

VII. RESEARCH NEEDS

The result of the release of boron trifluoride into an environment of normal humidity is the immediate production of a white mist. The compounds present in the mist have not been identified, but they are assumed to be the products of boron trifluoride hydration. Accurate identification of the actual compounds present is essential to the determination of boron trifluoride toxicity. In addition, the relationships among aerosol size, relative humidity, and chemical composition must be determined.

No adequate method to sample and analyze the environment for boron trifluoride gas and mist has been developed. Most attempted methods are cumbersome. However, some of the methods (fluoborate-ion-specific electrode and atomic absorption spectrophotometry) are potentially useful, and, therefore, an intensive program to verify a sampling and analysis method should be undertaken.

Torkelson et al [22] noted an increase in the fluoride levels in bones and teeth of rats exposed to boron trifluoride. There is some evidence that boron-fluoride compounds may be excreted more rapidly than they can be deposited. [72,73] The evidence suggests that boron trifluoride has less potential for deposition than the fluoride ion. Confirmation from new research is needed. At the same time, the possibility of boron deposition resulting from boron trifluoride exposures should be evaluated.

Animal inhalation studies have established the effects of boron trifluoride on the respiratory systems of several species. However, the animal data are not sufficient; long term tests are needed to properly

evaluate the effects of chronic exposure. Populations occupationally exposed to boron trifluoride have not been studied extensively. Epidemiologic surveys are needed to evaluate dose-response relationships for this compound. Additional testing is needed to investigate any carcinogenic, mutagenic, or teratogenic potential of boron trifluoride.

VIII. REFERENCES

1. Compressed Gas Association, Inc: Handbook of Compressed Gases. New York, Van Nostrand Reinhold Co, 1966, 398 pp
2. Martin DR: Boron trifluoride, in Kirk-Othmer Encyclopedia of Chemical Technology, ed 2. New York, Interscience Publishers, 1966, vol 9, pp 554-62
3. Clinton M: API Toxicological Review--Boron Trifluoride. New York, American Petroleum Institute, Dept of Safety, 1948, 3 pp
4. Booth HS, Martin DR: Boron Trifluoride and Its Derivatives. New York, John Wiley and Sons Inc, 1949, 315 pp
5. Plant observation reports and evaluation. Menlo Park, Calif, Stanford Research Institute, January 1976, 326 pp, (submitted to NIOSH under contract No. CDC-99-74-31)
6. Boron Trifluoride Monoethylamine Complex Technical BF₃C₂H₅NH₂, product information data sheet No. DA-34663. Morristown, NJ, Allied Chemical Corp, General Chemical Division, Baker and Adamson, 1958, 2 pp
7. Boron Trifluoride-Ethyl Ether Complex, product information technical bulletin BFTB-3. Morristown, NJ, Allied Chemical Corp, Specialty Chemicals Division, 1971, 4 pp
8. Boron Trifluoride Compressed Gas, product technical bulletin BFTB-1. Morristown, NJ, Allied Chemical Corp, Specialty Chemicals Division, 1971, 10 pp
9. Boron Fluoride Complexes, technical bulletin No. 4/73. Cleveland, Kewanee Oil Co, Harshaw Chemical Co, 1975, 2 pp
10. Boron Fluoride Phosphoric Acid (100%), bulletin No. 11375. Cleveland, Kewanee Oil Co, Harshaw Chemical Co, 1975, 1 p
11. Boron Fluoride Dihydrate, bulletin No. 11175. Cleveland, Kewanee Oil Co, Harshaw Chemical Co, 1975, 1 p
12. Boron Fluoride Monoethylamine Complex, bulletin No. 62475. Cleveland, Kewanee Oil Co, Harshaw Chemical Co, 1975, 1 p
13. Booth HS, Martin DR: Systems with boron trifluoride. J Am Chem Soc 64:2198-205, 1942
14. Greenwood NN, Martin RL: Boron trifluoride co-ordination compounds. Q Rev 8:1-39, 1954

15. Halbedel HS: Fluoboric acid and the fluoborates, in Kirk-Othmer Encyclopedia of Chemical Technology, ed 2. New York, Interscience Publishers, 1966, vol 9, pp 562-72
16. Bixby W, Almenas K: Measurements of neutron fluxes in strongly absorbing gaseous media. Nucl Technol 23:213-21, 1974
17. Sampson TE, Vincent DH: An absolute measurement of the efficiency of a BF₃ proportional counter. Nucl Instrum Methods 95:563-69, 1971
18. Miller GT, Kralik RJ, Belmore EA, Drury JS: Production of boron-10, in Proceedings of the 2nd United Nations International Conference on the Peaceful Uses of Atomic Energy, Geneva, September 1-13, 1958, pp 585-94
19. Marcovitch S: Volatile fluorine compounds for the control of insects. J Econ Entomol 35:288-89, 1942
20. Wamser CA: Equilibria in the system boron trifluoride-water at 25 [C]. J Am Chem Soc 73:409-16, 1951
21. Halbedel HS: Acid Fluorides and Safety, technical bulletin AF and S 673. Cleveland, Kewanee Oil Co, Harshaw Chemical Co, 11 pp
22. Torkelson TR, Sadek SE, Rowe VK: The toxicity of boron trifluoride when inhaled by laboratory animals. Am Ind Hyg Assoc J 22:263-70, 1961
23. Milby TH, Key MM, Gibson RL, Stokinger HE: Boron compounds, in Gafafer WM (ed): Occupational Diseases--A Guide to Their Recognition, Publication No. 1097. US Dept of Health, Education, and Welfare, Public Health Service, 1964, pp 95-97
24. Thenard LJ, Gay-Lussac JL: [Memorandum on fluoric acid.] Ann Chim (Paris) 69:204-20, 1809 (Fre)
25. Boron Trifluoride, in Stecher PG (ed): The Merck Index--An Encyclopedia of Chemicals and Drugs, ed 8. Rahway, NJ, Merck and Co Inc, 1968, pp 161-62
26. Mellor JW: Treatise on Inorganic and Theoretical Chemistry. London, Longmans, Green and Co, 1924, vol 5, p 122
27. Kirii VG: The health of workers producing polyisobutylene by the use of a boron fluoride catalyst. Biol Abstr 49:4068, 1966
28. Boron Trifluoride--Information Concerning Industrial Hygiene Practices, Allied Chemical Corp, Specialty Chemicals Division, Buffalo, NY, 1975, 109 pp
29. Spiegl CJ: Part A. Inhalation-toxicity studies of boron halides and certain fluorinated hydrocarbons, in Voegtlin C, Hodge HC (eds):

Pharmacology and Toxicology of Uranium Compounds. New York, McGraw-Hill Book Co Inc, 1953, pp 2291-321 (National Nuclear Energy series--Manhattan Project Technical section Division VI--Vol 1)

30. Weil CS, Horton C, Laush G: Tribnol--The Second 30-Day Toxicological Study at a Metered Concentration of 100 Parts Per Million--Toxic Effects of Inhalation by Laboratory Animals, report No. 359. Rochester, NY, University of Rochester, Division of Pharmacology, Inhalation Section, 1945, 60 pp
31. Weil CS, Horton C, Wilson H: Boron Trifluoride BF₃--A 30-Day Toxicological Study at a Metered Concentration of 15 Parts Per Million, report No. 409. Rochester, NY, University of Rochester, Division of Pharmacology, Inhalation Section, 1945, 73 pp
32. Horton CA, Weil CS: Part C. Determination of boron halides in air, in Voegtlin C, Hodge HC (eds): Pharmacology and Toxicology of Uranium Compounds. New York, McGraw-Hill Book Co Inc, 1953, pp 2328-34 (National Nuclear Energy series--Manhattan Project Technical section Division VI--Vol 1)
33. Kasparov AA, Kirii VG: [Toxicity of boron fluoride.] Farmakol Toksikol (Moscow) 35:369-72, 1972 (Rus)
34. National Institute for Occupational Safety and Health: Criteria for a Recommended Standard...Occupational Exposure to Inorganic Fluorides, HEW Publication No. (NIOSH) 76-103. Rockville, Md, US Dept of Health, Education, and Welfare, Public Health Service, Center for Disease Control, NIOSH, 1975, 191 pp
35. Kasparov AA, Kirii VG: [The problem of methods of determination of boron trifluoride in air.] Gig Sanit 37:57-59, 1972 (Rus)
36. Salyamon GS, Maslyukova AI: [Determination of microgram amounts of boron trifluoride and its hydrolysates in air.] Gig Sanit 36:65-67, 1971 (Rus)
37. Weil CS, Horton C, Wilson H: Boron Trifluoride Methyl Ether Complex--A 32-Day Toxicological Study at a Metered Concentration of 20 Parts Per Million, report No. 414. Rochester, NY, University of Rochester, Division of Pharmacology, Inhalation Section, 1945, 72 pp
38. Weil CS, Horton C, Laush G: 890--A 30-Day Toxicological Study at a Metered Concentration of 50 Parts Per Million--Toxic Effects of Inhalation by Laboratory Animals, report No. 325. Rochester, NY, University of Rochester, Division of Pharmacology, Inhalation Section, 1945, 69 pp
39. Calandra JC: The Evaluation of Four Analytical Methods for Determining Boron Trifluoride Concentrations, IBT No. 663-05569. Unpublished report submitted to NIOSH by Industrial Bio-Test Laboratories Inc, Northbrook, Ill, 1975, 30 pp

40. Hatcher JT, Wilcox LV: Colorimetric determination of boron using carmine. Anal Chem 22:567-69, 1950
41. Technique and Applications of Atomic Absorption. Norwalk, Conn, Perkin-Elmer Corp, 1975, 16 pp
42. Silverman L, Trego K: Colorimetric microdetermination of boron by the curcumin-acetone solution method. Anal Chem 25:1264-67, 1953
43. Wilcox LV: Determination of boron in plant material--An ignition-electrometric titration method. Ind Eng Chem (Anal Ed) 12:341-43, 1940
44. Wilcox LV: Electrometric titration of boric acid. Ind Eng Chem (Anal Ed) 4:38-39, 1932
45. Foote FJ: Determination of boron in waters--Method for direct titration of boric acid. Ind Eng Chem (Anal Ed) 4:39-42, 1932
46. Pflaum DJ, Wenzke HH: Determination of fluorine and boron in organic compounds. Ind Eng Chem (Anal Ed) 4:392-93, 1932
47. Committee on Biological Effects of Atmospheric Pollutants: Fluorides. Washington DC, National Academy of Sciences, National Research Council, Division of Medical Sciences, 1971, pp 1-295
48. Intersociety Committee: Tentative method of analysis for fluoride content of the atmosphere and plant tissues (manual methods), in Methods of Air Sampling and Analysis. Washington DC, American Public Health Association, 1972, pp 246-81
49. Elfers LA, Decker CE: Determination of fluoride in air and stack gas samples by use of an ion specific electrode. Anal Chem 40:1658-61, 1968
50. Neefus JD, Cholak J, Saltzman BE: The determination of fluoride in urine using a fluoride-specific ion electrode. Am Ind Hyg Assoc J 31:96-99, 1970
51. Greene AF: Determination of Total Fluoride in the Presence of Boron by the Fluoride Ion Selective Electrode, analytical method No. 35-0834. Cleveland, Kewanee Oil Co, Harshaw Chemical Co, 1974, 3 pp
52. Skaar OB: The effect of foreign ions on the spectrophotometric determination of boron with methylene blue. Anal Chim Acta 28:200-04, 1963
53. Carlson RM, Paul JL: Potentiometric determination of boron as tetrafluoroborate. Anal Chem 40:1292-95, 1968

54. I. Methods for the preparation of boron fluoride, in Topchiev AV, Zaugorodnil SV, Paushkin YM: Boron Fluoride and its Compounds as Catalysts in Organic Chemistry. New York, Pergamon Press Inc, 1959, pp 13-14
55. Tubich GE: New materials and processes create new liabilities for the foundry. Ind Med Surg 33:79-85, 1964
56. American Conference of Governmental Industrial Hygienists, Committee on Industrial Ventilation: Industrial Ventilation--A Manual of Recommended Practice, ed 14. Lansing, Mich, ACGIH, 1976, pp 1-1 to 14-8
57. Handling and Storage of Boron Trifluoride Tube Trailers, product technical bulletin BFTB-2. Morristown, NJ, Allied Chemical Corp, 1971, 13 pp
58. Boron trifluoride, in Braker W, Mossman AL: Matheson Gas Data Book, ed 5. East Rutherford, NJ, Will Ross Inc, Matheson Gas Products, 1971, pp 43-46
59. Hudswell F, Nairn JS, Wilkinson KL: Corrosion experiments with gaseous boron trifluoride. J Appl Chem 1:333-36, 1951
60. Boron Trifluoride Compressed Gas BF₃--Handling Information, technical bulletin TB-34692. Morristown, NJ, Allied Chemical Corp, General Chemical Division, Baker and Adamson, 1969, 4 pp
61. American Conference of Governmental Industrial Hygienists: Threshold limit values for 1957. Arch Ind Health 16:264, 1957
62. American Conference of Governmental Industrial Hygienists: Threshold limit values for 1960. Arch Environ Health 1:141, 1960
63. American Conference of Governmental Industrial Hygienists, Committee on Threshold Limit Values: Documentation of Threshold Limit Values. Cincinnati, ACGIH, 1962, pp 14-15
64. American Conference of Governmental Industrial Hygienists: Threshold Limit Values for 1963. Cincinnati, ACGIH, 1963, pp 1, 3, 16-17
65. American Conference of Governmental Industrial Hygienists: TLVs--Threshold Limit Values for Chemical Substances and Physical Agents in the Workroom Environment with Intended Changes for 1974. Cincinnati, ACGIH, 1974, pp 1-4, 11, 52-54
66. American Conference of Governmental Industrial Hygienists, Committee on Threshold Limit Values: Documentation of the Threshold Limit Values for Substances in Workroom Air, ed 3. Cincinnati, ACGIH, 1971, pp 26-27

67. Boron trifluoride, in Encyclopaedia of Occupational Health and Safety. Geneva, International Labour Office, 1972, vol 1, 1972, p 204
68. Commission for the Study of Harmful Substances Encountered During Work: [Maximum Concentrations at the Place of Work, communication No. 7.] Bonn, Bundesrepublik, Deutschland, German Research Assoc, 1971, vol 29,, p 10 (Ger)
69. [Official recognition of boron trifluoride as a toxic gas-- Ministerial decree, November 6, 1972.] Gazzetta Ufficiale della Repubblica Italiana 113:7818, 1972 (It)
70. Laboratory Waste Disposal Manual, rev. Washington, DC, Manufacturing Chemists Association, 1974, 176 pp
71. Safe Handling of Compressed Gases in Containers, CGA pamphlet P-1, ed 6. New York, Compressed Gas Association Inc, 1974, 15 pp
72. Largent EJ, Heyroth FF: The absorption and excretion of fluorides-- III. Further observations on metabolism of fluorides at high levels of intake. J Ind Hyg Toxicol 31:134-38, 1949
73. Largent EJ: Metabolism of inorganic fluorides, in Shaw JH (ed): Fluoridation as a Public Health Measure. Washington, DC, American Assoc for the Advancement of Science, 1954, pp 49-78

IX. APPENDIX I

MATERIAL SAFETY DATA SHEET

The following items of information which are applicable to a specific product or material shall be provided in the appropriate block of the Material Safety Data Sheet (MSDS).

The product designation is inserted in the block in the upper left corner of the first page to facilitate filing and retrieval. Print in upper case letters as large as possible. It should be printed to read upright with the sheet turned sideways. The product designation is that name or code designation which appears on the label, or by which the product is sold or known by employees. The relative numerical hazard ratings and key statements are those determined by the rules in Chapter V, Part B, of the NIOSH publication, An Identification System for Occupationally Hazardous Materials. The company identification may be printed in the upper right corner if desired.

(a) Section I. Product Identification

The manufacturer's name, address, and regular and emergency telephone numbers (including area code) are inserted in the appropriate blocks of Section I. The company listed should be a source of detailed backup information on the hazards of the material(s) covered by the MSDS. The listing of suppliers or wholesale distributors is discouraged. The trade name should be the product designation or common name associated with the material. The synonyms are those commonly used for the product, especially formal chemical nomenclature. Every known chemical designation or

competitor's trade name need not be listed.

(b) Section II. Hazardous Ingredients

The "materials" listed in Section II shall be those substances which are part of the hazardous product covered by the MSDS and individually meet any of the criteria defining a hazardous material. Thus, one component of a multicomponent product might be listed because of its toxicity, another component because of its flammability, while a third component could be included both for its toxicity and its reactivity. Note that a MSDS for a single component product must have the name of the material repeated in this section to avoid giving the impression that there are no hazardous ingredients.

Chemical substances should be listed according to their complete name derived from a recognized system of nomenclature. Where possible, avoid using common names and general class names such as "aromatic amine," "safety solvent," or "aliphatic hydrocarbon" when the specific name is known.

The "%" may be the approximate percentage by weight or volume (indicate basis) which each hazardous ingredient of the mixture bears to the whole mixture. This may be indicated as a range or maximum amount, ie, "10-40% vol" or "10% max wt" to avoid disclosure of trade secrets.

Toxic hazard data shall be stated in terms of concentration, mode of exposure or test, and animal used, eg, "100 ppm LC50-rat," "25 mg/kg LD50-skin-rabbit," "75 ppm LC man," or "permissible exposure from 29 CFR 1910.1000," or if not available, from other sources of publications such as the American Conference of Governmental Industrial Hygienists or the American National Standards Institute Inc. Flashpoint, shock sensitivity,

or similar descriptive data may be used to indicate flammability, reactivity, or similar hazardous properties of the material.

(c) Section III. Physical Data

The data in Section III should be for the total mixture and should include the boiling point and melting point in degrees Fahrenheit (Celsius in parentheses); vapor pressure, in conventional millimeters of mercury (mmHg); vapor density of gas or vapor (air = 1); solubility in water, in parts/hundred parts of water by weight; specific gravity (water = 1); percent volatiles (indicated if by weight or volume) at 70 degrees Fahrenheit (21.1 degrees Celsius); evaporation rate for liquids or sublimable solids, relative to butyl acetate; and appearance and odor. These data are useful for the control of toxic substances. Boiling point, vapor density, percent volatiles, vapor pressure, and evaporation are useful for designing proper ventilation equipment. This information is also useful for design and deployment of adequate fire and spill containment equipment. The appearance and odor may facilitate identification of substances stored in improperly marked containers, or when spilled.

(d) Section IV. Fire and Explosion Data

Section IV should contain complete fire and explosion data for the product, including flashpoint and autoignition temperature in degrees Fahrenheit (Celsius in parentheses); flammable limits, in percent by volume in air; suitable extinguishing media or materials; special firefighting procedures; and unusual fire and explosion hazard information. If the product presents no fire hazard, insert "NO FIRE HAZARD" on the line labeled "Extinguishing Media."

(e) Section V. Health Hazard Information

The "Health Hazard Data" should be a combined estimate of the hazard of the total product. This can be expressed as a TWA concentration, as a permissible exposure, or by some other indication of an acceptable standard. Other data are acceptable, such as lowest LD50 if multiple components are involved.

Under "Routes of Exposure," comments in each category should reflect the potential hazard from absorption by the route in question. Comments should indicate the severity of the effect and the basis for the statement if possible. The basis might be animal studies, analogy with similar products, or human experiences. Comments such as "yes" or "possible" are not helpful. Typical comments might be:

Skin Contact--single short contact, no adverse effects likely; prolonged or repeated contact, possibly mild irritation.

Eye Contact--some pain and mild transient irritation; no corneal scarring.

"Emergency and First Aid Procedures" should be written in lay language and should primarily represent first-aid treatment that could be provided by paramedical personnel or individuals trained in first aid.

Information in the "Notes to Physician" section should include any special medical information which would be of assistance to an attending physician including required or recommended preplacement and periodic medical examinations, diagnostic procedures, and medical management of overexposed employees.

(f) Section VI. Reactivity Data

The comments in Section VI relate to safe storage and handling of hazardous, unstable substances. It is particularly important to highlight instability or incompatibility to common substances or circumstances, such as water, direct sunlight, steel or copper piping, acids, alkalies, etc. "Hazardous Decomposition Products" shall include those products released under fire conditions. It must also include dangerous products produced by aging, such as peroxides in the case of some ethers. Where applicable, shelf life should also be indicated.

(g) Section VII. Spill or Leak Procedures

Detailed procedures for cleanup and disposal should be listed with emphasis on precautions to be taken to protect employees assigned to cleanup detail. Specific neutralizing chemicals or procedures should be described in detail. Disposal methods should be explicit including proper labeling of containers holding residues and ultimate disposal methods such as "sanitary landfill" or "incineration." Warnings such as "comply with local, state, and federal antipollution ordinances" are proper but not sufficient. Specific procedures shall be identified.

(h) Section VIII. Special Protection Information

Section VIII requires specific information. Statements such as "Yes," "No," or "If necessary" are not informative. Ventilation requirements should be specific as to type and preferred methods. Respirators shall be specified as to type and NIOSH or US Bureau of Mines approval class, ie, "Supplied air," "Organic vapor canister," etc. Protective equipment must be specified as to type and materials of construction.

(i) Section IX. Special Precautions

"Precautionary Statements" shall consist of the label statements selected for use on the container or placard. Additional information on any aspect of safety or health not covered in other sections should be inserted in Section IX. The lower block can contain references to published guides or in-house procedures for handling and storage. Department of Transportation markings and classifications and other freight, handling, or storage requirements and environmental controls can be noted.

(j) Signature and Filing

Finally, the name and address of the responsible person who completed the MSDS and the date of completion are entered. This will facilitate correction of errors and identify a source of additional information.

The MSDS shall be filed in a location readily accessible to employees exposed to the hazardous material. The MSDS can be used as a training aid and basis for discussion during safety meetings and training of new employees. It should assist management by directing attention to the need for specific control engineering, work practices, and protective measures to ensure safe handling and use of the material. It will aid the safety and health staff in planning a safe and healthful work environment and in suggesting appropriate emergency procedures and sources of help in the event of harmful exposure of employees.

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MATERIAL SAFETY DATA SHEET

I PRODUCT IDENTIFICATION		
MANUFACTURER'S NAME	REGULAR TELEPHONE NO. EMERGENCY TELEPHONE NO	
ADDRESS		
TRADE NAME		
SYNONYMS		
II HAZARDOUS INGREDIENTS		
MATERIAL OR COMPONENT	%	HAZARD DATA
III PHYSICAL DATA		
BOILING POINT, 760 MM HG		MELTING POINT
SPECIFIC GRAVITY (H ₂ O=1)		VAPOR PRESSURE
VAPOR DENSITY (AIR=1)		SOLUBILITY IN H ₂ O, % BY WT
% VOLATILES BY VOL.		EVAPORATION RATE (BUTYL ACETATE=1)
APPEARANCE AND ODOR		

IV FIRE AND EXPLOSION DATA				
FLASH POINT (TEST METHOD)			AUTOIGNITION TEMPERATURE	
FLAMMABLE LIMITS IN AIR, % BY VOL.		LOWER		UPPER
EXTINGUISHING MEDIA				
SPECIAL FIRE FIGHTING PROCEDURES				
UNUSUAL FIRE AND EXPLOSION HAZARD				
V HEALTH HAZARD INFORMATION				
HEALTH HAZARD DATA				
ROUTES OF EXPOSURE				
INHALATION				
SKIN CONTACT				
SKIN ABSORPTION				
EYE CONTACT				
INGESTION				
EFFECTS OF OVEREXPOSURE				
ACUTE OVEREXPOSURE				
CHRONIC OVEREXPOSURE				
EMERGENCY AND FIRST AID PROCEDURES				
EYES				
SKIN				
INHALATION				
INGESTION				
NOTES TO PHYSICIAN				

VI REACTIVITY DATA
CONDITIONS CONTRIBUTING TO INSTABILITY
INCOMPATIBILITY
HAZARDOUS DECOMPOSITION PRODUCTS
CONDITIONS CONTRIBUTING TO HAZARDOUS POLYMERIZATION
VII SPILL OR LEAK PROCEDURES
STEPS TO BE TAKEN IF MATERIAL IS RELEASED OR SPILLED
NEUTRALIZING CHEMICALS
WASTE DISPOSAL METHOD
VIII SPECIAL PROTECTION INFORMATION
VENTILATION REQUIREMENTS
SPECIFIC PERSONAL PROTECTIVE EQUIPMENT
RESPIRATORY (SPECIFY IN DETAIL)
EYE
GLOVES
OTHER CLOTHING AND EQUIPMENT

IX SPECIAL PRECAUTIONS

PRECAUTIONARY
STATEMENTS

OTHER HANDLING AND
STORAGE REQUIREMENTS

PREPARED BY _____

ADDRESS _____

DATE _____

X. TABLES

TABLE X-1

PHYSICAL PROPERTIES OF BORON TRIFLUORIDE

Molecular formula	BF ₃
Formula weight	67.81
Density at 760 mmHg and 0 C	3.077 g/l
Melting point	-127.1 C
Boiling point	-100.4 C
Solubility	
in ml of gas/ml of water	1,057 at 762 mmHg, 0 C
in g/100 g of water	369.4 at 6 C
Conversion factors	1 ppm = 2.77 mg/cu m 1 mg/cu m = 0.361 ppm

Adapted from Booth and Martin [4] and Martin [2]

TABLE X-2

REPRESENTATIVE BORON TRIFLUORIDE CATALYZED REACTIONS

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- (1) Synthesis of:
- | | |
|----------------------------|----------------------|
| (a) saturated hydrocarbons | (d) ketones |
| (b) alcohols | (e) ethers |
| (c) olefins | (f) sulfur compounds |
- (2) Condensation of:
- | | |
|-------------------------|--------------------------|
| (a) acid with olefin | (d) nitrite with alcohol |
| (b) acid with acetylene | (e) amide with alcohol |
| (c) acid with alcohol | (f) aldols |
- (3) Dehydration of:
- (a) alcohols
 - (b) acids
 - (c) ketones
- (4) Alkylation of:
- (a) paraffinic hydrocarbons
 - (b) alkyl esters
 - (c) aromatic hydrocarbons
- (5) Polymerization of:
- | | |
|-----------------|---------------------|
| (a) ethylene | (e) vinyl compounds |
| (b) propylene | (f) dienes |
| (c) butylene | (g) cyclic oxides |
| (d) isobutylene | |
- (6) Isomerization (Beckman rearrangement)
- (7) Epoxy resin curing
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Adapted from Booth and Martin [4]

TABLE X-3

USES AND PHYSICAL STATES OF SOME
REPRESENTATIVE BORON TRIFLUORIDE COMPLEXES

Complex	Physical State	Use	Reference
BF ₃ monoethylamine	Solid	Catalyst for epoxy resins	6 12
BF ₃ dihydrate	Liquid	Catalyst for organic reactions	11
BF ₃ diethyl ether	"	Catalyst for polymerizations, alkylations, and isomerizations; intermediate chemical	7
BF ₃ phenol	"	Catalyst	9
BF ₃ phosphoric acid	"	Polymerization and alkylation catalyst	10
BF ₃ piperidine	"	Epoxy catalyst	9
BF ₃ dimethyl aniline	"	"	9
BF ₃ methanol solution	"	Catalyst	8