

200 - 600 nm



# Evaluation of the Control of Reactivity and Longevity of Nano Scale Colloids by the Method of Colloid Manufacture

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#### **Current Practical Manufacturing Methods**

- ■Bottom Up Precipitation
  - Dominant Known Technology is Sodium Borohydride Reduction
  - Others?
- ■Bottom Down Attrition via Ball Milling
- Bottom Down Iron Oxides then Hydrothermal Reduction

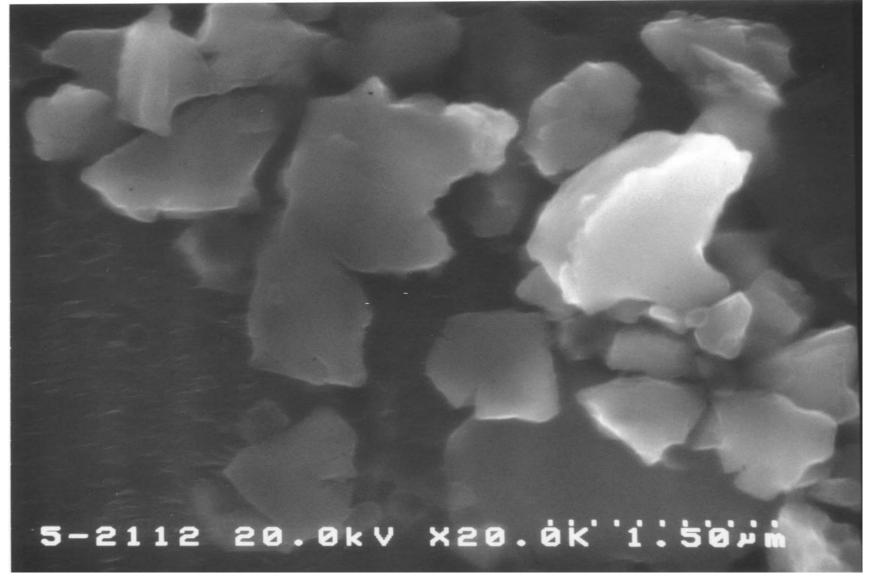


#### What Defines a Practical Manufacturing Technology

- Cost
- Capacity to be produced in ton lots
- Capacity to be produced in relatively short time frames

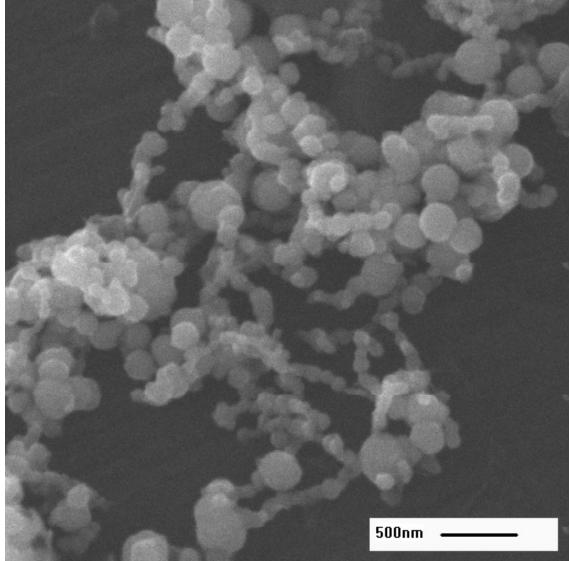


#### Top Down Nano-Scale Fe Colloids



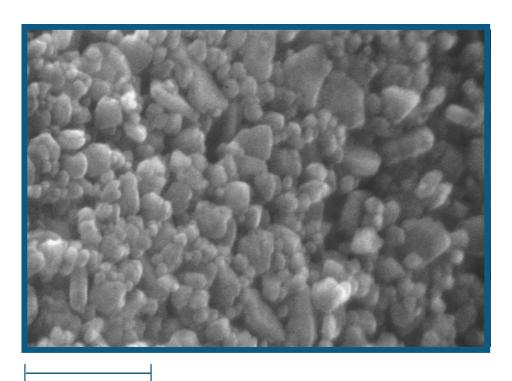


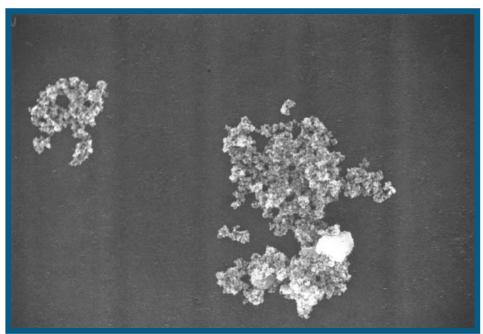
Bottom Up Nano Scale Fe Colloids (ARCADIS)





#### Another Bottom Up Iron





120 nm

10 µm



#### Key Performance Issues

- Transport and Delivery
  - Size the balance between gravitational settling
     and attractive forces 200 to 600 nm ideal
- Colloid Longevity
  - Passivation by dissolved inorganics in Water
  - Unproductive hydrogen generation
  - Kinetic Response

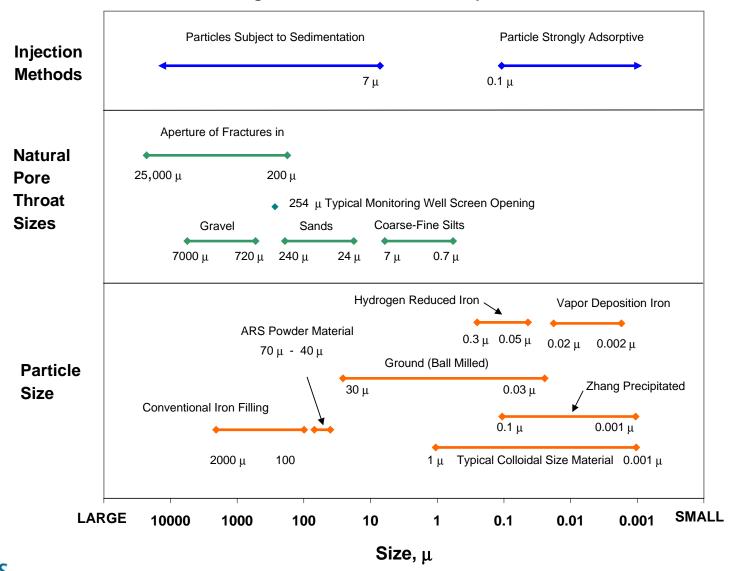


#### **Delivery**



#### Nanoscale Iron Particle Size Comparison

#### Size Ranges of Zero Valent Iron Compared to Pore Slot Size

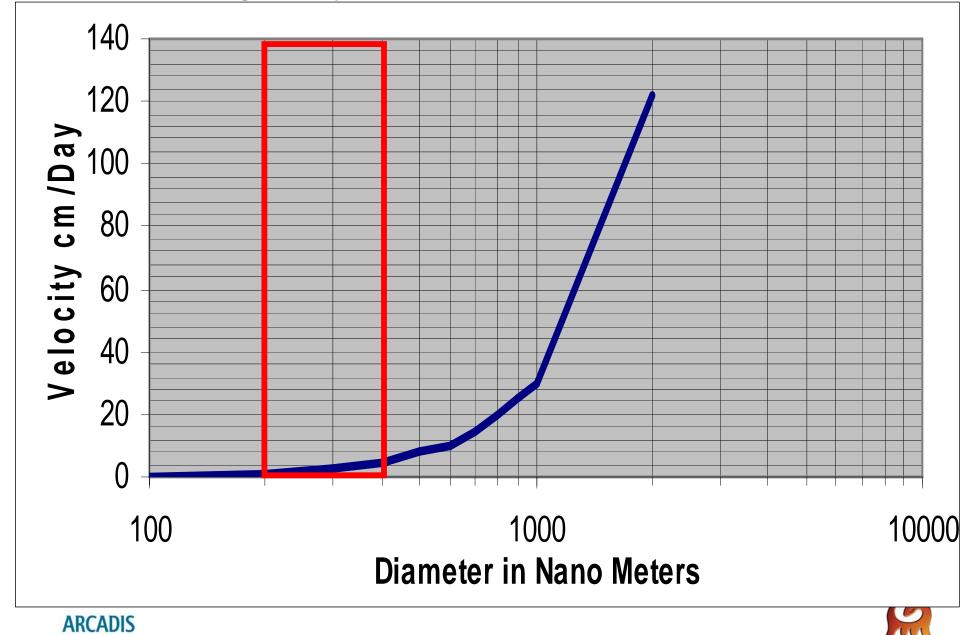


#### Point of Zero Charge - pH<sub>pznpc</sub> Binding or Dissociation of Protons

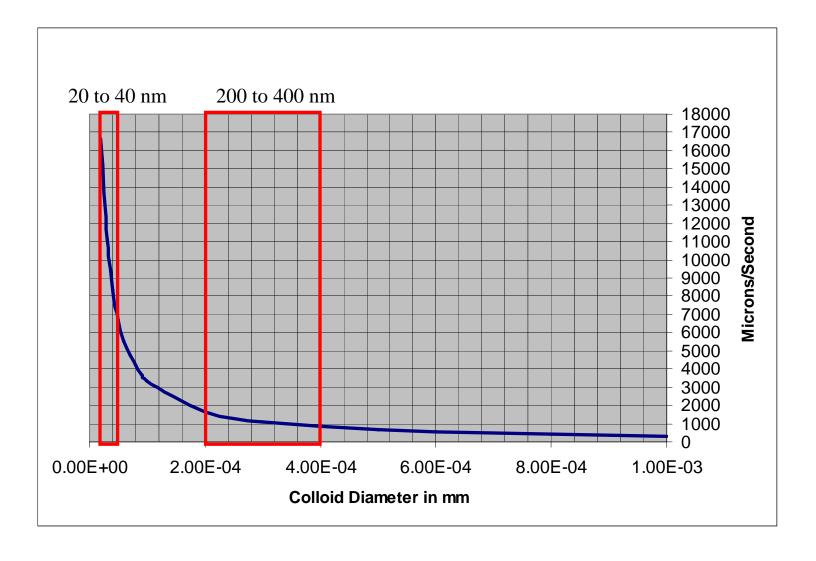
• $\alpha$ -Al <sub>2</sub> O <sub>3</sub>	9.1	• $\delta$ -MnO <sub>2</sub>	2.8
• $\alpha$ -Al(OH) <sub>3</sub>	5.0	• $\beta$ -MnO <sub>2</sub>	7.2
<ul><li>γ-AlOOH</li></ul>	8.2	• SiO <sub>2</sub>	2.0
<ul><li>CuO</li></ul>	9.5	<ul><li>ZrSiO<sub>4</sub></li></ul>	5
• $\alpha$ -Fe <sub>3</sub> O <sub>4</sub>	6.5	<ul><li>Feldspars</li></ul>	2-2.4
• α-FeOOH	7.8	<ul><li>Kaolinite</li></ul>	4.6
• $Fe_2O_3$	8.5	<ul><li>Montmorillonite</li></ul>	2.5
• Fe(OH) <sub>3</sub> (amorph)	8.5	<ul><li>Albite</li></ul>	2.0
<ul><li>MgO</li></ul>	12.4	<ul><li>Chrysotile</li></ul>	>10



#### Stokes Settling Velocity Vs. Fe Colloid Diameter



#### Colloid Velocity Due to Brownian Motion





#### Colloid Reactivity and Longevity



#### Environmental Impacts on ZVI Longevity

- Effect of high TDS
  - Sulfate and Soluble Carbonates
- Effect of water dissociation
- Effect of CVOC reactions



#### Intrinsic Controls on Colloid Longevity

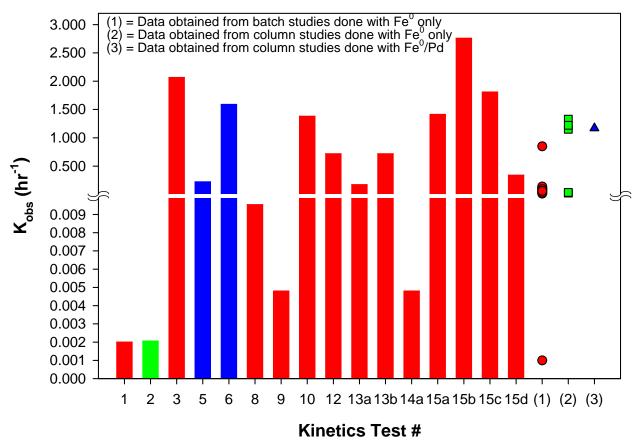
- Colloid structure
  - Particle morphology shape, pits
  - Particle crystal structure size of crystal domains, kinks, amorphous zones
- Control composition Secondary constituents in colloids
  - Catalysts
  - Manufacturing byproducts
- Modification of the colloid surface
  - Catalysts
  - Inorganic inhibitors
  - Polymers



#### **Kinetics Batch Test Results**

#### (higher values indicate short half-lives)

#### Experimental Data and Data from Literature

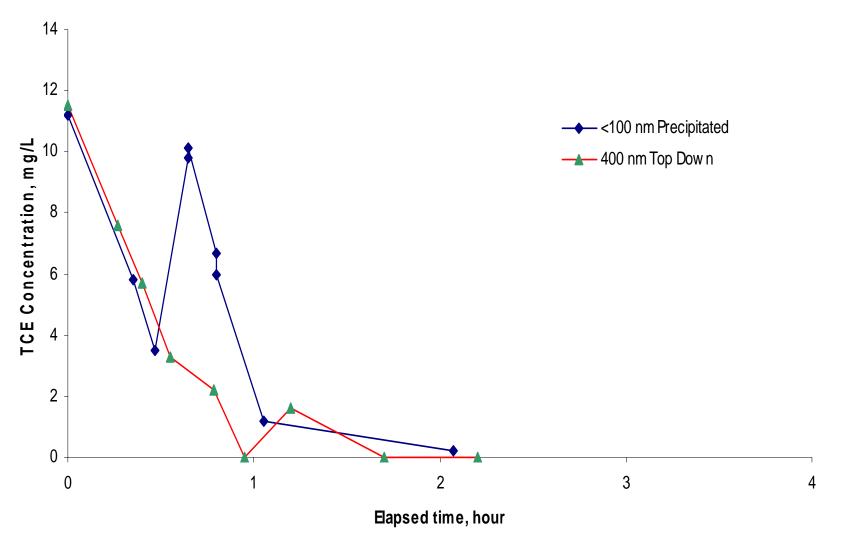


Red Bars are Vendor C Ball Milled

Green Bar is Vendor A Precipitated Blue Bars are Vendor B Precipitated

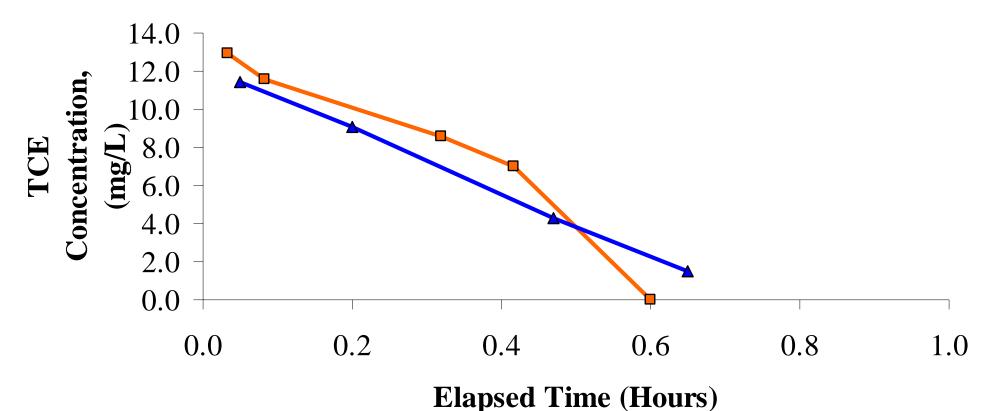


#### Reactivity of Top Down Vs. Bottom Up Colloids





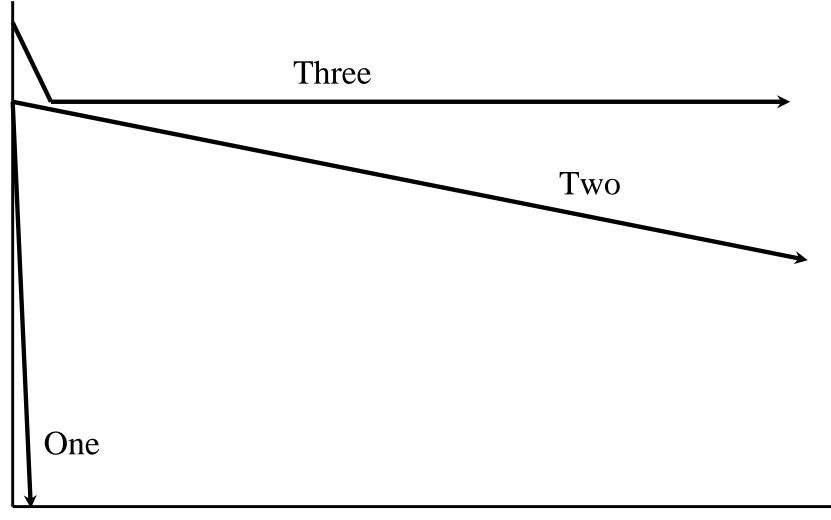
# BNI TCE Dechlorination Kinetics Stability of Ball Milled Colloids



**─** 7/10/02 BNI Batch **→** 10/17/02 BNI Batch



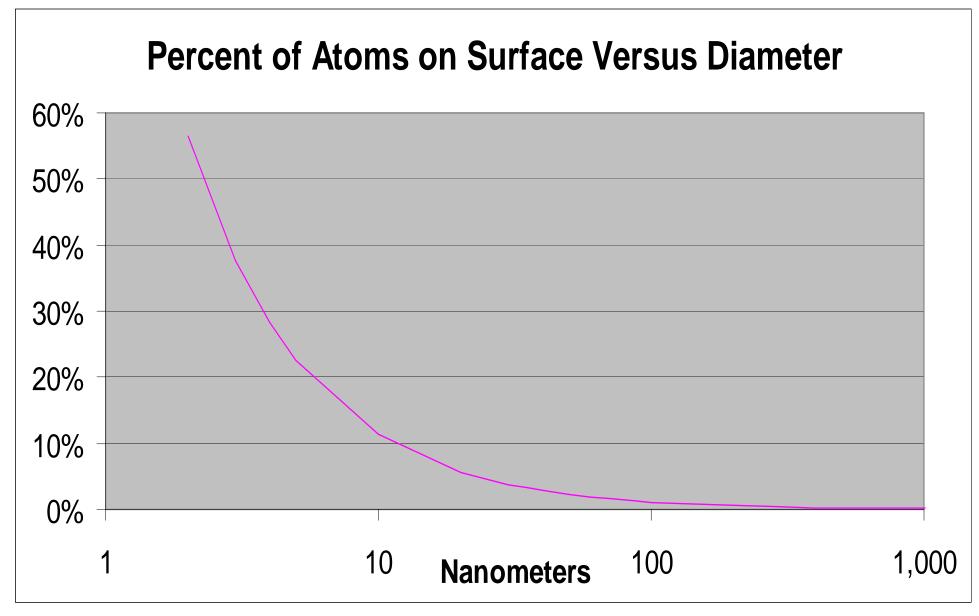
#### Three Classes of Colloid Reaction



#### What Causes Type Three Behavior?

- Oxidation during shipment or handling, surface coatings
  - Acid treatment does not remove effect
- The presence of by products from the manufacturing process that interfere with electron transfer
  - Borohydride leaves % concentrations of boron in the colloid
- Structural changes
  - Annealing or Ostwald Ripening
- Palladization restores reactivity





#### Variations in Iron Colloid Response

- Class One
  - Typical response from all early product runs
- Class Two
  - Effect due to size and chemical make up or structure colloids from 100 nm to 2 Microns
  - Becomes class one with palladization
- Class Three
  - Acid pretreatment has no effect
  - Repeated testing by independent labs as well
  - Becomes class one with palladization



## The Good News Type One and Type Two Each Have a Valuable Niche

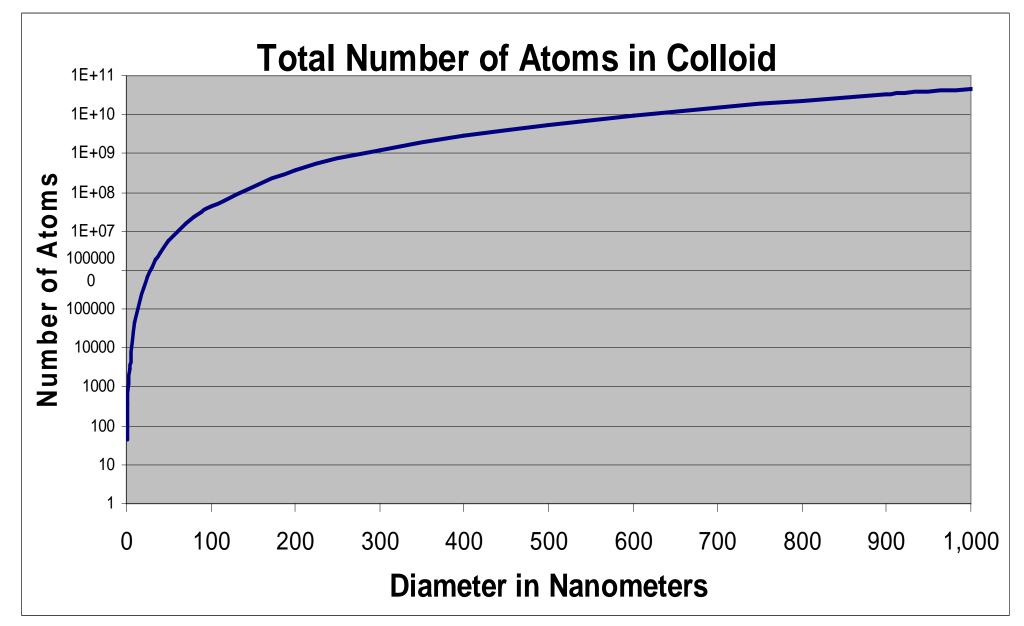
- Type one colloids are of value for treatment of DNAPL or high concentrations of adsorbed CVOC
  - Think of reductive version of chemical oxidation
  - The "Champaign effect" is observed with the most extreme examples
- Type two colloids are of value for use in reactive walls for the long term treatment dissolved CVOCs under natural flow conditions



### A New Technology with Unique Potential Problems

- We understand how to manipulate isolated molecular systems, chemical oxidation for example
- Efficient bacterial enzymatic pathways have been developed over several billion years
- Nano scale colloids are large assemblages of molecules subject to atomic forces with complex structure and a behavior that is in the process of definition







"Reality is that which, when you stop believing in it, does not go away"

The Bottom Line

Make Assumptions at Your Own

Risk