#### U. S. DEPARTMENT OF COMMERCE WASHINGTON

## National Bureau of Standards

## Certificate of Analyses

# Standard Sample 127A

### **SOLDER**

(30 Sn-70 Pb)

ANALYST	TIN SnCl <sub>2</sub> -KIO <sub>3</sub>	ANTIMONY	ARSENIC	BISMUTH	COPPER	NICKEL Colorimetric	SILVER	
1	$\begin{bmatrix} 30.03 \\ a30.03 \end{bmatrix}$	³ 0. 79	°0.132	₫ <b>0.</b> 039	°0.003	<sup>1</sup> 0. 002	°0.003	
2	30.01	g.78	h.128	a.036	¹.004	. 002		
3	30. 01	g.79	h.13	٠.04	<sup>1</sup> .004	k.002		
4	¹30. 08	₹.79	m. 138	n. 035	°.004	. 002	<u> </u>	
5	[P30. 04] [P30. 06]	g.78	h.125					
t	30.03	₹.79	h.125	r.035	°.004	.002	* . 004	
7	<sup>1</sup> 29. 99	t.79	<sup>u</sup> . 125	₹.040		w.002		
8	30.02	g.80	h.124	r.036	3.004			
9	30.05	g.79	h.130	d. 035	1.004	. 002		
10	30. 07	g.80	m. 131	*.031	y.004	. 002		
Average	30.03	0.79	0. 129	0.036	0.004	0.002	0.004	

<sup>a</sup> Tin separated by distillation from a 0.7-g sample, precipitated with cupferron and ignited to SnO<sub>2</sub>. See J. Research NBS 33, 307 (1944) RP1610.

<sup>b</sup> Antimony separated by distillation from a 5-g sample as described in J. Research NBS 21, 95 (1938) RP1116. Distillate treated with Hi.S. Antimonous sulfide dissolved and titrated with KMnO<sub>4</sub>.

<sup>c</sup> Sodium hypophosphite-ammonium molybdate-photometric method.

<sup>d</sup> Antimony, arsenic, and tin volatilized from a 1-g sample with HBr-HClO<sub>4</sub>. Bismuth determined by the thiourea-photometric method. See J. Research NBS 47, NO<sub>4</sub> 4 (Oct. 1951) RP2250.

<sup>e</sup> Dithizone-photometric method.

Polarographic method. Determination made by J. K. Taylor.

"Antimony reduced with H<sub>2</sub>SO<sub>3</sub> and titrated with KBrO<sub>3</sub>. See ASTM method E-46 or E-57.

A Arsenic separated by distillation and titrated with KBrO<sub>3</sub>.

Copper separated with metallic lead and determined photometrically with HBr.

Internal electrolysis. See ASTM method E-46.

A Spectrographic analysis.

Tin reduced with lead and titrated with iodine.

Bismuth separated by internal electrolysis and determined gravimetrically.

Deposition by internal electrolysis and titration with iodide-thiosulfate.

Tin reduced with nickel or lead and titrated with KIO<sub>3</sub>.

Tin reduced with nickel or lead and titrated with iodine.

- r Bismuth separated by internal electrolysis and determined photometrically with thiourea.
  Silver separated by internal electrolysis and determined photometrically as AgCl.
  'Titration with KMnO4.
  Ansenic separated by distillation and titrated with indire.
- iodine.
  \*Bismuth separated by internal electrolysis and deter-
- \*Bismuth separated by internal electrolysis and determined photometrically.

  \* Weighed as nickel dimethylglyoxime.

  \* Lead button, obtained from the fire assay of a 10-g sample, scorified and weighed. Bismuth determined in a sample of the button by the thiourea-photometric method.

  \* Copper determined photometrically on a 10-g sample after removal of tin and antimony by HBr-H<sub>2</sub>SO<sub>4</sub> treatment, and the other wifers.
- and lead as the sulfate.

#### List of Analysts

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- C. B. Craver, Roland Leoffler, and J. L. Pinkston, The Eagle-Picher Company of Texas, Dallas, Tex.
- 8. J. W. Claypool, Nassau Smelting and Refining Co., Tottenville, N. Y.
- 9. R. G. Ernst, United States Metals Refining Co., Carteret, N. J.
- 10. Silve Kallmann, Ledoux and Co., Inc., New York, N. Y.

The metal for the preparation of this standard was furnished by Federated Metals Division, American Smelting and Refining Co., and atomized by the Metals Disintegrating Co.

Washington 25, D. C., August 20, 1951.

A. V. Astin, Acting Director.