

U. S. DEPARTMENT OF COMMERCE

National Bureau of Standards

Certificate of Analyses

OF

STANDARD SAMPLE 10E

BESSEMER STEEL, 0.4% CARBON

ANALYST*	C	Mn	P	S	Si											
	Direct combustion	Bismuthate (FeSO ₄ -KMnO ₄)	Persulfate-Arsenite	Gravimetric (weighed as Mg ₂ P ₂ O ₇ after removal of arsenic)	Alkali-Molybdate ^a	Gravimetric (direct oxidation and final precipitation after reduction of iron)	Evolution with HCl (1:1) ZnS-Iodine (theoretical sulfur titer) ^b	Combustion	Sulfuric acid dehydration	COPPER H ₂ S-CuS-CuO	NICKEL Weighed as nickel dimethylglyoxime	CHROMIUM FeSO ₄ -KMnO ₄ titration	VANADIUM	MOLYBDENUM Colorimetric	ARSENIC	NITROGEN
1.	0.406	0.632	0.635	0.081	0.082	0.046	0.045		^d 0.068	0.035	0.020	0.024	^e 0.004	0.009	^f 0.004	^g 0.009
2.	.403	.635	.082	.081	.046	.045			.067	.034	.021	.023	ⁱ .005	.011	^k .006	
3.	.404	.631	.085	.084					^l 0.065	^m .030	.020	ⁿ .025	^t .004	.009		
4.	.407	.636	.083	.085	.048	.049			^p d. 066	^q .034	.021	^r .023	^s .004	.009	^u .004	^t .010
5.	.410	.635		.085			^u l. 050	^v .049	^w d. 065	^m .032	.020	^x .023	^y .003	.012		
6.	.406	.631		.081	.046		^v .047	^l d. 073	.029	.018	^z .022	^z .002	.010	^z 1. 003		
Averages	0.406	0.634	0.634	0.083	0.083	0.047	0.047	0.048	0.067	0.032	0.020	0.023	0.004	0.010	0.004	0.010
General average	0.406	0.634		0.083			0.047		0.067	0.032	0.020	0.023	0.004	0.010	0.004	0.010

* Precipitated at 40° C washed with a 1-percent solution of KNO_3 and titrated with alkali standardized by the use of National Bureau of Standards acid potassium phthalate and the ratio 23NaOH:1P.

^a Value obtained by standardizing the titrating solution by means of sodium oxalate through KMnO_4 and $\text{Na}_2\text{S}_2\text{O}_3$ and the use of the ratio 2I:1S.

^b Molybdenum-blue photometric method. See J. Research NBS 26, 405 (1941) RP1386.

^c Double dehydration.

^d Persulfate oxidation, potentiometric titration with ferrous ammonium sulfate solution.

^e Nitric acid oxidation, potentiometric titration with ferrous ammonium sulfate solution.

^f Determination made by K. D. Fleischer by the distillation-molybdenum-blue photometric method. See J. Research NBS 24, 7 (1940) RP1267.

^g Determination made by M. Marie Cron, by the vacuum-fusion method. See BS J. Research 7, 375 (1931) RP346.

^h Solution standardized by use of a standard steel.

ⁱ Ferrous sulfate-persulfate method.

^j Distillation-bromate titration method.

^k Perchloric acid dehydration.

^l Finished by electrolysis.

^m Perchloric acid-photometric method.

ⁿ Absorbed in cadmium chloride solution.

^o Nitric-sulfuric acid dehydration.

^p Finished by $\text{Na}_2\text{S}_2\text{O}_3$ titration.

^q Perchloric acid oxidation.

^h Distillation-As₂S₃ method.

ⁱ Solution in HCl (1:1). Distillation-titration method.

^u 5-g sample wrapped in filter paper and annealed for 1 hour at 700° C.

^v Sulfur gases absorbed in NaOH-H₂O₂ and excess NaOH titrated with H_2SO_4 .

^w Solution in $\text{HCl}-\text{HNO}_3$. Perchloric acid dehydration.

^x Cupferron-phosphotungstovanadate-photometric method.

^y Preliminary separation of chromium in a 10-g sample by precipitation with NaHCO_3 .

^z Vanadium separated from a 10-g sample as in footnote y, and determined by the H_2O_2 -colorimetric method.

¹ Distillation-As₂S₃-Ag₃AsO₄-KCNS titration method.

* LIST OF ANALYSTS

1. Ferrous Laboratory, National Bureau of Standards, Washington 25, D. C., John L. Hague in charge. Analysis by John P. Hewlett, Jr., James I. Shultz, Florence Yenchius, and Jewel Doran.
2. R. H. Rouse, Bethlehem Steel Corporation, Steelton, Pa.
3. Louis Singer and Alfred Weissler, Naval Research Laboratory, Washington, D. C.
4. S. S. Heide, W. A. Bass, and J. W. Lovin, Tennessee Coal, Iron & Railroad Co., Birmingham, Ala.
5. R. S. Gibbs and F. B. Clardy, Chemical Laboratory, Norfolk Navy Yard, Portsmouth, Va.
6. George Dreher, Rotary Electric Steel Co., Detroit, Mich.

The steel for the preparation of this standard was furnished by the Bethlehem Steel Co.

WASHINGTON, November 15, 1944.

LYMAN J. BRIGGS, Director.