

# Nanostructured Bulk Materials

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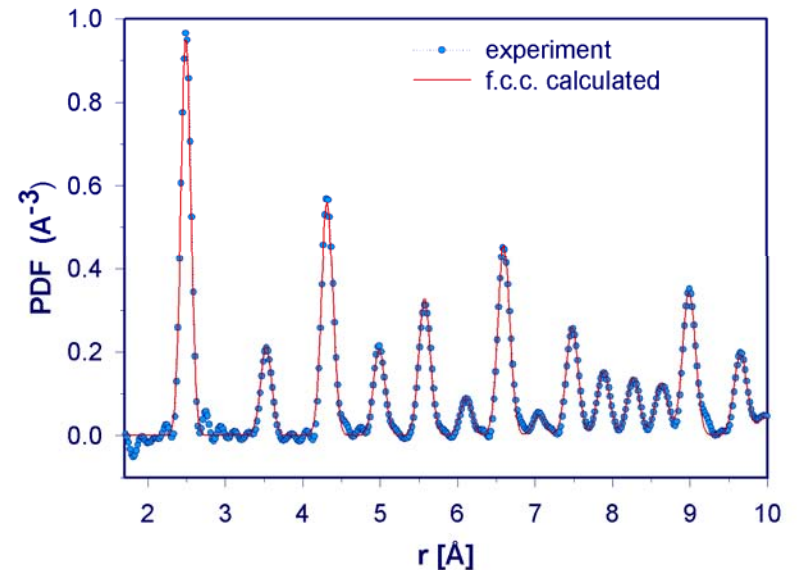
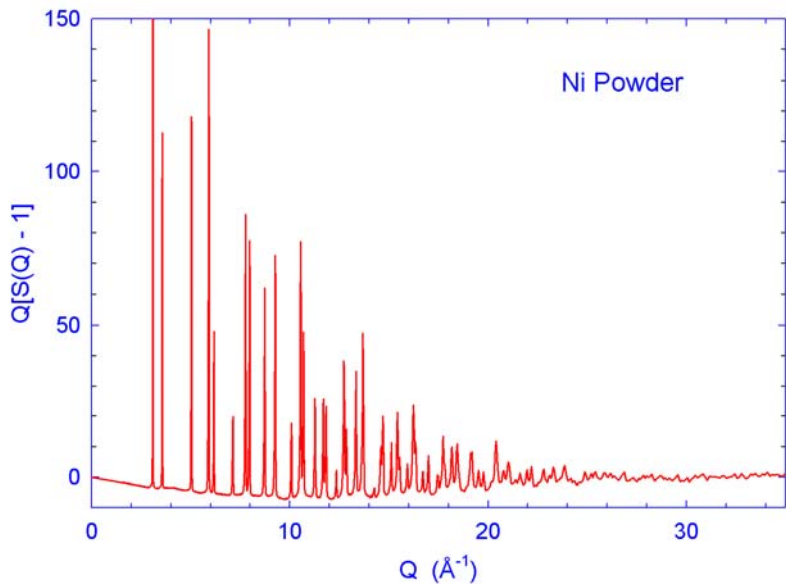
# Nanostructured Bulk Materials

- Possible Research Areas:
  - Nanophase transformations in bulk.
  - Nanoparticles.
  - Structure in fluids, solutions and glasses.

## Directly Involved SNS/HFIR Instruments

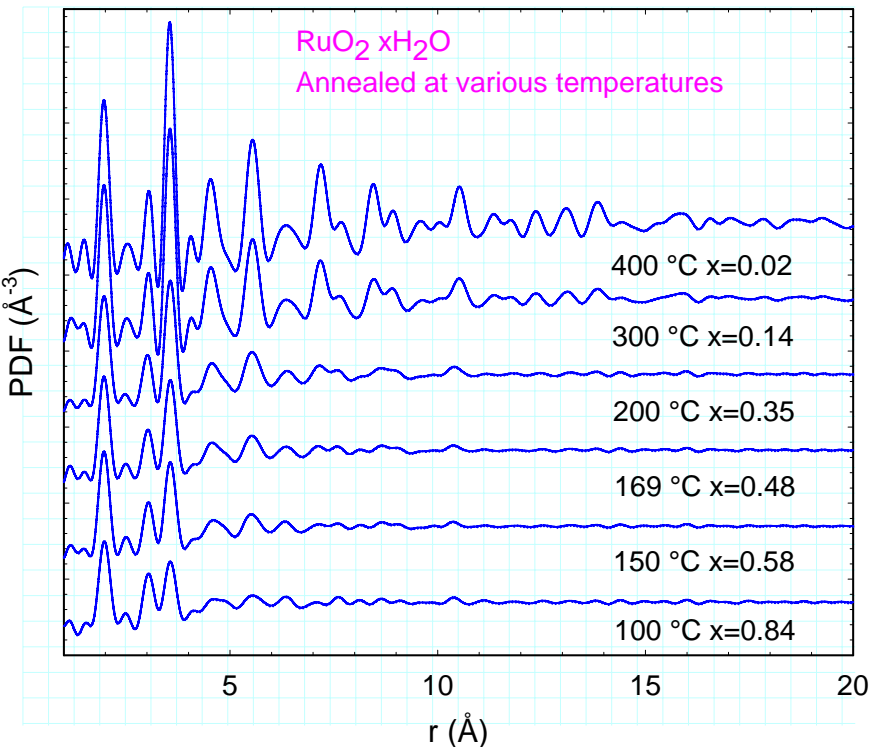
- NOMAD (high intensity disordered materials diffractometer; PDF)
- ARCS (high intensity wide angle chopper spectrometer; dynamic PDF)
- VULCAN/NRSF2 (in-situ strain measurement)
- POWGEN-3 (powder diffraction)
- SNAP (high pressure diffraction)

# Local Structure by Atomic Pair-Density Function (PDF)

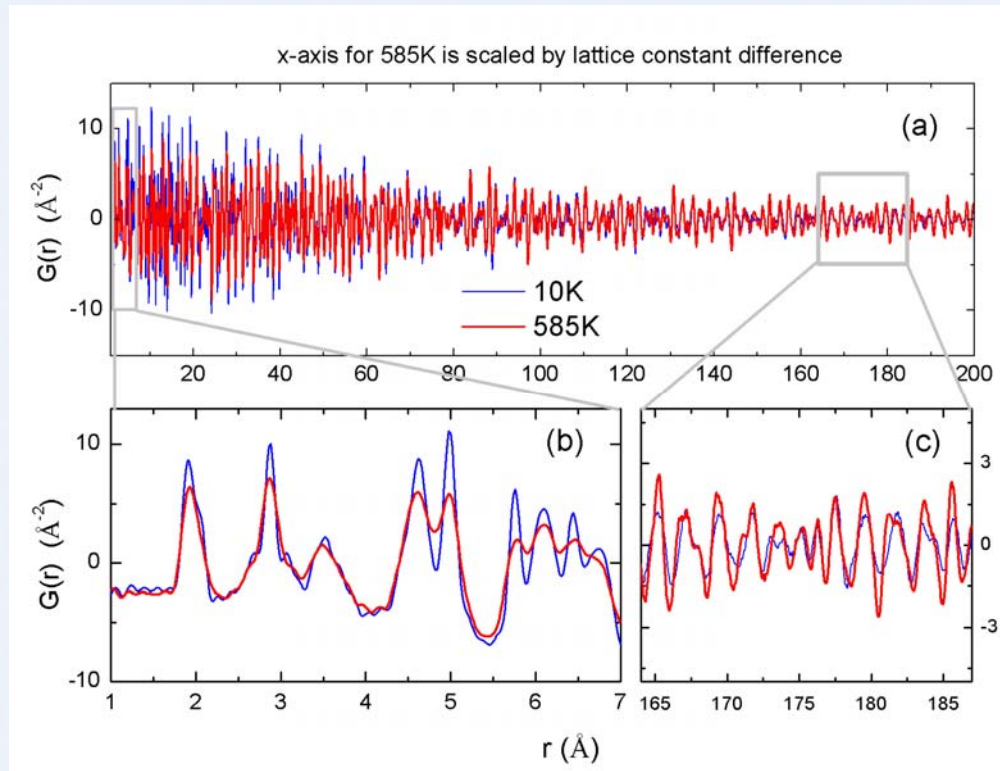


- **Distribution of distances between atoms, can describe local structural deviations.**
- **“Underneath the Bragg Peaks”, T. Egami and S. J. L. Billinge (Pergamon Press, Oxford, 2003).**

# Local Atomic Structure by PDF

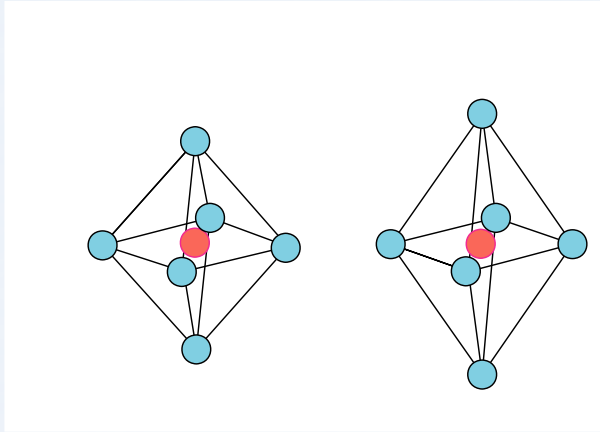


- The pair-density function (PDF) of nano-particle RuO<sub>2</sub>-H<sub>2</sub>O. It is possible to differentiate nano-particles from the amorphous structure.

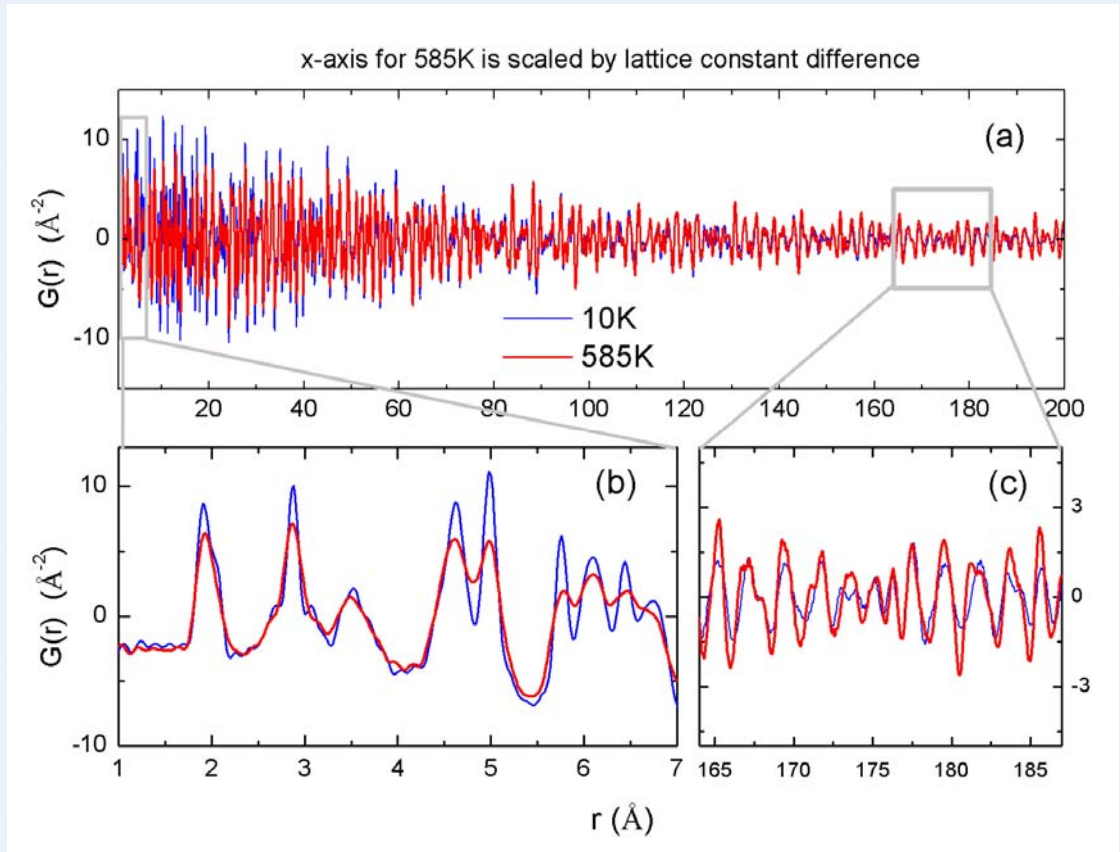


- By using a high-resolution pulsed neutron diffractometer it is now possible to determine the PDF up to 200 Å or more.

# Frustrated Orbital Ordering in $\text{LiNiO}_2$



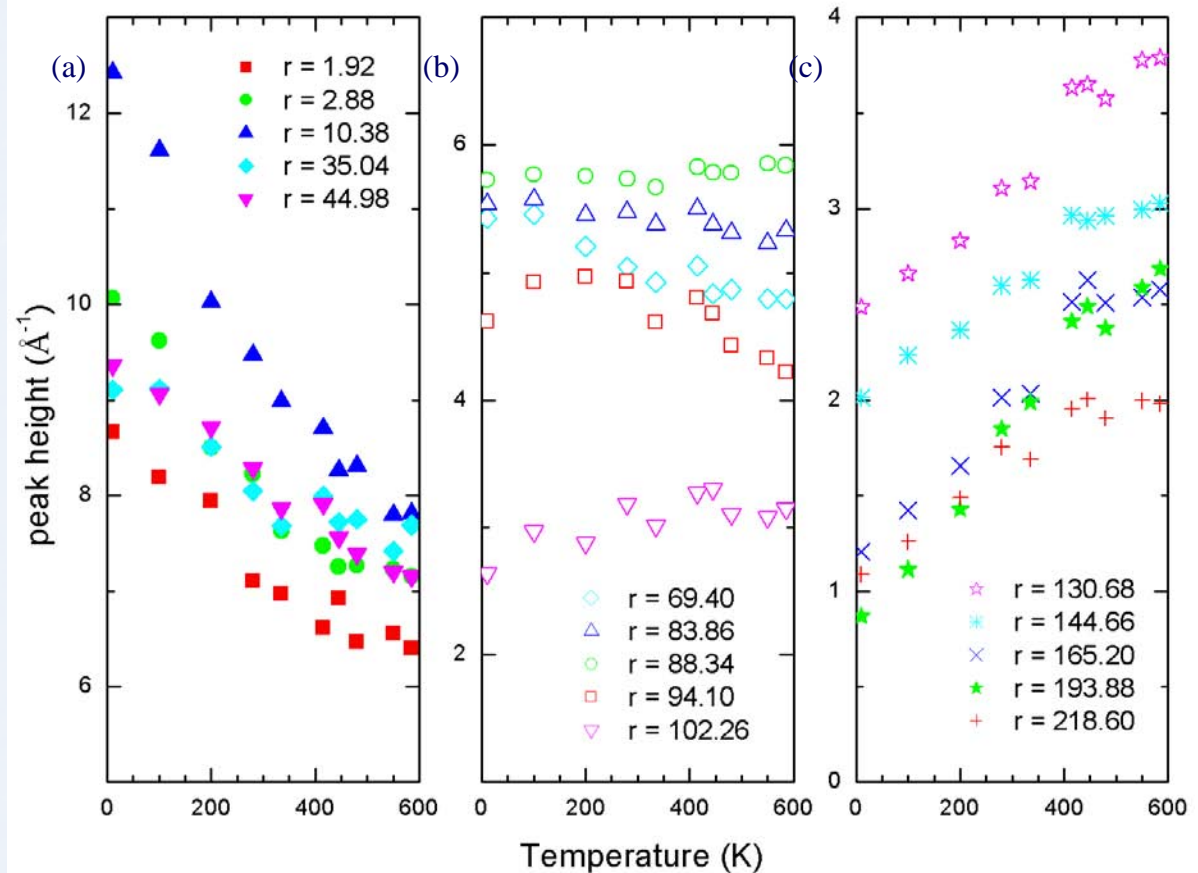
PDF Obtained with the NPDF,  
LANSCE



- The first peak of the PDF (Ni-O peak) is consistent with the  $z^2$ -type JT orbital state with 4 short, 2 long bonds.
- But there is no long-range JT distortion.

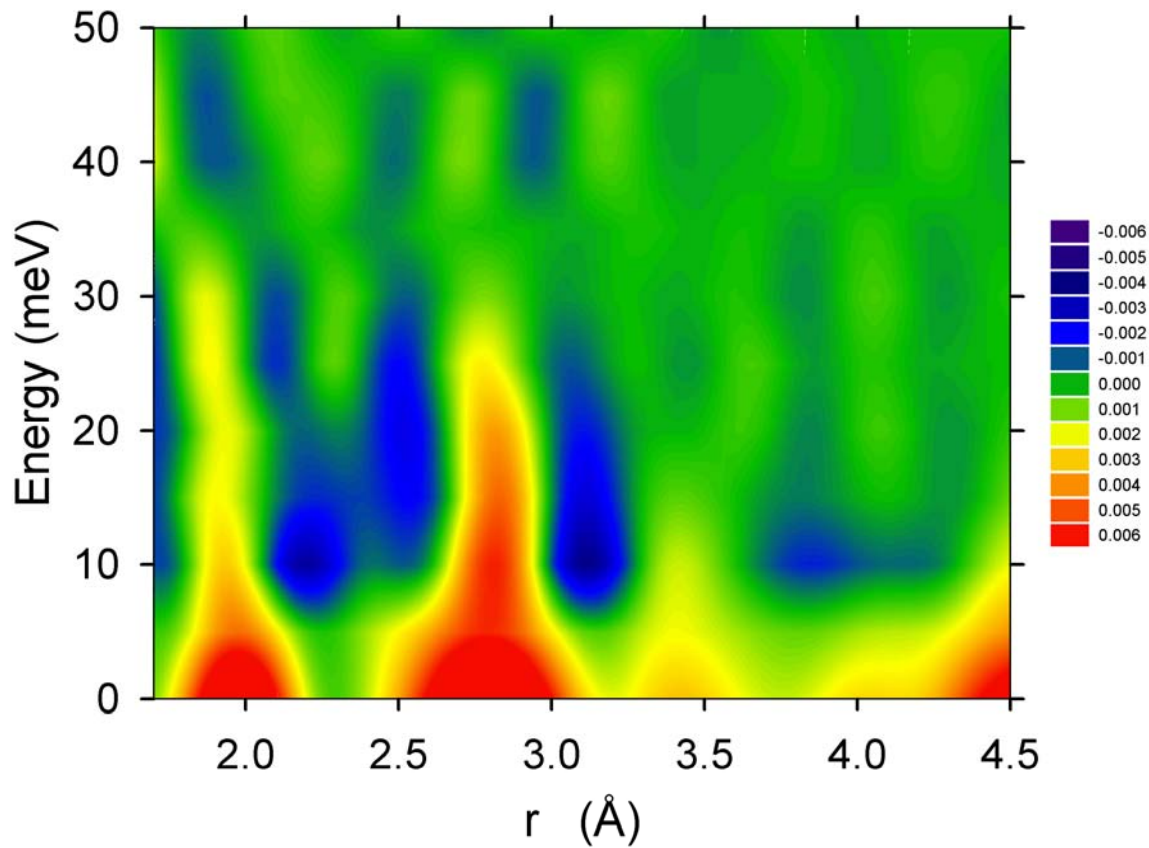
# Temperature Dependence

- For  $r < 60 \text{ \AA}$  peak height decreases with  $T$ .
- For  $r > 100 \text{ \AA}$  peak height increases with  $T$ .
- Local ordering vs. domain formation.



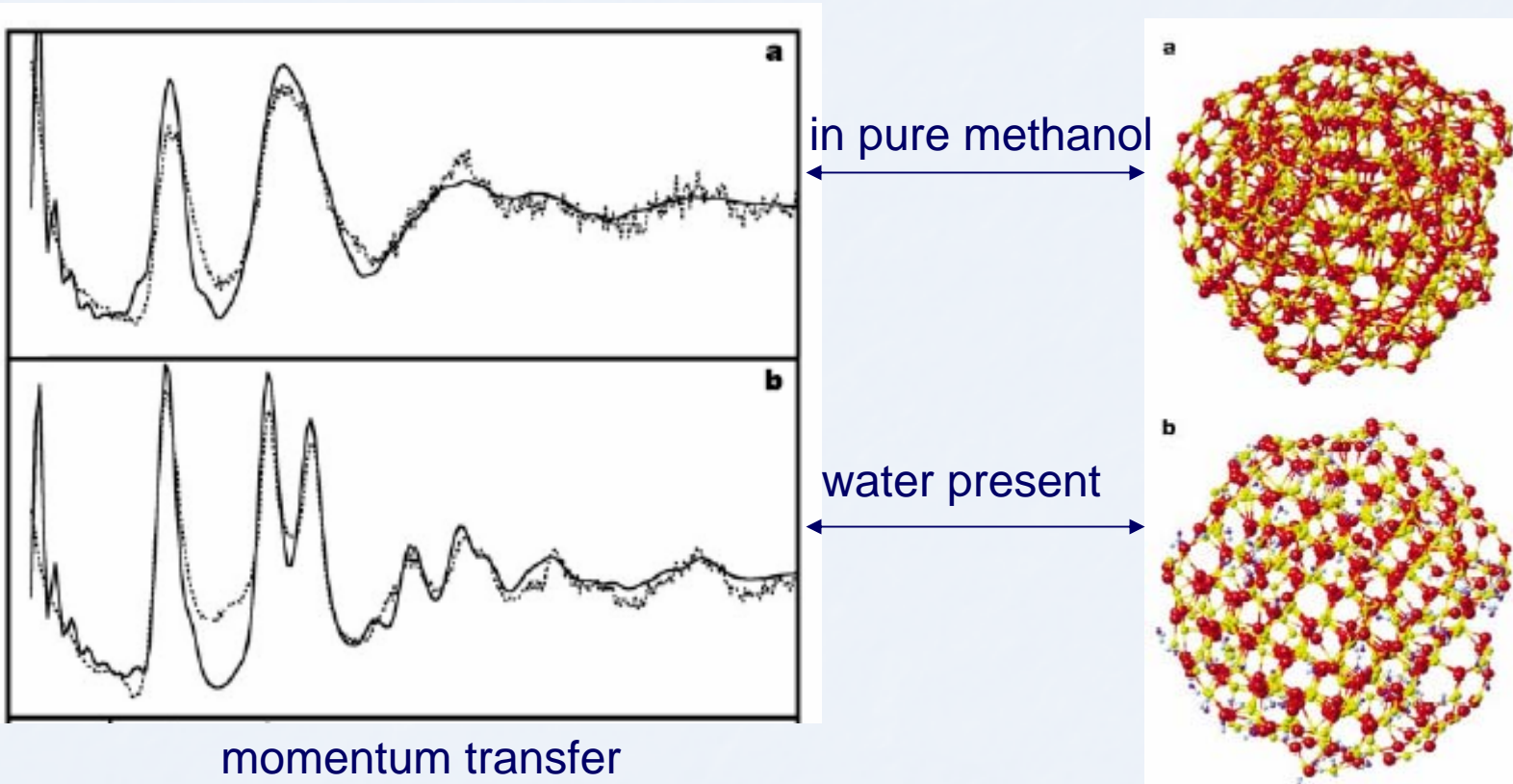


# Dynamic PDF of lead magnesium niobate (PMN) at $T = 590$ K



# Solvent-driven structure transformation in nanoparticles

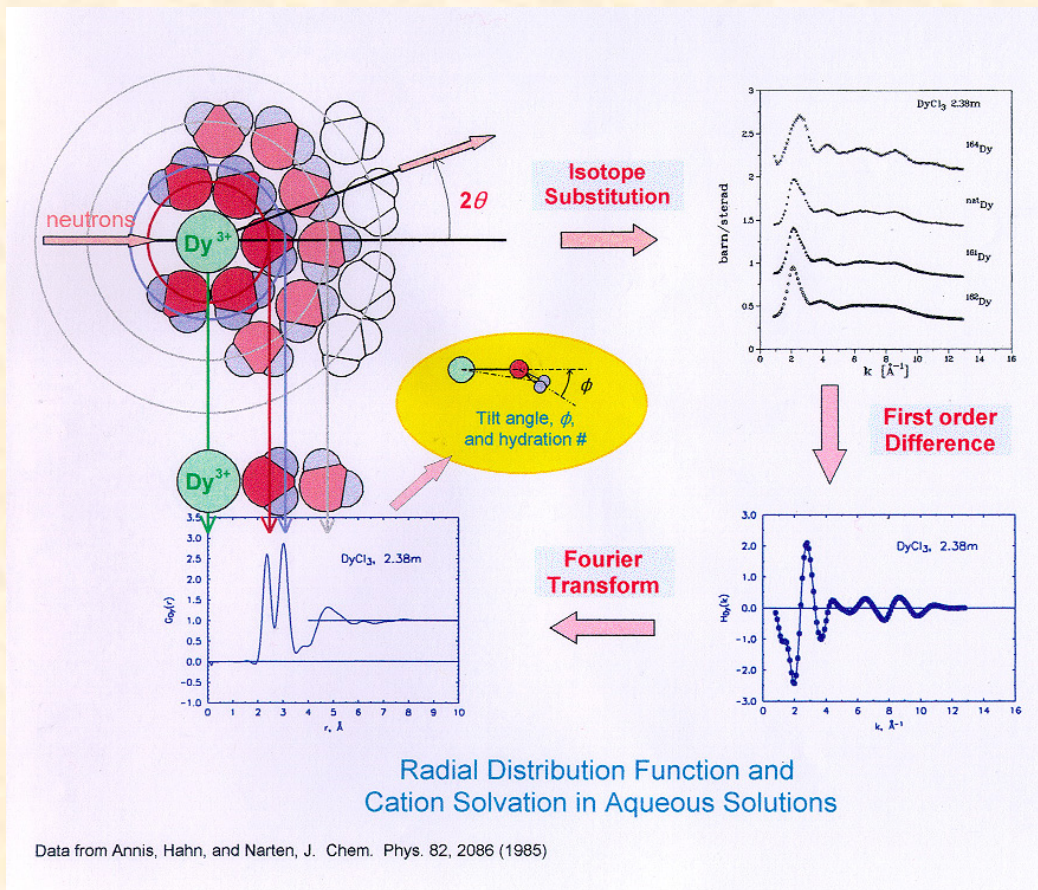
structure function



H. Zhang, B. Gilbert, F. Huang and J. Banfield:  
Nature **424** (2003) 1025

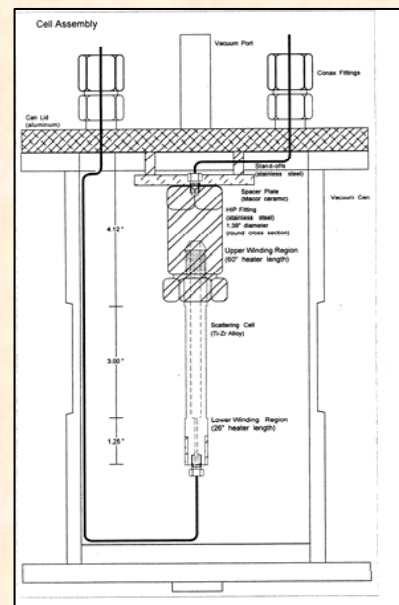


# Neutron Diffraction with Isotopic Substitution (NDIS) for Determining Hydration/Complexation Structure

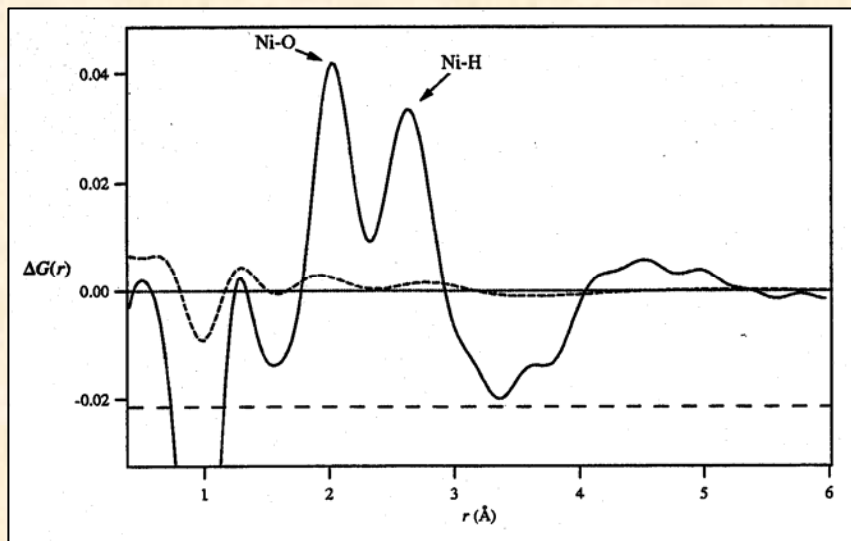


Match T,p,m for two solutions  
 $D_2O$  to minimize corrections  
 Differ in metal isotope  
 Special sample environment needed (null scattering)

Difference gives local environment  
 High stability needed  
 S/N dependent on system



# Structural results from neutron scattering from $\text{NiCl}_2$



Results at 298K for  $\text{NiCl}_2$  at 3.87m.

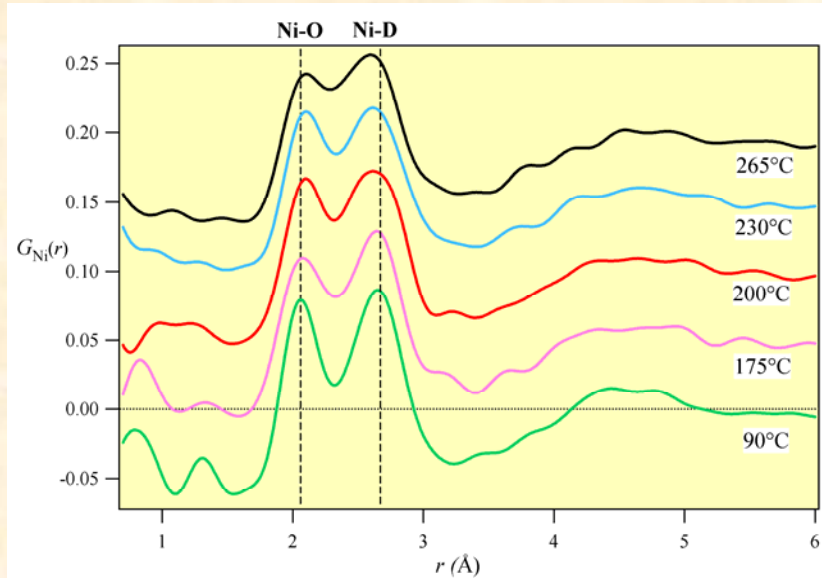
$N \approx 6$ ,  $r_{\text{Ni-O}} \approx 0.205 \text{ nm}$ ,  $r_{\text{Ni-H}} \approx 0.266 \text{ nm}$

From Badyal et al., *J. Neutron Res.*, 2002.

Results at higher temperatures for  $\text{NiCl}_2$

Apparent broadening and shifting in hydration peaks; no discernable association.

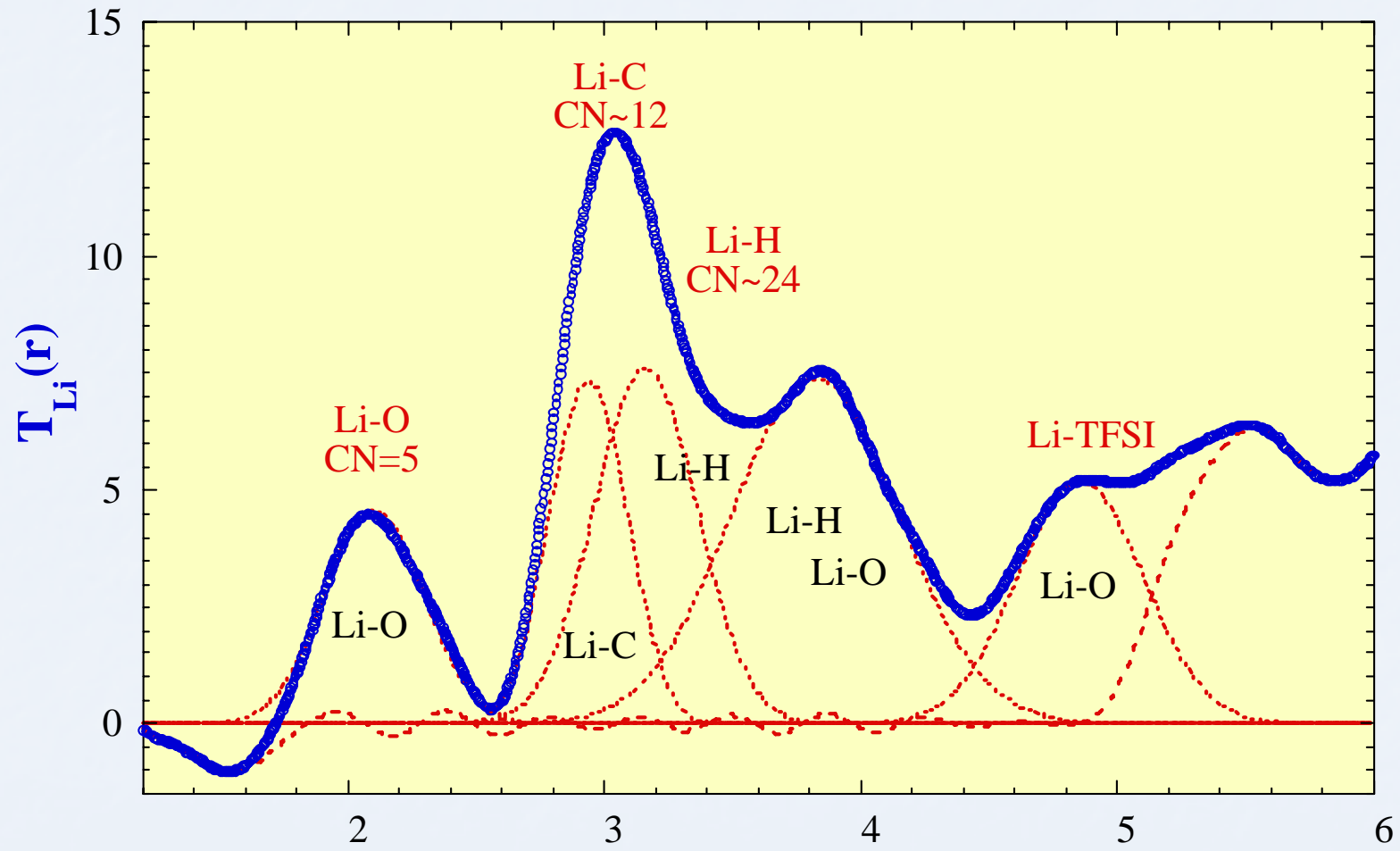
From Badyal and Simonson, *JCP*, 2003



# Polymer electrolytes

- A leading candidate for vehicle propulsion and secondary batteries in consumer market in coming decade
- Two relaxation processes [Angell 1992]:  
mechanical relaxation ( $\tau_s$ ) and electrical relaxation ( $\tau_\sigma$ )  
Decoupling index  $R^* = \tau_s(T_g) / \tau_\sigma(T_g)$
- **Optimizing performance involves max.  $R^*$ , min.  $T_g$**
- Maximizing  $R^*$  brings in question of configuration of migrating ions relative to polymer matrix
- Structure: Neutron Diffraction with Isotope Substitution (NDIS):  
H:D,  $^6\text{Li}$ :  $^7\text{Li}$ ,  $^{35}\text{Cl}$ :  $^{37}\text{Cl}$
- Dynamics: quasielastic neutron scattering (QENS)  
neutron spin-echo (NSE)

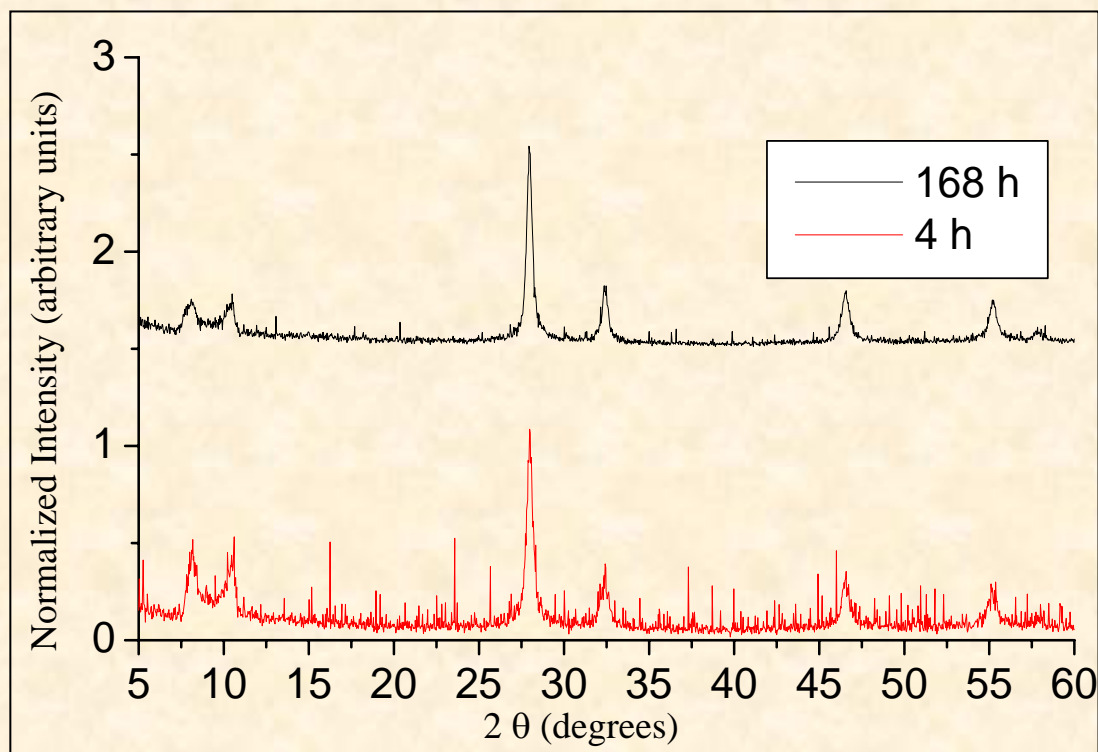
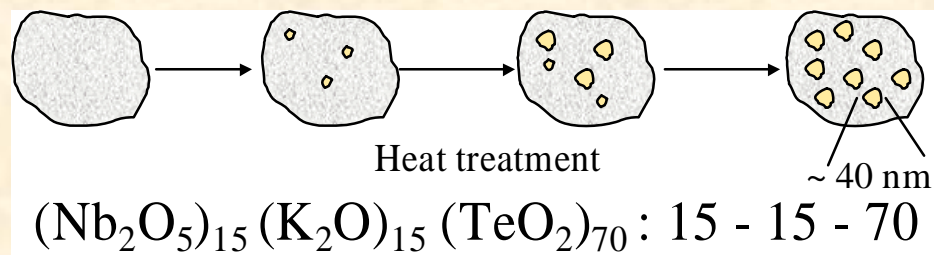
# Li difference pair correlation function of $\text{P(EO)}_{7.5}\text{LiTFSI}$





# Application to glass ceramics

Optically transparent glass ceramics can form on heat treatment of quenched glasses.



$\sim 250\%$  increase in S/N ratio for 168h heat treatment compared to 4h heat treatment.

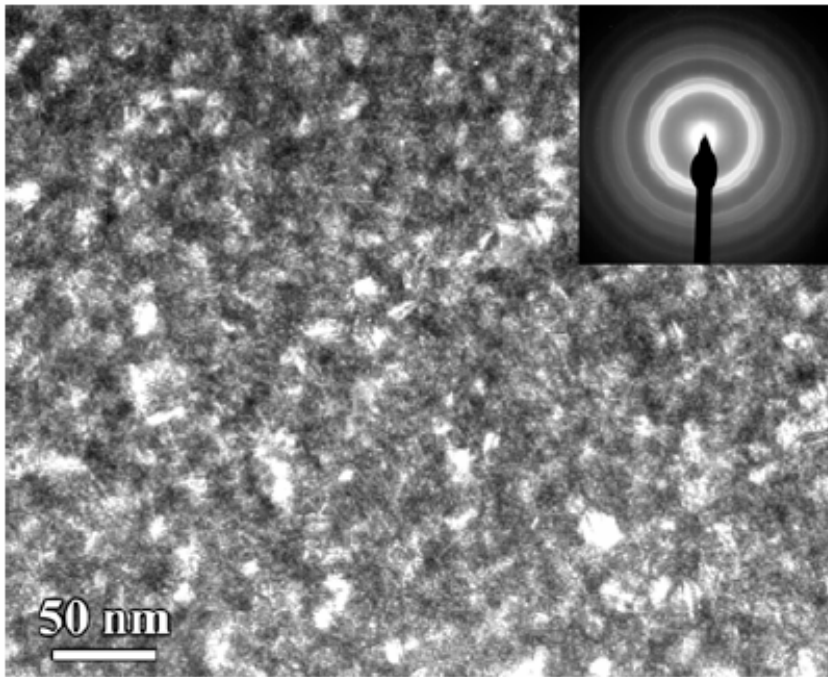
Limiting crystallite size  $\sim 80\text{nm}$ .

Growth faster than nucleation.

Data needed to evaluate models of kinetic behavior.

Courtesy of Prof. Josef Zwanziger,  
Dalhousie University

# Partially Crystallized Bulk Metallic Glass Contains High-density Nanometer Sized Crystallites

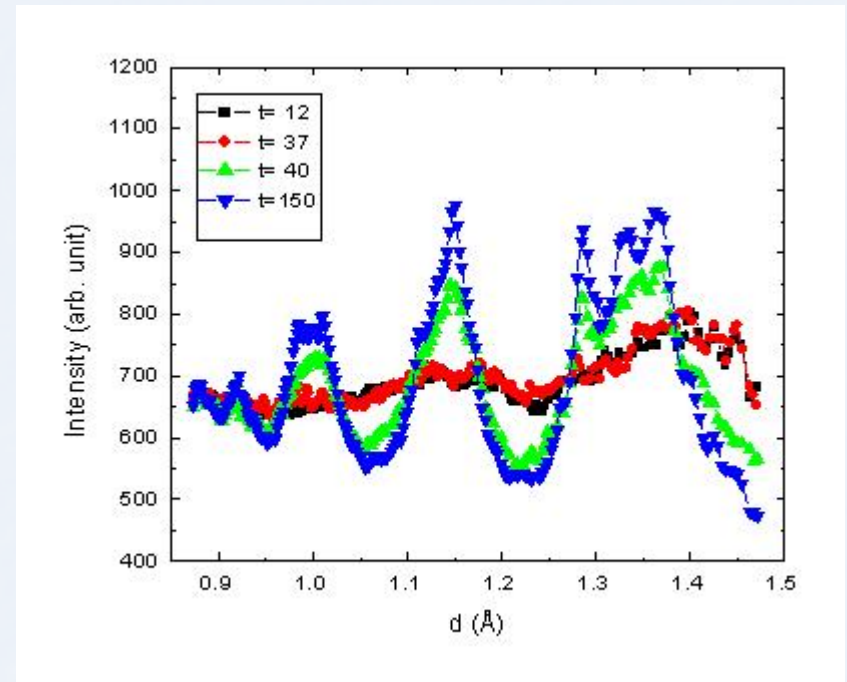
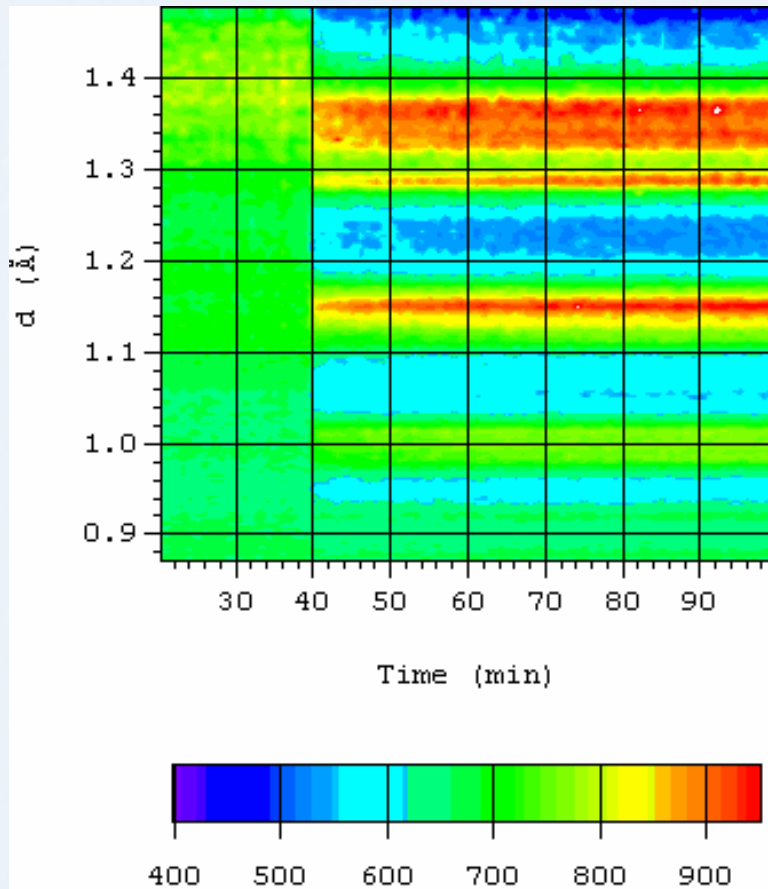


*Figure 4*  
Dark-field transmission electron microscopy image with diffraction pattern (inset) of Vit105, annealed for 15 h at 673 K (Pekarskaya *et al.* 2003).

- e.g., upon isothermal annealing
- Density  $10^{23}$ - $10^{24}$  m<sup>-3</sup>
- Crystallite size ~10 nm
- Even for temperatures close to  $T_g$



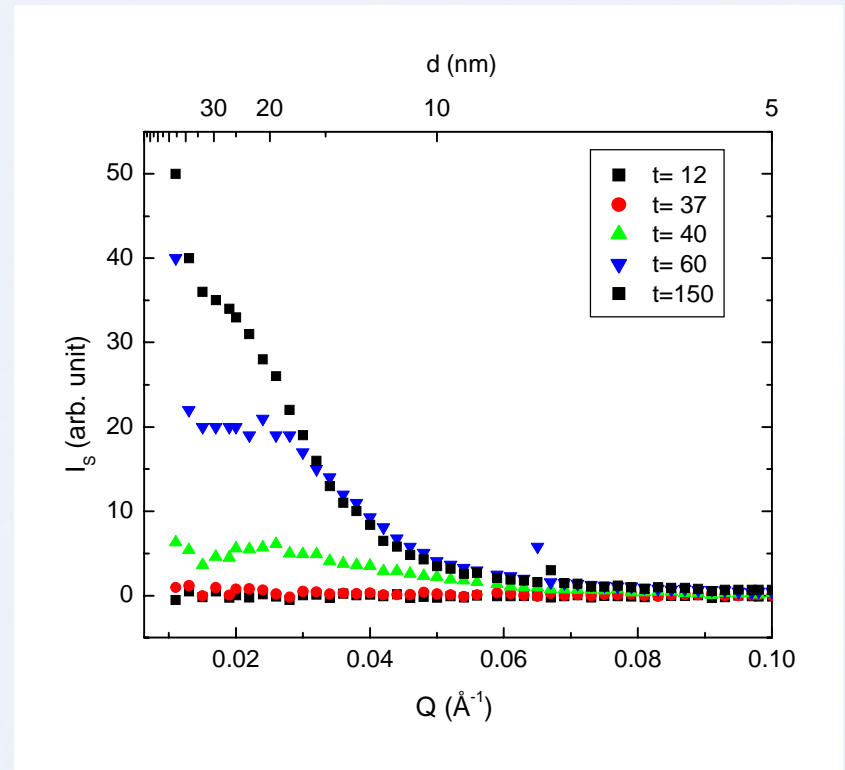
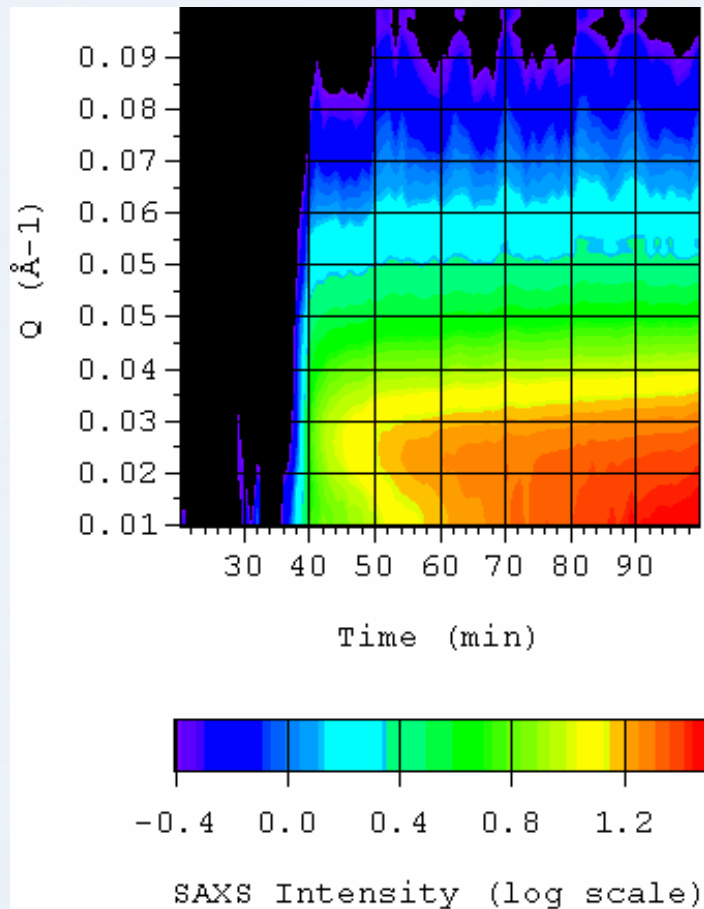
# An Abrupt Amorphous-to-Crystalline Transformation Was Observed at t=40 min



**Crystalline phase is primarily  $Zr_2Ni$**

**Our Sample Is BAM-11,  
 $Zr_{52.5}Cu_{17.9}Ni_{14.6}Al_{10}Ti_5$**

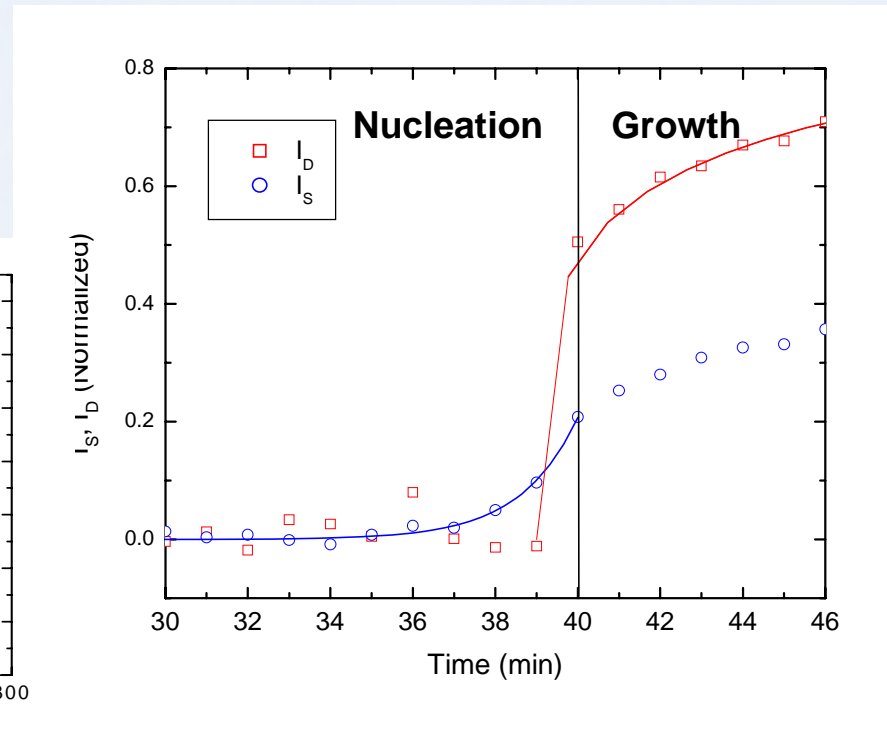
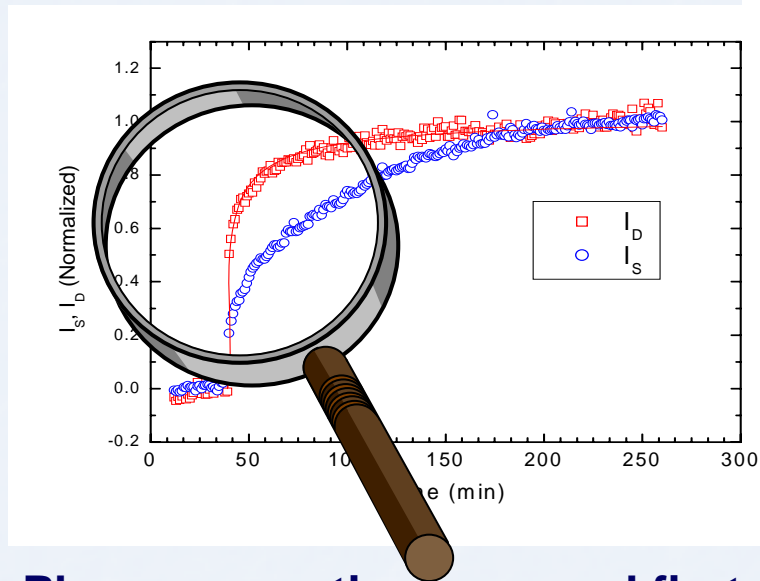
# Small Angle Scattering Data Show Phase Separation Prior to Crystallization



**Small angle scattering profile exhibits characteristic interference peak**

# Combined Data Show Phase Separation Before Crystallization

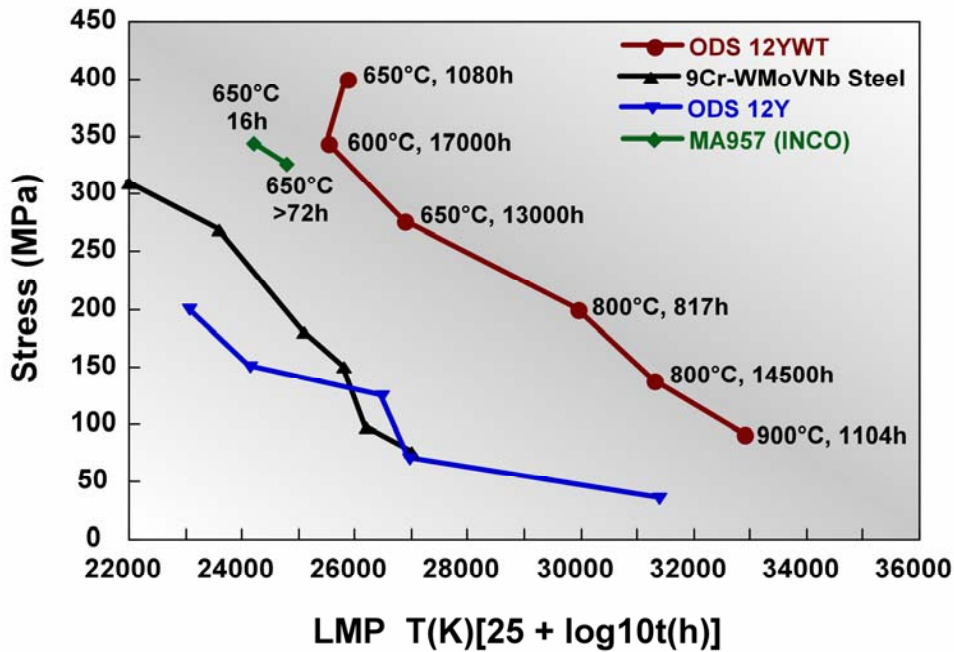
$$I_D = \sum |Y(t, d) - Y(t_0, d)|$$
$$I_S = \sum$$



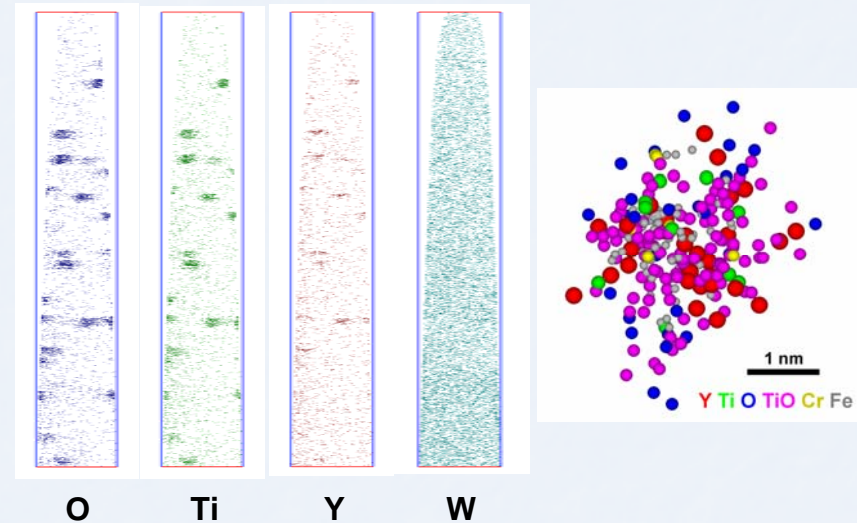
- Phase separation occurred first, setting stage for crystallization
- Rapid rise of small angle scattering intensity for  $t < 40$  min
- Slow growth of diffraction intensity for  $t > 40$  min.

Wang et al., PRL 2003

# Oxide-Dispersion-Strengthened Steel Containing Y-Ti-O Nanoclusters Exhibits Outstanding Mechanical Properties at High Temperature



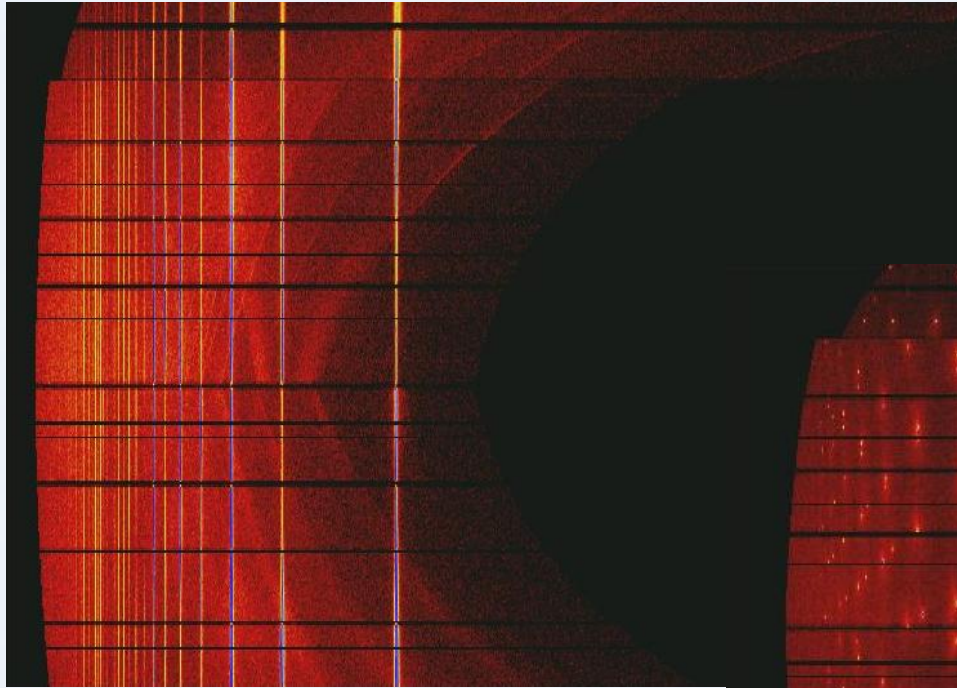
Atom probe showed that Y-Ti-O form nanoscale clusters that are highly stable even at high temperatures



Creep rate is reduced by six orders of magnitude in ODS steel containing Y-Ti-O. This degree of improvement represents a major breakthrough in the use of ferritic alloys for high temperature structural applications, which could tremendously benefit energy-generation systems

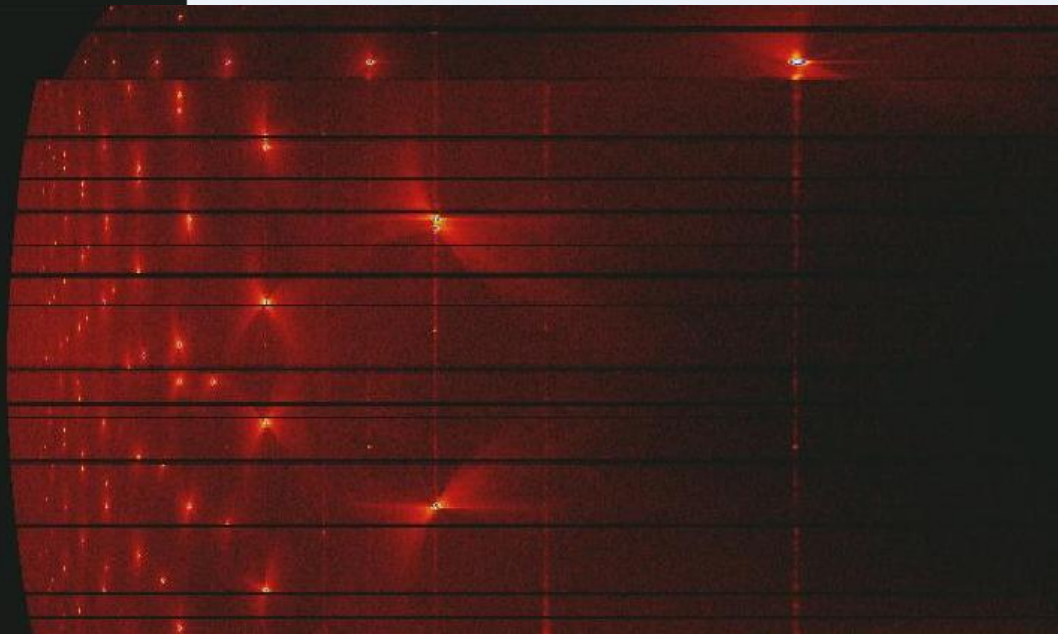
# In-situ Neutron Diffraction Studies to 1300°C Demonstrated the Thermal Stability of Mechanically-Alloyed ODS Steel

Room temperature diffraction patterns after heating to 1300°C



Mechanically alloyed ODS steel are stable after heating to 1300°C

Cast ODS steel recrystallized and grew into single crystals



Nanoscale control of materials structure via phase transformation



# CNMS Contribution to “Bulk Nanoscience”

## Sample preparation and characterization

- Nanostructured and nanoparticle materials

- Processing/treatment approaches

- Specialized sample labeling

  - H/D substitution in hydrogenous systems

  - Isotopes beyond H/D for enhanced contrast

- Enhanced access to additional ORNL capabilities

## Cooperation and collaboration

- Expertise in synthesis and characterization

- Nanomaterials Theory Institute