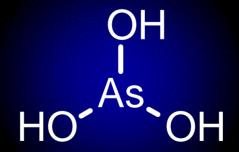
# Role of Microorganisms in the Speciation and Mobility of Arsenic

# Jim A Field, Wenjie Sun, Irail Cortinas and Reyes Sierra

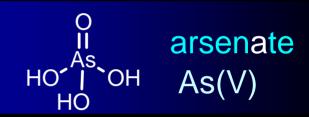
Department of Chemical and Environmental Engineering
The University of Arizona



# **Inorganic Arsenic**

#### Predominant Species of the Biogeochemical Cycle

arsenite As(III)



More toxic than As(V)

Analogue of phosphate

Reacts with R-SH

- Adsorbed by iron and aluminum oxides
- Adsorbed by iron oxides
- Less mobile

More mobile

# Organoarsenicals

# Methylated Forms are Less Predominant Species in the Biogeochemical Cycle

cid

H<sub>3</sub>C As CH<sub>3</sub> DMA(V)

Very Toxic

Pesticides

Metabolite

Metabolite

Unstable

## Landfill

#### **Methanogenesis in Landfill**

Phase I: Consumption of O<sub>2</sub>

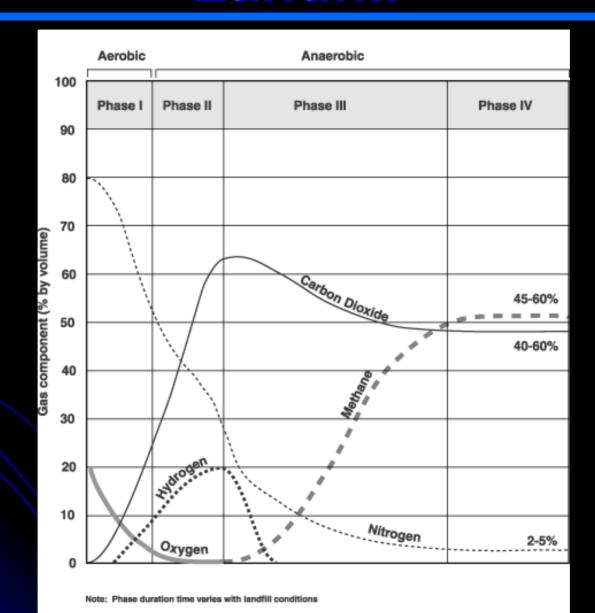
Phase II: Formation of organic acids, release NH<sub>4</sub><sup>+</sup>



Phase III: Conversion of organic acids to CH<sub>4</sub> and HCO<sub>3</sub><sup>-</sup>, continued release NH<sub>4</sub><sup>+</sup>, mildly alkaline pH



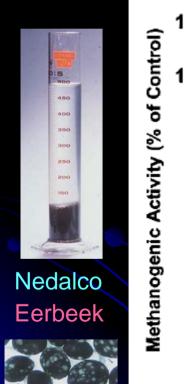
# Landfill

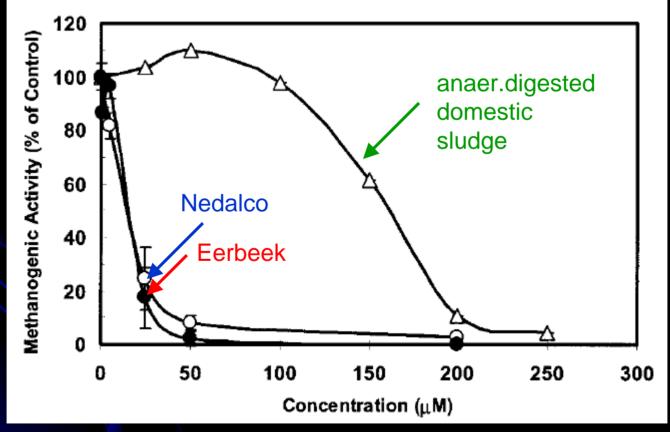


Source: EPA 1997

# Toxicity Arsenic to Methanogens

# Dose Response Curves of As(III) on Acetoclastic Methanogenesis







# **Toxicity Arsenic to Methanogens**

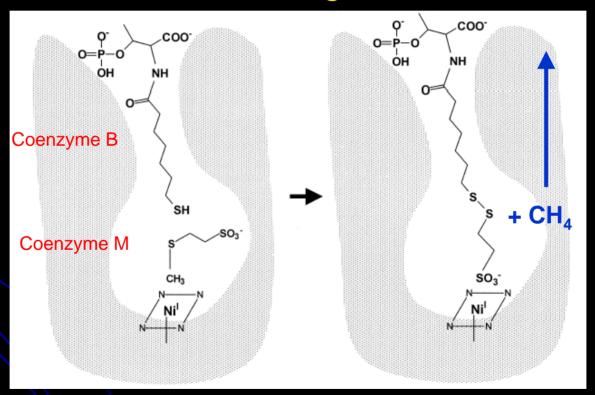
# 50% Inhibitory Concentrations of Arsenicals on Methanogensis

Compound	Substrate	IC50 (μM)	IC50 (ppm)
As(III)	acetate	15.5	1.16
As(III)	$H_2$	27.1	2.03
As(III)	lactate	4.4	0.33
As(V)	acetate	>500.0	
As(V)	$H_2$	>500.0	
MMA(III)	acetate	9.1	0.68
MMA(V)	acetate	>5,000.0	
DMA(V)	acetate	>5,000.0	

Sierra-Alvarez et al. 2004 AEM 70:5688

# **Toxicity Arsenic to Methanogens**

Sulfhydryl Groups are Central to the Biochemistry of Methanogens



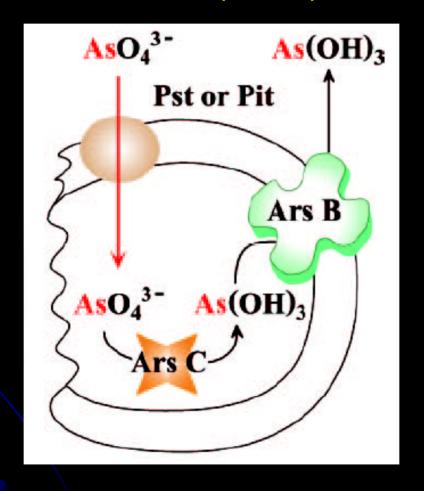
Methyl Reductase: F430 Complex

# Microorganisms use Two Strategies for Arsenate Reduction

Arsenate Reductase (ArsC) for Detoxification: As(V) is reduced to As(III) to facilitate pumping it from the cell without accidently pumping out phosphate

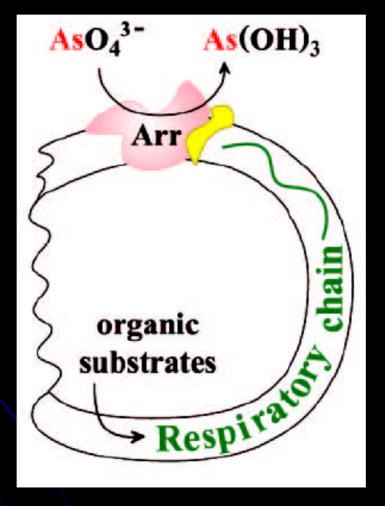
Dissimilatory Arsenate Reductase (ArrA): Arsenate is reduced as a terminal electron acceptor during the anoxic respiration coupled to the oxidation of simple substrates

Arsenate Reductase (ArsC): Detoxification



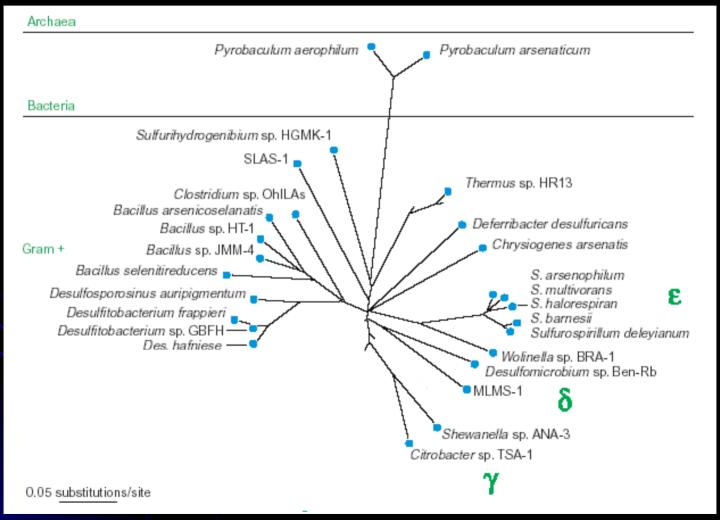
Silver & Phung 2005. AEM 71:599

Dissimilatory Arsenate Reductase (ArrA): Respiration



Silver & Phung 2005. AEM 71:599

#### Dissimilatory Arsenate Reducing Bacteria



#### Dissimilatory Arsenate Reducing Bacteria

Example of Reactions

**Complete oxidation organic matter** 

$$CH_3COOH + 4 AsO(OH)_3 \rightarrow 2 CO_2 + 4 As(OH)_3 + 2 H_2O$$
  
acetate  $As(V)$   $CO_2$   $As(III)$ 

Partial oxidation organic matter

CH<sub>3</sub>CHOHCOOH + 2 AsO(OH)<sub>3</sub> 
$$\rightarrow$$
 CO<sub>2</sub> + CH<sub>3</sub>COOH + 2 As(OH)<sub>3</sub> + H<sub>2</sub>O

lactate As(V) CO<sub>2</sub> acetate As(III)

#### Dissimilatory Arsenate Reducing Bacteria

Electron Donors Known to Support Dissimilatory Arsenate Reducing Microorganisms

Organic Acids Alcohol, Suga		Aromatic	Inorganic	
Lactate	Ethanol	Phenol	$H_2$	
Pyruvate	Glycerol	Benzoate	$H_2^-S$	
Fumurate	Glucose	Syringate		
Malate		Ferulate		
Succinate		Toluene		
Citrate				

**Formate** 

Acetate

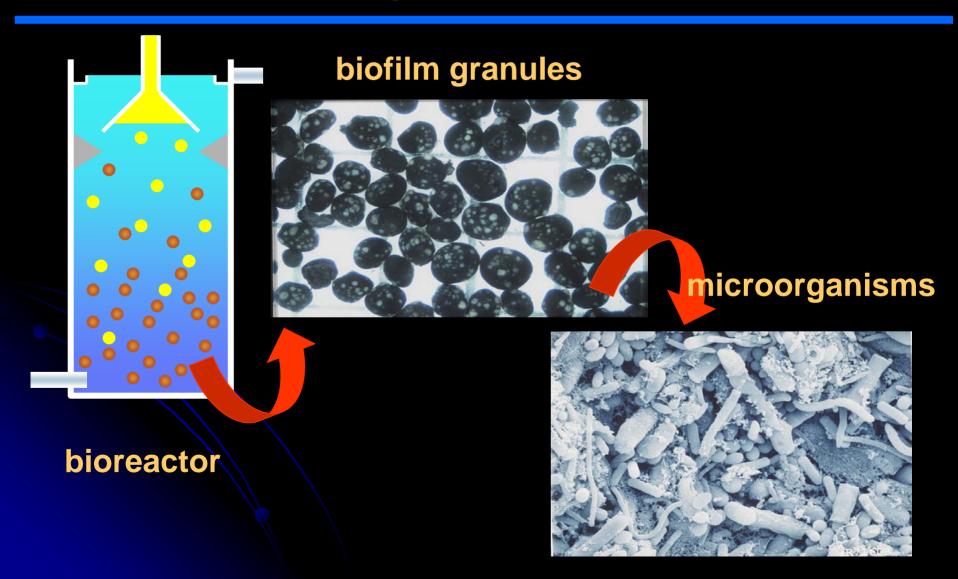
Butyrate

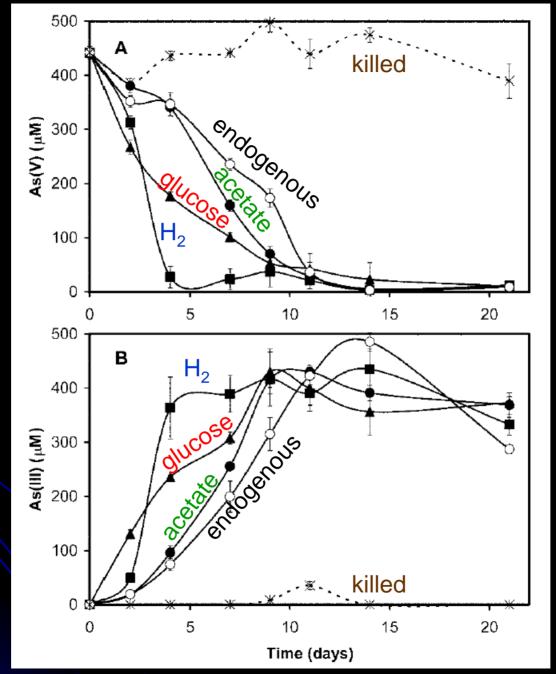
Niggemyer et al 2001. AEM 67:5568

Liu et al 2004. FEMS Microbiol. Ecol. 48:323

Hoeft et al 2004. AEM 70:2741

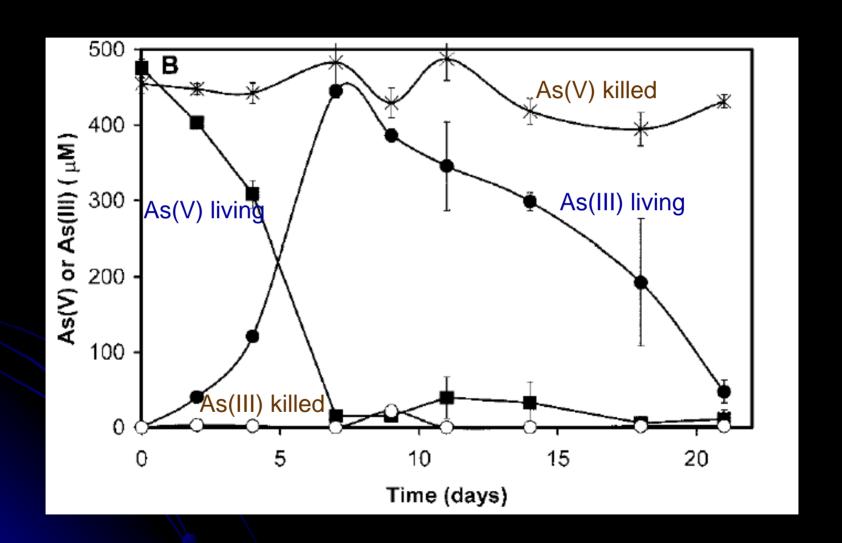
# Does Arsenate Reduction Occur in a Methanogenic Consortium?





Field et al 2004. Biodegradation 15:185

# Arsenate Reduction In a Methanogenic Consortium with Sulfate Added



# **Methylation of Arsenic**

#### Both Eukaryotes and Prokaryotes Methylate Arsenic

- Challenger Mechanism
- Enzymes Cyt19 in fungi, mammals ArsM in bacteria, archaea
- Methyl Donors S-adenosyl methionine Methyl-Vitamin B12

HO MMA(V) MMA(III) R-CH<sub>3</sub>

As(V)

As(III)

R-CH<sub>3</sub>

Bentley& Chasteen 2002. MMBR 66:250

## **Oxidation of Arsenite**

There are two physiologically distinct classes of Arsenite Oxidizers

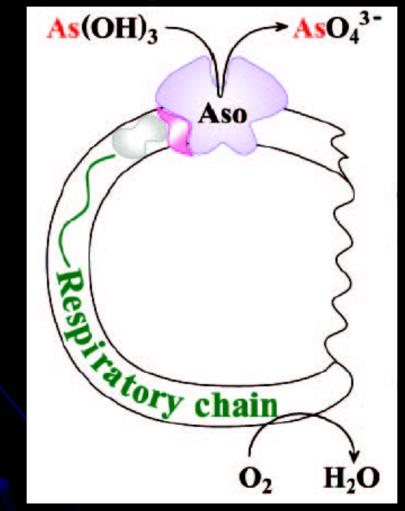
- Chemoorganoheterotrophic Arsenite Oxidizers: Detoxification Mechanism
- Chemolithoautotrophic Arsenite Oxidizers: Arsenite is used as an energy source (electron-donor); CO<sub>2</sub> is fixed

Oxic

Anoxic

 $NO_3^- + 2 H^+ + 5 As(OH)_3 \rightarrow N_2 + 5 AsO(OH)_3 + H_2O$ 

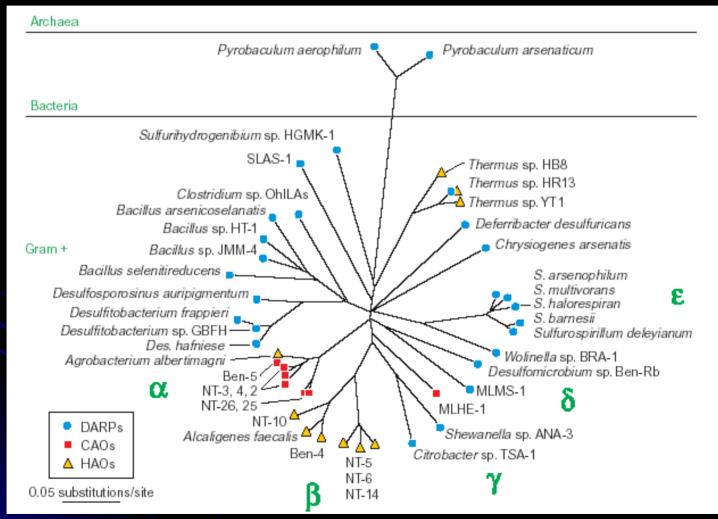
#### Respiratory Arsenite Oxidase



Silver & Phung 2005. AEM 71:599

### **Oxidation of Arsenite**

#### Arsenite Oxidizing Bacteria ( ▲ ■ )



# Anoxic Arsenite Oxidation by Municipal Anaerobic Digester Sludge

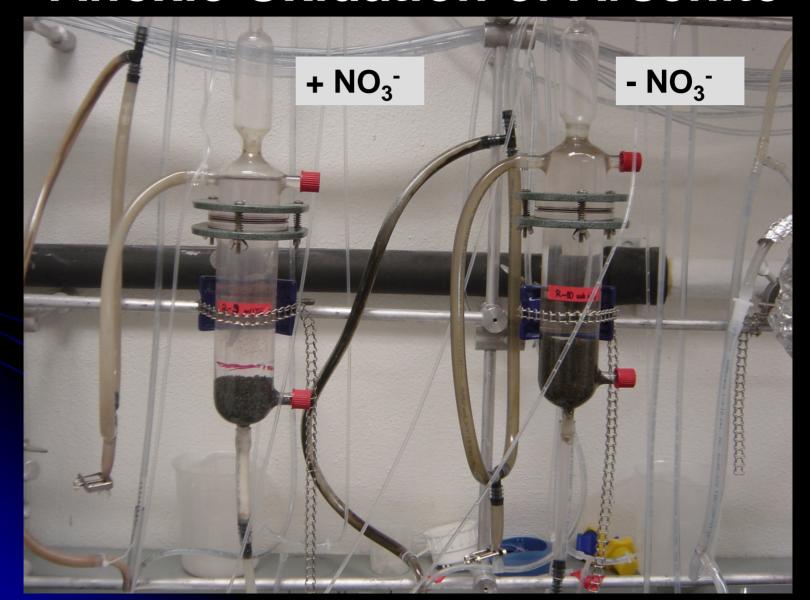
Formation of As(V) from 0.5 mM As(III)

#### **Anoxic Oxidation of Arsenite is Ubiquitous**

Sample	As(V) formation		Time <sup>†</sup>
	+NO <sub>3</sub> -	-NO <sub>3</sub> -	(d)
anaer. bioreactor sludge, distillery	+	_	< 4
anaer. bioreactor sludge, paper	-	-	
municipal anaer. digester sludge	+	-	10
thiosulfate-denitrification enrichment	+	-	10
municipal aerobic active sludge	-	-	
duck pond sediments	+	-	< 5
Winogradsky column sediment	+	-	< 5
Pinal Creek sediments (high Mn)	+	+	
groundwater	-	_	

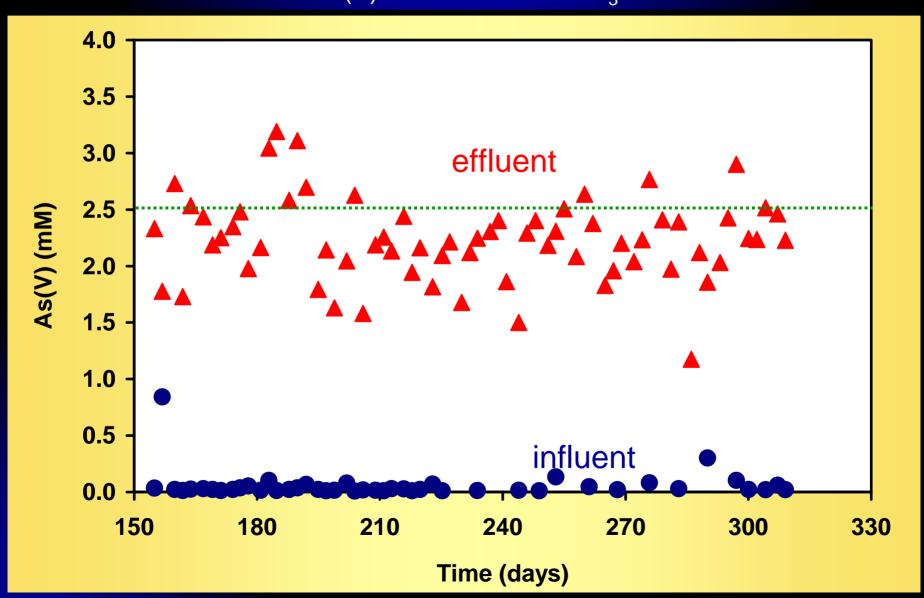
time to oxidize 0.5 mM As(III) to As(V) linked to denitrification

# **Bioreactor Test for Continuous Anoxic Oxidation of Arsenite**



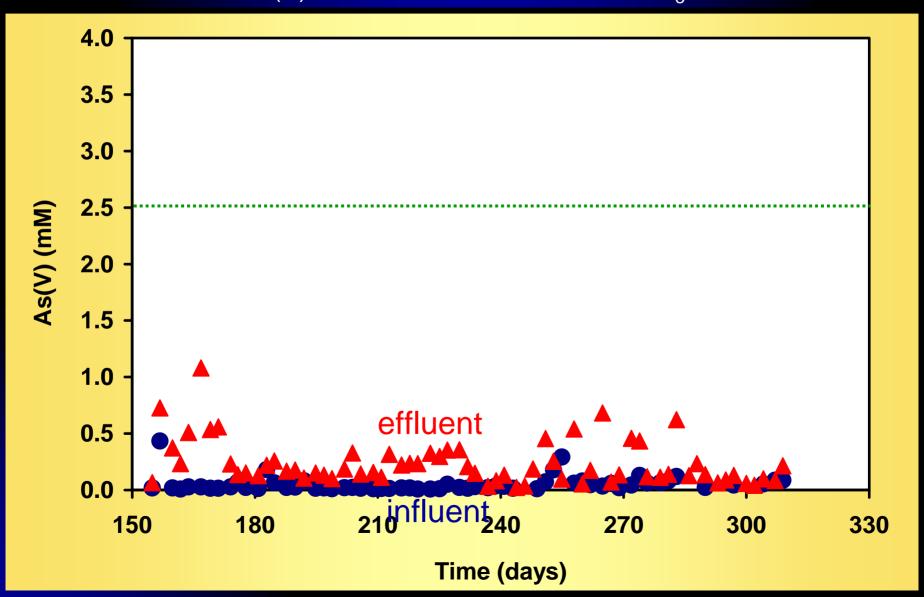
# **Bioreactor Results**

#### As(V) in Reactor Fed NO<sub>3</sub>-

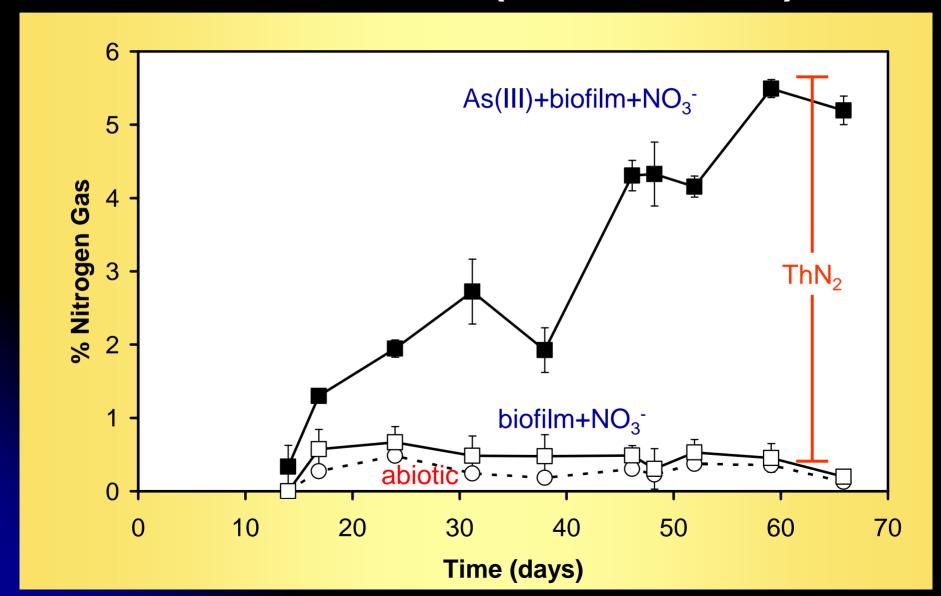


## **Bioreactor Results**

As(V) in Control Reactor not Fed NO<sub>3</sub>-

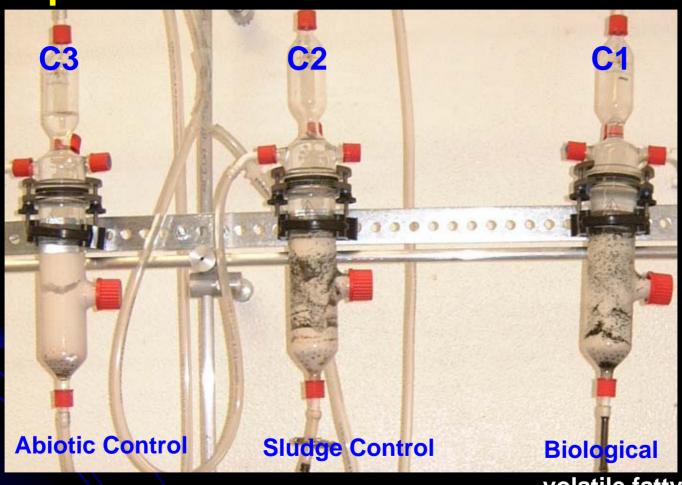


# As(III) Oxidation Linked to Complete Denitrification (3.5 mM AsIII)





#### **Experimental: Continuous Columns**



sludge

NH<sub>3</sub> + NaHCO<sub>3</sub>

volatile fatty acids sludge

NH<sub>3</sub> + NaHCO<sub>3</sub>

NH<sub>3</sub> + NaHCO<sub>3</sub>

#### **Experimental: Continuous Columns**

Activated Aluminum (AA) operated for 257d:

Sierra-Alvarez et al 2005 Water Research 39:199

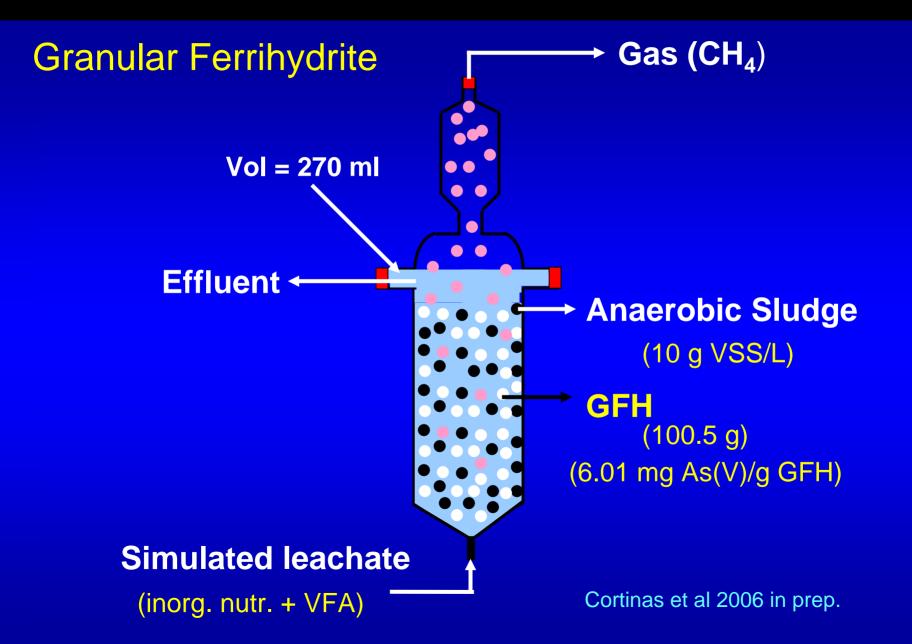
**Biologically Active Column** = 17.4% of arsenic released as freely soluble identified arsenic species; loss of 37.2% from AA

The predominate species released was As(III), 85% of soluble species

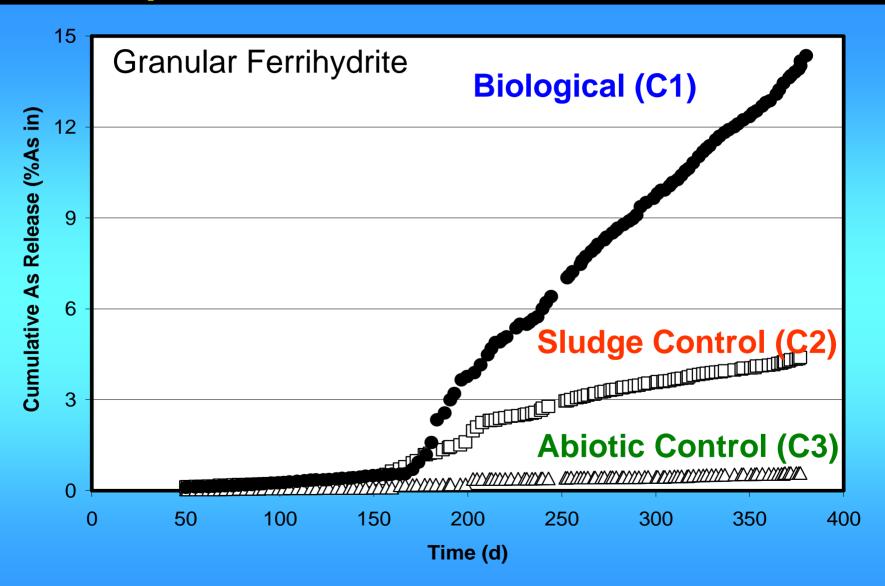
**Abiotic Column** = 3.4% of arsenic released as freely soluble identified arsenic species; loss of 7.6% from AA

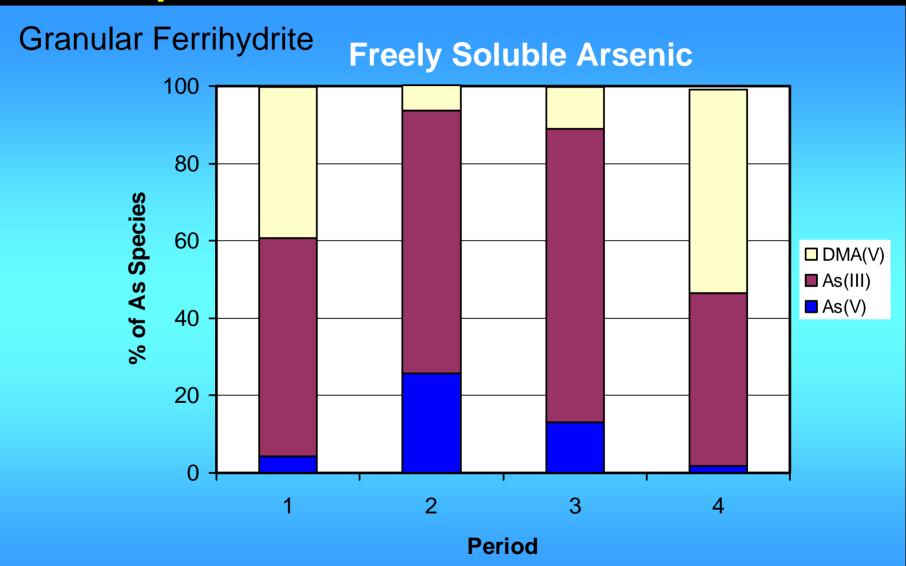
**Granular Ferrihydrite (GFH)** operated for 387d:

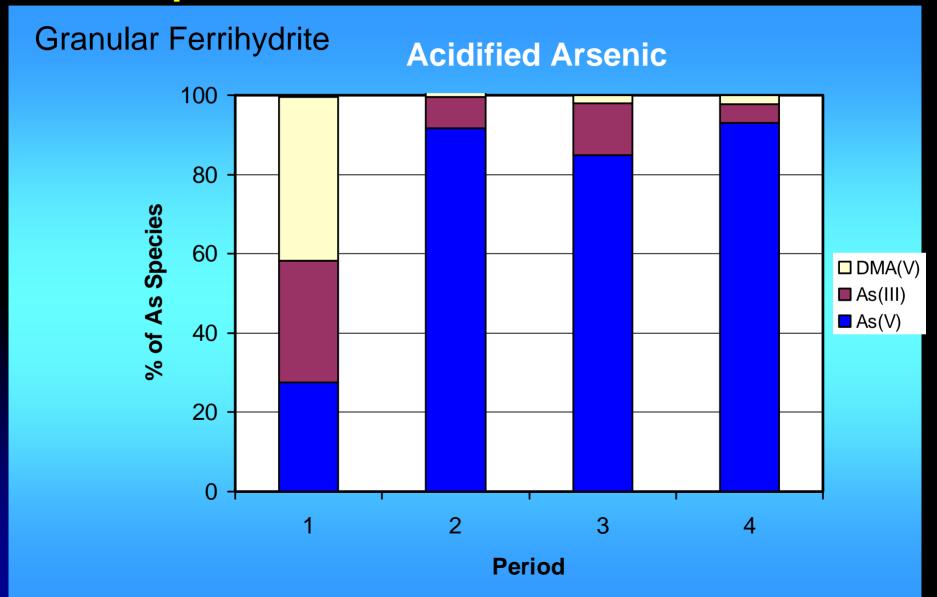
Results reported in next slides (Cortinas et al. 2006. In preparation)







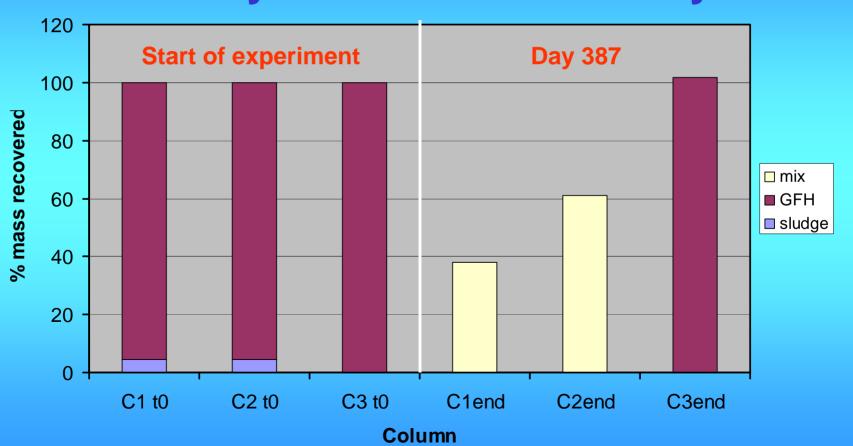




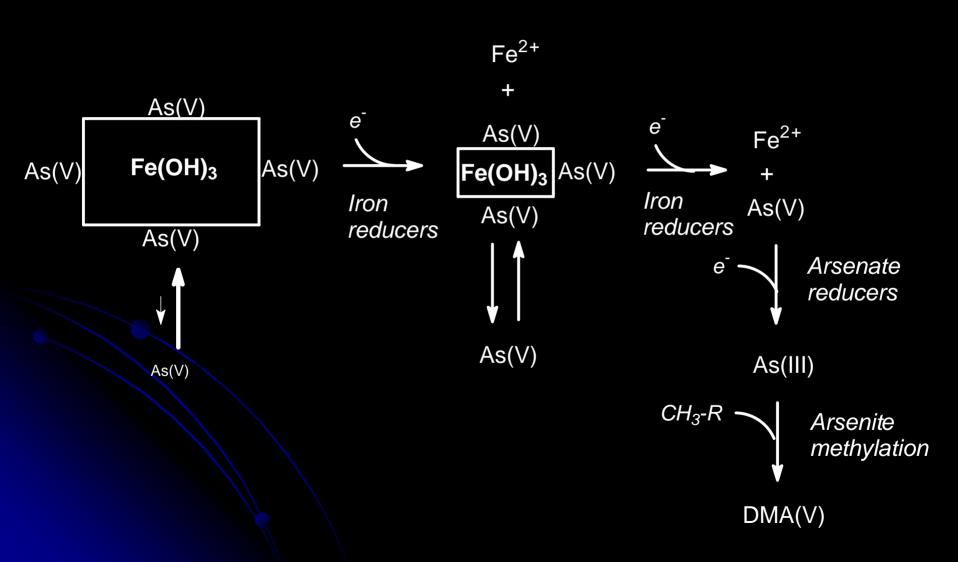
**Experimental: Continuous Columns** 

Granular Ferrihydrite

#### Recovery of Mass after 387 days

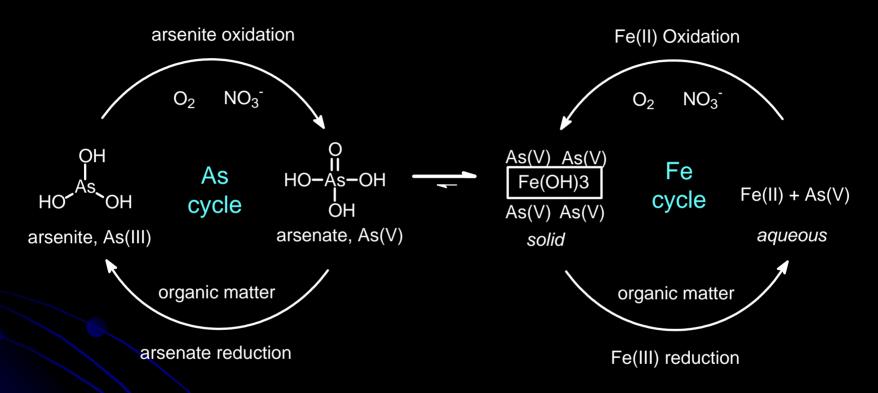


#### **Schematic Overview GFH Columns**



## **Conclusion 1**

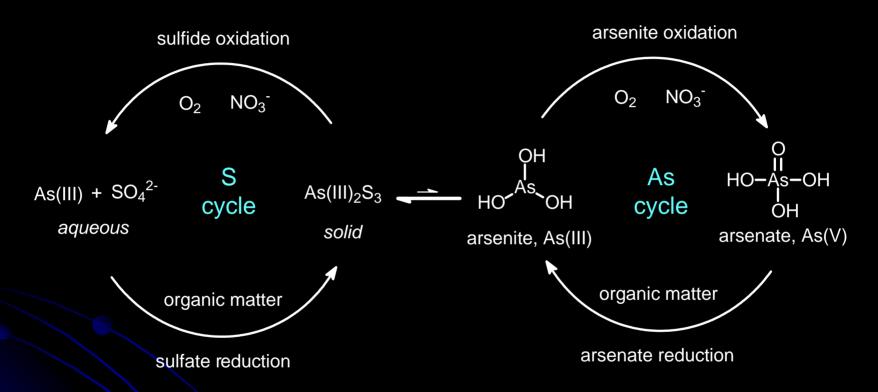
#### **Arsenic in Matrix with Fe/Al Oxide Minerals**



- Aerobic Conditions: Arsenic immobile
- Anaerobic Conditions: Arsenic mobile

### **Conclusion 2**

#### **Arsenic in Matrix with Sulfide Minerals**



- Aerobic Conditions: Arsenic mobile
- Anaerobic Conditions: Arsenic immobile

# **Graduate Students**

