



Environmental Contaminants and Climate Change in Alaska

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Abstract: Climate change is a global pollution issue, caused by anthropogenic greenhouse gases and aerosols, primarily CO₂, CH₄, carbon particles (“black” carbon) and N₂O. It will alter contaminant effects on Service-managed resources in Alaska by: 1) Influencing contaminant fate and transport, as chemicals more readily partition to air under warmer temperatures, which may increase the amount of atmospheric transport to the Arctic via poleward movement of warm air, and projected precipitation increases will scour these contaminants from the atmosphere, depositing them within watersheds; 2) Releasing contaminants from abiotic media through enhanced coastal and river erosion, more frequent and intense wildfires (releasing mercury), glacial melt surrendering pollutants deposited decades ago, and melting permafrost liberating toxins sequestered in ice; 3) Increased natural resource development and new shipping routes from projected sea ice losses and warmer Arctic temperatures, raising the risk of oil or hazardous materials spills in Alaska’s sensitive habitats; 4) Altering metabolism, uptake, cycling, and toxicity of pollutants such as mercury, which generally increase with temperature and heat stress. Collectively, these findings suggest that projected temperature increases will enhance contaminant influx to and within Alaskan ecosystems, particularly aquatic systems, and by stressing organisms increase their susceptibility to contaminant impacts. The Environmental Contaminants Program will support the Service’s collective efforts to address this daunting challenge by continuing to implement our core mission of evaluating and remediating contaminants exposure and effects on trust resources, and by working with others on oil spill preparation and response, eroding landfills, contaminated sites, and contaminants in subsistence foods.



Oiled birds from the Selendang Ayu spill, 2004

spills in broken ice conditions can’t be efficiently cleaned up using traditional methods like skimming (e.g., SL Ross et al. 1998). Reduced marine ice cover will **increase ocean-atmosphere gas exchange** and therefore deposition of contaminants like **PCBs and toxaphene**, which are still loading into the Arctic Ocean via the atmosphere (AMAP 2003).

References are listed on back of handouts.

Coastal and riverine erosion, surface water encroachment, and loss of permafrost may cause release of pollutants from **contaminated sites** like landfills, drilling mud pits, sewage lagoons, tailing ponds, and oil-contaminated sites (AMAP 2003). In particular, **shoreline erosion** has and will continue to impact **fuel delivery, storage, and pipeline systems** in Alaska. Dendritic drainage patterns that will allow more efficient transport of contaminants into ponds and lakes and possibly re-mobilize contaminants entrained in tundra soils (MacDonald et al. 2005).



Eroding landfill, Beaufort Sea shore, Arctic National Wildlife Refuge

Modeling predicts **increases in mean annual river discharge** of about 20% for the Yenisei, Lena, Mackenzie Rivers (MacDonald et al. 2005) and other large Arctic rivers. Contaminant loads are correlated and will also increase, including **mercury** (studied in the MacKenzie River; Leitch et al. 2007) and other contaminants (Carroll et al. 2008, Zhulidov et al. 2000).



Lower Marsh Fork River, Arctic National Wildlife Refuge

Melting glaciers (Hinzman et al. 2005) can release contaminants accumulated during years of deposition (Blais et al. 1998). **Glacial streams contain unusually high proportions of persistent organic pollutants in solution**, because of the low organic matter content and therefore low potential adsorption to suspended sediments. Contaminants in solution are much more **bioavailable** than adsorbed contaminants (Blais et al. 2001).

Mercury is toxic and increasing in Arctic biota. Continuing use of coal and other fossil fuels, especially in Asia (Pacyna et al. 2006), physical factors such as ice cover, permafrost melting and organic carbon cycling, and abiotic factors such as increasing primary productivity may be driving increased mercury in Arctic biota (MacDonald et al. 2005), including in fish used for subsistence (Carrie et al. 2010) and in freshwater ecosystems (Chetelat and Amyot 2009). Mercury becomes toxic and bioavailable through **methylation** in wetlands (Lindberg et al. 2002, Loseto et al. 2004, MacDonald et al. 2005) – something that will increase with more **flooding** (ACIA 2005). Simple erosion can increase mercury concentrations in waters, such as those draining from peat bogs (Jörnhagena et al. 2007). **Mercury** is also released from peat soils during **fires in boreal and tundra ecosystems**, which modeling suggests will increase (ACIA 2005, Turetsky et al. 2006), along with mercury concentrations in fish after fires (Kelly et al. 2006).

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