

Irradiation for the Novel Radiolytic Formation of Superalloy Nanoparticles



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Problem

Utilization of Novel NP Formation without Sintering or Oxidation
 Superalloys: Ni-based alloys that are hardened with refractory metals to higher temperatures while retaining superb mechanical strength e.g., γ' -phases Ni/(Co, Cr, Mo, or W) or γ'' -phases Ni₃(Al, Ti)

Synthesis of nanoparticles of alloy compositions allows for both nano- and bulk-scale applications of alloys.

Sintering of nanoparticles of an alloy allows for more defect-free bulk alloys, non-destruction of the refractory elements, and access to metastable phase spaces and morphologies.

Radiolysis is a room-temperature method to produce kinetically favorable, metastable alloy nanoparticles.

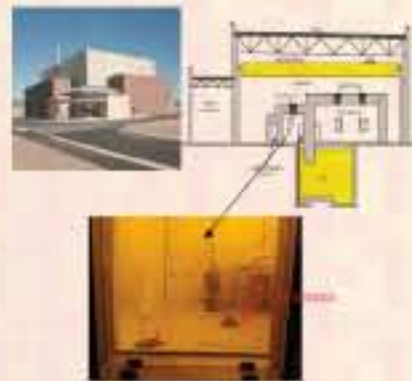
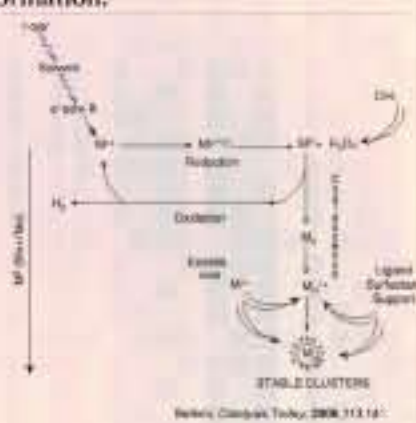
Ability to make homogenous NPs, independent of reaction size, means large quantities are possible.

Approach

Room Temp Radiolysis at SNL's GIF Facility

Radiolysis is a method by which metal ions are reduced in water by ionizing radiation: The dose rate dictates the [e⁻] in the reaction solution thereby affecting the chemistry of the NP formation.

Sandia's Gamma Irradiation Facility (GIF) is a ⁶⁰Co source: 1.345 x 10¹³ Ci, = 300 K rad/hr.

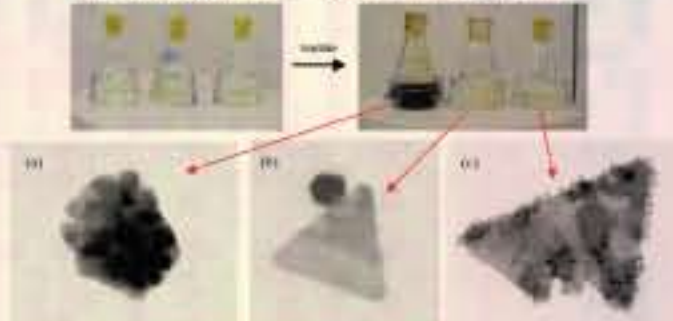


Results

Gold Nanoparticles, Morphology Determined by Dose Rate

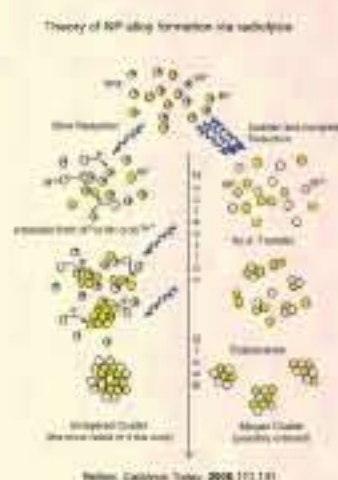
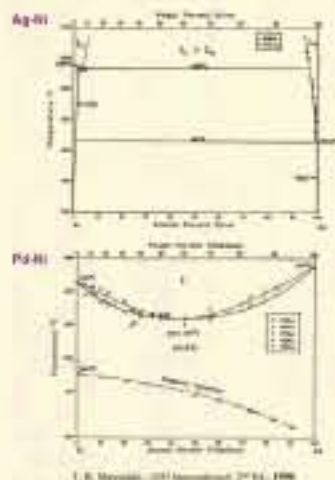
Reaction Conditions: 25-ml solutions in 100-ml vials
 HAuCl₄ (1000 ppm Au in dilute HNO₃), poly(vinyl alcohol) (PVA, MW of 48000), D₂O
 Purged solution with N₂, sealed and stored in dark
 Exposed solutions to irradiation, allowed to age and crystals to grow

(a) 770, (b) 72.3, (c) 7.07 rad/sec for 3 min dose (0.07 rad)

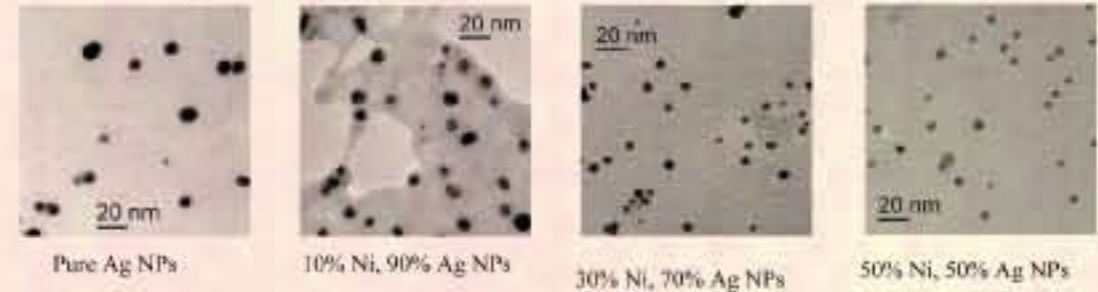


All Gold Nanoparticles Exhibit Single-Crystal Behavior (1000x)

Kinetically Driven Access to New Phase Spaces: Thermodynamically inaccessible or limited Ni-based NPs

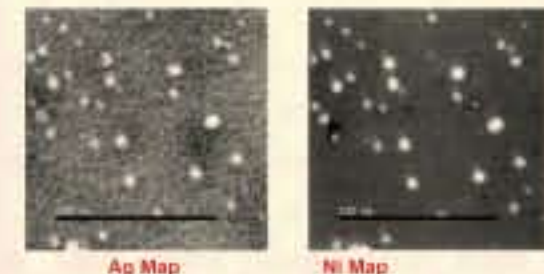
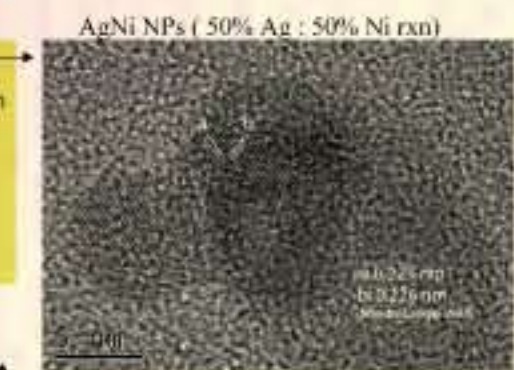
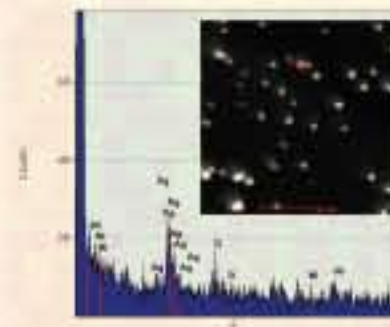


Reaction Composition Effect on NP sizes: TEM



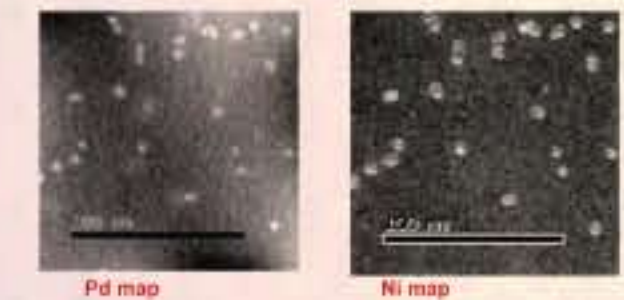
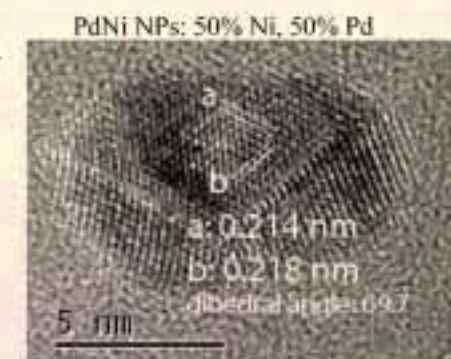
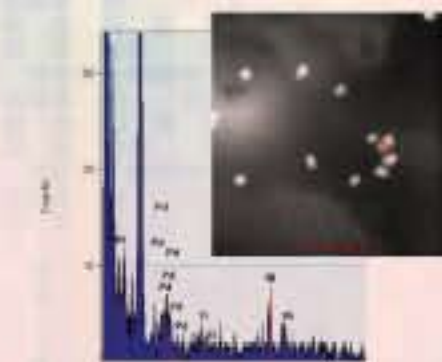
AgNi Superalloy Nanoparticles

• HRTEM (CINT):
 Ni (111) = 0.203 nm, Ag (111) = 0.236 nm
 Ag-Ni alloy based on 50% Ni = 0.22 nm
 • STEM & EDX: Single particle data indicates homogenous composition of Ag & Ni, confirmed by EELS maps



PdNi Superalloy Nanoparticles

• HRTEM (CINT) (twinned crystal)
 Ni (111) = 0.203 nm, Pd (111) = 0.225 nm
 Pd-Ni alloy based on 50% Ni = 0.214 nm
 • STEM & EDX: Single particle data indicates homogenous composition of Pd & Ni, confirmed by EELS maps



Significance

- Long-term Science for SNL Business SMUs (NTM, ERN) plus DOE National Security Mission and Strategic Goals.
- Defense, Energy, Science applications require SuperAlloys: lightweight, corrosion-resistant, sintered refractory materials (weapons casings & connects, aircraft, satellites, power plants, gas turbine engines & burners)
- Leverage future funding by DOE/NE (nuclear fuel alloys), DOE/H₂ (H₂ dissociative membranes), DOD/DARPA/DTRA