



Arsenic in the Landfill Environment: *Untested Ideas and Open Questions*

SBRP Arsenic Conference
Arsenic and Landfills: Protecting Water
Quality
Boston, MA
October 3-4, 2006

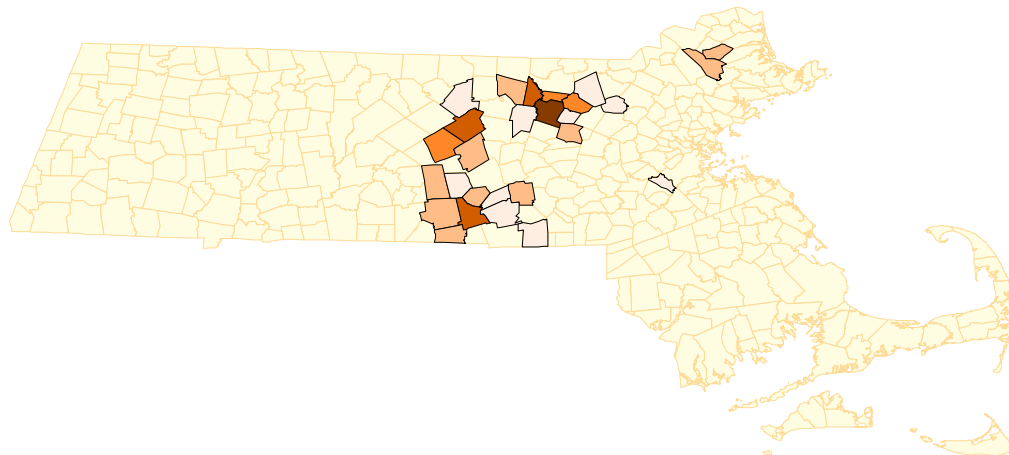
Bill Brandon, EPA NE

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- Region 1 Laboratory
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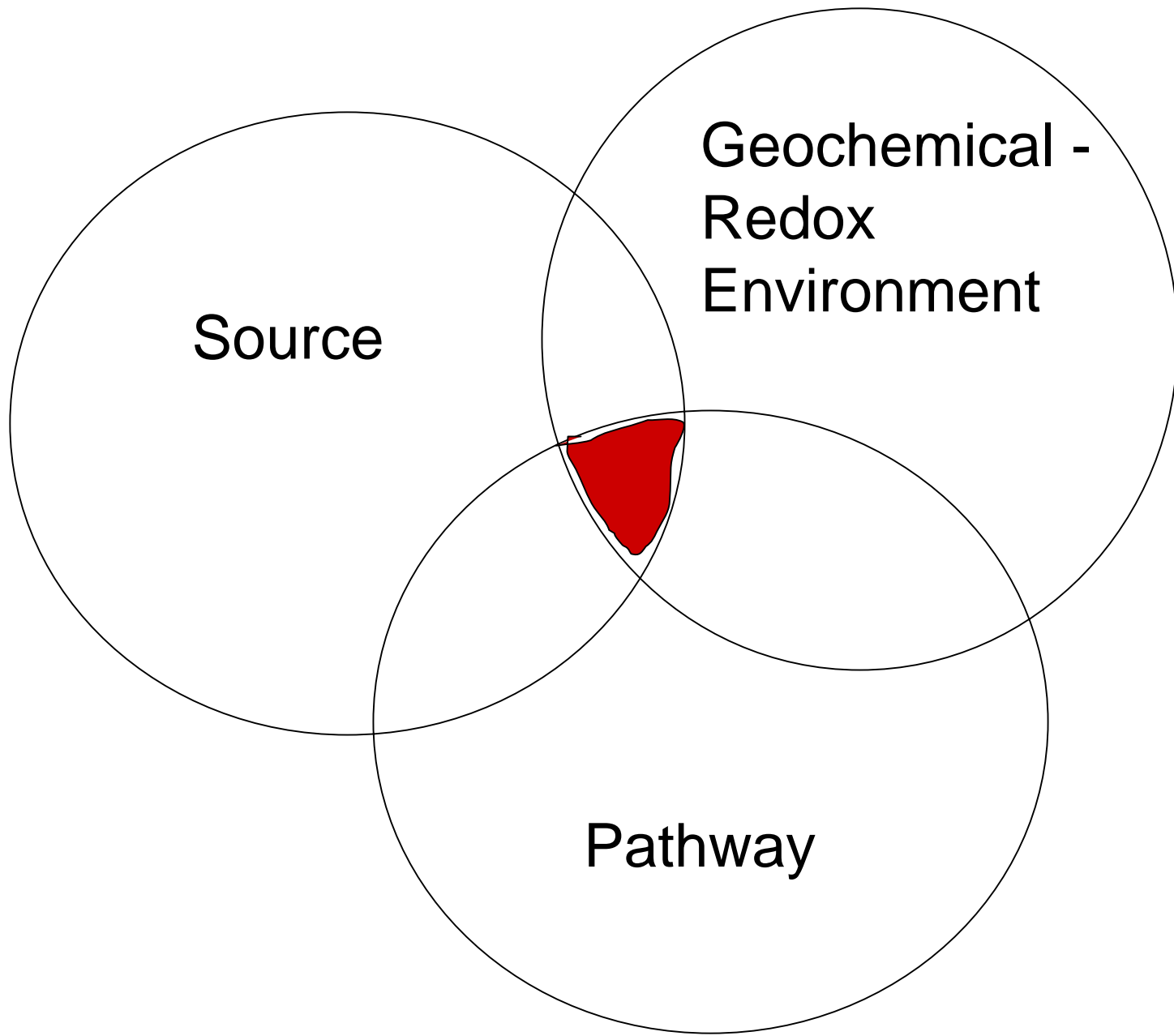
Introduction

- Central Mass. Landfill Averages on the order of 500 ppb....
- SHL (As up to **6000+** ppb)....
- What is Different about SHL ??



Introductory Questions

- Source = Waste Materials ?
- Source attributable to Particular Geologic Circumstances ?
- Both ?
- Or, Is this landfill just better characterized than the norm ??
- Does “Solvent Plume” paradigm hamper Conceptual Site Model (CSM) development ?
- Significance of Redox Environment ?



Source

Geochemical -
Redox
Environment

Pathway

Conceptual Site Models (CSMs)

- ‘Popular’ CSMs have emerged but none fully supported by data
- ‘Overlap’ of existing CSMs
 - Some elements common to more than one
- We don’t have all the answers!
 - “Challenge questions” posed

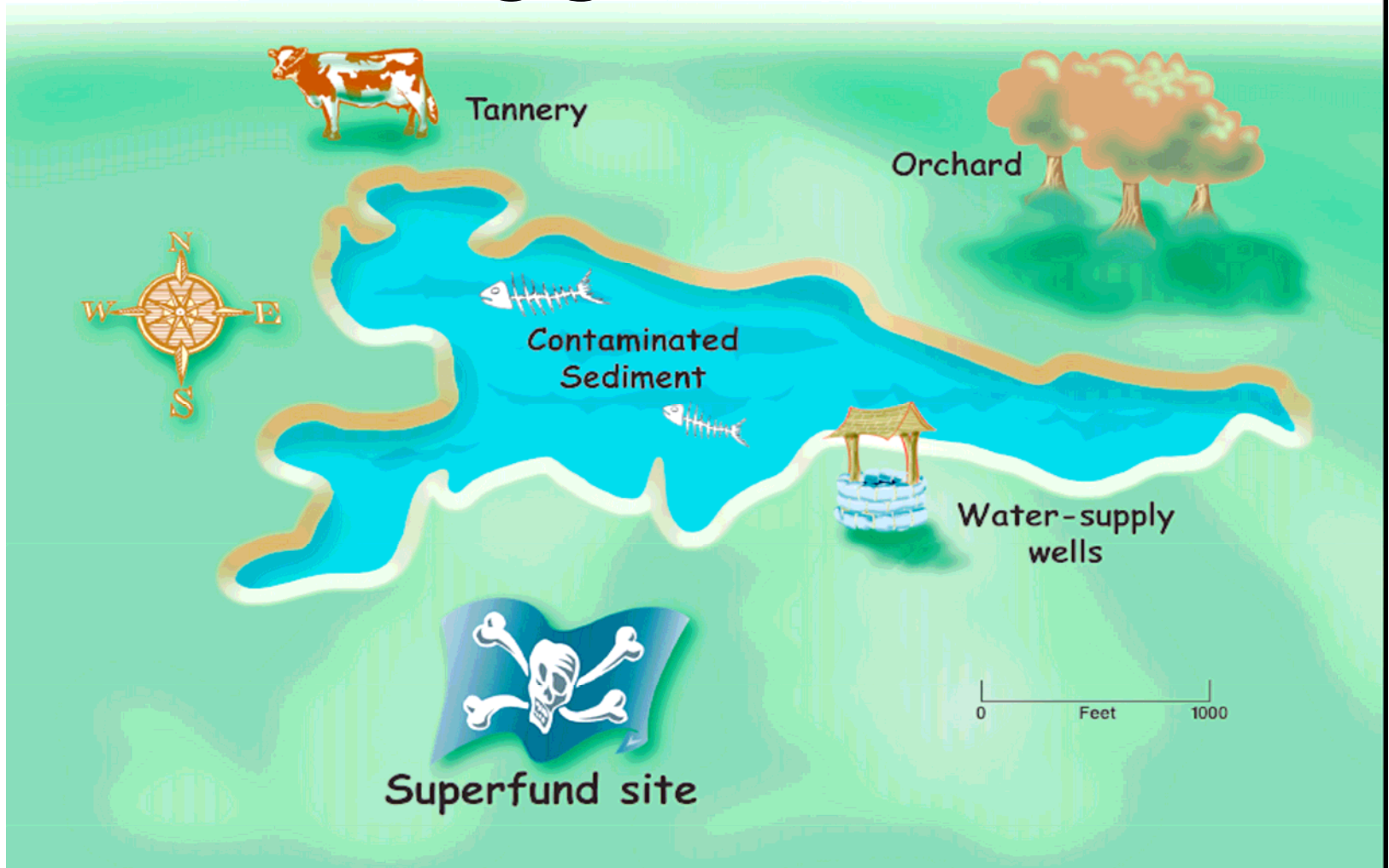
CSM Inputs

- Sources
- Geologic Characterization/Solid Phases
- Hydraulic Characterization/System
- Redox Environment/System
- Contaminant Migration Pathways
- Contaminant Transformation Pathways
- Intermediate Contaminant Fate - Accumulation
- Receptors – Risk Assessment
- Other ?

When MNA Isn't Working... (CSM Updates Needed)...

- increase in concentrations
- detection of contaminants outside of known plume boundary
- rate of decrease is not as expected
- changes in land and/or groundwater use

CSM - v.1



Relative Arsenic Concentrations in Grove and Plow Shop Ponds



0 255 510 1,020 1,530 2,040 Feet

0 80 160 320 480 640 Meters

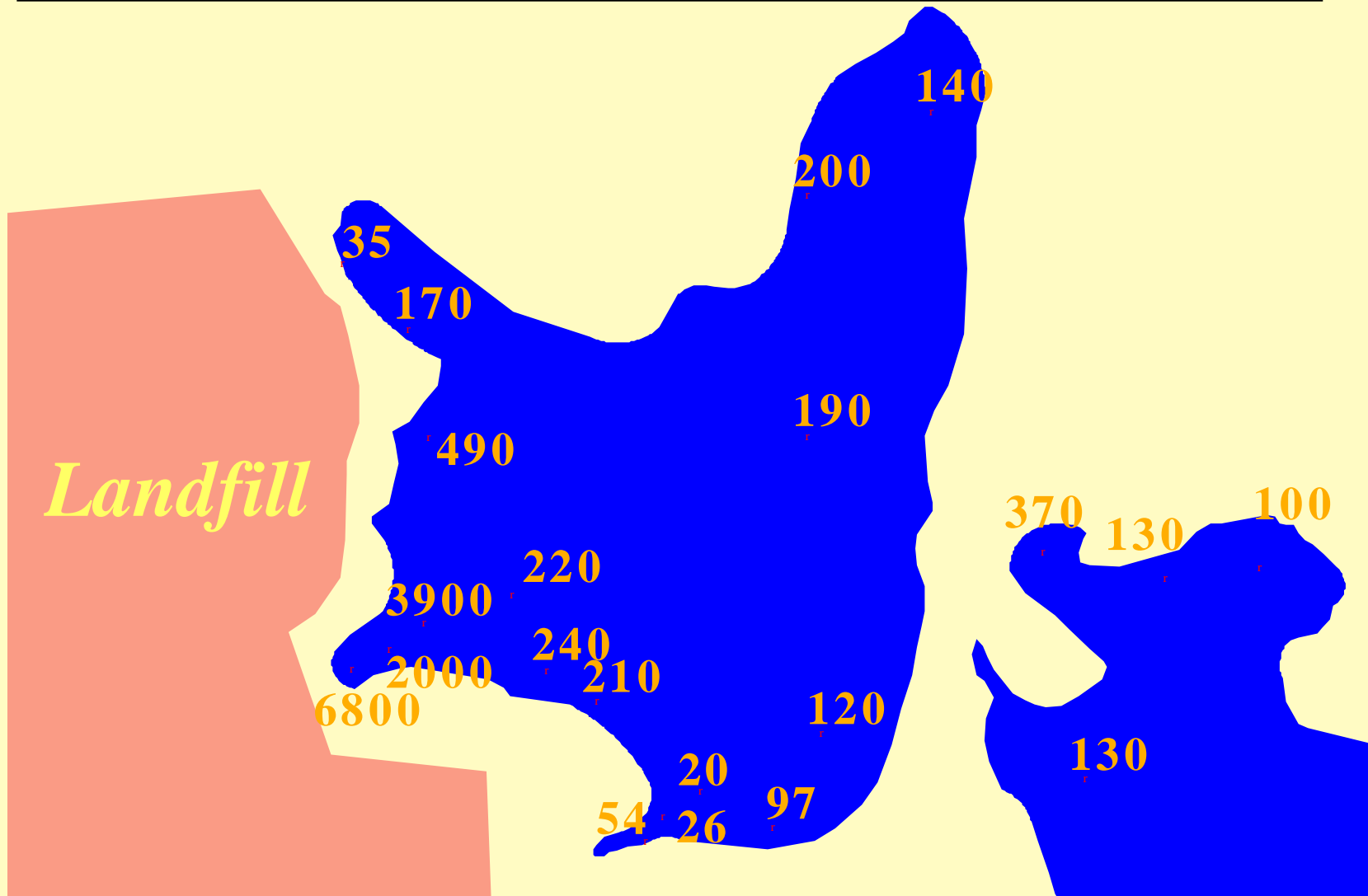
devensdsmarch04_mb_edit Events

ARSENIC





Arsenic in Bottom Sediments mg/Kg



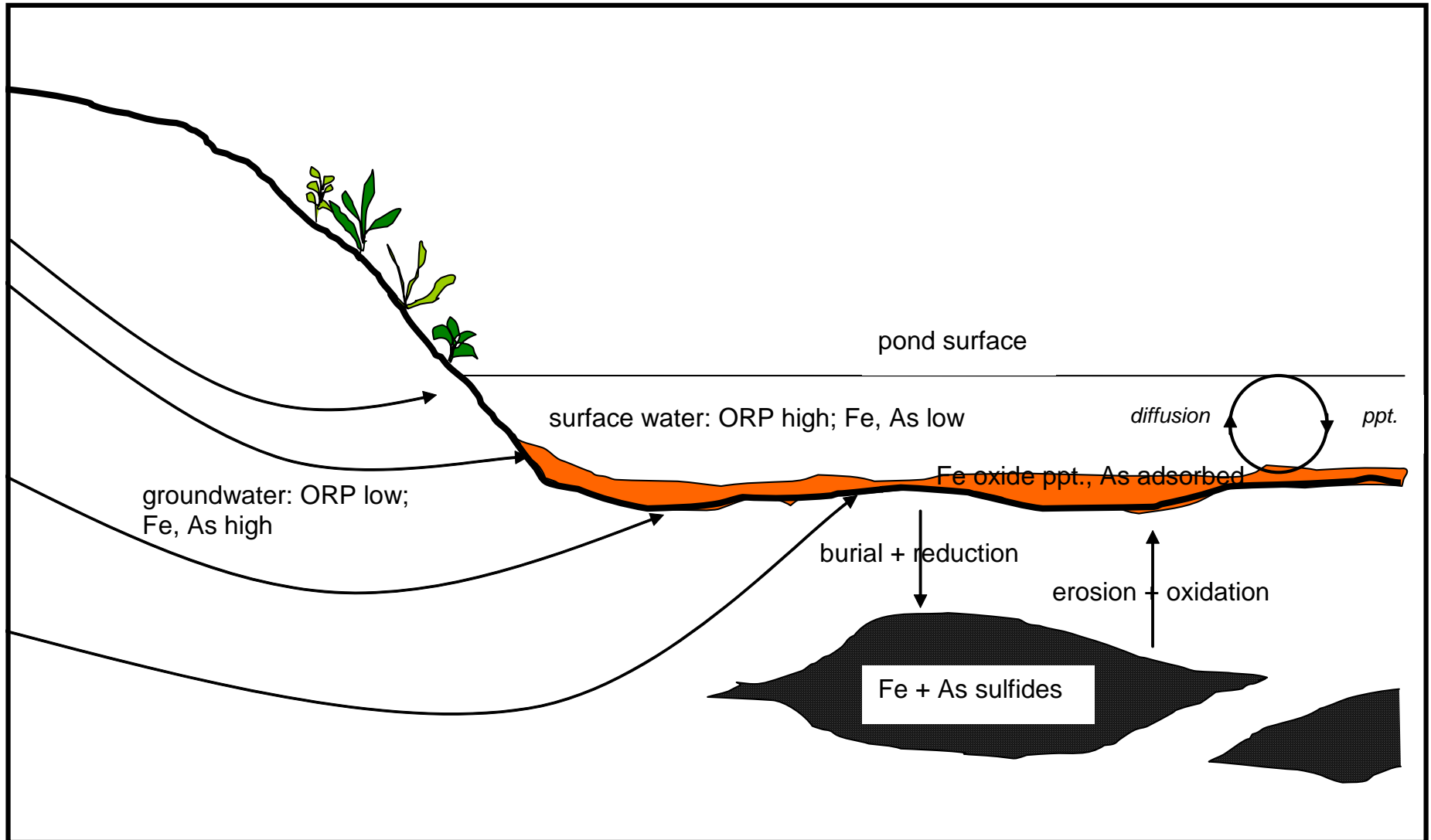
Toxicity Test Results



Working Conceptual Model #1

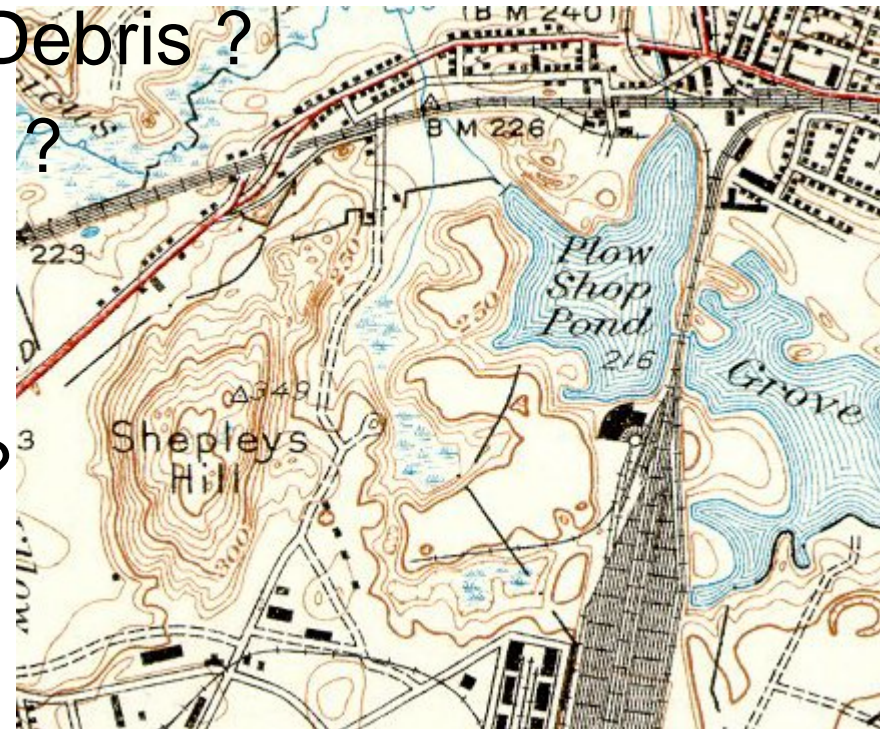
- Arsenic is present in some bedrock lithologies
- Glacial transport and postdepositional alteration redistribute Fe and As
- As is sorbed by hydrous ferric oxides (HFO) in overburden
- Landfills lower ORP of downgradient groundwater by oxidation of organics
- HFO dissolves (“reductive dissolution”), liberates arsenic

Unified Conceptual Model- Red Cove



Potential Anthropogenic Sources Need Additional Consideration

- Coal Ash (locomotives) ?
- CCA treated wood/Ash (on-site incinerator) ?
- Wall Board/Construction Debris ?
- Treatment Plant residuals ?
- Arsenical Pesticides ?
- Rat Poison (As_2O_3)?
- Other Hazardous Waste ?
- Other ?



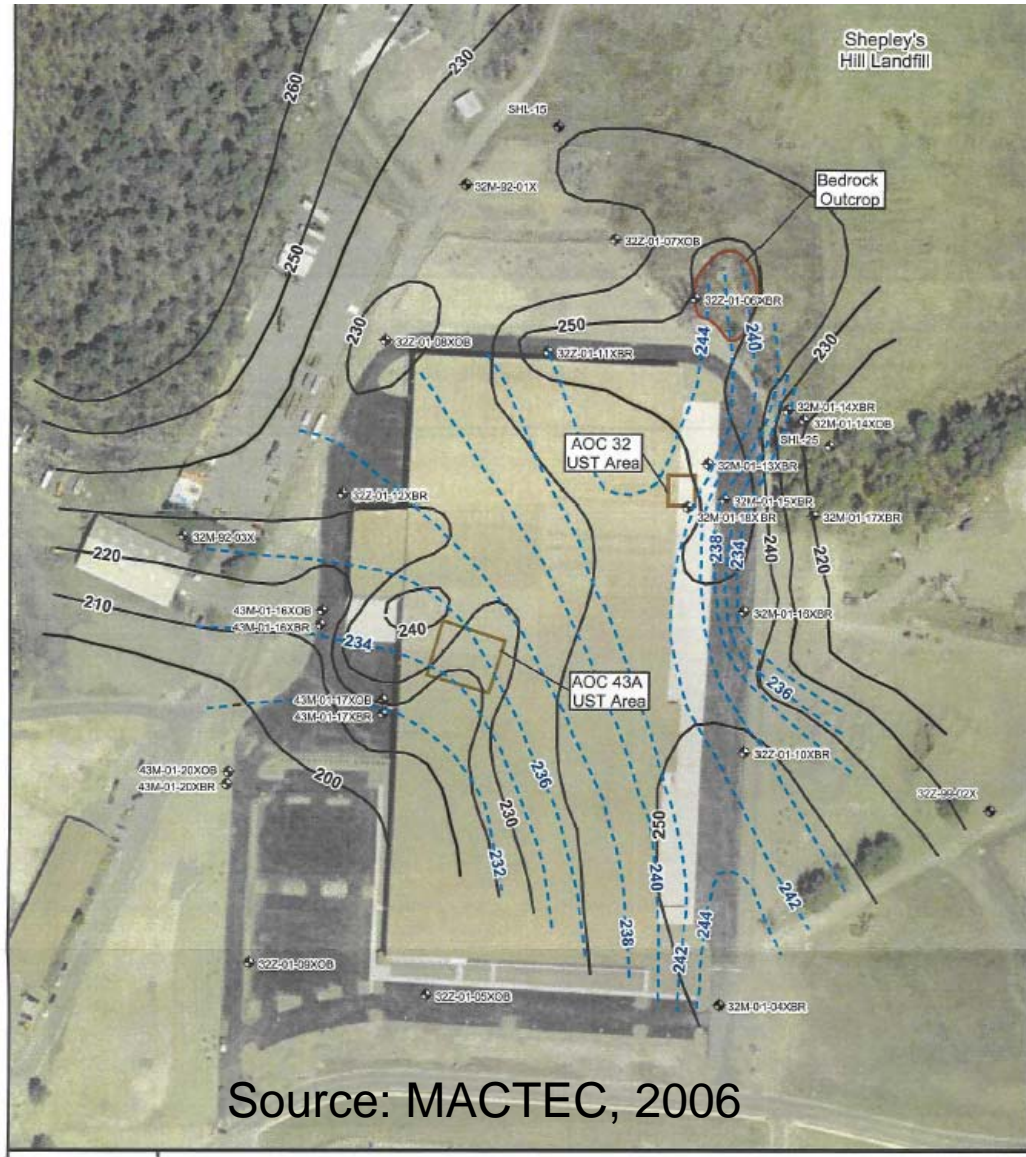
Potential Geologic Sources

- HFO coatings on outwash sand grains ?
- Rock Chips in outwash sands ?
- Bedrock Fractures Coatings (oxides) ?
- Disseminated Sulfides ?
- Mineralized zones (veins, shear zones, etc.) ?
- All of the above ?
- Effects of Blasting ?

Blasting Presents Fresh Exposures



Interpretive Bedrock Groundwater Surface Map, October 7, 2004



Pre-Blast Bedrock Exposures at SE Corner of Building Area





Hydrologic Issues

- Impoundments
- Cap Performance ?
- “Run-under” from Shepley’s Hill
- Pumping Wells
- Engineered Drainage
- Impervious Surfaces
 - Pavement
 - Landfill caps
- TIME SCALES !??



Storm Drain Installation



Fill Emplacement SW of Building Footprint

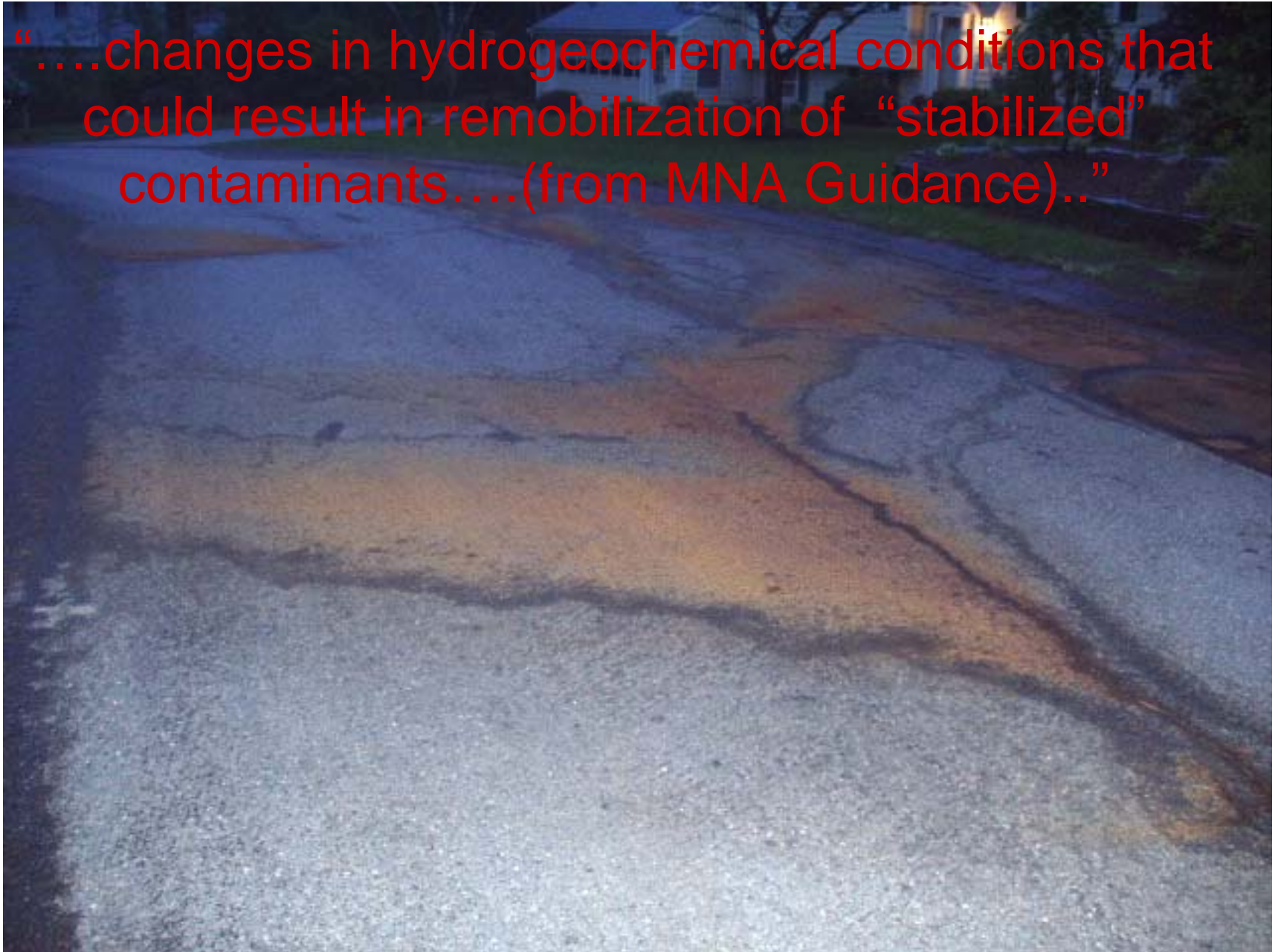


Hydrology Affects Geochemistry

(R. Ford, SFBR, 2006)

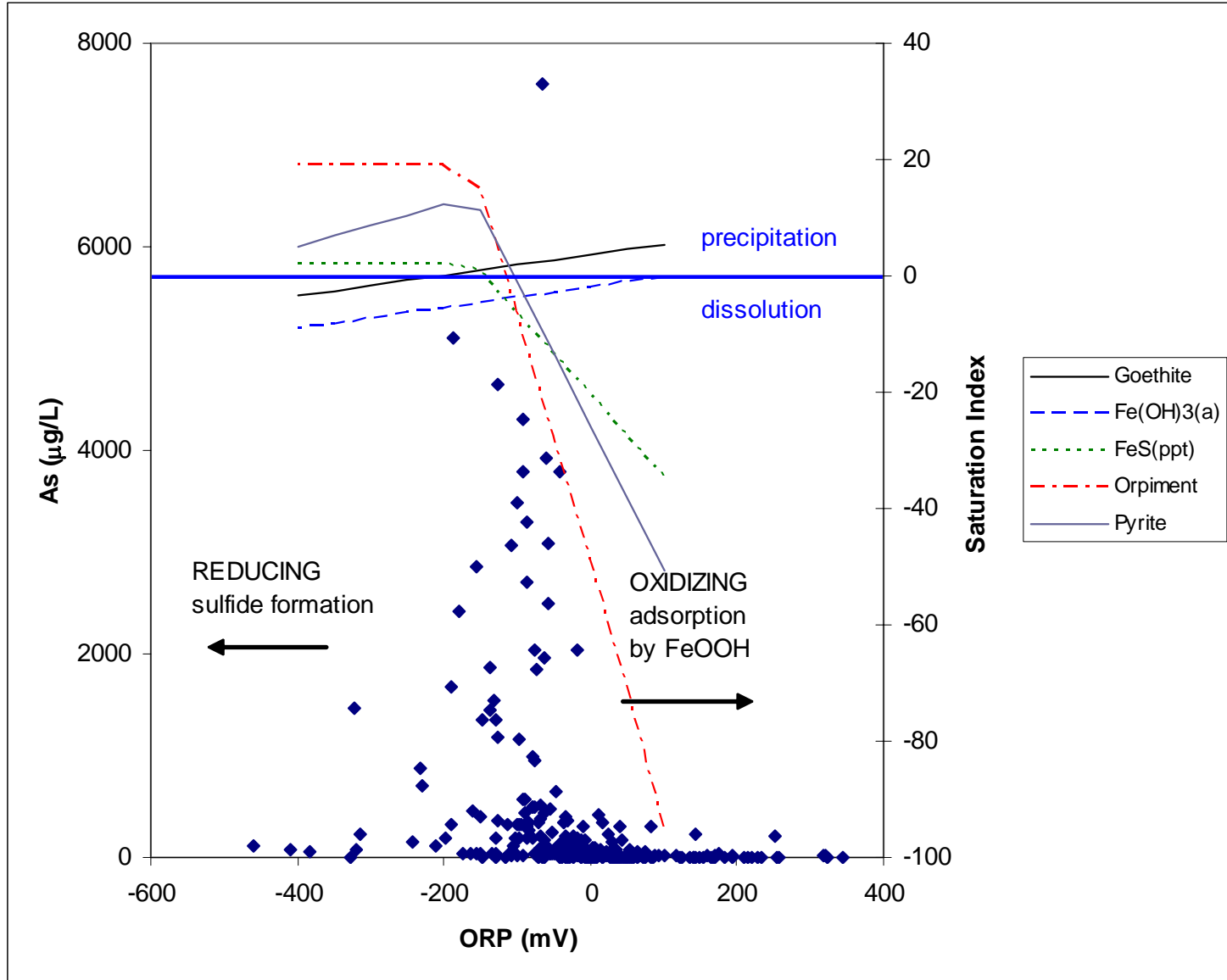
- Infiltration of precipitation during operational lifetime ('oxic reaction front')
- Fluctuations in water table for closed landfill
 - Cause alternation between oxidizing and reducing conditions
- Internal redox shifts within landfill mass during degradation of waste material coupled with seasonal fluctuations in microbial activity
- *Dynamic fluctuations will tend to maintain arsenic mobility*

“....changes in hydrogeochemical conditions that could result in remobilization of “stabilized” contaminants....(from MNA Guidance)..”

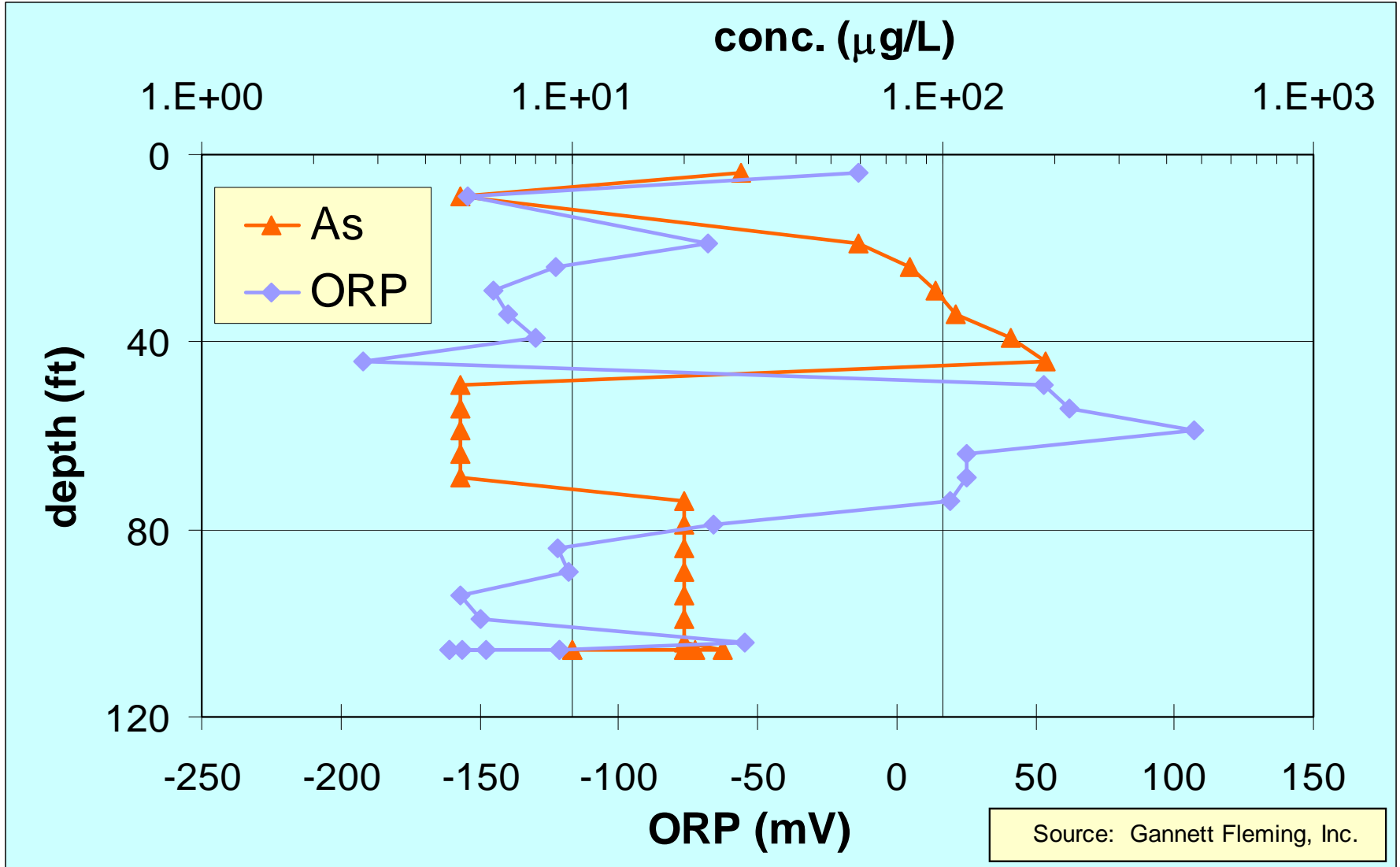


SHL Groundwater Geochemistry

DANGER ZONE

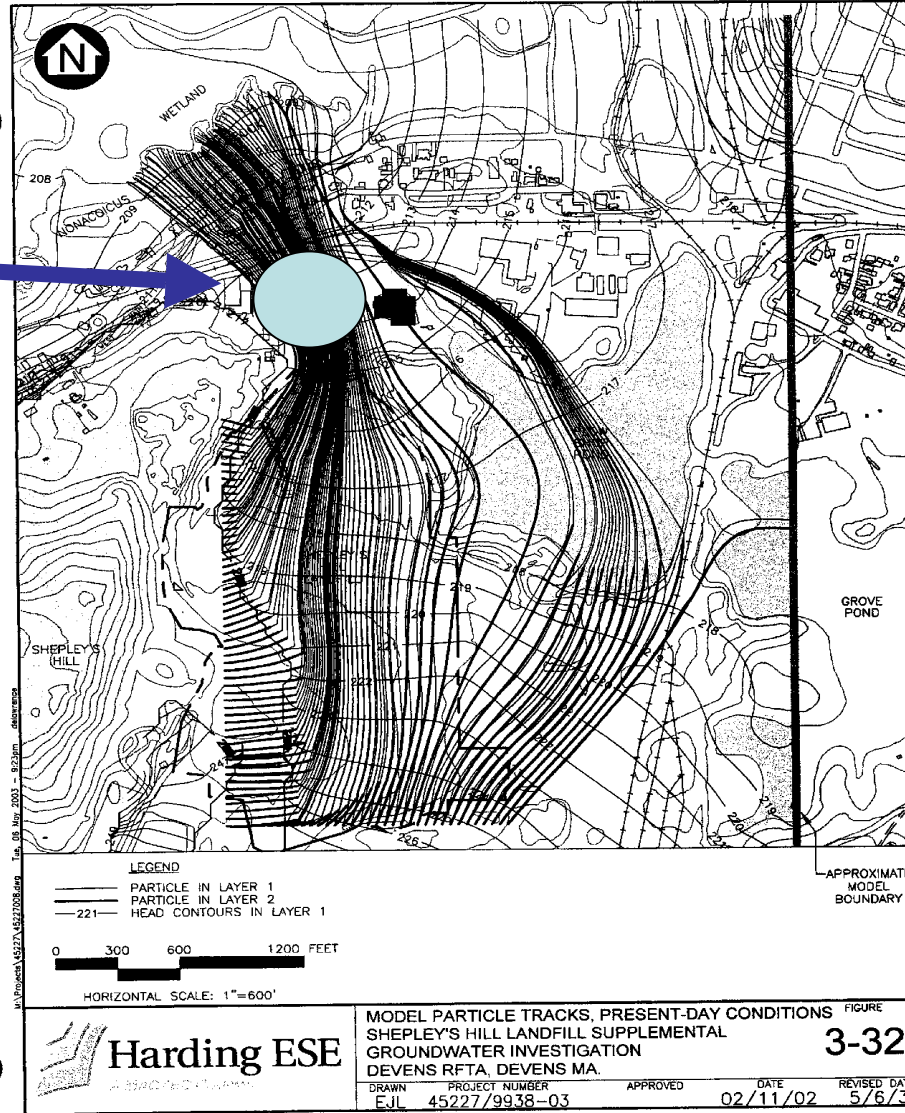


As and ORP in Ground Water



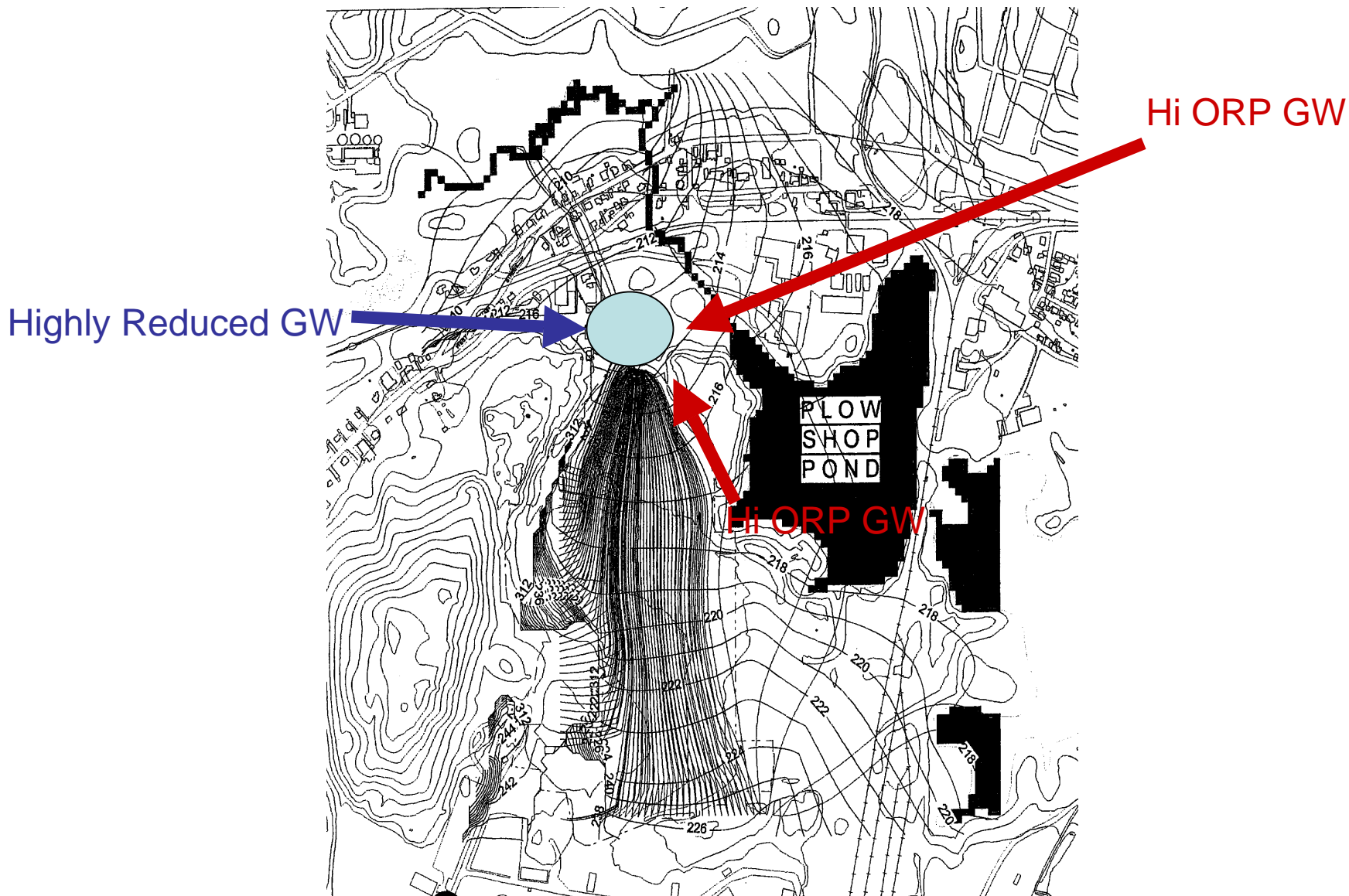
SHEPLEY'S HILL LANDFILL MODELED PARTICLE TRACKS, CURRENT CONDITIONS

Highly
Reduced GW

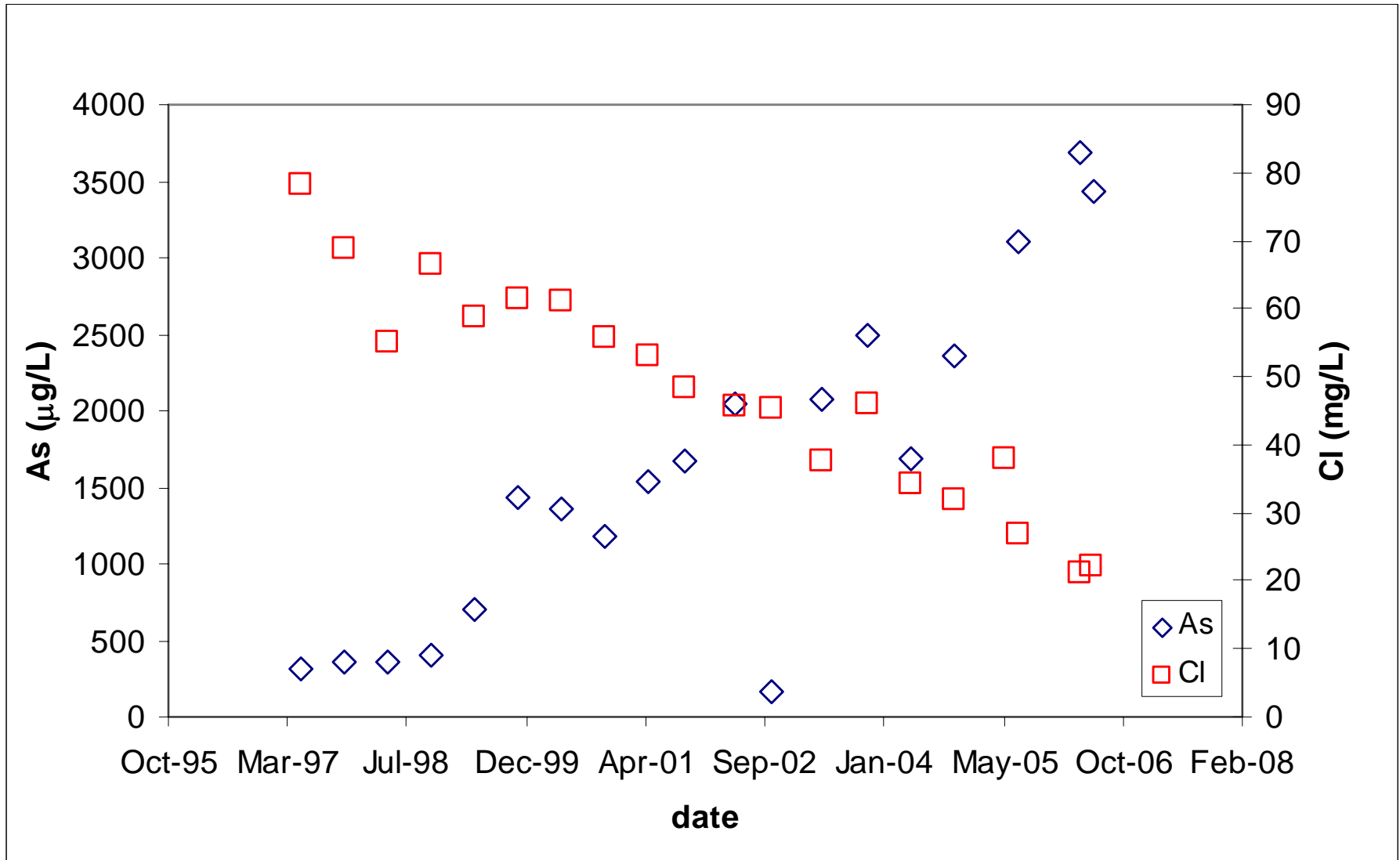


SHEPLEY'S HILL LANDFILL

MODELED PARTICLE TRACKS, RUN 106C, 50 gpm



SHM-96-22B



Characterization “Quality” Issue

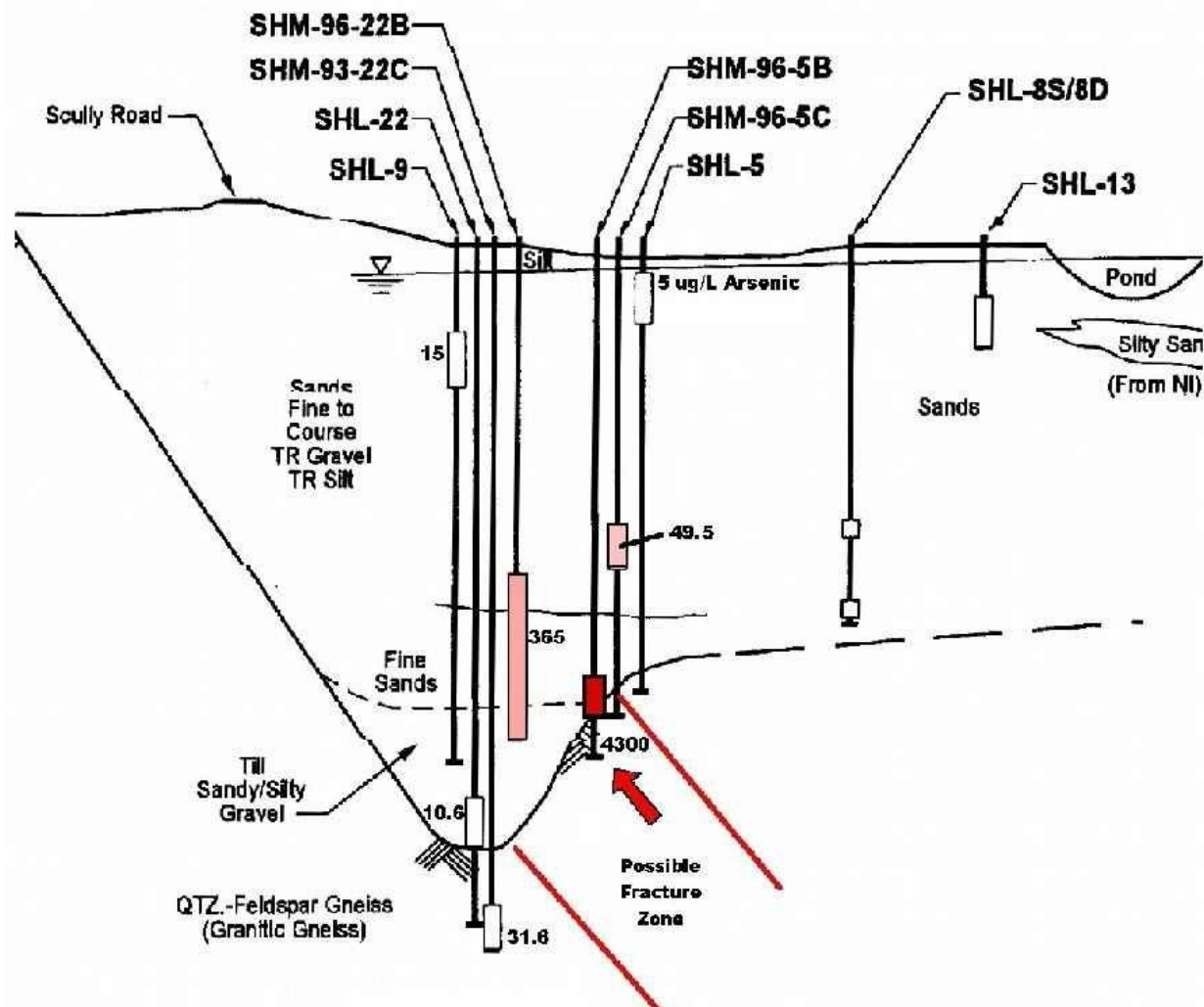
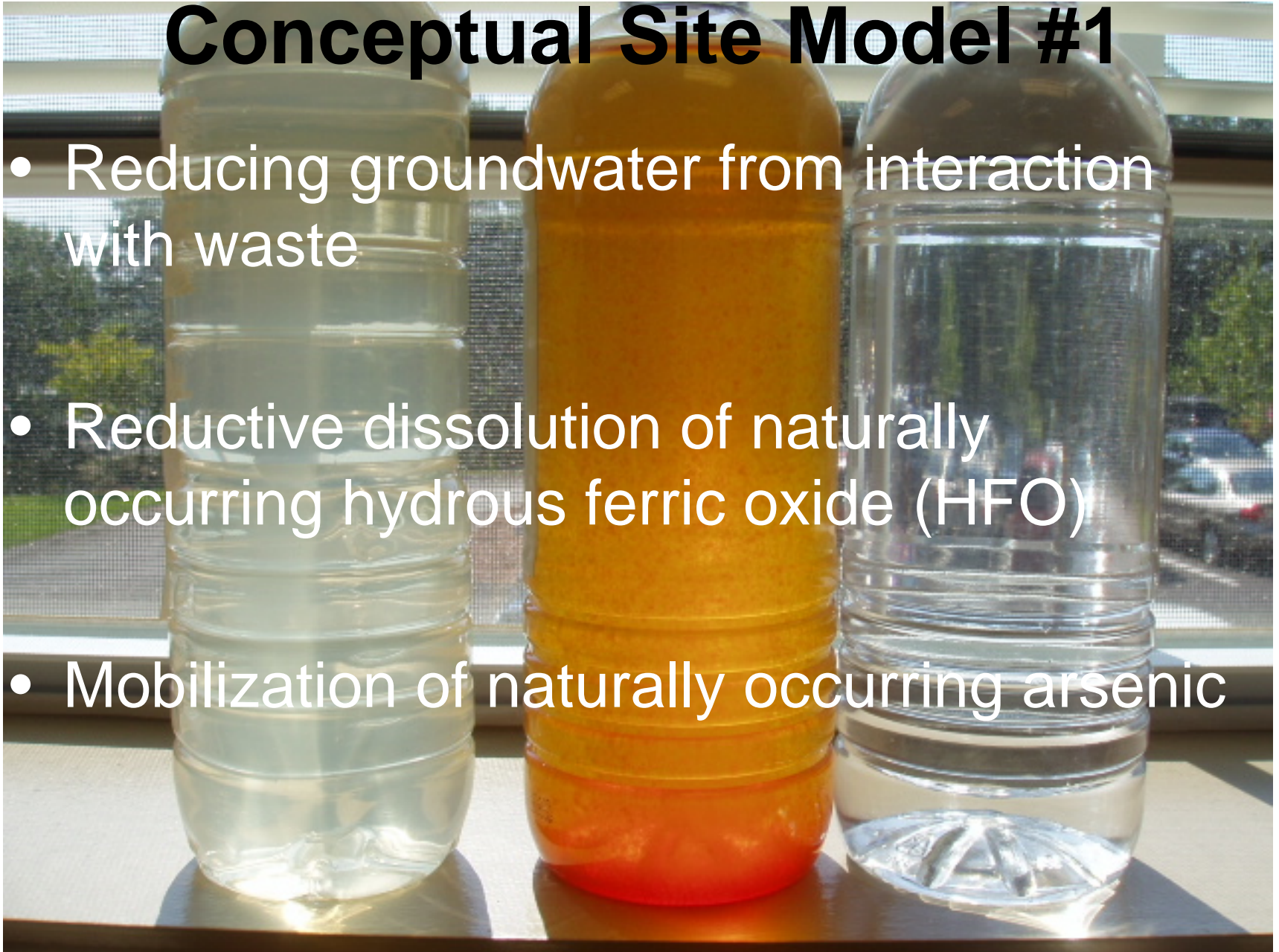


Figure 3. Spatial relationships and relative concentrations of arsenic-bearing wells on the northern boundary of Shepley's Hill Landfill.

Deeper red translates to greater arsenic concentration.

Conceptual Site Model #1

- Reducing groundwater from interaction with waste
- Reductive dissolution of naturally occurring hydrous ferric oxide (HFO)
- Mobilization of naturally occurring arsenic



Conceptual Site Model #1

Questions

- What data support/refute cause of reducing conditions (waste vs. naturally occurring)?
- If waste/GW interaction generates low-ORP 'plume'
 - Can 'plume' be mapped self-consistently?
 - Is 'plume' consistent with other leachate indicators, e.g. Cl, conductance?
- Why is As deep?
 - Density flows?
- Other tracers for waste interaction?
 - Rationalize SHM-96-22B (As increasing, Cl decreasing)

Conceptual Site Model #2

- Arsenic source present in waste
 - Suggested by unusually high As at SHL
 - Incomplete waste characterization
- Adds to naturally occurring As mobilized by reductive dissolution (CSM#1)



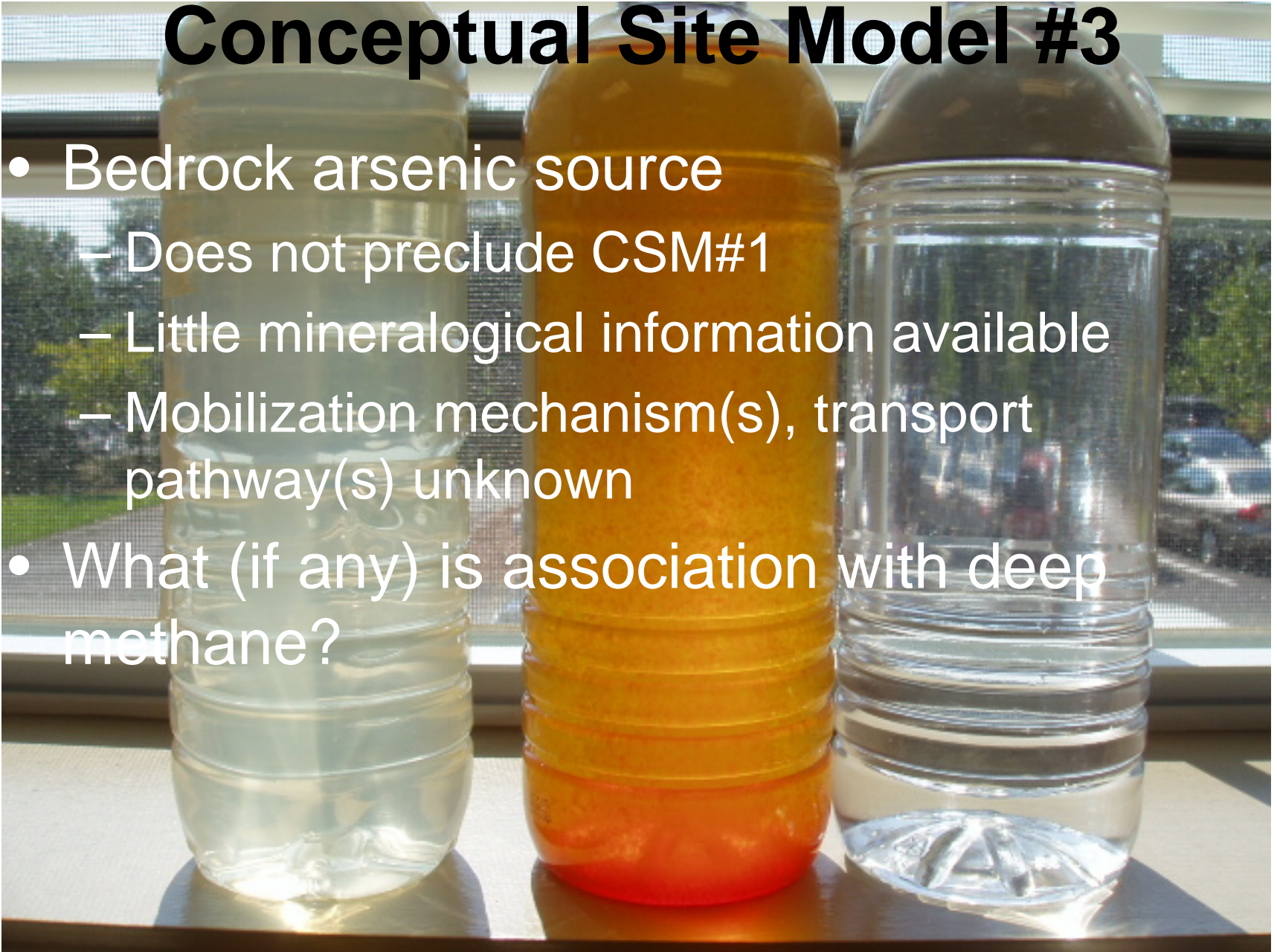
Conceptual Site Model #2

Questions

- What are candidate materials?
 - Coal ash: Where are associated trace elements? Why only elevated As?
 - Pesticides (e.g., As_2O_3): Why correlated with Fe in soil and groundwater? Is gross mass balance consistent with pesticide application? Surface application, now capped, cannot be continuing source to groundwater ?
- Why is high As found in deep GW, when waste interaction is shallow?

Conceptual Site Model #3

- Bedrock arsenic source
 - Does not preclude CSM#1
 - Little mineralogical information available
 - Mobilization mechanism(s), transport pathway(s) unknown
- What (if any) is association with deep methane?



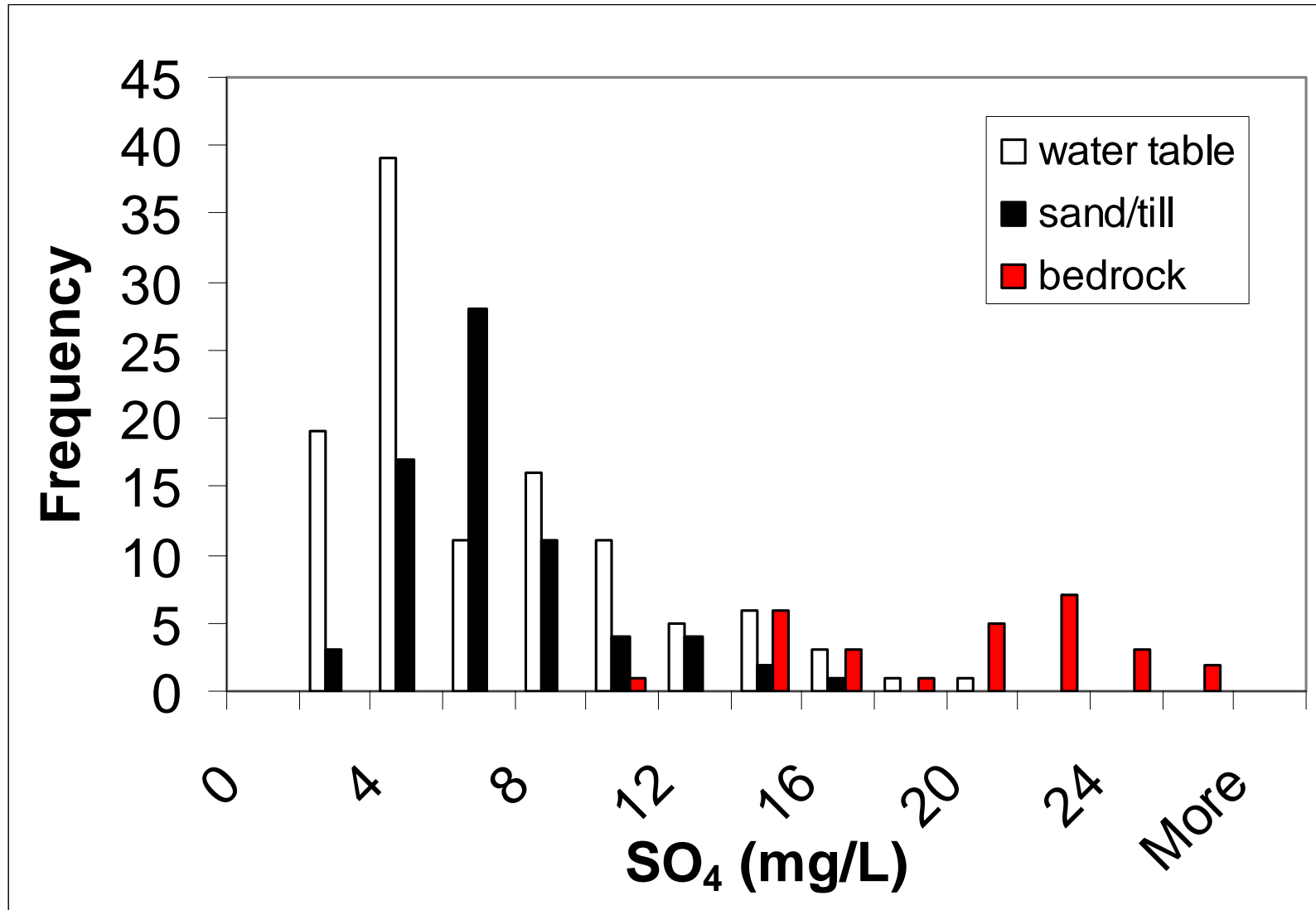
Overarching Questions Regarding Waste-Groundwater Interaction

- Little apparent geochemical signature of waste interaction
 - Relatively little physical interaction (small saturated volume)
 - Waste is relatively chemically unreactive
- General increase with depth: major-element concentration, TDS, sp. cond., alkalinity, etc.
 - Density?
 - Universal observation: deeper = older, longer residence time (Freeze and Cherry, 1979, p. 241)

A photograph of a bright blue sky filled with various cloud formations. The clouds range from small, wispy white clouds to larger, more dense grey and white cumulus clouds. The lighting suggests a bright, sunny day.

Your
Thoughts ?

LTMP Geochemical Data



Groundwater Characteristics

	Specific Conductance (mS/m)	Alkalinity (mg/L CaCO ₃)	Chloride (mg/L)	Na (mg/L)	Ca (mg/L)	SO ₄ (mg/L)
Wisconsin landfills [1]	284 – 1585	960 - 6845	180 - 2651	12 - 1630	200 - 2100	8.4 - 500
SHL [2]	2.1 – 148	3 - 670	1U - 65	2U - 48	1.8 - 140	1.1 - 20.9 [3]
Grove Pond [4]	21 - 364	20.2 - 182	0.5 - 111	16 – 54.3	2.9 – 74.5	0.1 – 44.2

[1] From Fetter, 1994; typical ranges of site medians

[2] SHL PMP/LTMP data

[3] Average of LTMP data, by well, 5/98-6/05

[4] From *Grove Pond Arsenic Investigation*, Gannett Fleming 2002

Arsenic Speciation

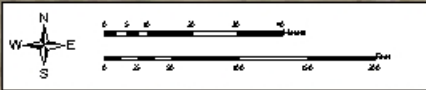
- Inorganic Species
 - As(III): H_3AsO_3^0
 - As(V): H_2AsO_4^- , HAsO_4^{2-}
- Organic Species
 - Monomethyl arsenic (MMA)
 - Dimethyl arsenic (DMA)
 - Arsenobetaine (AsB)

Which forms will be analyzed?

Questions Addressed by Arsenic Speciation Analysis

- Arsenic sources?
 - From waste?
 - Reductive dissolution of HFO?
- Carbon source (for organic species)?
 - C in waste vs. C in peat (“young” vs. “old”)?
- Risk issues
 - Bioavailability (organic or inorganic species)?
- How will data be used?
- Sampling and analytical considerations

Red Cove Sampling Locations



FLOW SHOP POND

AYER

- Legend**
- Pushpoint Sample Locations
 - Sediment Samples - Mar04
 - Sediment Samples - Feb05
 - Geoprobe
 - Surface Water Sampling - Nov04
 - Monitoring Wells



Red Cove

Sediment and Pore Water Arsenic

