



National Institute of Standards & Technology

Certificate of Analysis

Standard Reference Material 3131a

Spectrometric Solution Standard

Magnesium

Batch Code 491403

This Standard Reference Material (SRM) is intended for use in atomic absorption spectrometry, optical emission (plasma) spectrometry, spectrophotometry, or any other analytical technique that requires aqueous standard solutions for calibrating instruments. SRM 3131a is a single element solution prepared gravimetrically to contain 10 mg/mL of magnesium with a nitric acid concentration (V/V) of 10 percent. The certified value is based on a gravimetric procedure, i.e., weight per volume composition of the high-purity metal dissolved in NIST high-purity reagents. The uncertainty listed is based on gravimetric and volumetric uncertainties of the preparation and the effect of solvent transpiration through the container wall for one year. The density of the solution at 22 °C is 1.068 g/mL.

Metal	Concentration (mg/mL)	Source Purity, %	Acid Conc. (V/V) Approximate
Mg	10.00 ± 0.03	Mg metal (99.98)*	HNO ₃ , 10%

*This high-purity material was analyzed for gases by inert gas fusion and for metals by optical emission spectrometry. It was found to contain 0.0011 weight percent dissolved gases (O₂, N₂, H₂) and less than 100 µg/g total metallic impurities.

Procedures for Use

Stability: This certification is valid for one year from the shipping date, provided the SRM solution is kept tightly capped and stored under normal laboratory conditions. NIST will monitor the stability of representative solutions from this SRM lot, and if any changes occur that invalidate this certification, NIST will notify purchasers.

Preparation of Working Standard Solutions: All solutions should be brought to 22 ± 1 °C and all glass or plastic surfaces coming into contact with the SRM must have been previously cleaned. A working standard solution can be prepared from the SRM solution by serial dilution. Dilutions should be made with certified volumetric class A flasks and 5 or 10 mL class A pipets. All volumetric transfers of solutions should be performed using a proven analytical technique. Each dilution should be acidified with an appropriate high-purity acid and diluted to calibrated volume using high-purity water. The stability of the working standard solution will depend on the final acid concentration; therefore, care should be exercised to ensure that the final acid concentration of the dilution closely approximates that of the SRM. To achieve the highest accuracy, the analyst should prepare daily working solutions from 100 µg/mL dilutions of the original SRM solution.

Gaithersburg, MD 20899
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Thomas E. Gills, Chief
Standard Reference Materials Program

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SRM 3131a was prepared by T.A. Butler of the NIST Inorganic Analytical Research Division; atomic absorption and emission spectrometry analyses were made by T.A. Butler and J.A. Norris. Gas analyses were performed at Luvak, Inc. Boyleston, MA.

The technical and support aspects involved in the preparation, certification, and issuance of this SRM were coordinated through the Standard Reference Materials Program by J.S. Kane.