## Source-attribution for atmospheric mercury deposition: Where does the *mercury* in *mercury deposition* come from?

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Presentation given via speakerphone to a meeting of the Mercury Working Group, Office of Air Quality, Indiana Department of Environmental Management (IDEM) April 21, 2005

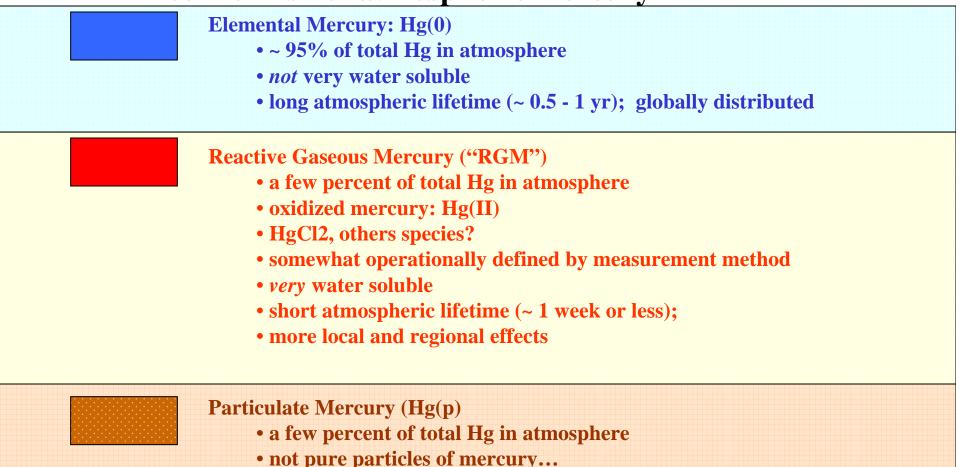
# For mercury, how important is atmospheric deposition relative to other loading pathways?

Estimates of the Percent of Great Lakes Loadings Attributable to the Atmospheric Deposition Pathway					
Pollutant	Lake Superior	Lake Michigan	Lake Huron	Lake Erie	Lake Ontario
DDT	97 <sup>a</sup>	98 <sup>a</sup>	97 <sup>a</sup>	22 <sup>a</sup>	31ª
Lead	97 <sup>a</sup> : 64 <sup>b</sup> : 69 <sup>d</sup>	99ª	98 <sup>a</sup>	46 <sup>a</sup>	73ª
Mercury	73 <sup>d</sup>	> 80 <sup>j</sup>	k	k	k
PCB's	$90^{a}$ ; ~ $95^{b,c}$ ; $82^{d}$	58ª	78 <sup>a</sup>	13 <sup>a</sup>	7 <sup>a</sup>
PCDD/F	~100° ~80 <sup>f</sup>	50-100° (PCDD) 5-35° (PCDF) 88 <sup>f</sup>	86 <sup>f</sup>	~40 <sup>f</sup>	5-35 (PCDD) <sup>e</sup> < 5 (PCDF) <sup>e</sup>
Benzo(a)pyrene	96 <sup>a</sup>	86ª	80 <sup>a</sup>	79 <sup>a</sup>	72 <sup>a</sup>
Hexachloro- benzene	99 <sup>f</sup>	95 <sup>r</sup>	96 <sup>f</sup>	> 17 <sup>f</sup>	40 <sup>f</sup>
Atrazine	97 <sup>h</sup>	~30 <sup>g</sup> ; 23 <sup>h</sup>	~20 <sup>h</sup>	~10-20 <sup>h</sup>	~5 <sup>h</sup>
Mirex	k	k	k	k	~5ª

#### **References and Notes**

(a) Strachan and Eisenreich (1988), percentages of total inputs; (b) Hoff *et al.* (1996); (c) Net loss of PCB's to the atmosphere of 1600 kg/year; total non-atmospheric inputs of approximately 70 kg/year; (d) Dolan *et al.* (1993); (e) Pearson *et al.* (1998); (f) Cohen *et al.* (1995); (g) Rygwelski et al. (1999); (h) Schottler and Eisenreich (1997); (j) Mason and Sullivan (1997); (k) no estimates could be found

### **Three "forms" of atmospheric mercury**



- (Hg compounds associated with atmospheric particulate)
- (ing compounds associated with atmospheric particular
- species largely unknown (in some cases, may be HgO?)
- moderate atmospheric lifetime (perhaps 1~ 2 weeks)
- local and regional effects
- bioavailability?

modeling the fate and transport of atmospheric mercury

#### Elemental Mercury: Hg(0) **Atmospheric Fate Processes for Hg Reactive Gaseous Mercury: RGM** Particulate Mercury: Hg(p) Upper atmospheric Polar sunrise halogen-mediated "mercury depletion events" cloud heterogeneous oxidation? Br CLOUD DROPLET "DRY" (low RH) Hg(II) reduced to Hg(0) by $SO_2$ ATMOSPHERE: Adsorption/ Hg(p) Hg(0) oxidized to RGM desorption of Hg(II) to by O<sub>3</sub>, H<sub>2</sub>O<sub>2</sub>, Cl<sub>2</sub>, OH, HCl /from soot Primary Hg(0) oxidized to dissolved Anthropogenic RGM by O<sub>3</sub>, OH, HOCI, OCI-**Emissions** Re-emission of natural Dry and Wet Deposition AND previously deposited anthropogenic mercury

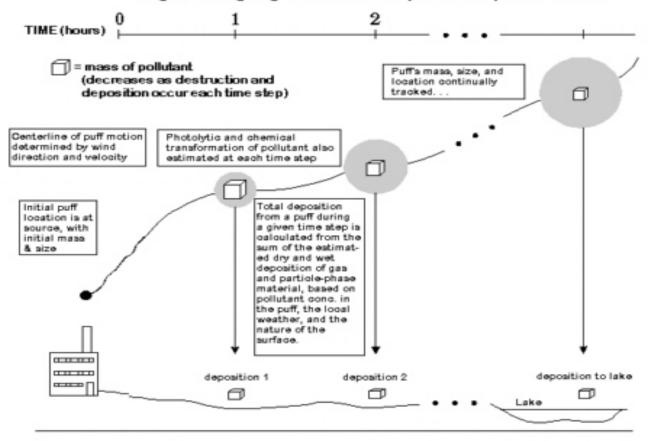
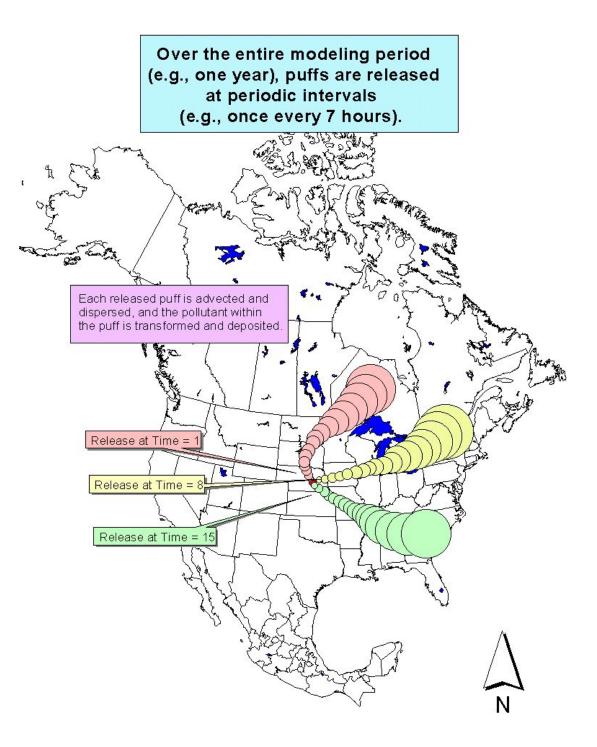
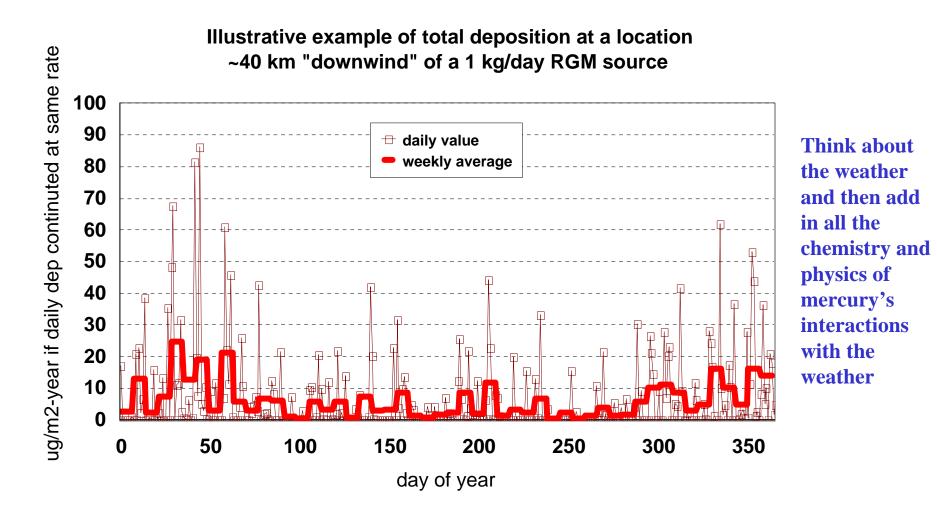


Figure 1. Lagrangian Puff Air Transport and Deposition Model



some illustrative modeling results

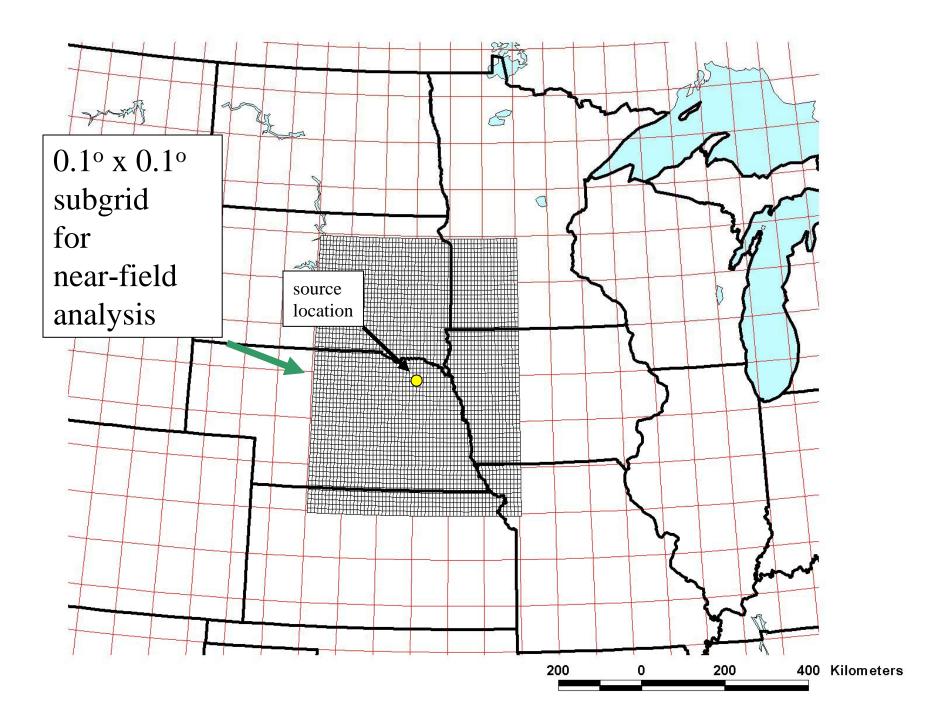
The impact of any given mercury emissions source on any receptor is highly variable, both in space and in time



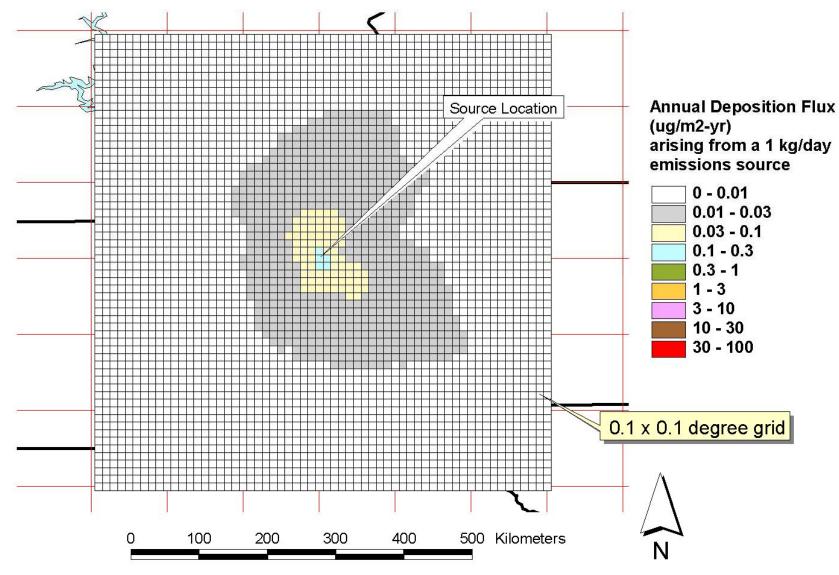
The impact of any given mercury emissions source on any receptor is highly dependent on the "type" of mercury emitted

- Elemental mercury Hg<sup>0</sup> is not readily dry or wet deposited, and its conversion to ionic Hg or Hg(p) is relatively slow
- Particulate mercury Hg(p) is moderately susceptible to dry and wet deposition
- Ionic mercury also called Reactive Gaseous Mercury or RGM – is very easily dry and wet deposited

**Conversion of RGM to Hg^0 in plumes?** 

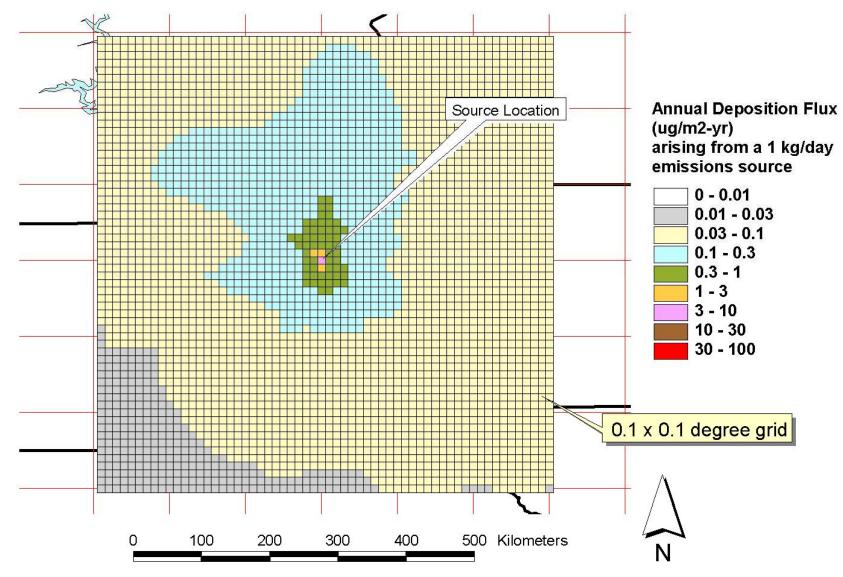


# Annual deposition summary for emissions of elemental Hg from a 250 meter high source

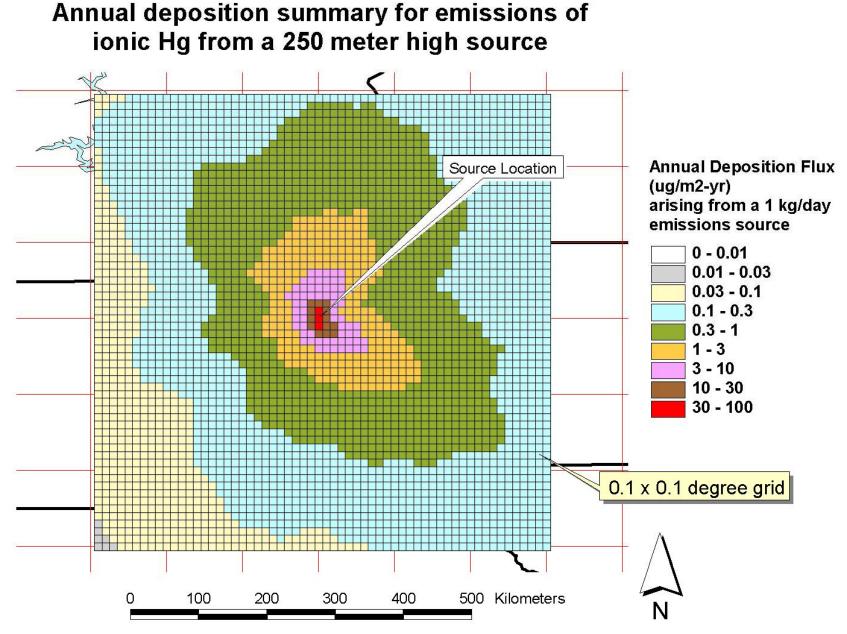


Hypothetical emissions source at lat = 42.5, long = -97.5; simulation for entire year 1996 using archived NGM meteorology (180 km resolution)

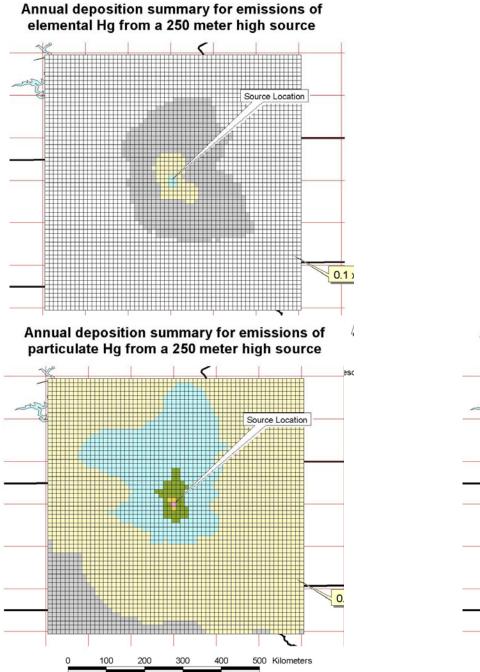
# Annual deposition summary for emissions of particulate Hg from a 250 meter high source



Hypothetical emissions source at lat = 42.5, long = -97.5; simulation for entire year 1996 using archived NGM meteorology (180 km resolution)

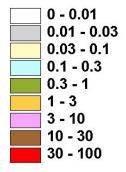


Hypothetical emissions source at lat = 42.5, long = -97.5; simulation for entire year 1996 using archived NGM meteorology (180 km resolution)

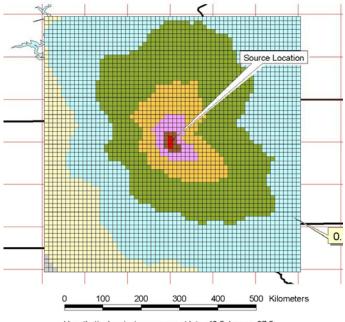


Hypothetical emissions source at lat = 42.5, long = -97.5; simulation for entire year 1996 using archived NGM meteorology (180 km  $\scriptstyle\rm I$ 

Annual Deposition Flux (ug/m2-yr) arising from a 1 kg/day emissions source



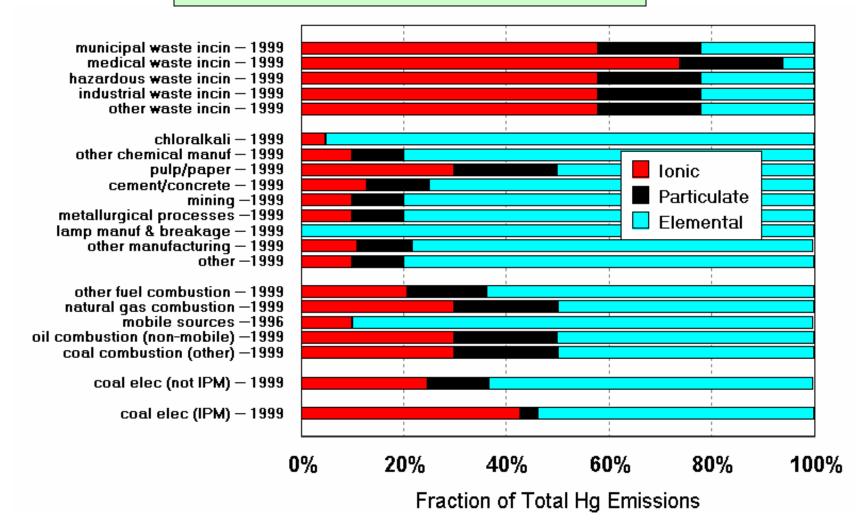
#### Annual deposition summary for emissions of ionic Hg from a 250 meter high source



Hypothetical emissions source at lat = 42.5, long = -97.5; simulation for entire year 1996 using archived NGM meteorology (180 km r

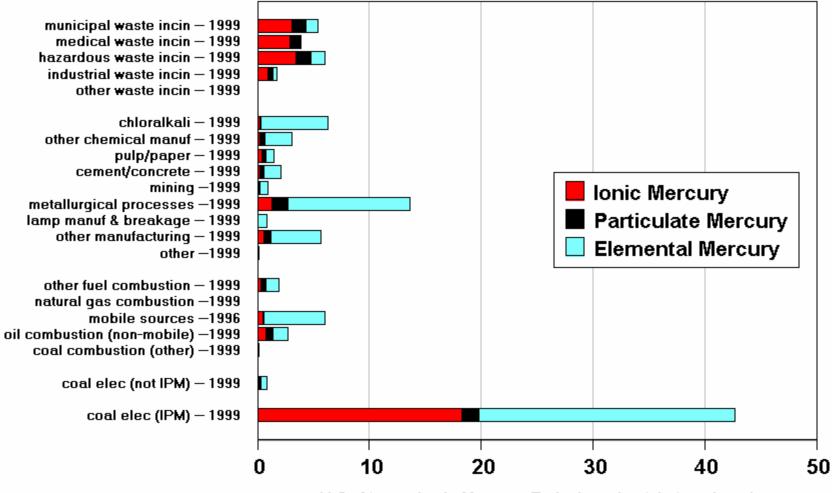
#### **Estimated Speciation Profile for 1999 U.S. Atmospheric Anthropogenic Mercury Emissions**

### **Very uncertain for most sources**



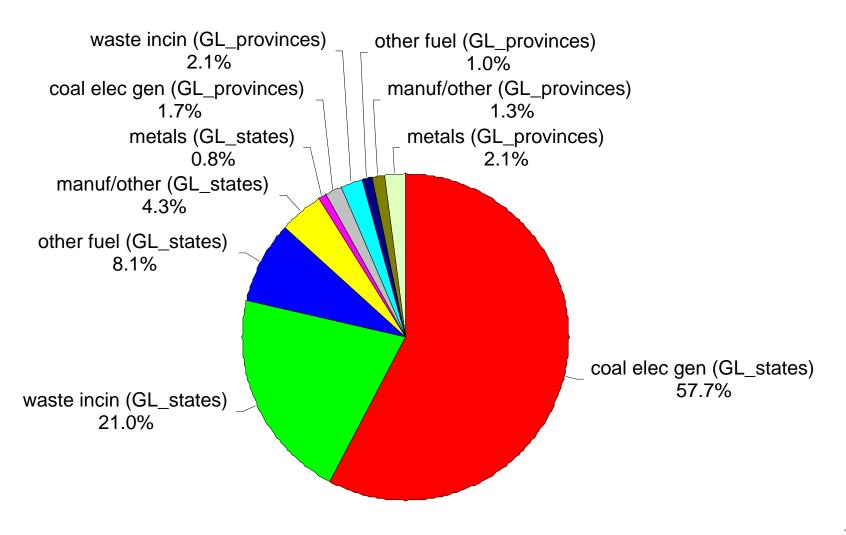
- Each type of source has a very different emissions speciation profile
- Even within a given source type, there can be big differences – depending on process type, fuels and raw materials, pollution control equipment, etc.

#### **Estimated 1999 U.S. Atmospheric Anthropogenic Mercury Emissions**

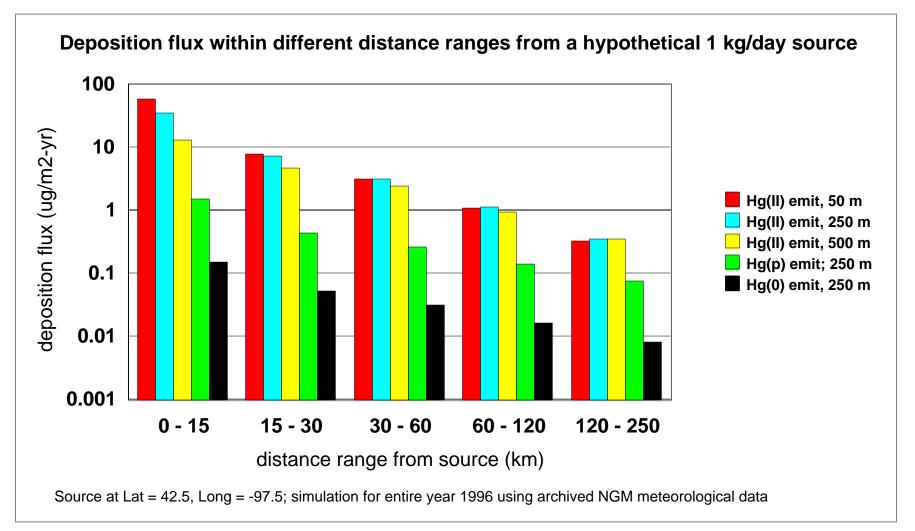


U.S. Atmospheric Mercury Emissions (metric tons/year)

Emissions of Ionic Mercury (RGM) from Different Anthropogenic Source Sectors in Great Lakes States and Provinces (~1999-2000) [Total RGM emissions = 13.4 metric tons/year]



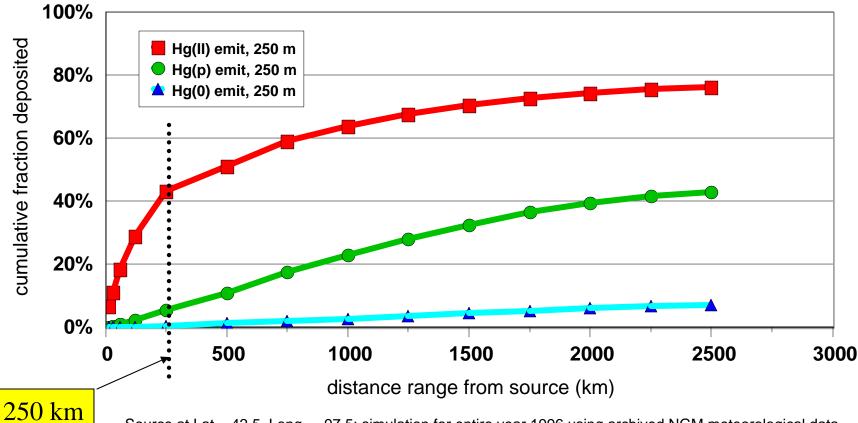
## There can be large local and regional impacts



Hypothesized rapid reduction of Hg(II) in plumes? If true, then dramatic impact on modeling results...<sub>20</sub>

# At the same time, medium to long range transport can't be ignored

Cumulative fraction deposited out to different distance ranges from a hypothetical source



Source at Lat = 42.5, Long = -97.5; simulation for entire year 1996 using archived NGM meteorological data

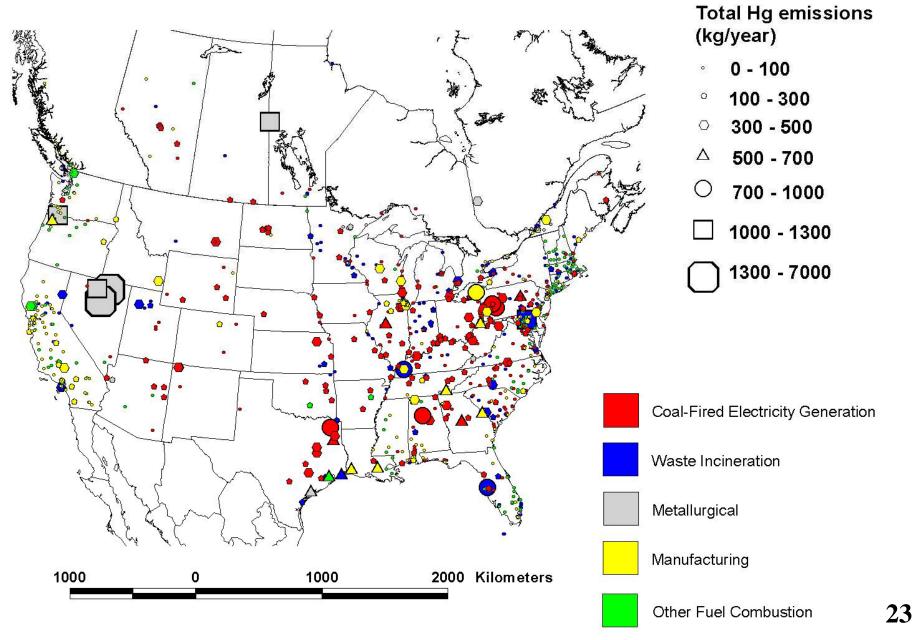
There are a lot of sources...

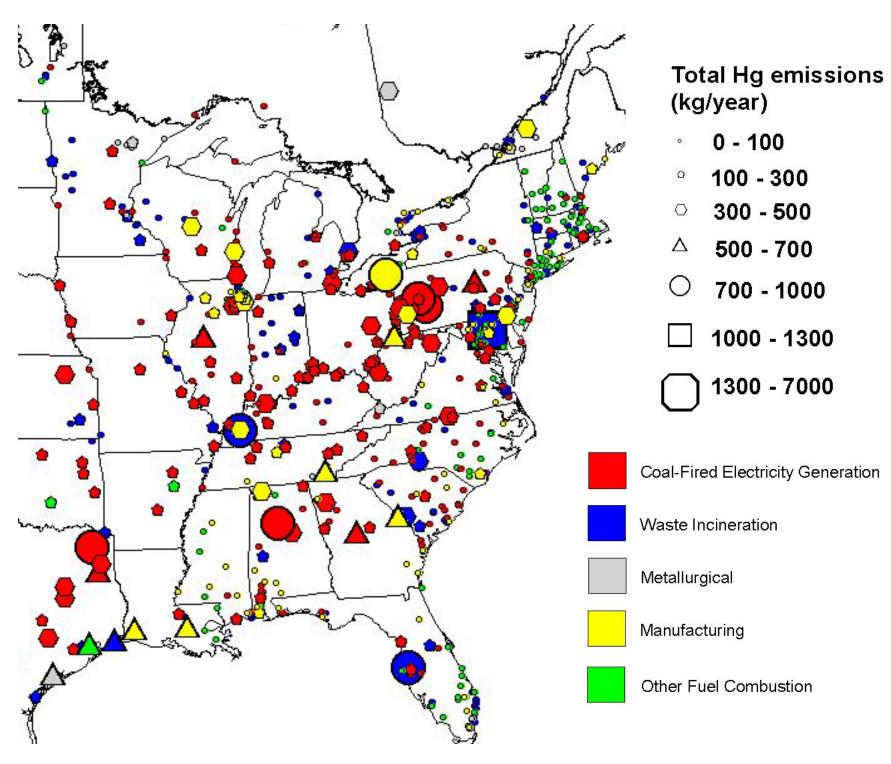
□ Large spatial and temporal variations

Each source emits mercury forms in different proportions

A lot of different sources can contribute significant amounts of mercury through atmospheric deposition to any given receptor

#### **Geographic Distribution of Largest Anthropogenic Mercury Emissions Sources in the U.S. (1999) and Canada (2000)**

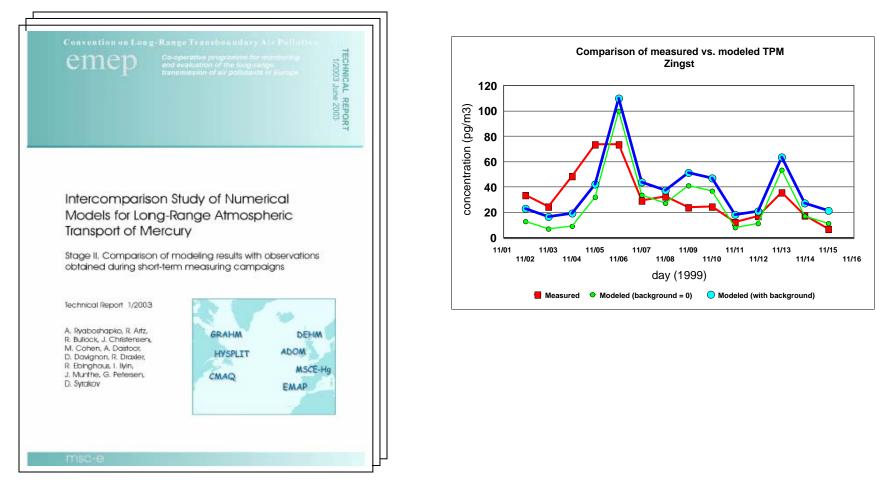




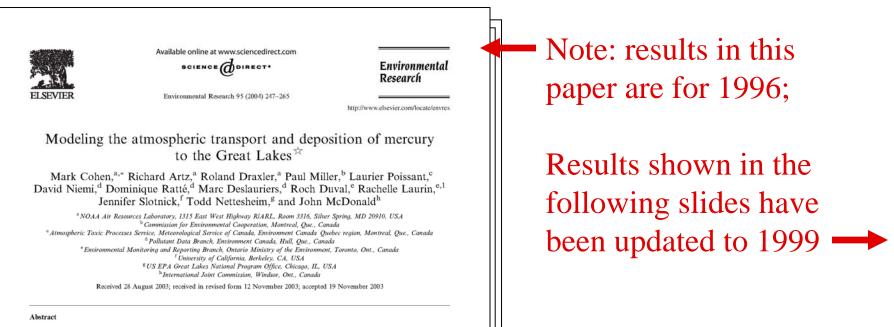
Getting the source-apportionment information we all want is difficult

- □ With measurements alone, generally impossible
- Coupling measurements with back-trajectory analyses yields only a little information
- Comprehensive fate and transport modeling "forward" from emissions to deposition – holds the promise of generating detailed source-receptor information

There are a lot of uncertainties in current comprehensive fate and transport models... Nevertheless, many models seem to be performing reasonably well, i.e., are able to explain a lot of what we see



# some source-receptor results for the Great Lakes



A special version of the NOAA HYSPLIT\_4 model has been developed and used to estimate the atmospheric fate and transport of mercury in a North American modeling domain. Spatial and chemical interpolation procedures were used to expand the modeling results and provide estimates of the contribution of each source in a 1996 anthropogenic US/Canadian emissions inventory to atmospheric mercury deposition to the Great Lakes. While there are uncertainties in the emissions inventories and ambient data

atmosphere mercury dep suitable for model evaluat the Great Lakes region ar from the Great Lakes cor significant contributions contributor to atmospher Published by Elsevier Inc

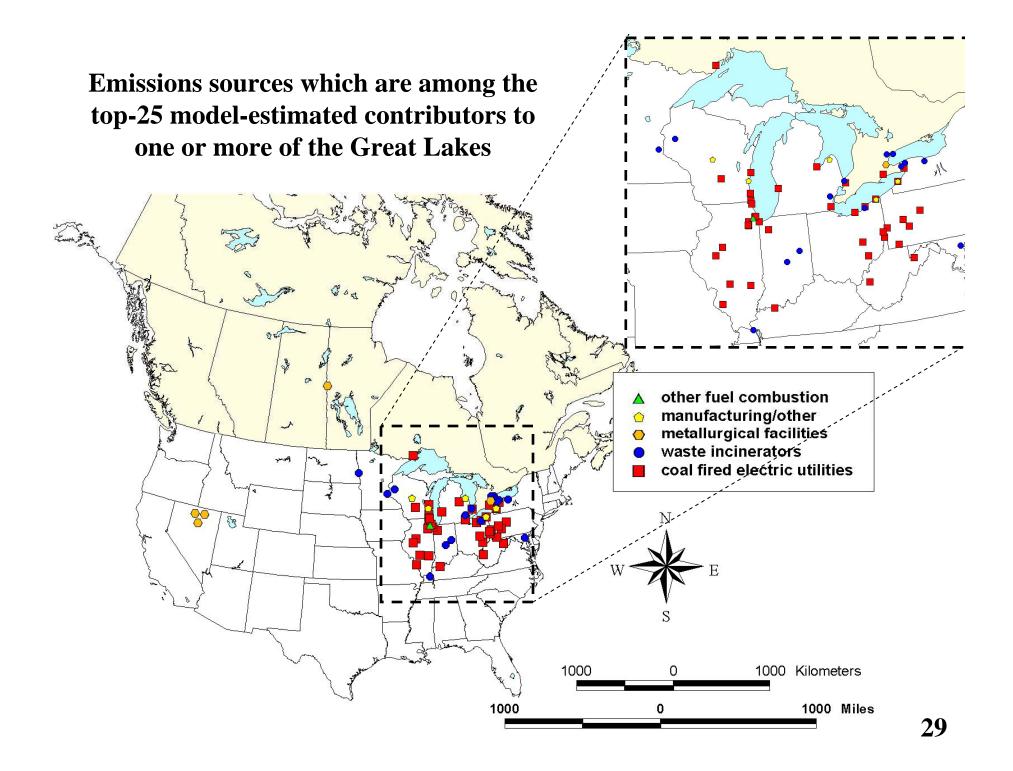
Keywords: Mercury; Atmo

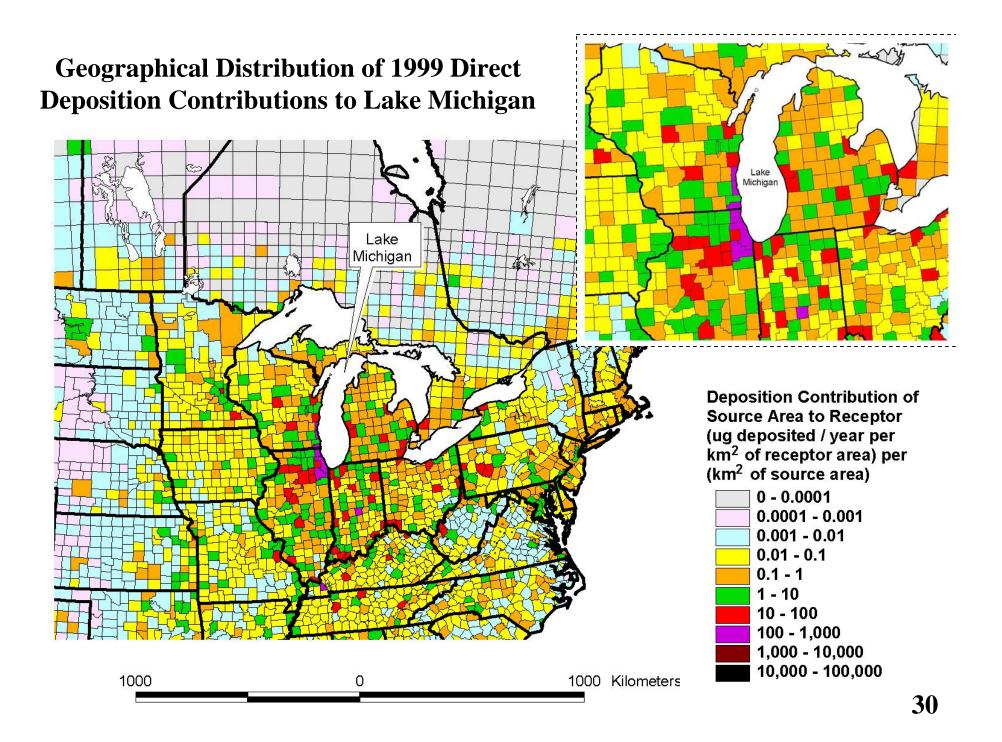
Mercury contaminat other ecosystems is in serious environmental human exposure to m tion, and significant po are believed to be cons levels of mercury (e., 2000). Historical disc production using the m to have caused large

<sup>4</sup> Supplementary data asso the online version, at doi = 1 \*Corresponding author. F *E-mail address:* mark.coh <sup>1</sup> Current address: DPRA Research, Concord, Ontario. Cohen, M., Artz, R., Draxler, R., Miller, P., Poissant, L., Niemi, D., Ratte, D., Deslauriers, M., Duval, R., Laurin, R., Slotnick, J., Nettesheim, T., McDonald, J. "Modeling the Atmospheric Transport and Deposition of Mercury to the Great Lakes." *Environmental Research* 95(3), 247-265, 2004.

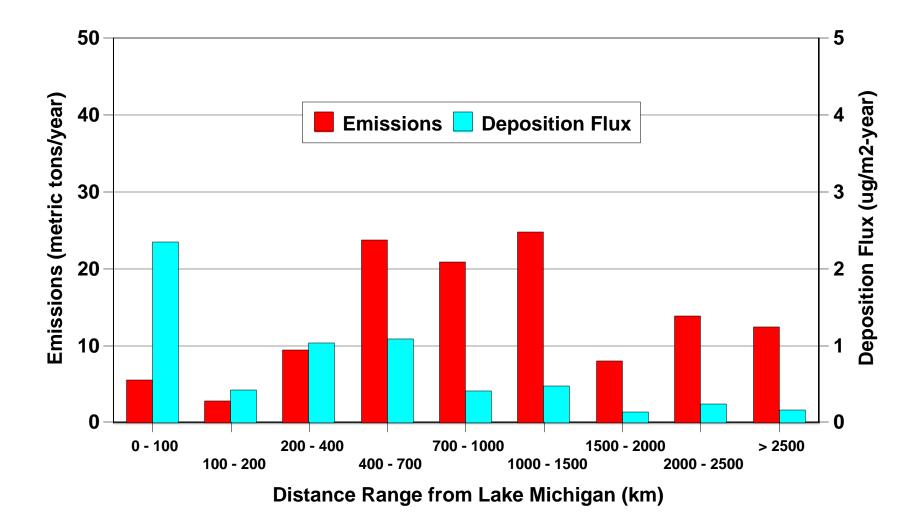
Note: Volume 95(3) is a Special Issue: "An Ecosystem Approach to Health Effects of Mercury in the St. Lawrence Great Lakes", edited by David O. Carpenter.

0013-9351/\$ - see front matter Published by Elsevier Ind doi:10.1016/j.envres.2003.11.007

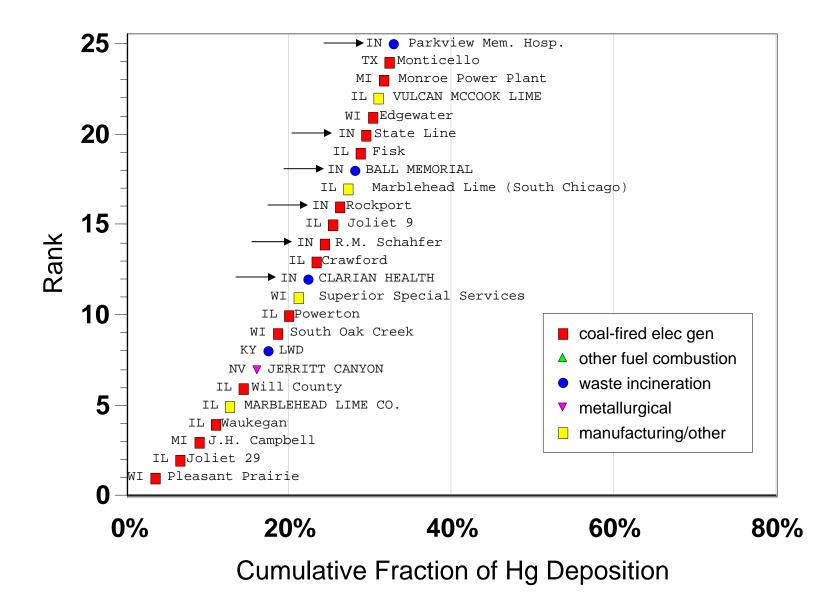




## Emissions and Deposition Contributions from Different Distance Ranges Away From Lake Michigan



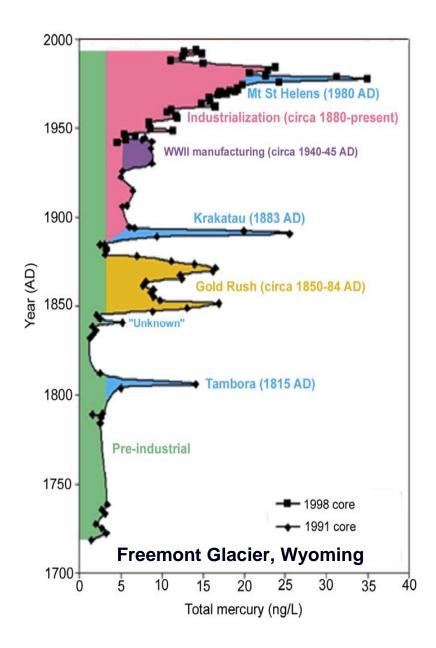




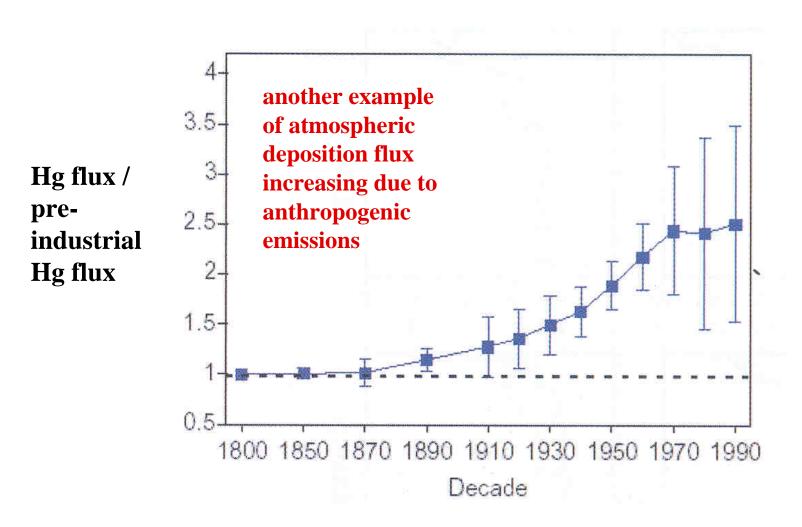
# sourceattribution – the "big picture"

# Natural vs. anthropogenic mercury?

Studies show that anthropogenic activities have typically increased bioavailable Hg concentrations in ecosystems by a factor of 2 – 10



source: USGS, Shuster et al., 2002

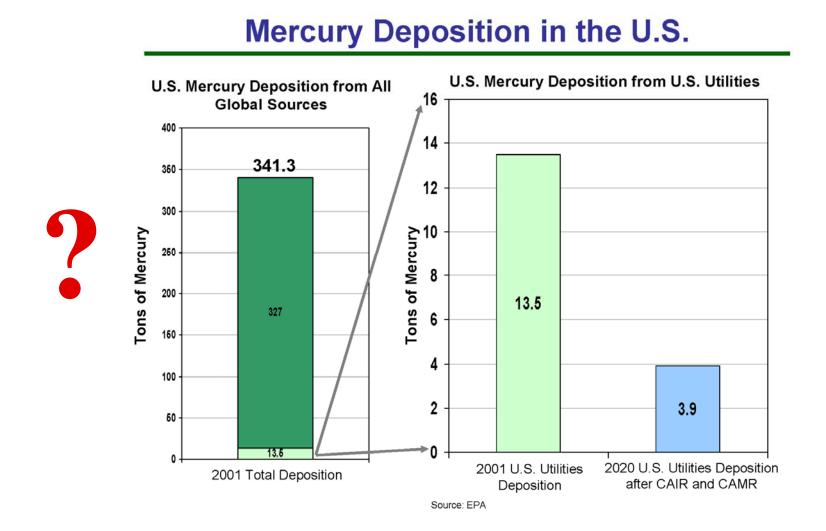


# Average mercury accumulation rate relative to pre-industrial (1800-1850) accumulation rate in five lakes in Northern Alaska (based on sediment cores)

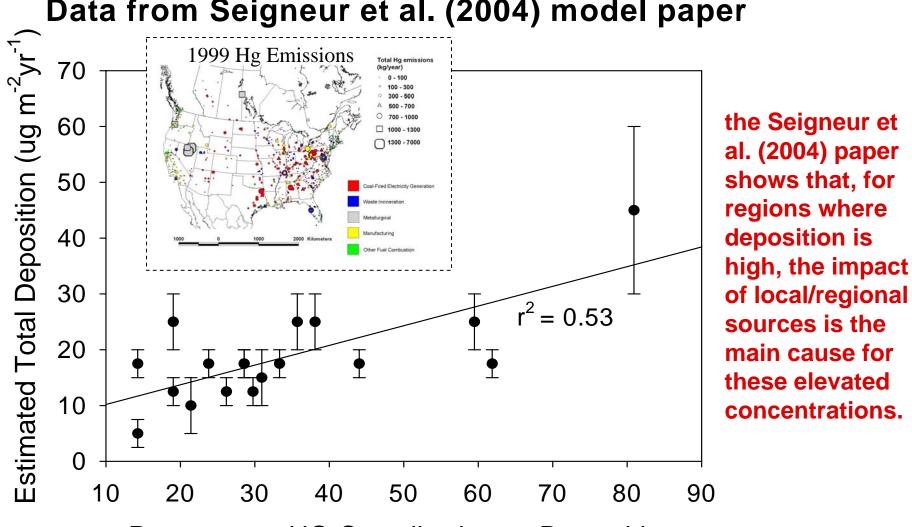
from Fitzgerald et al. (2005), "Modern and Historic Atmospheric Mercury Fluxes in Northern Alaska: Global Sources and Arctic Depletion" *Environ Sci Tech* **39**, 557-568

# What is the relative importance of global, national, regional, and local sources?

Data used by EPA to support recent Clean Air Mercury Rule



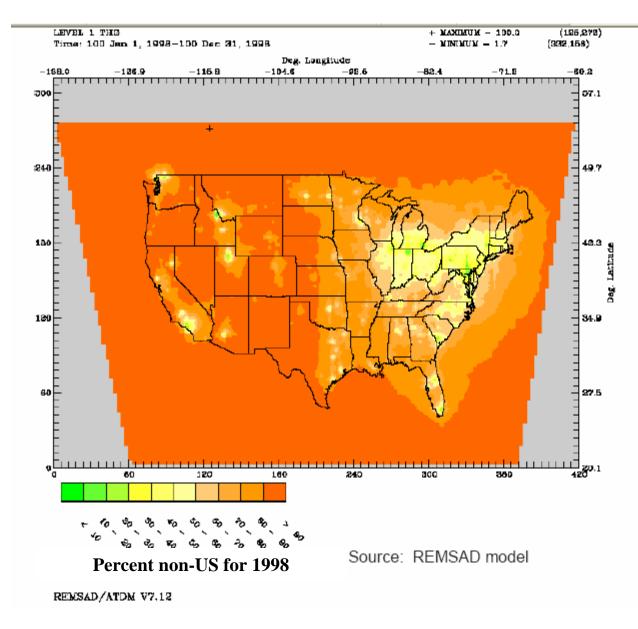
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Data from Seigneur et al. (2004) model paper

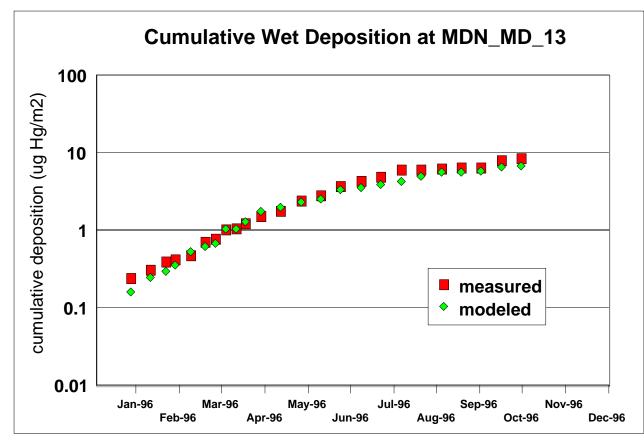
Percentage US Contribution to Deposition [Seigneur et al., (2004), "Global Source Attribution for Mercury Deposition in the United States", ES&T 38, 555-569.]

### Results from the EPA REMSAD Mercury model



- Based on this modeling approximately half of U.S. mercury deposition is from domestic anthropogenic sources and half is from other sources
- Domestic sources dominate deposition for large part of Eastern U.S.
- Global sources are dominant in the Western U.S.

Source: slide developed by Anne Pope for the Hg Roundtable conference call April 21, 2005 Modeled vs. Measured Wet Deposition at Mercury Deposition Network Site MD\_13 during 1996



HYSPLIT modeling has shown that in areas of significant local and regional anthropogenic sources, ambient measurements can be explained reasonably well by considering only these local and regional anthropogenic sources.

Source-apportionment answers depend a lot on <u>where you are</u>

## For areas without large emissions sources

- **the deposition may be relatively low,**
- but what deposition there is may largely come from natural and global sources

## **For areas** with large emissions sources

- **the deposition will be higher**
- and be more strongly influenced by these large emissions sources...

What is the relative importance of global, national, regional, and local sources?

Possible answers are emerging as our understanding improves, but there is no scientific consensus yet...

# Conclusions



Impacts are episodic & depend on form of mercury emitted

Modeling needed to get source-attribution information

(more!) Monitoring needed for model evaluation & refinement



Many uncertainties but useful model results are emerging – these HYSPLIT model results are being extended to include global & natural emissions, and re-emissions



The question of the relative importance of global vs. national vs. regional vs. local sources is complex – the answer depends on *location* and on *what model* one is using...