

Nitrogen Transformation Inhibitors and Controlled Release Urea

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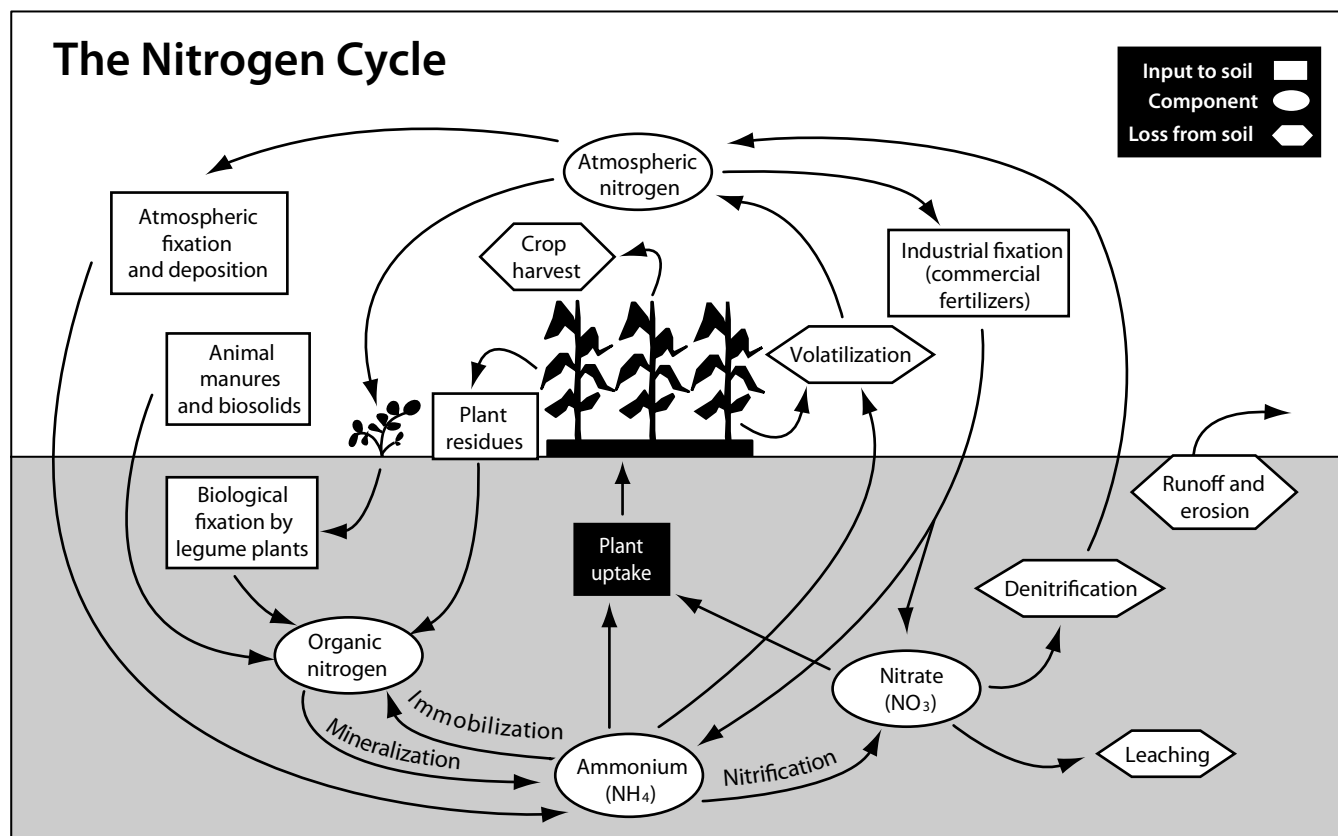
The soaring cost of fossil fuels is an indicator that nitrogen (N) fertilizer prices are going to remain high for the foreseeable future. With higher N prices, many producers are trying to evaluate the usefulness of several N additive products in their production systems. High N prices make these products more attractive because it takes fewer pounds of saved N to offset the price of the additive. Currently, there are three types of products being marketed that claim to improve nitrogen use efficiency: nitrification inhibitors, urease inhibitors, and controlled release fertilizer products. These products work by slowing one of the processes within the nitrogen cycle, thereby reducing N loss. Prior to purchase, producers should have a good understanding of how these products work in order to make informed decisions regarding their use.

Nitrification Inhibitors

Nitrification is the conversion of ammonium nitrogen ($\text{NH}_4\text{-N}$) to nitrate nitrogen ($\text{NO}_3\text{-N}$) in the soil (Figure 1). Depending on soil conditions, some inhibitors can slow this process by a few weeks. The most common, nitrapyrin (N-Serve®), has been commercially available for 30 years. It can be used with any N fertilizer that contains or produces (when applied to the soil) $\text{NH}_4\text{-N}$. Examples are anhydrous ammonia, urea, and urea-ammonium nitrate (UAN) solutions.

Inhibiting nitrification is important because nitrogen in the $\text{NH}_4\text{-N}$ form is held tightly by the soil particles and is not subject to leaching or denitrification loss. *Leaching* happens when $\text{NO}_3\text{-N}$ is moved deeper into the soil profile by moving water.

Figure 1. The nitrogen cycle showing components, inputs, losses, and transformations to soil nitrogen pools (adapted from a drawing provided by the Potash & Phosphate Institute and used with permission).



It is possible that soil $\text{NO}_3\text{-N}$ can be leached below the rooting zone and then become an environmental concern. *Denitrification* occurs when $\text{NO}_3\text{-N}$ is converted into a gas and escapes into the atmosphere. This reaction only happens when soil lacks oxygen or is largely water saturated. Depending on the amount of oxygen in the soil, the gas emitted is either in the nitrous oxide or nitrogen gas form. Nitrous oxide is considered a greenhouse gas, and emissions may be regulated in the future. Denitrification losses are most common on poorly drained soils saturated for many days during the spring.

Urease Inhibitors

When urea fertilizers are applied to the soil, an enzyme called urease begins their conversion to ammonia gas. If this conversion takes place below the soil surface, the ammonia is almost instantaneously converted to $\text{NH}_4\text{-N}$ which is bound to soil particles. If the conversion takes place on the soil surface or on surface residues, there is a potential for the ammonia gas to escape back into the atmosphere in a process called *ammonia volatilization*.

Volatilization losses depend on the environmental conditions at the time of application. Soil temperature, soil moisture, amount of surface residue, soil pH, and length of time between application and the first rain event or irrigation are all factors that determine the total amount of N that could be lost via volatilization. Nitrogen losses from fertilizer applied prior to May 1 are generally very low. After May 1, N loss is greatest, especially when urea is surface-applied to soils with high residue or vegetation (i.e., no-till corn or pastures), during warm, wet weather followed by a warm, breezy drying period.

Volatilization losses can be substantially reduced if a urease inhibitor is used with the fertilizer. The most common urease inhibitor is NBPT (N-[n-butyl] thiophosphoric triamide) sold under the trade name Agrotain[®]. Urease inhibitors reduce the activity of the urease enzyme for up to 14 days. As long as it rains during this 14-day period, the urea will be moved into the soil where it can be converted to $\text{NH}_4\text{-N}$ without the risk of volatilization.

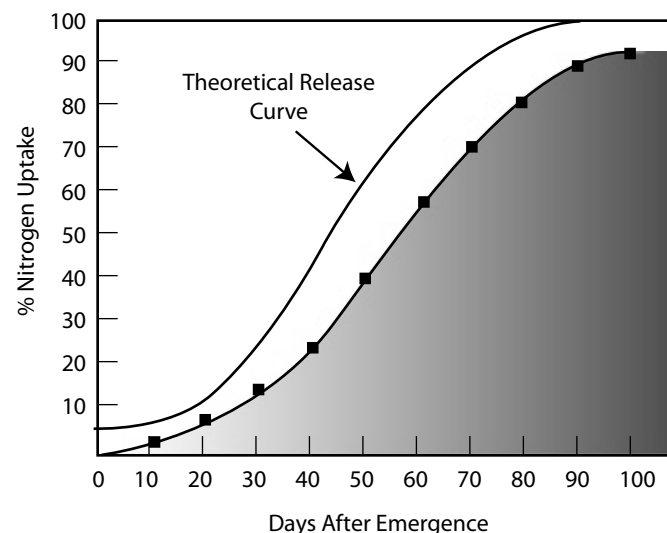
Controlled Release Urea

Controlled release fertilizer products have also been available for more than 30 years. Probably the best known of these products is sulfur-coated urea. A sulfur coating is applied to urea granules, and urea dissolves/diffuses through imperfections in the coating. By altering the thickness and number of imperfections in the coating, release characteristics can be controlled. Sulfur-coated urea was not a useful agronomic product in part because the cost of coating was high relative to the cost of the N fertilizer.

Recent advancements in polymer (plastic) technology have created a whole new type of controlled release fertilizer, the most common of which is polymer-coated urea (PCU). Polymer-coated urea has been used in the turf and horticultural industries for several years, but the cost of the materials prohibited their greater use in the agricultural market. Now Agrium Inc. has introduced a PCU called ESN[®] that is priced competitively in the agricultural market.

Modern polymers allow chemists to create release curves that closely match the uptake characteristics of target crops (Figure 2). The amount and rate of release is controlled by the thickness and other characteristics of the polymer

Figure 2. Typical nitrogen uptake curve for corn and a theoretical polymer-coated urea release curve.

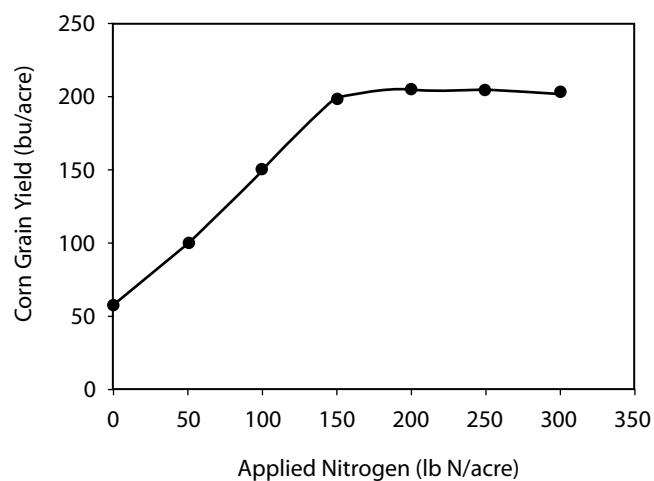


Agronomics of N Products

As mentioned earlier, there are several factors that should be considered before deciding whether these products are appropriate and economical in your specific production system.

Ideally, corn producers strive to apply the exact amount of N to reach maximum yield (example: 150 lb N/acre in Figure 3). If a farmer applied the optimal amount of fertilizer and used an inhibitor or PCU, N would be saved, but yield would not be increased. This is because maximum yield has already been obtained. In order for these new products to be agronomically useful, the producer must reduce the rate of applied nitrogen by the amount expected to be saved as a result of using the additive. To be economical, the cost of the saved N must exceed the price of the additive.

Figure 3. Typical nitrogen response curve for corn grown on well-drained soils in Kentucky.



The second consideration is the time of year fertilizer is being applied. Denitrification occurs primarily when the soil is water-saturated. Therefore, losses are usually highest for N applied early in the spring. Later side-dress applications usually result in very little denitrification loss since soil saturation is less likely. The total amount of nitrogen lost as a result of denitrification is a function of the number of days the soil remains saturated and the amount of nitrogen in the NO₃-N form (Table 1). Approximately 3 to 4% of the NO₃-N can be lost per day of soil saturation beyond two days.

Table 1. The percentage of fertilizer N in the NO₃-N form 0, 3, and 6 weeks after application.

N Source	Weeks After Application		
	0	3	6
	% of Fertilizer as NO ₃ -N		
Anhydrous ammonia (AA)	0	20	65
AA with N-Serve	0	10	50
Urea	0	50	75
Urea with N-Serve	0	30	70
UAN solutions	25	60	80
Ammonium nitrate	50	80	90

Volatilization losses are highest when the soil is warm (above 60°F), experiencing high evaporation rates, and/or when soil pH is greater than 7. In most years, temperatures become high enough to cause concern in early May. After this time, urea N contained in surface applications is more volatile. If the fertilizer is surface-applied and incorporated or if ¼ inch or more of rain is received within two days, volatilization losses will be minimal. If, on the other hand, it is not incorporated and no rain is received, loss can exceed 25% with an average of about 10% of the total. High surface residue levels also increase volatilization; therefore, maximum losses will be observed when urea is broadcast on no-till or pasture fields after May 1.

Polymer-coated urea, because of its slow release characteristics, offers farmers the option of early fertilizer application with a reduced risk of denitrification or leaching loss. There is still the potential for volatilization losses from this product because of its urea.

Inhibitor Research Results

Several research studies have been conducted across Kentucky to assess the performance of these products under our growing conditions. Wheat research in western Kentucky demonstrates that soils stay too warm for nitrification inhibitors to be effective, if all of the N fertilizer is applied in the fall (Table 2).

Table 2. Effect of time of application of urea and anhydrous ammonia N and the use of nitrification inhibitors on the yield of wheat in Kentucky (from Murdock 1985).

Nitrogen Treatment	Urea
	5-Yr Average
	Wheat Yield (bu/acre)
Check (no N)	31
Fall	46
Fall + Inhibitor	47
Spring	55

Nitrification inhibitors have, nevertheless, been shown to be effective for corn where N fertilizers were applied at planting at yield-limiting rates. In a study conducted in Bath, Lewis, and Lee counties, corn yield was increased by an average of 32 bu/acre when nitrapyrin (nitrification inhibitor) was applied with 75 lb N/acre as ammonium nitrate at planting (Frye et al., 1981). However, this study showed no significant yield increase with nitrapyrin at the 150 lb N/acre rate. These results demonstrate the need to reduce N application rates in order to get the benefit of the inhibitor. In general, Kentucky research has found economic benefits from nitrification inhibitors are more likely on poorly drained soils that tend to remain wet during the spring.

The urease inhibitor NBPT has also been effective in some Kentucky cropping systems. A study was conducted in Fayette County to assess the response of NBPT with surface applications of urea and urea ammonium nitrate (UAN) solution for fescue and corn production. Both crops were grown at yield-limiting N rates (140 lb N/acre/yr for fescue and 75 to 100 lb N/acre/yr for corn). Fescue yield was 13% lower when urea was used without NBPT but only 3% lower when UAN was used without NBPT. For corn, NBPT increased average grain yield by 14 bu/acre with urea and 6 bu/acre with UAN (Frye et al., 1990).

A product called SuperU® (marketed by Agrotain International) is urea with both NBPT and a nitrification inhibitor (dicyandiamide DCD). It is designed to prevent losses associated with surface applications of urea as well as leaching and denitrification losses. A two-year study in Meade County compared urea, ammonium nitrate, and SuperU effects on no-till corn yield. In both years of the study (1997-98), N was applied shortly after planting, and early-season rainfall was excessive. In both years, at yield-limiting rates of N, grain yield from SuperU was significantly higher than the yield from urea. In 1998, at the low fertilization rate, grain yield using SuperU was significantly higher than the yield from ammonium nitrate (Table 3). Results indicate that volatilization was a factor in both years, but leaching/denitrification losses were only a factor in 1998.

Table 3. Effect of N source and rate on corn yield in 1997 and 1998 (Wells et al., 1999).

N (lb/a)	1997			1998		
	AN	Urea	SuperU®	AN	Urea	SuperU®
	----- Corn Yield (bu/acre) -----					
0	----- 105 -----			----- 82 -----		
60				128	118	151
80	136	116	139			
120				161	138	159
160		143	136			
180				153	141	158

In 2004, a Hardin County study compared the performance of several N products in no-till corn production. All plots received a pre-plant application of 50 lb N/acre. Side-dress applications of different N fertilizer products were made at the 6-leaf stage. Excessive rainfall was received between planting and the side-dress application, but the soil was not saturated after the side-dress

treatments were applied, suggesting little potential for denitrification or leaching. Therefore, plots with the combination of NBPT and DCD were not statistically different ($p = 0.10$) from plots with only NBPT (Table 4). Weather and soil conditions were very conducive to ammonia volatilization losses. Yield was substantially lower for the urea and UAN treatments when compared to the ammonium nitrate application. When NBPT was added to urea and UAN, yield was increased by 43 and 29 bu/acre, respectively (Table 4). At this point, it is not clear why NBPT was less effective when used with UAN. The results may be an anomaly since similar studies have shown NBPT to be equally effective with urea and UAN. Our study will be repeated in 2005 to determine if this trend continues. Nitrogen loss was extremely high because less than optimal rates of N were used, but this study illustrates the loss potential in a no-till environment.

Table 4. Yield for side-dressed no-till corn in Hardin County, 2004.

Treatment ¹	Yield (bu/acre)
Check ²	117 d ³
Urea	158 c
Urea + NBPT	201 b
SuperU (urea + NBPT + DCD)	201 b
UAN	150 c
UAN + NBPT	179 bc
UAN + NBPT + DCD	175 bc
AN	192 b
Poly-coated urea (at planting)	177 bc
AN (130 lb N/acre)	239 a

¹ N side-dress rate was 50 lb N/acre.

² Check received 50 lb N/acre at planting and 0 lb N/acre side-dress.

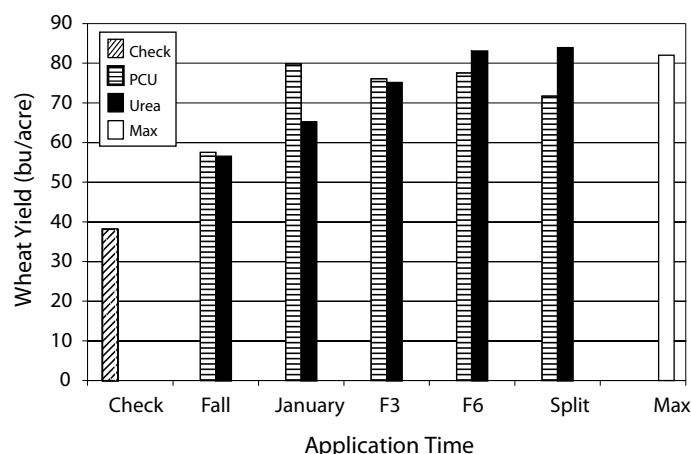
³ Numbers followed by the same letter are not significantly different.

Polymer-Coated Urea Research

The agriculturally priced polymer-coated urea (PCU) has only been available for the past two years. One potential use for this product is fall application on wheat, especially for wetter soils that might not support spreader trucks in the spring.

A wheat study begun in the fall of 2002 near Lexington evaluated PCU application timing on wheat. Treatments consisted of an unfertilized check, and 60 lb of N as urea or PCU broadcast at four application times (planting, January, Feekes Stage 3, and Feekes Stage 6). Also included were a split urea application of one-third at Feekes 3 and two-thirds at Feekes 6, and another receiving 20 lb urea-N/acre and 40 lb PCU-N/acre, all at Feekes 3. An additional treatment was included to determine the yield potential for the study site consisting of 30 lb urea-N/acre at Feekes 3 and 60 lb urea-N/acre at Feekes 6 (see Max at Figure 4). Like earlier studies that compared N with and without a ni-

Figure 4. The effect of N applied as polymer-coated urea (PCU) and urea on 2003-04 wheat grain yield in Lexington (LSD $p < 0.10 = 8$ bu/acre).



trification inhibitor, yields in this study were equivalent for PCU and urea applied in the fall (Figure 4). However, PCU applied in January produced significantly higher yields than urea applied in January. In fact, January PCU yield was not significantly different from that for the split application of urea. Therefore, part of the higher cost of PCU could be offset by one less fertilizer application. Applications of PCU later in the spring are not recommended because the slow release appears to reduce yield.

Studies have also been conducted in Lexington and Princeton to determine the effect of PCU on corn growth and yield. Study sites included both well-drained and somewhat poorly drained soils. For the well-drained soils, there was no yield benefit to using PCU. On the more poorly drained soils, PCU applied at or before planting increased yield significantly compared to urea applied at the same time. Polymer-coated urea was not, however, superior to a split application of urea (one-third at planting and two-thirds at the 6 to 8 leaf stage).

This latest wheat and corn PCU research has been conducted during the past two years, both considerably wetter than normal (especially early in the growing season). Additional research is needed to determine PCU effects in a more normal year.

Conclusions

There are several products that Kentucky farmers could use to help improve fertilizer nitrogen efficiency. These products are useful only in specific situations, so it is important to understand how they work and when they are most useful. It is also important to realize these products are designed to conserve nitrogen. Benefits will be realized only if the total N application rate is reduced by the amount of N estimated to be saved by using inhibitors or PCU.

References

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Trade Names

- Agrotain[®] is a registered trade name of Agrotain International.
- ESN[®] is a registered trade name of Agrium U.S. Inc.
- N-Serve[®] is a registered trade name of Dow AgroSciences.
- SuperU[®] is a registered trade name of Agrotain International.

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