

# Nighttime Tropospheric Chemistry

Nocturnal Reactions of  $\text{NO}_x$ ,  $\text{O}_3$ , VOC and Aerosol

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# Ozone (O<sub>3</sub>) and Air Quality

## Relevance

Health Impacts

Greenhouse gas

## Ingredients

Nitrogen Oxides (NO<sub>x</sub> = NO + NO<sub>2</sub>)

Produced mainly from fossil fuel combustion

VOC (Volatile Organic Compounds)

Both anthropogenic and natural sources

Sunlight !

Key steps = radical formation, NO<sub>2</sub> photolysis

## Scientific Requirements

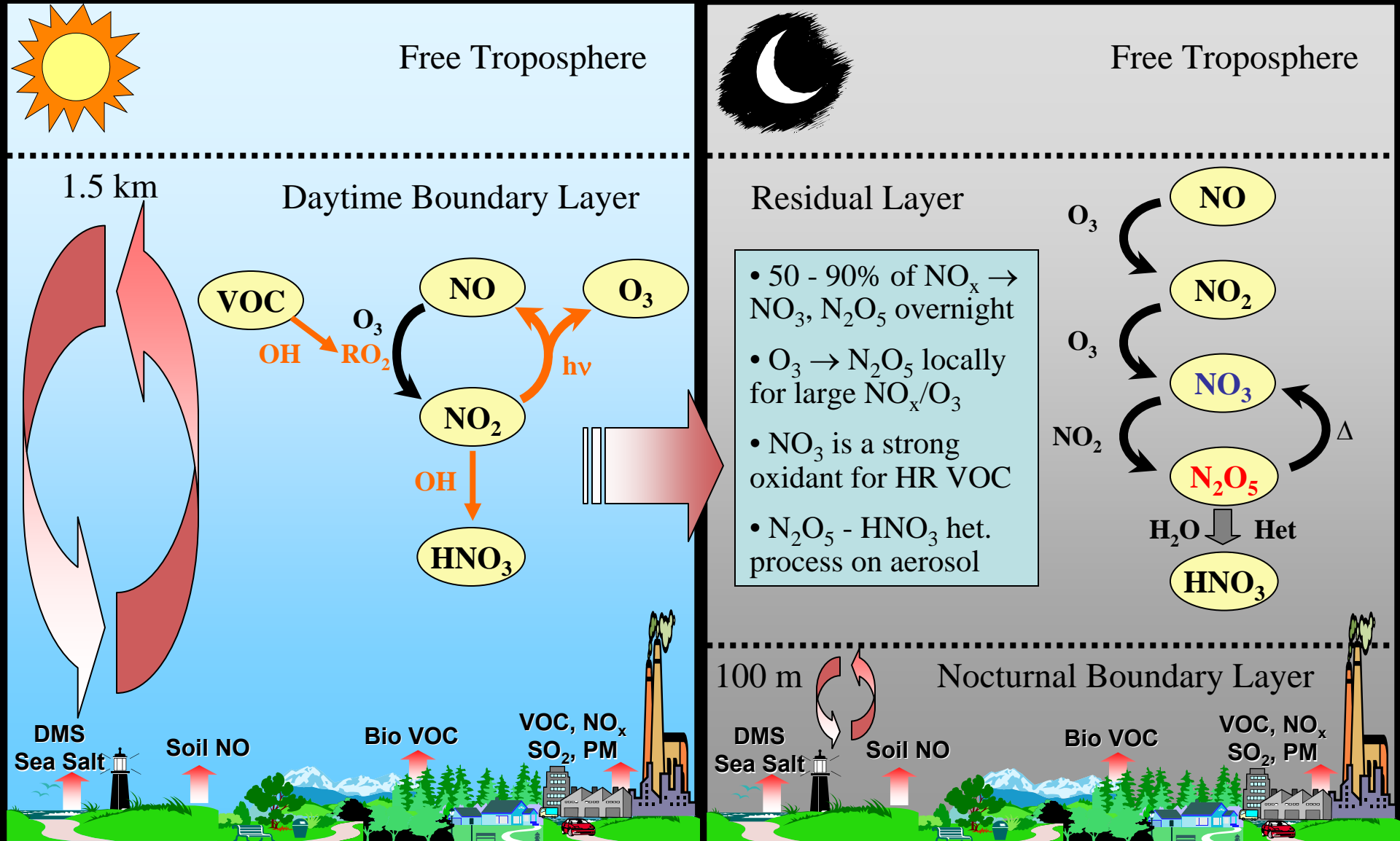
Emissions

Process Chemistry

Meteorology & Transport

QuickTime™ and a  
TIFF (LZW) decompressor  
are needed to see this picture.

# Diurnal Nitrogen Oxide Cycles



NO<sub>x</sub>, VOC, O<sub>3</sub> transformed at night

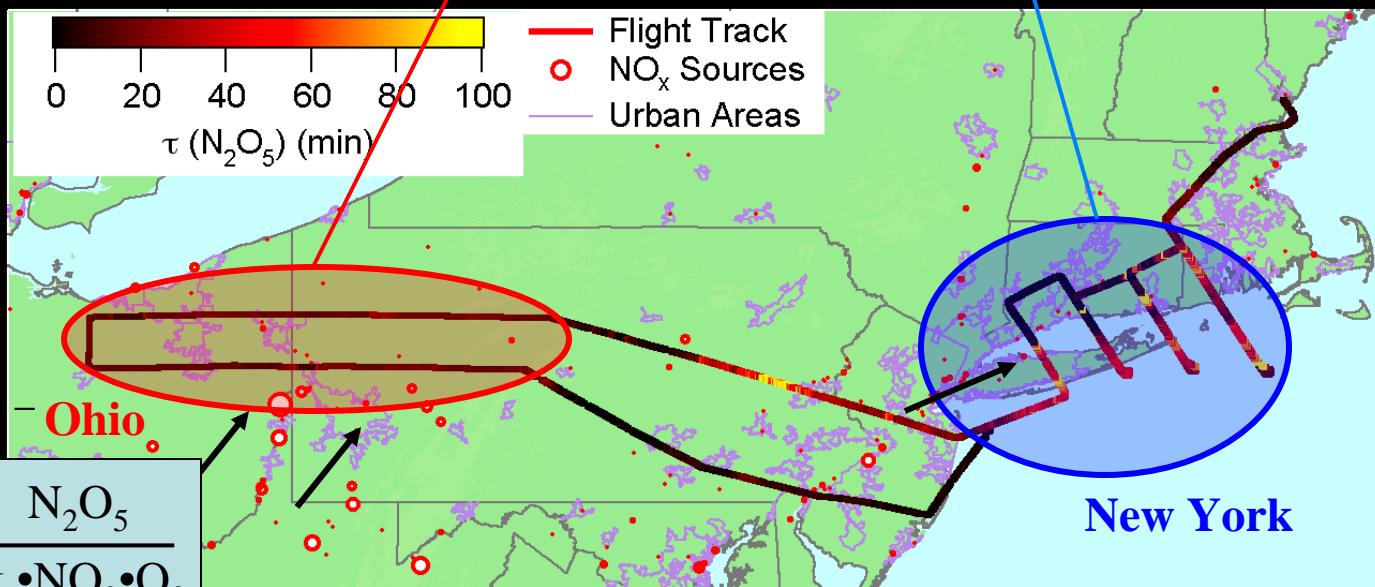
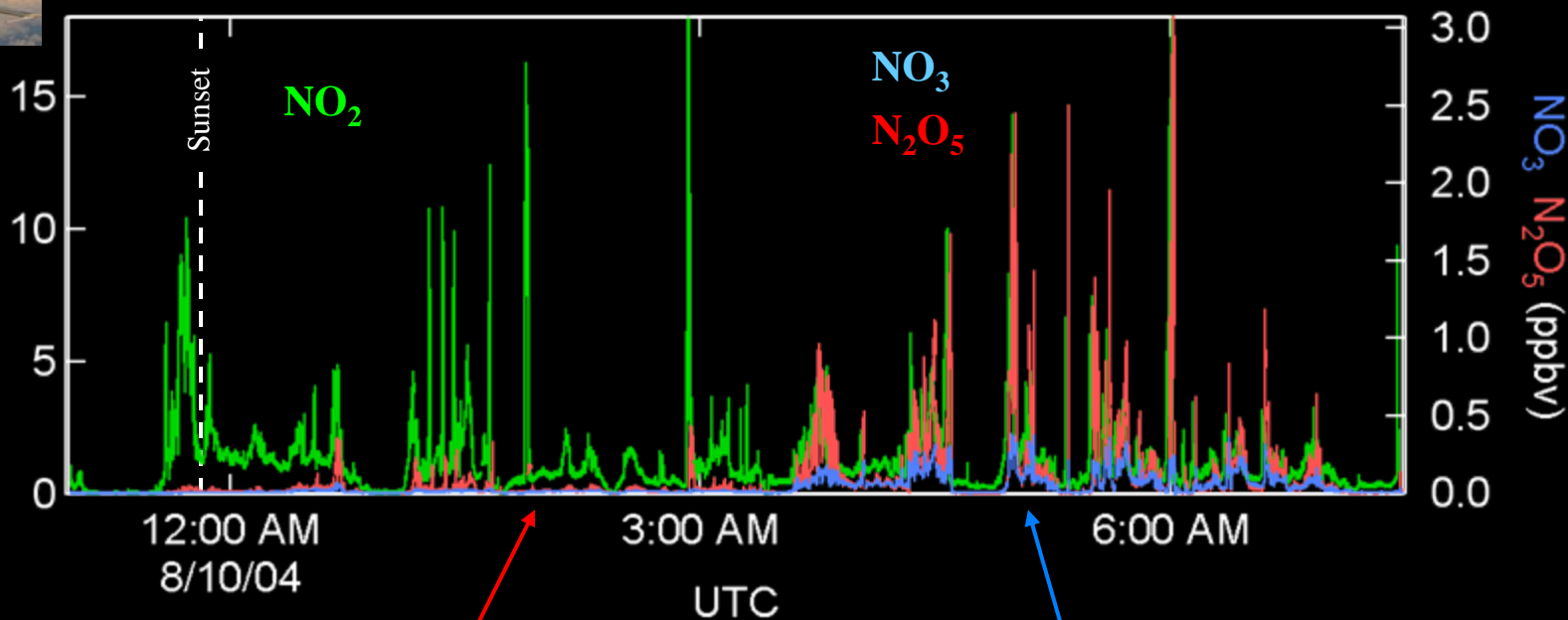
# Nighttime Chemistry - Topics

- Variability in the heterogeneous hydrolysis of  $\text{N}_2\text{O}_5$
- Halogen activation in polluted coastal or marine environments
- Oxidation of marine sulfur emissions in continental outflow
- Daytime vs. nighttime  $\text{NO}_x$  loss in the marine boundary layer
- Laboratory studies of key kinetic parameters
- $\text{NO}_x$ - driven oxidation of biogenic VOC and HR VOC
- Vertical stratification and nighttime chemical transformation
- Seasonal and regional variation of  $\text{NO}_x$  impacts - ozone vs. PM



# Regional Variability in N<sub>2</sub>O<sub>5</sub> Lifetimes

ICARTT  
2004  
NO<sub>2</sub>  
Tom Ryerson



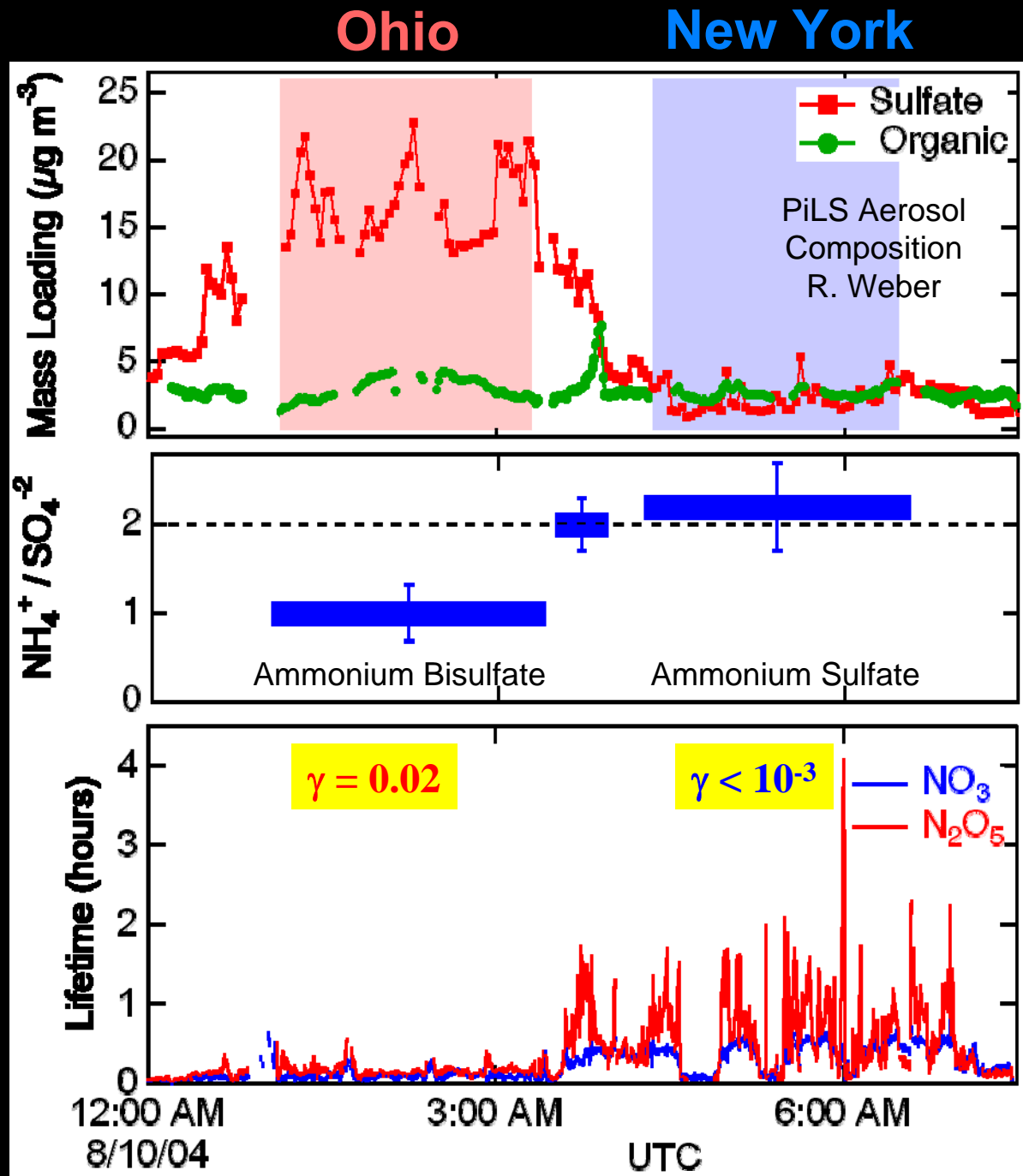
$$\tau(\text{N}_2\text{O}_5) = \frac{\text{N}_2\text{O}_5}{k_s \cdot \text{NO}_2 \cdot \text{O}_3}$$

# N<sub>2</sub>O<sub>5</sub> Uptake & Aerosol Composition

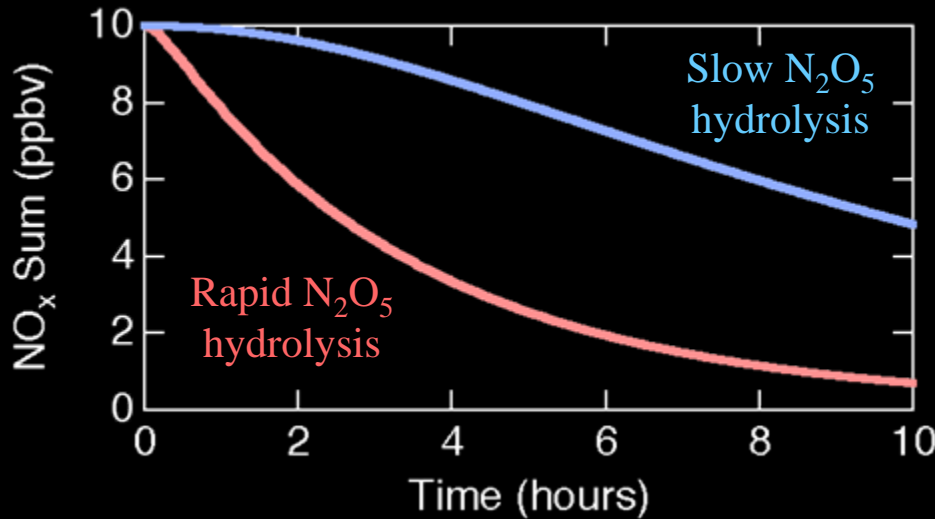
Ohio → New York Changes:

- Sulfate Loading
- Aerosol Organic Fraction
- Aerosol Acidity
- Relative Humidity

Nocturnal NO<sub>x</sub> lifetime strongly correlated to the aerosol loading, particularly sulfate

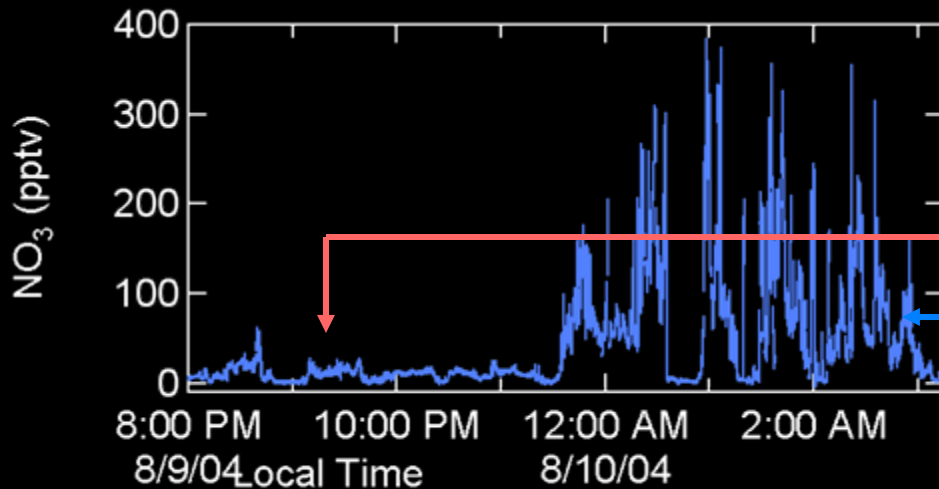
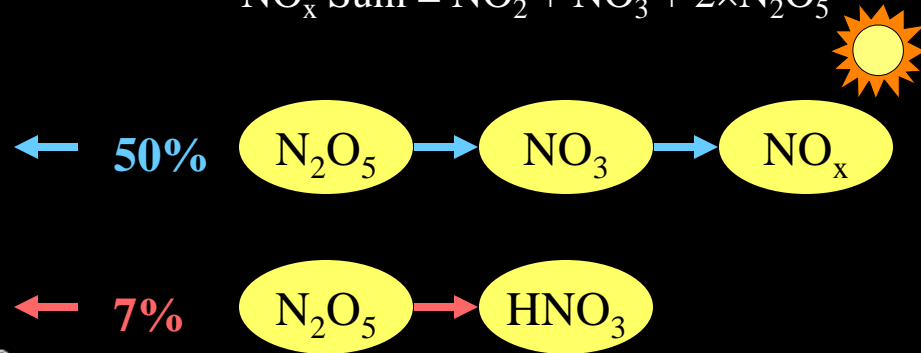


# N<sub>2</sub>O<sub>5</sub> Lifetimes and O<sub>3</sub> precursors



## NO<sub>x</sub> processing overnight

$$\text{NO}_x \text{ Sum} = \text{NO}_2 + \text{NO}_3 + 2 \times \text{N}_2\text{O}_5$$



## Nocturnal Oxidative Capacity

Reactive VOC (e.g. alkenes, aldehydes)

$\gamma = 0.02$   $\tau(\text{VOC}) \sim 60$  hours 15% loss

$\gamma < 0.001$   $\tau(\text{VOC}) \sim 6$  hours 82% loss

Variability in  $\tau(\text{N}_2\text{O}_5)$  affects the 2 components of photochemical ozone

# Payoffs - Variability in $\text{N}_2\text{O}_5$ hydrolysis

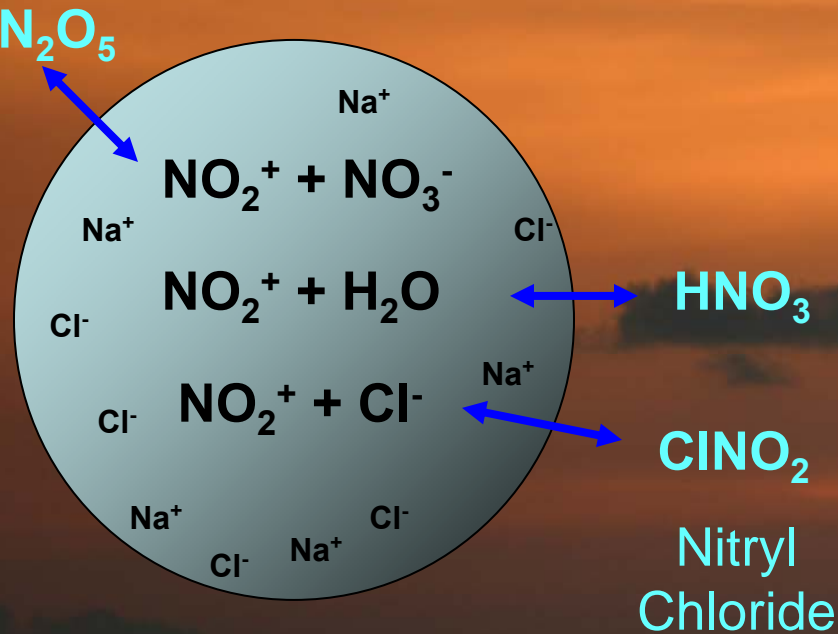
- Actual reactivity of  $\text{N}_2\text{O}_5$  on real atmospheric aerosol much different than suggested by laboratory studies on proxy substrates (e.g., inorganic salts)
- Uptake of  $\text{N}_2\text{O}_5$  to aerosol a key input to regional air quality models for both ozone (summer) and particulate matter (winter)

## Future

- Expand field database, develop parameterization of  $\text{N}_2\text{O}_5$  uptake from field measurements, compare to laboratory studies and implement in models



# Halogen Activation in Polluted Marine Environments



Behnke et al., JGR 1997  
 $\Phi(\text{ClNO}_2) \approx 1$ ,  $[\text{Cl}^-] \geq 1 \text{ M}$

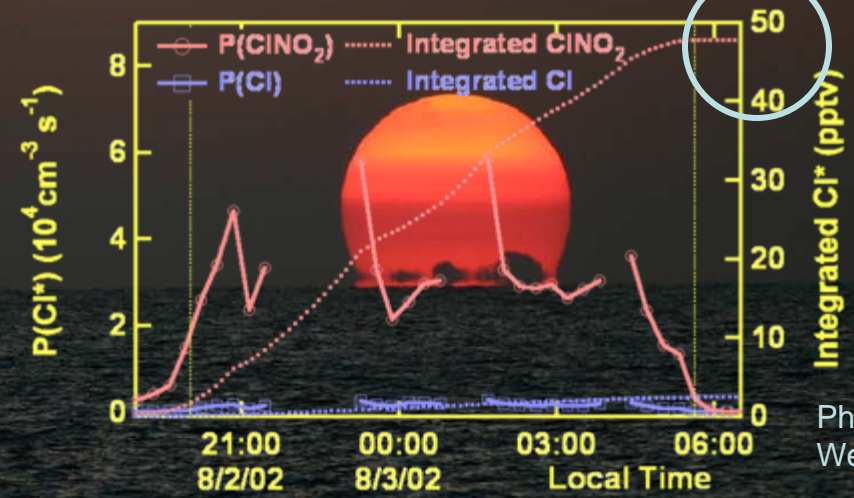
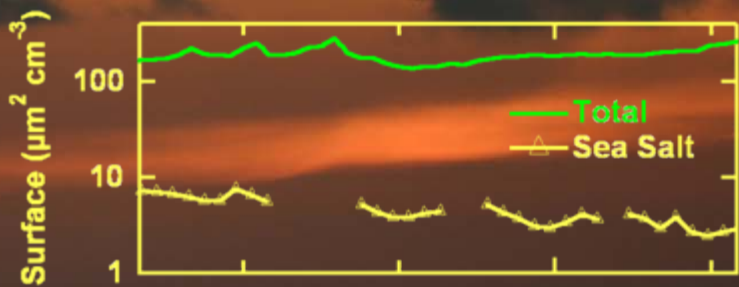
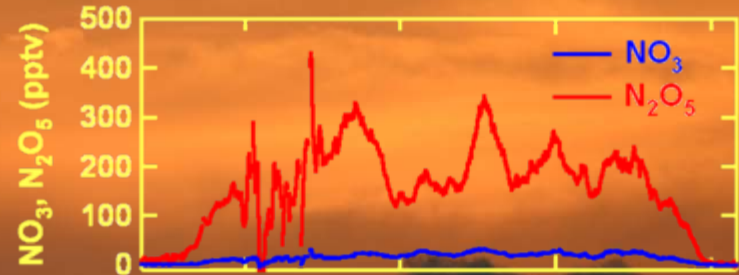
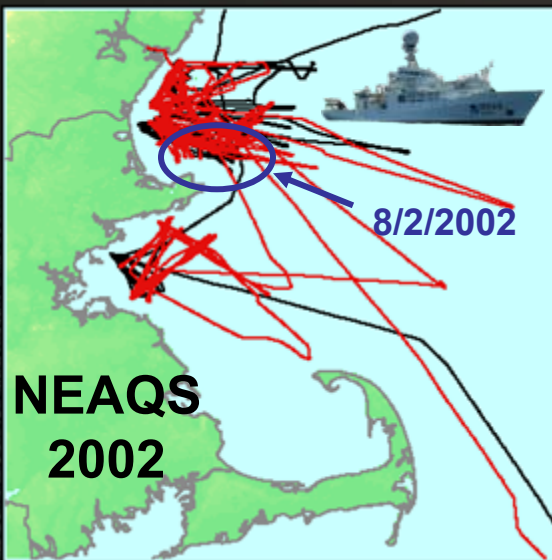


Photo: Dan Welsh-Bon

Simple calculation: Modest (50 pptv integrated) halogen activation even for strong  $\text{N}_2\text{O}_5$  production

# TexAQS / GoMACCS 2006: $N_2O_5$ & $ClNO_2$

July 26 - September 15, 2006

$ClNO_2$ : CIMS

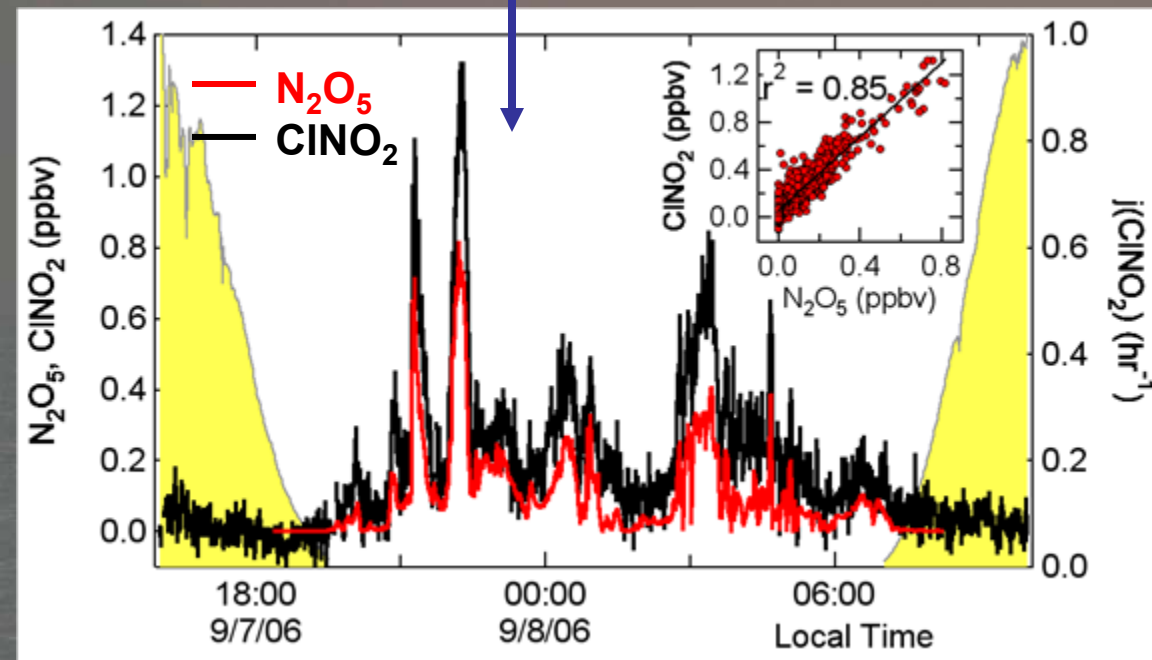
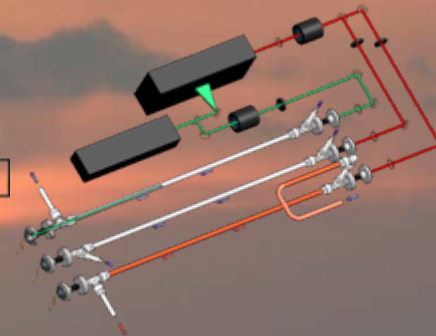
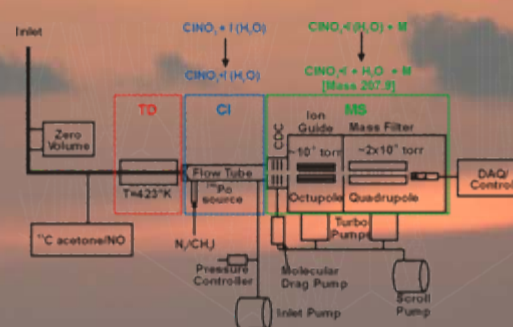


Jim Roberts

$N_2O_5$ : CRDS

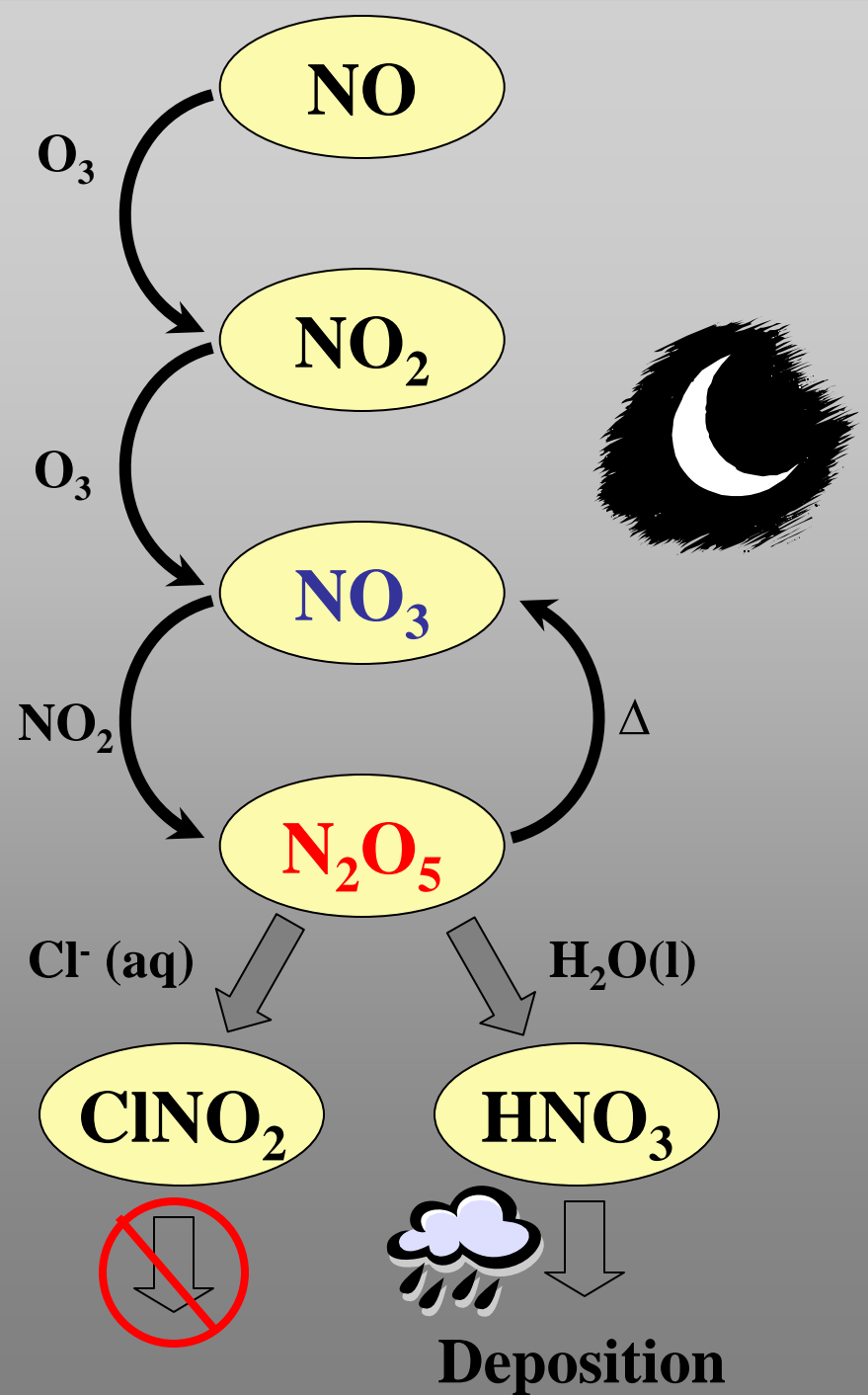


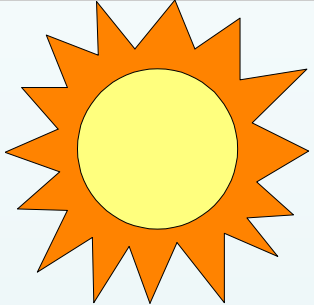
Hans Osthoff



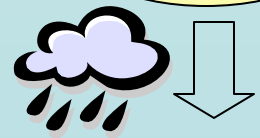
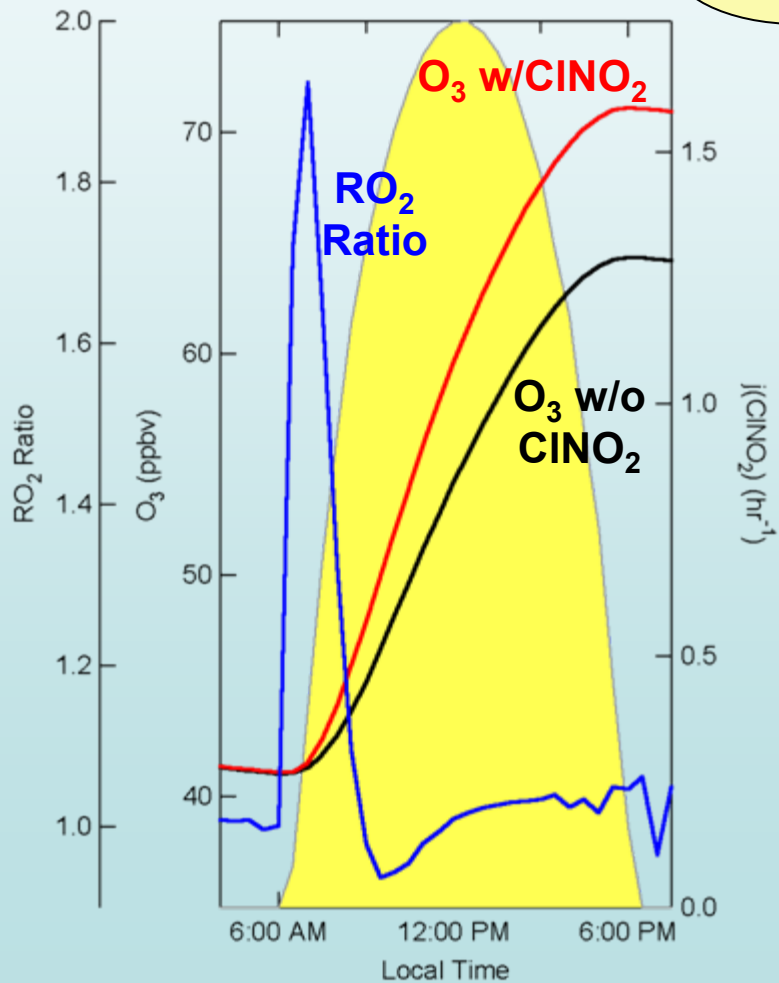
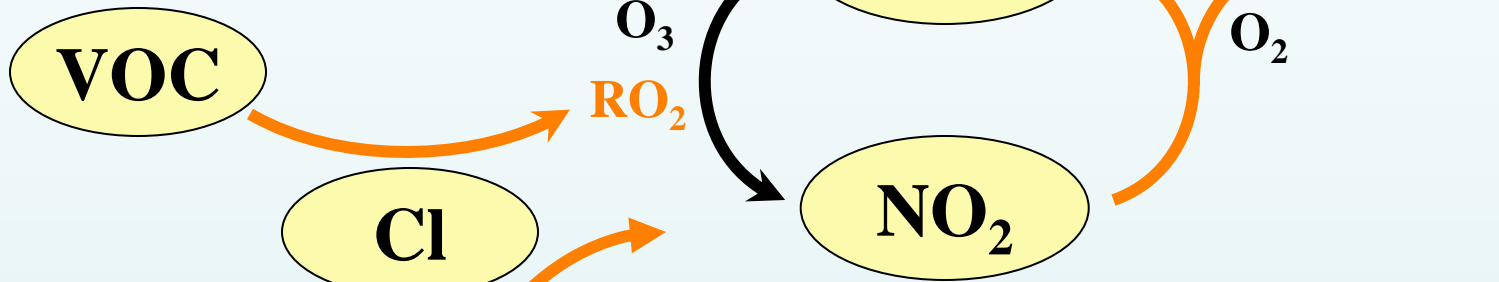
- Observed  $ClNO_2$  levels inconsistent with  $N_2O_5$  uptake on sea salt alone !
- Implies role for submicron aerosol
- High efficiency for  $ClNO_2$  production: up to  $\sim 50\%$  yield per  $NO_x$  oxidized

# Implications





# Implications



**Deposition**

# Payoffs - Halogen Activation

- Efficiency of nocturnal halogen activation from  $\text{N}_2\text{O}_5$  much greater than previously recognized
- May have regional effects on oxidant chemistry and ozone formation relevant for air quality modeling in coastal areas

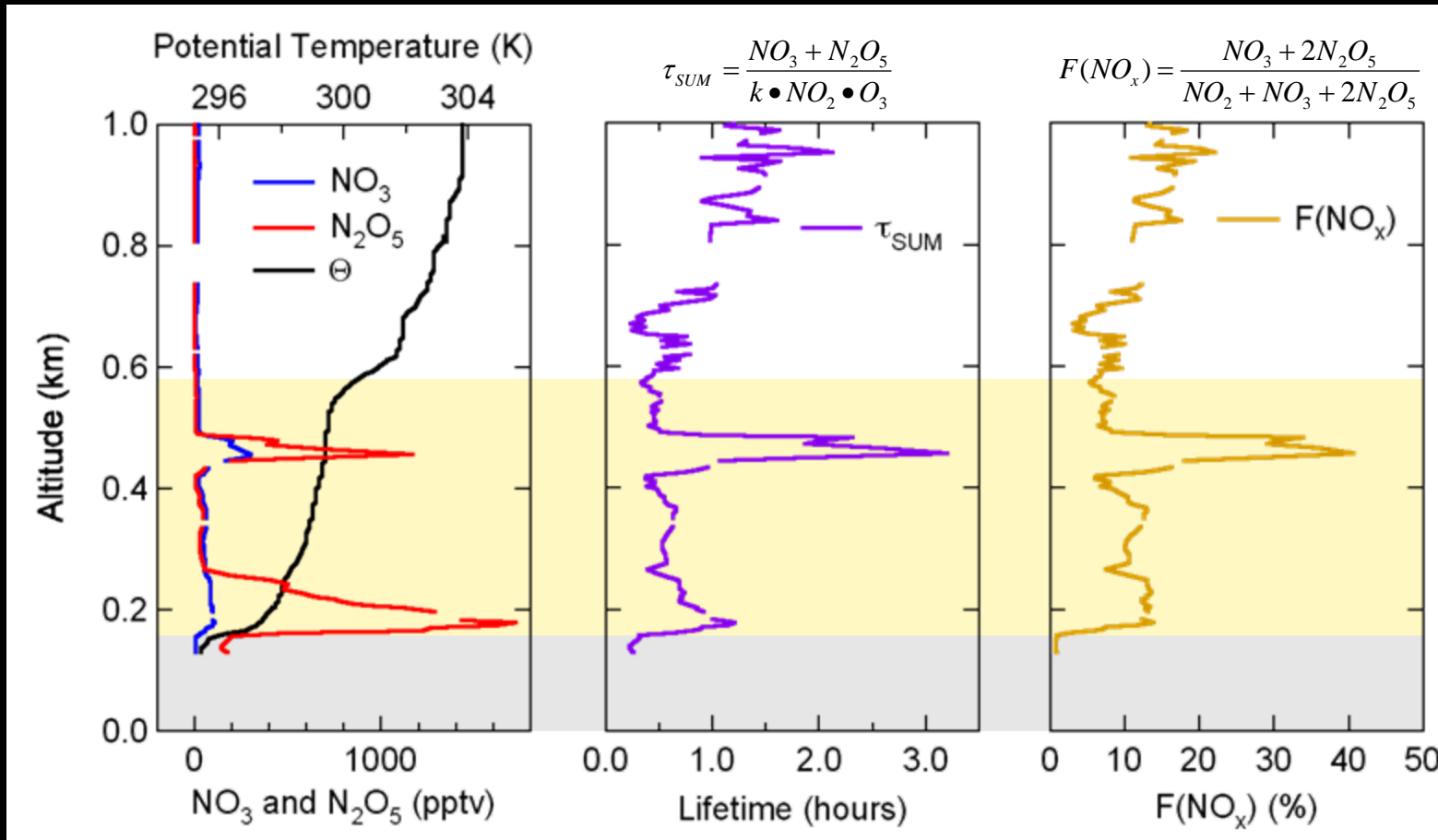
## Future

- Impact in winter coastal New England, the Arctic (2008) and California (2010)

# Vertical Stratification & Nighttime Chemistry

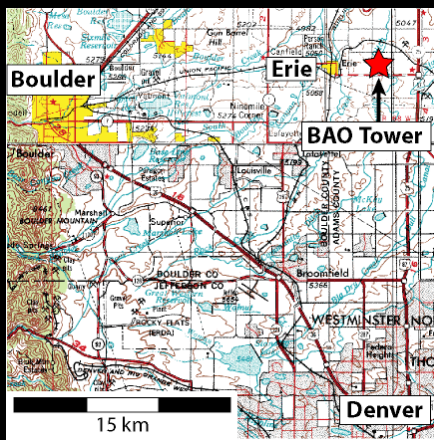


TexAQS  
2006

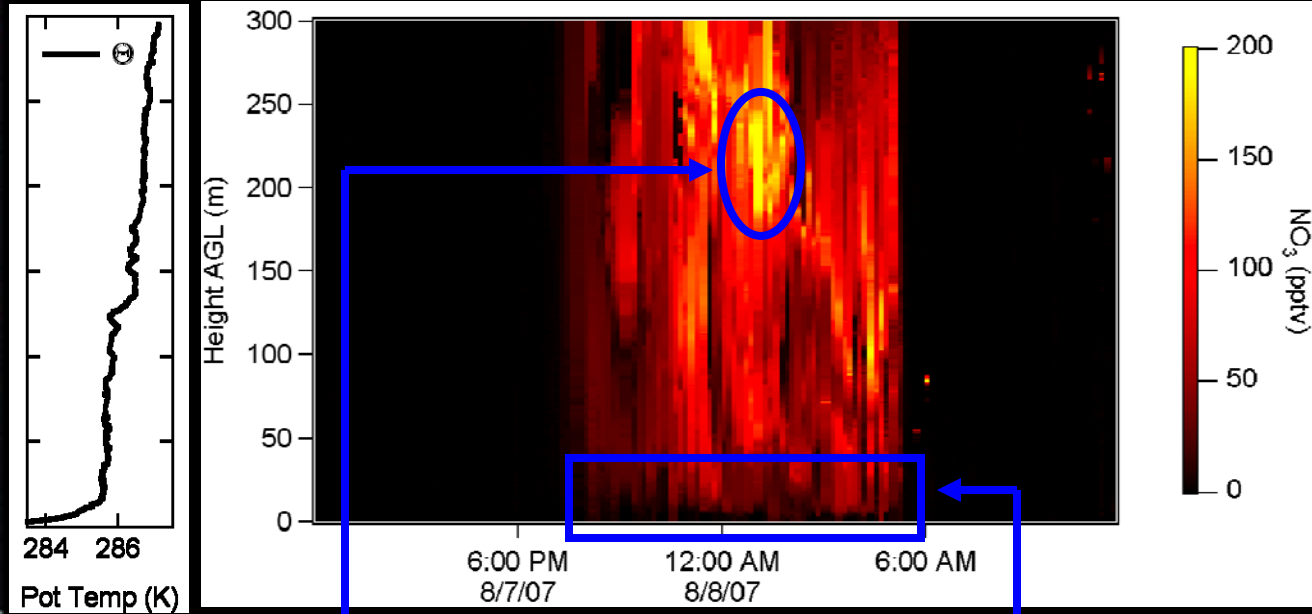


- NO<sub>x</sub> & VOC plumes occur in discrete layers at night
- Chemical transformation within different layers differs markedly

# Boundary Layer Vertical Profiles Erie (BAO) Tower



- 300 m w/ vertical resolution  $\sim 0.5$  m
- Movable carriage on *outside* with  $> 1$  ton payload
- Studies in 2004 (fall) and 2007 (summer)



- Large  $[\text{NO}_3]$  routinely observed aloft
- Often associated with complex layering

Surface layer commonly observed

# Payoffs - Vertical Profiles

- Large differences observed between measurements at surface level and aloft - aircraft, tower measurements change view of nighttime chemistry
- High resolution vertical data shed light on interaction between emissions, vertical stratification and chemistry

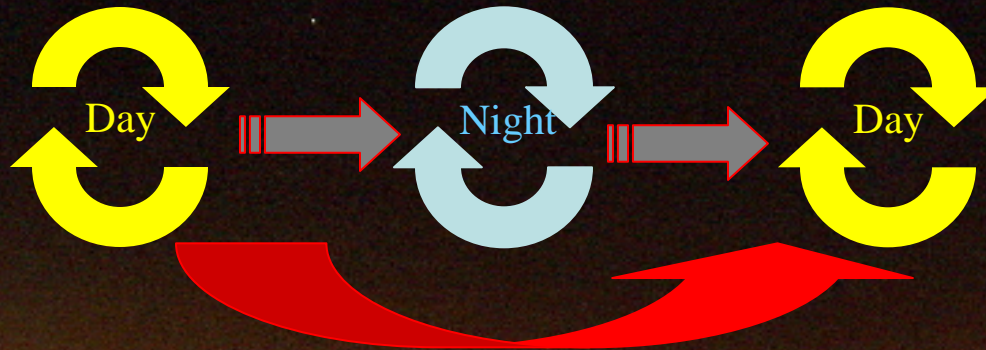
## Future

- Future: Incorporation of micrometeorology and modeling to data analysis; winter measurements



# Conclusions

## The Dark Side of Atmospheric Chemistry



“Photochemical”  $O_3$  & PM production depends on more than just *photo*-chemistry

Nocturnal reactions transform the mixture of  $NO_x$ , VOC, radicals,  $O_3$  and aerosol

Nighttime chemistry, transport and mixing are complex, interrelated problems