

# PROJECT facts

U.S. DEPARTMENT OF ENERGY  
OFFICE OF FOSSIL ENERGY  
NATIONAL ENERGY TECHNOLOGY LABORATORY

Gasification  
Technologies

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## OXYGEN CARRIER DEVELOPMENT FOR CHEMICAL LOOPING COMBUSTION

### Background

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Chemical looping combustion (CLC) is a flameless combustion technology where there is no direct contact between air and fuel. Theoretical analyses of the technology indicates that its use results in increased power efficiency compared to regular combustion. Another theoretical study indicates IGCLC showed better efficiency than that with IGCC. The products of CLC are  $\text{CO}_2$  and  $\text{H}_2\text{O}$ . Thus, once the steam is condensed, a relatively pure stream of  $\text{CO}_2$  is produced ready for sequestration. The combustion process does not produce  $\text{NO}_x$ . The production of a sequestration ready  $\text{CO}_2$  stream does not require any additional separation units and there is no energy penalty or reduction in power plant efficiency.

The majority of the work performed to date on CLC has been performed using methane as the fuel. There are only limited studies with oxygen carriers used to combust coal-derived synthesis gas. Those few studies that have been performed using coal-derived fuel gas suggest that more experimental information is needed.

### Accomplishments

A comprehensive literature search was performed to identify those chemical looping processes that involve coal gasification. Heat calculations for the reduction and oxidation reactions with synthesis gas were completed. Various oxygen carriers, including NiO on bentonite and CuO on bentonite have been prepared and evaluated as oxygen carriers at 700, 800 and 900 °C using thermal gravimetric analysis (TGA). Stable reactivity was observed over 10 cycles of oxidation with air and reduction with synthesis gas. Rate information was computed for reactions over both oxygen carriers. Particle size, temperature and pressure affected the rate. Performance of these oxygen carriers were evaluated in a high pressure flow reactor at 150 psi with synthesis gas at 700, 800 and 900 °C. Stable reactivity was observed over 3 high pressure cycles, and complete combustion of hydrogen and complete utilization of the metal oxide were observed. Novel nano-composite oxygen carriers were also developed in collaboration with University of Pittsburgh to improve the rates.

#### PARTNER

University of Pittsburgh





## Benefits

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Chemical looping combustion (CLC) can lead to increased power efficiency reduced combustion exergy. CLC produces a stream of combustion products that primarily consist of CO<sub>2</sub> and steam. A relatively pure stream of CO<sub>2</sub> that is sequestration ready can be produced by simply condensing the steam. This avoids the energy penalty traditional fossil fuel fired combustors must pay to produce a pure stream of CO<sub>2</sub>. Additionally, CLC avoids production of NO<sub>x</sub> that is produced in almost all other combustion processes.

Development of improved oxygen carriers could increase their mechanical strength leading to increased service life, improve the rate of oxidation and reduction steps, avoid carbon deposition, and improve the stability during multi cycle tests. These improvements could significantly improve the economics of CLC.