

# Standard Chemical Thermodynamic Properties of Polycyclic Aromatic Hydrocarbons and Their Isomer Groups. II. Pyrene Series, Naphthopyrene Series, and Coronene Series

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Received November 2, 1987; revised manuscript received June 28, 1988

The tables in our first paper on polycyclic aromatic hydrocarbons [J. Phys. Chem. Ref. Data 17, 241 (1988)] have been extended by calculating thermodynamic properties for the first four isomer groups in the pyrene series, the first three isomer groups in the naphthopyrene series, and the first three isomer groups in the coronene series. Successive isomer groups in each series differ by  $C_4H_2$ . Since chemical thermodynamic properties are known for only a limited number of polycyclic aromatic hydrocarbons, the properties of individual species have been estimated using Benson group values of Stein and Fahr for temperatures from 298.15 to 3000 K. Values of  $C_p^\circ$ ,  $S^\circ$ ,  $\Delta_f H^\circ$ , and  $\Delta_f G^\circ$  have been calculated in joules for a standard state pressure of 1 bar. The chemical thermodynamic properties of the individual species have also been calculated. The isomer group values provide a basis for extrapolating to higher carbon numbers where it is not feasible to consider individual molecular species.

Key words: polycyclic aromatic hydrocarbons; pyrene; naphthopyrene; coronene; Benson method; enthalpy of formation; heat capacity; isomer group thermodynamic properties; isomer mole fractions; thermodynamic properties.

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## I. Introduction

The preceding article in this series<sup>1</sup> presented thermodynamic tables for the benzene series isomer groups from  $C_6H_6$  to  $C_{26}H_{16}$  and for the individual species. This paper extends these tables to the next three isomer groups. Some of the needs for thermodynamic properties of polycyclic aromatic hydrocarbons have been indicated in the preceding article.

Dias<sup>2-6</sup> has developed a formula periodic table for benzenoid polycyclic aromatic hydrocarbons using graph theoretical principles. Molecular formulas in this periodic table represent isomer groups. In thinking about the chemical thermodynamic properties of polycyclic aromatic hydrocarbons it is convenient to think of them in terms of series of isomer groups in which successive isomer groups differ by  $C_4H_2$ . Table 1 gives the molecular formulas, numbers of rings, and numbers of isomers for the first four series. In the paper on the benzene series, we found that the thermodynamic properties of the higher isomer groups can be estimated by linear extrapolation from the properties of lower isomer groups. This paper is concerned with the pyrene, naphthopyrene, and coronene series. There is an infinite number of these series, and each has an infinite number of isomer groups.

The numbers of isomers in Table 1 come from Trinajstić and co-workers.<sup>7</sup> We are indebted to them for the computer generation of all the possible structures that correspond to these molecular formulas. The structural formulas of the isomers considered in this paper are given in Fig. 1. To simplify this figure only sigma bonds are shown. However, the thermodynamic properties of all these isomers are not estimated in this paper for the following reasons: (i) Some of the structures are diradicals, and we do not currently know how to estimate their properties—although they are expected to be unstable with respect to the nonradical isomers. (ii)

Some of the structures are helicenes in the sense that the hydrogen atoms overlap in the bay region, and this prevents them from being planar molecules; again we do not currently know how to estimate their properties—although they are expected to be unstable with respect to the nonradical isomers.

Table 2 gives the numbers of diradicals for the various isomer groups considered in this paper; they are identified in Trinajstić's tables by  $K = 0$ , since they do not have any Kekule structures. The identification numbers of the diradical isomers in Fig. 1 are as follows:

$C_{24}H_{14}$	13
$C_{28}H_{16}$	22,51,57,58,66,67
$C_{22}H_{12}$	2
$C_{26}H_{14}$	9
$C_{30}H_{16}$	27,47,53,54,55,56,64,65,66
$C_{28}H_{14}$	8
$C_{32}H_{16}$	15,25,41,44,45,46,48,53,54.

Table 2 also gives the numbers of helicenes that are omitted in the thermodynamic tables. Their identification numbers are as follows:

$C_{28}H_{16}$	3,24,35,36,41,42,48,57
$C_{30}H_{16}$	41,42.

The omission of diradicals and helicenes is not expected to affect the isomer group thermodynamic properties because these species have significantly higher standard Gibbs energies of formation, and therefore do not contribute significantly to the equilibrium population. The tables presented here are of the same type as those published earlier for alkanes,<sup>8</sup> alkylbenzenes,<sup>9</sup> alkenes,<sup>10</sup> alkylnaphthalenes,<sup>11</sup> alkylcyclopentanes and cyclohexanes,<sup>12</sup> alkynes,<sup>13</sup> thiols,<sup>14</sup> alkanols,<sup>15</sup> and benzene-series polycyclic aromatic hydrocarbons.<sup>1</sup>

TABLE 1. First four series of polycyclic aromatic hydrocarbons<sup>a</sup>

$N_C$	Benzene series	Pyrene series	Naphthopyrene series	Coronene series
6	$C_6H_6(1,1)$			
8				
10	$C_{10}H_8(2,1)$			
12				
14	$C_{14}H_{10}(3,2)$			
16		$C_{16}H_{10}(4,1)$		
18	$C_{18}H_{12}(4,5)$			
20		$C_{20}H_{12}(5,3)$		
22	$C_{22}H_{14}(5,12)$		$C_{22}H_{12}(6,3)$	
24		$C_{24}H_{14}(6,14)$		$C_{24}H_{12}(7,1)$
26	$C_{26}H_{16}(6,36)$		$C_{26}H_{14}(7,10)$	
28		$C_{28}H_{16}(7,68)$		$C_{28}H_{14}(8,9)$
30	$C_{30}H_{18}(7,118)$		$C_{30}H_{16}(8,67)$	
32		$C_{32}H_{18}(8,329)$		$C_{32}H_{16}(9,55)$
34	$C_{34}H_{20}(8,411)$		$C_{34}H_{18}(9,396)$	
36		$C_{36}H_{20}(9,1601)$		
38	$C_{38}H_{22}(8,1489)$			
40				

<sup>a</sup> The first number in parentheses is the number of rings and the second is the number of isomers, including the diradical benzenoid isomers.

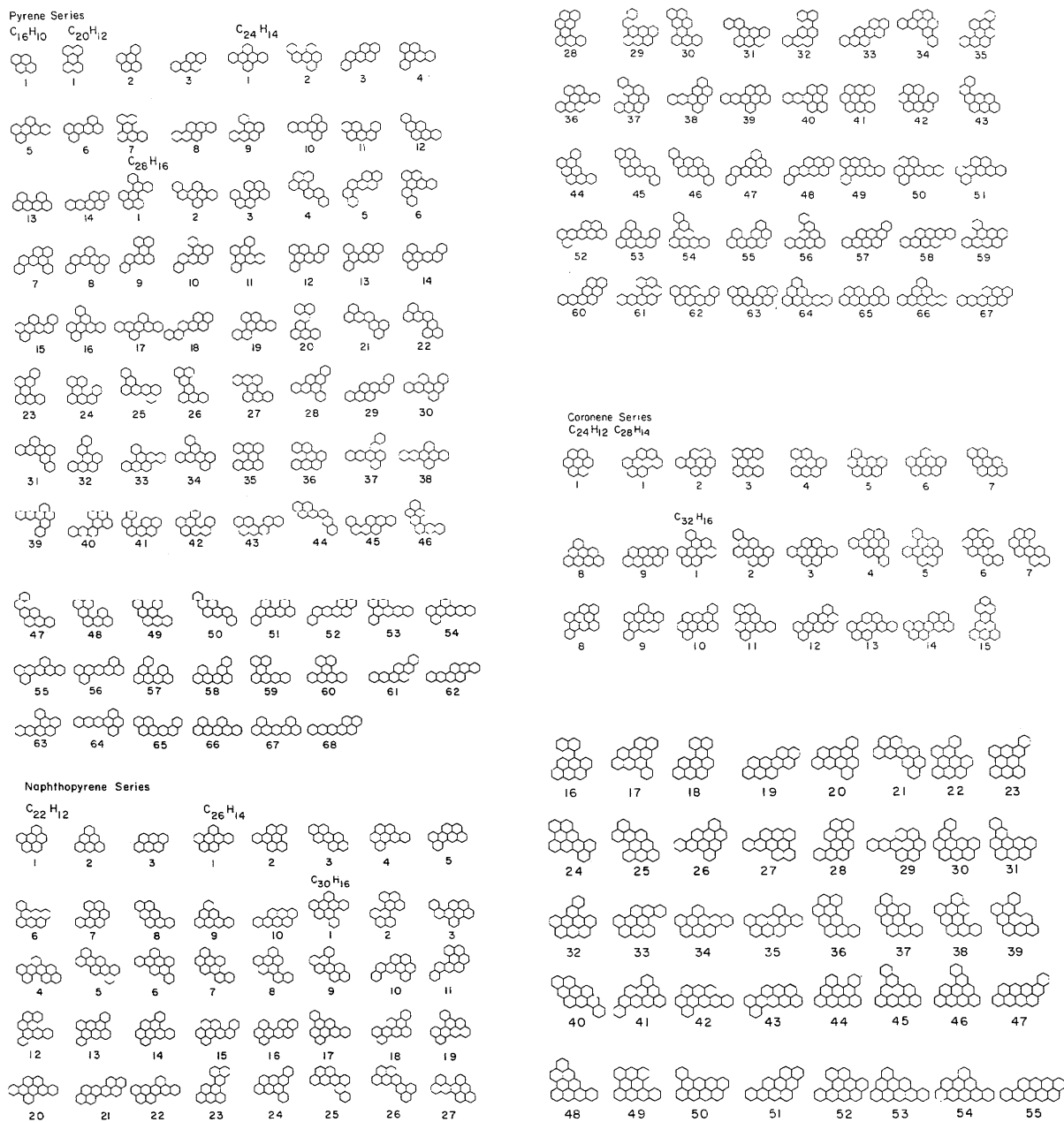


FIG 1. Structures of isomers in the pyrene, naphthopyrene, and coronene series.

## 2. Standard Thermodynamic Properties of Polycyclic Aromatic Isomer Groups

When isomers are in equilibrium, the standard Gibbs energy of formation  $\Delta_f G^\circ(I)$  of the isomer groups is defined by<sup>16,17</sup>

$$\Delta_f G^\circ(I) = -RT \ln \sum_{i=1}^{N_1} \exp(-\Delta_f G_i^\circ/RT), \quad (1)$$

where  $\Delta_f G_i^\circ$  is the standard Gibbs energy of formation of an individual isomer and  $N_1$  is the number of isomers in an isomer group. The equilibrium mole fractions  $r_i$  of various isomers in a group can be calculated using

$$r_i = \frac{y_i}{y_1} = \exp\{[\Delta_f G^\circ(I) - \Delta_f G_i^\circ]/RT\}, \quad (2)$$

where  $y_1$  is the sum of the mole fractions of the individual

TABLE 2. Numbers of isomers in thermodynamic tables

Formulas	Number of isomers	Number of diradicals	Number of helicenes	Number in tables
Pyrene series				
C <sub>16</sub> H <sub>10</sub>	1	0	0	1
C <sub>20</sub> H <sub>12</sub>	3	0	0	3
C <sub>24</sub> H <sub>14</sub>	14	1	0	13
C <sub>28</sub> H <sub>16</sub>	68	6	8	55 <sup>a</sup>
Naphthopyrene series				
C <sub>22</sub> H <sub>12</sub>	3	1	0	2
C <sub>26</sub> H <sub>14</sub>	10	1	0	9
C <sub>30</sub> H <sub>16</sub>	87	9	2	76
Coronene series				
C <sub>24</sub> H <sub>12</sub>	1	0	0	1
C <sub>28</sub> H <sub>14</sub>	9	1	0	8
C <sub>32</sub> H <sub>16</sub>	55	9	0	46

<sup>a</sup> One diradical is a helicene.

isomers. The corresponding equations for the other standard thermodynamic properties  $C_p^\circ(I)$ ,  $S^\circ(I)$ , and  $\Delta_f H^\circ(I)$  can be derived by differentiating Eq. (1) with respect to temperature.<sup>18</sup> When standard Gibbs energies of formation of isomer groups are used in equilibrium calculations for ideal gases the equilibrium mole fractions of isomer groups are obtained.

For the polycyclic aromatic hydrocarbons the standard thermodynamic properties for an isomer group are interrelated by

$$\Delta_f G^\circ(I) = \Delta_f H^\circ(I) - T [S^\circ(I) - n_C S_{\text{graphite}}^\circ - (n_H/2) S_{\text{H}_2}^\circ(g)], \quad (3)$$

where  $n_C$  is the number of carbon atoms and  $n_H$  is the number of hydrogen atoms.

### 3. Calculations of Standard Thermodynamic Properties of Polycyclic Aromatic Hydrocarbons using the Benson Method

Experimental data on the polycyclic aromatic hydrocarbons are limited to the smaller molecules. Fortunately enough data were available for Stein, Golden, and Benson<sup>19,20</sup> to obtain group values for C<sub>B</sub>-(H), C<sub>FR</sub>-(C<sub>FR</sub>)(C<sub>B</sub>)<sub>2</sub>, and C<sub>FR</sub>-(C<sub>FR</sub>)<sub>2</sub>(C<sub>B</sub>) groups; the group values of C<sub>FR</sub>-(C<sub>FR</sub>)<sub>3</sub> were evaluated from graphite. The significance of these symbols is described in Benson's book.<sup>19</sup> The average difference between predicted and measured  $\Delta_f H^\circ(298)$  for 11 polycyclic aromatic hydrocarbons was < 2 kcal mol<sup>-1</sup> and generally was within experimental uncertainties. More recently, Stein and Fahr<sup>21</sup> have provided Benson group values with  $C_p^\circ$  values up to 3000 K. Although these values are similar to those of Stein, Golden, and Benson,<sup>20</sup> two major changes were made: (i) heat capacity values were calculated using the harmonic oscillator-rigid rotor approximation, and (ii) values of  $C_p^\circ$  for the [C<sub>FR</sub>-(C<sub>FR</sub>)<sub>2</sub>] group were derived from pyrene frequencies<sup>22</sup> rather than from graphite. These values are given in Table A.1 in the Appendix.

The procedures used in the estimation of chemical thermodynamic properties of the gaseous polycyclic aromatic hydrocarbons have been described in the preceding article.<sup>1</sup> The latest values have been used for various thermodynamic quantities.<sup>23-25</sup> Comparisons between properties of polycyclic aromatic hydrocarbons calculated using the Benson method and given in the TRC Thermodynamic Tables<sup>26</sup> for benzene, naphthalene, anthracene, and phenanthrene up to 1500 K were given in the preceding paper. There is a serious shortage of experimental data on polycyclic aromatic hydrocarbons. The only additional species for which a literature comparison is possible is pyrene. Pedley, Naylor, and Kirby<sup>27</sup> have selected the best value of the enthalpy of formation of pyrene gas at 298.15 K to be that determined by Smith *et al.*,<sup>28</sup> 225.7 ± 1.2 kJ mol<sup>-1</sup>. The Benson method with the parameters we have used yields 230.5 kJ mol<sup>-1</sup>. Since Smith *et al.*, studied vapor pressures over a range of temperatures and Wong and Westrum<sup>29</sup> obtained entropies of solid pyrene from low-temperature calorimetric studies, Smith *et al.*,<sup>28</sup> were able to calculate chemical thermodynamic properties from 200 to 600 K. At the three temperatures at which their table overlaps ours, the average difference in  $C_p^\circ$  is 2.6 J K<sup>-1</sup> mol<sup>-1</sup>, in  $S^\circ$  is 0.9 J K<sup>-1</sup> mol<sup>-1</sup>, in  $\Delta_f H^\circ$  is 5.0 kJ mol<sup>-1</sup>, and in  $\Delta_f G^\circ$  is 5.0 kJ mol<sup>-1</sup>. These are all within the uncertainties that Benson states for his method, that is ± 4 J K<sup>-1</sup> mol<sup>-1</sup> for  $C_p^\circ$  and  $S^\circ$ , and ± 8 kJ mol<sup>-1</sup> for  $\Delta_f H^\circ$  and  $\Delta_f G^\circ$ .

All values of thermodynamic quantities in this article are for a standard state pressure of 1 bar (0.1 MPa) in accordance with the recommendation of the International Union of Pure and Applied Chemistry.<sup>30</sup>

### 4. Tables of Standard Thermodynamic Properties of Polycyclic Aromatic Isomer Groups in the Pyrene, Naphthopyrene, and Coronene Series

In Tables 3-8 all of the values have been estimated using the Benson method. Since the increments from one isomer group to the next are C<sub>4</sub>H<sub>2</sub> in each series, the increments in the properties are given for each series at each temperature. These increments provide a basis for a linear extrapolation of standard thermodynamic properties to polycyclic aromatic isomer groups to higher carbon number.

Table 7 gives  $H^\circ(I, T) - H^\circ(I, 298.15 \text{ K})$ , the standard enthalpy for an isomer group relative to the isomer group at 298.15 K. Table 8 gives values for  $H^\circ(I, T) - H^\circ(I, 298.15 \text{ K}) + \Delta_f H^\circ(I, 298.15 \text{ K})$ , the standard enthalpy for the isomer group relative to the elements of 298.15 K. This quantity allows the direct calculation of heat effects when the reactants and products are at different temperatures.

### 5. Equilibrium Mole Fractions Within Polycyclic Aromatic Isomer Groups

The equilibrium mole fractions within isomer groups calculated from standard Gibbs energies of formation are given in Table 9 for the ideal gas state. Since the uncertainties in  $\Delta_f G^\circ(I)$  and  $\Delta_f G^\circ_f$  are about the same, the uncertainty in

Table 3. Standard heat capacity at constant pressure (J/K mol) for three series of polycyclic aromatic hydrocarbons

T/K	298	300	500	700	1000	1500	2000	2500	3000
Pyrene series									
C16H10	204.8	206.0	333.5	418.5	491.6	551.7	581.6	598.8	609.7
C20H12	261.2	262.6	419.3	522.8	611.6	684.5	720.7	741.6	754.8
C24H14	350.7	353.0	537.0	632.5	730.0	814.7	857.1	881.7	897.1
C28H16	384.4	386.4	601.4	734.4	849.4	946.0	994.4	1022.3	1040.0
Naphthopyrene series									
C22H12	279.2	280.8	453.4	563.4	656.5	732.2	769.4	790.5	803.8
C26H14	334.4	336.3	548.4	678.3	781.2	865.1	907.3	931.6	946.9
C30H16	446.9	449.2	639.0	772.3	893.8	994.7	1044.5	1072.9	1090.7
Coronene series									
C24H12	290.5	292.4	480.0	598.8	698.7	778.8	817.4	839.1	852.5
C28H14	344.7	346.8	563.7	702.2	818.6	911.8	956.8	982.2	997.8
C32H16	396.6	399.0	664.4	825.7	947.5	1045.4	1094.6	1122.8	1140.4

Table 3a. Increments per C4H2

T/K	298	300	500	700	1000	1500	2000	2500	3000
Pyrene series									
C20-C16	56.3	56.6	85.8	104.4	120.0	132.8	139.1	142.8	145.1
C24-C20	89.5	90.5	117.7	109.7	118.4	130.2	136.5	140.1	142.3
C28-C24	33.7	33.4	64.4	101.9	119.5	131.3	137.2	140.6	142.8
Naphthopyrene series									
C26-C22	55.2	55.5	95.0	115.0	124.7	132.9	137.9	141.1	143.1
C30-C26	112.5	112.8	90.5	94.0	112.6	129.6	137.2	141.3	143.7
Coronene series									
C28-C24	54.2	54.4	83.7	103.4	119.9	133.1	139.5	143.1	145.3
C32-C28	51.9	52.3	100.7	123.6	128.9	133.6	137.8	140.7	142.6

Table 4. Standard entropy for polycyclic aromatic hydrocarbon isomer groups in  $\text{J/K mol}$ 

T/K	298	300	500	700	1000	1500	2000	2500	3000
Pyrene series									
C16H10	402.2	403.5	539.6	666.4	829.3	1041.6	1204.9	1336.7	1446.9
C20H12	467.2	468.8	640.9	799.9	1002.9	1266.6	1469.0	1632.2	1768.7
C24H14	521.5	523.7	754.8	951.5	1195.0	1509.2	1750.0	1944.1	2106.3
C28H16	617.9	620.3	870.3	1095.3	1378.4	1743.6	2023.1	2248.3	2436.4
Naphthopyrene series									
C22H12	478.7	480.4	664.9	835.8	1053.8	1336.3	1552.5	1726.7	1872.1
C26H14	543.7	545.8	769.4	976.8	1238.0	1572.8	1828.0	2033.3	2204.6
C30H16	609.1	611.8	889.4	1127.0	1424.9	1809.1	2102.8	2339.2	2536.5
Coronene series									
C24H12	487.2	489.0	684.6	866.7	1098.9	1399.6	1629.5	1814.4	1968.7
C28H14	562.2	564.4	794.9	1008.5	1280.6	1632.7	1901.8	2118.2	2298.7
C32H16	632.1	634.5	903.2	1155.2	1472.7	1878.0	2186.2	2433.7	2640.1

Table 4a. Increments per C4H2

T/K	298	300	500	700	1000	1500	2000	2500	3000
Pyrene series									
C20-C16	65.0	65.3	101.3	133.4	173.6	225.0	264.1	295.6	321.8
C24-C20	54.3	54.9	113.9	151.6	192.1	242.6	281.0	311.8	337.6
C28-C24	96.4	96.6	115.5	143.8	183.5	234.5	273.2	304.2	330.1
Naphthopyrene series									
C26-C22	65.0	65.4	104.5	140.9	184.1	236.5	275.5	306.7	332.6
C30-C26	65.3	66.0	119.9	150.2	187.0	236.3	274.8	305.8	331.8
Coronene series									
C28-C24	75.1	75.4	110.3	141.8	181.7	233.1	272.3	303.8	330.1
C32-C28	69.8	70.2	108.3	146.7	192.1	245.3	284.4	315.5	341.3

Table 5. Standard enthalpy of formation for polycyclic aromatic hydrocarbon isomer groups in kJ/mol

T/K	298	300	500	700	1000	1500	2000	2500	3000
Pyrene series									
C16H10	230.5	230.4	218.1	211.0	206.7	208.2	212.6	216.1	217.0
C20H12	280.2	280.0	266.8	259.7	259.6	266.4	266.4	271.9	274.0
C24H14	331.7	331.6	326.6	322.4	319.7	324.3	332.1	338.2	340.1
C28H16	401.4	401.2	388.8	382.7	380.3	386.6	396.0	402.9	405.0
Naphthopyrene series									
C22H12	301.7	301.6	289.2	282.9	280.0	283.9	290.3	294.7	295.0
C26H14	351.2	351.0	338.5	334.3	334.4	341.4	349.9	355.4	356.1
C30H16	405.4	405.3	398.3	392.9	390.8	397.6	407.2	413.9	415.0
Coronene series									
C24H12	322.7	322.5	309.4	302.7	299.4	302.8	308.4	311.4	309.7
C28H14	372.2	372.0	357.5	350.4	347.6	353.3	361.4	366.5	366.2
C32H16	417.0	416.8	402.4	399.0	401.1	410.9	421.2	427.3	427.1

Table 5a. Increments per C4H2

T/K	298	300	500	700	1000	1500	2000	2500	3000
Pyrene series									
C20-C16	49.5	49.6	48.8	48.7	49.3	51.5	53.9	55.8	57.0
C24-C20	51.6	51.6	59.7	62.8	63.6	64.6	65.7	66.3	66.1
C28-C24	69.5	69.6	62.2	60.2	60.6	62.3	63.8	64.7	64.8
Naphthopyrene series									
C26-C22	49.5	49.4	49.4	51.4	54.4	57.5	59.6	60.8	61.1
C30-C26	54.2	54.3	59.8	58.6	56.4	56.2	57.3	58.4	58.9
Coronene series									
C28-C24	49.5	49.4	48.1	47.7	48.2	50.5	53.0	55.1	56.5
C32-C28	44.8	44.8	44.9	48.6	53.5	57.6	59.8	60.8	60.9

Table 6. Standard Gibbs energy of formation for polycyclic aromatic hydrocarbon isomer groups

T/K	298	300	500	700	1000	1500	2000	2500	3000
Pyrene series									
C16H10	332.8	333.4	405.9	482.5	599.8	796.3	991.7	1187.3	1379.9
C20H12	408.9	409.7	500.2	595.0	739.6	980.9	1220.3	1459.6	1695.1
C24H14	490.0	491.0	599.1	708.9	875.2	1152.3	1427.1	1701.8	1972.2
C28H16	576.8	577.9	699.7	825.4	1015.9	1333.0	1647.2	1961.2	2270.2
Naphthopyrene series									
C22H12	430.2	431.0	521.5	616.0	760.1	1000.3	1238.7	1477.1	1712.0
C26H14	506.3	507.3	615.5	727.2	895.8	1175.2	1451.8	1728.3	2000.6
C30H16	586.8	587.9	711.6	838.0	1029.3	1347.4	1662.6	1977.6	2287.7
Coronene series									
C24H12	452.3	453.1	544.2	639.5	784.7	1026.9	1267.5	1508.3	1746.0
C28H14	525.2	526.1	633.4	745.3	915.3	1198.2	1478.7	1759.1	2035.5
C32H16	595.0	596.1	720.4	848.4	1040.8	1358.6	1672.9	1987.0	2296.5

Table 6a. Increments per C4H2

T/K	298	300	500	700	1000	1500	2000	2500	3000
Pyrene series									
C20-C16	76.1	76.2	94.3	112.5	139.8	184.6	228.6	272.3	315.2
C24-C20	81.1	81.3	98.9	113.9	135.6	171.4	206.8	242.2	277.1
C28-C24	86.8	86.9	100.6	116.5	140.7	180.7	220.1	259.4	298.0
Naphthopyrene series									
C26-C22	76.1	76.2	94.0	111.2	135.7	174.8	213.1	251.2	288.6
C30-C26	80.5	80.7	96.1	110.7	133.6	172.3	210.8	249.3	287.1
Coronene series									
C28-C24	72.9	73.1	89.2	105.7	130.5	171.3	211.2	250.8	289.6
C32-C28	69.8	69.9	86.9	103.1	125.5	160.4	194.3	227.9	261.0



Table 7. Standard enthalpy for polycyclic aromatic hydrocarbon isomer groups relative to isomer groups at 298.15 K in kJ/mol

T/K	298	300	500	700	1000	1500	2000	2500	3000
Pyrene series									
C16H10	.0	.4	54.8	130.7	268.3	531.1	815.4	1110.6	1413.0
C20H12	.0	.5	69.2	164.3	335.8	562.3	1014.7	1380.5	1754.8
C24H14	.0	.7	92.8	210.1	415.8	804.6	1223.9	1658.9	2103.8
C28H16	.0	.7	100.7	235.4	474.7	926.6	1413.2	1917.6	2433.5
Naphthopyrene series									
C22H12	.0	.5	74.8	177.4	361.8	711.5	1088.1	1478.3	1877.1
C26H14	.0	.6	90.0	214.0	434.6	848.9	1293.3	1753.3	2223.2
C30H16	.0	.8	110.9	252.9	504.7	980.1	1491.5	2021.1	2562.3
Coronene series									
C24H12	.0	.5	78.7	187.6	383.9	756.0	1156.2	1570.6	1993.7
C28H14	.0	.6	92.7	220.5	450.5	886.3	1354.9	1839.9	2335.2
C32H16	.0	.7	108.2	258.9	527.0	1028.4	1565.0	2119.6	2685.7

Table 7a. Increments per C4H2

T/K	298	300	500	700	1000	1500	2000	2500	3000
Pyrene series									
C20-C16	.0	.1	14.5	33.6	67.5	131.1	199.3	269.8	341.8
C24-C20	.0	.2	23.5	45.8	79.9	142.4	209.2	278.4	349.0
C28-C24	.0	.1	7.9	25.2	58.9	122.0	189.3	258.8	329.7
Naphthopyrene series									
C26-C22	.0	.1	15.2	36.6	72.8	137.3	205.2	275.0	346.0
C30-C26	.0	.2	20.9	39.0	70.1	131.2	198.2	267.8	339.1
Coronene series									
C28-C24	.0	.1	14.0	32.8	66.6	130.3	198.6	269.3	341.5
C32-C28	.0	.1	15.5	38.4	76.6	142.1	210.1	279.7	350.6

Table 8. Standard enthalpy for polycyclic aromatic hydrocarbon isomer groups relative to the elements at 298.15 K in kJ/mol

T/K	298	300	500	700	1000	1500	2000	2500	3000
Pyrene series									
C16H10	230.5	230.9	285.3	361.2	498.8	761.7	1045.9	1341.2	1643.5
C20H12	280.2	280.7	349.4	444.5	616.0	942.4	1294.8	1660.6	2035.0
C24H14	331.7	332.4	424.5	541.9	747.5	1136.4	1555.6	1990.6	2435.6
C28H16	401.4	402.1	502.0	636.7	876.0	1328.0	1814.5	2319.0	2834.9
Naphthopyrene series									
C22H12	301.7	302.2	376.5	479.1	663.6	1013.3	1389.8	1780.0	2178.8
C26H14	351.2	351.8	441.2	565.2	785.8	1200.1	1644.5	2104.4	2574.3
C30H16	405.4	406.2	516.4	658.3	910.1	1385.5	1896.9	2426.5	2967.7
Coronene series									
C24H12	322.7	323.2	401.4	510.3	706.6	1078.7	1478.9	1893.3	2316.4
C28H14	372.2	372.8	464.9	592.6	822.6	1258.4	1727.1	2212.1	2707.4
C32H16	417.0	417.7	525.2	675.9	944.0	1445.4	1981.9	2536.6	3102.7

Table 8a. Increments per C4H2

T/K	298	300	500	700	1000	1500	2000	2500	3000
Pyrene series									
C20-C16	49.6	49.8	64.1	83.3	117.2	180.8	248.9	319.5	391.5
C24-C20	51.6	51.7	75.1	97.4	131.5	193.9	260.8	330.0	400.6
C28-C24	69.6	69.7	77.5	94.8	128.5	191.6	258.9	328.4	399.3
Naphthopyrene series									
C26-C22	49.5	49.6	64.7	86.1	122.2	186.8	254.7	324.4	395.5
C30-C26	54.2	54.4	75.2	93.2	124.3	185.5	252.4	322.1	393.4
Coronene series									
C28-C24	49.5	49.6	63.4	82.3	116.1	179.8	248.1	318.8	390.9
C32-C28	44.8	44.9	60.3	83.2	121.4	186.9	254.9	324.5	395.3

Table 9. Equilibrium mole fractions within polycyclic aromatic hydrocarbon isomer groups

T/K	298	300	500	700	1000	1500	2000	2500	3000
Pyrene series									
C20H12									
1. perylene	.3228	.3226	.2910	.2616	.2314	.2036	.1887	.1796	.1734
2. benzo(e)pyrene	.6457	.6452	.5819	.5232	.4627	.4072	.3774	.3591	.3469
3. benzo(a)pyrene	.0315	.0322	.1271	.2152	.3059	.3893	.4339	.4613	.4797
C24H14									
1. dibenzo(fg,op)naphthacene	.7956	.7906	.7548	.6854	.6319	.5842	.5489	.5246	.5064
2. naphtho(8,1,2-ghi)chrysene	.0137	.0144	.1520	.2225	.2473	.2512	.2489	.2462	.2438
3. benzo(pqr)picene	.0019	.0020	.0122	.0145	.0139	.0129	.0125	.0123	.0123
4. dibenzo(a,e)pyrene	.0777	.0790	.1113	.0703	.0421	.0271	.0217	.0192	.0178
5. benzo(b)perylene	.0019	.0020	.0122	.0145	.0139	.0129	.0125	.0123	.0123
6. zethrene	.0137	.0144	.1520	.2225	.2473	.2512	.2489	.2462	.2438
7. benzo(a)perylene	.0019	.0020	.0122	.0145	.0139	.0129	.0125	.0123	.0123
8. dibenzo(b,def)chrysene	.0137	.0144	.1520	.2225	.2473	.2512	.2489	.2462	.2438
9. dibenzo(def,p)chrysene	.0009	.0010	.0061	.0072	.0070	.0065	.0062	.0062	.0062
10. dibenzo(de,qr)naphthacene	.0003	.0004	.0166	.0458	.0817	.1201	.1581	.1686	.1686
11. dibenzo(c,mno)chrysene	.0009	.0010	.0061	.0072	.0070	.0065	.0062	.0062	.0062
12. benzo(rst)pentaphene	.0009	.0010	.0061	.0072	.0070	.0065	.0062	.0062	.0062
14. naphtho(2,1,8-qr-a)naphthacene	.0000	.0000	.0013	.0030	.0046	.0062	.0072	.0079	.0085
C28H16									
1. tribenzo(f,ij,no)tetracene	.0288	.0297	.1348	.1769	.1790	.1619	.1488	.1399	.1334
2. tribenzo(a,fg,op)tetracene	.3258	.3252	.1975	.1118	.0610	.0349	.0260	.0218	.0195
4. naphtho(2,1,8-def)picene	.0080	.0081	.0216	.0230	.0202	.0167	.0149	.0140	.0135
5. dibenzo(a,pqr)picene	.0002	.0002	.0024	.0047	.0067	.0080	.0086	.0090	.0093
6. benzo(a)naphtho(8,1,2-fgh)tetracene	.0080	.0081	.0216	.0230	.0202	.0167	.0149	.0140	.0135
7. naphtho(1,2,3,4-bqr)picene	.0450	.0445	.0158	.0073	.0034	.0018	.0013	.0011	.0010
8. naphtho(1,2-b)perylene	.0080	.0081	.0216	.0230	.0202	.0167	.0149	.0140	.0135
9. naphtho(2,1-a)perylene	.0080	.0081	.0216	.0230	.0202	.0167	.0149	.0140	.0135
10. dibenzo(j,pqr)picene	.3258	.3252	.1975	.1118	.0610	.0349	.0260	.0218	.0195
11. benzo(fg)naphtho(3,2,1-op)tetracene	.0080	.0081	.0216	.0230	.0202	.0167	.0149	.0140	.0135
12. tribenzo(f,m,pqr)tetracene	.0080	.0081	.0216	.0230	.0202	.0167	.0149	.0140	.0135
13. dibenzo(f,pqr)picene	.0011	.0011	.0017	.0015	.0011	.0009	.0007	.0007	.0007
14. tribenzo(a,hi,lmn)tetracene	.0080	.0081	.0216	.0230	.0202	.0167	.0149	.0140	.0135
15. naphtho(1,2-e)perylene	.0080	.0081	.0216	.0230	.0202	.0167	.0149	.0140	.0135
16. dibenzo(a,e)perylene	.0080	.0081	.0216	.0230	.0202	.0167	.0149	.0140	.0135
17. dibenzo(fg,qr)pentacene	.0450	.0445	.0158	.0073	.0034	.0018	.0013	.0011	.0010
18. naphtho(8,1,2-cde)picene	.0011	.0011	.0017	.0015	.0011	.0009	.0007	.0007	.0007
19. benzo(c)naphtho(8,1,2-ghi)tetracene	.0080	.0081	.0216	.0230	.0202	.0167	.0149	.0140	.0135
20. benzo(a)zethrene	.0002	.0002	.0024	.0047	.0067	.0080	.0086	.0090	.0093
21. dibenzo(de,kl)pentaphene	.0005	.0006	.0009	.0007	.0006	.0004	.0004	.0004	.0003
23. dibenzo(a,n)perylene	.0080	.0081	.0216	.0230	.0202	.0167	.0149	.0140	.0135
25. dibenzo(de,ij)pentaphene	.0011	.0011	.0017	.0015	.0011	.0009	.0007	.0007	.0007
26. benzo(f)zethrene	.0011	.0011	.0017	.0015	.0011	.0009	.0007	.0007	.0007
27. dibenzo(a,j)perylene	.0014	.0015	.0295	.0728	.1183	.1548	.1711	.1797	.1845
28. dibenzo(h,rst)pentaphene	.0225	.0223	.0079	.0036	.0017	.0009	.0007	.0007	.0005
29. dibenzo(b,tuv)picene	.0011	.0011	.0017	.0015	.0011	.0009	.0007	.0007	.0005
30. dibenzo(b,j)perylene	.0080	.0081	.0216	.0230	.0202	.0167	.0149	.0140	.0135
31. dibenzo(fg,ij)pentaphene	.0223	.0223	.0079	.0036	.0017	.0009	.0007	.0005	.0005
32. phrentho(1,2,3,4-pqr)tetracene	.0014	.0015	.0295	.0728	.1183	.1548	.1711	.1797	.1845
33. benzo(c)naphtho(1,2,3,4-pqr)tetracene	.0080	.0081	.0216	.0230	.0202	.0167	.0149	.0140	.0135

Table 9. Equilibrium mole fractions within polycyclic aromatic hydrocarbon isomer groups -- continued

T/K	298	300	500	700	1000	1500	2000	2500	3000
34.	.0040	.0041	.0108	.0115	.0101	.0083	.0075	.0070	.0067
37.	.0011	.0011	.0017	.0015	.0011	.0009	.0007	.0007	.0007
38.	.0225	.0223	.0079	.0036	.0017	.0009	.0005	.0005	.0005
39.	.0002	.0002	.0024	.0047	.0067	.0080	.0086	.0090	.0093
40.	.0002	.0002	.0024	.0047	.0067	.0080	.0086	.0090	.0093
43.	.0000	.0000	.0002	.0003	.0004	.0004	.0005	.0005	.0005
44.	.0000	.0000	.0024	.0047	.0067	.0080	.0086	.0090	.0093
45.	.0000	.0000	.0003	.0010	.0022	.0038	.0049	.0058	.0064
46.	.0002	.0002	.0024	.0047	.0067	.0080	.0086	.0090	.0093
47.	.0007	.0007	.0147	.0364	.0592	.0774	.0856	.0898	.0923
49.	.0011	.0011	.0017	.0015	.0011	.0009	.0007	.0007	.0007
50.	.0000	.0000	.0002	.0003	.0004	.0004	.0004	.0005	.0005
52.	.0011	.0011	.0017	.0015	.0011	.0009	.0007	.0007	.0007
53.	.0011	.0011	.0017	.0015	.0011	.0009	.0007	.0007	.0007
54.	.0011	.0011	.0017	.0015	.0011	.0009	.0007	.0007	.0007
55.	.0000	.0000	.0002	.0003	.0004	.0004	.0004	.0005	.0005
56.	.0002	.0002	.0024	.0047	.0067	.0080	.0086	.0090	.0093
59.	.0007	.0007	.0147	.0364	.0592	.0774	.0856	.0898	.0923
60.	.0000	.0000	.0002	.0003	.0004	.0004	.0004	.0005	.0005
61.	.0000	.0000	.0002	.0003	.0004	.0004	.0004	.0005	.0005
62.	.0000	.0000	.0002	.0003	.0004	.0004	.0004	.0005	.0005
63.	.0002	.0002	.0024	.0047	.0067	.0080	.0086	.0090	.0093
64.	.0000	.0000	.0002	.0003	.0004	.0004	.0004	.0005	.0005
65.	.0000	.0000	.0002	.0003	.0004	.0004	.0004	.0005	.0005
68.	.0000	.0000	.0000	.0001	.0001	.0002	.0002	.0003	.0003
Naphthopyrene series									
C22H12									
1.	.9535	.9524	.8207	.7086	.6020	.5112	.4652	.4377	.4197
3.	.0465	.0476	.1793	.2914	.3980	.4888	.5348	.5623	.5803
C26H14									
1.	.6310	.6301	.4931	.3444	.2113	.1258	.0942	.0791	.0706
2.	.3155	.3151	.2465	.1722	.1057	.0629	.0471	.0395	.0353
3.	.0038	.0039	.0135	.0177	.0175	.0150	.0135	.0127	.0122
4.	.0154	.0157	.0539	.0706	.0698	.0601	.0542	.0508	.0488
5.	.0154	.0157	.0539	.0706	.0698	.0601	.0542	.0508	.0488
6.	.0027	.0029	.0735	.2242	.4098	.5583	6.204	6.511	6.678
7.	.0004	.0004	.0059	.0146	.0231	.0288	.0311	.0326	.0338
10.	.0004	.0004	.0059	.0146	.0231	.0288	.0311	.0326	.0338
C30H16									
1.	.2102	.2140	.3002	.2084	.1347	.0881	.0696	.0600	.0542
2.	.0002	.0002	.0112	.0336	.0653	.0978	.1145	.1234	.1281
3.	.0051	.0053	.0328	.0425	.0445	.0421	.0400	.0385	.0375
4.	.0051	.0053	.0328	.0425	.0445	.0421	.0400	.0385	.0375
5.	.0073	.0073	.0060	.0034	.0019	.0011	.0009	.0008	.0007
6.	.5948	.5867	.1099	.0325	.0115	.0047	.0030	.0023	.0020
7.	.0051	.0053	.0328	.0425	.0445	.0421	.0400	.0385	.0375
8.	.0051	.0053	.0328	.0425	.0445	.0421	.0400	.0385	.0375
9.	.0051	.0053	.0328	.0425	.0445	.0421	.0400	.0385	.0375
10.	.0007	.0007	.0026	.0025	.0025	.0022	.0020	.0019	.0019
11.	.0001	.0001	.0036	.0088	.0147	.0201	.0230	.0247	.0259
12.	.0051	.0053	.0328	.0425	.0445	.0421	.0400	.0385	.0375
13.	.0290	.0293	.0240	.0135	.0076	.0045	.0035	.0030	.0027

Table 9. Equilibrium mole fractions within polycyclic aromatic hydrocarbon isomer groups -- continued

T/K	298	300	500	700	1000	1500	2000	2500	3000
14.	.0051	.0053	.0328	.0429	.0445	.0421	.0400	.0385	.0375
15.	.0001	.0001	.0036	.0088	.0147	.0201	.0230	.0247	.0259
16.	.0001	.0001	.0036	.0088	.0147	.0201	.0230	.0247	.0259
17.	.0290	.0293	.0240	.0135	.0076	.0045	.0035	.0030	.0027
18.	.0290	.0293	.0240	.0135	.0076	.0045	.0035	.0030	.0027
19.	.0051	.0053	.0328	.0429	.0445	.0421	.0400	.0385	.0375
20.	.0290	.0293	.0240	.0135	.0076	.0045	.0035	.0030	.0027
21.	.0000	.0000	.0003	.0006	.0008	.0010	.0012	.0012	.0013
22.	.0007	.0007	.0026	.0028	.0025	.0022	.0020	.0019	.0019
23.	.0001	.0001	.0036	.0088	.0147	.0201	.0230	.0247	.0259
24.	.0007	.0007	.0026	.0028	.0025	.0022	.0020	.0019	.0019
25.	.0007	.0007	.0026	.0028	.0025	.0022	.0020	.0019	.0019
26.	.0000	.0000	.0003	.0006	.0008	.0010	.0012	.0012	.0013
28.	.0001	.0001	.0036	.0088	.0147	.0201	.0230	.0247	.0259
29.	.0001	.0001	.0036	.0088	.0147	.0201	.0230	.0247	.0259
30.	.0001	.0001	.0036	.0088	.0147	.0201	.0230	.0247	.0259
31.	.0007	.0007	.0026	.0028	.0025	.0022	.0020	.0019	.0019
32.	.0007	.0007	.0026	.0028	.0025	.0022	.0020	.0019	.0019
33.	.0004	.0004	.0013	.0014	.0013	.0011	.0010	.0010	.0009
34.	.0051	.0053	.0328	.0429	.0445	.0421	.0400	.0385	.0375
35.	.0051	.0053	.0328	.0429	.0445	.0421	.0400	.0385	.0375
36.	.0001	.0001	.0036	.0088	.0147	.0201	.0230	.0247	.0259
37.	.0000	.0000	.0003	.0006	.0008	.0010	.0012	.0012	.0013
38.	.0007	.0007	.0026	.0028	.0025	.0022	.0020	.0019	.0019
39.	.0004	.0004	.0013	.0014	.0013	.0011	.0010	.0010	.0009
40.	.0000	.0000	.0004	.0004	.0004	.0006	.0006	.0006	.0006
43.	.0007	.0007	.0026	.0028	.0025	.0022	.0020	.0019	.0019
44.	.0000	.0000	.0003	.0006	.0008	.0010	.0012	.0012	.0013
45.	.0000	.0000	.0003	.0006	.0008	.0010	.0012	.0012	.0013
46.	.0000	.0000	.0003	.0006	.0008	.0010	.0012	.0012	.0013
48.	.0000	.0000	.0003	.0006	.0008	.0010	.0012	.0012	.0013
49.	.0000	.0000	.0003	.0006	.0008	.0010	.0012	.0012	.0013
50.	.0007	.0007	.0026	.0028	.0025	.0022	.0020	.0019	.0019
51.	.0007	.0007	.0026	.0028	.0025	.0022	.0020	.0019	.0019
52.	.0000	.0000	.0003	.0006	.0008	.0010	.0012	.0012	.0013
57.	.0000	.0000	.0003	.0006	.0008	.0010	.0012	.0012	.0013
58.	.0000	.0000	.0003	.0006	.0008	.0010	.0012	.0012	.0013
59.	.0001	.0001	.0036	.0088	.0147	.0201	.0230	.0247	.0259
60.	.0037	.0038	.0268	.0371	.0402	.0394	.0380	.0370	.0362
61.	.0037	.0038	.0268	.0371	.0402	.0394	.0380	.0370	.0362
62.	.0000	.0000	.0002	.0002	.0002	.0004	.0004	.0004	.0004
63.	.0000	.0000	.0002	.0002	.0002	.0004	.0004	.0004	.0004
67.	.0000	.0000	.0000	.0001	.0003	.0005	.0007	.0008	.0009
Coronene series									
C28H14									
1.	.3042	.3041	.2919	.2777	.2594	.2383	.2251	.2162	.2099
2.	.1521	.1521	.1460	.1369	.1297	.1192	.1125	.1081	.1049
3.	.0761	.0760	.0730	.0694	.0649	.0596	.0563	.0540	.0525
4.	.3042	.3041	.2919	.2777	.2594	.2383	.2251	.2162	.2099
5.	.0074	.0076	.0039	.0057	.0085	.0139	.0294	.0541	.1049
6.	.1521	.1521	.1460	.1369	.1297	.1192	.1125	.1081	.1049
7.	.0037	.0038	.0159	.0286	.0429	.0570	.0647	.0694	.0725
9.	.0002	.0002	.0035	.0117	.0283	.0545	.0744	.0892	.1003
C32H16									

Table 9. Equilibrium mole fractions within polycyclic aromatic hydrocarbon isomer groups -- continued

T/K	298	300	500	700	1000	1500	2000	2500	3000
1.	.0016	.0017	.0232	.0535	.0736	.0759	.0725	.0694	.0582
2.	.0090	.0091	.0170	.0169	.0125	.0082	.0063	.0054	.0043
3.	.0023	.0023	.0042	.0042	.0031	.0020	.0016	.0014	.0012
4.	.0045	.0046	.0085	.0085	.0063	.0041	.0032	.0027	.0024
5.	.9571	.9563	.7934	.5478	.3046	.1553	.1043	.0808	.0677
6.	.0000	.0000	.0025	.0110	.0243	.0363	.0417	.0446	.0452
7.	.0000	.0000	.0025	.0110	.0243	.0363	.0417	.0446	.0452
8.	.0090	.0091	.0170	.0169	.0125	.0082	.0063	.0054	.0049
9.	.0090	.0091	.0170	.0169	.0125	.0082	.0063	.0054	.0049
10.	.0002	.0002	.0019	.0035	.0041	.0039	.0036	.0035	.0034
11.	.0002	.0002	.0019	.0035	.0041	.0039	.0036	.0035	.0034
12.	.0002	.0002	.0019	.0035	.0041	.0039	.0036	.0035	.0034
13.	.0002	.0002	.0019	.0035	.0041	.0039	.0036	.0035	.0034
14.	.0000	.0000	.0002	.0007	.0014	.0019	.0021	.0022	.0023
15.	.0000	.0000	.0002	.0007	.0014	.0019	.0021	.0022	.0023
16.	.0000	.0000	.0025	.0110	.0243	.0363	.0417	.0446	.0452
17.	.0001	.0001	.0009	.0017	.0021	.0020	.0017	.0017	.0017
18.	.0000	.0000	.0025	.0110	.0243	.0363	.0417	.0446	.0452
19.	.0000	.0000	.0002	.0007	.0014	.0019	.0021	.0022	.0023
20.	.0002	.0002	.0019	.0035	.0041	.0039	.0036	.0035	.0034
21.	.0000	.0000	.0001	.0004	.0007	.0009	.0010	.0011	.0012
22.	.0016	.0017	.0232	.0535	.0736	.0759	.0725	.0694	.0669
23.	.0002	.0002	.0019	.0035	.0041	.0039	.0036	.0035	.0034
24.	.0002	.0002	.0019	.0035	.0041	.0039	.0036	.0035	.0034
25.	.0002	.0002	.0019	.0035	.0041	.0039	.0036	.0035	.0034
26.	.0002	.0002	.0019	.0035	.0041	.0039	.0036	.0035	.0034
27.	.0002	.0002	.0019	.0035	.0041	.0039	.0036	.0035	.0034
28.	.0000	.0000	.0025	.0110	.0243	.0363	.0417	.0446	.0452
29.	.0000	.0000	.0001	.0004	.0007	.0009	.0010	.0011	.0012
30.	.0000	.0000	.0003	.0023	.0080	.0173	.0240	.0285	.0320
31.	.0000	.0000	.0002	.0007	.0014	.0019	.0021	.0022	.0023
32.	.0000	.0000	.0025	.0110	.0243	.0363	.0417	.0446	.0452
33.	.0000	.0000	.0025	.0110	.0243	.0363	.0417	.0446	.0452
34.	.0016	.0017	.0232	.0535	.0736	.0759	.0725	.0694	.0669
35.	.0016	.0017	.0232	.0535	.0736	.0759	.0725	.0694	.0669
36.	.0000	.0000	.0002	.0007	.0014	.0019	.0021	.0022	.0023
37.	.0000	.0000	.0025	.0110	.0243	.0363	.0417	.0446	.0452
38.	.0000	.0000	.0025	.0110	.0243	.0363	.0417	.0446	.0452
39.	.0000	.0000	.0002	.0007	.0014	.0019	.0021	.0022	.0023
40.	.0000	.0000	.0002	.0007	.0014	.0019	.0021	.0022	.0023
42.	.0000	.0000	.0002	.0007	.0014	.0019	.0021	.0022	.0023
43.	.0000	.0000	.0002	.0007	.0014	.0019	.0021	.0022	.0023
47.	.0000	.0000	.0000	.0001	.0005	.0009	.0012	.0014	.0016
48.	.0000	.0000	.0002	.0007	.0014	.0019	.0021	.0022	.0023
49.	.0000	.0000	.0002	.0007	.0014	.0019	.0021	.0022	.0023
50.	.0000	.0000	.0002	.0007	.0014	.0019	.0021	.0022	.0023
51.	.0000	.0000	.0002	.0007	.0014	.0019	.0021	.0022	.0023
52.	.0000	.0000	.0002	.0007	.0014	.0019	.0021	.0022	.0023
53.	.0000	.0000	.0002	.0007	.0014	.0019	.0021	.0022	.0023
55.	.0000	.0000	.0000	.0001	.0005	.0009	.0012	.0014	.0016

the difference is nearly independent of the relative values of the two parameters, but the absolute uncertainty does increase with temperature.

The substances in the tables are named according to Dias,<sup>6</sup> who follows the IUPAC Revised and Collected Recommendations for the Nomenclature of Organic Chemistry, 1978.<sup>31</sup>

Table 9 shows that there are significant changes in the distribution of isomers within an isomer group when the temperature is changed from 298 K to 3000 K. At higher temperatures, the relative stabilities are determined primarily by the entropy. The distribution becomes more uniform as the temperature is raised.

## 6. Standard Thermodynamic Properties of Individual Polycyclic Aromatic Hydrocarbons

The values of  $C_p^\circ$ ,  $S^\circ$ ,  $\Delta_f H^\circ$ , and  $\Delta_f G^\circ$  calculated using the Benson method for the polycyclic aromatic hydrocarbons are given in Tables 10 to 13 in joules for a standard state pressure of 1 bar.

## 7. Discussion

The increments in the standard Gibbs energy of formation for the pyrene isomer groups per  $C_4H_2$  are very close to those for the benzene series.<sup>1</sup> The increments for the naphthopyrene series are a little smaller, and those for the coron-

ene series are still smaller. However, Fig. 2 of the preceding article<sup>1</sup> provides a useful guide to the partial pressures of acetylene and hydrogen that will cause higher isomer groups to predominate at a given temperature.

It is interesting to observe that the zethrenes (number 6 of  $C_{24}H_{14}$ , numbers 20 and 26 of  $C_{28}H_{16}$ ) do not contribute significantly to the isomer group thermodynamic properties. These molecules are remarkable because two double bonds in the center of the molecules are fixed. For this reason we were not sure that they should be included in the current calculations. However, this bond fixation appears to have no influence on the aromatic character of the absorption spectra and the reactivity.<sup>32</sup>

In making equilibrium calculations on polycyclic aromatic hydrocarbons in flames it is convenient to use isomer groups because of the geometric increase in the number of isomers with carbon numbers. It is also convenient to deal with series, such as the three described here, because the standard thermodynamic properties are very nearly a linear function of the number of  $C_4H_2$  increments added to the leading member of the series. When one or more reactants are available to a system at a constant chemical potential, their terms can be removed from the fundamental equation of thermodynamics by use of a Legendre transform to obtain a new thermodynamic potential that is a minimum at equilibrium.<sup>33</sup> Thus a new thermodynamic potential for isomer groups of polycyclic aromatic hydrocarbons can be defined at specified chemical potentials of acetylene and hydrogen. This makes it possible to treat a series as a single species in an equilibrium calculation.

Table 10. Standard heat capacity at constant pressure for polycyclic aromatic hydrocarbons in J/K mol

T/K	298	300	500	700	1000	1500	2000	2500	3000
Pyrene series									
C16H10									
pyrene	204.8	206.0	333.5	418.5	491.6	551.7	581.6	598.8	609.7
C20H12									
1. perylene	257.7	259.0	414.8	519.3	609.4	683.4	720.1	741.2	754.6
2. benzo(e)pyrene	257.7	259.0	414.8	519.3	609.4	683.4	720.1	741.2	754.6
3. benzo(a)pyrene	257.7	259.0	414.8	519.3	609.4	683.4	720.1	741.2	754.6
C24H14									
1. dibenzo(fg,op)naphthacene	310.5	312.1	496.0	620.2	727.2	815.1	858.6	883.6	899.4
2. naphtho(8,1,2-ghi)chrysene	306.1	307.8	494.8	619.1	725.9	813.2	856.3	881.0	896.6
3. benzo(pqr)picene	310.5	312.1	496.0	620.2	727.2	815.1	858.6	883.6	899.4
4. dibenzo(a,e)pyrene	310.5	312.1	496.0	620.2	727.2	815.1	858.6	883.6	899.4
5. benzo(b)perylene	310.5	312.1	496.0	620.2	727.2	815.1	858.6	883.6	899.4
6. zethrene	310.5	312.1	496.0	620.2	727.2	815.1	858.6	883.6	899.4
7. benzo(a)perylene	306.1	307.8	494.8	619.1	725.9	813.2	856.3	881.0	896.6
8. dibenzo(b,def)chrysene	310.5	312.1	496.0	620.2	727.2	815.1	858.6	883.6	899.4
9. dibenzo(def,p)chrysene	306.1	307.8	494.8	619.1	725.9	813.2	856.3	881.0	896.6
10. dibenzo(de,qr)naphthacene	310.5	312.1	496.0	620.2	727.2	815.1	858.6	883.6	899.4
11. dibenzo(c,mno)chrysene	306.1	307.8	494.8	619.1	725.9	813.2	856.3	881.0	896.6
12. benzo(rst)pentaphene	310.5	312.1	496.0	620.2	727.2	815.1	858.6	883.6	899.4
14. naphtho(2,1,8-qr) naphthacene	310.5	312.1	496.0	620.2	727.2	815.1	858.6	883.6	899.4
C28H16									
1. tribenzo(f,ij,ro)tetraphene	354.5	356.5	574.8	718.9	842.3	943.0	992.5	1020.9	1038.7
2. tribenzo(a,fg,op)tetracene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
4. naphtho(2,1,8-def)picene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
5. dibenzo(a,pqr)picene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
6. benzo(a)naphtho(8,1,2-fgh)tetracene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
7. naphtho(1,2,3,4-bqr)picene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
8. naphtho(1,2-b)perylene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
9. naphtho(2,1-a)perylene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
10. dibenzo(jg)naphtho(3,2,1-op)tetracene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
11. benzo(fg)naphtho(3,2,1-op)tetracene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
12. tribenzo(f,m,pqr)tetraphene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
13. dibenzo(f,pqr)picene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
14. tribenzo(a,hi,mn)tetracene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
15. naphtho(1,2-e)perylene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
16. dibenzo(a,e)perylene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
17. dibenzo(fg,qr)pentacene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
18. naphtho(8,1,2-cde)picene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
19. benzo(c)naphtho(8,1,2-ghi)tetraphene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
20. benzo(a)zethrene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
21. dibenzo(de,kl)pentaphene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
23. dibenzo(a,n)perylene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
25. dibenzo(de,ij)pentaphene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
26. benzo(f)zethrene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
27. dibenzo(a,j)perylene	354.5	356.5	574.8	718.9	842.3	943.0	992.5	1020.9	1038.7
28. dibenzo(h,rst)pentaphene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
29. dibenzo(b,tuv)picene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2



Table 10. Standard heat capacity at constant pressure for polycyclic aromatic hydrocarbons in J/K mol -- continued

T/K	298	300	500	700	1000	1500	2000	2500	3000
30. dibenzo(b,j)perylene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
31. dibenzo(fg,ij)pentaphene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
32. phenanthro(1,2,3,4-pqr)tetracene	354.5	356.5	574.8	718.9	842.3	943.0	992.5	1020.9	1038.7
33. benzo(c)naphtho(1,2,3,4-cqr)tetracene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
34. dibenzo(b,c)perylene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
37. naphtho(2,1,8-fgh)pentacene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
38. dibenzo(fg,st)pentaphene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
39. benzo(de)naphtho(1,2,3-qr)tetracene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
40. dibenzo(de,uv)pentaphene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
43. benzo(c)naphtho(1,2,3-mno)tetracene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
44. naphtho(8,1,2-cde)pentaphene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
45. tribenzo(c,m,pqr)tetracene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
46. naphtho(8,1,2-ocq)pentacene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
47. dibenzo(o,rst)pentaphene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
49. naphtho(1,2,3,4-rst)pentaphene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
50. dibenzo(c,rst)pentaphene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
52. benzo(a)naphtho(2,1,8-hij)tetracene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
53. benzo(a)naphtho(8,1,2-cds)tetracene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
54. dibenzo(de,st)pentacene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
55. dibenzo(de,qr)pentacene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
56. heptazethrene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
59. naphtho(2,3-a)perylene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
60. dibenzo(a,f)perylene	354.5	356.5	574.8	718.9	842.3	943.0	992.5	1020.9	1038.7
61. benzo(vwx)hexaphene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
62. anthra 2,1,9-qla)tetracene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
63. phenanthro(9,10,11-hif)tetracene	358.9	360.8	576.1	720.0	843.7	944.9	994.8	1023.4	1041.5
64. dibenzo(de,uv)pentacene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
65. benzo(a)naphtho(8,1,2-lmn)tetracene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
68. naphtho(2,1,8-jkl)pentacene	363.3	365.1	577.3	721.0	845.0	946.8	997.1	1026.0	1044.2
Naphthopyrene series									
C22H12									
1. benzo(ghi)perylene	274.11	275.70	447.41	559.07	654.05	731.08	768.73	790.14	803.52
3. dibenzo(def,mno)chrysene	274.11	275.70	447.41	559.07	654.05	731.08	768.73	790.14	803.52
C26H14									
1. dibenzo(e,ghi)perylene	326.9	328.8	528.7	659.9	771.8	862.8	907.2	932.5	948.3
2. naphtho(1,2,3,4-ghi)perylene	326.9	328.8	528.7	659.9	771.8	862.8	907.2	932.5	948.3
3. dibenzo(cd,lm)perylene	326.9	328.8	528.7	659.9	771.8	862.8	907.2	932.5	948.3
4. dibenzo(b,ghi)perylene	326.9	328.8	528.7	659.9	771.8	862.8	907.2	932.5	948.3
5. naphtho(8,1,2-bcd)perylene	326.9	328.8	528.7	659.9	771.8	862.8	907.2	932.5	948.3
6. benzo(e)anthanthrene	322.5	324.4	527.4	658.9	770.5	860.9	905.0	930.0	945.6
7. dibenzo(a,ghi)perylene	322.5	324.4	527.4	658.9	770.5	860.9	905.0	930.0	945.6
8. benzo(a)anthanthrene	326.9	328.8	528.7	659.9	771.8	862.8	907.2	932.5	948.3
10. benzo(b)anthanthrene	326.9	328.8	528.7	659.9	771.8	862.8	907.2	932.5	948.3
C30H16									
1. benzo(e)naphtho(1,2,3,4-ghi)perylene	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
2. tetrabenzo(de,hi,mn,qr)tetracene	371.0	373.2	607.4	758.7	886.9	990.7	1041.2	1069.8	1087.7
3. benzo(ghi)naphtho(1,2-e)perylene	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
4. benzo(lm)naphtho(1,8-ab)perylene	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
5. terrylene	379.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
6. tribenzo(fg,ij,rst)pentaphene	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
7. benzo(ghi)naphtho(2,1-e)perylene	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
8. phenanthro(1,2,3,4-ghi)perylene	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4

Table 10. Standard heat capacity at constant pressure for polycyclic aromatic hydrocarbons 1r J/K mol -- continued

T/K	298	300	500	700	1000	1500	2000	2500	3000
9.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1093.2
10.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
11.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
12.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
13.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
14.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
15.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
16.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
17.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
18.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
19.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
20.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
21.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
22.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
23.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
24.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
25.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
26.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
27.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
28.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
29.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
30.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
31.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
32.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
33.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
34.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
35.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
36.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
37.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
38.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
39.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
40.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
41.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
42.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
43.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
44.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
45.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
46.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
47.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
48.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
49.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
50.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
51.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
52.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
53.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
54.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
55.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
56.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
57.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
58.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
59.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
60.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
61.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
62.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
63.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
64.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
65.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4
66.	373.8	381.8	609.9	760.8	889.6	994.4	1045.7	1074.9	1093.2
67.	375.4	377.5	608.7	759.7	888.3	992.6	1043.5	1072.4	1090.4

Coronene series

C24H12

1. coronene

C28H14

290.55 292.36 480.03 598.81 698.69 778.76 817.38 839.09 852.47

Table 10. Standard heat capacity at constant pressure for polycyclic aromatic hydrocarbons in J/K mol -- continued

T/K	298	300	500	700	1000	1500	2000	2500	3000
1.	343.4	345.4	561.3	699.6	816.5	910.4	955.9	981.5	997.3
2.	343.4	345.4	561.3	699.6	816.5	910.4	955.9	981.5	997.3
3.	343.4	345.4	561.3	699.6	816.5	910.4	955.9	981.5	997.3
4.	343.4	345.4	561.3	699.6	816.5	910.4	955.9	981.5	997.3
5.	343.4	345.4	561.3	699.6	816.5	910.4	955.9	981.5	997.3
6.	343.4	345.4	561.3	699.6	816.5	910.4	955.9	981.5	997.3
7.	343.4	345.4	561.3	699.6	816.5	910.4	955.9	981.5	997.3
8.	343.4	345.4	561.3	699.6	816.5	910.4	955.9	981.5	997.3
9.	343.4	345.4	561.3	699.6	816.5	910.4	955.9	981.5	997.3
C32H16									
1.	391.8	394.1	641.3	799.4	932.9	1040.2	1092.1	1121.3	1139.4
2.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
3.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
4.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
5.	391.8	394.1	641.3	799.4	932.9	1040.2	1092.1	1121.3	1139.4
6.	391.8	394.1	641.3	799.4	932.9	1040.2	1092.1	1121.3	1139.4
7.	391.8	394.1	641.3	799.4	932.9	1040.2	1092.1	1121.3	1139.4
8.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
9.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
10.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
11.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
12.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
13.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
14.	391.8	394.1	641.3	799.4	932.9	1040.2	1092.1	1121.3	1139.4
15.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
16.	391.8	394.1	641.3	799.4	932.9	1040.2	1092.1	1121.3	1139.4
17.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
18.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
19.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
20.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
21.	391.8	394.1	641.3	799.4	932.9	1040.2	1092.1	1121.3	1139.4
22.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
23.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
24.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
25.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
26.	391.8	394.1	641.3	799.4	932.9	1040.2	1092.1	1121.3	1139.4
27.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
28.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
29.	391.8	394.1	641.3	799.4	932.9	1040.2	1092.1	1121.3	1139.4
30.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
31.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
32.	391.8	394.1	641.3	799.4	932.9	1040.2	1092.1	1121.3	1139.4
33.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
34.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
35.	391.8	394.1	641.3	799.4	932.9	1040.2	1092.1	1121.3	1139.4
36.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
37.	391.8	394.1	641.3	799.4	932.9	1040.2	1092.1	1121.3	1139.4
38.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
39.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
40.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
41.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
42.	391.8	394.1	641.3	799.4	932.9	1040.2	1092.1	1121.3	1139.4
43.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
44.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
45.	391.8	394.1	641.3	799.4	932.9	1040.2	1092.1	1121.3	1139.4
46.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
47.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
48.	391.8	394.1	641.3	799.4	932.9	1040.2	1092.1	1121.3	1139.4
49.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
50.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
51.	391.8	394.1	641.3	799.4	932.9	1040.2	1092.1	1121.3	1139.4
52.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
53.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1
54.	391.8	394.1	641.3	799.4	932.9	1040.2	1092.1	1121.3	1139.4
55.	396.2	398.5	642.5	800.5	934.3	1042.1	1094.4	1123.9	1142.1

Table 11. Standard entropy for polycyclic aromatic hydrocarbons (in J/K mol)

T/K	298	300	500	700	1000	1500	2000	2500	3000
Pyrene series									
C16H10									
pyrene	402.2	403.5	539.6	666.4	829.3	1041.6	1204.9	1336.7	1445.3
C20H12									
1. perylene	456.8	458.4	628.3	785.9	987.9	1251.0	1453.1	1616.3	1752.7
2. benzo(e)pyrene	468.3	464.1	634.1	791.7	993.4	1256.7	1458.9	1622.0	1758.4
3. benzo(a)pyrene		469.9	639.9	797.4	999.4	1262.5	1464.7	1627.8	1763.2
C24H14									
1. dibenzo(fg,op)naphthacene	511.4	513.3	717.1	905.4	1146.5	1460.3	1701.4	1895.9	2058.5
2. naphtho(8,1,2-ghi)chrysene	551.4	553.3	755.9	943.8	1184.5	1497.7	1738.1	1932.1	2094.2
3. benzo(pqr)picene	522.9	524.8	728.6	916.9	1158.0	1471.8	1712.9	1907.4	2070.0
4. dibenzo(a,e)pyrene	522.9	524.8	728.6	916.9	1158.0	1471.8	1712.9	1907.4	2070.0
5. benzo(b)perylene	517.1	519.1	722.8	911.1	1152.2	1466.1	1707.1	1901.7	2064.2
6. zethrene	551.4	553.3	755.9	943.8	1184.5	1497.7	1738.1	1932.1	2094.2
7. benzo(a)perylene	517.1	519.1	722.8	911.1	1152.2	1466.1	1707.1	1901.7	2064.2
8. dibenzo(b,def)chrysene	551.4	553.3	755.9	943.8	1184.5	1497.7	1738.1	1932.1	2094.2
9. dibenzo(def,f)chrysene	517.1	519.1	722.8	911.1	1152.2	1466.1	1707.1	1901.7	2064.2
10. dibenzo(de,cr)naphthacene	517.1	519.1	722.8	911.1	1152.2	1466.1	1707.1	1901.7	2064.2
11. dibenzo(c,mro)chrysene	551.4	553.3	755.9	943.8	1184.5	1497.7	1738.1	1932.1	2094.2
12. benzo(rst)pentaphene	517.1	519.1	722.8	911.1	1152.2	1466.1	1707.1	1901.7	2064.2
14. naphtho(2,1,8-qr)naphthacene	522.9	524.8	728.6	916.9	1158.0	1471.8	1712.9	1907.4	2070.0
C28H16									
1. tribenzo(f,ij,no)tetraphene	628.7	630.9	866.1	1084.4	1363.8	1727.1	2005.9	2230.6	2418.5
2. tribenzo(a,fg,op)tetracene	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
4. naphtho(2,1,8-def)picene	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
5. dibenzo(a,pqr)picene	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
6. benzo(a)naphtho(8,1,2-fgh)tetracene	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
7. naphtho(1,2,3,4-bqr)picene	577.5	579.8	817.3	1036.3	1316.6	1681.2	1961.2	2187.0	2375.8
8. naphtho(1,2-b)perylene	577.5	579.8	817.3	1036.3	1316.6	1681.2	1961.2	2187.0	2375.8
9. naphtho(2,1-a)perylene	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
10. dibenzo(j,pcr)picene	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
11. benzo(fg)naphtho(3,2,1-op)tetracene	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
12. tribenzo(f,m,pqr)tetraphene	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
13. dibenzo(f,pcr)picene	577.5	579.8	817.3	1036.3	1316.6	1681.2	1961.2	2187.0	2375.8
14. tribenzo(a,hi,mn)tetracene	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
15. naphtho(1,2-e)perylene	571.7	574.0	811.6	1030.6	1310.8	1675.4	1955.4	2181.3	2370.1
16. dibenzo(a,e)perylene	577.5	579.8	817.3	1036.3	1316.6	1681.2	1961.2	2187.0	2375.8
17. dibenzo(fg,cr)pentacene	577.5	579.8	817.3	1036.3	1316.6	1681.2	1961.2	2187.0	2375.8
18. naphtho(8,1,2-cde)picene	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
19. benzo(a)zethrene	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
20. benzo(a)zethrene	571.7	574.0	811.6	1030.6	1310.8	1675.4	1955.4	2181.3	2370.1
21. dibenzo(de,kl)pentaphene	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
23. dibenzo(a,n)perylene	577.5	579.8	817.3	1036.3	1316.6	1681.2	1961.2	2187.0	2375.8
25. dibenzo(de,ij)pentaphene	577.5	579.8	817.3	1036.3	1316.6	1681.2	1961.2	2187.0	2375.8
26. benzo(f)zethrene	628.7	630.9	866.1	1084.4	1363.8	1727.1	2005.9	2230.6	2418.5
27. dibenzo(a,j)perylene	571.7	574.0	811.6	1030.6	1310.8	1675.4	1955.4	2181.3	2370.1
28. dibenzo(m,rst)pentaphene	577.5	579.8	817.3	1036.3	1316.6	1681.2	1961.2	2187.0	2375.8
29. dibenzo(b,tuv)picene	577.5	579.8	817.3	1036.3	1316.6	1681.2	1961.2	2187.0	2375.8

Table 11. Standard entropy for polycyclic aromatic hydrocarbons in J/K mol continued

T/K	298	300	500	700	1000	1500	2000	2500	3000
30.	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
31.	571.7	574.0	811.6	1030.6	1310.8	1675.4	1955.4	2181.3	2370.1
32.	634.5	636.7	871.9	1090.1	1369.5	1732.8	2011.6	2236.4	2424.2
33.	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
34.	600.2	602.5	838.8	1057.5	1337.3	1701.2	1980.6	2206.0	2394.3
37.	571.5	573.8	817.3	1036.3	1316.6	1681.2	1961.2	2187.0	2375.8
38.	571.7	574.0	811.6	1030.6	1310.8	1675.4	1955.4	2181.3	2370.1
39.	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
40.	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
43.	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
44.	577.5	579.8	817.3	1036.3	1316.6	1681.2	1961.2	2187.0	2375.8
45.	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
46.	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
47.	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
49.	628.7	630.9	866.1	1084.4	1363.8	1727.1	2005.9	2230.6	2418.5
50.	577.5	579.8	817.3	1036.3	1316.6	1681.2	1961.2	2187.0	2375.8
52.	577.5	579.8	817.3	1036.3	1316.6	1681.2	1961.2	2187.0	2375.8
53.	577.5	579.8	817.3	1036.3	1316.6	1681.2	1961.2	2187.0	2375.8
54.	577.5	579.8	817.3	1036.3	1316.6	1681.2	1961.2	2187.0	2375.8
55.	577.5	579.8	817.3	1036.3	1316.6	1681.2	1961.2	2187.0	2375.8
56.	571.7	574.0	811.6	1030.6	1310.8	1675.4	1955.4	2181.3	2370.1
59.	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
60.	628.7	630.9	866.1	1084.4	1363.8	1727.1	2005.9	2230.6	2418.5
61.	577.5	579.8	817.3	1036.3	1316.6	1681.2	1961.2	2187.0	2375.8
62.	577.5	579.8	817.3	1036.3	1316.6	1681.2	1961.2	2187.0	2375.8
63.	606.0	608.2	844.6	1063.2	1343.1	1707.0	1986.4	2211.7	2400.0
64.	571.7	574.0	811.6	1030.6	1310.8	1675.4	1955.4	2181.3	2370.1
65.	577.5	579.8	817.3	1036.3	1316.6	1681.2	1961.2	2187.0	2375.8
68.	577.5	579.8	817.3	1036.3	1316.6	1681.2	1961.2	2187.0	2375.8
Naphthopyrene series									
C22H12									
1.	477.75	479.45	662.23	832.05	1049.18	1331.04	1547.07	1721.11	1866.44
3.	477.75	479.45	662.23	832.05	1049.18	1331.04	1547.07	1721.11	1866.44
C26H14									
1.	538.1	540.1	756.7	957.3	1213.5	1546.1	1801.1	2006.5	2178.0
2.	532.3	534.4	751.0	951.5	1207.8	1540.4	1795.3	2000.7	2172.2
3.	526.6	528.6	745.2	945.7	1202.0	1534.6	1789.6	1995.0	2166.5
4.	538.1	540.1	756.7	957.3	1213.5	1546.1	1801.1	2006.5	2178.0
5.	538.1	540.1	756.7	957.3	1213.5	1546.1	1801.1	2006.5	2178.0
6.	538.1	540.1	756.7	957.3	1213.5	1546.1	1801.1	2006.5	2178.0
7.	566.6	568.6	784.0	984.2	1240.0	1572.0	1826.3	2031.2	2202.2
8.	538.1	540.1	756.7	957.3	1213.5	1546.1	1801.1	2006.5	2178.0
10.	538.1	540.1	756.7	957.3	1213.5	1546.1	1801.1	2006.5	2178.0
C30H16									
1.	621.2	623.5	872.7	1103.6	1398.6	1781.3	2074.6	2310.8	2508.0
2.	638.2	640.5	888.5	1119.0	1413.5	1795.6	2088.3	2324.0	2520.7
3.	621.2	623.5	872.7	1103.6	1398.6	1781.3	2074.6	2310.8	2508.0
4.	621.2	623.5	872.7	1103.6	1398.6	1781.3	2074.6	2310.8	2508.0
5.	581.2	583.5	834.0	1065.2	1360.6	1744.0	2037.8	2274.6	2472.3
6.	586.5	588.3	839.7	1071.0	1366.3	1749.7	2043.6	2280.3	2478.1
7.	621.2	623.5	872.7	1103.6	1398.6	1781.3	2074.6	2310.8	2508.0
8.	621.2	623.5	872.7	1103.6	1398.6	1781.3	2074.6	2310.8	2508.0

Table 11. Standard entropy for polycyclic aromatic hydrocarbons in J/K mol -- continued

T/K	288	300	500	700	1000	1500	2000	2500	3000
9.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
10.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
11.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
12.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
13.	586.9	589.3	839.7	1071.0	1366.3	1719.7	2043.6	2280.3	2478.1
14.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
15.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
16.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
17.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
18.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
19.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
20.	586.9	589.3	839.7	1071.0	1366.3	1719.7	2043.6	2280.3	2478.1
21.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
22.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
23.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
24.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
25.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
26.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
27.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
28.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
29.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
30.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
31.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
32.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
33.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
34.	586.9	589.3	839.7	1071.0	1366.3	1719.7	2043.6	2280.3	2478.1
35.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
36.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
37.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
38.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
39.	586.9	589.3	839.7	1071.0	1366.3	1719.7	2043.6	2280.3	2478.1
40.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
41.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
42.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
43.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
44.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
45.	586.9	589.3	839.7	1071.0	1366.3	1719.7	2043.6	2280.3	2478.1
46.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
47.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
48.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
49.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
50.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
51.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
52.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
53.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
54.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8
55.	586.9	589.3	839.7	1071.0	1366.3	1719.7	2043.6	2280.3	2478.1
56.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
57.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
58.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
59.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
60.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
61.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
62.	621.2	623.5	872.7	1103.6	1398.6	1751.3	2074.6	2310.8	2508.0
63.	615.4	617.8	867.0	1097.9	1392.8	1715.5	2068.8	2305.0	2502.3
64.	592.7	595.1	845.5	1076.7	1372.1	1735.5	2049.3	2286.1	2483.8

Coronene series

C24H12

1. coronene

C28H14

487.18 488.99 684.61 866.68 1098.93 1399.57 1629.48 1814.42 1968.67

Table 11. Standard entropy for polycyclic aromatic hydrocarbons in J/k mol -- continued

T/K	298	300	500	700	1000	1500	2000	2500	3000
1.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
2.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
3.	541.8	543.9	773.3	986.1	1257.5	1608.9	1877.7	2094.0	2274.5
4.	553.3	555.4	784.9	997.7	1269.0	1620.4	1889.3	2105.6	2286.0
5.	553.3	555.4	784.9	997.7	1269.0	1620.4	1889.3	2105.6	2286.0
6.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
7.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
8.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
9.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
10.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
11.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
12.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
13.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
14.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
15.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
16.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
17.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
18.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
19.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
20.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
21.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
22.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
23.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
24.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
25.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
26.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
27.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
28.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
29.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
30.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
31.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
32.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
33.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
34.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
35.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
36.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
37.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
38.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
39.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
40.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
41.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
42.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
43.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
44.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
45.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
46.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
47.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
48.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
49.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
50.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
51.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
52.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
53.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
54.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2
55.	547.5	549.7	779.1	991.9	1263.3	1614.7	1883.5	2099.8	2280.2

Table 12. Standard enthalpy of formation for polycyclic aromatic hydrocarbons in kJ/mol

T/K	298	300	500	700	1000	1500	2000	2500	3000
Pyrene series									
C16H10									
Pyrene									
	230.5	230.4	218.1	211.0	206.7	208.2	212.6	216.1	217.0
C20H12									
1. perylene	279.9	279.7	285.7	257.7	253.2	256.0	262.4	267.6	269.6
2. benzo(e)pyrene	279.9	279.7	285.7	257.7	253.2	256.0	262.4	267.6	269.6
3. benzo(a)pyrene	289.1	288.9	274.9	266.9	262.4	265.3	271.6	276.8	278.8
C24H14									
1. dibenzo(fg,op)naphthacene	329.2	329.1	313.3	304.4	299.7	303.9	312.3	319.2	322.2
2. naphtho(8,1,2-ghi)chrysene	351.2	351.1	334.8	325.7	320.7	324.1	331.4	337.1	338.8
3. benzo(pqr)picene	347.7	347.5	331.7	322.8	318.1	322.3	330.7	337.6	340.6
4. dibenzo(a,e)pyrene	338.5	338.3	322.5	313.6	308.9	313.1	321.5	328.4	331.4
5. benzo(b)perylene	338.5	338.3	322.5	313.6	308.9	313.1	321.5	328.4	331.4
6. zethrene	347.7	347.5	331.7	322.8	318.1	322.3	330.7	337.6	340.6
7. benzo(a)perylene	351.2	351.1	334.8	325.7	320.7	324.1	331.4	337.1	338.8
8. dibenzo(b,def)chrysene	347.7	347.5	331.7	322.8	318.1	322.3	330.7	337.6	340.6
9. dibenzo(def,gh)chrysene	351.2	351.1	334.8	325.7	320.7	324.1	331.4	337.1	338.8
10. dibenzo(de,qr)naphthacene	347.7	347.5	331.7	322.8	318.1	322.3	330.7	337.6	340.6
11. dibenzo(c,mmo)chrysene	360.5	360.3	344.0	334.9	329.9	333.3	340.6	346.3	348.0
12. benzo(rst)pentaphene	347.7	347.5	331.7	322.8	318.1	322.3	330.7	337.6	340.6
14. naphtho(2,1,3-qr)naphthacene	356.9	356.7	340.9	332.0	327.3	331.6	339.9	346.8	349.8
C28H16									
1. tribenzo(f,ij,no)tetraphene	413.4	413.2	394.7	384.5	379.0	382.9	391.2	397.4	398.8
2. tribenzo(a,fg,op)tetracene	400.6	400.4	382.4	372.4	367.2	372.0	381.3	388.7	391.4
3. naphtho(2,1,8-def)picene	409.8	409.6	391.6	381.6	376.4	381.2	390.5	397.9	400.6
4. dibenzo(a,pqr)picene	419.0	418.8	400.8	390.8	385.6	390.4	399.7	407.1	409.8
5. benzo(a)naphtho(8,1,2-fgh)tetracene	409.8	409.6	391.6	381.6	376.4	381.2	390.5	397.9	400.6
6. naphtho(1,2,3,4-bqr)picene	397.0	396.8	379.3	369.5	364.6	370.2	380.6	389.2	393.2
7. naphtho(1,2-b)perylene	397.0	396.8	379.3	369.5	364.6	370.2	380.6	389.2	393.2
8. naphtho(1,2-b)perylene	409.8	409.6	391.6	381.6	376.4	381.2	390.5	397.9	400.6
9. naphtho(2,1-a)perylene	409.8	409.6	391.6	381.6	376.4	381.2	390.5	397.9	400.6
10. dibenzo(j,pqr)picene	400.6	400.4	382.4	372.4	367.2	372.0	381.3	388.7	391.4
11. benzo(fg)naphtho(3,2,1-op)tetracene	409.8	409.6	391.6	381.6	376.4	381.2	390.5	397.9	400.6
12. tribenzo(f,m,pqr)tetraphene	409.8	409.6	391.6	381.6	376.4	381.2	390.5	397.9	400.6
13. dibenzo(f,pqr)picene	406.2	406.0	388.5	378.7	373.8	379.4	389.8	398.4	402.4
14. tribenzo(a,hi,mm)tetracene	406.2	406.0	388.5	378.7	373.8	379.4	389.8	398.4	402.4
15. naphtho(1,2-a)perylene	409.8	409.6	391.6	381.6	376.4	381.2	390.5	397.9	400.6
16. dibenzo(a,e)perylene	409.8	409.6	391.6	381.6	376.4	381.2	390.5	397.9	400.6
17. dibenzo(fg,qr)pentacene	397.0	396.8	379.3	369.5	364.6	370.2	380.6	389.2	393.2
18. naphtho(8,1,2-cde)picene	406.2	406.0	388.5	378.7	373.8	379.4	389.8	398.4	402.4
19. benzo(c)naphtho(8,1,2-ghi)tetraphene	409.8	409.6	391.6	381.6	376.4	381.2	390.5	397.9	400.6
20. benzo(a)zethrene	419.0	418.8	400.8	390.8	385.6	390.4	399.7	407.1	409.8
21. dibenzo(de,kl)pentaphene	406.2	406.0	388.5	378.7	373.8	379.4	389.8	398.4	402.4
22. dibenzo(a,n)perylene	409.8	409.6	391.6	381.6	376.4	381.2	390.5	397.9	400.6
23. dibenzo(de,ij)pentaphene	406.2	406.0	388.5	378.7	373.8	379.4	389.8	398.4	402.4
24. benzo(f)zethrene	406.2	406.0	388.5	378.7	373.8	379.4	389.8	398.4	402.4
25. benzo(f)zethrene	422.6	422.4	403.9	393.7	388.2	392.1	400.4	406.6	408.0
26. benzo(a,j)perylene	422.6	422.4	403.9	393.7	388.2	392.1	400.4	406.6	408.0
27. dibenzo(h,rst)pentaphene	397.0	396.8	379.3	369.5	364.6	370.2	380.6	389.2	393.2
28. dibenzo(h,rst)pentaphene	397.0	396.8	379.3	369.5	364.6	370.2	380.6	389.2	393.2
29. dibenzo(b,tuv)picene	406.2	406.0	388.5	378.7	373.8	379.4	389.8	398.4	402.4



Table 12. Standard enthalpy of formation for polycyclic aromatic hydrocarbons in kJ/mol -- continued

T/K	298	300	500	700	1000	1500	2000	2500	3000
30. dibenzo(b,j)perylene	409.8	409.6	391.6	381.6	376.4	381.2	390.5	397.9	400.6
31. dibenzo(fg,ij)pentaphene	397.0	396.8	379.3	369.5	364.6	370.2	380.6	389.2	393.2
32. phenanthro(1,2,3,4-pqr)tetracene	422.6	422.4	403.9	393.7	388.2	392.1	400.4	406.6	408.0
33. dibenzo(c)naphtho(1,2,3,4-pqr)tetracene	409.8	409.6	391.6	381.6	376.4	381.2	390.5	397.9	400.6
34. dibenzo(b,c)perylene	409.8	409.6	391.6	381.6	376.4	381.2	390.5	397.9	400.6
37. naphtho(2,1,8-gh)pentacene	406.2	406.0	388.5	378.7	373.8	379.4	389.8	398.4	402.4
38. dibenzo(fg,ist)pentaphene	397.0	396.8	379.3	369.5	364.6	370.2	380.6	389.2	393.2
39. benzo(de)naphtho(1,2,3-gr)tetracene	419.0	418.8	400.8	390.8	385.6	390.4	399.7	407.1	409.8
40. dibenzo(de,uv)pentaphene	419.0	418.8	400.8	390.8	385.6	390.4	399.7	407.1	409.8
43. dibenzo(c)naphtho(8,1,2-mno)tetracene	415.4	415.2	397.7	387.9	383.1	388.6	399.0	407.6	411.7
44. naphtho(8,1,2-cde)pentaphene	419.0	418.8	400.8	390.8	385.6	390.4	399.7	407.1	409.8
45. tribenzo(c,m,pqr)tetracene	428.2	428.0	410.0	400.0	394.8	399.6	408.9	416.3	419.0
46. naphtho(8,1,2-opq)pentacene	419.0	418.8	400.8	390.8	385.6	390.4	399.7	407.1	409.8
47. dibenzo(o,rst)pentaphene	422.6	422.4	403.9	393.7	388.2	392.1	400.4	406.6	408.0
49. naphtho(1,2,3,4-rst)pentaphene	406.2	406.0	388.5	378.7	373.8	379.4	389.8	398.4	402.4
50. dibenzo(c,rst)pentaphene	415.4	415.2	397.7	387.9	383.1	388.6	399.0	407.6	411.7
52. benzo(a)naphtho(2,1,8-hi)j)tetracene	406.2	406.0	388.5	378.7	373.8	379.4	389.8	398.4	402.4
53. benzo(a)naphtho(8,1,2-cde)tetracene	406.2	406.0	388.5	378.7	373.8	379.4	389.8	398.4	402.4
54. dibenzo(de,st)pentacene	406.2	406.0	388.5	378.7	373.8	379.4	389.8	398.4	402.4
55. dibenzo(de,qr)pentacene	406.2	406.0	388.5	378.7	373.8	379.4	389.8	398.4	402.4
56. heptazethrene	415.4	415.2	397.7	387.9	383.1	388.6	399.0	407.6	411.7
59. naphtho(2,3-a)perylene	419.0	418.8	400.8	390.8	385.6	390.4	399.7	407.1	409.8
60. dibenzo(a,f)perylene	422.6	422.4	403.9	393.7	388.2	392.1	400.4	406.6	408.0
61. benzo(vwx)hexaphene	415.4	415.2	397.7	387.9	383.1	388.6	399.0	407.6	411.7
62. anthra 2,1,9-qa)tetracene	415.4	415.2	397.7	387.9	383.1	388.6	399.0	407.6	411.7
63. phenanthro(9,10,1-hif)tetracene	419.0	418.8	400.8	390.8	385.6	390.4	399.7	407.1	409.8
64. dibenzo(de,uv)pentacene	415.4	415.2	397.7	387.9	383.1	388.6	399.0	407.6	411.7
65. benzo(a)naphtho(8,1,2-lm)tetracene	415.4	415.2	397.7	387.9	383.1	388.6	399.0	407.6	411.7
68. naphtho(2,1,8-jkl)pentacene	424.6	424.4	406.9	397.1	392.3	397.9	408.2	416.8	420.9
Naphthopyrene series									
C22H12									
1. benzo(ghi)perylene	301.29	301.13	287.53	280.17	276.32	279.45	285.42	289.50	289.66
3. dibenzo(def,mno)chrysene	310.50	310.34	296.73	289.38	285.52	288.65	294.62	298.70	298.87
C26H14									
1. dibenzo(e,ghi)perylene	350.7	350.5	335.1	326.9	322.8	327.3	335.3	341.1	342.3
2. naphtho(1,2,3,4-ghi)perylene	350.7	350.5	335.1	326.9	322.8	327.3	335.3	341.1	342.3
3. dibenzo(cd,lm)perylene	359.9	359.7	344.3	336.1	332.0	336.5	344.5	350.3	351.5
4. dibenzo(b,ghi)perylene	359.9	359.7	344.3	336.1	332.0	336.5	344.5	350.3	351.5
5. naphtho(8,1,2-bcd)perylene	359.9	359.7	344.3	336.1	332.0	336.5	344.5	350.3	351.5
6. benzo(e)anthanthrene	359.9	359.7	344.3	336.1	332.0	336.5	344.5	350.3	351.5
7. dibenzo(a,ghi)perylene	372.7	372.5	356.7	348.2	343.8	347.5	354.4	359.5	360.7
8. benzo(a)anthanthrene	369.1	368.9	353.5	345.3	341.2	345.7	353.7	359.5	360.7
10. benzo(b)anthanthrene	369.1	368.9	353.5	345.3	341.2	345.7	353.7	359.5	360.7
C30H16									
1. benzo(e)naphtho(1,2,3,4-ghi)perylene	412.8	412.6	395.1	385.7	381.1	386.2	395.0	401.3	402.3
2. tetrabenzo(de,hi,lm,qr)tetracene	434.8	434.6	416.6	407.0	402.1	406.3	414.2	418.9	419.9
3. benzo(ghi)naphtho(1,2-e)perylene	422.0	421.8	404.3	394.9	390.3	395.4	404.3	410.5	411.5
4. benzo(lm)naphtho(1,8-ab)perylene	422.0	421.8	404.3	394.9	390.3	395.4	404.3	410.5	411.5
5. terrylene	409.2	409.0	391.9	382.8	378.5	384.4	394.3	401.6	404.1
6. tribenzo(fg,ij,rst)pentaphene	400.0	399.8	382.7	373.6	369.3	375.2	385.1	392.6	394.9
7. benzo(ghi)naphtho(2,1-e)perylene	422.0	421.8	404.3	394.9	390.3	395.4	404.3	410.5	411.5
8. phenanthro(1,2,3,4-ghi)perylene	422.0	421.8	404.3	394.9	390.3	395.4	404.3	410.5	411.5

Table 12. Standard enthalpy of formation for polycyclic aromatic hydrocarbons in kJ/mol -- continued

T/K	298	300	500	700	1000	1500	2000	2500	3000
9.	422.0	421.8	404.3	394.9	390.3	395.4	404.3	410.5	411.5
10.	418.4	418.2	401.1	392.0	387.7	393.6	403.6	410.5	413.3
11.	431.2	431.0	413.5	404.1	398.5	404.6	413.5	419.7	420.7
12.	422.0	421.8	404.3	394.9	390.3	395.4	404.3	410.5	411.5
13.	409.2	409.0	381.9	382.8	378.5	384.4	394.3	401.8	404.1
14.	422.0	421.8	404.3	394.9	390.3	395.4	404.3	410.5	411.5
15.	431.2	431.0	413.5	404.1	398.5	404.6	413.5	419.7	420.7
16.	431.2	431.0	413.5	404.1	398.5	404.6	413.5	419.7	420.7
17.	409.2	409.0	391.9	382.8	378.5	384.4	394.3	401.8	404.1
18.	409.2	409.0	391.9	382.8	378.5	384.4	394.3	401.8	404.1
19.	409.2	409.0	391.9	382.8	378.5	384.4	394.3	401.8	404.1
20.	409.2	409.0	391.9	382.8	378.5	384.4	394.3	401.8	404.1
21.	427.6	427.4	410.3	401.2	396.9	402.8	412.8	420.2	422.5
22.	418.4	418.2	401.1	392.0	387.7	393.6	403.6	411.0	413.3
23.	431.2	431.0	413.5	404.1	398.5	404.6	413.5	419.7	420.7
24.	418.4	418.2	401.1	392.0	387.7	393.6	403.6	411.0	413.3
25.	418.4	418.2	401.1	392.0	387.7	393.6	403.6	411.0	413.3
26.	427.6	427.4	410.3	401.2	396.9	402.8	412.8	420.2	422.5
27.	431.2	431.0	413.5	404.1	398.5	404.6	413.5	419.7	420.7
28.	431.2	431.0	413.5	404.1	398.5	404.6	413.5	419.7	420.7
29.	431.2	431.0	413.5	404.1	398.5	404.6	413.5	419.7	420.7
30.	431.2	431.0	413.5	404.1	398.5	404.6	413.5	419.7	420.7
31.	418.4	418.2	401.1	392.0	387.7	393.6	403.6	411.0	413.3
32.	418.4	418.2	401.1	392.0	387.7	393.6	403.6	411.0	413.3
33.	418.4	418.2	401.1	392.0	387.7	393.6	403.6	411.0	413.3
34.	422.0	421.8	404.3	394.9	390.3	395.4	404.3	410.5	411.5
35.	422.0	421.8	404.3	394.9	390.3	395.4	404.3	410.5	411.5
36.	422.0	421.8	404.3	394.9	390.3	395.4	404.3	410.5	411.5
37.	427.6	427.4	410.3	401.2	396.9	402.8	412.8	420.2	422.5
38.	418.4	418.2	401.1	392.0	387.7	393.6	403.6	411.0	413.3
39.	418.4	418.2	401.1	392.0	387.7	393.6	403.6	411.0	413.3
40.	440.4	440.2	422.7	413.3	408.7	413.8	422.7	428.9	429.9
41.	418.4	418.2	401.1	392.0	387.7	393.6	403.6	411.0	413.3
42.	418.4	418.2	401.1	392.0	387.7	393.6	403.6	411.0	413.3
43.	427.6	427.4	410.3	401.2	396.9	402.8	412.8	420.2	422.5
44.	427.6	427.4	410.3	401.2	396.9	402.8	412.8	420.2	422.5
45.	427.6	427.4	410.3	401.2	396.9	402.8	412.8	420.2	422.5
46.	427.6	427.4	410.3	401.2	396.9	402.8	412.8	420.2	422.5
47.	418.4	418.2	401.1	392.0	387.7	393.6	403.6	411.0	413.3
48.	418.4	418.2	401.1	392.0	387.7	393.6	403.6	411.0	413.3
49.	427.6	427.4	410.3	401.2	396.9	402.8	412.8	420.2	422.5
50.	427.6	427.4	410.3	401.2	396.9	402.8	412.8	420.2	422.5
51.	418.4	418.2	401.1	392.0	387.7	393.6	403.6	411.0	413.3
52.	427.6	427.4	410.3	401.2	396.9	402.8	412.8	420.2	422.5
53.	427.6	427.4	410.3	401.2	396.9	402.8	412.8	420.2	422.5
54.	427.6	427.4	410.3	401.2	396.9	402.8	412.8	420.2	422.5
55.	427.6	427.4	410.3	401.2	396.9	402.8	412.8	420.2	422.5
56.	431.2	431.0	413.5	404.1	398.5	404.6	413.5	419.7	420.7
57.	422.8	422.6	405.1	395.7	391.1	396.2	405.1	411.4	412.3
58.	422.8	422.6	405.1	395.7	391.1	396.2	405.1	411.4	412.3
59.	422.8	422.6	405.1	395.7	391.1	396.2	405.1	411.4	412.3
60.	422.8	422.6	405.1	395.7	391.1	396.2	405.1	411.4	412.3
61.	440.4	440.2	422.7	413.3	408.7	413.8	422.7	428.9	429.9
62.	440.4	440.2	422.7	413.3	408.7	413.8	422.7	428.9	429.9
63.	436.8	436.6	419.5	410.4	406.1	412.0	422.0	429.4	431.7

Coronene series

C24H12

1. coronene

C28H14

322.70 322.54 309.40 302.66 299.41 302.84 308.40 311.37 309.74

Table 12. Standard enthalpy of formation for polycyclic aromatic hydrocarbons in kJ/mol -- continued

T/k	298	300	500	700	1000	1500	2000	2500	3000
1. dibenzo(hi,qr)anthanthrene	372.1	371.9	357.0	349.3	345.9	350.7	358.3	362.9	362.4
2. benzo(a)coronene	372.1	371.9	357.0	349.3	345.9	350.7	358.3	362.9	362.4
3. phenanthro(1,10,9,8-opqra)perylene	372.1	371.9	357.0	349.3	345.9	350.7	358.3	362.9	362.4
4. naphtho(8,1,2-efg)anthanthrene	381.3	381.1	366.2	358.6	355.1	359.9	367.5	372.1	371.6
5. dibenzo(cd,hi)anthanthrene	372.1	371.9	357.0	349.3	345.9	350.7	358.3	362.9	362.4
6. dibenzo(cd,fg)anthanthrene	381.3	381.1	366.2	358.6	355.1	359.9	367.5	372.1	371.6
7. naphtho(2,1,8-hij)anthanthrene	390.5	390.3	375.4	367.8	364.3	369.1	376.7	381.3	380.8
9. dibenzo(cd,lm)anthanthrene									
C32H16									
1. tribenzo(e,hi,qr)anthanthrene	434.2	434.0	416.9	408.2	404.2	409.6	418.0	423.2	422.4
2. tribenzo(b,hi,qr)anthanthrene	421.4	421.2	404.6	396.0	392.4	398.6	408.1	414.5	415.0
3. dibenzo(a,f)coronene	421.4	421.2	404.6	396.0	392.4	398.6	408.1	414.5	415.0
4. dibenzo(a,g)coronene	416.6	416.4	399.4	390.6	386.6	392.0	400.5	405.6	404.8
5. dibenzo(a,d)coronene	443.4	443.2	426.1	417.4	413.4	418.8	427.2	432.4	431.6
6. naphtho(1,2-a)coronene	443.4	443.2	426.1	417.4	413.4	418.8	427.2	432.4	431.6
7. naphtho(8,1,2-abc)peropyrene	421.4	421.2	404.6	396.0	392.4	398.6	408.1	414.5	415.0
8. tribenzo(b,fg,qr)anthanthrene	421.4	421.2	404.6	396.0	392.4	398.6	408.1	414.5	415.0
9. tribenzo(a,cd,fg)anthanthrene	430.6	430.4	413.8	405.2	401.6	407.8	417.3	423.7	424.2
10. tribenzo(a,cd,hi)anthanthrene	430.6	430.4	413.8	405.2	401.6	407.8	417.3	423.7	424.2
11. benzo(a)naphtho(8,1,2-efg)anthanthrene	430.6	430.4	413.8	405.2	401.6	407.8	417.3	423.7	424.2
12. benzo(ghi)terylene	430.6	430.4	413.8	405.2	401.6	407.8	417.3	423.7	424.2
13. benzo(qr)naphtho(3,2,1-hi)anthanthrene	430.6	430.4	413.8	405.2	401.6	407.8	417.3	423.7	424.2
14. benzo(bc,lm)peropyrene	439.8	439.6	423.0	414.4	410.8	417.0	426.5	432.9	433.4
16. d'naphtho(1,8-ab:8',1',2',3'-fghi)perylene	443.4	443.2	426.1	417.4	413.4	418.8	427.2	432.4	431.6
17. naphtho(1,2,3,4-ijkl)peropyrene	443.4	443.2	426.1	417.4	413.4	418.8	427.2	432.4	431.6
18. dibenzo(ghi,lm)naphtho(1,8-ab)perylene	439.8	439.6	423.0	414.4	410.8	417.0	426.5	432.9	433.4
19. naphtho(2,1,8-bcd)peropyrene	439.8	439.6	423.0	414.4	410.8	417.0	426.5	432.9	433.4
20. benzo(a)naphtho(2,1,8-cde)anthanthrene	439.8	439.6	423.0	414.4	410.8	417.0	426.5	432.9	433.4
21. dibenzo(bc,qr)peropyrene	430.6	430.4	413.8	405.2	401.6	407.8	417.3	423.7	424.2
22. tribenzo(e,hi,op)anthanthrene	430.6	430.4	413.8	405.2	401.6	407.8	417.3	423.7	424.2
23. benzo(fg)naphtho(1,8-ab)anthanthrene	430.6	430.4	413.8	405.2	401.6	407.8	417.3	423.7	424.2
24. benzo(fg)naphtho(1,8-ab)anthanthrene	430.6	430.4	413.8	405.2	401.6	407.8	417.3	423.7	424.2
26. tribenzo(b,hi,op)anthanthrene	430.6	430.4	413.8	405.2	401.6	407.8	417.3	423.7	424.2
27. benzo(b)naphtho(8,1,2-nop)anthanthrene	443.4	443.2	426.1	417.4	413.4	418.8	427.2	432.4	431.6
28. benzo(hi)naphtho(1,2,3-qr)anthanthrene	439.8	439.6	423.0	414.4	410.8	417.0	426.5	432.9	433.4
29. naphtho(2,3-a)coronene	439.8	439.6	423.0	414.4	410.8	417.0	426.5	432.9	433.4
30. benzo(cd)naphtho(1,2,3-hi)anthanthrene	452.6	452.4	435.3	426.6	422.6	428.0	436.4	441.6	440.8
31. benzo(cd)naphtho(3,2,1-hi)anthanthrene	439.8	439.6	423.0	414.4	410.8	417.0	426.5	432.9	433.4
32. benzo(cd)naphtho(3,2,1-fg)anthanthrene	443.4	443.2	426.1	417.4	413.4	418.8	427.2	432.4	431.6
33. tribenzo(a,hi,op)anthanthrene	443.4	443.2	426.1	417.4	413.4	418.8	427.2	432.4	431.6
34. tribenzo(cd,fg,kl)anthanthrene	434.2	434.0	416.9	408.2	404.2	409.6	418.0	423.2	422.4
35. tribenzo(cd,hi,kl)anthanthrene	434.2	434.0	416.9	408.2	404.2	409.6	418.0	423.2	422.4
36. benzo(a)naphtho(2,1,8-lmn)anthanthrene	439.8	439.6	423.0	414.4	410.8	417.0	426.5	432.9	433.4
37. tribenzo(cd,fg,ij)anthanthrene	443.4	443.2	426.1	417.4	413.4	418.8	427.2	432.4	431.6
38. benzo(a)naphtho(8,1,2-nop)anthanthrene	443.4	443.2	426.1	417.4	413.4	418.8	427.2	432.4	431.6
39. benzo(a)naphtho(2,1,8-hij)anthanthrene	439.8	439.6	423.0	414.4	410.8	417.0	426.5	432.9	433.4
40. benzo(a)naphtho(2,1,8-hij)anthanthrene	439.8	439.6	423.0	414.4	410.8	417.0	426.5	432.9	433.4
42. benzo(b)naphtho(2,1,8-lmn)anthanthrene	439.8	439.6	423.0	414.4	410.8	417.0	426.5	432.9	433.4
43. benzo(lm)naphtho(1,8-ab)anthanthrene	439.8	439.6	423.0	414.4	410.8	417.0	426.5	432.9	433.4
47. benzo(a)naphtho(8,1,2-klm)anthanthrene	443.4	443.2	426.1	417.4	413.4	418.8	427.2	432.4	431.6
49. tribenzo(a,fg,op)anthanthrene	439.8	439.6	423.0	414.4	410.8	417.0	426.5	432.9	433.4
50. benzo(b)naphtho(8,1,2-bcd)anthanthrene	439.8	439.6	423.0	414.4	410.8	417.0	426.5	432.9	433.4
51. benzo(b)naphtho(2,1,8-hij)anthanthrene	443.4	443.2	426.1	417.4	413.4	418.8	427.2	432.4	431.6
52. benzo(b)naphtho(8,1,2-efg)anthanthrene	443.4	443.2	426.1	417.4	413.4	418.8	427.2	432.4	431.6
55. benzo(b)naphtho(8,1,2-klm)anthanthrene	449.0	448.8	432.2	423.7	420.0	426.2	435.7	442.1	442.6

Table 13. Standard Gibbs energy for polycyclic aromatic hydrocarbons in kJ/mol

T/K	298	300	503	700	1000	1500	2000	2500	3000
Pyrene series									
C16H10									
Pyrene									
	332.8	333.4	405.9	482.5	599.8	796.3	991.7	1187.3	1379.9
C20H12									
1. perylene	411.7	412.5	505.3	602.8	751.8	1000.7	1248.0	1495.3	1738.8
2. benzo(e)pyrene	410.0	410.8	502.4	598.8	746.0	992.1	1236.5	1480.9	1721.5
3. benzo(a)pyrene	417.5	418.3	508.3	603.9	749.4	992.7	1234.2	1475.7	1713.4
C24H14									
1. dibenzo(fg,op)naphthacene	490.6	491.6	604.7	723.1	903.7	1205.2	1504.4	1803.3	2097.7
2. naphtho(8,1,2-ghi)chrysene	500.7	501.6	606.3	717.5	886.7	1169.3	1450.0	1730.7	2007.1
3. benzo(pqr)picene	505.6	506.5	617.4	733.5	910.6	1206.3	1499.7	1792.9	2081.5
4. dibenzo(a,e)pyrene	496.4	497.3	608.2	724.3	901.4	1197.1	1490.5	1783.7	2072.3
5. benzo(b)perylene	507.3	508.3	620.3	737.5	916.4	1215.0	1511.3	1807.3	2098.8
6. zetrirrene	500.7	501.6	606.3	717.5	886.7	1169.3	1450.0	1730.7	2007.1
7. benzo(a)perylene	500.7	501.6	606.3	717.5	886.7	1169.3	1450.0	1730.7	2007.1
8. dibenzo(b,def)chrysene	507.3	508.3	620.3	737.5	916.4	1215.0	1511.3	1807.3	2098.8
9. dibenzo(def,p)chrysene	500.7	501.6	606.3	717.5	886.7	1169.3	1450.0	1730.7	2007.1
10. dibenzo(de,qr)naphthacene	507.3	508.3	620.3	737.5	916.4	1215.0	1511.3	1807.3	2098.8
11. dibenzo(c,mm)chrysene	509.9	510.8	616.1	726.8	895.9	1178.5	1459.2	1739.9	2016.3
12. benzo(rst)pentaphene	507.3	508.3	620.3	737.5	916.4	1215.0	1511.3	1807.3	2098.8
14. naphtho(2,1,8-qr)naphthacene	514.8	515.7	626.3	742.7	919.8	1215.5	1509.0	1802.1	2090.7
C28H16									
1. tritlenzo(f,ij,no)tetraphene	535.6	536.6	707.3	835.2	1029.7	1354.7	1677.3	2000.1	2318.0
2. tritlenzo(a,fg,op)tetracene	579.5	580.6	706.3	837.9	1038.7	1373.8	1706.3	2038.7	2366.0
4. naphtho(2,1,8-def)picene	538.7	539.9	715.5	847.1	1047.9	1383.0	1715.6	2047.9	2375.2
5. dibenzo(a,pqr)picene	537.9	539.1	724.7	856.3	1057.1	1392.2	1724.8	2057.1	2384.4
6. benzo(a)naphtho(8,1,2-fgh)tetracene	538.7	539.9	715.5	847.1	1047.9	1383.0	1715.6	2047.9	2375.2
7. naphtho(1,2,3,4-bqr)picene	534.4	535.6	716.3	853.8	1062.6	1410.8	1756.1	2100.9	2440.4
8. naphtho(1,2-b)perylene	534.4	535.6	716.3	853.8	1062.6	1410.8	1756.1	2100.9	2440.4
9. naphtho(2,1-a)perylene	538.7	539.9	715.5	847.1	1047.9	1383.0	1715.6	2047.9	2375.2
10. dibenzo(j,pqr)picene	538.7	539.9	715.5	847.1	1047.9	1383.0	1715.6	2047.9	2375.2
11. benzo(fg)naphtho(3,2,1-op)tetracene	579.5	580.6	706.3	837.9	1038.7	1373.8	1706.3	2038.7	2366.0
12. tritlenzo(f,m,pqr)tetraphene	538.7	539.9	715.5	847.1	1047.9	1383.0	1715.6	2047.9	2375.2
13. dibenzo(f,pqr)picene	538.7	539.9	715.5	847.1	1047.9	1383.0	1715.6	2047.9	2375.2
14. tritlenzo(a,hi,nn)tetracene	533.6	534.8	726.0	863.0	1071.8	1420.0	1765.3	2110.1	2449.6
15. naphtho(1,2-e)perylene	538.7	539.9	715.5	847.1	1047.9	1383.0	1715.6	2047.9	2375.2
16. dibenzo(a,e)perylene	538.7	539.9	715.5	847.1	1047.9	1383.0	1715.6	2047.9	2375.2
17. dibenzo(fg,qr)pentacene	538.7	539.9	715.5	847.1	1047.9	1383.0	1715.6	2047.9	2375.2
18. naphtho(8,1,2-cde)picene	538.7	539.9	715.5	847.1	1047.9	1383.0	1715.6	2047.9	2375.2
19. benzo(c)naphtho(8,1,2-ghi)tetraphene	537.9	539.1	724.7	856.3	1057.1	1392.2	1724.8	2057.1	2384.4
20. benzo(a)zetrirrene	538.7	539.9	715.5	847.1	1047.9	1383.0	1715.6	2047.9	2375.2
21. dibenzo(de,kl)pentaphene	538.7	539.9	715.5	847.1	1047.9	1383.0	1715.6	2047.9	2375.2
22. dibenzo(a,n)perylene	533.6	534.8	726.0	863.0	1071.8	1420.0	1765.3	2110.1	2449.6
23. dibenzo(de,ij)pentaphene	533.6	534.8	726.0	863.0	1071.8	1420.0	1765.3	2110.1	2449.6
24. benzo(f)zetrirrene	533.6	534.8	726.0	863.0	1071.8	1420.0	1765.3	2110.1	2449.6
25. dibenzo(a,j)pentaphene	533.6	534.8	726.0	863.0	1071.8	1420.0	1765.3	2110.1	2449.6
26. dibenzo(a,j)perylene	534.8	535.8	717.1	844.4	1038.9	1363.9	1686.5	2009.3	2327.2
27. dibenzo(h,rst)pentaphene	536.2	537.3	719.7	857.8	1058.3	1419.4	1767.6	2115.3	2457.7
28. dibenzo(b,tuv)picene	533.6	534.8	726.0	863.0	1071.8	1420.0	1765.3	2110.1	2449.6

Table 13. Standard Gibbs energy for polycyclic aromatic hydrocarbons in kJ/mol --- continued

T/K	298	300	500	700	1000	1500	2000	2500	3000
30.	588.7	589.9	715.5	847.1	1047.9	1383.0	1715.6	2047.9	2375.2
31.	586.2	587.3	719.7	857.8	1068.3	1419.4	1787.6	2115.3	2457.7
32.	593.0	594.1	714.2	840.4	1033.2	1355.2	1675.0	1994.9	2309.9
33.	588.7	589.9	715.5	847.1	1047.9	1383.0	1715.6	2047.9	2375.2
34.	590.5	591.6	718.4	851.1	1053.6	1391.6	1727.1	2062.3	2392.5
37.	593.6	594.8	726.0	863.0	1071.8	1420.0	1765.3	2110.1	2449.6
38.	586.2	587.3	719.7	857.8	1068.3	1419.4	1767.6	2115.3	2457.7
39.	597.9	599.1	724.7	856.3	1057.1	1392.2	1724.8	2057.1	2384.4
40.	597.9	599.1	724.7	856.3	1057.1	1392.2	1724.8	2057.1	2384.4
43.	597.9	599.1	724.7	856.3	1057.1	1392.2	1724.8	2057.1	2384.4
44.	602.9	604.0	735.2	872.2	1081.0	1429.2	1774.5	2119.3	2458.8
45.	597.9	599.1	724.7	856.3	1057.1	1392.2	1724.8	2057.1	2384.4
46.	607.2	608.3	733.9	865.5	1066.3	1401.4	1734.0	2066.3	2393.6
47.	597.9	599.1	724.7	856.3	1057.1	1392.2	1724.8	2057.1	2384.4
49.	593.6	594.8	726.0	863.0	1071.8	1420.0	1765.3	2110.1	2449.6
50.	593.6	594.8	726.0	863.0	1071.8	1420.0	1765.3	2110.1	2449.6
52.	602.9	604.0	735.2	872.2	1081.0	1429.2	1774.5	2119.3	2458.8
53.	593.6	594.8	726.0	863.0	1071.8	1420.0	1765.3	2110.1	2449.6
54.	593.6	594.8	726.0	863.0	1071.8	1420.0	1765.3	2110.1	2449.6
55.	604.6	605.7	738.1	876.2	1086.8	1437.8	1786.0	2133.7	2476.1
56.	597.9	599.1	724.7	856.3	1057.1	1392.2	1724.8	2057.1	2384.4
59.	594.8	595.8	717.1	844.4	1038.9	1363.9	1686.5	2009.3	2327.2
60.	602.9	604.0	735.2	872.2	1081.0	1429.2	1774.5	2119.3	2458.8
61.	602.9	604.0	735.2	872.2	1081.0	1429.2	1774.5	2119.3	2458.8
62.	597.9	599.1	724.7	856.3	1057.1	1392.2	1724.8	2057.1	2384.4
63.	604.6	605.7	738.1	876.2	1086.8	1437.8	1786.0	2133.7	2476.1
64.	602.9	604.0	735.2	872.2	1081.0	1429.2	1774.5	2119.3	2458.8
65.	602.9	604.0	735.2	872.2	1081.0	1429.2	1774.5	2119.3	2458.8
68.	612.1	613.2	744.4	881.4	1090.2	1438.4	1783.7	2128.5	2468.0
Naphthopyrene series									
C22H12									
1.	430.28	431.08	521.91	617.13	762.49	1005.20	1246.22	1487.39	1725.09
3.	439.48	440.28	531.11	626.34	771.69	1014.41	1255.43	1496.60	1734.29
C26H14									
1.	507.4	508.4	618.4	733.4	908.7	1201.0	1491.0	1781.0	2066.7
2.	509.2	510.1	621.3	737.5	914.5	1209.7	1502.6	1795.4	2084.0
3.	520.1	521.1	633.4	750.7	929.4	1227.5	1523.3	1819.0	2110.5
4.	516.7	517.6	627.6	742.6	917.9	1210.2	1500.3	1790.2	2075.9
5.	516.7	517.6	627.6	742.6	917.9	1210.2	1500.3	1790.2	2075.9
6.	516.7	517.6	627.6	742.6	917.9	1210.2	1500.3	1790.2	2075.9
7.	520.9	521.9	626.4	735.9	903.2	1182.4	1459.7	1737.2	2010.7
8.	525.9	526.8	636.9	751.8	927.1	1219.4	1509.5	1799.4	2085.1
10.	525.9	526.8	636.9	751.8	927.1	1219.4	1509.5	1799.4	2085.1
C30H16									
1.	590.6	591.7	716.6	847.1	1046.0	1377.7	1706.9	2036.0	2360.4
2.	607.6	608.6	730.2	857.6	1052.0	1376.4	1698.6	2021.0	2338.9
3.	599.8	600.9	725.8	856.3	1055.2	1386.9	1716.1	2045.2	2369.6
4.	599.8	600.9	725.8	856.3	1055.2	1386.9	1716.1	2045.2	2369.6
5.	599.0	600.1	732.8	871.0	1081.4	1432.0	1779.7	2127.0	2469.4
6.	588.0	589.2	720.7	857.8	1066.4	1414.1	1758.9	2103.4	2442.9
7.	599.8	600.9	725.8	856.3	1055.2	1386.9	1716.1	2045.2	2369.6
8.	599.8	600.9	725.8	856.3	1055.2	1386.9	1716.1	2045.2	2369.6

Table 13. Standard Gibbs energy for polycyclic aromatic hydrocarbons in kJ/mol -- continued

T/K	298	303	500	700	1000	1500	2000	2500
9.	599.8	600.9	725.8	856.3	1055.2	1386.9	1716.1	2045.2
10.	604.7	605.3	736.3	872.2	1079.1	1423.9	1765.8	2107.4
11.	609.0	610.1	735.0	865.5	1064.4	1396.1	1725.3	2054.2
12.	599.8	600.3	729.8	856.3	1055.2	1386.9	1716.1	2045.2
13.	597.3	598.4	725.9	857.0	1075.6	1423.3	1768.1	2112.6
14.	599.8	600.3	725.8	856.3	1055.2	1386.9	1716.1	2045.2
15.	609.0	610.1	735.0	865.5	1064.4	1396.1	1725.3	2054.2
16.	609.0	610.1	735.0	865.5	1064.4	1396.1	1725.3	2054.2
17.	595.5	596.7	727.1	863.0	1069.9	1414.7	1756.6	2098.2
18.	595.5	596.7	727.1	863.0	1069.9	1414.7	1756.6	2098.2
19.	599.8	600.9	725.8	856.3	1055.2	1386.9	1716.1	2045.2
20.	599.8	600.9	725.8	856.3	1055.2	1386.9	1716.1	2045.2
21.	615.7	616.3	748.4	885.4	1094.0	1441.7	1786.5	2131.0
22.	604.7	605.3	736.3	872.2	1079.1	1423.9	1765.8	2107.4
23.	609.0	610.1	735.0	865.5	1064.4	1396.1	1725.3	2054.2
24.	604.7	605.3	736.3	872.2	1079.1	1423.9	1765.8	2107.4
25.	604.7	605.3	736.3	872.2	1079.1	1423.9	1765.8	2107.4
26.	613.9	615.1	745.5	881.4	1088.3	1433.1	1775.0	2116.6
27.	609.0	610.1	735.0	865.5	1064.4	1396.1	1725.3	2054.2
28.	609.0	610.1	735.0	865.5	1064.4	1396.1	1725.3	2054.2
29.	609.0	610.1	735.0	865.5	1064.4	1396.1	1725.3	2054.2
30.	609.0	610.1	735.0	865.5	1064.4	1396.1	1725.3	2054.2
31.	609.0	610.1	735.0	865.5	1064.4	1396.1	1725.3	2054.2
32.	604.7	605.3	736.3	872.2	1079.1	1423.9	1765.8	2107.4
33.	606.5	607.5	739.2	876.2	1084.8	1432.5	1777.3	2121.8
34.	606.5	607.5	739.2	876.2	1084.8	1432.5	1777.3	2121.8
35.	599.8	600.3	725.8	856.3	1055.2	1386.9	1716.1	2045.2
36.	599.8	600.3	725.8	856.3	1055.2	1386.9	1716.1	2045.2
37.	609.0	610.1	735.0	865.5	1064.4	1396.1	1725.3	2054.2
38.	613.9	615.1	745.5	881.4	1088.3	1433.1	1775.0	2116.6
39.	604.7	605.3	736.3	872.2	1079.1	1423.9	1765.8	2107.4
40.	606.5	607.5	739.2	876.2	1084.8	1432.5	1777.3	2121.8
41.	618.2	619.3	744.2	874.7	1073.6	1405.3	1734.5	2063.6
42.	604.7	605.3	736.3	872.2	1079.1	1423.9	1765.8	2107.4
43.	613.9	615.1	745.5	881.4	1088.3	1433.1	1775.0	2116.6
44.	615.7	616.3	748.4	885.4	1094.0	1441.7	1786.5	2131.0
45.	613.9	615.1	745.5	881.4	1088.3	1433.1	1775.0	2116.6
46.	613.9	615.1	745.5	881.4	1088.3	1433.1	1775.0	2116.6
47.	613.9	615.1	745.5	881.4	1088.3	1433.1	1775.0	2116.6
48.	604.7	605.3	736.3	872.2	1079.1	1423.9	1765.8	2107.4
49.	613.9	615.1	745.5	881.4	1088.3	1433.1	1775.0	2116.6
50.	613.9	615.1	745.5	881.4	1088.3	1433.1	1775.0	2116.6
51.	604.7	605.3	736.3	872.2	1079.1	1423.9	1765.8	2107.4
52.	613.9	615.1	745.5	881.4	1088.3	1433.1	1775.0	2116.6
53.	613.9	615.1	745.5	881.4	1088.3	1433.1	1775.0	2116.6
54.	615.7	616.3	748.4	885.4	1094.0	1441.7	1786.5	2131.0
55.	609.0	610.1	735.0	865.5	1064.4	1396.1	1725.3	2054.2
56.	600.7	601.8	726.6	857.1	1056.0	1387.7	1716.9	2046.0
57.	600.7	601.8	726.6	857.1	1056.0	1387.7	1716.9	2046.0
58.	600.7	601.8	726.6	857.1	1056.0	1387.7	1716.9	2046.0
59.	600.7	601.8	726.6	857.1	1056.0	1387.7	1716.9	2046.0
60.	600.7	601.8	726.6	857.1	1056.0	1387.7	1716.9	2046.0
61.	600.7	601.8	726.6	857.1	1056.0	1387.7	1716.9	2046.0
62.	600.7	601.8	726.6	857.1	1056.0	1387.7	1716.9	2046.0
63.	620.0	621.1	747.1	878.7	1079.3	1413.9	1746.0	2105.3
64.	620.0	621.1	747.1	878.7	1079.3	1413.9	1746.0	2105.3
65.	623.2	624.3	754.7	890.6	1097.5	1442.3	1784.2	2125.8
66.	623.2	624.3	754.7	890.6	1097.5	1442.3	1784.2	2125.8
67.	623.2	624.3	754.7	890.6	1097.5	1442.3	1784.2	2125.8

## Coronene series

C24H12

1. coronene

C28H14

452.29 453.09 544.25 639.55 784.75 1026.94 1267.46 1508.31 1745.97

Table 13. Standard Gibbs energy for polycyclic aromatic hydrocarbons in kJ/mol -- continued

T/K	293	300	500	700	1000	1500	2000	2500	3000
1.	529.5	530.4	640.8	755.8	931.0	1232.8	1512.3	1801.9	2087.6
2.	529.5	530.4	640.8	755.8	931.0	1232.8	1512.3	1801.9	2087.6
3.	531.2	532.1	637.9	759.9	936.7	1231.4	1500.8	1816.3	2104.9
4.	527.7	528.7	637.9	759.9	936.7	1231.4	1500.8	1816.3	2104.9
5.	536.9	537.4	641.1	761.0	934.4	1223.3	1510.5	1796.7	2079.5
6.	530.5	530.4	640.8	755.8	931.0	1222.8	1512.3	1801.9	2087.6
7.	538.7	539.6	650.0	765.1	940.2	1232.0	1521.5	1811.1	2096.8
9.	547.9	548.8	659.2	774.3	949.4	1241.2	1530.7	1820.3	2106.0
C32H16									
1.	610.9	612.0	736.0	865.4	1062.4	1390.8	1716.6	2042.5	2363.9
2.	606.6	607.8	737.3	872.1	1077.2	1418.6	1757.1	2095.5	2429.2
3.	610.1	611.2	743.1	880.2	1088.7	1435.9	1780.2	2124.3	2463.8
4.	608.3	609.5	740.2	876.2	1082.9	1427.2	1768.6	2109.9	2446.5
5.	595.1	596.2	721.3	851.9	1050.6	1381.9	1710.5	2039.3	2363.7
6.	620.1	621.2	745.2	874.6	1071.6	1400.0	1725.8	2051.7	2373.2
7.	620.1	621.2	745.2	874.6	1071.6	1400.0	1725.8	2051.7	2373.2
8.	606.6	607.8	737.3	872.1	1077.2	1418.6	1757.1	2095.5	2429.2
9.	606.6	607.8	737.3	872.1	1077.2	1418.6	1757.1	2095.5	2429.2
10.	615.8	617.0	746.5	881.4	1086.4	1427.8	1766.3	2104.7	2438.4
11.	615.8	617.0	746.5	881.4	1086.4	1427.8	1766.3	2104.7	2438.4
12.	615.8	617.0	746.5	881.4	1086.4	1427.8	1766.3	2104.7	2438.4
13.	620.1	621.2	745.2	874.6	1071.6	1400.0	1725.8	2051.7	2373.2
14.	620.1	621.2	745.2	874.6	1071.6	1400.0	1725.8	2051.7	2373.2
15.	618.7	619.8	744.9	874.6	1071.6	1400.0	1725.8	2051.7	2373.2
16.	626.2	627.3	755.7	890.6	1095.6	1437.0	1775.5	2113.9	2447.6
17.	626.2	627.3	755.7	890.6	1095.6	1437.0	1775.5	2113.9	2447.6
18.	626.2	627.3	755.7	890.6	1095.6	1437.0	1775.5	2113.9	2447.6
19.	626.2	627.3	755.7	890.6	1095.6	1437.0	1775.5	2113.9	2447.6
20.	615.8	617.0	746.5	881.4	1086.4	1427.8	1766.3	2104.7	2438.4
21.	615.8	617.0	746.5	881.4	1086.4	1427.8	1766.3	2104.7	2438.4
22.	615.8	617.0	746.5	881.4	1086.4	1427.8	1766.3	2104.7	2438.4
23.	615.8	617.0	746.5	881.4	1086.4	1427.8	1766.3	2104.7	2438.4
24.	615.8	617.0	746.5	881.4	1086.4	1427.8	1766.3	2104.7	2438.4
25.	615.8	617.0	746.5	881.4	1086.4	1427.8	1766.3	2104.7	2438.4
26.	615.8	617.0	746.5	881.4	1086.4	1427.8	1766.3	2104.7	2438.4
27.	620.1	621.2	745.2	874.6	1071.6	1400.0	1725.8	2051.7	2373.2
28.	620.1	621.2	745.2	874.6	1071.6	1400.0	1725.8	2051.7	2373.2
29.	626.8	627.9	758.6	894.6	1101.3	1445.6	1787.0	2128.3	2464.9
30.	626.8	627.9	758.6	894.6	1101.3	1445.6	1787.0	2128.3	2464.9
31.	620.1	621.2	745.2	874.6	1071.6	1400.0	1725.8	2051.7	2373.2
32.	620.1	621.2	745.2	874.6	1071.6	1400.0	1725.8	2051.7	2373.2
33.	620.1	621.2	745.2	874.6	1071.6	1400.0	1725.8	2051.7	2373.2
34.	610.9	612.0	736.0	865.4	1062.4	1390.8	1716.6	2042.5	2363.9
35.	610.9	612.0	736.0	865.4	1062.4	1390.8	1716.6	2042.5	2363.9
36.	620.1	621.2	745.2	874.6	1071.6	1400.0	1725.8	2051.7	2373.2
37.	620.1	621.2	745.2	874.6	1071.6	1400.0	1725.8	2051.7	2373.2
38.	620.1	621.2	745.2	874.6	1071.6	1400.0	1725.8	2051.7	2373.2
39.	625.0	626.2	755.7	890.6	1095.6	1437.0	1775.5	2113.9	2447.6
40.	625.0	626.2	755.7	890.6	1095.6	1437.0	1775.5	2113.9	2447.6
41.	625.0	626.2	755.7	890.6	1095.6	1437.0	1775.5	2113.9	2447.6
42.	625.0	626.2	755.7	890.6	1095.6	1437.0	1775.5	2113.9	2447.6
43.	625.0	626.2	755.7	890.6	1095.6	1437.0	1775.5	2113.9	2447.6
44.	625.0	626.2	755.7	890.6	1095.6	1437.0	1775.5	2113.9	2447.6
45.	625.0	626.2	755.7	890.6	1095.6	1437.0	1775.5	2113.9	2447.6
46.	625.0	626.2	755.7	890.6	1095.6	1437.0	1775.5	2113.9	2447.6
47.	625.0	626.2	755.7	890.6	1095.6	1437.0	1775.5	2113.9	2447.6
48.	625.0	626.2	755.7	890.6	1095.6	1437.0	1775.5	2113.9	2447.6
49.	625.0	626.2	755.7	890.6	1095.6	1437.0	1775.5	2113.9	2447.6
50.	625.0	626.2	755.7	890.6	1095.6	1437.0	1775.5	2113.9	2447.6
51.	625.0	626.2	755.7	890.6	1095.6	1437.0	1775.5	2113.9	2447.6
52.	620.1	621.2	745.2	874.6	1071.6	1400.0	1725.8	2051.7	2373.2
53.	634.2	635.4	764.9	899.8	1104.8	1446.2	1784.7	2123.1	2456.8

## 8. Acknowledgments

The calculations in this paper were made using a terminal connected to the IBM 370/3033N in the MIT Computer Center. Programs were written in APL and tables were printed on a Xerox 8700 Printer in the Computer Center. We are indebted to Professor J. R. Dias for many helpful discussions. This research was supported by a grant from the National Institute of Standards and Technology.

## 9. Nomenclature

$C_{\rho i}^{\circ}$	= standard heat capacity at constant pressure of isomer $i$ , $\text{J K}^{-1} \text{mol}^{-1}$
$C_{\rho}^{\circ}(\text{I})$	= standard heat capacity at constant pressure of isomer group I, $\text{J K}^{-1} \text{mol}^{-1}$
$\Delta_f G_i^{\circ}$	= standard Gibbs energy of formation of isomer $i$ , $\text{kJ mol}^{-1}$
$\Delta_f G^{\circ}(\text{I})$	= standard Gibbs energy of formation of isomer group I, $\text{kJ mol}^{-1}$
$H^{\circ}(\text{I}, T) - H^{\circ}(\text{I}, 298.15 \text{ K})$	= standard enthalpy for isomer groups relative to isomer groups at 298.15 K $\text{kJ mol}^{-1}$
$H^{\circ}(\text{I}, T) - H^{\circ}(\text{I}, 298.15 \text{ K}) + \Delta_f H^{\circ}(\text{I}, 298.15 \text{ K})$	= standard enthalpy for isomer groups relative to elements at 298.15 K, $\text{kJ mol}^{-1}$
$\Delta_f H_i^{\circ}$	= standard enthalpy of formation of isomer $i$ , $\text{kJ mol}^{-1}$
$\Delta_f H^{\circ}(\text{I})$	= standard enthalpy of formation of isomer group, I, $\text{kJ mol}^{-1}$
$n_{\text{C}}$	= number of carbon atoms in a molecule
$n_{\text{H}}$	= number of hydrogen atoms in a molecule
$N_i$	= number of isomers in an isomer group
$r_i$	= equilibrium mole fraction of species $i$ in an isomer group
$S_i^{\circ}$	= standard entropy of isomer $i$ , $\text{J K}^{-1} \text{mol}^{-1}$
$S^{\circ}(\text{I})$	= standard entropy of isomer group I, $\text{J K}^{-1} \text{mol}^{-1}$
TSN	= total symmetry number
$y_i$	= mole fraction of isomer $i$ within the isomer group
$y_{\text{I}}$	= mole fraction of isomer group I in a mixture

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1. Appendix Table A.1 Group Additivity Values of Stein and Fahr for Benzenoid Aromatic Hydrocarbons

Group	$\frac{\Delta H^\circ_{298}}{\text{kJ mol}^{-1}}$	$\frac{S^\circ_{298}}{\text{JK}^{-1}\text{mol}^{-1}}$	$C_p^\circ/\text{J K}^{-1}\text{mol}^{-1}$											
			300	400	500	600	800	1000	1250	1500	2000	2500	3000	
$C_B-(H)$	13.8	48.2	13.8	18.8	23.1	26.6	31.6	35.1	38.2	40.3	42.8	44.2	45	
$C_{FR}-(C_{FR})(C_B)2$	20.1	-20.9	12.6	15.4	17.5	19.3	22	23.7	25	25.8	26.6	27	27.2	
$C_{FR}-(C_{FR})2(C_B)$	15.5	-20.9	12.6	15.4	17.5	19.3	22	23.7	25	25.8	26.6	27	27.2	
$C_{FR}-(C_{FR})3$	6.1	7.6	8.4	13	16.3	18.5	21.1	22.4	23.2	23.7	24.2	24.5	24.6	