

# Amphipathic Oral Chelators and Radionuclide Contamination

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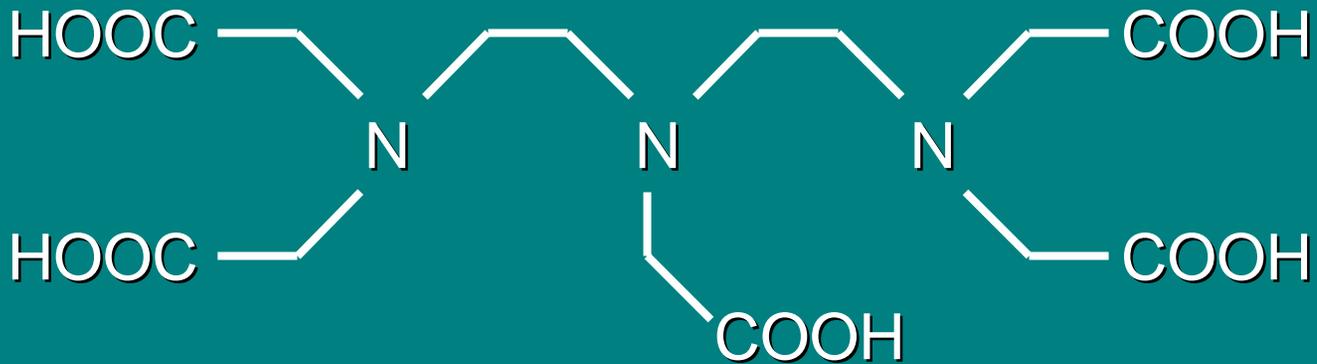
# Overview

- Chelation objectives and chelator design
- Possible advantages/disadvantages
- Excretion pathways
- Detection methods
- Binding affinities
- Efficacy: Pu, Am, U, Co, others
- Safety
- Ongoing studies:

# Chelator Design

- Orally active
- Polyaminocarboxylic acid family
- Effective for actinides (esp. Pu and Am)
- Broad efficacy for other metals
- Suitable for long-term use
- Low toxicity

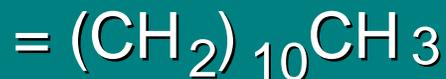
# DTPA



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# Amphipathic Triethylenetriaminepentaacetic (TT) Chelators



# Amphipathic Polyaminocarboxylic Acid Chelators

## Possible advantages?

- Orally available
- Better organ/tissue availability?
- Target organs?
- Select excretion pathways
- Different coordination than DTPA
- Safety
- Efficacy for Pu and Am

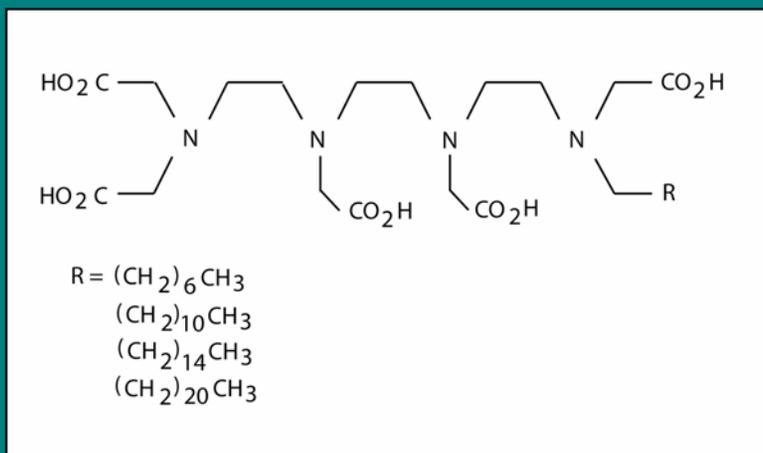
## Possible disadvantages?

- Bind in tissues?
- Amphipathic - toxicity?
- Translocation?
- Slower acting?

Polyaminocarboxylic acids can bind a broad range of metals and radionuclides

**Disadvantages:** When exposure is from one well characterized nuclide.

**Advantages:** When exposure is to multiple metals/nuclides, or composition not well known



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# Amphipathic Properties and Excretion Pathways

Biliary Excretion

Urinary Excretion

Liver



Bile



Feces

More Lipophilic  
[C<sub>22</sub>TT]

Less Lipophilic  
[C<sub>4</sub>TT]

Blood



Kidney



Urine



# Detection Methodologies Radionuclides

**Plutonium-239**



Scintillation Counting  
Alpha Spectroscopy  
Neutron-induced  
autoradiography (NIAR)  
Fission track analysis (FTA)

**Americium-241**



Alpha Spectroscopy  
Gamma Spectroscopy

# Detection Methodologies Non-Radioactive Elements

Uranium  
Cerium  
Cobalt  
Strontium  
Lead  
Iron  
Others...



Inductive Coupled Plasma  
Mass Spectroscopy (ICP-MS)

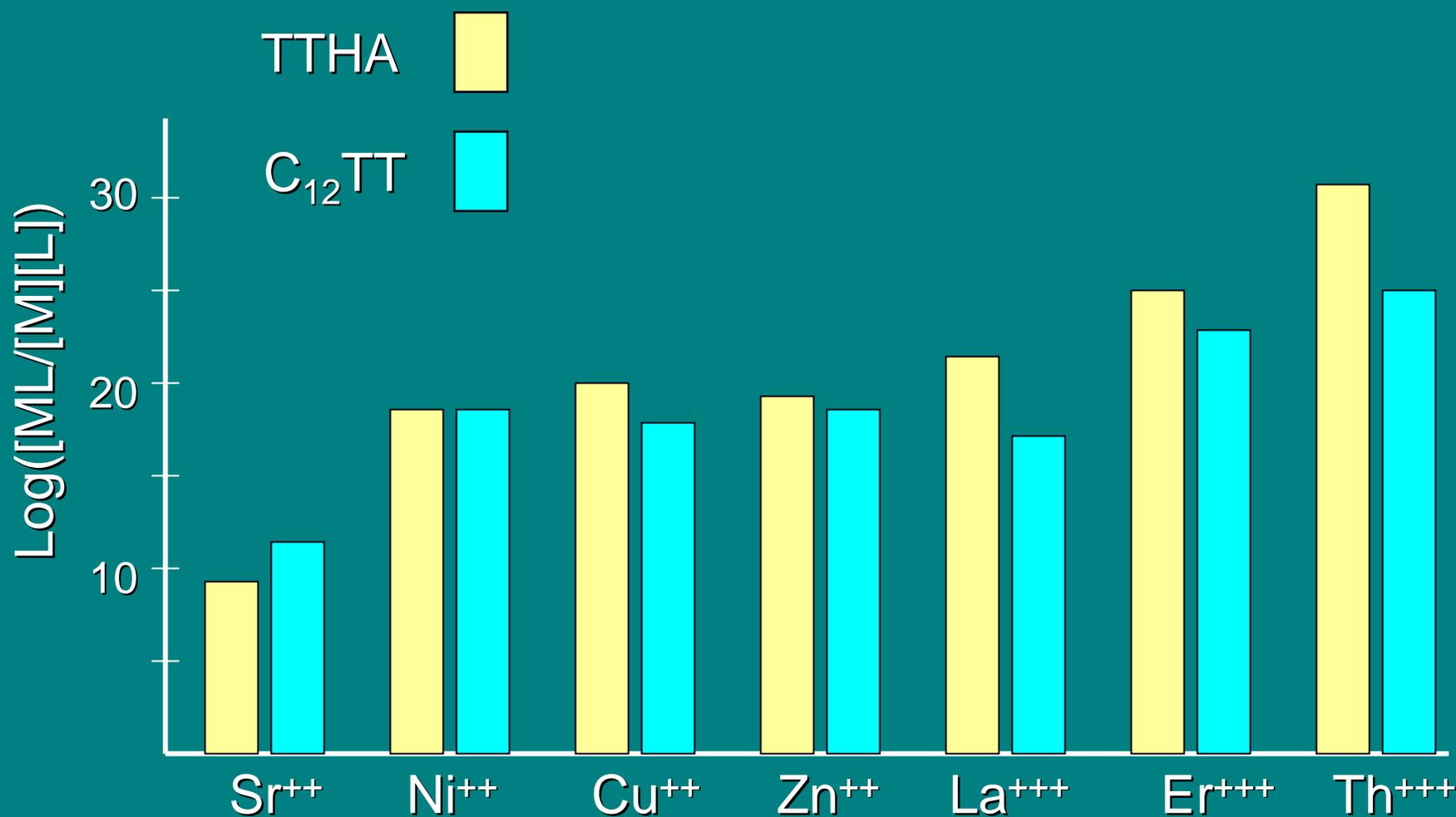
Other methods: Electron Paramagnetic Resonance (EPR)  
Atomic Absorption Spectroscopy (AAS)

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# Log Stability Constants



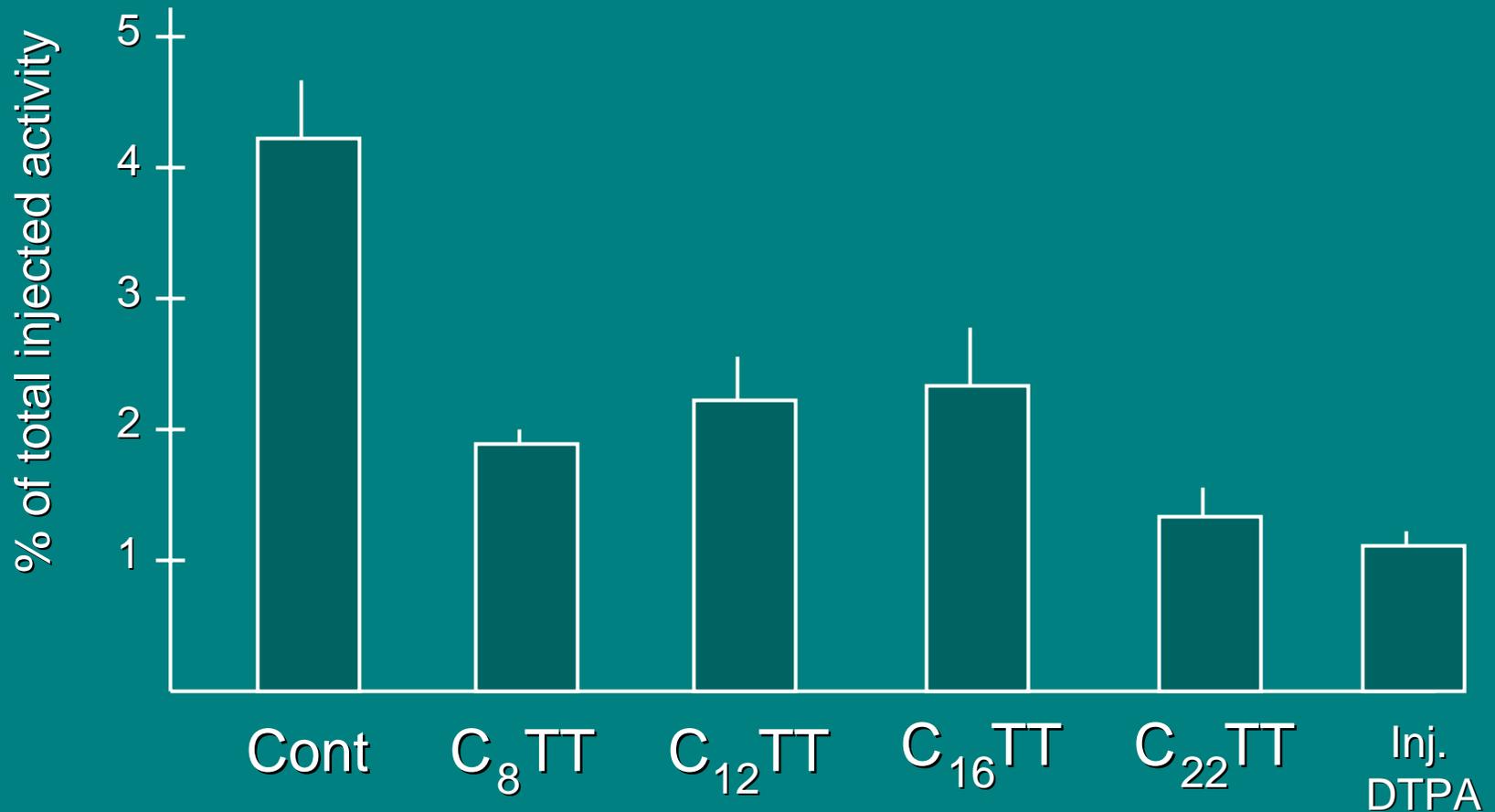
# Typical Animal Protocol

## Actinides

- Expose rat/mouse to metal or nuclide (systemic)
- Wait: 0-14 days
- Begin oral treatment with  $C_xTT$   
Intubation or add to food
- In-life measures: wt, food, water, feces, urine
- Organs: Bone, liver, kidney, spleen, other soft tissues

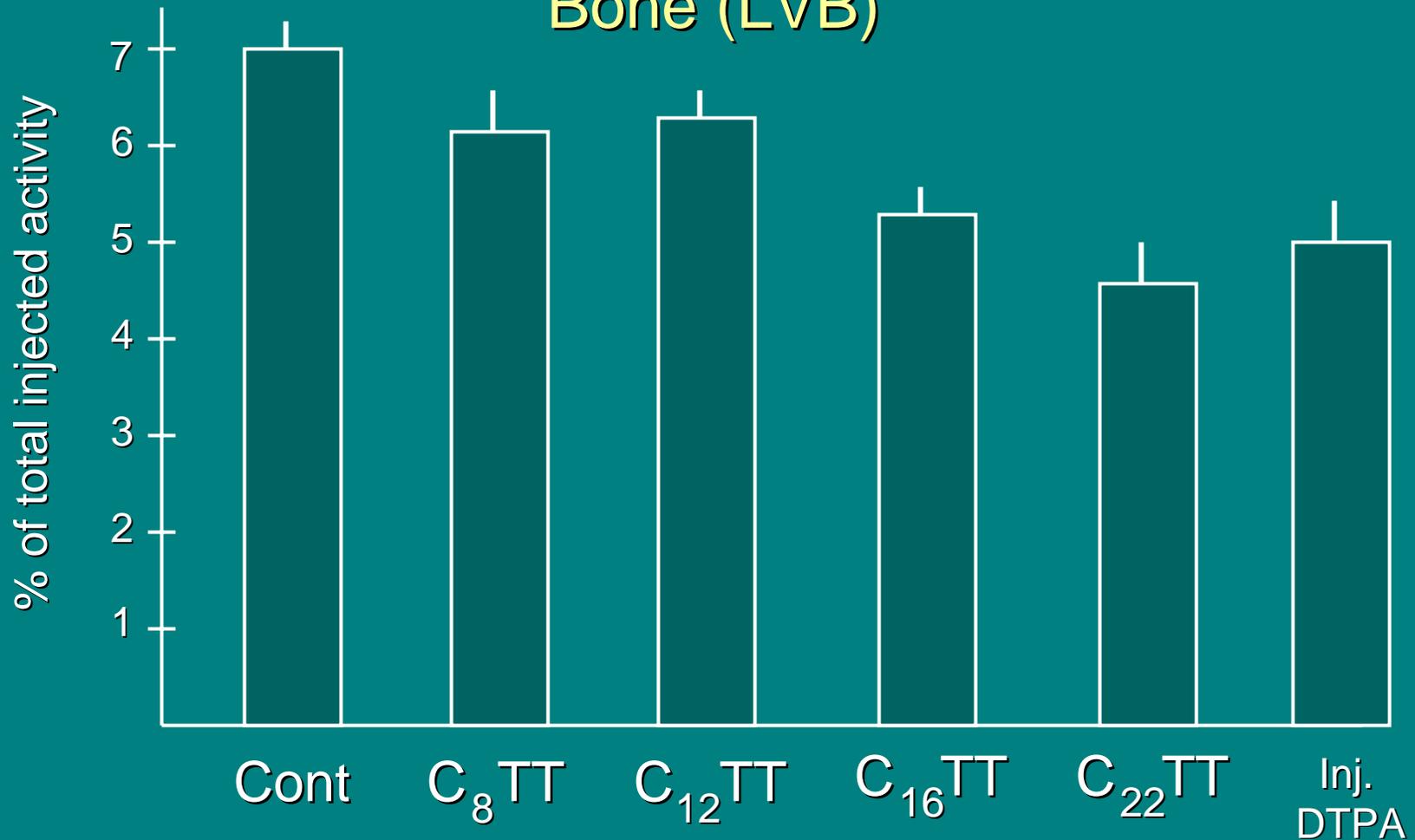
# C<sub>x</sub>TT Chelation of Pu<sup>239</sup>

14 d Pu-239 - 30 d C<sub>x</sub>TT with diet  
Liver



# C<sub>x</sub>TT Decorporation of Pu<sup>239</sup>

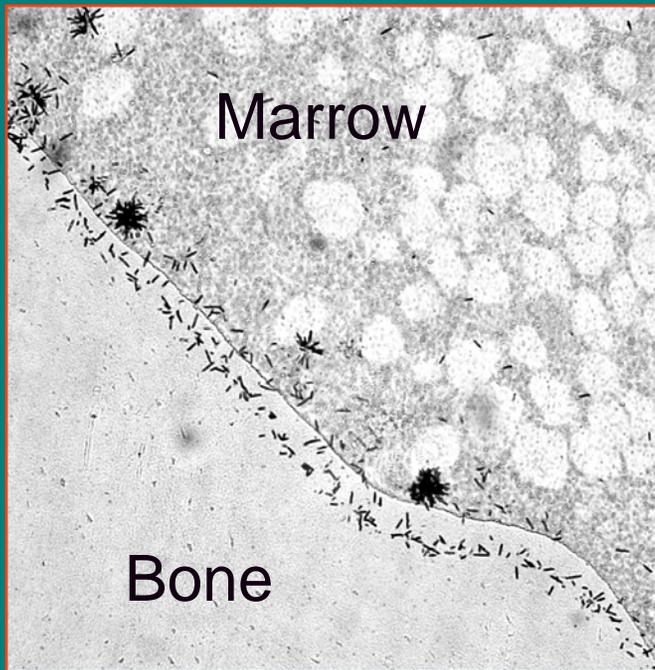
14 d Pu<sup>239</sup>; 30 d C<sub>x</sub>TT with diet  
Bone (LVB)



# Plutonium Deposition in Bone

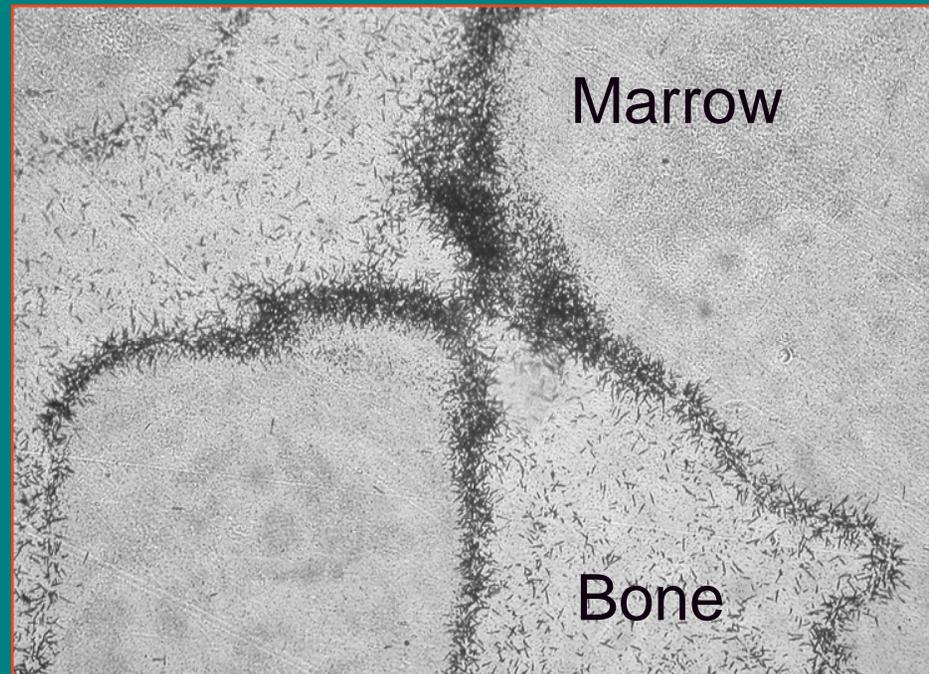
## Neutron-Induced AutoRadiography (NIAR)

Dog



2 Mo after injection

Human



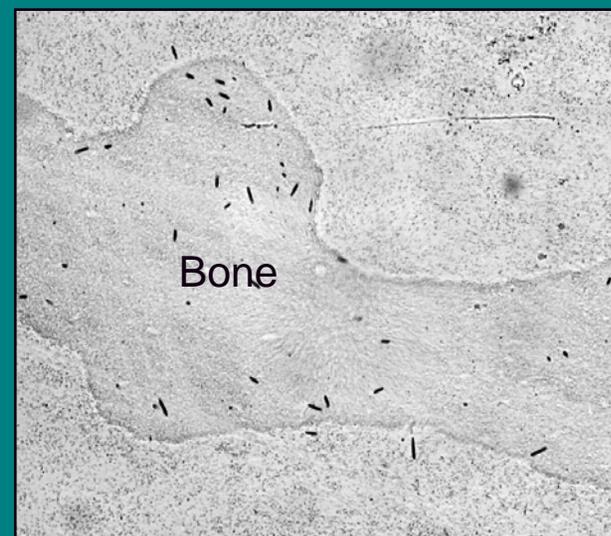
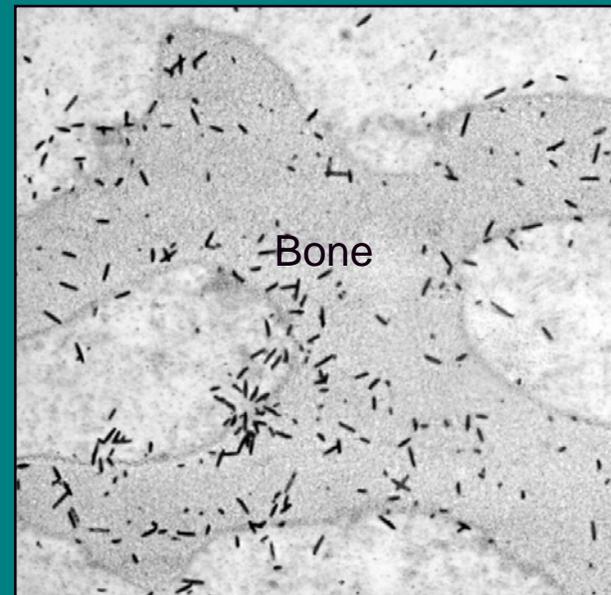
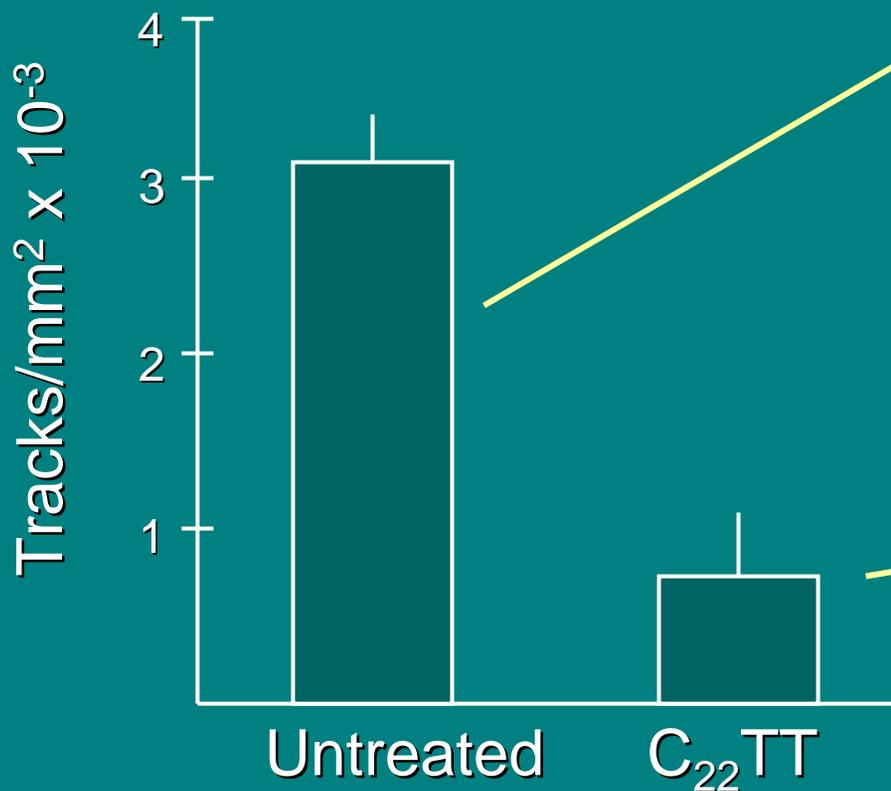
About 40 years of industrial exposure;  
With late in life liver disease

# C<sub>22</sub>TT on Redeposition of Pu<sup>239</sup> in Bone

90 d chelation; rat humerus metaphysis

Neutron flux:  $8 \times 10^{12} \text{ cm}^{-2} \text{ s}^{-1}$

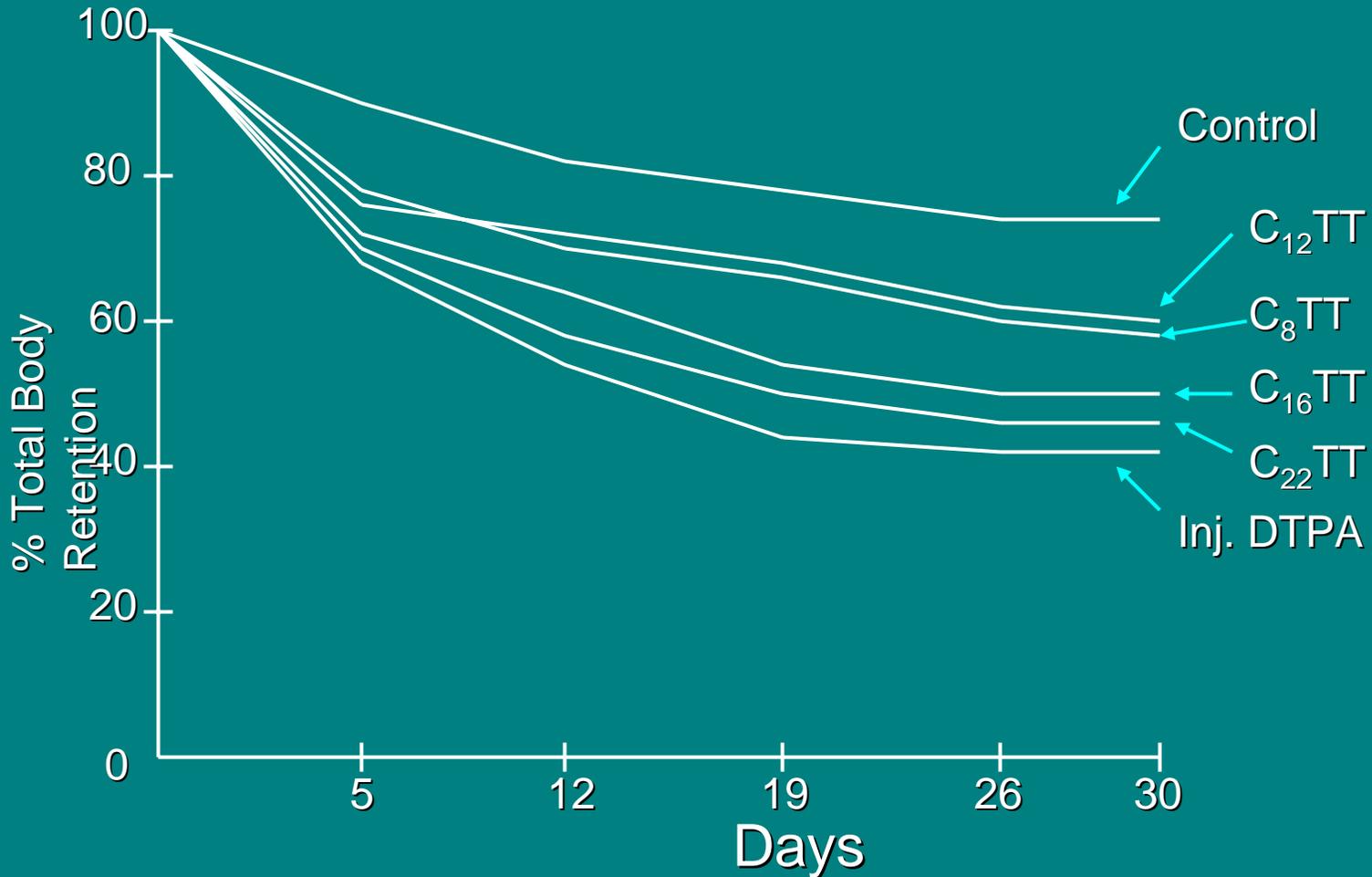
Neutron fluence:  $4 \times 10^{15}$



# CxTT Chelation Am<sup>241</sup>

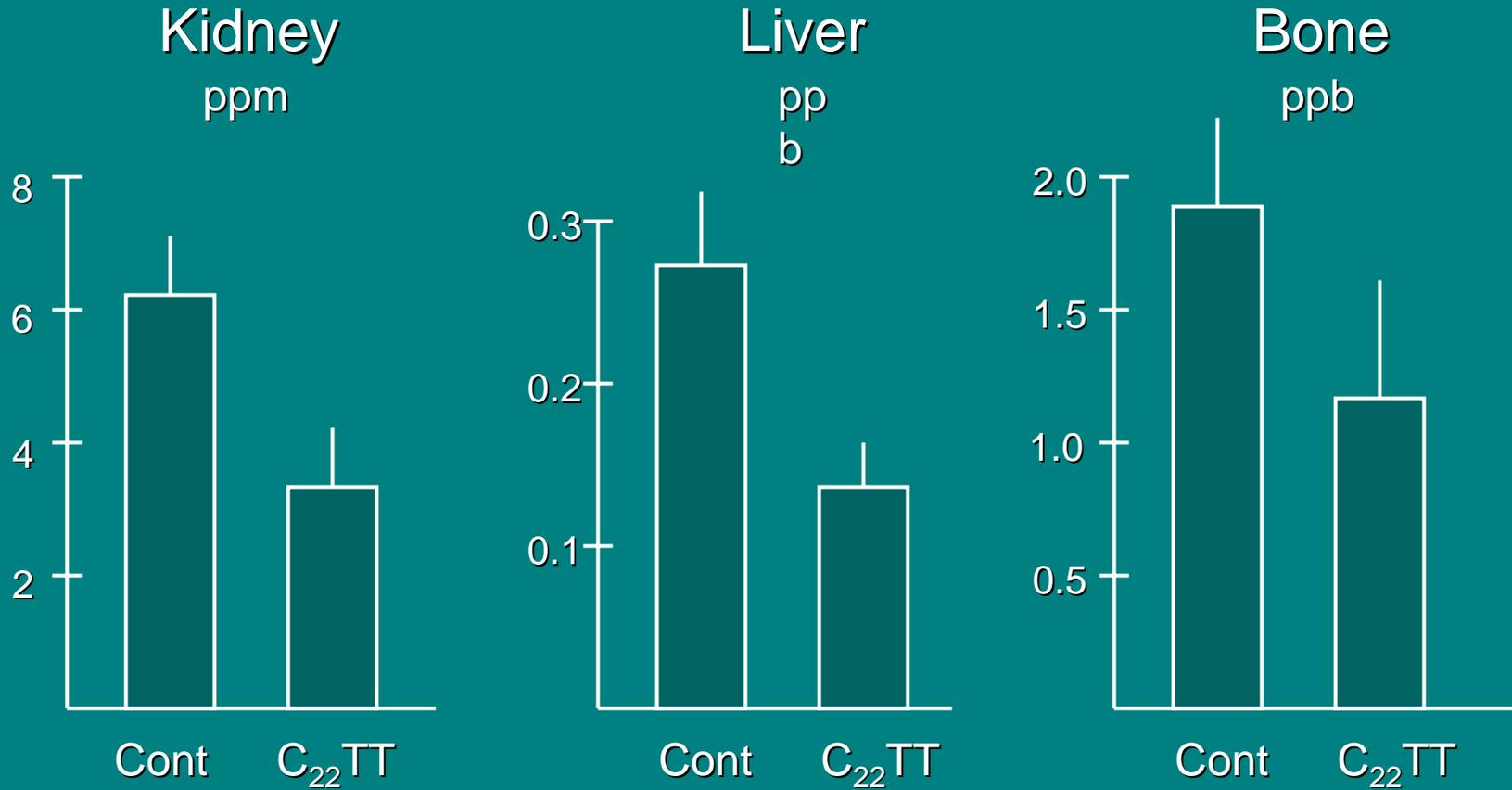
14 d Am-241 - 30 d CxTT with diet

Am-241 by Whole Body Gamma Spectroscopy



# Uranium - C<sub>22</sub>TT

10 d study: C<sub>22</sub>TT added to diet



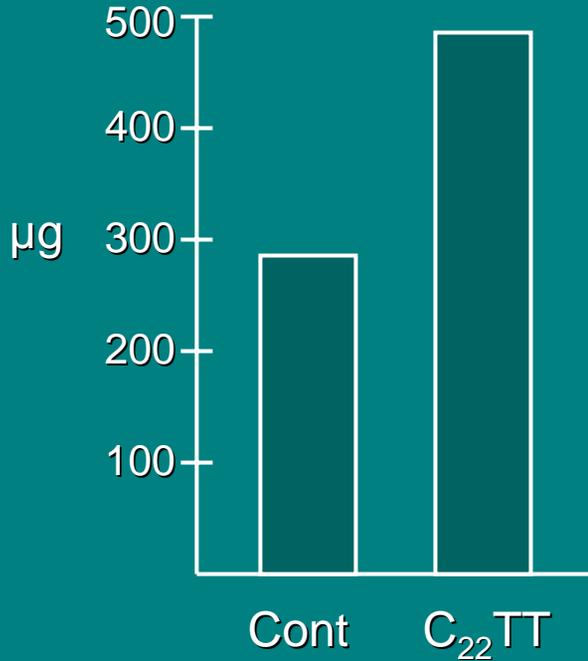
Preliminary data based on 3 rats per group

# Cobalt

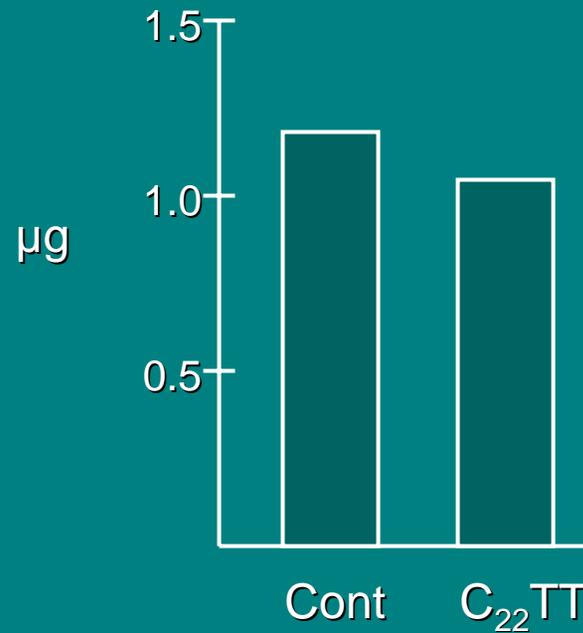
10 d, C<sub>22</sub>TT by gavage, Co by ICP-MS

Total 24 hr excretion

After 1st C<sub>22</sub>TT



Kidney Content



Preliminary data based on 3 rats per group

# Safety Issues with $C_xTT$

- Most experience is with  $C_{22}TT$
- No formal toxicity testing
- Observations to date:
  - Well tolerated, especially when added to diet
  - Food, water consumption normal
  - Zn and Ca salts not yet tested
  - May expect trace metal depletion
- Other effects:
  - Increased bone mass in aged rodent models
  - Better coat and appearance

# In Progress

- Bioavailability and PK
- Binding affinities - chain length
- More in vivo efficacy (U, Co, Ce, Sr)
- Bone/soft tissue Pu<sup>239</sup> distribution with chelation

# Summary

## Amphipathic polyaminocarboxylic acid chelators

- Orally active
  - Binding affinities are suitable for a number of
  - metals/nuclides
  - Amphipathic properties may be altered
- In vivo efficacy (oral administration) demonstrated for
- Pu, Am, U, Co, Pb, Fe (Studies in progress on Ce, Sr)
  - Zn, Ca salts not yet tested in animals
  - Relatively easy synthesis
  - Stable
- Well tolerated
- Some beneficial properties in aged animal models

# Research Team

Scott Miller, Ph.D.: PI: Actinide radiobiology

Gang Liu, Ph.D.: Metallo-organic chemistry

Melinda Krahenbuhl, Ph.D.: Nuclear engineering and  
Radiochemistry

Steven Kern, Ph.D.: Pharmacokinetics and preclinical  
development

William Johnson, Ph.D.: ICP-MS

Beth Bowman, M.S.: Animal services, lab management

\*Ray Lloyd, Ph.D.: Actinide radiobiology

\*\*Fred Bruenger, M.S.: Chelator  
chemistry, radiochemistry

\*Retired: July, 2007; now a consultant.

\*\*Deceased, April 2007