U. S. DEPARTMENT OF COMMERCE NATIONAL BUREAU OF STANDARDS WASHINGTON 25, D. C.

PROVISIONAL CERTIFICATE STANDARD SAMPLE 1055 COBALT CYCLOHEXANEBUTYRATE

(Standard for Determination of Cobalt in Petroleum Products)

This compound was prepared to insure material that is essentially free from other metals and has suitable solubility, compatibility, and uniformity for use in the preparation of a standard of cobalt in lubricating oils. The compound is being certified provisionally to one part per hundred of cobalt, pending further studies of analytical procedures, of loss of weight on drying, and of changes in weight when exposed to air. Every effort should be made to maintain a uniform procedure, by following the directions in this certificate.

CHEMICAL AND SPECTROGRAPHIC ANALYSES Procedure and Results of Chemical Analysis

Cobalt, percent - - - - 17.1 Nickel, percent - - - - 0.07

Cobalt and nickel were determined by ignition of a 1-g sample (dried for two hours over phosphorus pentoxide), wrapped in filter paper and covered with oxalic acid. The oxide was dissolved in hydrochloric acid, electrodeposited as cobalt from ammoniacal solution, and weighed as the metals. The oxide was also ignited under hydrogen, and weighed as the metals. Nickel was separated from a hydrochloric acid solution of the metals by anion-exchange, precipitated with dimethylglyoxime, and weighed. Analysts, B. B. Bendigo and L. A. Machlan.

Procedure and Results of Spectrographic Analysis

The compound was examined spectrographically for metallic impurities. A 5-mg sample of the compound was excited in a direct-current arc and the photographed spectrum was examined for the characteristic lines of 51 elements. Several impurities were found, but, with the possible exception of nickel, none is considered to be present in sufficient concentration to interfere with the intended use. The principal impurities are nickel, determined chemically with the result given above; tin, estimated to be less than 0.1 percent; and aluminum, copper, iron, and manganese, each estimated to be less than 0.01 percent. Analyst, Elizabeth K. Hubbard.

STABILITY. - Tests show that standard lubricating-oil solutions of this compound with concentrations of cobalt up to 500 ppm are stable for several weeks when prepared by the directions given below.

COMPATIBILITY. - Lubricating-oil solutions of this compound have been found to be compatible with lubricating-oil solutions of the other compounds in this series. Blends of several different compounds have been prepared by the procedures given in the provisional certificates for the other compounds. (Tests have not been carried out to insure compatibility with the various additives that may be in the oils to be analyzed.)

DIRECTIONS FOR PREPARING LUBRICATING-OIL SOLUTIONS OF COBALT CYCLOHEXANEBUTYRATE

Transfer approximately 0.3 g of this compound from the bottle to a small beaker and dry over phosphorus pentoxide in a desiccator for two hours. (Tightly close the bottle containing the remainder of the compound.) Quickly and accurately transfer 0.292 g of this dried salt to a weighed 200-ml flask. (This weight of salt is equivalent to 50 mg of cobalt.) Add 2 ml of xylene and 4 ml of 2-ethylhexylamine and heat the flask on a hot plate, with swirling and without charring, until a clear solution forms. Add to the hot solution 2 ml of 2-ethylhexanoic acid and 80 to 90 ml of lubricating oil and gently shake the flask to mix the contents. Allow the flask to cool to room temperature and add enough lubricating oil to bring the total weight of the contents of the flask to 100 ± 0.5 g. Stopper the flask and shake gently to insure a homogeneous solution. The concentration of cobalt in this solution is 500 ppm.

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