

PROJECT facts

U.S. DEPARTMENT OF ENERGY
OFFICE OF Fossil ENERGY
NATIONAL ENERGY TECHNOLOGY LABORATORY

Sequestration and
Gasification Technologies

04/2008



CARBON DIOXIDE HYDRATE PROCESS FOR GAS SEPARATION FROM A SHIFTED SYNTHESIS GAS STREAM

CONTACTS

Sean Plasynski

Sequestration Technology Manager
National Energy Technology
Laboratory
626 Cochran Mill Road
P.O. Box 10940
Pittsburgh, PA 15236
412-386-4867
sean.plasynski@netl.doe.gov

Gary J. Stiegel

Gasification Technology Manager
National Energy Technology
Laboratory
626 Cochran Mill Road
P.O. Box 10940
Pittsburgh, PA 15236
412-386-4499
gary.stiegel@netl.doe.gov

José D. Figueroa

Project Manager
National Energy Technology
Laboratory
626 Cochran Mill Road
P.O. Box 10940
Pittsburgh, PA 15236
412-386-4966
jose.figueroa@netl.doe.gov

Gerald Choi

Nexant
101 Second Street 10/FI.
San Francisco, CA 94105
415-369-1075
gnchoi@nexant.com

Background

One approach to de-carbonizing coal is to gasify it to form fuel gas consisting predominately of carbon monoxide and hydrogen. This fuel gas is sent to a shift conversion reactor where carbon monoxide reacts with steam to produce carbon dioxide (CO₂) and hydrogen. After scrubbing the CO₂ from the fuel, a stream of almost pure hydrogen stream remains, which can be burned in a gas turbine or used to power a fuel cell with essentially zero emissions. However, for this approach to be practical, it will require an economical means of separating CO₂ from mixed gas streams. Since viable options for sequestration or reuse of CO₂ are projected to involve transport through pipelines and/or direct injection of high pressure CO₂ into various repositories, a process that can separate CO₂ at high pressures and minimize recompression costs will offer distinct advantages. This project addresses the issue of CO₂ separation from shifted synthesis gas at elevated pressures.

The project is concerned with development of the low temperature SIMTECHE process, which utilizes the formation of CO₂ hydrates to remove CO₂ from a gas stream. Many people are familiar with methane hydrates but are unaware that, under the proper conditions, CO₂ forms similar hydrates. In Phase 1, a conceptual process flow scheme was developed. The thermodynamic limits of such a process were confirmed by equilibrium hydrate formation experiments for shifted synthesis gas compositions, and rapid hydrate formation kinetics were demonstrated in a bench-scale flow apparatus. Performance projections were then made for a few selected process configurations, and encouraging preliminary economics were developed.

Primary Project Goal

The goal of this project is to construct and operate a laboratory-scale unit utilizing the hydrate process for CO₂ separation.

Objectives

This project will investigate an innovative, proprietary, CO₂ capture technology that can be applied to integrated gasification combined cycle power plants and other industrial gasification facilities. The SIMTECHE CO₂-Hydrate Separation Process holds promise of not only greatly reducing CO₂ emissions but also reducing the costs and the energy penalty associated with the capture process.



PARTNERS

Nexant

Los Alamos National Laboratory

SIMTEC

COST

Total Project Value

\$14,385,000

DOE/Non-DOE Share

\$5,435,000 / \$0

Los Alamos National Laboratory DOE/Non-DOE Share

\$8,950,000 / \$0

ADDRESS

National Energy Technology Laboratory

1450 Queen Avenue SW
Albany, OR 97321-2198
541-967-5892

2175 University Avenue South
Suite 201
Fairbanks, AK 99709
907-452-2559

3610 Collins Ferry Road
P.O. Box 880
Morgantown, WV 26507-0880
304-285-4764

626 Cochran's Mill Road
P.O. Box 10940
Pittsburgh, PA 15236-0940
412-386-4687

One West Third Street,
Suite 1400
Tulsa, OK 74103-3519
918-699-2000

CUSTOMER SERVICE

1-800-553-7681

WEBSITE

www.netl.doe.gov

Objectives include:

- Experimental confirmation of the feasibility of the proposed CO₂ hydrate concept.
- Extending previously developed process modeling to the latest proposed concept for the SIMTEC process.
- Determining the ultimate reduction in CO₂ concentration that can be achieved and assessing the potential negative influence of H₂S and CH₄ on the process.
- Providing detailed design and operating data in preparation for field testing of a slipstream test unit at an industrial site.
- Assessing the impact of the experimental findings on the overall process economics and identifying critical properties and critical parameters.

Accomplishments

- Demonstrated the viability of low-temperature CO₂ separation from a mixed-gas stream through the formation of CO₂ hydrates.
- Potential 68 percent CO₂ removal was demonstrated during once-through operation at 1000 psi without promoters.
- Potential 90 percent CO₂ removal was demonstrated with promoters.
- Confirmed design residence time assumptions on both a kinetic and heat transfer basis.
- Engineering analysis showed that a two-stage Simteche process with a promoter and 90 percent CO₂ removal was most economic, and compared favorably with a two-stage Selexol process.

Benefits

The hydrate process will provide a high-pressure, low-temperature system for separating CO₂ from shifted synthesis gas in an economical manner. The process can be adapted to an existing gasification power plant for CO₂ separation in the production of synthesis gas. Overall, the process will result in a residual concentrated stream of hydrogen capable of fueling zero-emission power plants of the future and a concentrated CO₂ stream available for re-use or sequestration.

