

Vibrational and Electronic Energy Levels of Polyatomic Transient Molecules. Supplement B

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(Received 7 March 2002; accepted 9 May 2002; published 18 February 2003)

A summary is presented of recently published, critically evaluated experimental vibrational and electronic energy level data for approximately 1700 neutral and ionic transient molecules and high temperature species possessing from three to sixteen atoms. Although the emphasis is on species with lifetimes too short for study using conventional sampling techniques, there has been selective extension of the compilation to include data for isolated molecules of inorganic species such as the heavy-metal oxides, which are important in a wide variety of industrial chemical systems. Radiative lifetimes and the principal rotational constants are included. Observations in the gas phase, in molecular beams, and in rare-gas and diatomic molecule matrices are evaluated, and several thousand references are cited. The types of measurement surveyed include conventional and laser-based absorption and emission techniques, laser absorption with mass analysis, and photoelectron spectroscopy. © 2003 by the U.S. Secretary of Commerce on behalf of the United States. All rights reserved. [DOI: 10.1063/1.1497629]

Key words: *ab initio* calculations; density functional calculations; electronic energy levels; emission spectra; experimental data; free radicals; gas phase; high-temperature molecules; infrared spectra; laser-excited fluorescence; matrix isolation; molecular ions; photoelectron spectroscopy; polyatomic molecules; radiative lifetimes; Raman spectra; rotational constants; transient molecules; ultraviolet absorption; vibrational energy levels.

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1. Introduction

Most chemical processes—including not only laboratory and industrial chemical syntheses but also those which occur in flames, propellant systems, the initiation of energetic materials, atmospheric pollution, chemical vapor deposition, and plasma processing—consist of a complicated sequence of interrelated reactions in which neutral and charged molecular fragments play essential roles. Although these fragments are present in only very small concentration, they are highly chemically reactive. If a specific molecular fragment is removed from the system, as by introducing a scavenger molecule, the reactions in which that fragment participates stop. Other parts of the overall process continue, resulting in very significant changes in product distribution and yield.

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In the early studies of complex chemical processes, it was necessary to postulate mechanisms involving such transient intermediates, present in concentrations too small for direct detection. Conventional end product analysis aids in the selection of suitable mechanisms, but generally does not yield a complete description of the system. Consequently, the improvement of industrial chemical processes often is achieved by semiempirical experimentation. The determination of the detailed chemical mechanism would, in turn, permit the development of rational strategies for removing undesired products and enhancing the yield of the desired species.

In recent years, there has been great progress in the development of techniques suitable for monitoring chemical reaction intermediates. Molecular spectroscopy is especially well suited to this task. Optical detection can be used not only for gas-phase measurements, but also for studies of processes which occur on surfaces or in the condensed phase. It also permits remote sensing, an important advantage. A wide variety of recently developed laser-based spectroscopic detection schemes are not only highly sensitive but also space and time specific. Although the development of spectroscopy-based diagnostics for chemical reaction systems is in its infancy, already the laboratory application of sophisticated sampling and observation techniques has yielded a wealth of vibrational and electronic spectral data for reaction intermediates.

For many years, the most important source of vibrational and electronic energy level data for small polyatomic reaction intermediates was the compilation of spectroscopic data for small polyatomic molecules (3–12 atoms) given by Herzberg.¹ To meet the need for an updated, critically evaluated compilation, a series of publications^{2–4} have appeared in this journal, culminating in the publication in 1994 of a monograph⁵ which presented evaluated spectral data for more than 1550 small polyatomic transient molecules. In 1998, a supplement to that monograph appeared,⁶ adding data for approximately 900 molecules, more than half of which were not in the monograph. Electronic band origins, vibrational fundamentals in the ground and excited electronic states, and radiative lifetimes were given in both the monograph and its supplement. To aid in spectral identification, the principal rotational constants were also given in cm^{-1} to three decimal places. The vibrational and electronic energy level data from these tables forms a part of the material available from the Internet-based NIST Chemistry WebBook (<http://WebBook.nist.gov/chemistry/>). The WebBook format permits relatively frequent updates and several different types of data search.

The rapid growth in the scientific literature concerned with the spectroscopic study of transient molecules and with their detection in chemical reaction systems continues. Since the late 1997 cutoff in the data evaluation for the 1998 supplement,⁶ substantive new spectroscopic data have been published for many species included in it, and the first data have become available for more than 1200 other molecules. Progress in the spectroscopic characterization of transient species produced by the reaction of metal atoms with oxygen

and other small molecules has continued. This paper attempts to provide a comprehensive, critically evaluated summary of these new data for the vibrational and electronic energy levels of small polyatomic transient molecules, in order to support further research and technologies such as plasma processing and chemical vapor deposition.

2. Scope of Review

This review provides a critical evaluation of the vibrational and electronic spectral data available at the end of 2001 (or, for a few species, in early 2002) for approximately 1700 small polyatomic transient molecules. Of these, about 1225 are new; the remaining approximately 475 molecules were included in the monograph and/or the earlier supplement but have since been more extensively studied. Despite this rapid progress, many gaps remain in our knowledge of the energy levels of the species represented, and many new and potentially important transient molecules remain to be discovered.

Data have been selectively included for some molecules which are important in environmental and industrial chemical reaction systems but which can be studied only with difficulty using conventional sampling techniques because of the ease with which they decompose, rearrange, or polymerize. Also included are data derived from spectra of many high-temperature species, such as metal oxides, studied in molecular beams and in rare-gas matrices. These species include a number of binary and some ternary complexes involving a metal atom and a small inorganic molecule (H_2 , N_2 , O_2 , CO , NO). Examples of unusual chemical bonding, including side bonding to produce a cyclic intermediate, are found. Such complexes are expected to play important roles in catalysis. Because of the ubiquity of atmospheric impurities, these complexes may contribute widely to the chemistry of metal-atom reaction systems. A number of other molecules included in this database are among the first inorganic species to exhibit aromatic behavior. Moreover, there is a growing number of entries for small molecular cations and anions.

3. Types of Measurement

Studies in the gas phase offer the potential for the most precise, detailed measurements. Because of the high chemical reactivity of transient molecules, it is difficult to obtain gas-phase infrared spectra of them. The well known advantages of Fourier transform infrared measurements, coupled with sophisticated digital data handling procedures, have permitted the acquisition of gas-phase survey spectra for a number of transient molecules. Diode lasers and other laser-based techniques with limited tunability have been used to obtain high resolution spectra of individual vibrational transitions of these species.

During the past few years, the power of *ab initio* and density functional calculations for obtaining ground-state vibrational frequencies has grown dramatically. Relatively so-

phisticated calculations are now feasible for molecules in the size range covered by this compilation. Recently, Byrd and co-workers⁷ have compared the positions of 84 ground-state vibrational fundamentals for 33 small free radicals calculated at several different levels (e.g., Hartree–Fock, MP2, CCSD, B3LYP) using a variety of common basis sets with the experimentally determined positions. The results are expressed as averages of the numerical and absolute deviations and as the standard deviation for each calculational procedure. These deviations generally amount to several percent. Since anharmonicity is expected to be a major contributor to these deviations, Byrd and co-workers also attempted a comparison for the small set of frequencies for which the harmonic values are available. They reported the results only for the most sophisticated basis set, cc-pVTZ, used in their study. Their results for three different calculations are summarized in Fig. 1. The deviations obtained at the CCSD level and using the B3LYP density functional procedure are approximately 4%. The more sophisticated CCSD(T) calculations overcorrected for the anharmonicity by approximately 0.9%. Moreover, the spread in the deviations, expressed by the standard deviation, was greater than for the other two types of calculation. Of course, these conclusions may change as a larger data set of experimentally determined harmonic vibrational frequencies becomes available.

Although vibrational frequencies of ground-state molecular ions have frequently been estimated from structure in Rydberg transitions of the parent neutral species, such data are not included in this compilation, since many of these transitions have residual valence character, resulting in significant variations in vibrational frequencies from one Rydberg state to another.

As in the earlier compilations, spectral data obtained in rare gas and small covalent molecule matrices are included. The application of matrix isolation sampling for the stabili-

zation and spectroscopic study of neutral and ionic reaction intermediates has recently been reviewed.⁸ Because nitrogen and the rare gases are transparent through the entire infrared spectral region, matrix isolation measurements provide a potentially valuable survey tool. In these matrices, infrared absorptions are typically sharp, with half bandwidths between 0.1 cm^{-1} and 1 cm^{-1} . Rotational structure is, with few exceptions, quenched. Multiple trapping sites occur, often resulting in the appearance of several absorption maxima—usually one or two of which predominate—over a range of a few cm^{-1} .

Matrix shifts for covalently bonded molecules trapped in solid neon or argon often are quite small. A comparison⁹ of the positions of the ground-state vibrational fundamentals of over 200 diatomic molecules observed in the gas phase and in nitrogen and rare-gas matrices has shown that, typically, the smallest matrix shift occurs for neon matrix observations, with successively greater matrix shifts for the heavier rare gases and for nitrogen. This comparison has been updated—with similar conclusions—in the recent review⁸ for diatomic oxides, an important class of molecules for which considerable new data have become available since the earlier comparison. Except for very weakly bonded molecules and for the alkali metal and Group IIIa halides, matrix shifts of most diatomic molecules isolated in solid argon are less than 2%, and shifts for the molecules isolated in solid neon are even smaller. Similar conclusions result from the comparison of neon- and argon-matrix shifts for the ground-state vibrational fundamentals of larger molecules.⁸ Figure 1 includes a comparison of the observed average deviations (by analogy with the computational deviations, expressed as matrix value minus gas-phase value, in percent) for all of the the ground-state fundamental vibrations of transient molecules with from 2 to 16 atoms trapped in solid neon and argon for which the positions of the gas-phase band centers have been determined. These comparisons included 168 fundamental vibrations for the neon matrix and 426 for the argon matrix. The numerical average deviation of the neon-matrix band centers from their gas-phase counterparts is very small, and both the absolute average deviation and the standard deviation are well within 1%. The deviations for argon-matrix measurements are somewhat larger, and the average fundamental frequency is approximately 0.2% smaller than the gas-phase value. As is shown in Fig. 1, the average deviations obtained from the *ab initio* and density functional calculations are several times as large. While the results of such calculations are demonstrated to be a useful guide, where neon- or argon-matrix values for the ground-state vibrational frequencies of small transient molecules have been determined they are likely to provide a significantly better approximation to the gas-phase band centers. Interaction between both gas-phase and matrix-isolation spectroscopists and scientists developing new computational strategies for molecules with unpaired electrons, for which configuration interaction is important, promises to be mutually helpful.

For molecular ions, neon is the matrix of choice. Data for many small cation species trapped in solid neon are consistent with the matrix shift generalizations given above. How-

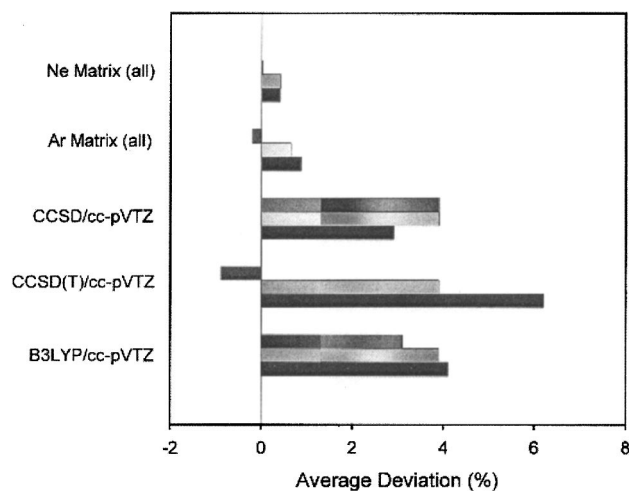


FIG. 1. Comparison of average deviations (matrix—gas, %) of transient molecules from 2 to 16 atoms trapped in neon and argon matrices with average deviations (calculated—observed, %) reported by Byrd *et al.*⁷ for harmonic vibrational fundamentals of small free radicals. In each group, top=num. av., middle=abs. av., and bottom=std. dev.

ever, some ions react with even the lighter rare gases. An extreme example is provided by HF^+ , the ground-state vibrational fundamental of which is shifted by some 700 cm^{-1} because of reaction to form NeHF^+ .¹⁰ Vibrational frequency shifts which result from proton sharing between rare-gas atoms or clusters and small protonated molecules have recently been reviewed by Bieske and Dopfer.¹¹ Because the proton affinity of helium is 80% as great as that of neon, such shifts can be appreciable even for helium. Although specific examples are not yet available, it is anticipated that some nonprotonated species may also interact strongly with neon. Polarization and charge-transfer interactions become successively more important for molecules isolated in the heavier rare gases. Moreover, charge delocalization sometimes occurs for ionic species trapped in the rare gases.^{12,13} The anomalously large matrix shift for ν_3 of ClHCl^- may be attributed to this phenomenon. Only a few comparisons are available for molecular cations observed both in the gas phase and trapped in solid argon. The absolute values of almost half of the observed matrix shifts are greater than 1%. Very few comparisons are possible for molecular anions. Where the frequency of a cation or an anion vibrational fundamental is similar in both argon- and neon-matrix observations, it is suggested that the deviation from the gas-phase band center is small. However, data are insufficient to confirm this generalization. A number of molecular anions have been generated in rare-gas (usually argon) matrices by charge transfer between a precursor molecule and an alkali metal atom. Recent studies of such species as CO_2^- and SO_2^- generated instead by photoionization and/or Penning ionization and trapped in solid neon indicate that shifts on the order of 50 cm^{-1} may be attributed to the relatively strong interaction of the anion with the nearby alkali metal cation. On the other hand, when the uncharged molecule has a relatively large electron affinity, as is true for C_2 and for NO_2 , charge transfer occurs at a relatively great separation, and a substantial fraction of the anion population may be trapped in sites in which interaction with the alkali metal cation is minimal.

Matrix shifts for vibrations associated with ionic bonds are often considerably larger than those associated with uncharged molecules or with intramolecular vibrations of molecular ions. Criteria for inclusion of data for species which include an ionic bond are exemplified by the selection process for the heavy-metal oxides. Often the stable dioxide structures include an M^+O_2^- species with significant covalent bond character for the attachment of M^+ , as evidenced by a substantial shift in the O_2^- stretching fundamental as M^+ is varied. Such species are included in the compilation. On the other hand, there is little evidence for substantial metal-atom participation in the vibrations characteristic of the O_3^- moiety of M^+O_3^- . Accordingly, spectral data are given for O_3^- , but not for M^+O_3^- .

Data are beginning to appear for molecules trapped in a hydrogen matrix. Insufficient information is available for generalization on the magnitude of matrix shifts in this medium. For the few species heretofore studied, including several transient molecules present in this compilation, the ma-

trix shifts have been comparable to those in a neon matrix.

Many other matrix materials have also been employed for spectroscopic studies. However, complications due to reaction or to relatively strong interaction (e.g., hydrogen bonding) of the transient molecule with the matrix frequently occur. Therefore, observations in such media as solid hydrocarbons and aqueous solutions and studies of condensed reaction products without an inert carrier have been excluded.

Because of the prevalence of electronic emission spectra and the sensitivity, rapid time response, and cumulative detection capability of the photographic plate in the visible and ultraviolet spectral regions, the study of the electronic spectra of gas-phase reaction intermediates has a comparatively long history. Flash photolysis has permitted the preparation of relatively high concentrations of transient species. Many electronic band systems of gas-phase transient molecules have been discovered through flash photolysis studies. More recently, a wide variety of laser-based techniques have also been used for electronic spectral observations, often with exceptionally high detection sensitivity. The spatial configuration of the laser beam makes possible crossed molecular beam-laser beam studies, providing an extremely powerful tool for studies of the energy levels of molecules in molecular beams. Laser-based techniques show great promise for application in the development of diagnostics for chemical reaction intermediates in the environment, the laboratory, and the industrial plant. Laser studies may be broadly classified according to whether the interaction of the molecule with the laser beam(s) is followed by photon or mass detection. Photon-based observations are amenable to remote sensing applications. Because pulsed lasers offer an exceptionally wide range of time specificity, they are very useful for determining radiative lifetimes and rates of elementary chemical reactions.

Much valuable information on the energy levels of molecular cations has been obtained from photoelectron spectroscopy. These tables include selective coverage of the voluminous literature on photoelectron spectroscopic measurements. The number of stable molecules which possess more than six atoms for which photoelectron spectra have been reported is too great to permit the inclusion of low-to-moderate resolution photoelectron spectral data for most molecular cations with more than six atoms. Those who need such data for larger molecules may find the reviews by Turner *et al.*,¹⁴ Rabalais,¹⁵ and Kimura *et al.*¹⁶ helpful. Several criteria are important in determining whether a given reference should be included in the present work. The first of these is resolution. In the few instances in which high resolution photoelectron data are available, these data are heavily weighted. Where direct spectroscopic observation is possible, the measurements generally are of considerably higher precision than are the photoelectron data, which are then omitted from the tables. A second criterion is the availability of adiabatic ionization potentials. In order to obtain information on the positions of electronic transitions from photoelectron spectral data, it is necessary to subtract the first ioniza-

tion potential from the energy of the photoelectron band. Where there is little change in the molecular geometry in the transition, the difference between the vertical ionization potentials gives a reasonable approximation to the position of the electronic transition. However, this is not the general case. Therefore, priority is given to papers which include adiabatic ionization potentials.

For most photoelectron spectroscopic transitions, structure has not been resolved. Many of these states are dissociative. Further information on the dissociation products can be obtained from values of the appearance potentials for various products in photoionization studies on the parent molecule. Such studies are beyond the scope of this review. The tables of ionization and appearance potentials by Lias and co-workers^{17,18} (also available in the NIST Chemistry WebBook) constitute a valuable source of information on the appearance potentials of photofragments.

The range of tunability of visible and ultraviolet lasers, like that of infrared lasers, is limited. Therefore, a preliminary survey using conventional gas-phase and/or matrix-isolation spectroscopic studies is often desirable. A comparison of the positions of the electronic band origins of diatomic molecules in the gas phase and in rare-gas and nitrogen matrices has been published.¹⁹ As in the determination of ground-state vibrational energy levels, neon is the matrix material of choice, with a sharp maximum at 0.0% in the distribution of matrix deviations for valence transitions of covalently bonded molecules. This generalization is supported by the recent study by Maier²⁰ of the electronic spectra of mass-selected carbon-chain species. In that work, neon-matrix spectra were used as a guide in the search for π - π electronic transitions of the gas-phase species. For these transitions of neutral and charged species of formula C_n , HC_n , HC_nH , HC_nN , and NC_nN , Maier found that almost all of the neon-matrix shifts were to longer wavelengths, with a magnitude less than approximately 150 cm^{-1} . In argon-matrix observations, most such band origins are shifted by less than 2% from the gas-phase values. At the somewhat higher temperatures often used for electronic spectral observations in matrices of the heavier rare gases or of nitrogen, relatively broad phonon bands become prominent. The blueshift of the phonon maximum from the zero-phonon line in absorption measurements, and the red shift in emission measurements, typically amount to approximately 1%–1.5%. Rydberg transitions of molecules in matrices often are greatly broadened and experience much larger shifts. Further details of the behavior of electronic transitions of matrix-isolated molecules have previously been discussed.^{3,20,21}

The radiative lifetime of a relatively large molecule isolated in a rare-gas matrix is frequently related to the radiative lifetime in the gas phase by a simple refractive index correction.²² In a neon matrix, such a correction typically decreases the radiative lifetime by about 15%. For such molecules, often intramolecular mechanisms for nonradiative energy transfer are available both in the gas phase and in the matrix. On the other hand, the density of excited states is

much lower for small molecules, and matrix shifts may alter perturbation interactions between strongly coupled electronic states, providing a path for nonradiative deactivation. In this circumstance fluorescence, which is prominent in the gas phase, may even be completely quenched in the matrix.

4. Guide to the Compilation

The goal of this paper is to extend the previously published compilation⁵ of critically evaluated vibrational and electronic spectroscopic data for small polyatomic transient molecules. The literature has been surveyed through December 2001; only limited addition of more recent data has been possible. Unfortunately, with a few exceptions it is not possible to include data for stable molecules. However, the spectra of many of these species are relatively well established, and sources of data such as the tables of Herzberg¹ and Shimanouchi²³ remain extremely useful. At the time of those compilations, it was difficult to ascertain the positions of band centers for very prominent fundamentals of molecules with dense rotational structure, such as the small inorganic halides. For such absorptions, it is important to seek more recent high resolution infrared measurements. In obtaining spectral identifications with the help of the present compilation, it is crucial that the possible contribution of the absorptions or emissions by a stable molecule also be considered.

Considerable effort has been expended to provide a critical evaluation of the data. However, for many species the available data are meager. The identities of some species have been proposed on the basis of chemical evidence. While such evidence may be quite compelling, it is not definitive. Many examples could be cited in which a spectrum was later reassigned to characteristic impurities in the sample. Where chemical evidence has provided a reasonable basis for the assignment of vibrational or electronic bands to a transient molecule, data have been included in this compilation, in the hope that further testing of the assignment will be facilitated.

While every effort has been made to make these tables as complete as possible, for various reasons omissions do occur. There remains some selectivity in the coverage of electronic spectral data for larger molecules. It is planned to support this database, with further selective extension, by the preparation of additional supplements. Data from the earlier tables have sometimes been omitted from this paper because more recent data dictate a reassignment or because there has been a subsequent refinement. An important example of this latter situation is the replacement of low resolution photoelectron spectral data by spectroscopic studies with appreciably higher resolution and greater precision. To check on the possibility of reassignment, check the Molecule Index (Sec. 9) to see whether the previously listed molecule is still included. Candidate molecules or energy levels may also have been inadvertently omitted. The NIST Chemistry WebBook is inherently more responsive than any archival publication can be, and should be checked for more recent information. Suggestions of additions or needed revisions to the data to be

included in subsequent extensions of this database are welcome, as are inquiries regarding new data added after the publication cutoff data for this compilation.

Molecular formulas are used in this compilation. In order to permit a compact index, an attempt has been made to provide as much structural information as possible in a minimal amount of space. This restriction is especially severe for larger molecules. The following formula abbreviations have been used:

<i>br</i>	bridged
<i>cyc</i>	cyclic. If parentheses follow, only the atoms enclosed in them are included in the ring.
<i>c</i>	<i>cis</i>
<i>t</i>	<i>trans</i>

Where heavy isotopic peaks are resolved, data are given for the most abundant isotopic species (e.g., ${}^7\text{Li}$, ${}^{11}\text{B}$, ${}^{35}\text{Cl}$, ${}^{79}\text{Br}$).

As in the earlier compilations, the tables are grouped by the number of atoms in the molecule and, secondarily, by the number of hydrogen atoms present. Molecules within a given section of the tables are arranged in the order of increasing number of valence electrons. For species with the same number of valence electrons, molecules with a simple chain of three heavy atoms are listed in the order, first, of the number of valence electrons in the central atom of the chain and, second, of the row which this atom occupies in the Periodic Table. For larger molecules, the sequence is somewhat arbitrary, but criteria of increasing molecular size and grouping in the Periodic Table (e.g., the order N, P, O, S) are used. Halogen-substituted species are placed immediately after the related hydrides. Data for molecules related to benzene are presented in a separate section. As in the earlier tables, data are included for both the normal and the fully deuterium-substituted molecule. However, only the hydrogen-containing species is listed in the index.

The heading for each electronic state gives its symmetry, the point group to which the molecule belongs in that electronic state, and, where available, references to the determination of a quantitative molecular structure. For C_{2v} molecules, there is potential ambiguity in the definition of the molecular symmetry axes. The convention in which the x axis is chosen perpendicular to the plane of the molecule, recommended by the Joint Commission for Spectroscopy of IAU and IUPAP,²⁴ has been adopted. Often this has required the interchange of published assignments of energy levels with B_1 and B_2 symmetry.

Most authors of papers on photoelectron spectroscopy have proposed assignments for the various photoelectron bands, using arguments based on molecular orbital theory and often on semiempirical or *ab initio* calculations. These assignments have been included in the present tables. Where several conflicting assignments have been given in the literature, an attempt has been made to choose the most satisfactory one. Generally, the assignments of photoelectron spectra have been made with the presumption that the point group to which the molecular cation belongs is the same in all of its

excited states. Structural data for these excited states are extremely rare. Therefore, the molecular point group which has been adopted in the analysis of the photoelectron spectrum is given in these tables. In practice, it is likely that there is some variation in excited-state molecular symmetries. Thus, a bent molecular ion may become linear in some of its excited states. For highly symmetric species, Jahn–Teller distortion may reduce the molecular symmetry.

The energy of the electronic transition follows the state designation and symmetry information. Where possible, T_0 , the energy separation between the electronic energy level of interest and the ground electronic, vibrational, and rotational states of the molecule, is given. However, where only low resolution data or photoelectron data are available, often only band maxima have been given in the literature. With photoelectron data, T_0 is derived by subtracting the value of the first ionization potential from that of the higher ionization potential which corresponds to the state of interest. When data for the first adiabatic ionization potential are available, the footnote phrase “from vertical ionization potential” implies that the first adiabatic ionization potential is known but that the higher ionization potential is measured to the peak maximum; the phrase “from vertical ionization potentials” implies that the energy difference between the higher and the first absorption maximum was used. If the first photoelectron transition has a gradual onset, a better value of the first ionization potential may have been obtained from photoionization data or from the extrapolation of Rydberg series in the spectrum of the parent molecule. Supplementary sources of data for the first ionization potential are cited in the tables. However, if the difference between the first adiabatic ionization potential obtained in the photoelectron spectrum and that obtained in other measurements amounts to only 10 meV or 20 meV, the photoelectron spectroscopic value is used, because of the advantage of a consistent set of measurements. Where threshold energies differ by one quantum in a vibrational progression, a best value for the ionization potential is chosen which coincides with the most probable position of the vibrationally unexcited transition. Because of inherent uncertainties in the determination of higher ionization potentials in many photoelectron spectral measurements, photoelectron peaks above about 18 eV are often omitted. Except where otherwise indicated, the units of all quantities in these tables are cm^{-1} . Error estimates are those of the authors of the original literature. The numbers in parentheses give these estimated errors in relation to the last digits of the electronic or vibrational frequency (e.g., 1234.567(89) $\equiv 1234.567 \pm 0.089$). Where the error includes a decimal point, the decimal point is retained inside the parentheses. When the uncertainty is not explicitly indicated, the value is given to the estimated number of significant figures. As in the tables of Herzberg,¹ T_0 values are given to the center of multiplet structure. For doublet states, the two components differ by $\pm A$ (the spin-orbit splitting constant), and the energy difference is measured from the average of the two bands, whereas for triplet states the three components fall at 0, $\pm A$ with respect to the position from which the band

energy is measured. This convention is also followed here unless specific states (e.g., ${}^2\Pi_{3/2}$) are given. However, in matrix isolation absorption and laser excitation studies only the lowest component is accessible. Except for transitions with relatively small values of A , this is also likely to be true for studies using cooled molecular beams. Often these latter studies give T_0 values for the lowest energy component with a precision better than that to which A is known.

The wavelength range (nm) in which various electronic transitions have been observed is also tabulated. This range is a composite of the values typical of absorption and emission observations. Laser-excited fluorescence studies often include both excitation and resolved emission measurements. Since the position of the band origin is given, ambiguity should not arise. For information on the range in which the band system is observed for a given type of measurement, see the original literature cited for that measurement technique.

The format of the vibrational tables is similar to that used in the earlier compilations. The vibrational numbering convention is that used by Herzberg.¹ Within a given symmetry species, vibrations are numbered starting with the highest frequency. The same convention is followed for deuterated species. Therefore, a given type of vibration may be numbered differently for the deuterated than for the unsubstituted molecule. For triatomic molecules, the bending vibration is always designated as ν_2 . For aromatic molecules, an alternate vibrational numbering scheme developed by Wilson²⁵ has often been used in the literature. Where both the Herzberg and the Wilson numbering schemes have been used for the published data, the Herzberg numbering is adopted, and the Wilson numbering is sometimes shown in parentheses. For a few species, only the Wilson numbering has been used. To avoid confusion, this is retained in the present tables, and the use of the Wilson numbering is indicated in a footnote. Where possible, the values of $\Delta G(\frac{1}{2})$, the separation between the $v=0$ and $v=1$ levels for the vibration of interest, have been used. The expression of uncertainties is similar to that described for electronic band origins. Where vibrational frequencies have been determined with a precision greater than two decimal places, the tabulated values have been rounded off. If a bending fundamental is split by Renner–Teller interaction, the position of the unperturbed fundamental is given. Where specific components of such a split fundamental have been studied, they may also be listed, with the transition designated in a footnote. For a more complete treatment of the Renner effect and definitions of the parameters included in many of these footnotes, see the discussion by Herzberg¹ and the references cited for the molecule of interest. A few of the species in these tables possess out-of-plane vibrations which have resolved inversion splitting structure. For these, the specific component for which the vibrational frequency is reported is designated in a footnote. Relative intensities of vibrational bands are dependent on the technique used for the measurement. When possible, the relative intensities of *ground-state infrared absorptions* are included. It is not feasible to give the corresponding rela-

tive intensities for other types of observation. Relative intensity abbreviations include:

vw	very weak
w	weak
m	medium
s	strong
vs	very strong
sh	shoulder
br	broad

Where radiative lifetimes have been measured, they are cited following the vibrational energy level table for the appropriate electronic state. τ_0 , the radiative lifetime of the vibrationless transition, is given whenever possible. If the lifetime is accessible only for excited vibrational states, the subscripts give the vibrational quantum numbers of the observed band.

When spin-orbit splitting occurs and the splitting constant, A , is known, it is included in the compilation.

Finally, as an aid in the recognition of vibrational bands and electronic band systems observed with comparatively high resolution, the principal rotational constants are summarized. Where possible, the values associated with the vibrationless transition (A_0, B_0, C_0) are given. Occasionally these values have not been determined, and the subscript gives the vibrational quantum numbers appropriate to the band for which the rotational constants have been measured. These constants are truncated at three decimal places. Often a far more detailed set of molecular constants, with much greater precision, has been derived from the analysis of high resolution spectra. Microwave spectroscopy is an important source of detailed, highly precise rotational data for molecules in their ground states. The references to the experimental literature, which are included in the compilation, should facilitate the location of such high resolution data.

5. Abbreviations

Many sophisticated laser techniques—frequently employing two or more laser beams—have been used for studies of transient molecules. The laser is frequently used both in the preparation of the transient molecule and in the detection scheme. For example, ions may be generated by multiphoton ionization and detected by absorption of radiation from a probe laser. Often the developers of such techniques have designated them by complicated acronyms. In these tables, an attempt has been made to avoid relatively lengthy and unfamiliar acronyms by designating only the generic type of detection, using the abbreviations defined below. (Velocity modulation, designated as a separate detection technique in the first of this series of data evaluations,² is widely used and is considered to be a measurement tool rather than a type of observation. The type of laser used for the absorption measurement in an infrared detection scheme employing velocity modulation is instead specified in these tables.)

AB	near infrared–visible-ultraviolet absorption
CC	color-center laser

CR	cavity ringdown
DL	diode laser absorption
DM	degenerate four-wave mixing
DPI	depletion photoionization
DR	double resonance
ED	electron diffraction
EF	electron-excited fluorescence
EM	near infrared-visible-ultraviolet emission
ESR	electron spin resonance
FD	fluorescence depletion
HFD	high frequency deflection
IB	ion beam
ID	ion drift, ion depletion (see specific reference)
IR	infrared absorption (conventional or Fourier transform)
LD	laser difference frequency
LF	laser-excited fluorescence (excitation and re-solved emission)
LMR	laser magnetic resonance
LS	laser Stark spectroscopy
MO	molecular orbital calculations
MPD	multiphoton dissociation
MPI	multiphoton ionization
MW	microwave and millimeter wave
ND	neutron diffraction
PD	electron photodetachment
PE	photoelectron spectroscopy
PEFCO	photoelectron-photon coincidence
T-PEFCO	threshold photoelectron-photon coincidence
PEPICO	photoelectron-photoion coincidence
PF	photofragment spectroscopy
PI	photoionization
PIFCO	photoion-photon coincidence
PIR	photoionization resonance
PRI	photoinduced Rydberg ionization
Ra	Raman
SEP	stimulated emission pumping
TF	tunable far-infrared laser
TPE	threshold photoelectron spectroscopy, including ZEKE detection
UV	near infrared-visible-ultraviolet absorption and emission

6. Acknowledgment

This work was supported in part by the Standard Reference Data Program of the National Institute of Standards and Technology.

7. General References

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8. Tables

8.1. H_3^+ , H_3 , and Triatomic Dihydrides H_3^+

\tilde{X} D_{3h}						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1'	1	Ring breathing	3178.18 ^a	gas	IR,PI,LD,EM	10,12,13,19,22
e'	2	Deformation	2521.42 ^a 2109.7T	gas H_2	LD,IR IR	1,4,15,22,23 24

$B_0=43.510$; $C_0=20.698$ LD^{1,4}DL⁴IR¹⁰EM²²

 H_2D^+

\tilde{X} C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1	1	Ring breathing	2992.51	gas	LD	3,6,11
	2	Deformation	2205.87	gas	LD,DL	7
b_2	3	Deformation	2335.45	gas	LD,DL	7

$A_0=43.438(2)$; $B_0=29.134$; $C_0=16.601$ LD,MW^{6,7,11}

 D_2H^+

\tilde{X} C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1	1	Ring breathing	2736.98	gas	LD	5,11,14
	2	Deformation	1968.17	gas	DL	8,14
b_2	3	Deformation	2078.43	gas	DL	8,14

$A_0=36.199$; $B_0=21.869$; $C_0=13.070$ LD^{5,11,14}DL^{8,14}

 D_3^+

\tilde{X} D_{3h}						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1'	1	Ring breathing	2300.84 ^a	gas	EM,DL	21
e'	2	Deformation	1834.67 ^a	gas	IB,DL	2,9

$B_0=21.810$; $C_0=10.533$ DL^{9,21}EM²¹

^aHot bands arising from ν_1 and ν_2 of H_3^+ have been observed,^{16,19} as have been the first¹⁷ and second^{18,20} overtones of ν_2 . Several of the corresponding hot bands and overtones of D_3^+ have also been reported.²¹

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 H_3

Higher Rydberg states have been detected using photoionization and field ionization. The ionization limit observed for vibrationally and rotationally unexcited H_3 from its $2p^2A_2'$ state is 29562.6(5).^{12,15,16} Ion depletion studies have also yielded frequencies for the ring breathing vibration of a number of these higher Rydberg states.²³ Near the lowest ionization threshold, predissociation has been found to be induced by very weak electric fields.²⁵ Rotational and vibrational interactions, autoionization, and predissociation in the np Rydberg manifold have been studied.²⁷

$3d^2A_1'$ D_{3h} Structure: EM ⁸						
$T_0^a=18511$	gas	EM ⁸ PF ²⁶				$3d-2p^2A_2''$ 568–615 nm
		EM ⁸				$3d-3p^2E'$ 3891–4456 cm^{-1}
$B_0=42.99$; $C_0=22.735$		EM ⁸				

$3d^2E''$ D_{3h} Structure: EM ⁸						
$T_0^a=18409$	gas	EM ⁸ PF ^{14,26}				$3d-2p^2A_2''$ 568–615 nm
		EM ⁸				$3d-3p^2E'$ 3891–4456 cm^{-1}

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1'	1	Ring breathing	3168 ^b	gas	PI	19,23
e'	2	Deformation	2518	gas	EM,PF	22

$B_0=42.99$; $C_0=22.735$ EM⁸

$3d\ ^2E'$ D_{3h} Structure: EM⁸
 $T_0^a = 18037$ gas EM⁸ $3d-2p^2A_2''$ 568–615 nm
 EM⁸ $3d-3p^2E'$ 3891–4456 cm⁻¹
 $B_0 = 42.99$; $C_0 = 22.735$ EM⁸

$3p\ ^2A_2''$ D_{3h} Structure: EM³
 $T_0^a = 17789$ gas EM^{2,3,8} $3p^2A_2''-2s^2A_1'$ 556–574 nm
 $\tau = 37(4)$ ns gas EM¹⁰
 $B_0 = 47.45$; $C_0 = 23.495$ EM⁸

$3s\ ^2A_1'$ D_{3h} Structure: EM⁶
 $T_0^a = 17600$ gas EM^{3,PF}^{13,14} $3s^2A_1'-2p^2A_2''$ 592–615 nm
 EM⁶ $3s^2A_1'-3p^2E'$ 3178–3847 cm⁻¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1'	1	Ring breathing	3212.1(3) ^b	gas	PI	19,23
e'	2	Deformation	2588(2)	gas	EM,PF	22

$B_0 = 44.19$; $C_0 = 22.676$ EM⁶

$3p\ ^2E'$ D_{3h} Structure: EM⁶
 $T_0^a = 13961$ gas EM^{2,4} $3p^2E'-2s^2A_1'$ 708–736 nm
 EM⁶ $3s^2A_1'-3p^2E'$ 3178–3847 cm⁻¹
 EM⁸ $3d-3p^2E'$ 3891–4456 cm⁻¹
 $\tau = 1.1(+0.2, -1.0)$ ns gas EM²¹
 $B_0 = 42.15$; $C_0 = 21.505$ EM⁶

$2p\ ^2A_2''$ D_{3h} Structure: EM⁶
 $T_0^a = 993$ gas EM^{3,6} $3s^2A_1'-2p^2A_2''$ 592–615 nm
 EM⁸ $3d-2p^2A_2''$ 568–615 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1'	1	Ring breathing	3255.38(3)	gas	PI	17–19
e'	2	Deformation	2618.34(3)	gas	PI	18

$B_0 = 44.58$; $C_0 = 22.288$ EM⁶

$\tau_{00} = 640(+300, -100)$ ns; $\tau_{10} = 740(+300, -100)$ ns gas PI²⁴PF²⁶

$2s\ ^2A_1'^c$ D_{3h} Structure: EM³
 gas EM^{2,3} $3p^2A_2''-2s^2A_1'$ 556–574 nm
 gas EM⁴ $3p^2E'-2s^2A_1'$ 708–736 nm
 $B_0 = 46.82$; $C_0 = 23.41$ EM³

H₂D

$3p\ ^2B_1$ C_{2v}
 $T_0^a = 17806.5$ gas EM³⁰
 $\tau = 29(3)$ ns gas EM²⁰
 $A = 42.75(50)$; $B = 28.6(5)$; $C = 16.5(5)$ EM³⁰

$3s\ ^2A_1$ C_{2v}
 $T_0^a = 16609.25$ gas EM³¹
 $\tau \approx 4$ ns gas EM²⁰
 $A = 43.99$; $B = 29.439$; $C = 16.954$ EM³¹

$3p\ ^2A_1, ^2B_2$ C_{2v}
 $\tau = 2.5(+0.3, -0.7)$ ns gas EM²¹

$2p\ ^2A_2$ C_{2v}
 $A = 44.125$; $B = 29.486$; $C = 16.841$ EM³¹

$2s\ ^2A_1$ C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2		2300(30)T	gas	EM	30
b_2	3		2450(30)T	gas	EM	30

$\tau \approx 0.4(2)$ ps gas EM³⁰

$A = 45.3(5)$; $B = 30.3(5)$; $C = 17.5(5)$ EM³⁰

D₂H

$3p\ ^2B_1$ C_{2v} Structure: EM⁹
 $T_0^a = 17834.4$ gas EM^{9,30} $3p^2B_1-2s^2A_1$ 560–673 nm
 $\tau_1 = 31.5(3.2)$ ns; $\tau_2 \approx 8.9$ ns gas EM²⁰
 $A = 37.8(4)$; $B = 22.6(4)$; $C = 13.9(4)$ EM³⁰

$3s\ ^2A_1$ C_{2v}
 $T_0^a = 16602.51$ gas EM³¹
 $\tau \approx 5$ ns gas EM²⁰
 $A = 37.7$; $B = 22.17$; $C = 13.69$ EM³¹

$3p\ ^2A_1, ^2B_2$ C_{2v} Structure: EM⁹
 gas EM³⁰ $3p^2(B_2, A_1)-2s^2A_1$ 709–860 nm
 $\tau = 5.0(7)$ ns gas EM²¹

$2p\ ^2A_2$ C_{2v}

$A = 36.9$; $B = 22.07$; $C = 13.58$ EM³¹

$2s\ ^2A_1$ C_{2v} Structure: EM⁹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Ring breathing	2950(20)	gas	EM	20
	2		2125(80)	gas	EM	30
b_2	3		2358(50)	gas	EM	30

$\tau \approx 0.5(2)$ ps gas EM³⁰

$A = 39.14(40)$; $B = 23.4(4)$; $C = 14.4(4)$ EM³⁰

D₃

The ionization limit of D₃ with respect to the vibrationless level of its $2p\ ^2A_2''$ state has been found to be 29602.²¹

$3d\ ^2A_1'$ D_{3h} Structure: EM⁸
 $T_0^a = 18530$ gas EM⁸ $3d-2p^2A_2''$ 569–601 nm
 EM⁸LD³³ $3d-3p^2E'$ 3772–4517 cm⁻¹
 $\tau = 12(1)$ ns gas EM^{9,20}
 $B_0 = 21.72(2)$; $C_0 = 10.91(2)$ EM⁸

$3d\ ^2E''$ D_{3h} Structure: EM⁸
 $T_0^a = 18433$ gas EM⁸ $3d-2p^2A_2''$ 569–601 nm
 EM⁸LD³³ $3d-3p^2E'$ 3772–4517 cm⁻¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1'	1	Ring breathing	2296.1	gas	PI	32

$\tau = 12(1)$ ns gas EM^{9,20}

$B_0 = 21.72(2); C_0 = 10.91(2) \text{ EM}^8$

$3d^2E''$ D_{3h} Structure: EM^8
 $T_0^a = 18098 \text{ gas EM}^8$ $3d-2p^2A_2''$ 569–601 nm
 EM^8LD^{33} $3d-3p^2E'$ 3772–4517 cm^{-1}
 $\tau = 12(1) \text{ ns gas EM}^{9,20}$
 $B_0 = 21.72(2); C_0 = 10.91(2) \text{ EM}^8$

$3p^2A_2''$ D_{3h} Structure: EM^3
 $T_0^a = 17872 \text{ gas EM}^{2,3,8,30}LF^7$ $3p^2A_2''-2s^2A_1'$ 553–651 nm
 $\tau_0 = 29(1) \text{ ns gas EM}^9$
 $B_0 = 22.73(6); C_0 = 10.68(2) \text{ EM}^8$

$3s^2A_1'$ D_{3h} Structure: EM^6
 $T_0^a = 17642 \text{ gas EM}^3$ $3s^2A_1'-2p^2A_2''$ 592–614 nm
 EM^6LD^{33} $3s^2A_1'-3p^2E'$ 3382–3768 cm^{-1}
 $\tau \approx 10 \text{ ns gas EM}^{20}$
 $B_0 = 21.98; C_0 = 12.41 \text{ EM}^6$

$3p^2E'$ D_{3h} Structure: EM^6
 $T_0^a = 14091 \text{ gas EM}^{2,4,21,30}LF^7$ $3p^2E'-2s^2A_1'$ 700–860 nm
 EM^6LD^{33} $3s^2A_1'-3p^2E'$ 3382–3768 cm^{-1}
 EM^8 $3d-3p^2E'$ 3772–4517 cm^{-1}

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1'	1	Ring breathing	2145T	gas	EM	21
e'	2	Deformation	1750T	gas	EM	4,21

$\tau = 17.5(2.0) \text{ ns gas EM}^{9,21}$
 $B_0 = 21.15; C_0 = 10.59 \text{ EM}^6$

$2p^2A_2''$ D_{3h} Structure: EM^6
 $T_0^a = 1052 \text{ gas EM}^{3,6}LF^7$ $3s^2A_1'-2p^2A_2''$ 592–614 nm
 EM^8 $3d-2p^2A_2''$ 569–601 nm

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1'	1	Ring breathing	2353.3	gas	PI	29,32
e'	2	Deformation	1900.9	gas	PI	29,32

$B_0 = 22.112; C_0 = 11.056 \text{ EM}^6$

$2s^2A_1^c$ D_{3h} Structure: EM^3
 $\text{gas EM}^{2,3}LF^7$ $3p^2A_2''-2s^2A_1'$ 553–569 nm
 $EM^{4,21}$ $3p^2E'-2s^2A_1'$ 700–765 nm

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1'	1	Ring breathing	2470(22)	gas	EM	4,20,21, 30
e'	2	Deformation	1884(22)	gas	EM	21,30

$B_0 = 23.09; C_0 = 11.544(6) \text{ EM}^{3,21}$

^aMeasured with respect to lowest bound state, $2s^2A_1'$. Structure observed¹ in the dissociation spectrum of H_2 has been reinterpreted⁵ as arising from the predissociation of H_3 ($2s^2A_1'$) into $H+H_2$. Unstructured emission observed¹¹ between 190 and 280 nm, with a maximum near 230 nm, upon charge transfer between K and H_3^+ or D_3^+ has been attributed to transitions originating in bound Rydberg states of H_3 or D_3 and terminating in the dissociative ground-state continuum. When charge transfer with Cs was studied,²⁸ emission between 200 and 400 nm was detected for all four isotopic species of H_3 . A double maximum for D_3 was interpreted as arising from emission to both sheets of the ground-state potential surface. Photofragment spectroscopy¹⁴ has placed the $2p^2A_2''$ state 5.563(20) eV above the ground-state $H+H_2$ dissociation limit.

^bObserved for $N=1$ rotational level.

^cPredissociated by vibronic interaction with the $2p^2E'$ repulsive ground state; linewidth is approximately 15 cm^{-1} for H_3 and 6 cm^{-1} for D_3 .²

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CeH₂⁺

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1	1	CeH ₂ s-stretch	1486.3	Ar	IR	1
b_2	3	CeH ₂ a-stretch	1445.9	Ar	IR	1

CeD₂⁺

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CeD ₂ s-stretch	1059.8	Ar	IR	1
b ₂	3	CeD ₂ a-stretch	1035.3	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **104**, 1640 (2000).

FeH₂

In an argon, krypton, or xenon matrix, three broad absorptions appear¹ between 400 and 450 nm. Irradiation at 440 nm results in photodecomposition, producing Fe+H₂.^{1,2}

$\tilde{X}^5\Delta_g$ D _{∞h} Structure: LMR ⁵						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π _u	2	Bend	335	Ar	IR	2
			322	Kr	IR	1
			323	Xe	IR	1
Σ _g ⁺	3	FeH a-stretch	1674.72	gas	LMR	5,7
			1660.8	Ar	IR	2,4
			1647	Kr	IR	1,2
			1636	Xe	IR	1,2

B₀ = 3.075 LMR^{5,7}

FeD₂

In krypton and xenon matrices, three broad absorptions appear¹ between 400 and 450 nm, each slightly shifted from their FeH₂ counterparts. Irradiation at 440 nm results in the formation of Fe+D₂.^{1,2}

T₀ = 9530(180) gas PE³

$\tilde{X}^5\Delta_g$ D _{∞h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π _u	2	Bend	221.14 ^a	gas	LMR	6
			235	Ar	IR	2
			232	Xe	IR	1
Σ _u ⁺	3	FeD a-stretch	1204.2	Ar	IR	2,4
			1195	Kr	IR	2
			1188	Xe	IR	1

B₀ = 1.542 LMR⁶

^{a5}Π₃-⁵Δ₄ component. ⁵Φ₃-⁵Δ₂ component at 226.06.⁶

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⁷H. Körsgen, W. Urban, and J. M. Brown, *J. Chem. Phys.* **110**, 3861 (1999).

cyc-PdH₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	HH stretch	2971.4	Ar	IR	2,3
			871.8	Ne	IR	3
	3	PdH s-stretch	865.3			
			950.0s	Ar	IR	2,3
b ₂	3	PdH a-stretch	960	Kr	IR	1
			894.5	Xe	IR	1
			885.5			
			1507.5w	Ar	IR	2,3

cyc-PdD₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	DD stretch	2169	Ar	IR	2,3
			670.9	Ne	IR	3
	3	PdD s-stretch	714.4	Ar	IR	2,3
			714	Kr	IR	1
b ₂	3	PdD a-stretch	1098.0	Ar	IR	2,3

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PtH₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	PtH s-stretch	2365.7	Ar	IR	2
b ₂	3	PtH a-stretch	2355.2	Ne	IR	2
			2348.9	Ar	IR	1,2
			2314	Kr	IR	1

PtD₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	PtD s-stretch	1697.7	Ar	IR	2
b ₂	3	PtD a-stretch	1688.2 1683.3	Ne Ar	IR IR	2 2

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CeH₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CeH ₂ s-stretch	1330.7	Ar	IR	1
b ₂	3	CeH ₂ a-stretch	1281.7	Ar	IR	1

CeD₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CeD ₂ s-stretch	943.7	Ar	IR	1
b ₂	3	CeD ₂ a-stretch	916.9	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **104**, 1640 (2000).

PrH₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		PrH ₂ stretch	1286.6	Ar	IR	1

PrD₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		PrD ₂ stretch	917.2	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **104**, 1640 (2000).

NdH₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b ₂	3	NdH ₂ a-stretch	1148.4	Ar	IR	1

NdD₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b ₂	3	NdD ₂ a-stretch	822.4	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **104**, 1640 (2000).

SmH₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	SmH ₂ s-stretch	1213.0	Ar	IR	1
b ₂	3	SmH ₂ a-stretch	1156.5	Ar	IR	1

SmD₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	SmD ₂ s-stretch	864.3	Ar	IR	1
b ₂	3	SmD ₂ a-stretch	827.2	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **104**, 1640 (2000).

EuH₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	EuH ₂ s-stretch	1211.7	Ar	IR	1
b ₂	3	EuH ₂ a-stretch	1155.6	Ar	IR	1

EuD₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	EuD ₂ s-stretch	863.6	Ar	IR	1
<i>b</i> ₂	3	EuD ₂ a-stretch	827.0	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **104**, 1640 (2000).

GdH₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	GdH ₂ s-stretch	1399.0	Ar	IR	1
<i>b</i> ₂	3	GdH ₂ a-stretch	1359.3	Ar	IR	1

GdD₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	GdD ₂ s-stretch	998.8	Ar	IR	1
<i>b</i> ₂	3	GdD ₂ a-stretch	973.0	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **104**, 1640 (2000).

TbH₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	TbH ₂ s-stretch	1445.3	Ar	IR	1
<i>b</i> ₂	3	TbH ₂ a-stretch	1391.1	Ar	IR	1

TbD₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	TbD ₂ s-stretch	1031.4	Ar	IR	1
<i>b</i> ₂	3	TbD ₂ a-stretch	996.0	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **104**, 1640 (2000).

DyH₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	DyH ₂ s-stretch	1247.1	Ar	IR	1
<i>b</i> ₂	3	DyH ₂ a-stretch	1193.1	Ar	IR	1

DyD₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	DyD ₂ s-stretch	888.6	Ar	IR	1
<i>b</i> ₂	3	DyD ₂ a-stretch	853.8	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **104**, 1640 (2000).

HoH₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	HoH ₂ s-stretch	1255.8	Ar	IR	1
<i>b</i> ₂	3	HoH ₂ a-stretch	1203.6	Ar	IR	1

HoD₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	HoD ₂ s-stretch	895.1	Ar	IR	1
<i>b</i> ₂	3	HoD ₂ a-stretch	861.3	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **104**, 1640 (2000).

ErH₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	ErH ₂ s-stretch	1271.0	Ar	IR	1
<i>b</i> ₂	3	ErH ₂ a-stretch	1217.3	Ar	IR	1

ErD₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	ErD ₂ s-stretch	905.7	Ar	IR	1
<i>b</i> ₂	3	ErD ₂ a-stretch	871.3	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **104**, 1640 (2000).

TmH₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	TmH ₂ s-stretch	1281.2	Ar	IR	1
<i>b</i> ₂	3	TmH ₂ a-stretch	1222.2	Ar	IR	1

TmD₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	TmD ₂ s-stretch	913.3	Ar	IR	1
<i>b</i> ₂	3	TmD ₂ a-stretch	874.8	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **104**, 1640 (2000).

YbH₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	YbH ₂ s-stretch	1276.1	Ar	IR	1
<i>b</i> ₂	3	YbH ₂ a-stretch	1218.6	Ar	IR	1

YbD₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	YbD ₂ s-stretch	910.0	Ar	IR	1
<i>b</i> ₂	3	YbD ₂ a-stretch	872.1	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **104**, 1640 (2000).

LuH₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	LuH ₂ s-stretch	1486.4	Ar	IR	1
<i>b</i> ₂	3	LuH ₂ a-stretch	1426.4	Ar	IR	1

LuD₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	LuD ₂ s-stretch	1061.3	Ar	IR	1
<i>b</i> ₂	3	LuD ₂ a-stretch	1021.5	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **104**, 1640 (2000).

PdH₂⁻

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_u^+	3	PdH a-stretch	1247.0	Ar	IR	1

PdD₂⁻

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_u^+	3	PdD a-stretch	912.9	Ne	IR	1
			908.5	Ar	IR	1

Reference

¹L. Andrews, X. Wang, M. E. Alikhani, and L. Manceron, J. Phys. Chem. A **105**, 3052 (2001).

CH₂

4p C_{2v}
T₀=74254 gas MPI³⁷

\tilde{D}
T₀=71592 gas AB¹MPI³⁶ $\tilde{D}-\tilde{X}$ 139.7 nm

\tilde{C}
T₀=70917 gas AB¹MPI³⁶ $\tilde{C}-\tilde{X}$ 141.0 nm

3d ³A₂ C_{2v} Structure: AB⁷
T₀=70634 gas AB¹MPI³⁶ 3d ³A₂- \tilde{X} 141.5 nm

Diffuse. First member of Rydberg series converging to 83851. Higher members observed (AB²) at 76553, 79241, and 80688.
B₀=6.89^a AB¹

3p C_{2v}
T₀=64126 gas MPI³⁷

\tilde{c} ¹A₁ gas AB³ $\tilde{c}-\tilde{a}$ 330–362 nm

\tilde{b} ¹B₁^b C_{2v} Structure: AB^{3,29}
T₀=11497(10) gas $\tilde{b}-\tilde{a}$ 465–1049 nm
AB^{1,3,27,44}LMR²¹LF^{33,38,40,42,43}DL⁴⁷⁻⁵⁰

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	Bend	570T	gas	AB	3

$\tau(0,14,0)=4.6(1)$ μ s LF^{9,39}

$\tau(0,16,0)=1.3(3)$ μ s LF¹¹
3.8(3) μ s LF³⁹

B₀₁₀=8.59(4) DL⁴⁷
Barrier to linearity=1617²⁹

\tilde{a} ¹A₁^b C_{2v} Structure: AB^{3,29,31}
T₀=3147(5) gas $\tilde{b}-\tilde{a}$ 465–1049 nm
AB^{1,3,27,28,44}LMR^{21,26,30}PE^{23,24}LF^{32,38,40,42,43,45}SEP^{32,34}DL⁴⁷⁻⁵⁰

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CH ₂ s-stretch	2806.01(7)	gas	LF,LD IR	10,20,31 40
	2	Bend	1352.6	gas	AB,LF	3,27,28 40
b ₂	3	CH ₂ a-stretch	2864.97(2)	gas	LD,IR	20,31

$\tau \approx 18$ s^c

A₀=20.118(2); B₀=11.205(2); C₀=7.069(2) AB^{3,27,28}
Barrier to linearity=8600(400) LF⁴⁵

\tilde{X} ³B₁ C_{2v} Structure: ESR⁴⁻⁶AB⁷LMR^{15,17}IR^{17,26}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	Bend	963.10	gas	LMR DL	12,16 19,25
b ₂	3	CH ₂ a-stretch	3190(5) ^d	gas	IR	31

A₀=73.811; B₀=8.450; C₀=7.184 IR²⁵
Barrier to linearity=1931(30)²⁶

CD₂

4p C_{2v}
T₀=74228 gas MPI³⁷

\tilde{D}
T₀=70947 gas AB¹MPI³⁶ $\tilde{D}-\tilde{X}$ 140.95 nm

\tilde{C}
T₀=71510 gas AB¹MPI³⁶ $\tilde{C}-\tilde{X}$ 139.8 nm

3d ³A₂ C_{2v} Structure: AB⁷
T₀=70591.7 gas AB¹MPI³⁶ 3d ³A₂- \tilde{X} 141.6 nm
B₀=3.595 AB¹

3p C_{2v}
T₀=64082 gas MPI³⁷

\tilde{b} ¹B₁^b C_{2v}
gas LF^{13,41} $\tilde{b}-\tilde{a}$ 510–610 nm
 $\tau(0,16,0)=6.0(7)$ μ s LF¹³

\tilde{a} ¹A₁^b C_{2v}
T₀=3140(50) gas PE²³LF⁴¹SEP⁴¹ $\tilde{b}-\tilde{a}$ 510–610 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	Bend	1005(1)	gas	LF	13

A₀=11.37(32); B₀=5.476(48); C₀=3.701(45) LF⁴¹SEP⁴¹

\tilde{X} ³B₁ C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	Bend	752.37	gas	DL	19

A₀=37.787; 1/2(B+C)₀=3.962; 1/2(B-C)₀=0.267 LMR^{18,22}MW⁴⁶

^aValue given for ¹³CH₂.

^bThe \tilde{a} ¹A₁ and \tilde{b} ¹B₁ states are perturbed by strong Renner–Teller interaction.^{13,14,29,44} They are also strongly perturbed by interaction with the \tilde{X} ³B₁ state.^{27,28,35}

^cCalculated value.²¹

^dFrom analysis of perturbations involving combination bands.³¹

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SiH₂

$\tilde{A}^1B_1^a$ C_{2v} Structure: AB^{1,2,16}
 $T_0 = 15547.77$ gas AB^{1,2,16}LF^{10,11,13} $\tilde{A}-\tilde{X}$ 480–650 nm
 Onset of predissociation into Si(¹D) + H₂ near 21450.¹¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	1990	gas	LF	13
	2	Bend	854	gas	AB,LF	1,11,13

$\tau_0 = 1.10(17)$ μ s gas LF^{6,7,11}
 $A_0 = 18.324(2)$; $B_0 = 4.900$; $C_0 = 3.766$ AB²
 Barrier to linearity $\cong 8000^3$

\tilde{a}^3B_1 C_{2v}
 $T_0 = 7340(240)^b$ gas PI⁸
 Barrier to predissociation into Si(³S) + H₂ between 17070 and 17690.¹¹

$\tilde{X}^1A_1^a$ C_{2v} Structure: AB^{1,2}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	1995.93 ^c	gas	LF,DL	10,13,15, 18
			1992.8 ^c	Ar	IR	5
			2013.8T	Kr	IR	17
	2	Bend	999.03	gas	AB,LF	2,4,9,13
			994.8	Ar	IR	5
			998.5	Kr	IR	17
b ₂	3	Asym. stretch	1992.82	gas	DL	18
			1973.3	Ar	IR	5
			1993.7T	Kr	IR	17

$A_0 = 8.099$; $B_0 = 7.024$; $C_0 = 3.703$ AB^{2,16}DL^{9,17}

SiD₂

$\tilde{A}^1B_1^a$ C_{2v}
 $T_0 = 15539.875(2)$ gas LF^{11,14} $\tilde{A}-\tilde{X}$ 463–652 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	Bend	618.55	gas	AB,LF	1,11,14

$A_0 = 9.629$; $B_0 = 2.456$; $C_0 = 1.926$ LF¹⁴
 $\tau_0 = 0.93(38)$ μ s gas LF¹¹

$\tilde{X}^1A_1^a$ C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	1444.6 ^c	Ar	IR	5
			1456.3T	Kr	IR	17
			731 ^d	gas	LF	11
	2	Bend	719.8	Ar	IR	5
			721	Kr	IR	17
			1439.1	Ar	IR	5
b ₂	3	Asym. stretch	1447T	Kr	IR	17

$A_0 = 4.334$; $B_0 = 3.519$; $C_0 = 1.919$ LF¹⁴

^aThe \tilde{A}^1B_1 and \tilde{X}^1A_1 states are perturbed by strong Renner–Teller interaction.³ The combined effects of Renner–Teller and spin-orbit interaction have been considered in detail by Ref. 12.

^bPossibly 6290(240).⁸

^cIn Fermi resonance with $2\nu_2$, observed for SiH₂ at 1987.69 (gas) and 1964.4 (Ar) and for SiD₂ at 1426.9 (Ar), adopting the reanalysis proposed by Ref. 18.

^d ω .

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GeH₂

\tilde{A}^1B_1 C_{2v} Structure: LF^{5,7}AB⁷
 $T_0 = 16325.54$ gas LF^{3-5,7}AB^{6,7} $\tilde{A}-\tilde{X}$ 489–850 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	1798.4	gas	LF	5
	2	Bend	783.0	gas	LF	3–5

$\tau_0 = 2.29(7) \mu\text{s}$ gas LF^{4,5}
 $A_0 = 16.363(11)$; $B_0 = 4.522$; $C_0 = 3.458(2)$ LF⁵AB⁷

\tilde{X}^1A_1 C_{2v} Structure: LF⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	1856	gas	LF	7
			1864wmT	Ar	IR	1,2
	2	Bend	916	gas	LF	3,7
			920wm	Ar	IR	1,2
b ₂	3	Asym. stretch	1887wmT	Ar	IR	1,2

$A_0 = 7.019(6)$; $B_0 = 6.529(3)$; $C_0 = 3.330(2)$ LF⁵AB^{6,7}

GeD₂

\tilde{A}^1B_1 C_{2v}
 $T_0 = 16324.34$ gas LF^{3,4,7} $\tilde{A}-\tilde{X}$ 500–850 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	1304	gas	LF	7
	2	Bend	561	gas	LF	3,4,7

$\tau_0 = 2.5(5) \mu\text{s}$ gas LF⁴

$A_0 = 8.348(3)$; $B_0 = 2.269$; $C_0 = 1.756$ LF⁷
 \tilde{X}^1A_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	1335	gas	LF	7
			1329vsT	Ar	IR	1
			1325vsT			
	2	Bend	657	gas	LF	3,7
			658m	Ar	IR	1
b ₂	3	Asym. stretch	1338msT	Ar	IR	1

$A_0 = 3.618(4)$; $B_0 = 3.273$; $C_0 = 1.699$ LF⁷

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PH₂⁺

\tilde{a}^3B_1 C_{2v}
 $T_0 \cong 5650$ gas PE¹⁻³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	Bend	888(80)	gas	PE	3

\tilde{X}^1A_1 C_{2v}

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NH₂

Rydberg series with members at 93054, 95753, 97193, and 98049, converging to NH₂⁺ (\tilde{A}^1A_1) at 100410. PI¹⁹

$\tilde{A}^2A_1(\Pi_u)^a$ C_{2v} Structure: AB^{1,4}
 $T_0 = 11122.23(5)$ gas AB^{1,8,22,31}LF^{6,21,25,26}EM^{25,31,33} $\tilde{A}-\tilde{X}$ 342–2700 nm
 Ne,Ar,Kr,Xe^b AB^{2,3,5,27} $\tilde{A}-\tilde{X}$ 344–880 nm
 N₂^b AB⁵ $\tilde{A}-\tilde{X}$ 480–620 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	3325	gas	AB	1
	2	Bend	1157.8	gas	AB,EM	31

$\tau_{090\Sigma} = 10.0(1.7) \mu\text{s}$ gas LF⁷
 $\tau_{080\Pi} = 10(3) \mu\text{s}$ gas LF¹⁷

Approximate ν^3 dependence.^{7,17} In another LF study,¹² τ varied from 25 to 46 μs for relatively unperturbed rotational sublevels, and there was a weaker

~100 μ s component associated with levels which are substantially perturbed.

$B_0 = 8.776(4)$ AB^{1,31}EM³¹
Barrier to linearity = 730¹⁴

$\tilde{X}^2B_1^a$		C _{2v}		Structure: AB ¹ IR ²⁹		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	3219.37	gas	LF,EM	6,15,16
			LD,IR	21,23		
b ₂	3	Asym. stretch	3220w ^c	N ₂	IR	5
			1497.32	gas	UV,LF	1,6,8–10
			1499m	N ₂	LMR IR	13,20
b ₂	3	Asym. stretch	3301.11	gas	IR	5
			LD,LF	16,21		
					IR	23

$A_0 = 23.693$; $B_0 = 12.952$; $C_0 = 8.173$ AB^{1,8}LMR¹³IR^{20,29,30}MW^{28,32}
Barrier to linearity = 12024¹⁴

ND₂

$\tilde{A}^2A_1(\Pi_u)^a$		C _{2v}				
gas	AB ¹			$\tilde{A}-\tilde{X}$ 500–680 nm		
Ar	AB ²⁷			$\tilde{A}-\tilde{X}$ 380–825 nm		

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	2520T	gas	AB	1

$B_0 = 4.41$ AB¹

$\tilde{X}^2B_1^a$		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	Bend	1108.75	gas	LMR	11,18
			1110m	N ₂	IR	5

$A_0 = 13.342$; $B_0 = 6.488$; $C_0 = 4.290$ AB^{1,18}LMR¹¹MW²⁴IR²⁹

^aThe \tilde{A}^2A_1 and \tilde{X}^2B_1 states are perturbed by strong Renner–Teller interaction.

^bA detailed comparison of the argon-matrix data with gas-phase data has been given in Refs. 22 and 27. Rotational structure is resolved in the rare-gas matrices. In nitrogen,⁵ bands are very broad and redshifted by approximately 400 cm⁻¹, with no evidence for rotational structure.

^cAssigned⁵ in matrix studies to ν_3 . Gas-phase observation of ν_1 at 3219.37 cm⁻¹ and demonstration¹⁶ that ν_1 is more intense than ν_3 dictate reassignment to ν_1 .

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PH₂

$\tilde{A}^2A_1^a$		C _{2v}		Structure: AB ⁴		
$T_0 = 18276.569$	gas	AB ^{1,4,6} EM ^{2,3,5} LF ^{22,23}		$\tilde{A}-\tilde{X}$ 360–880 nm		
18215(4)	Ar	AB ^{13,19}		$\tilde{A}-\tilde{X}$ 405–550 nm		

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	Bend	949.12	gas	UV	2,6
			949(7)	Ar	AB	13,19

$\tau = 4(1)$ μ s gas LF^{11,20}EM¹²
 $A_0 = 20.414(5)$; $B_0 = 5.607$; $C_0 = 4.293$ AB^{4,6}EM⁵LF^{22,23}
Barrier to linearity = 6840⁷

$\tilde{X}^2B_1^a$		C _{2v}		Structure: AB ⁴ MW ²⁴		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	2310(2)	gas	PE,Ra	10,21
			1101.91	gas	UV, LMR	2,4,14
			1103m	Ar	IR	13

$A_0 = 9.132$; $B_0 = 8.084$; $C_0 = 4.214$ AB^{4,16}LMR^{8,14,15}MW^{17,18}LF²²
Barrier to linearity = 25100⁷

PD₂

$\tilde{A}^2A_1^a$ C_{2v} Structure: AB¹EM^{2,3}
 $T_0=18282.1$ gas $\tilde{A}-\tilde{X}$ 360–880 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	Bend	689.5 665(25)	gas Ar	EM UV	2 13

$\tilde{X}^2B_1^a$ C_{2v}
 Vib. sym. No. Approximate type of mode cm⁻¹ Med. Type meas. Refs.

a_1	2	Bend	795.5 797w	gas Ar	EM IR	2,3 13
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$A_0=4.855$; $B_0=4.048$; $C_0=2.180$ AB⁹MW²⁴

^aThe \tilde{A}^2A_1 and \tilde{X}^2B_1 states are perturbed by strong Renner–Teller interaction.

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AsH₂

\tilde{A}^2A_1 C_{2v} Structure: AB¹
 $T_0=19907.8$ gas AB¹EM² $\tilde{A}-\tilde{X}$ 390–650 nm
 Predissociated above 23300¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	Bend	851.4	gas	AB	1

$\tau=130(20)$ ns gas EM²
 $A_{010}=19.48(1)$; $B_{010}=4.97(1)$; $C_{010}=3.71$ AB¹

\tilde{X}^2B_1 C_{2v} Structure: AB¹MW⁴
 Vib. sym. No. Approximate type of mode cm⁻¹ Med. Type meas. Refs.

a_1	2	Bend	981	gas	EM	2
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$A_0=7.550$; $B_0=7.163$; $C_0=3.615$ AB¹MW³LMR⁵

AsD₂

\tilde{A}^2A_1 C_{2v} Structure: AB¹
 $T_0=19904.9$ gas AB¹ $\tilde{A}-\tilde{X}$ 390–490 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	Bend	615.9	gas	AB	1

\tilde{X}^2B_1 C_{2v}
 $A_0=3.881$; $B_0=3.587$; $C_0=1.842$ MW⁴

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H₂O⁺

\tilde{B}^2B_2 C_{2v}
 $T_0=36757(12)$ gas PE⁹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	2968 ^a	gas	PE	9
	2	Bend	1596 ^a	gas	PE	9

$\tilde{A}^2A_1(\Pi_u)^b$ $D_{\infty h}$ Structure: PE^{4,7}EM⁷
 $T_{070}=13409.3$ gas EM^{1,5,17}AB^{14,18} $\tilde{A}-\tilde{X}$ 400–750 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	3547(16)	gas	PE	9
	2	Bend	876.8 ^b	gas	EM,PE	5,9

$\tau=10.5(1.0)$ μ s gas EF⁶
 $B_{070}=8.57$ EM⁵

$\tilde{X}^2B_1^b$ C_{2v} Structure: EM^{5,7}LMR⁸LD¹¹CC¹⁵

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	3212.86	gas	PE,LD	9,11
			3182.7	Ne	PI,CC	13,15
	2	Bend	1408.42	gas	IR	16
b_2	3	Asym. stretch	1401.7	Ne	EM,PE	1,5,9
			3259.04	gas	DL	12
			3219.5	Ne	IR	16

$A_0=29.036(2)$; $B_0=12.423$; $C_0=8.469$ LMR⁸LD¹¹DL¹²CC¹⁵
 Barrier to linearity $\leq 9187^{7,17}$

D₂O⁺

\tilde{B}^2B_2 C_{2v}
 $T_0=37430(50)$ gas PE^{2,4}
 38498(12) gas PE⁹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	2282 ^a	gas	PE	2,9
	2	Bend	1099 ^a	gas	PE	9

$\tilde{A}^2A_1(\Pi_u)^b$ $D_{\infty h}$ Structure: EM¹⁰
 $T_{050}^c=10456(30)$ gas PE^{2,4}EM¹⁰ $\tilde{A}-\tilde{X}$ 490–670 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	2531(8)	gas	PE	9
	2	Bend	640(9)	gas	PE	9

τ is $\sim 12\%$ greater than for H₂O^{+,3}

$\tilde{X}^2B_1^b$ C_{2v} Structure: EM¹⁰

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	2344(6)	gas	PE	2,4,9
			2326.7	Ne	IR	16
	2	Bend	1044.27(5)	gas	EM	10
b_2	3	Asym. stretch	1040.5	Ne	IR	16
			2392.7	Ne	IR	16

$A_0=16.03$; $B_0=6.240(3)$; $C_0=4.407(3)$ EM¹⁰

^aBest fit of simulated photoelectron spectrum.

^bThe $\tilde{A}^2A_1(\Pi_u)$ and \tilde{X}^2B_1 states are perturbed by strong Renner–Teller interaction.

^cVibrational numbering uncertain.

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H₂Cl⁺

\tilde{X} C_{2v} Structure: DL¹MW^{2,4}LD³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	2643.22	gas	LD	3
	2	Bend	1184.13	gas	DL	1
b_2	3	Asym. stretch	2630.14	gas	LD	3

$A_0=11.253$; $B_0=9.124$; $C_0=4.941$ DL¹MW^{2,4}LD³

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HXeH

\tilde{X} $D_{\infty h}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π_u	2	Bend	700.8	Xe	IR	1,4
Σ_u^+	3	Asym. stretch	1170.0T	Ne	IR	6
			1163.2T			
			1180.6	Xe	IR	1–5
			1165.9			

DXeD

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Π_u	2	Bend	513.5wT	Xe	IR	1
Σ_u^+	3	Asym. stretch	856.2 845.8	Xe	IR	1,4

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8.2. Triatomic Monohydrides

NaCH

$\tilde{X}^3\Sigma^-$ $C_{\infty v}$ Structure: MW^{1,2}
 $B_0 = 0.378$ MW^{1,2}

NaCD

$\tilde{X}^3\Sigma^-$ $C_{\infty v}$
 $B_0 = 0.334$ MW^{1,2}

References

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KCH

$\tilde{X}^3\Sigma^-$ $C_{\infty v}$ Structure: MW¹
 $B_0 = 0.250$ MW¹

KCD

$\tilde{X}^3\Sigma^-$ $C_{\infty v}$
 $B_0 = 0.221$ MW¹

Reference

- J. Xin and L. M. Ziurys, *J. Chem. Phys.* **110**, 3360 (1999).

YNH

$\tilde{B}^2\Sigma^+$ $C_{\infty v}$ Structure: LF^{2,3}
 $T_0 = 16992.60$ gas LF¹⁻³
 $B_0 = 0.330$ LF^{2,3} $\tilde{B}-\tilde{X}$ 585–650 nm

$\tilde{A}^2\Pi$ $C_{\infty v}$ Structure: LF³
 $T_0 = 15425.61$ gas LF³
 $A_0 = 443.76$ gas LF³
 $B_0 = 0.331$ LF³ $\tilde{A}-\tilde{X}$ 636–660 nm

$\tilde{A}''^2\Pi_{1/2}$ $C_{\infty v}$ Structure: LF³
 $T_0 = 14742.77$ gas LF³
 $\tau = 260(10)$ ns gas LF³
 $B_0 = 0.330$ LF³ $\tilde{A}''-\tilde{X}$ 675–680 nm

$\tilde{X}^2\Sigma^+$		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Π	2	Bend	510	gas	LF	1
Σ^+	3	YN stretch	810	gas	LF	1

$B_0 = 0.340$ gas LF^{2,3}

YND

$\tilde{B}^2\Sigma^+$ $C_{\infty v}$
 $T_0 = 16949.25$ gas LF¹⁻³
 $B_0 = 0.285$ LF^{2,3} $\tilde{B}-\tilde{X}$ 585–650 nm

$\tilde{A}^2\Pi$ $C_{\infty v}$
 $T_0 = 15396.05$ gas LF³
 $B_0 = 0.292$ LF³

$\tilde{X}^2\Sigma^+$		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ^+	1	ND stretch	2230	gas	LF	1
Π	2	Bend	400	gas	LF	1
Σ^+	3	YN stretch	770	gas	LF	1

$B_0 = 0.300$ LF^{2,3}

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HCC⁺

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	549.5	Ar	IR	1
Σ ⁺	3	CC stretch	1832.2	Ne	IR	1
			1820.4	Ar	IR	1

DCC⁺

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	CD stretch	2496.0	Ar	IR	1
Π	2	Bend	433.8	Ar	IR	1
Σ ⁺	3	CC stretch	1735.1	Ne	IR	1
			1724.6	Ar	IR	1

Reference

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MgOH

$\tilde{X} \ ^2\Sigma^+$		$C_{\infty v}^a$		Structure: MW ^{1,3}		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	188(2)H	gas	LF	2
Σ ⁺	3	MgO stretch	750(3)	gas	LF	2

$B_0 = 0.494$ MW¹

MgOD

$\tilde{X} \ ^2\Sigma^+$		$C_{\infty v}^a$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	130(2)H	gas	LF	2

$B_0 = 0.448$ MW^{1,3}

^aAnalysis⁴ of rotational transitions which arise from states of MgOH and MgOD in which ν_2 is vibrationally excited indicates that these species are quasilinear.

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MgSH

\tilde{X} C_s Structure: MW¹
 $A_0 = 9.640(8)$; $B_0 = 0.227$; $C_0 = 0.221$ MW¹

Reference

¹A. Taleb-Bendiab and D. Chomiak, Chem. Phys. Lett. **334**, 195 (2001).

SrOH

$\tilde{F} \ ^2\Pi$		$C_{\infty v}$					$\tilde{F}-\tilde{X}$ 297–304 nm	
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.		
Σ ⁺	3	SrO stretch	592T	gas	LF	16		

$A = 22$ gas LF¹⁶

$\tilde{E} \ ^2\Sigma^+$		$C_{\infty v}$					$\tilde{E}-\tilde{X}$ 308–334 nm	
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.		
Σ ⁺	3	SrO stretch	612T	gas	LF	16		

$\tilde{D} \ ^2\Sigma^+$		$C_{\infty v}$					$\tilde{D}-\tilde{X}$ 353–361 nm	
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.		
Σ ⁺	3	SrO stretch	630T	gas	LF	16		

$\tilde{C} \ ^2\Pi$		$C_{\infty v}$					$\tilde{C}-\tilde{X}$ 354–383 nm	
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.		
Π	2	Bend	255T (ω)	gas	LF	16		

$A = 24.52(4)$ gas LF¹⁶
 $B = 0.256$ LF¹⁶

$\tilde{B} \ ^2\Sigma^+$ $C_{\infty v}$ Structure: LF³
 $T_0 = 16377.505(1)$ gas CL²LF^{3,7,11} $\tilde{B}-\tilde{X}$ 580–611 nm
 Absorption maximum at 16553(15) in a krypton matrix.⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	400.84	gas	LF	3,11
Σ ⁺	3	SrO stretch	533.7	gas	LF	15

$B_0 = 0.252$ LF³

$\tilde{A}^2\Pi$ $C_{\infty v}$
 $T_0 = 14674.332(2)$ gas $CL^2LF^{6,7,12,13}$ $\tilde{A}-\tilde{X}$ 645–695 nm
 Absorption maximum at 14598(15) in a krypton matrix.⁴ An incompletely resolved absorption at 14777(15) may be contributed either by SrOH trapped in another matrix site or by the excitation of bending vibration in the \tilde{A} state.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	381.36	gas	LF	12
Σ^+	3	SrO stretch	542.6	gas	LF	6,13

$A = 263.93$ gas $LF^{6,13}$
 $\epsilon = -0.0791$ gas LF^{12}
 $B_0 = 0.254$ gas LF^6

$\tilde{X}^2\Sigma^+$ $C_{\infty v}$ Structure: LF^3MW^9

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	OH stretch	3766(10)	gas	LF	8
Π	2	Bend	363.69	gas	LF	3,6,11,16
Σ^+	3	SrO stretch	526.99	gas	LF	3,6,13,16
			479.3	Ar	IR	5

$B_0 = 0.249$ $LF^{3,6,10}MW^{9,10}$

SrOD

$\tilde{B}^2\Sigma^+$ $C_{\infty v}$ Structure: LF^3
 $T_0 = 16366.0983$ gas $LF^{3,14}$ $\tilde{B}-\tilde{X}$ 607–611 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	311(10)	gas	LF	3
Σ^+	3	SrO stretch	523.65	gas	LF	3,14

$B_0 = 0.228$ $LF^{3,14}$

$\tilde{X}^2\Sigma^+$ $C_{\infty v}$ Structure: LF^3

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	282(10)	gas	LF	3
Σ^+	3	SrO stretch	510(10)	gas	LF	3
			470.6	Ar	IR	5

$B_0 = 0.225$ $LF^{3,14}MW^9$

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SrSH

\tilde{C}^2A' C_s
 $T_0 = 15026$ gas LF^1 $\tilde{C}-\tilde{X}$ 642–690 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	3	SrS stretch	269	gas	LF	1

\tilde{B}^2A'' C_s
 $T_0 = 14815$ gas LF^1 $\tilde{B}-\tilde{X}$ 675–688 nm

\tilde{A}^2A' C_s
 $T_0 = 14293$ gas LF^1 $\tilde{A}-\tilde{X}$ 686–727 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	3	SrS stretch	270	gas	LF	1

\tilde{X}^2A' C_s Structure: MW^2

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	3	SrS stretch	270	gas	LF	1

$A_0 = 9.709(3)$; $B_0 = 0.096$; $C_0 = 0.095$ MW^2

SrSD

\tilde{X}^2A' C_s
 $A_0 = 4.993$; $B_0 = 0.094$; $C_0 = 0.092$ MW^2

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BaOH

$\tilde{D}^2\Sigma^+$ $C_{\infty v}$
 $T_0 = 23057$ gas LF^8 $\tilde{D}-\tilde{X}$ 413–454 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	290T	gas	LF	8
Σ^+	3	BaO stretch	563	gas	LF	8

$B_0 = 0.255(5)$ LF^8

$\tilde{C}^2\Pi$ $C_{\infty v}$

James and Sugden¹ first proposed the assignment to the $\tilde{C}^2\Pi-\tilde{X}^2\Sigma^+$ transition of BaOH emission bands near 500 nm in flames produced in the Ba/H₂/O₂/N₂ system. Laser-excited fluorescence studies on the jet-cooled molecule by Ref. 8 support this identification, but a detailed assignment of the BaOH bands in this spectral region has not yet been achieved.

 $\tilde{B}^2\Sigma^+$ $C_{\infty v}$

$T_0=13200.007(2)$ gas LF^{4,6} $\tilde{B}-\tilde{X}$ 710–757 nm
Absorption maximum at 13105(15) in a krypton matrix.²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	3	BaO stretch	461.0(3)	gas	LF	4

 $B_0=0.213$ LF⁴ $\tilde{A}^2\Pi$ $C_{\infty v}$

$T_0=11760(2)$ gas LF⁵ $\tilde{A}-\tilde{X}$ 827–874 nm
Absorption maximum at 11892(15) in a krypton matrix.²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	352	gas	LF	5
Σ^+	3	BaO stretch	458	gas	LF	5

 $A=635(1)$ LF⁴ $\tilde{A}'^2\Delta$ $C_{\infty v}$

gas LF⁵ $\tilde{A}'-X$ 848–888 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	342	gas	LF	5
Σ^+	3	BaO stretch	468	gas	LF	2

 $\tilde{X}^2\Sigma^+$ $C_{\infty v}$ Structure: LF⁴MW⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	341.6(6)	gas	LF	4,8
Σ^+	3	BaO stretch	492.4(8)	gas	LF	4,8
			430.1	Ar	IR	3

 $B_0=0.217$ LF⁴MW⁷

BaOD

 $\tilde{B}^2\Sigma^+$ $C_{\infty v}$

$T_0=13177.318(3)$ gas LF⁴ $\tilde{B}-\tilde{X}$ 730–759 nm
 $B_0=0.19$ gas LF⁴

 $\tilde{A}^2\Pi$ $C_{\infty v}$

$T_0=11754(2)$ gas LF⁵ $\tilde{A}-\tilde{X}$ 828–878 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	268	gas	LF	5
Σ^+	3	BaO stretch	451	gas	LF	5

 $\tilde{A}'^2\Delta$ $C_{\infty v}$

gas LF⁵ $\tilde{A}'-X$ 886–889 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	258	gas	LF	5
Σ^+	3	BaO stretch	469	gas	LF	5

 $\tilde{X}^2\Sigma^+$ $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	257.6(4)	gas	LF	4
Σ^+	3	BaO stretch	482.4(2)	gas	LF	4
			413.6	Ar	IR	3

 $B_0=0.196$ LF⁴MW⁷

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HScO

 \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	ScH stretch	1391.1	Ar	IR	1
	3	ScO stretch	922.3	Ar	IR	1

DScO

 \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	ScD stretch	1004.2	Ar	IR	1
	3	ScO stretch	922.3	Ar	IR	1

Reference

- L. Zhang, J. Dong, and M. Zhou, *J. Phys. Chem. A* **104**, 8882 (2000).

ScOH

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ^+	3	ScO stretch	765.6	Ar	IR	1

ScOD

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
	3	ScO stretch	750.5	Ar	IR	1

Reference

¹L. Zhang, J. Dong, and M. Zhou, *J. Phys. Chem. A*, **104**, 8882 (2000).

HYO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	YH stretch	1316.2	Ar	IR	1
	3	YO stretch	809.6	Ar	IR	1

DYO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	YD stretch	945.8	Ar	IR	1
	3	YO stretch	807.9	Ar	IR	1

Reference

¹L. Zhang, L. Shao, and M. Zhou, *Chem. Phys.* **272**, 27 (2001).

YOH

$\tilde{C}^1\Sigma^+$		$C_{\infty v}$				
$T_0 = 18508.6$		gas LF ¹				
		$\tilde{C}-\tilde{X}$ 508–540 nm				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Π	2	Bend	457	gas	LF	1
Σ^+	3	YO stretch	575	gas	LF	1

$B_0 = 0.271$ LF¹

$\tilde{B}^1\Pi^a$		$C_{\infty v}$				
$T_0 = 16448.5$		gas LF ¹				
		$\tilde{B}-\tilde{X}$ 539–613 nm				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ^+	3	YO stretch	565.20	gas	LF	1

$B_0 = 0.270$ LF¹

$\tilde{X}^1\Sigma^+$		$C_{\infty v}$				
		Structure: LF ¹				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Π	2	Bend	313.73	gas	LF	1
Σ^+	3	YO stretch	673.83(2) 662.2wT	gas Ar	LF IR	1 2

$B_0 = 0.290$ LF¹

YOD

$\tilde{C}^1\Sigma^+$		$C_{\infty v}$				
$T_0 = 18472.5$		gas LF ¹				
		$\tilde{C}-\tilde{X}$ 510–542 nm				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Π	2	Bend	339	gas	LF	1
Σ^+	3	YO stretch	568	gas	LF	1

$B_0 = 0.247$ LF¹

$\tilde{B}^1\Pi^a$		$C_{\infty v}$				
$T_0 = 16464.0$		gas LF ¹				
		$\tilde{B}-\tilde{X}$ 569–612 nm				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ^+	3	YO stretch	550.86	gas	LF	1

$B_0 = 0.246$ LF¹

$\tilde{X}^1\Sigma^+$		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Π	2	Bend	237.43(2)	gas	LF	1
Σ^+	3	YO stretch	655.34(3) 651.2wT	gas Ar	LF IR	1 2

$B_0 = 0.261$ LF¹

^aAt equilibrium, the molecule is bent and this state is split into A' and A'' states, but with a small barrier to linearity.

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HLaO

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	LaH stretch	1180.2	Ar	IR	1
	3	LaO stretch	758.6	Ar	IR	1

DLaO

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>			753.1	Ar	IR	1

Reference

¹L. Zhang, L. Shao, and M. Zhou, Chem. Phys. **272**, 27 (2001).

HTiO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	TiH stretch	1483.3	Ar	IR	1
	3	TiO stretch	968.8	Ar	IR	1

DTiO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	TiD stretch	1071.1	Ar	IR	1
	3	TiO stretch	967.9	Ar	IR	1

Reference

¹M. Zhou, L. Zhang, J. Dong, and Q. Qin, J. Am. Chem. Soc. **122**, 10680 (2000).

HZrO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	ZrH stretch	1544.7	Ar	IR	1
	3	ZrO stretch	908.8	Ar	IR	1

DZrO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	ZrD stretch	1110.1	Ar	IR	1
	3	ZrO stretch	907.4	Ar	IR	1

Reference

¹M. Zhou, L. Zhang, J. Dong, and Q. Qin, J. Am. Chem. Soc. **122**, 10680 (2000).

HHfO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	HfH stretch	1626.5	Ar	IR	1
	3	HfO stretch	902.9	Ar	IR	1

DHfO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	3	HfO stretch	900.9	Ar	IR	1

Reference

¹M. Zhou, L. Zhang, J. Dong, and Q. Qin, J. Am. Chem. Soc. **122**, 10680 (2000).

CuOH

\tilde{B}^1A'' C_s Structure: LF^{1,3}
 $T_0 = 18406.779$ gas CL¹LF^{1,3} $\tilde{B}-\tilde{X}$ 500–560 nm
 $A_0 = 25.898$; $B_0 = 0.382$; $C_0 = 0.376$ LF^{1,3}

\tilde{A}^1A' C_s Structure: LF⁴
 $T_0 = 15911.546$ gas LF⁴
 $A_0 = 18.316$; $B_0 = 0.388$; $C_0 = 0.378$ LF⁴

\tilde{X}^1A' C_s Structure: LF^{1,3}MW^{5,6}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	2	Bend	743(1)	gas	LF	1
			727.7	Ar	IR	2
	3	CuO stretch	632.7	Ar	IR	2

$A_0 = 23.039$; $B_0 = 0.392$; $C_0 = 0.385$ LF^{1,3}MW⁵

CuOD

\tilde{B}^1A'' C_s $\tilde{B}-\tilde{X}$ 500–560 nm
 $T_0=18422.335$ gas LF^{1,3}
 $A_0=14.235$; $B_0=0.354$; $C_0=0.344$ LF^{1,3}

\tilde{A}^1A' C_s
 $T_0=15911.095(3)$ gas LF⁴
 $A_0=10.714$; $B_0=0.361$; $C_0=0.346$ LF⁴

\tilde{X}^1A' C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	Bend	537(1)	gas	LF	1
			533.6	Ar	IR	2
	3	CuO stretch	635.1	Ar	IR	2

$A_0=12.426$; $B_0=0.366$; $C_0=0.354$ LF^{1,3}MW⁶

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AgOH

\tilde{X} C_s Structure: MW^{1,2}
 $A=22.22(3)$; $B=0.277$; $C=0.273$ MW¹

AgOD

\tilde{X} C_s MW²
 $A=11.92$; $B=0.259$; $C=0.253$ MW²

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- C. J. Whitham, H. Ozeki, and S. Saito, J. Chem. Phys. **110**, 11109 (1999).
- C. J. Whitham, H. Ozeki, and S. Saito, J. Chem. Phys. **112**, 641 (2000).

YbOH

$\tilde{A}^2\Pi$ $C_{\infty v}$ $\tilde{A}-\tilde{X}$ 520–580 nm
 $T_0=17998.62$ gas LF¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	357(5)	gas	LF	1
Σ^+	3	YbO stretch	573(5)	gas	LF	1

$A=1350T$ gas LF¹
 $B_0=0.253$ LF¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	339(5)	gas	LF	1
Σ^+	3	YbO stretch	529.34	gas	LF	1

$B_0=0.245$ LF¹

Reference

- T. C. Melville and J. A. Coxon, J. Chem. Phys. **115**, 6974 (2001).

HCC

$3p\sigma$ Rydberg state^a $C_{\infty v}$
 $T_0=72100(1300)$ gas MPI³⁴

$T_0=51387(25)T$ Ar AB⁸ 195–160 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CC stretch	2175(25)	Ar	AB	8
		Bend	630(25) ^b	Ar	AB	8

\tilde{B}^2A' C_s Structure: LF⁴⁶
 $T_0\leq 39157.4$ gas LF^{35,36,40,46}EM³⁹ $\tilde{B}-\tilde{X}$ 250–313 nm
 Approximately 30 emission bands of HCC have been assigned to transitions between a few common upper-state energy levels and HCC (\tilde{X}) with appreciable bending excitation. Analysis of these bands has yielded accurate vibrational and rotational constants for several excited ground-state vibrational levels. However, a definitive assignment of the upper-state energy levels has not been achieved.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	Bend	775T	gas	LF	46
	3	CC stretch	1221T	gas	LF	46

$\tau_0=287(13)$ ns gas LF⁴⁷
 $A_T=18.82(3)$; $B_T=1.278$; $C_T=1.195$ LF⁴⁶
 $T_0\leq 29360$ Ar AB^{2,8}

An absorption band system between 340 and 246 nm, with band spacings of approximately 2700, 1300 and 840 cm⁻¹, which has been observed on vacuum UV photolysis of C₂H₂ in an argon matrix is tentatively attributed to HCC.

$\tilde{A}^2\Pi$ $C_{\infty v}$
 $T_0^c=3692.61$ gas CC^{10,14}PE^{32,43}
 3685.8 Ne AB³⁸
 3732 Ar AB^{21,38}

In neon and argon matrices, a complicated absorption band system of HCC extends from approximately 3600 to 9000.^{21,38} This band system is extensively perturbed by high vibrational levels of the ground state. A few of the individual bands have been observed in the gas phase by color-center laser absorption^{10,14,29}, by high resolution emission spectroscopy,²⁴ or by time-resolved emission spectroscopy.³⁰ Because of the extensive perturbations and because of the high energy input in the gas-phase studies, high ground-state vibrational levels are prominent both in the gas phase^{10,14,19,26,27,32,37,40–42,46} and in neon and argon matrices.^{20,21,38}

Quasicontinuous 400–900 nm emission results on 136–110 nm photolysis of C₂H₂ or HCCBr in the gas phase.^{5,11,13,17} The fluorescence lifetimes vary from 6 to 20 μ s.^{11,13,17} Unstructured emission from 1 to 5 μ m has been detected¹⁸ upon 193-nm photolysis of gas-phase C₂H₂, with maximum intensity between 3600 and 5000. The HCC fluorescence resulting from the

193-nm photolysis of HCCBr extends from 500 nm to 5 μm ,¹⁸ with lifetime increasing from ca. 5 μs near 500 nm to ca. 60 μs near 4000. Unstructured HCC emission between 400 and 500 nm has also been observed¹⁵ on vacuum UV irradiation of C₂H₂ isolated in the solid rare gases.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	3	CC stretch	1706.2T	Ne	AB	38

$\bar{X}^2\Sigma^+$		$C_{\infty v}$	Structure: MW ³¹			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CH stretch	3298.85T	gas	CC	29
Π	2	Bend	371.60 ^d	gas	DL	26,27
Σ^+	3	CC stretch	1840.57	gas	DL	23
			1835.5	Ne	IR	38,45
			1846.2m	Ar	IR	1,3,20,21,45

$A \approx 10$ IR¹⁴

$B_{000} = 1.457$ MW^{4,6,7,9,48} LMR¹²

$B_{020} = 1.451$ LMR²⁸

DCC

3p σ Rydberg state^a $C_{\infty v}$
 $T_0 = 72100(1300)$ gas MPI³⁴
 $T_0 = 51493(25)$ T Ar AB⁸ 194–170 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CC stretch	2183(25)	Ar	AB	8
		Bend	520(25) ^b	Ar	AB	8

Fluorescence observed on laser excitation between 34500 and 40000 (250–290 nm) of gas-phase DCC from 2 to 10 μs after its formation by 193 nm photolysis of C₂D₂ has been attributed³⁵ to transitions of DCC between high vibrational levels of the ground state and an undetermined excited state, probably \bar{B}^2A' .

$\bar{A}^2\Pi$ $C_{\infty v}$
 $T_0 = 3594(50)$ T gas PE⁴³
 <3800 Ar AB²¹

A complicated absorption band system extends to approximately 9700 in neon and argon-matrix studies of DCC.^{21,38} As for HCC, the band system is extensively perturbed by high vibrational levels of the ground state. A few of the bands, some of which arise from excited ground-state vibrational energy levels, have been studied in the gas phase using infrared laser absorption^{22,29,33,44} and photoelectron spectroscopy.⁴³

$\bar{X}^2\Sigma^+$		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CD stretch	2537.1	Ne	IR	38
Π	2	Bend	288.21	gas	PE,LMR	32,44
Σ^+	3	CC stretch	1743.18	gas	DL	25
			1739.6	Ne	IR	38,45
			1746.3m	Ar	IR	1,3,20,21,45

$B_0 = 1.203$ MW¹⁶

^aTentative assignment.

^bObserved band spacing; $2\nu_2$ if upper state is linear.

^cThe $\bar{A}(000)$ band of HCC is strongly perturbed by high vibrational energy levels of the ground state which possess $^2\Pi$ vibronic symmetry. Accordingly, there is extensive mixing of the $^2\Pi$ vibronic levels in this spectral region, and designation of the band origin is approximate.^{10,14,38,41}
^dDerived from $(\nu_2 + \nu_3) - [(\nu_2 + \nu_3) - \nu_2]$. The detailed assignment of $(\nu_2 + \nu_3)$ is given in Ref. 27 and that of $(\nu_2 + \nu_3) - \nu_2$ in Ref. 26.

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HCSi

$\tilde{A}^2\Sigma^+$ $C_{\infty v}$ Structure: LF^{4,5}
 $T_0 = 11766.72$ gas AB³LF^{4,5,7,8}EM⁶ $\tilde{A}-\tilde{X}$ 605–1300 nm
 11749(3)^a Ne AB¹ $\tilde{A}-\tilde{X}$ 656–852 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	715	gas	LF	4,5
Σ ⁺	3	CSi stretch	1167.7	gas	AB,LF,EM	3,5,6
			1170(5)	Ne	AB	1

$B_0 = 0.635$ AB³LF⁵EM⁶

$\tilde{X}^2\Pi$ $C_{\infty v}$ Structure: LF^{4,5}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend (ω)	508.0(4)	gas	LF	7
		(κ ² Σ)	606.9	gas	LF	7
		(² Δ _{3/2})	576.0(2)	gas	LF	7
		(² Δ _{5/2})	505.4(2)	gas	LF	7
		(μ ² Σ)	442.7(2)	gas	LF	7
Σ ⁺	3	CSi stretch	1014.29	gas	AB,EM	3,6
			1010.4	Ar	IR	2

$A_0 = -69.81$ gas AB³LF^{4,5}
 $B_0 = 0.580$ AB³LF^{4,5}EM⁶

DCSi

$\tilde{A}^2\Sigma^+$ $C_{\infty v}$ Structure: LF^{5,7}
 $T_0 = 11709.47$ gas LF^{5,7} $\tilde{A}-\tilde{X}$ 617–1290 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	558	gas	LF	5
Σ ⁺	3	SiC stretch	1127	gas	LF	5

$B_0 = 0.543$ LF⁵

$\tilde{X}^2\Pi$ $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend (ω)	391.0(5)	gas	LF	7
		(κ ² Σ)	482.9	gas	LF	7
		(² Δ _{3/2})	460.3(2)	gas	LF	7
		(² Δ _{5/2})	389.8(2)	gas	LF	7
		(μ ² Σ)	350.2(2)	gas	LF	7
Σ ⁺	3	CSi stretch	980.5	gas	LF	7
			977.4	Ar	IR	2

$A_0 = -69.94$ gas LF⁵
 $B_0 = 0.499$ LF⁵

^aReassigned to HCSi by Ref. 3.

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HCGe

$\tilde{A}^2\Sigma^+$ $C_{\infty v}$ Structure: LF²
 $T_0 = 13901.83$ gas LF^{1,2} $\tilde{A}-\tilde{X}$ 555–730 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	638	gas	LF	2
Σ ⁺	3	CGe stretch	990	gas	LF	2

$B_0 = 0.480$ LF²

$\tilde{X}^2\Pi$ $C_{\infty v}$ Structure: LF²
 $A = -334.7$ gas LF^{1,2}
 $B_0 = 0.429$ LF²

DCGe

$\tilde{A}^2\Sigma^+$ $C_{\infty v}$ Structure: LF²
 $T_0 = 13845.33$ gas LF² $\tilde{A}-\tilde{X}$ 555–730 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	497	gas	LF	2
Σ ⁺	3	CGe stretch	952	gas	LF	2

$B_0 = 0.409$ LF²

$\tilde{X}^2\Pi$ $C_{\infty v}$
 $B_0 = 0.368$ LF²

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cyc-HSi₂

\tilde{A}^2B_1 C_{2v}
 $T_0 = 160T$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	SiSi stretch	520(20)	gas	PE	1

\tilde{X}^2A_1 C_{2v}

Reference

¹C. Xu, T. R. Taylor, G. R. Burton, and D. M. Neumark, *J. Chem. Phys.* **108**, 7645 (1998).

HCN⁺

\tilde{B} C_s
 $T_0 = 44140(10)$ gas PE^{2,4,7}
 PEPICO measurements⁷ indicate that this state is strongly predissociated. H⁺ and, at higher energies, CN⁺ have been detected.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CH stretch	1194(10)	gas	PE	7
	2	Bend	331(10)	gas	PE	7

$\tilde{A}^2\Sigma^+$ $C_{\infty v}$
 $T_0 = 3260(30)$ gas PE²TPE⁶

The \tilde{A} and \tilde{X} states of HCN⁺ are strongly coupled by vibronic interaction. A band assignment based on *ab initio* calculations³ is supported by the results of the TPE study.⁶

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CH stretch	3106(40)	gas	PE	2
Π	2	Bend	428(30)H	gas	PE	2
Σ^+	3	CN stretch	2098(30)	gas	PE	2

$\tilde{X}^2\Pi$ $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CH stretch	3070(30)	gas	PE	1
			3049.9	Ne	IR	5
Π	2	Bend	760(30)T ^a	gas	PE	1
Σ^+	3	CN stretch	1800(30)	gas	PE	1,2

DCN⁺

\tilde{B} C_s
 $T_0 = 44400(10)$ gas PE^{2,7}
 PEPICO measurements⁷ indicate that this state is strongly predissociated.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CD stretch	984(10)	gas	PE	7
	2	Bend	250(10)	gas	PE	7

$\tilde{A}^2\Sigma^+$ $C_{\infty v}$
 $T_0 = 3114(30)$ gas PE²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CD stretch	2566(40)	gas	PE	2
Π	2	Bend	323(30)H	gas	PE	2
Σ^+	3	CN stretch	1904(30)	gas	PE	2

$\tilde{X}^2\Pi$ $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CD stretch	2374.1	Ne	IR	5
Σ^+	3	CN stretch	1686(30)	gas	PE	2

^aRef. 2 gives 298(30) for HCN⁺ and 234(30) for HNC⁺.

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HHgCl

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	HgH stretch	2091.9	Ar	IR	1
			2076.0	Kr	IR	1
			2096.0	N ₂	IR	1
	2	Bend	546.9	Ar	IR	1
			543.1			
			540.8	Kr	IR	1
			537.3			
			546.7	N ₂	IR	1
			541.1			

DHgCl

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	HgD stretch	1500.5	Ar	IR	1
			1503.5	N ₂	IR	1

Reference

- ¹N. Legay-Sommaire and F. Legay, *Chem. Phys. Lett.* **314**, 40 (1999).

HTiO⁻ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	TiH stretch	1341.8	Ar	IR	1
	3	TiO stretch	933.4	Ar	IR	1

DTiO⁻ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	TiD stretch	978.3	Ar	IR	1
	3	TiO stretch	931.1	Ar	IR	1

Reference

¹M. Zhou, L. Zhang, J. Dong, and Q. Qin, *J. Am. Chem. Soc.* **122**, 10680 (2000).

HBO \tilde{X} $C_{\infty v}$ Structure: MW^{3,4,8}IR⁸

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	754.42	gas	DL	5
			756.1m	Ar	IR	1,6,7
Σ ⁺	3	BO stretch	1825.56	gas	DL	2
			1822.3s	Ar	IR	1,6,7

$B_0 = 1.308$ DL²MW^{3,4}

DBO \tilde{X} $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	BD stretch	2253.53	gas	DL,EM	8
			2258.9w	Ar	IR	1,7
Π	2	Bend	608.36	gas	EM	8
			607.6m	Ar	IR	1,7
Σ ⁺	3	BO stretch	1647.69	gas	DL,EM	8
			1650.2m	Ar	IR	1,7

$B_0 = 1.049$ gas MW^{4,8}

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HBCl⁺ \tilde{X} $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	3	BCl stretch	1121.57	gas	DL	1

$B_0 = 0.631$ DL¹

Reference

¹N. T. Hunt, Z. Liu, and P. B. Davies, *Mol. Phys.* **97**, 205 (1999).

HBrBr⁺ \tilde{X} $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	3	BBr stretch	937.57	gas	DL	1

$B_0 = 0.462$ DL¹

Reference

¹N. T. Hunt, D. Collet, Z. Liu, and P. B. Davies, *J. Chem. Phys.* **111**, 5905 (1999).

HCC⁻

Threshold for electron detachment from ground-state HCC⁻ = 23950(50) gas PE^{1,3}

 \tilde{X} $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	CH stretch	3318.5	Ar	IR	4
Π	2	Bend	505(20)	gas	PE	1
Σ ⁺	3	CC stretch	1800(20)	gas	PE	1
			1773.0	Ne	IR	2,4
			1770.5	Ar	IR	4

DCC⁻

Threshold for electron detachment from ground-state DCC⁻ = 23990(50) gas PE^{1,3}

 \tilde{X} $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	395(40)	gas	PE	1
Σ ⁺	3	CC stretch	1705(20)	gas	PE	1
			1675.7	Ne	IR	4
			1676.7	Ar	IR	4

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cyc-HSi₂⁻

Threshold for electron detachment from ground-state *cyc*-HSi₂⁻ = 18640(80) gas PE¹

Reference

- ¹C. Xu, T. R. Taylor, G. R. Burton, and D. M. Neumark, *J. Chem. Phys.* **108**, 7645 (1998).

HCO⁺

\tilde{X}		$C_{\infty v}$	Structure: MW ^{2-4,15}			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CH stretch	3088.74	gas	LD,CC	5,6,16,17
Π	2	Bend	829.72	gas	DL,MPI	10,11,18
Σ^+	3	CO stretch	2183.95	gas	DL	7,8,14

$B_0 = 1.488$ MW²⁻⁴

DCO⁺

\tilde{X}		$C_{\infty v}$	Structure: MW ^{4,5,9,12}			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CD stretch	2584.56	gas	DL	12
Π	2	Bend	666.0	gas	PE,MPI	1,13,19,20
Σ^+	3	CO stretch	1904.06	gas	DL	9

$B_0 = 1.201$ MW²⁻⁴

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HNC

\tilde{A}^a
 $T_0 = 32850$ gas AB⁷ $\tilde{A}-\tilde{X}$ 250-305 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CN stretch	1005	gas	UV	7

\tilde{X} $C_{\infty v}$ Structure: MW^{4,5,9,12}
 Higher vibrational levels of HNC (\tilde{X}) have been analyzed by Ref. 13.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	NH stretch	3652.66	gas	IR,EM	3,6,8,11,15-17
			3643.1	Ne	IR	10
			3620s	Ar	IR	2
			3583s N ₂	Ar	IR	1,2
			3598.6	Kr	IR	14
			3577.0	Xe	IR	14
Π	2	Bend	462.72	gas	IR,EM	8,17
			477s	Ar	IR	2
			535s N ₂	Ar	IR	1
			478.2	Kr	IR	14
			477.0	Xe	IR	14
			559s	N ₂	IR	2
Σ^+	3	NC stretch	2023.86	gas	IR,EM	8,15,17
			2025.4	Ne	IR	10
			2029w	Ar	IR	2
			2032w N ₂	Ar	IR	1
			2024.7	Kr	IR	14
			2021.0	Xe	IR	14
			2035w	N ₂	IR	2

$B_0 = 1.512$ MW^{4,18}IR⁸

DNC

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	ND stretch	2787.07	gas	IR	3,6
			2780.7	Ne	IR	10
			2769s	Ar	IR	2
			2733s N ₂	Ar	IR	1
			2728s	N ₂	IR	2
II	2	Bend	374s	Ar	IR	2
			413s N ₂	Ar	IR	1
			432s	N ₂	IR	2
Σ^+	3	NC stretch	1938.7	Ne	IR	10
			1940w	Ar	IR	2
			1940w N ₂	Ar	IR	1
			1937w	N ₂	IR	2

$B_0 = 1.273$ MW⁴

^aTentative identification.

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HOC⁺

\tilde{X}		$C_{\infty v}$ Structure: MW ¹⁻³				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	OH stretch	3268.03	gas	LD	4

$B_0 = 1.492$ MW^{1,2,5}LD⁴

DOC⁺

\tilde{X}
 $B_0 = 1.274$ MW^{3,5} $C_{\infty v}$

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HCO

$4p^2\Pi(A'')$ C_s
 $T_0 = 64073.5$ gas DR^{40,42}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	Bend	863.8(5)	gas	DR	40,42

$3p^2\Pi$ $C_{\infty v}$ Structure: MPI,LF^{40,46}
 $T_0 = 45540.1(3.3)$ gas MPI^{19,20,28} $3p^2\Pi - \tilde{X}$ 187–222 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
II	2	Bend	799.9(1.4) ^a	gas	MPI	28
Σ^+	3	CO stretch	2177(3)	gas	MPI	20

$A = 0.9(3)$; $\epsilon = 0.071$ gas MPI²⁸
 $B = 1.492(12)$ MPI²⁸

\tilde{B}^2A' C_s Structure: MPI,LF^{40,46}
 $T_0 = 38695.48$ gas
 EM⁶LF^{25,30,33,35,36,39,43}MPI²⁹SEP^{30,35} $\tilde{B} - \tilde{X}$ 235–475 nm
 38595(35) Ar AB^{5,11} $\tilde{B} - \tilde{X}$ 210–260 nm
 38567(35) CO AB⁵ $\tilde{B} - \tilde{X}$ 210–260 nm
 Lifetime measurements^{32,35,37,38} give evidence for predissociation. Large decrease in fluorescence quantum yield above 41465.³³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CH stretch	2596.4(2)	gas	EM,LF	6,25,29,
				Ar	MPI	30,33
				CO	AB	5,11
				gas	LF,MPI	25,29,30
						33
2	Mixed		1375(35)	Ar	AB	5,11
			1375(35)	CO	AB	5
			1065.9(2)	gas	LF,MPI	25,29,30
					33	
					5,11	
3	Mixed		1035(35)	Ar	AB	5,11
			1035(35)	CO	AB	5

$A_0 = 15.976$; $1/2(B_0 + C_0) = 1.151$; $1/4(B_0 - C_0) = 0.021$ LF^{35,43}
 $\tau_0 = 89(2)$ ns gas LF^{27,32,35,37,38}

$\tilde{A}^2A''(\Pi)$ $C_{\infty v}$
 $T_0 = 9297(3)$ gas $AB^{1,3,8}LF^{24,26}DM^{31}CR^{41}$ $\tilde{A}-\tilde{X}$ 460–860 nm

Bands with $K' > 0$ are diffuse.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CH stretch	3319(3)	gas	AB	1,3,8
	2	Bend	805	gas	AB	1,3,8
	3	CO stretch	1812.2	gas	AB	1,3,8

$\tau_{090} = 46(4)$ ns LF^{15}

$B_0 = 1.34$ $UV^{1,3,8}$

\tilde{X}^2A' C_s Structure: $MW^7 UV^8$

Vibrational term energies up to 20777 have been reported by Ref. 36.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.			
a'	1	CH stretch	2434.48	gas	LF,PE	17,18,25			
					DL,LD	22,26			
					IR	23			
					2483m	Ar	IR	5	
					2442.3	Xe	IR	45	
					2488m	CO	IR	4	
					1080.76	gas	Bend	UV,LS	1,3,8
								LMR	9,10
								LF	25,26
					1087s	Ar	IR	5	
					1076.5	Xe	IR	44,45	
					1090s	CO	IR	2,4	
					1868.17	gas	CO stretch	IR	12,23
								LMR	13,25
								IR	5
1863vs	Ar	IR	44,45						
1858.4	Xe	IR	44,45						
1856.6	CO	IR	2,4						
1861vs									

$A_0 = 24.329$; $B_0 = 1.494$; $C_0 = 1.399$ $UV^{1,3,8}MW^{16}$

DCO

$4p^2\Pi(A'')$ C_s
 $T_0 = 64033.3$ gas $DR^{40,42}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	Bend	677.3(5)	gas	DR	40,42

$3p^2\Pi$ $C_{\infty v}$
 $T_0 = 45444.0(3.6)$ gas $MPI^{20,28}$ $3p^2\Pi-\tilde{X}$ 187–229 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	645.5(1.6) ^a	gas	MPI	28
Σ^+	3	CO stretch	1900(5)	gas	MPI	20

$A = 0.9(3)$; $\epsilon = 0.069$ MPI^{28}

$B = 1.221(12)$ MPI^{28}

\tilde{B}^2A' C_s
 $T_0 = 38629.50(3)$ gas $MPI^{29}LF^{34,40,46}$ $\tilde{B}-\tilde{X}$ 236–450 nm
 38568(70) Ar AB^5 $\tilde{B}-\tilde{X}$ 200–260 nm
 38569(35) CO AB^5 $\tilde{B}-\tilde{X}$ 204–260 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CD stretch	1944.2(2)	gas	MPI,LF	29,40,42,46
			1211.8(8)	gas	MPI,LF	29,40,42,46
			1150(35)	Ar	AB	5
	3	Bend+CO stretch	1150(35)	CO	AB	5
			921.6(8)	gas	MPI,LF	29,40,42,46
			925(35)	Ar	AB	5
			925(35)	CO	AB	5

$A_0 = 9.172$; $B_0 = 1.093$; $C_0 = 0.974$ $LF^{40,46}$

$\tilde{A}^2A''(\Pi)$ $C_{\infty v}$
 $T_0 = 9162(3)$ gas $UV^{1,3,8}LF^{24,26}$ $\tilde{A}-\tilde{X}$ 460–860 nm
 Bands with $K' > 0$ are diffuse.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CD stretch	2547(2)	gas	UV	1,3,8
Π	2	Bend	641.7(7)	gas	UV	1,3,8

$B_0 = 1.10$ $UV^{1,3,8}$

\tilde{X}^2A' C_s
 Vibrational term energies up to 18186 have been reported by Ref. 42

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.		
a'	1	CD stretch	1909.77	gas	LMR,LF,SEP	14,26,40,42		
			1926s	Ar	IR	5		
			1937s	CO	IR	4		
			2	Bend	846.5	gas	UV,LF,SEP	1,3,8,26,40,42
					850s	Ar	IR	5
					852s	CO	IR	2,4
			3	CO stretch	1794.59	gas	LMR,LF,SEP	14,26,40,42
					1803m	Ar	IR	5
					1800m	CO	IR	2,4

$A_0 = 14.734$; $B_0 = 1.281$; $C_0 = 1.171$ $UV^{1,3,8}MW^{21}$

^a ω_2 .

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HCS

\tilde{X}^2A' C_s
 $B_{\text{eff}}=0.672$ MW¹

Reference

- ¹ H. Habara, S. Yamamoto, C. Ochsenfeld, M. Head-Gordon, R. I. Kaiser, and Y. T. Lee, *J. Chem. Phys.* **108**, 8859 (1998).

HCCI⁺

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1078	Ar	IR	1

Reference

- ¹ T. D. Fridgen, X. K. Zhang, J. M. Parnis, and R. E. March, *J. Phys. Chem. A* **104**, 3487 (2000).

HSC

\tilde{X} C_s Structure: MW¹
 $A_0=9.841$; $B_0=0.702$; $C_0=0.652$ MW¹

DSC

\tilde{X} C_s MW¹
 $A_0=5.201$; $B_0=0.682$; $C_0=0.600$ MW¹

Reference

- ¹ H. Habara and S. Yamamoto, *J. Chem. Phys.* **112**, 10905 (2000).

HSiO

\tilde{X} C_s Structure: MW¹
 $A_0=10.411$; $B_0=0.663$; $C_0=0.621$ MW¹

Reference

- ¹ M. Izuha, S. Yamamoto, and S. Saito, *J. Mol. Struct.* **413/414**, 527 (1997).

HSiS

\tilde{X}^2A' C_s Structure: MW¹
 $A_0=10.197$; $B_0=0.281$; $C_0=0.272$ MW¹

Reference

- ¹ F. X. Brown, S. Yamamoto, and S. Saito, *J. Mol. Struct.* **413/414**, 537 (1997).

HCF

$\tilde{E}^1A'(3p)$ C_s
 $T_0=62154(2)$ gas MPI¹⁵

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	Bend	1128(4)	gas	MPI	15
	3	CF stretch	1614(4)	gas	MPI	15

\tilde{A}^1A'' C_s Structure: $AB^1LF^{5,6}$
 $T_0 = 17277.47$ gas $AB^1CL^3LF^{5,10,16}$ $\tilde{A}-\tilde{X}$ 414–635 nm
 17320(15) Ar AB^2 $\tilde{A}-\tilde{X}$ 469–546 nm
 Evidence has been obtained^{8,9} for perturbation of the \tilde{A} state by high vibrational levels of the ground state and by the low-lying triplet state.

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	CH stretch	2852(2)T	gas	LF	16
	2	Bend	1021.26	gas	AB,LF	1,7
			1000(20)	Ar	AB	2
	3	CF stretch	1260(2)	gas	LF	16

$A_0 = 25.69$; $B_0 = 1.162$; $C_0 = 1.107$ AB^1LF^5
 $\tau_0 = 2.45(10)$ μs gas LF^4
 $\tau_1 = 2.57(16)$ μs ; $\tau_2 = 12.5(8)$ μs gas EM^{13}

\tilde{a}^3A'' C_s
 $T_0 = 5210(140)$ gas $PE^{12,14}$

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	2	Bend	1047(25)	gas	PE	14
	3	CF stretch	1232(25)	gas	PE	14

\tilde{X}^1A' C_s Structure: $AB^1LF^{5,6}SEP^{11}$

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	CH stretch	2643.04	gas	SEP	11
	2	Bend	1403.20	gas	AB,LF	1,7
			1406vw	Ar	IR	2
	3	CF stretch	1189(25)	gas	PE	12,14
			1181.5m	Ar	IR	2

$A_0 = 15.563$; $B_0 = 1.223$; $C_0 = 1.130$ $AB^1LF^5SEP^{11}DR^{17}$

DCF

$\tilde{E}^1A'(3p)$ C_s
 $T_0 = 62175(2)$ gas MPI^{15}

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	CD stretch	2095(5)	gas	MPI	15
	2	Bend	872(3)	gas	MPI	15
	3	CF stretch	1582(4)	gas	MPI	15

\tilde{A}^1A'' C_s
 $T_0 = 17293.426(3)$ gas CL^3LF^6 $\tilde{A}-\tilde{X}$ 460–585 nm

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	2	Bend	780(5)	gas	CL	3

$A_0 = 15.10$; $B_0 = 1.014$; $C_0 = 0.945$ LF^6

\tilde{a}^3A' C_s
 $T_0 < 5140(700)$ gas PE^{12}

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	2	Bend	834(25)	gas	PE	12,14
	3	CF stretch	1216(25)	gas	PE	12,14

\tilde{X}^1A' C_s

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	2	Bend	1046m	Ar	IR	2
	3	CF stretch	1193(25)	gas	PE	12,14
			1183m	Ar	IR	2

$A_0 = 8.828$; $B_0 = 1.120$; $C_0 = 0.990$ LF^6

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HCCI

\tilde{A}^1A'' C_s Structure: AB^8
 $T_0 = 12280.411(2)$ gas $AB^{1,8,9}LF^{4,7,13}$ $\tilde{A}-\tilde{X}$ 550–820 nm
 Ar AB^2 $\tilde{A}-\tilde{X}$ 570–750 nm

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	2	Bend	873.0	gas	AB	1,9
			855(50)	Ar	AB	2
	3	CCl stretch	987.4	gas	AB	9

Barrier to linearity = 2250¹

$B_0 = 0.609$ AB^8

\tilde{a}^3A'' C_s
 $T_0 = 1470(880)$ gas $PE^{5,6}$

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	3	CCl stretch	850(60)	gas	PE	5,6

\tilde{X}^1A'		C_s		Structure: $AB^1LF^{3,13}$		
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	2	Bend	1195	gas	LF	13
			1201wm	Ar	IR	2,11
			1201	Kr	IR	11
	3	CCl stretch	811.60	gas	DL,LF	10,13
			815s	Ar	IR	2,11
			817	Kr	IR	11

$A_0=15.759$; $B_0=0.605$; $C_0=0.581$ $AB^1LF^3MW^{12}$

DCCI

\tilde{A}^1A''		C_s		$\tilde{A}-\tilde{X}$ 550–820 nm		
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	2	Bend	657.2	gas	AB	1

$T_0=12274$ gas AB^1

\tilde{X}^1A'		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	3	CCl stretch	805s	Ar	IR	2

$A_0=8.75$; $B_0=0.557$; $C_0=0.525$ AB^1

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HCB_r

\tilde{A}^1A''		C_s		Structure: AB^4		
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	2	Bend	828.7	gas	LF	3,6
	3	CBr stretch	783T	gas	LF	3

$T_0=11972.43$ gas $LF^3AB^{4-6,8,10}$ $\tilde{A}-\tilde{X}$ 550–955 nm

Barrier to linearity approximately 13590 above $\tilde{X}(000)$.³
 $B_{av}=0.437$ AB^8

\tilde{a}^3A''		C_s		$T_0=2006(8)$ gas LF^9		
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	2	Bend	994	gas	LF	9
	3	CBr stretch	733	gas	PE,LF	2,9

\tilde{X}^1A'		C_s		Structure: $AB^{4,5,8}$		
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	2	Bend	1117.2	gas	LF,AB	9,10
	3	CBr stretch	676	gas	PE,LF	1,2,9

$A_0=15.534$; $B_0=0.429$; $C_0=0.417$ $AB^{4-6,8,10}$

DCBr

\tilde{A}^1A''		C_s		$\tilde{A}-\tilde{X}$ 550–970 nm		
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	2	Bend	608	gas	LF	3,6
	3	CBr stretch	737T	gas	LF	3,6

$T_0=11966.28$ gas $LF^3AB^{5,6,8,10}DL^7$

\tilde{X}^1A'		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	2	Bend	832.75	gas	AB,DL,LF	6,7,10,11
	3	CBr stretch	663	gas	LF	11

$A_0=17.99(3)$ AB^6DL^7
 $B_{av}=0.383$ AB^8

\tilde{X}^1A'		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	2	Bend	832.75	gas	AB,DL,LF	6,7,10,11
	3	CBr stretch	663	gas	LF	11

$A_0=8.618$; $B_0=0.394$; $C_0=0.376$ $AB^{5,6,8,10}$

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HSiF

\tilde{A}^1A'' C_s Structure: LF^{3-6}
 $T_0=23260.02$ gas LF^{2-6} $\tilde{A}-\tilde{X}$ 390–470 nm

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	SiH stretch	1547	gas	LF	5,6
	2	Bend	558	gas	LF	2,5,6
	3	SiF stretch	857	gas	LF	5,6

$\tau_0=175(6)$ ns gas $LF^{2,3}$
 $A_0=9.319$; $B_0=0.549$; $C_0=0.516$ LF^{3-5}
 Barrier to linearity= $9130(20)$ LF^5

\tilde{X}^1A' C_s Structure: $LF^{3,4,6,7}$

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	SiH stretch	1930(3)	gas	LF	7
			1913s	Ar	IR	1
	2	Bend	860.8(6)	gas	LF	2,7
3	SiF stretch	859m	Ar	IR	1	
		838(2)	gas	LF	7	
	834s	Ar	IR	1		

$A_0=7.58$; $B_0=0.564$; $C_0=0.524$ $LF^{3,4}$

DSiF

\tilde{A}^1A'' C_s Structure: LF^6
 $T_0=23338.723(6)$ gas LF^6 $\tilde{A}-\tilde{X}$ 383–453 nm

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	SiD stretch	1174.3	gas	LF	6
	2	Bend	424.8	gas	LF	6
	3	SiF stretch	854.4	gas	LF	6

$A_0=5.086$; $B_0=0.518$; $C_0=0.467$ LF^6

\tilde{X}^1A' C_s

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	SiD stretch	1402(4)	gas	LF	7
			1387m	Ar	IR	1
	2	Bend	640.7(5)	gas	LF	6,7
3	SiF stretch	638w	Ar	IR	1	
		839.6(6)	gas	IR	7	
	833m	Ar	IR	1		

$A_0=3.997$; $B_0=0.549$; $C_0=0.481$ LF^6

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HSiCl

\tilde{A}^1A'' C_s Structure: UV^1LF^2
 $T_0=20717.77$ gas $UV^1LF^{2,3}$ $\tilde{A}-\tilde{X}$ 410–600 nm

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	SiH stretch	1747.1	gas	LF	2
	2	Bend	563.9	gas	UV,LF	1,2
	3	SiCl stretch	532.3	gas	UV,LF	1,2

$\tau_0=432(20)$ ns gas LF^2
 $A_0=9.840(2)$; $B_0=0.247$; $C_0=0.240$ UV^1LF^2

\tilde{X}^1A' C_s Structure: $UV^1LF^{2,4}$

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	SiH stretch	1968.6(8)	gas	LF	2,4
	2	Bend	806.6(2)	gas	UV,LF	1,2,4
	3	SiCl stretch	522.6(4)	gas	UV,LF	1,2,4

$A_0=7.587$; $B_0=0.246$; $C_0=0.238$ UV^1LF^2

DSiCl

\tilde{A}^1A'' C_s Structure: UV^1LF^2
 $T_0=20773.43$ gas UV^1LF^2 $\tilde{A}-\tilde{X}$ 410–600 nm

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	SiD stretch	1300.8	gas	LF	2
	2	Bend	408.6	gas	UV,LF	1,2
	3	SiCl stretch	543.2	gas	LF	2

$\tau_0=437(20)$ ns gas LF^2
 $A_0=5.269$; $B_0=0.235$; $C_0=0.224$ UV^1LF^2

\tilde{X}^1A' C_s

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	SiD stretch	1434	gas	LF	2,4
	2	Bend	593.4	gas	LF	2,4
	3	SiCl stretch	518.2(5)	gas	LF	2,4

$A_0=3.970$; $B_0=0.240$; $C_0=0.226$ UV^1LF^2

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HSiBr

\tilde{A}^1A'' C_s Structure: UV^1LF^2
 $T_0 = 19902.851(7)$ gas UV^1LF^2 $\tilde{A}-\tilde{X}$ 429–620 nm

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	SiH stretch	1787.0	gas	LF	2
	2	Bend	535.3	gas	UV,LF	1,2
	3	SiBr stretch	416.5	gas	UV,LF	1,2

$A_0 = 9.897(2)$; $B_0 = 0.159$; $C_0 = 0.155$ UV^1LF^2
 $\tau_0 = 598(18)$ ns gas LF^2

\tilde{X}^1A' C_s Structure: $UV^1LF^{2,3}$

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	SiH stretch	1976	gas	LF	2,3
	2	Bend	772.4(3)	gas	UV,LF	1–3
	3	SiBr stretch	412.7(5)	gas	UV,LF	1–3

$A_0 = 7.576(8)$; $B_0 = 0.158$; $C_0 = 0.155$ UV^1LF^2

DSiBr

\tilde{A}^1A'' C_s
 $T_0 = 19953.677(5)$ gas LF^2 $\tilde{A}-\tilde{X}$ 464–502 nm

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	SiD stretch	1325.6	gas	LF	2
	2	Bend	375.9	gas	LF	2
	3	SiBr stretch	434.3	gas	LF	2

$\tau_0 = 586(21)$ ns gas LF^2
 $A_0 = 5.274(3)$; $B_0 = 0.151$; $C_0 = 0.146$ LF^2

\tilde{X}^1A' C_s

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	SiD stretch	1440	gas	LF	2,3
	2	Bend	562.8(2)	gas	LF	3
	3	SiBr stretch	407.8(7)	gas	LF	2,3

$A_0 = 3.956(3)$; $B_0 = 0.153$; $C_0 = 0.148$ LF^2

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- D. A. Hostutler, N. Ndiege, D. J. Clouthier, and S. W. Pauls, J. Chem. Phys. **115**, 5485 (2001).

HSiI

\tilde{A}^1A'' C_s Structure: AB^1
 $T_0 = 18259.02$ gas AB^1LF^2 $\tilde{A}-\tilde{X}$ 460–560 nm

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	SiH stretch	1852.5	gas	LF	2
	2	Bend	485.0	gas	AB,LF	1,2
	3	SiI stretch	335.7	gas	LF	2

$\tau_0 = 1.23(3)$ μs gas LF^2
 $A_0 = 9.807$; $B_0 = 0.118$; $C_0 = 0.116$ AB^1LF^2

\tilde{X}^1A' C_s Structure: AB^1LF^2

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	2	Bend	727	gas	AB	1
	2	SiI stretch	350.0	gas	LF	2

$A_0 = 7.557$; $B_0 = 0.118$; $C_0 = 0.116$ AB^1

DSiI

\tilde{A}^1A'' C_s
 $T_0 = 18302.96$ gas AB^1LF^2 $\tilde{A}-\tilde{X}$ 460–560 nm

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	SiD stretch	1356T	gas	LF	2
	2	Bend	368	gas	AB,LF	1,2
	3	SiI stretch	325	gas	LF	2

$\tau_0 = 1.22(9)$ μs gas LF^2
 $A_0 = 5.198$; $B_0 = 0.112$; $C_0 = 0.110$ LF^2

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HGeCl

\tilde{A}^1A'' C_s Structure: $LF^{3,4}$
 $T_0 = 21514.68$ gas $CL^2LF^{3,4}$ $\tilde{A}-\tilde{X}$ 439–520 nm

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	GeH stretch	1262.9(2)	gas	LF	4
	2	Bend	431.2(2)	gas	CL,LF	2–4
	3	GeCl stretch	398.8(2)	gas	LF	4

$\tau_0 = 548(19)$ ns gas LF^4
 $A_0 = 7.961(3)$; $B_0 = 0.151$; $C_0 = 0.147$ LF^4

\tilde{X}^1A'		C_s	Structure: LF ^{3,4}			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	GeH stretch	1862w	Ar	IR	1
	2	Bend	706	gas	CL	2
	3	GeCl stretch	439.2	gas	CL	2

$A_0=6.748(15)$; $B_0=0.149$; $C_0=0.146$ LF⁴

DGeCl

\tilde{A}^1A''		C_s	$\tilde{A}-\tilde{X}$ 441–462 nm			
$T_0=21614.48$		gas LF ⁴				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	GeD stretch	979.6(2)	gas	LF	4
	2	Bend	321.7(2)	gas	LF	4
	3	GeCl stretch	398.9(2)	gas	LF	4

$\tau_0=527(26)$ ns gas LF⁴

$A_0=4.145(3)$; $B_0=0.149$; $C_0=0.142$ LF⁴

\tilde{X}^1A'		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	GeD stretch	1343vw	Ar	IR	1

$A_0=3.435(14)$; $B_0=0.148$; $C_0=0.143$ LF⁴

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HGeBr

\tilde{A}^1A''		C_s	Structure: LF ^{2,3}			
$T_0=20660.30$		gas LF ^{2,3}	$\tilde{A}-\tilde{X}$ 450–500 nm			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	GeH stretch	1380.8(2)	gas	LF	3
	2	Bend	419.3(2)	gas	LF	2,3
	3	GeBr stretch	280.3(2)	gas	LF	3

$\tau_0=736(24)$ ns gas LF³

$A_0=8.184(4)$; $B_0=0.082$; $C_0=0.081$ LF³

\tilde{X}^1A'		C_s	Structure: LF ^{2,3}			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	GeH stretch	1858vs	Ar	IR	1
	2	Bend	695T	gas	LF	2
			701m	Ar	IR	1
3	GeBr stretch	283s	Ar	IR	1	

$A_0=6.736(14)$; $B_0=0.081$; $C_0=0.080$ LF³

DGeBr

\tilde{A}^1A''		C_s	$\tilde{A}-\tilde{X}$ 462–482 nm			
$T_0=20746.43$		gas LF ³				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	GeD stretch	1047.6(2)	gas	LF	3
	2	Bend	312.1(2)	gas	LF	3
	3	GeBr stretch	278.5(2)	gas	LF	3

$\tau_0=733(37)$ ns gas LF³

$A_0=4.210(4)$; $B_0=0.080$; $C_0=0.079$ LF³

\tilde{X}^1A'		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	GeD stretch	1336vs	Ar	IR	1
	2	Bend	502m	Ar	IR	1
	3	GeBr stretch	281ms	Ar	IR	1

$A_0=3.305(13)$; $B_0=0.080$; $C_0=0.078$ LF³

References

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HGeI

\tilde{A}^1A''		C_s	Structure: LF ¹			
$T_0=18929.29$		gas LF ¹	$\tilde{A}-\tilde{X}$ 482–529 nm			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	GeH stretch	1542.5(2)	gas	LF	1
	2	Bend	337.1(2)	gas	LF	1
	3	GeI stretch	203.4(2)	gas	LF	1

$\tau_0=1.515(4)$ μ s gas LF¹

$A_0=8.052$; $B_0=0.055$; $C_0=0.055$ LF¹

\tilde{X}^1A'		C_s	Structure: LF ¹			
$A_0=6.714$		$B_0=0.056$; $C_0=0.056$	LF ¹			

DGeI

\tilde{A}^1A''		C_s	$\tilde{A}-\tilde{X}$ 496–527 nm			
$T_0=19001.76$		gas LF ¹				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	GeD stretch	1144.2(2)	gas	LF	1
	2	Bend	259.7(2)	gas	LF	1
	3	GeI stretch	205.0(2)	gas	LF	1

$\tau_0=1.52(3)$ μ s gas LF¹

$A_0=4.158$; $B_0=0.055$; $C_0=0.054$ LF¹

\tilde{X}^1A'		C_s	LF ¹			
$A_0=3.424$		$B_0=0.056$; $C_0=0.055$	LF ¹			

Reference

¹W. W. Harper, C. M. Klusek, and D. J. Clouthier, *J. Chem. Phys.* **109**, 9300 (1998).

HON

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	OH stretch	3467.2wm	Ar	IR	1
	2	Bend	1095.6vs	Ar	IR	1

DON

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	OD stretch	2563.6wm	Ar	IR	1
	2	Bend	868.8s	Ar	IR	1
	3	ON stretch	1149.0vs	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, and M. De Marco, *Angew. Chem.* **111**, 113 (1999); *Angew. Chem. Int. Ed.* **38**, 108 (1999).

BiOH

$\tilde{A}_1 A'^a$ C_s						
$T_0 = 22782$ gas LF^2						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	3	BiO stretch	272	gas	LF	2
$\tau_0 = 0.92(3) \mu s$ gas LF^2						
$A_0 - 1/2(B_0 + C_0) = 23.99(5)$ LF^2						

$\tilde{a}_1 A'^a$ C_s						
$T_0 = 11326(50)$ gas LF^2						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	3	BiO stretch	520(70)	gas	LF	2

$\tilde{X}_3 A'^a$ C_s						
$T_0 = 6183.8(3)$ gas $CL^{1,2}$						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	3	BiO stretch	522	gas	LF	2
$A_0 - 1/2(B_0 + C_0) = 23.09$ LF^2						

$\tilde{X}_2 A'^a$ C_s
 $T_0 = 6171.4$ gas CL^1 $\tilde{X}_2 A'' - \tilde{X}_1 A'$ 1580–1677 nm
 $A_0 - 1/2(B_0 + C_0) = 22.6$ CL^1

$\tilde{X}_1 A'^a$ C_s						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	OH stretch	3669(50)	gas	LF	2
	3	BiO stretch	504.7(4)	gas	CL,LF	1,2
$A_0 - 1/2(B_0 + C_0) = 23.47(6)$ $CL^{1,2}$						

BiOD

$\tilde{A}_1 A'^a$ C_s						
$T_0 = 22815$ gas LF^2						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	3	BiO stretch	266	gas	LF	2
$A_0 - 1/2(B_0 + C_0) = 12.68(2)$ LF^2						
$\tilde{a}_1 A'^a$ C_s						
$T_0 = 11294(50)$ gas LF^2						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	3	BiO stretch	520(70)	gas	LF	2

$\tilde{X}_3 A'^a$ C_s
 $T_0 = 6179.3(2)$ gas $CL^{1,2}$
 $A_0 - (B_0 + C_0) = 12.37(2)$ LF^2

$\tilde{X}_2 A'^a$ C_s
 $T_0 = 6168.1$ gas CL^1 $\tilde{X}_2 A'' - \tilde{X}_1 A'$ 1582–1665 nm
 $A_0 - 1/2(B_0 + C_0) = 12.0$ CL^1

$\tilde{X}_1 A'^a$ C_s						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	OD stretch	2662(50)	gas	LF	2
	3	BiO stretch	492.6(5)	gas	CL,LF	1,2
$A_0 - 1/2(B_0 + C_0) = 12.49(2)$ $CL^1 LF^2$						

^aFine structure component of electronic state.

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HPCI

\tilde{A}^2A' C_s
 $T_0=21305(6)$ gas EM¹ $\tilde{A}-\tilde{X}$ 420–590 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	Bend	622(4)	gas	EM	1
	3	PCI stretch	529(8)	gas	EM	1

\tilde{X}^2A'' C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	Bend	868(2)	gas	EM	1

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HO₂

Broad, unstructured gas-phase absorption between 200 and 280 nm, with maximum near 205 nm.^{2,3,5,6,37}

\tilde{A}^2A' C_s
 $T_0=7029.688$ gas AB^{9,15}EM^{10,16,19,20,35,41} $\tilde{A}-\tilde{X}$ 1.13–2.12 μm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	OH stretch	3268.5	gas	EM	35
	2	Bend	1285	gas	EM	35
	3	OO stretch	929.068	gas	AB,EM	15,19,28,35

$A_0=20.486$; $B_0=1.021$; $C_0=0.968$ EM^{16,20,41}

\tilde{X}^2A'' C_s Structure: MW¹⁴UV²¹LMR²³ESR²³IR^{31,33}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	OH stretch	3436.20	gas	LD	26
			3415.1	Ne	IR	36
			3412.5 s ^a	Ar	IR	1,4,7
			3400	O ₂	IR	32
	2	Bend	1391.75	gas	DL,IR	24,39
			1397.8	Ne	IR	36
			1388.5 vs ^a	Ar	IR	1,4,7
	3	OO stretch	1392	O ₂	IR	32
			1097.63	gas	LMR	18,29
					DL,IR	29,38,39
1100.3			Ne	IR	36	
1101.1 s ^a			Ar	IR	1,4,7	
1109			O ₂	IR	32	

$A_0=20.356$; $B_0=1.118$; $C_0=1.056$ LMR^{8,11,12,18}MW^{7,13,17,25}EM¹⁶IR⁴⁰

DO₂

\tilde{A}^2A' C_s
 $T_0=7041.1(1)$ gas AB⁹EM^{10,19,21} $\tilde{A}-\tilde{X}$ 1.13–2.12 μm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	3	OO stretch	940(28)	gas	AB,EM	8,19

$A_0=11.147(7)$; $B_0=0.970$; $C_0=0.887$ EM²¹

\tilde{X}^2A'' C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	OD stretch	2549.22	gas	LD,DL	31
			2529.2	Ne	IR	36
			2529.5 m ^a	Ar	IR	1,4,7
			2521	O ₂	IR	32
			1020.16	gas	LMR,DL	22,33
	2	Bend	1027.3	Ne	IR	36
			1019.9 s ^a	Ar	IR	1,4,7
			1024	O ₂	IR	32
	3	OO stretch	1121.47	gas	LMR,DL	22,33
			1124.7	Ne	IR	36
1122.9 vw ^a			Ar	IR	7	

$A_0=11.194$; $B_0=1.056$;

$C_0=0.961$ MW^{14,27,34,42}EM^{21,42}LMR^{22,23,34}ESR²³

^aRefined value from unpublished Fourier transform spectra.

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HS₂

gas AB¹⁻³ 297–380 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	SH stretch	2500T	gas	AB	3
	2	Bend	900T	gas	AB	3
	3	SS stretch	600T	gas	AB	3

\tilde{A}^2A' C_s
 $T_0=7255(7)$ gas CL⁴ $\tilde{A}-\tilde{X}$ 950–2100 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	3	SS stretch	504(4)	gas	CL	4

$A_0=9.7(5)$ CL⁴

\tilde{X}^2A'' C_s Structure: MW^{5,8}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	SH stretch	2463T 2460T	Ar	IR	7
	2	Bend	904(8)	gas	CL	4
	3	SS stretch	903T 595(4)	Ar gas	IR CL	7 4

$A_0=9.906$; $B_0=0.267$; $C_0=0.259$ MW^{5,8}LMR⁶

DS₂

\tilde{A}^2A' C_s
 $T_0=7264(15)$ gas CL⁴ $\tilde{A}-\tilde{X}$ 950–2100 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	3	SS stretch	502(15)	gas	CL	4

\tilde{X}^2A'' C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	2	Bend	696(20)	gas	CL	4
	3	SS stretch	591(10)	gas	CL	4

$A_0=5.184$; $B_0=0.260$; $C_0=0.247$ MW^{5,8}

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HO₂⁻

Threshold for electron detachment from ground-state HO₂⁻ is 8790(50).^{1,2}

\tilde{X}^1A' C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	3	OO stretch	775(250)	gas	PE	1

DO₂⁻

Threshold for electron detachment from ground-state DO₂⁻ is 8790(140).¹

\tilde{X}^1A' C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	3	OO stretch	900(250)	gas	PE	1

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HOCl

\tilde{X}		C_s Structure: IR ^{2,14} MW ^{4,12}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	OH stretch	3609.48 3581	gas Ar	IR IR	2,6,7,11,15 3
	2	Bend	1238.62 1239	gas Ar	IR IR	1,5,6,8,9,11,13 3
	3	OCl stretch	724.36 728	gas Ar	IR IR	5,6,8,11,13 3

$A_0=20.464$; $B_0=0.504$; $C_0=0.491$ IR^{2,11,15,16}MW^{4,10}

DOCl

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	OD stretch	2665.58 2647	gas Ar	IR IR	1,2,11,14,18 3
	2	Bend	909.63 911	gas Ar	IR,DL IR	1,11,14,17 3
	3	OCl stretch	723.25 728	gas Ar	IR IR	11,14 3

$A_0=11.052$; $B_0=0.477$; $C_0=0.456$ IR^{2,14}MW^{4,12}

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HOBr

In the gas phase, a prominent absorption maximum near 280 nm (35700) has been assigned^{6,8,10,11} to HOBr.

In the gas phase, an absorption maximum near 350 nm (28600) has been assigned^{6,8,10,11} to HOBr.

In the gas phase, a relatively weak absorption maximum and a maximum in the production of OH near 457 nm (21900) is believed to arise from excitation of HOBr to a dissociative triplet state.⁹⁻¹¹

\tilde{X}		C_s Structure: MW ³ IR ⁷				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	OH stretch	3614.90 3590	gas Ar	IR IR	2,7 1
	2	Bend	1162.57 1164	gas Ar	IR IR	4 1
	3	OBr stretch	620.2 626.0	gas Ar	IR IR	2,4,5 1

$A_0=20.470$; $B_0=0.353$; $C_0=0.346$ MW³IR⁷

DOBr

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	OD stretch	2668.79 2652	gas Ar	IR IR	12 1
	2	Bend	854	Ar	IR	1
	3	OBr stretch	621.8	Ar	IR	1

$A_0=11.027$; $B_0=0.331$; $C_0=0.321$ MW³IR¹²

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HOI

In the gas phase, broad absorption maxima at 29380 (340 nm) and 24610 (406 nm) have been assigned⁴ to HOI.

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	OH stretch	3625.84	gas	IR	2,3
			3597m	N ₂	IR	1
	2	Bend	1069.8	gas	IR	2
1075			Ar	IR	1	
1103m			N ₂	IR	1	
3	OI stretch	577	Ar	IR	1	
		575m	N ₂	IR	1	

$A_0=20.935$; $B_0=0.279$; $C_0=0.275$ EM³

DOI

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	OD stretch	2653	N ₂	IR	1
	2	Bend	808	N ₂	IR	1
	3	OI stretch	571	N ₂	IR	1

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NeHF⁺ \tilde{X} $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	NeH stretch	292.4T	Ne	IR	1
Σ^+	3	HF stretch	2218.4	Ne	IR	1

NeDF⁺ \tilde{X} $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	3	DF stretch	1663.0	Ne	IR	1

Reference

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ClHCl⁻

An absorption maximum which appeared at 287 nm in argon-matrix studies² of the 122-nm photolysis of Ar:HCl or Ar:H₂O:Cl₂ samples and in argon-matrix studies⁶ of the electron bombardment of Ar:HCl samples has been assigned to an electronic transition of ClHCl⁻.

 \tilde{X} $D_{\infty h}$ Structure: DL⁵

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	263.4 ^b	Ne	IR	8
			259.3 ^{ab}	Ar	IR	1-3,7,9,10
			252.8 ^b	Kr	IR	7,9
			248.8 ^b	Xe	IR	7
Σ_u^+	3	Asym. stretch	722.90	gas	DL	5
			737.9	Ne	IR	8
			728.9			
			695.6s ^a	Ar	IR	1-3,7,9,10
			662.8	Kr	IR	7,9,10
			644.1	Xe	IR	7

$B_0=0.0974$ DL⁵

ClDCI⁻ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	267 ^{ab}	Ar	IR	1-3,9
			255.3 ^b	Kr	IR	7
Σ_u^+	3	Asym. stretch	496.2	Ne	IR	8
			489.3			
			463 ^a	Ar	IR	1-3,10
			437.7	Kr	IR	7,10

^aAttributed by Ref. 1 to the uncharged species. Reassigned to the anion by Ref. 2, and Ref. 3 demonstrated that the absorptions did not appear when the atoms were present but a supplementary high energy source suitable for inducing photoionization or electron transfer was not. The increase in the rate of isotopic exchange in the ³⁷Cl+H³⁵Cl reaction for vibrationally excited HCl⁴ indicates that there is a potential barrier, rather than a minimum, for the ClHCl neutral species.

^b($\nu_1 + \nu_3$) - ν_3 .

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HArF

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	HAr stretch	2020.8 ^a 2016.3 ^a 1969.5	Ar	IR	1,2
Π	2	Bend	697.0 ^a 693.5 ^a 687.0	Ar	IR	1,2
Σ^+	3	ArF stretch	435.7	Ar	IR	1

DArF

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	DAr stretch	1497.1 ^a 1493.8 ^a 1466.3	Ar	IR	1,2
Π	2	Bend	513.0	Ar	IR	1
Σ^+	3	ArF stretch	435.3	Ar	IR	1

^aRelatively great thermal stability.²

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¹L. Khriachtchev, M. Pettersson, N. Runeberg, J. Lundell, and M. Räsänen, *Nature (London)* **406**, 874 (2000).

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HKrCl

In a krypton matrix, a maximum in the photodecomposition cross section of HKrCl has been observed² near 35700 (280 nm).

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	HKr stretch	1476vs	Kr	IR	1
Π	2	Bend	544w	Kr	IR	1

DKrCl

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	DKr stretch	1106s	Kr	IR	1

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HXeCl

In a xenon matrix, an unstructured absorption maximum at 40650 (246 nm) has been assigned³ to HXeCl.

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	HXe stretch	1611.8 1664 1649	Ne Kr Xe	IR	2 1 1,3

DXeCl

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	DXe stretch	1172.1 1198	Ne Xe	IR	2 1

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¹M. Pettersson, J. Lundell, and M. Räsänen, *J. Chem. Phys.* **102**, 6423 (1995).

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³J. Ahokas, K. Vaskonen, J. Eloranta, and H. Kunttu, *J. Phys. Chem. A* **104**, 9506 (2000).

HXeBr

In a xenon matrix, an unstructured absorption maximum at 38170 (262 nm) has been assigned⁴ to HXeBr.

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	HXe stretch	1452.5 1524 1504vs	Ne Kr Xe	IR	2,3 1 1,4
Π	2	Bend	489w	Xe	IR	1

DXeBr

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	DXe stretch	1064.1 1100s	Ne Xe	IR	3 1

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HXeI

In a xenon matrix, an unstructure absorption maximum at 31750 (315 nm) has been assigned⁴ to HXeI.

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	HXe stretch	1193vs	Xe	IR	1–4
Π	2	Bend	450w	Xe	IR	1,3
Σ^+	3	XeI stretch	146T	Xe	IR	3

DXeI

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	DXe stretch	893vs	Xe	IR	1–3

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¹M. Pettersson, J. Lundell, and M. Räsänen, *J. Chem. Phys.* **102**, 6423 (1995).

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³J. Lundell, M. Pettersson, L. Khriachtchev, M. Räsänen, G. M. Chaban, and R. B. Gerber, *Chem. Phys. Lett.* **322**, 389 (2000).

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HAr₂⁺

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	237 ^a	Ar	IR	5
Σ_u^+	3	Asym. stretch	903.4wm ^b	Ar	IR	1,2,4–6

DAr₂⁺

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_u^+	3	Asym. stretch	644s ^b	Ar	IR	1–5,7

^a $(\nu_1 + \nu_3) - \nu_3$.

^bAssigned in Ref. 1 to vibration of uncharged H (or D) atoms trapped in the Ar lattice. Reassigned by Ref. 2 to the cation. Peak at 644 cm⁻¹ was prominent in deuteron radiolysis experiments.³ Reference 4 demonstrated that the absorptions did not appear when H or D atoms were present but a supplementary high energy source suitable for inducing photoionization or electron transfer was not.

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⁷T. D. Fridgen, X. K. Zhang, J. M. Parnis, and R. E. March, *J. Phys. Chem. A* **104**, 3487 (2000).

HKr₂⁺

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	155 ^a	Kr	IR	4,7,8
Σ_u^+	3	Asym. stretch	885.3T ^b 853.2m ^c	Ar Kr	IR	5,6 1–4,7,8

DKr₂⁺

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	165 ^a	Kr	IR	4,8
Σ_u^+	3	Asym. stretch	625T 606m ^c	Ar Kr	IR	5 1,4,8

^a $(\nu_1 + \nu_3) - \nu_3$.

^bReassignment proposed by Ref. 6.

^cAssigned in Ref. 1 to vibration of uncharged H (or D) atoms trapped in the Kr lattice. Reassigned by Ref. 2 to the cation. Reference 3 demonstrated that the 852 cm⁻¹ absorption did not appear when H atoms were present but a supplementary high energy source suitable for inducing photoionization or electron transfer was not.

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HXe₂⁺

\tilde{X}		D _{∞h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	1	Sym. stretch	124.4T ^{ab}	Ar	IR	2,3
			118.6T ^{ab}	Kr	IR	2,3
			111.9 ^b	Xe	IR	1,4
Σ _u ⁺	3	Asym. stretch	828.1T ^a	Ar	IR	2,3
			781.7T ^a	Kr	IR	2,3
			729.9	Xe	IR	1,4

DXe₂⁺

\tilde{X}		D _{∞h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	1	Sym. stretch	118.0 ^b	Xe	IR	1,4
Σ _u ⁺	3	Asym. stretch	538.1T	Kr	IR	2
			516.7	Xe	IR	1,4

^aReassignment proposed by Ref. 3.

^b(ν₁ + ν₃) - ν₃.

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Kr₂H^a

\tilde{A}
 T₀ = 64500 Kr AB^{4,6}
 In a Kr matrix, unstructured emission with maximum at 49600 (202 nm)¹⁻⁴
 τ = 48 ns Kr EM²

Kr₂D

\tilde{A}
 T₀ = 64800 Kr AB^{4,6}
 In an Ar matrix, unstructured emission with maximum at 50750 (197 nm)⁵
 In a Kr matrix, unstructured emission with maximum at 49700 (201 nm)^{2,4}
 τ = 58 ns Ar EM⁵
 46 ns Kr EM²

^aExcimer. Extensive delocalization of positive charge in Kr and Xe matrices.

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8.3. Triatomic Nonhydrides**Li₃**

\tilde{C}^2E'' D_{3h}^a
 T₀ = 21541 gas MPI³ $\tilde{C}-\tilde{X}$ 450–472 nm
 B = 0.57 MPI⁴

\tilde{A}^2E'' D_{3h}^a
 T₀ = 14575.5 gas MPI^{2,3,5,6}DR^{7,8} $\tilde{A}-\tilde{X}$ 660–706 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	Sym. stretch	261.4	gas	MPI	6
	2		112.6	gas	MPI	6

B = 0.57 MPI⁴. A_{ps} = 0.738; B_{ps} = 0.398; C_{ps} = 0.255 DR⁷

\tilde{X}^2E' D_{3h}^a Structure: MPI⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁ '	1	Sym. stretch	302	gas	MPI	3
			303	Xe	Ra	1

Barrier to pseudorotation = 26; pseudorotation frequency = 34 MPI³
 B = 0.584 MPI⁴. A_{ps} = 0.705; B_{ps} = 0.468; C_{ps} = 0.278 DR⁷

^aSubject to dynamic Jahn–Teller distortion.

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Na₃

$\tilde{D}^2E''(^2A_2)$ D_{3h}(C_{2v})^a
 gas MPI^{4,5,14,19}DPI¹²PE¹⁴ $\tilde{D}-\tilde{X}$ 410–440 nm
 Fragments into Na₂ + Na in less than 1 ps.¹⁹

$\tilde{C}^2E''(^2A_2)$ D_{3h}(C_{2v})^a
 T₀ = 20813 gas MPI^{2,4,6}DPI^{8,12} $\tilde{C}-\tilde{X}$ 467–481 nm
 Higher vibrational bands are predissociated.⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁ '	1	Sym. stretch	135	gas	MPIPF	8

Extensive vibronic structure has been tentatively assigned⁸ to energy levels derived from excitation of $\nu_2(e')$, perturbed by dynamic Jahn–Teller interaction.

$\tau_0 = 7(3)$ ns gas MPI⁷

$\tau_{\text{frag}} = 1.12$ ns for the lowest vibrational level. MPI²⁵

\tilde{B}''^2B_2

$T^b = 19200$ gas DPI¹²

\tilde{B}^2A_1' D_{3h}^a Structure: MPI²⁴
 $T_0 = 16124.46$ gas MPI^{1-6,17,20-24}DPI¹²TPE¹⁸DR²⁸ $\tilde{B}-\tilde{X}$ 550–625 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			128	gas	MPI,TPE	1,3,18
			105	gas	MPI,TPE	18
			74	gas	MPI,TPE	18

$\tau_{\tilde{B}''}(16255) = 14(5)$ ns gas MPI⁷

$\tau_{\tilde{B}'}(17418) = 7(3)$ ns gas MPI⁷

$A_0 = 0.082(2)$; $B_0 = 0.138(4)$; $C_0 = 0.051$ MPI^{22,24}

Vibronic pseudorotation occurs,³ with a beat structure period of approximately 3 ps.^{17,20,21}

$2^4E'$ D_{3h}
 $T_0 = 15791.39 + x$ He LF^{27,30} $2^4E' - 1^4A_2'$ 616–649 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e'	2	Asym. stretch	88.2	He	LF	27

$\tilde{A}^2E''(^2A_2)$ D_{3h}(C_{2v})^a Structure: MPI^{15,16}
 $T_0 = 14894.769(4)$ gas MPI^{1,2,4-6,11,13,15,16,26}DPI¹²DR²⁸ $\tilde{A}-\tilde{X}$ 658–675 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	Sym. stretch	150	gas	MPI	11
	2	Bend	127	gas	MPI	4,6,11,26
	3	Asym. stretch	47	gas	MPI	4,6,11

Vibronic structure assigned in Ref. 11. First three bands are partially localized; pseudorotation barrier=196.

$\tau_0 = 60(10)$ ns gas MPI⁷

$A_0 = 0.166$; $B_0 = 0.085$; $C_0 = 0.056$ MPI^{15,16}

\tilde{A}'^2A_1' D_{3h}
 $T_0 = 13200$ U gas DPI^{12,29}

$1^4A_2'$ D_{3h}
 $T_0 = x$ He LF^{27,30} $2^4E' - 1^4A_2'$ 616–649 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1'	1	Sym. stretch	40T	He	LF	30
e'	2	Asym. stretch	40T	He	LF	30

$\tilde{X}^2E'(2B_2)$ D_{3h}(C_{2v})^a Structure: MPI^{15,16}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	Sym. stretch	139	gas	MPI, SEP	4,6,9,10
	2	Bend	49.5	gas	MPI, SEP	4,6,9,10
	3	Asym. stretch	87	gas	MPI, SEP	6,9,10

$A_0 = 0.179$; $B_0 = 0.085$; $C_0 = 0.057$ MPI^{15,16,22}DR²⁸

^aDistorted by Jahn–Teller interaction.

^bBand maximum.

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K₃ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Sym. stretch	81.5	Kr	Ra	1
		Deformation	61.0	Kr	Ra	1

Reference

¹A. Kornath, R. Ludwig, and A. Zoerner, *Angew. Chem.* **110**, 1620 (1998); *Angew. Chem. Int. Ed.* **37**, 1575 (1998).

Rb₃ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Sym. stretch	53.9	Ar	Ra	1
		Deformation	38.3	Ar	Ra	1

Reference

¹A. Kornath, A. Zoerner, and R. Ludwig, *Inorg. Chem.* **38**, 4696 (1999).

Cs₃ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Sym. stretch	39.5	Ar	Ra	1
		Deformation	24.4	Ar	Ra	1

Reference

¹A. Kornath, A. Zoerner, and R. Ludwig, *Inorg. Chem.* **38**, 4696 (1999).

Hf₃

In an argon matrix, overlapping absorption maxima are observed¹ at 16500, 16390, 16260, and 16150 (606, 610, 615, and 619 nm, respectively).

\tilde{E} D_{3h}
 $T_0 = 785.4(2)$ Ar Ra¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'_1	1	Sym. stretch	150.3	Ar	Ra	1
e'	2	Deformation	108.7	Ar	Ra	1

\tilde{D} D_{3h}
 $T_0 = 642.8(5)$ Ar Ra¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'_1	1	Sym. stretch	149.5(9)	Ar	Ra	1
e'	2	Deformation	108.4(9)	Ar	Ra	1

\tilde{C}
 $T_0 = 609.6(5)$ Ar Ra¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			116.7(1.3)	Ar	Ra	1

\tilde{B}
 $T_0 = 413.4(6)$ Ar Ra¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	Sym. stretch	141.7(1.1)	Ar	Ra	1

\tilde{A} D_{3h}
 $T_0 = 319.0(3)$ Ar Ra¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'_1	1	Sym. stretch	152.3(4)	Ar	Ra	1
e'	2	Deformation	102.7(1.9)	Ar	Ra	1

$\tilde{X}E'$ D_{3h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'_1	1	Sym. stretch	142.5	Ar	Ra	1

Reference

¹H. Wang, Z. Hu, H. Haouari, R. Craig, Y. Liu, J. R. Lombardi, and D. M. Lindsay, *J. Chem. Phys.* **106**, 8339 (1997).

Ta₃ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	Sym. stretch	251.2	Ar	Ra	1

Reference

¹L. Fang, X. Shen, X. Chen, and J. R. Lombardi, *Chem. Phys. Lett.* **332**, 299 (2000).

Cr₃

In an argon matrix, an absorption maximum at 20960 (477 nm) behaves appropriately for assignment to Cr₃.^{1,2} The corresponding maximum appears in a krypton matrix at 20880 (479 nm).¹

\tilde{X} D_{3h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'_1	1	Sym. stretch	400	Ar	Ra	2
e'	2	Deformation	302	Ar	Ra	2

References

- ¹W. E. Klotzbücher and G. A. Ozin, *J. Am. Chem. Soc.* **100**, 2262 (1978).
²L. Fang, B. Davis, H. Lu, and J. R. Lombardi, *J. Phys. Chem. A* **105**, 9375 (2001).

Fe₃ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Sym. stretch	249T	Ar	Ra	1

Reference

- ¹T. L. Haslett, K. A. Bosnick, S. Fedrigo, and M. Moskovits, *J. Chem. Phys.* **111**, 6456 (1999).

Ru₃ $T_0 = 581.5$ Ar Ra¹ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	Sym. stretch	303.4	Ar	Ra	1

Reference

- ¹L. Fang, X. Shen, X. Chen, and J. R. Lombardi, *Chem. Phys. Lett.* **332**, 299 (2000).

Rh₃ $T_0 = 400$ Ar Ra¹ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		C_{2v}				
a_1	1	Sym. stretch	321.9(5)	Ar	Ra	1
	2	Bend	247.9(8)	Ar	Ra	1
b_2	3	Asym. stretch	259	Ar	Ra	1

Reference

- ¹L. Fang, X. Shen, X. Chen, and J. R. Lombardi, *J. Chem. Phys.* **113**, 7178 (2000).

Ag₃

In an argon matrix study with mass selection,^{8,12} an absorption maximum at 31150 (321 nm) behaves appropriately for assignment to Ag₃, as does an emission maximum at 26700 (374 nm). Irradiation of the deposit at 388 nm results in the replacement of this absorption maximum by peaks at 28570 (350 nm) and 27550 (363 nm) and by growth in the same fluorescence emission.¹² The new absorption maxima are attributed to Ag₃ trapped in more stable sites.

In a krypton matrix study,⁶ the corresponding absorption maximum lies at 30200 (331 nm), and there is an emission maximum at 26200 (381 nm). In mass-selection observations,⁶ there is also a less prominent absorption maximum near 27500 (364 nm).

²E'' D_{3h}^a
 $T_0 = 26969.0$ gas MPI^{4,5,9}LF⁷ ${}^2E'' - \tilde{X}$ 365–385 nm
 In an argon matrix study with mass selection,⁸ an absorption maximum at 25900 (386 nm) behaves appropriately for assignment to Ag₃. In a similar krypton matrix study,⁶ the absorption maximum appears at 24900 (402 nm).

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'_1	1	Sym. stretch	158.2	gas	MPI,LF	4,5,7,9
e'	2	Deformation	96	gas	MPI,LF	4,5,7,9
$\tau < 80$ ns gas LF ⁷						

In argon, krypton, and xenon matrices, an absorption maximum near 23700 (422 nm) has been attributed^{2,3,6} to Ag₃. In an argon-matrix study of mass-selected Ag₃,¹² the maximum at 23870 (419 nm) was replaced by one at 23470 (426 nm) on 388-nm irradiation of the deposit.

²A₁'? D_{3h}
 $T_0 = 19809(2)$ gas EM¹⁰ 495–505 nm
 In an argon matrix study with mass selection,⁸ an absorption maximum at 20300 (492 nm) and an emission maximum at 16100 (622 nm) behave appropriately for assignment to Ag₃. A subsequent study¹² of mass-selected Ag₃ isolated in an argon matrix suggested that an absorption maximum at 19880 (503 nm) and an emission maximum at 16450 (608 nm) are associated with a less stable site. That study also suggested the existence of a third site with absorption and emission maxima at 20490 (488 nm) and 16180 (618 nm), respectively. In a krypton matrix study with mass selection,⁶ absorption maxima at 21830 (458 nm) and 19500 (514 nm) and emission maxima at 17900 (560 nm) and 16000 (626 nm) also behave appropriately for assignment to Ag₃.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'_1	1	Sym. stretch	107 ^b	gas	EM	10
e'	2	Deformation	94 ^b	gas	EM	10

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
$\tilde{X} {}^2E'$ D_{3h}^a						
a'_1	1	Sym. stretch	121	gas	LF	7,9
			119	Ar	Ra	11
			120T	Kr	Ra	1
e'	2	Deformation	99	gas	LF	7,9

^aDistorted by Jahn–Teller interaction.

^b¹⁰⁷Ag₂¹⁰⁹Ag.

References

- ¹W. Schulze, H. U. Becker, R. Minkwitz, and K. Manzel, *Chem. Phys. Lett.* **55**, 59 (1978).
²W. Schulze, H. U. Becker, and H. Abe, *Chem. Phys.* **35**, 177 (1978).
³G. A. Ozin, H. Huber, and S. A. Mitchell, *Inorg. Chem.* **18**, 2932 (1979).
⁴P. Y. Cheng and M. A. Duncan, *Chem. Phys. Lett.* **152**, 341 (1988); *Chem. Phys. Lett.* **156**, 420 (1989).
⁵K. LaiHing, P. Y. Cheng, and M. A. Duncan, *Z. Phys. D* **13**, 161 (1989).
⁶W. Harbich, S. Fedrigo, F. Meyer, D. M. Lindsay, J. Lignieres, J. C. Rivoal, and D. Kreisler, *J. Chem. Phys.* **93**, 8535 (1990).
⁷A. M. Ellis, E. S. J. Robles, and T. A. Miller, *Chem. Phys. Lett.* **201**, 132 (1993).

- ⁸S. Fedrigo, W. Harbich, and J. Buttet, *J. Chem. Phys.* **99**, 5712 (1993).
⁹E. E. Wedum, E. R. Grant, P. Y. Cheng, K. F. Willey, and M. A. Duncan, *J. Chem. Phys.* **100**, 6312 (1994).
¹⁰T. Okazaki, Y. Saito, A. Kasuya, and Y. Nishina, *J. Chem. Phys.* **104**, 812 (1996).
¹¹T. L. Haslett, K. A. Bosnick, S. Fedrigo, and M. Moskovits, *J. Chem. Phys.* **111**, 6456 (1999).
¹²I. Rabin, W. Schulze, G. Ertl, C. Felix, C. Sieber, W. Harbich, and J. Buttet, *Chem. Phys. Lett.* **320**, 59 (2000).

Hg₃

In the gas phase, an excitation maximum at 44135 has been tentatively attributed² to Hg₃.

In an argon matrix, a weak, broad absorption and an excitation maximum at 39800 have been attributed¹ to Hg₃.

In the gas phase, fluorescence maxima at 24745, 22930, and 20000 have been tentatively attributed² to Hg₃.

In an argon matrix, a fluorescence maximum at 19800 has been attributed¹ to Hg₃.

References

- ¹C. Crépin and A. Tramer, *J. Chem. Phys.* **100**, 5475 (1994).
²J. Koperski, J. B. Atkinson, and L. Krause, *J. Mol. Spectrosc.* **187**, 181 (1998).

LiOLi

¹B₁ C_{2v}
 T₀ = 21331 gas MPI⁵LF⁶ ¹B₁- \tilde{X} 433–645 nm

$\tilde{X}^1\Sigma^+$		D _{∞h}	Structure: MPI ⁵			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	784(6) (ω)	gas	LF	6
Π_u	2	Bend	122(4) (ω)	gas	LF	6
			112	Kr	IR	2
Σ_u^+	3	Asym. stretch	964(7)H	gas	LF	6
			997.3	Ar	IR	4
			986.5	Kr	IR	1,2
			945.6	N ₂	IR	3

B₀ = 0.466(4) MPI⁵

References

- ¹D. White, K. S. Seshadri, D. F. Dever, D. E. Mann, and M. J. Linevsky, *J. Chem. Phys.* **39**, 2463 (1963).
²K. S. Seshadri, D. White, and D. E. Mann, *J. Chem. Phys.* **45**, 4697 (1966).
³R. C. Spiker, Jr., and L. Andrews, *J. Chem. Phys.* **58**, 702 (1973).
⁴L. Andrews, W. Saffell, and J. T. Yustein, *Chem. Phys.* **189**, 343 (1994).
⁵D. Bellert and W. H. Breckenridge, *J. Chem. Phys.* **114**, 2871 (2001).
⁶D. Bellert, D. K. Winn, and W. H. Breckenridge, *Chem. Phys. Lett.* **348**, 39 (2001).

cyc-Co₂N

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CoN s-stretch	746.8	Ar	IR	1
			745.4	N ₂	IR	1

Reference

- ¹L. Andrews, A. Citra, G. V. Chertihin, W. D. Bare, and M. Neurock, *J. Phys. Chem. A* **102**, 2561 (1998).

CoCoN

\tilde{X}		C ₅				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CoN stretch	1016.6T	N ₂	IR	1

Reference

- ¹L. Andrews, A. Citra, G. V. Chertihin, W. D. Bare, and M. Neurock, *J. Phys. Chem. A* **102**, 2561 (1998).

cyc-RhRhN

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	728.0	Ar	IR	1
			738.9	N ₂	IR	1

Reference

- ¹A. Citra and L. Andrews, *J. Phys. Chem. A* **103**, 3410 (1999).

RhRhN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	RhN stretch	975.6	Ar	IR	1
			971.5	N ₂	IR	1

Reference

- ¹A. Citra and L. Andrews, *J. Phys. Chem. A* **103**, 3410 (1999).

NiNiN

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	NiN stretch	1003.2 996.0	Ar N ₂	IR IR	1 1

Reference

¹L. Andrews, A. Citra, G. V. Chertihin, W. D. Bare, and M. Neurock, J. Phys. Chem. A **102**, 2561 (1998).

Pt₂N

\tilde{X} C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			732.1	N ₂	IR	1

Reference

¹A. Citra, X. Wang, W. D. Bare, and L. Andrews, J. Phys. Chem. A **105**, 7799 (2001).

PtPtN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1095.2	N ₂	IR	1

Reference

¹A. Citra, X. Wang, W. D. Bare, and L. Andrews, J. Phys. Chem. A **105**, 7799 (2001).

Pr₂N

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	PrN stretch	475.7	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **102**, 10238 (1998).

Nd₂N

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NdN stretch	477.4	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **102**, 10238 (1998).

Eu₂N

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	EuN stretch	451.3	N ₂	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **102**, 10238 (1998).

Gd₂N

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	GdN stretch	494.4	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **102**, 10238 (1998).

Tb₂N

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	TbN stretch	502.7 468.6	Ar N ₂	IR IR	1 1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 1311 (1999).

Dy₂N

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	DyN stretch	505.3	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 1311 (1999).

Ho₂N \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	HoN stretch	509.4	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 1311 (1999).

Lu₂N \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	LuN stretch	522.0	Ar	IR	1
			513.3	N ₂	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 1311 (1999).

ScOSc

 \tilde{X} D_{∞h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _u ⁺	3	Asym. stretch	892.9	Ar	IR	1

Reference

¹G. V. Chertihin, L. Andrews, M. Rosi, and C. W. Bauschlicher, Jr., J. Phys. Chem. A **101**, 9085 (1997).

CrOCr

 \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b ₂	3	Antisym. stretch	804T	Ar	IR	1

Reference

¹G. V. Chertihin, W. D. Bare, and L. Andrews, J. Chem. Phys. **107**, 2798 (1997).

MnOMn

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Asym. stretch	808.3	Ar	IR	1
			806.0	N ₂	IR	1

Reference

¹G. V. Chertihin and L. Andrews, J. Phys. Chem. A **101**, 8547 (1997).

cyc-Ir₂O \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	673.3	Ar	IR	1

Reference

¹A. Citra and L. Andrews, J. Phys. Chem. A **103**, 4182 (1999).

AgOAg

 \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Stretch	443.3T	Ar	IR	1,2

References

¹D. E. Tevault, R. R. Smardzewski, M. W. Urban, and K. Nakamoto, J. Chem. Phys. **77**, 577 (1982).

²A. Citra and L. Andrews, J. Mol. Struct. **489**, 95 (1999).

cyc-B₂C \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	B-B stretch	1392.8	Ar	IR	1

Reference

¹C. W. Larson and J. D. Presilla-Márquez, J. Chem. Phys. **111**, 1988 (1999).

ScCC

 \tilde{B} $T_0=8880(240)$ gas PE¹ \tilde{A} $T_0=2340(240)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		ScC stretch	550(60)	gas	PE	1

 \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁		ScC ₂ s-stretch	670(40)	gas	PE	1

Reference

¹X. Li and L.-S. Wang, J. Chem. Phys. **111**, 8389 (1999).

TiCC

 \tilde{D} $T_0=9410(500)$ gas PE¹ \tilde{C} $T_0=7240(290)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			720(50)	gas	PE	1

 \tilde{B} $T_0=4620(260)$ gas PE¹ \tilde{A} $T_0=3730(230)$ gas PE¹ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			560(50)	gas	PE	1

Reference

¹X.-B. Wang, C.-F. Ding, and L.-S. Wang, J. Phys. Chem. A **101**, 7699 (1997).

VCC

 \tilde{C} $T_0=13800(160)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		VC stretch	370(40)	gas	PE	1

 \tilde{B} $T_0=12020(160)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		VC stretch	480(40)	gas	PE	1

 \tilde{A} $T_0=3870(160)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		VC stretch	520(50)	gas	PE	1

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		VC stretch	550(40)	gas	PE	1

Reference

¹X. Li and L.-S. Wang, J. Chem. Phys. **111**, 8389 (1999).

NbCC

 \tilde{A}^4A_2 C_{2v} $T^a=9400(280)$ gas PE¹ \tilde{c}^2A_1 C_{2v} $T^a=8350(280)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2		560(50)	gas	PE	1

 \tilde{b}^2a_2 C_{2v} $T^a=6250(280)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2		540(50)	gas	PE	1

 \tilde{a}^2B_1 C_{2v} $T^a=3510(260)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2		550(40)	gas	PE	1

 \tilde{X}^4B_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2		530(30)	gas	PE	1

^aFrom vertical electron detachment energy.

Reference

¹H.-J. Zhai, S.-R. Liu, X. Li, and L.-S. Wang, J. Chem. Phys. **115**, 5170 (2001).

CrCC

\tilde{G}
 $T_0 = 14280(160)$ gas PE¹

\tilde{F}
 $T_0 = 11130(160)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CrC stretch	550(60)	gas	PE	1

\tilde{E}
 $T_0 = 10080(160)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CrC stretch	540(60)	gas	PE	1

\tilde{D}
 $T_0 = 8470(160)$ gas PE¹

\tilde{C}
 $T_0 = 6780(160)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CrC stretch	480(60)	gas	PE	1

\tilde{B}
 $T_0 = 5650(160)$ gas PE¹

\tilde{A}
 $T_0 = 3150(80)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CrC stretch	540(30)	gas	PE	1

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CrC stretch	510(30)	gas	PE	1

Reference

¹X. Li and L.-S. Wang, J. Chem. Phys. **111**, 8389 (1999).

MnCC

\tilde{B}
 $T_0 = 7580(160)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		MnC stretch	330(40)	gas	PE	1

\tilde{A}
 $T_0 = 5970(80)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		MnC stretch	450(40)	gas	PE	1

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		MnC stretch	520(30)	gas	PE	1

Reference

¹X. Li and L.-S. Wang, J. Chem. Phys. **111**, 8389 (1999).

FeCC

\tilde{B}
 $T_0 = 9520(160)$ gas PE²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		FeC stretch	420(60)	gas	PE	2

\tilde{A}
 $T_0 = 8070(160)$ gas PE^{1,2}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		FeC stretch	480(50)	gas	PE	2

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		C _{2v}				
a ₁		FeC ₂ s-stretch	560(30)	gas	PE	2

References

¹J. Fan and L.-S. Wang, J. Phys. Chem. **98**, 11814 (1994).

²X. Li and L.-S. Wang, J. Chem. Phys. **111**, 8389 (1999).

CoCC

\tilde{A}
 $T_0 = 7420(480)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		C _{2v}				
		CoC stretch	540(60)	gas	PE	1

Reference

¹X. Li and L.-S. Wang, J. Chem. Phys. **111**, 8389 (1999).

LiOSi

\tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	SiO stretch	998.5vs	Ar	IR	1
	2	Bend	249w	Ar	IR	1
	3	LiO stretch	668.4wm	Ar	IR	1

Reference

¹B. Tremblay, M. E. Alikhani, and L. Manceron, Chem. Phys. **218**, 37 (1997).

ScCC⁻

Threshold for electron detachment from ground-state

ScCC⁻ = 13310(240) gas PE¹

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
		ScC ₂ s-stretch	500T	gas	PE	1

Reference

¹X. Li and L.-S. Wang, J. Chem. Phys. **111**, 8389 (1999).

TiCC⁻

Threshold for electron detachment from ground-state

TiCC⁻ = 12440(160) gas PE¹

Reference

¹X.-B. Wang, C.-F. Ding, and L.-S. Wang, J. Phys. Chem. A **101**, 7699 (1997).

VCC⁻

Threshold for electron detachment from ground-state

VCC⁻ = 11460(160) gas PE¹

Reference

¹X. Li and L.-S. Wang, J. Chem. Phys. **111**, 8389 (1999).

NbCC⁻

Threshold for electron detachment from ground-state NbCC⁻ = 11130(200) gas PE¹

Reference

¹H.-J. Zhai, S.-R. Liu, X. Li, and L.-S. Wang, J. Chem. Phys. **115**, 5170 (2001).

CrCC⁻

Threshold for electron detachment from ground-state CrCC⁻ = 13150(80) gas PE¹

Reference

¹X. Li and L.-S. Wang, J. Chem. Phys. **111**, 8389 (1999).

MnCC⁻

Threshold for electron detachment from ground-state MnCC⁻ = 17100(80) gas PE¹

Reference

¹X. Li and L.-S. Wang, J. Chem. Phys. **111**, 8389 (1999).

FeCC⁻

Threshold for electron detachment from ground-state FeCC⁻ = 15980(80) gas PE^{1,2}

References

¹J. Fan and L.-S. Wang, J. Phys. Chem. **98**, 11814 (1994).

²X. Li and L.-S. Wang, J. Chem. Phys. **111**, 8389 (1999).

CoCC⁻

Threshold for electron detachment from ground-state CoCC⁻ = 13720(565) gas PE¹

Reference

¹X. Li and L.-S. Wang, J. Chem. Phys. **111**, 8389 (1999).

BeCN

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
	1	CN stretch	2183.1	Ar	IR	1
	3	BeC stretch	805.8	Ar	IR	1

Reference

¹D. V. Lanzisera and L. Andrews, J. Am. Chem. Soc. **119**, 6392 (1997).

BeNC

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
	1	NC stretch	2088.7	Ar	IR	1
	3	BeN stretch	938.4	Ar	IR	1

Reference

¹D. V. Lanzisera and L. Andrews, *J. Am. Chem. Soc.* **119**, 6392 (1997).

MgNC

$\tilde{A}^2\Pi$ $C_{\infty v}$
 $T_0=26084.01$ gas LF⁶

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend (ω)	193	gas	LF	6
		($\kappa^2\Sigma$)	206.6	gas	LF	6
Σ^+	3	MgN stretch	581.7	gas	LF	6

$A=36.93$ gas LF⁶
 $B_0=0.204$ LF⁶

$\tilde{X}^2\Sigma^+$ $C_{\infty v}$ Structure: MW^{2,7}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	NC stretch	2076.1	Ar	IR	4
Π	2	Bend	86T	gas	MW	3
Σ^+	3	MgN stretch	513.7	Ar	IR	4

$B_0=0.199$ MW^{1,5}LF⁶

References

- ¹K. Kawaguchi, E. Kagi, T. Hirano, S. Takano, and S. Saito, *Astrophys. J.* **406**, L39 (1993).
²M. A. Anderson and L. M. Ziurys, *Chem. Phys. Lett.* **231**, 164 (1994).
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⁴D. V. Lanzisera and L. Andrews, *J. Phys. Chem. A* **101**, 9666 (1997).
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⁶R. R. Wright and T. A. Miller, *J. Mol. Spectrosc.* **194**, 219 (1999).
⁷E. Kagi and K. Kawaguchi, *J. Mol. Spectrosc.* **199**, 309 (2000).

CaNC

\tilde{D}^2A' C_s
 $T_0=31515$ gas LF¹¹ $\tilde{D}-\tilde{X}$ 298–317 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	Bend	175	gas	LF	11
	3	CaN stretch	480	gas	LF	11

$\tilde{C}^2\Pi$ $C_{\infty v}$
 Unstructured absorption gas LF¹ $\tilde{C}-\tilde{X}$ 385–418 nm
 $\tau=165(38)$ ns gas LF¹

$\tilde{A}^2\Pi$ $C_{\infty v}$
 $T_0=16229.54$ gas LF^{1,3-6,9}CL² $\tilde{A}-\tilde{X}$ 572–670 nm
 $\tau(607\text{ nm})=40.8(1.5)$ ns gas LF¹
 $A=77.64$ gas LF^{6,9}
 $B_0=0.150$ LF^{6,9}

$\tilde{X}^2\Sigma^+$ $C_{\infty v}$ Structure: LF⁹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	NC stretch	2058.4	Ar	IR	10
Σ^+	3	CaN stretch	403	Ar	IR	10

$B_0=0.135$ gas LF^{4,6,9}MW^{7,8}

References

- ¹L. Pasternack and P. J. Dagdigian, *J. Chem. Phys.* **65**, 1320 (1976).
²N. Furio and P. J. Dagdigian, *Chem. Phys. Lett.* **115**, 358 (1985).
³L. C. Ellingboe, A. M. R. P. Bopegedera, C. R. Brazier, and P. F. Bernath, *Chem. Phys. Lett.* **126**, 285 (1986).
⁴C. J. Whitham, B. Soep, J.-P. Visticot, and A. Keller, *J. Chem. Phys.* **93**, 991 (1990).
⁵M. Douay and P. F. Bernath, *Chem. Phys. Lett.* **174**, 230 (1990).
⁶T. C. Steimle, D. A. Fletcher, K. Y. Jung, and C. T. Scurlock, *J. Chem. Phys.* **97**, 2909 (1992); *J. Chem. Phys.* **100**, 4025 (1994).
⁷T. C. Steimle, S. Saito, and S. Takano, *Astrophys. J.* **410**, L49 (1993).
⁸C. T. Scurlock, T. C. Steimle, R. D. Suenram, and F. J. Lovas, *J. Chem. Phys.* **100**, 3497 (1994).
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¹⁰D. V. Lanzisera and L. Andrews, *J. Phys. Chem. A* **101**, 9666 (1997).
¹¹G. M. Greetham and A. M. Ellis, *J. Chem. Phys.* **113**, 8945 (2000).

SrNC^a

\tilde{E}^2A' C_s
 $T_0=32768$ gas LF⁴ $\tilde{E}-\tilde{X}$ 294–305 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	Bend	179	gas	LF	4

\tilde{D}^2A' C_s
 $T_0=29114$ gas LF⁴ $\tilde{D}-\tilde{X}$ 322–345 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	Bend	170	gas	LF	4

\tilde{C} C_s
 $T_0=21580$ gas LF^{1,5}MPI⁵ $\tilde{C}-\tilde{X}$ 395–463 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	3	SrN stretch	337	gas	LF,MPI	5

$\tau=104.4(6.3)$ ns gas LF^{1,5}

$\tilde{A}, \tilde{B}^2\Pi, ^2\Sigma^+$ $C_{\infty v}$
 Unassigned structure gas LF^{1,2} $\tilde{A}, \tilde{B}-\tilde{X}$ 645–725 nm
 $\tau=51.2(6.2)$ ns gas LF¹
 $A(\tilde{A})=301$ gas LF²

$\tilde{X}^2\Sigma^+$ $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	NC stretch	2052.3	Ar	IR	3
Π	2	Bend	70T	gas	LF	5
Σ^+	3	SrN stretch	338(2)	Ar	IR	3

^aOriginally assigned to the SrCN structure.

References

- ¹L. Pasternack and P. J. Dagdigian, *J. Chem. Phys.* **65**, 1320 (1976).
²M. Douay and P. F. Bernath, *Chem. Phys. Lett.* **174**, 230 (1990).
³D. V. Lanzisera and L. Andrews, *J. Phys. Chem. A* **101**, 9666 (1997).
⁴G. M. Greetham and A. M. Ellis, *J. Chem. Phys.* **113**, 8945 (2000).
⁵G. M. Greetham and A. M. Ellis, *Chem. Phys. Lett.* **332**, 303 (2000).

BaNC^a

 $\tilde{C}^2\Pi$ $C_{\infty v}$

Unstructured absorption gas LF¹ $\tilde{C}-\tilde{X}\cong 500-629$ nm
 $\tau=229(13)$ ns gas LF¹

 \tilde{X} $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	NC stretch	2048.9	Ar	IR	2

^aElectronic spectrum originally assigned to the BaCN structure.

References

- ¹L. Pasternack and P. J. Dagdigian, *J. Chem. Phys.* **65**, 1320 (1976).
²D. V. Lanzisera and L. Andrews, *J. Phys. Chem. A* **101**, 9666 (1997).

FeNC

$T_0=27235.7(2)$ gas LF¹ 350–370 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	3	FeN stretch	523	gas	LF	1

$B_0=0.151$ LF¹

 $\tilde{X}^6\Delta$ $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	3	FeN stretch	468(15)	gas	LF	1

$B_0=0.144$ LF¹

Reference

- ¹J. Lie and P. J. Dagdigian, *J. Chem. Phys.* **114**, 2137 (2001).

CuCN

 \tilde{X} $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	3	CuC stretch	480(30)	gas	PE	1

Reference

- ¹A. I. Boldyrev, X. Li, and L.-S. Wang, *J. Chem. Phys.* **112**, 3627 (2000).

AgCN

 \tilde{X} $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	3	AgC stretch	390(30)	gas	PE	1

Reference

- ¹A. I. Boldyrev, X. Li, and L.-S. Wang, *J. Chem. Phys.* **112**, 3627 (2000).

ScCO⁺

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1962.4 1923.5	Ne Ar	IR IR	1 1

Reference

- ¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 2964 (1999).

YCO⁺

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1903.6	Ar	IR	1

Reference

- ¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 2964 (1999).

TiCO⁺

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2041.3	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 5259 (1999).

ReCO⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2102.1	Ne	IR	1

Reference

¹L. Andrews, M. Zhou, X. Wang, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **104**, 8887 (2000).

FeCO⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2123.0 2081.5	Ne Ar	IR IR	2 1

References

¹M. Zhou, G. V. Chertihin, and L. Andrews, *J. Chem. Phys.* **109**, 10893 (1998).

²M. Zhou, and L. Andrews, *J. Chem. Phys.* **110**, 10370 (1999).

RuCO⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2134.9	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 6956 (1999).

OsCO⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2106.0	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 6956 (1999).

CoCO⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2165.5	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7773 (1999).

RhCO⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2174.1	Ne	IR	1,2

References

¹M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **121**, 9171 (1999).

²M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7773 (1999).

IrCO⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2156.5	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7773 (1999).

NiCO⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2206.5 2176.3	Ne Ar	IR IR	1 1

Reference

¹B. Liang, M. Zhou, and L. Andrews, *J. Phys. Chem. A* **104**, 3905 (2000).

PdCO⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2206.4	Ne	IR	1

Reference

¹B. Liang, M. Zhou, and L. Andrews, *J. Phys. Chem. A* **104**, 3905 (2000).

PtCO⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2204.7	Ne	IR	1

Reference

¹B. Liang, M. Zhou, and L. Andrews, *J. Phys. Chem. A* **104**, 3905 (2000).

CuCO⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2234.4	Ne	IR	1
			2174.4T	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Chem. Phys.* **111**, 4548 (1999).

AgCO⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2233.1	Ne	IR	1

Reference

¹B. Liang and L. Andrews, *J. Phys. Chem. A* **104**, 9156 (2000).

AuCO⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2236.8	Ne	IR	1

Reference

¹B. Liang and L. Andrews, *J. Phys. Chem. A* **104**, 9156 (2000).

cyc-BCC

$T_0 = 11745(3)$ Ne AB³

$T_0 = 6296$ Ne AB³

 \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
α_1	2	CBC s-stretch	1196.8(1.0)	Ne	IR	3
			1194.6	Ar	IR	1,2

References

¹J. M. L. Martin, P. R. Taylor, J. T. Yustein, T. R. Burkholder, and L. Andrews, *J. Chem. Phys.* **99**, 12 (1993).

²J. D. Presilla-Márquez, C. W. Larson, P. G. Carrick, and C. M. L. Rittby, *J. Chem. Phys.* **105**, 3398 (1996).

³M. Wyss, M. Grutter, and J. P. Maier, *J. Phys. Chem. A* **102**, 9106 (1998).

BNB

$\bar{A} \ 2\Sigma_g^+$ D_{∞h}
 $T_0 = 6330(40)$ gas PE^{5,6}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1174(40)	gas	PE	5,6
Σ_u^+	3	Asym. stretch	2492(40)	gas	PE	5,6

In the argon-matrix study,² the $\nu_1 + \nu_3$ combination band at 1998.3, originally assigned to the cyclic isomer, and other bands between 3250 and 6150 are moderately intense, suggesting strong vibronic coupling to a low-lying excited electronic state. These bands are also seen in the gas-phase photoelectron spectrum,⁶ confirming that ground-state bands involving odd numbers of quanta of ν_3 are strongly coupled to the \bar{A} state.

 $\tilde{X} \ 2\Sigma_u^+$ C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1143(40)	gas	PE	5,6
			1116 ^a	Ar	IR	2
Σ_u^+	3	Asym. stretch	855(40)	gas	PE	5,6
			882.3s	Ar	IR	2-4
			910.4	N ₂	IR	1
			890.3			

^a($\nu_1 + \nu_3$) - ν_3 .

References

¹P. Hassanzadeh and L. Andrews, *J. Phys. Chem.* **96**, 9177 (1992).

²L. Andrews, P. Hassanzadeh, T. R. Burkholder, and J. M. L. Martin, *J. Chem. Phys.* **98**, 922 (1993).

³C. A. Thompson and L. Andrews, *J. Am. Chem. Soc.* **117**, 10125 (1995).

⁴S. Li, R. J. Van Zee, and W. Weltner, Jr., *Chem. Phys. Lett.* **262**, 298 (1996).

⁵K. R. Asmis, T. R. Taylor, and D. M. Neumark, *Eur. Phys. J. D* **9**, 257 (1999).

⁶K. R. Asmis, T. R. Taylor, and D. M. Neumark, *J. Chem. Phys.* **111**, 8838 (1999).

AINAI

\tilde{X} $D_{\infty h}$						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	544.9T	N ₂	IR	1
Σ_u^+	3	Asym. stretch	981.3 974.8 956.7	Ar N ₂	IR	1

Reference

¹L. Andrews, M. Zhou, G. V. Chertihin, W. D. Bare, and Y. Hannachi, J. Phys. Chem. A **104**, 1656 (2000).

GaNga

\tilde{X} $D_{\infty h}$						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_u^+	3	Asym. stretch	757.4T	N ₂	IR	1

Reference

¹M. Zhou and L. Andrews, J. Phys. Chem. A **104**, 1648 (2000).

InNIn

\tilde{X} $D_{\infty h}$						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_u^+	3	Asym. stretch	666.5	N ₂	IR	1

Reference

¹M. Zhou and L. Andrews, J. Phys. Chem. A **104**, 1648 (2000).

AICC

\tilde{A} $T_0=7910(320)$ gas PE ¹						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		AIC stretch	590(50)	gas	PE	1

\tilde{X} C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		AIC ₂ s-stretch	590(40)	gas	PE	1

Reference

¹A. I. Boldyrev, J. Simons, X. Li, and L.-S. Wang, J. Am. Chem. Soc. **121**, 10193 (1999).

AICSi

\tilde{A}
 $T_0=4200(640)$ gas PE¹

Reference

¹A. I. Boldyrev, J. Simons, X. Li, and L.-S. Wang, J. Am. Chem. Soc. **121**, 10193 (1999).

Al₂P

\tilde{B}^2B_1 C_{2v}
 $T^a=3770(260)$ gas PE¹

\tilde{A}^2A_1 C_{2v}
 $T^a=2560(260)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	421(15)	gas	PE	1

\tilde{X}^2B_2 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	Bend	70T	gas	PE	1

^aFrom vertical ionization potential.

Reference

¹H. Gómez, T. R. Taylor, and D. M. Neumark, J. Phys. Chem. A **105**, 6886 (2001).

Ga₂P

\tilde{B}
 $T^a=3230T$ gas PE³

\tilde{A}
 $T_0=2160(200)$ gas PE^{2,3}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	311	gas	PE	2,3

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	Bend	64	gas	PE	2
b_2	3	Asym. stretch	281.2	Ar	IR	1

^aFrom vertical photodetachment energy with respect to the neutral ground state.

References

- ¹S. Li, R. J. Van Zee, and W. Weltner, Jr., *J. Phys. Chem.* **97**, 11393 (1993).
²T. R. Taylor, K. R. Asmis, H. Gomez, and D. M. Neumark, *Eur. Phys. J. D* **9**, 317 (1999).
³T. R. Taylor, H. Gómez, K. R. Asmis, and D. M. Neumark, *J. Chem. Phys.* **115**, 4620 (2001).

In₂P

\tilde{A}
 $T_0 = 1860\text{T}$ gas PE^{2,3}

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	204(2)	gas	TPE	3
	2	Bend	47.0(3)	gas	TPE	3
b_2	3	Asym. stretch	249.3	Ar	IR	1

References

- ¹S. Li, R. J. Van Zee, and W. Weltner, Jr., *J. Phys. Chem.* **98**, 2275 (1994).
²C. Xu, E. de Beer, D. W. Arnold, C. C. Arnold, and D. M. Neumark, *J. Chem. Phys.* **101**, 5406 (1994).
³C. C. Arnold and D. M. Neumark, *Can. J. Phys.* **72**, 1322 (1994).

Ga₂As

\tilde{B}
 $T_0 = 2260\text{T}$ gas PE²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1		279	gas	PE	2

\tilde{A}
 $T_0 = 1690(320)$ gas PE²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1		200	gas	PE	2

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b_2	3	Asym. stretch	205.4	Ar	IR	1

References

- ¹S. Li, R. J. Van Zee, and W. Weltner, Jr., *J. Phys. Chem.* **97**, 11393 (1993).
²T. R. Taylor, H. Gómez, K. R. Asmis, and D. M. Neumark, *J. Chem. Phys.* **115**, 4620 (2001).

PdCN⁻

Threshold for electron detachment from ground-state PdCN⁻ = 20520(60) gas PE¹

Reference

- ¹S. A. Klopčič, V. D. Moravec, and C. C. Jarrold, *J. Chem. Phys.* **110**, 8986 (1999).

CuCN⁻

Threshold for electron detachment from ground-state CuCN⁻ = 11830(80) gas PE^{1,2}

References

- ¹A. I. Boldyrev, X. Li, and L.-S. Wang, *J. Chem. Phys.* **112**, 3627 (2000).
²Y. Negishi, T. Yasuike, F. Hayakawa, M. Kizawa, S. Yabushita, A. Nakajima, and K. Kaya, *J. Chem. Phys.* **113**, 1725 (2000).

CuNC⁻

Threshold for electron detachment from ground-state CuNC⁻ = 13150(1200) gas PE¹

Reference

- ¹Y. Negishi, T. Yasuike, F. Hayakawa, M. Kizawa, S. Yabushita, A. Nakajima, and K. Kaya, *J. Chem. Phys.* **113**, 1725 (2000).

AgCN⁻

Threshold for electron detachment from ground-state AgCN⁻ = 12810(80) gas PE¹

Reference

- ¹A. I. Boldyrev, X. Li, and L.-S. Wang, *J. Chem. Phys.* **112**, 3627 (2000).

ScCO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1851.4	Ne	IR	1
			1834.2	Ar	IR	1

Reference

- ¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 2964 (1999).

YCO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1874.1	Ar	IR	1
			1869.0			

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 2964 (1999).

TiCO

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1920.0 1887.8	Ne Ar	IR IR	2 1,2

References

¹G. V. Chertihin and L. Andrews, *J. Am. Chem. Soc.* **117**, 1595 (1995).

²M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 5259 (1999).

ZrCO

 \tilde{X} $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	1899.5	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **122**, 1531 (2000).

HfCO

 \tilde{X} $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	1868.6	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **122**, 1531 (2000).

VCO

 $\tilde{X}^6\Sigma$ $C_{\infty v}$ Structure: ESR²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	1930.6 1904.7 1890 1868	Ne Ar Kr Xe	IR IR IR IR	3 1,3 1 1

References

¹L. Hanlan, H. Huber, and G. A. Ozin, *Inorg. Chem.* **15**, 2592 (1976).

²R. J. Van Zee, S. B. H. Bach, and W. Weltner, Jr., *J. Phys. Chem.* **90**, 583 (1986).

³M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 5259 (1999).

CNbO

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NbO stretch	919.8	Ne	IR	1
		NbC stretch	783.7	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7785 (1999).

NbCO

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1932.0	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7785 (1999).

TaCO

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1865.2 1862.7 1831T 1819T	Ne Ar	IR IR	2 1

References

¹R. L. DeKock, *Inorg. Chem.* **10**, 1205 (1971).

²M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7785 (1999).

CrCO

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1975.3	Ar	IR	1,2

References

- ¹S. B. H. Bach, C. A. Taylor, R. J. Van Zee, M. T. Vala, and W. Weltner, Jr., *J. Am. Chem. Soc.* **108**, 7104 (1986).
²P. F. Souter and L. Andrews, *J. Am. Chem. Soc.* **119**, 7350 (1997).

MoCO

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1862.6	Ar	IR	1

Reference

- ¹P. F. Souter and L. Andrews, *J. Am. Chem. Soc.* **119**, 7350 (1997).

WCO

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1848.8	Ar	IR	1

Reference

- ¹P. F. Souter and L. Andrews, *J. Am. Chem. Soc.* **119**, 7350 (1997).

MnCO

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1950.7	Ne	IR	1
			1933.6	Ar	IR	1

Reference

- ¹L. Andrews, M. Zhou, X. Wang, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **104**, 8887 (2000).

ReCO

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1884.0	Ar	IR	1

Reference

- ¹L. Andrews, M. Zhou, X. Wang, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **104**, 8887 (2000).

FeCO

 $\tilde{a}^5\Sigma^-$ $C_{\infty v}$
 $T_0 = 1135(25)$ gas PE²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	1990(15)	gas	PE	2
Π	2	Bend	180(60)	gas	PE	2
Σ^+	3	FeC stretch	460(15)	gas	PE	2

 $\tilde{X}^3\Sigma^-$ $C_{\infty v}$ Structure: MW^{3,5}DL⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	1946.47 1933.7 1922.0T ^a	gas Ne Ar	PE,DL IR IR	1,2,4 8 7
Π	2	Bend	330(50)	gas	PE	2
Σ^+	3	FeC stretch	530(10)	gas	PE	2

 $B_0 = 0.146$ MW^{3,5,6}DL⁴

^aAs discussed in Ref. 8, isotopic shift disparities suggest that this peak may be contributed by an ArFeCO species or by FeCO in the low-lying $^3\Delta$ state, which may lie below the $^3\Sigma^-$ state for FeCO isolated in solid argon.

References

- ¹R. J. Ryther and E. Weitz, *J. Phys. Chem.* **95**, 9841 (1991).
²P. W. Villalta and D. G. Leopold, *J. Chem. Phys.* **98**, 7730 (1993).
³Y. Kasai, K. Obi, Y. Ohshima, Y. Endo, and K. Kawaguchi, *J. Chem. Phys.* **103**, 90 (1995).
⁴K. Tanaka, K. Sakaguchi, and T. Tanaka, *J. Chem. Phys.* **106**, 2118 (1997).
⁵K. Tanaka, M. Shirasaka, and T. Tanaka, *J. Chem. Phys.* **106**, 6821 (1997).
⁶E. Kagi, Y. Kasai, H. Ungerechts, and K. Kawaguchi, *Astrophys. J.* **488**, 776 (1997).
⁷M. Zhou, G. V. Chertihin, and L. Andrews, *J. Chem. Phys.* **109**, 10893 (1998).
⁸M. Zhou and L. Andrews, *J. Chem. Phys.* **110**, 10370 (1999).

RuCO

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1935.6 1917.7	Ne Ar	IR IR	1 1

Reference

- ¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 6956 (1999).

OsCO

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1972.6	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 6956 (1999).

CoCO

\tilde{X} $C_{\infty v}$						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	1973.9 1960.7 1957.3vs 1952 1944 1947 1941	Ne Ar Kr Xe	IR IR IR	3 1,2,4 1 1
Π	2	Bend	424.9w	Ar	IR	4
Σ^+	3	CoC stretch	579.2vw	Ar	IR	4

References

¹L. A. Hanlan, H. Huber, E. P. Kündig, B. R. McGarvey, and G. A. Ozin, *J. Am. Chem. Soc.* **97**, 7054 (1975).

²M. Zhou and L. Andrews, *J. Phys. Chem. A* **102**, 10250 (1998).

³M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7773 (1999).

⁴B. Tremblay, M. E. Alikhani, and L. Manceron, *J. Phys. Chem. A* **105**, 11388 (2001).

RhCO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2022.5 2008.0	Ne Ar	IR IR	2,3 1,3

References

¹G. A. Ozin and A. J. L. Hanlan, *Inorg. Chem.* **18**, 2091 (1979).

²M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **121**, 9171 (1999).

³M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7773 (1999).

IrCO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2024.5	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7773 (1999).

NiCO

$T_0 = 11050(240)$ gas PE²

\tilde{X} $C_{\infty v}$						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	1940(80) 2006.6 1994.5vs	gas Ne Ar	PE IR IR	2 5 1,3,4
Π	2	Bend	409.1vw	Ar	IR	3
Σ^+	3	NiC stretch	591.1vw	Ar	IR	3

References

¹R. L. DeKock, *Inorg. Chem.* **10**, 1205 (1971).

²A. E. Stevens, C. S. Feigerle, and W. C. Lineberger, *J. Am. Chem. Soc.* **104**, 5026 (1982).

³H. A. Joly and L. Manceron, *Chem. Phys.* **226**, 61 (1998).

⁴M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **120**, 11499 (1998).

⁵B. Liang, M. Zhou, and L. Andrews, *J. Phys. Chem. A* **104**, 3905 (2000).

PdCO

\tilde{X} $C_{\infty v}$						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	2140(60) 2056.4 2044.2vs 2045	gas Ne Ar Kr	PE IR IR IR	3 5 1,2,4 2
Π	2	Bend	615.7vw	Ar	IR	4
Σ^+	3	PdC stretch	350(40) 472.0w	gas Ar	PE IR	3 4

References

¹E. P. Kündig, M. Moskovits, and G. A. Ozin, *Can. J. Chem.* **50**, 3587 (1972).

²J. H. Darling and J. S. Ogden, *J. Chem. Soc., Dalton Trans.* 1079 (1973).

³S. A. Klopčič, V. D. Moravec, and C. C. Jarrold, *J. Chem. Phys.* **110**, 8986 (1999).

⁴B. Tremblay and L. Manceron, *Chem. Phys.* **250**, 187 (1999).

⁵B. Liang, M. Zhou, and L. Andrews, *J. Phys. Chem. A* **104**, 3905 (2000).

PtCO

\tilde{X} $C_{\infty v}$ Structure: MW ⁴						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	2065.5 2051.9	Ne Ar	IR IR	3 1,2
Π	2	Bend	916.8	Ar	IR	2
Σ^+	3	PtC stretch	580.8	Ar	IR	2

$B_0 = 0.111$ MW⁴

References

- ¹E. P. Kündig, D. McIntosh, M. Moskovits, and G. A. Ozin, *J. Am. Chem. Soc.* **95**, 7234 (1973).
²L. Manceron, B. Tremblay, and M. E. Alikhani, *J. Phys. Chem. A* **104**, 3750 (2000).
³B. Liang, M. Zhou, and L. Andrews, *J. Phys. Chem. A* **104**, 3905 (2000).
⁴C. J. Evans and M. C. L. Gerry, *J. Phys. Chem. A* **105**, 9659 (2001).

CuCO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CO stretch	2029.7	Ne	IR	4
			2010.4	Ar	IR	1–4
			2029.5	N ₂	IR	2
2	Bend	207.5	Ar	IR	3	
		3	CuC stretch	322.7	Ar	IR

References

- ¹H. Huber, E. P. Kündig, M. Moskovits, and G. A. Ozin, *J. Am. Chem. Soc.* **97**, 2097 (1975).
²S. Dobos and S. Nunziante Cesaro, *High Temp. Mater. Sci.* **37**, 81 (1997).
³B. Tremblay and L. Manceron, *Chem. Phys.* **242**, 235 (1999).
⁴M. Zhou and L. Andrews, *J. Chem. Phys.* **111**, 4548 (1999).

AuCO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2053.2	Ne	IR	1

Reference

- ¹B. Liang and L. Andrews, *J. Phys. Chem. A* **104**, 9156 (2000).

ThCO

In a neon matrix, rearranges to CThO when the sample is exposed to radiation in the 470–580 nm spectral region.¹

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1817.5	Ne	IR	1,2

References

- ¹M. Zhou, L. Andrews, J. Li, and B. E. Bursten, *J. Am. Chem. Soc.* **121**, 12188 (1999).
²J. Li, B. E. Bursten, M. Zhou, and L. Andrews, *Inorg. Chem.* **40**, 5448 (2001).

CThO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	ThO stretch	812.2	Ne	IR	1,2
	3	ThC stretch	617.7	Ne	IR	1,2

References

- ¹M. Zhou, L. Andrews, J. Li, and B. E. Bursten, *J. Am. Chem. Soc.* **121**, 12188 (1999).
²J. Li, B. E. Bursten, M. Zhou, and L. Andrews, *Inorg. Chem.* **40**, 5448 (2001).

CUO

$\tilde{a}^3\Phi$ $C_{\infty v}$
 When CUO is isolated in an argon matrix, a new absorption pattern results, attributed to a sufficiently large shift in the energy of this very low-lying excited state to place it below the $\tilde{X}^1\Sigma^+$ state of the free molecule.²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	Mixed	852.5	Ar	IR	2
	3	Mixed	804.3	Ar	IR	2

$\tilde{X}^1\Sigma^+$		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	UC stretch	1047.3	Ne	IR	1
	3	UO stretch	872.2	Ne	IR	1

References

- ¹M. Zhou, L. Andrews, J. Li, and B. E. Bursten, *J. Am. Chem. Soc.* **121**, 9712 (1999).
²L. Andrews, B. Liang, J. Li, and B. E. Bursten, *Angew. Chem. Int. Ed.* **39**, 4565 (2000).

UCO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1917.8	Ne	IR	3
			1893	Ar	IR	1–3

References

- ¹J. L. Slater, R. K. Sheline, K. C. Lin, and W. Weltner, Jr., *J. Chem. Phys.* **55**, 5129 (1971).
²T. J. Tague, Jr., L. Andrews, and R. D. Hunt, *J. Phys. Chem.* **97**, 10920 (1993).
³M. Zhou, L. Andrews, J. Li, and B. E. Bursten, *J. Am. Chem. Soc.* **121**, 9712 (1999).

ScNN

\tilde{X} $C_{\infty v}$						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	NN stretch	1902.0	Ar	IR	1

Reference

¹G. V. Chertihin, L. Andrews, and C. W. Bauschlicher, Jr., *J. Am. Chem. Soc.* **120**, 3205 (1998).

cyc-ScNN

\tilde{X} C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	NN stretch	1714.0T	Ar	IR	1

Reference

¹G. V. Chertihin, L. Andrews, and C. W. Bauschlicher, Jr., *J. Am. Chem. Soc.* **120**, 3205 (1998).

cyc-YNN

\tilde{X} C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	NN stretch	1763.4	Ar	IR	1

Reference

¹G. V. Chertihin, W. D. Bare, and L. Andrews, *J. Phys. Chem. A* **102**, 3697 (1998).

cyc-LaNN

\tilde{X} C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	NN stretch	1770.7	Ar	IR	1

Reference

¹G. V. Chertihin, W. D. Bare, and L. Andrews, *J. Phys. Chem. A* **102**, 3697 (1998).

TiNN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NN stretch	1847.1	Ar	IR	1,2

References

¹G. V. Chertihin and L. Andrews, *J. Phys. Chem.* **98**, 5891 (1994).

²G. P. Kushto, P. F. Souter, G. V. Chertihin, and L. Andrews, *J. Chem. Phys.* **110**, 9020 (1999).

cyc-TiNN

\tilde{X} C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	NN stretch	1125.9	Ar	IR	1

Reference

¹G. P. Kushto, P. F. Souter, G. V. Chertihin, and L. Andrews, *J. Chem. Phys.* **110**, 9020 (1999).

NTiN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		TiN ₂ a-stretch	867.3	Ar	IR	1

Reference

¹G. P. Kushto, P. F. Souter, G. V. Chertihin, and L. Andrews, *J. Chem. Phys.* **110**, 9020 (1999).

ZrNN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NN stretch	1797.5	Ar	IR	1

Reference

¹G. P. Kushto, P. F. Souter, G. V. Chertihin, and L. Andrews, *J. Chem. Phys.* **110**, 9020 (1999).

cyc-ZrNN

\tilde{X}						
C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	NN stretch	1022.8	Ar	IR	1

Reference

¹G. P. Kushto, P. F. Souter, G. V. Chertihin, and L. Andrews, *J. Chem. Phys.* **110**, 9020 (1999).

NZrN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		ZrN stretch	706.3	Ar	IR	1

Reference

¹G. P. Kushto, P. F. Souter, G. V. Chertihin, and L. Andrews, *J. Chem. Phys.* **110**, 9020 (1999).

HfNN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NN stretch	1793.8	Ar	IR	1

Reference

¹G. P. Kushto, P. F. Souter, G. V. Chertihin, and L. Andrews, *J. Chem. Phys.* **110**, 9020 (1999).

NHfN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		HfN stretch	708.1	Ar	IR	1

Reference

¹G. P. Kushto, P. F. Souter, G. V. Chertihin, and L. Andrews, *J. Chem. Phys.* **110**, 9020 (1999).

NVN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			820.1	N ₂	IR	1

Reference

¹L. Andrews, W. D. Bare, and G. V. Chertihin, *J. Phys. Chem. A* **101**, 8417 (1997).

VNN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NN stretch	1945.0	Ar	IR	1

Reference

¹L. Andrews, W. D. Bare, and G. V. Chertihin, *J. Phys. Chem. A* **101**, 8417 (1997).

cyc-VNN

\tilde{X}						
C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	NN stretch	1709.1	Ar	IR	1

Reference

¹L. Andrews, W. D. Bare, and G. V. Chertihin, *J. Phys. Chem. A* **101**, 8417 (1997).

NbNN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NN stretch	1914.5 1965.2T	Ar N ₂	IR IR	1 1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **102**, 9061 (1998).

cyc-NbN₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	NN stretch	1888.7	Ar	IR	1
			1815.6	N ₂	IR	1

Reference

¹M. Zhou and L. Andrews, J. Phys. Chem. A **102**, 9061 (1998).

NNbN

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	NbN ₂ s-stretch	918.6	Ar	IR	1
			914.9			
b_2	3	NbN ₂ a-stretch	685.4	Ar	IR	1
			680.8			

Reference

¹M. Zhou and L. Andrews, J. Phys. Chem. A **102**, 9061 (1998).

TaNN

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NN stretch	1952.8	Ar	IR	1
			1947.6T	N ₂	IR	1

Reference

¹M. Zhou and L. Andrews, J. Phys. Chem. A **102**, 9061 (1998).

cyc-TaN₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	NN stretch	1762.6	N ₂	IR	1

Reference

¹M. Zhou and L. Andrews, J. Phys. Chem. A **102**, 9061 (1998).

NTaN

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	TaN ₂ s-stretch	922.7	Ar	IR	1
b_2	3	TaN ₂ a-stretch	740.3	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, J. Phys. Chem. A **102**, 9061 (1998).

NCrN

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	956.1	N ₂	IR	1
b_2	3	Asym. stretch	875.7	N ₂	IR	1

Reference

¹L. Andrews, W. D. Bare, and G. V. Chertihin, J. Phys. Chem. A **101**, 8417 (1997).

NMoN

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	MoN s-stretch	975.1	N ₂	IR	1
b_2	3	MoN a-stretch	877.1	Ar	IR	1
			861.6	N ₂	IR	1
			860.6			

Reference

¹L. Andrews, P. F. Souter, W. D. Bare, and B. Liang, J. Phys. Chem. A **103**, 4649 (1999).

MoNN

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NN stretch	1938.7T	Ar	IR	1

Reference

¹L. Andrews, P. F. Souter, W. D. Bare, and B. Liang, J. Phys. Chem. A **103**, 4649 (1999).

NWN

\tilde{X} C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b_2	3	WN a-stretch	878.3	Ar	IR	1
			870.3	N ₂	IR	1
			865.1			

Reference

¹L. Andrews, P. F. Souter, W. D. Bare, and B. Liang, J. Phys. Chem. A **103**, 4649 (1999).

WNN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NN stretch	1894.3T	Ar	IR	1

Reference

¹L. Andrews, P. F. Souter, W. D. Bare, and B. Liang, J. Phys. Chem. A **103**, 4649 (1999).

NMnN

\tilde{X} C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b_2	3	Asym. stretch	858.7	N ₂	IR	1

Reference

¹L. Andrews, W. D. Bare, and G. V. Chertihin, J. Phys. Chem. A **101**, 8417 (1997).

cyc-MnNN

\tilde{X} C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	NN stretch	1819.3	Ar	IR	1

Reference

¹L. Andrews, W. D. Bare, and G. V. Chertihin, J. Phys. Chem. A **101**, 8417 (1997).

ReNN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NN stretch	1940.1	Ar	IR	1
			1944.7T	N ₂	IR	1

Reference

¹M. Zhou and L. Andrews, J. Phys. Chem. A **102**, 9061 (1998).

cyc-ReN₂

\tilde{X} C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	NN stretch	1894.7T	N ₂	IR	1

Reference

¹M. Zhou and L. Andrews, J. Phys. Chem. A **102**, 9061 (1998).

NRuN

\tilde{X} C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b_2	3	Asym. stretch	831.9	Ar	IR	1,2
			831.5	N ₂	IR	2

References

¹A. Citra and L. Andrews, J. Am. Chem. Soc. **121**, 11567 (1999).

²A. Citra and L. Andrews, J. Phys. Chem. A **104**, 1152 (2000).

NOsN

\tilde{X} C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b_2	3	Asym. stretch	900.4	Ar	IR	1,2
			901.1	N ₂	IR	1,2

References

¹A. Citra and L. Andrews, J. Am. Chem. Soc. **121**, 11567 (1999).

²A. Citra and L. Andrews, J. Phys. Chem. A **104**, 1152 (2000).

CoNN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NN stretch	2100.9	Ar	IR	1,2
			2109	N ₂	IR	2

References

- ¹G. A. Ozin and A. Vander Voet, *Can. J. Chem.* **51**, 637 (1973).
²L. Andrews, A. Citra, G. V. Chertihin, W. D. Bare, and M. Neurock, *J. Phys. Chem. A* **102**, 2561 (1998).

cyc-CoNN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NN stretch	1873.5	N ₂	IR	1

Reference

- ¹L. Andrews, A. Citra, G. V. Chertihin, W. D. Bare, and M. Neurock, *J. Phys. Chem. A* **102**, 2561 (1998).

RhNN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NN stretch	2153.3	Ar	IR	1,2

References

- ¹G. A. Ozin and A. Vander Voet, *Can. J. Chem.* **51**, 3332 (1973).
²A. Citra and L. Andrews, *J. Phys. Chem. A* **103**, 3410 (1999).

NRhN

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b ₂	3	RhN ₂ a-stretch	823.2	Ar	IR	1
			830.9	N ₂	IR	1

Reference

- ¹A. Citra and L. Andrews, *J. Phys. Chem. A* **103**, 3410 (1999).

NINN

\tilde{X} C _{∞v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	NN stretch	2089.2vs	Ar	IR, Ra	1-4
Π	2	Bend	357.0w	Ar	IR	4
Σ ⁺	3	NiN stretch	563.7wm	Ar	IR	3,4

References

- ¹H. Huber, E. P. Kündig, M. Moskovits, and G. A. Ozin, *J. Am. Chem. Soc.* **95**, 332 (1973).
²W. Klotzbücher and G. A. Ozin, *J. Am. Chem. Soc.* **97**, 2672 (1975).
³L. Andrews, A. Citra, G. V. Chertihin, W. D. Bare, and M. Neurock, *J. Phys. Chem. A* **102**, 2561 (1998).
⁴L. Manceron, M. E. Alikhani, and H. A. Joly, *Chem. Phys.* **228**, 73 (1998).

cyc-NiNN

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	NN stretch	1879.3	Ar	IR	1
			1902.7T	N ₂	IR	1
			1870.6T	N ₂	IR	1

Reference

- ¹L. Andrews, A. Citra, G. V. Chertihin, W. D. Bare, and M. Neurock, *J. Phys. Chem. A* **102**, 2561 (1998).

PdNN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NN stretch	2215	Ar	IR,Ra	1,2
		PdN stretch	378	Ar	IR	2

References

- ¹H. Huber, E. P. Kündig, M. Moskovits, and G. A. Ozin, *J. Am. Chem. Soc.* **95**, 332 (1973).
²W. Klotzbücher and G. A. Ozin, *J. Am. Chem. Soc.* **97**, 2672 (1975).

PtNN

\tilde{X} C _{∞v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	NN stretch	2169.5	Ne	IR	4
			2168.5	Ar	IR	1-4
			2170.7	Kr	IR	2
			2168.8			
			2173.0	N ₂	IR	4
Σ ⁺	3	PtN stretch	499.6	Ar	IR	4

References

- ¹E. P. Kündig, M. Moskovits, and G. A. Ozin, *Can. J. Chem.* **51**, 2710 (1973).
²D. W. Green, J. Thomas, and D. M. Gruen, *J. Chem. Phys.* **58**, 5453 (1973).
³W. Klotzbücher and G. A. Ozin, *J. Am. Chem. Soc.* **97**, 2672 (1975).
⁴A. Citra, X. Wang, W. D. Bare, and L. Andrews, *J. Phys. Chem. A* **105**, 7799 (2001).

NCeN

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ_u^+	3	Asym. stretch	790.9 756.4	Ar N_2	IR	1 1

Reference

- ¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **102**, 10238 (1998).

CeNN

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
		NN stretch	1569.1	Ar	IR	1

Reference

- ¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **102**, 10238 (1998).

NPrN

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ_u^+	3	Asym. stretch	809.6	N_2	IR	1

Reference

- ¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **102**, 10238 (1998).

NNdN

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ_u^+	3	Asym. stretch	778.4	N_2	IR	1

Reference

- ¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **102**, 10238 (1998).

NSmN

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ_u^+	3	Asym. stretch	738.1	N_2	IR	1

Reference

- ¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **102**, 10238 (1998).

SmNN

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
		NN stretch	1576.8	N_2	IR	1

Reference

- ¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **102**, 10238 (1998).

cyc-SmNN

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1	1	NN stretch	1744.9	Ar	IR	1

Reference

- ¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **102**, 10238 (1998).

cyc-EuNN

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1	1	NN stretch	1787.5	Ar	IR	1

Reference

- ¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **102**, 10238 (1998).

cyc-GdNN

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1	1	NN stretch	1876.9	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **102**, 10238 (1998).

GdNN

\tilde{X}						
C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NN stretch	1558.2	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **102**, 10238 (1998).

cyc-TbNN

\tilde{X}						
C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	NN stretch	1859.5 1846.1	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 1311 (1999).

cyc-DyNN

\tilde{X}						
C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	NN stretch	1809.2 1799.8 1794.8 1759.4 1755.5 1744.7	Ar N ₂	IR IR	1 1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 1311 (1999).

cyc-HoNN

\tilde{X}						
C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	NN stretch	1798.3 1754.7 1749.3	Ar N ₂	IR IR	1 1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 1311 (1999).

cyc-ErNN

\tilde{X}						
C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	NN stretch	1802 1769.7 1753.4 1748.5	Ar N ₂	IR IR	1 1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 1311 (1999).

cyc-TmNN

\tilde{X}						
C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	NN stretch	1799.5 1752.3	Ar N ₂	IR IR	1 1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 1311 (1999).

cyc-YbNN

\tilde{X}						
C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	NN stretch	1744.1	N ₂	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 1311 (1999).

cyc-LuNN

\tilde{X}						
C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	NN stretch	1845.3 1840.2 1836.0 1824.3 1852.0 1848.9 1846.0 1844.2 1839.5	Ar N ₂	IR IR	1 1

Reference

¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **103**, 1311 (1999).

NThN

\tilde{X} $D_{\infty v}$						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ_u^+	3	Asym. stretch	756.6	Ar	IR	1

Reference

¹G. P. Kushto, P. F. Souter, and L. Andrews, *J. Chem. Phys.* **108**, 7121 (1998).

ThNN

\tilde{X} $C_{\infty v}$						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ^+	1	NN stretch	1673.9	Ar	IR	1

Reference

¹G. P. Kushto, P. F. Souter, and L. Andrews, *J. Chem. Phys.* **108**, 7121 (1998).

NUN

\tilde{X} $D_{\infty h}$						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1008.3 ^a	Ar	IR	2
Σ_u^+	3	Asym. stretch	1076.6	Ne	IR	4
			1051.0	Ar	IR	1,2,5
			996T	N ₂	IR	2,3,5

^aCalculated from position of ¹⁴N¹⁵N absorption.

References

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²R. D. Hunt, J. T. Yustein, and L. Andrews, *J. Chem. Phys.* **98**, 6070 (1993).
³K. Sankaran, K. Sundararajan, and K. S. Viswanathan, *Bull. Mater. Sci.* **22**, 785 (1999).
⁴M. Zhou and L. Andrews, *J. Chem. Phys.* **111**, 11044 (1999).
⁵K. Sankaran, K. Sundararajan, and K. S. Viswanathan, *J. Phys. Chem. A* **105**, 3995 (2001).

cyc-ScNO⁺

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'			1159.8	Ne	IR	2
			1117.0	Ar	IR	1

References

- ¹G. P. Kushto, M. Zhou, L. Andrews, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **103**, 1115 (1999).
²L. Andrews and X. Wang, *J. Phys. Chem. A* **106**, 1196 (2002).

cyc-VNO⁺

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	NO stretch	1146.5	Ne	IR	1
			1143.7			
			1139.5	Ar	IR	1
			1137.0			

Reference

¹L. Andrews and X. Wang, *J. Phys. Chem. A* **106**, 1196 (2002).

NVO⁺

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
			1020.5	Ne	IR	1

Reference

¹L. Andrews and X. Wang, *J. Phys. Chem. A* **106**, 1196 (2002).

FeNO⁺

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
		NO stretch	1897.3	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **104**, 3915 (2000).

RuNO⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1918.0	Ne	IR	1

Reference¹A. Citra and L. Andrews, J. Phys. Chem. A **104**, 8689 (2000).**OsNO⁺** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1921.8	Ne	IR	1
			1901.5	Ar	IR	1

Reference¹A. Citra and L. Andrews, J. Phys. Chem. A **104**, 8689 (2000).**CoNO⁺** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO stretch	1957.5	Ne	IR	1

Reference¹M. Zhou and L. Andrews, J. Phys. Chem. A **104**, 3915 (2000).**RhNO⁺** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1957.5	Ne	IR	1

Reference¹A. Citra and L. Andrews, J. Phys. Chem. A **104**, 11897 (2000).**IrNO⁺** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1990.0	Ne	IR	1

Reference¹A. Citra and L. Andrews, J. Phys. Chem. A **104**, 11897 (2000).**NiNO⁺** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO stretch	2001.9	Ne	IR	1

Reference¹M. Zhou and L. Andrews, J. Phys. Chem. A **104**, 3915 (2000).**PdNO⁺** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1921.8	Ne	IR	1
			1921.6	Ar	IR	1

Reference¹A. Citra and L. Andrews, J. Phys. Chem. A **104**, 8160 (2000).**PtNO⁺** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	2019.6	Ne	IR	1
			2010.4	Ar	IR	1

Reference¹A. Citra and L. Andrews, J. Phys. Chem. A **104**, 8160 (2000).**CuNO⁺** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1907.9	Ne	IR	1

Reference¹M. Zhou and L. Andrews, J. Phys. Chem. A **104**, 2618 (2000).

AgNO⁺

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1910.9 1904.3	Ne Ar	IR IR	1 1

Reference

¹A. Citra and L. Andrews, *J. Phys. Chem. A* **105**, 3042 (2001).

NUO⁺

\tilde{X} C _{∞v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	UN stretch	1118.6	Ne	IR	1
	3	UO stretch	969.8	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Chem. Phys.* **111**, 11044 (1999).

BCC⁻

$\tilde{A}^1\Pi$ C _{∞v}						
T ₀ =23131(11)		Ne AB ¹			$\tilde{A}-\tilde{X}$ 361–432 nm	
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1		1535(11)	Ne	AB	1
	3		1081(11)	Ne	AB	1,2

$\tilde{X}^1\Sigma^+$ C _{∞v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	CC stretch	1936	Ne	IR	1

References

- ¹M. Wyss, M. Grutter, and J. P. Maier, *J. Phys. Chem. A* **102**, 9106 (1998).
²C. Léonard, D. Panten, P. Rosmus, M. Wyss, and J. P. Maier, *Chem. Phys.* **264**, 267 (2001).

BNB⁻

Threshold for electron detachment from ground-state BNB⁻ = 25000(40) gas PE^{3,4}

$\tilde{X}^1\Sigma_g^+$ D _{∞h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	1	Sym. stretch	1130(40)	gas	PE	4
Σ _u ⁺	3	Asym. stretch	1728(40) 1736.5	gas Ar	PE IR	4 1,2

References

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³K. R. Asmis, T. R. Taylor, and D. M. Neumark, *Eur. Phys. J. D* **9**, 257 (1999).
⁴K. R. Asmis, T. R. Taylor, and D. M. Neumark, *J. Chem. Phys.* **111**, 8838 (1999).

BCN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CN stretch	2170.8	Ar	IR	1
	3	BC stretch	800.0	Ar	IR	1

Reference

¹D. V. Lanzisera, L. Andrews, and P. R. Taylor, *J. Phys. Chem. A* **101**, 7134 (1997).

BNC

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NC stretch	2055.2	Ar	IR	1
	3	BN stretch	984.3	Ar	IR	1

Reference

¹D. V. Lanzisera, L. Andrews, and P. R. Taylor, *J. Phys. Chem. A* **101**, 7134 (1997).

AICC⁻

Threshold for electron detachment from ground-state AICC⁻ = 21380(240) gas PE¹

Reference

¹A. I. Boldyrev, J. Simons, X. Li, and L.-S. Wang, *J. Am. Chem. Soc.* **121**, 10193 (1999).

AICSi⁻

Threshold for electron detachment from ground-state AICSi⁻ = 20170(480) gas PE¹

Reference

¹A. I. Boldyrev, J. Simons, X. Li, and L.-S. Wang, *J. Am. Chem. Soc.* **121**, 10193 (1999).

AICN

$\tilde{A}^1\Pi$ C_s
 $T_0 = 28755.3(3)$ gas $LF^{2,3}$ $\tilde{A}-\tilde{X}$ 325–405 nm
 $B_0 = 0.181(4)$ LF^3

$\tilde{X}^1\Sigma^+$ $C_{\infty v}$						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ^+	1	CN stretch	1974.5(1.5)	gas	LF	2,3
			2143.7	Ar	IR	1
Π	2	Bend	131.9(1.3)	gas	LF	2,3
Σ^+	3	AIC stretch	523.5(0.7)	gas	LF	2,3

$B_0 = 0.168$ LF^3MW^4

References

- ¹D. V. Lanzisera and L. Andrews, J. Phys. Chem. A **101**, 9660 (1997).
²M. Fukushima, Chem. Phys. Lett. **283**, 337 (1998).
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AINC

$\tilde{A}^1\Pi$ $C_{\infty v}$
 $T_0 = 36389.1(2)$ gas LF^3 $\tilde{A}-\tilde{X}$ 265–279 nm

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Π	2	Bend (ω)	238	gas	LF	3
		($\kappa^1\Sigma^+$)	371	gas	LF	3
Σ^+	3	NC stretch	598	gas	LF	3

$\epsilon = +0.53$ LF^3
 $B_0 = 0.210$ LF^3

$\tilde{X}^1\Sigma^+$ $C_{\infty v}$						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ^+	1	NC stretch	2069(21)	gas	LF	3
			2051.4	Ar	IR	2
Π	2	Bend	100(6)	gas	LF	3
Σ^+	3	AIN stretch	557	gas	LF	3

$B_0 = 0.200$ MW^1

References

- ¹J. S. Robinson, A. J. Apponi, and L. M. Ziurys, Chem. Phys. Lett. **278**, 1 (1997).
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GaCN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
		CN stretch	2137.8	Ar	IR	1

Reference

- ¹D. V. Lanzisera and L. Andrews, J. Phys. Chem. A **101**, 9660 (1997).

GaNC

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
		NC stretch	2046.4	Ar	IR	1

Reference

- ¹D. V. Lanzisera and L. Andrews, J. Phys. Chem. A **101**, 9660 (1997).

InCN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
		CN stretch	2132.4	Ar	IR	1

Reference

- ¹D. V. Lanzisera and L. Andrews, J. Phys. Chem. A **101**, 9660 (1997).

InNC

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
		NC stretch	2051.1	Ar	IR	1

Reference

- ¹D. V. Lanzisera and L. Andrews, J. Phys. Chem. A **101**, 9660 (1997).

BNSi

\tilde{X} $C_{\infty v}$						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ^+	1	BN stretch	1528.0	N_2	IR	1

Reference

- ¹G. Meloni, S. Nunzianta Cesaro, and N. Sanna, Chem. Phys. Lett. **343**, 113 (2001).

Al₂P⁻

Threshold for electron detachment from ground-state Al₂P⁻ = 20280(160) gas PE¹

\tilde{X}^1A_1		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	450(20)	gas	PE	1

Reference

¹H. Gómez, T. R. Taylor, and D. M. Neumark, J. Phys. Chem. A **105**, 6886 (2001).

Ga₂P⁻

Threshold for electron detachment from ground-state Ga₂P⁻ = 20020(120) gas PE^{1,2}

\tilde{X}^1A_1		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	375(25)	gas	PE	1,2

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¹T. R. Taylor, K. R. Asmis, H. Gomez, and D. M. Neumark, Eur. Phys. J. D **9**, 317 (1999).
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In₂P⁻

Threshold for electron detachment from ground-state In₂P⁻ = 19360(8) gas PE^{1,3}TPE²

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	295T	gas	TPE	2
	2	Bend	100T	gas	TPE	2
b ₂	3	Asym. stretch	223T	gas	TPE	2

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¹C. Xu, E. de Beer, D. W. Arnold, C. C. Arnold, and D. M. Neumark, J. Chem. Phys. **101**, 5406 (1994).
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³K. R. Asmis, T. R. Taylor, and D. M. Neumark, Chem. Phys. Lett. **308**, 347 (1999).

Ga₂As⁻

Threshold for electron detachment from ground-state Ga₂As⁻ = 19820(120) gas PE¹

\tilde{X}^1A_1		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1		245	gas	PE	1

Reference

¹T. R. Taylor, H. Gómez, K. R. Asmis, and D. M. Neumark, J. Chem. Phys. **115**, 4620 (2001).

TICN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CN stretch	2125.7	Ar	IR	1

Reference

¹D. V. Lanzisera and L. Andrews, J. Phys. Chem. A **101**, 9660 (1997).

TINC

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NC stretch	2047.7	Ar	IR	1

Reference

¹D. V. Lanzisera and L. Andrews, J. Phys. Chem. A **101**, 9660 (1997).

Ga₂O

\tilde{C}
 T₀ = 36939(15) Ar AB⁶ $\tilde{C}-\tilde{X}$ 265–271 nm
 36226(15) Kr AB⁶ $\tilde{C}-\tilde{X}$ 270–276 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	GaO s-stretch	527	Ar	AB	6
	2	Bend	182	Ar	AB	6
			218	Kr	AB	6

\tilde{B}
 T₀ = 28328(15) Kr EM⁶ $\tilde{B}-\tilde{X}$ 353–403 nm

\tilde{A}
 T₀ = 20600(15) Kr EM⁶

\tilde{X}		D _{∞h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _u ⁺	3	Asym. stretch	822.3	Ar	IR	2,4,5,7
			809.4	N ₂	IR	1,3–5

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- T. R. Burkholder, J. T. Yustein, and L. Andrews, J. Phys. Chem. **96**, 10189 (1992).

In₂O

\tilde{C}	$T_0 = 32699(15)$ Ar AB ⁶	$\tilde{C}-\tilde{X}$ 300–306 nm
	31854(15) Kr AB ⁶	$\tilde{C}-\tilde{X}$ 306–314 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	Sym. stretch	576	Ar	AB	6
	2	Bend	139	Kr	AB	6

\tilde{B}_2	C _{2v}	$T_0 = 26524.53$ gas LF ⁸	$\tilde{B}_2-\tilde{X}$ 373–377 nm
		$\tau = 1.350(25)$ μ s gas LF ⁸	

\tilde{B}_1	C _{2v}	$T_0 = 25815$ gas LF ⁸	$\tilde{B}_1-\tilde{X}$ 380–388 nm
		25644(15) Kr AB ⁶	
		25535(15) Kr EM ⁶	

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	InO s-stretch	219.2	gas	AB	8
	2	Bend	94T	gas	AB	8

$\tau = 1.74(4)$ μ s gas LF⁸

\tilde{A}	$T_0 = 19260(15)$ Kr EM ⁶
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Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	InO s-stretch	222	gas	LF	8
Π_u	2	Bend	52H	gas	LF	8
Σ_u^+	3	InO a-stretch	745.4	Ne	IR	2
			735.2	Ar	IR	2,4,5,7
			728.2	Kr	IR	2
			722.4	N ₂	IR	1–5

References

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- A. J. Hinchcliffe and J. S. Ogden, J. Phys. Chem. **75**, 3908 (1971).
- A. J. Hinchcliffe and J. S. Ogden, J. Phys. Chem. **77**, 2537 (1973).
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- T. R. Burkholder, J. T. Yustein, and L. Andrews, J. Phys. Chem. **96**, 10189 (1992).
- N. M. Lakin, G. van den Hoek, I. R. Beattie, and J. M. Brown, J. Chem. Phys. **107**, 4439 (1997).

Ti₂O

Vib.	No.	Approximate type of mode	cm ⁻¹	Med.	Type	Refs.
	3	Asym. stretch	643.6	Ar	IR	2,4
			634.6	Kr	IR	2
			625.3	N ₂	IR	1–3

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C₃

${}^1\Sigma_u^+$	D _{∞h}	$T_0 = 52826(30)$ Ar AB ¹⁵	170–190 nm
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Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1080(30)	Ar	AB	15
Π_u	2	Bend	300(30)H	Ar	AB	15
Σ_u^+	3	Asym. stretch	780(30)H ^a	Ar	AB	15

$\tilde{B}^1\Delta_u$	D _{∞h}
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In 2-photon ionization studies of jet-cooled C₃, a complicated group of bands, all with rotational structure appropriate for a $\Sigma_u^+ - \Sigma_g^+$ vibronic transition arising from the \tilde{X} state, has been observed between 266 and 302 nm.²⁰ Lifetimes of these bands range from 0.4 to 2.5 μ s. These same bands, as well as bands at somewhat lower energies (possibly below the ionization threshold) and bands arising from a $\Sigma_u^+ - \Pi_g$ vibronic transition, have also been studied using LF measurements on cooled beams.^{23,47} The average B' values for the Σ and Π vibronic levels are 0.395(14) and 0.398(17), respectively.⁴⁷ In a neon matrix,^{8,39} absorptions attributable to the $\tilde{B}-\tilde{X}$ transition of C₃ are observed between 281 and 302 nm, with the first intense band at 33149(20), and in an argon matrix,³⁰ absorptions arising from this transition of C₃ are observed between 280 and 284 nm.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	940(60)	gas	LF	47
Π_u	2	Bend	222(20)H	Ne	AB	39
			195H	Ar	AB	30

In an argon matrix,³⁰ bands attributable to C₃ have been observed between 367 and 373 nm.

$\tilde{A}^1\Pi_u$	$D_{\infty h}$	Structure: UV ⁶
$T_0 = 24675.5$	gas	EM ^{1,2,6} AB ^{3,6,9} LF ^{17,23,33,35,36,42,43,46}
24640	Ne	AB ^{4,5,8} EM ⁵ LF ¹¹
24370 ^b	Ar	AB ^{4,5,7} LF ^{11,45}
24350	Kr	AB ⁷
23610	Xe	AB ^{4,7}
24635	N ₂	AB ⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1085.9	gas	AB,LF	6,33
			1094(6)	Ne	AB	5
			1093(6)	Ar	AB	5,7
			1090	Kr	AB	7
			1120	Xe	AB	7
			1050	N ₂	AB	7
Π_u	2	Bend	307.9 ^c	gas	AB	6
Σ_u^+	3	Asym. stretch	541.7 ^d	gas	LF	36,46

$\tau_0 = 200(10)$ ns gas LF^{12,13}
 In a neon or argon matrix,^{11,45} efficient intersystem crossing into the $\tilde{a}^3\Pi_u$ state occurs, and $\tau \leq 10$ ns.
 $B_0 = 0.430$ UV⁶

$\tilde{b}^3\Pi_g$	$D_{\infty h}$	Structure: DL,EM ²⁹
$T_0 = 23570(210)$	gas	DL ²⁹ EM ^{29,37}
23584 ^e	Ne	IR ³²
		$\tilde{b}-\tilde{a}$ 1530–1640 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π_u	2	Bend	345(40) ^f	gas	EM	37

$A = 14.77(24)$ EM³⁷
 $B_0 = 0.424$ DL²⁹EM²⁹

$\tilde{a}^3\Pi_u$	$D_{\infty h}$	Structure: DL,EM ²⁹
$T_0 = 17090(210)$	gas	DL ²⁹ EM ^{29,37} PE ³¹
17080	Ne	EM ⁵ LF ¹¹
		IR ³²
16930	Ar	EM ^{5,34} LF ⁴⁵
		IR ³²
		$\tilde{b}-\tilde{a}$ 1530–1640 nm
		$\tilde{a}-\tilde{X}$ 585–631 nm
		$\tilde{b}-\tilde{a}$ 1250–1538 nm
		$\tilde{a}-\tilde{X}$ 590–857 nm
		$\tilde{b}-\tilde{a}$ 1253–1544 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1156.5 ^g	Ne	IR	32
			1154.2 ^g	Ar	IR	32
Π_u	2	Bend	505(40) ^h	gas	EM	37
Σ_u^+	3	Asym. stretch	1449.53	gas	DL	38
			1454.1	Ne	IR	32
			1455.3	Ar	IR	32

$\tau \cong 0.02$ s Ne EM⁵
 $\cong 0.01$ s Ar LF⁴⁵
 $A = 15.16(4)$ EM³⁷
 $B_0 = 0.417$ DL^{29,38}EM²⁹

$\tilde{X}^1\Sigma_g^+$ $D_{\infty h}$ Structure: UV⁶
 This state of C₃ is highly anharmonic. The term values of many excited vibrational energy levels have been determined in SEP^{21,26,27} and LF²³ studies. Analysis of data from the SEP studies²⁸ using the semirigid bender model indicates that the molecule is linear in its ground state but that on excitation of ν_3 a potential barrier appears at the linear configuration.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1224.5	gas	AB,LF	9,17,23,33
			1226	Ne	EM	5
			1216.4(6)	Ar	IR,EM,LF	19,34,45
			1215.8T	Kr	IR	41
			1234 ^g	N ₂	IR	19
Π_u	2	Bend	63.41 ⁱ	gas	UV,DL	6,22,23
					SEP,TF	24,25
			70T ^j	Ne	AB	5
			73.0(3)H	Ar	LF	45
Σ_u^+	3	Asym. stretch	2040.02	gas	IR,DL,LF	16,18,23
			2042	Ne	IR	4
			2038.9s	Ar	IR	4,10,45
			2033.3	Kr	IR	41
			2036.0	H ₂	IR	40,44
			2031	N ₂	IR	19

$B_0 = 0.431$ UV⁶IR¹⁶DL¹⁸TF²⁵

^aAlternate assignment gives 1320.

^bIn the LF studies,^{11,45} a second site was observed with $T_0 = 24408$.

^c ω . Large Renner splitting, with $\epsilon = 0.537$.⁶ Detailed comparisons of gas-phase with neon- and argon-matrix band positions are given in Refs. 8 and 14. Reference 14 also gives a more detailed analysis of electronic orbital angular momentum effects in the gas-phase molecule.

^dIn the gas phase,³⁵ the $\tilde{A}(002) - \tilde{X}(000)$ transition appears at 26348.0, giving $2\nu_3 = 1672.5$, and the corresponding transition observed in a neon matrix⁸ yields $2\nu_3 = 1680$. Reference 46 reports a barrier of 284.3 in the double-minimum potential for the antisymmetric stretching fundamental of C₃ $\tilde{A}^1\Pi_u$.

^eSum of values for $\tilde{a}-\tilde{X}$ and $\tilde{b}-\tilde{a}$ transitions.

^f ω . Large Renner splitting, with $\epsilon = 0.447$.³⁷

^g $(\nu_1 + \nu_3) - \nu_3$.

^h ω . Large Renner splitting, with $\epsilon = 0.566$.³⁷

ⁱMost precise value with tunable far infrared laser spectrometer (TF).²⁵ $\cong 45$ in $\tilde{X}(011)$.²²

^jGreatly broadened in a rare-gas matrix by interaction with lattice modes.¹¹

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Si₃

\tilde{G} C_{2v} $\tilde{G}-\tilde{X}$ 330–390 nm
 $T_0=25753(13)T$ Ne AB⁶

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1			350T	Ne	AB	6

\tilde{F} C_{2v} $\tilde{F}-\tilde{X}$ 430–540 nm
 $T_0=19146(7)$ Ne AB⁶

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	360T	Ne	AB	6

\tilde{E} C_{2v} $\tilde{E}-X$ 530–580 nm
 $T_0=17250(6)$ Ne AB^{1,6}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	Bend	250T	Ne	AB	6

$\tilde{C}^1A_1^a$ C_{2v} $\tilde{C}-\tilde{X}$ 700–780 nm
 $T_0=14200(160)$ gas PE^{2,7}
 12839(3) Ne AB⁶

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	480(40) 470T	gas Ne	PE AB	2 6

$\tilde{B}^1B_1^b$ C_{2v} $\tilde{B}-\tilde{X}$ 700–780 nm
 $T_0=11549(160)$ gas PE⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	480(20)	gas	PE	7
	2	Bend	260(20)	gas	PE	7

$\tilde{c}^3B_1^c$ C_{2v} $\tilde{c}-\tilde{X}$ 700–780 nm
 $T_0=8880(80)$ gas PE^{2,7}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	480(20)	gas	PE	7
	2	Bend	260(20)	gas	PE	7

$\tilde{b}^3A_1^d$ C_{2v} $\tilde{b}-\tilde{X}$ 700–780 nm
 $T_0=7180(80)$ gas PE^{2,7}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	500(20)	gas	PE	7

\tilde{A}^1B_2 C_{2v} $\tilde{A}-\tilde{X}$ 700–780 nm
 $T_0=3630(80)$ gas PE^{2,7}EM³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	480(25)	gas	PE,EM	2,3

\tilde{a}^3A_2' D_{3h} $\tilde{a}-\tilde{X}$ 700–780 nm
 $T_0<350$ gas TPE⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1'	1	Sym. stretch	501(10)	gas	TPE	4
e'	2	Deformation	337(10)	gas	EM TPE	2–4

\tilde{X}^1A_1		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	550T	Ne	IR	5
			550.6	Ar	IR	5
			546.7	Kr	IR	5
b_2	3	Asym. stretch	525.5	Ne	IR	5
			525.1	Ar	IR	5
			523.1	Kr	IR	5

^aPhotoelectron band E (Ref. 7).

^bPhotoelectron band D (Ref. 7).

^cPhotoelectron band C (Ref. 7).

^dPhotoelectron band B (Ref. 7).

References

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- C. Xu, T. R. Taylor, G. R. Burton, and D. M. Neumark, *J. Chem. Phys.* **108**, 1395 (1998).

Sn₃

\tilde{D}
 $T_0=5890(800)$ gas PE¹

\tilde{C}
 $T_0=5400(80)$ gas PE¹

\tilde{B}
 $T_0=2980(160)$ gas PE¹

\tilde{A}
 $T_0=890(110)$ gas PE¹

Reference

- V. D. Moravec, S. A. Klopčič, and C. C. Jarrold, *J. Chem. Phys.* **110**, 5079 (1999).

CCN⁺

$\tilde{A}^1\Pi$ $C_{\infty v}$
 $T_0=21637(10)$ Ne AB¹ $\tilde{A}-\tilde{X}$ 427–462 nm

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ^+	1	“Asym.” stretch	1725(20)	Ne	AB	1
Σ^+	3	“Sym.” stretch	1223(20)	Ne	AB	1

Reference

- M. Wyss, E. Riaplov, J. P. Maier, M. Hochlaf, and P. Rosmus, *Helv. Chim. Acta* **84**, 1432 (2001).

CNC⁺

$\tilde{A}^1\Pi_u$ $D_{\infty h}$
 $T_0=30694(20)$ Ne AB² $\tilde{A}-\tilde{X}$ 301–326 nm

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1100T	Ne	AB	2

$\tilde{X}^1\Sigma_g^+$		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ_u^+	3	Asym. stretch	1974.07 1972.9	gas Ne	DL IR	1 2,3

$B_0=0.461$ DL¹

References

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ScCO⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
	1	CO stretch	1732.0	Ne	IR	1

Reference

- M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 2964 (1999).

TiCO⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
	1	CO stretch	1789.9	Ne	IR	1

Reference

- M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 5259 (1999).

VCO⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
	1	CO stretch	1806.7	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 5259 (1999).

CNbO⁻
 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NbO stretch	877.8	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7785 (1999).

CTaO⁻
 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		TaO stretch	902.0	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7785 (1999).

TaCO⁻
 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1700.9T	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7785 (1999).

MnCO⁻
 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1807.5	Ne	IR	1
			1789.4	Ar	IR	1

Reference

¹L. Andrews, M. Zhou, X. Wang, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **104**, 8887 (2000).

ReCO⁻

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1728.0	Ne	IR	1
			1704.2	Ar	IR	1

Reference

¹L. Andrews, M. Zhou, X. Wang, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **104**, 8887 (2000).

FeCO⁻

Threshold for electron detachment from ground-state FeCO⁻ = 9335(40) gas PE^{1,2}

 $\tilde{X}^4\Sigma^-$ C_{∞v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	CO stretch	1980(100)T	gas	PE	1
			1782.0	Ne	IR	4
			1770.3	Ar	IR	3
Π	2	Bend	230(40)	gas	PE	2
Σ ⁺	3	FeC stretch	465(10)	gas	PE	2

References

¹P. C. Engelking and W. C. Lineberger, *J. Am. Chem. Soc.* **101**, 5569 (1979).

²P. W. Villalta and D. G. Leopold, *J. Chem. Phys.* **98**, 7730 (1993).

³M. Zhou, G. V. Chertihin, and L. Andrews, *J. Chem. Phys.* **109**, 10893 (1998).

⁴M. Zhou and L. Andrews, *J. Chem. Phys.* **110**, 10370 (1999).

RuCO⁻
 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1792.8	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 6956 (1999).

OsCO⁻
 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1785.5	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 6956 (1999).

CoCO⁻

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	1820.2	Ne	IR	2
			1804.0	Ar	IR	1

References

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **102**, 10250 (1998).

²M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7773 (1999).

RhCO⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1828.6	Ne	IR	1,2
			1813.7	Ar	IR	2

References

¹M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **121**, 9171 (1999).

²M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7773 (1999).

IrCO⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1842.6	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7773 (1999).

NiCO⁻

Threshold for electron detachment from ground-state NiCO⁻ = 6490(100) gas PE¹

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	1860.6	Ne	IR	3
			1850.1	Ar	IR	2
			1847.0			

References

¹A. E. Stevens, C. S. Feigerle, and W. C. Lineberger, *J. Am. Chem. Soc.* **104**, 5026 (1982).

²M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **120**, 11499 (1998).

³B. Liang, M. Zhou, and L. Andrews, *J. Phys. Chem. A* **104**, 3905 (2000).

PdCO⁻

Threshold for electron detachment from ground-state PdCO⁻ = 4870(80) gas PE¹

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1909.0	Ne	IR	2

References

¹S. A. Klopčič, V. D. Moravec, and C. C. Jarrold, *J. Chem. Phys.* **110**, 8986 (1999).

²B. Liang, M. Zhou, and L. Andrews, *J. Phys. Chem. A* **104**, 3905 (2000).

PtCO⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1896.3	Ne	IR	1

Reference

¹B. Liang, M. Zhou, and L. Andrews, *J. Phys. Chem. A* **104**, 3905 (2000).

CuCO⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1746.2	Ne	IR	1
			1733.4	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Chem. Phys.* **111**, 4548 (1999).

CThO⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		ThO stretch	761.7	Ne	IR	1

Reference

¹J. Li, B. E. Bursten, M. Zhou, and L. Andrews, *Inorg. Chem.* **40**, 5448 (2001).

CUO⁻ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	UC stretch	929.3	Ne	IR	1
	3	UO stretch	803.3	Ne	IR	1

Reference

¹M. Zhou, L. Andrews, J. Li, and B. E. Bursten, *J. Am. Chem. Soc.* **121**, 9712 (1999).

UCO⁻ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1689.2	Ne	IR	1

Reference

¹M. Zhou, L. Andrews, J. Li, and B. E. Bursten, *J. Am. Chem. Soc.* **121**, 9712 (1999).

PtNN⁻ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NN stretch	2054.1	Ne	IR	1
			2045.8	Ar	IR	1
			2048.9	N ₂	IR	1

Reference

¹A. Citra, X. Wang, W. D. Bare, and L. Andrews, *J. Phys. Chem. A* **105**, 7799 (2001).

BeNO

 \tilde{X} C_{∞v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	BeNO s-stretch	913.8	Ar	IR	1

Reference

¹G. P. Kushto, F. Ding, B. Liang, X. Wang, A. Citra, and L. Andrews, *Chem. Phys.* **257**, 223 (2000).

MgNO

 \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	NO stretch	1461.3	Ar	IR	1

Reference

¹G. P. Kushto, F. Ding, B. Liang, X. Wang, A. Citra, and L. Andrews, *Chem. Phys.* **257**, 223 (2000).

NScO

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	ScO stretch	931.1	Ne	IR	2
			909.5	Ar	IR	1
	3	ScN stretch	480.8	Ne	IR	2
			471.1	Ar	IR	1

References

¹G. P. Kushto, M. Zhou, L. Andrews, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **103**, 1115 (1999).

²L. Andrews and X. Wang, *J. Phys. Chem. A* **106**, 1196 (2002).

ScNO

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1589.4	Ne	IR	2
			1563.3	Ar	IR	1

References

¹G. P. Kushto, M. Zhou, L. Andrews, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **103**, 1115 (1999).

²L. Andrews and X. Wang, *J. Phys. Chem. A* **106**, 1196 (2002).

cyc-ScNO

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>			860.8	Ne	IR	2
			865.5	Ar	IR	1
			644.4	Ne	IR	2
			644.1	Ar	IR	1

References

¹G. P. Kushto, M. Zhou, L. Andrews, and C. W. Bauschlicher, Jr., J. Phys. Chem. A **103**, 1115 (1999).

²L. Andrews and X. Wang, J. Phys. Chem. A **106**, 1196 (2002).

NTiO

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	TiO stretch	915.0	Ne	IR	3
			900.7s	Ar	IR	1,2
	2	Bend	248.2wm	Ar	IR	2
			732.0	Ne	IR	3
	3	TiN stretch	718.3s	Ar	IR	1,2

References

¹G. P. Kushto, M. Zhou, L. Andrews, and C. W. Bauschlicher, Jr., J. Phys. Chem. A **103**, 1115 (1999).

²L. Krim, C. Prot, E. M. Alikhani, and L. Manceron, Chem. Phys. **254**, 267 (2000).

³L. Andrews and X. Wang, J. Phys. Chem. A **106**, 1196 (2002).

TiNO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1614.7	Ar	IR	1

Reference

¹G. P. Kushto, M. Zhou, L. Andrews, and C. W. Bauschlicher, Jr., J. Phys. Chem. A **103**, 1115 (1999).

NZrO

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	ZrO stretch	861.3	Ne	IR	2
			844.2	Ar	IR	1
	3	ZrN stretch	687.9	Ne	IR	2
			673.3	Ar	IR	1

References

¹G. P. Kushto and L. Andrews, J. Phys. Chem. A **103**, 4836 (1999).

²L. Andrews and X. Wang, J. Phys. Chem. A **106**, 1196 (2002).

NHfO

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	HfO stretch	855.2	Ar	IR	1
	3	HfN stretch	685.3	Ar	IR	1

Reference

¹G. P. Kushto and L. Andrews, J. Phys. Chem. A **103**, 4836 (1999).

NVO

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	VN stretch	1007.0	Ne	IR	2
			998.1	Ar	IR	1
	3	VO stretch	912.0	Ne	IR	2
			906.4	Ar	IR	1

References

¹M. Zhou and L. Andrews, J. Phys. Chem. A **103**, 478 (1999).

²L. Andrews and X. Wang, J. Phys. Chem. A **106**, 1196 (2002).

VNO

\tilde{X} $C_{\infty v}$						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	NO stretch	1606.0	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, J. Phys. Chem. A **103**, 478 (1999).

cyc-VNO

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	NO stretch	1069.8	Ne	IR	2
			1075.7	Ar	IR	1

References

- ¹M. Zhou and L. Andrews, J. Phys. Chem. A **103**, 478 (1999).
²L. Andrews and X. Wang, J. Phys. Chem. A **106**, 1196 (2002).

NNbO

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	NbN stretch	977.3	Ar	IR	1
			975.3			
	3	NbO stretch	852.6	Ar	IR	1
			850.6			

Reference

- ¹M. Zhou and L. Andrews, J. Phys. Chem. A **102**, 10025 (1998).

NTaO

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	TaN stretch	967.6	Ar	IR	1
			965.9			
	3	TaO stretch	857.9	Ar	IR	1
			855.8			

Reference

- ¹M. Zhou and L. Andrews, J. Phys. Chem. A **102**, 10025 (1998).

NCrO

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CrN stretch	969.7	Ne	IR	2
			976.1	Ar	IR	1
	3	CrO stretch	869.1	Ne	IR	2
			866.2	Ar	IR	1

References

- ¹M. Zhou and L. Andrews, J. Phys. Chem. A **102**, 7452 (1998).
²L. Andrews and X. Wang, J. Phys. Chem. A **106**, 1196 (2002).

CrNO

\tilde{X} C _{∞v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	NO stretch	1637.5	Ne	IR	2
			1609.1	Ar	IR	1
	3	CrN stretch	533.3	Ar	IR	1

References

- ¹M. Zhou and L. Andrews, J. Phys. Chem. A **102**, 7452 (1998).
²L. Andrews and X. Wang, J. Phys. Chem. A **106**, 1196 (2002).

cyc-CrNO

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	NO stretch	1123.9	Ne	IR	2
			1108.8	Ar	IR	1
	2	CrN stretch	528.2vw	Ar	IR	1
	3	CrO stretch	478.0vw	Ar	IR	1

References

- ¹M. Zhou and L. Andrews, J. Phys. Chem. A **102**, 7452 (1998).
²L. Andrews and X. Wang, J. Phys. Chem. A **106**, 1196 (2002).

NMoO

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	MoN stretch	999.1	Ar	IR	1
	3	MoO stretch	846.5	Ar	IR	1

Reference

- ¹L. Andrews and M. Zhou, J. Phys. Chem. A. **103**, 4167 (1999).

MoNO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1620.6	Ar	IR	1

Reference

- ¹L. Andrews and M. Zhou, J. Phys. Chem. A. **103**, 4167 (1999).

NWO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	WN stretch	1016.3	Ar	IR	1
	3	WO stretch	907.3	Ar	IR	1

Reference

¹L. Andrews and M. Zhou, J. Phys. Chem. A. **103**, 4167 (1999).

NMnO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	MnN stretch	922.6	Ne	IR	2
	3	MnO stretch	932.3	Ar	IR	1
			872.7	Ne	IR	2
			874.0	Ar	IR	1

References

¹L. Andrews, M. Zhou, and D. W. Ball, J. Phys. Chem. A **102**, 10041 (1998).

²L. Andrews and X. Wang, J. Phys. Chem. A **106**, 1196 (2002).

MnNO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
		NO stretch	1754.5	Ne	IR	2
			1748.6T	Ar	IR	1

References

¹L. Andrews, M. Zhou, and D. W. Ball, J. Phys. Chem. A **102**, 10041 (1998).

²L. Andrews and X. Wang, J. Phys. Chem. A **106**, 1196 (2002).

cyc-MnNO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	NO stretch	1235.7	Ne	IR	2
			1236.8	Ar	IR	1
			1232.4			

References

¹L. Andrews, M. Zhou, and D. W. Ball, J. Phys. Chem. A **102**, 10041 (1998).

²L. Andrews and X. Wang, J. Phys. Chem. A **106**, 1196 (2002).

NReO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	ReN stretch	1052.9	Ar	IR	1
	3	ReO stretch	901.2	Ar	IR	1

Reference

¹L. Andrews, M. Zhou, and D. W. Ball, J. Phys. Chem. A **102**, 10041 (1998).

cyc-ReNO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	NO stretch	1137.2	Ar	IR	1

Reference

¹L. Andrews, M. Zhou, and D. W. Ball, J. Phys. Chem. A **102**, 10041 (1998).

FeNO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
		NO stretch	1766.0	Ne	IR	3
			1748.7	Ar	IR	1,3
			1746.8	N ₂	IR	2
			1731.0			

References

¹D. W. Ball and J. A. Chiarelli, J. Mol. Struct. **372**, 113 (1995).

²L. Andrews, G. V. Chertihin, A. Citra, and M. Neurock, J. Phys. Chem. **100**, 11235 (1996).

³M. Zhou and L. Andrews, J. Phys. Chem. A **104**, 3915 (2000).

cyc-FeNO

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	NO stretch	1342.2	Ne	IR	1
			1343.8	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **104**, 3915 (2000).

RuNO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1785.8	Ne	IR	1
			1765.9	Ar	IR	1

Reference

¹A. Citra and L. Andrews, *J. Phys. Chem. A* **104**, 8689 (2000).

NRuO

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	RuN stretch	991.5	Ne	IR	1
			991.8	Ar	IR	1
	3	RuO stretch	800.4	Ne	IR	1
			803.0	Ar	IR	1

Reference

¹A. Citra and L. Andrews, *J. Phys. Chem. A* **104**, 8689 (2000).

OsNO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1809.4	Ne	IR	1
			1789.1	Ar	IR	1

Reference

¹A. Citra and L. Andrews, *J. Phys. Chem. A* **104**, 8689 (2000).

NOsO

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	OsN stretch	1057.6	Ne	IR	1
			1052.0	Ar	IR	1
	3	OsO stretch	892.0	Ne	IR	1
			886.9	Ar	IR	1

Reference

¹A. Citra and L. Andrews, *J. Phys. Chem. A* **104**, 8689 (2000).

cyc-OsNO

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	NO stretch	1168.0	Ne	IR	1
			1132.6	Ar	IR	1

Reference

¹A. Citra and L. Andrews, *J. Phys. Chem. A* **104**, 8689 (2000).

CoNO

\tilde{X} $C_{\infty v}$						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	NO stretch	1794.2	Ne	IR	3
			1761.0	Ar	IR	1,3
			1771.4	N ₂	IR	2
	3	CoN stretch	620.1	Ar	IR	3

References

¹G. K. Ruschel, T. M. Nemetz, and D. W. Ball, *J. Mol. Struct.* **384**, 101 (1996).

²G. V. Chertihin, A. Citra, L. Andrews, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **101**, 8793 (1997).

³M. Zhou and L. Andrews, *J. Phys. Chem. A* **104**, 3915 (2000).

cyc-CoNO

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	NO stretch	1317.4	Ne	IR	1
			1284.2	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, J. Phys. Chem. A **104**, 3915 (2000).

RhNO

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1806.4	Ne	IR	2
			1775.1	Ar	IR	1
			1765.9	N ₂	IR	1

References

¹A. Citra and L. Andrews, J. Phys. Chem. A **103**, 3410 (1999).
²A. Citra and L. Andrews, J. Phys. Chem. A **104**, 11897 (2000).

NRhO

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	RhN stretch	967.8	Ne	IR	1
			964.2	Ar	IR	1
	3	RhO stretch	813.7	Ar	IR	1

Reference

¹A. Citra and L. Andrews, J. Phys. Chem. A **104**, 11897 (2000).

IrNO

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1851.1	Ne	IR	1
			1833.7	Ar	IR	1

Reference

¹A. Citra and L. Andrews, J. Phys. Chem. A **104**, 11897 (2000).

NiRO

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	IrN stretch	994.6	Ne	IR	1
			977.3	Ar	IR	1
	3	IrO stretch	868.9	Ne	IR	1
			850.6	Ar	IR	1

Reference

¹A. Citra and L. Andrews, J. Phys. Chem. A **104**, 11897 (2000).

NiNO

 \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	NO stretch	1680.1	Ne	IR	3
			1677.1	Ar	IR	1-3
	3	NiN stretch	608.4	Ar	IR	2,3

References

¹G. K. Ruschel, T. M. Nemetz, and D. W. Ball, J. Mol. Struct. **384**, 101 (1996).
²L. Krim, L. Manceron, and M. E. Alikhani, J. Phys. Chem. A **103**, 2592 (1999).
³M. Zhou and L. Andrews, J. Phys. Chem. A **104**, 3915 (2000).

cyc-NiNO

 \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	NO stretch	1292.6	Ne	IR	2
			1293.9	Ar	IR	1,2
	2		540.5	Ar	IR	1
	3		464.4	Ar	IR	1

References

¹L. Krim, L. Manceron, and M. E. Alikhani, J. Phys. Chem. A **103**, 2592 (1999).
²M. Zhou and L. Andrews, J. Phys. Chem. A **104**, 3915 (2000).

PdNO

 \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	NO stretch	1676.4	Ne	IR	1
			1661.3s	Ar	IR	1,2
	2	Bend	229.8vw	Ar	IR	2
	3	PdN stretch	522.1w	Ar	IR	2

References

¹A. Citra and L. Andrews, J. Phys. Chem. A **104**, 8160 (2000).
²L. Krim, E. M. Alikhani, and L. Manceron, J. Phys. Chem. A **105**, 7812 (2001).

PtNO \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1712.6 1677.0	Ne Ar	IR IR	1 1

Reference¹A. Citra and L. Andrews, J. Phys. Chem. A **104**, 8160 (2000).**CuNO** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1602.2 1587.1	Ne Ar	IR IR	1 1

Reference¹M. Zhou and L. Andrews, J. Phys. Chem. A **104**, 2618 (2000).**AgNO** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1711.8 1707.3 1680.3	Ne Ar	IR IR	1 1

Reference¹A. Citra and L. Andrews, J. Phys. Chem. A **105**, 3042 (2001).**NCeO** \tilde{X} $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CeO stretch	757.2s	Ar	IR	1
Σ^+	3	CeN stretch	690.3w	Ar	IR	1

Reference¹S. P. Willson, L. Andrews, and M. Neurock, J. Phys. Chem. A **104**, 3446 (2000).**CeNO** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1341.1	Ar	IR	1

Reference¹S. P. Willson, L. Andrews, and M. Neurock, J. Phys. Chem. A **104**, 3446 (2000).**NPrO** \tilde{X} $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	PrN stretch	900.8s	Ar	IR	1
Σ^+	3	PrO stretch	742.0m	Ar	IR	1

Reference¹S. P. Willson, L. Andrews, and M. Neurock, J. Phys. Chem. A **104**, 3446 (2000).**PrNO** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1359.3	Ar	IR	1

Reference¹S. P. Willson, L. Andrews, and M. Neurock, J. Phys. Chem. A **104**, 3446 (2000).**NNdO** \tilde{X} $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	Mixed	768.7s	Ar	IR	1
Σ^+	3	Mixed	661.4w	Ar	IR	1

Reference¹S. P. Willson, L. Andrews, and M. Neurock, J. Phys. Chem. A **104**, 3446 (2000).

NSmO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Mixed	736.9	Ar	IR	1

Reference

¹S. P. Willson, L. Andrews, and M. Neurock, *J. Phys. Chem. A* **104**, 3446 (2000).

NEuO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		EuO stretch	657.8	Ar	IR	1

Reference

¹S. P. Willson, L. Andrews, and M. Neurock, *J. Phys. Chem. A* **104**, 3446 (2000).

cyc-EuNO

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>			963.8	Ar	IR	1

Reference

¹S. P. Willson, L. Andrews, and M. Neurock, *J. Phys. Chem. A* **104**, 3446 (2000).

NGdO

\tilde{X} C _{∞v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺		GdO stretch	769.8	Ar	IR	1

Reference

¹S. P. Willson, L. Andrews, and M. Neurock, *J. Phys. Chem. A* **104**, 3446 (2000).

cyc-GdNO

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>			997.6	Ar	IR	1

Reference

¹S. P. Willson, L. Andrews, and M. Neurock, *J. Phys. Chem. A* **104**, 3446 (2000).

NTbO

\tilde{X} C _{∞v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	Mixed	770.6	Ar	IR	1
Σ ⁺	3	Mixed	674.2	Ar	IR	1

Reference

¹S. P. Willson, L. Andrews, and M. Neurock, *J. Phys. Chem. A* **104**, 3446 (2000).

cyc-TbNO

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>			1111.4	Ar	IR	1

Reference

¹S. P. Willson, L. Andrews, and M. Neurock, *J. Phys. Chem. A* **104**, 3446 (2000).

NDyO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		DyO stretch	784.9	Ar	IR	1

Reference

¹S. P. Willson, L. Andrews, and M. Neurock, *J. Phys. Chem. A* **104**, 3446 (2000).

cyc-DyNO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'			771.4	Ar	IR	1
			641.1	Ar	IR	1

Reference

¹S. P. Willson, L. Andrews, and M. Neurock, J. Phys. Chem. A **104**, 3446 (2000).

NHoO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
		HoO stretch	724.5	Ar	IR	1

Reference

¹S. P. Willson, L. Andrews, and M. Neurock, J. Phys. Chem. A **104**, 3446 (2000).

cyc-HoNO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'			764.7	Ar	IR	1

Reference

¹S. P. Willson, L. Andrews, and M. Neurock, J. Phys. Chem. A **104**, 3446 (2000).

NErO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
		ErO stretch	790.3	Ar	IR	1

Reference

¹S. P. Willson, L. Andrews, and M. Neurock, J. Phys. Chem. A **104**, 3446 (2000).

cyc-ErNO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'			760.1	Ar	IR	1

Reference

¹S. P. Willson, L. Andrews, and M. Neurock, J. Phys. Chem. A **104**, 3446 (2000).

NTmO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
		TmO stretch	795.0	Ar	IR	1

Reference

¹S. P. Willson, L. Andrews, and M. Neurock, J. Phys. Chem. A **104**, 3446 (2000).

NYbO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
		YbO stretch	782.0	Ar	IR	1

Reference

¹S. P. Willson, L. Andrews, and M. Neurock, J. Phys. Chem. A **104**, 3446 (2000).

NLuO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	LuO stretch	797.2	Ar	IR	1
	3	LuN stretch	425.6	Ar	IR	1

Reference

¹S. P. Willson, L. Andrews, and M. Neurock, J. Phys. Chem. A **104**, 3446 (2000).

NThO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	ThO stretch	784.2	Ne	IR	2
			760.3	Ar	IR	1
	3	ThN stretch	709.8	Ne	IR	2
			697.3	Ar	IR	1

References

- ¹G. P. Kushto and L. Andrews, *J. Phys. Chem. A* **103**, 4836 (1999).
²M. Zhou and L. Andrews, *J. Chem. Phys.* **111**, 11044 (1999).

NUO

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	UN stretch	1004.9	Ne	IR	4
			983.6	Ar	IR	1,2
			938	N ₂	IR	3
3	UO stretch	833.5	Ne	IR	4	
		818.9	Ar	IR	2	
		795	N ₂	IR	3	

References

- ¹D. W. Green and G. T. Reedy, *J. Chem. Phys.* **65**, 2921 (1976).
²G. P. Kushto, P. F. Souter, L. Andrews, and M. Neurock, *J. Chem. Phys.* **106**, 5894 (1997).
³K. Sankaran, K. Sundararajan, and K. S. Viswanathan, *Bull. Mater. Sci.* **22**, 785 (1999).
⁴M. Zhou and L. Andrews, *J. Chem. Phys.* **111**, 11044 (1999).

cyc-ScO₂⁺

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	OO stretch	892.9	Ar	IR	1
	2	OScO s-stretch	641.1	Ar	IR	1
b_2	3	OScO a-stretch	624.8	Ar	IR	1

Reference

- ¹C. W. Bauschlicher, Jr., M. Zhou, L. Andrews, J. R. Tobias Johnson, I. Panas, A. Snis, and B. O. Roos, *J. Phys. Chem. A* **103**, 5463 (1999).

cyc-YO₂⁺

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	OO stretch	816.3	Ar	IR	1
	2	OYO s-stretch	595.3	Ar	IR	1
b_2	3	OYO a-stretch	587.5	Ar	IR	1

Reference

- ¹L. Andrews, M. Zhou, G. V. Chertihin, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **103**, 6525 (1999).

cyc-LaO₂⁺

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	OO stretch	804.0	Ar	IR	1

Reference

- ¹L. Andrews, M. Zhou, G. V. Chertihin, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **103**, 6525 (1999).

OLaO⁺

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_u^-	3	Asym. stretch	689.3	Ar	IR	1

Reference

- ¹L. Andrews, M. Zhou, G. V. Chertihin, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **103**, 6525 (1999).

ONbO⁺

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	989.7	Ar	IR	1
			988.0			
b_2	3	Asym. stretch	938.4	Ar	IR	1
			937.1			

Reference

- ¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **102**, 8251 (1998).

OTaO⁺

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	993.1 991.5	Ar	IR	1,2
b ₂	3	Asym. stretch	938.8 937.4	Ar	IR	1,2

References

- ¹M. Zhou and L. Andrews, J. Phys. Chem. A **102**, 8251 (1998).
²M. Chen, X. Wang, L. Zhang, M. Yu, and Q. Qin, Chem. Phys. **242**, 81 (1999).

OPrO⁺

\tilde{X} D _{∞h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _u ⁺	3	Asym. stretch	939.8 914.1	Ne Ar	IR IR	1 1

Reference

- ¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

ONdO⁺

\tilde{X} D _{∞h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _u ⁺	3	Asym. stretch	894.5 869.5	Ne Ar	IR IR	1 1

Reference

- ¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

OUO⁺

\tilde{X} D _{∞h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _u ⁺	3	Asym. stretch	980.1 952.3	Ne Ar	IR IR	1 1

Reference

- ¹M. Zhou, L. Andrews, N. Ismail, and C. Marsden, J. Phys. Chem. A **104**, 5495 (2000).

NAIN

\tilde{X} D _{∞h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _u ⁺	3	Asym. stretch	656.9	N ₂	IR	1

Reference

- ¹L. Andrews, M. Zhou, G. V. Chertihin, W. D. Bare, and Y. Hannachi, J. Phys. Chem. A **104**, 1656 (2000).

NGaN

\tilde{X} D _{∞h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _u ⁺	3	Asym. stretch	586.4	N ₂	IR	1

Reference

- ¹M. Zhou and L. Andrews, J. Phys. Chem. A **104**, 1648 (2000).

NInN

\tilde{X} D _{∞h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _u ⁺	3	Asym. stretch	479.5	N ₂	IR	1

Reference

- ¹M. Zhou and L. Andrews, J. Phys. Chem. A **104**, 1648 (2000).

GaCO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1875.6 1866.5	Ar	IR	1

Reference

- ¹H.-J. Himmel, A. J. Downs, J. C. Green, and T. M. Greene, J. Phys. Chem. A **104**, 3642 (2000).

InCO

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1920.8	Ar	IR	1

Reference

¹H.-J. Himmel, A. J. Downs, J. C. Green, and T. M. Greene, *J. Phys. Chem. A* **104**, 3642 (2000).

AlOSi

 \tilde{X} C_{∞v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	"Sym." stretch	524.2	CH ₄	Ra	1
	3	"Asym." stretch	1012.7 1007.7	Ar CH ₄	IR IR	1 1

Reference

¹M. Junker, M. Friesen, and H. Schnöckel, *J. Chem. Phys.* **112**, 1444 (2000).

cyc-AIP₂ \tilde{A}^2A_1 C_{2v}T^a=5780(210) gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	Sym. stretch	425(10)	gas	PE	1

 \tilde{X}^2B_2 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	Sym. stretch	306(10)	gas	PE	1

^aFrom vertical ionization potential.

Reference

¹H. Gómez, T. R. Taylor, and D. M. Neumark, *J. Phys. Chem. A* **105**, 6886 (2001).

cyc-GaP₂ \tilde{B}^2B_1 C_{2v}T₀=21000(400) gas PE^{2,3}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	P-P stretch	500	gas	PE	2,3
	2	GaP ₂ s-stretch	234	gas	PE	2,3

 \tilde{A}^2A_1 C_{2v}T₀=8420(800) gas PE^{2,3}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	GaP ₂ s-stretch	328	gas	PE	2,3

 \tilde{X}^2B_2 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	P-P stretch	322T	Kr	IR	1
	2	GaP ₂ s-stretch	222 220.9	gas Ar	PE IR	2,3 1

References

¹S. Li, R. J. Van Zee, and W. Weltner, Jr., *J. Phys. Chem.* **97**, 11393 (1993).

²T. R. Taylor, K. R. Asmis, H. Gomez, and D. M. Neumark, *Eur. Phys. J. D* **9**, 317 (1999).

³T. R. Taylor, H. Gómez, K. R. Asmis, and D. M. Neumark, *J. Chem. Phys.* **115**, 4620 (2001).

cyc-InP₂ \tilde{A}^2A_1 C_{2v}T₀=10330(8) gas PE¹TPE²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	PInP s-stretch	287	gas	TPE	2

 \tilde{X}^2B_2 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	P=P stretch	479	gas	TPE	2
	2	PInP s-stretch	190.0(2)	gas	TPE	2

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cyc-GaAs₂ \tilde{A}^2A_1 C_{2v}T₀=5600(620) gas PE²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2		235	gas	PE	2

 \tilde{X}^2B_2 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1		231.0T	Ar	IR	1
	2		176	gas	PE	2
			174.1	Ar	IR	1

References

- ¹S. Li, R. J. Van Zee, and W. Weltner, Jr., *J. Phys. Chem.* **97**, 11393 (1993).
²T. R. Taylor, H. Gómez, K. R. Asmis, and D. M. Neumark, *J. Chem. Phys.* **115**, 4620 (2001).

 C_3^-

\tilde{E}
 $T_0 = 32003(3)$ gas PD⁷ $\tilde{E}-\tilde{X}$ 308–313 nm

\tilde{D}
 $T_0 = 29885(3)$ gas PD⁷ $\tilde{D}-\tilde{X}$ 327–335 nm
 29833(20) Ne AB⁷ $\tilde{D}-\tilde{X}$ 329–336 nm

$\tilde{C}^2\Sigma_u^+$ D_{∞h}
 $T_0 = 28507(3)$ gas 20PD^{6,7} $\tilde{C}-\tilde{X}$ 337–351 nm
 28660(7) Ne AB^{6,7} $\tilde{C}-\tilde{X}$ 335–349 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1140(6)	gas	PD	7
			1180(14)	Ne	AB	6
Π_u	2	Bend	445H	gas	PD	7

$\tilde{B}^2\Sigma_u^-$ D_{∞h}
 $T_0 = 25948(3)$ gas PD⁶ $\tilde{B}-\tilde{X}$ 355–386 nm
 25968(7) Ne AB⁶ $\tilde{B}-\tilde{X}$ 370–385 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1072(6)	gas	PD	6
			1062(14)	Ne	AB	6
Π_u	2	Bend (ω)	407	gas	PD	6

$B_0 = 0.416$ PD⁶

$\tilde{A}^2\Delta_u$ D_{∞h}
 $T_0 = 24743(3)$ gas PD⁶ $\tilde{A}-\tilde{X}$ 386–405 nm
 24810(7) Ne AB⁶ $\tilde{A}-\tilde{X}$ 385–403 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1140(6)	gas	PD	6
			1110(14)	Ne	AB	6
Π_u	2	Bend (ω)	410	gas	PD	6

$B_0 = 0.418$ PD⁶

$\tilde{b}^4\Pi_u$ D_{∞h}
 $T_0 = 22342(3)$ gas PD⁶ $\tilde{b}-\tilde{X}$ 429–448 nm
 22306(7) Ne AB⁶ $\tilde{b}-\tilde{X}$ 412–448 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	959(4)	gas	PD	6
			953(10)	Ne	AB	6

Threshold for electron detachment from ground-state $C_3^- = 15980(160)$ gas PE¹⁻³

$\tilde{X}^2\Sigma_g^+$		D _{∞h}	Structure: IR ⁵			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1075(100)	gas	PE	3
Σ_u^+	3	Asym. stretch	1721.8	Ar	IR	4,5

References

- ¹J. M. Oakes and G. B. Ellison, *Tetrahedron* **42**, 6263 (1986).
²S. Yang, K. J. Taylor, M. J. Craycraft, J. Conceicao, C. L. Pettiette, O. Cheshnovsky, and R. E. Smalley, *Chem. Phys. Lett.* **144**, 431 (1988).
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⁴J. Szczepanski, S. Ekern, and M. Vala, *J. Phys. Chem. A* **101**, 1841 (1997).
⁵J. Szczepanski, C. Wehlburg, and M. Vala, *J. Phys. Chem. A* **101**, 7039 (1997).
⁶M. Tulej, J. Fulara, A. Sobolewski, M. Jungen, and J. P. Maier, *J. Chem. Phys.* **112**, 3747 (2000).
⁷N. M. Lakin, F. Güthe, M. Tulej, M. Pachkov, and J. P. Maier, *Faraday Discuss.* **115**, 383 (2000).

 Sn_3^-

Threshold for electron detachment from ground-state $Sn_3^- = 18070(80)$ gas PE¹

Reference

- ¹V. D. Moravec, S. A. Klopčič, and C. C. Jarrold, *J. Chem. Phys.* **110**, 5079 (1999).

CCN

$\tilde{D}^2\Pi$ C_{∞v}
 $T_0 = 27322(15)$ Ne AB¹⁶

$\tilde{C}^2\Sigma^+$ C_{∞v}
 $T_0 = 26661.80$ gas AB¹LF¹⁴ $\tilde{C}-\tilde{X}$ 310–375 nm
 26645(15) Ne AB¹⁶
 Evidence for predissociation above 29100.¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	Stretch	1859.20	gas	AB,LF	1,14
Π	2	Bend	469.9(4)	gas	AB,LF	1,14
Σ^+	3	Stretch	1257(4) (ω)	gas	LF	14

$B_0 = 0.413$ AB¹LF¹⁴

$\tilde{B}^2\Sigma^-$ C_{∞v}
 $T_0 = 22413.25$ gas AB¹ $\tilde{B}-\tilde{X}$ 442–446 nm
 22430(10) Ne AB¹⁶
 22180 Ar AB³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	445T	gas	AB	1

$B_0 = 0.405$ AB¹

$\tilde{A}^2\Delta$	$C_{\infty v}$					
$T_0=21259.203$	gas	$AB^1LF^4EM^9$		$\tilde{A}-\tilde{X}$	376–471 nm	
21248(10)	Ne	AB^{16}				
21377	Ar	LF^2AB^3		$\tilde{A}-\tilde{X}$	373–550 nm	

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ^+	1	Stretch	1770.77	gas	AB	1
			1732(2)	Ar	LF	2
Π	2	Bend	457.1 (ω)	gas	AB,LF	1,6,14
Σ^+	3	Stretch	1241.64	gas	AB	1
			1225(2)	Ar	LF	2

$\tau=170$ ns Ar LF^2
 $A_{\text{eff}}=-0.807$ gas $AB^1LF^{4,6}$
 $B_0=0.414$ $AB^1LF^{4,6}DR^7$

$\tilde{X}^2\Pi$	$C_{\infty v}$					
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ^+	1	Stretch	1923.25	gas	LF,EM, LMR,DL	5,8,9,12, 10,11
			1717	Ar	LF	2
Π	2	Bend	319.78 (ω) 179.60 ($\mu^2\Sigma^-$)	gas	AB,LF,LMR	1,8,15 14,15
Σ^+	3	Stretch	1050.76	gas	LF,EM	5,8,9
			1066	Ar	LF	2

$A_0=42.539$, $\epsilon\omega_2=133.63$ gas $LF^8DL^{10,11}LMR^{12,15}$
 $B_0=0.398$ $AB^1LF^{4,6}DL^{10,11}LMR^{12,15}MW^{13}$

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SiCN

$\tilde{X}^2\Pi$	$C_{\infty v}$					
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ^+	1	CN stretch	2077.3m	Ar	IR	1
			3	SiC stretch	584.6vs	Ar

$A_{\text{eff}}=70.8$ gas MW^3
 $B_0=0.185$ $MW^{2,3}$

References

- G. Maier, H. P. Reisenauer, H. Egenolf, and J. Glatthaar, *Eur. J. Org. Chem.* **1998**, 1307.
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- M. C. McCarthy, A. J. Apponi, C. A. Gottlieb, and P. Thaddeus, *J. Chem. Phys.* **115**, 870 (2001).

CNC

$\tilde{B}^2\Sigma_u^-$	$D_{\infty h}$	Structure: AB^1			
$T_0=34802.33$	gas	AB^1		$\tilde{B}-\tilde{X}$	283–288 nm
34652(20)	gas	AB^3			
34602(20)	Ar ^a	AB^2		$\tilde{B}-\tilde{X}$	276–292 nm
34305(20)					

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Π_u	2	Bend	398	gas	AB	1
			385(20)H	Ar	AB	2

$B_0=0.443$ AB^1

$\tilde{A}^2\Delta_u$	$D_{\infty h}$	Structure: AB^1			
$T_0=30338.53$	gas	AB^1		$\tilde{A}-\tilde{X}$	325–332 nm
30294(20)	Ne	AB^3			
30048(20)	Ar	AB^2		$\tilde{A}-\tilde{X}$	324–333 nm

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Π_u	2	Bend	440	gas	AB	1

$A=0.33$ gas AB^1

$B_0=0.450$ AB^1

$\tilde{X}^2\Pi_g$	$D_{\infty h}$	Structure: AB^1				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Π_u	2	Bend	321 ^b	gas	AB	1
			1452.4	Ne	IR	3
Σ_u^+	3	Asym. stretch	1453s	Ar	IR	2

$A=26.41$; $\epsilon=0.549$ AB^1

$B_0=0.454$ AB^1

^aTwo prominent sites in argon matrix.

^bCalculated position of lowest frequency component ($^2\Sigma_u^-$) is 144 cm^{-1} . Moderately intense absorption at 134 cm^{-1} in an argon matrix is tentatively assigned to this transition.

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SiNC

In an argon matrix,¹ irradiation at 366 nm results in photoisomerization to SiCN.

$\tilde{X}^2\Pi$		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	NC stretch	1945.2vs	Ar	IR	1
	3	SiN stretch	629.1m	Ar	IR	1

$A_{\text{eff}}=58.8$ gas MW³
 $B_0=0.213$ MW²

References

- ¹G. Maier, H. P. Reisenauer, H. Egenolf, and J. Glatthaar, *Eur. J. Org. Chem.* **1998**, 1307.
²A. J. Apponi, M. C. McCarthy, C. A. Gottlieb, and P. Thaddeus, *Astrophys. J.* **536**, L55 (2000).
³M. C. McCarthy, A. J. Apponi, C. A. Gottlieb, and P. Thaddeus, *J. Chem. Phys.* **115**, 870 (2001).

SiNSi

$^2\Sigma_u^+$ $D_{\infty h}$ Structure: MPI¹
 $T_0=34314.29$ gas MPI¹ $^2\Sigma_u^+ - \tilde{X}$ 275–315 nm
 $B_0=0.113$ MPI¹

$\tilde{X}^2\Pi_{g,1/2}$		$D_{\infty h}$ Structure: MPI ¹				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_u^+	3	Asym. stretch	904.0	Ar	IR	2
			910.0	N ₂	IR	2

$B_0=0.112$ MPI¹

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- ¹D. J. Brugh and M. D. Morse, *Chem. Phys. Lett.* **267**, 370 (1997).
²G. Meloni, S. Nunziante Cesaro, and N. Sanna, *Chem. Phys. Lett.* **343**, 113 (2001).

SnCN

$\tilde{X}^2\Pi$		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	3	SnC stretch	380(10)	gas	PE	1

$A=2300(65)$ gas PE¹

Reference

- ¹V. D. Moravec and C. C. Jarrold, *J. Chem. Phys.* **113**, 1035 (2000).

NTiO⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		TiO stretch	791.4	Ne	IR	2
			784.3	Ar	IR	1

References

- ¹G. P. Kushto, M. Zhou, L. Andrews, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **103**, 1115 (1999).
²L. Andrews and X. Wang, *J. Phys. Chem. A* **106**, 1196 (2002).

NTaO⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		TaO stretch	819.1T	Ar	IR	1
			814.1T			

Reference

- ¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **102**, 10025 (1998).

CrNO⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO stretch	1519.9	Ne	IR	2
			1511.5	Ar	IR	1

References

- ¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **102**, 7452 (1998).
²L. Andrews and X. Wang, *J. Phys. Chem. A* **106**, 1196 (2002).

NReO⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	ReN stretch	1038.0T	Ar	IR	1
	3	ReO stretch	860.9T	Ar	IR	1

Reference

- ¹L. Andrews, M. Zhou, and D. W. Ball, *J. Phys. Chem. A* **102**, 10041 (1998).

CoNO⁻ $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1585.7	Ne	IR	1

Reference¹M. Zhou and L. Andrews, J. Phys. Chem. A **104**, 3915 (2000).**NiNO⁻** $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1454.7	Ne	IR	1

Reference¹M. Zhou and L. Andrews, J. Phys. Chem. A **104**, 3915 (2000).**PdNO⁻** $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1487.6	Ne	IR	1

Reference¹A. Citra and L. Andrews, J. Phys. Chem. A **104**, 8160 (2000).**PtNO⁻** $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1462.1	Ne	IR	1

Reference¹A. Citra and L. Andrews, J. Phys. Chem. A **104**, 8160 (2000).**AgNO⁻** $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1399.2	Ne	IR	1
			1392.3	Ar	IR	1

Reference¹A. Citra and L. Andrews, J. Phys. Chem. A **105**, 3042 (2001).**NCeO⁻** $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CeO stretch	624.8	Ar	IR	1

Reference¹S. P. Willson, L. Andrews, and M. Neurock, J. Phys. Chem. A **104**, 3446 (2000).**NPrO⁻** $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	PrN stretch	718.2	Ar	IR	1
	3	PrO stretch	612.3	Ar	IR	1

Reference¹S. P. Willson, L. Andrews, and M. Neurock, J. Phys. Chem. A **104**, 3446 (2000).**GaNO** $\tilde{\chi}$ $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	NO stretch	1578.5	Ar	IR	1

Reference¹L. Andrews, M. Zhou, and X. Wang, J. Phys. Chem. A **104**, 8475 (2000).**InNO** $\tilde{\chi}$ $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	NO stretch	1524.9	Ar	IR	1

Reference¹L. Andrews, M. Zhou, and X. Wang, J. Phys. Chem. A **104**, 8475 (2000).

TINO

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	NO stretch	1454.6	Ar	IR	1

Reference

¹L. Andrews, M. Zhou, and X. Wang, J. Phys. Chem. A **104**, 8475 (2000).

OScO

\tilde{C}
 $T_0=10330(800)$ gas PE¹

\tilde{B}
 $T_0=3870T$ gas PE¹

\tilde{A}
 $T_0=3070T$ gas PE¹

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	Sym. stretch	740(80)	gas	PE	1

Reference

¹H. Wu and L.-S. Wang, J. Phys. Chem. A **102**, 9129 (1998).

OYO

\tilde{C}
 $T_0=10330(540)$ gas PE¹

\tilde{B}
 $T_0=6050(470)$ gas PE¹

\tilde{A}
 $T_0=3710(765)$ gas PE¹

\tilde{X} C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	640(80)	gas	PE	1
			708.2	Ar	IR	2

References

¹H. Wu and L.-S. Wang, J. Phys. Chem. A **102**, 9129 (1998).

²L. Andrews, M. Zhou, G. V. Chertihin, and C. W. Bauschlicher, Jr., J. Phys. Chem. A **103**, 6525 (1999).

OLaO

\tilde{X} C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	569.8	Ar	IR	1

Reference

¹L. Andrews, M. Zhou, G. V. Chertihin, and C. W. Bauschlicher, Jr., J. Phys. Chem. A **103**, 6525 (1999).

OTiO

\tilde{A}^1B_2 $D_{\infty h}$
 $T_0=19360(1600)$ gas PE³
 18880 Ne EM¹ 529–621 nm

\tilde{a}^3B_2
 $T_0=15800(800)$ gas PE⁴

\tilde{X} C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	965T	gas	IR,PE	2,4
			962.0	Ne	IR	1
			946.9	Ar	IR	3
b_2	3	Asym. stretch	944	gas	IR	2
			934.8	Ne	IR	1
			917.1	Ar	IR	3

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¹N. S. McIntyre, K. R. Thompson, and W. Weltner, Jr., J. Phys. Chem. **75**, 3243 (1971).

²T. C. DeVore and T. N. Gallaher, High Temp. Sci. **16**, 269 (1983).

³G. V. Chertihin and L. Andrews, J. Phys. Chem. **99**, 6356 (1995).

⁴H. Wu and L.-S. Wang, J. Chem. Phys. **107**, 8221 (1997).

OZrO

\tilde{X} C_{2v} Structure: MW ²						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	884.3	Ar	IR	1
b_2	3	Asym. stretch	818.0	Ar	IR	1

$A_0=0.663$; $B_0=0.257$; $C_0=0.185$ MW²

References

¹G. V. Chertihin and L. Andrews, J. Phys. Chem. **99**, 6356 (1995).

²D. J. Brugh, R. D. Suenram, and W. J. Stevens, J. Chem. Phys. **111**, 3526 (1999).

OVO

\tilde{C}^2A_2 C_{2v}
 $T_0=21000(800)$ gas PE²

\tilde{B}^2B_2 C_{2v}
 $T_0=16140(800)$ gas PE²

\tilde{A}^2B_1 C_{2v}
 $T_0=4840(800)$ gas PE²

\tilde{X}^2A_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	970(40) 946.3w	gas Ar	PE IR	2 1
b_2	3	Asym. stretch	935.9s	Ar	IR	1

References

¹G. V. Chertihin, W. D. Bare, and L. Andrews, *J. Phys. Chem. A* **101**, 5090 (1997).

²H. Wu and L.-S. Wang, *J. Chem. Phys.* **108**, 5310 (1998).

ONbO

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	933.5	Ar	IR	1
b_2	3	Asym. stretch	875.9	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **102**, 8251 (1998).

OTaO

\tilde{B}
 $T_0=16232$ Ne AB¹ $\tilde{B}-\tilde{X}$ 558–616 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	2	Bend	281	Ne	AB	1

\tilde{A}
 $T_0=11615$ Ne AB¹ $\tilde{A}-\tilde{X}$ 716–861 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	Sym. stretch	937	Ne	AB	1
	2	Bend	285	Ne	AB	1

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	965.3	Ar	IR	1–3
b_2	3	Asym. stretch	912.2	Ar	IR	1–3

References

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OCrO

\tilde{B}
 $T^a=14680(430)$ gas PE¹⁰

\tilde{A}
 $T^a=7910(430)$ gas PE¹⁰

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	895(20) 920.8 914.4	gas Ne Ar	PE IR IR	6,10 9 7,8
	2	Bend	220(20)	gas	PE	6,10
b_2	3	Asym. stretch	974.9 965.4	Ne Ar	IR IR	9 1–5,7,8

^aFrom vertical electron detachment energies.

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¹⁰G. L. Gutsev, P. Jena, H.-J. Zhai, and L.-S. Wang, *J. Chem. Phys.* **115**, 7935 (2001).

OMoO

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	950.7 939.3	Ne Ar	IR IR	1,4 1–3
b_2	3	Asym. stretch	901.3 885.5	Ne Ar	IR IR	1,4 1–3

References

¹W. D. Hewett, Jr., J. H. Newton, and W. Weltner, Jr., *J. Phys. Chem.* **79**, 3640 (1975).

²M. J. Almond and A. J. Downs, *J. Chem. Soc., Dalton Trans.* 809 (1988).

³W. D. Bare, P. F. Souter, and L. Andrews, *J. Phys. Chem. A* **102**, 8279 (1998).

⁴M. Zhou and L. Andrews, *J. Chem. Phys.* **111**, 4230 (1999).

OWO

 $T_0 = 12672$ Ne AB¹ 605–790 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	972	Ne	AB	1
	2	Bend	300	Ne	AB	1

${}^3A_1?$
 $T_0 \geq 8010(100)$ C_{2v} gas PE⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	945(25)	gas	PE	4
	2	Bend	340(20)	gas	PE	4

${}^1B_1?$
 $T_0 \geq 6260(100)$ C_{2v} gas PE⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	970(15)	gas	PE	4

$\tilde{a} {}^3B_1$
 $T_0 \geq 2640(100)$ C_{2v} gas PE⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	995(25)	gas	PE	4
	2	Bend	320(15)	gas	PE	4

OMnO

$\tilde{X} {}^4B_1$ C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	816.4	Ar	IR	2
b_2	3	Asym. stretch	948.0s	Ar	IR	1,2
			969.6	N ₂	IR	2
			962.4			

References

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\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	OO stretch	1108.2	Ar	IR	1,2
			1108.4	N ₂	IR	2
	2	OMnO s-stretch	686.4s	Ar	IR	1,2
			686.6	N ₂	IR	2

References

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MnOO

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OO stretch	1224.5T	Ar	IR	1

Reference

¹G. V. Chertihin and L. Andrews, J. Phys. Chem. A **101**, 8547 (1997).

OReO

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	989.2	Ne	IR	1
			981.9	Ar	IR	1
b_2	3	Asym. stretch	941.0	Ne	IR	1
			931.7	Ar	IR	1

Reference

¹M. Zhou, A. Citra, B. Liang, and L. Andrews, J. Phys. Chem. A **104**, 3457 (2000).

ORuO

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b_2	3	Asym. stretch	911.9	Ne	IR	2
			902.1	Ar	IR	1,2

References

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OOsO

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b_2	3	Asym. stretch	957.3	Ne	IR	1
			949.9	Ar	IR	1

Reference

¹M. Zhou, A. Citra, B. Liang, and L. Andrews, *J. Phys. Chem. A* **104**, 3457 (2000).

OCoO

\tilde{X}		$D_{\infty h}$		Structure: ESR ¹		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	790(60)	gas	PD	3
Σ_u^+	3	Asym. stretch	945.4	Ar	IR	2
			978.6	N ₂	IR	2

References

¹R. J. Van Zee, Y. M. Hamrick, S. Li, and W. Weltner, Jr., *J. Phys. Chem.* **96**, 7247 (1992).

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³L.-S. Wang, cited by Ref. 2.

CoOO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OO stretch	1286.8	Ar	IR	1
			1284	N ₂	IR	1

Reference

¹G. V. Chertihin, A. Citra, L. Andrews, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **101**, 8793 (1997).

cyc-CoOO

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	OO stretch	984.9	Ar	IR	1
			1018.8	N ₂	IR	1

Reference

¹G. V. Chertihin, A. Citra, L. Andrews, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **101**, 8793 (1997).

ORhO

\tilde{X}		$D_{\infty h}$		Structure: ESR ¹		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	845T	Ar	IR	2
Σ_u^+	3	Asym. stretch	900.1	Ar	IR	2

References

¹R. J. Van Zee, Y. M. Hamrick, S. Li, and W. Weltner, Jr., *J. Phys. Chem.* **96**, 7247 (1992).

²A. Citra and L. Andrews, *J. Phys. Chem. A* **103**, 4845 (1999).

OIrO

\tilde{X}		$D_{\infty h}$		Structure: ESR ¹		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	960T	Ar	IR	2
Σ_u^+	3	Asym. stretch	929.0	Ar	IR	2

References

¹R. J. Van Zee, Y. M. Hamrick, S. Li, and W. Weltner, Jr., *J. Phys. Chem.* **96**, 7247 (1992).

²A. Citra and L. Andrews, *J. Phys. Chem. A* **103**, 4182 (1999).

cyc-IrOO

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	OO stretch	1022.7	Ar	IR	1

Reference

¹A. Citra and L. Andrews, *J. Phys. Chem. A* **103**, 4182 (1999).

ONiO

$\tilde{b}^1\Sigma_g^+$		$D_{\infty h}$				
$T_0=6210(240)$		gas	PE ²			
$\tilde{a}^1\Delta_g$		$D_{\infty h}$				
$T_0=3230(160)$		gas	PE ²			
$\tilde{X}^3\Sigma_g^-$		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	750(30)	gas	PE	2
Π_u	2	Bend	125.0	Ar	IR	3
Σ_u^+	3	Asym. stretch	954.9	Ar	IR	1,3
			959.5	N ₂	IR	1

References

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cyc-NiO₂

\tilde{A}
 $T_0 = 8000\text{T}$ gas PE³

\tilde{X}		C_{2v}	Structure: IR ¹			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	OO stretch	967.1s	Ar	IR	1,2,4
			970.1	N ₂	IR	2
b_2	3	ONiO s-stretch	538.3w	Ar	IR	2,4
			511.7m	Ar	IR	2,4

References

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cyc-PdO₂

\tilde{X}		C_{2v}	Structure: IR ¹			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	OO stretch	1030.1	Ne	IR	3
			1023.0	Ar	IR	1,2
			997.7	N ₂	IR	2
b_2	2	OPdO s-stretch	427w	Ar	IR	1

References

- ¹H. Huber, W. Klotzbücher, G. A. Ozin, and A. Vander Voet, Can. J. Chem. **51**, 2722 (1973).
²W. D. Bare, A. Citra, G. V. Chertihin, and L. Andrews, J. Phys. Chem. A **103**, 5456 (1999).
³X. Wang and L. Andrews, J. Phys. Chem. A **105**, 5812 (2001).

OPtO

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	900T	Ar	IR	1
Σ_u^+	3	Asym. stretch	958.7	Ne	IR	2
			961.8	Ar	IR	1,3
			953.3			

References

- ¹W. D. Bare, A. Citra, G. V. Chertihin, and L. Andrews, J. Phys. Chem. A **103**, 5456 (1999).
²X. Wang and L. Andrews, J. Phys. Chem. A **105**, 5812 (2001).
³D. Danset, L. Manceron, and L. Andrews, J. Phys. Chem. A **105**, 7205 (2001).

cyc-PtO₂

\tilde{X}		C_{2v}	Structure: IR ¹			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	OO stretch	930.0	Ne	IR	3
			928.0s	Ar	IR	1,2,4
b_2	3	OPtO a-stretch	512.3w	Ar	IR	4
			551.2wm	Ar	IR	4

References

- ¹H. Huber, W. Klotzbücher, G. A. Ozin, and A. Vander Voet, Can. J. Chem. **51**, 2722 (1973).
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⁴D. Danset, L. Manceron, and L. Andrews, J. Phys. Chem. A **105**, 7205 (2001).

OCuO

$T_0 = 20699$ Ne LF^{3,7} 432–540 nm
 20486.5 Ar LF^{1,3,7}
 20064 Kr LF⁷
 19686.9 Xe UV²LF⁷ 440–586 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	CuO s-stretch	611.5	Ne	LF	3,7
			627.5	Ar	LF	7
			625.9			
Π_u	2	Bend	608.1			
			618.9	Xe	AB,LF	2,7
			614.1	N ₂	LF	7
Σ_u^+	3	Asym. stretch	135.6	Ne	LF	3,7
			118	Xe	LF	7
			628	Xe	LF	7

\tilde{E}'
 $T_0 = 13640(160)$ gas PE⁵

\tilde{D}
 $T_0 = 11309.5$ Ne LF⁷
 11268.5 Xe LF⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	636.8	Ne	LF	7
Π_u	2	Bend	234	Ne	LF	7

\tilde{C}
 $T_0 = 11110.5$ Ne LF⁷
 11060 Xe LF⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	640.8	Ne	LF	7

\tilde{C}' $T_0=9680(160)$ gas PE⁵ \tilde{C}'' $T_0=6540(160)$ gas PE^{4,5} \tilde{B} $T_0=6683.4$ Xe LF⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	569.7	Xe	LF	7

 \tilde{B}' $T_0=5080(160)$ gas PE^{4,5}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			640(80)	gas	PE	4,5

 \tilde{A} $T_0=3850$ T Ne LF^{3,7}
3387.3 Xe LF⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	590	Ne	LF	3
			597.5	Xe	LF	7
Π_u	2	Bend	113	Xe	LF	7

 \tilde{A}' $T_0=2580(160)$ gas PE^{4,5}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			640(80)	gas	PE	4,5

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	CuO s-stretch	658	Ne	LF	3,7
			668	Ar	LF	1
			664.2	Xe	EM,LF	2,7
Π_u	2	Bend	193.4	Ne	LF	3,7
			161.5	Xe	LF	7
Σ_u^+	3	CuO a-stretch	823.0	Ar	IR	6
			826.7	N ₂	IR	6

References

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- ⁷N. Caspary, E. V. Savchenko, A. Thoma, A. Lammers, and V. E. Bondybey, Low Temp. Phys. **26**, 744 (2000).

AgOO

 \tilde{X} C_sStructure: IR¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	OO stretch	1102.5	Ne	IR	3
			1075.7	Ar	IR	1,2

References

- ¹D. E. Tevault, R. R. Smardzewski, M. W. Urban, and K. Nakamoto, J. Chem. Phys. **77**, 577 (1982).
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- ³X. Wang and L. Andrews, J. Phys. Chem. A **105**, 5812 (2001).

AuOO

In an argon matrix, relatively weak absorption maxima at 40980 (244 nm), 36230 (276 nm), and 31950 (313 nm) and a prominent, broad absorption maximum at 23640 (423 nm) have been attributed¹ to AuO₂.

 \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	OO stretch	1213.6	Ne	IR	3
			1093.8	Ar	IR	1,2

References

- ¹D. McIntosh and G. A. Ozin, Inorg. Chem. **15**, 2869 (1976).
- ²A. Citra and L. Andrews, J. Mol. Struct. **489**, 95 (1999).
- ³X. Wang and L. Andrews, J. Phys. Chem. A **105**, 5812 (2001).

OAuO

 \tilde{X} D_{∞h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_u^+	3	Asym. stretch	824.2	Ne	IR	2
			817.9	Ar	IR	1

References

- ¹A. Citra and L. Andrews, J. Mol. Struct. **489**, 95 (1999).
- ²X. Wang and L. Andrews, J. Phys. Chem. A **105**, 5812 (2001).

OCeO

$T_0 = 20067$	Ne	EM ¹	498–649 nm
19836	Ar	EM ¹	504–660 nm
19259	N ₂	EM ¹	519–633 nm
$\tau = 200(50)$ ms	Ar	EM ¹	

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	777w	Ne	IR,EM	1
			757.3w	Ar	IR,EM	1–3
			714	N ₂	IR,EM	1
b_2	3	Asym. stretch	759s	Ne	IR	1
			736.7s	Ar	IR	1–3
			719	N ₂	IR	1

References

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³S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

cyc-CeO₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	OO stretch	823.7	Ar	IR	1

Reference

- ¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

OPrO

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	711.7T	Ne	IR	3
			696.0T	Ar	IR	3
Σ_u^+	3	Asym. stretch	752.5	Ne	IR	3
			730.1	Ar	IR	1–3

References

- ¹R. L. DeKock and W. Weltner, Jr., J. Phys. Chem. **75**, 514 (1971).
²S. D. Gabelnick, G. T. Reedy, and M. G. Chasanov, J. Chem. Phys. **60**, 1167 (1974).
³S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

cyc-PrO₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	OO stretch	827T	Ar	IR	1

Reference

- ¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

ONdO

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	661.8T	Ar	IR	1
			737.6	Ne	IR	1
Σ_u^+	3	Asym. stretch	716.9	Ar	IR	1

Reference

- ¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

OSmO

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	602.5T	Ar	IR	1
			643.2	Ar	IR	1
Σ_u^+	3	Asym. stretch				

Reference

- ¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

OEuO

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b_2	3	Asym. stretch	622.8	Ar	IR	1

Reference

- ¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

OGdO

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b_2	3	Asym. stretch	635.5	Ar	IR	1

Reference

- ¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

OTbO

\tilde{X} C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	758.8	Ar	IR	2,3
b_2	3	Asym. stretch	718.8	Ar	IR	1-3

References

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²S. D. Gabelnick, G. T. Reedy, and M. G. Chasanov, *J. Chem. Phys.* **60**, 1167 (1974).
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ODyO

\tilde{X} $D_{\infty h}$						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	612.2T	Ar	IR	1
Σ_u^+	3	Asym. stretch	599.2	Ne	IR	1
			580.5	Ar	IR	1

Reference

- ¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **103**, 6972 (1999).

OHoO

\tilde{X} C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	668.2	Ne	IR	1
			649.2	Ar	IR	1
b_2	3	Asym. stretch	548.8	Ar	IR	1

Reference

- ¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **103**, 6972 (1999).

OTmO

\tilde{X} C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	706.6	Ar	IR	1
b_2	3	Asym. stretch	615.7	Ar	IR	1

Reference

- ¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **103**, 6972 (1999).

OYbO

\tilde{X} $D_{\infty h}$						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	632.1T	Ar	IR	1
Σ_u^+	3	Asym. stretch	627.7	Ar	IR	1

Reference

- ¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **103**, 6972 (1999).

OThO

\tilde{X} C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	808.4	Ne	IR	3
			787.3	Ar	IR	1,2
b_2	3	Asym. stretch	756.7	Ne	IR	3
			735.0	Ar	IR	1,2

References

- ¹S. D. Gabelnick, G. T. Reedy, and M. G. Chasanov, *J. Chem. Phys.* **60**, 1167 (1974).
²G. P. Kushto and L. Andrews, *J. Phys. Chem. A* **103**, 4836 (1999).
³M. Zhou and L. Andrews, *J. Chem. Phys.* **111**, 11044 (1999).

OUO

³H_g? $D_{\infty h}$
 When OUO is isolated in an argon or heavier rare-gas matrix, an absorption pattern which differs from that characteristic of a neon matrix results. This is attributed⁴ to a sufficiently large shift in this very low-lying excited state to place it below the ground state of the free molecule.

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ_u^+	3	Asym. stretch	776.1	Ar	IR	1-4
			768.0	Kr	IR	1

\tilde{X} $D_{\infty h}$						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ_u^+	3	Asym. stretch	914.8	Ne	IR	4

References

- ¹S. D. Gabelnick, G. T. Reedy, and M. G. Chasanov, *J. Chem. Phys.* **58**, 4468 (1973).
²R. D. Hunt and L. Andrews, *J. Chem. Phys.* **98**, 3690 (1993).
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⁴M. Zhou, L. Andrews, N. Ismail, and C. Marsden, *J. Phys. Chem. A* **104**, 5495 (2000).

AICO⁻

\tilde{X} C _{∞v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	CO stretch	1682.3	Ar	IR	1

Reference

¹L. Zhang, J. Dong, M. Zhou, and Q. Qin, J. Chem. Phys. **113**, 10169 (2000).

NAIO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	AIO stretch	1079.5	Ar	IR	1

Reference

¹L. Andrews, M. Zhou, and W. D. Bare, J. Phys. Chem. A **102**, 5019 (1998).

AINO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1644.3	Ar	IR	1
	3	AIN stretch	510.2	Ar	IR	1

Reference

¹L. Andrews, M. Zhou, and W. D. Bare, J. Phys. Chem. A **102**, 5019 (1998).

AION

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	ON stretch	1282.1	Ar	IR	1,2
	3	AIO stretch	566.7	Ar	IR	2

References

¹G. K. Ruschel and D. W. Ball, High Temp. Mater. Sci. **37**, 63 (1997).
²L. Andrews, M. Zhou, and W. D. Bare, J. Phys. Chem. A **102**, 5019 (1998).

cyc-AIP₂⁻

Threshold for electron detachment from ground-state *cyc*-AIP₂⁻ = 15600(56) gas PE¹

Reference

¹H. Gómez, T. R. Taylor, and D. M. Neumark, J. Phys. Chem. A **105**, 6886 (2001).

cyc-InP₂⁻

Threshold for electron detachment from ground-state InP₂⁻ = 13050(8) gas PE¹TPE²

\tilde{X}^1A_1 C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	PlnP s-stretch	227T	gas	TPE	2

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¹C. Xu, E. de Beer, D. W. Arnold, C. C. Arnold, and D. M. Neumark, J. Chem. Phys. **101**, 5406 (1994).
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cyc-GaAs₂⁻

Threshold for electron detachment from ground-state *cyc*-GaAs₂⁻ = 15280(270) gas PE¹

Reference

¹T. R. Taylor, H. Gómez, K. R. Asmis, and D. M. Neumark, J. Chem. Phys. **115**, 4620 (2001).

CCO

$\tilde{c}^1\Pi$ C_{∞v}
 T₀ = 22390(160) gas PF,PE¹⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	3	CC stretch	1234(140)	gas	PE	17

In an argon matrix,¹ a broad, unstructured absorption is observed near 500 nm, and CCO photodissociates on exposure of the sample to visible light.

$\tilde{A}^3\Pi$ C_{∞v} Structure: AB³
 T₀ = 11651.182 gas AB^{2,3,14}LF⁶DL^{10-12,14,20} $\tilde{A}-\tilde{X}$ 500–860 nm
 11650(3) Ne AB¹⁸
 11860 Ar AB⁵ $\tilde{A}-\tilde{X}$ 600–850 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	CO stretch	2045.7 2045(3)	gas Ne	AB AB	3 18
Π	2	Bend	594.75 ^a	gas	AB,DL	3,12
Σ ⁺	3	CC stretch	1283.60 1279(3)	gas Ne	AB,DL AB	3,14 18

The fluorescence decay pattern⁷ of CCO $\tilde{A}(101)$ and of higher vibronic levels is complex. There is a short-lived (~15 μs) component and a long-lived (333+105/-64 μs) component which is, in turn, nonexponential, sug-

gesting perturbation by the $\tilde{b}^1\Sigma^+$ and $\tilde{a}^1\Delta$ states, as well as by high ground-state vibrational levels.

$A = -35.36$; $\epsilon = -0.172$ gas AB^3DL^{10}
 $B_0 = 0.407$ AB^3DL^{10}

$\tilde{b}^1\Sigma^+$ $C_{\infty v}$
 $T_0 = 8190(145)$ gas PE^{15,17}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	1920(160)	gas	PE	15,17

$\tilde{a}^1\Delta$ $C_{\infty v}$
 $T_0 = 5270(140)$ gas PE^{15,17}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	1942.85	gas	PE,DL	15,17,19

$B_0 = 0.383$ DL¹⁹

$\tilde{X}^3\Sigma^-$ $C_{\infty v}$ Structure: AB³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	1970.86	gas	LF,DL	6,9,16
			1969s	Ar	IR	1,4
			1978 N ₂	Ar	IR	1
			1987	N ₂	IR	1
Π	2	Bend	379.53	gas	AB,DL	3,11
			381m	Ar	IR	1
Σ^+	3	CC stretch	1063	gas	LF	6
			1064w	Ar	IR	1
			1074 N ₂	Ar	IR	1
			1077	N ₂	IR	1

$B_0 = 0.385$ $AB^3MW^{8,13}$

^a ω .

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²⁰H. Abe, M. Mukai, M. Fujitaki, and N. Ohashi, *J. Mol. Spectrosc.* **195**, 317 (1999).

SiCO

$\tilde{A}^3\Pi?$ $C_{\infty v}$
 $T_0 = 24056(10)$ Ar AB¹ $\tilde{A}-\tilde{X}$ 365–416 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	1857(10)	Ar	AB	1
	3	SiC stretch	750(10)	Ar	AB	1

$\tilde{X}^3\Sigma^-?$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1898.1	Ar	IR	1,2

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²L. Zhang, J. Dong, and M. Zhou, *J. Chem. Phys.* **113**, 8700 (2000).

GeCO

\tilde{X} $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	1918.9	Ar	IR	2,3
			1908	Kr	IR	1
			1921.3	N ₂	IR	2
			1924.0	CO	IR	2

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SnCO

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1941.1	Ar	IR	2
			1921	Kr	IR	1

References

- ¹A. Bos, J. Chem. Soc., Chem. Commun. 26 (1972).
²L. Zhang, J. Dong, and M. Zhou, J. Chem. Phys. **113**, 8700 (2000).

CCS

$\tilde{A}^3\Pi$ $C_{\infty v}$
 $T_0 < 14663$ gas LF⁴ $\tilde{A}-\tilde{X}$ 600–690 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CC stretch	1760T	gas	LF	4

$\tau \cong 4 \mu s$ gas LF⁴
 $A = -122.4$ LF⁴
 $B = 0.213$ LF⁴

$\tilde{X}^3\Sigma^-$ $C_{\infty v}$ Structure: MW²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CC stretch	1671(3)	gas	LF	4
Π	2	Bend	267(3)	gas	LF	4
Σ^+	3	CS stretch	858(3)	gas	LF	4

$B_0 = 0.216$ MW¹⁻³

References

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NCN

$\tilde{a}^1\Delta_u$ $D_{\infty h}$
 $T_0 = 43530(80)$ gas AB⁶PF¹⁴ $\tilde{a}-\tilde{a}$ 250–290 nm
 In the gas phase,¹⁴ this state photodissociates into $N_2(X^1\Sigma_g^+) + C(^1\Delta)$.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1020T	gas	PF	14

$\tilde{c}^1\Pi_u$ $D_{\infty h}$ Structure: AB⁵
 $T_0 = 38200(80)$ gas AB⁵PF¹⁴ $\tilde{c}-\tilde{a}$ 330–334 nm
 In the gas phase,¹⁴ this state photodissociates into $N_2(X^1\Sigma_g^+) + C(^3P)$.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1160T	gas	AB	6

$\epsilon\omega_2 = -84.2$ gas AB⁵
 $B_0 = 0.395$ AB⁵

$\tilde{B}^3\Sigma_u^-$ $D_{\infty h}$
 $T_0 = 33488$ gas AB⁶PF¹⁴ $\tilde{B}-\tilde{X}$ 258–300 nm
 33100 Ar AB² $\tilde{B}-\tilde{X}$ 240–302 nm
 33215 N₂ AB² $\tilde{B}-\tilde{X}$ 240–301 nm

In the gas phase, the bands are diffuse,⁶ and photodissociation into $N_2(X^1\Sigma_g^+) + C(^3P)$ predominates.¹⁴ The threshold for photodecomposition into C+N₂ is observed in argon and nitrogen matrices^{2,4} near 280 nm.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1050T 1050(10)	gas Ar, N ₂	AB, PF AB	6,14 2
Π_u	2	Bend	540(20)	gas	PF	14

$\tilde{A}^3\Pi_u$ $D_{\infty h}$ Structure: AB¹LF¹⁰
 $T_0 = 30383.967$ gas AB¹LF^{10,12} $\tilde{A}-\tilde{X}$ 326–329 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1254(5)	gas	LF	8
Π_u	2	Bend	534T	gas	LF	12

$A = -37.57$; $\epsilon\omega_2 = -91.12$ gas AB¹LF¹²
 $\tau_0 = 183(6)$ ns gas LF⁸
 $B_0 = 0.397$ AB¹LF^{10,12}

$\tilde{b}^1\Sigma_g^+$ $D_{\infty h}$
 $T_0 = 13140(80)$ gas PE¹³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1120(50)	gas	PE	13

$\tilde{a}^1\Delta_g$ $D_{\infty h}$ Structure: AB⁵
 $T_0 = 8150(80)$ gas AB⁵PE¹³ $\tilde{c}-\tilde{a}$ 330–334 nm
 $B_0 = 0.399$ AB⁵

$\tilde{X}^3\Sigma_g^-$ $D_{\infty h}$ Structure: AB¹LF¹⁰

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1197 ^a	Ar	IR	4
Π_u	2	Bend	437T 423m	gas Ar	LF IR	12 2,4
Σ_u^+	3	Asym. stretch	1466.51 1475vs 1478vs	gas Ar N ₂	IR, LMR IR IR	9,11 2,4 2–4

$B_0 = 0.397$ AB¹LF^{9,10}IR⁹

^aFrequency deduced from weak combination with ν_3 which appears at 2672.

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CNN

An absorption which appears in a nitrogen matrix at 51070 when a high concentration of CNN is present has been tentatively attributed to this species.¹⁰

$\tilde{C}^3\Pi?$	$C_{\infty v}$					
$T_0 = 48540(50)$	Ar	AB ¹⁰		$\tilde{C}-\tilde{X}$	206 nm	
	49100(50)	N ₂	AB ¹⁰	$\tilde{C}-\tilde{X}$	203.7 nm	

$\tilde{B}^3\Sigma^-?$	$C_{\infty v}$					
$T_0 \leq 40985$	gas	PF ¹⁴		$\tilde{B}-\tilde{X}$	210–251 nm	
	39950	Ar	AB ¹⁰	$\tilde{B}-\tilde{X}$	210–251 nm	
	39850	N ₂	AB ¹⁰	$\tilde{B}-\tilde{X}$	210–251 nm	

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	"Sym." stretch	1000T	gas	PF	14
			990(40)	Ar,N ₂	AB	10
	3	"Asym." stretch	1455	gas	PF	14
			1450(40)	Ar,N ₂	AB	10

$\tilde{A}^3\Pi?$	$C_{\infty v}$					
$T_0 = 23850^a$	gas	AB ⁶ LF ¹¹ PF ¹⁴		$\tilde{A}-\tilde{X}$	395–420 nm	
	23750	Ne	AB ⁵	$\tilde{A}-\tilde{X}$	397–420 nm	
	23830					
	23597	Ar	AB ^{2,3} LF ^{8,9}	$\tilde{A}-\tilde{X}$	401–424 nm	
	23865	N ₂	AB ^{2,3}	$\tilde{A}-\tilde{X}$	396–419 nm	

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	"Sym." stretch	1386	gas	PF	14
			1325(10)	Ne	AB	5
			1322(2)	Ar	AB,LF	2,3,8,9
			1335(10)	N ₂	AB	2,3
Π	2	Bend	525(2)	Ar	LF	8
Σ^+	3	"Asym." stretch	1807(2)	Ar	LF	8

$\tau_0 = 250(30)$ ns	Ar	LF ^{8,9}
$A = -26.5^b$	gas	LF ¹¹
$A = 9; \epsilon = -0.07$	Ar	LF ⁸
$B_0 = 0.425(10)^a$		LF ¹¹

$\tilde{b}^1\Sigma^+$	$C_{\infty v}$					
$T_0 = 10690(120)$	gas	PE ¹³				

$\tilde{a}^1\Delta$	$C_{\infty v}$					
$T_0 = 6830(120)$	gas	PE ¹³				

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	3	"Asym." stretch	1600(120)	gas	PE	13

$\tilde{X}^3\Sigma^-$	$C_{\infty v}$					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	"Sym." stretch	1230(120)	gas	PE	13
			1235	Ne	EM	5
			1235	Ar	LF	8,12
			1241 N ₂	Ar	IR	3
			1252	N ₂	IR	3,4,7
Π	2	Bend	390(120)	gas	PE	13
			396	Ar	LF	8,12
			393 N ₂	Ar	IR	3
			394	N ₂	IR	4,7
Σ^+	3	"Asym." stretch	1425(120)	gas	PE	13
			1419 ^b	Ar	LF	12

$$B_0 = 0.414(10)^b \text{ LF}^{11}$$

^aApproximate value, used in simulation.

^bFor many years, a prominent infrared absorption of CNN which appeared at 2847 in an argon matrix was assigned as the highest frequency fundamental, with approximate C≡N stretching character. Recently, a fluorescence signal corresponding to approximately half that value has been detected.¹² Associated CASSCF calculations indicate that the infrared absorption intensity of $2\nu_3$ of CNN should be greater than that of the fundamental.

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SiNN

$\tilde{B}^3\Sigma^-?$	$C_{\infty v}$					
$T_0 = 32162(10)$	Ar	AB ¹		$\tilde{B}-\tilde{X}$	295–311 nm	
	31892(25)	Kr	AB ²	$\tilde{B}-\tilde{X}$	297–314 nm	

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	NN stretch	1672(10)	Ar	AB	1
			1671(25)	Kr	AB	2

$\tilde{A}^3\Pi?$	$C_{\infty v}$					
$T_0 \leq 27170(20)$	Ar AB ¹					$\tilde{A}-\tilde{X}$ 331–368 nm
	Kr AB ²					$\tilde{A}-\tilde{X}$ 333–360 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	3	SiN stretch	450T	Ar	AB	1
			450T	Kr	AB	2

$\tilde{X}^3\Sigma^-?$	$C_{\infty v}$					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	NN stretch	1731.6vs	Ar	IR	1,3
			1754.7vs	N ₂	IR	3
	3	SiN stretch	484.3vw	Ar	IR	1,3
			461.6vw	N ₂	IR	3

References

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³G. Maier, H. P. Reisenauer, and J. Glatthaar, *Organomet.* **19**, 4775 (2000).

cyc-SiN₂

\tilde{X}	C_{2v}					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	NN stretch	1475.6	N ₂	IR	1
	2	SiN s-stretch	716.0	N ₂	IR	1

Reference

- ¹G. Maier, H. P. Reisenauer, and J. Glatthaar, *Organomet.* **19**, 4775 (2000).

SnCN⁻

Threshold for electron detachment from ground-state SnCN⁻ = 15510(50) gas PE¹

$\tilde{X}^3\Sigma^-$	$C_{\infty v}$					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	3	SnC stretch	330(20)	gas	PE	1

Reference

- ¹V. D. Moravec and C. C. Jarrold, *J. Chem. Phys.* **113**, 1035 (2000).

CiLiCl

$\tilde{A}^2\Pi_u$	$D_{\infty h}$	
$T^a = 2140(450)$	gas	PE ¹
$A = 810$	gas	PE ¹

$\tilde{X}^2\Pi_g$	$D_{\infty h}$	
$A = 565(450)$	gas	PE ¹

^aFrom vertical detachment energies. First VDE(CiLiCl⁻) = 47760(320), or 5.92(4) eV gas PE¹.

Reference

- ¹X.-B. Wang, C.-F. Ding, L.-S. Wang, A. I. Boldyrev, and J. Simons, *J. Chem. Phys.* **110**, 4763 (1999).

CiNaCl

$\tilde{C}^2\Sigma_g^+$	$D_{\infty h}$	
$T^a = 3790(540)$	gas	PE ¹

$\tilde{B}^2\Sigma_u^+$	$D_{\infty h}$	
$T^a = 1370(580)$	gas	PE ¹

^aFrom vertical detachment energies. Because of relatively large uncertainties in \tilde{X} and \tilde{A} state assignments, measured from first VDE. First VDE(CiNaCl⁻) = 47280(480), or 5.86(6) eV gas PE¹.

Reference

- ¹X.-B. Wang, C.-F. Ding, L.-S. Wang, A. I. Boldyrev, and J. Simons, *J. Chem. Phys.* **110**, 4763 (1999).

BrLiBr

$\tilde{C}^2\Sigma_g^+$	$D_{\infty h}$	
$T^a = 6780(340)$	gas	PE ¹

$\tilde{B}^2\Sigma_u^+$	$D_{\infty h}$	
$T^a = 6210(340)$	gas	PE ¹

$\tilde{A}^2\Pi_u$	$D_{\infty h}$	
$T^a = 2820(540)$	gas	PE ¹
$A = 1780(540)$	gas	PE ¹

$\tilde{X}^2\Pi_g$	$D_{\infty h}$	
$A = 1940(540)$	gas	PE ¹

^aFrom vertical detachment energies. Measured from first VDE. First VDE(BrLiBr⁻) = 43730(240), or 5.42(3) eV.

Reference

- ¹X.-B. Wang, C.-F. Ding, L.-S. Wang, A. I. Boldyrev, and J. Simons, *J. Chem. Phys.* **110**, 4763 (1999).

BrNaBr

$\tilde{C}^2\Sigma_g^+$ $D_{\infty h}$
 $T^a=5890(580)$ gas PE¹

$\tilde{B}^2\Sigma_u^+$ $D_{\infty h}$
 $T^a=3870(540)$ gas PE¹

$\tilde{A}^2\Pi_u$ $D_{\infty h}$
 $T^a=730(580)$ gas PE¹
 $A=1290(400)$ gas PE¹

$\tilde{X}^2\Pi_g$ $D_{\infty h}$
 $A=1940(680)$ gas PE¹

^aFrom vertical detachment energies. First VDE(BrNaBr⁻)=43250(480), or 5.36(6) eV gas PE¹.

Reference

¹X.-B. Wang, C.-F. Ding, L.-S. Wang, A. I. Boldyrev, and J. Simons, J. Chem. Phys. **110**, 4763 (1999).

ILiI

$\tilde{C}^2\Sigma_g^+$ $D_{\infty h}$
 $T^a=9200(630)$ gas PE¹

$\tilde{B}^2\Sigma_u^+$ $D_{\infty h}$
 $T^a=8470(630)$ gas PE¹

$\tilde{A}^2\Pi_u$ $D_{\infty h}$
 $T^a=1290(640)$ gas PE¹
 $A=1860(340)$ gas PE¹

$\tilde{X}^2\Pi_g$ $D_{\infty h}$
 $A=2900(540)$ gas PE¹

^aFrom vertical detachment energies. First VDE(ILiI⁻)=39370(240), or 4.88(3) eV gas PE¹.

Reference

¹X.-B. Wang, C.-F. Ding, L.-S. Wang, A. I. Boldyrev, and J. Simons, J. Chem. Phys. **110**, 4763 (1999).

INaI

$\tilde{C}^2\Sigma_g^+$ $D_{\infty h}$
 $T^a=9600(760)$ gas PE¹

$\tilde{B}^2\Sigma_u^+$ $D_{\infty h}$
 $T^a=7660(730)$ gas PE¹

$\tilde{A}^2\Pi_u$ $D_{\infty h}$
 $T^a=1210(790)$ gas PE¹
 $A=1780(400)$ gas PE¹

$\tilde{X}^2\Pi_g$ $D_{\infty h}$
 $A=2100(680)$ gas PE¹

^aFrom vertical detachment energies. First VDE(INaI⁻)=39050(480), or 4.84(6) eV gas PE¹.

Reference

¹X.-B. Wang, C.-F. Ding, L.-S. Wang, A. I. Boldyrev, and J. Simons, J. Chem. Phys. **110**, 4763 (1999).

OScO⁻

Threshold for electron detachment from ground-state OScO⁻
 =18720(160) gas PE²

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type. meas.	Refs.
b_2	3	ScO ₂ a-stretch	722.5 705.4	Ar N ₂	IR IR	1,3 1,3

References

¹G. V. Chertihin, L. Andrews, M. Rosi, and C. W. Bauschlicher, Jr., J. Phys. Chem. A **101**, 9085 (1997).

²H. Wu and L.-S. Wang, J. Phys. Chem. A **102**, 9129 (1998).

³C. W. Bauschlicher, Jr., M. Zhou, L. Andrews, J. R. Tobias Johnson, I. Panas, A. Snis, and B. O. Roos, J. Phys. Chem. A **103**, 5463 (1999).

OYO⁻

Threshold for electron detachment from ground-state OYO⁻=16140(240) gas PE¹

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type. meas.	Refs.
a_1	1	Sym. stretch	702.0	Ar	IR	2
b_2	3	Asym. stretch	618.6	Ar	IR	2

References

¹H. Wu and L.-S. Wang, J. Phys. Chem. A **102**, 9129 (1998).

²L. Andrews, M. Zhou, G. V. Chertihin, and C. W. Bauschlicher, Jr., J. Phys. Chem. A **103**, 6525 (1999).

OLaO⁻

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type. meas.	Refs.
a_1	1	Sym. stretch	656.6	Ar	IR	1
b_2	3	Asym. stretch	559.2	Ar	IR	1

Reference

¹L. Andrews, M. Zhou, G. V. Chertihin, and C. W. Bauschlicher, Jr., J. Phys. Chem. A **103**, 6525 (1999).

OZrO⁻

Threshold for electron detachment from ground-state OZrO⁻ = 14520(3200) gas PE²

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type. meas.	Refs.
b ₂	3	Asym. stretch	761.4	Ar	IR	1

References

- ¹G. V. Chertihin and L. Andrews, *J. Phys. Chem.* **99**, 6356 (1995).
²O. C. Thomas, S. Xu, T. P. Lippa, and K. H. Bowen, *J. Cluster Sci.* **10**, 525 (1999).

OVO⁻

Threshold for electron detachment from ground-state OVO⁻ = 16380(80) gas PE²

\tilde{X}		C _{2v}				
T ₀ = 2500(80)T gas PE ²						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type. meas.	Refs.
b ₂	3	Asym. stretch	896.9 894.3	Ar	IR	1

References

- ¹G. V. Chertihin, W. D. Bare, and L. Andrews, *J. Phys. Chem. A* **101**, 5090 (1997).
²H. Wu and L.-S. Wang, *J. Chem. Phys.* **108**, 5310 (1998).

ONbO⁻

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type. meas.	Refs.
b ₂	3	Asym. stretch	854.1	Ar	IR	1

Reference

- ¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **102**, 8251 (1998).

OTaO⁻

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type. meas.	Refs.
b ₂	3	Asym. stretch	817.1	Ar	IR	1,2

References

- ¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **102**, 8251 (1998).
²M. Chen, X. Wang, L. Zhang, M. Yu, and Q. Qin, *Chem. Phys.* **242**, 81 (1999).

OCrO⁻

Threshold for electron detachment from ground-state OCrO⁻ = 19470(65) gas EB¹PE^{2,4}

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type. meas.	Refs.
a ₁	1	Sym. stretch	847.1	Ne	IR	3
b ₂	3	Asym. stretch	918.7	Ne	IR	3

References

- ¹G. D. Flesch, R. M. White, and H. J. Svec, *Int. J. Mass Spectrom. Ion Phys.* **3**, 339 (1969).
²P. G. Wenthold, K.-L. Jonas, and W. C. Lineberger, *J. Chem. Phys.* **106**, 9961 (1997).
³M. Zhou and L. Andrews, *J. Chem. Phys.* **111**, 4230 (1999).
⁴G. L. Gutsev, P. Jena, H.-J. Zhai, and L.-S. Wang, *J. Chem. Phys.* **115**, 7935 (2001).

OMoO⁻

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type. meas.	Refs.
a ₁	1	Sym. stretch	883.1	Ne	IR	1
b ₂	3	Asym. stretch	837.3	Ne	IR	1

Reference

- ¹M. Zhou and L. Andrews, *J. Chem. Phys.* **111**, 4230 (1999).

OWO⁻

Threshold for electron detachment from ground-state OWO⁻ ≤ 16120(80) gas PE¹

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type. meas.	Refs.
a ₁	1	Sym. stretch	952.3 946.3	Ne Ar	IR IR	2 2
b ₂	3	Asym. stretch	887.8 880.0	Ne Ar	IR IR	2 2

References

- ¹G. E. Davico, R. L. Schwartz, T. M. Ramond, and W. C. Lineberger, *J. Phys. Chem. A* **103**, 6167 (1999).
²M. Zhou and L. Andrews, *J. Chem. Phys.* **111**, 4230 (1999).

OMnO⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>b</i> ₂	3	Asym. stretch	858.2	Ar	IR	1
			866.7T	N ₂	IR	1

Reference

¹G. V. Chertihin and L. Andrews, J. Phys. Chem. A **101**, 8547 (1997).

OReO⁻

Threshold for electron detachment from ground-state OReO⁻ = 20200(800) gas PE²

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	3	Asym. stretch	893.8	Ne	IR	1
			885.5	Ar	IR	1

References

¹M. Zhou, A. Citra, B. Liang, and L. Andrews, J. Phys. Chem. A **104**, 3457 (2000).

²A. Pramann and K. Rademann, Chem. Phys. Lett. **343**, 99 (2001).

ORuO⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	3	Asym. stretch	860.6	Ne	IR	1
			851.8	Ar	IR	1

Reference

¹M. Zhou, A. Citra, B. Liang, and L. Andrews, J. Phys. Chem. A **104**, 3457 (2000).

OOsO⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	3	Asym. stretch	897.5	Ne	IR	1

Reference

¹M. Zhou, A. Citra, B. Liang, and L. Andrews, J. Phys. Chem. A **104**, 3457 (2000).

AgOO⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OO stretch	1030.7	Ne	IR	1

Reference

¹X. Wang and L. Andrews, J. Phys. Chem. A **105**, 5812 (2001).

OCeO⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	Sym. stretch	712.0	Ar	IR	1
<i>b</i> ₂	3	Asym. stretch	662.0	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

OPrO⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	Sym. stretch	665.0	Ar	IR	1
<i>b</i> ₂	3	Asym. stretch	667.6	Ne	IR	1
			653.8	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

ONdO⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>b</i> ₂	3	Asym. stretch	660.6	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

OSmO⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	676.4	Ar	IR	1
b ₂	3	Asym. stretch	575.5	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

OEuO⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	661.0	Ar	IR	1
b ₂	3	Asym. stretch	560.8	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

OGdO⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	685.9	Ar	IR	1
b ₂	3	Asym. stretch	589.4	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

OTbO⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	711.2	Ar	IR	1
b ₂	3	Asym. stretch	669.0	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 6972 (1999).

ODyO⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	693.9	Ar	IR	1
b ₂	3	Asym. stretch	591.2	Ne	IR	1
			574.6	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 6972 (1999).

OHoO⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	696.2	Ar	IR	1
b ₂	3	Asym. stretch	547.2	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 6972 (1999).

OErO⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	702.3	Ar	IR	1
b ₂	3	Asym. stretch	613.4	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 6972 (1999).

OYbO⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	701.2	Ar	IR	1
b ₂	3	Asym. stretch	604.2	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 6972 (1999).

OLuO⁻

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b_2	3	Asym. stretch	626.9T	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **103**, 6972 (1999).

OUO⁻

\tilde{X}		D_{zh}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_u^+	3	Asym. stretch	857.2	Ne	IR	1

Reference

¹M. Zhou, L. Andrews, N. Ismail, and C. Marsden, *J. Phys. Chem. A* **104**, 5495 (2000).

BO₂

$\tilde{B}^2\Sigma_u^+$		D_{zh}		Structure: AB ¹		
$T_0 = 24508.0$	gas	AB ¹	$\tilde{B}-\tilde{X}$	405–410 nm		
24529.5	Ne	AB ¹⁶				
24481	Ar	AB ²	$\tilde{B}-\tilde{X}$	408–412 nm		

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π_u	2	Bend	505	gas	AB	1
Σ_u^+	3	Asym. stretch	1410 ^a	gas	AB	1

$B_0 = 0.325$ AB¹

$\tilde{A}^2\Pi_u$		D_{zh}		Structure: UV ¹		
$T_0 = 18291.597$	gas	UV ¹ LF ^{3,5,14}	$\tilde{A}-\tilde{X}$	396–700 nm		
18162	Ne	AB ¹⁶ LF ¹⁶	$\tilde{A}-\tilde{X}$	419–1465 nm		
17915 ^b	Ar	AB ²	$\tilde{A}-\tilde{X}$	423–558 nm		

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	994	gas	UV	1
			1027	Ne	AB	16
Π_u	2	Bend (ω)	477.29	gas	UV,LF	1,14
		($\kappa^2\Sigma_u^-$)	633.15	gas	UV,LF	1,14
		($^2\Delta_{u,3/2}$)	588.21	gas	UV,LF	1,14
		($^2\Delta_{u,5/2}$)	442.07	gas	UV,LF	1,14
		($\mu^2\Sigma_u^+$)	404.96	gas	UV,LF	1,14
Σ_u^+	3	Asym. stretch	2357H	gas	UV	1
			2361H	Ne	AB	16

$\tau_0 = 91(4)$ ns gas LF^{6,9}

A systematic study of the dependence of τ on rotational and vibrational level has been given by Ref. 10.

$A = -101.281$; $\epsilon\omega_2 = -13.896$ gas UV¹LF¹⁴

$B_0 = 0.311$ UV¹LF^{13,14}

$\tilde{X}^2\Pi_g$		D_{zh}		Structure: UV ¹		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1056.4	gas	UV,LF	1,3–5
			1058	Ne	LF	16
Π_u	2	Bend (ω)	448.18	gas	UV,LF	1,3–5,14
		($\kappa^2\Sigma_u^-$)	633.80	gas	DL	11
			446.6wm	Ar	IR	15
			398.2m	Ar	IR	15
			493.4	Kr	IR	15
			394.6	Kr	IR	15
Σ_u^+	3	Asym. stretch	1278.26	gas	DL,IR	8,12
			1322H	Ne	LF	16
			1299.3vs	Ar	IR	15
			1282.8s	Ar	IR	15
			1274.6s	Ar	IR	2,15
			1300.7vs	Kr	IR	15
			1285.5s	Kr	IR	15
			1292.2vs	Xe	IR	15
			1279.0	O ₂	IR	15

$A = -148.6$; $\epsilon\omega_2 = -86.91$ gas UV¹LF^{3,7,14}

$B_0 = 0.329$ UV¹LF^{3,7,13,14}DL⁸IR¹²

^aEstimated from isotopic shifts.

^bIndependent analysis of the matrix spectrum not given. Each argon-matrix absorption is shifted to lower frequency by approximately 400 cm⁻¹ from the corresponding gas-phase R_1 branch band head.

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AINO⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO stretch	1380.6	Ar	IR	1

Reference

¹L. Andrews, M. Zhou, and W. D. Bare, *J. Phys. Chem. A* **102**, 5019 (1998).

cyc-TiO₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	OO stretch	1082.0	Ar	IR	1-3
			1089	N ₂	IR	1
			1081	O ₂	IR	1
	2	OTiO s-stretch	295.2s	Ar	IR	1-3
			296	N ₂	IR	1
			296	O ₂	IR	1

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OTiO

\tilde{X} D _{∞h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _u ⁺	3	Asym. stretch	698.0	Ar	IR	1

Reference

¹L. Andrews, G. P. Kushto, J. T. Yustein, E. Archibong, R. Sullivan, and J. Leszczynski, *J. Phys. Chem. A* **101**, 9077 (1997).

CCO⁻

Threshold for electron detachment from ground-state CCO⁻ is 18640(100) gas PE^{2,3}

$\tilde{A}^2\Sigma^+$ C _{∞v}						
T ₀ = 12234	Ne	AB ⁴			$\tilde{A}-\tilde{X}$ 541-818 nm	
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	CO stretch	2082(3)	Ne	AB	4
Π	2	Bend	656(3)	Ne	AB	4
Σ ⁺	3	CC stretch	1185(3)	Ne	AB	4

\tilde{X} C _{∞v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	CO stretch	1900T 1876.7(1.0)	gas Ne	PE IR	1,2 4

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SiCO⁻

\tilde{X} C _{∞v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	CO stretch	1706.7	Ar	IR	1

Reference

¹L. Zhang, J. Dong, and M. Zhou, *J. Chem. Phys.* **113**, 8700 (2000).

GeCO⁻

\tilde{X} C _{∞v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	CO stretch	1708.7 1705.2 1701.0	Ar	IR	1

Reference

¹L. Zhang, J. Dong, and M. Zhou, *J. Chem. Phys.* **113**, 8700 (2000).

SnCO⁻

\tilde{X} C _{∞v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	CO stretch	1729.1	Ar	IR	1

Reference

¹L. Zhang, J. Dong, and M. Zhou, *J. Chem. Phys.* **113**, 8700 (2000).

NCN⁻

Threshold for electron detachment from ground-state NCN⁻ = 20040(50) gas PE^{1,2}

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NCO

$\tilde{B}^2\Pi_{3/2}$	$C_{\infty v}$				
$T_0 = 31768.5$ T	gas	$UV^2LF^{18,30,31}PD^{27}$		$\tilde{B}-\tilde{X}$ 265–320 nm	
31616(25)	Ne	UV^3		$\tilde{B}-\tilde{X}$ 260–320 nm	
31437(25)	Ar	UV^3		$\tilde{B}-\tilde{X}$ 232–315 nm	
31339(25)	N_2	UV^3		$\tilde{B}-\tilde{X}$ 256–315 nm	

All vibrational states are predissociated.^{27,33}

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ^+	1	Stretch	2303	gas	UV	2
			2295(50)	Ne	UV	3
			2303(50)	Ar	UV	3
			1047	gas	UV	2
			1033(50)	Ne	UV	3
3	Stretch	1053(50)	Ar	UV	3	
		1025(50)	N_2	UV	3	

$\tau_0 = 63(3) - 152(5)$ ns for various rovibronic bands gas $LF^{13,30,31}$
 $A = -76.6$ gas LF^{18}
 $B_0 = 0.356$ LF^{18}

$\tilde{A}^2\Sigma^+$	$C_{\infty v}$		Structure: UV^{20}		
$T_0 = 22754.020(2)$	gas	$AB^1LF^{23,30,31}SEP^{26}EM^{28}$		$\tilde{A}-\tilde{X}$ 304–512 nm	
22800(10)	Ne	AB^3		$\tilde{A}-\tilde{X}$ 398–440 nm	
22712(2)	Ar	LF^8		$\tilde{A}-\tilde{X}$ 390–530 nm	
22956(10)	N_2	AB^3		$\tilde{A}-\tilde{X}$ 395–440 nm	

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ^+	1	Stretch	2338.0	gas	UV	1
			2325(20)	Ne	UV	3
			2332(4)	Ar	UV,LF	3,8
			2321(20)	N_2	UV	3
			680.8	gas	UV	1
II	2	Bend	673(20)	Ne	UV	3
			1289.3 ^a	gas	UV	1
Σ^+	3	Stretch	1270(20)	Ne	UV	3
			1291(4)	Ar	UV,LF	3,8

$\tau_0 = 435(10)$ ns gas $LF^{9,13}$
 350(30) ns gas $LF^{11,12}$
 170 ns Ar LF^8
 $B_0 = 0.402$ UV^1

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.			
Σ^+	1	Sym. stretch	1266.63(8) ^b	gas	LF,LMR,SEP	14,15,17,19,23,26,29			
			1275vw	Ar	IR,LF	3,8			
			1269.0	Xe	IR	32			
			534.06(3)	gas	UV,LF,SEP	7,15,23,29			
			669.28	gas	UV,LF	7,15,23			
II	2	Bend (ω) ($^2\Sigma^-$)	672	Ar	LF	8			
			($^2\Delta_{3/2}$)	628.45	gas	UV,LF	7,15,23		
			626	Ar	LF	8			
			($^2\Delta_{5/2}$)	534.64	gas	UV,LF	7,15,23		
			529.5	Ar	LF	8			
			($^2\Sigma^+$)	488.43	gas	UV,LF	7,15,23		
			487	Ar	IR,LF	3,8			
			Σ^+	3	Asym. stretch	1921.28 ^c	gas	LMR,LF,DL,EM,SEP	10,14,15,21,23,28,25
						1923m	Ar	IR,LF	3,8
						1921.6	Kr	IR	32
1917.8	Xe	IR				32			
1915.8									
1935	N_2	IR	3						

$A_0 = -95.589(3)$, $\epsilon\omega_2 = -78.37(3)$ gas $LF^{23}SEP^{26,29}EM^{28}$
 $B_0 = 0.390$ $UV^1MW^{16}LMR^{22}EM^{28}$

^aIn Fermi resonance with $2\nu_2$, at 1385.3.

^bFor $\tilde{X}^2\Pi_{1/2}$, 1362.87 gas $LF^{14,15,19,23,24}SEP^{26}$

^cFor $\tilde{X}^2\Pi_{1/2}$, 2017.7(5) gas $LF^{14,15,23,24}$

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NCS

$\tilde{B}^2\Sigma^+$ $C_{\infty v}$
 $T_0 = 26987.8$ gas $EM^1AB^2LF^{4,6}$ $\tilde{B}-\tilde{X}$ 353–485 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	2	Bend	343(10)	gas	AB	2
Σ^+	3	CS stretch	921.5	gas	LF	4

$\tau_{001} = 225(5)$ ns gas LF⁴
 $B_0 = 0.197$ AB²

$\tilde{A}^2\Pi$ $C_{\infty v}$
 $T_0 = 26054.56(1)$ gas $EM^1AB^2LF^{4,6}$ $\tilde{A}-\tilde{X}$ 337–417 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CN stretch	1920.61	gas	AB,LF	2,4
Π	2	Bend ($\kappa^2\Sigma^-$)	487.2	gas	LF	4
		($^2\Delta_{3/2}$)	445.3T	gas	LF	4
		($^2\Delta_{5/2}$)	363.7	gas	LF	4
		($\mu^2\Sigma^+$)	304.5	gas	LF	4
Σ^+	3	CS stretch	771.12	gas	LF	4

$\tau_0 = 160(5)$ ns gas LF^{3,4}EF⁵
 $A = -91.58(1)$; $|\epsilon\omega_2| = 103(5)$ gas AB²LF⁴
 $B_0 = 0.191$ AB²LF⁶

$\tilde{X}^2\Pi$ $C_{\infty v}$ Structure: AB²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CN stretch	1942.2	gas	LF	4
Π	2	Bend (ω)	376.0(3)	gas	AB,LF,SEP	2,4,6
		($\kappa^2\Sigma^-$)	704.84	gas	LF	4,6
		($^2\Delta_{3/2}$)	688.17	gas	LF	4,6
		($^2\Delta_{5/2}$)	375.1	gas	LF	4,6
		($\mu^2\Sigma^+$)	364.44	gas	LF	4,6
Σ^+	3	CS stretch	761.9 ^a	gas	SEP	6

$A = -327.6(2)$; $\epsilon\omega_2 = -60.06(44)$ gas AB²LF^{4,6}SEP⁶
 $B_0 = 0.204$ AB²LF^{4,6}MW⁷

^aIn Fermi resonance with $2\nu_2$.

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AsCO

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	1919.3	Ar	IR	1

Reference

- ¹L. Zhang, J. Dong, and M. Zhou, *Chem. Phys. Lett.* **335**, 334 (2001).

CO₂⁺

$\tilde{C}^2\Sigma_g^+$ $D_{\infty h}$ Structure: MP²⁴
 $T_0 = 45157(3)$ gas TPE²¹PE²³MP²⁴ $\tilde{C}-\tilde{A}$ 585–640 nm
 The band origin is perturbed by another state of $^2\Sigma_g^+$ symmetry at 45188 which has a rotational constant of 0.353.²⁴ This state may be an excited vibrational level of the \tilde{A} or \tilde{B} state.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1352(4) ^a	gas	PE	23
Π_u	2	Bend	614(4)	gas	TPE,PE	21,23
Σ_u^+	3	Asym. stretch	1567(4)	gas	PE	23

$B_0 = 0.395$ MP²⁴

$\tilde{B}^2\Sigma_u^+$ $D_{\infty h}$ Structure: EM⁹
 $T_0 = 34591.6^b$ gas $EM^{1,9,29,30}EF^{29}$ $\tilde{B}-\tilde{X}$ 287–291 nm
 Perturbations by the \tilde{A} state are considered in Refs. 14–16 and Ref. 29.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1284(10)	gas	TPE,PE	21,23,36
Π_u	2	Bend	590(10)	gas	EM,PE,TPE	9,23,36
Σ_u^+	3	Asym. stretch	1891(10)	gas	PE,EM,TPE	17,23,30 36

$\tau_0 = 140(7)$ ns gas T-PEFCO¹⁰PEFCO¹³LF¹⁶
 $B_0 = 0.380$ EM^{1,29}

$\tilde{A}^2\Pi_u$ $D_{\infty h}$ Structure: EM¹¹
 $T_0 = 28500.35$ gas EM^{2.6,11,30}LF^{27,30}PE³¹ $\tilde{A}-\tilde{X}$ 253–490 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1126	gas	EM,TPE	2,5,11 23,31,36
Π_u	2	Bend (ω)	461	gas	EM,PE	11,23,31
		($\kappa^2\Sigma_u^-$)	568.8	gas	EM,TPE	23,31,36
		($\mu^2\Sigma_u^+$)	440.9	gas	EM,TPE	23,31,36
Σ_u^+	3	Asym. stretch	2685.3	gas	LF,EM,TPE	30,36

$\tau_0 = 102(8)$ ns gas EF⁷T-PEFCO¹⁰
 124(6) ns gas PEFCO¹³HFD¹⁸
 $A = -95.51$ gas EM¹¹LF²⁷PE³¹
 $B_0 = 0.350$ EM^{2,11}LF²⁷

$\tilde{X}^2\Pi_g$ $D_{\infty h}$ Structure: EM^{2-5,9,11}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1244.3(3)	gas	EM,DL PE,TPE	4,5,8,12 22,31,35
Π_u	2	Bend (ω)	511.4(3)	gas	EM,DL PE,TPE	11,22,28, 31,35
		($\kappa^2\Sigma_u^-$)	719.17	gas	EM,DL,TPE	11,22,35
		($^2\Delta_{3/2}$)	668.04	gas	EM,DL,TPE	11,22,35
		($^2\Delta_{5/2}$)	511.60	gas	EM,DL,TPE	11,22,35
		($\mu^2\Sigma_u^+$)	467.26	gas	EM,DL,TPE	11,20,22,35
Σ_u^+	3	Asym. stretch	1423.08	gas	DL,PE,TPE	19,31,35
			1421.7	Ne	IR	25,32–34

$A = -161.02(6)$; $\epsilon\omega_2 = -98.8(3)$ gas EM^{1,9,11}DL^{20,22}TPE²⁸ (Reanalysis by Ref. 26 gives $A = -161.48(5)$ and $\epsilon\omega_2 = -100.4$)
 $B_0 = 0.380$ EM^{1,3,9,11}

^aCorrected for Fermi resonance.

^bMeasured from lowest rotational level of \tilde{X} state,²⁹ 34672.33.

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OCS⁺

$\tilde{C}^2\Sigma^+$ $C_{\infty v}$
 $T_0 = 54640(30)$ gas PI⁴PE¹⁰

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	2202(2)	gas	PE	10
Π	2	Bend	454(5)	gas	PE	10
Σ^+	3	CS stretch	926(5)	gas	PI,PE	4,10

$\tilde{B}^2\Sigma^+$ $C_{\infty v}$
 $T_0 = 39180(20)$ gas PI⁴PF⁹PE¹⁰

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	1850(8)	gas	PE	10
Π	2	Bend	515(3)	gas	PF,PE	9,10
Σ^+	3	CS stretch	829 ^a	gas	PF	9

$\tilde{A}^2\Pi_{3/2}$ $C_{\infty v}$
 $T_0 = 31404.099(7)$ gas EF¹LF⁸PF^{9,11} $\tilde{A}-\tilde{X}$ 282–432 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	2027	gas	PE,PF	10,11
Π	2	Bend	336(20)H	gas	PE	10
Σ^+	3	CS stretch	807 ^b	gas	PF	9,11

$\tau_0 = 93(9)$ ns^c gas PEFCO⁵
 $\tau_0(\omega = 3/2) = 105(3)$ ns; $\tau_0(\omega = 1/2) = 77(3)$ ns gas HFD⁶EF⁷
 $A = -111.8$ gas EF¹PF^{9,11}
 $B_0 = 0.186$ LF⁸PF^{9,11}

$\tilde{X}^2\Pi_{3/2}$		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	2066	gas	EF,PF,TPE	1,11,12
			2071.1	Ne	IR	13
Π	2	Bend	476(16)	gas	PE	10
		($\kappa^2\Sigma^-$)	830	gas	TPE	12
		($^2\Delta_{3/2}$)	792	gas	TPE	12
		($\mu^2\Sigma^+$)	419	gas	TPE	12
Σ^+	3	CS stretch	697 ^d	gas	PF,TPE	9,11,12
			702.5	Ne	IR	13

$A = -367.2$ gas EF¹PF⁹TPE¹²

$B_0 = 0.194$ LF⁸PF^{9,11}

^aRef. 10 gives value of 742(7).

^b817 for $\omega = 1/2$.^{9,11}

^cAbsence of emission from states above the $\tilde{A}^2\Pi$ band origin in photoionization experiments² suggested that the molecule is predissociated into CO+S⁺(⁴S^o), as was later confirmed.³ PEFCO studies⁵ have yielded the branching ratio for photoexcitation vs. predissociation for the transition origin, permitting an estimate of 550(50) ns for the radiative lifetime.

^d699.7 for $\omega = 1/2$.^{9,11}

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CS₂⁺

$\tilde{C}^2\Sigma_g^+$		$D_{\infty h}$		Structure: MP ¹⁵		
$T_0 = 49064$		gas	PI ⁵ PF ¹³ MP ¹⁵	$\tilde{C}-\tilde{B}$ 658–724 nm		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	652(2)	gas	PI,PE, PF,MP	5,9,11
						13–15,19
Π_u	2	Bend	348(9)	gas	PF,PE	13,14,19
Σ_u^+	3	Asym. stretch	1024(6)T	gas	PE	14

$\tau_0 = 11(2)$ ps gas MP¹⁵
 $B_0 = 0.111$ PF¹³MP¹⁵

$\tilde{B}^2\Sigma_u^+$ $D_{\infty h}$ Structure: EM¹
 $T_0 = 35238.01$ gas EM¹
 35270 35226 Ne LF⁷ $\tilde{B}-\tilde{X}$ 277–307 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	602	gas	EM,PE	3,19
Π_u	2	Bend	351(5)	gas	PE	14,19
Σ_u^+	3	Asym. stretch	1320(5)	gas	PE	14,19

$\tau_0 = 290(10)$ ns gas EF²PIFCO⁴PEFCO⁸UV¹²
 There is also a long-lifetime component, with $\tau = 1.44(22)$ μ s.^{8,12}

$B_0 = 0.108$ EM¹

$\tilde{A}^2\Pi_u$ $D_{\infty h}$ Structure: EM³
 $T_0 = 20975$ gas EM³LF²² $\tilde{A}-\tilde{X}$ 426–512 nm
 21010.6 Ne LF^{6,7,17,24} $\tilde{A}-\tilde{X}$ 400–642 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	613(5) ^{ab}	gas	PE	25
			551 ^b	Ne	LF	6,7,18,24
Π_u	2	Bend ($\kappa^2\Sigma_u^-$)	510(5)	gas	PE	25
		($^2\Delta_{u,5/2}$)	286(5)	gas	PE	25
			309H	Ne	LF	6,7,18,24
Σ_u^+	3	Asym. stretch	1723(5) ^a	gas	PE	25
			1644H	Ne	LF	7

$\tau = 4.09(19)$ μ s gas PIFCO⁴ID¹⁰UV¹²
 2.3(1) μ s Ne LF^{6,7}

$A = -177$ gas EM³PE^{14,19,25}LF²²

$B_0 = 0.101$ EM³

$\tilde{X}^2\Pi_g$		$D_{\infty h}$		Structure: EM ¹		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	617 ^b	gas	EM,TPE	3,16,21
			618 ^b	Ne	LF	6,7,17,18
Π_u	2	Bend	332	gas	TPE	16,21
		($^2\Delta_{u,5/2}$)	333.8	gas	TPE	21
		($\mu^2\Sigma_u^+$)	328.3	gas	TPE	21
			349H	Ne	LF	6,7,17,18
Σ_u^+	3	Asym. stretch	1188(8)	gas	EM,PE	3,14,19
			1207.1	Ne	LF,IR	6,7,17,18
						20,23,24
			1200.5	Ar	IR	23

$A = -440.39(3)$ gas EM^{1,2}TPE^{16,21}

$B_0 = 0.109$ EM¹

^aVibrational fundamentals given for $\tilde{A}^2\Pi_{3/2}$. For $\tilde{A}^2\Pi_{1/2}$, Ref. 25 gives $\nu_1 = 740(5)$ and $\nu_3 = 1870(5)$.

^bStrong Fermi resonance with $2\nu_2$.

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CICN⁺

$\tilde{C}^2\Sigma^+$ $C_{\infty v}$
 $T_0 = 54000(300)$ gas PE^{1,2}

$\tilde{B}^2\Pi_{3/2}$ $C_{\infty v}$ Structure: UV, PE³LF¹²
 $T_0 = 22515.54$ gas EF⁸LF¹⁰⁻¹²PE¹³ $\tilde{B}-\tilde{X}$ 365–569 nm
 22598(5) Ne AB⁶ $\tilde{B}-\tilde{X}$ 380–442 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CN stretch	2128.5(7)	gas	LF	10
Π	2	Bend	303.1(7)	gas	LF	10
Σ^+	3	CCl stretch	531.90	gas	LF, PE	10, 11, 13
			539(4)	Ne	AB	6

$\tau_1 = 205(40)$ ns gas EF³
 280(56) ns gas PEFCO⁵
 170(20) ns gas PIFCO^{4,7}

$\tau_2 = 900(100)$ ns gas EF³
 970(80) ns gas PIFCO⁷

Both lifetimes are dependent on extent of vibrational excitation.⁵

$A = -368(2)$ gas EF^{8,9}LF¹⁰

$B_0 = 0.177$ LF¹¹

$\tilde{A}^2\Sigma^+$ $C_{\infty v}$
 $T_0 = 11690(1)$ gas EF^{3,8}PE¹³ $\tilde{A}-\tilde{X}$ 843–881 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	3	CCl stretch	774(2)	gas	EF, PE	8, 13

$\tau = 4.4(1.0)$ μ s gas EF³

$\tilde{X}^2\Pi_{1/2}$ $C_{\infty v}$
 $T_0 = 276(2)$ gas EF^{3,8,9}PE¹³ $\tilde{A}-\tilde{X}$ 843–881 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	C≡N stretch	1914(2)	gas	EF, PE	8, 13
	3	CCl stretch	827(2)	gas	EF, PE	8, 13

$\tilde{X}^2\Pi_{3/2}$ $C_{\infty v}$ Structure: UV, PE³LF¹²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	C≡N stretch	1915(2)	gas	EF, LF, PE	8–10, 13
Π	2	Bend	376T	gas	LF	10
Σ^+	3	CCl stretch	827(2)	gas	EF, LF, PE	8–10, 13

$B_0 = 0.205$ LF¹¹

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CNN⁻

Threshold for electron detachment from ground-state
 $CNN^- = 14290(80)$ gas PE¹

Reference

- ¹E. P. Clifford, P. G. Wenthold, W. C. Lineberger, G. A. Petersson, K. M. Broadus, S. R. Kass, S. Kato, C. H. DePuy, V. M. Bierbaum, and G. B. Ellison, *J. Phys. Chem. A* **102**, 7100 (1998).

N₃

$\tilde{B}^2\Sigma_u^+$ D_{∞h} Structure: AB²
 $T_0^a = 36738.750(2)$ gas AB^{1,2}LF^{3,8}PD⁷ $\tilde{B}-\tilde{X}$ 260–273 nm

All bands above 37000 are diffuse. Fast beam photodissociation studies⁷ show that predissociation into N(²D)+N₂(X¹Σ_g⁺) occurs throughout the 260–273 nm spectral region.

An absorption maximum was observed at 272 nm in mixed argon-nitrogen matrix studies.⁶

$\tau \geq 0.37, \leq 20$ ns gas LF^{3,8}
 $B_0 = 0.432$ gas AB²

$\tilde{X}^2\Pi_g$		D _{∞h}		Structure: AB ² IR ⁶		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	1	Sym. stretch	1320T	gas	LF	3
			1287 ^b	N ₂ ^c	IR	4
Π _u	2	Bend	457T	gas	LF	3
			472.0	N ₂	IR	4,9
Σ _u ⁺	3	Asym. stretch	1644.68	gas	LMR,IR	5,6
			1636.1	Ar	IR	10
			1657.7	N ₂	IR	4,9,10

$A_{\text{eff}} = -71.3; \epsilon\omega_2 = -94.38$ gas AB²IR⁶
 $B_0 = 0.431$ AB²LMR⁵IR⁶

^aRevised value resulting from reanalysis by Ref. 6.

^b($\nu_1 + \nu_3$) - ν_3 .

^cMixed with argon.

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As₃

\tilde{X}^2E'' D_{3h}
 Jahn–Teller splitting occurs, with the ²A₂ and ²B₁ levels separated by ~2900 cm⁻¹ gas PE¹

Reference

- T. P. Lippa, S.-J. Xu, S. A. Lyapustina, J. M. Nilles, and K. H. Bowen, J. Chem. Phys. **109**, 10727 (1998).

N₂O⁺

$\tilde{C}^2\Sigma^+$ C_{∞v}
 $T_0 = 58245(32)$ gas PE^{1,8}PI⁵

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	Sym. stretch	1242(24)	gas	PE	1,8,21
Π	2	Bend	508(16)	gas	PE	8
Σ ⁺	3	Asym. stretch	2283(24)	gas	PE	1,8,21

$\tilde{B}^2\Pi$ C_{∞v}
 $T_0 = 38440(100)^a$ gas PE¹PF¹⁷ $\tilde{B}-\tilde{A}$ 538–866 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	Sym. stretch	960	gas	PE	1,13
Π	2	Bend	368	gas	PE	13

$\tilde{A}^2\Sigma^+$ C_{∞v} Structure: EM³PF¹⁷
 $T_0 = 28162.33$ gas EM^{3,16}PF^{6,11,12,18,20} $\tilde{A}-\tilde{X}$ 317–421 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	Sym. stretch	1345.38	gas	EM,PF	3,11,22
Π	2	Bend	614.45	gas	EM	3,16
Σ ⁺	3	Asym. stretch	2451.7	gas	EM	3

$\tau = 230(10)$ ns gas EF^{2,10}PIFCO⁴PEFCO⁷ID⁹EM¹⁴HFD¹⁵
 $B_0 = 0.433$ EM^{3,16}PF^{11,12}

$\tilde{X}^2\Pi$		C _{∞v}		Structure: EM ³		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	Sym. stretch	1126.28	gas	EM	3,22
			1135.5	Ne	IR	19
Π	2	Bend	452.42	gas	EM,PF	3,12,16
Σ ⁺	3	Asym. stretch	1737.37	gas	EM	3,22
			1741.4wm	Ne	IR	19

^aCalculated using the first ionization potential of 12.886(2) eV, from Ref. 5.

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²²C. E. Fellows and M. Vervloet, *Chem. Phys.* **264**, 203 (2001).

NiCl₂

In an argon matrix,¹¹ a relatively weak absorption has its onset near 41500.

UV3

$T_0 = 34652$ Ar AB^{5,10,11} 270–289 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	Sym. stretch	290(15)	Ar	AB	5,10,11

UV2

$T_0 = 30760$ Ar AB^{5,10,11} 298–325 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			250T	Ar	AB	5,10,11

UV1

gas AB^{1,3}LF^{19,21} 353–374 nm

Molecular beam studies of this spectral region with vibrational¹⁹ and rotational²¹ resolution reveal an extremely complicated pattern of transitions for which a definitive assignment has not been achieved. It has been suggested that several mutually perturbing electronic states of NiCl₂ lie in this spectral region.

$T_0 = 27166$ Ar AB^{5,10,11} 327–368 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	Sym. stretch	260(15)	Ar	AB	5,10,11

$\tau = 0.25(2)$ μ s gas LF¹⁹

³ Δ_g ?	D _{∞h}	Structure: LF ²⁰	
$T_0 = 21480.64$	gas AB ^{1,3,6,7} LF ^{16,17,20,22}		³ A _g - \tilde{X} 439–495 nm
20500(2) ^a	gas LF ²²		³ A _g - \tilde{A} 487–539 nm
21590	Ne AB ¹⁵		³ A _g - \tilde{X} 439–486 nm
21465	Ne EM ¹⁵		
19405	Ne EM ¹⁵		³ A _g - \tilde{A} 515–591 nm
21300	Ar AB ¹⁵		³ A _g - \tilde{X} 446–491 nm
21257	Ar EM ^{8,11}		
19537	Ar EM ^{8,11}		³ A _g - \tilde{A} 509–529 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	357.21 (ω)	gas	LF	16,17,20
			355T	Ne	AB	15
			350T	Ar	AB	15
Π_u	2	Bend	60.66 (ω)	gas	LF	17,20
Σ_u^+	3	Asym. stretch	530.95 (ω)	gas	LF	20

$\tau > 1.2$ μ s gas LF¹⁶
= 3.5(2) μ s Ne EM¹⁵

$B_0 = 0.055$ LF²⁰

In an argon matrix,⁸ weak emission appears between 582 and 583 nm.

In a neon matrix,¹⁵ structured absorption appears between 595 and 648 nm, and, in an argon matrix,¹⁵ between 610 and 632 nm.

In the gas phase,^{6,7} an absorption maximum appears near 14510 (689 nm).

In the gas phase,^{3,6,7} a relatively weak absorption maximum has been observed near 13000. In a neon matrix,¹⁵ absorption bands are seen at 12783 and 12808 (780–783 nm). In an argon matrix,¹⁵ three bands appear between 12470 and 12635 (791–802 nm).

In the gas phase,^{6,7} a relatively sharp absorption maximum has been observed at 11727.

In the gas phase,^{3,6,7} an absorption maximum has been observed near 4000. The absorption extends up to at least 8000.

\tilde{A} ³ Π_g	D _{∞h}	
$T_0 = 1572^a$	gas LF ²²	³ A _g - \tilde{A} 487–539 nm
2060	Ne EM,LF ¹⁵	³ A _g - \tilde{A} 515–591 nm
1720T	Ar EM ^{8,11}	³ A _g - \tilde{A} 509–529 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	366.3(8)	gas	LF	22
			370	Ne	EM,LF	15
			360	Ar	EM	8,11

\tilde{X} ³ Σ_g^- D_{∞h} Structure: LF²⁰

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	359.2(3)	gas	LF	14,16,20,22
			357	Ne	Ra	15
			360	Ar	EM,Ra	8,11,15
Π_u	2	Bend	87.28	gas	LF	14,17,20,22
			85	Ar	IR	9
Σ_u^+	3	Asym. stretch	520.4H	gas	IR,LF	2,4,14,20,22
			529.6	Ne	IR	12
			520.9	Ar	IR	5,12,13,18

$B_0 = 0.057$ LF²⁰

^a0⁺–0⁺ band separation.

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CuCl₂

In the gas phase, an absorption maximum has been observed³ near 44800.

$2^2\Pi_g$ $D_{\infty h}$
 $T_0^a = 19900(1000)$ gas PE¹⁶

$2^2\Sigma_u^+$ $D_{\infty h}$
 $T_0^a = 17800(900)$ gas PE¹⁶

$2^2\Pi_u$ $D_{\infty h}$
 $T_0 = 15500T$ gas AB^{1,3}LF^{7-12,15}PE¹⁶
 15416.0 Ne AB¹³LF¹³
 14920.5 Ar AB⁴LF¹³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			240T	Ne	AB	13
			255T	Ar	AB	13

$2^2\Delta_{g,3/2}$ $D_{\infty h}$

From maxima in the photoelectron spectrum¹⁶ of CuCl₂⁻, the origin of this transition is estimated at 8700 cm⁻¹. A revised estimate of 9450 cm⁻¹ has also been suggested.¹⁷

$T_0 = 9567.5$ Ne LF¹³
 9427T Ar LF¹³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	364.5	Ne	LF	13
			354.6T	Ar	LF	13
Π_u	2	Bend	80.2	Ne	LF	13

$2^2\Delta_{g,5/2}$ $D_{\infty h}$

In the gas phase, absorption^{1,3,6} has been reported between 760 and 1454 nm and emission^{5,6} between 1150 and 1460 nm, with the origin estimated at ≤ 6877 . From maxima in the photoelectron spectrum¹⁶ of CuCl₂⁻, the origin of this transition is estimated at 7100 cm⁻¹. A revised estimate of 7780 cm⁻¹ has also been suggested.¹⁷

$T_0 = 7893.5$ Ne LF¹³
 7753.1 Ar LF¹³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	360(15)	gas	AB	3
			360.6(ω)	Ne	LF	13
Π_u	2	Bend	88.76(ω)	Ne	LF	13

$2^2\Sigma_g^+$ $D_{\infty h}$
 $T_0 = 1910.9$ Ne LF¹³
 1616.5 Ar LF¹³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	389.4(ω)	Ne	LF	13
			391.7T (ω)	Ar	LF	13
Π_u	2	Bend	102.6(ω)	Ne	LF	13
			100.3 (ω)	Ar	LF	13

$2^2\Pi_{g,1/2}$ $D_{\infty h}$
 $T_0 = 474.1$ Ne LF¹³
 303.6T Ar LF¹³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	350.1 (ω)	Ne	LF	13
			344.5 (ω)	Ar	LF	13
Π_u	2	Bend	103.6(ω)	Ne	LF	13
			111.1 (ω)	Ar	LF	13

$\tilde{X}^2\Pi_{g,3/2}$ $D_{\infty h}$ Structure: LF^{8,9,15}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	371.69(ω)	gas	LF	7,9,11,12,14
			370.6 (ω)	Ne	LF	13
			367.3 (ω)	Ar	LF	13
Π_u	2	Bend	95.81 (ω)	gas	LF	10,12,14
			97.0 (ω)	Ne	LF	13
			98.6 (ω)	Ar	LF	13
Σ_u^+	3	Asym. stretch	525.90 (ω)	gas	IR,LF	2,7,9,11,12,14
			522.3 (ω)	Ne	LF	13
			514.4 (ω)	Ar	LF	13

$A = -253.9T$ gas LF¹⁵
 $B_0 = 0.058$ LF^{7-9,12,14}

^aFrom vertical detachment energies.

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CIBO

\tilde{X}		$C_{\infty v}$	Structure: MW,DL ²			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	B=O stretch	1972.18 1958s	gas Ar	DL IR	3 1
Π	2	Bend	404s	Ar	IR	1
Σ^+	3	B-Cl stretch	676.04 673wm	gas Ar	DL IR	2 1

$$B_0 = 0.174 \text{ MW,DL}^2$$

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BrBO

\tilde{X}		$C_{\infty v}$	Structure: MW,DL ²			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	B=O stretch	1937vs	Ar	IR	1
Π	2	Bend	374s	Ar	IR	1
Σ^+	3	B-Br stretch	535w	Ar	IR	1

$$B_0 = 0.120 \text{ MW}^2$$

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FBS

\tilde{X}		$C_{\infty v}$	Structure: MW ^{1,2}			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	B=S stretch	1644.38	gas	IR	3
Π	2	Bend	370(50)	gas	MW	1,2

$$B_0 = 0.165 \text{ MW}^{1,2} \text{IR}^3$$

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CIBS

\tilde{X}		$C_{\infty v}$	Structure: MW ¹ IR ⁵			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	BS stretch	1407.56	gas	IR	3-5
Π	2	Bend	300(40)	gas	MW	2
Σ^+	3	B-Cl stretch	529.90	gas	IR	3-5

$$B_0 = 0.093 \text{ MW}^{1,2} \text{IR}^5$$

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cyc-GaO₂⁻

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1			568.5	Ar	IR	1,2

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cyc-InO₂⁻

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1			462.4	Ar	IR	1,2

References

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cyc-TiO₂⁻

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1			473.9	Ar	IR	1

Reference

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NCO⁻

Threshold for electron detachment from ground-state NCO⁻ = 29120(40) gas PE²

\tilde{X}		$C_{\infty v}$ Structure: PE ²				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	3	Asym. stretch	2124.31	gas	DL	1
			2124.5T	Xe	IR	3

$B_0 = 0.384$ DL¹

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AsCO⁻

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	1756.4	Ar	IR	1

Reference

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cyc-CS₂

1B_1		C_{2v}				
$T_0 = 27607(5)$ Ar AB ²		${}^1B_1 - \tilde{X}$ 334–363 nm				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1			570T	Ar	AB	2

1B_2		C_{2v}				
$T_0 \leq 26586(5)$ Ar AB ²		${}^1B_2 - \tilde{X}$ 366–376 nm				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1			350T	Ar	AB	2

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CS ₂ s-stretch	876.5	Ar	IR	1
			881.3	N ₂	IR	1
b_2	3	CS ₂ a-stretch	517.7	Ar	IR	1
			520.9	N ₂	IR	1

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OCTe

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CO stretch	1970.7	Ar	IR	1
			1965.3	CO	IR	1

Reference

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SiS₂

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	514	CH ₄	Ra	2
Σ_u^+	3	Asym. stretch	918.0	Ar	IR	1

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GeO₂

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	GeO s-stretch	870.1	CH ₄	Ra	4
Π_u	2	Bend	195.5	Ar	IR	2
Σ_u^+	3	GeO a-stretch	1052.3	Ar	IR	2,3
			1061.6	N ₂	IR	1

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GeS₂

\tilde{X}		D _{∞h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	GeS s-stretch	474.7	CH ₄	Ra	3
Σ_u^+	3	GeS a-stretch	653.4	Ar	IR	1,2

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N₃⁻

Threshold for electron detachment from ground-state N₃⁻ is 21620(80).^{1,5}

$\tilde{X}^1\Sigma_g^+$		D _{∞h}	Structure: DL ^{2,3}			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_u^+	3	Asym. stretch	1986.47	gas	DL	2,3
			1991.9	Ar	IR	6
			2002.9	N ₂	IR	4,6

B₀ = 0.426 DL^{2,3}

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NO₂⁺

\tilde{D}^1B_2		C _{2v}				
T ₀ = 75485 gas		TPE ¹¹				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	1102	gas	TPE	11

\tilde{e}^3B_2		C _{2v}				
T ₀ = 74799 gas		PE ^{1,2,10} TPE ¹¹				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	1114	gas	PE,TPE	1,2,10,11

\tilde{d}^3A_1		C _{2v}				
T ₀ = 71622 gas		TPE ¹¹				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	1402	gas	TPE	11
	2	Bend	941	gas	TPE	11
b ₂	3	Asym. stretch	2054	gas	TPE	11

\tilde{C}^1B_1		C _{2v}				
T ₀ = 60902 gas		PE ^{2,10} TPE ¹¹				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	1009	gas	PE,TPE	1,2,10,11

\tilde{c}^3B_1		C _{2v}				
T ₀ = 60354 gas		PE ^{1,2,10} TPE ¹¹				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	1024	gas	PE,TPE	1,2,10,11

\tilde{B}^1B_2		C _{2v}				
T ₀ = 39166 gas		PE ^{1,2,10} TPE ¹¹				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	1038	gas	PE,TPE	2,10,11
	2	Bend	576	gas	PE,TPE	1,2,10,11
b ₂	3	Asym. stretch	1495	gas	TPE	11

\tilde{A}^1A_2		C _{2v}				
T ₀ = 36150 gas		PE ^{1,2,10} TPE ¹¹				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	963	gas	PE,TPE	1,2,10,11
	2	Bend	615	gas	TPE	11

\tilde{b}^3A_2		C _{2v}				
T ₀ = 32318 gas		PE ^{1,2,10} TPE ¹¹				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	Bend	685	gas	PE,TPE	1,2,10,11

\tilde{a}^3B_2 C_{2v}
 $T_0=26422$ gas PE^{1,2,10}TPE¹¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	Bend	639	gas	PE,TPE	1,2,10,11
$A=6.5$ TPE ¹¹						
$\tilde{X}^1\Sigma_g^+$		$D_{\infty h}$	Structure: TPE ^{4,5,7}			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1401.1	gas	TPE	8,9,11
			1362.4 ^a	Ne	IR	6
Π	2	Bend	627.7	gas	TPE	5,7-9,11
Σ_u^+	3	Asym. stretch	2376.5	gas	TPE	5,7,11
			2348.2	Ne	IR	6

$B_0=0.417$ TPE^{4,7}

^a $(\nu_1 + \nu_3) - \nu_3$.

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PNO

\tilde{X}	$C_{\infty v}$	Structure: MW ³				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	NO stretch	1756.65	gas	DL	2
			1754.7	Ar	IR	1
	3	PN stretch	865.2	Ar	IR	1

$B_0=0.206$ DL²MW³

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P₂O

\tilde{X}	$C_{\infty v}$					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	P=O stretch	1277.645	gas	DL	2,3
			1270.4	Ar	IR	1,5

$B_0=0.130$ DL³MW⁴

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XeC₂

Structured absorptions centered at 50500 (198 nm) and 23640 (423 nm) which appear when C₂H₂ isolated in solid xenon is subjected to 193 or 248 nm irradiation¹ are tentatively attributed to XeC₂.

\tilde{X}	$C_{\infty v}$					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CC stretch	1767.0	Xe	IR	1

Reference

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BCl₂

Emission which is observed between 200 and 260 nm on excitation of BCl₃ by radiation of wavelength shorter than 91 nm has been attributed⁵ to BCl₂.

\tilde{B}, \tilde{C}

Emission which is observed between 200 and 500 nm on excitation of BCl₃ by radiation of wavelength between 100 and 124 nm (9.99–12.27 eV) has been attributed^{8,9,11} to the $\tilde{B}-\tilde{X}$ and $\tilde{C}-\tilde{X}$ transitions of BCl₂. Other studies are consistent with contributions from two different emission systems, as summarized below.

Emission which is observed between 240 and 380 nm on excitation of BCl₃ by radiation of wavelength shorter than 118 nm^{4,5} and on electron impact⁶ has been attributed to BCl₂, as has been a similar chemiluminescence emission observed^{1,2} on reaction of H atoms with BCl₃. The electron impact study⁶ yielded a radiative lifetime of 1.65(20) μs and suggested that the lower state is the ground state of BCl₂.

Emission which is observed between 280 and 380 nm on excitation of BCl₃ by radiation of wavelength shorter than 124 nm^{4,5} and on electron impact⁶ has also been attributed to BCl₂.

\tilde{A}

Emission which is observed between 380 and 650 nm on excitation of BCl₃ by radiation of wavelength shorter than 138 nm^{4,5,8} and on electron impact⁶ has been attributed to the $\tilde{A}-\tilde{X}$ transition of BCl₂. The electron impact study indicated that the lower state is the ground state of BCl₂.

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	731w	Ar	IR	3
b_2	3	Asym. stretch	977.5 976.4 965.6vs	Ne Ar	IR	10 3,7

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AIF₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	755.3	Ar	IR	1,2
	2	Bend	253.7	Ar	IR	2
b_2	3	Asym. stretch	887.5	Ar	IR	1,2

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CO₂⁻

Maximum in the photoelectron spectrum of CO₂⁻ near 11300 (1.4 eV) gas PE¹

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	1253.8w	Ne	IR	6
	2	Bend	714.1w	Ne	IR	6
b_2	3	Asym. stretch	1658.3vs 1657.0	Ne Ar	IR	2,3,5,6 4

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FCO

\tilde{B}^a		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
$T_0 \leq 35741$	gas	AB ^{2,7,8}			$\tilde{B}-\tilde{X}$ 220–280 nm	
	35587	Ar	AB ³		$\tilde{B}-\tilde{X}$ 234–281 nm	
	35211	CO	AB ^{1,3}		$\tilde{B}-\tilde{X}$ 217–284 nm	

In the gas phase,² bands are diffuse, and the onset of predissociation is estimated⁶ to lie at a wavelength longer than 294 nm. In an argon matrix,³ the threshold for photodissociation into F+CO was observed near 280 nm.

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
			651T	gas	AB	2,7,8
			650T	Ar	AB	3
			650T	CO	AB	1,3

$\tilde{A} \ ^2\Pi(A'')$		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
$T_0 \leq 27000$	gas	CL ⁵ AB ^{7,8} CR ⁹			$\tilde{A}-\tilde{X}$ 280–455 nm	
$T_0 \geq 24000$	gas	AB ⁸				

Bands are somewhat broadened because of predissociation.⁹

$T_0 \leq 29586$	Ar	AB ³			$\tilde{A}-\tilde{X}$ 284–338 nm	
	29516	CO	AB ^{1,3}		$\tilde{A}-\tilde{X}$ 289–339 nm	

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Π	2	Bend	360HT 350HT 350HT	gas Ar CO	AB,CR AB AB	8,9 3 1,3
Σ^+	3	CF stretch	1080T 1050T 1050T	gas Ar CO	CR AB AB	9 3 1,3

$\tilde{X} \ ^2A'$		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	CO stretch	1861.64 1857vs 1855vs	gas Ar CO	DL IR IR	4 3 1
	2	Bend	627.5m 626m	Ar CO	IR IR	3 1
	3	CF stretch	1026.13 1023vs 1018s	gas Ar CO	DL IR IR	4 3 1

$A_0=6.38$; $B_0=0.382$; $C_0=0.360$ DL⁴MW¹⁰

^aThe designation of this state as \tilde{B} is arbitrary. See the discussion in Ref. 9.

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CICO

\tilde{X}		C_s		Structure: IR ²		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CO stretch	1884.59	gas	IR	4
			1876.7vs	Ar	IR	1-3
			1880vs	CO	IR	1
2	Bend	334.6	Ar	IR	2	
		3	CCl stretch	570.1s	Ar	IR
			570s	CO	IR	1

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CISiO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	SiO stretch	1160.9	Ar	IR	1
		SiCl stretch	509.4	Ar	IR	1

Reference

- ¹M. Junker and H. Schnöckel, *J. Chem. Phys.* **110**, 3769 (1999).

OCS⁻

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CO stretch	1646.4	Ne	IR	1
		CS stretch	718.2	Ne	IR	1

Reference

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CS₂⁻

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b_2	3	Asym. stretch	1159.4	Ne	IR	1,2
			1160.4	Ar	IR	2

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CCl₂⁺

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b_2	3	Asym. stretch	1213.5	Ne	IR	4
			1195.40	Ar	IR	1-3,5
			1197	Kr	IR	5

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CClBr⁺

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CCl stretch	1120.6	Ar	IR	1-3

References

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PO₂

${}^2B_1?$		C_{2v}				
$T_0 = 30393(2)$	gas	AB ¹ LF ^{3,11}			${}^2B_1-\tilde{X}$	268-600 nm
	Ar	AB ⁶			${}^2B_1-\tilde{X}$	292-301 nm

In LF studies,³ there was an apparently continuous background signal, with a maximum between 400 and 500 nm. The similarity of the behavior of this band system to that of the visible bands of NO₂ suggests that the quasicontinuum may be contributed by high vibrational levels of the ground state.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	Sym. stretch	933	gas	AB,LF	1,11
			942	Ar	AB	6
	2	Bend	389	gas	AB,LF	1,11

$\tau < 500$ ns gas LF^{3,11}
 $\tau_{\text{cont}} \approx 4.5$ μ s gas LF³

\tilde{X}^2A_1 C_{2v} Structure: AB¹MW,LMR²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	PO s-stretch	1076(12)	gas	MW,LF,LMR	2,3,11
	2	Bend	397(12)	gas	MW,LF,LMR	2,3,11
			386.4	Ar	IR	7
<i>b</i> ₂	3	PO a-stretch	1327.53	gas	DL	8,9
			1319.1	Ar	IR	4,5,7,10

$A_0 = 3.486$; $B_0 = 0.287$; $C_0 = 0.264$ MW,LMR²

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SNO

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	NO stretch	1522.8s ^a	Ar	IR	1-3
			1527.2	N ₂	IR	4
	3	NS stretch	790.2m	Ar	IR	1-3
			792.3	N ₂	IR	4

^aIn Fermi resonance with $2\nu_3$.

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SO₂⁺

\tilde{F}^2A_1 C_{2v}
 $T_0 = 62200(500)$ gas PE²

\tilde{E}^2B_1 C_{2v}
 $T_0 = 33090(20)$ gas PE^{2,6,7}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	Sym. stretch	960(10)	gas	PE	7
	2	Bend	444(10)	gas	PE	7

\tilde{D}^2A_1 C_{2v}
 $T_0 = 32190(50)$ gas PE^{1,2,6,7}PF⁵ $\tilde{D}-\tilde{X}$ 300-317 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	Sym. stretch	912(1)	gas	PE,PF	1,2,5-7
	2	Bend	411(60)	gas	PF	5

\tilde{C}^2B_2 C_{2v}
 $T_0 = 28670(50)$ gas PE^{1,2,6,7}PF⁵ $\tilde{C}-\tilde{B}$ 511-437 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	Sym. stretch	800(4)	gas	PE	6,7
	2	Bend	371(10)	gas	PF,PE	4-7

\tilde{B}^2B_2 C_{2v}
 $T_0 = 7034(80)$ gas PE^{1,6,7}PF^{4,5} $\tilde{C}-\tilde{B}$ 437-511 nm^a

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	2	Bend	465(9)	gas	PE,PF	1,4-7
<i>b</i> ₂	3	Asym. stretch	1320(8)HT	gas	PE	7

$\tau \approx 25$ μ s gas PF⁵

\tilde{A}^2A_2 C_{2v}
 $T_0 = 5156(65)$ gas PE^{1,6,7}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	Sym. stretch	981(60)	gas	PE	6,7
	2	Bend	353(7)	gas	PE	6,7
<i>b</i> ₂	3	Asym. stretch	202(13)	gas	PE	6,7

\tilde{X}^2A_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	2	Bend	404.2(5)	gas	PE	1,6,7
	3	Asym. stretch	1275.9	Ne	IR	8

Barrier to linearity ~ 3200 PE^{6,7}

^aAttributed by Ref. 5 to the $\tilde{C}-\tilde{A}$ transition.

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CF₂ \tilde{B}^a $T_0 \cong 72740$ gas AB¹⁰ $\tilde{B}-\tilde{X}$ 131–138 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	2	Bend	625	gas	AB	10

 \tilde{A}^1B_1 C_{2v} Structure: AB¹⁰

$T_0 = 37216$ gas EM¹AB^{2,3,5,10}LF^{20,42,43} $\tilde{A}-\tilde{X}$ 220–380 nm
 37219(2) Ne LF¹⁷
 36878(2) Ar AB^{4,6,16}EM¹⁶LF¹⁷ $\tilde{A}-\tilde{X}$ 210–346 nm
 37054(2) N₂ LF¹⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	1012.1(5)	gas	LF	42,43
	2	Bend	496.7(5)	gas	UV,LF	1–3,5,10,43
			496(2)	Ne	LF	17
			496(2)	Ar	AB,LF	4,6,16,17
			496(2)	N ₂	LF	17
b ₂	3	Asym. stretch	1180.2(5)	gas	LF	42,43

 $\tau_0 = 63.5(1.5)$ ns gas LF^{20,21,28,30,33,44}EM^{35,39}EF³⁷31 ns Ne LF¹⁷27 ns Ar LF¹⁷23 ns Kr LF¹⁷ $A_0 = 4.577$; $B_0 = 0.334$; $C_0 = 0.311$ AB¹⁰ \tilde{a}^3B_1 $T_0 = 19828$ gas CL^{18,19,22,24,29}PE^{34,35} $\tilde{a}-\tilde{X}$ 430–800 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	1020(30)	gas	PE	45
	2	Bend	517	gas	CL,PE	18,22,24,29,45

 $\tau \cong 1$ s gas CL¹⁹ \tilde{X}^1A_1 C_{2v} Structure: MW⁸AB^{9,10}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	1225.08	gas	DL,IR,PE	23,31,32,34,36,45
			1221.9	Ne	IR,LF	12,17,40
			1222vs	Ar	IR,LF	6,11,17
	2	Bend	666.25	gas	UV,PE	1,10,34,45
			668vw	Ar	DL	41
			668vw	Ar	IR,LF	6,11,17
b ₂	3	Asym. stretch	1114.44	gas	IR,DL	7,13,27,32,38
			1105.8	Ne	IR	12,40
			1102vs	Ar	IR	6,11

 $A_0 = 2.947$; $B_0 = 0.417$; $C_0 = 0.365$ MW^{8,15,26}AB^{9,10}

^aTentative assignment. This band system was associated with the $\tilde{C}-\tilde{X}$ transition in Ref. 10. Subsequent studies^{14,25} have dictated the reassignment to CF₃ of almost all of the bands between 136 and 160 nm which had tentatively been attributed¹⁰ to CF₂.

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CFCl

\tilde{A}^1A'' C_s Structure: LF^{9,10}
 $T_0=25284.0$ gas LF^{6,7,9-11} $\tilde{A}-\tilde{X}$ 342–466 nm
 24983 Ar AB¹LF^{2,3} $\tilde{A}-\tilde{X}$ 340–667 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CF stretch	1229(2) (ω)	gas	LF	11
	2	Bend	391	gas	LF	6,9–11
			392(1)	Ar	LF	3
3	CCl stretch	713	gas	LF	9–11	
		712(2)	Ar	LF	3	

$\tau_0=620(30)$ ns gas LF^{4,5,9,11}EM⁸
 330(20) ns Ar LF³

$A_{020}=4.354$; $B_{020}=0.185$; $C_{020}=0.177$ LF^{9,10}

\tilde{X}^1A' C_s Structure: LF^{9,10}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CF stretch	1154	gas	LF	6,7,9,10,11
			1146vs	Ar	IR	1
2	Bend		447	gas	LF	6,7,9,10,11
			442	Ar	LF	2,3
			753	gas	LF	6,7,9,10,11
3	CCl stretch		742s	Ar	IR	1

$A_0=2.349$; $B_0=0.214$; $C_0=0.196$ LF^{9,10}

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CFBr

\tilde{A}^1A'' C_s
 $T_0=20906$ gas LF³⁻⁵ $\tilde{A}-\tilde{X}$ 415–580 nm
 Ar LF¹ $\tilde{A}-\tilde{X}$ 442–535 nm

The barrier to dissociation is estimated⁵ to be 3250(150) above the origin of the \tilde{A} state.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CF stretch	1140	gas	LF	4,5
	2	Bend	340	gas	LF	4,5
			240(40)	Ar	LF	1
3	CBr stretch	645	gas	LF	4,5	

$\tau_{011}=3100(250)$ ns gas LF⁵

\tilde{X}^1A' C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CF stretch	1157vs	Ar	IR	2
	2	Bend	345	gas	LF	3,5
340(5)			Ar	LF	1	
3	CBr stretch	648	gas	LF	5	
		656s	Ar	IR	2	

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CCl₂

\tilde{A}^1B_1 C_{2v} Structure: LF¹⁷
 $T_0=17255.67(2)$ gas CL⁸LF^{7,9,10,15-18}EM¹¹ $\tilde{A}-\tilde{X}$ 400–800 nm
 17092 Ar AB^{1,3}LF⁴⁻⁶ $\tilde{A}-\tilde{X}$ 440–827 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	Sym. stretch	634.3	gas	LF	10,15,17
	2	Bend	624	Ar	LF	6
302.6			gas	LF	9,10,15,17	
304			Ar	AB,LF	1,3,6	

$\tau=3.81(30)$ μ s gas LF⁷;

$\tau_1=1.83(2)$ μ s, $\tau_2=3.72(6)$ μ s gas EM^{11,13}

$\tau=3.6$ μ s Ar LF⁶

$A_0=3.640$; $B_0=0.106$; $C_0=0.103$ LF¹⁷

\tilde{a}^3B_1		C_{2v}		Structure: MW ¹⁴ LF ¹⁷			
$T_0=970(800)$		gas	PE ²⁰				
\tilde{X}^1A_1		C_{2v}		Structure: MW ¹⁴ LF ¹⁷			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a_1	1	Sym. stretch	730.0	gas	PE,LF	12,17,20	
			725.6	Ne	IR	19	
			721wm	Ar	IR,LF	1-3,5,6	
	2	Bend	335.2	gas	PE,LF	12,17,20	
			333	Ar	LF	4-6	
			757.9	Ne	IR	19	
b_2	3	Asym. stretch	748vs	Ar	IR	1-3,21	
			756	Kr	IR	21	
			741				

$A_0=1.675$; $B_0=0.123$; $C_0=0.115$ MW^{14,22}LF^{16,17}

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CClBr

\tilde{A}		C_s		Structure: LF ⁵			
$T_0=16190$		gas	LF ^{5,6}	$\tilde{A}-\tilde{X}$ 497-600 nm			
16044		Ar	LF ^{3,4}	$\tilde{A}-\tilde{X}$ 540-776 nm			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a'	1	CCl stretch	684	Ar	LF	4	
			246	gas	LF	5	
	246	Ar	LF	4			
	2	Bend	246	Ar	LF	4	
			532	gas	LF	5	
	526	Ar	LF	4			
b_2	3	CBr stretch	532	gas	LF	5	
			526	Ar	LF	4	

$\tau=5.6(6)$ μ s Ar LF⁴

\tilde{X}		C_s		Structure: LF ⁶			
$T_0=15092.5(1.7)$		gas	LF ^{5,6}	$\tilde{A}-\tilde{X}$ 560-663 nm			
14962		Ar	LF ^{3,4}	$\tilde{A}-\tilde{X}$ 600-857 nm			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a'	1	CCl stretch	744	Ar	IR	1,2,7	
			262	gas	LF	6	
	260	Ar	LF	3,4			
	3	CBr stretch	618T	gas	LF	6	
			611	Ar	IR	1,2,7	

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CBr₂

\tilde{A}^1B_1		C_{2v}		Structure: LF ⁶			
$T_0=15092.5(1.7)$		gas	LF ^{5,6}	$\tilde{A}-\tilde{X}$ 560-663 nm			
14962		Ar	LF ^{3,4}	$\tilde{A}-\tilde{X}$ 600-857 nm			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a_1	1	Sym. stretch	474.8(1.2)	gas	LF	5,6	
			468	Ar	LF	4	
	2	Bend	185.5(4)	gas	LF	5,6	
			186	Ar	LF	4	

$\tau=14.5(1.5)$ μ s Ar LF⁴

\tilde{a}^3B_1		C_{2v}		Structure: LF ⁷			
$T_0=810(800)$		gas	PE ⁷				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a_1	1	Sym. stretch	525(20)	gas	PE	7	
			200(40)	gas	PE	7	

\tilde{X}^1A_1		C_{2v}		Structure: LF ⁷			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a_1	1	Sym. stretch	598	gas	LF	6	
			595w	Ar	IR	1,2	
	196	Ar	LF	3,4			
b_2	3	Asym. stretch	641s	Ar	IR	1,2	

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Cl₂

\bar{a}^1A_1 C_{2v}
T₀=480(800) gas PE¹

\bar{X}^3B_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	500(20)	gas	PE	1
	2	Bend	120(40)	gas	PE	1

Reference

¹R. L. Schwartz, G. E. Davico, T. M. Ramond, and W. C. Lineberger, *J. Phys. Chem. A* **103**, 8213 (1999).

SiF₂

\bar{B}^1B_2 C_{2v}
T₀=62280 gas UV¹¹MPI^{15,16,19} $\bar{B}-\bar{X}$ 158–165 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	790T	gas	AB,MPI	11,19
	2	Bend	320T	gas	AB	11

\bar{A}^1B_1 C_{2v} Structure: AB⁹
T₀=44113.9 gas EM^{1,2,20}AB^{5,9}LF^{14,17} $\bar{A}-\bar{X}$ 213–276 nm
≅43964 Ne AB⁷ $\bar{A}-\bar{X}$ 216–225 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	Bend	250.1(3)	gas	AB,LF	5,9,17,18
			253T	Ne	AB	7

τ=11.2(1.5) ns gas LF¹⁷EM²⁰
A₀=1.446; B₀=0.241; C₀=0.206 AB⁹

\bar{a}^3B_1 C_{2v} Structure: LF¹⁸
T₀=26319.478(6) gas EM^{10,20}MPI^{16,19}LF^{18,21} $\bar{a}-\bar{X}$ 364–420 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	Bend	278.2	gas	EM,LF,MPI	10,16,18

A₀=1.367; B₀=0.253; C₀=0.213 LF¹⁸

\bar{X}^1A_1 C_{2v} Structure: MW^{3,4}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	855.01	gas	IR,LF	6,13,17
			851s	Ne	IR	8
			843s	Ar	IR	7,8
	2	Bend	345	gas	MW,UV,LF,MPI	4,5,17,19
			343	Ar	IR	7
			870.40	gas	IR	6,13
b ₂	3	Asym. stretch	864.6s	Ne	IR	8
			855vs	Ar	IR	7,8

A₀=1.021; B₀=0.294; C₀=0.228 MW^{3,4}

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SiFCI

\bar{X} C_s Structure: MW¹
A₀=0.793; B₀=0.151; C₀=0.127 MW¹

Reference

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GeF₂

The high-temperature vapor of GeF₂ shows unstructured absorption between 136 and 156 nm, with a maximum near 146.3 nm.⁵

\bar{A}^1B_1 C_{2v}
T₀=43860.9 gas AB¹LF¹⁰EM¹¹ $\bar{A}-\bar{X}$ 215–265 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	623.8(6)	gas	LF	10
	2	Bend	159.6	gas	AB,LF	1,10

τ=9.3(1) ns gas EM¹¹

\tilde{a}^3B_1 C_{2v}
 $T_0=30582.1$ gas EM^{6,8,9,11}LF¹⁰ $\tilde{a}-\tilde{X}$ 300–380 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	673.1(5)	gas	LF	10
	2	Bend	192.2	gas	EM,LF	6,8,10

$\tau > 500$ ns gas EM¹¹

\tilde{X}^1A_1 C_{2v} Structure: IR²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	663	gas	IR	2
			655	Ne	IR	2
			648	Ar	IR,Ra	2,7
			653	N ₂	Ra	7
b_2	3	Asym. stretch	263(2)	gas	AB,EM	1,6,8
			692	gas	IR	2
b_2	3	Asym. stretch	685	Ne	IR	2
			676	Ar	IR	2

$A_0=0.513$; $B_0=0.262$; $C_0=0.173$ MW^{3,4}

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GeCl₂

\tilde{A}^1B_1 C_{2v}
 $T_0=30622(2)$ gas AB²LF¹¹EM¹³ $\tilde{A}-\tilde{X}$ 300–330 nm
 Structured absorption² is superposed on a continuum with maximum near 32280, presumably due to predissociation of GeCl₂ into GeCl+Cl. In the LF excitation spectrum,¹¹ there is a marked increase in the line density beyond approximately 31630.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	354	gas	LF	11
	2	Bend	104	gas	AB,LF	2,11

$\tau=89.7(6.8)$ ns gas EM¹⁰

\tilde{a}^3B_1 C_{2v}
 $T_0=22315(2)$ gas CL¹LF¹¹EM¹³ $\tilde{a}-\tilde{X}$ 400–490 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	393	gas	LF	11
	2	Bend	118	gas	CL,LF	1,11

$\tau=17.4(6)$ μ s gas EM¹⁰

\tilde{X}^1A_1 C_{2v} Structure: ED⁸MW¹²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	399	gas	CL,Ra	1,4
			398.6	Ar	IR	3,5,7,9
			390	N ₂	Ra	6
b_2	2	Bend	160(4)	gas	CL,AB,Ra	1,2,4
			163	N ₂	Ra	6
b_2	3	Asym. stretch	373.5	Ar	IR	3,5,7,9
			362	N ₂	Ra	6

$A_0=0.241$; $B_0=0.087$; $C_0=0.064$ MW¹²

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PO₂⁻

Threshold for electron detachment from ground-state
 $PO_2^- = 27600(80)$ gas PE²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	Bend	520T	gas	PE	2
b_2	3	PO a-stretch	1198.6	Ar	IR	1,3

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FPO

\bar{X}		C_s Structure: MW ^{3,5} IR ⁵				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	PO stretch	1297.54	gas	IR	4,5
			1292.2	Ar	IR	1
	2	Bend	412T	gas	IR	2,5
			416.0	Ar	IR	1
	3	PF stretch	819.57	gas	IR	2,4,5
			811.4	Ar	IR	1

$A_0 = 1.397$; $B_0 = 0.310$; $C_0 = 0.253$ MW^{3,5}IR^{4,5}

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CIPO

\bar{X}		C_s Structure: MW ⁷				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	PO stretch	1263.01	gas	IR,DL	5,6,8
			1258vs	Ar	IR	1-4
	2	Bend	308wm	Ar	IR	1
			492T	gas	IR	5
	3	PCI stretch	489vs	Ar	IR	1-3

$A_0 = 1.115$; $B_0 = 0.151$; $C_0 = 0.133$ DL⁶MW⁷

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BrPO

\bar{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	PO stretch	1257.84	gas	IR,DL	3,4
			1253.0	Ar	IR	1,2
	2	Bend	253.7	Ar	IR	1
			408	gas	IR	3
	3	PBr stretch	407.1	Ar	IR	1,2

$A_0 \cong 1.021$; $B_0 = 0.092$; $C_0 = 0.084$ DL⁴

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IPO

\bar{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	3	PO stretch	1253	gas	IR	1

Reference

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FPS

\bar{X}		C_s Structure: MW,IR ³				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	PF stretch	803.25	gas	IR	2,3
			791.4	Ar	IR	1
	2	Bend	313.6	Ar	IR	1
			726.27	gas	IR	3
	3	PS stretch	720.2	Ar	IR	1

$A_0 = 0.948$; $B_0 = 0.163$; $C_0 = 0.139$ MW^{2,3}

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PF₂⁺

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	PF stretch	980(30)	gas	PE	1,2
			1012.3	Ne	IR	3
b ₂	3	PF stretch	1049.3	Ne	IR	3

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CION

In an argon matrix, an absorption between 320 and 550 nm, with maximum at 22980 (417 nm) has been assigned² to CION.

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	NO stretch	1842.2vs	Ar	IR	1,2
			1852.1vs	N ₂	IR	1,2
	2	Bend	409.4w	Ar	IR	1,2
3	ClO stretch	405.2w	N ₂	IR	1,2	
		247T	Ar	IR	2	

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BrON

In an argon matrix, an absorption between 330 and 550 nm, with maximum at 22730 (440 nm) has been assigned¹ to BrON.

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	ON stretch	1820.0vs	Ar	IR	1
			1816.2vs	N ₂	IR	1
	2	Bend	350T	Ar	IR	1
3	BrO stretch	365.4	N ₂	IR	1	
		190T	N ₂	IR	1	

Reference

- ¹G. Maier, H. P. Reisenauer, and M. De Marco, *Chem. Eur. J.* **6**, 800 (2000).

SSO

An absorption band system between 190 and 230 nm has been attributed⁷ to SSO. However, an alternate assignment to the $\tilde{C}-\tilde{X}$ band system of SO₂ has been proposed.¹⁵

\tilde{C}^1A'		C _s Structure: AB ¹¹					
T ₀	29687.72	gas	AB ^{1,7,11} LF ^{12,13,16-18}	29285(20)	Xe	AB ⁴	$\tilde{C}-\tilde{X}$ 250-395 nm
							$\tilde{C}-\tilde{X}$ 280-342 nm
Predissociation limit between 31172 and 31307. AB ¹¹							

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	SO stretch	1033.9(6)	gas	AB,LF	11,13,16,18
	2	Bend	253.8(6)	gas	AB,LF	11,13,16,18
	3	SS stretch	410.6	gas	AB,LF	7,11,13,16-18
			415(20)	Xe	AB	4

$\tau = 66(4)$ ns gas LF¹⁷
A₀ = 1.016; B₀ = 0.149; C₀ = 0.130 AB¹¹LF¹⁷

\tilde{a}^3A'		C _s				
T ₀	13943	gas	AB ^{10,15} LF ¹³			$\tilde{a}-\tilde{X}$ 430-670 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	SO stretch	1089	gas	AB	10,15
	2	Bend	332	gas	AB	10,15
	3	SS stretch	505	gas	AB	10,15

\tilde{X}^1A'		C _s Structure: MW ^{2,6}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	SO stretch	1166.45	gas	IR,DL,LF	1,3,14,18
			1156.2	Ar	IR,Ra	8,9
	2	Bend	380	gas	LF	16,18
3	SS stretch	382	Ar	IR,Ra	8,9	
		679.1	gas	IR,LF	1,3,13,18	
		672.2	Ar	IR,Ra	8,9	

A₀ = 1.398; B₀ = 0.169; C₀ = 0.150 MW^{2,5,6} DL¹⁴

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SeO₂

¹B₂ C_{2v} Structure: AB¹⁰
 T₀=31957.4 gas AB^{4,9,10}EM⁴ 31065(20) Xe AB⁶ ¹B₂- \tilde{X} 225–345 nm
¹B₂- \tilde{X} 242–322 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	648.8	gas	AB	1,9
			620T	Xe	AB	6
	2	Bend	258	gas	AB	9

³B₂ C_{2v} Structure: AB¹¹
 T₀=23840 gas AB^{2,11}EM⁴ Kr EM⁶ Xe EM⁶ ³B₂- \tilde{X} 370–500 nm
³B₂- \tilde{X} 475–600 nm
³B₂- \tilde{X} 475–615 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	780	gas	AB	11
	2	Bend	199	gas	AB,EM	2,4,11
b ₂	3	Asym. stretch	863	gas	AB	11

$\tau < 200(100) \mu\text{s}$ Xe EM⁶

\tilde{X}^1A_1 C_{2v} Structure: MW⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	922.6	gas	AB,PE	9,11,12
			923.4	Ne	IR	5
			922.0w	Ar	IR	5,8,13
	2	Bend	364	gas	AB,IR	9,11,14
			366.0s	Ar	IR	8,13
			365T	Kr,Xe	EM	6
b ₂	3	Asym. stretch	968	gas	IR,AB	3,11,14
			971.2	Ne	IR	5
			970.2			
			965.3s	Ar	IR	5,8,13

A₀=0.962; B₀=0.289; C₀=0.222 MW⁷

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TeO₂

¹B₂? C_{2v}
 T₀=25526 gas LF⁴ 345–406 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	679 (ω)	gas	LF	4
	2	Bend	220 (ω)	gas	LF	4

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	823 (ω)	gas	LF	4
			826.4	Ne	IR	1
			822.6	Ar	IR	1
			831.7	N ₂	IR	2
	2	Bend	282 (ω)	gas	IR,LF	3,4
			294	N ₂	IR	2
b ₂	3	Asym. stretch	853	gas	IR	3
			844.8	Ne	IR	1
			839.4	Ar	IR	1
			848.3	N ₂	IR	2

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OCIO⁺

\tilde{D}^1A_2 C_{2v}
 T₀=62200(80) gas PE³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁			485(40)	gas	PE	3

\tilde{d}^3A_2 C_{2v}
 T₀=59460(80) gas PE¹⁻³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CIO s-stretch	605(40)	gas	PE	3

\tilde{C}^1B_2						
$T_0=39530(80)$		C_{2v} gas PE ¹⁻³				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CIO s-stretch	765(40)	gas	PE	3
$\tilde{B}^1B_1^a$						
$T_0=25410(80)$		C_{2v} gas PE ³				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CIO s-stretch	725(40)	gas	PE	3
$\tilde{A}^1A_2^a$						
$T_0=24040(80)$		C_{2v} gas PE ³				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CIO s-stretch	810(40)	gas	PE	3
$\tilde{c}^3B_1^a$						
$T_0=19690(80)$		C_{2v} gas PE ¹⁻³				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	Bend	485(40)	gas	PE	3
$\tilde{b}^3A_2^a$						
$T^b=20330(80)$		C_{2v} gas PE ³				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CIO s-stretch	765(40)	gas	PE	3
$\tilde{a}^3B_2^a$						
$T_0=16540(80)$		C_{2v} gas PE ¹⁻³				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CIO s-stretch	765(40)	gas	PE	3
	2	Bend	520(40)	gas	PE	2,3
\tilde{X}^1A_1						
		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CIO s-stretch	1015(40)	gas	PE	2,3
	2	Bend	520(40)	gas	PE	2,3

^aAssignments from Ref. 4.

^bFrom vertical ionization potential.

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CF₂⁻

Threshold for electron detachment from ground-state
CF₂⁻ = 1450(160) gas PE¹⁻³

\tilde{X}^2B_1						
		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CF stretch	860(30)	gas	PE	1

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CCl₂⁻

Threshold for electron detachment from ground-state
CCl₂⁻ = 12830(560) gas PE¹⁻³

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CBr₂⁻

Threshold for electron detachment from ground-state
CBr₂⁻ = 15170(560) gas PE¹

Reference

- ¹R. L. Schwartz, G. E. Davico, T. M. Ramond, and W. C. Lineberger, J. Phys. Chem. A **103**, 8213 (1999).

Cl₂⁻

Threshold for electron detachment from ground-state
Cl₂⁻ = 16860(560) gas PE¹

Reference

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PF₂

\tilde{M} C_{2v}
T₀ = 68505(5) gas MPI⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	972(7) ^a	gas	MPI	4
	2	Bend	345(3) ^a	gas	MPI	4

\tilde{L} C_{2v}
T₀ = 67922(5) gas MPI⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	956(2)	gas	MPI	4
	2	Bend	358(3)	gas	MPI	4

\tilde{K} C_{2v}
T₀ = 66763(5) gas MPI⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	956(2)	gas	MPI	4
	2	Bend	364(1)	gas	MPI	4

\tilde{J} C_{2v}
T₀ = 66118(5) gas MPI⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	981(4) ^a	gas	MPI	4
	2	Bend	366(1) ^a	gas	MPI	4

\tilde{I} C_{2v}
T₀ = 65958(5) gas MPI⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	980(4) ^a	gas	MPI	4
	2	Bend	365(3) ^a	gas	MPI	4

\tilde{H} C_{2v}
T₀ = 60962(5) gas MPI⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	1004(2)	gas	MPI	4

\tilde{G} C_{2v}
T₀ = 58184(5) gas MPI⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	998(2)	gas	MPI	4

\tilde{F}^2A_1 C_{2v}
T₀ = 55126(5) gas MPI⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	1008(2)	gas	MPI	4

\tilde{E}^2B_1 C_{2v}
T₀ = 51932(5) gas MPI⁴

The $\tilde{E}-\tilde{A}$ emission of gas-phase PF₂ is observed⁵ near 30800 (325 nm) when PF₃ is excited by 14.4 eV radiation.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	1016(2) ^a	gas	MPI	4
	2	Bend	408(2) ^a	gas	MPI	4

τ = 7.9 ns gas EM⁵

\tilde{C}^2A_1 C_{2v}
T₀ = 45000T gas EM⁵
τ = 14.7 ns gas EM⁵

\tilde{B}^2B_2 C_{2v}
T₀ = 23998(6) gas EM^{3,5} $\tilde{B}-\tilde{X}$ 400–520 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	505(2)	gas	EM	3
	2	Bend	219T	gas	EM	3

τ ≥ 1.9 ms gas EM³

\tilde{A}^2A_1 C_{2v}
gas EM⁵ $\tilde{A}-\tilde{X}$ 320–550 nm

\tilde{X}^2B_1 C_{2v} Structure: MW²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	841(4)	gas	EM	3
			852.1ms	Ar	IR	1
	2	Bend	366	gas	MW,EM	2,3
b ₂	3	Asym. stretch	848(24)	gas	MW	2
			831.4s	Ar	IR	1

A₀ = 0.933; B₀ = 0.310; C₀ = 0.232 MW²

^aω.

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PCl₂

$\tilde{E}(4p)$ C_{2v}
T₀=51320(10) gas MPI³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	600(15)	gas	MPI	3
	2	Bend	240(15)	gas	MPI	3

$\tilde{D}^2A_1(4s)$ C_{2v}
T₀=42760(15) gas MPI³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	620(20)	gas	MPI	3
	2	Bend	230(20)	gas	MPI	3

\tilde{B}^2B_2 C_{2v}
In the gas phase, unstructured emission at wavelengths longer than 320 nm, with a maximum at approximately 460 nm, has been attributed^{2,4} to PCl₂.
τ=29(6) μs gas EM²

\tilde{X}^2B_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	525(10)	gas	MPI	3
			525	Ar	IR	1
b ₂	3	Asym. stretch	452	Ar	IR	1

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ClOO

An unstructured absorption between 225 and 270 nm, with a maximum near 246 nm in the gas phase³⁻⁶ and near 250 nm in a neon⁸ or an argon⁷ matrix, has been assigned to ClOO, which photodecomposes in that spectral region.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	OO stretch	1443 ^a	gas	IR	4
			1442.5	Ne	IR	8,9
			1439.6			
			1442.5vs ^b	Ar	IR	2,7,9
			1416.7			
			1440.7 ^b	Kr	IR	9
			1412.1			
			1438	N ₂	IR	1
			1428			
			1436	O ₂	IR	7
			413.7	Ne	IR	8
			408.3vw ^b	Ar	IR	2,7
			3	ClO stretch	201.4	Ne
192.4w	Ar	IR			7	
203	O ₂	IR			7	

^aAbsorption maximum; spectral slit width 13 cm⁻¹.

^bPeaks at 1417, 435, and 227 cm⁻¹ in argon-matrix experiments and at 1412 in krypton-matrix experiments are contributed by ClOO trapped in a metastable site in the rare-gas lattice.⁷⁻⁹

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BrOO

In an argon matrix, a prominent absorption between 240 and 300 nm, with maximum at 38200 (262 nm), has been assigned³ to BrOO. Irradiation at 280 nm leads to photoisomerization to OBrO.⁴

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	OO stretch	1487.0	Ne	IR	5
			1485.1	Ar	IR	1-4

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SO₂⁻

Threshold for electron detachment from ground-state SO₂⁻ = 8930(65) gas PE^{2,3}

\tilde{X}^2B_1		C _{2v}	Structure: PE ³			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	944(48)	gas	PE	3
			990.8	Ne	IR	5
	2	Bend	985m Cs	Ar	IR,Ra	1,4
			990 Na	Ar	IR	1,4
			435(100)	gas	PE	3
b ₂	3	Asym. stretch	495wm Cs	Ar	IR	1,4
			495 Na	Ar	IR	1,4
			1086.2	Ne	IR	5
			1089.7	Ar	IR	4
			1081.8			
			1042s Cs	Ar	IR	1,4
			1041 Na	Ar	IR	1,4

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S₃⁻

Threshold for electron detachment from ground-state S₃⁻ = 17000(110) gas PE^{1,3}

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b ₂	3	Asym. stretch	594.2T	Ar	IR	2

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OF₂⁺

\tilde{E} C_{2v}
T^a = 51560(400) gas PE^{1,2}

\tilde{D} C_{2v}
T₀ ≅ 38650 gas PE^{1,2}

\tilde{C}^2A_2 C_{2v}
T₀ = 26880(120) gas PE¹⁻³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	OF ₂ s-stretch	900(50)	gas	PE	3
	2	Bend	480(50)	gas	PE	3

\tilde{A}^2B_2 C_{2v}
T₀ = 21950(120) gas PE¹⁻³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	Bend	440(30)	gas	PE	3

\tilde{X}^2B_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	OF s-stretch	1000(20)	gas	PE	1-3

^aFrom vertical ionization potential.

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Cl₂O⁺

\tilde{F}^2B_1 C_{2v}
T^a = 54380(320) gas PE¹

\tilde{E}^2A_1 C_{2v}
T^a = 46070(320) gas PE¹

\tilde{D}^2B_2 C_{2v}
T^a = 40020(320) gas PE¹

$\tilde{C}^2A_2^b$ C_{2v}
T₀ = 14970(70) gas PE^{1,2}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Sym. stretch	613(40)	gas	PE	2
	2	Bend	306T	gas	PE	2

$\tilde{B}^2A_1^b$ C_{2v}
 $T_0 \geq 12630(70)$ gas PE^{1,2}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	Bend	282(40)	gas	PE	2

$\tilde{A}^2B_2^b$ C_{2v}
 $T_0 = 9110(70)$ gas PE^{1,2}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	Bend	290(40)	gas	PE	2

$\tilde{X}^2B_1^b$ C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CLO s-stretch	678(40)	gas	PE	1,2
	2	Bend	274(40)	gas	PE	1,2

^aFrom vertical ionization potential.

^bFor assignment, see Ref. 3.

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OCIO

\tilde{E}
 $T_0 = 63774$ gas AB^{12,19,45} $\tilde{E}-\tilde{X}$ 148–157 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	Sym. stretch	1000T	gas	AB	12
	2	Bend	508	gas	AB	12,19

\tilde{D}
 $T_0 = 61430(20)$ gas AB^{12,19,42,45} $\tilde{D}-\tilde{X}$ 155–163 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	Sym. stretch	1051(20)	gas	AB	12,19,42
	2	Bend	521(20)	gas	AB	12,19,42

\tilde{C}^2A_1 C_{2v}
 $T_0 = 54689(20)$ gas AB^{12,19,42,45} $\tilde{C}-\tilde{X}$ 176–183 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	1020(20)	gas	AB	12,19,42

$^2A_1, ^2B_2$ C_{2v}

These two as yet undetected states are expected to lie close to the \tilde{A} state and to interact strongly with it. One or both of them may contribute to the much-studied predissociation of the \tilde{A} state.^{21,24,29,35,40,46–48}

\tilde{A}^2A_2 C_{2v}^a Structure: AB²⁵
 $T_0 = 21017.2$ gas AB^{1,2,5,14,25,31,32,34,35,45} LF^{15,16,50} MPI^{33,36} PF^{41,43,47}
 20991.3 Ne AB⁴⁸LF⁴⁸ $\tilde{A}-\tilde{X}$ 260–780 nm
 20827.8 Ar AB⁴⁸LF⁴⁸ $\tilde{A}-\tilde{X}$ 415–755 nm
 20684.8 Kr AB⁴⁸LF⁴⁸ $\tilde{A}-\tilde{X}$ 417–765 nm
 $\tilde{A}-\tilde{X}$ 425–748 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	708.0 707.9 716.0 712.5	gas Ne Ar Kr	AB,LF AB AB AB	5,14,34,50 48 48 48
	2	Bend	289.7 292.5 302.3 303.0	gas Ne Ar Kr	AB,LF AB AB AB	5,14,34,50 48 48 48
b_2	3	Asym. stretch	441.2 443.8H 448.6H	gas Ne Kr	AB AB AB	25,34 48 48

$\tau_1^b = 56(20)$ ps gas AB^{21,24}LF²⁹
 $\tau_2^c = 336(27)$ ps gas LF²⁹
 $A_0 = 1.057; B_0 = 0.311; C_0 = 0.240$ AB²⁵

\tilde{X}^2B_1 C_{2v} Structure: MW^{6,30}IR²⁶

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	945.59s 944.8m 947.6m	gas Ne Ar	IR,AB LF,Ra LMR DL,LS IR IR,Ra	3–5,14–16 20,23,26, 49 28,37,39 44,48 13,17,18 44,48
	2	Bend	944.1 940(2) 950(2) 447.70s	Kr Xe N ₂ gas	Ra,IR Ra Ra IR,AB	18,48 18 18 4,5,14–16, 22,25,38,39,49
			448.7m 451s 447s 447	Ne Ar Kr	IR IR,LF LF	44 13,17,18,44 18
b_2	3	Asym. stretch	1110.11vs 1107.6vs 1106.5vs 1100.8vs 1102.2	gas Ne Ar Kr	IR,Ra IR IR IR	3,14,26 38,39,49 44,48 13,17,44, 48 48

$A_0 = 1.737; B_0 = 0.332; C_0 = 0.278$ MW^{6–10,27}IR^{37,38}

^aDouble minimum in potential for antisymmetric stretch,^{11,25} with barrier height of 1153.³⁴

^bFor $F_1(J=N+1/2)$ spin states.

^cFor $F_2(J=N-1/2)$ spin states.

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OBrO

\tilde{C}^2A_2 C_{2v}
 $T_0 = 15863$ gas AB^{2,6,10} $\tilde{C}-\tilde{X}$ 388–645 nm
 15818.4(2) Ne AB⁸ $\tilde{C}-\tilde{X}$ 482–632 nm
 16785(20) Ar AB^{3,5} $\tilde{C}-\tilde{X}$ 400–650 nm

Irradiation of the argon-matrix deposit in this spectral region leads to photoisomerization to BrOO.⁵

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	641.5 ^a	gas	AB	2,6
			642.77(13) ^a	Ne	AB	8
			631T	Ar	AB	5
	2	Bend	210.7 ^a	gas	AB	2,6
			223.2(4) ^a	Ne	AB	8
			221T	Ar	AB	5
b_2	3	Asym. stretch	463H	Ne	AB	8

\tilde{X}^2B_1		C_{2v}	Structure: MW ^{4,7}			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	799.4	gas	AB,IR	6,9
			797.6	Ne	IR	8
			795.7	Ar	IR	3,5
	2	Bend	317.5	gas	AB	6
			317.0	Ar	IR	5
			848.6	gas	IR	4,9
b_2	3	Asym. stretch	848.4	Ne	IR	8
			845.2	Ar	IR	1,3,5

$A_0 = 0.935$; $B_0 = 0.275$; $C_0 = 0.212$ MW^{4,7}

^a ω .

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BrOCl

Gas-phase absorption maxima of BrOCl have been observed² at 36760 (272 nm) and 31250 (320 nm).

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	ClO stretch	675.9	Ar	IR	1
	3	BrO stretch	559.5	Ar	IR	1

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BrOBr

In the gas phase, a prominent absorption maximum near 50000 (200 nm) and a less prominent absorption maximum near 31800 (314 nm), with a broad shoulder which has its onset near 22700 (440 nm) and on which structure with separations of ~500 is partially resolved, have been assigned⁴ to BrOBr. A more recent gas-phase study⁶ supports the assignment of the absorption maximum near 31800 (314 nm) to BrOBr and proposes the assignment of absorption maxima at 28570 (350 nm), 19230 (520 nm), and 15150 (660 nm) to that species, as well.

In an argon matrix, very slow photoisomerization of BrOBr to BrBrO is observed⁸ on exposure of the sample to radiation between 16700 (600 nm) and 25000 (400 nm). Rapid photoisomerization occurs between 25000 and 28600 (350 nm).

In a nitrogen matrix, a strong absorption maximum at approximately 47000, a weaker shoulder near 31200, and a much weaker shoulder near 22100 have been attributed³ to electronic transitions of BrOBr.

\tilde{X}		C_{2v}		Structure: EXAFS ³ MW ^{5,7}		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	533	gas	IR	10
			526.1s	Ar	IR	1,2,8
			528	N ₂	IR	3
b_2	3	Asym. stretch	629	gas	IR	10
			621.4	Ne	IR	9
			623.4w	Ar	IR	2,8
			626	N ₂	IR	3

$A_0 = 1.108$; $B_0 = 0.046$; $C_0 = 0.044$ MW^{5,7}

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SF₂

$\tilde{J}^1B_1(5s)$		C_{2v}				
$T_0 = 68951$ gas		MPI ¹⁴				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	942(3)	gas	MPI	14

$\tilde{I}(3d)$		C_{2v}				
$T_0 = 68847$ gas		MPI ¹⁴				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	933(4)	gas	MPI	14

$\tilde{H}(3d)$		C_{2v}				
$T_0 = 68571$ gas		MPI ¹⁴				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	936(4)	gas	MPI	14

$\tilde{G}(3d)$		C_{2v}				
$T_0 = 68378$ gas		MPI ¹⁴				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	939(12)	gas	MPI	14

$\tilde{F}(4p)$		C_{2v}				
$T_0 = 63812$ gas		MPI ¹⁴				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	966(6)	gas	MPI	14
	2	Bend	417(23)	gas	MPI	14

$\tilde{E}(4p)$		C_{2v}				
$T_0 = 62025(30)$ gas		MPI ^{10,14}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	941(27)	gas	MPI	10,14
	2	Bend	413(12)	gas	MPI	10,14

\tilde{C}		C_{2v}				
gas MPI ^{10,14}		$\tilde{C}-\tilde{X}$ 165–175 nm				

$\tilde{B}^1B_1(4s)$		C_{2v}				
$T_0 = 54433(30)$ gas		MPI ^{10,14}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	991(12)	gas	MPI	10,14
	2	Bend	361(24)	gas	MPI	10,14

\tilde{B}'^1A_1 C_{2v}
 $T_0 = 38623$ gas MPI^{14,15} $\tilde{B}' - \tilde{X}$ 235–260 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	926(4)	gas	MPI	15

\tilde{A}
 $T_0 = 18100(1000)$ gas CL^{7,8,11} $\tilde{A} - \tilde{X}$ 550–850 nm
 Chemiluminescence in the reaction of F₂ with CS₂, originally assigned^{7,8} to FCS, has been reassigned⁹ to SF₂.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	2	Bend	243(6)	gas	CL	11

\tilde{X} C_{2v} Structure: MW^{1,2,4}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	838.53	gas	IR, CL	5,7–9,11
			834	Ne	IR	3
			832vs	Ar	IR	3,12
			825	N ₂	IR	3
			355(2)	gas	MW,CL	2,7–9,11
	2	Bend	358	Ne	IR	3
			358m	Ar	IR	3
			358	N ₂	IR	3
			813.04	gas	IR,CL	5,11
			807.5	Ne	IR	3
b_2	3	Asym. stretch	804vs	Ar	IR	3,12
			795	N ₂	IR	6

$A_0 = 0.898$; $B_0 = 0.307$; $C_0 = 0.228$ MW^{1,2,13}

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CISBr

\tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	SCI stretch	515.6	Ar	IR	1
			517.8	Kr	IR	1
			519.2	N ₂	IR	1
	3	SBr stretch	416.0	Ar	IR	1
			413.0	Kr	IR	1
			416.0	N ₂	IR	1

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OCIO⁻

Threshold for electron detachment from ground-state OCIO⁻ = 17310(20) gas PE^{2,3}

\tilde{X}^1A_1 C_{2v} Structure: PE²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	774(25)	gas	PE	2
			790 Cs	Ar	IR	1
	2	Bend	418 Cs	Ar	IR	1
b_2	3	Asym. stretch	823 Cs	Ar	IR	1

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FCIO

\tilde{X} C_s Structure: IR²MW³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CIO stretch	1037.69	gas	IR	2
			1038s	Ar	IR	1
	2	Bend	310H	gas	IR	2
	3	FCl stretch	315.2m	Ar	IR	1
			596.9	gas	IR	2
			593.5vs	Ar	IR	1

$A_0 = 1.223$; $B_0 = 0.278$; $C_0 = 0.226$ IR²MW³

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BrBrO

In an argon matrix, irradiation at 870 nm results in photoisomerization to BrOBr.³ Rapid photoisomerization also ensues on irradiation beyond 29400 (340 nm).⁴

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	BrO stretch	804.8	Ne	IR	5
			805.4	Ar	IR	1-4
	3	BrBr stretch	236	Ar	IR	1

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I₃ \tilde{C}

A band in the photoelectron spectrum of I₃⁻ 5700 (0.707 eV) above the band for I₃ (\tilde{X}) has been assigned² to I₃ (\tilde{C}).

 \tilde{B}

A band in the photoelectron spectrum of I₃⁻ 5190 (0.643 eV) above the band for I₃ (\tilde{X}) has been assigned² to I₃ (\tilde{B}).

 \tilde{A}

A band in the photoelectron spectrum of I₃⁻ 2480 (0.307 eV) above the band for I₃ (\tilde{X}) has been assigned^{1,2} to I₃ (\tilde{A}).

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	115(5)	gas	PE	1

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Cl₃⁻

In the gas phase, a photodetachment peak at 47360 (211 nm) has been assigned⁶ to the formation of Cl₃ (1²Π_g) + e. The doublet structure of this peak is attributed to spin-orbit interaction. This state of Cl₃ is expected to be bound.

In the gas phase, a broad photodetachment peak with maximum at 41700 (240 nm) has been assigned⁶ to the formation of Cl₃ ($\tilde{X}^2\Pi_u, 1^2\Sigma_u^+$) + e, where both states of Cl₃ are dissociative. An absorption maximum which appears at 39840 (251 nm) when CsCl is codeposited with Cl₂ in an argon

matrix³ and at 39530 (253 nm) when an Ar:Cl₂ sample is subjected to electron bombardment during deposition⁵ has also been assigned to Cl₃⁻.

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	225T Cs 253 Rb 258 K 276 Na	Ar Ar Ar Ar	Ra Ra Ra Ra	2 2 2 2
Σ_u^+	3	Asym. stretch	327 Cs 340 Rb 345 K 375 Na 374 ^a	Ar Ar Ar Ar Kr	IR,Ra IR,Ra IR,Ra IR,Ra IR	2 2 2 2 1

^aAttributed in Ref. 1 to the uncharged species. Reassigned by Ref. 4 to the anion.

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I₃⁻

Threshold for electron detachment from ground-state I₃⁻ = 34100(100) gas PE²

\tilde{X}		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	112(1) 113 Cs	gas Ar	Ra ^a Ra ^a	3 1

^aResonance Raman.

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8.4. Four-Atomic Trihydrides**HPd(H₂)**

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	PdH stretch	1900.8	Ar	IR	1

DPd(D₂)

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	PdD stretch	1366.4	Ar	IR	1

Reference

¹L. Andrews, X. Wang, M. E. Alikhani, and L. Manceron, J. Phys. Chem. A **105**, 3052 (2001).

PtH₃

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	PtH s-stretch	2064.7	Ar	IR	1
a''	5	PtH a-stretch	1802.4	Ne	IR	1
			1784.3	Ar	IR	1

PtD₃

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a''	5	PtD a-stretch	1293.8	Ar	IR	1

Reference

¹L. Andrews, X. Wang, and L. Manceron, J. Chem. Phys. **114**, 1559 (2001).

HAu(H₂)

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2		2170.1	Ne	IR	1
			2167.9			
			2173.6	Ar	IR	1
			2170.6			

DAu(D₂)

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2		1559.0	Ne	IR	1
			1559.3	Ar	IR	1
			1556.5	D ₂	IR	1
b ₂			939.6	D ₂	IR	1
			434.8	D ₂	IR	1

Reference

¹X. Wang and L. Andrews, J. Am. Chem. Soc. **123**, 12899 (2001).

CeH₃

\tilde{X} C _{3v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e	3	CeH ₃ stretch	1283.1	Ar	IR	1

CeD₃

\tilde{X} C _{3v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e	3	CeD ₃ stretch	918.0	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **104**, 1640 (2000).

NdH₃

\tilde{X} C _{3v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e	3	NdH ₃ stretch	1150.0	Ar	IR	1

NdD₃

\tilde{X} C _{3v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e	3	NdD ₃ stretch	824.6	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **104**, 1640 (2000).

SmH₃

\tilde{X} C _{3v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e	3	SmH ₃ stretch	1203.5	Ar	IR	1

SmD₃

\tilde{X}		C _{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>e</i>	3	SmD ₃ stretch	857.0	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **104**, 1640 (2000).

GdH₃

\tilde{X}		C _{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>e</i>	3	GdH ₃ stretch	1323.6	Ar	IR	1

GdD₃

\tilde{X}		C _{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>e</i>	3	GdD ₃ stretch	947.5	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **104**, 1640 (2000).

TbH₃

\tilde{X}		C _{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>e</i>	3	TbH ₃ stretch	1375.5	Ar	IR	1

TbD₃

\tilde{X}		C _{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>e</i>	3	TbD ₃ stretch	985.2	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **104**, 1640 (2000).

LuH₃

\tilde{X}		C _{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>e</i>	3	LuH ₃ stretch	1386.4	Ar	IR	1

LuD₃

\tilde{X}		C _{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>e</i>	3	LuD ₃ stretch	993.9	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **104**, 1640 (2000).

GaH₃

\tilde{X}		D _{3h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₂ ^{''}	2	OPLA	717.4ms	Ar	IR	1
<i>e</i> [']	3	GaH ₃ stretch	1923.2vs	Ar	IR	1,2
	4	Deformation	758.7s	Ar	IR	1,2

GaD₃

\tilde{X}		D _{3h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₂ ^{''}	2	OPLA	517.5s	Ar	IR	1
<i>e</i> [']	3	GaD ₃ stretch	1387.7vs	Ar	IR	1
	4	Deformation	544.0m	Ar	IR	1

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CH₃⁺

\tilde{A}^1E' D_{3h}
 $T_0 = 50510(280)$ gas PE²

\tilde{a}^3E' D_{3h}
 $T_0 = 39700(280)$ gas PE²

$\tilde{X}^1A'_1$ D_{3h} Structure: LD^{3,4}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_2''	2	OPLA	1359(7)	gas	PE,PD	1,2,6
e'	3	CH stretch	3108.38	gas	LD	3-5
	4	Deformation	1370(7)	gas	PD	6

$B_0 = 9.362$ LD³⁻⁵

CD₃⁺

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_2''	2	OPLA	1070(30)	gas	PE	2

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CH₃

$4f^2E'^a$ D_{3h}
 $T_0 = 72508$ gas MPI¹²
 Higher member of Rydberg series observed at 74961. MPI¹²

$4p^2A_2''$ D_{3h}
 $T_0 = 69853.44(13)$ gas MPI¹⁵
 $B_0 = 9.90$ gas MPI¹⁵

$3d^2A'_1$ D_{3h} Structure: AB²
 $T_0 = 66805$ gas AB^{1,2} $3d^2A'_1 - \tilde{X}$ 147-150 nm
 Ar AB³ $3d^2A'_1 - \tilde{X} \sim 150.3$ nm
 First member of Rydberg series converging to 79392(5). Higher members observed at 72326, 74851, 76256, 77090, and 77643. AB²
 $B_0 = 10.72(8)$ AB²

$3d^2E''$ D_{3h} Structure: AB²
 $T_0 = 66536$ gas AB^{1,2}MPI¹⁰ $3d^2E'' - \tilde{X}$ 144-150 nm
 Ar AB³ $3d^2E'' - \tilde{X} \sim 150.3$ nm
 Diffuse. First member of Rydberg series converging to 79392(5). Higher members observed at 72165, 74851, 76256, 77090, and 77643. AB²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_2''	2	OPLA	1372H	gas	AB MPI	2,10

$3p^2A_2''$ D_{3h} Structure: MPI¹²
 $T_0 = 59972$ gas MPI^{12,28}
 Higher member of Rydberg series observed at 69837. MPI¹²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1'	1	CH stretch	2914	gas	MPI	12
a_2''	2	OPLA	1334	gas	MPI	12

$B_0 = 9.51(7)$; $C_0 = 4.62(3)$ MPI²⁸

$3s^2A'_1$ D_{3h} Structure: AB²
 $T_0 = 46205$ gas AB^{1,2,7}Ra²⁴ $3s^2A'_1 - \tilde{X}$ 216 nm
 Diffuse. First member of Rydberg series converging to 79392(5). Next member observed at 71042. AB²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1'	1	CH stretch	2040T	gas	Ra	24

\tilde{X}^2A_2'' D _{3h} Structure: AB ² IR ^{9,18} CAR ^{23,26}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁ '	1	CH stretch	3004.43(2)	gas	CAR,Ra	13,16,23,26,32
a ₂ '	2	OPLA	606.453	gas	IR,DL	5,9,21
			617vs	Ne	IR	4
			603 ^b	Ar	IR	3,8
			611	N ₂	IR	3
e'	3	CH stretch	3160.821	gas	LD,CC	11,25,27
					CR,IR	30,34-36
			3162wm	Ne	IR	4
			3150	Ar	IR	6
			3171.4	H ₂	IR	29,33
			3170.6			
	4	Deformation	1396w	Ne	IR	4
			1398 ^c	Ar	IR	8
			1402.7	H ₂	IR	29,33
			1401.6			

$B_0 = 9.578$ AB²DL⁹LD³⁴IR³⁶; $C_0 = 4.743$ DL⁹LD³⁴

CD₃

$df^2E'^a$ D_{3h}
 $T_0 = 72431$ gas MPI¹²
 Higher member of Rydberg series observed at 74885. MPI¹²

$4p^2A_2''$ D_{3h}
 $T_0 = 69777.40(4)$ gas MPI¹⁵
 $B_0 = 4.846(2)$ gas MPI¹⁵

$3d^2A_1'$ D_{3h} Structure: AB²
 $T_0 = 66715$ gas AB^{1,2} $3d^2A_1' - \tilde{X}$ 145-150 nm
 Ar AB³ $3d^2A_1' - \tilde{X} \sim 150.3$ nm
 First member of Rydberg series converging to 79315(5). Higher members observed at 72296, 74781, 76181, 77023, 77562, and 77933. AB²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₂ '	2	OPLA	1040H	gas	AB	2

$B_0 = 5.14$ AB²

$3d^2E''$ D_{3h} Structure: AB²
 $T_0 = 66465$ gas AB^{1,2}MPI¹⁰ $3d^2E'' - \tilde{X}$ 146-150 nm
 Ar AB³ $3d^2E'' - \tilde{X} \sim 150.3$ nm
 Diffuse. First member of Rydberg series converging to 79315(5). Higher members observed at 72180, 74753, 76166, 77023, 77562, and 77933. AB²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₂ '	2	OPLA	1031H	gas	AB MPI	2,10

$3p^2A_2''$ D_{3h} Structure: MPI¹
 $T_0 = 59886$ gas MPI^{12,17}
 Higher members of Rydberg series observed at 69789, 73645, and 75557. MPI¹²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁ '	1	CD stretch	2031 ^d	gas	MPI	12,17
a ₂ '	2	OPLA	1032	gas	MPI	12,17

$B_0 = 4.76(2)$; $C_0 = 2.38$ MPI^{12,17}

$3s^2A_1'$ D_{3h} Structure: AB²
 $T_0 = 46629$ gas AB^{1,2,7}Ra²⁴ $3s^2A_1' - \tilde{X}$ 204-225 nm
 First member of Rydberg series converging to 79315(5). Higher members observed at 70910, 74246, 75869, and 76830. AB²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁ '	1	CD stretch	1684 ^d	gas	AB	7
a ₂ '	2	OPLA	1094	gas	AB,Ra	7,24

$B_0 = 4.42$ AB²

\tilde{X}^2A_2'' D _{3h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁ '	1	CD stretch	2157.5(2)	gas	Ra,CAR	19,20
a ₂ '	2	OPLA	457.81	gas	DL	14,18
			463s	Ne	IR	4
			453 ^b	Ar	IR	3,8
			463	N ₂	IR	3
e'	3	CD stretch	2381.09	gas	IR,DL	22,31
			2381w	Ne	IR	4
			2369	Ar	IR	6
	4	Deformation	1026vw	Ne	IR	4
			1029	Ar	IR	6

$B_0 = 4.802$ AB²DL¹⁴

^aTentative assignment.

^bBand center. Rotational structure assigned.⁸

^cR(0₀) transition.

^dApproximate value; perturbed by Fermi resonance.

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SiH₃

$\tilde{M}^2A_2''(6p)$ D_{3h}
T₀=60341 gas MPI⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₂ ''	2	OPLA	800(30)	gas	MPI	7

$\tilde{L}(5d)$ D_{3h}
T₀=59615(30)^c gas MPI⁸

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₂ ''	2	OPLA	839(26)	gas	MPI	8

$\tilde{J}'(4d)$ D_{3h}
T₀=57726(30) gas MPI⁸

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₂ ''	2	OPLA	835(26)	gas	MPI	8

$\tilde{J}^2A_2''(5p)$ D_{3h}
T₀=56929 gas MPI⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₂ ''	2	OPLA	821(4)	gas	MPI	7

$\tilde{I}'(4d)$ D_{3h}
T₀=56253(30) gas MPI⁸

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₂ ''	2	OPLA	814(25)	gas	MPI	8

$\tilde{D}^2A_1'(3d)$ D_{3h}
T₀=49787(30) gas MPI⁸

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₂ ''	2	OPLA	810(31)	gas	MPI	8

$\tilde{E}^2A_2''(4p)$ D_{3h}
T₀=48438 gas MPI^{5,7}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₂ ''	2	OPLA	796(7)	gas	MPI	5,7
e'	4	Deformation	870(5)H	gas	MPI	7

\tilde{A}^2A_1 C_{3v}
gas AB^{9,12} $\tilde{A}-\tilde{X}$ 205–256 nm

\tilde{X}^2A_1 C_{3v} Structure: ESR^{1,2}DL³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	Umbrella	727.94 ^a 721.05 ^b 733.6 732.4 730.0	gas gas Ar Kr	DL,MPI DL,MPI IR IR	3,6,7 3,6,7 13 11
e	3	SiH ₃ stretch	2185.2	gas	DL	10
	4	Deformation	929T	Kr	IR	11

Barrier to inversion=1935 gas PE⁴MPI⁷
B₀=4.763 DL³

SiD₃

$\tilde{P}^2A_2''(7p)$ D_{3h}
T₀=62002 gas MPI⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₂ ''	2	OPLA	615(3)	gas	MPI	7

$\tilde{N}^2E'(5f)$ D_{3h}
T₀=61005 gas MPI^{7,8}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₂ ''	2	OPLA	619(7)	gas	MPI	7

$\tilde{M}^2A_2''(6p)$ D_{3h}
T₀=60267 gas MPI⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₂ ''	2	OPLA	615(4)	gas	MPI	7

$\tilde{K}^2E'(4f)$ D_{3h}
T₀=58417 gas MPI^{7,8}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₂ ''	2	OPLA	602(10)	gas	MPI	8

$\tilde{J}'(4d)$		D _{3h}					
$T_0=57840(30)^c$		gas		MPI ⁸			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a_2''	2	OPLA	603(20)	gas	MPI	8	

$\tilde{J}^2A_2''(5p)$		D _{3h}					
$T_0=56874$		gas		MPI ⁷			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a_2''	2	OPLA	608(3)	gas	MPI	7	

$\tilde{I}'(4d)$		D _{3h}					
$T_0=56205(30)$		gas		MPI ⁸			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a_2''	2	OPLA	600(17)	gas	MPI	8	

$\tilde{H}^2E'(4p)$		D _{3h}					
$T_0\cong 50000$		gas		MPI ^{7,8}			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a_2''	2	OPLA	602(5)	gas	MPI	7	

$\tilde{D}^2A_1''(3d)$		D _{3h}					
$T_0=49685(30)^c$		gas		MPI ⁸			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a_2''	2	OPLA	600(28)	gas	MPI	8	

$\tilde{E}^2A_2''(4p)$		D _{3h}					
$T_0=48391$		gas		MPI ⁷			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a_1'	1	SiD ₃ stretch	1576(3)	gas	MPI	7	
a_2''	2	OPLA	589(3)	gas	MPI	7	
e'	4	Deformation	635(6)H	gas	MPI	7	

\tilde{X}^2A_1		C _{3v}					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a_1	2	Umbrella	545 ^a	gas	MPI	7	
			542 ^b	gas	MPI	7	
			545.5	Ar	IR	13	
			548.9	Kr	IR	11	
			547.2				
			545.8	D ₂	IR	13	

Barrier to inversion=1925 gas MPI⁷

^a1⁻-0⁺ transition.

^b1⁺-0⁻ transition.

^cExtrapolated value.

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NH₃⁺

\tilde{A}^2E D_{3h}
 $T_0=36470(100)$ gas PE^{3,4,12}

Broad, partially resolved vibrational structure has been discussed³ in terms of the expected Jahn-Teller distortion. Continuous background absorption may be associated with the formation of NH₂⁺, for which the threshold is ~44700,^{1,3} or NH⁺.

\tilde{X}^2A_2''		D _{3h}					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a_1'	1	NH stretch	3150(100)	gas	PE	5,10	
a_2''	2	OPLA	903.39	gas	PE,DL	3,7,11,12	
			916.8	Ne	IR	13	
e'	3	NH stretch	3388.65	gas	LD	6	
			3404.6	Ne	IR	13	
			1507.1	gas	PE,TPE	8-11	
4	Bend	1516.8	Ne	IR	13		

$B_0=10.644$; $C_0=5.247$ LD⁶DL⁷

ND₃⁺

\tilde{X}^2A_2''		D _{3h}					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a_1'	1	ND stretch	2453(56)	gas	PE	10	
a_2''	2	OPLA	684(4)	gas	PE	2,11	
			683.3	Ne	IR	13	
e'	3	ND stretch	2551.8	Ne	IR	13	
			1138(56)	gas	PE	10	
4	Bend	1114.6	Ne	IR	13		

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H₃O⁺

\tilde{X}		C _{3v}		Structure: CC ⁴ MW ¹⁹		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	OH stretch	3491.17 ^a	gas	LD	18
			3389.66 ^b	gas	LD	18
	2	Umbrella	954.40 ^a	gas	DL	2,3,5,7
e	3	OH stretch	525.82 ^b	gas	DL	5-7
			3535.56 ^{c,d}	gas	CC	1,4,10,15
	4	Deformation	3518.95 ^{c,e}	gas	CC	1,4,10,14,15
			1625.95 ^d	gas	DL	11
			1638.53 ^e	gas	DL	11

$B(0^+) = 11.254$; $C(0^+) = 6.129$; $B(0^-) = 11.054$; $C(0^-) = 6.211$
 IR, MW^{9,12,15} CC¹⁶ LD^{16,18}

D₃O⁺

\tilde{X}		C _{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	Umbrella	645.13 ^a	gas	DL	8
			438.39 ^b	gas	DL	8
e	3	OD stretch	2629.65 ^d	gas	LD	13
			2624.24 ^e	gas	LD	13

$B(0^+) = 5.675$; $C(0^+) = 3.15(5)$; $B(0^-) = 5.644$; $C(0^-) = 3.16(5)$
 DL⁸ LD¹³ MW¹⁹

^a1⁻-0⁺ transition.

^b1⁺-0⁻ transition. For H₃O⁺, 0⁻-0⁺ = 55.35 cm⁻¹,^{6,12} and for D₃O⁺, 15.36 cm⁻¹.^{8,17,19}

^c $\nu - C' \zeta + 7/4 \eta_K$.

^d1⁺-0⁺ transition.

^e1⁻-0⁻ transition.

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H₃S⁺

\tilde{X}		C _{3v}		Structure: MW ^{6,7}		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	SH stretch	2521.05	gas	LD	3
			1033.31	gas	DL	2
e	3	SH stretch	2525.74	gas	LD	1,3

$B_0 = 4.895$; $C_0 = 4.228$ LD^{1,3} DL² MW⁴

D₃S⁺

\tilde{X}		C _{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	SD stretch	1827.22	gas	DL	5
e	3	SD stretch	1838.62	gas	DL	5

$B_0 = 2.542$; $C_0 = 2:12(3)$ DL⁵ MW⁶

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8.5. Four-Atomic Dihydrides

cyc-Pd₂H₂

\tilde{X} D _{2h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _{2u}		PdH stretch	1345.8	Ne	IR	1
			1347.7	Ar	IR	1

cyc-Pd₂D₂

\tilde{X} D _{2h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _{2u}		PdD stretch	953.9	Ne	IR	1
			950.2	Ar	IR	1

Reference

¹L. Andrews, X. Wang, M. E. Alikhani, and L. Manceron, *J. Phys. Chem. A* **105**, 3052 (2001).

cyc-(CeH)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Asym. stretch	1128.0	Ar	IR	1

cyc-(CeD)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Asym. stretch	808.3	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **104**, 1640 (2000).

cyc-(PrH)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Asym. stretch	1135.6	Ar	IR	1

cyc-(PrD)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Asym. stretch	813.4	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **104**, 1640 (2000).

cyc-(GdH)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Asym. stretch	1201.9	Ar	IR	1

cyc-(GdD)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Asym. stretch	860.9	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **104**, 1640 (2000).

cyc-(TbH)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Asym. stretch	1212.1	Ar	IR	1

cyc-(TbD)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Asym. stretch	868.8	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **104**, 1640 (2000).

cyc-(LuH)₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Asym. stretch	1273.2	Ar	IR	1

cyc-(LuD)₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Asym. stretch	912.1	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **104**, 1640 (2000).

LiNH₂

\tilde{X} C_{2v} Structure: MW¹
 $A_0=13.161(2)$; $B_0=1.011$; $C_0=0.933$ MW¹

LiND₂

\tilde{X} C_{2v}
 $A_0=6.584$; $B_0=0.894$; $C_0=0.781$ MW¹

Reference

¹D. B. Grotjahn, P. M. Sheridan, I. A. Jihad, and L. M. Ziurys, *J. Am. Chem. Soc.* **123**, 5489 (2001).

NaNH₂

\tilde{X} C_{2v} Structure: MW²
 $A_0=12.933(3)$; $B_0=0.375$; $C_0=0.364$ MW¹

NaND₂

\tilde{X} C_{2v}
 $A_0=6.506$; $B_0=0.330$; $C_0=0.313$ MW²

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¹J. Xin, M. A. Brewster, and L. M. Ziurys, *Astrophys. J.* **530**, 323 (2000).

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HcBeH \tilde{X} $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	2	BeH stretch	2087.3	Ar	IR	1
	3	CBe stretch	983.7	Ar	IR	1
Π	4	Deformation	534.2	Ar	IR	1

DCBeD \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	2	BeD stretch	1593.5	Ar	IR	1
	3	CBe stretch	935.2	Ar	IR	1
Π	4	Deformation	430.2 427.8	Ar	IR	1

Reference

¹T. M. Greene, D. V. Lanzisera, L. Andrews, and A. J. Downs, *J. Am. Chem. Soc.* **120**, 6097 (1998).

MgCH₂ \tilde{X}^3B_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CH ₂ s-stretch	2786.8	Ar	IR	1
	3	MgC stretch	502.2	Ar	IR	1

MgCD₂ \tilde{X}^3B_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CD ₂ s-stretch	2040.3	Ar	IR	1
	3	MgC stretch	472.1	Ar	IR	1

Reference

¹W. D. Bare, A. Citra, C. Trindle, and L. Andrews, *Inorg. Chem.* **39**, 1204 (2000).

MgNH₂

\tilde{X}^2A_1 C_{2v} Structure: MW²
 $A_0=12.626$; $B_0=0.444$; $C_0=0.428$ MW^{1,2}

MgND₂

\tilde{X}^2A_1 C_{2v}
 $A_0=6.334$; $B_0=0.388$; $C_0=0.365$ MW²

References

- ¹P. M. Sheridan and L. M. Ziurys, *Astrophys. J.* **540**, L61 (2000).
²P. M. Sheridan and L. M. Ziurys, *Can. J. Phys.* **79**, 409 (2001).

CaNH₂

\tilde{C}^2A_1 C_{2v} Structure: LF^{2,7}
 $T_0=17375.167$ gas CL¹LF^{2,7} $\tilde{C}-\tilde{X}$ 575 nm
 $A=12.95$; $1/2(B+C)=0.302$; $1/4(B-C)=0.0018$ LF^{2,7}

\tilde{B}^2B_1 C_{2v} Structure: LF⁷
 $T_0=15885.28$ gas CL¹LF^{3-5,7,8} $\tilde{B}-\tilde{X}$ 620–650 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	3	CaN stretch	545.8(5)	gas	LF	4

$A_0=14.365$; $1/2(B_0+C_0)=0.302$; $1/4(B_0-C_0)=0.004$ LF^{5,7,8}

\tilde{A}^2B_2 C_{2v} Structure: LF⁷
 $T_0=15464.367$ gas CL¹LF³⁻⁶ $\tilde{A}-\tilde{X}$ 620–650 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	3	CaN stretch	549.5(1.0)	gas	LF	3,4

$A_0=11.449$; $B_0=0.307$; $C_0=0.299$ LF^{5,6}

\tilde{X}^2A_1 C_{2v} Structure: LF^{2,6,7}MW⁹
 Vib. sym. No. Approximate type of mode cm⁻¹ Med. Type meas. Refs.
 a_1 3 CaN stretch 524(10) gas LF 3
 $A=13.080(3)$; $B_0=0.300$; $C_0=0.293$ LF^{2,5,6}MW⁹

CaND₂

\tilde{X}^2A_1 C_{2v}
 $A_0=6.527$; $B_0=0.260$; $C_0=0.250$ MW⁹

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⁹M. A. Brewster and L. M. Ziurys, *J. Chem. Phys.* **113**, 3141 (2000).

SrNH₂

\tilde{C}^2A_1 C_{2v}
 $T_0=15862(10)$ gas CL¹LF² $\tilde{C}-\tilde{X}$ 632 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	3	SrN stretch	458(5)	gas	LF	2,3
b_1	4	OPLA	290(30)	gas	LF	3
b_2	6	Deformation	290(30)	gas	LF	3

\tilde{B}^2B_1 C_{2v} Structure: LF³
 $T_0=14689.32$ gas CL¹LF^{2,3} $\tilde{B}-\tilde{X}$ 670–725 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	3	SrN stretch	444(5)	gas	LF	2,3

$A_0=14.714$; $(B_0+C_0)=0.455$; $(B_0-C_0)=0.004$ LF³

\tilde{A}^2B_2 C_{2v} Structure: LF³
 $T_0=14230.62$ gas CL¹LF^{2,3} $\tilde{A}-\tilde{X}$ 670–725 nm
 $A_0=12.103$; $(B_0+C_0)=0.457$; $(B_0-C_0)=0.005$ LF³

\tilde{X}^2A_1 C_{2v} Structure: LF³MW⁴
 Vib. sym. No. Approximate type of mode cm⁻¹ Med. Type meas. Refs.
 a_1 1 NH₂ s-stretch 3293(10) gas LF 3
 a_1 2 NH₂ scissors 1678(10) gas LF 3
 a_1 3 SrN stretch 459(5) gas LF 2,3
 b_1 4 OPLA 320(30) gas LF 3
 b_2 6 Deformation 320(30) gas LF 3

$A_0=13.154(5)$; $B_0=0.226$; $C_0=0.222$ LF³MW⁴

SrND₂

\tilde{X}^2A_1 C_{2v}
 $A_0=6.557$; $B_0=0.194$; $C_0=0.188$ MW⁴

References

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³C. R. Brazier and P. F. Bernath, *J. Mol. Spectrosc.* **201**, 116 (2000).
⁴J. M. Thompsen, P. M. Sheridan, and L. M. Ziurys, *Chem. Phys. Lett.* **330**, 373 (2000).

HYOH⁺

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		YH stretch	1534.2	Ar	IR	1
		YO stretch	695.1	Ar	IR	1

DYOD⁺

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		YD stretch	1099.6	Ar	IR	1
		YO stretch	681.6	Ar	IR	1

Reference

¹L. Zhang, L. Shao, and M. Zhou, Chem. Phys. **272**, 27 (2001).

HCCH⁺

$\tilde{C}^2\Sigma^+$ D _{∞h} T ₀ =92460(80) gas PE ⁶						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	2	CC stretch	1370T	gas	PE	6

$\tilde{B}^2\Sigma_u^+$ D _{∞h} T ₀ =56380(80) gas PE ^{1,3}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	1	CH s-stretch	2500(20)	gas	PE	1,3
	2	CC stretch	1815(20)	gas	PE	1,3

τ < 14 fs gas PE³

$\tilde{A}^2A_g^a$ C _{2h} Structure: MPD ¹¹ T ₀ =39109.7(1.0) gas PE ³ MPD ¹¹ $\tilde{A}-\tilde{X}$ 240–255 nm						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a _g	1	CH s-stretch	2530(20)	gas	PE	3
	2	CC stretch	1730(20)	gas	PE	3
	3	Bend	793 ^b	gas	MPD	11
a _u	4	Torsion	282H ^b	gas	MPD	11
b _u	6	Bend	436H ^b	gas	MPD	11

τ₀ = 150 ps gas MPD¹¹
A₀ = 12.15; B₀ = 0.94(5) MPD¹¹

$\tilde{X}^2\Pi_u$ D _{∞h} Structure: LD ⁹						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	2	CC stretch	1829(3)	gas	PE	1,3
Σ _u ⁺	3	CH a-stretch	3135.98	gas	LD	5,9
			3137.6	Nc	IR	8,12
			3105.5	Ar	IR	4,8,12
Π _g	4	Deform.	694 ^c	gas	PE,TPE	7,10
Π _u	5	Deform.	837(12) ^c	gas	PE	3

A = -30.91(2) gas LD^{5,9}
B₀ = 1.105 LD^{5,9}

DCCD⁺

$\tilde{B}^2\Sigma_u^+$ D _{∞h} T ₀ =56655(80) gas PE ³						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	1	CC stretch	2275(20)	gas	PE	1,3
	2	CD s-stretch	1475(20)	gas	PE	1,3

\tilde{A}^2A_g C _{2h} T ₀ =39906(80) gas PE ³						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a _g	1	CD s-stretch	2280(20)	gas	PE	3
	2	CC stretch	1450(20)	gas	PE	3
	3	Bend	339(12)	gas	PE	3
b _u	6	Bend	516(12)	gas	PE	3

$\tilde{X}^2\Pi_u$ D _{∞h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	1	CD s-stretch	2572(16)	gas	PE	3
	2	CC stretch	1651(4)	gas	PE	1,3
Σ _u ⁺	3	CD a-stretch	2329.1	Ne	IR	8,12
			2311.5	Ar	IR	4,8,12
Π _u	5	Bend	702(12) ^c	gas	PE	3

^aThreshold for formation of HCC⁺ ≤ 48000 cm⁻¹,^{2,3}

^bAll three bending fundamentals have exceptionally large anharmonicities.
^cω₄ · ε₄ = 0.30,¹⁰ The identification^{7,10} of the Renner components of ν₄ (Π_g) of HCCH⁺ in the photoelectron spectrum of HCCH (\tilde{A}^1A_u) necessitates the reassignment to ν₅ (Π_u) of the peaks previously assigned³ to ν₄ of HCCH⁺ and DCCD⁺.

References

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¹¹Ch. Cha, R. Weinkauff, and U. Boesl, *J. Chem. Phys.* **103**, 5224 (1995).

¹²L. Andrews, G. P. Kushto, M. Zhou, S. P. Willson, and P. F. Souter, *J. Chem. Phys.* **110**, 4457 (1999).

HScOH

In an argon matrix,¹ photolyzes with 300–400 nm radiation, producing H₂ + ScO.

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		ScH stretch	1482.6	Ar	IR	1,2
		ScO stretch	713.0	Ar	IR	1,2

DScOD

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		ScD stretch	1068.5	Ar	IR	1,2
		ScO stretch	693.6	Ar	IR	1,2

References

¹J. W. Kauffman, R. H. Hauge, and J. L. Margrave, *J. Phys. Chem.* **89**, 3547 (1985).

²L. Zhang, J. Dong, and M. Zhou, *J. Phys. Chem. A* **104**, 8882 (2000).

HYOH

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		YH stretch	1399.1	Ar	IR	1
		YO stretch	641.5	Ar	IR	1

DYOD

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		YD stretch	1002.9	Ar	IR	1
		YO stretch	623.1	Ar	IR	1

Reference

¹L. Zhang, L. Shao, and M. Zhou, *Chem. Phys.* **272**, 27 (2001).

HLaOH

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		LaH stretch	1263.7	Ar	IR	1
		LaO stretch	576.4	Ar	IR	1

DLaOD

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		LaD stretch	905.8	Ar	IR	1
		LaO stretch	559.7	Ar	IR	1

Reference

¹L. Zhang, L. Shao, and M. Zhou, *Chem. Phys.* **272**, 27 (2001).

HTiOH

In an argon matrix,¹ photolyzes with 400–500 nm radiation, producing H₂ + TiO.

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		TiH stretch	1538.9	Ar	IR	1,2
		TiO stretch	699.7	Ar	IR	1

DTiOD

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		TiD stretch	1107.7	Ar	IR	1
		TiO stretch	697.3	Ar	IR	1

References

¹J. W. Kauffman, R. H. Hauge, and J. L. Margrave, *J. Phys. Chem.* **89**, 3547 (1985).

²M. Zhou, L. Zhang, J. Dong, and Q. Qin, *J. Am. Chem. Soc.* **122**, 10680 (2000).

H₂TiO $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		TiH ₂ s-stretch	1646.8	Ar	IR	1
		TiH ₂ a-stretch	1611.9	Ar	IR	1
		TiO stretch	1010.5	Ar	IR	1

D₂TiO $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		TiD ₂ a-stretch	1168.2	Ar	IR	1
		TiO stretch	1009.6	Ar	IR	1

Reference

¹M. Zhou, L. Zhang, J. Dong, and Q. Qin, *J. Am. Chem. Soc.* **122**, 10680 (2000).

HZrOH $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		ZrH stretch	1530.9	Ar	IR	1

DZrOD $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		ZrD stretch	1098.2	Ar	IR	1

Reference

¹M. Zhou, L. Zhang, J. Dong, and Q. Qin, *J. Am. Chem. Soc.* **122**, 10680 (2000).

H₂ZrO $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		ZrH ₂ s-stretch	1577.6	Ar	IR	1
		ZrH ₂ a-stretch	1539.4	Ar	IR	1
		ZrO stretch	924.7	Ar	IR	1

D₂ZrO $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		ZrD ₂ s-stretch	1130.3	Ar	IR	1
		ZrD ₂ a-stretch	1107.8	Ar	IR	1
		ZrO stretch	922.4	Ar	IR	1

Reference

¹M. Zhou, L. Zhang, J. Dong, and Q. Qin, *J. Am. Chem. Soc.* **122**, 10680 (2000).

H₂HfO $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		HfH ₂ s-stretch	1646.4	Ar	IR	1
		HfH ₂ a-stretch	1615.6	Ar	IR	1
		HfO stretch	921.0	Ar	IR	1

D₂HfO $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		HfD ₂ a-stretch	1158.1	Ar	IR	1
		HfO stretch	918.7	Ar	IR	1

Reference

¹M. Zhou, L. Zhang, J. Dong, and Q. Qin, *J. Am. Chem. Soc.* **122**, 10680 (2000).

HVOH

In an argon matrix,¹ photolyzes with radiation having a short wavelength cutoff of 400 nm.

 $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		VH stretch	1567.0	Ar	IR	1,2
		VO stretch	703.3T	Ar	IR	1
		Bend	414.5T	Ar	IR	1

DVOD

 $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		VD stretch	1128.4	Ar	IR	1,2
		VO stretch	696.6T	Ar	IR	1

References

¹J. W. Kauffman, R. H. Hauge, and J. L. Margrave, *J. Phys. Chem.* **89**, 3547 (1985).

²M. Zhou, J. Dong, L. Zhang, and Q. Zin, *J. Am. Chem. Soc.* **123**, 135 (2001).

H₂VO $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	VH s-stretch	1710.6	Ar	IR	1
	2	VO stretch	1029.4	Ar	IR	1
		VH a-stretch	1680.4	Ar	IR	1

D₂VO $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	VD s-stretch	1235.1	Ar	IR	1
	2	VO stretch	1029.1	Ar	IR	1
		VD a-stretch	1215.3	Ar	IR	1

Reference

¹M. Zhou, J. Dong, L. Zhang, and Q. Qin, *J. Am. Chem. Soc.* **123**, 135 (2001).

HNbOH

 $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	2	NbH stretch	1587.5	Ar	IR	1

DNbOD

 $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	2	NbD stretch	1141.2	Ar	IR	1

Reference

¹M. Zhou, J. Dong, L. Zhang, and Q. Qin, *J. Am. Chem. Soc.* **123**, 135 (2001).

H₂ONb $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		H ₂ O bend	1569.1	Ar	IR	1

D₂ONb $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		D ₂ O bend	1153.2	Ar	IR	1

Reference

¹M. Zhou, J. Dong, L. Zhang, and Q. Qin, *J. Am. Chem. Soc.* **123**, 135 (2001).

H₂NbO $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NbH s-stretch	1704.5	Ar	IR	1
	2	NbO stretch	976.9	Ar	IR	1
		NbH a-stretch	1665.2	Ar	IR	1

D₂NbO $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NbD s-stretch	1224.0	Ar	IR	1
	2	NbO stretch	976.9	Ar	IR	1
		NbD a-stretch	1197.7	Ar	IR	1

Reference

¹M. Zhou, J. Dong, L. Zhang, and Q. Qin, *J. Am. Chem. Soc.* **123**, 135 (2001).

H₂TaO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	TaH s-stretch	1780.4	Ar	IR	1
	2	TaO stretch	986.9	Ar	IR	1
		TaH a-stretch	1762.3	Ar	IR	1

D₂TaO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	2	TaO stretch	984.9	Ar	IR	1
		TaD a-stretch	1264.7	Ar	IR	1

Reference

¹M. Zhou, J. Dong, L. Zhang, and Q. Qin, *J. Am. Chem. Soc.* **123**, 135 (2001).

HMnOH

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	2	MnH stretch	1663.4	Ar	IR	1,2
		MnO stretch	648.1	Ar	IR	1,2

DMnOD

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	2	MnD stretch	1197.1	Ar	IR	1,2
		MnO stretch	628.5	Ar	IR	1,2

References

¹J. W. Kauffman, R. H. Hauge, and J. L. Margrave, *J. Phys. Chem.* **89**, 3541 (1985).

²M. Zhou, L. Zhang, L. Shao, W. Wang, K. Fan, and Q. Qin, *J. Phys. Chem. A* **105**, 5801 (2001).

HFeOH

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	2	FeH stretch	1731.9	Ar	IR	1,2
		FeO stretch	682.4	Ar	IR	1,2
		Bend	457.6	Ar	IR	1

DFeOD

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	2	FeD stretch	1245.3	Ar	IR	1,2
		FeO stretch	660.5	Ar	IR	1,2

References

¹J. W. Kauffman, R. H. Hauge, and J. L. Margrave, *J. Phys. Chem.* **89**, 3541 (1985).

²L. Zhang, M. Zhou, L. Shao, W. Wang, K. Fan, and Q. Qin, *J. Phys. Chem. A* **105**, 6998 (2001).

AlNH₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	2	NH ₂ scissors	1520.3	Ar	IR	1,2
	3	AlN stretch	726.3	Ar	IR	1,2
<i>b</i> ₁	4	OPLA	406.7	Ar	IR	1,2
<i>b</i> ₂	5	NH ₂ a-stretch	3495.1	Ar	IR	1,2

AlND₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	2	ND ₂ scissors	1137.7	Ar	IR	1,2
	3	AlN stretch	694.8	Ar	IR	1,2
<i>b</i> ₁	4	OPLA	314.6	Ar	IR	2

References

¹D. V. Lanzisera and L. Andrews, *J. Phys. Chem. A* **101**, 5082 (1997).

²H.-J. Himmel, A. J. Downs, and T. M. Greene, *J. Am. Chem. Soc.* **122**, 9793 (2000).

GaNH₂

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	NH ₂ scissors	1505.9	Ar	IR	1,2
	3	GaN stretch	589.3	Ar	IR	1,2
b ₁	4	OPLA	303.3	Ar	IR	2
b ₂	5	NH ₂ a-stretch	3471.6	Ar	IR	1,2

GaND₂

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	ND ₂ scissors	1132.6	Ar	IR	2
	3	GaN stretch	557.9	Ar	IR	2

References

- ¹H.-J. Himmel, A. J. Downs, and T. M. Greene, Chem. Commun. 871 (2000).
²H.-J. Himmel, A. J. Downs, and T. M. Greene, J. Am. Chem. Soc. **122**, 9793 (2000).

InNH₂

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	NH ₂ scissors	1498.1	Ar	IR	1
	3	InN stretch	498.7	Ar	IR	1
b ₁	4	OPLA	237.0	Ar	IR	1
b ₂	5	NH ₂ a-stretch	3481.7	Ar	IR	1
	6	NH ₂ rock	469.6	Ar	IR	1

InND₂

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	ND ₂ scissors	1116.3	Ar	IR	1
	3	InN stretch	480.6	Ar	IR	1

Reference

- ¹H.-J. Himmel, A. J. Downs, and T. M. Greene, J. Am. Chem. Soc. **122**, 9793 (2000).

H₂CSi

\tilde{B}^1B_2		C _{2v}		Structure: LF ⁴		$\tilde{B}-\tilde{X}$ 300–342 nm	
T ₀ = 29312.88		gas		AB ¹ LF ³⁻⁵			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a ₁	1	CH ₂ stretch	2997	gas	LF	4,5	
	2	CH ₂ scissors	1101.96	gas	AB,LF	1,4,5	
	3	CSi stretch	702.00	gas	AB,LF	1,3–5	
b ₁	4	OPLA	697	gas	LF	5	
b ₂	6	CH ₂ rock	731	gas	LF	4,5	

A₀ = 8.533; B₀ = 0.510; C₀ = 0.479 AB¹LF⁴

\tilde{X}^1A_1		C _{2v}		Structure: MW ² LF ⁴			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a ₁	1	CH ₂ s-stretch	2947(5)	gas	LF	6	
	2	CH ₂ scissors	1273(3)	gas	SEP	6	
	3	CSi stretch	933(3)	gas	SEP	6	
b ₁	4	OPLA	687(5)H	gas	SEP	6	
b ₂	6	CH ₂ rock	263(5)	gas	LF	6	

A₀ = 10.186; B₀ = 0.552; C₀ = 0.521 AB¹MW²LF⁴

D₂CSi

\tilde{B}^1B_2		C _{2v}		Structure: AB ¹ LF ^{4,5}		$\tilde{B}-\tilde{X}$ 308–342 nm	
T ₀ = 29237.35		gas		AB ¹ LF ^{4,5}			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a ₁	1	CD ₂ stretch	2180	gas	LF	4,5	
	2	CD ₂ scissors	831	gas	AB,LF	1,4,5	
	3	CSi stretch	681	gas	AB,LF	1,4,5	
	4	OPLA	547	gas	LF	5	
b ₂	6	CD ₂ rock	549	gas	LF	4,5	

A₀ = 4.286; B₀ = 0.443; C₀ = 0.400 LF⁴

\tilde{X}^1A_1		C _{2v}					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a ₁	1	CD ₂ s-stretch	2185(3)	gas	SEP	6	
	2	CD ₂ scissors	1054(3)	gas	SEP	6	
	3	CSi stretch	808(3)	gas	SEP	6	
b ₁	4	OPLA	539(3)H	gas	SEP	6	
b ₂	6	CD ₂ rock	216(5)	gas	LF	6	

A₀ = 5.084; B₀ = 0.468; C₀ = 0.425 LF⁴

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H₂CGe

\tilde{B}^1B_2 C_{2v} Structure: LF²
 $T_0=27330.42$ gas LF^{1,2} $\tilde{B}-\tilde{X}$ 354–367 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CH ₂ scissors	1024	gas	LF	2
	3	CGe stretch	548	gas	LF	1,2
b_1	4	OPLA	613	gas	LF	2
b_2	6	CH ₂ rock	692	gas	LF	2

$\tau=2 \mu\text{s}$ gas LF²
 $A_0=8.318$; $B_0=0.368$; $C_0=0.351$ LF²

\tilde{X}^1A_1 C_{2v} Structure: LF²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CH ₂ scissors	1238	gas	LF	3
	3	GeC stretch	782	gas	LF	3
b_1	4	OPLA	673	gas	LF	3
b_2	6	CH ₂ rock	351	gas	LF	3

$A_0=10.055$; $B_0=0.401$; $C_0=0.385$ LF²

D₂CGe

\tilde{B}^1B_2 C_{2v} Structure: LF²
 $T_0=27278.45$ gas LF² $\tilde{B}-\tilde{X}$ 354–367 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CD ₂ scissors	782	gas	LF	2
	3	CGe stretch	528	gas	LF	2
b_1	4	OPLA	465	gas	LF	2
b_2	6	CD ₂ rock	513	gas	LF	2

$A_0=4.179$; $B_0=0.315$; $C_0=0.293$ LF²

\tilde{X}^1A_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CD ₂ s-stretch	2089	gas	LF	3
	2	CD ₂ scissors	1000	gas	LF	3
	3	GeC stretch	694	gas	LF	3
b_1	4	OPLA	520	gas	LF	3
b_2	6	CD ₂ rock	282	gas	LF	3

$A_0=5.025$; $B_0=0.334$; $C_0=0.312$ LF²

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br-Si₂H₂

\tilde{X} C_{2v} Structure: MW^{1,2}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2		890.3w	Ar	IR	3
b_1	5		1475.6vw	Ar	IR	3
b_2	6		1092.8vs	Ar	IR	3

$A_0=5.244$; $B_0=0.243$; $C_0=0.240$ MW^{1,2}

br-Si₂D₂

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b_2	6		818	Ar	IR	3

$A_0=2.688$; $B_0=0.240$; $C_0=0.234$ MW²

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HCNH⁺

\tilde{X} $C_{\infty v}$ Structure: LD⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	NH stretch	3482.84	gas	LD	1,2
	2	CH stretch	3187.86	gas	LD	1,2
	3	CN stretch	2155.70	gas	DL	7,8
Π	4	HCN bend	801.59	gas	DL	5
	5	HNC bend	645.92	gas	DL	6

$B_0=1.236$ LD^{1,2}MW^{3,9}

DCND⁺

\tilde{X} $C_{\infty v}$
 $B_0=0.892$ MW⁹

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H₂CN

$\tilde{C}^2B_1^a$ C_{2v}
 $T_0=35620$ gas AB^{2,3,5} $\tilde{C}-\tilde{X}$ 280–285 nm
 35436(25) Ar AB⁶ $\tilde{C}-\tilde{X}$ 270–285 nm
 Gas-phase absorption is diffuse.⁵ Photolysis is observed in an argon matrix.⁶

$\tilde{B}^2A_1^a$ C_{2v}
 $T_0=35075^a$ gas AB^{2,3,5} $\tilde{B}-\tilde{X}$ 280–285 nm
 34990(25) Ar AB⁶ $\tilde{B}-\tilde{X}$ 270–285 nm
 All gas-phase bands are diffuse.⁵ Photolysis is observed in an argon matrix.⁶

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CH ₂ s-stretch	2774(50)T	Ar	AB	6
	2	CN stretch	1883(50)	Ar	AB	6
	3	CH ₂ scissors	1413(50)	Ar	AB	6

\tilde{X}^2B_2 C_{2v} Structure: ESR^{1,4}MW⁸

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CH ₂ s-stretch	2820T	gas	PE	7
	2	CN stretch	1725.4vw	Ar	IR	6
	3	CH ₂ scissors	1337T	gas	PE	7
b ₁			1336.6m	Ar	IR	6
			1332.7	Kr	IR	9
			1328.7	Xe	IR	9
	4	OPLA	954.1wm	Ar	IR	6
			950.3	Kr	IR	9
			3103.2m	Ar	IR	6
b ₂	5	CH ₂ a-stretch	912.8w	Ar	IR	6
			909.0	Kr	IR	9

$A_0=9.484$; $B_0=1.306$; $C_0=1.142$ MW⁸

D₂CN

$\tilde{C}^2B_1^a$ C_{2v}
 $T_0=35481^a$ gas AB^{2,5} $\tilde{C}-\tilde{X}$ 271–285 nm

$\tilde{B}^2A_1^a$ C_{2v}
 $T_0=35036^a$ gas AB^{2,5} $\tilde{B}-\tilde{X}$ 271–285 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	CN stretch	1894T	gas	AB	2,5
	3	CD ₂ scissors	1079T	gas	AB	2,5

\tilde{X}^2B_2 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CD ₂ s-stretch	2080T	gas	PE	7
		CD ₂ scissors	990T	gas	PE	7
				1073.4	Ar	IR
b ₁	4	OPLA	776	Ar	IR	6
b ₂	5	CD ₂ a-stretch	2427.5T	Ar	IR	6

^aTentative assignment.

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HCNH

\tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		CNH deform.	1217.9T	Kr	IR	2
			886.3T	Ar	IR	1
			882.8T	Kr	IR	2

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H₂CP

\tilde{X} C_{2v} Structure: MW¹
 $A_0=9.879$; $B_0=0.566$; $C_0=0.534$ MW¹

D₂CP

\tilde{X} C_{2v}
 $A_0=4.951$; $B_0=0.477$; $C_0=0.434$ MW¹

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H₂CO⁺
 \tilde{C}^2B_2 C_{2v}
 $T_0=43172(11)$ gas PE^{1,4,5}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1			1895T	gas	PE	4,5
			1412T	gas	PE	1,4,5

 \tilde{B}^2A_1 C_{2v}
 $T_0=39928(6)$ gas PE^{1,4,5}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CO stretch	1304(4)	gas	PE	1,4,5

 \tilde{A}^2B_1 C_{2v}
 $T_0=25929(5)$ gas PE^{1,4,5}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1			1488(4)	gas	PE	4,5
			1250(4)	gas	PE	1,4,5
b_1	4	OPLA	263(4)	gas	PE	4,5

 \tilde{X}^2B_2 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CH ₂ s-stretch	2580(4)	gas	PE	1,4,5
	2	CO stretch	1675(4)	gas	PE	4,5
	3	CH ₂ scissors	1210(4)	gas	PE	1,4,5,7
b_1	4	OPLA	919(24)	gas	PE	7
b_2	5	CH ₂ a-stretch	2718(24)	gas	PE	7
	6	CH ₂ deform.	777(4)	gas	PE	4,5,7

 $A_0=8.875(3)$; $B_0=1.31(1)$; $C_0=1.14(1)$ TPE⁶
D₂CO⁺
 \tilde{C}^2B_2 C_{2v}
 $T_0=44600(160)$ gas PE^{4,5}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1			1604T	gas	PE	4,5
			943T	gas	PE	1,4,5
			819T	gas	PE	4,5

 \tilde{B}^2A_1 C_{2v}
 $T_0=39816(5)$ gas PE^{1,4,5}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CO stretch	1311(4)	gas	PE	1,4,5
	3	CD ₂ scissors	957(4)	gas	PE	1,4,5

 \tilde{A}^2B_1 C_{2v}
 $T_0=25756(7)$ gas PE^{1,4,5}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2		1282(4)	gas	PE	1,4,5
	3		1064(4)	gas	PE	4,5
b_1	4	OPLA	777(4)	gas	PE	5

$\tau_0=64(22)\mu\text{s}$ gas PEPICO³

 \tilde{X}^2B_2 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CD stretch	1948(4)	gas	PE	1,4,5
	2	CO stretch	1657(4)	gas	PE	4,5
	3	CD ₂ scissors	920(4)	gas	PE	1,4,5
			648(4)	gas	PE	4,5

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InH₂Cl
 \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	InH ₂ s-stretch	1804.0	Ar	IR	1
	2	InH ₂ scissors	575.8	Ar	IR	1
	3	InCl stretch	343.4	Ar	IR	1
b_1	4	OPLA	541.4	Ar	IR	1
b_2	5	InH ₂ a-stretch	1820.3	Ar	IR	1
	6	HIInCl deform.	415.7	Ar	IR	1

InD₂Cl
 \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	InD ₂ s-stretch	1291.9	Ar	IR	1
	2	InD ₂ scissors	415.3	Ar	IR	1
	3	InCl stretch	339.3	Ar	IR	1
b_1	4	OPLA	389.1	Ar	IR	1
b_2	5	InD ₂ a-stretch	1310.9	Ar	IR	1
	6	DInCl deform.	298.2	Ar	IR	1

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H₂CS

$\tilde{E} \ 3p_z \ ^1B_2$ C_{2v}
 $T_0 = 55096$ gas AB^{8,19} $\tilde{E}-\tilde{X}$ 181.5 nm

$\tilde{D} \ 3p_y \ ^1A_1$ C_{2v}
 $T_0 = 53134$ gas AB^{8,19} $\tilde{D}-\tilde{X}$ 188.2 nm

$\tilde{C} \ 3s \ ^1B_2$ C_{2v} Structure: AB¹⁹
 $T_0 = 47110.821(9)$ gas AB^{1,8,15,19} $\tilde{C}-\tilde{X}$ 212.1 nm
 $A_0 = 8.557$; $B_0 = 0.603$; $C_0 = 0.562$ AB¹⁹

$\tilde{B} \ ^1A_1^a$ C_{2v}
 $T_0 = 45197$ gas AB^{8,18} $\tilde{B}-\tilde{X}$ 185–215 nm
 All but the first absorption band show evidence for predissociation.¹⁸

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	3	CS stretch	476	gas	AB	18
<i>b</i> ₁	4	OPLA	363H	gas	AB	18

$\tilde{A} \ ^1A_2$ C_{2v} Structure: AB^{10,21}LF²⁸
 $T_0 = 16394.628(4)$ gas AB^{7,9,10}LF^{24,28,32,33}DR³⁴ $\tilde{A}-\tilde{X}$ 440–610 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CH stretch	3033.4 ^b	gas	AB,LF	9,28
	2	CH ₂ scissors	1334.5 ^b	gas	LF	28
	3	CS stretch	819.7	gas	AB,LF	7,9,28
<i>b</i> ₁	4	OPLA	370.3	gas	AB,LF	9,28,32
<i>b</i> ₂	5	CH stretch	3054.9 ^b	gas	LF	28
	6	CH ₂ rock	785.2	gas	LF	28

$\tau_0 = 140(3) \mu\text{s}$ gas LF^{22,25}
 $A_0 = 9.434$; $B_0 = 0.538$; $C_0 = 0.509$ AB¹⁰LF^{24,28}

$\tilde{a} \ ^3A_2$ C_{2v} Structure: AB^{11,21}LF²⁹
 $T_0 = 14507.38$ gas AB^{7,11}LF^{20,24,29}CL^{26,30}DR³⁴ $\tilde{a}-\tilde{X}$ 610–800 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	2	CH ₂ scissors	1318.6	gas	AB,LF	11,29
	3	CS stretch	861.6	gas	AB,LF	11,23
	4	OPLA	326	gas	LF,CL	20,21,26,30,32
<i>b</i> ₂	6	CH ₂ rock	762.3	gas	LF	23

$\tau > 1.5 \text{ ms}$ gas LF^{22,25}
 $A_0 = 9.383$; $B_0 = 0.552$; $C_0 = 0.521$ AB¹¹LF²⁴

$\tilde{X} \ ^1A_1$		C_{2v}	Structure: MW ^{2,4,5} IR ^{3,14}			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CH stretch	2971.03	gas	IR	3,14
			2970w	Ar	IR	6,17
			2973w	N ₂	IR	6
<i>a</i> ₂	2	CH ₂ scissors	1455.50	gas	LF,IR	13,31
			1447	Ar	IR	17
<i>a</i> ₃	3	CS stretch	1059.20	gas	LS,IR	12,14
			1063w	Ar	IR	6,17
			1062w	N ₂	IR	6
<i>b</i> ₁	4	OPLA	990.19	gas	LS,IR	12,14
			993s	Ar	IR	6,17
			995s	N ₂	IR	6
<i>b</i> ₂	5	CH stretch	3024.61	gas	IR	3,14
			6	CH ₂ rock	991.01	gas
			988m	Ar	IR	6,17

$A_0 = 9.729$; $B_0 = 0.590$; $C_0 = 0.555$ MW^{2,4,5}AB¹⁰IR³¹

D₂CS

$\tilde{C} \ 3s \ ^1B_2$ C_{2v}
 $T_0 = 47325.563(4)$ gas AB^{8,19} $\tilde{C}-\tilde{X}$ 211.2 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CD stretch	1783	gas	AB	8,19
			2	CD ₂ scissors	746	gas

$A_0 = 4.350$; $B_0 = 0.510$; $C_0 = 0.456$ AB¹⁹

$\tilde{B} \ ^1A_1^a$ C_{2v}
 $T_0 \cong 45200$ gas AB¹⁸ $\tilde{B}-\tilde{X}$ 185–215 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	3	CS stretch	467	gas	AB	18
			4	OPLA	263H	gas

$\tilde{A} \ ^1A_2$ C_{2v}
 $T_0 = 16483.502(8)$ gas AB^{7,9,10} $\tilde{A}-\tilde{X}$ 440–610 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.		
<i>a</i> ₁	1	CD stretch	2139(2)	gas	AB	9		
			2	CD ₂ scissors	1013(2)	gas	AB	9
			3	CS stretch	771.3(5)	gas	AB	7,9
<i>b</i> ₁	4	OPLA	275.33	gas	AB	9		
<i>b</i> ₂	5	CD stretch	2324.85	gas	AB	9		
			6	CD ₂ rock	599(2)	gas	AB	9

$\tau_0 = 182 \mu\text{s}$ gas LF²⁵
 $A_0 = 4.736$; $B_0 = 0.458$; $C_0 = 0.417$ AB¹⁰

$\tilde{a} \ ^3A_2$ C_{2v} Structure: AB²¹
 $T_0 = 14613.54$ gas AB^{7,11}CL^{26,30} $\tilde{a}-\tilde{X}$ 610–815 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	2	CD ₂ scissors	1012	gas	AB,CL	11,30
			3	CS stretch	798	gas
<i>b</i> ₁	4	OPLA	219.4(5.0)	gas	AB,CL	21,26,30
<i>b</i> ₂	6	CD ₂ rock	572.6(5.0)	gas	CL	30

$A_0 = 4.716$; $B_0 = 0.469$; $C_0 = 0.426$ AB¹¹

\tilde{X}^1A_1		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CD stretch	2158.5	gas	IR	14
			2155m	Ar	IR	17
	2	CD ₂ scissors	1171.8	gas	IR	14
1167m			N ₂	IR	17	
3	CS stretch	936.13	gas	IR,LS	14,16	
		941vw	Ar	IR	6,17	
		939wm	N ₂	IR	6	
		781.2	gas	IR	14	
b_1	4	OPLA	783m	Ar	IR	6,17
			784s	N ₂	IR	6
			757.4	gas	IR	14
b_2	6	CD ₂ rock				

$A_0 = 4.883$; $B_0 = 0.497$; $C_0 = 0.450$ MW²AB¹⁰LF²⁷

^aBarrier to inversion ≈ 50 .¹⁸

^bFrom fit to combination bands.

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H₂CSe

\tilde{A}^1A_2		C_{2v}			Structure: LF ⁷	
$T_0 \equiv 13555$	gas	LF ^{5,6} CL ⁸			$\tilde{A}-\tilde{X}$ 685–833 nm	
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	3	C=Se stretch	662	gas	LF	6
b_1	4	OPLA	355T	gas	LF	6

$A^a = 9.015$; $B^a = 0.377$; $C^a = 0.360$ LF⁷

\tilde{a}^3A_2		C_{2v}			Structure: LF ⁹	
$T_0 = 12162.514$	gas	AB ¹ CL ^{3,8} LF ^{5,6,9}			$\tilde{a}-\tilde{X}$ 658–865 nm	
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CH ₂ scissors	1312	gas	LF	5,6
	3	C=Se stretch	704	gas	AB,CL,LF	1,3,5,6,8
b_1	4	OPLA	297 ^b	gas	AB,LF	1,5,6
b_2	6	HCS _e bend	812H	gas	LF	5,6

$A_0 = 9.320$; $B_0 = 0.391$; $C_0 = 0.375$ LF⁹

\tilde{X}^1A_1		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CH ₂ scissors	1460(30)	gas	CL	8
			860(10)	gas	CL	8
b_1	4	OPLA	906(10)	gas	CL	8
b_2	6	HCS _e bend	914(20)	gas	CL	8

$A_0 = 9.690$; $B_0 = 0.414$; $C_0 = 0.396$ MW^{2,4}LF⁷

D₂CSe

\tilde{A}^1A_2		C_s			Structure: LF ⁶	
$T_0 = 13631.4$	gas	LF ⁶			$\tilde{A}-\tilde{X}$ 671–734 nm	
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	3	C=Se stretch	619	gas	LF	6

$A_0 = 4.711$; $B_0 = 0.316$; $C_0 = 0.296$ LF⁷

\tilde{a}^3A_2		C_{2v}			Structure: LF ^{6,9,10}	
$T_0 = 12258.060$	gas	LF ^{6,9,10}			$\tilde{a}-\tilde{X}$ 661–815 nm	
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CD ₂ scissors	996	gas	LF	6
			667	gas	LF	6
b_1	4	OPLA	208 ^b	gas	LF	6
b_2	6	DCSe bend	563H	gas	LF	6

$A_0 = 4.688$; $B_0 = 0.327$; $C_0 = 0.306$ LF⁹

\tilde{X}^1A_1		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	3	C=Se stretch	789T	gas	LF	6

$A_0=4.865$; $B_0=0.344$; $C_0=0.321$ MW⁴LF^{7,9}

^a $v_4=1$.

^bFrom fit to double minimum potential. Barrier to inversion 13.1 for H₂CSe, 16.2 for D₂CSe.⁶

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t -N₂H₂

$\tilde{C}^1B_u^a$		C_{2h}				
$T_0=67894$	gas	AB ⁹		$\tilde{C}-\tilde{X}$	135–147 nm	

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_g	2	Bend	1180	gas	AB	9
	3	NN stretch	1849	gas	AB	9

\tilde{B}^1B_u		C_{2h}				
$T_0=57926.5$	gas	AB ^{2,9}		$\tilde{B}-\tilde{X}$	150–175 nm	

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_g	1	NH stretch	3092	gas	AB	9
	2	Bend	1180	gas	AB	2,9
	3	NN stretch	1875	gas	AB	2,9

$A_0=15.63$; $B_0=1.32$; $C_0=1.22$ AB⁹

\tilde{A}^1B_g		C_{2h}				
$T^c=23896$	gas	AB ^{4,5,8}		$\tilde{A}-\tilde{X}$	300–440 nm	

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_g	2	Bend	1215(15)	gas	AB	5
	3	NN stretch	1550(20)	gas	AB	5

\tilde{X}^1A_g		C_{2h}		Structure: IR ^{2,6,13}		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.

a_g	1	NH stretch	3128	N ₂	Ra	3
	2	NH bend	1583	N ₂	Ra	3
	3	N=N stretch	1529	N ₂	Ra	3
a_u	4	Torsion	1288.65	gas	IR	10,11,13
			1283	Ar	IR	7
			1286	N ₂	IR,Ra	1,3,7
b_u	5	NH stretch	3120.29	gas	IR	2,6,10,11
			3118	Ar	IR	7
			3137	N ₂	IR	7
	6	NH bend	1316.41	gas	IR	10,11,13
			1313	Ar	IR	7
			1321	N ₂	IR	3,7

$A_0=10.001$; $B_0=1.304$; $C_0=1.150$ IR^{6,10,11}

t -N₂D₂

\tilde{B}^1B_u		C_{2h}				
$T_0\equiv 58086^d$	gas	AB ²		$\tilde{B}-\tilde{X}$	159–172 nm	

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_g	2	Bend	950	gas	AB	2

\tilde{A}^1B_g		C_{2h}				
gas	AB ⁵			$\tilde{A}-\tilde{X}$	320–430 nm	

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_g	2	Bend	910(10)	gas	AB	5
	3	NN stretch	1440(20)	gas	AB	5

\tilde{X}^1A_g		C_{2h}				
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Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_g	2	ND bend	1215	N ₂	Ra	3
	3	N=N stretch	1539	N ₂	Ra	3
a_u	4	Torsion	948.30	gas	IR	12,13
			946	N ₂	IR	1,3,7
b_u	5	ND stretch	2315.05	gas	IR	6,12
			2308	N ₂	IR	7
	6	ND bend	968.70	gas	IR	12,13
			972	N ₂	IR	7

$A_0=6.025$; $B_0=1.089$; $C_0=0.920$ IR^{6,12}

^a4pπ Rydberg transition.

^b3pπ Rydberg transition.

^c₀¹ vibronic band origin.⁸

^d1–0 subband origin.

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H₂CF

5p Rydberg state C_{2v}
T₀=67265(10) gas MPI⁸

4p Rydberg state C_{2v}
T₀=63275(10) gas MPI⁸ 4p- \tilde{X} 147–158 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	CF stretch	1580(20)	gas	MPI	8
	3	CH ₂ scissors	1443(20)	gas	MPI	8
b ₁	4	OPLA	1259(20)	gas	MPI	8

3p Rydberg state C_{2v}
T₀=52863(10) gas MPI⁸ 3p- \tilde{X} 167–193 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	CF stretch	1575(20)	gas	MPI	8
	3	CH ₂ scissors	1420(20)	gas	MPI	8
b ₁	4	OPLA	1223(20)	gas	MPI	8

Threshold for photodecomposition, producing CF, observed⁵ near 280 nm in an argon matrix.

\tilde{X}^2B_1 C_{2v} Structure: ESR¹MW⁶

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	3	CF stretch	1170.42	gas	DL	7
			1163m	Ar	IR	2,3,5
b ₁	4	OPLA	300(30)	gas	MW	6
			260(30)	gas	MPI	8

A₀=8.846; B₀=1.032; C₀=0.925 LMR^{4,9}MW⁶

D₂CF

5p Rydberg state C_{2v}
T₀=67186(10) gas MPI⁸

4p Rydberg state C_{2v}
T₀=63195(10) gas MPI⁸ 4p- \tilde{X} 154–159 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CD ₂ s-stretch	2190(20)	gas	MPI	8
	2	CF stretch	1513(20)	gas	MPI	8
	3	CD ₂ scissors	1076(20)	gas	MPI	8
b ₁	4	OPLA	1004(20)	gas	MPI	8

3p Rydberg state C_{2v}
T₀=52786(10) gas MPI⁸ 3p- \tilde{X} 167–193 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CD ₂ s-stretch	2176(20)	gas	MPI	8
	2	CF stretch	1504(20)	gas	MPI	8
	3	CD ₂ scissors	1080(20)	gas	MPI	8
b ₁	4	OPLA	976(10)	gas	MPI	8

\tilde{X}^2B_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	CF stretch	1191m	Ar	IR	2,3,5
	3	CD ₂ scissors	1013w	Ar	IR	5
b ₁	4	OPLA	170(30)	gas	MPI	8

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H₂CCI

\tilde{X} C_{2v} Structure: ESR³MW⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	CH ₂ scissors	1391wm	Ar	IR	2,6
			827s	Ar	IR	1,2,6
			827	Kr	IR	6
b ₁	4	OPLA	829	Xe	IR	6
			402s	Ar	IR	1,2,6
			395	Kr	IR	6
			389	Xe	IR	6

A₀=9.152(3); B₀=0.532; C₀=0.502 MW⁴LMR⁵

D₂CCI

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	CD ₂ scissors	1045m	Ar	IR	1,2
			788m	Ar	IR	1,2,6
			791	Kr	IR	6
b ₁	4	OPLA	793	Xe	IR	6
			291m	Ar	IR	1,2

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H₂NS

\tilde{X}^2B_1		C _{2v}		Structure: MW ¹		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b ₁	4	OPLA	325T	gas	MW	1

A₀=11.123(5); B₀=0.538; C₀=0.513 MW¹

D₂NS

\tilde{X}^2B_1		C _{2v}		Structure: MW ¹		
A ₀ =5.605; B ₀ =0.466; C ₀ =0.430		MW ¹				

Reference

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HOOH⁺

\tilde{C}		C ₂		Structure: PE ¹⁻³		
T ₀ =49000(500)		gas				

\tilde{B}^2A		C ₂		Structure: PE ¹⁻³		
T ₀ =32430(220)		gas				

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a	2	OO stretch	1100(50)	gas	PE	3

\tilde{A}^2A		C ₂		Structure: PE ¹⁻³		
T ^a =16070(500)		gas				

\tilde{X}^2B		C ₂		Structure: PE,PI		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a		Deformation	1080(50)	gas	PE,PI	2-4

^aFrom vertical ionization potential. The first ionization potential of HOOH is taken to equal 10.63 eV, from the photoionization study of Ref. 4.

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HOOH

Continuous absorption, 120–300 nm.^{1,2,7,12}

\tilde{X}		C ₂		Structure: IR ^{4,19,21} MW ^{8,21}			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a	1	OH s-stretch	3617.95 ^a	gas	IR,Ra	3,4,10,17	
			3609.8	Ar	Ra	11	
	2	OH s-bend	1393.5	gas	Ra	10	
			1385	Ar	Ra	11	
	3	OO stretch	877.93 ^a	gas	IR	20,	
			865.94	gas	Ra, IR	10,20	
	4	Torsion	869	Ar	Ra	11	
			370.89 ^a	gas	IR	6,17,18,19	
	b	5	OH a-stretch	254.55	Ar	IR	9
				372br	Ar	IR	9
5		OH a-stretch	264	N ₂	IR	5,9	
			378vs,br	gas	IR	3,4,17	
5		OH a-stretch	3618.84 ^a	gas	IR	3,4,17	
			3610.66	Ar	IR	9,23	
5		OH a-stretch	3597.0 ^b	Ar	IR	9,23	
			3587.8 ^b	Kr	IR	23	
5		OH a-stretch	3583.6 ^b	Kr	IR	23	
			3574.0 ^b	Xe	IR	23	
5	OH a-stretch	3568.0 ^b	Xe	IR	23		
		3560.0 ^b	N ₂	IR	5,9		
6	OH a-bend	3587s	N ₂	IR	5,9		
		3582s	gas	IR,DL	3,16,24		
6	OH a-bend	1273.68 ^a	gas	IR,DL	3,16,24		
		1264.58	Ar	IR	9,23		
6	OH a-bend	1277.0ms ^b	Ar	IR	9,23		
		1270.9vs ^b	Kr	IR	23		
6	OH a-bend	1273.7 ^b	Kr	IR	23		
		1268.7 ^b	Xe	IR	23		
6	OH a-bend	1270.3 ^b	Xe	IR	23		
		1265.7 ^b	N ₂	IR	5,9		
6	OH a-bend	1294vs	N ₂	IR	5,9		

A₀=10.069; B₀=0.874; C₀=0.838 IR^{4,17,20}MW^{8,13-15,22}

DOOD

\tilde{X}		C ₂		Structure: IR ^{4,17,20} MW ^{8,13-15,22}			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a	1	OD s-stretch	2668	gas	Ra	10	
			2653.5	Ar	Ra	11	
	2	OD s-bend	1029	gas	Ra	10	
			1021.5	Ar	Ra	11	
	3	OO stretch	867	gas	Ra	10	
			871	Ar	Ra	11	
	4	Torsion	251	Ar	IR	9	
			286br	N ₂	IR	9	
	b	5	OD a-stretch	2661m	gas	IR	3
				2655.7	Ar	IR	9,23
5		OD a-stretch	2651.3	Ar	IR	9,23	
			2645.3	Kr	IR	23	
5		OD a-stretch	2645.2	Kr	IR	23	
			2639	Xe	IR	23	
5		OD a-stretch	2633.2	Ar	IR	9,23	
			2628.4	Ar	IR	9,23	
6		OD a-bend	2646	N ₂	IR	9	
			947s	gas	IR	3	
6	OD a-bend	951.3vs	Ar	IR	9,23		
		949.9	Kr	IR	23		
6	OD a-bend	947.8	Xe	IR	23		
		966vs	N ₂	IR	9		

^aTransitions to two lowest torsional levels associated with this fundamental are given.

^bRef. 23 attributes the doubling in the matrix absorptions to excitation from the two lowest torsional levels.

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HSSH

\tilde{X}		C ₂		Structure: MW,IR ^{13,14}		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a	1	SH s-stretch	2555.78	gas	IR	3,5
			2553.8	Ar	IR	18
	2	Deformation	883	gas	IR	3
			880.3	Ar	IR	18
3	SS stretch	515.92	gas	IR,MW	3,15	
		417.48	gas	IR	2,3,11,15	
b	5	SH a-stretch	2558.63	gas	IR	1,3,5,16
			2556.6	Ar	IR	18
6	Deformation	882.0	gas	IR	1,3	

$A_0=4.899$; $B_0=0.233$; $C_0=0.232$ MW^{4,8,9,12,17}IR^{10,12}

DSSD

\tilde{X}		C ₂				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a	4	Torsion	306	gas	IR	3
b	5	SD a-stretch	1863	gas	IR	3
		Deformation	646.4	gas	IR	3

$A_0=2.550$; $B_0=0.218$; $C_0=0.218$ MW^{4,6,7,13}IR¹³

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HFFH⁺

\tilde{X}		C _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _u	5	HF a-stretch	3064.7	Ne	IR	1

DFFD⁺

\tilde{X}		C _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _u	5	DF a-stretch	2287.6	Ne	IR	1

Reference

- C. L. Lugez, M. E. Jacox, and R. D. Johnson III, *J. Chem. Phys.* **110**, 5037 (1999).

HCICIH⁺

\tilde{X} C _{2h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _u	5	HCl a-stretch	2704.1	Ne	IR	1

DCICID⁺

\tilde{X} C _{2h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _u	5	DCl a-stretch	1962.2	Ne	IR	1

Reference

¹D. Forney, M. E. Jacox, and W. E. Thompson, J. Chem. Phys. **103**, 1755 (1995).

HBrBrH⁺

\tilde{X} C _{2h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _u	5	HBr a-stretch	2440.3	Ne	IR	1

DBrBrD⁺

\tilde{X} C _{2h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _u	5	DBr a-stretch	1756.4	Ne	IR	1

Reference

¹C. L. Lugez, M. E. Jacox, and W. E. Thompson, J. Chem. Phys. **105**, 3901 (1996).

HIH⁺

\tilde{X} C _{2h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _u	5	HI a-stretch	2143.0	Ne	IR	1

DIID⁺

\tilde{X} C _{2h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _u	5	DI a-stretch	1537.0	Ne	IR	1

Reference

¹C. L. Lugez, M. E. Jacox, and W. E. Thompson, J. Chem. Phys. **105**, 3901 (1996).

HXeOH

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	XeH stretch	1577.6	Xe	IR	1

DXeOD

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	XeD stretch	1141.2	Xe	IR	1

Reference

¹M. Pettersson, L. Khriachtchev, J. Lundell, and M. Räsänen, J. Am. Chem. Soc. **121**, 11904 (1999).

HXeSH

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		XeH stretch	1118.6	Xe	IR	1,2

DXeSD

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		XeD stretch	832.8	Xe	IR	2

References

¹L. Khriachtchev, M. Pettersson, E. Isoniemi, and M. Räsänen, J. Chem. Phys. **108**, 5747 (1998).

²M. Pettersson, J. Lundell, L. Khriachtchev, E. Isoniemi, and M. Räsänen, J. Am. Chem. Soc. **120**, 7979 (1998).

8.6. Four-Atomic Monohydrides

LiCCH

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
II	5	Deformation	136T	gas	MW	2

$B_0 = 0.352$ MW^{1,2}

LiCCD

\tilde{X}
 $B_0 = 0.321$ MW^{1,2}

References

- ¹D. B. Grotjahn, A. J. Apponi, M. A. Brewster, J. Xin, and L. M. Ziurys, *Angew. Chem.* **110**, 2824 (1998); *Angew. Chem. Int. Ed. Engl.* **37**, 2678 (1998).
²A. J. Apponi, M. A. Brewster, and L. M. Ziurys, *Chem. Phys. Lett.* **298**, 161 (1998).

NaCCH

$\tilde{X}^1\Sigma$		$C_{\infty v}$		Structure: MW ¹		
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
II	5	Deformation	102T	gas	MW	1

$B_0 = 0.150$ MW¹

NaCCD

$\tilde{X}^1\Sigma$
 $B_0 = 0.139$ MW¹

Reference

- ¹M. A. Brewster, A. J. Apponi, J. Xin, and L. M. Ziurys, *Chem. Phys. Lett.* **310**, 411 (1999).

KCCH

\tilde{X}
 $B_0 = 0.099$ MW^{1,2} Structure: MW^{1,2}

KCCD

\tilde{X}
 $B_0 = 0.092$ MW^{1,2}

References

- ¹J. Xin and L. M. Ziurys, *Astrophys. J.* **501**, L151 (1998).
²D. B. Grotjahn, A. J. Apponi, M. A. Brewster, J. Xin, and L. M. Ziurys, *Angew. Chem.* **110**, 2824 (1998); *Angew. Chem. Int. Ed. Engl.* **37**, 2678 (1998).

YbCCH⁺

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
II	5	YbCC bend	97(5)	gas	MPI	1

Reference

- ¹H.-P. Loock, A. Bérces, B. Simard, and C. Linton, *J. Chem. Phys.* **107**, 2720 (1997).

MgCCH

$\tilde{A}^2\Pi$ $C_{\infty v}$
 $T_0 = 22807$ gas LF^{1,3} $\tilde{A}-\tilde{X}$ 415–440 nm

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ^+		MgC stretch	552	gas	LF	1

$A = 36.4$ gas LF¹
 $B_0 = 0.175$ gas LF¹

$\tilde{X}^2\Sigma^+$		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ^+	2	CC stretch	1984.2	Ar	IR	2
	3	MgC stretch	496(2)	gas	LF	3
			491.8	Ar	IR	2
II	4	HCC bend	660.8	Ar	IR	2
	5	MgCC bend	143(2)	gas	LF	3

$B_0 = 0.166$ MW⁴

MgCCD

$\tilde{X}^2\Sigma^+$		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ^+	2	CC stretch	1856.4	Ar	IR	2
	3	MgC stretch	485.1	Ar	IR	2

References

- ¹G. K. Corlett, A. M. Little, and A. M. Ellis, *Chem. Phys. Lett.* **249**, 53 (1996).
²C. A. Thompson and L. Andrews, *J. Am. Chem. Soc.* **118**, 10242 (1996).
³G. K. Corlett, M. S. Beardah, and A. M. Ellis, *J. Mol. Spectrosc.* **185**, 202 (1997).
⁴M. A. Brewster, A. J. Apponi, J. Xin, and L. M. Ziurys, *Chem. Phys. Lett.* **310**, 411 (1999).

CaCCH

$\tilde{C}^2\Delta$ $C_{\infty v}$
 $T_0 = 21900\text{T}$ gas LF⁹ $\tilde{C}-\tilde{X}$ 440–452 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		HCC deform.	573(5)	gas	LF	9
		CaC stretch	394(5)	gas	LF	9
		CaCC deform.	99(5)	gas	LF	9

$\tilde{B}^2\Sigma^+$ $C_{\infty v}$
 $T_0 = 16600\text{T}$ gas LF⁵

$\tilde{A}^2\Pi$ $C_{\infty v}$
 $T_0 = 15525.393$ gas LF^{1-3,5-8,10} $\tilde{A}-\tilde{X}$ 640–665 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	5	CaCC bend	101.39	gas	LF	6

$A = 70.821$ gas LF^{1-3,5}
 $\epsilon_4\omega_4 = 1.85(4)$ gas LF¹⁰
 $\epsilon_5\omega_5 = 3.528(14)$ gas LF⁶
 $B_0 = 0.115$ LF^{2,3,5}

$\tilde{X}^2\Sigma^+$ $C_{\infty v}$ Structure: LF^{2,5}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	3	CaC stretch	397.40	gas	LF	1,8
Π	5	CaCC bend	102.94	gas	LF	6

$B_0 = 0.113$ LF^{2,3,5}MW⁴

References

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- M. Elhanine, R. Lawruszczuk, and B. Soep, Chem. Phys. Lett. **288**, 785 (1998).
- M. Li and J. A. Coxon, J. Mol. Spectrosc. **196**, 14 (1999).

SrCCH

$\tilde{B}'^2\Delta$ $C_{\infty v}$
 $T_0 \leq 21129.6$ gas LF³ $\tilde{B}'-\tilde{X}$ 457–475 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	3	SrC stretch	339	gas	LF	3

$A = 106$ gas LF³

$\tilde{A}^2\Pi$ $C_{\infty v}$
 $T_0 = 14176(10)$ gas LF¹ $\tilde{A}-\tilde{X}$ 685–725 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	3	SrC stretch	354(10)	gas	LF	1

$A = 275(10)$ gas LF¹

$\tilde{X}^2\Sigma^+$ $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	3	SrC stretch	343(10)	gas	LF	1
Π	5	SrCC bend	70(5)H	gas	LF	1

$B_0 = 0.083$ MW²

SrCCD

$\tilde{B}'^2\Delta$ $C_{\infty v}$
 $T_0 \leq 21103.6$ gas LF³ $\tilde{B}'-\tilde{X}$ 457–474 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	3	SrC stretch	334	gas	LF	3

$A = 104$ gas LF³

References

- A. M. R. P. Bopegedera, C. R. Brazier, and P. F. Bernath, Chem. Phys. Lett. **136**, 97 (1987).
- B. P. Nuccio, A. J. Apponi, and L. M. Ziurys, Chem. Phys. Lett. **247**, 283 (1995).
- G. M. Gretham and A. M. Ellis, J. Mol. Spectrosc. **206**, 198 (2001).

PdCCH

\tilde{A}
 $T_0 = 2820(1210)$ gas PE¹

\tilde{X} $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	2	CC stretch	1720(170)	gas	PE	1

Reference

- V. D. Moravec and C. C. Jarrold, J. Chem. Phys. **112**, 792 (2000).

YbCCH

$\tilde{A}^2\Pi_{3/2}$ $C_{\infty v}$
 $T_0 = 18250(5)$ gas LF,MPI¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	5	YbCC bend	99(5)	gas	MPI	1

$\tilde{A}^2\Pi_{1/2}$ $C_{\infty v}$
 $T_0 = 16850(5)$ gas LF,MPI¹ $\tilde{A}-\tilde{X}$ 587–594 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	5	YbCC bend	103(5)	gas	MPI	1

$\tilde{X}^2\Sigma^+$ $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	3	YbC stretch	328(20)	gas	LF	1
Π	5	YbCC bend	96(10)	gas	MPI	1

YbCCD

$\tilde{A}^2\Pi_{3/2}$ $C_{\infty v}$
 $T_0 = 18250(5)$ gas LF,MPI¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	5	YbCC bend	93(5)	gas	MPI	1

$\tilde{A}^2\Pi_{1/2}$ $C_{\infty v}$
 $T_0 = 16850(5)$ gas LF,MPI¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	5	YbCC bend	99(5)	gas	MPI	1

$\tilde{X}^2\Sigma^+$ $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	5	YbCC bend	94(5)	gas	LF,MPI	1

Reference

¹H.-P. Loock, A. Bérces, B. Simard, and C. Linton, *J. Chem. Phys.* **107**, 2720 (1997).

PdCCH⁻

Threshold for electron detachment from ground-state PdCCH⁻ = 15980(240) gas PE¹

Reference

¹V. D. Moravec and C. C. Jarrold, *J. Chem. Phys.* **112**, 792 (2000).

HBeCN

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CN stretch	2202.8	Ar	IR	1
	2	BeH stretch	2128.6	Ar	IR	1
	3	BeC stretch	800.1	Ar	IR	1
	4	H deformation	530.1	Ar	IR	1

DBeCN

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CN stretch	2194.2	Ar	IR	1
	2	BeD stretch	1642.1	Ar	IR	1
	3	BeC stretch	750.9	Ar	IR	1
	4	H deformation	436.1	Ar	IR	1
			431.5			

Reference

¹D. V. Lanzisera and L. Andrews, *J. Am. Chem. Soc.* **119**, 6392 (1997).

HBeNC

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	BeH stretch	2153.2	Ar	IR	1
	2	NC stretch	2085.7	Ar	IR	1
	3	BeN stretch	912.4	Ar	IR	1
	4	H deformation	525.4	Ar	IR	1
			521.7			

DBeNC

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NC stretch	2098.6	Ar	IR	1
	2	BeD stretch	1631.3	Ar	IR	1
	3	BeN stretch	895.6	Ar	IR	1
	4	D deformation	422.1	Ar	IR	1
			418.0			

Reference

¹D. V. Lanzisera and L. Andrews, *J. Am. Chem. Soc.* **119**, 6392 (1997).

HBCC

$\tilde{X}^1\Sigma^+$		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	BH stretch	2763.2	Ar	IR	2
	2	C≡C stretch	1995.3	Ar	IR	1,2

DBCC

$\tilde{X}^1\Sigma^+$		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	BD stretch	2173.3	Ar	IR	1
	2	C≡C stretch	1907.7	Ar	IR	1,2

References

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²L. Andrews, D. V. Lanzisera, P. Hassanzadeh, and Y. Hannachi, J. Phys. Chem. A **102**, 3259 (1998).

HBCN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	BH stretch	2664.8	Ar	IR	1
	2	CN stretch	2108.0	Ar	IR	1
	3	BC stretch	938.4	Ar	IR	1

DBCN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CN stretch	1971.2	Ar	IR	1
	3	BC stretch	908.6	Ar	IR	1

Reference

¹D. V. Lanzisera, L. Andrews, and P. R. Taylor, J. Phys. Chem. A **101**, 7134 (1997).

HBNC

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	2	NC stretch	1952.2	Ar	IR	1
	3	BN stretch	1046.4	Ar	IR	1

DBNC

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	3	BN stretch	1024.7	Ar	IR	1

Reference

¹D. V. Lanzisera, L. Andrews, and P. R. Taylor, J. Phys. Chem. A **101**, 7134 (1997).

cyc-HB(CN)

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	BH stretch	2708.6	Ar	IR	1
			1081.0	Ar	IR	1
			736.8	Ar	IR	1

cyc-DB(CN)

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	BD stretch	2037.9	Ar	IR	1
			1042.7	Ar	IR	1
			737.4T	Ar	IR	1

Reference

¹D. V. Lanzisera, L. Andrews, and P. R. Taylor, J. Phys. Chem. A **101**, 7134 (1997).

HC₃

\tilde{C}^2A''	C_s					
$T_0=23171$	gas	MPI ⁸				$\tilde{C}-\tilde{X}$ 424–432 nm
\tilde{B}^2A''	C_s					
$T_0=20538$	gas	MPI ⁸				$\tilde{B}-\tilde{X}$ 454–487 nm
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	4	CCH bend	783.5	gas	MPI	8
	5	C ₃ bend	493	gas	MPI	8

\tilde{A}^2A' C_s
 $T_0 = 19187$ gas MPI⁸ $\tilde{A}-\tilde{X}$ 445–521 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CH stretch	2800	gas	MPI	8
	2		1836	gas	MPI	8
	3		1091	gas	MPI	8
	4	CCH bend	881.3	gas	MPI	8
	5	C ₃ bend	460	gas	MPI	8

$A \approx 19.2$ MPI⁸

$\tilde{X}^2\Pi_{1/2}$ $C_{\infty v}$ Structure: MW⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CH stretch	3238.0w	Ar	IR	5,6
	2	C ₃ a-stretch	1832.6m 1824.7s	Ar	IR	1,5,6
	3	C ₃ s-stretch	1167br 1159.8w	Ar	IR	5,6
Π	4	HCC bend	28 ^a	gas	MW	4

$A = 14.44$ gas MW²⁻⁴
 $B_0 = 0.373$ MW²⁻⁴

DC₃

$\tilde{X}^2\Pi_{1/2}$ $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CH stretch	2424.0wmT	Ar	IR	5,6
	2	C ₃ a-stretch	1778.8sh 1771.5m	Ar	IR	1,5,6
	3	C ₃ s-stretch	1148br 1140.4w	Ar	IR	5,6

$A = 12.53$ gas MW⁴
 $B_0 = 0.337$ MW⁴

^{a2} Σ^u component.

References

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HCCSi

$\tilde{X}^2\Pi$ $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	2	CC stretch	1989.8	Ar	IR	1
	3	SiC stretch	636.0T	Ar	IR	1

$A_{\text{eff}} = 72.4$ MW³
 $B_0 = 0.181$ MW^{2,3}

DCCSi

$\tilde{X}^2\Pi$ $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	2	CC stretch	1874.9	Ar	IR	1
	3	SiC stretch	622.4T	Ar	IR	1

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- ³M. C. McCarthy, A. J. Apponi, C. A. Gottlieb, and P. Thaddeus, J. Chem. Phys. **115**, 870 (2001).

cyc-HSi₃

\tilde{X}^2B_2 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	3		398(20)	gas	PE	1

Reference

- ¹C. Xu, T. R. Taylor, G. R. Burton, and D. M. Neumark, J. Chem. Phys. **108**, 7645 (1998).

cyc-HSi₃⁻

Threshold for electron detachment from ground-state cyc-HSi₃⁻ = 20410(80) gas PE¹

Reference

- ¹C. Xu, T. R. Taylor, G. R. Burton, and D. M. Neumark, J. Chem. Phys. **108**, 7645 (1998).

HCCN

 ${}^3\Sigma^-?$

In the gas phase, a prominent absorption band system beginning at 340 nm has been attributed^{2,3} to HCCN, but has not been analyzed. An absorption band system assigned to HCCN was observed in an argon matrix between 240 and 340 nm, with band separations of approximately 1050.^{4,11} 254-nm irradiation of HCCN isolated in an argon matrix results in photoisomerization to HCNC.¹¹

$\tilde{X}^3\Sigma^-$		$C_{\infty v}^a$	Structure: ESR ¹ MW ⁵⁻⁷			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CH stretch	3246.66	gas	CC	8
			3229.4s	Ar	IR	4,11
			3213.3	N ₂	IR	11
2	CCN a-stretch	1734.9s	Ar	IR	4,11	
		1750.3	N ₂	IR	11	
		1178.6wm	Ar	IR	4,11	
3	CCN s-stretch	1189.2	N ₂	IR	11	
II	4	CCN deform.	383(20)	gas	CC,MW	8,9
			5	H deform.	128.91	gas

 $B_0 = 0.366$ MW^{5,7,9}LMR¹⁴

DCCN

$\tilde{X}^3\Sigma^-$		$C_{\infty v}^a$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CD stretch	2436.37	gas	LD	10
			2424.0ms	Ar	IR	4,11
			2432.3	N ₂	IR	11
2	CCN a-stretch	1729.0s	Ar	IR	4,11	
		1739.5	N ₂	IR	11	
		1149.2wm	Ar	IR	11	
3	CCN s-stretch	1148.5	N ₂	IR	11	
II	4	CCN deform.	367(15)	gas	MW	9
			5	D deform.	74.84	gas

 $B_0 = 0.330$ MW^{6,9}LMR¹³

^aQuasi-linear,^{6,7} with barrier to linearity of 256 for HCCN^{9,12,15} and of 229 for DCCN.^{10,15}

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¹³W. E. Jones, F. Sun, R. F. Curl, M. D. Allen, K. M. Evenson, and J. M. Brown, *Can. J. Phys.* **79**, 389 (2001).

¹⁴M. D. Allen, K. M. Evenson, and J. M. Brown, *J. Mol. Spectrosc.* **209**, 143 (2001).

¹⁵P. Y. Hung, F. Sun, N. T. Hunt, L. A. Burns, and R. F. Curl, *J. Chem. Phys.* **115**, 9331 (2001).

HCNC

In an argon matrix, absorption occurs between 200 and 240 nm, with maximum near 45900 (218 nm).¹

$T_0 = 32150$ Ar AB ¹		270–311 nm				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		NC stretch	1950T	Ar	AB	1
			400T	Ar	AB	1

In an argon matrix, weak absorption is observed between 600 and 1100 nm, with maximum near 11700 (855 nm) and a band spacing of approximately 520. Irradiation at wavelengths longer than 700 nm leads to isomerization to *cyc*-HC=NC.¹

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CH stretch	2834.5wm	Ar	IR	1
			2852.8	N ₂	IR	1
	2	N=C stretch	1859.5vs	Ar	IR	1
			1874.2	N ₂	IR	1
3	C–N stretch	1173.5m	Ar	IR	1	
		1182.9	N ₂	IR	1	
4	HCN bend	1080.5w	Ar	IR	1	
		1090.9	N ₂	IR	1	

DCNC

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CD stretch	2110.9wm	Ar	IR	1
			2124.8	N ₂	IR	1
	2	N=C stretch	1856.4vs	Ar	IR	1
			1871.2	N ₂	IR	1
3	C–N stretch	1082.9w	Ar	IR	1	
		1091.6	N ₂	IR	1	
4	DCN bend	904.5wm	Ar	IR	1	
		911.2	N ₂	IR	1	

Reference

- G. Maier, H. P. Reisenauer, and K. Rademacher, *Chem. Eur. J.* **4**, 1957 (1998).

cyc-HC=NC:

In an argon matrix, unstructured absorption occurs between 240 and 300 nm, with a maximum near 37900 (264 nm).¹

In an argon matrix, irradiation at wavelengths between ~350 and 450 nm leads to isomerization to HCCN.¹

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CH stretch	3140.2wm	Ar	IR	1
			3154.4	N ₂	IR	1
	2	C=N stretch	1571.0wm	Ar	IR	1
			1573.3	N ₂	IR	1
	3			1290.5s	Ar	IR
1289.6				N ₂	IR	1
4			1009.2m	Ar	IR	1
			1018.2	N ₂	IR	1
5			828.6vs	Ar	IR	1
			843.1	N ₂	IR	1
a''	6	H-CC OPLA	895.7w	Ar	IR	1
			903.2	N ₂	IR	1

cyc-DC=NC:

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CD stretch	2365.8wm	Ar	IR	1
			2376.3	N ₂	IR	1
	2	C=N stretch	1532.1wm	Ar	IR	1
			1533.3	N ₂	IR	1
	3			1268.5vs	Ar	IR
1268.9				N ₂	IR	1
4			957.9s	Ar	IR	1
			964.5	N ₂	IR	1
5	DCN deform.		668.2s	Ar	IR	1
			677.5	N ₂	IR	1
a''	6	D-CC OPLA	829.3wmT	Ar	IR	1
			843.1T	N ₂	IR	1

Reference

¹G. Maier, H. P. Reisenauer, and K. Rademacher, Chem. Eur. J. **4**, 1957 (1998).

HCNSi

In an argon matrix,¹ irradiation at wavelengths longer than 700 nm results in photoisomerization to cyc-(HCSiN).

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CH stretch	2753.5wm	Ar	IR	1
	2	CN stretch	1538.1wm	Ar	IR	1
	3	CH deform.	977.0vs	Ar	IR	1
	4	SiN stretch	734.8w	Ar	IR	1

DCNSi

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CD stretch	2059.0w	Ar	IR	1
	2	CN stretch	1485.9s	Ar	IR	1
	3	CD deform.	854.7vs	Ar	IR	1
	4	SiN stretch	665.3wm	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, H. Egenolf, and J. Glatthaar, Eur. J. Org. Chem. **1998**, 1307.

cyc-(HCSiN)

In an argon matrix,¹ irradiation at wavelengths longer than 395 nm or with a 366 nm laser leads to photoisomerization to HSiNC.

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CH stretch	2974.2m	Ar	IR	1
	2	CN stretch	1466.2wm	Ar	IR	1
	3	CH deform.	1124.7vs	Ar	IR	1
	4	SiC stretch	769.8vs	Ar	IR	1
	5	Ring deform.	635.5vs	Ar	IR	1
a''	6	CH deform.	820.6vs	Ar	IR	1

cyc-(DCSiN)

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	CN stretch	1452.2wm	Ar	IR	1
	3	CD deform.	944.0wm	Ar	IR	1
	4	SiC stretch	718.3m	Ar	IR	1
	5	Ring deform.	591.6vs	Ar	IR	1
	6	CD deform.	634.1w	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, H. Egenolf, and J. Glatthaar, Eur. J. Org. Chem. **1998**, 1307.

HSiNC

In an argon matrix,¹ irradiation at 254 nm leads to photoisomerization to HSiCN.

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	NC stretch	2039.5s	Ar	IR	1
	2	SiH stretch	2017.8vs	Ar	IR	1
	3	SiH deform.	867.9m	Ar	IR	1
	4	SiN stretch	622.0m	Ar	IR	1

DSiNC

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	NC stretch	2035.9vs	Ar	IR	1
	2	SiD stretch	1470.1m	Ar	IR	1
	3	SiD deform.	662.6m	Ar	IR	1
	4	SiN stretch	607.4wm	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, H. Egenolf, and J. Glatthaar, Eur. J. Org. Chem. **1998**, 1307.

HSiCN

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CN stretch	2142.7s	Ar	IR	1
	2	SiH stretch	2026.7vs	Ar	IR	1
	3	SiH deform.	826.4vs	Ar	IR	1
	4	SiC stretch	563.5wm	Ar	IR	1

DSiCN

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CN stretch	2143.2m	Ar	IR	1
	2	SiD stretch	1470.0vs	Ar	IR	1
	3	SiD deform.	642.1m	Ar	IR	1
	4	SiC stretch	547.3wm	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, H. Egenolf, and J. Glatthaar, Eur. J. Org. Chem. **1998**, 1307.

HCCP

\tilde{X}
 $B_0=0.188$ $C_{\infty v}$ MW¹

Structure: MW¹

DCCP

\tilde{X}
 $B_0=0.171$ $C_{\infty v}$ MW¹

Reference

¹I. K. Ahmad, H. Ozeki, and S. Saito, J. Chem. Phys. **107**, 1301 (1997).

HCBF

\tilde{X} $C_{\infty v}$						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CH stretch	3261.7	Ar	IR	1
	2	B=C stretch	1822.3	Ar	IR	1

DCBF

\tilde{X} $C_{\infty v}$						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CD stretch	2468.5	Ar	IR	1
	2	B=C stretch	1786.8	Ar	IR	1

Reference

¹D. V. Lanzisera and L. Andrews, J. Phys. Chem. A **104**, 9295 (2000).

HCBCI

\tilde{X} $C_{\infty v}$						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CH stretch	3251.2	Ar	IR	1
	2	B=C stretch	1691.1	Ar	IR	1

DCBCI

\tilde{X} $C_{\infty v}$						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CD stretch	2464.1	Ar	IR	1
	2	B=C stretch	1644.3	Ar	IR	1

Reference

¹D. V. Lanzisera and L. Andrews, *J. Phys. Chem. A* **104**, 9295 (2000).

HCBBr

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ^+	1	CH stretch	3248.2	Ar	IR	1
	2	B=C stretch	1665.1	Ar	IR	1

DCBBr

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ^+	1	CD stretch	2461.9	Ar	IR	1
	2	B=C stretch	1616.3	Ar	IR	1

Reference

¹D. V. Lanzisera and L. Andrews, *J. Phys. Chem. A* **104**, 9295 (2000).

HCCO

$\tilde{B}^2\Pi$		$C_{\infty v}$					$\tilde{B}-\tilde{X}$ 263–300 nm
$T_0 = 33464.75(5)$ gas PF ^{6,7} LF ^{8,9}							
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.	
Σ^+	3	CCO s-stretch	1037	gas	PF,LF	7–9	
Π	4	HCC deform. ($^2\Sigma$)	750.7	gas	LF	9	
		($^2\Sigma$)	564.3	gas	LF	9	
	5	CCO deform. ($^2\Sigma$)	382.0	gas	LF	9	
		($^2\Sigma$)	346.6	gas	LF	9	

$A_{\text{eff}} = -46.74(3)$ LF⁹
 $B_0 = 0.324$ PF⁷LF^{8,9}

Analysis of the submillimeter-wave spectrum¹ indicates that HCCO possesses a low-lying excited electronic state which, together with the ground state, is derived from a Π state by Renner–Teller interaction.

\tilde{X}^2A''		C_s Structure: MW ^{1,4}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	2	CCO a-stretch	2022.64	gas	DL	3
			2023.7	Ne	IR	5
			2019.5	Ar	IR	2
a''	6	CCH bend	494	gas	LF	8

$A_0 = 41.5(1.5)$; $B_0 = 0.363$; $C_0 = 0.359$ MW^{1,4}
 Barrier to linearity between 700 and 900 PE¹⁰

DCCO

$\tilde{B}^2\Pi$		$C_{\infty v}$				
$T_0 = 33448.99(5)$ gas LF ⁹						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Π	4	DCC deform. ($^2\Sigma$)	652.5	gas	LF	9
		($^2\Sigma$)	465.6	gas	LF	9
	5	CCO deform. ($^2\Sigma$)	380.0	gas	LF	9

$A_{\text{eff}} = -37.10(4)$ LF⁹
 $B_0 = 0.294$ LF⁹

\tilde{X}^2A''		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	2	CCO a-stretch	1995.2	Ne	IR	5
			1989.9	Ar	IR	2

$A_0 = 21.75(12)$; $B_0 = 0.331$; $C_0 = 0.325$ MW¹

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HNCO

\tilde{C}^2A''		C_s					$\tilde{C}-\tilde{X}$ 289–317 nm
$T_0 = 31521$ gas AB ^{2,3} PF ⁸							
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.	
a'		NCN s-stretch	1024	gas	AB,PF	2,3,8	

\tilde{B}^2A'		C_s Structure: AB ¹					$\tilde{B}-\tilde{X}$ 344–364 nm
$T_0 = 28994.1$ gas AB ¹ LF ⁴ PF ⁸							
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.	
a''	6	NCN deform.	560(10)	gas	PF	8	

$\tau_0 = 20(5)$ ns gas LF⁴
 $A_0 = 22.30$; $B_0 = 0.376$; $C_0 = 0.369$ AB¹LF⁴

\tilde{X}^2A'' C_s Structure: AB ¹						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	NH stretch	3297.5m	Ar	IR	6
	2	NCN a-stretch	1879(106)	gas	PE	7
			1843s	Ar	IR	6
3	NCN s-stretch	1049(162)	gas	PE	7	
		1146vs	Ar	IR	6	
a''	6	NCN deform.	440T	gas	LF	4

$A_0=21.18(11)$; $B_0=0.370$; $C_0=0.363$ AB¹LF⁴MW⁵

DNCN

\tilde{C}^2A'' C_s $T_0=31506$ gas AB ² PF ⁸ $\tilde{C}-\tilde{X}$ 307–317 nm						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		NCN s-stretch	1036	gas	PF	8

\tilde{X}^2A'' C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	NCN a-stretch	1879(106)	gas	PE	7
	3	NCN s-stretch	1049(162)	gas	PE	7

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HCNN

\tilde{B} $T_0=30500$ gas AB ^{1,2,4,5} $\tilde{A}-\tilde{X}$ 289–328 nm						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1048	gas	AB	4

\tilde{A}^2A' C_s $T_0=4830(720)$ gas PE ⁷						
\tilde{X}^2A'' C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CH stretch	3233.7wm	Ar	IR	3,6
			3229m	N ₂	IR	3
	2	CNN a-stretch	1780(77)	gas	PE	7
			1786.6s	Ar	IR	3,6
			1784s	Kr	IR	3
			1800s	N ₂	IR	3
			862.2vs	Ar	IR	3,6
	5	CNN deform.	860vs	Kr	IR	3
			871m	N ₂	IR	3
			863m			
485(70)	gas	PE	7			

DCNN

\tilde{A}^2A' C_s $T_0=4840(720)$ gas PE ⁷						
\tilde{X}^2A'' C_s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	CNN a-stretch	1790(74)	gas	PE	7
			1771vs	Ar	IR	3
	5	CNN deform.	725vs	Ar	IR	3
			460(70)	gas	PE	7

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HCCO⁻

Threshold for electron detachment from ground-state HCCO⁻ = 18860(65) gas PE^{1,2}

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HNCN⁻

Threshold for electron detachment from ground-state
 HNCN⁻ = 7330(20) gas PE¹

DNCN⁻

Threshold for electron detachment from ground-state
 DNCN⁻ = 7330(20) gas PE¹

Reference

¹E. P. Clifford, P. G. Wenthold, W. C. Lineberger, G. A. Petersson, and G. B. Ellison, *J. Phys. Chem. A* **101**, 4338 (1997).

HCNN⁻

Threshold for electron detachment from ground-state
 HCNN⁻ = 13600(50) gas PE¹

DCNN⁻

Threshold for electron detachment from ground-state
 DCNN⁻ = 13540(50) gas PE¹

Reference

¹E. P. Clifford, P. G. Wenthold, W. C. Lineberger, G. A. Petersson, K. M. Broadus, S. R. Kass, S. Kato, C. H. DePuy, V. M. Bierbaum, and G. B. Ellison, *J. Phys. Chem. A* **102**, 7100 (1998).

HNCO

Between 132.5 and 120 nm, continuous absorption, with some diffuse bands.⁵

Relatively sharp absorption bands at 72940, 73910, and 74680 (137.1, 135.3, and 133.9 nm), which may be contributed by a Rydberg transition.⁵

Between 147 and 163 nm, diffuse bands are superposed on a continuum, with maximum near 157 nm.⁵

Continuous absorption between 163 and 185 nm, with a maximum near 166 nm.⁵

Continuous absorption has its onset near 41000 (244 nm), and extends beyond 200 nm.⁴

\tilde{A}^1A'' C_s
 $T_0 = 32449(20)$ gas AB⁴PF¹⁸ $\tilde{A}-\tilde{X}$ 228–282 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	3	HNC bend	1192(10)	gas	PF	18
	4	NC stretch	1034(20)	gas	PF	18
	5	NCO bend	599(7)	gas	PF	18

$A \cong 4.37$; $B \cong 0.388$; $C \cong 0.357$ AB⁴

\tilde{X}^1A' C_s Structure: MW⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	NH stretch	3538.25s 3516.8wm 3505.7wm 3514.7 3500.6 3485.2 3483.0 3479.9	gas Ar Kr Xe	IR,Ra IR	1,2,12,14,15 12 17 17
	2	NCO a-stretch	2268.89vs 2259.0vs 2261.6 2254.4	gas Ar Kr Xe	IR,Ra IR	1,2,9,10,16 12,13 17 17
	3	NCO s-stretch	1327vw	gas	IR,Ra	1,2,16
	4	HNC, NCO bend	776.62wm 769.8wm 763.6 765.1 760.3	gas Ar Kr Xe	IR	6 12,13 17 17
	5	HNC, NCO bend	577.35w 573.7wm 572.9 571.3 574.2 570.3	gas Ar Kr Xe	IR	1,2,6 12,13 17 17
<i>a''</i>	6	Torsion	656.29 695.6 (1-0) 694.7 (1-0)	gas Kr Xe	IR	6,11 17 17

$A_0 = 30.638$; $B_0 = 0.369$; $C_0 = 0.364$ MW⁷IR¹⁴

DNCO

\tilde{X}^1A' C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	ND stretch	2637.20 2606.9m	gas Ar	IR	8 12
	2	NCO a-stretch	2235vs 2231.0vs	gas Ar	IR	3 12
	3	NCO s-stretch	1310	gas	IR	3
	4	DNC, NCO bend	578.6w	Ar	IR	12
	5	DNC, NCO bend	475.4w	Ar	IR	12
<i>a''</i>	6	Torsion	602.9	gas	IR	3

$A_0 = 17.09$; $B_0 = 0.344$; $C_0 = 0.336$ MW⁷

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¹⁷M. Pettersson, L. Khriachtchev, S. Jolkkonen, and M. Räsänen, *J. Phys. Chem. A* **103**, 9154 (1999).
¹⁸H. L. Berghout, F. F. Crim, M. Zyryanov, and H. Reisler, *J. Chem. Phys.* **112**, 6678 (2000).

HOCN

\tilde{X}		C _s		Structure: MO ²⁻⁴		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	OH stretch	3610	Ne	IR	5
			3569.6s	Ar	IR	1,6
			3536	Kr	IR	8
	2	C≡N stretch	3506s	N ₂	IR	1
			2302	gas	IR	9
			2294	Ne	IR	5
			2286.3vs	Ar	IR	1,6,7
			2291.1	Kr	IR	8
			2294s	N ₂	IR	1
	3	OH deform.	1227	Ne	IR	5
			1227.9s	Ar	IR	1,6,7
			1220.6	Kr	IR	8
			1241m	N ₂	IR	1
			1082	Ne	IR	5
			1081.3m	Ar	IR	1,6,7
4	C–O stretch	1088.6	Kr	IR	8	
		1098s	N ₂	IR	1	
		460wm	N ₂	IR	1	
5	OCN deform.					

DOCN

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	OD stretch	2635.0vs	Ar	IR	1,6
			2590sh	N ₂	IR	1
	2	C≡N stretch	2284.6vs	Ar	IR	1,6
			2292s	N ₂	IR	1
	3	OD deform.	1077.8ms	Ar	IR	1,6
			1093m	N ₂	IR	1
	4	C–O stretch	949.4m	Ar	IR	1,6
			957m	N ₂	IR	1
	5	OCN deform.	437wm	N ₂	IR	1

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HCNO

\tilde{A} C_s
 $T_0 \leq 35053$ gas AB⁸ $\tilde{A}-\tilde{X}$ 244–285 nm
 Underlying absorption continuum, with increasing intensity at shorter wavelengths.⁸

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Bend	345(5)	gas	AB	8
\tilde{X}		C _{∞v} ^a		Structure: MW,IR ¹³ MO ¹⁵		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CH stretch	3336.12	gas	IR	1,10,18
			3338	Ne	IR	14
			3317.2s	Ar	IR	17
			3311	N ₂	IR	17
			2195.76	gas	IR	1,2,10,18
	2	CNO a-stretch	2200	Ne	IR	14
			2192.7vs	Ar	IR	17
			2200	N ₂	IR	17
			1254.23	gas	IR	1,2,10,18
			1250	Ne	IR	14
	3	CNO s-stretch	1244.1m	Ar	IR	17
			1232	N ₂	IR	17
			537.18	gas	IR	1,2,9,16,18,19
			539	Ne	IR	14
	II	CNO bend	538.2w	Ar	IR	17
536.9w						
528			N ₂	IR	17	
224.10			gas	IR	9,11,18,19	
560			Ne	IR	17	
566.6m			Ar	IR	17	
582			N ₂	IR	17	

$B_0 = 0.383$ MW^{3,4}

DCNO

$\tilde{\chi}$		$C_{\infty v}^a$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CD stretch	2620.73	gas	IR	7
			2612.7vs	Ar	IR	17
	2	CNO a-stretch	2070.92	gas	IR	12
2063.2s			Ar	IR	17	
3	CNO s-stretch	1254	gas	IR	2,6	
		1218.5m	Ar	IR	17	
II	5	DCN bend	162.64	gas	IR	9,19
			418.5wm	Ar	IR	17

$$B_0 = 0.343 \text{ MW}^{3.5} \text{ IR}^7$$

^aQuasilinear. See discussion in Refs. 13, 15, and 17.

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HSCN

$\tilde{\chi}$		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
1	SH stretch		2581.0	Ar	IR	1
			2580.2	N ₂	IR	1
2	CN stretch		2182.3	Ar	IR	1
			2182.3	N ₂	IR	1
HSC bend			959.7	Ar	IR	1
			960.9	N ₂	IR	1

DSCN

$\tilde{\chi}$		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
1	CN stretch		2183.0	Ar	IR	1
			2183.8	N ₂	IR	1
2	SD stretch		1874.0	Ar	IR	1
			1873.6	N ₂	IR	1
DSC bend			692.8	Ar	IR	1
			688.5	N ₂	IR	1

Reference

¹M. Wierzejewska and Z. Mielke, *Chem. Phys. Lett.* **349**, 227 (2001).

HSNC

$\tilde{\chi}$		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
2	NC stretch		2064.2T	Ar	IR	1
			2065.9T	N ₂	IR	1

DSNC

$\tilde{\chi}$		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
2	NC stretch		2063.5T	Ar	IR	1
			2065.4T	N ₂	IR	1

Reference

¹M. Wierzejewska and Z. Mielke, *Chem. Phys. Lett.* **349**, 227 (2001).

t-HNNO

$\tilde{\chi}$		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	NH stretch	3254.0wm	Xe	IR	1
	2	NN stretch	1628.9vs 1627.1	Xe	IR	1
3	NO stretch		1296.2m 1294.5	Xe	IR	1
4	HNN bend		1214.7m 1213.4	Xe	IR	1
a''	6	Torsion	746.5wm	Xe	IR	1

Reference

¹S. L. Laursen, A. E. Delia, and K. Mitchell, *J. Phys. Chem. A* **104**, 3681 (2000).

c-HNNO

In a xenon matrix,¹ irradiation at 700 nm (14290) results in photoisomerization to *t*-HNNO.

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	NH stretch	3158vw	Xe	IR	1
	2	NN stretch	1623.7s 1622.3	Xe	IR	1
	3	NO stretch	1273.4wm	Xe	IR	1
<i>a''</i>	4	HNN bend	1166.9wm	Xe	IR	1
	6	Torsion	712.1	Xe	IR	1

Reference

¹S. L. Laursen, A. E. Delia, and K. Mitchell, *J. Phys. Chem. A* **104**, 3681 (2000).

HFCN

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CH stretch	3016w	Ar	IR	1,2
	2	C=N stretch	1672s	Ar	IR	1,2
	3	HCF deform.	1233w	Ar	IR	2
	4	CF stretch	1057vs	Ar	IR	1,2
<i>a''</i>	5	FCN bend	536m	Ar	IR	1,2
	6	OPLA	894w	Ar	IR	2

DFCN

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CD stretch	2252w	Ar	IR	1,2
	2	C=N stretch	1651m	Ar	IR	1,2
	3	CF stretch	1047vs	Ar	IR	1,2
	4	DCF deform.	922w	Ar	IR	2
	5	FCN bend	530wm	Ar	IR	1,2
<i>a''</i>	6	OPLA	735vw	Ar	IR	2

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¹R. D. Hunt and L. Andrews, *Inorg. Chem.* **26**, 3051 (1987).

²E. Ya. Misochko, I. U. Goldschleger, A. V. Akimov, and C. A. Wight, *J. Am. Chem. Soc.* **123**, 5156 (2001).

FCNH

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	NH stretch	3405wm	Ar	IR	1
	2	CN stretch	1700s	Ar	IR	1
	3	CF stretch	1172wm	Ar	IR	1
	4	Deformation	905vs	Ar	IR	1
	5	Deformation	595w	Ar	IR	1
<i>a''</i>	6	Torsion	734wm	Ar	IR	1

FCND

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	ND stretch	2532wm	Ar	IR	1
	2	CN stretch	1701vs	Ar	IR	1
	3	CF stretch	1142vs	Ar	IR	1
	4	Deformation	759s	Ar	IR	1
<i>a''</i>	6	Torsion	575wm	Ar	IR	1

Reference

¹E. Ya. Misochko, I. U. Goldschleger, A. V. Akimov, and C. A. Wight, *J. Am. Chem. Soc.* **123**, 5156 (2001).

HAICl₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	AIH stretch	1968s	Ar	IR	1,2
	2	AlCl s-stretch	481s	Ar	IR	1,2
<i>b</i> ₁	4	OPLA	472s	Ar	IR	1,2
<i>b</i> ₂	5	AlCl a-stretch	579m	Ar	IR	1,2
	6	H deformation	654vs	Ar	IR	1,2

DAICl₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	AID stretch	1430	Ar	IR	1
	2	AlCl s-stretch	478	Ar	IR	1
<i>b</i> ₁	4	OPLA	355	Ar	IR	1
<i>b</i> ₂	6	AlCl a-stretch	598	Ar	IR	1

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²J. Müller and B. Wittig, *Eur. J. Inorg. Chem.* **1998**, 1807.

HAlBr₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	AlH stretch	1952.5	Ar	IR	1
	2	AlBr s-stretch	366.0	Ar	IR	1
b ₁	4	OPLA	453.5	Ar	IR	1
	5	HAlBr deform.	634.5	Ar	IR	1
b ₂	6	AlBr a-stretch	476.0	Ar	IR	1

Reference

¹J. Müller and B. Wittig, Eur. J. Inorg. Chem. **1998**, 1807.

HGaCl₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	GaH stretch	2015.3s	Ar	IR	1,2
	2	GaCl ₂ s-stretch	414.3wm	Ar	IR	1,2
b ₁	4	OPLA	464.3ms	Ar	IR	1,2
	5	H deform.	607.5vs	Ar	IR	1,2
b ₂	6	GaCl ₂ a-stretch	437.3s	Ar	IR	1,2

DGaCl₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	GaD stretch	1450.2	Ar	IR	1
	2	GaCl ₂ s-stretch	414.5	Ar	IR	1
b ₂	5	Mixed	469.6	Ar	IR	1
	6	Mixed	420.4	Ar	IR	1

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¹R. Köppe, M. Tacke, and H. Schnöckel, Z. Anorg. Allg. Chem. **605**, 35 (1991).

²J. Müller and H. Sternkicker, J. Chem. Soc., Dalton Trans., 4149 (1999).

HGaBr₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	GaH stretch	1991.5s	Ar	IR	1
	2	GaBr ₂ s-stretch	287.5w	Ar	IR	1
b ₁	4	OPLA	445.5wm	Ar	IR	1
	5	HGaBr deform.	597.0vs	Ar	IR	1
b ₂	6	GaBr ₂ a-stretch	333.0m	Ar	IR	1

Reference

¹J. Müller and H. Sternkicker, J. Chem. Soc., Dalton Trans., 4149 (1999).

HInCl₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	InH stretch	1846.9	Ar	IR	1
	2	InCl ₂ s-stretch	369.1	Ar	IR	1
b ₁	4	OPLA	402.6	Ar	IR	1
	5	HInCl deform.	469.7	Ar	IR	1
b ₂	6	InCl ₂ a-stretch	358.5	Ar	IR	1

DInCl₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	InD stretch	1325.6	Ar	IR	1
	2	InCl ₂ s-stretch	368.8	Ar	IR	1
b ₁	4	OPLA	304.8	Ar	IR	1
	5	DInCl deform.	311.2	Ar	IR	1
b ₂	6	InCl ₂ a-stretch	385.1	Ar	IR	1

Reference

¹H.-J. Himmel, A. J. Downs, and T. M. Greene, J. Am. Chem. Soc. **122**, 922 (2000).

HCOCI

\tilde{A}^1A'' C _s						
T ₀ =32760 gas AB ^{5,6} CR ¹² $\tilde{A}-\tilde{X}$ 230–314 nm						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	CO stretch	1153.8	gas	AB	5
	4	CCl stretch	633.6	gas	AB	5
	5	CICO deform.	306.3	gas	AB	5
a''	6	Umbrella (OPLA)	779.5	gas	AB	5

Barrier to planarity=1642 gas AB⁵CR¹²

\tilde{X}^1A' C _s Structure: MW ^{2,4}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CH stretch	2929.18w	gas	IR	1,3,6
	2	CO stretch	1784.14vs	gas	IR	1,3,6,8
	3	CH bend	1307.21m	gas	IR	1,3,6,7
	4	CCl stretch	738.80vs	gas	IR	1,3,6,9
	5	CCl bend	457.02vw	gas	IR	1,6,9
a''	6	OPLA	932.15vw	gas	IR	1,3,6,11

A₀=2.601; B₀=0.205; C₀=0.190 MW^{2,4}IR⁷

DCOCI

\tilde{A}^1A'' C_s
 $T_0=32775.3$ gas AB⁵ $\tilde{A}-\tilde{X}$ 283–313 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	CO stretch	1092.0	gas	AB	5
	4	CCl stretch	633.4	gas	AB	5
	5	CICO deform.	303.1	gas	AB	5
a''	6	Umbrella (OPLA)	566.5	gas	AB	5

\tilde{X}^1A' C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CD stretch	2204.3	gas	IR	4,10
	2	CO stretch	1749.1	gas	IR	4,10
	3	CD bend	986.11	gas	IR	4,10
	4	CCl stretch	700.77	gas	IR	4,10,11
	5	CCl bend	455.78	gas	IR	10
a''	6	OPLA	781.50	gas	IR	10,11

$A_0=1.904$; $B_0=0.203$; $C_0=0.184$ MW^{2,4}DL¹⁰

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HCCl₂⁺

A broad, unstructured absorption observed near 250 nm in argon-matrix experiments² in which infrared absorptions of HCCl₂⁺ are prominent has been attributed to an excited state of HCCl₂⁺ which can undergo proton transfer to the matrix.

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CH stretch	3032.8w	Ar	IR	3,6
	2	CCl stretch	860(30)	gas	PE	4,5
			845w	Ar	IR	3
b_2	5	H deformation	1291m	Ar	IR	1–3,6
	6	CCl stretch	1044s	Ar	IR	1–3,6
			1038	Kr	IR	6

DCCl₂⁺

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CCl stretch	790(30)	gas	PE	4,5
b_2	5	CCl stretch	1122s	Ar	IR	1,2,6
	6	D deformation	864wm	Ar	IR	1,2,6

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HCCIBr⁺

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CCl stretch	994m	Ar	IR	1,2

DCCIBr⁺

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1077s	Ar	IR	1

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HKrCN

\tilde{X} $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	2	KrH stretch	1497.4	Kr	IR	1

DKrCN

\tilde{X} $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	2	KrD stretch	1109.0	Kr	IR	1

Reference

¹M. Pettersson, J. Lundell, L. Khriachtchev, and M. Räsänen, *J. Chem. Phys.* **109**, 618 (1998).

HXeCN

In a xenon matrix, an unstructured absorption maximum at 42740 (234 nm) has been assigned³ to HXeCN.

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	2	XeH stretch	1650	Kr	IR	1
			1623.8	Xe	IR	1-3

DXeCN

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	2	XeD stretch	1178.0	Xe	IR	1

References

¹M. Pettersson, J. Lundell, L. Khriachtchev, and M. Räsänen, *J. Chem. Phys.* **109**, 618 (1998).

²M. Pettersson, L. Khriachtchev, S. Jolkkonen, and M. Räsänen, *J. Phys. Chem. A* **103**, 9154 (1999).

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HXeNC

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	NC stretch	2048.5	Kr	IR	1
			2043.8	Xe	IR	1
	2	XeH stretch	1878	Kr	IR	1
			1851.0	Xe	IR	1-3

DXeNC

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	2	XeD stretch	1338.8	Xe	IR	1

References

¹M. Pettersson, J. Lundell, L. Khriachtchev, and M. Räsänen, *J. Chem. Phys.* **109**, 618 (1998).

²M. Pettersson, L. Khriachtchev, S. Jolkkonen, and M. Räsänen, *J. Phys. Chem. A* **103**, 9154 (1999).

³J. Ahokas, K. Vaskonen, J. Eloranta, and H. Kunttu, *J. Phys. Chem. A* **104**, 9506 (2000).

t-HONO

\tilde{B}

An intense, unstructured absorption between 200 and 275 nm, with a maximum near 46500 (215 nm), has been attributed¹³ to HONO.

\tilde{A}^1A'' C_s
 $T_0=26034$ gas $AB^{1-3,6,25}$ $\tilde{A}-\tilde{X}$ 315-385 nm
 Diffuse bands; predissociated into OH+NO.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		NO stretch	1117	gas	AB	6

\tilde{X}^1A' C_s Structure: MW^{9,11,12}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	OH stretch	3590.71m	gas	IR	4,5,8,16,17,21
			3572.6	Ar	IR	24
			3568.5			
	2	N=O stretch	3551.8	Kr	IR	27
			3558	N ₂	IR	7
			1699.76s	gas	IR,LS	4,5,8,14
					DL	16-18,23
			1689.1	Ar	IR	10,20,24
			1688.0			
			1684.0	Kr	IR	27
3	HON bend	1684	N ₂	IR	7	
		1263.21s	gas	IR,DL	4,5,8,16,17,19,23,26	
		1265.8	Ar	IR	24	
		1263.9				
		1265.3	Kr	IR	27	
		1298	N ₂	IR	7	
		790.12s	gas	IR	4,5,8,16,17,22,26	
		800.4	Ar	IR	10,20,24	
5	ONO bend	796.6				
		794.5	Kr	IR	27	
		815	N ₂	IR	7	
		595.6s	gas	IR	4,5,8,17	
		608.7	Ar	IR	24	
		606.7	Kr	IR	27	
		625	N ₂	IR	7	
		543.0m	gas	IR	4,5,8,17	
a''	6	Torsion	549.4	Ar	IR	10,24
			549.1	Kr	IR	27
			583	N ₂	IR	7

$A_0=3.099$; $B_0=0.418$; $C_0=0.367$ MW^{9,11,15}IR²¹

t-DONO

\tilde{A}^1A'' C_s
 $T_0=26050(10)$ gas $AB^{2,3,6}$ $\tilde{A}-\tilde{X}$ 315–385 nm
 Diffuse bands; predissociated into OD+NO.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		NO stretch	1147(20)	gas	AB	6

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	OD stretch	2651.13	gas	IR	4,5,8,17
			2620	N ₂	IR	7
	2	N=O stretch	1693.98	gas	IR	4,5,8,17
			1682	N ₂	IR	7
	3	DON bend	1012.68	gas	IR	4,5,8,17
			1030	N ₂	IR	7
	4	O–N stretch	736.27	gas	IR	4,5,8,17
			769	N ₂	IR	7
	5	ONO bend	590.4	gas	IR	4,5,8,17
			618	N ₂	IR	7
a''	6	Torsion	416.1	gas	IR	4,8,17
			444	N ₂	IR	7

$A_0=2.981$; $B_0=0.389$; $C_0=0.344$ MW^{9,11}

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c-HONO \tilde{B}

An intense, unstructured absorption between 200 and 275 nm, with a maximum near 46500 (215 nm), has been attributed¹³ to HONO.

\tilde{A}^1A'' C_s
 $T_0=26320$ gas $AB^{1-3,6}$ $\tilde{A}-\tilde{X}$ 315–385 nm
 Diffuse bands; predissociated into OH+NO.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		NO stretch	1107	gas	AB	6

\tilde{X}^1A' C_s Structure: MW^{10,11}
 141(35) higher in energy than *t*-HONO (\tilde{X}). MW¹²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	OH stretch	3426.20w	gas	IR	4,5,8,16,17,21
			3412.4	Ar	IR	9,22
			3410.7			
			3399.7	Kr	IR	24
			3410	N ₂	IR	7
	2	N=O stretch	1640.52m	gas	IR,LS	4,5,8,14,16,21
			1634.0	Ar	IR	9,19,22
			1632.8			
			1629.0	Kr	IR	24
			1633	N ₂	IR	7
	3	HON bend	1315.2	Kr	IR	24
	4	O–N stretch	851.94s	gas	IR,DL	4,5,8,16–18,20,23
			853.1	Ar	IR	9,19,22
			850.2			
			851.9	Kr	IR	24
			865	N ₂	IR	7
	5	ONO bend	609.0w	gas	IR	8,17
			608	Ar	IR	9,19
			616.6	Kr	IR	24
a''	6	Torsion	638.5m	gas	IR	4,5,8,17
			638.4	Ar	IR	9,19,22
			635.0	Kr	IR	24
			658	N ₂	IR	7

$A_0=2.805$; $B_0=0.439$; $C_0=0.379$ MW^{10,15}DL¹⁸IR²¹

c-DONO

\tilde{A}^1A'' C_s
 gas $AB^{2,3}$ $\tilde{A}-\tilde{X}$ 315–385 nm
 Diffuse bands; predissociated into OD+NO.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	4	O–N stretch	660T	gas	AB	6

\tilde{X}^1A'		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	OD stretch	2525	gas	IR	4,5,8
			2518	N_2	IR	7
	2	N=O stretch	1625	gas	IR	5,8
			1612	N_2	IR	7
3	DON bend	1008	gas	IR	8	
4	O-N stretch	813.50	gas	IR	4,5,8,17	
		828	N_2	IR	7	
5	ONO bend	601	gas	IR	8	
a''	6	Torsion	508.2	gas	IR	4,8,17
			522	N_2	IR	7

$A_0=2.362$; $B_0=0.430$; $C_0=0.363$ MW¹⁰

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HNOO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	NH stretch	3165.5w	Xe	IR	1
	2	HNO bend	1489.8	Ar	IR	1
			1485.5wm	Xe	IR	1
3	NO stretch	1097.7	Ar	IR	1	
		1096.6				
4	OO stretch	1092.3vs	Xe	IR	1	
		1058.0	Ar	IR	1	
		1054.5m	Xe	IR	1	
a''	6	Torsion	776.3	Ar	IR	1
			764.0ms	Xe	IR	1

Reference

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c-HOPO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	OH stretch	3550.7	Ar	IR	1
	2	P=O stretch	1252.6	Ar	IR	1
	4	P-O stretch	851.07	gas	DL	2
a''	6	Torsion	841.5	Ar	IR	1
			523.9	Ar	IR	1

$A_0=1.239(3)$; $B_0=0.318$; $C_0=0.254$ DL²

c-DOPO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	OD stretch	2620.4	Ar	IR	1
	2	P=O stretch	1253.0	Ar	IR	1

References

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HCF₂

\tilde{F} 3p Rydberg state C_{2v}
 $T_0=49312(10)$ gas MPI⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CF stretch	1365(8)	gas	MPI	4
	3	CF ₂ scissors	660(20)	gas	MPI	4
b_1	4	OPLA	1022(1)	gas	MPI	4

\tilde{X} C_s Structure: ESR¹MPI⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	CF s-stretch	1164s	Ar	IR	2,3
	3	Umbrella	949(10)	gas	MPI	4
a''	5	HCF deform.	1317m	Ar	IR	2,3
	6	CF a-stretch	1173vs	Ar	IR	2,3

Barrier to inversion=2800(500) gas MPI⁴
 $A_0=2.242$; $B_0=0.368$; $C_0=0.320$ MW⁵

DCF₂

\tilde{F} 3p Rydberg state C_{2v}
 $T_0=49323(10)$ gas MPI⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CF stretch	1300(21)	gas	MPI	4
	3	CF ₂ scissors	650(15)	gas	MPI	4
b_1	4	OPLA	864(2)	gas	MPI	4

\tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	CF s-stretch	1143m	Ar	IR	2,3
	3	Umbrella	794(4)	gas	MPI	4
a''	5	CF a-stretch	1214s	Ar	IR	2,3
	6	DCF deform.	933wm	Ar	IR	2,3

References

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- ²T. G. Carver and L. Andrews, J. Chem. Phys. **50**, 5100 (1969).
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HCCl₂

3d Rydberg state C_{2v}
 $T_0=54024(10)$ gas MPI³ $3d-\tilde{X}$ 179–185 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CCl ₂ s-stretch	845(10)	gas	MPI	3

\tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a''	5	HCCl deform.	1226m 1222	Ar Kr	IR IR	1,4 4
	6	CCl ₂ a-stretch	902vs 896	Ar Kr	IR IR	1,4 4

DCCl₂

3d Rydberg state C_{2v}
 $T_0=53980(10)$ gas MPI³ $3d-\tilde{X}$ 180–185 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CCl ₂ s-stretch	814(10)	gas	MPI	3

\tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a''	5	DCCl deform.	974vs	Ar	IR	1,2
	6	CCl ₂ a-stretch	814s	Ar	IR	1,2

References

- ¹T. G. Carver and L. Andrews, J. Chem. Phys. **50**, 4235 (1969).
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- ³G. R. Long and J. W. Hudgens, J. Phys. Chem. **91**, 5870 (1987).
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HCCIBr

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	2	H deformation	1196m	Ar	IR	1,2
	3	CCl stretch	866s	Ar	IR	1,2

References

- ¹T. G. Carver and L. Andrews, J. Chem. Phys. **50**, 4235 (1969).
- ²T. D. Fridgen, X. K. Zhang, J. M. Parnis, and R. E. March, J. Phys. Chem. A **104**, 3487 (2000).

HO₃

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	OH stretch	3361	Ar	IR	1
		OH deform.	1223	Ar	IR	1
		O ₃ deform.	566	Ar	IR	1

DO₃ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	OD stretch	2486	Ar	IR	1
		OD deform.	940	Ar	IR	1
		O ₃ deform.	552	Ar	IR	1

Reference

¹B. Nelander, A. Engdahl, and T. Svensson, Chem. Phys. Lett. **332**, 403 (2000); Chem. Phys. Lett. **339**, 295 (2001).

8.7. Four-Atomic Nonhydrides**K₄** \tilde{X} D_{2h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a _g		Sym. stretch	93.0	Kr	Ra	1
		Deformation	30.0	Kr	Ra	1
b _{3g}		Deformation	61.5	Kr	Ra	1

Reference

¹A. Kornath, R. Ludwig, and A. Zoerner, Angew. Chem. **110**, 1620 (1998); Angew. Chem. Int. Ed. **37**, 1575 (1998).

Ag₄

Mass-selected Ag₄ trapped in an argon matrix has a prominent fluorescence band at 21830 (458 nm) which is strongest on excitation of the molecule at 24690 (405 nm), 33440 (299 nm), and 36630 (273 nm).

Reference

¹C. Félix, C. Sieber, W. Harbich, J. Buttet, I. Rabin, W. Schulze, and G. Ertl, Chem. Phys. Lett. **313**, 105 (1999).

Nb₃O⁺ \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	3	Nb ₃ deformation	312	gas	TPE	1

Reference

¹D.-S. Yang, M. Z. Zgierski, D. M. Rayner, P. A. Hackett, A. Martinez, D. R. Salahub, P.-N. Roy, and T. Carrington, Jr., J. Chem. Phys. **103**, 5335 (1995).

V₃O \tilde{A} T₀ = 3350(30) gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	3		310(15)	gas	PE	1

 \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	VO stretch	750(20)	gas	PE	1
	2		415(15)	gas	PE	1
	3		340(15)	gas	PE	1

Reference

¹S. M. E. Green, S. Alex, N. L. Fleischer, E. L. Millam, T. P. Marcy, and D. G. Leopold, J. Chem. Phys. **114**, 2653 (2001).

Nb₃O \tilde{B} T₀ = 6190(120) gas PE² \tilde{A} T₀ = 4300(100) gas PE² \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	NbO stretch	710(15)	gas	PE	2
	3	Nb ₃ deformation	320	gas	TPE,PE	1,2

References

¹D.-S. Yang, M. Z. Zgierski, D. M. Rayner, P. A. Hackett, A. Martinez, D. R. Salahub, P.-N. Roy, and T. Carrington, Jr., J. Chem. Phys. **103**, 5335 (1995).

²S. M. E. Green, S. Alex, N. L. Fleischer, E. L. Millam, T. P. Marcy, and D. G. Leopold, J. Chem. Phys. **114**, 2653 (2001).

Ta₃O \tilde{B} T₀ = 3870(50) gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			250(15)	gas	PE	1

\tilde{A}
 $T_0=3180(40)$ gas PE¹

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	TaO stretch	710(15)	gas	PE	1
	3		225(15)	gas	PE	1

Reference

¹S. M. E. Green, S. Alex, N. L. Fleischer, E. L. Millam, T. P. Marcy, and D. G. Leopold, *J. Chem. Phys.* **114**, 2653 (2001).

V₃O⁻

Threshold for electron detachment from ground-state V₃O⁻ = 9830(65) gas PE¹

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	VO stretch	770(20)	gas	PE	1
	3		355(20)	gas	PE	1

Reference

¹S. M. E. Green, S. Alex, N. L. Fleischer, E. L. Millam, T. P. Marcy, and D. G. Leopold, *J. Chem. Phys.* **114**, 2653 (2001).

Nb₃O⁻

Threshold for electron detachment from ground-state Nb₃O⁻ = 11240(50) gas PE¹

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	3		300(20)	gas	PE	1

Reference

¹S. M. E. Green, S. Alex, N. L. Fleischer, E. L. Millam, T. P. Marcy, and D. G. Leopold, *J. Chem. Phys.* **114**, 2653 (2001).

Ta₃O⁻

Threshold for electron detachment from ground-state Ta₃O⁻ = 12770(80) gas PE¹

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	3		215(20)	gas	PE	1

Reference

¹S. M. E. Green, S. Alex, N. L. Fleischer, E. L. Millam, T. P. Marcy, and D. G. Leopold, *J. Chem. Phys.* **114**, 2653 (2001).

cyc-(ScN)₂⁺

\tilde{X} D_{2h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _{2u}			807.3	Ar	IR	1

Reference

¹G. V. Chertihin, L. Andrews, and C. W. Bauschlicher, Jr., *J. Am. Chem. Soc.* **120**, 3205 (1998).

cyc-(YN)₂⁺

\tilde{X} D_{2h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _{2u}			731.5	Ar	IR	1

Reference

¹G. V. Chertihin, W. D. Bare, and L. Andrews, *J. Phys. Chem. A* **102**, 3697 (1998).

CaI₃

\tilde{C} D_{3h}
 $T_0=12260(290)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			440(50)	gas	PE	1

\tilde{B} D_{3h}
 $T_0=5410(510)$ gas PE¹

\tilde{A} C_{2v}
 $T_0=1050(360)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			610(60)	gas	PE	1

\tilde{X}^2B_2 C_{2v}

Reference

¹A. I. Boldyrev, J. Simons, X. Li, W. Chen, and L.-S. Wang, *J. Chem. Phys.* **110**, 8980 (1999).

cyc-SiAl₃

\tilde{C} C_{2v}
 $T^a = 6540(1030)$ gas PE¹

\tilde{B} C_{2v}
 $T^a = 1210(760)$ gas PE¹

\tilde{X}, \tilde{A} C_{2v}

^aFrom vertical electron detachment energies.

Reference

¹X. Li, H.-F. Zhang, L.-S. Wang, A. E. Kuznetsov, N. A. Cannon, and A. I. Boldyrev, *Angew. Chem. Int. Ed.* **40**, 1867 (2001).

cyc-GeAl₃

\tilde{C} C_{2v}
 $T^a = 7660(1620)$ gas PE¹

\tilde{B} C_{2v}
 $T^a = 1050(970)$ gas PE¹

\tilde{X}, \tilde{A} C_{2v}

^aFrom vertical electron detachment energies.

Reference

¹X. Li, H.-F. Zhang, L.-S. Wang, A. E. Kuznetsov, N. A. Cannon, and A. I. Boldyrev, *Angew. Chem. Int. Ed.* **40**, 1867 (2001).

cyc-SnAl₃

\tilde{C} C_{2v}
 $T^a = 8550(580)$ gas PE¹

\tilde{B} C_{2v}
 $T^a = 1780(360)$ gas PE¹

\tilde{X}, \tilde{A} C_{2v}

^aFrom vertical electron detachment energies.

Reference

¹X. Li, H.-F. Zhang, L.-S. Wang, A. E. Kuznetsov, N. A. Cannon, and A. I. Boldyrev, *Angew. Chem. Int. Ed.* **40**, 1867 (2001).

cyc-PbAl₃

\tilde{C} C_{2v}
 $T^a = 10300(510)$ gas PE¹

\tilde{B} C_{2v}
 $T^a = 4600(180)$ gas PE¹

\tilde{A} C_{2v}
 $T^a = 440(230)$ gas PE¹

\tilde{X} C_{2v}

^aFrom vertical electron detachment energies.

Reference

¹X. Li, H.-F. Zhang, L.-S. Wang, A. E. Kuznetsov, N. A. Cannon, and A. I. Boldyrev, *Angew. Chem. Int. Ed.* **40**, 1867 (2001).

cyc-ScC₃

\tilde{A} C_{2v}
 $T_0 = 3390(160)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CScC s-stretch	600(50)	gas	PE	1

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CScC s-stretch	560(30)	gas	PE	1

Reference

¹L.-S. Wang and X. Li, *J. Chem. Phys.* **112**, 3602 (2000).

cyc-TiC₃

\tilde{C}
 $T_0 = 8660(190)$ gas PE¹

\tilde{B}
 $T_0 = 6730(190)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			500(60)	gas	PE	1

\tilde{A}
 $T_0 = 5120(190)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			500(60)	gas	PE	1

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			650(30)	gas	PE	1

Reference

¹X.-B. Wang, C.-F. Ding, and L.-S. Wang, *J. Phys. Chem. A* **101**, 7699 (1997).

cyc-VC₃

\tilde{B}						
$T_0=12180(240)$		C_{2v}	gas PE ¹			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CVC s-stretch	600(50)	gas	PE	1
\tilde{A}						
$T_0=2580(160)$		C_{2v}	gas PE ¹			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CVC s-stretch	450(50)	gas	PE	1
\tilde{X}						
		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CVC s-stretch	600(30)	gas	PE	1

Reference

¹L.-S. Wang and X. Li, J. Chem. Phys. **112**, 3602 (2000).

cyc-NbC₃

\tilde{C}						
$T^a=12620(230)$		C_{2v}	gas PE ¹			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1			840(50)	gas	PE	1
\tilde{B}						
$T^a=9710(260)$		C_{2v}	gas PE ¹			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1			610(30)	gas	PE	1

^aFrom vertical electron detachment energy.

Reference

¹H.-J. Zhai, S.-R. Liu, X. Li, and L.-S. Wang, J. Chem. Phys. **115**, 5170 (2001).

cyc-CrC₃

\tilde{G}						
$T_0=14040(240)$		C_{2v}	gas PE ¹			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CCrC s-stretch	500(50)	gas	PE	1
\tilde{E}						
$T_0=10650(240)$		C_{2v}	gas PE ¹			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CCrC s-stretch	540(30)	gas	PE	1
\tilde{D}						
$T_0=7750(160)$		C_{2v}	gas PE ¹			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CCrC s-stretch	560(60)	gas	PE	1
\tilde{C}						
$T_0=3630(160)$		C_{2v}	gas PE ¹			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CCrC s-stretch	540(30)	gas	PE	1
\tilde{B}						
$T_0=2500(160)$		C_{2v}	gas PE ¹			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CCrC s-stretch	560(60)	gas	PE	1
\tilde{A}						
$T_0=1210(160)$		C_{2v}	gas PE ¹			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CCrC s-stretch	560(60)	gas	PE	1

Reference

¹L.-S. Wang and X. Li, J. Chem. Phys. **112**, 3602 (2000).

cyc-MnC₃

\tilde{X}						
		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CMnC s-stretch	490(20)	gas	PE	1

Reference

¹L.-S. Wang and X. Li, J. Chem. Phys. **112**, 3602 (2000).

cyc-FeC₃

\tilde{B}						
$T_0=5490(160)$		C_{2v}	gas PE ²			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CFeC s-stretch	500(30)	gas	PE	2

\tilde{A} C_{2v}
 $T_0=4760(160)$ gas PE^{1,2}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CFeC s-stretch	350(50)T	gas	PE	2

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CFeC s-stretch	480(40)	gas	PE	2

References

- ¹J. Fan, L. Lou, and L.-S. Wang, J. Chem. Phys. **102**, 2701 (1995).
²L.-S. Wang and X. Li, J. Chem. Phys. **112**, 3602 (2000).

cyc-CoC₃

\tilde{C} C_{2v}
 $T_0=8310T$ gas PE¹

\tilde{B} C_{2v}
 $T_0=2420T$ gas PE¹

\tilde{A} C_{2v}
 $T_0=1210T$ gas PE¹

Reference

- ¹L.-S. Wang and X. Li, J. Chem. Phys. **112**, 3602 (2000).

cyc-NiC₃

\tilde{A} C_{2v}
 $T_0=3230T$ gas PE¹

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CNiC s-stretch	480(60)	gas	PE	1

Reference

- ¹L.-S. Wang and X. Li, J. Chem. Phys. **112**, 3602 (2000).

Fe₂CO

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1897.7	Ar	IR	1,2
			1884	Kr	IR	1

References

- ¹C. H. F. Peden, S. F. Parker, P. H. Barrett, and R. G. Pearson, J. Phys. Chem. **87**, 2329 (1983).
²M. Zhou, G. V. Chertihin, and L. Andrews, J. Chem. Phys. **109**, 10893 (1998).

cyc-(ScN)₂

\tilde{X} D_{2h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b_{2u}			772.2	Ar	IR	1
b_{3u}			672.9	Ar	IR	1

Reference

- ¹G. V. Chertihin, L. Andrews, and C. W. Bauschlicher, Jr., J. Am. Chem. Soc. **120**, 3205 (1998).

cyc-(YN)₂

\tilde{X} D_{2h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b_{2u}			710.4	Ar	IR	1
b_{3u}			586.9	Ar	IR	1

Reference

- ¹G. V. Chertihin, W. D. Bare, and L. Andrews, J. Phys. Chem. A **102**, 3697 (1998).

cyc-(LaN)₂

\tilde{X} D_{2h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b_{2u}			652.3	Ar	IR	1
b_{3u}			529.9	Ar	IR	1

Reference

- ¹G. V. Chertihin, W. D. Bare, and L. Andrews, J. Phys. Chem. A **102**, 3697 (1998).

cyc-(TiN)₂

\tilde{X} D_{2h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b_{1u}			766.3	Ar	IR	1
b_{2u}			680.2	Ar	IR	1

Reference

¹G. P. Kushto, P. F. Souter, G. V. Chertihin, and L. Andrews, J. Chem. Phys. **110**, 9020 (1999).

cyc-(ZrN)₂

\tilde{X}		D _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _{1u}			771.2	Ar	IR	1
			757.2			
b _{2u}			675.2	Ar	IR	1

Reference

¹G. P. Kushto, P. F. Souter, G. V. Chertihin, and L. Andrews, J. Chem. Phys. **110**, 9020 (1999).

cyc-(HfN)₂

\tilde{X}		D _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _{1u}			775.7	Ar	IR	1
b _{2u}			682.7	Ar	IR	1

Reference

¹G. P. Kushto, P. F. Souter, G. V. Chertihin, and L. Andrews, J. Chem. Phys. **110**, 9020 (1999).

cyc-(CoN)₂

\tilde{X}		D _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _{2u}	5		711.1	N ₂	IR	1
b _{3u}	6		537.3	N ₂	IR	1

Reference

¹L. Andrews, A. Citra, G. V. Chertihin, W. D. Bare, and M. Neurock, J. Phys. Chem. A **102**, 2561 (1998).

cyc-(RhN)₂

\tilde{X}		D _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			748.1	Ar	IR	1
			756.7	N ₂	IR	1

Reference

¹A. Citra and L. Andrews, J. Phys. Chem. A **103**, 3410 (1999).

cyc-(NiN)₂

\tilde{X}		D _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _{2u}			710.1	Ar	IR	1
			723.3	N ₂	IR	1

Reference

¹L. Andrews, A. Citra, G. V. Chertihin, W. D. Bare, and M. Neurock, J. Phys. Chem. A **102**, 2561 (1998).

Pt₂NN

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			2132.3	Ar	IR	1
			491.9	Ar	IR	1

Reference

¹A. Citra, X. Wang, W. D. Bare, and L. Andrews, J. Phys. Chem. A **105**, 7799 (2001).

PtNNPt

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			716.8	Ne	IR	1
			672.1T	Ar	IR	1
			713.8	N ₂	IR	1

Reference

¹A. Citra, X. Wang, W. D. Bare, and L. Andrews, J. Phys. Chem. A **105**, 7799 (2001).

cyc-(CeN)₂

\tilde{X}		D _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _{2u}	5		542.8	Ar	IR	1
			526.2	N ₂	IR	1
b _{3u}	6		667.6	Ar	IR	1
			639.6	N ₂	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **102**, 10238 (1998).

cyc-(PrN)₂

\tilde{X}		D _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>b</i> _{2u}	5		544.9	Ar	IR	1
			531.1	N ₂	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **102**, 10238 (1998).

cyc-(NdN)₂

\tilde{X}		D _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>b</i> _{2u}	5		546.7	Ar	IR	1
			530.3	N ₂	IR	1
<i>b</i> _{3u}	6		679.2	Ar	IR	1
			657.6	N ₂	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **102**, 10238 (1998).

cyc-(SmN)₂

\tilde{X}		D _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>b</i> _{2u}	5		553.9	Ar	IR	1
			533.9	N ₂	IR	1
			526.7			
<i>b</i> _{3u}	6		686.7	Ar	IR	1
			666.5	N ₂	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **102**, 10238 (1998).

cyc-(GdN)₂

\tilde{X}		D _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>b</i> _{2u}	5		563.5	Ar	IR	1
			545.6	N ₂	IR	1
			538.5			
<i>b</i> _{3u}	6		690.2	Ar	IR	1
			688.3			
			686.8			
			668.5	N ₂	IR	1
			663.8			

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **102**, 10238 (1998).

cyc-(TbN)₂

\tilde{X}		D _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>b</i> _{2u}	5		569.3	Ar	IR	1
			567.8			
			550.9	N ₂	IR	1
<i>b</i> _{3u}	6		698.8	Ar	IR	1
			669.4	N ₂	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 1311 (1999).

cyc-(DyN)₂

\tilde{X}		D _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>b</i> _{3u}	6		701.9	Ar	IR	1
			678.2	N ₂	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 1311 (1999).

cyc-(HoN)₂

\tilde{X}		D _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>b</i> _{2u}	5		551.9	N ₂	IR	1
<i>b</i> _{3u}	6		701	Ar	IR	1
			678.6	N ₂	IR	1

Reference

¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **103**, 1311 (1999).

cyc-(ErN)₂

\tilde{X}		D _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _{3u}	6		678.3	N ₂	IR	1

Reference

¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **103**, 1311 (1999).

cyc-(TmN)₂

\tilde{X}		D _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _{2u}	5		566.5	N ₂	IR	1
			563.4			
b _{3u}	6		684.2	N ₂	IR	1

Reference

¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **103**, 1311 (1999).

cyc-(LuN)₂

\tilde{X}		D _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _{2u}	5		589.9	Ar	IR	1
			587.7			
			569.6	N ₂	IR	1
b _{3u}	6		703.9	Ar	IR	1
			701.0			
			675.2	N ₂	IR	1

Reference

¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **103**, 1311 (1999).

cyc-(ThN)₂

\tilde{X}		D _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _{2u}	5		719.0	Ar	IR	1
b _{3u}	6		605.6	Ar	IR	1

Reference

¹G. P. Kushto, P. F. Souter, and L. Andrews, *J. Chem. Phys.* **108**, 7121 (1998).

BNBB

$^5\Sigma^-$		C _{∞v}				
T ₀ = 7790(160) gas PE ^{1,2}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺			1879(60)	gas	PE	1,2
			1403(60)	gas	PE	2

$^3\Pi$		C _{∞v}				
T ₀ = 6660(110) gas PE ^{1,2}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺			1863(60)	gas	PE	1,2
			758(60)	gas	PE	1,2

$\tilde{X}^1\Sigma^+$ C_{∞v}

References

¹K. R. Asmis, T. R. Taylor, and D. M. Neumark, *Eur. Phys. J. D* **9**, 257 (1999).

²K. R. Asmis, T. R. Taylor, and D. M. Neumark, *J. Chem. Phys.* **111**, 10491 (1999).

BCCB

\tilde{X}		D _{∞h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _u ⁺	3	BC a-stretch	955.0	Ar	IR	1

Reference

¹J. D. Presilla-Márquez, P. G. Carrick, and C. W. Larson, *J. Chem. Phys.* **110**, 5702 (1999).

AICCAI

\tilde{C}
T^a = 15490(830) gas PE²

\tilde{B}
T^a = 12420(630) gas PE²

\tilde{A}
T^a = 10730(470) gas PE²

\tilde{X}		D _{∞h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _u ⁺	3	AIC a-stretch	605.1	Ar	IR	1

^aFrom vertical electron detachment energies.

References

- ¹G. V. Chertihin, L. Andrews, and P. R. Taylor, *J. Am. Chem. Soc.* **116**, 3513 (1994).
²N. A. Cannon, A. I. Boldyrev, X. Li, and L.-S. Wang, *J. Chem. Phys.* **113**, 2671 (2000).

CAI₃⁻

Threshold for electron detachment from ground-state CAI₃⁻ = 20650(160) gas PE¹

Reference

- ¹A. I. Boldyrev, J. Simons, X. Li, W. Chen, and L.-S. Wang, *J. Chem. Phys.* **110**, 8980 (1999).

cyc-SiAl₃⁻

Vertical electron detachment energy from ground-state SiAl₃⁻ = 22030(640) gas PE¹

Reference

- ¹X. Li, H.-F. Zhang, L.-S. Wang, A. E. Kuznetsov, N. A. Cannon, and A. I. Boldyrev, *Angew. Chem. Int. Ed.* **40**, 1867 (2001).

cyc-GeAl₃⁻

Vertical electron detachment energy from ground-state GeAl₃⁻ = 21780(730) gas PE¹

Reference

- ¹X. Li, H.-F. Zhang, L.-S. Wang, A. E. Kuznetsov, N. A. Cannon, and A. I. Boldyrev, *Angew. Chem. Int. Ed.* **40**, 1867 (2001).

cyc-SnAl₃⁻

Vertical electron detachment energy from ground-state SnAl₃⁻ = 21460(320) gas PE¹

Reference

- ¹X. Li, H.-F. Zhang, L.-S. Wang, A. E. Kuznetsov, N. A. Cannon, and A. I. Boldyrev, *Angew. Chem. Int. Ed.* **40**, 1867 (2001).

cyc-PbAl₃⁻

Vertical electron detachment energy from ground-state PbAl₃⁻ = 20040(160) gas PE¹

Reference

- ¹X. Li, H.-F. Zhang, L.-S. Wang, A. E. Kuznetsov, N. A. Cannon, and A. I. Boldyrev, *Angew. Chem. Int. Ed.* **40**, 1867 (2001).

NAI₃

\tilde{a} C_{2v}
 $T^a = 14680(690)$ gas PE¹

\tilde{X} D_{3h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>e'</i>	3	NAI ₃ a-stretch	773.1 777.9 770.3	Ar N ₂	IR IR	2 2

^aFrom vertical electron detachment energy.

References

- ¹S. K. Nayak, B. K. Rao, P. Jena, X. Li, and L.-S. Wang, *Chem. Phys. Lett.* **301**, 379 (1999).
²L. Andrews, M. Zhou, G. V. Chertihin, W. D. Bare, and Y. Hannachi, *J. Phys. Chem. A* **104**, 1656 (2000).

NGa₃

\tilde{X} D_{3h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>e'</i>	3	NGa ₃ a-stretch	666.2 656.0	N ₂	IR	1

Reference

- ¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **104**, 1648 (2000).

NIn₃

\tilde{X} D_{3h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>e'</i>	3	NIn ₃ a-stretch	604.7 596.3	N ₂	IR	1

Reference

- ¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **104**, 1648 (2000).

NTI₃

\tilde{X} D_{3h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>e'</i>	3	NTI ₃ a-stretch	551.9 544.2	N ₂	IR	1

Reference

¹M. Zhou and L. Andrews, J. Phys. Chem. A **104**, 1648 (2000).

cyc-ScC₃⁻

Threshold for electron detachment from ground-state *cyc*-ScC₃⁻ = 13230(160) gas PE¹

Reference

¹L.-S. Wang and X. Li, J. Chem. Phys. **112**, 3602 (2000).

cyc-TiC₃⁻

Threshold for electron detachment from ground-state *cyc*-TiC₃⁻ = 12590(120) gas PE¹

Reference

¹X.-B. Wang, C.-F. Ding, and L.-S. Wang, J. Phys. Chem. A **101**, 7699 (1997).

cyc-VC₃⁻

Threshold for electron detachment from ground-state *cyc*-VC₃⁻ = 11860(160) gas PE¹

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁		CVC s-stretch	420(50)	gas	PE	1

Reference

¹L.-S. Wang and X. Li, J. Chem. Phys. **112**, 3602 (2000).

cyc-NbC₃⁻

Threshold for electron detachment from ground-state *cyc*-NbC₃⁻ = 13600(160) gas PE¹

Reference

¹H.-J. Zhai, S.-R. Liu, X. Li, and L.-S. Wang, J. Chem. Phys. **115**, 5170 (2001).

cyc-CrC₃⁻

Threshold for electron detachment from ground-state *cyc*-CrC₃⁻ = 12100(160) gas PE¹

Reference

¹L.-S. Wang and X. Li, J. Chem. Phys. **112**, 3602 (2000).

cyc-MnC₃⁻

Threshold for electron detachment from ground-state *cyc*-MnC₃⁻ = 15090(160) gas PE¹

Reference

¹L.-S. Wang and X. Li, J. Chem. Phys. **112**, 3602 (2000).

cyc-FeC₃⁻

Threshold for electron detachment from ground-state *cyc*-FeC₃⁻ = 13550(160) gas PE^{1,2}

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁		CFeC s-stretch	330(40)	gas	PE	2

References

¹J. Fan, L. Lou, and L.-S. Wang, J. Chem. Phys. **102**, 2701 (1995).

²L.-S. Wang and X. Li, J. Chem. Phys. **112**, 3602 (2000).

cyc-CoC₃⁻

Threshold for electron detachment from ground-state *cyc*-CoC₃⁻ = 12510(480) gas PE¹

Reference

¹L.-S. Wang and X. Li, J. Chem. Phys. **112**, 3602 (2000).

cyc-NiC₃⁻

Threshold for electron detachment from ground-state *cyc*-NiC₃⁻ = 11210(400) gas PE¹

Reference

¹L.-S. Wang and X. Li, J. Chem. Phys. **112**, 3602 (2000).

BNBB⁻

Threshold for electron detachment from ground-state BNBB⁻ = 16930(280) gas PE¹

²Π C_{∞v}
T₀ = 1220(80) gas PE¹

$\tilde{X}^4\Sigma^-$ C_{∞v}

Reference

¹K. R. Asmis, T. R. Taylor, and D. M. Neumark, J. Chem. Phys. **111**, 10491 (1999).

Al₃N⁻

Threshold for electron detachment from ground-state Al₃N⁻ = 7750(650) gas PE¹

Reference

¹S. K. Nayak, B. K. Rao, P. Jena, X. Li, and L.-S. Wang, Chem. Phys. Lett. **301**, 379 (1999).

BCCC

\tilde{X} $C_{\infty v}$						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	C ₃ stretch	2002.1T	Ar	IR	1
	2	BC stretch	1512.5	Ar	IR	1

Reference

¹J. D. Presilla-Márquez, P. G. Carrick, and C. W. Larson, J. Chem. Phys. **110**, 5702 (1999).

CAISi₂

\tilde{B}
T^a=8550(870) gas PE¹

\tilde{A}
T^a=4030(460) gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			400(60)T	gas	PE	1

\tilde{X}^2A_1 C_{2v}

^aFrom vertical electron detachment energies.

Reference

¹A. I. Boldyrev, X. Li, and L.-S. Wang, J. Phys. Chem. A **104**, 5358 (2000).

AICCAI⁻

Threshold for electron detachment from ground-state quasilinear AICCAI⁻ = 5160(400) gas PE¹

Reference

¹N. A. Cannon, A. I. Boldyrev, X. Li, and L.-S. Wang, J. Chem. Phys. **113**, 2671 (2000).

cyc-Al₂C₂⁻

Vertical detachment energy from ground-state cyc-Al₂C₂⁻ to form quasilinear AICCAI⁻ = 12750(400) gas PE¹

Reference

¹N. A. Cannon, A. I. Boldyrev, X. Li, and L.-S. Wang, J. Chem. Phys. **113**, 2671 (2000).

Al₃O

\tilde{C}^2A_1 C_{2v}
T₀ = 20250(800) gas PE¹

\tilde{B}^2B_2 C_{2v}
T₀ = 15400(800) gas PE¹MO²

\tilde{A}^2A_1 C_{2v}
T₀ = 11130(700) gas PE¹MO²

\tilde{X}^2B_2 C_{2v}

References

¹H. Wu, X. Li, X.-B. Wang, C.-F. Ding, and L.-S. Wang, J. Chem. Phys. **109**, 449 (1998).

²T. K. Ghanty and E. R. Davidson, J. Phys. Chem. A **103**, 2867 (1999).

ScOScO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			866.0	Ar	IR	1

Reference

¹G. V. Chertihin, L. Andrews, M. Rosi, and C. W. Bauschlicher, Jr., J. Phys. Chem. A **101**, 9085 (1997).

cyc-(ScO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			699.1	Ar	IR	1
			688.1	N ₂	IR	1
			647.8	Ar	IR	1
			635.5	N ₂	IR	1

Reference

¹G. V. Chertihin, L. Andrews, M. Rosi, and C. W. Bauschlicher, Jr., J. Phys. Chem. A **101**, 9085 (1997).

VOVO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		-OVO a-stretch	1042.4	Ar	IR	1

Reference

¹G. V. Chertihin, W. D. Bare, and L. Andrews, J. Phys. Chem. A **101**, 5090 (1997).

cyc-(VO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			668.1	Ar	IR	1
			504.3	Ar	IR	1

Reference

¹G. V. Chertihin, W. D. Bare, and L. Andrews, *J. Phys. Chem. A* **101**, 5090 (1997).

NbONbO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			997.6	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **102**, 8251 (1998).

cyc-(TaO)₂

\tilde{X} D _{2h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		TaO stretch	689.1	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **102**, 8251 (1998).

CrOCrO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			996.6	Ne	IR	2
			984.3	Ar	IR	1

References

¹G. V. Chertihin, W. D. Bare, and L. Andrews, *J. Chem. Phys.* **107**, 2798 (1997).

²M. Zhou and L. Andrews, *J. Chem. Phys.* **111**, 4230 (1999).

cyc-(MnO)₂

\tilde{X} D _{2h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			601.0	Ar	IR	1
			603.2	N ₂	IR	1
			506.8	Ar	IR	1

Reference

¹G. V. Chertihin and L. Andrews, *J. Phys. Chem. A* **101**, 8547 (1997).

MnOMnO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			847.3	Ar	IR	1

Reference

¹G. V. Chertihin and L. Andrews, *J. Phys. Chem. A* **101**, 8547 (1997).

cyc-(FeO)₂

\tilde{A}						
$T_0 = 3550(400)$ gas PE ⁶						
\tilde{X} D _{2h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b_{2u}			670(70)	gas	PE	6
			660.6	Ar	IR	3,4,7
			657.3	N ₂	IR	5
b_{3u}			517.4	Ar	IR	1-4
			535.5	N ₂	IR	5

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³L. Andrews, G. V. Chertihin, A. Ricca, and C. W. Bauschlicher, Jr., *J. Am. Chem. Soc.* **118**, 467 (1996).

⁴G. V. Chertihin, W. Saffel, J. T. Yustein, L. Andrews, M. Neurock, A. Ricca, and C. W. Bauschlicher, Jr., *J. Phys. Chem.* **100**, 5261 (1996).

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⁶H. Wu, S. R. Desai, and L.-S. Wang, *J. Am. Chem. Soc.* **118**, 5296 (1996).

⁷G. V. Chertihin, A. Citra, L. Andrews, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **101**, 8793 (1997).

cyc-(CoO)₂

\tilde{X}						
D _{2h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>b</i> _{2u}			685.2	Ar	IR	1
			706.0	N ₂	IR	1
<i>b</i> _{3u}			469.6	Ar	IR	1
			494.6	N ₂	IR	1

Reference

¹G. V. Chertihin, A. Citra, L. Andrews, and C. W. Bauschlicher, Jr., J. Phys. Chem. A **101**, 8793 (1997).

CoOCoO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			944.0	Ar	IR	1

Reference

¹G. V. Chertihin, A. Citra, L. Andrews, and C. W. Bauschlicher, Jr., J. Phys. Chem. A **101**, 8793 (1997).

cyc-(RhO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			562.4T	Ar	IR	1

Reference

¹A. Citra and L. Andrews, J. Phys. Chem. A **103**, 4845 (1999).

cyc-(PtO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			550.9	Ar	IR	1

Reference

¹W. D. Bare, A. Citra, G. V. Chertihin, and L. Andrews, J. Phys. Chem. A **103**, 5456 (1999).

cyc-(CeO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CeO stretch	575.0	Ar	IR	1
		CeO stretch	483.6	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

cyc-(PrO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		PrO stretch	713	Ne	IR	1
			699.2	Ar	IR	1
		PrO stretch	605.8	Ar	IR	1
			593.7			

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

cyc-(NdO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NdO stretch	600.0	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

cyc-(SmO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		SmO stretch	564.4	Ar	IR	1
		SmO stretch	422.6	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

cyc-(EuO)₂

\tilde{X}						
D _{2h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>b</i> _{2u}			530.5	Ar	IR	1
<i>b</i> _{3u}			444.6	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

cyc-(TbO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			607.2	Ar	IR	1
			532.7	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 6972 (1999).

cyc-(DyO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			609.6	Ar	IR	1
			522.2	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 6972 (1999).

cyc-(TmO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			553.7	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 6972 (1999).

cyc-(YbO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			590.9T	Ar	IR	1
			540.1T	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 6972 (1999).

cyc-(LuO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			569.6T	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 6972 (1999).

cis-CAISi₂⁻

Vertical electron detachment energy from ground-state *cis*-CAISi₂⁻ = 20490(320) gas PE¹

Reference

¹A. I. Boldyrev, X. Li, and L.-S. Wang, J. Phys. Chem. A **104**, 5358 (2000).

Al₂N₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		AlN stretch	651.4	Ar	IR	1
			648.2	N ₂	IR	1

Reference

¹L. Andrews, M. Zhou, G. V. Chertihin, W. D. Bare, and Y. Hannachi, J. Phys. Chem. A **104**, 1656 (2000).

Al₂P₂

\tilde{b}						
C _{2v}						
T ^a = 10250(830) gas PE ¹						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	PP s-stretch	420(15)	gas	PE	1

\tilde{a}^3A_2		C_{2v}				
$T^a=4680(830)$		gas		PE ¹		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	AIP s-stretch	320(12)	gas	PE	1

\tilde{X}^1A_g D_{2h}

^aFrom vertical ionization potential.

Reference

¹H. Gómez, T. R. Taylor, and D. M. Neumark, *J. Phys. Chem. A* **105**, 6886 (2001).

Al₃O⁻

Threshold for electron detachment from ground-state Al₃O⁻ = 12670(480) gas PE¹MO²

References

¹H. Wu, X. Li, X.-B. Wang, C.-F. Ding, and L.-S. Wang, *J. Chem. Phys.* **109**, 449 (1998).
²T. K. Ghanty and E. R. Davidson, *J. Phys. Chem. A* **103**, 2867 (1999).

C₄

$^3\Sigma_u^-$ $D_{\infty h}$
 $T_0=26384.9(2)$ gas CR¹⁵
 26323(15) Ne AB¹¹ $^3\Sigma_u^- - \tilde{X}$ 334–379 nm
 $^3\Sigma_u^- - \tilde{X}$ 325–380 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1821	gas	CR	15
			1822(20)	Ne	AB	11
			869	gas	CR	15
Π_g	2	C–C stretch	903(20)	Ne	AB	11
			370(20)H	Ne	AB	11
Π_u	5	Bend	175(20)H	Ne	AB	11

$B_0=0.157$ CR¹⁵

$^1\Pi_g$ $D_{\infty h}$
 $T_0=11380$ gas PE¹⁴

$^1\Pi_u$ $D_{\infty h}$
 $T_0=9360(160)T$ gas PE¹⁴

$^3\Pi_u$ $D_{\infty h}$
 $T_0=7500(160)$ gas PE¹⁴

$^3\Pi_g$ $D_{\infty h}$
 $T_0=6620(160)$ gas PE¹⁴

$^1\Sigma_g^+$ $D_{\infty h}$
 $T_0=4030(160)T$ gas PE¹⁴

$^1\Delta_g$ $D_{\infty h}$
 $T_0=2680(120)$ gas PE^{6,14}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	2032(50)	gas	PE	14
Π_g	4	Bend	331(50)	gas	PE	14

$\tilde{X}^3\Sigma_g^-$		$D_{\infty h}$		Structure: ESR ^{2,5} DL ^{3,7}		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	2057(50)	gas	PE	6,14
			1548.61	gas	DL	3
			1547.0	Ne	IR	8,10,13
Σ_u^+	3	Asym. stretch	1543.4	Ar	IR	1
			1539.5	Kr	IR	12
			323(50)	gas	PE,DL	6,9,14
Π_g	4	Bend	160(4) ^b	gas	DL	7
Π_u	5	Bend	172.4	Ar	IR	4
			170.4			

$B_0=0.166$ DL^{4,8}

^aESR measurements^{2,5} suggest that C₄ may deviate by a few degrees from the linear structure.

^bEstimated from *l*-doubling parameter for ν_5 .

References

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¹⁴C. Xu, G. R. Burton, T. R. Taylor, and D. M. Neumark, *J. Chem. Phys.* **107**, 3428 (1997).
¹⁵H. Linnartz, O. Vaizert, T. Motylewski, and J. P. Maier, *J. Chem. Phys.* **112**, 9777 (2000).

SiC₃

\tilde{a} $C_{\infty v}$
 $T_0=2210(25)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π		Deformation	140(25)	gas	PE	1

$\tilde{X}^3\Sigma^-$ $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	C ₃ a-stretch	1980(20)	gas	PE	1
		C ₃ s-stretch	1320(20)	gas	PE	1
Π	5	Deformation	130T	gas	PE	1

Reference

¹G. E. Davico, R. L. Schwartz, and W. C. Lineberger, *J. Chem. Phys.* **115**, 1789 (2001).

cyc-SiC₃ (prolate)

\tilde{X} C_{2v} Structure: MW^{1,2}
A = 1.266; B = 0.210; C = 0.180 MW^{1,2}

References

¹M. C. McCarthy, A. J. Apponi, and P. Thaddeus, *J. Chem. Phys.* **110**, 10645 (1999).

²A. J. Apponi, M. C. McCarthy, C. A. Gottlieb, and P. Thaddeus, *J. Chem. Phys.* **111**, 3911 (1999).

cyc-SiC₃ (oblate)

\tilde{X} C_{2v} Structure: MW¹
A = 0.416; B = 0.378; C = 0.198 MW¹

Reference

¹M. C. McCarthy, A. J. Apponi, and P. Thaddeus, *J. Chem. Phys.* **111**, 7175 (1999).

Si₄

¹B_{1u} D_{2h}
T₀ = 23635T^a gas EM³ ¹B_{1u}- \tilde{X} 418–448 nm
21432(9) Ne AB^{1,7} ¹B_{1u}- \tilde{X} 410–470 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a _g	1		431(14)	Ne	AB	7
	2		310(25)	gas	EM	3
			317(14)	Ne	AB	1,7

³B_{1u} D_{2h}
T₀ = 15570(160) gas PE^{2,8}TPE⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a _g	1		450(20)	gas	PE,TPE	2,4,8

³B_{1g} D_{2h}
T₀ = 13800(160) gas PE⁸

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a _g	1		355(20)	gas	PE	8

³B_g C_{2h}
T₀ = 12100T gas PE⁸

¹B_{3u} D_{2h}
T₀ = 11050(80) gas PE^{2,8}TPE⁴
10807(2) Ne AB⁷

¹B_{3u}- \tilde{X} 790–930 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a _g			300(8)	gas	TPE,PE	4,8
			300(6)	Ne	AB	7

³B_{3u} D_{2h}
T₀ = 6580(80) gas PE^{2,8}TPE⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a _g	2		312	gas	PE,TPE	2,4,8

\tilde{X} ¹A_g D_{2h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a _g	1		470	N ₂	Ra	5
	2		380(20)	gas	PE	2,8
			345	N ₂	Ra	5
b _{1u}	4		501.9	Ne	IR	6
			500.9	Ar	IR	6
			498.5	Kr	IR	6

^aAttributed by Refs. 1 and 3 to Si₃.

References

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⁸C. Xu, T. R. Taylor, G. R. Burton, and D. M. Neumark, *J. Chem. Phys.* **108**, 1395 (1998).

PtNNN

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		N ₃ a-stretch	2104.4	N ₂	IR	1
		N ₃ s-stretch	1389.5	N ₂	IR	1

Reference

¹A. Citra, X. Wang, W. D. Bare, and L. Andrews, *J. Phys. Chem. A* **105**, 7799 (2001).

NNPtN \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NN stretch	2280.0	N ₂	IR	1
		PtN stretch	854.7	N ₂	IR	1

Reference

¹A. Citra, X. Wang, W. D. Bare, and L. Andrews, *J. Phys. Chem. A* **105**, 7799 (2001).

OScCO⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2221.8	Ar	IR	1
		ScO stretch	965.5	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **120**, 13230 (1998).

OScOC⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	OC stretch	2068.1	Ar	IR	1
		OSc stretch	974.1	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **120**, 13230 (1998).

ScOCO⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2105.7	Ar	IR	1
		ScO stretch	976.4	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **120**, 13230 (1998).

OYCO⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2206.0	Ar	IR	1
			2203.3			
		YO stretch	857.2	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **120**, 13230 (1998).

OYOC⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	OC stretch	2078.0	Ar	IR	1
		OY stretch	869.1	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **120**, 13230 (1998).

YOCO⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2106.0	Ar	IR	1
		YO stretch	871.9	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **120**, 13230 (1998).

OTiCO⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2190.4	Ar	IR	1
			2185.5			
		TiO stretch	1004.3	Ar	IR	1
			1003.0			

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 2066 (1999).

OTiOC⁺

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2065.2	Ar	IR	1
		TiO stretch	1012.9	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 2066 (1999).

OVCO⁺

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2205.4	Ar	IR	1
			2203.3			
		VO stretch	1020.0	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 2066 (1999).

OVOC⁺

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2059.3	Ar	IR	1
		VO stretch	1029.6	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 2066 (1999).

OCrCO⁺

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2175.5	Ar	IR	1

Reference

¹M. Zhou, B. Liang, and L. Andrews, *J. Phys. Chem. A* **103**, 2013 (1999).

OMnCO⁺

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2173.0	Ar	IR	1
		MnO stretch	851.9	Ar	IR	1

Reference

¹M. Zhou, B. Liang, and L. Andrews, *J. Phys. Chem. A* **103**, 2013 (1999).

OThCO⁺

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2009.9	Ne	IR	1
		ThO stretch	896.4	Ne	IR	1

Reference

¹L. Andrews, M. Zhou, B. Liang, J. Li, and B. E. Bursten, *J. Am. Chem. Soc.* **122**, 11440 (2000).

OUCO⁺

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2073.0	Ne	IR	1
		UO stretch	881.2	Ne	IR	1

Reference

¹L. Andrews, M. Zhou, B. Liang, J. Li, and B. E. Bursten, *J. Am. Chem. Soc.* **122**, 11440 (2000).

Al₂P₂⁻

Threshold for electron detachment from ground-state Al₂P₂⁻ = 17350(800) gas PE¹

Reference

¹H. Gómez, T. R. Taylor, and D. M. Neumark, *J. Phys. Chem. A* **105**, 6886 (2001).

C₄⁻

Threshold for electron detachment from ground-state C₄⁻ = 31320(80) gas PE^{1,2,7}

(3) ²Π_u D_{∞h}
 T₀ = 28717 gas MPD¹⁰ (3) ²Π_u- \tilde{X} 336–348 nm
 28868 Ne AB⁶ (3) ²Π_u- \tilde{X} 335–347 nm

(2) ${}^2\Pi_u$ $D_{\infty h}$
 $T_0 = 25989$ gas MPD¹⁰ (2) ${}^2\Pi_u-\tilde{X}$ 347–385 nm
 26069(14) Ne AB⁶ (2) ${}^2\Pi_u-\tilde{X}$ 372–384 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1		2056	gas	MPD	10
	2		710	gas	MPD	10
			755(20)	Ne	AB	6
Π_g	4		374H	gas	MPD	10
			349(10)H	Ne	AB	6
Π_u	5		268H	gas	MPD	10
			271(10)H	Ne	AB	6

$\tilde{C} {}^2\Pi_u$ $D_{\infty h}$
 $T_0 = 21871.53$ gas PD⁴MPD¹⁰ $\tilde{C}-\tilde{X}$ 379–460 nm
 21896(5) Ne AB³LF⁵ $\tilde{C}-\tilde{X}$ 410–527 nm
 21720 Ar AB⁸ $\tilde{C}-\tilde{X}$ 416–461 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1		1729H	gas	MPD	10
	2		752.4	gas	PD	4
			759(5)	Ne	AB	3
			763	Ar	AB	8
Π_u	5	Deformation	223H	gas	PD	4

$A_0 = -37(9)$ gas PD⁴
 $B_0 = 0.156$ PD⁴

$\tilde{B} {}^2\Sigma_u^+$ $D_{\infty h}$
 $T_0 = 10789(2)$ Ne AB⁶ $\tilde{B}-\tilde{X}$ 672–927 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1		2056(5)	Ne	AB	6
	2		930(4)	Ne	AB	6
Π_u	5		259(2)H	Ne	AB	6

$\tilde{A} {}^2\Sigma_g^+$ $D_{\infty h}$
 Ne AB⁶ $\tilde{A}-\tilde{X}$ 907–1206 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	2		817(3)	Ne	AB	6
Π_u	5		275(2)	Ne	AB	6

$\tilde{X} {}^2\Pi_g$ $D_{\infty h}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1		2047(20)	Ne	LF	5
	2		936(20)	Ne	LF	5
Σ_u^+	3	Asym. stretch	1699.8	Ar	IR	9
Π_g	4		396(20)H	Ne	LF	5

$A_0 = -39(9)$ gas PD⁴
 $B_0 = 0.167$ PD⁴

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- ¹⁰M. Tulej, D. A. Kirkwood, G. Maccaferri, O. Dopfer, and J. P. Maier, Chem. Phys. **228**, 293 (1998).

SiC₃⁻

Threshold for electron detachment from ground-state linear SiC₃⁻ = 22810(60) gas PE¹

Reference

- ¹G. E. Davico, R. L. Schwartz, and W. C. Lineberger, J. Chem. Phys. **115**, 1789 (2001).

Si₄⁻

Threshold for electron detachment from ground-state Si₄⁻ = 17180(80) gas PE^{1,4}TPE²

${}^2B_{1u}$ D_{2h} ${}^2B_{1u}-\tilde{X}$ 780–875 nm
 $T_0 = 11460(3)$ Ne AB³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_g	1		352(4)	Ne	AB	3
	2		278(4)	Ne	AB	3

$\tilde{X} {}^2B_{2g}$ D_{2h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_g	2		365	gas	TPE	2

References

- ¹T. N. Kitsopoulos, C. J. Chick, A. Weaver, and D. M. Neumark, J. Chem. Phys. **93**, 6108 (1990).

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Ge₄⁻

Threshold for electron detachment from ground-state Ge₄⁻ = 15650(80) gas PE¹

Reference

¹G. R. Burton, C. Xu, C. C. Arnold, and D. M. Neumark, J. Chem. Phys. **104**, 2757 (1996).

Sn₄⁻

Threshold for electron detachment from ground-state Sn₄⁻ = 16460(80) gas PE¹

Reference

¹V. D. Moravec, S. A. Klopčič, and C. C. Jarrold, J. Chem. Phys. **110**, 5079 (1999).

OScCO

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CO stretch	1873.4	Ar	IR	1
		ScO stretch	894.1	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, J. Am. Chem. Soc. **120**, 13230 (1998).

cyc-(ScOC)O

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OCO a-stretch	1763.4	Ar	IR	1
		Ring stretch	778.0	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, J. Am. Chem. Soc. **120**, 13230 (1998).

cyc-(COSc)O

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CO stretch	1613.9	Ar	IR	1
		ScO stretch	909.6	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, J. Am. Chem. Soc. **120**, 13230 (1998).

OYCO

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CO stretch	1861.5	Ar	IR	1
		YO stretch	796.5	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, J. Am. Chem. Soc. **120**, 13230 (1998).

cyc-(YOC)O

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OCO a-stretch	1773.1	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, J. Am. Chem. Soc. **120**, 13230 (1998).

cyc-(COY)O

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CO stretch	1614.5	Ar	IR	1
		YO stretch	804.4	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, J. Am. Chem. Soc. **120**, 13230 (1998).

OTiCO

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1884.0	Ne	IR	2
			1866.7	Ar	IR	1,2
		TiO stretch	973.1	Ne	IR	2
			952.8	Ar	IR	1,2

References

¹G. V. Chertihin and L. Andrews, J. Am. Chem. Soc. **117**, 1595 (1995).
²M. Zhou and L. Andrews, J. Phys. Chem. A **103**, 2066 (1999).

cyc-(COTi)O

\bar{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1539.0	Ne	IR	1
			1532.3	Ar	IR	1
		TiO stretch	990.6	Ne	IR	1
			965.0	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, J. Phys. Chem. A **103**, 2066 (1999).

OZrCO

\bar{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CO stretch	1837.8	Ar	IR	1
	2	ZrO stretch	878.9	Ar	IR	1

Reference

¹L. Zhang, X. Wang, M. Chen, and Q.-Z. Qin, Chem. Phys. **254**, 231 (2000).

OVCO

\bar{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1881.1	Ar	IR	1
		VO stretch	974.8	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, J. Phys. Chem. A **103**, 2066 (1999).

cyc-(COV)O

\bar{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1516.1	Ar	IR	1
		VO stretch	990.0	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, J. Phys. Chem. A **103**, 2066 (1999).

ONbCO (I)

\bar{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CO stretch	1902.6	Ar	IR	1
	2	NbO stretch	960.1	Ar	IR	1

Reference

¹M. Chen, X. Wang, L. Zhang, and Q. Qin, J. Phys. Chem. A **104**, 7010 (2000).

ONbCO (II)

\bar{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CO stretch	1821.6	Ar	IR	1
	2	NbO stretch	915.6	Ar	IR	1

Reference

¹M. Chen, X. Wang, L. Zhang, and Q. Qin, J. Phys. Chem. A **104**, 7010 (2000).

OTaCO

\bar{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CO stretch	1887.3	Ar	IR	1
	2	TaO stretch	967.3	Ar	IR	1

Reference

¹X.-F. Wang, M.-H. Chen, L.-N. Zhang, and Q.-Z. Qin, J. Phys. Chem. A **104**, 758 (2000).

OCrCO

\bar{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CO stretch	2014.4	Ar	IR	1
	2	CrO stretch	866.3	Ar	IR	1

Reference

¹P. F. Souter and L. Andrews, J. Am. Chem. Soc. **119**, 7350 (1997).

CrOCO \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OCO a-stretch	1735.6	Ar	IR	1
		CrO stretch	721.0 716.1	Ar	IR	1

Reference¹P. F. Souter and L. Andrews, J. Am. Chem. Soc. **119**, 7350 (1997).**OMoCO** \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CO stretch	1847.0	Ar	IR	1
	2	MoO stretch	951.8	Ar	IR	1

Reference¹P. F. Souter and L. Andrews, J. Am. Chem. Soc. **119**, 7350 (1997).**OWCO** \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CO stretch	1879.0	Ar	IR	1
	2	WO stretch	969.6 483.8	Ar	IR	1

Reference¹P. F. Souter and L. Andrews, J. Am. Chem. Soc. **119**, 7350 (1997).**OMnCO** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2082.5	Ar	IR	1
		MnO stretch	869.9	Ar	IR	1

Reference¹M. Zhou, B. Liang, and L. Andrews, J. Phys. Chem. A **103**, 2013 (1999).**OFeCO** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2037.1	Ar	IR	1
		FeO stretch	872.8	Ar	IR	1

Reference¹M. Zhou, B. Liang, and L. Andrews, J. Phys. Chem. A **103**, 2013 (1999).**OCoCO** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2026.6	Ar	IR	1
		CoO stretch	783.8	Ar	IR	1

Reference¹M. Zhou, B. Liang, and L. Andrews, J. Phys. Chem. A **103**, 2013 (1999).**ONiCO** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2086.6	Ar	IR	1

Reference¹M. Zhou, B. Liang, and L. Andrews, J. Phys. Chem. A **103**, 2013 (1999).**OThCO** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	1778.4	Ne	IR	1

Reference¹L. Andrews, M. Zhou, B. Liang, J. Li, and B. E. Bursten, J. Am. Chem. Soc. **122**, 11440 (2000).

OUCO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
1		CO stretch	1806.9	Ne	IR	2
			1799.6	Ar	IR	1
2		UO stretch	823.2	Ne	IR	2
			804.4	Ar	IR	1

References

¹T. J. Tague, Jr., L. Andrews, and R. D. Hunt, *J. Phys. Chem.* **97**, 10920 (1993).

²L. Andrews, M. Zhou, B. Liang, J. Li, and B. E. Bursten, *J. Am. Chem. Soc.* **122**, 11440 (2000).

NNBN

\tilde{A}		$\tilde{A}-\tilde{X}$ 348–380 nm				
$T_0=26350$ Ar AB ⁴						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Deformation	480	Ar	AB	4

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
1		NN stretch	2091.7	Ar	IR	2
			2100(2) N ₂	Ar	IR	3
			2124.8	N ₂	IR	1
2		BN stretch	1802.0	Ar	IR	2
			1803(2) N ₂	Ar	IR	3
			1806.1	N ₂	IR	1
3		N ₂ -BN stretch	760.3w	Ar	IR	2
			750.4	N ₂	IR	1

References

¹P. Hassanzadeh and L. Andrews, *J. Phys. Chem.* **96**, 9177 (1992).

²L. Andrews, P. Hassanzadeh, T. R. Burkholder, and J. M. L. Martin, *J. Chem. Phys.* **98**, 922 (1993).

³I. A. Al-Jihad, B. Liu, C. J. Linnen, and J. V. Gilbert, *J. Phys. Chem. A* **102**, 6220 (1998).

⁴M. J. Travers and J. V. Gilbert, *J. Phys. Chem. A* **104**, 3780 (2000).

AINNN

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	N ₃ a-stretch	2144.0	N ₂	IR	1
	2	N ₃ s-stretch	1386.0	N ₂	IR	1
	3	AIN stretch	509.7	N ₂	IR	1

Reference

¹L. Andrews, M. Zhou, G. V. Chertihin, W. D. Bare, and Y. Hannachi, *J. Phys. Chem. A* **104**, 1656 (2000).

GaNNN

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	N ₃ a-stretch	2096.8	N ₂	IR	1
	2	N ₃ s-stretch	1328.3	N ₂	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **104**, 1648 (2000).

InNNN

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	N ₃ a-stretch	2074.5	N ₂	IR	1
	2	N ₃ s-stretch	1323.9	N ₂	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **104**, 1648 (2000).

TINNN

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	N ₃ a-stretch	2048.8	N ₂	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **104**, 1648 (2000).

TIOTIO

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		TIO stretch	726.9	Ar	IR	1
		TIO stretch	655.9	Ar	IR	1

Reference

¹L. Andrews, G. P. Kushto, J. T. Yuste, E. Archibong, R. Sullivan, and J. Leszczynski, *J. Phys. Chem. A* **101**, 9077 (1997).

cyc-TiO₂Ti

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			422	Ar	IR	1,2
			422.7	N ₂	IR	1

References

- ¹B. J. Kelsall and K. D. Carlson, *J. Phys. Chem.* **84**, 951 (1980).
²L. Andrews, G. P. Kushto, J. T. Yustein, E. Archibong, R. Sullivan, and J. Leszczynski, *J. Phys. Chem. A* **101**, 9077 (1997).

NCCP

\tilde{X}			Structure: MW ¹			
$B_0=0.090$	$C_{\infty v}$	MW ¹				

Reference

- ¹L. Bizzocchi, C. Degli Esposti, and P. Botschwina, *J. Chem. Phys.* **113**, 1465 (2000).

CCCO

\tilde{X}						
$C_{\infty v}$		Structure: MW ^{4,7}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1		2257.22	gas	IR	8
			2243vs	Ar	IR	1,5-7,10
			2237.2	Xe	IR	9
	2		1907.0m	Ar	IR	5,7
			1904.4	Xe	IR	9
	3		939.1vw	Ar	IR	7
II	4		580m	Ar	IR	5,7
			571.9	Xe	IR	9

$B_0=0.160$ MW²⁻⁴IR⁸

References

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⁹G. Maier and C. Lutz, *Eur. J. Org. Chem.* **1998**, 769.
¹⁰M. Dibben, J. Szczepanski, C. Wehlburg, and M. Vala, *J. Phys. Chem. A* **104**, 3584 (2000).

CCCS

In an argon matrix, a weak absorption maximum at 26460 (378 nm) has been assigned² to CCCS.

$\tilde{X}^1\Sigma$						
$C_{\infty v}$		Structure: MW ³				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1		2058.22	gas	DL	5
			2047.6vs	Ar	IR	2,6
			1533.2wm	Ar	IR	2
	3		725.6w	Ar	IR	2

$B_0=0.096$ MW^{1,4}

References

- ¹S. Yamamoto, S. Saito, K. Kawaguchi, N. Kaifu, H. Suzuki, and M. Ohishi, *Astrophys. J.* **317**, L119 (1987).
²G. Maier, J. Schrot, H. P. Reisenauer, and R. Janoschek, *Chem. Ber.* **124**, 2617 (1991).
³Y. Ohshima and Y. Endo, *J. Mol. Spectrosc.* **153**, 627 (1992).
⁴F. J. Lovas, R. D. Suenram, T. Ogata, and S. Yamamoto, *Astrophys. J.* **399**, 325 (1992).
⁵S. Takano, J. Tang, and S. Saito, *J. Mol. Spectrosc.* **178**, 194 (1996).
⁶J. Szczepanski, R. Hodyss, J. Fuller, and M. Vala, *J. Phys. Chem. A* **103**, 2975 (1999).

SiNNSi

\tilde{X}						
$D_{\infty h}$						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		SiN stretch	1153.3	N ₂	IR	1

Reference

- ¹G. Maier, H. P. Reisenauer, and J. Glatthaar, *Organomet.* **19**, 4775 (2000).

cyc-(SiN)₂

\tilde{X}						
C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b_2		SiN stretch	900.9	N ₂	IR	1

Reference

- ¹G. Maier, H. P. Reisenauer, and J. Glatthaar, *Organomet.* **19**, 4775 (2000).

SiNSiN

\tilde{X}						
$C_{\infty v}$						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		NSi stretch	1374.7	N ₂	IR	1
		SiN stretch	1164.4	N ₂	IR	1

Reference

¹G. Maier, H. P. Reisenauer, and J. Glatthaar, *Organomet.* **19**, 4775 (2000).

OTaCO⁻
 $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1838.1	Ar	IR	1

Reference

¹X.-F. Wang, M.-H. Chen, L.-N. Zhang, and Q.-Z. Qin, *J. Phys. Chem. A* **104**, 758 (2000).

OCrCO⁻
 $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1831.1	Ar	IR	1
		CrO stretch	825.5	Ar	IR	1

Reference

¹M. Zhou, B. Liang, and L. Andrews, *J. Phys. Chem. A* **103**, 2013 (1999).

OMnCO⁻
 $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1810.0	Ar	IR	1
		MnO stretch	810.1	Ar	IR	1

Reference

¹M. Zhou, B. Liang, and L. Andrews, *J. Phys. Chem. A* **103**, 2013 (1999).

OFeCO⁻
 $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1806.3	Ar	IR	1
		FeO stretch	814.8	Ar	IR	1

Reference

¹M. Zhou, B. Liang, and L. Andrews, *J. Phys. Chem. A* **103**, 2013 (1999).

OCoCO⁻
 $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1849.2	Ar	IR	1
		CoO stretch	807.9	Ar	IR	1

Reference

¹M. Zhou, B. Liang, and L. Andrews, *J. Phys. Chem. A* **103**, 2013 (1999).

CoCO₂⁻
 $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO ₂ a-stretch	1693.5	Ar	IR	1
		CO ₂ s-stretch	1228.4	Ar	IR	1
		CO ₂ deform.	721.9	Ar	IR	1

Reference

¹M. Zhou, B. Liang, and L. Andrews, *J. Phys. Chem. A* **103**, 2013 (1999).

ONiCO⁻
 $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1881.4	Ar	IR	1

Reference

¹M. Zhou, B. Liang, and L. Andrews, *J. Phys. Chem. A* **103**, 2013 (1999).

NiCO₂⁻
 $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO ₂ a-stretch	1684.8	Ar	IR	1
		CO ₂ s-stretch	1226.1	Ar	IR	1
		CO ₂ deform.	723.9	Ar	IR	1

Reference

¹M. Zhou, B. Liang, and L. Andrews, *J. Phys. Chem. A* **103**, 2013 (1999).

OCuCO⁻ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1943.4	Ar	IR	1

Reference¹M. Zhou, B. Liang, and L. Andrews, *J. Phys. Chem. A* **103**, 2013 (1999).**CuCO₂⁻** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO ₂ a-stretch	1713.4	Ar	IR	1
		CO ₂ s-stretch	1234.2	Ar	IR	1
		CO ₂ deform.	697.0	Ar	IR	1

Reference¹M. Zhou, B. Liang, and L. Andrews, *J. Phys. Chem. A* **103**, 2013 (1999).**OZnCO⁻** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2084.4	Ar	IR	1

Reference¹M. Zhou, B. Liang, and L. Andrews, *J. Phys. Chem. A* **103**, 2013 (1999).**t-OCCO⁺** \tilde{B} gas PD²

Superposed on continuum.

 $\tilde{B}-\tilde{X}$ 270–330 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a_g</i>			475(20)	gas	PD	2

 $\tilde{X}^2\tilde{B}_u$ C_{2h} Structure: ESR, MO¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a_g</i>	1	CO s-stretch	2076T ^a	Ne	IR	4
			526(30)T	gas	PD	2
<i>b_u</i>	5	CO a-stretch	2056.6	Ne	IR	3–5

^aCalculated using observed values for asymmetrically substituted species.**References**

- ¹L. B. Knight, Jr., J. Steadman, P. K. Miller, D. E. Bowman, E. R. Davidson, and D. Feller, *J. Chem. Phys.* **80**, 4593 (1984).
²S. C. Ostrander, L. Sanders, and J. C. Weisshaar, *J. Chem. Phys.* **84**, 529 (1986).
³M. E. Jacox and W. E. Thompson, *Res. Chem. Intermed.* **12**, 33 (1989).
⁴W. E. Thompson and M. E. Jacox, *J. Chem. Phys.* **95**, 735 (1991).
⁵M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 2964 (1999).

N₄⁺ \tilde{A}, \tilde{B}

A broad, unstructured absorption^{1–3} between 270 and 650 nm, with a maximum near 330 nm, leads to the formation of N₂⁺ + N₂. Detection of the fluorescence of N₂⁺ (B) at the higher energies in this range suggests that the potential energy surface for a bound excited state of N₄⁺ which correlates with N₂⁺ (B) has an avoided crossing with the surface for the dissociative A state.³

 $\tilde{X}^2\Sigma_u$ D_{∞h} Structure: ESR, MO⁴DL^{7,8}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	N≡N s-stretch	2283T ^a	Ne	IR	6
Σ_u^+	3	N≡N a-stretch	2234.51	gas	DL	7,8
			2237.6	Ne	IR	5,6

*B*₀ = 0.112 DL^{7,8}^aCalculated using observed values for asymmetrically ¹⁵N-substituted species.**References**

- ¹G. P. Smith and L. C. Lee, *J. Chem. Phys.* **69**, 5393 (1978).
²M. F. Jarrold, A. J. Illies, and M. T. Bowers, *J. Chem. Phys.* **81**, 214 (1984).
³S. C. Ostrander and J. C. Weisshaar, *Chem. Phys. Lett.* **129**, 220 (1986).
⁴L. B. Knight, Jr., K. D. Johannessen, D. C. Cobranchi, E. A. Earl, D. Feller, and E. R. Davidson, *J. Chem. Phys.* **87**, 885 (1987).
⁵M. E. Jacox and W. E. Thompson, *Res. Chem. Intermed.* **12**, 33 (1989).
⁶W. E. Thompson and M. E. Jacox, *J. Chem. Phys.* **93**, 3856 (1990).
⁷T. Ruchti, T. Speck, J. P. Connelly, E. J. Bieske, H. Linnartz, and J. P. Maier, *J. Chem. Phys.* **105**, 2591 (1996).
⁸T. Speck, T. Ruchti, H. Linnartz, and J. P. Maier, *J. Mol. Spectrosc.* **185**, 425 (1997).

cyc-(O₂Sc)O \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OO stretch	1109.5w	Ar	IR	1
			922.0w	Ar	IR	1
		ScO stretch	909.0s	Ar	IR	1,2
			466.3w	Ar	IR	1,2

References

- ¹G. V. Chertihin, L. Andrews, M. Rosi, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **101**, 9085 (1997).

²C. W. Bauschlicher, Jr., M. Zhou, L. Andrews, J. R. Tobias Johnson, I. Panas, A. Snis, and B. O. Roos, *J. Phys. Chem. A* **103**, 5463 (1999).

VO₃

\tilde{B}^2B_2 C_{2v}
T₀=6370(400) gas PE¹

\tilde{A}^2A_1 C_{2v}
T₀=4760(400) gas PE¹

Reference

¹H. Wu and L.-S. Wang, *J. Chem. Phys.* **108**, 5310 (1998).

CrO₃

\tilde{B}
T^a=14280(230) gas PE³

\tilde{A}
T^a=8390(230) gas PE³

\tilde{a}
T^a=7660(230) gas PE³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CrO ₃ s-stretch	890(60)	gas	PE	3
	3	CrO stretch	991.3	Ne	IR	2
			968.4	Ar	IR	1

^aFrom vertical electron detachment energies.

References

¹G. V. Chertihin, W. D. Bare, and L. Andrews, *J. Chem. Phys.* **107**, 2798 (1997).

²M. Zhou and L. Andrews, *J. Chem. Phys.* **111**, 4230 (1999).

³G. L. Gutsev, P. Jena, H.-J. Zhai, and L.-S. Wang, *J. Chem. Phys.* **115**, 7935 (2001).

MoO₃

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	MoO stretch	976T	Ne	IR	1
e	3	MoO stretch	923.4	Ne	IR	1,3
			915.8	Ar	IR	2

References

¹W. D. Hewett, Jr., J. H. Newton, and W. Weltner, Jr., *J. Phys. Chem.* **79**, 2640 (1975).

²W. D. Bare, P. F. Souter, and L. Andrews, *J. Phys. Chem. A* **102**, 8279 (1998).

³M. Zhou and L. Andrews, *J. Chem. Phys.* **111**, 4230 (1999).

WO₃

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e	3	WO stretch	924.3	Ne	IR	4
			918.3	Ar	IR	1-3
			916	Kr	IR	1

References

¹D. W. Green and K. M. Ervin, *J. Mol. Spectrosc.* **89**, 145 (1981).

²M. J. Almond and A. J. Downs, *J. Chem. Soc., Dalton Trans.* 809 (1988).

³W. D. Bare, P. F. Souter, and L. Andrews, *J. Phys. Chem. A* **102**, 8279 (1998).

⁴M. Zhou and L. Andrews, *J. Chem. Phys.* **111**, 4230 (1999).

(cyc-O₂Mn)O

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	OO stretch	1092.2	Ar	IR	1
		MnO stretch	886.9	Ar	IR	1

Reference

¹G. V. Chertihin and L. Andrews, *J. Phys. Chem. A* **101**, 8547 (1997).

ReO₃

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e	3	ReO ₃ stretch	962.0	Ne	IR	1
			953.4	Ar	IR	1

Reference

¹M. Zhou, A. Citra, B. Liang, and L. Andrews, *J. Phys. Chem. A* **104**, 3457 (2000).

FeO₃

\tilde{A}
 $T_0 = 4440(160)$ gas PE²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			850(50)	gas	PE	2

 \tilde{X} D_{3h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			850(50)	gas	PE	2
			975.8T	Ar	IR	1

References

- ¹G. V. Chertihin, W. Saffel, J. T. Yustein, L. Andrews, M. Neurock, A. Ricca, and C. W. Bauschlicher, Jr., *J. Phys. Chem.* **100**, 5261 (1996).
²H. Wu, S. R. Desai, and L.-S. Wang, *J. Am. Chem. Soc.* **118**, 5296 (1996); *J. Am. Chem. Soc.* **118**, 7434 (1996).

RuO₃ \tilde{X} D_{3h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>e'</i>	3	RuO ₃ stretch	901.1	Ne	IR	2
			893.3	Ar	IR	1,2

References

- ¹J. G. Kay, D. W. Green, K. Duca, and G. L. Zimmerman, *J. Mol. Spectrosc.* **138**, 49 (1989).
²M. Zhou, A. Citra, B. Liang, and L. Andrews, *J. Phys. Chem. A* **104**, 3457 (2000).

OsO₃ \tilde{X} D_{3h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>e'</i>	3	OsO ₃ stretch	966.1	Ne	IR	1
			959.1	Ar	IR	1

Reference

- ¹M. Zhou, A. Citra, B. Liang, and L. Andrews, *J. Phys. Chem. A* **104**, 3457 (2000).

OCooO \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OO stretch	1331.5T	Ar	IR	1

Reference

- ¹G. V. Chertihin, A. Citra, L. Andrews, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **101**, 8793 (1997).

cyc-(O₂Co)O \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OO stretch	1090.0	Ar	IR	1
			783.0	Ar	IR	1

Reference

- ¹G. V. Chertihin, A. Citra, L. Andrews, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **101**, 8793 (1997).

ORhOO \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OO stretch	1367.3	Ar	IR	1

Reference

- ¹A. Citra and L. Andrews, *J. Phys. Chem. A* **103**, 4845 (1999).

PtO₃ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	3	OPtO a-stretch	833.4T	Ar	IR	1

Reference

- ¹W. D. Bare, A. Citra, G. V. Chertihin, and L. Andrews, *J. Phys. Chem. A* **103**, 5456 (1999).

cyc-(O₂Pt)O \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	OO stretch	1174.2	Ar	IR	1

Reference

- ¹W. D. Bare, A. Citra, G. V. Chertihin, and L. Andrews, *J. Phys. Chem. A* **103**, 5456 (1999).

OOPtO

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OO stretch	1402.1	Ar	IR	1

Reference

¹W. D. Bare, A. Citra, G. V. Chertihin, and L. Andrews, *J. Phys. Chem. A* **103**, 5456 (1999).

UO₃ \tilde{X} C_{2v}Structure: IR²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	UO stretch	843.5	Ar	IR	1,2
	2	UO ₂ s-stretch	760.3	Ne	IR	5
			745.6	Ar	IR	1–5
b ₂	5	UO ₂ a-stretch	740.7	Kr	IR	1
			865.3	Ne	IR	5
			852.6	Ar	IR	1–5
			848.1	Kr	IR	1

References

¹S. D. Gabelnick, G. T. Reedy, and M. G. Chasanov, *J. Chem. Phys.* **58**, 4468 (1973).

²S. D. Gabelnick, G. T. Reedy, and M. G. Chasanov, *J. Chem. Phys.* **59**, 6397 (1973).

³R. D. Hunt and L. Andrews, *J. Chem. Phys.* **98**, 3690 (1993).

⁴K. Sankaran, K. Sundararajan, and K. S. Viswanathan, *Bull. Mater. Sci.* **22**, 785 (1999).

⁵M. Zhou, L. Andrews, N. Ismail, and C. Marsden, *J. Phys. Chem. A* **104**, 5495 (2000).

(cyc-NNB)Cl

In an argon matrix, a broad absorption with maximum at 24510 (408 nm) has been tentatively assigned² to (cyc-NNB)Cl.

 \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Ring breathing	1677	Ar	IR	1

References

¹M. J. Travers, E. L. Eldenburg, and J. V. Gilbert, *J. Phys. Chem. A* **103**, 9661 (1999).

²M. J. Travers and J. V. Gilbert, *J. Phys. Chem. A* **104**, 3780 (2000).

OCCS

In an argon matrix, a structured absorption between 280 and 320 nm with a maximum at 32680 (306 nm) has been assigned¹ to OCCS.

 \tilde{X} C_{∞v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	OCC stretch	2156.0vs	Ar	IR	1
	2	OCCS stretch	1505.2m	Ar	IR	1
	3	OCCS stretch	685T	Ar	IR	1
Π	4	Bend	452.2wm	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, and R. Ruppel, *Angew. Chem.* **109**, 1972 (1997); *Angew. Chem. Int. Ed. Engl.* **36**, 1862 (1997).

Ge₂O₂ \tilde{X} D_{2h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a _g	1		692.5	Ar	Ra	4
	2	Ge–Ge breathing	400(60)	gas	PE	3
b _{1g}	3		328.0	Ar	Ra	4
			469.0	Ar	Ra	4
b _{1u}	4		173T	Ar	Ra	4
b _{2u}	5	GeO stretch	667.7	Ar	IR	1,2,4
			667	N ₂	IR	1
b _{3u}	6	GeO stretch	601.7	Ar	IR	1,2,4
			599	N ₂	IR	1

References

¹J. S. Ogden and M. J. Ricks, *J. Chem. Phys.* **52**, 352 (1970).

²L. Andrews and M. McCluskey, *J. Mol. Spectrosc.* **154**, 223 (1992).

³J. B. Nicholas, J. Fan, H. Wu, S. D. Colson, and L.-S. Wang, *J. Chem. Phys.* **102**, 8277 (1995).

⁴A. Zumbusch and H. Schnöckel, *J. Chem. Phys.* **108**, 8092 (1998).

NOCN

 \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CN stretch	2086.0vw	Ar	IR	1
	2	NO stretch	1836.9vs	Ar	IR	1
	3	Bend	485.4w	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, J. Eckwert, M. Naumann, and M. De Marco, *Angew. Chem.* **109**, 1795 (1997); *Angew. Chem. Int. Ed. Engl.* **36**, 1707 (1997).

CNNO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	CN stretch	1997.5m ^a 1967.1m ^a	Ar	IR	1
	2	NO stretch	1681.0vs	Ar	IR	1
	3	Deformation	723.4m	Ar	IR	1
	4	NN stretch	261.0ms	Ar	IR	1

^aIn Fermi resonance with ($\nu_2 + \nu_4$).

Reference

¹G. Maier, H. P. Reisenauer, J. Eckwert, M. Naumann, and M. De Marco, *Angew. Chem.* **109**, 1795 (1997); *Angew. Chem. Int. Ed. Engl.* **36**, 1707 (1997).

NCNO

An absorption maximum has been reported^{3,9} at 216 nm, with absorption extending beyond 200 nm.

A weaker absorption maximum occurs near 270 nm, with a long wavelength threshold near 400 nm.⁹

\tilde{A}^1A''		C_s		Structure: PF ¹¹		
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
$T_0 = 11339$ gas AB ^{1,3,7} PF ¹¹ $\tilde{A}-\tilde{X}$ 540–971 nm						
Threshold for photodissociation into CN and NO at 17085. ^{9,10} Extensively perturbed by interaction with high vibrational levels of the ground state. ¹¹						
a'	1	C≡N stretch	1956	gas	PF	11
	2	N=O stretch	1485	gas	PF	11
	3	C–N stretch	918	gas	PF	11
	4	NCN bend	543	gas	AB,PF	7,11
	5	CNO bend	212.5	gas	AB,PF	7,11
a''	6	Torsion	411	gas	PF	11

$\tau_{rad} \approx 14 \mu s$ gas LF^{8,12}

$\tau_{fluor} > 40 \mu s$ for all levels below D_0 (17085) LF¹²

$A = 4.76(2)$; $B = 0.167(3)$ PF¹¹

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	C≡N stretch	2170.0	gas	IR	6
			2163.0s	Ar	IR	13
	2	N=O stretch	1501.0	gas	IR	6
			1498.5vs	Ar	IR	13
			820.0	gas	IR	3,6
	3	C–N stretch	809.5s	Ar	IR	13
			809.5s	Ar	IR	13
	4	NCN bend	588.5	gas	IR	6
			576.1	Ar	IR	13
	5	CNO bend	212.0(2)	gas	IR	5
264.2			gas	IR	5	
a''	6	Torsion	264.2	gas	IR	5

$A_0 = 2.709$; $B_0 = 0.180$; $C_0 = 0.168$ MW^{2,4}IR⁵

References

- ¹P. Horsewood and G. W. Kirby, *Chem. Commun.* 1139 (1971).
²R. Dickinson, G. W. Kirby, J. G. Sweeny, and J. K. Tyler, *Chem. Commun.* 241 (1973).
³E. A. Dorko and L. Buelow, *J. Chem. Phys.* **62**, 1869 (1975).
⁴R. Dickinson, G. W. Kirby, J. G. Sweeny, and J. K. Tyler, *J. Chem. Soc., Faraday Trans. 2* **74**, 1393 (1978).
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⁹I. Nadler, J. Pfab, H. Reisler, and C. Wittig, *J. Chem. Phys.* **81**, 653 (1984).
¹⁰I. Nadler, H. Reisler, M. Noble, and C. Wittig, *Chem. Phys. Lett.* **108**, 115 (1984).
¹¹M. Noble, I. Nadler, H. Reisler, and C. Wittig, *J. Chem. Phys.* **81**, 4333 (1984).
¹²C. X. W. Qian, H. Reisler, and C. Wittig, *Chem. Phys. Lett.* **139**, 175 (1987).
¹³G. Maier, H. P. Reisenauer, J. Eckwert, M. Naumann, and M. De Marco, *Angew. Chem.* **109**, 1795 (1997); *Angew. Chem. Int. Ed. Engl.* **36**, 1707 (1997).

NCPO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
	1	CN stretch	2165	gas	IR	1
	2	PO stretch	1385	gas	IR	1

Reference

¹A. W. Allaf, M. Kassem, M. Alibrahim, and I. Boustani, *J. Mol. Struct.* **478**, 193 (1999).

CICNC

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	NC stretch	1943.6vs	Ar	IR	1,2
	2	CN stretch	1023.7m	Ar	IR	1,2
	3	CNC deform.	769.5wm	Ar	IR	2

References

- ¹D. E. Milligan, M. E. Jacox, and A. M. Bass, *J. Chem. Phys.* **43**, 3149 (1965).
²G. Maier, A. Bothur, J. Eckwert, and H. P. Reisenauer, *Chem. Eur. J.* **4**, 1964 (1998).

CICCN

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CN stretch	2114sT	Ar	IR	1,2
			995msT	Ar	IR	1,2

References

- ¹D. E. Milligan, M. E. Jacox, and A. M. Bass, *J. Chem. Phys.* **43**, 3149 (1965).
²G. Maier, A. Bothur, J. Eckwert, and H. P. Reisenauer, *Chem. Eur. J.* **4**, 1964 (1998).

BrCNC

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	NC stretch	1922.3vs	Ar	IR	1,2
			1925.7	N ₂	IR	2
	2	CN stretch	1034.3m	Ar	IR	1,2
			1038.5	N ₂	IR	2
	3	CNC deform.	672.4wm	Ar	IR	2

References

- ¹D. E. Milligan, M. E. Jacox, and A. M. Bass, *J. Chem. Phys.* **43**, 3149 (1965).
²G. Maier, A. Bothur, J. Eckwert, and H. P. Reisenauer, *Chem. Eur. J.* **4**, 1964 (1998).

BrCCN

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CN stretch	2101.1mT	Ar	IR	1,2
			984.5vsT	Ar	IR	1,2

References

- ¹D. E. Milligan, M. E. Jacox, and A. M. Bass, *J. Chem. Phys.* **43**, 3149 (1965).
²G. Maier, A. Bothur, J. Eckwert, and H. P. Reisenauer, *Chem. Eur. J.* **4**, 1964 (1998).

N₄

\tilde{X}		T_d				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
f_2	3		936.7T	N ₂	IR	1

Reference

- ¹J. P. Zheng, J. Waluk, J. Spanget-Larsen, D. M. Blake, and J. G. Radziszewski, *Chem. Phys. Lett.* **328**, 227 (2000).

P₄

\tilde{X}^1A_1		T_d Structure: IR ⁶				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1		600.5	gas	Ra	2,4,5
e	2		360.8	gas	Ra	2-5
f_2	3		466.29	gas	IR,Ra	1,2,4-6
			465.8	Ar	IR	7

$B_0=0.113$ IR⁶

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YO₃⁻

\tilde{X}		C_{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e	3	YO ₃ stretch	530.8	Ar	IR	1

Reference

- ¹L. Andrews, M. Zhou, G. V. Chertihin, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **103**, 6525 (1999).

VO₃⁻

Threshold for electron detachment from ground-state VO₃⁻ = 35180(400) gas PE¹

Reference

- ¹H. Wu and L.-S. Wang, *J. Chem. Phys.* **108**, 5310 (1998).

NbO₃⁻

\tilde{X}		C_{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>e</i>	3	NbO ₃ stretch	817.1	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, J. Phys. Chem. A **102**, 8251 (1998).

TaO₃⁻

\tilde{X}		C_{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>e</i>	3	TaO ₃ stretch	807.0	Ar	IR	1,2

References

¹M. Zhou and L. Andrews, J. Phys. Chem. A **102**, 8251 (1998).

²M. Chen, X. Wang, L. Zhang, M. Yu, and Q. Qin, Chem. Phys. **242**, 81 (1999).

CrO₃⁻

Threshold for electron detachment from ground-state CrO₃⁻ = 29530(160) gas PE²

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CrO ₃ stretch	943.9	Ne	IR	1

References

¹M. Zhou and L. Andrews, J. Chem. Phys. **111**, 4230 (1999).

²G. L. Gutsev, P. Jena, H.-J. Zhai, and L.-S. Wang, J. Chem. Phys. **115**, 7935 (2001).

ReO₃⁻

Threshold for electron detachment from ground-state ReO₃⁻ = 29000(800) gas PE²

\tilde{X}		D_{3h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>e'</i>	3	ReO ₃ stretch	923.7	Ne	IR	1

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PrO₃⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>e</i>	3	PrO ₃ stretch	585.3 573.9	Ne Ar	IR IR	1 1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

GdO₃⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		GdO stretch	566.1	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 3171 (1999).

TbO₃⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		TbO stretch	596.1T	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **103**, 6972 (1999).

UO₂F

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		UO ₂ a-stretch	871.7 867.8	Ar	IR	1

Reference

¹P. F. Souter and L. Andrews, J. Mol. Struct. **412**, 161 (1997).

***t*-OCCO⁻**

In solid neon, threshold for electron detachment <18000.^{1,2}

\tilde{X}		C _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a _g	3	Bend	524 ^a	Ne	IR	2
b _u	5	CO a-stretch	1517.7	Ne	IR	1,2,5
			1515.5	Ar	IR	3,4

^a(ν₃ + ν₅) - ν₅. Tentative assignment of combination band.

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CICNO⁺

$\tilde{A}^2\Pi$ C_{∞v}
T^a=28400(800) gas PE¹

^aFrom vertical ionization potentials.

Reference

- T. Pasinszki and N. P. C. Westwood, J. Phys. Chem. A **102**, 4939 (1998).

ONNO⁺ a

In a neon matrix, photoisomerizes to the higher energy isomer or is promoted to a metastable state which has its ν₅ absorption^{10,13,15} at 1424.1 when the deposit is exposed to 540–580 nm mercury-arc radiation.¹⁵ The reverse conversion occurs on slight warming of the deposit or on exposure of the deposit to tungsten-lamp radiation of wavelength longer than 780 nm.¹⁵

In the gas phase, dissociates into NO+NO⁺, with onset at 10700(1000) and maximum at 16400(1000). AB^{2-5,7}PE⁸

\tilde{X}		C _{2h} (C _{2v})				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a _g (a ₁)	1	NO s-stretch	2090.8	gas	PI,PE	6,8
			2064.7 ^b	Ne	TPE	11,12,16
			2054.2 ^b	Ar	IR	10,13,15
					IR	14
	2		317.4	gas	TPE	16
	3		118.5	gas	TPE	11,12,16
a _u (a ₂)	4	Torsion	131.5	gas	TPE	16
b _u (b ₂)	5	NO a-stretch	1618.2	gas	TPE	12,16
			1619.2	Ne	IR	10,12,13,15
			1593.3	Ar	IR	1,9,14
	6	Deformation	327.2	gas	TPE	11,16

^aIt is uncertain whether the *cis*- or the *trans*- isomer is the lower energy structure. *Ab initio* calculations suggest that the energy separation between the two isomers is small. The observation of the ν₁ absorption only for the noncentrosymmetric isotopomers^{10,13} is consistent with the selection rules

for the *trans*- but not the *cis*- conformation. However, a possible exception has been discussed by Ref. 15.

^b(ν₁ + ν₅) - ν₅.

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CIBNCI

\tilde{X}		C _{∞v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	BN stretch	2034	Ar	IR	1

Reference

- L. A. Johnson, S. A. Sturgis, I. A. Al-Jihad, B. Liu, and J. V. Gilbert, J. Phys. Chem. A **103**, 686 (1999).

FAIO₂

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1		1077.3	Ar	IR	1
b ₂	5		781.8	Ar	IR	1

Reference

- J. Bahlo, H.-J. Himmel, and H. Schnöckel, Angew. Chem. Int. Ed. **40**, 4696 (2001).

F₂C=C:

In an argon matrix, absorption maxima at 43860 (228 nm) and 39370 (254 nm) have been assigned² to F₂C=C:

\tilde{a}^3A_2 C_{2v}
 $T_0=7455(70)$ gas PE¹

\tilde{X}^1A_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	C=C stretch	1670(25)	gas	PE	1
			1672m	Ar	IR	2
	2	CF ₂ s-stretch	905(25)	gas	PE	1
918m			Ar	IR	2	
3	CF ₂ scissors	510(25)	gas	PE	1	
		511w	Ar	IR	2	
b_1	4	OPLA	551w	Ar	IR	2
b_2	5	CF ₂ a-stretch	1267vs	Ar	IR	2
			6	CF ₂ rock	334T	Ar

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²J. Breidung, H. Bürger, C. Kötting, R. Kopitzky, W. Sander, M. Senzlobler, W. Thiel, and H. Willner, *Angew. Chem.* **109**, 2072 (1997); *Angew. Chem. Int. Ed. Engl.* **36**, 1983 (1997).

CICNO

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
1	CNO a-stretch	2219	gas	IR	2	
		2281.4	Ar	IR	1	
		2261.7				
2	CNO s-stretch	1343T	gas	IR	2	
		1326.3	Ar	IR	1	

Barrier to linearity=167.³

References

¹G. Maier and J. H. Teles, *Angew. Chem.* **99**, 152 (1987); *Angew. Chem. Int. Ed. Engl.* **26**, 155 (1987).

²T. Pasinski and N. P. C. Westwood, *J. Phys. Chem. A* **102**, 4939 (1998).

³H. Lichau, C. W. Gillies, J. Z. Gillies, S. C. Ross, B. P. Winnewisser, and M. Winnewisser, *J. Phys. Chem. A* **105**, 10065 (2001).

BrCNO

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
CNO a-stretch			2211	gas	IR	2
			2271.3	Ar	IR	1
			2252.5			
CNO s-stretch			1321	gas	IR	2
			1305.6	Ar	IR	1

Barrier to linearity=131.³

References

¹G. Maier and J. H. Teles, *Angew. Chem.* **99**, 152 (1987); *Angew. Chem. Int. Ed. Engl.* **26**, 155 (1987).

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c-(NO)₂

A diffuse gas-phase absorption with onset below 38500 (>260 nm) and maximum at 48800 (205 nm) has been assigned^{4,7} to c-(NO)₂.

\tilde{X} C_{2v} Structure: MW⁹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a_1	1	NO s-stretch	1868.25	gas	IR,DL,Ra	3,13,15,24,26	
			1867.2	Ne	IR	14,16,17,27	
			1863.4m	Ar	IR	1,21	
			1867.0	N ₂	IR	2,17,23	
			1862m	CO ₂	IR,Ra	1,8	
			1866	NO	IR,Ra	5,6,10	
	2	N··N stretch	239.36	gas	IR,Ra	22,25,26	
			242.8	Ar	IR	21	
			244.5	N ₂	IR	23	
			262	CO ₂	Ra	8	
			266	NO	IR,Ra	5,6,10	
			134.50	gas	IR,Ra	22,25,26	
3	NNO s-bend	175.5T	Ar	IR	21		
		117T	gas	IR	22		
a_2	4	Torsion	97T	NO	IR,Ra	5,6,10	
			1789.09	gas	IR,DL	3,11,12	
b_2	5	NO a-stretch	1780.6	Ne	IR	14,16,17,27	
			1778.7				
			1776.3s	Ar	IR	1,21	
			1779.9	N ₂	IR	2,17,23	
			1768s	CO ₂	IR	1	
			1762	NO	IR,Ra	6,10	
			429.14vw	gas	IR	22,25	
			6	NNO a-bend			

$A_0=0.862$; $B_0=0.187$; $C_0=0.154$ MW^{9,20}IR¹⁵

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t-(NO)₂

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a_g</i>	1	Sym. stretch	1861.1vw	N ₂	IR	4
<i>b_u</i>	5	Asym. stretch	1762.5	Ne	IR	3,7
			1760.6			
			1747.1	Ar	IR	5
			1760.0	N ₂	IR	2,4,5,6
			1740	CO ₂	IR	1

References

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⁷L. Andrews and M. Zhou, *J. Chem. Phys.* **111**, 6036 (1999).

BF₃⁺

$\tilde{E}^2A'_1$ D_{3h}
*T*₀ ≅ 47800 gas PE³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ' ₁	1	BF stretch	800T	gas	PE	3

τ = 10 ns gas EM⁵

\tilde{D}^2E' D_{3h}
*T*₀ = 34860(160) gas PE^{1,4,7}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ' ₁	1	BF stretch	760(20)	gas	PE	1,4,7

Additional structure reported by Ref. 4 has been attributed to vibronic coupling.

$\tilde{C}^2A''_2$ D_{3h}
*T*₀ = 27110(80) gas PE^{1,2,4,7}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ' ₁	1	BF stretch	775(20)	gas	PE	2,4,7

\tilde{B}^2E' D_{3h}
*T*₀ = 10890(240) gas PE^{1,2,4,7}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ' ₁	1	BF stretch	770(60)	gas	PE	1,4,7

\tilde{A}^2E'' D_{3h}
*T*₀ = 5890(240) gas PE^{1,4,7}

$\tilde{X}^2A'_2$ D_{3h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>e</i> '	3	BF ₃ stretch	1661.6	Ne	IR	6

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BCl₃⁺

$\tilde{E}^2A'_1$ D_{3h}
*T*₀ = 49200(120) gas PE^{1,8}TPE⁶

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ' ₁	1	BCl stretch	403(16)	gas	PE, TPE	1,6,8

\tilde{D}^2E' D_{3h}
*T*₀ = 29700(120) gas PE^{1,8}TPE⁶

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ' ₁	1	BCl stretch	420(20)	gas	PE	8

Emission which is observed between about 300 and 650 nm on excitation of BCl₃ by radiation of wavelength shorter than 81 nm (15.36 eV) arises from the \tilde{D} state of BCl₃⁺.³⁻⁵ The emission between 330 and 420 nm has been attributed to the $\tilde{D}-\tilde{X}$ transition of BCl₃⁺, and that between 420 and 580 nm, with vibrational spacings of ~445, to the $\tilde{D}-\tilde{A}$, \tilde{B} transitions.³ The T-PEPICO measurements of Ref. 6 indicate that the quantum yield for these emissions is ≤0.04 and that fragmentation into BCl₂⁺ + Cl predominates.

A broad absorption with maximum at 320 nm (31200) which appears on argon-resonance photolysis of BCl_3 isolated in an argon matrix and which can be destroyed by prolonged exposure of the sample to 340–600 nm radiation has been assigned² to the $\tilde{D}-\tilde{X}$ transition of BCl_3^+ .

$A = 1600(80)$ gas TPE⁶
 $\tau = 8.9(2)$ ns gas EM⁹

\tilde{C}^2A_2'' D_{3h}
 $T_0 = 21780(120)$ gas PE^{1,8}TPE⁶

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1'	1	BCl stretch	403(16)	gas	PE, TPE	1,6,8

\tilde{B}^2E' D_{3h}
 $T^a = 9280(120)$ gas PE^{1,8}TPE⁶
 $A = 800\text{T}$ gas TPE⁶

\tilde{A}^2E'' D_{3h}
 $T_0 = 5080(120)$ gas PE^{1,8}TPE⁶

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e'	4	Deformation	320T	gas	PE	8

\tilde{X}^2A_2' D_{3h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e'	3	BCl stretch	1103.8 1090	Ne Ar	IR IR	7 2

^aFrom vertical ionization potential.

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BBr₃⁺

\tilde{F}
 $T \cong 53500(480)$ gas PE¹

\tilde{E}^2A_1' D_{3h}
 $T_0 = 49380(560)$ gas PE^{1,4}

\tilde{D}^2E' D_{3h}
 $T_0^a = 25500(400)$ gas PE^{1,4}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1'	1	BBr stretch	250(20)	gas	PE	4

Emission between 240 and 390 nm, with a lifetime ≤ 16 ns, which is observed when gas-phase BBr_3 is excited by radiation of wavelength shorter than 89 nm has been assigned³ to the $\tilde{D}-\tilde{X}$ transition of BBr_3^+ .

A broad absorption with maximum at 355 nm (28200) which appears on argon-resonance photolysis of BBr_3 isolated in an argon matrix and which can be destroyed by prolonged exposure of the sample to 340–600 nm radiation has been assigned² to the $\tilde{D}-\tilde{X}$ transition of BBr_3^+ .

$A = 2300(160)$ gas PE^{1,4}

\tilde{C}^2A_2'' D_{3h}
 $T_0 = 18440(480)$ gas PE^{1,4}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1'	1	BBr stretch	260(20)	gas	PE	4

\tilde{B}^2E' D_{3h}
 $T^b = 9680(480)$ gas PE^{1,4}

\tilde{A}^2E'' D_{3h}
 $T_0^a = 5000(400)$ gas PE^{1,4}
 $A \cong 1130$ gas PE¹

\tilde{X}^2A_2' D_{3h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e'	3	BBr stretch	930	Ar	IR	2

^aOnset of transition.

^bFrom vertical ionization potential.

References

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BI₃⁺

$\tilde{E}^2A'_1$ D_{3h}
 $T_0 = 47200(400)$ gas PE^{1,2}

\tilde{D}^2E' D_{3h}
 $T_0^a = 24450(480)$ gas PE^{1,2}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'_1	1	BI stretch	160(20)	gas	PE	2
A = 4030(80) gas PE ^{1,2}						

$\tilde{C}^2A''_2$ D_{3h}
 $T_0 = 17840(400)$ gas PE^{1,2}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'_1	1	BI stretch	180(20)	gas	PE	2

\tilde{B}^2E'' D_{3h}
 $T_0^a = 8310(400)$ gas PE^{1,2}
 A = 810(160) gas PE¹

\tilde{A}^2E' D_{3h}
 $T_0^a = 4840(400)$ gas PE^{1,2}
 A = 1450(160) gas PE¹

$\tilde{X}^2A'_2$ D_{3h}

^aOnset of transition.

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NNO₂⁻

\tilde{B}^2A_2 C_{2v}
 Photoelectron spectra² of NNO₂⁻ indicate that at both 532 and 355 nm dissociation into N₂O + O⁻ occurs. Fast ion beam translational spectroscopy studies⁶ indicate that a structured absorption with onset near 580 nm arises from excitation to the \tilde{B} state. Two different dissociation channels occur in this energy range; dissociation on the ground-state potential surface leads to the formation of N₂O + O⁻, and excited-state dissociation results in the formation of NO + NO⁻. When 266 and 213 nm photons are used, the photoelectron spectra⁵ suggest that electron detachment to form excited states of uncharged NNO₂ occurs.

\tilde{X}^2B_2 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	NN stretch	1359.6wm	Ne	IR	4
			1355w	Ar	IR	1,3,4
	2	NO ₂ s-stretch	950T	gas	PE	5
			1004.5wm	Ne	IR	4
b_2	5	NO ₂ a-stretch	1008w	Ar	IR	1,3,4
			1199.3m	Ne	IR	4,7
			1205.5m	Ar	IR	1,3

References

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t-(NO)₂⁻

In gas-phase photoelectron spectra¹ taken at 532 nm, evidence was obtained for both electron detachment and photodissociation into NO + NO⁻. Maximum in the photoelectron spectrum at 23640(400), or 2.93(5) eV gas PE³

\tilde{X} C_{2h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b_u	5	Asym. stretch	1227.5	Ne	IR	4,6
			1221.0	Ar	IR	2,5

References

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- ⁶L. Andrews and M. Zhou, J. Chem. Phys. **111**, 6036 (1999).

c-(NO)₂⁻

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	NO s-stretch	1300.3	Ar	IR	1
b_2	5	NO a-stretch	1225.1	Ne	IR	3
			1222.7	Ar	IR	1,2
	6	NNO deform.	884.4	Ar	IR	1

References

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NO₃

\tilde{B}^2E' D_{3h}
 $T_0 = 15089$ gas AB^{1,2,4-7}LF^{8,9,17,21} $\tilde{B}-\tilde{X}$ 450–795 nm
 All bands are diffuse.^{2,7,21} The threshold for the production of NO+O₂ ≤ 16780, and that for the production of O+NO₂ is 17090(20).²⁰ Above this second threshold, the predominant photodissociation products are O(³P) and NO₂.^{19,20}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'_1	1	Sym. stretch	930	gas	AB	2,6
			1450	gas	AB	6
			850	gas	AB	6

$\tau_0 = 340(20)$ μs gas LF¹⁰

\tilde{A}^2E'' D_{3h} Structure: DL²²
 $T_0 = 7061(8)$ gas PE¹⁴IR²³
 A high resolution diode laser study has been reported²² for a band at 7602.58, contributed by excitation to the $^2A''_1$ vibronic component of $\nu_4 = 1$ in the \tilde{A} state.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'_1	1	Sym. stretch	804(4)T	gas	PE	14
a''_2	2	OPLA	683(8)T	gas	IR	23
e'	4	Deformation	541(8)T	gas	PE	14

$B_{0001} = 0.433$; $C_{0001} = 0.216$ DL²²

$\tilde{X}^2A'_2$ D_{3h} Structure: DL¹¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'_1	1	Sym. stretch	1050	gas	LF,PE	8,9,14,17
a''_2	2	OPLA	762.33	gas	IR,LF	12,17
e'	3	NO stretch ^a	1492.39	gas	LF,DL	8,9,11,17
					IR	12
	4	Deformation ^b	360	gas	LF,PE	8,9,14,17

$B_0 = 0.459$ DL^{11,13}IR^{12,23}

^aAnomalous structure appears in the high resolution spectrum of this band because of the vibronic coupling with the \tilde{B}^2E' state.¹⁶ Coupling with this state and with the \tilde{A}^2E'' state leads to the appearance of infrared absorptions of several prominent overtone and combination bands between 1900 and 2600.^{15,18,23} Among these bands, also observed in fluorescence,^{8,17} is one near 2020, attributed by Ref. 17 to the occurrence of pseudorotation on the ground-state potential surface, but by Ref. 15 to intensification of the fourth overtone of ν_4 , which has appreciable “negative” anharmonicity, as a result of vibronic interaction with the \tilde{A} and \tilde{B} states.

^bCoupled to the \tilde{B}^2E' state through this mode.¹⁴

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c-OSNO

\tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	NO stretch	1450.8	Ar	IR	1
			1454.4	N ₂	IR	1
			1154.9	Ar	IR	1
	2	SO stretch	1156.1	N ₂	IR	1

Reference

- M. Bahou and Y.-P. Lee, J. Chem. Phys. **115**, 10694 (2001).

t-OSNO

\tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	NO stretch	1456.0	Ar	IR	1
			1459.0	N ₂	IR	1
			1181.2	Ar	IR	1
	2	SO stretch	1178.0	N ₂	IR	1

Reference

- M. Bahou and Y.-P. Lee, J. Chem. Phys. **115**, 10694 (2001).

PO₃

\tilde{C}^2E' ^a D_{3h}
 $T_0 = 14378\text{T}$ Ar AB^{1,3} $\tilde{B}-\tilde{X}$ 589–696 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'_1	1	PO ₃ s-stretch	913(10)	Ar	IR	1
e'	4	Deformation	525(10)	Ar	IR	1

\tilde{B}^2A_2 D_{3h}^b
 $T^c = 8790(680)$ gas PE²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e'	4	Deformation	500(80)	gas	PE	2

$\tilde{X}^2A'_2$ D_{3h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'_1	1	Sym. stretch	1000(100)	gas	PE	2

^aTentative assignment, by analogy with NO₃.

^bUpper component of ²E" state, which is split by Jahn–Teller interaction.

^cFrom vertical ionization potential.

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t-O₄⁺

In the gas phase, the high frequency tail of a photodissociation continuum, resulting in the formation of O₂⁺+O₂, has been observed^{1–3} between 450 and 680 nm.

\tilde{X} C_{2h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_g	1	O=O s-stretch	1644.1 ^a	Ne	IR	4,5
b_u	5	O=O a-stretch	1164.4	Ne	IR	4,5
			1118.6	Ar	IR	5

^a($\nu_1 + \nu_5$) – ν_5 .

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cyc-O₄⁺

\tilde{X} C_{2h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_g	1	O=O s-stretch	1628.3 ^a	Ne	IR	1–3
b_{1u}	5	O=O a-stretch	1320.3	Ne	IR	1–3
			1331.4	Ar	IR	3
			1328.9			

^a($\nu_1 + \nu_5$) – ν_5 .

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CCl₃⁺

\tilde{X} D_{3h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e'	3	CCl stretch	1045.9	Ne	IR	4
			1037	Ar	IR	1–3

References

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Cl₂SiO

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Si=O stretch	1240s	Ar	IR	1,2
	2	SiCl s-stretch	501.1m	Ar	IR	1,2
b_1	4	OPLA	280ms	Ar	IR	1
b_2	5	SiCl a-stretch	637.5vs	Ar	IR	1,2
	6	SiCl ₂ rock	269m	Ar	IR	1

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F₂SiS

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	SiF ₂ s-stretch	996vs	Ar	IR	1
	2	SiS stretch	638vw	Ar	IR	1
	3	SiF ₂ scissors	337wm	Ar	IR	1
b ₁	4	OPLA	296m	Ar	IR	1
b ₂	5	SiF ₂ a-stretch	969s	Ar	IR	1
	6	FSiS deform.	247w	Ar	IR	1

Reference

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c-OONO⁻

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	N=O stretch	1458.3	Ar	IR	1
	3		806.1	Ar	IR	1

Reference

¹B. Liang and L. Andrews, *J. Am. Chem. Soc.* **123**, 9848 (2001).

t-OONO⁻

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	N=O stretch	1433.3	Ar	IR	1
	2		983.2	Ar	IR	1

Reference

¹B. Liang and L. Andrews, *J. Am. Chem. Soc.* **123**, 9848 (2001).

BrNO₂

In the gas phase, unstructured absorption maxima have been observed⁵ at 50250 (199 nm), 40490 (247 nm), and 26880 (372 nm).

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	NO ₂ s-stretch	1294.40	gas	IR	3–5,8
			1291.0vs	Ar	IR	1,2,5,7
			1291	O ₂	IR	4
2	NO ₂ deform.	787	gas	IR	3–5	
		782.7s	Ar	IR	2,5,7	
		784	O ₂	IR	4	
3	NBr stretch	281.8w	Ar	IR	5,7	
		280.7w				
b ₁	4	OPLA	605	gas	IR	5
			605.7	Ar	IR	5,7
b ₂	5	NO ₂ a-stretch	1666.98	gas	IR	3–6
			1659.6vs	Ar	IR	2,5,7
			1655	O ₂	IR	4
6	NO ₂ wag	290Tsh	Ar	IR	5	

A₀=0.443; B₀=0.102; C₀=0.083 IR^{6,8}

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c-BrONO

In the gas phase, *c*-BrONO has unstructured absorption maxima at 43860 (228 nm) and 31650 (316 nm).²

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	N=O stretch	1650.7vs	Ar	IR	1
			862.6w	Ar	IR	1
			573.5m	Ar	IR	1
			420.2wm	Ar	IR	1
a''	6	Torsion	368.2w	Ar	IR	1

References

- ¹D. Scheffler and H. Willner, *Inorg. Chem.* **37**, 4500 (1998).
²J. B. Burkholder and J. J. Orlando, *Chem. Phys. Lett.* **317**, 603 (2000).

***t*-BrONO**

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	N=O stretch	1723.4vs	Ar	IR	1-4
	2	ONO deform.	835.9m	Ar	IR	2-4
	3	BrO stretch	586.9vs	Ar	IR	1-4
	4	O-N stretch	391.2wm	Ar	IR	2-4
a''	6	Torsion	150HT	Ar	IR	3,4

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 PO_3^-

Threshold for electron detachment from ground-state $PO_3^- = 39940(480)$ gas PE²

\tilde{X}		D_{3h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_2''	2	OPLA	480.3	Ar	IR	1,3
e'	3	PO_3 stretch	1273.3	Ar	IR	1,3

References

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²X.-B. Wang and L.-S. Wang, *Chem. Phys. Lett.* **313**, 179 (1999).
³C. W. Bauschlicher, Jr., M. Zhou, and L. Andrews, *J. Phys. Chem. A* **104**, 3566 (2000).

 PO_2Cl

\tilde{X}		C_{2v}		Structure: MW ²		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	PO_2 s-stretch	1122	Ar	IR	1
	2	PCI stretch	586	Ar	IR	1
b_2	5	PO_2 a-stretch	1429	Ar	IR	1

$A_0 = 0.296$; $B_0 = 0.141$; $C_0 = 0.095$ MW²

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²B. Brupbacher-Gatehouse, *J. Am. Chem. Soc.* **122**, 4171 (2000).

 F_2PO^+

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	PO stretch	1473.2wmT	Ne	IR	1
b_2	5	PF stretch	1135.5wT	Ne	IR	1

Reference

- ¹C. L. Lugez, K. K. Irikura, and M. E. Jacox, *J. Chem. Phys.* **108**, 8381 (1998).

 CCl_3

$\tilde{M}^2E'(4d)^a$		D_{3h}				
$T_0 = 57733(10)$		gas MPI ⁹				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_2''	2	OPLA	542(3)	gas	MPI	9

$\tilde{L}^2A_2'(4p)$		D_{3h}				
$T_0 = 56409(10)$		gas MPI ⁹				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_2''	2	OPLA	533(15)	gas	MPI	9

$\tilde{K}^2E'(4p)$		D_{3h}				
$T_0 = 56236(10)$		gas MPI ⁹				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_2''	2	OPLA	526(16)	gas	MPI	9

$\tilde{J}^2A_1'(4s)$		D_{3h}				
$T_0 = 53471(10)$		gas MPI ⁹				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_2''	2	OPLA	530(20)	gas	MPI	9

$\tilde{G}^2E'(3d)^a$		D_{3h}				
$T_0 = 51218(10)$		gas MPI ⁹				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_2''	2	OPLA	520(17)	gas	MPI	9

$\tilde{F}^2A_2''(3p)$		D_{3h}				
$T_0 = 47868(10)$		gas MPI ⁹				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_2''	2	OPLA	528(3)	gas	MPI	9

$\tilde{E}^2E'(3p)$ D_{3h}
 $T_0=47170(10)$ gas MPI⁹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'_1	1	Sym. stretch	544(6)	gas	MPI	9
a''_2	2	OPLA	509(21)	gas	MPI	9

$A=33(5)$ gas MPI⁹

$\tilde{C}^2A'_1(3s)$ D_{3h}
 gas AB^{8,11} $\tilde{C}-\tilde{X}$ 265–195 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a''_2	2	OPLA	569(63)	gas	AB	11

Photodissociation of CCl₃ in a pulsed supersonic beam to produce CCl₂ + Cl has been observed¹⁰ at 308 nm.

A broad emission observed in radiofrequency discharges between 420 and 700 nm, with a maximum near 490 nm, has been attributed⁷ to a transition between two electronically excited states of CCl₃.

\tilde{X}^2A_1 C_{3v} Structure: ESR⁵

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_2	2	Umbrella	290 ^b 251	gas	MPI	9
e	3	CCl stretch	908.5 898vs	Ne Ar	IR IR	12 1–4,6

^aTentative symmetry assignment.

^bInversion doublet. Barrier to inversion=460(40) gas MPI⁹

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SiF₃

Unstructured emission bands between 290 and 340 nm and between 350 and 800 nm which appear with varying relative intensities on photoexcitation of SiF₄ by radiation at 99.1, 95.5, or 92.2 nm have been attributed^{4,8} to SiF₃. The 350–800 nm band may correspond to the unstructured emission centered at 632 nm which has been observed^{3,5} in the flowing afterglow of an SiF₄ discharge and in the reaction of F atoms with Si and which has been tentatively assigned to SiF₃. The lifetime for the emission between 380 and 650 nm has been found to be 3.9(7) ns.⁸ A third emission band, between 240 and 280 nm, becomes more prominent at higher excitation energies. This latter band occurs in the same spectral region as the 210–260 nm emission band system observed in a discharge through SiF₄. Although that band system was initially assigned² to SiF₃, subsequent studies⁶ have demonstrated that it is entirely contributed by SiF₂.

\tilde{X}		C_{3v}	Structure: MW ⁹			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	SiF stretch	834.9 832s	Ne Ar	IR IR	7 1
	2	Umbrella	406s	Ar	IR	1
e	3	SiF stretch	958.6 954vs	Ne Ar	IR IR	7 1
	4	Deformation	290wm	Ar	IR	1

$B_0=0.250$ MW⁹

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F₂NO

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	NO stretch	1573	Ar	IR	1
	2	NF ₂ s-stretch	761	Ar	IR	1
	3	NF ₂ scissors	705	Ar	IR	1
a''	5	NF ₂ a-stretch	813	Ar	IR	1
	6	NF ₂ deform.	553	Ar	IR	1

Reference

- E. Ya. Misochko, A. V. Akimov, I. U. Goldschleger, and C. A. Wight, *J. Am. Chem. Soc.* **120**, 11520 (1998).

PF₃⁺

\tilde{E}^2E C_{3v}
 $T_0^a \geq 61500(200)$ gas PE^{1,2}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	PF stretch	660(30)	gas	PE	2
	2	Umbrella	360(30)	gas	PE	2

\tilde{D}^2A_1 C_{3v}
 $T_0^a \geq 55000(200)$ gas PE^{1,2}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	PF stretch	690(30)	gas	PE	2
	2	Umbrella	395(30)	gas	PE	2

\tilde{C}^2E C_{3v}
 $T_0^a \geq 45500(200)$ gas PE^{1,2}

\tilde{B}^2E C_{3v}
 $T^{ab} \geq 39300(600)$ gas PE^{1,2}

\tilde{A}^2A_2 C_{3v}
 $T_0^a \geq 31220(500)$ gas PE^{1,2}

\tilde{X}^2A_1 C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	PF stretch	940.0ms	Ne	IR	5
	2	Umbrella	475(30)	gas	PE	1,2
e	3	PF stretch	465.1ms	Ne	IR	5
			1097.5s	Ne	IR	5

^aIn accord with recent photoionization studies,^{3,4} the first ionization potential of PF₃ is taken to be ≤ 11.44 eV.

^bFrom vertical ionization potential.

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- 5 C. L. Lugez, K. K. Irikura, and M. E. Jacox, *J. Chem. Phys.* **108**, 8381 (1998).

PCl₃⁺

\tilde{F}^2A_1 C_{3v}
 $T^a = 67050(320)$ gas PE^{1,3}

\tilde{E}^2E C_{3v}
 $T^a = 37840(320)$ gas PE¹⁻³

In the gas phase, an emission which results on vacuum ultraviolet excitation of PCl₃ which has its maximum near 550 nm, has been attributed⁴ to the $\tilde{E}-\tilde{C}$ transition of PCl₃⁺. Emission between 320 and 450 nm is attributed⁴ to the $\tilde{E}-\tilde{B}$ and $\tilde{E}-\tilde{A}$ transitions.

\tilde{D}^2A_1 C_{3v}
 $T^a = 30010(320)$ gas PE¹⁻³

\tilde{C}^2E C_{3v}
 $T^a = 19850(320)$ gas PE¹⁻³

\tilde{B}^2E C_{3v}
 $T^a = 12020(320)$ gas PE¹⁻³

\tilde{A}^2A_2 C_{3v}
 $T^a = 9600(320)$ gas PE¹⁻³

\tilde{X}^2A_1 C_{3v}

^aFrom vertical ionization potentials.

References

- 1 P. A. Cox, S. Evans, A. F. Orchard, N. V. Richardson, and P. J. Roberts, *Faraday Discuss. Chem. Soc.* **54**, 26 (1972).
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PBr₃⁺

\tilde{E}^2E C_{3v}
 $T^a = 33560(320)$ gas PE¹⁻³

Unstructured emission between 500 and 600 nm which appears on vacuum ultraviolet excitation of gas-phase PBr₃ has been attributed⁴ to the $\tilde{E}-\tilde{C}$ transition of PBr₃⁺. Emission between 320 and 450 nm has been attributed⁴ to the $\tilde{E}-\tilde{B}$ and $\tilde{E}-\tilde{A}$ transitions.

\tilde{D}^2A_1 C_{3v}
 $T^a = 25580(320)$ gas PE¹⁻³

\tilde{C}^2E C_{3v}
 $T^a = 14760(320)$ gas PE¹⁻³

\tilde{B}^2E C_{3v}
 $T^a = 8390(320)$ gas PE¹⁻³
 $A = 2660(320)$ gas PE¹⁻³

\tilde{A}^2A_2 C_{3v}
 $T^a = 5240(320)$ gas PE¹⁻³

\tilde{X}^2A_1 C_{3v}

^aFrom vertical ionization potentials.

References

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cyc-O₄⁻

An unstructured absorption which appears near 270 nm in Ar:O₂ samples in which an alkali metal is also present and which grows on controlled warmup of the sample has been attributed⁶ to M⁺O₄⁻.

Threshold for photodestruction near 900 nm, and increasing cross section for photodestruction, probably by photodetachment, in the 850–400 nm spectral region.⁷ Photoelectron studies⁸ suggest that both photodetachment and photodissociation occur at 532 and 355 nm.

\tilde{X}		D _{2h}	Structure: IR ¹¹ MO ¹²			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a _g	1	O=O stretch	1292 K	Ar	IR	9
	2	O··O stretch	287 Cs	Ar	Ra	5
			298 Rb	Ar	Ra	5
			305 K	Ar	Ra	4
b _{1u}	5	O=O a-stretch	973.1	Ne	IR	10
			953.8	Ar	IR	11
			1001 Cs	Ar	IR	3
			992 Rb	Ar	IR	2
			993 K	Ar	IR	2,9
			1001 Na	Ar	IR	1,3
			991 Na	Ar	IR	1,3
b _{2u}	6	Asym. bend	990.8	N ₂	IR	11
			131 K	Ar	IR	9

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¹¹G. V. Chertihin and L. Andrews, *J. Chem. Phys.* **108**, 6404 (1998).
¹²A. J. A. Aquino, P. R. Taylor, and S. P. Walch, *J. Chem. Phys.* **114**, 3010 (2001).

CISO₂

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	SO ₂ s-stretch	1098.2s	Ar	IR	1
			1098.5s	Kr	IR	1
	2	OSO bend	497.7ms	Ar	IR	1
a''	5	SO ₂ a-stretch	497.0ms	Kr	IR	1
			456.1m	Ar	IR	1
			454.2m	Kr	IR	1
			1309.6vs	Ar	IR	1
			1309.5vs	Kr	IR	1

Reference

- ¹M. Bahou, S.-F. Chen, and Y.-P. Lee, *J. Phys. Chem. A* **104**, 3613 (2000).

F₂SO⁺

\tilde{E}		C _s				
T ₀ = 47120(320)		gas	PE ¹			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	SF ₂ stretch	705(40)	gas	PE	1
	4	SF ₂ scissors	390(40)	gas	PE	1
\tilde{D}^2A'		C _s				
T ^a = 38570(320)		gas	PE ^{1,2}			
\tilde{C}^2A''		C _s				
T ^a = 34800(1000)		gas	PE ^{1,2}			
\tilde{B}^2A''		C _s				
T ₀ = 18960(320)		gas	PE ^{1,2}			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	SO stretch	1180(40)	gas	PE	1,2
	2	SF ₂ s-stretch	790(40)	gas	PE	1,2
	4	SF ₂ scissors	350(40)	gas	PE	1,2
\tilde{A}^2A''		C _s				
T ^a = 15330(500)		gas	PE ^{1,2}			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	SO stretch	1000T	gas	PE	2
\tilde{X}^2A'		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	3	FSO s-deform.	420(40)	gas	PE	1,2
			455.6	Ne	IR	3

^aFrom vertical ionization potential.

References

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F₂PO⁻

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	PO stretch	1148.8wmT	Ne	IR	1
	2	PF stretch	685.7wT	Ne	IR	1

Reference

- ¹C. L. Lugez, K. K. Irikura, and M. E. Jacox, J. Chem. Phys. **108**, 8381 (1998).

ClO₃⁻

Threshold for electron detachment from ground-state ClO₃⁻ = 34290(800) gas PE¹

Reference

- ¹X.-B. Wang and L.-S. Wang, J. Chem. Phys. **113**, 10928 (2000).

ClOOCl

An unstructured gas-phase absorption with maximum at 40800 (245 nm) has been assigned^{1,4-6,8} to ClOOCl. Cl atoms have been detected⁶ on irradiation of ClOOCl in this band.

In an argon matrix, photodecomposition occurs at wavelengths shorter than 360 nm (27800).⁷

\tilde{X}		C ₂		Structure: MW ³		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a	1	OO stretch	750	gas	IR	4,9
			754.0m	Ar	IR	2,7
	2	ClO s-stretch	560	gas	IR	4
543.0m			Ar	IR	7	
b	5	ClO a-stretch	653	gas	IR	4,9
			647.7vs	Ar	IR	2,7
	6	Deformation	418.5w	Ar	IR	7

A₀ = 0.437; B₀ = 0.080; C₀ = 0.071 MW³

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⁹A. S. Brust, F. Zabel, and K. H. Becker, Geophys. Res. Lett. **24**, 1395 (1997).

ClClO₂

In the gas phase, prominent, unstructured absorption maxima appear²⁻⁴ at 44200 and 33800 (226 and 296 nm). In a neon matrix, unstructured absorption maxima are observed^{1,3,4} at 42700 and 34100 (234 and 293 nm), with half band widths of 40 and 45 nm, respectively.

In an argon matrix, the threshold⁴ for photodecomposition into Cl+OClO, followed by cage recombination to form ClOClO, lies near 16400 (610 nm). The maximum for this process⁵ lies near 33300 (300 nm).

\tilde{X}		C _s		Structure: MW ^{6,7}		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	ClO ₂ s-stretch	1041.5(5)s	gas	IR	2,3
			1041.2s	Ne	IR	1,3
			1040.7	Ar	IR	5
2	ClO ₂ scissors	522.7(5)wm	gas	IR	2,3	
		522.5wm	Ne	IR	1,3	
		440.5(5)s	gas	IR	2,3	
3	ClCl stretch	440.4s	Ne	IR	1,3	
		271.4wm	Ne	IR	1,3	
		1218.2(5)vs	gas	IR	2,3	
a''	5	ClO ₂ a-stretch	1216.4vs	Ne	IR	1,3
			1213.2	Ar	IR	5
			251.4vw	Ne	IR	1,3
6	ClO ₂ rock					

A₀ = 0.315; B₀ = 0.120; C₀ = 0.093 MW^{6,7}

References

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⁶H. S. P. Müller and E. A. Cohen, J. Phys. Chem. A **101**, 3049 (1997).
⁷H. S. P. Müller, E. A. Cohen, and D. Christen, J. Chem. Phys. **110**, 11865 (1999).

SF₃⁺

\tilde{X}		C _{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	SF stretch	944.2	Ne	IR	1
	2	Deformation	540.7T	Ne	IR	1
e	3	SF stretch	969.0	Ne	IR	1

Reference

¹C. L. Lugez, M. E. Jacox, R. A. King, and H. F. Schaefer III, J. Chem. Phys. **108**, 9639 (1998).

PF₃⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b ₂	5	PF stretch	470.9	Ne	IR	1

Reference

¹C. L. Lugez, K. K. Irikura, and M. E. Jacox, J. Chem. Phys. **108**, 8381 (1998).

8.8. Five-Atomic Tetra- and Trihydrides**Pd(H₂)₂**

\tilde{X} D _{2d}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b ₂	4	HH stretch	3038	Ar	IR	1
	5	PdH stretch	792.4	Ne	IR	1
			778.0	Ar	IR	1

Pd(D₂)₂

\tilde{X} D _{2d}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b ₂	4	DD stretch	2160	Ar	IR	1
	5	PdD stretch	597.5	Ne	IR	1
			585.7	Ar	IR	1

Reference

¹L. Andrews, X. Wang, M. E. Alikhani, and L. Manceron, J. Phys. Chem. A **105**, 3052 (2001).

CeH₄

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Sym. stretch	1500.5	Ar	IR	1
		Asym. stretch	1441.8	Ar	IR	1

CeD₄

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Sym. stretch	1070.2	Ar	IR	1
		Asym. stretch	1032.1	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **104**, 1640 (2000).

NdH₄

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Sym. stretch	1384.6	Ar	IR	1
		Asym. stretch	1304.3	Ar	IR	1
			1300.7			

NdD₄

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Sym. stretch	986.3	Ar	IR	1
		Asym. stretch	935.5	Ar	IR	1
			931.6			

Reference

¹S. P. Willson and L. Andrews, J. Phys. Chem. A **104**, 1640 (2000).

SmH₄

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Asym. stretch	1331.4	Ar	IR	1
			1328.7			

SmD₄

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Asym. stretch	954.1	Ar	IR	1
			952.1			

Reference

¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **104**, 1640 (2000).

DyH₄ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Asym. stretch	1386.6	Ar	IR	1

DyD₄ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Asym. stretch	992.9	Ar	IR	1

Reference

¹S. P. Willson and L. Andrews, *J. Phys. Chem. A* **104**, 1640 (2000).

CH₄⁺

\tilde{C}^2A_1 T_d
 $T_0 = 78870(160)^a$ gas PE^{3,6}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CH stretch	2190(80)	gas	PE	6

\tilde{B}^b $C_s?$
 $T^c \cong 19240$ gas PE^{1-3,5,6}

\tilde{A}^b $C_s?$
 $T_0 \leq 13350^d$ gas PE^{1-3,5,6}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1300(100)	gas	PE	5

$\tilde{X}^2B_2^b$ C_{2v} Structure: ESR⁷MO^{8,10}TPE¹¹⁻¹³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1700(100) ^d	gas	PE	5
			1200(100)	gas	PE	3,5

Evidence has been obtained¹¹ for the occurrence of pseudorotation.

CD₄⁺

\tilde{C}^2A_1 T_d
 $T_0 = 79400(200)$ gas PE⁶

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CD stretch	1460(80)	gas	PE	3,6

$\tilde{X}^2B_2^b$ C_{2v}
 $A_0 = 3.368(59)$; $B_0 = 2.640(108)$; $C_0 = 1.967(54)$ TPE¹³

^aBased on adiabatic ionization potential of 12.615(10) eV for CH₄^{4,5} and of 12.658(15) eV for CD₄.⁹

^bResulting from Jahn-Teller distortion of the ground ²F₂ state of CH₄⁺.

^cFrom vertical ionization potential.

^dTwo progressions, with onset near 3670.

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NH₄

$3p^2F_2$ T_d
 $T_0 \cong 15078^{ab}$ gas EM^{1,2,4} $3p^2F_2 - 3s^2A_1$ 663.5 nm Diffuse.

$3s^2A_1$ T_d Structure: TPE⁸
 gas EM^{1,2} TPE⁸ $3p^2F_2 - 3s^2A_1$ 663.5 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	NH stretch	2552T	gas	EM	1,4
e	2	Deformation	1581T	gas	EM	1,4

ND₄

$3p^2F_2$ T_d
 $T_0^a = 14828.285(4)$ gas EM^{1,2,4} AB^{3,5} $3p^2F_2 - 3s^2A_1$ 675 nm
 Three weak bands have been observed⁴ 775, 1138, and 1722 cm⁻¹ above the band origin. However, the assignment of these bands has not yet been established.

$B_0 = 2.937$ gas TPE⁸

$3s^2A_1$ T_d
 gas EM^{1,2,4,7} $3p^2F_2 - 3s^2A_1$ 675 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	ND stretch	1960T	gas	EM	1,4
e	2	Deformation	1080.25(7)	gas	EM	7

$\tau \approx 30 \mu s$ gas AB⁵
 $B_0 = 2.856(4)$ gas TPE⁸

^aMeasured with respect to the lowest Rydberg state, $3s^2A_1$. The ground state is dissociative.

^bEstimated⁵ by scaling of data for ND₄.

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LiCH₃

\tilde{X}^1A_1 C_{3v} Structure: MW²⁻⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CH ₃ s-stretch	2780	Ar	IR	1
	2	CH ₃ umbrella	1158	Ar	IR	1
	3	CLi stretch	530T	Ar	IR	1
e	4	CH ₃ a-stretch	2820	Ar	IR	1
	5	CH ₃ deformation	1387	Ar	IR	1
	6	HCLi deformation	408.5	Ar	IR	1

$B_0 = 0.765$ MW^{2,3}

LiCD₃

\tilde{X}^1A_1 C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CD ₃ s-stretch	2030	Ar	IR	1
	2	CD ₃ umbrella	883	Ar	IR	1
	3	CLi stretch	510	Ar	IR	1
e	5	CD ₃ deformation	1027	Ar	IR	1
	6	DCLi deformation	319	Ar	IR	1

$B_0 = 0.643$ MW^{2,3}

References

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- ²D. B. Grotjahn, T. C. Pesch, J. Xin, and L. M. Ziurys, J. Am. Chem. Soc. **119**, 12368 (1997).
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NaCH₃

\tilde{X}^1A_1 C_{3v} Structure: MW^{2,3}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CH ₃ s-stretch	2760s	N ₂	IR	1
	2	CH ₃ s-deform.	1092w	N ₂	IR	1
	3	CNa stretch	298s	N ₂	IR	1
e	4	CH ₃ a-stretch	2805s	N ₂	IR	1
	5	CH ₃ a-deform.	1384w	N ₂	IR	1
	6	HCNa bend	362s	N ₂	IR	1

$B_0 = 0.316$ MW^{2,3}

NaCD₃

\tilde{X}^1A_1 C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CD ₃ s-stretch	2019m	N ₂	IR	1
	2	CD ₃ s-deform.	836m	N ₂	IR	1
	3	CNa stretch	278m	N ₂	IR	1
e	4	CD ₃ a-stretch	2123m	N ₂	IR	1
	5	CD ₃ a-deform.	972w	N ₂	IR	1
	6	DCNa bend	285s	N ₂	IR	1

$B_0 = 0.266$ MW^{2,3}

References

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KCH₃

\tilde{X} C_{3v} Structure: MW²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CH ₃ s-stretch	2732s	N ₂	IR	1
	2	CH ₃ s-deform.	1053w	N ₂	IR	1
	3	CK stretch	280	N ₂	IR	1
e	4	CH ₃ a-stretch	2775s	N ₂	IR	1
	5	CH ₃ a-deform.	1384w	N ₂	IR	1
	6	HCK bend	307m	N ₂	IR	1

$B_0 = 0.205$ MW²

KCD₃

\tilde{X} C _{3v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CD ₃ s-stretch	1994m	N ₂	IR	1
	2	CD ₃ s-deform.	807m	N ₂	IR	1
	3	CK stretch	259s	N ₂	IR	1
<i>e</i>	4	CD ₃ a-stretch	2101m	N ₂	IR	1
	5	CD ₃ a-deform.	967w	N ₂	IR	1
	6	DCK bend	237m	N ₂	IR	1

$$B_0 = 0.170 \text{ MW}^2$$

References

- ¹K. Burczyk and A. J. Downs, J. Chem. Soc., Dalton Trans. 2351 (1990).
²D. B. Grotjahn, T. C. Pesch, M. A. Brewster, and L. M. Ziurys, J. Am. Chem. Soc. **122**, 4735 (2000).

MgCH₃⁺

\tilde{X}^1A_1 C _{3v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	2	CH ₃ umbrella	1020	gas	TPE	1
	3	MgC stretch	516	gas	TPE	1
<i>e</i>	6	HCMg deform.	673	gas	TPE	1

Reference

- ¹T. A. Barckholtz, D. E. Powers, T. A. Miller, and B. E. Bursten, J. Am. Chem. Soc. **121**, 2576 (1999).

ZnCH₃⁺

\tilde{X}^1A_1 C _{3v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	2	CH ₃ umbrella	1109	gas	TPE	1
	3	ZnC stretch	482	gas	TPE	1
<i>e</i>	6	HCZn deform.	760	gas	TPE	1

Reference

- ¹T. A. Barckholtz, D. E. Powers, T. A. Miller, and B. E. Bursten, J. Am. Chem. Soc. **121**, 2576 (1999).

CdCH₃⁺

\tilde{X}^1A_1 C _{3v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	2	CH ₃ deformation	1089(2)	gas	TPE	1
	3	CdC stretch	422(2)	gas	TPE	1
<i>e</i>	5	CH ₃ deformation	1386(2)	gas	TPE	1
	6	Deformation	752(2)	gas	TPE	1

Reference

- ¹S. I. Panov, D. E. Powers, and T. A. Miller, J. Chem. Phys. **108**, 1335 (1998).

BeCH₃

\tilde{X} C _{3v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	2	CH ₃ s-deform.	1180.6 1178.9	Ar	IR	1
	3	BeC stretch	851.7	Ar	IR	1

BeCD₃

\tilde{X} C _{3v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	2	CD ₃ s-deform.	962.1	Ar	IR	1
	3	BeC stretch	765.6	Ar	IR	1

Reference

- ¹T. M. Greene, D. V. Lanzisera, L. Andrews, and A. J. Downs, J. Am. Chem. Soc. **120**, 6097 (1998).

CH₂BeH

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	2	BeH stretch	2071.9	Ar	IR	1
	3	CH ₂ scissors	1388.5	Ar	IR	1
<i>b</i> ₁	5	H ₂ CBe OPLA	660.0	Ar	IR	1
<i>b</i> ₂	9	HCBBe deform.	577.6T	Ar	IR	1

CD₂BeD

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	BeD stretch	1578.2	Ar	IR	1
b ₁	5	D ₂ CBe OPLA	541.8	Ar	IR	1
b ₂	8	CD ₂ deform.	577.4	Ar	IR	1

Reference

¹T. M. Greene, D. V. Lanzisera, L. Andrews, and A. J. Downs, J. Am. Chem. Soc. **120**, 6097 (1998).

BaCH₃

\tilde{X}^2A_1 C_{3v} Structure: MW¹
B₀ = 0.174 MW¹

Reference

¹J. Xin, J. S. Robinson, A. J. Apponi, and L. M. Ziurys, J. Chem. Phys. **108**, 2703 (1998).

MgCH₃

\tilde{A}^2E C_{3v} Structure: LF¹
T₀ = 20030.296(2) gas LF^{1,3}MPI⁴ $\tilde{A}-\tilde{X}$ 463–557 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	CH ₃ umbrella	997.5	gas	LF,MPI	3,4
	3	MgC stretch	464	gas	MPI	4
e	6		669.1(2)	gas	LF,MPI	3,4

τ₀ = 60(20) ns gas LF³
A = 28.591 gas LF³
A₀ = 4.992; B₀ = 0.365 LF¹

\tilde{X}^2A_1		C _{3v} Structure: LF ¹ MW ²				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2		509	gas	LF	3
e	6		1072	gas	LF	3

A₀ = 5.222; B₀ = 0.367 LF^{1,3}MW²

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CaCH₃

\tilde{B}^2A_1 C_{3v}
T₀ = 16003(10) gas LF¹ $\tilde{B}-\tilde{X}$ 620–630 nm

\tilde{A}^2E C_{3v}
T₀ = 14743.382^a gas LF^{1,2,4} $\tilde{A}-\tilde{X}$ 630–730 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	CH ₃ umbrella	1048(10)	gas	LF	1
	3	CaC stretch	413(10)	gas	LF	1
e	6	CaCH deform.	391(5)H	gas	LF	1

A = 72.71 gas LF^{1,2,4}
A₀ = 5.386; B₀ = 0.254 LF^{2,4}

\tilde{X}^2A_1 C_{3v} Structure: LF²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	CH ₃ umbrella	1085(10)	gas	LF	1
	3	CaC stretch	419(10)	gas	LF	1
e	6	CaCH deform.	319(5)H	gas	LF	1

A₀ = 5.448; B₀ = 0.252 LF^{2,4}MW³DR⁵

^aPredissociated above ~16200.¹

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⁵K. C. Namiki and T. C. Steimle, J. Chem. Phys. **110**, 11309 (1999).

ZnCH₃

$\tilde{C}^2A_1^a$ C_{3v}
T₀ = 36510 gas AB¹ $\tilde{C}-\tilde{X}$ 260–274 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	CH ₃ deform.	950T	gas	AB	1

\tilde{A}^2E C_{3v} Structure: LF⁶
T₀ = 24082.82 gas AB¹EM²LF^{5,6}MPI⁷ $\tilde{A}-\tilde{X}$ 379–437 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	CH ₃ deform.	1060	gas	AB,LF MPI	1,4,5,7
	3	ZnC stretch	467	gas	LF,MPI	5,7
e	6	HCZn deform.	749	gas	MPI	7

τ = 40(3) ns gas EM³LF⁴
A = 253 gas AB¹EM^{2,3}LF^{4,5}
A₀ = 4.949; B₀ = 0.317 LF⁶

\tilde{X}^2A_1		C_{3v}		Structure: LF ⁶		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CH ₃ deform.	1064	gas	EM,LF	3–5
	3	ZnC stretch	445	gas	LF	4,5
e	6	ZnCH deform.	315T	gas	LF	5

$$A_0 \cong 5.15; B_0 = 0.310 \text{ LF}^6$$

^aAssignment to \tilde{C} state suggested⁴ by analogy with ZnH.

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- T. M. Cerny, X. Q. Tan, J. M. Williamson, E. S. J. Robles, A. M. Ellis, and T. A. Miller, *J. Chem. Phys.* **99**, 9376 (1993).
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CdCH₃

\tilde{C}^2A_1		C_{3v}		$\tilde{C}-\tilde{X}$ 264–287 nm		
$T_0 = 34916$ gas AB ¹						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CH ₃ deform.	960T	gas	AB	1

$\tilde{A}^2E_{1/2}$		C_{3v}		Structure: LF ⁸		
$T_0 = 22514$ gas AB ¹ EM ²⁻⁵ LF ⁶⁻⁹ MPI ¹⁰ FD ¹¹				$\tilde{A}-\tilde{X}$ 400–458 nm		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CH ₃ deform.	1019(3)	gas	LF,MPI	7,8,10,11
	3	CdC stretch	400(3)	gas	LF,MPI	6–8,10,11
e	6	Deformation	638(2)	gas	MPI,FD	10,11

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CH ₃ deform.	1019(3)	gas	LF,MPI	7,8,10,11
	3	CdC stretch	400(3)	gas	LF,MPI	6–8,10,11
e	6	Deformation	638(2)	gas	MPI,FD	10,11

$\tau = 70(4)$ ns gas EM^{2,4}LF⁷
 $A = 1008$ gas FD¹¹
 $A_0 = 4.87; B_0 = 0.245 \text{ LF}^8\text{MPI}^{10}$

\tilde{X}^2A_1		C_{3v}		Structure: LF ⁸		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CH ₃ deform.	1000(3)	gas	EM,LF	5–7
	3	CdC stretch	355(3)	gas	EM,LF	5–7

$A_0 = 5.157; B_0 = 0.234 \text{ LF}^8$

References

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¹⁰S. I. Panov, D. E. Powers, and T. A. Miller, *J. Chem. Phys.* **108**, 1335 (1998).

¹¹M. B. Pushkarsky, T. A. Barckholtz, and T. A. Miller, *J. Chem. Phys.* **110**, 2016 (1999).

GaCH₃

\tilde{X}		C_{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CH ₃ stretch	2986.9	Ar	IR	1,2
	2	CH ₃ umbrella	1147.9	Ar	IR	1–3
	3	GaC stretch	476.2s	Ar	IR	1–3
e	5	CH ₃ deform.	1403.9	Ar	IR	1,3

GaCD₃

\tilde{X}		C_{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CD ₃ stretch	2200.4	Ar	IR	1,2
	2	CD ₃ umbrella	902.8	Ar	IR	1,2
e	5	CD ₃ deform.	1025.4	Ar	IR	1,2

References

¹H.-J. Himmel, A. J. Downs, T. M. Greene, and L. Andrews, *Chem. Commun.* 2243 (1999).

²H.-J. Himmel, A. J. Downs, T. M. Greene, and L. Andrews, *Organomet.* **19**, 1060 (2000).

³J. Müller, H. Sternkicker, U. Bergmann, and B. Atakan, *J. Phys. Chem. A* **104**, 3627 (2000).

InCH₃

\tilde{X}		C_{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CH ₃ stretch	2905.2	Ar	IR	1,2
	2	CH ₃ umbrella	1115.3	Ar	IR	1,2
	3	InC stretch	422.1	Ar	IR	1,2
e	4	CH ₃ stretch	2976.2	Ar	IR	1,2
	5	CH ₃ deform.	1424.0	Ar	IR	1,2

InCD₃

\tilde{X} C _{3v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	CD ₃ umbrella	868.1	Ar	IR	1,2

References

- ¹H.-J. Himmel, A. J. Downs, T. M. Greene, and L. Andrews, Chem. Commun. 2243 (1999).
²H.-J. Himmel, A. J. Downs, T. M. Greene, and L. Andrews, Organomet. **19**, 1060 (2000).

AlNH₃⁺

\tilde{X} C _{3v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	3	AlN stretch	333	gas	TPE	1
e	6	Deformation	557	gas	TPE	1

AlND₃⁺

\tilde{X} C _{3v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	3	AlN stretch	325(5)	gas	MPI	2

References

- ¹D.-S. Yang and J. Miyawaki, Chem. Phys. Lett. **313**, 514 (1999).
²Z. J. Jakubek and B. Simard, J. Chem. Phys. **112**, 1733 (2000).

GaNH₃⁺

\tilde{X} C _{3v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	3	GaN stretch	266	gas	TPE	1

Reference

- ¹S. Li, G. K. Rothschof, D. Pillai, B. R. Sohnlein, B. M. Wilson, and D.-S. Yang, J. Chem. Phys. **115**, 7968 (2001).

InNH₃⁺

\tilde{X} C _{3v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	3	InN stretch	234	gas	TPE	1

Reference

- ¹G. K. Rothschof, J. S. Perkins, S. Li, and D.-S. Yang, J. Phys. Chem. A **104**, 3178 (2000).

AlNH₃

In an argon matrix, an absorption maximum at 23360 (428 nm) can be assigned³ to AlNH₃. Irradiation near this frequency leads to the formation of HAlNH₂.

$\tilde{X}^2E_{1/2}$ C _{3v} (C _s) ^a						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	NH ₃ a-stretch	3447.1	Ar	IR	3
	4	NH ₃ umbrella	1131.4	Ar	IR	3
	6	AlN stretch	227	gas	TPE	1
a''	8	NH ₃ a-deform.	1593.6	Ar	IR	3

A = 58 gas TPE¹

AlND₃

\tilde{C}^2E C _{3v}						
T ₀ = 21185(5) gas MPI ² $\tilde{C}-\tilde{X}$ 406–472 nm						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	3	AlN stretch	428(2) (ω)	gas	MPI	2
e	6	Deformation	594(12)T (ω)	gas	MPI	2

\tilde{B}^2A_1 C _{3v}						
T ₀ = 18532.5(7) gas MPI ² $\tilde{B}-\tilde{X}$ 468–542 nm						

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	3	AlN stretch	316.0(8) (ω)	gas	MPI	2

$\tilde{X}^2E_{1/2}$ C _{3v} (C _s) ^a						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	4	ND ₃ umbrella	972.6	Ar	IR	3
	6	AlN stretch	325(5)	gas	MPI	2

A = 55.8(7) gas MPI²

^aJahn–Teller interaction leads to a slight distortion of the molecule, with resultant C_s symmetry. Vibrational fundamentals are numbered appropriately for the distorted molecule.

References

- ¹D.-S. Yang and J. Miyawaki, Chem. Phys. Lett. **313**, 514 (1999).
²Z. J. Jakubek and B. Simard, J. Chem. Phys. **112**, 1733 (2000).

³H.-J. Himmel, A. J. Downs, and T. M. Greene, *J. Am. Chem. Soc.* **122**, 9793 (2000).

HAlNH₂

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	NH ₂ s-stretch	3476.4	Ar	IR	2
	3	AlH stretch	1761.7	Ar	IR	1,2
	4	NH ₂ scissors	1533.6	Ar	IR	1,2
	5	AlN stretch	778.7	Ar	IR	1,2
	6	NH ₂ rock	705.2	Ar	IR	1,2
	7	AlH deform.	482.2	Ar	IR	2
	a''	8	NH ₂ OPLA	483.8	Ar	IR
9		AlH OPLA	393.8	Ar	IR	2

DAIND₂

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	ND ₂ s-stretch	2595.6	Ar	IR	2
	3	AID stretch	1282.7	Ar	IR	1,2
	4	ND ₂ scissors	1151.4	Ar	IR	1,2
	5	AlN stretch	748.3	Ar	IR	1,2
	6	ND ₂ rock	549.8	Ar	IR	1,2
a''	8	ND ₂ OPLA	346.5	Ar	IR	2
	9	AID OPLA	304.6	Ar	IR	2

References

- ¹D. V. Lanzisera and L. Andrews, *J. Phys. Chem. A* **101**, 5082 (1997).
²H.-J. Himmel, A. J. Downs, and T. M. Greene, *J. Am. Chem. Soc.* **122**, 9793 (2000).

GaNH₃

In an argon matrix, an absorption maximum at 22730 (440 nm) can be assigned^{1,2} to GaNH₃. Irradiation near this frequency leads to the formation of HGaNH₂.

\tilde{X}		$C_{3v}(C_s)^a$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	NH ₃ a-stretch	3441.5	Ar	IR	2
	4	NH ₃ umbrella	1104.2	Ar	IR	1,2
	6	GaN stretch	161	gas	TPE	3
a''	8	NH ₃ a-deform.	1580.7	Ar	IR	2

GaND₃

\tilde{X}		$C_{3v}(C_s)^a$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	4	ND ₃ umbrella	824.9	Ar	IR	2

^aJahn–Teller interaction leads to a slight distortion of the molecule, with resultant C_s symmetry. Vibrational fundamentals are numbered appropriately for the distorted molecule.

References

- ¹H.-J. Himmel, A. J. Downs, and T. M. Greene, *Chem. Commun.* 871 (2000).
²H.-J. Himmel, A. J. Downs, and T. M. Greene, *J. Am. Chem. Soc.* **122**, 9793 (2000).
³S. Li, G. K. Rothschof, D. Pillai, B. R. Sohnlein, B. M. Wilson, and D.-S. Yang, *J. Chem. Phys.* **115**, 7968 (2001).

HGaNH₂

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	3	GaH stretch	1721.8	Ar	IR	1,2
	4	NH ₂ scissors	1528.7	Ar	IR	1,2
	5	NH ₂ rock	746.2	Ar	IR	1,2
a''	6	GaN stretch	668.5	Ar	IR	1,2
	8	NH ₂ OPLA	494.1	Ar	IR	1,2
	9	GaH OPLA	210.9	Ar	IR	1,2

DGaND₂

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	3	GaD stretch	1249.5	Ar	IR	2
	4	ND ₂ scissors	1145.0	Ar	IR	2
	5	GaN stretch	626.6	Ar	IR	2
	6	ND ₂ rock	563.5	Ar	IR	2

References

- ¹H.-J. Himmel, A. J. Downs, and T. M. Greene, *Chem. Commun.* 871 (2000).
²H.-J. Himmel, A. J. Downs, and T. M. Greene, *J. Am. Chem. Soc.* **122**, 9793 (2000).

InNH₃

In an argon matrix, an absorption maximum at 22990 (435 nm) can be assigned¹ to InNH₃. Irradiation near this frequency leads to the formation of HInNH₂.

\tilde{X}		C _{3v} (C _s)				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	NH ₃ s-stretch	3424.4	Ar	IR	1
	4	NH ₃ umbrella	1082.9	Ar	IR	1
	6	InN stretch	141	gas	TPE	2

InND₃

\tilde{X}		C _{3v} (C _s) ^a				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	4	ND ₃ umbrella	816.0	Ar	IR	1

^aJahn–Teller interaction leads to a slight distortion of the molecule, with resultant C_s symmetry. Vibrational fundamentals are numbered appropriately for the distorted molecule.

References

- ¹H.-J. Himmel, A. J. Downs, and T. M. Greene, *J. Am. Chem. Soc.* **122**, 9793 (2000).
²G. K. Rothschof, J. S. Perkins, S. Li, and D.-S. Yang, *J. Phys. Chem. A* **104**, 8178 (2000).

HInNH₂

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	NH ₂ s-stretch	3463.5	Ar	IR	1
	3	InH stretch	1533.8	Ar	IR	1
	4	NH ₂ scissors	1512.9	Ar	IR	1
	5	NH ₂ rock	709.0	Ar	IR	1
	6	InN stretch	564.8	Ar	IR	1

DInND₂

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	3	ND ₂ scissors	1122.2	Ar	IR	1
	4	InD stretch	1108.8	Ar	IR	1
	5	InN stretch	528.8	Ar	IR	1
	6	ND ₂ rock	522.9	Ar	IR	1

Reference

- ¹H.-J. Himmel, A. J. Downs, and T. M. Greene, *J. Am. Chem. Soc.* **122**, 9793 (2000).

AIPH₃

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	2	PH s-stretch	2285.5	Ar	IR	1
	3	PH ₃ a-deform.	1101.2	Ar	IR	1
	4	PH ₃ s-deform.	974.7	Ar	IR	1

AIPD₃

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	2	PD s-stretch	1659.4	Ar	IR	1
	3	PD ₃ a-deform.	793.7	Ar	IR	1
	4	PD ₃ s-deform.	718.3	Ar	IR	1

Reference

- ¹H.-J. Himmel, A. J. Downs, and T. M. Greene, *Inorg. Chem.* **40**, 396 (2001).

HAIPH₂

In argon-matrix experiments,¹ an absorption maximum at 18180 (550 nm) can be assigned to HAIPH₂.

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	3	AIH stretch	1768.2	Ar	IR	1
	4	PH ₂ scissors	1159.4	Ar	IR	1
	5	PH ₂ wag	727.1	Ar	IR	1
	7	HAIP deform.	403.9	Ar	IR	1

DAIPD₂

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	3	AID stretch	1288.4	Ar	IR	1
	5	PD ₂ wag	532.3	Ar	IR	1

Reference

- ¹H.-J. Himmel, A. J. Downs, and T. M. Greene, *Inorg. Chem.* **40**, 396 (2001).

H₂AlPH \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	2	AlH ₂ a-stretch	1874.7	Ar	IR	1
	3	AlH ₂ s-stretch	1866.1	Ar	IR	1
	4	AlH ₂ bend	765.9	Ar	IR	1
	5	HPAl bend	606.3	Ar	IR	1
	6	AlH ₂ wag	569.0	Ar	IR	1

D₂AlPD \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	2	AlD ₂ a-stretch	1373.5	Ar	IR	1
	3	AlD ₂ s-stretch	1345.6	Ar	IR	1
	4	AlD ₂ bend	564.5	Ar	IR	1
	5	DPAl bend	482.8	Ar	IR	1

Reference

¹H.-J. Himmel, A. J. Downs, and T. M. Greene, *Inorg. Chem.* **40**, 396 (2001).

GaPH₃ \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	2	PH s-stretch	2280.8	Ar	IR	1
	3	PH ₃ a-deform.	1108.2	Ar	IR	1
	4	PH ₃ s-deform.	973.6	Ar	IR	1

GaPD₃ \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	2	PD s-stretch	1660.3	Ar	IR	1
	3	PD ₃ a-deform.	795.9	Ar	IR	1
	4	PD ₃ s-deform.	717.7	Ar	IR	1

Reference

¹H.-J. Himmel, A. J. Downs, and T. M. Greene, *Inorg. Chem.* **40**, 396 (2001).

HGaPH₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	3	GaH stretch	1721.4	Ar	IR	1
	4	PH ₂ scissors	1060.9	Ar	IR	1
	7	HGaP deform.	428.2	Ar	IR	1

DGaPD₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	3	GaD stretch	1244.7	Ar	IR	1

Reference

¹H.-J. Himmel, A. J. Downs, and T. M. Greene, *Inorg. Chem.* **40**, 396 (2001).

H₂GaPH \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	2	GaH ₂ a-stretch	1897.5	Ar	IR	1
	3	GaH ₂ s-stretch	1893.3	Ar	IR	1
	4	GaH ₂ bend	738.9	Ar	IR	1
	5	HPGa bend	646.5	Ar	IR	1
			644.8			
	6	GaH ₂ wag	454.8	Ar	IR	1

D₂GaPD \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	2	GaD ₂ a-stretch	1372.8	Ar	IR	1
	3	GaD ₂ s-stretch	1360.1	Ar	IR	1
	4	GaD ₂ bend	528.8	Ar	IR	1
	5	DPGa bend	486.4	Ar	IR	1

Reference

¹H.-J. Himmel, A. J. Downs, and T. M. Greene, *Inorg. Chem.* **40**, 396 (2001).

InPH₃ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	4	PH ₃ a-deform.	1105.7	Ar	IR	1
	6	PH ₃ s-deform.	974.4	Ar	IR	1

InPD₃ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	6	PD ₃ s-deform.	717.6	Ar	IR	1

Reference

¹H.-J. Himmel, A. J. Downs, and T. M. Greene, *Inorg. Chem.* **40**, 396 (2001).

HInPH₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	2	PH ₂ a-stretch	2299.4	Ar	IR	1
	3	InH stretch	1546.4	Ar	IR	1

DInPD₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	3	InD stretch	1114.9	Ar	IR	1

Reference

¹H.-J. Himmel, A. J. Downs, and T. M. Greene, *Inorg. Chem.* **40**, 396 (2001).

H₂InPH \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	5	HPIn bend	674.7	Ar	IR	1

Reference

¹H.-J. Himmel, A. J. Downs, and T. M. Greene, *Inorg. Chem.* **40**, 396 (2001).

C₂H₃**Rydberg state** $T_0 = 59410$ gas AB³ 164–169 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1306	gas	AB	3

Unstructured gas-phase absorption starting near 42000 (238 nm) and continuously increasing to the 225 nm observation limit has been assigned⁶ to an in-plane $\pi^*(2a'') - \pi(1a'')$ transition of C₂H₃.

 \tilde{A}^2A'' $T_0 = 20042$ gas AB¹CR^{7,8}PD⁹ $\tilde{A} - \tilde{X}$ 400–530 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	5	Mixed	1249(2)	gas	PD	9
	6	CC stretch	1183(2)	gas	AB,PD	1,9
	7	CCH bend	934(2)	gas	AB,PD	1,9
a''	8	Torsion	1168(2)	gas	PD	9
	9	OPLA	836(2)	gas	PD	9

 \tilde{X} C_s^a

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	7		674(2)	gas	PD	9
a''	8	Mixed OPLA	895.16	gas	DL	4
			895.4	Ne	IR	5
			900	Ar	IR	2

$A_0 = 7.913$; $B_0 = 1.083$; $C_0 = 0.949$ DL⁴

C₂D₃ \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a''	8	Mixed OPLA	704	Ar	IR	2

^aRapid tunneling, giving effective C_{2v} symmetry.⁴

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CH₃N

\tilde{A}^3E C_{3v} Structure: EM⁴
 $T_0 = 31830.913(12)$ gas AB¹EM^{1,2,4,5,7}LF^{8,10} $\tilde{A}-\tilde{X}$ 288–356 nm
 31576(20) N₂ AB^{3,6} $\tilde{A}-\tilde{X}$ 281–317 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CH ₃ stretch	2943	gas	LF	8,10
	2	CH ₃ deform.	1239	gas	LF	8
			1166T	N ₂	AB	6
3	CN stretch	758(4)	gas	UV,LF	1,2,8	
		755(22)	N ₂	AB	3,6	
<i>e</i>	5	CH ₃ deform.	1500T ^a	gas	LF	8
	6	CH ₃ rock	728(4)	gas	EM	5

$\tau_0 = 415(8)$ ns gas LF^{9,10}
 $A = -22.52$ gas EM^{4,7}
 $A_0 = 5.423$; $B_0 = 0.845$ EM^{4,7}

\tilde{a}^1E C_{3v}
 $T_0 = 10905(90)$ gas PE¹¹

\tilde{X}^3A_2 C_{3v} Structure: EM^{4,7}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CH ₃ s-stretch	2943(4)	gas	EM	2,5
	2	CH ₃ deform.	1349(4)	gas	EM	2,5
			1040(4)	gas	EM,LF	2,5,10
3	CN stretch	1029	N ₂	AB	3	
		2989(4)	gas	EM	5	
<i>e</i>	4	CH ₃ a-stretch	2989(4)	gas	EM	5
	5	CH ₃ deform.	1490(4)	gas	EM	5
	6	CH ₃ rock	903(8)	gas	EM	2,5

$A_0 = 5.61$; $B_0 = 0.929$ EM^{4,7}

CD₃N

\tilde{A}^3E C_{3v}
 $T_0 = 31774.158(2)^b$ gas AB¹EM^{2,4,5} $\tilde{A}-\tilde{X}$ 294–365 nm
 31516(30) N₂ AB^{3,6} $\tilde{A}-\tilde{X}$ 277–318 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	2	CD ₃ deform.	1044T	N ₂	AB	6
	3	CN stretch	759(4)	gas	UV	1,2
695T			N ₂	AB	6	
<i>e</i>	6	CD ₃ rock	579(4) ^c	gas	EM	2

$B_0 = 0.691$ EM⁴

\tilde{X}^3A_2 C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	2	CD ₃ deform.	941(4)	gas	EM	5
	3	CN stretch	1110(4)	gas	EM	2,5
<i>e</i>	6	CD ₃ rock	749(8)	gas	EM	2,5

$B_0 = 0.744$ EM⁴

^aFrom combination bands.

^bCalculated assuming $A(\text{CD}_3\text{N}) = A(\text{CH}_3\text{N})$.

^cObserved as sequence band.

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CH₃O⁺

$^1E, ^3E$ C_{3v}
 $T^a = 35700\text{T}$ gas PE²

\tilde{b}^1A_1 C_{3v}
 $T_0 = 23720(100)$ gas PE²

\tilde{a}^1E C_{3v}
 $T_0 = 13880(100)$ gas PE²

\tilde{X}^3A_2 C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	2	CO stretch	1950(60)	gas	PE	2

CD₃O⁺

\tilde{X} C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CD ₃ s-stretch	2400T	gas	PI	1

^aFrom vertical ionization potential.

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CH₃S⁺ $T^a = 44100\text{T}$ gas PE³ $T^a = 35000\text{T}$ gas PE³ ${}^1E, {}^3E$ C_{3v}
 $T^a = 21460(100)$ gas PE³ $\tilde{b} {}^1A_1$ C_{3v}
 $T_0 = 8550(100)$ gas PE³ $\tilde{a} {}^1E$ C_{3v}
 $T_0 = 5240(100)$ gas PE³ $\tilde{X} {}^3A_2$ C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2		1100(60)	gas	PE	3
	3	CS stretch	733(5)	gas	PI,TPE	1,2

CD₃S⁺ \tilde{X} C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	3	CS stretch	730(60)	gas	PI	1

^aFrom vertical ionization potential.**References**

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CH₃O

$\tilde{A} {}^2A_1$ C_{3v} Structure: LF^{22,23}
 $T_0 = 31644.2$ gas EM^{1,2,11,12} AB⁵LF^{6,8,18,19,22-24,27-29,31} $\tilde{A}-\tilde{X}$ 271-421 nm
 31291(3) Ar LF²¹ $\tilde{A}-\tilde{X}$ 270-420 nm
 Evidence for predissociation above 35437 (3793 above the origin of the \tilde{A} state),^{13,29,32} with CH₃ + O the principal products.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CH ₃ stretch	2947.8	gas	LF	31
	2	Umbrella	1289.3	gas	LF	19,27,31
			1308(4)	Ar	LF	21
3	CO stretch	662.4	gas	AB,EM,LF	5,12,19,31	
		657(2)	Ar	LF	21	
e	4	CH ₃ stretch	3077.8T	gas	LF	31
			1403.0	gas	LF	19,31
	5	CH ₂ scissors	1410(3)	Ar	LF	21
			929.5	gas	LF	31
6	HCO deform.					

$\tau = 2.57(13) \mu\text{s}$ gas EM^{2,9}LF^{4,14,15,16,18,20,32}
 $A_0 = 4.981(3)$; $B_0 = 0.743$ LF^{22-24,28}

$\tilde{X} {}^2E$ C_{3v}^a Structure: LMR^{3,7}MW^{10,17}LF²²
 High vibrational levels have been studied using SEP.^{25,28,30}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CH ₃ stretch	2840T ^b	gas	LF	19,27
			1412(4)T ^c	gas	LF	27
		1406(2)T	Ar	LF	21	
e	3	CO stretch	1047	gas	LF,EM	6,9,12,19,27,30
			1044(2)	Ar	LF	21
	4	CH ₃ stretch	2778	gas	SEP	30
			2758(3)	Ar	LF	21
	5	CH ₂ scissors (E) (A ₁)	1465(10)	gas	PE	33,35
			914	gas	SEP	30
6	HCO deform. (E) (A ₁)	1210(10)	gas	PE	33,35	
		651.5	gas	LF	19,26,27	
				SEP,PE	30,33,35	

$A = -61.97(7)$ gas LMR⁷MW¹⁰EM^{11,12}LF^{19,22,23,27}

$A_0 = 5.206(4)$; $B_0 = 0.932$ LMR⁷MW^{10,17}LF^{22,23}

CD₃O

$\tilde{A} {}^2A_1$ C_{3v}
 $T_0 = 31557$ gas LF^{6,19,34}EM¹²FD³⁴ $\tilde{A}-\tilde{X}$ 269-410 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CD ₃ stretch	1889T	gas	LF	34
	2	CD ₃ umbrella	970	gas	LF	19,34
			664	gas	EM,LF	12,19,34
e	4	CD ₃ stretch	2325T	gas	LF	34
			1047	gas	LF	19,34
	6	DCO deform.	697	gas	LF	34

$\tau_0 = 2.8 \mu\text{s}$ gas LF³⁴

$\tilde{X} {}^2E$ C_{3v}^a

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CO stretch	1000T	gas	LF	19
			893T	gas	LF	19
e	5	CD ₂ scissors (E)	1174	gas	LF,EM	6,12,19,33
			920(15)	gas	PE	33,35
6	DCO deform. (E) (A ₁)	496	gas	LF	19	

$A = -56(2)$ gas EM¹²

$B_0 = 0.740$ MW¹⁷

^aSomewhat distorted by Jahn-Teller coupling.

^bRef. 30 indicates that reassignment is necessary.

^cRef. 30 favors assignment at 1359.

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CH₂OH

$\tilde{B}^2A''(3p)$ C_s
 $T_0 = 41065(3)$ gas MPI^{3,4,7,8}AB⁵ $\tilde{B}-\tilde{X}$ 217–244 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	4	CO stretch	1621(7)	gas	MPI	3,4,7
	5	CH ₂ scissors	1465(7)	gas	MPI	3,7
	6	HCOH deform.	1357(7)	gas	MPI	3,7
	7	HCOH deform.	1107(7)	gas	MPI	3,7
<i>a''</i>	9	Torsion	993(8)	gas	MPI	7

Predissociates, with lifetime of 0.5(1) ps gas MPI⁸

$\tilde{A}^2A'(3s)$ C_s
 $T_0 = 35050$ gas AB⁵ $\tilde{A}-\tilde{X}$ 243–285 nm
 Threshold for photodecomposition into H₂CO+H near 280 nm.^{1,2}

\tilde{X}^2A'' C_s^a

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	OH stretch	3650m	Ar	IR	1,2
			3637m	N ₂	IR	1
	4	CH ₂ scissors	1459w	Ar	IR	2
<i>a''</i>	5	OH deform.	1334m	Ar	IR	1,2
	6	CO stretch	1176(7)	gas	MPI	7
			1183vs	Ar	IR	1,2
			1183s	N ₂	IR	1
<i>a''</i>	7	HCOH deform.	1048s	Ar	IR	1,2
			1056m	N ₂	IR	1
	8	Torsion	420m	Ar	IR	1,2
<i>a''</i>			482m	N ₂	IR	1
	9	H ₂ CO OPLA	234(5)	gas	MPI	7

CD₂OD

$\tilde{B}^2A''(3p)$ C_s
 $T_0 = 40913$ gas MPI^{3,7}AB⁶ $\tilde{B}-\tilde{X}$ 216–244 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	4	CO stretch	1565	gas	MPI,AB	3,6,7
	5	CD ₂ scissors	1109	gas	MPI,AB	3,6,7
<i>a''</i>	8	D ₂ CO OPLA	952(23)	gas	MPI	7
	9	Torsion	723(23)	gas	MPI	7

$\tilde{A}^2A'(3s)$ C_s
 $T_0 = 35124$ gas AB⁶
 Threshold for photodecomposition into D₂CO+D near 280 nm.^{1,2}

\tilde{X}^2A'' C_s^a

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	OD stretch	2694wm	Ar	IR	2
			2682m	N ₂	IR	1
	4	CO stretch	1221(4)	gas	MPI	7
			1223m	Ar	IR	2
			1222m	N ₂	IR	1
<i>a''</i>	5	CD ₂ scissors	1020(5)	gas	MPI	7
			1041m	Ar	IR	2
	7		765wm	Ar	IR	2
<i>a''</i>	8	Torsion	329(23)	gas	MPI	7
	9	D ₂ CO OPLA	177(23)	gas	MPI	7

^aRapid interconversion of nonplanar forms.⁷

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CH₃S \tilde{B}^2A_2

$T_0 = 45620$ gas AB^{1,11}PF¹³
Dissociates into CH₃ + S.¹³

 \tilde{A}^2A_1

C_{3v} Structure: LF⁸
 $T_0 = 26396.8$ gas EM²LF^{6,8,10}PF¹³ $\tilde{A}-\tilde{X}$ 315–530 nm
Predissociation threshold = 27324.¹³ In an argon matrix, CH₂SH is formed.⁵

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	CH ₃ umbrella	1098(2)	gas	LF,PF	10,13
	3	CS stretch	401(2)	gas	EM,LF	2,6,8,10
e	6	HCS deform.	635(10)	gas	PF	13
					LF	10

$\tau_0 = 1090(55)$ ns gas LF^{9,10,15}
 $A_0 = 5.343(47)$; $B_0 = 0.346$ LF⁸

 \tilde{X}^2E

C_{3v} Structure: MW⁷LF⁸

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CH ₃ stretch	2960(30)	gas	PE	14
	2	CH ₃ umbrella	1313(5)	gas	PD,LF	4,6,10
	3	CS stretch	727(3)	gas	EM,PE	2–4,6
e	4	CH ₃ stretch	2706T	gas	PD,LF	10,14
					LF	10
					LF	10
e	6	HCS deform.	586T	gas	LF	10
					LF	10

$A = -259.1$ gas LF^{8,10}TPE¹²PF¹³PE¹⁴
 $A_0 = 5.68(4)$; $B_0 = 0.450$ MW⁷LF⁸

CD₃S \tilde{A}^2A_1

C_{3v} $T_0 = 26574$ gas LF⁶PF¹³ $\tilde{A}-\tilde{X}$ 339–378 nm
Predissociation threshold = 28089 PF¹³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	CD ₃ umbrella	837(1)	gas	LF	6
	3	CS stretch	395(1)	gas	LF,PF	6,13

$\tau_0 = 0.45(11)$ μ s gas LF⁶

 \tilde{X}^2E

C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	CD ₃ umbrella	1100(50)	gas	PD	4
	3	CS stretch	667(1)	gas	PD,LF	4,6
e	5	DCS deform.	780(30)H	gas	PD	4

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CH₃Cl⁺ \tilde{C}^2A_1

C_{3v} $T^a = 82400(900)$ gas PE²

 \tilde{B}^2E

C_{3v} $T^a = 33170(900)$ gas PE^{1–4,7}

Position of first maximum is given. A Jahn–Teller splitting of ~5000 is observed.^{4,7}

A weak, broad absorption with onset near 400 nm (25000) and maximum at 335 nm (29800) which appears on argon-resonance photolysis of CH₃Cl isolated in an argon matrix and which is destroyed by exposure of the sample to 290–1000 nm radiation has been attributed⁶ to the $\tilde{B}-\tilde{X}$ and $\tilde{A}-\tilde{X}$ transitions of CH₃Cl⁺.

 \tilde{A}^2A_1

C_{3v} $T_0 = 20260(900)$ gas PE^{1–4,7}

 \tilde{X}

C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH ₃ deform.	1550(50)	gas	PE	3,4
				Ne	IR	8
				Ne	IR	8
		CH ₃ deform.	1239.5wm	Ne	IR	8
				gas	PE	2,4
		CH ₃ umbrella	1073(50)	gas	PE	2,4
				gas	PE,TPE	3,4,9
		CH ₃ rock	870(50)	gas	PE,TPE	3,4,9
				Ne	IR	8
		CH ₃ deform.	649.4wm	Ne	IR	8
				gas	PE,TPE	3,4,9
		CCl stretch	626.9wm			
		CCl stretch	654(50)	gas	PE,TPE	3,4,9

CD₃Cl⁺ \tilde{X}

C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CD ₃ stretch	2013.4m	Ne	IR	8
				Ne	IR	8
		CD ₃ deform.	975.2w	Ne	IR	8
				Ne	IR	8
		CD ₃ deform.	927.9wm	Ne	IR	8
				Ne	IR	8
		CCl stretch	628.9wm	Ne	IR	8

^aFrom vertical ionization potential.

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CH₃I⁺

\tilde{C}^2A_2 C_{3v}
T^a = 81200(900) gas PE¹

\tilde{B}^2E C_{3v}
T₀ = 35180(900) gas PE¹⁻³

A Jahn-Teller splitting of ~5600 is observed.¹⁻³ (Onset of the transition is given.)

\tilde{A}^2A_1 C_{3v}
T₀ = 16884 gas PE¹⁻³PF⁶⁻¹¹

A broad, unstructured absorption with onset near 420 nm (23800) and with maximum at 373 nm (26800) which appears on argon-resonance photolysis of CH₃I isolated in solid argon and which has a photo-decomposition threshold between 500 and 650 nm has been assigned⁵ to the $\tilde{A}-\tilde{X}$ transition of CH₃I⁺.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CH ₃ stretch	2810	gas	PF	7
	2	CH ₃ umbrella	1192	gas	PF	9-11
	3	CI stretch	294.2	gas	PE,PF	3,9-11

A^b = 5.07(5); B^b = 0.185 PF^{6,8}

\tilde{X}^2E C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CH ₃ stretch	2930(5)	gas	TPE	12
	2	CH ₃ umbrella	1254(5) ^c	gas	PF,TPE	1-3,9,12
	3	CI stretch	478(5) ^d	gas	PE,TPE	1,3,12
e	4	CH ₃ stretch	3036(5)	gas	PE,TPE	3,4,12
	5	Deformation	1405H	gas	TPE	12
	6	CH ₃ rock	860(5) ^e	gas	PE,TPE	3,12

A = 5050 gas PE¹⁻⁴PF^{9,11}TPE¹²

CD₃I⁺

\tilde{A}^2A_1 C_{3v}
T₀ = 16982 gas PE,PF^{7,9-11}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CD ₃ stretch	2100	gas	PF	9
	2	CD ₃ umbrella	918	gas	PF	10,11
	3	CI stretch	276.3	gas	PF	10,11

\tilde{X}^2E

C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CD ₃ stretch	2168T	gas	PF,TPE	11,12
	2	CD ₃ umbrella	952(5)	gas	PF,TPE	7,12
	3	CI stretch	454(5)	gas	PF,TPE	11,12
e	4	CD ₃ stretch	2189HT	gas	TPE	12
	5	Deformation	959HT	gas	TPE	12
	6	CD ₃ rock	650(5)	gas	TPE	12

A = 5036 gas PF¹¹TPE¹²

^aFrom vertical ionization potential.

^bFrom study of band at 16978 in $\tilde{A}-\tilde{X}^2E_{1/2}$ transition.

^c1245(5) in $\tilde{X}^2E_{1/2}$.

^d465(5) in $\tilde{X}^2E_{1/2}$.

^e830(5) in $\tilde{X}^2E_{1/2}$.

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CH₃O⁻

Threshold for electron detachment from ground-state CH₃O⁻ = 12680(30) gas PE^{1,3,4}

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	CH ₃ umbrella	1075(100)	gas	PE	1

CD₃O⁻

Threshold for electron detachment from ground-state CD₃O⁻ = 12580(30) gas PE¹⁻⁴

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	CD ₃ umbrella	950(120)	gas	PE	1,2

References

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CH₃S⁻

Threshold for electron detachment from ground-state CH₃S⁻ = 15060(30) gas PD²PE^{1,3,4}

\tilde{X} C _{3v} Structure: MO ²						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	3	CS stretch	410(30)	gas	PE	4

References

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PH₃O

\tilde{X} C _{3v} Structure: MW ²						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	PH ₃ stretch	2359.0 ^a	Ar	IR	1
	2	PO stretch	1240.2	Ar	IR	1
	3	PH ₃ umbrella	1143.5	Ar	IR	1
e	4	PH ₃ stretch	2371.5	Ar	IR	1
	5	PH ₃ deform.	1114.3	Ar	IR	1
	6	HPO deform.	853.0	Ar	IR	1

B₀ = 0.581 MW²

PD₃O

\tilde{X} C _{3v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	PD ₃ stretch	1721.1	Ar	IR	1
	2	PO stretch	1217.3	Ar	IR	1
	3	PD ₃ umbrella	843.6	Ar	IR	1
e	6	DPO deform.	655.9	Ar	IR	1

B₀ = 0.487 MW²

^aIn Fermi resonance with ν₂ + ν₃.

References

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8.9. Five-Atomic Dihydrides**HMnOMnH**

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		MnH stretch	1638.5	Ar	IR	1,2
		MnOMn stretch	870.5	Ar	IR	1,2

DMnOMnD

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		MnD stretch	1175.3	Ar	IR	1,2
		MnOMn stretch	870.3	Ar	IR	1,2

References

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HFeOFeH

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		FeH stretch	1708.2	Ar	IR	1,2
		FeO stretch	911.8	Ar	IR	1,2

DFeOFeD

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		FeD stretch	1228.7	Ar	IR	1,2
		FeO stretch	911.7	Ar	IR	1,2

References

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BH₂CN \bar{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		C≡N stretch	2092.2	Ar	IR	1

Reference

¹D. V. Lanzisera and L. Andrews, *J. Phys. Chem. A* **101**, 824 (1997).

CH₂NB \bar{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1830.5	Ar	IR	1

Reference

¹D. V. Lanzisera and L. Andrews, *J. Phys. Chem. A* **101**, 824 (1997).

cyc-C₃H₂

Photodecomposition threshold in an argon matrix near 360 nm; linear C₃H₂ formed.^{1,6,7}

\bar{X}^a	C _{2v}	Structure: MW ^{4,5}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	3		1277.37	gas	IR	8
			1278.6	Ar	IR	1,6,7
			1277.7			
	4		887.1 ^b	Ar	IR	1,6,7
b ₁	6		1063.6 ^b	Ar	IR	1,6,7
b ₂	8		787.8 ^b	Ar	IR	1,6,7

A₀ = 1.171; B₀ = 1.075; C₀ = 0.559 MW^{2-5,9}

cyc-C₃D₂ \bar{X}^a C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	3		1261.8	Ar	IR	7

^aAssigned in accord with *ab initio* calculations of Ref. 10.

^bDid not diminish in intensity on mercury-arc irradiation at wavelengths longer than 345 nm,⁷ calling into question the assignment to cyc-C₃H₂.

References

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H₂C=C=C:

Photoisomerization to HCCCH occurs at 254 nm.^{1,2}

\bar{C}^1A_1	C _{2v}	
T ₀ = 39063	Ne AB ⁹	$\bar{C}-\bar{X}$ 212–256 nm
38650(160)	Ar AB ⁸	$\bar{C}-\bar{X}$ 213–259 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁		C ₃ s-stretch	1000T	Ne	AB	9
			900T	Ar	AB	8

\bar{B}^1B_1	C _{2v}	
T ₀ = 16426	Ne AB ⁹	$\bar{B}-\bar{X}$ 403–609 nm
T ₀ ≤ 18720(80)	Ar AB ⁸	$\bar{B}-\bar{X}$ 382–535 nm

The spectral region near the origin of this transition is complicated by extensive vibronic interactions.⁹

In an argon matrix, randomization of carbon-13 labelling occurs on irradiation at wavelengths longer than 22500 (444 nm).⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	C ₃ a-stretch	2120	Ne	AB	9
			985	Ne	AB	9
			1000T	Ar	AB	8

\bar{A}^1A_2	C _{2v}	
T ₀ ≤ 13975	Ne AB ⁹	$\bar{A}-\bar{X}$ 619–716 nm
13960	Ar AB ⁸	

\bar{a}^3B_1	C _{2v}	
T ₀ = 10390(70)	gas PE ⁶	

\bar{X}^1A_1	C _{2v}	Structure: MW ⁴				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CH ₂ s-stretch	3059.6	Ar	IR	1,2
			3049.5			
			1963.2	Ar	IR	1,2
	2	C ₃ a-stretch	1952.2			
	3	CH ₂ scissors	1449.3	Ar	IR	1,2
			1446.9			
b ₁	5	H ₂ CC OPLA	1003.0	Ar	IR	1,2
			999.2			
b ₂	8	CH ₂ rock	1025.0	Ar	IR	1

A₀ = 9.633; B₀ = 0.353; C₀ = 0.340 MW^{3,5}

D₂C=C=C:

\tilde{X}^1A_1		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CD ₂ s-stretch	2212.5 2200.5	Ar	IR	1
	2	C ₃ a-stretch	1944.4 1933.4	Ar	IR	1
	3	CD ₂ scissors +C ₃ s-stretch	1208.7	Ar	IR	1
	4	C ₃ s-stretch + CD ₂ scissors	950.8	Ar	IR	1
b_1	5	D ₂ CC OPLA	803.2 800.3	Ar	IR	1
b_2	8	CD ₂ rock	832.6	Ar	IR	1
			829.2			

$A_0=4.842$; $B_0=0.314$; $C_0=0.294$ MW⁴

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CH₂MgF

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		MgF stretch	772.9	Ar	IR	1

CD₂MgF

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		MgF stretch	762.9	Ar	IR	1

Reference

- W. D. Bare, A. Citra, C. Trindle, and L. Andrews, *Inorg. Chem.* **39**, 1204 (2000).

CH₂MgCl

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		MgCl stretch	663.0	Ar	IR	1

Reference

- W. D. Bare, A. Citra, C. Trindle, and L. Andrews, *Inorg. Chem.* **39**, 1204 (2000).

CH₂MgBr

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1			650.9	Ar	IR	1

CD₂MgBr

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1			626.4	Ar	IR	1

Reference

- W. D. Bare, A. Citra, C. Trindle, and L. Andrews, *Inorg. Chem.* **39**, 1204 (2000).

CH₂BO

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		BCO a-stretch	1946.6	Ar	IR	1
		Deformation	663.6	Ar	IR	1

CD₂BO

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		BCO a-stretch	1945.2	Ar	IR	1
		Deformation	554.8	Ar	IR	1

Reference

- D. V. Lanzisera and L. Andrews, *J. Phys. Chem. A* **101**, 1482 (1997).

H₂C=C=C: ⁻

\tilde{D}^2B_1 C_{2v}
 $T_0=27754(20)$ gas PD⁵
 (Feshbach resonance) $\tilde{D}-\tilde{X}$ 338–360 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2		1776(30)	gas	PD	5
	4		1309(30)	gas	PD	5

\tilde{C}^2A_1 C_{2v}
 $T_0=20055(5)$ gas PD⁵
 (Feshbach resonance) $\tilde{C}-\tilde{X}$ 445–500 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	C ₃ a-stretch	1860(25)	gas	PD	5
	4		1245(25)	gas	PD	5
b ₁	6	H ₂ CC OPLA	400(12)	gas	PD	5

\tilde{B}^2A'' C_s
 $T_0=18255(5)$ gas PD⁵
 (Feshbach resonance) $\tilde{B}-\tilde{X}$ 496–548 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	C ₃ a-stretch	1896(12)	gas	PD	5
	4		1089(12)	gas	PD	5
	5		677(12)	gas	PD	5
	6		527(12)	gas	PD	5
	8		883(12)	gas	PD	5
b ₁	9		375(7)	gas	PD	5

Threshold for electron detachment from ground-state H₂C=C=C: ⁻
 = 14470(65) gas PE^{1,2}

\tilde{A}^2A_1 C_{2v}
 (Dipole-Bound State)
 $T_0=14284.420(5)$ gas PD³ $\tilde{A}-\tilde{X}$ 615–700 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	C ₃ a-stretch	1956	gas	PD	4,5
	4	C ₃ s-stretch	1111	gas	PD	4,5
b ₁	6	H ₂ CC OPLA	221.45	gas	PD	4,5
b ₂	9		276.69	gas	PD	4

$A_0=9.651$; $1/2(B_0+C_0)=0.346$ PD³

\tilde{X}^2B_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b ₁	6	H ₂ CC OPLA	385.43	gas	PD	4
b ₂	9		309.24	gas	PD	4

$A_0=9.731$; $B_0=0.344$; $C_0=0.332$ PD³

D₂C=C=C: ⁻

\tilde{C}^2A_1 C_{2v}
 $T_0=20090(20)$ gas PD⁵
 (Feshbach resonance) $\tilde{C}-\tilde{X}$ 456–500 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	C ₃ a-stretch	1830(30)	gas	PD	5
	4		990(30)	gas	PD	5
b ₁	6	D ₂ CC OPLA	340(30)	gas	PD	5

\tilde{B}^2A'' C_s
 $T_0=18273(10)$ gas PD⁵
 (Feshbach resonance) $\tilde{B}-\tilde{X}$ 504–547 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	4		990(15)	gas	PD	5
	5		609(15)	gas	PD	5
	6		505(15)	gas	PD	5
	8		847(15)	gas	PD	5
	9		323(15)	gas	PD	5

\tilde{A}^2A_1 C_{2v}
 $T_0=14246(2)$ gas PD⁵
 (Dipole-bound state) $\tilde{A}-\tilde{X}$ 618–702 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	C ₃ a-stretch	1920(10)	gas	PD	5
	4		936(3)	gas	PD	5
b ₁	6	D ₂ CC OPLA	188(3)	gas	PD	5

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H₂CCN

\tilde{X}^2B_1 C_{2v} Structure: MW⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b ₁	5	H ₂ CC OPLA	663.79	gas	PE,DL	1,3

$A_0=9.506(4)$; $B_0=0.348$; $C_0=0.329$ MW^{2,4}

D₂CCN

\tilde{X}^2B_1		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>b</i> ₁	5	D ₂ CC OPLA	538 ^a	gas	PE	1

$A_0=4.768$; $B_0=0.303$; $C_0=0.284$ MW⁴

^aFrom computer fit.

References

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²S. Saito, S. Yamamoto, W. M. Irvine, L. M. Ziurys, H. Suzuki, M. Ohishi, and N. Kaifu, *Astrophys. J.* **334**, L113 (1988).

³Y. Sumiyoshi, K. Tanaka, and T. Tanaka, *J. Chem. Phys.* **104**, 1839 (1996).

⁴S. Saito and S. Yamamoto, *J. Chem. Phys.* **107**, 1732 (1997).

H₂CCP

\tilde{X}^2B_1		C_{2v}		Structure: MW,MO ¹		
$A_0=9.634(3)$; $B_0=0.182$; $C_0=0.179$ MW ¹						

Reference

¹I. K. Ahmad, H. Ozeki, S. Saito, and P. Botschwina, *J. Chem. Phys.* **109**, 4252 (1998).

Y(OH)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		YO ₂ a-stretch	597.9	Ar	IR	1

Y(OD)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		YO ₂ a-stretch	578.0	Ar	IR	1

Reference

¹L. Zhang, L. Shao, and M. Zhou, *Chem. Phys.* **272**, 27 (2001).

La(OH)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		LaO ₂ s-stretch	519.3	Ar	IR	1
		LaO ₂ a-stretch	491.2	Ar	IR	1

La(OD)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		LaO ₂ s-stretch	517.3	Ar	IR	1
		LaO ₂ a-stretch	488.2	Ar	IR	1

Reference

¹L. Zhang, L. Shao, and M. Zhou, *Chem. Phys.* **272**, 27 (2001).

Mn(OH)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		MnO ₂ a-stretch	709.0	Ar	IR	1

Mn(OD)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		MnO ₂ a-stretch	693.6	Ar	IR	1

Reference

¹M. Zhou, L. Zhang, L. Shao, W. Wang, K. Fan, and Q. Qin, *J. Phys. Chem. A* **105**, 5801 (2001).

Fe(OH)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OFeO stretch	735.5	Ar	IR	1,2

Fe(OD)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OFeO stretch	721.7	Ar	IR	1,2

References

¹J. W. Kauffman, R. H. Hauge, and J. L. Margrave, *J. Phys. Chem.* **89**, 3541 (1985).

²L. Zhang, M. Zhou, L. Shao, W. Wang, K. Fan, and Q. Qin, *J. Phys. Chem. A* **105**, 6998 (2001).

H₂CBF

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CH ₂ s-stretch	3050.2	Ar	IR	1
	2		1762.2	Ar	IR	1
			917.4	Ar	IR	1
b ₁		755.0	Ar	IR	1	
		589.5	Ar	IR	1	

D₂CBF

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁			1741.5	Ar	IR	1
b ₁			671.8	Ar	IR	1

Reference

¹D. V. Lanzisera and L. Andrews, J. Phys. Chem. A **104**, 9295 (2000).

H₂CBCl

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		BC stretch	1605.7	Ar	IR	1
		Deformation	686.9	Ar	IR	1

D₂CBCl

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CD ₂ stretch	2232.4	Ar	IR	1
		BC stretch	1570.3	Ar	IR	1
		Deformation	611.1	Ar	IR	1

Reference

¹D. V. Lanzisera and L. Andrews, J. Phys. Chem. A **104**, 9295 (2000).

H₂CBBr

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH ₂ stretch	3045.1	Ar	IR	1
		BC stretch	1576.0	Ar	IR	1

D₂CBBr

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CD ₂ stretch	2224.7	Ar	IR	1
		BC stretch	1541.0	Ar	IR	1

Reference

¹D. V. Lanzisera and L. Andrews, J. Phys. Chem. A **104**, 9295 (2000).

HN=C=NH

\tilde{X}		C ₂		Structure: MW ⁶		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a	2	NCN s-stretch	1285(20) ^a	gas	IR	2
			1275 ^a	Ar	IR	1
	5	NCN deform.	537m	Ar	IR	1
b	7	NCN a-stretch	2104.7	gas	IR	2
			2097s	Ar	IR	1
	8	NH deform.	890(10)	gas	IR	2
			886vs	Ar	IR	1

A₀ = 12.650; B₀ = 0.346; C₀ = 0.346 IR, MW³⁻⁵

DN=C=ND

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a	1	ND stretch	2545s ^b	Ar	IR	1
	5		471m	Ar	IR	1
b	6	NCN a-stretch	2545s ^b	Ar	IR	1
	7		2107vs	Ar	IR	1
	8		752s	Ar	IR	1

A₀ = 7.055; B₀ = 0.303; C₀ = 0.302 MW⁶

^aCalculated from (ν₂ + ν₈) combination band.

^bBoth ND-stretching frequencies presumed equal.

References

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- ²M. Birk and M. Winnewisser, Chem. Phys. Lett. **123**, 386 (1986).
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- ⁶W. Jabs, M. Winnewisser, S. P. Belov, F. Lewen, F. Maiwald, and G. Winnewisser, Mol. Phys. **97**, 213 (1999).

cyc-H₂COC:

In an argon or a nitrogen matrix, a weak, unstructured absorption maximum¹ at 30860 (324 nm) is associated with the isomerization of this species (ox-iranylidene) to ketene.

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	2	Ring deform.	1470.8m	N ₂	IR	1
	3	CH ₂ scissors	1375.1ms	N ₂	IR	1
	4	CH ₂ wag	1078.5w	N ₂	IR	1
	5	Ring deform.	830.1vs	N ₂	IR	1
	6	Ring deform.	778.1wm	N ₂	IR	1

cyc-D₂COC:

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	2	Ring deform.	1419.4s ^a 1400.6s	N ₂	IR	1
	3	CD ₂ scissors	1050.2wm	N ₂	IR	1
	4	CD ₂ wag	960.3ms	N ₂	IR	1
	5	Ring deform.	780.1vs	N ₂	IR	1
	6	CD ₂ wag	706.3vs	N ₂	IR	1

^aIn Fermi resonance with 2ν₆.**Reference**¹G. Maier, H. P. Reisenauer, and M. Cibulka, *Angew. Chem.* **111**, 110 (1999); *Angew. Chem. Int. Ed.* **38**, 105 (1999).**CH₂=CF**Isomerized to *c*-CHF=CH by 355 nm radiation.¹

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	2	CH ₂ stretch	2778wT	Ar	IR	1
	3	C=C stretch	1654vs	Ar	IR	1
	4	CH ₂ scissors	1355w	Ar	IR	1
	5	CF stretch	1133s	Ar	IR	1
	6	CH ₂ rock	930wm	Ar	IR	1
	<i>a''</i>	8	OPLA	798m	Ar	IR

CD₂=CF

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	2	CD ₂ stretch	2221w	Ar	IR	1
	3	C=C stretch	1625vs	Ar	IR	1
	4	CF stretch	1138s	Ar	IR	1
	5	CD ₂ scissors	973wm	Ar	IR	1
	6	CD ₂ rock	797wm	Ar	IR	1
	<i>a''</i>	8	OPLA	630wm	Ar	IR

Reference¹I. U. Goldschleger, A. V. Akimov, E. Ya. Misochko, and C. A. Wight, *J. Mol. Spectrosc.* **205**, 269 (2001).***c*-CHF=CH**Partially converted to CH₂=CF by 532 nm radiation.²

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CH stretch	3193wm	Ar	IR	2
	2	CH stretch	3026w	Ar	IR	2
	3	C=C stretch	1631s	Ar	IR	2
	4	HCF scissors	1623vs	Ar	IR	1
			1263wm	Ar	IR	2
			1211wm	Ar	IR	1
	5	CF stretch	1068vs	Ar	IR	1,2
	6	C=CH deform.	698m	Ar	IR	2
	<i>a''</i>	7	C=CF deform.	678s	Ar	IR
462m				Ar	IR	1
785s				Ar	IR	1,2
8	HFC=C OPLA	Torsion	643m	Ar	IR	2
			631vs	Ar	IR	1

***c*-DCF=CD**

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CD stretch	2406wm	Ar	IR	2
	2	CD stretch	2248w	Ar	IR	2
	3	C=C stretch	1572vs	Ar	IR	2
			1564s	Ar	IR	1
			1068s	Ar	IR	1,2
	5	Mixed	935wm	Ar	IR	2
6	C=CD deform.	633wm	Ar	IR	2	
<i>a''</i>	8	DFC=C OPLA	624m	Ar	IR	1
			660w	Ar	IR	2
			485m	Ar	IR	1

References¹M. E. Jacox, *Chem. Phys.* **53**, 307 (1980).²I. U. Goldschleger, A. V. Akimov, E. Ya. Misochko, and C. A. Wight, *J. Mol. Spectrosc.* **205**, 269 (2001).

H₂NCO

In a xenon matrix,¹ the photodecomposition threshold lies between 365 and 405 nm.

$\tilde{X} \quad C_s$						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	NH ₂ a-stretch	3518.5	Xe	IR	1
	3	CO stretch	1811.8	Xe	IR	1
			1806.5			
			1796.6			
			1794.2			
4		1555.8	Xe	IR	1	
5		1214.4	Xe	IR	1	

D₂NCO

$\tilde{X} \quad C_s$						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	ND ₂ a-stretch	2634.4	Xe	IR	1
	3	CO stretch ^a	1800.2	Xe	IR	1
			1797.8			
			1796.3			
		1793.9				

^aAbsorption of HDNCO could not be distinguished from that of D₂NCO.

Reference

¹M. Pettersson, L. Khriachtchev, S. Jolkkonen, and M. Räsänen, *J. Phys. Chem. A* **103**, 9154 (1999).

HCOOH⁺

$\tilde{E}^2A'' \quad C_s$
 $T_0 = 49700(320)$ gas PE^{1,3}

$\tilde{D}^2A' \quad C_s$
 $T_0 = 45500(320)$ gas PE^{1,3}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>		CO stretch	1300(40)	gas	PE	1,2

$\tilde{C}^2A'' \quad C_s$
 $T_0 \cong 32800$ gas PE^{1,3}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>			940	gas	PE	1

$\tilde{B}^2A' \quad C_s$
 $T_0 \cong 23200$ gas PE^{1,3}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>			940	gas	PE	1

$\tilde{A}^2A'' \quad C_s$
 $T_0 = 8495(25)$ gas PE^{1,3,4}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>			1398(20)	gas	PE	4
			1029(20)	gas	PE	1,2,4
			574(20)	gas	PE	4

$\tilde{X}^2A' \quad C_s$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	3	CO stretch	1495(20)	gas	PE	1,2,4
	5		1196(20)	gas	PE	4
	7		510(20)	gas	PE	4

DCOOD⁺

$\tilde{D}^2A' \quad C_s$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>		CO stretch	1210(40)	gas	PE	2
		COD bend ?	880(40)	gas	PE	2

$\tilde{A}^2A'' \quad C_s$
 $T_0 = 8430(25)$ gas PE⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>			1085(20)	gas	PE	2,4
			521(20)	gas	PE	4

$\tilde{X}^2A' \quad C_s$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>		CO stretch	1472(20)	gas	PE	2,4
		COD bend	965(20)	gas	PE	2,4
			447(20)	gas	PE	4

^aFrom vertical ionization potential.

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HInCl(OH) \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OH stretch	3649.3	Ar	IR	1
		InH stretch	1857.2	Ar	IR	1
			810.0	Ar	IR	1
			606.4	Ar	IR	1
			451.6	Ar	IR	1
			428.3	Ar	IR	1
			348.5	Ar	IR	1

Reference

¹H.-J. Himmel, A. J. Downs, and T. M. Greene, *J. Am. Chem. Soc.* **122**, 922 (2000).

H₂CSS

In an argon matrix, there is a prominent absorption maximum at 28100 (356 nm).¹

 \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CH ₂ stretch	3126.3vw	Ar	IR	1
	2	CH ₂ stretch	3010.4w	Ar	IR	1
	3	CH ₂ scissors	1366.4m	Ar	IR	1
	4	CS stretch	970.9m	Ar	IR	1
	5	CH ₂ rock	910.2m	Ar	IR	1
	6	SS stretch	622.8s	Ar	IR	1
<i>a''</i>	8	CH ₂ wag	756.8s	Ar	IR	1

Reference

¹G. Mloston, J. Romański, H. P. Reisenauer, and G. Maier, *Angew. Chem. Int. Ed.* **40**, 393 (2001).

cyc-H₂CS₂ \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CH ₂ stretch	2986.7m	Ar	IR	1
	2	CH ₂ scissors	1412.0w	Ar	IR	1
<i>b</i> ₁	7	CH ₂ rock	928.3m	Ar	IR	1
<i>b</i> ₂	8	CH ₂ wag	1053.0m	Ar	IR	1
	9	Ring stretch	581.0s	Ar	IR	1

Reference

¹G. Mloston, J. Romański, H. P. Reisenauer, and G. Maier, *Angew. Chem. Int. Ed.* **40**, 393 (2001).

H₂CCI-Cl

A broad absorption with maximum near 360 nm (28000) which appears¹⁻³ on vacuum ultraviolet photolysis of CH₂Cl₂ isolated in solid argon and a much weaker absorption near 620 nm (16000) have been attributed³ to this product. Exposure of the sample to radiation in the spectral region of either of these absorptions leads to destruction of H₂Cl-Cl and to re-formation of CH₂Cl₂.

 \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH ₂ stretch	3143.3w	Ar	IR	3,4
		CH ₂ stretch	3023.5wm	Ar	IR	3,4
			3019	Kr	IR	4
		CH ₂ scissors	1404.9wm	Ar	IR	3,4
			1405	Kr	IR	4
			1402	Xe	IR	4
			1193sT	Ar	IR	1,2
		CCl stretch	958.5wm	Ar	IR	3,4
			965	Kr	IR	4
			763.2vs	Ar	IR	1-4
			775	Kr	IR	4
			776	Xe	IR	4

D₂CCI-Cl \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CD ₂ stretch	2375.5vw	Ar	IR	3
		CD ₂ stretch	2220.7wm	Ar	IR	3,4
			2220	Kr	IR	4
			2212	Xe	IR	4
		CD ₂ scissors	1083.4ms	Ar	IR	2-4
			1082	Xe	IR	4
		CCl stretch	886.1m	Ar	IR	3,4
		CD ₂ wag	603.1vs	Ar	IR	1-4
			613	Xe	IR	4

References

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²B. J. Kelsall and L. Andrews, *J. Mol. Spectrosc.* **97**, 362 (1983).

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⁴T. D. Fridgen, X. K. Zhang, J. M. Parnis, and R. E. March, *J. Phys. Chem. A* **104**, 3487 (2000).

H₂CCI-Br

A broad absorption with maximum near 382 nm (26200) which appears¹ on 230- or 240-nm irradiation of CH₂ClBr isolated in solid argon has been attributed¹ to this product. Exposure of the sample to 360 nm radiation or to radiation of wavelength longer than 570 nm leads to destruction of H₂CCI-Br and to re-formation of CH₂ClBr.

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
		CH ₂ stretch	3038.6w	Ar	IR	1
		CH ₂ scissors	1388.3wm	Ar	IR	1,2
		CCl stretch	931.2wm	Ar	IR	1,2
		CH ₂ wag	726.6vs	Ar	IR	1,2

D₂CCI-Br

\tilde{X} C_s						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
		CD ₂ stretch	2232.3wm	Ar	IR	1
		CD ₂ scissors	1069.6s	Ar	IR	1
		CCl stretch	868.8m	Ar	IR	1
		CD ₂ wag	573.3vs	Ar	IR	1

References

- ¹G. Maier, H. P. Reisenauer, J. Hu, L. J. Schaad, and B. A. Hess, Jr., *J. Am. Chem. Soc.* **112**, 5117 (1990).
²T. D. Fridgen, X. K. Zhang, J. M. Parnis, and R. E. March, *J. Phys. Chem. A* **104**, 3487 (2000).

H₂O₃

\tilde{X} C_2						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
<i>a</i>	1	OH s-stretch	3529.6	Ar	IR	1
	2	HOO s-bend	1347.4	Ar	IR	1
	3	O ₃ s-stretch	821.0	Ar	IR	1
	4	O ₃ deform.	509.1	Ar	IR	1
	5	Sym. torsion	346.4	Ar	IR	1
<i>b</i>	6	OH a-stretch	3529.6	Ar	IR	1
	7	HOO a-bend	1359.1	Ar	IR	1
	8	O ₃ a-stretch	776.3	Ar	IR	1
	9	Asym. torsion	387.0	Ar	IR	1

D₂O₃

\tilde{X} C_2						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
<i>a</i>	1	OH s-stretch	2610.4	Ar	IR	1
	5	Sym. torsion	273.5	Ar	IR	1
<i>b</i>	6	OD a-stretch	2610.4	Ar	IR	1
	7	DOO a-bend	1007.3	Ar	IR	1
	8	O ₃ a-stretch	762.6	Ar	IR	1
	9	Asym. torsion	301.6	Ar	IR	1

Reference

- ¹A. Engdahl and B. Nelander, *Science* **295**, 482 (2002).

***t*-HSSSH**

\tilde{X} C_2 Structure: MW ²						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
		SH a-stretch	2542	gas	IR	1
		SSH a-bend	860	gas	IR	1
		S ₃ a-stretch	480	gas	IR	1

$A_0=0.470$; $B_0=0.092$; $C_0=0.079$ MW^{1,2}

***t*-DSSSD**

\tilde{X} C_2						
$A_0=0.426$; $B_0=0.087$; $C_0=0.076$ MW ²						

References

- ¹M. Liedtke, A. H. Saleck, K. M. T. Yamada, G. Winnewisser, D. Cremer, E. Kraka, A. Dolgner, J. Hahn, and S. Dobos, *J. Phys. Chem.* **97**, 11204 (1993).
²M. Liedtke, K. M. T. Yamada, G. Winnewisser, and J. Hahn, *J. Mol. Struct.* **413/414**, 265 (1997).

***c*-HSSSH**

\tilde{X} C_s Structure: MW ⁵						
Estimated ⁴ to be 87 cm^{-1} (0.25 kcal/mol) higher in energy than <i>t</i> -HSSSH.						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
		SH s-stretch	2548	gas	IR	4
		SSH a-bend	865	gas	IR	4
		S ₃ a-stretch	480	gas	IR	4
		S ₃ bend	240(30)	gas	MW	3

$A_0=0.470$; $B_0=0.092$; $C_0=0.079$ MW¹⁻⁵

***c*-DSSSD**

\tilde{X} C_s						
$A_0=0.425$; $B_0=0.087$; $C_0=0.077$ MW ⁵						

References

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⁴M. Liedtke, A. H. Saleck, K. M. T. Yamada, G. Winnewisser, D. Cremer, E. Kraka, A. Dolgner, J. Hahn, and S. Dobos, *J. Phys. Chem.* **97**, 11204 (1993).
⁵M. Liedtke, K. M. T. Yamada, G. Winnewisser, and J. Hahn, *J. Mol. Struct.* **413/414**, 265 (1997).

8.10. Five-Atomic Monohydrides

PdCCNH

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1800(200)	gas	PE	1

Reference

¹V. D. Moravec and C. C. Jarrold, *J. Chem. Phys.* **112**, 792 (2000).

PdCCNH⁻

Threshold for electron detachment from ground-state PdCCNH⁻ = 17510(240) gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1600(200)	gas	PE	1

Reference

¹V. D. Moravec and C. C. Jarrold, *J. Chem. Phys.* **112**, 792 (2000).

C₄H

Absorptions between 250 and 300 nm which were attributed by Ref. 1 to C₄H have since been reassigned (D. Forney and J. P. Maier, private communication) to C₆H₂.

$\tilde{B}^2\Pi$ C_{∞v}
T₀ = 24033.432(6) gas LF¹⁰ $\tilde{B}-\tilde{X}$ 400–417 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	5	HCC bend	344.2 ^a	gas	LF	10
	6	CCC bend	189.3 ^b	gas	LF	10

A_{SO} = -22.1 gas LF¹⁰
B_{eff} = 0.150 LF¹⁰

Groups of absorptions observed in an argon matrix¹ between 13408 and 13906, 17629 and 17939, 21972 and 25667, and 26867 and 32104 are tentatively attributed to C₄H.

$\tilde{A}^2\Pi$ C_{∞v}
T₀ ≤ 468 gas PE¹¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	2	C≡C stretch	2081T	gas	PE	11
Π	7	CCC bend	201T	gas	PE	11

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	CH stretch	3307.4w	Ar	IR	7
	2	C≡C stretch	2083.9wm	Ar	IR	7
	3	C≡C stretch	2063.7	Ne	IR	8,12
			2060.6ms	Ar	IR	1,7,12
	4	C–C stretch	960(50)	gas	PE	11
Π	7	CCC bend	131T	gas	MW	6
			226T		PE	11

B₀ = 0.165 MW²⁻⁵

C₄D

$\tilde{B}^2\Pi$ C_{∞v}
T₀ = 24099.191(2) gas LF¹⁰ $\tilde{B}-\tilde{X}$ 400–415 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	5	DCC bend	295.3(5) ^a	gas	LF	10
	6	CCC bend	183.4 ^b	gas	LF	10

A_{SO} = -19.40(41) gas LF¹⁰
B_{eff} = 0.140 LF¹⁰

An absorption at 17685 and a group of bands between 26925 and 30883 in the argon-matrix observations¹ are tentatively attributed to C₄D.

$\tilde{A}^2\Pi$ C_{∞v}
T₀ ≤ 468 gas PE¹¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	2	C≡C stretch	2073L	gas	PE	11
Π	7	CCC bend	185T	gas	PE	11

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	CD stretch	2579.3w	Ar	IR	7
	2	C≡C stretch	2056.5wm	Ar	IR	7
	3	C≡C stretch	2052.9	Ne	IR	8,12
			2049.6ms	Ar	IR	1,7,12
	4	C–C stretch	895(50)	gas	PE	11
Π	7	CCC bend	210T	gas	PE	11

^aω. ε = -0.6566 for C₄H and -0.6921(8) for C₄D.¹⁰

^bω. ε = -0.01578 for C₄H and -0.01877 for C₄D.¹⁰

References

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¹¹T. R. Taylor, C. Xu, and D. M. Neumark, *J. Chem. Phys.* **108**, 10018 (1998).

¹²L. Andrews, G. P. Kushto, M. Zhou, S. P. Willson, and P. F. Souter, *J. Chem. Phys.* **110**, 4457 (1999).

Si₄H

\tilde{X}^2A'		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'			310(20)	gas	PE	1

Reference

¹C. Xu, T. R. Taylor, G. R. Burton, and D. M. Neumark, *J. Chem. Phys.* **108**, 7645 (1998).

HC≡C–C≡N⁺

\tilde{C}		C _s				
T ₀ = 48570(160) gas PE ¹						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1320(40)	gas	PE	1

$\tilde{B}^2\Pi$		C _{∞v}				
T ₀ = 19600(160) gas PE ¹ 19374(43) Ne AB ² $\tilde{B}-\tilde{X}$ 474–516 nm						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺			1940(40)	gas	PE	1
	4	C–C stretch	810(40)	gas	PE	1
			820(60)	Ne	AB	2

$\tilde{A}^2\Sigma^+$		C _{∞v}				
T ₀ = 15650(160) gas PE ¹						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	4	C–C stretch	860(40)	gas	PE	1

$\tilde{X}^2\Pi$		C _{∞v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	CH stretch	3196.5m	Ne	IR	3
	2		2180(40)	gas	PE	1
			2175.8w	Ne	IR	3
	3		1852.8s	Ne	IR	3

DC≡C–C≡N⁺

$\tilde{X}^2\Pi$		C _{∞v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	CD stretch	2499.9wm	Ne	IR	3
	2		2114.8wm	Ne	IR	3
	3		1795.6s	Ne	IR	3

References

¹C. Baker and D. W. Turner, *Proc. R. Soc. London, Ser. A* **308**, 19 (1968).

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³A. M. Smith-Gicklhorn, M. Lorenz, R. Kolos, and V. E. Bondybey, *J. Chem. Phys.* **115**, 7534 (2001).

C₄H⁻

Threshold for electron detachment from ground-state C₄H⁻ = 28710(120) gas PE¹

$\tilde{X}^1\Sigma^+$		C _{∞v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	7	Deformation	210(50)	gas	PE	1

C₄D⁻

Threshold for electron detachment from ground-state C₄D⁻ = 28660(120) gas PE¹

$\tilde{X}^1\Sigma^+$		C _{∞v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Π	7	Deformation	169(50)	gas	PE	1

Reference

¹T. R. Taylor, C. Xu, and D. M. Neumark, *J. Chem. Phys.* **108**, 10018 (1998).

Si₄H⁻

Threshold for electron detachment from ground-state Si₄H⁻ = 21620(80) gas PE¹

Reference

¹C. Xu, T. R. Taylor, G. R. Burton, and D. M. Neumark, *J. Chem. Phys.* **108**, 7645 (1998).

HC≡C-N≡C

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CH stretch	3338.96s	gas	IR	1,3
	2	N≡C stretch	2218.64s	gas	IR	1,3
	3	C≡C stretch	2036.66s	gas	IR	1,3
	4	C-C stretch	938.09w ^a	gas	IR	3
II	5	HCC bend	620.63vs	gas	IR	1,3
	6		430T	gas	IR	3
	7		207T	gas	IR	3

$B_0 = 0.166$ MW^{1,2,5}

DC≡C-N≡C

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CD stretch	2617.39s	gas	IR	3
	2	N≡C stretch	2157.58s	gas	IR	3
	3	C≡C stretch	1965.48s	gas	IR	3
	4	C-C stretch	923.58w ^a	gas	IR	3
II	5	DCC bend	480.17	gas	IR	3
	6		425T	gas	IR	3

$B_0 = 0.153$ MW^{1,4}IR³

^aDeperturbed value.

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HNCCC

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	NH stretch	3562m	Ar	IR	2
	2	NC stretch	2205vs	Ar	IR	2
	3	Skel. stretch	1905w	Ar	IR	2

Quasilinear. $B_0 = 0.156$ MW¹

DNCCC

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	ND stretch	2611m	Ar	IR	2
	2	NC stretch	2162vs	Ar	IR	2
	3	Skel. stretch	1890vwT	Ar	IR	2

$B_0 = 0.147$ MW¹

References

- ¹Y. Hirahara, Y. Ohshima, and Y. Endo, *Astrophys. J.* **403**, L83 (1993).
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HNCCN⁺

\tilde{X}		$D_{\infty h}$	Structure: MW ⁴			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	NH stretch	3448.27	gas	LD	1

$B_0 = 0.148$ LD¹MW^{2,5}

DNCCN⁺

\tilde{X}		$D_{\infty h}$
$B_0 = 0.139$		MW ³

References

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HCICCO

\tilde{X}		C_s	Structure: MW ¹			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	CO stretch	2157.19	gas	DL	3
			2141.4vs	Xe	IR	2
	3	HCCI deform.	1293w	Xe	IR	2
	4	CC stretch	1107w	Xe	IR	2
	5	CCl stretch	842.2vw	Xe	IR	2
a''	8	HCICC OPLA	747	Xe	IR	2

$A_0 = 1.206$; $B_0 = 0.101$; $C_0 = 0.093$ MW¹

DCICCO

\tilde{X} C_s
 $A_0=0.974; B_0=0.101; C_0=0.091$ MW¹

References

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BF₂OH

\tilde{X}		C_s		Structure: MW ¹		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	OH stretch	3712.5s	Ne	IR	2
	2	BF ₂ , BO stretch	1464.3vs	Ne	IR	2
	3	BF ₂ a-stretch	1414.9vs	Ne	IR	2
	4	BOH deform.	959.1s	Ne	IR	2
	5	BF ₂ , BO stretch	879.3wm	Ne	IR	2
	7	Deformation	447.5m	Ne	IR	2
a''	8	OPLA	680.9ms	Ne	IR	2
	9	Torsion	520.8ms	Ne	IR	2

$A_0=0.344; B_0=0.337; C_0=0.170$ MW¹

BF₂OD

\tilde{X}		C_s		Structure: MW ¹		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	OD stretch	2737.2m	Ne	IR	2
	2	BF ₂ , BO stretch	1456.8s	Ne	IR	2
	3	BF ₂ a-stretch	1400.9	Ne	IR	2
	4	BF ₂ , BO stretch	879.3wm	Ne	IR	2
	5	BOD deform.	751.1wm	Ne	IR	2
	6	Deformation	481.5wm	Ne	IR	2
	7	Deformation	413.4	Ne	IR	2
a''	8	OPLA	680.4	Ne	IR	2

$A_0=0.343; B_0=0.314; C_0=0.164$ MW¹

References

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HONO₂

In the gas phase, continuous absorption begins near 330 nm, with a maximum near 38500 (260 nm), then increasing absorption out to the 190 nm observation limit.^{4,5,31} Between 330 and 220 nm, photodissociation to OH + NO₂ predominates.^{6,16,33,39} However, at 193 nm photodissociation to HONO+O is the major process.³³ Vacuum ultraviolet spectral observations^{7,12} show an absorption maximum near 54900 (182 nm), a more prominent maximum near 73500 (136 nm), and overlapping absorption bands at higher energies. The onset of the photoproduction of OH ($A^2\Sigma$) + NO₂ has been observed¹² at 68000 (147.5 nm).

\tilde{X}		C_s		Structure: MW ^{1,3,43}		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	OH stretch	3550.0m	gas	IR	2
			3522.3	Ar	IR	28
			3519.3			
			3491.8	N ₂	IR	8,36
	2	NO ₂ a-stretch	1709.57vs	gas	IR,DL	2,9,11,20,21,23,35
			1699.4	Ar	IR	28
			1696.2			
			1698.3	N ₂	IR	8,36
	3	Mixed	1325.74s	gas	IR,DL	2,19,26,27,40,41
			1321.4	Ar	IR	28
			1318.7			
			1346.1	N ₂	IR	8,36
	4	Mixed	1303.52vs	gas	IR,DL	2,26,27,40,41
			1304.4	Ar	IR	28
			1311.9	N ₂	IR	8,36
	5	ON stretch	879.11s	gas	IR,DL	2,18,32,34,37,40,41
			896.9	Ar	IR	2
			889.5			
			903.1	N ₂	IR	8,36
	6	NO ₂ scissors	646.83w	gas	IR	2,25,46
			656.6	Ar	IR	36
			664.1	N ₂	IR	8,36
	7	NO ₂ rock	580.30w	gas	IR	2,25,44
			588.0	Ar	IR	36
			597.5	N ₂	IR	8,36
a''	8	ONO ₂ OPLA	763.15s	gas	IR	2,25,44,45
			763.6	Ar	IR	28
			767.7	N ₂	IR	8,36
	9	Torsion	458.23m	gas	IR	2,15,24,47
			479	N ₂	IR	8

$A_0=0.434; B_0=0.404; C_0=0.209$ MW^{1,10,13,14,17,22}IR^{11,24,25}

DONO₂

$\bar{\chi}$	C _s					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	OD stretch	2621.5m	gas	IR	2
			2601.5	Ar	IR	36
			2599.1			
	2	NO ₂ a-stretch	2576.5	N ₂	IR	8,36
			1688.36vs	gas	IR	2,42
			1678.6	Ar	IR	36
	3	NO ₂ s-stretch	1674.9			
			1669.2	N ₂	IR	8,36
			1308.4vs	gas	IR	2
	4	DON bend	1310.4	Ar	IR	36
			1311.1	N ₂	IR	8,36
			1013.6m	gas	IR	2
5	ON stretch	1012.2	Ar	IR	36	
		1032.4	N ₂	IR	8,36	
		888.0s	gas	IR	2	
6	NO ₂ scissors	894.2	Ar	IR	36	
		884.5				
		906.0	N ₂	IR	8,36	
7	NO ₂ rock	642.14s	gas	IR	2,38	
		660	N ₂	IR	8	
		541.58w	gas	IR	2,38	
a''	8	ONO ₂ OPLA	559.9	N ₂	IR	8,36
			762.87s	gas	IR	2,29
			763.7	Ar	IR	36
9	Torsion	767.7	N ₂	IR	8,36	
		343.85m	gas	IR	2,15,30	
		361	N ₂	IR	8	

A₀=0.433; B₀=0.377; C₀=0.201 MW^{1,43}IR³⁸

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***t,perp*-HOONO**

In an argon matrix, an absorption maximum near 45500 (220 nm) has been assigned³ to *t,perp*-HOONO.

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
1		OH stretch	3545.5w	Ar	IR	1-5
			3563.3 ^a	Ar	IR	1-4
			3541.7	N ₂	IR	1,2,5
2		N=O stretch	1703.6s	Ar	IR	1-5
			1708.3 ^a	Ar	IR	1-5
			1701.4	N ₂	IR	1,2,5
3		HOO bend	1364.4m	Ar	IR	1-5
			1372.7 ^a	Ar	IR	1-5
			1394.9	N ₂	IR	1,2,5
4		OO stretch	952.0m	Ar	IR	1-5
			957.4 ^a	Ar	IR	1-5
			960.5	N ₂	IR	1,2,5
5		ONO bend	772.8m	Ar	IR	1-5
			782.9 ^a	Ar	IR	1-5
			793.6	N ₂	IR	1,2

***t,perp*-DOONO** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
1		OD stretch	2615.4	Ar	IR	1,2
			2633.5 ^a	Ar	IR	1,2
			2622.3	N ₂	IR	2
2		N=O stretch	1703.7	Ar	IR	1,2
			1708.3 ^a	Ar	IR	1,2
			1701.8	N ₂	IR	2
3		DOO bend	1089.7	Ar	IR	1,2
			1091.7 ^a	Ar	IR	1,2
			1090.7	N ₂	IR	2
4		OO stretch	950.3	Ar	IR	1,2
			955.7 ^a	Ar	IR	1,2
			957.3	N ₂	IR	2
5		ONO bend	772.1	Ar	IR	1,2
			782.0 ^a	Ar	IR	1,2
			790.3	N ₂	IR	2

^aLess stable matrix site.

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- ¹B.-M. Cheng, J.-W. Lee, and Y.-P. Lee, *J. Phys. Chem.* **95**, 2814 (1991).
- ²W.-J. Chen, W.-J. Lo, B.-M. Cheng, and Y.-P. Lee, *J. Chem. Phys.* **97**, 7167 (1992).
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HXeNCO \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	NCO a-stretch	2148.3	Xe	IR	1
	2	XeH stretch	1788.1	Xe	IR	1

DXeNCO \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	NCO a-stretch	2145.4	Xe	IR	1
	2	XeD stretch	1298.8	Xe	IR	1

Reference

- ¹M. Pettersson, L. Khriachtchev, J. Lundell, S. Jolkkonen, and M. Räsänen, *J. Phys. Chem. A* **104**, 3579 (2000).

HCF₃⁺ \tilde{F}^2A_1 C_{3v}
T^a=85360(400) gas PE^{1,3,4}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CH stretch	2660(80)	gas	PE	4
	3	CF ₃ stretch	1050(80)	gas	PE	4

 $\tilde{D}, \tilde{E}^2E, ^2A_1$ C_{3v}
T^a≅54400 gas PE¹⁻⁴

Broad emission bands between about 230 and 580 nm which appear on excitation of HCF₃ by synchrotron radiation in the 48–62 nm spectral region have been attributed^{5,6,8} to transitions of HCF₃⁺. A peak in this emission near 290 nm corresponds to the $\tilde{D}-\tilde{B}$ transition, and a shoulder near 430 nm to the $\tilde{D}-\tilde{C}$ transition.⁸

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁			480(80)	gas	PE	1
<i>τ</i> = 12.6(8) ns gas TPEFCO ⁷						

 \tilde{C}^2E C_{3v}
*T*₀=26220(400) gas PE¹⁻⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	2	CF ₃ stretch	1050(80)	gas	PE	1-4
	3	CF ₃ umbrella	550(80)	gas	PE	1,3,4

 \tilde{B}^2E C_{3v}
T^a=18800(1000) gas PE¹⁻⁴ \tilde{A}^2A_2 C_{3v}
T^a=13200(1000) gas PE¹⁻⁴ \tilde{X}^2A_1 C_{3v}

DCF₃⁺

\tilde{C}^2E		C_{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CF ₃ stretch	1050(80)	gas	PE	3
	3	CF ₃ umbrella	500(80)	gas	PE	3

^aFrom vertical ionization potential.

References

- ¹A. W. Potts, H. J. Lempka, D. G. Streets, and W. C. Price, *Philos. Trans. R. Soc. London, Ser. A* **268**, 59 (1970).
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8.11. Five-Atomic Nonhydrides**Ag₅**

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1			162	Ar	Ra	1
			136	Ar	Ra	1
			100	Ar	Ra	1
			68	Ar	Ra	1
b_1			174	Ar	Ra	1
			126	Ar	Ra	1
			80	Ar	Ra	1

Reference

- ¹T. L. Haslett, K. A. Bosnick, and M. Moskovits, *J. Chem. Phys.* **108**, 3453 (1998).

Y₃C₂⁺

$\tilde{X}^1A'_1$		D_{3h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'_1	2	Y ₃ stretch	228.8(4)	gas	TPE	1
e'	5	Y ₃ deformation	86	gas	TPE	1

Reference

- ¹D.-S. Yang, M. Z. Zgierski, and P. A. Hackett, *J. Chem. Phys.* **108**, 3591 (1998).

Nb₃C₂⁺

$\tilde{X}^1A'_1$		D_{3h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'_1	2	Nb ₃ s-stretch	339.2(4)	gas	TPE	1
e'	5	Nb ₃ deformation	258.8(4)	gas	TPE	1

Reference

- ¹D.-S. Yang, M. Z. Zgierski, A. Bérces, P. A. Hackett, P.-N. Roy, A. Martinez, T. Carrington, Jr., D. R. Salahub, R. Fournier, T. Pang, and C. Chen, *J. Chem. Phys.* **105**, 10663 (1996).

Y₃C₂

\tilde{X}^2B_1		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	4	Y ₃ deformation	82.4(4)	gas	TPE	1
b_1	7	Y ₃ deformation	24.5(2)	gas	TPE	1

Reference

- ¹D.-S. Yang, M. Z. Zgierski, and P. A. Hackett, *J. Chem. Phys.* **108**, 3591 (1998).

Nb₃C₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Nb ₃ s-stretch	326.3(6)	gas	TPE	1
		Nb ₃ deformation	237.6(3)	gas	TPE	1
		Nb ₃ deformation	82.7(3)	gas	TPE	1

Reference

- ¹D.-S. Yang, M. Z. Zgierski, A. Bérces, P. A. Hackett, P.-N. Roy, A. Martinez, T. Carrington, Jr., D. R. Salahub, R. Fournier, T. Pang, and C. Chen, *J. Chem. Phys.* **105**, 10663 (1996).

Nb₃N₂⁺

\tilde{X}^1A_1		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		Nb ₃ s-deformation	257	gas	TPE	1

Reference

¹D.-S. Yang, M. Z. Zgierski, A. Bérces, P. A. Hackett, A. Martinez, and D. R. Salahub, *Chem. Phys. Lett.* **227**, 71 (1997).

Al₅

\tilde{D}
 $T^a = 17710(650)$ gas PE¹

\tilde{C}
 $T^a = 14160(330)$ gas PE¹

\tilde{B}
 $T^a = 5240(180)$ gas PE¹

\tilde{A}
 $T^a = 2860(140)$ gas PE¹

\tilde{X} C_{2v}

^aFrom vertical ionization potentials.

Reference

¹G. D. Geske, A. I. Boldyrev, X. Li, and L.-S. Wang, *J. Chem. Phys.* **113**, 5130 (2000).

Al₅⁻

Threshold for electron detachment from ground-state Al₅⁻ = 18440(80) gas PE¹

Reference

¹G. D. Geske, A. I. Boldyrev, X. Li, and L.-S. Wang, *J. Chem. Phys.* **113**, 5130 (2000).

Al₄C

\tilde{C}
 $T^a = 5240(680)$ gas PE¹

\tilde{B}
 $T^a = 2820(680)$ gas PE¹

\tilde{A}
 $T^a = 1690(680)$ gas PE¹

\tilde{X} T_d

^aFrom vertical electron detachment energies.

Reference

¹X. Li, L.-S. Wang, A. I. Boldyrev, and J. Simons, *J. Am. Chem. Soc.* **121**, 6033 (1999).

Al₄Si

\tilde{C} C_{2v}
 $T_0 = 8380(290)$ gas PE¹

\tilde{B} C_{2v}
 $T^a = 6280T$ gas PE¹

\tilde{A} C_{2v}

$T_0 = 4270(290)$ gas PE¹

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁			220T	gas	PE	1

^aFrom vertical electron detachment energy.

Reference

¹A. I. Boldyrev, X. Li, and L.-S. Wang, *Angew. Chem. Int. Ed.* **39**, 3307 (2000).

Al₄Ge

\tilde{C} C_{2v}
 $T_0 = 7960(290)$ gas PE¹

\tilde{B} C_{2v}
 $T^a = 6340T$ gas PE¹

\tilde{A} C_{2v}
 $T_0 = 4400(510)$ gas PE¹

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁			160T	gas	PE	1

^aFrom vertical electron detachment energy.

Reference

¹A. I. Boldyrev, X. Li, and L.-S. Wang, *Angew. Chem. Int. Ed.* **39**, 3307 (2000).

Al₄C⁻

Vertical electron detachment energy from ground-state Al₄C⁻ = 21380(480) gas PE¹

The slow onset of electron detachment is interpreted¹ as supporting a planar structure for the anion, contrasted with a tetrahedral structure for the uncharged species.

Reference

¹X. Li, L.-S. Wang, A. I. Boldyrev, and J. Simons, *J. Am. Chem. Soc.* **121**, 6033 (1999).

Al₄Si⁻

Threshold for electron detachment from ground-state Al₄Si⁻ = 17110(160) gas PE¹

Reference

¹A. I. Boldyrev, X. Li, and L.-S. Wang, *Angew. Chem. Int. Ed.* **39**, 3307 (2000).

Al₄Ge⁻

Threshold for electron detachment from ground-state Al₄Ge⁻
= 17060(160) gas PE¹

Reference

¹A. I. Boldyrev, X. Li, and L.-S. Wang, *Angew. Chem. Int. Ed.* **39**, 3307 (2000).

Al₄N

\tilde{A} D_{4h}
T^a=8800(350) gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			330T	gas	PE	1

The photoelectron spectrum¹ of Al₄N⁻ is consistent with the occurrence of an almost isoenergetic isomer of Al₄N that has C_{2v} symmetry.

\tilde{X} D_{4h}

^aFrom vertical electron photodetachment energy.

Reference

¹S. K. Nayak, B. K. Rao, P. Jena, X. Li, and L.-S. Wang, *Chem. Phys. Lett.* **301**, 379 (1999).

CAI₃Si

\tilde{C} C_{2v}
T^a=10650(520) gas PE¹

\tilde{B} C_{2v}
T^a=4920(470) gas PE¹

\tilde{A} C_{2v}
T^a=1210(520) gas PE¹

\tilde{X} C_{2v}

^aFrom vertical electron detachment energies.

Reference

¹L.-S. Wang, A. I. Boldyrev, X. Li, and J. Simons, *J. Am. Chem. Soc.* **122**, 7681 (2000).

CAI₃Ge

\tilde{C} C_{2v}
T^a=11700(460) gas PE¹

\tilde{B} C_{2v}
T^a=5570(400) gas PE¹

\tilde{A} C_{2v}
T^a=1610(460) gas PE¹

\tilde{X} C_{2v}

^aFrom vertical electron detachment energies.

Reference

¹L.-S. Wang, A. I. Boldyrev, X. Li, and J. Simons, *J. Am. Chem. Soc.* **122**, 7681 (2000).

TiC₄

\tilde{B}
T₀=6050(360) gas PE¹

\tilde{A}
T₀=1500(510) gas PE¹

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			440(40)	gas	PE	1

Reference

¹X.-B. Wang, C.-F. Ding, and L.-S. Wang, *J. Phys. Chem. A* **101**, 7699 (1997).

TiC₄⁻

Threshold for electron detachment from ground-state
TiC₄⁻ = 12050(160) gas PE¹

Reference

¹X.-B. Wang, C.-F. Ding, and L.-S. Wang, *J. Phys. Chem. A* **101**, 7699 (1997).

Al₄N⁻ (D_{4h})

Threshold for electron detachment from Al₄N⁻ (singlet, D_{4h})=18720(240) gas PE¹

Reference

¹S. K. Nayak, B. K. Rao, P. Jena, X. Li, and L.-S. Wang, *Chem. Phys. Lett.* **301**, 379 (1999).

Al₄N⁻ (C_{2v})

Threshold for electron detachment from Al₄N⁻ (triplet, C_{2v})=13720(240) gas PE¹

Reference

¹S. K. Nayak, B. K. Rao, P. Jena, X. Li, and L.-S. Wang, *Chem. Phys. Lett.* **301**, 379 (1999).

CAI₃Si⁻

Vertical electron detachment energy from ground-state CAI₃Si⁻
= 23240(400) gas PE¹

Reference

¹L.-S. Wang, A. I. Boldyrev, X. Li, and J. Simons, *J. Am. Chem. Soc.* **122**, 7681 (2000).

CAI₃Ge⁻

Vertical electron detachment energy from ground-state CAI₃Ge⁻ = 22430(320) gas PE¹

Reference

¹L.-S. Wang, A. I. Boldyrev, X. Li, and J. Simons, *J. Am. Chem. Soc.* **122**, 7681 (2000).

C₅

$T_0 = 41967(35)T$ Ne AB¹³ 220–240 nm
44228T Ar AB⁸ 211–226 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1813	Ar	AB	8
			695(40)	Ne	AB	13
			654	Ar	AB	8

¹Π_u D_{∞h}
 $T_0 = 19566.3(2)$ gas CR¹⁸ ¹Π_u- \tilde{X} 493–411 nm
19599(8) Ne AB¹³ ¹Π_u- \tilde{X} 490–510 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			675	gas	CR	18
			697(10)	Ne	AB	13
Σ _g ⁺	2	Sym. stretch	530	gas	CR	18
			542(10)	Ne	AB	13
Π _u	7	Bend	139H	gas	CR	18
			143(5)H	Ne	AB	13

\tilde{a} D_{∞h}
 $T^a = 16060(80)$ gas PE¹⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	2	Sym. stretch	779(10)	gas	TPE	9,10
			776T	Ar	IR	1
Σ _u ⁺	3	Asym. stretch	2169.441	gas	IR,DL	4–7
			2166.4	Ne	IR	2,12,13,16
			2164	Ar	IR	1–3
			2157.0	Kr	IR	15
			2164.5	H ₂	IR	14
	4	Asym. stretch	1444.3	Ne	IR	12,16
			1446.6	Ar	IR	11
			1443.2	Kr	IR	15
Π _g	5	Bend	218(13)T	gas	DL,TPE	7,9,10
Π _u	6	Bend	535(10)	gas	TPE	9,10
Π _u	7	Bend	118(3)T	gas	DL,TPE	7,9,10

$B_0 = 0.0853$ IR⁴DL^{5–7}

^aFrom vertical ionization potentials.

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¹⁶S. Tam, M. Macler, and M. E. Fajardo, *J. Chem. Phys.* **106**, 8955 (1997).
¹⁷M. Kohno, S. Suzuki, H. Shiromaru, and Y. Achiba, *J. Chem. Phys.* **110**, 3781 (1999).
¹⁸T. Motylewski, O. Vaizert, T. F. Giesen, H. Linnartz, and J. P. Maier, *J. Chem. Phys.* **111**, 6161 (1999).

C₄Si

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	CC stretch	2095.46	gas	DL	3
			2080.1	Ar	IR	2

$B_0 = 0.051$ MW^{1,4}DL³

References

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SiCCCSi

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			420(25)	gas	PE	3

$\tilde{X}^1\Sigma_g^+$ D _{∞h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	C ₃ s-stretch	1520(25)	gas	PE	3
	2	SiC s-stretch	490(25)	gas	PE	3
Σ_u^+	3	C ₃ a-stretch	1968.19	gas	DL	2
	4	SiC a-stretch	898.9w	Ar	IR	1

$B_0=0.0316$ DL²

References

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Si₅

$T_0=9360(400)$ gas PE¹

³B₁ C_{2v}
 $T_0=4680(400)$ gas PE¹

\tilde{X}^1A_1' D _{3h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1'			233(20)	gas	PE	1

Reference

- ¹C. Xu, T. R. Taylor, G. R. Burton, and D. M. Neumark, *J. Chem. Phys.* **108**, 1395 (1998).

GeCCCGe

\tilde{X} D _{∞h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_u	3	C ₃ a-stretch	1920.7	Ar	IR	1

Reference

- ¹D. L. Robbins, C. M. L. Rittby, and W. R. M. Graham, *J. Chem. Phys.* **114**, 3570 (2001).

BeOBeNO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		BeO stretch	1484.5 1483.0	Ar	IR	1

Reference

- ¹G. P. Kushto, F. Ding, B. Liang, X. Wang, A. Citra, and L. Andrews, *Chem. Phys.* **257**, 223 (2000).

Sc(CO)₂⁺

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO a-stretch	1926.0	Ne	IR	1

Reference

- ¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 2964 (1999).

Fe(CO)₂⁺

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2134.0	Ne	IR	1

Reference

- ¹M. Zhou and L. Andrews, *J. Chem. Phys.* **110**, 10370 (1999).

Co(CO)₂⁺

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2168.9	Ne	IR	1

Reference

- ¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7773 (1999).

Rh(CO)₂⁺

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2184.7	Ne	IR	1,2

References

- ¹M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **121**, 9171 (1999).
²M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7773 (1999).

Ir(CO)₂⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2153.8	Ne	IR	1

Reference¹M. Zhou and L. Andrews, J. Phys. Chem. A **103**, 7773 (1999).**Ni(CO)₂⁺** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO a-stretch	2205.3	Ne	IR	1

Reference¹B. Liang, M. Zhou, and L. Andrews, J. Phys. Chem. A **104**, 3905 (2000).**Pd(CO)₂⁺** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO a-stretch	2210.5	Ne	IR	1

Reference¹B. Liang, M. Zhou, and L. Andrews, J. Phys. Chem. A **104**, 3905 (2000).**Pt(CO)₂⁺** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO a-stretch	2210.3	Ne	IR	1

Reference¹B. Liang, M. Zhou, and L. Andrews, J. Phys. Chem. A **104**, 3905 (2000).**Cu(CO)₂⁺** \tilde{X} D_{∞h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2230.4	Ne	IR	1

Reference¹M. Zhou and L. Andrews, J. Chem. Phys. **111**, 4548 (1999).**Ag(CO)₂⁺** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2234.6	Ne	IR	1

Reference¹B. Liang and L. Andrews, J. Phys. Chem. A **104**, 9156 (2000).**Au(CO)₂⁺** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2233.4	Ne	IR	1

Reference¹B. Liang and L. Andrews, J. Phys. Chem. A **104**, 9156 (2000).**Ga₂P₃** \tilde{X}^2A_2'' D_{3h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁ '			213	gas	PE	1

Reference¹T. R. Taylor, H. Gómez, K. R. Asmis, and D. M. Neumark, J. Chem. Phys. **115**, 4620 (2001).**Ga₂As₃** \tilde{X}^2A_2'' D_{3h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁ '			193	gas	PE	1

Reference¹T. R. Taylor, H. Gómez, K. R. Asmis, and D. M. Neumark, J. Chem. Phys. **115**, 4620 (2001).

Al₃O₂

\tilde{C}
 $T_0 = 23400(800)$ gas PE¹

\tilde{B}
 $T_0 = 19930(940)$ gas PE¹

\tilde{A}
 $T_0 = 10970(630)$ gas PE¹

Reference

¹H. Wu, X. Li, X.-B. Wang, C.-F. Ding, and L.-S. Wang, *J. Chem. Phys.* **109**, 449 (1998).

C₅⁻

\tilde{C} D_{∞h}
 $T_0 = 27785(3)$ gas PD⁹ $\tilde{C}-\tilde{X}$ 337–360 nm
 27847(15) Ne AB⁶ $\tilde{C}-\tilde{X}$ 311–360 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	1		1792(6)	gas	PD	9
			1756(20)	Ne	AB	6
2			754(6)	gas	PD	9
			724(20)	Ne	AB	6
	7	Deformation	168H	gas	PD	9

\tilde{B}
 $T_0 = 24563(3)$ gas PD⁹ $\tilde{B}-\tilde{X}$ 372–407 nm
 24540(20) Ne AB⁹ $\tilde{B}-\tilde{X}$ 372–408 nm

Threshold for electron detachment from ground-state C₅⁻ = 23020(10) gas TPE¹PE²

$\tilde{A} \ ^2\Pi_g$ D_{∞h}
 $T_0 = 20164$ gas MPD^{3,7} $\tilde{A}-\tilde{X}$ 446–496 nm
 20200(8) Ne AB⁶ $\tilde{A}-\tilde{X}$ 457–495 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	1		1649	gas	MPD	7
			1679(9)	Ne	AB	6
	2		729	gas	MPD	3,7
			737(8)	Ne	AB	6

$\tilde{X} \ ^2\Pi_u$ D_{∞h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _u ⁺	3	Asym. stretch	1822.3	Ne	IR	5
			1831.8	Ar	IR	4,8
Π _u	7	Bend	200T	gas	TPE	1

A = 22T gas TPE¹

References

¹T. N. Kitsopoulos, C. J. Chick, Y. Zhao, and D. M. Neumark, *J. Chem. Phys.* **95**, 5479 (1991).

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⁹N. M. Lakin, F. Güthe, M. Tulej, M. Pachkov, and J. P. Maier, *Faraday Discuss.* **115**, 383 (2000).

SiCCCSi⁻

Threshold for electron detachment from ground-state SiCCCSi⁻ = 14250(100) gas PE¹

$\tilde{X} \ ^2\Pi$ D_{∞h}
 A = 60(30) PE¹

Reference

¹X. Duan, L. W. Burggraf, D. E. Weeks, G. E. Davico, R. L. Schwartz, and W. C. Lineberger, *J. Chem. Phys.* **116**, 3601 (2002).

Si₅⁻

Threshold for electron detachment from ground-state Si₅⁻ = 20900(160) gas PE¹

Reference

¹C. Xu, T. R. Taylor, G. R. Burton, and D. M. Neumark, *J. Chem. Phys.* **108**, 1395 (1998).

Sc(CO)₂

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁		CO stretch	1851.4	Ar	IR	1
b ₂		CO stretch	1716.3	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 2964 (1999).

Ti(CO)₂

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CO s-stretch	1893.1	Ne	IR	1
b ₂		CO a-stretch	1799.3	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 5259 (1999).

Zr(CO)₂

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CO stretch	1855.5	Ne	IR	1
b ₂		CO stretch	1785.1	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **122**, 1531 (2000).

OZrCCO

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CCO a-stretch	2046.4	Ne	IR	1
		ZrO stretch	902.3	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **122**, 1531 (2000).

Hf(CO)₂

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CO stretch	1850.9	Ne	IR	1
b ₂		CO stretch	1806.7	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **122**, 1531 (2000).

OHfCCO

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CCO a-stretch	2063.5	Ne	IR	1
		HfO stretch	897.2	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **122**, 1531 (2000).

V(CO)₂

\tilde{X}		C _{2v} Structure: ESR ²				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CO s-stretch	1944.0	Ne	IR	3
b ₂		CO a-stretch	1844.2	Ne	IR	3
			1820	Ar	IR	1
			1816	Kr	IR	1
			1814	Xe	IR	1

References

¹L. Hanlan, H. Huber, and G. A. Ozin, *Inorg. Chem.* **15**, 2592 (1976).

²R. J. Van Zee, S. B. H. Bach, and W. Weltner, Jr., *J. Phys. Chem.* **90**, 583 (1986).

³M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 5259 (1999).

Nb(CO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1847.2	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7785 (1999).

ONbCCO

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2065.5	Ne	IR	1
		NbO stretch	950.9	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7785 (1999).

Ta(CO)₂

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1918.3	Ne	IR	1
		CO stretch	1838.6	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7785 (1999).

OTaCCO

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2087.5	Ne	IR	1
		TaO stretch	965.7	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, J. Phys. Chem. A **103**, 7785 (1999).

Mn(CO)₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO s-stretch	1972.8	Ne	IR	1
		CO a-stretch	1851.5	Ne	IR	1
			1832.5	Ar	IR	1

Reference

¹L. Andrews, M. Zhou, X. Wang, and C. W. Bauschlicher, Jr., J. Phys. Chem. A **104**, 8887 (2000).

Re(CO)₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1905.4	Ne	IR	1
			1895.9	Ar	IR	1

Reference

¹L. Andrews, M. Zhou, X. Wang, and C. W. Bauschlicher, Jr., J. Phys. Chem. A **104**, 8887 (2000).

Fe(CO)₂ ${}^3B_2^a$ C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CO stretch	1984.8w	Ar	IR	3
b ₂		CO stretch	1879.2	Ar	IR	3

 $\tilde{X} \ 3\Sigma_g^-$ D_{∞h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _u ⁺	3	CO a-stretch	1928.18	gas	IR,DL	1,2,5
			1917.1	Ne	IR	4

B₀ = 0.0472 DL^{2,5}

^aIt has been proposed³ that in an argon matrix this low-lying state occurs below the ${}^3\Sigma_g^-$ state.

References

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²K. Tanaka, Y. Tachikawa, and T. Tanaka, Chem. Phys. Lett. **281**, 285 (1997).
³M. Zhou, G. V. Chertihin, and L. Andrews, J. Chem. Phys. **109**, 10893 (1998).
⁴M. Zhou and L. Andrews, J. Chem. Phys. **110**, 10370 (1999).
⁵K. Tanaka, Y. Tachikawa, K. Sakaguchi, T. Hikida, and T. Tanaka, J. Chem. Phys. **111**, 3970 (1999).

Ru(CO)₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO a-stretch	1955.3	Ne	IR	1
			1915.3 ^a	Ar	IR	1

^aIt is likely that for Ru(CO)₂, as for Fe(CO)₂, the ground electronic state of the molecule trapped in an argon matrix is not the same as that of the molecule trapped in a neon matrix.¹

Reference

¹M. Zhou and L. Andrews, J. Phys. Chem. A **103**, 6956 (1999).

Os(CO)₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO a-stretch	1975.2	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, J. Phys. Chem. A **103**, 6956 (1999).

Co(CO)₂ \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b ₂	7	CO stretch	1945.6	Ne	IR	3
			1920.8	Ar	IR	1,2

References

- ¹L. A. Hanlan, H. Huber, E. P. Kundig, B. R. McGarvey, and G. A. Ozin, J. Am. Chem. Soc. **97**, 7054 (1975).
²M. Zhou and L. Andrews, J. Phys. Chem. A **102**, 10250 (1998).
³M. Zhou and L. Andrews, J. Phys. Chem. A **103**, 7773 (1999).

Rh(CO)₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2031.0	Ne	IR	2,3
			2014.6	Ar	IR	1,3

References

- ¹G. A. Ozin and A. J. L. Hanlan, *Inorg. Chem.* **18**, 2091 (1979).
²M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **121**, 9171 (1999).
³M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7773 (1999).

Ir(CO)₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	2014.5	Ne	IR	1

Reference

- ¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7773 (1999).

Ni(CO)₂ \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CO stretch	2100(80)	gas	PE	2
			2089.7	Ne	IR	5
			2071.5w	Ar	IR	1,4
	2		461.5Tvw	Ar	IR	4
	3		449.5Tvw	Ar	IR	4
b ₁	6		398.9vw	Ar	IR	4
b ₂	7	CO stretch	1978.9	Ne	IR	5
			1965.5vs	Ar	IR	1,3,4
	8	NiC stretch	515.5	Ar	IR	1,3,4
	9		335vw	Ar	IR	4

References

- ¹R. L. DeKock, *Inorg. Chem.* **10**, 1205 (1971).
²A. E. Stevens, C. S. Feigerle, and W. C. Lineberger, *J. Am. Chem. Soc.* **104**, 5026 (1982).
³M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **120**, 11499 (1998).
⁴L. Manceron and M. E. Alikhani, *Chem. Phys.* **244**, 215 (1999).
⁵B. Liang, M. Zhou, and L. Andrews, *J. Phys. Chem. A* **104**, 3905 (2000).

Pd(CO)₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO a-stretch	2065.8	Ne	IR	2
			2051.6	Ar	IR	1

References

- ¹B. Tremblay and L. Manceron, *Chem. Phys.* **250**, 187 (1999).
²B. Liang, M. Zhou, and L. Andrews, *J. Phys. Chem. A* **104**, 3950 (2000).

Pt(CO)₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO a-stretch	2053.2	Ne	IR	2
			2039.8	Ar	IR	1
		PtCO deform.	801.9	Ar	IR	1
		PtC a-stretch	383.4	Ar	IR	1

References

- ¹L. Manceron, B. Tremblay, and M. E. Alikhani, *J. Phys. Chem. A* **104**, 3750 (2000).
²B. Liang, M. Zhou, and L. Andrews, *J. Phys. Chem. A* **104**, 3950 (2000).

Cu(CO)₂ \tilde{X} D_{∞h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1904.4	Ne	IR	2
			1890.7	Ar	IR	1,2

References

- ¹H. Huber, E. P. Kündig, M. Moskovits, and G. A. Ozin, *J. Am. Chem. Soc.* **97**, 2097 (1975).
²M. Zhou and L. Andrews, *J. Chem. Phys.* **111**, 4548 (1999).

Ag(CO)₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1858.7	Ne	IR	2
			1842.0	Ar	IR	1
			1827.5			
			1798.8	Kr	IR	1
			1790.0			
			1802.0	Xe	IR	1
			1790.8			

References

- ¹D. McIntosh and G. A. Ozin, *J. Am. Chem. Soc.* **98**, 3167 (1976).
²B. Liang and L. Andrews, *J. Phys. Chem. A* **104**, 9156 (2000).

Au(CO)₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1943.7	Ne	IR	1

Reference

- ¹B. Liang and L. Andrews, *J. Phys. Chem. A* **104**, 9156 (2000).

Th(CO)₂ \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CO s-stretch	1827.7	Ne	IR	1
b ₂		CO a-stretch	1775.6	Ne	IR	1

Reference

- ¹J. Li, B. E. Bursten, M. Zhou, and L. Andrews, *Inorg. Chem.* **40**, 5448 (2001).

OThCCO \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CO stretch	2048.6vs	Ne	IR	1
	2	CC stretch	1353.6vw	Ne	IR	1
		ThO stretch	822.5w	Ne	IR	1

Reference

- ¹J. Li, B. E. Bursten, M. Zhou, and L. Andrews, *Inorg. Chem.* **40**, 5448 (2001).

U(CO)₂ \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CO stretch	1840.2	Ne	IR	1
b ₂	7	CO stretch	1790.8	Ne	IR	1

Reference

- ¹M. Zhou, L. Andrews, J. Li, and B. E. Bursten, *J. Am. Chem. Soc.* **121**, 9712 (1999).

OUCCO \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	CO stretch	2051.5	Ne	IR	1
			1361.8	Ne	IR	1
		UO stretch	841.0	Ne	IR	1

Reference

- ¹M. Zhou, L. Andrews, J. Li, and B. E. Bursten, *J. Am. Chem. Soc.* **121**, 9712 (1999).

NNPtNN \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NN stretch	2206.9	Ne	IR	2
			2195.4	Ar	IR	1,2
			2205.7	N ₂	IR	1,2

References

- ¹D. W. Green, J. Thomas, and D. M. Gruen, *J. Chem. Phys.* **58**, 5453 (1973).
²A. Citra, X. Wang, W. D. Bare, and L. Andrews, *J. Phys. Chem. A* **105**, 7799 (2001).

NNThNN \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NN stretch	1895.5	Ar	IR	1

Reference

- ¹G. P. Kushto, P. F. Souter, and L. Andrews, *J. Chem. Phys.* **108**, 7121 (1998).

OVOVO \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1026.2	Ar	IR	1

Reference

¹G. V. Chertihin, W. D. Bare, and L. Andrews, *J. Phys. Chem. A* **101**, 5090 (1997).

Co₂O₃ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			952.5T	Ar	IR	1

Reference

¹G. V. Chertihin, A. Citra, L. Andrews, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **101**, 8793 (1997).

Ni₂O₃ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			962.3	Ar	IR	1

Reference

¹A. Citra, G. V. Chertihin, L. Andrews, and M. Neurock, *J. Phys. Chem. A* **101**, 3109 (1997).

Cu₂O₃ \tilde{B}

$T_0=6290(400)$ gas PE¹

 \tilde{A}

$T_0=3870(400)$ gas PE¹

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			995.3	Ar	IR	2
			640T	gas	PE	1

References

¹L.-S. Wang, H. Wu, S. R. Desai, and L. Lou, *Phys. Rev. B* **53**, 8028 (1996).

²G. V. Chertihin, L. Andrews, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **101**, 4026 (1997).

Ga₂P₃⁻

Threshold for electron detachment from ground-state Ga₂P₃⁻ = 24130(20) gas PE¹

Reference

¹T. R. Taylor, H. Gómez, K. R. Asmis, and D. M. Neumark, *J. Chem. Phys.* **115**, 4620 (2001).

Ga₂As₃⁻

Threshold for electron detachment from ground-state Ga₂As₃⁻ = 22450(200) gas PE¹

Reference

¹T. R. Taylor, H. Gómez, K. R. Asmis, and D. M. Neumark, *J. Chem. Phys.* **115**, 4620 (2001).

Al₃O₂⁻

Threshold for electron detachment from ground-state Al₃O₂⁻ = 17590(480) gas PE¹

Evidence for the participation of a second electronic state of Al₃O₂⁻ in its photoelectron spectrum is presented by Ref. 2.

References

¹H. Wu, X. Li, X.-B. Wang, C.-F. Ding, and L.-S. Wang, *J. Chem. Phys.* **109**, 449 (1998).

²T. K. Ghanty and E. R. Davidson, *J. Phys. Chem. A* **103**, 8985 (1999).

C₄S \tilde{B}

$T_0 \cong 41700T$ Ar AB¹

 $\tilde{A}^3\Sigma^-$

$T_0 = 22380(10)$ Ne AB⁴
22220T Ar AB¹

$\tilde{A}-\tilde{X}$ 400–447 nm
 $\tilde{A}-\tilde{X}$ 380–450 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1		2022(10)	Ne	AB	4
	2		1595(10)	Ne	AB	4
	3		1060(10)	Ne	AB	4
	4		584(10)	Ne	AB	4

 $\tilde{X}^3\Sigma^-$ $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1746.8	Ar	IR	3

$B_0 = 0.051$ MW²

References

¹G. Maier, J. Schrot, and H. P. Reisenauer, *Chem. Ber.* **124**, 2613 (1991).

²Y. Hirahara, Y. Ohshima, and Y. Endo, *Astrophys. J.* **408**, L113 (1993).

³J. Szczepanski, R. Hodyss, J. Fuller, and M. Vala, *J. Phys. Chem. A* **103**, 2975 (1999).

⁴E. Riaplov, M. Wyss, N. M. Lakin, and J. P. Maier, *J. Phys. Chem. A* **105**, 4894 (2001).

Sn(CN)₂

\tilde{X}^1A_1 C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁		SnC stretch	370(60)	gas	PE	1

Reference

¹V. D. Moravec and C. C. Jarrold, *J. Chem. Phys.* **113**, 1035 (2000).

Ti(CO)₂⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CO s-stretch	1742.1	Ne	IR	1
b ₂		CO a-stretch	1677.5	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 5259 (1999).

Zr(CO)₂⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CO stretch	1745.9	Ne	IR	1
b ₂		CO stretch	1682.2	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **122**, 1531 (2000).

Hf(CO)₂⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CO stretch	1747.4	Ne	IR	1
b ₂		CO stretch	1681.0	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **122**, 1531 (2000).

V(CO)₂⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CO s-stretch	1777.9	Ne	IR	1
b ₂		CO a-stretch	1670.6	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 5259 (1999).

Nb(CO)₂⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CO stretch	1768.6	Ne	IR	1
b ₂	7	CO stretch	1656.7	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7785 (1999).

Ta(CO)₂⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1773.6T	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7785 (1999).

Mn(CO)₂⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1756.2	Ne	IR	1
			1754.9	Ar	IR	1

Reference

¹L. Andrews, M. Zhou, X. Wang, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **104**, 8887 (2000).

Re(CO)₂⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1788.7	Ne	IR	1

Reference

¹L. Andrews, M. Zhou, X. Wang, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **104**, 8887 (2000).

Fe(CO)₂⁻

${}^2A_1^a$ C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁		CO s-stretch	1815.0w	Ar	IR	1
<i>b</i> ₂		CO a-stretch	1721.9	Ar	IR	1

$\tilde{X} {}^2\Pi_u$ D _{∞h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_u^+		CO a-stretch	1732.9	Ne	IR	2

^aIt has been proposed² that in an argon matrix this low-lying state occurs below the ²Π_u state.

References

¹M. Zhou, G. V. Chertihin, and L. Andrews, *J. Chem. Phys.* **109**, 10893 (1998).
²M. Zhou and L. Andrews, *J. Chem. Phys.* **110**, 10370 (1999).

Ru(CO)₂⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CO s-stretch	1830.6	Ne	IR	1
			1834.2	Ar	IR	1
<i>b</i> ₂		CO a-stretch	1769.6	Ne	IR	1
			1756.9	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 6956 (1999).

Os(CO)₂⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CO s-stretch	1898.8	Ne	IR	1
<i>b</i> ₂		CO a-stretch	1783.5	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 6956 (1999).

Co(CO)₂⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CO stretch	1847.1	Ne	IR	2
			1860.2	Ar	IR	1
<i>b</i> ₂	7	CO stretch	1789.2	Ne	IR	2
			1768.9	Ar	IR	1

References

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **102**, 10250 (1998).
²M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7773 (1999).

Rh(CO)₂⁻

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CO stretch	1900.4	Ne	IR	1,2
<i>b</i> ₂	7	CO stretch	1816.7	Ne	IR	1,2
			1799.4	Ar	IR	2

References

¹M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **121**, 9171 (1999).
²M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7773 (1999).

Ir(CO)₂⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1818.1	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 7773 (1999).

Ni(CO)₂⁻

Threshold for electron detachment from ground-state Ni(CO)₂⁻
= 5190(110) gas PE¹

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO a-stretch	1813.8	Ne	IR	3
			1801.7	Ar	IR	2

References

¹A. E. Stevens, C. S. Feigerle, and W. C. Lineberger, *J. Am. Chem. Soc.* **104**, 5026 (1982).

²M. Zhou and L. Andrews, *J. Am. Chem. Soc.* **120**, 11499 (1998).

³B. Liang, M. Zhou, and L. Andrews, *J. Phys. Chem. A* **104**, 3905 (2000).

Pd(CO)₂⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO a-stretch	1832.3	Ne	IR	1

Reference

¹B. Liang, M. Zhou, and L. Andrews, *J. Phys. Chem. A* **104**, 3905 (2000).

Pt(CO)₂⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO a-stretch	1844.0	Ne	IR	1

Reference

¹B. Liang, M. Zhou, and L. Andrews, *J. Phys. Chem. A* **104**, 3905 (2000).

Cu(CO)₂⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1793.9	Ne	IR	1
			1780.8	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Chem. Phys.* **111**, 4548 (1999).

Th(CO)₂⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1719.1	Ne	IR	1

Reference

¹J. Li, B. E. Bursten, M. Zhou, and L. Andrews, *Inorg. Chem.* **40**, 5448 (2001).

U(CO)₂⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1661.2	Ne	IR	1

Reference

¹M. Zhou, L. Andrews, J. Li, and B. E. Bursten, *J. Am. Chem. Soc.* **121**, 9712 (1999).

Pt(NN)₂⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NN stretch	1869.0	Ne	IR	1
			1862.5	N ₂	IR	1

Reference

¹A. Citra, X. Wang, W. D. Bare, and L. Andrews, *J. Phys. Chem. A* **105**, 7799 (2001).

Al(CO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CO s-stretch	1995.3m	Ar	IR	2-4
			1988.7			
			1988	Kr	IR	1
			2003.0	N ₂	IR	3
			497.0w	Ar	IR	2
b ₁		Wag	234.7vw	Ar	IR	2
b ₂		CO a-stretch	1907.6vs	Ar	IR	2-4
			1891.0			
			1890	Kr	IR	1
			1912.8	N ₂	IR	3
		AlCO deform.	281.0w	Ar	IR	2

References

- ¹A. J. Hinchliffe, J. S. Ogden, and D. D. Oswald, *J. Chem. Soc., Chem. Commun.* 388 (1972).
²C. Xu, L. Manceron, and J. P. Perchard, *J. Chem. Soc., Faraday Trans.* **89**, 1291 (1993).
³A. Feltrin, M. Guido, and S. Nunziante Cesaro, *Vib. Spectrosc.* **8**, 175 (1995).
⁴L. Zhang, J. Dong, M. Zhou, and Q. Qin, *J. Chem. Phys.* **113**, 10169 (2000).

Ga(CO)₂

\tilde{X}		C _{2v}							
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.			
a ₁	1	CO s-stretch	2016.8	Ar	IR	1,2			
			2011.8						
			2009.2						
			2006.1				Kr	IR	2
b ₂		CO a-stretch	2040.2	N ₂	IR	1			
			1933.6				Ar	IR	2
			1920.4						
			1912.4						
			1909.3						
1938.7	N ₂	IR	1						

References

- ¹A. Feltrin, M. Guido, and S. Nunziante Cesaro, *Vib. Spectrosc.* **8**, 175 (1995).
²H.-J. Himmel, A. J. Downs, J. C. Green, and T. M. Greene, *J. Phys. Chem. A* **104**, 3642 (2000).

In(CO)₂

\tilde{X}		C _{2v}							
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.			
		CO s-stretch	2029.2	Ar	IR	1			
			1980.2				Ar	IR	1
			483w				Ar	IR	1

Reference

- ¹H.-J. Himmel, A. J. Downs, J. C. Green, and T. M. Greene, *J. Phys. Chem. A* **104**, 3642 (2000).

Sn(CN)₂⁻

Threshold for electron detachment from ground-state Sn(CN)₂⁻ = 21160(32) gas PE¹

Reference

- ¹V. D. Moravec and C. C. Jarrold, *J. Chem. Phys.* **113**, 1035 (2000).

Ni(NO)₂⁺

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO a-stretch	1926.2	Ne	IR	1

Reference

- ¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **104**, 3915 (2000).

C₄O⁻

$\tilde{B}^2\Pi$		C _{∞v}				
T ₀ = 22610(10)		Ne AB ¹		$\tilde{B}-\tilde{X}$ 417–443 nm		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	4		676(10)	Ne	AB	1

$\tilde{A}^2\Sigma^+$		C _{∞v}				
T ₀ = 11434(3)		Ne AB ¹		$\tilde{A}-\tilde{X}$ 870–875 nm		

Reference

- ¹E. Riaplov, M. Wyss, N. M. Lakin, and J. P. Maier, *J. Phys. Chem. A* **105**, 4894 (2001).

V(NO)₂

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	NO s-stretch	1736.8	Ar	IR	1
b ₂		NO a-stretch	1627.8	Ne	IR	2
			1614.5	Ar	IR	1

References

- ¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 478 (1999).
²L. Andrews and X. Wang, *J. Phys. Chem. A* **106**, 1196 (2002).

Nb(NO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO stretch	1583.6	Ar	IR	1

Reference

- ¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **102**, 10025 (1998).

Ta(NO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO stretch	1574.8	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **102**, 10025 (1998).

Cr(NO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO stretch	1628.8	Ne	IR	2
			1623.3	Ar	IR	1

References

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **102**, 7452 (1998).

²L. Andrews and X. Wang, *J. Phys. Chem. A* **106**, 1196 (2002).

Mo(NO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO stretch	1642.7	Ar	IR	1

Reference

¹L. Andrews and M. Zhou, *J. Phys. Chem. A* **103**, 4167 (1999).

W(NO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO stretch	1639.2	Ar	IR	1

Reference

¹L. Andrews and M. Zhou, *J. Phys. Chem. A* **103**, 4167 (1999).

Mn(NO)₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	NO stretch	1744.7	Ar	IR	1
b ₂		NO stretch	1706.2	Ne	IR	2
			1693.0	Ar	IR	1

References

¹L. Andrews, M. Zhou, and D. W. Ball, *J. Phys. Chem. A* **102**, 10041 (1998).

²L. Andrews and X. Wang, *J. Phys. Chem. A* **106**, 1196 (2002).

Re(NO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO stretch	1651.6	Ar	IR	1

Reference

¹L. Andrews, M. Zhou, and D. W. Ball, *J. Phys. Chem. A* **102**, 10041 (1998).

Fe(NO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO s-stretch	1810.8	Ne	IR	2
			1798.1	Ar	IR	2
		NO a-stretch	1744.6	Ne	IR	2
			1731.6	Ar	IR	1,2

References

¹G. K. Ruschel, T. M. Nemetz, and D. W. Ball, *J. Mol. Struct.* **384**, 101 (1996).

²M. Zhou and L. Andrews, *J. Phys. Chem. A* **104**, 3915 (2000).

Ru(NO)₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO s-stretch	1805.3	Ne	IR	1
		NO a-stretch	1721.4	Ne	IR	1
			1709.6	Ar	IR	1

Reference

¹A. Citra and L. Andrews, *J. Phys. Chem. A* **104**, 8689 (2000).

Os(NO)₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
1		NO s-stretch	1834.7 1823.8 1815.8 1811.1	Ne	IR	1
		NO a-stretch	1726.9 1723.2 1717.9 1712.5	Ne Ar	IR IR	1 1

Reference¹A. Citra and L. Andrews, *J. Phys. Chem. A* **104**, 8689 (2000).**Co(NO)₂** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
1		NO s-stretch	1827.2	Ar	IR	1
		NO a-stretch	1749.1 1737.6	Ne Ar	IR IR	1 1
		CoN stretch	581.5	Ar	IR	1

Reference¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **104**, 3915 (2000).**Rh(NO)₂** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
1		NO s-stretch	1801.8	Ar	IR	1
		NO a-stretch	1746.9 1736.5	Ne Ar	IR IR	1 1

Reference¹A. Citra and L. Andrews, *J. Phys. Chem. A* **104**, 11897 (2000).**Ir(NO)₂** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO a-stretch	1743.2	Ar	IR	1

Reference¹A. Citra and L. Andrews, *J. Phys. Chem. A* **104**, 11897 (2000).**Ni(NO)₂** \tilde{X} D_{∞h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _u ⁺	3	NO a-stretch	1762.0 1750.2vs	Ne Ar	IR IR	2 1-3
		NiN a-stretch	524.4wm	Ar	IR	2,3
Π _u	6	Bend	623.9vw	Ar	IR	3

References¹G. K. Ruschel, T. M. Nemetz, and D. W. Ball, *J. Mol. Struct.* **384**, 101 (1996).²M. Zhou and L. Andrews, *J. Phys. Chem. A* **104**, 3915 (2000).³M. E. Alikhani, L. Krim, and L. Manceron, *J. Phys. Chem. A* **105**, 7817 (2001).**Pd(NO)₂** \tilde{X} D_{∞h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _u ⁺	3	NO a-stretch	1753.0 1739.0vs	Ne Ar	IR IR	1 1,2
		PdN a-stretch	433.5w	Ar	IR	2
Π _u	6	Bend	494.5vw	Ar	IR	2

References¹A. Citra and L. Andrews, *J. Phys. Chem. A* **104**, 8160 (2000).²M. E. Alikhani, L. Krim, and L. Manceron, *J. Phys. Chem. A* **105**, 7817 (2001).**Pt(NO)₂** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO a-stretch	1776.9T 1763.8	Ne Ar	IR IR	1 1

Reference¹A. Citra and L. Andrews, *J. Phys. Chem. A* **104**, 8160 (2000).**Cu(NO)₂** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO a-stretch	1637.2 1611.6	Ne Ar	IR IR	1 1
		CuN a-stretch	615.1vw 608.6	Ne Ar	IR IR	1 1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **104**, 2618 (2000).

Al(CO)₂⁻

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CO s-stretch	1803.9	Ar	IR	1
b ₂	7	CO a-stretch	1741.0	Ar	IR	1

Reference

¹L. Zhang, J. Dong, M. Zhou, and Q. Qin, *J. Chem. Phys.* **113**, 10169 (2000).

Si(CO)₂

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CO s-stretch	1995.9w	Ar	IR	2
b ₂	7	CO a-stretch	1928.3	Ar	IR	1,2

References

¹R. R. Lembke, R. F. Ferrante, and W. Weltner, Jr., *J. Am. Chem. Soc.* **99**, 416 (1977).

²L. Zhang, J. Dong, and M. Zhou, *J. Chem. Phys.* **113**, 8700 (2000).

Ge(CO)₂

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CO s-stretch	2000.8	Ar	IR	1
b ₂	7	CO a-stretch	1934.8	Ar	IR	1

Reference

¹L. Zhang, J. Dong, and M. Zhou, *J. Chem. Phys.* **113**, 8700 (2000).

Sn(CO)₂

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CO s-stretch	1996.6	Ar	IR	1
b ₂	7	CO a-stretch	1904.3	Ar	IR	1

Reference

¹L. Zhang, J. Dong, and M. Zhou, *J. Chem. Phys.* **113**, 8700 (2000).

NCCNO

In the gas phase, absorption has been observed³ between 210 and 290 nm, with a maximum at 251.7 nm (39730).

\tilde{X}		C _{∞v}					Structure: MW ⁴	
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.		
Σ	1	CNO a-stretch	2328vs	gas	IR	3		
			2356vs	Ar	IR	1		
	2	CN stretch	2192m	gas	IR	3		
			2192w	Ar	IR	1		
3	CNO s-stretch	1442s	gas	IR	3			
		1445wm	Ar	IR	1			
4	CC stretch	714.75vw	gas	IR	2,3			
		717vw	Ar	IR	1			
II	5	CNO bend	422T	gas	IR	3		
	6	NCC bend	403.93w	gas	IR	2,3		
			407vw	Ar	IR	1		
7	CCN bend	80.52	gas	IR,MW	3,5			

B₀ = 0.077 IR^{2,5}MW^{4,5}

References

¹G. Maier and J. H. Teles, *Angew. Chem.* **99**, 152 (1987); *Angew. Chem. Int. Ed. Engl.* **26**, 155 (1987).

²B. Guo, T. Pasinszki, N. P. C. Westwood, K. Zhang, and P. F. Bernath, *J. Chem. Phys.* **105**, 4457 (1996).

³T. Pasinszki and N. P. C. Westwood, *J. Phys. Chem.* **100**, 16856 (1996).

⁴Th. Brupbacher, R. K. Bohn, W. Jäger, M. C. L. Gerry, T. Pasinszki, and N. P. C. Westwood, *J. Mol. Spectrosc.* **181**, 316 (1997).

⁵H. Lichau, S. C. Ross, M. Lock, S. Albert, B. P. Winnewisser, M. Winnewisser, and F. C. De Lucia, *J. Phys. Chem. A* **105**, 10080 (2001).

Cr(NO)₂⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO stretch	1463.1	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **102**, 7452 (1998).

Co(NO)₂⁻

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO a-stretch	1593.8	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **104**, 3915 (2000).

Ni(NO)₂⁻ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO a-stretch	1592.2	Ne	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **104**, 3915 (2000).

Cu(NO)₂⁻ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO a-stretch	1565.2	Ne	IR	1
			1551.2	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **104**, 2618 (2000).

Si(N₂)₂ \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	NN s-stretch	2045.0	N ₂	IR	1
b ₂	7	NN a-stretch	1912.9	N ₂	IR	1

Reference

¹G. Maier, H. P. Reisenauer, and J. Glatthaar, *Organomet.* **19**, 4775 (2000).

(cyc-O₂V)O₂ \tilde{C}

T₀ = 16140(800) gas PE²

 \tilde{B}

T₀ = 11300(800) gas PE²

 \tilde{A}

T₀ = 8070(800) gas PE²

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OO stretch	1121.9	Ar	IR	1
			1126.4	N ₂	IR	1
		OVO s-stretch	975.3	Ar	IR	1
		OVO a-stretch	974.1s	Ar	IR	1
			968.8	N ₂	IR	1
			555.6	Ar	IR	1
			561.8	N ₂	IR	1
			506.9	Ar	IR	1,3

References

¹G. V. Chertihin, W. D. Bare, and L. Andrews, *J. Phys. Chem. A* **101**, 5090 (1997).

²H. Wu and L.-S. Wang, *J. Chem. Phys.* **108**, 5310 (1998).

³M. Zhou and L. Andrews, *J. Phys. Chem. A* **102**, 8251 (1998).

(cyc-O₂Nb)O₂ \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OO stretch	1109.3	Ar	IR	1
		ONbO s-stretch	945.9	Ar	IR	1
			943.5			
		ONbO a-stretch	903.6	Ar	IR	1
			900.7			
			511.3	Ar	IR	1

Reference

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **102**, 8251 (1998).

(cyc-O₂Ta)O₂ \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OO stretch	1095.7	Ar	IR	1,2
		OTaO s-stretch	950.5	Ar	IR	1,2
			946.3			
		OTaO a-stretch	894.5	Ar	IR	1,2
			889.4			
			524.2	Ar	IR	1

References

¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **102**, 8251 (1998).

²M. Chen, X. Wang, L. Zhang, M. Yu, and Q. Qin, *Chem. Phys.* **242**, 81 (1999).

(cyc-O₂Mn)O₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OO stretch	1130	Ar	IR	1
		OMnO a-stretch	974.9	Ar	IR	1

Reference¹G. V. Chertihin and L. Andrews, *J. Phys. Chem. A* **101**, 8547 (1997).**(cyc-O₂Re)O₂** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OReO s-stretch	992.6	Ne	IR	1
			992.4	Ar	IR	1
		OReO a-stretch	969.8	Ne	IR	1
			964.3	Ar	IR	1
		OO stretch	878.6	Ne	IR	1
			882.4	Ar	IR	1

Reference¹M. Zhou, A. Citra, B. Liang, and L. Andrews, *J. Phys. Chem. A* **104**, 3457 (2000).**(cyc-O₂Fe)O₂** \tilde{A} $T_0=4360(160)$ gas PE²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			850(50)	gas	PE	2

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1095.4	Ar	IR	1
			1097.6	O ₂	IR	1
			920(50)	gas	PE	2
			968.9	Ar	IR	1
			970.2	O ₂	IR	1

References¹G. V. Chertihin, W. Saffel, J. T. Yustein, L. Andrews, M. Neurock, A. Ricca, and C. W. Bauschlicher, Jr., *J. Phys. Chem.* **100**, 5261 (1996).²H. Wu, S. R. Desai, and L.-S. Wang, *J. Am. Chem. Soc.* **118**, 5296 (1996); *J. Am. Chem. Soc.* **118**, 7434 (1996).**(cyc-O₂Ru)O₂** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		ORuO s-stretch	946.9	Ne	IR	1
			940.2	Ar	IR	1
		ORuO a-stretch	928.5	Ne	IR	1
			920.7	Ar	IR	1

Reference¹M. Zhou, A. Citra, B. Liang, and L. Andrews, *J. Phys. Chem. A* **104**, 3457 (2000).**RuO₄** \tilde{X} T_d

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
f_2	3	RuO ₄ stretch	922.2	Ne	IR	2
			916.6	Ar	IR	1,2

References¹D. W. Green, J. G. Kay, G. L. Zimmerman, and B. A. Balko, *J. Mol. Spectrosc.* **138**, 62 (1989).²M. Zhou, A. Citra, B. Liang, and L. Andrews, *J. Phys. Chem. A* **104**, 3457 (2000).**(cyc-O₂Os)O₂** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OOsO s-stretch	1004.8	Ar	IR	1
		OOsO a-stretch	970.0	Ar	IR	1
		OO stretch	898.1	Ar	IR	1

Reference¹M. Zhou, A. Citra, B. Liang, and L. Andrews, *J. Phys. Chem. A* **104**, 3457 (2000).**OOCO₂** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OCoO a-stretch	954.7	Ar	IR	1

Reference

¹G. V. Chertihin, A. Citra, L. Andrews, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **101**, 8793 (1997).

(cyc-O₂Co)O₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OC ₂ O a-stretch	897.3	Ar	IR	1
		OC ₂ O s-stretch	797.3	Ar	IR	1

Reference

¹G. V. Chertihin, A. Citra, L. Andrews, and C. W. Bauschlicher, Jr., *J. Phys. Chem. A* **101**, 8793 (1997).

O₂RhO₂ \tilde{X}

		D _{2d}	Structure: IR ¹			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OO stretch	1048.4	Ar	IR	1,2
			1045	Xe	IR	1
			1038	O ₂	IR	1

References

¹A. J. L. Hanlan and G. A. Ozin, *Inorg. Chem.* **16**, 2848 (1977).
²A. Citra and L. Andrews, *J. Phys. Chem. A* **103**, 4845 (1999).

O₂PdO₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OO stretch	1112.5	Ne	IR	3
			1110.8	Ar	IR	1,2
		PdO stretch	504w	Ar	IR	1

References

¹H. Huber, W. Klotzbücher, G. A. Ozin, and A. Vander Voet, *Can. J. Chem.* **51**, 2722 (1973).
²W. D. Bare, A. Citra, G. V. Chertihin, and L. Andrews, *J. Phys. Chem. A* **103**, 5456 (1999).
³X. Wang and L. Andrews, *J. Phys. Chem. A* **105**, 5812 (2001).

O₂PtO₂ \tilde{X}

		D _{2h}	Structure: IR ²			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OO stretch	1056.0	Ne	IR	3
			1051.3	Ar	IR	1,2

References

¹H. Huber, W. Klotzbücher, G. A. Ozin, and A. Vander Voet, *Can. J. Chem.* **51**, 2722 (1973).
²W. D. Bare, A. Citra, G. V. Chertihin, and L. Andrews, *J. Phys. Chem. A* **103**, 5456 (1999).
³X. Wang and L. Andrews, *J. Phys. Chem. A* **105**, 5812 (2001).

ReO₄⁻ \tilde{X}

		T _d				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
f ₂	3	ReO ₄ stretch	914.5	Ne	IR	1
			907.2	Ar	IR	1

Reference

¹M. Zhou, A. Citra, B. Liang, and L. Andrews, *J. Phys. Chem. A* **104**, 3457 (2000).

FeO₄⁻

Threshold for electron detachment from ground-state FeO₄⁻ = 26620(320) gas PE¹

Reference

¹H. Wu, S. R. Desai, and L.-S. Wang, *J. Am. Chem. Soc.* **118**, 5296 (1996); *J. Am. Chem. Soc.* **118**, 7434 (1996).

C₂O₃⁻ \tilde{X}

		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CO stretch	1793.7	Ar	IR	1
a''		CO ₂ a-stretch	1701.7	Ar	IR	1

Reference

¹M. Zhou, L. Zhang, M. Chen, and Q. Qin, *J. Chem. Phys.* **112**, 7089 (2000).

UO₂F₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		UO ₂ a-stretch	940.5	Ar	IR	1

Reference

¹P. F. Souter and L. Andrews, *J. Mol. Struct.* **412**, 161 (1997).

F₂CCO

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CCO a-stretch	2161.6m	Ar	IR	1
	2	CCO s-stretch	1426.8vs	Ar	IR	1
b ₂	7	CF ₂ a-stretch	1274.4s	Ar	IR	1

Reference

¹C. Kötting, W. Sander, M. Senzlober, and H. Bürger, *Chem. Eur. J.* **4**, 1611 (1998).

O₂CNO⁻

\tilde{X}^1A' C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	NO stretch	1495.8	Ar	IR	1
	2	CO ₂ s-stretch	1310.4	Ar	IR	1
a''		CO ₂ a-stretch	1713.6	Ar	IR	1

Reference

¹M. Zhou, L. Zhang, and Q. Qin, *J. Am. Chem. Soc.* **122**, 4483 (2000).

O₂N-NO

In nitrogen- and neon-matrix studies,^{4,8} excitation in the region of a weak absorption between 700 and 900 nm, with maximum near 720 nm, leads to isomerization forming O=N-O-N=O.

\tilde{X} C _s Structure: MW ³						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	N=O stretch	1829.54 1830.2s 1831.5 1840m 1867vs	gas Ne Ar N ₂ O ₂ NO	IR IR IR IR IR,Ra	1,5,7,10 8 11,12 4,11 2 6
	2	NO ₂ a-stretch	1652 1643.3s 1630.1 1630vs 1596s 1593	gas Ne Ar N ₂ O ₂ NO	IR IR IR IR IR IR,Ra	5 8 11,12 4,11 2 6
	3	NO ₂ s-stretch	1304.25 1302.5vs 1296.2 1302s 1303 1298	gas Ne Ar N ₂ O ₂ NO	IR IR IR IR IR IR,Ra	1,5,9 8 11,12 4,11 2 6
	4	NO ₂ deform.	773 773.1wm 769.2 776wm 788 787	gas Ne Ar N ₂ O ₂ NO	IR IR IR IR IR IR,Ra	1,5 8 11,12 4,11 2 6
	5	NO wag	627	NO	Ra	6
	6	N-N stretch	241 266	gas NO	IR Ra	5 6
	7	NO ₂ wag	205	NO	Ra	6
a''	8	NNO ₂ OPLA	414 420w 405 70	gas N ₂ NO NO	IR IR Ra Ra	5 4 6 6
	9	Torsion	63	gas	IR	5

A₀=0.415; B₀=0.141; C₀=0.105 MW³IR^{9,10}

References

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- ²W. G. Fateley, H. A. Bent, and B. Crawford, Jr., *J. Chem. Phys.* **31**, 204 (1959).
- ³A. H. Brittain, A. P. Cox, and R. L. Kuczkowski, *Trans. Faraday Soc.* **65**, 1963 (1969).
- ⁴E. L. Varetti and G. C. Pimentel, *J. Chem. Phys.* **55**, 3813 (1971).
- ⁵C. H. Bibart and G. E. Ewing, *J. Chem. Phys.* **61**, 1293 (1974).
- ⁶E. M. Nour, L.-H. Chen, and J. Laane, *J. Phys. Chem.* **87**, 1113 (1983).
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- ⁹A. M. Andrews, J. L. Domenech, G. T. Fraser, and W. J. Lafferty, *J. Mol. Spectrosc.* **163**, 428 (1994).
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- ¹¹T. G. Koch and J. R. Sodeau, *J. Chem. Soc., Faraday Trans.* **92**, 2347 (1996).
- ¹²C.-I Lee, Y.-P. Lee, X. Wang, and Q.-Z. Qin, *J. Chem. Phys.* **109**, 10446 (1998).

O=N-O-N=O

In nitrogen-matrix studies,¹ absorption maxima were observed at 398, 381, and 363 nm. Excitation in the 370–480 nm spectral region resulted in photoisomerization into O₂N-NO.^{1,3}

\bar{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	N=O s-stretch	1740	NO	Ra	2
			973.6m	Ne	IR	3
	2	NON s-stretch	971.0	Ar	IR	4,5
			969w	N ₂	IR	1
			973m	NO	IR,Ra	2
3	NON bend	366s	N ₂	IR	1	
		395	NO	Ra	2	
		210	NO	Ra	2	
a ₂	5	Torsion	210	NO	Ra	2
b ₁	6	Torsion	140	NO	Ra	2
b ₂	7	N=O a-stretch	1697.2vs	Ne	IR	3
			1688.6	Ar	IR	4,5
			1690s	N ₂	IR	1
	8	O=NO a-bend	1687vs	NO	IR	2
			1661w	N ₂	IR	1
			704.6	Ar	IR	5
			704vw	N ₂	IR	1
	9	NON a-stretch	705vw	NO	IR	2
			380	Ar	IR	5
			387m	N ₂	IR	1,5

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t,c-ONONO

\bar{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	<i>t</i> -N=O stretch	1704.5	Ar	IR	1,2
			1665.7	Ar	IR	1,2
	3	ONO a-deform.	877.8	Ar	IR	1,2
			243.0	Ar	IR	1,2

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OPOPO

\bar{X} C ₂						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1253.4	Ar	IR	1,2
			1247.8	Ar	IR	1,2
			859.3	Ar	IR	1,2

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CO₄⁻

The gas-phase photodestruction cross section of CO₄⁻ (O₂C·O₂⁻) is small over the 840–350 nm range.^{1–4} In a neon matrix,⁵ the onset of photodestruction occurs near 260 nm, with evidence for the detachment of O or O⁻.

\bar{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CO ₂ a-stretch	1895.2vs	Ne	IR	5
			1891.5	Ar	IR	5,6
2	CO ₂ s-stretch	1256.5s	Ne	IR	5	
		1257.0	Ar	IR	5,6	
		697.1m	Ne	IR	5	
4	CO ₂ scissors	691.2	Ar	IR	5,6	

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c-FC(O)OO

\bar{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	C=O stretch	1913.2vs	Ne	IR	1
			1904.5vs	Ar	IR	1
			1202.1s	Ne	IR	1
			1196.7s	Ar	IR	1
			1078.1wm	Ne	IR	1
			1071.9wm	Ar	IR	1
			867.1wm	Ne	IR	1
			863.6wm	Ar	IR	1
			661.0w	Ne	IR	1
			659.2w	Ar	IR	1
			552.7w	Ne	IR	1
			550.3vw	Ar	IR	1
			346.2vw	Ne	IR	1
a''	8	OPLA	708.7w	Ne	IR	1
			704.2w	Ar	IR	1

Reference

- ¹S. Sander, H. Pernice, and H. Willner, *Chem. Eur. J.* **6**, 3645 (2000).

t-FC(O)OO

A gas-phase absorption between 200 and 285 nm, with maximum at 43100 (232 nm), has been assigned^{1,2} to FC(O)OO. This absorption maximum has been observed³ at 42550 (235 nm) in a neon matrix.

\tilde{X}	C_s					
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	C=O stretch	1925.9s	Ne	IR	3
			1919.2	Ar	IR	3
	2	CF stretch	1175.0vs	Ne	IR	3
			1168.3	Ar	IR	3
	3	OO stretch	1090.1m	Ne	IR	3
			1084.8	Ar	IR	3
	4	C-O stretch	900.8m	Ne	IR	3
895.7			Ar	IR	3	
5	OCO deform.	680.2w	Ne	IR	3	
		678.4	Ar	IR	3	
6	FCO deform.	520.9w	Ne	IR	3	
		518.5	Ar	IR	3	
a''	7	COO deform.	335.7vw	Ne	IR	3
			712.0wm	Ne	IR	3
8	OPLA	707.4	Ar	IR	3	

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CF₃N

3E	C_{3v}					
$T_0=28250$	Ar	AB ¹	342–354 nm			
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1			1160T	Ar	AB	1
			500T	Ar	AB	1

Reference

- ¹N. P. Gritsan, I. Likhovtorik, Z. Zhu, and M. S. Platz, *J. Phys. Chem. A* **105**, 3039 (2001).

ClONO₂

The onset of continuous absorption by gas-phase ClONO₂ occurs near 25000 (400 nm).³ Except for two possible shallow maxima, the absorption continues to rise out to the 185 nm cutoff of the observations.

\tilde{X}	C_s					
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	NO ₂ a-stretch	1736.9vs	gas	IR,Ra	1,2,4,13 21
			1727.6s	Ar	IR	9,11,16
			1734.0s	N ₂	IR	11,16
			1730.0vs	O ₂	IR	7,16
			1292.84vs	gas	IR,Ra	1,2,4,13 18,21,23
			1285.8vs	Ar	IR	9,11,16
	2	NO ₂ s-stretch	1291.4vs	N ₂	IR	11,16
			1289.7s	O ₂	IR	7,16
			809.4s	gas	IR,Ra	1,2,4,13 21
			801.6m	Ar	IR	9,11,16
			809.7wm	N ₂	IR	11,16
			808.1m	O ₂	IR	7,16
3	ClO stretch	780.22ms	gas	IR,Ra	1,2,4 DL	
		775.6m	Ar	IR	9,11,16	
		778.6m	N ₂	IR	11,16	
		776.3m	O ₂	IR	7,16	
		563.1s	gas	IR,Ra	1,2,21	
		560.9ms	Ar	IR	9,11,16	
4	Mixed	563.2wm	N ₂	IR	11,16	
		559.0m	O ₂	IR	7,16	
		434.0m	gas	IR,Ra	1,2,21	
		426	Ar	IR	9,11	
		436.5w	N ₂	IR	11	
		435.0	O ₂	IR	7	
5	Mixed	273.3vww	gas	IR,Ra	1,2,21	
		711.0w	gas	IR	1,21	
		710.8w	Ar	IR	9,11,16	
		708.4w	N ₂	IR	11,16	
		704.6wm	O ₂	IR	7,16	
		120.16	gas	IR,Ra	1,2,6,8 21	
a''	7	ClO rock	711.0w	gas	IR	1,21
			710.8w	Ar	IR	9,11,16
8	NO ₂ wag	708.4w	N ₂	IR	11,16	
		704.6wm	O ₂	IR	7,16	
9	Torsion	120.16	gas	IR,Ra	1,2,6,8 21	

$A_0=0.404$; $B_0=0.093$; $C_0=0.075$ MW^{5,10,22}DL²³
 Torsional barrier=1900(100) gas IR^{6,8}

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SO₄

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		S=O stretch	1443s	Ar	IR	1,2
		S=O stretch	1273s	Ar	IR	1,2
		O-O stretch	925wm	Ar	IR	1
			777wm	Ar	IR	1
			611m	Ar	IR	1
			498sh	Ar	IR	1
			490sh	Ar	IR	1

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CF₃O

 \tilde{B}
 $T_0 \leq 34000$ Ne AB⁸

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁			495(10)	Ne	AB	8

\tilde{A}^2A_1 C_{3v} Structure: LF⁶
 $T_0 = 28531.26(2)$ gas LF³⁻⁶ $\tilde{A}-\tilde{X}$ 330–400 nm
 28517(3) Ne AB⁸ $\tilde{A}-\tilde{X}$ 310–351 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	2		819(5)	Ne	AB	8
	3		635.0	gas	LF	5,6
			632(5)	Ne	AB	8
<i>e</i>	4	CF ₃ a-stretch	1254(5)	Ne	AB	8
	5	CF ₃ deform.	626.1	gas	LF	6
			632(5)	Ne	AB	8
	6	FCO deform.	426.3	gas	LF	3–6
			434(5)	Ne	AB	8

$\tau_0 = 30.1$ ns gas LF³
 $A_0 = 0.191$; $B_0 = 0.195$ LF^{5,6}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CO stretch	1215 (ω) 1273ms 1260ms	gas Ne Ar	LF IR IR	3,4,7 8 8
	2	CF ₃ s-stretch	977 (ω) 891w 894w	gas Ne Ar	LF IR IR	3,4,7 8 8
	3	CF ₃ umbrella	527 (ω) 621w 622w	gas Ne Ar	LF IR IR	7 8 8
<i>e</i>	4	CF ₃ a-stretch	1215s (a'') 1207vs (a') 1207s ^b (a'') 1199vs (a')	Ne Ne Ar Ar	IR IR IR IR	8 8 8 8
	5	CF ₃ deform.	600 (ω) 666w (a') 663w (a')	gas Ne Ar	LF IR IR	7 8 8
	6	FCO deform.	465 (ω)	gas	LF	3,4,7

$A = -41(5)$ gas LF⁵
 $B_0 = 0.198$ LF⁵

^aSignificant Jahn–Teller interaction.⁷ Reference 8 has proposed that the unusually large shifts in the vibrational fundamentals of CF₃O isolated in neon and argon matrices from their gas-phase positions may be a consequence of an interaction between CF₃O and Ne or Ar which increases the Jahn–Teller stabilization energy sufficiently to “lock in” the distorted structure.

^bReferences 1 and 2 report 1221.6.

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CF₃S

\tilde{A}^2A_1 C_{3v} Structure: LF³
 $T_0 = 26393.16(2)$ gas LF¹⁻³FD^{1,2} $\tilde{A}-\tilde{X}$ 339–380 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CF stretch	990.5	gas	LF,FD	2,3
	2	CF ₃ umbrella	737.7	gas	LF,FD	2,3
	3	CS stretch	310.2	gas	LF,FD	1–3
<i>e</i>	5	CF ₃ deformation	548	gas	LF,FD	2,3
	6	CF ₃ rock	281.4	gas	LF,FD	2,3

$\tau_0 = 2.74(14)$ μ s gas LF^{1,2}
 $A_0 = 0.189$; $B_0 = 0.103$ LF³

\tilde{X}^2E^a		C_{3v}	Structure: LF^3			
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1	1	CF_3 s-stretch	1142(ω)	gas	LF	4
	2	CS stretch	765(ω)	gas	LF	4
	3	CF_3 umbrella	449(ω)	gas	LF	4
e	5	CF_3 deform.	536(ω)	gas	LF	4
	6	FCS deform.	320(ω)	gas	LF	4

$A_0 = -160$ gas LF^2

$A_0 = 0.189$; $B_0 = 0.111$ LF^3

^aSignificant Jahn–Teller interaction.⁴

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CF_4^+

\tilde{D}^2A_1 T_d Structure: PE,EF^7EM^{14}
 $T_0^a = 78830(160)$ gas $PE^{2,3,5}TPEPICO^{15}$
 gas EF^6 $\tilde{D}-\tilde{C}$ 350–420 nm
 Broad, unstructured emission maxima at 189 and 160 nm (52900 and 62500) which appear on He^+ or electron impact on CF_4 have been interpreted as arising from the $\tilde{D}-\tilde{B}$ and $\tilde{D}-\tilde{A}$ transitions of CF_4^+ , respectively.⁶

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1	1	CF stretch	800(1)	gas	PE,EF	3,5,6

$\tau = 2.1(2)$ ns gas EF^1

$B_0 = 0.180(3)^b$ EF^7

\tilde{C}^2F_2 T_d Structure: PE,EF^7EM^{14}
 $T_0^a = 51230(160)$ gas $PE^{2,3,5}TPEPICO^{15}$
 $T_0(\tilde{D}-\tilde{C}) = 27662.2$ gas EF^6EM^{14} $\tilde{D}-\tilde{C}$ 350–420 nm
 Broad, unstructured emission maxima at 290 and 230 nm (34500 and 43500) which result from He^+ or electron impact⁶ on CF_4 and from exposure of CF_4 to synchrotron radiation^{10,16} of wavelength shorter than 57.4 nm have been interpreted as arising from the $\tilde{C}-\tilde{A}$ and $\tilde{C}-\tilde{X}$ transitions of CF_4^+ , respectively.

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1	1	CF stretch	727.4	gas	PE,EF EM	2,3,5,6 14

$A = +16(1)$ EF^{6-8}

$\tau = 9.0(9)$ ns gas $EF^1EM^{9,11}TPEFCO^{13,15}$

$B_0 = 0.169$ EM^{14}

\tilde{B}^2E

$T_0^a = 23800(1000)$ gas PE^{2-5}

Dissociates into $CF_3^+ + F$, probably by internal conversion to the \tilde{A}^2F_2 state. TPEPICO¹²

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1	1	CF stretch	810(80)	gas	PE	3–5
e		Deformation	500(100)	gas	PE	3–5

\tilde{A}^2F_2

$T_0^a = 14100(1000)$ gas PE^{2-5}

Direct dissociation into $CF_3^+ + F$. TPEPICO¹²

\tilde{X}^2F_1

Direct dissociation into $CF_3^+ + F$. TPEPICO¹²

^aMeasured with respect to onset of first photoelectron band, estimated by Ref. 3 at 15.35 eV.

^bFrom computer simulation of emission bands.

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CF_3Cl^+

\tilde{F}^2E C_{3v}

$T_0^a \leq 66130(400)$ gas $PE^{2,4}$

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1	2	CF_3 umbrella	589(80)	gas	PE	4
	3	CCl stretch	420(80)	gas	PE	4

\tilde{E}^2A_1 C_{3v}
 $T_0^a = 60420(400)$ gas PE^{2,4}

Broad emission bands in the 190–600 nm spectral region^{8,9,11} which appear on excitation of CF₃Cl by synchrotron radiation with energy greater than 19 eV have been attributed^{9,11} to transitions arising from the \tilde{E} state of CF₃Cl⁺. Discrete structure between 190 and 270 nm is contributed by the $B-X$ and $A-X$ transitions of CF.¹¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CF ₃ umbrella	637(80)	gas	PE	4

$\tau = 9.4(1.8)$ ns gas TPEFCO¹⁰

\tilde{D}^2E C_{3v}
 $T_0^a = 39720(400)$ gas PE^{1,2,4}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CF ₃ umbrella	726(80)	gas	PE	4
	3	CCl stretch	387(80)	gas	PE	4
e	4	CF ₃ stretch	1130(80)	gas	PE	4

\tilde{C}^2E C_{3v}
 $T^{ab} = 34610(400)$ gas PE^{1,2,4}

Broad, unstructured absorption with onset near 400 nm (25000) and maximum at 295 nm (33900) which appears on argon-resonance photolysis of CF₃Cl isolated in solid argon and which has a photodecomposition threshold at a wavelength longer than 340 nm has been assigned⁷ to the $\tilde{C}-\tilde{X}$ transition of CF₃Cl⁺.

\tilde{B}^2A_2 C_{3v}
 $T^{ab} = 26950(400)$ gas PE^{1,2,4}

\tilde{A}^2A_1 C_{3v}
 $T^{ab} = 22110(400)$ gas PE^{1,2,4}

\tilde{X}^2E C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CF stretch	1299vs	Ar	IR	5,6
		CCl stretch	734wm	Ar	IR	5,6
		Deformation	460ms	Ar	IR	5,6
			451m			
		Deformation	416m	Ar	IR	5,6

^aThe first ionization potential of CF₃Cl is taken as 12.42(4) eV, the mean of the values reported in the photoionization studies of Refs. 2 and 3.

^bFrom vertical ionization potential.

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SiF₄⁺

\tilde{D}^2A_1 T_d Structure: PE,EF⁹
 $T^{ab} = 50800(200)$ gas PE²TPE^{13,15}EM¹⁴
 EF⁷EM^{8,10}

$\tilde{D}-\tilde{C}$ 530–590 nm
 $\tilde{D}-\tilde{C}$ band origin measured at 18146.8 in emission studies on a cooled beam.⁷

Continuous emission between 570 and 730 nm, with a maximum near 610 nm (16400), has been assigned to the $\tilde{D}-\tilde{C}$ transition.^{8,10,12}

Broad, unstructured emission maxima at 370 and 304 nm (27000 and 32900) which appear on ion, electron, or photon impact on SiF₄ have been interpreted as arising from the $\tilde{D}-\tilde{B}$ and $\tilde{D}-\tilde{A}$ transitions of SiF₄⁺, respectively.^{6,8}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	SiF stretch	743.4(5)	gas	EF TPE	7,13,15
e	2	Deformation	258(40)	gas	TPE	13
f_2	3	SiF stretch	1178(40)	gas	TPE	13
	4	Deformation	379(12)	gas	TPE	15

$\tau = 9.30(4)$ ns gas EF⁷EM^{12,14}TPEFCO¹³
 $B_0 = 0.136(1)^c$ EF⁹

\tilde{C}^2F_2 T_d^d Structure: PE,EF⁹
 $T_0^a = 33130(100)$ gas PE^{2,3,5}TPE^{13,15}
 EF⁷EM^{8,10}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	SiF stretch	706.6(5)	gas	TPE,EF	3,5,7,13,15
e	2	Deformation	159.0(5)	gas	EF	7
f_2	4	Deformation	431.0(5)	gas	TPE,EF	3,5,7,13,15

$A = +6.9(2)$ EF^{7,9,11}
 $B_0 = 0.132^c$ PE,EF⁹

\tilde{B}^2E
 $T_0^a = 22580(100)$ gas PE^{2,3,5}TPE¹⁵

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	SiF stretch	685(50)	gas	PE	5

\tilde{A}^2F_2 T_d
 $T_0^a = 17000(1000)$ gas PE^{2,3,5}TPE¹⁵

\tilde{X}^2F_1 T_d

^aMeasured with respect to a first ionization potential of 15.19 eV, estimated⁴ by extrapolation of the photoionization efficiency curve for SiF₄.

^bFrom vertical ionization potential.

^cFrom Franck–Condon analysis of the photoelectron spectrum and computer simulation of the $\tilde{D}-\tilde{C}$ emission.

^dDynamic Jahn–Teller distortion, probably to C_{3v}.^{7,9}

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SiCl₄⁺

\tilde{D}^2A_1 T_d
 $T_0 = 48900(400)$ gas PE^{1,2}TPE^{8,11}
 Fragments to SiCl₂⁺ T-PEPICO⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	SiCl stretch	379(16)	gas	TPE	8
			339(12)	gas	TPE	11

 \tilde{C}^2F_2

$T_0 = 26620(160)$ gas PE^{1,2}TPE^{8,11}
 A broad, unstructured emission with maxima at 410 and 570 nm (24400 and 17500) has been assigned^{5,6,9} to the $\tilde{C}-\tilde{X}$ and $\tilde{C}-\tilde{A}$ transitions, respectively, of gas-phase SiCl₄⁺. Using pulsed synchrotron excitation^{6,10} of SiCl₄, as well as TPEFCO measurements,⁸ a radiative lifetime of 39(2) ns has been determined for the \tilde{C} state of SiCl₄⁺.

A broad absorption with maximum at 475 nm (21000) which appears on argon-resonance photolysis of SiCl₄ isolated in an argon matrix has been assigned⁴ to the $\tilde{C}-\tilde{X}$ transition of SiCl₄⁺. The energy difference is attributed to structural relaxation in the argon matrix. The absorption can be destroyed by exposure of the sample to 290–1000 nm radiation.

 \tilde{B}^2E

$T^a = 13880(400)$ gas PE^{1,2}TPE¹¹

 \tilde{A}^2F_2

$T_0 = 7750(160)$ gas PE^{1,2}TPE¹¹
 Splitting between $G_{3/2}$ and $E_{5/2}$ spin-orbit components = 920(12) gas TPE¹¹

 \tilde{X}^2F_1

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		SiCl ₂ a-stretch	717T	Ar	IR	4

^aFrom vertical ionization potential.

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- ¹¹L. Cooper, L. G. Shpinkova, D. M. P. Holland, and D. A. Shaw, *Chem. Phys.* **270**, 363 (2001).

SiBr₄⁺

\tilde{D}^2A_1 T_d
 $T^a = 52550(20)$ gas PE¹TPE³

\tilde{C}^2F_2 T_d
 $T^a = 25450(20)$ gas PE¹TPE³

A broad, unstructured emission with maxima at 435 and 550 nm (23000 and 18200) has been assigned² to the $\tilde{C}-\tilde{X}$ and $\tilde{C}-\tilde{A}$ transitions, respectively, of gas-phase SiBr₄⁺, excited by electron impact. Using pulsed synchrotron excitation of SiBr₄, a radiative lifetime of 47.6(3) ns has been determined² for the \tilde{C} state of SiBr₄⁺.

Splitting between $G_{3/2}$ and $E_{5/2}$ spin-orbit levels ~1570.³

\tilde{B}^2E T_d
 $T^a = 11560(80)$ gas PE¹TPE³

\tilde{A}^2F_2 T_d
 $T^a = 5740(20)$ gas PE¹TPE³

Splitting between $G_{3/2}$ and $E_{5/2}$ spin-orbit levels ~3950.³

\tilde{X}^2F_1 T_d

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
f ₂	2	SiBr stretch	444T	gas	TPE	3

Splitting between $E_{1/2}$ and $G_{3/2}$ spin-orbit levels ~2400, with superposed structure arising from Jahn–Teller interaction.³

^aFrom vertical ionization potentials.

References

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GeF₄⁺

\tilde{D}^2A_1 T_d
 $T_0^{\text{ab}} = 45300(1000)$ gas PE^{2,4}
 EF⁶

$\tilde{D}-\tilde{C}$ 390–420 nm
 Broad, unstructured emission maxima at 290 and 255 nm (34500 and 39200) which appear on ion impact on GeF₄ have been interpreted as arising from the $\tilde{D}-\tilde{B}$ and $\tilde{D}-\tilde{A}$ transitions of GeF₄⁺, respectively.⁵ On 21.4 eV synchrotron excitation of GeF₄,⁸ not only these two bands but also emission maxima at 400 and 230 nm (25000 and 43500), arising from the $\tilde{D}-\tilde{C}$ and $\tilde{D}-\tilde{X}$ transitions of GeF₄⁺, respectively, appear.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	GeF stretch	644.3	gas	EF	6

$\tau = 5.02(1)$ ns gas EM⁸

$\tilde{C}^2F_2^c$
 $T_0^a = 20330(240)$ gas PE¹⁻⁴
 EF⁶

$\tilde{D}-\tilde{C}$ 390–420 nm

$\tilde{D}-\tilde{C}$ band origin measured at 25064.0 in emission studies on a cooled beam.⁶

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	GeF stretch	620.8	gas	EF	6
e	2	Deformation	82.8	gas	EF	6
f_2	4	Deformation	288.3	gas	EF	6

$A = -18.6^d$ EF⁶

\tilde{B}^2E
 $T_0^{\text{ab}} = 11210(320)$ gas PE¹⁻⁴

\tilde{A}^2F_2
 $T_0^{\text{ab}} = 7020(320)$ gas PE¹⁻⁴

\tilde{X}^2F_1

^aFirst ionization potential taken to be 15.69(2) eV, as in Ref. 1.

^bFrom vertical ionization potential.

^cDistorted by Jahn–Teller interaction.

^dTentative value.

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GeCl₄⁺

\tilde{D}^2A_1 T_d
 $T_0 = 51070(400)$ gas PE^{1,2}TPE⁵

\tilde{C}^2F_2 T_d
 $T_0 = 21620(240)$ gas PE^{1,2}TPE⁵

A broad, unstructured emission with maxima at 495 and 615 nm (20200 and 16300) has been assigned^{3,4,6} to the $\tilde{C}-\tilde{X}$ and $\tilde{C}-\tilde{A}$ transitions, respectively, of gas-phase GeCl₄⁺. Using pulsed synchrotron excitation of GeCl₄, a radiative lifetime of 65.9(1.4) ns has been determined^{4,7} for the \tilde{C} state of GeCl₄⁺.

\tilde{B}^2E T_d
 $T_0 = 8880(1100)$ gas PE^{1,2}TPE⁵

\tilde{A}^2F_2 T_d
 $T_0 = 5650(1100)$ gas PE^{1,2}TPE⁵

\tilde{X}^2F_1 T_d

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- ¹J. C. Green, M. L. H. Green, P. J. Joachim, A. F. Orchard, and D. W. Turner, *Philos. Trans. R. Soc. London, Ser. A* **268**, 111 (1970).
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SO₄⁻

Threshold for electron detachment from ground-state SO₄⁻ = 41150(800) gas PE¹

Reference

- ¹X.-B. Wang, J. B. Nicholas, and L.-S. Wang, *J. Phys. Chem. A* **104**, 504 (2000).

Cl₂CCI-Cl

A strong, broad absorption with maximum near 415 nm (24000) which appears on argon-resonance photolysis of CCl₄ isolated in an argon matrix was originally assigned^{2,3} to the $\tilde{C}-\tilde{X}$ transition of CCl₄⁺. Subsequent argon-matrix studies⁴ showed that this absorption can also be produced by 220 or 193 nm irradiation of the sample. Ab initio calculations are consistent with the alternate assignment of this band and of the associated infrared absorption pattern to Cl₂CCI-Cl. The product absorptions are destroyed, and the absorptions of ordinary CCl₄ are intensified, by exposure of the sample to radiation of wavelength longer than 360 nm.

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	CCl stretch	1018.2	Ne	IR	5
			1019.7m	Ar	IR	1,4
	2	CCl stretch	503.9	Ne	IR	5
			501.9m	Ar	IR	1,4
3	Deformation	373.9wm	Ar	IR	1,4	
		291.2wm	Ar	IR	1,4	
4	Deformation	246.4wm	Ar	IR	4	
		931.6	Ne	IR	5	
a''	7	CCl stretch	929.1vs	Ar	IR	1,4

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 PF_4^+

\tilde{X}		T_d				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
f_2	3	PF stretch	1174.4	Ne	IR	1

Reference

- ¹C. L. Lugez, K. K. Irikura, and M. E. Jacox, *J. Chem. Phys.* **108**, 8381 (1998).

 ClO_4^-

Threshold for electron detachment from ground-state $ClO_4^- = 42360(800)$ gas PE¹

Reference

- ¹X.-B. Wang and L.-S. Wang, *J. Chem. Phys.* **113**, 10928 (2000).

 $Cl_2CCl-Cl^-$

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	CCl stretch	899.3	Ne	IR	2
a''	7	CCl stretch	848.3	Ne	IR	2
			851	Ar	IR	1

References

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²C. L. Lugez, M. E. Jacox, and R. D. Johnson III, *J. Chem. Phys.* **109**, 7147 (1998).

 CCl_4^-

\tilde{X}		C_{3v}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
e	4	CCl_3 stretch	757.9	Ne	IR	1

Reference

- ¹C. L. Lugez, M. E. Jacox, and R. D. Johnson III, *J. Chem. Phys.* **109**, 7147 (1998).

 F_3PO^-

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	1	PO stretch	1200.0mT	Ne	IR	1
			768.0m	Ne	IR	1
a''	7	PF stretch	614.7ms	Ne	IR	1

Reference

- ¹C. L. Lugez, K. K. Irikura, and M. E. Jacox, *J. Chem. Phys.* **108**, 8381 (1998).

 PF_4

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1	1	PF stretch	910.7mT	Ne	IR	1
b_1	6	PF stretch	948.2mT	Ne	IR	1
b_2	8	PF stretch	849.1sT	Ne	IR	1

Reference

- ¹C. L. Lugez, K. K. Irikura, and M. E. Jacox, *J. Chem. Phys.* **108**, 8381 (1998).

 SF_4^+

$T^a = 51600(800)$ gas PE¹⁻³

$T^a = 46800(400)$ gas PE¹⁻³

$T^a = 43000(400)$ gas PE¹⁻³

$\tilde{C}, \tilde{D}^2B_2, ^2A_1$ C_{2v}
 $T^a = 31460(400)$ gas PE¹⁻³

$\tilde{A}, \tilde{B}^2A_2, ^2B_1$ C_{2v}
 $T^a = 25580(400)$ gas PE¹⁻³

\tilde{X}^2A_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b_1	6	SF stretch	975.4	Ne	IR	4
b_1	8	SF stretch	698.3	Ne	IR	4

^aFrom vertical ionization potential. The adiabatic first ionization potential of SF₄ is estimated to be 11.90 eV.³

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- 4 C. L. Lugez, M. E. Jacox, R. A. King, and H. F. Schaefer III, *J. Chem. Phys.* **108**, 9639 (1998).

PF₄⁻

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	PF stretch	775.6wT	Ne	IR	1
b_1	6	PF stretch	517.0m	Ne	IR	1
b_2	8	PF stretch	730.0mT	Ne	IR	1

Reference

- 1 C. L. Lugez, K. K. Irikura, and M. E. Jacox, *J. Chem. Phys.* **108**, 8381 (1998).

SF₄⁻

\tilde{X} C_{4v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e	6		540.7	Ne	IR	1

Reference

- 1 C. L. Lugez, M. E. Jacox, R. A. King, and H. F. Schaefer III, *J. Chem. Phys.* **108**, 9639 (1998).

8.12. Six-Atomic Molecules

CH₅⁺

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		HH stretch	2950T	gas	LD	1

Reference

- 1 E. T. White, J. Tang, and T. Oka, *Science* **284**, 135 (1999).

CH₃BeH

\tilde{X} C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	BeH stretch	2062.3 2059.4	Ar	IR	1
	3	CH ₃ s-deform.	1205.4	Ar	IR	1
e	7	CH ₃ rock	704.1	Ar	IR	1
	8	BeH deform.	443.0 435.4	Ar	IR	1

CD₃BeD

\tilde{X} C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	BeD stretch	1570.5 1568.0	Ar	IR	1
	3	CD ₃ s-deform.	979.3	Ar	IR	1
	4	BeC stretch	748.0	Ar	IR	1
e	7	CD ₃ rock	580.7 579.8	Ar	IR	1

Reference

- 1 T. M. Greene, D. V. Lanzisera, L. Andrews, and A. J. Downs, *J. Am. Chem. Soc.* **120**, 6097 (1998).

CH₃MgH

\tilde{X} C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	MgH stretch	1560.3 1552.2	Ar	IR	1,2
	3	CH ₃ s-deform.	1127.7	Ar	IR	1
e	7	CH ₃ rock	547.9 545.4 542.8	Ar	IR	1

CD₃MgD

\tilde{X}		C _{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	MgD stretch	1136.8	Ar	IR	1,2
	3	CD ₃ s-deform.	877.1	Ar	IR	1
	4	CMg stretch	506.9	Ar	IR	1
e	7	CD ₃ rock	423.8	Ar	IR	1
			422.9			
			421.1			

References

¹T. M. Greene, D. V. Lanzisera, L. Andrews, and A. J. Downs, J. Am. Chem. Soc. **120**, 6097 (1998).

²W. D. Bare and L. Andrews, J. Am. Chem. Soc. **120**, 7293 (1998).

CH₃HgH

\tilde{X}		C _{3v}					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a ₁	1	CH ₃ stretch	2921.2w	Ar	IR	1,2	
		HgH stretch	1954.9vs	Ar	IR	1,2	
	3	CH ₃ deform.	1191.7vw	N ₂	IR	3	
		HgC stretch	534.0w	Ar	IR	1,2	
e	5		530.3	N ₂	IR	3	
		CH ₃ stretch	2990.6mT	Ar	IR	1	
		CH ₃ deform.	1425.0w	Ar	IR	1,2	
		CH ₃ rock	779.8	Ar	IR	1,2	
	8			777.9			
				778.4	N ₂	IR	3
				777.4			
				774.3			
		526.5	Ar	IR	1,2		
		527.2	N ₂	IR	3		
		525.2					

CD₃HgD

\tilde{X}		C _{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CD ₃ stretch	2127.7w	Ar	IR	1,2
		HgD stretch	1400.7vs	Ar	IR	1,2
	3	CD ₃ deform.	1024.5vwT	N ₂	IR	3
e	4	HgC stretch	487.0	Ar	IR	2
			485.6	N ₂	IR	3
	5	CD ₃ stretch	2233.9m	Ar	IR	1
		CD ₃ deform.	1041.1wT	Ar	IR	1,2
	7	CD ₃ rock	593.5m	Ar	IR	1,2
			592.1			
			592.8	N ₂	IR	3
			589.3			
8	CHgD deform.	374.5	Ar	IR	2	

References

¹N. Legay-Sommaire and F. Legay, Chem. Phys. Lett. **217**, 97 (1994).

²T. M. Greene, L. Andrews, and A. J. Downs, J. Am. Chem. Soc. **117**, 8180 (1995).

³N. Legay-Sommaire and F. Legay, Chem. Phys. **211**, 367 (1996).

SiH₃HgH

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		HgH stretch	1888.5	Ar	IR	1
			1865.4	Kr	IR	1
			1873.7	N ₂	IR	1
			867.8	Ar	IR	1
			864.4	Kr	IR	1
			867.8	N ₂	IR	1

SiD₃HgD

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		HgD stretch	1354.7	Ar	IR	1
			1338.2	Kr	IR	1
			1344.4	N ₂	IR	1
			645.9	Ar	IR	1
			643.2	Kr	IR	1
			646.5	N ₂	IR	1

Reference

¹N. Legay-Sommaire and F. Legay, J. Phys. Chem. A **102**, 8759 (1998).

CH₃GaH

In an argon matrix, an absorption with maximum at 45900 (218 nm) behaves appropriately for assignment to CH₃GaH.¹

In an argon matrix, a weak, broad (100 nm FWHM) absorption at 16700 (600 nm) has also been attributed¹ to CH₃GaH. Irradiation of the sample in the region of this absorption leads to photodestruction of CH₃GaH.

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH stretch	2970	Ar	IR	1
		CH stretch	2914	Ar	IR	1
		GaH stretch	1719.7	Ar	IR	1–4
			1712.5	Kr	IR	2
		CH ₃ deform.	1432	Ar	IR	1
		CH ₃ deform.	1176.4	Ar	IR	1,2
			1153.7	Kr	IR	2
		CH ₃ rock	753.1	Ar	IR	1–4
			747.8	Kr	IR	2
		CH ₃ rock	642	Ar	IR	1
		GaC stretch	528.6	Ar	IR	1–4
			520.0	Kr	IR	2
		Deformation	475.5	Ar	IR	3,4

CD₃GaD

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CD stretch	2223.9	Ar	IR	1,4
		CD stretch	2137	Ar	IR	1
		GaD stretch	1243.6	Ar	IR	1,2,4
			1239.0	Kr	IR	2
		CD ₃ deform.	1026	Ar	IR	1
		CD ₃ deform.	747.0T ^a	Kr	IR	2
		CD ₃ rock	578.2	Ar	IR	1,2,4
		GaC stretch	525.0T ^b	Ar	IR	2

^aRef. 1 gives 899 for an argon matrix.

^bRef. 1 gives 484.

References

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³H.-J. Himmel, A. J. Downs, T. M. Greene, and L. Andrews, *Chem. Commun.* 2243 (1999).
⁴H.-J. Himmel, A. J. Downs, T. M. Greene, and L. Andrews, *Organomet.* **19**, 1060 (2000).

CH₃InH

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	3	InH stretch	1545.9	Ar	IR	1,2
	6	CH ₃ rock	697.3	Ar	IR	1,2

CD₃InD

\tilde{X} C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	6	InD stretch	1115.0	Ar	IR	2

References

- ¹H.-J. Himmel, A. J. Downs, T. M. Greene, and L. Andrews, *Chem. Commun.* 2243 (1999).
²H.-J. Himmel, A. J. Downs, T. M. Greene, and L. Andrews, *Organomet.* **19**, 1060 (2000).

H₂AlNH₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	NH ₂ s-stretch	3499.7	Ar	IR	1,2
	2	AlH ₂ s-stretch	1891.0	Ar	IR	1,2
	3	NH ₂ scissors	1541.6	Ar	IR	1,2
	4	AlN stretch	818.7	Ar	IR	1,2
	5	AlH ₂ scissors	755.0	Ar	IR	1,2
<i>b</i> ₁	7	AlH ₂ OPLA	608.7	Ar	IR	1,2
	8	NH ₂ OPLA	518.3	Ar	IR	1,2
<i>b</i> ₂	10	AlH ₂ a-stretch	1899.3	Ar	IR	1,2
	11	NH ₂ rock	769.8	Ar	IR	1,2

D₂AIND₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	3	ND ₂ scissors	1159.5	Ar	IR	2
	5	AlD ₂ scissors	548.4	Ar	IR	2
<i>b</i> ₁	7	AlD ₂ OPLA	450.4	Ar	IR	2
	8	ND ₂ OPLA	397.7	Ar	IR	2
<i>b</i> ₂	10	AlD ₂ a-stretch	1384.2	Ar	IR	2
	11	ND ₂ rock	611.0	Ar	IR	2

References

- ¹H.-J. Himmel, A. J. Downs, and T. M. Greene, *Chem. Commun.* 871 (2000).
²H.-J. Himmel, A. J. Downs, and T. M. Greene, *J. Am. Chem. Soc.* **122**, 9793 (2000).

H₂GaNH₂

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	NH ₂ s-stretch	3413.4	Ar	IR	1,2
	2	GaH ₂ s-stretch	1970.8T	Ar	IR	1,2
	3	NH ₂ scissors	1530.4	Ar	IR	1,2
	4	GaH ₂ scissors	779.6	Ar	IR	1,2
	5	GaN stretch	706.2	Ar	IR	1,2
<i>b</i> ₁	7	GaH ₂ OPLA	567.7	Ar	IR	1,2
	8	NH ₂ OPLA	304.9	Ar	IR	1,2
<i>b</i> ₂	9	NH ₂ a-stretch	3510.7	Ar	IR	1,2
	10	GaH ₂ a-stretch	1970.8T	Ar	IR	1,2
	11	NH ₂ rock	782.8	Ar	IR	1,2

D₂GaND₂

$\tilde{\chi}$		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	GaD ₂ s-stretch	1407.7	Ar	IR	2
	3	ND ₂ scissors	1150.9	Ar	IR	2
	4	GaN stretch	667.8	Ar	IR	2
	5	GaD ₂ scissors	568.8	Ar	IR	2
b ₁	7	GaD ₂ OPLA	405.9	Ar	IR	2
b ₂	10	GaD ₂ a-stretch	1419.1	Ar	IR	2
	11	ND ₂ rock	605.1	Ar	IR	2

References

¹H.-J. Himmel, A. J. Downs, and T. M. Greene, Chem. Commun. 871 (2000).

²H.-J. Himmel, A. J. Downs, and T. M. Greene, J. Am. Chem. Soc. **122**, 9793 (2000).

H₂InNH₂

$\tilde{\chi}$		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	3	NH ₂ scissors	1506.6	Ar	IR	1,2
	4	InN stretch	616.3	Ar	IR	1,2
b ₂	10	InH ₂ a-stretch	1805.9	Ar	IR	1,2
	11	NH ₂ rock	733.3	Ar	IR	1,2

D₂InND₂

$\tilde{\chi}$		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	4	InD ₂ scissors	474.6	Ar	IR	2
b ₂	10	InD ₂ a-stretch	1299.0	Ar	IR	2

References

¹H.-J. Himmel, A. J. Downs, and T. M. Greene, Chem. Commun. 871 (2000).

²H.-J. Himmel, A. J. Downs, and T. M. Greene, J. Am. Chem. Soc. **122**, 9793 (2000).

CH₂=SiH₂

In an Ar or N₂ matrix, absorption maximum at 258 nm.¹⁻⁴ On irradiation at 254 nm, photoisomerizes to CH₃SiH.^{2,4}

$\tilde{\chi}^a$		C _{2v}					Structure: MW ⁵	
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.		
a ₁		SiH s-stretch	2219m	Ar	IR	1-3		
			2214m	N ₂	IR	2,4		
	CH ₂ scissors	1350w	Ar	IR	2,3			
		1350w	N ₂	IR	2,4			
		Si=C stretch	985w	Ar	IR	1-3		
		985w	N ₂	IR	1,2,4			
SiH ₂ scissors	927w	Ar	IR	1-3				
	927w	N ₂	IR	2,4				
b ₁	CH ₂ wag	741s	Ar	IR	1-3			
		747s	N ₂	IR	2,4			
b ₂	SiH a-stretch	2239m	Ar	IR	1-3			
		2235m	N ₂	IR	2,4			
	CH ₂ rock	817s	Ar	IR	1-3			
		817s	N ₂	IR	2,4			

A₀=3.493; B₀=0.493; C₀=0.432 MW⁵

CH₂=SiD₂

In an Ar matrix, absorption maximum at 259 nm.^{1,4}

$\tilde{\chi}^a$		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	SiD s-stretch	1600m	Ar	IR	1,3	
		1600	N ₂	IR	4	
	CH ₂ scissors	1335	Ar	IR	3	
		1352	N ₂	IR	4	
b ₁	Si=C stretch	952w	Ar	IR	1,3	
		952	N ₂	IR	4	
	CH ₂ wag	719s	Ar	IR	1,3	
		725	N ₂	IR	4	
b ₂	SiD ₂ wag	396w	Ar	IR	1,3	
		396	N ₂	IR	4	
	SiD a-stretch	1635m	Ar	IR	1,3	
		1635	N ₂	IR	4	
CH ₂ rock	759s	Ar	IR	1,3		
	760	N ₂	IR	4		

^aAssigned using density functional theory calculations of Ref. 6.

References

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²H. P. Reisenauer, G. Mihm, and G. Maier, Angew. Chem. **94**, 864 (1982); Angew. Chem. Int. Ed. Engl. **21**, 854 (1982).

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⁴G. Maier, G. Mihm, H. P. Reisenauer, and D. Littmann, Chem. Ber. **117**, 2369 (1984).

⁵S. Bailleux, M. Bogey, J. Demaison, H. Bürger, M. Senzlobler, J. Breitung, W. Thiel, R. Fajgar, and J. Pola, J. Chem. Phys. **106**, 10016 (1997).

⁶V. N. Khabashesku, K. N. Kudin, and J. L. Margrave, J. Mol. Struct. **443**, 175 (1998).

CH₃GeH

\bar{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	3	GeH stretch	1798.6	Ar	IR	1
	5	CH ₃ rock	1201.2	Ar	IR	1
	6	Deformation	868.8	Ar	IR	1
<i>a''</i>	8	CH stretch	2891.6	Ar	IR	1
	11	GeC stretch	535.6	Ar	IR	1

Reference

¹V. N. Khabashesku, K. N. Kudin, J. Tamás, S. E. Bogdanov, J. L. Margrave, and O. M. Nefedov, *J. Am. Chem. Soc.* **120**, 5005 (1998).

CH₃SH⁺

\bar{C}^2A'' C_s
T^a=49820(160) gas PE¹⁻³

\bar{B}^2A' C_s
T^a=34000(160) gas PE¹⁻³

\bar{A}^2A' C_s
T^a=21170(160) gas PE¹⁻³

\bar{X}^2A''		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>		CH ₃ umbrella	1250(80)	gas	PE	2
		SH bend	782(3)T	gas	TPE	6
		CS stretch	687(3)	gas	TPE,PI	1,2,4-6
<i>a''</i>		Torsion	178(3)T	gas	TPE	6

^aFrom vertical ionization potential. The first ionization potential of CH₃SH is taken as 9.553 eV, as in the two-photon pulsed field ionization photoelectron spectroscopic study of Ref. 6.

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CH₂NH₂

$\bar{C}(3p)$		C _s				
T ₀ =31300 gas AB ¹		$\bar{C}-\bar{X}$ 270-320 nm				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>		NH ₂ scissors	1700	gas	AB	1
		Deformation	850	gas	AB	1
		Deformation	600	gas	AB	1

$\bar{B}(3p)$		C _s				
T ₀ =27130 gas AB ¹		$\bar{B}-\bar{X}$ 320-368 nm				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>			830T	gas	AB	1

Reference

¹T. la Cour Jansen, I. Trabjerg, S. Rettrup, P. Pagsberg, and A. Sillesen, *Acta Chem. Scand.* **53**, 1054 (1999).

cyc-(CH)₂BH

\bar{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	2	BH stretch	2639.6	Ar	IR	1,2
	4	Ring deform.	1175.3	Ar	IR	1,2
<i>b</i> ₁	7	Deformation	832.5	Ar	IR	1,2
	8		653.0	Ar	IR	1,2
<i>b</i> ₂	10		1169.5	Ar	IR	1,2

cyc-(CD)₂BD

\bar{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	2	BD stretch	2006.1	Ar	IR	1,2
	4	Ring deform.	1126.0	Ar	IR	1,2
<i>b</i> ₁	8	Deformation	481.6	Ar	IR	1,2

References

- ¹D. V. Lanzisera, P. Hassanzadeh, Y. Hannachi, and L. Andrews, *J. Am. Chem. Soc.* **119**, 12402 (1997).
- ²L. Andrews, D. V. Lanzisera, P. Hassanzadeh, and Y. Hannachi, *J. Phys. Chem. A* **102**, 3259 (1998).

H₂BCCH

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	3	C≡C stretch	2058.1	Ar	IR	1
b ₂	9	BH ₂ a-stretch	2593.3	Ar	IR	1

D₂BCCD

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁			1832.5	Ar	IR	1
b ₂	9	BD ₂ a-stretch	1945.9	Ar	IR	1

Reference

¹L. Andrews, D. V. Lanzisera, P. Hassanzadeh, and Y. Hannachi, *J. Phys. Chem. A* **102**, 3259 (1998).

HBCCH₂

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	BH stretch	2757.7	Ar	IR	1
	3	CC stretch	1859.8	Ar	IR	1

DBCCD₂

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	BD stretch	2168.2	Ar	IR	1
	3	CC stretch	1774.6	Ar	IR	1

Reference

¹L. Andrews, D. V. Lanzisera, P. Hassanzadeh, and Y. Hannachi, *J. Phys. Chem. A* **102**, 3259 (1998).

cyc-C₃H₃⁺

\tilde{X}		D _{3h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e'	4	CH stretch	3130	Ne	IR	1

cyc-C₃D₃⁺

\tilde{X}		D _{3h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e'	4	CD stretch	2344	Ne	IR	1

Reference

¹M. Wyss, E. Riaplov, and J. P. Maier, *J. Chem. Phys.* **114**, 10355 (2001).

CH₂CCH⁺

\tilde{A}^1A'		C _{2v}				
T ₀ = 37330(30)		Ne	AB ³	$\tilde{A}-\tilde{X}$ 246–268 nm		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	4		1629(50)	Ne	AB	3
	11		667(50)	Ne	AB	3

\tilde{a}^3A_2 C_{2v}
T₀ ≅ 13900 gas PE¹

\tilde{X}^1A_1		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CH stretch	3319	gas	TPE	2
	2	CH ₂ s-stretch	3130	gas	TPE	2
	3	C≡C stretch	2093 ^a	gas	TPE	2
			2080	Ne	IR	3
	4	CH ₂ scissors	1465	gas	TPE	2
	5		1106T	gas	TPE	2
b ₁	6	CH ₂ OPLA	1183	gas	TPE	2
	7	CCH deform.	910T	gas	TPE	2

CD₂CCD⁺

\tilde{A}^1A'		C _{2v}				
T ₀ = 37430(30)		Ne	AB ³	$\tilde{A}-\tilde{X}$ 246–267 nm		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	4		1501(50)	Ne	AB	3
	11		641(50)	Ne	AB	3

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		C≡C stretch	1955	Ne	IR	3

^aReassigned by Ref. 3.

References

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³M. Wyss, E. Riaplov, and J. P. Maier, *J. Chem. Phys.* **114**, 10355 (2001).

CH₂BNH \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		BN stretch	1881.0	Ar	IR	1

Reference

- ¹D. V. Lanzisera and L. Andrews, *J. Phys. Chem. A* **101**, 824 (1997).

CH₃NB \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NB stretch	1981.1	Ar	IR	1

Reference

- ¹D. V. Lanzisera and L. Andrews, *J. Phys. Chem. A* **101**, 824 (1997).

CH₂CCH

In the gas phase, absorption between 230 and 280 nm, with a maximum at 41320 (242 nm), has been assigned⁹ to CH₂CCH.

\tilde{B}^2A''	C _s					
T ₀ =30109U	gas	AB ¹ CR ¹²	$\tilde{A}, \tilde{B}-\tilde{X}$	290–353 nm		
29146(17)	Ne	AB ¹³	$\tilde{B}-\tilde{X}$	294–343 nm		
	Ar	AB ²	$\tilde{A}, \tilde{B}-\tilde{X}$	288–359 nm		

All bands in the gas-phase spectrum are diffuse.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	5	C–C stretch	961(10)	gas	AB	1
			966(25)	Ne	AB	13
			965(10)	Ar	AB	2
	6	CCH bend	661(25)	Ne	AB	13
	8	C ₃ deform.	301(25)	Ne	AB	13

\tilde{A}^2A'	C _s					
T ₀ =28409(16)	Ne	AB ¹³	$\tilde{A}-\tilde{X}$	345–352 nm		

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	8	C ₃ deform.	285(25)	Ne	AB	13

\tilde{X}^2B_1	C _{2v}					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH stretch	3322.29	gas	CC	5,11
			3308.8m	Ar	IR	2,4
		CH ₂ wag	687.18	gas	DL	6,8
			686.5m	Ar	IR	2
		C ₃ deformation	490T	gas	PE	3,7
			483.5m	Ar	IR	2,4

A₀=9.608; B₀=0.318; C₀=0.307 CC⁵DL⁸MW¹⁰

CD₂CCD

\tilde{B}^2A''	C _s					
T ₀ =29189(17)	Ne	AB ¹³	$\tilde{B}-\tilde{X}$	310–343 nm		

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	6	CCD bend	547(25)	Ne	AB	13
	8	C ₃ deform.	301(25)	Ne	AB	13

\tilde{X}^2B_1	C _{2v}					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CD stretch	2546.8m	Ar	IR	2,4
			552.9m	Ar	IR	2
			479.8	Ar	IR	4
			350T	gas	PE	7

References

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¹²D. B. Atkinson and J. W. Hudgens, *J. Phys. Chem. A* **103**, 4242 (1999).
¹³M. Wyss, E. Riaplov, and J. P. Maier, *J. Chem. Phys.* **114**, 10355 (2001).

SiH₂CCH

\tilde{X}	C _s					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	3	CC stretch	2055.6	Ar	IR	1
	4	SiH ₂ scissors	926.8	Ar	IR	1

SiD₂CCD

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	CC stretch	1927.2	Ar	IR	1
	4	SiD ₂ scissors	701.9	Ar	IR	1

Reference

¹D. S. Han, C. M. L. Rittby, and W. R. M. Graham, *J. Chem. Phys.* **109**, 8355 (1998).

CaOCH₃

\tilde{C}^2E C_{3v}
 $T_0 = 21750T$ gas LF³ $\tilde{C}-\tilde{X}$ 446–462 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CaO stretch	440(5)	gas	LF	3
		Skel. deform.	150(5)	gas	LF	3

\tilde{B}^2A_1 C_{3v}
 $T_0 = 17682.431$ gas CL¹LF^{1,2,4,5} $\tilde{B}-\tilde{X}$ 525–590 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	3	CO stretch	1150(5)	gas	LF	1
	4	CaO stretch	491(5)	gas	LF	1,2
e	8	CaOC bend	168(5)	gas	LF	1,2

$B_0 = 0.118$ LF⁵

\tilde{A}^2E C_{3v}
 $T_0 = 15930(10)$ gas CL¹LF^{1,2} $\tilde{A}-\tilde{X}$ 605–635 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	3	CO stretch	1140(5)	gas	LF	1, 2
	4	CaO stretch	500(10)	gas	LF	1,2
e	8	CaOC bend	145(5)	gas	LF	1

$A = 68(5)$ gas LF^{1,2}

\tilde{X}^2A_1 C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	3	CO stretch	1156(5)	gas	LF	2
	4	CaO stretch	488(5)	gas	LF	1,2
e	8	CaOC bend	144(5)	gas	LF	1,2

$A_0 = 5.448T$; $B_0 = 0.116$ DR⁴LF⁵

CaOCD₃

\tilde{B}^2A_1 C_{3v}
 $T_0 = 17674(5)$ gas LF¹ $\tilde{B}-\tilde{X}$ 528–600 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	3	CO stretch	1156(5)	gas	LF	1
	4	CaO stretch	476(5)	gas	LF	1
e	8	CaOC bend	166(5)	gas	LF	1

\tilde{A}^2E C_{3v}
 $T_0 = 15935(10)$ gas LF¹ $\tilde{A}-\tilde{X}$ 584–630 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	3	CO stretch	1156(5)	gas	LF	1
	4	CaO stretch	480(5)	gas	LF	1
e	8	CaOC bend	140(5)	gas	LF	1

$A = 72(5)$ gas LF¹

\tilde{X}^2A_1 C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	4	CaO stretch	467(5)	gas	LF	1
e	8	CaOC bend	142(5)	gas	LF	1

References

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CH₃BO

\tilde{X} C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CBO a-stretch	1972.4	Ar	IR	1
		CH ₃ umbrella	1305.0	Ar	IR	1
e		Deformation	896.8 895.6	Ar	IR	1

CD₃BO

\tilde{X} C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CBO a-stretch	1972.0	Ar	IR	1
e		Deformation	750.0	Ar	IR	1

Reference

¹D. V. Lanzisera and L. Andrews, *J. Phys. Chem. A* **101**, 1482 (1997).

CH₂BOH
 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1748.3	Ar	IR	1

CD₂BOD
 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1724.9	Ar	IR	1

Reference

¹D. V. Lanzisera and L. Andrews, *J. Phys. Chem. A* **101**, 1482 (1997).

CH₂CCH⁻

Threshold for electron detachment from ground-state CH₂CCH⁻ = 7410(65) gas PE^{1,2}

 \tilde{X}
C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		C ₃ a-stretch	1857.9	Ne	IR	3
			687.0	Ne	IR	3

CD₂CCD⁻

Threshold for electron detachment from ground-state CD₂CCD⁻ = 7380(65) gas PE²

 \tilde{X}
C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		C ₃ a-stretch	1802.6	Ne	IR	3

References

- ¹J. M. Oakes and G. B. Ellison, *J. Am. Chem. Soc.* **105**, 2969 (1983).
²M. S. Robinson, M. L. Polak, V. M. Bierbaum, C. H. DePuy, and W. C. Lineberger, *J. Am. Chem. Soc.* **117**, 6766 (1995).
³D. Forney, M. E. Jacox, C. L. Lugez, and W. E. Thompson, *J. Chem. Phys.* **115**, 8418 (2001).

CH₃BCl
 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		B–C stretch	1172.7	Ar	IR	1

CD₃BCl
 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		B–C stretch	1161.0	Ar	IR	1

Reference

¹D. V. Lanzisera and L. Andrews, *J. Phys. Chem. A* **104**, 9295 (2000).

CH₃BBr
 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		B–C stretch	1142.4	Ar	IR	1

CD₃BBr
 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		B–C stretch	1161.0	Ar	IR	1

Reference

¹D. V. Lanzisera and L. Andrews, *J. Phys. Chem. A* **104**, 9295 (2000).

CH₂CHO \tilde{B}^2A'' C_s

$T_0 = 28784.09(1)$ gas AB¹LF^{2,4,6,12,16}FD¹⁰PF¹³ $\tilde{B}-\tilde{X}$ 300–405 nm
 The failure to detect fluorescence on excitation above 30000 suggests² the onset of predissociation near 330 nm. In the gas phase,¹³ both CH₃ and H atoms are produced above 28700. In the argon matrix experiments,⁵ the threshold for the photodecomposition of CH₂CHO to produce CH₃ + CO was observed between 280 and 300 nm.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a'	4	CO stretch	1621	gas	FD,PF LF,DR	10,12,13 14	
	5	CH ₂ scissors	1405.23	gas	FD,PF LF,DR	10,12,13 14,16	
	6	OCH bend	1274	gas	PF,LF,DR	13,14	
	7	CH ₂ rock	1121.65	gas	LF,FD PF,DR	2,6,10,12 13,14,16	
	8	CC stretch	917.10	gas	LF,FD PF,DR	2,6,10,12 13,14,16	
	9	CCO bend	449	gas	LF,FD PF,DR	2,6,10,12 13,14	
	a''	10	HCO wag	595H	gas	LF,DR	14
		11	CH ₂ wag	436H	gas	PF,LF,DR	13,14
		12	Torsion	274H	gas	PF,LF,DR	13,14

 $\tau_0 = 190(2)$ ns gas LF^{11,14} $A_0 = 2.103(4)$; $B_0 = 0.344(1)$; $C_0 = 0.296(1)$ LF⁶ \tilde{A}^2A' C_s $T_0 = 8190(120)$ gas AB³PE¹⁵ $\tilde{A}-\tilde{X}$ 1000–1250 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	4	CC stretch	1580(120)	gas	PE	15
	5	CH ₂ scissors	1350(120)	gas	PE	15
	9	CCO bend	460(120)	gas	PE	15

 \tilde{X}^2A'' C_s Structure: MW^{8,9}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a'	4	CCO stretch	1543 1542m HF 1525m	gas Ar	LF IR	2,6,12,14 5	
	5	CH ₂ scissors	1486	gas	LF	14	
	6	OCH deform.	1366 1375m HF	gas Ar	LF IR	12,14 5	
	7	CC stretch	1143	gas	LF,PD	2,6,7,14	
	8	CH ₂ rock	957	gas	LF	14	
	9	CCO bend	500	gas	LF,PD	2,6,7,14	
	a''	10	CHO wag	703H	gas	LF	14
		11	CH ₂ wag	557H	gas	LF	14
		12	Torsion	404H	gas	LF	14

 $A_0 = 2.224$; $B_0 = 0.382$; $C_0 = 0.326$ LF⁶MW^{8,17}CD₂CDO \tilde{B}^2A'' C_s $T_0 = 28840$ gas LF²PF¹³ $\tilde{B}-\tilde{X}$ 300–411 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'			980	gas	LF	2
			768	gas	LF	2

 \tilde{X}^2A'' C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		C=O stretch	1540	gas	LF	2
			1513ms HF	Ar	IR	5
			1223w HF	Ar	IR	5
		CC stretch	1050	gas	LF	2
			CCO bend	445	gas	LF

 $A_0 = 1.442$; $B_0 = 0.336$; $C_0 = 0.272$ MW⁹

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CH₃MgF \tilde{X} C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CH ₃ s-stretch	2907.2	Ar	IR	1
	2	CH ₃ s-deform.	1127.2	Ar	IR	1
	3	CMgF a-stretch	755.8	Ar	IR	1
e	5	CH ₃ a-stretch	2958.5	Ar	IR	1
	7	CH ₃ rock	560.2	Ar	IR	1

CD₃MgF

$\tilde{\chi}$		C _{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CD ₃ s-stretch	2211.0	Ar	IR	1
	2	CD ₃ s-deform.	880.8	Ar	IR	1
	3	CMgF a-stretch	744.5	Ar	IR	1
<i>e</i>	5	CD ₃ a-stretch	2232.6	Ar	IR	1
	7	CD ₃ rock	432.3 431.3	Ar	IR	1

Reference

¹W. D. Bare and L. Andrews, J. Am. Chem. Soc. **120**, 7293 (1998).

HMgCH₂F

$\tilde{\chi}$		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		MgH stretch	1547T	Ar	IR	1

Reference

¹W. D. Bare and L. Andrews, J. Am. Chem. Soc. **120**, 7293 (1998).

CH₃MgCl

$\tilde{\chi}$		C _{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CH ₃ s-stretch	2910.4	Ar	IR	1
	2	CH ₃ s-deform.	1127.5	Ar	IR	1
	3	CMgCl a-stretch	627.1	Ar	IR	1
<i>e</i>	5	CH ₃ a-stretch	2951.3	Ar	IR	1
	7	CH ₃ rock	556.6	Ar	IR	1

Reference

¹W. D. Bare and L. Andrews, J. Am. Chem. Soc. **120**, 7293 (1998).

HMgCH₂Cl

$\tilde{\chi}$		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		MgH stretch	1555.4T	Ar	IR	1

Reference

¹W. D. Bare and L. Andrews, J. Am. Chem. Soc. **120**, 7293 (1998).

CH₃MgBr

$\tilde{\chi}$		C _{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CH ₃ s-stretch	2910.6	Ar	IR	1
	2	CH ₃ s-deform.	1125.4	Ar	IR	1
	3	CMgBr a-stretch	599T	Ar	IR	1
<i>e</i>	5	CH ₃ a-stretch	2950.9	Ar	IR	1
	7	CH ₃ rock	555.3	Ar	IR	1

CD₃MgBr

$\tilde{\chi}$		C _{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CD ₃ s-stretch	2123.5	Ar	IR	1
	2	CD ₃ s-deform.	881.2	Ar	IR	1
	3	CMgBr a-stretch	575T	Ar	IR	1
<i>e</i>	5	CD ₃ a-stretch	2138.5	Ar	IR	1
	7	CD ₃ rock	429.0	Ar	IR	1

Reference

¹W. D. Bare and L. Andrews, J. Am. Chem. Soc. **120**, 7293 (1998).

CH₃MgI

$\tilde{\chi}$		C _{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CH ₃ s-stretch	2910.3	Ar	IR	1
	2	CH ₃ s-deform.	1125br	Ar	IR	1
	3	CMgI a-stretch	588T	Ar	IR	1
<i>e</i>	5	CH ₃ a-stretch	2950br	Ar	IR	1
	7	CH ₃ rock	558.2	Ar	IR	1

Reference

¹W. D. Bare and L. Andrews, J. Am. Chem. Soc. **120**, 7293 (1998).

HSc(OH)₂

$\tilde{\chi}$						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		ScH stretch	1418.3wT	Ar	IR	2
		ScO ₂ a-stretch	729.6T	Ar	IR	1,2

DSc(OD)₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		ScO ₂ a-stretch	713.3T	Ar	IR	1,2

References

¹J. W. Kauffman, R. H. Hauge, and J. L. Margrave, *J. Phys. Chem.* **89**, 3547 (1985).

²L. Zhang, J. Dong, and M. Zhou, *J. Phys. Chem. A* **104**, 8882 (2000).

HY(OH)₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		YH stretch	1368.6	Ar	IR	1
		YO ₂ a-stretch	616.9	Ar	IR	1

DY(OD)₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		YD stretch	981.6	Ar	IR	1
		YO ₂ a-stretch	593.4	Ar	IR	1

Reference

¹L. Zhang, L. Shao, and M. Zhou, *Chem. Phys.* **272**, 27 (2001).

HLa(OH)₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		LaH stretch	1229.0	Ar	IR	1
		LaO ₂ a-stretch	554.3	Ar	IR	1

DLa(OD)₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		LaD stretch	880.3	Ar	IR	1
		LaO ₂ a-stretch	534.5	Ar	IR	1

Reference

¹L. Zhang, L. Shao, and M. Zhou, *Chem. Phys.* **272**, 27 (2001).

CH₂CHO⁻**Dipole-Bound State C_s** $T_0 = 14712.747(5)$ gas PD^{1,3,4}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		C-C stretch	1143	gas	PD	3
		CCO bend	499	gas	PD	3
a''		Torsion	102H	gas	PD	3

$A_0 = 2.221(2)$; $B_0 = 0.376$; $C_0 = 0.320$ PD^{3,4}

Threshold for electron detachment from ground-state CH₂CHO⁻ = 14480(120) gas PE^{2,6}

 \tilde{X} C_s Structure: PD³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		CCO bend	525.82	gas	PD	3,4
a''		Torsion	375T	gas	PD	3

$A_0 = 2.494$; $B_0 = 0.362$; $C_0 = 0.316$ PD^{3,4}

CD₂CDO⁻**Dipole-Bound State C_s** $T_0 = 14665.97(5)$ gas PD^{1,3,5}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'			1382(10)	gas	PD	1
			981(10)	gas	PD	1
		CCO bend	437(10)	gas	PD	1
a''		Torsion	80HT	gas	PD	1

$A_0 = 1.419(3)$; $B_0 = 0.330$; $C_0 = 0.268$ PD^{3,5}

Threshold for electron detachment from ground-state CD₂CDO⁻ = 14660(240) gas PE²

 \tilde{X} C_s $A_0 = 1.554(1)$; $B_0 = 0.319$; $C_0 = 0.264$ PD³**References**

¹R. L. Jackson, P. C. Hiberty, and J. I. Brauman, *J. Chem. Phys.* **74**, 3705 (1981).

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CH₃S₂⁺

\tilde{c}^3A' C_s
 $T_0=25740(230)$ gas PE¹

\tilde{B}^1A'' C_s
 $T_0=16860(230)$ gas PE¹

\tilde{b}^3A'' C_s
 $T_0=13390(230)$ gas PE¹

\tilde{A}^1A'' C_s
 $T_0=10570(230)$ gas PE¹

\tilde{a}^3A'' C_s
 $T_0=5890(230)$ gas PE¹

\tilde{X}^1A' C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		SS stretch	600(60)	gas	PE	1

¹From vertical ionization potential.

Reference

¹M. Ge, J. Wang, Z. Sun, X. Zhu, and D. Wang, *J. Chem. Phys.* **114**, 3051 (2001).

CH₃O₂

An unstructured gas-phase absorption between 200 and 300 nm, with a maximum at 235 nm, has been assigned^{1,3,4,6,7,9} to CH₃O₂. In the gas phase, CH₃O₂ photolyzes on exposure to 248-nm radiation,⁸ and, in an argon matrix, on exposure to 254-nm radiation.⁵

\tilde{A}^2A' C_s
 $T_0=7382.8(5)$ gas AB²CR¹⁰PE¹¹ $\tilde{A}-\tilde{X}$ 7375–9149 cm⁻¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		Mixed	1005(10)	gas	PE	11
		Mixed	896(9)	gas	AB,PE	2,11

\tilde{X}^2A'' C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CH ₃ stretch	3032	Ar	IR	12
	2	CH ₃ stretch	2954	Ar	IR	12
	3	CH ₃ deform.	1448s	Ar	IR	5,12
	4	CH ₃ deform.	1410w	Ar	IR	5,12
	5	Mixed	1180m	Ar	IR	5,12
	6	Mixed	1124(5)	gas	PE	11
			1109wm	Ar	IR	5,12
			902s	Ar	IR	5,12
a''	7	CO stretch	902s	Ar	IR	5,12
	8	COO bend	482(9)	gas	PE	11
a''	9	CH ₃ stretch	492wm	Ar	IR	5,12
	10	CH ₃ deform.	3024	Ar	IR	12
		1434vs	Ar	IR	5,12	

CD₃O₂

\tilde{A}^2A' C_s
 $T_0=7372.6(5)$ gas CR¹⁰PE¹¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		Mixed	975(10)	gas	PE	11
		Mixed	840(10)	gas	PE	11

\tilde{X}^2A'' C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CD ₃ stretch	2280wm	Ar	IR	12
	2	CD ₃ stretch	2172m	Ar	IR	5,12
	3	CD ₃ deform.	1123(7)	gas	PE	11
			1144ms	Ar	IR	5,12
			1076s	Ar	IR	5,12
			1050wm	Ar	IR	12
			941wm	Ar	IR	12
			822vs	Ar	IR	5,12
a''	7	CO stretch	822vs	Ar	IR	5,12
	8	COO bend	440(7)	gas	PE	11
			446wm	Ar	IR	5,12
a''	9	CD ₃ stretch	2273s	Ar	IR	12
	10	CD ₃ deform.	1046ms	Ar	IR	5,12
	11	CD ₃ rock	860wm	Ar	IR	5,12

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CH₃SCI⁺

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			631(60)	gas	PI	1

Reference

- B.-M. Cheng, E. P. Chew, C.-P. Liu, J.-S. K. Yu, and C.-H. Yu, *J. Chem. Phys.* **110**, 4757 (1999).

CH₃O₂⁻

Threshold for electron detachment from ground-state
CH₃O₂⁻ = 9370(40) gas PE¹

Reference

¹S. J. Blanksby, T. M. Ramond, G. E. Davico, M. R. Nimlos, S. Kato, V. M. Bierbaum, W. C. Lineberger, G. B. Ellison, and M. Okumura, *J. Am. Chem. Soc.* **123**, 9585 (2001).

CICH₂OH \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OH stretch	3646	gas	IR	3
			3591	Ar	IR	1
		CH ₂ a-stretch	2981	Ar	IR	1
			2971			
		CH ₂ s-stretch	2913	Ar	IR	1
		COH bend	1374	gas	IR	2,3
			1393	Ar	IR	1
		CH ₂ wag	1318	gas	IR	2,3
			1323	Ar	IR	1
			1319			
			1231	Ar	IR	1
			1176	gas	IR	3
		CH ₂ twist	1114	Ar	IR	1
		CO stretch	1083	gas	IR	2,3
			1096	Ar	IR	1
		CH ₂ rock	960	gas	IR	2,3
			959	Ar	IR	1
		CCl stretch	697	gas	IR	3
			669	Ar	IR	1
		CICO deform.	469	Ar	IR	1
			463			
		OH torsion	372	Ar	IR	1
			368			

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CH₃SCI \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			513(60)	gas	PI	1
			244(60)	gas	PI	1

Reference

¹B.-M. Cheng, E. P. Chew, C.-P. Liu, J.-S. K. Yu, and C.-H. Yu, *J. Chem. Phys.* **110**, 4757 (1999).

HCCCNH⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	NH stretch	3513.96	gas	LD	1
	3	CN stretch	2315.14	gas	DL	2

$B_0 = 0.144$ LD¹MW^{3,4}

References

¹S. K. Lee and T. Amano, *Astrophys. J.* **323**, L145 (1987).
²K. Kawaguchi, M. Kajita, K. Tanaka, and E. Hirota, *J. Mol. Spectrosc.* **144**, 451 (1990).
³K. Kawaguchi, Y. Kasai, S. Ishikawa, M. Ohishi, N. Kaifu, and T. Amano, *Astrophys. J.* **420**, L95 (1994).
⁴C. A. Gottlieb, A. J. Apponi, M. C. McCarthy, P. Thaddeus, and H. Linhart, *J. Chem. Phys.* **113**, 1910 (2000).

HOCH=C=C:

In an argon matrix, the threshold for rearrangement to propynal (HC≡CCHO) lies between 380 and 360 nm.¹

On broad-band irradiation, short-lived absorptions, attributed to the *cisoid*-rotamer, appear² at 1999.6 and 1254.3.

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		C ₃ a-stretch	1992.5vs	Ar	IR	1,2
		COH bend	1460.8wm	Ar	IR	1,2
		Mixed	1280.5w	Ar	IR	2
		Deformation	1223.1wm	Ar	IR	1,2
		Skel. stretch	1016.3wm	Ar	IR	1,2

References

¹B. J. Ortman, R. H. Hauge, J. L. Margrave, and Z. H. Kafafi, *J. Phys. Chem.* **94**, 7973 (1990).
²J. Szczepanski, S. Ekern, and M. Vala, *J. Phys. Chem.* **99**, 8002 (1995).

CH₂CFO \tilde{B}^2A''

$T_0 = 29867.4$ gas LF¹⁻⁴ $\tilde{B}-\tilde{X}$ 307-335 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	3	CO stretch	1790	gas	LF	1,4
	4	CH ₂ scissors	1409	gas	LF	4
	5	CF stretch	1253	gas	LF	1,4
	6	CH ₂ rock	911	gas	LF	1,4
	7	CC stretch	874	gas	LF	1,4
	8	FCO bend	537	gas	LF	1,4
	9	CCO bend	421	gas	LF	1,4

$\tau_0 = 81.0(5)$ ns gas LF¹
 $1/2(A_0 + B_0) = 0.345(4)$; $C_0 = 0.181(4)$ LF²

\tilde{X}^2A'' C_s						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	3	CO stretch	1724	gas	LF	1
	4	CH ₂ scissors	1475	gas	LF	4
	5	CF stretch	1211	gas	LF	1
	6	CH ₂ rock	906T	gas	LF	1
	7	CC stretch	847	gas	LF	1
	8	FCO bend	584	gas	LF	1
	9	CCO bend	416T	gas	LF	1

$$1/2(A_0 + B_0) = 0.360(6); C_0 = 0.185(6) \quad LF^2$$

CD₂CFO

\tilde{B}^2A'' C_s						
$T_0 = 29867$ gas		LF^4		$\tilde{B}-\tilde{X}$ 316–335 nm		
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	3	CO stretch	1772	gas	LF	4
	4	CF stretch	1241	gas	LF	4
	5	CD ₂ scissors	1073	gas	LF	4
	6	CC stretch	827	gas	LF	4
	7	CD ₂ rock	783	gas	LF	4
	8	CCF bend	530	gas	LF	4
	9	CCO bend	370	gas	LF	4

$$\tau_0 = 78(11) \text{ ns} \quad \text{gas} \quad LF^4$$

\tilde{X}^2A'' C_s						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'	3	CO stretch	1735	gas	LF	4
	4	CF stretch	1248	gas	LF	4
	5	CD ₂ scissors	1043	gas	LF	4
	6	CC stretch	863	gas	LF	4
	7	CD ₂ rock	774T	gas	LF	4
	8	CCF bend	597	gas	LF	4
	9	CCO bend	370	gas	LF	4

References

- ¹M. Furubayashi, I. Bridier, S. Inomata, N. Washida, and K. Yamashita, *J. Chem. Phys.* **106**, 6302 (1997).
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- ³M. Yamaguchi, M. Furubayashi, S. Inomata, and N. Washida, *Chem. Phys. Lett.* **298**, 93 (1998).
- ⁴S. Inomata, M. Furubayashi, T. Imamura, N. Washida, and M. Yamaguchi, *J. Chem. Phys.* **111**, 6356 (1999).

t-CHCICHO

In the gas phase, a fluorescence spectrum with origin at 29040 has been attributed¹ to *t*-CHCICHO.

Reference

- ¹N. Washida, M. Furubayashi, T. Imamura, I. Bridier, and A. Miyoshi, *J. Chem. Phys.* **107**, 6998 (1997).

CH₂=SiCl₂

In an Ar matrix, absorption maximum at 246 nm.¹

\tilde{X}^a C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1		Si=C stretch	1008m	Ar	IR	1
b_2		CH ₂ rock	732s	Ar	IR	1
		SiCl ₂ a-stretch	593m	Ar	IR	1

^aAssigned using density functional theory calculations of Ref. 2.

References

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HCCl₂OH

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
		OH stretch	3611	gas	IR	2
		COH bend	1388	gas	IR	1,2
		CH a-rock	1221	gas	IR	1,2
		CO stretch	1105	gas	IR	1,2
			1003	gas	IR	1
		CCl ₂ a-stretch	740	gas	IR	2

References

- ¹G. S. Tyndall, T. J. Wallington, M. D. Hurley, and W. F. Schneider, *J. Phys. Chem.* **97**, 1576 (1993).
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C₅H

$$\tilde{X}^2\Pi \quad C_{\infty v}$$

$$A_{\text{eff}} = 24.20 \quad MW^3$$

$$B_0 = 0.080 \quad MW^{1.3}$$

C₅D

$$\tilde{X}^2\Pi \quad C_{\infty v}$$

$$A_{\text{eff}} = 24.2 \quad MW^2$$

$$B_0 = 0.075 \quad MW^2$$

References

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SiCCCCH
 $\tilde{X}^2\Pi$
 $B=0.047$ MW¹ C_{∞v}
Reference
¹M. C. McCarthy, A. J. Apponi, C. A. Gottlieb, and P. Thaddeus, *J. Chem. Phys.* **115**, 870 (2001).
CCCHCN
 \tilde{X} C_s
 $A_0=1.067$; $B_0=0.098$; $C_0=0.089$ MW¹
Reference
¹M. C. McCarthy, A. J. Apponi, V. D. Gordon, C. A. Gottlieb, P. Thaddeus, T. D. Crawford, and J. F. Stanton, *J. Chem. Phys.* **111**, 6750 (1999).
HCCCCN
 \tilde{X} C_{∞v}
 $B_0=0.077$ MW¹
Reference
¹J. Tang, Y. Sumiyoshi, and Y. Endo, *Chem. Phys. Lett.* **315**, 69 (1999).
(cyc-HC₃)CN
 \tilde{X} C_s
 $A_0=1.144$; $B_0=0.117$; $C_0=0.106$ MW¹
Reference
¹M. C. McCarthy, J.-U. Grabow, M. J. Travers, W. Chen, C. A. Gottlieb, and P. Thaddeus, *Astrophys. J.* **513**, 305 (1999).
NCCH=C=O

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CH stretch	3076.7w	Ar	IR	2
	2	CN stretch	2240.2w	Ar	IR	1,2
	3	CCO a-stretch	2162.8vs	Ar	IR	1,2
	4	CH bend	1364.4w	Ar	IR	2
<i>a''</i>	10	CH bend	552.8wm	Ar	IR	2

NCCD=C=O

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CD stretch	2273.7	Ar	IR	2
	2	CN stretch	2235.3	Ar	IR	2
	3	CCO a-stretch	2155.8	Ar	IR	2
	4		1325.5	Ar	IR	2
<i>a''</i>	10	CD bend	539.3	Ar	IR	2

References
¹D. W. J. Moloney, M. W. Wong, R. Flammang, and C. Wentrup, *J. Org. Chem.* **62**, 4240 (1997).

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c-CHFCFO
 In the gas phase, a fluorescence spectrum with origin at 31644 has been attributed¹ to *c*-CHFCFO.
References
¹N. Washida, M. Furubayashi, T. Imamura, I. Bridier, and A. Miyoshi, *J. Chem. Phys.* **107**, 6998 (1997).
t-CHFCFO
 In the gas phase, a fluorescence spectrum with origin at 31270 has been attributed¹ to *t*-CHFCFO.
References
¹N. Washida, M. Furubayashi, T. Imamura, I. Bridier, and A. Miyoshi, *J. Chem. Phys.* **107**, 6998 (1997).
CHCICFO
 In the gas phase, a fluorescence spectrum with origin at 30715 has been tentatively attributed¹ to CHCICFO.
References
¹N. Washida, M. Furubayashi, T. Imamura, I. Bridier, and A. Miyoshi, *J. Chem. Phys.* **107**, 6998 (1997).

HOONO₂

In the gas phase, the threshold for unstructured absorption lies near 330 nm. This absorption increases steadily to a maximum at or beyond 190 nm.^{4-7,10,12}

\bar{X}		Structure: MW ⁸				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	OH stretch	3540.1wm	gas	IR	1,4,12
	2	NO ₂ a-stretch	1728.3vs	gas	IR	1,3,7,12
			1726.2vs	Ar	IR	14
	3	OH bend	1396.9wm	gas	IR	1,3,4,7,12,15
	4	NO ₂ s-stretch	1304.2s	gas	IR	1-4,7,12
			1292.7m	Ar	IR	14
	5	O-O stretch	941.0w	gas	IR	1,3,11,12
			922.1w	gas	IR	1,3
919.2w			gas	IR	1,3,11	
6	NO ₂ scissors	802.54m	gas	IR,DL	1-4,7,9,12,13	
7	N-O stretch	648	gas	IR	13	
8	NO ₂ rock	466	gas	IR	13	
a''	10	NO ₂ wag	722	gas	IR	11,13
	12	NO ₂ torsion	145(6)	gas	MW	8

A₀ = 0.400; B₀ = 0.156; C₀ = 0.113 gas MW⁸

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HSO₄

\bar{X}
T₀ = 5240(1130) gas PE¹

References

- X.-B. Wang, J. B. Nicholas, and L.-S. Wang, *J. Phys. Chem. A* **104**, 504 (2000).

CCl₃OH

\bar{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		OH stretch	3604	gas	IR	1
		COH bend	1311	gas	IR	1
		CO stretch	1113	gas	IR	1
		CCl ₃ stretch	784	gas	IR	1

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HSO₄⁻

Threshold for electron detachment from ground-state HSO₄ = 38320(800) gas PE¹

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HOOCIO₂

\bar{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a	1	OH stretch	3526.7	Ar	IR	1
	2	HOO bend	1282.6	Ar	IR	1
	3	ClO ₂ a-stretch	1207.4	Ar	IR	1
	4	ClO ₂ s-stretch	1046.4	Ar	IR	1
	5	OO stretch	893.2	Ar	IR	1

DOOCIO₂

\bar{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a	1	OD stretch	2604.7	Ar	IR	1
	2	ClO ₂ a-stretch	1207.4	Ar	IR	1
	3	ClO ₂ s-stretch	1048.3	Ar	IR	1
	4	DOO bend	956.7	Ar	IR	1
	5	OO stretch	889.7	Ar	IR	1

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Al₅C

\bar{C} C_{2v}
 $T_0 = 11540(650)$ gas PE¹

\bar{B} C_{2v}
 $T_0 = 3630(580)$ gas PE¹

\bar{A} C_{2v}
 $T_0 = 1690(520)$ gas PE¹

References

¹A. I. Boldyrev, J. Simons, X. Li, and L.-S. Wang, J. Chem. Phys. **111**, 4993 (1999).

Al₅C⁻

Threshold for electron detachment from ground-state Al₅C=21060(320) gas PE¹

References

¹A. I. Boldyrev, J. Simons, X. Li, and L.-S. Wang, J. Chem. Phys. **111**, 4993 (1999).

TiC₅

\bar{A}
 $T_0 = 7640(560)$ gas PE¹

\bar{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			240(60)	gas	PE	1

References

¹X.-B. Wang, C.-F. Ding, and L.-S. Wang, J. Phys. Chem. A **101**, 7699 (1997).

TiC₅⁻

Threshold for electron detachment from ground-state TiC₅⁻=14100(400) gas PE¹

References

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C₆

(2) $^3\Sigma_u^-$ $D_{\infty h}$
 $T_0 = 42108(35)$ Ne AB¹⁵

(5) $^3\Pi_u$ $D_{\infty h}$
 $T_0 = 40090(32)T$ Ne AB¹⁵ $^3\Pi_u - \bar{X}$ 241-249 nm

In an argon matrix, the growth behavior of the 1952.5 cm⁻¹ infrared absorption on sample warmup has been tentatively correlated with that of an absorption maximum near 246 nm.⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	3	Stretch	644(45)	Ne	AB	15

$^3\Sigma_u^-$ $D_{\infty h}$
 $T_0 = 19558(5)$ Ne AB^{10,15} $^3\Sigma_u^- - \bar{X}$ 430-512 nm
 19220 Ar AB¹²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1		2096(5)	Ne	AB	10
	2		1566(5)	Ne	AB	10
	3		640(5)	Ne	AB	10

$^3\Pi_u$ $D_{\infty h}$
 $T_0 = 8880(160)$ gas PE¹⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	3	Stretch	640(50)	gas	PE	14

$^3\Sigma_u^+$ $D_{\infty h}$
 $T_0 = 6850(120)$ gas PE¹⁴

$^1\Delta_g$ $D_{\infty h}$
 $T_0 = 1340(120)$ gas PE₁₄

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	2016(50)	gas	PE	14
	2	Sym. stretch	1686(50)	gas	PE	14

\bar{X} $^3\Sigma_g^-$ $D_{\infty h}$ Structure: ESR²DL⁸

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	2061(10)	gas	TPE,PE	6,14
			2050T	Ar	IR	11
	2	Sym. stretch	1694(50)	gas	PE	14
	3	Stretch	637(50)	gas	PE	14
Σ_u^+	4	Asym. stretch	1959.86	gas	DL	8
			1958.7	Ne	IR	9,10,13
			1952.0	Ar	IR	3,4,7
			1951.2	Kr	IR	11
			1956.8	H ₂	IR	13
	5	Asym. stretch	1199.4	Ne	IR	9
			1197.3	Ar	IR	1,3,7
			1197.0	Kr	IR	11
Π_g	7	Bend	246(50)H	gas	PE	5,14
Π_u	9	Bend	90(50)H	gas	PE	14

$B_0 = 0.048$ DL⁸

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cyc-C₆

\tilde{X}		D _{3h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e'	4		1694.9	Ar	IR	1,2

References

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SiC₄Si

\tilde{a}		D _{∞h}				
T ₀ = 888(32)		gas PE ²				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	1		1790(20)	gas	PE	2
	2		990(20)	gas	PE	2
Σ _u ⁺	4	C=C stretch	1807.4	Ar	IR	1
	5	Si-C stretch	719.1	Ar	IR	1

$\tilde{X}^3\Sigma^-$		D _{∞h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	1		1870(20)	gas	PE	2
	2		1200(20)	gas	PE	2
Σ _u ⁺	4	C=C stretch	1807.4	Ar	IR	1
	5	Si-C stretch	719.1	Ar	IR	1

References

- 1 J. D. Presilla-Márquez, C. M. L. Rittby, and W. R. M. Graham, *J. Chem. Phys.* **106**, 8367 (1997).
- 2 G. E. Davico, R. L. Schwartz, and W. C. Lineberger, *J. Chem. Phys.* **115**, 1789 (2001).

Si₆

\tilde{A}		D _{3h}				
T ₀ = 6450T		gas PE ³				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			323(20)	gas	PE	3

\tilde{X}^1A_{1g}		D _{4h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a _{1g}			458	N ₂	Ra	1
			300	N ₂	Ra	1
b _{1g}			386	N ₂	Ra	1
b _{2g}			252	N ₂	Ra	1
e _g			404	N ₂	Ra	1
e _u			462.9	Ne	IR	2
			460.9	Ar	IR	2
			458.5	Kr	IR	2

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C₆⁻

Threshold for electron detachment from ground-state C₆⁻ = 33725(10) gas PE^{1,2,9}TPE³

\tilde{F}		D _{∞h}				
T ₀ = 33680T		gas PE ³				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	1		2058(5)	gas	PE	3
	2		1307(5)	gas	PE	3
	3		480(2)	gas	PE	3
Σ _u ⁺	5		837HT	gas	PE	3
Π _g ⁺	7		195HT	gas	PE	3
Π _u ⁺	8		313HT	gas	PE	3
	9		93HT	gas	PE	3

(3) ² Π _g		D _{∞h}				
T ₀ = 22517(10)		Ne AB ⁸				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	2		1861(17)	Ne	AB	8
	3		626(15)	Ne	AB	8
Π _u	9		157(7)H	Ne	AB	8

(2) ${}^2\Pi_g$ $D_{\infty h}$
 $T_0=20064(8)$ Ne AB⁸ 470–499 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	3		597(12)	Ne	AB	8
Π_g	7		237(6)H	Ne	AB	8
Π_u	9		152(6)H	Ne	AB	8

$\tilde{C}^2\Pi_g$ $D_{\infty h}$
 $T_0=16476$ gas PD⁵MPD¹¹PE¹² $\tilde{C}-\tilde{X}$ 445–629 nm
 16458(5) Ne AB⁴ $\tilde{C}-\tilde{X}$ 539–608 nm
 16239 Ar AB⁶ $\tilde{C}-\tilde{X}$ 573–616 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	C≡C stretch	2052	gas	PD	5
			2064(5)	Ne	AB	4
	2	C≡C stretch	1767	gas	PD	5
			1817(5)	Ne	AB	4
	3	C–C stretch	605	gas	PD,MPD	5,11
			607(5)	Ne	AB	4
			605T	Ar	AB	6
Π_g	8		245H	gas	PD	5
Π_u	9		110H	gas	PD	5

Internal conversion to vibrationally excited levels of the \tilde{B} and \tilde{A} states occurs in 730(50) fs gas PE¹²

$\tilde{A}^2\Sigma_g^+$ $D_{\infty h}$
 $T_0=9352(2)$ Ne AB⁸ $\tilde{A}-\tilde{X}$ 737–1070 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1		2111(4)	Ne	AB	8
	2		1912(4)	Ne	AB	8
	3		651(3)	Ne	AB	8
Π_g	7		238(2)H	Ne	AB	8
Π_u	9		128(2)H	Ne	AB	8

$\tilde{X}^2\Pi_u$ $D_{\infty h}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	2083	Ar	Ra	10
	2	Sym. stretch	1775	Ar	Ra	10
	3	C–C stretch	564	gas	PD	5
			634	Ar	Ra	10
Σ_u^+	4	Asym. stretch	1938.5	Ne	IR	7
			1936.7	Ar	IR	6,10
Π_g	7		220T	gas	PE	3
	8		201H	gas	PD	5
			234H	Ar	Ra	10
Π_u	9		111T	gas	PE	3

$A = -29(2)$ gas PE³

References

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SiC₄Si⁻

Threshold for electron detachment from ground-state SiC₄Si⁻ = 20520(50) gas PE¹

Reference

¹G. E. Davico, R. L. Schwartz, and W. C. Lineberger, J. Chem. Phys. **115**, 1789 (2001).

C₅N

${}^2\Pi$ $C_{\infty v}$
 $T_0=21259(9)$ Ne AB² ${}^2\Pi-\tilde{X}$ 427–471 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		CC,CN stretch	2122(13)	Ne	AB	2
		C–C stretch	1515(13)	Ne	AB	2
		C–C stretch	652(13)	Ne	AB	2
Π		Deformation	135(7)H	Ne	AB	2

$\tilde{X}^2\Sigma$ $C_{\infty v}$
 $B_0=0.0468$ MW¹

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NCC≡CCN⁺

$\tilde{D}^2\Pi_u$ $D_{\infty h}$
 $T_0=25500(160)$ gas PE^{1,3}

$\tilde{C}^2\Sigma_u^+$ $D_{\infty h}$
 $T_0=18720(160)$ gas PE^{1,3}

$\tilde{B}^2\Sigma_g^+$ $D_{\infty h}$
 $T_0=17430(160)$ gas PE^{1,3}

$\tilde{A}^2\Pi_g$	$D_{\infty h}$					
$T_0 = 16780.31$	gas	$EF^2LF^3AB^6$				$\tilde{A}-\tilde{X}$ 528–720 nm
16679	Ne	LF^4				$\tilde{A}-\tilde{X}$ 530–816 nm
16709	Ne	LF^4AB^5				

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ_g^+	1	C≡N stretch	2151(3)	gas	LF	3
			2154(2)	Ne	LF	4
	2	C≡C stretch	2099(3)	gas	LF	3
	3	C–C stretch	2094(2)	Ne	LF	4
			696(3)	gas	LF	3
Π		Bend	511(2)	Ne	LF	4
			440H	Ne	AB	5

$\tau = 13(2)$ ns gas EF^2PEFCO^3

$A_0 = -45(10)$ gas AB^6

$B_0 = 0.044$ AB^6

$\tilde{X}^2\Pi_u$	$D_{\infty h}$					
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ_g^+	1	C≡N stretch	2210(10)	gas	EF	2,3
			2220	Ne	LF	4
	2	C≡C stretch	1930(10)	gas	EF	2,3
	3	C–C stretch	1942	Ne	LF	4
			570(10)	gas	EF	2,3
Σ_u^+	4		613	Ne	LF	4
			2015	Ne	LF,IR	4,7
			1208 ^a	Ne	LF	4

$A_0 = -45(10)$ gas AB^6

$B_0 = 0.045$ AB^6

^aFrom overtones and combination bands.

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C₅N⁻

$\tilde{X}^1\Sigma$	$C_{\infty v}$					
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ^+			2116	Ne	IR	1

Reference

- M. Grutter, M. Wyss, and J. P. Maier, *J. Chem. Phys.* **110**, 1492 (1999).

C₅O

$\tilde{X}^1\Sigma^+$	$C_{\infty v}$	Structure: MW ¹				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
			2251.7T	Ar	IR	2

$B_0 = 0.046$ MW¹

References

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C₅S

$\tilde{X}^1\Sigma^+$	$C_{\infty v}$					
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
			2124.5	Ar	IR	2

$B_0 = 0.031$ MW¹

References

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NCCCNC

\tilde{X}	$C_{\infty v}$	Structure: MW ³				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
Σ^+	1	C≡C stretch	2295.72	gas	IR	3
			2287.1s	Ar	IR	1,2
	2	–C≡N stretch	2209.62	gas	IR	3
			2203.6s	Ar	IR	1,2
	3	–N≡C stretch	2052.98	gas	IR	3
			2044.8vs	Ar	IR	1,2
	4	C–C, C–N a-stretch	1202.3vw	Ar	IR	1,2
			610.1vwT	Ar	IR	1,2
	5	C–C, C–N s-stretch				

$B_0 = 0.047$ MW^{3,4}

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Al₃O₃ \tilde{B} $T_0 = 18560(800)$ gas PE¹ \tilde{A} $T_0 = 6450(440)$ gas PE¹ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			610(60) ^a	gas	PE	1

^aRef. 1 presents evidence for the existence of an isomer, for which the vibrational spacing in the first photoelectron band is 720(60). A more detailed assignment has been proposed by Ref. 2.

References

¹H. Wu, X. Li, X.-B. Wang, C.-F. Ding, and L.-S. Wang, *J. Chem. Phys.* **109**, 449 (1998).

²T. K. Ghanty and E. R. Davidson, *J. Phys. Chem. A* **103**, 8985 (1999).

Cr₂O₄ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			716.2	Ar	IR	1
			643.1	Ar	IR	1

Reference

¹G. V. Chertihin, W. D. Bare, and L. Andrews, *J. Chem. Phys.* **107**, 2798 (1997).

Al₃O₃⁻

Threshold for electron detachment from ground-state Al₃O₃⁻ = 22590(160) gas PE¹

Evidence for the participation of a second electronic state of Al₃O₃⁻ in its photoelectron spectrum is presented by Ref. 2.

References

¹H. Wu, X. Li, X.-B. Wang, C.-F. Ding, and L.-S. Wang, *J. Chem. Phys.* **109**, 449 (1998).

²T. K. Ghanty and E. R. Davidson, *J. Phys. Chem. A* **103**, 8985 (1999).

(SiNN)₂ \tilde{X} C_{2h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>b_u</i>		NN a-stretch	2080	N ₂	IR	1

Reference

¹G. Maier, H. P. Reisenauer, and J. Glatthaar, *Organomet.* **19**, 4775 (2000).

OCNCO

In an argon matrix, an absorption maximum at 45450 (220 nm) has been assigned¹ to OCNCO.

 \tilde{X} C_{2h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a_g</i>	1	NCO stretch	2235T	Ar	IR	1
	2	NCO stretch	1460T	Ar	IR	1
	3	NN stretch	840T	Ar	IR	1
<i>a_u</i>	6	OPLA bend	533.5w	Ar	IR	1
<i>b_u</i>	9	NCO stretch	2208	gas	IR	1
			2200.6vs	Ar	IR	1
	11	Bend	661.0w	Ar	IR	1

Reference

¹G. Maier, M. Naumann, H. P. Reisenauer, and J. Eckwert, *Angew. Chem.* **108**, 1800 (1996); *Angew. Chem. Int. Ed. Engl.* **35**, 1696 (1996).

(GeO)₃ \tilde{X} D_{3h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a₁'</i>	1		475.0	Ar	Ra	3
	2		305.0	Ar	Ra	3
<i>e'</i>	5	GeO stretch	828.6s	Ar	IR,Ra	1-3
			824	N ₂	IR	1
	6	GeO stretch	438.2w	Ar	IR,Ra	1,3
			440	N ₂	IR	1
	7		193.0	Ar	Ra	3

References

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²L. Andrews and M. McCluskey, *J. Mol. Spectrosc.* **154**, 223 (1992).

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O₂CCO₂⁺

Photodestruction of gas-phase O₂CCO₂⁺ occurs in the near infrared and visible spectral regions, with onset³ below 9400 and maximum^{1,2} near 15400 (650 nm).

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO ₂ a-stretch	2130.8	Ne	IR	4,5
			1274.4	Ne	IR	4,5
			1274.0	Ar	IR	4

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- ¹G. P. Smith, P. C. Cosby, and J. T. Moseley, *J. Chem. Phys.* **67**, 3818 (1977).
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⁵W. E. Thompson and M. E. Jacox, *J. Chem. Phys.* **111**, 4487 (1999).

S₂CCS₂⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CS ₂ a-stretch	1385.2	Ne	IR	1
			1379.7	Ar	IR	1

Reference

- ¹M. Zhou and L. Andrews, *J. Chem. Phys.* **112**, 6576 (2000).

(NO)₃⁺ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO stretch	1790T	gas	PF	1

Reference

- ¹A. Mouhandes and A. J. Stace, *J. Chem. Phys.* **111**, 9517 (1999).

O₂CCO₂⁻

Maximum in the photoelectron spectrum of (CO₂)₂⁻ near 20170 (2.5 eV) gas PE¹⁻³

 \tilde{X} D_{2d}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b ₂	5	CO ₂ s-stretch	1189.2	Ne	IR	4,6,7
			1184.7	Ar	IR	5
	6	Deformation	679.2T	Ne	IR	6,7
e	7	CO ₂ a-stretch	1852.4	Ne	IR	4,6,7
			1856.7	Ar	IR	5

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- ⁷W. E. Thompson and M. E. Jacox, *J. Chem. Phys.* **111**, 4487 (1999).

FAI(O₂)₂ \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2		892.4	Ar	IR	1
b ₂	11		705.3	Ar	IR	1

Reference

- ¹J. Bahlo, H.-J. Himmel, and H. Schnöckel, *Angew. Chem. Int. Ed.* **40**, 4696 (2001).

c-(NO)₃⁻

Maximum in the photoelectron spectrum of c-(NO)₃⁻ at 29850(160), or 3.70(2) eV gas PE¹

 \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1369.9	Ne	IR	2,3
			1001.9	Ne	IR	2
			693.8	Ne	IR	2

References

- ¹T. Tsukuda, M. Saeki, L. Zhu, and T. Nagata, *Chem. Phys. Lett.* **295**, 416 (1998).
²C. L. Lugez, W. E. Thompson, M. E. Jacox, A. Snis, and I. Panas, *J. Chem. Phys.* **110**, 10345 (1999).
³L. Andrews and M. Zhou, *J. Chem. Phys.* **111**, 6036 (1999).

N₂O₄

In the gas phase, absorption increases between 300 and 180 nm, with some indication of maxima near 265 and 190 nm.⁷

A broad gas-phase absorption with onset near 380 nm has its maximum near 340 nm.⁷

\tilde{X}		Structure: ED ¹ IR ^{10,11}					
D _{2h}							
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
<i>a_g</i>	1		1383(3)	Ne	Ra	8	
			1383	Ar	Ra	5	
			1387(3)	Xe	Ra	8	
	2		807(3)	Ne	Ra	8	
			813	Ar	Ra	5	
			815(3)	Xe	Ra	8	
	3		265(3)	Ne	Ra	8	
			262	Ar	Ra	5	
			257(3)	Xe	Ra	8	
<i>a_u</i>	4	82T	gas	IR ^a	6,12		
<i>b_{1g}</i>	5	1718(3)	Xe	Ra	8		
	6	480T	gas	IR ^a	6,12		
		498(3)	Ne	Ra	8		
		485(3)	Xe	Ra	8		
<i>b_{1u}</i>	7	425	gas	IR	6		
<i>b_{2g}</i>	8	657(3)	Xe	Ra	8		
<i>b_{2u}</i>	9	1756.76	gas	IR,DL	6,10,13–15		
		1749.2s	Ar	IR	2,3,9		
		1735s					
		1761	N ₂	IR	4		
		1737					
		1750	O ₂	IR	3		
		1735					
		265T	gas	IR	11		
		1261.08	gas	IR,DL	2,6,11,13–15		
		1257.0s	Ar	IR	2,3,9		
	12	1261	N ₂	IR	4		
		1261	O ₂	IR	3		
		755.37 ^b	gas	IR	6,13		
		747.85 ^b	gas	IR	6,13,15		
		755sh	Ar	IR	3		
		745.8					
		751	N ₂	IR	4		
		755	O ₂	IR	3		
		746					

$A_0=0.218$; $B_0=0.122$; $C_0=0.078$ IR^{10,11}DL¹⁴
Barrier to internal rotation=1900(200) gas IR¹²

^aFrom analysis of sequence bands near 540 cm⁻¹.

^bFermi resonance between ν_{12} and $\nu_6 + \nu_{10}$.

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¹⁵M. Hepp, R. Georges, M. Herman, J.-M. Flaud, and W. J. Lafferty, *J. Mol. Struct.* **517/518**, 171 (2000).

O₂PPO₂

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		PO ₂ a-stretch	1473.1	Ar	IR	1–3
		PO ₂ s-stretch	1158.1	Ar	IR	1–3
		PO ₂ deform.	479.4	Ar	IR	1,3

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C₂F₄⁺

$\tilde{H}, \tilde{I}^2B_{3g}, ^2B_{3u}$		D _{2h}				
$T_0=73020(320)$		gas PE ^{1,4} TPE ⁵				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a_g</i>	2	CF stretch	806(40)	gas	PE,TPE	1,5

\tilde{G}^2B_{2g}		D _{2h}				
$T^a=65190(320)$		gas PE ^{2–4}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a_g</i>	2	CF s-stretch	766(40)	gas	PE,TPE	1,5
	3	CF ₂ scissors	330(80)	gas	PE	1

\tilde{F}^2B_{1u}		D _{2h}				
$T_0=59460(400)$		gas PE ^{1–4} TPE ⁵				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a_g</i>	2	CF s-stretch	766(40)	gas	PE,TPE	1,5
	3	CF ₂ scissors	330(80)	gas	PE	1

$\tilde{B}, \tilde{C}, \tilde{D}, \tilde{E}^2A_g, ^2B_{2u}, ^2A_u, ^2B_{1g}$ D_{2h}
 $T^a \approx 52000$ gas PE^{1–4}

\tilde{A}^2B_{3g} D_{2h}
 $T^a=46880(320)$ gas PE^{1–4}

\tilde{X}^2B_{3u}		D _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a_g</i>	1	C=C stretch	1686(40)	gas	PE,TPE	1–3,5
	2	CF s-stretch	766(40)	gas	PE,TPE	1–3,5
	3	CF ₂ scissors	371(40)	gas	PE,TPE	2,3,5

^aFrom vertical ionization potential.

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⁵G. K. Jarvis, K. J. Boyle, C. A. Mayhew, and R. P. Tuckett, *J. Phys. Chem. A* **102**, 3230 (1998).

UOF₄

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		UO stretch	834.7	Ar	IR	1

Reference

¹P. F. Souter and L. Andrews, *J. Mol. Struct.* **412**, 161 (1997).

CF₃O₂

In the gas phase, an absorption maximum near 47620 (210 nm) has been assigned^{5,6} to CF₃O₂. In a neon matrix, the corresponding absorption maximum has been observed⁷ at 48430 (206.5 nm).

\tilde{A}^2A' C_s
T₀ = 6656(3) gas CR⁸

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2	CO stretch	1192(5)T	gas	CR	8
	3	OO stretch	980(5)	gas	CR	8

\tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CF ₃ a-stretch	1309.0vs	Ne	IR	7
			1303.9vs	Ar	IR	2-4,7
	2	CF ₃ s-stretch	1178.3ms	Ne	IR	7
			1172.5s	Ar	IR	1-4,7
	3	O-O stretch	1099.6m	Ne	IR	7
			1093.0m	Ar	IR	1-4,7
	4	CO stretch	872.7w	Ne	IR	7
			869.2w	Ar	IR	2,7
	5	CF ₃ s-deform.	694.8w	Ne	IR	7
			693.2m	Ar	IR	2-4,7
	6	CF ₃ a-deform.	495.0w	Ne	IR	7
			491.2	Ar	IR	7
	7	COO deform.	449.6vw	Ne	IR	7
			447.6vw	Ar	IR	2,7
	8	CF ₃ rock	289.6vw	Ne	IR	7
			286.8vw	Ar	IR	2,7
a''	9	CF ₃ a-stretch	1266.1s	Ne	IR	7
			1261.8vs	Ar	IR	2-4,7
	10	CF ₃ a-deform.	596.9w	Ne	IR	7
			595.2w	Ar	IR	2,4,7
	11	CF ₃ rock	402.7vw	Ne	IR	7
			401.2	Ar	IR	7

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(SO₂)₂^{-a}

Threshold for electron detachment from ground-state (SO₂)₂⁻ = 15300(1600) gas PE^{2,4}; vertical detachment energy = 22270(160) gas PE⁶

Gas-phase absorption maxima^{1,3,5} at 25000 (400 nm) and 16700 (600 nm) result from the photodissociation of (SO₂)₂⁻ into SO₂ + SO₂⁻.

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		SO stretch	1267.4	Ne	IR	7
			1265.9			
		SO stretch	1256.6	Ne	IR	7
			1256.1			
			1015.7	Ne	IR	7
			622.7	Ne	IR	7

^aTwo or more nearly isoenergetic structures may contribute to the observed spectrum.

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SF₅⁺

\tilde{X} D_{3h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₂ ⁺	3	SF stretch	1008.2vs	Ne	IR	1
	4	Deformation	633.2m	Ne	IR	1
e ⁺	5	SF stretch	1074.8vs	Ne	IR	1
	6	Deformation	579.6m	Ne	IR	1

Reference

¹C. L. Lugez, M. E. Jacox, R. A. King, and H. F. Schaefer III, *J. Chem. Phys.* **108**, 9639 (1998).

PF₅⁻

\tilde{X}		C _{4v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	PF stretch	815.6ms	Ne	IR	1
	3		455.9mT	Ne	IR	1
e	7	PF stretch	735.0vs	Ne	IR	1

Reference

¹C. L. Lugez, K. K. Irikura, and M. E. Jacox, *J. Chem. Phys.* **108**, 8381 (1998).

SF₅

\tilde{X}		C _{4v}	Structure: ESR ^{1,2}			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	Eq. s-stretch	884.5	Ar	IR	5
	3	SF stretch	553.8	Ne	IR	6
e	7		552m	Ar	IR	3,4
		SF a-stretch	817.0	Ne	IR	6
			813.1vs	Ar	IR	3-5

References

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SF₅⁻

\tilde{X}		C _{4v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	SF stretch	795.8	Ne	IR	4
			795.5m	Ar	IR	2,3
e	7	SF stretch	602.9	Ne	IR	4
			596s ^a	Ar	IR	1,2
			8	Deformation	470.5	Ne
470w	Ar	IR			2	

^aThis absorption was attributed to SF₆⁻ by Ref. 1. Reassignment to SF₅⁻ is dictated by the close correspondence, discussed in Ref. 2, of these three absorptions to peaks observed for solid CsSF₅.

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8.13. Seven-Atomic Molecules

Pd(H₂)₃

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		HH stretch	2909	Ar	IR	1
		PdH stretch	745.7	Ne	IR	1
			730.1	Ar	IR	1
		Deformation	319.5	Ar	IR	1

Pd(D₂)₃

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		DD stretch	2070	Ar	IR	1
		PdD stretch	560.6	Ne	IR	1
			549.4	Ar	IR	1
		Deformation	236.6	Ar	IR	1

Reference

¹L. Andrews, X. Wang, M. E. Alikhani, and L. Manceron, *J. Phys. Chem. A* **105**, 3052 (2001).

CH₃GaH₂

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	3	GaH ₂ s-stretch	1898.0	Ar	IR	1
a''	11	GaH ₂ a-stretch	1892.0	Ar	IR	1

Reference

¹J. Müller, H. Sternkicker, U. Bergmann, and B. Atakan, *J. Phys. Chem. A* **104**, 3627 (2000).

C₂H₅**3p Rydberg state**

In the gas phase, an absorption with maximum at 205 nm has been assigned^{6,7} to the $3p-\tilde{X}$ transition of C₂H₅. The detection of this band in MPI studies⁹ is consistent with the assignment to a Rydberg state.

3s Rydberg state

In the gas phase, a broad, unstructured absorption with maximum at 246 nm has been assigned^{1,3,6,7} to the $3s-\tilde{X}$ transition of C₂H₅.

\tilde{X}		C _s	Structure: MO ⁴			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	CH ₂ s-stretch	3037.02	gas	LD	13	
		3033m	Ar	IR	2,4,5,8	
		3032.6	H ₂	IR	11	
	CH ₃ s-stretch	2920m	Ar	IR	4,5	
		2842s	Ar	IR	2,4,5,8	
		1440m	Ar	IR	2,4,5,8	
	CH ₂ deform.	1383	Ar	IR	8	
		1366m	Ar	IR	2,4,5,8	
	CC stretch	1138w	Ar	IR	4,5,8	
		1025	Ar	IR	8	
CCH ₂ umbrella	528.12	gas	DL	10,12		
	540vs	Ar	IR	2,4,5,8		
<i>a''</i>	CH ₂ a-stretch	3128.69	gas	LD	13	
		3112s	Ar	IR	2,4,5,8	
	CH ₃ a-stretch	3122.8	H ₂	IR	11	
		2987s	Ar	IR	2,4,5	
	CH ₃ deform.	2984.3	H ₂	IR	11	
		1440m	Ar	IR	2,4,5,8	
	H deform.	1175m	Ar	IR	4,5,8	

$$A_0 = 9.629; (B_0 + C_0)/2 = 0.730; (B_0 - C_0)/4 = 0.014 \quad \text{DL}^{12}$$

C₂D₅

\tilde{X}		C _s	Structure: MO ⁴			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	CD ₂ s-stretch	2199mT	Ar	IR	4,5	
		2094m	Ar	IR	4,5	
	CD ₃ s-stretch	2092.2	H ₂	IR	14	
		2048m	Ar	IR	4,5	
	2-CD stretch	2050.0	H ₂	IR	14	
		1070mT	Ar	IR	4,5	
	CD ₃ deform.	1035mT	Ar	IR	4,5	
		398vs	Ar	IR	4,5	
	CCD ₂ umbrella	2249mT	Ar	IR	4,5	
		2170sT	Ar	IR	4,5	
<i>a''</i>	CD ₂ a-stretch	1041mT	Ar	IR	4,5	
		2350T	H ₂	IR	14	
	CD ₃ a-stretch	2343T	H ₂	IR	14	
		2060.7T	H ₂	IR	14	
	CD ₃ deform.	1188.7	H ₂	IR	14	

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cyc-BHCHCH₂

\tilde{X}		C ₁	Structure: MO ⁴			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			2621.6	Ar	IR	1
			864.3	Ar	IR	1
			637.6	Ar	IR	1

cyc-BDCDCD₂

\tilde{X}		C ₁	Structure: MO ⁴			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1962.2	Ar	IR	1
			650.9T	Ar	IR	1
			530.4	Ar	IR	1

Reference

- L. Andrews, D. V. Lanzisera, P. Hassanzadeh, and Y. Hannachi, J. Phys. Chem. A **102**, 3259 (1998).

H₂BCCH₂

\tilde{X}		C ₂	Structure: MO ⁴				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
			BH ₂ a-stretch	2568.7	Ar	IR	1
			C=C stretch	1770.6	Ar	IR	1
				1347.1	Ar	IR	1
				847.7w	Ar	IR	1
				826.5	Ar	IR	1

D₂BCCD₂

\tilde{X}		C ₂	Structure: MO ⁴				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
			BD ₂ a-stretch	1933.5	Ar	IR	1
			C=C stretch	1732.1	Ar	IR	1
				650.9T	Ar	IR	1

Reference

¹L. Andrews, D. V. Lanzisera, P. Hassanzadeh, and Y. Hannachi, *J. Phys. Chem. A* **102**, 3259 (1998).

CH₃CCH⁺

\tilde{C}
 $T^a = 55100(800)$ gas PE¹

\tilde{B}
 $T^a = 39000(800)$ gas PE¹

\tilde{A}
 $T_0 = 26790(100)$ C_{3v} gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	4	CH ₃ deform.	1290(40)	gas	PE	1

\tilde{X}^2E C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1	CH stretch	3226.7T	Ne	IR	3
	2	CH stretch	2767T	gas	PI	2
	3	CC stretch	1996(38)	gas	PE,PI	2
	5	CC stretch	940(40)	gas	PE	1
e	9	CCH deform.	586(74)	gas	PI	2
	10	Deformation	289(65)	gas	PI	2

^aFrom vertical ionization potential.

References

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H₂CCCH⁺

\tilde{B}
 $T^a = 61400(800)$ gas PE¹

\tilde{A}^2E D₂
 $T_0 = 35638(32)$ gas PE¹⁻⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1320(6)	gas	PE	3,4
			1030(6)	gas	PE	3,4

\tilde{X}^2E D₂ Structure: PE^{2,3}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH ₂ stretch	3047.3wm	Ne	IR	5
		CH ₂ stretch	2956.0m	Ne	IR	5
		CH ₂ deform.	1314.4s	Ne	IR	5
			880.4m	Ne	IR	5
b ₁	4	Torsion	745(5)	gas	PE	3,4
			738.4wm	Ne	IR	5

D₂CCCD₂⁺

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CD ₂ stretch	2295.5wm	Ne	IR	5
		CD ₂ stretch	2177.2wm	Ne	IR	5
			2169.6wm			
		CD ₂ deform.	962.4m	Ne	IR	5
			724.6wm	Ne	IR	5
b ₁	4	Torsion	565(50)	gas	PE	1
			583.6wm	Ne	IR	5

^aFrom vertical ionization potential.

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cyc-CH₂CH₂Si

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	CH ₂ deform.	1417.1w	Ar	IR	1
	3	CC stretch	1011.9wm	Ar	IR	1
	4	CH ₂ deform.	869.6w	Ar	IR	1
	5	Ring deform.	601.4wm	Ar	IR	1
b ₁	9	CH stretch	3032.0wm	Ar	IR	1
	10	CH ₂ deform.	719.4w	Ar	IR	1
b ₂	12	CH stretch	2966.3wm	Ar	IR	1
	13	CH ₂ deform.	1376.7w	Ar	IR	1
	14	CH ₂ deform.	919.4wm	Ar	IR	1
	15	Ring deform.	592.1vs	Ar	IR	1

Reference

- ¹G. Maier, H. P. Reisenauer, and H. Egenolf, *Eur. J. Org. Chem.* **1998**, 1313.

H₂CCHCH:

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH ₂ wag	786.6T	Xe	IR	1

Reference

- ¹G. Maier, C. Lautz, and S. Senger, *Chem. Eur. J.* **6**, 1467 (2000).

H₂CCHSiH

\bar{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	1	CH ₂ stretch	3056.1w	Ar	IR	1
	2	CH stretch	2997.0w	Ar	IR	1
	3	CH stretch	2976.5w	Ar	IR	1
	4	SiH stretch	1979.4vs	Ar	IR	1
	5	Mixed	1566.9w	Ar	IR	1
	6	CH ₂ deform.	1388.2wm	Ar	IR	1
	8	CH ₂ deform.	1004.8w	Ar	IR	1
	9	Deform.	812.7wm	Ar	IR	1
	10	Deform.	665.2wm	Ar	IR	1
	<i>a''</i>	12	CH deform.	1009.3w	Ar	IR
13		CH ₂ deform.	985.2wm	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, and H. Egenolf, *Eur. J. Org. Chem.* **1998**, 1313.

H₂CCSiH₂

\bar{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CH stretch	3008.2	Ar	IR	1
	2	SiH stretch	2205.9	Ar	IR	1
	4	CH ₂ deform.	1371.0w	Ar	IR	1
	5	SiH ₂ deform.	930.3vs	Ar	IR	1
	8	CH stretch	3069.1	Ar	IR	1
<i>b</i> ₂	12	SiH stretch	2221.6	Ar	IR	1
	13	CH ₂ deform.	917.6vs	Ar	IR	1
	14	CSiH ₂ deform.	656.2vs	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, and H. Egenolf, *Eur. J. Org. Chem.* **1998**, 1313.

CH₃BNH

\bar{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NH stretch	3690.5	Ar	IR	1
		BN stretch	1940.2	Ar	IR	1

Reference

¹D. V. Lanzisera and L. Andrews, *J. Phys. Chem. A* **101**, 824 (1997).

CH₃NBH

\bar{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NB stretch	1949.2	Ar	IR	1

Reference

¹D. V. Lanzisera and L. Andrews, *J. Phys. Chem. A* **101**, 824 (1997).

CH₂BNH₂

\bar{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		BN stretch	1743.5	Ar	IR	1

Reference

¹D. V. Lanzisera and L. Andrews, *J. Phys. Chem. A* **101**, 824 (1997).

CH₃CNH⁺

\bar{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NH stretch	3527.29	gas	LD	1,2

$B_0 = 0.287 \text{ LD}^{1,2} \text{ MW}^3$

References

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CH₃CHO⁺

\bar{E}		C_s				
$T_0 = 47930(400)$		gas	PE ^{1,2}			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>		CO stretch	1200(40)	gas	PE	2

\bar{C}, \bar{D}
 $T_0 = 37840(400)$ gas C_s PE^{1,2}

\bar{B}
 $T_0 = 28160(400)$ gas C_s PE^{1,2}

\tilde{A} C_s
 $T_0 = 18560(240)$ gas PE^{1,2}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		CO stretch	1210(40)	gas	PE	1,2
		CCO deform.	440(40)	gas	PE	1,2

\tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		CH stretch	2815(10)	gas	PE	3
		CH stretch	2570(40)	gas	PE	2
			1428(10)	gas	PE	3
		CH ₃ deform.	1315(10)	gas	PE	1-3
		CH deform.	1100(40)	gas	PE	1,2
			1032(10)	gas	PE	3
a''	10		823(10)	gas	PE	3
	14	CH ₃ rock	363(10)	gas	PE	3
	15	CH ₃ rock	742(10)	gas	PE	1-3
		Torsion	145(10)	gas	PE	3

CD₃CDO⁺

\tilde{A} C_s
 $T_0 = 18560T$ gas PE²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		CO stretch	1210(40)	gas	PE	2
		CCO bend	440(40)	gas	PE	2

\tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		CH stretch	1890(40)	gas	PE	2
		CO stretch	1550(40)	gas	PE	2
			960(40)	gas	PE	2
			830(40)	gas	PE	2
		CCO deform.	320(40)	gas	PE	2

References

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³H.-T. Kim and S. L. Anderson, J. Chem. Phys. **114**, 3018 (2001).

syn-CH₂=CHOH

\tilde{X} C_s Structure: MW^{1,3}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	OH stretch	3633.5m	gas	IR	5,6
			3620s	Ar	IR	2,4
	2	CH stretch	3121.7(2)	gas	IR	5
	5		C=C stretch ^a	1663m	gas	IR
			1626m			
			1662vs	Ar	IR	2,4
			1622s			
	6		1411.8vw	gas	IR	5,6
			1409.4	Ar	IR	4
	7		1300.0vw	gas	IR	5,6
			1326w	Ar	IR	4
	8		1259.7m	gas	IR	5,6
			1300.2m	Ar	IR	4
	9	CO stretch+ OH deform. ^b	1118vs	gas	IR	5,6
			1078vs			
		1121s	Ar	IR	2,4	
		1079vs				
10		947.6w	gas	IR	5	
		943.4w	Ar	IR	4	
11		486.1w	Ar	IR	4	
a''	12		960m	gas	IR	6
			971.4m	Ar	IR	4
13	H ₂ C=C OPLA	816.66s	gas	IR	5,6	
		813.7s	Ar	IR	2,4	
		698.9w	gas	IR	5	
		698vw	Ar	IR	4	
15	Torsion	388H	gas	IR	6	
		413s	Ar	IR	2,4	

$A_0 = 1.990$; $B_0 = 0.352$; $C_0 = 0.299$ MW^{1,3}IR⁶

syn-CD₂=CDOD

\tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		OD stretch	2677	Ar	IR	2
		C=C stretch	1590	Ar	IR	2
			1584			
		CO stretch+	926	Ar	IR	2
		OD deform.	922			
a''		D ₂ C=C OPLA	651	Ar	IR	2
		Torsion	310	Ar	IR	2

$A_0 = 1.342$; $B_0 = 0.312$; $C_0 = 0.253$ MW³

^aFermi resonance between ν_5 and $2\nu_{13}$.

^bFermi resonance between ν_9 and $(\nu_{14} + \nu_{15})$.

References

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anti-CH₂=CHOH

Relative intensities of microwave lines for the two isomers of CH₂=OH indicate that the *anti*-isomer lies 380(50), or 4.5(6) kJ/mol, above the *syn*-isomer.¹

\tilde{X} C_s
A₀=2.097; B₀=0.349; C₀=0.299 MW¹

Reference

¹M. Rodler, J. Mol. Spectrosc. **114**, 23 (1985).

***t*-CH₃OSiH**

\tilde{X}		C _s					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
<i>a'</i>	3	SiH stretch	1929.1vs 1964.7vs	Ar	IR	1	
	4	CH ₂ deform.	1473.6vw	Ar	IR	1	
	5	CH ₃ deform.	1443.9w	Ar	IR	1	
	6	CH ₃ deform.	1186.2wm	Ar	IR	1	
	7	Mixed	1084.8s	Ar	IR	1	
	8	SiH deform.	859.4ms	Ar	IR	1	
	9	SiO stretch	749.3wm	Ar	IR	1	
	<i>a''</i>	12	CH ₂ deform.	1457.6w	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, and H. Egenolf, Monatsh. Chem. **130**, 227 (1999).

***c*-CH₃OSiH**

\tilde{X}		C _s					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
<i>a'</i>	3	SiH stretch	1849.6vs 1910.7vs	Ar	IR	1	
	4	CH ₂ deform.	1461.3w	Ar	IR	1	
	5	CH ₃ deform.	1438.7w	Ar	IR	1	
	6	CH ₃ deform.	1174.7w	Ar	IR	1	
	7	Mixed	1096.2s	Ar	IR	1	
	8	SiH deform.	881.4w	Ar	IR	1	
	9	SiO stretch	741.8m	Ar	IR	1	
	<i>a''</i>	12	CH ₂ deform.	1452.6w	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, and H. Egenolf, Monatsh. Chem. **130**, 227 (1999).

CH₃CNO

\tilde{X}		C _{3v}		Structure: MW ^{1,2}		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	CH ₃ s-stretch	2943w	gas	IR	5
	2	CNO a-stretch	2311vs 2309	gas Ar	IR	5 4
	3	CH ₃ s-deform.	1394 1381	gas Ar	IR	5 4
	4	CNO s-stretch	1348 1332	gas Ar	IR	5 4
	5	CC stretch	785 780	gas Ar	IR	5 4
<i>e</i>	6	CH ₃ a-stretch	3022vwT	gas	IR	5
	7	CH ₃ a-deform.	1453vw	gas	IR	5
	8	CH ₃ rock	1034vwT	gas	IR	5
	9	CNO bend	477vw	gas	IR	5

B₀=0.131 MW¹⁻³

CD₃CNO

\tilde{X}		C _{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CNO a-stretch	2297	Ar	IR	4
		CNO s-stretch	1341	Ar	IR	4

References

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HC₅H

${}^3\Sigma_u^-$		D _{∞h}		Structure: MW ¹		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	2	C≡C s-stretch	1817(5)	Ne	AB	1
	3		751(5)	Ne	AB	1

T₀=23033(5) Ne AB¹ ³Σ_u⁻- \tilde{X} 390–435 nm

Reference

¹J. Fulara, P. Freivogel, D. Forney, and J. P. Maier, J. Chem. Phys. **103**, 8805 (1995).

(*cyc*-HC=CHC)=C=C:

\tilde{X} C_{2v}
A₀=1.067; B₀=0.118; C₀=0.106 MW¹

(cyc-DC=CDC)=C=C:
 \tilde{X} C_{2v}
 $A_0=0.808; B_0=0.112; C_0=0.098$ MW¹
Reference
¹C. A. Gottlieb, M. C. McCarthy, V. D. Gordon, J. M. Chakan, A. J. Ap-
 poni, and P. Thaddeus, *Astrophys. J.* **509**, L141 (1998).
HCCCH=C=C:
 \tilde{X} C_s
 $A_0=1.061; B_0=0.096; C_0=0.088$ MW¹
DCCCD=C=C:
 \tilde{X} C_s
 $A_0=0.844; B_0=0.090; C_0=0.081$ MW¹
Reference
¹C. A. Gottlieb, M. C. McCarthy, V. D. Gordon, J. M. Chakan, A. J. Ap-
 poni, and P. Thaddeus, *Astrophys. J.* **509**, L141 (1998).
H₂C₅:
 \tilde{X} C_{2v}
 $A_0=9.26; B_0=0.077; C_0=0.076$ MW¹
Reference
¹M. C. McCarthy, M. J. Travers, A. Kovács, W. Chen, S. E. Novick, C. A.
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(cyc-HC₃)CCH
 \tilde{X} C_s
 $A_0=1.155; B_0=0.114; C_0=0.104$ MW¹
(cyc-DC₃)CCD
 \tilde{X} C_s
 $A_0=0.994; B_0=0.104; C_0=0.094$ MW¹
Reference
¹M. J. Travers, M. C. McCarthy, C. A. Gottlieb, and P. Thaddeus, *Astro-
 phys. J.* **483**, L135 (1997).
HSiCCCH
 \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>		CH stretch	3328.0w	Ar	IR	1
		CC stretch	2155.7m	Ar	IR	1
		SiH stretch	2029.1vs	Ar	IR	1
		CC stretch	2016.4s	Ar	IR	1
		CC stretch	999.6w	Ar	IR	1
		SiH deform.	806Tvs	Ar	IR	1
<i>a''</i>		SiC stretch	466.0w	Ar	IR	1
		CCC deform.	593.7w	Ar	IR	1

DSiCCCD
 \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>		CD stretch	2591.4m	Ar	IR	1
		CC stretch	2118.2vs	Ar	IR	1
		CC stretch	1936.3vw	Ar	IR	1
		SiD stretch	1474.2s	Ar	IR	1
		CC stretch	985.1w	Ar	IR	1
		SiH deform.	628Ts	Ar	IR	1
		SiC stretch	456.5vw	Ar	IR	1

Reference
¹G. Maier, H. P. Reisenauer, and A. Meudt, *Eur. J. Org. Chem.* **1998**, 1285.
Si(CCH)₂
 \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁		CH stretch	3312.9s	Ar	IR	1
			3296.7			
		CC stretch	2012.5w	Ar	IR	1
<i>b</i> ₁			2004.3			
		SiC stretch	605.9w	Ar	IR	1
			602.4			
<i>b</i> ₂		CH deform.	630.0vs	Ar	IR	1
			618.9			
<i>b</i> ₂		CH stretch	3312.9s	Ar	IR	1
			3296.7			
		CC stretch	2007.3s	Ar	IR	1
			1999.5			
<i>b</i> ₂		CH deform.	621.6m	Ar	IR	1
			618.9			

Si(CCD)₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CD stretch	2578.4w 2567.2	Ar	IR	1
		CC stretch	1894.2w 1885.8	Ar	IR	1
		SiC stretch	597.9s	Ar	IR	1
b_2		CD stretch	2578.4w 2567.2	Ar	IR	1
		CC stretch	1888.9vs 1861.1	Ar	IR	1
			612.6vs	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, and A. Meudt, Eur. J. Org. Chem. **1998**, 1285.

H₂CCCCN

\tilde{X}
 $B_{\text{eff}}=0.073$ C_{2v}
MW¹

Reference

¹W. Chen, M. C. McCarthy, M. J. Travers, E. W. Gottlieb, M. R. Munrow, S. E. Novick, C. A. Gottlieb, and P. Thaddeus, Astrophys. J. **492**, 849 (1998).

cyc-(HC=CHN=C=N)

In an argon matrix,¹ irradiation of cyc-(CH=CHN=C=N) at wavelengths longer than 570 nm leads to the formation of (cyc-HC=CHN)CN.

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	4	CH deform.	1077.8w	Ar	IR	1
	5	NCN s-stretch	979.2s 967.7s	Ar	IR	1
b_1	6	NCN deform.	718.1m	Ar	IR	1
	9	OPLA	672.9s 670.0s	Ar	IR	1
b_2	12	NCN a-stretch	1707.3vs 1684.7vs	Ar	IR	1
	13	CH a-deform.	1207.3m 1196.7m	Ar	IR	1

Reference

¹G. Maier and J. Endres, Chem. Eur. J. **5**, 1590 (1999).

(cyc-HC=CHN)CN

In an argon matrix,¹ a prominent absorption at 41000 (244 nm) has been assigned to (cyc-HC=CHN)CN. Exposure of the deposit to 254-nm radiation leads to the formation of H₂C=C=NCN.

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CH s-stretch	3235.1w	Ar	IR	1
			3233.6w	N ₂	IR	1
	2	CN stretch	2207.4s	Ar	IR	1
			2217.6vs 2207.4vs	N ₂	IR	1
	3	C=C stretch	1712.6w	Ar	IR	1
			1714.1w	N ₂	IR	1
	4	N-C stretch	1051.8m	Ar	IR	1
			1049.0s	N ₂	IR	1
5	CH deform.	960.0w	Ar	IR	1	
		961.8w	N ₂	IR	1	
6	Mixed	803.2w	Ar	IR	1	
		803.2w	N ₂	IR	1	
7	Mixed	613.6m	Ar	IR	1	
		618.8w	N ₂	IR	1	
8	Mixed	562.8m	Ar	IR	1	
		580.7m 571.3m	N ₂	IR	1	
a''	10	CH a-stretch	3185.7s	Ar	IR	1
			3183.1s			
			3183.7s	N ₂	IR	1
			3181.6s			
11	CH deform.	958.0w	Ar	IR	1	
		636.7vw	Ar	IR	1	
13	NCN deform.	635.0w	N ₂	IR	1	

Reference

¹G. Maier and J. Endres, Chem. Eur. J. **5**, 1590 (1999).

H₂C=C=NCN

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CH s-stretch	3044.7w	Ar	IR	1
			3035.4w	N ₂	IR	1
2	CN stretch	2267.0m	Ar	IR	1	
		2249.7m 2232.8m 2253.8m 2235.9m	N ₂	IR	1	
3	CCN a-stretch	2048.9vs	Ar	IR	1	
		2041.8vs	N ₂	IR	1	
5	Mixed	1222.0vw	Ar	IR	1	
		1221.6w	N ₂	IR	1	
6	Mixed	857.9vw	Ar	IR	1	
		862.5vw	N ₂	IR	1	
7	CH ₂ wag	733.5m	Ar	IR	1	
		749.9m 747.3m	N ₂	IR	1	
8	Skel. deform.	596.3vw	Ar	IR	1	
		601.4vw	N ₂	IR	1	
a''	11	CH a-stretch	3136.4vw	Ar	IR	1
			3123.3vw	N ₂	IR	1

Reference

¹G. Maier and J. Endres, Chem. Eur. J. 5, 1590 (1999).

cyc-(C=NCHNCH)

\tilde{X}^3A'' C _s						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	3	Mixed	1452.0s	Ar	IR	1
	4	Mixed	1358.9s	Ar	IR	1
	5	CH a-deform.	1250.6w	Ar	IR	1
	6	CH s-deform.	1210.3m	Ar	IR	1
	7	NCN deform.	1114.0s	Ar	IR	1
	8	Mixed	988.6s	Ar	IR	1
			981.1s			
	9	C-C: stretch	971.9m	Ar	IR	1
	10	NCN deform.	864.1w	Ar	IR	1
a''	13	CH deform.	810.0m	Ar	IR	1
	14	Ring deform.	516.1m	Ar	IR	1

Reference

¹G. Maier and J. Endres, Eur. J. Org. Chem. 2535 (2000).

HC-N=CHCN

In an argon matrix, an absorption maximum at 36100 (277 nm) can be assigned to HC-N=CHCN.¹

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	3	CN stretch	2216.1w	Ar	IR	1
			2216.3	N ₂	IR	1
	4	NC stretch	1980.9s	Ar	IR	1
			1973.2s			
			1985.0s	N ₂	IR	1
			1975.4s			
	8	CH deform.	848.0w	Ar	IR	1
			847.3w	N ₂	IR	1
	9	:CH deform.	730.7vs	Ar	IR	1
			713.2vs			
			735.7vs	N ₂	IR	1
			713.9vs			
	10	:CH deform.	575.5w	Ar	IR	1
			576.3w	N ₂	IR	1
	12	Skel. deform.	442.8m	Ar	IR	1
			449.3m	N ₂	IR	1

Reference

¹G. Maier and J. Endres, Eur. J. Org. Chem. 2535 (2000).

cyc-(HC=NCH)CN

In an argon matrix, an absorption maximum at 38900 (257 nm) can be assigned to cyc-(HC=NCH)CN.¹

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	4	C=N stretch	1676.2m	Ar	IR	1
			1671.5m	N ₂	IR	1
	6	Mixed	1214.2m	Ar	IR	1
			1209.4m	N ₂	IR	1
	7	=CH deform.	1025.1w	Ar	IR	1
			1028.1w	N ₂	IR	1
	8	CH deform.	977.1m	Ar	IR	1
			980.7m	N ₂	IR	1
	9	CH deform.	917.7s	Ar	IR	1
			919.4s	N ₂	IR	1
	10	=CH deform.	768.0m	Ar	IR	1
			750.7m	N ₂	IR	1
	11	Ring deform.	708.4w	Ar	IR	1
	12	Mixed	561.3vw	Ar	IR	1
			562.8vw	N ₂	IR	1
	13	CCN deform.	533.9vw	Ar	IR	1
			535.1vw	N ₂	IR	1

Reference

¹G. Maier and J. Endres, Eur. J. Org. Chem. 2535 (2000).

HNC=C=CHCN \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NH stretch	3331.6w	Ar	IR	1
	3	CN stretch	2238.9w	Ar	IR	1
	4	CCN stretch	2070.4vs	Ar	IR	1
	7	Mixed	950.5vw	Ar	IR	1
	8	NH deform.	890.7s	Ar	IR	1
	9	Mixed	830.9vw	Ar	IR	1
	11	Mixed	598.0vw	Ar	IR	1

Reference

¹G. Maier and J. Endres, Eur. J. Org. Chem. 2535 (2000).

C₆H $\tilde{B}^2\Pi$

C_{∞v}
 $T_0 = 18990.1(3)$ gas CR^{8,10} $\tilde{B}-\tilde{X}$ 484–528 nm
 18854(5) Ne AB⁷ $\tilde{B}-\tilde{X}$ 444–530 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	3	C≡C stretch	2076(5)	Ne	AB	7
	5	C≡C stretch	1630.8	gas	CR	8
			1638(5)	Ne	AB	7
	6	C-C stretch	659.5	gas	CR	8
			658(5)	Ne	AB	7

$A_{\text{eff}} = -23.69$ gas CR^{10,11}
 $B_0 = 0.046$ CR^{8,10,11}

$\tilde{A}^2\Sigma^+$ $C_{\infty v}$
 $T_0 = 1460(50)$ gas PE⁹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	2	C≡C stretch	2202(70)	gas	PE	9

$\tilde{X}^2\Pi$ $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		C≡C stretch	1962.2T 1953.4	Ne Ar	IR IR	7 6

$A_{\text{eff}} = -15.11$ gas MW^{3-5,11,12}
 $B_0 = 0.046$ MW^{1-5,11,12}

C₆D

$\tilde{B}^2\Pi$ $C_{\infty v}$
 $T_0 = 19041.26$ gas CR^{8,10,11} $\tilde{B} - \tilde{X}$ 524–526 nm
 18900(10) Ne AB⁸
 $A_{\text{eff}} = -24.08$ gas CR¹¹
 $B_0 = 0.044$ CR¹¹

$\tilde{A}^2\Sigma^+$ $C_{\infty v}$
 $T_0 = 1452(50)$ gas PE⁹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	2	C≡C stretch	2202(70)	gas	PE	9

$\tilde{X}^2\Pi$ $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		C≡C stretch	1862.4	Ar	IR	6

$A_{\text{eff}} = -15.13$ MW¹¹
 $B_0 = 0.044$ MW¹¹

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SiC₅H

$\tilde{X}^2\Pi$ $C_{\infty v}$
 $B = 0.029$ MW¹

Reference

- M. C. McCarthy, A. J. Apponi, C. A. Gottlieb, and P. Thaddeus, *J. Chem. Phys.* **115**, 870 (2001).

H(C≡C)₂CN⁺

$\tilde{C}^2\Pi$ $C_{\infty v}$
 $T_0 = 27350(160)$ gas PE¹

$\tilde{B}^2\Sigma^+$ $C_{\infty v}$
 $T_0 = 21860(160)$ gas PE¹

$\tilde{A}^2\Pi$ $C_{\infty v}$
 $T_0 = 17178.04$ gas EF¹AB⁴ $\tilde{A} - \tilde{X}$ 580–670 nm
 17098(20) Ne AB²LF³ $\tilde{A} - \tilde{X}$ 494–820 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+			1870(160)	gas	EF	1
	3	C≡C stretch	2006(5)	Ne	AB	2
	4	CN stretch	1849(5)	Ne	AB	2
	5	C–C stretch	1187(5)	Ne	AB	2
	6	C–C stretch	600(160) 584(5)	gas Ne	EF AB	1 2

$\tau_0 = 15(2)$ ns gas EF¹
 $A_0 = -32.58(41)$ gas AB⁴
 $B_0 = 0.044$ AB⁴

$\tilde{X}^2\Pi$ $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1	CH stretch	3403.4	Ne	LF	3
	2	C≡N stretch	2190(10) 2194.6	gas Ne	EF LF	1 3
	3	C≡C stretch	2070(10)	gas	EF	1
	4	C≡C stretch	2070.2	Ne	LF	3
	5	C–C stretch	1912.7 1220(10) 1213	Ne gas Ne	LF EF LF	3 1 3
	6	C–C stretch	630(10) 625	gas Ne	EF LF	1 3

$A_0 = -35.71(41)$ gas AB⁴
 $B_0 = 0.045$ AB⁴

D(C≡C)₂CN⁺

$\tilde{A}^2\Pi$ $C_{\infty v}$
 $T_0 = 17188.50$ gas AB⁴
 $A_0 = -33.38(57)$ gas AB⁴
 $B_0 = 0.042$ AB⁴

$\tilde{X}^2\Pi$ $C_{\infty v}$
 $A_0 = -36.47(57)$ AB⁴
 $B_0 = 0.043$ AB⁴

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⁴W. E. Sinclair, D. Pfluger, and J. P. Maier, J. Chem. Phys. **111**, 9600 (1999).

C₆H⁻

Threshold for electron detachment from ground-state
 C₆H⁻ = 30730(120) gas PE¹

C₆D⁻

Threshold for electron detachment from ground-state
 C₆D⁻ = 30700(120) gas PE¹

Reference

- ¹T. R. Taylor, C. Xu, and D. M. Neumark, J. Chem. Phys. **108**, 10018 (1998).

H(C≡C)₂NC

\tilde{X} C_{∞v}
 B₀ = 0.0467 MW¹

Reference

- ¹P. Botschwina, A. Heyl, W. Chen, M. C. McCarthy, J.-U. Grabow, M. J. Travers, and P. Thaddeus, J. Chem. Phys. **109**, 3108 (1998).

C₇

$^1\Sigma_u^+$ D_{∞h}
 T₀ = 39556(30)T Ne AB⁷ $^1\Sigma_u^+ - \tilde{X}$ 246–253 nm
 40470 Ar AB⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			979(30)	Ne	AB	7
			428(30)	Ne	AB	7

T₀ = 32765(22)T Ne AB⁷ 246–306 nm
 34360T Ar AB⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1256(25)	Ne	AB	7
			695(25)	Ne	AB	7

$^1\Pi_u$ D_{∞h}
 T₀ = 18440(7) Ne AB⁷ $^1\Pi_u - \tilde{X}$ 485–543 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			2170(10)	Ne	AB	7
			1873(10)	Ne	AB	7
			568(10)	Ne	AB	7
			478(10)	Ne	AB	7

\tilde{a} D_{∞h}
 T^a = 13800(80) gas PE¹¹

$\tilde{X}^1\Sigma_g^+$ D_{∞h} Structure: DL¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	3	Sym. stretch	548(90)	gas	PE	5
			582T	Ar	IR	9
Σ_u^+	4	Asym. stretch	2138.315	gas	DL	1,2
			2134.6	Ne	IR	6,7,10
			2127.8	Ar	IR	8
			2120.4	Kr	IR	9
	5	Asym. stretch	1898.376	gas	DL	3
			1897.5	Ne	IR	6,7,10
			1894.3	Ar	IR	8
			1889.3	Kr	IR	9
Π_u	7	Bend	496(110)T	gas	PE	5

B₀ = 0.031 DL¹⁻³

^aFrom vertical ionization potentials.

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⁴J. Szczepanski and M. Vala, J. Phys. Chem. **95**, 2792 (1991).
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⁹J. Szczepanski, S. Ekern, C. Chapo, and M. Vala, Chem. Phys. **211**, 359 (1996).
¹⁰S. Tam, M. Macler, and M. E. Fajardo, J. Chem. Phys. **106**, 8955 (1997).
¹¹M. Kohno, S. Suzuki, H. Shiromaru, and Y. Achiba, J. Chem. Phys. **110**, 3781 (1999).

C₆Si

\tilde{X} C_{∞v}
 B₀ = 0.020 MW^{1,2} Structure: MW²

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²V. D. Gordon, E. S. Nathan, A. J. Apponi, M. C. McCarthy, and P. Thaddeus, J. Chem. Phys. **113**, 5311 (2000).

Si₇

\tilde{A}
 $T_0 = 9200(400)$ gas PE⁴

		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'_1	1		435	N ₂	Ra	1
	2		385(20)	gas	PE	3,4
e'_1			358	N ₂	Ra	1
			422.4	Ar	IR	2
e'_2			420.4	Kr	IR	2
			340	N ₂	Ra	1
			289	N ₂	Ra	1

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C₇⁻

Threshold for electron detachment from ground-state
 $C_7^- = 27090(115)$ gas PE^{1,2}

$\tilde{C} \ ^2\Pi_u$ $D_{\infty h}$
 $T_0 = 35231(25)$ Ne AB⁶ $\tilde{C} - \tilde{X}$ 278–284 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	3		491(25)	Ne	AB	6

$\tilde{B} \ ^2\Pi_u$ $D_{\infty h}$
 $T_0 = 20298$ gas MPD^{5,7,9} $\tilde{B} - \tilde{X}$ 447–493 nm
 20314(8) Ne AB⁶ $\tilde{B} - \tilde{X}$ 380–493 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	2038	gas	MPD	5,7,9
			2058(10)	Ne	AB	6
			1569(9)	Ne	AB	6
	2		532	gas	MPD	7,9
	3		575(8)	Ne	AB	6

$\tilde{A} \ ^2\Pi_u$ $D_{\infty h}$
 $T_0 = 15930$ gas MPD^{5,7,9,10} $\tilde{A} - \tilde{X}$ 496–627 nm
 15954(5) Ne AB⁶ $\tilde{A} - \tilde{X}$ 495–627 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1870	gas	MPD	5,7,9,10
			1877(6)	Ne	AB	6
			1452	gas	MPD	7,9,10
	2		1457(5)	Ne	AB	6
			543	gas	MPD	5,7,9,10
	3		562(5)	Ne	AB	6

$A = 0.6(4)$ MPD¹⁰

		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_u^+	5	Asym. stretch	1736.4	Ne	IR	4
			1734.8	Ar	IR	3,8

$A = 27.4(2)$ MPD¹⁰

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- N. M. Lakin, M. Pachkov, M. Tulej, J. P. Maier, G. Chambaud, and P. Rosmus, *J. Chem. Phys.* **113**, 9586 (2000).

Si₇⁻

Threshold for electron detachment from ground-state
 $Si_7^- = 14930(160)$ gas PE¹

Reference

- C. Xu, T. R. Taylor, G. R. Burton, and D. M. Neumark, *J. Chem. Phys.* **108**, 1395 (1998).

C₅N₂

$\tilde{A} \ ^3\Sigma_u^-$ $D_{\infty h}$
 $T_0 = 22832.7$ gas CR² $\tilde{A} - \tilde{X}$ 403–438 nm
 22762.9(2)T Ne AB¹LF¹
 22737.3(2)T Ar AB¹LF¹ $\tilde{A} - \tilde{X}$ 375–610 nm
 22731.8(2)T N₂ AB¹LF¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1		1958	gas	CR	2
			1961.3(2)T	Ne	AB	1
			1960.5(2)	Ar	AB	1
	8		601.2(2)T	Ar	AB	1
			508.5(2)T	Ar	AB	1

$B_0 = 0.028$ CR²

$\tilde{X}^3\Sigma_g^-$		D_{zh}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
1			2070.7(2)T	Ne	LF	1
			2071.4(2)T	Ar	LF	1
			2071.6(2)T	N ₂	LF	1
2			2030.7(2)T	Ne	LF	1
			2030.9(2)T	Ar	LF	1
			2032.0(2)T	N ₂	LF	1
3			1639.2(2)T	Ne	LF	1
			1639.5(2)T	Ar	LF	1
			1639.7(2)T	N ₂	LF	1
4			871.7(2)T	Ne	LF	1
			870.9(2)T	Ar	LF	1
5			860.8(2)T	Ne	LF	1
			859.3(2)T	Ar	LF	1
6			804.7(2)T	Ne	LF	1
			801.5(2)T	Ar	LF	1
			753.3(2)T	Ne	LF	1
7			748.5(2)T	Ar	LF	1
			603.2(2)T ^a	Ne	LF	1
			609.6(2)T	Ar	LF	1
8			601.9(2)T	Ar	LF	1
			527.7(2)T ^a	Ne	LF	1
			530.5(2)T	Ar	LF	1
			525.2(2)T	Ar	LF	1
9			521.9(2)T	Ar	LF	1

$$B_0 = 0.028 \text{ CR}^2$$

^aCenter of split transition.

References

- ¹A. M. Smith, C. Engel, A. Thoma, G. Schallmoser, B. E. Wurfel, and V. E. Bondybey, *Chem. Phys.* **184**, 233 (1994).
²H. Linnartz, O. Vaizert, P. Cias, L. Grüter, and J. P. Maier, *Chem. Phys. Lett.* **345**, 89 (2001).

Al₃O₄

\tilde{A}
 $T_0 = 12260(970) \text{ gas PE}^1$

Reference

- ¹H. Wu, X. Li, X.-B. Wang, C.-F. Ding, and L.-S. Wang, *J. Chem. Phys.* **109**, 449 (1998).

Fe₂O₅

\tilde{A}
 $T_0 = 1450(400) \text{ gas PE}^1$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			790(60)	gas	PE	1

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			750(50)	gas	PE	1

Reference

- ¹H. Wu, S. R. Desai, and L.-S. Wang, *J. Am. Chem. Soc.* **118**, 5296 (1996);
J. Am. Chem. Soc. **118**, 7434 (1996).

Al₃O₄⁻

Threshold for electron detachment from ground-state
 $\text{Al}_3\text{O}_4^- = 28880(650) \text{ gas PE}^1$

Reference

- ¹H. Wu, X. Li, X.-B. Wang, C.-F. Ding, and L.-S. Wang, *J. Chem. Phys.* **109**, 449 (1998).

Fe₂O₅⁻

Threshold for electron detachment from ground-state
 $\text{Fe}_2\text{O}_5^- = 31300(320) \text{ gas PE}^1$

Reference

- ¹H. Wu, S. R. Desai, and L.-S. Wang, *J. Am. Chem. Soc.* **118**, 5296 (1996);
J. Am. Chem. Soc. **118**, 7434 (1996).

V(NO)₃

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO a-stretch	1724.0	Ne	IR	2
			1715.1	Ar	IR	1

References

- ¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **103**, 478 (1999).
²L. Andrews and X. Wang, *J. Phys. Chem. A* **106**, 1196 (2002).

Nb(NO)₃

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO stretch	1673.2	Ar	IR	1

Reference

- ¹M. Zhou and L. Andrews, *J. Phys. Chem. A* **102**, 10025 (1998).

Ta(NO)₃ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO stretch	1676.9	Ar	IR	1

Reference¹M. Zhou and L. Andrews, J. Phys. Chem. A **102**, 10025 (1998).**Mn(NO)₃** \tilde{X} C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	NO stretch	1824.1	Ar	IR	1
<i>e</i>		NO stretch	1721.8	Ne	IR	2
			1713.2	Ar	IR	1
		MnNO deform.	594.1	Ar	IR	1
			591.0			
		MnN stretch	534.3	Ar	IR	1

References¹L. Andrews, M. Zhou, and D. W. Ball, J. Phys. Chem. A **102**, 10041 (1998).²L. Andrews and X. Wang, J. Phys. Chem. A **106**, 1196 (2002).**Fe(NO)₃** \tilde{X} C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	NO stretch	1794.6	Ar	IR	1
<i>e</i>		NO stretch	1757.8	Ne	IR	1
			1742.6	Ar	IR	1
		FeN stretch	513.4	Ar	IR	1

Reference¹M. Zhou and L. Andrews, J. Phys. Chem. A **104**, 3915 (2000).**Ru(NO)₃** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO stretch	1741.1	Ne	IR	1
			1738.9	Ar	IR	1

Reference¹A. Citra and L. Andrews, J. Phys. Chem. A **104**, 8689 (2000).**Os(NO)₃** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO stretch	1745.5	Ar	IR	1

Reference¹A. Citra and L. Andrews, J. Phys. Chem. A **104**, 8689 (2000).**Co(NO)₃** \tilde{X} C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	1	NO stretch	1825.6	Ne	IR	1
			1814.6	Ar	IR	1
		CoN stretch	579.3	Ar	IR	1
<i>e</i>		NO stretch	1782.1	Ne	IR	1
			1770.1	Ar	IR	1
		CoN stretch	493.8	Ar	IR	1

Reference¹M. Zhou and L. Andrews, J. Phys. Chem. A **104**, 3915 (2000).**Rh(NO)₃** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO stretch	1757.7	Ne	IR	1
			1749.0	Ar	IR	1

Reference¹A. Citra and L. Andrews, J. Phys. Chem. A **104**, 11897 (2000).**Ir(NO)₃** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		NO stretch	1753.7	Ne	IR	1
			1745.5	Ar	IR	1

Reference¹A. Citra and L. Andrews, J. Phys. Chem. A **104**, 11897 (2000).

Al_2O_5^-

Threshold for electron detachment from ground-state $\text{Al}_2\text{O}_5^- = 30260(240)$ gas PE¹

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		Sym. stretch	1170(60)	gas	PE	1

Reference

¹S. R. Desai, H. Wu, C. M. Rohlfing, and L.-S. Wang, *J. Chem. Phys.* **106**, 1309 (1997).

 $\text{O}_2\text{N-O-NO}_2$

In the gas phase, continuous absorption begins near 420 nm, and has its maximum beyond 200 nm.^{11,15}

 \tilde{X} C_2^a Structure: ED⁶MW¹³DL¹⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
<i>a</i>	1	NO ₂ a-stretch	1720vs	gas	IR	1,2,5,8	
			1742.4vs	Ar	IR	12	
			1745	N ₂	IR	3	
			1752	CO ₂	IR	1	
			1338m	gas	IR	2,8	
	2	NO ₂ s-stretch	1339.8wm	Ar	IR	12	
			1305	N ₂	IR	3	
			1300	O ₂	IR	4	
			1316	CO ₂	IR	1	
			860m	gas	IR	2,8	
	3		863.1w	Ar	IR	12	
	<i>b</i>	9	NO ₂ a-stretch	1720vs	gas	IR	1,2,5,8
				1702.7s	Ar	IR	7,12
				1704	N ₂	IR	3
1704				O ₂	IR	4	
1700				CO ₂	IR	1	
10		NO ₂ s-stretch	1245.9s	gas	IR,DL	1,2,5,8,11,14	
			1243.0m	Ar	IR	7,12	
			1247	N ₂	IR	3	
			1241	O ₂	IR	4	
			1248	CO ₂	IR	1	
11			743.4s	gas	IR	1,2,5,8,11	
			737s	Ar	IR	7,12	
			739	N ₂	IR	3	
			736	O ₂	IR	4	
			719	CO ₂	IR	1	
12		614mT	gas	IR	2		
		639.7w	Ar	IR	12		
13		557s	gas	IR	2,5		
		569.4m	Ar	IR	12		
14		353vsT	gas	IR	2,9		
		343.9vs	Ar	IR	12		
15		50w,brT	gas	IR	9		

$A_0=0.221$; $B_0=0.063$; $C_0=0.060$ MW¹⁰

^aRef. 2 analyzed the spectrum in terms of a C_{2v} structure. However, the electron diffraction measurements of Ref. 6, analyzed using a dynamical model, and the low rotational temperature microwave and diode laser spectral observations of Refs. 13 and 14 suggest instead a C_2 structure. The vibrational assignment has been revised to correspond with the results of *ab initio* calculations given in Ref. 13.

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- M. H. Harwood, J. B. Burkholder, and A. R. Ravishankara, *J. Phys. Chem. A* **102**, 1309 (1998).

 SF_6^+ \tilde{F}^2A_{1g}

$T_0^a = 94450(100)$ gas PE^{2,3,6}TPE⁷
Fragments into SF_2^+ . T-PEPICO⁵

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	Sym. stretch	690(40)	gas	PE	6,7
e_g	2		460T	gas	PE TPE	6,7

 \tilde{E}^2F_{1u}

$T_0^a = 57620(100)$ gas PE^{2,3,6}TPE⁷
Fragments into SF_3^+ . T-PEPICO⁵

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_g	1	Sym. stretch	540(40)	gas	PE TPE	6,7

 \tilde{D}^2F_{2g}

$T_0^a = 33250(150)$ gas PE^{1-3,6}TPE⁷
Fragments into SF_5^+ , SF_4^+ , and SF_3^+ . T-PEPICO⁵

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_g	1	Sym. stretch	590(40)	gas	PE TPE	3,6,7

 \tilde{C}^2E_g

$T_0^a \leq 24360(100)$ gas PE^{1-3,6,8}TPE⁷
Low vibrational levels fragment into $\text{SF}_5^+ + \text{F}$, higher levels yield SF_4^+ . T-PEPICO⁵

$\tilde{A}, \tilde{B} \ ^2F_{1u}, \ ^2F_{2u}$
 $T_0^a = 12780(240)$ gas PE^{1-3,6,8}TPE⁷

 Direct dissociation into SF₅⁺ + F. T-PEPICO^{4,5}
 $\tilde{X} \ ^2F_{1g}$

 Direct dissociation into SF₅⁺ + F. T-PEPICO^{4,5}
^aMeasured with respect to photoionization threshold, 15.116 eV, determined by Ref. 7.

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 SF_6^-

\tilde{X}		O_h				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
t_{1u}	3		619.8T	Ne	IR	1

Reference

- C. L. Lugez, M. E. Jacox, R. A. King, and H. F. Schaefer III, *J. Chem. Phys.* **108**, 9639 (1998).

8.14. Eight-Atomic Molecules

 $GaBH_6$

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	BH ₂ s-stretch	2482m	gas	IR	1
			2482m	N ₂	IR	1
			2465m			
	2	GaH ₂ s-stretch	1982w	gas	IR	1
			1965m	N ₂	IR	1
			1937m	gas	IR	1
			1925m	N ₂	IR	1
4	Mixed	1435s	gas	IR	1	
		1432s	N ₂	IR	1	
		1422s				
		1130m	gas	IR	1	
5	BH ₂ scissors	1125m	N ₂	IR	1	
		1120m				
		1105s				
6	GaH ₂ scissors	729m	gas	IR	1	
		730s	N ₂	IR	1	
		468m	gas	IR	1	
7	Mixed	460s	N ₂	IR	1	
		452m				
		2007m	gas	IR	1	
b_1	10	Mixed	2018s	N ₂	IR	1
			1315w	gas	IR	1
11	Mixed	1330w	N ₂	IR	1	
		1290m				
		782vw	gas	IR	1	
		770w	N ₂	IR	1	
		544m	gas	IR	1	
13	GaH ₂ wag	570m	N ₂	IR	1	
		560m				
		2558m	gas	IR	1	
b_2	14	BH ₂ a-stretch	2558m	N ₂	IR	1
			2536m			
15	GaH ₂ a-stretch	2005m	gas	IR	1	
		1992s,br	N ₂	IR	1	
		930vw	gas	IR	1	
		620w	gas	IR	1	
16	BH ₂ rock	630w	N ₂	IR	1	
		612w				
17	GaH ₂ rock	630w	gas	IR	1	
		612w	N ₂	IR	1	

 $GaBD_6$

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	BD ₂ s-stretch	1823m	gas	IR	1
			1428w	gas	IR	1
			1383m	gas	IR	1
			1060s	gas	IR	1
			850w	gas	IR	1
			527m	gas	IR	1
			412s	gas	IR	1
b_1	10	Mixed	1450m	gas	IR	1
			980w	gas	IR	1
			421s	gas	IR	1
b_2	14	BD ₂ a-stretch	1933.5m	gas	IR	1
			1447m	gas	IR	1
			670w	gas	IR	1
			442m	gas	IR	1
			442m	gas	IR	1

Reference

¹A. J. Downs, T. M. Greene, E. Johnsen, P. T. Brain, C. A. Morrison, S. Parsons, C. R. Pulham, D. W. H. Rankin, K. Aarset, I. M. Mills, E. M. Page, and D. A. Rice, *Inorg. Chem.* **40**, 3482 (2001).

ZnC₂H₅⁺

\tilde{X}^1A'		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	6	CH ₃ bend	1380(3)	gas	TPE	1
	7	Wag	1107(3)	gas	TPE	1
	8	CC stretch	1015(3)	gas	TPE	1
	9	Wag	933(3)	gas	TPE	1
	10	ZnC stretch	424(3)	gas	TPE	1
	11	ZnCC bend	217(3)	gas	TPE	1
<i>a''</i>	17	CH ₂ rock	683(3)	gas	TPE	1
	18	CH ₃ torsion	238(3)	gas	TPE	1

Reference

¹M. B. Pushkarsky, V. L. Stakhursky, and T. A. Miller, *J. Phys. Chem. A* **104**, 9184 (2000).

ZnC₂H₅

\tilde{A}^2A'		C_s				
$T_0=22515(5)$		gas	LF ¹ MPI ²	$\tilde{A}-\tilde{X}$ 425–475 nm		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	7	Wag	1087(3)	gas	MPI	2
	8	CC stretch	1011(5)	gas	LF,MPI	1,2
	10	ZnC stretch	424(5)	gas	LF,MPI	1,2
	11	ZnCC bend	245(5)	gas	LF,MPI	1,2
<i>a''</i>	18	CH ₃ torsion	218HT	gas	MPI	2

\tilde{X}^2A'		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>	7	Wag	1109(5)T	gas	LF	1
	8	CC stretch	987(5)T	gas	LF	1
	9	Wag	915(5)T	gas	LF	1
	10	ZnC stretch	387(5)	gas	LF	1
	11	ZnCC bend	180(5)	gas	LF	1

References

¹S. J. Pooley and A. M. Ellis, *J. Mol. Spectrosc.* **185**, 48 (1997).

²M. B. Pushkarsky, V. L. Stakhursky, and T. A. Miller, *J. Phys. Chem. A* **104**, 9184 (2000).

CdC₂H₅

\tilde{A} C_s
The absence of structure in the excitation spectrum of this transition suggests¹ that predissociation occurs in the \tilde{A} state of CdC₂H₅.
 $T_0=21374(5)$ gas LF¹ $\tilde{A}-\tilde{X}$ 467–488 nm

\tilde{X}^2A'		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>			880(5)	gas	LF	1
	10	CdC stretch	278(5)	gas	LF	1
	11	CdCC bend	157(5)	gas	LF	1

Reference

¹A. J. Bezant and A. M. Ellis, *J. Mol. Spectrosc.* **185**, 54 (1997).

H₂BC₂H₃

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		BH ₂ a-stretch	2570.2	Ar	IR	1
			1536.8	Ar	IR	1
			1414.1w	Ar	IR	1
			1213.9	Ar	IR	1
			1031.3w	Ar	IR	1

D₂BC₂D₃

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1472.5	Ar	IR	1

Reference

¹L. Andrews, D. V. Lanzisera, P. Hassanzadeh, and Y. Hannachi, *J. Phys. Chem. A* **102**, 3259 (1998).

CH₃B=CH₂

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		B=C stretch	1631.2	Ar	IR	1

Reference

¹L. Andrews, D. V. Lanzisera, P. Hassanzadeh, and Y. Hannachi, *J. Phys. Chem. A* **102**, 3259 (1998).

GaNH₂CH₃⁺

\tilde{X}^1A'		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		NH ₃ rock	515T	gas	TPE	1
		GaN stretch	299 (ω)	gas	TPE	1
		Bend	124 (ω)	gas	TPE	1

Reference

¹S. Li, G. K. Rothschof, D. Pillai, B. R. Sohnlein, B. M. Wilson, and D.-S. Yang, *J. Chem. Phys.* **115**, 7968 (2001).

CH₂CHCH₂⁺

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	5	CH ₂ s-rock	1239(7)	gas	TPE	2
	6	C ₃ s-stretch	1027(7)	gas	TPE	2
	7	C ₃ bend	431(7)	gas	PE,TPE	1,2
a_2	9	CH ₂ a-twist	612(7)	gas	TPE	2
b_1	10	CH OPLA	1078(7)	gas	TPE	2
	12	CH ₂ s-twist	274(7)	gas	TPE	2
b_2	17	C ₃ a-stretch	1236(7)	gas	TPE	2

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¹F. A. Houle and J. L. Beauchamp, *J. Am. Chem. Soc.* **100**, 3290 (1978).

²T. Gilbert, I. Fischer, and P. Chen, *J. Chem. Phys.* **113**, 561 (2000).

GaNH₂CH₃

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		GaNc bend	93	gas	TPE	1

Reference

¹S. Li, G. K. Rothschof, D. Pillai, B. R. Sohnlein, B. M. Wilson, and D.-S. Yang, *J. Chem. Phys.* **115**, 7968 (2001).

CH₂CHCH₂

8s C_{2v}
 $T_0=63398(5)$ gas MPI²⁵

7s C_{2v}
 $T_0=62620(5)$ gas MPI²⁵

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	7	C ₃ bend	429(5)	gas	MPI	25

6s C_{2v}
 $T_0=61337(5)$ gas MPI²⁵

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	7	C ₃ bend	439(5)	gas	MPI	25

4s C_{2v}
 $T_0=53708(5)$ gas MPI²⁵

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	6	C ₃ stretch	1034(5)	gas	MPI	25
	7	C ₃ bend	429(5)	gas	MPI	25

3d C_{2v}
 $T_0=52114(5)$ gas MPI²⁵

\tilde{D}^2B_2 C_{2v}
 $T_0=41557.8(5)$ gas MPI¹³Ra¹⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_2	9	CH ₂ a-twist	596	gas	Ra	14
b_1	12	CH ₂ s-twist	564	gas	Ra	14

$\tau_0=12$ ps gas MPI²⁰

\tilde{C}^2B_1 C_{2v} Structure: MPI¹⁸
 $T_0=40305.5(5)$ gas AB²MPI^{11,13,18} $\tilde{C}-\tilde{X}$ 220–250 nm
In an argon matrix, a very prominent absorption maximum at 213 nm, with a shoulder at 220 nm, has been assigned⁵ to this transition of CH₂CHCH₂.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	7	C ₃ bend	385(2)	gas	MPI	11,18

$\tau_0=15$ ps gas MPI²⁰

$A_0\cong 1.619$; $B_0\cong 0.351$; $C_0\cong 0.288$ MPI¹¹

$\tilde{B}^2A_1(3s)$ C_{2v}^a Structure: MPI¹⁸
 $T_0=40056.5(5)$ gas MPI^{7,8,13,18}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	7	C ₃ bend	379(2)	gas	MPI	7,8,18
a_2	9	CH ₂ a-twist	596(2)	gas	MPI	18
b_1	12	CH ₂ s-twist	491(7)T	gas	MPI	8

$\tau_0=20$ ps gas MPI²⁰

\tilde{A}^2B_1 C_2
 $T_0=24485$ gas AB¹CR²⁴ $\tilde{A}-\tilde{X}$ 370–410 nm
24480 Ar AB⁵ $\tilde{A}-\tilde{X}$ 360–410 nm
Diffuse bands.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a			1241	gas	AB,CR	1,24
			1005	gas	AB,CR	1,24
			908	gas	AB,CR	1,24
			359	gas	AB,CR	1,24

\tilde{X}^2A_2		C_{2v}		Structure: DL ¹²		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CH ₂ s-stretch	3113.98	gas	LD,CC	21,22
			3109wm	Ar	IR	4-6,26
	2	CH stretch	3052w	Ar	IR	4-6,26
			3027vw	Ar	IR	26
	4	CH ₂ scissors	1488(4)	gas	Ra	10,15
			1478w	Ar	IR	4-6,26
	5	Mixed	1245(3)	gas	Ra	10,15
1242vw			Ar	IR	4-6,26	
6	Mixed	1066(4)	gas	Ra	10,15	
		990(20)	gas	PE	19	
7	C ₃ bend	427(4)	gas	MPI,Ra	7,8,10,15	
9	CH ₂ a-twist	549(4)	gas	MPI,Ra	8,15,16	
b_1	10	CH OPLA	968H	gas	Ra	10
			983.6wm	Ar	IR	4-6,9,26
11	CH ₂ s-wag	801.72	gas	DL,PE	12,19	
		801.1vs	Ar	IR	3-6,9,26	
12	CH ₂ s-twist	518(4)	gas	MPI,Ra	7,8,10,15	
		510.1wm	Ar	IR	4-6,9,26	
b_2	13	CH ₂ a-stretch	3110.60	gas	LD,CC	21-23
			3107w	Ar	IR	26
	14	CH stretch	3020w	Ar	IR	4-6,26
			1464wm	Ar	IR	4-6,26
	16	Mixed	1390vw	Ar	IR	4-6,26
			1182	Ar	IR	6,26

$A_0=1.802$; $B_0=0.346$; $C_0=0.290$ DL¹²LD^{21,23}

CD₂CD₂

\tilde{D}^2B_2		C_{2v}	
$T_0=41532.1(5)$	gas	MPI ¹³	
\tilde{C}^2B_1		C_{2v}	
$T_0=40286.7(5)$	gas	MPI ^{13,18}	

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	7	C ₃ bend	321(2)	gas	MPI	18

$\tilde{B}^2A_1(3s)$		C_{2v}^a				
$T_0=40096(2)$	gas	MPI ^{7,13,18}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	7	C ₃ bend	316(2)	gas	MPI	7,18
a_2	9	CD ₂ a-twist	429(2)	gas	MPI	18

\tilde{A}^2B_1		C_{2v}				
$T_0=24745$	gas	AB ¹				
Diffuse bands.						
$\tilde{A}-\tilde{X}$ 360-405 nm						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1			1155	gas	AB	1
			981	gas	AB	1
			823	gas	AB	1

\tilde{X}^2A_2		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	2	CD stretch	2285w	Ar	IR	6,26
			2215w	Ar	IR	6
3	CD stretch	1272(8)	1263	gas	Ra	17
			1263	Ar	IR	6,26
4	CD ₂ scissors	1020(8)	1018w	gas	Ra	17
			1018w	Ar	IR	6,26
5	Mixed	835(20)	835(20)	gas	MPI,PE	7,19
			844vw	Ar	IR	26
6	Mixed	350(8)	350(8)	gas	MPI,Ra	7,17-19
			350(8)	PE		
9	CD ₂ torsion	372H	372H	gas	Ra	17
			762H	gas	Ra	17
b_1	10	OPLA	767w	Ar	IR	26
			678(10)H	gas	PR	19
11	Deform.	646.5vs	646.5vs	Ar	IR	6,9,26
			403(8)	gas	Ra,MPI	17,18
12	Deform.	2209	2209	Ar	IR	6,26
			1387wm	Ar	IR	26
b_2	14	CD ₂ stretch	1062w	Ar	IR	6,26
			1062w	Ar	IR	6,26
15	CD ₂ scissors	900w	900w	Ar	IR	26
			900w	Ar	IR	26

^aEvidence presented in Refs. 14 and 18 suggests that this state may be slightly nonplanar.

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C₂H₃NH₂

\tilde{X}		C_1	Structure: MW ³⁻⁵ IR ⁵			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	5	CH stretch	2987 2976	gas	IR	2
	6	C=C stretch	1672 1668	gas	IR	2
	7	NH ₂ scissors	1625T	gas	IR	2
	8	CH ₂ scissors	1454	gas	IR	2
	10	C-N stretch	1260 1248	gas	IR	2
	11	NH ₂ rock	1084 1078	gas	IR	2
	12		1046 1039	gas	IR	2
	14		812s 805s	gas	IR	2
	15		674.44	gas	IR	2,4,5,7
	17		469.91	gas	IR	4,5,7
	18	NH ₂ wag	614.76 ^b 569.58 ^c 331.92 ^d 286.71 ^e	gas	IR	2,4,5,7 2,4,5,7 4,5,7 4,5,7

$A_0 = 1.879$; $B_0 = 0.335$; $C_0 = 0.286$ MW^{1,3,5}

^aThe barrier to inversion about the N atom is 356(3) cm^{-1,3-5} leading to splittings in a number of the vibrational band centers. The 0⁻-0⁺ transition has been observed^{5,6} in the far IR at 45.5 cm⁻¹.

^b1⁻-0⁺.

^c1⁻-0⁻.

^d1⁺-0⁺.

^e1⁺-0⁻.

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CH₃CH₂O⁺

$T^a = 22190(120)$ gas PE¹

\tilde{b}^1A'' C_s
 $T^a = 9280(120)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'			1170(60)	gas	PE	1

\tilde{a}^1A' C_s
 $T^a = 5320(120)$ gas PE¹

\tilde{X}^3A'' C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'			1260(60)	gas	PE	1

^aFrom vertical ionization potentials.

Reference

- ¹J. Wang, Z. Sun, X. Zhu, M. Ge, and D. Wang, *Chem. Phys. Lett.* **340**, 98 (2001).

CH₃CHOH⁺

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1600(30)	gas	PE	1

CD₃CDOD⁺

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1590(30)	gas	PE	1

Reference

- ¹J. M. Dyke, A. P. Groves, E. P. F. Lee, and M. H. Zamanpour Niavaran, *J. Phys. Chem. A* **101**, 373 (1997).

C₂H₅S⁺

\tilde{B}^3A' C_s
 $T^a = 34940(240)$ gas PE²

\tilde{A}^3A'' C_s
 $T^a = 26380(240)$ gas PE²

\tilde{b}^1A'' C_s
 $T^a = 11460(240)$ gas PE²

\tilde{a}^1A' C_s
 $T^a = 9920(240)$ gas PE²

\tilde{X}^3A'' C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		CS stretch	670(30)	gas	PE	1,2

^aFrom vertical ionization potentials.

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C₂H₅O

\tilde{B}^2A' C_s
 $T_0 = 29210.207(6)$ gas EM^{1,2,4}LF^{3,5,6} $\tilde{B}-\tilde{X}$ 310–500 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>		CO stretch	585H	gas	LF	5,6

$\tau_0 = 1.7(2)$ μ s gas LF³EM⁴
 $A_0 = 1.117(2)$; $B_0 = 0.311(2)$; $C_0 = 0.268(2)$ LF⁶

\tilde{A}^2A' C_s
 $T_0 = 355(10)$ gas PE⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>			685T	gas	PE	7

\tilde{X}^2A'' C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>		CH ₂ wag	1370	gas	LF	3
		CH ₂ scissors	1295(10)	gas	PE	7
		C–O stretch	1067	gas	LFEM	3–5
		C–C stretch	875 ^a	gas	LF	3
		CCO bend	442	gas	LF	3

$A_0 = 1.318(2)$; $B_0 = 0.321(2)$; $C_0 = 0.284(3)$ LF⁶

C₂D₅O

\tilde{A}^2A' C_s
 $T_0 = 275(10)$ gas PE⁷

^aAssignment to overtone of CCO bend cannot be excluded.

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C₂H₅S

\tilde{B} C_s
 $T_0 = 23519.6$ gas LF^{1,2} $\tilde{A}-\tilde{X}$ 390–600 nm
 Predissociation occurs.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>		CH ₂ wag	1158.9	gas	LF	2
		CC stretch	1054.6	gas	LF	2
		CH ₃ wag	862.8	gas	LF	2
		CS stretch	420.5	gas	LF	1,2
		CCS bend	256.0	gas	LF	2
<i>a''</i>		CH ₂ twist	1203.3	gas	LF	2
		CH ₂ rock	718.4	gas	LF	2

$\tau = 75$ ns gas LF¹

\tilde{A}^2A' C_s
 $T_0 = 240(30)$ gas PE³

\tilde{X}^2A'' C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>		CH ₃ s-stretch	2950T	gas	LF	2
		CH ₂ bend	1470(10)	gas	LF	2
		CH ₂ wag	1257(10)	gas	LF	2
		CC stretch	1075(10)	gas	LF	2
		CH ₃ wag	957(15)	gas	LF	2
		CS stretch	672.4	gas	LF	1,2
		CCS bend	296.0	gas	LF	2
<i>a''</i>		CH ₃ a-stretch	3050T	gas	LF	2
		CH ₂ twist	1290(10)	gas	LF	2
		CH ₃ wag	890(10)	gas	LF	2
		CH ₂ rock	478.3	gas	LF	2
		Torsion	271.9	gas	LF	2

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C₂H₅O⁻

Threshold for electron detachment from ground-state C₂H₅O⁻ = 13810(30) gas PE^{1–3}

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1250(400)	gas	PE	1

C₂D₅O⁻

Threshold for electron detachment from ground-state C₂D₅O⁻ = 13710(30) gas PE¹⁻³

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1200(400)	gas	PE	1

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HFeCH=C=CH₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		C ₃ a-stretch	1832.0	Ar	IR	1,2
		FeH stretch	1737.8	Ar	IR	1,2
		CH ₂ scissors	1413.8	Ar	IR	1
		CH ₂ wag	820.2	Ar	IR	1,2
		FeC stretch	431.4	Ar	IR	1,2
			428.7			

DFeCH=C=CD₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		C ₃ a-stretch	1799.0	Ar	IR	1,2
		FeD stretch	1250.0	Ar	IR	1,2
		CD ₂ wag	585.5	Ar	IR	1,2

References

- ¹D. W. Ball, R. G. S. Pong, and Z. H. Kafafi, *J. Am. Chem. Soc.* **115**, 2864 (1993).
- ²D. W. Ball, R. G. S. Pong, and Z. H. Kafafi, *J. Phys. Chem.* **98**, 10720 (1994).

HCCCH₂FeH \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH stretch	3314.4	Ar	IR	1,2
		FeH stretch	1697.5	Ar	IR	1,2

DCCCD₂FeD \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		FeD stretch	1204.3	Ar	IR	1,2

References

- ¹D. W. Ball, R. G. S. Pong, and Z. H. Kafafi, *J. Am. Chem. Soc.* **115**, 2864 (1993).
- ²D. W. Ball, R. G. S. Pong, and Z. H. Kafafi, *J. Phys. Chem.* **98**, 10720 (1994).

HFeCCCH₃ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH ₃ s-stretch	2881.8	Ar	IR	1,2
		CC stretch	2117.1	Ar	IR	1,2
			2114.7			
			2110.5			
		FeH stretch	1721.9	Ar	IR	1,2
		CH ₃ deform.	1373.8	Ar	IR	2
		CH ₃ rock	984.9	Ar	IR	1,2
		C-C stretch	911.0	Ar	IR	1,2

DFeCCCD₃ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CC stretch	2113.5	Ar	IR	1,2
			2109.5			
		CD ₃ s-stretch	2036.9	Ar	IR	2
		FeD stretch	1239.0	Ar	IR	1,2
			1236.2			
			1234.9			
		CD ₃ deform.	1113.2	Ar	IR	1,2
			1112.2			
		C-C stretch	897.0	Ar	IR	1,2
			895.1			

References

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- ²D. W. Ball, R. G. S. Pong, and Z. H. Kafafi, *J. Phys. Chem.* **98**, 10720 (1994).

CH₃FeCCH

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH stretch	3328.1	Ar	IR	1
		CH ₃ stretch	2963.8	Ar	IR	1
		CC stretch	1977.4	Ar	IR	1,2
			1975.5			
			1965.5			
		CH ₃ deform.	1455.6	Ar	IR	1
			1439.4			
		CH ₃ s-deform.	1163.7	Ar	IR	1,2
			1161.9			
		CH bend	669.7	Ar	IR	1,2
		FeC stretch	566.0	Ar	IR	1,2
			564.7			
		CH ₃ rock	551.0	Ar	IR	1,2
			540.8			

CD₃FeCCD

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CC stretch	1864.3	Ar	IR	1,2
			1861.4			
			1851.5			
			1850.8			
		CD ₃ s-deform.	912.5	Ar	IR	2
		FeC stretch	520.4	Ar	IR	1,2
			519.2			
		CD ₃ rock	421.8	Ar	IR	2
			416.4			

References

- ¹D. W. Ball, R. G. S. Pong, and Z. H. Kafafi, *J. Am. Chem. Soc.* **115**, 2864 (1993).
²D. W. Ball, R. G. S. Pong, and Z. H. Kafafi, *J. Phys. Chem.* **98**, 10720 (1994).

cyc-C₄H₄

In an argon matrix,⁸ absorption is weak between 420 and 250 nm and is accompanied by dissociation to form C₂H₂. Below 250 nm, the absorption rises abruptly.

\tilde{X}						
D _{2h} Structure: IR ³⁻⁵ Ra ⁷						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a _g		CH stretch	3140(5)	Ne,Ar	Ra	7
		C=C stretch	1678(5)	Ne,Ar	Ra	7
		CH deform.	1059(5)	Ne,Ar	Ra	7
		CH deform.	989(5)	Ne,Ar	Ra	7
b _{1u}		CH stretch	3124wm	Ar	IR	5,6,8
		C=C stretch	1527	Ne	IR	6
			1526w	Ar	IR	4-6,8
		CH deform.	1028vw	Ar	IR	5
b _{2g}		CH OPLA deform.	531(5)	Ne,Ar	Ra	7
b _{2u}		CH stretch	3105wm	Ar	IR	5,6,8
		CH deform.	1244	Ne	IR	6
			1242s	Ar	IR	1-6,8
			1237.7	Xe	IR	9
		Ring deform.	719wm	Ar	IR	4-6,8
b _{3g}		CH stretch	3093(5)	Ne,Ar	Ra	7
		Ring deform.	723(5)	Ne,Ar	Ra	7
b _{3u}		CH OPLA	576	Ne	IR	6
			569vs	Ar	IR	1-6,8
			573.3	Xe	IR	9

cyc-C₄D₄

\tilde{X}						
D _{2h}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b _{1u}		C=C stretch	1456w	Ar	IR	4
b _{2u}		CD deform.	1043wm	Ar	IR	4
		Ring deform.	609wm	Ar	IR	4
b _{3u}		CD OPLA	421vs	Ar	IR	4

References

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⁴S. Masamune, F. A. Souto-Bachiller, T. Machiguchi, and J. E. Bertie, *J. Am. Chem. Soc.* **100**, 4889 (1978).
⁵G. Maier, *Angew. Chem.* **100**, 317 (1988); *Angew. Chem. Int. Ed. Engl.* **27**, 309 (1988).
⁶A. M. Orendt, B. R. Arnold, J. G. Radziszewski, J. C. Facelli, K. D. Malsch, H. Strub, D. M. Grant, and J. Michl, *J. Am. Chem. Soc.* **110**, 2648 (1988).
⁷B. R. Arnold, J. G. Radziszewski, A. Campion, S. S. Perry, and J. Michl, *J. Am. Chem. Soc.* **113**, 692 (1991).
⁸B. R. Arnold and J. Michl, *J. Phys. Chem.* **97**, 13348 (1993).
⁹G. Maier and C. Lautz, *Eur. J. Org. Chem.* **1998**, 769.

H₂C=C=CHCH:

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		C ₃ a-stretch ^a	1942	Ar	IR	1
		C ₃ a-stretch ^a	1881	Ar	IR	1
			877	Ar	IR	1

^aIt is likely that more than one stereoisomer contributes to the observed absorptions.

Reference

¹R. Wrobel, W. Sander, D. Cremer, and E. Kraka, *J. Phys. Chem. A* **104**, 3819 (2000).

H₃C(cyc-CSiCH)

In an argon matrix, irradiation at 313 nm results in isomerization to H₃CCCSiH.¹

 $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH stretch	3014.9wm	Ar	IR	1
		CH ₃ stretch	2973.4m 2963.6	Ar	IR	1
		CH ₃ stretch	2929.5m 2915.3	Ar	IR	1
		CH ₃ stretch	2850.0wm	Ar	IR	1
		C=C stretch	1562.8wm	Ar	IR	1
		CH ₃ deform.	1440.5wm	Ar	IR	1
		CH ₃ deform.	1433.9wm	Ar	IR	1
		CH ₃ deform.	1360.9w	Ar	IR	1
		CH ₃ rock	1140.4wm	Ar	IR	1
		CH ₃ rock	1016.9w	Ar	IR	1
		CH bend	964.1wm	Ar	IR	1
		CC stretch	953.3s	Ar	IR	1
		CH bend	823.4m	Ar	IR	1
		Si-(CH) stretch	744.0vs	Ar	IR	1
		Si-(CCH ₃) stretch	576.5m	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, J. Jung, H. Pacl, and H. Egenolf, *Eur. J. Org. Chem.* **1998**, 1297.

cyc-CH₂CHCHSi

In an argon matrix, irradiation at 405 or 366 nm results in isomerization to H₂CCHCHSi, whereas irradiation at 313 nm results in isomerization to H₂CCCHSiH.¹

 $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH stretch	3062.6w	Ar	IR	1
		CH ₂ stretch	3017.3w	Ar	IR	1
		CH ₂ stretch	2972.0m	Ar	IR	1
		CH stretch	2931.5w	Ar	IR	1
		CH ₂ scissors	1422.1w	Ar	IR	1
		CH deform.	1405.0w	Ar	IR	1
		CH deform.	1247.7vw	Ar	IR	1
		CH ₂ twist	1129.2wm	Ar	IR	1
		CH deform.	1028.6wm	Ar	IR	1
		CC stretch	993.0w	Ar	IR	1
		CH ₂ wag	970.2wm	Ar	IR	1
		CH deform.	879.2m	Ar	IR	1
		CH deform.	823.2s	Ar	IR	1
		SiC stretch	765.0vs	Ar	IR	1
		SiCC deform.	719.8wm	Ar	IR	1
		CCC deform.	549.3w	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, J. Jung, H. Pacl, and H. Egenolf, *Eur. J. Org. Chem.* **1998**, 1297.

H₃CCHCSi $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		SiCC stretch	1681.4	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, J. Jung, H. Pacl, and H. Egenolf, *Eur. J. Org. Chem.* **1998**, 1297.

H₂CCHCHSi

In an argon matrix, irradiation at 313 nm results in isomerization to H₂CCCHSiH.¹

 $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH ₂ stretch	3103.4m	Ar	IR	1
		CH ₂ stretch	3052.5w	Ar	IR	1
		CH stretch	3040.3vw	Ar	IR	1
		CH stretch	3010.6w	Ar	IR	1
		C=C stretch	1594.6m	Ar	IR	1
		CC stretch	1167.5m	Ar	IR	1
		CH bend	979.7m	Ar	IR	1
		CH ₂ wag	881.0vs 877.3	Ar	IR	1
		CH ₂ twist	541.2w	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, J. Jung, H. Pacl, and H. Egenolf, *Eur. J. Org. Chem.* **1998**, 1297.

H₃CCCSiH

In an argon matrix, irradiation at 313 or 254 nm results in isomerization to H₃CCHCSi and H₃CSiCCH.¹

 $\tilde{\chi}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH ₃ stretch	2982.0w	Ar	IR	1
		CH ₃ stretch	2933.0w	Ar	IR	1
		CH ₃ stretch	2860w	Ar	IR	1
		C≡C stretch	2151.1s 2142.7	Ar	IR	1
		SiH stretch	1970.8vs 1964.1	Ar	IR	1
		CH ₃ deform.	1442.5w	Ar	IR	1
		CH ₃ deform.	1437.6vw	Ar	IR	1
		SiH bend	826.7m	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, J. Jung, H. Pacl, and H. Egenolf, Eur. J. Org. Chem. **1998**, 1297.

H₃CSiCCH

In an argon matrix, irradiation at wavelengths longer than 385 nm results in isomerization to H₂CSiHCCH.¹

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH stretch	3308.6vs	Ar	IR	1
		C≡C stretch	2001.2ms	Ar	IR	1
		CH ₃ deform.	1430.0w	Ar	IR	1
		CH ₃ deform.	1218.8m	Ar	IR	1
		Si-CH ₃ stretch	620.2s	Ar	IR	1
		Si-CCH stretch	588.4vs	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, J. Jung, H. Pacl, and H. Egenolf, Eur. J. Org. Chem. **1998**, 1297.

H₂CCCHSiH

In an argon matrix, irradiation at 313 or 254 nm results in isomerization to H₃CCHCSi and H₃CSiCCH.¹

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH ₂ stretch	3023.9vw	Ar	IR	1
		CH stretch	2980.2vw	Ar	IR	1
		SiH stretch	1993.8vs	Ar	IR	1
			1988.3			
		CCC stretch	1915.5vs	Ar	IR	1
			1908.0			
		SiCC stretch	1184.4wm	Ar	IR	1
			1179.4			
		SiH bend	860.4wm	Ar	IR	1
		CH ₂ wag	798.4m	Ar	IR	1
		HSiCH deform.	682.3wm	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, J. Jung, H. Pacl, and H. Egenolf, Eur. J. Org. Chem. **1998**, 1297.

H₂CSiHCCH

In an argon matrix, irradiation at 313 nm results in isomerization to H₃CSiCCH.¹

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		≡CH stretch	3314.5vs	Ar	IR	1
		SiH stretch	2224.0s	Ar	IR	1
		C≡C stretch	2047.5s	Ar	IR	1
		CH ₂ scissors	1331.6m	Ar	IR	1
		SiH bend	858.2vs	Ar	IR	1
		CH ₂ wag	716.2m	Ar	IR	1
		CH ₂ rock	689.5s	Ar	IR	1
		≡CH bend	669.2ms	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, J. Jung, H. Pacl, and H. Egenolf, Eur. J. Org. Chem. **1998**, 1297.

H₃Si(cyc-CSiCH) \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH stretch	3017.2w	Ar	IR	1
		SiH stretch	2190T	Ar	IR	1
		CC stretch	1498.8wm	Ar	IR	1
			1484.0			
		CH deform.	1029.3	Ar	IR	1
			1026.6			
		SiH deform.	953.1wm	Ar	IR	1
		SiH deform.	921.8T	Ar	IR	1
		SiH deform.	919.3vs	Ar	IR	1
		SiC stretch	840.8wm	Ar	IR	1
		SiC stretch	749.7wm	Ar	IR	1
		SiCC deform.	665.7wm	Ar	IR	1
		Mixed	650.6wm	Ar	IR	1
			655.5			

D₃Si(cyc-CSiCD) \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		SiD stretch	1600T	Ar	IR	1
		CC stretch	1472.5w	Ar	IR	1
		CD deform.	911.2w	Ar	IR	1
		SiC stretch	803.2ms	Ar	IR	1
		SiD deform.	695.4	Ar	IR	1
		SiD deform.	694.4vs	Ar	IR	1
		SiD deform.	677.9w	Ar	IR	1
		SiC stretch	652.0w	Ar	IR	1
		SiCC deform.	522.0w	Ar	IR	1
		CD deform.	508.3	Ar	IR	1
			505.8			

Reference

¹G. Maier, H. P. Reisenauer, and A. Meudt, Eur. J. Org. Chem. **1998**, 1291.

HSiCCSiH₃

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		SiH stretch	2200Tw	Ar	IR	1
		SiH stretch	2030.2vw	Ar	IR	1
			2006.6			
		SiH deform.	938.3vw	Ar	IR	1
		SiH deform.	922.5vs	Ar	IR	1
		SiH deform.	808.0vw	Ar	IR	1
		SiC stretch	784.0s	Ar	IR	1
			780.0			
		Mixed	680.2vw	Ar	IR	1

DSiCCSiD₃

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		SiD stretch	1600Tw	Ar	IR	1
		SiD stretch	1477.0vw	Ar	IR	1
			1458.7			
		SiC stretch	782.8vs	Ar	IR	1
		SiD deform.	686.1s	Ar	IR	1
		SiD deform.	622.0vw	Ar	IR	1
		Mixed	531.0vw	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, and A. Meudt, Eur. J. Org. Chem. **1998**, 1291.

HCCSiSiH₃

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH stretch	3294.1wm	Ar	IR	1
			3291.1			
		SiH ₃ stretch	2168T	Ar	IR	1
		SiH ₃ deform.	856.0vs	Ar	IR	1
			854.5			
		HCC deform.	690.1wm	Ar	IR	1
		SiC stretch	586.0m	Ar	IR	1

Reference

¹G. Maier, H. P. Reisenauer, and A. Meudt, Eur. J. Org. Chem. **1998**, 1291.

H₂C₅H

\tilde{X} C_{2v}
 $1/2(B_0 + C_0) = 0.072$ MW¹

Reference

¹W. Chen, S. E. Novick, M. C. McCarthy, and P. Thaddeus, J. Chem. Phys. **109**, 10190 (1998).

C₆H₂⁺

$\tilde{B}^2\Pi_u$ $D_{\infty h}$
 $T_0 = 27350(160)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			650(80)	gas	PE	1

$\tilde{A}^2\Pi_g$ $D_{\infty h}$
 $T_0 = 16654.69$ gas EF^{2,3}LF³AB^{6,7} $\tilde{A}-\tilde{X}$ 485–725 nm
 16570 Ne LF⁴AB⁵ $\tilde{A}-\tilde{X}$ 524–843 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	CH stretch	3243(2)T	gas	LF	3
	2	C≡C stretch	2053(2)	gas	LF	3
			2065.7	Ne	LF,AB	4,5
	3	C≡C stretch	1880(2)	gas	LF	3
			1890.2	Ne	LF,AB	4,5
	4	C–C stretch	617.8	gas	LF,EF	2,3,7
			620	Ne	LF,AB	4,5
Π_g	10	Skel. bend	244T	gas	LF	3
			248H	Ne	LF	4

$\tau_0 = 17(2)$ ns gas EF²
 $A_0 = -28.41(28)$ AB⁶
 $B_0 = 0.044$ AB⁶

$\tilde{X}^2\Pi_u$ $D_{\infty h}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	2	C≡C stretch	2182(2)	gas	EF	2,3
			2181.4	Ne	LF	4
	3	C≡C stretch	1903(2)	gas	EF	2,3
			1903.4	Ne	LF	4
	4	C–C stretch	632(2)	gas	EF	2,3
			635H	Ne	LF	4
Σ_u^+	6	C≡C stretch	2014T	Ne	LF	4
Π_g	10	Skel. bend	228(2)T	gas	LF	3
			230T	Ne	LF	4

$A_0 = -31.40(28)$ AB⁶
 $B_0 = 0.045$ AB⁶

C₆D₂⁺

$\tilde{A} \ ^2\Pi_g$ D_{∞h}
 $T_0 = 16686.51$ gas AB⁶
 $A_0 = -28.40(49)$ AB⁶
 $B_0 = 0.040$ AB⁶

$\tilde{X} \ ^2\Pi_u$ D_{∞h}
 $A_0 = -31.31(49)$ AB⁶
 $B_0 = 0.041$ AB⁶

References

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- ³D. Klapstein, R. Kuhn, J. P. Maier, M. Ochsner, and W. Zambach, *J. Phys. Chem.* **88**, 5176 (1984).
- ⁴A. M. Smith, J. Agreiter, M. Härtle, C. Engel, and V. E. Bondybey, *Chem. Phys.* **189**, 315 (1994).
- ⁵P. Freivogel, J. Fulara, D. Lessen, D. Forney, and J. P. Maier, *Chem. Phys.* **189**, 335 (1994).
- ⁶W. E. Sinclair, D. Pfluger, H. Linnartz, and J. P. Maier, *J. Chem. Phys.* **110**, 296 (1999).
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H₂C₆:

\tilde{X} C_{2v}
 $A_0 = 8.95$; $B_0 = 0.0450$; $C_0 = 0.0447$ MW¹

Reference

- ¹M. C. McCarthy, M. J. Travers, A. Kovács, W. Chen, S. E. Novick, C. A. Gottlieb, and P. Thaddeus, *Science* **275**, 518 (1997).

F₂C=(cyc-C₃H₂) \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁		C=C stretch	1906.9wm	Ar	IR	1
		Ring stretch	1563.2wm	Ar	IR	1
		Deformation	1198.1vs	Ar	IR	1
		Breathing	705w	Ar	IR	1
<i>b</i> ₁		H ₂ deform.	623.6wm	Ar	IR	1
<i>b</i> ₂		CF ₂ a-stretch	1215s	Ar	IR	1
		HCCCH rock	1038.3w	Ar	IR	1

Reference

- ¹C. Kötting, W. Sander, and M. Senzlober, *Chem. Eur. J.* **4**, 2360 (1998).

F₂C=CHCCH \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1731.1vs	Ar	IR	1
			1348.1ms	Ar	IR	1
			1223.3m	Ar	IR	1
			914.2wm	Ar	IR	1
			659.0w	Ar	IR	1

Reference

- ¹C. Kötting, W. Sander, and M. Senzlober, *Chem. Eur. J.* **4**, 2360 (1998).

F₂C=C=C=CH₂ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			2114.3s	Ar	IR	1
			1703.5vs	Ar	IR	1
			1404.8m	Ar	IR	1
			1239.4s	Ar	IR	1
			1167.1m	Ar	IR	1

Reference

- ¹C. Kötting, W. Sander, and M. Senzlober, *Chem. Eur. J.* **4**, 2360 (1998).

C₇H

$\tilde{X} \ ^2\Pi_{1/2}$ C_{∞v}
 $A_{\text{eff}} = 26.17$; $B_0 = 0.029$ MW^{1,2}

C₇D

$\tilde{X} \ ^2\Pi_{1/2}$ C_{∞v}
 $A_{\text{eff}} = 26.15$; $B_0 = 0.028$ MW^{1,2}

References

- ¹M. J. Travers, M. C. McCarthy, C. A. Gottlieb, and P. Thaddeus, *Astrophys. J.* **465**, L77 (1996).
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SiC₆H

$\tilde{X} \ ^2\Pi$ C_{∞v}
 $B = 0.019$ MW¹

Reference

- ¹M. C. McCarthy, A. J. Apponi, C. A. Gottlieb, and P. Thaddeus, *J. Chem. Phys.* **115**, 870 (2001).

HC₆N

$\tilde{A}^3\Sigma^-$ $C_{\infty v}$
 $T_0 = 21208.60(5)$ gas CR²
 21181(10) Ne AB² $\tilde{A}-\tilde{X}$ 432–472 nm
 $\tilde{A}-\tilde{X}$ 432–472 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		CC stretch	1942.5(5)	gas	CR	2
			1950(15)	Ne	AB	2
Π		Bend	234HT	Ne	AB	2

$\tilde{X}^3\Sigma^-$ $C_{\infty v}$
 $B_0 = 0.028$ MW¹

DC₆N

$\tilde{A}^3\Sigma^-$ $C_{\infty v}$
 $T_0 = 21282.10(5)$ gas CR²
 21239(10) Ne AB² $\tilde{A}-\tilde{X}$ 431–471 nm
 $\tilde{A}-\tilde{X}$ 431–471 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		CC stretch	1926.0	gas	CR	2
			1935(15)	Ne	AB	2
Π		Bend	172HT	Ne	AB	2

References

- ¹V. D. Gordon, M. C. McCarthy, A. J. Apponi, and P. Thaddeus, *Astrophys. J.* **540**, 2886 (2000).
²O. Vaizert, T. Motylewski, M. Wyss, E. Riaplov, H. Linnartz, and J. P. Maier, *J. Chem. Phys.* **114**, 7918 (2001).

(cyc-HC₃)CCCN

\tilde{X} C_s
 $A_0 = 1.150$; $B_0 = 0.035$; $C_0 = 0.034$ MW¹

Reference

- ¹M. C. McCarthy, J.-U. Grabow, M. J. Travers, W. Chen, C. A. Gottlieb, and P. Thaddeus, *Astrophys. J.* **513**, 305 (1999).

Nb₈

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			180(15)	gas	PE	1

Reference

- ¹T. P. Marcy and D. G. Leopold, *Int. J. Mass Spectrom.* **195/196**, 653 (2000).

Nb₈⁻

Threshold for electron detachment from ground-state Nb₈⁻ = 12210(65) gas PE¹

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			165(20)	gas	PE	1

Reference

- ¹T. P. Marcy and D. G. Leopold, *Int. J. Mass Spectrom.* **195/196**, 653 (2000).

Ag₈

The resonant two-photon ionization of helium clusters containing Ag₈ shows a narrow peak at 31950 (3.96 eV) and a broader peak at 32190 (3.99 eV).³

In solid neon, the corresponding absorption maximum for Ag₈ has been observed² at 31550 (3.91 eV). In solid argon, this maximum appears¹ at 31390 (3.89 eV), and a fluorescence maximum has been detected⁴ at 31220 (3.87 eV).

References

- ¹W. Harbich, S. Fedrigo, and J. Buttet, *Z. Phys. D* **26**, 138 (1993).
²W. Harbich, Y. Belyaev, R. Kleiber, and J. Buttet, *Surf. Rev. Lett.* **3**, 1147 (1996).
³F. Federmann, K. Hoffmann, N. Quaa, and J. P. Toennies, *Eur. Phys. J. D* **9**, 11 (1999).
⁴C. Félix, C. Sieber, W. Harbich, J. Buttet, I. Rabin, W. Schulze, and G. Ertl, *Phys. Rev. Lett.* **86**, 2992 (2001).

C₈

(2) $^3\Sigma_u^-$ $D_{\infty h}$
 $T_0 = 36071(26)$ Ne AB⁹

$T_0 = 32934(22)$ Ne AB⁹ $D_{\infty h}$ 285–304 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	3		1639(34)	Ne	AB	9
	4		473(32)	Ne	AB	9

$^3\Sigma_u^-$ $D_{\infty h}$
 $T_0 = 15630(5)$ Ne AB^{3,9} $^3\Sigma_u^- - \tilde{X}$ 550–640 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1		2075(5)	Ne	AB	3
	4		471(5)	Ne	AB	3

$^3\Pi_g$ $D_{\infty h}$
 $T_0 = 8310(160)$ gas PE⁸

$^3\Sigma_u^+$ $D_{\infty h}$
 $T_0 = 6290(160)$ gas PE⁸

$1\Sigma_g^+$ $D_{\infty h}$
 $T_0 = 930(120)$ gas PE⁸

$1\Delta_g$ $D_{\infty h}$
 $T_0 = 570(120)$ gas PE⁸

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1903(50)	gas	PE	8
	3	Sym. stretch	1355(50)T	gas	PE	8

$\tilde{X}^3\Sigma_g^-$ $D_{\infty h}$ Structure: ESR¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	Sym. stretch	1928(50)	gas	PE	8
	3	Sym. stretch	1361(50)	gas	PE	8
Σ_u^-	5		2067.8	Ne	IR	3,6,7
			2063.9	Ar	IR	4
			2065.3	Kr	IR	5
	6		1707.8	Ne	IR	4-6
			1705.6	Ar	IR	4
		1706.0T	Kr	IR	5	

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- M. Grutter, M. Wyss, E. Riaplov, J. P. Maier, S. D. Peyerimhoff, and M. Hanrath, *J. Chem. Phys.* **111**, 7397 (1999).

cyc-C₈

\tilde{X} D_{4h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e_u	12	Stretch	1844.2	Ar	IR	1,2

References

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- S. L. Wang, C. M. L. Rittby, and W. R. M. Graham, *J. Chem. Phys.* **112**, 1457 (2000).

C₈⁻

Threshold for electron detachment from ground-state
 $C_8^- = 35330(50)$ gas PE^{1,2,6}

(2) $2\Pi_u$ $D_{\infty h}$
 $T_0 = 16305$ gas PD⁴MPD⁸ 446-614 nm
 16295(5) Ne AB⁷ 438-614 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1		2104	gas	MPD	8
			2125(10)	Ne	AB	7
	2		1821(10)	Ne	AB	7
			469(8)	Ne	AB	7
Π_g	8		487(5)H	Ne	AB	7

$\tilde{C}^2\Pi_u$ $D_{\infty h}$
 $T_0 = 13014$ gas PD⁴MPD⁸ $\tilde{C}-\tilde{X}$ 639-774 nm
 12933(5) Ne AB³ $\tilde{C}-\tilde{X}$ 665-775 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1		2087	gas	PD	4
	2		2039	gas	MPD	8
			2055(5)	Ne	AB	3
	3		959(5)	Ne	AB	3
4			450	gas	PD, MPD	4,8
			475(5)	Ne	AB	3

$2\Sigma_u^+$ $D_{\infty h}$
 $T_0 = 9545(2)$ Ne AB⁷ 817-1048 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1		2010(4)	Ne	AB	7
	2		1956(4)	Ne	AB	7

$\tilde{X}^2\Pi_g$ $D_{\infty h}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_u^+	6		1796.0	Ne	IR	5

References

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- M. Tulej, D. A. Kirkwood, M. Pachkov, and J. P. Maier, *Astrophys. J.* **506**, L69 (1998).

SiC₇

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	1		2100.8	Ar	IR	1

Reference

¹X. D. Ding, S. L. Wang, C. M. L. Rittby, and W. R. M. Graham, *J. Phys. Chem. A* **104**, 3712 (2000).

C₇N

$^2\Pi$		$C_{\infty v}$				
$T_0=17185(6)$		Ne AB ¹		$^2\Pi-\tilde{X}$ 482–582 nm		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		CC,CN stretch	2086(8)	Ne	AB	1
		C–C stretch	1630(8)	Ne	AB	1
		C–C stretch	739(8)	Ne	AB	1
		C–C stretch	523(8)T	Ne	AB	1
Π		Deformation	134(5)H	Ne	AB	1

Reference

¹M. Grutter, M. Wyss, and J. P. Maier, *J. Chem. Phys.* **110**, 1492 (1999).

NC(C≡C)₂CN⁺

$\tilde{E}^2\Pi_u$		$D_{\infty h}$				
$T_0=30420(160)$		gas PE ¹				
$\tilde{D}^2\Pi_g$		$D_{\infty h}$				
$T_0=25580(160)$		gas PE ¹				
$\tilde{B}, \tilde{C}^2\Sigma_g^+, ^2\Sigma_u^+$		$D_{\infty h}$				
$T_0=22190(160)$		gas PE ¹				
$\tilde{A}^2\Pi_u$		$D_{\infty h}$				
$T_0=15245.74$		gas EF ¹ AB ⁴ CR ⁴		$\tilde{A}-\tilde{X}$ 630–770 nm		
		15130 ^a Ne AB ² LF ³		$\tilde{A}-\tilde{X}$ 565–1510 nm		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	C≡C s-stretch	2475(5)T	Ne	AB	2
	2	CN s-stretch	1940(80)	gas	PE	1
			2029	Ne	AB,LF	2,3
	3	C–C s-stretch	1726(5)	Ne	AB	2
	4	C–C s-stretch	449	Ne	AB,LF	2,3

$B_0=0.019$ AB⁴CR⁴

$\tilde{X}^2\Pi_g$		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1	C≡N s-stretch	2180(10)	gas	EF	1
			2185	Ne	LF	3
	2	C≡C s-stretch	2100(10)	gas	EF	1
			2094	Ne	LF	3
	3	C–C s-stretch	1360(10)	gas	EF	1
			1354	Ne	LF	3
	4	C–C s-stretch	460(10)	gas	EF	1
			458	Ne	LF	3

$B_0=0.019$ AB⁴CR⁴

^aRef. 3 reported a second trapping site with $T_0=15160$.

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C₇N⁻

$^1\Sigma^+$		$C_{\infty v}$				
$T_0=36496(27)$		Ne AB ¹		$^1\Sigma^+-\tilde{X}$ 249–274 nm		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		CC,CN stretch	2054(38)	Ne	AB	1
		C–C stretch	1599(38)	Ne	AB	1
		C–C stretch	1069(38)	Ne	AB	1
		C–C stretch	472(38)	Ne	AB	1
$\tilde{X}^1\Sigma^+$		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+			2124	Ne	IR	1
			2074	Ne	IR	1

Reference

¹M. Grutter, M. Wyss, and J. P. Maier, *J. Chem. Phys.* **110**, 1492 (1999).

C₇O

\tilde{X}		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			2244.2vsT	Ar	IR	2
			2198.3sT	Ar	IR	2

$B_0=0.019$ MW¹

References

- ¹T. Ogata, Y. Ohshima, and Y. Endo, *J. Am. Chem. Soc.* **117**, 3593 (1995).
²M. Dibben, J. Szczepanski, C. Wehlburg, and M. Vala, *J. Phys. Chem. A* **104**, 3584 (2000).

F₂C=C=CF₂: \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		C ₃ a-stretch (cis)	2006.8	Ar	IR	1
		C ₃ a-stretch (trans)	1995.6ms	Ar	IR	1
		C ₃ s-stretch	1444.4wm	Ar	IR	1
		CF ₂ a-stretch	1268.0m	Ar	IR	1
		:CF stretch (cis)	1194.5	Ar	IR	1
		:CF stretch (trans)	1186.5s	Ar	IR	1
		=CF stretch	1174.2vs	Ar	IR	1

Reference

- ¹C. Kötting, W. Sander, and M. Senzlober, *Chem. Eur. J.* **4**, 2360 (1998).

F₂C=(cyc-C₃F₂) \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁		Skel. breathing	2000.1wm	Ar	IR	1
			1241.9m	Ar	IR	1
			874.5ms	Ar	IR	1
b ₂		FCCF a-stretch	1317.8vs	Ar	IR	1
			CF ₂ a-stretch	1252.8ms	Ar	IR

Reference

- ¹C. Kötting, W. Sander, and M. Senzlober, *Chem. Eur. J.* **4**, 2360 (1998).

8.15. Hydrocarbons With More Than Eight Atoms**t-H₂C=CH-CH=CH₂⁺** \tilde{E} T₀=49210(80) gas PE^{1,5}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1130(80)	gas	PE	5

 \tilde{D} T^a=39000(80) gas PE⁵ \tilde{C} T₀=33000(80) gas PE^{1,5}

An absorption maximum which appears at 33890 (295 nm) in an argon matrix has been attributed² to *t*-butadiene cation.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1290(80)	gas	PE	5

 \tilde{B} T^a=25250(80) gas PE^{1,5} \tilde{A} T₀=18720(80) gas PE^{1,5}

A relatively weak absorption maximum which appears at 18720 (534 nm) in an argon matrix has been attributed² to *t*-butadiene cation.

 \tilde{X} C_{2h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.		
a _g	4	C=C stretch	1620(2)	gas	PE,TPI	5-7		
	5		1479(2)	gas	TPI	6		
	6		1281(2)	gas	TPI	6		
	7		1259(2)	gas	PE,TPI	5-7		
	8		923(2)	gas	TPI	6		
	9		511(2)	gas	PE,TPI	5-7		
	a _u		11	1000.9	Ar	IR	3,4	
			b _g	14	1000(2)	gas	TPI	6
				15	901(2)	gas	TPI	6
b _u	17	3125.3	Ar	IR	3,4			
	18	3073.7	Ar	IR	3,4			
	19	3026.3	Ar	IR	3,4			
	20	1477.5	Ar	IR	3,4			
	21	1330T	Ar	IR	3,4			
	22	1251.1	Ar	IR	3,4			
	23	1006.1	Ar	IR	3,4			

t-D₂C=CD-CD=CD₂⁺ \tilde{X} C_{2h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a _u	11		739.6	Ar	IR	3,4
b _u	17		2358.1	Ar	IR	3,4
	18		2276.1	Ar	IR	3,4
	19		2215.4	Ar	IR	3,4
	20		1319.1	Ar	IR	3,4
	21		1047.2	Ar	IR	3,4
	22		996.1	Ar	IR	3,4

^aFrom vertical ionization potential.

References

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⁶J. D. Hofstein and P. M. Johnson, *Chem. Phys. Lett.* **316**, 229 (2000).

⁷J. Liu and S. L. Anderson, *J. Chem. Phys.* **114**, 6618 (2001).

H₂CCHC(CH₃):

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1454.6m	Xe	IR	1
			1421.5m	Xe	IR	1
			1402.8w	Xe	IR	1
			1354.4vw	Xe	IR	1
			874.6m	Xe	IR	1
			767.1s	Xe	IR	1

Reference

¹G. Maier, C. Lautz, and S. Senger, *Chem. Eur. J.* **6**, 1467 (2000).

cyc-C₅H₅

$\tilde{A} \ ^2A_2''$ D_{5h} Structure: LF⁹
 $T_0 = 29572.166(2)$ gas AB^{1,3,4}LF^{6,8,11,12} $\tilde{A}-\tilde{X}$ 306–395 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1'	2		1055	gas	LF	11,12
e_1''	8	CH deform.	166H	gas	AB,LF	1,3,4,9,11,12
e_2'	11		1019	gas	LF	11,12
	12		797	gas	AB,LF	4,9,11,12

$\tau_0 = 64.9(5)$ ns gas LF^{7,11}
 $B_0 = 0.286$; $C_0 = 0.144$ LF^{8,9}

$\tilde{X} \ ^2E_1''$ D_{5h} Structure: ESR²AB⁴LF⁹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1'	2		1071	gas	LF	12
a_2'	3		1244	gas	LF	12
a_2''	4		681	gas	LF	12
			661vs	Ar	IR	10
e_1'	5	CH stretch	3079wm	Ar	IR	10
	6		1383m	Ar	IR	10
	7		1001	gas	LF	12
e_1''	8		766H	gas	LF	12
e_2'	10		1320	gas	LF	12
	11		1041	gas	LF	12
	12		872	gas	PD,LF	5,6,9,12
e_2''	13		861H	gas	LF	12
	14		576H	gas	LF	12

$B_0 = 0.296$; $C_0 = 0.148$ LF^{8,9}

cyc-C₅D₅

$\tilde{A} \ ^2A_2''$ D_{5h}
 $T_0 = 29819.434(2)$ gas AB⁴LF^{9,12} $\tilde{A}-\tilde{X}$ 319–365 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1'	2		994	gas	LF	12
e_1''	8	CD deform.	116H	gas	AB,LF	4,12
e_2'	11		798T	gas	LF	12
e_2''	14		296H	gas	LF	12

$B_0 = 0.233$; $C_0 = 0.117$ LF⁹

$\tilde{X} \ ^2E_1''$ D_{5h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1'	2		1045	gas	LF	12
a_2''	4		562H	gas	LF	12
e_1''	8		616H	gas	LF	12
e_2'	10		1353	gas	LF	12
	11		861	gas	LF	12
e_2''	13		743H	gas	LF	12
	14		414H	gas	LF	12

$B_0 = 0.240$; $C_0 = 0.120$ LF⁹

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o-C₆H₄

(o-Benzynes)

Absorption maxima at 40650 and 50250 (246 and 199 nm) have been reported^{6,7} for o-benzyne isolated in an argon matrix.

An absorption maximum at 26320 (380 nm) has been assigned^{6,7} to o-benzyne isolated in an argon matrix. In a neon matrix, this absorption maximum appears at 28730 (348 nm).⁸

$\tilde{a} \ ^3B_2$ C_{2v}
 $T_0 = 13140(50)$ gas PE^{4,9}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1			1520(40)	gas	PE	9
			1395(30)	gas	PE	4,9
			1275(30)	gas	PE	9
			560(20)	gas	PE	4,9

\tilde{X}^1A_1		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CH stretch	3094wm	Ne	IR	8
			3088	N ₂	IR	2
	2	CH stretch	3071w	Ne	IR	8
			3	C=C stretch	1860(15)	gas
	1846w	Ne			IR	8
	4	Mixed	1415vw	Ne	IR	8
			5	Mixed	1271w	Ne
	6	CH deform.			1055wm	Ne
			1053m	Ar	IR	1
7	Ring stretch	1056	N ₂	IR	2,5	
		1010(40)	gas	PE	4,9	
8	Ring stretch	1039m	Ne	IR	8	
		1038m	Ar	IR	1	
9	Ring deform.	1039	N ₂	IR	2,5	
		982wm	Ne	IR	8	
14	CH wag	838vw	Ne	IR	8	
		15	CH wag	737s	Ne	IR
736vs	Ar			IR	1	
16	Ring torsion	743	N ₂	IR	2,5	
		388wm	Ne	IR	8	
17	CH stretch	3086m	Ne	IR	8	
		18	CH stretch	3049vw	Ne	IR
19	Ring stretch			1451m	Ne	IR
		1451m	Ar	IR	1	
20	Mixed	1448	N ₂	IR	2,5	
		1394wm	Ne	IR	8	
21	Mixed	1395	N ₂	IR	5	
		1355	N ₂	IR	5	
22	Mixed	1307vw	Ne	IR	8	
		1094w	Ne	IR	8	
23	Ring deform.	849ms	Ne	IR	8	
		849s	Ar	IR	1	
24	Ring deform.	847	N ₂	IR	2,5	
		472vs	Ne	IR	8	
24	Ring deform.	469vs	Ar	IR	1	
		472	N ₂	IR	2,5	

$A_0=0.233$; $B_0=0.190$; $C_0=0.105$ MW³

$o-C_6D_4$

\tilde{a}^3B_2		C_{2v}				
$T_0=13180(100)$		gas	PE ^{4,9}			
Vib.	No.	Approximate	cm ⁻¹	Med.	Type	Refs.
a_1			1515(30)	gas	PE	9
			1265(30)	gas	PE	4,9
			550(30)	gas	PE	4

\tilde{X}^1A_1		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CD stretch	2311wm	Ne	IR	8
			2293	N ₂	IR	2
	2	CD stretch	2295vw	Ne	IR	8
			3	C=C stretch	1860(15)	gas
	1844vw	Ne			IR	8
	4	Mixed	1364w	Ne	IR	8
			5	Mixed	1198w	Ne
	6	Ring stretch			980(20)	gas
			995wm	Ne	IR	8
7	CD deform.	853w	Ne	IR	8	
		8	Ring stretch	792wm	Ne	IR
9	Ring deform.			580(20)	gas	PE
		579vw	Ne	IR	8	
14	CD wag	679vw	Ne	IR	8	
		15	CD wag	571m	Ne	IR
16	Ring torsion			336w	Ne	IR
		17	CD stretch	2314wm	Ne	IR
18	CD stretch			2285vw	Ne	IR
		19	Ring stretch	1411w	Ne	IR
20	Mixed			1293w	Ne	IR
		1293	N ₂	IR	2	
21	Mixed	1112vw	Ne	IR	8	
		1108	N ₂	IR	2	
22	Mixed	875vw	Ne	IR	8	
		23	Ring deform.	790m	Ne	IR
792	N ₂			IR	2	
24	Ring deform.	469vs	Ne	IR	8	
		471	N ₂	IR	2	

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$m-C_6H_4$

(*m*-Benzene)

\tilde{a}		C_{2v}				
$T_0=7350(110)$		gas	PE ²			
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1			970(15)	gas	PE	2

\bar{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1		CH stretch	3037w	Ar	IR	1
		Mixed	1402wm	Ar	IR	1
			300T	gas	PE	2
b_1		CH OPLA	824wm	Ar	IR	1
		CH OPLA	751ms	Ar	IR	1
b_2		Mixed	1486wm	Ar	IR	1
		HCC deform.	936m	Ar	IR	1
		Ring deform.	547vs	Ar	IR	1

 $m-C_6D_4$

\bar{a}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1			955(20)	gas	PE	2

\bar{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1			300T	gas	PE	2

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 $p-C_6H_4$ *(p*-Benzynes)

\bar{a}		D_{2h}				
$T_0=1330(130)$		gas PE ²				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_g	4		995(20)	gas	PE	2
	5		610(15)	gas	PE	2

\bar{X}		D_{2h}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_g	4		990(20)	gas	PE	2
	5		635(20)	gas	PE	2
	10		1403w	Ne	IR	1
b_{1u}	11		976w	Ne	IR	1
			980s	Ar	IR	1
	12	Ring deform.	918w	Ne	IR	1
b_{2u}	16		1331w	Ne	IR	1
	17		1207w	Ne	IR	1
b_{3u}	23	CH wag	721s	Ne	IR	1
			725s	Ar	IR	1
	24	Ring torsion	435w	Ne	IR	1

 $p-C_6D_4$

\bar{a}		D_{2h}				
$T_0=1350(130)$		gas PE ²				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_g	4		945(40)	gas	PE	2
	5		590(20)	gas	PE	2

\bar{X}		D_{2h}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_g	5		615(20)	gas	PE	2
b_{1u}	12	Ring deform.	767	Ar	IR	1

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 $o-C_6H_4^-$

Threshold for electron detachment from ground-state $o-C_6H_4^- = 4550(55)$ gas PE^{1,2}

\bar{X}^2B_2		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1			600(30)	gas	PE	1,2

 $o-C_6D_4^-$

Threshold for electron detachment from ground-state $o-C_6D_4^- = 4490(65)$ gas PE^{1,2}

\bar{X}^2B_2		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1			580(30)	gas	PE	1

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***m*-C₆H₄⁻**

Threshold for electron detachment from ground-state *m*-C₆H₄⁻ 6870(90) in the gas phase, estimated from collision-induced dissociation branching ratios.¹

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			885(20)	gas	PE	1

***m*-C₆D₄⁻** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			910(40)	gas	PE	1

Reference

¹P. G. Wenthold, R. R. Squires, and W. C. Lineberger, *J. Am. Chem. Soc.* **120**, 5279 (1998).

***p*-C₆H₄⁻**

Threshold for electron detachment from ground-state *p*-C₆H₄⁻ = 10210(65) gas PE¹

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			615(30)	gas	PE	1

***p*-C₆D₄⁻**

Threshold for electron detachment from ground-state *p*-C₆D₄⁻ = 10170(65) gas PE¹

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			605(30)	gas	PE	1

Reference

¹P. G. Wenthold, R. R. Squires, and W. C. Lineberger, *J. Am. Chem. Soc.* **120**, 5279 (1998).

C₆H₅⁺ \tilde{a}^3B_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁		CH stretch	2790(100)	gas	PE	1

 \tilde{X}^1A_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>b</i> ₁		CH deform.	713	Ar	IR	2
<i>b</i> ₂		CH stretch	3110	Ar	IR	2

C₆D₅⁺ \tilde{a}^3B_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁		CH stretch	2370(110)	gas	PE	1

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C₆H₅ \tilde{C}^2B_2 C_{2v}*T*₀ = 47281 Ar AB¹¹ \tilde{B}^2A_1 C_{2v}

A broad gas-phase absorption with maximum near 40820 (245 nm) has been attributed⁷ to C₆H₅. It may correspond with a broad, unstructured absorption maximum observed¹¹ at 42535 (235 nm) in an argon matrix.

 \tilde{A}^2B_1 C_{2v}*T*₀ = 18908 gas AB¹
19589 Ar AB¹¹ $\tilde{A}-\tilde{X}$ 440–530 nm

This weak absorption is overlapped by more prominent absorption of C₆H₅O.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			896	gas	AB	1
			722	gas	AB	1
			571	gas	AB	1

Gas-phase photoelectron spectra⁹ suggest that there is an excited electronic state of C₆H₅ at or below 13700.

\tilde{X}^2A_1		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CH s-stretch	3086w	Ar	IR,Ra	6,8,10,12,13,14
	2	CH s-stretch	3072vw	Ar	IR,Ra	10,12,13,14
	3	CH s-stretch	3037vw	Ar	IR,Ra	13,14
	4	Ring stretch	1497vw	Ar	IR,Ra	13,14
	5	Ring stretch	1441wm	Ar	IR,Ra	8,10,12,13,14
	6	CH deform.	1154vw	Ar	IR,Ra	13,14
	7	Ring deform.	1027wm	Ar	IR,Ra	6,8,10,12,13,14
	8	Ring breathing	997vw	Ar	IR,Ra	13,14
	9	Ring deform.	968(15)	gas	PE	9
			976w	Ar	IR,Ra	8,10,13,14
	10	Ring deform.	600(10)	gas	PE	9
			605w	Ar	IR,Ra	8,10,13,14
a_2	11	OPLA CH deform.	945T	Ar	Ra	14
	12	OPLA CH deform.	816T	Ar	Ra	14
b_1	14	CH deform.	990vw	Ar	IR,Ra	10,13,14
	15	CH deform.	874w	Ar	IR,Ra	10,13,14
	16	CH deform.	706vs	Ar	IR,Ra	2-6,8,10,12-14
	17	CH deform.	657w	Ar	IR,Ra	8,12-14
	18	Deformation	416w	Ar	IR,Ra	10,13,14
b_2	19	CH a-stretch	3071wm	Ar	IR,Ra	8,10,12-14
	20	CH a-stretch	3060vw	Ar	IR,Ra	8,10,13,14
	21	Ring stretch	1593w	Ar	IR,Ra	13,14
	22	Ring stretch	1432wm	Ar	IR,Ra	6,8,10,12-14
	23	CH deform.	1310vw	Ar	IR,Ra	13,14
	24	Ring deform.	1283vw	Ar	IR,Ra	13,14
	25	CH deform.	1159vw	Ar	IR,Ra	13,14
	26	Ring deform.	1063w	Ar	IR,Ra	8,10,13,14
	27	Ring stretch	587vw	Ar	IR,Ra	8,10,13,14

C₆D₅

\tilde{X}^2A_1		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1	CD stretch	2292	Ar	IR	13
	2	CD stretch	2290	Ar	IR	13
	3	CD stretch	2282	Ar	IR	13
	4		1494	Ar	IR	13
	5		1314	Ar	IR	13
	6		960(20)	gas	PE	9
			851	Ar	IR	13
	7		803	Ar	IR	13
	10		588(10)	gas	PE	9
			590	Ar	IR	13
b_1	17	CD deform.	517s	Ar	IR	2,5,13
	18		382	Ar	IR	13
b_2	19	CD stretch	2271	Ar	IR	13
	20	CD stretch	2264	Ar	IR	13
	21		1561	Ar	IR	13
	22		1312	Ar	IR	13
	23		1297	Ar	IR	13
	26		806	Ar	IR	13
	27		547	Ar	IR	13

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C₆H₆⁺

\tilde{F}^2A_{1g}		D_{6h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_{1g}	1 ^b (2)	CH stretch	2790(100)	gas	PE	1
	2(1)	Ring stretch	960(10)	gas	PE	1,16

\tilde{E}^2B_{1u}		D_{6h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_{1g}	1 ^b (2)	CH stretch	1694(8)	gas	PE	16

\tilde{D}^2B_{2u}		D_{6h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_{1g}	1 ^b (2)	CH stretch	1694(8)	gas	PE	16

\tilde{C}^2E_{1u}		D_{6h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_{1g}	1 ^b (2)	CH stretch	1694(8)	gas	PE	16

\tilde{B}^2A_{2u}		D_{6h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e_{2g}	18 (6)	Ring deform.	508T	gas	PE	16

\tilde{B}^2A_{2u}		D_{6h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e_{2g}	18 (6)	Ring deform.	508T	gas	PE	16

In the gas phase, the mass-selected ion-dip spectrum of C₆H₆⁺ shows a broad, unstructured absorption with onset near 19000 and increasing in intensity up to the detection limit of 24000.¹⁰

A broad, unstructured absorption near 24000 may have been contributed by the $\tilde{B}-\tilde{X}$ transition of C₆H₆⁺ produced by vacuum-ultraviolet photolysis of benzene isolated in a neon matrix.²

$T_0 \approx 19840$ Ar LF³AB⁴ 420–547 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e_{2g}	18 ^b (6)	Ring deform.	600(30)T	Ar	AB	4

\tilde{A}^2E_{2g} D_{6h}
 $T_0 = 18117$ gas $PF^9PRI^{13,15}PE^{16}$ $\tilde{A}-\tilde{X}$ 510–550 nm
 In an argon matrix, a weak, sharp absorption at 18100 has been attributed² to a vibronically allowed transition in the excitation of $C_6H_6^+$ to the \tilde{A} state.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_{1g}	1 ^b (2)	CH stretch	2856(10)	gas	PE	1,16
	2 (1)	Ring stretch	976(8) ^c	gas	PF,PE	9,16
a_{2u}	4 (11)	CH bend (E_{2u})	506	gas	PRI	15
e_{1u}	14 (18)	CH bend (B_{1u})	551	gas	PRI	15
e_{2g}	16 (8)	Ring stretch	1520(100)T	gas	PE	1
	17 (9)	CH bend	1140 ^e	gas	PF	9
e_{2u}	18 (6)	Ring deform.	617 ^d	gas	PE,PRI	1,15,16
	19 (17)	CH bend (E_{2u})	635	gas	PF,PRI	9,15
		(A_{2u})	579			
20 (16)	Ring deform. (E_{2u})	300	gas	PF,PRI	9,15	
	(A_{2u})	228				

\tilde{X}^2E_{1g} D_{6h} Structure: TPE¹¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_{1g}	1 ^b (2)	CH stretch	2960(100)T	gas	PE	1
	2 (1)	Ring stretch	966	gas	PE,MPI	1,6,8,16
b_{2g}	8 (4)	Ring deform.	422 ^e	gas	PE,MPI	6,15
e_{1g}	11 (10)	CH bend	835(15)	gas	MPI	8
e_{2g}	16 (8)	Ring stretch	1557(8) ^f	gas	PE	1,6,16
		(B_{2g})	1480(10)T	Ar	LF	3
	17 (9)	CH bend	1230(15) ^f	gas	PE, MPI	1,6,8
18 (6)	Ring deform. (E_{1g})	676.4 ^f	gas	TPE,PE	1,6,8,12, MPI	15,16
	(B_{2g})	630(10)T	Ar	LF	3	
	(B_{1g})	365.4 ^g	gas	TPE,PE	12,15,16 MPI	
e_{2u}	20 (16)	Ring deform. (E_{1u})	345.4 ^h	gas	TPE,MPI	6,12,15
		(B_{1g})	300(10)	gas	PE,MPI	6,8,15,16

$B_0 = 0.189$; $C_0 = 0.095$ $PE^7TPE^{11}TPI^{14}$

$C_6D_6^+$

\tilde{F}^2A_{1g} D_{6h}
 $T^a \approx 62000$ gas $PE^{1,17}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_{1g}	1 ^b (2)	CD stretch	2240(100)	gas	PE	1,17
	2 (1)	Ring stretch	920(100)	gas	PE	1,17

\tilde{E}^2B_{1u} D_{6h}
 $T^a \approx 50000$ gas $PE^{1,17}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_{1g}	1 ^b (2)	CD stretch	1610T	gas	PE	1,17

\tilde{B}^2A_{2u} D_{6h}
 $T_0 \approx 19930^c$ Ar LF^3AB^4 $\tilde{C}-\tilde{X}$ 470–545 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
e_{2g}	18 ^b (6)	Ring deform.	590(60)T	Ar	AB	4

\tilde{A}^2E_{2g} D_{6h}
 $T^a = 18362$ gas $PE^{1,17}PRI^{15}$
 In an argon matrix, a weak, sharp absorption at 18215 has been attributed³ to a vibronically allowed transition in the excitation of $C_6D_6^+$ to the \tilde{B} state.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_{1g}	1 ^b (2)	CD stretch	2140(100)	gas	PE	1
a_{2u}	4 (11)	CD bend	486	gas	PRI	15
		CD bend (B_{1u})	451	gas	PRI	15
e_{1u}	14 (18)	(B_{2u})	425			
		Ring stretch	1450(100)	gas	PE	1,17
e_{2g}	17 (9)	CD bend	870(100)	gas	PE	1,17
		Ring deform.	569	gas	PE,PRI	1,15,17
e_{2u}	19 (17)	CD bend (E_{2u})	604	gas	PRI	15
		(A_{2u})	546			
20 (16)	Ring deform. (E_{2u})	218	gas	PRI	15	
	(A_{2u})	151				

\tilde{X}^2E_{1g} D_{6h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_{1g}	1 ^b (2)	CD stretch	2330(100)	gas	PE	1
		Ring stretch	928(20)	gas	PE	1,6,17
b_{2g}	8 (4)	Ring deform.	351(20)	gas	PE	6
		Ring stretch	1565(100)	gas	PE	1
e_{2g}	16 (8)	Ring stretch	1460(10)T	Ar	LF	3
		Ring deform. (E_{1g})	634 ^f	gas	PE,MPI	1,6,15,17
		(B_{2g})	590(10)T	Ar	LF	3
18 (6)	Ring deform. (E_{1g})	(B_{2g})	356	gas	MPI	15
		(B_{1g})	335 ⁱ	gas	PE,MPI	6,15
		(B_{1g})	262	gas	PE,MPI	6,15
e_{2u}	20 (16)	Ring deform. (E_{1u})	262	gas	PE,MPI	6,15

$B = 0.155$; $C = 0.076$ TPI^{14}

^aFrom vertical ionization potentials. The first ionization potential of benzene is taken to equal 74555.0(4), or 9.2405 eV, from Ref. 7.

^bIn order to be consistent with other vibrational numberings in these tables, the vibrational numbering of Herzberg is used here. However, many authors use instead the vibrational numbering of Wilson [E. B. Wilson, Jr., Phys. Rev. **45**, 706 (1934)]. This latter numbering is indicated in parentheses.

^cFrom analysis of combination bands.

^dRef. 16 gives 652(8).

^eRef. 16 gives 468(8).

^f $j = \pm 1/2$.

^g $j = +3/2$.

^h $j = -3/2$.

ⁱ $j = 3/2$.

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HC₇H

³Σ_u⁻ D_{∞h}
 T₀ = 19818.0 gas CR^{2,3} ³Σ_u⁻- \tilde{X} 459–505 nm
 19812(5) Ne AB¹ ³Σ_u⁻- \tilde{X} 410–506 nm
 19770(5)

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	2	C≡C s-stretch	1954.5(2)	gas	CR	3
			1961(5)	Ne	AB	1
	4		647(5)	Ne	AB	1
Π		CH bend	286HT	Ne	AB	1

DC₇D

³Σ_u⁻ D_{∞h}
 T₀ = 19943.2 gas CR² ³Σ_u⁻- \tilde{X} 457–502 nm
 19932(5) Ne AB¹ ³Σ_u⁻- \tilde{X} 410–503 nm
 19891(5)

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	2	C≡C s-stretch	1926.6(2)	gas	CR	3
			1935(5)	Ne	AB	1
	4		551(5)	Ne	AB	1
Π		CD bend	202HT	Ne	AB	1

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(cyc-HC₃)(C≡C)₂H

\tilde{X} C_s
 A₀ = 1.158; B₀ = 0.035; C₀ = 0.034 MW¹

(cyc-DC₃)(C≡C)₂D

\tilde{X} C_s
 A₀ = 0.997(2); B₀ = 0.033; C₀ = 0.032 MW¹

Reference

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3,5-C₆H₃CH₃

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		CH ₃ deform.	1533wm	Ar	IR	1
		Ring deform.	1462ms	Ar	IR	1
		CH wag	824m	Ar	IR	1
		CH wag	768m	Ar	IR	1
		Deformation	570wm	Ar	IR	1
a''		CH ₃ deform.	1570wm	Ar	IR	1
		Mixed	1000wm	Ar	IR	1
		Ring deform.	543vs	Ar	IR	1

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C₆H₅CH

In an argon matrix, absorption maxima have been observed^{2,3} at 240 and 245 nm. A weaker absorption maximum has also been observed³ at 300 nm.

In an argon matrix, structured absorption has been observed¹⁻³ between 372 and 430 nm.

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	3		3073ms	Ar	IR	1-3
	5		3060wm	Ar	IR	3
	6		3035wm	Ar	IR	3
	7		1562vw	Ar	IR	3
	8		1538w	Ar	IR	3
			1505(5)m	Ar	IR	1
	9		1458wm	Ar	IR	1,3
	10		1428wm	Ar	IR	1,3
			1390(5)m	Ar	IR	1
	11		1325vw	Ar	IR	3
	12		1290vw	Ar	IR	3
	13		1264w	Ar	IR	3
			1210(5)wm	Ar	IR	1
	14		1160vw	Ar	IR	3
	15		1149vw	Ar	IR	3
	16		1091w	Ar	IR	3
	17		1017w	Ar	IR	1,3
	18		976w	Ar	IR	3
			945(5)wm	Ar	IR	1
	19		840wm	Ar	IR	3
	20		788wm	Ar	IR	3
	21		612vw	Ar	IR	3
			550(5)wm	Ar	IR	1
	22		525vw	Ar	IR	3

a''	23	343wm	Ar	IR	3
	26	884wm	Ar	IR	1,3
	27	835vw	Ar	IR	3
	28	741vs	Ar	IR	1-3
	29	670m	Ar	IR	1-3
	30	497wm	Ar	IR	3
	31	443ms	Ar	IR	1-3
	32	392vw	Ar	IR	3

C₆D₅CD

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	2		2289m	Ar	IR	3
	3		2271vs	Ar	IR	3
	5		2257w	Ar	IR	3
	6		2245wm	Ar	IR	3
	7		1518vw	Ar	IR	3
	8		1485vw	Ar	IR	3
	9		1338wm	Ar	IR	3
	10		1329w	Ar	IR	3
	11		1256w	Ar	IR	3
	14		944w	Ar	IR	3
	15		854w	Ar	IR	3
	16		837vw	Ar	IR	3
	17		842w	Ar	IR	3
	18		823wm	Ar	IR	3
	19		762m	Ar	IR	3
	20		658m	Ar	IR	3
	21		585w	Ar	IR	3
	23		295wm	Ar	IR	3
a''	24		815w	Ar	IR	3
	26		740m	Ar	IR	3
	28		630m	Ar	IR	3
	29		525vs	Ar	IR	3
	30		429w	Ar	IR	3
	31		354s	Ar	IR	3
	32		339wm	Ar	IR	3

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cyc-C₇H₆**(1,2,4,6-Cycloheptatetraene)**

In an argon matrix, the onset of absorption occurs near 390 nm, with increasing absorption out to the limit of the measurements, near 220 nm.^{2,4}

\tilde{X}		C ₂				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a	1		3048wm	Ar	IR	4
	2		3041ms	Ar	IR	4
	3		3021w	Ar	IR	4
	4		1527w	Ar	IR	4
	5		1422w	Ar	IR	1,4
	6		1364wm	Ar	IR	1,4
	7		1211vw	Ar	IR	4
	8		1104w	Ar	IR	4
	9		1012vw	Ar	IR	4
	10		966wm	Ar	IR	1,4
	11		913wm	Ar	IR	1,2,4
	12		826vw	Ar	IR	4
	13		798vw	Ar	IR	4
	14		689m	Ar	IR	1-4
	15		581m	Ar	IR	1,2,4
	16		404m	Ar	IR	1,4
	17		302wm	Ar	IR	4
b	18		3038vs	Ar	IR	1,2,4
	20		3013wm	Ar	IR	1,2,4
	21		1823wm	Ar	IR	1-4
	22		1579w	Ar	IR	4
	23		1380wm	Ar	IR	1-4
	24		1285w	Ar	IR	4
	25		1189w	Ar	IR	1,4
	26		1019w	Ar	IR	4
	27		952w	Ar	IR	4
	28		874wm	Ar	IR	4
	29		772s	Ar	IR	1-4
	30		679vs	Ar	IR	1,2,4
	31		482w	Ar	IR	4
	32		362wm	Ar	IR	4
	33		298w	Ar	IR	4

cyc-C₇D₆

\tilde{X}		C ₂				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a	2		2263ms	Ar	IR	4
	3		2234vw	Ar	IR	4
	4		1499w	Ar	IR	4
	5		1312m	Ar	IR	4
	6		1221wm	Ar	IR	4
	7		915w	Ar	IR	4
	8		869w	Ar	IR	4
	9		828vw	Ar	IR	4
	11		765wm	Ar	IR	4
	12		724m	Ar	IR	4
	13		654wm	Ar	IR	4
	14		598wm	Ar	IR	4
	15		507m	Ar	IR	4
	16		394m	Ar	IR	4
	17		266wm	Ar	IR	4
b	18		2275vs	Ar	IR	4
	19		2254ms	Ar	IR	4
	20		2221wm	Ar	IR	4
	21		1800wm	Ar	IR	4
	22		1528w	Ar	IR	4
	23		1184wm	Ar	IR	4
	24		1006vw	Ar	IR	4
	25		892vw	Ar	IR	4
	26		843wm	Ar	IR	4
	27		823vw	Ar	IR	4
	29		660vs	Ar	IR	4
	30		543s	Ar	IR	4
	31		435w	Ar	IR	4
	32		314wm	Ar	IR	4
	33		262w	Ar	IR	4

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C₆H₅CH₂ \tilde{E}

A gas-phase absorption maximum near 230 nm has been attributed⁹ to the first Rydberg transition of C₆H₅CH₂.

 \tilde{D}

A gas-phase absorption between 260 and 245 nm, with maximum near 255 nm, has been attributed⁹ to the $\tilde{D}-\tilde{X}$ transition of C₆H₅CH₂. This band may have been excited in the MPI studies of Ref. 16. In a neon matrix, the counterpart of this absorption maximum has been observed near 245 nm.⁶

 \tilde{C}^2A_2

$T_0=32760$ gas AB^{2,5,7} $\tilde{C}-\tilde{X}$ 291–309 nm
 32730 Ne AB⁶

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	7a	C-CH ₂ stretch	1145	gas	AB	7
	18a	CH deform.	968	gas	AB	7
	1	Ring breathing	931	gas	AB	7
	12a	Ring deform.	804	gas	AB	7
	6a	Ring deform.	432	gas	AB	7
b ₂	6b	Ring deform.	525	gas	AB	7
	18b	CH deform.	276	gas	AB	7

 $\tilde{B}^2B_1^a$

$T_0=22850$ T gas AB¹¹EM¹¹LF^{22,24}MPI²⁸
 Vibronically coupled to modes of b₂ symmetry in the \tilde{A} state.^{11,22,24,27,28}
 A^b=0.179; B^b=0.088; C^b=0.059 gas LF²⁴

 \tilde{A}^2A_2

$T_0=22001.5$ gas EM^{1,3,5,10}AB⁴LF^{12,20-22,26}MPI²⁸ $\tilde{A}, \tilde{B}-\tilde{X}$ 429–471 nm
 22003 Ne AB⁶ $\tilde{A}, \tilde{B}-\tilde{X}$ 429–455 nm
 21862 Ar LF¹³ $\tilde{A}, \tilde{B}-\tilde{X}$ 430–510 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
6a	Ring deform.	Ring deform.	910	Ar	LF	13
			770T	gas	LF	20
			798	Ar	LF	13
			437	gas	AB,LF	11,18,20,22
			433	Ne	AB	6
			456	Ar	LF	13
			388 ^c	gas	AB	11,17,18
					LF	22
			402	Ne	AB	6
			328 ^c	gas	AB	11,17,18
6b	Ring deform.	Ring deform.	344	Ne	AB	6
					LF	22

$\tau_{0(1)}=0.4 \mu\text{s}$; $\tau_{0(2)}=1.85 \mu\text{s}$ gas LF^{15,20,21}
 A^d=0.180; B^d=0.088; C^d=0.059 EM^{10,17,19}LF^{22,24}

 $\tilde{X}^2B_1^a$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a ₁	8a	CH stretch	3069w	Ar	IR	25	
			Ring stretch	1603	gas	EM	5,8,18
			C-CH ₂ stretch	1510(25)	gas	PE	23,29
	19a	Ring stretch	1469m	Ar	IR	25	
			1431	gas	EM	5,8,18	
			1423	Ar	LF	13	
			1409m	Ar	IR	25	
	7a	CH ₂ deform.	1258	gas	EM,LF	5,7,18,20	
			1264m	Ar	IR	25	
			1181	gas	EM	5,8	
	9a	CH deform.	1046	gas	EM	18	
	18a	CH deform.	987.4	gas	EM,LF	5,8,10,18,20	
			982	Ar	LF	13	
	12a	Ring deform.	830	gas	EM,LF	5,8,18,20	
	6a	Ring deform.	524	gas	EM,LF	5,8,10,20	
				PE,MPI	23,28,29		
a ₂	17a	CH deform.	520	Ar	LF	13	
			963	gas	EM	18	
	10a	CH deform.	860	gas	EM	8,18	
	16a	Ring deform.	393	gas	EM	8	
	b ₁	CH ₂ wag	882.0w	Ar	IR	25	
			762.0vs	Ar	IR	25	
			710.9w	Ar	IR	25	
			667.0s	Ar	IR	25	
			465.0s	Ar	IR	25	
	b ₂	16b	Ring deform.	430	gas	EM	8
3111w				Ar	IR	25	
8b		Ring stretch	1549	gas	EM	5,8,18	
			1530	Ar	LF	13	
9b		CH deform.	1446m	Ar	IR	25	
			1305w	Ar	IR	25	
			1152	gas	EM	5,8,18	
15		CH deform.	1089	gas	EM	5,8	
	1015w		Ar	IR	25		
6b	Ring deform.	948.1vw	Ar	IR	25		
		615	gas	EM,LF	5,8,10,18		
				MPI	20,28		
				LF	13		
18b	CH deform.	356	gas	EM,LF	5,8,18,20		
				MPI	28		
		357	Ar	LF	13		

A₀=0.184; B₀=0.090; C₀=0.060 EM^{10,17,19}LF²⁴

C₆D₅CD₂ $\tilde{B}^2B_1^a$

$T_0=22455(10)$ gas AB¹¹EM¹¹

 \tilde{A}

$T_0=22093.7$ gas EM^{5,10} $\tilde{A}-\tilde{X}$ 434–502 nm
 21962 Ar LF¹³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
6a	Ring deform.	Ring deform.	844	Ar	LF	13
			731	Ar	LF	13
			404	Ar	LF	13

$\tau \approx 1340$ ns gas LF¹⁴

$\tilde{X}^2B_1^a$		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	8a	Ring stretch	1593	gas	EM	5,8
	19a	Ring stretch	1327	gas	EM	5,8
			1323	Ar	LF	13
	7a	C-CD ₂ stretch	1204	gas	EM	5,8
	1	Ring breathing	945.7	gas	EM	5,8,10
			945	Ar	LF	13
	9a	CD deform.	895	gas	EM	5,8
	18a	CD deform.	848	gas	EM	8
	12a	Ring stretch	791	gas	EM	5,8
	6a	Ring deform.	497.5	gas	EM	5,8,10
495			Ar	LF	13	
a_2	10a	CD deform.	750	gas	EM	8
	16a	Ring deform.	305	gas	EM	8
b_1	16b	Ring deform.	376	gas	EM	8
b_2	8b	Ring stretch	1495	gas	EM	5,8
			1490	Ar	LF	13
	15	CD deform.	848	gas	EM	5,8
	6b	Ring deform.	589.1	gas	EM	5,8,10
588			Ar	LF	13	
	18b	CD deform.	305	gas	EM	5,8
			303	Ar	LF	13

^aIn many of the references concerned with this molecule, the x axis is chosen in the molecular plane, resulting in an interchange of the B_1 and B_2 representations. However, the international convention established in 1955 is followed in the symmetry designations given here. In all of the references, the vibrational numbering follows that introduced by Wilson [E. B. Wilson, Jr., Phys. Rev. **45**, 706 (1934)]. In view of this unanimity, the common vibrational numbering is retained.

^bFrom analysis of band at 22326.368(2).

^cData presented in Ref. 22 suggest that this level is mixed with a nearby a_1 level of the \tilde{B}^2B_1 state.

^dFrom analysis of band at 22432.395(2).

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$C_6H_5CH_2^-$

Threshold for electron detachment from ground-state $C_6H_5CH_2^-$ = 7360(50) gas PE^{2,3}

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			945	gas	MPD	1
			514(15)	gas	PE	3

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C_8H

$^2\Pi_{3/2}$ $C_{\infty v}$
 $T_0 = 15973.5(2)$ gas CR⁵ $^2\Pi-\tilde{X}$ 554–626 nm
 15848(5) Ne AB¹ $^2\Pi-\tilde{X}$ 515–631 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	3		2077.6(3)	gas	CR	5
			2080(5)	Ne	AB	1
	5		1591.2(3)	gas	CR	5
			1607(5)	Ne	AB	1
	6		783(5)	Ne	AB	1

$A = -36.8(2)$ gas CR⁵

$\tilde{A}^2\Sigma^+$ $C_{\infty v}$
 $T_0 = 2041(50)$ gas PE⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		C≡C stretch	2016(70)	gas	PE	4

$\tilde{X}^2\Pi_{3/2}$ $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		C≡C stretch	1661(50)	gas	PE	4

$A_{\text{eff}} = -19.33$ MW^{2,3,6}
 $B_0 = 0.0196$ MW^{2,3,6}

C₈D

${}^2\Pi_{3/2}$ $D_{\infty h}$
 $T_0 = 16004.0(3)$ gas CR⁵
 15873(5) Ne AB⁵ ${}^2\Pi-\tilde{X}$ 568–625 nm
 ${}^2\Pi-\tilde{X}$ 572–630 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	3	C≡C stretch	2073.8(6)	gas	CR	5
			2064(7)	Ne	AB	5
	5	C–C stretch	1592.3(4)	gas	CR	5
			1604(7)	Ne	AB	5

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C₈H⁻

Threshold for electron detachment from ground-state
 $C_8H^- = 32000(80)$ gas PE¹

${}^1\Sigma^+$ $C_{\infty v}$
 $T_0 = 35174(25)$ Ne AB² ${}^1\Sigma^+ - \tilde{X}$ 275–285 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		C–C stretch	720(35)	Ne	AB	2
		C–C stretch	464(35)	Ne	AB	2

$\tilde{X} {}^1\Sigma^+$ $C_{\infty v}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		CC stretch	2106	Ne	IR	2
		CC stretch	2021	Ne	IR	2

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- ¹T. R. Taylor, C. Xu, and D. M. Neumark, *J. Chem. Phys.* **108**, 10018 (1998).
- ²M. Grutter, M. Wyss, and J. P. Maier, *J. Chem. Phys.* **110**, 1492 (1999).

C₈H₂⁺

$\tilde{C} {}^2\Pi_u$ $D_{\infty h}$
 $T_0 = 31300(200)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			400T	gas	PE	1

$\tilde{B} {}^2\Pi_g$ $D_{\infty h}$
 $T_0 = 24600(200)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+		C≡C stretch	1960(80)	gas	PE	1
			1380(80)	gas	PE	1

$\tilde{A} {}^2\Pi_u$ $D_{\infty h}$
 $T_0 = 14143.18$ gas PE¹EF²EM⁵CR⁵
 13975 Ne LF³AB⁴ $\tilde{A}-\tilde{X}$ 625–845 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	3	C≡C stretch	1950(80)	gas	PE	1
			2029.5	Ne	LF,AB	3,4
	5	C–C stretch	463	Ne	LF,AB	3,4

$\tau_0 \leq 6$ ns gas EF²
 $B_0 = 0.019$ EM⁵CR⁵

$\tilde{X} {}^2\Pi_g$ $D_{\infty h}$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	2	C≡C stretch	2100(80)	gas	PE	1
			2141.0	Ne	LF	3
			2040.5	Ne	LF	3
	5	C–C stretch	472.2	Ne	LF	3
Π	17	Deform.	88.5HT	Ne	LF	3

$B_0 = 0.019$ EM⁵CR⁵

C₈D₂⁺

$\tilde{A} {}^2\Pi_u$ $D_{\infty h}$
 $T_0 = 14169.52$ gas EM⁵CR⁵
 $B_0 = 0.017$ EM⁵CR⁵

$\tilde{X} {}^2\Pi_g$ $D_{\infty h}$
 $B_0 = 0.018$ EM⁵CR⁵

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cyc-C₈H₆

\tilde{a}
 $T_0 = 5710(50)$ gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CC stretch	2060(20)	gas	PE	1
			1490(20)	gas	PE	1

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁		CC stretch	2185(20)	gas	PE	1
			1500(20)	gas	PE	1
			295(20)	gas	PE	1

Reference

¹P. G. Wenthold and W. C. Lineberger, *J. Am. Chem. Soc.* **119**, 7772 (1997).

***cyc*-C₈H₆⁻**

Threshold for electron detachment from ground-state *cyc*-C₈H₆⁻ = 5710(50) gas PE¹

Reference

¹P. G. Wenthold and W. C. Lineberger, *J. Am. Chem. Soc.* **119**, 7772 (1997).

***p*-HC=C₆H₄=CH**

In a nitrogen matrix,¹ this species has strong bands at 27400 (365 nm), 25710 (389 nm), 23920 (418 nm) and 21740 (460 nm), as well as a long, regularly structured tail out to 700 nm.

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH stretch	3051.0w	N ₂	IR	1
			1258.2wm	N ₂	IR	1
			1248.2wm	N ₂	IR	1
			1239.0wm	N ₂	IR	1
			1237.5wm			
			956.6w	N ₂	IR	1
			847.4wm	N ₂	IR	1
			844.5m			
			826.6wm	N ₂	IR	1
			817.4vs	N ₂	IR	1
			815.4m			
			814.4m			
			768.9wm	N ₂	IR	1
			629.8wm	N ₂	IR	1
			618.6s	N ₂	IR	1
			615.1s			
			612.5m			
			457.3ms	N ₂	IR	1
455.0wm						

Reference

¹W. Subhan, P. Rempala, and R. S. Sheridan, *J. Am. Chem. Soc.* **120**, 11528 (1998).

C₉

¹Σ_u⁺ D_{∞h}
 T₀ = 33895(25) Ne AB⁹
 32480^a Ar AB² ¹Σ_u⁺ - \tilde{X} 292–295 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			306(30)	Ne	AB	9

T₀ = 26777(15)T Ne AB⁹ 324–374 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	1		2535(20)	Ne	AB	9
	3		1041(20)	Ne	AB	9
	4		443(20)	Ne	AB	9

\tilde{a} D_{∞h}
 T^a = 10730(80) gas PE¹⁵

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	3	Sym. stretch	1258(50)	gas	PE	6
	4	Sym. stretch	484(48)	gas	PE	6
Σ _u ⁺	5	Asym. stretch	448T	Ar	IR	12
			2079.67	gas	DL	11
			2081.1	Ne	IR	12,13
			2078.1	Ar	IR	12
			2079.0T	Kr	IR	12
	6	Asym. stretch	2014.28	gas	DL	3,7
			2010.0	Ne	IR	12–14
			1998.0 ^b	Ar	IR	1,2,5,12
			1994.2	Kr	IR	12
	7	Asym. stretch	2007.3	H ₂	IR	10,14
1602.8			Ne	IR	12	
1601.0			Ar	IR	8,12	
		1600.8	Kr	IR	12	

B₀ = 0.0143 DL^{3,7,11}

^aFrom vertical ionization potentials.

^bPreviously assigned to C₈. Reassignment to C₉ dictated by arguments presented in Ref. 4.

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C₉⁻

Threshold for electron detachment from ground-state C₉⁻ = 29720(80) gas PE^{1,2}

$\tilde{D}^2\Pi_g$ D_{∞h}
 T₀ = 34536(24) Ne AB⁶ $\tilde{D}-\tilde{X}$ 281–290 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	3		983(25)	Ne	AB	6
	4		474(24)	Ne	AB	6

$\tilde{C}^2\Pi_g$ D_{∞h}
 T₀ = 29334 gas MPD⁷
 29446(17) Ne AB⁶ $\tilde{C}-\tilde{X}$ 295–340 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	1		2177(19)	Ne	AB	6
	2		1666(19)	Ne	AB	6
	4		435(17)	Ne	AB	6

$\tilde{B}^2\Pi_g$ D_{∞h}
 T₀ = 16462 gas MPD^{5,7,9}
 16468(5) Ne AB⁶ $\tilde{B}-\tilde{X}$ 530–608 nm
 $\tilde{B}-\tilde{X}$ 528–608 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	1		1991	gas	MPD	7,9
			2001(6)	Ne	AB	6

$\tilde{A}^2\Pi_g$ D_{∞h}
 T₀ = 13068 gas MPD^{5,7,9}
 13082(3) Ne AB⁶

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺			1200(40)	gas	MPD	5

$\tilde{X}^2\Pi_u$ D_{∞h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _u ⁺	6	Asym. stretch	1692.6	Ne	IR	4
			1686.7	Ar	IR	3,8
		Asym. stretch	1583.3	Ar	IR	3,8

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cum-C₉H⁻ ^a

¹A' C_s
 T₀ = 29014(10) gas MPD¹ ¹A'- \tilde{X} 326–345 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		C=C stretch	1668(10)	gas	MPD	1

^aThis isomer of C₉H⁻, with cumulenonic bonding, lies at a somewhat higher energy than the acetylenic isomer.

Reference

- ¹M. Tulej, F. Güthe, M. Schnaiter, M. V. Packhov, D. A. Kirkwood, J. P. Maier, and G. Fischer, *J. Phys. Chem. A* **103**, 9712 (1999).

acet-C₉H⁻

³Σ_v⁻ C_{∞v}
 T₀ = 27630(10) gas MPD¹ ³Σ_v⁻- \tilde{X} 337–362 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺		C≡C stretch	1978(10)	gas	MPD	1
		C–C stretch	680(10)	gas	MPD	1

Reference

- ¹M. Tulej, F. Güthe, M. Schnaiter, M. V. Packhov, D. A. Kirkwood, J. P. Maier, and G. Fischer, *J. Phys. Chem. A* **103**, 9712 (1999).

HC₉H

³Σ_u⁻ D_{∞h}
 T₀ = 17184.6 gas CR^{2,3}
 17218(5) Ne AB¹ ³Σ_u⁻- \tilde{X} 521–582 nm
 17160(5) ³Σ_u⁻- \tilde{X} 433–583 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	3	C≡C s-stretch	1976.0(2)	gas	CR	2,3
			1969(5)	Ne	AB	1
			770(5)	Ne	AB	1

DC₉D

${}^3\Sigma_u^-$	$D_{\infty h}$				
$T_0 = 17262.5$	gas	CR ^{2,3}		${}^3\Sigma_u^- - \tilde{X}$	519–580 nm
17292(5)	Ne	AB ¹		${}^3\Sigma_u^- - \tilde{X}$	432–581 nm
17236(5)					

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	3	C≡C s-stretch	1968.4(2)	gas	CR	2,3
			1966(5)	Ne	AB	1

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cyc-(HC₃)(C≡C)₃H

\tilde{X}	C_s				
$A_0 = 1.157T$;	$B_0 = 0.015$;	$C_0 = 0.015$	MW ¹		

Reference

- ¹M. C. McCarthy, M. J. Travers, W. Chen, C. A. Gottlieb, and P. Thaddeus, *Astrophys. J.* **498**, L89 (1998).

C₁₀

(2) ${}^3\Sigma_u^-$	$D_{\infty h}$				
$T_0 = 29985(18)$	Ne	AB ²			324–334 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	5		401(25)	Ne	AB	2

$T_0 = 27925(16)$	$D_{\infty h}$	Ne	AB ²			349–358 nm
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Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	5		364(22)	Ne	AB	2

$\tilde{X} {}^3\Sigma_g^-$	$D_{\infty h}$				
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Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_u^+			2074.5	Ne	IR	1
			1915.4	Ne	IR	1

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cyc-C₁₀

${}^1\Sigma_u^+$	D_{10h}				
$T_0 = 31636(20)$	Ne	AB ¹			308–316 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_g			374(30)	Ne	AB	1

Reference

- ¹M. Grutter, M. Wyss, E. Riaplov, J. P. Maier, S. D. Peyerimhoff, and M. Hanrath, *J. Chem. Phys.* **111**, 7397 (1999).

C₁₀⁻

REMPD studies⁴ suggest the formation of cyclic species of C₁₀⁻ when the anion is generated at low laser fluence.

(2) ${}^2\Pi_g$	$D_{\infty h}$				
$T_0 = 13596(5)$	Ne	AB ^{1,3}			447–736 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1		2022(7)	Ne	AB	1,3
	2		1913(7)	Ne	AB	3
	5		400(6)	Ne	AB	1,3

$\tilde{C} {}^2\Pi_g$	$D_{\infty h}$				
$T_0 = 10380(5)$	gas	MPD ⁴			
	10338(5)	Ne	AB ¹		$\tilde{C} - \tilde{X}$ 804–968 nm
	10189(10)	Ar	AB ⁵		

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	2		2092(5)	Ne	AB	1
	3		1992(5)	Ne	AB	1
	4		488(5)	Ne	AB	1
	5		400(5)	gas	MPD	4
			372(5)	Ne	AB	1

$\tilde{B} {}^2\Sigma_g^+$	$D_{\infty h}$				
$T_0 = 8964(2)$	Ne	AB ³			

$\tilde{X} {}^2\Pi_u$	$D_{\infty h}$				
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Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_u^+			2094.5	Ne	IR	2

References

- ¹P. Freivogel, J. Fulara, M. Jakobi, D. Forney, and J. P. Maier, *J. Chem. Phys.* **103**, 54 (1995).
²P. Freivogel, M. Grutter, D. Forney, and J. P. Maier, *Chem. Phys.* **216**, 401 (1997).
³P. Freivogel, M. Grutter, D. Forney, and J. P. Maier, *J. Chem. Phys.* **107**, 4468 (1997); *J. Chem. Phys.* **108**, 2261 (1998).
⁴M. Ohara, M. Suwa, T. Ishigaki, H. Shiromaru, Y. Achiba, and W. Krätschmer, *J. Chem. Phys.* **109**, 1329 (1998).
⁵J. Szczepanski, J. Fuller, S. Ekern, and M. Vala, *Spectrochim. Acta A* **57**, 775 (2001).

C₁₀H

²Π_{3/2} C_{∞v}
 T₀ = 14000 gas CR² 2Π- \tilde{X} 622–715 nm
 13852(5) Ne AB¹ 2Π- \tilde{X} 627–722 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	3		2084(2)	gas	CR	2
			2075(2)			
			2097(5)	Ne	AB	1
	5		1520(5)T	Ne	AB	1

B₀ = 0.010 MW³

C₁₀D

²Π_{3/2} C_{∞v}
 T₀ = 14020 gas CR² 2Π- \tilde{X} 621–714 nm
 13866(4) Ne AB² 2Π- \tilde{X} 626–722 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	3		2082.2	gas	CR	2
			2094(6)	Ne	AB	2

References

- 1 P. Freivogel, J. Fulara, M. Jakobi, D. Forney, and J. P. Maier, *J. Chem. Phys.* **103**, 54 (1995).
- 2 H. Linnartz, T. Motylewski, and J. P. Maier, *J. Chem. Phys.* **109**, 3819 (1998).
- 3 C. A. Gottlieb, M. C. McCarthy, M. J. Travers, J.-U. Grabow, and P. Thaddeus, *J. Chem. Phys.* **109**, 5433 (1998).

cum-C₁₀H⁻ a

\tilde{A}
 T₀ = 27847(10) gas MPD¹ \tilde{A} - \tilde{X} 281–359 nm
 28050(15) Ne AB¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		C=C stretch	1678(10)	gas	MPD	1

^aThis isomer of C₁₀H⁻, with cumulenic bonding, lies at a somewhat higher energy than the acetylenic isomer, and either the \tilde{X} or the \tilde{A} state may be nonlinear.

Reference

- 1 D. A. Kirkwood, M. Tulej, M. V. Pachkov, M. Schnaiter, F. Güthe, M. Grutter, M. Wyss, J. P. Maier, and G. Fischer, *J. Chem. Phys.* **111**, 9280 (1999).

acet-C₁₀H⁻

¹Σ⁺ C_{∞v}
 T₀ = 30600(15) Ne AB^{1,2} ¹Σ⁺- \tilde{X} 302–327 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺		CC stretch	2363(15)	gas	MPD	2
			2394(20)	Ne	AB	2
		C–C stretch	480(15)	gas	MPD	2
			553(27)	Ne	AB	1,2

\tilde{X} ¹Σ⁺ C_{∞v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺		CC stretch	2040	Ne	IR	1

References

- 1 M. Grutter, M. Wyss, and J. P. Maier, *J. Chem. Phys.* **110**, 1492 (1999).
- 2 D. A. Kirkwood, M. Tulej, M. V. Pachkov, M. Schnaiter, F. Güthe, M. Grutter, M. Wyss, J. P. Maier, and G. Fischer, *J. Chem. Phys.* **111**, 9280 (1999).

C₁₁

¹Σ_u⁺ D_{∞h}
 T₀ = 29732(20) Ne AB² ¹Σ_u⁺- \tilde{X} 323–337 nm
 28248 Ar AB⁷

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺			1166(30)	Ne	AB	2
			872(30)	Ne	AB	2
			528(30)	Ne	AB	2
			240(30)	Ne	AB	2

T₀ = 22245(10)T Ne AB² 399–450 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	1		2440(15)	Ne	AB	2
	2		2298(15)	Ne	AB	2
	5		357(15)	Ne	AB	2

\tilde{a} D_{∞h}
 T^a = 9520(80) gas PE⁶

\tilde{X} ¹Σ_g⁺ D_{∞h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _u ⁺		Sym. stretch	440T	gas	PE	1
		Asym. stretch	1938.6	Ne	IR	3
			1946.0(8)	Ar	IR	4,5
			1937.3	Kr	IR	5
			1942.6	N ₂	IR	5
	8	Asym. stretch	1853.4	Ne	IR	3
			1856.7(8)	Ar	IR	4,5
			1849.9	Kr	IR	5
			1854.8	N ₂	IR	5
	9	Asym. stretch	1357.0w	N ₂	IR	5

^aFrom vertical ionization potentials.

References

- ¹D. W. Arnold, S. E. Bradforth, T. N. Kitsopoulos, and D. M. Neumark, *J. Chem. Phys.* **95**, 8753 (1991).
²D. Forney, P. Freivogel, M. Grutter, and J. P. Maier, *J. Chem. Phys.* **104**, 4954 (1996).
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⁵L. Lapinski and M. Vala, *Chem. Phys. Lett.* **300**, 195 (1999).
⁶M. Kohno, S. Suzuki, H. Shiromaru, and Y. Achiba, *J. Chem. Phys.* **110**, 3781 (1999).
⁷J. Szczepanski, J. Fuller, S. Ekern, and M. Vala, *Spectrochim. Acta A* **57**, 775 (2001).

C₁₁⁻

Threshold for electron detachment from ground-state C₁₁⁻ = 31570(65) gas PE^{1,2}

REMPD studies⁵ suggest the formation of cyclic species of C₁₁⁻ when the anion is generated at low laser fluence.

$\tilde{D}^2\Pi_u$ D_{∞h}
 T₀ = 29884(18) Ne AB⁴ $\tilde{D}-\tilde{X}$ 320–335 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	4		765(18)	Ne	AB	4
	5		344(18)	Ne	AB	4

$\tilde{C}^2\Pi_u$ D_{∞h}
 T₀ = 25151(13) Ne AB⁴ $\tilde{C}-\tilde{X}$ 392–398 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	5		355(13)	Ne	AB	4

$\tilde{B}^2\Pi_u$ D_{∞h}
 T₀ = 13850(5) gas MPD⁵
 13906(4) Ne AB⁴ $\tilde{B}-\tilde{X}$ 496–722 nm
 $\tilde{B}-\tilde{X}$ 627–720 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	1		2070(5)	gas	MPD	5
			2032(5)	Ne	AB	4
			1882(4)	Ne	AB	4
	5		580(5)	gas	MPD	5

$\tilde{A}^2\Pi_u$ D_{∞h}
 T₀ = 11050(10) gas MPD^{3,5}
 11042(2) Ne AB⁴ $\tilde{A}-\tilde{X}$ 788–905 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	5		570(5)	gas	MPD	3,5

References

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cum-C₁₁H⁻ a

¹A' C_s
 T₀ = 26742(10) gas MPD¹

^aThis isomer of C₁₁H⁻, with cumulenic bonding, lies at a somewhat higher energy than the acetylenic isomer.

Reference

- ¹M. Tulej, F. Güthe, M. Schnaiter, M. V. Packhov, D. A. Kirkwood, J. P. Maier, and G. Fischer, *J. Phys. Chem. A* **103**, 9712 (1999).

acet-C₁₁H⁻

³Σ⁻ C_{∞v}
 T₀ = 24288(10) gas MPD¹ ³Σ⁻ - \tilde{X} 391–412 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺		CC stretch	1234(10)	gas	MPD	1
		CC stretch	725(10)	gas	MPD	1

Reference

- ¹M. Tulej, F. Güthe, M. Schnaiter, M. V. Packhov, D. A. Kirkwood, J. P. Maier, and G. Fischer, *J. Phys. Chem. A* **103**, 9712 (1999).

HC₁₁H

³Σ_u⁻ D_{∞h}
 T₀ = 15294.9 gas CR²
 15302(5) Ne AB¹ ³Σ_u⁻ - \tilde{X} 579–654 nm
 15268(5) ³Σ_u⁻ - \tilde{X} 519–656 nm
 15251(5)

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺		C≡C s-stretch	1944.5(3)	gas	CR	2
			1973(5)	Ne	AB	1
		C≡C s-stretch	1885(5)	Ne	AB	1

DC₁₁D

³Σ_u⁻ D_{∞h}
 T₀ = 15345.5 gas CR²
 15294(5) Ne AB¹ ³Σ_u⁻ - \tilde{X} 578–652 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺		C≡C s-stretch	1944.1(3)	gas	CR	2

References

- ¹J. Fulara, P. Freivogel, D. Forney, and J. P. Maier, *J. Chem. Phys.* **103**, 8805 (1995).
²C. D. Ball, M. C. McCarthy, and P. Thaddeus, *J. Chem. Phys.* **112**, 10149 (2000).

C₁₂

$\tilde{X}^3\Sigma_g^-$		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_u^+	7		2140.6T	Ar	IR	2
	8		2003.9T	Ne	IR	1
			1997.2	Ar	IR	2
	9		1818.0	Ar	IR	2

References

- ¹P. Freivogel, M. Grutter, D. Forney, and J. P. Maier, *Chem. Phys.* **216**, 401 (1997).
²X. D. Ding, S. L. Wang, C. M. L. Rittby, and W. R. M. Graham, *J. Chem. Phys.* **112**, 5113 (2000).

cyc-C₁₂

$(2) ^1E_{1u}$		D_{6h}				
$T_0=30112(18)$		Ne	AB ¹			290–332 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_g			1882(28)	Ne	AB	1
			300(26)	Ne	AB	1

$(1) ^1E_{1u}$		D_{6h}				
$T_0=13893(4)$		Ne	AB ¹			626–720 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_g			2071(7)	Ne	AB	1
			1907(7)	Ne	AB	1

Reference

- ¹M. Grutter, M. Wyss, E. Riaplov, J. P. Maier, S. D. Peyerimhoff, and M. Hanrath, *J. Chem. Phys.* **111**, 7397 (1999).

C₁₂⁻

$(2) ^2\Pi_u$		$D_{\infty h}$				
$T_0=11636(5)$		Ne	AB ^{1,3}			642–860 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1		2014(6)	Ne	AB	3
	2		1923(6)	Ne	AB	1,3
	4		1463(5)	Ne	AB	3
	6		320(4)	Ne	AB	1,3

$\tilde{C}^2\Pi_u$		$D_{\infty h}$				
$T_0=8100(100)$		gas	MPD ⁴			$\tilde{C}-\tilde{X}$ 1100–1240 nm
		8006(5)	Ne	AB ¹		$\tilde{C}-\tilde{X}$ 1200–1250 nm
		7906(10)	Ar	AB ⁵		

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	6		323(5)	Ne	AB	1

$\tilde{X}^2\Pi_g$		$D_{\infty h}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_u^+			2012.6	Ne	IR	2
			1819.3	Ne	IR	2

References

- ¹P. Freivogel, J. Fulara, M. Jakobi, D. Forney, and J. P. Maier, *J. Chem. Phys.* **103**, 54 (1995).
²P. Freivogel, M. Grutter, D. Forney, and J. P. Maier, *Chem. Phys.* **216**, 401 (1997).
³P. Freivogel, M. Grutter, D. Forney, and J. P. Maier, *J. Chem. Phys.* **107**, 4468 (1997).
⁴M. Ohara, D. Kasuya, H. Shiromaru, and Y. Achiba, *J. Phys. Chem. A* **104**, 8622 (2000).
⁵J. Szczepanski, J. Fuller, S. Ekern, and M. Vala, *Spectrochim. Acta A* **57**, 775 (2001).

C₁₂H

$^2\Pi_{3/2}$		$C_{\infty v}$				
$T_0=12492(5)$		Ne	AB ¹			$^2\Pi-\tilde{X}$ 685–801 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	3		2089(5)	Ne	AB	1
	5		1395(5)T	Ne	AB	1

$$B_0=0.0058 \text{ MW}^2$$

References

- ¹P. Freivogel, J. Fulara, M. Jakobi, D. Forney, and J. P. Maier, *J. Chem. Phys.* **103**, 54 (1995).
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cum-C₁₂H⁻ ^a

\tilde{A}		$D_{\infty h}$				
$T_0=25654(10)$		gas	MPD ¹			$\tilde{A}-\tilde{X}$ 377–390 nm
		25806(15)	Ne	AB ¹		

^aThis isomer of C₁₂H⁻, with cumulenic bonding, lies at a somewhat higher energy than the acetylenic isomer, and either the \tilde{X} or the \tilde{A} state may be nonlinear.

Reference

- ¹D. A. Kirkwood, M. Tulej, M. V. Pachkov, M. Schnaiter, F. Güthe, M. Grutter, M. Wyss, J. P. Maier, and G. Fischer, *J. Chem. Phys.* **111**, 9280 (1999).

acet-C₁₂H⁻

$1\Sigma^+$	$C_{\infty v}$					
$T_0=26802(10)$	gas	MPD ²			$1\Sigma^+-\tilde{X}$	339–373 nm
26961(15)	Ne	AB ^{1,2}			$1\Sigma^+-\tilde{X}$	315–372 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		CC stretch	2100(15)	gas	MPD	2
			2034(20)	Ne	AB	1,2
		C–C stretch	776(15)	gas	MPD	2
			737(20)	Ne	AB	1,2

$\tilde{X}^1\Sigma^+$	$C_{\infty v}$					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		CC stretch	1986	Ne	IR	1

References

- ¹M. Grutter, M. Wyss, and J. P. Maier, J. Chem. Phys. **110**, 1492 (1999).
²D. A. Kirkwood, M. Tulej, M. V. Pachkov, M. Schnaiter, F. Güthe, M. Grutter, M. Wyss, J. P. Maier, and G. Fischer, J. Chem. Phys. **111**, 9280 (1999).

HC₁₂H

$1\Sigma_u^+$	$D_{\infty h}$					
$T_0=38971(30)$	Ne	AB ¹			$1\Sigma_u^+-X$	232–257 nm
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+		CC stretch	2029(40)	Ne	AB	1

Reference

- ¹M. Grutter, M. Wyss, J. Fulara, and J. P. Maier, J. Phys. Chem. A **102**, 9785 (1998).

HC₁₂H⁻

(3) $2\Pi_g$	$D_{\infty h}$					
$T_0=31506(20)$	Ne	AB ¹				
(2) $2\Pi_g$	$D_{\infty h}$					
$T_0=13526(4)$	Ne	AB ¹			(2) $2\Pi_g-\tilde{X}$	522–740 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+		CC stretch	2301(7)	Ne	AB	1
		CC stretch	2014(7)	Ne	AB	1
		C–C stretch	464(6)	Ne	AB	1

(1) $2\Pi_g$	$D_{\infty h}$					
$T_0=12814(3)$	Ne	AB ¹			(1) $2\Pi_g-\tilde{X}$	650–781 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+		CC stretch	2140(6)	Ne	AB	1
		CC stretch	1994(6)	Ne	AB	1
		C–C stretch	550(5)	Ne	AB	1

Reference

- ¹M. Grutter, M. Wyss, J. Fulara, and J. P. Maier, J. Phys. Chem. A **102**, 9785 (1998).

C₁₃

$1\Sigma_u^+$	$D_{\infty h}$					
$T_0=26341(15)$	Ne	AB ²			$1\Sigma_u^+-\tilde{X}$	370–380 nm
24969	Ar	AB ⁴				

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			669(20)	Ne	AB	2
			424(20)	Ne	AB	2
			207(20)	Ne	AB	2

$T_0=18761(7)T$	Ne	AB ²				
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Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+	1		2409(10)	Ne	AB	2
	2		2258(10)	Ne	AB	2
	6		307(10)	Ne	AB	2

\tilde{a}	$D_{\infty h}$					
$T^a=7750(80)$	gas	PE ³				

$\tilde{X}^1\Sigma_g^+$	$D_{\infty h}$				Structure: DL ¹	
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Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_u^+			1808.96	gas	DL	1

$B_0=0.0047$	DL ¹					
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^aFrom vertical ionization potentials.

References

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²D. Forney, P. Freivogel, M. Grutter, and J. P. Maier, J. Chem. Phys. **104**, 4954 (1996).
³M. Kohno, S. Suzuki, H. Shiromaru, and Y. Achiba, J. Chem. Phys. **110**, 3781 (1999).
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C₁₃⁻

$\tilde{B}^2\Pi_g$ $D_{\infty h}$
 $T_0 = 11300(100)$ gas MPD² $\tilde{B}-\tilde{X}$ 813–885 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			400T	gas	MPD	2

$\tilde{A}^2\Pi_g$ $D_{\infty h}$
 $T_0 = 9500(100)$ gas MPD² $\tilde{A}-\tilde{X}$ 917–1053 nm
 9502(2) Ne AB¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			300T	gas	MPD	2

References

- ¹M. Wyss, M. Grutter, and J. P. Maier, Chem. Phys. Lett. **304**, 35 (1999).
²M. Ohara, D. Kasuya, H. Shiromaru, and Y. Achiba, J. Phys. Chem. A **104**, 8622 (2000).

C₁₃H

$\tilde{X}^2\Pi_{1/2}$ $C_{\infty v}$
 $B_0 = 0.0046$ MW¹

Reference

- ¹C. A. Gottlieb, M. C. McCarthy, M. J. Travers, J.-U. Grabow, and P. Thaddeus, J. Chem. Phys. **109**, 5433 (1998).

cum-C₁₃H⁻ ^a

$^1A'$ C_s
 $T_0 = 24726(10)$ gas MPD¹

^aThis isomer of C₁₃H⁻, with cumulenic bonding, lies at a somewhat higher energy than the acetylenic isomer.

Reference

- ¹M. Tulej, F. Güthe, M. Schnaiter, M. V. Packhov, D. A. Kirkwood, J. P. Maier, and G. Fischer, J. Phys. Chem. A **103**, 9712 (1999).

acet-C₁₃H⁻

$^3\Sigma^-$ $C_{\infty v}$
 $T_0 = 22200(10)$ gas MPD¹ $^3\Sigma^- - \tilde{X}$ 410–450 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		C≡C stretch	2144(10)	gas	MPD	1
		C–C stretch	739(10)	gas	MPD	1
		C–C stretch	617(10)	gas	MPD	1

Reference

- ¹M. Tulej, F. Güthe, M. Schnaiter, M. V. Packhov, D. A. Kirkwood, J. P. Maier, and G. Fischer, J. Phys. Chem. A **103**, 9712 (1999).

HC₁₃H

$^3\Sigma_u^-$ $D_{\infty h}$
 $T_0 = 13916.4(3)$ gas CR²
 13901(5) Ne AB¹ $^3\Sigma_u^- - \tilde{X}$ 562–722 nm
 13866(5)
 13852(5)

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ_g^+		C≡C s-stretch	1967(5)	Ne	AB	1
		C≡C s-stretch	1904(5)	Ne	AB	1
			619(5)	Ne	AB	1

DC₁₃D

$^3\Sigma_u^-$ $D_{\infty h}$
 $T_0 = 13951.3(3)$ gas CR²

References

- ¹J. Fulara, P. Freivogel, D. Forney, and J. P. Maier, J. Chem. Phys. **103**, 8805 (1995).
²C. D. Ball, M. C. McCarthy, and P. Thaddeus, J. Chem. Phys. **112**, 10149 (2000).

C₁₄

\tilde{A}
 $T_0 = 19010$ N₂ LF¹ $\tilde{A}-\tilde{X}$ 526–640 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			437	N ₂	LF	1
			108	N ₂	LF	1

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			2218	N ₂	Ra	1
			2089	N ₂	Ra	1
			1926	N ₂	Ra	1
			1807	N ₂	Ra	1
			1655	N ₂	Ra	1
			1627	N ₂	Ra	1
			1394	N ₂	Ra	1
			1211	N ₂	Ra	1
			851	N ₂	Ra	1
			829	N ₂	Ra	1
			282	N ₂	Ra	1

Reference

¹G. A. Rechtsteiner, C. Felix, A. K. Ott, O. Hampe, R. P. Van Duyne, M. F. Jarrold, and K. Raghavachari, *J. Phys. Chem. A* **105**, 3029 (2001).

cyc-C₁₄¹E_{1u}T₀=28837(17) Ne AB¹

Reference

¹M. Grutter, M. Wyss, E. Riplov, J. P. Maier, S. D. Peyerimhoff, and M. Hanrath, *J. Chem. Phys.* **111**, 7397 (1999).

C₁₄⁻

(2) ²Π_g D_{∞h}
T₀=10200(100) gas MPD³
10202(2) Ne AB^{1,2}

723–981 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	1		1810(4)	Ne	AB	2

²C̃²Π_g D_{∞h}
T₀=6900(100) gas MPD³
6849(5) Ne AB¹
6782 Ar AB⁴

C̃–X̃ 1333–1450 nm
C̃–X̃ 1402–1460 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	7		283(2)	Ne	AB	1

References

- ¹P. Freivogel, J. Fulara, M. Jakobi, D. Forney, and J. P. Maier, *J. Chem. Phys.* **103**, 54 (1995).
²P. Freivogel, M. Grutter, D. Forney, and J. P. Maier, *J. Chem. Phys.* **107**, 4468 (1997).
³M. Ohara, D. Kasuya, H. Shiromaru, and Y. Achiba, *J. Phys. Chem. A* **104**, 8622 (2000).
⁴J. Szczepanski, J. Fuller, S. Ekern, and M. Vala, *Spectrochim. Acta A* **57**, 775 (2001).

C₁₄H

²Π_{3/2} C_{∞v}
T₀=11554(5)T Ne AB¹
B₀=0.0037 MW²

References

- ¹P. Freivogel, J. Fulara, M. Jakobi, D. Forney, and J. P. Maier, *J. Chem. Phys.* **103**, 54 (1995).
²C. A. Gottlieb, M. C. McCarthy, M. J. Travers, J.-U. Grabow, and P. Thaddeus, *J. Chem. Phys.* **109**, 5433 (1998).

cum-C₁₄H⁻ a²Ã

T₀=23763(10) gas MPD¹
23837(15) Ne AB¹

Ã–X̃ 369–421 nm
Ã–X̃ 361–420 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CC stretch	1996(15)	gas	MPD	1
			1982(20)	Ne	AB	1
		C=C stretch	1747(15)	gas	MPD	1
		C–C stretch	1280(15)	gas	MPD	1
			1244(20)	Ne	AB	1

^aThis isomer of C₁₄H⁻, with cumulenenic bonding, lies at a somewhat higher energy than the acetylenic isomer, and either the X̃ or the Ã state may be nonlinear.

Reference

¹D. A. Kirkwood, M. Tulej, M. V. Pachkov, M. Schnaiter, F. Güthe, M. Grutter, M. Wyss, J. P. Maier, and G. Fischer, *J. Chem. Phys.* **111**, 9280 (1999).

acet-C₁₄H⁻¹Σ⁺

C_{∞v}
T₀=23763(10) gas MPD^{1,3}
23837(15) Ne AB¹⁻³

¹Σ⁺–X̃ 369–421 nm
¹Σ⁺–X̃ 360–420 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺	1	CH stretch	3127(7)	gas	MPD	1
			3136(20)	Ne	AB	1
	3		2276(7)	gas	MPD	1
			2238(20)	Ne	AB	1,2
			1996(15)	gas	MPD	1,3
			1982(20)	Ne	AB	1–3
		CC stretch	1838(15)	gas	MPD	3
			1597(15)	gas	MPD	3
			1280(15)	gas	MPD	1,3
			1244(20)	Ne	AB	1–3
			579(15)	gas	MPD	1,3
			539(20)	Ne	AB	1,2
			490(15)T	gas	MPD	1,3
			463(20)T	Ne	AB	1–3

²X̃¹Σ⁺C_{∞v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺		CC stretch	1958	Ne	IR	2
		CC stretch	1888	Ne	IR	2

References

- ¹D. A. Kirkwood, H. Linnartz, M. Grutter, O. Dopfer, T. Motylewski, M. Pachkov, M. Tulej, M. Wyss, and J. P. Maier, *Faraday Discuss.* **109**, 109 (1998).
²M. Grutter, M. Wyss, and J. P. Maier, *J. Chem. Phys.* **110**, 1492 (1999).
³D. A. Kirkwood, M. Tulej, M. V. Pachkov, M. Schnaiter, F. Güthe, M. Grutter, M. Wyss, J. P. Maier, and G. Fischer, *J. Chem. Phys.* **111**, 9280 (1999).

HC₁₄H

¹Σ_u⁺ D_{∞h}
T₀ = 36377(37) Ne AB¹ ¹Σ_u⁺-X̃ 236–275 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺		CC stretch	2040(37)	Ne	AB	1

Reference

¹M. Grutter, M. Wyss, J. Fulara, and J. P. Maier, J. Phys. Chem. A **102**, 9785 (1998).

HC₁₄H⁻

(3) ²Π_u D_{∞h}
T₀ = 29070(17) Ne AB¹

(2) ²Π_u D_{∞h}
T₀ = 12309(4) Ne AB¹ (2)²Π_u-X̃ 639–813 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺		CH stretch	3335(7)	Ne	AB	1
		CC stretch	2205(7)	Ne	AB	1

(1) ²Π_u D_{∞h}
T₀ = 11633(3) Ne AB¹ (1)²Π_u-X̃ 734–860 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺		CC stretch	1976(5)	Ne	AB	1

Reference

¹M. Grutter, M. Wyss, J. Fulara, and J. P. Maier, J. Phys. Chem. A **102**, 9785 (1998).

C₁₅

¹Σ_u⁺ D_{∞h}
T₀ = 23832(12) Ne AB¹ ¹Σ_u⁺-X̃ 410–420 nm
22371 Ar AB³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			277(20)	Ne	AB	1

T₀ = 16090(5)T Ne AB¹ 611–622 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	7		266(7)	Ne	AB	1

\tilde{a} D_{∞h}
T^a = 6370(80) gas PE²

^aFrom vertical ionization potentials.

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- ¹D. Forney, P. Freivogel, M. Grutter, and J. P. Maier, J. Chem. Phys. **104**, 4954 (1996).
²M. Kohno, S. Suzuki, H. Shiromaru, and Y. Achiba, J. Chem. Phys. **110**, 3781 (1999).
³J. Szczepanski, J. Fuller, S. Ekern, and M. Vala, Spectrochim. Acta A **57**, 775 (2001).

C₁₅⁻

$\tilde{B}^2\Pi_u$ D_{∞h}
T₀ = 10100(100) gas MPD² $\tilde{B}-\tilde{X}$ 833–990 nm

$\tilde{A}^2\Pi_u$ D_{∞h}
T₀ = 8300(100) gas MPD² $\tilde{A}-\tilde{X}$ 1123–1205 nm
8316 Ne AB¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			300T	gas	MPD	2

References

- ¹M. Wyss, M. Grutter, and J. P. Maier, Chem. Phys. Lett. **304**, 35 (1999).
²M. Ohara, D. Kasuya, H. Shiromaru, and Y. Achiba, J. Phys. Chem. A **104**, 8622 (2000).

cum-C₁₅H⁻ ^a

¹A' C_s
T₀ = 23284(10) gas MPD¹

^aThis isomer of C₁₅H⁻, with cumulenic bonding, lies at a somewhat higher energy than the acetylenic isomer.

Reference

- ¹M. Tulej, F. Güthe, M. Schnaiter, M. V. Packhov, D. A. Kirkwood, J. P. Maier, and G. Fischer, J. Phys. Chem. A **103**, 9712 (1999).

acet-C₁₅H⁻

³Σ⁻ C_{∞v}
T₀ = 19434(10) gas MPD¹ ³Σ⁻-X̃ 498–515 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺		C–C stretch	618(10)	gas	MPD	1

Reference

- ¹M. Tulej, F. Güthe, M. Schnaiter, M. V. Packhov, D. A. Kirkwood, J. P. Maier, and G. Fischer, J. Phys. Chem. A **103**, 9712 (1999).

C₁₆

\tilde{X}		D _{∞h} ?				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			2200	N ₂	Ra	1
			2096	N ₂	Ra	1
			1920	N ₂	Ra	1
			1372T	N ₂	Ra	1
			735	N ₂	Ra	1
			646	N ₂	Ra	1
			468	N ₂	Ra	1
			249	N ₂	Ra	1

Reference

¹A. K. Ott, G. A. Rechtsteiner, C. Felix, O. Hampe, M. F. Jarrold, and R. P. Van Duyne, *J. Chem. Phys.* **109**, 9652 (1998).

C₁₆⁻

(2) ² Π _u		D _{∞h}				
T ₀ = 9100(100)		gas MPD ²				
(1) ² Π _u		D _{∞h}				
T ₀ = 5800(100)		gas MPD ²				
	5784(5)	Ne	AB ¹		² Π _u - \tilde{X}	1587–1725 nm
	5732(2)	Ar	AB ³		² Π _u - \tilde{X}	1658–1730 nm
					² Π _u - \tilde{X}	1668–1745 nm
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ _g ⁺	8		245(5)	Ne	AB	1
			263(3)	Ar	AB	3

References

¹P. Freivogel, J. Fulara, M. Jakobi, D. Forney, and J. P. Maier, *J. Chem. Phys.* **103**, 54 (1995).

²M. Ohara, D. Kasuya, H. Shiromaru, and Y. Achiba, *J. Phys. Chem. A* **104**, 8622 (2000).

³J. Szczepanski, J. Fuller, S. Ekern, and M. Vala, *Spectrochim. Acta A* **57**, 775 (2001).

8.16. Non-Hydrocarbons With More Than Eight Atoms**CH₃BNCH₃**

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			2060.1	Ar	IR	1
			2058.9			

Reference

¹D. V. Lanzisera and L. Andrews, *J. Phys. Chem. A* **101**, 824 (1997).

C₂H₅ZnH

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH stretch	2977.7	Ar	IR	1
		CH stretch	2974.0	Ar	IR	1
		CH stretch	2960.2	Ar	IR	1
		ZnH stretch	1869.3	Ar	IR	1
		CH ₃ deform.	1465.4	Ar	IR	1
		CH ₂ bend	1436.9	Ar	IR	1
		CH ₃ deform.	1378.0	Ar	IR	1
		CC stretch	960.3	Ar	IR	1
		CH ₂ wag	829.3	Ar	IR	1
			827.2			
		CZn stretch	545.8	Ar	IR	1

Reference

¹V. A. Bracken, N. Legay-Sommaire, and J. G. McCaffrey, *J. Phys. Chem. A* **101**, 9863 (1997).

(CH₃)₂GaH

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	3	GaH stretch	1869.5	Ar	IR	1
b ₁	16	CH ₃ umbrella	1202.5	Ar	IR	1
	18	GaC ₂ a-stretch	589.5	Ar	IR	1
	19	CGaH deform.	556.0	Ar	IR	1
b ₂	22	CH ₃ deform.	729.0	Ar	IR	1
	23	H deform.	415.5	Ar	IR	1

Reference

¹J. Müller, H. Sternkicker, U. Bergmann, and B. Atakan, *J. Phys. Chem. A* **104**, 3627 (2000).

CH₃SiH=CH₂

In the gas phase, absorption between 220 and 300 nm, with a maximum at 38460 (260 nm) has been assigned⁷ to CH₃SiH=CH₂. In an argon matrix, an absorption maximum at 38460 (260 nm) has also been assigned^{1,3,4} to this product.

\bar{X}^a		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		CH stretch	3018w	Ar	IR	5
		CH stretch	2976w	Ar	IR	5
		SiH stretch	2181.1s	Ar	IR	1-6
		CH ₃ deform.	1411.7w	Ar	IR	3,5,6
		CH ₂ scissors	1297.4w	Ar	IR	1,3-6
		CH ₃ deform.	1254.2s	Ar	IR	1-6
		Si=C stretch	989.1s	Ar	IR	1-6
		SiH deform.	878.4s	Ar	IR	1-3,5,6
		Deformation	810.3vs	Ar	IR	1-6
		Si-C stretch	677.9wm	Ar	IR	3,5,6
	a''		CH ₃ deform.	1393.6w	Ar	IR
		CH ₂ wag	713.7m	Ar	IR	1-6
		CH deform.	615.9m	Ar	IR	1,3-6

^aAssigned using the density functional theory calculations of Ref. 8.

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- V. N. Khabashesku, K. N. Kudin, and J. L. Margrave, *J. Mol. Struct.* **443**, 175 (1998).

(CH₃)₂Si=CH₂

\bar{X}^a		C_{2v} Structure: ED ³				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		CH stretch	2991.6w	Ar	IR	7
		CH stretch	2968.7w	Ar	IR	7
		CH ₂ scissors	1413.6w	Ar	IR	7
		CH ₃ rock	1265.2w	Ar	IR	7
		Si=C stretch	1003.8s	Ar	IR	1,2,4,5,7
		CH ₃ rock	818.2m	Ar	IR	1,4,5,7
		Si-C s-stretch	625.1wm	Ar	IR	7
b_1		CH stretch	2896.5w	Ar	IR	7
		CH ₃ rock	1423.3wm	Ar	IR	7
b_2		CH ₂ wag	643.9m	Ar	IR	1,2,4,5,7
		CH stretch	2862.0w	Ar	IR	7
		CH ₃ rock	1261.6wm	Ar	IR	1,4,5,7
		Deformation	826.2vs	Ar	IR	1,2,4,5,7
	Si-C a-stretch	711.6w	Ar	IR	7	

(CD₃)₂Si=CD₂

\bar{X}^a		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		Si=C stretch	1112.0ms	Ar	IR	6
		CD ₃ deform.	1002.5sh	Ar	IR	6
		Mixed	866.5m	Ar	IR	6
b_1		CD ₃ rock	651.2m	Ar	IR	6
		CD ₂ wag	501.6s	Ar	IR	6
b_2		CD ₃ deform.	1005.0m	Ar	IR	6
		SiC a-stretch	732.0vs	Ar	IR	6

^aAssigned using the density functional theory calculations of Ref. 8.

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- V. N. Khabashesku, K. N. Kudin, and J. L. Margrave, *J. Mol. Struct.* **443**, 175 (1998).

(CH₃)₂Ge=CH₂

\bar{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1		=CH ₂ a-stretch	3008.0	Ar	IR	1
		CH s-stretch	2973.3	Ar	IR	1
		CH ₂ scissors	1348.3	Ar	IR	1
		Mixed	847.3	Ar	IR	1
		Ge=C stretch	818.8	Ar	IR	1
		Ge-C s-stretch	576.0	Ar	IR	1
	b_1		CH a-stretch	2927.9	Ar	IR
		CH ₃ deform.	1416.8	Ar	IR	1
b_2		CH ₂ wag	596.0	Ar	IR	1
		CH a-stretch	2874.0	Ar	IR	1
		CH ₃ rock	1241.7	Ar	IR	1
		Deformation	804.4	Ar	IR	1
	Ge-C a-stretch	580.1	Ar	IR	1	

Reference

- V. N. Khabashesku, K. N. Kudin, J. Tamás, S. E. Boganov, J. L. Margrave, and O. M. Nefedov, *J. Am. Chem. Soc.* **120**, 5005 (1998).

(CH₃)₂Si=CHCH₃

\tilde{X}^a		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>		CH stretch	3015.3w	Ar	IR	1,2,4
		CH stretch	2980m	Ar	IR	1,2
		CH stretch	2933.1wm	Ar	IR	1,2,4
		CH stretch	2869.0wm	Ar	IR	1,2,4
		CH ₃ deform.	1465.2wm	Ar	IR	1,2,4
		CH ₃ deform.	1378.8wm	Ar	IR	1,2,4
		CH bend	1314.9w	Ar	IR	1,2,4
		CH ₃ rock	1260.2m	Ar	IR	1,2,4
		C-C stretch	1116.1w	Ar	IR	1,2,4
		Deformation	984.9m	Ar	IR	1,2,4
		Mixed	882.7ms	Ar	IR	1,2,4
		CH ₃ rock	794.5vs	Ar	IR	1-4
		Mixed	771.9wm	Ar	IR	4
		SiC a-stretch	712m	Ar	IR	1,2
			708sh			
<i>a''</i>		SiC s-stretch	607.7w	Ar	IR	1-4
		Deformation	358m	Ar	IR	1,2
		CH stretch	2965m	Ar	IR	1,2
		CH stretch	2901.8wm	Ar	IR	1,2,4
			2897.5wm			
		CH ₃ deform.	1407.8wm	Ar	IR	1,2,4
	CH ₃ rock	808	Ar	IR	1,2	
	CH OPLA	645.7wm	Ar	IR	1-4	
		644.1wm				

^aAssigned using the density functional calculations of Ref. 5.

References

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SiC₉

$\tilde{X}^3\Sigma$		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	4		1935.8	Ar	IR	1

Reference

- X. D. Ding, S. L. Wang, C. M. L. Rittby, and W. R. M. Graham, *J. Chem. Phys.* **110**, 11214 (1999).

FCCCF=CFCCF

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁			2337.3vs	Ne	IR	1
			1678.0w	Ne	IR	1
			1072.1vs	Ne	IR	1
<i>b</i> ₂			1398.5wm	Ne	IR	1
			912.5wm	Ne	IR	1

Reference

- H. H. Wenk, A. Balster, W. Sander, D. A. Hrovat, and W. T. Borden, *Angew. Chem. Int. Ed.* **40**, 2295 (2001).

***p*-ClCC₆H₄CCl**

In an argon matrix,² a structured absorption maximum at 22730 (440 nm) has been assigned to *p*-ClCC₆H₄CCl. In a nitrogen matrix,¹ bands between 21640 (462 nm) and 24330 (411 nm) are contributed by this species.

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1582s	Ar	IR	2
			1565s	N ₂	IR	1
			1386w	Ar	IR	2
			1388w	N ₂	IR	1
			1323w	N ₂	IR	1
			1190m	Ar	IR	2
			1117w	N ₂	IR	1
			1019m	Ar	IR	2
			958m	Ar	IR	2
			958m	N ₂	IR	1
			939w	Ar	IR	2
			940w	N ₂	IR	1
			918s	Ar	IR	2
			922m	N ₂	IR	1
			918s			
			896w	Ar	IR	2
			897w	N ₂	IR	1
			805m	Ar	IR	2
			809m	N ₂	IR	1
			807m			
			709w	Ar	IR	2
			708m	N ₂	IR	1
			689m	Ar	IR	2
			687m	N ₂	IR	1
			600w	Ar	IR	2
			600m	N ₂	IR	1
			464w	N ₂	IR	1

References

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- H. Tomioka, K. Komatsu, T. Nakayama, and M. Shimizu, *Chem. Lett.* 1291 (1993).

CaOC₂H₅

$\tilde{C}^2\Delta$ $C_{\infty v}^a$
 $T_0 = 21650\text{T}$ gas LF² $\tilde{C}-\tilde{X}$ 449–464 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CaO stretch	370(5)	gas	LF	2
		OCC bend	360(5)	gas	LF	2
		CaOC bend	80(5)	gas	LF	2

$\tilde{A}^2\Pi$ $C_{\infty v}^a$
 $T_0 = 15890(5)$ gas LF¹ $\tilde{A}-\tilde{X}$ 623–635 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CaOC bend	531(5)	gas	LF	1
		CaO stretch	402(5)	gas	LF	1

$A = 63(5)$ gas LF¹

$\tilde{X}^2\Sigma^+$ $C_{\infty v}^a$

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CaOC bend	512(5)	gas	LF	1
		CaO stretch	385(5)	gas	LF	1
		CaOC bend	85(5)	gas	LF	1

^aSymmetry given with respect to the spectroscopically active CaOC group.

References

¹C. R. Brazier, L. C. Ellingboe, S. Kinsey-Nielsen, and P. F. Bernath, *J. Am. Chem. Soc.* **108**, 2126 (1986).

²M. Elhanine, R. Lawruszczuk, and B. Soep, *Chem. Phys. Lett.* **288**, 785 (1998).

C₆H₅N≡N⁺

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		N≡N stretch	2327vs	Ar	IR	1

Reference

¹M. Winkler and W. Sander, *Angew. Chem. Int. Ed.* **39**, 2014 (2000).

H(C≡C)₃CN⁺

$\tilde{A}^2\Pi$ $C_{\infty v}$
 $T_0 = 14925.42$ gas AB³
 14836 Ne AB¹LF² $\tilde{A}-\tilde{X}$ 594–820 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	5	CN stretch	1969(5)	Ne	AB	1
	6	C–C stretch	1578(5)	Ne	AB	1
	7	C–C stretch	1017(5)	Ne	AB	1
	8	C–C stretch	459(5)	Ne	AB	1

$B_0 = 0.0188$ AB³

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+	2	C≡N stretch	2180.2	Ne	LF	2
	3	C≡C stretch	2159.6	Ne	LF	2
	4	C≡C stretch	2128.5	Ne	LF	2
	5	C≡C stretch	1938.3	Ne	LF	2
	6	C–C stretch	957.2	Ne	LF	2
	7	C–C stretch	881.0	Ne	LF	2
	8	C–C stretch	464.5	Ne	LF	2

$B_0 = 0.0190$ AB³

D(C≡C)₃CN⁺

$\tilde{A}^2\Pi$ $C_{\infty v}$
 $T_0 = 14937.31$ gas AB³
 $B_0 = 0.0183$ AB³

$\tilde{X}^2\Pi$ $C_{\infty v}$
 $B_0 = 0.0183$ AB³

References

¹D. Forney, P. Freivogel, J. Fulara, and J. P. Maier, *J. Chem. Phys.* **102**, 1510 (1995).

²A. M. Smith, J. Agreiter, and V. E. Bondybey, *Chem. Phys. Lett.* **244**, 379 (1995).

³W. E. Sinclair, D. Pfluger, D. Verdes, and J. P. Maier, *J. Chem. Phys.* **112**, 8899 (2000).

H(C≡C)₃NC

\tilde{X} $C_{\infty v}$
 $B_0 = 0.0194$ MW¹

Reference

¹P. Botschwina, A. Heyl, W. Chen, M. C. McCarthy, J.-U. Grabow, M. J. Travers, and P. Thaddeus, *J. Chem. Phys.* **109**, 3108 (1998).

C₉N

$^2\Pi$ $C_{\infty v}$
 $T_0 = 14802(4)$ Ne AB¹ $^2\Pi-\tilde{X}$ 529–676 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		CC,CN stretch	2082(6)	Ne	AB	1
		C–C stretch	1480(6)	Ne	AB	1

Reference

¹M. Grutter, M. Wyss, and J. P. Maier, *J. Chem. Phys.* **110**, 1492 (1999).

C₉N⁻
 $^1\Sigma^+$ $C_{\infty v}$
 $T_0 = 30656(19)$ Ne AB¹ $^1\Sigma^+ - \tilde{X}$ 294–327 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		CC,CN stretch	2099(27)	Ne	AB	1
		C–C stretch	1283(27)	Ne	AB	1
		C–C stretch	448(27)	Ne	AB	1

$\tilde{X}^1\Sigma^+$		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		CC,CN stretch	2033	Ne	IR	1

Reference
¹M. Grutter, M. Wyss, and J. P. Maier, J. Chem. Phys. **110**, 1492 (1999).
C₁₁N
 $^2\Pi$ $C_{\infty v}$
 $T_0 = 13207(4)$ Ne AB¹ $^2\Pi - \tilde{X}$ 654–758 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		CC,CN stretch	2070(6)	Ne	AB	1
		C–C stretch	1318(6)	Ne	AB	1

Reference
¹M. Grutter, M. Wyss, and J. P. Maier, J. Chem. Phys. **110**, 1492 (1999).
C₁₁N⁻
 $^1\Sigma^+$ $C_{\infty v}$
 $T_0 = 26295(14)$ Ne AB¹ $^1\Sigma^+ - \tilde{X}$ 346–381 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		CC,CN stretch	2074(20)	Ne	AB	1
		C–C stretch	429(20)	Ne	AB	1

Reference
¹M. Grutter, M. Wyss, and J. P. Maier, J. Chem. Phys. **110**, 1492 (1999).
HC₁₀CN⁻
 $(3)^2\Pi$ $C_{\infty v}$
 $T_0 = 32895(22)$ Ne AB¹

$(2)^2\Pi$		$C_{\infty v}$				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		CH stretch	3338(9)	Ne	AB	1
		CC stretch	2297(9)	Ne	AB	1
		CC stretch	1974(9)	Ne	AB	1

 $(1)^2\Pi$ $C_{\infty v}$
 $T_0 = 13466(4)$ Ne AB¹ $(1)^2\Pi - \tilde{X}$ 566–743 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+			2117(7)	Ne	AB	1

Reference
¹M. Grutter, M. Wyss, J. Fulara, and J. P. Maier, J. Phys. Chem. A **102**, 9785 (1998).
C₁₃N
 $^2\Pi$ $C_{\infty v}$
 $T_0 = 12085(3)$ Ne AB¹ $^2\Pi - \tilde{X}$ 750–828 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ^+		C–C stretch	1238(4)	Ne	AB	1

Reference
¹M. Grutter, M. Wyss, and J. P. Maier, J. Chem. Phys. **110**, 1492 (1999).
C₁₃N⁻
 $^1\Sigma^+$ $C_{\infty v}$
 $T_0 = 22619(10)$ Ne AB¹
Reference
¹M. Grutter, M. Wyss, and J. P. Maier, J. Chem. Phys. **110**, 1492 (1999).

HC₁₂CN⁻(3) ²Π C_{∞v}
T₀ = 29762(18) Ne AB¹(2) ²Π C_{∞v}
T₀ = 12692(3) Ne AB¹ (2)²Π- \tilde{X} 624–788 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺		CH stretch	3326(7)	Ne	AB	1
		CC stretch	2173(7)	Ne	AB	1
		CC stretch	1902(7)	Ne	AB	1

(1) ²Π C_{∞v}
T₀ = 12038(3) Ne AB¹ (1)²Π- \tilde{X} 711–831 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
Σ ⁺		CC stretch	2019(6)	Ne	AB	1

Reference¹M. Grutter, M. Wyss, J. Fulara, and J. P. Maier, *J. Phys. Chem. A* **102**, 9785 (1998).**Cl₃TiOCH₃** \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH ₃ stretch	3006	Ar	IR	1
		CH ₃ stretch	2945	Ar	IR	1
		CH ₃ deform.	1449	Ar	IR	1
		C–O stretch	1152vs	Ar	IR	1
		CH ₃ rock	1109	Ar	IR	1
		CH ₃ rock	1095	Ar	IR	1
		TiO stretch	636	Ar	IR	1
		TiCl stretch	486vs	Ar	IR	1

Cl₃TiOCD₃ \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CD ₃ stretch	2237	Ar	IR	1
		CD ₃ stretch	2086	Ar	IR	1
		C–O stretch	1196	Ar	IR	1
		CD ₃ deform.	1047	Ar	IR	1
		CD ₃ rock	859	Ar	IR	1
		CD ₃ rock	855	Ar	IR	1
		TiO stretch	603	Ar	IR	1
		TiCl stretch	498	Ar	IR	1

Reference¹B. S. Ault and J. B. Everhart, *J. Phys. Chem.* **100**, 15726 (1996).**CH₂COCH₃** \tilde{B} T₀ = 27282.7(5) gas LF¹⁻³ \tilde{B} - \tilde{X} 331–369 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1668(25) ^a	gas	LF	3
			1303(25) ^a	gas	LF	3
			829(25) ^a	gas	LF	3
			454(25) ^a	gas	LF	3

τ₀ = 120(10) ns gas LF¹^aA series of low-frequency torsional transitions is superposed on each vibronic band.³**References**¹S. Williams, E. Zingher, and J. C. Weisshaar, *J. Phys. Chem. A* **102**, 2297 (1998).²N. Washida, S. Inomata, and M. Furubayashi, *J. Phys. Chem. A* **102**, 7924 (1998).³S. Williams, L. B. Harding, J. F. Stanton, and J. C. Weisshaar, *J. Phys. Chem. A* **104**, 10131 (2000).**CH₃CHCHO** \tilde{B} T₀ = 29090.1(5)T gas LF^{1,2} \tilde{B} - \tilde{X} 330–347 nmτ₀ = 190(10) ns gas LF¹**References**¹S. Williams, E. Zingher, and J. C. Weisshaar, *J. Phys. Chem. A* **102**, 2297 (1998).²N. Washida, S. Inomata, and M. Furubayashi, *J. Phys. Chem. A* **102**, 7924 (1998).**n-C₃H₇O** \tilde{B} T₀ = 28634 gas LF¹⁻³ \tilde{B} - \tilde{X} 335–520 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			969	gas	LF	3
		C ₃ stretch	818	gas	LF	3
			796	gas	LF	3
			785	gas	LF	3
			647	gas	LF	3
		CO stretch	582	gas	LF	2,3
		CCO deformation	311	gas	LF	3
		C ₃ deformation	260	gas	LF	3
		CH ₃ torsion	203	gas	LF	3
		Skeletal flex	149	gas	LF	3

τ = 900(50) ns gas LF^{1,2} \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CO stretch	1000(50)	gas	LF	1,2

References

- ¹J. Bai, H. Okabe, and M. K. Emadi-Babaki, *J. Photochem. Photobiol., A: Chem.* **50**, 163 (1989).
²Ch. Mund, Ch. Fockenberg, and R. Zellner, *Ber. Bunsenges. Phys. Chem.* **102**, 709 (1998).
³C. C. Carter, J. R. Atwell, S. Gopalakrishnan, and T. A. Miller, *J. Phys. Chem. A* **104**, 9165 (2000).

 $(\text{CH}_3)_2\text{CHO}$

\tilde{B} C_s
 $T_0 = 27171$ gas $\text{EM}^{1,4}\text{LF}^{2,3,5,7}$ $\tilde{B}-\tilde{X}$ 330–540 nm

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'			1212	gas	LF	7
			1152	gas	LF	7
			938	gas	LF	7
		C_3 stretch	860	gas	LF	7
		$C-O$ stretch	574.4 (ω)	gas	LF	2,3,5,7
		CCO s-deformation	466	gas	LF	7
		CCO a-deform.	449	gas	LF	7
		C_3 deformation	378	gas	LF	7
		CH_3 s-torsion	357	gas	LF	7
		CH_3 a-torsion	344	gas	LF	7
			290(10)	gas	LF	5

$\tau = 0.64(9)$ μs gas EM^4

\tilde{A}^2A'' C_s
 $T_0 = 1225(65)$ gas PE^6

\tilde{X}^2A' C_s

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'			960(20)	gas	EM,LF	4–6
		Skel. s-deform.	500(15)	gas	PE	6
		CCC bend	375(25)	gas	PE	6

References

- ¹K. Ohbayashi, H. Akimoto, and I. Tanaka, *J. Phys. Chem.* **81**, 798 (1977).
²R. J. Balla, H. H. Nelson, and J. R. McDonald, *Chem. Phys.* **99**, 323 (1985).
³S. C. Foster, Y.-C. Hsu, C. P. Damo, X. Liu, C.-Y. Kung, and T. A. Miller, *J. Phys. Chem.* **90**, 6766 (1986).
⁴J. Bai, H. Okabe, and J. B. Halpern, *Chem. Phys. Lett.* **149**, 37 (1988).
⁵Ch. Mund, Ch. Fockenberg, and R. Zellner, *Ber. Bunsenges. Phys. Chem.* **102**, 709 (1998).
⁶T. M. Ramond, G. E. Davico, R. L. Schwartz, and W. C. Lineberger, *J. Chem. Phys.* **112**, 1158 (2000).
⁷C. C. Carter, J. R. Atwell, S. Gopalakrishnan, and T. A. Miller, *J. Phys. Chem. A* **104**, 9165 (2000).

 $(\text{CH}_3)_2\text{CHO}^-$

Threshold for electron detachment from ground-state $(\text{CH}_3)_2\text{CHO}^- = 14900(30)$ gas $\text{PE}^{1,2}$

\tilde{X}^1A' C_s

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a'		CCC bend	330(25)	gas	PE	2

References

- ¹G. B. Ellison, P. C. Engelking, and W. C. Lineberger, *J. Phys. Chem.* **86**, 4873 (1982).
²T. M. Ramond, G. E. Davico, R. L. Schwartz, and W. C. Lineberger, *J. Chem. Phys.* **112**, 1158 (2000).

 $(\text{CH}_3)_2\text{Ge=O}$

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1		CH_3 rock	1239.1w	Ar	IR	1
		Ge=O stretch	943.0vs	Ar	IR	1
		CH_3 rock	857.4m	Ar	IR	1
b_1		CH_3 deform.	1441.2w	Ar	IR	1
		CH_3 rock	794.8m	Ar	IR	1
b_2		CH_3 rock	1234.0w	Ar	IR	1
		CH_3 rock	770.7m	Ar	IR	1
		GeC a-stretch	605.0s	Ar	IR	1

Reference

- ¹V. N. Khabashesku, S. E. Boganov, K. N. Kudin, J. L. Margrave, and O. M. Nefedov, *Organomet.* **17**, 5041 (1998).

 $(\text{CH}_3)_2\text{Ge=S}$

\tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.	
a_1	3	CH_3 deform.	1407	Ar	IR	2	
	4	CH_3 deform.	1237	Ar	IR	2	
	5	CH_3 rock	850s	Ar	IR	1,2	
	6	Ge=S stretch	605vs	Ar	IR	1,2	
	7	GeC s-stretch	516m	Ar	IR	1,2	
	b_1	14	CH_3 a-deform.	1390m	Ar	IR	1,2
		15	CH_3 deform.	761m	Ar	IR	1,2
b_2	21	CH_3 deform.	1229m	Ar	IR	1,2	
	22	CH_3 rock	809	Ar	IR	2	
	23	GeC a-stretch	574	Ar	IR	2	

References

- ¹J. Barrau, V. Balaji, and J. Michl, *Organomet.* **8**, 2034 (1989).
²V. N. Khabashesku, S. E. Boganov, P. S. Zuev, O. M. Nefedov, J. Tamás, A. Gömöry, and I. Besenyi, *J. Organomet. Chem.* **402**, 161 (1991).

CH₃CHCOCH₃

\tilde{B}
gas LF¹ $\tilde{B}-\tilde{X}$ 331–366 nm

Reference

¹N. Washida, S. Inomata, and M. Furubayashi, J. Phys. Chem. A **102**, 7924 (1998).

(CH₃)₂CCHO

\tilde{B}
gas LF¹ $\tilde{B}-\tilde{X}$ 334–340 nm

Reference

¹N. Washida, S. Inomata, and M. Furubayashi, J. Phys. Chem. A **102**, 7924 (1998).

(CH₃)₂CCOCH₃

\tilde{B}
gas LF¹ $\tilde{B}-\tilde{X}$ 330–364 nm

Reference

¹N. Washida, S. Inomata, and M. Furubayashi, J. Phys. Chem. A **102**, 7924 (1998).

1-C₄H₉O

\tilde{B} C_s
 $T_0=28649$ gas LF¹ $\tilde{B}-\tilde{X}$ 335–350 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>			710	gas	LF	1
			612	gas	LF	1
			512	gas	LF	1
			444	gas	LF	1
			334	gas	LF	1
			324	gas	LF	1
			271	gas	LF	1
		CH ₃ torsion		207	gas	LF

Reference

¹C. C. Carter, J. R. Atwell, S. Gopalakrishnan, and T. A. Miller, J. Phys. Chem. A **104**, 9165 (2000).

2-C₄H₉O

\tilde{B}
 $T_0=26757$ gas LF^{1,2} $\tilde{B}-\tilde{X}$ 350–380 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CC stretch	918	gas	LF	2
		CC stretch	729	gas	LF	2
		CO stretch	559	gas	LF	1,2
			470	gas	LF	2

$\tau \approx 85$ ns gas LF¹

References

¹C. Wang, L. G. Shemesh, W. Deng, M. D. Lilien, and T. S. Dibble, J. Phys. Chem. A **103**, 8207 (1999).
²C. C. Carter, J. R. Atwell, S. Gopalakrishnan, and T. A. Miller, J. Phys. Chem. A **104**, 9165 (2000).

(CH₃)₃CO

\tilde{B} C_{3v}
 $T_0=25861$ gas LF^{1,2,4,5} $\tilde{B}-\tilde{X}$ 333–387 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			975	gas	LF	5
			855	gas	LF	5
			760	gas	LF	5
		CO stretch	546 (ω)	gas	LF	1,2,4,5
			311	gas	LF	1,2,4,5

$\tau \approx 1.5(1)$ μ s gas LF⁴

\tilde{X}^2E C_{3v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a'</i>		CO stretch	1205(15)	gas	PE	3
		C ₃ umbrella	435(15)	gas	PE	3

References

¹M. Blitz, M. J. Pilling, S. H. Robertson, and P. W. Seakins, Phys. Chem. Chem. Phys. **1**, 73 (1999).
²C. Wang, L. G. Shemesh, W. Deng, M. D. Lilien, and T. S. Dibble, J. Phys. Chem. A **103**, 8207 (1999).
³T. M. Ramond, G. E. Davico, R. L. Schwartz, and W. C. Lineberger, J. Chem. Phys. **112**, 1158 (2000).
⁴Ch. Lotz and R. Zellner, Phys. Chem. Chem. Phys. **2**, 2353 (2000).
⁵C. C. Carter, J. R. Atwell, S. Gopalakrishnan, and T. A. Miller, J. Phys. Chem. A **104**, 9165 (2000).

(CH₃)₃CO⁻

Threshold for electron detachment from ground-state (CH₃)₃CO⁻ = 15400(30) gas PE^{1,2}

\tilde{X}		C _{3v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		C ₃ umbrella	440(25)	gas	PE	2

References

- ¹G. B. Ellison, P. C. Engelking, and W. C. Lineberger, *J. Phys. Chem.* **86**, 4873 (1982).
²T. M. Ramond, G. E. Davico, R. L. Schwartz, and W. C. Lineberger, *J. Chem. Phys.* **112**, 1158 (2000).

C₉O

\tilde{X}		C _{∞v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			2239.5wmT	Ar	IR	2
			2040.7vsT	Ar	IR	2

B₀ = 0.010 MW¹

References

- ¹T. Ogata, Y. Ohshima, and Y. Endo, *J. Am. Chem. Soc.* **117**, 3593 (1995).
²M. Dibben, J. Szczepanski, C. Wehlburg, and M. Vala, *J. Phys. Chem. A* **104**, 3584 (2000).

cyc-(N=CHN=CHC)=C=O

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	1	CH stretch	3036.1vw	Ar	IR	1
	3	CCO stretch	2152.7vs	Ar	IR	1
	4	Mixed	1493.5m	Ar	IR	1
	5	C=C stretch	1402.0vw	Ar	IR	1
	6	Mixed	1359.0m	Ar	IR	1
	7	Mixed	1278.3w	Ar	IR	1
	8	CH deform.	1197.5vw	Ar	IR	1
	9	Mixed	1177.8w	Ar	IR	1
	10	Mixed	1077.3w	Ar	IR	1
	11	Ring deform.	946.4m	Ar	IR	1
a''	18	Ring deform.	627.1vw	Ar	IR	1
	19	Ring deform.	616.3vw	Ar	IR	1
	20	CCO deform.	537.4vw	Ar	IR	1

Reference

- ¹G. Maier and J. Endres, *Eur. J. Org. Chem.* 2535 (2000).

Cl₂V(O)OCH₃

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH ₃ a-stretch	2940	Ar	IR	1
		CH ₃ a-stretch	2933	Ar	IR	1
		CH ₃ s-stretch	2827	Ar	IR	1
		CH ₃ a-deform.	1447	Ar	IR	1
		CH ₃ a-deform.	1444	Ar	IR	1
		CH ₃ deform.	1430	Ar	IR	1
		CH ₃ rock	1152	Ar	IR	1
		CH ₃ rock	1121	Ar	IR	1
		C-O stretch	1069vs	Ar	IR	1
		V=O stretch	1030vs	Ar	IR	1
		V-O stretch	674s	Ar	IR	1
		VCl ₂ stretch	502vs	Ar	IR	1
		VCl ₂ stretch	446	Ar	IR	1

Cl₂V(O)OCD₃

\tilde{X}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CD ₃ a-stretch	2186	Ar	IR	1
		CD ₃ a-stretch	2179	Ar	IR	1
		CD ₃ s-stretch	2050	Ar	IR	1
		CD ₃ deform.	1085	Ar	IR	1
		C-O stretch	1066	Ar	IR	1
		CD ₃ a-deform.	1050	Ar	IR	1
		V=O stretch	1030	Ar	IR	1
		CD ₃ rock	897	Ar	IR	1
		CD ₃ rock	887	Ar	IR	1
		V-O stretch	646	Ar	IR	1
		VCl ₂ stretch	502	Ar	IR	1
		VCl ₂ stretch	443	Ar	IR	1

Reference

- ¹B. S. Ault, *J. Phys. Chem. A* **103**, 11474 (1999).

C₂H₅O₂

An unstructured gas-phase absorption between 200 and 290 nm, with maximum at 240 nm, has been assigned^{1,3-5,7,8} to C₂H₅O₂.

\tilde{A}		T ₀ = 7593(6) gas AB ² PE ⁹					$\tilde{A}-\tilde{X}$ 1175-1317 nm	
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.		
		OO stretch	918(9)	gas	AB,PE	2,9		
		Skel. bend	178(10)	gas	PE	9		

\tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH ₂ a-stretch	3016wm	Ar	IR	6
		CH ₂ scissors	1474m	Ar	IR	6
		CH ₂ deform.	1451sh	Ar	IR	6
		CH ₂ deform.	1389vs	Ar	IR	6
		CH ₂ deform.	1380sh	Ar	IR	6
		CH ₂ wag	1351m	Ar	IR	6
		CH ₂ twist	1242w	Ar	IR	6
		CH ₂ rock	1136m,br	Ar	IR	6
		OO stretch	1089(16)	gas	PE	9
			1112ms	Ar	IR	6
		CC stretch	1009s	Ar	IR	6
		CO stretch	838m	Ar	IR	6
		CH ₂ rock	800m	Ar	IR	6
		Skel. bend	499vs	Ar	IR	6
		Skel. bend	234(9)	gas	PE	9

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- S. J. Blanksby, T. M. Ramond, G. E. Davico, M. R. Nimlos, S. Kato, V. M. Bierbaum, W. C. Lineberger, G. B. Ellison, and M. Okumura, *J. Am. Chem. Soc.* **123**, 9585 (2001).

 $C_2H_5O_2^-$

Threshold for electron detachment from ground-state $C_2H_5O_2^- = 9570(30)$ gas PE¹

Reference

- S. J. Blanksby, T. M. Ramond, G. E. Davico, M. R. Nimlos, S. Kato, V. M. Bierbaum, W. C. Lineberger, G. B. Ellison, and M. Okumura, *J. Am. Chem. Soc.* **123**, 9585 (2001).

 $t-C_4H_9O_2$

A gas-phase absorption between 210 and 300 nm, with maximum near 240 nm, has been attributed² to $t-C_4H_9O_2$.

\tilde{A}^2A' C_s
T₀ = 7800(90) gas PE⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'		OO stretch	930(90)	gas	PE	4
			240(90)	gas	PE	4

 \tilde{X}^2A'' C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
		CH ₃ rock	1187(2)vs	Ar	IR	3
		CH ₃ rock	1139(2)m	Ar	IR	3
		OO stretch	1130(90)	gas	PE	4
			1124(2)s	Ar	IR	3
		CC stretch	808(2)ms	Ar	IR	3
			760(2)	gas	IR	1
		CO stretch	730(2)m	Ar	IR	3
			693.7(5)	gas	IR	1
		Skel. bend	539(2)ms	Ar	IR	3
		Skel. bend	403(2)wm	Ar	IR	3
		Skel. bend	361(2)m	Ar	IR	3
		Skel. bend	337(2)m	Ar	IR	3
			245(90)	gas	PE	4

References

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 $t-C_4H_9O_2^-$

Threshold for electron detachment from ground-state $t-C_4H_9O_2^- = 9650(90)$ gas PE¹

Reference

- E. P. Clifford, P. G. Wenthold, R. Gareyev, W. C. Lineberger, C. H. DePuy, V. M. Bierbaum, and G. B. Ellison, *J. Chem. Phys.* **109**, 10293 (1998).

 $(cyc-C_5H_4)O_2$

(Cyclopentadiene Dioxirane)^a

 \tilde{X}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1395s	Ar	IR	1,2
			1385vs	Ar	IR	1,2
			1184m	Ar	IR	1,2
			1179w	Ar	IR	1,2
			1142m	Ar	IR	1
			1023w	Ar	IR	1
			983w	Ar	IR	1
			938vw	Ar	IR	1
		C ₅ H ₄ deform.	895vs	Ar	IR	1,2
		C ₅ H ₄ deform.	741s	Ar	IR	1,2

^aThese absorptions were assigned by Ref. 1 to cyclopentadienone-O-oxide. However, Ref. 2 demonstrated that they are instead contributed by the dioxirane isomer.

References

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²I. R. Dunkin and C. J. Shields, *J. Chem. Soc., Chem. Commun.* 154 (1986).

O=C₆H₂=O

(Dehydrobenzoquinone)

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2	C≡C stretch	2285(20)	gas	PE	1
	3		1685(15)	gas	PE	1
	4		1535(15)	gas	PE	1
	5		1240(20)	gas	PE	1
	8		465(15)	gas	PE	1
	9		405(15)	gas	PE	1

Reference

- ¹G. E. Davico, R. L. Schwartz, T. M. Ramond, and W. C. Lineberger, *J. Am. Chem. Soc.* **121**, 6047 (1999).

O=C₆H₂=O⁻

Threshold for electron detachment from ground-state O=C₆H₂=O⁻ = 15000(40) gas PE¹

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	8		460(20)	gas	PE	1
	9		260(20)	gas	PE	1

Reference

- ¹G. E. Davico, R. L. Schwartz, T. M. Ramond, and W. C. Lineberger, *J. Am. Chem. Soc.* **121**, 6047 (1999).

CF₃OONO₂

In the gas phase,³ the onset of continuous absorption occurs near 340 nm. The intensity of this absorption increases out to the measurement limit, near 190 nm.

\tilde{X} C ₁						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	1	NO ₂ a-stretch	1764s	gas	IR	1-3
			1762s	Ar	IR	3
	2	NO ₂ s-stretch	1314sh	gas	IR	3
			1308m	Ar	IR	3
	3	CF ₃ s-stretch	1303s	gas	IR	1-3
			1290s	Ar	IR	3
	4	CF ₃ a-stretch	1243vs	gas	IR	1-3
			1238vs	Ar	IR	3
	5	CF ₃ a-stretch	1192vs	gas	IR	1-3
			1187s	Ar	IR	3
	6	OO stretch	958wm	gas	IR	1-3
			960wm	Ar	IR	3
	7	CO stretch	860w	gas	IR	3
			880w	Ar	IR	3
	8	NO ₂ scissors	792ms	gas	IR	1-3
			787m	Ar	IR	3
	9	ONO ₂ OPLA	710wm	gas	IR	1-3
			708w	Ar	IR	3
	11	CF ₃ a-deform.	674wm	gas	IR	1,3
			676w	Ar	IR	3
	12	CF ₃ a-deform.	608wm	gas	IR	1,3
			609w	Ar	IR	3
	13	NO ₂ rock	565w	gas	IR	1,3
			570w	Ar	IR	3
	14	N-OO stretch	495wm	gas	IR	1,3
			498w	Ar	IR	3
	15	CF ₃ rock	445vw	gas	IR	1,3
			450w	Ar	IR	3
	16	CF ₃ rock	380w	gas	IR	1,3
			286vw	gas	IR	1,3
	17	OO-NO ₂ deform.	286vw	gas	IR	1,3
			256w	gas	IR	1,3
	18	OO-CF ₃ deform.	256w	gas	IR	1,3

References

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²J. Chen, V. Young, T. Zhu, and H. Niki, *J. Phys. Chem.* **97**, 11696 (1993).
³R. Kopitzky, H. Willner, H.-G. Mack, A. Pfeiffer, and H. Oberhammer, *Inorg. Chem.* **37**, 6208 (1998).

8.17. Molecules Related to Benzene

3,5-C₆H₃F

\tilde{X} C _{2v}						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁		CF stretch	1440vs	Ar	IR	1
		Mixed	1184ms	Ar	IR	1
		CC s-bend	884m	Ar	IR	1
b ₁		CH wag	810m	Ar	IR	1
		CH wag	767wm	Ar	IR	1
		Mixed	532wm	Ar	IR	1
b ₂		Mixed	949wm	Ar	IR	1
		Skel. deform.	543s	Ar	IR	1

Reference

¹W. Sander and M. Exner, *J. Chem. Soc., Perkin Trans. 2*, 2285 (1999).

***o*-C₆F₄**

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁			1495.9wm	Ar	IR	1
			1459.1m	Ar	IR	1
			1289.0vw	Ar	IR	1
			1076.2m	Ar	IR	1
<i>b</i> ₂			1073.5wm			
			1488.6vs	Ar	IR	1
			1483.4m			
			981.4wm	Ar	IR	1
			972.7ms			
			586.6w	Ar	IR	1
		465.6w	Ar	IR	1	

Reference

¹H. H. Wenk and W. Sander, *Chem. Eur. J.* **7**, 1837 (2001).

***m*-C₆F₄**

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁			1823.8m	Ne	IR	1
			1817.7wm			
			1534.2vs	Ne	IR	1
			952.3vs	Ne	IR	1
			488.2w	Ne	IR	1
<i>b</i> ₁			1605.4vs	Ne	IR	1
			1497.0wm	Ne	IR	1
<i>b</i> ₂			1273.3wm	Ne	IR	1
			1268.5wm			
			992.7ms	Ne	IR	1
			981.3m			
			649.1w	Ne	IR	1
			511.6w	Ne	IR	1

Reference

¹H. H. Wenk and W. Sander, *Chem. Eur. J.* **7**, 1837 (2001).

***p*-C₆F₄**

\tilde{X}		D _{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>b</i> _{1u}			1406.6m	Ne	IR	1
			1117.3ms	Ne	IR	1
<i>b</i> _{2u}			1516.1s	Ne	IR	1
			1501.7s			
			924.6s	Ne	IR	1

Reference

¹H. H. Wenk, A. Balster, W. Sander, D. A. Hrovat, and W. T. Borden, *Angew. Chem. Int. Ed.* **40**, 2295 (2001).

2,4-C₆H₃OH

In an argon matrix,^{1,2} the first absorption region has a shoulder at 28090 (356 nm), maxima at 29070 and 31450 (344 and 318 nm), and a shoulder at 32900 (304 nm). The second absorption region has maxima at 39060 (256 nm) and 42020 (238 nm).

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> '	1	OH stretch	3612.0m	Ar	IR	1,2
	6		1516.3s	Ar	IR	1,2
	7		1429.0w	Ar	IR	1,2
	8		1368.2m	Ar	IR	1,2
	9		1290.1w	Ar	IR	1,2
	10		1259.4m	Ar	IR	1,2
	11		1209.1m	Ar	IR	1,2
	12	OH deform.	1157.6m	Ar	IR	1,2
	13		1128.6vw	Ar	IR	1,2
	14		971.0w	Ar	IR	1,2
	15		877.0w	Ar	IR	1,2
	16		641.2s	Ar	IR	1,2
	17		518.8s	Ar	IR	1,2
<i>a</i> ''	22		694.7m	Ar	IR	1,2

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***m*-C₆F₄I**

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> '			1636.8w	Ne	IR	1
			1562.3wm	Ne	IR	1
			1484.4vs	Ne	IR	1
			1416.6wm	Ne	IR	1
			1073.4ms	Ne	IR	1
			1030.4m	Ne	IR	1
			837.0wm	Ne	IR	1

Reference

¹H. H. Wenk and W. Sander, Chem. Eur. J. **7**, 1837 (2001).

***p*-C₆F₄I**

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁			1574.4w	Ne	IR	1
			1428.3wm	Ne	IR	1
			1352.2vw	Ne	IR	1
			1188.1wm	Ne	IR	1
			833.6wm	Ne	IR	1
b ₂			1471.7vs	Ne	IR	1
			1259.4w	Ne	IR	1
			1138.4vw	Ne	IR	1
			941.9m	Ne	IR	1
			692.9wm	Ne	IR	1

Reference

¹H. H. Wenk, A. Balster, W. Sander, D. A. Hrovat, and W. T. Borden, Angew. Chem. Int. Ed. **40**, 2295 (2001).

C₆H₅F⁺

$\tilde{H}, \tilde{I}^2B_2, ^2B_1$		C _{2v}				
T ^a =57900(1000) gas PE ²						
\tilde{G}^2A_1		C _{2v}				
T ^a =48250(1000) gas PE ²						
\tilde{F}^2B_2		C _{2v}				
T ^a =43400(1000) gas PE ²						
\tilde{E}^2B_2		C _{2v}				
T ₀ =37680(160) gas PE ²						
\tilde{D}^2A_1		C _{2v}				
T ^a =30500(1000) gas PE ²						
\tilde{C}^2B_1		C _{2v}				
T ^a =24800(1000) gas PE ²						
		Ar		AB ^{1,3}		
				C \tilde{C} - \tilde{X} 410–435 nm		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁			510	Ar	AB	3
\tilde{B}^2B_1		C _{2v}				
T ₀ =21075 gas PE ² PF ⁴ PRI ⁸						
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	9a		1152	gas	PF,PRI	4,8
	12		695	gas	PRI	8
	6a		458	gas	PF,PRI	4,8
a ₂	16a		236	gas	PRI	8
b ₁			366	gas	PRI	8
	11		207	gas	PRI	8
b ₂	6b		520	gas	PRI	8
	18b		430	gas	PRI	8

\tilde{A}^2A_2 C_{2v}
T^a=4680(1000) gas PE²

\tilde{X}^2B_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	8a		1619	gas	MPI,TPE	5,6
			1370(40)	gas	PE	2
			1164(2)	gas	MPI,TPE,TPI	5–7
	1		983	gas	MPI,TPE	5,6
	18a		959	gas	MPI,TPE,TPI	5–7
	12		795(2)	gas	MPI,TPE,TPI	4–7
	6a		500(2)	gas	TPE,TPI	2,4,6,7
a ₂	16a		341	gas	TPI	8
b ₁	11		181(2)	gas	TPI	7
	6b		505(2)	gas	MPI,TPE,TPI	5–7
b ₂	18b		400(2)	gas	MPI,TPE,TPI	5–7

^aFrom vertical ionization potential.

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C₆H₅Cl⁺

\tilde{K}^2A_1 C_{2v}
T₀=63450(25) gas PE¹⁶

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁		CH stretch	2905(10)	gas	PE	16
	(1)		970(10)	gas	PE	16
	(12)		565T	gas	PE	16

\tilde{J}^2A_1 C_{2v}
T^a=55950(800) gas PE¹⁶

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁		CH stretch	2820(10)	gas	PE	16

\tilde{I}^2B_2 C_{2v}
T^a=50700(100) gas PE¹⁶

\tilde{H}^2B_2 C_{2v}
T^a=45460(800) gas PE¹⁶

\tilde{G}^2A_1 C_{2v}
T^a=42150(100) gas PE¹⁶

\tilde{F}^2B_1 C_{2v}
 $T^a=33520(100)$ gas PE^{3,16}
 $T_0=33410(10)$ Ar AB⁶ $\tilde{F}-\tilde{X}$ 287–300 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1			710(10)	Ar	AB	6

\tilde{E}^2B_2 C_{2v}
 $T^a=30930(800)$ gas PE^{3,16}

\tilde{D}^2A_1 C_{2v}
 $T^a=25850(100)$ gas PE^{1,3,16}

\tilde{C}^2B_1 C_{2v}
 $T^0=21240(25)$ gas PE^{1,3,16}
 20750(10) Ar AB^{4,6} $\tilde{C}-\tilde{X}$ 434–482 nm
 Photodissociation into C₆H₅⁺ + Cl occurs.^{2,7,8,10}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1			1370(10)	Ar	AB	6
	(1)		936(10)	gas	PE	3,16
			900(10)	Ar	AB	6
	(12)		734(10)	gas	PE	16
			370(10)	Ar	AB	6

\tilde{B}^2B_1 C_{2v}
 $T^0=18270(25)$ gas PE^{1,3,16}PF^{7,8,10}MPI¹⁵
 Ar AB⁶ $\tilde{B}-\tilde{X}$ 485–580 nm
 $\tilde{B}-\tilde{X}$ 498–534 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	5 (19a)		1488	gas	PRI	15
	6 (9a)		1263	gas	PRI	15
	7 (7a)		1131	gas	PRI	15
	8 (1)		1010	gas	PRI	15
			970(10)	gas	PE,PF	3,7,16
			930(10)	Ar	AB	6
	9 (18a)		866	gas	PRI	15
	10 (12)		636	gas	PF,PRI	7,15
	11 (6a)		340(10)	gas	PE,PF,PRI	3,7,15,16
a_2	14 (16a)		223	gas	PRI	15
b_1	15 (5)		730	gas	PRI	15
	19 (16b)		218	gas	PRI	15
b_2	29 (6b)		562	gas	PRI	15
	30 (18b)		329	gas	PRI	15

\tilde{A}^2A_2 C_{2v}
 $T^a=5170(25)$ gas PE^{1,3,16}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	(1)		1000(10)	gas	PE	3,16

\tilde{X}^2B_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	(8a)		1540(10)	gas	PE	16
	5 (19a)		1429(2)	gas	PE,TPE	3,13
	6 (9a)		1200(2)	gas	MPI,TPE,TPI	9,12–14
	7 (7a)		1116(2)	gas	MPI,TPE,TPI	9,12–14,16
	8 (18a)		995(2)	gas	MPI,TPE,TPI	9,12–14
	9 (1)		975(2)	gas	MPI,TPE,TPI	9,12–14
	10 (12)		716(2)	gas	MPI,TPE,TPI	9,12–14
	11 (6a)		422(2)	gas	TPE,PF,MPI	3,5,7
						11–14,16
a_2	14 (16a)		348(2)	gas	MPI,PE	11–14
b_1	15 (5)		909(20)	gas	TPE	12
	16		812(20)	gas	TPE	12
	17 (10b)		603(20)	gas	TPE	12
	18 (4)		558(20)	gas	TPE	12
	19 (16b)		394(2)	gas	TPE,TPI	12–14
	20		154(10)	gas	MPI,PE	11,12
b_2	29 (6b)		531(2)	gas	MPI,PE	9,11–14
	30 (18b)		311(2)	gas	MPI,PE	11–13

C₆D₅Cl⁺

\tilde{X}^2B_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	11		413(20)	gas	TPE	12

^aFrom vertical ionization potential.

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C₆H₅Br⁺

\tilde{K}^2A_1 C_{2v}
T₀=63170(100) gas PE²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1		920(10)	gas	PE	2

\tilde{J}^2A_1 C_{2v}
T^a=53300T gas PE²

\tilde{I}^2B_2 C_{2v}
T^a=49530(100) gas PE²

\tilde{H}^2B_2 C_{2v}
T^a=44690(100) gas PE²

\tilde{G}^2A_1 C_{2v}
T^a=40980(100) gas PE²

\tilde{F}^2B_1 C_{2v}
T^a=31540T gas PE²

\tilde{E}^2B_2 C_{2v}
T^a=29760(100) gas PE²

\tilde{D}^2A_1 C_{2v}
T^a=24040(100) gas PE²

\tilde{C}^2B_1 C_{2v}
T₀=17730(25) gas PE²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	12		605(10)	gas	PE	2
	6a		234(10)	gas	PE	2

\tilde{B}^2B_2 C_{2v}
T₀=13250(25) gas PE²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1		968(10)	gas	PE	2
	12		621(10)	gas	PE	2

\tilde{A}^2A_2 C_{2v}
T₀=5420(25) gas PE²

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	12		775(10)	gas	PE	2
	6a		258(10)	gas	PE	2

\tilde{X}^2B_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	8a		1530(30)	gas	MPI,PE	1
	1		1017(10)	gas	MPI,PE	1,2
	18a		980(30)	gas	MPI,PE	1
	6a		331(10)	gas	MPI,PE	1,2
b ₂	9b		1080(30)	gas	MPI,PE	1
	6b		540(30)	gas	MPI,PE	1

^aFrom vertical ionization potential.

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C₆H₅I⁺

\tilde{K}^2A_1 C_{2v}
T^a=64110(100) gas PE¹

\tilde{J}^2B_2 C_{2v}
T^a=53620(800) gas PE¹

\tilde{I}^2A_1 C_{2v}
T^a=50640(100) gas PE¹

\tilde{H}^2B_2 C_{2v}
T^a=45470(100) gas PE¹

\tilde{G}^2A_1 C_{2v}
T^a=39340(100) gas PE¹

\tilde{F}^2B_1 C_{2v}
T^a=31840(800) gas PE¹

\tilde{E}^2B_2 C_{2v}
T^a=29340(100) gas PE¹

\tilde{D}^2A_1 C_{2v}
T^a=23280(100) gas PE¹

\tilde{C}^2B_1 C_{2v}
T₀=14420(25) gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	12		637(10)	gas	PE	1
	6a		202(10)	gas	PE	1

\tilde{B}^2B_2 C_{2v}
T₀=8200(25) gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	1		968(10)	gas	PE	1
	12		621(10)	gas	PE	1

\tilde{A}^2A_2 C_{2v}
T₀=6060(25) gas PE¹

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	12		686(10)	gas	PE	1
	6a		210(10)	gas	PE	1

\tilde{X}^2B_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	7a		1017(10)	gas	PE	1
	6a		282(10)	gas	PE	1

^aFrom vertical ionization potential.

Reference

¹D. M. P. Holland, D. Edvardsson, L. Karlsson, R. Maripuu, K. Siegbahn, A. W. Potts, and W. von Niessen, *Chem. Phys.* **253**, 133 (2000).

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1,2,3,4-C₆H₂F₄⁺

\tilde{B}^2B_1 C_{2v}
 $T_0 = 23291.7$ gas EF¹EM²LF^{3,7} $\tilde{B}-\tilde{X}$ 415–465 nm
 23192 Ne LF⁴ $\tilde{B}-\tilde{X}$ 387–480 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2		1539	Ne	LF	4
	4		1355	Ne	LF	4
	6		1172	Ne	LF	4
	7		1060	Ne	LF	4
	8		654	gas	LF	3
			668	Ne	LF	4
	9		433	gas	EM,LF	2,3,10
			442	Ne	LF	4
	10		324	gas	LF	7,10
			324	Ne	LF	4
	11		271	gas	EM,LF	2,3,10
		277	Ne	LF	4	

$\tau_0 = 50(3)$ ns gas EF¹PIFCO⁵PEFCO⁶
 44(2) ns Ne LF⁹

\tilde{A}^2A_2 C_{2v}
 $T_0 =$ gas PE¹ $\tilde{B}-\tilde{A}$ 520–600 nm
 2767T Ne LF⁸

\tilde{X}^2B_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2		1665	gas	EM,EF	2,7
			1665	Ne	LF	4
			1228	gas	EM	2
	6		1198	Ne	LF	4
	7		1083	gas	EF	7
			1082	Ne	LF	4
	8		678	gas	EF	7
			680	Ne	LF	4
			593	gas	EM	2
	9		444	gas	EM,LF,EF	2,3,7,10
			443	Ne	LF	4
	10		343	gas	EF,LF	7,10
			340	Ne	LF	4
	11		271	gas	EM,LF,EF	2,3,7,10
			273	Ne	LF	4

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1,2,3,5-C₆H₂F₄⁺

\tilde{B}^2B_1 C_{2v}
 $T_0 = 23323.7$ gas EF¹EM²LF^{3,8} $\tilde{B}-\tilde{X}$ 400–470 nm
 23232 Ne LF⁵ $\tilde{B}-\tilde{X}$ 400–502 nm
 22903 Ar LF⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2		1580	Ne	LF	5
			1582	Ar	LF	4
	3		1445T	Ne	LF	5
			1536T	Ar	LF	4
	5		1264	gas	LF	3
			1259	Ne	LF	5
			1263	Ar	LF	4
	6		1140	Ne	LF	5
	8		783	Ne	LF	5
			783	Ar	LF	4
	9		569	gas	LF	3,11
			573	Ne	LF	5
	10		576	Ar	LF	4
			424	gas	EM,LF	2,3,11
			428	Ne	LF	5
			431	Ar	LF	4
	11		301	gas	EM,LF	2,3,11
			305	Ne	LF	5
			307	Ar	LF	4

$\tau_0 = 50(3)$ ns gas EF¹PIFCO⁶PEFCO⁷
 43(2) ns Ne LF¹⁰

\tilde{A}^2A_2 C_{2v}
 $T_0 =$ gas EF¹ $\tilde{B}-\tilde{A}$ 520–600 nm
 2442 Ne LF⁹

\tilde{X}^2B_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	2		1650	gas	EM,EF,LF	2,8,11
			1647	Ne	LF	5
	3		1458	gas	EF	8
			1449	Ne	LF	5
	5		1318	gas	EM,EF	2,8
			1305	Ne	LF	5
	8		785	Ne	LF	5
	9		584	gas	EF,LF	8,11
			581	Ne	LF	5
			586	Ar	LF	4
	10		424	gas	EM,EF,LF	2,8,11
			426	Ne	LF	5
			429	Ar	LF	4
	11		305	gas	EM,EF,LF	2,8,11
			303	Ne	LF	5
			307	Ar	LF	4

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1,2,4,5-C₆H₂F₄⁺

\tilde{B}^2B_{1u} D_{2h}
 $T_0 = 24443.5$ gas EF¹EM²LF^{3,8,12}PI¹¹ $\tilde{B}-\tilde{X}$ 385–500 nm
 24352 Ne LF⁵ $\tilde{B}-\tilde{X}$ 385–510 nm
 24072 Ar LF⁴

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a _g	2		1536	Ne	LF	5
			1536	Ar	LF	4
	3		1392	Ar	LF	4
			725	Ne	LF	5
	4		722	Ar	LF	4
			460	gas	LF	3,12
			465	Ne	LF	5
	5		470	Ar	LF	4
			274	gas	LF	3,12
			276	Ne	LF	5
	6		279	Ar	LF	4

$\tau_0 = 32(2)$ ns gas EF¹PIFCO⁶PEFCO⁷
 30(2) ns Ne LF¹⁰

\tilde{A}^2B_{3g} D_{2h}
 $T_0 = \text{gas}$ EF¹ $\tilde{B}-\tilde{A}$ 550–650 nm
 6115 Ne LF⁹ $\tilde{B}-\tilde{A}$ 545–625 nm

\tilde{X}^2B_{2g} D_{2h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a _g	2		1558	gas	EF,LF	8,12
			1558	Ne	LF	5
	3		1477	gas	EF	8
			1476	Ne	LF	5
	4		726	Ne	LF	5
			482	gas	EM,EF,PI,LF	2,8,11,12
	5		485	Ne	LF	5
			480	Ar	LF	4
			287	gas	EF,LF	8,12
	6		287	Ne	LF	5
			289	Ar	LF	4
	b _{1u}		183H	gas	PI	11

1,2,4,5-C₆D₂F₄⁺

\tilde{B}^2B_{1u} D_{2h}
 $T_0 = 24400$ Ne LF⁵ $\tilde{B}-\tilde{X}$ 370–465 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a _g	2		1531	Ne	LF	5
	3		1383T	Ne	LF	5
	4		722	Ne	LF	5
	5		466	Ne	LF	5
	6		276	Ne	LF	5

$\tau_0 = 38(3)$ ns gas PEFCO⁷

\tilde{X} D_{2h}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a _g	2		1542	Ne	LF	5
	3		1472	Ne	LF	5
	4		706	Ne	LF	5
	5		480	Ne	LF	5
	6		286	Ne	LF	5

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C₆H₅CHCl

$T_0 = 22044.5$ gas LF¹MPI¹ 431–467 nm

\tilde{X}^2A'' C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	6b		612.0	gas	MPI	1
	6a		358.4	gas	MPI	1

Reference

- J. Yao and E. R. Bernstein, *J. Chem. Phys.* **107**, 3352 (1997).

C₆H₅CHBr $T_0 = 21952.6$ gas LF¹

444–456 nm

Reference¹J. Yao and E. R. Bernstein, *J. Chem. Phys.* **107**, 3352 (1997).**C₆H₅CCl₂**gas LF¹MPI¹

439–466 nm

 \tilde{X} C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	6a		406.3	gas	MPI	1
b ₂	6b		639.3	gas	MPI	1

Reference¹J. Yao and E. R. Bernstein, *J. Chem. Phys.* **107**, 3352 (1997).**(2-FC₆H₄)CH₂** \tilde{A}
 $T_0 = 21924$ gas EM^{1,4}LF² $\tilde{A}-\tilde{X}$ 435–524 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1022	gas	LF	2
			704	gas	LF	2
			501	gas	LF	2
			386	gas	LF	2

 $\tau_0 = 1220(40)$ ns gas LF² \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	7 (8a)	C-CH ₂ stretch	1570	gas	PE,EM	3,4
	13 (14)		1285	gas	EM	4
	14 (7a)		1264	gas	PE,EM	3,4
	15 (13)		1208	gas	EM	4
	17 (18a)		1119	gas	EM	4
	18 (18b)		1033	gas	EM	4
	20		882	gas	LF	2
	21 (1)	Ring breathing	757	gas	PE,EM	3,4
	22 (6a)	Ring deform.	575	gas	PE,EM	3,4
	23 (6b)	Ring deform.	516	gas	EM	4
	24 (9b)	CH ₂ rock	436	gas	PE,EM	3,4
	25 (15)		304	gas	EM	4
a''	(5)		969	gas	EM	4
	(16a)		535	gas	EM	4
	(16b)		424	gas	EM	4
	(10a)		266	gas	EM	4

References¹T. F. Bindley, A. T. Watts, and S. Walker, *Trans. Faraday Soc.* **60**, 1 (1964).²T. R. Charlton and B. A. Thrush, *Chem. Phys. Lett.* **125**, 547 (1986).³J. B. Kim, P. G. Wenthold, and W. C. Lineberger, *J. Phys. Chem. A* **103**, 10833 (1999).⁴S. K. Lee and S. K. Lee, *J. Phys. Chem. A* **105**, 3034 (2001).**(3-FC₆H₄)CH₂** \tilde{A} $T_0 = 21691$ gas EM^{1,4}LF² $\tilde{A}-\tilde{X}$ 445–498 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			734	gas	LF	2
			466	gas	LF	2

 $\tau_0 = 597(15)$ ns gas LF² \tilde{X} C_s

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	7 (19b)	C-CH ₂ stretch	1599	gas	PE,EM	3,4
	9 (8b)		1567	gas	EM	4
	10 (8a)		1557	gas	EM	4
	11 (19a)		1428	gas	EM	4
	14 (14)		1288	gas	EM	4
	16 (9b)		1142	gas	EM	4
	19 (12)		998	gas	EM	4
	20 (7b)		930	gas	EM	4
	21 (1)		741	gas	LF,PE,EM	2–4
	22 (6a)		530	gas	EM	4
	23 (6b)	Ring deform.	506	gas	EM,LF,PE	1–4
	24 (15)		453	gas	EM	4

References¹T. F. Bindley, A. T. Watts, and S. Walker, *Trans. Faraday Soc.* **60**, 1 (1964).²T. R. Charlton and B. A. Thrush, *Chem. Phys. Lett.* **125**, 547 (1986).³J. B. Kim, P. G. Wenthold, and W. C. Lineberger, *J. Phys. Chem. A* **103**, 10833 (1999).⁴S. K. Lee and B. U. Ahn, *Chem. Phys. Lett.* **321**, 25 (2000).**(4-FC₆H₄)CH₂** \tilde{A}^2A_2 C_{2v} $T_0 = 21524.0(2)$ gas EM^{1,3,5,6}LF² $\tilde{A}-\tilde{X}$ 429–475 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1760	gas	LF	2
	(7a)		1228	gas	LF	4
	(13)		1194	gas	LF	2,4
	(18a)		846	gas	LF	4
	(1)		813	gas	LF	2,4
	(6b)		524	gas	LF	4
	(6a)		411	gas	LF	4
	(9b)		398	gas	LF	4
	(15)		222	gas	LF	4

 $\tau_0 = 532(5)$ ns gas LF^{2,4}

\tilde{X}^2B_1		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	4 (8a)	C-CH ₂ stretch	1568	gas	EM,PE	6,7
	(7a)		1262	gas	EM	6
	(13)		1243	gas	EM	6
	11 (1)	Ring deform.	835	gas	EM,LF	3,4,6
	12 (12)		754	gas	EM	6
	13 (6a)		453	gas	EM,LF,PE	1-4,6,7
b_2	(6b)	Ring deform.	632T	gas	EM	6
	(9b)		420T	gas	EM	6
	(15)		323T	gas	EM	6

References

- ¹T. F. Bindley, A. T. Watts, and S. Walker, *Trans. Faraday Soc.* **60**, 1 (1964).
²T. R. Charlton and B. A. Thrush, *Chem. Phys. Lett.* **125**, 547 (1986).
³C. Cossart-Magos and D. Cossart, *Mol. Phys.* **65**, 627 (1988).
⁴M. Fukushima and K. Obi, *J. Chem. Phys.* **93**, 8488 (1990).
⁵D. Y. Baek and S. K. Lee, *Bull. Korean Chem. Soc.* **19**, 1359 (1998).
⁶S. K. Lee and D. Y. Baek, *Chem. Phys. Lett.* **301**, 407 (1999).
⁷J. B. Kim, P. G. Wenthold, and W. C. Lineberger, *J. Phys. Chem. A* **103**, 10833 (1999).

(2-CIC₆H₄)CH₂

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	7	C-CH ₂ stretch	1540(40)	gas	PE	1
	13	Deformation	1250(40)	gas	PE	1
	22	Ring deform.	560(20)	gas	PE	1

Reference

- ¹J. B. Kim, P. G. Wenthold, and W. C. Lineberger, *J. Phys. Chem. A* **103**, 10833 (1999).

(3-CIC₆H₄)CH₂

\tilde{A}		C_s				
$T_0=21194.8$		gas	EM ²	$\tilde{A}-\tilde{X}$ 471-503 nm		
\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	7	C-CH ₂ stretch	1520(40)	gas	PE	1
	13 (3)		1287	gas	EM	2
	14 (13)		1237	gas	EM	2
	18 (12)	Ring deform.	984	gas	EM	2
	19 (5)		959	gas	EM	2
	21 (1)		686	gas	PE,EM	1,2
	22 (6a)		525	gas	PE,EM	1,2
	23 (7b)		413	gas	EM	2
	24 (15)		394	gas	EM	2

References

- ¹J. B. Kim, P. G. Wenthold, and W. C. Lineberger, *J. Phys. Chem. A* **103**, 10833 (1999).
²S. K. Lee and S. Y. Chae, *J. Phys. Chem. A* **105**, 5808 (2001).

(4-CIC₆H₄)CH₂

\tilde{B}^2B_1		C_{2v}							
$T_0=21733.8(5)$		gas	LF ¹ EM ³	$\tilde{A}, \tilde{B}-\tilde{X}$ 436-515 nm					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.			
a_1	(12)	C-CH ₂ stretch	1194	gas	LF	1			
			1076	gas	LF	1			
			960	gas	LF	1			
			779	gas	LF	1			
			340	gas	LF	1			
			242	gas	LF	1			
			b_2	(6b)	Ring deform.	623	gas	LF	1
						122	gas	LF	1

\tilde{A}^2A_2		C_{2v}				
$T_0=21638.7(5)$		gas	LF ¹ EM ³	$\tilde{A}, \tilde{B}-\tilde{X}$ 436-515 nm		
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	(12)	C-CH ₂ stretch	726	gas	LF	1

\tilde{X}^2B_1		C_{2v}					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a_1	4 (8a)	C-CH ₂ stretch	1544	gas	LF,EM,PE	2-4	
			(1)	1079	gas	LF,EM	2,3
			(6a)	811	gas	LF,EM	2,3
			(12)	638	gas	LF,EM	1,3
				559	gas	LF	2
				380	gas	EM,PE	3,4
b_2	13 (7a)	Ring deform.	8b	1529	gas	EM	3
			3	1306	gas	EM	3
			15	257	gas	EM	3
				209	gas	LF	1

References

- ¹M. Fukushima and K. Obi, *Chem. Phys. Lett.* **248**, 269 (1996).
²M. Fukushima, K. Saito, and K. Obi, *J. Mol. Spectrosc.* **180**, 389 (1996).
³S. K. Lee and D. Y. Baek, *Chem. Phys. Lett.* **304**, 39 (1999).
⁴J. B. Kim, P. G. Wenthold, and W. C. Lineberger, *J. Phys. Chem. A* **103**, 10833 (1999).

(2-BrC₆H₄)CH₂

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	7	C-CH ₂ stretch	1540(40)	gas	PE	1
	13	Deformation	1230(40)	gas	PE	1
	22	Ring deform.	550(20)	gas	PE	1

Reference

¹J. B. Kim, P. G. Wenthold, and W. C. Lineberger, *J. Phys. Chem. A* **103**, 10833 (1999).

(3-BrC₆H₄)CH₂

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	7	C-CH ₂ stretch	1520(40)	gas	PE	1
	22	Ring deform.	525(20)	gas	PE	1

Reference

¹J. B. Kim, P. G. Wenthold, and W. C. Lineberger, *J. Phys. Chem. A* **103**, 10833 (1999).

(4-BrC₆H₄)CH₂

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	4	C-CH ₂ stretch	1510(40)	gas	PE	1
	13	Ring deform.	290(20)	gas	PE	1

Reference

¹J. B. Kim, P. G. Wenthold, and W. C. Lineberger, *J. Phys. Chem. A* **103**, 10833 (1999).

(2,6-F₂C₆H₃)CH₂

\tilde{A}^2A_2 C_{2v}
 $T_0=21774$ gas EM¹ $\tilde{A}-\tilde{X}$ 459–488 nm

\tilde{X}^2B_2		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1		696	gas	EM	1
	6a		470	gas	EM	1
	9a		332	gas	EM	1
b_2	14		1268	gas	EM	1

Reference

¹S. K. Lee and D. Y. Baek, *J. Phys. Chem. A* **104**, 5219 (2000).

C₆F₅CH₂

\tilde{A}^2A_2 C_{2v}
 $T_0=21857$ gas EM¹ $\tilde{A}-\tilde{X}$ 457–473 nm

\tilde{X}^2B_2		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1		550	gas	EM	1
	6a		456	gas	EM	1
	18a		334	gas	EM	1
b_1	9a		277	gas	EM	1
	4		727	gas	EM	1
	16b		584	gas	EM	1
	11		219	gas	EM	1

Reference

¹S. K. Lee and D. Y. Baek, *Chem. Phys. Lett.* **311**, 36 (1999).

(2-FC₆H₄)CH₂⁻

Threshold for electron detachment from ground-state (2-FC₆H₄)CH₂⁻ = 8800(65) gas PE¹

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	24	CH ₂ rock	425(40)	gas	PE	1

Reference

¹J. B. Kim, P. G. Wenthold, and W. C. Lineberger, *J. Phys. Chem. A* **103**, 10833 (1999).

(3-FC₆H₄)CH₂⁻

Threshold for electron detachment from ground-state (3-FC₆H₄)CH₂⁻ = 9465(65) gas PE¹

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	22	Ring deform.	510(20)	gas	PE	1

Reference

¹J. B. Kim, P. G. Wenthold, and W. C. Lineberger, *J. Phys. Chem. A* **103**, 10833 (1999).

(4-FC₆H₄)CH₂⁻

Threshold for electron detachment from ground-state (4-FC₆H₄)CH₂⁻
= 7560(65) gas PE¹

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	13	Ring deform.	420(20)	gas	PE	1

Reference

¹J. B. Kim, P. G. Wenthold, and W. C. Lineberger, J. Phys. Chem. A **103**, 10833 (1999).

(2-ClC₆H₄)CH₂⁻

Threshold for electron detachment from ground-state (2-ClC₆H₄)CH₂⁻
= 10140(65) gas PE¹

Reference

¹J. B. Kim, P. G. Wenthold, and W. C. Lineberger, J. Phys. Chem. A **103**, 10833 (1999).

(3-ClC₆H₄)CH₂⁻

Threshold for electron detachment from ground-state (3-ClC₆H₄)CH₂⁻
= 10260(65) gas PE¹

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	22	Ring deform.	490(20)	gas	PE	1

Reference

¹J. B. Kim, P. G. Wenthold, and W. C. Lineberger, J. Phys. Chem. A **103**, 10833 (1999).

(4-ClC₆H₄)CH₂⁻

Threshold for electron detachment from ground-state (4-ClC₆H₄)CH₂⁻
= 9470(65) gas PE¹

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	13	Ring deform.	365(20)	gas	PE	1

Reference

¹J. B. Kim, P. G. Wenthold, and W. C. Lineberger, J. Phys. Chem. A **103**, 10833 (1999).

(2-BrC₆H₄)CH₂⁻

Threshold for electron detachment from ground-state (2-BrC₆H₄)CH₂⁻
= 10550(65) gas PE¹

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	22	Ring deform.	530(30)	gas	PE	1

Reference

¹J. B. Kim, P. G. Wenthold, and W. C. Lineberger, J. Phys. Chem. A **103**, 10833 (1999).

(3-BrC₆H₄)CH₂⁻

Threshold for electron detachment from ground-state (3-BrC₆H₄)CH₂⁻
= 10545(65) gas PE¹

\tilde{X}		C _s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	22	Ring deform.	500(20)	gas	PE	1

Reference

¹J. B. Kim, P. G. Wenthold, and W. C. Lineberger, J. Phys. Chem. A **103**, 10833 (1999).

(4-BrC₆H₄)⁻

Threshold for electron detachment from ground-state (4-BrC₆H₄)CH₂⁻
= 9915(65) gas PE¹

\tilde{X}		C _{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a ₁	13	Ring deform.	265(40)	gas	PE	1

Reference

¹J. B. Kim, P. G. Wenthold, and W. C. Lineberger, J. Phys. Chem. A **103**, 10833 (1999).

(4-NC-C₆H₄)CH₂

\tilde{A}^2B_1 C_{2v}
T₀ = 20743.9 gas LF¹EM² $\tilde{A}-\tilde{X}$ 445–580 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
b ₂	6b		516	gas	LF	1

$\tau_0 = 400\text{T ns}$ gas LF¹
A₀ = 0.172; B₀ = 0.032; C₀ = 0.028 LF¹

\tilde{X}^2B_1		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
a_1	20a	CH stretch	3133	gas	EM	2
		CN stretch	2225	gas	LF,EM	1,2
	8a		1563	gas	LF,EM	1,2
	7a		1210	gas	EM	2
	9a		1149	gas	LF,EM	1,2
	1		825	gas	LF,EM	1,2
				513	gas	LF
	6a		407	gas	EM	2
a_2	17a		951	gas	EM	2
b_1	16b		427	gas	EM	2
b_2	8b		1521	gas	EM	2
			642	gas	LF,EM	1,2
			355	gas	EM	2

$A_0=0.182$; $B_0=0.032$; $C_0=0.028$ LF¹

References

- ¹M. Fukushima, K. Saito, and K. Obi, *J. Mol. Spectrosc.* **180**, 389 (1996).
²S. K. Lee and B. U. Ahn, *Chem. Phys. Lett.* **320**, 601 (2000).

$p-(CF_2)_2C_6H_4$

In an argon matrix, $p-(CF_2)_2C_6H_4$ has prominent absorption maxima at 36100 (277 nm), 37600 (266 nm), and 50500 (198 nm).¹ When irradiated at 248 nm, rearrangement to 4,5-F₂C₇H₄=CF₂ occurs.

\tilde{X}		D_{2h}				
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.
b_{1u}			1693vs	Ar	IR	1
			1423wm	Ar	IR	1
			1080ms	Ar	IR	1
			956w	Ar	IR	1
			678wm	Ar	IR	1
b_{2u}			1590vw	Ar	IR	1
			1341m	Ar	IR	1
			1267ms	Ar	IR	1
			1118w	Ar	IR	1
b_{3u}			800w	Ar	IR	1

Reference

- ¹H. H. Wenk, W. Sander, A. Leonov, and A. de Meijere, *Eur. J. Org. Chem.* 3287 (1999).

4,5-F₂C₇H₄=CF₂

\tilde{X}		C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.		
a_1			1698s	Ar	IR	1		
			1608vs	Ar	IR	1		
			1217s	Ar	IR	1		
			1046s	Ar	IR	1		
			870wm	Ar	IR	1		
			693m	Ar	IR	1		
			777wm	Ar	IR	1		
		b_1		1442m	Ar	IR	1	
		b_2			1345wm	Ar	IR	1
					1293s	Ar	IR	1
	1070wm			Ar	IR	1		
		719m	Ar	IR	1			

Reference

- ¹H. H. Wenk, W. Sander, A. Leonov, and A. de Meijere, *Eur. J. Org. Chem.* 3287 (1999).

C₆H₅NH₂⁺

\tilde{A}^2A_2		C_{2v}						
$T_0=10570(80)$		gas	PE ⁷					
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.		
a_1	1		840(40)	gas	PE	7		
	6a		524(10)	gas	PE	7		
\tilde{X}^2B_1		C_{2v}						
Vib. sym.	No.	Approximate type of mode	cm^{-1}	Med.	Type meas.	Refs.		
a_1			1635	Ar	IR	8		
		8a ^a		1594	gas	TPE	2,5,6	
				1515	Ar	IR	8	
				1483	Ar	IR	8	
		19a		1436(10)T	gas	PE	2	
			13		1385	gas	TPE	2,5,6
		9a		1188	gas	TPE,PI	2,4-6,9	
		18a		996	gas	TPE,TPI	5,6,9	
		12	Ring s-stretch	982	gas	PI,TPE	1,2,4-6,9	
		1		814	gas	PI,TPE	1,2,4,6,9	
		6a	N-ring s-str.	522	gas	PI,TPE	1-6,9	
		a_2	16a	C-N torsion	577H	gas	TPE	6
					356	gas	PI,TPE	2-6
		b_1		17b	963T ^b	gas	TPE	6
					790 ^b	gas	TPE	6
11				785	Ar	IR	8	
	Inversion			658	gas	PI,TPE	2,3,5,6,9	
	656(2)			Ar	IR	8		
16b				629	gas	TPE	6	
4	Ring deform.			445	gas	TPE,TPI	6,9	
10b				179	gas	PI,TPE	3,5,6,9	
b_2	6b				582	gas	TPE	6
					550H	gas	TPE	5,6

C₆D₅ND₂⁺

\tilde{X}^2B_1		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	8a		1563	gas	TPI	9
	12		948	gas	TPI	9
	18a		847	gas	TPI	9
	1		738	gas	TPI	9
	6a		513	gas	TPI	9
<i>a</i> ₂	16a		309	gas	TPI	9
<i>b</i> ₁	5		831	gas	TPI	9
		Inversion	465(10)	gas	PI	3
	10b		157	gas	TPI	9
<i>b</i> ₂	14		1341	gas	TPI	9

^aVibrational numbering and assignments parallel those given by Chernoff and Rice [J. Chem. Phys. **70**, 2511 (1979)] for aniline.

^bFrom combination band.

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***o*-C₆H₄FNH₂⁺**

\tilde{X}		C_s					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
<i>a'</i>	3	CH deform.	1243(10)	gas	TPI	1	
	7a	CF stretch	1209(10)	gas	TPI	1	
	9a	CH deform.	1145(25)	gas	TPI	1	
	18b	CH deform.	1093(20)	gas	TPI	1	
	18a	CH deform.	988(10)	gas	TPI	1	
	5	CH deform.	962(10)	gas	TPI	1	
	12	C ₃ deform.	855(10)	gas	TPI	1	
	1	Ring breathing	754(10)	gas	TPI	1	
	6a	C ₃ deform.	515(10)	gas	TPI	1	
	9b	CF deform.	421(10)	gas	TPI	1	
	15	CN deform.	308(10)	gas	TPI	1	
	<i>a''</i>	17a	CH deform.	920(10)	gas	TPI	1
			NH ₂ wag	716(10)	gas	TPI	1
		11	CF deform.	682(10)	gas	TPI	1
		10a	CH deform.	562(10)	gas	TPI	1
4		C ₃ deform.	466(10)	gas	TPI	1	
16a		C ₃ deform.	256(10)	gas	TPI	1	
	10b	CN deform.	154(10)	gas	TPI	1	

Reference

- J. L. Lin and W. B. Tzeng, Phys. Chem. Chem. Phys. **2**, 3759 (2000).

***p*-C₆H₄FNH₂⁺**

\tilde{X}		C_{2v}					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
<i>a</i> ₁		NH ₂ scissors	1730(10)	gas	TPI	1	
	8a	Ring stretch	1615(10)	gas	TPI	1	
	13	CN stretch	1471(10)	gas	TPI	1	
	7a	CF stretch	1352(10)	gas	TPI	1,2	
	18a	CH deform.	968(10)	gas	TPI	1,2	
	12	C ₃ deform.	836(10)	gas	TPI	1,2	
	1	Ring breathing	763(10)	gas	TPI	1,2	
	6a	C ₃ deform.	452(10)	gas	TPI	1,2	
	<i>a</i> ₂	17b	CH deform.	867(10)	gas	TPI	1
	<i>b</i> ₁	5	CH deform.	939(10)	gas	TPI	1
			NH ₂ wag	737(10)	gas	TPI	1,2
		16b	C ₃ deform.	502(10)	gas	TPI	1
	<i>b</i> ₂	19b	Ring stretch	1512	gas	TPI	1,2
		3	CH deform.	1410(10)	gas	TPI	1
		18b	CH deform.	1165(10)	gas	TPI	1,2
6b		C ₃ deform.	634(10)	gas	TPI	1,2	
15		CN deform.	371(10)	gas	TPI	1,2	

References

- W. B. Tzeng and J. L. Lin, J. Phys. Chem. A **103**, 8612 (1999).
- J. L. Lin and W. B. Tzeng, J. Chem. Phys. **115**, 743 (2001).

***p*-C₆H₄CINH₂⁺**

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
<i>a</i> ₁	9a	CH deform.	1195(10)	gas	TPI	1
	1	C ₃ deform.	1136(10)	gas	TPI	1
	18a	CH deform.	1005(10)	gas	TPI	1
	6a	C ₃ deform.	821(10)	gas	TPI	1
	12	C ₃ deform.	664(10)	gas	TPI	1
	7a	Mixed	380(10)	gas	TPI	1
	<i>b</i> ₁		NH deform.	1041(10)	gas	TPI
5		CH deform.	976(10)	gas	TPI	1
17b		CH deform.	842(10)	gas	TPI	1
<i>b</i> ₂		NH ₂ wag	724(10)	gas	TPI	1
	16b	C ₃ deform.	494(10)	gas	TPI	1
	14	CC stretch	1269(10)	gas	TPI	1
	6b	C ₃ deform.	621(10)	gas	TPI	1

Reference

- J. L. Lin and W. B. Tzeng, J. Chem. Phys. **113**, 4109 (2000).

C₆H₅O

\tilde{E}^2B_1 C_{2v}
 In the gas phase, a prominent absorption maximum near 43100 (232 nm) has been attributed^{5,8,9} to C₆H₅O.
 T₀=41800 Ar AB¹¹

\tilde{D}^2A_1 C_{2v}

In the gas phase, a broad band between about 33300 (300 nm) and 37000 (270 nm) has been attributed^{5,8,9} to C_6H_5O .

$T_0 = 33900$ Ar AB¹¹

 \tilde{C}^2B_1 C_{2v}

gas AB^{1,2}

$T_0 = 25175(10)$ Ar AB^{4,6,10,11}

$\tilde{C}-\tilde{X}$ 380–395 nm

$\tilde{C}-\tilde{X}$ 351–397 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1462(20)	Ar	AB	4
			1140(20)	Ar	AB	4
			920(20)	Ar	AB	4

 \tilde{B}^2A_1 C_{2v}

$T_0 \cong 16360$ gas AB^{2,3}

15930(10) Ar AB^{4,6,10,11}

$\tilde{B}-\tilde{X}$ 559–612 nm

$\tilde{B}-\tilde{X}$ 573–629 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			500T	gas	AB	3
			504(10)	Ar	AB	4

 \tilde{A}^2B_2 C_{2v}

$T_0 = 8550(40)$ gas PE⁷

8900T Ar AB¹¹

 \tilde{X}^2B_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1 (20a)	CH stretch	3090vw	Ar	IR	10
	2 (13)	CH stretch	3065w	Ar	IR	10
	3 (2)	CH stretch	3018vw	Ar	IR	10
	4 (8a)	CC stretch	1550ms	Ar	IR	10
	5 (7a)	CO stretch	1490(25)	gas	PE	7
			1481m	Ar	IR	10
	6 (19a)	Mixed	1397w	Ar	IR	10
	7 (9a)	CH bend	1167w	Ar	IR	10
	8 (12)	Mixed	1038vw	Ar	IR	10
	9 (18a)	CCC bend	977w	Ar	IR	10
	10 (1)	Mixed	813w	Ar	IR	10
	11 (6a)	CCC bend	515(15)	gas	PE	7
			520w	Ar	IR	10
b_1	15 (5)	HCCH torsion	1016vw	Ar	IR	10
	16 (17b)	Mixed	898wm	Ar	IR	10
	17 (4)	Mixed	784s	Ar	IR	10
	18 (11)	Mixed	635vs	Ar	IR	10
	19 (16b)	Mixed	472vw	Ar	IR	10
b_2	21 (20b)	CH stretch	3074wm	Ar	IR	10
	22 (7b)	CH stretch	3054vw	Ar	IR	10
	23 (19b)	Mixed	1515m	Ar	IR	10
	24 (8b)	Mixed	1441vw	Ar	IR	10
	25 (14)	Mixed	1318wm	Ar	IR	10
	26 (3)	Mixed	1266wm	Ar	IR	10
	27 (15)	Mixed	1140w	Ar	IR	10
	28 (9b)	Mixed	1072m	Ar	IR	10
	29 (6b)	CCC bend	616w	Ar	IR	10
	30 (18b)	CO bend	446wm	Ar	IR	10

 C_6D_5O \tilde{C}^2B_1 C_{2v}

$T_0 = 25240(10)$ Ar AB^{4,10}

$\tilde{C}-\tilde{X}$ 349–397 nm

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
			1349(20)	Ar	AB	4
			809(20)	Ar	AB	4

 \tilde{B}^2A_1 C_{2v}

$T_0 = 15973(10)$ Ar AB¹¹

 \tilde{X}^2B_1 C_{2v}

Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	1 (20a)	CD stretch	2331vw	Ar	IR	10
	2 (13)	CD stretch	2298w	Ar	IR	10
	3 (2)	CD stretch	2284vw	Ar	IR	10
	4 (8a)	CC stretch	1509s	Ar	IR	10
	5 (7a)	CO stretch	1467ms	Ar	IR	10
	6 (19a)	Mixed	1244wm	Ar	IR	10
	7 (18a)	CCC bend	944w	Ar	IR	10
	8 (12)	Mixed	902vw	Ar	IR	10
	9 (9a)	CD bend	844vw	Ar	IR	10
	10 (1)	Mixed	724w	Ar	IR	10
	11 (6a)	CCC bend	505w	Ar	IR	10
b_1	15 (5)	DCCD torsion	860w	Ar	IR	10
	16 (17b)	Mixed	775wm	Ar	IR	10
	17 (4)	Mixed	660m	Ar	IR	10
	18 (11)	Mixed	489s	Ar	IR	10
	19 (16b)	Mixed	406w	Ar	IR	10
b_2	21 (20b)	CD stretch	2326vw	Ar	IR	10
	22 (7b)	CD stretch	2281vw	Ar	IR	10
	23 (19b)	Mixed	1449wm	Ar	IR	10
	24 (8b)	Mixed	1334wm	Ar	IR	10
	25 (14)	Mixed	1255m	Ar	IR	10
	26 (3)	Mixed	1032w	Ar	IR	10
	27 (15)	Mixed	836w	Ar	IR	10
	28 (9b)	Mixed	824w	Ar	IR	10
	30 (18b)	CO bend	423w	Ar	IR	10

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1,2-C₆F₄>CO

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1			1904.4m	Ar	IR	1
			1610.5w	Ar	IR	1
			1470.7s	Ar	IR	1
			1288.7w	Ar	IR	1
			1108.4wm	Ar	IR	1
			884.6w	Ar	IR	1
			548.1w	Ar	IR	1
b_1		580.6vw	Ar	IR	1	
b_2		1605.6w	Ar	IR	1	
		1494.4vs	Ar	IR	1	
		1050.2wm	Ar	IR	1	
		818.1wm	Ar	IR	1	
		691.0w	Ar	IR	1	
		623.2w	Ar	IR	1	

Reference

¹H. H. Wenk and W. Sander, Chem. Eur. J. **7**, 1837 (2001).

***o*-C₆H₄(OH)₂⁺**

\tilde{X}		C_s				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a'	18b		987	gas	TPI	1
	1		755	gas	TPI	1
	6a		585	gas	TPI	1
	6b		507	gas	TPI	1
a''		Torsion	504H	gas	TPI	1

Reference

¹M. Gerhards, S. Schumm, C. Unterberg, and K. Kleinermanns, Chem. Phys. Lett. **294**, 65 (1998).

***m*-C₆H₄(OH)₂⁺ (C_s)**

\tilde{X}		C_s					
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.	
a'		OH stretch	3582	gas	PIR	1	
		OH stretch	3569	gas	PIR	1	
	12		974	gas	TPI	1	
	7b		926	gas	TPI	1	
	1		736	gas	TPI	1	
	6a		522	gas	TPI	1	
	6b		489	gas	TPI	1	
	15		441	gas	TPI	1	
	9a		343	gas	TPI	1	
	a''	16a		638	gas	TPI	1
		4		597	gas	TPI	1
			Torsion	577	gas	TPI	1
			Torsion	540	gas	TPI	1
		16b		395H	gas	TPI	1
		10a		198H	gas	TPI	1
10b			187	gas	TPI	1	

Reference

¹M. Gerhards, C. Unterberg, and S. Schumm, J. Chem. Phys. **111**, 7966 (1999).

***m*-C₆H₄(OH)₂⁺ (C_{2v})**

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
a_1	12		971	gas	TPI	1
	1		733	gas	TPI	1
	6a		526	gas	TPI	1
	9a		343	gas	TPI	1
a_2	16a		629	gas	TPI	1
	10a		196H	gas	TPI	1
b_1	4		592	gas	TPI	1
	16b		392H	gas	TPI	1
	10b		185	gas	TPI	1
b_2		OH stretch	3576	gas	PIR	1
	7b		917	gas	TPI	1
	6b		495	gas	TPI	1
	15		427	gas	TPI	1

Reference

¹M. Gerhards, C. Unterberg, and S. Schumm, J. Chem. Phys. **111**, 7966 (1999).

***p*-C₆H₄(OH)₂⁺ (C_{2v})**

\tilde{X}		C_{2v}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	13	CO stretch	1292T	gas	TPI	1
	9a	CH deform.	1174	gas	TPI	1
	18b	Mixed	1055T	gas	TPI	1
	17b	CH deform.	875T	gas	TPI	1
	1	Ring breathing	834	gas	TPI	1
	12	Ring deform.	724T	gas	TPI	1
	6a	Ring deform.	456	gas	TPI	1
	15	CO deform.	365	gas	TPI	1

Reference

¹J. L. Lin, L. C. L. Huang, and W. B. Tzeng, J. Phys. Chem. A **105**, 11455 (2001).

***p*-C₆H₄(OH)₂⁺ (C_{2h})**

\tilde{X}		C_{2h}				
Vib. sym.	No.	Approximate type of mode	cm ⁻¹	Med.	Type meas.	Refs.
	13	CO stretch	1294T	gas	TPI	1
	9a	CH deform.	1175	gas	TPI	1
	17b	CH deform.	882T	gas	TPI	1
	1	Ring breathing	837	gas	TPI	1
	6a	Ring deform.	459	gas	TPI	1
	9b	CO deform.	439	gas	TPI	1

Reference

Formula

Structure/Name

References

¹J. L. Lin, L. C. L. Huang, and W. B. Tzeng, *J. Phys. Chem. A* **105**, 11455 (2001).

9. Molecule Index

In the *References* column, unprefix numbers give the page on which the table appears in the monograph.⁵ Numbers prefixed by "A" give the page number in Supplement A,⁶ and numbers prefixed by "B" give the page number in the present work.

Formula	Structure/Name	References
AgAuCu	CuAgAu	60
AgCu ₂	Cu ₂ Ag	59
AgHO	AgOH	B28
AgNO ⁺	AgNO ⁺	B78
AgNO	AgNO	B93
AgNO ⁻	AgNO ⁻	B102
AgO ₂	AgOO	A178,B108
AgO ₂ ⁻	AgOO ⁻	B118
Ag ₂ O	AgOAg	B55
Ag ₃	Ag ₃	60,A161,B52
Ag ₄	Ag ₄	B200
Ag ₅	Ag ₅	B277
Ag ₈	Ag ₈	B360
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AlClO	OAlCl	85
AlCl ₂	AlCl ₂	90,A193
AlCl ₂ H	HAICl ₂	169,B193
AlCl ₃ ⁺	AlCl ₃ ⁺	203
AlFO	FAIO	85
AlFO ₂	FAIO ₂	B232
AlFO ₄	FAI(O ₂) ₂	B330
AlF ₂	AlF ₂	A192,B134
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AlH ₂ N	AlNH ₂	B168
AlH ₂ O	HAIOH	139
AlH ₃	AlH ₃	A212
AlH ₃ N ⁺	AlNH ₃ ⁺	B251
AlH ₃ N	AlNH ₃	B251
AlH ₃ N	HAlNH ₂	B252
AlH ₃ P	AlPH ₃	B253
AlH ₃ P	HAIPH ₂	B253
AlH ₃ P	H ₂ AlPH	B254
AlH ₄ ⁻	AlH ₄ ⁻	A266
AlH ₄ N	H ₂ AlNH ₂	B310
AlH ₄ Si	SiH ₃ AlH	307
AlI ₂	AlI ₂	A193
AlNO	NAIO	B111
AlNO	AlNO	B111
AlNO	AlON	B111
AlNO ⁻	AlNO ⁻	B120
AlN ₂	NAlN	B97

AlN ₃	AlN ₃	B222
AlOSi	AlOSi	B98
AlO ₂	<i>cyc</i> -AlO ₂	75
AlO ₂	OAlO	75,A183
AlO ₂ ⁻	OAlO ⁻	A189
AlO ₃	OAlOO	190
AlP ₂	<i>cyc</i> -AlP ₂	B98
AlP ₂ ⁻	<i>cyc</i> -AlP ₂ ⁻	B111
Al ₂ H ₂	<i>cyc</i> -Al ₂ H ₂	A219
Al ₂ H ₂	(<i>cyc</i> -AlHAl)	A219
Al ₂ H ₂	HAIAIH	A220
Al ₂ N	AlNAI	B63
Al ₂ N ₂	Al ₂ N ₂	B213
Al ₂ O	Al ₂ O	63
Al ₂ O ₂	(AlO) ₂	184,A248
Al ₂ O ₂ ⁻	(AlO) ₂ ⁻	A249
Al ₂ O ₃	Al ₂ O ₃	283,A297
Al ₂ O ₃ ⁻	Al ₂ O ₃ ⁻	A297
Al ₂ O ₄	Al ₂ O ₄	A321
Al ₂ O ₄ ⁻	Al ₂ O ₄ ⁻	A321
Al ₂ O ₅	Al ₂ O ₅	A331
Al ₂ O ₅ ⁻	Al ₂ O ₅ ⁻	B347
Al ₂ P	Al ₂ P	B63
Al ₂ P ⁻	Al ₂ P ⁻	B80
Al ₂ P ₂	Al ₂ P ₂	B213
Al ₂ P ₂ ⁻	Al ₂ P ₂ ⁻	B217
Al ₂ S ₃	Al ₂ S ₃	A297
Al ₃	Al ₃	62,A162
Al ₃ Ge	<i>cyc</i> -GeAl ₃	B202
Al ₃ Ge ⁻	<i>cyc</i> -GeAl ₃ ⁻	B208
Al ₃ N	NAI ₃	B208
Al ₃ N ⁻	Al ₃ N ⁻	B209
Al ₃ O	Al ₃ O	B210
Al ₃ O ⁻	Al ₃ O ⁻	B214
Al ₃ O ₂	Al ₃ O ₂	B283
Al ₃ O ₂ ⁻	Al ₃ O ₂ ⁻	B288
Al ₃ O ₃	Al ₃ O ₃	B329
Al ₃ O ₃ ⁻	Al ₃ O ₃ ⁻	B329
Al ₃ O ₄	Al ₃ O ₄	B345
Al ₃ O ₄ ⁻	Al ₃ O ₄ ⁻	B345
Al ₃ Pb	<i>cyc</i> -PbAl ₃	B202
Al ₃ Pb ⁻	<i>cyc</i> -PbAl ₃ ⁻	B208
Al ₃ Si	<i>cyc</i> -SiAl ₃	B202
Al ₃ Si ⁻	<i>cyc</i> -SiAl ₃ ⁻	B208
Al ₃ Sn	<i>cyc</i> -SnAl ₃	B202
Al ₃ Sn ⁻	<i>cyc</i> -SnAl ₃ ⁻	B208
Al ₄ Ge	Al ₄ Ge	B278
Al ₄ Ge ⁻	Al ₄ Ge ⁻	B279
Al ₄ N	Al ₄ N	B279
Al ₄ N ⁻	Al ₄ N ⁻ (D _{4h})	B279
Al ₄ N ⁻	Al ₄ N ⁻ (C _{2v})	B279
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Al ₄ Si ⁻	Al ₄ Si ⁻	B278
Al ₅	Al ₅	B278
Al ₅ ⁻	Al ₅ ⁻	B278

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
ArBeO	ArBeO	A188	BClH ⁺	HBCl ⁺	B32
ArBeO ₂	ArBeO ₂	A253	BClH ₂	H ₂ BCl	142
ArBrKr	ArKrBr	123	BClN ₂	(<i>cyc</i> -NNB)Cl	B228
ArBrXe	ArXeBr	123	BClO	CIBO	84,A189,B130
ArClKr	ArKrCl	122	BClS ⁺	CIBS ⁺	74,A183
ArDKr	ArKrD	57	BClS	CIBS	85,A189,B130
ArFH	HArF	B47	BCl ₂ ⁺	BCl ₂ ⁺	A189
ArFKr	ArKrF	121,A210	BCl ₂	BCl ₂	89,A192,B133
ArFXe	ArXeF	121	BCl ₂ H ⁺	HBCl ₂ ⁺	166
ArHXe	ArXeH	57	BCl ₂ H	HBCl ₂	A238
Ar ₂ Cl	Ar ₂ Cl	122	BCl ₂ H ₂ N ⁺	NH ₂ BCl ₂ ⁺	332
Ar ₂ H ⁺	HAr ₂ ⁺	56,A159,B48	BCl ₂ N	CIBNCl	B232
Ar ₂ H	Ar ₂ H	56	BCl ₃ ⁺	BCl ₃ ⁺	201,A255,B234
Ar ₂ F	Ar ₂ F	121,A210	BCl ₃ ⁻	BCl ₃ ⁻	A261
AsBr ₃ ⁺	AsBr ₃ ⁺	221	BCl ₄ ⁻	BCl ₄ ⁻	302
AsClO	ClAsO	104	BFH ⁺	HBF ⁺	37
AsCl ₃ ⁺	AsCl ₃ ⁺	221	BFH ₂	H ₂ BF	142
AsFH ₂	AsH ₂ F	152	BFO	FBO	84,A188
AsF ₂	AsF ₂	109,A203	BFS ⁺	FBS ⁺	74
AsF ₂ H	HAsF ₂	179	BFS	FBS	85,B130
AsF ₂ H ₃	AsH ₃ F ₂	325	BF ₂ ⁺	BF ₂ ⁺	A189
AsF ₃ ⁺	AsF ₃ ⁺	221	BF ₂	BF ₂	89,A191
AsGa ₂	Ga ₂ As	A165,B64	BF ₂ H ⁺	HBF ₂ ⁺	165
AsGa ₂ ⁻	Ga ₂ As ⁻	B80	BF ₂ H	HBF ₂	A238
AsHO ₃	HOAsO ₂	273	BF ₂ HO	BF ₂ OH	B274
AsH ₂	AsH ₂	23,B20	BF ₂ H ₂ N ⁺	NH ₂ BF ₂ ⁺	332
AsH ₃ ⁺	AsH ₃ ⁺	129	BF ₂ O	F ₂ BO	200
AsH ₃ O	H ₃ AsO	243	BF ₃ ⁺	BF ₃ ⁺	201,A255,B234
AsH ₃ O	H ₂ AsOH	243	BF ₃ ⁻	BF ₃ ⁻	A261
AsIn ₂	In ₂ As	A165	BF ₄ ⁻	BF ₄ ⁻	302
As ₂ Ga	<i>cyc</i> -GaAs ₂	A172,B98	BGaH ₆ ⁺	GaBH ₆ ⁺	A332
As ₂ Ga ⁻	<i>cyc</i> -GaAs ₂ ⁻	B111	BGaH ₆	GaBH ₆	B348
As ₃	As ₃	B127	BHN	HNB	31,A141
As ₃ Ga ₂	Ga ₂ As ₃	B282	BHO	HBO	35,B32
As ₃ Ga ₂ ⁻	Ga ₂ As ₃ ⁻	B288	BHO ₂	HOBO	162
As ₄ ⁺	As ₄ ⁺	187	BHS ⁺	HBS ⁺	31,A141
As ₄ O	As ₄ O	284	BHS	HBS	35
As ₄ O	br-As ₄ O	284	BH ₂	BH ₂	17
AuCu ₂	Cu ₂ Au	60	BH ₂ N	HBNH	137,A222
AuH ₃	HAu(H ₂)	B154	BH ₂ N	H ₂ NB	A223
AuO ₂	AuOO	A178,B108	BH ₃	BH ₃	124,A212
AuO ₂	OAuO	B108	BH ₃ ⁻	BH ₃ ⁻	125
Au ₃	Au ₃	60,A161	BH ₃ N	HBNH ₂	A269
BBrH ⁺	HBBr ⁺	B32	BH ₃ O	H ₂ BOH	233
BBrH ₂	H ₂ BBr	142	BH ₃ O ₃	B(OH) ₃	A326
BBrO	BrBO	85,B130	BH ₃ S	H ₂ BSH	233
BBrS	BrBS	85	BH ₄ ⁻	BH ₄ ⁻	A266
BBr ₂ ⁺	BBr ₂ ⁺	A189	BH ₄ K	KBH ₄	A308
BBr ₂	BBr ₂	89,A192	BH ₄ Li	LiBH ₄	305,A307
BBr ₂ H ⁺	HBBr ₂ ⁺	166	BH ₄ N ⁺	H ₂ BNH ₂ ⁺	307
BBr ₂ H	HBBr ₂	A238	BH ₄ N	H ₂ BNH ₂	309
BBr ₂ H ₂ N ⁺	NH ₂ BBr ₂ ⁺	332	BH ₄ Na	NaBH ₄	305,A307
BBr ₃ ⁺	BBr ₃ ⁺	202,B235	BH ₄ Sr	SrBH ₄	306
BCaH ₄	CaBH ₄	305	BI ₂	BI ₂	89
BClF ₃ ⁻	BF ₃ Cl ⁻	302	BI ₃ ⁺	BI ₃ ⁺	202,B236

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
BNSi	BNSi	B79	BeO ₃	BeO ₃	A250
BN ₂	NBN	67	BeO ₄	O ₂ BeO ₂	A297
BN ₂	BNN	69	Be ₂ H	BeBeH	A136
BN ₃	NNBN	184,B222	Be ₂ H ₂	HBeBeH	A217
BO ₂	BO ₂	73,B120	Be ₂ H ₂ O	HBeOBeH	A275
BO ₂ ⁻	BO ₂ ⁻	84	Be ₂ H ₃	HBeHBeH	A267
BO ₂ S	OBSO	190	Be ₂ H ₄	HBeH ₂ BeH	A308
BS ₂	BS ₂	74,A182	Be ₂ NO ₂	BeOBeNO	B281
B ₂ Cl ₄ ⁺	B ₂ Cl ₄ ⁺	348	Be ₂ N ₂	BeBeNN	A242
B ₂ F ₄ ⁺	B ₂ F ₄ ⁺	347	Be ₂ N ₂	BeNNBe	A242
B ₂ HN	HBNB	A232	Be ₂ N ₂	(BeN) ₂	A242
B ₂ H ₂	HBBH	A219	Be ₂ N ₄	NNBeBeNN	A317
B ₂ H ₂ N ₂	HNBBNH	A315	Be ₂ O	BeOBe	A163
B ₂ H ₃ N	H ₂ BNBH	A311	Be ₂ O ₂	<i>cyc</i> -(BeO) ₂	A242
B ₂ H ₄ ⁺	B ₂ H ₄ ⁺	306	Be ₂ O ₂	BeOBeO	A242
B ₂ H ₅ ⁺	B ₂ H ₅ ⁺	359	Be ₂ O ₃	OBeOBeO	A294
B ₂ H ₆ ⁺	B ₂ H ₆ ⁺	A332	BiHO	BiOH	B42
B ₂ H ₇ N	B ₂ H ₅ NH ₂	398	Bi ₃	Bi ₃	82
B ₂ N	BNB	62,A165,B62	Bi ₄	Bi ₄	189
B ₂ N ⁻	BNB ⁻	63,A165,B78	Bi ₄ ⁻	Bi ₄ ⁻	197
B ₂ N ₂	BBNN	181	BrClH ⁻	ClHBr ⁻	54
B ₂ N ₂	BNBN	181	BrClO	BrClO	A209
B ₂ N ₂	<i>cyc</i> -(BN) ₂	181	BrClO	ClBrO	A209
B ₂ O	BBO	63	BrClO	BrOCl	A207,B151
B ₂ O	BOB	63	BrClO ₂	BrClO ₂	A263
B ₂ O ₂ ⁺	B ₂ O ₂ ⁺	183	BrClO ₂	BrOCIO	A264
B ₂ O ₂	B ₂ O ₂	184	BrClO ₂	ClOBrO	A264
B ₃	B ₃	A162	BrClS	ClSBr	B152
B ₃ F ₃ H ₃ N ₃ ⁺	<i>cyc</i> -(FBNH) ₃ ⁺	399	BrCl ₂ ⁻	ClBrCl ⁻	119
B ₃ N	BNBB	B207	BrCl ₂ ⁻	ClClBr ⁻	119
B ₃ N ⁻	BNBB ⁻	B209	BrFH ⁻	FHBr ⁻	54
BaHO	BaOH	30,B24	BrFS	FSBr	A208
BaH ₂ N	BaNH ₂	133	BrF ₂	BrF ₂	117
BaH ₂ O	HBaOH	135	BrF ₂ ⁻	FBrF ⁻	118
BaH ₂ O ₂	Ba(OH) ₂	251	BrF ₂ ⁻	FFBr ⁻	118
BaO ₂	OBaO	A175	BrF ₂ P ⁺	PF ₂ Br ⁺	220
BaO ₂	<i>cyc</i> -BaO ₂	A175	BrF ₃ ⁺	BrF ₃ ⁺	226
Ba ₂ O ₂	<i>cyc</i> -(BaO) ₂	A243	BrF ₃ Si ⁺	SiF ₃ Br ⁺	295
Ba ₂ O ₂	BaOBaO	A243	BrF ₅ ⁺	BrF ₅ ⁺	358
BeBr ₂	BeBr ₂	84	BrFeH	HFeBr	35
BeCl ₂	BeCl ₂	84	BrGeH	HGeBr	45,B41
BeF ₂	BeF ₂	83	BrGeH ₂	H ₂ GeBr	150
BeHO	BeOH	A138	BrGeH ₃ ⁺	GeH ₃ Br ⁺	241
BeH ₂	BeH ₂	A125	BrHI ⁻	BrHI ⁻	55
BeH ₂ O	HBeOH	A222	BrHO ⁺	HOBr ⁺	A155
BeH ₂ O ₂	Be(OH) ₂	A281	BrHO	HOBr	52,A156,B45
BeI ₂	BeI ₂	84	BrHSi	HSiBr	44,A151,B40
BeKrO	KrBeO	A188	BrHXe	HXeBr	A158,B47
BeNO	BeNO	B87	BrH ₂ N ⁺	H ₂ NBr ⁺	152
BeN ₂ O ₂	NNBeO ₂	A296	BrH ₂ P	H ₂ PBr	A229
BeN ₄	NNBeNN	A294	BrH ₃ Si ⁺	SiH ₃ Br ⁺	241
BeN ₄	NN(<i>cyc</i> -BeN ₂)	A294	BrKrXe	KrXeBr	123
BeOXe	XeBeO	A188	BrKr ₂	Kr ₂ Br	123
BeO ₂ ⁺	OBeO ⁺	A171	BrNO ⁺	BrNO ⁺	94
BeO ₂	OBeO	A174	BrNO	BrON	B143

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
BrNO ₂	BrNO ₂	213,A260,B239	CAISi	AlCSi	B63
BrNO ₂	<i>c</i> -BrONO	B239	CAISi ⁻	AlCSi ⁻	B78
BrNO ₂	<i>t</i> -BrONO	213,A260,B240	CAISi ₂	CAISi ₂	B210
BrNO ₃ ⁺	BrONO ₂ ⁺	A301	CAISi ₂ ⁻	<i>cis</i> -CAISi ₂ ⁻	B213
BrNO ₃	BrONO ₂	289,A303	CAI ₂	Al ₂ C	A164
BrNS ⁺	NSBr ⁺	95	CAI ₃	CAI ₃	B201
BrNS	NSBr	104	CAI ₃ ⁻	CAI ₃ ⁻	B208
BrN ₃ ⁺	BrN ₃ ⁺	197	CAI ₃ Ge	CAI ₃ Ge	B279
BrNeXe	NeXeBr	123	CAI ₃ Ge ⁻	CAI ₃ Ge ⁻	B280
BrOP	BrPO	103,B142	CAI ₃ Si	CAI ₃ Si	B279
BrOPS	BrP(O)S	214	CAI ₃ Si ⁻	CAI ₃ Si ⁻	B279
BrO ₂	BrOO	110,A204,B147	CAI ₄	Al ₄ C	B278
BrO ₂	OBRO	114,A206,B150	CAI ₄ ⁻	Al ₄ C ⁻	B278
BrO ₂ P	PO ₂ Br	214	Al ₅ C	Al ₅ C	B325
BrPS	BrPS	104	Al ₅ C ⁻	Al ₅ C ⁻	B325
BrS ₂	SSBr	111	CA ₅ N	AsCN	71
BrXe ₂	Xe ₂ Br	123	CA ₅ O	AsCO	B123
Br ₂ Cl ⁻	BrClBr ⁻	119	CA ₅ O ⁻	AsCO ⁻	B131
Br ₂ Cl ⁻	ClBrBr ⁻	119	CAuO ⁺	AuCO ⁺	B62
Br ₂ F	BrBrF	117	CAuO	AuCO	B68
Br ₂ F ₂	Br ₂ F ₂	227	CBF ₂ NO	F ₂ BNCO	A321
Br ₂ GaH	HGaBr ₂	B194	CBN	BCN	B78
Br ₂ Ge ⁺	GeBr ₂ ⁺	92	CBN	BNC	B78
Br ₂ GeH ₂ ⁺	GeH ₂ Br ₂ ⁺	263	CBO	BCO	68
Br ₂ H ⁻	BrHBr ⁻	55,A157	CBO ₂	OBCO	186
Br ₂ HN ⁺	HNBr ₂ ⁺	179	CBS ₂	SBCS	A249
Br ₂ H ₂ ⁺	HBrBrH ⁺	B179	CB ₂	<i>cyc</i> -B ₂ C	B55
Br ₂ H ₂ Si ⁺	SiH ₂ Br ₂ ⁺	262	CBa ₂ N	BaNC	B60
Br ₂ Li	BrLiBr	B115	CBe ₂ N	BeCN	B58
Br ₂ N	NBr ₂	108	CBe ₂ N	BeCN	B58
Br ₂ Na	BrNaBr	B116	CBeO ₂	OCBeO	A247
Br ₂ O	BrOBr	114,A207,B151	CBeO ₂	COBeO	A248
Br ₂ O	BrBrO	116,A209,B153	CBe ₂ O	BeBeCO	A241
Br ₂ OP	OPBr ₂	221	CBrCl ⁺	CClBr ⁺	91,B135
Br ₂ P	PBr ₂	109,A203	CBrCl	CClBr	99,B139
Br ₂ S ⁺	SBr ₂ ⁺	112	CBrCl ₂ ⁺	CCl ₂ Br ⁺	211
Br ₂ S	SBr ₂	115	CBrCl ₂	CCl ₂ Br	218
Br ₂ S ₂ ⁺	S ₂ Br ₂ ⁺	224	CBrClF ₂ ⁻	CF ₂ ClBr ⁻	303
Br ₂ S ₂	SSBr ₂	226	CBrF	CFBr	98,B138
Br ₂ Se ⁺	SeBr ₂ ⁺	113	CBrF ₂ ⁺	CF ₂ Br ⁺	210
Br ₂ Si	SiBr ₂	101	CBrF ₂	CF ₂ Br	216
Br ₃ ⁻	Br ₃ ⁻	119	CBrF ₂ ⁻	CF ₂ Br ⁻	225
Br ₃ P ⁺	PBr ₃ ⁺	221,B242	CBrF ₃ ⁺	CF ₃ Br ⁺	291
Br ₃ PO ⁺	Br ₃ PO ⁺	299	CBrF ₃ ⁻	CF ₃ Br ⁻	303
Br ₃ PS ⁺	Br ₃ PS ⁺	299	CBrF ₄	CF ₃ BrF	356
Br ₃ Sb ⁺	SbBr ₃ ⁺	222	CBrN ⁺	BrCN ⁺	80,A187
Br ₄ Ge ⁺	GeBr ₄ ⁺	297	CBrN	BrNC	88
Br ₄ Si ⁺	SiBr ₄ ⁺	296,B305	CBrNO ⁺	BrCNO ⁺	A252
CAgN	AgCN	B60	CBrNO ⁺	BrNCO ⁺	193
CAgN ⁻	AgCN ⁻	B64	CBrNO	BrCNO	199,A253,B233
CAgO ⁺	AgCO ⁺	B62	CBrNO	BrNCO	198
CAIN	AlCN	B79	CBrNS ⁺	BrSCN ⁺	194
CAIN	AlNC	B79	CBrNS	BrSCN	199
CAIO	AlCO	68,A172	CBrNSe ⁺	BrSeCN ⁺	195
CAIO ⁻	AlCO ⁻	B111	CBr ₂ ⁺	CBr ₂ ⁺	91

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
CBr_2	CBr_2	100,B139	CCl_3F^-	CFCl_3^-	303
CBr_2^-	CBr_2^-	B145	CCl_3NO^+	CCl_3NO^+	353
CBr_2Cl^+	CClBr_2^+	211	CCl_3NO_4	$\text{CCl}_3\text{OONO}_2$	426
CBr_2Cl	CClBr_2	218	CCl_3O_2	CCl_3O_2	354
CBr_2F^+	CFBr_2^+	210	CCl_4^+	CCl_4^+	294,A305
CBr_2F	CFBr_2	216	CCl_4	$\text{Cl}_2\text{CCl-Cl}$	302,B306
CBr_2F_2^+	CF_2Br_2^+	292	CCl_4^-	$\text{Cl}_2\text{CCl-Cl}^-$	B307
CBr_2F_2^-	CF_2Br_2^-	303	CCl_4^-	CCl_4^-	B307
CBr_3^+	CBr_3^+	211	CCoO^+	CoCO^+	B61
CBr_3	CBr_3	218	CCoO	CoCO	A167,B67
CBr_3F^+	CFBr_3^+	293	CCoO^-	CoCO^-	B86
CBr_3F^-	CFBr_3^-	303	CCoO_2	OCoCO	B221
CBr_4^+	CBr_4^+	294	CCoO_2^-	OCoCO^-	B224
CCaN	CaNC	62,A164,B59	CCoO_2^-	CoCO_2^-	B224
CCaNO	CaNCO	182	CCrO	CrCO	A166,B65
CClF	CFCl	98,B138	CCrO_2^+	OCrCO^+	B217
CClFOS	ClFCSO (I)	288	CCrO_2	OCrCO	B220
CClFOS	ClFCSO (II)	288	CCrO_2	CrOCO	B221
CClFS^+	FClCS^+	205	CCrO_2^-	OCrCO^-	B224
CClF_2^+	CF_2Cl^+	210	CCuN	CuCN	B60
CClF_2	CF_2Cl	216	CCuN^-	CuCN^-	B64
CClF_2NO^+	CF_2ClNO^+	353	CCuN^-	CuNC^-	B64
CClF_2NO_4	$\text{CF}_2\text{ClOONO}_2$	425	CCuO^+	CuCO^+	B62
CClF_3^+	CF_3Cl^+	290,B303	CCuO	CuCO	A167,B68
CClF_3^-	CF_3Cl^-	303	CCuO^-	CuCO^-	B86
CClF_4	CF_3ClF	356	CCuO_2^-	OCuCO^-	B225
CCIN^+	ClCN^+	79,B126	CCuO_2^-	CuCO_2^-	B225
CCIN	ClNC	88	CFI	CFI	99
CCINO^+	ClCNO^+	B232	CFI_2^+	CFI_2^+	210
CCINO^+	ClNCO^+	193	CFI_2	CFI_2	217
CCINO	ClCNO	199,B233	CFN^+	FCN^+	79
CCINO	ClNCO	198	CFN	FNC	88
CCINS^+	ClSCN^+	194	CFNO	FNCO	198,A253
CCINS	ClSCN	198	CFNO_5	FC(O)OONO_2	A342
CCINSe^+	ClSeCN^+	195	CFNS^+	FSCN^+	193
CCIO	ClCO	90,B135	CFN_2	FNCN	192
CCIP^+	ClCP^+	81	CFO^+	FCO^+	86
CCIP	ClCP	A190	CFO	FCO	90,A193,B134
CCIS	ClCS	105	CFO_2	FCO_2	203,A256
CCl_2^+	CCl_2^+	91,B135	CFO_2^-	FCO_2^-	212,A259
CCl_2	CCl_2	99,B138	CFO_3	<i>c</i> - FC(O)OO	B300
CCl_2^-	CCl_2^-	107,B145	CFO_3	<i>t</i> - FC(O)OO	287,B301
CCl_2F^+	CFCl_2^+	210	CFP^+	FCP^+	81
CCl_2F	CFCl_2	216	CFP	FCP	A190
CCl_2FNO^+	CFCl_2NO^+	353	CFS	FCS	A194
CCl_2FNO_4	$\text{CFCl}_2\text{OONO}_2$	426	CF_2^+	CF_2^+	91
CCl_2F_2^+	CF_2Cl_2^+	292,A304	CF_2	CF_2	97,A196,B137
CCl_2F_2^-	CF_2Cl_2^-	303	CF_2^-	CF_2^-	107,B145
CCl_2I	CCl_2I	218	CF_2I^+	CF_2I^+	210
CCl_2O^+	Cl_2CO^+	204	CF_2I	CF_2I	216
CCl_2S^+	Cl_2CS^+	206	CF_2N	F_2CN	204
CCl_2Se	Cl_2CSe	212	CF_2NOP^+	PF_2NCO^+	349
CCl_3^+	CCl_3^+	210,B238	CF_2NP^+	PF_2CN^+	285
CCl_3	CCl_3	217,B240	CF_2NPS^+	PF_2NCS^+	350
CCl_3F^+	CFCl_3^+	293	CF_2N_2^+	CF_2N_2^+	285

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
CF ₂ O ⁺	F ₂ CO ⁺	204	CHCaO ₂	HCOOCa	268
CF ₂ OS	F ₂ CSO	288	CHCl ⁺	HCCl ⁺	B36
CF ₂ O ₂	<i>cyc</i> -F ₂ CO ₂	A302	CHCl	HCCl	42,A149,B37
CF ₂ S ⁺	F ₂ CS ⁺	205	CHCl ⁻	HCCl ⁻	47
CF ₂ Se ⁺	F ₂ CSe ⁺	206	CHClF ⁺	HCFCl ⁺	171
CF ₃ ⁺	CF ₃ ⁺	210,A258	CHClF	HCFCl	176
CF ₃ ⁻	CF ₃ ⁻	215,A261	CHClF ₂ ⁺	HCF ₂ Cl ⁺	275
CF ₃ I ⁺	CF ₃ I ⁺	A263	CHClO ⁺	HCOCI ⁺	169
CF ₃ I ⁻	CF ₃ I ⁻	291	CHClO	HCOCI	170,A239,B194
CF ₃ IO	CF ₃ IO	303	CHCl ₂ ⁺	HCCl ₂ ⁺	171,B195
CF ₃ IO	CF ₃ IO	356	CHCl ₂	HCCl ₂	177,B199
CF ₃ IO	CF ₃ OI	356	CHCl ₂ F ⁺	HCFCl ₂ ⁺	275
CF ₃ N	CF ₃ N	B301	CHCl ₂ NO ₄	CHCl ₂ OONO ₂	425
CF ₃ NO ⁺	CF ₃ NO ⁺	353	CHCl ₃ ⁺	HCCl ₃ ⁺	275
CF ₃ NO ₃	CF ₃ ONO ₂	382,A342	CHCl ₃ O	CCl ₃ OH	B324
CF ₃ NO ₄	CF ₃ OONO ₂	A368,B395	CHF ⁺	HCF ⁺	41
CF ₃ O	CF ₃ O	289,A303,B302	CHF	HCF	42,A148,B36
CF ₃ O ⁻	CF ₃ O ⁻	302	CHF ⁻	HCF ⁻	47
CF ₃ O ₂	CF ₃ O ₂	354,B332	CHF ⁺	HCF ⁺	171
CF ₃ S	CF ₃ S	A303,B302	CHFI	HCFI	177
CF ₄ ⁺	CF ₄ ⁺	290,A303,B303	CHFN	HFCN	168,B193
CF ₄ I	CF ₃ IF	356	CHFN	FCNH	B193
CF ₄ O ⁺	CF ₃ OF ⁺	355	CHFO ⁺	HFCO ⁺	168
CF ₄ OS	CF ₃ OSF	372	CHFS ⁺	HFCS ⁺	A237
CFeN	FeNC	B60	CHF ₂ ⁺	HCF ₂ ⁺	170
CFeO ⁺	FeCO ⁺	B61	CHF ₂	HCF ₂	176,B199
CFeO	FeCO	63,A167,B66	CHF ₂ N ⁺	CF ₂ NH ⁺	A289
CFeO ⁻	FeCO ⁻	67,B85	CHF ₂ N	CF ₂ =NH	270,A289
CFeO ₂	OFeCO	B221	CHF ₂ N	<i>c</i> -HFC=NF	271
CFeO ₂ ⁻	OFeCO ⁻	B224	CHF ₂ P	CF ₂ =PH	271
CFe ₂ O	Fe ₂ CO	B204	CHF ₃ ⁺	HCF ₃ ⁺	274,B276
CGaN	GaCN	B79	CHGe	HCGe	B30
CGaN	GaNC	B79	CHI	HCI	43
CGaO	GaCO	B97	CHI ⁻	HCI ⁻	48
CGeO	GeCO	A180,B112	CHIO	HCOI	A240
CGeO ⁻	GeCO ⁻	B121	CHI ₂	HCl ₂	178
CHBBr	HCBBr	B188	CHK	KCH	B22
CHBCl	HCBCl	B187	CHKrN	HKrCN	B195
CHBF	HCBF	B187	CHN ⁺	HCN ⁺	33,B31
CHBN	HBCN	B183	CHN ⁺	HNC ⁺	34
CHBN	HBNC	B183	CHN	HNC	38,A145,B33
CHBN	<i>cyc</i> -HB(CN)	B183	CHNO ⁺	HNCO ⁺	160
CHBeN	HBeCN	B182	CHNO ⁺	HCNO ⁺	161
CHBeN	HBeNC	B182	CHNO	HNCO	163,A236,B190
CHBr	HCBBr	43,A149,B38	CHNO	HOCN	163,B191
CHBr ⁻	HCBBr ⁻	47	CHNO	HCNO	164,B191
CHBrCl ⁺	HCClBr ⁺	171,B195	CHNOXe	HXeNCO	B276
CHBrCl	HCClBr	178,B199	CHNP	HPCN	160
CHBrClF ⁺	HCFClBr ⁺	A292	CHNS ⁺	HNCS ⁺	160
CHBrF ⁺	HCFBr ⁺	171	CHNS	HSCN	B192
CHBrF	HCFBr	177	CHNS	HSNC	B192
CHBrO	HCOBr	170,A239	CHNSi	HCNSi	B186
CHBr ₂ ⁺	HCBBr ₂ ⁺	172	CHNSi	<i>cyc</i> -(HCSiN)	B186
CHBr ₂	HCBBr ₂	178	CHNSi	HSiNC	B187
CHBr ₃ ⁺	HCBBr ₃ ⁺	276	CHNSi	HSiCN	B187

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
CHNXe	HXeCN	B196	CH ₂ CIN	CH ₂ =NCl	258
CHNXe	HXeNC	B196	CH ₂ CINO ₄	CH ₂ CIOONO ₂	425
CHN ₂	HCNN	161,A235,B189	CH ₂ CIP	CH ₂ =PCl	258
CHN ₂	HNCN	160,A235,B188	CH ₂ Cl ₂ ⁺	CH ₂ Cl ₂ ⁺	261,A285
CHN ₂ ⁻	HCNN ⁻	B190	CH ₂ Cl ₂	H ₂ CCl-Cl	263,B269
CHN ₂ ⁻	HNCN ⁻	B190	CH ₂ Cl ₂ O	HCCl ₂ OH	339,B322
CHNa	NaCH	B22	CH ₂ Cl ₂ O ₂	CHCl ₂ OOH	368
CHO ⁺	HCO ⁺	37,A145,B33	CH ₂ Cl ₂ Si	CH ₂ =SiCl ₂	337,B322
CHO ⁺	HOC ⁺	39,B34	CH ₂ Co	CoCH ₂	A219
CHO	HCO	40,A146,B34	CH ₂ Cr	CrCH ₂	A218
CHOP	HPCO	165	CH ₂ Cu	CuCH ₂	132
CHOS ⁺	HOCS ⁺	165,A236	CH ₂ F ⁺	H ₂ CF ⁺	147
CHO ₂ ⁺	HOCO ⁺	165	CH ₂ F	H ₂ CF	149,B176
CHO ₂	<i>c</i> -HOCO	167	CH ₂ FI ⁺	CH ₂ FI ⁺	261
CHO ₂	<i>t</i> -HOCO	167,A236	CH ₂ FMg	CH ₂ MgF	B263
CHO ₂	HCO ₂	A237	CH ₂ FO ₂	CH ₂ FO ₂	339
CHO ₂ ⁻	HCO ₂ ⁻	A230	CH ₂ F ₂ ⁺	CH ₂ F ₂ ⁺	260,A284
CHO ₂ Sr	HCOOSr	268	CH ₂ Fe	FeCH ₂	131
CHO ₃	HC(O)OO	270	CH ₂ Fe	HFeCH	132
CHP ⁺	HCP ⁺	34	CH ₂ Ge	H ₂ CGe	B170
CHS ⁺	HCS ⁺	38,A145	CH ₂ I	H ₂ CI	150
CHS	HCS	B36	CH ₂ I ₂ ⁺	CH ₂ I ₂ ⁺	262
CHS	HSC	B36	CH ₂ I ₂	H ₂ CI-I	265
CHS ₂	<i>t</i> -HS ₂ CS	167	CH ₂ Mg	MgCH ₂	B162
CHS ₂	HCS ₂	168	CH ₂ N ⁺	HCNH ⁺	138,B170
CHSi	HCSi	B30	CH ₂ N	H ₂ CN	140,B171
CHTi	TiCH	A137	CH ₂ N	HCNH	B171
CHV	VCH	A137	CH ₂ N ⁻	H ₂ CN ⁻	143
CHW	WCH	A137	CH ₂ NO	H ₂ NCO	B268
CH ₂ ⁺	CH ₂ ⁺	18	CH ₂ NO	H ₂ CNO	A283
CH ₂	CH ₂	18,A130,B16	CH ₂ NOSr	HCONHSr	326
CH ₂ ⁻	CH ₂ ⁻	21	CH ₂ NO ₂	CH ₂ NO ₂	331,A316
CH ₂ B	HBCH	A221	CH ₂ NO ₂ ⁻	CH ₂ NO ₂ ⁻	337
CH ₂ BBr	H ₂ CBBr	B266	CH ₂ N ₂ ⁺	CH ₂ N ₂ ⁺	253
CH ₂ BCl	H ₂ CBCl	B266	CH ₂ N ₂ ⁺	<i>cyc</i> -CH ₂ N ₂ ⁺	254
CH ₂ BF	H ₂ CBF	B266	CH ₂ N ₂ ⁺	NH ₂ CN ⁺	253
CH ₂ BN	BH ₂ CN	B262	CH ₂ N ₂	HN=C=NH	254,A282,B266
CH ₂ BN	CH ₂ NB	B262	CH ₂ N ₂	HCNNH	A282
CH ₂ BO	CH ₂ BO	B263	CH ₂ N ₂ O	H ₂ NNCO	331
CH ₂ B ₂	HBCBH	245,A275	CH ₂ Ni	NiCH ₂	A219
CH ₂ Be	HCB _e H	B162	CH ₂ O ⁺	H ₂ CO ⁺	140,A223,B172
CH ₂ Br ⁺	H ₂ CBr ⁺	147	CH ₂ OS ⁺	H ₂ CSO ⁺	258
CH ₂ Br	H ₂ CBr	150,A228	CH ₂ OS	H ₂ CSO	259,A284
CH ₂ BrCl	H ₂ CCl-Br	264,B269	CH ₂ O ₂ ⁺	HCOOH ⁺	257,B268
CH ₂ BrF ⁺	CH ₂ FBr ⁺	261	CH ₂ O ₂	<i>cyc</i> -H ₂ CO ₂	A283
CH ₂ BrI	H ₂ CBr-I	265	CH ₂ O ₃	HC(O)OOH	336
CH ₂ BrMg	CH ₂ MgBr	B263	CH ₂ P	H ₂ CP	B171
CH ₂ Br ₂ ⁺	CH ₂ Br ₂ ⁺	261	CH ₂ S ⁺	H ₂ CS ⁺	141
CH ₂ Br ₂	H ₂ CBr-Br	264	CH ₂ S	H ₂ CS	143,A224,B173
CH ₂ CaNO	HCONHCa	326	CH ₂ S ⁻	H ₂ CS ⁻	149
CH ₂ Cl ⁺	H ₂ CCl ⁺	147	CH ₂ S ₂	H ₂ CSS	B269
CH ₂ Cl	H ₂ CCl	149,A228,B176	CH ₂ S ₂	<i>cyc</i> -H ₂ CS ₂	B269
CH ₂ ClF ⁺	CH ₂ FCI ⁺	261	CH ₂ S ₂	<i>t</i> -HCSSH	259
CH ₂ ClI	H ₂ CCl-I	264	CH ₂ S ₂	<i>c</i> -HCSSH	259
CH ₂ ClMg	CH ₂ MgCl	B263	CH ₂ Se ⁺	H ₂ CSe ⁺	141

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
CH ₂ Se	H ₂ CSe	145,A226,B174	CH ₃ IO	ICH ₂ OH	323
CH ₂ Si	H ₂ CSi	138,A223,B169	CH ₃ In	InCH ₃	B250
CH ₂ Zn	ZnCH ₂	131	CH ₃ K	KCH ₃	230,B247
CH ₂ Zn	HZnCH	131	CH ₃ Li	LiCH ₃	B247
CH ₃ ⁺	CH ₃ ⁺	124,A213,B156	CH ₃ Mg ⁺	MgCH ₃ ⁺	B248
CH ₃	CH ₃	125,A214,B156	CH ₃ Mg	MgCH ₃	A267,B249
CH ₃ ⁻	CH ₃ ⁻	129	CH ₃ N ⁺	CH ₂ NH ⁺	233
CH ₃ Al	AlCH ₃	A268	CH ₃ N	CH ₃ N	234,A269,B256
CH ₃ B	H ₂ C=BH	A268	CH ₃ N	CH ₂ NH	234
CH ₃ BBr	CH ₃ BBr	B316	CH ₃ NO ⁺	CH ₃ NO ⁺	318
CH ₃ BCl	CH ₃ BCl	B316	CH ₃ NO ⁺	CH ₂ NOH ⁺	318
CH ₃ BN	CH ₂ BNH	B314	CH ₃ NO ⁺	HCONH ₂ ⁺	318
CH ₃ BN	CH ₃ NB	B314	CH ₃ NO	CH ₃ NO	320
CH ₃ BO ⁺	BH ₃ CO ⁺	312	CH ₃ NO	CH ₂ NOH	320,A314
CH ₃ BO	CH ₃ BO	B315	CH ₃ NO ₂	<i>c</i> -CH ₂ (NO)OH	367
CH ₃ BO	CH ₂ BOH	B316	CH ₃ NO ₂	<i>t</i> -CH ₂ (NO)OH	367
CH ₃ Ba	BaCH ₃	B249	CH ₃ NO ₄	CH ₃ OONO ₂	425
CH ₃ BaO	BaOCH ₃	315	CH ₃ NO ₅	HOCH ₂ OONO ₂	425
CH ₃ Be	BeCH ₃	B248	CH ₃ NS ⁺	HCSNH ₂ ⁺	319
CH ₃ Be	CH ₂ BeH	B248	CH ₃ Na	NaCH ₃	229,B247
CH ₃ Br ⁺	CH ₃ Br ⁺	239,A274	CH ₃ O ⁺	CH ₃ O ⁺	235,B256
CH ₃ Br ⁺	H ₂ CBrH ⁺	A274	CH ₃ O ⁺	CH ₂ OH ⁺	235,A270
CH ₃ BrF	CH ₃ BrF	324	CH ₃ O	CH ₃ O	236,A271,B257
CH ₃ BrMg	CH ₃ MgBr	B318	CH ₃ O	CH ₂ OH	237,A272,B258
CH ₃ Ca	CaCH ₃	230,A267,B249	CH ₃ O ⁻	CH ₃ O ⁻	242,B260
CH ₃ CaO	CaOCH ₃	314,B315	CH ₃ OSr	SrOCH ₃	315
CH ₃ CaS	CaSCH ₃	315	CH ₃ O ₂	CH ₃ O ₂	321,B320
CH ₃ Cd ⁺	CdCH ₃ ⁺	B248	CH ₃ O ₂ ⁻	CH ₃ O ₂ ⁻	B321
CH ₃ Cd	CdCH ₃	231,A268,B250	CH ₃ O ₃	HOCH ₂ O ₂	368
CH ₃ Cl ⁺	CH ₃ Cl ⁺	239,A273,B259	CH ₃ P ⁺	CH ₂ PH ⁺	233
CH ₃ Cl ⁺	H ₂ CClH ⁺	A274	CH ₃ S ⁺	CH ₃ S ⁺	235,A270,B257
CH ₃ ClF	CH ₃ ClF	324	CH ₃ S ⁺	CH ₂ SH ⁺	235
CH ₃ ClMg	CH ₃ MgCl	B318	CH ₃ S	CH ₃ S	237,A272,B259
CH ₃ ClMg	HMgCH ₂ Cl	B318	CH ₃ S	CH ₂ SH	238
CH ₃ ClO ⁺	CH ₃ OCl ⁺	322	CH ₃ S ⁻	CH ₃ S ⁻	242,B261
CH ₃ ClO	ClCH ₂ OH	323,B321	CH ₃ SSr	SrSCH ₃	315
CH ₃ ClO ₂	CH ₂ ClOOH	368	CH ₃ S ₂ ⁺	CH ₃ S ₂ ⁺	B320
CH ₃ ClS ⁺	CH ₃ SCl ⁺	B320	CH ₃ S ₂	CH ₃ S ₂	322
CH ₃ ClS	CH ₃ SCl	B321	CH ₃ S ₂ ⁻	CH ₃ S ₂ ⁻	322
CH ₃ ClSi	CH ₂ =SiHCl	321	CH ₃ Si	CH ₂ SiH	232
CH ₃ ClSi	CH ₃ SiCl	321	CH ₃ Si	CH ₃ Si	232
CH ₃ Cl ₂ O ₂ V	Cl ₂ V(O)OCH ₃	B393	CH ₃ Si ⁻	CH ₂ SiH ⁻	233
CH ₃ Cl ₃ OTi	Cl ₃ TiOCH ₃	B390	CH ₃ Si ⁻	CH ₃ Si ⁻	233
CH ₃ F ⁺	CH ₃ F ⁺	238	CH ₃ Sr	SrCH ₃	230,A267
CH ₃ F ⁺	H ₂ CFH ⁺	A273	CH ₃ Te	CH ₃ Te	238
CH ₃ FI	CH ₃ IF	324	CH ₃ Zn ⁺	ZnCH ₃ ⁺	B247
CH ₃ FMg	CH ₃ MgF	B317	CH ₃ Zn	ZnCH ₃	230,A267,B249
CH ₃ FMg	HMgCH ₂ F	B318	CH ₄ ⁺	CH ₄ ⁺	227,B246
CH ₃ FO ⁺	CH ₃ OF ⁺	322	CH ₄ B	H ₂ CBH ₂	A309
CH ₃ FO	CH ₃ OF	322	CH ₄ B	CH ₃ BH	A310
CH ₃ Ga	GaCH ₃	B250	CH ₄ BN	CH ₃ BNH	B336
CH ₃ I ⁺	CH ₃ I ⁺	239,B260	CH ₄ BN	CH ₃ NBH	B336
CH ₃ IMg	CH ₃ MgI	B318	CH ₄ BN	CH ₂ BNH ₂	B336
CH ₃ IO	CH ₃ IO	323	CH ₄ Be	CH ₃ BeH	B308
CH ₃ IO	CH ₃ OI	323	CH ₄ Cd	CH ₃ CdH	A309

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
CH ₄ Co	CH ₃ CoH	A308	CMoO ₂	OMoCO	B221
CH ₄ Fe	CH ₃ FeH	306	CNO ⁺	NCO ⁺	71
CH ₄ Ga	CH ₃ GaH	307,A310,B309	CNO	CNO	82
CH ₄ Ge	CH ₃ GeH	B312	CNO	NCO	76,A184,B122
CH ₄ Hg	CH ₃ HgH	A309,B309	CNO ⁻	NCO ⁻	85,B131
CH ₄ In	CH ₃ InH	B310	CNOP	NCPO	B229
CH ₄ Mg	CH ₃ MgH	B308	CNOSr	SrNCO	182
CH ₄ N ⁺	CH ₂ NH ₂ ⁺	310	CNO ₃ ⁻	O ₂ CNO ⁻	B299
CH ₄ N	CH ₂ NH ₂	B312	CNP	PCN	70
CH ₄ Ni	CH ₃ NiH	306	CNPd ⁻	PdCN ⁻	B64
CH ₄ NiO	HNiOCH ₃	362	CNS	NCS	77,B123
CH ₄ NiO	CH ₃ NiOH	362	CNS ⁻	NCS ⁻	86
CH ₄ O ⁺	CH ₃ OH ⁺	310	CNSi	SiCN	B100
CH ₄ OSi	HCH ₃ SiO	363	CNSi	SiNC	B101
CH ₄ OSi	<i>t</i> -CH ₃ OSiH	B338	CNSn	SnCN	B101
CH ₄ OSi	<i>c</i> -CH ₃ OSiH	B338	CNSn ⁻	SnCN ⁻	B115
CH ₄ O ₂	CH ₂ (OH) ₂	A325	CNSr	SrNC	62,B59
CH ₄ O ₃	HO ₂ CH ₂ OH	379	CNTI	TiCN	B80
CH ₄ S ⁺	CH ₃ SH ⁺	310,B312	CNTI	TiNC	B80
CH ₄ Si ⁺	CH ₂ =SiH ₂ ⁺	308	CN ₂	NCN	70,A180,B113
CH ₄ Si	CH ₂ =SiH ₂	309,A310,B311	CN ₂	CNN	71,B114
CH ₄ Si	CH ₃ SiH	309	CN ₂ ⁻	NCN ⁻	B121
CH ₄ Zn	CH ₃ ZnH	A309	CN ₂ ⁻	CNN ⁻	B126
CH ₅ ⁺	CH ₅ ⁺	B308	CN ₂ O ⁺	ONCN ⁺	186
CH ₅ BO	H ₂ B=OCH ₃	374	CN ₂ O	NCNO	188,B229
CH ₅ BS	H ₂ BSCH ₃	A335	CN ₂ O	NOCN	B228
CH ₅ Ga	CH ₃ GaH ₂	B333	CN ₂ O	CNNO	B229
CH ₅ GaN ⁺	GaNH ₂ CH ₃ ⁺	B350	CN ₂ S ⁺	NSCN ⁺	A249
CH ₅ GaN	GaNH ₂ CH ₃	B350	CN ₂ S	NSCN	A252
CH ₆ BN	H ₂ B=NHCH ₃	398	CN ₄ ⁺	N ₃ CN ⁺	278
CH ₆ OSi	CH ₃ SiH ₂ OH	416	CNbO	CNbO	B65
CHfO	HfCO	B65	CNbO	NbCO	B65
CIN ⁺	ICN ⁺	81	CNbO ⁻	NbCO ⁻	B85
CIN	INC	A191	CNbO ₂	ONbCO (I)	B220
CINO ⁺	INCO ⁺	193	CNbO ₂	ONbCO (II)	B220
CINO	INCO	198	CNiO ⁺	NiCO ⁺	B61
CINS ⁺	ISCN ⁺	195	CNiO	NiCO	A167,B67
CINS	ISCN	199	CNiO ⁻	NiCO ⁻	A171,B86
Cl ₂	Cl ₂	B140	CNiO ₂	ONiCO	B221
Cl ₂ ⁻	Cl ₂ ⁻	B145	CNiO ₂ ⁻	ONiCO ⁻	B224
Cl ₃	Cl ₃	218	CNiO ₂ ⁻	NiCO ₂ ⁻	B224
CInN	InCN	B79	COOs ⁺	OsCO ⁺	B61
CInN	InNC	B79	COOs	OsCO	B66
CInO	InCO	B98	COOs ⁻	OsCO ⁻	B85
CIrO ⁺	IrCO ⁺	B61	COP	PCO	77
CiRO	IrCO	B67	COPd ⁺	PdCO ⁺	B61
CiRO ⁻	IrCO ⁻	B86	COPd	PdCO	A167,B67
CMgN	MgCN	A164	COPd ⁻	PdCO ⁻	B86
CMgN	MgNC	62,A164,B59	COPt ⁺	PtCO ⁺	B62
CMnO	MnCO	B66	COPt	PtCO	A167,B67
CMnO ⁻	MnCO ⁻	B85	COPt ⁻	PtCO ⁻	B86
CMnO ₂ ⁺	OMnCO ⁺	B217	CORe ⁺	ReCO ⁺	B61
CMnO ₂	OMnCO	B221	CORe	ReCO	B66
CMnO ₂ ⁻	OMnCO ⁻	B224	CORe ⁻	ReCO ⁻	B85
CMoO	MoCO	B66	CORh ⁺	RhCO ⁺	B61

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
CORh	RhCO	B67	CO ₂ V	<i>cyc</i> -(COV)O	B220
CORh ⁻	RhCO ⁻	B86	CO ₂ W	OWCO	B221
CORu ⁺	RuCO ⁺	B61	CO ₂ Y ⁺	OYCO ⁺	B216
CORu	RuCO	B66	CO ₂ Y ⁺	OYOC ⁺	B216
CORu ⁻	RuCO ⁻	B85	CO ₂ Y ⁺	YOCO ⁺	B216
COS ⁺	OCS ⁺	78,A185,B124	CO ₂ Y	OYCO	B219
COS ⁻	OCS ⁻	B135	CO ₂ Y	<i>cyc</i> -(YOC)O	B219
COSc ⁺	ScCO ⁺	B60	CO ₂ Y	<i>cyc</i> -(COY)O	B219
COSc	ScCO	A166,B64	CO ₂ Zn ⁻	OZnCO ⁻	B225
COSc ⁻	ScCO ⁻	B84	CO ₂ Zr	OZrCO	B220
COSi	SiCO	70,B112	CO ₃	CO ₃	199
COSi ⁻	SiCO ⁻	B121	CO ₃ ⁻	CO ₃ ⁻	203
COSn	SnCO	A180,B112	CO ₃ Ti	O ₂ TiCO	A296
COSn ⁻	SnCO ⁻	B121	CO ₃ U	O ₂ UCO	A297
COTa	TaCO	A166,B65	CO ₄ ⁻	CO ₄ ⁻	287,B300
COTa ⁻	CTaO ⁻	B85	CS ₂ ⁺	CS ₂ ⁺	79,A186,B125
COTa ⁻	TaCO ⁻	B85	CS ₂	<i>cyc</i> -CS ₂	B131
COTe	OTe	B131	CS ₂ ⁻	CS ₂ ⁻	A194,B135
COTh	ThCO	B68	CSi ₂	Si ₂ C	67
COTh	CThO	B68	CSi ₃	Si ₃ C	181
COTh ⁻	CThO ⁻	B86	C ₂ AgO ₂ ⁺	Ag(CO) ₂ ⁺	B282
COTi ⁺	TiCO ⁺	B60	C ₂ AgO ₂	Ag(CO) ₂	B286
COTi	TiCO	A166,B65	C ₂ Al	AICC	B63
COTi ⁻	TiCO ⁻	B84	C ₂ Al ⁻	AICC ⁻	B78
COU	CUO	A168,B68	C ₂ AlO ₂	Al(CO) ₂	B291
COU	UCO	A168,B68	C ₂ AlO ₂ ⁻	Al(CO) ₂ ⁻	B295
COU ⁻	CUO ⁻	B87	C ₂ Al ₂	AICCAI	A242,B207
COU ⁻	UCO ⁻	B87	C ₂ Al ₂ ⁻	AICCAI ⁻	B210
COV	VCO	A166,B65	C ₂ Al ₂ ⁻	<i>cyc</i> -Al ₂ C ₂ ⁻	B210
COV ⁻	VCO ⁻	B84	C ₂ AuO ₂ ⁺	Au(CO) ₂ ⁺	B282
COW	WCO	B66	C ₂ AuO ₂	Au(CO) ₂	B287
COY ⁺	YCO ⁺	B60	C ₂ B	<i>cyc</i> -BCC	62,A164,B62
COY	YCO	B64	C ₂ B ⁻	BCC ⁻	B78
COZr	ZrCO	B65	C ₂ BFN ₂ O ₂	FB(NCO) ₂	A341
CO ₂ ⁺	CO ₂ ⁺	77,A185,B123	C ₂ B ₂	BCCB	B207
CO ₂ ⁻	CO ₂ ⁻	90,A193,B134	C ₂ BeO ₂	Be(CO) ₂	A293
CO ₂ Sc ⁺	OScCO ⁺	B216	C ₂ BrCl ⁺	ClCCBr ⁺	191
CO ₂ Sc ⁺	OScOC ⁺	B216	C ₂ BrF ₃	CF ₃ CBr	354
CO ₂ Sc ⁺	ScOCO ⁺	B216	C ₂ BrN	BrCNC	B230
CO ₂ Sc	OScCO	B219	C ₂ BrN	BrCCN	189,B230
CO ₂ Sc	<i>cyc</i> -(ScOC)O	B219	C ₂ Br ₂ ⁺	C ₂ Br ₂ ⁺	191
CO ₂ Sc	<i>cyc</i> -(COSc)O	B219	C ₂ ClF ₃ ⁺	CF ₂ =CFCl ⁺	352
CO ₂ Ta	OTaCO	B220	C ₂ ClF ₃	CF ₃ CCl	354
CO ₂ Ta ⁻	OTaCO ⁻	B224	C ₂ ClN	ClCNC	B229
CO ₂ Th ⁺	OThCO ⁺	B217	C ₂ ClN	ClCCN	189,B230
CO ₂ Th	OThCO	B221	C ₂ Cl ₂ ⁺	C ₂ Cl ₂ ⁺	190
CO ₂ Ti ⁺	OTiCO ⁺	B216	C ₂ Cl ₂ F ₂ ⁺	CF ₂ =CCl ₂ ⁺	352
CO ₂ Ti ⁺	OTiOC ⁺	B217	C ₂ Cl ₂ O ⁺	Cl ₂ CCO ⁺	285
CO ₂ Ti	OTiCO	A248,B219	C ₂ Cl ₂ O	Cl ₂ CCO	285
CO ₂ Ti	<i>cyc</i> -(COTi)O	B220	C ₂ Cl ₂ O ₂ ⁺	<i>t</i> -(ClCO) ₂ ⁺	348
CO ₂ U ⁺	OUCO ⁺	B217	C ₂ Cl ₄ ⁺	C ₂ Cl ₄ ⁺	353
CO ₂ U	OUCO	A248,B222	C ₂ Co	CoCC	B57
CO ₂ V ⁺	OVCO ⁺	B217	C ₂ Co ⁻	CoCC ⁻	B58
CO ₂ V ⁺	OVOC ⁺	B217	C ₂ CoO ₂ ⁺	Co(CO) ₂ ⁺	B281
CO ₂ V	OVCO	B220	C ₂ CoO ₂	Co(CO) ₂	B285

Formula	Structure/Name	References	Formula	Structure/Name	References
C ₂ CoO ₂ ⁻	Co(CO) ₂ ⁻	B290	C ₂ HN	HCNC	B185
C ₂ Cr	CrCC	B57	C ₂ HN	<i>cyc</i> -HC=NC:	B186
C ₂ Cr ⁻	CrCC ⁻	B58	C ₂ HNO	HCOCN	269,A287
C ₂ CuO ₂ ⁺	Cu(CO) ₂ ⁺	B282	C ₂ HNO ₂	O=C=C=NOH	A317
C ₂ CuO ₂	Cu(CO) ₂	B286	C ₂ HNPd	PdCCNH	B271
C ₂ CuO ₂ ⁻	Cu(CO) ₂ ⁻	B291	C ₂ HNPd ⁻	PdCCNH ⁻	B271
C ₂ F ₂ ⁺	C ₂ F ₂ ⁺	190	C ₂ HNS ⁺	HCSCN ⁺	A287
C ₂ F ₂	FCCF	197	C ₂ HNS	HCSCN	A288
C ₂ F ₂	F ₂ C=C:	197,B232	C ₂ HN ₂ ⁺	HNCCN ⁺	268,B273
C ₂ F ₂ ⁻	F ₂ C=C: ⁻	203	C ₂ HNa	NaCCH	B180
C ₂ F ₂ N ₂	F ₂ C=C=N=N	347	C ₂ HO	HCCO	156,A234,B188
C ₂ F ₂ O	F ₂ CCO	B299	C ₂ HO ⁻	HCCO ⁻	B189
C ₂ F ₂ O ₂ ⁺	<i>t</i> -(FCO) ₂ ⁺	348	C ₂ HP	HCCP	B187
C ₂ F ₃	C ₂ F ₃	287	C ₂ HPd	PdCCH	B181
C ₂ F ₃ O	CF ₃ CO	A322	C ₂ HPd ⁻	PdCCH ⁻	B182
C ₂ F ₄ ⁺	C ₂ F ₄ ⁺	351,B331	C ₂ HS	HCCS	157,A234
C ₂ F ₄	CF ₃ CF	354	C ₂ HS	HSCC	157
C ₂ F ₅	C ₂ F ₅	371	C ₂ HSi	HCCSi	A233,B184
C ₂ F ₅ P	CF ₂ =PCF ₃	382	C ₂ HSr	SrCCH	155,A231,B181
C ₂ Fe	FeCC	A164,B57	C ₂ HYb ⁺	YbCCH ⁺	B180
C ₂ Fe ⁻	FeCC ⁻	A164,B58	C ₂ HYb	YbCCH	B181
C ₂ FeO ₂ ⁺	Fe(CO) ₂ ⁺	B281	C ₂ H ₂ ⁺	HCCH ⁺	133,A221,B164
C ₂ FeO ₂	Fe(CO) ₂	B285	C ₂ H ₂	H ₂ C=C:	137
C ₂ FeO ₂ ⁻	Fe(CO) ₂ ⁻	B290	C ₂ H ₂ ⁻	H ₂ C=C: ⁻	139
C ₂ DFe	FeCCD	A232	C ₂ H ₂ Al	<i>cyc</i> -HC=CHAl	A276
C ₂ GaO ₂	Ga(CO) ₂	B292	C ₂ H ₂ Al	HCCAlH	A276
C ₂ GeO ₂	Ge(CO) ₂	B295	C ₂ H ₂ Al	(C ₂ H ₂)Al	A277
C ₂ H ⁺	HCC ⁺	B23	C ₂ H ₂ B	<i>cyc</i> -HC=CHB	A276
C ₂ H	HCC	32,A142,B28	C ₂ H ₂ B	<i>cyc</i> -HC=CBH	A276
C ₂ H ⁻	HCC ⁻	37,B32	C ₂ H ₂ B	HCCBH	A276
C ₂ HAl	HCCAl	A233	C ₂ H ₂ Be	HBeCCH	A275
C ₂ HB	HBCC	A232,B183	C ₂ H ₂ Br	CHBr=CH	257
C ₂ HB	HCCB	A232	C ₂ H ₂ ClF ⁺	CH ₂ =CFCl ⁺	334
C ₂ HB	<i>cyc</i> -HBC ₂	A232	C ₂ H ₂ ClF ⁺	<i>c</i> -CHF=CHCl ⁺	334
C ₂ HBe	BeCCH	A231	C ₂ H ₂ ClF ⁺	<i>t</i> -CHF=CHCl ⁺	334
C ₂ HBr ⁺	HCCBr ⁺	158	C ₂ H ₂ CIN ⁺	CH ₂ CICN ⁺	331
C ₂ HCa	CaCCH	155,A231,B181	C ₂ H ₂ ClO	<i>t</i> -CHClCHO	B322
C ₂ HCl ⁺	HCCCl ⁺	158	C ₂ H ₂ Cl ₂ ⁺	CH ₂ =CCl ₂ ⁺	335
C ₂ HClFO	CHClCFO	B323	C ₂ H ₂ Cl ₂ ⁺	<i>c</i> -CHCl=CHCl ⁺	335
C ₂ HClF ₂ ⁺	CHCl=CF ₂ ⁺	340	C ₂ H ₂ Cl ₂ ⁺	<i>t</i> -CHCl=CHCl ⁺	336
C ₂ HClO	HCICCO	269,B273	C ₂ H ₂ F	CH ₂ =CF	B267
C ₂ HCl ₃ ⁺	CHCl=CCl ₂ ⁺	341	C ₂ H ₂ F	<i>c</i> -CHF=CH	257,B267
C ₂ HF ⁺	HCCF ⁺	158	C ₂ H ₂ FN ⁺	CH ₂ FCN ⁺	330
C ₂ HF	HFC=C:	162	C ₂ H ₂ FO	CH ₂ CFO	A316,B321
C ₂ HF ⁻	HFC=C: ⁻	168	C ₂ H ₂ FO ⁻	CH ₂ COF ⁻	337
C ₂ HFO	HFCCO	269,A288	C ₂ H ₂ F ₂ ⁺	CH ₂ =CF ₂ ⁺	332
C ₂ HF ₂ O	<i>c</i> -CHF ₂ CFO	B323	C ₂ H ₂ F ₂ ⁺	<i>c</i> -CHF=CHF ⁺	333
C ₂ HF ₂ O	<i>t</i> -CHF ₂ CFO	B323	C ₂ H ₂ F ₂ ⁺	<i>t</i> -CHF=CHF ⁺	333
C ₂ HF ₃ ⁺	CHF=CF ₂ ⁺	340	C ₂ H ₂ Fe	HFeCCH	245
C ₂ HFe ⁻	FeCCH ⁻	A232	C ₂ H ₂ Ga	(C ₂ H ₂)Ga	A277
C ₂ HI ⁺	HCCI ⁺	159	C ₂ H ₂ In	(C ₂ H ₂)In	A277
C ₂ HK	KCCH	B180	C ₂ H ₂ Li	LiC ₂ H ₂	244
C ₂ HLi	LiCCH	B180	C ₂ H ₂ N	H ₂ CCN	250,A280,B264
C ₂ HMg	MgCCH	A231,B180	C ₂ H ₂ N	H ₂ CNC	250
C ₂ HN	HCCN	156,A234,B185	C ₂ H ₂ N ⁻	H ₂ CCN ⁻	252

Formula	Structure/Name	References	Formula	Structure/Name	References
C ₂ H ₂ N ⁻	H ₂ CNC ⁻	252	C ₂ H ₃ O ₃	CH ₃ COO ₂	380,A339
C ₂ H ₂ N ₂ ⁺	HN=CHCN ⁺	327	C ₂ H ₃ P ⁺	CH ₃ CP ⁺	314
C ₂ H ₂ N ₂ ⁺	H ₂ C=NCN ⁺	327	C ₂ H ₃ P	CH ₃ CP	316
C ₂ H ₂ N ₂	<i>c</i> -HN=CHCN	327	C ₂ H ₃ Si	SiH ₂ CCH	B314
C ₂ H ₂ N ₂	<i>t</i> -HN-CHCN	328	C ₂ H ₄ ⁺	C ₂ H ₄ ⁺	307
C ₂ H ₂ N ₂	H ₂ C=NCN	328	C ₂ H ₄ Al	AlC ₂ H ₄	361
C ₂ H ₂ N ₂ O ₂	HON=CHCNO	381	C ₂ H ₄ B	<i>cyc</i> -BHCHCH ₂	B334
C ₂ H ₂ N ₂ O ₂	HON=CHNCO	381	C ₂ H ₄ B	H ₂ BCCH ₂	B334
C ₂ H ₂ Ni	(C ₂ H ₂)Ni	245	C ₂ H ₄ F	FCH ₂ CH ₂	364,A325
C ₂ H ₂ Ni	NiC=CH ₂	245	C ₂ H ₄ Fe	HFeC ₂ H ₃	360
C ₂ H ₂ O ⁺	H ₂ CCO ⁺	250,A280	C ₂ H ₄ FeO	CH ₂ =CHFeOH	378
C ₂ H ₂ O	HCCOH	255	C ₂ H ₄ FeO	<i>cyc</i> -C ₂ H ₄ OFe	378
C ₂ H ₂ O	<i>cyc</i> -H ₂ COC:	B266	C ₂ H ₄ In	InC ₂ H ₄	361
C ₂ H ₂ OS	CHOCHS	330	C ₂ H ₄ Li	C ₂ H ₄ Li	360
C ₂ H ₂ O ₂ ⁺	(HCO) ₂ ⁺	329	C ₂ H ₄ Li ₂	1,2-C ₂ H ₄ Li ₂	377
C ₂ H ₂ O ₂	HOC≡COH	A315	C ₂ H ₄ N ⁺	CH ₃ CNH ⁺	362,B336
C ₂ H ₂ O ₃	(HCO) ₂ O	368	C ₂ H ₄ O ⁺	CH ₃ CHO ⁺	B336
C ₂ H ₂ P	H ₂ CCP	B265	C ₂ H ₄ O ⁺	CH ₂ =CHOH ⁺	362,A324
C ₂ H ₂ S ⁺	H ₂ CCS ⁺	253	C ₂ H ₄ O	<i>syn</i> -CH ₂ =CHOH	363,B337
C ₂ H ₂ S	H ₂ CCS	255,A282	C ₂ H ₄ O	<i>anti</i> -CH ₂ =CHOH	B338
C ₂ H ₂ S	HCCSH	256	C ₂ H ₄ O ₃	1,2,3- <i>cyc</i> -C ₂ H ₄ O ₃	422,A366
C ₂ H ₂ S	<i>cyc</i> -C ₂ H ₂ S	256	C ₂ H ₄ O ₃	1,2,4- <i>cyc</i> -C ₂ H ₄ O ₃	422,A367
C ₂ H ₂ S ₂	HS-CH=C=S	331	C ₂ H ₄ O ₃	<i>c</i> -HCOOCH ₂ OH	423
C ₂ H ₂ Se	H ₂ CCSe	A283	C ₂ H ₄ O ₃	<i>t</i> -HCOOCH ₂ OH	423
C ₂ H ₂ Si	<i>cyc</i> -HC=CHSi	A278	C ₂ H ₄ S ⁺	CH ₃ CHS ⁺	363
C ₂ H ₂ Si	HCCSiH	A279	C ₂ H ₄ S	CH ₃ CHS	363,A324
C ₂ H ₂ Si	H ₂ C=C=Si:	A279	C ₂ H ₄ Si	<i>cyc</i> -CHCHSiH ₂	A324
C ₂ H ₂ Si	<i>cyc</i> -H ₂ SiCC	A249	C ₂ H ₄ Si	<i>cyc</i> -CH ₂ CH ₂ Si	B335
C ₂ H ₃ ⁺	C ₂ H ₃ ⁺	232,A268	C ₂ H ₄ Si	H ₂ CCHSiH	B336
C ₂ H ₃	C ₂ H ₃	232,A269,B255	C ₂ H ₄ Si	H ₂ CCSiH ₂	B336
C ₂ H ₃ As	HCCAsH ₂	A313	C ₂ H ₄ Si ₂	H ₃ Si(<i>cyc</i> -CSiCH)	B357
C ₂ H ₃ As	CH ₃ CA _s	A313	C ₂ H ₄ Si ₂	HSiCCSiH ₃	B358
C ₂ H ₃ B	<i>cyc</i> -(CH) ₂ BH	B312	C ₂ H ₄ Si ₂	HCCSiSiH ₃	B358
C ₂ H ₃ B	H ₂ BCCH	B313	C ₂ H ₅ ⁺	C ₂ H ₅ ⁺	359
C ₂ H ₃ B	HBCCH ₂	B313	C ₂ H ₅	C ₂ H ₅	359,A323,B334
C ₂ H ₃ CaO ₂	CH ₃ COOCa	379	C ₂ H ₅ B	H ₂ BC ₂ H ₃	B349
C ₂ H ₃ Cl ⁺	CH ₂ =CHCl ⁺	319	C ₂ H ₅ B	CH ₃ B=CH ₂	B349
C ₂ H ₃ ClO	CH ₃ OCCL	366	C ₂ H ₅ BrS	CH ₃ SBrCH ₂	418
C ₂ H ₃ F ⁺	CH ₂ =CHF ⁺	319	C ₂ H ₅ CaO	CaOC ₂ H ₅	B388
C ₂ H ₃ F ₂	CH ₃ CF ₂	A327	C ₂ H ₅ Cd	CdC ₂ H ₅	B349
C ₂ H ₃ N ⁺	CH ₃ CN ⁺	312	C ₂ H ₅ ClS	CH ₃ SClCH ₂	418
C ₂ H ₃ N ⁺	CH ₃ NC ⁺	313	C ₂ H ₅ IO	C ₂ H ₅ IO	A364
C ₂ H ₃ N ⁺	CH ₂ =C=NH ⁺	313	C ₂ H ₅ IO	C ₂ H ₅ OI	A364
C ₂ H ₃ N ⁺	<i>cyc</i> -(HC=NCH ₂) ⁺	A311	C ₂ H ₅ IO ₂	C ₂ H ₅ IO ₂	A364
C ₂ H ₃ N	CH ₂ =C=NH	316,A312	C ₂ H ₅ IS	CH ₃ SICH ₂	418
C ₂ H ₃ N	HCNCH ₂	A312	C ₂ H ₅ N	<i>c</i> -CH ₃ CH=NH	374,A336
C ₂ H ₃ N	<i>cyc</i> -(HC=NCH ₂)	A313	C ₂ H ₅ N	<i>t</i> -CH ₃ CH=NH	374
C ₂ H ₃ NO	CH ₃ CNO	366,B338	C ₂ H ₅ N	C ₂ H ₃ NH ₂	375,A336,B352
C ₂ H ₃ NO	CH ₃ OCN	A326	C ₂ H ₅ N	H ₂ C=NCH ₃	375
C ₂ H ₃ NO	HOCH ₂ CN	366	C ₂ H ₅ NO ₂	<i>c</i> -CH ₃ CH(NO)OH	424
C ₂ H ₃ NO ₂	CH ₃ ONCO	380	C ₂ H ₅ NO ₄	C ₂ H ₅ OONO ₂	425
C ₂ H ₃ O	CH ₃ CO	317	C ₂ H ₅ O ⁺	CH ₃ CHOH ⁺	B352
C ₂ H ₃ O	CH ₂ CHO	317,A313,B317	C ₂ H ₅ O ⁺	CH ₃ CH ₂ O ⁺	B352
C ₂ H ₃ O ⁻	CH ₂ CHO ⁻	320,A314,B319	C ₂ H ₅ O	C ₂ H ₅ O	376,A337,B353
C ₂ H ₃ O ₂ Sr	CH ₃ COOSr	380	C ₂ H ₅ O	CH ₃ CHOH	376

Formula	Structure/Name	References	Formula	Structure/Name	References
C ₂ H ₅ O	HOCH ₂ CH ₂	376,A337	C ₂ N ₂ ⁺	NCCN ⁺	183
C ₂ H ₅ O ⁻	C ₂ H ₅ O ⁻	B353	C ₂ N ₂ ⁺	CNCN ⁺	184
C ₂ H ₅ O ₂	C ₂ H ₅ O ₂	420,B393	C ₂ N ₂	CNCN	185,A248
C ₂ H ₅ O ₂ ⁻	C ₂ H ₅ O ₂ ⁻	B394	C ₂ N ₂	CNNC	185
C ₂ H ₅ O ₃	HOCH ₂ CH ₂ O ₂	424	C ₂ N ₂ O ⁺	NCCNO ⁺	A295
C ₂ H ₅ S ⁺	CH ₃ SCH ₂ ⁺	A336	C ₂ N ₂ O ⁺	NCNCO ⁺	279
C ₂ H ₅ S ⁺	C ₂ H ₅ S ⁺	B352	C ₂ N ₂ O	NCCNO	283,A297,B295
C ₂ H ₅ S	C ₂ H ₅ S	376,A337,B353	C ₂ N ₂ O	NCNCO	283
C ₂ H ₅ Zn ⁺	ZnC ₂ H ₅ ⁺	B349	C ₂ N ₂ O ₂ ⁺	ONCCNO ⁺	A320
C ₂ H ₅ Zn	ZnC ₂ H ₅	B349	C ₂ N ₂ O ₂	ONCCNO	A321
C ₂ H ₆ BN	CH ₃ BNCH ₃	B385	C ₂ N ₂ O ₂	OCNNCO	B329
C ₂ H ₆ Ge	(CH ₃) ₂ Ge	400	C ₂ N ₂ S ⁺	S(CN) ₂ ⁺	280
C ₂ H ₆ GeO	(CH ₃) ₂ Ge=O	B391	C ₂ N ₂ S ⁺	NCNCS ⁺	280
C ₂ H ₆ GeS	(CH ₃) ₂ Ge=S	B391	C ₂ N ₂ S	NCNCS	284
C ₂ H ₆ Hg	C ₂ H ₅ HgH	A357	C ₂ N ₂ S ₂ ⁺	(SCN) ₂ ⁺	346
C ₂ H ₆ N ₂ O	CH ₃ NHCH ₂ NO(A)	424	C ₂ N ₂ Se ⁺	Se(CN) ₂ ⁺	281
C ₂ H ₆ N ₂ O	CH ₃ NHCH ₂ NO(B)	424	C ₂ N ₂ Sn	Sn(CN) ₂	B289
C ₂ H ₆ OSi	(CH ₃) ₂ SiO	416	C ₂ N ₂ Sn ⁻	Sn(CN) ₂ ⁻	B292
C ₂ H ₆ OSi	CH ₂ =Si(OH)CH ₃	417	C ₂ Nb	NbCC	B56
C ₂ H ₆ OSi	CH ₃ OSiCH ₃	417	C ₂ Nb ⁻	NbCC ⁻	B58
C ₂ H ₆ OSi	CH ₃ OSiH=CH ₂	417	C ₂ NbO ₂	Nb(CO) ₂	B284
C ₂ H ₆ O ₂ Si	(CH ₃ O) ₂ Si	421	C ₂ NbO ₂	ONbCCO	B284
C ₂ H ₆ O ₂ Si	CH ₃ SiOOCH ₃	421	C ₂ NbO ₂ ⁻	Nb(CO) ₂ ⁻	B289
C ₂ H ₆ O ₂ Si	CH ₃ SiHOHCHO	A365	C ₂ Nb ₃ ⁺	Nb ₃ C ₂ ⁺	B277
C ₂ H ₆ O ₃	HOCH ₂ CH ₂ OOH	424	C ₂ Nb ₃	Nb ₃ C ₂	B277
C ₂ H ₆ O ₄	CH ₃ O ₄ CH ₃	424	C ₂ NiO ₂ ⁺	Ni(CO) ₂ ⁺	B282
C ₂ H ₆ Si	(CH ₃) ₂ Si	399	C ₂ NiO ₂	Ni(CO) ₂	B286
C ₂ H ₆ Si	CH ₃ SiH=CH ₂	399,B385	C ₂ NiO ₂ ⁻	Ni(CO) ₂ ⁻	B291
C ₂ H ₆ Sn	(CH ₃) ₂ Sn	400	C ₂ O	CCO	69,A179,B111
C ₂ H ₆ Zn ⁺	(CH ₃) ₂ Zn ⁺	A357	C ₂ O ⁻	CCO ⁻	76,A183,B121
C ₂ H ₆ Zn	C ₂ H ₅ ZnH	B385	C ₂ OS	OCCS	B228
C ₂ H ₇ ⁺	C ₂ H ₇ ⁺	383	C ₂ O ₂ ⁺	<i>t</i> -OCCO ⁺	186,B225
C ₂ H ₇ ⁺	<i>br</i> -C ₂ H ₇ ⁺	383	C ₂ O ₂ ⁻	<i>t</i> -OCCO ⁻	190,B232
C ₂ H ₇ BO	H ₂ B=OC ₂ H ₅	408	C ₂ O ₂ Os	Os(CO) ₂	B285
C ₂ H ₇ BS	H ₂ BSC ₂ H ₅	A361	C ₂ O ₂ Os ⁻	Os(CO) ₂ ⁻	B290
C ₂ H ₇ Ga	(CH ₃) ₂ GaH	B385	C ₂ O ₂ Pd ⁺	Pd(CO) ₂ ⁺	B282
C ₂ H ₈ BN	H ₂ B=N(CH ₃) ₂	399	C ₂ O ₂ Pd	Pd(CO) ₂	B286
C ₂ H ₈ OSi	(CH ₃) ₂ SiHOH	417	C ₂ O ₂ Pd ⁻	Pd(CO) ₂ ⁻	B291
C ₂ HfO ₂	Hf(CO) ₂	B284	C ₂ O ₂ Pt ⁺	Pt(CO) ₂ ⁺	B282
C ₂ HfO ₂	OHfCCO	B284	C ₂ O ₂ Pt	Pt(CO) ₂	B286
C ₂ HfO ₂ ⁻	Hf(CO) ₂ ⁻	B289	C ₂ O ₂ Pt ⁻	Pt(CO) ₂ ⁻	B291
C ₂ I ₂ ⁺	C ₂ I ₂ ⁺	192	C ₂ O ₂ Re	Re(CO) ₂	B285
C ₂ InO ₂	In(CO) ₂	B292	C ₂ O ₂ Re ⁻	Re(CO) ₂ ⁻	B290
C ₂ IrO ₂ ⁺	Ir(CO) ₂ ⁺	B282	C ₂ O ₂ Rh ⁺	Rh(CO) ₂ ⁺	B281
C ₂ IrO ₂	Ir(CO) ₂	B286	C ₂ O ₂ Rh	Rh(CO) ₂	B286
C ₂ IrO ₂ ⁻	Ir(CO) ₂ ⁻	B290	C ₂ O ₂ Rh ⁻	Rh(CO) ₂ ⁻	B290
C ₂ Mn	MnCC	B57	C ₂ O ₂ Ru	Ru(CO) ₂	B285
C ₂ Mn ⁻	MnCC ⁻	B58	C ₂ O ₂ Ru ⁻	Ru(CO) ₂ ⁻	B290
C ₂ MnO ₂	Mn(CO) ₂	B285	C ₂ O ₂ Sc ⁺	Sc(CO) ₂ ⁺	B281
C ₂ MnO ₂ ⁻	Mn(CO) ₂ ⁻	B289	C ₂ O ₂ Sc	Sc(CO) ₂	B283
C ₂ N ⁺	CCN ⁺	B84	C ₂ O ₂ Si	Si(CO) ₂	283,B295
C ₂ N ⁺	CNC ⁺	67,B84	C ₂ O ₂ Sn	Sn(CO) ₂	B295
C ₂ N	CCN	68,A173,B99	C ₂ O ₂ Ta	Ta(CO) ₂	B284
C ₂ N	CNC	69,B100	C ₂ O ₂ Ta	OTaCCO	B285
C ₂ NP	NCCP	B223	C ₂ O ₂ Ta ⁻	Ta(CO) ₂ ⁻	B289

Formula	Structure/Name	References	Formula	Structure/Name	References
C ₂ O ₂ Th	Th(CO) ₂	B287	C ₃ Fe ⁻	<i>cyc</i> -FeC ₃ ⁻	A242,B209
C ₂ O ₂ Th	OThCCO	B287	C ₃ Ge ₂	GeCCCCGe	B281
C ₂ O ₂ Th ⁻	Th(CO) ₂ ⁻	B291	C ₃ H	HC ₃	156,A233,B183
C ₂ O ₂ Ti	Ti(CO) ₂	B283	C ₃ H	<i>cyc</i> -HC ₃	156,A233
C ₂ O ₂ Ti ⁻	Ti(CO) ₂ ⁻	B289	C ₃ HCl	<i>cyc</i> -C ₃ HCl	A288
C ₂ O ₂ U	U(CO) ₂	B287	C ₃ HCl	HCCCCl	A288
C ₂ O ₂ U	OUCCO	B287	C ₃ HCl	HCIC=C=C:	A288
C ₂ O ₂ U ⁻	U(CO) ₂ ⁻	B291	C ₃ HFe	FeC=C=CH	A286
C ₂ O ₂ V	V(CO) ₂	B284	C ₃ HFe	FeCCCH	A286
C ₂ O ₂ V ⁻	V(CO) ₂ ⁻	B289	C ₃ HFe ⁻	FeC=C=CH ⁻	A286
C ₂ O ₂ Zr	Zr(CO) ₂	B284	C ₃ HFe ⁻	FeCCCH ⁻	A286
C ₂ O ₂ Zr	OZrCCO	B284	C ₃ HN ⁺	HCCCN ⁺	266,B272
C ₂ O ₂ Zr ⁻	Zr(CO) ₂ ⁻	B289	C ₃ HN ⁺	HCCNC ⁺	267
C ₂ O ₃ ⁻	C ₂ O ₃ ⁻	B298	C ₃ HN	HNCCC	267,B273
C ₂ O ₄ ⁺	O ₂ CCO ₂ ⁺	B329	C ₃ HN	HCCNC	267,B273
C ₂ O ₄ ⁻	O ₂ CCO ₂ ⁻	B330	C ₃ HNO	NCCH=C=O	B323
C ₂ S	CCS	70,B113	C ₃ HF ₃ S	CF ₃ H(<i>cyc</i> -CCS)	381
C ₂ S ₂	SCCS	188	C ₃ HO	HCCCO	268,A286
C ₂ S ₃	S(CS) ₂	284	C ₃ HS	HCCCS	A287
C ₂ S ₄ ⁺	S ₂ CCS ₂ ⁺	B330	C ₃ H ₂ ⁺	<i>cyc</i> -C ₃ H ₂ ⁺	248
C ₂ Sc	ScCC	B56	C ₃ H ₂	HCCCH	249,A278
C ₂ Sc ⁻	ScCC ⁻	B58	C ₃ H ₂	<i>cyc</i> -C ₃ H ₂	248,B262
C ₂ Si	SiCC	66,A170	C ₃ H ₂	H ₂ C=C=C:	248,A277,B262
C ₂ Si ₂	Si ₂ C ₂	181,A245	C ₃ H ₂	HCCH=C:	249
C ₂ Si ₃	Si ₃ C ₂	A293	C ₃ H ₂ ⁻	H ₂ C=C=C: ⁻	A280,B264
C ₂ Ti	TiCC	B56	C ₃ H ₂ N ⁺	HCCCNH ⁺	326,A315,B321
C ₂ Ti ⁻	TiCC ⁻	B58	C ₃ H ₂ N ₂	<i>cyc</i> -(HC=CHN=C=N)	B340
C ₂ V	VCC	B56	C ₃ H ₂ N ₂	(<i>cyc</i> -HC=CHN)CN	B340
C ₂ V ⁻	VCC ⁻	B58	C ₃ H ₂ N ₂	H ₂ C=C=NCN	B340
C ₂ Xe	XeC ₂	B133	C ₃ H ₂ N ₂	<i>cyc</i> -(C=NCHNCH)	B341
C ₂ Y	YCC	A164	C ₃ H ₂ N ₂	HC-N=CHCN	B341
C ₂ Y ₃ ⁺	Y ₃ C ₂ ⁺	B277	C ₃ H ₂ N ₂	<i>cyc</i> -(HC=NCH)CN	B341
C ₂ Y ₃	Y ₃ C ₂	B277	C ₃ H ₂ N ₂	HNC=C=CHCN	B341
C ₃	C ₃	65,A168,B81	C ₃ H ₂ O ⁺	H ₂ C=C=C=O ⁺	327
C ₃ ⁻	C ₃ ⁻	68,A173,B99	C ₃ H ₂ O	HOCH=C=C:	328,B321
C ₃ B	BCCC	B210	C ₃ H ₂ O	H ₂ C=C=C=O	328
C ₃ BrN ⁺	BrCCCN ⁺	282	C ₃ H ₂ O ₂	<i>t</i> -HCCCHOO	A327
C ₃ ClN ⁺	ClCCCN ⁺	281	C ₃ H ₂ O ₂	<i>c</i> -HCCCHOO	A327
C ₃ Cl ₂	<i>cyc</i> -C ₃ Cl ₂	A299	C ₃ H ₂ O ₂	HCC(<i>cyc</i> -CHOO)	A328
C ₃ Cl ₂	ClCCCCl	A299	C ₃ H ₂ O ₃	<i>cyc</i> -(HOC=COHC)=O	A340
C ₃ Cl ₂	Cl ₂ C=C=C:	A299	C ₃ H ₂ S	H ₂ C=C=C=S	329
C ₃ Co	<i>cyc</i> -CoC ₃	B204	C ₃ H ₂ Se	H ₂ C=C=C=Se	329
C ₃ Co ⁻	<i>cyc</i> -CoC ₃ ⁻	B209	C ₃ H ₂ Se	HCC-CHSe	329
C ₃ Cr	<i>cyc</i> -CrC ₃	B203	C ₃ H ₃ ⁺	<i>cyc</i> -C ₃ H ₃ ⁺	B313
C ₃ Cr ⁻	<i>cyc</i> -CrC ₃ ⁻	B209	C ₃ H ₃ ⁺	CH ₂ CCH ⁺	312,B313
C ₃ FN ⁺	FCCCN ⁺	281	C ₃ H ₃	CH ₂ CCH	312,A311,B314
C ₃ F ₂	<i>cyc</i> -C ₃ F ₂	A299	C ₃ H ₃ ⁻	CH ₂ CCH ⁻	A311,B316
C ₃ F ₂	FCCCF	A299	C ₃ H ₃ ⁻	CH ₃ CC ⁻	A312
C ₃ F ₂	F ₂ C=C=C:	A299	C ₃ H ₃ Br ⁺	CH ₃ CCBr ⁺	365
C ₃ F ₂ O	F ₂ C=C=C=O	347	C ₃ H ₃ Cl ⁺	CH ₃ CCCl ⁺	365
C ₃ F ₂ O	<i>cyc</i> -(CF=CFC)=O	347	C ₃ H ₃ N ⁺	C ₂ H ₃ CN ⁺	A325
C ₃ F ₆ ⁺	C ₃ F ₆ ⁺	405	C ₃ H ₃ N	H ₂ C=(<i>cyc</i> -CN=CH)	A326
C ₃ F ₇	<i>n</i> -C ₃ F ₇	405	C ₃ H ₄ ⁺	H ₂ CCCH ₂ ⁺	362,A323,B335
C ₃ F ₇	<i>i</i> -C ₃ F ₇	406	C ₃ H ₄ ⁺	CH ₃ CCH ⁺	B335
C ₃ Fe	<i>cyc</i> -FeC ₃	A242,B203	C ₃ H ₄	H ₂ CCHCH:	B335

Formula	Structure/Name	References	Formula	Structure/Name	References
C ₃ H ₄ Fe	HFeCH=C=CH ₂	B354	C ₃ Ni	<i>cyc</i> -NiC ₃	B204
C ₃ H ₄ Fe	HCCCH ₂ FeH	B354	C ₃ Ni ⁻	<i>cyc</i> -NiC ₃ ⁻	B209
C ₃ H ₄ Fe	HFeCCCH ₃	B354	C ₃ O	CCCO	186,B223
C ₃ H ₄ Fe	CH ₃ FeCCH	B355	C ₃ O ₂ ⁺	C ₃ O ₂ ⁺	279
C ₃ H ₄ Li	CH ₃ CCHLi	377	C ₃ S	CCCS	186,A249,B223
C ₃ H ₄ NO ₂	CH ₃ C(=NO)CHO	A367	C ₃ Sc	<i>cyc</i> -ScC ₃	B202
C ₃ H ₄ O	<i>cyc</i> -(H ₂ COC)=CH ₂	379	C ₃ Sc ⁻	<i>cyc</i> -ScC ₃ ⁻	B209
C ₃ H ₄ O	CH ₃ CCOH	A338	C ₃ Si	SiC ₃	B214
C ₃ H ₄ S	HCH ₃ (<i>cyc</i> -CCS)	379	C ₃ Si	<i>cyc</i> -SiC ₃ (prolate)	B215
C ₃ H ₄ Si	H ₃ C(<i>cyc</i> -CSiCH)	B356	C ₃ Si	<i>cyc</i> -SiC ₃ (oblate)	B215
C ₃ H ₄ Si	<i>cyc</i> -CH ₂ CHCHSi	B356	C ₃ Si ⁻	SiC ₃ ⁻	B218
C ₃ H ₄ Si	H ₃ CCHCSi	B356	C ₃ Si ₂	SiCCCSi	278,A293,B280
C ₃ H ₄ Si	H ₂ CCHCHSi	B356	C ₃ Si ₂ ⁻	SiCCCSi ⁻	B283
C ₃ H ₄ Si	H ₃ CCCSiH	B356	C ₃ Ti	<i>cyc</i> -TiC ₃	B202
C ₃ H ₄ Si	H ₃ CSiCCH	B357	C ₃ Ti ⁻	<i>cyc</i> -TiC ₃ ⁻	B209
C ₃ H ₄ Si	H ₂ CCCHSiH	B357	C ₃ V	<i>cyc</i> -VC ₃	B203
C ₃ H ₄ Si	H ₂ CSiHCCH	B357	C ₃ V ⁻	<i>cyc</i> -VC ₃ ⁻	B209
C ₃ H ₅ ⁺	CH ₂ CHCH ₂ ⁺	372,B350	C ₄	C ₄	181,A245,B214
C ₃ H ₅ ⁺	<i>cyc</i> -C ₃ H ₅ ⁺	372	C ₄ ⁻	C ₄ ⁻	183,A246,B217
C ₃ H ₅	CH ₂ CHCH ₂	372,A334,B350	C ₄ Br ₂ ⁺	Br(CC) ₂ Br ⁺	345
C ₃ H ₅ ⁻	CH ₂ CHCH ₂ ⁻	A335	C ₄ Cl ₂ ⁺	Cl(CC) ₂ Cl ⁺	345
C ₃ H ₅ N	H ₂ C=CH-CH=NH	408	C ₄ F ₂ ⁺	F(CC) ₂ F ⁺	344
C ₃ H ₅ N	H ₂ C=C=NCH ₃	408	C ₄ F ₄	F ₂ C=C=CFCF:	B363
C ₃ H ₅ N	H ₂ C=CHN=CH ₂	409	C ₄ F ₄	F ₂ C=(<i>cyc</i> -C ₃ F ₂)	B363
C ₃ H ₅ N	<i>cyc</i> -C ₃ H ₅ N	409	C ₄ F ₆ O	CF ₃ CCOCF ₃	418
C ₃ H ₅ O	CH ₂ COCH ₃	B390	C ₄ F ₆ O	(CF ₃) ₂ (<i>cyc</i> -CCO)	418
C ₃ H ₅ O	CH ₃ CHCHO	B390	C ₄ Fe ⁻	FeC ₄ ⁻	A292
C ₃ H ₆ O	<i>t</i> -CH ₃ C-OCH ₃	415	C ₄ H	C ₄ H	266,A286,B271
C ₃ H ₆ O	<i>c</i> -CH ₃ C-OCH ₃	415	C ₄ H ⁻	C ₄ H ⁻	B272
C ₃ H ₆ S	(CH ₃) ₂ CS	415	C ₄ HBr ⁺	H(CC) ₂ Br ⁺	339
C ₃ H ₇ ⁺	1-C ₃ H ₇ ⁺	383	C ₄ HCl ⁺	H(CC) ₂ Cl ⁺	339
C ₃ H ₇ ⁺	2-C ₃ H ₇ ⁺	383	C ₄ HF ⁺	H(CC) ₂ F ⁺	339
C ₃ H ₇	CH ₃ CH ₂ CH ₂	383	C ₄ HF ⁻	FeCCCCH ⁻	A316
C ₃ H ₇	(CH ₃) ₂ CH	383	C ₄ HI ⁺	H(CC) ₂ I ⁺	340
C ₃ H ₇ N	CH ₃ CH=CHNH ₂	409	C ₄ HN	CCCHCN	B323
C ₃ H ₇ N	C ₂ H ₃ NHCH ₃	409	C ₄ HN	HCCCCN	B323
C ₃ H ₇ O	<i>n</i> -C ₃ H ₇ O	416,B390	C ₄ HN	(<i>cyc</i> -HC ₃)CN	B323
C ₃ H ₇ O	(CH ₃) ₂ CHO	416,B391	C ₄ HO	HCCCCO	A316
C ₃ H ₇ O ⁻	(CH ₃) ₂ CHO ⁻	B391	C ₄ HS	HCCCCS	A317
C ₃ H ₇ O ₂	(CH ₃) ₂ CHO ₂	420	C ₄ HSi	SiCCCCH	B323
C ₃ H ₈ Ge	(CH ₃) ₂ Ge=CH ₂	B386	C ₄ H ₂ ⁺	C ₄ H ₂ ⁺	325
C ₃ H ₈ O ₂ Si	(CH ₃) ₂ SiOHCHO	A365	C ₄ H ₂	H ₂ CCCC:	326,A315
C ₃ H ₈ Si ⁺	(CH ₃) ₂ Si-CH ₂ ⁺	400	C ₄ H ₂ F ₂	F ₂ C=(<i>cyc</i> -C ₃ H ₂)	B359
C ₃ H ₈ Si	(CH ₃) ₂ Si=CH ₂	401,B386	C ₄ H ₂ F ₂	F ₂ C=CHCCH	B359
C ₃ H ₉ Al	(CH ₃) ₃ Al	400	C ₄ H ₂ F ₂	F ₂ C=C=C=CH ₂	B359
C ₃ H ₉ NSi	(CH ₃) ₃ SiN	410	C ₄ H ₂ N	H ₂ CCCCN	B340
C ₃ IN ⁺	ICCCN ⁺	282	C ₄ H ₂ N ₂ O	<i>cyc</i> -(N=CHN=CHC) =C=O	B393
C ₃ Mn	<i>cyc</i> -MnC ₃	B203	C ₄ H ₂ Si	HSiCCCCH	B339
C ₃ Mn ⁻	<i>cyc</i> -MnC ₃ ⁻	B209	C ₄ H ₂ Si	Si(CCH) ₂	B339
C ₃ N	CCCN	183,A247	C ₄ H ₄ ⁺	<i>cyc</i> -C ₄ H ₄ ⁺	377,A337
C ₃ N ₂	C(CN) ₂	278,A295	C ₄ H ₄ ⁺	H ₂ C=(<i>cyc</i> -C ₃ H ₂) ⁺	377
C ₃ N ₂ O ⁺	CO(CN) ₂ ⁺	343	C ₄ H ₄	<i>cyc</i> -C ₄ H ₄	378,A338,B355
C ₃ N ₂ O ₂	(CN) ₂ COO	A330	C ₄ H ₄	H ₂ C=(<i>cyc</i> -C ₃ H ₂)	377
C ₃ Nb	<i>cyc</i> -NbC ₃	B203	C ₄ H ₄	C ₂ H ₃ CH=C:	A338
C ₃ Nb ⁻	<i>cyc</i> -NbC ₃ ⁻	B209			

Formula	Structure/Name	References	Formula	Structure/Name	References
C ₄ H ₄	H ₂ C=C=CHCH:	B355	C ₄ Si ₂ ⁻	SiC ₄ Si ⁻	B327
C ₄ H ₄ ⁻	C ₂ H ₃ CH=C: ⁻	A338	C ₄ Ti	TiC ₄	B279
C ₄ H ₄ CaN	Ca(C ₄ H ₄ N)	410	C ₄ Ti ⁻	TiC ₄ ⁻	B279
C ₄ H ₄ CdN	Cd(C ₄ H ₄ N)	412	C ₅	C ₅	277,A292,B280
C ₄ H ₄ MgN	Mg(C ₄ H ₄ N)	410	C ₅ ⁻	C ₅ ⁻	278,A293,B283
C ₄ H ₄ NSr	Sr(C ₄ H ₄ N)	411	C ₅ Br ₄	<i>cyc</i> -C ₅ Br ₄	A357
C ₄ H ₄ NZn	Zn(C ₄ H ₄ N)	411	C ₅ Br ₄ O ₂	<i>cyc</i> -C ₅ Br ₄ O-1-O	A366
C ₄ H ₅ N	H ₂ C=(<i>cyc</i> -CN=CCH ₃)	A361	C ₅ Cl ₄	<i>cyc</i> -C ₅ Cl ₄	A357
C ₄ H ₆ ⁺	<i>t</i> -H ₂ C=CH-CH=CH ₂ ⁺	A342,B363	C ₅ Cl ₄ O ₂	<i>cyc</i> -C ₅ Cl ₄ O-1-O	A365
C ₄ H ₆	H ₂ CCHC(CH ₃):	B364	C ₅ F ₄ ⁺	CF ₃ (CC) ₂ F ⁺	407
C ₄ H ₆	C(CH ₂) ₃	A343	C ₅ F ₆ O	(CF ₃) ₂ (<i>cyc</i> -CCO)	418
C ₄ H ₆ ⁻	C(CH ₂) ₃ ⁻	A343	C ₅ F ₈ O	CF ₃ (<i>cyc</i> -CCO)C ₂ F ₅	419
C ₄ H ₆ Ge	<i>cyc</i> -C ₄ H ₄ GeH ₂	A359	C ₅ H	C ₅ H	A316,B322
C ₄ H ₆ Ge	<i>cyc</i> -C ₄ H ₆ Ge:	A359	C ₅ HN ⁺	H(C≡C) ₂ CN ⁺	369,A328,B342
C ₄ H ₆ Ge	<i>cyc</i> -C ₄ H ₅ GeH (1,3)	A360	C ₅ HN	H(C≡C) ₂ NC	B343
C ₄ H ₆ Ge	<i>cyc</i> -C ₄ H ₅ GeH (1,4)	A360	C ₅ HSi	SiC ₅ H	B342
C ₄ H ₆ S	(CH ₃) ₂ (<i>cyc</i> -CCS)	419	C ₅ H ₂ ⁺	C ₅ H ₂ ⁺	A327
C ₄ H ₆ Si	<i>cyc</i> -C ₄ H ₄ SiH ₂	A358	C ₅ H ₂	HC ₅ H	A327,B338
C ₄ H ₆ Si	<i>cyc</i> -C ₄ H ₆ Si:	A358	C ₅ H ₂	(<i>cyc</i> -HC=CHC)=C=C:	B338
C ₄ H ₆ Si	<i>cyc</i> -C ₄ H ₅ SiH (1,3)	A358	C ₅ H ₂	HCCCH=C=C:	B339
C ₄ H ₆ Si	<i>cyc</i> -C ₄ H ₅ SiH (1,4)	A359	C ₅ H ₂	H ₂ C ₅ :	B339
C ₄ H ₇ ⁺	H ₂ CC(CH ₃)CH ₂ ⁺	A343	C ₅ H ₂	(<i>cyc</i> -HC ₃)CCH	B339
C ₄ H ₇ ⁺	<i>cyc</i> -C ₄ H ₇ ⁺	A343	C ₅ H ₃	H ₂ C ₅ H	B358
C ₄ H ₇	CH ₃ CHCH=CH ₂	A343	C ₅ H ₃ Br ⁺	CH ₃ (CC) ₂ Br ⁺	406
C ₄ H ₇	H ₂ CC(CH ₃)CH ₂	A343	C ₅ H ₃ Cl ⁺	CH ₃ (CC) ₂ Cl ⁺	406
C ₄ H ₇ ⁻	H ₂ CC(CH ₃)CH ₂ ⁻	A344	C ₅ H ₃ F ₂ N ⁺	<i>cyc</i> -C ₅ H ₃ F ₂ N ⁺	412
C ₄ H ₇ N	(CH ₃) ₂ C=C=NH	412	C ₅ H ₃ N	<i>cyc</i> -C ₅ H ₃ N	412
C ₄ H ₇ N	CH ₃ CC-NHCH ₃	412	C ₅ H ₄ ⁺	CH ₃ (CC) ₂ H ⁺	385
C ₄ H ₇ O	CH ₃ CHCOCH ₃	B392	C ₅ H ₄	<i>cyc</i> -C ₅ H ₄	385
C ₄ H ₇ O	(CH ₃) ₂ CCHO	B392	C ₅ H ₄ Cl	<i>cyc</i> -C ₅ H ₄ Cl	407
C ₄ H ₈ Si	<i>cyc</i> -C ₂ H ₂ Si(CH ₃) ₂	A358	C ₅ H ₄ F	<i>cyc</i> -C ₅ H ₄ F	407
C ₄ H ₉ ⁺	<i>t</i> -C ₄ H ₉ ⁺	384	C ₅ H ₄ O	(<i>cyc</i> -C ₅ H ₄)O	419
C ₄ H ₉	<i>n</i> -C ₄ H ₉	384,A344	C ₅ H ₄ O ₂	(<i>cyc</i> -C ₅ H ₄)O ₂	B394
C ₄ H ₉	<i>i</i> -C ₄ H ₉	384	C ₅ H ₄ O ₂	<i>cyc</i> -C ₅ H ₄ O-1-O	422,A365
C ₄ H ₉	<i>t</i> -C ₄ H ₉	384	C ₅ H ₅	<i>cyc</i> -C ₅ H ₅	385,A344,B364
C ₄ H ₉ O	1-C ₄ H ₉ O	B392	C ₅ H ₅ Ca	CaC ₅ H ₅	402,A360
C ₄ H ₉ O	2-C ₄ H ₉ O	B392	C ₅ H ₅ Cd	CdC ₅ H ₅	403
C ₄ H ₉ O	(CH ₃) ₃ CO	B392	C ₅ H ₅ Mg	MgC ₅ H ₅	401
C ₄ H ₉ O ⁻	(CH ₃) ₃ CO ⁻	B393	C ₅ H ₅ Sr	SrC ₅ H ₅	402
C ₄ H ₉ O ₂	<i>t</i> -C ₄ H ₉ O ₂	421,B394	C ₅ H ₅ Zn	ZnC ₅ H ₅	402
C ₄ H ₉ O ₂ ⁻	<i>t</i> -C ₄ H ₉ O ₂ ⁻	B394	C ₅ H ₆ ⁺	<i>cyc</i> -C ₅ H ₆ ⁺	A345
C ₄ H ₁₀ Si	(CH ₃) ₂ Si=CHCH ₃	401,B387	C ₅ H ₆ Se	(CH ₂) ₃ C=C=Se	419
C ₄ I ₂ ⁺	I(CC) ₂ I ⁺	345	C ₅ H ₆ Si ⁺	C ₅ SiH ₆ ⁺	404
C ₄ N ₂ ⁺	NCC≡CCN ⁺	343,A319,B327	C ₅ H ₆ Si	C ₅ SiH ₆	404
C ₄ N ₂	NCCCNC	343,A320,B328	C ₅ H ₆ Si	C ₅ SiH ₆ (Dewar)	405
C ₄ N ₂	CNCCNC	343	C ₅ H ₉ O	(CH ₃) ₂ CCOCH ₃	B392
C ₄ N ₂ O	(CN) ₂ CCO	A330	C ₅ H ₁₁	<i>n</i> -C ₅ H ₁₁	386,A346
C ₄ O	C ₄ O	278,A294	C ₅ H ₁₁	(CH ₃) ₃ CCH ₂	386
C ₄ O ⁻	C ₄ O ⁻	B292	C ₅ N	C ₅ N	A319,B327
C ₄ OS	C ₄ OS	344	C ₅ N ⁻	C ₅ N ⁻	B328
C ₄ O ₂	C ₄ O ₂	344	C ₅ N ₂	C ₅ N ₂	A330,B344
C ₄ S	C ₄ S	278,A294,B288	C ₅ O	C ₅ O	A320,B328
C ₄ S ₂	C ₄ S ₂	344	C ₅ OS	C ₅ OS	370
C ₄ Si	C ₄ Si	277,A293,B280	C ₅ O ₂	C ₅ O ₂	370
C ₄ Si ₂	SiC ₄ Si	A318,B326	C ₅ S	C ₅ S	343,B328

Formula	Structure/Name	References	Formula	Structure/Name	References
C ₅ S ₂	C ₅ S ₂	370	C ₆ H ₄ Br	<i>m</i> -C ₆ H ₄ Br	428
C ₅ Ti	TiC ₅	B325	C ₆ H ₄ Br	<i>p</i> -C ₆ H ₄ Br	428
C ₅ Ti ⁻	TiC ₅ ⁻	B325	C ₆ H ₄ Cl	<i>o</i> -C ₆ H ₄ Cl	427
C ₆	C ₆	342,A317,B325	C ₆ H ₄ Cl	<i>m</i> -C ₆ H ₄ Cl	427
C ₆	<i>cyc</i> -C ₆	B326	C ₆ H ₄ Cl	<i>p</i> -C ₆ H ₄ Cl	427
C ₆ ⁻	C ₆ ⁻	342,A318,B326	C ₆ H ₄ Cl ₂ ⁺	1,3-C ₆ H ₄ Cl ₂ ⁺	431
C ₆ Br ₃ F ₃ ⁺	sym-C ₆ F ₃ Br ₃ ⁺	439	C ₆ H ₄ Cl ₂ ⁺	1,4-C ₆ H ₄ Cl ₂ ⁺	431
C ₆ ClF ₅ ⁺	C ₆ F ₅ Cl ⁺	439	C ₆ H ₄ F	<i>o</i> -C ₆ H ₄ F	427
C ₆ Cl ₃ F ₃ ⁺	sym-C ₆ F ₃ Cl ₃ ⁺	439	C ₆ H ₄ F	<i>m</i> -C ₆ H ₄ F	427
C ₆ F ₄	<i>o</i> -C ₆ F ₄	B396	C ₆ H ₄ F	<i>p</i> -C ₆ H ₄ F	427
C ₆ F ₄	<i>m</i> -C ₆ F ₄	B396	C ₆ H ₄ F ₂ ⁺	1,2-C ₆ H ₄ F ₂ ⁺	429
C ₆ F ₄	<i>p</i> -C ₆ F ₄	B396	C ₆ H ₄ F ₂ ⁺	1,3-C ₆ H ₄ F ₂ ⁺	429
C ₆ F ₄	FC ₆ CF=CF ₆ CF	B387	C ₆ H ₄ F ₂ ⁺	1,4-C ₆ H ₄ F ₂ ⁺	430,A369
C ₆ F ₄ I	<i>m</i> -C ₆ F ₄ I	B396	C ₆ H ₄ F ₂ O ⁺	2,5-F ₂ C ₆ H ₃ OH ⁺	444
C ₆ F ₄ I	<i>p</i> -C ₆ F ₄ I	B397	C ₆ H ₄ F ₂ O ⁺	3,5-F ₂ C ₆ H ₃ OH ⁺	444
C ₆ F ₅ N	C ₆ F ₅ N:	442,A371	C ₆ H ₄ I	<i>o</i> -C ₆ H ₄ I	428
C ₆ F ₅ N	<i>bicyc</i> -C ₆ F ₅ N	A361	C ₆ H ₄ N	(<i>cyc</i> -C ₅ H ₄)CN	413
C ₆ F ₆ ⁺	CF ₃ (CC) ₂ CF ₃ ⁺	307	C ₆ H ₄ N ₂ O ₂	1,2-C ₆ H ₄ (NO) ₂	446
C ₆ F ₆ ⁺	C ₆ F ₆ ⁺	438	C ₆ H ₄ O	2,4-C ₆ H ₃ OH	426,B396
C ₆ H	C ₆ H	369,A328,B341	C ₆ H ₄ S ₂	<i>p</i> -C ₆ H ₄ S ₂	422
C ₆ H ⁻	C ₆ H ⁻	B343	C ₆ H ₅ ⁺	C ₆ H ₅ ⁺	387,B367
C ₆ HF ₅ ⁺	C ₆ HF ₅ ⁺	437	C ₆ H ₅	C ₆ H ₅	388,A346,B367
C ₆ HF ₅ O ⁺	C ₆ F ₅ OH ⁺	445	C ₆ H ₅ ⁻	C ₆ H ₅ ⁻	388
C ₆ HN ⁺	HC ₅ CN ⁺	A340	C ₆ H ₅ Br ⁺	C ₆ H ₅ Br ⁺	B399
C ₆ HN	HC ₆ N	B360	C ₆ H ₅ Cl ⁺	C ₆ H ₅ Cl ⁺	428,A368,B397
C ₆ HN	(<i>cyc</i> -HC ₃)CCC	B360	C ₆ H ₅ F ⁺	C ₆ H ₅ F ⁺	428,B397
C ₆ HSi	SiC ₆ H	B359	C ₆ H ₅ I ⁺	C ₆ H ₅ I ⁺	B399
C ₆ H ₂ ⁺	C ₆ H ₂ ⁺	380,A339,B358	C ₆ H ₅ N	C ₆ H ₅ N	442,A370
C ₆ H ₂	H ₂ C ₆ :	A340,B359	C ₆ H ₅ N	3-CH(<i>cyc</i> -C ₅ H ₄ N)	413
C ₆ H ₂ F ₄ ⁺	1,2,3,4-C ₆ H ₂ F ₄ ⁺	435,B400	C ₆ H ₅ N	<i>cyc</i> -C ₆ H ₅ N	413
C ₆ H ₂ F ₄ ⁺	1,2,3,5-C ₆ H ₂ F ₄ ⁺	436,B400	C ₆ H ₅ N ⁻	C ₆ H ₅ N ⁻	442,A371
C ₆ H ₂ F ₄ ⁺	1,2,4,5-C ₆ H ₂ F ₄ ⁺	437,B401	C ₆ H ₅ N ₂ ⁺	C ₆ H ₅ NN ⁺	B388
C ₆ H ₂ F ₄ O ⁺	2,3,5,6-F ₄ C ₆ HOH ⁺	445	C ₆ H ₅ O	C ₆ H ₅ O	443,A372,B407
C ₆ H ₂ O ₂	O=C ₆ H ₂ =O	B395	C ₆ H ₅ O ⁻	C ₆ H ₅ O ⁻	446
C ₆ H ₂ O ₂ ⁻	O=C ₆ H ₂ =O ⁻	B395	C ₆ H ₅ S	C ₆ H ₅ S	443
C ₆ H ₃ ClF ₂ ⁺	1,3,5-C ₆ H ₃ ClF ₂ ⁺	434	C ₆ H ₆ ⁺	C ₆ H ₆ ⁺	388,A347,B368
C ₆ H ₃ Cl ₂ F ⁺	1,3,5-C ₆ H ₃ Cl ₂ F ⁺	434	C ₆ H ₆ ⁺	CH ₃ (CC) ₂ CH ₃ ⁺	390
C ₆ H ₃ Cl ₃ ⁺	1,3,5-C ₆ H ₃ Cl ₃ ⁺	434	C ₆ H ₆ CIN ⁺	<i>p</i> -C ₆ H ₄ CINH ₂ ⁺	B407
C ₆ H ₃ F	3,5-C ₆ H ₃ F	B395	C ₆ H ₆ F	C ₆ H ₆ F	440
C ₆ H ₃ F ₂ N	2,6-C ₆ H ₃ F ₂ N:	A371	C ₆ H ₆ FN ⁺	<i>o</i> -C ₆ H ₄ FNH ₂ ⁺	B407
C ₆ H ₃ F ₂ N	<i>bicyc</i> -C ₆ H ₃ F ₂ N	A361	C ₆ H ₆ FN ⁺	<i>p</i> -C ₆ H ₄ FNH ₂ ⁺	B407
C ₆ H ₃ F ₃ ⁺	1,2,3-C ₆ H ₃ F ₃ ⁺	432	C ₆ H ₆ Li	LiC ₆ H ₆	440
C ₆ H ₃ F ₃ ⁺	1,2,4-C ₆ H ₃ F ₃ ⁺	432	C ₆ H ₆ N	C ₆ H ₅ NH	443
C ₆ H ₃ F ₃ ⁺	1,3,5-C ₆ H ₃ F ₃ ⁺	433	C ₆ H ₆ O ⁺	C ₆ H ₅ OH ⁺	444,A373
C ₆ H ₃ F ₃ O ⁺	2,3,4-F ₃ C ₆ H ₂ OH ⁺	445	C ₆ H ₆ O ₂ ⁺	<i>o</i> -C ₆ H ₄ (OH) ₂ ⁺	B409
C ₆ H ₃ F ₃ O ⁺	2,4,5-F ₃ C ₆ H ₂ OH ⁺	445	C ₆ H ₆ O ₂ ⁺	<i>m</i> -C ₆ H ₄ (OH) ₂ ⁺ (C _s)	B409
C ₆ H ₃ N ⁺	CH ₃ (CC) ₂ CN ⁺	413	C ₆ H ₆ O ₂ ⁺	<i>m</i> -C ₆ H ₄ (OH) ₂ ⁺ (C _{2v})	B409
C ₆ H ₄ ⁺	C ₆ H ₄ ⁺	386	C ₆ H ₆ O ₂ ⁺	<i>p</i> -C ₆ H ₄ (OH) ₂ ⁺ (C _{2v})	B409
C ₆ H ₄	<i>o</i> -C ₆ H ₄	386,B364	C ₆ H ₆ O ₂ ⁺	<i>p</i> -C ₆ H ₄ (OH) ₂ ⁺ (C _{2h})	B409
C ₆ H ₄	<i>m</i> -C ₆ H ₄	A346,B365	C ₆ H ₇	CH ₃ (<i>cyc</i> -C ₅ H ₄)	391
C ₆ H ₄	<i>p</i> -C ₆ H ₄	B366	C ₆ H ₇ Ca	CaC ₅ H ₄ CH ₃	403
C ₆ H ₄ ⁻	<i>o</i> -C ₆ H ₄ ⁻	387,B366	C ₆ H ₇ Cd	CdC ₅ H ₄ CH ₃	404
C ₆ H ₄ ⁻	<i>m</i> -C ₆ H ₄ ⁻	B367	C ₆ H ₇ Mg	MgC ₅ H ₄ CH ₃	403
C ₆ H ₄ ⁻	<i>p</i> -C ₆ H ₄ ⁻	B367	C ₆ H ₇ Zn	ZnC ₅ H ₄ CH ₃	404
C ₆ H ₄ Br	<i>o</i> -C ₆ H ₄ Br	428	C ₆ H ₇ N ⁺	C ₆ H ₅ NH ₂ ⁺	443,A372,B406

Formula	Structure/Name	References	Formula	Structure/Name	References
C ₆ H ₈ ⁺	<i>t</i> -CH ₂ (CH) ₄ CH ₂ ⁺	391	C ₇ H ₆ F ⁻	(2-FC ₆ H ₄)CH ₂ ⁻	B404
C ₆ H ₈ Si	1-CH ₃ C ₅ SiH ₅	405	C ₇ H ₆ F ⁻	(3-FC ₆ H ₄)CH ₂ ⁻	B404
C ₆ H ₁₀	(CH ₃) ₃ CCH=C:	A348	C ₇ H ₆ F ⁻	(4-FC ₆ H ₄)CH ₂ ⁻	B405
C ₆ H ₁₀ ⁻	(CH ₃) ₃ CCH=C ⁻	A348	C ₇ H ₇ ⁺	<i>cyc</i> -C ₇ H ₇ ⁺	392
C ₆ N ₂ ⁺	NC(CC) ₂ CN ⁺	414,A341,B362	C ₇ H ₇ ⁺	C ₆ H ₅ CH ₂ ⁺	392,A349
C ₆ O	C ₆ O	370,A330	C ₇ H ₇	C ₆ H ₅ CH ₂	392,A349,B372
C ₆ O ₂ S ₄	C ₆ S ₄ O ₂	426	C ₇ H ₇	<i>cyc</i> -C ₇ H ₇	394
C ₆ S ₆	C ₆ S ₆ (A)	426	C ₇ H ₇ ⁻	C ₆ H ₅ CH ₂ ⁻	394,B373
C ₆ S ₆	C ₆ S ₆ (B)	427	C ₇ H ₈ ⁺	C ₆ H ₅ CH ₃ ⁺	394
C ₆ Si	C ₆ Si	B343	C ₇ N	C ₇ N	B362
C ₇	C ₇	369,A329,B343	C ₇ N ⁻	C ₇ N ⁻	B362
C ₇ ⁻	C ₇ ⁻	370,A329,B344	C ₇ N ₂ ⁺	NCC ₅ CN ⁺	A362
C ₇ F ₄ O	1,2-C ₆ F ₄ >CO	B409	C ₇ O	C ₇ O	A341,B362
C ₇ F ₈ ⁺	C ₆ F ₅ CF ₃ ⁺	442	C ₇ O ₂	C ₇ O ₂	422
C ₇ H	C ₇ H	A340,B359	C ₇ Si	SiC ₇	B362
C ₇ HN ⁺	H(C≡C) ₃ CN ⁺	A362,B388	C ₈	C ₈	381,A340,B360
C ₇ HN ⁺	H(C≡C) ₃ NC	B388	C ₈	<i>cyc</i> -C ₈	B361
C ₇ H ₂ ⁺	C ₇ H ₂ ⁺	A349	C ₈ ⁻	C ₈ ⁻	381,A340,B361
C ₇ H ₂	HC ₇ H	A349,B370	C ₈ H	C ₈ H	A351,B373
C ₇ H ₂	(<i>cyc</i> -HC ₃)(C≡C) ₂ H	B370	C ₈ H ⁻	C ₈ H ⁻	B374
C ₇ H ₂ F ₅	C ₆ F ₅ CH ₂	B404	C ₈ HN ⁺	HC ₇ CN ⁺	A362
C ₇ H ₃ F ₅ ⁺	C ₆ F ₅ CH ₃ ⁺	441	C ₈ H ₂ ⁺	C ₈ H ₂ ⁺	A351,B374
C ₇ H ₄ O	(<i>cyc</i> -C ₅ H ₄)CCO	419	C ₈ H ₄ Cl ₂	<i>p</i> -ClCC ₆ H ₄ CCl	B387
C ₇ H ₄ O	<i>cyc</i> -C ₆ H ₄ C=O	419	C ₈ H ₄ F ₄	<i>p</i> -(CF ₂) ₂ C ₆ H ₄	B406
C ₇ H ₅ Cl	C ₆ H ₅ CCl	441	C ₈ H ₄ F ₄	4,5-F ₂ C ₇ H ₄ =CF ₂	B406
C ₇ H ₅ Cl	(2-CIC ₆ H ₄)CH	441	C ₈ H ₆	<i>p</i> -HC=C ₆ H ₄ =CH	B375
C ₇ H ₅ Cl	<i>cyc</i> -1-C ₇ H ₅ Cl	408	C ₈ H ₆	<i>cyc</i> -C ₈ H ₆	B374
C ₇ H ₅ ClO	C ₆ H ₅ OCCl	446	C ₈ H ₆ ⁻	<i>cyc</i> -C ₈ H ₆ ⁻	B375
C ₇ H ₅ Cl ₂	C ₆ H ₅ CCl ₂	B402	C ₈ H ₆ N	(4-NC-C ₆ H ₄)CH ₂	A370,B405
C ₇ H ₅ F	C ₆ H ₅ CF	440	C ₈ H ₈	<i>o</i> -(CH ₂) ₂ C ₆ H ₄	396
C ₇ H ₅ F	<i>cyc</i> -1-C ₇ H ₅ F	407	C ₈ H ₈	<i>m</i> -(CH ₂) ₂ C ₆ H ₄	A351
C ₇ H ₅ F ₂	(2,6-F ₂ C ₆ H ₃)CH ₂	B404	C ₈ H ₈	<i>p</i> -(CH ₂) ₂ C ₆ H ₄	A352
C ₇ H ₅ F ₃ ⁺	sym-C ₆ H ₂ F ₃ CH ₃ ⁺	441	C ₈ H ₈	C ₆ H ₅ CCH ₃	396
C ₇ H ₅ N ⁺	C ₂ H ₅ (CC) ₂ CN ⁺	414	C ₈ H ₈	<i>m</i> -CH ₃ C ₆ H ₄ CH:	395
C ₇ H ₆	3,5-C ₆ H ₃ CH ₃	B370	C ₈ H ₈	<i>o</i> -CH ₃ C ₆ H ₄ CH:	395
C ₇ H ₆	C ₆ H ₅ CH	391,B370	C ₈ H ₈	<i>p</i> -CH ₃ C ₆ H ₄ CH:	395
C ₇ H ₆	<i>cyc</i> -C ₇ H ₆	391,B371	C ₈ H ₈	1-CH ₃ (<i>cyc</i> -C ₇ H ₅)	396
C ₇ H ₆	<i>cyc</i> -C ₇ H ₆ :	392	C ₈ H ₈	4-CH ₃ (<i>cyc</i> -C ₇ H ₅)	397
C ₇ H ₆ Br	C ₆ H ₅ CHBr	B402	C ₈ H ₈	5-CH ₃ (<i>cyc</i> -C ₇ H ₅)	397
C ₇ H ₆ Br	(2-BrC ₆ H ₄)CH ₂	B403	C ₈ H ₈ ⁻	<i>m</i> -(CH ₂) ₂ C ₆ H ₄	A352
C ₇ H ₆ Br	(3-BrC ₆ H ₄)CH ₂	B404	C ₈ N ₂ ⁺	NC(C≡C) ₃ CN ⁺	A362
C ₇ H ₆ Br	(4-BrC ₆ H ₄)CH ₂	B404	C ₈ O	C ₈ O	A364
C ₇ H ₆ Br ⁻	(2-BrC ₆ H ₄)CH ₂ ⁻	B405	C ₉	C ₉	398,A352,B375
C ₇ H ₆ Br ⁻	(3-BrC ₆ H ₄)CH ₂ ⁻	B405	C ₉ ⁻	C ₉ ⁻	398,A352,B376
C ₇ H ₆ Br ⁻	(4-BrC ₆ H ₄)CH ₂ ⁻	B405	C ₉ H	C ₉ H	A353
C ₇ H ₆ Cl	C ₆ H ₅ CHCl	B401	C ₉ H ⁻	<i>cum</i> -C ₉ H ⁻	B376
C ₇ H ₆ Cl	(2-CIC ₆ H ₄)CH ₂	B403	C ₉ H ⁻	<i>acet</i> -C ₉ H ⁻	B376
C ₇ H ₆ Cl	(3-CIC ₆ H ₄)CH ₂	B403	C ₉ HN ⁺	H(C≡C) ₄ CN ⁺	A362
C ₇ H ₆ Cl	(4-CIC ₆ H ₄)CH ₂	A370,B403	C ₉ H ₂ ⁺	C ₉ H ₂ ⁺	A353
C ₇ H ₆ Cl ⁻	(2-CIC ₆ H ₄)CH ₂ ⁻	B405	C ₉ H ₂	HC ₉ H	A353,B376
C ₇ H ₆ Cl ⁻	(3-CIC ₆ H ₄)CH ₂ ⁻	B405	C ₉ H ₂	(<i>cyc</i> -HC ₃)(C≡C) ₃ H	B377
C ₇ H ₆ Cl ⁻	(4-CIC ₆ H ₄)CH ₂ ⁻	B405	C ₉ N	C ₉ N	B388
C ₇ H ₆ F	(2-FC ₆ H ₄)CH ₂	A369,B402	C ₉ N ⁻	C ₉ N ⁻	B389
C ₇ H ₆ F	(3-FC ₆ H ₄)CH ₂	A369,B402	C ₉ N ₂ ⁺	NCC ₇ CN ⁺	A363
C ₇ H ₆ F	(4-FC ₆ H ₄)CH ₂	A370,B402	C ₉ O	C ₉ O	A364,B393

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
C ₉ Si	SiC ₉	B387	C ₁₅ H ⁻	<i>cum</i> -C ₁₅ H ⁻	B384
C ₁₀	C ₁₀	A353,B377	C ₁₅ H ⁻	<i>acet</i> -C ₁₅ H ⁻	B384
C ₁₀	<i>cy</i> c-C ₁₀	B377	C ₁₆	C ₁₆	B385
C ₁₀ ⁻	C ₁₀ ⁻	A354,B377	C ₁₆ ⁻	C ₁₆ ⁻	A358,B385
C ₁₀ H	C ₁₀ H	A354,B378	CaHO	CaOH	27,A139
C ₁₀ H ⁻	<i>cum</i> -C ₁₀ H ⁻	B378	CaHS	CaSH	28,A140
C ₁₀ H ⁻	<i>acet</i> -C ₁₀ H ⁻	B378	CaH ₂	CaH ₂	14
C ₁₀ HN ⁺	HC ₉ CN ⁺	A363	CaH ₂ N	CaNH ₂	132,A220,B163
C ₁₀ H ₂ ⁺	C ₁₀ H ₂ ⁺	A354	CaH ₂ O	HCaOH	134
C ₁₀ N ₂ ⁺	NC(C≡C) ₄ CN ⁺	A363	CaH ₂ O ₂	Ca(OH) ₂	251
C ₁₁	C ₁₁	398,A354,B378	CaN ₃	CaN ₃	182
C ₁₁ ⁻	C ₁₁ ⁻	398,B379	CaO ₂	OCaO	A174
C ₁₁ H	C ₁₁ H	A354	CaO ₂	<i>cy</i> c-CaO ₂	A174
C ₁₁ H ⁻	<i>cum</i> -C ₁₁ H ⁻	B379	CaO ₄	O ₂ CaO ₂	A298
C ₁₁ H ⁻	<i>acet</i> -C ₁₁ H ⁻	B379	Ca ₂ H ₂	Ca ₂ H ₂	130
C ₁₁ HN ⁺	H(C≡C) ₅ CN ⁺	A363	Ca ₂ H ₂ O	HCaOCaH	246
C ₁₁ HN ⁻	HC ₁₀ CN ⁻	B389	Ca ₂ H ₂ O	HCa ₂ OH	247
C ₁₁ H ₂ ⁺	C ₁₁ H ₂ ⁺	A355	Ca ₂ H ₄	HCaH ₂ CaH	305
C ₁₁ H ₂	HC ₁₁ H	A355,B379	Ca ₂ O ₂	<i>cy</i> c-(CaO) ₂	A243
C ₁₁ N	C ₁₁ N	B389	Ca ₂ O ₂	CaOCaO	A243
C ₁₁ N ⁻	C ₁₁ N ⁻	B389	Ca ₃ H ₂	Ca ₃ H ₂	244
C ₁₁ N ₂ ⁺	NCC ₉ CN ⁺	A363	CdH ₂	CdH ₂	A127
C ₁₂	C ₁₂	A355,B380	CdH ₂ O	HCdOH	A222
C ₁₂	<i>cy</i> c-C ₁₂	B380	CdO ₂	OCdO	A179
C ₁₂ ⁻	C ₁₂ ⁻	A355,B380	Cd ₂ H	CdCdH	A136
C ₁₂ H	C ₁₂ H	A355,B380	Cd ₂ O ₂	CdOCdO	A244
C ₁₂ H ⁻	<i>cum</i> -C ₁₂ H ⁻	B380	CeH ₂ ⁺	CeH ₂ ⁺	B11
C ₁₂ H ⁻	<i>acet</i> -C ₁₂ H ⁻	B381	CeH ₂	CeH ₂	B13
C ₁₂ H ₂	HC ₁₂ H	B381	CeH ₃	CeH ₃	B154
C ₁₂ H ₂ ⁻	HC ₁₂ H ⁻	B381	CeH ₄	CeH ₄	B245
C ₁₂ HN ⁺	HC ₁₁ CN ⁺	A363	CeNO	NCeO	B93
C ₁₂ H ₂ ⁺	C ₁₂ H ₂ ⁺	A355	CeNO	CeNO	B93
C ₁₂ N ₂ ⁺	NC(C≡C) ₅ CN ⁺	A363	CeNO ⁻	NCeO ⁻	B102
C ₁₃	C ₁₃	A356,B381	CeN ₂	NCeN	B74
C ₁₃ ⁻	C ₁₃ ⁻	B382	CeN ₂	CeNN	B74
C ₁₃ H	C ₁₃ H	B382	CeO ₂	OCeO	A179,B109
C ₁₃ H ⁻	<i>cum</i> -C ₁₃ H ⁻	B382	CeO ₂	<i>cy</i> c-CeO ₂	B109
C ₁₃ H ⁻	<i>acet</i> -C ₁₃ H ⁻	B382	CeO ₂ ⁻	OCeO ⁻	B118
C ₁₃ HN ⁺	H(C≡C) ₆ CN ⁺	A363	Ce ₂ H ₂	<i>cy</i> c-(CeH) ₂	B161
C ₁₃ HN ⁻	HC ₁₂ CN ⁻	B390	Ce ₂ N ₂	<i>cy</i> c-(CeN) ₂	B205
C ₁₃ H ₂ ⁺	C ₁₃ H ₂ ⁺	A356	Ce ₂ O ₂	<i>cy</i> c-(CeO) ₂	B212
C ₁₃ H ₂	HC ₁₃ H	A356,B382	ClFH ⁻	FHCl ⁻	53
C ₁₃ N	C ₁₃ N	B389	ClFN	NFCl	108,A201
C ₁₃ N ⁻	C ₁₃ N ⁻	B389	ClFO	FCIO	116,B152
C ₁₄	C ₁₄	B382	ClFO ₂ S ⁺	FCISO ₂ ⁺	301
C ₁₄	<i>cy</i> c-C ₁₄	B383	ClFO ₃ ⁺	FCIO ₃ ⁺	300
C ₁₄ ⁻	C ₁₄ ⁻	A356,B383	ClFS	FSCI	115,A208
C ₁₄ H	C ₁₄ H	A356,B383	ClFSi	SiFCl	B140
C ₁₄ H ⁻	<i>cum</i> -C ₁₄ H ⁻	B383	ClFXe	XeClF	121
C ₁₄ H ⁻	<i>acet</i> -C ₁₄ H ⁻	B383	ClF ₂	ClF ₂	116
C ₁₄ H ₂ ⁺	C ₁₄ H ₂ ⁺	A356	ClF ₂ ⁻	FCIF ⁻	117
C ₁₄ H ₂	HC ₁₄ H	B384	ClF ₂ ⁻	FFCl ⁻	118
C ₁₄ H ₂ ⁻	HC ₁₄ H ⁻	B384	ClF ₂ P ⁺	PF ₂ Cl ⁺	220
C ₁₅	C ₁₅	A357,B384	ClF ₃ ⁺	ClF ₃ ⁺	226
C ₁₅ ⁻	C ₁₅ ⁻	B384	ClF ₃ P ⁻	PClF ₃ ⁻	304

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
ClF ₃ S	SClF ₃	304	ClO ₄ ⁻	ClO ₄ ⁻	B307
ClF ₃ Si ⁺	SiF ₃ Cl ⁺	295	CIPS	CIPS	104
ClF ₄ Si ⁻	SiF ₄ Cl ⁻	357	ClS ₂	SSCl	111,A205
ClFeH	HFeCl	35	ClXe ₂	Xe ₂ Cl	122
ClGaH ₂	GaH ₂ Cl	143	Cl ₂ Cu	CuCl ₂	B129
ClGeH	HGeCl	45,B40	Cl ₂ F	ClCIF	116
ClGeH ₂	H ₂ GeCl	150	Cl ₂ F ⁻	ClFCl ⁻	118
ClGeH ₃ ⁺	GeH ₃ Cl ⁺	241	Cl ₂ F ⁻	FCICI ⁻	118
ClHHg	HHgCl	B31	Cl ₂ F ₂	Cl ₂ F ₂	227
ClHI ⁻	ClHI ⁻	55	Cl ₂ F ₂ S	SCl ₂ F ₂	304
ClHKr	HKrCl	A158,B47	Cl ₂ F ₂ Si ⁺	SiF ₂ Cl ₂ ⁺	295
ClHO ⁺	HOCl ⁺	51	Cl ₂ F ₃ Si ⁻	SiF ₃ Cl ₂ ⁻	357
ClHO	HOCl	52,A155,B45	Cl ₂ Ga	GaCl ₂	90
ClHO ₂	HOCIO	A241	Cl ₂ GaH	HGaCl ₂	169,B194
ClHO ₂	HClO ₂	A241	Cl ₂ Ga ₂ H ₄	H ₂ GaCl ₂ GaH ₂	A339
ClHO ₄	HOOCIO ₂	B324	Cl ₂ Ge ⁺	GeCl ₂ ⁺	92
ClHP	HPCl	B43	Cl ₂ Ge	GeCl ₂	102,A198,B141
ClHSSi	HClSi=S	172	Cl ₂ GeH ₂ ⁺	GeH ₂ Cl ₂ ⁺	263
ClHSi	HSiCl	44,A150,B39	Cl ₂ GeS	Cl ₂ GeS	211
ClHXe	HXeCl	A158,B47	Cl ₂ H ⁻	ClHCl ⁻	54,A157,B46
ClH ₂ ⁺	H ₂ Cl ⁺	26,B21	Cl ₂ HIn	HInCl ₂	B194
ClH ₂ In	InH ₂ Cl	B172	Cl ₂ HN ⁺	HNCI ₂ ⁺	179
ClH ₂ InO	HInCl(OH)	B269	Cl ₂ H ₂ ⁺	HClClH ⁺	B179
ClH ₂ N ⁺	H ₂ NCl ⁺	152	Cl ₂ H ₂ Si ⁺	SiH ₂ Cl ₂ ⁺	262
ClH ₂ P	H ₂ PCl	A229	Cl ₂ Li	ClLiCl	B115
ClH ₃ Si ⁺	SiH ₃ Cl ⁺	240	Cl ₂ Mn ⁺	MnCl ₂ ⁺	73
ClIO	OICl	116	Cl ₂ N	NCl ₂	108
ClIO ₂	ClIO ₂	226	Cl ₂ Na	ClNaCl	B115
ClIO ₂	IClO ₂	A264	Cl ₂ Ni ⁺	NiCl ₂ ⁺	73
ClIO ₂	IOClO	A264	Cl ₂ Ni	NiCl ₂	B128
ClKrNe	NeKrCl	122	Cl ₂ O ⁺	Cl ₂ O ⁺	111,B148
ClKrXe	KrXeCl	122	Cl ₂ O	ClClO	116,A208
ClKr ₂	Kr ₂ Cl	122	Cl ₂ OP	OPCl ₂	220
ClNO ⁺	ClNO ⁺	93	Cl ₂ OS ⁺	Cl ₂ SO ⁺	223
ClNO	ClON	B143	Cl ₂ OSi	Cl ₂ SiO	211,B238
ClNO ₂ ⁺	ClNO ₂ ⁺	208	Cl ₂ O ₂	ClOOCl	225,A263,B244
ClNO ₂	ClNO ₂	212,A259	Cl ₂ O ₂	ClClO ₂	225,A263,B244
ClNO ₂	<i>c</i> -ClONO	213,A259	Cl ₂ O ₂	ClOClO	A264
ClNO ₂	<i>t</i> -ClONO	213,A259	Cl ₂ O ₂ S ⁺	Cl ₂ SO ₂ ⁺	301
ClNO ₃ ⁺	ClONO ₂ ⁺	A301	Cl ₂ O ₂ Si	<i>cyc</i> -Cl ₂ SiO ₂	287
ClNO ₃	ClONO ₂	288,A302,B301	Cl ₂ O ₃	Cl ₂ O ₃	302,A307
ClNO ₅	O ₂ ClONO ₂	371	Cl ₂ O ₄	ClOClO ₃	355,A322
ClNS ⁺	NSCl ⁺	95	Cl ₂ O ₆	O ₃ ClOClO ₂	382
ClN ₃ ⁺	ClN ₃ ⁺	196	Cl ₂ P	PCl ₂	108,A203,B147
ClOP	ClPO	103,B142	Cl ₂ S ⁺	SCl ₂ ⁺	112
ClOSi	ClSiO	B135	Cl ₂ SSi	Cl ₂ SiS	221
ClO ₂ ⁺	OCIO ⁺	106,B144	Cl ₂ S ₂ ⁺	S ₂ Cl ₂ ⁺	224
ClO ₂	OCIO	113,A205,B149	Cl ₂ S ₂	SSCl ₂	226
ClO ₂	ClOO	110,A204,B147	Cl ₂ Se ⁺	SeCl ₂ ⁺	112
ClO ₂ ⁻	OCIO ⁻	115,B152	Cl ₂ Se ₂ ⁺	Se ₂ Cl ₂ ⁺	224
ClO ₂ P	PO ₂ Cl	213,B240	Cl ₂ Si ⁺	SiCl ₂ ⁺	92
ClO ₂ S	ClSO ₂	B243	Cl ₂ Si	SiCl ₂	101,A197
ClO ₃	ClO ₃	A262	Cl ₂ Xe	XeCl ₂	121
ClO ₃ ⁻	ClO ₃ ⁻	B244	Cl ₂ Zn ⁺	ZnCl ₂ ⁺	72
ClO ₄	OCIO ₃	A306	Cl ₃ ⁻	Cl ₃ ⁻	119,B153

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
Cl ₃ FS	SiCl ₃ F	304	CrN ₂ O ₂	Cr(NO) ₂	B293
Cl ₃ FSi ⁺	SiFCl ₃ ⁺	296	CrN ₂ O ₂ ⁻	Cr(NO) ₂ ⁻	B295
Cl ₃ F ₂ Si ⁻	SiF ₂ Cl ₃ ⁻	357	CrO ₂	OCrO	A175,B104
Cl ₃ Ge	GeCl ₃	219	CrO ₂ ⁻	OCrO ⁻	A182,B117
Cl ₃ HOSi	SiCl ₃ OH	342	CrO ₃	CrO ₃	B226
Cl ₃ HSi ⁺	HSiCl ₃ ⁺	276	CrO ₃ ⁻	CrO ₃ ⁻	B231
Cl ₃ N ⁺	NCl ₃ ⁺	219	Cr ₂ H	CrCrH	26
Cl ₃ NO ₃ Si	SiCl ₃ ONO ₂	382	Cr ₂ H ⁻	CrCrH ⁻	26
Cl ₃ NO ₄ Si	SiCl ₃ OONO ₂	426	Cr ₂ O	CrOCr	B55
Cl ₃ OP ⁺	Cl ₃ PO ⁺	298	Cr ₂ O ₂	CrOCrO	B211
Cl ₃ O ₂ P	OPCl ₂ OCl	355	Cr ₂ O ₄	Cr ₂ O ₄	B329
Cl ₃ P ⁺	PCl ₃ ⁺	220,B242	Cr ₃	Cr ₃	B51
Cl ₃ PS ⁺	Cl ₃ PS ⁺	299	Cs ₃	Cs ₃	B51
Cl ₃ Sb ⁺	SbCl ₃ ⁺	222	CuHO	CuOH	31,B27
Cl ₃ Si ⁺	SiCl ₃ ⁺	211	CuH ₂ N	CuNH ₂	133
Cl ₃ Si	SiCl ₃	218	CuH ₂ O	HCuOH	137
Cl ₄ FSi ⁻	SiFCl ₄ ⁻	358	CuH ₃ N	HCuNH ₂	232
Cl ₄ Ge ⁺	GeCl ₄ ⁺	297,A306,B306	CuNO ⁺	CuNO ⁺	B77
Cl ₄ Si ⁺	SiCl ₄ ⁺	296,A306,B305	CuNO	CuNO	B93
Cl ₄ SiO	SiCl ₃ OCl	356	CuN ₂ O ₂	Cu(NO) ₂	B294
Cl ₅ P ⁺	PCl ₅ ⁺	357	CuN ₂ O ₂ ⁻	Cu(NO) ₂ ⁻	B296
CoH ₂	CoH ₂	15,A127	CuO ₂ ⁺	CuOO ⁺	A172
CoH ₂ ⁻	CoH ₂ ⁻	17	CuO ₂	OCuO	A178,B107
CoH ₂ O	HCoOH	136	CuO ₂ ⁻	CuOO ⁻	A178
CoNO ⁺	CoNO ⁺	B77	CuO ₂ ⁻	OCuO ⁻	A182
CoNO	CoNO	B91	CuO ₂ ⁻	CuOO ⁻	A182
CoNO	<i>cyc</i> -CoNO	B91	CuO ₃	CuO ₃	A251
CoNO ⁻	CoNO ⁻	B102	CuO ₃ ⁻	CuO ₃ ⁻	A252
CoN ₂	CoNN	B73	CuO ₄	O ₂ CuO ₂	A299
CoN ₂	<i>cyc</i> -CoNN	B73	Cu ₂ H ₂	Cu ₂ H ₂	131
CoN ₂ O ₂	Co(NO) ₂	B294	Cu ₂ O	CuOCu	A163
CoN ₂ O ₂ ⁻	Co(NO) ₂ ⁻	B295	Cu ₂ O ⁻	CuOCu ⁻	A166
CoN ₃ O ₃	Co(NO) ₃	B346	Cu ₂ O ₂	Cu ₂ O ₂	A244
CoO ₂	OCoO	B106	Cu ₂ O ₂ ⁻	Cu ₂ O ₂ ⁻	A246
CoO ₂	CoOO	B106	Cu ₂ O ₃	Cu ₂ O ₃	A294,B288
CoO ₂	<i>cyc</i> -CoOO	B106	Cu ₂ O ₃ ⁻	Cu ₂ O ₃ ⁻	A295
CoO ₃	OCoOO	B227	Cu ₂ O ₄	Cu ₂ O ₄	A320
CoO ₃	<i>cyc</i> -(O ₂ Co)O	B227	Cu ₂ O ₄ ⁻	Cu ₂ O ₄ ⁻	A320
CoO ₄	OOCoo ₂	B297	Cu ₃	Cu ₃	59,A161
CoO ₄	(<i>cyc</i> -O ₂ Co)O ₂	B298	Cu ₃ H ₂	Cu ₃ H ₂	244
Co ₂ N	<i>cyc</i> -Co ₂ N	B53	DyH ₂	DyH ₂	B14
Co ₂ N	CoCoN	B53	DyH ₄	DyH ₄	B246
Co ₂ N ₂	<i>cyc</i> -(CoN) ₂	B205	DyNO	NDyO	B94
Co ₂ O ₂	<i>cyc</i> -(CoO) ₂	B212	DyNO	<i>cyc</i> -DyNO	B95
Co ₂ O ₂	CoOCoo	B212	DyN ₂	<i>cyc</i> -DyNN	B75
Co ₂ O ₃	Co ₂ O ₃	B288	DyO ₂	ODyO	B110
CrH ₂	CrH ₂	15	DyO ₂ ⁻	ODyO ⁻	B119
CrH ₂ O	HCrOH	136	Dy ₂ N	Dy ₂ N	B54
CrH ₃	CrH ₃	123	Dy ₂ N ₂	<i>cyc</i> -(DyN) ₂	B206
CrH ₃ O ₂	HCr(OH) ₂	316	Dy ₂ O ₂	<i>cyc</i> -(DyO) ₂	B213
CrNO	NCrO	B89	ErH ₂	ErH ₂	B15
CrNO	CrNO	B89	ErNO	NErO	B95
CrNO	<i>cyc</i> -CrNO	B89	ErNO	<i>cyc</i> -ErNO	B95
CrNO ⁻	CrNO ⁻	B101	ErN ₂	<i>cyc</i> -ErNN	B75
CrN ₂	NCrN	B71	ErO ₂ ⁻	OErO ⁻	B119

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
Er ₂ N ₂	<i>cyc</i> -(ErN) ₂	B207	F ₂ H ⁻	FHF ⁻	53,A156
EuH ₂	EuH ₂	B13	F ₂ HN ⁺	HNF ₂ ⁺	178
EuNO	NEuO	B94	F ₂ HP ⁺	HPF ₂ ⁺	179
EuNO	<i>cyc</i> -EuNO	B94	F ₂ H ₂ ⁺	HFFH ⁺	B178
EuN ₂	<i>cyc</i> -EuNN	B74	F ₂ H ₂ Si ⁺	SiH ₂ F ₂ ⁺	262
EuO ₂	OEuO	B109	F ₂ H ₃ P	PH ₃ F ₂	324,A315
EuO ₂ ⁻	OEuO ⁻	B119	F ₂ I ⁻	FIF ⁻	118
Eu ₂ N	Eu ₂ N	B54	F ₂ I ⁻	FFI ⁻	118
Eu ₂ O ₂	<i>cyc</i> -(EuO) ₂	B213	F ₂ I ₂	I ₂ F ₂	227
FFeH	HFeF	35	F ₂ Kr ⁺	KrF ₂ ⁺	117
FFe ₂ H	HFe ₂ F	155	F ₂ Kr	KrF ₂	120,A209
FGaO	OGaF	85	F ₂ Mg ⁺	MgF ₂ ⁺	A181
FGeH ₃ ⁺	GeH ₃ F ⁺	241	F ₂ Mg	MgF ₂	84
FHI ⁻	FHI ⁻	54	F ₂ N ⁺	NF ₂ ⁺	103,A199
FHN	HNF	48,A152	F ₂ N	NF ₂	107,A201
FHNe ⁺	NeHF ⁺	B46	F ₂ N ⁻	NF ₂ ⁻	A207
FHO ⁺	HOF ⁺	51	F ₂ NO	F ₂ NO	B241
FHO	HOF	51	F ₂ N ₂ ⁺	<i>t</i> -N ₂ F ₂ ⁺	207
FHSi	HSiF	43,A150,B39	F ₂ O ⁺	OF ₂ ⁺	111,B148
FH ₂ ⁺	H ₂ F ⁺	26	F ₂ OP ⁺	F ₂ PO ⁺	B240
FH ₂ N ⁺	NH ₂ F ⁺	151	F ₂ OP ⁻	F ₂ PO ⁻	B244
FH ₂ N	NH ₂ F	153	F ₂ OS ⁺	F ₂ SO ⁺	223,B243
FH ₂ P	H ₂ PF	153,A228	F ₂ OSi	F ₂ SiO	211
FH ₃ N ₂	NH ₂ NHF	324	F ₂ O ₂	FOOF	225
FH ₃ Si ⁺	SiH ₃ F ⁺	240	F ₂ O ₂ S ⁺	F ₂ SO ₂ ⁺	300
FI ₂	IIF	117	F ₂ O ₂ Si	<i>cyc</i> -F ₂ SiO ₂	287
FKrXe	KrXeF	121,A211	F ₂ O ₂ U	UO ₂ F ₂	B299
FKr ₂	Kr ₂ F	121,A210	F ₂ O ₂ Xe	XeO ₂ F ₂	304
FNO ⁺	FNO ⁺	93	F ₂ O ₃ Xe	XeO ₃ F ₂	358
FNO	FON	104	F ₂ P ⁺	PF ₂ ⁺	104,B143
FNO ₂ ⁺	FNO ₂ ⁺	208	F ₂ P	PF ₂	108,A202,B146
FNO ₂	FONO	212	F ₂ S ⁺	SF ₂ ⁺	112
FNO ₂ ⁺	FONO ₂ ⁺	A301	F ₂ S	SF ₂	114,A207,B151
FNO ₃ ⁺	FONO ₂ ⁺	A302	F ₂ SSi	F ₂ SiS	B239
FNS ⁺	NSF ⁺	94	F ₂ S ₂ ⁺	F ₂ SS ⁺	223
FN ₂ ⁺	FNN ⁺	87	F ₂ S ₂ ⁺	FSSF ⁺	224
FN ₃ ⁺	FN ₃ ⁺	196	F ₂ Se ⁺	SeF ₂ ⁺	112
FN ₃	FN ₃	199	F ₂ Se	SeF ₂	A208
FNeXe	NeXeF	121	F ₂ Se ₂	FSeSeF	A264
FNe ₂	Ne ₂ F	121	F ₂ Se ₂	SeSeF ₂	A264
FOP	FPO	103,B142	F ₂ Si ⁺	SiF ₂ ⁺	91
FOS	FSO	111	F ₂ Si	SiF ₂	100,A197,B140
FO ₂	FOO	109	F ₂ Xe ⁺	XeF ₂ ⁺	117
FO ₂ S ⁻	FSO ₂ ⁻	225	F ₂ Xe	XeF ₂	120,A209
FO ₂ U	UO ₂ F	B231	F ₃ ⁻	F ₃ ⁻	117
FO ₃ S ⁺	FSO ₃ ⁺	289	F ₃ HSi ⁺	HSiF ₃ ⁺	276
FO ₃ S	FSO ₃	300	F ₃ N ⁺	NF ₃ ⁺	219,A262
FPS	FPS	104,B142	F ₃ NO ⁺	F ₃ NO ⁺	298
FS ₂	SSF	A205	F ₃ NS ⁺	F ₃ NS ⁺	298
FXe ₂	Xe ₂ F	122,A211	F ₃ OP ⁺	F ₃ PO ⁺	298
F ₂ Ga	GaF ₂	A193	F ₃ OP ⁻	F ₃ PO ⁻	B307
F ₂ Ge ⁺	GeF ₂ ⁺	92	F ₃ OS ⁻	SOF ₃ ⁻	304
F ₂ Ge	GeF ₂	101,A198,B140	F ₃ O ₂ S ⁻	SO ₂ F ₃ ⁻	358
F ₂ GeH ₂ ⁺	GeH ₂ F ₂ ⁺	263	F ₃ P ⁺	PF ₃ ⁺	219,B242
F ₂ GeO	F ₂ GeO	212	F ₃ P ⁻	PF ₃ ⁻	B245

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
F ₃ PS ⁺	F ₃ PS ⁺	299	Fe ₂ H ₂ O	HFeOFeH	247,B261
F ₃ S ⁺	SF ₃ ⁺	B244	Fe ₂ H ₂ O	HFe ₂ OH	247
F ₃ S	SF ₃	226	Fe ₂ H ₃ N	HFe ₂ NH ₂	311
F ₃ Sb ⁺	SbF ₃ ⁺	222	Fe ₂ N	<i>cyc</i> -Fe ₂ N	A162
F ₃ Si ⁺	SiF ₃ ⁺	A258	Fe ₂ N	FeFeN	A162
F ₃ Si	SiF ₃	218,A262,B241	Fe ₂ O	FeOFe	A163
F ₃ Si ⁻	SiF ₃ ⁻	A263	Fe ₂ O ⁻	FeOFe ⁻	A166
F ₄ Ge ⁺	GeF ₄ ⁺	297,B306	Fe ₂ O ₂	<i>cyc</i> -(FeO) ₂	A244,B211
F ₄ OU	UOF ₄	B332	Fe ₂ O ₂ ⁻	Fe ₂ O ₂ ⁻	A246
F ₄ OXe ⁺	XeOF ₄ ⁺	359	Fe ₂ O ₃	Fe ₂ O ₃	A294
F ₄ P ⁺	PF ₄ ⁺	B307	Fe ₂ O ₃ ⁻	Fe ₂ O ₃ ⁻	A295
F ₄ P	PF ₄	B307	Fe ₂ O ₄	Fe ₂ O ₄	A320
F ₄ P ⁻	PF ₄ ⁻	304,B308	Fe ₂ O ₄ ⁻	Fe ₂ O ₄ ⁻	A320
F ₄ P ₂ ⁺	P ₂ F ₄ ⁺	355	Fe ₂ O ₅	Fe ₂ O ₅	B345
F ₄ P ₂	PF ₃ =PF	355	Fe ₂ O ₅ ⁻	Fe ₂ O ₅ ⁻	B345
F ₄ S ⁺	SF ₄ ⁺	A307,B307	Fe ₃	Fe ₃	B52
F ₄ S ⁻	SF ₄ ⁻	B308	GaHO	GaOH	36
F ₄ Si ⁺	SiF ₄ ⁺	294,A305,B304	GaH ₂	GaH ₂	18,A129
F ₄ Xe ⁺	XeF ₄ ⁺	305	GaH ₂ N	GaNH ₂	B169
F ₅ I ⁺	IF ₅ ⁺	358	GaH ₂ O	HGaOH	139,A223
F ₅ P ⁺	PF ₅ ⁺	356	GaH ₃	GaH ₃	A213,B155
F ₅ P ⁻	PF ₅ ⁻	B333	GaH ₃ N ⁺	GaNH ₃ ⁺	B251
F ₅ S ⁺	SF ₅ ⁺	B332	GaH ₃ N	GaNH ₃	B252
F ₅ S	SF ₅	358,B333	GaH ₃ N	HGaNH ₂	B252
F ₅ S ⁻	SF ₅ ⁻	359,B333	GaH ₃ P	GaPH ₃	B254
F ₅ Si ⁻	SiF ₅ ⁻	357	GaH ₃ P	HGaPH ₂	B254
F ₆ S ⁺	SF ₆ ⁺	A331,B347	GaH ₃ P	H ₂ GaPH	B254
F ₆ S ⁻	SF ₆ ⁻	B348	GaH ₄ N	H ₂ GaNH ₂	B310
FeHI	HFeI	35	GaInO	InGaO	64
FeH ₂	FeH ₂	15,A126,B12	GaNO	GaNO	B102
FeH ₂ ⁻	FeH ₂ ⁻	17	GaN ₂	NGaN	B97
FeH ₂ O	HFeOH	136,B168	GaN ₃	GaN ₃ N	B222
FeH ₂ O ₂	Fe(OH) ₂	251,B265	GaO ₂	<i>cyc</i> -GaO ₂	75
FeH ₃	FeH ₃	A211	GaO ₂	OGaO	75
FeH ₃ N	HFeNH ₂	231	GaO ₂ ⁻	<i>cyc</i> -GaO ₂ ⁻	B130
FeNO ⁺	FeNO ⁺	B76	GaO ₃	OGaOO	190
FeNO	FeNO	A171,B90	GaP ₂	<i>cyc</i> -GaP ₂	A172,B98
FeNO	<i>cyc</i> -FeNO	B91	GaSb ₂	GaSb ₂	A172
FeNO ₂	NFeO ₂	A249	Ga ₂ H ₂	<i>cyc</i> -Ga ₂ H ₂	A220
FeN ₂	NFeN	A166	Ga ₂ H ₂	HGaGaH	A220
FeN ₂	FeNN	A267	Ga ₂ H ₆ ⁺	Ga ₂ H ₆ ⁺	A333
FeN ₂ O	N ₂ FeO	A248	Ga ₂ H ₆	Ga ₂ H ₆	372,A333
FeN ₂ O ₂	Fe(NO) ₂	B293	Ga ₂ N	GaN ₂ Ga	B63
FeN ₂ O ₂	N ₂ FeO ₂	A296	Ga ₂ O	Ga ₂ O	64,B80
FeN ₃ O ₃	Fe(NO) ₃	B346	Ga ₂ O ₂	GaOGaO	185
FeO ₂	OFeO	A176	Ga ₂ O ₂	<i>cyc</i> -GaO ₂ Ga	185
FeO ₂	FeOO	A176	Ga ₂ O ₃	Ga ₂ O ₃	283
FeO ₂	<i>cyc</i> -FeO ₂	A177	Ga ₂ P	Ga ₂ P	A165,B63
FeO ₂ ⁻	OFeO ⁻	A182	Ga ₂ P ⁻	Ga ₂ P ⁻	B80
FeO ₃	FeO ₃	A250,B227	Ga ₂ P ₃	Ga ₂ P ₃	B282
FeO ₃	(<i>cyc</i> -O ₂ Fe)O	A251	Ga ₂ P ₃ ⁻	Ga ₂ P ₃ ⁻	B288
FeO ₃ ⁻	FeO ₃ ⁻	A252	Ga ₂ Sb	Ga ₂ Sb	A165
FeO ₄	(<i>cyc</i> -O ₂ Fe)O ₂	A298,B297	Ga ₃ H ₂	HGa ₃ H	A275
FeO ₄	(<i>cyc</i> -O ₂ Fe)OO	A298	Ga ₃ N	NGa ₃	B208
FeO ₄ ⁻	FeO ₄ ⁻	B298	GdH ₂	GdH ₂	B14

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
GdH ₃	GdH ₃	B155	HNO ⁻	HNO ⁻	48
GdNO	NGdO	B94	HNOS ⁺	HNSO ⁺	169
GdNO	<i>cyc</i> -GdNO	B94	HNOS	<i>t</i> -HONS	174
GdN ₂	<i>cyc</i> -GdNN	B74	HNOS	<i>t</i> -HSNO	174
GdN ₂	GdNN	B75	HNOS	<i>c</i> -HSNO	174
GdO ₂	OGdO	B109	HNOS	<i>c</i> -HNSO	175,A240
GdO ₂ ⁻	OGdO ⁻	B119	HNOS	<i>t</i> -HNSO	175
GdO ₃ ⁻	GdO ₃ ⁻	B231	HNOS	<i>c</i> -HOSN	176
Gd ₂ H ₂	<i>cyc</i> -(GdH) ₂	B161	HNO ₂	<i>t</i> -HONO	172,B196
Gd ₂ N	Gd ₂ N	B54	HNO ₂	<i>c</i> -HONO	173,B197
Gd ₂ N ₂	<i>cyc</i> -(GdN) ₂	B206	HNO ₂	HNOO	B198
GeHI	HGeI	B41	HNO ₃ ⁺	HNO ₃ ⁺	270
GeH ₂	GeH ₂	20,A132,B18	HNO ₃	HONO ₂	271,A289,B274
GeH ₂ I ₂ ⁺	GeH ₂ I ₂ ⁺	263	HNO ₃	<i>c,c</i> -HOONO	A291
GeH ₂ O	H ₂ GeO	146	HNO ₃	<i>t,perp</i> -HOONO	272,A291,B276
GeH ₂ O	HGeOH	146	HNO ₄	HOONO ₂	341,A317,B324
GeH ₂ O ₃	H ₂ GeO ₃	338	HNSc	ScNH	A138
GeH ₃ ⁺	GeH ₃ ⁺	125	HNSi	HNSi	38,A146
GeH ₃	GeH ₃	128	HNSi	HSiN	A145
GeH ₃ I ⁺	GeH ₃ I ⁺	242	HNY	YNH	27,B22
GeH ₄ ⁺	GeH ₄ ⁺	228	HN ₂ ⁺	HN ₂ ⁺	39
GeH ₄ O	GeH ₄ OH	311	HN ₂ O ⁺	HONN ⁺	165
GeH ₄ S ⁺	GeH ₃ SH ⁺	311	HN ₂ O	<i>t</i> -HNNO	B192
GeI ₂ ⁺	GeI ₂ ⁺	92	HN ₂ O	<i>c</i> -HNNO	B193
GeOS	OGeS	86	HN ₃ ⁺	HN ₃ ⁺	161
GeO ₂	GeO ₂	86,B131	HNaO ⁺	NaOH ⁺	27
GeO ₂ ⁻	GeO ₂ ⁻	A194	HNiO	NiOH	30
GeS ₂	GeS ₂	86,B132	HOP	HPO	46
Ge ₂ H ₂ O	HGe ₂ OH	256	HOS ⁺	HSO ⁺	A152
Ge ₂ O ₂	Ge ₂ O ₂	188,A251,B228	HOS	HSO	50,A154
Ge ₂ O ₂ ⁻	Ge ₂ O ₂ ⁻	A252	HOSc	HScO	B25
Ge ₂ O ₃ ⁻	Ge ₂ O ₃ ⁻	A300	HOSc	ScOH	B26
Ge ₃	Ge ₃	A171	HOSi ⁺	HOSi ⁺	39
Ge ₃ ⁻	Ge ₃ ⁻	A173	HOSi	HSiO	B36
Ge ₃ O ₃	(GeO) ₃	346,B329	HOSr	SrOH	29,A140,B23
Ge ₄	Ge ₄	A246	HOTi	HTiO	B27
Ge ₄ ⁻	Ge ₄ ⁻	B219	HOTi ⁻	HTiO ⁻	B32
HHfO	HHfO	B27	HOXe	XeOH	53
HIO ⁺	HIO ⁺	A155	HOY	YOH	B26
HIO	HOI	52,A156,B46	HOY	HYO	B26
HISi	HSiI	44,B40	HOYb	YbOH	B28
HIXe	HXeI	A158,B48	HOZn	ZnOH	31,A141
HI ₂ ⁻	IHI ⁻	55,A158	HOZr	HZrO	B27
HInO	InOH	37,A144	HO ₂ ⁺	HO ₂ ⁺	47
HKO ⁺	KOH ⁺	27	HO ₂	HO ₂	49,A153,B43
HKrXe	KrXeH	57	HO ₂ ⁻	HO ₂ ⁻	51,B44
HKr ₂ ⁺	HKr ₂ ⁺	56,A159,B48	HO ₂ P	<i>c</i> -HOPO	174,B198
HKr ₂	Kr ₂ H	57,B49	HO ₃	HO ₃	B199
HLaO	HLaO	B27	HO ₃ P	HOPO ₂	273
HLi ₂	Li ₂ H	26,A136	HO ₃ P	HOPO	273
HMgO	MgOH	27,A138,B23	HO ₃ P	HP(O ₂)O	273
HMgS	MgSH	B23	HO ₃ S	HOSO ₂	277
HNO ⁺	HNO ⁺	41	HO ₃ Sb	HSbO ₃	274
HNO	HNO	45,A151	HO ₃ Sb	HOSbO ₂	274
HNO	HON	B42	HO ₄ S	HSO ₄	B324

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
HO ₄ S ⁻	HSO ₄ ⁻	B324	H ₂ N ₂ ⁺	<i>t</i> -N ₂ H ₂ ⁺	142
HPS ₂	HPSS	A241	H ₂ N ₂	<i>t</i> -N ₂ H ₂	147,A227,B175
HPS ₃	HSPS ₂	A292	H ₂ N ₂	H ₂ NN	148
HSSi	HSiS	B36	H ₂ N ₂ O	NH ₂ NO	260
HSSr	SrSH	29,B24	H ₂ N ₂ O ₂	NH ₂ NO ₂	337
HS ₂	HS ₂	50,A154,B44	H ₂ Nb	NbH ₂	A128
HS ₂ ⁻	HS ₂ ⁻	53	H ₂ NbO	HNbOH	B167
HSi ₂	<i>cyc</i> -HSi ₂	B31	H ₂ NbO	H ₂ ONb	B167
HSi ₂ ⁻	<i>cyc</i> -HSi ₂ ⁻	B33	H ₂ NbO	H ₂ NbO	B167
HSi ₃	<i>cyc</i> -HSi ₃	B184	H ₂ Nd	NdH ₂	B13
HSi ₃ ⁻	<i>cyc</i> -HSi ₃ ⁻	B184	H ₂ Ni	NiH ₂	16,A127
HSi ₄	Si ₄ H	B272	H ₂ Ni ⁻	NiH ₂ ⁻	17
HSi ₄ ⁻	Si ₄ H ⁻	B272	H ₂ NiO	HNiOH	136
HXe ₂ ⁺	HXe ₂ ⁺	56,B49	H ₂ NiO ₂	Ni(OH) ₂	252
HXe ₂	Xe ₂ H	57	H ₂ Ni ₂ O	HNi ₂ OH	248
HZn ₂	ZnZnH	A136	H ₂ O ⁺	H ₂ O ⁺	24,A134,B20
H ₂ Hf	HfH ₂	A126	H ₂ OP	H ₂ PO	151,A228
H ₂ HfO	H ₂ HfO	B166	H ₂ OP	HPOH	151
H ₂ Hg	HgH ₂	16,A128	H ₂ OS	HSOH	154
H ₂ Hg ₂	HHgHgH	A218	H ₂ OSc	HScOH	135,B165
H ₂ Ho	HoH ₂	B14	H ₂ OSi	HSiOH	146
H ₂ IP	H ₂ PI	A229	H ₂ OSi	H ₂ SiO	145,A226
H ₂ I ₂ ⁺	IIH ⁺	B179	H ₂ OSn	HSnOH	146
H ₂ I ₂ Si ⁺	SiH ₂ I ₂ ⁺	263	H ₂ OSn ₂	HSn ₂ OH	257
H ₂ In	InH ₂	A129	H ₂ OSr	HSrOH	134
H ₂ InN	InNH ₂	B169	H ₂ OTa	H ₂ TaO	B168
H ₂ InO	HInOH	139	H ₂ OTi	HTiOH	135,B165
H ₂ LaO	HLaOH	B165	H ₂ OTi	H ₂ TiO	B166
H ₂ LaO ₂	La(OH) ₂	B265	H ₂ OV	HVOH	135,B166
H ₂ LiN	LiNH ₂	B162	H ₂ OV	H ₂ VO	B167
H ₂ Lu	LuH ₂	B15	H ₂ OXe	HXeOH	B179
H ₂ Lu ₂	<i>cyc</i> -(LuH) ₂	B162	H ₂ OY ⁺	HYOH ⁺	B164
H ₂ Mg	MgH ₂	14,A125	H ₂ OY	HYOH	B165
H ₂ MgN	MgNH ₂	B162	H ₂ OZn	HZnOH	A222
H ₂ MgO	HMgOH	134	H ₂ OZr	HZrOH	B166
H ₂ MgO ₂	Mg(OH) ₂	A282	H ₂ OZr	H ₂ ZrO	B166
H ₂ Mg ₂	HMgMgH	A217	H ₂ O ₂ ⁺	HOOH ⁺	152,B177
H ₂ Mg ₂	<i>br</i> -(MgH) ₂	A218	H ₂ O ₂	HOOH	153,A229,B177
H ₂ Mg ₂ O	HMgOMgH	246	H ₂ O ₂ S	HSO ₂ H	265
H ₂ Mg ₂ O	HMg ₂ OH	246	H ₂ O ₂ Si	HSiOOH	260
H ₂ Mn	MnH ₂	15	H ₂ O ₂ Y	Y(OH) ₂	B265
H ₂ Mn ⁻	MnH ₂ ⁻	17	H ₂ O ₃	H ₂ O ₃	B270
H ₂ MnO	HMnOH	136,B168	H ₂ O ₃ Si	H ₂ SiO ₃	338
H ₂ MnO ₂	Mn(OH) ₂	B265	H ₂ P ⁺	PH ₂ ⁺	21,B18
H ₂ Mn ₂ O	HMn ₂ OH	247	H ₂ P	PH ₂	22,A1344,B19
H ₂ Mn ₂ O	HMnOMnH	247,B261	H ₂ PS	HPSH	A228
H ₂ Mo	MoH ₂	16	H ₂ Pd	<i>cyc</i> -PdH ₂	B12
H ₂ N ⁺	NH ₂ ⁺	20,A132	H ₂ Pd ⁻	PdH ₂ ⁻	B15
H ₂ N	NH ₂	21,A133,B18	H ₂ Pd ₂	<i>cyc</i> -Pd ₂ H ₂	B161
H ₂ N ⁻	NH ₂ ⁻	26	H ₂ Pr	PrH ₂	B13
H ₂ NNa	NaNH ₂	B162	H ₂ Pr ₂	<i>cyc</i> -(PrH) ₂	B161
H ₂ NO ⁺	H ₂ NO ⁺	148	H ₂ Pt	PtH ₂	A127,B12
H ₂ NO	H ₂ NO	151	H ₂ S ⁺	H ₂ S ⁺	24,A135
H ₂ NS	H ₂ NS	B177	H ₂ SXe	HXeSH	B179
H ₂ NSr	SrNH ₂	133,B163	H ₂ S ₂ ⁺	HSSH ⁺	153

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
H ₂ S ₂	HSSH	155,A230,B178	H ₃ PS	H ₂ PSH	A275
H ₂ S ₃	<i>c</i> -HSSSH	A285,B270	H ₃ Pd	HPd(H ₂)	B153
H ₂ S ₃	<i>t</i> -HSSSH	A285,B270	H ₃ Pt	PtH ₃	B154
H ₂ Sb	SbH ₂	23	H ₃ S ⁺	H ₃ S ⁺	130,A217,B160
H ₂ Se ⁺	H ₂ Se ⁺	25	H ₃ Sb ⁺	SbH ₃ ⁺	129
H ₂ Si ⁺	SiH ₂ ⁺	18,A130	H ₃ Si ⁺	SiH ₃ ⁺	124,A213
H ₂ Si	SiH ₂	19,A131,B17	H ₃ Si	SiH ₃	126,A215,B158
H ₂ Si ⁻	SiH ₂ ⁻	21	H ₃ Si ⁻	SiH ₃ ⁻	129
H ₂ Si ₂	<i>br</i> -Si ₂ H ₂	138,A223,B170	H ₃ Sm	SmH ₃	B154
H ₂ Si ₂	Si(H)SiH	138	H ₃ Tb	TbH ₃	B155
H ₂ Sm	SmH ₂	B13	H ₃ Th	ThH ₃	A212
H ₂ Tb	TbH ₂	B14	H ₃ Ti	TiH ₃	A211
H ₂ Tb ₂	<i>cyc</i> -(TbH) ₂	B161	H ₃ U	UH ₃	A212
H ₂ Te ⁺	H ₂ Te ⁺	25	H ₃ Zr	ZrH ₃	A211
H ₂ Th	ThH ₂	A128	H ₄ Hf	HfH ₄	A265
H ₂ Ti	TiH ₂	14,A126	H ₄ HgSi	SiH ₃ HgH	B309
H ₂ Tm	TmH ₂	B15	H ₄ InN	H ₂ InNH ₂	B311
H ₂ U	UH ₂	A128	H ₄ Mg ₂	HMgH ₂ MgH	A308
H ₂ U ₂	<i>cyc</i> -U ₂ H ₂	A128	H ₄ N ⁺	NH ₄ ⁺	229,A266
H ₂ V	VH ₂	14	H ₄ N	NH ₄	229,A266,B246
H ₂ Xe	HXeH	A136,B21	H ₄ N ₂ ⁺	N ₂ H ₄ ⁺	311
H ₂ Yb	YbH ₂	B15	H ₄ N ₂ O	NH ₂ NHOH	364
H ₂ Zn	ZnH ₂	16,A127	H ₄ Nd	NdH ₄	B245
H ₂ Zn ₂	HZnZnH	130,A218	H ₄ OSi	SiH ₃ OH	311
H ₂ Zr	ZrH ₂	A126	H ₄ P ₂ ⁺	P ₂ H ₄ ⁺	311
H ₃ ⁺	H ₃ ⁺	11,A123,B9	H ₄ Pd	Pd(H ₂) ₂	B245
H ₃	H ₃	12,A123,B9	H ₄ SSi ⁺	SiH ₃ SH ⁺	310
H ₃ Hf	HfH ₃	A211	H ₄ Si ⁺	SiH ₄ ⁺	228
H ₃ ISi ⁺	SiH ₃ I ⁺	241	H ₄ Sm	SmH ₄	B245
H ₃ In	InH ₃	A213	H ₄ Th	ThH ₄	A265
H ₃ InN ⁺	InNH ₃ ⁺	B251	H ₄ Ti	TiH ₄	227,A265
H ₃ InN	InNH ₃	B253	H ₄ U	UH ₄	A265
H ₃ InN	HInNH ₂	B253	H ₄ U ₂	U ₂ H ₄	A308
H ₃ InP	InPH ₃	B255	H ₄ Zr	ZrH ₄	A265
H ₃ InP	HInPH ₂	B255	H ₅ ⁺	H ₅ ⁺	227
H ₃ InP	H ₂ InPH	B255	H ₅ NSi	SiH ₃ NH ₂	360
H ₃ LaO ₂	HLa(OH) ₂	B319	H ₅ O ₂ ⁺	H ₅ O ₂ ⁺	360,A323
H ₃ Lu	LuH ₃	B155	H ₅ Si ⁺	SiH ₅ ⁺	A307
H ₃ Mo	MoH ₃	124	H ₆ OSi ₂	SiH ₃ SiH ₂ OH	417
H ₃ N ⁺	NH ₃ ⁺	128,A217,B159	H ₆ Pd	Pd(H ₂) ₃	B333
H ₃ NNi	HNiNH ₂	231	H ₇ O ₃ ⁺	H ₇ O ₃ ⁺	414
H ₃ NO ⁺	NH ₂ OH ⁺	242	H ₉ O ₄ ⁺	H ₉ O ₄ ⁺	414,A364
H ₃ NSi	HSiNH ₂	235	HfNO	NHfO	B88
H ₃ NSi	SiH ₂ NH	A270	HfN ₂	HfNN	B70
H ₃ N ₂ ⁺	N ₂ H ₃ ⁺	236	HfN ₂	NHfN	B70
H ₃ Nd	NdH ₃	B154	HfO ₂	OHfO	A176
H ₃ O ⁺	H ₃ O ⁺	129,B160	HfO ₂ ⁻	OHfO ⁻	A182
H ₃ OP	PH ₃ O	242,B261	HfO ₃	OHfOO	A250
H ₃ OP	<i>c</i> -H ₂ POH	243	Hf ₂ N ₂	<i>cyc</i> -(HfN) ₂	B205
H ₃ OSb	H ₃ SbO	243	Hf ₂ O ₂	<i>cyc</i> -(HfO) ₂	A244
H ₃ OSb	H ₂ SbOH	244	Hf ₃	Hf ₃	B51
H ₃ O ₂ Sc	HSc(OH) ₂	B318	Hg ₃	Hg ₃	B53
H ₃ O ₂ Y	HY(OH) ₂	B319	HoNO	NHoO	B95
H ₃ O ₃ P	(HO) ₂ HPO	368	HoNO	<i>cyc</i> -HoNO	B95
H ₃ P ⁺	PH ₃ ⁺	129	HoN ₂	<i>cyc</i> -HoNN	B75

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
HoO ₂	OHoO	B110	LaO ₂	OLaO	B103
HoO ₂ ⁻	OHoO ⁻	B119	LaO ₂ ⁻	OLaO ⁻	B116
Ho ₂ N	Ho ₂ N	B55	La ₂ N ₂	<i>cyc</i> -(LaN) ₂	B204
Ho ₂ N ₂	<i>cyc</i> -(HoN) ₂	B206	LiNO ₃	<i>c</i> -LiOONO	A295
IKrXe	KrXeI	123	LiNO ₃	<i>t</i> -LiOONO	A295
INO	INO	102	LiNa ₃	LiNa ₃	180
INO ₂	INO ₂	213	LiOSi	LiOSi	B58
INO ₃	IONO ₂	289	Li ₂ Na ₂	Li ₂ Na ₂	180
INS ⁺	NSI ⁺	95	Li ₂ O	LiOLi	B53
INS	NSI	105	Li ₃	Li ₃	58,A159,B49
IOP	IPO	B142	Li ₄	Li ₄	180
IO ₂	OIO	114,A207	Li ₆	Li ₆	342
IO ₂	IOO	A204	Li ₇	Li ₇	369
IO ₂ ⁻	OIO ⁻	115	Li ₈	Li ₈	381
IXe ₂	Xe ₂ I	123	LuNO	NLuO	B95
I ₂ Li	ILiI	B116	LuN ₂	<i>cyc</i> -LuNN	B75
I ₂ Na	INaI	B116	LuO ₂ ⁻	OLuO ⁻	B120
I ₂ S	SI ₂	115	Lu ₂ N	Lu ₂ N	B55
I ₂ Si ⁺	SiI ₂ ⁺	92	Lu ₂ N ₂	<i>cyc</i> -(LuN) ₂	B207
I ₃	I ₃	B153	Lu ₂ O ₂	<i>cyc</i> -(LuO) ₂	B213
I ₃ ⁻	I ₃ ⁻	119,B153	MgNO	MgNO	B87
InNO	InNO	B102	MgO ₂	OMgO	A174
InN ₂	NInN	B97	MgO ₂	<i>cyc</i> -MgO ₂	A174
InN ₃	InNNN	B222	MgO ₄	O ₂ MgO ₂	A298
InO ₂	<i>cyc</i> -InO ₂	75	Mg ₂ O ₂	MgOMgO	A243
InO ₂	OInO	75	MnNO	NMnO	B90
InO ₂ ⁻	<i>cyc</i> -InO ₂ ⁻	B130	MnNO	MnNO	B90
InP ₂	<i>cyc</i> -InP ₂	B98	MnNO	<i>cyc</i> -MnNO	B90
InP ₂ ⁻	<i>cyc</i> -InP ₂ ⁻	B111	MnN ₂	NMnN	B72
In ₂ N	InNIn	B63	MnN ₂	<i>cyc</i> -MnNN	B72
In ₂ O	In ₂ O	64,B81	MnN ₂ O ₂	Mn(NO) ₂	B293
In ₂ O ₂	InOInO	185	MnN ₃ O ₃	Mn(NO) ₃	B346
In ₂ O ₂	<i>cyc</i> -InO ₂ In	185	MnO ₂	OMnO	B105
In ₂ O ₃	In ₂ O ₃	283	MnO ₂	<i>cyc</i> -MnO ₂	B105
In ₂ P	In ₂ P	A165,B64	MnO ₂	MnOO	B105
In ₂ P ⁻	In ₂ P ⁻	B80	MnO ₂ ⁻	OMnO ⁻	B118
In ₂ Sb	In ₂ Sb	A165	MnO ₃	(<i>cyc</i> -O ₂ Mn)O	B226
In ₃ N	NIn ₃	B208	MnO ₄	(<i>cyc</i> -O ₂ Mn)O ₂	B297
IrNO ⁺	IrNO ⁺	B77	Mn ₂ O	MnOMn	B55
IrNO	IrNO	B92	Mn ₂ O ₂	<i>cyc</i> -(MnO) ₂	B211
IrNO	NIrO	B92	Mn ₂ O ₂	MnOMnO	B211
IrN ₂ O ₂	Ir(NO) ₂	B294	Mn ₃	Mn ₃	61
IrN ₃ O ₃	Ir(NO) ₃	B346	MoNO	NMoO	B89
IrO ₂	OIrO	B106	MoNO	MoNO	B89
IrO ₂	<i>cyc</i> -IrOO	B106	MoN ₂	NMoN	B71
Ir ₂ O	<i>cyc</i> -Ir ₂ O	B55	MoN ₂	MoNN	B71
KNO ₂	<i>cyc</i> -KNO ₂	A247	MoN ₂ O ₂	Mo(NO) ₂	B293
KNO ₂	<i>t</i> -KNO ₂	A247	MoO ₂	OMoO	A175,B104
KNO ₃	<i>c</i> -KOONO	A296	MoO ₂ ⁻	OMoO ⁻	B117
KNO ₃	<i>t</i> -KOONO	A296	MoO ₃	MoO ₃	A250,B226
K ₃	K ₃	B51	NNaO ₂	<i>cyc</i> -NaNO ₂	A247
K ₄	K ₄	B200	NNaO ₂	<i>t</i> -NaONO	A247
LaN ₂	<i>cyc</i> -LaNN	B69	NNaO ₃	<i>c</i> -NaOONO	A295
LaO ₂ ⁺	<i>cyc</i> -LaO ₂ ⁺	B96	NNaO ₃	<i>t</i> -NaOONO	A296
LaO ₂ ⁺	OLaO ⁺	B96	NNbO	NNbO	B89

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
NNdO	NNdO	B93	NOV	NVO	B88
NNd ₂	Nd ₂ N	B54	NOV	VNO	B88
NNiO ⁺	NiNO ⁺	B77	NOV	<i>cyc</i> -VNO	B88
NNiO	NiNO	B92	NOW	NWO	B90
NNiO	<i>cyc</i> -NiNO	B92	NOYb	NYbO	B95
NNiO ⁻	NiNO ⁻	B102	NOZr	NZrO	B88
NNi ₂	NiNiN	B54	NO ₂ ⁺	NO ₂ ⁺	87,A190,B132
NOOs ⁺	OsNO ⁺	B77	NO ₂ ⁻	NO ₂ ⁻	102
NOOs	OsNO	B91	NO ₂ S	<i>c</i> -OSNO	B237
NOOs	NOsO	B91	NO ₂ S	<i>t</i> -OSNO	B237
NOOs	<i>cyc</i> -OsNO	B91	NO ₂ U	NUO ₂	A249
NOP	PNO	88,B133	NO ₃ ⁺	NO ₃ ⁺	A255
NOPd ⁺	PdNO ⁺	B77	NO ₃	NO ₃	207,A257,B237
NOPd	PdNO	B92	NO ₃ ⁻	NO ₃ ⁻	A259
NOPd ⁻	PdNO ⁻	B102	NO ₃ ⁻	<i>c</i> -OONO ⁻	B239
NOPr	NPrO	B93	NO ₃ ⁻	<i>t</i> -OONO ⁻	B239
NOPr	PrNO	B93	NP ₂	PNP	A187
NOPr ⁻	NPrO ⁻	B102	NPr ₂	Pr ₂ N	B54
NOPt ⁺	PtNO ⁺	B77	NPt ₂	Pt ₂ N	B54
NOPt	PtNO	B93	NPt ₂	PtPtN	B54
NOPt ⁻	PtNO ⁻	B102	NRh ₂	<i>cyc</i> -RhRhN	B53
NORe	NReO	B90	NRh ₂	RhRhN	B53
NORe	<i>cyc</i> -ReNO	B90	NS ₂ ⁺	SNS ⁺	A191
NORe ⁻	NReO ⁻	B101	NS ₂	NS ₂	94,A195
NORh ⁺	RhNO ⁺	B77	NS ₂	NSS	94
NORh	RhNO	B92	NS ₂ ⁻	NS ₂ ⁻	102
NORh	NRhO	B92	NS ₃	NSSS	208
NORu ⁺	RuNO ⁺	B77	NSe ₂ ⁺	SeNSe ⁺	A191
NORu	RuNO	B91	NSe ₂	NSe ₂	A195
NORu	NRuO	B91	NSe ₂	SeSeN	A195
NOS ⁺	SNO ⁺	A191	NSe ₃	SeSeNSe	A258
NOS	SNO	93,A194,B136	NSe ₃	SeSeSeN	A258
NOS	NSO	94	NSi ₂	SiNSi	A174,B101
NOS ₂	SSNO	A258	NTb ₂	Tb ₂ N	B54
NOSc ⁺	<i>cyc</i> -ScNO ⁺	B76	NTl ₃	NTl ₃	B208
NOSc	NScO	B87	N ₂ Nb	NbNN	B70
NOSc	ScNO	B87	N ₂ Nb	<i>cyc</i> -NbN ₂	B71
NOSc	<i>cyc</i> -ScNO	B88	N ₂ Nb	NNbN	B71
NOSe	SeNO	A195	N ₂ NbO ₂	Nb(NO) ₂	B292
NOSm	NSmO	B94	N ₂ Nb ₃ ⁺	Nb ₃ N ₂ ⁺	B277
NOTa	NTaO	B89	N ₂ Nd	NNdN	B74
NOTa ⁻	NTaO ⁻	B101	N ₂ Nd ₂	<i>cyc</i> -(NdN) ₂	B206
NOTb	NTbO	B94	N ₂ Ni	NiNN	B73
NOTb	<i>cyc</i> -TbNO	B94	N ₂ Ni	<i>cyc</i> -NiNN	B73
NOTh	NThO	B96	N ₂ NiO ₂ ⁺	Ni(NO) ₂ ⁺	B292
NOTi	NTiO	B88	N ₂ NiO ₂	Ni(NO) ₂	B294
NOTi	TiNO	B88	N ₂ NiO ₂ ⁻	Ni(NO) ₂ ⁻	B296
NOTi ⁻	NTiO ⁻	B101	N ₂ Ni ₂	<i>cyc</i> -(NiN) ₂	B205
NOTl	TlNO	B103	N ₂ O ⁺	N ₂ O ⁺	83,A187,B127
NOTm	NTmO	B95	N ₂ OS	SNNO	A254
NOU ⁺	NUO ⁺	B78	N ₂ O ₂ ⁺	ONNO ⁺	195,A252,B232
NOU	NUO	A172,B96	N ₂ O ₂	<i>c</i> -(NO) ₂	200,A254,B233
NOU	UNO	A172	N ₂ O ₂	<i>t</i> -(NO) ₂	200,A254,B234
NOV ⁺	<i>cyc</i> -VNO ⁺	B76	N ₂ O ₂	NNO ₂	A254
NOV ⁺	NVO ⁺	B76	N ₂ O ₂ ⁻	NNO ₂ ⁻	207,A256,B236

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
N ₂ O ₂ ⁻	<i>t</i> -(NO) ₂ ⁻	207,A257,B236	N ₂ Sm	NSmN	B74
N ₂ O ₂ ⁻	<i>c</i> -(NO) ₂ ⁻	B236	N ₂ Sm	SmNN	B74
N ₂ O ₂ Os	Os(NO) ₂	B294	N ₂ Sm	<i>cyc</i> -SmNN	B74
N ₂ O ₂ Pd	Pd(NO) ₂	B294	N ₂ Sm ₂	<i>cyc</i> -(SmN) ₂	B206
N ₂ O ₂ Pt	Pt(NO) ₂	B294	N ₂ Ta	TaNN	B71
N ₂ O ₂ Re	Re(NO) ₂	B293	N ₂ Ta	<i>cyc</i> -TaN ₂	B71
N ₂ O ₂ Rh	Rh(NO) ₂	B294	N ₂ Ta	NTaN	B71
N ₂ O ₂ Ru	Ru(NO) ₂	B293	N ₂ Tb	<i>cyc</i> -TbNN	B75
N ₂ O ₂ S	ON-NSO	286	N ₂ Tb ₂	<i>cyc</i> -(TbN) ₂	B206
N ₂ O ₂ S	ONSNO	286	N ₂ Th	NThN	B76
N ₂ O ₂ Ta	Ta(NO) ₂	B293	N ₂ Th	ThNN	B76
N ₂ O ₂ Ti	O ₂ TiN ₂	A296	N ₂ Th ₂	<i>cyc</i> -(ThN) ₂	B207
N ₂ O ₂ U ₂	(NUO) ₂	A320	N ₂ Ti	TiNN	A166,B69
N ₂ O ₂ V	V(NO) ₂	B292	N ₂ Ti	<i>cyc</i> -TiNN	B69
N ₂ O ₂ W	W(NO) ₂	B293	N ₂ Ti	NTiN	B69
N ₂ O ₃	O ₂ N-NO	286,A300,B299	N ₂ Ti ₂	<i>cyc</i> -(TiN) ₂	B204
N ₂ O ₃	O=N-O-N=O	286,A300,B299	N ₂ Tm	<i>cyc</i> -TmNN	B75
N ₂ O ₃	<i>t, c</i> -ONONO	B300	N ₂ Tm ₂	<i>cyc</i> -(TmN) ₂	B207
N ₂ O ₄ ⁺	N ₂ O ₄ ⁺	349	N ₂ U	NUN	63,B76
N ₂ O ₄	N ₂ O ₄	350,A321,B330	N ₂ V	NVN	B70
N ₂ O ₄	N ₂ O ₄ (V _d)	350	N ₂ V	VNN	B70
N ₂ O ₄	ONO-NO ₂ (D)	351	N ₂ V	<i>cyc</i> -VNN	B70
N ₂ O ₄	ONO-NO ₂ (D')	351	N ₂ W	NWN	B72
N ₂ O ₅	O ₂ N-O-NO ₂	370,A331,B347	N ₂ W	WNN	B72
N ₂ Os	NOsN	B72	N ₂ Y	<i>cyc</i> -YNN	B69
N ₂ Pd	PdNN	B73	N ₂ Y ₂ ⁺	<i>cyc</i> -(YN) ₂ ⁺	B201
N ₂ Pr	NPrN	B74	N ₂ Y ₂	<i>cyc</i> -(YN) ₂	B204
N ₂ Pr ₂	<i>cyc</i> -(PrN) ₂	B206	N ₂ Yb	<i>cyc</i> -YbNN	B75
N ₂ Pt	PtNN	B73	N ₂ Zr	ZrNN	B69
N ₂ Pt ⁻	PtNN ⁻	B87	N ₂ Zr	<i>cyc</i> -ZrNN	B70
N ₂ Pt ₂	Pt ₂ NN	B205	N ₂ Zr	NZrN	B70
N ₂ Pt ₂	PtNNPt	B205	N ₂ Zr ₂	<i>cyc</i> -(ZrN) ₂	B205
N ₂ Pu	NPuN	A168	N ₃ ⁺	N ₃ ⁺	72,A181
N ₂ Re	ReNN	B72	N ₃	N ₃	82,B127
N ₂ Re	<i>cyc</i> -ReN ₂	B72	N ₃ ⁻	N ₃ ⁻	87,B132
N ₂ Rh	RhNN	B73	N ₃ NbO ₃	Nb(NO) ₃	B345
N ₂ Rh	NRhN	B73	N ₃ O ₃ ⁺	(NO) ₃ ⁺	B330
N ₂ Rh ₂	<i>cyc</i> -(RhN) ₂	B205	N ₃ O ₃ ⁻	<i>c</i> -(NO) ₃ ⁻	B330
N ₂ Ru	NRuN	B72	N ₃ O ₃ Os	Os(NO) ₃	B346
N ₂ S ⁺	NNS ⁺	83	N ₃ O ₃ Rh	Rh(NO) ₃	B346
N ₂ S	NNS	88	N ₃ O ₃ Ru	Ru(NO) ₃	B346
N ₂ S ₂ ⁺	N ₂ S ₂ ⁺	196	N ₃ O ₃ Ta	Ta(NO) ₃	B346
N ₂ S ₂	NS-SN	200	N ₃ O ₃ V	V(NO) ₃	B345
N ₂ S ₄ ⁺	N ₂ S ₄ ⁺	349	N ₃ P ₃	(PN) ₃	346
N ₂ S ₄	<i>cyc</i> -N ₂ S ₄	351	N ₃ Pt	PtNNN	B215
N ₂ Sc	ScNN	B69	N ₃ Pt	NNPtN	B216
N ₂ Sc	<i>cyc</i> -ScNN	B69	N ₃ S ₃ ⁺	S ₃ N ₃ ⁺	347
N ₂ Sc ₂ ⁺	<i>cyc</i> -(ScN) ₂ ⁺	B201	N ₃ Sr	SrN ₃	183
N ₂ Sc ₂	<i>cyc</i> -(ScN) ₂	B204	N ₃ Tl	TlNNN	B222
N ₂ Se ₂	NSeNSe	A255	N ₄ ⁺	N ₄ ⁺	187,A250,B225
N ₂ Si	SiNN	72,B114	N ₄	N ₄	B230
N ₂ Si	<i>cyc</i> -SiN ₂	B115	N ₄ Pt	NNPtNN	B287
N ₂ Si ₂	SiNNSi	B223	N ₄ Pt ⁻	Pt(NN) ₂ ⁻	B291
N ₂ Si ₂	<i>cyc</i> -(SiN) ₂	B223	N ₄ Si	Si(N ₂) ₂	B296
N ₂ Si ₂	SiNSiN	B223	N ₄ Si ₂	(SiNN) ₂	B329

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
N ₄ Th	NNThNN	B287	O ₂ P	PO ₂	93,A194,B135
Na ₃	Na ₃	58,A160,B49	O ₂ P ⁻	PO ₂ ⁻	103,A199,B141
Na ₄	Na ₄	180	O ₂ P ₂	(PO) ₂	200
Na ₅	Na ₅	277	O ₂ Pb	OPbO	A190
Na ₆	Na ₆	342	O ₂ Pb	<i>cyc</i> -PbO ₂	A190
Na ₇	Na ₇	369	O ₂ Pb ₂	Pb ₂ O ₂	A252
Na ₈	Na ₈	381	O ₂ Pb ₂	PbPb(O ₂)	A252
NbO ₂ ⁺	ONbO ⁺	B96	O ₂ Pd	<i>cyc</i> -PdO ₂	A177,B107
NbO ₂	ONbO	B104	O ₂ Pr ⁺	OPrO ⁺	B97
NbO ₂ ⁻	ONbO ⁻	B117	O ₂ Pr	OPrO	A179,B109
NbO ₃ ⁻	NbO ₃ ⁻	B231	O ₂ Pr	<i>cyc</i> -PrO ₂	B109
NbO ₄	(<i>cyc</i> -O ₂ Nb)O ₂	B296	O ₂ Pr ⁻	OPrO ⁻	B118
Nb ₂ O ₂	NbONbO	B211	O ₂ Pr ₂	<i>cyc</i> -(PrO) ₂	B212
Nb ₃	Nb ₃	A162	O ₂ Pt	OPtO	B107
Nb ₃ O ⁺	Nb ₃ O ⁺	B200	O ₂ Pt	<i>cyc</i> -PtO ₂	A177,B107
Nb ₃ O	Nb ₃ O	B200	O ₂ Pt ₂	<i>cyc</i> -(PtO) ₂	B212
Nb ₃ O ⁻	Nb ₃ O ⁻	B201	O ₂ Pu	OPuO	A179
Nb ₈	Nb ₈	B360	O ₂ Re	OReO	B105
Nb ₈ ⁻	Nb ₈ ⁻	B360	O ₂ Re ⁻	OReO ⁻	B118
NdO ₂ ⁺	ONdO ⁺	B97	O ₂ Rh	ORhO	B106
NdO ₂	ONdO	B109	O ₂ Rh ₂	<i>cyc</i> -(RhO) ₂	B212
NdO ₂ ⁻	ONdO ⁻	B118	O ₂ Ru	ORuO	B105
Nd ₂ O ₂	<i>cyc</i> -(NdO) ₂	B212	O ₂ Ru ⁻	ORuO ⁻	B118
NiO ₂	ONiO	A177,B106	O ₂ S ⁺	SO ₂ ⁺	96,A195,B136
NiO ₂	<i>cyc</i> -NiO ₂	A177,B107	O ₂ S	SOO	A199
NiO ₂	NiOO	A177	O ₂ S ⁻	SO ₂ ⁻	110,B148
NiO ₂ ⁻	ONiO ⁻	A182	O ₂ Sc ⁺	<i>cyc</i> -ScO ₂ ⁺	B96
NiO ₂ ⁻	<i>cyc</i> -NiO ₂	A182	O ₂ Sc	OScO	B103
NiO ₃	ONiOO	A251	O ₂ Sc ⁻	OScO ⁻	B116
NiO ₃	<i>cyc</i> -(O ₂ Ni)O	A251	O ₂ Sc ₂	ScOScO	B210
NiO ₄	O ₂ NiO ₂	A298	O ₂ Sc ₂	<i>cyc</i> -(ScO) ₂	B210
Ni ₂ O	NiNiO	A163	O ₂ Se	SeO ₂	A200,B144
Ni ₂ O ₂	<i>cyc</i> -(NiO) ₂	A244	O ₂ Se	SeOO	A201
Ni ₂ O ₂	NiONiO	A244	O ₂ Se ⁻	SeO ₂ ⁻	A204
Ni ₂ O ₃	Ni ₂ O ₃	B288	O ₂ Se ₂	<i>cyc</i> -(SeOSe)=O	A261
Ni ₃	Ni ₃	61	O ₂ Si	SiO ₂	86
OP ₂	P ₂ O	88,A191,B133	O ₂ Si ⁻	SiO ₂ ⁻	A194
OP ₄	P ₄ O	284	O ₂ Si ₂	Si ₂ O ₂	188
OP ₄	br-P ₄ O	284	O ₂ Sm	OSmO	B109
OPb ₂	PbOPb	A181	O ₂ Sm ⁻	OSmO ⁻	B119
OSSi	OSiS	86	O ₂ Sm ₂	<i>cyc</i> -(SmO) ₂	B212
OS ₂ ⁺	SSO ⁺	97	O ₂ Sn	OSnO	A189
OS ₂	SSO	105,A199,B143	O ₂ Sn ₂	Sn ₂ O ₂	A251
OS ₂ ⁻	SSO ⁻	110	O ₂ Sr	OSrO	A174
OSc ₂	ScOSc	B55	O ₂ Sr	<i>cyc</i> -SrO ₂	A175
OSeSi	OSiSe	A189	O ₂ Sr ₂	<i>cyc</i> -(SrO) ₂	A243
OSe ₂	SeSeO	A201	O ₂ Ta ⁺	OTaO ⁺	B97
OTa ₃	Ta ₃ O	B200	O ₂ Ta	OTaO	A176,B104
OTa ₃ ⁻	Ta ₃ O ⁻	B201	O ₂ Ta ⁻	OTaO ⁻	B117
OTi ₂	TiOTi	A163	O ₂ Ta ₂	<i>cyc</i> -(TaO) ₂	B211
OTl ₂	Tl ₂ O	65,B81	O ₂ Tb	OTbO	A179,B110
OV ₃	V ₃ O	B200	O ₂ Tb ⁻	OTbO ⁻	B119
OV ₃ ⁻	V ₃ O ⁻	B201	O ₂ Tb ₂	<i>cyc</i> -(TbO) ₂	B213
O ₂ Os	OOsO	B105	O ₂ Te	TeO ₂	B144
O ₂ Os ⁻	OOsO ⁻	B118	O ₂ Th	OThO	A179,B110

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
O ₂ Ti	OTiO	A176,B103	O ₃ Tb ⁻	TbO ₃ ⁻	B231
O ₂ Ti ⁻	OTiO ⁻	A181	O ₃ Ti	OTiOO	A250
O ₂ Ti ₂	<i>cyc</i> -(TiO) ₂	A243	O ₃ U	UO ₃	A251,B228
O ₂ Tl	<i>cyc</i> -TlO ₂	B121	O ₃ V	VO ₃	B226
O ₂ Tl	OTlO	B121	O ₃ V ⁻	VO ₃ ⁻	B230
O ₂ Tl ⁻	<i>cyc</i> -TlO ₂ ⁻	B131	O ₃ V ₂	OVOVO	B287
O ₂ Tl ₂	TlOTlO	B222	O ₃ W	WO ₃	A250,B226
O ₂ Tl ₂	<i>cyc</i> -TlO ₂ Tl	B223	O ₃ Y ⁻	YO ₃ ⁻	B230
O ₂ Tm	OTmO	B110	O ₃ Zr	OZrOO	A250
O ₂ Tm ₂	<i>cyc</i> -(TmO) ₂	B213	O ₄ ⁺	<i>t</i> -O ₄ ⁺	209,A258,B238
O ₂ U ⁺	OUO ⁺	B97	O ₄ ⁺	<i>cyc</i> -O ₄ ⁺	209,A258,B238
O ₂ U	OUO	A179,B110	O ₄ ⁻	<i>cyc</i> -O ₄ ⁻	222,B243
O ₂ U ⁻	OUO ⁻	B120	O ₄ Os	(<i>cyc</i> -O ₂ Os)O ₂	B297
O ₂ V	OVO	B104	O ₄ P ₂	O ₂ PPO ₂	B331
O ₂ V ⁻	OVO ⁻	B117	O ₄ Pd	O ₂ PdO ₂	A298,B298
O ₂ V ₂	VOVO	B210	O ₄ Pt	O ₂ PtO ₂	A298,B298
O ₂ V ₂	<i>cyc</i> -(VO) ₂	B211	O ₄ Re	(<i>cyc</i> -O ₂ Re)O ₂	B297
O ₂ W	OWO	A175,B105	O ₄ Re ⁻	ReO ₄ ⁻	B298
O ₂ W ⁻	OWO ⁻	B117	O ₄ Rh	O ₂ RhO ₂	A298,B298
O ₂ Y ⁺	<i>cyc</i> -YO ₂ ⁺	B96	O ₄ Ru	(<i>cyc</i> -O ₂ Ru)O ₂	B297
O ₂ Y	OYO	B103	O ₄ Ru	RuO ₄	B297
O ₂ Y ⁻	OYO ⁻	B116	O ₄ S	SO ₄	289,B302
O ₂ Yb	OYbO	B110	O ₄ S ⁻	SO ₄ ⁻	B306
O ₂ Yb ⁻	OYbO ⁻	B119	O ₄ S ₂ ⁻	(SO ₂) ₂ ⁻	B332
O ₂ Yb ₂	<i>cyc</i> -(YbO) ₂	B213	O ₄ Si ₂	Si ₂ O ₄	A321
O ₂ Zn	OZnO	A178	O ₄ Si ₂ ⁻	Si ₂ O ₄ ⁻	A321
O ₂ Zn ₂	ZnOZnO	A244	O ₄ Ta	(<i>cyc</i> -O ₂ Ta)O ₂	B296
O ₂ Zr	OZrO	A176,B103	O ₄ V	(<i>cyc</i> -O ₂ V)O ₂	B296
O ₂ Zr ⁻	OZrO ⁻	A182,B117	PS ₂	PS ₂	95
O ₂ Zr ₂	<i>cyc</i> -(ZrO) ₂	A243	PS ₂ ⁻	PS ₂ ⁻	104
O ₃ ⁺	O ₃ ⁺	95	P ₂ S	P ₂ S	88
O ₃ ⁻	O ₃ ⁻	109,A203	P ₂ S ₄	P ₂ S ₄	351
O ₃ Os	OsO ₃	B227	P ₃	P ₃	82
O ₃ P	PO ₃	209,B238	P ₄ ⁺	P ₄ ⁺	187
O ₃ P ⁻	PO ₃ ⁻	B240	P ₄	P ₄	B230
O ₃ P ₂	OPOPO	B300	P ₄ S	<i>cyc</i> -P ₄ S	284
O ₃ Pb	OPb(O ₂)	A253	Pd ₃	Pd ₃	61
O ₃ Pr ⁻	PrO ₃ ⁻	B231	Pt ₃	Pt ₃	61
O ₃ Pt	PtO ₃	B227	Rb ₃	Rb ₃	B51
O ₃ Pt	<i>cyc</i> -(O ₂ Pt)O	B227	Rh ₃	Rh ₃	B52
O ₃ Pt	OOPtO	B228	Ru ₃	Ru ₃	B52
O ₃ Re	ReO ₃	B226	S ₂ Si	SiS ₂	86,B131
O ₃ Re ⁻	ReO ₃ ⁻	B231	S ₃	S ₃	105,A200
O ₃ Rh	ORhOO	B227	S ₃ ⁻	S ₃ ⁻	110,B148
O ₃ Ru	RuO ₃	B227	S ₄	S ₄	214,A260
O ₃ S ⁺	SO ₃ ⁺	209	S ₄	SS ₃	214
O ₃ S	OSOO	A260	Sb ₃	Sb ₃	82
O ₃ S ⁻	SO ₃ ⁻	223	Sb ₃ ⁻	Sb ₃ ⁻	88
O ₃ Sc	<i>cyc</i> -(O ₂ Sc)O	B225	Sb ₄	Sb ₄	189
O ₃ Se	SeO ₃	214,A260	Sb ₄ ⁻	Sb ₄ ⁻	197
O ₃ Se	OSeOO	A261	Sc ₃	Sc ₃	61
O ₃ Si	SiO ₃	A253	Se ₂ Si	SiSe ₂	A189
O ₃ Si ₂ ⁻	Si ₂ O ₃ ⁻	A300	Se ₃	Se ₃	106
O ₃ Si ₃	(SiO) ₃	346	Se ₄	Se ₄	214
O ₃ Ta ⁻	TaO ₃ ⁻	B231	Se ₄	SeSe ₃	215

<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>	<i>Formula</i>	<i>Structure/Name</i>	<i>References</i>
Si ₃	Si ₃	67,A170,B83	Sn ₃	Sn ₃	B84
Si ₃ ⁻	Si ₃ ⁻	68,A173	Sn ₃ ⁻	Sn ₃ ⁻	B99
Si ₄	Si ₄	182,A245,B215	Sn ₄ ⁻	Sn ₄ ⁻	B219
Si ₄ ⁻	Si ₄ ⁻	183,A247,B218	Ta ₃	Ta ₃	B51
Si ₅	Si ₅	B281	Ta ₄	Ta ₄	A241
Si ₅ ⁻	Si ₅ ⁻	B283	Te ₃	Te ₃	106
Si ₆	Si ₆	A318,B326	Te ₄	Te ₄	215
Si ₇	Si ₇	A329,B344	Te ₄	TeTe ₃	215
Si ₇ ⁻	Si ₇ ⁻	B344			