UNITED STATES DEPARTMENT OF COMMERCE WASHINGTON 25, D.C.

National Bureau of Standards Certificate of Analyses

Standard Sample 172 Ferroboron

ANALYST	BORON	SILICON	CARBON	ALUMINUM
1	{ a 13.68 b 13.70 }	· 3.57	0, 238	{
2	f 13.73	s 3.72	. 245	e. 046
3	• 13.63	° 3.66	. 23	
4	a 13. 75	° 3.71	. 233	e. 050 h. 055
5	a 13. 57	o 3.51	. 222	h. 060
erage	13.68	3.63	0. 234	0. 053

^{*}One-g sample fused with Na₂O₂. Melt dissolved in dilute HCl and an aliquot (=0.2 g) passed through a column of a strong acid type ion exchange resin. Eluate nearly neutralized, and refluxed to remove CO₂. Solution adjusted to pH 6.9, invert sugar or mannitol added, and solution titrated potentiometrically to pH 6.9 with 0.1N NaOH standardized on H₃BO₃. See method E31-59T, Am. Soc. Testing Materials.

b Distillation-titration method.

° Na₂O₂ fusion-H₂SO₄ dehydration. See ASTM method E31-59T.
d Na₃O₅ fusion. Melt leached with water. Solution boiled to decompose peroxide. An aliquot of the supernatant liquid acidified. Iron and the like precipitated with cupferron. Cupferron, in the filtrate, destroyed with HNO₃ and HClO₄. Aluminum precipitated with NH₄OH, weighed as Al₂O₃, and corrected for P₂O₅ and Fe₂O₃.
e As in footnote d to destruction of excess cupferron, and

aluminum then determined by aurintricaryboxylic acid (Aluminon) photometric method. See ASTM method E31-59T.

f Na₂CO₃—KNO₂ fusion. Melt dissolved in HCl. Iron, etc., precipitated with CaCO₂. Solution filtered. Boron titrated in the filtrate.

g Na₂CO₃—KNO₃ fusion. HCl dehydration.

h Eriochrome Cyanin-R photometric method.

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The alloy for the preparation of this standard was furnished by the Molybdenum Corp. of America.

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