

UNITED STATES
NUCLEAR WASTE TECHNICAL REVIEW BOARD

SPRING BOARD MEETING

May 15, 2007

Crown Plaza Hotel
1480 Crystal Drive
Arlington, Virginia 22202

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Dr. David Duquette
Dr. B. John Garrick, Chair, NWTRB
Dr. George Hornberger
Dr. Andrew Kadak
Dr. Ronald Latanision
Dr. Ali Mosleh
Dr. William Murphy
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P R O C E E D I N G S

8:00 a.m.

GARRICK: Good morning.

On behalf of the Nuclear Waste Technical Revenue Board, let me welcome you to our second meeting in 2007. The meeting is completing with a couple of our major meetings, something we didn't know about when we were scheduling and planning the meeting. But, I think, still, we are going to be able to have a very interesting day.

As you know, our custom is to introduce the Board members individually. We are going to depart from that process a little bit today, and just identify the Board members and announce to the audience that we have gone to a little different practice here of supplying resumes or bios of each of the Board members, as well as each of the speakers. And, those will be available to anybody who wishes to have one at the table in the back of the room.

But, I would like to at least allow you to be able to match the face with the name, so I am going to name each of the Board members, have them raise their hand, so we can do that.

I'm Chairman of it, and my name is John Garrick, and the first one I'd like to have raise their hand is Mark Abkowitz. Howard Arnold. Thure Cerling. David Duquette. George Hornberger. Andy Kadak. Ron Latanision. Ali Mosleh.

1 William Murphy. And, Henry Petroski. We will be
2 introducing the speakers as they speak.

3 I would like to recognize one distinguished group
4 that's with us today, and that's the Advisory Committee on
5 Nuclear Waste, which I understand is now the Advisory
6 Committee on Nuclear Waste and Materials, something the
7 current Chairman and Committee was able to get through,
8 something the past Chairman and Committee was pushing for,
9 but was unable to get through, so, we congratulate the
10 Committee for its change. That's something that's very much
11 needed in the Nuclear Regulatory Commission.

12 What I'd like to do now is, as a practice, to
13 review some of the areas of major interest to the Board, not
14 necessarily our current total priority list. We are usually
15 influenced by the recent meetings, and topical meetings as
16 well. I'll try to highlight a couple of things that we are
17 having to follow, and have a great deal of interest in, and
18 I'll do it in accordance with the categories that we have
19 more or less associated our issues with, namely the
20 preclosure period, the postclosure period, and the
21 integration issues associated with the Project.

22 And, of course, as has been for quite some time,
23 during the preclosure period, transportation and the design
24 of the surface facilities stand out as areas of great
25 interest to the Board. The Board was updated on both

1 subjects at its last meeting in January, and we expect to
2 return to these topics again in the near future, as more
3 information becomes available.

4 The Board has also been particularly interested in
5 the Department of Energy's proposal to adopt a transportable,
6 aging, and disposal canister. The TAD canister system could
7 reduce the number of times individual assemblies are handled
8 because the canister and its contents would be handled in
9 pretty much a single action. This could also improve
10 facility throughput at the repository and reduce the
11 potential for accidents during handling operations.

12 The TAD canister system also has the potential to
13 simplify the design and reduce the cost of repository surface
14 facilities. However, as we have pointed out in our letter,
15 in our reports, the success of TADs is very dependent on such
16 uncertainties as the actual startup date of the repository,
17 the availability of a Nevada branch rail line, and decisions
18 on such spent fuel operations as the on-site storage at
19 nuclear power plants and proposed spent fuel storage on pads
20 at the repository site.

21 Consideration of these and other factors concerning
22 the handling of spent nuclear fuel will be necessary to
23 demonstrate the real value of the TADs concept.

24 Regarding postclosure performance, the primary
25 interest of the Board is establishing a sound technical basis

1 for a realistic assessment of the radiation doses that could
2 result from the repository. During past Board meetings, we
3 have often identified areas within the Total System
4 Performance Assessment that could include varying, and
5 sometimes unknown, degrees of conservatism. While this may
6 be acceptable for purposes of licensing, we have indicated
7 several times that a more realistic assessment would help
8 generate a sense of confidence, both within the technical
9 community and the general public, confidence that the
10 facility would indeed perform its intended function.

11 An important component of a realistic dose
12 assessment is knowledge of the chemical conditions within the
13 repository system, especially in the near-field environment.
14 The physical chemistry model of the waste mobilization
15 process in the near field is a continuing area of interest to
16 the Board. As a matter of fact, today, we have invited a
17 guest speaker to speak about the use of depleted uranium
18 oxide as a chemical barrier. And, our purpose here in
19 inviting this presentation is not to promote the use of
20 depleted uranium in a Yucca Mountain repository. It is to
21 improve our fundamental understanding of the chemistry in the
22 near field as a basis for enhancing our confidence in the
23 dose assessment of the repository. This afternoon, we will
24 also have a presentation by a representative of the Yucca
25 Mountain project on the evaluation of the near-field

1 chemistry.

2 A third major area of interest to the Board is
3 integration, both within the project and with other entities,
4 such as nuclear power plant operators that must interact with
5 the repository, or with the project. Many of our
6 presentations today support the Board's choice to review the
7 overall integration of project activities in addition to our
8 interest in the technical validity of the specific activities
9 or analyses discussed in each presentation.

10 A key factor in integration is the thermal
11 management strategy that will be adopted for the Yucca
12 Mountain repository. The subject of one of our presentations
13 today, the proposed Global Nuclear Energy Partnership, has
14 the potential to significantly affect thermal management, and
15 we are especially interested in exploring how that proposal
16 could affect design and operation of the overall waste
17 management system.

18 Let me talk a little bit about the agenda itself.
19 As is customary, we will begin with an overview, both of the
20 overall Office of Civilian Radioactive Waste Management
21 program and, more specifically, the Yucca Mountain project.
22 That overview will be followed by the presentation I just
23 mentioned on use of depleted uranium oxide as a chemical
24 barrier. Following a break, we will hear presentations on
25 the Inyo County drilling program and on the second generation

1 waste package design work being done at the University of
2 Nevada, Reno.

3 After lunch, we will have a presentation on the
4 Global Nuclear Energy Partnership and, particularly, the
5 waste streams that might be generated by the facilities
6 proposed for the Partnership. Then, we will begin the Yucca
7 Mountain presentations with talks on saturated zone testing
8 and on near-field chemistry. And, following the mid-
9 afternoon break, we will have a presentation on waste package
10 design and prototype development, and then an update on
11 probabilistic volcanic hazards analysis.

12 Now, as usual, following the presentations, we have
13 scheduled time for public comment, an aspect of our meeting
14 that is very important to us. If you would like to comment
15 at that time, please enter your name on the sign-up sheet at
16 the table near the entrance to the room. Of course, written
17 comments are always welcome, and will be made part of the
18 record.

19 In that connection, we have been asked to
20 accommodate some public statements prior to our afternoon
21 break to allow some people to attend another meeting. So,
22 that will be a slight departure from the agenda as it now
23 reads.

24 Some of you have asked about questioning during the
25 course of the presentations. Our preference is for you to

1 write down your questions and submit them to either Davonya
2 Barnes or Linda Coultry in the back of the room, and we will
3 cover as many as we can during the discussion period.

4 As we get into the presentation and discussion part
5 of our meeting, it is important that everybody understand or
6 is clear about the distinction between member opinions and
7 official Board positions. Board meetings are very
8 spontaneous. We want them that way. We want the exchange to
9 be as free and as open as it can possibly be. We express
10 ourselves pretty much as we like, and that is the process
11 we'd like to continue. So, when Board members speak that
12 way, or extemporaneously, it is important to realize that we
13 are speaking on our own behalf, and not on behalf of the
14 Board. And, we'll try to distinguish between the two
15 situations.

16 Also, we urge that all of you put your pagers and
17 your cell phones on the silent mode.

18 And, I have a reminder here too, especially for the
19 Board members, that it's very important when you speak, to
20 speak into the microphone, and not turn away from the
21 microphone while you're making your comments. And, you also
22 have to push a button in this case to speak. There is a
23 speak button. If it's like this one, it's on the right side.

24 All right, I think with that, we will proceed with
25 our presentations, and our first presentation today to give

1 us the overview is going to be Chris Kouts of the Department
2 of Energy. Chris?

3 KOUTS: Thank you, Dr. Garrick.

4 I have to say it's somewhat de javu, not only being
5 before the Board, but as I was mentioning to attendees from
6 the State of Nevada in walking over from the Metro, this is
7 where I started my federal career in Crystal City working for
8 the Environmental Protection Agency over 30 years ago. So,
9 it was kind of a strange feeling walking over here, but
10 nonetheless, it's the Ides of May, and I'm very happy to
11 present a status as to where the program is.

12 I'm going to give you a status of the key project
13 issues, and an overview of the areas of senior management
14 attention. I will say that our Director, Ward Sproat very
15 much wanted to be here, but Ward's duties essentially
16 required him to be out in Nevada this week, and he sends his
17 apologies, and I'm his substitute. I hope I will be able to
18 answer any questions, and provide the information to the
19 Board that they're interested in in terms of the status of
20 the program.

21 The Board is fairly familiar, I believe, with the
22 schedule of the program. It has not changed. We are still
23 working very diligently to complete and submit a license
24 application to the Nuclear Regulatory Commission no later
25 than June 30th of next year. That will require certain steps

1 along the way. We hope to finalize our design by November of
2 this year, have LSN, the licensing support network
3 certification no later than the end of December of 2007, in
4 order to meet the NRC requirements.

5 We're working on a supplemental EIS right now, and
6 we hope to have that also completed prior to the time that we
7 submit the license application.

8 The subsequent licensing dates are all basically in
9 the hands of the Nuclear Regulatory Commission, and you are
10 probably aware of the three year statutory requirement for
11 the NRC to make a decision, and they can go to Congress,
12 obviously, for another year, but, essentially, these dates,
13 except for the rail lines, are essentially contingent on the
14 ability of the Department and its application, to satisfy NRC
15 issues and concerns in relation to the licensing process.

16 Also, as we should always caveat when we talk about
17 2017 in terms of the start date of the repository, that's
18 contingent on several factors. One is adequate funding for
19 the program, and the other is enactment of the
20 administration's legislation that was resubmitted earlier
21 this year.

22 If we could go to the next slide, I think the Board
23 is familiar with these. Our Director has established four
24 strategic objectives, the first of which I've already
25 mentioned, which is to submit a license application no later

1 than June 30th of next year.

2 The second one is one that he has a personal
3 interest in, which is essentially to put the program in a
4 position such that the skills and culture needed to design,
5 license and manage the construction and operation of the
6 project, if you will, with safety, quality and cost-
7 effectiveness, are in place. And, he's basically attacked
8 this issue with a great deal of interest and zeal, and I
9 think we're seeing changes within the program. We've had a
10 lot of discussions internally into the program, within the
11 program about where we need to be, not only at the point that
12 we submit a license application, but also at the point where
13 we begin construction and operation, and that's a continuing
14 effort on his part, and will be as long as he's with us.

15 The next strategic objective is one that I am
16 routing very hard for since I generally have been the lead
17 witness in these spent fuel trials, I've testified in all
18 trials to date, there are nine of them. Someone has asked me
19 whether or not I enjoy it, and I think my perspective is
20 after the first couple of trials, the novelty really wears
21 off. There are 55 more trials in the pipeline. We have four
22 settlement agreements. I believe some of the Board staff has
23 asked for some statistics in terms of what those settlement
24 agreements cover, and so forth. But, this is certainly one
25 area that I would love to see some progress made.

1 And, the last I think is to develop and implement a
2 comprehensive national spent fuel transportation plan. I
3 believe Gary Lanthrum, who directs that area of the program,
4 is working very hard on that area.

5 If we can go to the next slide, which is areas of
6 management attention. One of the reasons our Director is not
7 here is that he's conducting a progress status review, which
8 he does on a regular basis. In addition to that, he has set
9 up a group within the program, which I'm a member of, which
10 is a Licensing Strategy Team, who reviews issues related to
11 licensing strategy. We meet on a regular basis, and discuss
12 issues about how the program needs to proceed. And, I think
13 that's helped the process a great deal from the license
14 application standpoint.

15 I believe from an organizational standpoint, he's
16 also embarked on taking some lessons from the private sector
17 and requiring each office director to develop business plans
18 in terms of what milestones and what we hope to accomplish,
19 how those milestones affect other areas of the program. And,
20 we work that very hard within the program.

21 He's also initiated the--actually, the hiring of a
22 substantial greater amount of federal staff, which we're in
23 the process of doing right now. He feels very strongly that
24 we need a much more active federal staff, and we're hiring
25 people on a regular basis in order to do that within our

1 ceiling limits.

2 He's also concerned about developing the management
3 team that we already have, and we've spent a great deal of
4 time on our different aspects of that with the individual
5 directors.

6 From a culture standpoint, he's very concerned
7 about quality, and that's--as he's indicated on many
8 occasions, that quality, schedule and safety are not mutually
9 exclusive. He believes that they can be done at the same
10 time. And, his basic issue with that is if a manager is
11 having problems, he needs to ask for help. And, every
12 manager who needs help gets it, and hopefully, we're moving
13 along to address quality issues.

14 In addition, the corrective action program, which
15 is one that, again, is an item of special attention for the
16 program, is receiving a lot of effort, and trying to
17 streamline the process, and also make sure that when we do
18 make a corrective action, that it deals with the issue and
19 goes to the true root causes of the situation, so we don't
20 have recurrences of the same issue.

21 From a Congressional standpoint, Ward has spent a
22 great deal of time on Capitol Hill educating staff and
23 members of Congress on where the program is and where it's
24 headed, and he's, from his perspective, building a great deal
25 of credibility for the program. And, certain Congress is

1 learning a great deal from those meetings.

2 The status of the budget, which everyone follows
3 very closely who follows this program, our FY '08 budget
4 request is \$494 million. As you're probably aware, we
5 received about \$100 million less than we requested for FY
6 '07, and that's required some reductions in staff. And,
7 that's been the history of the program. Typically, our
8 requests are not fully met by Congress, and the program has
9 to adjust when it finally gets its mark.

10 Status of the revision to the EPA standard?
11 There's really nothing new. That's not a departmental issue.
12 We are the regulated body, not the--we don't have any true
13 influence in relation to the issuance of those regulations.
14 Nonetheless, that's working its way within the government,
15 and we're hopeful to have that out sometime in the near
16 future.

17 The e-mail issue, the USGS e-mail issue, hopefully
18 now is somewhat behind us. It was a significant cost to the
19 program in terms of the resources that it took to address it,
20 and to redo the work. Hopefully, we're beyond that now, and
21 hopefully, we will not have another occurrence.

22 Dr. Garrick mentioned the Board's interest in TADs,
23 and since I'm responsible for implementing TADs, I put a
24 slide in here of my own just to give you an update as to
25 where we are. As you know, we issued a performance

1 specification, a preliminary performance specification on
2 November 29th of last year, which allowed us to kick off a
3 process to have basically the four vendors identified and the
4 slide to develop proof of concept designs to see whether or
5 not that performance specification works.

6 We received those proof of concept reports, and we
7 have reviewed them, accepted them, and paid the contractors
8 for their efforts. And, we have also, as part of that
9 process, asked for input from the vendors on ways that this
10 specification could be improved, and we're evaluating that
11 right now. The next steps in terms of implementation of TADs
12 will be the issuance of a final specification, and the
13 initiation of the next procurement to vendors to allow them
14 to develop our designs under Part 71 and 72, and submit them
15 to the NRC, generally, about the same time that we're
16 submitting a license application for the repository.

17 I'd like to end up with basically a recitation of
18 the Director's expectations of the program team. And, this
19 is a presentation that he's given, or a slide that he's used
20 in internal discussions with the staff.

21 First of all, compliance with quality assurance
22 requirements is not optional. Safety, quality and schedule
23 discipline are not mutually exclusive, as I mentioned
24 earlier. The corrective action process is the process to be
25 used for fixing problems, and I think the program is slowly

1 turning to making that happen. We're directed to work across
2 organizations as a team to bring our A game, and I guess
3 that's in reference to our Tiger Woods. But, we're not
4 playing too much golf in the program right now. To be
5 accountable for your performance, and integrity is a
6 prerequisite of working on this program.

7 And, with that, I'd be happy to answer any
8 questions that the Board might have as to the status of the
9 program.

10 GARRICK: Questions from the Board? Andy?

11 KADAK: Kadak. You mentioned a number of areas, but you
12 didn't give us any information about where they stand. For
13 example, the EPA rule you said will be issued shortly. Can
14 you give us more definition about when that is?

15 KOUTS: We hope to have it issued shortly. It's not
16 within the control of the Department as to when that will be
17 issued. That's basically an intra-governmental process at
18 this point.

19 KADAK: Are you guys part of that discussion?

20 KOUTS: We are part of it, but we don't control it.

21 KADAK: But any sense of--are we talking a year from
22 now, two years from now? I mean, this is pretty important,
23 don't you think?

24 KOUTS: Yes, it is. I can't really give you a time
25 frame on that.

1 KADAK: Okay. On the question of TADs, you said you got
2 the preconceptual designs, or whatever you--concepts. Now
3 what?

4 KOUTS: Well, as I just mentioned, the next step will be
5 to issue a final specification. If you looked, originally,
6 we issued a preliminary specification.

7 KADAK: Right.

8 KOUTS: The next step then would be initiation of a
9 procurement for the vendors to develop designs and submit
10 them to the NRC under Part 71 and 72.

11 KADAK: When would that occur?

12 KOUTS: I won't say specifically, but it's going to be
13 in the very near future.

14 KADAK: And, your expectation for having TADs available
15 for utilities is by when?

16 KOUTS: Assuming that we go forward with the program as
17 it's currently planned, the earlier we could have them
18 available to utilities would be about in the 2011 time frame.

19 KADAK: Two quick short ones. Is the DOE now working at
20 identifying and developing interim storage facilities?

21 KOUTS: No, it is not.

22 KADAK: Okay. Last problem. Slide 9, what is--these
23 are obviously very good goals. Now, why is it so difficult
24 to do these things?

25 KOUTS: I think the program, to an extent, has done

1 these things in the past. I think what we're looking at is
2 to raise the organizational behavior to hopefully all the
3 same level in all areas, and I think that's the best way that
4 I think I can characterize it. I think that it's been a long
5 struggle within the program. I've been in it for over 20
6 years, and the understanding of the licensing process, the
7 understanding of the CAP program, as you can see, we had some
8 issues with USGS in terms of quality assurance, I think it
9 took those speed bumps, if you will, to get people's
10 attention and to make them understand that you need to follow
11 procedures. And, that's the fact of the matter. We're
12 trying to elevate the performance of the program to be what
13 we believe will be acceptable to the NRC from a licensee
14 standpoint, and from an applicant's standpoint.

15 GARRICK: Henry?

16 PETROSKI: Petroski, Board.

17 I'm interested in the corrective action process.
18 Could you elaborate on that a little bit, and, say, give an
19 example of a problem that is being fixed, and also explain
20 how problems are identified to be subject to this corrective
21 process?

22 KOUTS: Okay. Well, briefly, we have a corrective
23 action within the program where anyone in the program can
24 write what we call a CR, and at that point in time, those--
25 that CR is evaluated by a screen team to understand its

1 validity and so forth, and then it's graded, and then it's
2 assigned to an individual manager. And, that manager
3 essentially is responsible for evaluating it, if necessary,
4 go through a root cause discussion, and at that point in
5 time--at every step along the way, his actions are reviewed
6 by other people to make sure that that corrective action and
7 that condition report, if you will, the original CR, is
8 addressed thoroughly, and such that this condition report
9 would not occur if the aspects of the corrective action plan
10 are fully implemented.

11 What has happened in the past is that where a
12 condition report was submitted to the system, and a manager
13 would develop a corrective action, that corrective action
14 would be taken, but in many cases, the corrective action did
15 not get to the root cause of the problem. It was not a
16 comprehensive analysis enough to make sure that that
17 condition report could not occur again. And, what we're
18 doing now is we're essentially re-evaluating all our
19 condition reports and making sure that all our corrective
20 action plans truly are comprehensive, and will make sure that
21 it's not a continuing problem in the program. That's taken a
22 lot of time to go through, a lot of discussion with the
23 managers, a lot of talk in terms of does this corrective
24 action plan really address the root cause issue.

25 So, that's, to give you an example, quality

1 assurance is a concern within the program, it's certainly a
2 concern at the level of the Director, and he owns that
3 condition report. And, it will be his corrective action plan
4 in terms of implementing that to hopefully bring the quality
5 assurance program and process up to the level that we need to
6 be as an applicant and as a licensee.

7 PETROSKI: How long does this process take?

8 KOUTS: It depends on the individual subject. In some
9 cases, they could be very minor. In other cases, they're
10 more substantial, and it can take hopefully no more than
11 several months to resolve. But, in the case of we've had
12 some very long-standing CRs within the program, and that's
13 one of the issues that we're trying to resolve, some of which
14 are over a year old, and we're trying to do the best we can
15 in order to close those as quickly as we can.

16 PETROSKI: Thank you.

17 GARRICK: Mark?

18 ABKOWITZ: Abkowitz, Board.

19 Chris, I was wondering if you could comment on the
20 implications of the budget that DOE received with respect to
21 the planning and the area of transportation. It was one of
22 the four priorities, but my understanding is that it has
23 taken a disproportionate share of the cuts that have come
24 along. Could you elaborate on that, please?

25 KOUTS: That's essentially a management decision and a

1 value judgment within the program. The driving force right
2 now behind the program is to, first and foremost, to submit a
3 quality license application to the NRC, and basically, all
4 other areas of the program are of secondary importance in
5 terms of trying to accomplish that.

6 Transportation is, like other areas of the program
7 that is not getting the funding that was originally intended,
8 obviously, because we had a \$100 million reduction. In terms
9 of a disproportionate amount, you know, all I can say is that
10 my areas of the program also took reductions, and other areas
11 have to look very carefully at reductions also. So, yes,
12 transportation was reduced, and typically when we get into
13 that kind of situation, you can't have transportation unless
14 you have a place to send it to. So, you've got to make sure
15 that first and foremost that the repository and the license
16 application is receiving the necessary resources in order to
17 accomplish its objective.

18 GARRICK: Ron?

19 LATANISION: Latanision, Board.

20 Could you give us an update on the transition in
21 terms of lead labs, the impact in terms of staffing,
22 scheduling with reference to the license application, et
23 cetera? What impact, if any, has this had, and is the
24 transition more or less complete at this point?

25 KOUTS: My perspective, and Claudia Newbury is nodding

1 to me, yes, the transition is complete. With any transition
2 like that where you're taking a large component of work away
3 from one contractor and giving it to a lab, obviously, there
4 are issues. I think we have worked through that. Sandia
5 National Lab, who is our lead lab, is fully at work, working
6 on their aspects of the LA, which is basically postclosure
7 performance, and they are coordinating closely with Bechtel
8 in relation to the development of the license application,
9 because it all has to come together and make sense from a
10 preclosure and a postclosure standpoint.

11 So, a simple answer to your question is the
12 transition is complete. There were, I will call them speed
13 bumps along the way, but I think Sandia has the resources
14 that it needs this year, as Bechtel does, in order to
15 accomplish what they need to accomplish. And, all reports
16 are that it's working smoothly at this point.

17 GARRICK: Bill?

18 MURPHY: Bill Murphy, Board.

19 I'm curious about the status of the performance
20 assessment calculations to support the license application.
21 Could you mention where that stands? Are new data being
22 collected, or new models being developed, or calculations
23 being conducted, or sensitivity analyses being done? At what
24 stage is that?

25 KOUTS: The answer to all your questions is yes.

1 Essentially, I believe all the data feeds, or most of the
2 data feeds for the TSPA are essentially complete, and, in the
3 very near future, we'll start essentially running the model
4 and doing the sensitivity studies that we need in order for
5 our safety case in the LA. So, that's ongoing, and it's a
6 continuous process, and will be a continuous process until we
7 submit the LA.

8 GARRICK: Chris, the Board is under the impression that
9 there are some schedule slippages, and that this is one of
10 the agenda items this week in Nevada on the status of the
11 project. Can you give us kind of a heads up on what
12 slippages we're talking about, and what might be the possible
13 impact?

14 KOUTS: Well, the slippages that you are referring to,
15 from my perspective, don't impact the June 30th date in any
16 way. We'd like to try to submit the LA as early as we can,
17 and basically, the plan is to have the LA done early next
18 year. So, to the extent that you are hearing about
19 slippages, it's slippages associated with an early next year
20 submittal of the LA, not the June 30th date, no later than
21 June 30th that the Director has committed to Congress.

22 All I can say is this is typical project
23 management. Whenever you have a great deal of activities
24 being done by various groups, you've got to manage the
25 process. And, in one case, if someone is behind, you need to

1 find out what the reason behind that is, whether there are
2 resources involved, whether there are decisions that need to
3 be made. And, that's essentially what goes on at our monthly
4 project reviews, which actually is going on today in Las
5 Vegas. We go through in detail all the different components
6 of the, especially of the license application, and identify
7 what areas and what deliverables, where they are, and whether
8 or not they're behind, whether or not they're going to catch
9 up in the near term, and so forth.

10 But, the simple answer to your question is although
11 you may have heard that we're behind, that's not in any way
12 impacting a June 30th submission date for the license
13 application.

14 GARRICK: Thank you. Are there any questions from the
15 Staff?

16 (No response.)

17 GARRICK: All right. Thank you very much.

18 I guess our next speaker will be Ray Wymer. Ray is
19 a consultant, retired from the Oak Ridge National Laboratory,
20 and he's going to talk to us about depleted uranium oxide as
21 a chemical barrier. Welcome, Ray.

22 WYMER: Thank you, John.

23 Thanks for the opportunity to make this
24 presentation this morning. It's a little bit off the
25 principal target of this group, but I think it's a chemistry

1 of some interest to you, and it's sort of, you might view the
2 information I'm going to present as another arrow in your
3 quiver that you'll have available to you when you want to
4 fire a bolt at the Department of Energy.

5 So, give me the first slide.

6 GARRICK: We don't do that.

7 WYMER: I would never accuse you of that, John.

8 I wanted to point out that this talk is based on
9 Russian work. It was started under the International Science
10 and Technology Center, which is a program of the U.S.
11 Department of State. It was created almost specifically for
12 the purpose of occupying former nuclear weapons, Russian
13 scientists in non-weapons research, but nonetheless, with a
14 nuclear application. And, my role in this was as a technical
15 advisor to this group, made a couple of visits to Russia,
16 looked at their facilities, and gave them insights that they
17 did not have on what the Yucca Mountain repository was all
18 about, and some of the problems that might need attention.
19 So, I didn't do any honest work. I just simply followed what
20 the Russians were doing and advised them.

21 The next viewgraph, that is something that is
22 familiar to certainly all of the Board, maybe not to all of
23 the members of the audience, it's a picture of the cross-
24 section of one of the proposed drifts in the Yucca Mountain
25 repository. And, the purpose of this viewgraph is to point

1 out some of the places that depleted uranium might be used in
2 the Yucca Mountain repository, depleted uranium oxide.

3 One application is as a backfill around the waste
4 package and the drip shield. Another application might be as
5 material to fill in inside the waste package, packed in and
6 around the fuel elements themselves. And, it also is meant
7 to show that there are various places where there are various
8 kinds of metals in the repository, in particular, around the
9 waste package, or as part of the waste package, as part of
10 the drip shield, and perhaps as part of the invert at the
11 bottom of the drift.

12 Next one, please. Okay, now, the source of
13 depleted uranium is the UF₆ that's stored in 14 ton
14 containers, mostly at Paducah and Portsmouth. Now, it's all
15 that's shipped off the Oak Ridge site, the K25 site. There's
16 about 700,000 metric tons of UF₆ in those cylinders, and as I
17 speak, there are two conversion plants being built, one at
18 Portsmouth, and one at Paducah, Portsmouth, Ohio and Paducah,
19 Kentucky, to convert that UF₆ to mainly the U308. The
20 process is based on what is fundamentally a French process
21 for conversion of UF₆ to oxide.

22 The way the process runs is it produces about 80
23 percent U308 and about 20 percent UO₂. It could be fairly--
24 the way we've talked to the people at Paducah, it could
25 fairly simply be changed to produce completely UO₂, if that

1 were desirable.

2 I should point out that recently, there's been a
3 suggestion that that depleted uranium might be re-enriched.
4 You know, it runs about .35 percent uranium 235. There's
5 still some good U235 in there. As UF6, it could be picked up
6 by the U.S. Enrichment Corporation, or one of the two
7 enrichment groups that's currently building enrichment
8 plants, to extract additional uranium 235 without having to
9 go to all the cost of producing UF6. So, there may be some
10 economic advantage to this, and that's being looked at.

11 Next one, please. Now, these are what I consider
12 to be the primary factors in the release of radioactivity,
13 and you'll see how these fit in with the UO2 discussion
14 shortly. Most important, of course, at the top of the list
15 is water. If you don't get water into the waste package and
16 into the fuel, you don't have a problem. So, the ingress of
17 water, one way or another, is of primary importance.

18 The next comes the potential failure of the drip
19 shields, which is, as some of you may know, is the titanium
20 palladium alloy that surrounds the waste packages. Next is
21 the breach of the waste packages themselves, which are
22 thought not to be breached very readily since they have Alloy
23 22 surrounding them, which is very resistant to corrosion.

24 Next, if you get through that barrier, then it's
25 the breach of the fuel rods themselves, some of which will

1 have failed after a few thousand years and they'll have
2 cracks in them. Then, the fuel itself, the fuel pellets will
3 dissolve, or if it's the 10 percent of the content of the
4 repository that's devoted to vitrified waste from the
5 plutonium production sites, it will be glass that would
6 dissolve.

7 And, then, the next, the transport of radioactivity
8 in water, either as dissolved radioactivity, ionic or as
9 colloidal, either primary colloids or pseudo colloids. And,
10 pseudo colloids are colloids that are sort of the second-hand
11 ways of producing colloids. You have something like ions
12 that will form a colloid, and then one of the fissure
13 products will sorb on that, and that's a pseudo colloid.
14 And, that transports just like a colloid, just like it were a
15 primary colloid.

16 And, then, the engineered and natural barriers,
17 some of which are listed above, if they fail, and finally the
18 primary factor in release is the chemistry actually in the
19 drift in the unsaturated zone and in the saturated zone. So,
20 you have these, these are the primary factors.

21 The next viewgraph, please? As far as things being
22 chemically important in the repository in connection with
23 DUO2 applications, application of DUO2 in the repository, the
24 water, of course, is very important and its amount and
25 composition is important. I'll talk just a little bit about

1 the composition of the water in a minute. Of course, air is
2 important, and the most important things in air, in my
3 judgment, are the oxygen and the carbon dioxide present in
4 the air, and in the water.

5 And, in temperature, anything having to do with
6 chemical reactions is temperature dependent. And, radiation,
7 of course, the radiation produces, as most of you know,
8 various kinds of oxidizing species, hydrogen peroxide
9 radicals, and potentially, possibly, the production of
10 nitrogen oxides from the nitrogen and oxygen in the drift to
11 produce ultimately nitrite or nitrate. And, then, of course,
12 the materials of construction, which I mentioned a little bit
13 earlier, being a lot of metals in there which are
14 intrinsically reducing agents.

15 Next viewgraph? Okay, starting at the top, the
16 importance of air is that air will get into the drift in
17 several different ways. Barometric pumping as the barometric
18 pressure goes up and down, there will be either vacuums or
19 pressures induced on the other rock, and it will pump air
20 into the drift. Of course, diffusion of air down through the
21 cracks in the Yucca Mountain, and water itself, of course,
22 will have dissolved air in it, and most important will be, I
23 think, the carbon dioxide that's present in the air as far as
24 the chemistry of uranium oxide dioxide is concerned.

25 And, I want to point out that there will be,

1 despite the fact that the repository is inherently an
2 oxidizing environment, there will be local reducing
3 conditions due to the presence of metal. So, there are tons
4 of iron in the repository, and, of course, there's a lot of
5 metal around the waste package. So, at a long time scale,
6 you know, thousands, tens of thousands of years, even some of
7 these metals that are very resistant to corrosion can be
8 expected to corrode eventually, somewhat.

9 Next viewgraph, please. As far as the temperature
10 is concerned, as you all are well aware, as long as the drift
11 itself is above the boiling point of water, water will not
12 get into the drift. It will be held out. And, that will be
13 for a thousand years, give or take a few hundred years.

14 The solubility of the UO_2 , depleted uranium oxide,
15 is of course temperature dependent. And, as you will see
16 later, the sorption of radionuclides, and specifically, I'll
17 be talking about neptunium and technetium, is temperature
18 dependent, quite strongly temperature dependent, and is
19 obvious, I think, corrosion of the metals and of all of the
20 ingredients in the repository, they are temperature
21 dependent. And, the chemistry itself, both with respect to
22 the chemical reactions that take place and the kinetics, the
23 equilibrium are shifted as a function of temperature,
24 depending on things like complexation formation, dissociation
25 and dissociation of complexing species. These things are all

1 temperature dependent. So, temperature is a big actor in the
2 repository, as I'm sure you're aware.

3 Next? This is data that was obtained in Russia,
4 and these are the principal investigators. I don't have
5 information, I do not know information specifically on all
6 the analytical techniques that were employed, although they
7 are very competent and they have cutting edge technology,
8 mainly with equipment that we have bought for them in
9 connection with this program, and other programs. But, they
10 are very well set up to carry out the kind of work that you
11 will see discussed.

12 Next viewgraph, please? Well, you have to talk
13 about the water that's coming into the drift, and as you go
14 through the literature, you will find several different rack-
15 ups of the composition of J-13 water, which is sort of
16 standardized. It's been chosen in order to be able to fix on
17 some kind of a composition of water, and it pretty much has
18 this composition. You will find some tables that vary a
19 little bit from this. Also, there are wells that have been
20 dug in and around the proposed repository, which have
21 compositions that are not the same as this. But, in order to
22 have something to work with, J-13 well water had sort of been
23 settled upon. It's not to be believed wholesale, I don't
24 think, that this is the exact composition of the water that
25 will be dripping into the repository.

1 The most important thing, in my judgment, on this
2 listing is the carbonate ion concentration. It's fairly high
3 in millimoles per liter, and it has a very profound effect on
4 the chemistry of everything that goes on in connection with
5 the UO₂ and its potential use in the repository.

6 Next viewgraph? The Russian study, three samples,
7 the most important features of which was the temperature at
8 which it was prepared. Samples one, two and three are
9 progressively higher in the preparation temperature. This
10 has a distinct effect on the behavior of the UO₂, on behavior
11 of its chemistry, and in particular, on the behavior of the
12 surface chemistry of the UO₂. And, as you will see as I go
13 through this, it's not the bulk of composition of the UO₂
14 that really is the deciding factor, it's the concentration of
15 the surface layer, what's the composition of the surface
16 layer with respect to uranium 4 and uranium 6.

17 Next viewgraph? This merely shows you measurements
18 of the particle size distribution for anybody who is
19 interested in boring into that detail. They're very similar
20 for all three of these samples, annealed at all three
21 temperatures.

22 Next viewgraph? I found this to be a fascinating
23 piece of information. This represents the permeability of
24 thin layers of UO₂ with respect to distilled water and J-13.
25 I see the distilled water didn't come out. It's above, a

1 little bit above the J-13 solution. But, you see, after half
2 a year, or so, the water, J-13 water does not go through a
3 thin layer, a millimeter or so thick layer of UO₂, and this
4 is the low temperature annealed UO₂. So, I thought this was
5 a fairly important observation, the fact that it inhibits the
6 movement of water pretty appreciably, I would say, although
7 to what extent it happens over 10,000 years is another story.

8 HORNBERGER: Ray, can you tell us a little bit about
9 what that actually is? What kind of filtration rate? How is
10 this experiment being done?

11 WYMER: How does it--well, this is in centimeters per
12 day, and the experiment was performed on a cylinder of UO₂,
13 about a millimeter thick, and J-13 water was just simply put
14 on top with no more pressure than just that exerted by the
15 weight of the water.

16 Next viewgraph, please? Well, a very important
17 consideration is the solubility of the UO₂, and this is the
18 phase ratio of the amount of UO₂ with respect to the volume
19 that it's shaken up with. And, both distilled water, which
20 we call DW here, and J-13 water were used, and it was
21 filtered through three different porosity filters to make
22 sure we weren't looking at colloidal UO₂ instead of dissolved
23 UO₂.

24 And, this is an extremely chemically complex
25 system. I hope that will come out in the next viewgraph as

1 we go along.

2 Next viewgraph, please? This just simply shows how
3 fast the solubility measurements came to equilibrium, and you
4 can see that the low fired UO₂ in J-13 water had a solubility
5 up here at around almost 5 parts per million, whereas, the
6 higher fired material was less soluble under these
7 conditions.

8 Next viewgraph? This is the intermediate fired and
9 the highest fired UO₂ were separated out and look at in J-13
10 well water at Page 7, and it sort of repeats what was in the
11 other viewgraph, but in a little bit more detail. The
12 equilibrium was essentially reached in a couple of weeks as
13 far as these measurements are concerned.

14 Next viewgraph? Now, this shows that there was an
15 effect of particle size in the water. The fairly porous
16 shoulder showed a fairly high uranium content in this low
17 fired material, and as you went to the very fine porosity
18 filter, you got down to a factor of 20 lower, which means
19 that there was some very finely divided UO₂ from slow fired
20 stuff. It was also true of the other two samples, the higher
21 fired samples, but not to the same extent. They pretty much
22 reached where they were going to get to with the 50 nanometer
23 filter. So, you had to be very careful with these
24 measurements that you're not mistaking solubility for just
25 extremely fine particles or colloiddally dispersed particles.

1 Next viewgraph, please? I just put this in because
2 I think it will help you understand a little bit later some
3 of the sorption studies of neptunium that you will see. This
4 shows the chemical--these are thermodynamic calculations, but
5 these kinds of plots are common in the literature. You all
6 have seen a lot of them if you've gone into this field at
7 all. This is in J-13 well water, so there's a little bit of
8 fluoride in the well water. That shows up here at the lower
9 pHs, and the formation of UNO complexes. You manage to get
10 to pHs where the carbonate ion can exist, where it's not
11 dripping off the CO₂ because of the acidity, then you begin
12 to form carbonate species, and these will be the species that
13 are present on the surface of the UO₂, not in the bulk of the
14 UO₂. And, I think that they will play a big part in the
15 explanation of some of the chemistry absorption.

16 Next viewgraph, please? Okay, we get in now a
17 little bit to what's happening on the surface of the UO₂
18 samples. When it just says UO₂, that means the low fired
19 stuff, and you see the function of pH here, how much U⁴⁺ and
20 how much U⁶⁺ do you have on the surface, even though if you
21 were to do a bulk analysis, it would still look like UO₂.
22 But, the Russians did measurements that just looked
23 specifically at the surface layers. And, when you do that,
24 then you see that the valent uranium on the surface varies as
25 a function of pH a little bit.

1 Next viewgraph, please? This is another way of
2 presenting that data. It shows the distributions of forms of
3 valent states for the low fired and the high fired UO₂, and
4 you can see that the blue here is the U(IV), and it goes away
5 as the pH goes up.

6 Next viewgraph, please? This shows uranium
7 solubility, the theory versus the experiment. You can see
8 that the low fired material here are--the black line is the
9 theoretical solubility, the red line is theoretical
10 solubility of U(IV) and U(VI) respectively, and the dots are
11 the experimental data. So, they lie fairly close to the red
12 line, which is the theoretical solubility of Uranium(VI).
13 And, you can see that as the pH goes up, the solubility goes
14 down.

15 Next viewgraph? This is meant to show that if you
16 reduce, if you keep the uranium under reducing conditions,
17 then the solubility remains low, at low pHs. But, it will be
18 essentially the same once you reach a pH of about 4 ½ or 5.
19 But, this does show that the presence of the uranium in an
20 oxidized condition is a good deal higher, and is under
21 reducing conditions, these are very low solubilities to start
22 with in millimoles here. So, there is an effect. You can
23 keep the system reduced, and reduce the solubility.

24 Next viewgraph, please? Now, I want to get into
25 the sorption of neptunium. I wanted to point out that what

1 I'm going to be talking about is the sorption of neptunium on
2 UO₂, and its surface oxidized species. But, there's a very
3 good discussion of the speciation of neptunium in spent fuel,
4 not with regard to sorption on UO₂. It was done at MIT in
5 the Nuclear Engineering Department. This is the, if you
6 wanted to access this, this is the number you go to.

7 Next viewgraph, please? This is a viewgraph meant
8 to point out something that most of you knew. It was taken
9 from John Kessler, who sent this to me from EPRI, and I
10 wanted to get away from either the NRC information or the DOE
11 information, and present an independent view of the
12 situation. And, it's pretty close to both the NRC and the
13 DOE model.

14 What it shows is the importance of neptunium with
15 respect to contributing dose at the site boundary, which of
16 course is what this whole thing is all about. The dose at
17 the site boundary is the big deal. It doesn't much matter
18 what happens in between if the dose at the site boundary is
19 acceptable. But, under some conditions, it may not be. So,
20 this is the proposed, if you're standard, this line shows
21 what the dose would be without neptunium, and this shows what
22 the dose would be with neptunium, according to John Kessler's
23 calculations. And, these are the contributors to the site
24 boundary of technetium, iodine, plutonium, and this solid
25 line is the total dose, of course. So, I just put that in

1 there--oh, and this is John's model of the failure of the
2 fuel pin cladding. Once the water gets that far into the
3 repository, then he assumes this "S" shaped curve. This is
4 cumulative failure, and so adding something like 10 to the
5 5th years, it's all failed, in his assumption. This is an
6 assumption, of course. As you all know, that's one of the
7 big buggles of this whole business, is nobody knows for sure
8 what's going to happen in 10,000 years. You're taking this
9 based on the significant extrapolations.

10 Next viewgraph, please? Now, this shows how the
11 valence states of neptunium in normal environment air changes
12 as a function of pH. And, so, there is a distribution of
13 valence states, according to the Russian data, as a function
14 of pH.

15 Next viewgraph, please? Now, we're getting at the
16 sort of the mechanistic look at this thing. In acidic
17 solutions, it's assumed by the Russian workers, based on
18 their experiments, that Neptunium(IV) absorbs in acidic
19 solutions as Neptunium(IV) when there is a mixture of the
20 uranium valence states out in the solution phase. But as you
21 go to neutral solutions and higher, you get an oxidized
22 uranium film, which is both, you could call it U₄O₉, a well
23 recognized uranium species, also could be considered 3UO₂ and
24 1UO₃ if you want to not violate valence rules. And, so,
25 there's an oxidized layer which has the average composition

1 of uranium 2.25 oxidized, and on that surface, it's Neptunium
2 (V) and not Neptunium(IV) that absorbs, and it's basically
3 just Neptunium(VI) out in the neutral solutions.

4 Now, the thing that doesn't show here is the fact
5 that there are carbonate species of both the uranium on the
6 surface, and of the neptunium, which is an indication of the
7 extreme complexity chemically of the system.

8 Next viewgraph, please? This is a plot of the
9 sorption of Neptunium(V) onto the various--well, two of the
10 oxidation states, the lowest and the intermediate oxidation
11 state, in distilled water, J-13 water, inert atmosphere, and
12 in air. And, as you can see, the sorption approaches, of
13 neptunium, approaches 100 percent and in the case of J-13
14 well water and the case of nitrogen for the lower fired
15 material. The sorption is not nearly as high for the higher
16 fired material, and that's because, as pointed out earlier,
17 the surface is much more stable to producing the higher
18 oxidation state of the uranium on the surface. And, you can
19 see these are run over long periods of time.

20 But, the important point here is that you can get,
21 with J-13 well water, you can get to 100 percent sorption of
22 Neptunium(V) on the surface, which I think is significant.

23 Next viewgraph, please? This is on de-ionized
24 water in the presence of nitrogen, and again, you can see
25 that the sorption in this case of the UO₂ is lower fired,

1 which again shows up at 100 percent in the presence of
2 nitrogen. And, it's not as good as these higher fired
3 materials. So, if you were to use this as a sorbant, you
4 would want to use the low fired material.

5 Next viewgraph? This is a function of pH, and
6 neptunium sorption here for Neptunium(V) is not as high, it's
7 a higher pH. This is not totally consistent with the other
8 results, but nonetheless, it's what they reported. And, I'm
9 not sure to what they attribute the difference, but I thought
10 you needed to know that there is some variation in the
11 results with time, with experiment.

12 Next viewgraph? This shows the sorption again of
13 Neptunium(V), after equilibrium has reached, and you can see
14 that the sorption again is about 100 percent, except at this
15 higher pH where it drops down, so there's a strong pH
16 dependency, which you saw in the last viewgraph, stronger
17 than you might expect, based on the equilibrium for the
18 carbonate complexation. And, the higher fired material again
19 absorbs less well than the lower fired material.

20 The J-13 well water did not come out as well as the
21 previous slide showed. So, lower pHs are desirable if you're
22 going to get good sorption of the neptunium.

23 Next viewgraph? This shows the effect of
24 temperature, shown here 20 degrees, 70 degrees, and 95
25 degrees for the two higher fired materials. And, the

1 important point here is that as the temperature goes up, the
2 sorption goes up. And, the repository conditions for quite a
3 long time when the temperature is well above 20 degrees or 70
4 degrees, there will be pretty good sorption of the neptunium
5 on the UO₂. And, this shows that in about a day, it's
6 reached saturation, 100 percent sorption. So, there's an
7 advantage of having a hot repository for as long as--with
8 respect to neptunium sorption for as long as it can be kept
9 hot.

10 Next viewgraph? This was just, I think, meant to
11 point out the influence of the carbonate ion. When you get
12 to 100th molar HCL, you're going to certainly suppress the
13 presence of carbonate ion in solution, and as you can see,
14 over time, the sorption just drops off. So, if you get too
15 acidic, like 100th molar, a pH of 2, then you get practically
16 no sorption, which suggests that it's a carbonate mediated
17 sorption process.

18 Next viewgraph? Now, I want to spend just the last
19 couple minutes talking about technetium. Technetium, as you
20 know, pertechnetate ion is the stable species, and if you
21 were to dissolve spent fuel in an oxidizing environment, the
22 species that would go into solution would be the singly
23 charged pertechnetate ion, which is very mobile in the
24 environment, it's a lot like the perchloride ion, it forms
25 few complexes, it doesn't attach to things that are in the

1 soil well at all, and it tends to move, it would tend to move
2 with the water.

3 However, in a repository environment where there
4 are reducing agents locally, you could produce
5 Technetium(IV), so that's what the Russians concentrated on,
6 was if you in fact produce Technetium(IV) perhaps by making
7 the invert out of iron, and with all the other iron that's in
8 the repository besides, you know, there's iron in the walls
9 of the drift, there's various kinds of metal alloys in the
10 waste package, so, you could have a reducing condition in the
11 water that comes in. And, so, if you produce Technetium(IV),
12 I don't assure that you would, but if you did, then the
13 following information should hold true.

14 The next viewgraph, please? This shows that
15 Technetium(IV) sorption is a function of pH in a nitrogen
16 atmosphere, and as the pH goes up, the sorption goes up. So,
17 there's an advantage in having a reducing environment.
18 Uranium dioxide itself is a mild reducing agent, wanting to
19 go from Uranium(IV) to Uranium(VI). Whether or not this
20 would have any influence has not yet been sorted out, sorted
21 out by the Russians. But, at any rate I take this to be
22 somewhat encouraging, that the pHs of about those that are
23 going to be present in the repository, you're likely to get
24 very good sorption of the technetium, which if it didn't hold
25 it forever, would certainly hold it a long time. It would

1 slow it down. It would slow down its release and transports
2 through the environment.

3 Next viewgraph? This is meant to show that it
4 doesn't make any difference. If you had Technetium(IV) to
5 start with, whether or not you're going to have a reducing
6 agent, does not change anything. It's already Technetium(IV)
7 and it's not going anywhere. Iron 2 and hydrazine were both
8 put into the solution, and you can see as you get to the pH
9 of the repository, you're not going to get much free
10 Technetium(IV) in the repository.

11 And, finally, this is sort of a summary that the
12 sorption increases as pH increases, and at pH 7 or above, the
13 sorption of Technetium(IV) is essentially 100 percent.

14 Next? Because the conversion process at Paducah
15 and Portsmouth produces about 80 percent UO₃ instead of 100
16 percent UO₂, we asked the Russians to take a look at sorption
17 on U₃O₈. That has not yet been completed.

18 That's it of my presentation. I want to stress
19 that I'm not advocating that we pour a lot of UO₂ in the
20 repository. I merely wanted to point out to you that there
21 is UO₂ present in the spent fuel, of course, quite
22 independent of what you might additionally put into the
23 repository, and this is just meant to give you a little
24 insight into some of the chemistry that might take place in
25 the repository under various conditions that are likely to

1 exist in the repository as a function of oxidizing conditions
2 as a function of composition of the water, as a function of
3 pH and a function of temperature, that all these factors that
4 are present and variables in the repository influence the
5 sorption of these two key species with respect to dose at the
6 site boundary, namely technetium and neptunium. And, that's
7 my message for today.

8 Final viewgraph should say The End.

9 GARRICK: Okay, thanks, Ray. We have a number of
10 questions. Ron, do you want to start it?

11 LATANISION: Yes, Latanision, Board.

12 Ray, I don't see a number on this slide, so I'm
13 looking at the--

14 WYMER: It's in the right-hand corner--I'm sorry if it
15 isn't there, but--okay, well, just tell me--

16 LATANISION: Well, it's the slide that shows that thin
17 layers of UO₂ become impermeable. On Page 3, but I don't
18 see--

19 WYMER: What about it?

20 LATANISION: Not the right one. Let's try--

21 WYMER: What's your question?

22 LATANISION: It's the slide that shows impermeability to
23 water.

24 WYMER: Yes, right, this one.

25 LATANISION: Yes. I don't know which number it is.

1 There you go, just passed it. There it is.

2 WYMER: Yeah.

3 LATANISION: How is the UO₂ introduced onto the surface,
4 and what is its state? Is it glassy, is it--what--

5 WYMER: No, it's a powder. It's present as a powder, of
6 the particle size distribution that the preceding slide
7 showed. It was that powder that was put on as a thin layer
8 onto a porous disk, and then it was through that layer of
9 that powder that the water was allowed to flow. And, after a
10 couple, three months, it stopped coming through.

11 LATANISION: Was the powder treated in some way?

12 WYMER: The powder was fired at 650 degrees, and it was
13 just simply put on. It was the same material that you'd use
14 in all these other experiments I showed the viewgraphs of,
15 the very same material.

16 LATANISION: It's certainly very intriguing to see that
17 kind of data for oxide materials. That's not common.

18 WYMER: As you probably noticed from the distribution,
19 it was finely divided material, and the speculation is that
20 it formed a hydrous layer, or a hydrous oxide layer right on
21 the surface of those tiny particles, which essentially
22 plugged the pores.

23 LATANISION: Yes, you're going in the direction that I
24 was going to head, and that was whether there was an
25 evaluation of the change in surface chemistry of these

1 powders as a function of time as well?

2 WYMER: They did not get that far. They merely made
3 this observation.

4 LATANISION: Well, it's very intriguing.

5 WYMER: I thought it was also very intriguing.

6 LATANISION: Thank you.

7 WYMER: I think it has implications with respect not
8 only to the fact to maybe putting UO₂ into the waste package
9 surrounding the spent fuel, but also with respect to the UO₂
10 in the fuel pin, which while it's not the same as this low
11 fired material, might in fact see some of this effect.

12 LATANISION: Latanision, Board.

13 I mean, I guess my concern would be, however, that
14 if the powder is swelling, and, therefore, limiting the
15 permeability of water, that could present another problem
16 that would not be so attractive.

17 WYMER: And, the Russians did not do any experiments on
18 volume change of the material. I would not expect that it
19 would swell a lot. I think it would be just very much
20 surface, and not--there would have to be some swelling,
21 obviously, or it wouldn't close the pores.

22 LATANISION: Yes. Thank you.

23 GARRICK: Howard?

24 ARNOLD: Arnold, Board.

25 I'm confused, I guess. Every time I've asked

1 questions about the behavior of the spent fuel, the answers
2 come back, well, once the cladding is breached, then the
3 pellets dissolve because it's an acid environment. And, I
4 guess I'm not hearing that here. I'm hearing an entirely
5 different story. But, I'm not knowledgeable enough to
6 understand the difference. Are you implying that the UO2
7 pellets themselves will survive?

8 WYMER: I think there are some of the things that you've
9 seen with these powders that will also happen on the surface
10 of the UO2, depending on the degree of fragmentation, I would
11 suppose. Of course, the UO2 pellets are all cracked. But, I
12 don't think you would see an effect of the UO2 pellets
13 anywhere approaching the kind of things that you see, or to
14 the degree that you see on these powders. I don't think that
15 the--I don't know what the repository environment would be
16 with respect to acidity. You see all kinds of discussions of
17 whether or not radiolysis would produce nitric or nitrates.

18 ARNOLD: Maybe I misspoke. Maybe I should have said
19 oxidizing environments. But, yeah, the question is the same
20 one, are you implying a different behavior of the UO2 pellets
21 than what I've heard in the past?

22 WYMER: I'm saying it's conceivable. But, they did not
23 look at UO2 pellets, which are much higher fired, and they
24 are much denser than this material. You might get some, on a
25 crack, as it starts to dissolve in the water that hits the

1 crack, you might get some exfoliation where you form a powder
2 material that would have some of the characteristics in this
3 work, yes. But, we don't know.

4 GARRICK: Thure?

5 CERLING: Cerling, Board.

6 On Slide 13, you showed this nice distribution of
7 particle sizes, and so on, and then later, you showed a slide
8 that showed that there was some colloidal material being
9 absorbed when you are doing the filtering.

10 WYMER: Right.

11 CERLING: And, did they have a sense of whether that
12 colloidal particle was re-suspension of this original
13 material, or was it new and different, I guess, different
14 than the original material?

15 WYMER: I think the assumption was that it was just this
16 material suspending. There were no chemical treatments of it
17 to alter its chemistry. They just simply looked at the
18 particle size, and filtered the particles, and some of the
19 particles are down at .19 microns. That's pretty small
20 stuff. So, the filter they used was small enough to catch
21 most of it, and as you saw from the viewgraph, all three of
22 the samples at the very smallest pore filter had about the
23 same value, or solubility.

24 GARRICK: David?

25 DUQUETTE: Duquette, Board.

1 I'm not as mathematically challenged as my material
2 science colleagues. Can I look at Slide Number 2? This
3 looks like a very different concept than we've been talking
4 about, with the exception of the titanium drip shield and the
5 steel lined tunnel. Is this your suggestion? Is this a
6 Russian suggestion? This is, I mean, even the package itself
7 that's a Cermet waste package, would be a totally different
8 package that we've been talking about so far.

9 WYMER: This is based on work that was done by the
10 Russians, and supported by the U.S. Department of State, and
11 they are, the Russians are looking at a Cermet as the waste
12 package. This viewgraph was made for a different selling job
13 than today. Of course, it's Alloy 22 for you people. So,
14 this stuff down the left-hand side, especially that which
15 shows the waste package, is different.

16 The filler material, the component of backfill that
17 it shows there is really meant to be a Richard's Barrier,
18 which as some of you know, is a coarse material, topped with
19 a very fine material, which by capillary action, redirects
20 the water around the waste package and down the side. And,
21 our thought was that if you made that out of UO₂, you'd get a
22 double advantage. You'd not only get the Richard's Barrier
23 advantage of diverting the water around the waste package,
24 but any water that came in would become saturated with UO₂,
25 thereby, greatly reducing almost to zero the driving force

1 for dissolving the UO₂ in the pellet. You'd already have a
2 saturated UO₂ solution of depleted uranium by the time that
3 water got to the pellets, so, there would be very little
4 driving force except for the difference in the chemical
5 activity of the UO₂ pellet, based on the fact that it's now
6 about 5 percent fission products, and it's fired for
7 temperature. So, this would change the chemical activity a
8 little. But, the solubility would be greatly reduced because
9 you already have a saturated solution of UO₂. So, it has a
10 double, sort of a double whammy effect here, you decrease the
11 solubility of the pellet, you brought it up against that, and
12 if you had a Richard's Barrier, you'd also divert it.

13 The advantage of the UO₂ saturation of the water
14 before it gets to the defective pin would also be present in
15 the case of packing UO₂ around the spent fuel inside the
16 waste package, which would be a much easier thing to
17 accomplish than putting in a Richard's Barrier in a drift.
18 That's not a simple thing.

19 DUQUETTE: Duquette, Board.

20 Why on earth would you put the titanium drip
21 shield, given the engineering problems and cost of doing
22 that, if you look at two pieces of data? One is that you're
23 going to basically use the UO₂ surrounding the package to
24 absorb water, and secondly, I think you just showed data that
25 indicated once it's saturated, it doesn't pass water anymore.

1 WYMER: That's right. And, I think that the--it's not
2 out of the question that the drip shield could be left out of
3 this repository. You know, these are personal opinions, but
4 it's a very expensive thing, and probably more expensive to
5 make those drip shields with this titanium palladium alloy
6 than it would be to pack depleted uranium oxide around the
7 spent fuel inside the package. Now, there's a question
8 there, though, in that you have provided an insulator. If
9 you pack the UO₂ around the spent fuel, you effectively
10 packed an insulator around the spent fuel, and it will get
11 hot. So, you might want to store the fuel on a pad for a
12 while before you put it into the repository and pack UO₂
13 around it.

14 DUQUETTE: Thank you.

15 GARRICK: Bill?

16 MURPHY: Bill Murphy, Board.

17 I'm fascinated by the chemistry of UO₂ added to the
18 repository environment, and I think that--

19 WYMER: I knew you would be, Bill.

20 MURPHY: I think a lot of the effects that have been
21 seen in these experiments are probably a consequence of the
22 oxidation, rather aggressive oxidation of the UO₂, and the
23 formation of secondary uranyl mineral phases. And, you
24 showed one diagram, this one, which is on composite Page 6,
25 showing concentration approaching U(VI) solubility. I don't

1 know what phase that might be. Maybe it's Shopite or some
2 uranyl hydrate.

3 But, the experimental data on UO₂ dissolution and
4 natural urananite dissolution and spent fuel dissolution
5 shows that under the oxidizing conditions of Yucca Mountain,
6 it will oxidize fast. And, the depleted uranium that one
7 would put in a repository, when exposed to the ambient
8 conditions, would tend to oxidize fast. The surfaces would
9 become coated with secondary uranyl minerals, which could
10 very well explain why it becomes impermeable in a period of
11 time. There's a large volume increase upon hydration and
12 oxidation of the UO₂.

13 So, I'm a little dubious of the relevance of the
14 sorption data on reduced UO₂ surface, because by the time any
15 of the matrix radionuclides could even encounter a buffer of
16 UO₂, the fuel itself would have to be oxidized. The pathways
17 of ingress of oxidants and water and CO₂ would have been
18 established along the transport pathways within the EBS. So,
19 I think while there may be a potential benefit in terms of
20 diffusion barriers, or sorption characteristics of placing
21 UO₂ in the repository, it also is a, practical matter, just a
22 place to get rid of the depleted uranium, which is of some
23 value.

24 I think that its function is going to depend on how
25 it behaves after it's been oxidized, to a large extent.

1 WYMER: I know that this is a field of your expertise,
2 Bill, these secondary phases. I think that a consideration
3 that might come into play is that you get a surface layer,
4 which then greatly mediates the subsequent reactions, rather
5 than the entire material being oxidized so quickly.

6 But, yes, everything you say is right. We just
7 simply do not know the answers to these things.

8 GARRICK: Thure?

9 CERLING: Cerling, Board.

10 One of the things that struck me in this
11 presentation was, you know, the extreme fine nature of this
12 stuff. It's all, you know, sort of one micron and finer.
13 Are they proposing to do further experiments on materials
14 that are larger in size to be more practical to--

15 WYMER: Yeah, the size is up to as high as 7 or 8
16 microns. But, you're right, they're very fine. There's a
17 sad footnote to this, in that while the Department of State
18 indicates an interest in continuing support of this, the work
19 is in fact directed out of Oak Ridge National Laboratory, and
20 people like myself follow the work, and advise them on
21 experiments to perform that might be of interest. While the
22 Department of State has continued to indicate possibility of
23 support for this work, the Department of Energy has decided
24 to withdraw the funding from ORNL, so there is no U.S.
25 oversight of the work anymore. As a consequence, I see,

1 personally see very little reason to continue to fund the
2 Russians to do the work if nobody on this side of the ocean
3 is telling them what to do.

4 GARRICK: Garrick, Board.

5 Ray, has there been any engineering studies done on
6 just what it would take to implement such a process? I'm not
7 thinking so much of redesigning the waste package with
8 Cermet, or whatever, but just using it as an appropriate
9 backfill. Has anybody looked at this from the standpoint of
10 practicality and how it would be engineered?

11 WYMER: No. The only thing that's been looked at, and
12 that only in a very cursory fashion, is what I mentioned,
13 namely the fact that it would be an insulator and the
14 temperature would go up. You'd have to decide whether or not
15 the spent fuel pins would tolerate the temperature without
16 failing faster than they would ordinarily. But, there have
17 been no cost analyses or engineering studies.

18 GARRICK: Yes. That was going to be my second question,
19 was the trade-off between the accelerated deterioration of
20 the fuel as a result of the increased heat with the
21 advantages of the depleted uranium.

22 WYMER: No, this has not been done. The money has not
23 been provided to carry out anywhere near that comprehensive a
24 study. This is just meant to provide some interesting
25 chemical insights that may have some relevance to the

1 repository.

2 GARRICK: Has there been any attempt to quantify the
3 effect of just the UO₂ in the fuel with respect to this type
4 of chemistry?

5 WYMER: No, they have not done any work with spent fuel
6 at all in this program.

7 GARRICK: Andy, and then George?

8 KADAK: Kadak.

9 I guess I'm intrigued more by the holdup of
10 neptunium and technetium as a useful application here. I
11 wouldn't put any of this stuff in the canisters because that
12 would complicate things even more, as you suggest. But,
13 suppose--I remember at WIPP they have these bags of things on
14 top of the plutonium canisters or barrels to absorb water, or
15 whatever it is that control pH. Have you looked at something
16 like that here? And, if Bill's concern is correct, namely
17 the stuff oxidizing creates this layer of impermeability may
18 not be helpful, but clearly, putting this on the ground, or
19 in some place in the bottom of these drifts might be helpful.

20 WYMER: Yes, I think that's exactly right. I did not go
21 into any of the other potential applications because of the
22 emphasis of this Board being specifically on Yucca Mountain.
23 But, yes, there are other potential applications that might
24 be more practical actually than using this in the repository
25 itself.

1 KADAK: I would suggest using it in the repository if
2 and when the canisters are breached and begin to dissolve,
3 and it falls onto this absorber, if you will, if it can be
4 maintained.

5 WYMER: Well, that's essentially what we were talking
6 about.

7 KADAK: Well, no, you put in the cans, and then you also
8 put it as a backfill. I'm talking about something more
9 simple, putting it on the invert, or making the--

10 WYMER: Sure. Sure. There are a number of ways
11 potentially to apply this. We don't know how effective they
12 would be. This is just interesting chemistry that has some
13 potential application. But, yeah, there's a lot of things
14 you could think of, and that's a good one. On the invert
15 would be a good place to put it, as a matter of fact.

16 GARRICK: George?

17 HORNBERGER: This stuff that the Russians worked with
18 has a huge specific surface area.

19 WYMER: It really does.

20 HORNBERGER: Surface chemistry effect. And, also, it's
21 pretty clear that you would never backfill with something in
22 the micron size range.

23 WYMER: No, you wouldn't.

24 HORNBERGER: So, then, given all that, the question is
25 is there some extra advantage over backfilling with a larger

1 aggregate, depleted uranium, versus some other getter?

2 WYMER: I think that the principal advantage might be
3 towards that you would saturate the incoming solution with
4 UO₂. That's the big thing that I can see. I think that's
5 pretty big.

6 HORNBERGER: What do you anticipate the basic effect
7 would be? I mean, you would certainly reduce then the rate
8 at which the uranium is dissolved. But, do you have a common
9 ion effect that would slow down release of other nuclides?

10 WYMER: I don't know. You get into the complexities of
11 the chemistry because a lot of this dissolved uranium will
12 have carbonate on it, and maybe some fluorides with the
13 chlorides, as I showed there, because these are minor
14 constituents in the repository, in the rock and in the J-13
15 water. So, I think the complexity gets out of hand, and
16 that's probably one of the major chemistry problems with the
17 TSPA, is the chemistry is just out of hand. It's far too
18 complex to really model realistically. You're going to have
19 to step back a step and take a grosser look and look for
20 overall moderating influences rather than detailed chemical
21 effects, I would think.

22 GARRICK: Any other questions from the Board? From the
23 Staff? David?

24 DIODATO: Diodato, Staff.

25 Dr. Wymer, I appreciate your presentation. I was

1 intrigued by your assertion of temperature influence on
2 sorbtion. And, you have the slide you showed, I think it's
3 about Slide 32.

4 WYMER: Yes.

5 DIODATO: So, would this experiment, can you tell me a
6 little bit like was the possibility of out gassing of CO₂ in
7 this experiment?

8 WYMER: There was no purging with an inert gas, or
9 anything of that, to carry the CO₂ out. And, at a pH of
10 around 7, or so, you would not expect out gassing of the CO₂.
11 It would be there as a carbonate ion.

12 DIODATO: So, a constant pH was maintained during this
13 experiment?

14 WYMER: Yeah, as far as I know.

15 DIODATO: I mean, you could think of if CO₂ was out
16 gassing, then what would happen to the solution chemistry.

17 WYMER: It wasn't boiled, and it wasn't purged. So,
18 there was no deliberate out gassing attempted.

19 DIODATO: All right. So, this is a constant pH, as far
20 as you know?

21 WYMER: Yes. Well, so far as pH stays constant as you
22 run the temperature up.

23 DIODATO: Well, that was my question, really.

24 WYMER: Yes, I don't know.

25 DIODATO: Okay, thanks.

1 WYMER: Would be my answer.

2 DIODATO: I appreciate that. Thanks.

3 GARRICK: Any other questions from the Staff? Board?

4 (No response.)

5 GARRICK: Thank you. Thank you very much, Ray.

6 Our agenda says we're to a point of a break. I'd
7 like to break until about 10 o'clock.

8 (Whereupon, a brief recess was taken.)

9 GARRICK: Before we turn the microphone over to John
10 Bredehoeft, I want to indicate that there's going to be some
11 minor changes in the program for this afternoon. All the
12 subjects that are on your agenda will be covered, but we're
13 going to change the order of a couple of them. And, in
14 particular, the 1:50 p.m. Saturated Zone Testing paper will
15 be given second, after the Near-Field Chemistry. So, Brady
16 and Reimus are going to just switch positions.

17 The other thing is that we're going to crowd a
18 little public comment time space in just after the Saturated
19 Zone Testing, and before the break.

20 With that, we will turn the microphone over to John
21 Bredehoeft, and John is from his own group called The
22 Hydrodynamics Group, and will give us a progress report on
23 the Inyo County Drilling Program.

24 BREDEHOEFT: Thank you, John.

25 I want to acknowledge my colleagues, Mike King, my

1 partner, and Chris Fridrich, geologist with the U.S.
2 Geological Survey, who has done a lot of the geologic mapping
3 for us in the area.

4 Next slide, please? Now, we are the oversight
5 contractor, technical contractor, for Inyo County. Inyo
6 County, as many of you know, is in California. And, when we
7 think about the repository issues in Inyo County, we're
8 concerned mainly with the potential for the movement by
9 groundwater from the repository basically to Death Valley, to
10 the springs in Death Valley. That's our principal concern,
11 and our program has been directed at looking at that concern.
12 And, I will try to lay out the justification for the program,
13 thinking behind it, in the next half our, or so, and show you
14 some of the things that we've done.

15 Next slide, please? Now, we've just completed a
16 well in this area right here. Actually, we've got two wells.
17 We have completed one well to the carbonate aquifer, and then
18 we drilled a second well into the Amargosa Valley fill into a
19 reasonably permeable limestone at about 600 feet. The
20 carbonate aquifer well is completed at 2800 feet into a
21 permeable zone in the carbonate aquifer.

22 Now, what's kind of interesting, I'll point it out
23 to you, is the repository is here. We have one well at the
24 repository that went into the carbonate aquifer, UE-25P1. It
25 penetrated the carbonate aquifer. And, there was an oil well

1 that was drilled somewhere about here, drilled and plugged.
2 And, that's the extent of the drilling to the carbonate
3 aquifer in the Amargosa area. So, this happens to be now the
4 second observation well that we have to the carbonate aquifer
5 after UE-25P1.

6 So, it's interesting that we're concerned about an
7 aquifer here for which we have very little really subsurface
8 information. We have other information. But, you know, in
9 terms of drilling, there isn't a lot.

10 Next slide, please? This is a log of the hole.
11 This is the Amargosa Valley fill. We got into the Paleozoics
12 here at 2400 feet, right here, and then drilled to 2800 feet,
13 where we got quite a lot of water, lost the hole, actually,
14 the first time around, and ran out of money. Then, we went
15 back early this spring. We first lined the hole to this
16 depth. We had to go in again and line the hole all the way
17 to the top of the Paleozoics, then completed this as an
18 observation hole. We still have not pumped this hole enough
19 to get it all cleaned up yet, get good water samples, and
20 bring it on line as an observation well. We'll do that in
21 June. But, we're not at that point yet.

22 KADAK: Could you help us with that? Where is the water
23 table in that slide?

24 BREDEHOEFT: The water table is within 100 feet of the
25 land surface. It's right here. It's in an elevation of

1 about 2180.

2 KADAK: And, it goes down how far?

3 BREDEHOEFT: Excuse me?

4 KADAK: And, how far does it go down?

5 BREDEHOEFT: The well, we completed--

6 KADAK: No, I mean, is there water all the way from that
7 depth that you mentioned?

8 BREDEHOEFT: Yes, to 2800 feet.

9 KADAK: It's full of water?

10 BREDEHOEFT: It's saturated.

11 KADAK: Okay.

12 BREDEHOEFT: All the way down. So, the water table is
13 here.

14 In fact, not very far from this location as you go
15 east through the Amargosa River Valley, the Valley is pretty
16 wet, and there's standing water in the Valley a lot of the
17 time. So, the water table is very near the land surface as
18 you go maybe two miles, or so, a mile and a half to the
19 northeast of the particular location.

20 All right, next slide? Now, I want to review for
21 you some of the thinking about how groundwater occurs in
22 Nevada. When you go back to before 1950, the idea was that
23 if you took any particular valley, there would be valley fill
24 in this valley, and if you looked at the valley, it would be
25 more or less a self-contained aquifer system, so that there

1 was recharge from the mountain range. The recharge went into
2 the valley, and then there would be discharge in the playa
3 area. That discharge would also be in the form of free
4 edified vegetation that went to the water table. So, these
5 were thought to be pretty much self-contained systems. That
6 was the paradigm when you thought about Nevada hydrology in
7 the 1950's, or so.

8 Now, there were a group of people, Ike Winograd--
9 Ike is here in the crowd--working at the Nevada Test Site.
10 And, Ike, Al Klept and his partner, Bill Thorasen, and they
11 came to the realization that at least in the vicinity of the
12 Test Site, that the valleys were underlain by this permeable
13 aquifer, the carbonate aquifer, and the hypothesis was that
14 this carbonate aquifer provided a kind of French drain
15 underneath the valleys, and integrated the groundwater system
16 from one valley to the next. And, that was the hypothesis
17 that they developed in the 1958, the late Fifties basically,
18 working, as I say, at the Nevada Test Site.

19 And, it turns out that hypothesis was very quickly
20 picked up by most of the hydrologists working in Nevada at
21 the time. So, there really wasn't--and, it's interesting to
22 look back at this because they put this hypothesis out, and
23 people jumped on it right away, said yes, that's probably
24 what's going on. And, there was also a realization that some
25 of these valleys, you could not account for the recharge and

1 the discharge. There were discrepancies between the amount
2 of water that's coming in, and the amount of water that was
3 coming out of the valley. In other words, the valleys, not
4 all of these valleys appeared to be self-contained.

5 So, this carbonate aquifer then was immediately
6 looked at as this underlying large aquifer system which
7 tended to integrate many of the valleys in eastern and
8 southern Nevada. Now, this carbonate terrain is huge.

9 Next slide, please. Oh, back up just a second.
10 One of the evidences that Winograd used, since there were not
11 any drill holes to sort of say--you know, you couldn't trace
12 this aquifer by looking at the drill holes, but the things
13 they looked at were the water chemistry. And, they could see
14 from the water chemistry that the springs and the water from
15 wells that were developed in the Nevada Test Site had similar
16 chemistry, and you could see that these are a group of
17 springs here in the area of the Amargosa Desert and the
18 Nevada Test Site, and you can see they all group rather
19 closely together here, and, this cluster of, geochemical
20 cluster, and that was one of their stronger evidences that
21 this kind of under drain was occurring, and that things were
22 moving, the groundwater was moving through the carbonate.

23 All right, now, the next slide, this shows the
24 extent of this carbonate aquifer, and as I started to say to
25 you before, it's huge. Here's California. I've got to be

1 careful here. Here's the Utah line. And, you can see we're
2 looking at most of eastern Nevada, and a goodly portion of
3 western Utah, all underlain by this carbonate aquifer, and
4 integrating many of the valleys in that part of the world
5 into this sort of large aquifer system that underlies the
6 whole area.

7 Now, this aquifer system, as you can imagine, it's
8 so huge that it's broken up into kind of sub-compartments so
9 that you, you know, it's not all one big aquifer. There are
10 compartments then in the aquifer itself. So, there's one
11 over in here that supplies the springs in the muddy river
12 area, and then we sort of look at this area down in here
13 where we're interested, and this is kind of one big area here
14 where we're looking at the carbonate system as one entity.

15 Okay, next slide, please. Now, I want to come back
16 to this one hole that we have at the Nevada Test Site. This
17 is a kind of schematic cross-section of that hole. This is
18 the hole, UE25-P1. It's kind of the carbonate, Paleozoic
19 carbonates here at about 1300 meters before the land surface.
20 And, then, there was quite a large section of carbonate rock
21 that was drilled down to about here. So, there was roughly
22 700 meters of carbonate that was drilled in this particular
23 well.

24 And, it's interesting that it encountered this Fran
25 Ridge fault zone right at the top of the Paleozoics, and it

1 at least appears to have encountered it right at the top of
2 the Paleozoics. And, as in the usual situation, the
3 Paleozoics here in this well were highly permeable. The
4 other thing that was interesting at this well particularly is
5 when you went into the Paleozoics, the water levels jumped by
6 approximately 20 meters. So, we've got a 20 meter increase
7 in hydraulic head between the overlying welded tuff material
8 here, and the Paleozoic aquifer.

9 Now, there was some discussion about how--let me
10 back up just a second. This water was relatively hot. It
11 was 55 degrees, approximately 55 degrees centigrade. So,
12 there was some discussion about was some of this head that we
13 were seeing, difference in head attributable to the fact that
14 we were filling the hole with lighter water, hotter water,
15 therefore, less dense, and, in fact, you could probably in an
16 extreme case, maybe account for 5 meters of head difference
17 from the fact that you had this hotter water in the
18 Paleozoics. But, still, there was still 15 meters, or at
19 least 15 meters of hydraulic, higher hydraulic head in the
20 carbonate here than there was in the overlying tuff aquifer.

21 Now, that's important because what you're saying to
22 yourself is, you know, the potential here is for groundwater
23 to move out of the Paleozoic aquifer into the overlying tuff
24 aquifer, not for groundwater to move downward from the tuff
25 aquifer into the Paleozoic. So, in fact, this higher

1 hydraulic head is protecting the Paleozoic aquifer from
2 downward movement of contaminants.

3 Now, there were a lot of interesting things about
4 this Paleozoic aquifer. It had a very good earth tide, and
5 the earth tide was analogous to the M-2, which is the lunar
6 earth tide, the biggest one when you do the composition of
7 the tidal components, the M-2, which is the big lunar 70
8 diurnal tide, and this particular well had an amplitude of
9 2.05 centimeters, which is easily observable. You know,
10 you're at 4 centimeters up and down attributable to the moon.
11 And, from that, you can calculate transmissivities. You can
12 also calculate storage coefficients.

13 The storage coefficient calculated from that was
14 extremely low. It was 10^{-10} , which we get for
15 Paleozoic--confined aquifers like this, we normally see
16 storage coefficients of something in the order of 10^{-6} to the
17 minus 6, but this turned out to be 10^{-10} .
18 That's extremely low. That suggested that the porosity of
19 this material was very low, 10^{-5} , which is a
20 very low porosity.

21 It was also highly permeable. They did a whole
22 bunch of packer testing in this section of the hole, and it
23 turned out there was 10 meters in here that had a
24 transmissivity, which is the permeability, average
25 permeability times the thickness, but there was a 10 meter

1 section that had a permeability, or transmissivity, excuse
2 me, of about 69 meters squared per day. I'll come back and
3 try to put that in context. I know it doesn't mean much,
4 just a number at this point. But, anyway, the hole was
5 interesting.

6 All right, next slide, please? Now, as some of you
7 probably know, there was two models being put together, one
8 for the Nevada Test Site and one for Yucca Mountain, and they
9 were modeling essentially this area of southern Nevada, and
10 parts of California. A decision was made that that was
11 redundant, and, therefore, the two models were combined into
12 a single model, and it was decided that the USGS would
13 complete this model. So, the USGS has basically built a
14 model for this area here, and they've quite a large effort
15 into that, maybe ten man years, maybe more than that, maybe
16 as much as twenty man years into this model.

17 So, that model has integrated what we know of the
18 regional geology, and pretty much the regional hydrology.
19 So, it seems to me that this is a database which one ought to
20 take advantage of in sort of looking at this particular
21 region. And, we've done that.

22 Next slide? Now, one of the problems with trying
23 to model this area is the geology is pretty complicated, and
24 you can see here's a cross-section block diagram through an
25 area right in this area in the eastern part of the area they

1 modeled. And, you can see it's all broken up. So, there are
2 all these fault zones in here, and it displaces the geology
3 rather badly. So, if you try to follow any one unit in here,
4 you see that the faulting breaks it up pretty badly.

5 Now, the question is how to model that kind of
6 system. And, what they decided was, in the next slide, that
7 what they would do is basically slice this system into 16
8 layers. So, there are 16 slices here, and the slices are
9 pretty arbitrary, so they basically took the land surface,
10 made 16 slices, and then decided in each slice, basically, in
11 each cell for the model, what geology would be there.

12 So, this is not the typical kind of model where you
13 would say follow the carbonate. Rather, it's an arbitrarily
14 defined slice, and then you put the geology in there that the
15 slice intersects. So, that allows you to create the model,
16 but it makes some complications when you're trying to
17 interpret the model, because you don't know whether when
18 you're in this layer down here, let's say Layer 15, whether
19 you're in the basement or whether you're in the Paleozoics.
20 You've got to look all the time to see where you are.

21 So, this model, even though it integrates all this
22 stuff rather nicely, it is difficult to use when you start to
23 interpret the results. And, as you can see, it's 16 layers,
24 it's about 250 kilometers across, 300 kilometers this
25 direction. The cell size in the model is 1500 meters by 1500

1 meters, so it's 160 by 194 by 16. There is roughly a half a
2 million nodes, cells in the model. And, the model has
3 convergence problems. I'm running this model, but it's
4 sticky when you're trying to make runs with it.

5 HORNBERGER: John, is it fully 3-D, or quasi 3-D?

6 BREDEHOEFT: It's fully 3-D.

7 All right, now, one of the things we did with the
8 model is to say to ourselves, okay, let's look at how much
9 carbonate rock is in the model, and this is the model area
10 here, this area on top, and this is the thickness of the
11 carbonate that's in the model, taken largely from the
12 geology. What they did is they got the geologists together,
13 created a whole bunch of cross-sections through the area, and
14 then connected those cross-sections and created these sort of
15 isopacks of what the carbonate would look like.

16 This is 5000 meters of carbonate. So, you can see
17 that there are lots of areas in here where you've got more
18 than 5000 meters of carbonate. You've got a tremendous
19 thickness of this carbonate. But, it's not everywhere. So,
20 there are also areas in here where there's no carbonate. So,
21 anyway, we've got this very thick carbonate, which underlies
22 much of the area.

23 Next slide, please? And, one of the things you can
24 also do with the model is it can output the water table, and
25 this happens to be the water table. This is Death Valley.

1 Death Valley is right in here, and then the Funeral Mountains
2 are through here, and this is the Amargosa Desert back in
3 here, and there's this sort of flat area in here in the
4 Amargosa Desert, with Yucca Mountain being right in here.
5 And, Ash Meadows, which is another outcrop of carbonate rock,
6 is basically there.

7 So, next slide? All right, now, one of the things
8 that Chris Fridrich has done for us is to map the carbonate
9 rock in the Funeral Mountains, and this is the PCA, this
10 color here, pinkish color is the carbonate rock in the
11 southern part of the Funeral Mountains. So, you can see we
12 have this area here is all--the mountain range is basically
13 composed of carbonate rock. It has this rather interesting
14 older material stuck in the middle here. So, there's
15 carbonate rock here, carbonate rock all around, and then this
16 carbonate rock area here.

17 The Big Springs and Death Valley occur right in
18 this area. So, they're on sort of the southwest side of the
19 Funeral Mountains, and the hypothesis that the water is
20 moving through the carbonate rock and the Funeral Mountains
21 to supply the Death Valley Springs.

22 So, next slide, please? So, one of the things that
23 Chris did in his mapping of the system was to project the
24 base of the carbonate rock in the Funeral Mountain area. So,
25 this is the bottom now of the carbonate rock, and he did that

1 by looking at this cross-section. This is the carbonate rock
2 here, PCA, here, here and here, and he basically looked at
3 these faults where they outcrop, and then projected them into
4 the subsurface. And, by projecting them to the subsurface,
5 you can then create this bottom of the carbonate rock, and
6 then we can hypothesize that that's the bottom of the
7 aquifer.

8 So, we're looking now at a map of the bottom of the
9 aquifer. So, it seemed to me one of the interesting things
10 to do was to say to yourself okay, can we, given that this is
11 the bottom of the aquifer, and we've got carbonate rock in
12 here most everywhere, can we create a model which will flow
13 water through this system, and supply the Death Valley
14 springs, and we did that.

15 Next slide? And, yes, it's feasible to do that.
16 What's kind of interesting about it is that the water has to
17 go through this narrows right in here. This is this sort of
18 older material that sticks up through the carbonate rock here
19 that I showed you on the original mapping. So, the water has
20 to be channeled through here, and then down to along the
21 Furnace Creek fault zone here. And, the most difficult
22 string to create is this one, Nevares here, where there's
23 this long finger of carbonate that sticks out to Navares.
24 It's actually in the carbonate.

25 The other springs here are actually in a syncline

1 of alluvial material that's associated with the Furnace Creek
2 fault zone. But, we've got a model. The model looks like
3 it's pretty reasonable, the kind of transmissivities that
4 we've got, that we solve for are out of the model, look like
5 they're kind of medium values for what's the transmissivity.

6 And, I'll show you in a minute what the transmissivity
7 values look like.

8 We recreated the spring flow. These are the
9 elevation of the springs. This one is very anomalous. But,
10 you can see these are the bigger springs, Travertine, Texas,
11 and then Nevares here, and then we've got a couple of smaller
12 ones. These are what we estimate the flow to be. We don't
13 have really good records on what the springs are, but we can
14 recreate them with the model reasonably well, so this is what
15 the model does. So, it's a pretty good fit.

16 Next slide? All right, now, I should tell you that
17 Las Vegas is having water supply problems. It probably
18 doesn't come as a surprise to you. You know, it's growing,
19 is a city of a million, and it's growing like mad. And, its
20 water supply comes from the Colorado River at the moment.
21 They're entitled to 300,000 acre feet a year from the
22 Colorado River. They've been taking that for some period of
23 time, but they're looking for additional water. And, so,
24 what they decided was that they are going to look to this
25 carbonate aquifer and see if they can get another 150,000

1 acre feet of water.

2 They've gone as far north as Ely, Nevada, and they
3 are going to pump groundwater in the vicinity of Ely, pipe
4 that groundwater to Las Vegas. Ely is almost 200 miles north
5 of Las Vegas. So, we're talking about a groundwater system
6 in the vicinity of Ely, and a very large transportation
7 system from that area back to Las Vegas. The costs are in
8 the billions, but they've got 60,000 acre feet out of their
9 first application in Spring Valley. So, they probably will
10 get this water.

11 At any rate, as part of that, they put together a
12 dataset for all of the transmissivity determinations that we
13 have from the carbonate aquifer through this entire very
14 large area in Nevada and western Utah, and this is the
15 frequency distribution of the carbonate transmissivities
16 here.

17 The transmissivity is the ability of the well,
18 basically, to supply water. And, that probably doesn't mean
19 a whole lot to you. We've got 218 samples in this frequency
20 distribution. But, let me try to put this in some context.

21 If you say to yourself okay, a good well is a well
22 that I can pump 100 gallons a minute with 100 feet of draw-
23 down, so I'll suggest to you that that's a pretty good well.
24 That happens here, right here, so 85 percent of the time I
25 can get a well better than that from the carbonate aquifer.

1 So, in other words, only 15 percent of the time is it less
2 productive than that. And, when you look down in here at the
3 4 or 5 percent level, these are the wells that are not very
4 permeable at all. So, then, you have to say to yourself,
5 okay, we've got this aquifer out here, is it permeable
6 everywhere. And, it looks to me like--I have asked this
7 question to all my colleagues, hydrologist in Nevada who deal
8 with the system, and it looks to me like if you're willing to
9 drill 1000 feet or so of carbonate aquifer, you're likely to
10 be getting yourself a pretty good well, at least a well
11 that's better than this 15 percent. You will be somewhere up
12 in here.

13 Now, looking at the model that we created for the
14 Funeral Mountains, we got a mean value right about here for
15 the transmissivity. It was 100 meters squared per day. So,
16 that's what--it was a mean value. The value that we
17 determined at UE-25P1 was 69 meters squared per day. You
18 know, that's very close to the mean. It's somewhere right in
19 here. So, again, we're kind of a mean value for what we got
20 from the one well that we've tested in this flow system.

21 So, my argument is that this thing is probably
22 permeable, reasonably permeable everywhere, and, you know,
23 there's quite a lot of it.

24 So, the next slide? All right, so, taking a
25 smaller area here, here's Yucca Mountain, the Amargosa

1 Valley, and Death Valley, and looking again at the carbonate,
2 this is the carbonate that underlies this portion--this
3 portion that I've highlighted here--this is the carbonate
4 aquifer that underlies it, the carbonate thickness of rock
5 that underlies it. And, again, you see this is 5,000 meters.
6 So, we've got carbonate in here--well, let me just--this is
7 the Furnace Creek fault, which is the southwest side of the
8 Funeral Mountains, and you can see that most of the Amargosa
9 Valley looks like, at least the geologic interpretation is
10 that we've got more than 5,000 meters of carbonate rock.
11 There's a lot of carbonate rock.

12 So, then, the next slide? This is a gravity survey
13 that was recently completed across the Amargosa Valley, and
14 you can see this is the Amargosa Valley fill, and then
15 beneath the fill is this carbonate rock again. And, again,
16 you see that we've got carbonate rock, it looks like it's
17 everywhere in there, at least from the gravity data. And,
18 again, you see that the Amargosa Valley here is pretty badly
19 broken up, but the carbonate rock is there. And, again, here
20 is the Funeral Mountains with the carbonate rock outcropping.
21 This goes from the Funeral Mountains to Devil's Hole Ash
22 Meadows, again, across Amargosa Valley.

23 So, next slide? So, I said to myself, okay, let's
24 do something really simple. Let's say to ourselves the
25 aquifer is there, carbonate is there, and we'll take the

1 carbonate as described in the USGS model, so this is the area
2 underlain by carbonate rock in the USGS model. And, we say
3 to ourselves, okay, we've got this carbonate rock, let's give
4 it a mean transmissivity so it's everywhere has the mean
5 value that comes from our frequency distribution, 100 meters
6 squared per day, and create a model for moving water through
7 the carbonate rock. And, we'll say to ourselves we've got
8 transmissivity, a good one aquifer, one layer aquifer, so,
9 carbonate rock with a mean transmissivity, and create a
10 model. And, that's what this represents.

11 So, here's Yucca Mountain right up in here. Here's
12 Nevares Spring. This is the Furnace Creek fault zone across
13 the southwest edge of the Funeral Mountains right in here.
14 And, you can see groundwater is moving in this direction from
15 up here, down towards the discharge of these springs in
16 Nevares area, and also from the Spring Mountains here through
17 Ash Meadows towards the Nevares area.

18 All right. So, we've got this one layer model, and
19 then we say to ourselves, okay, suppose we start with some
20 contaminant at Yucca Mountain, put a contaminant in here, and
21 ask ourselves how long will it take for that contaminant to
22 go from Yucca Mountain, we're assuming we've got the
23 contaminant now, and we just put it in the aquifer, how long
24 will it take for that contaminant to come from the vicinity
25 of Yucca Mountain to Nevares Springs. And, these are years.

1 So, the calculation suggests we get it there in
2 less than 50 years. Now, of course, that calculation depends
3 on what porosity you put in the model. The velocity of
4 groundwater movement depends upon the permeability over the
5 porosity, times the gradient of the hydraulic head. And,
6 what you assign here for the porosity is critical to how fast
7 this velocity is, and, of course, how fast the contaminants
8 are going to move. So, this calculation is based on a
9 porosity of 10^{-3} , and you can argue that's too
10 high or too low. As I suggested to you a few minutes ago,
11 the porosity that we determined at UE-25P1 may be as low as
12 10^{-5} . So, we're a couple of orders of magnitude
13 higher than what we saw at UE-25P1.

14 So, what it's saying is, you know, if the stuff
15 gets to the carbonate aquifer, it's pretty well gone. You're
16 going to see it in the biosphere reasonably quickly.

17 Okay, next slide, please? Okay, so, now, we come
18 back to UE-25P1, and you say to yourself, all right, there is
19 this upward hydraulic gradient, you know, we've got 15 meters
20 at least of upward hydraulic gradient between the carbonate
21 aquifer and the overlying welded tuff aquifer. So, you know,
22 that's certainly a barrier to any kind of contaminant
23 movement from the repository down to the carbonate aquifer.
24 But, then, you have to ask yourself the question, you know,
25 how permanent is that upward gradient. Because, you know,

1 one of the more ephemeral things about groundwater hydrology
2 is hydraulic head. When you start producing, and you start
3 developing groundwater, hydraulic head is the thing that you
4 lower, and you lower this regionally, and you do it really
5 quickly when you're developing. So, now, we've come back and
6 said to ourselves okay, so what could happen to this
7 hydraulic head.

8 So, anyway, the next slide? So, one of the
9 thoughts was to say to ourselves okay, let's take the USGS
10 model, as I said, they've put a lot of work into this model,
11 let's take the USGS model, project it out into the future,
12 and see what happens to the hydraulic head.

13 So, I have done that, and this is their 1998
14 hydraulic head map, and that includes a fair amount of
15 development. So, this was calibrated, and it was calibrated
16 for development in the Amargosa area, and in the Pahrump
17 area. And, they got what they think certainly is a
18 reasonable calibration. So, I said okay, let's run it out
19 for 1000 years into the future, and see what happens to the
20 hydraulic head.

21 So, the next slide? The next slide shows the
22 drawdowns that we would anticipate, using their model now,
23 1998 to 2998, almost 3000. And, this again is the--these are
24 the wells. Now, what I did is I just took their pumping
25 that's in the model at the moment, and you can ask yourself

1 well, how good is that pumping? It turns out that the State
2 of Nevada, for every valley in Nevada, has a potential yield,
3 and they say to you, the state engineer says you cannot
4 exceed the potential yield--or, I shouldn't say that. He
5 tries to restrict the development so that it does not exceed
6 the potential yield. That's a better statement. Whether he
7 does or not, is open to some question.

8 So, anyway, for the Amargosa Valley area, this area
9 in here, the potential yield quoted at the moment is 24,000
10 acre feet a year. The state measures, they do an inventory
11 each year of how much pumping is out there. They are
12 estimating at the moment somewhere around 12,000 to 13,000
13 acre feet a year. But, I looked at the USGS model, and it
14 looks to me like there's somewhere in the neighborhood of 20
15 to 23,000 acre feet a year in the USGS model. So, the USGS
16 model, as far as I can see at the moment, has about--it's
17 within 10 percent of what the state's yield is.

18 So, this, assuming, you know, you've got to make
19 some kind of assumption about how much development you're
20 going to allow, but this is pretty close to what the state
21 would currently allow, within 5 percent, actually. So,
22 anyway, this is a thousand year calculation. Yucca Mountain,
23 you can see, is somewhere up in here. And, we're calculating
24 here for Layer 8, which is right about the top of the
25 carbonate, about 10 meters of decline. So, if we had 15

1 meters there to start with, we're calculating that there will
2 be something in the order of 10 meters of drawdown, and we've
3 lost two-thirds of what that upward hydraulic gradient is.

4 Now, you know, that's kind of iffy, too, because
5 maybe you could say to yourself, well, the welded tuffs are
6 coming down as well as the carbonate, so maybe they're coming
7 down together, and you still have this relative difference
8 between the carbonate rocks and the welded tuffs. But,
9 you're making a projection here way into the future, and it's
10 terribly uncertain, and you're looking at a facet of the
11 hydrology that's probably the most ephemeral. And, you're
12 looking at an area that's probably going to develop. I mean,
13 it's almost--it's difficult to see that the Amargosa Valley
14 will not develop, particularly when you look at Las Vegas and
15 the spill-over into Pahrump, and now you're seeing a spill-
16 over into the Amargosa Valley. And, nothing seems to be
17 there stopping it. In fact, the federal government looks
18 like it's going to sell 55,000 acres in the Amargosa Valley
19 to private enterprise. So, it seems to me it's iffy whether
20 this upward gradient remains or not.

21 I think that's my last slide. So, we can now draw
22 some conclusions. You know, our thinking is that the whole
23 area is underlain by this carbonate aquifer. It's highly
24 transmissive. It has a low porosity and high groundwater
25 velocity. The aquifer extends to Death Valley. You know, if

1 stuff gets in there, it's going to get to the biosphere. I
2 don't think it's--and, it's going to go quickly. I don't
3 think there's much debate about--well, it seems to me that's
4 what the data suggests.

5 Then, you're counting on this upward head gradient
6 now as a barrier, and that upward head gradient is vulnerable
7 to future development. So, I think that's where we stand.

8 Thank you very much.

9 GARRICK: Questions from the Board? George?

10 HORNBERGER: John, let's see, it must be 15 years ago,
11 or more, that you and I were on an NRC committee, groundwater
12 at Yucca Mountain, and how high can it rise, and I recall
13 that one of the recommendations we made in there was that DOE
14 might consider better characterizing the carbonate in the
15 vicinity of Yucca Mountain. Has anyone done any drilling
16 except Inyo County?

17 BREDEHOEFT: Well, Inyo County did all that early
18 warning drilling. Some of those wells got water in the
19 bottom that looked like it was carbonate, carbonate
20 geochemistry. But, they did not get carbonate rock. So--
21 excuse me, Nye County, the Nye County drilling. But, Nye
22 County didn't get to the carbonate rock. So, the only
23 program that's done any drilling to the carbonate rock is
24 Inyo County. That's it. There has been nothing at the Test
25 Site. We've got one well at the Test Site. That's it.

1 HORNBERGER: So, how confident do you think we are that
2 there is an upward gradient pervasive throughout that huge
3 area of the carbonates that you showed us?

4 BREDEHOEFT: George, I don't think we're confident at
5 all. I think we're confident that it probably exists within
6 the vicinity of the repository. But, let's go back. One of
7 the things that's interesting, George, is that you see--
8 notice how steep these gradients are? And, that's that
9 really steep gradient that's suggested at the Test Site, at
10 Yucca Mountain, and, so, you know, I think you can say to
11 yourself with some confidence that probably that upward
12 gradient exists in the vicinity of the repository. But, I
13 would be reluctant to suggest that it extends, you know, even
14 out into the valley very far. So, you're thinking in terms
15 of, you know, maybe an area up in here where you've got high
16 gradient.

17 HORNBERGER: Of course, the real question then is that
18 you showed a pathline directly from the repository, but the
19 projections would be for the contaminants to move south.

20 BREDEHOEFT: Well, we still saw some upward gradient in
21 our drilling down here. You know, we're seeing a few feet of
22 upward gradient between the carbonate rock and the overlying
23 valley filled material, Amargosa Valley fill material. But,
24 you know, that's way down here. What happens in here? We
25 don't know. The only thing you could go with is you could go

1 back to the model and say to yourself, okay, how much
2 difference is there in the model. But, you know, we're
3 extrapolating based on very little information.

4 GARRICK: Yes, Bill?

5 MURPHY: Bill Murphy, Board.

6 This is fascinating to me. I really enjoyed your
7 talk. You alluded early to Ike Winograd's use of geochemical
8 data to infer inter-basinal flow. And, I happen to have
9 visited the springs on the western side of the Funeral
10 Mountains a couple of years ago, and I was astounded by
11 extreme differences in the water chemistry in springs that
12 are very close to one another. Some precipitating sulfur and
13 others, it looked like they're perfectly oxidizing, good,
14 high quality water. And, I wonder if you've made use of the
15 spring water chemistry to constrain your carbonate flow
16 model?

17 BREDEHOEFT: Yes, let's back up a second and let me
18 comment a little bit on that. We looked at the water
19 chemistry obviously. Now, this spring, Nevares, comes, we
20 think, is pretty well associated with the carbonate. It
21 comes out through alluvium, but there's carbonate rock very
22 close to where the alluvium is. So, that spring looks like
23 it's very carbonate dominated.

24 These springs, notice these--okay, so the carbonate
25 terminates along here, and then we have a syncline in here,

1 and these are the two bigger springs, Texas and Travertine,
2 and those are coming out of alluvial material. Again, they
3 have pretty much carbonate signatures.

4 Now, there is a very peculiar spring down here,
5 Navel, which is kind of sitting out here by itself, and that
6 chemistry looks rather different, very different in fact.
7 And, then, you have these two really small springs up in
8 here, which--but, again, I'm pretty sure those two springs, I
9 haven't looked at the geochemistry, but I'm pretty sure those
10 two springs look much like--this whole set here looks pretty
11 much the same. This one doesn't. It's very different. It's
12 very different. The origin of that one, it seems to me, is--
13 well, I've got questions about the origin of that one.

14 GARRICK: John, I'd like to ask you a whole bunch of
15 questions, but I'm afraid it would reveal my lack of
16 knowledge.

17 But, one thing that I'm curious about, aside from
18 the work that you've done, are you aware of any basin and
19 range community master plan with respect to the drawdown of
20 these aquifers, and how that matches with the time at which
21 there would be any contamination anywhere in the aquifers?

22 BREDEHOEFT: Well, obviously, the control of the
23 development of the water resources in Nevada is up to the
24 state engineer of Nevada. So, the state engineer of Nevada
25 has a plan for the development of water resources in Nevada,

1 and they are proceeding according to that plan.

2 That plan includes development in most of these
3 valleys. And, that has--there is no relationship to the
4 repository, zero.

5 GARRICK: Yes. But, what I'm getting at is when you try
6 to ask yourself does this really matter, you have to deal
7 with the question of timing. And, if you're talking about
8 contaminating something tens of thousands, hundreds of
9 thousands, millions of years into the future, what are we
10 talking about here?

11 BREDEHOEFT: Well, let me say this. It seems to me
12 perfectly clear that there is a lot of water in this
13 carbonate aquifer. This carbonate aquifer is now being
14 attacked by the City of Las Vegas for water supply, and
15 they're reaching out a couple hundred miles to the north.
16 Pahrump and Nye County are looking at the carbonate aquifer
17 as a water supply for Pahrump Valley and the local area in
18 the Amargosa Valley. So, I think you have every expectation
19 that the carbonate aquifer will be developed for water
20 supply.

21 GARRICK: Yes. And, I guess the question is the timing
22 of that, because--

23 BREDEHOEFT: John, that's going to happen quickly.

24 GARRICK: Yes. And, that's good. That's good.

25 BREDEHOEFT: Why is that good? Because that seems to me

1 will destroy this upward gradient, almost inevitably. That
2 will destroy the upward gradient.

3 GARRICK: Well, it depends upon how much of a drawdown
4 we're talking about. All I'm trying to better understand is
5 if you have an aquifer and it is likely that the benefit of
6 that aquifer is for the next few hundred years rather than
7 for the next tens of thousands of years--

8 BREDEHOEFT: No, that's not the argument. The argument
9 is--it goes like this. These valleys have a certain amount
10 of recharge.

11 GARRICK: Yes.

12 BREDEHOEFT: That recharge is discharged, and they say
13 much of that discharge goes for non-beneficial use.
14 Freadofites, which are basically brush, for the most part, in
15 this part of the world, sagebrush, creosote bush, various
16 different freadofites, and what you're going to do is you're
17 going to draw the water table down, get rid of those
18 freadofites, take some water out of storage, but ultimately,
19 the sort of plan that the State has is that every one of
20 these valleys will come into some kind of future equilibrium,
21 and that equilibrium can be maintained indefinitely.

22 GARRICK: Yes. But, it's kind of important to sort of
23 understand, I would guess, what that equilibrium condition
24 is.

25 BREDEHOEFT: Yes, that's very true.

1 GARRICK: Okay.

2 BREDEHOEFT: And, John, the only way you can understand
3 that equilibrium is--let me put it to you this way. The only
4 way to predict that equilibrium into the future is to say to
5 yourself, well, let's take the--I'm not saying this well.
6 The only tool we have to predict that equilibrium into the
7 future are these models.

8 GARRICK: You said at the outset that the subsurface,
9 hydrology was not nearly as well known as you like, or
10 something to that effect.

11 BREDEHOEFT: We have very little--we have two holes to
12 the carbonate, and one oil well hole that was plugged.

13 GARRICK: I'm just kind of surprised with the basin and
14 range being such a water oriented area in terms of trying to
15 get it, steal it, do whatever they can, that there isn't more
16 systematic and integrated, if you wish, between the regions
17 investigation, the long-term resources here.

18 BREDEHOEFT: The State wouldn't agree with that
19 statement.

20 GARRICK: Okay.

21 BREDEHOEFT: The State feels like they've done a good
22 job in identifying the potential yield of each one of these
23 basins, and the State engineer is attempting the best he can
24 to maintain the development within that potential yield.
25 And, that potential yield is designed to maintain the system

1 indefinitely.

2 GARRICK: Well, but one of the most interesting
3 observations that I heard today from you was that these
4 aquifers are not isolated.

5 BREDEHOEFT: That's correct.

6 GARRICK: So, it's much more than the State that has to
7 really be involved here to understand these, it seems to me.

8 BREDEHOEFT: Like Utah.

9 GARRICK: Yes, like Utah, like California. All I'm
10 getting at is that I'm trying to more clearly understand if
11 there's a problem here from the standpoint of Yucca Mountain.
12 And that really still seems to me to depend on the time
13 dependence of events, the demographics, and the one thing
14 that you pointed out that I never fully appreciated, namely
15 that the lack of--I thought this was pretty much a--so that's
16 the thing that I was trying to get at.

17 Yes, Bill, you had a question?

18 MURPHY: Bill Murphy.

19 Yes, I'm also interested in the relevance to Yucca
20 Mountain, and I am not sure, but George noted that the
21 contaminant transport models show that the contaminants move
22 to the south in the tuffaceous aquifer and in the alluvium.
23 And are there any models for contaminant transport from Yucca
24 Mountain that show that it gets anywhere near the carbonate
25 Aquifer? There's a kilometer of tuff you have to go through

1 first.

2 BREDEHOEFT: I haven't looked at those models.

3 HORNBERGER: But, they have an upward gradient, so they
4 don't even consider the carbonate. I mean, your whole point
5 is that if you switch that and you now have a downward
6 gradient, you now have a driving force to move the
7 contaminants downward. You still do have to get through the
8 tuff. There's no doubt about that.

9 GARRICK: Andy?

10 KADAK: Kadak.

11 I guess I'm trying to understand the basis for
12 these models. You have two data points, as best I can tell
13 from your earlier discussion, and you were able to create a
14 model based on those two data points?

15 BREDEHOEFT: Well, let's back up a second. That's not
16 quite right. Let's go to the second to the last slide. No,
17 further. There. The next one. Okay, so they have put
18 together a 16 layer model. Okay? So, you've got a 16 layer
19 model.

20 KADAK: Now, who is they?

21 BREDEHOEFT: USGS.

22 KADAK: All right. But, they had data from where?

23 BREDEHOEFT: Okay, so they have data on--this is their
24 array of pumping wells, and these pumping wells are pumping
25 from the valley fill aquifer. Okay? So, you're pumping here

1 from the valley fill aquifer, and I haven't shown everything.
2 This is Pahrump down here, and they're pumping from the,
3 again, from valley fill there. And, there are more wells
4 spread around. This is a gold mine up in the Beatty area.
5 So, you've got wells spread through this area. So, you have
6 data up in this section, so you've got data from the shallow
7 system, from the valley fill system, and then in some cases,
8 from the welded tuff section.

9 For example, you have that Nye County data, Nye
10 County drilled all those wells along 395 here. So, that data
11 is there. So, it's not as if you don't have data. You do
12 have data, shallower in the system. But, when it comes to
13 the deeper portion of the system, then we don't have much
14 data. So, in the deeper portion of the system, you are
15 dependent upon your ideas of what the geology looks like, and
16 we're extrapolating the geology from the outcrop areas into
17 the subsurface based on our based ideas of the geology and
18 the geophysics. And, then, you're fitting this model based
19 on data that's mostly collected in this shallower portion of
20 the system. So, it's not as if you don't have any data.
21 They have probably something of the order, I've forgotten the
22 number, but there's a couple hundred wells, for which we
23 have--not a couple hundred--there's probably a hundred wells,
24 for which we've got pretty good histories, Twentieth Century
25 histories for what happened to those wells during that

1 period.

2 KADAK: But, these are, as you said, shallow, and the
3 thing that you were trying to model was the deep ones; right?

4 BREDEHOEFT: That's right. For the deep system, we
5 don't have much.

6 KADAK: So, I'm just trying to test the veracity of your
7 comment relative to in 50 years, you'll get stuff from Yucca
8 in these wells. Where does that come from?

9 BREDEHOEFT: Well, you can question the veracity of
10 that. What I'm saying to you is the geology suggests that
11 there is 5 kilometers of carbonate rock down there. I'm
12 saying to you that 200 wells that we've got that penetrated
13 the carbonate rock suggest that it's highly permeable. We've
14 got one well that suggests it's very--that the porosity is
15 very low. So, if you combine, you've got big thicknesses of
16 carbonate, you've got reasonable permeability based on a
17 couple hundred wells, and then you don't know what the
18 porosity is. So, it's uncertain.

19 GARRICK: Okay, John, thank you very much.

20 We're now coming to our next speaker, and that's
21 going to be Sam Armijo from the University of Nevada, Reno.
22 He's going to talk about work that he has done on a second
23 generation waste package design.

24 Sam, Welcome.

25 ARMIJO: Thank you, John.

1 First, I'd like to thank the Committee for inviting
2 me and our team to present this paper. I'd like to
3 acknowledge our co-authors, my co-authors, Professor Mizra,
4 Dr. Karr, and also Professor George Danko of UNR, whose
5 multiplex code was used in the thermal analysis in this work.

6 I'd like to make the point that the views we are
7 expressing are the views of our team, and not necessarily the
8 views of the University of Nevada, Reno.

9 First slide, please? This Committee is well aware
10 of the standard of reference of what we would term a
11 Generation I design for Yucca Mountain, consisting of the
12 waste package--316 nuclear grade inner waste package, outer
13 waste package of Alloy 22, and a titanium drip shield. And,
14 we use the terminology 2nd Generation because we want to
15 distinguish between this design, which is a reference design,
16 which will get licensed, with a design that might be
17 considered after the initial licensing.

18 And, there's an analogy between the licensing of
19 Yucca Mountain initial design, and future waste or operated,
20 and future packages. The analogy is with nuclear power
21 plants in the United States. As you all know, the initial
22 plants that we built in the United States had a lot of
23 conservatism in them. And, in recent years, the industry and
24 NRC has taken advantage of that excess conservatism and
25 amended the licenses of the nuclear plants, increasing the

1 lifetime of many plants from 40 years to 60 years, and there
2 will be more license extensions as well.

3 In addition, the nuclear plants had sufficient
4 conservatism that they could be upgraded in power. So, we
5 have some plants in the United States that have been upgraded
6 by as much as 20 percent in rated power.

7 With respect to the analog for the waste package,
8 we have, for years, we have been introducing new fuel designs
9 into our light water reactors base, and primarily, the
10 purpose of those introductions are better performance and
11 better economics. So, we believe that that sort of thinking
12 would be valid for Yucca Mountain if you had something worth
13 changing to.

14 Next slide, please? So, the 2nd Generation concept
15 presupposes the initial, the current licensing application
16 will be approved, and the system will begin to operate. And,
17 at some point in time, could be amended to do the following
18 things. First, permit operation of the facility at higher
19 temperatures. Permit use of larger, lower cost waste
20 packages. And, I'll get into details of what that waste
21 package would look like. And, eliminate the titanium drip
22 shields.

23 The benefits of such a move would be to delay the
24 risk of aqueous corrosion by extending the post-closure
25 period by thousands of years, and maintaining a dry

1 environment. The second benefit would be to reduce the
2 number of waste packages. If we use the nuclear fuel to
3 create the additional heat in the drifts, then, of course,
4 we'd require fewer waste packages.

5 We propose eliminating a lot of the conservatism in
6 the current waste package design. I believe, and our team
7 believes, that there insufficient credit has been taken for
8 all of the materials that are protecting the fuel, including
9 the pellet itself, the fuel cladding, the 316 nuclear grade
10 material, the outer package, and, of course, the mountain
11 itself.

12 Now, we have done some very simple estimates of
13 what this could save if this approach was proven to be
14 valuable, and it's in the billions of dollars. Our simple
15 little calculation demonstrated \$5 billion would be pretty
16 easy to achieve in savings.

17 Finally, and I think since Yucca Mountain is sort
18 of a moving target with GNEP and with TADs and a number of
19 other things changing, this design is robust enough to
20 actually be compatible with all those concepts.

21 Next chart? This is a chart, and I borrowed
22 liberally from various Yucca Mountain project publications
23 and presentations, and I'll use those through this talk, I
24 don't want to imply that we generated this ourselves. This
25 chart demonstrates, or is supposed to show the range of

1 vulnerability to Alloy 22 to localized corrosion. And, as
2 you can tell from the chart, the red box is the range of
3 vulnerability. If you stay below 85 degrees centigrade and
4 below 50 percent relative humidity, this material should not
5 be vulnerable to any type of localized corrosion.

6 On the other hand, if you stay above 120 degrees
7 centigrade and less than 50 percent--say above 120 degrees
8 centigrade, it wouldn't be susceptible to any type of
9 localized corrosion. My personal opinion is that this is a
10 very conservative criteria, and just to make sure that
11 everybody understands, I believe Alloy 22 is a superb
12 material. I think it's just a little too expensive.

13 Next chart? Okay, our first step is to get the
14 heat. We do this by increasing the number of assemblies that
15 go into each package. In the case of the BWR packages, we
16 could fit in up to 64 assemblies by a very small increase in
17 the outer diameter, and save about a 1000 waste packages, and
18 their related titanium drip shield.

19 With the same diameter change in the PWR assembly,
20 we could increase the number of assemblies from 21 to 29.
21 Now, of course, that all requires the whole system of work,
22 we have not studied the mountain itself. It's way beyond the
23 scope of our study.

24 The Alloy 22 and the titanium drip shields would be
25 replaced by a material called Core-10. It's commonly known

1 in the industry, in the construction industry, as weathering
2 steel developed by the United States Steel Company, and is
3 widely used for unprotected steel applications for wet and
4 dry environments, alternating wet and dry environments. This
5 material is well understood, and has a lot of favorable
6 characteristics.

7 We would replace, we propose replacing the Alloy 22
8 and the drip shield with a very thick Core-10 steel by ASTM
9 grade of A588. And, I'll show you later, that's a very--with
10 that thickness, it's sufficient to meet the strength
11 requirements of the packages and the drip shields.

12 Also, we need, in order to get the temperatures
13 that we'd like to extend the dry period, we need to add
14 backfill. So, in our calculations we'll describe, we add
15 about 20 centimeters of backfill to get the temperatures we
16 want.

17 Just for reference, just on a raw material cost,
18 and this is from a little Alloy calculator from the London
19 Metals Exchange, Core-10 would be about 21 cents a pound,
20 compared to Alloy 22 about 10, and titanium in the \$13 to \$16
21 range. Now, that does not include the fabrication costs.
22 And, one thing about low alloy carbon steel is they do not
23 require complex post-low heat treatment, shop peening
24 techniques, laser peening techniques, and are very
25 fabricable.

1 Next slide? This just gives you a little idea of
2 what Core-10 steel looks like. There are a variety of
3 different steels. We picked A588, but there's no reason to
4 believe that the others wouldn't work just as well. And,
5 they are superior to regular carbon steel as far as corrosion
6 resistance. They form a dense protective oxide film, and it
7 lasts--well, we don't know how long it lasts. But, our
8 estimation is it will grow very, very slowly over the time
9 period of interest. You can tell it's about 98 percent iron
10 with a little bit of copper, a little bit of chromium, nickel
11 and vanadium. All of these elements contribute to the
12 oxidation resistance of this material.

13 Next slide? This is kind of a busy chart, and I
14 don't want to waste your time on it, except to point out that
15 the reference--let's take the BWR reference design. We have
16 the 316 nuclear grade, and its mechanical properties, the
17 Alloy 22, and the titanium drip shield. And, these are the
18 thicknesses of the various components of the package. We
19 eliminate the drip shield, and we keep the 316 nuclear grade
20 exactly the same. And, so, we wind up with a net of 15
21 centimeters thickness as compared to 17--I'm sorry--15 versus
22 compared to 22.

23 But, when you do take into account the thickness
24 and the mechanical properties, the combination of the 10
25 centimeter wall, A588 plus the nuclear grade, will meet the--

1 will meet or exceed the strength requirements of the
2 titanium, plus the Alloy 22.

3 Okay, next slide? Now, this is a chart from one of
4 the early Yucca Mountain studies that actually got us
5 interested in this idea. And, the reference is down there.
6 I don't remember the author. But, basically what was done,
7 this was for, I think, about a 30 year ventilation period,
8 and then closure. And, there were two temperature pulses,
9 one without backfill, and one with quite a lot of backfill.
10 I don't recall how much backfill was put in there.

11 But, what interested us was that even though you
12 can't see it, what this backfilling process did is extend the
13 dry period above 120 degrees centigrade, and that's our goal,
14 to stay above 120 degrees centigrade as long as possible,
15 because in that condition, the packages dry and there's no
16 risk from aqueous corrosion of any type. Below 120 degrees,
17 you can get into aqueous corrosion phenomenon, but we want to
18 avoid that, and delay that for as long as possible. If we
19 do, of course, we have taken advantage of radioactive decay
20 and minimized the risk of any potential breach of the
21 package.

22 There was a problem here that if you take the waste
23 package surface temperature up to 400 centigrade, you would
24 exceed the fuel cladding maximum temperature of 400
25 centigrade that's the current guidance for storage from the

1 NRC. So, we started to look at ways to achieve a high waste
2 package surface temperature without exceeding the 400 degrees
3 C. fuel cladding temperature limit.

4 Now, we were quite impressed with ourselves until
5 we discovered prior work done by the project about eleven
6 years ago, which in this case, the project was looking at
7 carbon steel as an outer package, again, with 316 or some
8 other stainless steel as the inner package. This was work
9 done by Henschel, and what he did is he took--there was an
10 analysis of what the--this is on a linear scale now, and what
11 he did is he took an earlier analysis of what the thermal
12 pulse would be, and he just added 100 degrees centigrade to
13 every time step, and he came up with this extreme temperature
14 history curve. And, he came up with an equation that
15 described that curve.

16 Then, he calculated, based on existing data of the
17 oxidation behavior of Core-10--not even Core-10 steel, just
18 plain carbon steel, had no alloying elements at all, and
19 there is quite a bit of data on the behavior of this material
20 in dry air as a function of temperature. It corrodes by
21 parabolic mechanism, which diffusion controlled, and had the
22 well-defined Arrhenius equation, which describes it over a
23 temperature range of probably 400 centigrade up to as high as
24 900. But, we're of course interested only in the lower
25 temperature range.

1 What he did with this extreme temperature history,
2 he calculated that in 5000 years, you would only get about 1
3 ½ millimeters of metal loss. That's 1 ½ millimeters out of
4 10 centimeters. So, it's very small metal loss. He also
5 added a conservatism in the treatment, where he said okay, at
6 every 20 microns of oxidation, we'll assume that the oxide
7 cracks, and is no longer protective, so what you'll have is a
8 whole series of parabolic oxidation steps each time a new
9 passive film grows to protect the material.

10 So, that was pretty good calculated oxidation
11 resistance. We wanted to learn more about that. But, we
12 wanted to learn it based on our concept of a higher
13 temperature package, but without exceeding the 400 C. limit.
14 This chart shows our calculation using Professor Danko's
15 multiplex code for a 2nd Generation design, containing 29 PWR
16 assemblies, and 20 centimeters of backfill.

17 As you can see, we can keep below the 400
18 centigrade, this is the fuel cladding temperature, and that's
19 for assemblies in the middle of the package. But, we can
20 only achieve somewhere around 280 degrees C. outer maximum
21 package surface temperature.

22 But, the important thing--I should say we would
23 like to do more work because we have some ideas to minimize
24 the Delta T between the package surface and the cladding
25 maximum temperature, but we have not done that work.

1 What is important at about 100 is that we can
2 extend the dry period to about 6000 years by this technique,
3 and we think that's a valuable benefit because it gives us,
4 you know, about 4000 years more of radioactive decay to
5 minimize the risk in case we had a breach of the package.

6 We can skip this one. Our calculation only went
7 out to 5000. This is on a linear scale. So, we just
8 extrapolated to where our 6000 time period is.

9 The next one is a key chart. And, this is the
10 well-known radio toxicity as a function of time curve. And,
11 this concept takes the referenced Yucca Mountain package, and
12 extends the dry period from about 2000 years to about 6000
13 years. Now, it has reduced the impact of actinides, but it
14 hasn't brought them down to the natural uranium ore standard.

15 But, more important things are going on than this
16 study, and that is GNEP. GNEP will take out many, if not
17 all, of the actinides, and change this curve substantially,
18 and will bring us to a much lower, shorter time, that we need
19 to worry about. And, so, it's my hope that GNEP moves
20 forward and that we make use of this valuable resource,
21 rather than just burying it forever.

22 So, with that concept, we tried to visualize what
23 will happen using the temperatures that we calculate, and the
24 oxidation that we know for carbon steels, and this is what we
25 would expect, assuming 60 years of ventilated operation

1 preclosure, and low humidity environment, and we would reach
2 150 C. peak surface temperature. Now, this is a little bit
3 of an artifact because in our calculation, we added backfill
4 even during the preclosure. So, it won't get that hot, but
5 if it did, if we added backfill right on top of the waste
6 package, we would reach the 150 degrees C. So, it would
7 oxidize a little bit in the first 60 years.

8 The next chart, from 60 to 6000 years, we know that
9 water--it's possible, although it's never been demonstrated
10 unless somebody can correct me, it's possible that water will
11 be coming down, but it will evaporate in the environment
12 above the drip shield--I mean, above the waste package. I
13 didn't show the backfill itself. I didn't know how to do
14 that. So, in the backfill, water will evaporate or be
15 combined with the backfill material, but it will never reach
16 the package surface. And, during that time, we calculate,
17 and very good agreement with Henschel's work, about .01 to .2
18 centimeters of oxide thickness. It may be small, but it
19 turns out we believe it--we'll tell you why we think it's
20 important.

21 The next chart is okay, we get past 6000 years.
22 The package has really cooled down. It's now down in the
23 range from 120 to 85 degrees centigrade. The environment is,
24 we assume, very humid. And, there is some possibility of
25 some liquid phase reaching the surface of the package. Well,

1 the first thing that will happen is the liquid phase will
2 boil dry. It will evaporate and leave an oxide--a mineral
3 scale on top of the iron oxide scale. But, you know, the
4 issue is would that damage the package? And, of course, that
5 depends on how much water is reaching the package. We think
6 it's going to be very little water. And, so, that's not a
7 great concern.

8 With time, next slide, we actually could have
9 liquid form and stay there as a liquid, either as a
10 deliquescent salt saturated liquid, or just well water
11 liquid. And, again, the damage to the package depends on how
12 much liquid there is, and what are the characteristics of the
13 oxides that are protecting the package. And, I want to
14 emphasize the "may" because I think that's central to the
15 whole discussion of Yucca Mountain, how much water can really
16 reach the package.

17 And, even more importantly, if the package
18 breached, how could it really, mechanistically, how could you
19 get it through the 316 stainless steel container, get it
20 through the zirconium alloy cladding, which I know is a very,
21 very corrosion resistant material, dissolve some UO₂, and
22 then come out through this same tortuous path and contaminate
23 Nevada. I just think that's unrealistic.

24 So, why is this material--why do we think Core-10
25 is such a great material? Well, just for comparison, Alloy

1 22 is a great material. It's great for just about any
2 environment you can think of, sea water, salt water, chemical
3 industry, you name it, but what protects it is a very thin
4 passive film. And, you can argue whether that's nanometers
5 or angstroms, but it's a very thin passive film, and in the
6 right environments, if it's fractured or damaged, it will
7 reform and protect the material. In the wrong environments,
8 the passive film could fail, and you can get into localized
9 corrosion, or stress corrosion cracking, or other mechanisms.

10 This material is protected by two mechanisms.
11 First, we will form a passive oxide film. We know that forms
12 because of the parabolic nature of the corrosion. In
13 addition, over that long period of time, we form a very
14 thick, we're talking of the order of a .1 millimeter to 1
15 millimeter iron oxide, it's a dense oxide scale, but it could
16 be cracked. But, the fact of the matter, on top of that, is
17 a mineral deposit. And, so, any water, unless it comes in in
18 large quantities, will have to get through these layers,
19 somehow damaged passive layer, and somehow attack the metal.
20 And, I believe that's really an additional barrier, which
21 makes a lower cost material do the job of a higher cost
22 material.

23 Now, what happens when temperatures get really low?
24 Of course, we can't duplicate Yucca Mountain times and
25 temperatures in our laboratory. As everyone who has worked

1 on this project knows, it's one of our biggest problems, how
2 do you predict long service life for our model? How do you
3 do accelerated testing? And, we're going to try and study
4 that in our future work at Nevada.

5 But, we do have artifacts from history that tell us
6 a little bit about how the material behaves in the event you
7 have it exposed in a water containing environment.

8 Next slide? These are iron spikes dug up in
9 Scotland. The longest spike is about 15 inches, and they are
10 nails, and there were--apparently, the Romans were having a
11 lot of trouble in 87 A.D., and they had to leave England and
12 Scotland, and go pass by the rebellious Gauls. And, they had
13 all this iron and they couldn't transport it back, and, so,
14 they buried it. And, anybody who has been to Scotland knows
15 that is not a dry Yucca Mountain type environment. It's wet
16 and dry.

17 Well, it was forgotten. Nobody--someone, I don't
18 know how they found it, dug it up after 1900 years. And,
19 these were just wrought iron. These were not alloy steel.
20 They were not even good carbon steel. They were just iron
21 spikes. So, these unprotected, unalloyed nails, suffered
22 negligible corrosion during this totally unprotected
23 exposure. We would expect a Core-10 material in a Yucca
24 Mountain environment to perform far, far better than that.

25 So, in conclusion, and I didn't go into how we came

1 up with the cost savings. That's in the paper, and it was
2 published in Nuclear Engineering and Design last year. We
3 think eliminating waste packages takes out about \$2 billion,
4 and that's based on a--well, maybe we can get into that in
5 the questions and answers if there's interest. But, it's a
6 lot of money.

7 Maintaining the dry environment and using more
8 spent fuel helps create these savings, replacing the Alloy 22
9 with a much cheaper and much more fabricable material, which
10 in our opinion, will do the job. That's A588, or one of
11 those classes of materials. And, of course, eliminating the
12 drip shield.

13 Our analysis shows we can achieve higher waste
14 package surface temperatures without exceeding the peak
15 cladding temperature of 400 C., and that will extend the dry
16 period by thousands of years. And, we believe that our
17 literature data on oxidation and corrosion of these steels,
18 plus the archeological data from things like these wrought
19 iron artifacts, supports another look at carbon steel for the
20 Yucca Mountain application.

21 Thank you.

22 GARRICK: Thank you. I should have announced at the
23 beginning that due to our mix-up, we don't yet have the
24 viewgraphs, or the copies of the viewgraphs for this
25 presentation. But, we are getting them, and they will be

1 made available.

2 David, do you have a question?

3 DUQUETTE: Duquette, Board.

4 I think the Gauls that the Romans went after were
5 probably my ancestors. But, with a little less levity, I'm
6 sure you're aware that the Core-10 steel in wet ionic, wet
7 electrolytes, doesn't behave any better than regular carbon
8 steel, and, so, it can be used, for example, on surfaces of
9 buildings, and things like that, but if you attempt to use
10 it, for example, for guard rails, has failed entirely because
11 it corrodes off at the ground level at the same rate as
12 regular carbon steel does.

13 ARMIJO: Yeah, but--

14 DUQUETTE: Let me finish my comments. So, that's number
15 one. Number two, the environment you're talking about does
16 contain both chloride and nitrate and the Alloy 22, the
17 project would like to take credit for the nitrate, but in
18 this environment with carbon steel of any kind, the nitrate
19 and chloride are going to act synergistically, neither one is
20 going to protect the other from happening. They're both
21 going to add to the ionic concentration of the water. Your
22 temperature of 120 flies in the face of what we've been told
23 about the mountain, which we'll see deliquescence of 150 to
24 160. I don't know of any work presently, and I presume
25 you're going to be doing it, that looks at the corrosion by

1 the way of Core-10 at 150 or 160 degrees Celsius. I don't
2 think there's very much work out there on that. Even using
3 the Arrhenius equation, I think you're going to have some
4 problems putting that together with your environments.

5 So, I think it's a nice idea. Of course, it's
6 revisiting something that was done a long time ago, because
7 the original package was going to be carbon steel, and you
8 simply let it corrode away. You simply took the allowance
9 for it, and let it corrode away. But, I don't think Core-10
10 is going to be the answer in this particular case, because of
11 the specific environment that's a fairly strong electrolyte,
12 and you will have--well, it's almost surely going to have
13 some deliquescence. Maintain the temperature at a higher
14 temperature to guarantee evaporation is probably a good idea,
15 period, if you're going to use carbon steel.

16 But, I think some of your arguments would have to
17 be proven because I just don't know what the oxide is going
18 to be like at 150. We do know that in the mountain today,
19 that carbon steel is corroding rather rapidly, as we speak,
20 and that some of it had corroded at quite a high rate, to the
21 point where it's, I believe, had to be replaced, or at least
22 had to be retired because of--but, things like rock bolts are
23 corroding, and there are some doors that seal off parts of
24 the vault, and those are corroding. And, so, there's
25 corrosion going on in the mountain at room temperature at the

1 present time of carbon steel.

2 So, I don't have a problem with using this if we're
3 going to use a corrosion allowance, and allow the container
4 to disappear. But, I don't think it's going to be a
5 replacement for Alloy 22.

6 ARMIJO: I'd like to respond to that.

7 DUQUETTE: Sure.

8 ARMIJO: First of all, I assume this is a corrosion
9 allowance method. You know, I don't expect that this
10 material would be unharmed and pristine, and I really don't
11 believe it's only going to have a tenth of a millimeter of
12 oxide at the end of even 10,000 years. But, I know there's
13 plenty of margin, and I know that this material is not
14 acceptable to many of the localized corrosion phenomenon that
15 affect the stainless steels.

16 And, then, if we could just go to the backup Slide
17 21? The thing that stimulated us a lot was the issue of how
18 much water are we really worried about? If Yucca Mountain
19 was a swamp, I could see the argument. We're talking about
20 long periods of dry, maybe some periods of wet and dry
21 operation. If it's going to be submerged, then Yucca
22 Mountain is not the right material--not the right place for
23 this material. So, again, nobody knows for sure, but the
24 international peer review that was done in year 2000 asked
25 the question and made the statement, the natural dripping of

1 groundwater hadn't been observed, and they also made the
2 point that there was a great evaporation potential from the
3 waste packages from the decay heat, and that would decay, you
4 know, boil away up to a thousand liters a year per container.
5 And, even at 10,000 years, it would boil away 100 liters a
6 year.

7 Do we expect 100 liters a year per container to be
8 hitting the waste package? I don't--I hope that's not the
9 expectation. And, I hope it's not greater than that.

10 The other point, this was made by the Secretary of
11 Energy in 2005 when he recommended that the President approve
12 the Yucca Mountain facility, and he repeated the point that
13 there are very small amounts of water, and, you know, there's
14 no indication that water is actually dripping into the
15 tunnels.

16 I just say that our concern is overstating the
17 threat, you know, water is corrosive, you know, water is
18 corrosive to Core-10 just like any other materials. But,
19 it's a matter of how much, and how much and how long, and if
20 you have a generous corrosion allowance, you could dissolve
21 that whole carbon steel thing away. I doubt it would ever
22 happen, but you'd still have the package, the fuel protected
23 by 316, and the zirconium alloy cladding. And, I've been in
24 the zirconium alloy business for a long time, and that is a
25 very corrosion resistant material. It's not going to fall

1 apart spontaneously. There are mechanisms that have to act
2 for the cladding to fail, even at those long times. And,
3 they're not going to be much of a threat.

4 But, the whole idea of this thing is can we make
5 Yucca Mountain perform its mission at a lower cost, and this
6 was our contribution to it. In the event that GNEP moves
7 forward, I think the use of Alloy 22 will be more in
8 question, because the lifetime duty is going to be much
9 shorter. So, that's really my response.

10 GARRICK: Go ahead.

11 LATANISION: Latanision, Board.

12 I would generally agree with Dave Duquette's
13 concerns, Sam. And, I would also add that it is known that
14 with constructional steels, hot nitrates will induce stress
15 corrosion cracking. And, so while we have this perversity of
16 nature right now in this context that, well, the current
17 generation of alloys, C-22, the external of the waste
18 package, we're looking at taking advantage of nitrates in
19 terms of reducing the potential for localized corrosion. You
20 would have the converse situation with a constructional steel
21 that you would have to worry about another form of localized
22 corrosion, and perhaps a more severe form in terms of stress
23 corrosion cracking.

24 ARMIJO: Yes, I agree with you, Ron. That's amenable to
25 testing. You know, if you came up with some realistic or

1 even aggressive, a more conservative environment with
2 nitrates, you know, the first thing you would do is you'd
3 test wells. But, these steels, even though they are
4 weldable, don't require the heat treatment to reduce the
5 residual stresses that would be the cause of the cracking.
6 You know, it would be just testing to see if it is
7 susceptible.

8 But, you know, again, I keep going back to the
9 amounts of water and the concentrations, and I just think
10 it's hard for me to see a real threat to a lot of packages,
11 unless Yucca Mountain is floating in water. And, so, I think
12 if that's the case, then this would not be the material to
13 use.

14 GARRICK: Any other questions? From the Staff?

15 (No response.)

16 GARRICK: Well, this is wonderful. The speakers have
17 been great this morning. We're right on time. Thank you
18 very much, and we will adjourn until 1 o'clock.

19 (Whereupon, the lunch recess was taken.)

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AFTERNOON SESSION

1

2 GARRICK: I want to remind you again of a couple of
3 minor changes in our program. One is the changing around of
4 the two presentations on Saturated Zone Testing, and the
5 Near-Field Chemistry. And, the other is to make sure we
6 allow a few minutes before the first break to have some
7 public comments from people that are going to have to leave
8 early.

9 And, now we're going to hear from Jim Laidler of
10 Argonne National Laboratory, who is going to tell us a little
11 bit of what the Board has been very anxious to hear about,
12 the Global Nuclear Energy Partnership. Jim?

13 LAIDLER: I've been selected to provide the light after
14 lunch entertainment on something non-controversial. So, I'll
15 talk about GNEP.

16 I'm not going to cover the motherhood things about
17 GNEP even though I believe in them very strongly. But,
18 suffice it to say that GNEP is about ensuring the
19 sustainability of nuclear energy in the future. It's about
20 closing the nuclear fuel cycle. And, it's about a complex
21 system of light water reactors and fast spectrum reactors.
22 I'm not going to talk today about the closure of the fast
23 reactor fuel cycle, because that's something that's off into
24 the future. But, our immediate priorities are on the
25 treatment of light water reactor spent fuel. And, so, I want

1 to talk about the perceptions that we have about what the
2 waste management aspects of the GNEP separations are.

3 I think we can just charge into it, so let me show
4 you the first slide. There are projections of significant
5 growth in nuclear power in the world. By 2050, we may have
6 something like 1100 reactors in operation. There are a
7 number of reprocessing plants currently in operation.
8 Perhaps the most recognizable are the French plant at
9 LaHague, which has the capacity, name plate capacity of
10 around 1600 tons per year, the new Rokkasho-mura plant in
11 Japan, the Sellafield plants of the UK. But most
12 importantly, I wanted to point out some of the planned
13 reprocessing plants in the world.

14 The Chinese are talking about building a UP3 type
15 reprocessing plant maybe by 2020, 2025. The Russians are
16 planning a replacement for the Mayak plant, which would go
17 into operation in roughly 2025, at a scale of 1000 tons per
18 year. And, in the GNEP program, we are looking at the
19 deployment of a reprocessing plant that we call the
20 Consolidated Fuel Treatment Center, the CFTC, and have that
21 in operation by sometime between 2020 and 2025 at a very
22 large scale, a scale that's really dictated by the present
23 generation rate of light water reactor spent fuel.

24 So, we're talking about some major new
25 installations in the world, in addition to those that are

1 shown on here as planned plants. We anticipate that the
2 French will construct a replacement plant for their two
3 LaHague facilities, and that that would come on line in 2035
4 or 2040.

5 The main point of bringing this up is that we are
6 trying to exert our national influence on what those plants
7 do. And, in GNEP, one of our precepts is that we will not
8 separate pure plutonium, and we want the rest of the world to
9 do the same. And, so, what we are aiming to do in the GNEP
10 program is to demonstrate that a reprocessing plant can be
11 operated efficiently, economically, and not act like a PUREX
12 plant, and not separate pure plutonium.

13 Next, please? And, this is what PUREX looks like.
14 If you're not familiar with the process, it involves the
15 chopping and nitric acid dissolution of spent fuel.
16 Currently, in the commercial plants, there are a number of
17 releases during that operation. Iodine is released to the
18 ocean in France and the U.K. Tritium, Krypton and Carbon-14
19 and CO2 are released to the atmosphere.

20 The PUREX process then follows the chopping and
21 dissolution step. This is a co-extraction of uranium and
22 plutonium. There is a separation of the pure stream of
23 uranium, and a pure stream of plutonium. In the French
24 plant, at least, the plutonium is sent as the nitrate
25 solution to precipitation step where plutonium oxide PuO₂ is

1 produced for use in mixed oxide fuel thermal recycle.

2 Uranium is converted to UO₃, and presently, it's
3 stored in France. There has been some recycle of the
4 uranium, both in the enrichment plant, and in the thermal
5 reactors. But, the balance of the fission products and the
6 minor actinides, neptunium, americium and curium, with all
7 the fission products then, other fission products, are sent
8 to a vitrification step, where they are put into glass and
9 packaged for ultimate disposal in a geologic repository.

10 Next, please? And, we have issues with that
11 process. Even though it's been around for a long, long time,
12 50 years or more, and has been successfully refined over the
13 years, and optimized, it does produce a pure stream of
14 plutonium. And, that is contrary to our national policy,
15 which states that we will not use plutonium in the civil fuel
16 cycle.

17 Minor actinides in the PUREX process are sent to
18 the waste, which increases the volume and radiotoxicity of
19 the waste material. The heat generators, the cesium and
20 strontium, the americium and curium, go into the waste
21 stream. So, there is no benefits to repository heat
22 management. And, frankly, you can make a number of minor
23 modifications to the PUREX process and call it proliferation
24 resistant, but any good chemist can change that process very
25 easily to extract a pure plutonium stream.

1 Next, please? So, we began in the GNEP program to
2 design a separations process that was really aimed at
3 optimizing waste management. Maybe the first time that
4 that's ever been done in the nuclear industry, but we had a
5 lot of bad examples that we could point to. One, of course,
6 is all that high level tank waste. And, so, one of our going
7 in criteria was we'll generate no high level liquid waste
8 that requires extended tank storage.

9 We want to limit the emissions from the plant
10 because at the size that we're talking about, the releases
11 could be significant, and might make the plant a little hard
12 to license. So, we want to recover the iodine, krypton,
13 tritium and Carbon 14.

14 We also want to hold the cost down. And, as you
15 can see, we're working at counter-purposes, because by doing
16 all the things that help waste management, we add complexity
17 to the reprocessing plant. And, so, we have to be very, very
18 careful that the things that we do are the right things, so
19 that we don't drive ourselves out of business. We want to
20 keep the increase in busbar cost of electricity to more than
21 just a few mils, four or five mils would be nice, to
22 accomplish the reprocessing of the spent fuel.

23 We want to get very efficient recovery of the
24 important radionuclides. For example, iodine and technetium,
25 we have a recovery target of 95 percent. Now, that's about a

1 factor of 20 reduction in the source term. We think we can
2 do better, and we're working on that.

3 We want to get about a ten-fold reduction in high-
4 level waste volume relative to direct disposal. And, in the
5 integrated process, that is, the whole plant process, we're
6 targeting for more than 99.9 percent recovery of the
7 transuranics and the cesium and strontium.

8 Next? And, this is our reference process that we
9 call UREX+1a. And, you will see where this came from in just
10 a few moments. We begin by--and, this is something only a
11 chemist or a chemical engineer could love, so I'm not going
12 to waste a lot of time on it, other than to say that we have
13 a lot of process steps that are targeted at specific
14 radionuclide to removal. We have a four step--four
15 extraction step system, where we successively remove cesium
16 and strontium, the transuranic elements together with
17 Lanthanide fission products, and then we separate the
18 transuranics from the Lanthanides. This is a so-called group
19 extraction process, where we're pulling out all the
20 transuranics in one stream, and then sending that stream to
21 fuel fabrication for irradiation as fuel in a fast reactor.
22 This is not a thermal recycle operation.

23 Next? We've tested this process. It works pretty
24 well. We get very high recovery efficiencies. Of course,
25 this is that lab scale, so we need to demonstrate this at a

1 larger scale in the future. But, we do get very high
2 purification of the transuranics. One of our major concerns
3 is the removal of the lanthanide fission products, because we
4 have evidence that they can cause fuel cladding chemical
5 interaction in fast reactor fuel. The purities of these
6 recovered elements are at least three-ninths, in some cases
7 more than that.

8 Next, please? Now, this is what we call the suite
9 of UREX+ Processes. If you recall, the initial flow sheet I
10 showed, the first extraction step is called UREX, where we
11 separate out the uranium. And, then, all the other process
12 steps that are added just make it UREX+ those other steps.
13 And, we've designed the processes so that they can be tailor
14 made to fit whatever our national policy evolves into. And,
15 so, the first ones, the UREX 1 and 1a, are designed for a
16 group extraction for irradiation of the transuranics as a
17 collection in a fast reactor.

18 The others are designed for what we call
19 heterogeneous recycle, where we split the transuranics into
20 two groups, plutonium and neptunium as a fast reactor fuel,
21 americium and curium, perhaps even separating the americium
22 and curium for use as targets, or in the case of curium, to
23 store the curium until it decays.

24 I just want to emphasize the reason for having
25 these heterogeneous recycle systems is that in the

1 homogeneous case where we do the group extraction, it forces
2 you into remote fabrication, hot cell fabrication of the
3 fuel, which is, we think, an expensive proposition. So,
4 we're looking at heterogeneous recycle, this kind, because
5 you can fabricate that fuel, the plutonium and neptunium
6 fuel, in a glove box, and the costs then become comparable to
7 what, for example, the French are doing at their Melax plant.

8 Eventually, we can evolve into a group extraction
9 system, once we develop the technologies that can make remote
10 fuel fabrication economical.

11 Next? Now, I wanted to show you this flow sheet
12 again, because now I'm highlighting the different waste
13 streams in red here. In the initial step, we perform a low
14 temperature voloxidation with just heating the fuel in an
15 oxygen partial pressure to convert the UO₂ to a higher oxide,
16 which expands, opens up the cladding, fragments the fuel, and
17 if it's done at the right temperature, will release only the
18 tritium. We want to do it in the dry step before the tritium
19 gets into the liquid phase and goes all through the plant and
20 becomes very difficult to recover. So, there's our first
21 recovery step, to get the tritium out.

22 Then, in the dissolution process, we'll remove the
23 xenon, krypton, CO₂. And, in addition, there is a certain
24 amount of fuel, material in the fuel, that will not dissolve
25 readily in nitric acid, and forms undissolve solids, or

1 sludge, if you will, at the bottom of the dissolver vessel.
2 Our intention is to recover that sludge, to wash it to make
3 sure that we've got the transuranics out of it to the
4 greatest extent that we can. And then also left behind in
5 the dissolver operation are the cladding hulls. And, that's
6 a valuable material. So, we want to recover the cladding
7 hulls, wash them with a combination of nitric and
8 hydrofluoric acids, make them non-transuranic, less than 100
9 nanocuries per gram, and use that as one of our waste
10 streams, in fact, to form the matrix for a metal alloy that
11 contains the sludge, the undissolved solids, and the
12 technetium that we recover at very high efficiency, in the
13 UREX process step.

14 We convert, in this step, we convert the technetium
15 to a metallic form, combine it with the hulls and the sludge,
16 and make a metal alloy out of zirconium. So, we've got to
17 locally very strong reducing environment that retains that
18 metallic technetium in the metallic state, where it has
19 limited solubility and limited mobility in groundwater. And,
20 it's contained in the matrix. It's pretty corrosion
21 resistant.

22 We recover the iodine in this step, in the
23 dissolution step as well. I'll show you what we're going to
24 do with that. The cesium and strontium I mentioned, our
25 intention is to place the cesium and strontium in a dedicated

1 waste form in some facility. We have not arrived at a
2 decision on what the facility ought to be, it could be a
3 remote part of the repository, the Yucca Mountain repository,
4 and just let it decay. It doesn't have to be in an exotic
5 container because after 300 years, it's going to be gone.
6 So, we're looking at different ways to immobilize that
7 product.

8 The TRUEX process separates the transuranics and
9 lanthanides from the other fission products, and those
10 fission products in that stream are largely the transition
11 metals. They don't make a very good glass, but they do make
12 a pretty good metal alloy. And, so, we're thinking about
13 combining those transition metal fission products with out
14 alloy that immobilizes the technetium as well.

15 The next step, TALSPEAK, separates transuranics
16 from the lanthanide fission products. The lanthanides make a
17 very nice glass, and, so, that's our present intention for
18 that.

19 Let me see the next slide, please. I mentioned the
20 alternatives of heterogeneous recycle. And, I've said these
21 things. Because of the need to move ahead and not get hung
22 up on a prohibitively expensive fabrication technology, we're
23 going to process, or are looking at a process, let's say,
24 that's called UREX+3. And, this is one where we recycle
25 uranium, plutonium, neptunium as fast reactor fuel. Separate

1 americium and curium and perhaps the americium from curium,
2 that's a fairly difficult separation, but it's feasible, and
3 transmute the americium in probably an epithermal region of a
4 reactor, a fast reactor. And, then, separate, if we can, the
5 curium and just let it decay. It will decay to plutonium and
6 americium, and then that could be recycled.

7 Next? And, here's what UREX+3 looks like. It
8 looks like it's getting a little more complicated, and, sure,
9 it is. We've got two product streams now, but the other
10 products are exactly the same as coming out of UREX+1a. The
11 only difference really is that we've got U-Pu-Np stream and
12 an americium and curium stream as our products separately.

13 Next, please? So, I wanted to talk a little bit
14 about what we envision as the waste forms, and you will see
15 that we are talking about a variety of different waste forms.
16 I'm not sure there's anything wrong with that, because we can
17 tailor those very precisely to do just exactly what we want
18 them to do.

19 The tritium will collect as water, and put it in
20 grout. The cladding hulls, the bulk of the cladding hulls we
21 will wash and then just compact. A portion, maybe 10 percent
22 of the cladding hulls, are diverted for use as the matrix
23 alloy for the metal waste form where we immobilize the
24 technetium and the undissolved solids, the sludge, and the
25 transition metal fission products.

1 Technetium is recovered as a metal, and that's done
2 by absorbing technetium pertechnetate on an anion exchange
3 resin, and paralyzing that resin to convert the oxide
4 to a metal. Now, we have done experiments with that, and we
5 find it forms nice little beads that are free flowing and
6 metallic. So, we think we've got something there.

7 And, I should point out that that's one of the big
8 considerations in all of our process design work, and that's
9 the chemistry, the extraction process is fairly easy. It's
10 designing the plant to operate with a variety of different
11 products to be able to recycle the reagents, to be able to
12 manage the effluent, and to efficiently produce the desired
13 products.

14 The xenon, krypton, we would like to separate those
15 two, the xenon from the krypton. Xenon is not a problem. It
16 can be vented, because it's all stable xenon. But, the
17 krypton 85, we're going to try to capture, and we think we
18 can immobilize that in a zeolite like structure.

19 Carbon-14 is simple. We'll just pass that through
20 a scrubber, a caustic scrubber, and make carbonate out of it.

21 Next, please? Iodine, conventional method is to
22 trap it in a silver-coated zeolite. We may convert it to a
23 more stable compound, like potassium iodate.

24 Uranium, we recover at a very high level of purity,
25 contaminants in the part per billion range, and it's pure

1 enough that we could re-enrich it. The uranium coming out of
2 our light water reactors is worth re-enriching. It may have
3 some 236 in it, but that may be a perfect source of
4 fissionable or fuel material for our international partners,
5 so that they don't have to get into the enrichment and
6 reprocessing business.

7 Cesium and strontium, we plan to go into decay
8 storage. Now, whether that's near surface or deep burial has
9 not been determined yet. We're looking at various options
10 there.

11 And, then, I mentioned the residual fission
12 products. Lanthanides, we'd put into glass, the transition
13 metals into a metal alloy.

14 Now, this is all a work in progress. We've really
15 only scratched the surface on how best to prepare these waste
16 forms, and that's one of our main areas of emphasis this year
17 and into 2009. But, our intention is to have the best
18 possible set of waste forms that meet the objectives of the
19 national program.

20 The next one? These are the estimated volumes.
21 Now, this table is different than the one that's in your
22 handouts, so please, I'm assuming that the Staff will make
23 this correction available to you. They certainly have it
24 available now. But, we've gone through a major exercise
25 recently on trying to get our best possible estimate of waste

1 volumes, and you can see here, Alan Croft hates this column.
2 I don't know that it's all going to be high-level waste.
3 Some of these things are fairly benign and may not need to go
4 into a repository.

5 But, what I'm showing here on the far right column,
6 is the unpackaged volume of these different waste forms. The
7 bulk of the volume comes from cladding hulls. The next
8 largest contributor is the uranium losses. We're now
9 projecting a 99 percent recovery efficiency for uranium--I'm
10 sorry--99.9 percent, and once we lose that additional tenth
11 of a percent, it amounts to a lot of material. So, that
12 we're presently projecting to go into glass. I think with a
13 reasonably conservative estimate for the waste loading, we
14 can probably get it down to the lower number here, 1 cubic
15 meter per 100 tons.

16 Just for comparison, that 100 tons of untreated LWR
17 spent fuel has a volume of about, unpackaged volume of around
18 45 cubic meters. And, if we're smart and do the right
19 things, then we get at least a five-fold reduction in
20 unpackaged waste volume. And, then, a significant reduction
21 when we come to putting it into canisters for ultimate
22 disposal, because we're no longer, in this case, limited by
23 heat generation. So, we can cram these waste forms in pretty
24 compactly. So, we're thinking that probably a factor of 14
25 to 20 reduction in packaged waste volume.

1 There's another option here, and that is to look at
2 the disposition of the zircaloy cladding. We have some
3 industrial firms that may be interested--I think it's in the
4 next slide--to look at recycling of zirconium. We could go
5 into a chlorination process, make volatile zirconium
6 chloride, and then go into the coal process, recover pure
7 zirconium and recycle it. And, we've got some companies,
8 Wachang (phonetic) is one of them, who have said that they
9 will be delighted to accept that material for refabrication
10 of cladding.

11 Now, the question is how long do we continue
12 refabricating, or fabricating zircaloy cladding if we're
13 going to deploy fast reactors. And, I don't know. It
14 depends on the mix of reactors that we have in our system,
15 and we'll probably have light water reactors for a long, long
16 time. They work pretty well.

17 I guess the bottom line is whether we do that
18 recovery and recycling of zirconium is an open question.
19 Now, it may be that a ten-fold reduction in waste volume is
20 just fine. The cladding hulls and the fuel assembly hardware
21 are not really significant contributors to the heat load
22 imposed on the repository. So, it may be that we could just
23 pack them in tightly and say fare thee well to them.

24 I believe that's the end of my presentation. Thank
25 you very much.

1 GARRICK: Questions? Andy?

2 KADAK: Thank you. You know, our interest obviously is
3 in the waste forms, and as you know, we are struggling even
4 with what we know to be the waste form in terms of how it
5 will be, you know, dissolved, or whatever in the repository.
6 How do we deal with that? Because the waste forms aren't
7 really defined into just sort of, I guess, analyses, we don't
8 know what the packaging requirements are, and we certainly
9 don't know how these waste forms will dissolve in water, if
10 you will. What do you advise the Board to do, given where
11 you're going with this, or potentially?

12 LAIDLER: Be patient.

13 KADAK: We have plenty of patience.

14 LAIDLER: We are working very diligently to produce
15 waste forms that are realistic. We've done a lot of
16 preparation with surrogates. But, now, we're trying to get
17 into, because now we're beginning to do demonstrations with
18 the processes and larger scale, we're accumulating more
19 actual fission products, and part of our demonstrations in
20 the next few years will be to complete the fabrication of the
21 waste forms, and then get into the testing of their behavior
22 in a representative environment.

23 The fact that we have a lot of different waste
24 forms may become an issue. I don't know. But, I'm convinced
25 that we can make them cheaper than if--this collection of

1 waste forms cheaper than if we had to put it all in glass.

2 And, that's one of our drivers, is to reduce cost.

3 GARRICK: Garrick, Board.

4 One of the issues in repositories is waste
5 acceptance criteria. And, going in the direction that this
6 goes, with large, much larger number of waste forms, there's
7 going to have to be a lot of attention paid to what
8 constitutes a reasonable criteria. Is that right?

9 LAIDLER: Well, absolutely. And, we think that you can
10 use the same kind of tests that are presently available as
11 measures of the behavior of these different waste forms.
12 We're not really getting that far afield from what's already
13 being produced. The metal waste form may be something new,
14 but there are ways to test that, and we're pretty sure that
15 that's going to be a winner as far as retention of the
16 technetium and the other fission products.

17 GARRICK: Besides proliferation, what are a couple of
18 the other principal drivers for GNEP?

19 LAIDLER: Well, we're trying to prevent proliferation by
20 non-weapon states, and, so, the organization of the program
21 is such that there would be a so-called group of supplier
22 nations, which consist of the current weapon states, plus
23 Japan, and those are countries that have established and
24 significant nuclear infrastructures.

25 Now, what we're trying to come up with is a system

1 where we would provide an assured fuel supply to any other
2 country as long as that country agrees to forego enrichment
3 of uranium and reprocessing of spent fuel. So, that's the
4 motherhood part of GNEP.

5 The other intentions are to make sure that we have
6 a strong nuclear energy program in the U.S. One of the
7 advertised features of the program is to be sure that we have
8 optimized the waste management so that we have the need for
9 only one repository in this century.

10 GARRICK: Andy?

11 KADAK: Just a follow up on that last comment. The
12 Board has heard that depending upon how you design the
13 repository, it could be as many as four to eight times the
14 current legislated capacity. And, I've heard this comment
15 made many times about only one repository. But, when you
16 look at your waste volumes, and you mentioned the hulls and
17 the sludge, depending upon how efficient you are, that number
18 could be much larger than the number that you pointed out.
19 In fact, if you do it wrong, it could be larger in terms of
20 total waste volume requiring geological disposal. What is
21 the sensitivity of your 99.9 to that question?

22 LAIDLER: Well, we're actually giving ourselves quite a
23 cushion. These processes can be made very, very efficient.
24 And, the interesting thing about a reprocessing plant is that
25 the separations part, the chemical separations part, is

1 really the easy part.

2 KADAK: That is the easy part. It's the other stuff.

3 LAIDLER: It's the other stuff, yeah, it's the head end
4 and back end of the process. And, that's where we have to
5 place our emphasis. And, we're counting on input from the
6 industry, and as you probably know, there's a funding
7 opportunities announcement that recently was published to
8 seek industrial engagement in the program.

9 KADAK: One other question. What are you going to do
10 with all the separated uranium? It says storage, but what
11 does that actually mean?

12 LAIDLER: Well, there are two options. One is to re-
13 enrichment, which we can do, because it does have value.
14 It's over natural levels. The other is that it can become a
15 blanket material for fast reactors in the future to breed
16 plutonium.

17 KADAK: You said the magic "B" word, and it wasn't
18 burner, and in your opening comments, you talked about
19 sustainability, and it's pretty clear that just the burner
20 reactors are not sustainable, and you would need to do
21 breeding, I would assume, as part of the long-term future for
22 this program.

23 LAIDLER: I think everybody recognizes that ultimately
24 we have to go to breeder economy. The burner reactor, as I
25 see it, is really the evolutionary step toward that.

1 KADAK: Can you do breeders without separating
2 plutonium?

3 LAIDLER: You can do the same kind of group transuranic
4 separation, or a plutonium, neptunium separation.

5 KADAK: So, yes?

6 LAIDLER: Yes, the answer is yes. But, breeders imply
7 reprocessing. There's no way around that.

8 KADAK: Yes.

9 GARRICK: Ron, and then Howard?

10 LATANISION: Latanision, Board.

11 Accepting for the moment that the chemistry can be
12 performed in the way you describe it, what is the path
13 forward in terms of geo-politics, I suppose, for lack of a
14 better word? This whole concept is derived on the notion
15 that other nations would become partners in this exercise.
16 So, in the best of all worlds, what is your path forward?
17 When would you see something like this taking form and
18 handling waste?

19 LAIDLER: We have been engaging a number of other
20 countries in the past year and a half. We have met with all
21 of the potential supplier nations, and at the ministerial
22 level, gotten their agreement and their support and their
23 commitment to go ahead with the partnership. We had a
24 meeting in Vienna with, I don't remember, something over 30
25 different countries, to get their opinions on participation

1 in the partnership. And, most of them are favorable. There
2 are some countries who are suspicious of it, that we'd be
3 forming a Cartel that would dictate prices, like OPEC, and we
4 tried to assure them that the supplier countries, the
5 potential supplier countries are very strong competitors, and
6 the intention is to have it be a competitive market, that the
7 client countries, or the user nations, can shop around for
8 the best price. And, each supplier nation would set its own
9 price, depending on what they want to do. And, the processes
10 may be slightly different. The only constant is going to be
11 no plutonium separation.

12 LATANISION: Just a follow up. Given all that, what's
13 the time frame that you look at for all this to come about?

14 LAIDLER: Well, I'll be a little bit older than I am
15 now. I'd give it 20 years.

16 LATANISION: 20 years.

17 ARNOLD: Arnold, Board.

18 I have a similar question. What year closer to
19 then do we have to make a major decision on building the
20 first major facility of this program? Because that's when
21 it's going to get crunchy.

22 LAIDLER: Exactly. Yes. We have selected a June of
23 2008 decision point by the Secretary on how to proceed with
24 GNEP, whether to go just R&D, or to go full blast into large
25 scale facility construction. It's hard to say what form that

1 would take, but our target is to have, if we go to an
2 industrial scale plant, to have it in operation by 2020.

3 ARNOLD: Which would mean a decision to build it would
4 have had to have been yesterday.

5 LAIDLER: Almost. If we go in in June of '08, we're
6 probably a year and a half late.

7 ARNOLD: So, you're saying fiscal '09 would have to have
8 construction at least anticipated, and environmental impacts
9 and licensing and all that?

10 LAIDLER: Well, we're working on an environmental impact
11 statement at the moment. We've had expressions of interest
12 by a number of industrial concerns. We know that there are
13 some companies out there that would just love to build a big
14 reprocessing plant. In fact, my company would like to build
15 one.

16 GARRICK: Jim, you mentioned a 2020 production plan. Is
17 there a plan to have a pilot plant between 2008 and 2020?

18 LAIDLER: What a good question. I think not. If we're
19 on a 2020 schedule, we don't have time. We're operating
20 pretty much in the French mode right now. The French claim
21 that piloting at a few kilograms per year or a few kilogram
22 batches is adequate to go up to 1600 tons per year. I'd feel
23 a lot more comfortable if we had a pilot plant, but I don't
24 think we can do it on that schedule. We are now at the one
25 to five kilogram batch in size.

1 GARRICK: Andy?

2 KADAK: What part of this international agreement is to
3 take spent fuel from other nations and to reprocess it and
4 ship them back fresh fuel, I assume, or reprocessed fuel? Is
5 that correct?

6 LAIDLER: Yes.

7 KADAK: Do we keep the waste, or does the waste go back
8 to the country?

9 LAIDLER: It depends. If it's the French participation
10 in the program, and the French have a law that prohibits the
11 acceptance of waste from other countries. So, they would
12 have to find another of the partner, the supplier partners,
13 to find a repository for the--or a disposition path for the
14 waste stream. And, that is something that's not resolved.
15 What we're trying to do in our program is show that the waste
16 that we produce is comparatively benign, and maybe in fact
17 the French could change their law.

18 KADAK: But, the U.S. will retain possession of the
19 high-level waste?

20 LAIDLER: No, that decision has not been made. But, I'm
21 going to make it good enough that you could put it in my back
22 yard.

23 KADAK: The high-level waste? Really? Because,
24 clearly, that's a question about capacity of the repository
25 and needing only one. If we're going to be the sort of host

1 of whatever deals we can make for reprocessing, those numbers
2 may change a little bit in terms of only needing one
3 repository.

4 LAIDLER: Yes, if you look at the--there's been a number
5 of studies strictly based on heat management in the
6 repository, and it can be fairly convincingly shown, if
7 that's the only criteria, is heat management, that you can
8 expand the effective capacity of the repository, of Yucca
9 Mountain, by maybe 200 times.

10 KADAK: 200 times? So, it wouldn't be a problem for us
11 to store international spent--waste, I should say, not spent
12 fuel?

13 LAIDLER: I'm not going to get into that one.

14 KADAK: But, that's a question, an important one.

15 LAIDLER: Sure, it's a question, and it's a policy issue
16 that we have to deal with. It's not going to be dealt with
17 by the technicians like myself.

18 KADAK: When do you think the fuel for these burner
19 reactors will be ready to put in a fast reactor for reliable
20 operation? These actinide fuels.

21 LAIDLER: One of the facilities that is part of GNEP is
22 called the Advanced Fuel Cycle Facility, and that is
23 basically a research laboratory on a grand scale, where we
24 would have the capability to fabricate the fast reactor fuel,
25 either from--for the heterogeneous case or for even the

1 homogeneous case. But, that will be ready by about 2017.

2 KADAK: With tested fuels that are suitable for, quote,
3 unquote, burning?

4 LAIDLER: Well, that's another problem. We don't have a
5 fast reactor in this country. We can test fast reactor fuels
6 in thermal reactors with some filtering, you know, to get a
7 sort of hard spectrum. But, our alternatives are go to the
8 Russians or the Japanese who do have operating fast reactors.

9 GARRICK: Okay, Ron, and then Carl DiBella.

10 LATANISION: Just a short question on budgets and cost.
11 Do you have a sense of what it would cost to put all of this
12 together and, given that sense, what is the current budget
13 for this effort?

14 LAIDLER: The current budget in '07 is, what, 230
15 million. The request for '08 is 405 million. And, once we
16 get into the industrial scale plants, of course, it goes on
17 into major system acquisitions and line items. So, I don't
18 know, there have been estimates for a plant the size we're
19 talking about, the 2500 ton per year plant, with fuel
20 fabrication, that combined total may be 15 billion, or so.
21 It's a big business.

22 GARRICK: Carl?

23 DI BELLA: Carl DiBella, Board Staff.

24 First of all, as a chemical engineer, I want to
25 thank you for those wonderful blackboard diagrams. I've got

1 a question on your overhead Number 5, just a clarification.

2 LAIDLER: Okay.

3 DI BELLA: It's about costs, and you said added fuel
4 cycle costs, and then you said four or five mils is sort of
5 what you're shooting for. What is that added to? I mean,
6 today's fuel cycle cost has a disposal fee as part of it.
7 Are you adding even to that, or--

8 LAIDLER: It's a mil per kilowatt hour.

9 DI BELLA: Yes.

10 LAIDLER: What I'm thinking in my non-professional
11 estimate is it's probably--we're probably going to add four
12 or five mils per kilowatt hour to the busbar cost. But, if
13 you look at--well, I'm from Chicago, and COMED just increased
14 their rates by almost 40 percent, so I'm paying way more than
15 that right now. I don't think you'll even see that effect.

16 DI BELLA: Thank you.

17 GARRICK: Thank you, Jim.

18 LAIDLER: You're certainly welcome.

19 GARRICK: We've been looking forward to this
20 presentation.

21 All right, our next presentation is going to be
22 from Pat Brady, a senior scientist at Sandia National
23 Laboratory, and he's going to report on work to characterize
24 the near-field chemistry of the repository.

25 BRADY: Okay, we're going to shift gears a little bit

1 here from the chemistry of the waste that's going to go into
2 the repository, and instead, focus on chemistry of the water
3 that will move through the overlying rock and possibly come
4 into contact with that waste.

5 For the next 20 minutes, I'm going to describe for
6 you the near-field chemistry model. This is a rough outline.
7 Essentially what I'm going to do is build a hydrologic model,
8 graft onto that a rock-based geochemistry model, consider
9 what that implies for the chemistry of the fluids going into
10 the repository. And, then, I will say a few words about
11 validation of that model at the end.

12 All right, I'm going to talk about the top, the
13 near-field chemistry model, which is--this takes the water
14 from the PTn/TSw contact and moves it down into the drift.
15 Inside of the drift, a model called the physical and chemical
16 environment model, the P&CE model, takes the seepage
17 compositions, evaporates them, and then considers their
18 impact on the drip shield and the waste package. Corrosion
19 is considered in a different model.

20 Downstream further is the consideration of the
21 engineered barrier system, flow and transport. This is where
22 we take those fluid chemistries, put them in contact with
23 seepage that comes out of the waste package, and then
24 consider the overall controls on radionuclide mobility.

25 All right, the first and most important point to be

1 made here is that this is not work I did. This is largely
2 the work done by Charles Bryan and Kate Helean. Charles
3 Bryan is here with us today. I hope you will take the
4 opportunity to talk to him later on as we go and explore the
5 particular details of the model.

6 Now, conceptually, what we do to calculate what
7 hits the engineered stuff is we take water, move it from the
8 PTn/TSw contact, down into the drift as seepage. Now, once
9 the water gets there, we do an explicit consideration of the
10 ratios of, for example, calcium to alkalinity to make
11 predictions as to what type of phases are going to form,
12 whether it's calcite or something else. And, that's done by
13 classical geochemical divide approach.

14 Ultimately, after you get the water through this
15 cascade of chemical processes, you are able to make some
16 estimations as to the chemistry that finally hits the
17 engineered materials, and ultimately contacts the interior of
18 the breached waste form.

19 I'm not going to talk about this section today,
20 because I believe you all have all heard about this.
21 Instead, I'm going to focus on this part up here, water
22 moving through the tuff, because this is where we've come up
23 with a new model, and this is what I'd like to focus on
24 today.

25 Now, conceptually, you can imagine--well,

1 conceptually, the processes that affect the chemistry of
2 water moving down towards the repository are fairly simple.
3 The water is going to stay in contact with something that's
4 basically two-thirds feldspar, one-third silica polymorphs,
5 quartzs, Cristobalite, Opal CT, what have you, and a very
6 small amount of clays and zeolites, plus, there's some
7 calcite.

8 Now, that means that, in essence, we've got a
9 reaction going like this. As you go down from the contact
10 down towards the repository, the feldspar goes down, as the
11 primary minerals alter to more stable secondary phases, clays
12 and zeolites. And, the process, you would expect that the
13 components of the feldspars would accumulate in solution,
14 potassium, sodium, some silica, such that the TDS would go up
15 a little bit. There's some--this is kind of an over
16 simplification, the silica, those values are also going to be
17 controlled by interaction with the other silica phases as we
18 go down.

19 What complicates this otherwise straightforward
20 picture is the temperature gradient being changed by the
21 thermal pulse. If we take our present day temperature
22 gradient, and then you put a thermal pulse through it, you'd
23 make it less steep as you move out this direction, and then
24 you'd come back down. And, that's going to affect all these
25 reactions through here. Now, what I'm going to do is show

1 you how we do a hydrological geochemical model to anticipate
2 what that net impact is.

3 All right, I'm going to do this in two steps.
4 First of all, I'll describe the hydrologic model in this
5 viewgraph, and then the geochemical model.

6 What we do is we take the repository footprint
7 here. We've got seven drifts. Along each of those drifts,
8 we have 16 locations. Okay, so, we've got 7 times 16 times
9 41 time slices. Essentially what we want to do is calculate
10 what the temperature as a function of distance up above the
11 repository is, at ultimately 4592 spots. Okay, this comes to
12 us from the multi-scale model. This doesn't come out of our
13 calculations here. But, the first input we have is a
14 temperature history as you go through the repository.

15 And, up there on the northeast corner of the slide,
16 you can see one of the--conceptually, what it looks like.
17 You have--there's ambient. As time goes on, and the thermal
18 pulse moves through the rock, the temperature gradient moves
19 out, and it comes back through.

20 In your handouts, you will see these things are
21 contoured. That's the units of time there.

22 All right, so, the first step of the modeling is we
23 take a thermal model generated on the southwest portion of
24 this graph, and we graft onto that a percolation flux. Okay,
25 since we can predict the temperatures from the multi-scale

1 model as you go up, now we want to take water and move it
2 down through that thermal gradient. And, that comes
3 independently of this model from the percolation
4 calculations.

5 Now, we have--I'm going to show you in just a
6 moment how that varies over time. But, what you will see in
7 the southeast corner of this slide is where we have combined
8 the water moving through the thermal gradient, and we've
9 tried to predict the arrival times of water as it comes into
10 the drift. And, we've got it for five different percolation
11 fluxes here. Each of these unattached points are the water
12 arrival times. The lines here that are anchored by a data
13 point right there, those were calculated independently with
14 FEM, a finite element heat and mass transfer code that does
15 particle tracking and interaction between the fractures and
16 the matrix.

17 The first point to be made here is that there is a
18 pretty good agreement between what we have done, which is
19 essentially a plug flow approximation. We have just taken
20 percolation fluxes that are independently calculated, and
21 moved them through a thermal gradient.

22 Now, the plug flow approximation hits pretty close
23 to what you'd get with the more elaborate FEHM calculation.
24 Typically, the disagreement is less than 10 percent. The
25 disagreement is larger at the higher fluxes.

1 So, to cap the hydrologic model, we take the plug
2 flow results, these unattached numbers, we multiply them by
3 that deviation, just to bring us up to what the more
4 elaborate FEHM model would predict, and we basically have our
5 fluid travel times, more importantly, the fluid residence
6 times as they go down through the rock.

7 Now, the other application of the close agreement
8 there that I would like to emphasize is the fact that those
9 numbers are fairly close suggests that equilibration or
10 exchange between the matrix and the fractures happens more
11 rapidly than the actual vertical movement.

12 Okay, so, up to this point, there has been no
13 chemistry. This is when the chemistry starts to come in.
14 The primary reaction that you see in the rock, like I said in
15 Slide Number 3, is feldspar altering to clays plus zeolites.
16 We need to come up with a rate of feldspar degradation. And,
17 the way we do it is by looking at the rock itself. These are
18 mineralogic abundances in the units we care about.
19 Basically, there's roughly, I'll call it 2 percent smectite
20 plus zeolite, about half a percent, or a little bit less of
21 sorptive zeolite. We can add those up and you have about 2.5
22 grams per kilogram of rock is alteration product.

23 If we assume that aluminum is conserved, as you
24 turn feldspar into clays plus zeolites, we can use the
25 alteration product abundances to back calculate out the

1 feldspar, the amount of feldspar that's been dissolved. In
2 other words, we just assume local conservation of alumina.
3 That is the least soluble component of the feldspar.

4 And, so, we can calculate how many mols that 2.5
5 grams equals. And, so, we end up with the amount of feldspar
6 in the rock today. To turn that into a rate, we need to
7 spread that over the age of the rock. The TSw is 12.8
8 million years old. So, we could take the modal abundance,
9 divide it by that number, and come up with a maximum feldspar
10 degradation rate. And, that's not the value we use, though.
11 5.9, 4 times 10 to the minus 9 mols per kilogram per year.
12 That's the highest value we could expect. That would be--
13 that is a rate calculated if you assume that all this stuff
14 came out of the mountain, and it sat there at ambient
15 temperatures for 12.8 million years, which really isn't the
16 case.

17 The temperatures quickly decayed, as the unit
18 cooled, and then as the overlying rocks were erupted and
19 adjacent bodies were erupted, temperatures went back up, and
20 then came back down again.

21 So, the first thing we have to compensate for is
22 the fact that temperatures over the lifetime of the rock had
23 not always been ambient. So, when you do that, we end up
24 with a rate that's about a third of that. It's 1.7 times 10
25 to the minus 9 mols per kilogram of rock per year.

1 Now, we're pushing water down through a column.
2 This temperature is changing. We've established what the
3 base rate of feldspar degradation is, the rate at which the
4 components of feldspar have turned into solution. To be able
5 to anticipate the rate at which the feldspar dissolves into
6 those packets of water over time, we have to be able to
7 predict the feldspar degradation rate of temperature. And,
8 for that, we need an activation energy.

9 The activation energy, this is taken from
10 literature values, 49 kilojoules per mol, it's a little bit
11 over 10 kJ per mol. Although there's an enormous debate
12 about what silicate weathering rates are in nature, and
13 whether you measure them in the field or in the lab, you tend
14 to see--the activation energies you measure in the lab tend
15 to be fairly close to the ones you measure in the field.

16 So, what we do is we take that number and we use it
17 to scale our feldspar degradation rates up temperature. And,
18 what you see here is, and I've done it just an Excel
19 spreadsheet calculation, from 25 degrees up to 95 degrees,
20 you get an increase in the silicate of a feldspar degradation
21 rate of about a factor of 50 or 60.

22 All right, so, at this point, we have a hydrologic
23 model that moves fluids through a thermal gradient. We have
24 a baseline feldspar degradation rate that we know is a
25 function of temperature. We map that onto the first one, and

1 we can calculate at a point in the repository, at a point in
2 time, how much feldspar--how much potential feldspar, how
3 much dissolved feldspar is going to be seeping into the
4 drift.

5 Now, that unwieldy description, the amount of
6 feldspar seeking into the drift, is called the WRIP, the
7 water/rock interaction parameter. That's nothing more than
8 the number of mols of feldspar dissolved into a liter of
9 water. And, I've plotted here the WRIP value for one of the
10 median thermal paths as a function of time. All of the lines
11 are rated out because these are cumulative distribution
12 functions over the percolation fluxes.

13 If you take the central one, that would correspond
14 to 7 millimeters per year for the present, 11 for the
15 Monsoon, 11 for the glacial, and 21 for the post-10K.

16 Now, what this graph shows you is if you were
17 sitting in a drift for a million years, and you were
18 collecting water that seeped through that point, and you
19 measured how much feldspar dissolved along the way, those are
20 the numbers you'd see starting from zero, up to the -5,
21 peaking, and going.

22 Note that the--if you look, go down on these
23 cumulative distribution functions, you're going to be
24 increasing flux. So, what that says is that the faster the
25 percolation, the lower the mean residence time of the water

1 in contact, which means less feldspar degradation.

2 Okay. I've only got three or four slides left. At
3 this point, I've talked about water moving through the rock,
4 and rock being dissolved into that water. Neither of those
5 are used. TSPA needs more than that for the calculation of
6 the corrosion potential, for the assessment of is this
7 chemistry good or bad for corrosion. We need the aqueous
8 concentrations of the components in the rock. And, the way
9 that is done is shown up here. Now that we know about--since
10 we can estimate the amount of feldspar for every packet, now
11 we have to back calculate what that does to the water
12 chemistry.

13 And, the way that is done is with EQ3, EQ6 reaction
14 path code, which makes our lives somewhat easier. But, I
15 want to show you pretty much what the boundary conditions of
16 that calculation are. Most notably, we assume equilibrium
17 with calcite. That's the additional constraint on our water
18 chemistries. There is between .01 to .41 percent calcite in
19 the TSw. If you look at geothermal systems throughout the
20 world, typically, you see that most waters stay close to
21 equilibrium with calcite. So, this helps us when we balance
22 out where the elements go.

23 We use a similar boundary condition for silica.
24 Again, there's no shortage of silica, and silica containing
25 phases in the TSw. It's not immediately obvious which ones

1 control the solubility of the amount of silica in solution,
2 which is kind of what we see in geothermal zones. If you
3 look in the southwest corner, you will see with temperature,
4 the measured silica concentration is 10 to track in between
5 some of the silica polymorphs.

6 We use amorphous silica to set the aqueous silica
7 concentrations, which probably leads to an over estimate in
8 silica. When it comes to the reactions that control the pH,
9 the alkalinity, things that we really need for the corrosion
10 assessment, silica is not that important. So, this little
11 bit of uncertainty isn't a huge one.

12 All right, if we take our feldspar degradation
13 reaction to the equilibria with silica and calcite, and what
14 you see with increasing alteration are the following three
15 reactions. Starting off, the feldspar dissolves and you end
16 up growing stellarite, a calcium bearing zeolite. You
17 increase this calcite--silica is also used up, as is calcite,
18 and you produce--basically, you shift from consuming just
19 silica to silica plus calcite. As the water/rock interaction
20 parameter goes up, the amount of silicate weathering
21 increases. You start moving into the formation of clay
22 minerals, which is shown down at the bottom.

23 Next slide? Okay, at this point, this bounds our
24 fluid chemistries. One of the things that is not bounded by
25 this calculation and which is indeed very difficult to bound

1 is what are the partial pressures of CO₂ and equilibrium with
2 the fluid as they come into the drift. This is very
3 important because it tells us something about the inherent
4 buffer capacity of the fluids.

5 We have two bounding approaches to give us a
6 minimum and a maximum in-drift CO₂ level. If you look down
7 at the bottom, you will see the minimum in-drift CO₂ level
8 calculation. And, that's estimated if we take the seepage
9 water that we've calculated independently. We bring it into
10 the drift, and evaporate it. The H₂O vapor takes up a large
11 volume of the drift, leaving a little bit of space left over,
12 at the same time contributing CO₂ to that. So, that part
13 started off with 10 to the minus 3 atmosphere, CO₂ the
14 ambient. It picked up a little bit more as the water was
15 evaporated. That's the minimum CO₂ calculation.

16 The maximum CO₂ is just assume that the seepage as
17 it moved down through the TSw, it carried with it its own
18 PCO₂. PCO₂'s go down because of the silicate degradation,
19 they go up when clay minerals form. We keep track of both of
20 those, and the maximum CO₂ level is just that, that CO₂.

21 Next slide, please? Okay, so, in a nutshell,
22 you've seen what the model is. You've got a plug flow,
23 hydrologic calculation, followed by a field based feldspar
24 weathering rate, one put on top of the other, for two
25 different bounding CO₂ scenarios.

1 How do we validate the model? We validate it in
2 three different ways. This is the--well, two of them are
3 done, a third is in the process of being done. The first one
4 was to compare the feldspar degradation rate with
5 independently derived degradation rates inside the TSw.
6 Second approach is to examine the evolution of the PTn
7 waters. And, the third is to look at the drift scale test
8 results. I'm only going to talk about the first one here.

9 All right, what's shown are data from Bryan
10 Marshall at the USGS. What they have measured are pore water
11 compositions and rock compositions, strontium 87/86 ratios in
12 the rock and in the pore fluids. The strontium comes from
13 the feldspars. These are just the pore fluids. Imagine, if
14 you will, a fairly vertical--imagine if you pushed this whole
15 thing out to here, and you had a vertical line. That would
16 be the strontium 87/86 in the rock.

17 What this means is that as you go down in depth,
18 what happens is the slanting of the pore water trajectory, if
19 that's the rock and that's the pore water--I'm exaggerating,
20 but as my elbows get closer, that's reflecting the strontium
21 87/86 exchange with the rock that occurs in dissolution. The
22 steeper the--let's see--the pointier my elbows are, the
23 higher the radius. If you had a vertical--if you had a semi-
24 vertical one, that would suggest that the approach of the
25 pore water to the rock is very, very slow. And, the rates

1 would be, that you'd calculate for mass balance, would be
2 low.

3 Doing it like this would give you a fast rate.
4 That gives you an idea of how the calculation is done. It's
5 a lot more complicated than that, but you get the gist of it.

6 All right, so, the end result is we've got--we can
7 calculate an 87/86 exchange rate that can be converted into a
8 feldspar degradation rate, an in situ feldspar degradation
9 rate, and we can compare it with ours.

10 This is the value we used. I want to point out the
11 units here. Molls per second per kilogram of tuff, and this
12 is logs, rates get faster this direction towards me, and they
13 get slower towards you all over there. All right, so that's
14 the value we get. The uncertainties that go into this are--
15 well, there are uncertainties in the model analysis that gave
16 us the amount of clays and zeolites. There's uncertainties
17 in the water/rock ratios. So, you get about an order of
18 magnitude spread around it.

19 From these two boreholes, you calculate rates that
20 are about a factor of five faster. The uncertainty here
21 comes from--Charles, you're going to have to correct me if I
22 say this wrong--a parametric uncertainty of--there is a
23 parametric uncertainty in the percolation fluxes. The extra
24 uncertainty here, that reflects uncertainty in the leaching
25 of the strontium out of the rock.

1 Okay, the upshot, though, is that--we're fairly
2 close--a factor of five in the silicate degradation world?
3 That's quite notable.

4 All right, next slide? Okay, so, let me summarize
5 all this. The near-field chemistry model, it's what we use
6 to build the major element composition of the seepage fluids.
7 That, we pulled out of independent thermal field calculations
8 and percolation fluxes. We've grafted onto that a rock based
9 feldspar degradation rate that gives us seepage chemistries.
10 And, it looks like we got it right the first time out of the
11 gate, because so far, our validation efforts, in particular
12 the strontium 87/86 comparison, looks like we're in the
13 ballpark for both the reaction mechanism and the rate.

14 Next time I talk, or Charles talks, we'll talk
15 about the drift scale test in the PTn waters. I think that's
16 my last slide.

17 GARRICK: Ron?

18 LATANISION: What have you learned in this study that
19 has implications related to corrosion of the waste package?
20 You made a comment at the outset that this--I think you made
21 a comment at the outset that this is part of the motivation.

22 BRADY: Well, that was the goal, was the come up with a
23 better prediction for what those values were. The values are
24 being written up right now in the AMRs, so I don't know if
25 we're at liberty to say. Charles?

1 BRYAN: Well, of course, also within that AMR, we talk
2 about our new pore water selection scheme, which in general
3 yields waters that have higher nitrate to chloride ratios
4 than the previous several year old results did. I think we
5 have a better understanding now of the changes that are
6 occurring as the water percolates downward, especially with
7 respect to pH and the PCO₂. So, I think what we have
8 developed is a little better understanding of the general
9 geochemical processes that are going to affect the water over
10 time.

11 LATANISION: Just a clarifying comment. This had no
12 functional affect on the chloride/nitrate ratio, ionic ratio,
13 which is clearly of importance in terms of--

14 BRYAN: I can address that as well. We are actually
15 sampling chloride and nitrate ratios differently this time
16 around. What we are doing is we have divided the waters, the
17 pore waters, into four different groups, and now in TSPA,
18 we're actually sampling the four different groups. Okay.
19 We're sampling the chloride to nitrate ratio. It's measured
20 over all of the waters within those groups. So, we're no
21 longer just using a single chloride to nitrate ratio to
22 represent each group. So, I think we're doing a better job
23 of incorporating the uncertainty in chloride to nitrate ratio
24 that we believe exists.

25 BRADY: And, Charles referred to the four different

1 water types. The chloride/nitrate ratios are right here.
2 That may be part of the answer to your question.

3 Let me make a point here. The calculation I did--
4 excuse me--that Charles did that I described, that's only
5 going to tell you the silica--well, it's going to tell you
6 the major element compositions as they come into the drift.
7 Chloride and nitrate don't show up in any of those reactions
8 for feldspar dissolution. The water comes in and they're
9 just along for the ride. So, when you do this calculation,
10 it's two parts. One, to get the major element chemistry as
11 it moves and goes down the pipe. But, two, is to get the
12 right waters in the first place because that's where the
13 chloride to nitrate ratios are going to come from. And, this
14 was a summary slide that Charles hit the higher points of. I
15 believe it's in your packets.

16 GARRICK: Thure?

17 CERLING: Cerling, Board.

18 I think you have answered some of my unasked
19 questions, but it had to do with your modeling was to
20 basically understand feldspar chemical dissolution. But,
21 that doesn't give you any anions except bicarbonate, and so
22 it seems that what you're trying to describe is just
23 confirmation that the water chemistry you're observing is
24 compatible with the fluxes. And, so, you're trying to get an
25 estimate of flux; is that right?

1 BRADY: That's part of it.

2 CERLING: Which will then give you other anions that are
3 so important to Ron's question?

4 BRADY: Yes, those anions start off, they go into the
5 calculation at the beginning, and there are no reactions that
6 happen to them until you get inside the drift. But, pH and
7 the buffer capacity of the fluids that hit the waste package,
8 those are also important, and those do come from, as you
9 said, bicarbonate. Those are one of the outputs from the
10 feldspar weathering reaction. So, it's a little bit more of
11 the feldspar. It carries a little bit more of a load.

12 BRYAN: With respect to the changes, the feldspar
13 alteration does change what the waters eventually evolve to
14 when we evaporate them within the drift. Even though we're
15 assuming equilibrium with calcite, calcium concentrations in
16 the water do go down, mainly because bicarbonate goes up.
17 So, for some of those four pore waters, for instance, if you
18 take them initially and evaporate them, you will end up with
19 a calcium chloride brine. As you increase the amount of
20 feldspar dissolution, those waters evolve into sodium,
21 potassium carbonate, and chloride and nitrate brines. So,
22 the composition of the brine does change. The composition of
23 the final end product, once that brine drips onto the waste
24 package and evaporates, changes potentially with the amount
25 of feldspar dissolution.

1 GARRICK: George?

2 HORNBERGER: Pat, you mentioned that your calculations
3 depend on the hydrologic model going through a thermal
4 gradient. And, you mentioned that you're basing this on the
5 multi-scale model. We've seen some results recently that
6 would question how good that multi-scale model is, because
7 it's a two dimensional model and it doesn't take into account
8 vapor flow along the drift, and you may have significant
9 amounts of vapor entering the drift. How do you think that
10 would affect your calculations?

11 BRADY: I'm going to pitch that one to Charles.

12 BRYAN: First of all, I wanted to expand on what Pat
13 said. We're not using the multi-scale model results. We're
14 using results which are compatible with those, actually model
15 results that are consistent with their approach used in the
16 condensation and convection MR. Basically, we're using that
17 model just for generating the thermal field around the drift,
18 which is a conduction only thermal field.

19 Within the drift, we're sampling the upper and
20 lower bounds for PCO₂, and then in the TSPA model, we'll
21 sample between those two. One of the primary reasons we're
22 sampling a range is because we don't know the degree to
23 which--how well we're modeling water vapor within the drift
24 or how much oxygen will diffuse into the drift, or CO₂ will
25 diffuse into the drift at the ends. Because we don't have a

1 good three dimensional model, we think we're bounding the CO2
2 range, but we're not trying to predict a single value to
3 represent that range.

4 GARRICK: Bill?

5 MURPHY: Bill Murphy, Board.

6 Pat, as you know, these are problems that I've
7 thought a lot about and am quite interested in, and, so, I'm
8 curious to see your results, and I'm not sure quite where to
9 start with my comments. But, I think I'll summarize some of
10 the observations of the ambient system. Well, first of all,
11 I commend you for trying to characterize the ambient system.
12 I think that's the key to developing a water chemistry model
13 that could potentially be extrapolated to characterize the
14 thermal conditions. And, so, looking at the feldspar rates
15 under ambient conditions, and considering the reactions, I
16 think that's an appropriate first step. Unless you can model
17 the ambient system, I don't think you have a chance of
18 modeling the perturbed system, frankly.

19 And, so, perhaps I'm anticipating some of your
20 future work on PTn waters, but one of the characteristics of
21 the unsaturated zone groundwater chemistries reported by Al
22 Yang, and others, is that there's really a very wide
23 diversity in cation ratios. The ionic strengths are all
24 about the same, but there's really a big diversity in these,
25 and that's a peculiarity that I've wondered about a lot.

1 But, we saw more or less the same variations in the seepage
2 that came into the south ramp in the winter of 2007, once
3 again quite a big range of cation ratios.

4 And, that's led me to think that a lot of the
5 unsaturated zone water chemistry is controlled very early in
6 the soil zone, or in the vitric part of the Tiva Canyon, and
7 if you look at the mountain, the vulnerable phase among the
8 rock materials is the glass. It's the volcanic glass that's
9 been altered to zeolites a depth, and even higher up in the
10 mountain. I think glass dissolution has a predominant
11 control on unsaturated zone groundwater chemistry, even
12 though it exists--it doesn't exist in the Topopah Spring,
13 particularly, but up above that.

14 Also, a lot of the components that enter the
15 unsaturated zone are introduced by reactions in the soil
16 zone, dissolution of dust and coleche, and the waters that
17 infiltrate are very highly conditioned by that infiltration.
18 And, I think the south ramp data in relation to the
19 unsaturated zone water chemistry tends to bear that notion
20 out. So, that's one observation.

21 I'd like to see your model compared to unsaturated
22 zone chemistry and to rationalize the sorts of--

23 BRADY: Let me point out that that was foremost in our
24 minds when we set it up, because you're right, there's glass
25 up above the TSw, and glass is far more reactive, and you

1 would expect that the fingerprint of the fluids that
2 ultimately make it through the TSw are going to be--they are
3 going to reflect that interaction. That's why when you see
4 that backup slide, the four different waters, those are all
5 TSw waters. They aren't PTn waters. In other words, they
6 implicitly carry that fingerprint of glass interaction with
7 them.

8 So, in a way, that was our attempt, to make certain
9 that the glass got into the water before we sent it to the
10 TSw where there wasn't glass. But, I agree with you about
11 glass being the important thing.

12 MURPHY: Somewhat along those same lines, in making the
13 effort to calibrate your model to the observations of the
14 system, there's some smectite and I thought it was a very
15 nice piece of work to estimate average alteration rates on
16 the basis of the smectite and zeolite, and I think you
17 produced a realistic looking number.

18 The predominant alteration phases, though, are
19 mordenite, clinoptilolite, and I'm aware of the challenges
20 and characterizing the properties of those for this kind of
21 modeling, but I think it would be nice to see the actual
22 phases that occur in the mountain represented in the model,
23 as opposed perhaps to stellerite.

24 BRADY: Obviously, the thermodynamic properties of those
25 minerals become very murky, but, yeah, I agree with you.

1 MURPHY: And, one final observation is I think you
2 probably answered this in addressing why you decreased your
3 feldspar alteration rate due to natural thermal
4 perturbations. I think that petrographers at Los Alamos
5 always say that a great deal of what alteration you see at
6 Yucca Mountain occurred very shortly after the volcanic rocks
7 were erupted during the cooling of the rocks, and, so, a
8 calculation of an average rate may be fine as an average
9 rate. But, there could be--a lot of it could have occurred
10 early.

11 In contrast, we have the very nice rate of
12 precipitation data from the calcite or opal, the opal dating
13 from Jim Pace's rates of reaction.

14 BRADY: These are all good points, Bill, and you'd think
15 the limiting case would be that it all happened right after
16 the rock was formed, in which case the rate is zero. But,
17 you can correct me as soon as I--but the fact that there are
18 a couple of items that kind of gave us confidence that we
19 were in the ballpark. First of all, it's the strontium 87/86
20 comparison.

21 Also, when you convert those rates to a mols per
22 centimeter squared per second value, I want to say it's about
23 10 to the minus 20, which is sort of the low end that we've
24 measured on plagioclase in Hawaii and other places. And, so,
25 I think we're in the ballpark on it.

1 BRYAN: I agree with you. But, the actual range, if you
2 consider that all of the alteration that we observed occurred
3 in 12 million years under ambient conditions, that would give
4 you one rate. The other bounding rate would be to assume
5 that the rocks had been at elevated temperature, 100 degrees
6 C. for 12 million years, in which case, you could calculate a
7 rate from the observed alteration, which would be a 100
8 degrees C. rate.

9 Now, we know from the activation energy that the
10 100 degrees C. rate can only be a factor of 60 above the
11 ambient rate. So, what we actually have is, at most, a
12 factor of 60 uncertainty in the actual rate.

13 Now, when you consider that most of these rocks did
14 form, most of the alteration did form early, you can actually
15 significantly reduce that. And, that's what we did in the
16 model. So, we actually have on the order of a factor of five
17 or eight in uncertainty in the actual rate, let me include
18 that.

19 So, we have considered the fact that the majority
20 of the alteration did form at elevated temperature early in
21 the history of the rock. Early being actually about 3 to 4
22 million years. The rock remained hot for several million
23 years because apparently, according to Bryan Marshall's
24 modeling, intrusion at greater depth.

25 MURPHY: Okay, let me make one more point, please.

1 Prefacing this by saying I think it's important to
2 demonstrate that you can characterize and understand the
3 controls on the chemistry and the ambient system, the real
4 effect is how is this going to affect corrosion and waste
5 form dissolution and perhaps hydrologic characteristics of
6 the system in the long term. And, as was pointed out before,
7 what you really need, what's used in PA is ionic strength and
8 pH and CO₂, and those are related to this in some ways.

9 But, also, I'm curious about the effects of the
10 thermal period on the hydrologic characteristics of the near-
11 field. Can you calculate during the thermal excursion that
12 may last a thousand or a few thousand years, if the
13 permeabilities or porosities, will there be a silica cap
14 formed, will there be channelized flow developed because of
15 the precipitation and dissolution? Are there coupled
16 hydrochemical effects that you can draw from your model that
17 really lead us to evaluating how the emplaced wastes are
18 going to run to respond?

19 BRADY: We don't, but we could, to a point.
20 Specifically, we could calculate the masses of the feldspar
21 dissolved, minus other stuff formed. I think it's going to
22 be murky though when you spread that out in fractures, or
23 matrix porosity. So, we haven't gone down that path.

24 GARRICK: Okay, I think we are going to have to move on.
25 Thanks, Pat.

1 Our next speaker is Paul Reimus. Paul is a senior
2 scientist at Los Alamos National Laboratory and he will give
3 us a presentation on testing in the saturated zone. So, we
4 go from the near-field to the far-field.

5 REIMUS: Yes, we'll move about 300 meters down now.

6 Next slide? This is the outline of my presentation
7 today. First, I'll give an overview of the new Nye County
8 early warning drilling program wells that have been drilled
9 since 2005, basically since the last time we had an
10 informational meeting with the Board. These are the Phase V
11 wells in the Nye County program.

12 I will also provide an update of the
13 hydrostratigraphic framework model, which is the underpinning
14 for the saturated zone flow and transport models for Yucca
15 Mountain, and the site scale models. But, I will provide an
16 overview of the hydraulic and tracer test results at Nye
17 County Site 22. These tests were in progress a couple years
18 ago when there was last discussions with the Board on
19 saturated zone testing, and these are all now concluded
20 tests, and interpretations are being documented in an AMR.

21 And, finally, I'll discuss some innovative methods
22 of identifying flowing intervals and measuring ambient flow
23 velocities in Nye County wells that have been implemented in
24 the last year and a half, primarily through the Nye County
25 program. All these activities have benefited from or been a

1 direct result of close cooperation between the DOE-funded Nye
2 County early warning drilling program, the DOE-University of
3 Nevada Cooperative Agreement, and Yucca Mountain Project
4 scientists. And, I should also add OSTI, your science and
5 technology program scientists have been involved in this as
6 well, although there's certainly some overlap with the Yucca
7 Mountain folks there, particularly with this last bullet,
8 that was an OSTI inspired investigation.

9 Next slide? Okay, the blue circles on this map are
10 the locations of the four new Nye County early warning
11 drilling program wells in Phase V. There's actually a fifth
12 one over at Site 22 added to two other--or three other wells
13 that were there. The two wells south of Highway 95 here, 33-
14 P and 32-P, were drilled at the location of magnetic
15 anomalies, looking for volcanic explanations for the magnetic
16 anomalies. 13-P was drilled up to the northwest here. There
17 was an additional well also drilled at Site 24 that I'll talk
18 about in a little bit more detail.

19 The two circled locations here, Sites 24 and 22,
20 are the locations of tests that I'll be discussing in this
21 presentation.

22 Okay, first, I wanted to briefly go over changes to
23 the hydrostratigraphic framework models since 2004. In 2004,
24 the base case model was based on what was called the HFM-19,
25 which had built into it regional model efforts from the mid

1 to late Nineties. The 2006 base case model basically
2 incorporates all the observations in Nye County wells through
3 Phase IV. So, the updates, particularly the southern portion
4 of the site model area are largely due to the Nye County
5 drilling program.

6 And, then, of course, the boundaries and other
7 changes to the model have come about primarily as a result of
8 changes to the USGS regional model that occurred in 2002 and
9 2004.

10 I do want to point out that the major change here,
11 this is a slice at the water table, basically, over the site
12 model domain in both cases. The big change in terms of
13 performance of the saturated zone is that whereas in the 2004
14 model, we had fractured volcanics, all these darker colors
15 are fractured volcanic units. The units are shown over here
16 in this legend, and that legend is expanded and explained a
17 little bit better in the very last slide of your handout.
18 But, fractured volcanics basically were in the flow pathway,
19 almost all the way up to Highway 95. Now, the Nye County
20 drilling program has identified at least a third or so of the
21 flow path from the repository horizon to the 18 kilometer
22 boundary, is in saturated alluvium at the water table.

23 Let's see, I also wanted to point out these black
24 lines here are particle tracks derived from the northern part
25 of the repository, the southern part of the repository and

1 the central part of the repository in the current site scale
2 transport model. And, if I may, I just wanted to point out
3 that Site 22 where the tracer testing and hydraulic testing
4 that I'll talk about next, occurred, is located right along
5 the project flow pathway from the model, basically right
6 along Forty Mile Wash, just a few miles north of the 18
7 kilometer boundary.

8 Next viewgraph? Okay, the testing at Nye County
9 Site 22 involved both hydraulic tests and tracer tests. The
10 objectives were to test and validate conceptual models of
11 flow and radionuclide transport in the alluvium south of
12 Yucca Mountain, the saturated alluvium, and to obtain
13 estimates of all these various flow and transport parameters.

14 The site layout is shown here in the lower left,
15 22S is the large diameter production well. It had four
16 screened intervals in it, and 22PA, 22PB and 22PC are all
17 dual nested piezometers that are completed to different
18 intervals. There's about 650 feet, or so, of saturated
19 alluvium at this location. It's underlain by what's
20 classified as a non-welded volcanic breccia.

21 The hydraulic testing involved isolated zone
22 pumping of each of the four zones in 22S, looking at
23 responses and recoveries in the observation wells, the
24 piezometers, and the tracer testing was conducted in Zone 2
25 with 22S as a production well, and both 22PA and PC as tracer

1 injection wells and cross-hole tests.

2 Okay, this slide summarizes the methodology for
3 hydraulic test interpretation in the testing at 22S. The
4 observation wells drawdowns and recoveries were analyzed in
5 both the pumped interval as well as intervals above and below
6 the pumped zone. This was a little bit of a departure and an
7 improvement over previous modeling that had been done, which
8 had primarily focused on looking at responses in the pumped
9 zone.

10 Three-aquifer semi-analytical solution was used,
11 basically simultaneously solved for the drawdowns, or
12 simultaneously fitted the drawdowns and recoveries in three
13 intervals, the pumped interval and intervals above and below,
14 and you can see here an example of fits to the drawdown and
15 recovery data for the test in Zone 2.

16 And, what came out of that was obtaining estimates
17 of storativity and transmissivity in either three aquifers or
18 in the case of the upper or lower zone, it would be two
19 aquifers, the two-aquifer analytical solution was used. And,
20 also, vertical hydraulic communication parameter was
21 estimated in the intervening layers between the zones.

22 Next viewgraph? This viewgraph summarizes the flow
23 conceptualization and the parameter estimates that came out
24 of the hydraulic testing. A lot of numbers here. I'll try
25 to summarize really quickly by saying that the upper two

1 intervals essentially behaved as a combined aquifer that had
2 weak vertical anisotropy that was a hydraulic conductivity, a
3 horizontal hydraulic conductivity to vertical conductivity
4 ratio of about two to one, based on the test interpretations.

5 The horizontal hydraulic conductivity composite
6 over this upper portion of the alluvium is about 10 meters
7 per day. The next interval down definitely exhibited more
8 confined behavior. There appears to be some sort of a
9 confining layer or semi-confining layer between Zones 2 and
10 3. And, then, again, there's confinement between the bottom
11 of the alluvium, or the lower zone in the alluvium, and the
12 volcanic breccia.

13 If you look at the entire alluvium as a composite
14 here, you get about a five meter per day horizontal hydraulic
15 conductivity. Again, ten meters a day in the upper portion.
16 And, just for comparison, the calibrated site scale flow
17 model has a horizontal hydraulic conductivity at this
18 location of about 17 meters per day. So, within a factor of
19 two, or so, of the estimate obtained from hydraulic testing,
20 and that's assuming a 10 to 1 vertical anisotropy ratio,
21 which we do observe deep, but not shallow at this location.

22 HORNBERGER: Can you give me an idea of how thick the
23 what you have marked as aquifers are?

24 REIMUS: The total saturated thickness of the alluvium
25 here is 650 feet. But, you're talking about the individual

1 screens? The screens are on the order of 100 feet, or so, 60
2 to 100 feet. The gravel packs for the screens are maybe
3 slightly longer.

4 Okay, moving on into the tracer test results.
5 There were two single well tracer tests conducted in 22S.
6 They both involved the use of two conservative tracers with
7 different diffusion coefficients, a halite fluorinated
8 benzoate. Iodide was the halite in both cases. The
9 difference between the two tests was the amount of time that
10 the tracers were allowed to sit in the formation in the
11 aquifer before being pumped back out. On the left here, we
12 see the tracer responses, the break-through curves,
13 normalized to injection mass, as a function of time, after
14 three days of sitting in the aquifer. On the right-hand
15 side, we have the responses after 30 days of sitting in the
16 aquifer, and you will see there is a difference.

17 And, what was interesting here is that the iodide
18 the tracer with the larger diffusion coefficient actually had
19 the higher normalized concentration in both of these tests.
20 The fact that there is a difference in the concentrations of
21 the tracers does indicate some dual porosity character or
22 diffusion into stagnant water. However, the fact that the
23 iodide is coming through at a higher concentration also
24 suggested that diffusion distance, or diffusion time scale,
25 was fairly short. And, we would expect this dual porosity

1 behavior that's observed at these sort of time scales to
2 convert over, or transfer over to more of a single porosity
3 behavior over a much longer time and distance scales.

4 Next viewgraph? Okay, the responses of the tracers
5 in the single well tests were also used to estimate specific
6 discharge in the alluvium at Site 22. That was done by
7 comparing and analyzing the responses of the same tracer and
8 the two different tests. You see iodide here, the responses
9 in each test shown with a linear scale on the left, with a
10 log scale on the right. The differences in the peak arrival
11 times, the mean arrival times, and the arrival times
12 associated with a high recovery, nominally about 97 percent,
13 as I recall, were used to estimate flow velocities, and, in
14 turn, estimate specific discharges. Those estimates are
15 shown right below here. You get slightly different values
16 depending on whether you analyze the peak or the mean or the
17 high recovery times.

18 Those values range from .5 to 5.4 meters a year.
19 You get the higher estimates with the high recovery time
20 difference. By comparison, single well testing at Site 19-D,
21 which is located a few kilometers further southwest along
22 Forty Mile Wash, range from 1.2 to 9.4 meters a year. And,
23 also, by comparison, other specific discharge estimates are
24 listed at the bottom here. The discharge estimated
25 independently, just using head data along Forty Mile Wash and

1 the hydraulic conductivity estimates from the hydraulic
2 testing in Zone 2 range from 3 to 12 meters a year, and the
3 estimate from the site-scale flow model is about 21 meters a
4 year.

5 Okay, there were two cross-hole tracer tests that
6 were conducted at the site. I'm showing here the results in
7 terms of normalized concentrations of tracers as a function
8 of time. From the first test, these are all normalized to
9 injection mass, so we can compare them fairly, the first
10 thing I want to point out is the red curve is the responsive
11 of 2,6 difluoral benzoate from 22-PC. It was the only tracer
12 injected into 22-PC, which was located almost due east of the
13 production well. The other tracers all came from 22-PA
14 located the same distance away from the production well, but
15 due north. Clearly, there's a faster response coming from
16 the north direction as opposed to the east direction. We see
17 clear evidence of flow anisotropy from those responses, or
18 from the comparison of those responses.

19 We also see that there is a separation between the
20 2,4,5 trifloural benzoate in this case, and bromide as a
21 halide. Again, that is indicative of dual porosity behavior.
22 There is some uncertainty in the normalized concentrations of
23 not only the bromide, but the trifloural benzoate as well,
24 primarily because there were some discrepancies between
25 masses that were measured to be injected, and concentrations

1 of the injection concentrations. But, even taking into
2 account that uncertainty, we basically see what appear to be
3 dual porosity effects with either the bromide being
4 significantly lower in concentration than the benzoate,
5 indicative of diffusion to stagnant porosity, or we see a
6 slightly delayed peak, which is also indicative of diffusion
7 to stagnant porosity, just a little bit less stagnant
8 porosity, or shorter time and distance scales.

9 The lithium response, lithium was used as a
10 reactive tracer here, injected with all the other tracers.
11 It's very attenuated in concentration relative to the
12 conservative tracers. However, it does peak early. This
13 behavior is also consistent with a dual porosity system.
14 And, finally, carboxide modified latex microspheres were used
15 as colloid surrogate tracers in this test, and you can see
16 their response is quite low. But, nevertheless, there was a
17 recovery from which colloid transport parameters could be
18 estimated.

19 Next? Okay, this just shows a quick summary of the
20 interpretive aspects of this first cross-hole tracer test.
21 By taking the two cases, the two end member cases of a
22 minimum difference between the floual benzoate and bromide
23 break-through curves and the maximum difference between the
24 normalized break-through curves, there were two sets of
25 interpretations done. Both interpretations involve invoking

1 three flow pathways to fit the response curve. That was
2 driven in part by the fact that it was just not possible to
3 get a good fit with the single advective dispersive pathway.
4 But, it was also driven in part by the fact that just looking
5 at the derivative of the concentrations of either the bromide
6 or the trifloural benzoate as a function of time, exhibited
7 multiple peaks, which is suggestive of multiple tracer
8 arrivals. So, there were a few different lines of evidence
9 to suggest multiple pathway behavior. And, at least two of
10 the pathways involved invoking dual porosity character of the
11 system. And, in the case of the maximum difference between
12 the benzoate and the bromide, all three pathways had dual
13 porosity character.

14 Next viewgraph? Okay, the second cross-hole tracer
15 test is shown here, the responses of the tracers. In this
16 test, only iodide and Perrhenate were used as tracers. They
17 were both injected at 22-PA, the same interval that the other
18 tracers were injected into, with the exception of that 26-
19 DFBA. Perrhenate was used in this case as a surrogate for
20 pertechnetate, as was mentioned this morning, pertechnetate
21 is expected to be the form that technetium 99 will be in in
22 the saturated zone, at least it will be under oxidizing
23 conditions, and Perrhenate is just below pertechnetate in the
24 periodic table, so it serves as a very good surrogate, has
25 similar redox behavior and general chemical behavior. So, it

1 was used as a surrogate for pertechnetate.

2 And, we see the two responses of the tracers. A
3 few things to note here, first of all, when you compare these
4 responses to the bromide response in the first test, you see
5 there is quite a difference, a much higher normalized peak
6 concentration and the shorter tail. Basically, what appeared
7 to be the case here is that the third flow pathway that was
8 activated in the first test was not activated in this test.

9 The second point of interest is the Perrhenate has
10 a lower peak concentration and a longer tail, ever so slight,
11 but nevertheless apparent, than the iodide, which does
12 suggest that there's some sort of retardation behavior of the
13 Perrhenate in the system. This might be explained by
14 diffusion, but the problem with that explanation is
15 literature values of Perrhenate diffusivities are lower than
16 iodide, so in that case, if it was all diffusion, you would
17 expect the Perrhenate to have a higher concentration than the
18 iodide. In this case, it doesn't, which does tend to point
19 toward some sort of a retardation mechanism, or sorption
20 mechanism.

21 Next viewgraph? Okay, this table summarizes all
22 the parameter estimates obtained from tracer testing. I
23 won't go into all the details here. You can read them in
24 your handouts. Each of these has a lower bound and an upper
25 bound, and in a few cases, there's best estimates provided.

1 I did want to point out some of the comparisons of
2 these estimates with what's coming out of the saturated zone
3 flow and transport modeling in cases where you can make a
4 direct comparison. The best estimate of effective flow
5 porosity that we get out of the tracer test is about .12 in
6 the alluvium at this location. The flow porosity
7 distribution in saturated zone modeling assumes that there's
8 a normal distribution of flow porosities that are sampled,
9 and they have a mean of .18. So, the best estimate is a
10 little lower, but keep in mind also that the tracer tests
11 were interpreted assuming dual porosity behavior, but it was
12 dual porosity behavior with very short time and distance
13 scales of diffusion. So, if you accounted for the stagnant
14 porosity, as would be expected to be valid, over longer time
15 and distance scales, this estimate of flow porosity would go
16 up to around the 20 percent level.

17 Horizontal anisotropy, we get a ratio of north-
18 south to east-west horizontal flow anisotropy of about 3 to
19 1. There is no explicit horizontal flow anisotropy in the
20 flow and transport models, but there is a zone of enhanced
21 permeability along Forty Mile Wash in the direction that we
22 see the principal axis of conductivity. And, the conceptual
23 model, flow and transport models, is that the alluvium
24 behaves as a porous medium, and basically, these tracer tests
25 confirm that assumption, although we do see dual porosity

1 behavior, again, the time and distance scales of diffusion
2 are very short. So, over longer--the performance assessment
3 type time scales, that should be, a single porosity
4 assumption should be pretty reasonable.

5 Okay, going back to the outline, I'll hit on the
6 last topic, these innovative methods of measuring ambient
7 flow velocities and identifying flowing intervals in Nye
8 County wells.

9 Okay, a couple different methods were employed,
10 inspired by, as I said, the OSTI program. I really want to
11 give Barry Friefeld of Lawrence Berkeley Laboratory a lot of
12 credit here. He was the one that was the PI on this project
13 for OSTI, or science and technology, and worked closely with
14 the Nye County folks doing this.

15 The idea with flowing electrical conductivity
16 logging, if you're not familiar with it, I won't go into all
17 the details, but the basic idea is you pump water out of a
18 well, run it through a de-ionizing unit, and reinject it at
19 the bottom of the well. And essentially try to fill up the
20 well with de-ionized water by running this circulation, and
21 you look at the extent to which you can do that replacement
22 of de-ionized water as one indicator of where there's cross-
23 flow occurring in the well. And, then, you can also stop
24 this recirculation and watch the ambient formation water
25 replace the de-ionized water in the well over time. And, you

1 do this, of course, by logging up and down with the simple
2 conductivity probe, and looking at how the profiles develop
3 over time.

4 Next one? And, these are the results from 24-PB,
5 which was one of the new Phase V Nye County wells completed
6 just last year. This well was completed primarily in
7 fractured volcanics. The water table was right at the base
8 of the alluvium, right at the top of the Bullfrog Tuff that
9 occurred at this location. And, there's some Fran Tuff below
10 the Bullfrog Tuff.

11 What you see here are profiles over time and also
12 model fits to those profiles of conductivity as a function of
13 depth. The blue line at the bottom, or the blue dashed line
14 at the bottom of all these curves is basically the baseline
15 profile that was obtained when trying to recirculate de-
16 ionized water into the bottom of the well at eight gallons a
17 minute. So, this essentially was as good as the well could
18 be de-ionized at eight gallons a minute. And, you can see as
19 you're flowing up, there is this one interval here at about
20 230 meters where you essentially go from being somewhat de-
21 ionized to hardly de-ionized at all. So, there's clearly a
22 lot of inflow occurring at this location, a fairly high flow
23 zone is identified.

24 Then, the pump was shut off and these other
25 profiles that go upward in time reflect the flushing of the

1 de-ionized water out of the borehole by the formation water.
2 We can see there's an inflow zone here. It doesn't really
3 show up very well from the baseline profile, but you can see
4 it as you watch the de-ionized water get flushed out. Flow
5 rates are backed out of these model fit calculations that are
6 shown here, the solid lines, and they indicated a cross-flow
7 at this particular depth here, this 230 meter depth where
8 there is this high flow zone of about two gallons a minute
9 cross-flow in the well, which was very interesting to see.

10 Next viewgraph? Okay, the other technique that's
11 been employed is called distributed thermal sensor logging.
12 The idea here is you run a constant wattage heater down the
13 length, the entire length of a borehole, and then also equip
14 the borehole with a sensor array that can look at and record
15 temperatures as a function of time over the entire length of
16 the well. And, this, as opposed to the electrical
17 conductivity logging, which is done in an open hole, at least
18 in this case it was, this is more amenable to being done in a
19 shut-in borehole, where you eliminate the possibility of
20 upward and downward flow in the borehole, by either plugging
21 the borehole or packing it off.

22 The idea here is you heat up the well bore for two
23 or three days, and in all these cases that were done by Nye
24 County, it was a two day heating period, and you look at
25 specific locations in the well, where the temperature

1 increase tends to lag, and take those as indicators of cross-
2 flow, where the water that's flowing in the vicinity of the
3 well bore is carrying heat away. So, anyplace in the well
4 that the temperature increase looks like it's lagging, or
5 once you have heated the well and you look at the cooling,
6 where the cooling occurs fastest, would be indicators of flow
7 zones. And, the higher the flow, the more lag you would tend
8 to see.

9 Next one? These are profiles from this same well I
10 just talked about, 24-PB. This is now with a grouted-in
11 borehole as opposed to an open borehole a couple months later
12 than the flowing electrical conductivity logs. We see here
13 in the dark blue line, the baseline temperature profile
14 before any heating, and then we see a whole bunch of curves
15 here going up on the plot in the upper direction on the plot,
16 are going out in time during the heating period. And, it
17 goes up to 48 hours, I believe, and we see that in this high
18 flow zone, or what was identified as a high flow zone, with
19 the flowing electrical conductivity logging at about 230
20 meters is clearly lagging in heating up. So, that's another
21 indicator that this is a high flow zone. And, again, doing
22 some thermal calculations, it's a little bit less directive a
23 measurement, but by doing some thermal calculations, heat
24 transfer calculations basically, you can deduce a flow rate
25 in this interval.

1 Next one? Okay, Nye County has used this thermal
2 logging method in several wells, six or seven by now, I
3 believe. And, I did want to show the heating profiles in 22-
4 PB, which was one of the wells at Site 22, where the
5 hydraulic and tracer tests were conducted. It's kind of
6 interesting here, this is the baseline profile and here are
7 the heating profiles. It's interesting here that the lagging
8 seems to be a little greater in the temperature increase down
9 a little deeper than the zone that was tested for tracers.
10 So, that might indicate there's a little bit more flow there,
11 although it is always a little uncertain to interpret these
12 thermal logs. They're dependent not only on water flow, but
13 also on thermal conductivity, porosity of the rock, how
14 uniform the well bore completion is, things like that all
15 factor in. But, nevertheless, it's interesting to see that
16 there's a possible indication of higher flow, slightly lower
17 than the tracer test interval.

18 Okay, finally to summarize and conclude, the
19 hydrostratigraphic framework model updates that I showed were
20 based on both Nye County wells that have been put in in the
21 last ten years, and the regional flow model updates in 2006.
22 These have resulted in much greater predicted transport of
23 radionuclides through the alluvium in the performance
24 assessment model than in the saturated zone, in particular.
25 That's important because alluvium has a lot more surface area

1 for sorption, and also flow velocities tend to be a lot
2 lower. You have a higher effective flow porosities in the
3 alluvium. So, flow velocities tend to be slower.

4 The Site 22 hydraulic tests exhibits vertical
5 anisotropy shallow, and stronger vertical anisotropy deep.
6 Just comparing with the site scale flow model, we get a good
7 comparison for vertical anisotropy. When I say vertical
8 anisotropy, again, I mean ratio of horizontal hydraulic
9 conductivity to vertical hydraulic conductivity. You get
10 good agreement deep. We see more of the uniform, or more of
11 the homogeneous system shallow.

12 The composite horizontal hydraulic conductivity is
13 5 to 10 meters per day based on the hydraulic testing, versus
14 about 17 meters a day coming out of the site-scale calibrated
15 flow model.

16 The tracer tests indicate a dual porosity system
17 with short diffusion, distance and time scales, single
18 porosity system is assumed in site-scale saturated zone
19 transport model. However, these are not incompatible because
20 the dual porosity behavior again appears to have very short
21 diffusion scales that would definitely transition over to
22 single porosity behavior over longer time scales.

23 Effective flow porosity estimates from pressure
24 testing, .12 versus a mean of about .18 in the site-scale
25 transport model.

1 Specific discharge estimates ranged from .5 to 5.4
2 meters a year, versus about 20 meters a year in the site-
3 scale flow model. And, the flowing electrical conductivity
4 logging and the differential thermal perturbation logging do
5 indicate a high flow zone in the Bullfrog Tuff at Site 24.
6 And, they show promise for possibly being applied at other
7 locations as well.

8 And, I think that was my last viewgraph, except for
9 the legend for the hydrostratigraphic framework map.

10 GARRICK: All right, questions? Yes, Thure?

11 CERLING: Cerling, Board.

12 If you go to Slide 4, I was just curious because it
13 seems like there's some large changes between these, but not
14 until--getting that last slide where we can see it--the
15 legend, and really not knowing exactly what all these are,
16 are the differences between the two maps really significant
17 in terms of flow characteristics for most areas? I know you
18 pointed out one area where it's in alluvium, and it was in
19 volcanics. But, in some areas, 18 has been switched to a 19,
20 and so on. Are there really significant differences as that
21 appears to be?

22 REIMUS: Well, by far the most significant difference is
23 the increased amount of alluvium in the flow pathway. That's
24 really been recognized primarily as a result of the Nye
25 County early warning drilling program. I do, as backup

1 slides, have a few vertical cross-sections, and, you know,
2 those show some differences. There's some changes in the
3 number of units that are used in the different models and a
4 little bit of difference in the mapping of the units from the
5 old framework model to the new framework model. But, the
6 changes for the most part are not great with respect to
7 affecting performance, other than this increased amount of
8 alluvium.

9 There's certainly been a lot of increased
10 understanding of the structure out there, and that's been
11 incorporated into the later version of the hydrostratigraphic
12 framework model. So, if you're interested in seeing those
13 vertical cross-sections, we could pull those up, I think.

14 CERLING: Well, I guess more as a follow-on question,
15 what would you, if you had your wish list of what you could
16 do to improve certain areas, you know, where is improvement
17 needed? Because clearly, there was, you know, what you said
18 is one very big improvement. Are there other areas where you
19 could possibly improve things with an additional well placed
20 boreholes?

21 REIMUS: Well, certainly there are. There is still
22 what's called an alluvial, or alluvium uncertainty zone, I
23 guess is what it's called, but it's basically a zone of
24 uncertainty in this region here where the water table
25 transitions from volcanics to alluvium. This is the way it's

1 shown in the current hydrostratigraphic framework model, but
2 there's definitely an area that I'm roughly outlining here
3 with this laser pointer that is considered a somewhat
4 uncertain zone where that transition occurs.

5 There has been some discussion of trying to put a
6 well in over here. Unfortunately, that's on the western side
7 of Forty Mile Wash, but still on the Test Site, which is a
8 bit more of a logistical challenge for locating a well. But,
9 certainly some more information there would help. That's
10 probably the big thing.

11 I guess the other thing is there is, in the flow
12 model, this zone of enhanced permeability that I referred to
13 along Forty Mile Wash, which, you know, could probably be
14 characterized a little bit better if there were some more
15 wells. Right now, all the wells are pretty much right along
16 Forty Mile Wash, so you really can't look at a contrast
17 between properties of the alluvium in the wash zone versus a
18 little bit away from the wash zone.

19 GARRICK: Ron, and then George?

20 LATANISION: Latanision, Board.

21 If we could go to Slide 21? Physically, what does
22 dual porosity system mean? What does that mean, physically?

23 REIMUS: Okay. Yes, dual porosity system is a system
24 that has basically porosity that's flowing, and porosity
25 that's full of water, but not flowing. So, essentially, the

1 flowing porosity is your primary porosity for transporting
2 contaminants, radionuclides, and the stagnant porosity, or
3 what's often referred to as the secondary porosity, is
4 porosity that is in mass transfer communication by diffusion
5 with the primary porosity, but it's not actually flowing.
6 So, it's essentially acting as a storage porosity that
7 tracers or contaminants can diffuse into and sorb within
8 surfaces in that porosity, and then they have to diffuse back
9 out to a continued transporting along the pathway.

10 LATANISION: How would you distinguish such a site
11 physically? I mean, can you look at a particular type of
12 pore and determine that it's going to be essentially a trap
13 for water, and the other a mobile? I mean, what's the
14 distinction?

15 REIMUS: In alluvium system, you know, you certainly
16 have internal grain porosity that would be dead-end porosity
17 that should behave, you would expect to behave as this
18 secondary porosity that's not flowing. However, there's
19 varying degrees, I guess of dual porosity character. You
20 could think of, in a layered system, which these alluvial
21 deposits certainly are, you know, you can have layers of
22 coarser material that have a much higher hydraulic
23 conductivity than other layers that are like clay, much finer
24 materials. And, so, those layers may actually--those layers
25 of a clay, finer grained material, may have a slow enough

1 flow velocity through them that they essentially act as a
2 secondary porosity that behaves more as a storage porosity
3 than part of the flowing porosity.

4 LATANISION: That helps. If we could go to Slide 20
5 next for just one moment? Your first bullet does not sound
6 like particularly good news. Am I reading it correctly?

7 REIMUS: What doesn't sound like good news?

8 LATANISION: Well, I mean, greater predicted transport;
9 is that a characteristic that gives you comfort or gives you
10 concern?

11 REIMUS: Well, I'm sorry, when I say greater, I'm
12 talking about a greater portion of the flow pathway is in
13 alluvium. Don't take that to mean that there's faster
14 transport through the whole system, but, no, it's poorly
15 worded, I guess. It is intended to convey the idea that the
16 new hydrostratigraphic framework model has a much greater
17 portion of alluvium in the flow pathways.

18 LATANISION: One of the hazards of the English language,
19 isn't it? Okay, good. Thank you.

20 HORNBERGER: As Thure said, you could have said it as
21 less transport in the tuffs.

22 REIMUS: Yes.

23 HORNBERGER: The FEC logging, was that done only in the
24 Bullfrog? You didn't log any of the alluvial wells?

25 REIMUS: Well, the water table was at the base of the

1 alluvium, basically right at the top of the Bullfrog in that
2 well.

3 HORNBERGER: But, that's the only well that was logged
4 that way?

5 REIMUS: No, there was one other well that was logged
6 that way, 32-P was logged that way. It was one of the wells
7 that was drilled where there's a volcanic anomaly. That had
8 some very interesting results, which I didn't present, some
9 interesting what appeared to be flow up out of the--let me
10 back up. In that well, there was a basalt layer intersected
11 by the well, which was sort of suggested by the magnetics,
12 and alluvium above and below that basalt layer. And, a
13 screen was put into the well, both in the basalt layer and in
14 the alluvium above and below the basalt. It would appear to
15 be the case there that there was flow coming from the
16 alluvium below and the alluvium above, into the basalt, and
17 then flowing out in the basalt. And, that's a very
18 preliminary result that hasn't been analyzed to the extent
19 that the result from 24-PB has.

20 HORNBERGER: So, for 24-PB, Barry calculated specific
21 discharges. You didn't present those?

22 REIMUS: Right.

23 HORNBERGER: Do you recall offhand what the numbers are?

24 REIMUS: No, I don't. I don't recall offhand what they
25 are. He primarily calculated linear velocities based on a

1 number of assumptions, and then tried to back out a specific
2 discharge.

3 HORNBERGER: My recollection is they are 10 to 100 times
4 bigger than the ones you cite for the alluvium wells.

5 REIMUS: Well, of course, again, this is in, you know,
6 fractured volcanics, not alluvium. But, as far as where the
7 linear velocities--I mean, if you take the specific discharge
8 that's assumed, or the range of specific discharges that are
9 assumed in the performance assessment parameter
10 distributions, and then you also look at the ranges of
11 effective flow porosities in the fractured volcanics, those
12 linear velocities that you back calculate out of like to 24-P
13 result, actually do fall within--they're at the upper end of
14 the range you would expect, but they do fall within the range
15 of that combination of parameters. You know, it's the
16 combination of the specific discharge and the flow porosity
17 that give you the linear velocity. And, it's certainly
18 within the range that's considered in TSPA.

19 HORNBERGER: So, if we go to Slide 19, you showed a
20 thermal log for? Now, this is the alluvium?

21 REIMUS: This is all--well, except for Screen 4, is this
22 volcanic breccia. But, yes, this is all alluvium.

23 HORNBERGER: And, so, the Screen 3 results, as you
24 pointed out, suggest that you have a higher flow zone there.
25 So, have you done the calculation? I mean, my recollection

1 is that you said the calibrated flow model had a specific
2 discharge of something like 20 and you were estimating from
3 the tracer test something on the order of 10. Is that just a
4 factor of two, or is it more than that?

5 REIMUS: I haven't done that calculation.

6 HORNBERGER: And, you can't eyeball it? Your eyeball
7 isn't that well calibrated?

8 REIMUS: Well, you know, as I said before, it's--you
9 know, I think before you write down a number, you have to
10 also consider the uncertainties associated with the well
11 completion, and the different, you know, porosities
12 encountered at the different depths. And, so, there is some
13 uncertainty associated with it, but no, and, in fact, it's
14 kind of interesting that Screen 3 actually was a poorer
15 producer than Screen 2 in terms of hydraulic conductivity
16 determined from hydraulic testing, just from looking at the
17 responses, the cross-hole, hydraulic responses. So, that
18 actually contradicts this thermal perturbation look at the
19 well. You, of course, have to recognize again, too, this is
20 a very local measurement. I mean, we're talking
21 interrogating only, you know, tens of centimeters at most
22 into the formation from the well bore, whereas, a cross-hole
23 hydraulic test is, you know, in this case, 18 meters. And,
24 then, the flow model is trying to do things at, you know,
25 kilometer scales.

1 HORNBERGER: Yeah, but I mean your interpretation of the
2 thermal isn't that you're just interrogating ten centimeters.
3 You're interrogating the flow into the borehole.

4 REIMUS: You're interrogating flow, in this case, around
5 the borehole. This is actually, you know, a closed off
6 borehole.

7 HORNBERGER: Yes. Right.

8 REIMUS: But, yeah.

9 HORNBERGER: To go back to Ron's question about the
10 greater predicted radionuclide transport through the
11 alluvium, the thrust of that slide is you now have a longer
12 predicted path through the alluvium?

13 REIMUS: Right.

14 HORNBERGER: And, did I hear you correctly; is it about
15 30 percent longer?

16 REIMUS: Well, it's about 30 percent or so of the total
17 flow path away.

18 HORNBERGER: Oh, of the total flow path?

19 REIMUS: right. However, previously, there was no
20 alluvium in the flow path up to the 18 kilometer boundaries.
21 So, yeah, it's 30 percent.

22 HORNBERGER: My recollection is from performance
23 assessments, that that can be pretty important in terms of
24 calculation of radionuclide transport, because that's where
25 you get the sorption of things like neptunium; is that

1 correct?

2 REIMUS: Right. It's very true, not only do you get
3 much greater surface area for sorption, but, you know, that
4 30 percent--I don't know the exact numbers, but I'm sure that
5 that 30 percent of the distance accounts for a much larger
6 percentage of the travel time because of the higher flow
7 porosity.

8 HORNBERGER: Right. And, so, these new results are
9 being fed into the TSPA?

10 REIMUS: Yes.

11 HORNBERGER: As we speak?

12 REIMUS: Yes.

13 HORNBERGER: Okay, thank you.

14 GARRICK: All right. Any other questions? I want to
15 move to the public comment as soon as we can. Dave, go
16 ahead.

17 DIODATO: Diodato, Staff.

18 Thanks for your talk, Paul. On Page 4, you had
19 the--on Slide 4 there, you had the two different hydrologic
20 flow models. I just want to look at this for a second. On
21 the left, if we look at the particle tracks, you know, in
22 some of the previous models, they would--Busted Butte, the
23 particle tracks would have to divert around Busted Butte, and
24 then reconverge. And, here, it seems like they don't see
25 Busted Butte at all. It's going to go ripping right through.

1 So, that's one observation I'd like you to help me
2 understand.

3 And, the other part of it is on the 2006 hydrologic
4 flow model, framework model, those particle tracks there
5 don't reflect that flow model. They reflect the old; is that
6 correct? I mean, they're identical, so it seems like--

7 REIMUS: Yes, I'm sorry, I should have mentioned that.
8 Those particle tracks are both actually from the 2006 flow
9 model. I just showed--I just superimposed them both on the
10 old hydrostratigraphic framework model and the new HFM. So,
11 those are both--those are identical particle tracks from the
12 2006 model.

13 DIODATO: So, what change, I guess that would suggest to
14 me that the Busted Butte hydraulic parameters of the Busted
15 Butte now don't have that much of an influence on the
16 particle tracks. It looks like it really just kind of rips
17 right through, doesn't really see it, you know?

18 REIMUS: Yes, I don't have a good answer for that. I'm
19 not directly involved in the flow modeling. I actually don't
20 recall the diversion around Busted Butte. Maybe I could ask
21 Charles Bryan to come up and answer that one.

22 DIODATO: It's been noted that--so, then on Slide 11,
23 you had three pathways postulated here, but then in a
24 subsequent test, you only had two active pathways. So,
25 what's your hypothesis of why in one test, you could see

1 three flowing pathways, and in a subsequent test, the same
2 wells, only two pathways would show up?

3 REIMUS: Yes, good question. And, that caused a lot of
4 pondering for sure. Can we go into my backup slides? I
5 don't remember which one it is, but I do have a backup slide
6 that addressed that.

7 DIODATO: We'd like to get these backup slides, by the
8 way, just so we could have them.

9 REIMUS: Two back. That one, okay. Okay, so this is a
10 few working hypotheses. First of all, something I didn't
11 mention was that the tracer solution that was injected in the
12 first test had a very high density. It was basically, you
13 know, on the order of 5 percent more dense than the
14 groundwater, which is pretty hefty, you know, we're talking a
15 few hundred thousand parts per million concentration of
16 tracers. The second test was much, much lower
17 concentrations, so there certainly was a difference in the
18 density of the tracer solution, even though it was the same
19 volume injected, and it was chased with the same amount of
20 water, and the pumping rates were all the same. Everything
21 was the same in the two tests, except for that.

22 So, clearly, density driven flow is something that
23 you would look for as part of the explanation, because you
24 would definitely expect it to occur in the first test, and
25 not in the second test.

1 And, so, there's a couple explanations here. The
2 first site, the explanation on the left here really just
3 accounts for the two pathways that we see, which, you know,
4 appear to be the case in both tests. There appeared to be
5 two flow pathways contributing to the transport in both the
6 first and the second test. It was the third pathway that was
7 different between the two.

8 One possibility for explaining the two pathways is
9 that we did only have a very small amount of tracer solution,
10 and then a small amount of water that chased it out of the
11 borehole. If there happened to be a higher conductivity
12 channel somewhat near the injection borehole, but not exactly
13 directly intersecting it, it's possible we could have pushed
14 a small amount of tracer into this high K channel. That
15 would have, in effect, broken off and flowed quickly toward
16 the production well. And, then, what remained behind that
17 wasn't initially pushed into that high K flow pathway would
18 have eventually bled into that, and resulted in the second
19 peak.

20 The other possibility in the cross-section view is
21 that there--you know, this is clearly a layered system. This
22 is 100 foot interval. We could have had a high conductivity
23 narrow layer, with a small amount of tracer mass, and that
24 first flow pathway only had about 5 percent of the tracer
25 mass in it. A small amount of mass could have transported

1 quickly through a high K channel, and the majority of the
2 rest of the mass would have moved through a much lower K
3 stratification or portion of the system.

4 I speculate that the third flow pathway occurred in
5 the first test, but not in the second test, because there was
6 quite a bit of tracer mass that sank out the bottom of the
7 borehole because of that density contrast. It would have
8 eventually deluded and worked its way back to the production
9 well under the influence of the pumping. But, that is my
10 explanation for why there's a third pathway in that first
11 test, but not in the second test. There wasn't that density
12 contrast at all in the second test. It was basically the
13 same ionic strength almost as the groundwater. So, that's a
14 couple of speculative explanations.

15 DIODATO: Okay, thank you.

16 GARRICK: All right, I think we're going to move on
17 because we want to make sure that people that have to leave
18 at 4 o'clock have an opportunity to make the comment we had
19 promised that they could make. So, if we have to, we can
20 come back to this. But, thanks very much.

21 And, as we promised, I guess, Judy, you were going
22 to make your comment, give us the benefit of your comment at
23 this time.

24 TREICHEL: Judy Treichel, Nuclear Waste Task Force. I'm
25 not sure how beneficial this is. But, it may be interesting.

1 One of the things that is so amazing, and I know
2 I've been doing this for as long as the Board has been
3 around, is the way things change, and the fact that the site
4 that was recommended is so very different from the site
5 that's headed, they say, toward licensing. And, there just
6 seems to be a sort of a chaotic thing going on here, and I
7 thought part of that really showed up in the GNEP
8 presentation, because it always seems like something that you
9 don't know too much about really looks good. And, I think
10 that that was really clear when Mr. Laidler said that he
11 would just as soon have the waste from that in his back yard.
12 And, I always have the urge to ask what his address is.

13 But, we just--we're just racing toward something
14 that's 20 years old, and there's such a disconnect between
15 site recommendation and site licensing now. And, I'm not
16 going to be here for the PVHA presentation, but in part of
17 that, it says that they're going to get the results out of
18 that in June '08, which I guess will come careening in along
19 with the license application, or probably, the way things go,
20 even after that time.

21 But, since the time that this started, when DOE was
22 promising that if every single rule at the time wasn't met,
23 they were going to walk away, we were talking about inches of
24 transport, groundwater travel time during centuries, and now
25 we're talking about meters per year. And, I know it depends

1 upon if you're talking about alluvium or if you're talking
2 about the carbonate aquifer, but when you talk to the guys
3 that worked for a long time at the Nevada Test Site when they
4 were doing a lot of drilling out there, you hear stories
5 about really rapidly moving water, and you hear stories from
6 people who have been around Amargosa Valley for a long time
7 about the various temperatures in the water, and all of the
8 questions. And, we just keep coming up with so many more
9 questions, and I think the Board really needs to be thinking
10 about that.

11 And, I guess I get really offended as a Nevadan who
12 like almost every other Nevadan, really doesn't want this
13 thing to happen, and to be looking at it and not to be even
14 an alarmist in order to be alarmed about it. The discussion
15 that was carried on here about the water use when you had the
16 Inyo County presentation regarding the carbonate aquifer, and
17 the discussion about well, what year does that hit. Okay,
18 great. Let a lot of people drink for a while, and then they
19 can give it over to the Yucca Mountain project, and that's
20 bizarre. We have been battling with EPA, who of course still
21 has not put out a standard, over the us versus them kind of
22 thing. And, you'd have one standard for the people now, and
23 one standard for the people way down the road.

24 Well, I think this is the epitome of that, that you
25 would have water available for people now, and then the

1 people in the future would have to accommodate or include
2 Yucca Mountain in their ability to be able to drink water.
3 And that's where Nevada has really been upset over this
4 thing, and that's where we're always going to be upset about
5 it, and that's why we're not going to quit until we win, no
6 matter how long it takes.

7 And, this situation just gets set up for that. The
8 battle lines are continually drawn, and there's always this
9 line in the sand, and we're not going to cross it. So, it
10 would seem to me that the Board, in looking at the scientific
11 validity of this whole thing, really needs to look at the
12 changes over time, and not give them a pass on this. A lot
13 of the stuff you're hearing is crazy, and it should not go
14 any further than presentations here.

15 Thank you.

16 GARRICK: Thank you. Thank you, Judy.

17 Steve, were you going to make a comment, or--

18 FRISHMAN: Steve Frishman, State of Nevada.

19 I just wanted to go a little further on this
20 question of the discussion that went on over future water use
21 in Nevada. John, I got the impression that you thought that
22 there was something going on in the water planning area in
23 Nevada and California that was somehow detrimental to this
24 program. And, that's not the way it goes. It's Nevada's
25 sole responsibility, and California's, to plan the use of

1 their water. And, it's perfectly reasonable in this part--or
2 in southern Nevada especially, and now we're beginning to see
3 in Inyo County as well, it's perfectly reasonable to assume
4 that there is going to be very rapid growth, and the basis of
5 that rapid growth is going to be available water, primarily
6 the carbonate aquifer.

7 And, to somehow imply that we in those two states
8 don't have the right to plan for that future development and
9 the right to use that water as our laws permit and as our
10 societal needs require, to somehow imply that it's wrong to
11 do that in the face of the fact that somebody else wants to
12 possibly contaminate this aquifer is just totally outrageous.
13 And, I think it shows maybe the--it's symptomatic of the
14 level of concern that we've had all along in Nevada about the
15 fact that there is some federal right that we're not supposed
16 to interfere with, when, in fact, we have our own rights.

17 I recall when I was working in Texas and Guthsmith
18 (phonetic) County was under consideration. And, it was,
19 well, the second richest farm county, agricultural county in
20 the country at the time, and there was a lot of concern about
21 the salt pile that would result from mining out a repository
22 in the salt beds beneath Guthsmith County. It would have
23 been a large salt mound. And, in that area, people were
24 growing wheat, growing sugar beets. The immediate area was,
25 in fact, the seed farm that developed the genetic red winter

1 wheat, which is the staple of wheat in this country. And,
2 people started saying well, this salt pile is going to result
3 in increasing salts in our groundwater. We already have that
4 problem from irrigation. We have salt build-up and we have
5 to deal with it. Now, you're bringing salt to the surface to
6 be blown around, and the Department of Energy's answer was
7 well, you people should grow more salt tolerant crops.

8 And, I see the same kind of thing in the assumption
9 that Nevada and Southern California don't have the right to
10 have a planning perspective over a very long period of time
11 over the use of their water, and that somehow, we do not
12 have--or somehow, we're preempted because there's some
13 greater plan. That just can't happen, and it isn't going to
14 happen. And, I think it's particularly disturbing to see
15 that outside of the Board's responsibility, there's even
16 concern about what our water resources planning is in the
17 State of Nevada, and in the State of California, and how we
18 would intend and attempt to protect those water resources for
19 their currently legal uses.

20 Thank you.

21 GARRICK: Yes, just on that. My question was only
22 informational. I was raising the question about what studies
23 had been performed more than I was concerned about who should
24 do the studies. And, the fact that these aquifers were as
25 interconnected as they are, which was something that was not

1 generally discussed in the past, would seem to indicate that
2 the regional boundaries are quite extensive here, and was
3 there not some consideration of that in the studies that were
4 either underway or being planned, and that's all I was really
5 trying to better understand.

6 FRISHMAN: Well, there's a difference between the
7 regional studies you were talking about, and the implication
8 that I got, which was regional--the concern over whose
9 responsibility and right are involved in regional water
10 planning.

11 GARRICK: I would never get into a State's rights or--
12 no, that was not--

13 FRISHMAN: Well, you damned near started one.

14 GARRICK: That was not my perspective at all.

15 KADAK: Kadak. I was trying to connect the morning
16 presentation with the afternoon presentation. And, I didn't
17 see water--I didn't see how they were connected, frankly.
18 The water flow paths looked like they were, as I think George
19 said, going south, and the other one was going like
20 southwest. Is there any clarity in where this water from the
21 Yucca Mountain area is going? Perhaps both of you can answer
22 that.

23 FRISHMAN: Well, I think in the specifics of where it's
24 going, there's a lot to either be agreed with or not agreed
25 with in the last presentation. It's certainly new

1 information, or a compilation of newer information than was
2 out there even for the 2004 model. But, there's also, at
3 least as was talked about this morning, this sort of unknown
4 about the relationship between the carbonate aquifer, the
5 tuff aquifer and the alluvial aquifer. And, we've really,
6 through all these years, as was pointed out, we've really
7 only had one data point for the higher head in the carbonate
8 aquifer. It's become sort of a core in everything all the
9 way to transport.

10 Now, there's apparently another data point that
11 suggests, but doesn't show the same level, but I think what
12 it all comes to is we have a pretty good understanding that
13 the discharge point for water that flows under Yucca Mountain
14 is Franklin Lake Playa, and also very likely springs in Death
15 Valley. And, the exact movement of that water, I don't think
16 we have enough data on. But, I think it also relates to some
17 unknowns between--in the relationships between those three
18 aquifers, and a much greater unknown about the carbonate
19 aquifer in that particular area.

20 So, in some discussion that we're going to have
21 with the California Energy Commission in a couple weeks,
22 we're going to be in this same discussion again, where Inyo
23 County's study is going to be discussed. I'm going to be
24 talking about some things, and it's--there are two different
25 perspectives. But, we're talking about maybe not the same

1 water, and we're talking about two discharge points, one of
2 which I feel pretty confident is an important discharge
3 point. The other I think is growing in importance as being a
4 discharge point, and I think the discussion about the head
5 reduction is an important one, because it makes the
6 possibility of the Death Valley discharge case probably as
7 important as the Franklin Lake Playa discharge case if you
8 are within--at least within the calculational range that it
9 is possible.

10 So, it's probably that there is confusion to some
11 extent about discharge, but I don't think it's as much as
12 saying it's either there or there. I think it is very likely
13 both, but for very different reasons. And, anybody else can
14 try to explain that if they want.

15 REIMUS: This is Paul Reimus.

16 I guess I just wanted to point out that the flow
17 pathways or the particle track pathways shown this morning,
18 and the ones that I showed, I think are actually completely
19 compatible. We're talking about two different depths in the
20 flow system. We were talking about the carbonate aquifer
21 this morning. Now, if I'm not mistaken, I believe a particle
22 was just put into the carbonate aquifer below the repository
23 this morning, and flowed through the carbonate aquifer, based
24 on, you know, the regional flow model, straight toward Death
25 Valley. Is that correct?

1 So, that was all carbonate, and in the Yucca
2 Mountain models that my particle tracks, or the particle
3 tracks in my presentation reflected those particles never see
4 the carbonate aquifer. There's that strong upward gradient
5 that was discussed this morning. The particles, once they
6 hit the saturated zone, they stay in the fractured volcanics,
7 relatively shallow actually. There's a vertical upward
8 gradient within the volcanics as well. And, these two
9 aquifers, it's fairly well recognized, at least in the
10 vicinity of Yucca Mountain, are quite separated from each
11 other. I mean, you wouldn't have this 20 meters of head
12 difference if there was good communication between the two.

13 I mean, we are talking about two different things
14 here that aren't incompatible based on the understanding of
15 the flow system and how separated the volcanics are from the
16 carbonate aquifer, at least in the vicinity of Yucca
17 Mountain.

18 KADAK: Is there a graphic that you can show us that
19 shows where these two aquifers are relative to where the
20 waste is?

21 REIMUS: Actually, the best one I have at least was in
22 my backup slides, showing the vertical cross-sections, if you
23 wanted to see that.

24 KADAK: Perhaps. I mean, I'm trying to--you're talking
25 two links; right?

1 REIMUS: One high, one low. And one is going towards
2 the southwest towards California, and the other one is going
3 sort of south, from what he showed; right?

4 KADAK: I'm sorry, I'm confused, but maybe that one
5 slide will help me. Can you find it? Is it that one there?

6 REIMUS: There we go. So, this is a vertical cross-
7 section now of the base case model in 2004. This was
8 actually an alternative model in 2004, which by then, you
9 know, certainly was recognized. There was differences based
10 on the Nye County program. And, then, this is the 2006 based
11 case. But, this blue color here is basically the carbonate
12 aquifer, and the dark blue is up-thrusted carbonates. But,
13 the carbonate aquifer is very deep. The water table is this
14 line here.

15 This actually, the earlier model actually was cut
16 off at the water table. The water table is the dashed line
17 going across here in the two lower figures. So, the blue and
18 the dark blue is the carbonate aquifer, and that certainly is
19 the main aquifer in a regional sense that John was talking
20 about this morning, and it underlies all these volcanic
21 units. The repository is here. The 18 kilometer boundary is
22 here. Death Valley is way out of this cross-section here.
23 But, anyway, all the particles in a vertical profile coming
24 out of the repository are going to hit the water table and
25 because of that strong upward gradient, are predicted to

1 remain shallow, go through the various volcanic units until
2 they hit this alluvium out here. And, then transport in the
3 alluvium to the boundary, and they remain separated from the
4 carbonate because of that upward gradient.

5 KADAK: So, your suggestion is that even if the
6 repository hundreds of thousands of years from now, or
7 sooner, starts dissolving the waste, you're staying that it
8 will stay in that upper aquifer?

9 REIMUS: That's what all the data suggests, that, or is
10 used in the saturated zone flow model, the site-scale
11 saturated zone flow model. There's both the head data, the
12 geochemical data, certainly suggests those aquifers are very
13 compartmentalized with respect to having any communication
14 with each other. And, so, that's basically what the data
15 supports.

16 KADAK: Is the flow path in those two aquifers the same
17 direction, or is one, as you suggest, going south and the
18 other one, as the earlier speaker suggested, going to Death
19 Valley?

20 REIMUS: I certainly can't speak to the deeper one,
21 especially in the vicinity of the repository, where, as was
22 mentioned, there's only a couple of wells. But, you know,
23 that one basically reflects the regional flow model that the
24 USGS has done. It reflects, you know, the information in
25 that regional flow model, the deeper one, and exactly what

1 direction the flow goes in the vicinity of Yucca Mountain, I
2 can't speak intelligently to.

3 I can speak to the model that the project has
4 developed, which is based on water levels and geochemistry, a
5 number of different lines of evidence in the volcanic units
6 and in the alluvium to the south.

7 KADAK: Okay, thank you.

8 GARRICK: Okay, thank you. All right, I think we're
9 going to take a 15 minute break. We're not off schedule as
10 much as it sounds, because we have had probably the
11 substantial part of our public statement, so I think we'll be
12 all right. 15 minute break.

13 (Whereupon, a brief recess was taken.)

14 GARRICK: Our next presenter is Mark Johnson from
15 Bechtel SAIC. Mark is the Project Engineer for Subsurface
16 Engineering. So, without further ado, here's Mark.

17 JOHNSON: Thank you, Dr. Garrick.

18 Yes, I'm going to talk a little bit about the waste
19 package design, prototyping. I'm also going to show in a
20 little bit of the prototyping, what's going on up at INEL,
21 about a 50,000 foot high-level, take a look at some of the
22 things they develop there.

23 Just as an outline, we're going to go over the
24 design requirement for the waste package, description and
25 major features associated with the waste package, we'll go

1 over some of the changes that have gone to the waste package
2 with the implementation of the TAD concept, touch on the
3 design code for the waste package, and then I'll hit on the
4 prototyping programs.

5 Okay, from design requirements, we have both
6 preclosure and postclosure design requirements. Obviously,
7 we have to fulfill safe loading of the canister, the
8 canistered spent fuel and high-level waste. It's
9 transportable within the surface facilities, it has to be
10 transportable and placed in the underground. We have to
11 safely and remotely close it. That's closure cell.
12 Retrievable, and it's got to meet its preclosure safety
13 requirements, that's breaches from drops, some handling
14 equipment, and rock fall scenarios, run-away scenario in the
15 emplacement vehicle, that sort of thing. We'll talk a little
16 bit about preclude criticality, however, that's really
17 transferred over to the TAD, and not the waste package
18 anymore.

19 And, postclosure. It is important to barrier
20 capability and it must meet long-term dose performance
21 requirements.

22 Some features here. We talked about it earlier
23 today in the other presentations, but, we have an Alloy 22
24 outer corrosion barrier. We have sleeves on each end. They
25 provide stiffness and also contact points. One of the

1 changes with the new surface handling facility, the TAD,
2 there is no longer removable trunnion collars here. It's all
3 handled via the pallet. This is the inner stainless steel
4 316 inner vessel. Also, there's a two lid instead of the old
5 three lid design that's new to this.

6 We touched on it briefly. We removed the internals
7 from the fuel bearing waste packages. We increased the
8 volume to take the TAD. We changed the number of
9 configurations down from ten different configurations, there
10 was a BWR, there was a PWR, there was a 12-PWR, down to six
11 configurations, and the criticality is, like I mentioned,
12 part of the TAD now.

13 We removed the inner Alloy 22 lid. We also added a
14 shield plug to the design of the waste package. Both the TAD
15 and the naval waste packages have integral shield plugs to
16 them. In order to get the dose level down on the closure
17 cell equipment, consistent with those two, a shield plug has
18 been added to the DOE high-level waste bearing waste
19 packages. And, I mentioned we removed the trunnion collars
20 to facilitate the handling approach in the surface
21 facilities.

22 This is just a quick view of the different
23 configurations. Right now, the 21 44-BWR TAD and the naval
24 long are identical. That will probably change as TAD
25 information comes in from the vendors. But, right now,

1 that's what we're using. We've got the DOE waste package,
2 and then also the naval short as well.

3 For criticality control imposed on all canisters.
4 Waste package doesn't have that feature anymore. Neutron
5 poisons in the DOE canisters are tailored to whatever type of
6 DOE fuel there is, and that comes with the canister.

7 The TAD is performance based into the spec., and it
8 can either be neutron absorber plates, tubes, of borated
9 stainless steel, and it can do more of an analytical method
10 and show the postclosure, how they meet criticality. I'm
11 probably going to focus on this one. And, that is all
12 provided in the performance specification for the TAD that
13 was discussed this morning.

14 An exploded view. This is the TAD naval waste
15 package, real quick, exploded view of this, you've got the
16 inner lid, with the spread ring. There's a couple fillet
17 welds on it. The outer lid has a full depth closure weld on
18 it. Thicknesses, the outer corrosion barrier is a 1 inch
19 thickness, the inner vessel is 2 inches, and those are
20 consistent with the lids. Basically, the same thing, except
21 you're seeing here the integrated shield plug that is on the
22 DOE package. It provides a shielding on top of these, and
23 that varies between 8 or 9 inches, depending on the waste
24 package configuration. I think the 2 MCO2 HLW is the 8 inch,
25 and then the five packs are both the 9 inch shield plug, but

1 the configurations are the same.

2 One of the things we're working on now is putting a
3 taper on this to make remote placement a little easier, and
4 then as well as where we place this. The old concept, or
5 current concept, I should say, is to place these lids on in
6 the closure cell. Since this is a lot thicker, and the
7 shield plug integrated with the lid, the INEL equipment
8 doesn't quite handle the weight of that, so we're looking at
9 placing this in the--using the canister transfer machine in
10 the facility, and then taking it to closure cell, where the
11 spread ring will be spread out and the fillet welds made.

12 Codes and standards that we're applying to the
13 design of this, I think it was in 2005, BSC along with the
14 code consultant, put together a position paper on the code
15 approach to the fabrication. The inner vessel is fabricated
16 to the ASME Code, Section III, Division 1, Subsection NC, and
17 we'll have an N Code stamp affixed to it.

18 Outer corrosion barrier, basically the same code,
19 Division 1, NC, however, it won't be N stamped. Materials
20 are specified in Section II, and NDE Section V, and welding
21 to Section IX.

22 Part of the reasons we've gone that route, you
23 know, in the position paper, we discuss--there's a lot of
24 different things, but I think your heavy hitters is
25 Subsection NC, this is precedent for NRC licensing of

1 basically storage casks as of late. There is also more
2 vendors that are qualified to do NC work than such that the
3 NB, a little stricter subsection, and that's why we've gone
4 that route.

5 From the prototyping on the waste package and
6 components, program objectives are to develop and confirm the
7 fabrication methods. I'll go over some lessons learned that
8 we've got with the first prototype. Inform design of design
9 alternatives. Part of the program allows us to change some
10 things with the prototype, is we go out and investigate how
11 well they worked, and then if it's successful, we can roll
12 that back into the design of the waste package. We want to
13 look at some commercial vendor capability. It's hard to do
14 with sending procurements of one waste package out there, but
15 we're basically getting the waste package out there, getting
16 companies familiar with fabricating them, and the issues with
17 it.

18 Along the same lines, we're developing qualified
19 vendors, gives us an idea of future costs and fabrication
20 durations. Prototyping costs are naturally a little higher,
21 but it gives us an idea of the ballpark, especially when you
22 start buying Alloy 22. If you're familiar with the market
23 and nickel right now, it's pretty expensive stuff. We look
24 at process variability. We want to go out, have a couple
25 different manufacturers prototype the same waste package, and

1 we'll compare them and see how much variation there is. And,
2 we also plan on using these to do start-up testing, factor
3 acceptance testing, provide for training, as well as we do
4 some destructive testing of the prototypes to get some
5 information on residual stress, those sort of things, and get
6 that information over to the lab.

7 I touched on a little bit informing design. We
8 figure out tolerances, how well things, we can fabricate, we
9 provide samples for destructive testing and ensure handling
10 techniques are achievable. This is just a little picture of
11 the prototype we just finished up here last January.

12 I have here the different things--we'll go over
13 what we can do with these prototypes. We have, I don't know
14 if you guys have seen the transport and emplacement vehicle,
15 but I'll have a little animation here that shows an
16 emplacement activity. That vehicle picks up the waste
17 package in the surface facility and transports it all the way
18 down underground. Maybe it's a newer concept you haven't
19 seen yet. But, with these prototypes, we can check different
20 things. We can look at receipt inspection to look at hey, we
21 shipped this thing across the country, how much damage did we
22 get to it. We can use these waste packages for handling in
23 the surface facilities, as well as in here. We can use them
24 to test fit run and loading TADs into them, loading the naval
25 package, loading the DOE waste.

1 We can test them, maneuver them around on the
2 emplacement pallet, and handling the waste package after
3 closure, emplacement. And, then, also shipment to the
4 vendors to do some extensive factory acceptance testing.

5 Go ahead and click the video. This will show you
6 the--it's basically an emplacement activity. It comes in,
7 opens the doors. You're going to see the bed plate rolled
8 out. That's the mobile shielding plate. It's lowering the
9 waste package, backing off. It will close, and exit the
10 emplacement drift. But, the reason I'm pointing that out in
11 here is that needs a pretty extensive factory acceptance
12 test. I would like to test it with actual waste packages.
13 So, those are the sort of things we can use these prototype,
14 after we have manufactured it, for. Also, in the surface
15 facility, they have a rather unique piece of equipment that
16 grabs that, tilts it up, so we can provide the different
17 waste packages to these vendors to support some extensive
18 factory acceptance testing.

19 This is a picture of the annealing process of the
20 first prototype. It was a 21 PWR, so it had the fuel basket
21 in it. That's the annealing of the outer corrosion barrier.
22 It's essentially upside down, so it's going in with the open
23 end. What you see here is a snorkel to get the steam out of
24 that as it is placed into the clench tank. And, my numbers
25 on the top of my head, I think we heated it up at about 100

1 degrees an hour, up to 2050 plus or minus 50. I can't
2 remember the whole time, but the quench time was supposed to
3 be under two minutes, and the water was at an ambient
4 temperature, to get us the results we wanted for the
5 annealing.

6 It finished in January of this year, and it's
7 presently being stored at the fabricator awaiting
8 implementation of the testing program. This first waste
9 package has a life to be tested, cut up and some results for
10 the science people.

11 Residual stress measurements. These are the
12 testing we're going to do to it. They're going to measure
13 stress fields due to the solution heat treating. We want to
14 look at, after we test it, is it consistent, did we have any
15 dimensional changes through shipping.

16 Mechanical and corrosion properties. We're going
17 to damage this waste package. We're going to actually go
18 measure the energy, the impact with different, you know, ball
19 bearings, different things to try to figure out how does
20 damage that may happen in the facility affect the residual
21 stress, and then maybe hopefully develop a criteria, saying,
22 if in mechanical handling, it gets dinged up, what type of
23 marring is acceptable. That's part of the planned testing.

24 Some samples are being provided for corrosion
25 testing. We're also planning on heating it up and then

1 placing the lids to see if there's any problems with placing
2 the lids on a waste package that's at the temperature it
3 would be after it's loaded with waste, and then the lids are
4 going to be placed on it.

5 Some lessons learned. On the welding, had pretty
6 good results with the welding, and didn't get any defects on
7 it. Did find out we missed a fillet weld when they did the
8 code analysis at the fabricator, and, so, that's been added.

9 On the machining, we found the guy that--the
10 machining sub tried to go a little too fast, he broke some
11 tools, and had to slow it down. It took a lot longer than
12 they anticipated. So, we've added that. We don't tell them
13 how to machine it, but we add in that machining Alloy 22
14 takes more time, to consider that in your schedules. And,
15 then, we give some recommendations, but we want to make sure
16 we're not trying to take the responsibility for manufacturing
17 this away from the vendor.

18 One of the things we had thought was the ability to
19 machine the inside of the outer corrosion barrier after the
20 annealing. When we annealed it--well, just go to the next
21 one, and I'll get to that. When we annealed it, we had some
22 distortion. We put essentially spacers in it to keep it from
23 deforming, but we still had some distortion at the top. We
24 originally planned on the vendor having the capability to
25 machine it out if that was the case. Before we annealed it,

1 we placed the inner package and it fit. Afterwards, we
2 measured and knew it wasn't going to fit, so we had to do
3 some machining to get it in there.

4 So, we have added extra fabrication stock to the
5 outer corrosion barrier to allow some machining. But, the
6 vendors out there, not everybody has the ability to reach all
7 the way in. Now, the distortion we noticed was at the outer
8 trunnion sleeve. We're going to try to anneal the outer
9 trunnion sleeve this time separately prior to installation.
10 We think that the stress that was in that actually deformed
11 it during the annealing.

12 Also, we found out we have the thermocouples lined
13 up in there that we had a few hot spots where the steam vapor
14 wasn't getting removed by the snorkels. So, we're going to
15 introduce a spray system in there to help mix that up and
16 remove those steam pockets.

17 Follow-on prototyping, we've had a change. Like I
18 said, the first one was a bare fuel type. We're going with
19 TAD-bearing, which is also right now identical to navy, and
20 then also the high-level waste. We do have the ability to
21 change this around as we see fit, but we wanted to go with
22 three of the similar ones so we can look at some vendor,
23 different vendor manufacturing variances with the same type
24 waste package.

25 This one now is planned to go out for--the design

1 documents are done, the spec is done, the code spec is done,
2 probably go out for bid in the 2008 time frame.

3 On the component side, we do have the pallet and
4 the drip shield that we do plan to prototype. Right now,
5 we're looked at two pallets, long and short. I don't think
6 pallet is too complicated, but they will be used to support
7 factory acceptance testing and start-up testing for the waste
8 packages.

9 And, the drip shields, two of them, we want to
10 confirm the connection/interlocking feature, as well as
11 support the factory acceptance test and start-up testing for
12 the drip shield emplacement gantry.

13 This one just hit the street. We're going out to
14 procure the mock-ups to support the actual closing
15 demonstration at the INL facility for the closure system.
16 These are going to be full diameter, but much shorter height
17 mock-ups to support that, and we're scheduled to complete
18 procurement of that and have it to INL by July 2008 to
19 support their testing of the system in October of 2008.

20 I'm going to give a brief overview of the closure
21 cell prototyping. This one is not under my area of
22 responsibility, but I'm going to hit some of the high points
23 of it.

24 This prototype doesn't just look at the welding
25 system. It also has in it the NDE, the plasticity

1 burnishing, the purge, and inerting of the waste package.
2 So, it's more than just welding. It's the whole system.
3 They've got the robotic arms. They have they call it the
4 remote handling machine. And, also, equipment for different
5 sort of recovery. But, they're going to go over the welding,
6 NDE, leak testing. They also are putting together the
7 process ops for that, and this unit would eventually
8 hopefully support start-up testing, and also be a training
9 unit to train operations staff, then operating the actual
10 system.

11 This is a picture of the current welding end
12 effector for the closure weld. It's going to be mounted on
13 the end of the robotic arm. The plan is to have two of these
14 arms that will each be doing 180 degrees of the weld.

15 Just an example of the end effector for the NDE.
16 The visual inspection was on the welding end effector, but
17 this one does the eddy current and the ultrasonic. It is
18 also attached to that same robot arm I showed in the last
19 slide. As you can see here, this is a little mock up of the
20 top of the closure weld of the waste package, and how it
21 travels along the waste package.

22 Inerting and leak testing. The inner lid has a
23 purge port on it, so the plan is to purge and fill the cavity
24 space between the TAD and the inner vessel with helium, and
25 then they leak test using this tool, leak test to see if they

1 have any leaks. You can see essentially in this little blow-
2 up, which is actually a TAD, but there's your spread ring,
3 you fill it welds, you've got a gasket here and here, and
4 eventually pull a vacuum on it, looking for any helium
5 leakage.

6 So, in summary, we went over the waste package
7 design requirements, how the waste package is changed to
8 accommodate a TAD. The criticality has been moved out of the
9 waste package into the TAD as a specification on the vendors.
10 And, then, we went over the prototyping program, talking
11 about the first one, the plan for the future, and then just
12 briefly hit on the closure cell.

13 So, with that, I can take some questions.

14 GARRICK: Okay, Henry, and then Howard?

15 PETROSKI: Petroski, Board.

16 I'm pleased to see that this is underway. I was
17 wondering how long you expect to be doing this? You said you
18 contemplate about six prototypes. I'd like to know how long
19 that might go?

20 JOHNSON: From the waste package side, the plan as
21 scheduled now is basically from now to the 2011 time frame.
22 It's all dependent on funding, but that's what's in the plan.
23 Your pallet is probably in the 9 to 11 time frame, and your
24 drip shield is coming after that. That's currently what we
25 have in the plan.

1 PETROSKI: Now, this first prototype that you describe,
2 and you indicated that you would cut samples from that for
3 corrosion testing. Will those samples incorporate the damage
4 that you will inflict on the prototype?

5 JOHNSON: I don't think they do for the corrosion
6 testing. Those were going to be separate. So, you're saying
7 inflict the damage, take the residual stress and then give
8 them to corrosion?

9 PETROSKI: Well, there's some concern. I'll let the
10 corrosion people speak to it more directly, but certain types
11 of damage may affect how this sample responds to corrosion
12 testing. So, I would think it would be important.

13 JOHNSON: That's a good point. I'll take a note of
14 that. We get the requirements from that. We develop the
15 testing spec jointly with Sandia, their waste package,
16 corrosion people, to try to just say hey, what would you like
17 from this, and we can get it for you. It's no problem. We
18 cut coupons, tell a size, how many you want, and that sort of
19 thing. So, if that's something that they would like, that's
20 not a problem.

21 PETROSKI: I'm glad to hear that you'll take that under
22 consideration. What is the cost of this program projected?

23 JOHNSON: Claudia, is that something I can--

24 NEWBURY: Newbury, DOE. If you know an answer, I--

25 JOHNSON: Well, they each--I'm going to go, I don't know

1 the total, but I can give you a unit cost. The budgeted cost
2 and the actual cost was pretty much the same of each one, is
3 around a million dollars for a waste package. The pallet in
4 the 200k range, and the drip shield I want to say is in the
5 800k range. Well, testing program, it's going out for bid,
6 so I'm not going to discuss what that one is.

7 PETROSKI: Well, it's still relatively small potatoes
8 compared to the whole project, and I would hope it would be
9 given high priority for having its funding maintained.

10 JOHNSON: Thanks. Next question?

11 ARNOLD: Arnold, Board.

12 You talked about doing marring tests, and then
13 measuring the effect on residual stresses just in the same
14 vein as what Henry was talking about. My question is what is
15 the original--I mean, what is your specification on the
16 finish itself when it's delivered to you?

17 JOHNSON: I'm going to pull that number out of my head.

18 ARNOLD: Do you require a special polishing and--

19 JOHNSON: Part of the program is looking at the
20 polishing. We've got, one of the things we're going to test
21 is a frit versus electro polishing. I can't remember the
22 number on what type of mill finish we want. But, it wasn't
23 anything out of ordinary. We do want a matt gray, because
24 that's what we have modeled for thermal on the emussivity.
25 But, the number on the smoothness escapes me.

1 ARNOLD: This goes back to the corrosion issue?

2 JOHNSON: One issue that we're working with the lab on
3 now is that oxide film left after the annealing. That is
4 what they would like us to remove. They don't really have a
5 preference. The requirement there, they're putting in what
6 we call our postclosure parameters document, is a removal of
7 it without any specifics. And, when we talk about a strip
8 blasting, a concern a customer has is the incremental cost of
9 frit blasting 11,000-and some waste packages. So, they're
10 looking at do we really need to do this? What are the
11 effects of that oxide layer on the long-term corrosion? But,
12 in the prototyping program, we've added in the ability to go
13 and say hey, on this one, we want you to frit blast it so we
14 can see about what it takes, what kind of surface does it
15 leave.

16 A lot of things have been thrown around, walnut
17 shell, I think that's gone by the wayside because of the
18 organics that are impinging on it, and things like that.
19 And, electro-polishing is a little overboard for what they
20 need, and that's kind of a spendy little process step to get
21 a really nice finish on it. But, right now, I think the one
22 we're going to go forward with in this prototype just to try
23 it out is a frit blasting.

24 ARNOLD: All right, thanks.

25 GARRICK: Ron, David, and Mark?

1 LATANISION: I had another question, but just a follow-
2 up on Howard's. So, there is a plan to cut the type--cut
3 sections and explore their corrosion resistance relative to
4 what's been done in laboratory samples?

5 JOHNSON: I'm not familiar with what the lab wants to
6 do. They have asked us for coupons from this one to go do
7 testing. So, that's what I'm providing them. But, I'd have
8 to have someone from Sandia, Neal Brown, or somebody, to say
9 what they're planning to do with it. I'm just saying how
10 many do you want, and we'll have them for you, and you can do
11 with it what you want. So, I can't answer your question. I
12 can get an answer for you, but--

13 LATANISION: Yes, I'd like to have that, because I think
14 it does make a difference. I mean, all of the laboratory
15 tests have presumably been done on polished surfaces, and
16 we're now looking at the real world, and the real world isn't
17 a polished surface.

18 JOHNSON: Right. And, this prototype did not have that
19 heavy oxide layer removed from the annealing process.

20 LATANISION: Okay. And, the question I wanted--

21 JOHNSON: Are these actions getting captured? Because I
22 don't want to--Okay. I don't want to tell you something and
23 not deliver.

24 LATANISION: Right. I couldn't tell from your
25 discussion whether post-weld anneal, stress relief anneal is

1 part of a plan.

2 JOHNSON: The only weld on the outer corrosion barrier
3 that's done post-annealing is the final closure.

4 LATANISION: Yes.

5 JOHNSON: In the past, they have talked about either a
6 laser peening or a low plasticity burnishing. I did a value
7 engineering study. Both methods were reasonable. I think
8 low plasticity burnishing won out. INEL is now taking that.
9 It is going to be doing that in that mock-up to do that low
10 plasticity burnishing. So, yes, a good point, sorry I missed
11 that.

12 LATANISION: Okay, thank you.

13 DUQUETTE: Could I go to Slide 14, please? I just want
14 to understand what's happened here. This is everything in
15 place, obviously not the waste, but everything in place,
16 including the baskets, and it's been totally sealed, it's
17 welded at this point?

18 JOHNSON: No, this is just the outer corrosion barrier.
19 That's the only one that's annealed. The baskets and the
20 inner stainless steel vessel are separate. They are not
21 annealed. So, this is your Alloy 22 that's basically, you
22 can see the trunnion collars on it, and they weren't machined
23 yet, but basically, I think the only thing left to do on this
24 was the machining of the trunnion collars to put their
25 receiver grooves in for that.

1 DUQUETTE: Okay. I was confused by the fact that it
2 said PWR absorber plate waste package.

3 JOHNSON: Sorry. That was the configuration, but this
4 is just the OCB.

5 DUQUETTE: Right. Can you tell me something about the
6 kind of distortion that you got, and you indicated you're
7 going to try to machine distortions out, so when you put the
8 inner package--slide the inner package--

9 JOHNSON: That's one method. The other is try to
10 prevent it up front. So, we are going to look at both ways.
11 But, yeah, I can get you exact details on what sort of
12 distortions. Essentially, though, they were primarily in
13 this area right here, and we thought it was induced from that
14 thick trunnion collar on there. That's just one of the
15 speculations. So, we're going to go play with it.

16 But, I'll tell you what. I'll send you the exact
17 dimensions that it was off. It was significant enough where
18 before the annealing, the inner package was placed, it went
19 right inside. Afterwards, it wouldn't fit. So, we had to
20 machine it to get it to fit.

21 DUQUETTE: Okay. Duquette, Board.

22 Just to clarify for the record, since we are making
23 a record, it's 2150-F, not 2150-C.

24 JOHNSON: Oh, excuse me, did I say C? Yes, it was
25 actually 2050-F.

1 DUQUETTE: No, you said 2050. You didn't indicate which
2 kind of temperature it was. But, I saw a couple of eyebrows
3 raised because you would melt it at about 2050 C.

4 JOHNSON: You're right, it is F.

5 GARRICK: Mark?

6 ABKOWITZ: Abkowitz, Board.

7 First of all, I wanted to share my sentiments with
8 Dr. Petroski's about what you're doing here with the
9 prototype program. I know it's something the Board has been
10 anxious about for some time, and it's good to see the plans
11 and how it's coming into fruition.

12 My question had to do with the whole process as
13 you're getting into prototyping. And, I guess the easiest
14 way to ask the question is what happens when one of your
15 prototypes fails a test?

16 JOHNSON: That's part of prototyping. So, we've learned
17 something that didn't work. I would say that probably the
18 distortion on that was when we were discussing, you know,
19 they don't fit, how are we going to fix this, was originally
20 thought to be a failed test. We got it fixed, but you learn
21 from your failings as much as you do from your successes.
22 That would be rolled right into the next one. So, we haven't
23 had that happen yet, but that's kind of my answer, is, well,
24 why did it fail, we've got to go figure that out. How can we
25 prevent that failure? And, then, let's take that

1 information, roll it into the next one.

2 ABKOWITZ: Is there a formal process for corrective
3 action when there is a prototype failure? Can you describe
4 for me the inner workings of the Department when a prototype
5 fails, and how--

6 JOHNSON: I'd be speculating.

7 ABKOWITZ: --it gets rectified? Because you will have
8 failures, obviously.

9 JOHNSON: Sure. Sure. As part of the program, we would
10 probably document one way to handle that would be--I don't
11 know if it would be captured in the CAP system, or your
12 typical non-conformance report. Non-conformance report would
13 be justified as, you know, accept as is, rework, that sort of
14 thing. Your non-conformance reports are then rolled into the
15 next spec on how you, you know, essentially just like
16 something that came to the field and didn't conform, you
17 fixed it, and then what your fix is is as built into your
18 specification or drawings or as into this case, the next
19 prototype. Or, if it was the last one, the final prototyping
20 spec that we will go out and procure with.

21 You could also handle it with a condition report
22 system, but, you know, the NCR process is in place to capture
23 fabricated items or constructed item, non-conformances, which
24 this would fall under. So, that's kind of my speculative
25 answer of how we would handle it. Put it like any piece of

1 equipment coming to a facility that didn't meet its spec'ed
2 requirements.

3 GARRICK: Any other questions?

4 (No response.)

5 GARRICK: Okay, thank you. Thank you very much.

6 JOHNSON: Thank you for your time.

7 GARRICK: Okay, our final speaker for today will be
8 Kevin Coppersmith, who is a consultant to Sandia National
9 Laboratories, and he will report on activities to update the
10 probabilistic volcanic hazard analysis, and also give us some
11 background information about the original PVHA. Okay?

12 COPPERSMITH: My talk will actually entail both, the
13 probabilistic volcanic hazard analysis done in 1996, which
14 continues to be the basis for the license application. I'll
15 call that PVHA '96. As well as an update which is underway,
16 and I'll go through the process that's being followed for
17 both of those.

18 Next? In the presentation, I'm going to talk a
19 little bit about formal expert elicitation methodologies, for
20 those who are not familiar, the steps that are involved in
21 any formal structured expert elicitation, in particular,
22 drawing on the guidance that we follow in the Yucca Mountain
23 project. Review the context in terms of the technical
24 information available for the PVHA done in '96, as well as
25 information we now have for the update, and to go through the

1 activities for the PVHA update. We are well along on that
2 process, in fact, had a workshop that ended I think about 72
3 hours ago.

4 Next?

5 KADAK: Could I just ask why are you doing this again?
6 What prompted this--

7 COPPERSMITH: I have a slide that will summarize that.
8 It's basically new information that became available
9 subsequent to the '96 study.

10 KADAK: Was there a volcanic eruption somewhere nearby,
11 or what are we talking about?

12 COPPERSMITH: No, that would be more profound
13 information.

14 KADAK: Okay. I want to gauge the significance of your
15 presentation.

16 COPPERSMITH: Yes, it's definitely new data, it's
17 aeromag data, and I'll discuss that.

18 People have asked the question though on stability.
19 It was a question that was asked I think by Leon Reiter at
20 the workshop a few days ago, asked for long-term stability,
21 since this was done after ten years, what will lead to
22 stability over the next ten years, and those types of
23 questions have been asked.

24 When it comes to looking at the types of guidance
25 that we follow on this study, the PVHA, we look to a couple

1 of documents. First is the Kotra et al. It's a branch
2 technical position developed by the NRC specifically for this
3 project, provides the steps that are needed, and should be
4 followed in a formal structured expert elicitation.

5 We also lean on and use the so-called SSHAC
6 guidance study that was sponsored by NRC, EPRI and DOE, that
7 has very similar overall steps, but some differences in the
8 roles that experts play on the panel. And, I will talk a
9 little bit about that.

10 The applications of these methodologies have been
11 two, the probabilistic seismic hazard analysis that was
12 completed in '98, and the PVHA that was completed in '96.

13 Next? In terms of the steps of the elicitation,
14 this is from the NRC branch technical position, and, for
15 those that are familiar with the decision analysis
16 literature, these are common steps in studies of this type.
17 We begin with a discussion and description of the objectives
18 of the assessment. Of course, selection of experts.
19 Decomposition, identification and decomposition of the issues
20 into the salient elements that allow for assessment by the
21 expert panel. A big part of the studies these days is the
22 assembly and dissemination of a database. I'll talk a bit
23 about that in the PVHA.

24 Pre-elicitation training, to have the experts
25 familiar with many cognitive biases and other issues related

1 to the probability assessments. Elicitation of judgments
2 which can occur in a private setting, interview setting.
3 Post-elicitation feedback, which is the stage that we just
4 completed last week, and the update, is very important,
5 provides opportunity for the experts to understand the
6 implications of their assessments and identify the most
7 important issues. Aggregation, which of course is a big part
8 of any expert study, is the combination of the assessments of
9 the panel, and final documentation.

10 Going to the SSHAC guidance, most of the steps that
11 are recommended are identical. I wanted to point out a
12 couple of differences related to the roles of the experts
13 themselves. This particular panel spent a lot of time
14 dealing with the issue of the roles of experts. Do they
15 represent individual representatives, are they
16 representatives from the larger technical community? If so,
17 how can a panel represent that larger community?

18 And, the bottom line evaluation, or bottom line
19 recommendations in the SSHAC study was that members of a
20 panel of this type should act as evaluators, as opposed to
21 proponents. Proponents are a much more common scientific
22 role that we're used to playing, that we could be a proponent
23 or advocate of a particular technical position, we often
24 publish that position and talk about it at professional
25 meetings, and we look to the larger community to provide a

1 critique and review of that position. And, we allow others
2 to develop other positions. And, we have an opportunity then
3 for discussion and dialogue and disagreement.

4 The role of an evaluator, though, on a panel of
5 this type is to capture that range of views, to listen to
6 proponents and advocates of different positions, and
7 ultimately to capture that range of the community
8 distribution, if you will, if they were to have gone through
9 the same process.

10 So, this issue, and of course the technical
11 facilitator/integrator is the terminology that SSHAC uses for
12 those who have to or are responsible for facilitating that
13 process. But, in the end, also for integrating the
14 assessments across the panel.

15 So, this role of evaluators is ones that we have
16 implemented in the PVHA. Ultimately when we finish, the
17 representation of their range of uncertainty will be deemed
18 to be representative of the larger technical community.

19 KADAK: Excuse me. Could I just ask the selection of
20 experts, that's the toughest part of this solicitation
21 process, because as people are people, many of them have
22 positions, especially if they're volcanologists and
23 seismologists. I mean, there aren't that many of them,
24 frankly, and I'm just wondering how you were able to find
25 some that were not proponents of something, because they have

1 all written papers likely, and they all have views about the
2 frequency, consequences of these kinds of events. So, how
3 did you create evaluators from this subset?

4 COPPERSMITH: Well, we do have--you're right, there
5 aren't that many. We identified about 70 in this particular
6 case, and we ended up with a panel of ten. Part of the
7 charge to them, that has to be part of their overall scope,
8 is the ability to act as an evaluator, be able to put on--

9 KADAK: That's the charge. Now, how did you--out of the
10 70--

11 COPPERSMITH: The selection process is asking for their
12 colleagues, as well as them individually, if they were able
13 to play that role. So, it is part of what we ask them to do
14 and they are aware of that coming into this.

15 KADAK: And, this is a different group than the first
16 group?

17 COPPERSMITH: I'll show the two panels that we, over ten
18 years, we lost two of them, passed away, we lost two more due
19 to retirement, and we added two to replace. So, we have
20 presently eight on the current panel.

21 But, the issue of--you're right, the expert
22 selection process is very important.

23 Next? In terms of what we're covering in the PVHA,
24 I want to make it clear, if we go to the risk triplet as
25 being these three things, we're dealing with only the first

1 two. What can occur and how likely is it to occur? So, it's
2 loosely called the probability part of the igneous issue.
3 The third is the consequences, so, it would be what are the
4 consequences, given a volcanic or igneous feature intersects
5 the repository, either a dike intersecting or a conduit for
6 eruption.

7 So, what can occur is broken down typically into
8 these types of things, either intrusions, volcanic dikes that
9 would intersect the repository drifts, and, if so, those are
10 described in terms of their dimensions, geometry, complexity,
11 and so on, and eruptions, which basically is the intersection
12 of a conduit, volcanic conduit with the repository tunnels.
13 The implications there are actual eruption through the
14 repository, and deposition of volcanic ash and other
15 materials at a more distant site. So, these two things are
16 what can occur in terms of nature, either the igneous
17 eruption or the intrusion.

18 Next? And, looking at how likely, the volcanic
19 experts divide their time between spatial models that deal
20 with the relative likelihood spatially of different volcanic
21 features occurring. This would be a relative intensity of
22 future events. Obviously, it's based largely on the pattern
23 of past events, but also other information, spatial
24 information. And, secondly, temporal models that deal with
25 the likelihood and time of occurrence of igneous features.

1 Common models are homogeneous Poissonian models, but as I'll
2 talk about, there are other episodic models, models with
3 memory, time volume models also come into play.

4 In this whole process, throughout aleatory
5 variability and epistemic uncertainty are captured. The
6 aleatory variability are those things that truly vary in
7 nature and are not reducible with additional information.
8 And, of course, uncertainties that are epistemic are those
9 that are knowledge based, with additional information, would
10 be reduced.

11 Part of the epistemic uncertainty element here, are
12 alternative conceptual models about how the system works,
13 different temporal models, whether or not it's episodic or
14 not. Those, we capture and quantify and wait, and include in
15 the assessment. So, we are rather than considering
16 alternative conceptual models and choosing one, we
17 incorporate both of them into the assessment.

18 Next? So, let me talk about PVHA '96. The purpose
19 of this study was to have a probabilistic assessment of
20 volcanic hazard at Yucca Mountain, with particular emphasis
21 on the quantification of uncertainties. And, this was done
22 in the '94, '95 time frame.

23 The product was the probability distribution of the
24 annual frequency of intersection of a basaltic dike with the
25 repository footprint. That is, and that product continues to

1 be, the basis for the inputs to the TSPA and for the license
2 application.

3 Next? These are the steps and the methodology.
4 You can track them through with the guidance that I talked
5 about earlier. In addition to developing data, compilations
6 and dissemination, we also had an opportunity to go out in
7 the field a couple of times with the panel for them to have
8 first-hand observations, and to look at the features in the
9 region, in Crater Flat and some of the volcanic features
10 nearby.

11 Part of the process here is bringing together this
12 group, this diverse group, but also bringing together a
13 series of data experts, resource experts, but also
14 proponents, and we had a wonderful time listening to and
15 juxtaposing proponents who could offer their alternative
16 views of exactly the same data. But, in some cases, they are
17 different views of different datasets. For example, age
18 dating at that time of some of the features of the Lathrop
19 Wells volcano to the south of the site, was a hotly contested
20 discussion at that time. It has since subsided over the ten
21 years with additional data. But, this was an opportunity for
22 the panel to listen to and consider the pros and cons of the
23 different geochronologic techniques themselves, and the data
24 that have been gathered.

25 Next? This is the panel. It's a mix of

1 researchers, academics and some consultants.

2 Next? And, this is the result of the PVHA '96
3 assessment. This was the product. The annual frequency of
4 intersection, shown here as a probability mass function. You
5 can see the individual expert assessments are shown here,
6 with their means and medians and fit to 95th percentile,
7 entire probability distribution across the panel here, that
8 is used in the TSPA. The mean of that distribution is 1.6
9 times 10 to the minus 8. That is the annual frequency of a
10 dike intersecting the repository.

11 Now, that probability distribution then goes on to
12 its application and consequences. Given that this happens,
13 what are the effects, and so on.

14 KADAK: If you took out the highest and the lowest, what
15 would the result be?

16 COPPERSMITH: It's fairly--actually, the mean estimate
17 is fairly stable, but we are using the entire probability
18 distribution. You can see it's a good, you know, the spread
19 here is three orders of magnitude, 10 to the minus 7, or 10
20 to the minus 10, not a surprise. This is a--when you look at
21 the basic problem here, our basic problem of course in
22 modeling this area is the dearth of volcanic features.

23 For those that have done PVHA's in other parts of
24 the world, and volcanic fields that have 300, 400, over a
25 thousand centers, the uncertainties are significantly

1 reduced. The mean hazard, of course, is significantly
2 higher. So, the disadvantage of having very few data points
3 is a broader spread in distribution. But, the mean estimates
4 are lower.

5 Next? This gets to the question, why did you redo
6 it? Well, three years after the PVHA was complete in '99,
7 the USGS did an aeromagnetic survey, and some ground magnetic
8 data had also been gathered by the center and others working
9 out there, and those surveys identified anomalies that
10 existed out in the areas that were alluvium, covered by
11 alluvium, down in the Amargosa Valley, over in Jackass Flats,
12 and those anomalies were--their origin was not known. Some
13 of them clearly looked like they were dike bolts or it looked
14 like they would be magnetized bodies, presumably the salt
15 bodies at depth. Others, not so, and at that point, the DOE
16 and those of us involved in the PVHA, did an analysis of the
17 potential impact that this new data would have on the PVHA
18 '96.

19 We looked at those anomalies, and assessed whether
20 or not they would be added to the existing number of events
21 that the experts had had, looked at the implications to the
22 temporal models primarily, and a bit of the spatial models.
23 And, the bottom line conclusion in that sensitivity study is
24 they would all be a marginal effect on the mean estimate of
25 dike intersection frequency.

1 That was sent to the NRC, and the NRC did a review,
2 and they disagreed, said that basically, yes, what you've
3 done would only lead to a marginal assessment. But, in fact,
4 this information needs to be interpreted by experts, and not
5 by the Department of Energy, but by the experts themselves,
6 not so much because they would disagree with you, but in fact
7 the information may lead to alternative conceptual models
8 that you're not able to consider, it might change the review
9 of what's happening from a process point of view. And the
10 bottom line assessment is that we did not provide an adequate
11 technical basis for the conclusions that in fact this led to
12 an insignificant difference.

13 So, DOE made a regulatory commitment to complete
14 several things: a program of field studies, they go out and
15 gather additional geophysical data, to drill several of the
16 anomalies and to age date those, and look at the chemistry,
17 and to go through a process of updating the PVHA. And, that
18 process is what we are in now. And, that will, the planned
19 end of that actually is--it will occur in June of 2008, which
20 happens to be about the same time the license application
21 goes in. The PVHA '96 is supportable and defensible and will
22 continue to be the basis, but we'll have the results of this
23 study at that time.

24 Next? So, let me talk a bit--I think jump back
25 one. Oh, I'm sorry, go ahead. Let me talk a little bit

1 about the PVHA update. As I mentioned, two of the members of
2 the original '96 panel have passed away. Two other members
3 have declined continued participation. They decided to take
4 retirement seriously. And, we replaced those with two other
5 members, and went through the process of expert selection.

6 Next? Let me talk a little bit about the
7 aeromagnetic survey and drilling, because it's all part of
8 the same package and commitment to carry this out. A high
9 resolution aeromagnetic survey was carried out, with
10 resolution significantly better than the original USGS
11 survey. Seven of the anomalies were drilled to look at,
12 particularly looking at whether or not in fact they were
13 related to the salt, and if so, how deep, are they buried by
14 alluvium? The depth of burial of course can give an
15 indication of age, various types of age dating, potassium
16 Argonne. The focus on Argonne, Argonne and geochemical
17 analyses were done, and obviously all of this information
18 provides information that can be used in assessments of the
19 age of these buried anomalies, the alignment of vents, the
20 nature and geometry of the sources, and so on.

21 Next? This is just an example of the aeromagnetic,
22 the high resolution aeromagnetic survey that was done. You
23 can see the--this is the repository footprint up here, Yucca
24 Mountain. You can see the sort of distinctive nature of the
25 known sub-aerial, like Lathrop Wells Volcanic Center, Red

1 Cone, Black Cone over here. These are some of the anomalies,
2 down in Amargosa Desert, G,F and H, for example, O,N,M and L
3 over here, some anomalies up in this area. And, the drilling
4 as well as a re-examination of the existing boreholes, was
5 done really with these in mind. What are these anomalies?
6 Are there buried basalts over in Jackass Flats, which was
7 determined to be a very important assessment, and so on.

8 Next? This is a map that provides an overall
9 summary of the interpretation coming out of the aeromag
10 interpretation and the drilling, as well as age dating. And,
11 the bottom line is that the assessments, these are shown in
12 red, are quaternary volcanic centers that are known. Shown
13 in the pink are those that are Pliocene in age, typically
14 about 3.7, 3.8 million years old, and Miocene, interpreted
15 Miocene basalts are shown in green. And, we have now some
16 drill hole information that we can use for interpreting these
17 older features in Jackass Flats. More information to tell us
18 about whether or not these anomalies were in fact tuff or
19 basalt, and this is the interpretation that comes from that.

20 The other advantage of the high resolution surveys
21 is we were able to make interpretations of the faulting
22 pattern as well, and it turns out that much of the location
23 and the features related to these younger centers are related
24 to the shallow pattern of faulting. And, that is, in fact,
25 part of the assessment that is being made now.

1 Next? So, the issues that are being addressed in
2 the update are these. We talked a bit about spatial
3 evaluations before. Source zones were used previously. Now,
4 there's more emphasis on the concept of smoothing the
5 locations of past events, depending on their age.

6 Next? Temporal evaluation. These types of
7 assessments that range from simple homogeneous type models,
8 to those that have either a time aspect to them or a volume
9 aspect, or are non-homogeneous nature.

10 Next? And, the event definition part of it is in
11 fact very important as well. The intrusive event geometry,
12 what do these dikes look like, in terms of whether or not
13 they occur singly or in groups. Also, they consider the
14 potential impact of the repository opening itself on the
15 localization of intrusive events.

16 Next? In terms of extrusive event geometry, we
17 looked at the potential for conduits developing, how many,
18 what are their geometries, how large would they be, all
19 assessments that ultimately can be used in subsequent
20 modeling. And, because of the potential change, we make
21 these assessments for both the 10,000 year and 1 million year
22 time period.

23 Next? As I mentioned previously, one of the big
24 issues, or one of the big steps involved in one of these
25 assessments is developing a database, a uniform database for

1 all the experts. And, of course, over the last ten years,
2 with computational ability, GIS, and so on, our ability to
3 develop and deliver databases to the experts is much
4 enhanced, and particularly having some national labs
5 involved, like LANL, they have been able to develop some
6 wonderful maps and other layered products for the experts.
7 And, this is just one example of an isostatic gravity map
8 that's been imposed, that has--on a digital topographic base,
9 with all of the faults identified, the interpreted
10 aeromagnetic anomalies identified, and so on.

11 Next? In addition to the available information in
12 the Yucca Mountain region proper, information was also
13 developed at a number of analog sites throughout the southern
14 great basin. Information that could be useful to the experts
15 in all of these ways, in terms of helping them to understand
16 and make assessments of future events at Yucca Mountain, we
17 developed information at a number of analog locations.

18 Next? This is a listing of those analogs. We also
19 had a field trip out to many of these sites so they could see
20 first-hand the information of these analogs.

21 Next? This is an example of--one of the
22 assessments that we asked them for is an assessment of
23 conduit geometry at depth. Of course, the repository depth
24 would be about here. This is an example of an older 8.8
25 million year conduit that's been identified and actually is

1 exposed in a large cliff sequence that shows the nature and
2 geometry of the feeding conduit to this particular feature.

3 This type of analog, as well as there's many
4 others, of the type they consider in making their assessments
5 of what would happen at the Yucca Mountain area.

6 Next? This is where we are in the project. May
7 10th and 11th, Thursday and Friday of last week, we had our
8 feedback workshop, which provided back to the experts
9 information that came from their interviews, their
10 elicitations. We presented information that identified the
11 issues, the important issues that they have to deal with in
12 finalizing their assessments, which will go on over the next
13 couple of months. And, then, we'll get into final hazard
14 calculations and aggregation of the expert assessments and
15 documentation.

16 Next? I just want to show a couple of examples of
17 the types of information that we talked about last week.
18 This is an example. One of the experts has looked at and
19 divided spatially the Amargosa Desert volcanic domain, he
20 calls it, from the Yucca Mountain fault domain. Two areas
21 that are separated in terms of the tectonic manifestation of
22 extension, either accommodated through a dike injection and
23 developed in the volcanic features here, or through fault
24 displacement and extensional faulting in the faulted domain.

25 Next? This is an example of some of the

1 assessments that were made on individual characteristics,
2 dike length in this case. Dike length is an example of
3 what's almost purely an aleatory variability. They expect
4 almost all dikes--dikes of all these lengths to occur, and
5 this is the relative frequency of occurrence of those dikes.
6 The maximum dike length tends to be an uncertainty, tends to
7 be an epistemic assessment. And, it is usually treated that
8 way.

9 Next? This is another example of the number of
10 conduits that might develop along a dike, and this particular
11 expert makes it a conditional assessment on the length of the
12 dike. The longer dikes, the larger number of conduits that
13 are part of his assessment.

14 Next? This is an example of feedback. One of the
15 issues here, I won't get into the details of kernel
16 smoothing, you're probably all aware of that, but one of the
17 assessments that needs to be made, if you are going to use a
18 kernel smoother and smooth the locations of past events, is
19 you need to consider the smoothing distance, or essentially
20 the standard deviation of that kernel. And, the assessment
21 is difficult to make without looking at the potential
22 influence, and this is an example of two smoothing distances,
23 a kernel of 5 kilometers, and a larger kernel that would
24 obviously lead to a more uniform map, less topography.

25 What this says is that the past location of events

1 provides more resolving power in the future location than a
2 kernel that's longer like this. These are assessments that
3 are typically in a seismic hazard analysis as well.

4 Next? Another example of sensitivity, the expert
5 was considering two alternative conceptual models, one that
6 says that I will wait. The location of future events by the
7 inverse of their age, in other words, the younger they are
8 the more likely the future events will be near them. This is
9 another model that says that the volume of those events is
10 really a strong discriminator, and the larger events, the
11 future events will be closer to the larger. And, so, there
12 was examining the potential influence of different weights on
13 those alternatives, either a fifty-fifty weighting or a
14 seventy-five twenty-five weighting. In this case, not much
15 difference in the predicted spatial intensity.

16 Next? And, finally, this is a case where the
17 actual centers that are being used in the smoothing are
18 different. Either going to use quaternary centers only as in
19 this case, or the Pliocene only, or the combination of the
20 two. And, you can see the implications here in terms of the
21 predicted spatial intensity. This is the type of
22 information, I don't see either Bill Melson or Leon Reiter
23 here, but they were there on Thursday and Friday, and we went
24 through about two days of this type of evaluation.

25 Next? So, in summary, the methodology that's been

1 followed is consistent with our guidance, NRC guidance, as
2 well as the guidance that's been developed by other groups.
3 We're taking advantage and the PVHA update of the lessons
4 learned and the opportunities for refinement of the basic
5 methodology. But, the process, just like it was in '96, is
6 structured around workshops and expert interaction. This is
7 part of the overall elicitation process. And, we will be
8 documenting this in fiscal year 2008 during the license
9 application review.

10 Thank you.

11 GARRICK: Okay, thank you. Thank you very much. Any
12 questions, please? Yes, Andy?

13 KADAK: Kadak.

14 I'm trying to understand the significance of this
15 again. And, how much of the expert solicitation is databased
16 or subjective, in the sense that if your best estimate is an
17 event occurred 1.X number of--10 million years ago or 100
18 million years ago.

19 COPPERSMITH: The oldest events considered are usually
20 about 9 million years ago. The ones considered very closely,
21 occurred say in the last 3 million years.

22 KADAK: 3 million years, you had a couple of dikes
23 occur; is that correct?

24 COPPERSMITH: In this area, over that time period,
25 probably more like about 15, 10 to 15.

1 KADAK: 15 dikes of meters or kilometers in length; is
2 that correct?

3 COPPERSMITH: What you see are evidence of a volcano
4 itself. You see the actual cinder cone. You do not see the
5 plumbing system in terms of the actual dikes.

6 KADAK: But, these anomalies are the dikes; is that
7 right? No?

8 COPPERSMITH: No, the anomalies are actually the buried
9 cinder cone.

10 KADAK: Buried cinder cone. So, tell me how this
11 relates to the dike question?

12 COPPERSMITH: Well, all of these features presumably,
13 they're all basaltic, are fed by dikes. So, the question is,
14 in our case, we are looking at fundamentally of dikes for
15 that reason. They are the most important now. When along a
16 dike, if it localizes down to, and you watch this process
17 happen in real eruptions, if it localizes down to an eruptive
18 center, that is usually the location where the cinder cone or
19 the volcano will develop. So, dikes can be longer than the
20 localization process. We're asking for both. We're asking
21 for the dike, the location and orientation and length of
22 dikes, but also conduits. Where will they be? What is their
23 dimension?

24 Remember all this is occurring at the repository
25 depth, where understood in the hazard at the repository

1 horizon, which is say about 300 meters below the surface.
2 So, the features that would exist at that depth are either
3 going to be a dike or a conduit on a dike, and those are the
4 two features we're concerned about.

5 KADAK: So, your expectation is at some point within the
6 million years, there will be some confluence of lava activity
7 below the mountain that will erupt in and disrupt the storage
8 of this stuff.

9 COPPERSMITH: It's a hard question to answer. The
10 hazard--right now, getting back to where the--what is the
11 process right now? We know that Yucca Mountain is composed
12 of volcanic tuff. We know that there were large caldera
13 complexes to the north, and that process of large scale, you
14 know, salicic volcanism occurred many years ago, say 12
15 million years ago. As we moved into this process--and,
16 again, I'm not saying anything that the experts haven't. In
17 my real life, I work on earthquakes. I'm portraying what the
18 experts tell me. As we move into a few million years past
19 and starting at about 9 million years ago, we start seeing
20 nothing but basaltic volcanism. We started seeing lava
21 pores, and that type of thing.

22 As we moved into about 3.8 million years ago, we
23 see, and still have at the surface, some of these basaltic
24 volcanoes, and there we do see, since about 3.7 million years
25 old, we see them dissected. The cones are gone, but we see

1 some of the plumbing system and we see some of the dikes that
2 gave rise to those.

3 If we move to the quaternary, say a million year
4 centers, now we see nothing but the constructional features.
5 We see the cinder cones, and so on, that are out in Crater
6 Flat and are down at Lathrop Wells. So, the process of
7 tectonically what's happened is we move from large scale
8 salicic volcanism to smaller scale basaltic volcanism. And,
9 that change not only in its, you know, geochemical nature,
10 but in volume, and so on, is being used by many of the
11 experts in their temporary modeling. They have time volume
12 models that take into account what happens over time.

13 KADAK: So, the answer to my question is?

14 COPPERSMITH: Okay, so what can happen at the repository
15 in general will be basaltic volcanism, which basaltic would
16 look like the types of volcanoes that we see in Crater Flat,
17 and--

18 KADAK: The little cones--or, actually, big cones
19 probably.

20 COPPERSMITH: Right. The little cones. I mean, the
21 volumes of these per event, just looking at quaternary, are
22 very small geologically.

23 KADAK: I mean, is this something that we really should
24 be concerned about at Yucca Mountain, I guess is what--

25 COPPERSMITH: I think we have to be. The information is

1 such that it could have a large consequence, and the
2 probabilities are high enough that it's something that needs
3 to be considered.

4 KADAK: Okay.

5 GARRICK: Ali, and then Bill.

6 MOSLEH: Mosleh, Board.

7 I'm trying to understand this frequency graph that
8 you have on Slide 11. Is this--the numbers coming from the
9 experts, do they represent some sort of calculation,
10 computation, and, therefore, a compound event?

11 COPPERSMITH: This is a single event. These are
12 definitely--

13 MOSLEH: Single.

14 COPPERSMITH: The numbers that you see here, the
15 distribution and again these are 5th to 95th fractals or
16 percentiles, they're entire distribution take into account
17 their spatial and temporal models--

18 MOSLEH: Oh, so it's an aggregate of--

19 COPPERSMITH: Yes.

20 MOSLEH: So, a related then question is that is this, I
21 think from what you were saying before, there is such a
22 notion that we have in the seismic hazard, it's frequency
23 versus magnitude.

24 COPPERSMITH: Exactly.

25 MOSLEH: And, this represents--

1 COPPERSMITH: This is not the same. The temporal model
2 would be the frequency, magnitude analogy to PSH.

3 MOSLEH: Okay.

4 COPPERSMITH: PSHA then goes to the next step saying
5 given that an earthquake happens, where does it happen. How
6 far away is it, and the result is then a hazard curve, which
7 is the frequency of exceeding ground motion levels. Okay,
8 this goes to that next step also. The spatial model says the
9 relative likelihood of it occurring at different locations,
10 given that a volcano occurs, so it's a conditional spatial
11 intensity, it combines that with the temporary model in an
12 absolute sense says how many occur, what is the rate. So,
13 the convolution of the two is what you have here.

14 MOSLEH: Yes.

15 COPPERSMITH: Actually, I've got to say there is a
16 third, which is given that an event occurs somewhere, it has
17 to have dimensions, it has to be close enough that the
18 dimensions of the dike would intersect the repository,
19 because this is intersection frequency.

20 MOSLEH: That's right. So, compounding events.

21 COPPERSMITH: So, roughly, that third part is like
22 attenuation of law in PSHA.

23 MOSLEH: Okay. So, given that, how do you go from the
24 ranges provided by the expert to do that aggregate
25 distribution? Is that--

1 COPPERSMITH: We have an explicit, or goal at the
2 beginning of this study that we will be in a position to
3 defend equal weights. The SSHAC guidance spends a lot of
4 time on aggregation methodology.

5 MOSLEH: And, you follow--

6 COPPERSMITH: We have people, George (inaudible), Peter
7 Morris, and others on the panel who spent a lot of time on
8 what we call integration in that study, and the goal from the
9 beginning is to provide a basis for being able to defend
10 equal weights. So, the distribution of a common database,
11 exposure to all of the same series of proponents and
12 advocates, a process of training so that they understand the
13 issues related with probability and coding, and so on, in
14 other words, we--and, the fact they're acting as evaluators,
15 that whole process leads to ultimately our decision to
16 equally weight. We leave open, and SSHAC guidance, the issue
17 is left open then in fact you may be in a position where you
18 need to provide differential weights, and we leave open that
19 option. But, in fact, in this case, we have provided,
20 combined them through a process of equal weighting.

21 MOSLEH: So, it looks like you have, you know, a set of
22 ranges from the experts, and then you end up with the
23 frequency distribution that you have--

24 COPPERSMITH: Actually, they're not ranges. They are a
25 series of distributions, just like--

1 MOSLEH: Oh, is that right? Okay. Okay, And, then,
2 you sample from them to generate--

3 COPPERSMITH: Exactly. It's the combination of the two
4 with equal weights. The ten in this case.

5 MURPHY: Bill Murphy, Board.

6 This is really interesting, and I was paying close
7 attention to Chuck Connor's work about the time the first
8 elicitation was done, and so I'm real curious about what's
9 changed, and tell me if my impression is right. The new
10 volcanoes that have been identified are all in the northern
11 Amargosa Desert in the old Miocene basalts out in Jackass
12 Flat, and ten years ago, the problem was that there was an
13 extremely steep gradient in the probability across the
14 repository horizon, because there were volcanoes in Crater
15 Flat, and none to the east. And, so, it was very hard to pin
16 the probability for the repository because it was on that
17 very steep probability gradient.

18 COPPERSMITH: Well, if you notice the plots that I
19 showed for sensitivity, we're still on a gradient.

20 MURPHY: I know, and plots like Slide 29.

21 COPPERSMITH: Right.

22 MURPHY: And, that makes me wonder when I see plots like
23 Slide 29 and the locations of the triangles, which I guess--

24 COPPERSMITH: Those are the--

25 MURPHY: I don't see them for the Miocene basalts in

1 Crater Flat.

2 COPPERSMITH: The most important part of the new data
3 collection part, with the analysis of the aeromag as well as
4 the drilling information, is the absence of Pliocene or
5 quaternary centers in Jackass Flat. That was postulated at
6 the time of '96, it was discussed, and perhaps that would be
7 something that could be there. Everything has been done
8 since shows a factor of--what's over there on Miocene, and in
9 most cases, you can see the distribution of Miocene events.
10 Their volumes are much larger. In most cases, the experts
11 give much lower weight to the spatial distribution of Miocene
12 features in predictions of the future.

13 MURPHY: Okay, thank you.

14 GARRICK: George?

15 HORNBERGER: Kevin, just to follow up on that. So, my
16 understanding is that the investigation or the aeromag and
17 the drilling has not produced any surprises that might lead
18 someone like me to expect that there was going to be a
19 tremendous change in the PVHA?

20 COPPERSMITH: That's right. But, if the next question
21 is what will the answer--

22 HORNBERGER: No, I know better than to ask that. I know
23 better than to ask it. I just--

24 COPPERSMITH: I'm asked that daily. So, I thought maybe
25 you--

1 HORNBERGER: No, I phrased my question--

2 COPPERSMITH: I think in terms of the new data, to me,
3 the strongest impact from what I am seeing in the elicitation
4 we've done so far is the lowering of the number of undetected
5 events, in other words, the potential for events that exist
6 up there, but we just don't see any case. That's simply due
7 to higher resolution, period, either drilling, aeromag,
8 mapping. The other part is I think a better and maybe more
9 sophisticated modeling of temporary aspects, homogeneous
10 Poisson processes were believed to be a default last time.
11 People on the panel now that have studied around the world
12 say we see much evidence for temporally clustered activity.
13 And, when they take those models here, they say hey,
14 potentially we see the same thing at about a million years,
15 about 3.7. You know, that type of modeling is much more
16 sophisticated than it was a decade ago. Now, the effect of
17 that, I don't know.

18 HORNBERGER: The other question I have there, sort of
19 falling on what Andy was trying to get at. We have a lot of
20 tunnels at Yucca Mountain, and they've been mapped, and all
21 the faults. How many dikes have been counted?

22 COPPERSMITH: Right now, there are no dikes that have
23 been encountered.

24 HORNBERGER: And, how does that enter into estimating
25 the probability of a dike intersecting?

1 COPPERSMITH: Well, the youngest and closest dike that
2 has been mapped is the Solitario Canyon dike, which sits to
3 the north. I don't know, one of those--go to the next slide.

4 HORNBERGER: And, what age is Solitario Canyon?

5 COPPERSMITH: That doesn't have it. It's about 10
6 million years old, it's arguably between 10 and 11, depending
7 on which lab. But, that is a--that was located up about
8 right here, and that of course is subject to a lot of
9 discussion, and was part of our field trip. And, the bottom
10 line, though, is that features that old are rarely used to
11 make assessments of future distribution of igneous features,

12 As you go back, literally as you go back, the
13 processes change. And, so, it's considered by all on the
14 panel, but it has very low weight in terms of spatial
15 distribution in the future.

16 GARRICK: Well, let me ask a question as a practitioner
17 actually. I want to know if you have observed the same thing
18 that we observed many years ago before this process was
19 formalized as much as it is now, and that is that we
20 discovered that this business of trying to calibrate the
21 expert, and this goes to Andy's question as well, is not
22 nearly as important as understanding the evidence supporting
23 the expert's opinion. As you expose the expert's evidence
24 and supporting database, or as the various experts begin to
25 converge on the same database, there was a convergence of

1 opinions and this tends to confirm the Bayesian concept that
2 given the same evidence, we're all basically wired the same,
3 and, therefore, we will assign the same probabilities. Did
4 you observe this kind of phenomenon?

5 COPPERSMITH: Well, number one, the criteria of both
6 that Jack has and the branch technical position has for when
7 you carry out these types of elicitations, has to do with
8 large uncertainties that are very significant to start with,
9 and that are not amenable to reduction, significant reduction
10 with new data collection. And, this is your classic case.
11 We obviously can gather information related to location and
12 nature of past events, but can do very little more than that.
13 So, uncertainties are significant in this case. They have
14 been shown to be potentially significant in the TSPA as well.

15 The process of convergence, SSHAC spends a lot of
16 time on consensus. And, of course, we do not push for that
17 process. But, what we do see is in the course of
18 consideration of proponent views, for example, and the
19 discussion that follows, we see quite a bit of convergence in
20 terms of unintended differences in interpretation, in
21 definition. Ultimately, when we get down to exactly their
22 assessments of what event types we're dealing with and the
23 types of models they'll implement, there's still a good bit
24 of divergence and their view of the world in terms of what is
25 happening out here, temporally and spatially. And, I think

1 that is reflected in this three orders of magnitude in this
2 intersection frequency.

3 GARRICK: But, the idea is not to push for convergence.

4 COPPERSMITH: Right.

5 GARRICK: The idea is to come to an understanding of
6 what the basis of their technical--

7 COPPERSMITH: I want to go back to that point, because I
8 just had an argument in Switzerland a few months ago with a
9 fellow who likes strong differential weights on the basis of
10 calibration tests of all types. And, I take strong issue
11 with that. In fact, in SSHAC, we spent about a year on this
12 issue with those on the panel who were experts in this
13 particular area.

14 There is a large burden that's associated with the
15 process of education, of data dissemination, of interaction
16 that goes along with this type of process. It's expensive
17 and it takes a long time. If we are able to bring in
18 experts, give them a questionnaire and get scores on almanac
19 type questions that we could then use in the aggregation
20 scheme, with strong differential weights, it would be an easy
21 life. But, it isn't the way this process works.

22 In fact, the discussion and interaction process is
23 just as important as any other mathematical aggregation
24 scheme. So, we spent a lot of time on what's called the
25 behavioral aggregation, and, in fact, they will work out and

1 argue some of the unintended differences in the process.

2 GARRICK: Okay, Andy? How is your brain wired?

3 KADAK: My brain is wired. Slide 11 again, I think just
4 as a follow-up to this, it appears to me that I don't see
5 much of the science of volcanism coming out--

6 COPPERSMITH: You don't see any science in a plot like
7 this. This is a calculated result.

8 KADAK: Okay. Now, you said you were a--now, for me, it
9 would be really good to understand how these things actually
10 occur. And, do we see any evidence of it occurring in this
11 area?

12 COPPERSMITH: Let me just show--give you an idea what a
13 typical model looks like for one expert.

14 KADAK: Okay.

15 COPPERSMITH: And, I can't get all the complexities for
16 those--anyone here who was there last week, and have some
17 sense of the nature of these models. Let's start out with a
18 description of the spatial distribution of future events.
19 Well, how do you do that? Well, they decide first which past
20 events they're going to use. So, they study and provide
21 tables and evaluations of the location and age and use of
22 past events, quaternary events, the Pliocene events, or they
23 consider the Miocene events. That provides the basic
24 database, looking back and now, I'm going to need that to
25 look forward.

1 The spatial distribution of future events is what
2 we care about, not the past. They use that past in different
3 ways. One is to say it will occur uniformly over some zone.
4 What is that zone, what's the tectonic basis for that zone,
5 or other bases? In some cases, they will say this is
6 tectonically part of this same trough, it responds
7 structurally in the same way. Others will say this is part
8 of an isotopic zone that tells me--its isotopic signature
9 tells me about what's going on at source depths, depths for
10 the user of 60 to 80 kilometers below the surface. Others
11 will have different reasons for their zonation.

12 They could then within those zones, talk about the
13 spatial distribution. Is it uniform? Is it non-uniform?
14 Will it follow the location of past events? They can have
15 all those. If it's non-uniform, they can then define a
16 spatial smoothing operator that tells you how uniform or non-
17 uniform it will be. They can describe that distribution.

18 Those all go into the relevant spatial intensity of
19 future events. I haven't even talked about their likelihood.
20 So, the temporal evaluation tells us something about that.
21 So, they sit down and say okay, is it uniform over time?
22 Well, now, they need to study past events, study other
23 locations around the world where they've looked, and apply
24 different models, either a Poissonian model that they have
25 their events defined, or we can have those uniform over

1 different time frames. They can have non-uniform, episodic
2 models that accelerate for a period of time, go for a long-
3 term Poissonian rate, and accelerate.

4 KADAK: But, there's no data.

5 COPPERSMITH: Yes, they have data.

6 KADAK: To support these things?

7 COPPERSMITH: They have data. They started out with the
8 events in the region.

9 KADAK: Like a million years ago, they knew how these
10 things were working on?

11 COPPERSMITH: These events that are defined over in many
12 cases the Pliocene and quaternary, provide a basis for that
13 assessment. Our chief scientist won't--

14 KADAK: Well, let me finish to the point. You do
15 seismic hazard analysis; right?

16 COPPERSMITH: Yes.

17 KADAK: Are you comfortable with numbers, 10 to the
18 minus 8, 10 to the minus 9, 10 to the minus 11 for some
19 event, like a seismic event?

20 COPPERSMITH: Right now, seismic events are different.
21 The seismic--the maximum earthquakes on faults occur much
22 more frequently.

23 KADAK: Right. And, that's where your comfort zone is.
24 But, are you comfortable with numbers of that order of
25 magnitude?

1 COPPERSMITH: Yes, because these are not directly
2 assessed. These are the product of a series of assessments.
3 For example, the average recurrence rate for volcanism in
4 this area is about 3 to 500,000 years. That's the average
5 rate, about a half a million years. So--

6 KADAK: And, where was the last one again?

7 COPPERSMITH: The last one was about 80,000 years ago.

8 KADAK: Where?

9 COPPERSMITH: Down at Lathrop Wells.

10 KADAK: And, there was like a--

11 COPPERSMITH: Yes, volcano.

12 KADAK: --Discovery Channel volcano?

13 COPPERSMITH: They are assumed, because all the cinder
14 is being trucked away as we speak. So, the recurrence part
15 of this by its nature the average recurrence rate is hundreds
16 of thousands of years to start with. Now, those are in the
17 more active areas. And, now, we have to deal with the
18 probability of the fact those areas, or features there are
19 occurring up closer to the mountain, and the probabilities
20 get lower still.

21 SWIFT: This is Peter Swift from Sandia.

22 Do we have Slide 15? No, 16, one more. The
23 reddish orange dots on here, Yucca Mountain is the yellow
24 area, and the reddish orange dots, and I see three of them,
25 one, and then--yeah, you had three of them, there's a big one

1 down there. There's a little cone up at the north end, that
2 one up there, too. Those are the actual cinder cone
3 volcanoes you see at the land surface from the crest. Those
4 are, unarguably, real volcanoes. The one down at the bottom,
5 Lathrop Wells, that one there, that one is about 80,000 years
6 old. And, the simplest possible model, geologic model for
7 recurrence of volcanoes, would be to draw a circle with Yucca
8 Mountain at the center, and capture those three or four red
9 dots, determine their age, and you get an aerial frequency.
10 That would be pretty much an uninformed, but we'd be well
11 within the range with what the experts came up with, but it
12 will be an uninformed model. So, we asked volcanology
13 experts to do a better job of drawing the circle and counting
14 the volcanoes inside it basically.

15 COPPERSMITH: Well put. Some of the others, just while
16 this is up, this is before--these anomalies here, for
17 example, showed up in the aeromagnetic survey. This one was
18 drilled and found to be basalt, 3.8 million years old. These
19 are model anomalies at the same depth, and interpreted to
20 have the same age. So, now, we have information on these we
21 can use. Likewise, this anomaly has been drilled. It's
22 again, another Pliocene basalt. So, these are the events in
23 the region that are used, they are past events, but they are
24 used to develop spatial models for future. It's an extremely
25 low probability.

1 SWIFT: I agree, but I think the components, it's easy
2 to show how the components lead to that, derived probability,
3 and the largest component that drops the probability the most
4 is the long recurrence interval. Typical recurrence
5 interval, I don't know if we have anyone here from the
6 Armenian plant. We worked on a plant in Armenia, and they
7 have, you know, recurrence intervals that are on the order of
8 several hundred years, maybe a thousand.

9 GARRICK: Okay, any other questions? Board? Staff?
10 Audience? Going, going--okay, well, thank you very much.
11 And, I also want to--pardon? Did somebody want a question?

12 FITZPATRICK: If you don't mind, I'll be real quick.
13 Charles Fitzpatrick from the State of Nevada.

14 I was just wondering with an eight person panel,
15 and six of them with the same--that were on the '96 panel,
16 the first one took from '94 to '96, and this one has taken
17 from 2004 to 2008. And, it looks like the expert's work will
18 be done in July of '07. Couldn't it be ready sooner than a
19 year from now?

20 COPPERSMITH: The short answer is no. The reason it
21 took longer is all the data collection, the drilling,
22 aeromatic, and so on, that were done this time as opposed to
23 the last time. No, we have every code that's used has to be
24 qualified. We have to go through a process to button this
25 all up so that it's completely QA.

1 FITZPATRICK: The second quick question was I believe I
2 read in the first one in '96, that the experts were strictly
3 limited to a 10,000 year window because of the EPA standard
4 at that time. Was any such limit provided in this exercise
5 as far as the scope?

6 COPPERSMITH: We are doing both 10,000 and 1 million
7 years on this one.

8 GARRICK: Okay, thank you. Thank you very much.

9 I also want to thank all of the presenters today.
10 We did an excellent job of conforming to the 50 percent rule,
11 50 percent briefing time, and 50 percent question time. And,
12 we got done on time, and we appreciate that very much.

13 And, unless somebody has an announcement or wants
14 to make a comment, I think we are in a position to adjourn
15 this meeting.

16 Not hearing any, we are adjourned. Thank you.

17 (Whereupon, at 5:43 p.m., the meeting was
18 adjourned.)

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C E R T I F I C A T E

I certify that the foregoing is a correct transcript of the Spring Board Meeting of the Nuclear Waste Technical Review Board held on May 15, 2007 in Arlington, Virginia taken from the electronic recording of proceedings in the above-entitled matter.

May 28, 2007

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