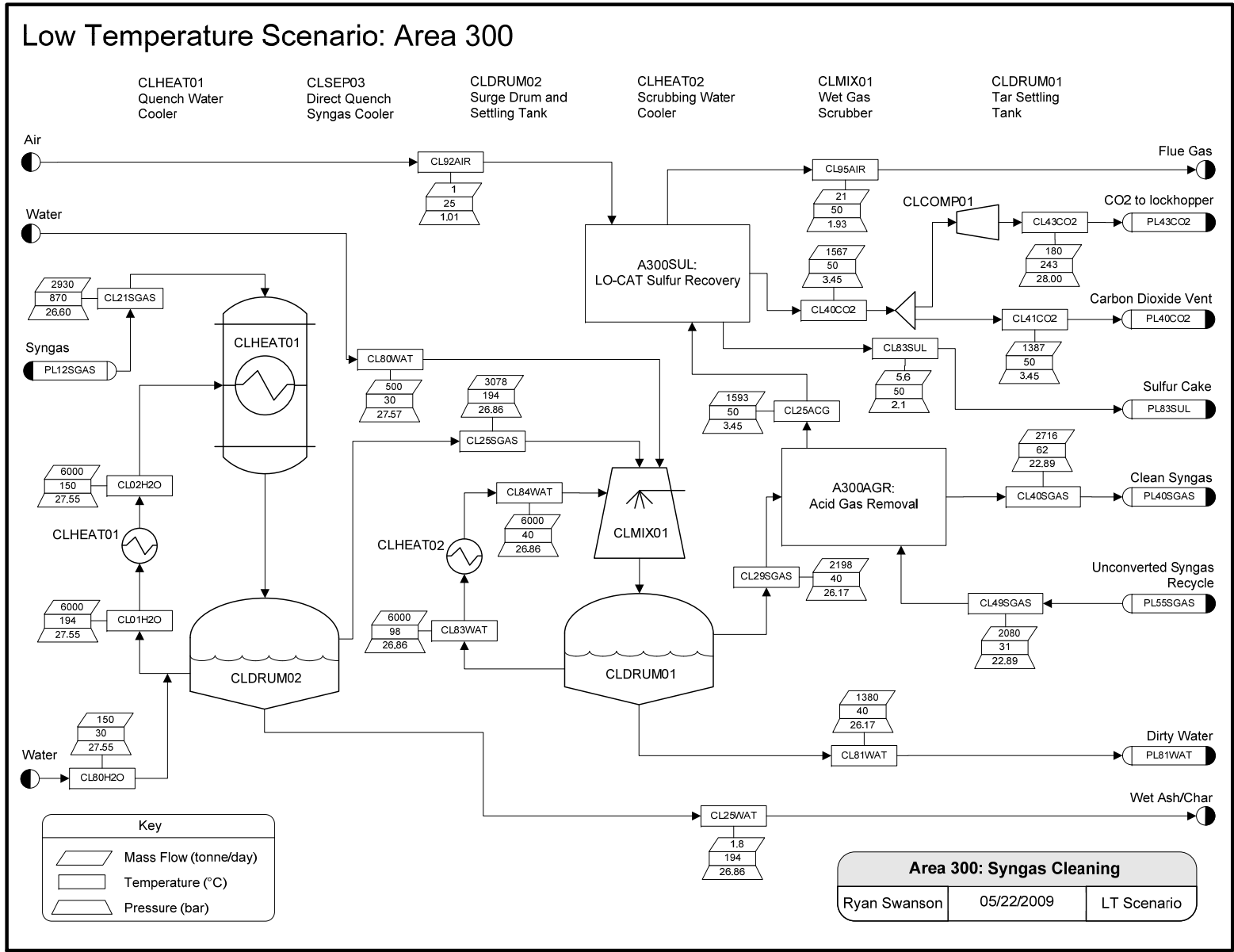


Figure D-14. Syngas cleaning area process flow diagram for LT scenario

Low Temperature Scenario: Area 300AGR

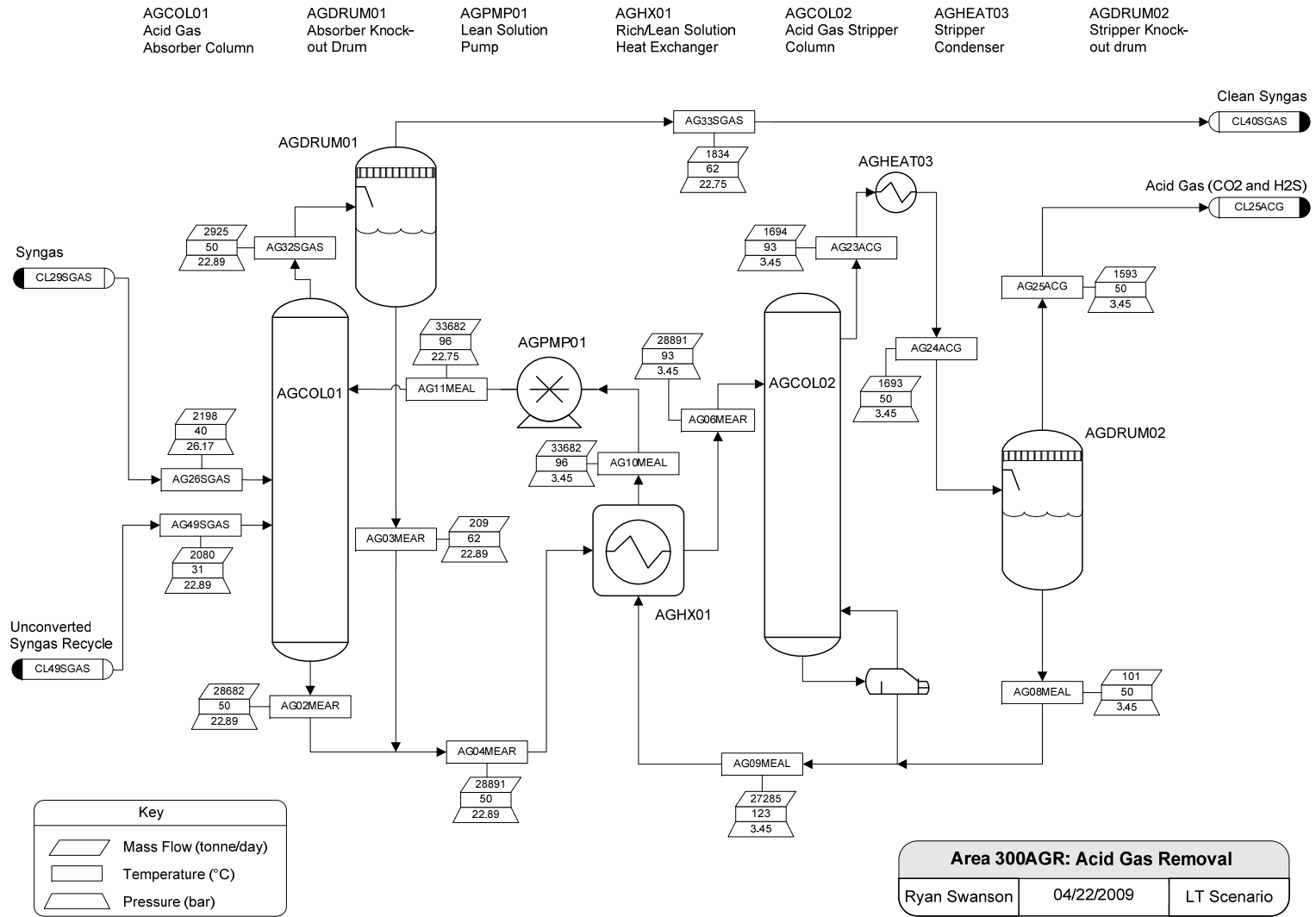


Figure D-15. Acid gas removal area process flow diagram for LT scenario

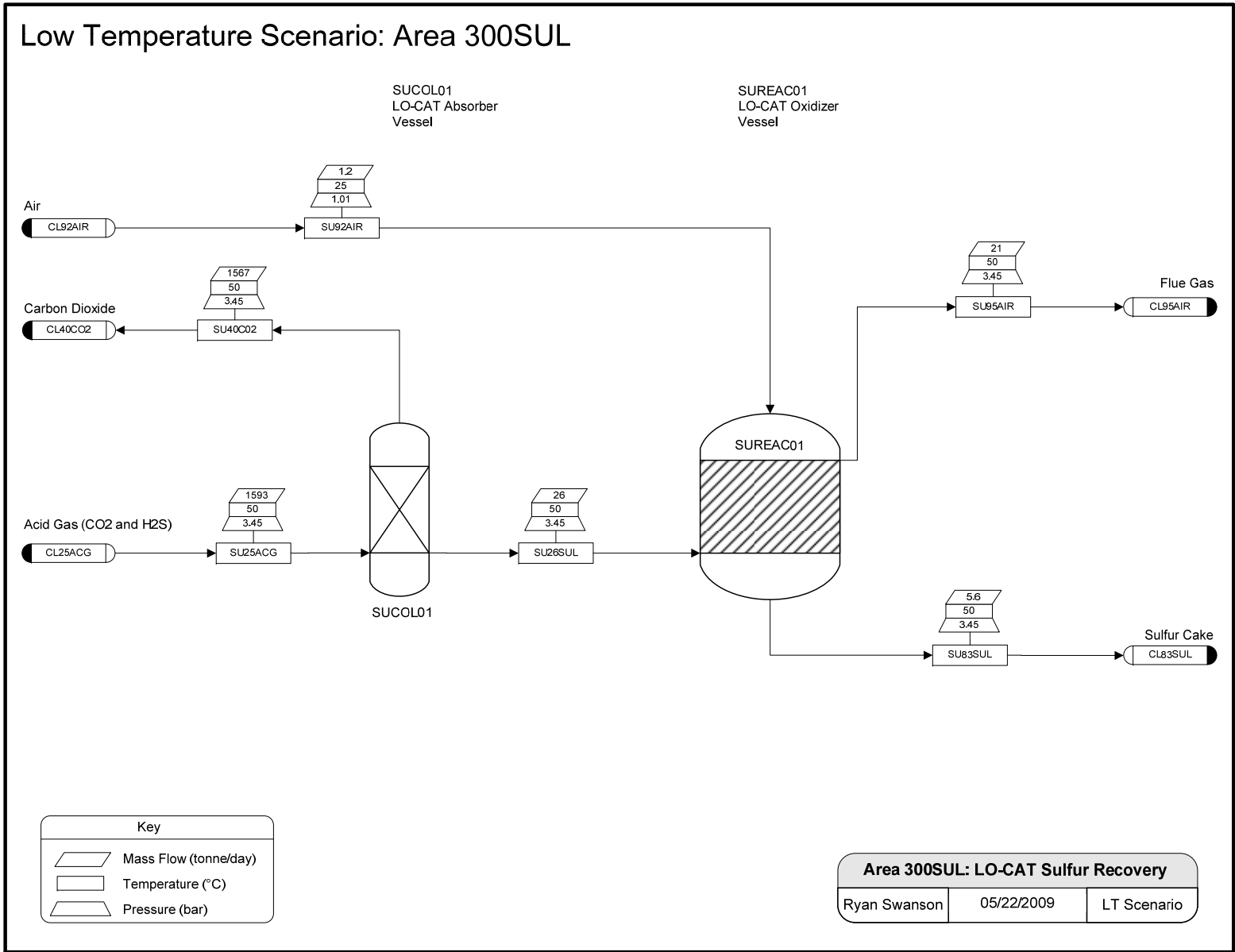


Figure D-16. Sulfur recovery process flow diagram for LT scenario

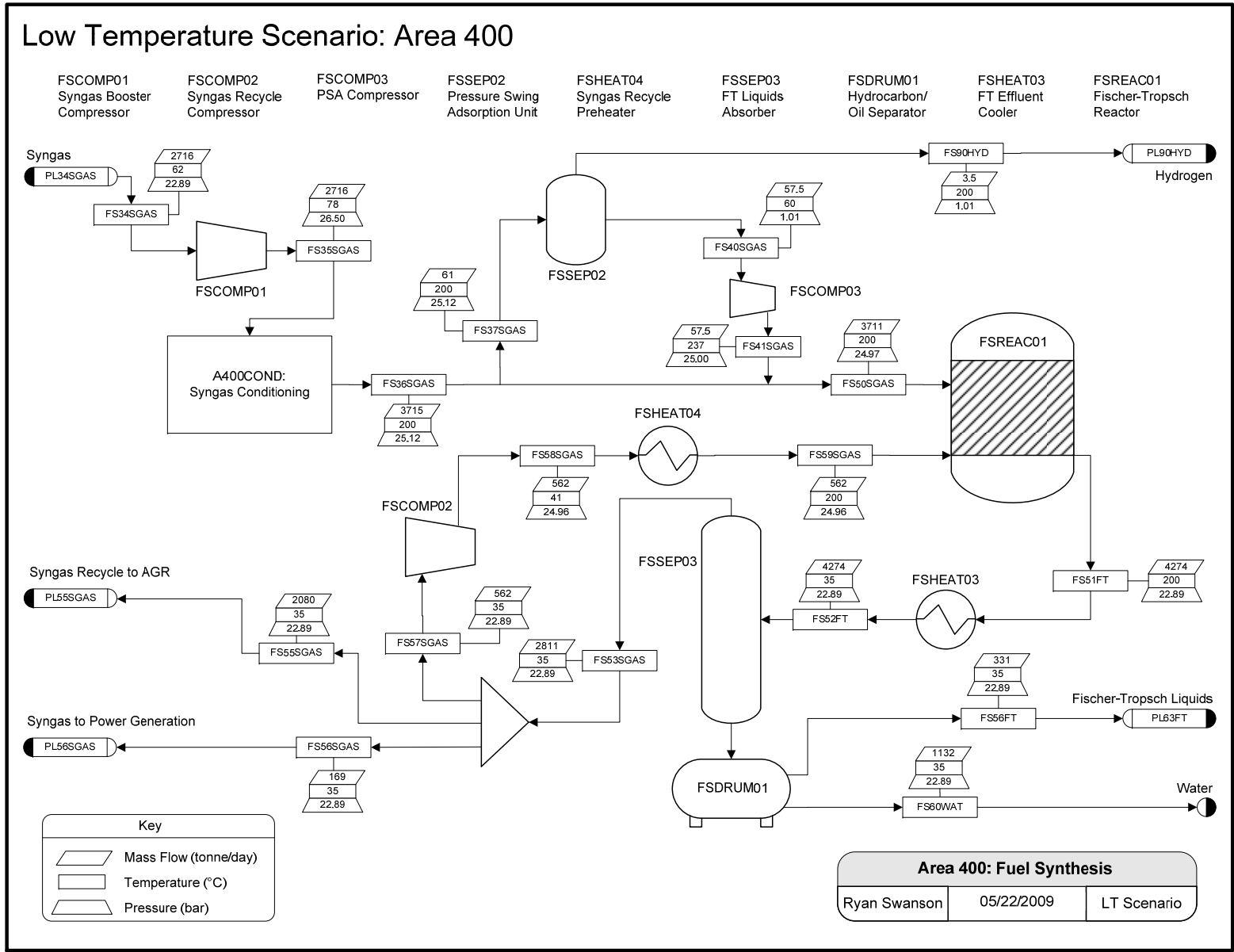


Figure D-17. Fuel synthesis area process flow diagram for LT scenario

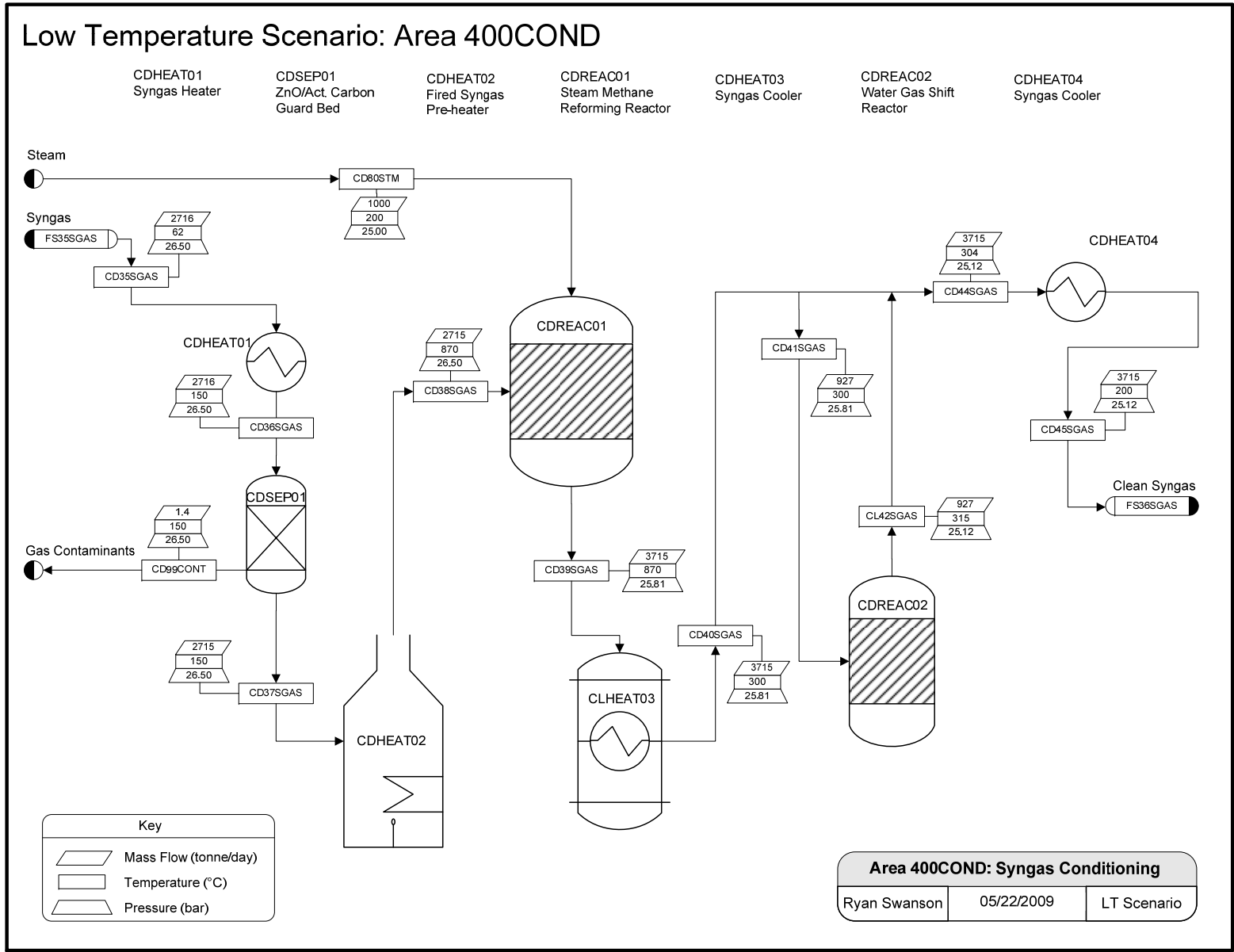


Figure D-18. Syngas conditioning area process flow diagram for LT scenario

Low Temperature Scenario: Area 500

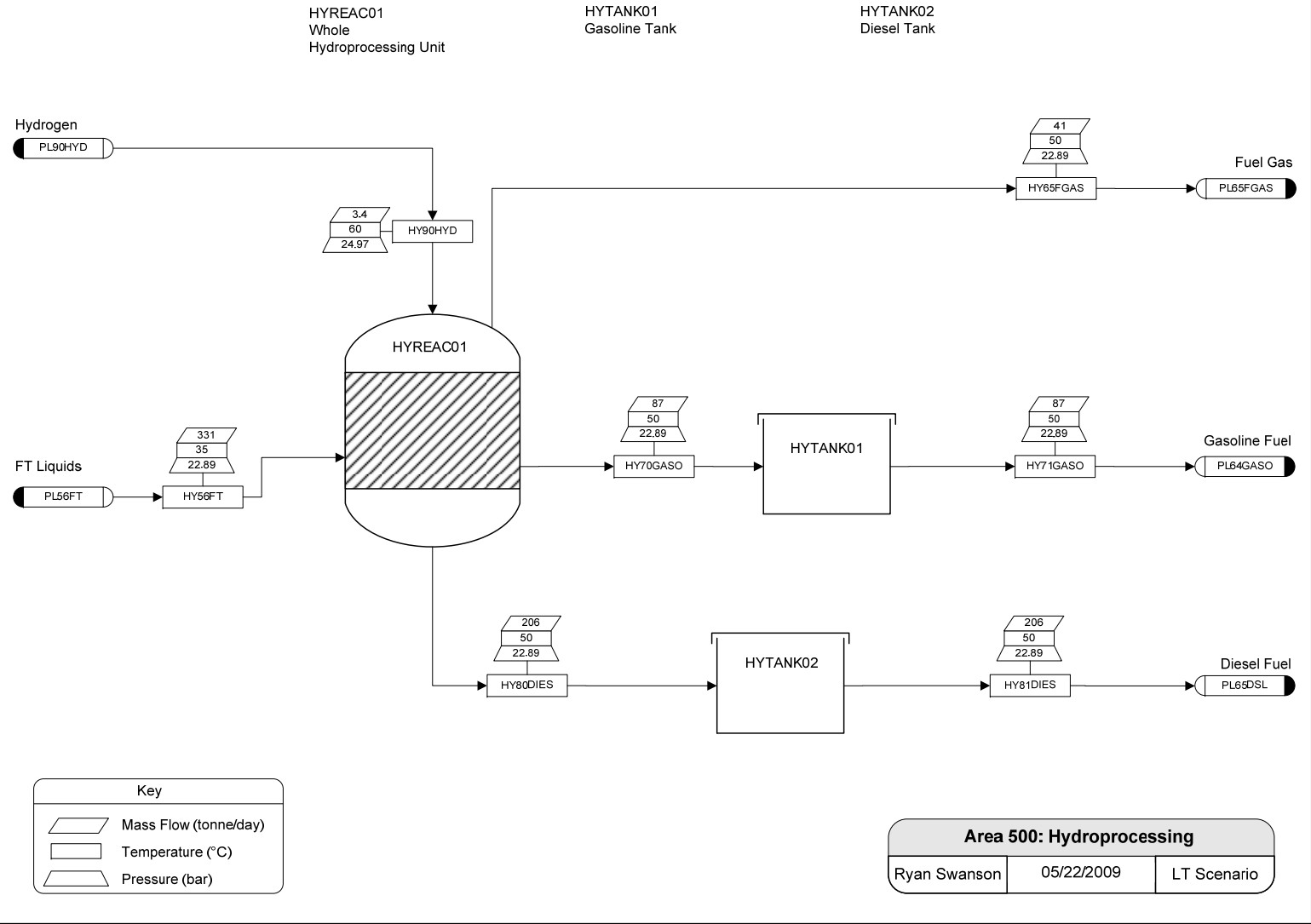


Figure D-19. Hydroprocessing area process diagram for LT scenario

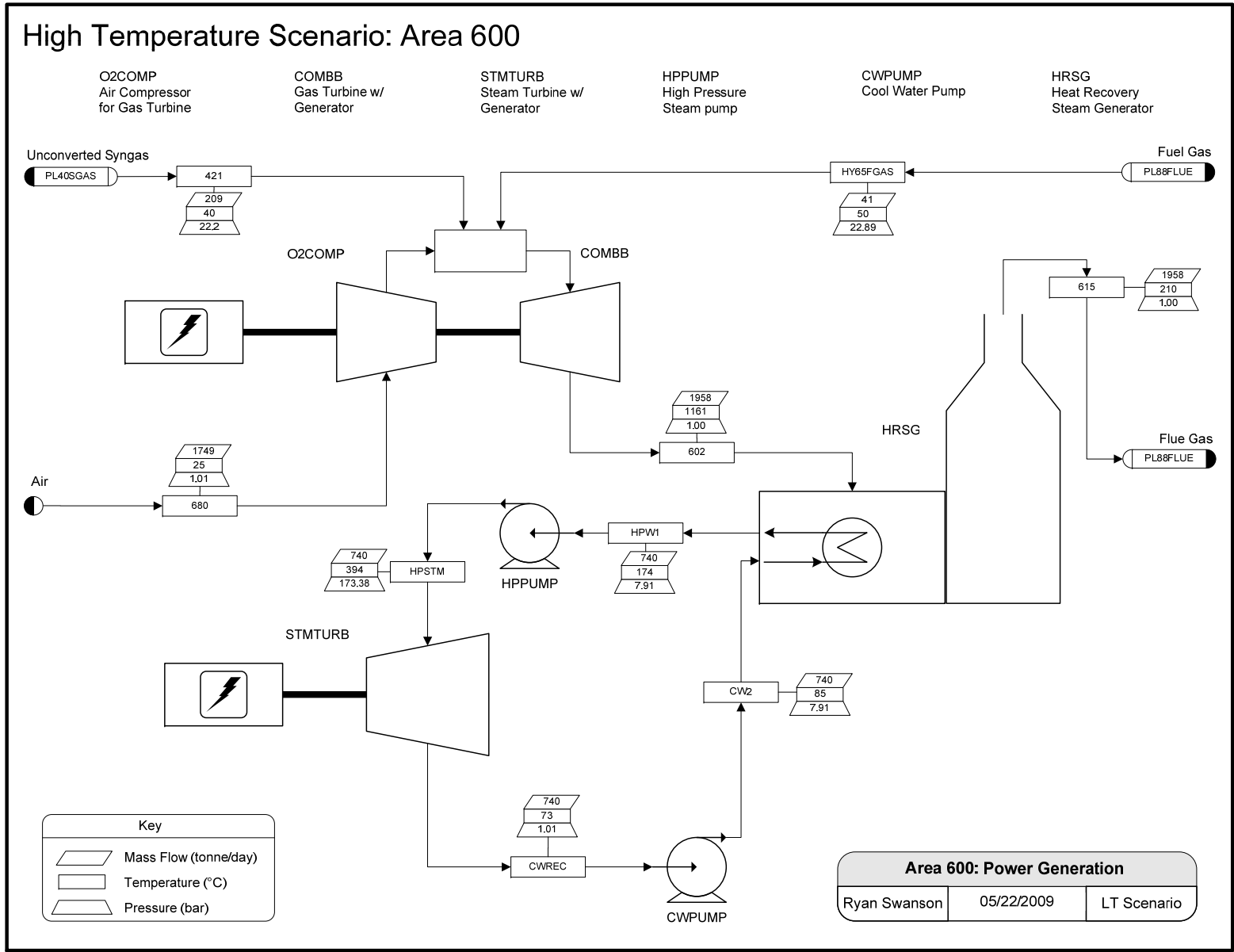


Figure D-20. Power generation area process flow diagram for LT scenario

High Temperature Scenario: Area 700

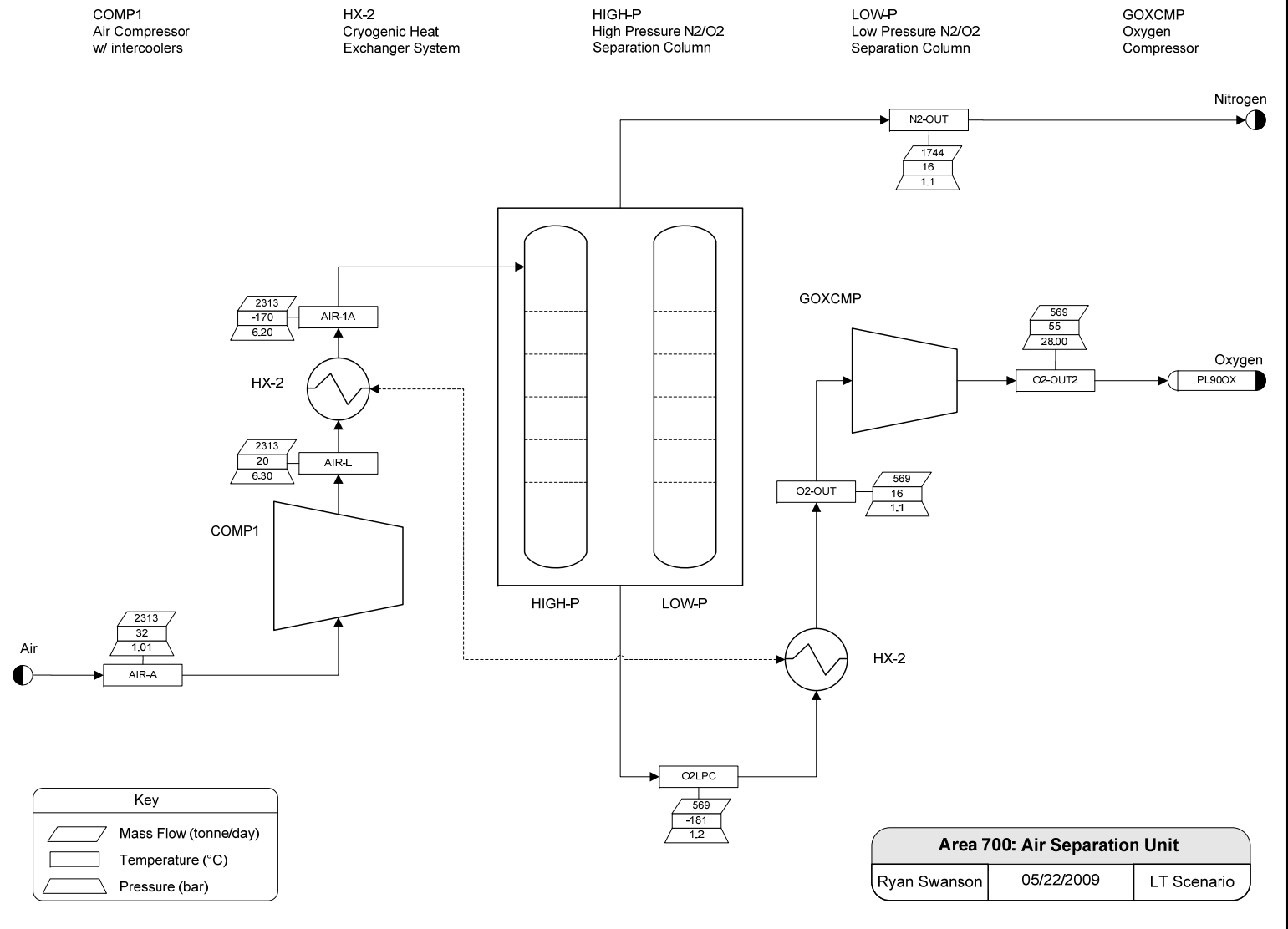


Figure D-21. Air separation unit process flow diagram for LT scenario

Appendix E. Stream Data

Table E-2. Preprocessing Area Stream Data for HT Scenario

HT A100	CH00BMAS	CH02BMAS	CH03BMAS	CH90BMAS	DR05BMAS	DR815TM	DR845TM	DR92WAT	GR05BMAS	GR08BMAS	GR09BMAS
Temperature (C)	25	25	25	25	90	200	120	120	90	90	90
Pressure (bar)	1.01	1.01	1.01	1.01	1.01	1.98	1.98	1.98	1.01	1.01	1.01
Vapor Fraction	0.00	0.00	0.00	0.00	0.00	0.00	1.00	1.00	0.00	0.00	0.00
Volume Flow** (m ³ /sec)	0.43	0.43	0.43	0	0.14	50.53	42.34	4.70	0.14	0.14	0
Mole Flow** (kmol/hr)	1542	1542	1542	0	513.97	9251	9251	1028	513.97	513.97	0
Mass Flow (tonnes/day)	2667	2709	2667	42.20	2222	4000	4000	444.44	2240	2222	17.60
H2O	666.67	666.67	666.67	0	222.22	0	0	444.44	222.22	222.22	0
CO	0	0	0	0	0	0	0	0	0	0	0
H2	0	0	0	0	0	0	0	0	0	0	0
CO2	0	0	0	0	0	0	0	0	0	0	0
O2	0	0	0	0	0	0	0	0	0	0	0
N2	0	0	0	0	0	0	0	0	0	0	0
CH4	0	0	0	0	0	0	0	0	0	0	0
C2H6	0	0	0	0	0	0	0	0	0	0	0
C2H4	0	0	0	0	0	0	0	0	0	0	0
C2H2	0	0	0	0	0	0	0	0	0	0	0
C3	0	0	0	0	0	0	0	0	0	0	0
C4	0	0	0	0	0	0	0	0	0	0	0
H2S	0	0	0	0	0	0	0	0	0	0	0
NH3	0	0	0	0	0	0	0	0	0	0	0
TAR	0	0	0	0	0	0	0	0	0	0	0
SULFUR	0	0	0	0	0	0	0	0	0	0	0
CARBON	0	0	0	0	0	0	0	0	0	0	0
CHAR	0	0	0	0	0	0	0	0	0	0	0
STEAM	0	0	0	0	0	4000	4000	0	0	0	0
SO2	0	0	0	0	0	0	0	0	0	0	0
NO2	0	0	0	0	0	0	0	0	0	0	0
MEA	0	0	0	0	0	0	0	0	0	0	0
AIR	0	0	0	0	0	0	0	0	0	0	0
WAXES	0	0	0	0	0	0	0	0	0	0	0
C5	0	0	0	0	0	0	0	0	0	0	0
C6	0	0	0	0	0	0	0	0	0	0	0
C7	0	0	0	0	0	0	0	0	0	0	0
C8	0	0	0	0	0	0	0	0	0	0	0
C9	0	0	0	0	0	0	0	0	0	0	0
C10	0	0	0	0	0	0	0	0	0	0	0
C11	0	0	0	0	0	0	0	0	0	0	0
C12	0	0	0	0	0	0	0	0	0	0	0
C13	0	0	0	0	0	0	0	0	0	0	0
C14	0	0	0	0	0	0	0	0	0	0	0
C15	0	0	0	0	0	0	0	0	0	0	0
C16	0	0	0	0	0	0	0	0	0	0	0
C17	0	0	0	0	0	0	0	0	0	0	0
C18	0	0	0	0	0	0	0	0	0	0	0
C19	0	0	0	0	0	0	0	0	0	0	0
C20	0	0	0	0	0	0	0	0	0	0	0
COS	0	0	0	0	0	0	0	0	0	0	0
AR	0	0	0	0	0	0	0	0	0	0	0
BIOMASS	2000	2042	2000	42.20	2000	0	0	0	2018	2000	17.60
ASH	0	0	0	0	0	0	0	0	0	0	0
SOOT	0	0	0	0	0	0	0	0	0	0	0
SLAG	0	0	0	0	0	0	0	0	0	0	0

**Volumetric and Mole flow values do not include biomass, ash, soot, or slag

Table E-3. Gasification Area Stream Data for HT Scenario

HT A200	GS02SLAG	GS06BMS	GS09CAIR	GS11BMS	GS12CO2	GS12FLUE	GS14SGAS	GS18STM	GS19STM	GS19OX	GS19CO2	GS19STM	GS19SGAS	SL01SLAG	SL02SGAS
Temperature (C)	50	90	25	90	250	220	45	200	120	149	92	200	203	1300	1300
Pressure (bar)	26.62	1.01	1.01	1.01	28.00	1.03	23.58	1.98	1.98	28.00	28.00	28.00	25.93	26.62	26.62
Vapor Fraction	0.00	0.00	1.00	0.53	1.00	1.00	1.00	1.00	1.00	1.00	1.00	0.00	1.00	0.00	1.00
Volume Flow** (m ³ /sec)	0	0.14	2.33	30.59	0.07	4.33	0.02	50.53	42.34	0.34	0.04	0.02	3.50	0	11.23
Mole Flow** (kmol/hr)	0	513.97	343.36	6974	175.05	393.42	71.82	9251	9251	957.07	158.25	2220	8308	0	8192
Mass Flow (tonnes/day)	114.00	2222	237.82	2222	180.00	266.91	29.08	4000	4000	743.70	166.90	960.00	3869	114.00	3825
H2O	0	222.22	0	222.22	2.50	27.89	0	0	0	0	0	960.00	1038	0	988.43
CO	0	0	0	0	0.47	0	11.24	0	0	0	0.45	0	1457	0	1457
H2	0	0	0	101.20	0	0	1.83	0	0	0	0	0	122.88	0	122.88
CO2	0	0	0	0	175.21	40.00	2.40	0	0	0	166.45	0	1184	0	1184
O2	0	0	55.97	812.60	0	10.28	0	0	0	700.00	0	0	0	0	0
N2	0	0	181.85	16.00	1.78	181.85	0	0	0	0	0	0	17.68	0	17.68
CH4	0	0	0	0	0.01	0	0.87	0	0	0	0	0	0.02	0	0.02
C2H6	0	0	0	0	0.01	0	1.46	0	0	0	0	0	0	0	0
C2H4	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C2H2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C3	0	0	0	0	0.01	0	1.93	0	0	0	0	0	0	0	0
C4	0	0	0	0	0.02	0	2.29	0	0	0	0	0	0	0	0
H2S	0	0	0	0	0	0	0	0	0	0	0	0	4.50	0	4.50
NH3	0	0	0	0	0	0	0	0	0	0	0	0	0.11	0	0.11
TAR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
SULFUR	0	0	0	4.40	0	0	0	0	0	0	0	0	0	0	0
CARBON	0	0	0	945.60	0	0	0	0	0	0	0	0	0	0	0
CHAR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
STEAM	0	0	0	0	0	0	0	4000	4000	0	0	0	0	0	0
SO2	0	0	0	0	0	0.02	0	0	0	0	0	0	0	0	0
NO2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
MEA	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
AIR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
WAXES	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C5	0	0	0	0	0	0	0.03	0	0	0	0	0	0	0	0
C6	0	0	0	0	0	0	0.03	0	0	0	0	0	0	0	0
C7	0	0	0	0	0	0	0.02	0	0	0	0	0	0	0	0
C8	0	0	0	0	0	0	0.02	0	0	0	0	0	0	0	0
C9	0	0	0	0	0	0	0.02	0	0	0	0	0	0	0	0
C10	0	0	0	0	0	0	0.02	0	0	0	0	0	0	0	0
C11	0	0	0	0	0	0	0.02	0	0	0	0	0	0	0	0
C12	0	0	0	0	0	0	0.01	0	0	0	0	0	0	0	0
C13	0	0	0	0	0	0	0.01	0	0	0	0	0	0	0	0
C14	0	0	0	0	0	0	0.01	0	0	0	0	0	0	0	0
C15	0	0	0	0	0	0	0.01	0	0	0	0	0	0	0	0
C16	0	0	0	0	0	0	0.01	0	0	0	0	0	0	0	0
C17	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C18	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C19	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C20	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
COS	0	0	0	0	0	0	0.02	0	0	0	0	0	0.30	0	0.30
AR	0	0	0	0	0	6.87	6.87	0	0	43.70	0	0	43.70	0	43.70
BIOMASS	0	2000	0	0	0	0	0	0	0	0	0	0	0	0	0
ASH	0	0	0	120.00	0	0	0	0	0	0	0	0	0	0	0
SOOT	0	0	0	0	0	0	0	0	0	0	0	0	0	0	6.00
SLAG	114.00	0	0	0	0	0	0	0	0	0	0	0	0	114.00	0

**Volumetric and Mole flow values do not include biomass, ash, soot, or slag

Table E-4. Syngas Cleaning Area Stream Data for HT Scenario

HT A300	CL25ACG	CL25SGAS	CL28SGAS	CL28SCHS	CL29SGAS	CL32SGAS	CL40CO2	CL41CO2	CL42CO2	CL49SGAS	CL81WAT	CL82WAT	CL83SUL	CL83WAT	CL90STM	CL92AIR	CL96AIR	G584STM	SC93STM	SL01H2O	SL02SGAS	SL03H2O	SL80H2O
Temperature (C)	50	203	291	240	60	40	53	53	250	45	30	40	50	60	190	25	50	62	250	203	1300	203	30
Pressure (bar)	3.45	25.93	24.82	24.82	24.82	24.82	3.45	3.45	28.00	23.58	24.82	24.82	1.93	24.82	10.00	1.01	1.93	22.75	25.86	26.62	26.62	25.93	26.62
Vapor Fraction	1.00	1.00	1.00	1.00	0.69	1.00	1.00	1.00	1.00	1.00	0.00	0.00	0.00	0.00	1.00	1.00	0.93	1.00	1.00	0.00	1.00	0.00	0.00
Volume Flow** (m ³ /sec)	3.69	1.08	2.01	4.54	2.07	1.86	3.70	0.38	0.07	1.64	0	0.02	0	0.02	1.30	0.01	0.01	3.45	0.54	0.07	11.23	0	0
Mole Flow** (kmol/hr)	1728	2576	3847	9579	9579	6419	1716	175.05	175.05	5260	115.64	3276	13.46	2313	1271	2.07	2.58	10089	1271	9251	8192	0	115.64
Mass Flow (tonnes/day)	1773	1199	1749	4418	4418	2967	1765	180.00	180.00	2130	50.00	1501	7.20	1000	549.62	1.59	2.32	3377	549.62	4000	3825	6.00	50.00
H2O	27.18	321.91	591.14	1308	1308	9.19	24.46	2.50	2.50	0	50.00	1348	4.04	1000	549.62	0	0.45	45.53	549.62	4000	988.43	0	50.00
CO	4.62	451.69	15.74	1021	1021	999.01	4.62	0.47	0.47	823.54	0	22.10	0	0	0	0	0	1818	0	0	1457	0	0
H2	0	38.09	69.47	154.26	154.26	154.26	0	0	0	134.08	0	0	0	0	0	0	0	288.34	0	0	122.88	0	0
CO2	1718	367.08	1052	1869	1869	1741	1718	175.21	175.21	175.74	0	128.26	0	0	0	0	0	190.01	0	0	1184	0	0
O2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	1.59	0.02	0	0	0	0	0	0
N2	17.41	5.48	5.48	17.68	17.68	17.43	17.41	1.78	1.78	0	0	0.26	0	0	0	0	0	0	0	0	17.68	0	0
CH4	0.08	0.01	0.01	0.02	0.02	0.02	0.08	0.01	0.01	63.47	0	0	0	0	0	0	0	63.41	0	0	0.02	0	0
C2H6	0.11	0	0	0	0	0	0.11	0.01	0.01	107.01	0	0	0	0	0	0	0	106.90	0	0	0	0	0
C2H4	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C2H2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C3	0.11	0	0	0	0	0	0.11	0.01	0.01	141.56	0	0	0	0	0	0	0	141.45	0	0	0	0	0
C4	0.17	0	0	0	0	0	0.17	0.02	0.02	167.55	0	0	0	0	0	0	0	167.38	0	0	0	0	0
H2S	3.39	1.40	1.40	4.50	4.50	3.48	0	0	0	0	0	1.02	0	0	0	0	0.03	0.03	0	0	4.50	0	0
NH3	0	0.03	0.03	0.11	0.11	0.01	0	0	0	0	0	0.10	0	0	0	0	0	0	0	0	0.11	0	0
TAR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
SULFUR	0	0	0	0	0	0	0	0	0	0	0	0	3.16	0	0	0	0	0	0	0	0	0	0
CARBON	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
CHAR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
STEAM	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
SO2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
NO2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
MEA	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
AIR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
WAXES	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C5	0	0	0	0	0	0	0	0	0	2.29	0	0	0	0	0	0	0	2.29	0	0	0	0	0
C6	0	0	0	0	0	0	0	0	0	2.46	0	0	0	0	0	0	0	2.46	0	0	0	0	0
C7	0	0	0	0	0	0	0	0	0	1.15	0	0	0	0	0	0	0	1.15	0	0	0	0	0
C8	0	0	0	0	0	0	0	0	0	1.18	0	0	0	0	0	0	0	1.18	0	0	0	0	0
C9	0.08	0	0	0	0	0	0	0	0	1.19	0	0	0	0	0	0	0	0.08	1.11	0	0	0	0
C10	0.03	0	0	0	0	0	0	0	0	1.19	0	0	0	0	0	0	0	0.03	1.16	0	0	0	0
C11	0.23	0	0	0	0	0	0	0	0	1.16	0	0	0	0	0	0	0	0.23	0.83	0	0	0	0
C12	0.13	0	0	0	0	0	0	0	0	0.54	0	0	0	0	0	0	0	0.13	0.13	0	0	0	0
C13	0.04	0	0	0	0	0	0	0	0	0.52	0	0	0	0	0	0	0	0.04	0.02	0	0	0	0
C14	0.01	0	0	0	0	0	0	0	0	0.50	0	0	0	0	0	0	0	0.01	0	0	0	0	0
C15	0	0	0	0	0	0	0	0	0	0.49	0	0	0	0	0	0	0	0	0	0	0	0	0
C16	0	0	0	0	0	0	0	0	0	0.47	0	0	0	0	0	0	0	0	0	0	0	0	0
C17	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C18	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C19	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C20	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
COS	0.08	0.09	0.09	0.30	0.30	0.17	0	0	0	1.12	0	0.13	0	0	0	0	0	0.08	1.20	0	0	0.30	0
AR	1.22	13.55	13.55	43.70	43.70	42.66	0	0	0	502.93	0	1.03	0	0	0	0	1.22	544.37	0	0	43.70	0	0
BIOMASS	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
ASH	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
SOOT	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	6.00	6.00	0
SLAG	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0

**Volumetric and Mole flow values do not include biomass, ash, soot, or slag

Table E-5. Acid Gas Removal and Sulfur Recovery Areas Stream Data for HT Scenario

HT A300AGR	AG02MEAR	AG03MEAR	AG04MEAR	AG06MEAR	AG08MEAL	AG09MEAL	AG10MEAL	AG11MEAL	AG23ACG	AG24ACG	AG25ACG	AG31SGAS	AG32SGAS	AG33SGAS	AG49SGAS	A300SUL	SU25ACG	SU26SUL	SU40CO2	SU63SUL	SU92AIR	SU93AIR	SU94AIR	SU95AIR	
Temperature (C)	58	62	58	86	50	123	96	96	86	50	50	40	58	62	45										
Pressure (bar)	22.75	22.75	22.75	3.45	3.45	3.45	3.45	20.68	3.45	3.45	3.45	24.82	22.75	22.75	23.58		3.45	3.45	3.45	1.93	1.01	2.07	1.93	1.93	
Vapor Fraction	0.00	0.00	0.00	0.01	0.00	0.00	0.00	0.00	0.98	0.83	1.00	1.00	0.93	1.00	1.00		1.00	0.49	1.00	0.00	1.00	1.00	1.00	0.93	
Volume Flow** (m ³ /sec)	0.480	0	0.480	1.590	0	0.480	0.470	0.470	4.820	3.690	3.690	1.860	3.410	3.450	1.640		3.690	0.010	3.700	0	0.010	0.010	0.010	0.010	
Mole Flow** (kmol/hr)	84877	699.000	85576	85576	342.110	83848	83848	83848	2070	2070	1728	6419	10788	10089	5260		1728	11.910	1716	13.460	2.070	2.070	2.070	2.580	
Mass Flow (tonnes/day)	42560	308.340	42868	42868	155.170	41096	41096	41096	1928	1928	1773	2967	3685	3377	2130		1773	7.930	1765	7.200	1.590	1.590	1.590	2.320	
H2O	33956	297.460	34254	34254	144.090	34226	34226	34226	171.270	171.270	27.180	9.190	342.990	45.530	0		27.180	2.720	24.460	4.040	0	0	0	0.450	
CO	0	4.640	4.640	4.640	0.010	0.010	0.010	0.010	4.630	4.630	4.620	999.010	1823	1818	823.540		4.620	0	4.620	0	0	0	0	0	
H2	0	0	0	0	0	0	0	0	0	0	0	154.260	288.340	288.340	134.080		0	0	0	0	0	0	0	0	
CO2	1725	1.650	1727	1727	8.690	8.690	8.690	8.690	1727	1727	1718	1741	191.660	190.010	175.740		1718	0	1718	0	0	0	0	0	
O2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	1.590	1.590	1.590	0.020	
N2	17.430	0	17.430	17.430	0.020	0.020	0.020	0.020	17.430	17.430	17.410	17.430	0	0	0		17.410	0	17.410	0	0	0	0	0	
CH4	0	0.080	0.080	0.080	0	0	0	0	0.080	0.080	0.080	0.020	63.490	63.410	63.470		0.080	0	0.080	0	0	0	0	0	
C2H6	0	0.110	0.110	0.110	0	0	0	0	0.110	0.110	0.110	0	107.010	106.900	107.010		0.110	0	0.110	0	0	0	0	0	
C2H4	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0	
C2H2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0	
C3	0	0.110	0.110	0.110	0	0	0	0	0.110	0.110	0.110	0	141.560	141.450	141.560		0.110	0	0.110	0	0	0	0	0	
C4	0	0.170	0.170	0.170	0	0	0	0	0.170	0.170	0.170	0	167.550	167.380	167.550		0.170	0	0.170	0	0	0	0	0	
H2S	3.450	0	3.450	3.450	0.060	0.060	0.060	0.060	3.450	3.450	3.390	3.480	0.030	0.030	0		3.390	3.390	0	0	0	0	0	0.030	
NH3	0.010	0	0.010	0.010	0.010	0.010	0.010	0.010	0.010	0.010	0	0.010	0	0	0		0	0	0	0	0	0	0	0	
TAR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0	
SULFUR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	3.160	0	0	0	0	
CARBON	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0	
CHAR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0	
STEAM	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0	
SO2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0	
NO2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0	
MEA	6858	0	6858	6858	0	6858	6858	6858	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0	
AIR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0	
WAXES	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0	
C5	0	0	0	0	0	0	0	0	0	0	0	0	2.290	2.290	2.290		0	0	0	0	0	0	0	0	
C6	0	0	0	0	0	0	0	0	0	0	0	0	2.460	2.460	2.460		0	0	0	0	0	0	0	0	
C7	0	0	0	0	0	0	0	0	0	0	0	0	1.150	1.150	1.150		0	0	0	0	0	0	0	0	
C8	0	0	0	0	0	0	0	0	0	0	0	0	1.180	1.180	1.180		0	0	0	0	0	0	0	0	
C9	0	0.080	0.080	0.080	0.010	0.010	0.010	0.010	0.080	0.080	0.080	0	1.190	1.110	1.190		0.080	0.080	0	0	0	0	0	0.080	
C10	0	0.030	0.030	0.030	0	0	0	0	0.030	0.030	0.030	0	1.190	1.160	1.190		0.030	0.030	0	0	0	0	0	0.030	
C11	0	0.330	0.330	0.330	0.100	0.100	0.100	0.100	0.320	0.320	0.230	0	1.160	0.830	1.160		0.230	0.230	0	0	0	0	0	0.230	
C12	0	0.410	0.410	0.410	0.280	0.280	0.280	0.280	0.410	0.410	0.130	0	0.540	0.130	0.540		0.130	0.130	0	0	0	0	0	0.130	
C13	0	0.500	0.500	0.500	0.460	0.460	0.460	0.460	0.500	0.500	0.040	0	0.520	0.020	0.520		0.040	0.040	0	0	0	0	0	0.040	
C14	0	0.500	0.500	0.500	0.490	0.490	0.490	0.490	0.490	0.490	0.500	0.010	0	0.500	0		0.010	0.010	0	0	0	0	0	0.010	
C15	0	0.490	0.490	0.490	0.490	0.490	0.490	0.490	0.490	0.490	0	0	0.490	0	0.490		0	0	0	0	0	0	0	0	
C16	0	0.470	0.470	0.470	0.470	0.470	0.470	0.470	0.470	0.470	0	0	0.470	0	0.470		0	0	0	0	0	0	0	0	
C17	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0	
C18	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0	
C19	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0	
C20	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0	
COS	0	0.080	0.080	0.080	0	0	0	0	0.080	0.080	0.080	0.170	1.290	1.200	1.120		0.080	0.080	0	0	0	0	0	0.080	
AR	0	1.220	1.220	1.220	0	0	0	0	1.220	1.220	1.220	42.660	545.590	544.370	502.930		1.220	1.220	0	0	0	0	0	1.220	
BIOMASS	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0	
ASH	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0	
SOOT	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0	
SLAG	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0	

**Volumetric and Mole flow values do not include biomass, ash, soot, or slag

Table E-6. Fuel Synthesis Area Stream Data for HT Scenario

HT A400	FS44SGAS	FS35SGAS	FS36SGAS	FS37SGAS	FS40SGAS	FS42SGAS	FS43SGAS	FS44FT	FS44SGAS	FS45FT	FS45SGAS	FS46FT	FS46SGAS	FS47SGAS	FS48SGAS	FS49SGAS	FS50FT	FS50SGAS	FS51SGAS	FS52SGAS	FS60WAT	FS69HYD	FS90HYD	FS99CONT	
Temperature (C)	62	76	200	200	30	30	417	200	202	35	202	43	0	45	45	45	35	35	45	50	200	35	30	200	
Pressure (bar)	22.75	26.00	26.00	24.97	1.00	1.00	24.97	23.58	24.97	23.58	24.96	23.58	0.00	23.58	23.58	23.58	22.20	23.58	24.96	24.96	22.20	1.00	1.00	24.97	
Vapor Fraction	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	0.86	1.00	0.00	1.00	1.00	1.00	1.00	0.00	1.00	1.00	1.00	0.00	1.00	1.00	1.00	
Volume Flow** (m³/sec)	3.45	3.16	4.29	4.47	1.26	0.63	0.06	4.84	4.45	2.71	5.86	0.02	2.74	0.02	1.64	0.11	0.01	1.00	0.96	1.42	0.01	0.63	0.63	0	
Mole Flow** (kmol/hr)	10089	10089	10089	10089	179.78	89.51	89.51	10438	9998	10438	13198	1575	8887	71.82	5260	355.46	89.49	3199	3199	3199	1483	90.27	90.27	0.04	
Mass Flow (tonnes/day)	3377	3377	3377	3377	60.18	55.81	55.81	4668	3372	4668	4668	1069	3599	29.08	2130	143.95	427.14	1296	1296	1296	641.34	4.37	4.37	0.03	
H2O	45.53	45.53	45.53	45.53	0.81	0.81	0.81	642.07	45.53	642.07	45.53	642.07	0	0	0	0	0	0	0	0	0	641.34	0	0	0
CO	1818	1818	1818	1818	32.40	32.40	32.40	1391	1818	1391	2319	0	1391	11.24	823.54	55.65	0	500.87	500.87	500.87	0	0	0	0	0
H2	288.34	288.34	288.34	288.34	5.14	0.77	0.77	225.34	283.97	225.34	365.52	0	226.52	1.83	134.08	9.06	0	81.55	81.55	81.55	0	4.37	4.37	0	0
CO2	190.01	190.01	190.01	190.01	3.39	3.39	3.39	296.89	190.01	296.89	296.89	0	296.90	2.40	175.74	11.88	0	106.88	106.88	106.88	0	0	0	0	0
O2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
N2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
CH4	63.41	63.41	63.41	63.41	1.13	1.13	1.13	107.33	63.41	107.33	102.02	0	107.23	0.87	63.47	4.29	0	38.60	38.60	38.60	0	0	0	0	0
C2H6	106.90	106.90	106.90	106.90	1.91	1.91	1.91	180.95	106.90	180.95	171.99	0	180.79	1.46	107.01	7.23	0	65.08	65.08	65.08	0	0	0	0	0
C2H4	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C2H2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C3	141.45	141.45	141.45	141.45	2.52	2.52	2.52	239.37	141.45	239.37	227.54	0	239.16	1.93	141.56	9.57	0	86.10	86.10	86.10	0	0	0	0	0
C4	167.38	167.38	167.38	167.38	2.98	2.98	2.98	283.31	167.38	283.31	269.28	0	283.06	2.29	167.55	11.32	0	101.90	101.90	101.90	0	0	0	0	0
H2S	0.03	0.03	0.03	0.03	0	0	0	0	0	0	0	0	0	0.00	0	0	0	0	0	0	0	0	0	0	0.03
NH3	0	0	0	0	0	0	0	0	0	0	0	0	0	0.00	0	0	0	0	0	0	0	0	0	0	0
TAR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
SULFUR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
CARBON	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
CHAR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
STEAM	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
SO2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
NO2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
MEA	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
AIR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
WAXES	0	0	0	0	0	0	0	170.22	0	170.22	0	170.22	0	0	0	0	0	170.22	0	0	0	0	0	0	0
C5	2.29	2.29	2.29	2.29	0.04	0.04	0.04	19.36	2.29	19.36	3.69	15.49	3.87	0.03	2.29	0.15	15.49	1.39	1.39	1.39	0	0	0	0	0
C6	2.46	2.46	2.46	2.46	0.04	0.04	0.04	20.81	2.46	20.81	3.96	16.65	4.16	0.03	2.46	0.17	16.65	1.50	1.50	1.50	0	0	0	0	0
C7	1.15	1.15	1.15	1.15	0.02	0.02	0.02	19.49	1.15	19.49	1.85	17.54	1.95	0.02	1.15	0.08	17.54	0.70	0.70	0.70	0	0	0	0	0
C8	1.18	1.18	1.18	1.18	0.02	0.02	0.02	19.99	1.18	19.99	1.90	17.99	2.00	0.02	1.18	0.08	17.99	0.72	0.72	0.72	0	0	0	0	0
C9	1.11	1.11	1.11	1.11	0.02	0.02	0.02	20.11	1.11	20.11	1.83	18.10	2.01	0.02	1.19	0.08	18.10	0.72	0.72	0.72	0	0	0	0	0
C10	1.16	1.16	1.16	1.16	0.02	0.02	0.02	20.13	1.16	20.13	1.88	18.12	2.01	0.02	1.19	0.08	18.12	0.72	0.72	0.72	0	0	0	0	0
C11	0.83	0.83	0.83	0.83	0.01	0.01	0.01	19.58	0.83	19.58	1.54	17.62	1.96	0.02	1.16	0.08	17.62	0.71	0.71	0.71	0	0	0	0	0
C12	0.13	0.13	0.13	0.13	0	0	0	18.16	0.13	18.16	0.45	17.25	0.91	0.01	0.54	0.04	17.25	0.33	0.33	0.33	0	0	0	0	0
C13	0.02	0.02	0.02	0.02	0	0	0	17.58	0.02	17.58	0.33	16.70	0.88	0.01	0.52	0.04	16.70	0.32	0.32	0.32	0	0	0	0	0
C14	0	0	0	0	0	0	0	17.01	0	17.01	0.31	16.16	0.85	0.01	0.50	0.03	16.16	0.31	0.31	0.31	0	0	0	0	0
C15	0	0	0	0	0	0	0	16.39	0	16.39	0.30	15.57	0.82	0.01	0.49	0.03	15.57	0.29	0.29	0.29	0	0	0	0	0
C16	0	0	0	0	0	0	0	15.72	0	15.72	0.28	14.93	0.79	0.01	0.47	0.03	14.93	0.28	0.28	0.28	0	0	0	0	0
C17	0	0	0	0	0	0	0	14.75	0	14.75	0	14.75	0	0	0	0	14.75	0	0	0	0	0	0	0	0
C18	0	0	0	0	0	0	0	14.05	0	14.05	0	14.05	0	0	0	0	14.05	0	0	0	0	0	0	0	0
C19	0	0	0	0	0	0	0	13.35	0	13.35	0	13.35	0	0	0	0	13.35	0	0	0	0	0	0	0	0
C20	0	0	0	0	0	0	0	12.64	0	12.64	0	12.64	0	0	0	0	12.64	0	0	0	0	0	0	0	0
COS	1.20	1.20	1.20	1.20	0.02	0.02	0.02	1.88	1.20	1.88	1.88	0	1.88	0.02	1.12	0.08	0	0.68	0.68	0.68	0	0	0	0	0
AR	544.37	544.37	544.37	544.37	9.70	9.70	9.70	850.25	544.37	850.25	850.25	0	849.66	6.87	502.93	33.99	0	305.88	305.88	305.88	0	0	0	0	0
BIOMASS	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
ASH	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
SOOT	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
SLAG	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0

**Volumetric and Mole flow values do not include biomass, ash, soot, or slag

Table E-7. Hydroprocessing, Power Generation, and Air Separation Areas Stream Data for HT Scenario

HT A500	HY50FT	HY65FGAS	HY70GASO	HY71GASO	HY80DIES	HY81DIES	HY90HYD	HT A600	421	602	615	680	CW2	CWREC	HPSTM1	HPSTM2	HT A700	AIR-1A	AIR-A	AIR-L	N2-OUT	O2-LFC	O2-OUT	O2-OUT2
Temperature (C)	35	35	35	36	35	37	30		45	1144	273	30	85	73	170	565		-170	32	20	16	-177	16	68
Pressure (bar)	22.20	22.20	22.20	1.03	22.20	1.03	1.00		23.58	1.00	1.00	1.01	7.91	0.30	7.91	173.38		6.20	1.01	6.30	1.10	1.88	1.10	29.97
Vapor Fraction	0.00	1.00	0.00	0.00	0.00	0.00	1.00		1.00	1.00	1.00	1.00	0.00	0.97	0.00	0.00		1.00	1.00	1.00	1.00	1.00	0.98	1.00
Volume Flow** (m ³ /sec)	0.01	0.02	0	0	0.01	0.01	0.63		0.11	117.68	45.37	22.37	0.01	49.13	0.01	0.01		1.33	29.07	4.49	19.73	1.04	5.58	0.24
Mole Flow** (kmol/hr)	89.49	74.55	41.08	41.08	48.97	48.97	90.27		355.46	3595	3595	3237	1939	1939	1939	1939		3968	4177	4177	3255	921.35	921.35	921.35
Mass Flow (tonnes/day)	427.14	52.77	112.62	112.62	266.11	266.11	4.37		143.95	2439	2439	2242	838.24	838.23	838.23	838.23		2758	2903	2903	2189	714.26	714.26	714.26
H2O	0	0	0	0	0	0	0		0	233.43	233.43	0	838.24	838.23	838.23	838.23		0	0	0	0	0	0	0
CO	0	0	0	0	0	0	0		55.65	0	0	0	0	0	0	0		0	0	0	0	0	0	0
H2	0	0	0	0	0	0	4.37		9.06	0	0	0	0	0	0	0		0	0	0	0	0	0	0
CO2	0	0	0	0	0	0	0		11.88	352.24	352.24	0	0	0	0	0		0	0	0	0	0	0	0
O2	0	0	0	0	0	0	0		0	104.25	104.25	527.27	0	0	0	0		638.58	672.19	672.19	0	672.19	672.19	672.19
N2	0	0	0	0	0	0	0		0	1715	1715	1715	0	0	0	0		2084	2194	2194	2189	4.91	4.91	4.91
CH4	0	14.94	0	0	0	0	0		4.29	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C2H6	0	0	0	0	0	0	0		7.23	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C2H4	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C2H2	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C3	0	37.83	0	0	0	0	0		9.57	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C4	0	0	0	0	0	0	0		11.32	0	0	0	0	0	0	0		0	0	0	0	0	0	0
H2S	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
NH3	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
TAR	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
SULFUR	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
CARBON	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
CHAR	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
STEAM	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
SO2	0	0	0	0	0	0	0		0	0.08	0.08	0	0	0	0	0		0	0	0	0	0	0	0
NO2	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
MEA	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
AIR	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
WAXES	170.22	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C5	15.49	0	0	0	0	0	0		0.15	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C6	16.65	0	0	0	0	0	0		0.17	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C7	17.54	0	0	0	0	0	0		0.08	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C8	17.99	0	112.62	112.62	0	0	0		0.08	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C9	18.10	0	0	0	0	0	0		0.08	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C10	18.12	0	0	0	0	0	0		0.08	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C11	17.62	0	0	0	0	0	0		0.08	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C12	17.25	0	0	0	0	0	0		0.04	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C13	16.70	0	0	0	0	0	0		0.04	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C14	16.16	0	0	0	0	0	0		0.03	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C15	15.57	0	0	0	0	0	0		0.03	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C16	14.93	0	0	0	266.11	266.11	0		0.03	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C17	14.75	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C18	14.05	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C19	13.35	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C20	12.64	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
COS	0	0	0	0	0	0	0		0.08	0	0	0	0	0	0	0		0	0	0	0	0	0	0
AR	0	0	0	0	0	0	0		33.99	33.99	33.99	0	0	0	0	0		35.42	37.28	37.28	0.12	37.16	37.16	37.16
BIOMASS	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
ASH	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
SOOT	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
SLAG	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0

**Volumetric and Mole flow values do not include biomass, ash, soot, or slag

Low-Temperature Scenario

Table E-8. Overall Plant Stream Data for LT Scenario

HT Overall Plant	PL00BMAS	PL00BMAS	PL00CAIR	PL21SGAS	PL34SGAS	PL41CO2	PL43CO2	PL52FLUE	PL55SGAS	PL56FT	PL58SGAS	PL60ASH	PL65FGAS	PL7GASO	PL81DIES	PL81STM	PL81WAT	PL83SUI	PL84STM	PL86FLUE	PL90NYD	PL90OX	PL90STM	PL92WAT	PL96STM	PL98AIR	PL92NTGN	PL98AIR	
Temperature (C)	25	90	100	870	62	50	243	200	32	35	32	0	50	50	50	200	40	50	120	344	60	149	300	120	204	32	16	30	
Pressure (bar)	1.01	1.01	1.01	27.55	22.89	3.45	28.00	1.00	22.89	22.89	22.89	1.00	22.89	22.89	22.89	1.98	26.86	2.07	1.98	1.00	1.01	22.00	25.00	1.98	22.00	1.01	1.10	1.01	
Vapor Fraction	0.00	0.00	1.00	1.00	1.00	1.00	1.00	1.00	0.00	1.00	0.00	1.00	0.00	1.00	0.00	1.00	0.00	0.00	0.00	1.00	1.00	1.00	1.00	1.00	0.00	1.00	1.00	1.00	
Volume Flow** (m³/sec)	0.43	0.14	16.89	4.76	2.40	2.84	0.07	22.48	1.50	0.01	0.12	0	0.02	0	50.53	0.02	0	42.34	40.84	0.53	0.32	1.15	4.70	0.01	23.17	15.73	17.42		
Mole Flow** (kmol/hr)	1542	513.97	1985	4939	7066	1334	170.42	2053	4869	69.40	394.75	0	57.67	31.78	37.88	9251	2804	5.56	9251	2863	69.96	722.81	2313	1028	814.33	3328	2594	2522	
Mass Flow (tonnes/day)	2667	2222	1375	2930	2706	1380	180.00	1471	2071	330.42	167.90	118.88	40.83	87.12	205.86	4000	1388	3.10	4000	1955	3.38	561.66	1000	444.44	352.09	2313	1744	1746	
H2O	666.67	222.22	0	413.42	30.18	20.50	0	27.16	0	0.09	0	0	0	0	0	0	0	1060	2.40	0	189.23	0	0	1000	444.44	352.09	0	0	0
CO	0	0	0	797.86	1575	0	0	795.10	0	64.47	0	0	0	0	0	0	0	15.27	0	0	0	0	0	0	0	0	0	0	0
H2	0	0	0	47.75	168.38	0	0	120.71	0	9.79	0	0	0	0	0	0	0	0.07	0	0	0	3.38	0	0	0	0	0	0	0
CO2	0	0	0	1427	167.80	1359	180.00	239.52	559.11	0	45.33	0	0	0	0	0	0	263.09	0	0	318.35	0	0	0	0	0	0	0	0
O2	0	0	320.19	0	0	0	148.99	0	0	0	0	0	0	0	0	0	0	0	0	0	80.99	0	528.66	0	0	0	535.65	0	410.78
N2	0	0	1055	0	0	0	1055	0	0	0	0	0	0	0	0	0	0	0	0	0	1336	0	0	0	0	0	1748	1744	1336
AR	0	0	0	33.00	411.50	0	0	380.61	0	30.86	0	0	0	0	0	0	0	0.69	0	0	30.86	0	33.00	0	0	0	29.71	0.10	0
CH4	0	0	0	103.81	149.06	0	0	52.01	0	4.22	0	11.56	0	0	0	0	0	4.58	0	0	0	0	0	0	0	0	0	0	0
C2H6	0	0	0	21.82	28.49	0	0	13.55	0	1.10	0	0	0	0	0	0	0	5.75	0	0	0	0	0	0	0	0	0	0	0
C2H4	0	0	0	44.96	36.05	0	0	0.36	0	0.03	0	0	0	0	0	0	0	8.11	0	0	0	0	0	0	0	0	0	0	0
C6H6	0	0	0	4.10	39.50	0	0	36.54	0	2.96	0	0	0	0	0	0	0	0.98	0	0	0	0	0	0	0	0	0	0	0
C3	0	0	0	0	54.18	0	0	58.62	0	4.75	0	29.27	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C4	0	0	0	0	34.31	0	0	41.73	0	3.38	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
H2S	0	0	0	4.51	0.13	0	0	0	0	0	0	0	0	0	0	0	0	1.87	0	0	0	0	0	0	0	0	0	0	0
NH3	0	0	0	19.09	1.52	0	0	0.28	0	0.02	0	0	0	0	0	0	0	17.36	0	0	0	0	0	0	0	0	0	0	0
TAR	0	0	0	10.57	0	0	0	0	0	0	0	0	0	0	0	0	0	10.57	0	0	0	0	0	0	0	0	0	0	0
SULFUR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
CARBON	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
STEAM	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	4000	0	0	4000	0	0	0	0	0	0	0	0	0	0
SO2	0	0	0	0	0	0	0	0.30	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
NO2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0.06	0	0	0	0	0	0	0	0
MEA	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
AIR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
WAXES	0	0	0	0	0	0	0	0	0	131.46	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C5	0	0	0	0	2.21	0	0	2.21	11.92	0.18	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C6	0	0	0	0	2.37	0	0	2.37	12.82	0.19	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C7	0	0	0	0	1.11	0	0	1.11	13.53	0.09	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C8	0	0	0	0	1.14	0	0	1.14	13.88	0.09	0	0	87.12	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C9	0	0	0	0	1.08	0	0	1.15	13.95	0.09	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C10	0	0	0	0	1.13	0	0	1.15	13.98	0.09	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C11	0	0	0	0	0.50	0	0	0.54	13.86	0.04	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C12	0	0	0	0	0.44	0	0	0.53	13.54	0.04	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C13	0	0	0	0	0.34	0	0	0.51	13.10	0.04	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C14	0	0	0	0	0.20	0	0	0.49	12.57	0.04	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C15	0	0	0	0	0	0	0	0.46	11.93	0.04	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C16	0	0	0	0	0.04	0	0	0.45	11.48	0.04	0	0	0	0	205.86	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C17	0	0	0	0	0	0	0	0	11.39	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C18	0	0	0	0	0	0	0	0	10.85	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C19	0	0	0	0	0	0	0	0	10.31	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C20	0	0	0	0	0	0	0	0	9.76	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
BIOMASS	2000	2000	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
ASH	0	0	0	1.01	0	0	0	0.01	0	0	0	118.88	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
CHAR	0	0	0	0.81	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0

**Volumetric and Mole flow values do not include biomass, ash, or char

Table E-9. Preprocessing Area Stream Data for LT Scenario

HT A100	CH00B/MAS	CH02B/MAS	CH03B/MAS	CH09B/MAS	DR05B/MAS	DR01S/TM	DR04S/TM	DR02W/AT	GR06B/MAS	GR09B/MAS	GR09B/MAS
Temperature (C)	25	25	25	0	90	200	120	120	90	90	0
Pressure (bar)	1.01	1.01	1.01	0.00	1.01	1.98	1.98	1.98	1.01	1.01	0.00
Vapor Fraction	0.00	0.00	0.00	0.00	0.00	1.00	1.00	1.00	0.00	0.00	0.00
Volume Flow** (m ³ /sec)	0.43	0.43	0.43	0	0.14	50.53	42.34	4.70	0.14	0.14	0
Mole Flow** (kmol/hr)	1542	1542	1542	0	513.97	9251	9251	1028	513.97	513.97	0
Mass Flow (tonnes/day)	2667	3181	2667	514.02	2222	4000	4000	444.44	3019	2222	796.81
H2O	666.67	666.67	666.67	0	222.22	0	0	444.44	222.22	222.22	0
CO	0	0	0	0	0	0	0	0	0	0	0
H2	0	0	0	0	0	0	0	0	0	0	0
CO2	0	0	0	0	0	0	0	0	0	0	0
O2	0	0	0	0	0	0	0	0	0	0	0
N2	0	0	0	0	0	0	0	0	0	0	0
AR	0	0	0	0	0	0	0	0	0	0	0
CH4	0	0	0	0	0	0	0	0	0	0	0
C2H6	0	0	0	0	0	0	0	0	0	0	0
C2H4	0	0	0	0	0	0	0	0	0	0	0
C6H6	0	0	0	0	0	0	0	0	0	0	0
C3	0	0	0	0	0	0	0	0	0	0	0
C4	0	0	0	0	0	0	0	0	0	0	0
H2S	0	0	0	0	0	0	0	0	0	0	0
NH3	0	0	0	0	0	0	0	0	0	0	0
TAR	0	0	0	0	0	0	0	0	0	0	0
SULFUR	0	0	0	0	0	0	0	0	0	0	0
CARBON	0	0	0	0	0	0	0	0	0	0	0
STEAM	0	0	0	0	0	4000	4000	0	0	0	0
SO2	0	0	0	0	0	0	0	0	0	0	0
NO2	0	0	0	0	0	0	0	0	0	0	0
MEA	0	0	0	0	0	0	0	0	0	0	0
AIR	0	0	0	0	0	0	0	0	0	0	0
WAXES	0	0	0	0	0	0	0	0	0	0	0
C5	0	0	0	0	0	0	0	0	0	0	0
C6	0	0	0	0	0	0	0	0	0	0	0
C7	0	0	0	0	0	0	0	0	0	0	0
C8	0	0	0	0	0	0	0	0	0	0	0
C9	0	0	0	0	0	0	0	0	0	0	0
C10	0	0	0	0	0	0	0	0	0	0	0
C11	0	0	0	0	0	0	0	0	0	0	0
C12	0	0	0	0	0	0	0	0	0	0	0
C13	0	0	0	0	0	0	0	0	0	0	0
C14	0	0	0	0	0	0	0	0	0	0	0
C15	0	0	0	0	0	0	0	0	0	0	0
C16	0	0	0	0	0	0	0	0	0	0	0
C17	0	0	0	0	0	0	0	0	0	0	0
C18	0	0	0	0	0	0	0	0	0	0	0
C19	0	0	0	0	0	0	0	0	0	0	0
C20	0	0	0	0	0	0	0	0	0	0	0
BIOMASS	2000	2514	2000	514.02	2000	0	0	0	2797	2000	796.81
ASH	0	0	0	0	0	0	0	0	0	0	0
CHAR	0	0	0	0	0	0	0	0	0	0	0

**Volumetric and Mole flow values do not include biomass, ash, or char

Table E-10. Gasification Area Stream Data for LT Scenario

HT A200	CB52FLUE	CB61ASH	CB62ASH	CY14SGAS	CY15CHAR	CY16CHAR	GS05SGAS	GS07SGAS	GS08CAIR	GS11BMAS	GS19CHAR	GS20SGAS	GS43CO2	GS46CO2	GS52FLUE	GS81STM	GS84STM	GS90OX	GS91STM
Temperature (C)	1200	0	0	870	0	0	871	870	100	96	0	870	243	100	200	200	120	149	204
Pressure (bar)	1.00	1.00	1.00	27.57	27.57	27.55	28.00	27.58	1.01	1.01	27.55	27.55	28.00	27.58	1.00	1.98	1.98	22.00	22.00
Vapor Fraction	1.00	0.00	0.00	1.00	0.00	0.00	1.00	1.00	1.00	0.00	0.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	0.00
Volume Flow** (m ³ /sec)	69.99	0	0	2.38	0	0	4.69	4.76	16.89	0	0	4.76	0.07	0.05	22.48	50.53	42.34	0.32	0.01
Mole Flow** (kmol/hr)	2053	0	0	2470	0	0	4931	4939	1985	513.97	0	4939	170.42	161.90	2053	9251	9251	722.81	814.33
Mass Flow (tonnes/day)	1471	17.82	118.88	1477	95.38	12.09	3136	3145	1375	2222	214.95	2930	180.00	171.00	1471	4000	4000	561.66	352.09
H2O	27.16	0	0	206.71	0	0	413.42	413.42	0	222.22	0	413.42	0	0	27.16	0	0	0	352.09
CO	0	0	0	398.93	0	0	797.86	797.86	0	0	0	797.86	0	0	0	0	0	0	0
H2	0	0	0	23.87	0	0	47.75	47.75	0	0	0	47.75	0	0	0	0	0	0	0
CO2	239.52	0	0	713.66	0	0	1418	1427	0	0	0	1427	180.00	171.00	239.52	0	0	0	0
O2	148.99	0	0	0	0	0	0	0	320.19	0	0	0	0	0	148.99	0	0	528.66	0
N2	1055	0	0	0	0	0	0	0	1055	0	0	0	0	0	1055	0	0	0	0
AR	0	0	0	16.50	0	0	33.00	33.00	0	0	0	33.00	0	0	0	0	0	33.00	0
CH4	0	0	0	51.90	0	0	103.81	103.81	0	0	0	103.81	0	0	0	0	0	0	0
C2H6	0	0	0	10.91	0	0	21.82	21.82	0	0	0	21.82	0	0	0	0	0	0	0
C2H4	0	0	0	22.48	0	0	44.96	44.96	0	0	0	44.96	0	0	0	0	0	0	0
C6H6	0	0	0	2.05	0	0	4.10	4.10	0	0	0	4.10	0	0	0	0	0	0	0
C3	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C4	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
H2S	0	0	0	2.25	0	0	4.51	4.51	0	0	0	4.51	0	0	0	0	0	0	0
NH3	0	0	0	9.54	0	0	19.09	19.09	0	0	0	19.09	0	0	0	0	0	0	0
TAR	0	0	0	5.29	0	0	10.57	10.57	0	0	0	10.57	0	0	0	0	0	0	0
SULFUR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
CARBON	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
STEAM	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	4000	4000	0	0
SO2	0.30	0	0	0	0	0	0	0	0	0	0	0	0	0	0.30	0	0	0	0
NO2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
MEA	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
AIR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
WAXES	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C5	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C6	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C7	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C8	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C9	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C10	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C11	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C12	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C13	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C14	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C15	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C16	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C17	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C18	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C19	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C20	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
BIOMASS	0	0	0	0	0	0	0	0	0	2000	0	0	0	0	0	0	0	0	0
ASH	0.01	17.82	118.88	7.19	52.75	6.69	119.90	119.90	0	0	118.89	1.01	0	0	0.01	0	0	0	0
CHAR	0	0	0	5.81	42.63	5.40	96.87	96.87	0	0	96.06	0.81	0	0	0	0	0	0	0

**Volumetric and Mole flow values do not include biomass, ash, or char

Table E-11. Syngas Cleaning Area Stream Data for LT Scenario

HT A300	CL01H2O	CL02H2O	CL21SGAS	CL25ACG	CL25SGAS	CL25WAT	CL28SGAS	CL40CO2	CL40SGAS	CL41CO2	CL43CO2	CL49SGAS	CL80H2O	CL80WAT	CL81WAT	CL83SUL	CL83WAT	CL84WAT	CL92AIR	CL95AIR
Temperature (C)	194	150	870	50	194	0	40	50	62	50	243	32	30	30	40	50	98	40	50	50
Pressure (bar)	27.55	27.55	27.55	3.45	27.55	27.55	26.86	3.45	22.89	3.45	28.00	22.89	27.57	27.57	26.86	2.07	27.55	27.55	2.07	2.07
Vapor Fraction	0.00	0.00	1.00	1.00	1.00	0.00	1.00	1.00	1.00	1.00	1.00	1.00	0.00	0.00	0.00	0.00	0.00	0.00	1.00	1.00
Volume Flow** (m ³ /sec)	0.08	0.08	4.76	3.28	2.03	0	0.98	3.21	2.40	2.84	0.07	1.50	0	0.01	0.02	0	0.07	0.07	0	0.10
Mole Flow** (kmol/hr)	13877	13877	4939	1538	5286	0	3638	1505	7066	1334	170.42	4869	346.93	1156	2804	5.56	13877	13877	0.45	27.74
Mass Flow (tonnes/day)	6000	6000	2930	1586	3078	1.82	2190	1560	2706	1380	180.00	2071	150.00	500.00	1388	3.10	6000	6000	0.35	22.89
H2O	6000	6000	413.42	22.78	563.42	0	3.78	20.50	30.18	20.50	0	0	150.00	500.00	1060	2.40	6000	6000	0	0.27
CO	0	0	797.86	3.07	797.86	0	782.59	0	1575	0	0	795.10	0	0	15.27	0	0	0	0	3.07
H2	0	0	47.75	0	47.75	0	47.68	0	168.38	0	0	120.71	0	0	0.07	0	0	0	0	0
CO2	0	0	1427	1539	1427	0	1164	1539	167.80	1359	180.00	559.11	0	0	263.09	0	0	0	0	0
O2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0.35	0
N2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
AR	0	0	33.00	1.42	33.00	0	32.31	0	411.50	0	0	380.61	0	0	0.69	0	0	0	0	1.42
CH4	0	0	103.81	2.18	103.81	0	99.23	0	149.06	0	0	52.01	0	0	4.58	0	0	0	0	2.18
C2H6	0	0	21.82	1.11	21.82	0	16.07	0	28.49	0	0	13.55	0	0	5.75	0	0	0	0	1.11
C2H4	0	0	44.96	1.14	44.96	0	36.85	0	36.05	0	0	0.36	0	0	8.11	0	0	0	0	1.14
C6H6	0	0	4.10	0.15	4.10	0	3.11	0	39.50	0	0	36.54	0	0	0.98	0	0	0	0	0.15
C3	0	0	0	4.29	0	0	0	0	54.18	0	0	58.62	0	0	0	0	0	0	0	4.29
C4	0	0	0	6.73	0	0	0	0	34.31	0	0	41.73	0	0	0	0	0	0	0	6.73
H2S	0	0	4.51	2.48	4.51	0	2.64	0	0.13	0	0	0	0	0	1.87	0	0	0	0	1.73
NH3	0	0	19.09	0.42	19.09	0	1.73	0	1.52	0	0	0.28	0	0	17.36	0	0	0	0	0.42
TAR	0	0	10.57	0	10.57	0	0	0	0	0	0	0	0	0	10.57	0	0	0	0	0
SULFUR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
CARBON	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
STEAM	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
SO2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
NO2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
MEA	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
AIR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
WAXES	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C5	0	0	0	0	0	0	0	0	2.21	0	0	2.21	0	0	0	0	0	0	0	0
C6	0	0	0	0	0	0	0	0	2.37	0	0	2.37	0	0	0	0	0	0	0	0
C7	0	0	0	0	0	0	0	0	1.11	0	0	1.11	0	0	0	0	0	0	0	0
C8	0	0	0	0	0	0	0	0	1.14	0	0	1.14	0	0	0	0	0	0	0	0
C9	0	0	0	0.07	0	0	0	0	1.08	0	0	1.15	0	0	0	0	0	0	0	0.07
C10	0	0	0	0.02	0	0	0	0	1.13	0	0	1.15	0	0	0	0	0	0	0	0.02
C11	0	0	0	0.03	0	0	0	0	0.50	0	0	0.54	0	0	0	0	0	0	0	0.03
C12	0	0	0	0.07	0	0	0	0	0.44	0	0	0.53	0	0	0	0	0	0	0	0.07
C13	0	0	0	0.10	0	0	0	0	0.34	0	0	0.51	0	0	0	0	0	0	0	0.10
C14	0	0	0	0.09	0	0	0	0	0.20	0	0	0.49	0	0	0	0	0	0	0	0.09
C15	0	0	0	0	0	0	0	0	0	0	0	0.46	0	0	0	0	0	0	0	0
C16	0	0	0	0.02	0	0	0	0	0.04	0	0	0.45	0	0	0	0	0	0	0	0.02
C17	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C18	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C19	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
C20	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
BIOMASS	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
ASH	0	0	1.01	0	0	1.01	0	0	0	0	0	0	0	0	0	0	0	0	0	0
CHAR	0	0	0.81	0	0	0.81	0	0	0	0	0	0	0	0	0	0	0	0	0	0

**Volumetric and Mole flow values do not include biomass, ash, or char

Table E-12. Acid Gas Removal and Sulfur Recovery Areas Stream Data for LT Scenario

HT A300AGR	AG02MEAR	AG03MEAR	AG04MEAR	AG06MEAR	AG08MEAR	AG09MEAR	AG10MEAR	AG11MEAR	AG23ACG	AG24ACG	AG25ACG	AG26SGAS	AG32SGAS	AG33SGAS	AG49SGAS	A300SUL	SU25ACG	SU26SUL	SU40CO2	SU68SUL	SU92AIR	SU95AIR	
Temperature (C)	50	62	50	93	50	123	80	80	93	3.45	3.45	3.45	26.86	22.89	22.89	32							
Pressure (bar)	22.89	22.89	22.89	3.45	3.45	3.45	3.45	26.00	3.45	3.45	3.45	26.86	22.89	22.89	22.89		3.45	3.45	3.45	2.07	2.07	2.07	
Vapor Fraction	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	1.00	0.88	1.00	1.00	0.93	1.00	1.00		1.00	0.82	1.00	0.00	1.00	1.00	
Volume Flow** (m³/sec)	0.31	0	0.31	0.33	0	0.31	0.30	0.30	4.24	3.28	3.28	0.98	2.30	2.40	1.50		3.28	0.06	3.21	0	0	0.10	
Mole Flow** (kmol/hr)	54067	449.42	54516	54516	207.83	52978	52978	52978	1746	1746	1538	3638	7515	7066	4869		1538	33.30	1505	5.56	0.45	27.74	
Mass Flow (tonnes/day)	28640	208.71	28848	28848	101.05	27263	27263	27263	1687	1687	1586	2190	2915	2706	2071		1586	25.65	1560	3.10	0.35	22.89	
H2O	20923	181.16	21104	21104	82.74	21081	21081	21081	105.52	105.52	22.78	3.78	211.34	30.18	0		22.78	2.28	20.50	2.40	0	0.27	
CO	0	3.07	3.07	3.07	0	0	0	0	3.07	3.07	3.07	782.59	1578	1575	795.10		3.07	3.07	0	0	0	3.07	
H2	0	0	0	0	0	0	0	0	0	0	0	47.68	168.38	168.38	120.71		0	0	0	0	0	0	
CO2	1551	4.53	1556	1556	16.15	16.15	16.15	16.15	1556	1556	1539	1164	172.33	167.80	559.11		1539	0	1539	0	0	0	
O2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0.35	0	
N2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	
AR	0	1.42	1.42	1.42	0	0	0	0	1.42	1.42	1.42	32.31	412.92	411.50	380.61		1.42	1.42	0	0	0	1.42	
CH4	0	2.19	2.19	2.19	0.01	0.01	0.01	0.01	2.19	2.19	2.18	99.23	151.24	149.06	52.01		2.18	2.18	0	0	0	2.18	
C2H6	0	1.13	1.13	1.13	0.02	0.02	0.02	0.02	1.13	1.13	1.11	16.07	29.62	28.49	13.55		1.11	1.11	0	0	0	1.11	
C2H4	0	1.16	1.16	1.16	0.01	0.01	0.01	0.01	1.16	1.16	1.14	36.85	37.21	36.05	0.36		1.14	1.14	0	0	0	1.14	
C6H6	0	0.15	0.15	0.15	0	0	0	0	0.15	0.15	0.15	3.11	39.65	39.50	36.54		0.15	0.15	0	0	0	0.15	
C3	0	4.44	4.44	4.44	0.15	0.15	0.15	0.15	4.44	4.44	4.29	0	58.62	54.18	58.62		4.29	4.29	0	0	0	4.29	
C4	0	7.42	7.42	7.42	0.69	0.69	0.69	0.69	7.41	7.41	6.73	0	41.73	34.31	41.73		6.73	6.73	0	0	0	6.73	
H2S	2.51	0.01	2.52	2.52	0.04	0.04	0.04	0.04	2.52	2.52	2.48	2.64	0.13	0.13	0		2.48	2.48	0	0	0	1.73	
NH3	0	0.49	0.49	0.49	0.07	0.07	0.07	0.07	0.49	0.49	0.42	1.73	2.01	1.52	0.28		0.42	0.42	0	0	0	0.42	
TAR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	
SULFUR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	
CARBON	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	
STEAM	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	
SO2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	
NO2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	
MEA	6163	0	6163	6163	0	6163	6163	6163	0	0	0	0	0	0	0		0	0	0	0	0	0	
AIR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	
WAXES	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	
C5	0	0	0	0	0	0	0	0	0	0	0	0	2.21	2.21	2.21		0	0	0	0	0	0	
C6	0	0	0	0	0	0	0	0	0	0	0	0	2.37	2.37	2.37		0	0	0	0	0	0	
C7	0	0	0	0	0	0	0	0	0	0	0	0	1.11	1.11	1.11		0	0	0	0	0	0	
C8	0	0	0	0	0	0	0	0	0	0	0	0	1.14	1.14	1.14		0	0	0	0	0	0	
C9	0	0.07	0.07	0.07	0	0	0	0	0.07	0.07	0.07	0	1.15	1.08	1.15		0.07	0.07	0	0	0	0.07	
C10	0	0.02	0.02	0.02	0	0	0	0	0.02	0.02	0.02	0	1.15	1.13	1.15		0.02	0.02	0	0	0	0.02	
C11	0	0.04	0.04	0.04	0	0	0	0	0.04	0.04	0.03	0	0.54	0.50	0.54		0.03	0.03	0	0	0	0.03	
C12	0	0.08	0.08	0.08	0.02	0.02	0.02	0.02	0.08	0.08	0.07	0	0.53	0.44	0.53		0.07	0.07	0	0	0	0.07	
C13	0	0.17	0.17	0.17	0.08	0.08	0.08	0.08	0.17	0.17	0.10	0	0.51	0.34	0.51		0.10	0.10	0	0	0	0.10	
C14	0	0.29	0.29	0.29	0.20	0.20	0.20	0.20	0.29	0.29	0.09	0	0.49	0.20	0.49		0.09	0.09	0	0	0	0.09	
C15	0	0.46	0.46	0.46	0.46	0.46	0.46	0.46	0.46	0.46	0	0	0.46	0	0.46		0	0	0	0	0	0	
C16	0	0.41	0.41	0.41	0.38	0.38	0.38	0.38	0.41	0.41	0.02	0	0.45	0.04	0.45		0.02	0.02	0	0	0	0.02	
C17	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	
C18	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	
C19	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	
C20	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	
BIOMASS	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	
ASH	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	
CHAR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0		0	0	0	0	0	0	

**Volumetric and Mole flow values do not include biomass, ash, or char

Table E-13. Fuel Synthesis Area Stream Data for LT Scenario

HT A400	FS34SGAS	FS35SGAS	FS36SGAS	FS37SGAS	FS40SGAS	FS41GAS	FS50SGAS	FS51FT	FS52FT	FS53SGAS	FS54FT	FS55SGAS	FS56FT	FS56SGAS	FS57SGAS	FS58SGAS	FS59SGAS	FS60WAT	FS60HYD
Temperature (C)	62	78	200	200	60	235	200	200	35	32	35	32	35	32	32	41	200	35	60
Pressure (bar)	22.89	26.50	25.12	25.12	1.01	25.00	24.96	22.89	22.89	22.89	22.89	22.89	22.89	22.89	22.89	25.30	24.96	22.89	1.01
Vapor Fraction	1.00	1.00	1.00	1.00	0.88	1.00	1.00	1.00	0.72	1.00	0.00	1.00	0.00	1.00	1.00	1.00	1.00	0.00	1.00
Volume Flow** (m ³ /sec)	2.40	2.18	4.44	0.07	0.65	0.05	5.02	4.39	2.08	2.02	0.01	1.50	0.01	0.12	0.41	0.38	0.58	0.01	0.53
Mole Flow** (kmol/hr)	7066	7066	10153	166.97	97.01	97.01	11399	9268	9268	6579	70.21	4869	69.40	394.75	1316	1316	1316	2619	69.96
Mass Flow (tonnes/day)	2706	2706	3705	60.93	57.55	57.55	4261	4261	4261	2798	330.77	2071	330.42	167.90	559.66	559.66	559.66	1132	3.38
H2O	30.18	30.18	671.91	11.05	11.05	11.05	671.91	1133	1133	0	0.43	0	0.09	0	0	0	0	1132	0
CO	1575	1575	1576	25.92	25.92	25.92	1791	1074	1074	0	795.10	0	64.47	214.89	214.89	214.89	0	0	0
H2	168.38	168.38	242.13	3.98	0.60	0.60	271.37	163.12	163.12	163.12	0	120.71	0	9.79	32.62	32.62	32.62	0	3.38
CO2	167.80	167.80	604.44	9.94	9.94	9.94	755.55	755.55	755.55	755.55	0	559.11	0	45.33	151.11	151.11	151.11	0	0
O2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
N2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
AR	411.50	411.50	411.50	6.77	6.77	6.77	514.36	514.36	514.36	514.34	0	380.61	0	30.86	102.87	102.87	102.87	0	0
CH4	149.06	149.06	52.13	0.86	0.86	0.86	66.19	70.29	70.29	70.29	0	52.01	0	4.22	14.06	14.06	14.06	0	0
C2H6	28.49	28.49	7.72	0.13	0.13	0.13	11.38	18.30	18.30	18.30	0	13.55	0	1.10	3.66	3.66	3.66	0	0
C2H4	36.05	36.05	0.39	0.01	0.01	0.01	0.49	0.49	0.49	0.49	0	0.36	0	0.03	0.10	0.10	0.10	0	0
C6H6	39.50	39.50	39.50	0.65	0.65	0.65	49.38	49.38	49.38	49.38	0	36.54	0	2.96	9.88	9.88	9.88	0	0
C3	54.18	54.18	54.18	0.89	0.89	0.89	70.02	79.16	79.16	79.22	0	58.62	0	4.75	15.84	15.84	15.84	0	0
C4	34.31	34.31	34.31	0.56	0.56	0.56	45.59	56.43	56.43	56.40	0	41.73	0	3.38	11.28	11.28	11.28	0	0
H2S	0.13	0.13	0.01	0	0	0	0.01	0.01	0.01	0.01	0	0	0	0	0	0	0	0	0
NH3	1.52	1.52	0.30	0	0	0	0.38	0.38	0.38	0.38	0	0.28	0	0.02	0.08	0.08	0.08	0	0
TAR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
SULFUR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
CARBON	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
STEAM	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
SO2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
NO2	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
MEA	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
AIR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
WAXES	0	0	0	0	0	0	0	131.46	131.46	0	131.46	0	131.46	0	0	0	0	0	0
C5	2.21	2.21	2.21	0.04	0.04	0.04	2.80	14.91	14.91	2.98	11.92	2.21	11.92	0.18	0.60	0.60	0.60	0	0
C6	2.37	2.37	2.37	0.04	0.04	0.04	3.01	16.02	16.02	3.20	12.82	2.37	12.82	0.19	0.64	0.64	0.64	0	0
C7	1.11	1.11	1.11	0.02	0.02	0.02	1.41	15.03	15.03	1.50	13.53	1.11	13.53	0.09	0.30	0.30	0.30	0	0
C8	1.14	1.14	1.14	0.02	0.02	0.02	1.45	15.42	15.42	1.54	13.88	1.14	13.88	0.09	0.31	0.31	0.31	0	0
C9	1.08	1.08	1.08	0.02	0.02	0.02	1.39	15.50	15.50	1.55	13.95	1.15	13.95	0.09	0.31	0.31	0.31	0	0
C10	1.13	1.13	1.13	0.02	0.02	0.02	1.44	15.54	15.54	1.55	13.98	1.15	13.98	0.09	0.31	0.31	0.31	0	0
C11	0.50	0.50	0.50	0.01	0.01	0.01	0.65	14.59	14.59	0.73	13.86	0.54	13.86	0.04	0.15	0.15	0.15	0	0
C12	0.44	0.44	0.44	0.01	0.01	0.01	0.59	14.26	14.26	0.71	13.54	0.53	13.54	0.04	0.14	0.14	0.14	0	0
C13	0.34	0.34	0.34	0.01	0.01	0.01	0.47	13.79	13.79	0.69	13.10	0.51	13.10	0.04	0.14	0.14	0.14	0	0
C14	0.20	0.20	0.20	0	0	0	0.34	13.23	13.23	0.66	12.57	0.49	12.57	0.04	0.13	0.13	0.13	0	0
C15	0	0	0	0	0	0	0.13	12.55	12.55	0.63	11.93	0.46	11.93	0.04	0.13	0.13	0.13	0	0
C16	0.04	0.04	0.04	0	0	0	0.16	12.08	12.08	0.60	11.48	0.45	11.48	0.04	0.12	0.12	0.12	0	0
C17	0	0	0	0	0	0	0	11.39	11.39	0	11.39	0	11.39	0	0	0	0	0	0
C18	0	0	0	0	0	0	0	10.85	10.85	0	10.85	0	10.85	0	0	0	0	0	0
C19	0	0	0	0	0	0	0	10.31	10.31	0	10.31	0	10.31	0	0	0	0	0	0
C20	0	0	0	0	0	0	0	9.76	9.76	0	9.76	0	9.76	0	0	0	0	0	0
BIOMASS	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
ASH	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
CHAR	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0

**Volumetric and Mole flow values do not include biomass, ash, or char

Table E-14. Syngas Conditioning Area Stream Data for LT Scenario

HT A400COND	CD36SGAS	CD36SGAS	CD37SGAS	CD38SGAS	CD39SGAS	CD40SGAS	CD41SGAS	CD42SGAS	CD44SGAS	CD45SGAS	CD80STM	CD81STM	CD98CONT
Temperature (C)	78	150	150	870	870	300	300	474	363	200	300	870	150
Pressure (bar)	26.50	26.50	26.50	26.50	25.81	25.81	25.81	25.12	25.12	25.12	25.00	27.00	26.50
Vapor Fraction	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00	1.00
Volume Flow** (m ³ /sec)	2.18	2.64	2.64	7.08	10.44	5.24	1.83	2.46	5.98	4.44	1.15	2.26	0
Mole Flow** (kmol/hr)	7066	7066	7063	7063	10153	10153	3554	3554	10153	10153	2313	2313	3.12
Mass Flow (tonnes/day)	2706	2706	2705	2705	3705	3705	1297	1297	3705	3705	1000	1000	1.34
H2O	30.18	30.18	30.18	30.18	850.65	850.65	297.73	118.99	671.91	671.91	1000	1000	0
CO	1575	1575	1575	1575	1854	1854	648.81	370.91	1576	1576	0	0	0
H2	168.38	168.38	168.38	168.38	222.13	222.13	77.75	97.75	242.13	242.13	0	0	0
CO2	167.80	167.80	167.80	167.80	167.80	167.80	58.73	495.37	604.44	604.44	0	0	0
O2	0	0	0	0	0	0	0	0	0	0	0	0	0
N2	0	0	0	0	0	0	0	0	0	0	0	0	0
AR	411.50	411.50	411.50	411.50	411.50	411.50	144.02	144.02	411.50	411.50	0	0	0
CH4	149.06	149.06	149.06	149.06	52.13	52.13	18.25	18.25	52.13	52.13	0	0	0
C2H6	28.49	28.49	28.49	28.49	7.72	7.72	2.70	2.70	7.72	7.72	0	0	0
C2H4	36.05	36.05	36.05	36.05	0.39	0.39	0.14	0.14	0.39	0.39	0	0	0
C6H6	39.50	39.50	39.50	39.50	39.50	39.50	13.83	13.83	39.50	39.50	0	0	0
C3	54.18	54.18	54.18	54.18	54.18	54.18	18.96	18.96	54.18	54.18	0	0	0
C4	34.31	34.31	34.31	34.31	34.31	34.31	12.01	12.01	34.31	34.31	0	0	0
H2S	0.13	0.13	0.01	0.01	0.01	0.01	0	0	0.01	0.01	0	0	0.12
NH3	1.52	1.52	0.30	0.30	0.30	0.30	0.11	0.11	0.30	0.30	0	0	1.21
TAR	0	0	0	0	0	0	0	0	0	0	0	0	0
SULFUR	0	0	0	0	0	0	0	0	0	0	0	0	0
CARBON	0	0	0	0	0	0	0	0	0	0	0	0	0
STEAM	0	0	0	0	0	0	0	0	0	0	0	0	0
SO2	0	0	0	0	0	0	0	0	0	0	0	0	0
NO2	0	0	0	0	0	0	0	0	0	0	0	0	0
MEA	0	0	0	0	0	0	0	0	0	0	0	0	0
AIR	0	0	0	0	0	0	0	0	0	0	0	0	0
WAXES	0	0	0	0	0	0	0	0	0	0	0	0	0
C5	2.21	2.21	2.21	2.21	2.21	2.21	0.77	0.77	2.21	2.21	0	0	0
C6	2.37	2.37	2.37	2.37	2.37	2.37	0.83	0.83	2.37	2.37	0	0	0
C7	1.11	1.11	1.11	1.11	1.11	1.11	0.39	0.39	1.11	1.11	0	0	0
C8	1.14	1.14	1.14	1.14	1.14	1.14	0.40	0.40	1.14	1.14	0	0	0
C9	1.08	1.08	1.08	1.08	1.08	1.08	0.38	0.38	1.08	1.08	0	0	0
C10	1.13	1.13	1.13	1.13	1.13	1.13	0.40	0.40	1.13	1.13	0	0	0
C11	0.50	0.50	0.50	0.50	0.50	0.50	0.18	0.18	0.50	0.50	0	0	0
C12	0.44	0.44	0.44	0.44	0.44	0.44	0.16	0.16	0.44	0.44	0	0	0
C13	0.34	0.34	0.34	0.34	0.34	0.34	0.12	0.12	0.34	0.34	0	0	0
C14	0.20	0.20	0.20	0.20	0.20	0.20	0.07	0.07	0.20	0.20	0	0	0
C15	0	0	0	0	0	0	0	0	0	0	0	0	0
C16	0.04	0.04	0.04	0.04	0.04	0.04	0.01	0.01	0.04	0.04	0	0	0
C17	0	0	0	0	0	0	0	0	0	0	0	0	0
C18	0	0	0	0	0	0	0	0	0	0	0	0	0
C19	0	0	0	0	0	0	0	0	0	0	0	0	0
C20	0	0	0	0	0	0	0	0	0	0	0	0	0
BIOMASS	0	0	0	0	0	0	0	0	0	0	0	0	0
ASH	0	0	0	0	0	0	0	0	0	0	0	0	0
CHAR	0	0	0	0	0	0	0	0	0	0	0	0	0

**Volumetric and Mole flow values do not include biomass, ash, or char

Table E-15. Hydroprocessing, Power Generation, and Air Separation Areas Stream Data for LT Scenario

HT A500	HY56FT	HY65FGAS	HY70GASO	HY71GASO	HY80DIES	HY81DIES	HY90HYD	A600	421	602	615	690	CW2	CWREC	HPW1	HPSTM	A700	AIR-1A	AIR-A	AIR-L	N2-OUT	O2-LPC	O2-OUT	O2-OUT2
Temperature (C)	35	50	50	50	50	50	60		32	1161	344	30	85	23	170	393		-170	32	20	16	35	16	55
Pressure (bar)	22.89	22.89	22.89	22.89	22.89	22.89	1.01		22.89	1.00	1.00	1.01	7.91	0.02	7.91	173.38		6.20	1.01	6.30	1.10	23.08	1.10	27.00
Vapor Fraction	0.00	1.00	0.00	0.00	0.00	0.00	1.00		1.00	1.00	1.00	1.00	0.00	0.89	0.00	1.00		1.00	1.00	1.00	1.00	1.00	1.00	1.00
Volume Flow** (m ³ /sec)	0.01	0.02	0	0	0	0	0.53		0.12	94.88	40.84	17.42	0.01	442.16	0	0.03		1.06	23.17	3.58	15.73	0.22	4.45	0.21
Mole Flow** (kmol/hr)	69.40	57.67	31.78	31.78	37.88	37.88	69.96		394.75	2863	2863	2522	1709	1709	1709	1709		3162	3328	3328	2594	734.20	734.20	734.20
Mass Flow (tonnes/day)	330.42	40.83	87.12	87.12	205.86	205.86	3.38		167.90	1955	1955	1746	740.10	740.10	740.10	740.10		2198	2313	2313	1744	569.18	569.18	569.18
H2O	0.09	0	0	0	0	0	0		0	189.23	189.23	0	740.10	740.10	740.10	740.10		0	0	0	0	0	0	0
CO	0	0	0	0	0	0	0		64.47	0	0	0	0	0	0	0		0	0	0	0	0	0	0
H2	0	0	0	0	0	0	3.38		9.79	0	0	0	0	0	0	0		0	0	0	0	0	0	0
CO2	0	0	0	0	0	0	0		45.33	318.35	318.35	0	0	0	0	0		0	0	0	0	0	0	0
O2	0	0	0	0	0	0	0		0	80.99	80.99	410.78	0	0	0	0		508.87	535.65	535.65	0	535.65	535.65	535.65
N2	0	0	0	0	0	0	0		0	1336	1336	1336	0	0	0	0		1661	1748	1748	1744	3.91	3.91	3.91
AR	0	0	0	0	0	0	0		30.86	30.86	30.86	0	0	0	0	0		28.22	29.71	29.71	0.10	29.61	29.61	29.61
CH4	0	11.56	0	0	0	0	0		4.22	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C2H6	0	0	0	0	0	0	0		1.10	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C2H4	0	0	0	0	0	0	0		0.03	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C6H6	0	0	0	0	0	0	0		2.96	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C3	0	29.27	0	0	0	0	0		4.75	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C4	0	0	0	0	0	0	0		3.38	0	0	0	0	0	0	0		0	0	0	0	0	0	0
H2S	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
NH3	0	0	0	0	0	0	0		0.02	0	0	0	0	0	0	0		0	0	0	0	0	0	0
TAR	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
SULFUR	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
CARBON	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
STEAM	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
SO2	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
NO2	0	0	0	0	0	0	0		0	0.06	0.06	0	0	0	0	0		0	0	0	0	0	0	0
MEA	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
AIR	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
WAXES	131.46	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C5	11.92	0	0	0	0	0	0		0.18	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C6	12.82	0	0	0	0	0	0		0.19	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C7	13.53	0	0	0	0	0	0		0.09	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C8	13.88	0	87.12	87.12	0	0	0		0.09	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C9	13.95	0	0	0	0	0	0		0.09	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C10	13.98	0	0	0	0	0	0		0.09	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C11	13.86	0	0	0	0	0	0		0.04	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C12	13.54	0	0	0	0	0	0		0.04	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C13	13.10	0	0	0	0	0	0		0.04	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C14	12.57	0	0	0	0	0	0		0.04	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C15	11.93	0	0	0	0	0	0		0.04	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C16	11.48	0	0	0	205.86	205.86	0		0.04	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C17	11.39	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C18	10.85	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C19	10.31	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
C20	9.76	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
BIOMASS	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
ASH	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0
CHAR	0	0	0	0	0	0	0		0	0	0	0	0	0	0	0		0	0	0	0	0	0	0

**Volumetric and Mole flow values do not include biomass, ash, or char

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14. ABSTRACT (Maximum 200 Words) This study compares capital and production costs of two biomass-to-liquid production plants based on gasification. The first biorefinery scenario is an oxygen-fed, low-temperature (870°C), non-slugging, fluidized bed gasifier. The second scenario is an oxygen-fed, high-temperature (1,300°C), slugging, entrained flow gasifier. Both are followed by catalytic Fischer-Tropsch synthesis and hydroprocessing to naphtha-range (gasoline blend stock) and distillate-range (diesel blend stock) liquid fractions. Process modeling software (Aspen Plus) is utilized to organize the mass and energy streams and cost estimation software is used to generate equipment costs. Economic analysis is performed to estimate the capital investment and operating costs. Results show that the total capital investment required for nth plant scenarios is \$610 million and \$500 million for high-temperature and low-temperature scenarios, respectively. Product value (PV) for the high-temperature and low-temperature scenarios is estimated to be \$4.30 and \$4.80 per gallon of gasoline equivalent (GGE), respectively, based on a feedstock cost of \$75 per dry short ton. Sensitivity analysis is also performed on process and economic parameters. This analysis shows that total capital investment and feedstock cost are among the most influential parameters affecting the PV.					
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