



**Fermi National Accelerator Laboratory**

P.O. Box 500, Batavia, Illinois 60510

FERMILAB 86/37

1104100

UC-41

# **Site Environmental Report**

## **for Calendar Year 1985**

**May 1, 1986**

**Samuel I. Baker**



Operated by Universities Research Association  
Under Contract with the United States Department of Energy

Fermi National Accelerator Laboratory  
P. O. Box 500, Batavia, Illinois 60510

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SITE ENVIRONMENTAL REPORT  
For Calendar Year 1985

by  
Samuel I. Baker  
May 1, 1986

Laboratory Work  
by  
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TABLE OF CONTENTS

<u>Section</u>	<u>TEXT</u>	<u>Page</u>
1.	Introduction.....	1
2.	Summary.....	9
3.	Environmental Program Information.....	14
3.1	Summary of Environmental Monitoring Performed in CY-1985..	14
3.2	Environmental Program Description.....	16
3.3	Environmental Radiation Monitoring.....	19
3.3.1	Penetrating Radiation.....	20
3.3.1.1	Muons.....	21
3.3.1.2	Neutrons.....	23
3.3.1.3	Gamma Rays.....	23
3.3.2	Airborne Radioactivity.....	24
3.3.3	Waterborne Radioactivity.....	26
3.3.3.1	Water Sample Collection.....	27
3.3.3.2	Results of Analyses.....	31
3.3.3.2.1	Tritium.....	32
3.3.3.2.2	Beryllium.....	36
3.3.3.2.3	Other Radionuclides.....	36
3.3.3.3	Sediment and Vegetation Sampling.....	37
3.3.3.4	Soil Activation.....	39
3.4	Environmental Monitoring for Non-radioactive Pollutants...	42
3.4.1	Domestic Water Supplies.....	42
3.4.2	Industrial Water Ponding Systems.....	48

TABLE OF CONTENTS

<u>Section</u>		<u>Page</u>
	<u>TEXT</u>	
3.4.3	Other Lakes and Ponds.....	49
3.4.4	Sewage Treatment.....	52
3.4.5	Chemical Treatment of Water Systems.....	53
3.4.5.1	Dalapon.....	53
3.4.5.2	Chlorine.....	53
3.4.5.3	Aquazine.....	54
3.4.5.4	Heavy Metals and Other Toxic Materials.....	54
3.5	Environmental Permits.....	56
3.6	Assessments and Impact Statements.....	57
3.7	Summary of Significant Environmental Activities.....	58
3.8	Summary of Hydrogeology.....	59
3.9	Evaluation of Environmental Impacts.....	61
3.9.1	Assessments of Potential Radiation Dose to the Public.....	61
3.9.2	Assessment of Non-radioactive Pollutant Releases.....	66
3.9.3	Potential Impact of Other Toxic Substances.....	67
3.9.3.1	Pesticides.....	67
3.9.3.2	Polychlorinated Biphenyls.....	69
3.9.3.3	Hazardous Wastes.....	69
3.9.3.4	Heavy Metals.....	71
3.9.3.5	Chlorides.....	71
4.	Quality Assurance in CY-1985.....	72
4.1	Quality Control.....	72

TABLE OF CONTENTS

<u>Section</u>		<u>Page</u>
	<u>TEXT</u>	
4.1.1	Analytical Procedures at Teledyne.....	72
4.1.2	Quality Assurance Samples.....	75
4.2	Additional Quality Assurance Efforts.....	76
5.	References.....	82
6.	Acknowledgements.....	88
7.	Distribution List.....	89

TABLE OF CONTENTS

	<u>Page</u>
<u>TABLES</u>	
Table 1	Summary of Radioactivity Released to the Off-Site Environment in CY-1985..... 11
Table 2	Tritium Detected in On-Site Water Samples..... 33
Table 3	CY-1985 Sediment Sampling Results..... 37
Table 4	CY-1985 Vegetation Sampling Results..... 38
Table 5	CY-1985 Ground Water Surveillance Wells On-Site Heavy Metals Data..... 44
Table 6	CY-1985 Ground Water Surveillance Wells On-Site Water Quality Data..... 45
Table 7	CY-1985 Ground Water Surveillance Wells On-Site Organics Survey Data..... 46
Table 8	CY-1985 Ground Water Surveillance Wells On-Site Trihalomethane Data for Chlorinated Systems..... 47
Table 9	Site Wide Water Quality Report for CY-1985..... 50
Table 10	Village Sewage Treatment Plant Monthly Averages Report for CY-1985..... 51
Table 11	Incremental Population Data in Vicinity of Fermilab, 1980..... 62
Table 12	Summary of Population Exposures for CY-1985 Within an 80 km (50 mi) Radius of Fermilab..... 65
Table 13	Specifications for the Analyses of Accelerator-Produced Radionuclides in Water..... 73
Table 14	Quality Assurance Results for Fermilab..... 76
Table 15	Quality Assurance Results for Teledyne - First Half of CY-1985..... 77
Table 16	Quality Assurance Results for Teledyne - Second Half of CY-1985..... 78

TABLE OF CONTENTS

	<u>Page</u>
<u>ILLUSTRATIONS</u>	
Figure 1	General Features..... 3
Figure 2	Location of Fermilab and Population Concentrations Within 80 km (50 mi)..... 4
Figure 3	80 km Population Distribution - 1980..... 5
Figure 4	Well Sample Locations..... 7
Figure 5	Fermilab Site..... 13
Figure 6	Surface Water Sample Locations..... 17
Figure 7	Penetrating Radiation (Muon) Directions..... 22
Figure 8	Target Tube Underdrain Discharges..... 34
Figure 9	Site Water Flow Map..... 35
Figure 10	Cross Sectional View at CO Abort Dump..... 40
Figure 11	Ground Water Level Contours..... 60
Figure 12	8 km Population Distribution - 1980..... 63

1. Introduction

This report summarizes the environmental activities and their results at Fermi National Accelerator Laboratory (Fermilab) for Calendar Year (CY-) 1985.

The facility consists of a series of proton accelerators which became operational in 1972, producing higher energy protons than any other accelerator. As a result of accelerator improvements, the original design energy of 200 GeV (billion electron volts) was gradually raised and operation at 400 GeV was routine between 1976 and 1982 using conventional magnets. At that time a ring of superconducting magnets was added which doubled the energy while using less electrical power.

The primary purpose of the installation is fundamental research in high energy physics. At the present time the protons are being extracted from the superconducting synchrotron and directed to fixed targets. Collisions of protons and antiprotons each having 1 TeV (1000 GeV) are planned. The antiproton source was tested in CY-1985 as well as a large detector. Head-on collisions of 800 GeV protons and antiprotons were observed. In addition, cancer patients are being treated using neutrons released by the interactions of 66 MeV protons from the Linac (linear accelerator), the second stage of the series of accelerators.

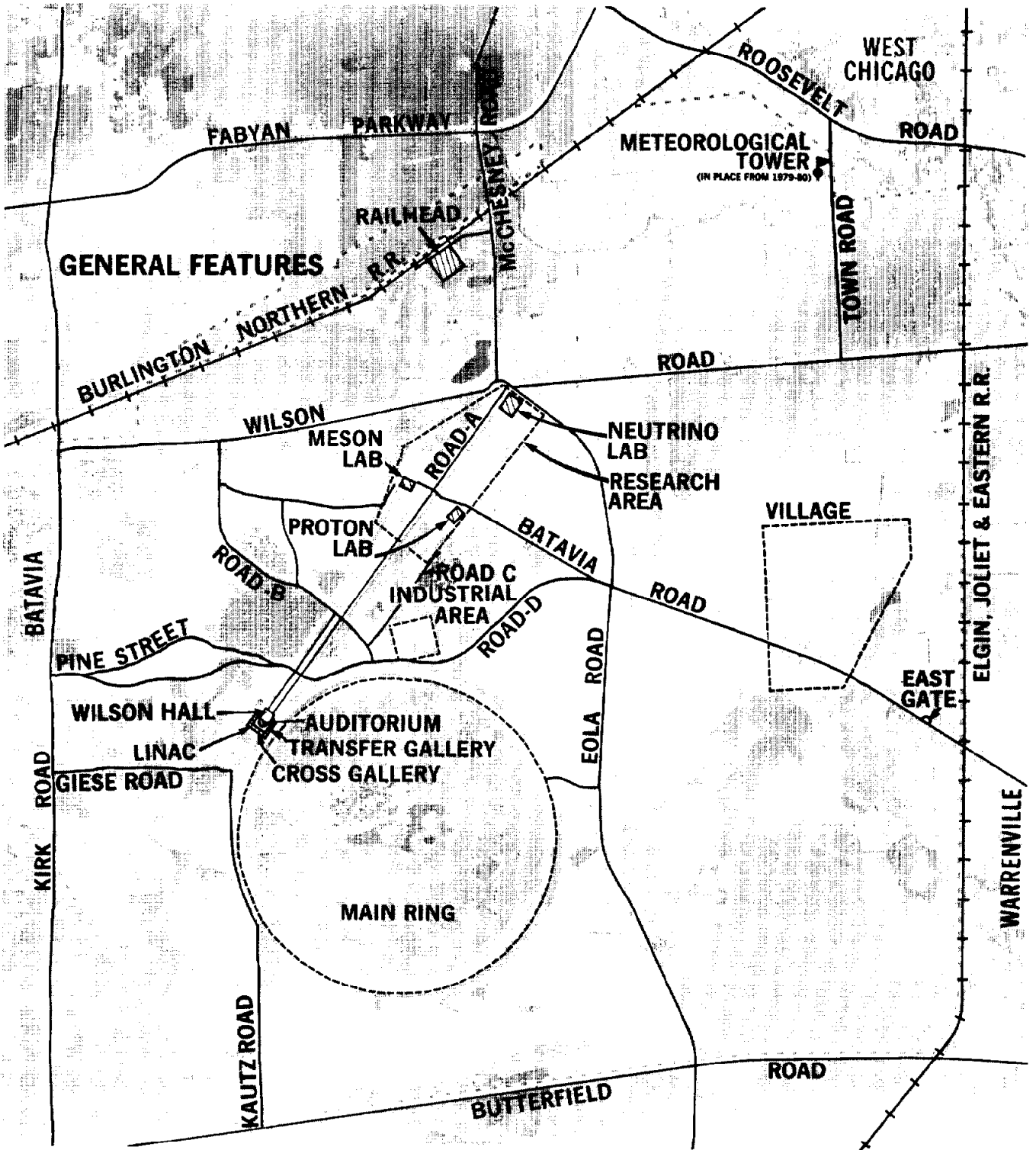


The proton beam extracted from the 2 km (1.2 mi) diameter main accelerator is taken to three different experimental areas on-site. These are the Meson, Neutrino and Proton Labs located in the Research Area (Fig. 1). All three of these areas received proton beams for the first time in 1972. Radioactivity is produced as a result of the interaction of the accelerated protons with matter. Operation of the accelerator produces some airborne radioactivity as well as some radiation which penetrates the shielding material. Also, some radioactivation occurs in the soil and in the water used to cool beam components. A thorough evaluation has been made of the on-site discharges as well as the potential for off-site releases of radioactive and non-radioactive effluents. An extensive monitoring program is being carried out to verify that radiation exposures as well as non-radioactive releases are far below the permissible limits.

Fermilab is located in the greater Chicago area (Fig. 2) on a 27.5 km<sup>2</sup> (10.6 mi<sup>2</sup>) tract of land in an area which is rapidly changing from farming to residential use. There are many municipalities in the vicinity, resulting in a distinct pattern of increasing population concentration eastward toward Chicago (Fig. 3). Within a 3 km (2 mi) distance from the Laboratory boundaries, Batavia (pop. 13,550),<sup>1</sup> Warrenville (pop. 9,402),<sup>2</sup> and West Chicago (pop. 13,109)<sup>2</sup> can be found (Fig. 2). The terrain is relatively flat as a result of past glacial action.

The two major environmental features near the Laboratory are the Fox River to the west, which flows south through Batavia with an average of 1900 million liters (500 million gallons) per day, and the west branch of

Figure 1



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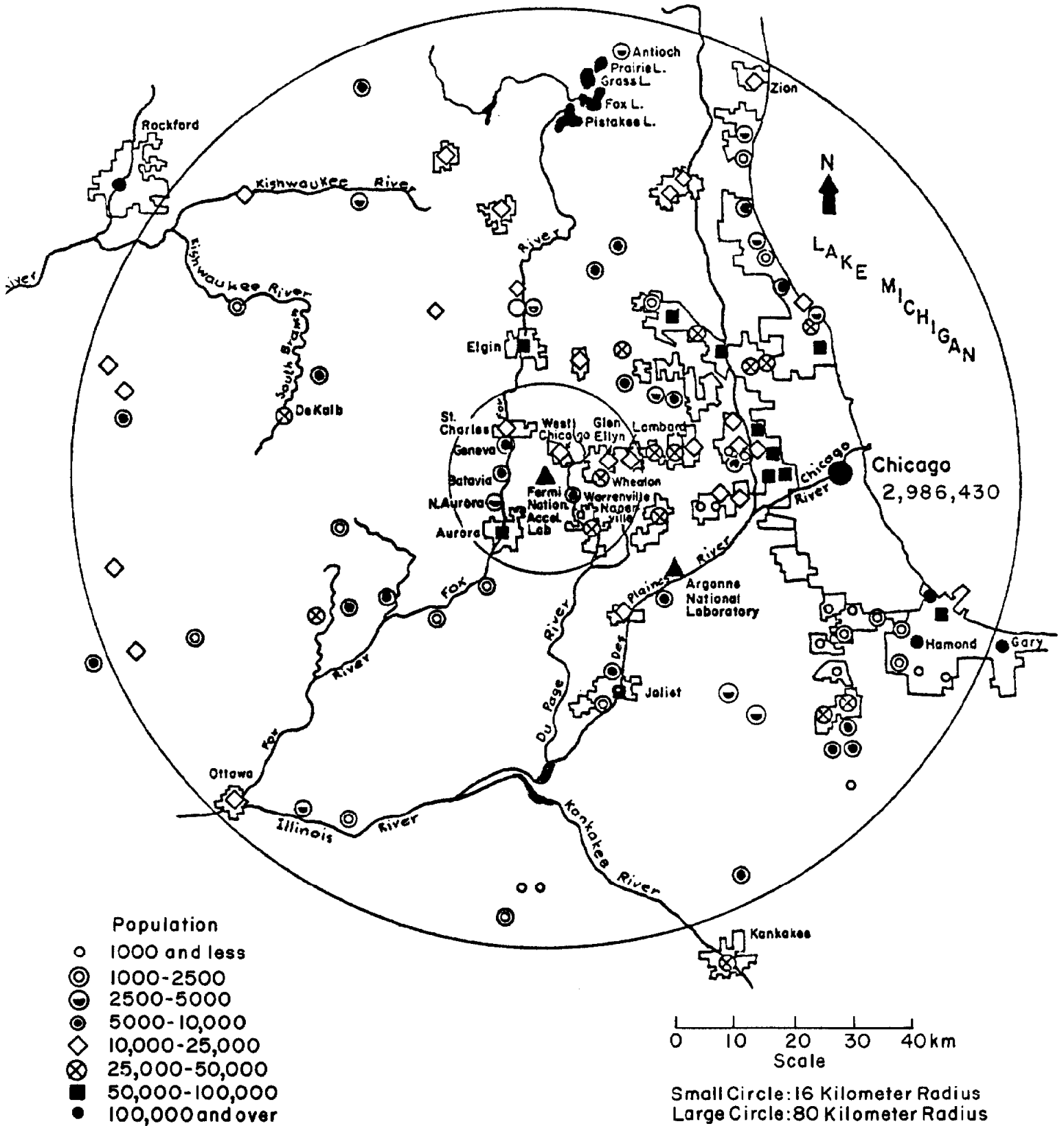
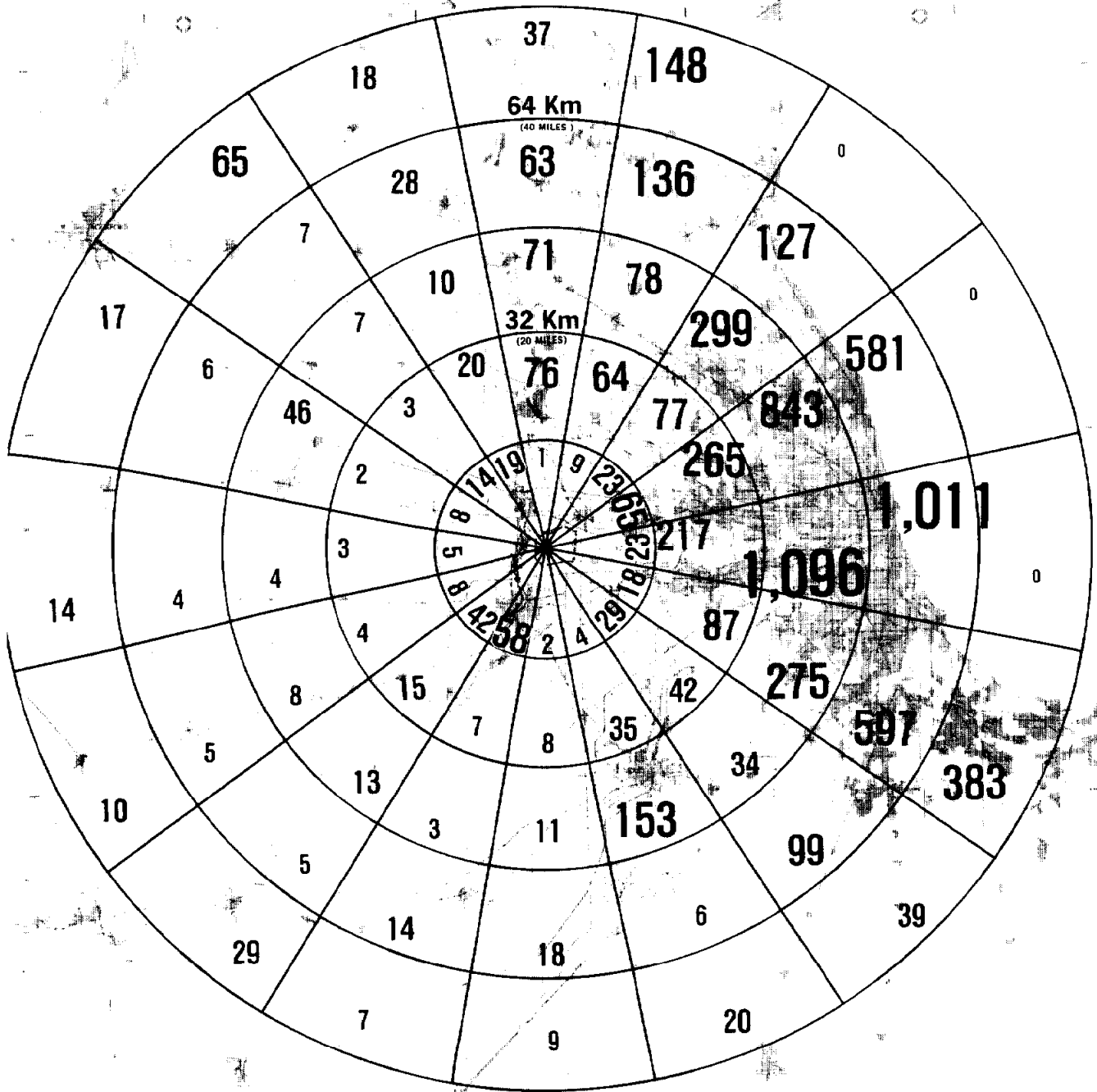


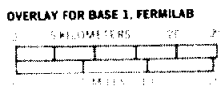
Figure 2 - Location of Fermilab and Population Concentrations within 80 km (50 mi.)

Figure 3



**NOTE: NUMBERS REPRESENT  
NEAREST 1000.**

**POPULATION  
DISTRIBUTION -1980**  
(1980 CENSUS)



REGIONAL AREA SURROUNDING FERMI  
NATIONAL ACCELERATOR LABORATORY

U.S. DEPARTMENT OF ENERGY  
WASHINGTON, D.C. 20545

PREPARED IN 1982  
PREPARED BY  
FOR DOE  
BY E. G. G.

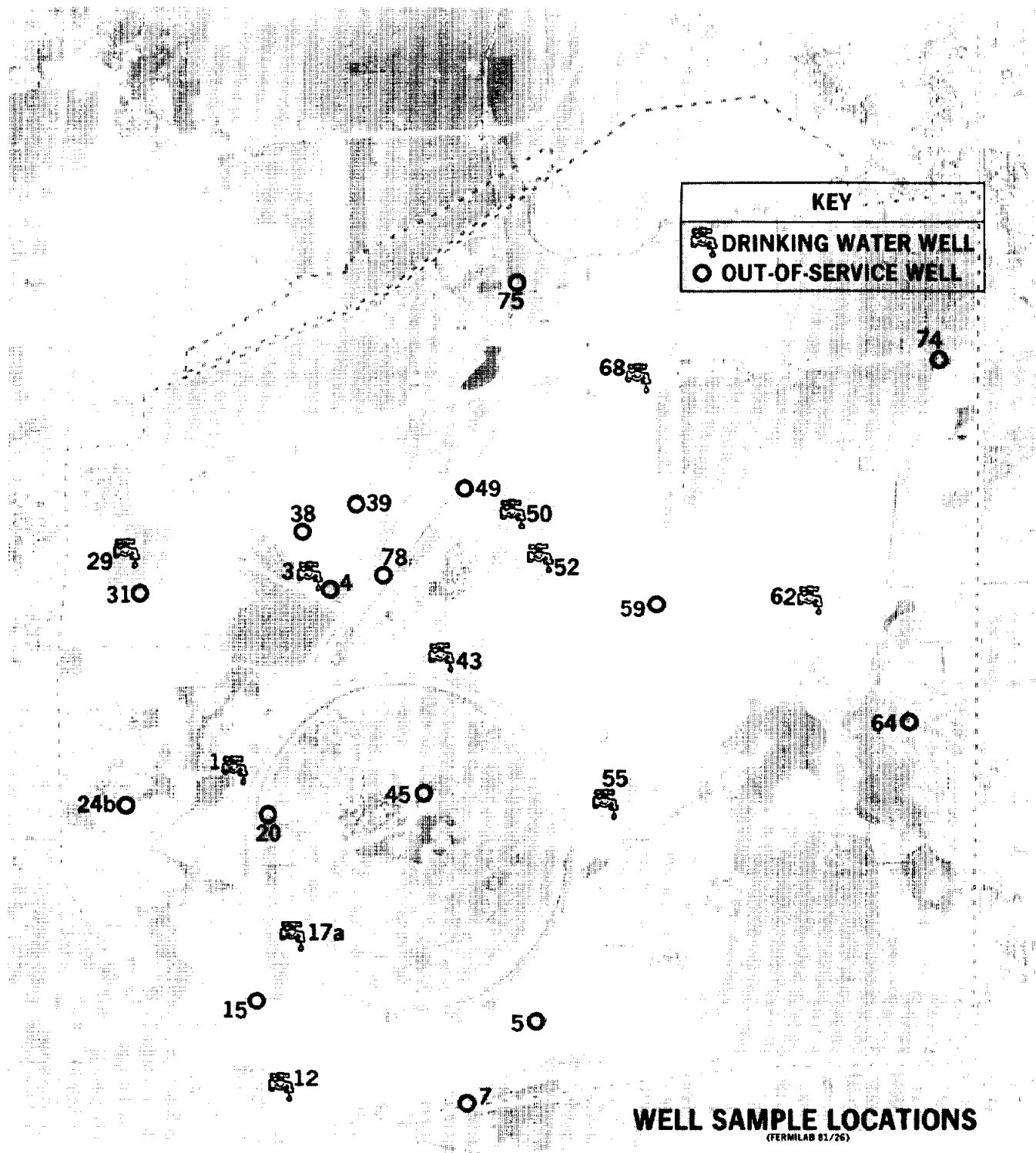
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the DuPage River which passes east of the site flowing south with an average of 265 million liters (70 million gallons) per day through Warrenville (Fig. 2). The rainfall on-site during 1985 was 93 cm (36.7 in).<sup>3</sup> The land on the site is relatively flat with the highest area, elevation 244 m (800 ft) above mean sea level (MSL), near the western boundary and the lowest point, elevation 218 m (715 ft), above MSL, toward the southeast. The drainage of the groundwater and most of the surface water is toward the southeastern corner of the Laboratory, toward the DuPage River. A somewhat smaller amount drains to the southwest, toward the Fox River. The drinking water in many of the surrounding communities comes from deep wells usually drilled 360 m (1200 ft) deep into the Cambrian/Ordovician aquifer system.<sup>4</sup> Also, there are many individual private wells drilled into the shallow silurian aquifer system around 30 m (100 ft) below the surface.

The drinking water used on the Fermilab site comes from the shallow Silurian dolomite aquifer.<sup>4</sup> These wells (primarily 1, 3, and 62 in Fig. 4) collect water from 20 to 70 m (65 to 220 ft) below the surface. The surface cooling waters are used for cooling the accelerator and some experimental area facilities through heat exchangers. The surface cooling water supply is augmented as necessary by pumping water from the nearby Fox River.

The land within the site boundary was primarily farm land before the State of Illinois acquired the site for Fermilab although the small village of Weston (population around 600 at that time) was located on the eastern

Figure 4



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side (Fig. 1). Much of the land, approximately  $10 \text{ km}^2$  (2500 acres), has remained in crop production, primarily corn. About  $1.7 \text{ km}^2$  (428 acres) has been planted in native prairie vegetation. The village of Weston has provided residences for visiting scientists as well as support facilities for the research program.

2. Summary

Fermilab began an extended period of high energy physics research using the superconducting accelerator in January 1985. The fixed target program continued until the end of August 1985. Extraction of  $1.5 \times 10^{18}$  800 GeV protons to the Experimental Areas was achieved. Then a period of antiproton studies commenced and continued until the middle of October 1985. The first head-on collision of an 800 GeV proton and an 800 GeV antiproton was observed at the end of this period. Approximately  $1 \times 10^{17}$  120 GeV protons were accelerated to produce antiprotons during these studies. In addition,  $3 \times 10^{17}$  120 to 150 GeV protons were accelerated for studies during the fixed target running period. The total number of protons accelerated in 1985 was  $1.9 \times 10^{18}$ . This total is 10% of the annual average of  $1.9 \times 10^{19}$  for conventional magnet operation from 1978 through 1982. Thus, environmental monitoring in CY-1985 was done to investigate the impact of the higher energies from the superconducting accelerator and to check on effects resulting from previous accelerator operations.

During CY-1985 there was one abnormal occurrence which had an impact on the facility and its operation. A 100 megavolt-amp (MVA) pulsed-power transformer failed on January 24, 1985. This transformer was the primary one for stepping down the incoming 345 kilovolts to 13.8 kilovolts for accelerator use. The transformer casing cracked and a small fraction of the oil leaked out onto the transformer pad. Most of the oil was recovered by pumping into drums. It is estimated that less than 380 l (100 gallons) entered the environment. There was no known adverse environmental impact



as a result. Fortunately, a spare transformer was available and the high energy research program was able to continue.

The maximum potential radiation exposure at the site boundary during CY-1985 (fence line assuming 24 hr/day occupancy) was 1.5 mrem compared to 0.8 mrem last year<sup>5</sup> and to an average exposure of 3 mrem per year during conventional magnet operations from 1978 to 1982. The maximum individual potential exposure to the general population would be essentially the same because the decrease in dose rate is small between the site boundary and the location of that individual. The exposure is 1.5 percent of the new standard of 100 mrem per year for chronic exposure to the general population (Section 5).

The total potential radiation exposure to the general off-site population from Fermilab operations during CY-1985 was 1.2 person-rem compared to 4.5 person-rem last year. The primary source of potential exposure was penetrating radiation from muons again this year. All exposure was from external radiation, as was the case in the past. Thus, the 50 year dose commitment from operations in 1985 is expected to be the same as the exposure received in 1985.

A summary of off-site releases of radioactive effluents in CY-1985 is given in Table 1. Airborne radioactivity was released across the site boundary from the stack ventilating a Neutrino Area enclosure where the proton beam strikes a target. The total release in CY-1985 was 150 Ci. The radionuclide released was <sup>11</sup>C with 20 minute half-life. No tritiated

water was evaporated as a means of disposal in CY-1985. The off-site release of tritium in surface water totaled approximately 165 mCi, about the same as last year's release. The primary source of tritium in that surface water was tritiated water discharging from an underdrain system beneath a target and dump system. The target was the primary target in the Neutrino Area. That target received most of the protons accelerated at Fermilab. After the CY-1982 operating period ended, the target was moved to a new location with a different underdrain system. Thus, the tritium released in CY-1985 was essentially from operations before CY-1983.

Table 1  
Summary of Radioactivity Released to the  
Off-Site Environment in CY-1985

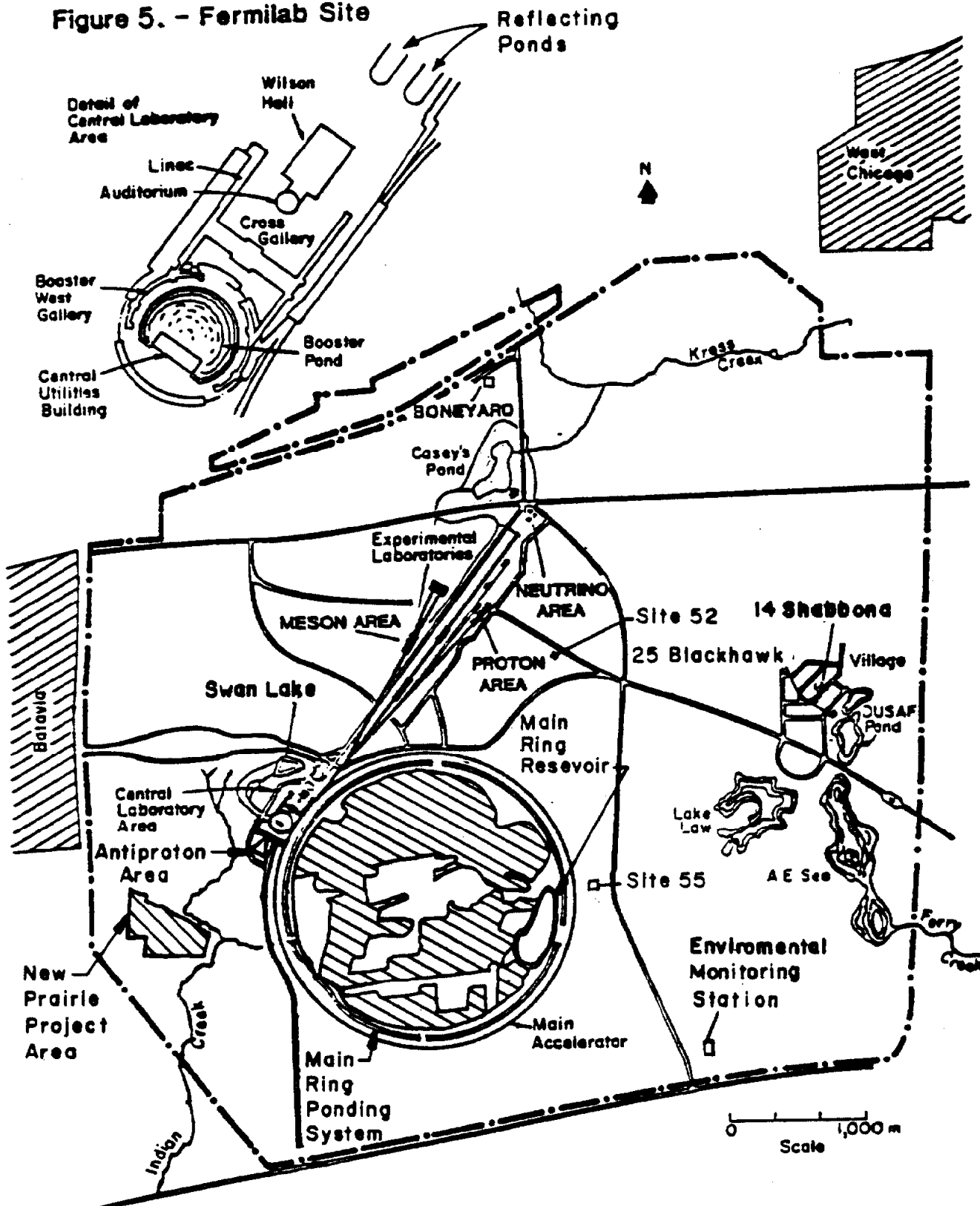
<u>Release Point</u>	<u>Radionuclide</u>	<u>Pathway</u>	<u>Release in Curies</u>
NS1 Enclosure	$^{11}\text{C}$	Air	150
Debonding Oven	$^3\text{H}$	Air	0.003
Kress Creek Spillway	$^3\text{H}$	Water	0.17

The sewage treatment plant, a single-stage oxidation pond, in the Village (Fig. 1) exceeded permit limits for biochemical oxygen demand and suspended solids in CY-1985. Eighteen of the excursions in CY-1985 resulted from algal blooms and excessive rainfall on the 0.2 km (5 acre) pond rather than directly from improperly treated sewage. In addition, there was one report of excessive fecal coliform. There was no known adverse environmental impact on the on-site ponds downstream (Fig. 5) or off the site from these excursions. The State of Illinois and the

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U.S. Environmental Protection Agency were notified of these permit violations. The Illinois Environmental Protection Agency inspected the plant on November 5, 1985 and recommended that steps be taken to achieve compliance. The Laboratory has made plans to connect to the City of Warrenville sewage collection system and to stop using the Village oxidation pond for sewage treatment.

Figure 5. - Fermilab Site



Prairie Restoration Project Areas



3. Environmental Program Information

3.1 Summary of Environmental Monitoring Performed in CY-1985

Fermilab performed extensive environmental monitoring in CY-1985 on three types of accelerator-produced radiation: penetrating, airborne, and waterborne. The penetrating radiation of concern was muons. Neutrons and gamma-rays were also monitored. The airborne radionuclide was  $^{11}\text{C}$  and the primary waterborne radionuclide was  $^3\text{H}$  (tritium). The Department of Energy (DOE) regulations requiring this monitoring are found in DOE Order 5480.1A, Chapter XI. The penetrating radiation measurements were made primarily using a mobile environmental radiation laboratory (MERL), a vehicle with detection equipment. A network of 120 fixed detectors with continuous data recording was also used.

For airborne effluents a continuously operating stack monitor recorded the concentration released from the stack. For waterborne effluents a meter recorded the volume of water discharged. Monthly water samples were analyzed for tritium concentration. The fraction of the year the water left the site was determined by weekly inspections of the spillway.

The data on radioactive waterborne effluents was reported to the Department of Energy via the Effluent and On-Site Discharge Information Systems operated for the Department of Energy by EG&G, Idaho. The extremely low site boundary airborne radionuclide concentrations have been exempted from those reporting requirements.

Additional monitoring for radionuclides in sediment and vegetation on the site has been done to investigate other possible pathways to the off-site environment.

The results have been a small percentage of applicable standards in every case. In particular, the highest off-site penetration radiation level was 1.5% of the relevant standard. The highest airborne radionuclide concentration was 0.1% of the standard and the highest waterborne concentration was less than 0.6% of the standard. See Section 5 for applicable standards.

Monitoring for chemical pollutants in drinking water systems on the site is done biennially. Wells and, in cases where distribution systems are extensive, distribution systems were sampled in CY-1985. The results indicated good water quality in all the wells sampled. The only parameters which exceeded the standard in any sample were iron, total dissolved solids, and in one well, sulfate. The iron is believed to be an indication of rust in the plumbing in those cases rather than iron in the aquifer.

Chlorine is tested every work day in the chlorinated drinking water supplies on the site. Routine monthly sampling includes fecal coliform and pH. No fecal coliform has been detected. There was one test showing no chlorine in the Main Site Water Supply in March 1985. All other tests were satisfactory.

Creeks and ponds are sampled semiannually for pH, dissolved oxygen, biochemical oxygen demand, suspended solids, and fecal coliform. Results met standards for waters in general use in CY-1985.

The Village sewage treatment plant effluent is sampled monthly for pH, biochemical oxygen demand, suspended solids, and fecal coliform. There were 19 permit violations in CY-1985.

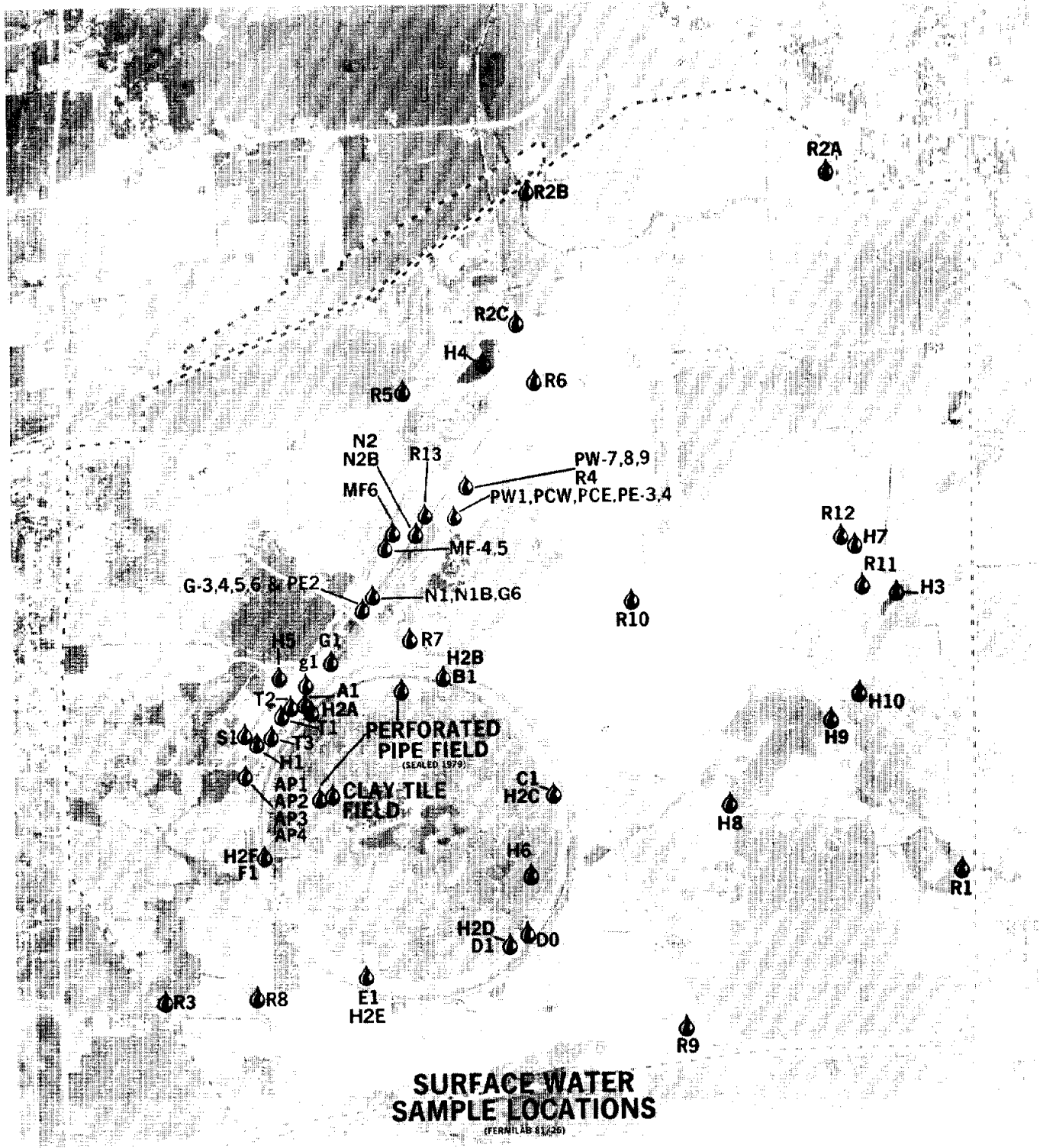
A coring was made to a depth of 1.5 m (5 ft) in the CUB Tile Field (clay tile field in Fig. 6) to obtain samples for heavy metal and salt analyses in CY-1985. Heavy metals have decreased from results in CY-1982. The peak chloride concentration is approximately three times higher than in CY-1982. Salt,  $^{7}\text{Be}$ , and low concentrations of heavy metals are released to the tile field from the regeneration of ion exchange resins. These resins are used to remove impurities from closed loop water systems.

### 3.2 Environmental Program Description

Since its inception, Fermilab has endeavored to protect and enhance the environment. For over ten years a prairie restoration project has been in progress on the 1.6 km<sup>2</sup> (400 acre) plot inside the main accelerator ring (Main Ring in Fig. 1). Recently the prairie project was expanded to include an area outside the ring (Fig. 5). In another effort to enhance the environment, farm houses were moved from their original locations to a site at the south end of the Village (Fig. 1) and renovated to provide housing for scientists performing experiments at Fermilab rather than

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Figure 6



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abandoned and allowed to deteriorate. Some farm wells were maintained for monitoring and others were properly sealed to prevent inadvertant contamination of the aquifer. Ponds and lakes were created to control surface run off and provide cooling water for the accelerator and experimental areas. Over 40,000 trees have been planted to improve the environment. In addition, strong emphasis has been placed on the control of chemical and radioactive materials as potential sources of environmental pollution. Adequate shielding has been provided for preventing exposure from penetrating radiation.

The Fermilab environmental radiological monitoring program follows, in general, the guidance given in the Department of Energy (DOE) report A Guide for Environmental Radiological Surveillance at DOE Installations.<sup>6</sup> This includes adherence to the standards given in DOE orders, in particular, DOE Order 5480.1A, Chapter XI, which pertains to permissible doses due to radioactive releases, and gives guidance on maintaining exposures to as low as reasonably achievable (ALARA).<sup>7</sup> In addition, the environmental monitoring is supplemented by effluent monitoring following, in general, the guidance given in the Department of Energy (DOE) report A Guide for Effluent Radiological Measurements at DOE Installations.<sup>8</sup>

The emphasis has been placed on potential environmental exposure pathways appropriate to high energy physics laboratories. These pathways include external exposure and internal exposure. The external exposure is from direct penetrating radiation and airborne short-lived <sup>11</sup>C. The internal exposure is from <sup>3</sup>H and <sup>22</sup>Na in water, primarily potential

drinking water. There is one unique characteristic at Fermilab which requires consideration. That is the use of large volumes of sand and gravel in two locations to assist in stopping the high energy protons and secondary particles. Although the groundwater beneath these two areas is protected by membranes impervious to water and by underdrain systems to collect the water, radiological monitoring of soil and water is necessary. See Section 3.3.3.4.

Monitoring results are also reported for non-radioactive pollutants. Included as pollutants are pesticides used in weed, insect, rodent, and algae control. In addition, results are included from monitoring the performance of the sewage treatment plant (Village Oxidation Pond) on-site. Discharges of suspended solids and measurements of biological oxidation demand in effluent from this plant have frequently exceeded permit limits. See Section 3.4.4.

### 3.3 Environmental Radiation Monitoring

The three types of accelerator-produced radiation which receive extensive environmental monitoring are discussed below - penetrating radiation, airborne radioactivity, and waterborne radioactivity. These radiations may have direct pathways to the off-site population. Other more indirect pathways, such as through the food chain, have received much less attention to date. The decision on what to monitor is based on the type of operation, radionuclides released, potential hazard, and monitoring results from this and other high energy physics laboratories.

### 3.3.1 Penetrating Radiation

Operation of the accelerator at current energies and intensities results in production of some penetrating radiation (primarily muons) outside the shielding. Although the shielding has been designed to be adequate for this operation, monitoring for purposes of determining actual radiation levels both on and off the site is necessary.

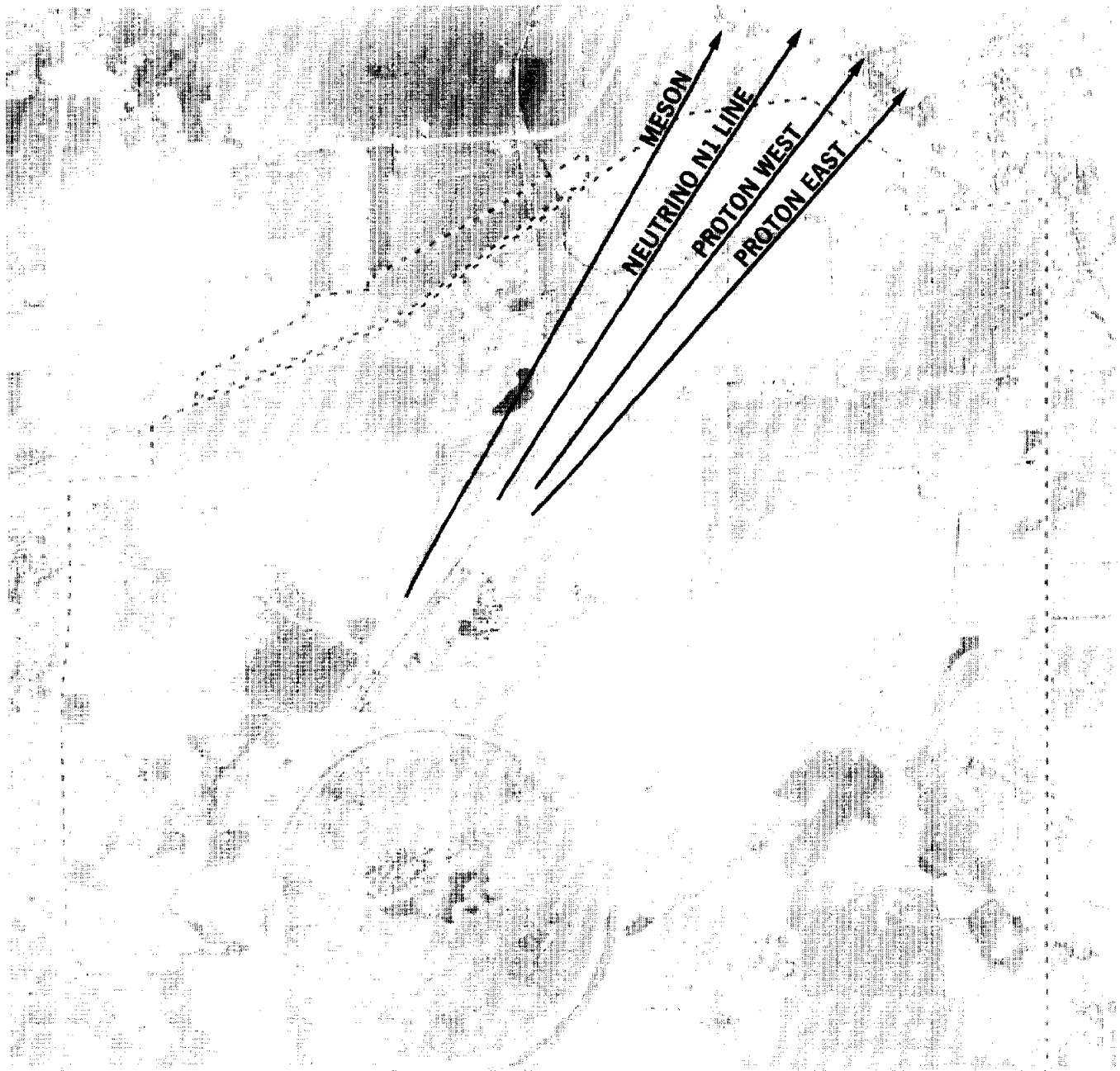
A large network of detectors was used to monitor penetrating radiation. At the end of CY-1985 there were approximately 120 detectors deployed around the site for the main purpose of protecting on-site personnel. The majority of these detectors were connected to a data logger which automatically recorded the radiation levels for subsequent examination.<sup>9</sup> Because the intensity was low in CY-1985 only four detectors were used primarily for environmental radiation monitoring. One was a large volume, 110 liter, ionization chamber (called a Hippo) for gamma-ray and charged particle detection. It was located near the Boneyard at the Railhead (Fig. 5). Two of the remaining three detectors were large scintillation counters. One was located near the site boundary (Environmental Monitoring Station in Fig. 5). The other was located at Site 52 near the experimental laboratories (Fig. 5). The last was a tissue-equivalent ion chamber located at 14 Shabbona in the Village (Fig. 5).

The Mobile Environmental Radiation Laboratory (MERL) was used in the past for determining the exposure levels at the site boundary and for locating the source and direction of penetrating radiation such as muons and neutrons.<sup>10,11,12,13</sup> The MERL is a four-wheel-drive vehicle equipped with two 20 cm x 20 cm (8 in x 8 in) scintillation counters, one approximately 15 cm (6 in) behind the other, for muon detection. It also has a dePangher "long counter" for neutron detection.<sup>14</sup>

### 3.3.1.1 Muons

Measurements of muons from the Meson, Neutrino, and Proton Areas were made in CY-1985 while the accelerator was delivering 800 GeV protons. The directions of penetrating muons are shown in Figure 7. The muons which penetrate the earth shielding can travel far beyond the site boundary through the air before stopping. Therefore, measurements were made both on and off the site. The site boundary muon dose rates for CY-1985 were then determined from these measurements and the numbers of protons incident on the targets at 800 GeV. The maximum fence line annual dose based on 24 hour per day occupancy was 1.5 mrem for CY-1985 for the Meson Area. Because the dose rate varied inversely with distance and the distance to the nearest off-site individual was small compared to the distance from the muon source, the maximum individual dose rate was approximately the same as the fence line dose in CY-1985. The fence line annual dose from the Proton West beam line was 0.3 mrem for CY-1985. The doses from the other beam lines were negligible.

Figure 7



**PENETRATING RADIATION  
(MUON) DIRECTIONS**  
(FERMILAB REPORT TM-1061 1981)



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### 3.3.1.2 Neutrons

Neutrons penetrated the shielding in the most easterly of the external experimental areas (Proton East line in Fig. 5) in the Proton Area in CY-1982.<sup>13,16</sup> However, in CY-1983 additional shielding was added to this area resulting in negligible site boundary dose rates from neutrons since that time.

### 3.3.1.3 Gamma Rays

The primary radioactive waste storage area on-site - the Boneyard - is also the primary source of off-site gamma radiation. Activated accelerator components and shielding, primarily iron and concrete, are stored at the Boneyard for future disposal or reuse following radioactive decay. As shown in Fig. 5, the Boneyard, which is a secure area, lies close to the site boundary. On the north side there is an earth berm to prevent any direct radiation from leaving the site. Shielding has been provided above and on all sides of those radioactive materials which would produce high radiation levels without shielding. This was done to protect Fermilab workers as well as reduce the off-site dose. Radiation levels at the site boundary closest to the Boneyard were at background levels in CY-1985.

### 3.3.2 Airborne Radioactivity

Radioactivation of air in measurable concentrations will occur wherever the proton beam or the spray of secondary particles resulting from its interactions with matter passes through the air. Along most proton beam lines (paths of the protons from the accelerator) the protons travel inside evacuated pipes. Thus, radioactivation of air is now usually caused by secondary particles. Monitoring of such activation is carried out for purposes of personnel exposure control. Under no circumstances is the off-site concentration of airborne radioactivity expected to approach the limits for uncontrolled areas. During CY-1985 the primary target in the Neutrino Area was in operation. This target has been the source of radioactive gas, primarily  $^{11}\text{C}$ , which was produced by interaction of secondary particles from this target with air. The concentration at the stack was recorded using a continuous air monitor which detected the positron from the decay of 20 min half-life  $^{11}\text{C}$ . The site boundary concentration was calculated using a gaussian plume diffusion model<sup>16</sup> with neutral wind conditions<sup>17</sup> and average wind speed. In CY-1985 the average wind speed at O'Hare International Airport, Chicago, Illinois was 4.6 m/sec (10.3 mi/hr).<sup>19</sup> Fermilab is about 43 km (27 mi) from the airport and the terrain between them is relatively flat. The site boundary  $^{11}\text{C}$  dose for CY-1985 was 0.026 mrem. See Section 3.3.3.3 for vegetation sampling results near airborne release points. Radionuclides are still present from earlier operations as well as from operations in CY-1985.

1985

A debonding oven was placed in operation in CY-1979 in the Industrial Area (Fig. 1). Its purpose is to debond magnets by decomposing the epoxy adhesives at high temperatures. Most of these magnets are radioactive and have failed during accelerator operations. The gaseous effluent was measured during the acceptance test on June 8, 1979 conducted for the Illinois EPA and contained only  $^3\text{H}$  at very low levels. The tests were primarily performed to measure non-radioactive emissions. The test utilized a typical 6 m (20 ft) long magnet reading 0.8 mrem/hr at 0.3 m (1 ft) from the surface and 8 mrem/hr in the bore tube where the protons traveled. The total amount of  $^3\text{H}$  released from this magnet was 160  $\mu\text{Ci}$  at a stack concentration of  $1.3 \times 10^{-8}$   $\mu\text{Ci}/\text{m}^3$  or about 20 percent of the Concentration Guide (Section 5) corresponding to 500 mrem per year. The stack is approximately 10 m (30 ft) high. Using the Gaussian plume diffusion model<sup>17</sup> with neutral wind conditions<sup>18</sup> gives a negligible percentage of the applicable Concentration Guide at the site boundary.

The number of radioactive magnets debonded in CY-1985 was 14 corresponding to a total release of 3 mCi of  $^3\text{H}$  into the air. The radioactivity in the magnets was similar to that in the 1979 test, thus the 1979 data are still valid. Two magnets had radiation levels of 2 mrem/hr at 0.3 m (1 ft). Those releases were determined by multiplying 160  $\mu\text{Ci}$  by  $2/0.8$ , the ratio of dose rates. In CY-1985 the wind conditions were similar to those in past years.



A water evaporator was placed in operation in CY-1981 at the Boneyard (Fig. 4) to dispose of tritiated water collected from closed loop cooling systems. No tritiated water was evaporated in CY-1985.

### 3.3.3 Waterborne Radioactivity

During accelerator operations, some radioactivation of the soil will occur.<sup>20,21</sup> Leaching of these radionuclides into the groundwater provides a possible mechanism for transport of Fermilab-produced radionuclides into the surface run-off waters and aquifer. Hence, a broad program of groundwater monitoring for radioactivity is maintained. Measurements are also made of on-site concentrations of radionuclides in Fermilab surface waters and in closed loop (recirculating) cooling systems which are sources of potential off-site releases.

Water samples are collected periodically on-site and from surface waters off-site. They are analyzed for the presence of those radionuclides which are produced in and leachable from Fermilab soils in measurable quantities.<sup>20</sup> This group of radionuclides also includes those produced in water directly. Analyses are made for  $^3\text{H}$ ,  $^7\text{Be}$ ,  $^{22}\text{Na}$ ,  $^{45}\text{Ca}$ ,  $^{54}\text{Mn}$  and  $^{60}\text{Co}$ . The latter is hardly leachable (approximately 0.1 percent); however, it has been detected in discharges during regeneration of water treatment resins. These ion exchange resins are used to remove impurities from water in closed loop systems.

Water samples were collected from the following types of wells on-site:

1. Farm Wells - Approximately 30 m Deep - 38 Samples
2. Fermilab Water Supplies - Approximately 70 m Deep - 5 Samples
3. Fermilab Deep Well Emergency Supply - 436 m Deep - 1 Sample

The wells routinely sampled are shown in Figure 4. Water samples were also collected from sumps, creeks, and rivers. All surface and groundwater samples collected were analyzed by Teledyne Isotopes, Inc., 1500 Frontage Road, Northbrook, Illinois 60062. Each monthly shipment included at least one sample containing accelerator-produced radionuclides in known amounts to check the accuracy of the assays. See Section 4 on quality assurance.

#### 3.3.3.1 Water Sample Collection

To obtain water samples from wells not in regular use, the wells are pumped for a sufficient length of time to insure that the water standing in the pipe has been pumped out before a sample is taken. The water in the pipe could conceivably have been there since the last time a sample was taken. Normally, the pipe volume is pumped several times before sampling. Water samples from sumps, creeks and other surface waters are normally collected by dipping a bottle well below the surface. Several of the sumps

inside normally locked enclosures are sampled by remotely operated peristaltic pumps or the sump pumps themselves. Also, in CY-1984 meters were added to record the operating time of sump pumps which pumped radioactive water.

The water sampling schedule is based on the following rationale:

1. Wells 38/39\*, 43, 49, & 78 are sampled quarterly because they are closest to the areas of maximum soil activation (near targets and dumps) and are in the direction the water is expected to flow in the aquifer.
2. Wells 1, 5, 17A, 20, and 45 are sampled semiannually because they are near the accelerator.
3. The remaining wells are sampled annually because they are near the site boundary or serve as back-ups to more frequently sampled wells or as drinking water supplies.
4. The one deep well is sampled annually to look for long-term trends or changes in percolation down to that level.

---

\*38 and 39 are close to each other and sample the same region of the aquifer. Each is sampled semiannually. See Figure 4.

5. The MF5, N1, N2, and PW8 sumps are sampled bimonthly because they are the closest to the areas of maximum soil activation. See Figure 6.
  
6. The MF4 sump and the N1 Retention Pit are sampled quarterly because the MF4 sump collects water from a region with less activity than that of the MF5 sump (outside the impervious membrane instead of inside) and the N1 Retention Pit does not have a pump in it even though it collects from a region of higher activity than the N1 sump. The N1 Retention Pit water is monitored and disposed of properly.
  
7. The other sumps are sampled less often with the frequency based on the tritium concentration found there in the past.
  
8. The creeks are routinely sampled three times a year and Kress Creek is sampled monthly whenever water from the Laboratory flows over the spillway into the creek. Ferry and Indian Creek were sampled three times and Kress Creek was sampled five times in CY-1985.
  
9. Ponds and ditches with a potential for receiving radioactive water are sampled annually.
  
10. The Fox River and west branch of the DuPage River which receive run-off from Fermilab are sampled annually.

11. The closed loop cooling systems which cool targets and dumps are sampled with a frequency which depends on the level of radioactivity. Operating systems having concentrations greater than 0.01  $\mu\text{Ci}/\text{m}\ell$  are sampled quarterly ( 333% of the relevant standard in Section 4). Those having concentrations between 0.001 and 0.01  $\mu\text{Ci}/\text{m}\ell$  (33 and 333% of the standard) are sampled semiannually. Those between 0.00001 and 0.001  $\mu\text{Ci}/\text{m}\ell$  (0.33 and 33%) are sampled annually. The total number of closed loop samples is approximately 20 per year.
  
12. The ion exchange resin regeneration systems are routinely sampled for analysis on-site. Semiannually one of these samples is sent to an outside laboratory for analysis as part of the quality assurance program. The regeneration systems remove radionuclides such as  $^7\text{Be}$ ,  $^{54}\text{Mn}$ , and  $^{60}\text{Co}$  as well as calcium and other non-radioactive impurities from the resins which function to keep conductivity of closed loop water systems low. Analyses are performed on-site for samples from every regeneration sending radioactive effluent to the Central Utilities Building (CUB) tile field inside the Main Ring. The line to the tile field was frozen from December 19, 1985 until the end of the year. During that time there were three regenerations. The discharge from these was sent to the Booster Pond. Samples were taken and results of the analyses are given in Section 3.3.3.2.2.
  
13. Several samples are collected annually to look for radioactivity leached from activated steel.

### 3.3.3.2 Results of Analyses

All current Fermilab water sampling locations for detection of accelerator-produced activity are shown in Figs. 4 and 6. Not all locations need to be sampled every year. See Section 3.3.3.1 above. No accelerator-produced radionuclides were reported in five water samples taken from Kress Creek (R2A in Fig. 6) and three samples each from Indian Creek and Ferry Creek. No accelerator-produced radionuclides have ever been detected in the water from the creeks and rivers. Thus, the results are not included in Table 2. All water samples with detected activity are reported in Table 2 with the exception of the sample containing radium and thorium from the deep well. River water samples were obtained once during CY-1985 from the Fox River in Aurora and from the west branch of the DuPage River in Warrenville (Fig. 2). Neither river is utilized as a drinking water supply.

The Village water supply (62 in Fig. 4) is the Laboratory's only community water supply. EPA regulations require analysis of community water supply samples every four years for radionuclides.<sup>22</sup> Quarterly water samples were collected and a composite analyzed for naturally occurring as well as accelerator-produced radionuclides in CY-1985. No activity was found. See Section 4.

1985

3.3.3.2.1 Tritium

The results for on-site tritium measurements yielding detectable levels in surface waters (Fig. 6) are given in Table 2. All other sampling points were at background levels. The sumps collect waters from around the footings of the buildings and enclosures. This water is considered surface water. Only aquifers are called ground waters. The total off-site release in surface waters was 165 mCi of tritium this year, about the same as the 150 mCi released last year. Only 17% of the discharge from the N1 sump left the site, as discussed below. The total amount of radioactivity discharged from that sump was less than in the immediately preceding years (Fig. 8). The off-site release occurred at less than 0.2 percent of the Concentration Guide (Section 5) and made a negligible contribution to the potential off-site dose. Detailed reports of off-site effluent releases and on-site discharges are made via the Department of Energy Effluent and On-Site Discharge Information Systems, EG&G, Idaho.

The surface water from the experimental areas flows into Casey's pond (Fig. 5) except during wet seasons. Then, the pond fills up and barricades are placed at the two entrances to the pond to keep the water from flooding the pump room. When these barriers, called stop logs, are in place, the water bypasses the pond and leaves the site via Kress Creek (Figs. 5 and 9). This was the case for approximately 17% of the year in CY-1985. There were no discharges of radioactivity from a closed loop water system leak in CY-1985.

Tritium Detected in On-Site Water Samples

Tritium Concentration C ( $\mu\text{Ci}/\text{mL}$ )\*

Collection Point	Number of Samples Collected	C Max	C Max Error	C Min	C Min Error	C Mean	Percentage of Relevant Standard
B1 Sump	1	$4.1 \times 10^{-6}$	$1.4 \times 10^{-6}$	$4.1 \times 10^{-6}$	$1.4 \times 10^{-6}$	$4.1 \times 10^{-6}$	0.041
D1 Sump	1	$9.0 \times 10^{-6}$	$3.0 \times 10^{-7}$	$9.0 \times 10^{-6}$	$3.0 \times 10^{-7}$	$9.0 \times 10^{-6}$	0.09
G4 Sump	2	$9.1 \times 10^{-5}$	$3.2 \times 10^{-6}$	$8.9 \times 10^{-5}$	$7.0 \times 10^{-7}$	$9.0 \times 10^{-5}$	0.9
G5 Sump	2	$1.1 \times 10^{-5}$	$1.7 \times 10^{-6}$	$4.6 \times 10^{-6}$	$4.0 \times 10^{-7}$	$8.1 \times 10^{-6}$	0.081
G7 Sump	3	$3.4 \times 10^{-4}$	$1.7 \times 10^{-6}$	$4.2 \times 10^{-6}$	$1.3 \times 10^{-6}$	$1.2 \times 10^{-4}$	1.2
MF4 Sump	6	$4.7 \times 10^{-6}$	$1.5 \times 10^{-6}$	$<3.0 \times 10^{-6}$	---	$7.8 \times 10^{-7}$	0.0078
MF5 Sump	6	$5.9 \times 10^{-5}$	$2.8 \times 10^{-6}$	$4.7 \times 10^{-5}$	$7.0 \times 10^{-7}$	$5.3 \times 10^{-5}$	0.53
N1 Sump	6	$3.3 \times 10^{-4}$	$4.3 \times 10^{-6}$	$8.7 \times 10^{-5}$	$9.0 \times 10^{-7}$	$1.8 \times 10^{-4}$	1.8
N2 Sump**	12	$6.6 \times 10^{-4}$	$6.9 \times 10^{-6}$	$8.8 \times 10^{-5}$	$1.1 \times 10^{-6}$	$3.7 \times 10^{-4}$	3.7
N2B Sump	2	$5.9 \times 10^{-6}$	$1.4 \times 10^{-6}$	$<3.0 \times 10^{-6}$	---	$3.0 \times 10^{-6}$	0.03
G3 Sump	1	$2.5 \times 10^{-5}$	$4.0 \times 10^{-7}$	$2.5 \times 10^{-5}$	$4.0 \times 10^{-7}$	$2.5 \times 10^{-5}$	0.25
NE8 Sump	2	$2.7 \times 10^{-5}$	$1.6 \times 10^{-6}$	$7.1 \times 10^{-6}$	$2.0 \times 10^{-7}$	$1.7 \times 10^{-5}$	0.17
NM2 Sump	2	$3.2 \times 10^{-5}$	$1.7 \times 10^{-6}$	$<3.0 \times 10^{-6}$	---	$1.6 \times 10^{-5}$	0.16
PW7 Sump	2	$4.1 \times 10^{-6}$	$1.4 \times 10^{-6}$	$<3.0 \times 10^{-6}$	---	$2.0 \times 10^{-6}$	0.02
PW8 Sump	2	$6.5 \times 10^{-6}$	$1.4 \times 10^{-6}$	$<3.0 \times 10^{-6}$	---	$3.3 \times 10^{-6}$	0.033
PW9 Sump	2	$5.5 \times 10^{-6}$	$1.4 \times 10^{-6}$	$<3.0 \times 10^{-6}$	---	$2.7 \times 10^{-6}$	0.027

\*C Max is the highest concentration detected in any sample from that location and C Min is the lowest. C Mean is the average for all samples from one location.

\*The high concentrations in this sump are assumed to be due to accumulation during the period of time that the sump was not pumping automatically. This period of time spanned about 75% of the year. The sump was pumped about four times during that period.

5211



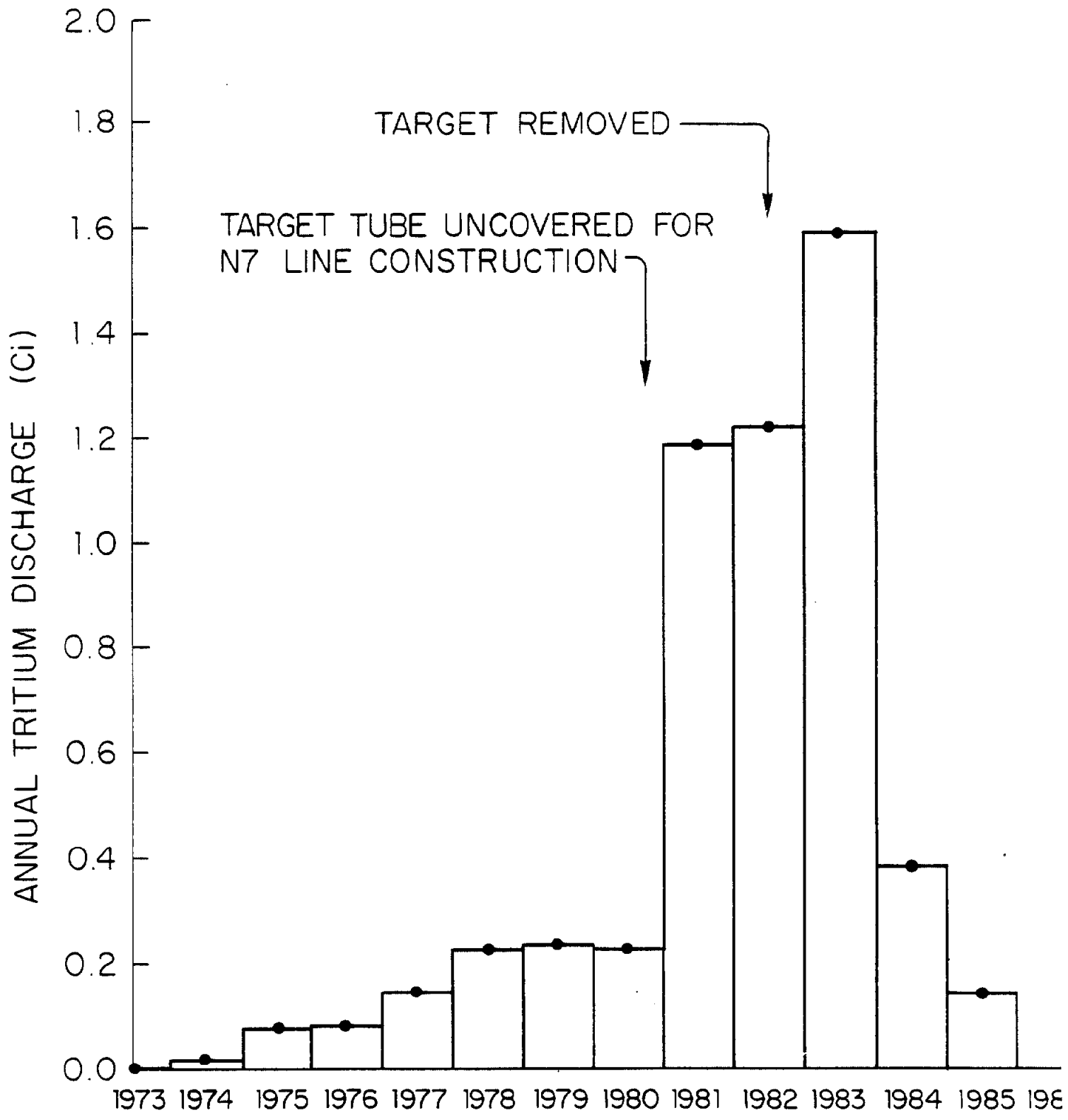
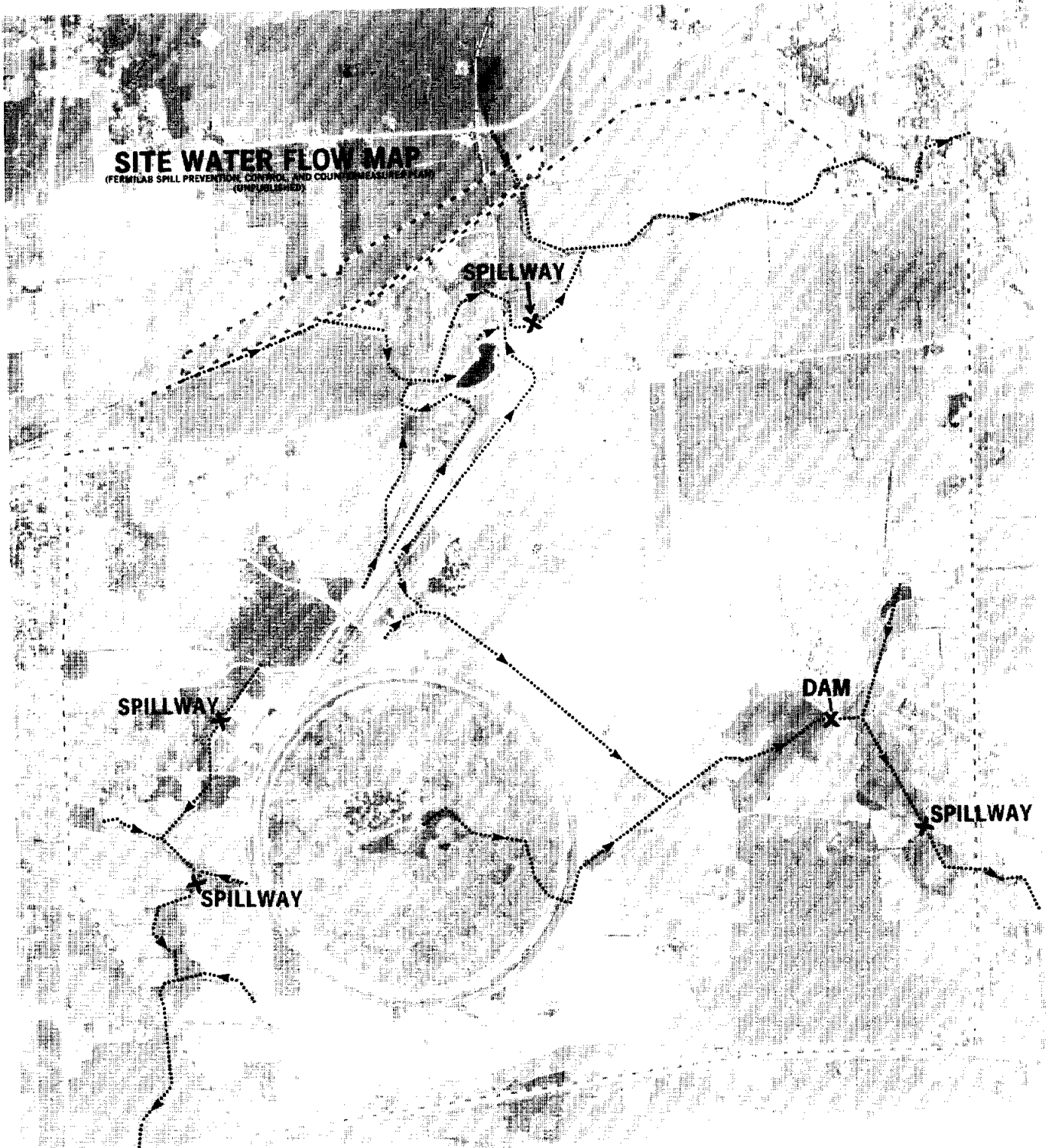


Figure 8. Target Tube Underdrain Discharges

1785

Figure 9



OVERLAY FOR BASE 4, FERMILAB

PREPARED IN 1982



FERMI NATIONAL  
ACCELERATOR LABORATORY

DRB

1985

### 3.3.3.2.2 Beryllium

Concurrent with the production of  $^3\text{H}$  with 12 year half-life is the production of  $^7\text{Be}$  with 53 day half-life in the closed cooling water systems. The  $^7\text{Be}$  is chemically active and is easily removed from the water by the resins used to maintain water purity. These resins are regenerated in two separate systems located at the Central Utilities Building (Fig. 5). The effluent from these two systems is sent to a clay tile field inside the main accelerator (Fig. 7). There it percolates into the soil about 60 cm (2 ft) below the surface. The short half-life of  $^7\text{Be}$  and its strong chemical affinity with the soil ensure that the release will place no burden on the environment. The total amount of  $^7\text{Be}$  discharged to the tile field in CY-1985 was 25 millicuries. During the latter part of December 1985 the tile field was frozen and  $^7\text{Be}$  was discharged into the Booster Pond (Fig. 5). The total amount of  $^7\text{Be}$  discharged was 0.5 millicurie at an average concentration of  $1.3 \times 10^{-6}$   $\mu\text{Ci}/\text{ml}$  or 0.03% of the applicable concentration guide.

### 3.3.3.2.3 Other Radionuclides

Tests were also made for radium and thorium in our deep well (4 in Fig. 4) to look for any long-term changes in percolation rates to deep-lying aquifers. The results were consistent showing no changes, as has been the case in the past.

3.3.3.3 Sediment and Vegetation Sampling

Sediment and vegetation samples were taken near discharge points for radioactive effluents. The results for sediment samples are given in Table 3. The vegetation results are given in Table 4. The samples were taken of the top centimeter of sediment in the ditches. Dry weights were obtained by weighing the samples after baking in an oven for at least 24 hours at 115°C.

The presence of  $^{137}\text{Cs}$  (Table 3) indicates fallout from previous atmospheric nuclear testing. The  $^{60}\text{Co}$  should be from fallout or accelerator-produced. The  $^7\text{Be}$  could be from cosmic ray production or accelerator-produced. The radionuclides  $^{22}\text{Na}$  and  $^{54}\text{Mn}$  are only accelerator-produced.

TABLE 3  
CY-1985 Sediment Sampling Results

<u>Location</u>	<u>Concentration (pCi/g dry weight)</u>				
	$^7\text{Be}$	$^{22}\text{Na}$	$^{54}\text{Mn}$	$^{60}\text{Co}$	$^{137}\text{Cs}$
Ferry Creek	--	--	--	--	0.91±0.08
Indian Creek	--	--	--	0.11±0.03	0.32±0.05
Kress Creek	--	--	--	--	0.11±0.04
MF5 Sump	--	0.3±0.1	0.13±0.03	--	0.05±0.03
N1 Sump	--	0.2±0.1	0.16±0.06	0.12±0.04	0.14±0.04
N2 Sump	2.0±1.6	0.07±0.05	0.05±0.05	0.09±0.03	--
PW8 Sump	--	--	--	0.06±0.01	--
T3 Sump	--	1.1±.1	0.27±0.06	0.10±0.04	--

An annual vegetation sampling program was initiated in CY-1978. Vegetation samples were taken near the  $^{11}\text{C}$  exhausts in the Neutrino Area (N1 in Fig. 7) in addition to vegetation samples in areas with waterborne radioactivity. See Table 4.

TABLE 4  
CY-1985 Vegetation Sampling Results

<u>Location</u>	<u>Concentration (pCi/g dry weight)</u> (pCi/ml of soil moisture for $^3\text{H}$ )				
	$^3\text{H}$	$^7\text{Be}$	$^{22}\text{Na}$	$^{54}\text{Mn}$	$^{60}\text{Co}$
Ferry Creek	--	--	--	--	--
Indian Creek	--	--	--	--	--
Kress Creek	--	--	--	0.1±0.1	--
MF5 Sump	--	19±3	2.6±0.2	0.4±0.1	0.2±0.0
N1 Sump	--	15±4	5.4±0.3	0.3±0.1	0.4±0.1
N2 Sump	--	17±3	0.5±0.1	0.1±0.1	--
PW8 Sump	--	9±2	--	--	--
T3 Sump	--	24±4	0.2±0.1	0.3±0.1	--
N1 Labyrinth Stack	49±2	49±4	--	--	--
N1 Muon Line Stack	22±1	12±3	--	--	--
N1 Spur Stack	85±2	91±6	--	--	--
CUB Tile Field	--	--	--	--	--

The peak concentrations for vegetation sampling are based on the dry weight of the sample. The results from the analyses of the vegetation samples indicated small concentrations of radionuclides similar to those seen in the past.<sup>16</sup> In other samples based on previous results<sup>25</sup> the radionuclide  $^7\text{Be}$  is expected to be present as surface contamination - from air while other radionuclides are most likely incorporated into the plants. The vegetation contained small quantities of  $^7\text{Be}$ . The amounts of radioactivity are so low that consumption of the vegetation by animals in the human food chain would be permissible.

#### 3.3.3.4 Soil Activation

Because the percolation rates for water in Fermilab soils are calculated to be very low - less than 1 m (3 ft) per year<sup>26</sup> - analyses of well waters do not provide the early warning desired for detection of accelerator-produced radioactivity in the groundwater. To provide such a warning soil samples were taken from the vicinity of targets and other locations where proton interactions result in some radioactivation of the soil. Many radionuclides were detected but because the major long-lived ones leachable from Fermilab soils were  $^3\text{H}$  and  $^{22}\text{Na}$ , quantitative measurements were made only on those.<sup>19</sup>

In CY-1984 a hole was drilled at 45° to sample the soil below the lowest underdrains beneath the Neutrino Area primary target tube.<sup>4</sup> The target in this 2 m (6 ft) diameter tube has received most of the protons accelerated at Fermilab. The purpose of the hole was to look for radionuclides, primarily  $^3\text{H}$  and  $^{22}\text{Na}$ , which might have been leached and escaped from the water collection system. No evidence was found for large quantities of radionuclides migrating downward toward the aquifer.<sup>5</sup>

In CY-1983 a new target and dump system was put into operation to abort any errant protons inside the Main Ring tunnel. The well shielded dump was placed just outside the tunnel (near C1 in Fig. 6). It was provided with a sampling underdrain which normally is not pumped (Fig. 10). The drains adjacent to the Main Ring tunnel lead to sump pits equipped with pumps, hence water in the region around the tunnel and above the

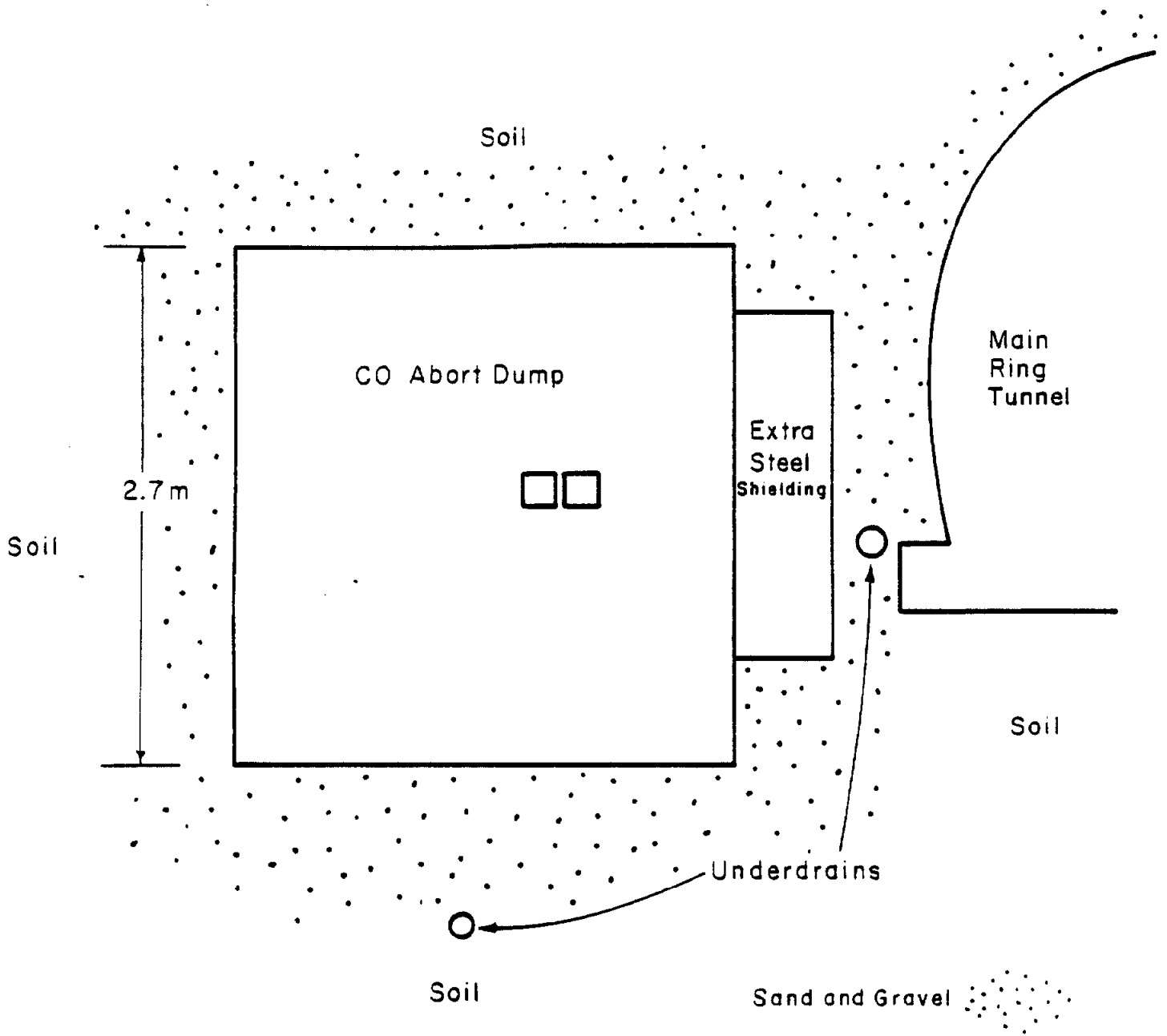


Figure 10. Cross Sectional View at CO Abort Dump

underdrains is normally kept free of standing water. The region below the Main Ring drains is not.

The abort dump itself is sealed watertight. Drains inside have remained dry. The part of the dump below the Main Ring tunnel drains sits in water, permitting leaching of tritium produced in the sand and gravel surrounding the dump (Fig. 10). Water samples from the underdrain beneath the dump contain tritium and  $^{22}\text{Na}$ . The concentrations are below the DOE Concentration Guides for release to surface waters (Section 5). We have begun a program of pumping approximately 2000 liters (approximately 500 gallons) of water from the sampling underdrain periodically to keep the concentration low.

Comparing the predicted radioactivity based on calculations using the computer program CASIM<sup>24</sup> with the observed radioactivity, we find good agreement. The total leachable activity outside the dump assuming the aborts occurred at the full energy of 800 GeV was predicted to be 0.13 (+.26, -.09) mCi for  $^3\text{H}$  and 0.01 (+.02, -0.006) mCi for  $^{22}\text{Na}$  for  $1.8 \times 10^{17}$  protons aborted in CY-1985. The  $^3\text{H}$  activity pumped out was 0.1 mCi and the  $^{22}\text{Na}$  activity was 0.006 mCi. These amounts of radioactivity would not exceed drinking water standards if allowed to percolate down to the aquifer.<sup>21</sup>



3.4 Environmental Monitoring for Non-radioactive Pollutants

3.4.1 Domestic Water Supplies

The primary drinking water supply at Fermilab is provided by two wells pumping from an aquifer approximately 70 m (220 ft) deep. One (1 in Fig. 4) is located in the Central Laboratory Area and the other (62 in Fig. 4) supplies the separate Village system. A third well (3 in Fig. 4) pumps from the same aquifer and supplies water to the Main Site System when demand exceeds the capacity of the Central Laboratory well (1 in Fig. 4).

These wells have chlorination systems and our water laboratory conducts tests for pH and fecal coliform monthly. The chlorine level in the chlorinated drinking water supplies is tested each work day. Test results conformed to Illinois standards during 1985. Samples of the Main Site (Central Laboratory Area) and Village Supplies are independently analyzed by the Illinois Environmental Protection Agency quarterly for total coliform per 100 ml. No coliform was found. Our average use from the Central Laboratory well (1 in Fig. 4) was approximately 412,000 l/day (109,000 gal/day) during 1985, an increase of 25% from 1984.<sup>5</sup> A negligible amount of water was drawn from the backup well (3 in Fig. 4). The average use from the Village well (62 in Fig. 4) was 179,000 l/day (47,000 gal/day) during CY-1985, an increase of 6% from 1984.

Wells are monitored biennially to determine compliance with State of Illinois regulations for non-radioactive pollutants such as heavy metals.<sup>25</sup> Eleven potable water wells (Fig. 4), two wells (20 and 45 in Fig. 4) near

the CUB tile field (Fig. 5), and one well near the Industrial Area (43 in Fig. 4) were sampled directly from the wells CY-1985.<sup>6</sup> Also, samples were taken from the water taps of the distribution systems of three of the 14. The water from all wells except the one in the industrial area was analyzed for nine metals including chromium, iron, lead, mercury, and zinc. Most of the samples were also analyzed for sodium. Total dissolved solids, chloride, fluoride, sulfate, and nitrate plus nitrite were also measured as well as pH and cyanide. Water from the chlorinated systems was analyzed for trihalomethanes: chloroform, dichlorobromomethane, dibromochloromethane and bromoform. Water from the industrial area well and other wells in areas where solvents are used were analyzed for ammonium, trichloroethane, trichloroethylene, and total organic carbon. Also, a few wells were sampled for benzene and gasoline. The analyses were performed by Aqualab, Inc., 850 W. Bartlett Road, Bartlett, Illinois 60103. The results are found in Tables 5-8.

Fermilab's community drinking water supply, the Village well #62, met all drinking water standards (Section 5). Several other wells were above the standard for iron. This is probably from the plumbing. One well, #3, was above the standards for manganese and total dissolved solids as well as iron. However, all these wells including the community water supply are exempt from the standards for iron and manganese. These standards only apply to community water supplies serving larger populations.

TABLE 5

CY-1985 Ground Water Surveillance Wells  
On-Site Heavy Metals Data

Well Number	Ag	Cd	Cr	Cu	Fe	Hg	Mn	Pb	Zn
	----- mg/l -----								
1	<0.001	<0.001	<0.001	<0.001	0.68	<0.001	0.010	<0.01	0.005
1*	<0.001	<0.001	<0.001	<0.001	0.75	<0.001	0.011	<0.01	0.004
1**				0.017	1.71				0.180
3	<0.001	<0.001	<0.001	<0.206	2.13	<0.001	0.205	<0.01	0.092
3**				0.007	0.30				0.012
17A	<0.001	<0.001	<0.001	0.005	1.50	<0.001	0.010	<0.01	0.149
20	<0.001	<0.001	<0.001	0.075	1.74	<0.001	<0.001	<0.01	0.424
29	<0.001	<0.001	<0.001	0.003	0.83	<0.001	0.01	<0.01	0.502
45	<0.001	<0.001	<0.001	0.014	3.10	<0.001	0.010	<0.01	0.148
50	<0.001	<0.001	<0.001	<0.001	1.80	<0.001	0.018	<0.01	0.072
52	<0.001	<0.001	<0.001	0.004	2.20	<0.001	0.035	<0.01	0.313
55	<0.001	<0.001	<0.001	<0.001	2.00	<0.001	0.017	<0.01	0.066
56	<0.001	<0.001	<0.001	<0.001	3.10	<0.001	0.025	<0.01	0.260
58	<0.001	<0.001	<0.001	<0.001	0.45	<0.001	<0.001	<0.01	0.673
62	<0.001	<0.001	<0.001	0.004	0.60	<0.001	0.013	<0.01	0.015
62**	<0.001	<0.001	<0.001	0.052	0.27	<0.001			0.041
68	<0.001	<0.001	<0.001	<0.001	1.54	<0.001	0.009	<0.01	2.20
Standard	0.05	0.010	0.05	5.0	1.0	0.002	0.15	0.05	5.0

\*Duplicate Sample

\*\*Sample from distribution system

1985

TABLE 6

CY-1985 Ground Water Surveillance Wells  
On-Site Water Quality Data

Well Number	TDS+	Chloride	Cyanide	Fluoride mg/l	Nitrogen++	Sulfate	pH	Sodium
1	490	16	<0.001	0.8	<0.05	149	7.63	31.4
1*	523	12	<0.001	0.8	<0.05	161	7.55	30.5
1**	533							
3	530	18	<0.001	0.7	<0.05	164	7.55	30.5
3**	797							
17A	537	22	<0.001	0.8	<0.05	108	7.43	25.5
20	440	2	<0.001	1.0	<0.05	38		
29	830 <sup>f</sup>	24	<0.001	0.9	<0.05	450 <sup>ff</sup>	7.58	49.0
45	440	6	<0.001	0.6	<0.05	108		
50	530	32	<0.001	0.6	<0.05	37	7.49	
52	517	22	<0.001	0.6	<0.05	109	7.64	19.1
55	443	30	<0.001	0.5	<0.05	62	7.45	19.1
56	340	2	<0.001	0.7	<0.05	<5.0	7.41	17.4
58	330	< 1	<0.001	0.7	<0.05	18	7.40	19.4
62 <sup>fff</sup>	423	8	<0.001	0.8	<0.05	69	7.42	19.8
62**	450							
68	420	8	<0.001	0.8	<0.05	76	7.64	31.4
Standard <sup>fff</sup>	500	250	0.1	1.8	10	250	6.5 to 9 <sup>∞</sup>	-

- \*Duplicate Sample
- \*\*Sample from distribution system
- +Total dissolved solids
- ++Nitrate plus nitrite
- <sup>f</sup>Treated before drinking
- <sup>ff</sup>Community Drinking Water Supply Only
- <sup>∞</sup>General Standard

TABLE 7  
CY-1985 Ground Water Surveillance Wells  
On-Site Organics Survey Data

Well Number	Ammonium	Benzene	Gasoline mg/l	Trichloro- ethane	Trichloro- ethylene	TOC*
1	0.49		<10.0	<0.001	<0.001	2.7
1**	0.81	<0.001	<10.0	<0.001	<0.001	3.2
3	<0.05	<0.001	<10.0	<0.001	<0.001	2.9
17A	0.82			<0.001	<0.001	3.3
29	1.14					
43	0.79		<10.0	<0.001	<0.001	2.6
50				<0.001	<0.001	8.0
55	0.36		<10.0	<0.001	<0.001	2.9
56	0.39		<10.0			
58	0.42					2.6
62	0.48	<0.001	<10.0	<0.001	<0.001	2.5
68	0.76					
Standard	1.5†	0.0††	---	0.2††	0.005††	---

\* Total organic carbon  
 \*\*Duplicate sample  
 † General Standard  
 ††Proposed EPA Standard

TABLE 8

CY-1985 Ground Water Surveillance Wells  
On-Site Trihalomethane Data for Chlorinated Systems

Well Number	Chloroform	Dichlorobromo- methane	Dibromochloro- methane	Bromofurm
	----- mg/l -----			
1	<0.001	<0.001	<0.001	<0.001
1*	<0.001	0.015	0.006	<0.001
62*†	<0.001	0.007	0.003	<0.001
Standard†	0.1	0.1	0.1	0.1

\*Sample from distribution system after chlorination

†Community Drinking Water Supply Only

Well #29 has a sulfate problem and ion exchange resins are used to treat the water. Several other wells are just above the standard for total dissolved solids. Well #50, which has had high coliform levels in the past, has a high total organic carbon value compared to other wells. This might indicate organic infiltration. This well is behind a former farm house. Thus, there were many potential sources of organic pollutants.

#### 3.4.2 Industrial Water Ponding Systems

There are several water systems used for cooling magnets and for fire protection:

The Industrial Cooling Water (ICW) System consists of Casey's Pond (Fig. 5) at the end of the Neutrino Beam Line and underground mains to fire hydrants and sprinkler systems throughout the Central Laboratory Area and Experimental Areas. Casey's Pond is supplied by surface drainage and can be supplied by pumping from the Fox River. The pond holds 68,000,000 l (18,000,000 gal).

The Swan Lake/Booster Pond System (Fig. 5) is used for cooling purposes at the Central Utilities Building (CUB). Water is pumped from the Booster Pond into a ditch in which it runs by way of a small West Pond into Swan Lake. The water is then returned to the Booster Pond by a return ditch. Water is also pumped from Swan Lake to NS1 Service Building (N1 in Fig. 7) for cooling purposes, from which it returns by a surface ditch. This system can be supplied water from the ICW System and it overflows into Indian Creek (Figs. 5 and 9).

The Main Ring Ponding System consists of a series of interconnecting canals completely encircling the interior of the Main Ring with a large reservoir pond inside the Main Ring Ponding System (Fig. 5). This water is used in heat exchangers at the Service Buildings for cooling the Main Ring magnets. The system is generally supplied by surface drainage, although make-up water can be pumped from Casey's Pond. The system overflows into Lake Law (Figs. 5 and 9).

The water in these systems normally meets the quality requirements of water in general use in Illinois (Section 5).

#### 3.4.3 Other Lakes and Ponds

Surface drainage from the eastern portion of the site flows into Lake Law, DUSAF Pond and the AE Sea (Figs. 5 and 9). The chlorinated effluent from the Village sewage treatment plant oxidation pond (just north of DUSAF Pond) also flows into DUSAF Pond. These lakes and ponds are accessible to the public, and they are the head waters of Ferry Creek.

Semiannual tests are made of water samples taken where the three creeks leave the site (R1, R2A, and R3 in Fig. 7), as well as from Casey's Pond and the Fox River. Results of the tests for pH, dissolved oxygen, five-day biochemical oxygen demand, (BOD5), suspended solids, and fecal coliform for 1985 are found in Table 9. Tests for fecal coliform bacteria are made monthly in our water laboratory. Levels above 200 were found in Kress Creek and Ferry Creek the last few months of the year. The



TABLE 9  
Site Wide Water Quality Report for CY-1985

	pH		DO mg/ℓ		BOD5 mg/ℓ		Susp. Solids mg/ℓ		Fecal Coliform mg/ℓ	
	April	Sept.	April	Sept.	April	Sept.	April	Sept.	April	Sept.
Ferry Creek	8.2	7.9	10.6	8.4	7.3	7.8	54	97	52	**
Kress Creek	8.1	7.8	11.4	9.2	2.1	0.1	2	65	30	**
Indian Creek	8.1	7.8	10.8	7.2	1.8	2.9	16	15	42	52
Casey's Pond	8.0	8.4	8.7	9.4	2.1	5.0	9	53	2	20
Fox River	8.4	8.4	9.8	8.3	5.0	8.5	26	50	48	68
General 28 Standards	6.9 - 9.0		Not less than 5.0 at any time		*		*		Mean of 200	

\*There are standards for effluent from treatment works or waste water sources, but no general standards.

\*\*Too numerous to count individually.

TABLE 10

Village Sewage Treatment Plant

Monthly Averages Report for CY-1985

Parameter	Permit Limit	Jan	Feb	Mar	Apr	May	Jun	Jul	Aug	Sept	Oct	Nov	Dec
pH	6-9	7.6	7.0	8.1	8.6	7.8	6.8	7.6	7.6	7.5	8.9	8.0	8.0
BOD5 mg/ℓ	10	9.4	2.2	10.0	16.5*	14.2*	29*	20.8*	12.8*	30.0*	6.0	58*	7.0
Suspended Solids mg/ℓ	12	12	17*	32*	17*	34*	17*	32*	25*	32*	45*	28*	15*
Fecal Col. #/100 mℓ	400	0	**	0	0	0	0	0	0	0	0	0	0

\*Violation Report filed

\*\*Too numerous to count, but chlorine residual was 0.3 ppm when sample was taken. Therefore, the coliform test result is questionable.

explanations for the high readings have not been found. Fecal coliform bacteria are found in recent deposits of fecal material from warm-blooded animals. They serve as an indicator for pathogens which can multiply under similar conditions. Thus, the Laboratory will continue to check the levels, search for the sources of nutrients, and look for any impacts.

#### 3.4.4 Sewage Treatment

An authorization permit to discharge under the National Pollutant Discharge Elimination System (NPDES) was obtained for the Village Oxidation Pond (just north of DUSAF Pond in Fig. 5) in 1979.<sup>27</sup> It recently was renewed and will expire on September 1, 1990 (See Section 3.5). Monthly testing results for 1985 are in Table 10.

The Main Site sewer system was connected to the City of Batavia system June 26, 1979 and has been delivering sewage to the Batavia sewage treatment plant since that time.

The NPDES permit for the Village sewer system granted in 1985 has limits for 30 days average BOD5 and suspended solids of 10 mg/l and 12 mg/l, respectively. The Village system exceeded the limit for suspended solids 11 times in CY-1985, in spite of treatment with Aquazine to control algae. The limit for BOD5 was exceeded seven times. See Table 10. These results are reported to the U. S. Environmental Protection Agency quarterly and to the Illinois Environmental Protection Agency semiannually. Fermilab plans to connect to the City of Warrenville sewage collection system to

achieve compliance with regulations.

### 3.4.5 Chemical Treatment of Water Systems

Some chemical treatment of our various water systems is required each year to control the growth of algae and aquatic weeds. Only EPA registered agents are administered by trained personnel following the manufacturer's directions.

#### 3.4.5.1 Dalapon

No Dalapon was applied in CY-1985.

#### 3.4.5.2 Chlorine

In addition to the routine chlorination of the domestic water systems, the swimming pool and the Village Oxidation Pond, a chlorination system for the Swan Lake cooling pond system has proved successful in helping to eliminate the need for chromate treatment of the cooling towers. Chlorine is added to the cooling water for a period of 30 minutes four times a day at a rate which results in a chlorine concentration of 0.5 ppm as the cooling water leaves the equipment. Only one piece of equipment within the plant is chlorinated at a time. Consequently the concentration of chlorine entering the Swan Lake system is significantly reduced from 0.5 ppm.

### 3.4.5.3 Aquazine

The Village Oxidation Pond was treated three times in CY-1985 in an attempt to control algae growth and reduce suspended solids. The total quantity of Aquazine used was 204 kg (450 lbs). Aquazine was also used to treat the Main Ring Ponding System the Swan Lake/Booster Pond System and the reflecting ponds by Wilson Hall in CY-1985 (Fig. 5). Approximately 540 kg (1200 lbs) of Aquazine was applied to the Main Ring Ponding System, 284 kg (628 lbs) was applied to the Swan Lake/Booster Pond System, and 43 kg (96 lbs) was applied to the reflecting ponds.

### 3.4.5.4 Heavy Metals and Other Toxic Materials

The continued success of the Swan Lake cooling pond system again made it possible to eliminate the use of chromates in 1985. Although it was necessary to use the cooling towers during the warm summer months, it was not necessary to treat the towers with chromate compounds. The chlorinated Swan Lake cooling pond water was passed through the cooling system and a biodispersant, Nalco 7348, was added which lifted deposits from the metal surfaces so they could be oxidized by the chlorine. The rate of application was 3.6 kg (8 lbs) per day with a peak concentration of 20 mg/l. Nalco 7348 is a polyglycol manufactured by Nalco Chemical Company, 2901 Butterfield Road, Oak Brook, Illinois 60521. Another Nalco product, Nalco 7387, was applied continuously to maintain less than 1 mg/l with a peak total phosphorus concentration of 1.3 mg/l. The rate of application was the same per day as for Nalco 7348. Nalco 7387 is an organophosphorus

1985

compound which prevents scale formation. It does not have the toxic properties of organic phosphorus esters found in some restricted-use pesticides.<sup>29</sup>

One ethylene glycol leak occurred from an underground pipe in a closed-loop water system in CY-1985. Approximately 190 l (500 gal) of a 50% solution spilled into the soil surrounding the pipe near the PS-5 Service Building (R4 in Fig. 7). The bad section of pipe was replaced. Ethylene glycol is biodegradable in contact with soil.

Trace amounts of heavy metals and copious quantities of sodium chloride are discharged into the CUB Tile Field (clay tile field in Fig. 6) inside the Main Ring. Copper is the primary heavy metal. It is an impurity removed by the ion exchange resins used to keep the conductivity of closed loop water systems low. These mixed-bed resins are regenerated using hydrochloric acid and sodium hydroxide. When the two chemicals combine after traversing the resins, salt is formed. Trace amounts of <sup>7</sup>Be are also removed (Section 3.3.3.2.2).

In CY-1985 a coring was made in the tile field to a depth of 1.5 m (5 ft). The samples were analyzed for zinc, copper, lead, chloride, sulfate, and total chromium. The results were compared with results from a coring made in CY-1982. Zinc concentration decreased from a peak of 650 mg/kg in 1982 to only 51 mg/kg in 1985. Copper decreased from 62 to 50 mg/kg. Lead decreased from 20 to less than 10 mg/kg, sulfate decreased from 440 to 38 mg/kg, and total chromium remained essentially the same at 22 to 26 mg/kg.

The only parameter to increase was chloride which peaked at 1100 mg/kg compared to 410 mg/kg in CY-1982. The peak occurred in the sample starting 1 m (39") below the surface. The clay tile is not more than 0.6 m (2 ft) below the surface. Thus, chloride is moving downward away from the surface. The impact is expected to be negligible (Section 3.9.3.5).

### 3.5 Environmental Permits

As mentioned in Section 3.4.4 Fermilab has a NPDES permit (ILO025941) for discharge of sewage from the Village Oxidation Pond. This permit expires September 1, 1990. The State of Illinois Environmental Protection Agency (IEPA) and the U.S. Environmental Protection Agency (EPA) were notified of all permit violations in CY-1985. Following a November 5, 1985 inspection of the pond, the IEPA indicated that these excursions require actions to come into compliance. The Laboratory has chosen to pursue connection to the City of Warrenville sewage collection system. The latter delivers sewage to a City of Naperville sewage treatment plant which has ample capacity to treat the additional sewage. The noncompliance has had negligible environmental impact on the on-site ponds (Fig. 5) or off the site.

The magnet debonding oven (Section 3.3.2) has an Illinois Environmental Protection Agency permit (I.D. No. 089801AAD D/O-1) which expires May 7, 1989. There have been no cases of noncompliance.

Fermilab has an interim permit (USEPA I.D. No. IL6890030046) to operate a hazardous waste storage facility. This permit was issued by the U.S. Environmental Protection Agency and will expire November 1, 1988. The facility is in compliance with regulations. Regulated chemical wastes are stored in the facility. Examples are hazardous wastes, polychlorinated biphenyls (PCBs), and used oil. Only wastes generated by Fermilab are stored at the facility for proper disposal elsewhere in the future.

Emco Wheaton coaxial vapor recovery systems have been installed on all gasoline dispensing equipment at Fermilab under a permit (Identification Number 089801-AAD) issued by the Illinois Environmental Protection Agency. The permit expires on February 13, 1991. There have been no compliance problems with the systems.

### 3.6 Assessments and Impact Statements

No formal environmental assessments and no environmental impact statements were prepared in either draft or final form during the year at Fermilab. However, there is an on-going program in place to routinely evaluate new projects and modifications to existing operations and facilities to determine if there is a significant potential for impact. Also, see Section 3.9 for the evaluation of operations conducted in CY-1985.



3.7 Summary of Significant Environmental Activities

In the early 1970's Fermilab began a prairie restoration project on the approximately 1.6 km<sup>2</sup> (400 acre) plot inside the main accelerator (Fig. 4). Three endangered species of prairie plants (Cypripedium Candidum, Iliamna remota, and Petalostemum foliosum) and one threatened plant (Filipendula rubra) have been introduced into the plot. Also, in CY-1984 some Fermilab land (0.11 km<sup>2</sup> or 28 acres) outside this plot has been plowed and seeded with prairie plants (Fig. 5). There are very few remnants of the original Illinois prairie left. The Fermilab restoration is one of the largest prairie sites in the country. The harvesting of seeds is done by volunteers and the environmental aspects receive the attention of a prairie committee consisting of laboratory personnel and outside university representatives. Fermilab conducts routine burning of the prairie restoration areas with assistance from the prairie committee and volunteers.

In CY-1985 a new building was completed for storing hazardous waste. This is the first heated building at the Hazardous Waste Storage Facility (Site 55 in Fig. 5). Wastes which freeze are now stored in this building. Also, drums of volatile solvents and gases which could overpressure in the hot summer sun are stored inside. The building is equipped with a hood for processing small quantities of special chemicals and separate containment areas for acids and bases.

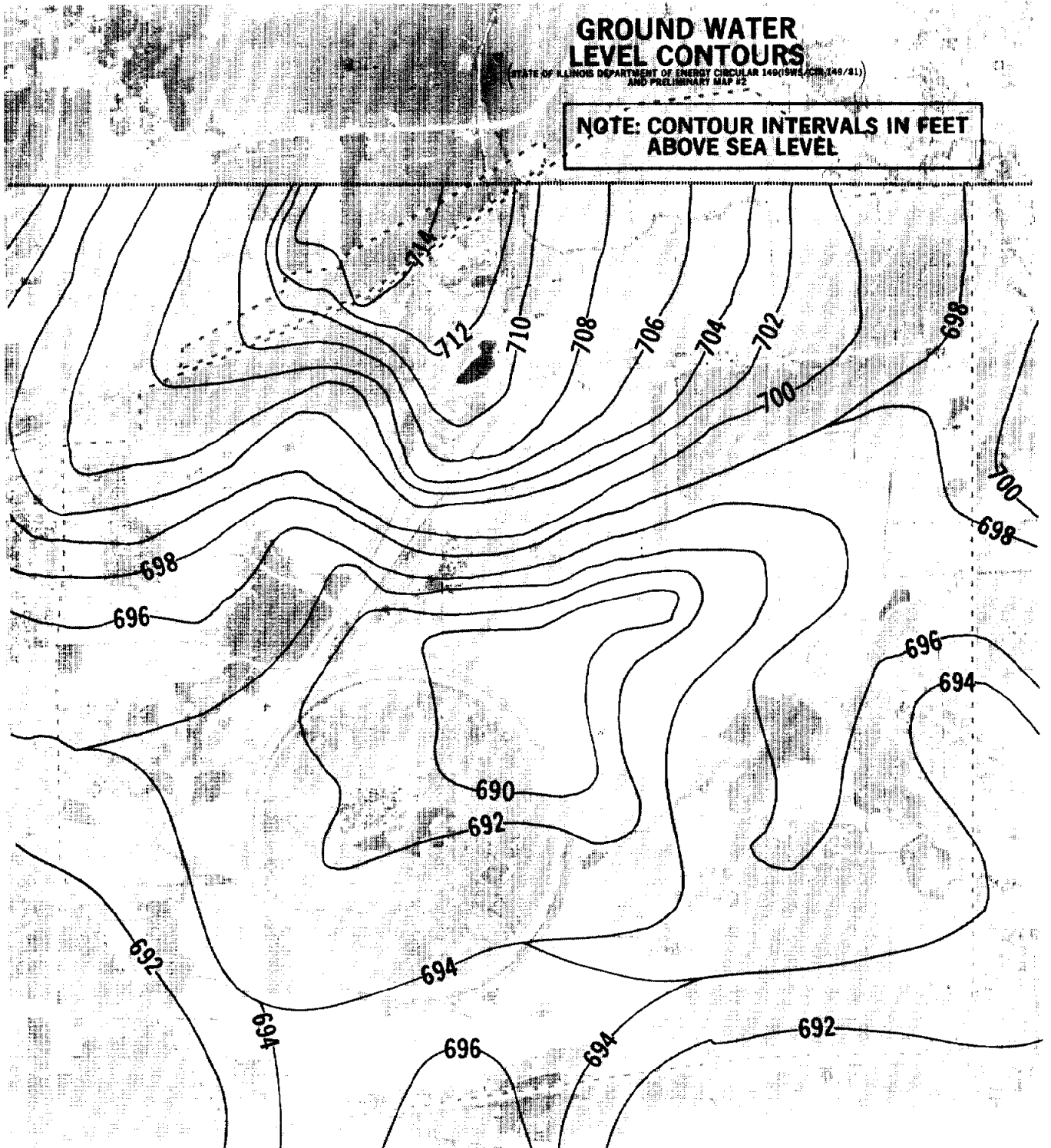
On January 24, 1985 a large transformer failed, cracking its case and spilling a small fraction of its oil on the transformer pad. Quick emergency response prevented an adverse environmental impact. As a result of that experience, it was determined that there was need for additional equipment and training. A second spill control trailer was purchased and equipped with appropriate control materials. Also, additional spill response personnel were trained.

Most of the oil spilled was contained on the transformer pad and pumped into drums for later disposal by incineration. A small quantity (less than 380 l or 100 gal) entered the gravel containment area surrounding the transformer pad. Subsequently when the water was released from the containment area, a thin film of oil was seen in the ditch adjacent to the Master Substation (near 4 in Fig. 4).

### 3.8 Summary of Hydrogeology

The Fermilab site has thick glacial till consisting primarily of low permeability clay.<sup>30</sup> This clay forms a barrier to the downward percolation of any water containing radioactivity. Beneath the clay the first layer of rock is a dolomite of Silurian age.<sup>4</sup> Its fractured upper 3 m (10 ft) carries sufficient water for individual farm needs. The water level contours for this aquifer are shown in Fig. 11. Note that the water from the Experimental Areas flows toward Well 1, the primary on-site drinking water supply (Fig. 4). Groundwater leaves the site and flows southwest toward the Fox River and southeast toward the West Branch of the DuPage River.

Figure 11



OVERLAY FOR BASE 4, FERMILAB

PREPARED IN 1982

FERMI NATIONAL  
ACCELERATOR LABORATORY

FIG. 11

1985

Beneath the silurian dolomite are older formations laid down by sedimentation during the Cambrian and Ordovician periods when the region was under sea water. These consist of dolomite and sandstone with perhaps some shale. The sandstone aquifer at approximately 400 m (1333 ft) below the surface provides sufficient volumes of water for local municipal water supplies.

The spillways should be noted as shown on the map of surface water drainage (Fig. 9). In the event of an accidental spill, backup efforts will be concentrated at those points to stop the flow of any hazardous substance if it cannot be contained closer to the discharge point.

### 3.9 Evaluation of Environmental Impacts

#### 3.9.1 Assessments of Potential Radiation Dose to the Public

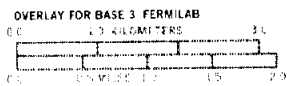
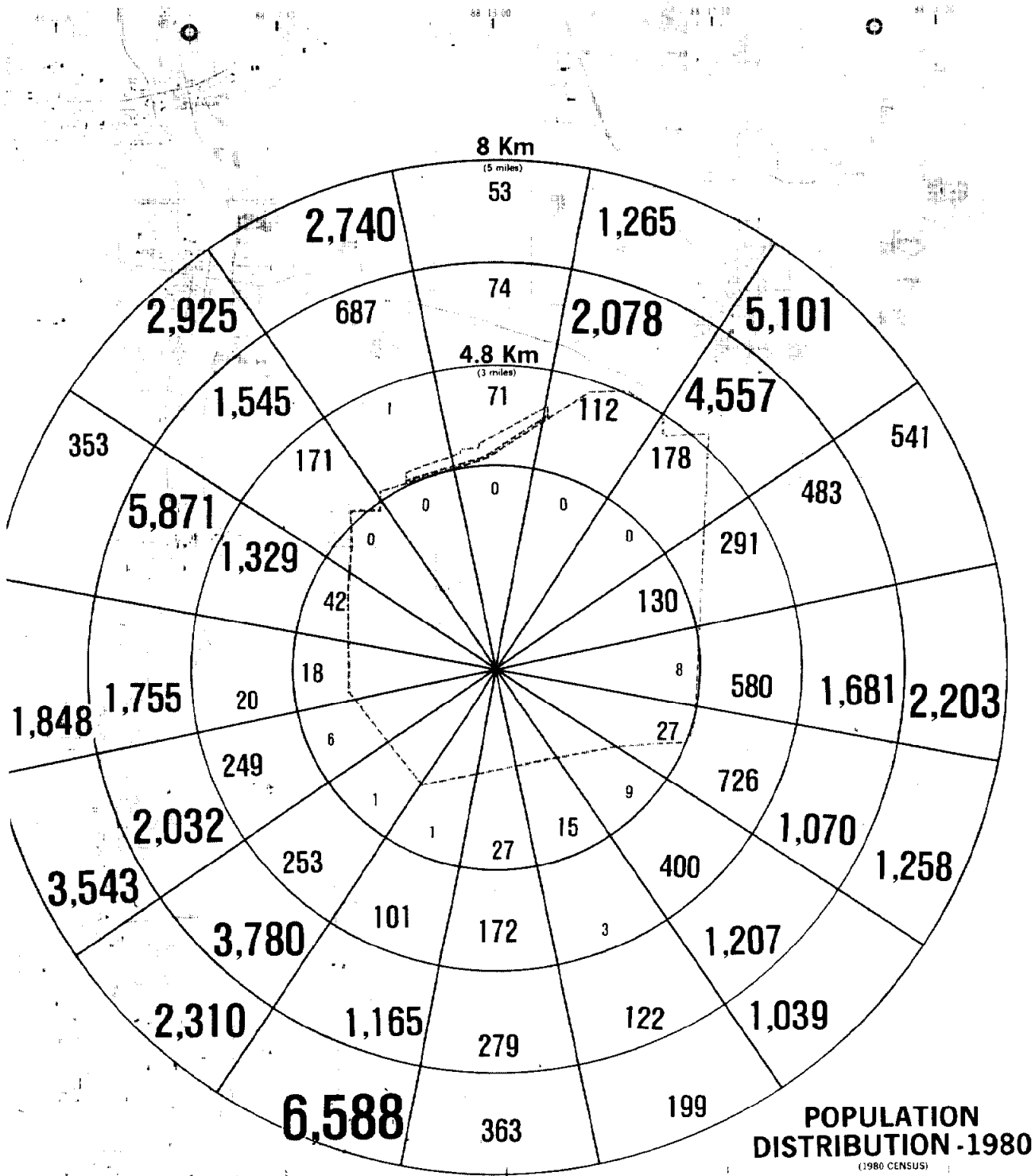
Fermi National Accelerator Laboratory is located in the densely populated Chicago Area. There are about eight million people living within 80 km (50 mi) of the site (Fig. 3).<sup>31</sup> There are 326,645 people within 16 km (10 mi) of the center of the main accelerator based on the 1980 census results compared to 265,677 counted in the 1970 census.<sup>32</sup> The detailed distribution of population as a function of distance and direction from Fermilab is given in Table 11.<sup>31</sup> The population distribution close to Fermilab is shown in Figure 12. The estimated increase in population from 1980 to 1985 is 9% within 16 km (10 mi) of Fermilab based on county and local city population estimates.<sup>1,2</sup>

TABLE 11

## Incremental Population Data in Vicinity of Fermilab, 1980

DISTANCE, KILOMETERS FROM CENTER OF MAIN RING	LATITUDE = 41.832					LONGITUDE = 88.251				
	0-8	8-16	16-32	32-48	48-64	64-80	80-97	97-113	113-128	
DISTANCE, MILES	0-5	5-10	10-20	20-30	30-40	40-50	50-60	60-70	70-80	
DIRECTION										
N	198	1110	77247	75658	63188	37183	30696	28459	149892	
NNE	3455	5821	68274	76075	120930	145415	100858	173092	87495	
NE	9836	12718	78701	292724	139718	0	0	0	0	
ENE	1445	63784	263526	840460	551913	0	0	0	0	
E	4472	18423	218631	1107254	924752	0	0	33317	56442	
ESE	3081	15075	92242	268040	597113	379986	196888	78056	17600	
SE	2655	25167	37956	34405	106938	38944	24651	11963	10027	
SSE	339	3262	44203	148699	7962	21154	70503	10828	13195	
S	841	1336	8604	10301	17011	11089	6640	4354	11967	
SSW	7055	49656	8635	3492	17420	6373	25217	24588	10469	
SW	6344	35851	13598	15566	5317	30917	36362	13671	13226	
WSW	5030	2205	5578	6322	4509	10930	8474	11704	12175	
W	3641	971	2941	5339	5111	13693	8445	28768	49103	
WNW	7595	851	3018	42762	6723	21231	40449	13891	37012	
NW	4641	9607	3297	7974	7358	65288	157549	71682	28229	
NNW	3428	15152	22722	10674	29830	17952	29399	24276	58430	
TOTAL	65656	260989	949171	2945745	2605793	800155	736131	528649	555262	
CUMULATIVE TOTAL	65656	326645	1275816	4221561	6827354	7627509	8363640	8892289	9447551	

Figure 12



FERMI NATIONAL ACCELERATOR  
LABORATORY AND VICINITY

STANLEY N. BRILLIANT  
PHYSICIAN  
LABORATORY

PREPARED IN 1982  
FOR THE  
DOE  
OFFICE OF ENERGY RESEARCH AND DEVELOPMENT

1215

The dose rate at the site boundary from Fermilab operations was primarily from muons. Earlier measurements of muons showed they went in one direction (toward the northeast). The maximum annual dose rate at the site boundary and the annual dose to the off-site individual receiving the maximum dose from Fermilab operations are essentially the same. This is because the change in distance is small to the site boundary to off-site housing compared to the change in dose rate with distance. The total dose to the individual is 1.5 mrem for CY-1985. The point where that exposure occurred is along the path muons traveled which originated in the Meson Area. This is approximately 1.2 percent of the background radiation dose.<sup>31</sup>

The radiation exposure to the general population from operation of Fermilab in CY-1985 was 1.2 person-rem. This exposure was from muons and airborne radioactivity. This is to be compared with a total of approximately one million person-rem to the population within 80 km (50 mi) from natural background radioactivity.<sup>32,33</sup> Based on typical United States radiation exposures from diagnostic x-rays, medical treatments, and other artificial sources an additional 500,000 person-rem would be expected for the population in the Chicago area with 80 km (50 mi) of Fermilab in CY-1985.<sup>33</sup>

The exposure from muons was determined by starting with the dose to the maximum individual near the site boundary and calculating dose versus distance from the point on-site where the penetrating radiation (Section 3.3.1) originated to 80 km (50 mi) from the site using the inverse

square of the distance and summing over the appropriate numbers of individuals. The dose was received by individuals living only in portions of the north-northeast and northeast sectors. See Table 12.

The magnet debonding oven was used to debond 14 radioactive magnets in CY-1985. The resulting  $^3\text{H}$  release from the debonding oven stack had negligible impact.

TABLE 12

Summary of Population Exposures for CY-1985  
Within an 80 km (50 mi) Radius of Fermilab

<u>Source</u>	<u>Contributions to Population Exposures (person-rem)</u>
Penetrating Radiation from Proton Area	0.5
Penetrating Radiation from Meson Area	0.6
Airborne Radioactivity from All Areas	<u>0.1</u>
TOTAL	1.2

Several of the closed loop cooling systems were drained during the extended shutdown in CY-1982-3. These were at levels where potential off-site releases, from these loops, would be detectable but not hazardous. The tritiated water was evaporated.<sup>4</sup> Some releases of radioactive water occurred from sumps collecting water from under areas where protons interacted. About 17% of this volume of water left the site while Casey's Pond (Fig. 4), the reservoir receiving water from discharges in the three



external areas to which protons are delivered, was full. The mean concentration of tritium during the period of release was less than one percent of the Concentration Guide for prolonged exposure to the general population. Also, drinking water in the area is taken from wells rather than from the creek receiving the discharge. Hence, the dose from the release is negligible.

The D0 abort system was replaced by a new abort system at C0 in CY-1983, featuring a well-shielded dump (at C1 in Fig. 6). Therefore, no additional soil activation is expected near D0 in the future. The soil was sampled near D0 in CY-1983.<sup>5</sup> Based on those sampling results, no environmental impact is expected from the D0 abort system. The soil activation associated with the new system at C0 should be much lower than at D0. Some  $^3\text{H}$  and  $^{22}\text{Na}$  has been detected in the underdrain beneath the new abort dump. The amount of radioactivity is consistent with the amount expected from design calculations. See Section 3.3.3.4.

A 45° hole was drilled and the soil was sampled beneath the primary target in the Neutrino Area in CY-1984. No evidence was found for movement of radionuclides toward the aquifer (Section 3.3.3.4).

### 3.9.2 Assessment of Non-radioactive Pollutant Releases

Although it was necessary to chemically treat some waters with aquazine to control the growth of algae and weeds during CY-1985, efforts were made to keep these treatments as low as possible in order to protect

wildlife and fish. Minimum amounts were used and the aquazine is biodegradable. No environmental impact was expected. There is a program to look for persistent chemicals in the Fermilab environment periodically.

There were no activities during CY-1985 which created problems with respect to non-radioactive airborne effluents. Heating is accomplished by use of natural gas, liquefied propane gas, or electricity. The bulk of the heating is supplied by natural gas fired boilers located in the Central Utilities Building. These boilers are adjusted annually to maintain proper combustion efficiency.

### 3.9.3 Potential Impact of Other Toxic Substances

#### 3.9.3.1 Pesticides

In addition to the water treatments mentioned in Section 3.6.3, the following EPA registered herbicides, insecticides and rodenticides were applied by trained personnel following the manufacturer's instructions:

Chipco Turf Kleen, containing 16.1% 2,4-D and 16.2% 2-(2-methyl-4 chlorophenoxy) proponic acid, was applied to turf areas around Wilson Hall, the east reflecting pond and Swan Lake (Fig. 5) to control broad leaf weeds. Approximately 15 l (4 gal) was applied to 1.2 hectares (3 acres) in CY-1985.

Corn was planted by licensees in CY-1985 on 9.2 km<sup>2</sup> (2268 acres). Licensees are persons who pay the Laboratory for use of a portion of the land on the site for agricultural purposes. Herbicides were applied as follows: 5300 l (1400 gal) of Lasso and 2050 kg (4500 lb) of Aatrex 4L. The application of these was supervised by Fermilab.

Spike 80-W, EPA Registration No. 1471-97, was applied to control weeds around electrical substations, parking lots, hardstand (crushed limestone) areas, air conditioner pads, and service buildings. Approximately 124 kg (264 lb) diluted in 17,600 l (4700 gal) of water was applied to 0.14 km<sup>2</sup> (41 acres) during CY-1985.

For mosquito control, an ultra low volume application of CYTHION Premium Grade Malathion was performed at five different times. Approximately 11 l (2.8 gal) of CYTHION were used in each application and the following areas were covered: Village and Sauk Circle just south of the Village (Fig. 1), Sites 29, 38, and 58 (29, 38, and south of 55 in Fig. 4), and the Meson, Proton and Neutrino experimental areas (Fig. 5).

Confrac Rat and Mouse Bait, EPA #12455-36, a rodenticide, was placed in pan-type feeders inside approximately 40 outdoor electrical substations to reduce rodent nesting in this high voltage equipment. Approximately 5.4 kg (12 lbs) was used in CY-1985.

The services of a contract exterminator, licensed by the State of Illinois and using EPA registered pesticides, was retained during CY-1985 for the control of miscellaneous pests found in kitchens, laboratories and living areas throughout the site.

### 3.9.3.2 Polychlorinated Biphenyls

An inventory of polychlorinated biphenyls (PCBs) is maintained. PCB inspections are performed and reports made to the U.S. Environmental Protection Agency as called for in the regulations.<sup>34</sup> The PCB status Report as of January 1, 1985 listed 60 PCB transformers and 2,017 large capacitors in use or storage for future use. These PCB items have been labeled as required. These totals differ from last year's totals because one PCB transformer was properly disposed of, and three operational transformers were sent to Stanford Linear Accelerator Center, and a number of PCB capacitors were disposed of by incineration in an EPA - approved incinerator.<sup>4</sup> The large transformer that failed in CY-1985 (Section 3.7) had only 1.3 ppm PCB in its oil.

### 3.9.3.3 Hazardous Wastes

Significant progress was made during 1985 with respect to identification, collection and disposal of hazardous waste in an environmentally acceptable manner. Responsibility for this program was assigned to the Safety Section in CY-1979 and a hazardous waste handling and storage facility was developed at Site 55 (Fig. 5). This facility is

roofed and fenced, and has a hardstand and two concrete containment areas. An additional facility with concrete containment area for PCBs was developed at Site 3 where the Environmental Monitoring Station is located (Fig. 5). This facility is for inside storage of hazardous materials which are for future use. In CY-1982 a PCB storage building was constructed at Site 55 which is much farther from the site boundary than Site 3. Off-site impact from a potential airborne release of PCBs was greatly reduced when most of the PCB items were removed from Site 3. In CY-1984 a heated chemical waste storage building was added at Site 55. This facility was completed in FY-1985 and has a hood and an indoor shower and eye wash. It also has indoor containment areas to segregate acids and bases. Typical wastes are solvents, oils, laboratory chemicals, asbestos, acids, and bases.

Over the years it has been the practice to deposit excess materials such as lumber, concrete, building materials and earth on the Meson Area shielding hill (north of the Meson Lab, Fig. 1) to scatter muons and provide some additional shielding. To assure that none of these materials are hazardous to the environment and none will contribute to the contamination of surface or ground waters, a program to control such deposition was developed during 1979. Rules have been promulgated and responsibility for access and control has been assigned to the Roads and Grounds group. The Safety Section monitors this program. In 1982 burial of wood, paper and other wood products on the hill was halted.

#### 3.9.3.4 Heavy Metals

Copper sulfate is no longer used to treat the ponding systems. There was no evidence of any further impact from the treatment in CY-1981.<sup>36</sup> Copper solution from the etching of printed circuit boards was disposed of as hazardous waste or recycled. Chromate treatment of the cooling towers has been replaced by biodegradable treatments. Only trace amounts of copper were released in the CUB Tile Field. The copper came from the regeneration of resins used on copper closed loop water systems. Thus, the environmental impact from heavy metals released in CY-1985 should be negligible.

#### 3.9.3.5 Chlorides

The potential environmental impact of release of chlorides from the resin regeneration process into the CUB clay tile field (Fig. 6) has been evaluated. The process uses sodium hydroxide and hydrochloric acid, yielding sodium chloride (salt) and water. Assuming the salt released in one year all ends up in the nearest drinking water well (W1 in Fig. 4) and is diluted in the water normally pumped from that well for one year, the concentration would be less than 25% of the applicable limit of 250 mg/l. See Section 5. Thus, the environmental impact should be minimal. A similar result was found for the impact from salt applied to Fermilab roads in the winter.

4. Quality Assurance in CY-1985

4.1 Quality Control

Water samples collected in CY-1985 were analyzed by Teledyne Isotopes, Inc., 1500 Frontage Road, Northbrook, Illinois 60062. In addition, such samples were counted at the Fermilab Nuclear Counting Laboratory. Tritium and  $^{45}\text{Ca}$  analyses were done only by Teledyne Isotopes, Inc. since Fermilab does not have the necessary liquid scintillation counting system. Each shipment to Teledyne included at least one sample prepared at Fermilab containing known amounts of several of the accelerator-produced radionuclides. Known concentrations of tritium were included in every shipment.

4.1.1 Analytical Procedures at Teledyne

Teledyne Isotopes, Inc. analyzes water samples using essentially the same procedures as described previously.<sup>35</sup> Liquid scintillation counting is done using a Beckman Instruments Inc. LS-230 refrigerated system. A 1 ml aliquot of the sample is placed in 10 or 15 ml of the scintillator "InstaGel," manufactured by Packard Instrument Co., Inc., 2200 Warrenville Rd., Downers Grove, IL. 60515.

The samples were subjected to the appropriate one of the following analyses:

Type 1a: Test for  $^3\text{H}$  (tritium),  $^7\text{Be}$ ,  $^{22}\text{Na}$ ,  $^{45}\text{Ca}$ ,  $^{54}\text{Mn}$ , and  $^{60}\text{Co}$  at surface water sensitivities. See Table 13.

TABLE 13  
Specifications for the Analyses of  
Accelerator-Produced Radionuclides in Water

Radio-nuclide	CONCENTRATION GUIDE FOR POPULATION (µCi/ml)			SPECIFIED SENSITIVITY AND PRECISION* (µCi/ml)	
	Occasional Exposure	Prolonged Period of Exposure	Community Water System	Surface Water	Ground Water
<sup>3</sup> H	1 x 10 <sup>-2</sup>	2 x 10 <sup>-3</sup>	2 x 10 <sup>-5</sup>	3 x 10 <sup>-6</sup>	1 x 10 <sup>-6</sup>
<sup>7</sup> Be	5 x 10 <sup>-3</sup>	1 x 10 <sup>-3</sup>	4 x 10 <sup>-5</sup>	5 x 10 <sup>-7</sup>	5 x 10 <sup>-7</sup>
<sup>22</sup> Na	5 x 10 <sup>-5</sup>	1 x 10 <sup>-5</sup>	4 x 10 <sup>-7</sup>	3 x 10 <sup>-7</sup>	2 x 10 <sup>-8</sup>
<sup>45</sup> Ca	3 x 10 <sup>-4</sup>	6 x 10 <sup>-5</sup>	2.4 x 10 <sup>-6</sup>	3 x 10 <sup>-7</sup>	6 x 10 <sup>-9</sup>
<sup>54</sup> Mn	2.5 x 10 <sup>-4</sup>	5 x 10 <sup>-5</sup>	2 x 10 <sup>-6</sup>	1 x 10 <sup>-7</sup>	7 x 10 <sup>-8</sup>
<sup>60</sup> Co	2.5 x 10 <sup>-5</sup>	5 x 10 <sup>-6</sup>	2 x 10 <sup>-7</sup>	1 x 10 <sup>-7</sup>	2 x 10 <sup>-8</sup>

\*: The precision and sensitivity are stated for the 68% confidence level (one standard deviation). The precision required is the value specified or ±10 percent, whichever is the lesser precision. The sensitivity is taken to be the minimum concentration which can be detected within the 68 percent confidence level.



Type 2a: Test for all of the above at ground water sensitivity plus total radium (the sum of  $^{223}\text{Ra}$ ,  $^{224}\text{Ra}$ , and  $^{226}\text{Ra}$ ) and total thorium (the sum of  $^{228}\text{Th}$  and  $^{232}\text{Th}$ ).

Type 3a: Chemical separation of  $^{45}\text{Ca}$  before its determination; otherwise the same as Type 1a.

Type 4a:  $^3\text{H}$  only, at surface water sensitivity.

Type 5a: Chemical separation of  $^{45}\text{Ca}$  and analysis for  $^{45}\text{Ca}$  only, using surface water sensitivity.

Type 6a: The same as Type 1a except at ground water sensitivity.

Type 7a: The same as Type 4a except at ground water sensitivity following distillation.

Type 8a: Test for gross alpha, gross beta,  $^3\text{H}$ ,  $^{131}\text{I}$ , and  $^{134}\text{Cs}$  at groundwater sensitivity. This analysis is performed on Fermilab's one community water system and on other drinking water systems on-site which supply water to more than 25 people during the work day.

Type 9a: Test for Sr-90 only, at groundwater sensitivity.

Separate analyses of two aliquots from the same sample bottle is indicated by changing the letter "a" to the letter "b" on the Type designation.

The specifications for the above analyses are given in Table 8.

#### 4.1.2 Quality Assurance Samples

During CY-1985 Fermilab participated in the DOE Environmental Measurements Laboratory (EML) quality assurance program.<sup>36</sup> Results are given in Table 14. Also, Fermilab sent quality assurance samples monthly to Teledyne Isotopes, Inc., who analyzed Fermilab water samples independently. The vendor's agreement with the known concentrations was close to the precision specified by Fermilab (Table 13) except for <sup>45</sup>Ca, which is difficult to assay in the presence of other beta-emitting radionuclides. See Tables 15 & 16.

TABLE 14

Quality Assurance Results for Fermilab

<u>Sample Date</u>	<u>Radio-nuclide</u>	<u>Percentage of Concentration Guide for Surface Waters* (%)</u>	<u>Prepared Concentration</u>	<u>Ratio of Fermilab Result to Prepared Concentration</u>
5/85	<sup>7</sup> Be	†	5.72 x 10 <sup>-3</sup> μCi	0.90
				0.89
	<sup>54</sup> Mn	†	4.50 x 10 <sup>-4</sup> μCi	0.88
				0.87
	<sup>60</sup> Co	†	5.00 x 10 <sup>-4</sup> μCi	0.92
				0.91
	<sup>54</sup> Mn	1.4	3.42 x 10 <sup>-6</sup> μCi/ml	1.02
<sup>60</sup> Co	**	1.09 pCi/g	0.91	
	<sup>60</sup> Co	19.6	4.91 x 10 <sup>-6</sup> μCi/ml	1.07
11/85	<sup>7</sup> Be	†	4.42 x 10 <sup>-3</sup> μCi	0.93
				1.00
	<sup>54</sup> Mn	†	4.80 x 10 <sup>-4</sup> μCi	0.94
				0.99
	<sup>60</sup> Co	†	4.54 x 10 <sup>-4</sup> μCi	0.95
				1.01
	<sup>54</sup> Mn	1.8	4.43 x 10 <sup>-6</sup> μCi/ml	0.99
<sup>60</sup> Co	19.3	4.82 x 10 <sup>-6</sup> μCi/ml	1.07	

\*Occasional Exposure (Table 13)

\*\*Vegetation Sample

†Air Filter: Two Filters Provided

4.2 Additional Quality Assurance Efforts

The scope of the environmental protection program at Fermilab has broadened over the years. The Laboratory has doubled in personnel from the number employed when the first proton beam was extracted from the main accelerator in 1972. Regulations have changed. Much more emphasis is now

TABLE 15

Quality Assurance Results for Teledyne

First Half of CY-1985

Sample Number	Radio-nuclide	Percentage of Concentration Guide for On-Site Surface Waters* %	Prepared Concentration (μCi/ml)	Ratio of Teledyne Result to Prepared Concentration
8501	<sup>3</sup> H	0.06	$5.7 \times 10^{-6}$	1.25
	<sup>7</sup> Be	0.05	$2.5 \times 10^{-6}$	0.84
	<sup>22</sup> Na	3.8	$1.9 \times 10^{-6}$	1.12
	<sup>45</sup> Ca	0.7	$2.1 \times 10^{-6}$	0.45
	<sup>54</sup> Mn	0.7	$1.8 \times 10^{-6}$	0.98
	<sup>60</sup> Co	15.2	$3.8 \times 10^{-6}$	0.94
8502	<sup>3</sup> H	0.06	$5.7 \times 10^{-6}$	1.02
	<sup>7</sup> Be	0.03	$1.3 \times 10^{-6}$	**
	<sup>22</sup> Na	3.8	$1.9 \times 10^{-6}$	1.05
	<sup>45</sup> Ca	0.6	$1.8 \times 10^{-6}$	3.73+
	<sup>54</sup> Mn	1.6	$4.0 \times 10^{-6}$	1.0
	<sup>60</sup> Co	36.8	$9.2 \times 10^{-6}$	1.08
8503	<sup>3</sup> H	0.03	$2.9 \times 10^{-6}$	1.4
8504	<sup>3</sup> H	1.8	$1.8 \times 10^{-4}$	1.03
	<sup>7</sup> Be	0.08	$3.9 \times 10^{-6}$	0.95
	<sup>60</sup> Co	14.8	$3.7 \times 10^{-6}$	1.03
EML 8505	<sup>3</sup> H	0.2	$1.9 \times 10^{-5}$	1.05
	<sup>54</sup> Mn	1.4	$3.4 \times 10^{-6}$	1.22
	<sup>60</sup> Co	19.6	$4.9 \times 10^{-6}$	1.07
8505	<sup>45</sup> Ca	1.0	$2.9 \times 10^{-6}$	0.37
	<sup>60</sup> Co	36.0	$9.0 \times 10^{-6}$	1.07
8506	<sup>3</sup> H	3.5	$3.5 \times 10^{-4}$	0.93
8507	<sup>3</sup> H	9.1	$9.1 \times 10^{-4}$	0.94
	<sup>7</sup> Be	0.4	$2.1 \times 10^{-5}$	1.08
	<sup>22</sup> Na	17.2	$8.6 \times 10^{-6}$	1.21
	<sup>45</sup> Ca	5.0	$1.5 \times 10^{-5}$	0.45
	<sup>54</sup> Mn	2.4	$5.9 \times 10^{-6}$	1.13
	<sup>60</sup> Co	72.0	$1.8 \times 10^{-5}$	1.19

\*See Table 13

\*\*Vendor reported results as less than the required sensitivity, i.e.,  $3 \times 10^{-6}$  μCi/ml, for these isotopes

+Chemical separation method not used to determine <sup>45</sup>Ca results.

TABLE 16

Quality Assurance Results for Teledyne

Second Half of CY-1985

Sample Number	Radio-nuclide	Percentage of Concentration Guide for On-Site Surface Waters* (%)	Prepared Concentration ( $\mu\text{Ci}/\text{ml}$ )	Ratio of Teledyne Result to Prepared Concentration
8508	$^3\text{H}$	0.03	$2.8 \times 10^{-6}$	1.10
	$^7\text{Be}$	0.02	$1.0 \times 10^{-6}$	1.11
	$^{22}\text{Na}$	1.7	$8.4 \times 10^{-7}$	1.21
	$^{45}\text{Ca}$	0.3	$8.9 \times 10^{-7}$	0.67
	$^{54}\text{Mn}$	2.2	$5.5 \times 10^{-6}$	1.04
	$^{60}\text{Co}$	14.0	$3.5 \times 10^{-6}$	1.04
8509	$^3\text{H}$	0.02	$1.5 \times 10^{-6}$	**
	$^7\text{Be}$	0.02	$9.5 \times 10^{-7}$	0.95
	$^{22}\text{Na}$	0.3	$1.7 \times 10^{-7}$	1.17
	$^{45}\text{Ca}$	0.1	$3.1 \times 10^{-7}$	0.80
	$^{54}\text{Mn}$	0.4	$1.1 \times 10^{-6}$	0.94
	$^{60}\text{Co}$	14.0	$3.5 \times 10^{-6}$	1.10
8510	$^3\text{H}$	0.03	$2.7 \times 10^{-6}$	1.44
8511	$^3\text{H}$	0.03	$3.4 \times 10^{-6}$	1.05
EML 8511	$^3\text{H}$	0.2	$1.9 \times 10^{-5}$	1.01
	$^{54}\text{Mn}$	1.8	$4.4 \times 10^{-6}$	1.16
	$^{60}\text{Co}$	19.2	$4.8 \times 10^{-6}$	1.01

\*See Table 13

\*\*Vendor reported results as less than the required sensitivity, i.e.,  $3 \times 10^{-6}$   $\mu\text{Ci}/\text{ml}$ , for these isotopes

being placed on control of hazardous wastes and other non-radioactive pollutants.

As the inventory of radionuclides with potential for release to the environment has grown, the environmental monitoring program has expanded. More remote pathways have been explored and additional sampling points have been added. Special investigations have been made to measure concentrations and to detect any unexpected movement of radionuclides. These investigations would provide any necessary early warnings, giving time to take action before an off-site problem occurs. No such warnings have resulted to date.

Fermilab has number of closed water systems which build up inventories of radionuclides, primarily tritium. These are sampled periodically to provide information useful for spill control. The precautions taken are imposed based upon the potential environmental impact. Once the concentration exceeds that which can be released according to DOE regulations (Section 5), then a spill plan is written and becomes part of the Laboratory's Spill Prevention, Control, and Countermeasures Plan (SPCC Plan).

When spills occur from closed loop cooling systems, sump pumps are shut off in the vicinity and samples taken to determine whether or not the water in the sump pits can be released. One of the lessons learned from spills is that a leak of water into a vacuum system in a radiation area can result in much higher tritium concentrations in the water pumped out than

expected. The water vapor removes tritium from components. Other radionuclides are not removed, so a gamma-ray survey meter does not detect the problem.

The Laboratory has recently strengthened its environmental review program. All new projects requiring project directives and/or affecting land management on the site receive an comprehensive environmental review. The review program includes considerations such as threatened and endangered species, cultural resources, wetlands, and floodplains, specifically addressed in the National Environmental Policy Act (NEPA).

Groundwater protection from organic chemicals has become a concern in recent years. Testing for several of these has been added to our sampling and analysis program. In addition, the Laboratory has strengthened its waste collection and auditing programs. Hazardous waste generators on the site are now required by regulation<sup>37</sup> to have a waste reduction program. The Laboratory documents these efforts in an annual report to the State of Illinois. This report also gives the quantities and types of hazardous waste generated, stored on-site, and disposed of off the site. The Laboratory does not have an on-site hazardous waste disposal facility.

All radioactive waste disposed of in CY-1985 from Fermilab was sent to a Department of Energy burial facility at Richland, Washington.

As a result of the transformer failure and oil spill, the Laboratory strengthened its spill control program in CY-1985. A new trailer of larger capacity was purchased to permit a larger variety of spill control materials to be carried to the spill site. Reserve spill control materials were collected at one location for ease in finding them in an emergency. Additional response personnel were trained both at the advisory and worker levels.



5. References

The appropriate Radiation Protection Standard for penetrating radiation applied to individuals in uncontrolled areas was taken from the DOE Order 5480.1A, Chapter XI.<sup>6</sup> The annual dose for whole body exposure is 500 mrem when applied to occasional exposures such as might occur during an accident. The appropriate standard for a prolonged period of exposure of the general population is 100 mrem/yr. Off-site exposures from routine Fermilab operations will be considered to be for prolonged periods. On-site exposures of individual members of the general population will be considered occasional exposures.

The Concentration Guides used in the analyses of the surface water samples for radioactivity were taken from the draft revision to DOE Order 5480.1, Chapter XI, (replacing Table II, Column 2). Derived Concentration Guides (DCG) - Concentrations of Radionuclides in Water and Air that could be Continuously Consumed or Inhaled, Respectively, and Not Exceed an Effective Dose Equivalent of 100 mrem/year. The specifications are given in Table 13. These Derived Concentrations Guides are based on guidance given in International Commission on Radiological Protection (ICRP) Publications 23, 26, and 30, Pergamon Press, New York. For tritium the Derived Concentration Guide is  $2 \times 10^{-7}$   $\mu\text{Ci}/\text{m}\ell$ . For  $^{11}\text{C}$  the Concentration Guide, for prolonged exposures,  $1.2 \times 10^{-8}$   $\mu\text{Ci}/\text{m}\ell$ , was taken from the calculations by Yamaguchi<sup>38</sup> because immersion dose must be included for  $^{11}\text{C}$ .

The Concentration Guide used in the analyses of ground water samples for tritium was taken from the U. S. Environmental Protection Agency regulations for community drinking water systems.<sup>22</sup> The maximum contamination level permitted for tritium is  $2 \times 10^{-5}$   $\mu\text{Ci/ml}$  and corresponds to an annual exposure of 4 mrem if one uses the supply as one's sole drinking water source. Note that this is 25 times more stringent than the DOE regulation for a prolonged period of exposure of the general population, which is 100 mrem/year. The Concentration Guides for the other radionuclides in Fermilab's analyses of groundwater samples have been determined by dividing the Derived Concentration Guides (DCG) in the draft revision to DOE Order 5484.1 by 25 (Table 13). The specified sensitivity and precision of the analyses have been reduced to well below these Concentration Guides (to at most 10% of the Guide).

The Air and Water Pollution Standards for non-radioactive pollutants were taken from Chapters 2 and 3 of the State of Illinois Pollution Control Board Rules and Regulations.<sup>25</sup> The waters on-site were considered to be in the "general use" category. The value for total hexavalent chromium for general water quality is 0.05 mg/l. The Standards for total copper at the discharge point and for general water quality are 1.0 and 0.02 mg/l respectively, for silver are 0.1 and 0.005 mg/l respectively, and for cyanide are 0.025 mg/l for both. The maximum contaminant level for chloride in water for general use is 500 mg/l and the level of total dissolved solids is 1000 mg/l. In public drinking water the standards for chloride and total dissolved solids are 250 mg/l and 500 mg/l, respectively.<sup>39</sup> The Air Quality Standards limit the release for oxides of

nitrogen to 136 g (0.3 lbs) per 252 million calories (per million btu's) of actual heat input in any one hour. Release of sulfur dioxide shall not exceed 2000 ppm.<sup>25</sup>

The appropriate regulations for PCBs and hazardous wastes are found in the U. S. Code of Federal Regulations 40 CFR 761 and 40 CFR 260-265, respectively. The concentration limit is 2 ppm for human consumption of fish.<sup>40</sup>

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7.

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