

DEPARTMENT OF COMMERCE

Bureau of Standards
Certificate of Analyses

OF

STANDARD SAMPLE No. 101

18 CHROMIUM-8 NICKEL STEEL

ANALYSTS	C			Mn	P	S	Si	Ni	Cr			NITROGEN $\text{N}_{\text{in}} = 0.005\%$						
	CARBON 1. Direct combustion $725^{\circ}-740^{\circ}\text{ C}.$	2. Direct combustion at $1,100^{\circ}-1,200^{\circ}\text{ C. with}$ red lead as accelerator	3. Direct combustion at $1,100^{\circ}-1,200^{\circ}\text{ C. with}$ tin as accelerator	MANGANESE 1. Bismuthate ($\text{FeSO}_4 \cdot \text{KMnO}_4$)	PHOSPHORUS 1. Alkali-molybdate ^a	2. Gravimetric (weighed as MgPO_4 after re- moval of arsenic)	SULPHUR 1. Granimetric (direct ox- idation and precipi- tation in reduced solution)	2. Evolution with HCl ($\text{I}-\text{I}$), ZnS -iodine (theoretical sulphur nitrate ^b)	SILICON 1. Sulphuric acid dehy- dration	COPPER 1. $\text{H}_2\text{S-CuS-CuO}$	NICKEL Weighed as nickel di- methylglyoxime	CHROMIUM $\text{FeSO}_4 \cdot \text{KMnO}_4$ titration	VANADIUM	MOLYBDENUM				
	1	0.064	0.061	0.059	0.555 ^c	0.010 ^d	0.010	0.010 ^e	0.012	0.766 ^f	0.054	8.44	17.53 ^g	0.045 ^h	0.005	0.028		
2		.061		.57 ^b	.008 ⁱ		.014 ⁱ		.760	.062 ^j	8.45 ^j	17.66						
3		.060		.559 ^k	.010	.010	.012 ^j	.011 ^{m,n}	.757 ^f	.053 ^o	8.49 ^o	17.52						
4		.060	.058	.555 ^p	.012 ^q			.015 ⁿ	.75		8.40	17.50 ^g						
5			.059	.561	.011 ^d			.012 ^m	.76 ^f	.058 ^o	8.43	17.61	.047 ^r	.005				
6		.064	.063	.548 ^s	.013 ⁱ		.013 ^j	.015 ⁿ	.771 ^f	.050 ^t	8.43	17.56						
				.059	.56 ^s	.012 ^d		.012	.76	.06	8.48 ^u	17.60						
8			.058	.056	.559 ^c	.010 ⁱ			.015	.778		8.48	17.57 ^g					
9				.059 ^u	.548 ^g	.012	.011	.015 ⁱ	.014 ⁿ	.75 ^f	.051 ^o	8.44	17.56 ^g	.041 ^r	.006			
10					.552 ^s	.013		.012	.011	.777 ^f		8.44	17.60					
11					.545 ^p	.010				.765 ^v			17.54					
12					.548 ^s	.012		.015	.760 ^f	.052 ^o	8.47	17.62						
13					.55	.013 ^c			.015	.757	.060	8.43	17.52		.006			
14					.561 ^w	.011		.012		.771	.052 ^o	8.39	17.51					
15					.064	.062	.557 ^x	.013 ^s		.013	.766 ^f		8.42	17.55				
Averages					.063 ^z	.061	.059	.555	.011	.010	.013	.763	.055	8.44	17.56	.044	.006	.028
General Averages												.763	.055	8.44	17.56	.044	.006	.028

^a Precipitated at 40°C., washed with a 1 percent solution of KNO_3 and titrated with alkali standardized by use of Bureau of Standards standard acid potassium phthalate and the 23:1 ratio.

^b Value obtained by standardization of titrating solution against sodium oxalate through KMnO_4 and $\text{Na}_2\text{S}_2\text{O}_8$.

^c Chromium separated by precipitating with NaHCO_3 before the bismuthate oxidation.

^d Solution in mixture of equal volumes of HCl and HNO_3 , silica separated by dehydration with HClO_4 before precipitating with molybdate.

^e Meissner's method.

^f Perchloric acid dehydration.

^g Potentiometric titration.

^h Chromium separated by precipitating with ZnO before oxidation of the manganese with lead peroxide.

ⁱ Solution in a mixture of equal volumes of HCl and HNO_3 .

^j KCN titration.

^k The bulk of the iron removed by extraction with ether, chromium removed by a persulphate-NaOH separation, and the manganese determined by the bismuthate method.

^l Solution in a mixture of 2 parts HNO_3 ; 1 part HCl .

^m Dissolved in concentrated HCl .

ⁿ Absorbed in CdCl_2 solution.

^o Finished by electrolysis.

^p Volhard-arsenite method.

^q Titration solution standardized by means of standard steel.

^r Iron and vanadium precipitated with cupferron. Precipitate ignited and vanadium determined by the persulphate method.

^s ZnO separation.

^t Colorimetric.

^u Burned with low carbon (0.05C) steel.

^v HCl dehydration.

^w Arsenite titration.

^x ZnO separation, persulphate oxidation, and potentiometric titration with sodium arsenite.

^y Sample dissolved in nitric and hydrofluoric acids.

^z Recommended value.

- 1. Ferrous Laboratory, Bureau of Standards, H. A. Bright in charge; analysis by R. M. Fowler and J. C. Redmond.
- 2. C. M. Johnson, The Crucible Steel Co. of America, Pittsburgh, Pa.
- Thomas Fudge, The International Nickel Co. (Inc.), Bayonne, N. J.
- 4. C. H. McCollam, The Timken Steel & Tube Co., Canton, Ohio.
- 5. R. J. Price, The Electro Metallurgical Co., New York, N.Y.
- 6. E. B. Welch, Firth Sterling Steel Co., McKeesport, Pa.
- 7. H. N. Austin, The Babcock & Wilcox Tube Co., Beaver Falls, Pa.

- 8. P. L. Tyson, The Carpenter Steel Co., Reading, Pa.
- 9. J. A. Wiley, The Midvale Co., Philadelphia, Pa.
- 10. R. G. Schaeffer, General Motors Corporation, Research Laboratory, Detroit, Mich.
- 11. F. M. Portz, Republic Steel Corporation, Massillon, Ohio.
- 12. C. K. Mitchell, W. B. Coleman & Co., Philadelphia, Pa.
- 13. W. T. Hartley, Ludlum Steel Co., Dunkirk, N.Y.
- 14. H. H. Craver, Pittsburgh Testing Laboratory, Pittsburgh, Pa.
- 15. C. P. Larrabee, The American Sheet and Tin Plate Co., Pittsburgh, Pa.

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