

U. S. DEPARTMENT OF COMMERCE

National Bureau of Standards

Certificate of Analyses

OF

STANDARD SAMPLE 135

CHROMIUM-MOLYBDENUM STEEL

ANALYST*	C	Mn	P	S	Si	Cr	Mo								
	<i>Direct combustion</i>	Bismuthate ($\text{FeSO}_4 \cdot \text{KMnO}_4$)	Persulfate-Arsenite	Gravimetric (weighed as MgPO_4 after removal of arsenic)	Alkali-Molybdate ^a	Gravimetric (direct oxidation and final precipitation after reduction of arsenic)	<i>Evolution (HCl sp gr 1.18-ZnS-iodine theoretical sulfur titer)^b</i>	<i>Combustion</i>	<i>Pechloric acid dehydration</i>	<i>COPPER</i> $\text{H}_2\text{S}-\text{CuS}-\text{CuO}$	<i>NICKEL</i> Weighed as nickel dimethylglyoxime	<i>FesO₄-KMnO₄ titration</i>	<i>VANADIUM</i>	<i>Gravimetric</i>	<i>Colorimetric</i>
1.	0.092	c. 0.460	0.016	d. 0.016	0.010	0.009	e. 0.384	0.080	0.086	f. 5.16	0.010	g. 0.577	h. 0.024		
2.	.093	0.453	.018	.018	.008	.009	j. 398	.077	.079	k. 15		l. 563	m. .025		
3.							m. 380			n. 15		o. 57	p. 0.56		
4.	.095	c. 46	n. 015	.010	o. n. 009		.377	p. 074	q. 085	k. 14		r. 59	s. .025		
	.097	.45	.018	.010	.011		.376	.085	q. 09	5. 18			.59		
	.097	s. 461	.019	t. 007			.375	u. 074	.079	5. 14		v. 582	w. .58	x. .025	
	.088	w. 453	.017	.013			.385	x. 07	.084	f. 5.14		y. 572			
8.	.090	.468	.020		.010		j. 392	.076	.083	5. 17		v. 56			
9.	.094	s. 461	.017		n. 012		e. 383	c. 077	q. 074	k. 5.15		r. 59	s. .023		
10.	.097	.457	.017	.012		z ² . 379	z ³ . 070	q. 088	k. 5.15		z ⁴ . 575	.57			
Averages General	0.094	0.457	0.459	0.017	0.010	0.010	0.383	0.076	0.083	5. 15	0.010	0.571	0.58	0.024	
Average	0.094	0.458	0.017			0.010	0.383	0.076	0.083	5. 15	0.010	0.575	0.024		

* Precipitated at 40° C, washed with a 1-percent solution of KNO_3 , and titrated with alkali standardized by the use of National Bureau of Standards acid potassium phthalate and the ratio 23NaOH:1P.

^a Value obtained by standardizing the titrating solution by means of sodium oxalate through KMnO_4 and Na_2S_2O_3 , and use of the ratio 2I:1S.

^b Chromium removed by precipitation with ZnO .

^c Colorimetric method. See J. Research NBS 26, 405 (1941) RP1386.

^d Double dehydration.

^e Persulfate oxidation and potentiometric titration with ferrous ammonium sulfate solution standardized with recrystallized potassium dichromate.

^f Alpha-benzoinoxime method. See BS J. Research 9, 1 (1932) RP453.

^g Determination made by M. Marie Cron, by the

vacuum-fusion method. See BS J. Research 7, 375 (1931) RP346.

^h Chromium removed by precipitation with Na_2HCO_3 .

ⁱ Sulfuric acid dehydration.

^j Perchloric acid oxidation.

^k Weighed lead molybdate.

^l Double dehydration in HClO_4 . Residues fused with Na_2S_2O_7 . Sulfuric acid added to partially cooled melt and solution fumed.

^m Titrating solution standardized by the use of a standard steel.

ⁿ Solution in HCl (1:1).

^o Copper precipitated with KCNS. Finished by KI-Na_2S_2O_3 titration.

^p Glyoxime precipitate titrated with KCN.

^q Solution-distillation method. Sample dissolved in HCl (1:1).

^s Chromium volatilized as CrO_2Cl_2 .

^t Meineke method.

^u Copper precipitated with Na_2S_2O_3 . Finished by electrolysis.

^v Solution-distillation method. See Ind. Eng. Chem., Anal. Ed. 11, 303 (1932).

^w Bismuthate-arsenite.

^x Copper precipitated with H_2S . Finished by KI-Na_2S_2O_3 titration.

^y $\text{H}_2\text{S}-\text{MoO}_3-\text{MoO}_4$.

^z Chromium removed as lead perchlorate.

^{aa} Distillation-titration. Solution and digestion in H_2SO_4 containing H_3PO_4 and selenium.

^{ab} Sulfuric-perchloric acid dehydration.

^{ac} Copper precipitated with KCNS. Finished by titration with KCN .

^{ad} Cyanate-cinchonine-thiosulfate method. See Iron Age 132 No. 2, 16 (1933).

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The steel for the preparation of this standard was furnished by the Crucible Steel Company of America.