



U.S. Department of Energy  
Office of Civilian Radioactive Waste Management



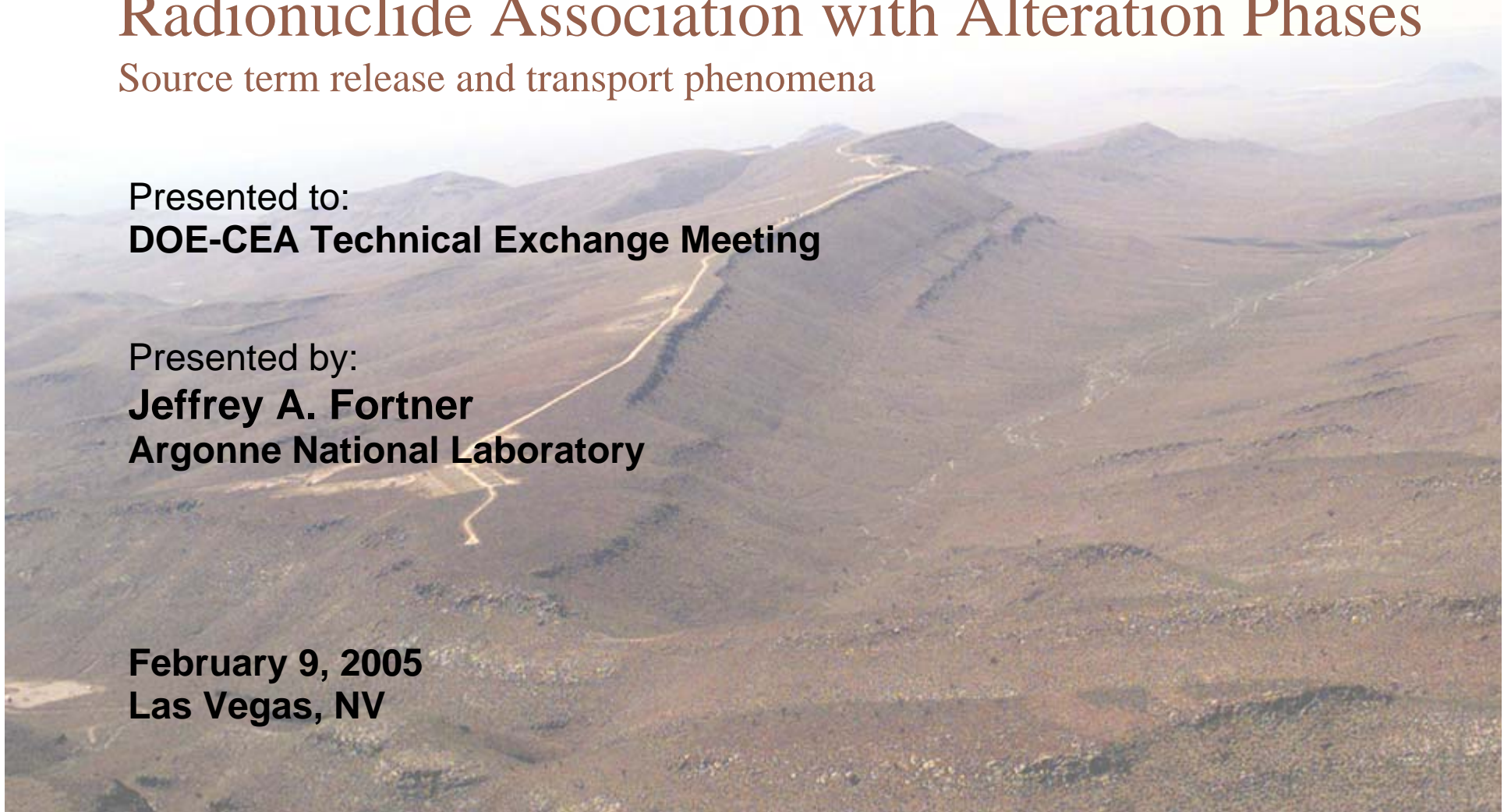
# Radionuclide Association with Alteration Phases

Source term release and transport phenomena

Presented to:  
**DOE-CEA Technical Exchange Meeting**

Presented by:  
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**Las Vegas, NV**



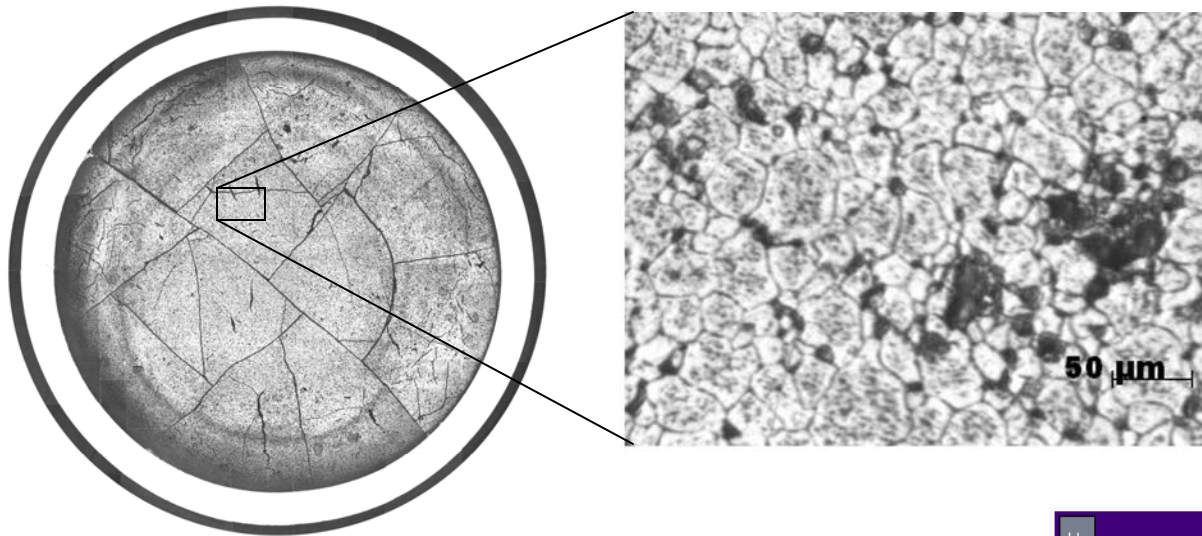
# Objectives

- **What is known about solid-state chemistry of trace radionuclides in a uranium-dominated system?**
- **Will sequestration of radionuclides reduce dissolved concentrations in the environment?**
- **Can one predict, from initial conditions of spent fuel in a repository, the retention of these radionuclides into alteration phases?**



# Structure of Spent Nuclear Fuel (SNF)

Mix of grains, grain boundaries, fission gas, and “gap” regions



*Images courtesy  
Hanchung Tsai,  
Argonne National Lab*



H																	He	
Li	Be											B	C	N	O	F	Ne	
Na	Mg											Al	Si	P	S	Cl	Ar	
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn	
Fr	Ra	Ac																
			Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu		
			Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr		

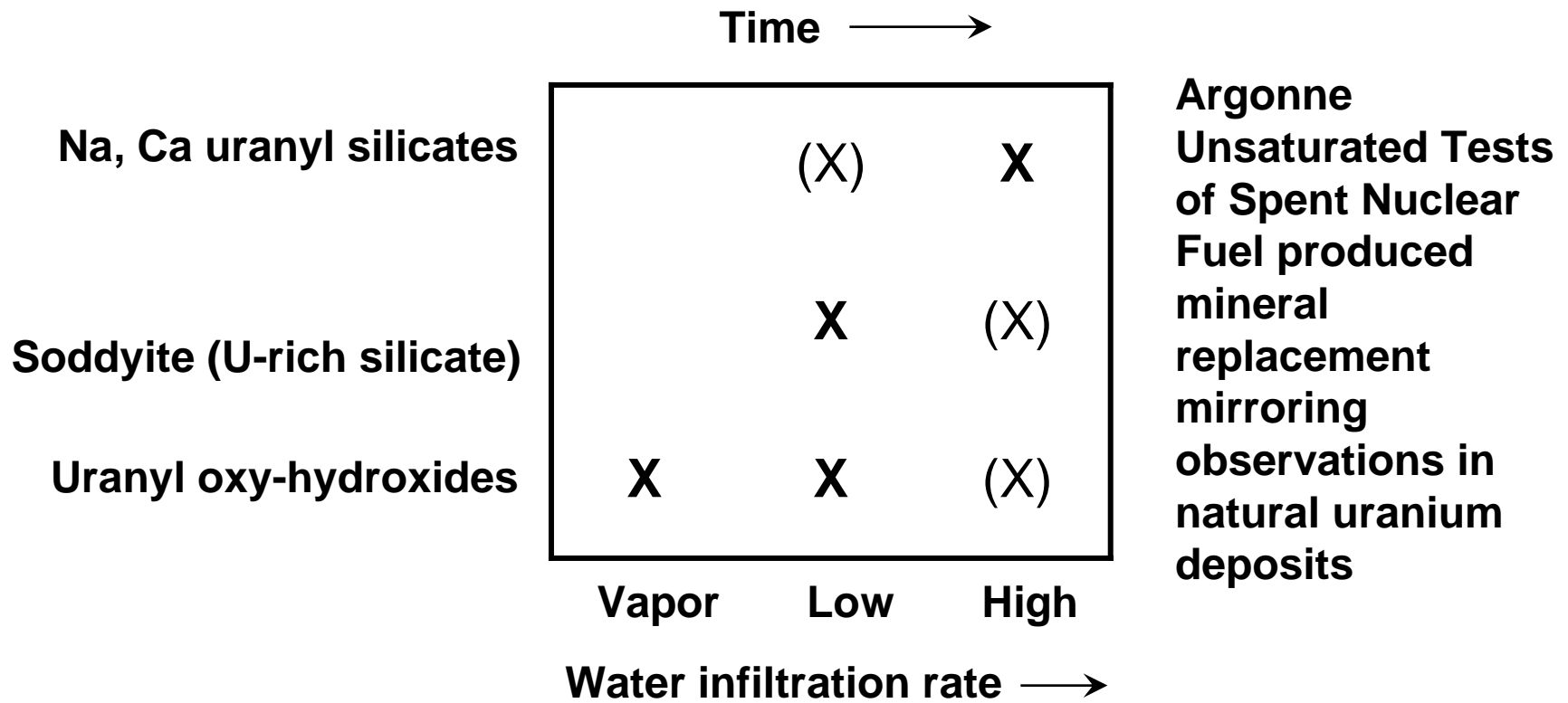


# Radionuclides in Fuel Alteration Phases

- **Uranium oxide fuel in a waste repository is expected to behave analogously to natural uranium deposits**
- **However...man-made elements (Np, Tc, Pu, etc.) have few natural analogs.**
  - **What happens to the neptunium, plutonium, technetium as spent nuclear fuel corrodes?**
    - ◆ **Can laboratory measurements predict behavior over geologic scales of time and distance?**



# Alteration Phase Paragenesis



After Finch *et al.*, 1999





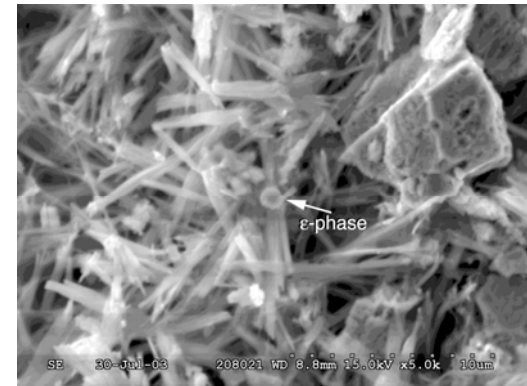
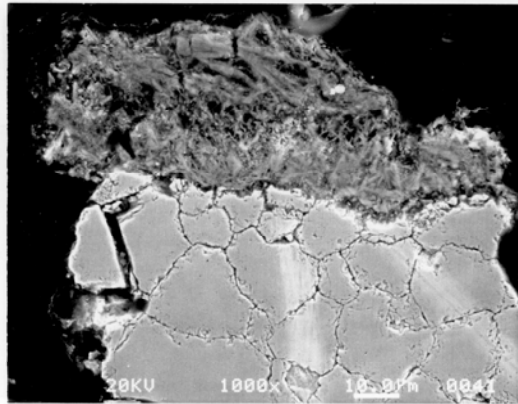
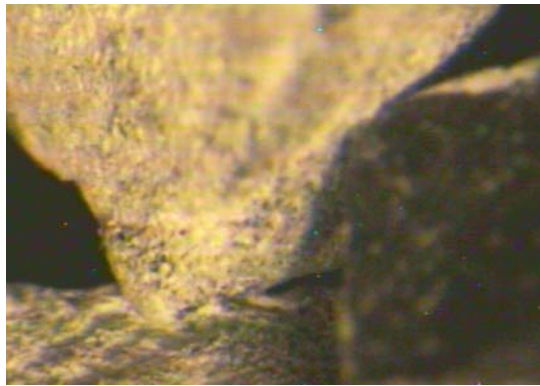
# Np Sequestration: Background

- **Burns et al. (1997)** noted similar crystal chemistry of neptunyl [ $\text{Np(V)O}_2^+$ ] and uranyl [ $\text{U(VI)O}_2^{+2}$ ] ions and hypothesized Np sequestration into U minerals.
- **Burns et al. (2004)** demonstrated Np incorporation into *Na-compreignacite* ( $\text{Na}_2[(\text{UO}_2)_3\text{O}_2(\text{OH})_3]_2(\text{H}_2\text{O})_7$ ) and *uranophane* ( $\text{Ca}(\text{UO}_2)_2(\text{SiO}_3\text{OH})_2(\text{H}_2\text{O})_5$ ) - but not *meta-schoepite* ( $\text{UO}_3 \cdot 2\text{H}_2\text{O}$ ) or  $\beta\text{-(UO}_2)(\text{OH})_2$ .
- **Buck et al. (2004)** demonstrated Np incorporation for *studtite* [ $(\text{UO}_2)(\text{O}_2)(\text{H}_2\text{O})_2](\text{H}_2\text{O})_2$  and *uranophane*.
- **Douglas et al. (2005)** added Np to *Na-boltwoodite* ( $\text{Na}(\text{UO}_2)(\text{SiO}_3\text{OH}) \cdot 1.5(\text{H}_2\text{O})_5$ ).
- **Finch and Kropf (2004)** documented Np substitution into  $\alpha\text{-U}_3\text{O}_8$  up to  $\text{Np}_{0.33}\text{U}_{2.67}\text{O}_8$ .
  - *Role of charge balancing*



# Other evidence for Np in alteration phases

- Argonne unsaturated tests on oxide spent fuel released Np ~ congruently with U (which formed copious alteration phases)

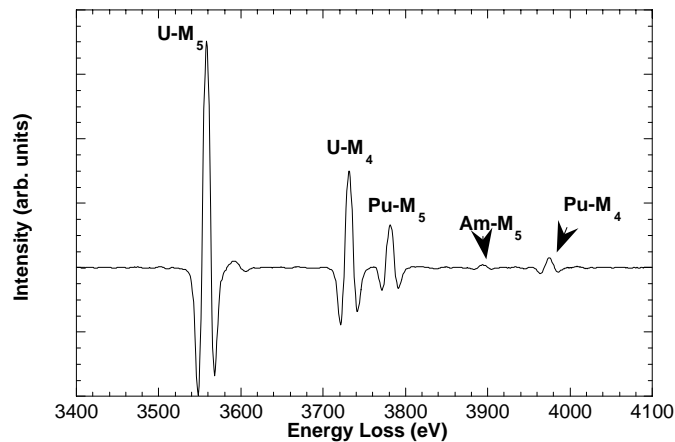
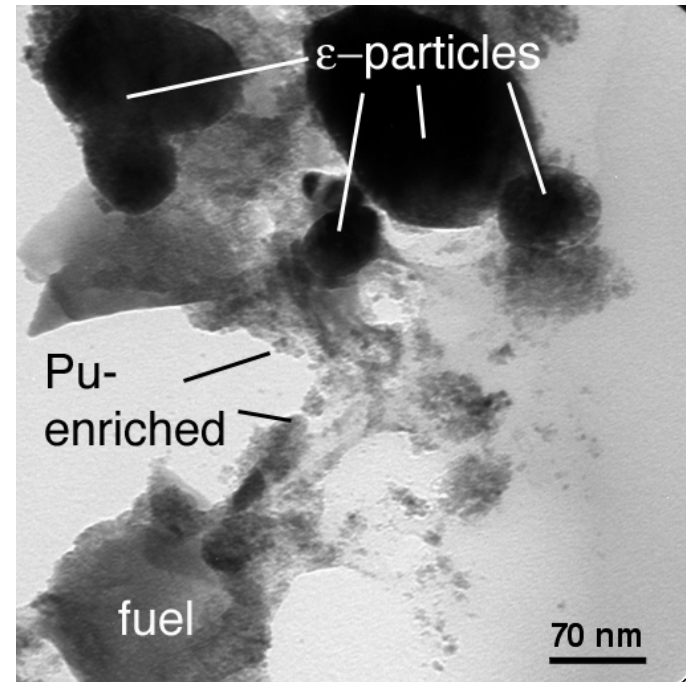


Light and electron micrographs of corroded SNF



# Pu enrichment in rind on corroded SNF

- Less-soluble Pu, Am, lanthanides, Zr, etc. retained in rind on fuel surface
- Described in detail by Buck *et al.*, 2004
- Note surviving 5-metal  $\epsilon$ -particles





# Trace element x-ray fluorescence spectroscopy at the Advanced Photon Source



# The Bent Laue X-ray Energy Analyzer

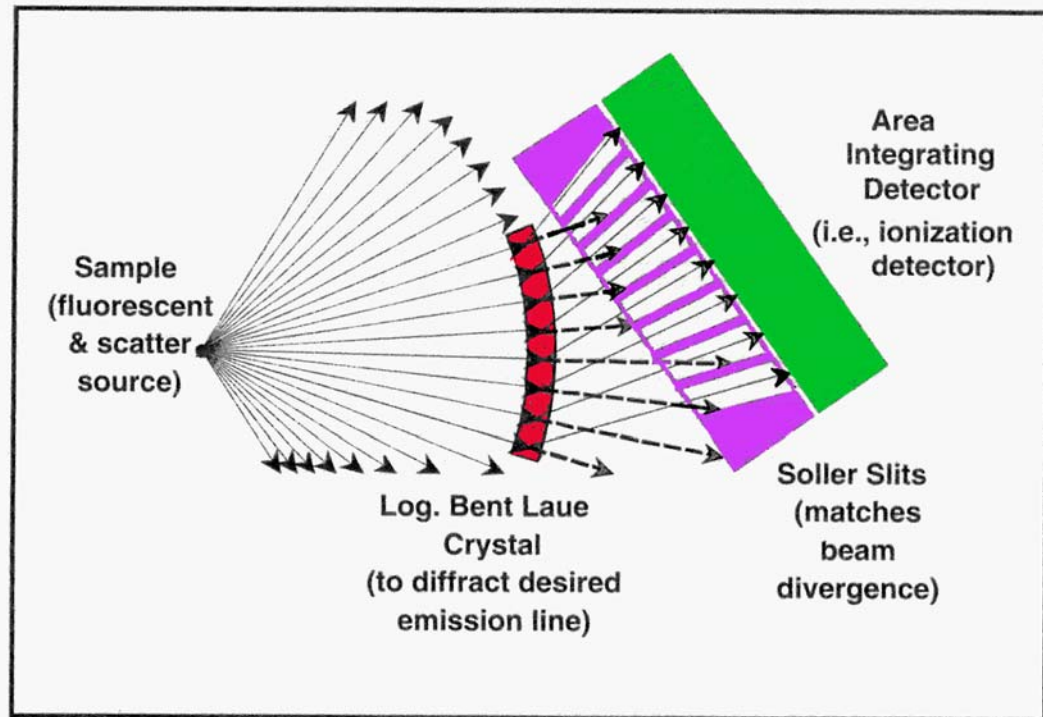
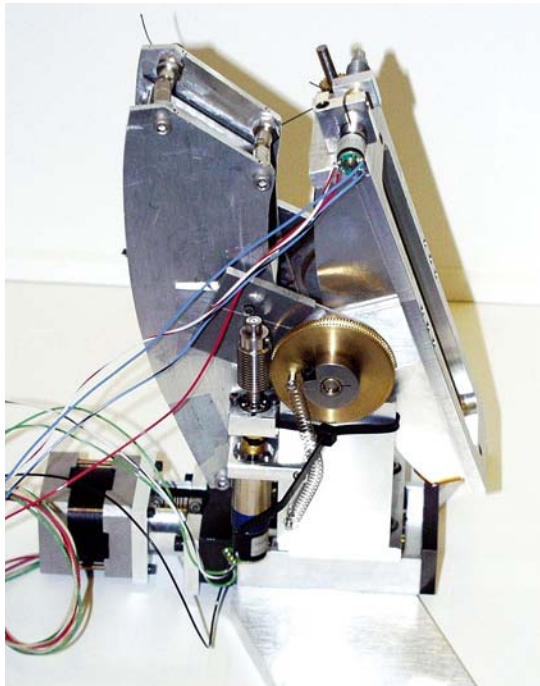
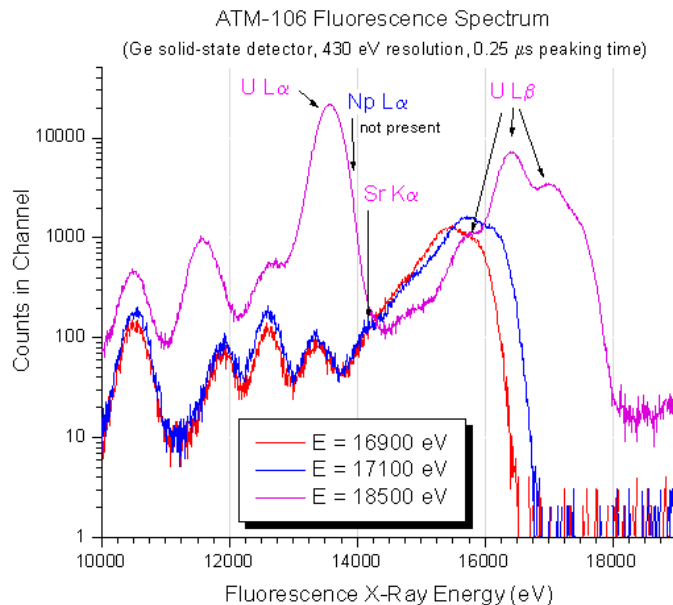


FIG. 1. A sketch of the "logarithmic bent Laue analyzer" concept.

Karanfil, L. D. Chapman, G. B. Bunker, Z. Zhong, R. Fischetti, C. U. Segre, and B. A. Bunker, 2001

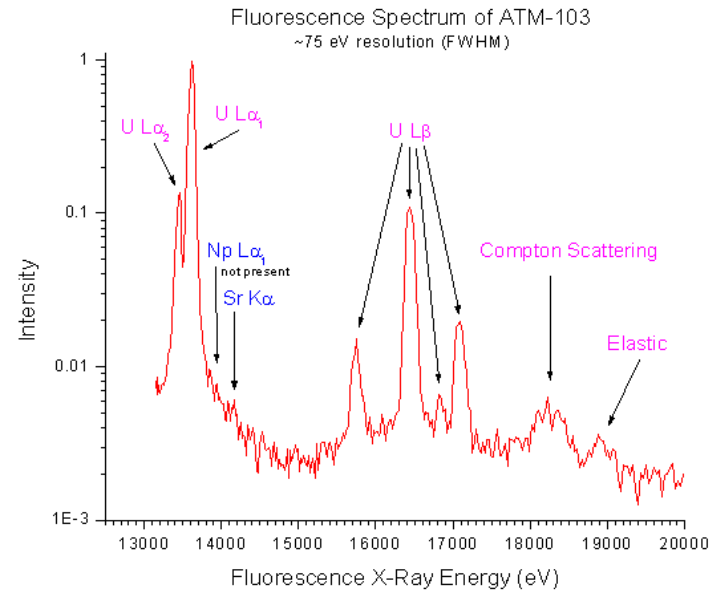


# Altered CSNF: Bent Laue Analyzer XRF



Fluorescence spectra from a fuel uranyl alteration phase using a conventional solid-state detector.

Note the large background at the Np fluorescence energy (13.945 keV) owing to the presence of copious uranium.



Fluorescence spectrum from a fuel uranyl alteration phase using the bent Laue analyzer.

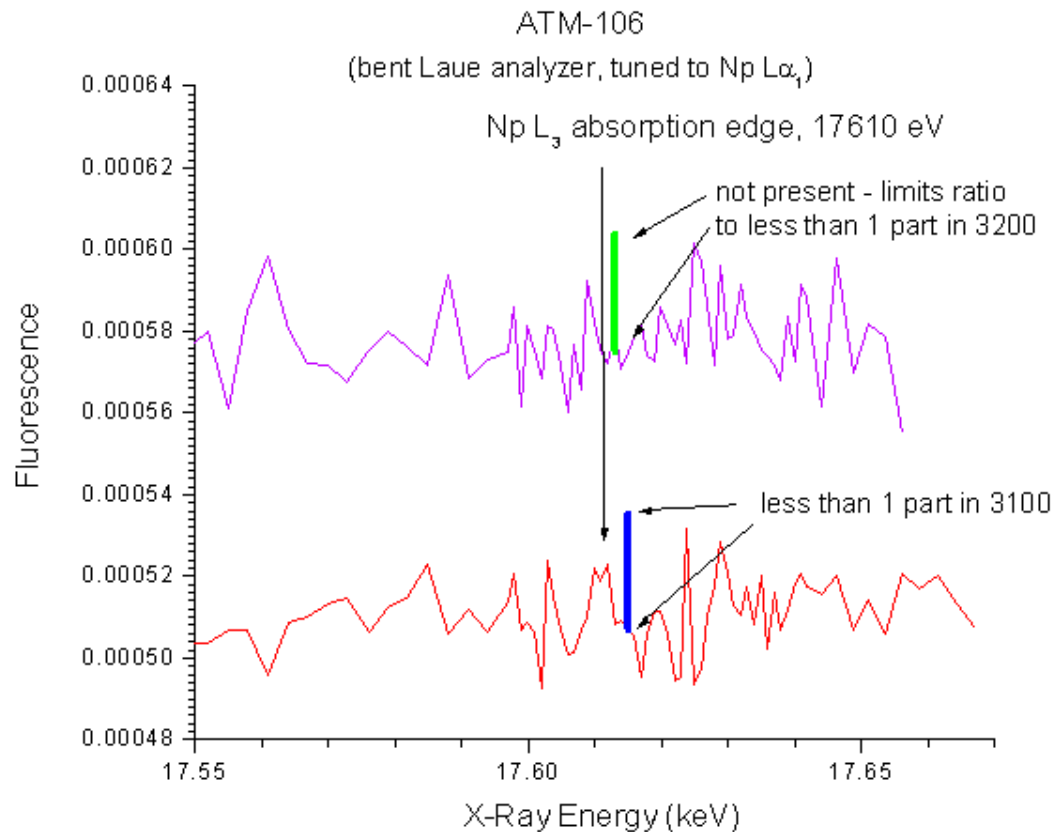
Note the greatly improved resolution and background rejection at the Np-L<sub>3</sub> energy.





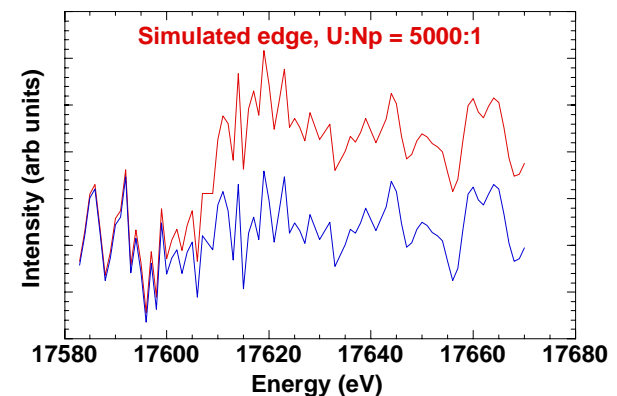
# Upper limits on Np into Schoepite

## Vapor-reacted fuel alteration phases



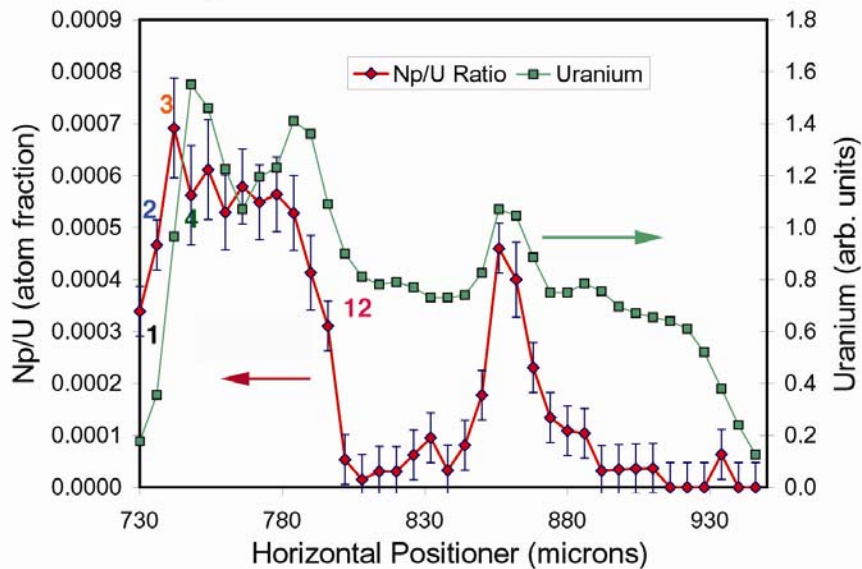
- Peak-to-peak error bars represent 1 part neptunium in 3000 parts uranium.

- Detailed analyses indicate that neptunium is present at less than 1 part in 5000.



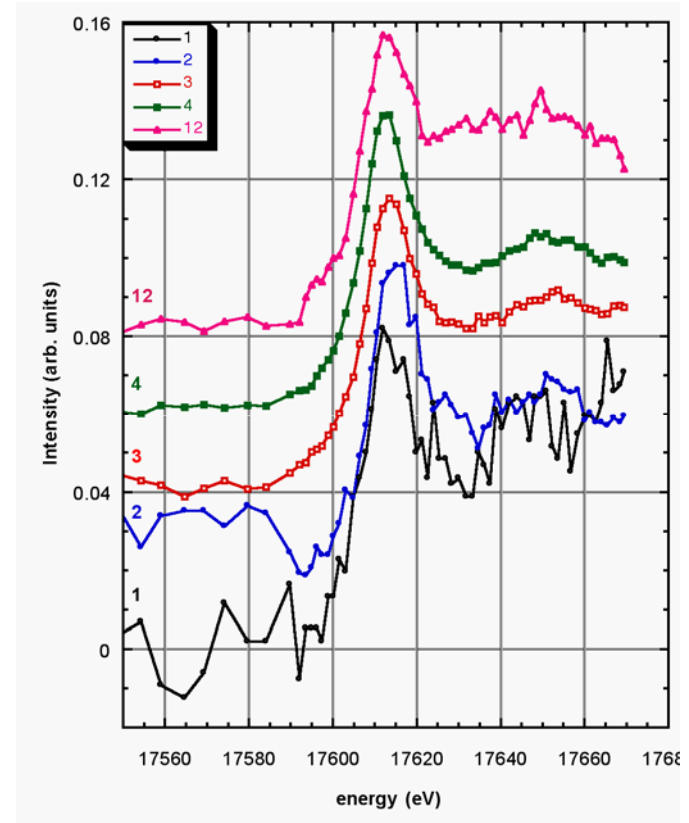
# Np in corroded CSNF with alteration products

## Np/U fluorescence line scans



**(sodium/silicate) groundwater-reacted fuel with alteration phases. Still, very little (*but not zero!*) Np in altered region**

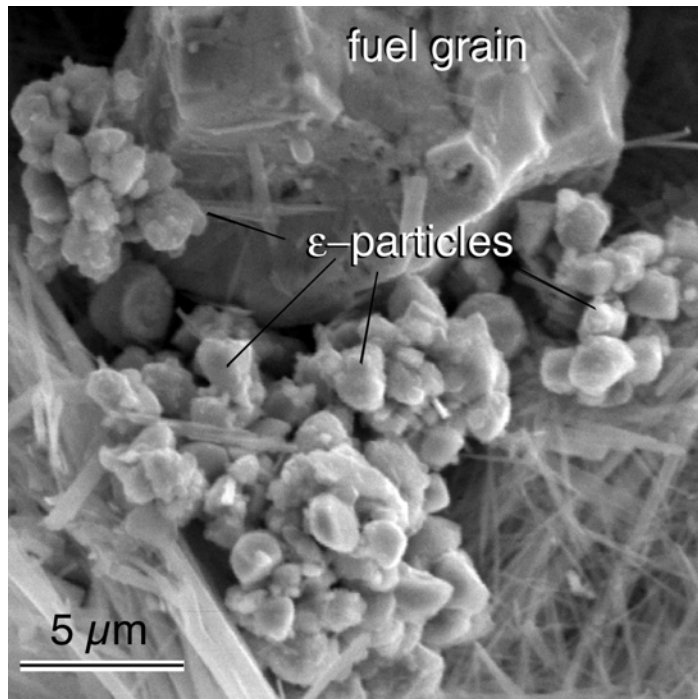
## Np XANES





# Are $\epsilon$ -phase particles corroding?

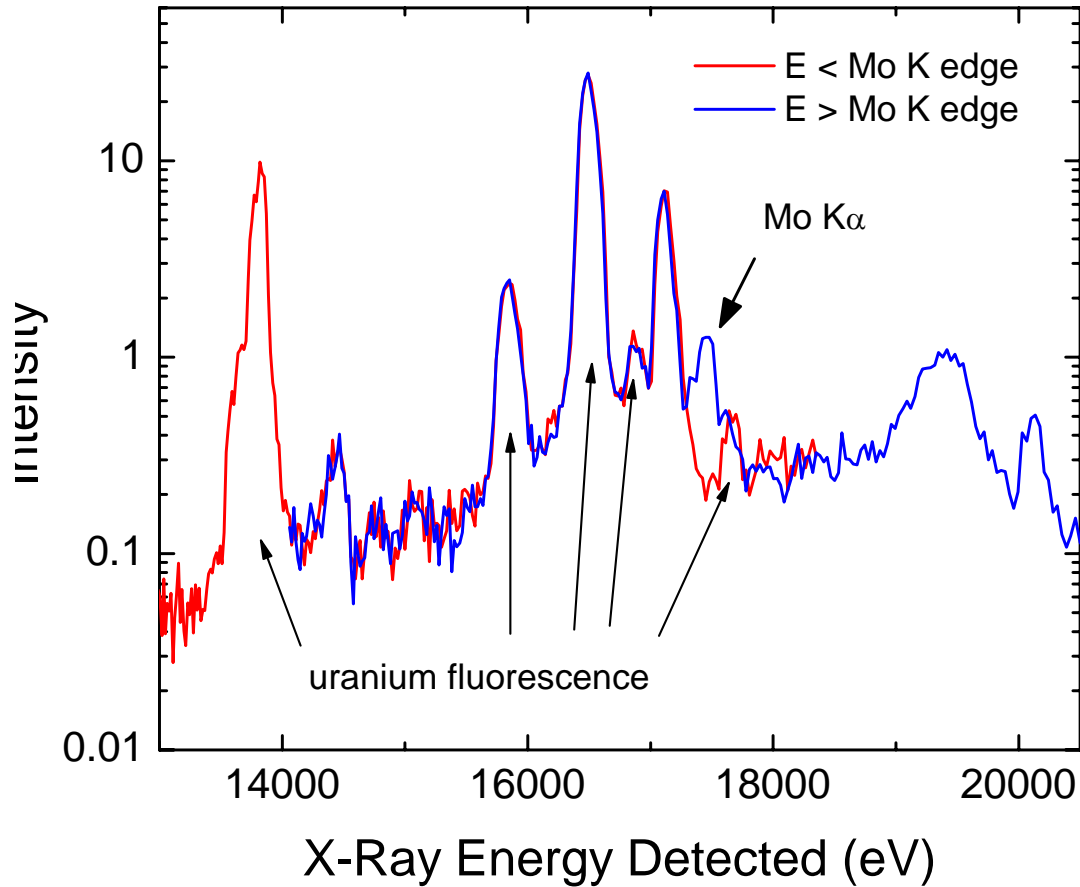
## Tc-bearing metal alloy particles



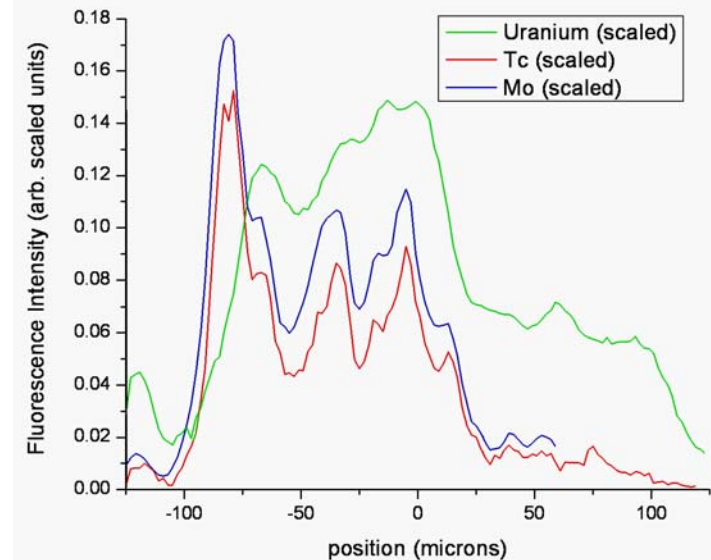
Scanning electron micrograph showing Mo- and Tc-rich  $\epsilon$ -particles concentrated on the surface of a corroded  $\text{UO}_2$  fuel grain, with acicular uranium (VI) silicates nearby. The  $\epsilon$ -particles have persisted 9 years of exposure to dripping groundwater,  $90^\circ\text{C}$ , 100% RH



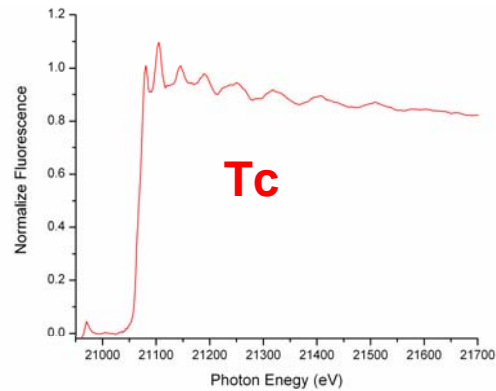
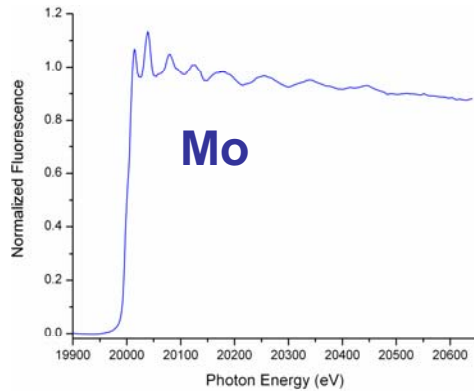
# Mo fluorescence



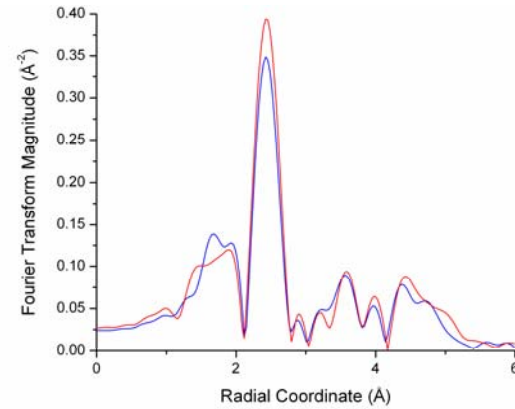
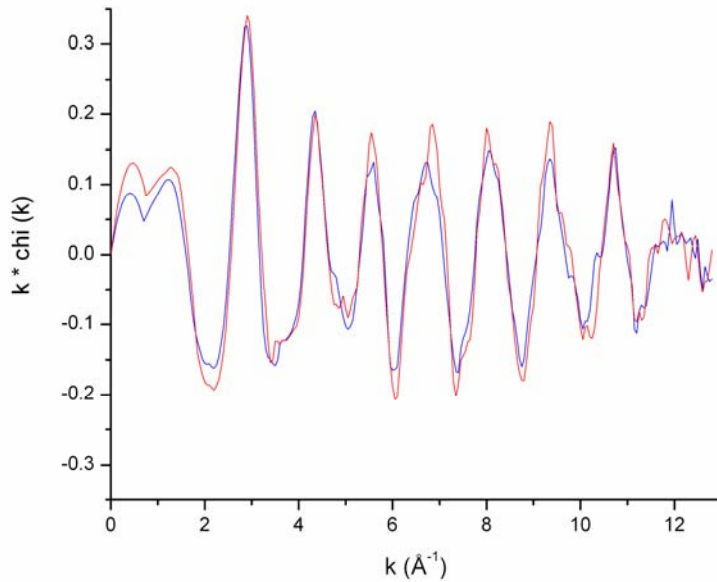
- CSNF with energy above and below the Mo Kedge
- Five U L $\beta$  lines visible
- Mo K $\alpha$  fortuitously between uranium lines



# Mo and Tc EXAFS



Spectra from a concentrated region show metallic alloy persisting despite corrosion of CSNF.



Fine structure analysis : Mo and Tc



# Observations

- **Np will likely be incorporated into uranyl alteration phases- *however*, Np(IV) in fuel may be stable at the CSNF corrosion potential.**
  - Delayed onset of U/Np phase formation.
  - $\text{NpO}_2$  may be an estimator of dissolved concentration.
- **Tc in metallic phases may likewise be slow to oxidize while fuel is intact.**
- **Pu and other sparingly soluble elements likely to form distinct phases.**



# Acknowledgements

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- **Jeff Emery and Mark Clark for Hot Cell facilities work.**

