Physicists and astronomers almost always use very small or very large numbers in the calculations or measurements.

Scientific notation is the best, and most compact, way to work with very large and small numbers.

This activity will review how to divide numbers in this form.


$$
\begin{aligned}
& 1.9 \times 10^{33} \text { grams }
\end{aligned}
$$

## Scientific Notation simplifies calculations with large or small numbers.

| Scientific Notati provides an easy | Here's how to do it! |  |
| :---: | :---: | :---: |
|  | State the problem: | $7.5 \times 10^{-11} / 4.5 \times 10^{+20}$ |
|  | Group the factors: | $(7.5 / 4.5) \times\left(10^{-11} / 10^{+20}\right)$ |
|  | Subtract the expon | $1.67 \times 10^{(-11-(+20))}$ |
| Now you try! | Answer: | $1.67 \times 10^{-31}$ |

Perform these divisions using scientific notation.

1) Density of a star in grams/cc:

$$
2.1 \times 10^{33} / 3.1 \times 10^{32}
$$

2) Speed of solar wind in $\mathrm{km} / \mathrm{sec}$ :
$1.47 \times 10^{8} / 2.7 \times 10^{5}$
3) Density of a proton in grams/cc:

$$
1.64 \times 10^{-24} / 3.7 \times 10^{-38}
$$

4) Speed of Sun around Milky Way in centimeters/second:

$$
2.7 \times 10^{23} / 6.3 \times 10^{15}
$$

