

## QUALITY ASSURANCE PROJECT PLAN (QAPjP) And QA Report for Pacific 2001

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**2. Team Members**

Corinne Schiller – York University  
Jean Francois Gouin – York University

**3. Measurement Program**

Tropospheric gas phase ammonia, nitrous acid and/or hydrogen peroxide

**4. Measurement Species and Units**

NH<sub>3</sub> (ammonia) – ppbv  
HONO (nitrous acid) – ppbv  
H<sub>2</sub>O<sub>2</sub> (hydrogen peroxide – ppbv

**5. Representative Size Range (if PM)**

N/A

**6. Measurement Platform (surface, airborne)**

Tunable Diode Laser Spectroscopy ~ 5 m above ground on trailer

**7. Measurement Sites (surface only)**

Langley

**8. Measurement Objective(s)**

Tropospheric gas phase Ammonia, nitrous acid and/or hydrogen peroxide using tunable diode laser absorption spectroscopy

**9. Measurement Details**

**9.1. Field Measurements**

**9.1.1. Measurement Principle**

Tunable Diode Laser Spectroscopy  
HONO ~1280 cm<sup>-1</sup>  
NH<sub>3</sub> ~ 950 cm<sup>-1</sup>  
H<sub>2</sub>O<sub>2</sub> ~ 1280 cm<sup>-1</sup>

**9.1.2. Instrumentation (Manufacturer/Model)**

The tunable diode laser spectrometer was originally a TAMS 150 built by Unisearch, which has since been modified by Corinne Schiller and Jean Francois Gouin at York University. The White cell incorporates 2 vertical corner cubes, resulting in 8 rows and thereby doubling the path length.

The White cell mirrors have been recoated with super silver resulting in a reflectance greater than 99.8%. The program has been completely rewritten in Labview with a Real Time data acquisition board. The electronics have incorporated a Profile temperature/laser controller as well as new GE lock in amplifiers.

**9.1.3. Flow System**

Air will be sampled from a manifold starting approximately 2 m above the trailer. The manifold has an internal diameter of approximately 2 inches. The air will be drawn down the manifold at a flow rate of approximately 50 l/min. The flow through the manifold should be laminar with a Reynolds number of approximately 1350. From the center of this flow, ~ 10 l/min will be sampled into the White cell through a glass inlet with aperture. A short length of Teflon tubing will be located on the low-pressure side to relieve tension.

**9.1.4. Inlet Height Above Ground (if surface)**

~ 2 meters above trailer therefore about 5 meters above the ground

**9.1.5. Nominal Flow Rate**

Flow down manifold ~ 50 l/min

Flow into White Cell ~ 10 l/min

**9.1.6. Flow Measurement/Control**

The flow through the manifold will be controlled by a 50 l/min mass flow controller, while the flow through the white cell will be controlled by an aperture in the glass inlet. The zero air for calibrations and backgrounds will be controlled with a 20 l/min flow controller. The flow through the permeation devices will be controlled by rotometers and the flow of HCl for HONO calibration will be controlled with a 50 ml/min flow controller.

**9.1.7. Flow Temperature and Pressure**

The flow controllers will be calibrated with a Gillian Flow measurement system to STP.

**9.1.8. Sampling Times/Period/Frequency**

The tunable diode laser will operate continuously, obtaining 1 minute averages of 2 species. Backgrounds will be taken every 3 minutes and calibrations once per hour

**9.1.9. Sampling Methods**

N/A

**9.1.10. Filter Type/Coating Type/Reagent Type**

N/A

**9.1.11. Planned Changes to Instruments or Methods During Study**

N/A

**9.2. Laboratory Measurements (If Applicable)**

**9.2.1. Laboratory Name and Address**

York University – Geoff Harris Lab

**9.2.2. Analytical Method(s)**

For periodic verification of our calibration standards

Nitrous Acid: Bubble gas sample through  $10^{-5}$  NaOH solution

Analyze using Ion Chromatography

Ammonia: Bubble gas sample through  $10^{-5}$  HCl solution

Analyze using Ion Chromatography

H<sub>2</sub>O<sub>2</sub> Bubble solution through titanium tetrachloride

Measure color change with a Spec 20.

**9.2.3. Sample Extraction or Work-up**

N/A

**9.2.4. Analytical Detection Limits**

Laboratory

NO<sub>2</sub><sup>-</sup> by IC = <1 nmoles/ml (do not require better)

NH<sub>4</sub><sup>+</sup> by IC = <1 nmoles/ml (do not require better)

H<sub>2</sub>O<sub>2</sub> by titanium tetrachloride =  $4 \times 10^{-5}$  g (~ 2% absorbance by spectrometer)

**10. Quality Assurance/Quality Control**

**10.1. Field Quality Assurance/Quality Control**

**10.1.1. Traceability**

Nitrite –volumetric dilution of a primary standard dried NaNO<sub>2</sub> (purity >99.99%) – Sigma Aldrich

Ammonium – volumetric dilution of a primary standard of ammonium sulfate – purity >99.99% - Sigma Aldrich H<sub>2</sub>O<sub>2</sub> - Quantitative reaction with titanium tetrachloride resulting in a colored product whose extinction coefficient has been determined quantitatively. [W. Pilz, I. Johann; J. Environ. Chem., 3, 257, 1974. – Extinction Coefficient of 735 M<sup>-1</sup> cm<sup>-1</sup>]

**10.1.2. Calibration**

Calibration of the TDLAS will be carried out by the addition of the molecule of interest from a permeation device being diluted into zero air very near where the air is drawn off the manifold.

Calibration of the permeation devices has been described in section 10.2.2

**10.1.3. Zeros and spans**

Zero air will be used for the zero on the TDLAS

**10.1.4. Blanks**

Verifying calibration concentrations will result in 2 blanks used to determine possible contamination of the samples with shipping.

**10.1.5. Field Quality Control procedures**

Calibration of permeation devices will be sampled into pre-pipetted sealed vials.

**10.1.6. Precision determination**

Precision will be determined by the variability of subsequent measurements of known concentrations of calibration gas in zero air introduced to the instrument.

**10.1.7. Comparison with other measurements**

No other ammonia measurements will be made at the Langley site. There may be a chemical method for HONO measurement by the AES group. No other measurements of H<sub>2</sub>O<sub>2</sub> are expected to be made at the Langley site.

**10.1.8. Inspections and Audits**

No external audits will be carried out.

**10.2. Laboratory Quality Assurance/Quality Control**

**10.2.1. Traceability**

Nitrite –volumetric dilution of a primary standard dried NaNO<sub>2</sub> (purity >99.99%) – Sigma Aldrich  
Ammonium Chloride – volumetric dilution of a primary standard – purity >99.999% - sigma Aldrich  
H<sub>2</sub>O<sub>2</sub> a spectroscopic method – (W. Pilz, I. Johann; J. Environ. Chem., 3, 257, 1974. – Extinction Coefficient of 735 M<sup>-1</sup> cm<sup>-1</sup>)

**10.2.2. Calibration procedures**

A 5 - point calibration in duplicate will be run for every 20 samples for both anions and cations measured by IC

**10.2.3. Blanks**

2 types of blanks will be run  
Calibration blank – run from the same water that the calibration standards were made with  
Collection Blank – a blank of the material samples were collected in will be returned from the field and analyzed the same as a sample

**10.2.4. Other lab QC**

Latex gloves will be worn while for standard generation. Weights of dried primary standards will be determined with a calibrated analytical balance with an accuracy of 0.0001 g. New volumetric glassware that has never

been dried will be used for calibration standards. Disposable tip pipettes will be used to measure 15 ml into sterile clean disposable vials for collection in the field.

**10.2.5. Precision determination**

Standards and samples will be run in triplicate to determine reproducibility and minimize error

**10.2.6. Comparison with other methods**

IC determinations of nitrite and ammonia are standard procedures. Determination of hydrogen peroxide by reaction with titanium tetrachloride is the accepted technique for low concentration determination of hydrogen peroxide. [Reference]

**10.2.7. Audits**

No external audits will be carried out

**11. Data Management and Quality Control**

**11.1. Raw Data Recording**

Raw 1 - minute spectra from the TDLAS will be recorded on computer as well as a calculated concentration from a fit with the previous calibration.

**11.2. Final Data Reporting**

1 minute data will be recorded, but data will be presented as either 10, 30 or 60 minute data

**11.3. Data Quality Control and Validation**

Only Valid Data will be reported.

**11.4. Validity Flags**

The following NARSTO flags will be used:

- V0 Valid value
- V1 Valid value but comprised wholly or partially of below-MDL data
- M1 Missing value because no value is available
- M2 Missing value because invalidated by data originator

**11.5. Below Method Detection Limit Values**

A macro will be written to flag values that are below the determined detection limit

**11.6. Derived Parameters**

N/A

**11.7. Explanation of Zero or Negative Data**

Data below the detection limit will be flagged with V1

## **12. Data Quality Objectives (Pre-Study)**

### **12.1. Accuracy**

The accuracy has yet to be determined. Accuracies of about 2% should be attainable.

### **12.2. Precision**

Precisions of better than 5% are expected

### **12.3. Comparability**

Since we do not expect very well mixed air masses for HONO and ammonia, there may be some spatial variability associated with these measurements, however this variability should be within ~10%

### **12.4. Representativeness**

The measurements at the Langley site will be representative of processed air pollution in which secondary pollutants, such as hydrogen peroxide, will have formed. We are close to the source region for ammonia, so there may be large variability in this point source measurement. HONO is produced on black carbon particles in the presence of NO<sub>2</sub> and water, and is lost through both wet and dry deposition; therefore the concentration is likely to relate to local conditions only.

### **12.5. Completeness**

Following the initial start up period, we expect between 22 and 23 hours per day coverage.

### **12.6. Other Quality Information**

Unaware of any qualifiers at this time

End of Pre-Study QAPjP

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Start of Post-Study QA Report

13. Significant Changes to Site, Instruments or Methods During Study

14. Post-study Data Quality Indicators (DQIs)

14.1.1. Accuracy

14.1.2. Precision

14.1.3. Comparability

14.1.4. Representativeness

14.1.5. Completeness

14.2. Blank correction (describe whether done and method used):

14.3. Other Quality Information

15. References: