



NANOTECHNOLOGY CHARACTERIZATION LABORATORY

Prepared for C-Sixty Inc., a subsidiary of
Carbon Nanotechnologies, Inc.

Functionalized Fullerenes
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Executive Summary

The objective of the C-Sixty, Inc. - NCL collaboration is to aid C-Sixty, Inc. in identifying the most promising candidate from a series of fullerene based antioxidants, and to investigate potential applications related to cancer therapy. Water-soluble, functionalized fullerenes have been shown to protect against a broad range of oxidative injuries, including organ reperfusion injury and ionizing radiation. The antioxidant candidates submitted for testing were AF1 (NCL16), AF3 (NCL17), C3 (NCL19), DF1 (NCL42), and DF1-mini (NCL45) (see Figure 1). NCL studies conducted on these candidates can be divided into physico-chemical characterization, *in vitro* toxicology, *in vitro* chemoprotection studies, and *in vitro* immunology. Of the fullerene derivatives, DF1 and DF1-mini were selected for more thorough characterization due to their therapeutic potential.

Physico-Chemical Characterization

Physico-chemical characterization of these samples included Dynamic Light Scattering (DLS) to measure the hydrodynamic size, Reversed Phase High Performance Liquid Chromatography (RP HPLC) and Capillary Electrophoresis (CE) for purity, and Matrix Assisted Laser Desorption/Ionization-Time of Flight Mass Spectrometry (MALDI-TOF MS) for molecular weight and purity. In addition, the zeta potential was also measured for DF1. Unlike the parent compound (C60), which is hydrophobic and insoluble in aqueous buffers, the derivatized fullerenes have a variable degree of aqueous solubility that increases their potential for biological applications. Since there is no currently available method for the detection and quantitation of the fullerene derivatives in matrix, a capillary electrophoresis approach was developed for the analysis of C3 and DF1 in both standard solutions and serum matrix.

Hydrodynamic sizes for these materials, measured by batch-mode DLS, were 4.8 nm for DF1, and 4.0 nm for DF1-mini. Molecular weight analysis by MALDI-TOF Mass spectrometry was in agreement with the proposed structures. Purity as assessed by RP-HPLC and MALDI-TOF MS shows that while both DF1 and DF1-mini contain impurities, the latter (DF1-mini) seems to be relatively purer than its next generation derivative (DF1).

In Vitro Toxicology

The biocompatibility of the functionalized fullerenes was evaluated in porcine renal proximal tubule (LLC-PK1) and human leukemia (HL60) cell lines. AF1 and C3 functionalized fullerenes were found to be minimally cytotoxic, while AF3 and DF1 were nontoxic to LLC-PK1 cells. In HL-60 cells, both DF1 and C3 were found to be cytotoxic. Cell cycle analysis demonstrated that DF1 primarily targeted HL60 cells in the G1 phase. Treatment of renal proximal tubule (LLC-PK1) cells and Sprague-Dawley primary hepatocytes with C3 or AF1 increased levels of reactive oxygen species (ROS), while DF1 decreased ROS levels dose dependently. The observation that DF1 is selective in its cytotoxicity, in that it is toxic to the human leukemia cell line (HL-60), but not the porcine renal proximal tubule cell line, warrants further investigation. Based on this biocompatibility assessment, DF1 was chosen as a promising candidate for therapeutic applications and is therefore the focus of the cisplatin chemoprotection studies which follow.

***In Vitro* Immunological Characterization**

This is a study of the effects of fullerene derivatives on the integrity of red blood cells, platelet aggregation, and formation of granulocyte-macrophage colonies (CFU-GM).. AF1 and AF3 demonstrated hemolytic properties on human erythrocytes, while C3 and DF1 did not affect the integrity of erythrocytes. None of the tested fullerene derivatives induced platelet aggregation, however C3 and DF1 suppressed aggregation of platelets induced by collagen. None of the tested fullerene derivatives were myelosuppressive or capable of preventing cisplatin-induced toxicity to bone marrow haematopoietic stem cells.

Nanoparticle Descriptions

The fullerenes are a family of carbon allotropes named after Richard Buckminster Fuller. A dendrofullerene is a fullerene derivative with an attached branched architecture. An amphiphilic fullerene is a fullerene derivative with an attached hydrophilic moiety to render solubility both in aqueous and organic solvents. The fullerene derivative samples (Figure 1) submitted to the NCL by C-Sixty, Inc. are:

NCL16: An amphiphilic fullerene C60 derivative commonly called AF1.

NCL17: An amphiphilic fullerene C60 derivative commonly called AF3.

NCL19: A tris-malonic acid C60 derivative commonly called C3.

NCL42: A dendrofullerene C60 derivative with branched architecture and 18 carboxylic acid terminal groups, commonly called DF1.

NCL45: A dendrofullerene derivative with 6 carboxylic acid terminal groups, referred to here as DF1-mini.

The purpose of this study was to determine a lead antioxidant candidate. Samples which showed the most therapeutic promise received the most extensive characterization.

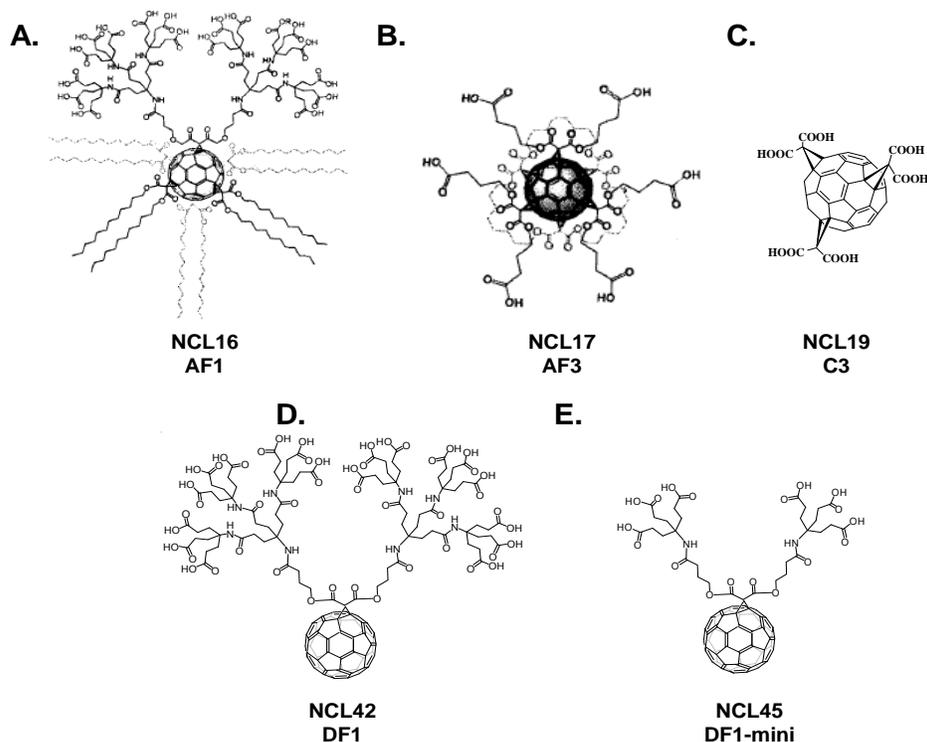


Figure 1. Theoretical molecular structures of the C-Sixty fullerene derivatives and their NCL number designations.

Physico-Chemical Characterization

Section Summary

The main purpose of this collaboration was to identify the most promising candidate from a series of fullerene based antioxidants. Based on preliminary discussions with C-Sixty Inc., it was decided to focus on DF1 and DF1-mini as the leading candidates. Physico-chemical characterization of these two samples included Dynamic Light Scattering (DLS) to measure the hydrodynamic size, Reversed Phase High Performance Liquid Chromatography (RP HPLC) and Capillary Electrophoresis (CE) for purity and Matrix Assisted Laser Desorption/Ionization-Time of Flight Mass Spectrometry (MALDI-TOF MS) for molecular weight and purity. In addition, the zeta potential was also measured for DF1. Unlike the parent compound (C60), which is hydrophobic and insoluble in aqueous buffers, the derivatized fullerenes have variable degree of aqueous solubility that increases their potential for biological applications. Since there is no current available method for the detection and quantitation of the fullerene derivatives in matrix, a capillary electrophoresis approach for the analysis of C3 and DF1 in both standard solutions and a serum matrix was developed. The quantitation of the particles was linear from 0-500 µg/mL with quantitation limits ranging from 0.6 to 6 µg/mL.

Hydrodynamic sizes for these materials, measured by batch-mode DLS, were 4.8 nm for DF1, and 4.0 nm for DF1-mini. Molecular weight analysis by MALDI-TOF Mass spectrometry was in agreement with the proposed structures. Purity as assessed by RP-HPLC and MALDI-TOF MS shows that while both DF1 and DF1-mini contain impurities, the latter (DF1-mini) seems to be relatively purer than its next generation derivative (DF1).

Hydrodynamic Size/Size Distribution via Dynamic Light Scattering (DLS)

Design and Methods

Hydrodynamic size (diameter) of DF1 and DF1-mini were measured in 10 mM NaCl using DLS at 25° C. An instrument with back scattering detector was used for these measurements in batch mode (no fractionation). Samples were weighed and dissolved in 10 mM NaCl to give a final concentration of 1 mg/mL. Samples were filtered through a 0.02 μ filter before taking a minimum of nine measurements.

Results

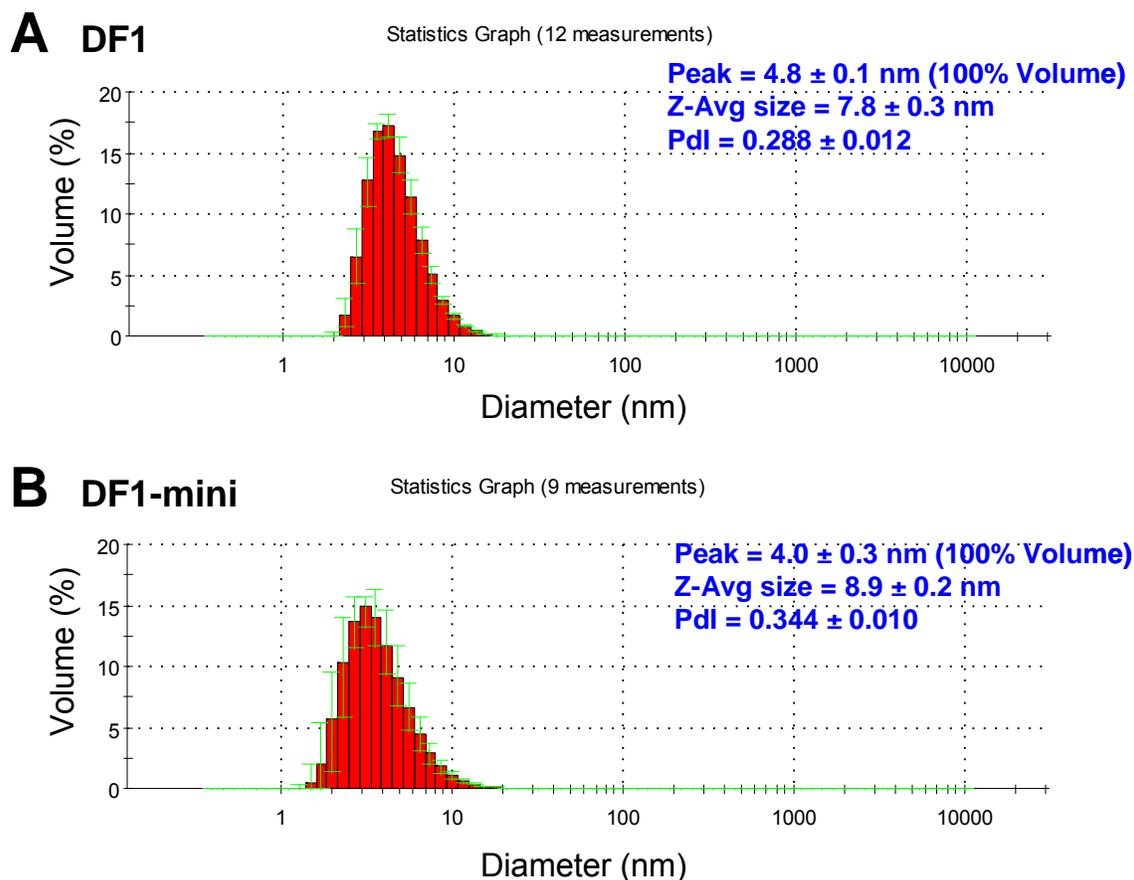


Figure 2. Volume distribution plot for (A) DF1 and (B) DF1-mini in 10 mM NaCl.

Analysis and Conclusions

The Z-avg size, which is more sensitive to larger size species in the sample, of DF1 is 7.8 nm. The Z-avg size of DF1-mini is 8.9 nm. There is only a slight difference in the volume-weighted size (denoted as 'Peak' in Figure 2) between the two conjugates, with DF1 being slightly larger (4.8 nm) than DF1-mini (4.0 nm) as expected. This data also suggests that under these buffer conditions, there is little or no aggregation in these samples.

Ultra Violet–Visible (UV-Vis) Absorbance Spectra

Design and Methods

UV-Vis spectra were recorded using a Thermo Electron Evolution 300 spectrophotometer (Waltham, MA). Samples were prepared in HPLC-grade water and measured in quartz microcuvettes (with light path length, $b = 10$ mm, QS109.004, Hellma, Plainview, NY).

Results

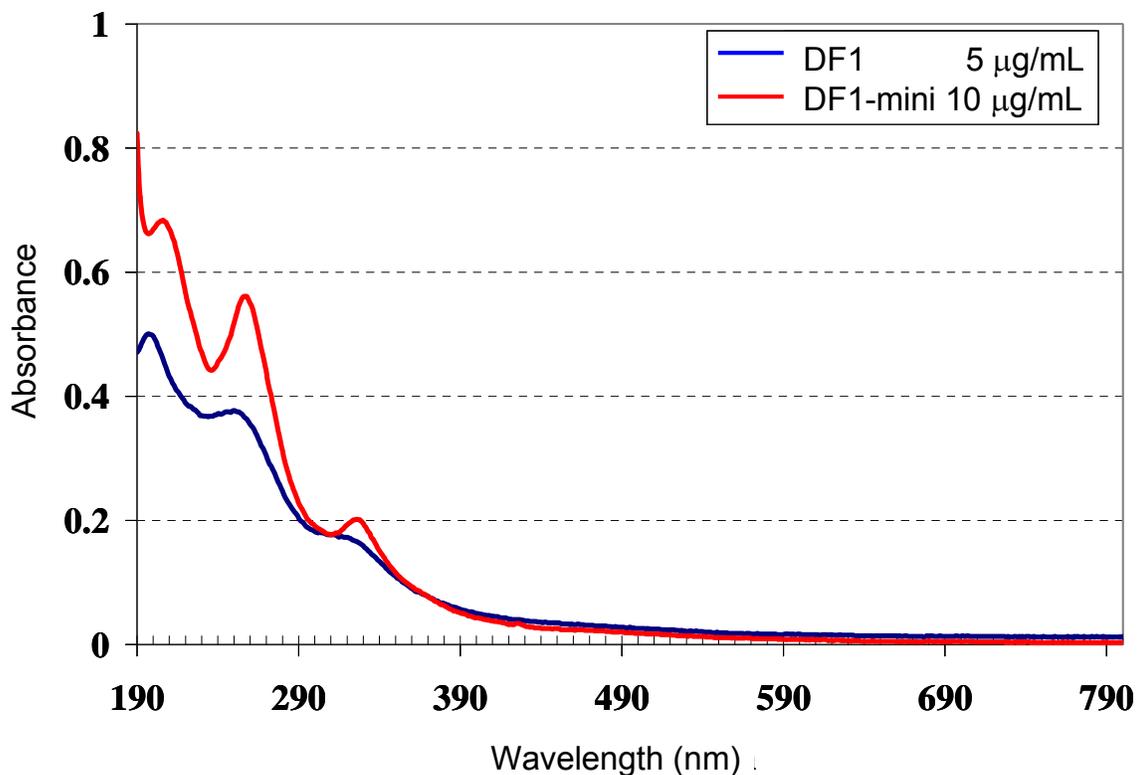


Figure 3. UV-Vis spectra for DF1 and DF1-mini.

Analysis and Conclusions

DF1 and DF1-mini have similar UV-Vis absorption spectra as expected. The peaks at 256 nm and 325 nm are at characteristic wavelengths for absorbance of the fullerene ring and are at approximately the same wavelengths (as expected) for DF1 and DF1-mini. The peaks around 200 nm are attributed to absorbance in the dendrimer branches. There is a small shift in the λ_{\max} of this peak between the spectra of the DF1 and DF1-mini samples. The λ_{\max} of the lowest wavelength peak in the DF1 spectrum is at a lower wavelength than that of the analogous peak in the DF1-mini spectrum.

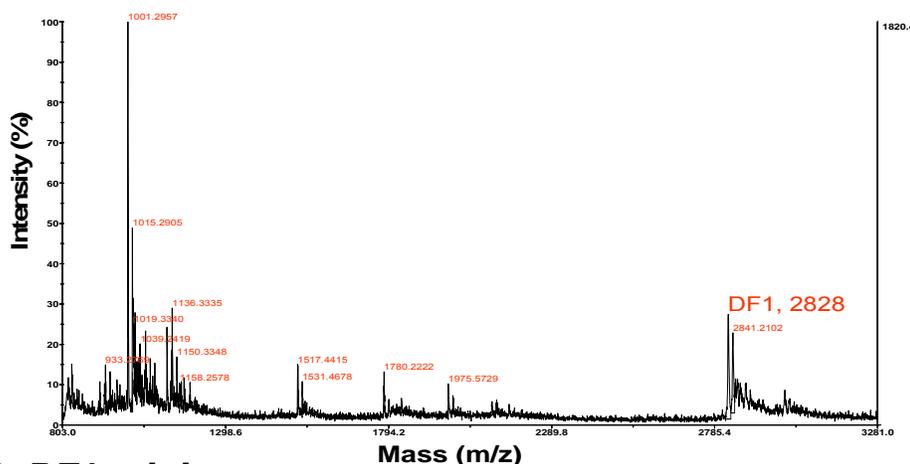
Molecular Weight Measurement by Mass Spectrometry (MS)

Design and Methods

Mass spectra were obtained for DF1 and DF1-Mini using matrix assisted laser desorption/ionization-time of flight mass spectrometry (MALDI-TOF MS) in DHB matrix, at 10mg/mL and CH₃CN/H₂O =3/7 (v/v) in positive-ion mode.

Results

A DF1



B DF1-mini

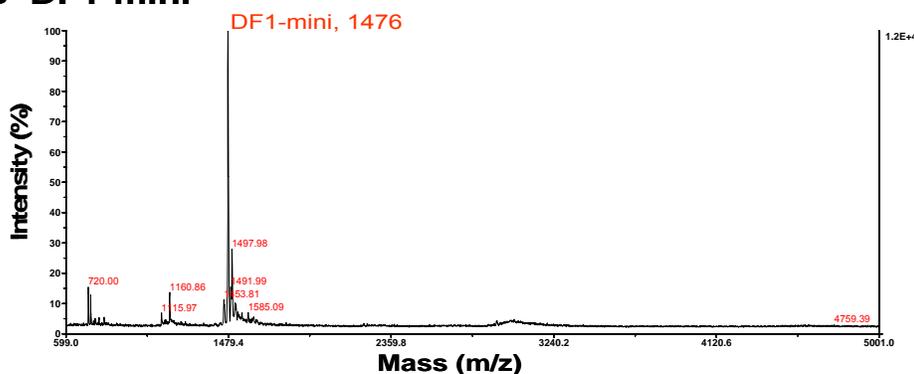


Figure 4. Mass spectra for (A) DF1 and (B) DF1-mini.

Analysis and Conclusions

The theoretical molecular weights of DF1 and DF1-mini are 2828 Da and 1452 Da, respectively. The experimental molecular weight determined from the mass spectra were 2827 Da for DF1 and 1476 Da (which corresponds to 1452 Da plus the mass of a sodium ion) for DF1-mini. The spectrum of the DF1 sample (A) contains several peaks in addition to the peak corresponding to DF1, suggesting the sample contains impurities or fragments due to MS conditions. Minor peaks (e.g., the peak at 720 Da which corresponds to a fullerene ring without any dendritic branches) are also present in the spectrum of DF1-mini (B) and suggest that minor impurities are also present in this sample.

Reversed-Phase HPLC

Design and Methods

Reversed-phase high performance liquid chromatography (RP-HPLC) is a separation technique used for determining the purity of a sample. It is based on the partitioning of the sample molecules with the mobile phase and the stationary hydrophobic phase. The chromatographic system used here consisted of a degasser (Agilent G1379A, Palo Alto, CA), capillary pump (Agilent G1378A), micro well-plate autosampler (Agilent G1377A), Zorbax 300SB-C18 column (1.0 mm ID x 150 mm, 3.5 μm , Agilent), and a diode array detector (Agilent G1315B). The mobile phase consisted of water/acetonitrile (HPLC Grade, 0.14 % (w/v) trifluoroacetic acid) at varying volume percentages with a flow rate of 50 $\mu\text{L}/\text{min}$. The elution profile is illustrated in red and the chromatogram is shown in blue in Figure 5. The injected sample had a concentration of 1mg/mL in a volume of 5 μL (in HPLC water). The eluted sample was detected at 254 nm. Samples were run in triplicate.

Results

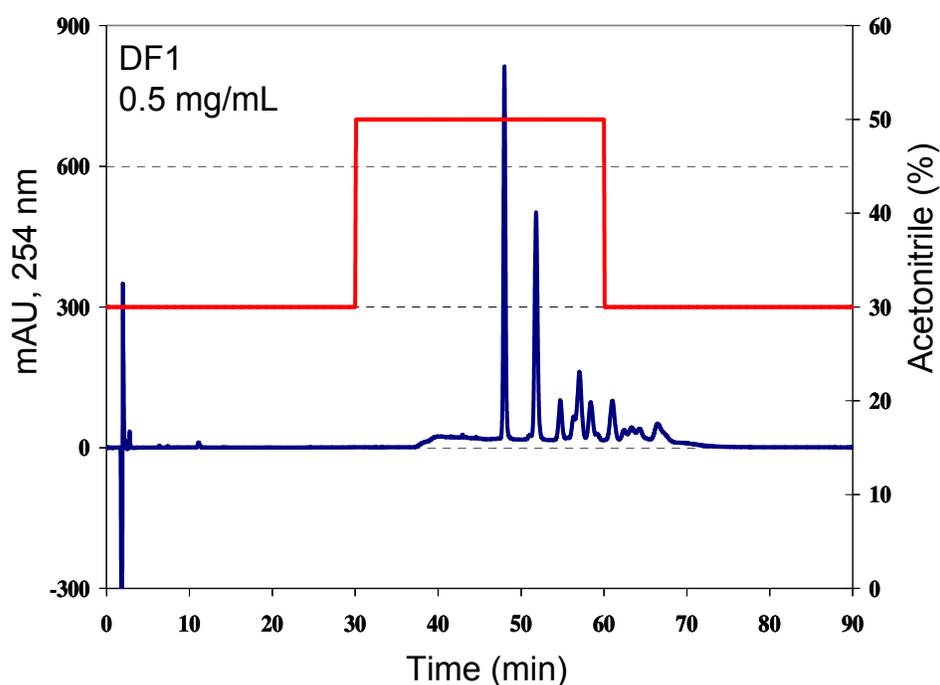


Figure 5. RP-HPLC chromatogram for DF1. See Table 1 for peak retention times and percent areas.

Peak #	Retention Time (min)	% Area	Std Dev (N=3)
1	48.0	29.7	0.4
2	51.1	0.7	0.2
3	51.8	26.3	0.1
4	54.7	4.8	0.1
5	56.4	2.7	0.2
6	57.0	10.4	0.1
7	58.4	5.8	0.3
8	59.2	0.8	0.2
9	61.0	6.8	0.3
10	62.5	1.7	0.1
11	63.3	2.7	0.2
12	64.3	2.4	0.1
13	66.5	5.2	0.3

Table 1. The retention times, percentage of the area and standard deviation (N = 3), for DF1 as determined by RP-HPLC (Figure 5).

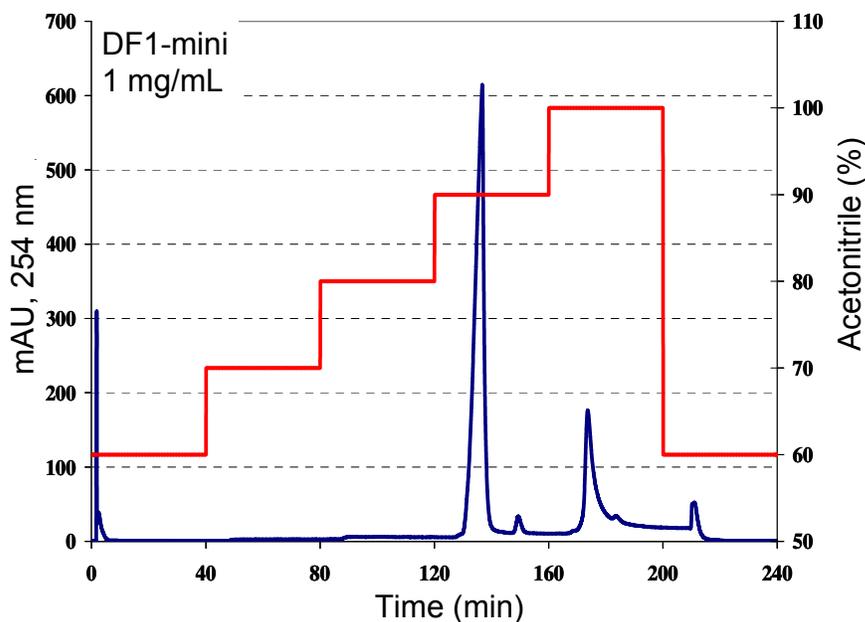


Figure 6. The RP-HPLC chromatogram and elution profile for DF1-mini. See Table 2 for peak retention times and percent areas.

Peak #	Retention Time (min)	% Area	Std Dev (N=3)
1	136.8	75.7	6.8
2	149.3	1.5	0.2
3	173.7	19.3	6.3
4	183.7	0.4	0.1
5	210.6	3.0	0.5

Table 2. The retention times, percentage of the area and standard deviation (N = 3), for DF1-mini as determined by RP-HPLC (Figure 6).

Analysis and Conclusions

For DF1, the UV spectra (data not shown) for the peaks (see Figure 5) highlighted in cyan in Table 1 are consistent with that of DF1 measured with a UV-Vis spectrophotometer. The UV spectral analysis on the remaining peaks showed similar absorption peaks but were slightly shifted to lower wavelengths (blue-shifted by ~10 nm). The UV absorption at these wavelengths is indicative of the presence of fullerene and not free dendritic branches. This data is consistent with published reports by Gharbi et al. (*Anal. Chem.* **2003**, 75, 4217-4222).

For DF1-Mini, the UV spectra (data not shown) for the peaks (see Figure 6) highlighted in cyan in Table 2 are consistent with that of DF1-mini measured with a UV-Vis spectrophotometer. The UV spectral analysis on the remaining two peaks showed a major peak at 225 nm and a minor peak at 254 nm suggesting that these impurities contain the fullerene component. DF1-mini seems to be relatively purer than the next generation derivative (DF1). This is also consistent with mass spectrometry analysis on these two derivatives. Further analysis with LC-MS would provide additional details on the species present as impurities in these two samples.

Zeta Potential

Design and Methods

Samples were diluted in 10 mM NaCl to give a 2 mg/mL final concentration (pH 5.4). A folded capillary cell was used to make these measurements. A minimum of ten measurements were taken for each sample at 25 °C. The applied voltage was 90 V.

Results

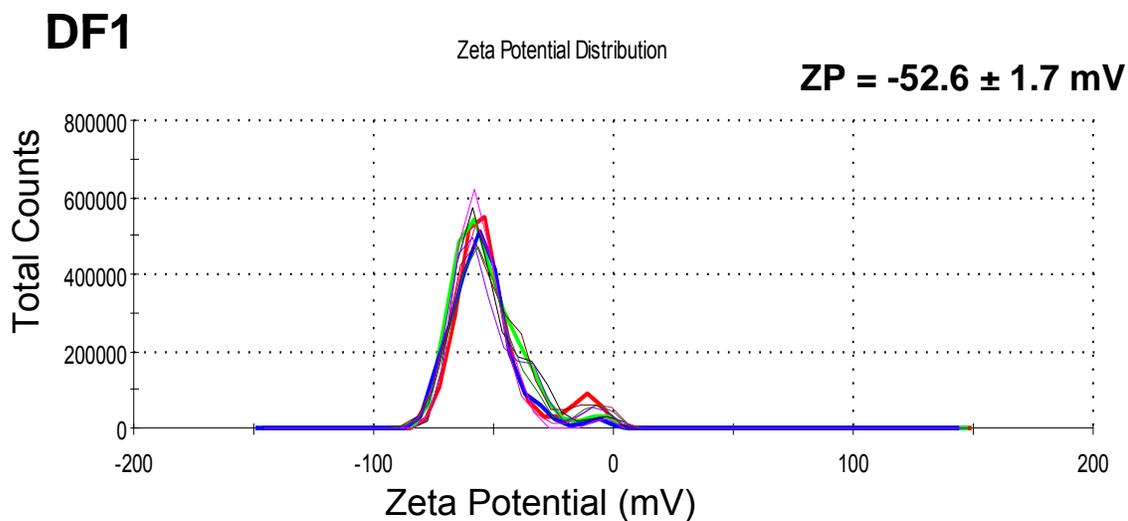


Figure 7. The zeta potential distribution derived from the electrophoretic mobility for DF1 in 10 mM NaCl.

Analysis and Conclusions

The measured zeta potential (-53.6 mV) is negative as expected for a fullerene ring with carboxylated dendritic branches.

Capillary Electrophoresis (CE)

Design and Methods

Capillary Electrophoresis (CE) is a powerful chromatographic technique that separates analytes on the basis of electrophoretic mobility differences. Mobility is determined by the mass to charge ratio of the analyte. CE has high separation efficiencies, high sensitivity, short run times and high automation capability. Sample concentration 0.1 mg/mL in water, capillary Sample preparation: all stock samples were prepared in 50% acetonitrile as 1mg/mL solutions and stored at -40 °C. The stock solutions were diluted 10-fold with water before capillary electrophoresis (0.1mg/mL final concentration). Running conditions: capillary: 60cm ×50 µm I.D; buffer: 10mM sodium tetraborate; separation voltage: 22kV; injection pressure: 20s/0.5 psi/20s; detector: UV (wavelength 200nm). The capillary was rinsed sequentially with 1N sodium hydroxide, water, and the CE buffer between runs.

Results

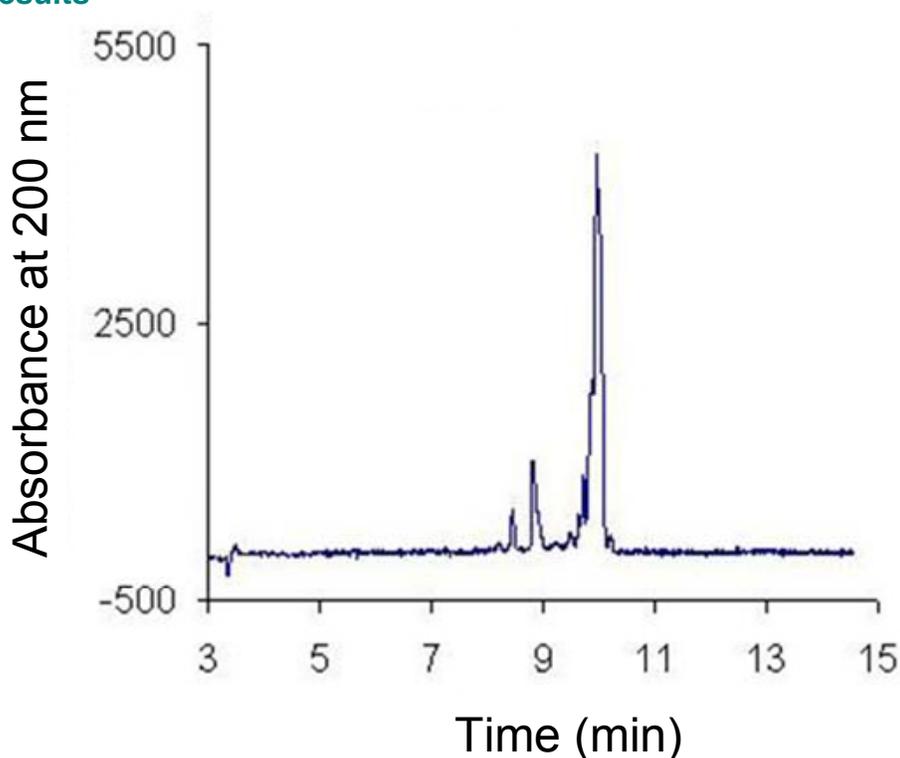


Figure 8. Electropherogram of DF1.

Analysis and Conclusions

The chromatogram shows a major peak eluting at ~10 minutes with several minor peaks eluting before. This suggests that DF1 contains impurities and is consistent with the RP-HPLC data.

Capillary Electrophoresis (CE) in Serum Matrix

Design and Methods

To facilitate *in vivo* studies for the detection and quantitation of the fullerene derivatives in matrix, a CE method was developed for serum incubated samples. Capillary electrophoresis was performed using a Beckman MDQ system (Fullerton, CA, USA) equipped with a photo-diode-array detector. All electropherograms were plotted using the 260 nm channel data. Separations were carried out at room temperature using 50 μm inner diameter fused-silica capillaries with lengths of 40 or 60 cm ($L_d=30$ or 50 cm). Samples were injected by applying pressure (0.5 psi) for 20 seconds. The field strength for separation was approximately 350 V/cm. Stock solutions of DF1 and C3 were prepared at a final concentration of 1 mg/mL in 50% acetonitrile/H₂O and stored at -40 °C until used. Prior to CE analysis, aliquots of the stock solutions were lyophilized and reconstituted in PBS or human serum. Nanoparticle samples prepared in human serum were incubated at room temperature for at least 3 hours before use. All serum samples were diluted five-fold with H₂O or SDS solutions of various concentrations before CE analysis. For Capillary Zone Electrophoresis (CZE) and Micellar Electro Kinetic Chromatography (MEKC), a capillary was sequentially washed for one minute with 1 N NaOH, H₂O, and separation buffers between sample injections. A dynamic coating procedure was used for CZE experiments with suppressed electro-osmotic flow.

Results

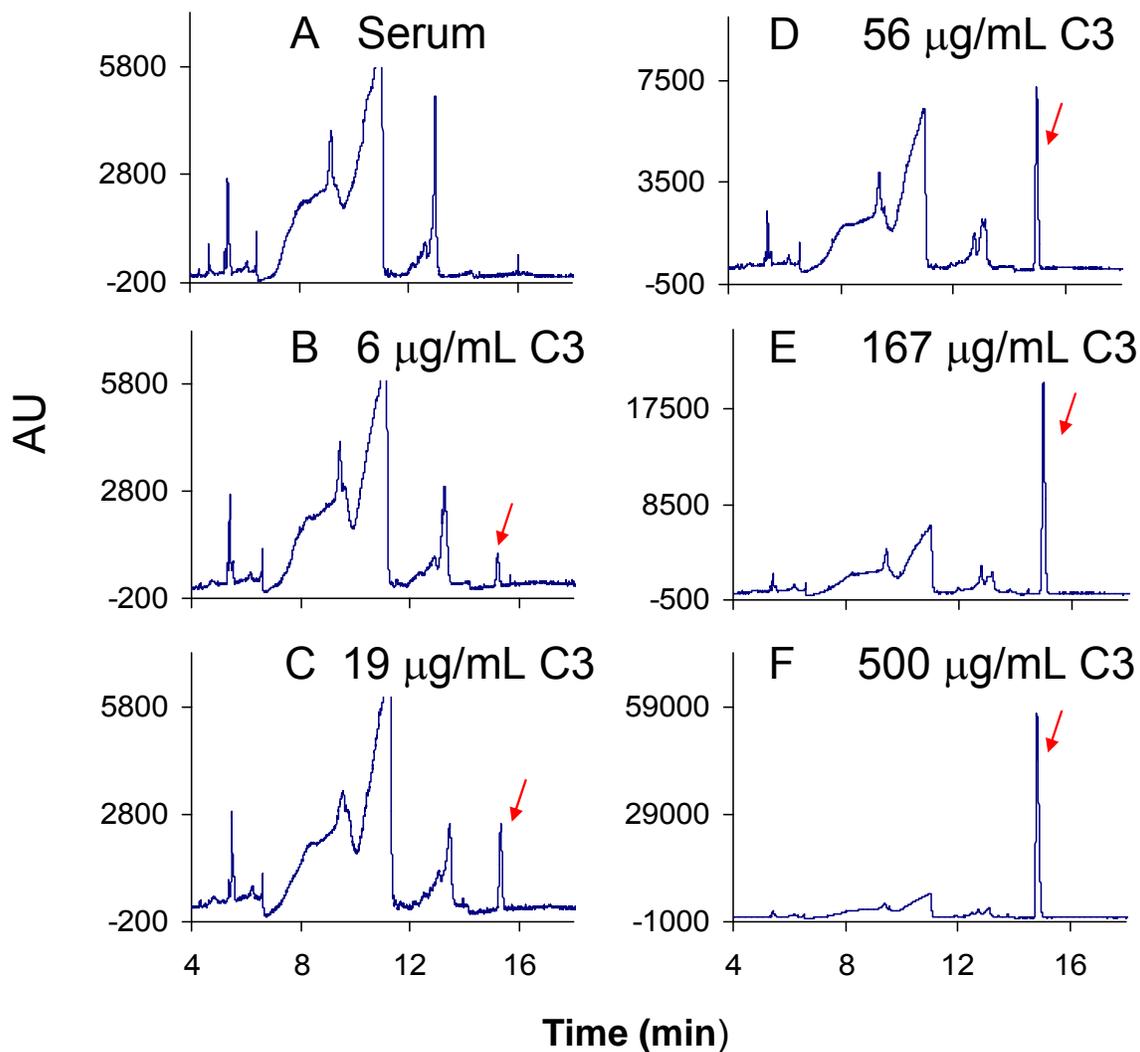


Figure 9. Calibration of carboxyfullerene (C3) in serum using micellar electrokinetic chromatography. Capillary: 50 µm x 40 cm; buffer: 10 mM sodium tetraborate (pH 9.2) with 150 mM SDS; voltage: +14 kV. Serum samples were diluted 5-fold with water and the C3 concentrations refer to the undiluted samples.

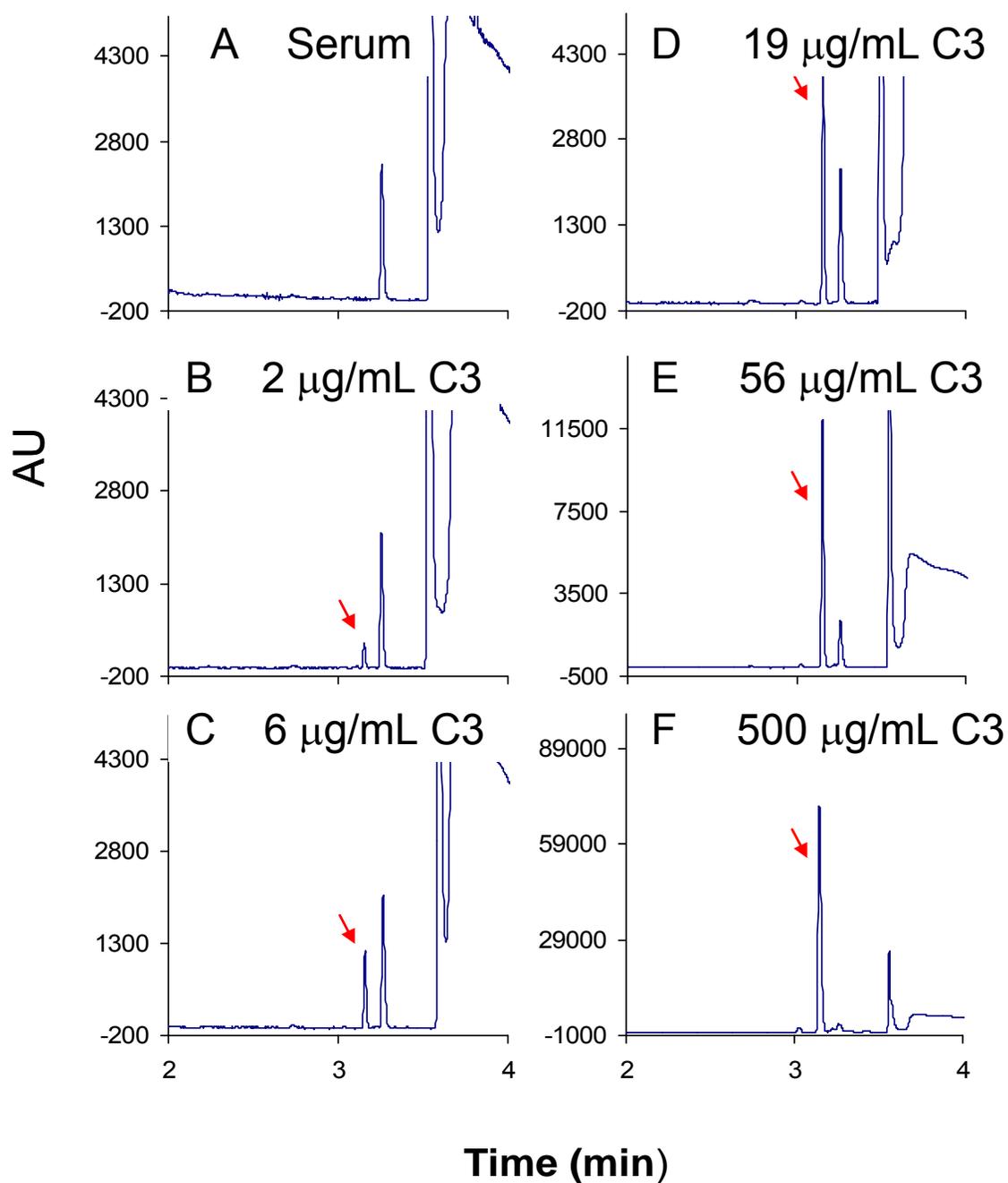


Figure 10. Calibration of carboxyfullerene (C3) in serum using capillary zone electrophoresis. Capillary: 50 μm x 40 cm with dynamic coating; buffer: 40 mM sodium phosphate (pH 7.4); voltage: -14 kV. Serum samples were diluted 5-fold with SDS solution (80 mM final concentration). The C3 concentrations refer to the undiluted samples.

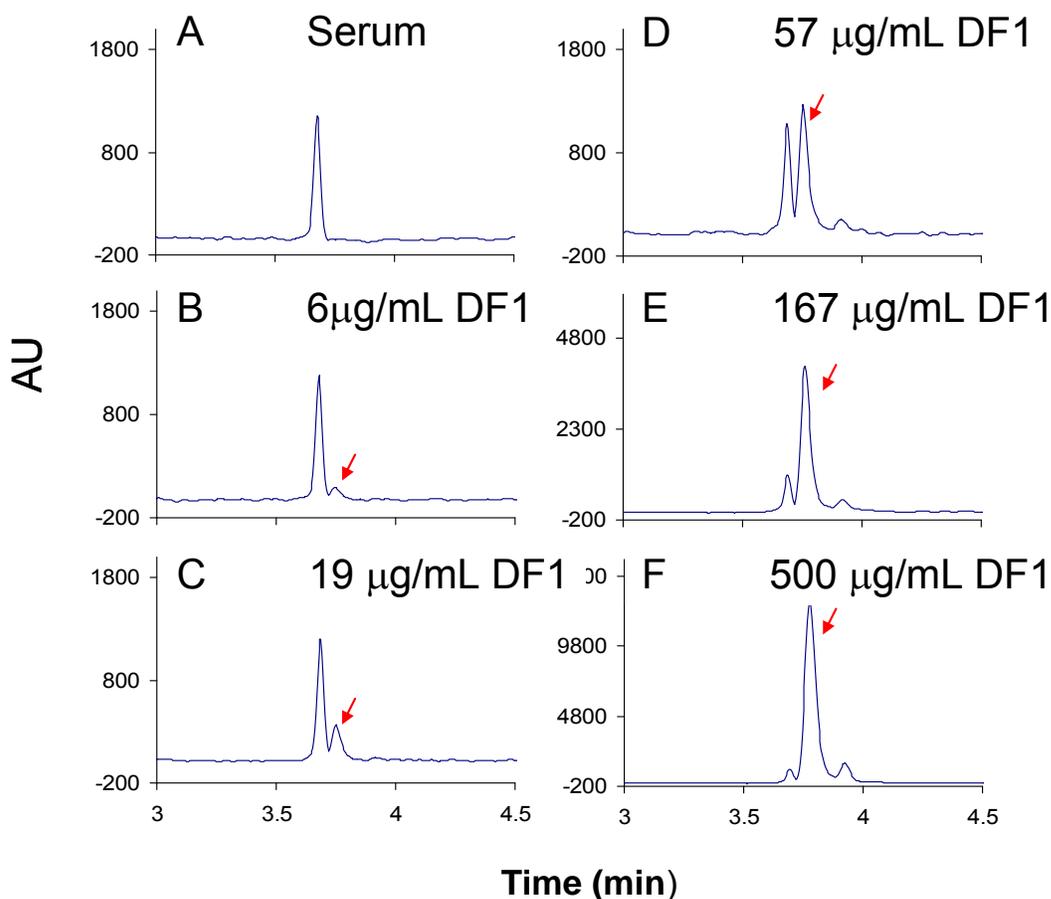


Figure 11. Calibration of dendrofullerene (DF1) in serum using capillary zone electrophoresis. Capillary: 50 μm x 40 cm with dynamic coating; buffer: 40 mM sodium tetraborate (pH 9.2); voltage: -14 kV. Serum samples were diluted 5-fold with SDS solution (20 mM final concentration). The DF1 concentrations refer to the undiluted samples.

Analysis and Conclusions

The calibration curve of C3 in serum using the optimized condition is shown in Figure 9. The peak area versus the C3 concentration is linear from 0-500 $\mu\text{g/mL}$ ($R^2 = 0.9998$) with a quantitation limit (S/N=3) of approximately 2 $\mu\text{g/mL}$. The recovery of the assay, calculated by comparing the peak area of 100 $\mu\text{g/mL}$ C3 spiked in human serum to that in PBS, is approximately 108%.

The CZE calibration of C3 in human serum treated with 80 mM SDS is shown in Figure 10. The peak area versus the concentration plot is linear from 0-500 $\mu\text{g/mL}$ ($R^2 = 0.999$) with a quantitation limit (S/N =3) of approximately 0.6 $\mu\text{g/mL}$. The recovery of the assay, calculated by comparing the peak area of 0.1 mg/ml C3 spiked in human serum to that in PBS, is approximately 104%.

The DF1 calibration in serum with the borate buffer after addition of 20 mM SDS prior to CZE analysis is shown in Figure 11. The peak area versus the concentration plot is linear from 0 to 500 $\mu\text{g/mL}$ ($R^2 = 0.997$), with a quantitation limit of detection approximately 6 $\mu\text{g/mL}$.

(S/N=3). By comparing the peak area of DF1 in PBS to that in serum, the recovery of the DF1 using CZE with the 20 mM SDS treatment protocol is approximately 60.

This CE method can be utilized both for quality control purposes, to test batch-to-batch variability, and for the detection and quantitation of the two fullerene derivatives studied in matrix. This method can be further utilized for other similar nanomaterial samples with minor modifications.

In Vitro Toxicology

Section Summary

The biocompatibility of the functionalized fullerenes was evaluated in porcine renal proximal tubule (LLC-PK1) and human leukemia (HL60) cell lines. AF1 and C3 functionalized fullerenes were found to be minimally cytotoxic, while AF3 and DF1 were nontoxic to LLC-PK1 cells, as determined by the MTT, LDH and trypan blue dye-exclusion assays (Figures 12 - 15, Table 3). The 48 hour IC₅₀ values for AF1 and C3 in LLC-PK1 cells were 0.216 (0.044-0.388, 95% CI) mg/mL and 0.143 (0.100 – 0.187, 95% CI) mg/mL, respectively (Figure 14, Table 3). In HL-60 cells, both DF1 and C3 were found to be cytotoxic, as determined by the trypan blue dye-exclusion assay, at the 0.25 mg/mL concentration tested (Figure 15). Cell cycle analysis demonstrated that DF1 primarily targeted HL60 cells in the G1 phase (Figure 16). Treatment of renal proximal tubule (LLC-PK1) cells and Sprague-Dawley primary hepatocytes with 0.004-1.0 mg/mL of C3 or AF1 increased levels of reactive oxygen species (ROS) in a linear fashion, while DF1 decreased ROS levels dose dependently (Figures 17 and 18). Since functionalized fullerenes have previously been shown to have the potential to induce oxidative stress (Sayes et al., 2004), and AF1 and C3 were found to increase ROS generation, it is likely that oxidative stress plays a role in the mechanism of AF1 and C3 cytotoxicity. The observation that DF1 is selective in its cytotoxicity, in that it is toxic to the human leukemia cell line (HL-60), but not the porcine renal proximal tubule cell line, warrants further investigation. Based on this biocompatibility assessment, DF1 was chosen as a promising candidate for therapeutic applications and is therefore the focus of the cisplatin chemoprotection studies which follow. The internalization and subcellular localization of the fluorescent AF3 fullerene derivative in LLC-PK1 cells was examined by confocal microscopy. Confocal microscopy data suggests that AF3 is internalized and distributes to mitochondria.

LLC-PK1 LDH and MTT Cytotoxicity Assays (GTA-1)

Design and Methods

The objective of this study was to determine if AF1, AF3, DF1, or C3 were cytotoxic to porcine renal proximal tubule (LLC-PK1) cells.

Cytotoxicity was determined as described in the NCL method, LLC-PK1 Kidney Cytotoxicity Assay (GTA-1). Briefly, test materials were solubilized via sonication and diluted to the desired assay concentrations in cell culture media (0.004-1 mg/mL). Cells were plated in a 96-well, microtiter plate format. Cells were preincubated for 24 hours prior to test material addition, reaching an approximate confluence of 80%. Cells were then exposed to test material for 6, 24, and 48 hours in the dark, and cytotoxicity was determined using the MTT cell viability and LDH membrane integrity assays. In order to estimate IC_{50} values, the MTT cytotoxicity assay dose-response curves were fit to a sigmoidal Hill equation, $E = E_{max} \frac{C^{\gamma}}{C^{\gamma} + IC_{50}^{\gamma}}$, using nonlinear regression analysis (Figure 14) (WinNonlin, Pharsight Corp., Mountain View, CA).

Results

The maximum concentration of AF1, AF3, DF1 and C3 tested in the LLC-PK1 cytotoxicity study was 1.0 mg/mL. Neither AF3 nor DF1 produced any measurable loss of cell viability, as determined by the MTT assay, or loss of membrane integrity, as determined by LDH assay at the concentrations tested (Data not shown). Treatment of cells with AF1 or C3, however, resulted in a dose-responsive loss of cell viability (Figures 12 and 13), as measured by MTT assay, and a minor, non-dose responsive loss of membrane integrity (Figure 12 and 13), as measured by LDH leakage. The IC_{50} values, estimated from the nonlinear regression analysis of the MTT assay dose-response curves, were 0.143 (0.100 – 0.187, 95% CI) mg/mL and 0.216 (0.044-0.388, 95% CI) mg/mL, for AF1 and C3, respectively (Figure 14, Table 3).

AF1, AF3 and DF1 were determined to be nontoxic, while C3 was determined to be minimally toxic, to LLC-PK1 cells under these assay conditions. DF1 and C3 were found to be minimally toxic to HL60 cells under these assay conditions.

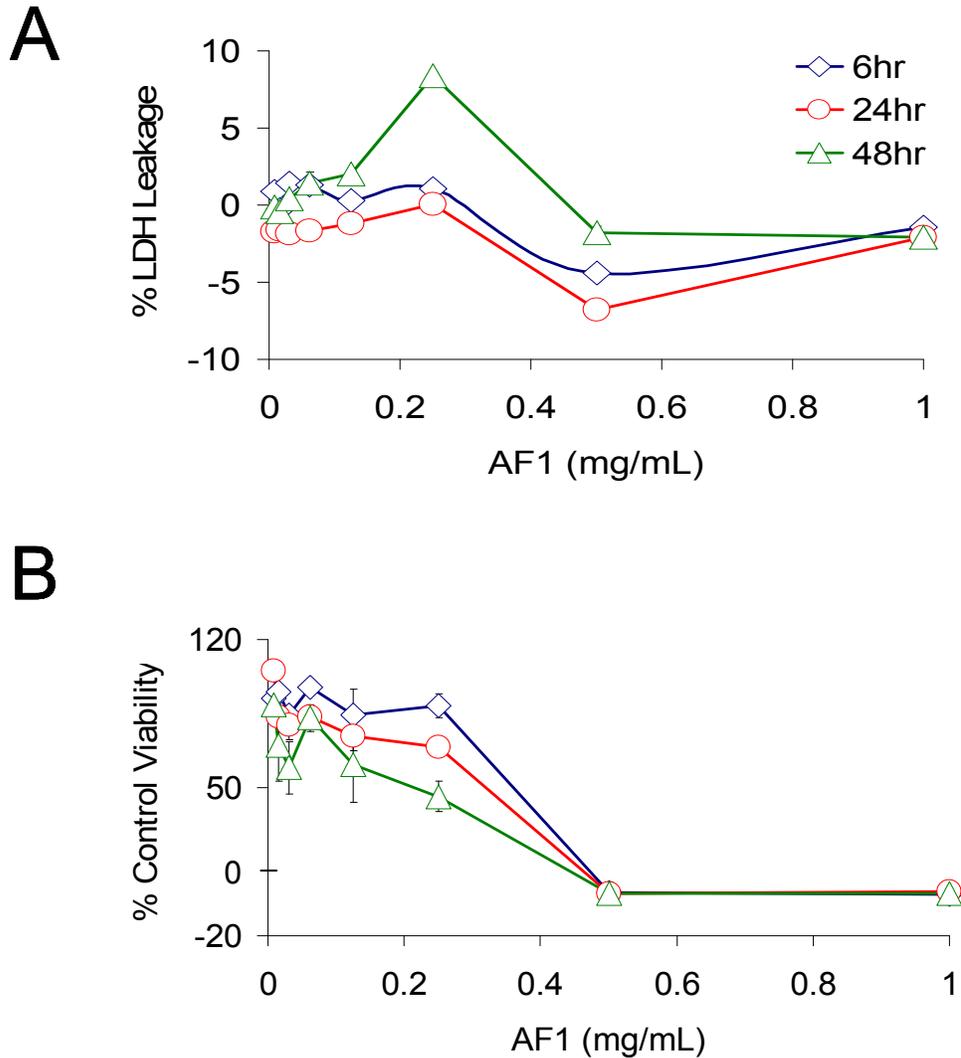


Figure 12. Cytotoxicity of AF1 assayed in LLC-PK1 cells. Porcine renal proximal tubule cells were treated for 6, 24, and 48 hours with 0.004-1.0 mg/mL of test sample. Cytotoxicity at each time point was determined by the LDH (A) and MTT (B) assays, as described in the LLC-PK1 Kidney Cytotoxicity Assay (GTA-1). The data points are the mean \pm the standard error, with $N=3$.

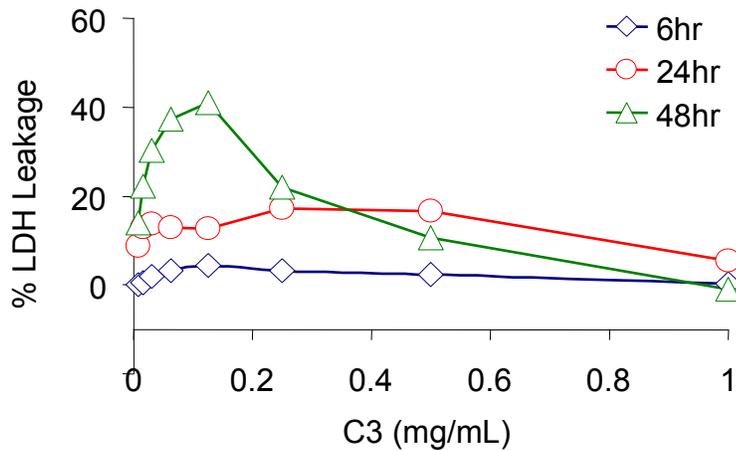
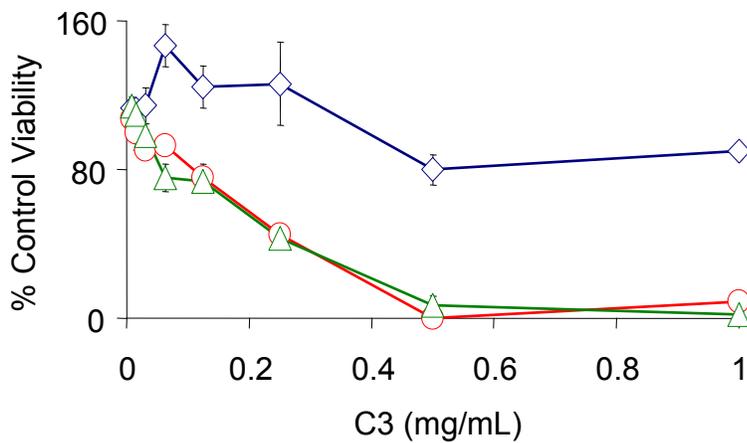
A**B**

Figure 13. C3 cytotoxicity assayed in LLC-PK1 cells. Porcine renal proximal tubule cells were treated for 6, 24, and 48 hours with 0.004-1.0 mg/mL of test sample. Cytotoxicity at each time point was determined by the LDH (A) and MTT (B) assays, as described in the LLC-PK1 Kidney Cytotoxicity Assay (GTA-1). The data points are the mean \pm the standard error, with $N=3$.

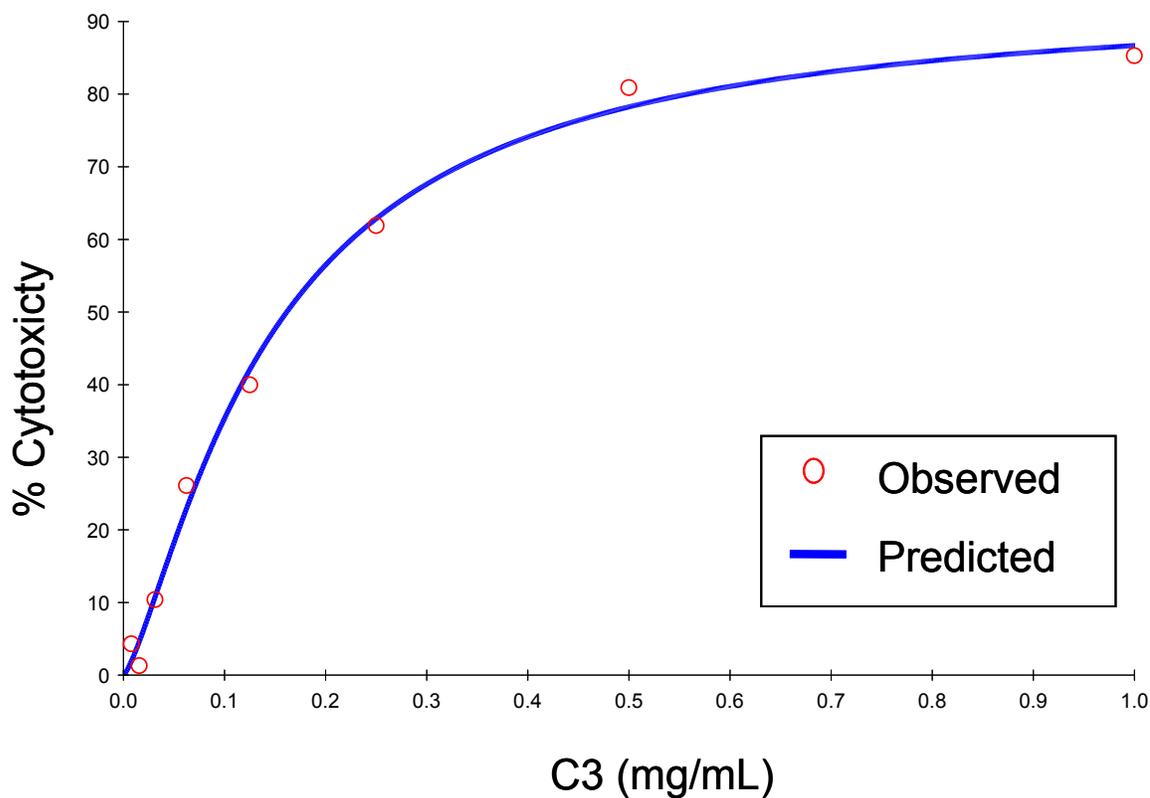


Figure 14. Nonlinear Regression of the Dependence of Percent Cytotoxicity on C3 Concentration. This graph displays the nonlinear fit of the 48 hour C3 MTT data to a sigmoidal Hill equation using the WinNonlin analysis software. The percent cytotoxicity versus the C3 concentration is displayed. The IC_{50} value determined from the fit is 0.145 (0.100 – 0.187, 95% CI) mg/mL.

Sample	IC ₅₀ (mg/mL)
C3	0.143(0.100-0.187)
DF1	Nontoxic
AF1	0.216 (0.44-0.388)
AF3	Nontoxic

Table 3. MTT cytotoxicity assay in LLC-PK1 cells. LLC-PK1 cells were treated for 48 hours, with 0.004-1.0 mg/mL of test sample. Cytotoxicity was determined by the MTT assay, as described in LLC-PK1 Kidney Cytotoxicity Assay (GTA-1). IC₅₀ values were determined by nonlinear regression analysis of the 48 hour MTT cytotoxicity dose-response curves. Data is presented as IC₅₀ (95% CI).

Analysis and Conclusions

AF3 and DF1 were determined to be nontoxic in the LLC-PK1 cells, under these assay conditions. AF1 and C3 were determined to be minimally toxic by both MTT and LDH assay.

LLC-PK1 and HL-60 Trypan Blue Cytotoxicity Assay

Design and Methods

The objective of this study was to determine if AF1, AF3, DF1, or C3 were cytotoxic to porcine renal proximal tubule (LLC-PK1) or human leukemia (HL-60) cell lines.

Cytotoxicity was determined by the trypan blue dye exclusion assay. Test materials were solubilized via sonication and diluted to the 0.25 mg/mL test concentration. LLC-PK1 and HL60 cells were plated in 35mm, 6-well plates. Cells were preincubated for 24 hours prior to test material addition, reaching an approximate confluence of 80%. Cells were then exposed to the test material for 24 hours in the dark, and cytotoxicity was determined using the trypan blue dye exclusion cell viability assay. Briefly, trypsinized LLC-PK1 and HL60 cells were diluted 1:10 with 0.08% trypan blue in PBS, stained for 5 minutes, and counted with a hemocytometer. Cell viability was determined by counting the number of live cells, which exclude the dye, versus the number of dead cells, which are stained.

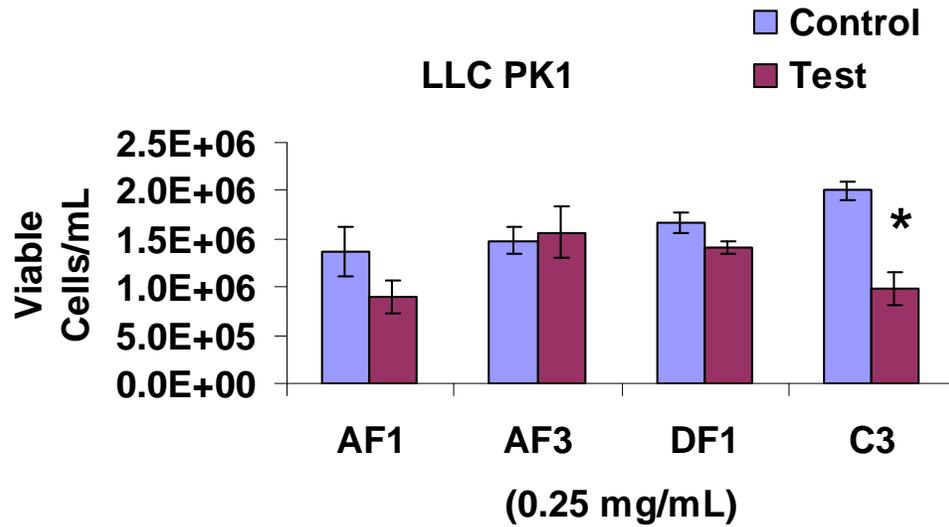
Results

The test concentration of AF1, AF3, DF1 and C3 utilized in the LLC-PK1 and HL60 cell trypan blue cytotoxicity assays was 0.25 mg/mL. Treatment of LLC-PK1 cells with AF1, AF3 or DF1 for 24 hours, did not produce a significant loss of cell viability, as determined by the trypan blue dye-exclusion assay (Figure 15A). However, treatment of LLC-PK1 cells with C3, at the same test concentration and exposure period, resulted in a significant loss of cell viability compared to media treated control (Figure 15A). Treatment of HL60 cells with either DF1 or C3, resulted in a significant loss of cell viability (Figure 15B), as measured by the trypan blue exclusion assay.

Analysis and Conclusions

AF1, AF3 and DF1 were determined to be nontoxic, while C3 was determined to be minimally toxic, to LLC-PK1 cells under these assay conditions. DF1 and C3 were found to be minimally toxic to HL60 cells under these assay conditions.

A



B

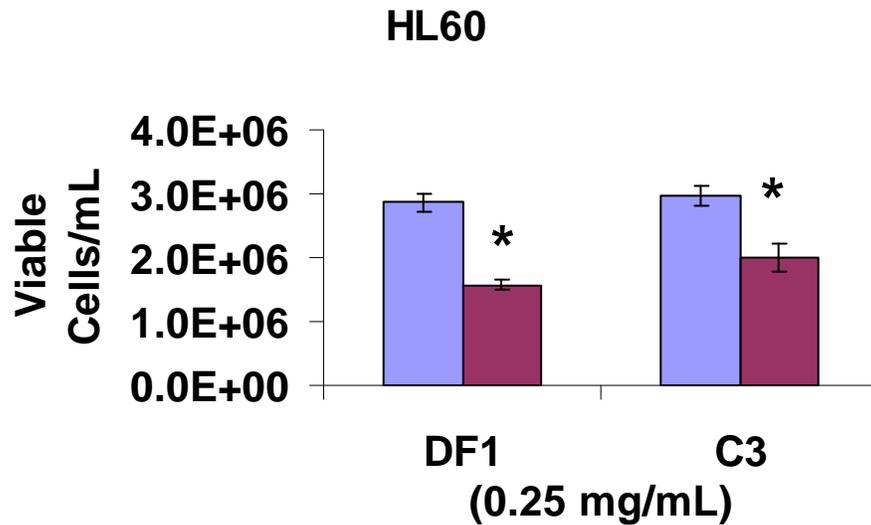


Figure 15. Trypan blue dye-exclusion assay in LLC-PK1 and HL60 cells. LLC-PK1 (A) and HL-60 (B) cells were treated for 24 hours with 0.25 mg/mL of test samples. Cytotoxicity was determined by the trypan blue dye-exclusion assay, as described in methods. The data points are the mean \pm the standard error, with $N=3$.

Cell Cycle Effects of DF1 in HL60 Cells

Design and Methods

The purpose of this study was determine if DF1 perturbs cell cycle progression in human leukemia (HL60) cells.

Flow cytometry analysis of the cell cycle in HL60 cells was conducted using propidium iodide nucleic acid staining. Briefly, DF1 was diluted to the 0.25 mg/mL test concentration in cell culture media. HL60 cells were plated in 35mm, 6-well format. Cells were preincubated for 24 hours prior to treatment. Cells were treated for 24 hours with 0.25 mg/mL of DF1 in the dark. Following the treatment period, cells were washed with PBS, fixed with ice cold ethanol, and stained with 3 μ M propidium iodide for 15 minutes at room temperature. Following staining, cells were analyzed by flow cytometry (ex. 488 nm and em. 617).

Results

Treatment of the HL60 cells with 0.25 mg/mL DF1, resulted in an increase in dead/apoptotic cells ($11 \pm 1.5\%$ vs. $35 \pm 1.9\%$) and a decrease in G1 ($45 \pm 0.2\%$ vs. $25 \pm 0.8\%$) and G2/M (19 ± 0.9 vs. 14 ± 0.4) phase cells (Figure 16).

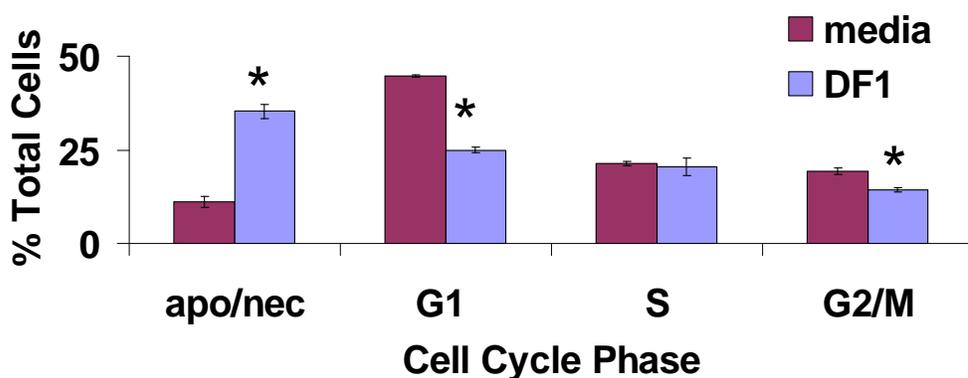


Figure 16. The effect of DF1 on the cell cycle of HL60 cells. HL60 cells were treated for 24 hours with 0.25 mg/mL of DF1. Cell cycle was determined by propidium iodide nucleic acid staining and flow cytometry. Data represents the mean \pm the standard error, $N=3$, $*P < 0.05$.

Analysis and Conclusions

DF1 appears to primarily target HL60 cells in the G1 phase of the cell cycle.

LC-PK1 and Sprague-Dawley Primary Hepatocyte Reactive Oxygen Species Assay (GTA-7)

Design and Methods

The objective of this study was to determine if AF1, AF3, DF1, or C3 influence reactive oxygen species (ROS) generation in porcine renal proximal tubule (LLC-PK1) and human hepatocarcinoma cells (Hep G2) cell lines.

Reactive oxygen species were measured as described in the Hepatocyte Primary ROS Assay (GTA-7). Briefly, test materials were solubilized via sonication and diluted to the 0.004-1 mg/mL test concentrations in cell culture media. LLC-PK1 cells and Sprague Dawley primary hepatocytes were plated in 96-well, microtiter plate format. Cells were preincubated for 24 hours prior to initial staining, reaching an approximate confluence of 80%. Cells were then incubated in the dark for 30 minutes at 37°C with 40 µM DCFH, the redox active dye. The wells were then washed with media, and treated with test samples in the dark. Microtiter plate fluorescence was measured at multiple times from 0-4 hours on a spectrophotometer at excitation 485 nm and emission 530 nm.

Results

AF3 was found to fluoresce at the assay wavelengths, and due to this interference was not included in the ROS assay. Treatment of LLC-PK1 cells with AF1 and C3 resulted in a supralinear dose-responsive increase in ROS generation (Figure 17A and C). Treatment of LLC-PK1 cells with DF1, however, resulted in a dose-responsive decrease in ROS generation (Figure 17B).

Treatment of SD primary hepatocytes with AF1 and C3 resulted in a dose responsive increase in ROS generation (Figure 18A and C). Treatment SD primary hepatocytes with DF1, however, resulted in a dose responsive decrease in ROS generation.

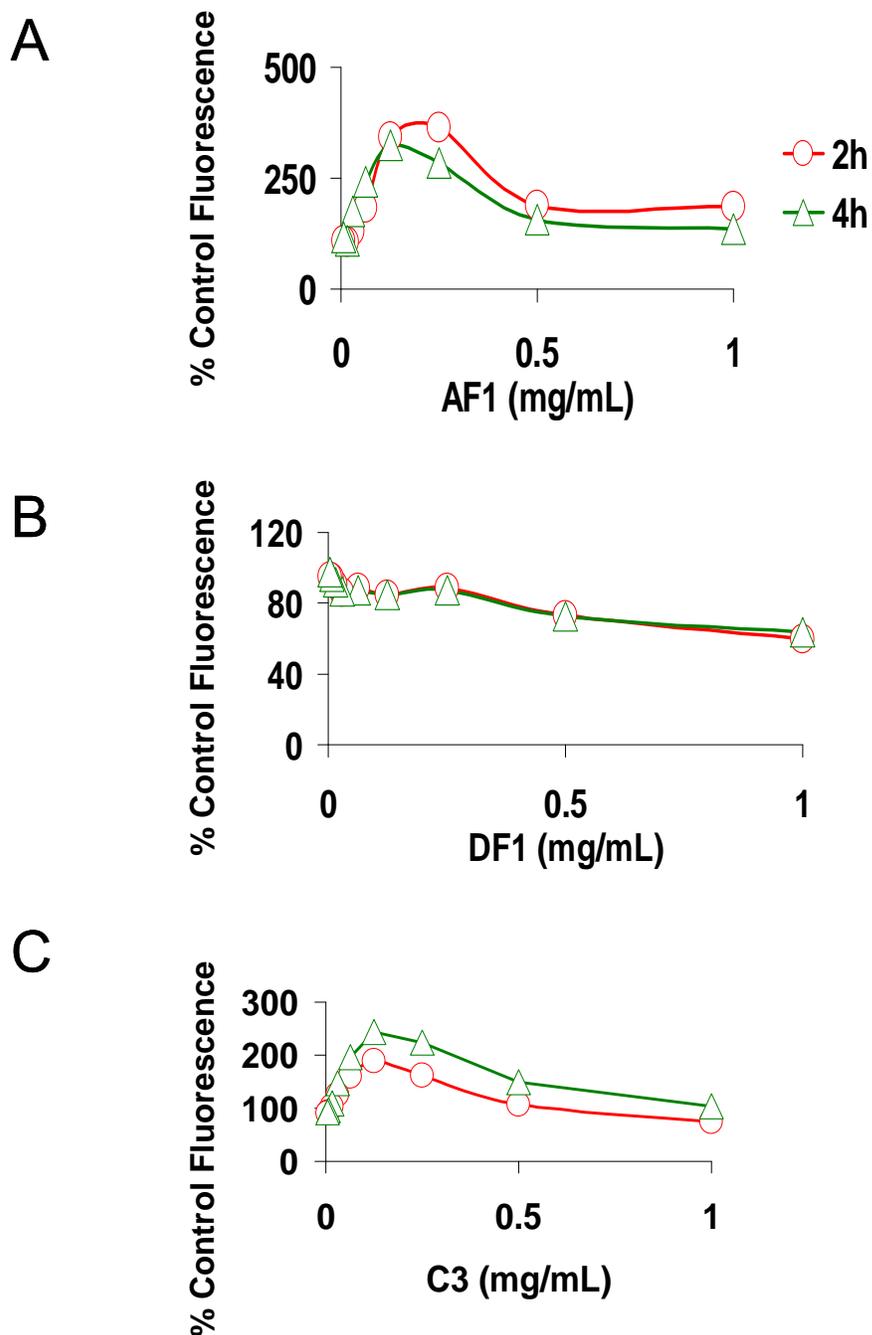


Figure 17. ROS Assay in LLC-PK1 cells. LLC-PK1 cells were treated for 2 and 4 hours with 0.004-1 mg/mL of AF1 (A) DF1 (B), or C3 (C). ROS was determined as described in described in the Hepatocyte Primary ROS Assay (GTA-7), except that LLC-PK1 cells were used instead of Hep G2 cells. Data represents the mean \pm the standard error, $N=3$.

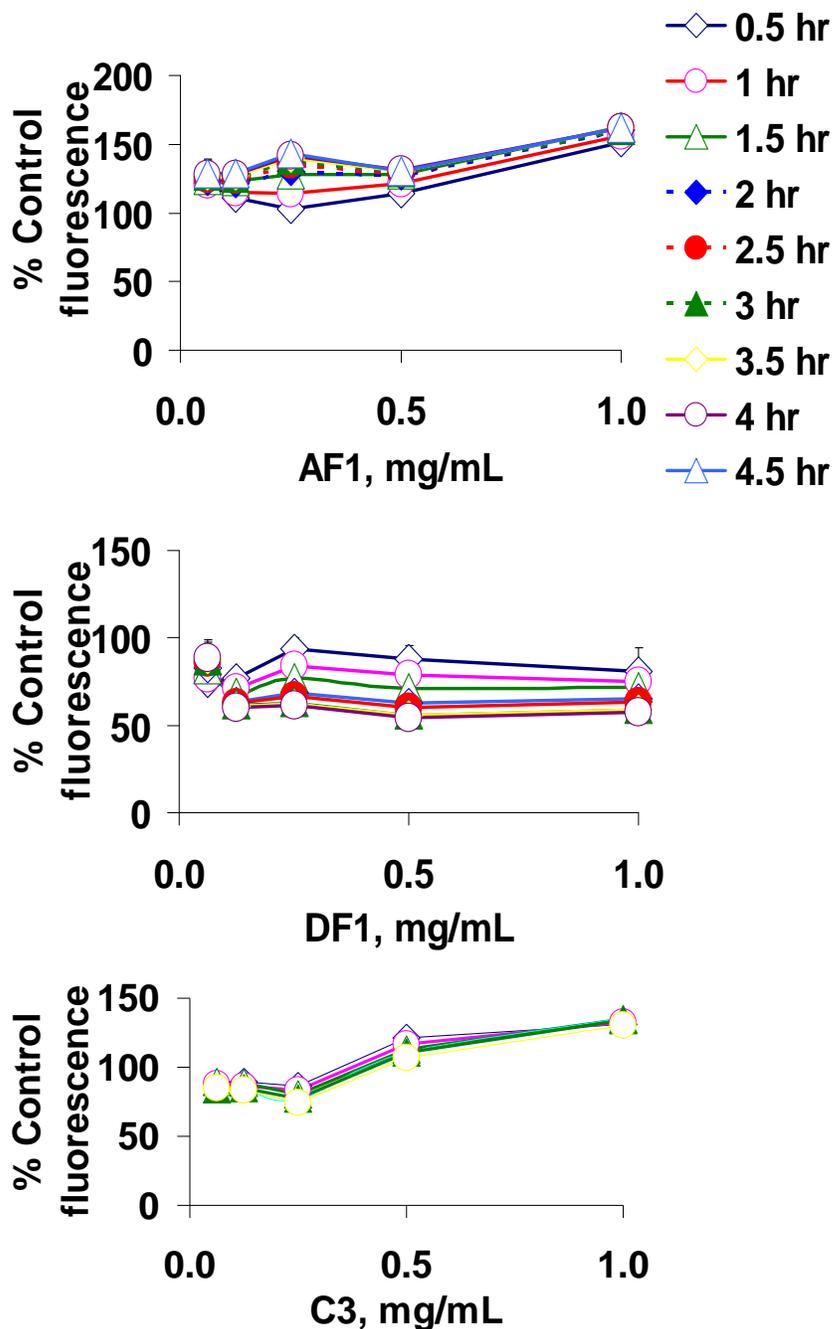


Figure 18. ROS Assay in SD Primary Hepatocytes. SD primary hepatocytes were treated for 0-4 hours with 0.004-1 mg/mL of AF1 (A), DF1 (B), or C3 (C). ROS were measured as described in the Hepatocyte Primary ROS Assay (GTA-7). Data represents the mean \pm the standard error with $N=3$.

Analysis and Conclusions

AF1 and C3 were prooxidant, while DF1 was antioxidant in LLC-PK1 cells and SD primary hepatocytes, under these assay conditions. The redox capacity of AF3 could not be determined by the ROS assay method, due to assay interference.

***In Vitro* Immunological Characterization**

Section Summary

This is a study of the effects of the C-Sixty, Inc. fullerene derivatives on the integrity of red blood cells, platelet aggregation, and formation of granulocyte-macrophage colonies (CFU-GM). The analyses were performed using three NCL immunotoxicity assays: ITA-1 (assay for hemolysis), ITA-2 (platelet aggregation) and ITA-3 (CFU-GM assay). The procedures for these NCL assays are detailed in the August 2006 NCL *In Vitro* Protocols. AF1 and AF3 demonstrated hemolytic properties on human erythrocytes, while C3 and DF1 did not affect the integrity of erythrocytes. None of the tested fullerene derivatives induced platelet aggregation, however C3 and DF1 suppressed aggregation of platelets induced by collagen. None of the tested fullerene derivatives were myelosuppressive or capable of preventing cisplatin-induced toxicity to bone marrow haematopoietic stem cells.

Nanoparticle Hemolytic Properties (ITA-1)

Design and Methods

The objective of this study was to evaluate four fullerene derivatives: AF1, AF3, DF1, and C3 in terms of their effects on the integrity of red blood cells. Stock solutions with theoretical concentration of 1mg/mL were used for each C60 derivative. The resulting final nanoparticle concentration was 125 $\mu\text{g/mL}$. NCL protocol ITA-1 was followed. Poly-L-lysine (PLL) and polyethyleneglycol (PEG) were used as positive and negative controls, respectively.

Results

At tested concentrations, AF1 resulted in >85% hemolysis. AF3 was also hemolytic (% hemolysis >10), while C3 and DF1 did not disturb the integrity of red blood cells (Figure 19).

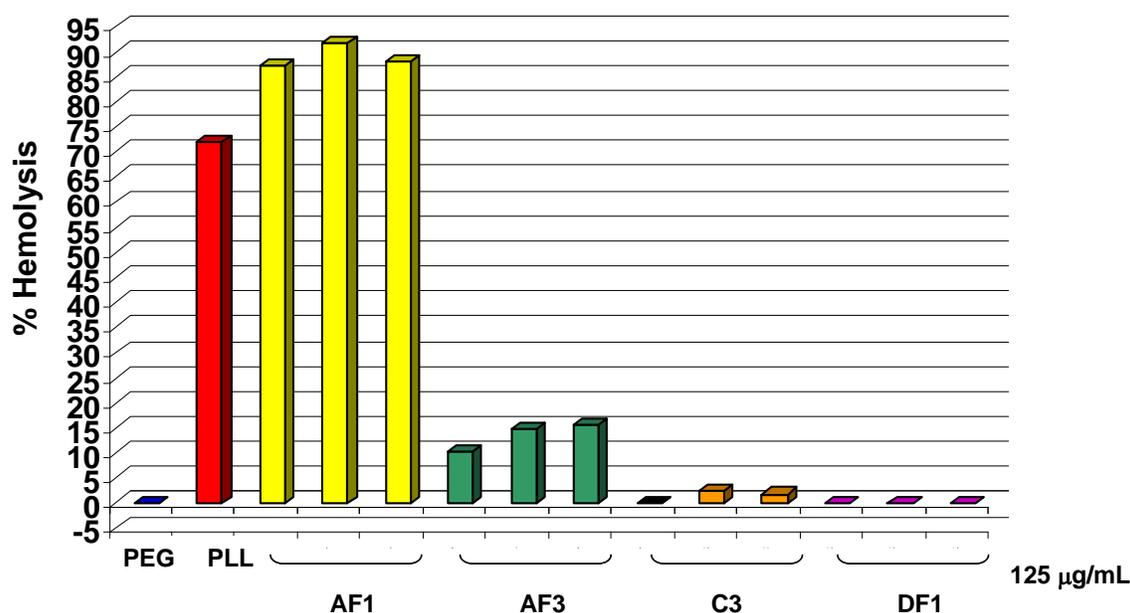


Figure 19. Analysis of nanoparticle hemolytic properties (ITA-1). 125 $\mu\text{g/mL}$ samples of AF1, AF3, C3 and DF1 were tested for effects on the integrity of red blood cells. Three independent samples were prepared for each nanoparticle sample and analyzed in duplicate (%CV<20). Each bar represents the mean of duplicate results. Poly-L-lysine (PLL) and polyethyleneglycol (PEG) were used as positive and negative controls, respectively. AF1 and AF3 were hemolytic, while C3 and DF1 did not disturb the integrity of red blood cells.

Analysis and conclusions

AF1 and AF3 are capable of inducing damage to red blood cells *in vitro*. C3 and DF1 do not have hemolytic properties when tested *in vitro*.

Nanoparticle Ability to Induce Platelet Aggregation (ITA-2)

Design and Methods

The objective of the study was to evaluate four C60 derivatives: AF1, AF3, DF1, and C3, in terms of their effects on human platelet aggregation *in vitro*. Stock solutions with concentrations of 1 mg/mL were used for each C60 derivative. The final nanoparticle concentrations after dilution according to the ITA-2 protocol were 200 $\mu\text{g/mL}$. Three independent samples were prepared for each nanoparticle formulation and analyzed in duplicate (%CV<20%). NCL protocol ITA-2 was followed. Collagen was used as positive control.

Results

At tested concentrations, none of the C60 derivatives induced platelet aggregation (all tested fullerene derivatives showed % aggregation <10). When platelet aggregation was induced by collagen, AF1 did not interfere with this process. AF3 resulted in approximately a 30% decrease in collagen-induced platelet aggregation. C3 and DF1 prevented platelet aggregation caused by collagen (Figure 20).

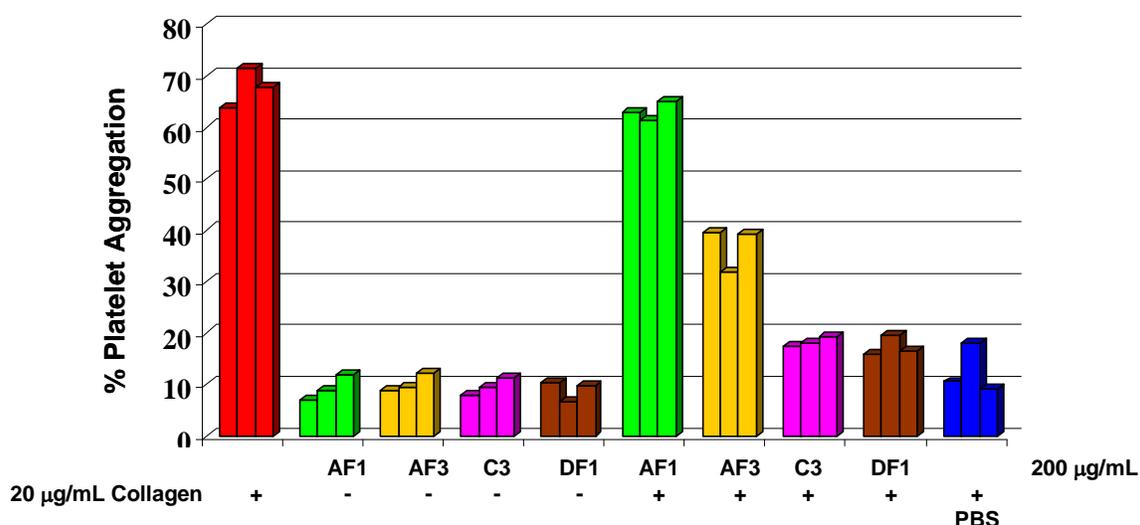


Figure 20. Analysis of nanoparticle ability to induce platelet aggregation (ITA-2).

AF1, AF3, C3 and DF1 at a final concentration of 200 $\mu\text{g/mL}$ were used to evaluate potential particle effects on the cellular component of the blood coagulation cascade. Collagen was used as positive control. The effects of the fullerene derivatives on collagen induced aggregation were studied by combining collagen with the nanoparticle treatments. Three independent samples were prepared for each nanoparticle formulation and analyzed in duplicate (%CV<20%). Bars represent the mean of duplicate results. None of the tested formulations induced platelet aggregation. AF1 did not interfere with collagen-induced aggregation, AF3 resulted in approximately a 30% decrease in collagen-induced platelet aggregation, and DF1 and C3 prevented aggregation of platelets in response to collagen.

Analysis and Conclusions

None of the C60 derivatives induced platelet aggregation. AF3 resulted in a decrease in collagen-induced platelet aggregation. C3 and DF1 prevented collagen-induced platelet aggregation.

Nanoparticle Toxicity to Bone Marrow Cells (ITA-3)

Design and Methods

The objective of this study was to evaluate the four C60 derivatives: AF1, AF3, DF1, and C3, in terms of their myelosuppressive effects. Since previous toxicology studies indicated DF1 was chemoprotective in kidney cell models, we also examined protection from cisplatin-induced myelosuppression. Stock solutions with theoretical concentrations of 1 mg/mL were used for each C60 derivative. The final nanoparticle concentrations of all formulations were 50 µg/mL. NCL protocol ITA-2 was followed. Hematopoietic stem cells were isolated from the bone marrow of 8-12 week old C57BL/6 mice. The cells were cultured for 12 days in MethoCult media supplemented with cytokines in the presence of PBS (a negative control), cisplatin (a positive control), and the C60 derivatives (to test for myelosuppressive properties) or a combination of cisplatin and the C60 derivatives (to test for chemoprotective properties). Two independent samples of each formulation were prepared and analyzed in duplicate.

Results

None of C60 derivatives was toxic to hematopoietic stem cells as indicated by undisturbed granulocyte and macrophage colonies formation (Figure 21). At tested concentrations, none of the C60 derivatives was capable of reducing or preventing cisplatin's toxicity to granulocyte and macrophages precursors (Figure 21).

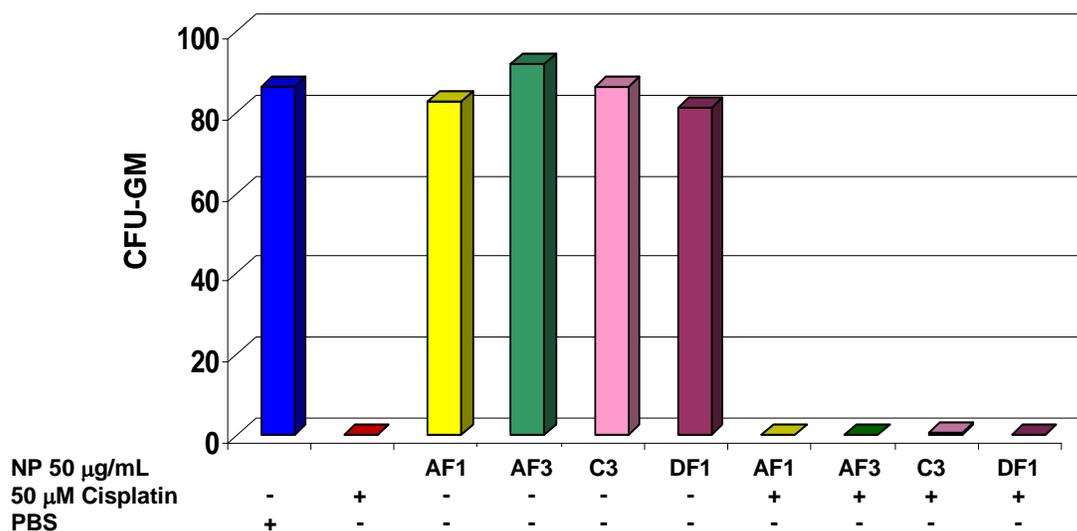


Figure 21. Analysis of nanoparticle toxicity to bone marrow cells (ITA-3). Two independent samples were prepared and analyzed in duplicate. Each bar represents the mean of duplicate responses (%CV <25). Results indicate that none of the tested particles was myelosuppressive or capable of protecting granulocyte-macrophage precursors from cisplatin cytotoxicity. NP stands for nanoparticle. All nanoparticle samples had concentrations of 50 µg/mL

Analysis and Conclusions

None of C60 derivatives was toxic to hematopoietic stem cells or protective from cisplatin's toxicity to granulocyte and macrophages precursors

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Abbreviations

2D PAGE Two-Dimensional Polyacrylamide gel Electrophoresis
⁵¹Cr ⁵¹ Chromium
 ABN Abnormal Plasma Standard
 AFFF Asymmetrical Field Force Fractionation
 APTT Activated Partial Thromboplastin Time
 ATCC American Type Culture Collection
 AUC Area Under the Curve
 CBA Cytometric Bead Array
 CE Capillary Electrophoresis
 CFU-GM Colony Forming Units-Granulocyte Macrophage
 Cisp Cisplatin
 CP Cisplatin
 CV Coefficient of Variation
 CVF Cobra Venom Factor
 DHB 2,5-dihydroxybenzoic acidDI
 (H₂O) Deionized H₂O
 DLS Dynamic Light Scattering
 E:T ratio Effector: Target Ratio
 ELISA Enzyme-Linked Immunosorbent Assay
 FBS Fetal Bovine Serum
 FDA Food and Drug Administration
 HEP-G2 Human Hepatocarcinoma Cells
 HPLC High Performance Liquid Chromatography
 IL InterleukinK562 Human Erythroleukemia Cell Line
 kDa Kilodalton
 kV KiloVolt
 LAL Limulus Activation Lysate
 LDH Lactate Dehydrogenase
 LEI Laboratory of Experimental Immunology
 LLC-PK1 Porcine Renal Proximal Tubule Cells
 LPS Lipopolysaccharide
 MALDI-TOF Matrix Assisted Laser Desorption/Ionization – Time of Flight
 mg Milligram
 mL Milliliter
 mm Millimeter
 MS Mass Spectroscopy
 MTT 3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2H-tetrazolium bromide
 N Normal plasma standard
 NCL Nanotechnology Characterization Laboratory
 NIST National Institute of Standards and Technology
 NO Nitrous Oxide
 PBMC Peripheral Blood Mononuclear Cells
 PBS Phosphate Buffered Saline
 PHA-M Phytohemagglutinin-M
 psi Pounds per Square Inch
 PT Prothrombin Time
 RDG Rayleigh-Debye- Gans
 RAW 264.7 Mouse Leukemic Macrophage Monocyte Cell Line

ROS Reactive Oxygen Species
RT-CES Real Time- Cell Electronic Sensing
SAIC Science Applications International Corporation
SPGR Spoiled Gradient Echo Sequences
Unt Untreated
UV-Vis Ultra Violet- Visible
v/v volume/volume

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