



# **Real-Time Water-Quality Monitoring for Water Security Applications**

**General Meeting of the Federal Remediation  
Technologies Roundtable, Arlington, VA, Dec. 7, 2005**

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# Real-Time Water-Quality Monitoring for Water Security

- Can water-quality monitoring increase security?
- What sensors should be used?
- How reliable are the available sensors?
- What are the maintenance and replacement intervals?

GAO

Testimony  
Before the Subcommittee on Environment  
and Hazardous Materials, Committee on  
Energy and Commerce, House of  
Representatives

For Release on Delivery  
Expected at 12:30 p.m. EDT  
Thursday, September 30, 2004

## DRINKING WATER

### Experts' Views on How Federal Funding Can Best Be Spent To Improve Security

Statement of John B. Stephenson, Director  
Natural Resources and Environment



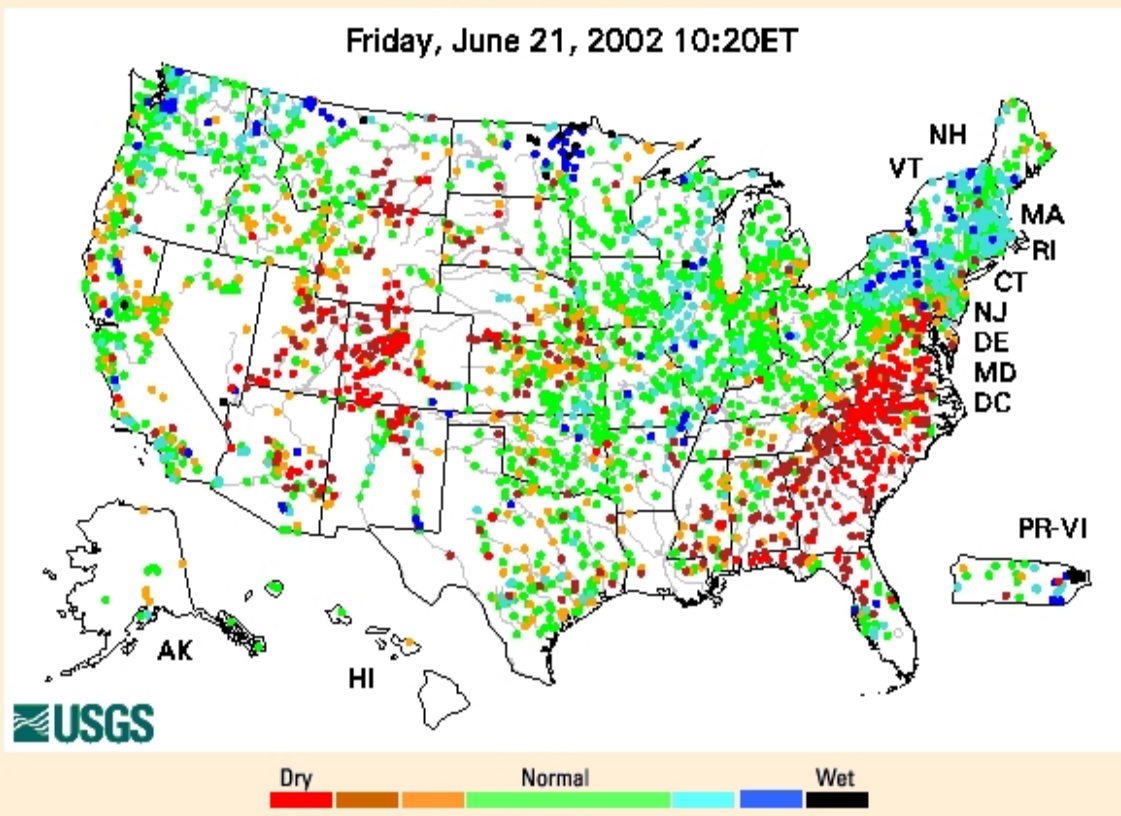
***“The need to develop near real-time monitoring technologies*** which would be particularly useful in quickly detecting contaminants in water that has already left the treatment plant for the consumer, ***has by far the strongest support”***

***Nearly 100%*** of experts consider this a high priority.



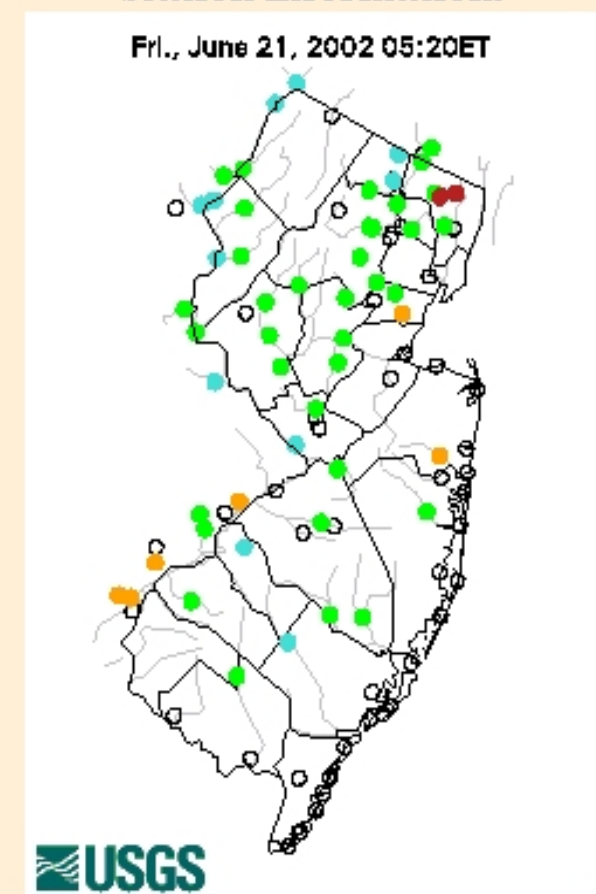
# Existing USGS real-time systems

## Map of real-time streamflow compared to historical streamflow for the day of the year (United States)



## Daily Streamflow Conditions

Select a site to retrieve data and station information.



# Objectives of the Overall Water Security Research Program

- To develop a real-time water-quality monitoring system for drinking water safety and security
- To evaluate available sensors for use in such a system
- To install and test the system in water distribution systems

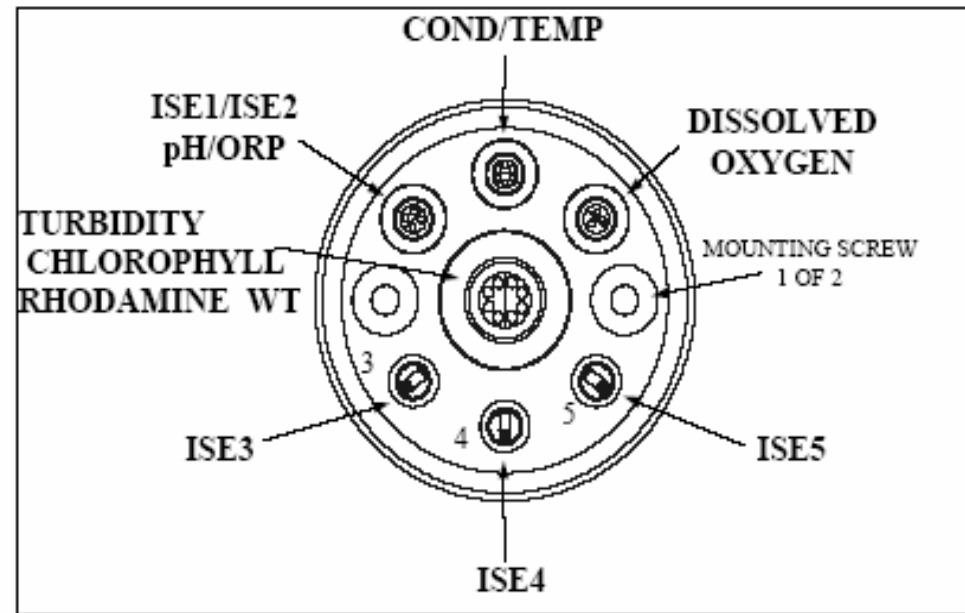
# Site Installations

- 5 distribution-system sites established
- Free Chlorine, Specific Conductance, pH, Oxidation-Reduction Potential, and temperature measured at all sites
- Total Organic Carbon and UV/VIS (on loan) at one site
- Additional sites based on model results to be installed (locations optimized for public health protection) – pending logistical issues

# Water-Quality Monitoring System



YSI 6920 Sonde Bulkhead: Probe Assemblage







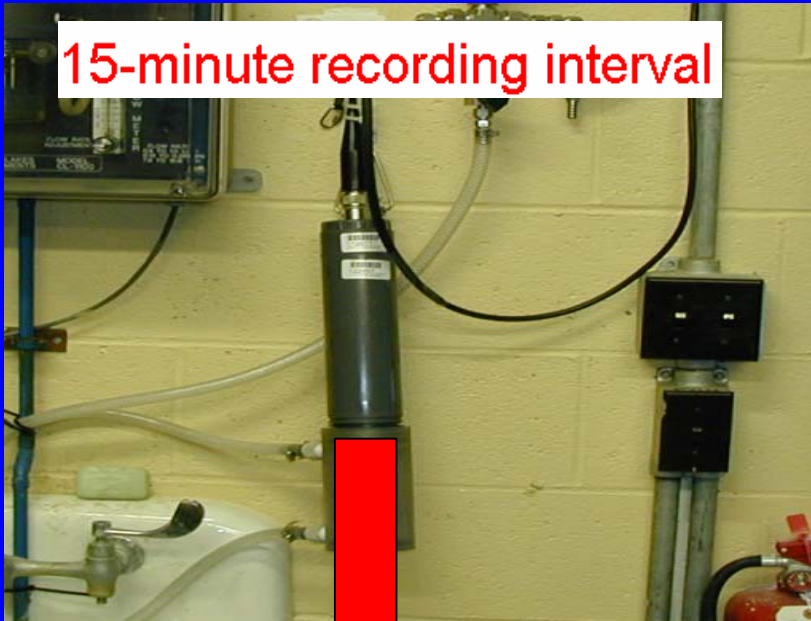
## Typical site installation

- Sensor locations: Based on distribution-system modeling
  - Water utility facilities, pumping stations, homes, government buildings, hospitals...
- Sensors: free chlorine, oxidation-reduction potential, specific conductance, pH, temperature

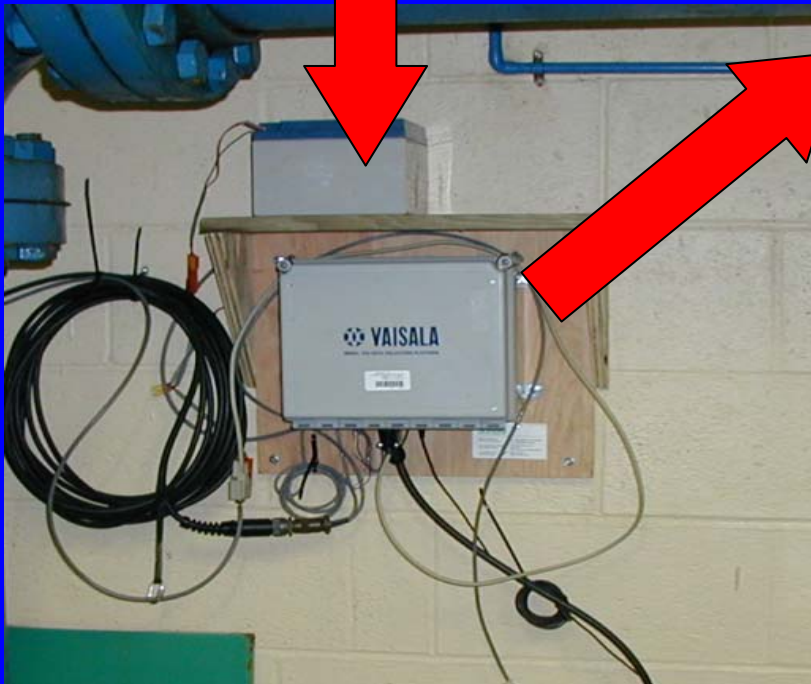


# Data telemetry and management : Sensors data transmitted to USGS secure webpage using satellite telemetry

15-minute recording interval



1 hour transmission



**USGS**  
 District Access  
 Water Resources

Data Category: Realtime  
 Geographic Area: New Jersey go

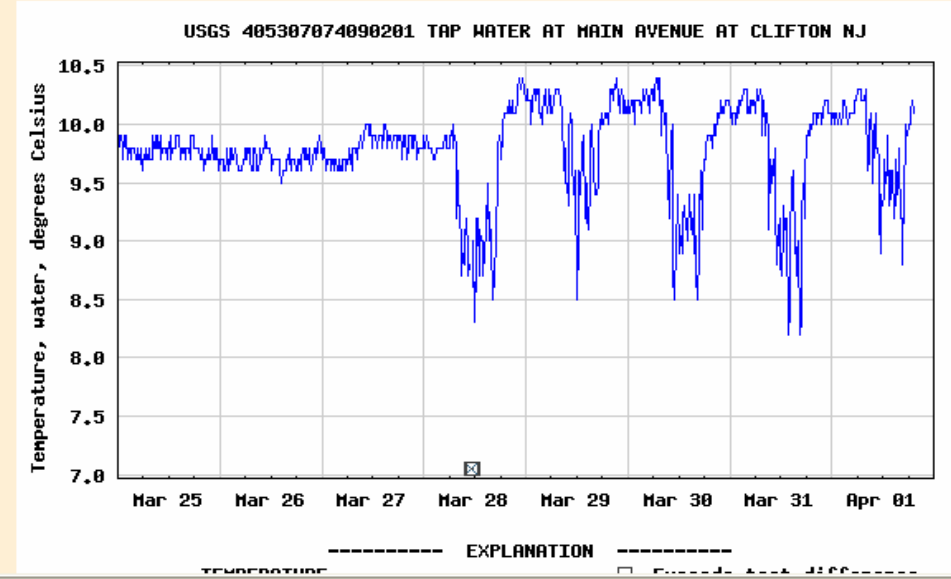
**USGS 405307074090201 TAP WATER AT MAIN AVENUE AT CLIFTON NJ (internal access only)**  
PROVISIONAL DATA SUBJECT TO REVISION

Available data for this site Realtime GO

<b>Available Parameters</b> All 9 parameters available at this site 00010 Temperature, water (DD 01) 00095 Specific cond at 25C (DD 02) 00300 Dissolved oxygen (DD 03)	<b>Output format</b> Graph	<b>Days</b> 7 (1-31)	<input type="button" value="get data"/>
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**Temperature, water, degrees Celsius**

Most recent value: 10.1 04-01-2005 19:15



**USGS**  
 District Access  
 Water Resources

Data Category: Realtime  
 Geographic Area: New Jersey go

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 00300 Dissolved oxygen (DD 03)

**Output format**  
 Table

**Days**  
 7  
 (1-31)

get data

Date / Time	Temperature, water, deg C (DD 01)	Oxidation reduction potential, mV (DD 16)	Specific conductance, uS/cm 25 degC (DD 02)	Dissolved oxygen, mg/L (DD 03)	Dissolved oxygen, percent of saturation (DD 04)	pH, water, unfltrd field, std units (DD 05)	Turbidity, IR LED light, det ang 90 deg, FNU (DD 06)	DCP battery voltage volts (DD 07)	Estimate of DCP trans. power, dB (DD 09)
03/25/2005 00:00	9.7	654	411	13.3	110	8.50	.0	13.7	
03/25/2005 00:15	9.9	648	410	13.2	110	8.51	.1		
03/25/2005 00:21									46.0
03/25/2005 00:30	9.9	643	411	13.3	110	8.52	.1		
03/25/2005 00:45	9.9	644	410	13.3	110	8.51	.2		
03/25/2005 01:00	9.8	645	412	13.3	110	8.51	.1	13.7	
03/25/2005 01:15	9.7	645	412	13.3	110	8.51	.1		
03/25/2005 01:21									46.0
03/25/2005 01:30	9.9	640	411	13.3	110	8.52	.0		
03/25/2005 01:45	9.9	638	410	13.3	110	8.53	.0		

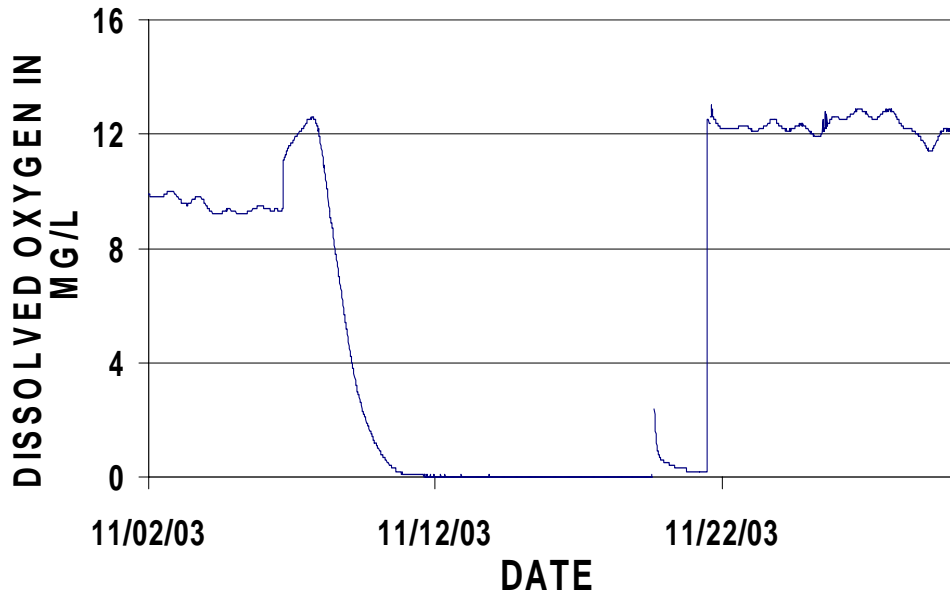


# Some Realities of Field Work with Sensors

- Often heard questions and comments:
  - Did anyone bring the manual for this thing?
  - Oh, No, not again!
  - Do we have a spare?
  - Did I really spend (X) years in college to do this?
  - If one more thing goes wrong I will miss (fill in favorite 9:00 pm TV show).

# Example 3: Dissolved Oxygen Sensor Membrane Replacement

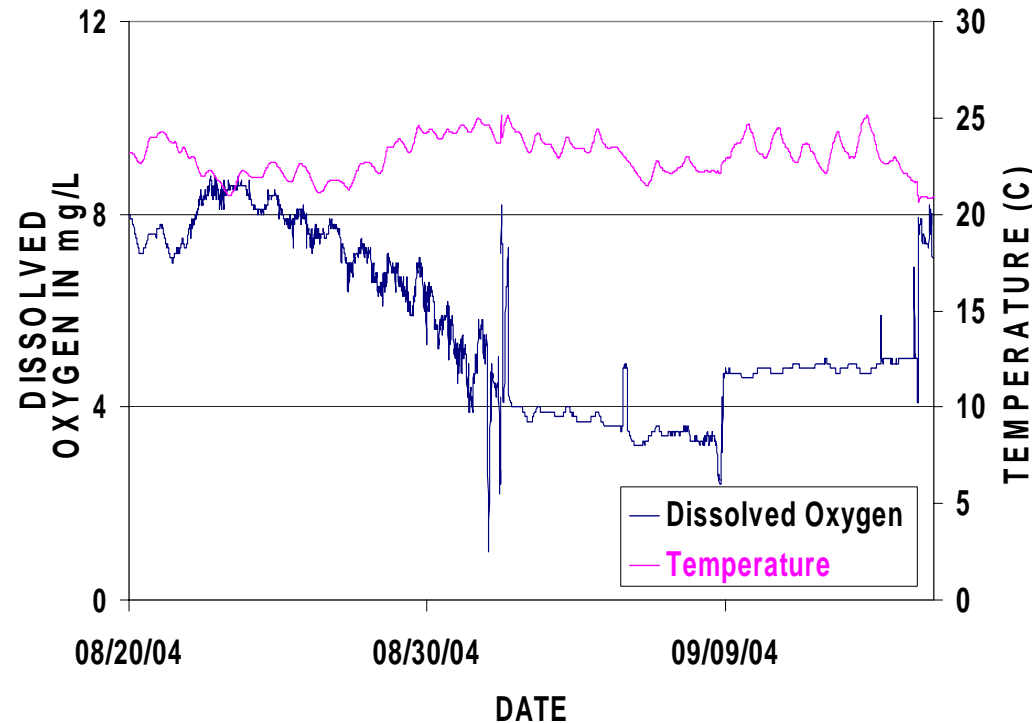
## Site SW1 Dissolved Oxygen Data



Sudden decline in response indicates membrane needs to be replaced.

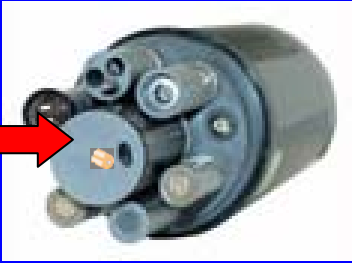
Gradual decline in response, noticeable as higher-than-normal calibration drift. Membrane needs to be

## Site SW1 Dissolved Oxygen Data





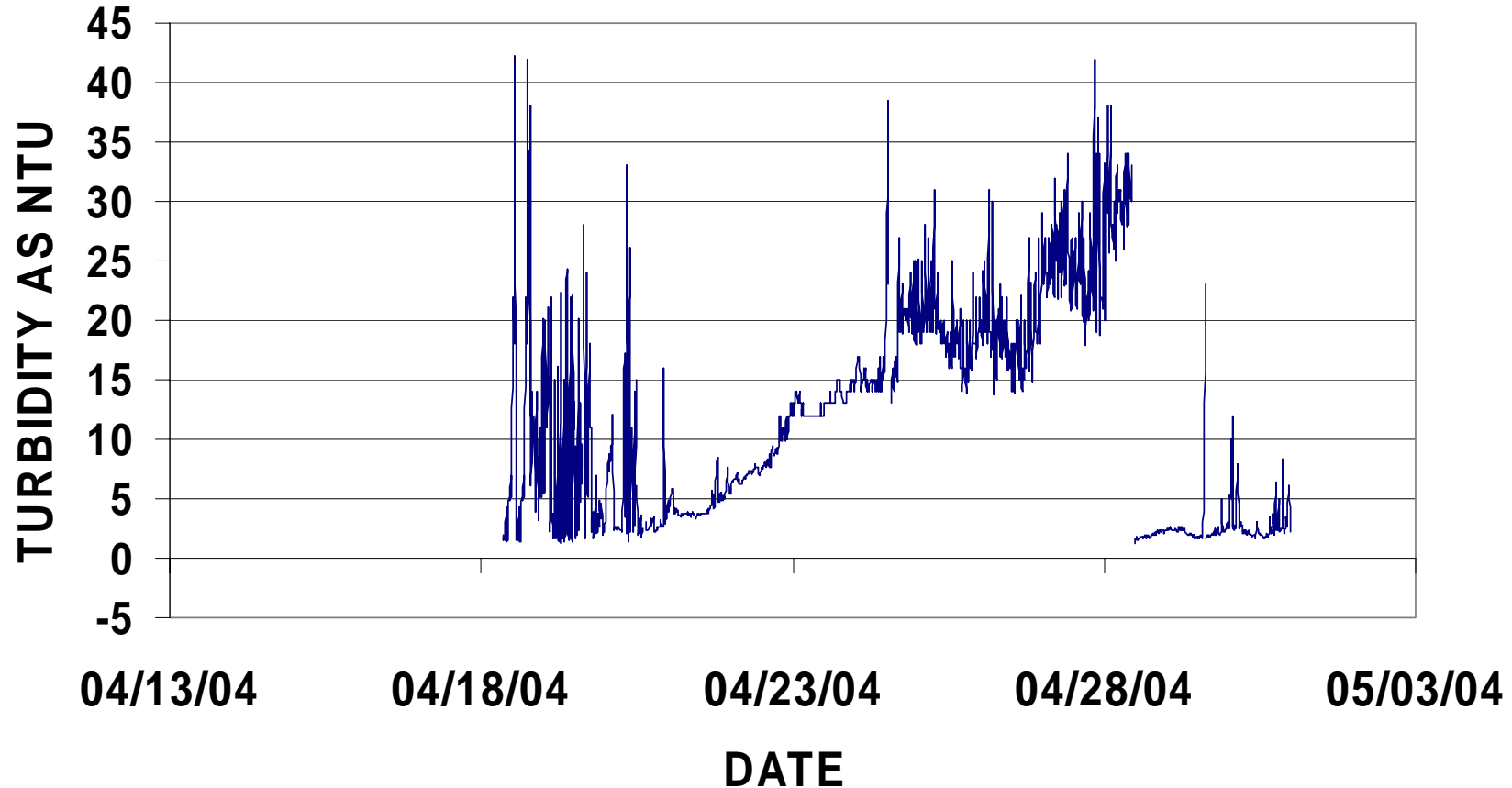
# Example 4: Turbidity Sensor Fouling



Wiper keeps particles from building up on window.

Removal of wiper causes inaccurately high readings

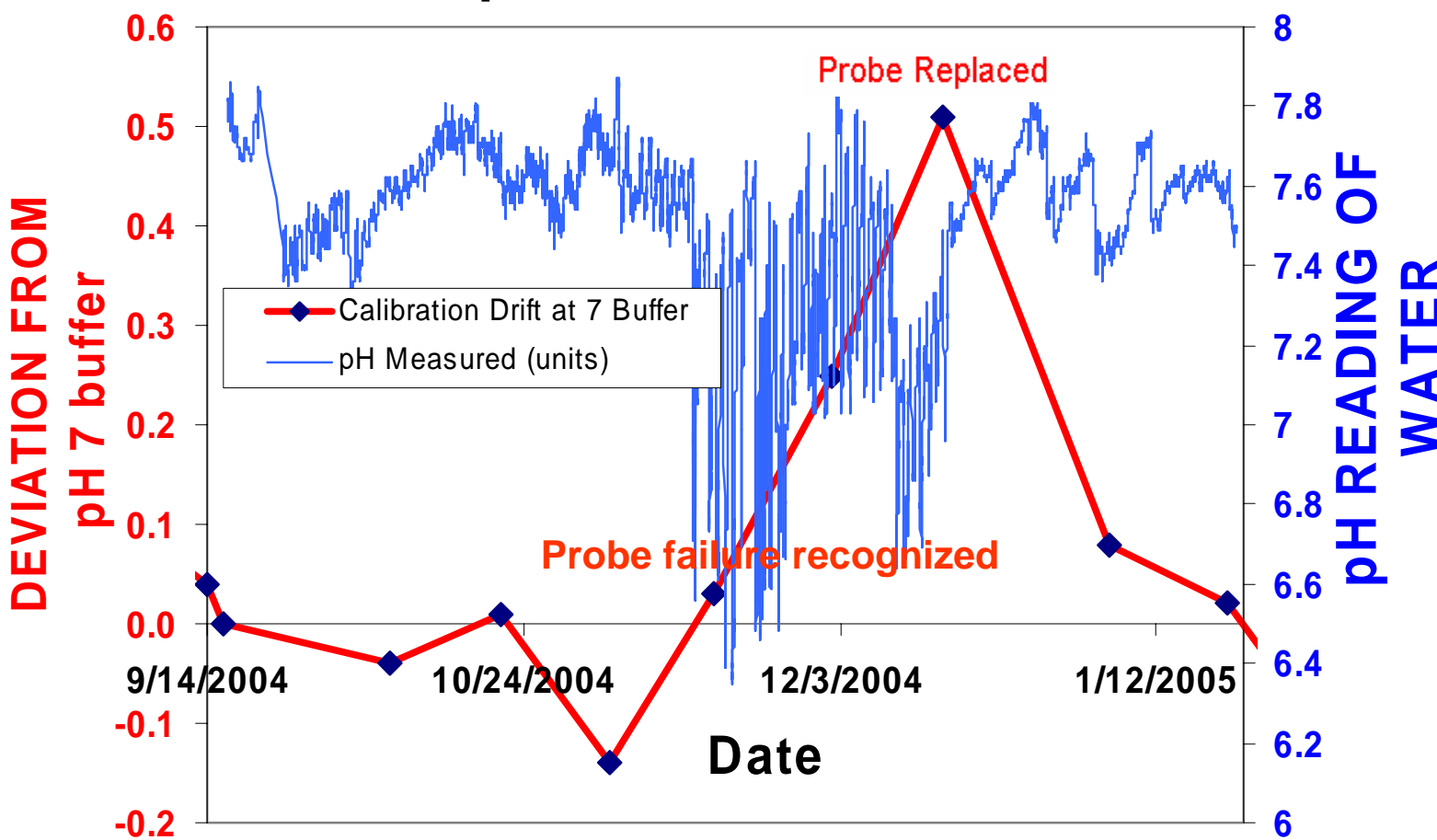
## Site SW 1 Turbidity: Sensor Fouling



# Example 1: pH Sensor that Needs to be Replaced

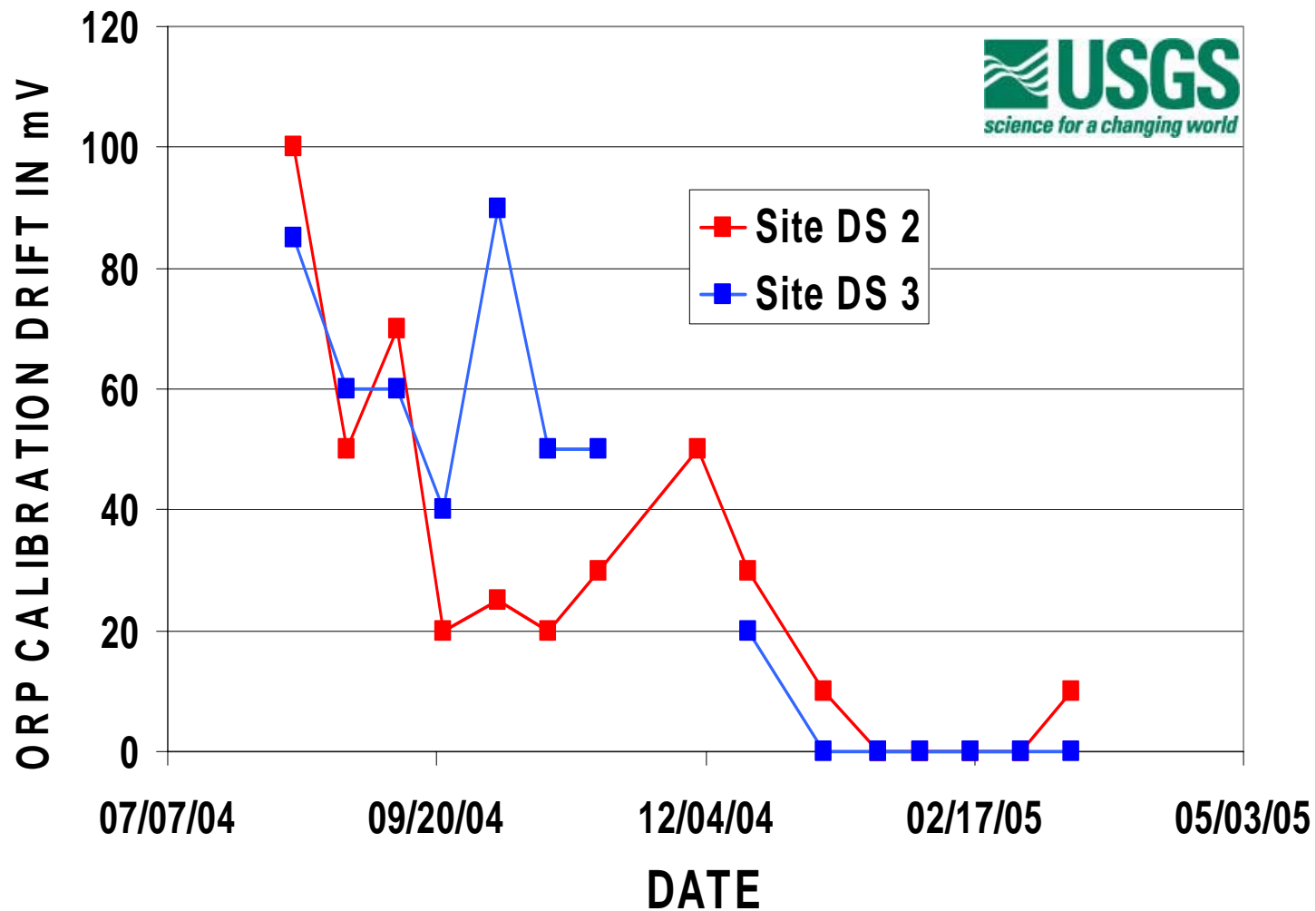
As probe signal degrades, response in buffer **deviates farther from standard value.**

## Site DS3 pH Probe: Calibration Drift



Result: Data from 11/04 – 12/04 **not accurate**, and variability not representative of true system conditions

# Example 2: Oxidation-Reduction-Potential (ORP) Sensor Performance: Calibration Drift of New Sensors



New sensors:  
Response (mV  
reading) increases  
after deionized  
water rinse

**Possible reason:**

Platinum electrode  
builds up an **oxide  
coating** which is  
removed by de-  
ionized water.

Over time coating  
becomes resistant  
to rinsing, reducing  
post-cleaning  
calibration drift

What does this mean? Must account for **post-cleaning variability** on newer probes!











**SIEVERS 900**  
On-Line TOC Analyzer

CE

SIEVERS

For the





**VAISALA**  
MODEL 6600 DATA COLLECTION PLATFORM

**STOP**  
DO NOT OPEN THIS CASE  
UNLESS YOU ARE  
A QUALIFIED ELECTRICIAN  
OR A LICENSED ELECTRICIAN

**POWER SONIC**  
MODEL 6600

**6600**  
MODEL 6600

POWER SUPPLY

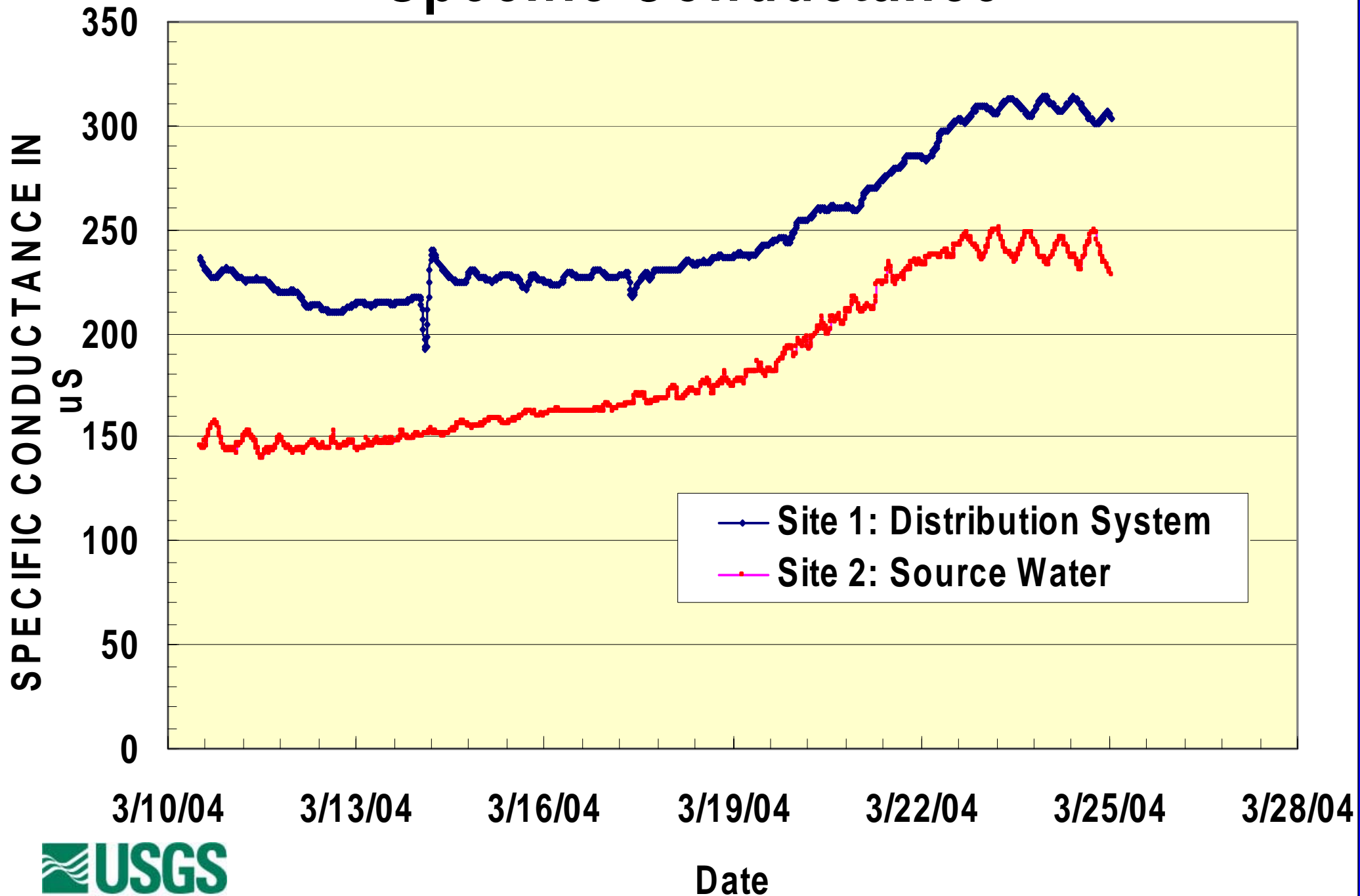
# Summary of Observations of the Sensors Tested

- **pH:** Mostly stable and accurate. Requires replacement after 1-2 years. Drift noticed over time, electrode requires reconditioning every 4-5 months.
- **SC sensor:** excellent performance
- **ORP sensor:** very accurate after initial “break-in” time
- **Chlorine residual sensor:** still beta testing, needs membrane replacement every 1-2 months. Needs calibration every 2 weeks.

# Sensor Performance: Key Issues

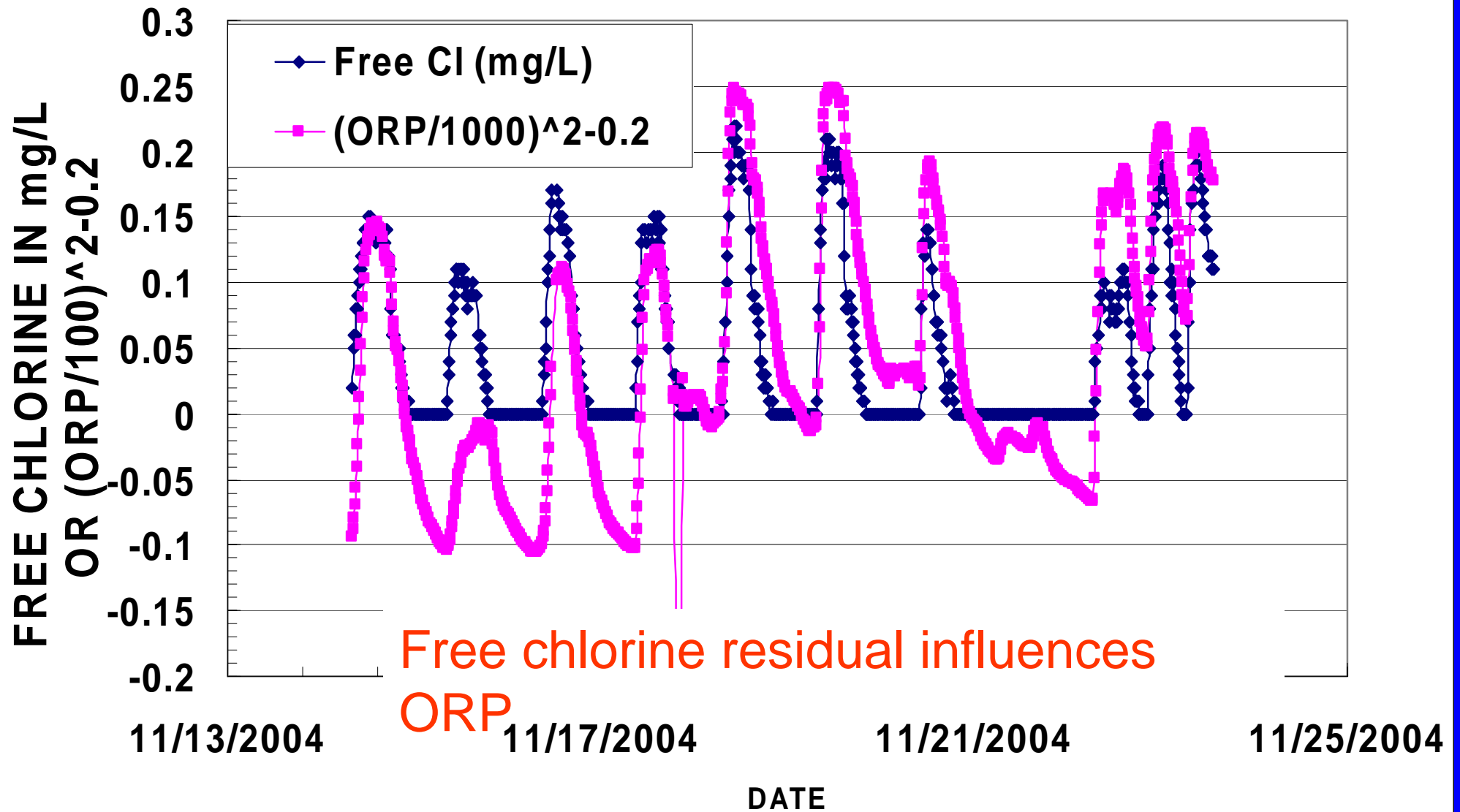
- **The need for sensor maintenance or replacement is not always obvious by data observation alone:**
  - Data from a nonresponsive sensor may look reasonable
- **Sensors must be maintained properly:**
  - **Calibration drift** affects data accuracy and precision
  - Quality assurance includes **assessing and correcting data** to reflect drift
  - **Shorter maintenance intervals** and more frequent calibrations reduce chances of significant sensor drift and **improve data accuracy**

# Specific Conductance





# RELATION BETWEEN FREE RESIDUAL CHLORINE AND ORP, DISTRIBUTION SITE 3



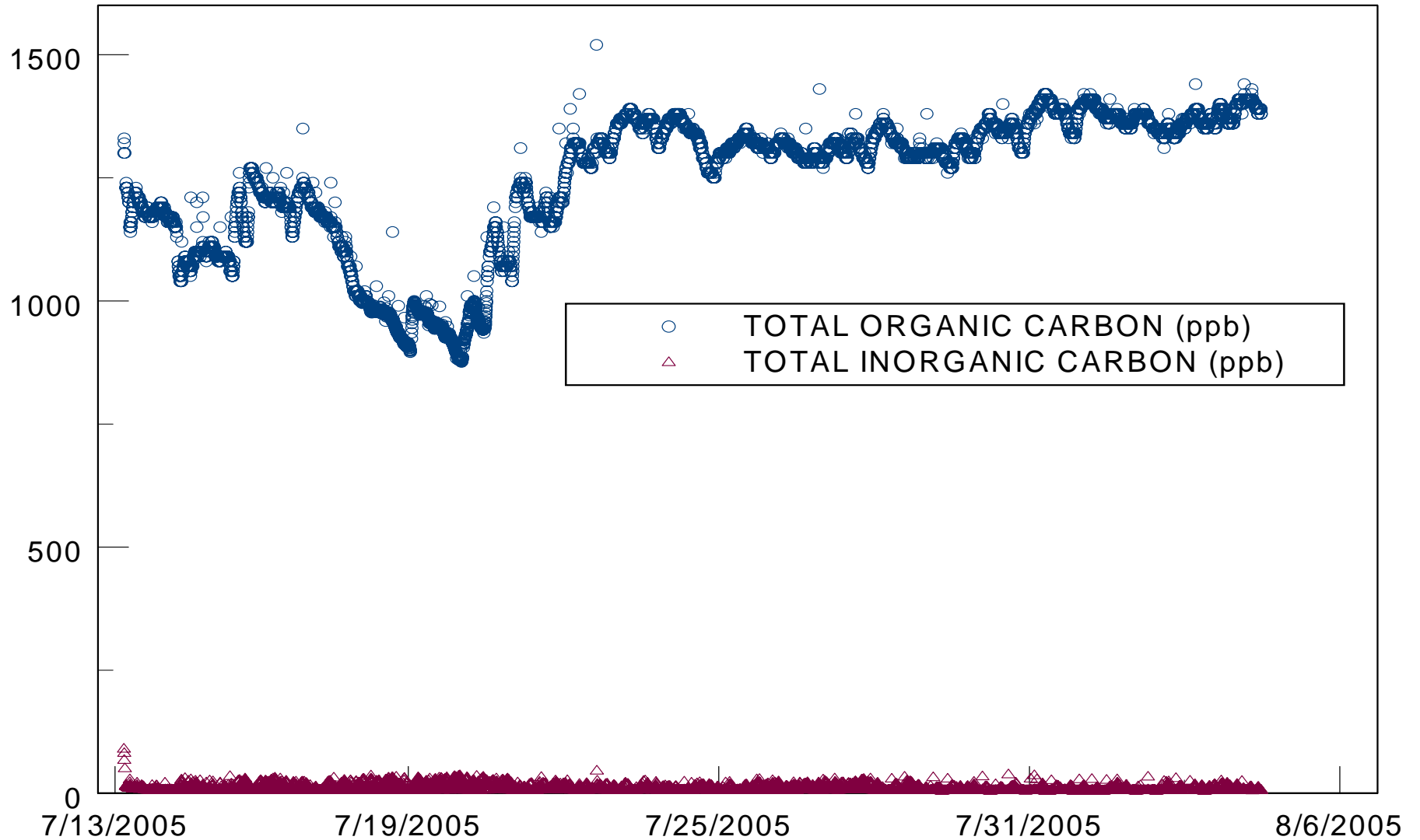
# Current Work: Characterizing water-quality variability in a distribution system

- **USEPA**
  - Conduct controlled laboratory experiments
  - Select sites based on distribution-system modeling
- **Sandia National Laboratories (SNL)**
  - Develop distribution-system and sensor network models
- **USGS**
  - Establish and operate a network of field sites
  - Collect and manage water quality data
  - Prepare interpretive reports
- **Cooperating Water Utility**
  - Provide and allow access to distribution system sites
  - Provide distribution system description and model
  - Support the field effort by preparing water and electrical connections and drains

# New Technologies Being Tested At Site DW3

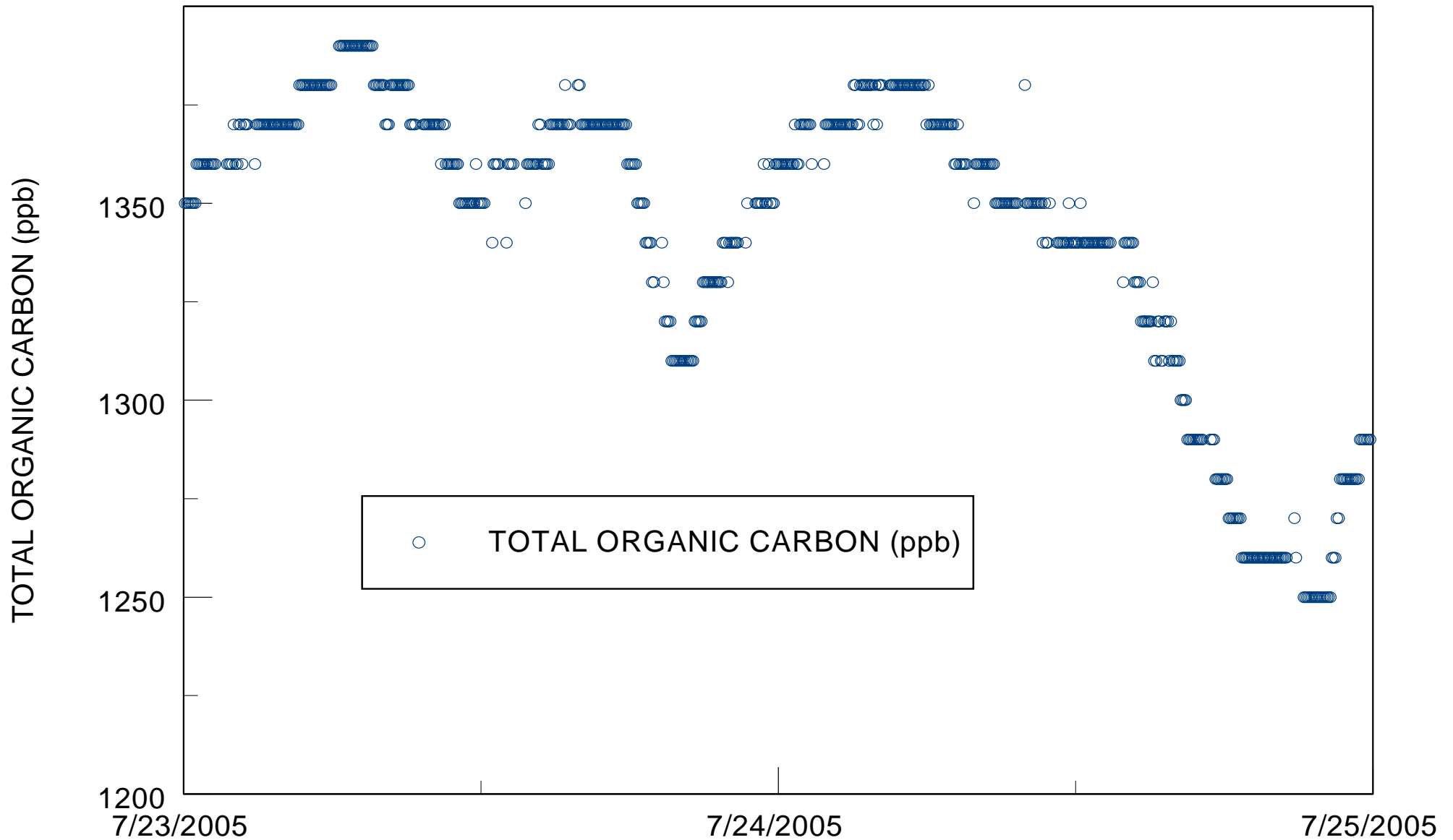
- First commercial use of **chlorine probe** in a YSI multiparameter water-quality monitoring system
- **Total organic and inorganic carbon analyzer** (General Electric)
- **UV-VIS spectrophotometer** with software for estimating water-quality parameter values and detecting unexpected changes in water quality (Scan Co.)

# Total Organic and Inorganic Carbon in Drinking Water, Site DW3



**Trends observed: Most carbon is organic, concentration is within a narrow range (800-1600 ppb), some outliers are present**

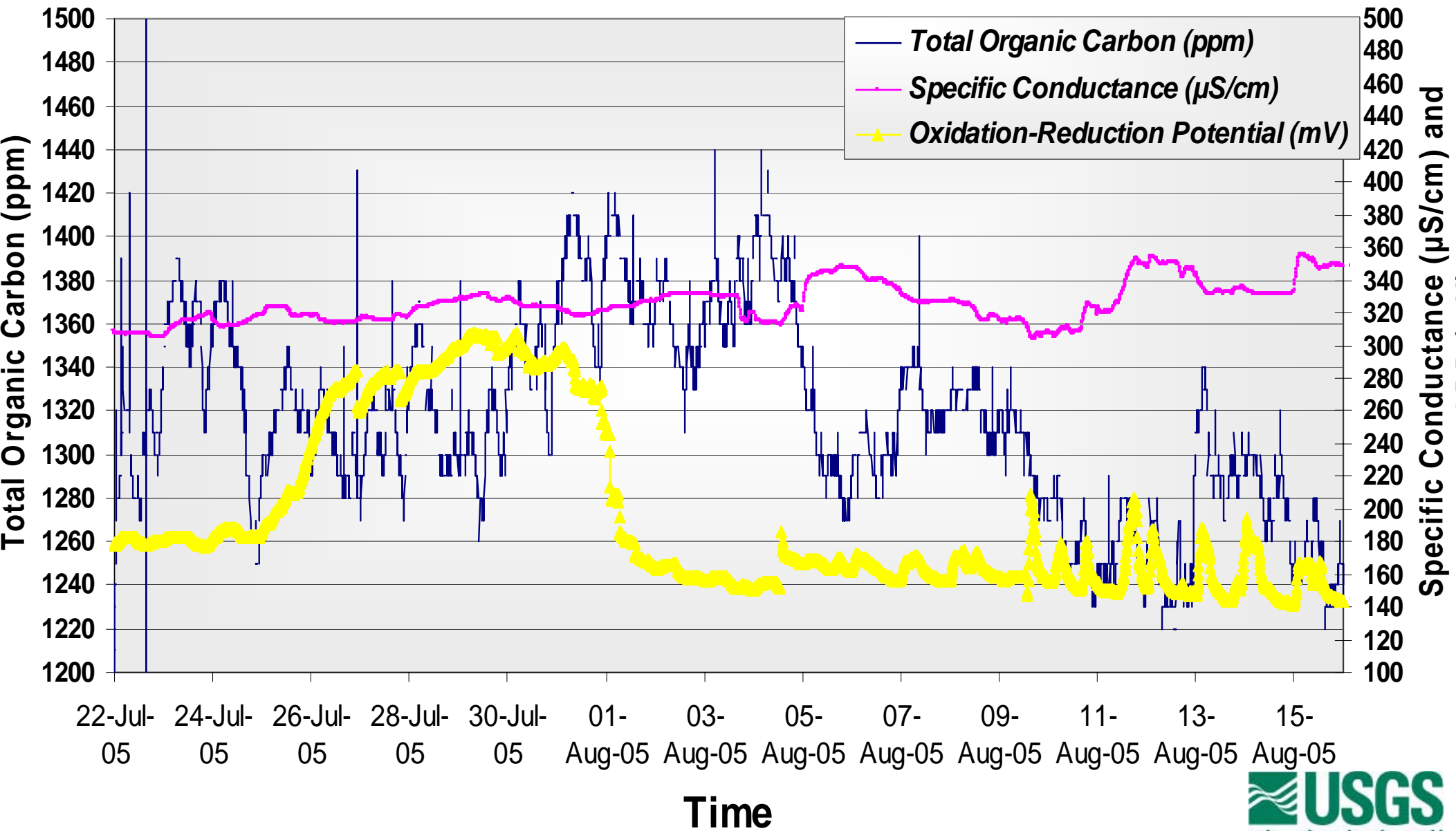
# Total Organic Carbon in Drinking Water, Site DW3



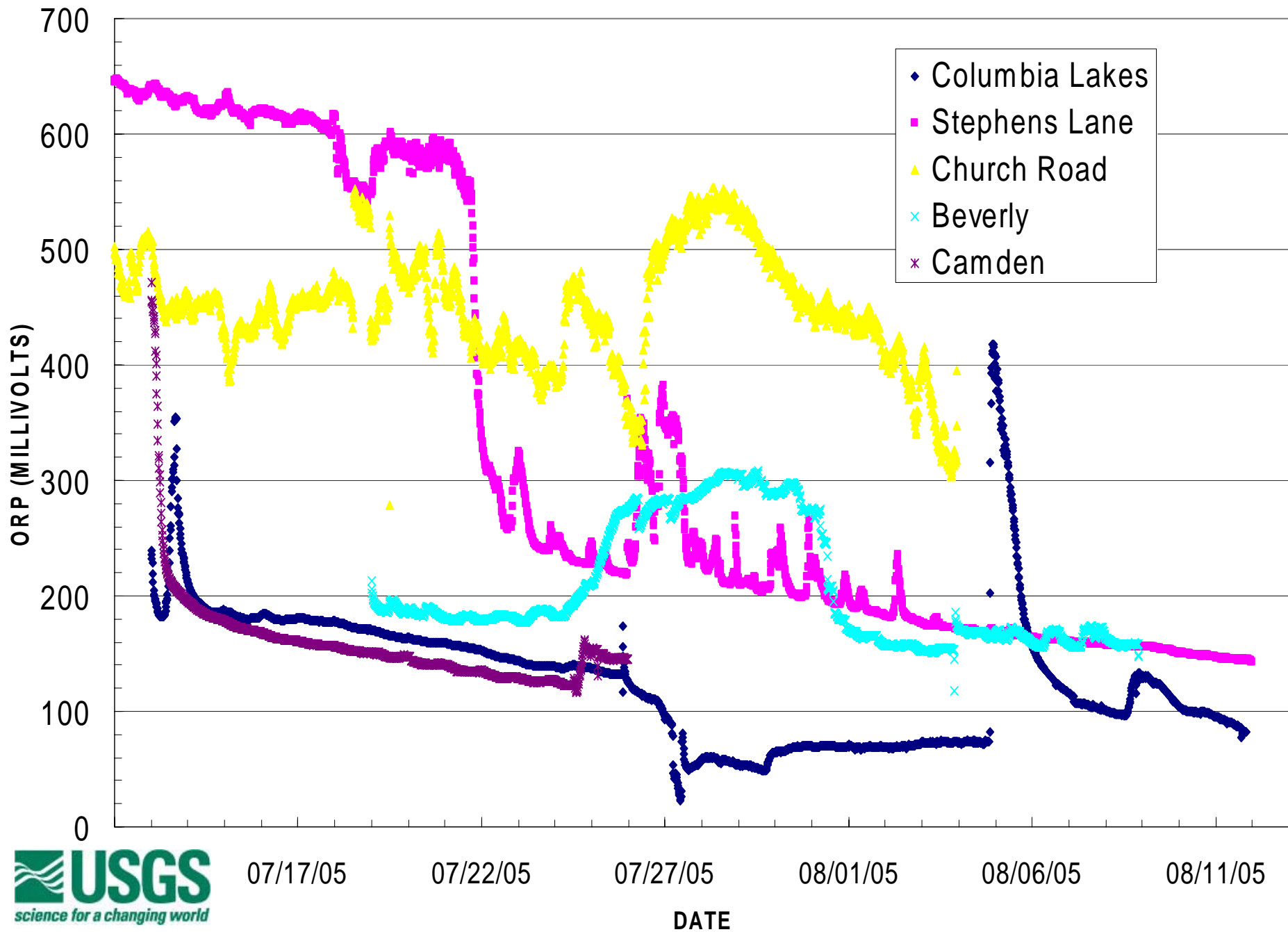
**Trends observed: Little variability in 2-day period, twice-daily peak concentrations resembling tidal pattern, “step function” concentration pattern**



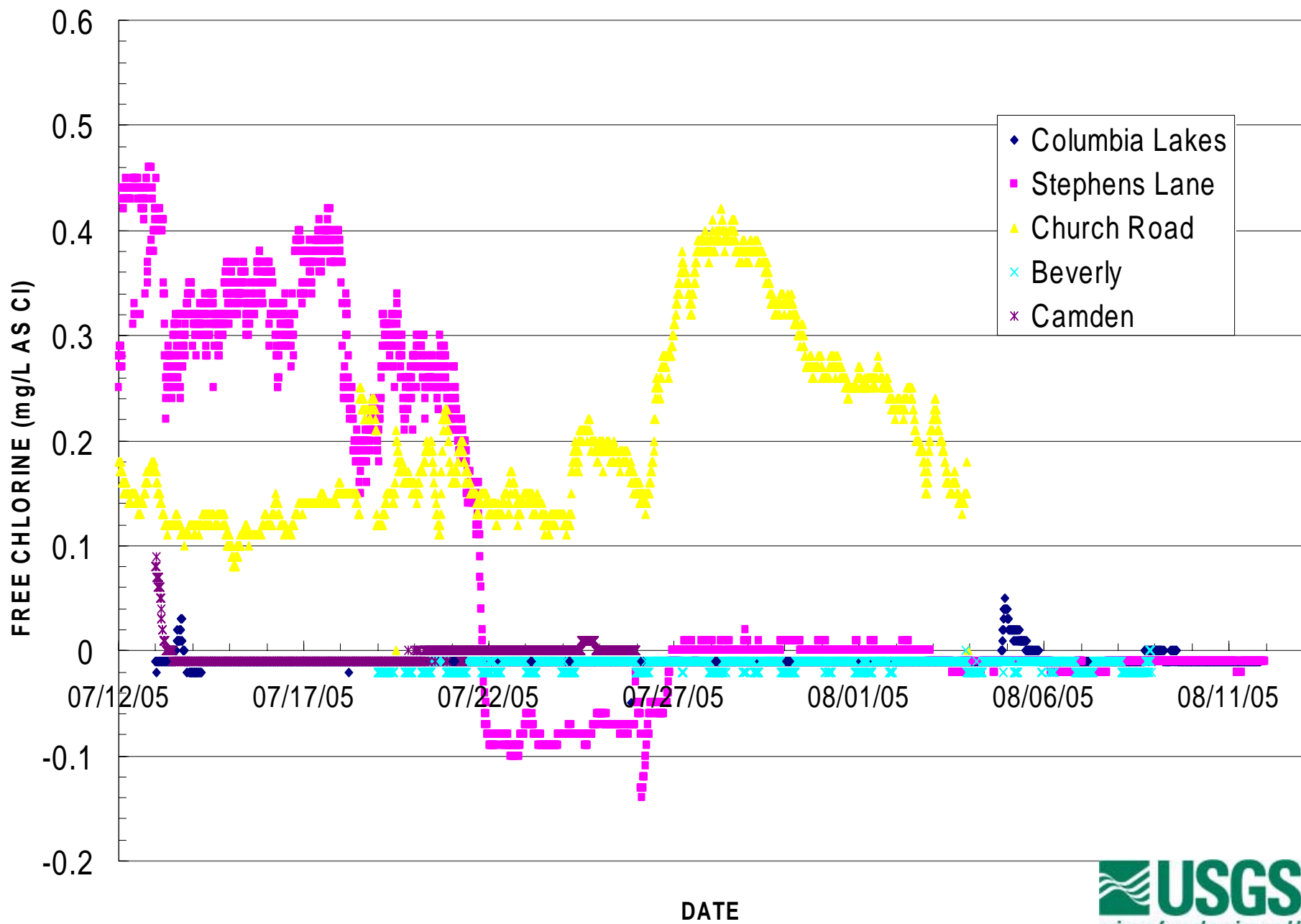
# Specific Conductance, Total Organic Carbon, and Oxidation Reduction Potential - in the absence of Chlorine Residual



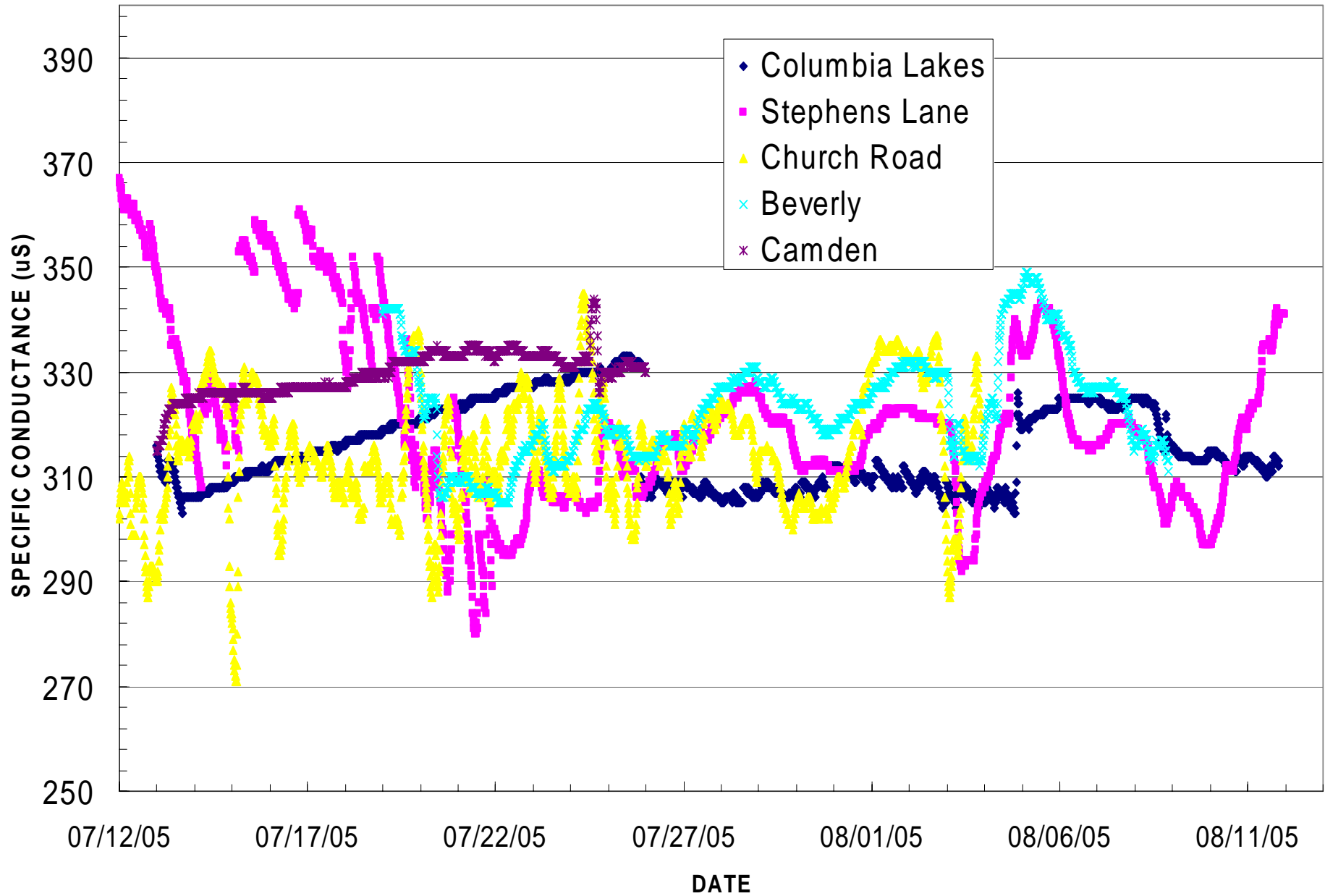
# Oxidation-Reduction Potential (ORP) at Five Sites



# Free Chlorine Residual at Five Sites



# Specific Conductance at Five Sites



# Analysis of Water-Quality Variability

- Spatially
  - Age of water
  - Distance between monitoring sites
  - Type of water (SW, GW, mixed)
- Temporally
  - 15-minute intervals (or more frequent if needed)
  - Hourly
  - Daily
  - Weekly
  - Monthly
  - Seasonally
  - Annually

# Explanation of the Density Diagram

A “smoothed histogram”, showing the shape of a data set.

**X axis:** The difference between each measurement (e.g., pH) and the mean of measured values (moving average) within a time increment (15 minutes, 4 hours or 24 hours).

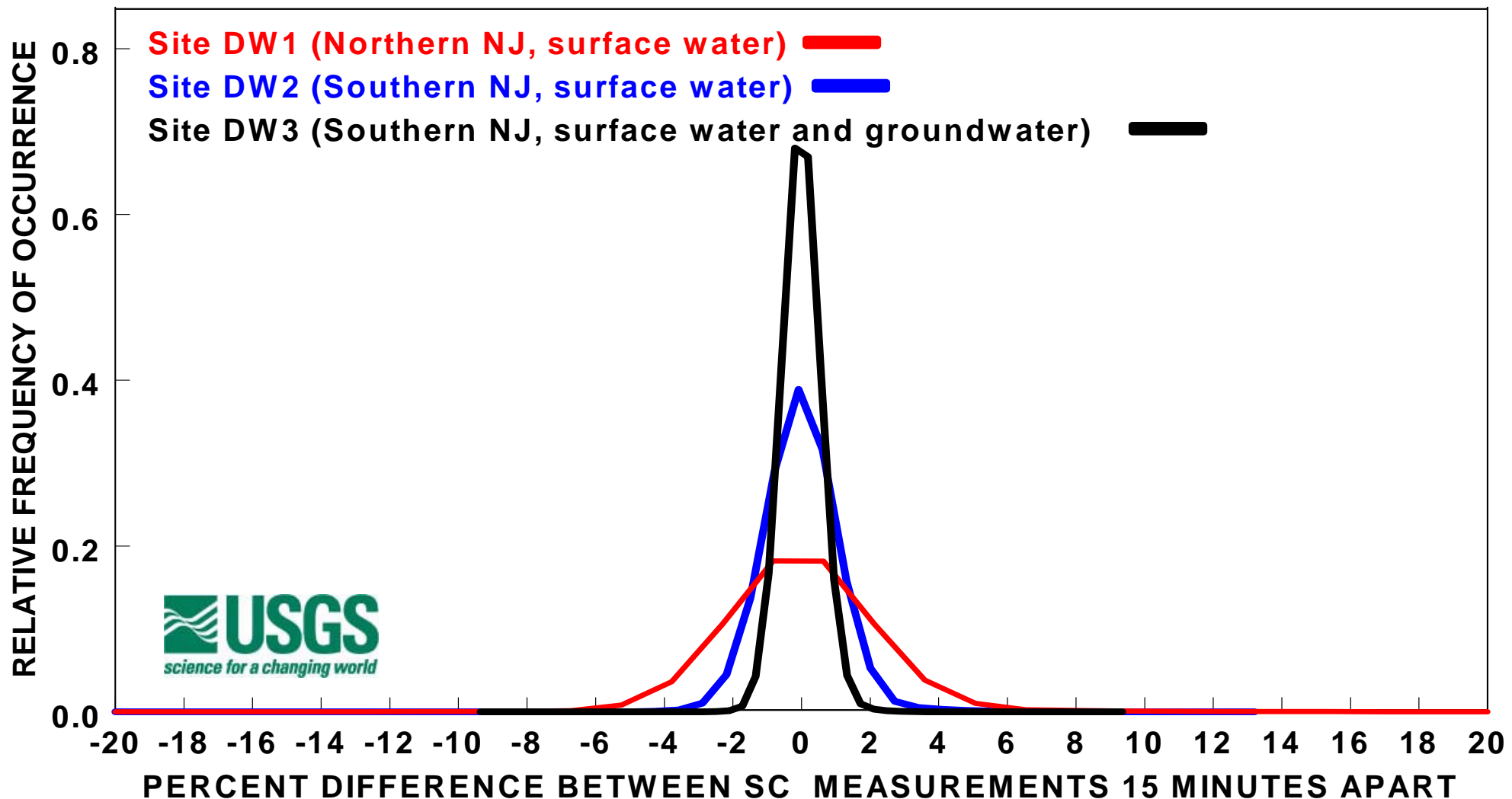
**Y axis:** Density, or relative frequency of occurrence, of a range of X values

**Evaluating density diagrams:** Relative magnitudes of density values (not the actual values) are most informative for understanding the shape of the data.

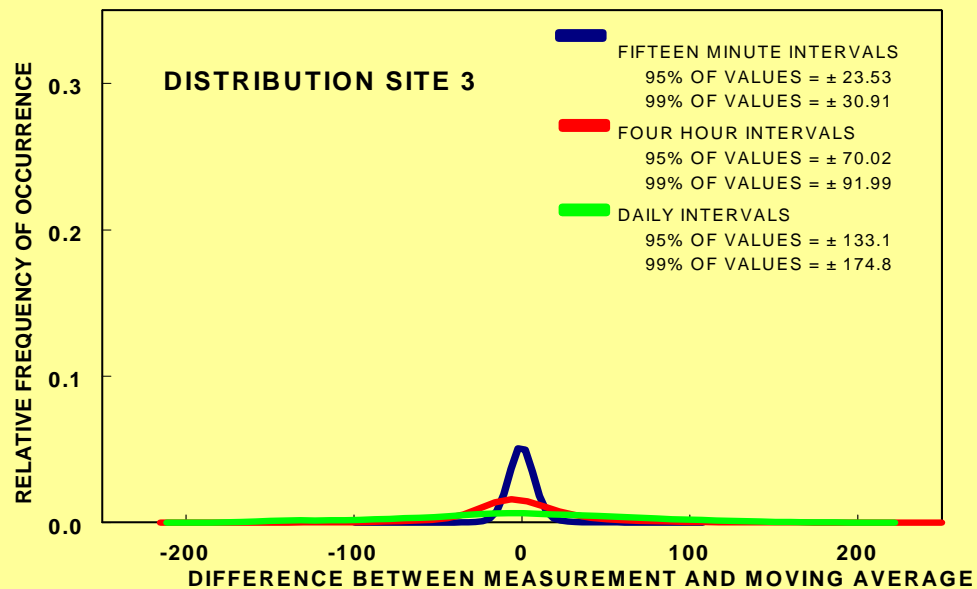
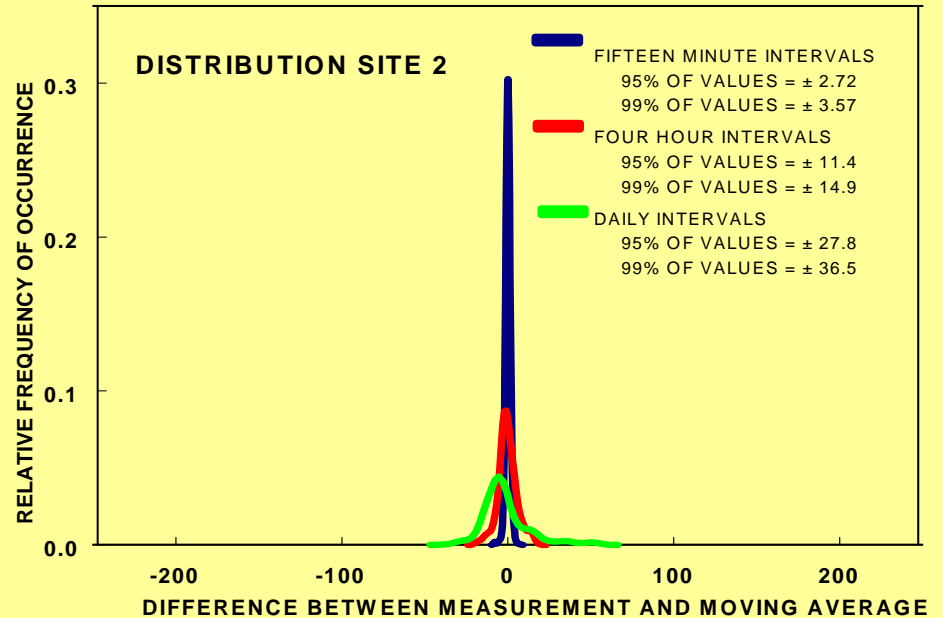
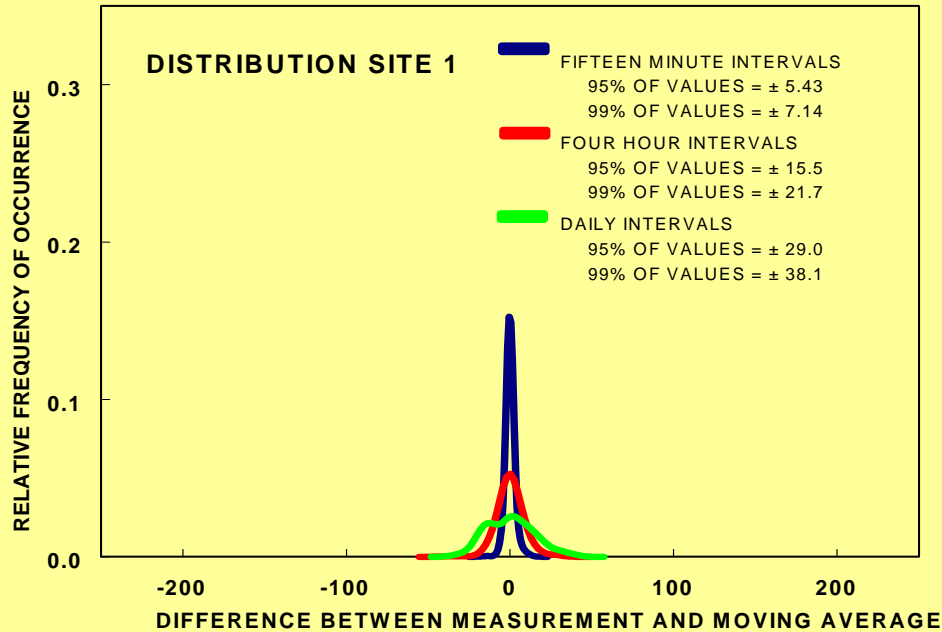


# Comparison of Temporal Water-Quality Variability Among 3 Distribution-System Sites

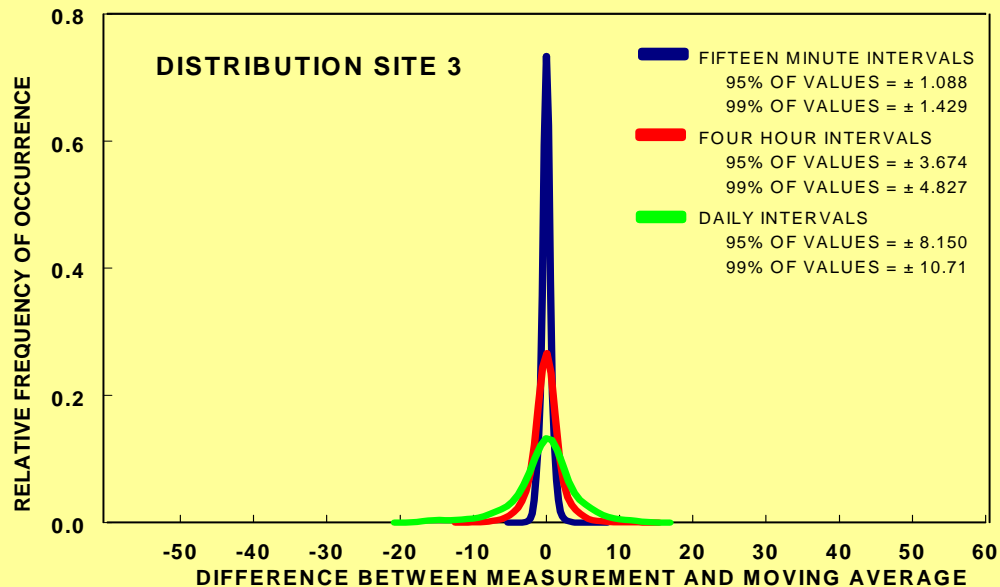
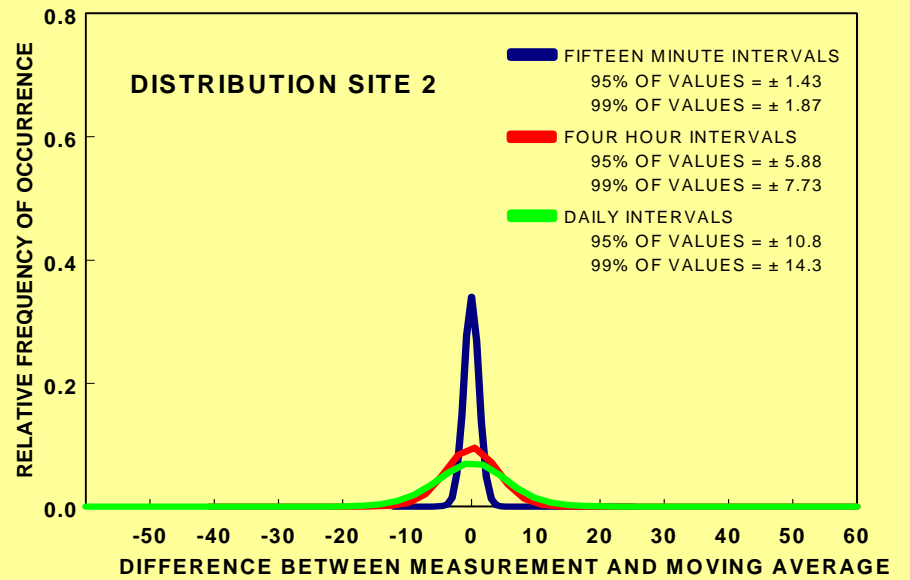
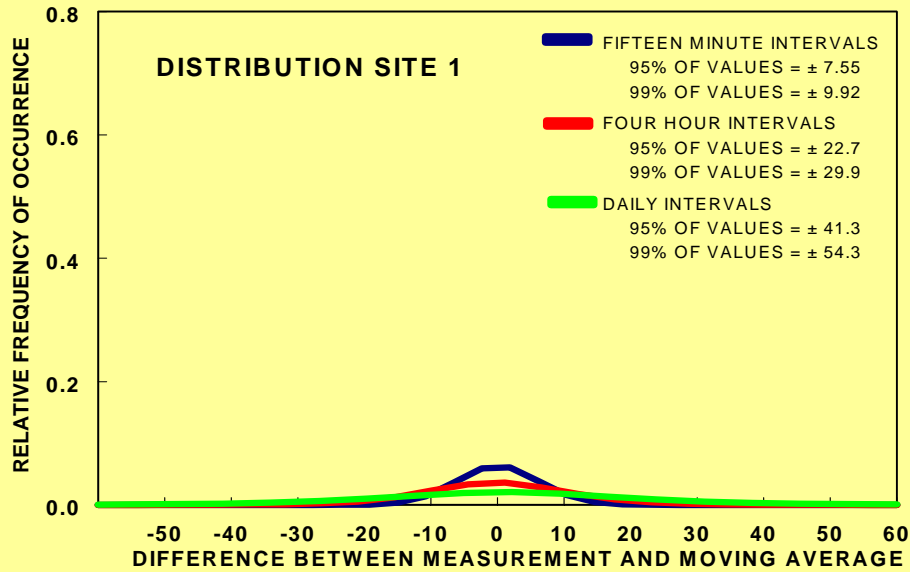
## SPECIFIC CONDUCTANCE (SC, $\mu\text{s}/\text{cm}$ )



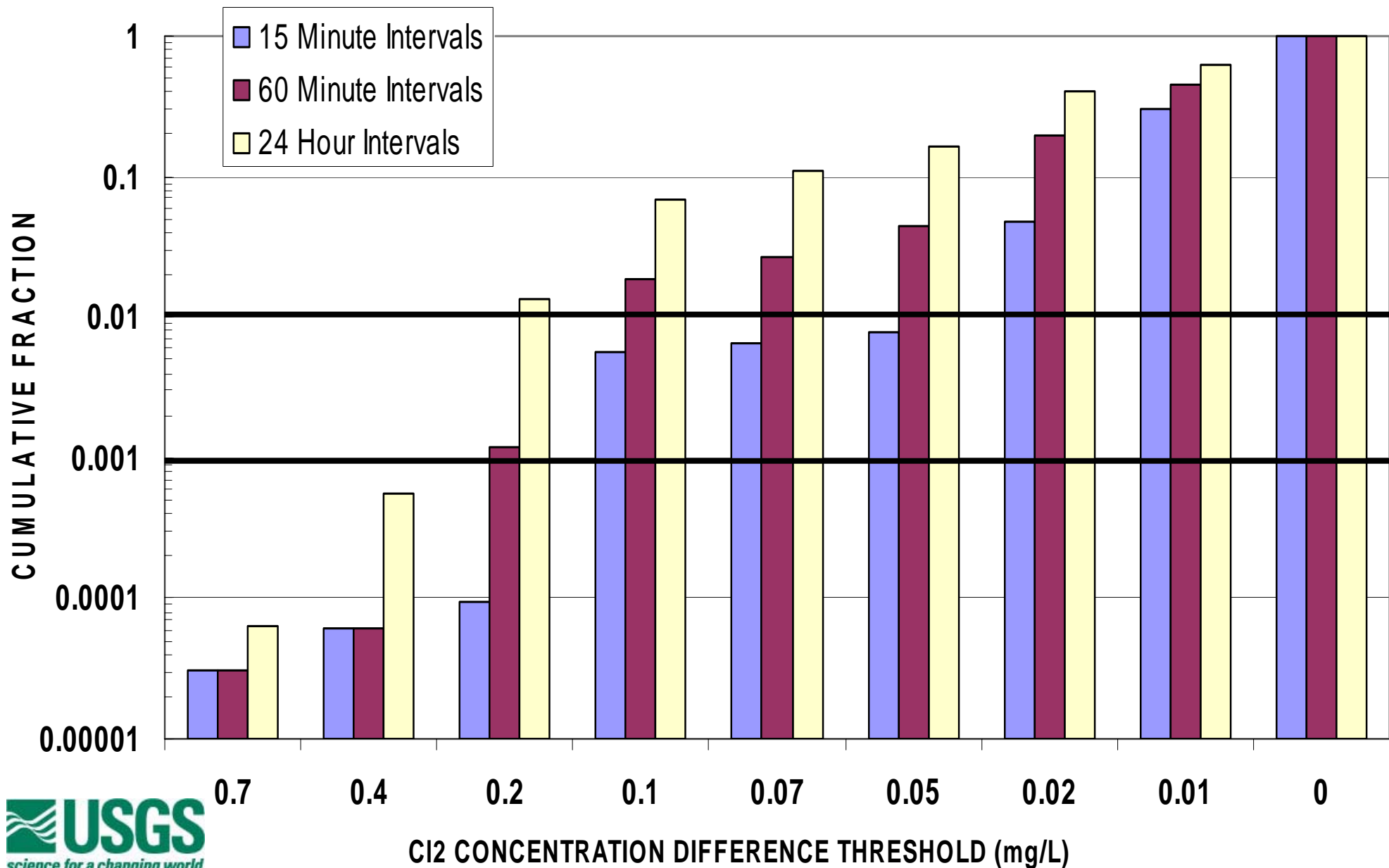
# OXIDATION-REDUCTION POTENTIAL (ORP, mV): DISTRIBUTION OF DIFFERENCES BETWEEN MEASUREMENTS AND MOVING AVERAGES OVER 3 TIME INTERVALS



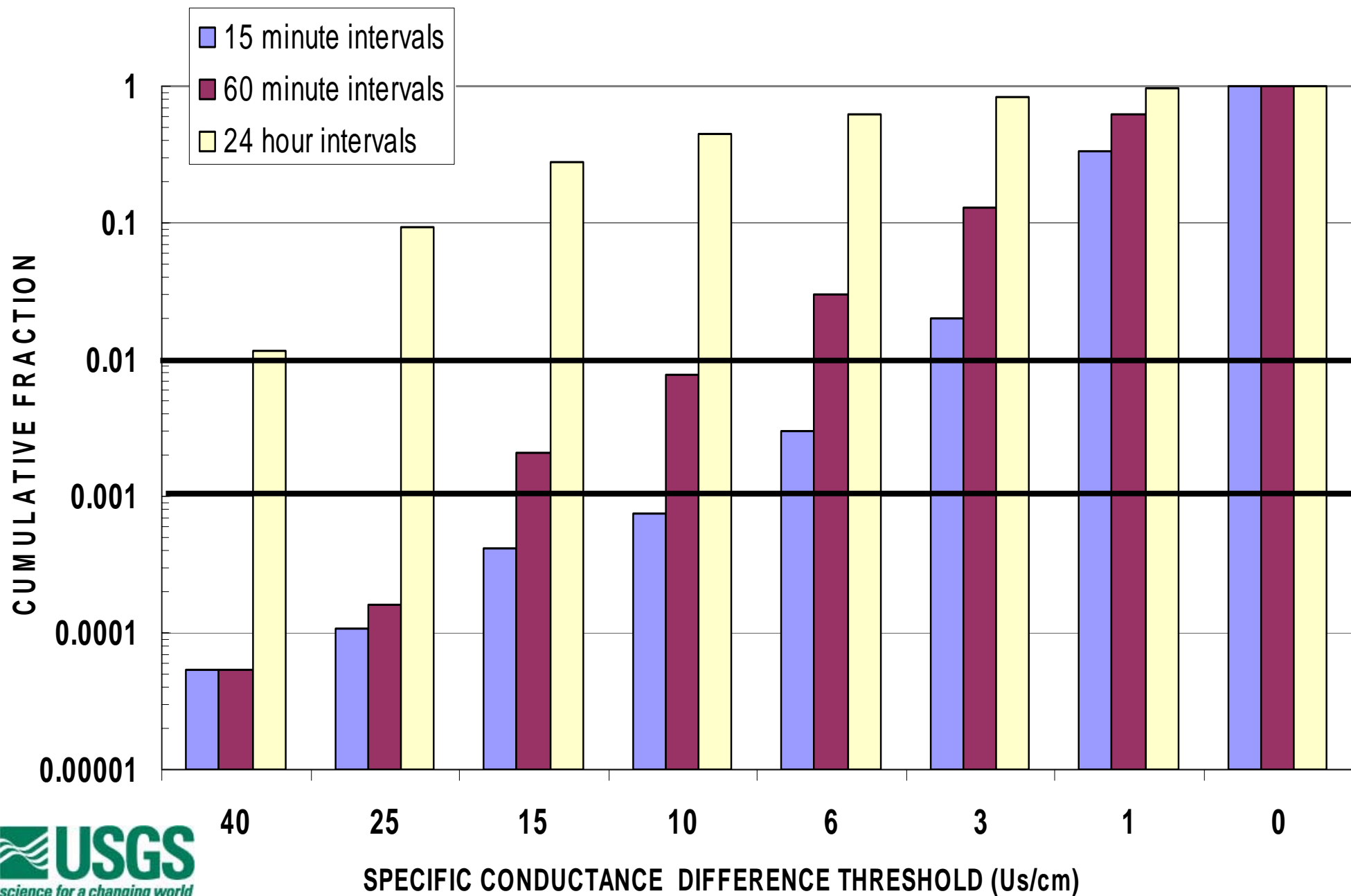
# SPECIFIC CONDUCTANCE (uS/cm): DISTRIBUTION OF DIFFERENCES BETWEEN MEASUREMENTS AND MOVING AVERAGES OVER 3 TIME INTERVALS



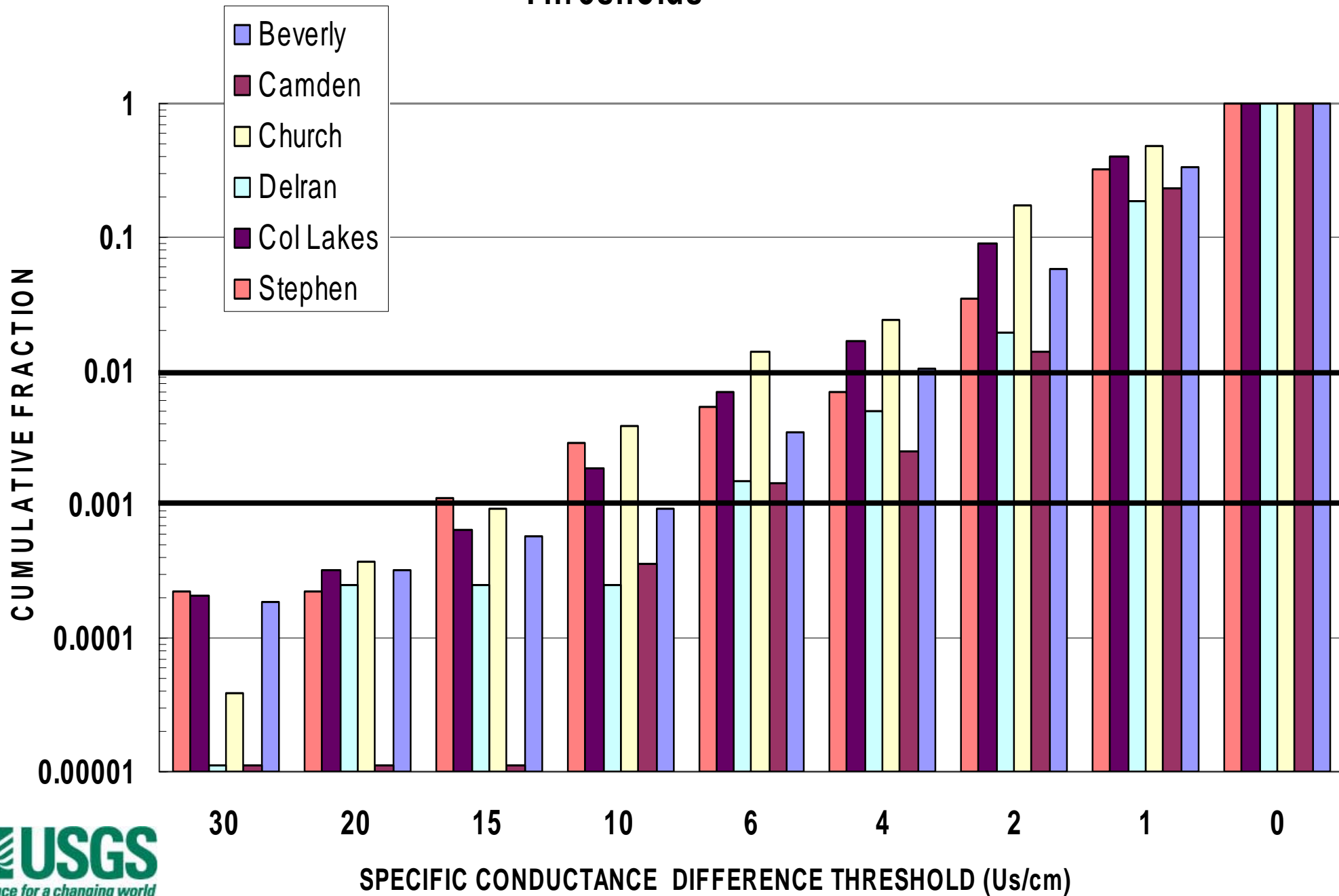
# Beverly Free Chlorine Data Through 09/07/05: Cumulative Fractions of Samples Having Consecutive-Concentration Differences At or Above Thresholds



# Beverly Specific Conductance Data Through 09/07/05: Cumulative Fractions of Samples Having Consecutive-Concentration Differences At or Above Thresholds

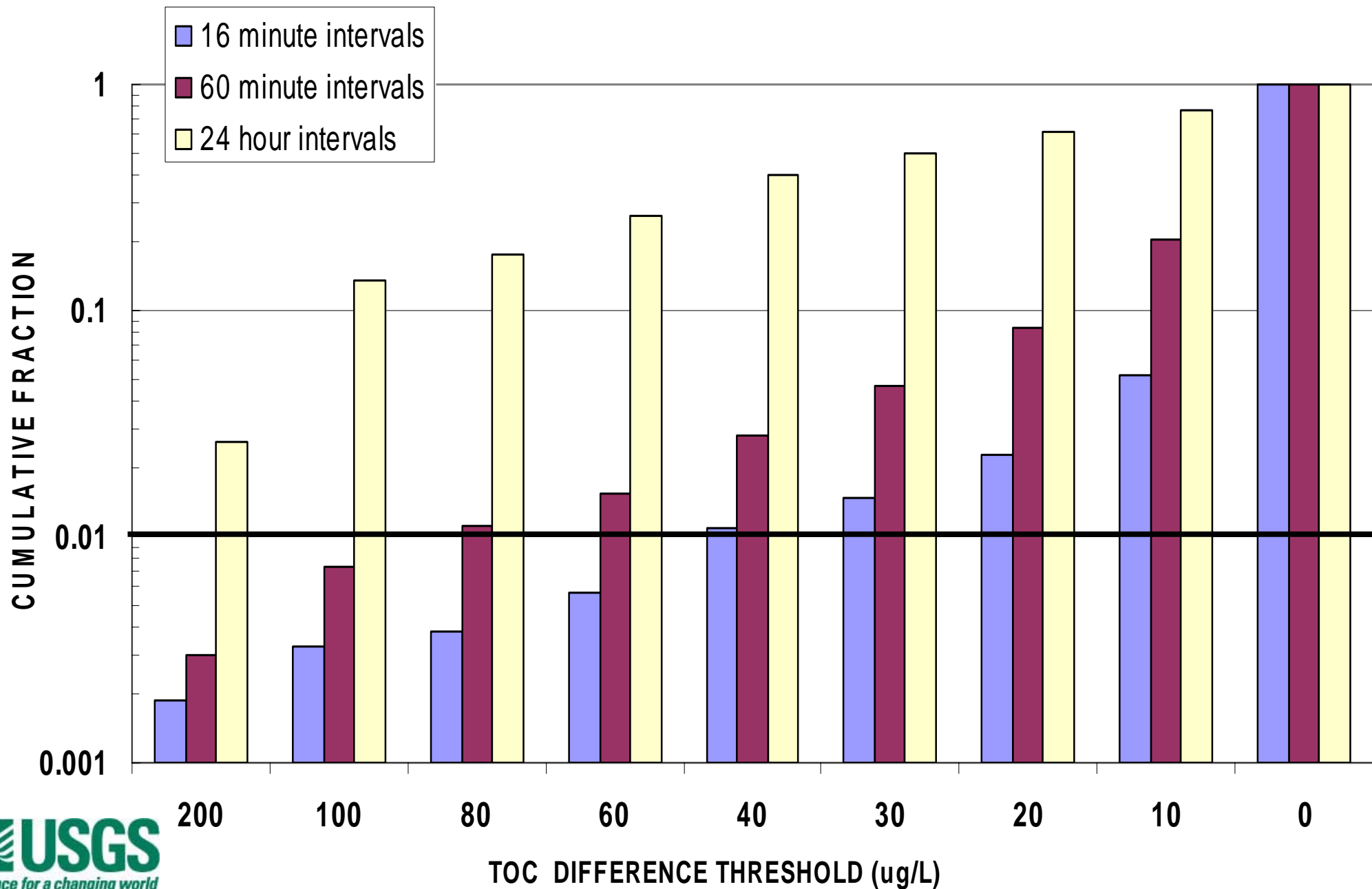


# Specific Conductance Data Through 09/07/05: Cumulative Fractions of Samples Having Consecutive-Concentration Differences At or Above Thresholds





# Beverly Total Organic Carbon (TOC) Data Through 09/07/05: Cumulative Fractions of Samples Having Consecutive-Concentration Differences At or Above Thresholds



# Long Term Program Objectives

- **Distribution system sites will serve as practical test beds for sensors and EWS components in the field**
- **Potential model calibration effort using a tracer in a sub-area of the distribution system**
- **Support collaborative efforts of the EPA/NHSRC and Sandia National Laboratories (SNL) to characterize water-quality variability for distribution systems**

# What will be learned?

- **Use of hydrologic and water-quality model to select sensor location**
- **Natural variability of water-quality characteristics in a distribution system**
- **Capabilities, limitations and maintenance requirements of sensors**

# Summary

- Overall Program Goal: Design and test a network of water-quality sensors for drinking water safety and security
- Sensors have been selected, site design and telemetry were developed, and water-quality data are being collected at 6 sites
- Real-time data will be used to determine the temporal and spatial variability of water quality in a distribution system
- ***Most important question remaining:*** Will introduction of contaminants of concern cause detectable changes in water quality?