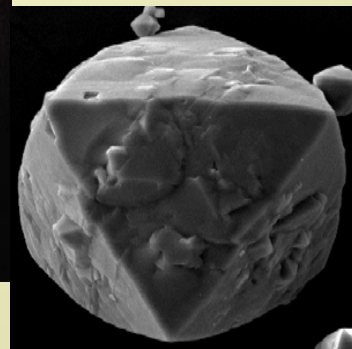




**Argonne**  
NATIONAL  
LABORATORY

*... for a brighter future*

# ***Chemical Sciences and Engineering***



**Establishing the relationship between structure and function  
at different length and time scales under reaction conditions**

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***J. Penner-Hahn (U. Mich.)***



U.S. Department  
of Energy

UChicago ►  
Argonne<sub>LLC</sub>

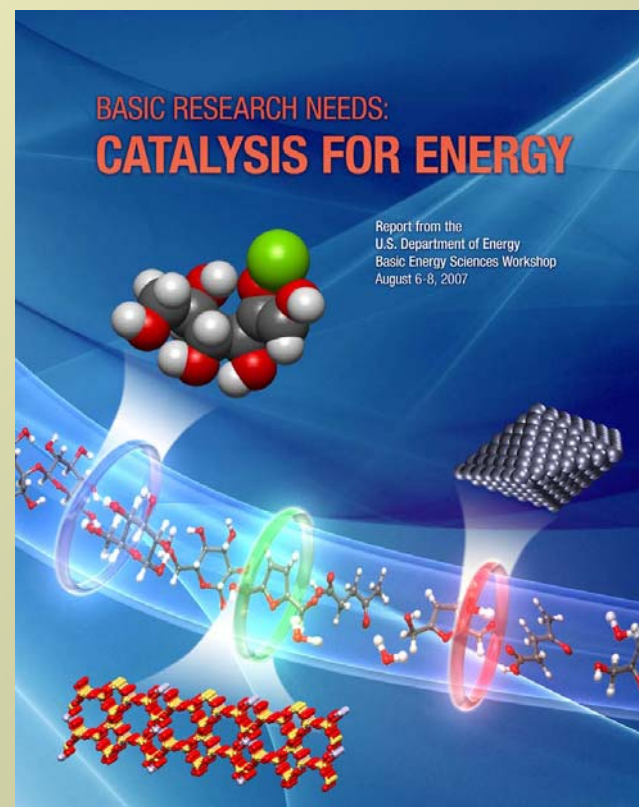


**Office of  
Science**  
U.S. DEPARTMENT OF ENERGY

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## *Impact of Field*

- **Heterogeneous Catalysis**
  - Production of fuels and chemicals
  - Critical to future, secure energy solutions
  - Essential for environmental protection
- **Fuel Cells**
  - Efficient electro-chemical power
- **Solution Chemistry**
  - Solubility/transport of heavy metals
  - Solid-liquid interfacial partitioning

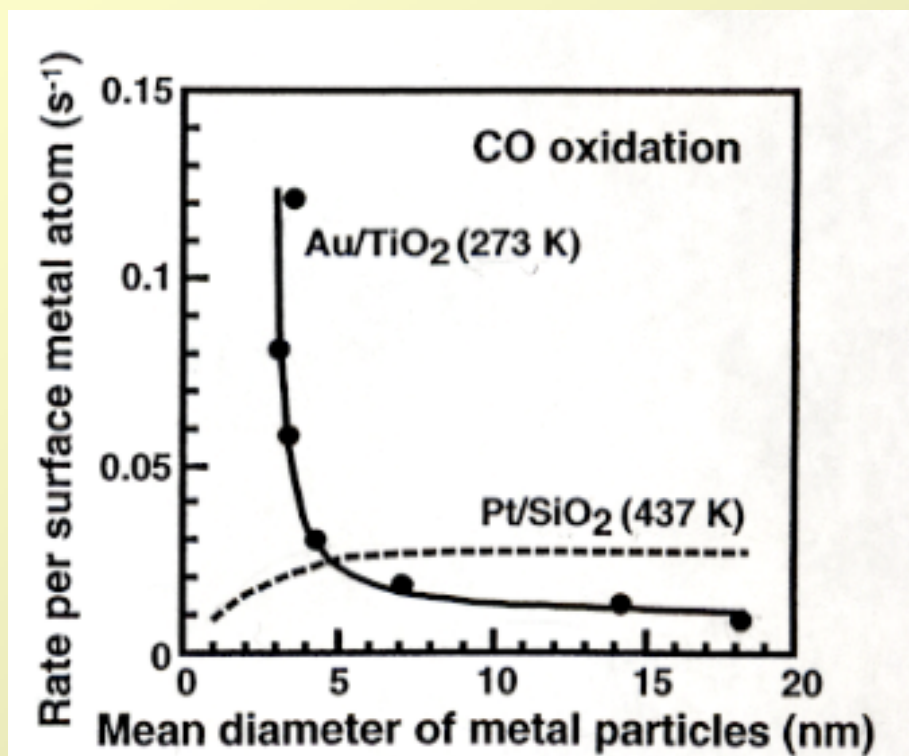


## *Today's Key Techniques*

- **Essential to characterize materials under realistic reaction conditions**
  - Reactive gases (Hydrocarbons, H<sub>2</sub>, CO, O<sub>2</sub>, NO<sub>x</sub>, H<sub>2</sub>S, etc.)
  - High Temperature (>800°C)
  - High pressure (1-100 bar)
  - Gas and liquid phase
  - Corrosive environments
- **Spectroscopy (EXAFS, XANES)**
- **Diffraction (XRD/PDF)**
- **Scattering (SAXS)**

## Catalysis of Au Nanoparticles

- Historically, these are very low activity catalysts
- Au foil does not chemisorb  $H_2$ , CO or  $O_2$
- Highly active for low temperature CO oxidation

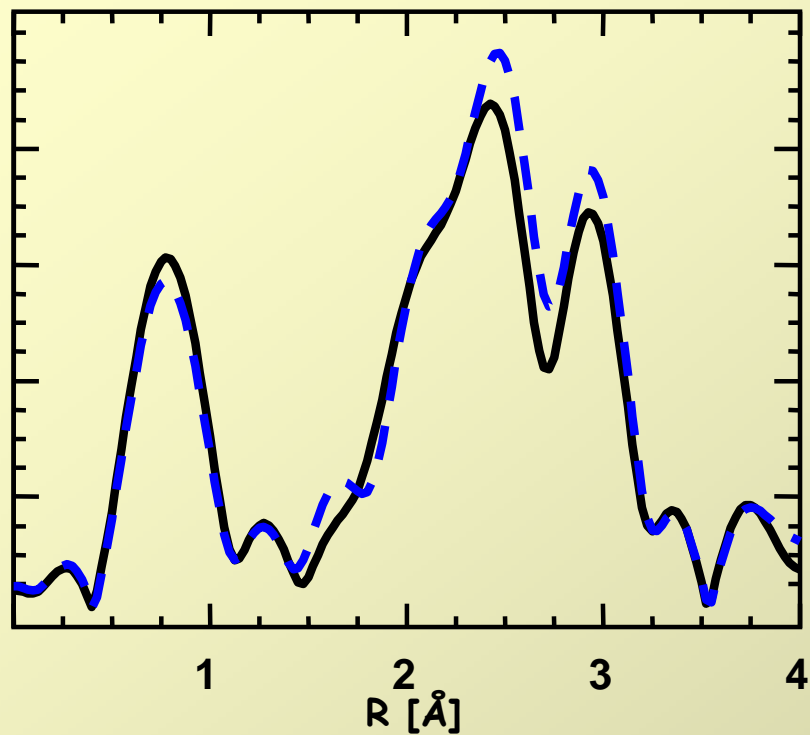


### Additional reactions

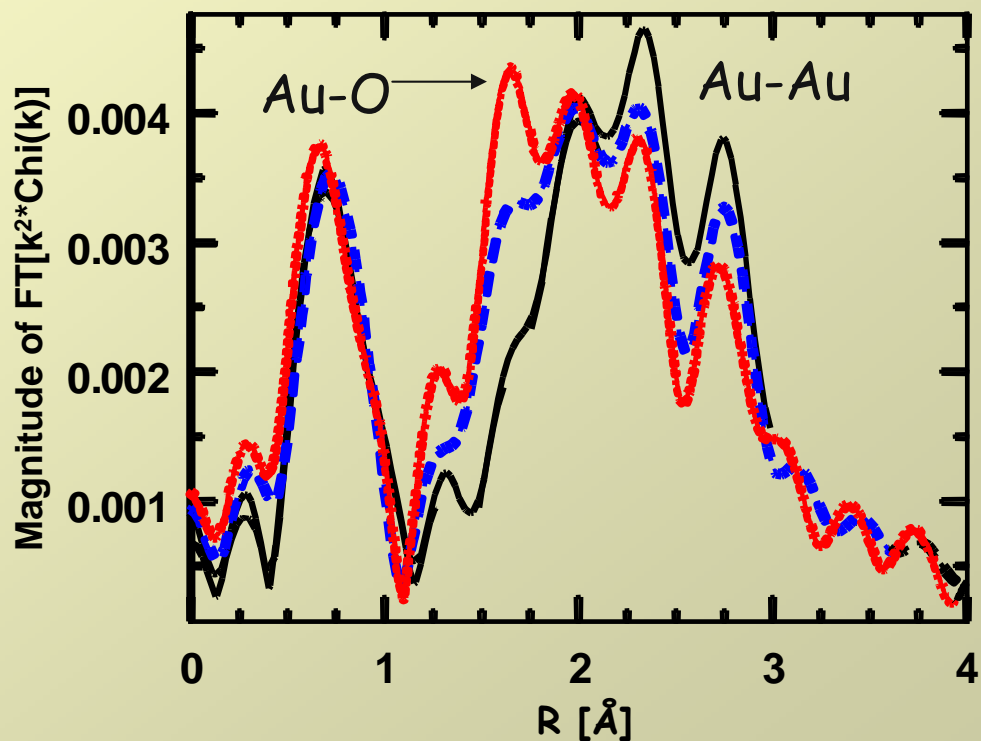
Selective CO( $H_2$ ) oxidation  
LT Water gas shift  
DeNO<sub>x</sub>  
Decomposition of halogen  
compounds  
Selective hydrogenation  
 $H_2O_2$  synthesis from  $H_2$  and  $O_2$

# EXAFS of Oxidized Au

3.2 wt % Au/SiO<sub>2</sub>  
CN = 8.7, R = 2.85 Å

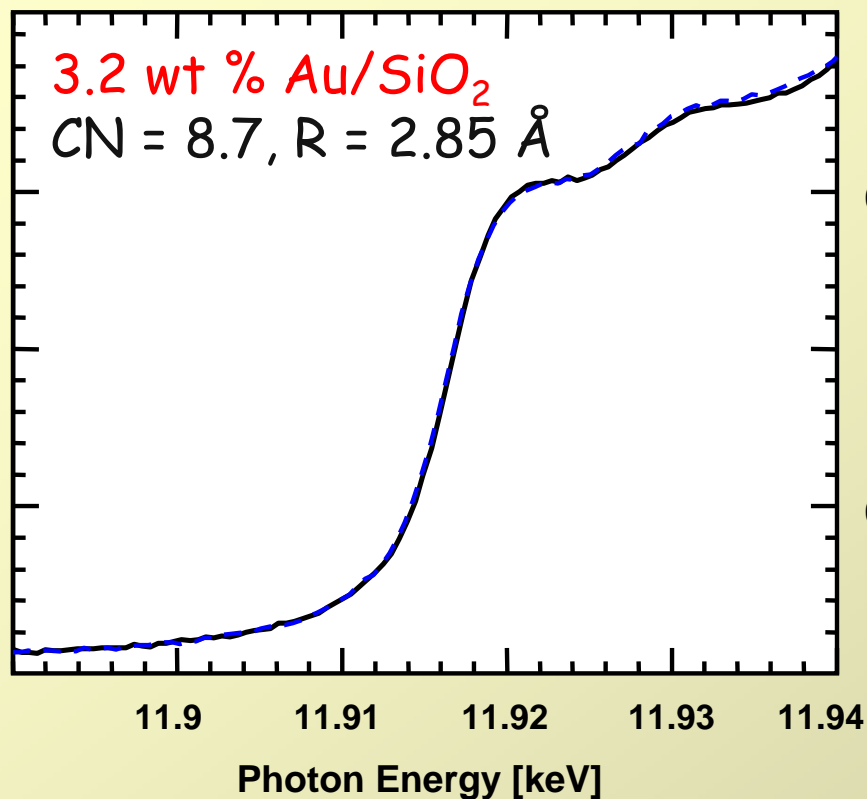


1.3% Au/Al<sub>2</sub>O<sub>3</sub>  
CN = 4.2, R = 2.72 Å

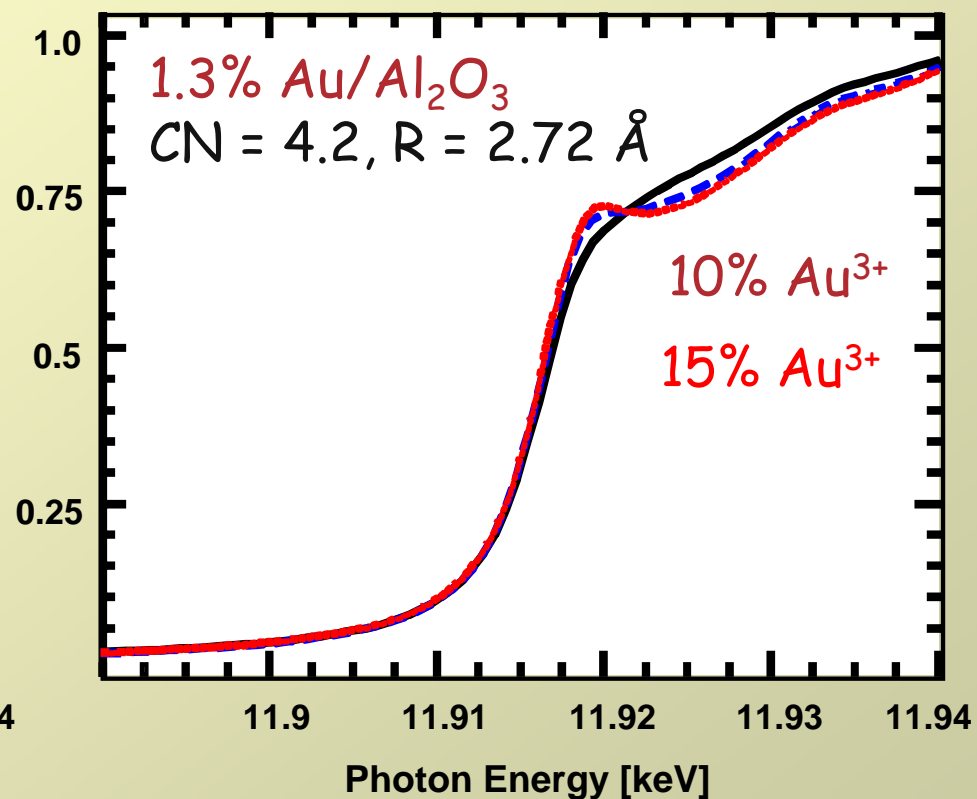


Black: Reduced in H<sub>2</sub> at 250°C  
Blue: Followed by air at RT  
Red: Followed by air at 225°C

## Reactivity of Au Particles to O<sub>2</sub>

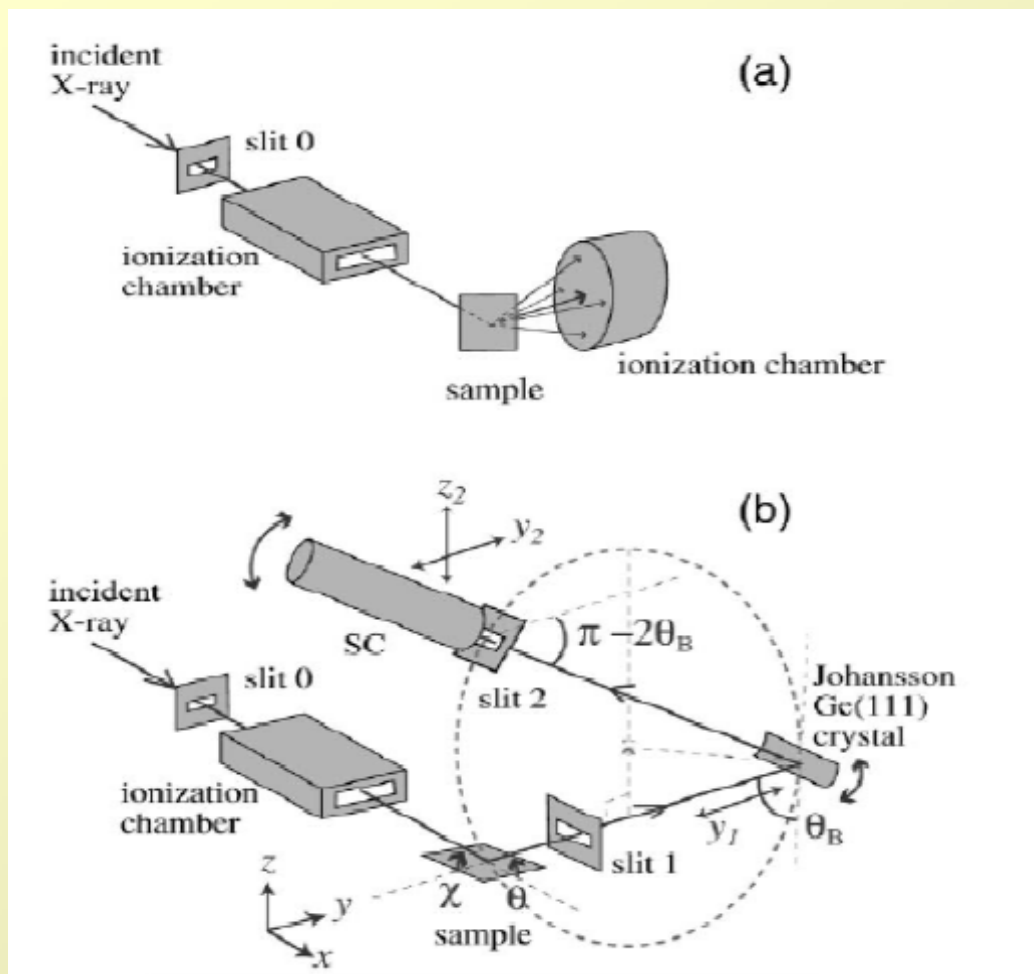


Black: Reduced in H<sub>2</sub> at 250°C  
Blue: Followed by air at RT  
Red: Followed by air at 225°C



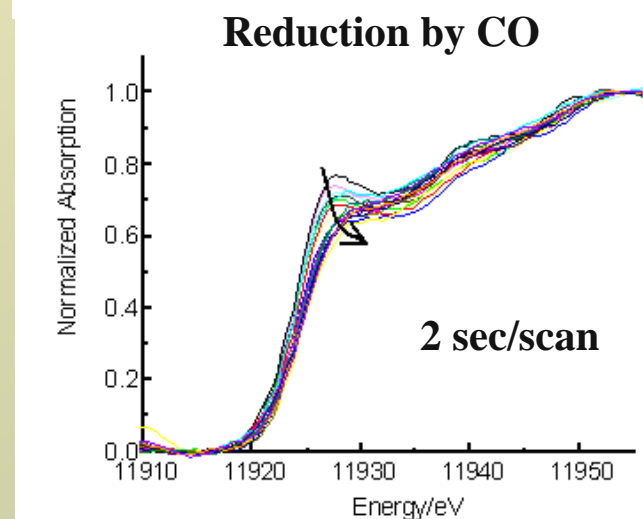
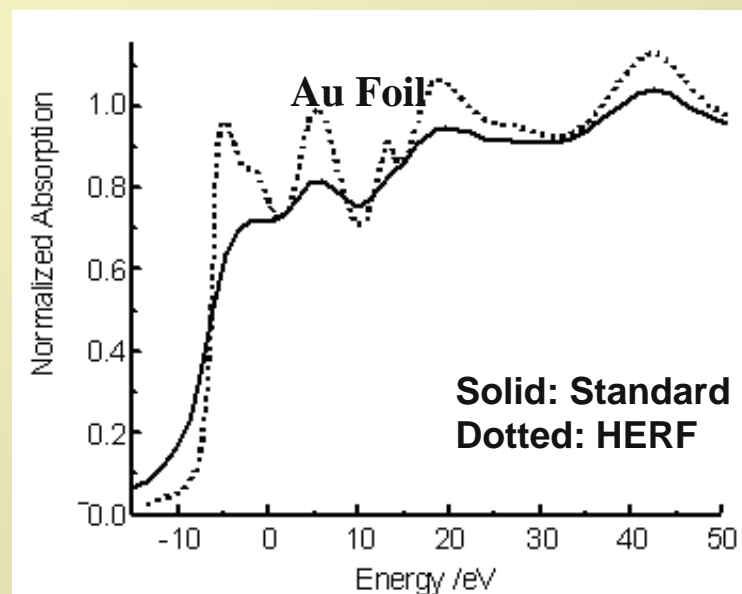
Miller, et al.  
J. Catal., 240, 222-234 (2006)

# High Energy Resolution Fluorescence (HERF) XANES (ESRF)



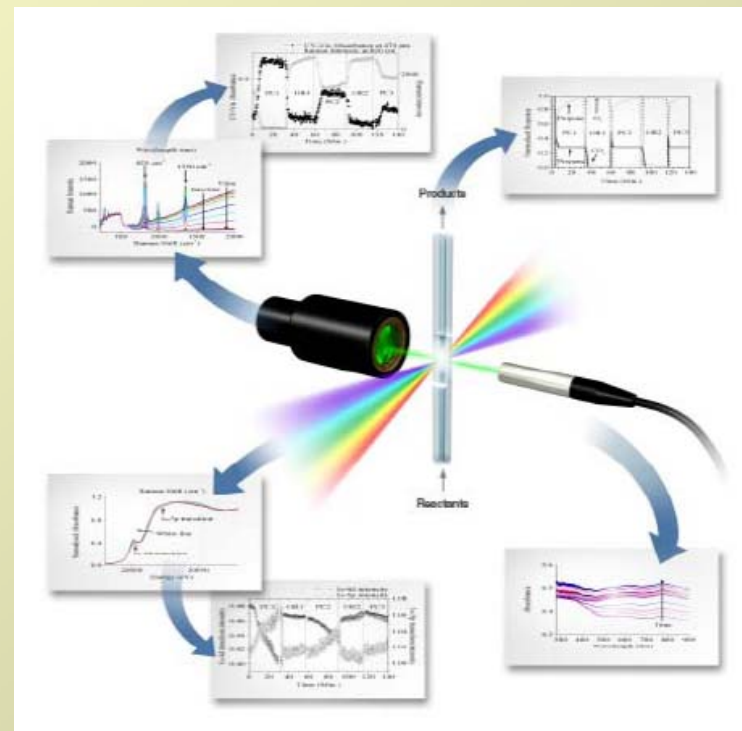
van Bokhoven (ETH), et al.

Angw. Chem. Int. Ed., 45, 4651 (2006)



## ***Desirable New in situ Capabilities***

- **Combined techniques: metals and adsorbates**
  - Synchrotron (EXAFS, XRD, PDF, SAXS)
  - Non-synchrotron (kinetics, IR, Raman, UV-Vis, etc.)
- **Higher signal to noise**
  - Fast acquisition (1  $\mu$ sec) XAFS
  - Fast kinetics
  - Lower concentrations
  - Unprecedented structural detail
- **High-energy-resolution fluorescence XANES**
- **Surface selective structural methods**
- **More detailed electronic information, e.g., energy and occupancy of d-orbitals**

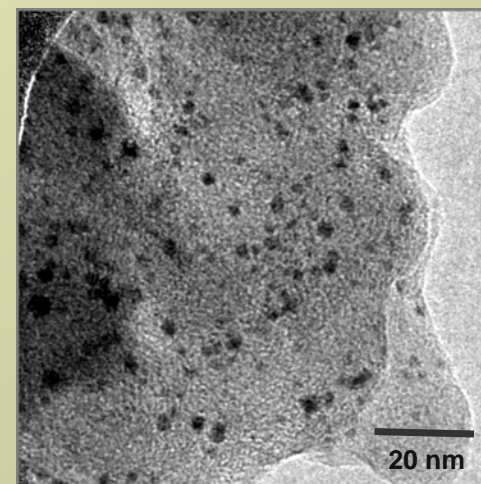
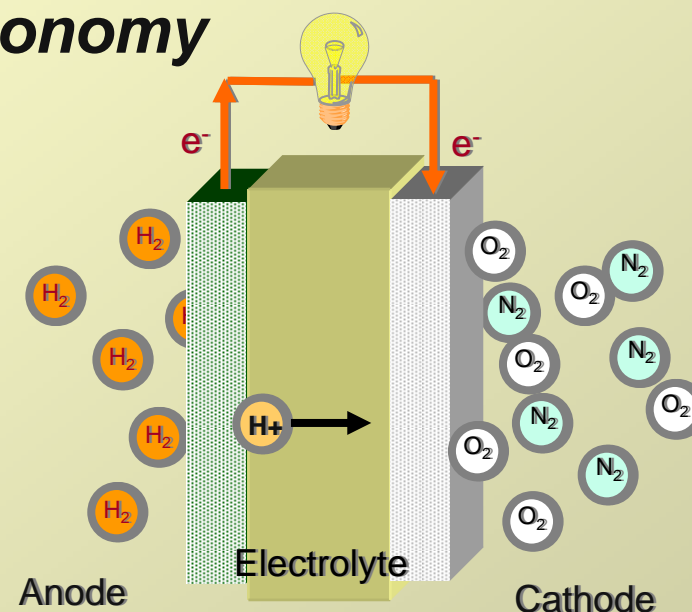


**Combined Laser Raman,  
UV-Vis, EXAFS and Kinetics  
(B. Weckhuysen, Utrecht)**

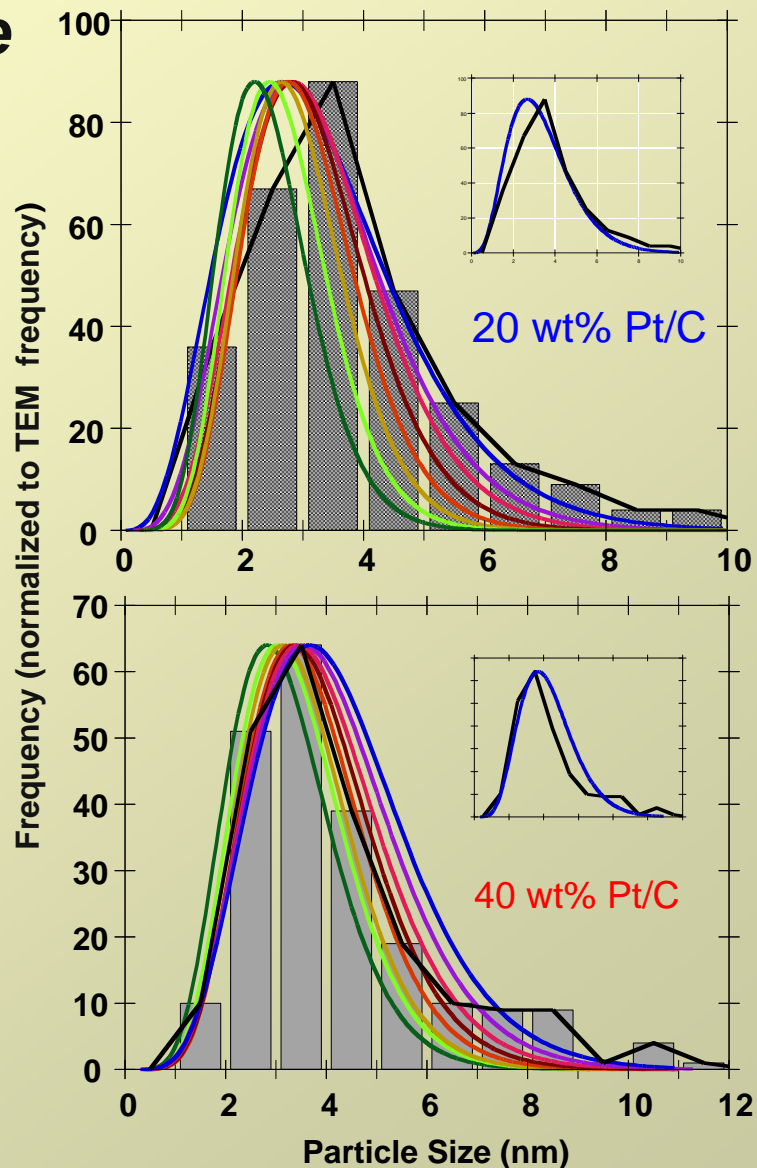
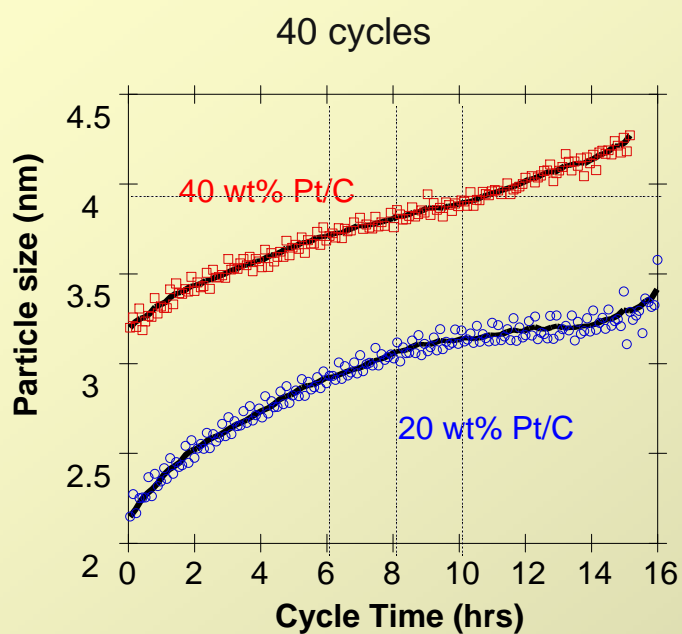


# Electro-catalysts for the Hydrogen Economy

- Hydrogen Fuel Cells are an efficient, new energy source for automotive power
- Current state-of-the-art fuel cell electrocatalysts are nanoparticles of Pt or Pt alloys but have:
  - Have insufficient catalytic activity
  - Composed of expensive metals
  - Lose of performance with time
- Significant improvements can only be made through a fundamental understanding of:
  - Identification of the active site and reaction intermediates
  - Understanding the structural changes that occur during degradation

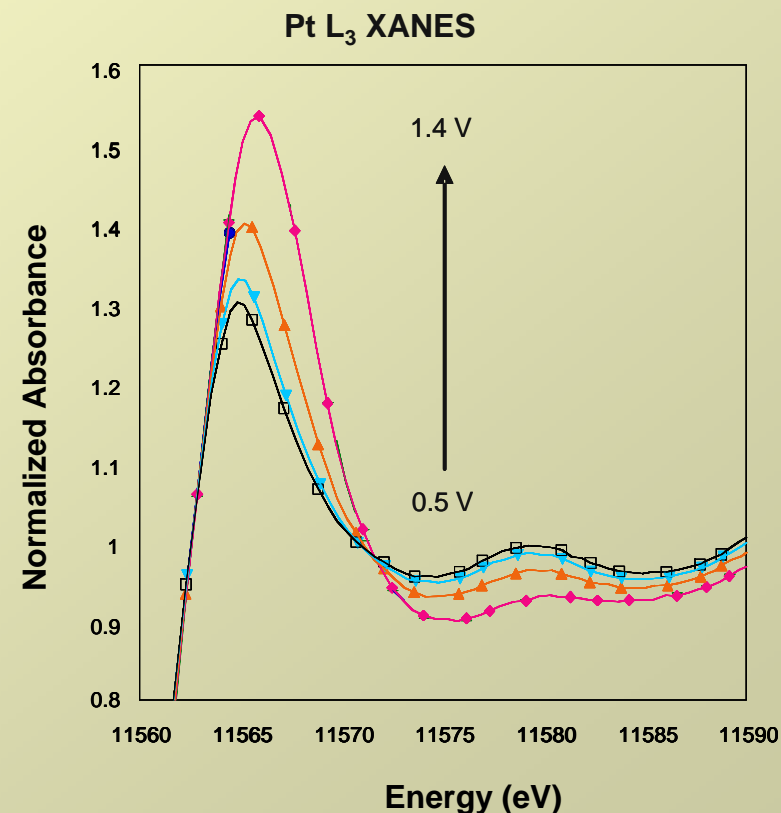


# SAXS studies shows Pt particle growth with cycling



## *Needs for understanding Pt catalyst degradation*

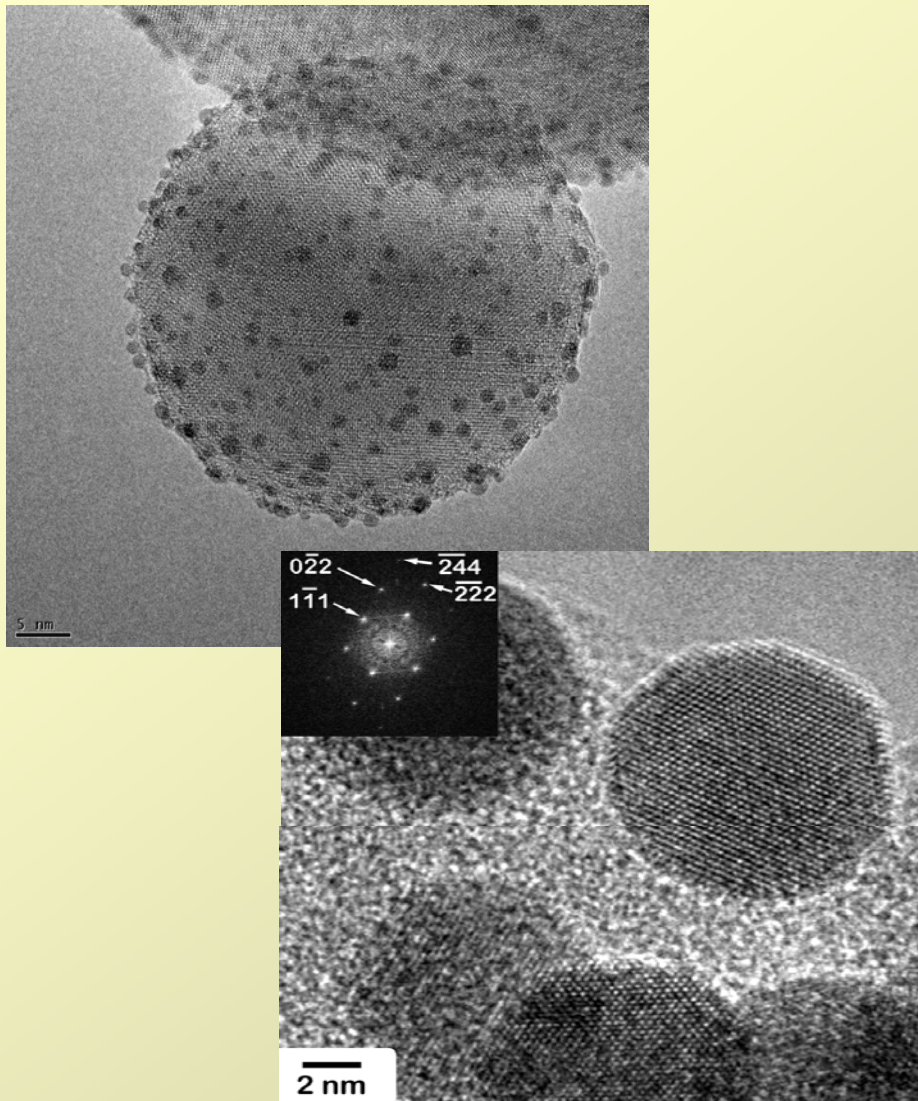
- Determine the structure of the **surface and sub-surface atomic** layers as a function of potential and relationship to dissolution
- Structural determination at **microsecond time-resolution** during operation
- Determine the structure of the soluble Pt species at **ppb** levels



## ***Advanced characterization of Fuel Cells***

- Simultaneous **spatio-temporal resolved** ( $\mu\text{m}$  and  $\mu\text{-sec}$ ) structural, electronic, and particle size characterization for a wide range of metals (Cr-Cu, 6-9 keV, to Pd, 24 keV)
- Resolution of binding of oxygen and oxygenated intermediates on surface sites of 2-6 nm bimetallic nanoparticles
- EXAFS on high-Z element K-edges (e.g., Pt at 78.4 KeV)
  - Deeper penetration through thick layers of low-Z materials
  - Longer exposure times due to **less damage**
  - More accurate determination of metal **oxidation state**
- Oxidation state and structure of the metal species in aqueous electrolyte at **<10 ppb**

# Imaging and Diffraction of Pt Nano-particles In Situ, Atomic Resolution TEM (Purdue)

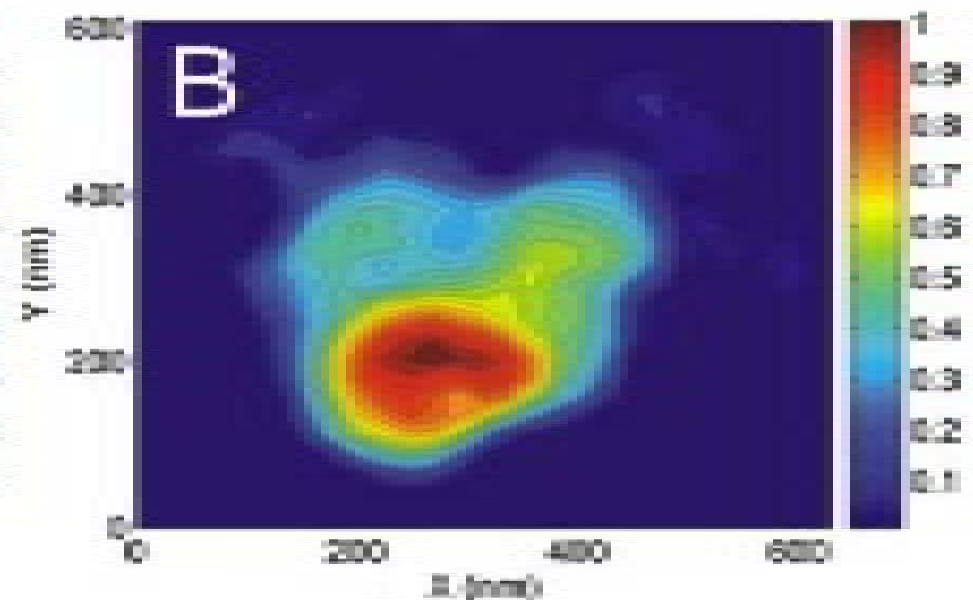
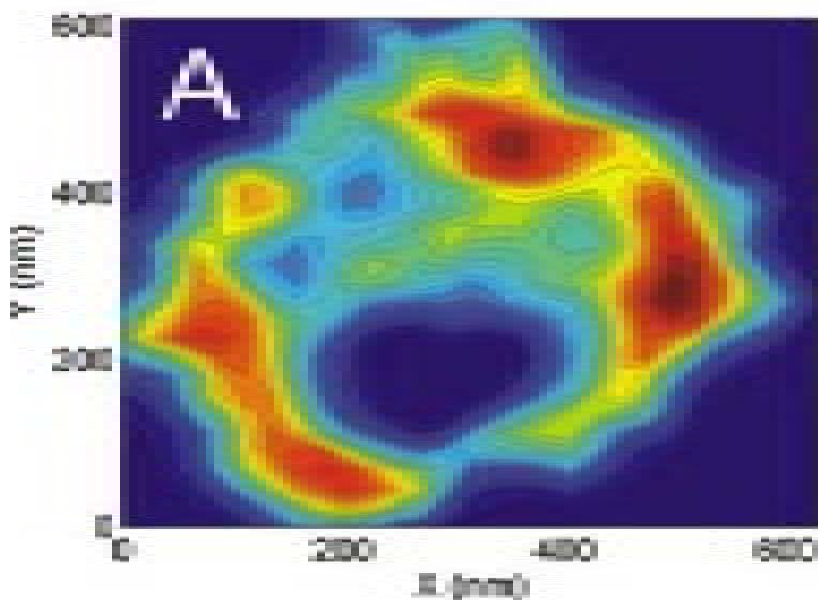
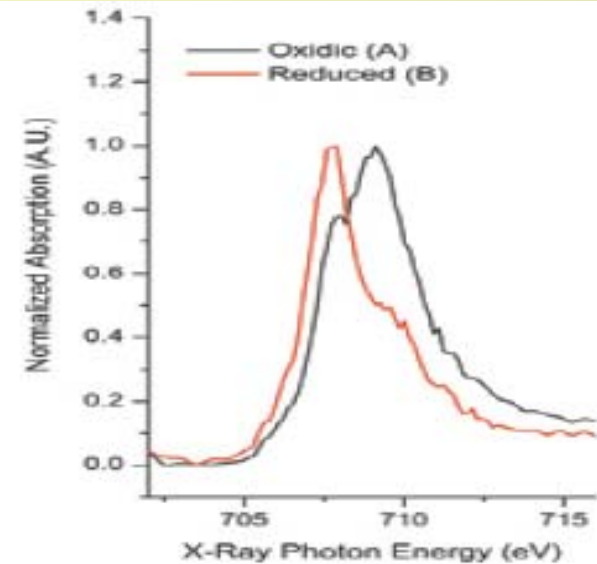


- X-ray imaging on Pt Nano-particles
  - Morphology and changes with reaction conditions
  - Inter-atomic distances
  - Novel bimetallic structures
  - Surface structures
  - Oxidation states
- Diffraction of Nano-particles?
  - Mixed phases
  - Alloy structure
  - Novel bimetallic structures
  - Structural changes under reaction conditions

# In-situ reduction of $Fe_2O_3$ Single Particle (BESSY)

In-Situ Micro-reactor:  
1 bar, **750K**  
Reduction +  
Fischer-Tropsch  
(CO+H<sub>2</sub>)

Frank de Groot  
Utrecht U, Netherlands



## *Imaging Needs for Catalysis*

- **Analysis under reaction conditions**
- **Atomic resolution**
- **Elemental composition**
- **Chemical information (oxidation states, DOS)**
- **Structural identification and quantification of multiple phases and complex (novel) structures**
- **Time resolved analysis, i.e., dynamic imaging**

## ***Closing Thoughts: Chemical Sciences and Engineering***

- **Energy demands are expected to double in the next 25 years**
- **New energy solutions are driven by new materials with novel properties**
- **Future materials will need to have rates several orders of magnitude higher than existing materials**
- **Future chemical reactions will need significantly higher selectivity**
  - **Higher yields**
  - **Fewer, less costly separations**
- **These advances will result from deeper understanding of the structural and electronic properties of the active site and how these interact with the reactants**
- **Synchrotron methods will be critical to the success of this future**