



National Institute of Standards & Technology

Certificate

Standard Reference Material 4222C Radioactivity Standard for Liquid Scintillation Counting

Radionuclide	Carbon-14- <u>n</u> -hexadecane
Source identification	4222C
Source description	5-ml of solution in NIST borosilicate-glass ampoule ⁽¹⁾ *
Solution composition	¹⁴ C-labeled <u>n</u> -hexadecane in inactive <u>n</u> -hexadecane ⁽²⁾
Radioactivity concentration	$5.402 \times 10^4 \text{ Bq g}^{-1}$
Reference time	1200 EST September 3, 1990
Overall uncertainty	0.81 percent ⁽³⁾
Photon-emitting impurities	None observed ⁽⁴⁾
Measuring instrument	Liquid-scintillation counter ⁽⁵⁾
Half life	$5760 \pm 50 \text{ years}$ ⁽⁶⁾

This standard reference material was prepared in the Center for Radiation Research, Ionizing Radiation Division, Radioactivity Group, Dale D. Hoppes, Group Leader.

Gaithersburg, MD 20899
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*Notes on back

NOTES

- (1) Approximately five milliliters of solution. Ampoule specifications:
- | | |
|----------------------|------------------------|
| body diameter | 16.5 ± 0.5 mm |
| wall thickness | 0.60 ± 0.04 mm |
| barium content | less than 2.5 percent |
| lead oxide content | less than 0.02 percent |
| other heavy elements | trace quantities |
- (2) The density of n-hexadecane is $0.7709 \pm 0.001 \text{ g/cm}^3$ at 25.0 °C.
- (3) The overall uncertainty was formed by taking three times the quadratic combination of standard deviations of the mean, or approximations thereof, for the following:
- | | |
|---|--------------|
| a) liquid-scintillation measurements for
11 vials times 5 runs | 0.03 percent |
| b) standardization utilizing efficiency
tracing technique ⁽⁵⁾ | 0.20 percent |
| c) gravimetric measurements | 0.15 percent |
| d) quenching in the liquid scintillator | 0.10 percent |
- (4) Limits of detection for photon-emitting impurities are:
- $0.02 \text{ } \gamma\text{s}^{-1}$ between 100 and 1900 keV.
- (5) The liquid-scintillation counter was standardized using a ³H radioactivity standard, by comparing the theoretical and observed spectra for ¹⁴C and ³H, as described in Coursey, B.M., et al. *Appl. Radiat. Isot.*, 37, 405 (1986).
- (6) Mann, W.B., Marlow, W.F., and Hughes, E.E., *Int. J. Appl. Radiat. Isot.* 11, 57 (1961).

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