UNITED STATES DEPARTMENT OF COMMERCE NATIONAL BUREAU OF STANDARDS WASHINGTON 25, D.C.

National Bureau of Standards Certificate

Standard Sample 17 Sucrose

Moisture	less than	0.01%
Ash		0.003%
Reducing substances, estimated as invert sugar	less than	0.02%

Each 100 ml of a normal sucrose solution contains 26.000 g of dried substance, weighed with brass weights in air (760 mm pressure, 20 °C, 50 percent relative humidity). At 20 °C, this solution in a 200-mm polariscope tube reads 100 °S (International Sugar Degrees). The illumination is white light filtered through a 15-mm layer of a 6-percent solution of potassium dichromate. The International Sugar Scale was defined and adopted by the International Commission for Uniform Methods of Sugar Analysis at the Eighth Session, Amsterdam, 1932 [Intern. Sugar J. 35, 17 (1933); NBS Circ. 440, p. 79, p. 775 (1942)].

The rotation in circular degrees of the normal sucrose solution observed in a 200-mm polariscope tube, for wavelength 5461 Å is 40.763° and for wavelength 5892.5 Å is 34.617°.

The specific rotations of sucrose for the normal solution are:

$$[\alpha]$$
 $\begin{array}{ccc} 20 \text{ °C} \\ 5461\text{ Å} \end{array} = 78.342\text{ °} \qquad 26 \text{ g per } 100 \text{ ml} \end{array}$

[a]
$${20 \text{ °C} \atop 5892.5\text{Å}} = 66.529\text{°}$$
 26 g per 100 ml

[Bul. BS 13, 67 (1916) S268; NBS Circ. 440, p. 82 (1942).]

For the Director

Chief, Organic Chemistry Section Physical Chemistry Division

Lot No. 6340 April 1963

GPO 683779