National Bureau of Standards Certificate of Analysis

Standard Reference Material 8j

Bessemer Steel (Simulated), 0.1% Carbon

	С	Mn	P	S	Si	Cu	Ni	Cr	V	Мо
ANALYST	Combustion- chromatographic	Persulfate- Arsenite	Spectrophoto- metric	Combustion Titration	Perchloric acid dehydration	Spectrophoto- metric	Spectrophoto- metric			Spectrophoto- metric
1	0.081a	0.505b	0.098	0.077c	0.059d	0.022e	0.114	0.048f	0.015g	0.037
2	.080	.500	.095 ^h	.076	.048d	.018e	.110 ⁱ	$\{.048^{j}\}$ $\{.046^{k}\}$.016 ^k	.038
3	.082	.505	$\left\{ \begin{array}{c} .094^{l} \\ .095^{h} \end{array} \right\}$.077¢	$\{.056^{d}\}\$.020m	$\left\{ .116^{n} \right\} $.046°	.014p	.041
4	.080	{.509 .508 ^q }	$\left\{ \begin{array}{c} .095^{\rm h} \\ .094 \end{array} \right\}$.078°	$\left\{ .063^{ m d} \atop .065^{ m r} ight\}$.020m	$\left\{ .110^{n} \right\}$.046s	.015t	.038
Average	0.081	0.505	0.095	0.077	0.058	0.020	0.113	0.047	0.015	0.038

^al-g sample, tin-coated copper accelerator. bPotentiometric titration.

c_{1-g} sample burned in oxygen at 1425 C and sulfur dioxide absorbed in starch-iodide solution. Iodine liberated from iodide by titration, during the combustion,

with standard KIO₃ solution. dDouble dehydration.

^eDiethyldithiocarbamate photometric method. ^fChromium separated from the bulk of the iron in a 10-g sample by NaHCO₃ hydrolysis, oxidized with peroxydisulfate, and titrated potentiometrically with ferrous ammonium sulfate.

^gVanadium separated as in (f), oxidized with HNO₃ and titrated potentiometrically with ferrous ammonium sul-

hAlkali-molybdate.

Same value obtained by gravimetric method.

Diphenylcarbazide photometric method.

Atomic absorption spectrometry.
Gravimetric — Mg₂P₂O₇.

MNeocuproine photometric method.

Dimethylglyoxime gravimetric method.

Peroxydisulfate oxidation — Fe(NH₄)(SO₄)₂ — KMnO₄

titration.
PFeSO₄(NH₄)₂ S₂O₈-KMnO₄ titration.
qKIO₄ photometric method.
Silicomolybdate photometric method. Schromium oxidized with peroxydisulfate and titrated amperometrically with ferrous ammonium sulfate. tKMnO₄-KNO₂-Urea-Fe(NH₄)₂(SO₄)₂.

The material for the preparation of this standard was prepared at the Carpenter Technology Corporation, Reading, Pennsylvania.

The overall direction and coordination of the technical measurements leading to certification were performed under the chairmanship of O. Menis and J. I. Shultz.

The technical and support aspects involved in the preparation, certification, and issuance of this Standard Reference Material were coordinated through the Office of Standard Reference Materials by R. E. Michaelis.

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J. Paul Cali, Chief Office of Standard Reference Materials

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