



**207<sup>TH</sup> MEETING OF THE  
ELECTROCHEMICAL SOCIETY**

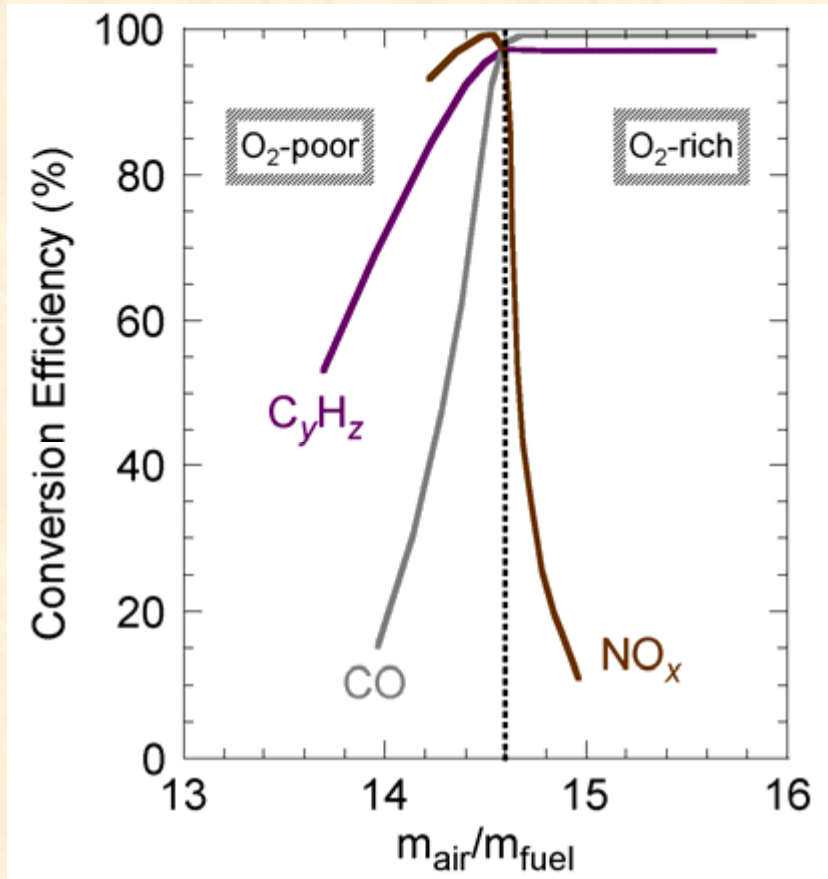
# All-Oxide “Total NO<sub>x</sub>” Sensing Elements

**Dave West, Fred Montgomery, Tim Armstrong**

Oak Ridge National Laboratory

# Motivation for NO<sub>x</sub> sensor development

## Three-way catalyst (TWC) efficiency†



- ❖ Diesel and lean-burn exhausts are  $O_2$ -rich, TWC cannot remove  $NO_x$ .
- ❖ Proposed remediation approaches: Lean  $NO_x$  traps (LNT) and selective catalytic reduction (SCR).
- ❖ Both will require  $NO_x$  sensors.
- ❖ Operating conditions ("rough"):  $T \sim 600^\circ\text{C}$ ,  $\sim 10\text{--}1000 \text{ ppm}_V NO_x$ , varying  $[O_2]$ .‡
- ❖ " $NO_x$ " is a mixture of NO and  $NO_2$  (primarily NO), but usually cannot assume  $[NO]/[NO_2]$ .
- ❖ May need NO or  $NO_2$ -selective, and/or "total  $NO_x$ " sensors for  $[NO_x]$  characterization.

†Woestman and Logothetis, *The Industrial Physicist*, 1, (1995).

‡Menil, Coillard, and Lucat, *Sensors and Actuators B*, 67, (2000).

# Experimental approach and methodology

## ❖ General approach

- ◆ Sensing elements using YSZ substrates.
  - ❖ Co-planar interdigitated oxide electrodes.
  - ❖ Electrodes identical in composition.
- ◆ Use applied DC electrical signal ("bias").<sup>†</sup>

## ❖ Methodology

- ◆ Electrodes applied by screen printing and thermal treatment.
- ◆ Tube furnace used to simulate elevated temperature service.
- ◆ Typical operating conditions:
  - ❖  $550\text{ °C} \leq T \leq 650\text{ °C}$ .
  - ❖ Mixtures of NO or NO<sub>2</sub>, O<sub>2</sub> (7–20 vol%), and N<sub>2</sub>.

## ❖ Performance metrics

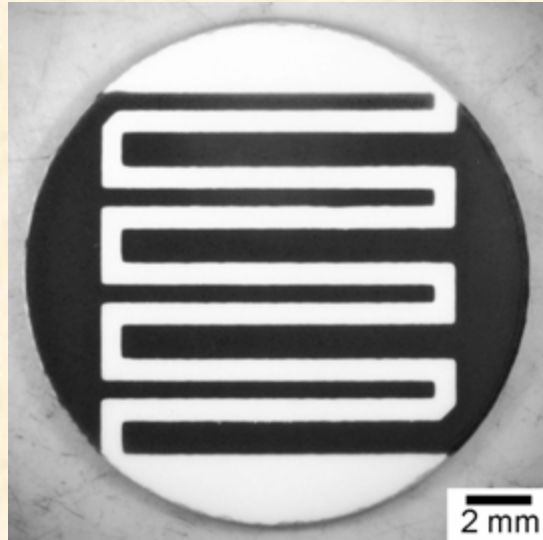
- ◆ NO<sub>x</sub> responses.
- ◆ Effect of varying [O<sub>2</sub>] & T.

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<sup>†</sup>Miura *et al.*, *Solid State Ionics*, **65**, pp. 283-90 (1999);  
Ho *et al.*, *Journ. Cer. Soc. Jpn.*, **104**, pp. 995-9 (1996);  
Grilli *et al.*, *Journ. Electrochem. Soc.*, **148**, pp. H98-102 (2001).

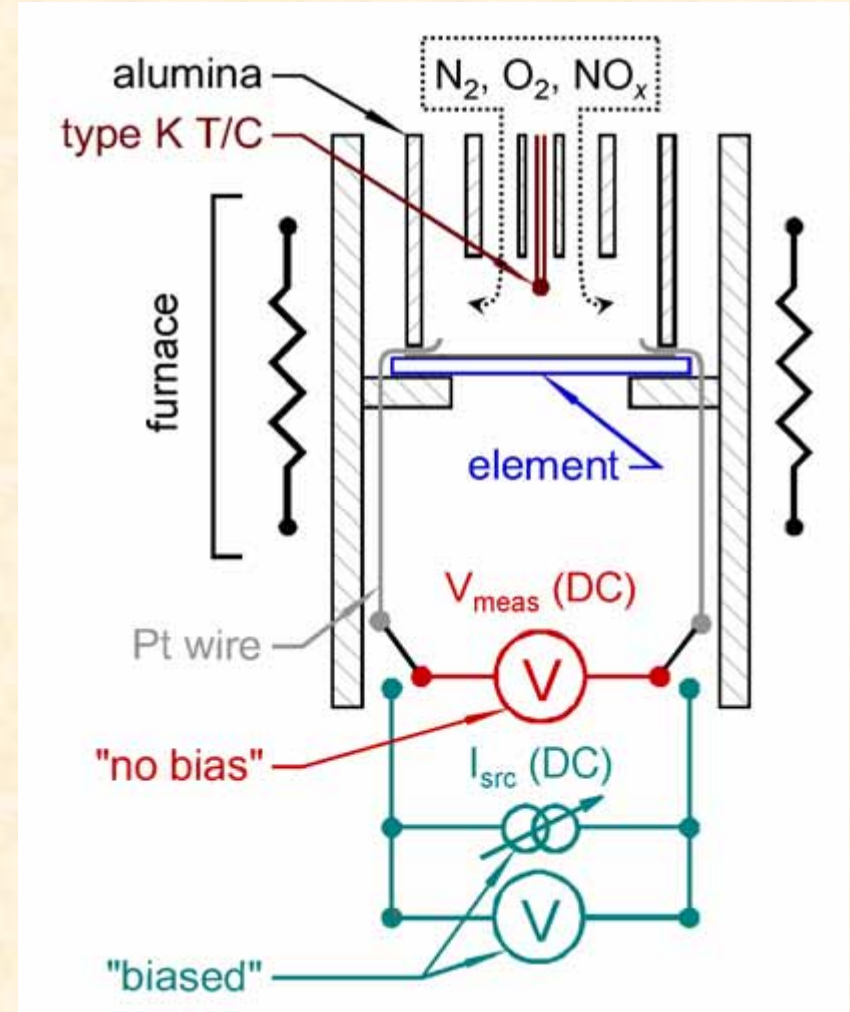
# Sensing element geometry, test fixture schematic

## Element geometry



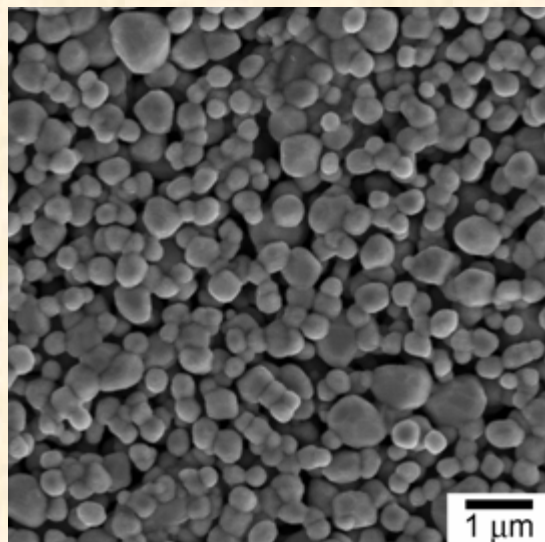
- ❖ YSZ (Tosoh TZ8YS) substrate
  - ◆ Tape cast and laminated
  - ◆  $T_{\text{sinter}} = 1400\text{ }^{\circ}\text{C}$
  - ◆  $d \sim 16\text{ mm}$ ,  $t \sim 1\text{ mm}$
- ❖ Screen-printed oxide electrodes
  - ◆ Usually  $\text{La}_{0.85}\text{Sr}_{0.15}\text{CrO}_3$  (LSC)<sup>†</sup>
  - ◆  $T_{\text{fire}} = 1200\text{ }^{\circ}\text{C}$

## Schematic of test fixture

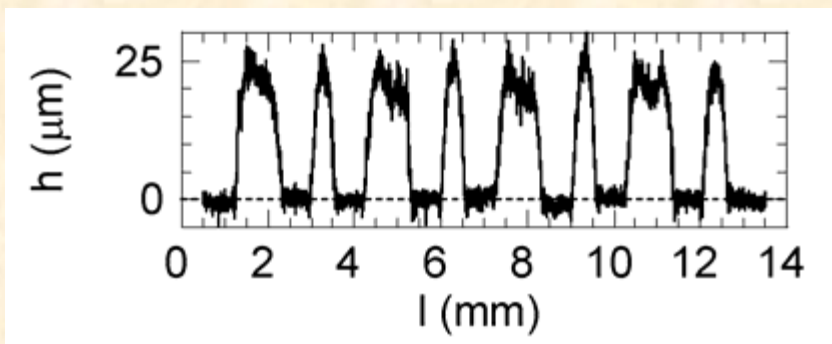


# Microstructure, thickness, phases present

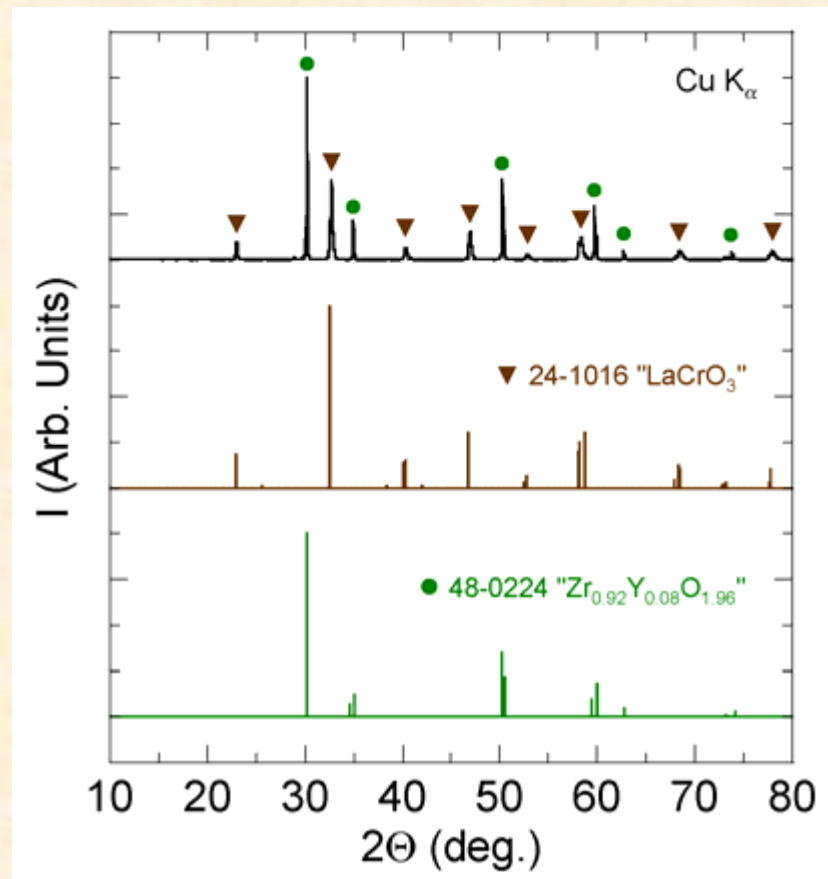
Electrode surface



Profilometer trace

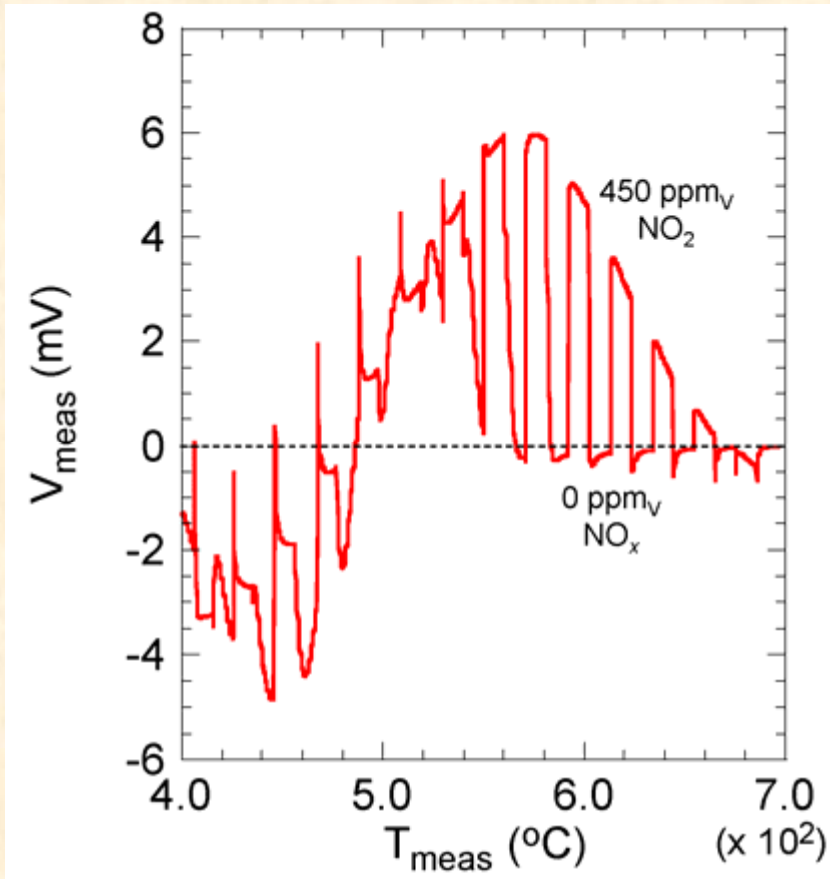


XRD of electroded surface

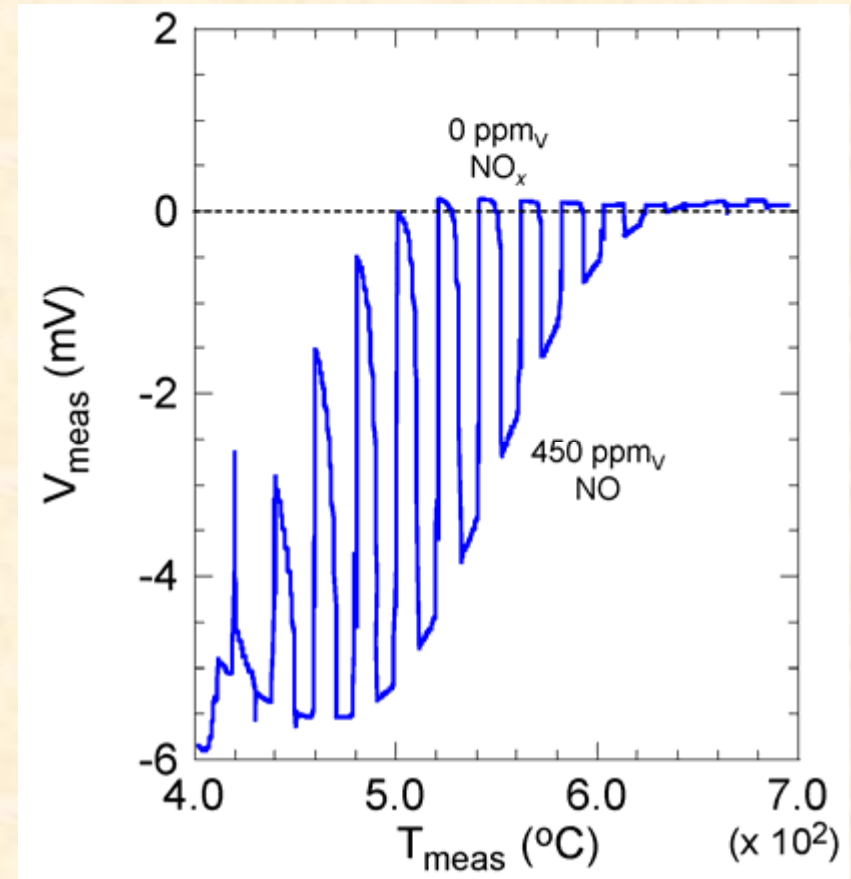


# Only small voltages in "mixed-potential" mode

input NO<sub>2</sub>



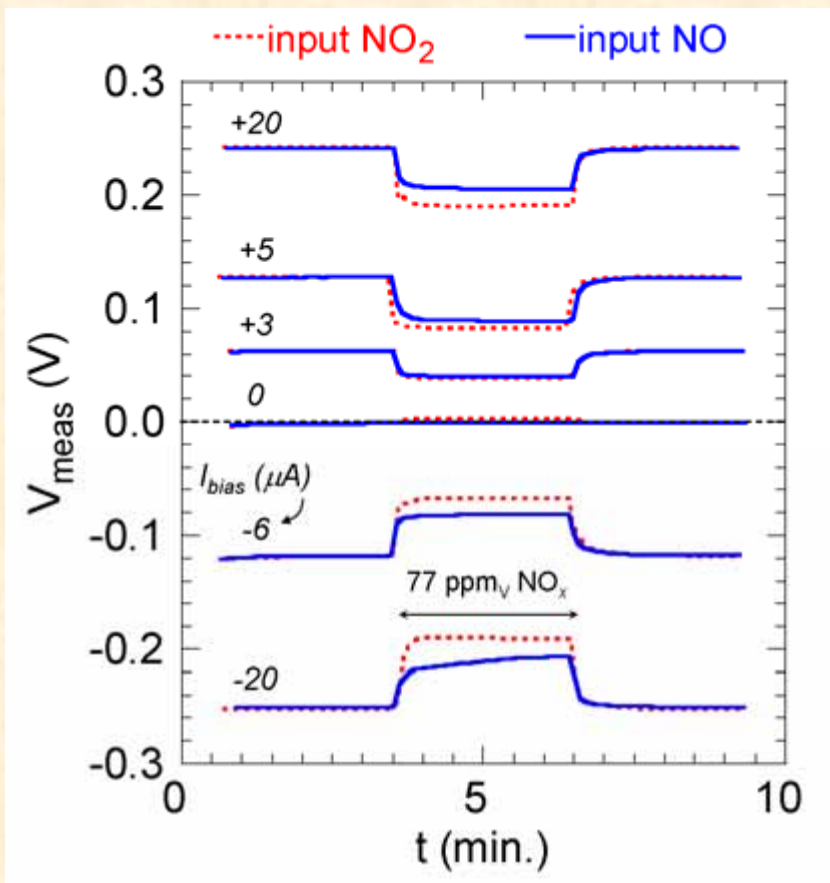
input NO



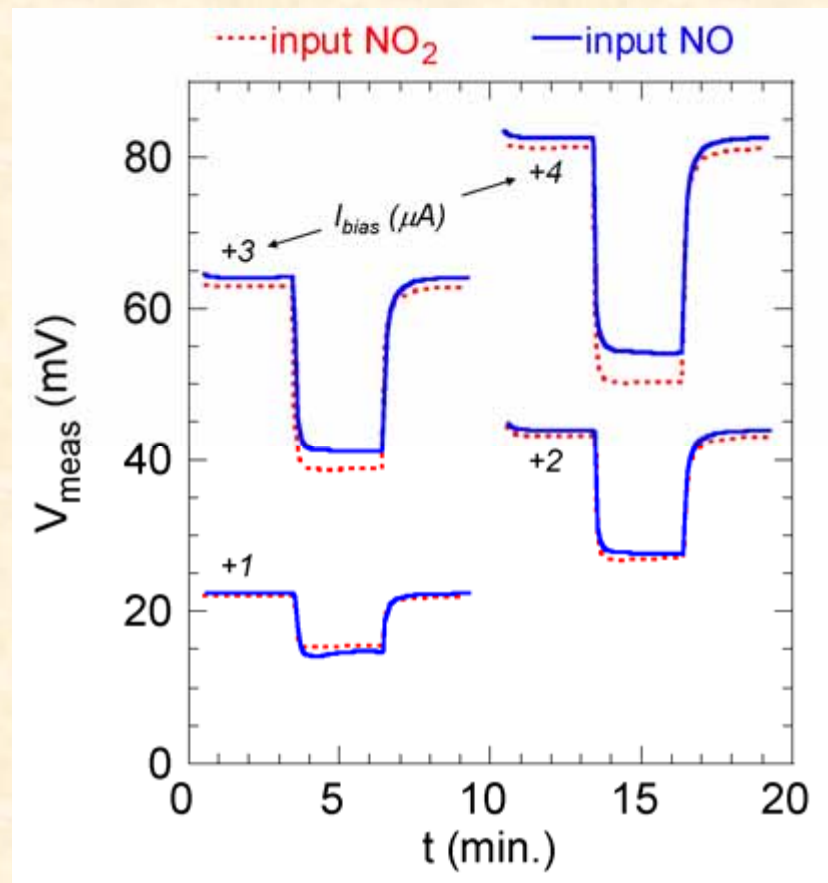
- ⇒ 5 min. NO<sub>x</sub> pulses (7 vol% O<sub>2</sub>) while ramping T at 2 deg./min.
- ⇒ Behavior with input NO and NO<sub>2</sub> distinctly different.

# Applied current influences NO<sub>x</sub> responses

$$-20 \mu\text{A} \leq I_{\text{app}} \leq +20 \mu\text{A}$$



$$+1 \mu\text{A} \leq I_{\text{app}} \leq +4 \mu\text{A}$$

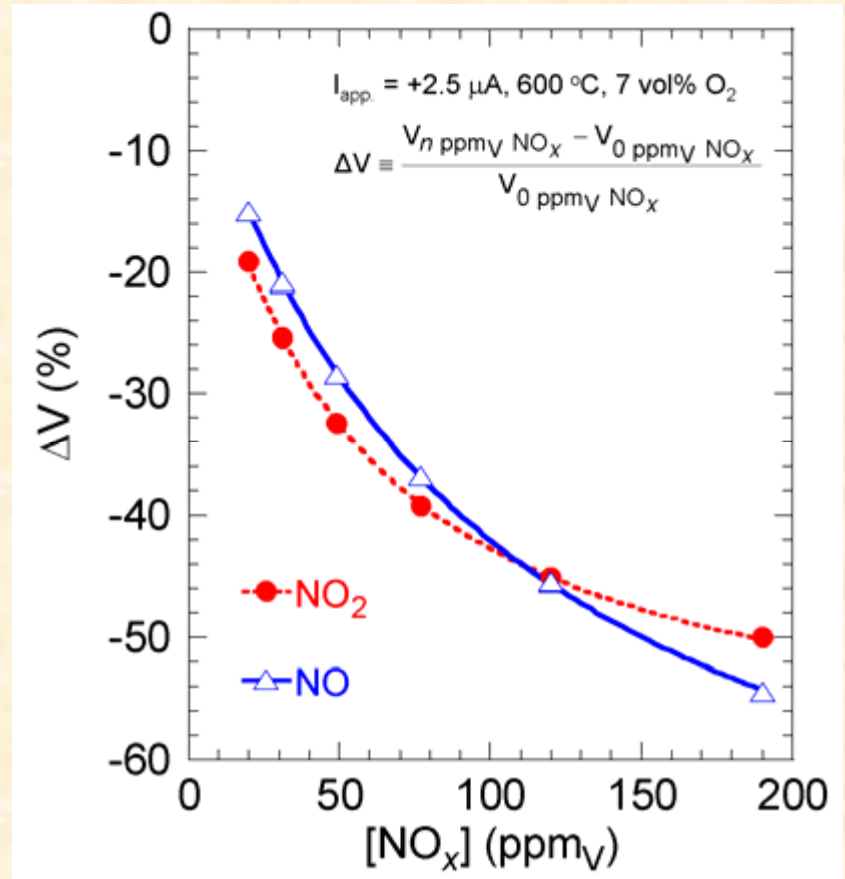
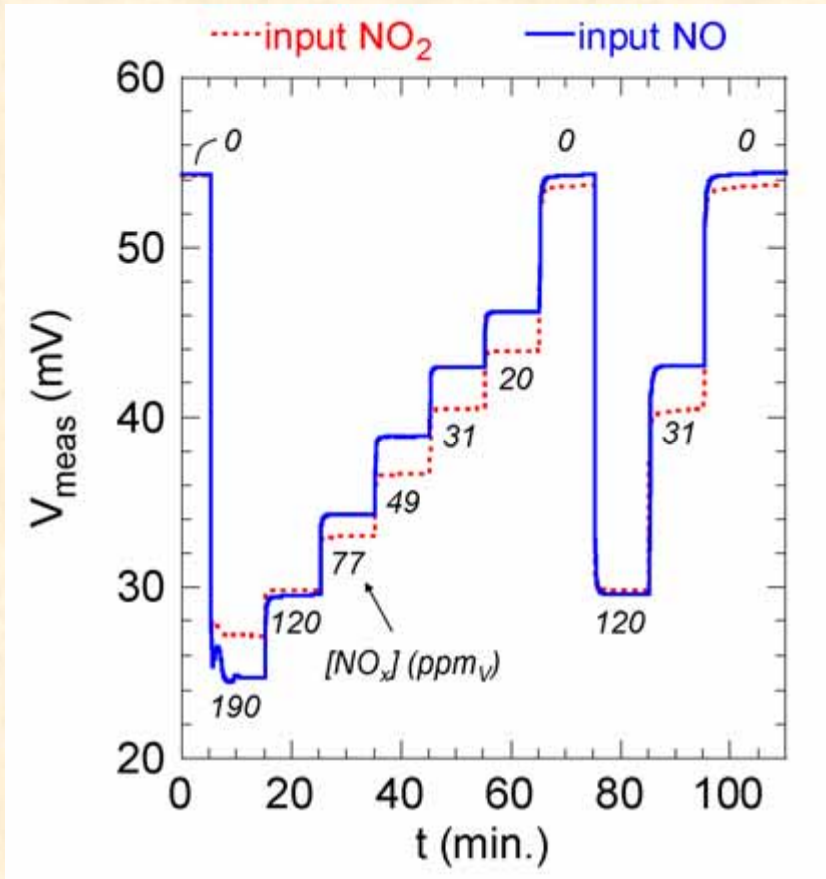


⇒  $T = 600 \text{ }^\circ\text{C}$ ,  $7 \text{ vol}\% \text{ O}_2$ .

# Symmetric response $20 \text{ ppm}_V \leq [\text{NO}_x] \leq 200 \text{ ppm}_V$

$$V_{\text{meas}} = f(t)$$

$$\Delta V = f([\text{NO}_x])$$



$\Rightarrow 0.8 \text{ ppm}_V: \Delta V_{\text{NO}} = -0.5\%, \Delta V_{\text{NO}_2} = -0.7\%$ .

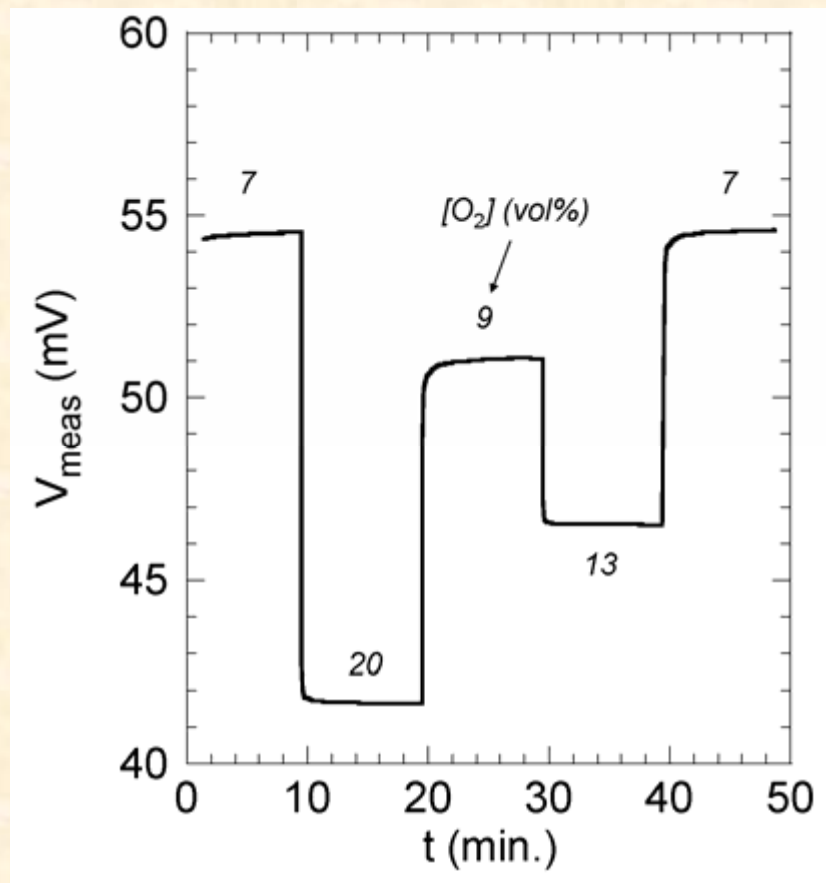
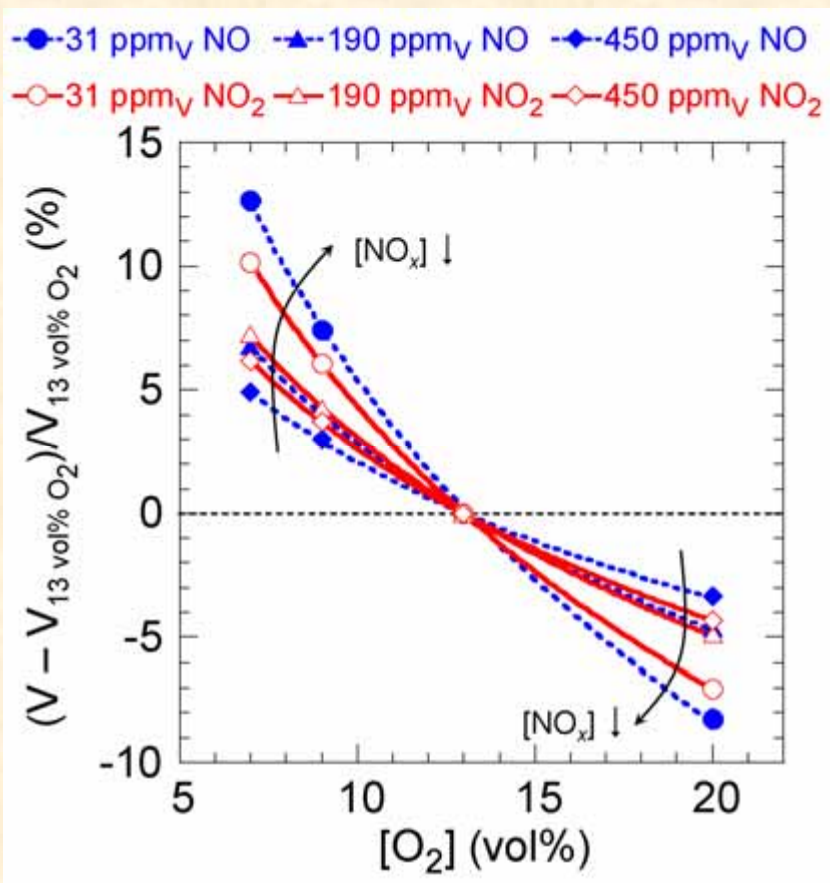
$\Rightarrow 1500 \text{ ppm}_V: \Delta V_{\text{NO}} = -80\%, \Delta V_{\text{NO}_2} = -68\%$ .



# [O<sub>2</sub>] dependence a decreasing function of [NO<sub>x</sub>]

$$V_{\text{meas}} = f([\text{O}_2])$$

$$V_{\text{meas}} = f(t) \text{ (0 ppm}_V \text{ NO}_x\text{)}$$

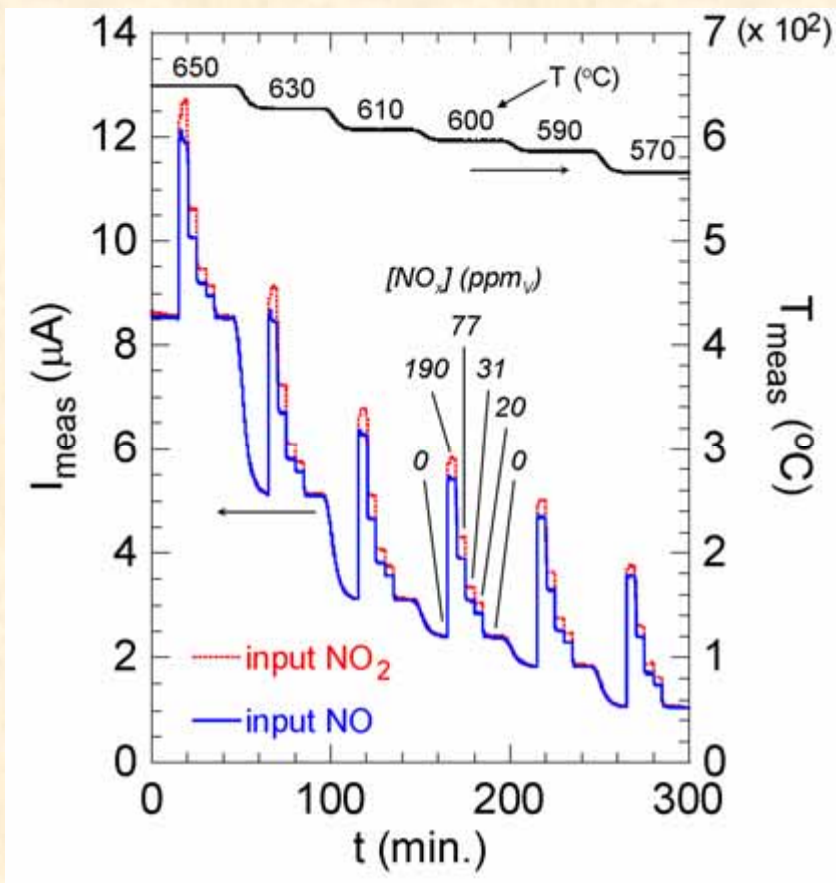


⇒  $I_{\text{app}} = +2.5 \mu\text{A}$ ,  $T = 600 \text{ }^\circ\text{C}$ .

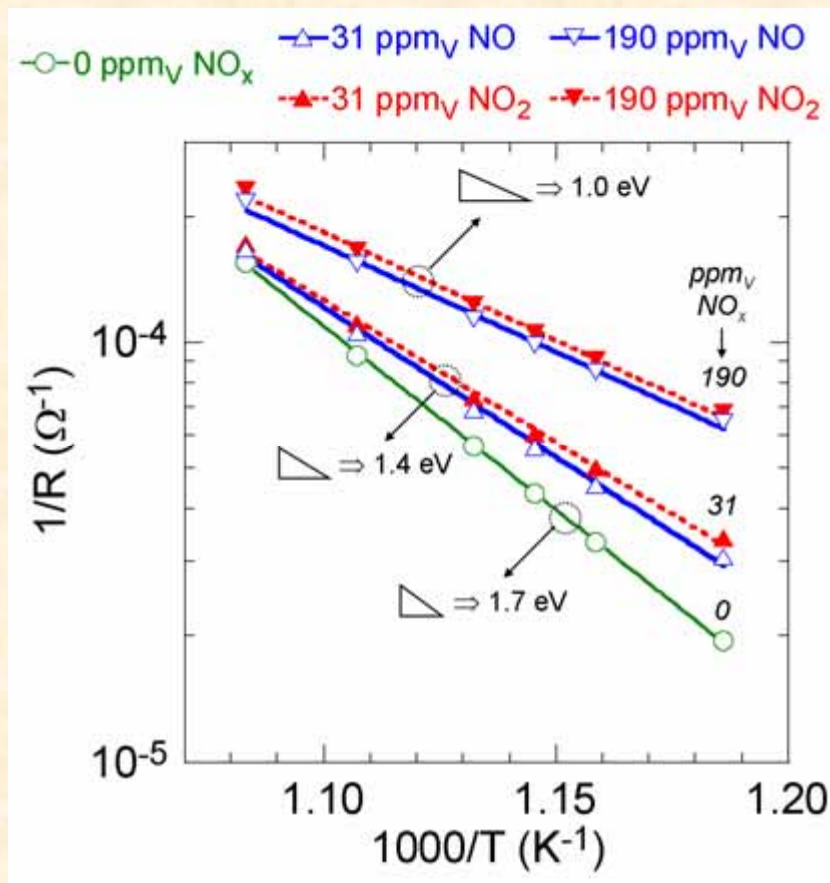
⇒ Fits to  $V_{\text{meas}} = f([\text{O}_2])$  logarithmic.

# Element resistance varies exponentially with T

$$I_{\text{meas}}, T_{\text{meas}} = f(t)$$

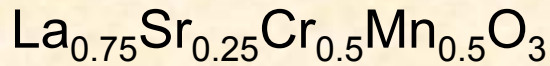


"Arrhenius" plot  $\ln(G) = f(1/T)$

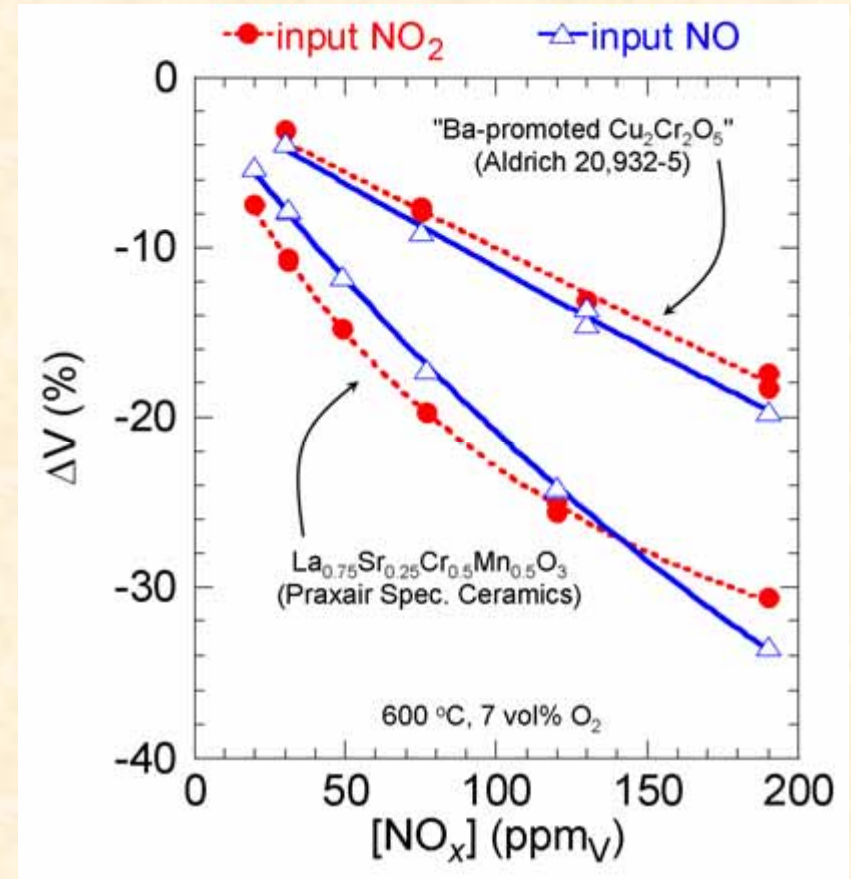
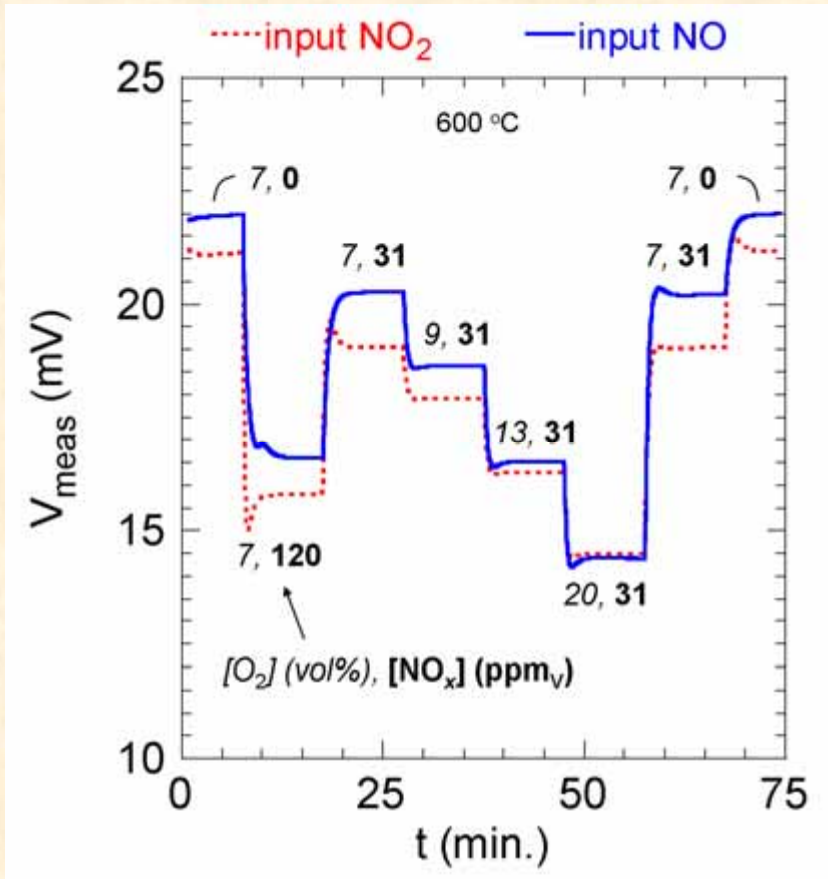


$\Rightarrow$  +55 mV applied voltage, 7 vol%  $\text{O}_2$ .

# Oxides other than LSC can yield "total NO<sub>x</sub>"



$$\Delta V = f([\text{NO}_x])$$



⇒ La<sub>0.85</sub>Ba<sub>0.15</sub>CrO<sub>3</sub> and La<sub>0.8</sub>Sr<sub>0.2</sub>FeO<sub>3</sub> could not yield "total NO<sub>x</sub>".

# Summary and conclusions

- ❖ Focus of investigation
  - ◆ Sensing elements with compositionally identical oxide electrodes.
- ❖ Observations ( $550\text{ }^{\circ}\text{C} \leq T \leq 650\text{ }^{\circ}\text{C}$ ,  $7\% \leq [\text{O}_2] \leq 20\%$ )
  - ◆ Use of applied electrical stimulus (bias) yields:
    - ❖ Equal response to NO and NO<sub>2</sub> in fairly narrow bias range.
    - ❖ Symmetry of response 20-200 ppm<sub>v</sub>.
    - ❖ [O<sub>2</sub>] sensitivity that is a decreasing function of [NO<sub>x</sub>].
    - ❖ Signal *levels* exponential function of T, *changes* due to NO<sub>x</sub> decrease with T.
    - ❖ Several different oxides have shown "total NO<sub>x</sub>" behavior.
- ❖ Conclusion and future outlook
  - ◆ Elements may be useful in "total NO<sub>x</sub>" sensor.
  - ◆ Areas for further investigation:
    - ❖ Selectivity.
    - ❖ Stability of response and electrode materials in exhaust environment.

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  - ◆ YSZ substrates & screen printing inks.
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