



Université Blaise Pascal







# Remote sensing of Arctic clouds and aerosol acidification effects

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## **Importance of Arctic clouds for climate change**

- Play an essential role in the Arctic and Global climate system
- They affect the hydrologic cycle and radiation balance
- Ice clouds occur at 60-70% of the time per year (Shupe 2010)
- Their representation in models is still not validated

 Cloud feedbacks are a key source of uncertainty for GCMs' simulations related to climate sensitivity (*Dufresne et Bony 2008*)

Aerosol/cloud/radiation interactions are poorly understood

## **AWAC4 Algorithm**

Artic Winter Aerosol and Cloud Classification from Cloudsat and CALIPSO





Classification according to the sensitivity of the CLOUDSAT radar

<u>Source</u>: Grenier et al., 2009

## **AWAC4 Algorithm**

Artic Winter Aerosol and Cloud Classification from Cloudsat and CALIPSO





Classification according to the sensitivity of the CLOUDSAT radar

Based on β<sub>532</sub> and χ
Validated using ground data from Zepplin
→ Indicator of sulphate in non-cloudy aera

#### <u>Source</u>: Grenier and Blanchet, 2010

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## **The SIFI effect**

Sulphate-Induced Freezing Inhibition

 Sulphates may inhibit the freezing of coated IFN at cold T C intervals (Eastwood et al 2009)

→ may result in ice clouds with fewer and larger ice crystals with a reduced competition for the water vapor with fewer IFN (Grenier et al 2009)



## **AWAC4** Algorithm

Artic Winter Aerosol and Cloud Classification from Cloudsat and CALIPSO

Distribution Test → no evidence of a strong local SIFI effect

#### Sulphate proxy limitations

(Sensitive to other aerosol mixtures (sea salt ...), above-cloud instead of in-cloud proxy ...)

#### **Classification algorithm limitations**

(Assumptions in the radar IWC retrieval, attenuation of the lidar beam by liquid droplets, ...)

#### SIFI effect strength

(Increase in CCN caused by sulphates could mask the SIFI effect, ...)

<u>Source</u>: Grenier and Blanchet, 2010

#### Averaged over Winter 08 & 09 (D,J,F) for different layers in Arctic Aerosol index TIC-28 fraction





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## **Experimental approach**

Investigation of the SIFI effect is performed using observations from satellite and <u>airborne</u> instruments as follow :

- 1. Compare ice cloud properties before and after an anthropogenic and/or volcanic perturbation event
- 2. Investigate the chemical composition of aerosols for cloud microphysics analysis
- 3. Using the particle dispersion model FLEXPART to study pollution events (detected by CALIPSO or IASI satellites)

## **1. Compare ice cloud properties**

#### ISDAC (April 2008)

- Ice and mixed-phase arctic clouds
- Barrow-Fairbanks (Alaska)
- Aircraft Convair-580 from NRC (Canada)
- Probes: 2-DS, 2-DC, 2-DP, Rosemount Icing Detector, PCASP ...





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#### April 1, 2008 23:13:41 – 23:45:41



High [small ice crytsals] looks like a TIC-1/2A

#### April 15, 2008 00:55:40 – 01:17:24



Low [large ice crystals] looks like a TIC-2B







Arctic circle

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## 2. Investigate the chemical composition of aerosols

#### PCASP size range [0.12-3µm]

300cm<sup>-3</sup> threshold clean vs polluted (*Peng et al. 2002*)

#### Future work :

The IFN and acidic information of aerosols are needed to make a better assessment on IN nucleation capability.



#### April 1, 2008 23:13:41 – 23:32:11

# 3. Using the particle dispersion model (Hysplit, Flexpart)

### To study origin of observed air masses to be linked with pollution (SO<sub>2</sub>) sources

April 15, 2008 00:55:40 - 01:17:24



Examination of the CALIPSO satellite tracks, which intersects the back trajectories in the region away from the airborne measurements

In april 2008, sustained eruptive activity was recorded at Cleveland and Veniaminof in the Aleutian Islands.

## Conclusion

ISDAC measurements are considered as:

- Ice clouds in clean conditions
- Ice clouds occurred within the polluted conditions

These clouds occurred in various environmental conditions (e.g. changing RH<sub>Ice</sub> and T<sub>a</sub> C)

Results appear to be consistent with the SIFI effect's hypothesis (e.g. increasing acidic coating results in decreasing IN but increasing snow)

#### Further work is needed and will include:

 $\rightarrow$  Additional data set related to the chemical composition of aerosols.

→ Examination of the CALIPSO satellite observations, which intersect the airmass back trajectories and will provide additional information on aerosol type/acidifications.

→ New observations from PIC3 project will be used for further validations of SIFI effect.

## References

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## **Sulphates form and properties**



Represent the High Arctic dominant species in terms of abundance and mass.

Are hydrophilic : attract water molecules to form **solution droplets**, hence are good cloud condensation nuclei (CCN).

Are scatterers : reflect solar light and laser beams.

Internally mix : covers 80 % of other aerosols.

Prevents ice-forming nuclei (IFN) from initiating ice formation. This is termed the sulphate-induced freezing inhibition or SIFI effect.



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#### **SIFI from laboratory experiments** (ice nucleation modes) no IFN YES needed vapor Vapo no droplet Deposition YES needed vapor Contact Immersion MAYBE / freezing vapor Condensation Vapor **7**5 / freezing ?

vapor

## **Onset of Ice Nucleation on Kaolinite Particles Coated** with (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>



• All experiments obtained with a surface area of  $10^{-4}$ to 10<sup>-3</sup> cm<sup>2</sup>. o At 245 K and 240 K,  $(NH_4)_2SO_4$  coated kaolinite particles are poor ice nuclei. • At 236 K, the coated particles are just as efficient nucleating ice at as uncoated kaolinite particles. • Why the big difference  $H_2SO_4$ between and  $(NH_{4})_{2}SO_{4}$  coatings ? o May be related to the phase of the coatings.