

A Framework for the Analysis of Localized Corrosion at the Proposed Yucca Mountain Repository

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Presented by:
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Performance**
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Introduction

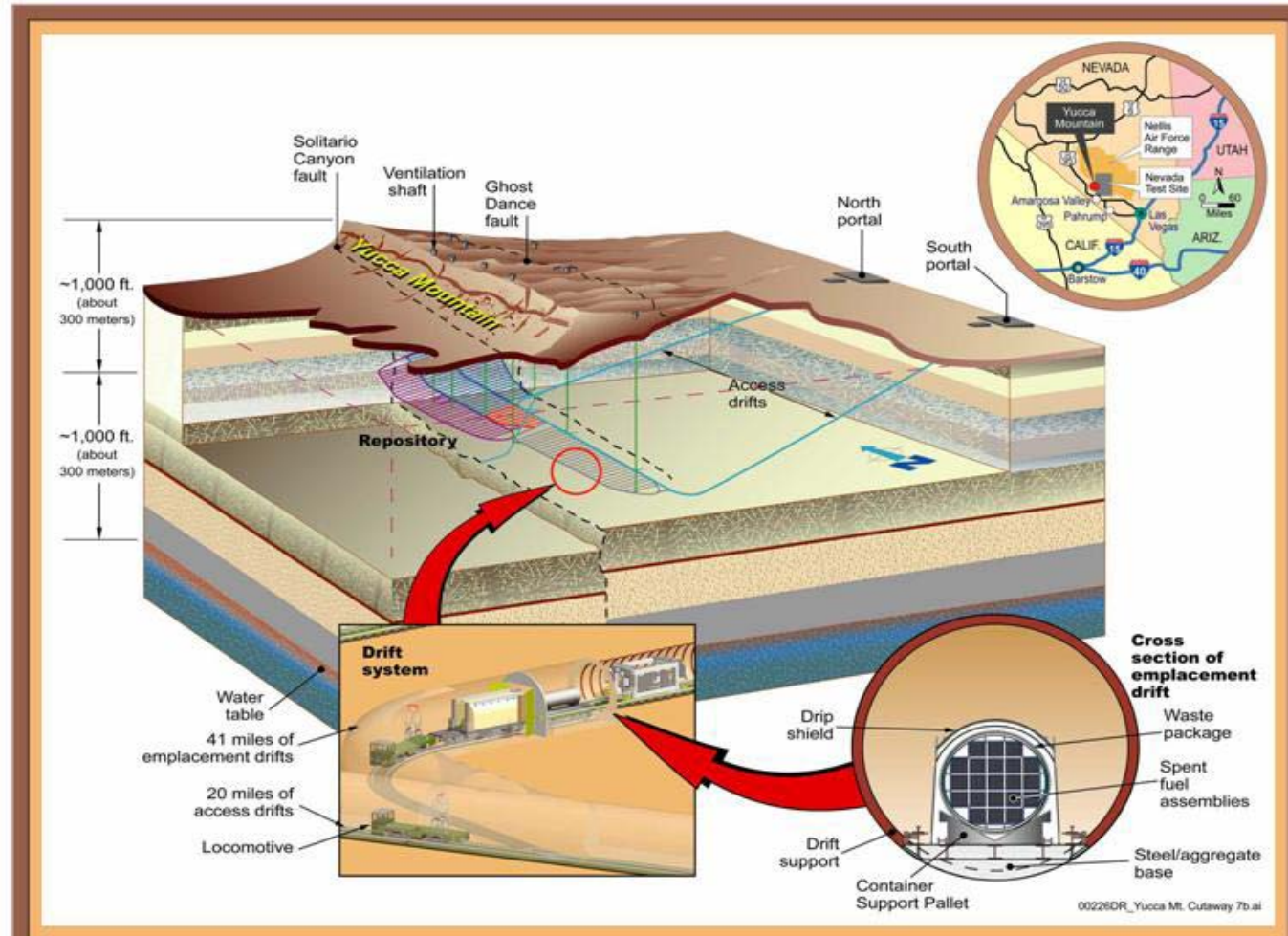
- **Corrosion is a primary determinant of waste package performance at the proposed Yucca Mountain Repository**
 - **The most likely degradation process**
 - **Controls the delay time for radionuclide transport from the waste package**
 - **Determines when packages will be penetrated and the shape size and distribution of those penetrations**
- **In this presentation a framework for the analysis of localized corrosion is presented and demonstrated for a scenario**
 - **Water chemistry of mixed salt solutions (sodium chloride-potassium nitrate)**
 - **Time-temperature-relative humidity profiles for a hot, mid and cool temperature waste package**

Methodology for Determination of Materials Performance

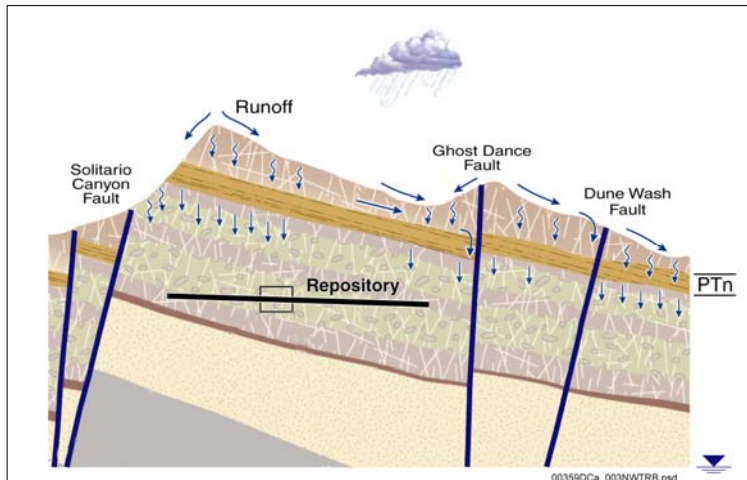
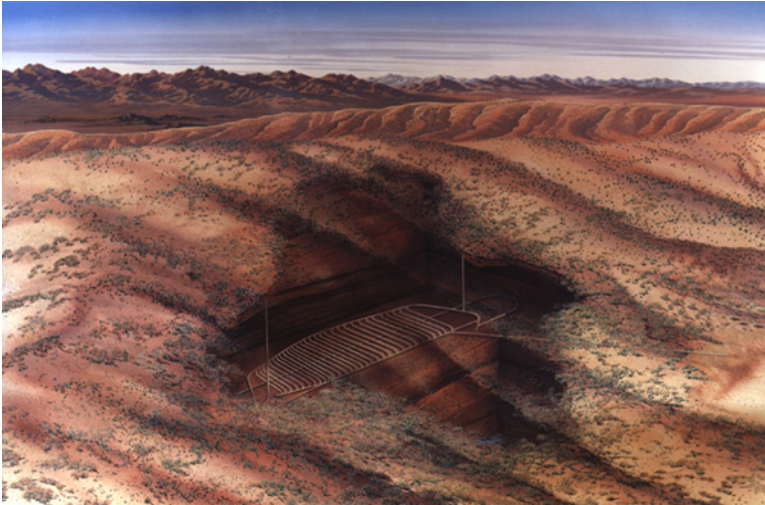
- **Materials performance at the proposed Yucca Mountain Repository is amenable to a familiar and effective analytical methodology**
 - **Widely accepted in the energy, transportation and other industries**
- **Three components comprise the analysis**
 - **Definition of the performance requirements**
 - **Determination of the operating conditions to which materials will be exposed**
 - **Selection of materials of construction that perform well in those conditions**
- **A special feature of the proposed Repository is the extremely long time frame of interest, i.e. 10,000's of years and longer**
 - **Time evolution of the environment in contact with waste package surfaces**
 - **Time evolution of corrosion damage that may result**

The Proposed Yucca Mountain Repository

Repository Reference Design Concept

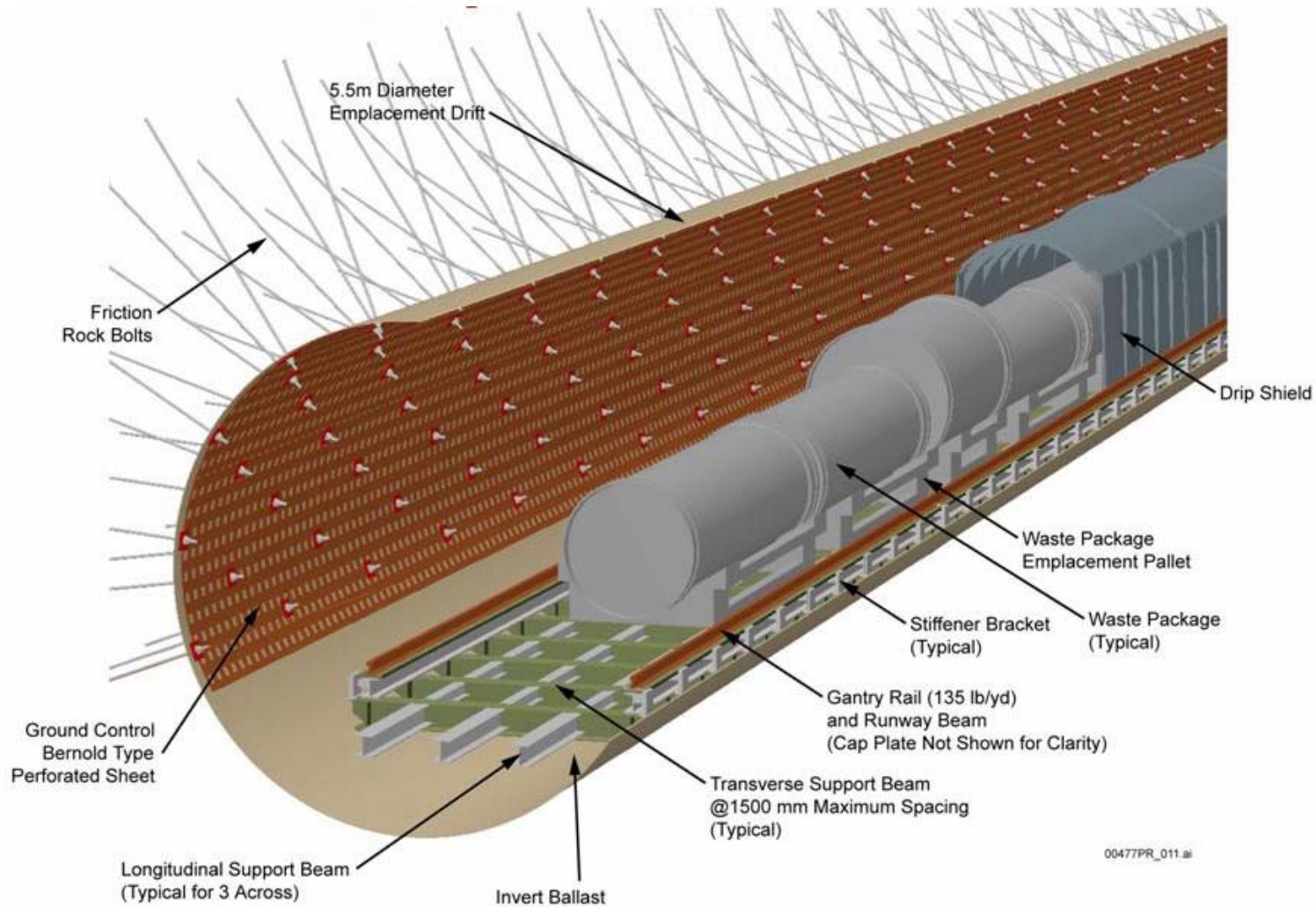


The Proposed Yucca Mountain Repository



- Proposed Repository is about 300 m below the surface and 300 m above the water table
- Unsaturated zone, i.e. fractures and pores in rock are partially filled with water
- Desert area with about 18 cm of rain per year
- Atmospheric pressure
- Ambient waters are dilute and near neutral pH
- Concentrated waters can form by condensation, deliquescence and evaporation

Proposed Emplacement Drift



Background on Ni-Cr-Mo Alloys

- **Alloy 22 belongs to a family of Ni-Cr-Mo alloys**
 - **Earlier alloys include C-276 and C-4 and later alloys include Inconel 686, Alloy 59, Hastelloy C-2000 and MAT-21**
 - **Alloy 22 (N06022) is a solid solution of Ni, Cr, Mo and W as the main alloying elements**
 - **Cr-Mo-W in Alloy 22 act synergistically to provide resistance to localized corrosion such as crevice corrosion**
- **Large industrial equipment in service for many years in harsh environments without corrosion**
 - **Alloy 22 has great toughness and over 50% elongation before failure**
 - **Can be hot or cold formed and is weldable by many methods**
 - **Can be fabricated into large structures and components**

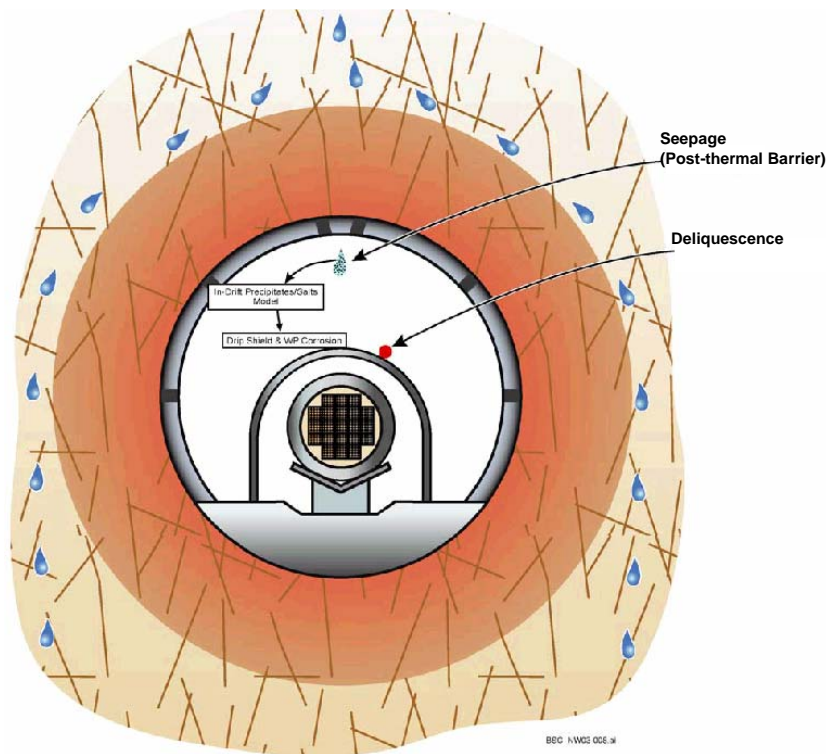
Corrosion Resistance is Crucial to Waste Package Performance

- Radionuclides are fully isolated if there are no penetrations
 - > Even penetrated package can limit radionuclide movement
- Corrosion rates of passive metals are extremely low
 - > Realistic rates are less than 1 $\mu\text{m}/\text{yr}$ (a millionth of a meter per year) and much less
 - > Alloy 22 layer is 2-cm thick (a stack of 12 U.S. quarters)
- Corrosion rates of approximately 0.01 $\mu\text{m}/\text{year}$ are measured in exposures of over 5-years at the Long Term Test Facility at Lawrence Livermore National Laboratory



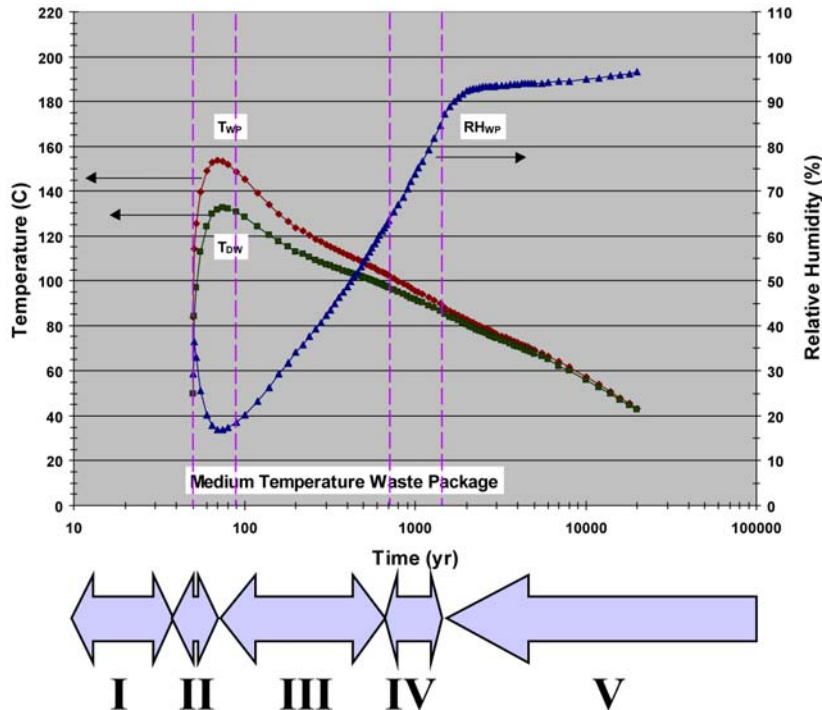
16,000 to 160,000 years to penetrate the thickness of one U.S. quarter for a corrosion rate of 0.1 to 0.01 $\mu\text{m}/\text{yr}$

Attributes of the Proposed Yucca Mountain Repository



- One long, slow heating/cooling cycle
 - > Packages cool to ambient over several thousands of years
- Waste packages on support pallets
 - > No immersion in waters
- No moving parts
- Low heat fluxes, slow heating and cooling, and modest thermal gradients
- Radiation effects at waste package surface negligible after a few hundred years
- Limited amount of water moving through the rock
- Limited salts and minerals carried into drifts by incoming water and dust

Relevant Time Periods for Corrosion



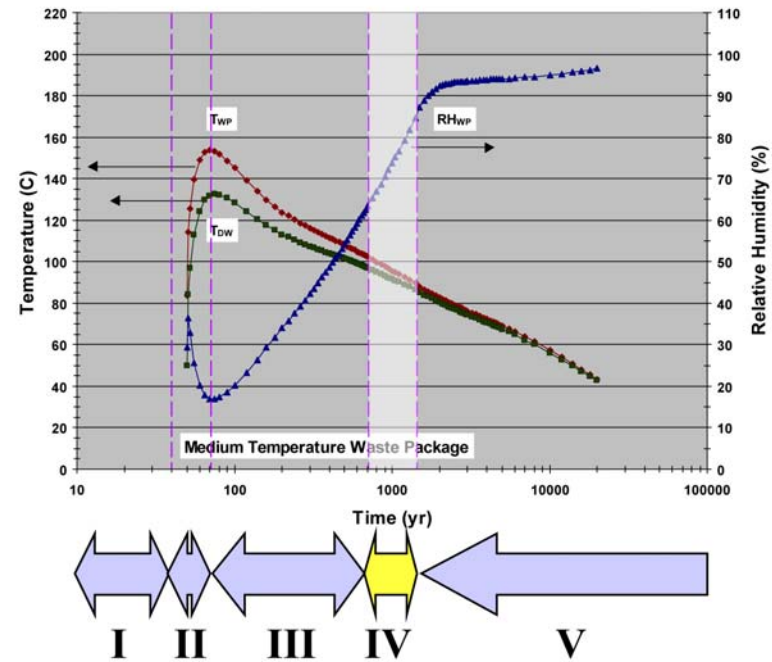
- I - **Emplacement of waste packages and preclosure**
 - > Start to Year 50
- II - **Heat Up after closure**
 - > Year 50 to ~65
- III - **Cool down/Thermal Barrier (drift wall above boiling temperature)**
 - > Year ~65 to 750
- IV - **Cool Down/Dripping and Seepage Possible**
 - > Year 750 to 1375
- V - **Waste Packages below Critical Temp for Corrosion**
 - > Year 1375 and beyond

- This scenario is for Temp-RH shown above
 - > Waste Package at 101°C when Drift Wall cooled to 96°C
 - > Critical Corrosion Temp 90°C

- Periods are determined by
 - > Temperature-RH conditions
 - > Time when drift wall reaches 96°C
 - > Critical Corrosion Temp for Alloy 22

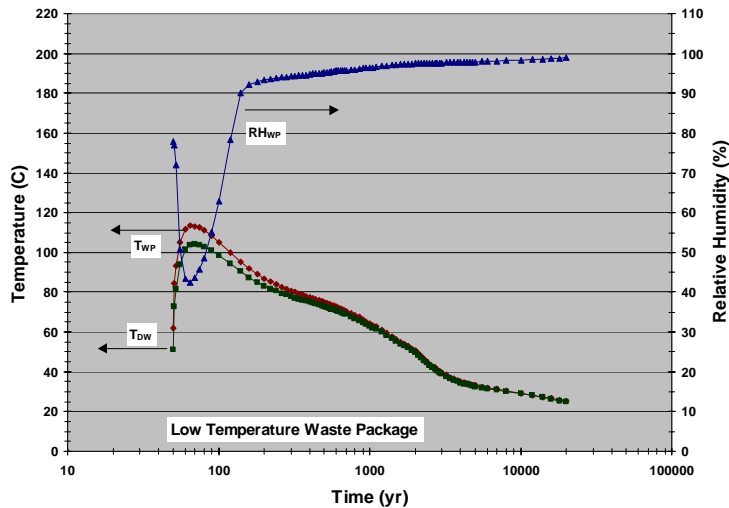
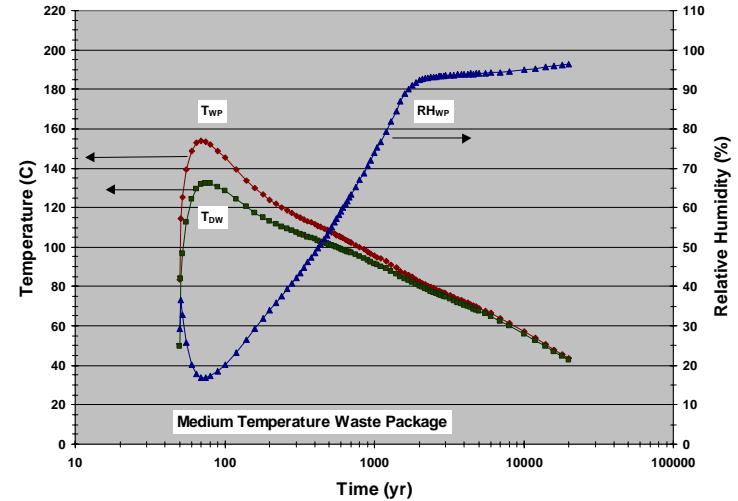
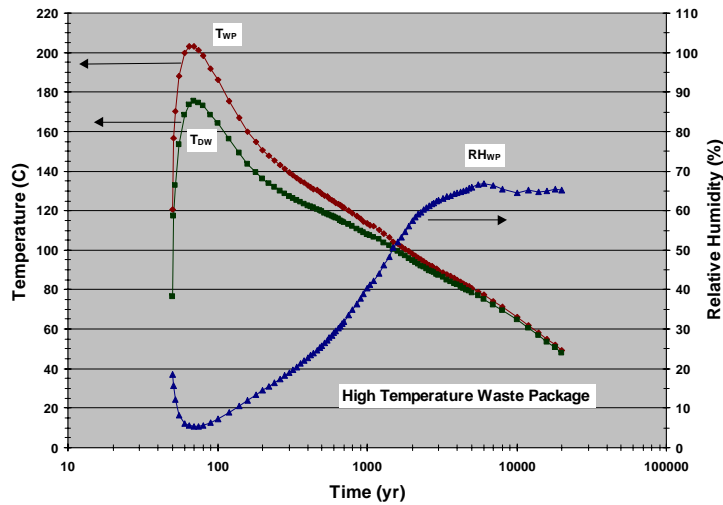
Period IV-Dripping and Seepage Possible

- When drift wall is below boiling temperature (96°C), dripping/seepage can occur
- Dripping/seepage can contact waste package surface
 - > Where both capillary barrier and drip shield are inoperative
 - > And dripping location is in aligned with drip shield penetration
- When these conditions are met
 - > If waste package temperature above critical corrosion temperature
 - > Then follow decision-tree analysis for local corrosion damage evolution



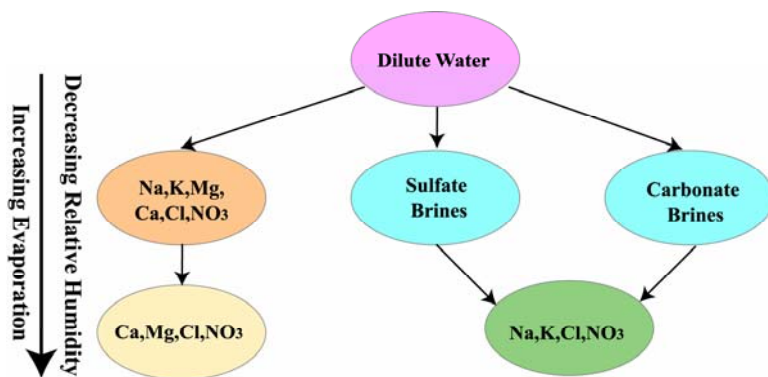
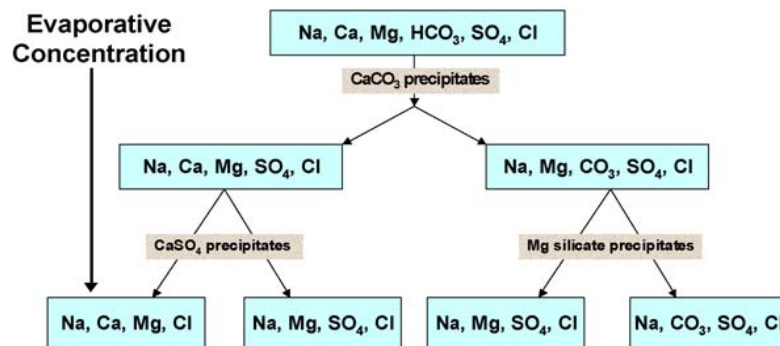
- Drift wall is below boiling at year 750
 - > Waste Package at 101°C
 - > Relative humidity 65%
- Waste Package is at 90°C at year 1375
 - > Relative Humidity 84%

Period IV Conditions for Mid, Hot and Cool Waste Packages



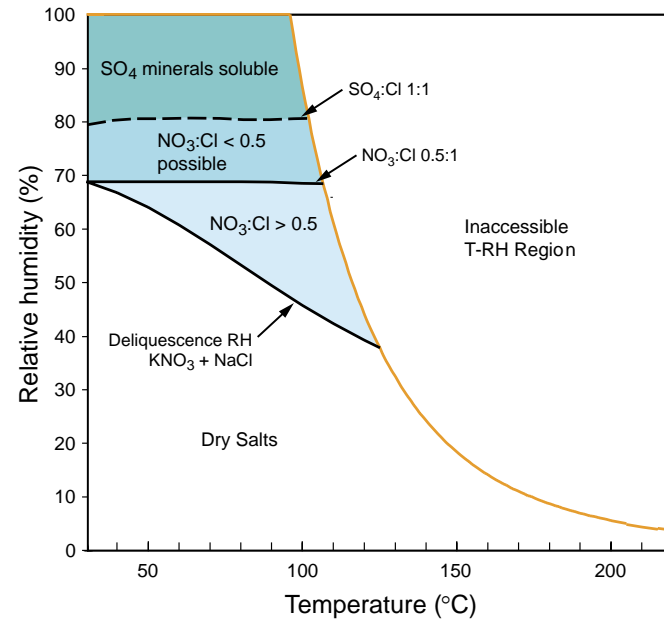
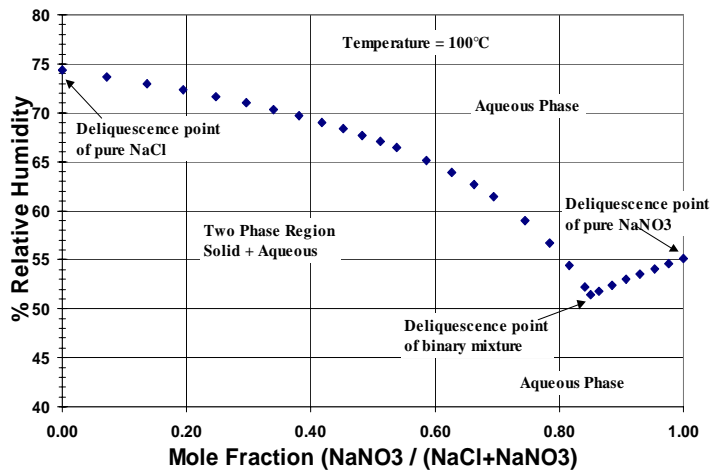
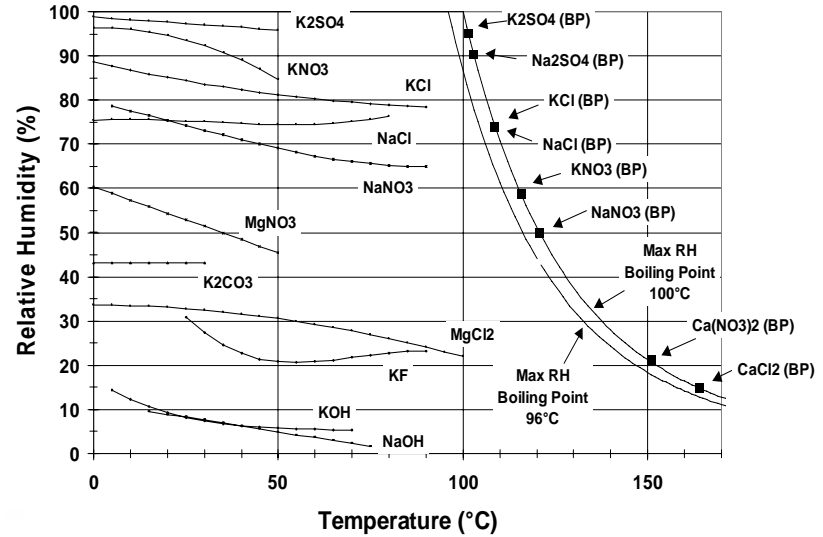
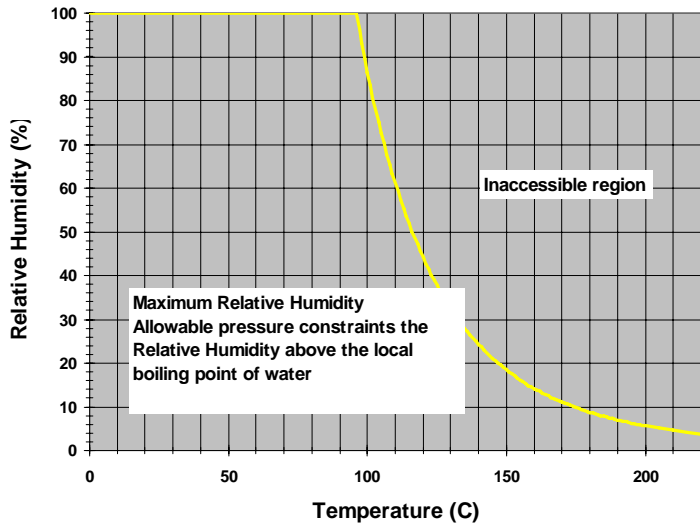
Drift Wall 96°C	Year	Waste Package Temp °C	Relative Humidity	Waste Package at 90°C
Mid WP	700	101	65	1325
Hot WP	1850	99	56	3000
Cool WP	62	102	72	125

Chemical Divide Processes Determine the Categories of Waters

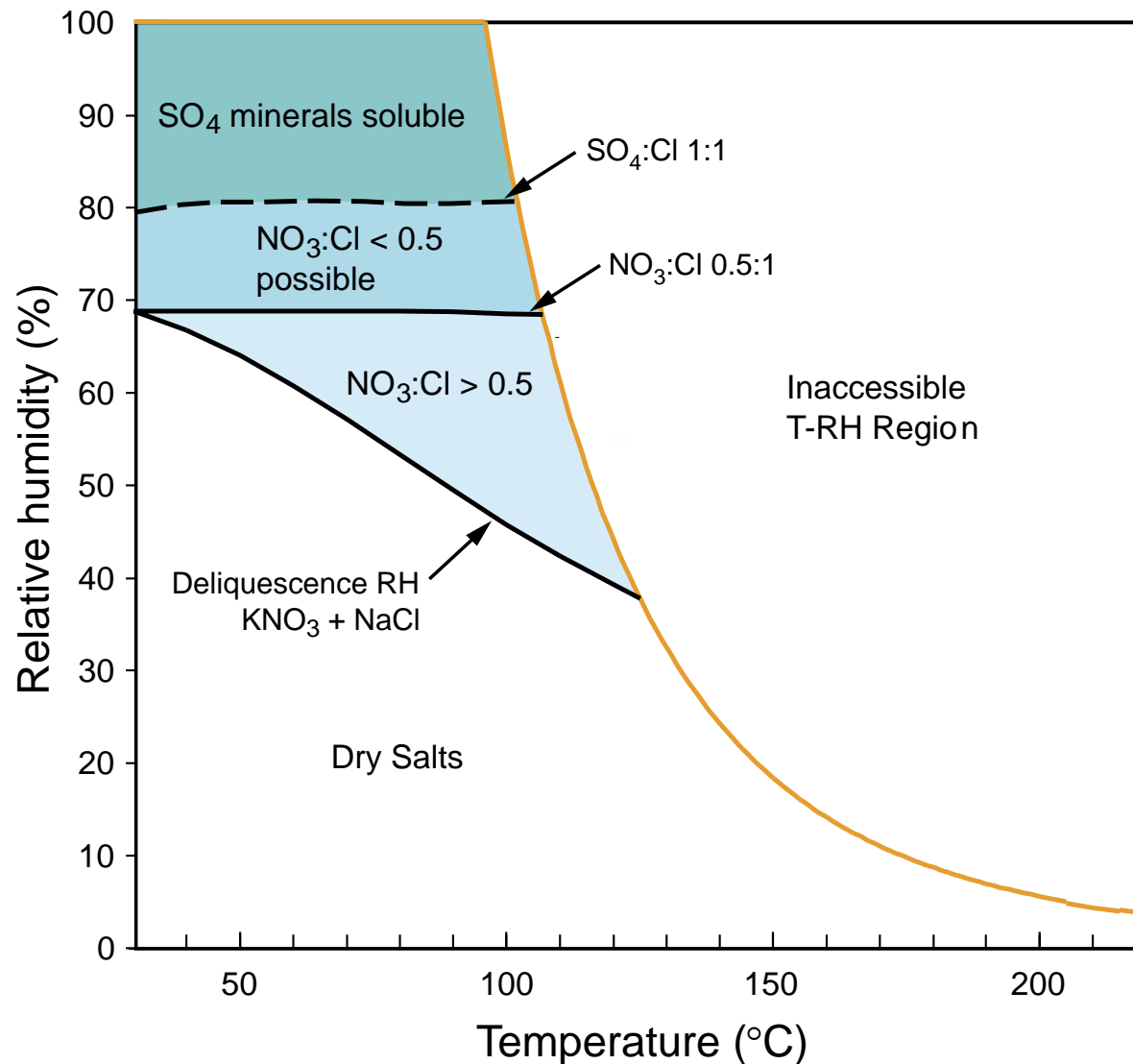


- **Ambient Waters:**
 - > Dilute solutions
 - > Na-Ca-Mg-HCO₃-CO₃-Cl-NO₃-SO₄
 - > Near neutral pH
- **Waters can be concentrated**
 - > Modified during movement
 - > Thermal-chemical processes
- **Modifications on waste package surface**
- **Chemical and electrochemical processes**

Solution Chemistry Principles

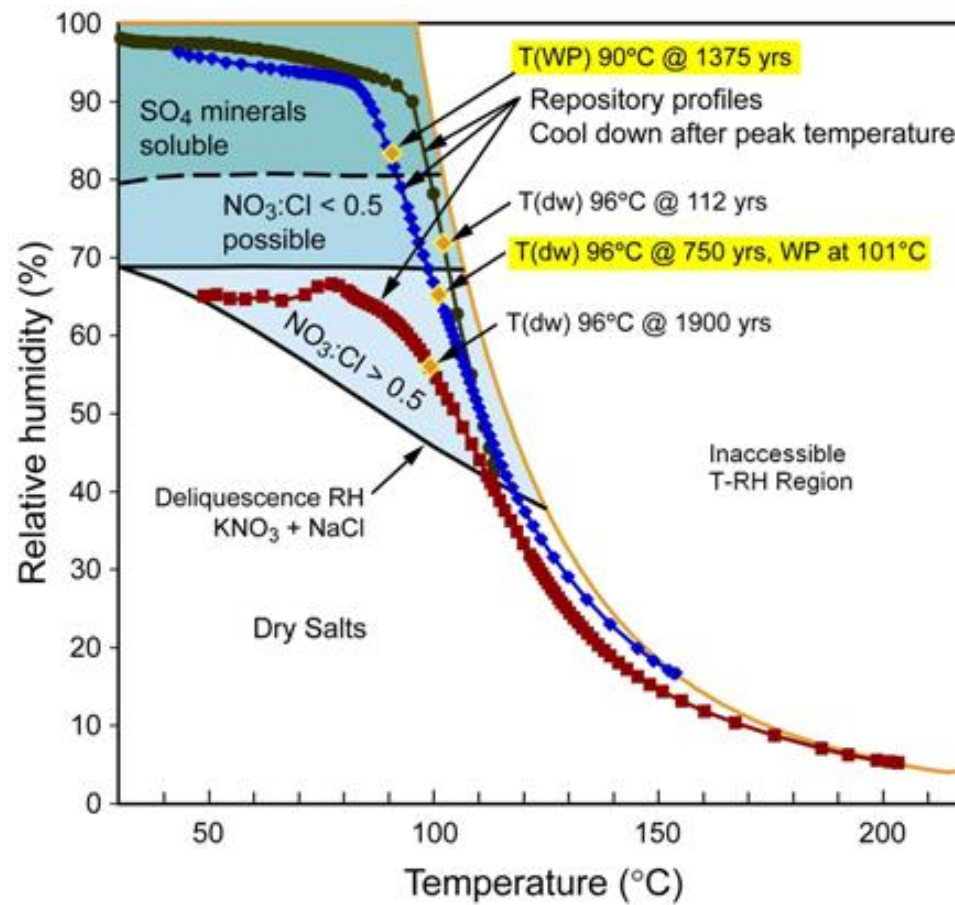


Constraints on Water Compositions for Sodium and Potassium Salts

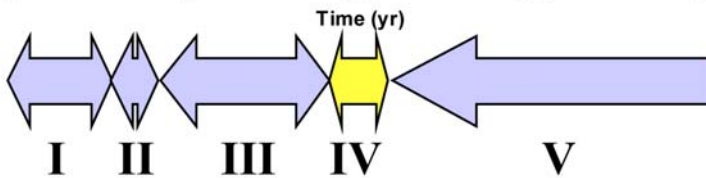
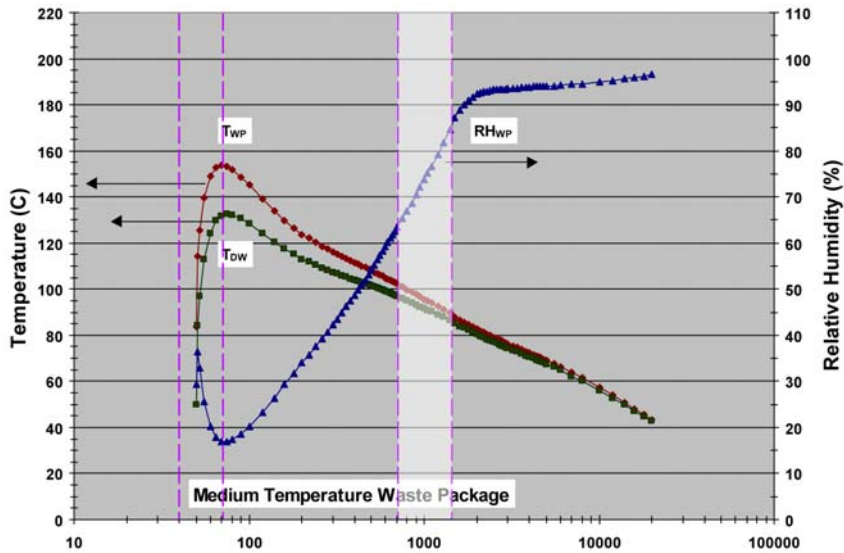


Water Chemistry Scenarios for Waste Package

- T-RH Profiles Related to Brine Solution Compositions for Sodium and Potassium Base Salts

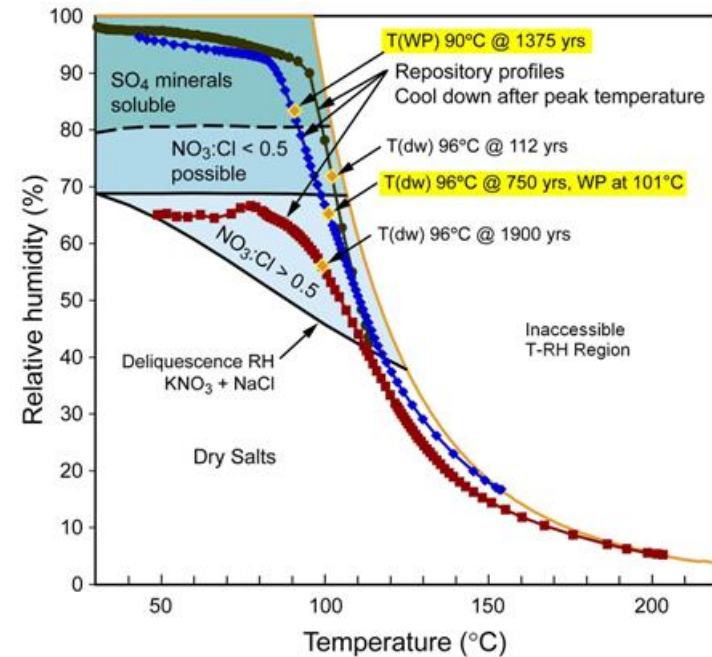


Period IV Analysis of T-RH-Solution Composition



Drift wall 96°C at 750 years;
 Waste Package at 101°C;
 Relative Humidity 65%

Critical Corrosion Temp 90°C
 at year 1375; Relative Humidity 85%



The Temp-RH at any time fixes the possible waters. Can follow the trajectory with time

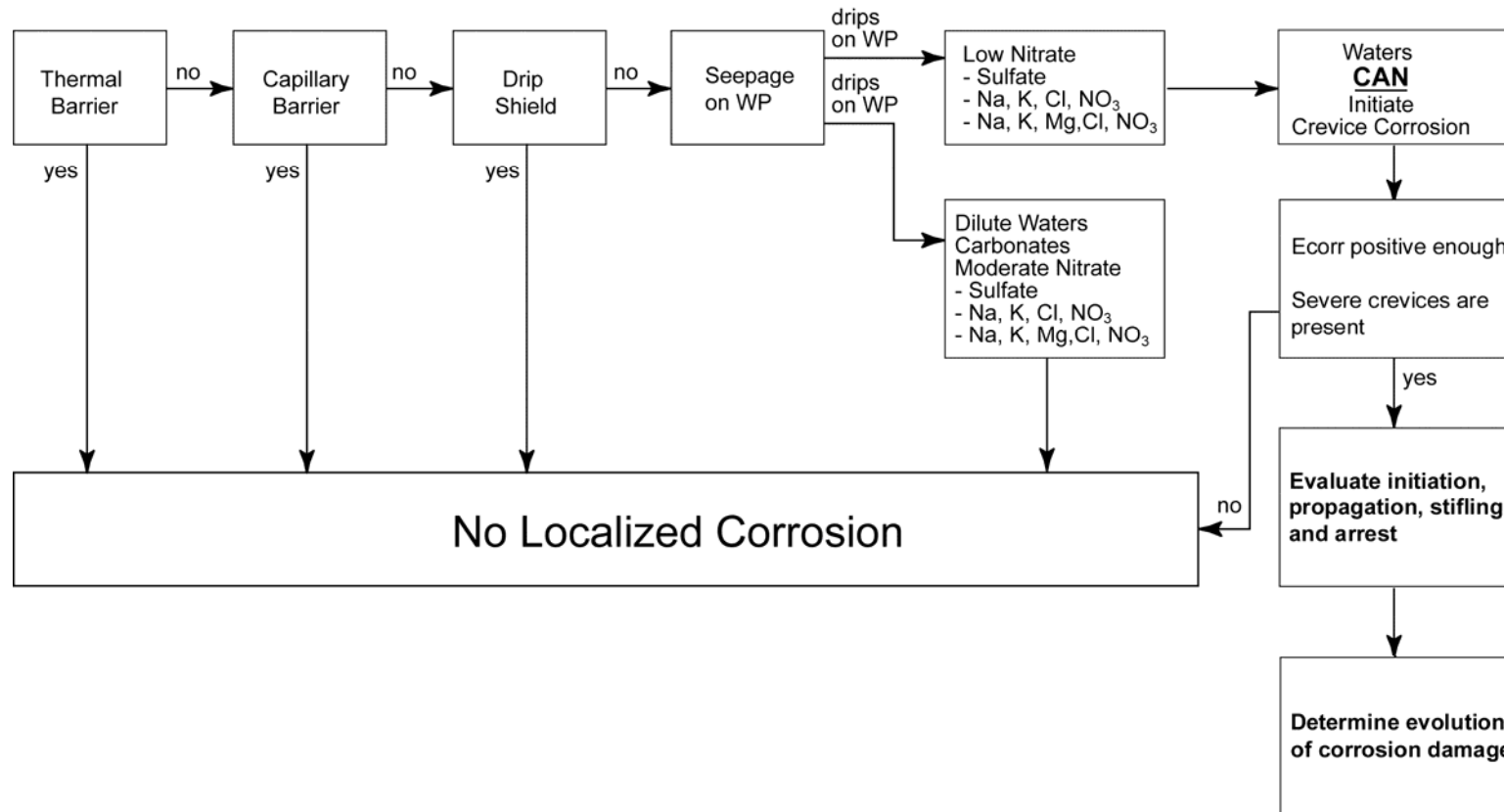
Number of non-corrosive solutions; Sodium chloride with low nitrate solutions can be corrosive

Decision-Tree Analysis

- **A decision-tree for localized corrosion**
 - **Are environments and crevices present to induce localized corrosion?**
 - **Consider conditions in moist layers of particulate and deposits**
 - **If localized corrosion initiates, will it persist?**
 - **Consider stifling and arrest processes as the corrosion proceeds**
 - **What amount of metal penetration occurs?**
 - **What is the size and distribution of corrosion sites?**

Decision-Tree Analysis

- A decision-tree for localized corrosion



Summary

- **Presented a framework for the analysis of localized corrosion**
- **Demonstrated the analysis for a scenario**
 - **Water chemistry of mixed salt solutions**
 - **Time-temperature-relative humidity profiles for waste packages**
- **Localized corrosion on waste packages is restricted to finite time periods**
 - **Corrosion conditions at key time periods in proposed Repository**
 - **Corrosion analysis during period IV-cool down/dripping and seepage**
- **Decision-tree analysis for corrosion damage evolution**
 - **For those time periods when localized corrosion can be supported**
 - **Based upon the temperature and possible water chemistries**
 - **Apply decision-tree analysis to determine the evolution of corrosion damage**