

5. PRODUCTION, IMPORT/EXPORT, USE, AND DISPOSAL

5.1 PRODUCTION

Nickel ranks 24th in order of abundance in the earth's crust, with an average concentration of 0.0086%. Its crustal concentration varies from >0.0001 to >0.3%. Economically exploitable ore deposits typically contain 1–4% nickel. The concentration of nickel increases towards the center of the earth, and nickel is estimated to comprise 0.22% of the earth's mantle and 5.8% of its core (Duke 1980a). Therefore, the nickel concentration is estimated to be 2% by weight when averaged over all of the Earth, making it the fifth most abundant element after iron, oxygen, magnesium, and silicon. Nickel is found combined with iron in meteorites; the nickel content ranges from 5 to 50% (Duke 1980a; Mastromatteo 1986). It is also found in sea floor nodules (Mastromatteo 1986).

Nickel ores are of two general types: magmatic sulfide ores, which are mined underground, and lateritic hydrous nickel silicates or garnierites, which are surface mined (Duke 1980a; Warner 1984).

The most important nickel sulfide-arsenide deposits are in hydrothermal veins associated with mafic (i.e., rich in magnesium and iron) and ultramafic igneous rock. These ores typically contain 1–3% nickel. Pentlandite (Ni,Fe)₉S₈ is the principle ore. Pentlandite often occurs along with the iron mineral pyrrhotite and the copper mineral chalcopyrite, and part of the smelting and refining process separates the copper and iron from the nickel. The ore is concentrated by physical means (i.e., flotation and magnetic separation) after crushing. One of the largest sulfidic nickel deposits is in Sudbury, Ontario, Canada. Nickeliferous sulfide deposits are also found in Thompson, Manitoba, and Voisey's Bay, Labrador, Canada; South Africa; Russia (primarily Siberia); Finland; western Australia; and Minnesota (Ademec and Kihlgren 1967; Duke 1980a; Kuck 2002).

The lateritic hydrous nickel silicate ores are formed by the weathering of rocks rich in iron and magnesium in humid tropical areas. The repeated processes of dissolution and precipitation lead to a uniform dispersal of the nickel that is not amenable to concentration by physical means; therefore, these ores are concentrated by chemical means such as leaching. Lateritic ores are less well defined than sulfide ores. The nickel content of lateritic ores is similar to that of sulfide ore and typically ranges from 1 to 3% nickel. Important lateritic deposits of nickel are located in Cuba, New Caledonia, Indonesia, Guatemala, the Dominican Republic, the Philippines, Brazil, and Spain. Fossil nickeliferous laterite

5. PRODUCTION, IMPORT/EXPORT, USE, AND DISPOSAL

deposits are found in Oregon, Greece, and the former Soviet Union, where humid, tropical climates prevailed in the past. Lateritic deposits constitute the largest nickel reserves (Ademec and Kihlgren 1967; Antonsen and Springer 1967; Duke 1980a). Thirty-five percent of known nickel reserves are in the United States, followed by Russia at 11% (USGS 2003).

Sulfide ores are processed by a number of pyrometallurgical processes: roasting, smelting, and converting. During these processes, sulfur and iron are removed to yield a sulfur-deficient copper-nickel matte. Especially after roasting and converting, the nickel in the matte may consist primarily of nickel subsulfide. After physical separation of the copper and nickel sulfides, the nickel is refined electrochemically or by the carbonyl process. The treatment of the matte depends on the end use of the nickel. Alternatively, the sulfide can be roasted to form a nickel oxide sinter that is used directly in steel production.

Lateritic ore is processed by pyrometallurgical or hydrometallurgical processes. In the pyrometallurgical process, sulfur is generally added to the oxide ore during smelting, usually as gypsum or elemental sulfur, and an iron-nickel matte is produced. The smelting process that does not include adding sulfur produces a ferronickel alloy, containing $\leq 50\%$ nickel, which can be used directly in steel production. Hydrometallurgical techniques involve leaching with ammonia or sulfuric acid, after which the nickel is selectively precipitated (Duke 1980b; IARC 1990; Tien and Howson 1981; Warner 1984). Alloys, such as stainless steels, are produced by melting primary metals and scrap in large arc furnaces and adjusting the carbon content and concentration of alloying metals to the desired levels. More information on the mining, smelting, and refining of nickel can be found in Duke (1980b), Tien and Howson (1981), and Warner (1984).

Domestic primary nickel production in the United States ceased in 1986 (Chamberlain 1985; Kirk 1988a) with the closing of the Hanna mine and smelter in Riddle, Oregon, and the AMAX refinery in Braithwaite, Louisiana. However, Glenbrook Nickel Company purchased the Riddle, Oregon, facility in 1989 and had reactivated the mining and smelting operation, but then decommissioned both the mining and smelting operations in 2000. World mine production of nickel in 2002 was estimated at 1,340,000 metric tons, which matched the production level reported for 2001 (Kuck 2002). Secondary nickel production from scrap is a major source of nickel for industrial applications. In 1988, an estimated 59,609 and 3,700 short tons (51,355 and 3,357 metric tons, respectively) of nickel were produced from ferrous and nonferrous scrap, respectively. Nickel recovery from scrap is estimated by using the gross weight of the scrap and a weighted average nickel content (e.g., 7.5% for stainless steel). The secondary

5. PRODUCTION, IMPORT/EXPORT, USE, AND DISPOSAL

recovery from ferrous scrap was considerably higher and the recovery from nonferrous scrap was considerably lower than for the previous 7 years in which the annual recovery of nickel from ferrous and nonferrous scrap ranged from 30,034 to 389,265 short tons (27,246–353,139 metric tons) and from 8,392 to 19,776 short tons (7,613–17,940 metric tons), respectively. The production of refined nickel in 1993 has been estimated as 220,700, 346,800, 176,200, 52,100, and 96,300 short tons (200,200, 314,600, 159,800, 47,300, and 87,400 metric tons, respectively) for North America, Europe, Asia, Africa, and Australia, respectively (ABMS 1994). In 2002, 1,210,000 metric tons of refined nickel were produced. Of this total production, there were 259,000 metric tons in the form of ferronickel, 678,000 metric tons as the metal, 97,000 metric tons as the oxide sinter, 17,800 metric tons as chemicals, and 154,000 metric tons as unspecified production (Kuck 2002). The distribution of world plant production of refined nickel was 20.3%, Russia (Commonwealth of Independent States); 16.4%, Europe; 12.9%, Japan; 11.9%, Canada; 15.8%, Australia/New Caledonia; 4.5%, China; 4.0%, Africa; 3.7%, Colombia; 3.3%, Cuba, and 7.2%, Brazil, Dominican Republic, Indonesia, and Venezuela (Kuck 2002). The reported world consumption of refined nickel was 1,150,800 metric tons in 2001, up from 997,800 metric tons in 1997 (ABMS 2002). In 2002, demand for primary nickel in the Western World was 1,032,000 metric tons, up from 968,700 metric tons in 2001 (Kuck 2002).

Tables 5-1 and 5-2 list the facilities that produced, imported, processed, or used nickel and its compounds, respectively, in 2001 according to reports made to the EPA under the requirements of Section 313 of the Emergency Planning and Community Right-to-Know Act of 1986, which were subsequently published in the Toxic Chemical Release Inventory (TRI) (TRI02 2004). Companies were required to report if they produced, imported, or processed $\geq 25,000$ pounds of nickel and its compounds or used $> 10,000$ pounds. Also included in Tables 5-1 and 5-2 are the maximum amount of nickel and its compounds, respectively, that these facilities had on site and whether nickel was produced, processed, or used by the facility.

5.2 IMPORT/EXPORT

In 2002, the United States imported 130,000 metric tons of nickel, including 121,000 metric tons of unwrought metal (97,200 metric tons of cathodes, pellets, briquets and shot; 12,300 metric tons of ferronickel; 6,970 metric tons of powder and flakes, 1,230 metric tons of metallurgical-grade oxide; 1,280 metric tons of catalysts; and 1,590 metric tons of salts), 6,080 metric tons of stainless steel scrap, and 3,030 metric tons of nickel waste and scrap (Kuck 2002). In 2002, Canada supplied the largest share of primary nickel, 59,400 metric tons (46%). Russia was the second largest exporter of primary nickel to the United States with 24,200 metric tons (19%) followed by Australia and Norway with 10,400 and

5. PRODUCTION, IMPORT/EXPORT, USE, AND DISPOSAL

Table 5-1. Facilities that Produce, Process, or Use Nickel Metal

State ^a	Number of facilities	Minimum amount on site in pounds ^b	Maximum amount on site in pounds ^b	Activities and uses ^c
AL	73	0	499,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
AR	47	0	99,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13
AZ	41	0	9,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13
CA	146	0	99,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
CO	39	100	9,999,999	1, 2, 3, 4, 5, 6, 7, 8, 10, 11, 12
CT	78	0	99,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
DC	2	10,000	999,999	8
DE	15	100	999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12
FL	30	0	499,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12
GA	57	0	999,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
HI	1	10,000	99,999	10
IA	71	0	9,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
ID	12	0	999,999	1, 3, 5, 7, 8, 10, 12
IL	145	0	499,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
IN	158	0	499,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
KS	41	0	49,999,999	1, 2, 3, 5, 7, 8, 9, 10, 11, 12, 13, 14
KY	85	0	999,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
LA	42	0	49,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
MA	71	0	499,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12
MD	44	0	9,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13
ME	13	1,000	99,999,999	1, 3, 7, 8, 12
MI	154	0	49,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
MN	69	0	9,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
MO	59	0	99,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
MS	17	0	9,999,999	1, 2, 3, 6, 7, 8, 10, 12
MT	8	100	99,999	1, 2, 3, 5, 6, 8, 10, 11
NC	66	0	99,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
ND	9	1,000	999,999	2, 3, 5, 7, 8, 9, 12
NE	39	0	9,999,999	1, 2, 3, 5, 6, 7, 8, 9, 10, 11, 12, 13
NH	19	0	49,999,999	3, 7, 8, 9, 11, 12
NJ	86	0	49,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13
NM	13	100	999,999	2, 3, 6, 7, 8, 10, 11
NV	27	100	9,999,999	1, 2, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13
NY	93	0	49,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13
OH	204	0	99,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
OK	74	0	99,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
OR	45	0	499,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13
PA	237	0	499,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
PR	6	0	99,999	2, 3, 7, 8, 10, 11, 12
RI	36	0	999,999	1, 2, 3, 4, 6, 7, 8, 9, 10, 11, 12

5. PRODUCTION, IMPORT/EXPORT, USE, AND DISPOSAL

Table 5-1. Facilities that Produce, Process, or Use Nickel Metal

State ^a	Number of facilities	Minimum amount on site in pounds ^b	Maximum amount on site in pounds ^b	Activities and uses ^c
SC	77	0	9,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
SD	9	1,000	999,999	1, 5, 7, 8, 9, 11
TN	113	0	499,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
TX	137	0	9,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
UT	40	100	9,999,999	1, 2, 3, 4, 5, 6, 7, 8, 10, 11, 12, 13
VA	49	0	49,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
VT	15	1,000	999,999	2, 4, 8, 11, 12
WA	31	0	999,999	1, 2, 3, 4, 5, 7, 8, 9, 10, 11, 12, 13
WI	114	0	99,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 14
WV	35	0	499,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13
WY	9	100	99,999	1, 4, 8, 9, 10, 12, 13

Source: TRI02 2004 (Data are from 2002)

^aPost office state abbreviations used

^bAmounts on site reported by facilities in each state

^cActivities/Uses:

- | | | |
|--------------------------|--------------------------|-----------------------------|
| 1. Produce | 6. Impurity | 11. Chemical Processing Aid |
| 2. Import | 7. Reactant | 12. Manufacturing Aid |
| 3. Onsite use/processing | 8. Formulation Component | 13. Ancillary/Other Uses |
| 4. Sale/Distribution | 9. Article Component | 14. Process Impurity |
| 5. Byproduct | 10. Repackaging | |

5. PRODUCTION, IMPORT/EXPORT, USE, AND DISPOSAL

Table 5-2. Facilities that Produce, Process, or Use Nickel Compounds

State ^a	Number of facilities	Minimum amount on site in pounds ^b	Maximum amount on site in pounds ^b	Activities and uses ^c
AK	7	10,000	9,999,999	1, 2, 3, 5, 7, 10, 11, 12, 13
AL	83	0	9,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
AR	46	0	99,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
AZ	54	100	99,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
CA	163	0	499,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
CO	16	100	999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 11, 12
CT	81	0	499,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13
DC	4	10,000	99,999	1, 3, 5, 6, 8, 10, 11
DE	19	0	9,999,999	1, 2, 3, 5, 6, 7, 8, 9, 10, 12, 13, 14
FL	52	0	999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
GA	79	0	9,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
HI	5	0	999,999	1, 5, 10, 12
IA	42	0	49,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
ID	20	1,000	49,999,999	1, 2, 3, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
IL	184	0	499,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
IN	143	0	9,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
KS	35	0	9,999,999	1, 2, 3, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
KY	82	100	499,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
LA	73	0	9,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
MA	50	100	9,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13
MD	37	0	49,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 12, 13
ME	12	0	999,999	1, 5, 8, 9, 11, 12, 13
MI	149	0	499,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
MN	66	100	999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
MO	63	0	499,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
MS	37	0	9,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
MT	16	0	9,999,999	1, 2, 3, 4, 5, 6, 8, 9, 10, 11, 12, 13, 14
NC	73	0	9,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
ND	4	1,000	9,999	1, 5, 12, 13, 14
NE	19	100	999,999	1, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
NH	12	0	99,999	1, 5, 7, 8, 9
NJ	91	0	499,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
NM	19	100	499,999,999	1, 3, 4, 5, 8, 9, 10, 11, 12, 13, 14
NV	33	100	10,000,000,000	1, 2, 3, 5, 6, 7, 8, 9, 10, 12, 13, 14
NY	110	0	499,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
OH	218	0	99,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
OK	42	100	999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
OR	38	100	49,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 11, 12, 13
PA	229	0	499,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14

5. PRODUCTION, IMPORT/EXPORT, USE, AND DISPOSAL

Table 5-2. Facilities that Produce, Process, or Use Nickel Compounds

State ^a	Number of facilities	Minimum amount on site in pounds ^b	Maximum amount on site in pounds ^b	Activities and uses ^c
PR	19	100	999,999	1, 2, 5, 6, 7, 8, 10, 11, 12, 13
RI	39	0	49,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12
SC	61	0	9,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
SD	1	10,000	99,999	1, 5, 9, 13
TN	100	0	499,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
TX	194	0	49,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
UT	26	100	49,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
VA	44	0	999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
VI	4	10,000	999,999	2, 10, 11
WA	28	0	999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13,
WI	75	0	9,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
WV	45	0	99,999,999	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14
WY	9	100	99,999	1, 2, 3, 4, 5, 9, 10, 12, 13

Source: TRI02 2004 (Data are from 2002)

^aPost office state abbreviations used

^bAmounts on site reported by facilities in each state

^cActivities/Uses:

- | | | |
|--------------------------|--------------------------|-----------------------------|
| 1. Produce | 6. Impurity | 11. Chemical Processing Aid |
| 2. Import | 7. Reactant | 12. Manufacturing Aid |
| 3. Onsite use/processing | 8. Formulation Component | 13. Ancillary/Other Uses |
| 4. Sale/Distribution | 9. Article Component | 14. Process Impurity |
| 5. Byproduct | 10. Repackaging | |

5. PRODUCTION, IMPORT/EXPORT, USE, AND DISPOSAL

8,550 metric tons, respectively. The 130,000 metric tons of nickel imported in 2002 was down from the 167,000 and 144,000 metric tons imported in 2000 and 2001, respectively (Kuck 2001, 2002). From 1999 to 2001, nickel imports as a percentage of consumption decreased from 63 to 46%, with a slight increase to 48% in 2002 (USGS 2003).

The amount of exported nickel dropped sharply in 1986 to 15,217 short tons (13,805 metric tons) from 35,245 short tons (31,974 metric tons) the previous year (Kirk 1988a), which coincided with the cessation of primary nickel production in the United States. The nickel content of exported primary and secondary nickel in 2002 was 45,900 metric tons, most of which (39,400 metric tons) was in the form of stainless steel scrap and waste scrap (Kuck 2002).

5.3 USE

Nickel is primarily used in alloys because it imparts to a product such desirable properties as corrosion resistance, heat resistance, hardness, and strength. Nickel alloys are often divided into categories depending on the primary metal with which they are alloyed and their nickel content. Copper-nickel alloys (e.g., Monel alloys) are used for industrial plumbing, marine equipment, petrochemical equipment, heat exchangers, pumps, and electrodes for welding. Coinage metal contains 75% copper and 25% nickel. Nickel-chromium alloys (e.g., Nichrome) are used for heating elements. Nickel-iron-chromium alloys (e.g., Inconel) provide strength and corrosion resistance over a wide temperature range. Hastelloy alloys, which contain nickel, chromium, iron, and molybdenum, provide oxidation and corrosion resistance for use with acids and salts. Nickel-based superalloys have the required high-temperature strength and creep and stress resistance for use in gas-turbine engines. Nickel silvers, and nickel alloys with zinc and copper, have an attractive white color and are used for coatings on tableware and as electrical contacts. Raney nickel, 50% aluminum and 50% nickel, is used as a catalyst in hydrogenation reactions. Large amounts of nickel are alloyed with iron to produce alloy steels, stainless steels, and cast irons. Stainless steel may contain as much as 25–30% nickel, although 8–10% nickel is more typical. Alloy steels generally contain 0.3–5% nickel. In addition to imparting characteristics such as strength, toughness, corrosion resistance, and machinability, some applications make use of nickel's magnetic characteristics. Most permanent magnets are made of alloys of iron and nickel (Tien and Howson 1981).

Nickel salts are used in electroplating, ceramics, pigments, and as catalysts. Sinter nickel oxide is used as charge material in the manufacture of alloy steel and stainless steel. Nickel is also used in nickel-cadmium (NiCd) and nickel-metal hydride (NiMH) batteries.

5. PRODUCTION, IMPORT/EXPORT, USE, AND DISPOSAL

The distribution of nickel consumption by use in 2002 was as follows: stainless and heat-resistant steel, 61%; nickel-copper and copper-nickel alloys, 4%; other nickel alloys, 13%; electroplating, 6%; superalloys, 9%; and other, 7%. Other uses include cast iron; chemicals and chemical use; electric, magnet, expansion alloys; steel alloys, other than stainless steel; batteries; and ceramics. Forty-six percent of primary nickel consumption in 2002 was for the production of stainless steel and low-nickel steels, and 33% was used for the production of superalloys and related nickel-based alloys (Kuck 2002).

5.4 DISPOSAL

Little information concerning the disposal of nickel and its compounds is found in the literature. Much of the nickel used in metal products (e.g., stainless steel, nickel plate, various alloys) is recycled, which is evident from the fact that 53% of nickel consumption in 2002 was derived from secondary scrap (Kuck 2002). According to the 2002 TRI, 86% of the 29,698,967 pounds (13,483,331 kg) of nickel and nickel compounds released on-site is released to land (see Section 6.1) (TRI02 2004). In addition, >14 million pounds of nickel were transferred to off-site locations that year with about 90% being recycled. Steel and other nickel-containing items discarded by households and commercial establishments are generally recycled, landfilled, or incinerated along with normal commercial and municipal trash.

Nickel is removed from electroplating wastes by treatment with hydroxide, lime, and/or sulfide to precipitate the metal (HSDB 2004). Adsorption with activated carbon, activated alumina, and iron filings is also used for treating nickel-containing waste water. Ion exchange is also used for nickel removal and recovery.

Nickel and its compounds have been designated as toxic pollutants by EPA pursuant to Section 307(a)(1) of the Federal Water Pollution Control Act (40 CFR 401.15). As such, permits are issued by the states under the National Pollutant Discharge Elimination System (NPDES) for discharges of nickel that meet the applicable requirements (40 CFR 401.12).