

LBNL's Bill Collins

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Research Highlights . . .

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Carbon Joins the magnetic club

The exclusive club of magnetic elements officially has a new member—carbon. Using a proton beam and advanced x-ray techniques, researchers at DOE's Stanford Linear Accelerator Center in collaboration with colleagues from Lawrence Berkeley National Laboratory and the University of Leipzig in Germany have finally put to rest doubts about carbon's ability to be made magnetic. Harnessing the magnetic properties of carbon could one day revolutionize a range of fields from nanotechnology to electronics. Magnetism, which forms the basis of information storage and processing in computer hard drives, could be employed in novel ways in tomorrow's electronic devices.

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Sandia sensor checks water for biological pathogens around the clock

In a multiyear research agreement with Tenix Investments, researchers at DOE's Sandia National Laboratories have developed an unattended water sensor capable of constantly monitoring water for biological pathogens including biotoxins, bacteria, viruses and protozoa. The sensor was tested at a large San Francisco Bay-area water utility for more than a year before being pronounced a success, and is now in place at a municipal water station in Arizona for additional observation and adjustments. After operating for at least three months in Arizona, Sandia and its partners hope to take the system to an **Environmental Protection Agency facility** or the US Army's Edgewood Chemical Biological Center for further testing.

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Particles hold on to mass

In the crushing atmosphere at the core of some stars, the building blocks of matter are thought to boast less mass than those trapped in the less-dense matter found here on Earth. At DOE's Jefferson Lab, scientists are attempting to catch these particles in the act of shedding mass as they experience a dense environment. Theory predicts that this loss may be seen in rho mesons, particles built of a quark and an antiquark. A recent experiment produced and measured rho mesons inside dense nuclei at high energy. Thus far, however, JLab physicists have seen no indication of particle mass loss.

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Electrolyzer Demo for Hydrogen Production

DOE's Savannah River National Laboratory has achieved a significant milestone in the development of an efficient, economical process for generating large quantities of hydrogen to fuel the nation's future. The laboratory completed a successful 100-hour long demonstration of the sulfur dioxide depolarized electrolyzer designed and fabricated by SRNL to produce hydrogen from water. This electrolyzer is a key component of the Hybrid Sulfur (HyS) thermochemical water-splitting process. The DOE Office of Nuclear Energy is funding development of the HyS Process as a means of generating hydrogen fuel using next-generation nuclear reactors. The test showed that the electrolyzer can successfully operate continuously without significant loss of performance. Work will now be directed at developing larger scale electrolyzers and testing them in a fully integrated HyS process.

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DOE Pulse highlights work being done at the Department of Energy's national laboratories. DOE's laboratories house world-class facilities where more than 30,000 scientists and engineers perform cuttingedge research spanning DOE's science, energy, national security and environmental quality missions. DOE Pulse (www.ornl. gov/news/pulse/) is distributed every two weeks. For more information, please contact Jeff Sherwood (jeff.sherwood@ hq.doe.gov, 202-586-5806).

Oak Ridge reactor resumes with a 'chill factor'

he research reactor at DOE's Oak Ridge National Laboratory is back in action and better than ever. After \$70 million in renovations and more than a year of meticulous system checks, ORNL's High Flux Isotope Reactor is running at its peak power of 85 megawatts.

Built in 1966, HFIR is internationally known as a neutron source for materials studies and isotope production. The reactor returns with a suite of new experiment instruments, beam lines to channel neutrons, a new beryllium reflector, and other upgrades.

In October, powerful refrigeration systems were added to cool the reactor's neutron beams to minus 425 degrees Fahrenheit. The intense cold slows the neutrons and lengthens their wavelength, allowing scientists to study "soft" materials such as proteins and polymers and to analyze materials with certain magnetic properties.

The restart marks HFIR's 408th cycle. Experiments scheduled for this summer include

- Experiments to create new materials with beneficial properties, based on polymer nano-composites, which are "hard" nanoparticles surrounded by a "soft" polymer matrix.
- Studies to make crystals from membrane proteins, which determine interaction and communication between living cells, to better understand the membrane proteins' structure and function.
- Examination of how high-pressure carbon dioxide is absorbed by and migrates through different types of coal to help develop new, more efficient ways to sequester CO2 to reduce greenhouse gas emissions.

Neutrons are vital to research in physics, chemistry, engineering and other materials-related fields. At room temperature, they are ideal for use in special instruments to illuminate the atomic structure and dynamics of hard, dense materials.

HFIR's cold source will complement the capabilities of ORNL's recently completed Spallation Neutron Source, the world's premier neutron science facility.

The fully instrumented HFIR will include 15 state-of-the-art neutron-scattering instruments, seven designed exclusively for cold neutron experiments; new computer control systems; and a new guide hall facility. Particularly prominent in the guide hall are the two new small-angle neutron scattering instruments, each terminating in a 70-foot long evacuated cylinder containing a large moveable neutron detector.

The reactor also produces radioisotopes used in nuclear medicine. HFIR is the only domestic source of californium-252, an isotope used in industrial analysis.

> Submitted by DOE's Oak Ridge National Laboratory

BERKELEY LAB'S BILL COLLINS: CLIMATE MODELS WITH CLOUT

"Climate models until now have either built a 'sim-Earth,' one without people, to see how natural causes affected climate in the past, or they have tried to predict the future by putting an expensive physics wrapper around socioeconomic assumptions," says leading climate modeler Bill Collins, who joined Lawrence Berkeley



Bill Collins

National Laboratory's Earth Sciences Division this April to form a new Department of Climate Science.

What Collins wants is something quite different, a model founded on basic science "that can tell you what you really want to know at the regional level and what you can do about it. That's never been done until now."

Collins set out to be an astrophysicist, with a B.A. from Princeton and a 1988 Ph.D. from the University of Chicago. By then, he says, he was already attracted to problems that evolved on a shorter time scale than that of the cosmos "and were directly relevant to society." Climate change fit the bill and led to postgraduate work at the Scripps Institution of Oceanography.

Collins joined Berkeley Lab from the NSF's National Center for Atmospheric Research in Boulder, Colo., where he led the development of the Community Climate System Model version 3 (CCSM3), a climate model sponsored by NSF and DOE that ranks among the world's most powerful. CCSM3's ability to simulate atmospheric, oceanic, and terrestrial processes underpinned the February report from Working Group I of the UN's Intergovernmental Panel on Climate Change, of which Collins was a lead author.

Collins's new kind of climate model will integrate cutting-edge science with computer expertise, drawing on the Lab's Computational Research Division and on researchers at UC Berkeley and other universities and national laboratories. His goal is to form "the first climate group I know of to work closely with large research teams on new strategies for addressing climate change."

Submitted by DOE's Lawrence Berkeley National Laboratory