

Standard Chemical Thermodynamic Properties of Alkyl-naphthalene Isomer Groups

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The chemical thermodynamic properties of alkyl-naphthalene isomer groups for $C_{10}H_8$ and $C_{11}H_{10}$ in the ideal gas phase have been calculated from 298.15 to 1000 K from tables of Stull, Westrum, and Sinke. In the absence of literature data on all isomers of higher isomer groups, the properties of isomers of $C_{12}H_{12}$ to $C_{14}H_{16}$ have been calculated using Benson group values. A new Benson group value for the 1,8-dimethyl steric hindrance has been calculated from recent experimental data. The increments in isomer group properties per carbon atom have been calculated to show the extent to which thermodynamic properties of higher isomer groups may be obtained by linear extrapolation. Equilibrium mole fractions within isomer groups have been calculated for the ideal gas state from 298.15 to 1000 K. Values of C_p° , S° , $\Delta_f H^\circ$, and $\Delta_f G^\circ$ are given for all species from $C_{10}H_8$ to $C_{14}H_{16}$ with energy units of joules for a standard state pressure of 1 bar.

Key words: alkyl-naphthalenes; Benson method; enthalpy of formation; entropy; Gibbs energy of formation; heat capacity; isomer group thermodynamic properties; isomer mole fractions; thermodynamic properties.

Contents

1. Introduction.....	821	3. Standard heat capacity at constant pressure for alkyl-naphthalene isomer groups in J/K mol ...	824
2. Standard Thermodynamic Properties of Alkyl-naphthalene Isomer Groups	822	4. Standard entropy for alkyl-naphthalene isomer groups in J/K mol.....	824
3. Calculations of Standard Thermodynamic Properties of Alkyl-naphthalenes Using the Benson Method	822	5. Standard enthalpy of formation for alkyl-naphthalene isomer groups in kJ/mol	824
4. Tables of Standard Thermodynamic Properties of Alkyl-naphthalene Isomer Groups	823	6. Standard Gibbs energy of formation for alkyl-naphthalene isomer groups in kJ/mol	824
5. Equilibrium Mole Fractions Within Alkyl-naphthalene Isomer Groups	825	7. Standard enthalpy for alkyl-naphthalene isomer groups relative to isomer groups at 298.15 K in kJ/mol	825
6. Standard Thermodynamic Properties of Individual Alkyl-naphthalene Species	828	8. Standard enthalpy for alkyl-naphthalene isomer groups relative to the elements at 298.15 K in kJ/mol.....	825
7. Discussion	837	9. Equilibrium mole fractions within alkyl-naphthalene isomer groups	826
8. Nomenclature	837	10. Standard heat capacity at constant pressure for alkyl-naphthalenes in J/K mol	829
9. Acknowledgments.....	837	11. Standard entropy for alkyl-naphthalenes in J/K mol	831
10. References	837	12. Standard enthalpy of formation for alkyl-naphthalenes in kJ/mol.....	833
		13. Standard Gibbs energy of formation for alkyl-naphthalenes in kJ/mol.....	835

List of Tables

1. Numbers of isomers of alkyl-naphthalenes	822
2. Root-mean-square deviations between alkyl-naphthalene thermodynamic properties from Stull, Westrum, and Sinke and from the Benson method.....	823

1. Introduction

Since the number of isomers of alkyl-naphthalenes increases rapidly with carbon number, it is convenient to use

isomer group thermodynamic properties for cases in which all isomers in the isomer group are in equilibrium. The tables presented here are of the same type as those published for the alkanes,¹ alkylbenzenes,² and alkenes.³

Milligan, Becker, and Pitzer⁴ calculated the standard thermodynamic properties of the alkyl-naphthalenes in the ideal gas state through $C_{12}H_{12}$, except for 1,8-dimethyl-naphthalene, using the method of increments. Some further

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small changes were made by Stull, Westrum, and Sinke⁵ in publishing their book in 1969. We have added tables to include all isomers through C₁₄H₁₆ using the Benson method.^{6,7} Stein, Golden, and Benson⁸ have extended this method to the prediction of standard gas-phase thermodynamic properties for polycyclic aromatic hydrocarbons consisting of six-membered rings. They showed that the group additivity method reproduced the measured $\Delta_f H^\circ(298.15 \text{ K})$ values for substituted naphthalenes within 5 kJ mol⁻¹.

2. Standard Thermodynamic Properties of Alkyl-naphthalene Isomer Groups

When isomers are in equilibrium, the standard Gibbs energy of formation $\Delta_f G^\circ(\text{I})$ of the isomer group is defined by^{9,10}

$$\Delta_f G^\circ(\text{I}) = -RT \ln \left[\sum_{i=1}^{N_I} \exp(-\Delta_f G_i^\circ/RT) \right], \quad (1)$$

where $\Delta_f G_i^\circ$ is the standard Gibbs energy of formation of an individual isomer and N_I is the number of isomers in the group, including stereoisomers. The equilibrium mole fractions r_i of various isomers in a group can be calculated using

$$r_i = y_i/y_1 = \exp\{[\Delta_f G^\circ(\text{I}) - \Delta_f G_i^\circ]/RT\}, \quad (2)$$

where y_1 is the sum of the mole fractions of the individual isomers. The corresponding equations for the other standard thermodynamic properties $C_p^\circ(\text{I})$, $S^\circ(\text{I})$, and $\Delta_f H^\circ(\text{I})$ can be derived by differentiating Eq. (1) with respect to temperature.¹¹ When standard Gibbs energies of formation of isomer groups are used to calculate equilibrium constants for reactions of ideal gases the equilibrium expression is written in terms of equilibrium mole fractions of isomer groups.

For the alkyl-naphthalenes, the standard chemical thermodynamic properties for an isomer group are interrelated by

$$\Delta_f G^\circ(\text{I}) = \Delta_f H^\circ(\text{I}) - T[S^\circ(\text{I}) - nS_{\text{graphite}}^\circ - (n-6)S_{\text{H}_2(\text{g})}^\circ], \quad (3)$$

where n is the number of carbon atoms.

To calculate the chemical thermodynamic properties for an isomer group, a term must be included for each molecular species, including stereoisomers. The numbers of chiral centers and isomers of the alkyl-naphthalenes are shown in Table 1. Rather than having a line in a thermodynamic table for each stereoisomer, this article follows the standard practice of giving properties of racemates. The $R \ln 2$ is added to the calculated standard entropy and $RT \ln 2$ is subtracted from the standard Gibbs energy of formation of one of the

chiral forms at each temperature. Table 1 also summarizes the numbers of lines in the tables in this article.

3. Calculations of Standard Thermodynamic Properties of Alkyl-naphthalenes Using the Benson Method

Since the literature values are incomplete on the C₁₂H₁₂ alkyl-naphthalenes, the isomer group properties for C₁₂H₁₂ have been calculated using the Benson method. Since the increments per carbon atom in isomer group properties are still changing at C₁₂H₁₂, this method has also been used to calculate properties for all isomers of C₁₃H₁₄ and C₁₄H₁₆.

In order to make these calculations the structure of each alkyl-naphthalene species was divided into C(H)₃(C), C(H)₂(C)₂, C(H)(C)₃, C(C)₄, C(C_B)(C)(H)₂, C(C_B)(C)₂(H), C(C_B)(C)₃, C_B(H), C_B(C), and C_{BF}(C_B)₂(C_{BF}); the group contributions from Ref. 7 were used. In addition the total symmetry number (TSN), number of optical isomers (OPT), and *ortho* corrections were identified. *Gauche* corrections were made for *sec*- and *isobutyl*-naphthalene. Naphthalene has a symmetry number of 4; 2,3,6,7-tetramethylnaphthalene has a symmetry number of 4 × 3⁴; and 1,5-dimethylnaphthalene has a symmetry number of 2 × 3². The report by Davies, Syverud, and Steiner¹² was very helpful in calculating symmetry numbers.

We made the first estimates of the thermodynamic properties of the various species of C₁₂H₁₂, C₁₃H₁₄, and C₁₄H₁₆ using Benson group values published in 1976.⁷ Gammon pointed out to the authors that there is steric hindrance in 1,8-dimethylnaphthalene that is not provided for in the 1976 Benson values. The enthalpies of combustion of 1,8-dimethylnaphthalene, 2,3-dimethylnaphthalene, 2,6-dimethylnaphthalene, and 2,7-dimethylnaphthalene were measured by Good,¹³ who pointed out that the enthalpy of formation of the crystalline 1,8-dimethylnaphthalene is about 7.5 kcal mol⁻¹ more positive than those for 2,6-dimethylnaphthalene and 2,7-dimethylnaphthalene. The enthalpies of vaporization of these substances were measured by Osborn and Douslin,¹⁴ and their heat capacities and enthalpies of transition were measured by Finke, Messerly, Lee, Osborn, and Douslin.¹⁵ To remedy this deficiency in our estimates, an additional Benson group was introduced to provide for the 1,8 steric hindrance. The group value for $\Delta_f H_{298}^\circ$ for this group was taken to be 5.88 kcal mol⁻¹ because this is the difference between the extrapolated value for the 1,8-dimethylnaphthalene and the prediction by the Benson method without this correction. The Benson method gives the correct values for $\Delta_f H^\circ$ for 2,6-dimethylnaphthalene and 2,7-dimethylnaphthalene within their uncertainties. The experimental entropies of the 2,6- and 2,7-dimethylnaphthalenes from 340 to 410 K agree within experimental error and extrapolate linearly to the S° estimated by the Benson method. The experimental entropies of the 1,8-dimethylnaphthalene are definitely lower and extrapolate linearly to an S° value about 9.3 J K⁻¹ mol⁻¹ lower than that estimated by the Benson method. Therefore the entropy assigned to the 1,8 repulsion was taken to be -2.22 cal K⁻¹ mol⁻¹. The experimental data do not indicate any heat ca-

Table 1. Numbers of isomers of alkyl-naphthalenes

	Number of lines	One Chiral Center	Total isomers
C10H8	1	0	1
C11H10	2	0	2
C12H12	12	0	12
C13H14	32	0	32
C14H16	110	2	112

Table 2. Root mean square deviations between alkylnaphthalene thermodynamic properties from Stull, Westrum, and Sinke and from the Benson method

T/K	298	300	400	500	600	700	800	900	1000
Standard heat capacity at constant pressure in J/K mol									
C10H8	1.66	1.63	.80	1.70	.57	1.24	2.32	1.67	1.51
C11H10	2.36	2.40	5.66	7.05	5.91	4.01	2.56	2.47	4.99
C12H12	4.31	4.36	7.91	9.71	8.67	6.80	5.25	4.68	6.77
C13H14	4.59	4.63	6.98	7.63	6.22	3.96	6.35	2.81	5.56
C14H16	2.93	2.99	5.97	7.23	5.59	3.28	1.52	1.87	5.69
Standard entropy in J/K mol									
C10H8	3.07	3.08	2.97	3.29	3.54	3.48	3.19	2.92	2.90
C11H10	3.31	3.28	2.20	1.15	1.22	1.74	2.06	2.27	2.62
C12H12	8.13	8.09	6.56	4.91	3.72	3.07	2.76	2.64	2.73
C13H14	3.42	3.41	2.87	3.31	4.06	4.62	4.45	4.36	4.70
C14H16	1.77	1.74	1.29	2.01	3.00	3.61	3.86	3.99	4.32
Standard enthalpy of formation in kJ/mol									
C10H8	.33	.35	.18	.29	.36	.35	.11	.13	.07
C11H10	.80	.81	.57	.50	.97	1.47	1.73	1.94	2.36
C12H12	1.71	1.69	1.39	1.02	1.28	1.90	2.39	2.82	3.40
C13H14	1.85	1.85	1.43	.81	.49	.81	.65	.69	1.06
C14H16	2.95	2.96	2.66	2.06	1.55	1.15	1.10	1.07	.92
Standard Gibbs energy of formation in kJ/mol									
C10H8	.71	.70	1.04	1.30	1.72	2.00	2.37	2.62	2.84
C11H10	.78	.76	.93	1.10	1.16	1.23	1.26	1.32	1.40
C12H12	1.66	1.67	2.22	2.73	3.12	3.46	3.66	3.88	4.06
C13H14	1.76	1.77	1.90	2.12	2.56	2.89	3.40	3.74	4.11
C14H16	2.83	2.84	2.84	2.96	3.27	3.50	3.97	4.24	4.54

capacity corrections for the 1,8 interaction. In the revised calculations given here these values were also applied to 4,5 methyl groups in trimethyl and tetramethyl naphthalenes. In the absence of any experimental data on 1,8 methyl-ethyl or other interactions, this group value was also applied whenever there were 1,8 or 4,5 alkyl groups in C₁₃H₁₄ and C₁₄H₁₆.

The matrix of numbers of contributions was matrix multiplied by a matrix of the Benson values to obtain for each isomer the sum of the contributions to $\Delta_f H^\circ_{298}$, S°_{int298} , C°_{P300} , C°_{P400} , C°_{P500} , C°_{P600} , C°_{P800} , and C°_{P1000} . In further steps in the calculation the heat capacity values were fit to the equation

$$C_p^\circ = \alpha + \beta T + \gamma T^2, \quad (4)$$

and the values of α , β , and γ were used to calculate C_p° , S° , and $\Delta_f H^\circ$ from 298.15 to 1000 K,

$$S^\circ = S^\circ_0 + \alpha \ln T + \beta T + (\gamma/2)T^2 - R \ln(\text{TSN/OPT}), \quad (5)$$

$$\begin{aligned} \Delta_f H^\circ = & \Delta_f H^\circ_0 + \alpha T + (\beta/2)T^2 + (\gamma/3)T^3 \\ & - n(H^\circ - H^\circ_{298})_{\text{graph}} \\ & - (n-6)(H^\circ - H^\circ_{298})_{\text{H}_2}. \end{aligned} \quad (6)$$

The values of $\Delta_f G^\circ$ at various temperatures were then calculated using Eq. (3).

Table 2 shows how well the chemical thermodynamic properties calculated using the Benson method agree with the values from Stull, Westrum, and Sinke⁷ for C₁₀H₈, C₁₁H₁₀, 11 of the 12 isomers of C₁₂H₁₂, five of the 32 isomers of C₁₃H₁₄ and two of the 112 isomers of C₁₄H₁₆. The differences between the literature and estimated values at each temperature were squared, divided by the number of pairs of values, and the square root was taken. For C₁₀H₈ this yields the magnitude of the deviations, and for the higher isomer

groups it yields the root-mean-square deviations as a function of temperature.

4. Tables of Standard Thermodynamic Properties of Alkylnaphthalene Isomer Groups

Since the International Union of Pure and Applied Chemistry has recently recommended that thermodynamic data be given in SI units for a standard state pressure of 1 bar (10⁵ Pa), this has been done for the tables in this article. The change in standard state pressure from 1 atm to 1 bar does not affect C_p° and $\Delta_f H^\circ$, but the standard entropy of an ideal gas is increased by $R \ln(1.01325) = 0.109 \text{ J K}^{-1} \text{ mol}^{-1}$ at any temperature and the standard Gibbs energy of formation is reduced by $[RT \ln(1.01325)]\delta$, where δ is the net increase ($7 - n$) in moles of gas in the formation reaction.¹⁶

In Tables 3-8, the Stull, Westrum, and Sinke tables have been used to calculate the isomer group properties for C₁₀H₈ and C₁₁H₁₀, and the Benson method has been used to calculate the isomer group properties for C₁₂H₁₂ to C₁₄H₁₆. For each property the increments in going from one carbon number to the next are provided. Thus, of the four increments per carbon atom, the last two have been calculated using the Benson alone. These increments provide a basis for a linear extrapolation of standard thermodynamic properties of alkane isomer groups to higher carbon numbers.

Table 7 gives $H^\circ(I,T) - H^\circ(I,298.15 \text{ K})$, the standard enthalpy for an isomer group relative to the isomer group at 298.15 K. Table 8 gives values for $H^\circ(I,T) - H^\circ(I,298.15 \text{ K}) + \Delta_f H^\circ(I,298.15 \text{ K})$, the standard enthalpy for the isomer group relative to the elements at 298.15 K. This quantity allows the direct calculation of heat effects when the reactants and products are at different temperatures.

Table 3. Standard heat capacity at constant pressure for alkylnaphthalene isomer groups in J/K mol

T/K	C10H8	C11H10	C12H12	C13H14	C14H16
298.15	132.55	159.87	181.5	208.7	244.7
300.00	133.43	160.85	182.6	209.8	246.1
400.00	179.20	211.78	235.8	269.9	313.8
500.00	218.11	255.38	283.1	323.1	370.2
600.00	249.66	290.88	324.4	369.2	417.9
700.00	275.18	320.06	359.7	408.3	458.4
800.00	296.10	344.08	388.9	440.6	492.1
900.00	313.42	363.97	412.1	466.1	518.9
1000.00	327.94	380.87	429.1	484.9	538.7

Table 3a. Increments per carbon atom

T/K	C11-C10	C12-C11	C13-C12	C14-C13
298.15	27.32	21.7	27.1	36.1
300.00	27.42	21.7	27.3	36.3
400.00	32.57	24.0	34.1	43.9
500.00	37.27	27.7	40.0	47.1
600.00	41.22	33.5	44.8	48.7
700.00	44.87	39.6	48.7	50.1
800.00	47.98	44.8	51.7	51.4
900.00	50.55	48.1	54.1	52.7
1000.00	52.93	48.3	55.8	53.8

Table 4. Standard entropy for alkylnaphthalene isomer groups in J/K mol

T/K	C10H8	C11H10	C12H12	C13H14	C14H16
298.15	335.75	384.61	433.9	473.1	506.8
300.00	336.59	385.63	435.0	474.4	508.3
400.00	381.40	439.06	494.9	543.1	588.6
500.00	425.71	491.16	552.7	609.2	664.9
600.00	468.38	540.92	608.1	672.3	736.7
700.00	508.84	588.04	660.8	732.2	804.3
800.00	546.96	632.38	710.8	788.9	867.7
900.00	582.86	674.08	758.0	842.3	927.3
1000.00	616.66	713.35	802.4	892.5	983.1

Table 4a. Increments per carbon atom

T/K	C11-C10	C12-C11	C13-C12	C14-C13
298.15	48.86	49.3	39.2	33.7
300.00	49.04	49.4	39.4	33.9
400.00	57.66	55.9	48.2	45.5
500.00	65.45	61.6	56.4	55.7
600.00	72.54	67.2	64.2	64.5
700.00	79.20	72.8	71.4	72.1
800.00	85.42	78.5	78.1	78.8
900.00	91.23	84.0	84.3	85.0
1000.00	96.69	89.0	90.1	90.6

TABLE 5. Standard enthalpy of formation for alkylnaphthalene isomer groups

T/K	C10H8	C11H10	C12H12	C13H14	C14H16
298.15	150.96	116.37	83.9	51.0	19.4
300.00	150.83	116.20	83.7	50.8	19.2
400.00	144.56	109.04	75.0	41.2	9.8
500.00	139.49	103.19	67.5	33.2	2.1
600.00	135.39	98.51	61.5	26.9	-4.0
700.00	132.21	94.95	56.8	22.2	-8.4
800.00	129.75	92.24	53.5	19.1	-11.4
900.00	127.95	90.35	51.3	17.1	-13.1
1000.00	126.90	89.34	49.9	16.1	-13.9

Table 5a. Increments per carbon atom

T/K	C11-C10	C12-C11	C13-C12	C14-C13
298.15	-34.59	-32.5	-32.9	-31.6
300.00	-34.63	-32.5	-32.9	-31.6
400.00	-35.52	-34.0	-33.8	-31.5
500.00	-36.30	-35.7	-34.3	-31.1
600.00	-36.88	-37.1	-34.6	-30.8
700.00	-37.26	-38.1	-34.6	-30.6
800.00	-37.50	-38.7	-34.5	-30.4
900.00	-37.59	-39.0	-34.2	-30.2
1000.00	-37.56	-39.4	-33.8	-30.0

TABLE 6. Standard Gibbs energy of formation for alkylnaphthalene isomer groups in kJ/mol

T/K	C10H8	C11H10	C12H12	C13H14	C14H16
298.15	223.69	215.21	208.9	204.9	204.0
300.00	224.15	215.82	209.6	205.9	205.1
400.00	249.58	250.18	253.0	259.1	268.6
500.00	276.47	286.16	298.3	314.5	334.2
600.00	304.21	323.14	345.1	371.4	401.3
700.00	332.65	360.93	392.8	429.2	469.2
800.00	361.42	399.08	441.0	487.6	537.6
900.00	390.49	437.57	489.5	546.2	606.3
1000.00	419.77	476.26	538.3	605.1	675.2

Table 6a. Increments per carbon atom

T/K	C11-C10	C12-C11	C13-C12	C14-C13
298.15	-8.48	-6.4	-3.9	-1.9
300.00	-8.34	-6.2	-3.7	-1.8
400.00	5.9	2.8	6.1	9.5
500.00	9.69	12.2	16.2	19.7
600.00	18.93	22.0	26.3	29.9
700.00	28.28	31.8	36.5	39.9
800.00	37.66	41.9	46.6	50.0
900.00	47.07	52.0	56.7	60.1
1000.00	56.49	62.0	66.8	70.1

Table 7. Standard enthalpy for alkylnaphthalene isomer groups relative to isomer groups at 298.15 K in kJ/mol

T/K	C10H8	C11H10	C12H12	C13H14	C14H16
298.15	.00	.00	.0	.0	.0
300.00	.26	.29	.3	.4	.5
400.00	15.81	18.88	21.3	24.4	28.6
500.00	35.71	42.24	47.3	54.1	62.8
600.00	59.09	69.56	77.7	88.8	102.3
700.00	85.40	100.19	112.0	127.7	146.2
800.00	113.96	133.38	149.5	170.2	193.7
900.00	144.43	168.77	189.5	215.6	244.3
1000.00	176.62	206.13	231.7	263.3	297.3

Table 7a. Increments per carbon atom

T/K	C11-C10	C12-C11	C13-C12	C14-C13
298.15	.00	.0	.0	.0
300.00	.03	.0	.1	.1
400.00	3.07	2.4	3.1	4.1
500.00	6.54	5.1	6.8	8.7
600.00	10.46	8.2	11.1	13.5
700.00	14.79	11.8	15.8	18.4
800.00	19.43	16.1	20.8	23.5
900.00	24.35	20.8	26.1	28.7
1000.00	29.51	25.5	31.6	34.0

Table 8. Standard enthalpy for alkylnaphthalene isomer groups relative to the elements at 298.15 K in kJ/mol

T/K	C10H8	C11H10	C12H12	C13H14	C14H16
298.15	150.96	116.37	83.9	51.0	19.4
300.00	151.22	116.66	84.2	51.4	19.9
400.00	166.77	135.25	105.2	75.4	48.0
500.00	186.67	158.61	131.2	105.1	82.3
600.00	210.05	185.93	161.6	139.8	121.7
700.00	236.36	216.56	195.9	178.7	165.6
800.00	264.92	249.75	233.4	221.2	213.2
900.00	295.38	285.14	273.5	266.6	263.8
1000.00	327.58	322.50	315.6	314.3	316.7

Table 8a. Increments per carbon atom

T/K	C11-C10	C12-C11	C13-C12	C14-C13
298.15	-34.59	-32.5	-32.9	-31.6
300.00	-34.56	-32.4	-32.9	-31.5
400.00	-31.52	-30.0	-29.8	-27.5
500.00	-28.05	-27.4	-26.1	-22.9
600.00	-24.13	-24.3	-21.8	-18.1
700.00	-19.80	-20.7	-17.1	-13.2
800.00	-15.16	-16.4	-12.1	-8.1
900.00	-10.24	-11.7	-6.8	-2.9
1000.00	-5.08	-6.9	-1.3	2.5

5. Equilibrium Mole Fractions Within Alkylnaphthalene Isomer Groups

The equilibrium mole fractions within isomer groups calculated from standard Gibbs energies of formation are given in Table 9 for the ideal gas state. Since the uncertainties in $\Delta_f G^\circ(\text{I})$ and $\Delta_f G_i^\circ$ are about the same, the uncertainty in the difference is nearly independent of the relative values of the two parameters, but the absolute uncertainty does increase with temperature. Since the nearly constant uncertainty is in the exponent in the calculation, the equilibrium mole fractions of the isomers at a given temperature are uncertain by the same factor, whether they are large or small. The usual equation for the propagation of variance indicates that the equilibrium mole fractions are uncertain by about 15% at the lower temperature and 10% at the higher temperatures. This makes it difficult to indicate the uncertainties in the table. It could be done by using exponential notation, but this makes it difficult to compare the mole fractions of various isomers.

Compounds are named in tables according to the IUPAC Revised and Collected Recommendations for the Nomenclature of Organic Chemistry, 1978.¹⁷ In naming species with two or more ring hydrogens substituted, the name of the largest substituent is given first to clarify the relationships between the various structural isomers. Naphthalene has four possible starting positions, each one being the carbon adjacent to the fused ring carbons. Because of the many possible combinations for names, the group appearing first in the alphabetically arranged name has priority if two groups are in equivalent positions. For example, 1-ethyl-8-methyl is used. However, names were made to yield the lowest possible numbers when there were more than two substituents. For example, 8-ethyl-1,2-dimethylnaphthalene is used.

Table 9 shows that in general alkylnaphthalenes with branched side chains have very low equilibrium mole fractions, especially at low temperatures. At higher temperatures there is in general a more even distribution of the mole fractions between the less branched isomers, reflecting the fact that entropies are more similar than the enthalpies of formation.

6. Standard Thermodynamic Properties of Individual Alkyl-naphthalene Species

The values of C_p° , S° , $\Delta_f H^\circ$, and $\Delta_f G^\circ$ for all the alkyl-naphthalene species through $C_{14}H_{16}$ are given in Tables 10–

13 in joules for a standard state pressure of 1 bar. The values for $C_{10}H_8$ and $C_{11}H_{10}$ have been converted from the tables of Stull, Westrum, and Sinke⁵ and the values for $C_{12}H_{12}$ through $C_{14}H_{16}$ have been calculated using the Benson method.⁷ The values for chiral forms are for the racemates.

Table 10. Standard heat capacity at constant pressure for alkyl naphthalenes in J/K mol

T/K	298.15	300	400	500	600	700	800	900	1000
C10H8									
naphthalene	132.55	133.43	179.20	218.11	249.66	275.18	296.10	313.42	327.94
C11H10									
1-methylnaphthalene	159.54	160.54	212.30	256.27	292.00	321.16	345.10	364.80	381.62
2-methylnaphthalene	159.79	160.75	211.29	254.68	289.99	319.16	343.21	363.21	380.16
C12H12									
1,2-dimethylnaphthalene	185.5	186.5	239.1	285.7	326.2	360.7	389.2	411.7	428.1
1,3-dimethylnaphthalene	180.4	181.4	233.9	280.5	321.3	356.1	385.1	408.2	425.5
1,4-dimethylnaphthalene	180.4	181.4	233.9	280.5	321.3	356.1	385.1	408.2	425.5
1,5-dimethylnaphthalene	180.4	181.4	233.9	280.5	321.3	356.1	385.1	408.2	425.5
1,6-dimethylnaphthalene	180.4	181.4	233.9	280.5	321.3	356.1	385.1	408.2	425.5
1,7-dimethylnaphthalene	180.4	181.4	233.9	280.5	321.3	356.1	385.1	408.2	425.5
1,8-dimethylnaphthalene	180.4	181.4	233.9	280.5	321.3	356.1	385.1	408.2	425.5
2,3-dimethylnaphthalene	185.5	186.5	239.1	285.7	326.2	360.7	389.2	411.7	428.1
2,6-dimethylnaphthalene	180.4	181.4	233.9	280.5	321.3	356.1	385.1	408.2	425.5
2,7-dimethylnaphthalene	180.4	181.4	233.9	280.5	321.3	356.1	385.1	408.2	425.5
1-ethylnaphthalene	182.1	183.2	237.6	285.6	327.1	362.0	390.6	412.6	428.1
2-ethylnaphthalene	182.1	183.2	237.6	285.6	327.1	362.0	390.6	412.6	428.1
C13H14									
1,2,3-trimethylnaphthalene	213.7	214.8	272.1	322.9	367.3	405.2	436.7	461.6	480.1
1,2,4-trimethylnaphthalene	208.6	209.7	266.9	317.8	362.3	400.6	432.6	458.2	477.6
1,2,5-trimethylnaphthalene	208.6	209.7	266.9	317.8	362.3	400.6	432.6	458.2	477.6
1,2,6-trimethylnaphthalene	208.6	209.7	266.9	317.8	362.3	400.6	432.6	458.2	477.6
1,2,7-trimethylnaphthalene	208.6	209.7	266.9	317.8	362.3	400.6	432.6	458.2	477.6
1,2,8-trimethylnaphthalene	208.6	209.7	266.9	317.8	362.3	400.6	432.6	458.2	477.6
1,3,5-trimethylnaphthalene	203.5	204.6	261.7	312.6	357.3	396.0	428.5	454.8	475.0
1,3,6-trimethylnaphthalene	203.5	204.6	261.7	312.6	357.3	396.0	428.5	454.8	475.0
1,3,7-trimethylnaphthalene	203.5	204.6	261.7	312.6	357.3	396.0	428.5	454.8	475.0
1,3,8-trimethylnaphthalene	203.5	204.6	261.7	312.6	357.3	396.0	428.5	454.8	475.0
1,4,5-trimethylnaphthalene	203.5	204.6	261.7	312.6	357.3	396.0	428.5	454.8	475.0
1,4,6-trimethylnaphthalene	203.5	204.6	261.7	312.6	357.3	396.0	428.5	454.8	475.0
1,6,7-trimethylnaphthalene	208.6	209.7	266.9	317.8	362.3	400.6	432.6	458.2	477.6
2,3,6-trimethylnaphthalene	208.6	209.7	266.9	317.8	362.3	400.6	432.6	458.2	477.6
1-ethyl-2-methylnaphthalene	210.3	211.4	270.6	322.8	368.1	406.5	438.0	462.6	480.2
1-ethyl-3-methylnaphthalene	205.2	206.3	265.4	317.6	363.1	401.9	433.9	459.1	477.6
1-ethyl-4-methylnaphthalene	205.2	206.3	265.4	317.6	363.1	401.9	433.9	459.1	477.6
1-ethyl-5-methylnaphthalene	205.2	206.3	265.4	317.6	363.1	401.9	433.9	459.1	477.6
1-ethyl-6-methylnaphthalene	205.2	206.3	265.4	317.6	363.1	401.9	433.9	459.1	477.6
1-ethyl-7-methylnaphthalene	205.2	206.3	265.4	317.6	363.1	401.9	433.9	459.1	477.6
1-ethyl-8-methylnaphthalene	205.2	206.3	265.4	317.6	363.1	401.9	433.9	459.1	477.6
2-ethyl-1-methylnaphthalene	210.3	211.4	270.6	322.8	368.1	406.5	438.0	462.6	480.2
2-ethyl-3-methylnaphthalene	210.3	211.4	270.6	322.8	368.1	406.5	438.0	462.6	480.2
2-ethyl-6-methylnaphthalene	205.2	206.3	265.4	317.6	363.1	401.9	433.9	459.1	477.6
2-ethyl-7-methylnaphthalene	205.2	206.3	265.4	317.6	363.1	401.9	433.9	459.1	477.6
3-ethyl-1-methylnaphthalene	205.2	206.3	265.4	317.6	363.1	401.9	433.9	459.1	477.6
6-ethyl-1-methylnaphthalene	205.2	206.3	265.4	317.6	363.1	401.9	433.9	459.1	477.6
7-ethyl-1-methylnaphthalene	205.2	206.3	265.4	317.6	363.1	401.9	433.9	459.1	477.6
1-propylnaphthalene	205.1	206.3	266.7	319.9	366.1	405.2	437.1	461.9	479.6
2-propylnaphthalene	205.1	206.3	266.7	319.9	366.1	405.2	437.1	461.9	479.6
1-isopropylnaphthalene	204.0	205.2	266.3	320.2	366.8	406.1	438.1	462.8	480.2
2-isopropylnaphthalene	204.0	205.2	266.3	320.2	366.8	406.1	438.1	462.8	480.2
C14H16									
1,2,3,4-tetramethylnaphthalene	241.8	243.0	305.1	360.2	408.4	449.7	484.1	511.6	532.2
1,2,3,5-tetramethylnaphthalene	236.8	238.0	299.8	355.0	403.4	445.1	480.0	508.2	529.7
1,2,3,6-tetramethylnaphthalene	236.8	238.0	299.8	355.0	403.4	445.1	480.0	508.2	529.7
1,2,3,7-tetramethylnaphthalene	236.8	238.0	299.8	355.0	403.4	445.1	480.0	508.2	529.7
1,2,3,8-tetramethylnaphthalene	236.8	238.0	299.8	355.0	403.4	445.1	480.0	508.2	529.7
1,2,4,5-tetramethylnaphthalene	231.7	232.9	294.6	349.8	398.4	440.5	475.9	504.8	527.1
1,2,4,6-tetramethylnaphthalene	231.7	232.9	294.6	349.8	398.4	440.5	475.9	504.8	527.1
1,2,4,7-tetramethylnaphthalene	231.7	232.9	294.6	349.8	398.4	440.5	475.9	504.8	527.1
1,2,4,8-tetramethylnaphthalene	231.7	232.9	294.6	349.8	398.4	440.5	475.9	504.8	527.1
1,2,5,6-tetramethylnaphthalene	236.8	238.0	299.8	355.0	403.4	445.1	480.0	508.2	529.7
1,2,5,7-tetramethylnaphthalene	231.7	232.9	294.6	349.8	398.4	440.5	475.9	504.8	527.1
1,2,5,8-tetramethylnaphthalene	231.7	232.9	294.6	349.8	398.4	440.5	475.9	504.8	527.1
1,2,6,7-tetramethylnaphthalene	236.8	238.0	299.8	355.0	403.4	445.1	480.0	508.2	529.7
1,2,6,8-tetramethylnaphthalene	231.7	232.9	294.6	349.8	398.4	440.5	475.9	504.8	527.1
1,2,7,8-tetramethylnaphthalene	236.8	238.0	299.8	355.0	403.4	445.1	480.0	508.2	529.7
1,3,5,7-tetramethylnaphthalene	226.6	227.8	289.4	344.6	393.4	435.8	471.8	501.4	524.6
1,3,5,8-tetramethylnaphthalene	226.6	227.8	289.4	344.6	393.4	435.8	471.8	501.4	524.6
1,3,6,7-tetramethylnaphthalene	231.7	232.9	294.6	349.8	398.4	440.5	475.9	504.8	527.1
1,3,6,8-tetramethylnaphthalene	226.6	227.8	289.4	344.6	393.4	435.8	471.8	501.4	524.6
1,4,5,8-tetramethylnaphthalene	226.6	227.8	289.4	344.6	393.4	435.8	471.8	501.4	524.6
1,4,6,7-tetramethylnaphthalene	231.7	232.9	294.6	349.8	398.4	440.5	475.9	504.8	527.1
2,3,6,7-tetramethylnaphthalene	236.8	238.0	299.8	355.0	403.4	445.1	480.0	508.2	529.7
1-ethyl-2,3-dimethylnaphthalene	238.4	239.7	303.5	360.0	409.2	451.0	485.4	512.5	532.3
1-ethyl-2,4-dimethylnaphthalene	233.4	234.6	298.3	354.9	404.2	446.4	481.3	509.1	529.7
1-ethyl-2,5-dimethylnaphthalene	233.4	234.6	298.3	354.9	404.2	446.4	481.3	509.1	529.7
1-ethyl-2,6-dimethylnaphthalene	233.4	234.6	298.3	354.9	404.2	446.4	481.3	509.1	529.7
1-ethyl-2,7-dimethylnaphthalene	233.4	234.6	298.3	354.9	404.2	446.4	481.3	509.1	529.7
1-ethyl-2,8-dimethylnaphthalene	233.4	234.6	298.3	354.9	404.2	446.4	481.3	509.1	529.7

Table 11. Standard entropy of alkylnaphthalenes in J/K mol

T/K	298.15	300	400	500	600	700	800	900	1000
C10H8									
naphthalene	335.75	336.59	381.40	425.71	468.38	508.84	546.96	582.86	616.66
C11H10									
1-methylnaphthalene	377.55	378.55	432.02	484.28	534.24	581.52	625.99	667.83	707.16
2-methylnaphthalene	380.14	381.15	434.49	486.46	536.08	583.07	627.29	668.88	708.08
C12H12									
1,2-dimethylnaphthalene	412.8	414.0	475.0	533.5	589.2	642.2	692.3	739.5	783.8
1,3-dimethylnaphthalene	419.6	420.7	480.2	537.5	592.4	644.6	694.1	740.8	784.8
1,4-dimethylnaphthalene	413.8	414.9	474.5	531.8	586.6	638.8	688.3	735.1	779.0
1,5-dimethylnaphthalene	413.8	414.9	474.5	531.8	586.6	638.8	688.3	735.1	779.0
1,6-dimethylnaphthalene	419.6	420.7	480.2	537.5	592.4	644.6	694.1	740.8	784.8
1,7-dimethylnaphthalene	419.6	420.7	480.2	537.5	592.4	644.6	694.1	740.8	784.8
1,8-dimethylnaphthalene	404.5	405.6	465.2	522.5	577.3	629.5	679.0	725.8	769.7
2,3-dimethylnaphthalene	407.1	408.2	469.2	527.7	583.5	636.4	686.5	733.7	778.0
2,6-dimethylnaphthalene	413.8	414.9	474.5	531.8	586.6	638.8	688.3	735.1	779.0
2,7-dimethylnaphthalene	413.8	414.9	474.5	531.8	586.6	638.8	688.3	735.1	779.0
1-ethylnaphthalene	421.0	422.1	482.4	540.7	596.5	649.6	699.9	747.3	791.6
2-ethylnaphthalene	421.0	422.1	482.4	540.7	596.5	649.6	699.9	747.3	791.6
C13H14									
1,2,3-trimethylnaphthalene	443.8	445.1	514.9	581.2	644.1	703.6	759.9	812.8	862.5
1,2,4-trimethylnaphthalene	450.5	451.8	520.1	585.3	647.2	706.0	761.7	814.2	863.5
1,2,5-trimethylnaphthalene	450.5	451.8	520.1	585.3	647.2	706.0	761.7	814.2	863.5
1,2,6-trimethylnaphthalene	450.5	451.8	520.1	585.3	647.2	706.0	761.7	814.2	863.5
1,2,7-trimethylnaphthalene	450.5	451.8	520.1	585.3	647.2	706.0	761.7	814.2	863.5
1,2,8-trimethylnaphthalene	441.2	442.5	510.8	576.0	637.9	696.7	752.4	804.9	854.2
1,3,5-trimethylnaphthalene	457.3	458.5	525.3	589.3	650.4	708.4	763.5	815.5	864.6
1,3,6-trimethylnaphthalene	457.3	458.5	525.3	589.3	650.4	708.4	763.5	815.5	864.6
1,3,7-trimethylnaphthalene	457.3	458.5	525.3	589.3	650.4	708.4	763.5	815.5	864.6
1,3,8-trimethylnaphthalene	448.0	449.2	516.1	580.0	641.1	699.1	754.2	806.2	855.3
1,4,5-trimethylnaphthalene	448.0	449.2	516.1	580.0	641.1	699.1	754.2	806.2	855.3
1,4,6-trimethylnaphthalene	457.3	458.5	525.3	589.3	650.4	708.4	763.5	815.5	864.6
1,6,7-trimethylnaphthalene	450.5	451.8	520.1	585.3	647.2	706.0	761.7	814.2	863.5
2,3,6-trimethylnaphthalene	450.5	451.8	520.1	585.3	647.2	706.0	761.7	814.2	863.5
1-ethyl-2-methylnaphthalene	451.9	453.2	522.3	588.4	651.4	711.1	767.5	820.6	870.3
1-ethyl-3-methylnaphthalene	458.6	459.9	527.5	592.5	654.5	713.5	769.3	822.0	871.3
1-ethyl-4-methylnaphthalene	458.6	459.9	527.5	592.5	654.5	713.5	769.3	822.0	871.3
1-ethyl-5-methylnaphthalene	458.6	459.9	527.5	592.5	654.5	713.5	769.3	822.0	871.3
1-ethyl-6-methylnaphthalene	458.6	459.9	527.5	592.5	654.5	713.5	769.3	822.0	871.3
1-ethyl-7-methylnaphthalene	458.6	459.9	527.5	592.5	654.5	713.5	769.3	822.0	871.3
1-ethyl-8-methylnaphthalene	449.4	450.6	518.2	583.2	645.2	704.2	760.0	812.7	862.1
2-ethyl-1-methylnaphthalene	451.9	453.2	522.3	588.4	651.4	711.1	767.5	820.6	870.3
2-ethyl-3-methylnaphthalene	451.9	453.2	522.3	588.4	651.4	711.1	767.5	820.6	870.3
2-ethyl-6-methylnaphthalene	458.6	459.9	527.5	592.5	654.5	713.5	769.3	822.0	871.3
2-ethyl-7-methylnaphthalene	458.6	459.9	527.5	592.5	654.5	713.5	769.3	822.0	871.3
3-ethyl-1-methylnaphthalene	458.6	459.9	527.5	592.5	654.5	713.5	769.3	822.0	871.3
6-ethyl-1-methylnaphthalene	458.6	459.9	527.5	592.5	654.5	713.5	769.3	822.0	871.3
7-ethyl-1-methylnaphthalene	458.6	459.9	527.5	592.5	654.5	713.5	769.3	822.0	871.3
1-propylnaphthalene	460.4	461.7	529.4	594.8	657.3	716.8	773.0	826.0	875.7
2-propylnaphthalene	460.4	461.7	529.4	594.8	657.3	716.8	773.0	826.0	875.7
1-isopropylnaphthalene	449.2	450.4	518.0	583.4	646.0	705.5	761.9	815.0	864.8
2-isopropylnaphthalene	449.2	450.4	518.0	583.4	646.0	705.5	761.9	815.0	864.8
C14H16									
1,2,3,4-tetramethylnaphthalene	469.0	470.5	549.0	623.2	693.2	759.3	821.7	880.4	935.4
1,2,3,5-tetramethylnaphthalene	481.5	482.9	560.0	633.0	702.1	767.5	829.3	887.5	942.2
1,2,3,6-tetramethylnaphthalene	481.5	482.9	560.0	633.0	702.1	767.5	829.3	887.5	942.2
1,2,3,7-tetramethylnaphthalene	481.5	482.9	560.0	633.0	702.1	767.5	829.3	887.5	942.2
1,2,3,8-tetramethylnaphthalene	472.2	473.6	550.7	623.7	692.8	758.2	820.0	878.2	932.9
1,2,4,5-tetramethylnaphthalene	478.9	480.4	556.0	627.8	695.9	760.6	821.8	879.6	934.0
1,2,4,6-tetramethylnaphthalene	488.2	489.6	565.3	637.0	705.2	769.9	831.1	888.9	943.3
1,2,4,7-tetramethylnaphthalene	488.2	489.6	565.3	637.0	705.2	769.9	831.1	888.9	943.3
1,2,4,8-tetramethylnaphthalene	478.9	480.4	556.0	627.8	695.9	760.6	821.8	879.6	934.0
1,2,5,6-tetramethylnaphthalene	475.7	477.2	554.3	627.2	696.3	761.7	823.5	881.7	936.5
1,2,5,7-tetramethylnaphthalene	488.2	489.6	565.3	637.0	705.2	769.9	831.1	888.9	943.3
1,2,5,8-tetramethylnaphthalene	478.9	480.4	556.0	627.8	695.9	760.6	821.8	879.6	934.0
1,2,6,7-tetramethylnaphthalene	481.5	482.9	560.0	633.0	702.1	767.5	829.3	887.5	942.2
1,2,6,8-tetramethylnaphthalene	478.9	480.4	556.0	627.8	695.9	760.6	821.8	879.6	934.0
1,2,7,8-tetramethylnaphthalene	466.4	467.9	545.0	617.9	687.0	752.4	814.2	872.5	927.2
1,3,5,7-tetramethylnaphthalene	489.2	490.6	564.7	635.3	702.6	766.5	827.1	884.5	938.5
1,3,5,8-tetramethylnaphthalene	485.7	487.1	561.2	631.8	699.1	763.0	823.6	880.9	935.0
1,3,6,7-tetramethylnaphthalene	488.2	489.6	565.3	637.0	705.2	769.9	831.1	888.9	943.3
1,3,6,8-tetramethylnaphthalene	479.9	481.3	555.4	626.1	693.3	757.2	817.8	875.2	929.3
1,4,5,8-tetramethylnaphthalene	464.8	466.2	540.4	611.0	678.2	742.2	802.8	860.1	914.2
1,4,6,7-tetramethylnaphthalene	482.4	483.9	559.5	631.3	699.5	764.1	825.3	883.1	937.5
2,3,6,7-tetramethylnaphthalene	469.9	471.4	548.5	621.5	690.6	756.0	817.8	876.0	930.7
1-ethyl-2,3-dimethylnaphthalene	482.9	484.3	562.2	636.2	706.3	772.6	835.1	893.9	949.0
1-ethyl-2,4-dimethylnaphthalene	489.6	491.0	567.4	640.2	709.4	774.9	836.9	895.3	950.1
1-ethyl-2,5-dimethylnaphthalene	489.6	491.0	567.4	640.2	709.4	774.9	836.9	895.3	950.1
1-ethyl-2,6-dimethylnaphthalene	489.6	491.0	567.4	640.2	709.4	774.9	836.9	895.3	950.1
1-ethyl-2,7-dimethylnaphthalene	489.6	491.0	567.4	640.2	709.4	774.9	836.9	895.3	950.1
1-ethyl-2,8-dimethylnaphthalene	480.3	481.8	558.1	630.9	700.1	765.7	827.6	886.0	940.8

Table 12. Standard enthalpy of formation for alkyl naphthalenes in kJ/mol

T/K	298.15	300	400	500	600	700	800	900	1000
C10H8									
naphthalene	150.96	150.83	144.56	139.49	135.39	132.21	129.75	127.95	126.90
C11H10									
1-methylnaphthalene	116.86	116.69	109.54	103.76	99.20	95.73	93.14	91.34	90.42
2-methylnaphthalene	116.11	115.94	108.74	102.84	98.07	94.43	91.63	89.66	88.58
C12H12									
1,2-dimethylnaphthalene	86.1	86.0	77.6	70.5	64.6	60.1	56.9	54.7	53.2
1,3-dimethylnaphthalene	83.8	83.6	74.7	67.0	60.7	55.7	52.0	49.4	47.7
1,4-dimethylnaphthalene	83.8	83.6	74.7	67.0	60.7	55.7	52.0	49.4	47.7
1,5-dimethylnaphthalene	83.8	83.6	74.7	67.0	60.7	55.7	52.0	49.4	47.7
1,6-dimethylnaphthalene	83.8	83.6	74.7	67.0	60.7	55.7	52.0	49.4	47.7
1,7-dimethylnaphthalene	83.8	83.6	74.7	67.0	60.7	55.7	52.0	49.4	47.7
1,8-dimethylnaphthalene	108.4	108.2	99.3	91.6	85.3	80.3	76.6	74.0	72.3
2,3-dimethylnaphthalene	86.1	86.0	77.6	70.5	64.6	60.1	56.9	54.7	53.2
2,6-dimethylnaphthalene	83.8	83.6	74.7	67.0	60.7	55.7	52.0	49.4	47.7
2,7-dimethylnaphthalene	83.8	83.6	74.7	67.0	60.7	55.7	52.0	49.4	47.7
1-ethylnaphthalene	96.9	96.7	88.1	80.8	75.0	70.6	67.5	65.4	64.1
2-ethylnaphthalene	96.9	96.7	88.1	80.8	75.0	70.6	67.5	65.4	64.1
C13H14									
1,2,3-trimethylnaphthalene	55.1	54.9	45.7	37.8	31.4	26.4	22.9	20.6	19.1
1,2,4-trimethylnaphthalene	52.7	52.5	42.8	34.4	27.4	22.0	18.1	15.3	13.6
1,2,5-trimethylnaphthalene	52.7	52.5	42.8	34.4	27.4	22.0	18.1	15.3	13.6
1,2,6-trimethylnaphthalene	52.7	52.5	42.8	34.4	27.4	22.0	18.1	15.3	13.6
1,2,7-trimethylnaphthalene	52.7	52.5	42.8	34.4	27.4	22.0	18.1	15.3	13.6
1,2,8-trimethylnaphthalene	77.3	77.1	67.4	59.0	52.0	46.6	42.7	39.9	38.2
1,3,5-trimethylnaphthalene	50.3	50.1	39.9	30.9	23.5	17.6	13.2	10.1	8.0
1,3,6-trimethylnaphthalene	50.3	50.1	39.9	30.9	23.5	17.6	13.2	10.1	8.0
1,3,7-trimethylnaphthalene	50.3	50.1	39.9	30.9	23.5	17.6	13.2	10.1	8.0
1,3,8-trimethylnaphthalene	74.9	74.7	64.5	55.5	48.1	42.2	37.8	34.7	32.6
1,4,5-trimethylnaphthalene	74.9	74.7	64.5	55.5	48.1	42.2	37.8	34.7	32.6
1,4,6-trimethylnaphthalene	50.3	50.1	39.9	30.9	23.5	17.6	13.2	10.1	8.0
1,6,7-trimethylnaphthalene	52.7	52.5	42.8	34.4	27.4	22.0	18.1	15.3	13.6
2,3,6-trimethylnaphthalene	52.7	52.5	42.8	34.4	27.4	22.0	18.1	15.3	13.6
1-ethyl-2-methylnaphthalene	65.8	65.6	56.2	48.2	41.8	36.9	33.6	31.4	29.9
1-ethyl-3-methylnaphthalene	63.4	63.2	53.3	44.7	37.8	32.5	28.7	26.1	24.4
1-ethyl-4-methylnaphthalene	63.4	63.2	53.3	44.7	37.8	32.5	28.7	26.1	24.4
1-ethyl-5-methylnaphthalene	63.4	63.2	53.3	44.7	37.8	32.5	28.7	26.1	24.4
1-ethyl-6-methylnaphthalene	63.4	63.2	53.3	44.7	37.8	32.5	28.7	26.1	24.4
1-ethyl-7-methylnaphthalene	63.4	63.2	53.3	44.7	37.8	32.5	28.7	26.1	24.4
1-ethyl-8-methylnaphthalene	88.0	87.8	77.9	69.3	62.4	57.1	53.3	50.7	49.0
2-ethyl-1-methylnaphthalene	65.8	65.6	56.2	48.2	41.8	36.9	33.6	31.4	29.9
2-ethyl-3-methylnaphthalene	65.8	65.6	56.2	48.2	41.8	36.9	33.6	31.4	29.9
2-ethyl-6-methylnaphthalene	63.4	63.2	53.3	44.7	37.8	32.5	28.7	26.1	24.4
2-ethyl-7-methylnaphthalene	63.4	63.2	53.3	44.7	37.8	32.5	28.7	26.1	24.4
3-ethyl-1-methylnaphthalene	63.4	63.2	53.3	44.7	37.8	32.5	28.7	26.1	24.4
6-ethyl-1-methylnaphthalene	63.4	63.2	53.3	44.7	37.8	32.5	28.7	26.1	24.4
7-ethyl-1-methylnaphthalene	63.4	63.2	53.3	44.7	37.8	32.5	28.7	26.1	24.4
1-propylnaphthalene	76.2	76.0	66.1	57.8	51.2	46.2	42.7	40.4	38.9
2-propylnaphthalene	76.2	76.0	66.1	57.8	51.2	46.2	42.7	40.4	38.9
1-isopropylnaphthalene	70.4	70.2	60.2	51.9	45.3	40.4	37.0	34.8	33.4
2-isopropylnaphthalene	70.4	70.2	60.2	51.9	45.3	40.4	37.0	34.8	33.4
C14H16									
1,2,3,4-tetramethylnaphthalene	24.1	23.8	13.8	5.1	-1.9	-7.3	-11.0	-13.5	-15.0
1,2,3,5-tetramethylnaphthalene	21.7	21.4	10.9	1.7	-5.8	-11.7	-15.9	-18.8	-20.5
1,2,3,6-tetramethylnaphthalene	21.7	21.4	10.9	1.7	-5.8	-11.7	-15.9	-18.8	-20.5
1,2,3,7-tetramethylnaphthalene	21.7	21.4	10.9	1.7	-5.8	-11.7	-15.9	-18.8	-20.5
1,2,3,8-tetramethylnaphthalene	46.3	46.0	35.5	26.3	18.8	12.9	8.7	5.8	4.1
1,2,4,5-tetramethylnaphthalene	43.9	43.7	32.6	22.9	14.8	8.5	3.8	.6	-1.5
1,2,4,6-tetramethylnaphthalene	19.3	19.0	8.0	-1.7	-9.8	-16.1	-20.8	-24.0	-26.1
1,2,4,7-tetramethylnaphthalene	19.3	19.0	8.0	-1.7	-9.8	-16.1	-20.8	-24.0	-26.1
1,2,4,8-tetramethylnaphthalene	43.9	43.7	32.6	22.9	14.8	8.5	3.8	.6	-1.5
1,2,5,6-tetramethylnaphthalene	21.7	21.4	10.9	1.7	-5.8	-11.7	-15.9	-18.8	-20.5
1,2,5,7-tetramethylnaphthalene	19.3	19.0	8.0	-1.7	-9.8	-16.1	-20.8	-24.0	-26.1
1,2,5,8-tetramethylnaphthalene	43.9	43.7	32.6	22.9	14.8	8.5	3.8	.6	-1.5
1,2,6,7-tetramethylnaphthalene	21.7	21.4	10.9	1.7	-5.8	-11.7	-15.9	-18.8	-20.5
1,2,6,8-tetramethylnaphthalene	43.9	43.7	32.6	22.9	14.8	8.5	3.8	.6	-1.5
1,2,7,8-tetramethylnaphthalene	46.3	46.0	35.5	26.3	18.8	12.9	8.7	5.8	4.1
1,3,5,7-tetramethylnaphthalene	16.9	16.7	5.0	-5.2	-13.7	-20.5	-25.6	-29.2	-31.6
1,3,5,8-tetramethylnaphthalene	41.5	41.3	29.6	19.4	10.9	4.1	-1.0	-4.6	-7.0
1,3,6,7-tetramethylnaphthalene	19.3	19.0	8.0	-1.7	-9.8	-16.1	-20.8	-24.0	-26.1
1,3,6,8-tetramethylnaphthalene	41.5	41.3	29.6	19.4	10.9	4.1	-1.0	-4.6	-7.0
1,4,5,8-tetramethylnaphthalene	66.1	65.9	54.3	44.0	35.5	28.7	23.6	20.0	17.6
1,4,6,7-tetramethylnaphthalene	19.3	19.0	8.0	-1.7	-9.8	-16.1	-20.8	-24.0	-26.1
2,3,6,7-tetramethylnaphthalene	21.7	21.4	10.9	1.7	-5.8	-11.7	-15.9	-18.8	-20.5
1-ethyl-2,3-dimethylnaphthalene	34.8	34.5	24.2	15.5	8.5	3.3	-.4	-2.7	-4.2
1-ethyl-2,4-dimethylnaphthalene	32.4	32.1	21.3	12.1	4.6	-1.2	-5.2	-8.0	-9.7
1-ethyl-2,5-dimethylnaphthalene	32.4	32.1	21.3	12.1	4.6	-1.2	-5.2	-8.0	-9.7
1-ethyl-2,6-dimethylnaphthalene	32.4	32.1	21.3	12.1	4.6	-1.2	-5.2	-8.0	-9.7
1-ethyl-2,7-dimethylnaphthalene	32.4	32.1	21.3	12.1	4.6	-1.2	-5.2	-8.0	-9.7
1-ethyl-2,8-dimethylnaphthalene	57.0	56.7	45.9	36.7	29.2	23.4	19.4	16.6	14.9

Table 12. Standard enthalpy of formation for alkylnaphthalenes in kJ/mol--Continued

T/K	298.15	300	400	500	600	700	800	900	1000
C14H16									
1-ethyl-3,5-dimethylnaphthalene	30.0	29.8	18.4	8.7	.6	-5.6	-10.1	-13.2	-15.2
1-ethyl-3,6-dimethylnaphthalene	30.0	29.8	18.4	8.7	.6	-5.6	-10.1	-13.2	-15.2
1-ethyl-3,7-dimethylnaphthalene	30.0	29.8	18.4	8.7	.6	-5.6	-10.1	-13.2	-15.2
1-ethyl-3,8-dimethylnaphthalene	54.6	54.4	43.0	33.3	25.2	19.0	14.5	11.4	9.4
1-ethyl-4,5-dimethylnaphthalene	54.6	54.4	43.0	33.3	25.2	19.0	14.5	11.4	9.4
1-ethyl-4,6-dimethylnaphthalene	30.0	29.8	18.4	8.7	.6	-5.6	-10.1	-13.2	-15.2
1-ethyl-6,7-dimethylnaphthalene	32.4	32.1	21.3	12.1	4.6	-1.2	-5.2	-8.0	-9.7
2-ethyl-1,3-dimethylnaphthalene	34.8	34.5	24.2	15.5	8.5	3.3	-.4	-2.7	-4.2
2-ethyl-1,4-dimethylnaphthalene	32.4	32.1	21.3	12.1	4.6	-1.2	-5.2	-8.0	-9.7
2-ethyl-1,5-dimethylnaphthalene	32.4	32.1	21.3	12.1	4.6	-1.2	-5.2	-8.0	-9.7
2-ethyl-1,6-dimethylnaphthalene	32.4	32.1	21.3	12.1	4.6	-1.2	-5.2	-8.0	-9.7
2-ethyl-1,7-dimethylnaphthalene	32.4	32.1	21.3	12.1	4.6	-1.2	-5.2	-8.0	-9.7
2-ethyl-1,8-dimethylnaphthalene	57.0	56.7	45.9	36.7	29.2	23.4	19.4	16.6	14.9
2-ethyl-3,6-dimethylnaphthalene	32.4	32.1	21.3	12.1	4.6	-1.2	-5.2	-8.0	-9.7
3-ethyl-1,2-dimethylnaphthalene	34.8	34.5	24.2	15.5	8.5	3.3	-.4	-2.7	-4.2
3-ethyl-1,5-dimethylnaphthalene	30.0	29.8	18.4	8.7	.6	-5.6	-10.1	-13.2	-15.2
3-ethyl-1,6-dimethylnaphthalene	30.0	29.8	18.4	8.7	.6	-5.6	-10.1	-13.2	-15.2
3-ethyl-1,7-dimethylnaphthalene	30.0	29.8	18.4	8.7	.6	-5.6	-10.1	-13.2	-15.2
3-ethyl-1,8-dimethylnaphthalene	54.6	54.4	43.0	33.3	25.2	19.0	14.5	11.4	9.4
3-ethyl-2,6-dimethylnaphthalene	32.4	32.1	21.3	12.1	4.6	-1.2	-5.2	-8.0	-9.7
4-ethyl-1,2-dimethylnaphthalene	32.4	32.1	21.3	12.1	4.6	-1.2	-5.2	-8.0	-9.7
4-ethyl-1,5-dimethylnaphthalene	54.6	54.4	43.0	33.3	25.2	19.0	14.5	11.4	9.4
4-ethyl-1,6-dimethylnaphthalene	30.0	29.8	18.4	8.7	.6	-5.6	-10.1	-13.2	-15.2
5-ethyl-1,2-dimethylnaphthalene	32.4	32.1	21.3	12.1	4.6	-1.2	-5.2	-8.0	-9.7
5-ethyl-1,3-dimethylnaphthalene	30.0	29.8	18.4	8.7	.6	-5.6	-10.1	-13.2	-15.2
5-ethyl-1,4-dimethylnaphthalene	54.6	54.4	43.0	33.3	25.2	19.0	14.5	11.4	9.4
6-ethyl-1,2-dimethylnaphthalene	32.4	32.1	21.3	12.1	4.6	-1.2	-5.2	-8.0	-9.7
6-ethyl-1,3-dimethylnaphthalene	30.0	29.8	18.4	8.7	.6	-5.6	-10.1	-13.2	-15.2
6-ethyl-1,4-dimethylnaphthalene	30.0	29.8	18.4	8.7	.6	-5.6	-10.1	-13.2	-15.2
6-ethyl-1,7-dimethylnaphthalene	32.4	32.1	21.3	12.1	4.6	-1.2	-5.2	-8.0	-9.7
6-ethyl-2,3-dimethylnaphthalene	32.4	32.1	21.3	12.1	4.6	-1.2	-5.2	-8.0	-9.7
7-ethyl-1,2-dimethylnaphthalene	32.4	32.1	21.3	12.1	4.6	-1.2	-5.2	-8.0	-9.7
7-ethyl-1,3-dimethylnaphthalene	30.0	29.8	18.4	8.7	.6	-5.6	-10.1	-13.2	-15.2
7-ethyl-1,6-dimethylnaphthalene	32.4	32.1	21.3	12.1	4.6	-1.2	-5.2	-8.0	-9.7
8-ethyl-1,2-dimethylnaphthalene	57.0	56.7	45.9	36.7	29.2	23.4	19.4	16.6	14.9
8-ethyl-1,3-dimethylnaphthalene	54.6	54.4	43.0	33.3	25.2	19.0	14.5	11.4	9.4
1-methyl-2-propylnaphthalene	45.2	45.0	34.2	25.1	17.9	12.5	8.7	6.3	4.8
1-methyl-3-propylnaphthalene	42.8	42.6	31.3	21.7	14.0	8.1	3.9	1.1	-.7
1-methyl-4-propylnaphthalene	42.8	42.6	31.3	21.7	14.0	8.1	3.9	1.1	-.7
1-methyl-5-propylnaphthalene	42.8	42.6	31.3	21.7	14.0	8.1	3.9	1.1	-.7
1-methyl-6-propylnaphthalene	42.8	42.6	31.3	21.7	14.0	8.1	3.9	1.1	-.7
1-methyl-7-propylnaphthalene	42.8	42.6	31.3	21.7	14.0	8.1	3.9	1.1	-.7
1-methyl-8-propylnaphthalene	67.4	67.2	55.9	46.3	38.6	32.7	28.5	25.7	23.9
2-methyl-1-propylnaphthalene	45.2	45.0	34.2	25.1	17.9	12.5	8.7	6.3	4.8
2-methyl-3-propylnaphthalene	45.2	45.0	34.2	25.1	17.9	12.5	8.7	6.3	4.8
2-methyl-6-propylnaphthalene	42.8	42.6	31.3	21.7	14.0	8.1	3.9	1.1	-.7
2-methyl-7-propylnaphthalene	42.8	42.6	31.3	21.7	14.0	8.1	3.9	1.1	-.7
3-methyl-1-propylnaphthalene	42.8	42.6	31.3	21.7	14.0	8.1	3.9	1.1	-.7
6-methyl-1-propylnaphthalene	42.8	42.6	31.3	21.7	14.0	8.1	3.9	1.1	-.7
7-methyl-1-propylnaphthalene	42.8	42.6	31.3	21.7	14.0	8.1	3.9	1.1	-.7
1-butyl-naphthalene	55.6	55.4	44.2	34.8	27.3	21.7	17.8	15.3	13.8
2-butyl-naphthalene	55.6	55.4	44.2	34.8	27.3	21.7	17.8	15.3	13.8
1,2-diethylnaphthalene	45.5	45.2	34.7	25.9	18.9	13.8	10.3	8.0	6.7
1,3-diethylnaphthalene	43.1	42.9	31.8	22.5	15.0	9.4	5.4	2.8	1.1
1,4-diethylnaphthalene	43.1	42.9	31.8	22.5	15.0	9.4	5.4	2.8	1.1
1,5-diethylnaphthalene	43.1	42.9	31.8	22.5	15.0	9.4	5.4	2.8	1.1
1,6-diethylnaphthalene	43.1	42.9	31.8	22.5	15.0	9.4	5.4	2.8	1.1
1,7-diethylnaphthalene	43.1	42.9	31.8	22.5	15.0	9.4	5.4	2.8	1.1
1,8-diethylnaphthalene	67.7	67.5	56.4	47.1	39.6	34.0	30.0	27.4	25.7
2,3-diethylnaphthalene	45.5	45.2	34.7	25.9	18.9	13.8	10.3	8.0	6.7
2,6-diethylnaphthalene	43.1	42.9	31.8	22.5	15.0	9.4	5.4	2.8	1.1
2,7-diethylnaphthalene	43.1	42.9	31.8	22.5	15.0	9.4	5.4	2.8	1.1
1-t-butyl-naphthalene	43.6	43.4	32.3	23.4	16.5	11.7	8.6	6.8	5.8
2-t-butyl-naphthalene	43.6	43.4	32.3	23.4	16.5	11.7	8.6	6.8	5.8
1-isopropyl-2-methylnaphthalene	39.4	39.1	28.3	19.3	12.1	6.7	3.1	.7	-.7
1-isopropyl-3-methylnaphthalene	37.0	36.7	25.4	15.8	8.1	2.3	-1.8	-4.5	-6.2
1-isopropyl-4-methylnaphthalene	37.0	36.7	25.4	15.8	8.1	2.3	-1.8	-4.5	-6.2
1-isopropyl-5-methylnaphthalene	37.0	36.7	25.4	15.8	8.1	2.3	-1.8	-4.5	-6.2
1-isopropyl-6-methylnaphthalene	37.0	36.7	25.4	15.8	8.1	2.3	-1.8	-4.5	-6.2
1-isopropyl-7-methylnaphthalene	37.0	36.7	25.4	15.8	8.1	2.3	-1.8	-4.5	-6.2
1-isopropyl-8-methylnaphthalene	61.6	61.3	50.0	40.4	32.7	26.9	22.8	20.1	18.4
2-isopropyl-1-methylnaphthalene	39.4	39.1	28.3	19.3	12.1	6.7	3.1	.7	-.7
2-isopropyl-3-methylnaphthalene	39.4	39.1	28.3	19.3	12.1	6.7	3.1	.7	-.7
2-isopropyl-6-methylnaphthalene	37.0	36.7	25.4	15.8	8.1	2.3	-1.8	-4.5	-6.2
2-isopropyl-7-methylnaphthalene	37.0	36.7	25.4	15.8	8.1	2.3	-1.8	-4.5	-6.2
3-isopropyl-1-methylnaphthalene	37.0	36.7	25.4	15.8	8.1	2.3	-1.8	-4.5	-6.2
6-isopropyl-1-methylnaphthalene	37.0	36.7	25.4	15.8	8.1	2.3	-1.8	-4.5	-6.2
7-isopropyl-1-methylnaphthalene	37.0	36.7	25.4	15.8	8.1	2.3	-1.8	-4.5	-6.2
1-secbutyl-naphthalene	53.1	52.9	41.6	32.2	24.8	19.3	15.5	13.1	11.6
2-secbutyl-naphthalene	53.1	52.9	41.6	32.2	24.8	19.3	15.5	13.1	11.6
1-isobutyl-naphthalene	49.6	49.3	38.1	28.7	21.2	15.7	11.9	9.5	8.0
2-isobutyl-naphthalene	49.6	49.3	38.1	28.7	21.2	15.7	11.9	9.5	8.0

Table 13. Standard Gibbs energy of formation for alkyl naphthalenes in kJ/mol

T/K	298.15	300	400	500	600	700	800	900	1000
C10H8									
naphthalene	223.69	224.15	249.58	276.47	304.21	332.65	361.42	390.49	419.77
C11H10									
1-methylnaphthalene	217.82	218.41	253.47	290.17	327.83	366.28	405.07	444.19	483.52
2-methylnaphthalene	216.28	216.90	251.72	288.16	325.61	363.89	402.56	441.55	480.76
C12H12									
1,2-dimethylnaphthalene	217.4	218.2	263.6	310.9	359.6	409.1	459.2	509.6	560.2
1,3-dimethylnaphthalene	213.0	213.8	258.6	305.4	353.8	403.0	452.9	503.1	553.6
1,4-dimethylnaphthalene	214.7	215.5	260.9	308.3	357.2	407.0	457.5	508.3	559.4
1,5-dimethylnaphthalene	214.7	215.5	260.9	308.3	357.2	407.0	457.5	508.3	559.4
1,6-dimethylnaphthalene	213.0	213.8	258.6	305.4	353.8	403.0	452.9	503.1	553.6
1,7-dimethylnaphthalene	213.0	213.8	258.6	305.4	353.8	403.0	452.9	503.1	553.6
1,8-dimethylnaphthalene	242.1	242.9	289.2	337.6	387.4	438.1	489.6	541.3	593.3
2,3-dimethylnaphthalene	219.1	219.9	265.9	313.8	363.0	413.1	463.8	514.8	566.0
2,6-dimethylnaphthalene	214.7	215.5	260.9	308.3	357.2	407.0	457.5	508.3	559.4
2,7-dimethylnaphthalene	214.7	215.5	260.9	308.3	357.2	407.0	457.5	508.3	559.4
1-ethylnaphthalene	225.7	226.4	271.1	317.7	365.6	414.4	463.8	513.4	563.2
2-ethylnaphthalene	225.7	226.4	271.1	317.7	365.6	414.4	463.8	513.4	563.2
C13H14									
1,2,3-trimethylnaphthalene	217.8	218.8	274.9	333.1	392.8	453.4	514.7	576.2	638.1
1,2,4-trimethylnaphthalene	213.4	214.4	269.9	327.6	387.0	447.3	508.4	569.8	631.5
1,2,5-trimethylnaphthalene	213.4	214.4	269.9	327.6	387.0	447.3	508.4	569.8	631.5
1,2,6-trimethylnaphthalene	213.4	214.4	269.9	327.6	387.0	447.3	508.4	569.8	631.5
1,2,7-trimethylnaphthalene	213.4	214.4	269.9	327.6	387.0	447.3	508.4	569.8	631.5
1,2,8-trimethylnaphthalene	240.8	241.8	298.2	356.9	417.1	478.4	540.4	602.7	665.4
1,3,5-trimethylnaphthalene	209.0	210.0	264.9	322.1	381.2	441.2	502.1	563.3	624.9
1,3,6-trimethylnaphthalene	209.0	210.0	264.9	322.1	381.2	441.2	502.1	563.3	624.9
1,3,7-trimethylnaphthalene	209.0	210.0	264.9	322.1	381.2	441.2	502.1	563.3	624.9
1,3,8-trimethylnaphthalene	236.4	237.3	293.2	351.4	411.3	472.3	534.1	596.3	658.8
1,4,5-trimethylnaphthalene	236.4	237.3	293.2	351.4	411.3	472.3	534.1	596.3	658.8
1,4,6-trimethylnaphthalene	209.0	210.0	264.9	322.1	381.2	441.2	502.1	563.3	624.9
1,6,7-trimethylnaphthalene	213.4	214.4	269.9	327.6	387.0	447.3	508.4	569.8	631.5
2,3,6-trimethylnaphthalene	213.4	214.4	269.9	327.6	387.0	447.3	508.4	569.8	631.5
1-ethyl-2-methylnaphthalene	226.1	227.0	282.4	339.8	398.8	458.7	519.3	580.0	641.1
1-ethyl-3-methylnaphthalene	221.7	222.6	277.4	334.4	393.0	452.6	512.9	573.6	634.5
1-ethyl-4-methylnaphthalene	221.7	222.6	277.4	334.4	393.0	452.6	512.9	573.6	634.5
1-ethyl-5-methylnaphthalene	221.7	222.6	277.4	334.4	393.0	452.6	512.9	573.6	634.5
1-ethyl-6-methylnaphthalene	221.7	222.6	277.4	334.4	393.0	452.6	512.9	573.6	634.5
1-ethyl-7-methylnaphthalene	221.7	222.6	277.4	334.4	393.0	452.6	512.9	573.6	634.5
1-ethyl-8-methylnaphthalene	249.0	250.0	305.7	363.6	423.2	483.7	545.0	606.5	668.4
2-ethyl-1-methylnaphthalene	226.1	227.0	282.4	339.8	398.8	458.7	519.3	580.0	641.1
2-ethyl-3-methylnaphthalene	226.1	227.0	282.4	339.8	398.8	458.7	519.3	580.0	641.1
2-ethyl-6-methylnaphthalene	221.7	222.6	277.4	334.4	393.0	452.6	512.9	573.6	634.5
2-ethyl-7-methylnaphthalene	221.7	222.6	277.4	334.4	393.0	452.6	512.9	573.6	634.5
3-ethyl-1-methylnaphthalene	221.7	222.6	277.4	334.4	393.0	452.6	512.9	573.6	634.5
6-ethyl-1-methylnaphthalene	221.7	222.6	277.4	334.4	393.0	452.6	512.9	573.6	634.5
7-ethyl-1-methylnaphthalene	221.7	222.6	277.4	334.4	393.0	452.6	512.9	573.6	634.5
1-propylnaphthalene	234.0	234.9	289.5	346.3	404.7	464.0	523.9	584.2	644.7
2-propylnaphthalene	234.0	234.9	289.5	346.3	404.7	464.0	523.9	584.2	644.7
1-isopropylnaphthalene	231.5	232.5	288.2	346.1	405.6	466.1	527.2	588.5	650.1
2-isopropylnaphthalene	231.5	232.5	288.2	346.1	405.6	466.1	527.2	588.5	650.1
C14H16									
1,2,3,4-tetramethylnaphthalene	219.9	221.1	288.5	358.1	429.4	501.7	574.8	648.1	721.7
1,2,3,5-tetramethylnaphthalene	213.8	215.0	281.1	349.8	420.2	491.6	563.9	636.4	709.3
1,2,3,6-tetramethylnaphthalene	213.8	215.0	281.1	349.8	420.2	491.6	563.9	636.4	709.3
1,2,3,7-tetramethylnaphthalene	213.8	215.0	281.1	349.8	420.2	491.6	563.9	636.4	709.3
1,2,3,8-tetramethylnaphthalene	241.2	242.4	309.5	379.0	450.3	522.7	595.9	669.4	743.2
1,2,4,5-tetramethylnaphthalene	236.8	237.9	304.5	373.6	444.5	516.6	589.6	662.9	736.6
1,2,4,6-tetramethylnaphthalene	209.4	210.6	276.1	344.3	414.4	485.5	557.6	630.0	702.7
1,2,4,7-tetramethylnaphthalene	209.4	210.6	276.1	344.3	414.4	485.5	557.6	630.0	702.7
1,2,4,8-tetramethylnaphthalene	236.8	237.9	304.5	373.6	444.5	516.6	589.6	662.9	736.6
1,2,5,6-tetramethylnaphthalene	215.5	216.7	283.5	352.7	423.6	495.7	568.5	641.6	715.1
1,2,5,7-tetramethylnaphthalene	209.4	210.6	276.1	344.3	414.4	485.5	557.6	630.0	702.7
1,2,5,8-tetramethylnaphthalene	236.8	237.9	304.5	373.6	444.5	516.6	589.6	662.9	736.6
1,2,6,7-tetramethylnaphthalene	213.8	215.0	281.1	349.8	420.2	491.6	563.9	636.4	709.3
1,2,6,8-tetramethylnaphthalene	236.8	237.9	304.5	373.6	444.5	516.6	589.6	662.9	736.6
1,2,7,8-tetramethylnaphthalene	242.9	244.1	311.8	381.9	453.8	526.8	600.5	674.6	749.0
1,3,5,7-tetramethylnaphthalene	206.7	207.9	273.5	341.7	412.0	483.5	555.9	628.7	701.9
1,3,5,8-tetramethylnaphthalene	232.4	233.5	299.5	368.1	438.7	510.5	583.3	656.5	730.1
1,3,6,7-tetramethylnaphthalene	209.4	210.6	276.1	344.3	414.4	485.5	557.6	630.0	702.7
1,3,6,8-tetramethylnaphthalene	234.1	235.3	301.8	371.0	442.2	514.6	587.9	661.7	735.8
1,4,5,8-tetramethylnaphthalene	263.2	264.4	332.4	403.1	475.8	549.7	624.5	699.8	775.5
1,4,6,7-tetramethylnaphthalene	211.1	212.3	278.5	347.2	417.8	489.6	562.2	635.2	708.5
2,3,6,7-tetramethylnaphthalene	217.2	218.4	285.8	355.5	427.1	499.7	573.1	646.8	720.9
1-ethyl-2,3-dimethylnaphthalene	226.5	227.6	293.6	362.0	432.0	503.0	574.7	646.7	718.9
1-ethyl-2,4-dimethylnaphthalene	222.1	223.2	288.6	356.6	426.2	496.9	568.4	640.2	712.3
1-ethyl-2,5-dimethylnaphthalene	222.1	223.2	288.6	356.6	426.2	496.9	568.4	640.2	712.3
1-ethyl-2,6-dimethylnaphthalene	222.1	223.2	288.6	356.6	426.2	496.9	568.4	640.2	712.3
1-ethyl-2,7-dimethylnaphthalene	222.1	223.2	288.6	356.6	426.2	496.9	568.4	640.2	712.3
1-ethyl-2,8-dimethylnaphthalene	249.4	250.6	317.0	385.8	456.4	528.0	600.5	673.2	746.2

Table 13. Standard Gibbs energy of formation for alkylnaphthalenes in kJ/mol--Continued

T/K	298.15	300	400	500	600	700	800	900	1000
C14H16									
1-ethyl-3,5-dimethylnaphthalene	217.7	218.8	283.7	351.1	420.4	490.8	562.1	633.8	705.8
1-ethyl-3,6-dimethylnaphthalene	217.7	218.8	283.7	351.1	420.4	490.8	562.1	633.8	705.8
1-ethyl-3,7-dimethylnaphthalene	217.7	218.8	283.7	351.1	420.4	490.8	562.1	633.8	705.8
1-ethyl-3,8-dimethylnaphthalene	245.0	246.2	312.0	380.3	450.6	521.9	594.2	666.7	739.6
1-ethyl-4,5-dimethylnaphthalene	245.0	246.2	312.0	380.3	450.6	521.9	594.2	666.7	739.6
1-ethyl-4,6-dimethylnaphthalene	217.7	218.8	283.7	351.1	420.4	490.8	562.1	633.8	705.8
1-ethyl-6,7-dimethylnaphthalene	222.1	223.2	288.6	356.6	426.2	496.9	568.4	640.2	712.3
2-ethyl-1,3-dimethylnaphthalene	226.5	227.6	293.6	362.0	432.0	503.0	574.7	646.7	718.9
2-ethyl-1,4-dimethylnaphthalene	222.1	223.2	288.6	356.6	426.2	496.9	568.4	640.2	712.3
2-ethyl-1,5-dimethylnaphthalene	222.1	223.2	288.6	356.6	426.2	496.9	568.4	640.2	712.3
2-ethyl-1,6-dimethylnaphthalene	222.1	223.2	288.6	356.6	426.2	496.9	568.4	640.2	712.3
2-ethyl-1,7-dimethylnaphthalene	222.1	223.2	288.6	356.6	426.2	496.9	568.4	640.2	712.3
2-ethyl-1,8-dimethylnaphthalene	249.4	250.6	317.0	385.8	456.4	528.0	600.5	673.2	746.2
2-ethyl-3,6-dimethylnaphthalene	222.1	223.2	288.6	356.6	426.2	496.9	568.4	640.2	712.3
3-ethyl-1,2-dimethylnaphthalene	226.5	227.6	293.6	362.0	432.0	503.0	574.7	646.7	718.9
3-ethyl-1,5-dimethylnaphthalene	217.7	218.8	283.7	351.1	420.4	490.8	562.1	633.8	705.8
3-ethyl-1,6-dimethylnaphthalene	217.7	218.8	283.7	351.1	420.4	490.8	562.1	633.8	705.8
3-ethyl-1,7-dimethylnaphthalene	217.7	218.8	283.7	351.1	420.4	490.8	562.1	633.8	705.8
3-ethyl-1,8-dimethylnaphthalene	245.0	246.2	312.0	380.3	450.6	521.9	594.2	666.7	739.6
3-ethyl-2,6-dimethylnaphthalene	222.1	223.2	288.6	356.6	426.2	496.9	568.4	640.2	712.3
4-ethyl-1,2-dimethylnaphthalene	222.1	223.2	288.6	356.6	426.2	496.9	568.4	640.2	712.3
4-ethyl-1,5-dimethylnaphthalene	245.0	246.2	312.0	380.3	450.6	521.9	594.2	666.7	739.6
4-ethyl-1,6-dimethylnaphthalene	217.7	218.8	283.7	351.1	420.4	490.8	562.1	633.8	705.8
5-ethyl-1,2-dimethylnaphthalene	222.1	223.2	288.6	356.6	426.2	496.9	568.4	640.2	712.3
5-ethyl-1,3-dimethylnaphthalene	217.7	218.8	283.7	351.1	420.4	490.8	562.1	633.8	705.8
5-ethyl-1,4-dimethylnaphthalene	245.0	246.2	312.0	380.3	450.6	521.9	594.2	666.7	739.6
6-ethyl-1,2-dimethylnaphthalene	222.1	223.2	288.6	356.6	426.2	496.9	568.4	640.2	712.3
6-ethyl-1,3-dimethylnaphthalene	217.7	218.8	283.7	351.1	420.4	490.8	562.1	633.8	705.8
6-ethyl-1,4-dimethylnaphthalene	217.7	218.8	283.7	351.1	420.4	490.8	562.1	633.8	705.8
6-ethyl-1,7-dimethylnaphthalene	222.1	223.2	288.6	356.6	426.2	496.9	568.4	640.2	712.3
6-ethyl-2,3-dimethylnaphthalene	222.1	223.2	288.6	356.6	426.2	496.9	568.4	640.2	712.3
7-ethyl-1,2-dimethylnaphthalene	222.1	223.2	288.6	356.6	426.2	496.9	568.4	640.2	712.3
7-ethyl-1,3-dimethylnaphthalene	217.7	218.8	283.7	351.1	420.4	490.8	562.1	633.8	705.8
7-ethyl-1,6-dimethylnaphthalene	222.1	223.2	288.6	356.6	426.2	496.9	568.4	640.2	712.3
8-ethyl-1,2-dimethylnaphthalene	249.4	250.6	317.0	385.8	456.4	528.0	600.5	673.2	746.2
8-ethyl-1,3-dimethylnaphthalene	245.0	246.2	312.0	380.3	450.6	521.9	594.2	666.7	739.6
1-methyl-2-propylnaphthalene	234.4	235.5	300.8	368.5	437.9	508.3	579.4	650.8	722.5
1-methyl-3-propylnaphthalene	230.0	231.1	295.8	363.0	432.0	502.2	573.1	644.4	715.9
1-methyl-4-propylnaphthalene	230.0	231.1	295.8	363.0	432.0	502.2	573.1	644.4	715.9
1-methyl-5-propylnaphthalene	230.0	231.1	295.8	363.0	432.0	502.2	573.1	644.4	715.9
1-methyl-6-propylnaphthalene	230.0	231.1	295.8	363.0	432.0	502.2	573.1	644.4	715.9
1-methyl-7-propylnaphthalene	230.0	231.1	295.8	363.0	432.0	502.2	573.1	644.4	715.9
1-methyl-8-propylnaphthalene	257.3	258.5	324.1	392.2	462.2	533.3	605.1	677.3	749.8
2-methyl-1-propylnaphthalene	234.4	235.5	300.8	368.5	437.9	508.3	579.4	650.8	722.5
2-methyl-3-propylnaphthalene	234.4	235.5	300.8	368.5	437.9	508.3	579.4	650.8	722.5
2-methyl-6-propylnaphthalene	230.0	231.1	295.8	363.0	432.0	502.2	573.1	644.4	715.9
2-methyl-7-propylnaphthalene	230.0	231.1	295.8	363.0	432.0	502.2	573.1	644.4	715.9
3-methyl-1-propylnaphthalene	230.0	231.1	295.8	363.0	432.0	502.2	573.1	644.4	715.9
6-methyl-1-propylnaphthalene	230.0	231.1	295.8	363.0	432.0	502.2	573.1	644.4	715.9
7-methyl-1-propylnaphthalene	230.0	231.1	295.8	363.0	432.0	502.2	573.1	644.4	715.9
1-butynaphthalene	242.2	243.4	307.9	374.9	443.7	513.5	584.1	655.0	726.1
2-butynaphthalene	242.2	243.4	307.9	374.9	443.7	513.5	584.1	655.0	726.1
1,2-diethylnaphthalene	234.7	235.9	301.2	368.8	438.1	508.3	579.3	650.5	721.9
1,3-diethylnaphthalene	230.4	231.5	296.2	363.3	432.3	502.2	573.0	644.0	715.3
1,4-diethylnaphthalene	232.1	233.2	298.5	366.2	435.7	506.3	577.6	649.2	721.1
1,5-diethylnaphthalene	232.1	233.2	298.5	366.2	435.7	506.3	577.6	649.2	721.1
1,6-diethylnaphthalene	230.4	231.5	296.2	363.3	432.3	502.2	573.0	644.0	715.3
1,7-diethylnaphthalene	230.4	231.5	296.2	363.3	432.3	502.2	573.0	644.0	715.3
1,8-diethylnaphthalene	259.4	260.6	326.8	395.5	465.9	537.4	609.6	682.2	755.0
2,3-diethylnaphthalene	236.5	237.6	303.5	371.7	441.5	512.4	583.9	655.7	727.7
2,6-diethylnaphthalene	232.1	233.2	298.5	366.2	435.7	506.3	577.6	649.2	721.1
2,7-diethylnaphthalene	232.1	233.2	298.5	366.2	435.7	506.3	577.6	649.2	721.1
1-t-butynaphthalene	241.6	242.8	311.0	381.8	454.2	527.5	601.4	675.6	749.9
2-t-butynaphthalene	241.6	242.8	311.0	381.8	454.2	527.5	601.4	675.6	749.9
1-isopropyl-2-methylnaphthalene	231.9	233.1	299.4	368.3	438.8	510.4	582.6	655.1	727.9
1-isopropyl-3-methylnaphthalene	227.5	228.7	294.4	362.8	433.0	504.3	576.3	648.7	721.3
1-isopropyl-4-methylnaphthalene	227.5	228.7	294.4	362.8	433.0	504.3	576.3	648.7	721.3
1-isopropyl-5-methylnaphthalene	227.5	228.7	294.4	362.8	433.0	504.3	576.3	648.7	721.3
1-isopropyl-6-methylnaphthalene	227.5	228.7	294.4	362.8	433.0	504.3	576.3	648.7	721.3
1-isopropyl-7-methylnaphthalene	227.5	228.7	294.4	362.8	433.0	504.3	576.3	648.7	721.3
1-isopropyl-8-methylnaphthalene	254.9	256.1	322.8	392.1	463.2	535.4	608.4	681.6	755.2
2-isopropyl-1-methylnaphthalene	231.9	233.1	299.4	368.3	438.8	510.4	582.6	655.1	727.9
2-isopropyl-3-methylnaphthalene	231.9	233.1	299.4	368.3	438.8	510.4	582.6	655.1	727.9
2-isopropyl-6-methylnaphthalene	227.5	228.7	294.4	362.8	433.0	504.3	576.3	648.7	721.3
2-isopropyl-7-methylnaphthalene	227.5	228.7	294.4	362.8	433.0	504.3	576.3	648.7	721.3
3-isopropyl-1-methylnaphthalene	227.5	228.7	294.4	362.8	433.0	504.3	576.3	648.7	721.3
6-isopropyl-1-methylnaphthalene	227.5	228.7	294.4	362.8	433.0	504.3	576.3	648.7	721.3
7-isopropyl-1-methylnaphthalene	227.5	228.7	294.4	362.8	433.0	504.3	576.3	648.7	721.3
1-secbutynaphthalene	241.4	242.6	307.6	375.2	444.5	514.9	586.0	657.4	729.1
2-secbutynaphthalene	241.4	242.6	307.6	375.2	444.5	514.9	586.0	657.4	729.1
1-isobutylnaphthalene	239.6	240.7	306.3	374.5	444.4	515.4	587.1	659.1	731.3
2-isobutylnaphthalene	239.6	240.7	306.3	374.5	444.4	515.4	587.1	659.1	731.3

7. Discussion

The chemical thermodynamic properties of the polycyclic aromatic hydrocarbons are of interest in connection with soot formation, carcinogenesis, and coal conversion processes. Enthalpies of formation of large polycyclic aromatic hydrocarbons have been estimated by Stein, Golden, and Benson.⁸ Shaw, Golden, and Benson¹⁸ have summarized thermodynamic properties of dihydronaphthalenes, tetrahydronaphthalenes, hexahydronaphthalenes, and other polycyclic compounds related to coal. Stein¹⁹ has used existing and new predictive methods to discuss high temperature chemical equilibria of polycyclic aromatic hydrocarbons.

8. Nomenclature

$C_{P_i}^\circ$	= standard heat capacity at constant pressure of isomer i , $\text{J K}^{-1} \text{mol}^{-1}$
$C_P^\circ(\text{I})$	= standard heat capacity at constant pressure of isomer group I, $\text{J K}^{-1} \text{mol}^{-1}$
$\Delta_f G_i^\circ$	= standard Gibbs energy of formation of isomer i , kJ mol^{-1}
$\Delta_f G^\circ(\text{I})$	= standard Gibbs energy of formation of isomer group I, kJ mol^{-1}
$H^\circ(\text{I}, T) - H^\circ(\text{I}, 298.15 \text{ K})$	= standard enthalpy for isomer groups relative to isomer groups at 298.15 K, kJ mol^{-1}
$H^\circ(\text{I}, T) - H^\circ(\text{I}, 298.15 \text{ K}) + \Delta_f H^\circ(\text{I}, 298.15 \text{ K})$	= standard enthalpy for isomer groups relative to elements at 298.15 K, kJ mol^{-1}
$\Delta_f H_i^\circ$	= standard enthalpy of formation of isomer i , kJ mol^{-1}
$\Delta_f H^\circ(\text{I})$	= standard enthalpy of formation of isomer group I, kJ mol^{-1}
n	= number of carbon atoms in a molecule
N_i	= number of isomers in an isomer group
OPT	= number of optical isomers
r_i	= equilibrium mole fraction of species i in an isomer group
S_i°	= standard entropy of isomer i , $\text{J K}^{-1} \text{mol}^{-1}$
$S^\circ(\text{I})$	= standard entropy of isomer group I, $\text{J K}^{-1} \text{mol}^{-1}$
TSN	= total symmetry number
y_i	= mole fraction of isomer i within the isomer group
y_I	= mole fraction of isomer group I in a mixture

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