APPENDIX E OTHER GLYCOL ETHERS

Table E-1.—Methyl-based glycol ethers, amides, etc.*

CAS [†] number	Name	
3610-27-3	Methoxytriethylene glycol acetate	
3121-61-7	2-Methoxy-ethanol acrylate	
24493-59-2	Methoxytriethylene glycol methacrylate	
45103-58-0	2-(2 Methoxyethoxy) ethyl methacrylate	
6976-93-8	2-Methoxyethyl methacrylate	
68133-26-6	N-(5-Amino-2-methoxyphenyl)-beta-alanine, 2-methoxyethyl ester	
68133-25-5	N-(2-Methoxy-5-nitrophenyl)-beta-alanine, 2-methoxyethyl ester	
16501-01-2	1,2-Benzenedicarboxylic acid, mono (2-methoxyethyl) ester	
117-82-8	1,2-Benzenedicarboxylic acid, bis (2-methoxyethyl) ester	
106-00-3	Hexanedioic acid, bis (2-methoxyethyl) ester	
140-05-6	Ethylene glycol monomethyl ether acetylricinoleate	
111-10-4	Ethylene glycol monomethyl ether oleate	
111-07-9	Hexadecanoic acid, 2-methoxyethyl ester	
70703-47-8	Acetylated ethylene glycol monomethylether hydroxystearate	
6522-67-4	N-[5-(Acetylamino)-4-[(2-bromo-4,6-dinitrophenyl)azo]-2 methoxyphenyl]-beta-alanine, 2-methoxyethyl ester	
51248-73-8	N-[4-[2-Chloro-4-nitrophenyl)azo]-3-(acetylamino)phenyl]-N-(2-cyanoethyl)-beta-alanine, 2-methoxyethyl ester	
49744-35-6	Aminobenzoic acid, 2-(2-methoxyethoxy) ethyl ester	
	(Continued	

^{*}Adapted from TSCA [1977] and SANSS/CIS [1988].
†Chemical Abstracts Service.

Table E-1 (Continued).—Methyl-based glycol ethers, amides, etc.

CAS number	Name	
49744-26-5	4-[5-Cyano-1-ethyl-1,6-dihydro-2-hydroxy-4-methyl-6-oxy-3-pyridinyl)azo] benzoic acid, (2-methoxyethoxy) ethyl ester	
42861-47-2	N-Ethyl-N(4-(2-bromo-4,6-dinitrophenyl)azo)-5-(acetylamino)-2-methoxyphenyl)-beta-alanine, 2-methoxyethyl ester	
42228-65-9	N-[5-(Acetylamino)-2-methoxyphenyl]-N-ethyl-beta-alanine, 2-methoxyethy ester	
18016-42-7	Cholesteryl 2-(2-methoxyethoxy) ethyl carbonate	
40228-74-8	Cholesteryl methoxy ethyl carbonate	
68479-79-8	N,N'-(4,8-Dihydroxy-9,10-dioxo-1, 5-anthracenediyl) bis [beta-alanine], bis(2-methoxyethyl) ester	
1616-88-2	Carbamic acid, 2-methoxyethyl ester	
10143-22-3	Carbamic acid, bis(hydroxymethyl)-, 2-methoxyethyl ester	
50883-78-8	Carbamic acid, dimethyl-, 2-methoxyethyl ester	
16672-66-5	2-Methoxyethyl methylocarbamate	
14983-42-7	Boric acid, tris (2-methoxyethyl) ester	
6163-73-1	Phosphoric acid, methoxyethyl ester	
42372-33-8	1-Naphthalenesulfonic acid, 6-diazo-5, 6-dihydro-5-oxo-, 2-methoxyethyl este	
17178-10-8	2-Methoxyethyl p-toluenesulfonate	
71550-36-2	1-Naphthalenesulfonic acid, 6-diazo-5, 6-dihydro-5-oxo-, 2-(2-methoxyethoxy) ethyl ester	

Table E-2.—Butyl-based glycol ether esters, amides, etc.*

CAS [†] number Name		
124-17-4	Butoxyethoxyethyl acetate	
112-07-2	Butoxyethyl acetate	
7251-90-3	Acrylic acid, 2-butoxyethoxy ester	
5330-17-6	2-Butoxyethyl chloroacetate	
27447-53-6	2-Butoxyethyl mercaptoacetate	
68797-46-6	Propanic acid, 2-chloro-, 2-butoxyethyl ester	
20442-06-2	2-Butoxyethyl butanoate	
20442-11-9	2-Butoxyethyl pelargonate	
109-37-5	Ethylene glycol monobutyl ether laurate	
109-38-6	Butoxyglycol stearate	
109-39-7	Ethylene glycol monobutyl ether oleate	
65520-45-8	Butanedioic acid, di-2-[2-(2-butoxyethoxy) ethoxy] ethyl ester	
65520-42-5	Pentanedioic acid, di-2-[2-(-2-butoxyethoxy) ethoxy] ethyl ester	
65520-46-9	Hexanedioic acid, di-2-[2-(2-butoxyethoxy) ethoxy] ethyl ester	
141-18-4	Adipic acid, bis (ethylene glycol monobutyl ether) ester	
141-17-3	Adipic acid, bis (diethylene glycol monobutyl ether) ester	
63021-23-8	Nonanedioic acid, bis (2-butyoxyethyl) ester	
70900-47-9	Nonanedioic acid, bis [2-(2-butoxyethoxy) ethyl] ester	
70900-46-8	Decanedioic acid, bis [2-(2-butoxyethoxy) ethyl] ester	
141-19-5	Decanedioic acid, bis (2-butoxyethyl) ester	
117-83-9	1,2-Benzenedicarboxylic acid, bis (2-butoxyethyl) ester	
16672-39-2	Phthalic acid, bis [2-(2-butoxyethoxy) ethyl] ester	
70900-48-0	1,2,4-Benzenetricarboxylic acid, tris [2-(2-butoxyethoxy) ethyl] ester	
62778-23-8	Cholest-5-en-3-ol (3-beta)-, 2-butoxyethyl carbonate	
5451-76-3	Benzoic acid, 2-butoxyethyl ester	

^{*}Adapted from TSCA [1977] and SANSS/CIS [1988].
†Chemical Abstracts Service.

Table E-2 (Continued).—Butyl-based glycol ether esters, amides, etc.

CAS number	Name
6661-54-7	Ethanol, 2-butoxy-, 4-methylbenzesulfonate
53404-31-2	2-(2,4-Dichlorophenoxy) propionic acid, butoxyethanol ester
19480-43-4	Acetic acid, (4-chloro-2-methylphenoxy)-, 2-butoxyethyl ester
30387-70-3	2,4,5-Trichlorophenoxypropionic acid n-butylglycol ester
32357-46-3	2,4-Dichlorophenoxybutyric acid, butoxyethyl ester
72152-95-5	Carbamic acid, [5-isocyanato-1,3,3-trimethylcyclohexyl) methyl]-, 2-butoxyethyl ester
78-51-3	Butoxyethyl phosphate
7332-46-9	2-(2-Butoxyethoxy) ethanol phosphate
64051-22-5	2-Butoxyethanol, hydrogen phosphate, diethylamine salt
14260-97-0	Dibutoxyethyl phosphate
64051-23-6	2-Butoxyethanol, dihydrogen phosphate, bis(diethylamine) salt
68133-43-7	Ethanol, 2-butoxy-, dihydrogen phosphate, dipotassium salt
14260-98-1	Butoxyethylphosphate

Table E-3.—Branched glycol ether esters, carbamates, etc.*

CAS [†] number	Name
68413-83-2	N,N-Dimethylol isopropoxyethyl carbamate
67952-46-9	Isopropoxyethyl carbamate
67952-44-7	Carbamic acid, (hydroxymethyl)-, 2-(1-methylethoxy) ethyl ester
16006-09-0	Carbamic acid, (2-isobutoxyethyl) ester
16005-83-7	Carbamic acid, bis (hydroxymethyl)-, 2-isobutoxyethyl ester
1464-69-3	2-Methyl-2-propenoic acid, 2-(ethenyloxy ethyl ether))
16839-48-8	Methacrylic acid, 2-(allyloxy) ethyl ester

^{*}Adapted from TSCA [1977] and SANSS/CIS [1988].
†Chemical Abstracts Service.

Table E-4.—Longer chain (butyl) glycol ether esters, phosphates, etc. *

CAS [†] number	Name	
20207-36-7	Lauric acid, 2-(hexyloxy) ethyl ester	
3538-36-1	Ethanol, 2-(hexyloxy)-, hydrogen phosphate	
64051-25-8	Ethanol, 2-(hexyloxy)-, dihydrogen phosphate, cmpd. with N-ethylethanamine (1:2)	
64051-24-7	Ethanol, 2-(hexyloxy)-, hydrogen phosphate, cmpd. with N-ethylethanamine (1:1)	
63294-54-2	Ethanol, 2-(hexyloxy)-, dihydrogen phosphate	
68757-58-4	Propanoic acid, 2,2-dimethyl-, 2-(hexyloxy) ethyl ester	
3694-74-4	Ethanol, 2-(tetradecyloxy)-, hydrogen sulfate, sodium salt	
56049-85-5	Triethylene glycol monohexadecyl ether sulfate ammonium salt	
55901-67-2	2-Decyloxyethyl sodium sulfate	
61894-66-4	2-Decyloxyethyl hydrogen sulfate	
14858-61-8	2-(Octadecyloxy) ethyl sodium sulfate	
14858-54-9	2-(Hexadecyloxy) ethyl sodium sulfate	
13150-00-0	Ethanol, 2-[2-(dodecyloxy) ethoxy] ethoxy]-, hydrogen sulfate sodium salt	
15826-16-1	Ethylene glycol monododecyl ether sulfate sodium salt	
25446-80-4	Triethylene glycol monomyristyl ether sodium sulfate	
25446-78-0	Sodium tridecyl tri(oxyethyl) sulfate	
3088-31-1	Diethylene glycol monododecyl ether sodium sulfate	
67923-90-4	Ethanol, 2-[2-(decyloxy)ethoxy]-, hydrogen sulfate, ammonium sa	
66104-67-4	2-Butenedioic acid (2)-, mono[2-[2-(dodecyloxy) ethoxy] ethoxy] ethyl]ester	
65138-77-4	Ethanol, 2-[20(tridecyloxy) ethoxy]-, dihydrogen phosphate	
65087-01-6	Ethanol, 2-[2-(tridecyloxy) ethoxy]-, hydrogen phosphate	
57119-83-2	2-[2-[[4-[(2-Bromo-4,6-dinitrophenyl)azo]-1-naphthyl]amino] ethoxy] ethano acetate (ester)	
57119-69-4	Ethanol, 2-[2-[[4-[(2-chloro-4,6-dinitrophenyl)azo]-1-naphthalenyl] amino]ethoxy]-, acetate (ester)	

*Adapted from TSCA [1977] and SANSS/CIS [1988].

†Chemical Abstracts Service.

Table E-4 (Continued).—Longer chain (butyl) glycol ether esters, phosphates, etc.

CAS number	Name	
21116-11-0	Ethanol, 2-[p-[[4-[(p-hydroxyphenyl)azo]-o-tolyl]azo] phenoxy]-, 1-(hydrogen sulfate), monosodium salt	
65993-31-5	Dicyclopentyloxyethyl acrylate	
66710-97-2	2-Propenoic acid, (1-methylethylidene) bis [(2,6-dibromo-4,1-phenylene) oxy-2, 1-ethanediyl] ester	
68400-37-3	7-Amino-4-hydroxy-3-[[4-[2-(sulfooxy)ethoxy]phenyl]azo]-2-napthalenesul-fonic acid	
68586-19-6	2-Propenoic acid, 2-methyl-, 2-[[2,3,3a,4,7,7a (or 3a,4,5,6,7,7a)-hexahydro-4,7-methano-1H-indenyl]oxy] ethyl ester	
70865-23-5	3-[[4-[2-(Sulfoxy)ethoxy]phenyl]azo][1,1'-biphenyl]-4-ol, monosodium salt	
71701-31-0	2-Naphthalenesulfonic acid, 4-hydroxy-3-[[4-[2-(sulfooxy)ethoxy]phenyl]azo] -7-[(2,5,6-trichloro-4-pyrimidinyl)amino]-, disodium salt	

Table E-5.—Phenoxy ethanol-based glycol esters, maleates, etc.*

CAS [†] number	Name
68141-05-0	Benzenoctadecanoic acid, 2-phenoxyethyl ester
103-60-6	Ethanol, 2-phenoxy-, isobutyrate
10534-77-7	Di(phenoxyethyl)maleate
46841-90-1	Mono(phenoxyethyl)maleate
48145-04-6	Ethanol, 2-phenoxy-, acrylate
58214-96-3	Butanoic acid, 3-methyl-, 2-phenoxyethyl ester
10595-06-9	2-Propenoic acid, 2-methyl-, 2-phenoxyethyl ester
23495-12-7	Propionic acid, 2-phenoxyethyl ester
23495-13-8	Pentanoic acid, 2-phenoxyethyl ester
23511-70-8	Butanoic acid, 2-phenoxyethyl ester
67845-81-2	Ethanol, 2-(2-phenoxyethoxy)-, benzoate
65379-23-9	Ethanol, 2-phenoxy-, dihydrogen phosphate, dipotassium salt

^{*}Adapted from TSCA [1977] and SANSS/CIS [1988].
†Chemical Abstracts Service.

Table E-6.—Substituted phenoxy ethanol glycol esters, acetates, etc.*

CAS [†] number	Name	
6807-11-0	Ethanol, 2-(4-methylphenoxy)-, acetate	
63217-11-8	Ethanol, 2-[2-(4-dodecylphenoxy) ethoxy]-dihydrogen phosphate	
52368-50-0	Decanoic acid, 2-[2-(nonylphenoxy) ethoxy] ethyl ester	
7347-19-5	2-(2,4,6-Tribromophenoxy) ethyl acrylate	
40184-38-1	2-(4'-Aminophenoxy) ethyl hydrogen sulfate	
68140-43-2	m-(2-Acetoxyethoxy) phenol	
56744-60-6	2-Propenoic acid, 2-methyl-, (1-methylethylidene) bis (4,1-phenyleneoxy-2, 1-ethanediyloxy-2,1-ethanediyl) ester	
56361-55-8	2-Propenoic acid, (1-methylethylidene) bis (4,1-phenyleneoxy-2, 1-ethanediyloxy-2, 1-ethanediyl) ester	
24447-78-7	2-Propenoic acid, (1-methylethylidene) bis (4,1-phenyleneoxy-2,1-ethanediyl) ester	
65133-66-6	2-Propenoic acid, 2-methyl-, 2-[4-[1-methyl-1-[4-[2-[2-[(2-methyl-1-oxo-2-propenyl)oxy] ethoxy]ethoxy] phenyl]ethyl[phenoxy] ethyl ester	

^{*}Adapted from TSCA [1977] and SANSS/CIS [1988].
†Chemical Abstracts Service.

Table E-7.—Glycol ether acetals*

CAS [†] number	Name	
5405-88-9	Diethylene glycol monomethyl ether formal	
71563-31-0	Propionaldehyde, bis (2-methoxyethyl acetal)	
71808-63-4	Butyraldehyde, bis (2-methoxyethyl acetal)	
71808-62-3	Isovaleraldehyde, bis (2-methoxyethyl acetal)	
71808-60-1	3,3-di (beta-Methoxyethoxy)-2-butanone	
71808-59-8	Isobutyraldehyde, bis(2-methoxyethyl acetal)	

^{*}Adapted from TSCA [1977] and SANSS/CIS [1988].
†Chemical Abstracts Service.

Table E-8.—Alkyl glycol ethers (methyl)*

CAS [†] number	Name	
52788-79-1	Diethylene glycol methyl tert-butyl ether	
112-49-2	1,2 bis(2-Methoxyethoxy)-ethane (Glyme-3)	
111-96-6	bis(2-Methoxyethyl) ether	
110-71-4	Dimethoxyethane	
7382-32-3	2-Butoxyethyl 2-methoxymethyl ether	
19685-21-3	Methyl triethylene glycol allyl ether	
66728-50-5	1-tert-Butoxy-2-methoxyethane	
54303-31-0	3-(2-Methoxyethoxy)-propanenitrile	
52808-36-3	2-(2-Methoxyethoxy) ethyl chloride	
3970-21-6	2-Methoxyethoxymethyl chloride	

^{*}Adapted from TSCA [1977] and SANSS/CIS [1988].
†Chemical Abstracts Service.

Table E-9.—Phenoxy-based glycol ethers*

CAS [†] number Name		
68385-79-5	N-[3-Amino-4-(2-methoxyethoxy) phenyl] acetamide	
67674-33-3	1-[(2-Methoxy)ethoxy]-2,4-dinitrobenze	
63810-51-5	2-(2-Methoxyethoxy)-4-nitrobenzenamine	
63810-54-8	2-(2-Methoxyethoxy)-5-nitrobenzenamine	
68703-73-1	N-[3-(Diethylamino)-4-(2-methoxyethoxy)phenyl] propanamide	
68703-72-0	N-[3-(Diethylamino)-4-(2-methoxyethoxy)phenyl] propionamide	
68703-71-9	N-[3-(Diethylamino)-4-(2-methoxyethoxy)phenyl] acetamide	
71230-65-4	N-[3-Amino-4-(2-methoxyethoxy)phenyl] propionamide	
71077-38-8	N-[3-(Ethylamino)-4-(2-methoxyethoxy)phenyl] acetamide	
71077-37-7	4-[2-Methoxyethoxy]-1,3-benzenediamine	
72175-36-1	2-[2-(2-Methoxyethoxy) ethoxy]-9,10-anthracenedione	
17869-10-2	1-Amino-4-hydroxy-2-(2-methoxyethoxy)-9,10-anthracenedione	
67846-62-2	N-[2-[(2-Chloro-4,6-dinitrophenyl)azo]-5-(ethylamino)-4-(2-methoxyethoxy phenyl] propanamide	
65059-45-2	1,4-Diamino-9,10-dihydro-N-[3-(2-methoxyethoxy)propyl]-9,10-dioxo-2,3 anthracenedicarboximide	
68597-67-5	N-[2-[(2-Chloro-4,6-dinitrophenyl)azo]-5-(ethylamino)-4-(2-methoxyethoxy) phenyl] acetamide	
71889-11-7	N-[2-[(2 Bromo-4,6-dinitrophenyl)azo]-5-(diethylamino)-4-(2-methoxyethoxy phenyl] propionamide	
71889-12-8	N-[2-Bromo-4,6-dinitrophenyl)azo]-5-(diethylamino)-4-(2-methoxyethoxy) phenyl] acetamide	
72066-86-5	N-[2-[(2-Bromo-4,6-dinitrophenyl)azo]-5-(ethylamino)-4-(2-methoxyethoxy) phenyl] acetamide	
72066-87-6	N-[2-[(2-Bromo-4,6-dinitrophenyl)azo]-5-(ethylamino)-4-(2-methoxyethoxy) phenyl] propionamide	
75198-92-4	1-Amino-4-[[2-[[[2-chloro-4-(2-methoxyethoxy)1,3,5-triazin-6-yl] amino]methyl]-4-methyl-6-sulfophenyl]amino]-2-anthraquinonesulfonic acid disodium salt	
73398-97-7	4-[[4-[[5-Cyano-2-[(8-methoxyoctyl)amino]-6-[(3-methoxy-propyl)amino]-4-methyl-3-pyridinyl]azo]-2,5-di-methylphenyl]azo]-N-[3-(2-phenoxyethoxy propyl] benzamide	

^{*}Adapted from TSCA [1977] and SANSS/CIS [1988].
†Chemical Abstracts Service.

Table E-10.—Ethoxy-ethanol based glycol ethers*

CAS [†] number	Name	
41771-35-1	1-Chloro-2-(2-ethoxyethoxy) ethane	
629-14-1	Ethylene glycol diethyl ether	
112-36-7	bis (2-Ethoxyethyl) ether	
10143-53-0	Diethylene glycol ethylvinyl ether	
51422-54-9	Ethylene glycol tert-butyl ethyl ether	
52788-80-4	Diethylene glycol ethyl-tert-butyl ether	

^{*}Adapted from TSCA [1977] and SANSS/CIS [1988].
†Chemical Abstracts Service.

Table E-11.—Siloxy glycol ethers

CAS* number	Name
2157-45-1	Silicic acid (H4SiO4), tetrakis (2-methoxyethyl) ester
1067-53-4	Tris (2-methoxyethoxy) vinyl silane
45117-69-9	Methylvinylbis (2-methoxyethyl) silane
57069-48-4	[(3-Methacryloxy)propyl] tris (2-methoxyethoxy) silane
17903-05-8	Tris(2-methoxyethoxy) phenyl silane
17980-64-2	Tris(2-methoxyethoxy) methyl silane
73545-23-0	N-[2-[[2-[[2-[(2-Aminoethyl) amino] ethyl] amino] ethyl] amino] ethyl]-N ¹¹¹¹ - [3-[tris(2-methoxyethoxy) silyl] propyl]nonanamide hydrochloride
24685-89-0	Ethanol, 2-(2-methoxyethoxy)-, tetraester with silicic acid (H4SiO4)
18407-94-8	Ethanol, 2-ethoxy-, tetraester with silicic acid (H4SiO4)
18765-38-3	Ethanol, 2-butoxy-, tetraester with silicic acid (H4SiO4)
68400-59-9	4,7,10-Trioxaundecyldimethylsilyl chloride

^{*}Chemical Abstracts Service.

Table E-12.—Ethoxy-ethanol based glycol ether esters, amides, etc.

CAS [*] number	Name	
112-15-2	Ethanol, 2-(2-ethoxyethoxy)-, acetate	
54396-97-3	Propanoic acid, 2-methyl-, 2-ethoxyethyl ester	
106-74-1	2-Propenoic acid, 2-ethoxyethyl ester	
7328-17-8	2-Propenoic acid, 2-(2-ethoxyethoxy) ethyl ester	
2370-63-0	2-Propenoic acid, 2-methyl-, 2-ethoxyethyl ester	
39670-09-2	Triethylene glycol monoethyl ether acrylate	
104-28-9	2-Propenoic acid, 3-(4-methoxyphenyl)-, 2-ethoxyethyl ester	
106-13-8	2-Ethoxyethyl dodecanoate	
67906-29-0	Octadecanoic acid, 10-hydroxy-9-sulfo-, 1-(2-ethoxyethyl) ester, monosodium salt	
68134-05-4	9-Octadecenoic acid, 2-ethoxyethyl ester	
37460-43-8	Ethanol, 2-ethoxy-, 4-nitrobenzoate	
605-54-9	Diethylglycol phthalate	
624-10-2	Sebacic acid, bis (2-ethoxyethyl) ester	
109-44-4	Hexanedioic acid, bis (2-ethoxyethyl) ester	
15484-00-1	2-(2-Ethoxyethoxy) ethyl carbonate	
68214-66-4	Carbamic acid, [2-[(2-chloro-4-nitrophenyl)azo]-5-(diethylamino)phenyl]-, 2-ethoxyethyl ester	
628-65-9	Carbamic acid, (2-ethoxyethyl) ester	
21578-97-2	2-Ethoxyethyl N-(7-hydroxynaphth-1-yl) carbamate	
70146-08-6	Carbamic acid, [3-(diethylamino)phenyl]-, 2-ethoxyethyl ester	

^{*}Chemical Abstracts Service.

Table E-13.—Acids and salts of glycol ethers

CAS* number	Name
67990-17-4	2-[(2-Butoxy)ethoxy] acetic acid, sodium salt
67990-18-5	2-[2-[(2-Hexyloxy)ethoxy]ethoxy] acetic acid, sodium salt
7420-07-7	Butoxyethoxypropionic acid
3139-99-9	2-Methoxyethanol, sodium salt
38321-18-5	Sodium 2-(2-butoxyethoxy) ethanolate
28099-67-4	bis (2-Methoxethoxy) calcium
14064-03-0	Magnesium ethoxyethoxide
52663-57-7	Sodium 2-butoxyethoxide
4084-36-0	Ethoxyethanol, compound with trifluoroborane
109-86-4	beta-Methoxyethanol
111-77-3	Diethylene glycol methyl ether
112-35-6	Methoxytriethylene glycol
111-90-9	Carbitol or 2-(2-ethoxyethoxy) ethanol
112-50-5	Ethoxytriethylene glycol
4353-29-1	3,6,9,12,15-Pentaoxaheptadecan-1-ol
2807-30-9	Ethylene glycol mono-N-propyl ether
6881-94-3	2-(2-Propoxyethoxy) ethanol
109-59-1	beta-Hydroxyethyl isopropyl ether
111-45-5	Ethylene glycol monoallyl ether
33065-62-2	1-(2-Hydroxyethoxy)-3-(2-propenyloxy)-2-propanol
3973-18-0	Ethylene glycol monopropargyl ether
111-76-2	Butoxyethanol
112-34-5	Butoxy diethylene glycol
143-22-6	Triethylene glycol mono-n-butyl ether
4439-24-1	Ethylene glycol isobutyl ether
18912-80-6	Diethylene glycol monoisobutyl ether
1606-85-5	bis(Hydroxyethyl) ether butynediol
	(Continued)

^{*}Chemical Abstracts Service.

Table E-13 (Continued).—Acids and salts of glycol ethers

AS number	Name
112-25-4	Ethylene glycol monohexyl ether
112-59-4	Diethylene glycol hexyl ether
25961-89-1	Triethylene glycol monohexyl ether
16394-44-8	2,2'-(1,4-Cyclohexylenedioxy) diethanol
1559-35-9	Ethylene glycol ethylhexyl ether
1559-37-1	Triethylene glycol 2-ethylhexyl ether
4536-30-5	Ethylene glycol monolauryl ether or lauryl alcohol oxy ethanol
3055-93-4	Lauryl alcohol mono (oxyethylene) ethanol
3055-94-5	Dodecyl triethylene glycol ether
14663-73-1	2-[2(Tridecyloxy) ethoxy] ethanol
4403-12-7	2-[2-[2-(Tridecyloxy) ethoxy] ethoxy] ethanol
56049-80-0	Diethylene glycol monopentadecyl ether
628-89-7	Ethylene glycol mono (2-chloroethyl) ether
68003-29-2	bis [2-(2-Hydroxyethoxy) ethyl] octylamine
53815-85-3	2-(2-(1-Naphthalenylamino) ethoxy) ethanol
1704-62-7	Dimethylaminoethoxyethanol
112-33-4	Diethylene glycol mono (aminopropyl) ether
140-82-9	Diethylaminoethoxyethanol
929-06-6	Aminoethoxyethanol
68141-01-5	2-[2-(3-Aminopropoxy) ethoxy] ethanol hydroxyacetic acid salt
68156-16-1	2-[2-(3-Aminopropoxy) ethoxy] ethanol hydrochloride
622-08-2	Benzyl hydroxyethyl ether
122-99-6	beta-Hydroxethyl phenyl ether
104-68-7	Diethylene glycol phenyl ether
711-82-0	Ethylene glycol alpha-naphthyl ether
901-44-0	Bisphenol A bis (2-hydroxyethyl) ether
15149-10-7	beta-Hydroxyethyl p-methylphenyl ether
104-39-2	2-[2-(p-Tolyloxy) ethoxy] ethanol
104-38-1	p-Di(2-hydroxyethoxy)benzene

Table E-13 (Continued).—Acids and salts of glycol ethers

CAS number	Name
102-40-9	m-bis (2-Hydroxyethoxy) benzene
6382-07-6	2-(p-tert-Pentylphenoxy) ethanol
20427-84-3	Diethylene glycol p-nonylphenyl ether
27176-93-8	Diethylene glycol mono(nonylphenyl) ether
61886-41-7	2-(p-Aminophenoxy) ethanol hydrochloride
18790-97-1	2-[2-[2-(p-Aminophenoxy) ethoxy] ethoxy] ethanol
66422-95-5	2-(2,4-Diaminophenoxy) ethanol dihydrochloride
16365-27-8	2-(p-Nitrophenoxy) ethanol
63134-26-9	2-[2-[4-Nitrophenoxy) ethoxy] ethoxy] ethanol
1892-43-9	p-Chlorophenyl glycol ether
60593-02-4	Hydroxyethyl pentabromophenyl ether
15480-00-9	2-(o-Chlorophenoxy) ethanol
23976-66-1	2-(2,4,6-Tribromophenoxy) ethanol
70715-17-2	2-[3-(6-Methyl-2-pyridinyl) propoxy] ethanol
64346-25-4	2-[(2,2,6,6-Tetramethyl-4-piperidinyl) oxy] ethanol
65104-24-7	2-[2-[[4-[(2-Bromo-4,6-dinitrophenyl) azo]-1-naphthalenyl] amino] ethoxy] ethanol
68039-37-2	2-[(3a,4,5,6,7,7a-Hexahydro-4,7-methano-1H-inden-5-yl)oxy] ethanol
67906-59-6	2-[4-[(4-Amino-5-methoxy-o-tolyl) azo] phenoxy] ethanol
57119-91-2	2-[2-[[4-[(2-Chloro-4,6-dinitrophenyl) azo]-1-napthalenyl] amino] ethoxy] ethanol
2192-20-3	2-[2[[4-(p-Chloro-alpha-phenylbenzyl)-1-piperazinyl] ethoxy] ethanol dihydrochloride
7070-15-7	beta-Hydroxyethyl isobornyl ether
4162-45-2	2,2'-[Isopropylidenebis [(2,6-dibromo-p-phenylene)oxy]] diethanol
2831-60-9	2-(2,4-Dinitrophenoxy) ethanol

Table E-14.—Miscellaneous glycol ethers*

CAS [†] number	Name
72403-65-7	Chromate (5-), bis [4-[[6-[[4-chloro-6-(2-ethoxyethoxy)-1,3,5-triazin-2-yl] amino]-1-hydroxy-3-sulfo-2-naphthalenyl] azo]-1-hydroxy-7-nitro-1-naphthalenesulfonato(4-)]-, pentasodium
71673-20-6	1-[2-(2-Ethoxyethoxy)ethyl]-2,2,4-trimethyl-1,2,3,4-tetrahydroquinoline
71673-19-3	7-Nitro-1-[2-(2-ethoxyethoxy) ethyl]-2,2,4-trimethyl-1,2,3.4- tetrahydro-quinoline
71673-14-8	7-Amino-1-[2-(2-ethoxyethoxy) ethyl]-2,2,4-trimethyl-1,2,3,4- tetrahydro-quinoline
71637-13-7	7-Acetamido-6-(2-bromo-4,6-dinitrophenylazo)-1-[2-(2-ethoxyethoxy) ethyl]-1,2,3,4-tetrahydro-22,4-trimethylquinoline
71637-12-6	7-Acetamido-6-(2-cyano-4,6-dinitrophenylazo)-1-[2-(2-ethoxyethoxy) ethyl]-1,2,3,4-tetrahydro-2,2,4-trimethylquinoline
71673-02-4	7-Acetamido -1-[2-(2-ethoxyethoxy) ethyl]-2,2,4-trimethyl -1,2,3,4- tetrahydroquinoline
70210-27-4	1-Amino-4-[[3-[[4-chloro-6-(2-ethoxyethoxy) -1,3,5-triazin-2-yl] amino] -2,4,6 rimethyl -5-sulfophenyl] amino]-9,10-dihydro-9,10-dioxo-2-anthracenesulfonicacid, disodium salt
65208-31-3	N-[2-[(2,6-Dibromo-4-nitrophenyl) azo] -5-[[2-(2-ethoxyethoxy) ethyl] ethylamino]phenyl] acetamide
65916-12-3	4-[(2,6-Dicyano-4-nitrophenol) azo]-N-[2-(2-ethoxyethoxy)ethyl] -N-ethyl-3-acetamidoaniline
67338-58-3	N-[2-(2-Ethoxyethoxy) ethyl] -N-ethyl-m-acetamidoaniline
23119-35-9	1,5-Dihydroxy-4,8-diamino-2-[4-(2-ethoxyethoxy) phenyl] anthraquinone
55993-15-2	2-[2-(2-Ethoxyethoxy) ethyl]-6-hydroxy -5-[(2-methyl 4-nitrophenyl) azo]-1H-benz[de] isoquinoline -1,3 (2H)-dione
68298-23-7	Propanoic acid, 3-(2-butoxyethoxy)-, sodium salt
68140-97-6	Propanoic acid, 3-(2-butoxyethoxy)-, potassium salt
68140-96-5	3-(2-Butoxyethoxy) propanenitrile
52788-78-0	Diethylene glycol butyl tert-butyl ether
1120-23-6	2-Butoxyethyl 2-chloroethyl ether
764-99-8	Diethylene glycol divinyl ether
143-29-3	bis (Butoxyethoxy) methane
143-29-3	bis (Butoxyethoxyethoxy) methane (Continue

^{*}Adapted from TSCA [1977] and SANSS/CIS [1988].
†Chemical Abstracts Service.

Table E-14 (Continued).—Miscellaneous glycol ethers

Butoxyethoxy propanol
Dibutoxy diethylene glycol
alpha-[2-[2-)n-Butoxy)ethoxy] ethoxy] -4,5-methylene dioxy-2-propyltoluene
Ethylene glycol diallyl ether
Diethylene glycol bis (allyl) ether
1-[2-[2-Propenyloxy)ethoxy] butane
1-(2-(1,1-Dimethylethoxy)ethoxy) butane
Triethylene glycol dichloride
Ethylene glycol bis (chloromethyl ether)
1-[2-[2-(2-Chloroethoxy) ethoxy]-4-octyl benzer.e
Diisobutylphenoxyethoxyethylchloride
1-[2-(2-Chloroethoxy)ethoxy]-4-(1,1,3,3-tetramethylbutyl) benzene
di (3-Aminopropoxy) ethane
Diethylene glycol bis (3-aminopropyl) ether
N,N-Dimethyl-2-[2-[4-(1,1,3,3-tetramethyl butyl) phenoxy]ethoxy - ethanamin
3,3'-(Ethylenedioxy) bis (propylammonium) adipate
Diisobutylphenoxyethoxyethyldimethylamine
Diisobutylcresoxyethoxyethyldimethylamine
Ethylene diglycidyl ether
Diethylene glycol bis (2-cyanoethyl) ether
Ethylene glycol bis (2-cyanoethyl) ether
(Diisobutylphenoxy) ethoxyethoxyethane sulfonic acid, sodium salt
2-[2-[[2,2,4(or 2,4,4)-Trimethylpentyl]phenoxy]ethoxy] ethanesulfonic acid, sodium salt
N,N-Dimethyl-2-[2-[2-methyl-4-(1,1,3,3-tetramethylbutyl) phenoxy]ethoxy] ethanamine
1,2-bis (3-Hydroxyphenoxy) ethane
Ethylene glycol diphenyl ether
1,2-bis (4-Carboxyphenoxy) ethane
Benzyldiisobutyl [2-(2-phenoxyethoxy)ethyl] ammonium chloride

Table E-14 (Continued).—Miscellaneous glycol ethers

CAS number Name		
14417-67-5	1,2-bis (Pyridinomethoxy) ethane dichloride	
17418-59-6	1-Amino-4-hydroxy-2-(2-phenoxyethoxy) anthraquinone	
41312-86-1	1,4-Diamino-2-chloro-3-(2-phenoxyethoxy) anthraquinone	
63833-78-3	5-[(2-Cyano-4-nitrophenyl) azo]-6-[2-hydroxyethyl) amino]-4-methyl-2-[[3-(2-phenoxyethoxy) propyl] amino]-3-pyridinecarbonitrile	
63281-10-7	5-[[2-Chloro-4-(methylsulfonyl) phenyl] azo]-4-methyl-2,6-bis [[3-(2-phenoxyethoxy) propyl] amino]-3-pyridinecarbonitrile	
63281-05-0	4-Methyl-2,6-bis [[3-(2-phenoxyethoxy) propyl] amino]-5-[[4-(phenylazo) phenyl] azo]-3-pyridinecarbonitrile	
63281-03-8	5-[[2-Chloro-4-(phenylzap) phenyl] azo]-4-methyl-2,6-bis[[3-(2-phenoxyethoxy) propyl] amino]-3-pyridinecarbonitrile	
68299-27-4	Nonabromomonochloro-1,2-diphenoxyethane	
121-54-0	p-Diisobutyl(phenoxyethoxy) ethyl] dimethylbenzylammonium chloride	
61262-53-1	1,1'- [1,2-Ethanediylbis (oxy)] bis [2,3,4,5,6-pentabromo-benzene	
23421-22-9	4,4"-[Oxybis(ethyleneoxy)]bis [2-hydroxy benzophenone]	
37853-59-1	1,2-bis(2,4,6-Tribromophenoxy) ethane	
67923-87-2	2-[2-[0-(Octylphenoxy) ethoxy] ethoxy] ethanesulfonic acid, sodium salt	
72953-52-7	1-Amino-2-[2-(bromophenoxy) ethoxy]-4-hydroxyanthraquinone	
72953-51-61	1-Amino-2-[2-(dibromophenoxy) ethoxy] -4-hydroxyanthraquinone	

APPENDIX F

BACKGROUND OF METHODS USED FOR ANALYSIS OF EAA and MAA IN URINE

Smallwood et al. [1984] developed a method for analyzing the glycol ether metabolites EAA and MAA in urine. The method was based on (1) methylene chloride extraction of "spiked" acidified human urine, (2) pentafluorobenzyl bromide (PFBB) derivitization, and (3) gas chromatography analysis using flame ionization detection (FID). Urine (1 ml) was adjusted to pH 2 with HCl and extracted three times with methylene chloride. Phase transfer catalysis (a combination of ion-pair extraction and fluoroanhydride derivitization) was done by adding alkaline tetrabutylammonium hydrogen sulfate and PFBB to the methylene chloride extract. The mixture was rotated for 2 hr. Gas chromatography was employed to analyze 5 μ l of the methylene chloride layer (bottom layer) using FID and a 6 ft × 1/4 in (4-mm id) glass column (packed with 1.95% QF-1 and 1.5% OV-17 on 80/100 mesh Supelcoport). Detection limits for MAA and EAA were 11.4 and 5.0 mg/liter of urine, respectively. Average recoveries (and relative standard deviations) were 78% (0.17) for MAA and 91% (0.14) for EAA.

Groeseneken et al. [1986a] developed a method for determining MAA and EAA in urine based on lyophilization of urine samples followed by derivitization with diazomethane. After adjustment of urine specimens to pH 8 to 8.5 with KOH, 1 ml of urine and 50 µg of 2-furonic acid (FA) (internal standard) were added, and the sample was lyophilized. The dry residue was redissolved in 1 ml methylene chloride with added HCl and derivitized with diazomethane in methylene chloride. Gas chromatographic analysis using FID was performed on a fused silica capillary column (CP WAX 57 CB, 25 m × 0.33 mm id) with a split ratio of 10:1. The detection limits of MAA and EAA were 0.15 and 0.07 mg/liter of urine, respectively. Mean recoveries of MAA, EAA, and FA added to "blank" urine samples were 31.4%, 62.5%, and 58.4%, respectively; the recoveries of MAA and EAA were well correlated with those of the internal standard. Day-to-day variability for MAA and EAA was 6.0% and 6.4%, respectively; the corresponding within-day variability was 6.2% and 8.9%.

^{*}See References beginning on page 262.

Smallwood et al. [1988] developed and validated a method for analysis of EAA in urine. Two ml of urine, along with potassium carbonate, tetrabutylammonium hydrogen sulfate, methylene chloride, and PFBB were added to a screw-top culture tube. After 2 hr of mixing on a rotator at 60 rpm, the tube was heated for 20 min in a 50°C water bath. Additional mixing at room temperature, removal of the upper aqueous layer, and washing of the lower methylene chloride layers with distilled water removed unreacted reagents. The methylene chloride extract was dried with anhydrous sodium sulfate and loaded into an autosampler vial. Automated gas chromatographic analysis using FID was conducted with the use of a 6 ft × 4 mm id glass column packed with 4% SE-30 and 6% OV-210 on 100/120 mesh Chromosorb WHP. Standards were prepared in pooled urine. The analytical range for EAA was 5 to 100 mg/liter of urine; the limit of detection was 4 mg/liter; and the limit of quantitation was 7 mg/liter. Within-day variation was 0.5% to 1.8%, and day-to-day variation was 3.0% to 4.7%. Sample stability was confirmed for at least 8 months when specimens were stored at -20°C. The authors stated that the method could also be used for MAA and butoxyacetic acid (BAA) in urine. Preliminary data were presented in the paper indicating that the technique has the potential for assessing EGEE exposure in shipyard painters who use paints containing EGEE.

Groeseneken et al. [1989b] observed that MAA appeared in the chromatogram of control subjects not exposed to EGME. Further investigation revealed that the diazomethane procedure was producing MAA by reacting with the hydroxyl group of naturally occurring glycolic acid. Groeseneken et al. [1989b] further evaluated the existing methods for determining alkoxyacetic acids and concluded that the phase transfer catalysis procedures had the required specificity, without the production of artifacts, but lacked sufficient sensitivity to detect these metabolites at low occupational exposure concentrations. On the other hand, the methods utilizing diazomethane derivitization had the required sensitivity but lacked the specificity. Therefore, Groeseneken et al. [1989a] developed an improved method that combined the best attributes of the two basic existing methods.

The procedure developed by Groeseneken et al. [1989b] was described as follows. Urine was adjusted to pH 7; 1-ml aliquots were placed in small vials with 3-chloropropionic acid (internal standard) and lyophilized overnight. The dry residue was redissolved in methanol containing PFBB, and the vials were capped. The vials were heated at 90°C for 3 hr. After cooling, sample cleanup was done by adding distilled water and extracting the pentafluorobenzyl-esters (PFB-esters) with methylene chloride. The methylene chloride extract was analyzed by gas chromatography using FID. A fused silica capillary column was used (CP Sil 5, 25 m × 0.32 mm id, 0.21 µm film thickness) with a split ratio of 5:1. Temperature programming was employed. All PFB-esters showed baseline resolution; retention times of 6.53 min (MAA), 7.77 min (EAA), and 8.59 min (internal standard) were observed. A typical gas chromatographic run, including cool-down and equilibration times, required about 30 min.

Optimization studies were done for reagent concentrations as well as for urinary pH and reaction time. After correction for the partial solubility of methylene chloride in the 50:50 methanol:urine phase, recoveries of alkoxyacetic acids from urine averaged 95.0% (MAA), 94.8% (EAA), and 95.1% (BAA). The yield for the derivitization reaction averaged 99.5%

(MAA) and 101.8% (EAA). Standard curves were set up for urine and were linear over the range of 0.1 to 200 mg/liter. The limit of detection, at a signal-to-noise ratio of 5, for the two acids was 0.03 mg/liter. Precision of the method, calculated from triplicate injections of 40 urine samples, averaged 3.5% (RSD), ranging from 1.1% at 25 mg/liter to 20% at 0.1 mg/liter.

NIOSH has not validated the Groeseneken et al. methods [1986a, 1989b].

APPENDIX G

GUIDELINES FOR BIOLOGICAL MONITORING

G.1 Monitoring

The frequency of biological monitoring should be tied to work practices and the use of the glycol ethers. Dermal absorption of glycol ethers can be significant. Compliance with the NIOSH RELs without a biological monitoring evaluation may not protect workers from the potential adverse effects of glycol ethers.

Urine samples should be evaluated for alkoxyacetic acid metabolites using the method of Groeseneken et al. [1989b]* or an equivalent method. Expression of results as milligrams of metabolite per gram of creatinine (mg/g creatinine) is suggested.

Factors that may affect the urinary levels of EAA and MAA include ethanol consumption (which lowers urinary metabolite levels) and dermal contact, heavy work, and nonoccupational exposures (all of which raise urinary levels).

Urine sample collection times are specific for the individual glycol ethers:

- EGEE and EGEEA: Urine samples should be collected at the beginning of the shift on the last working day of the workweek. This specimen would represent the integrated weekly exposure (dermal and inhalation).
- EGME and EGMEA: Urine samples should be collected preshift on the first day of
 the workweek following a typical workweek of exposure. This specimen would
 reflect the integrated exposure (dermal and inhalation) from the previous week.

Measurable concentrations of EAA or MAA in the urine are an indication of uptake of the respective glycol ethers by either inhalation or skin exposure. The concentrations reflect nonoccupational as well as occupational exposure and are not likely to correlate with the NIOSH RELs. If concentrations of EAA and MAA exceed the estimated guidelines below, then exposure to glycol ethers has occurred, but not necessarily at concentrations above the NIOSH RELs. A thorough industrial hygiene evaluation, with specific emphasis on possible dermal absorption, should be conducted to determine the source of exposure. The following guidelines are suggested until better documented guidelines are developed.

See References beginning on page 262.

- 1. The presence of EAA in urine specimens (collected as specified) above a concentration of approximately 5 mg EAA/g creatinine is evidence for a single EGEE and/or EGEEA inhalation exposure corresponding to an 8-hr exposure to 0.5 ppm EGEE and/or EGEEA at 60 W of exercise. This value was extrapolated from 4-hr experimental exposures at 5 ppm at 60-W workload to an 8-hr exposure at 0.5 ppm at 60-W workload [Groeseneken et al. 1986c, 1987b] using the principle of superposition [Gibaldi and Perrier 1982].
- 2. The presence of MAA in urine specimens (collected as specified) above a concentration of approximately 1 μg/ml (equivalent to approximately 0.8 mg/g creatinine) is evidence of inhalation-only exposure to EGME and/or EGMEA. This value was extrapolated from 4-hr laboratory exposures at 5.1 ppm EGME at rest to 8-hr exposure to 0.5 ppm at 60-W workload [Groeseneken et al. 1989a]. The excretion of MAA was relatively constant 4 hr after the end of a 4-hr exposure. Therefore, doubling the urinary EAA concentration after a 4-hr exposure is a reasonable estimate of urinary values following an 8-hr exposure (based on Figure 1 [Groesenken et al. 1989a]). Extrapolation to 0.1 ppm may produce MAA levels below the detection limit of the method. There are no data on occupational exposures, but based on EGEE data and on the strong possibility of concurrent dermal absorption of EGME, concentrations of MAA may be higher than 0.8 mg/g creatinine in urine specimens collected from workers.

G.2 Justification For Recommendations

Biological monitoring for glycol ether exposure is recommended, even though no validated guidelines can be provided as to the relationship between airborne exposure to glycol ethers and the alkoxyacetic acid urinary metabolites. The alkoxyacetic acid metabolites (EAA and MAA) are not only an index of exposure or uptake of EGEE or EGME by the worker, but they are also an index of potential adverse health effects from these glycol ethers.

Dermal absorption may be a major route of exposure to EGME and EGEE and their respective acetates. The potential exists for absorption of glycol ether vapors through wet skin.

The influence of workload is significant for inhalation exposure. Doubling the workload results in twice the uptake of the glycol ethers.