U. S. Department of Commerce
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## Certificate

## Standard Reference Material 4222 Beta-Ray Standard n-Hexadecane-1-Carbon-14

This standard consists of approximately 3 grams of carbon-14 labeled n-hexadecane in a flame-sealed glass ampoule.

The activity in disintegrations per second per gram of solution as of June 1967, was

 $*3.93_2 \times 10^4 \pm 3.0_5\%$ \*.

Aliquots of the solution from which this standard was prepared, were converted to sodium carbonate-<sup>14</sup>C and compared, by liquid-scintillation counting, with the National Bureau of Standards sodium carbonate-<sup>14</sup>C standard.

The value reported above is based on 37 liquid-scintillation comparisons, and the uncertainty,  $\pm 3.0_5$  percent, is the sum of the computed standard error,  $0.5_5$  percent, at the 99.73-percent confidence level (3.23 $\sigma$ ), the estimated limits of the systematic error,  $\pm 1.0_0$  percent, in the hexadecane-sodium carbonate conversion, and the overall uncertainty,  $\pm 1.5_0$  percent, assigned to the National Bureau of Standards sodium carbonate-<sup>14</sup>C standard.

The beta-ray and gamma-ray spectra were examined with anthracene-crystal and NaI(Tl)-crystal spectrometers and no radioactive impurities were observed.

A half life of 5730±40 years was adopted as the best value at the Fifth Radiocarbon Dating Conference, Cambridge, England, 1962 [Nature 195, 984 (1962)].

This standard was prepared and calibrated in the Institute for Basic Standards, Radiation Physics Division, by members of the Radioactivity Section, W. B. Mann, Chief.