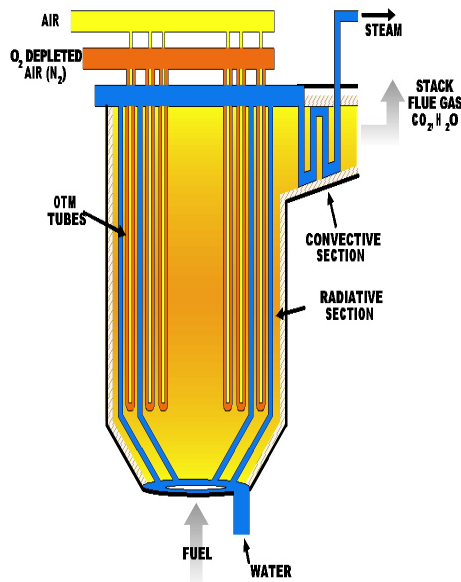


## 3.0 CAPTURING AND SEQUESTERING CARBON DIOXIDE

### 3.1. GEOLOGIC SEQUESTRATION

#### 3.1.1 CO<sub>2</sub> CAPTURE AND SEPARATION

##### Technology Description



### Oxyfuel Technology

Integrates air separation, using O<sub>2</sub> Transport Membrane (OTM) and oxygen combustion

Fossil- and biomass-based energy conversion processes convert hydrocarbon materials (i.e., substances consisting mostly of carbon and hydrogen) into carbon dioxide and water while releasing energy. The goal of CO<sub>2</sub> capture and separation is to produce relatively pure CO<sub>2</sub> from these processes, preferably at pressures suitable for transport, storage, or reuse.

#### System Concepts

- *Post-combustion capture.* A chemical or physical separation process extracts CO<sub>2</sub> from the flue gas of a conventional air-fired combustion process. CO<sub>2</sub> is present in concentrations ranging from 3% to 12%. The focus is on technology for retrofitting or repowering existing power plants and industrial processes.
- *Oxy-fuel combustion.* Pure oxygen rather than air is charged to the combustion chamber, producing a flue gas of CO<sub>2</sub> and water. A portion of the CO<sub>2</sub> is recycled and mixed with the oxygen to absorb heat and control the reaction temperature.
- *Precombustion decarbonization.* The hydrocarbon feedstock is gasified to produce a synthesis gas made up primarily of hydrogen and carbon dioxide. The CO<sub>2</sub> is separated from the hydrogen before it is combusted or charged to a fuel cell.
- There are other advanced-system concepts in which fuel processing and CO<sub>2</sub> capture are integrated into a single stage using, for example, membranes or reduction-oxidation agents.

#### Representative Technologies

- The conventional technology for post-combustion capture (removing CO<sub>2</sub> from flue gas) is amine scrubbing. A solution of amine and water is contacted with flue gas. The amine and the CO<sub>2</sub> undergo a chemical reaction forming a rich amine that is soluble in the water. The rich amine solution is pumped to a desorber where it is heated, reversing the reaction and releasing pure CO<sub>2</sub> gas. The recovered amine is recycled to the flue-gas contactor.
- Other technologies for post-combustion capture include cryogenic distillation, membranes (polymer, ceramic, palladium, mine, or ionic liquid coated), carbon absorbents, sodium absorbents, hydrides, and lithium silicate.

**Technology Status/Applications**

- Amine systems are used in numerous industrial applications to capture CO<sub>2</sub> from flue gas for use as a commodity chemical. Cryogenic and carbon absorbent systems have been built commercially.
- Other post-combustion capture technologies are being developed at the laboratory and pilot scale.

**Current Research, Development, and Demonstration****RD&D Goals**

- The metrics and goals for CO<sub>2</sub> capture research are focused on reducing the cost and energy penalty, because analysis shows that CO<sub>2</sub> capture drives the cost of sequestration systems. Similarly, the goals and metrics for carbon storage and measurement, monitoring, and mitigation (MM&V) are focused on permanence and safety. All three research areas work toward the overarching program goal of 90% CO<sub>2</sub> capture, with 99% storage permanence at less than 20% increase in the cost of energy services by 2007, and less than 10% by 2012.

**RD&D Challenges**

- CO<sub>2</sub> exists in air-combustion flue gas at low concentration, 3-12 volume percent, which makes post-combustion removal very expensive using current approaches.
- Flue gas contains reactive impurities that can adversely affect CO<sub>2</sub> capture systems.
- Transport and/or storage systems may require highly pure CO<sub>2</sub> product, which would increase cost.
- Loss of CO<sub>2</sub> temperature and pressure across the capture system.
- Wide-scale application of any one capture technology will be difficult, due to real estate constraints, availability of pollutant-control equipment, resource limitations, and economic and load demographics.
- Significant cost associated with compressing CO<sub>2</sub> for transportation and storage.

**RD&D Activities**

- Laboratory-scale experiments with advanced amines, ceramic membranes, high-temperature polymer membranes, vortex gas/liquid separator, ammonium and sodium bicarbonate, carbon absorbents, ionic liquids, aqueous ammonia, and electrochemical pumps.
- Pilot-scale tests with a novel oxy-fuel boiler, a CO<sub>2</sub>/water hydrate process, a sodium-based CO<sub>2</sub> sorbent, aqueous K<sub>2</sub>CO<sub>2</sub> promoted by piperazine, an oxygen-fired circulating fluidized bed, and a metal reduction-oxidation power generation process.

**Recent Progress**

- During a short three-year period, a strong portfolio of research projects for the existing and future power-generation fleet has been developed by DOE with more than 40% private-sector cost-share.
- The international community has been successfully engaged through participation in the International Energy Association Greenhouse Gas Programme, the CO<sub>2</sub> Capture Project with the European Commission and other international participants, and other collaborations with Canada, Australia, and Japan.
- Analysis by the Carbon Sequestration Regional Partnership on the likely and probable application of various technologies to different point sources has been conducted and will continue in Phase II (utility and nonutility).

**Commercialization and Deployment Activities**

- Roughly 15 Mt/yr of CO<sub>2</sub> is captured from anthropogenic emissions sources in the United States and used as a commodity chemical.

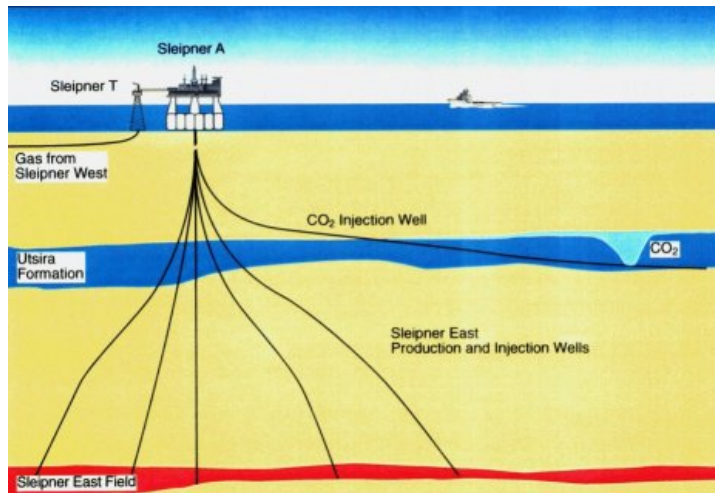
**Market Context**

- Development of approaches for economically decarbonizing fossil fuels will allow the carbon-free production of electricity and hydrogen, and will take advantage of an existing fossil fuel infrastructure that accounts for more than 80% of the energy consumed in the United States and internationally.

## 3.1.2 CO<sub>2</sub> STORAGE IN GEOLOGIC FORMATIONS

### Technology Description

#### Sleipner North Sea Project



Large amounts of CO<sub>2</sub> (about a billion tons per year) may need to be stored as a part of a future global atmospheric stabilization strategy. CO<sub>2</sub> can be injected into depleting oil fields, gas fields, and unmineable coal-bearing formations to enhance resource recovery. A portion of the CO<sub>2</sub> remains underground, although current industry practices are geared strongly toward minimizing the CO<sub>2</sub> left underground – and little or no attention is paid to the CO<sub>2</sub> that is not recovered. R&D is focused on revamping conventional enhanced oil recovery, gas recovery, and enhanced coalbed methane processes so that they can serve a dual purpose: resource recovery and CO<sub>2</sub> storage. Other high permeability formations filled with brine, organic-rich shale beds, and other nonconventional geologic structures have potentially enormous CO<sub>2</sub> storage capacities. Research is focused on learning more about these formations and developing the capabilities needed to use them as CO<sub>2</sub> repositories.

#### System Concepts

- CO<sub>2</sub> is captured from a large anthropogenic point source, and transported and injected into a depleting oil field, unmineable coal seam, saline formation, depleting gas field, shale, or other geologic structure amenable to CO<sub>2</sub> storage. There are five different mechanisms that can trap CO<sub>2</sub> in geologic formations:
  - Structural trapping: A layer or “cap” of impermeable rock that overlies the formation of porous rock into which the CO<sub>2</sub> is injected prevents upward flow of CO<sub>2</sub>. This is the mechanism that caused natural deposits of crude oil, natural gas, and CO<sub>2</sub>.
  - Capillary trapping: The surface of sandstone and other rocks preferentially adheres to saline water versus CO<sub>2</sub>. If there is enough saline water within a pore (75-90% of the pore volume), it will form a capillary plug that traps the residual CO<sub>2</sub> within the pore space.
  - Dissolution in saline water: CO<sub>2</sub> is soluble in saline water, and will dissolve in solution on contact.
  - Mineralization: Over longer periods of time, dissolved CO<sub>2</sub> can react with minerals in the formation to form solid carbonates. There may be ways to enhance this reaction.
  - Adsorption: Coal and other organically rich formations will preferentially absorb CO<sub>2</sub> onto carbon surface as a function of reservoir pressure. In some cases, such as coal beds, CO<sub>2</sub> displaces methane, which can be recovered to enhance economics.
- In an oil field, the CO<sub>2</sub> displaces the oil in place and also dissolves in the oil, decreasing the oil viscosity and enabling more of it to be recovered. A portion of the injected CO<sub>2</sub> remains stored in a reservoir as a free gas, in brine or oil solution, or in carbonate minerals.
- Leakage of sequestered CO<sub>2</sub> back to the surface may occur through faults, active or abandoned wells, and microseepage.

**Representative Technologies**

- Natural gas storage fields provide experience of injecting significant quantities of gas into geologic formations.
- Technologies will borrow extensively from the petroleum industry in the areas of drilling simulation; completion of injection wells; processing, compression, and pipeline transport of gases, including acid gases; operational experience of CO<sub>2</sub> injection for enhanced oil recovery and natural gas storage; and subsurface reservoir engineering and characterization.
- Enhanced coal bed methane recovery using nitrogen.

**Technology Status/Applications**

- The Mount Simon reservoir underlying Illinois, Indiana, Michigan, Kentucky, and Pennsylvania has been approved for industrial waste disposal and underlies a region with numerous fossil energy power plants.
- Industry has experience with more than 400 wells for injecting industrial wastes into saline formations.
- The petroleum technology is readily adaptable to subsurface CO<sub>2</sub> storage.

**Current Research, Development, and Demonstration****RD&D Goals**

- Develop domestic CO<sub>2</sub> underground storage repositories capable of accepting around a billion tons of CO<sub>2</sub> per year.
- Demonstrate that CO<sub>2</sub> storage underground is safe and environmentally acceptable, and an acceptable GHG mitigation approach.
- Demonstrate an effective business model for CO<sub>2</sub> enhanced oil recovery and enhanced coalbed methane, where significantly more CO<sub>2</sub> is permanently stored than under current practices.
- Develop cost-effective methods to survey large land areas and locate zones of potential CO<sub>2</sub> leakage.
- Provide monitoring techniques that can reliably evaluate the stability, capacity, rate of leakage, and permanence of carbon dioxide stored in geologic formations.
- Develop publicly accepted monitoring protocols.

**RD&D Challenges**

- Develop the capability to inject CO<sub>2</sub> into saline formations with low permeability.
- Harness geochemical reactions to enhance containment.
- Develop injection practices that preserve cap integrity.
- Develop an understanding of the properties of shales and other unconventional hydrocarbon-bearing formations that determine how they will react to CO<sub>2</sub> injection.
- Develop the ability to track CO<sub>2</sub> transport.
- Develop field practices that optimize CO<sub>2</sub> storage and resource recovery.
- Develop the ability to predict the CO<sub>2</sub> storage capacity and potential resource recovery of a particular formation.
- Develop models that are able to simulate the fate and transport of CO<sub>2</sub> in geologic formations and along potential migration pathways.
- Develop the ability to track the fate and transport of injected CO<sub>2</sub>.
- Develop methods to locate well bores, with and without casing, that might potentially leak.
- Develop surface and near-surface monitoring technologies that will allow public demonstration of the safety of CO<sub>2</sub> storage.
- Develop a better understanding of the chemistry of coal and CO<sub>2</sub>, and conduct comprehensive R&D program on all physical and chemical aspects of CO<sub>2</sub> interactions with reservoir phases.

**RD&D Activities**

- Study geochemical reactions involving CO<sub>2</sub> in a laboratory.
- Study the natural analogs of geochemical CO<sub>2</sub> conversion. Study rock samples from CO<sub>2</sub> bearing geologic formations to better understand in situ geochemical/geobiological reactions.
- Develop CO<sub>2</sub> tracking technology, e.g., sonic, chemical tracers.
- Study CO<sub>2</sub> transport in the Sleipner Vest gas field, via the International Energy Agency's Greenhouse Gas Programme.

- Assessment of techniques for finding abandoned wells near a potential sequestration site.
- Develop models to simulate migration of CO<sub>2</sub> through multiple subsurface formations.
- Novel injection techniques to increase CO<sub>2</sub> storage in saline formations.
- CO<sub>2</sub> storage in coal beds. Laboratory measurements of CO<sub>2</sub>/CH<sub>4</sub> sorption and coal swelling under confined and unconfined conditions, ARI and industry consortium, commercial-scale field demonstration in the San Juan Basin; Consol – horizontal drilling, Alabama geologic survey, screening model for Black Warrior.
- Apply surface and near-surface monitoring techniques such as surface CO<sub>2</sub> flux, injection tracers in soil-gas, and changes in shallow aquifer chemistry for CO<sub>2</sub> leakage.
- Study chemical reactions involving CO<sub>2</sub> and cement from new or existing wells as a possible source of leakage.
- CO<sub>2</sub> storage in oil reservoirs. Weyburn, reservoir mapping, West Pearl Queen, CO<sub>2</sub> monitoring and simulation.
- Airborne reconnaissance of a 38,000-acre EOR/sequestration site to locate wells, faults, and other potential CO<sub>2</sub> leakage zones.
- Other RD&D activities in DOE, Australia, the European Union, Japan, etc.

#### **Recent Progress**

- Major saline formations underlying the United States have been identified.
- Initiated a pilot-scale test of CO<sub>2</sub> storage in a depleted oil reservoir.
- Completed pilot-scale injection of CO<sub>2</sub> into the Frio deep saline aquifer and field tested near-surface monitoring techniques.
- Initiated several field tests with key industrial companies participating and providing cost-share: Consol Inc. CBM,-Appalachia ARI, CBM-San Juan Basin; Strata Production C. – Permian Basin; Pan Canadian Resources EOR-Canada.

#### **Commercialization and Deployment Activities**

- Since 1999, Statoil has been injecting CO<sub>2</sub> at a rate of 1 Mt/yr into the Sleipner Vest gas field in a sandstone aquifer 1,000m beneath the North Sea.
- About 70 oil fields worldwide use CO<sub>2</sub> for enhanced oil recovery.
- Another project uses CO<sub>2</sub> from Dakota Gasification for enhanced oil recovery in the Weyburn field in Canada. CO<sub>2</sub> is transported via pipeline.
- The pipeline enables extensive use of CO<sub>2</sub> for enhanced coal bed methane recovery in the San Juan basin.
- There are plans for using CO<sub>2</sub> for enhanced oil recovery in Kansas, using CO<sub>2</sub> from ethanol production.
- Planned test in Kansas using landfill gas for enhanced coal bed methane recovery.

#### **Market Context**

- Development of approaches for economically decarbonizing fossil fuels will allow the carbon-free production of electricity and hydrogen, and will take advantage of an existing fossil fuel infrastructure that accounts for more than 80% of the energy consumed in the United States and internationally.

### 3.1.3 NOVEL SEQUESTRATION SYSTEMS

<b>Technology Description</b>
<p>In the long term, CO<sub>2</sub> capture can be integrated with geologic storage and/or conversion. Many CO<sub>2</sub> conversion reactions are attractive but too slow for economic chemical processes.</p> <p><b>System Concepts</b></p> <ul style="list-style-type: none"> <li>• Using impurities in captured CO<sub>2</sub> (e.g., SO<sub>x</sub>, NO<sub>x</sub>) or additives enhances geologic storage. This is a possible opportunity to combine CO<sub>2</sub> emissions reduction and criteria pollutant-emissions reduction.</li> <li>• Conducting reactions on CO<sub>2</sub> while it is being stored underground can alleviate the problem with slow kinetics.</li> <li>• In situ mineralization is an important trapping mechanism, that there may be ways to enhance it.</li> <li>• Rejected heat from electricity generation and CO<sub>2</sub> compression can help drive CO<sub>2</sub> conversion process.</li> <li>• Air capture of CO<sub>2</sub>.</li> </ul> <p><b>Representative Technologies</b></p> <ul style="list-style-type: none"> <li>• Capture of CO<sub>2</sub> from flue gas and algal conversion to biomass.</li> <li>• Capture of CO<sub>2</sub>, storage in a geologic formation, and in situ biological conversion to methane.</li> </ul> <p><b>Technology Status/Applications</b></p> <ul style="list-style-type: none"> <li>• Conceptual.</li> </ul>
<b>Current Research, Development, and Demonstration</b>
<p><b>RD&amp;D Goals</b></p> <ul style="list-style-type: none"> <li>• Demonstrate viable chemical or biological conversion approaches at the laboratory scale.</li> <li>• Develop robust conceptual designs for integrated capture, storage, and conversion systems.</li> </ul> <p><b>RD&amp;D Challenges</b></p> <ul style="list-style-type: none"> <li>• CO<sub>2</sub> conversion reaction kinetics are slow, energy requirements are high.</li> <li>• For biological in situ CO<sub>2</sub> conversion, must provide food and remove waste.</li> <li>• Truly novel concepts may be required to meet the ultimate “stretch” goals of the program. Technology breakthroughs could come from concepts associated with areas not normally related to traditional energy technologies (e.g., nanotechnology). Tapping areas where current researchers do not have an energy mindset will require new approaches for soliciting proposals for R&amp;D projects.</li> </ul> <p><b>RD&amp;D Activities</b></p> <ul style="list-style-type: none"> <li>• Laboratory and pilot-scale experiments with biological and chemical conversion.</li> <li>• Conceptual studies of integrated systems and in situ CO<sub>2</sub> conversion.</li> </ul>
<b>Recent Success</b>
<ul style="list-style-type: none"> <li>• Several cost-shared research projects have been initiated.</li> </ul>
<b>Commercialization and Deployment Activities</b>
<ul style="list-style-type: none"> <li>• None.</li> </ul>