

## Temporal and Spatial Patterns of Oxygen Depletion in Reservoirs

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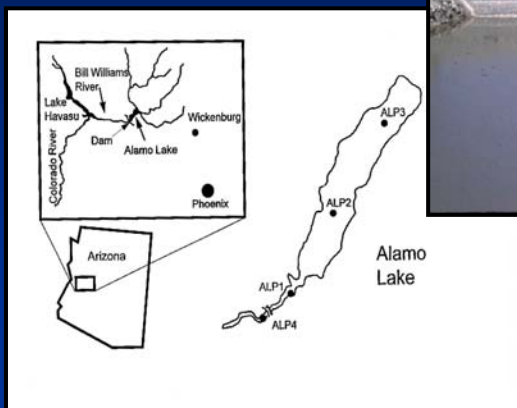
The production and consumption of dissolved oxygen results in concentration fluctuations at both temporal and spatial scales. Temporal and spatial development of density gradients, due primarily to changes in vertical temperatures in the water column, results in vertical and longitudinal zonation in aquatic systems. Extinction of light with depth results in the establishment of two primary vertical zones. In the upper, photic zone net oxygen production typically exceeds consumption. In a lower zone where photosynthesis is limited and oxygen production is exceeded. This results in net consumption of dissolved oxygen often followed by the establishment of hypoxic or anoxic conditions. The extent and the rate of oxygen depletion can be used to evaluate the “health” of the aquatic system and provide information critical for design guidance for remediation. Factors contributing to the establishment of these zones will be presented for selected reservoirs, assessment methods will be described, and implications for in-lake management techniques will be discussed.

# Temporal and Spatial Gradients In Dissolved Oxygen Concentrations Assessment Techniques

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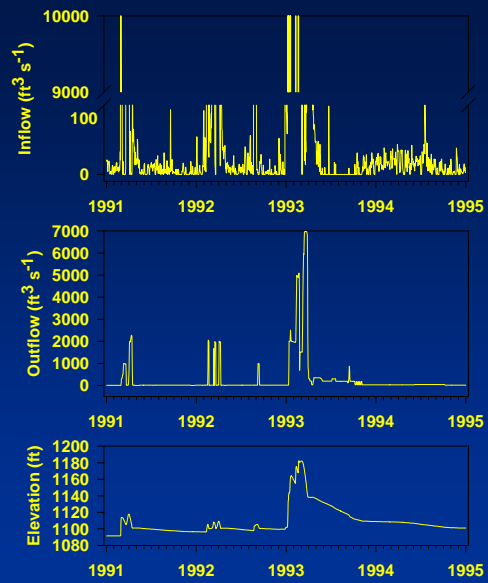
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## Alamo Lake, AZ Case Study: Flood Control Project in Arid Region



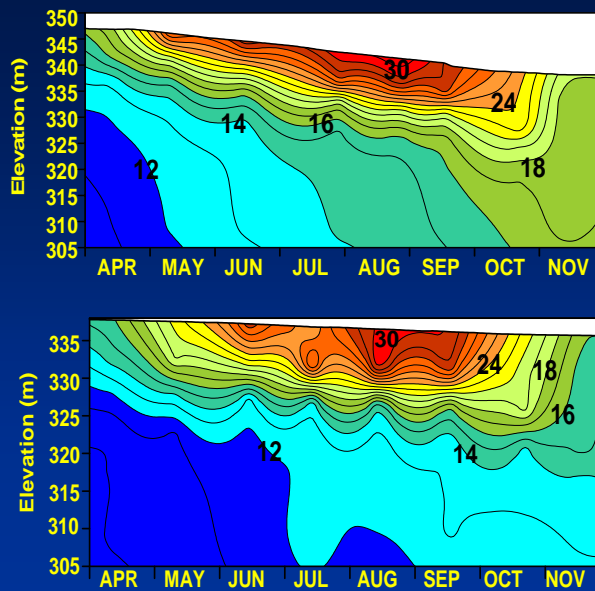
## Hydrologic Information

- 1993 was a very wet year with high surface elevation and discharge
- 1994 was a very dry year with low discharge



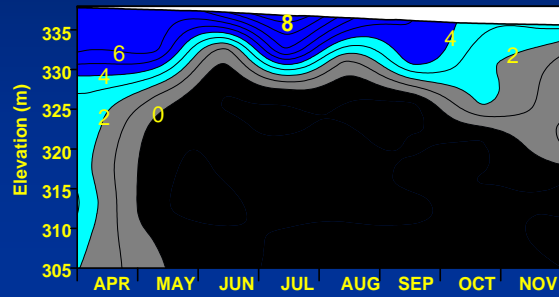
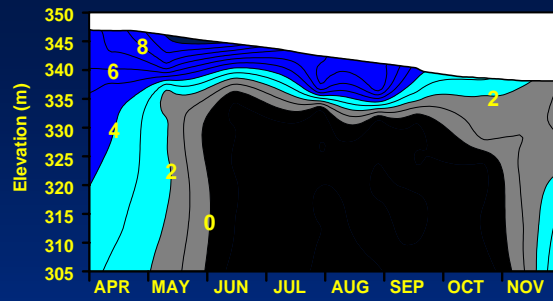
## Thermal Structure

- 1993 (top)
- 1994 (bottom)

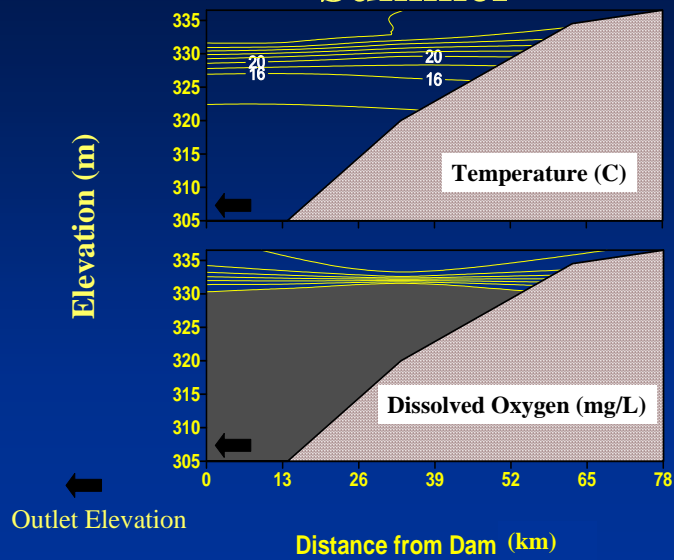


## Dissolved Oxygen Distribution

- 1993 (top)
- 1994 (bottom)



## Longitudinal Gradients - Late Summer



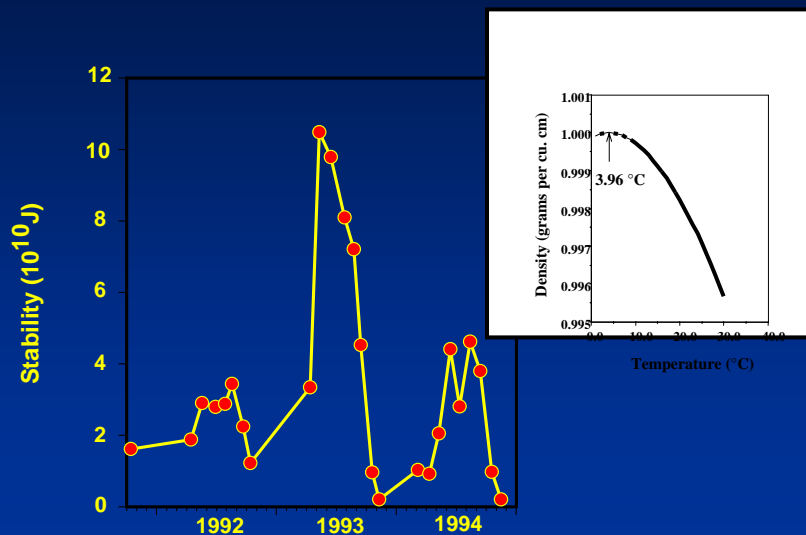
# Stability

$$\text{Stability (J)} = PEM - PES$$

$$PE = g \sum_{i=1}^n p_i V_i h_i$$

Davis, J.M. 1980. Water Services, 84, 497-504.  
Meyer, E.B. 1991. IEB, E-91-1, USAE, WES  
Hutchinson, G.E. 1957 and Birge, E.A. 1915

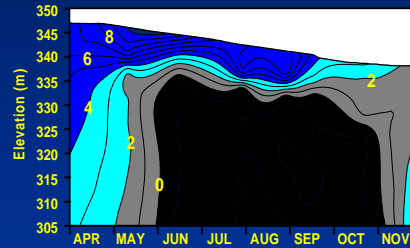
# Stability



# Dissolved Oxygen Dynamics

- Hypolimnetic Oxygen Deficit Rates  
Walker, W.W., 1987. TR EL-81-9, USAE

## PROFILE



- Oxygen Consumption Rates  
Dark Bottle Respiration  
Measurement

## PROFILE

**Areal or Volumetric Rates  
Valid for Oxidic Conditions**

**Requires Morphometry,  
Area/Volume Curves**

## Bottle Method

**Discrete Rate  
Applicable to Oxidic  
or Anoxic Conditions**

**Requires Multiple  
Sampling Depths**

# Dissolved Oxygen Consumption

## PROFILE

0.04 - 0.11 mg L<sup>-1</sup> day<sup>-1</sup>

## Oxygen Consumption Method

0.05 - 0.51 mg L<sup>-1</sup> day<sup>-1</sup>

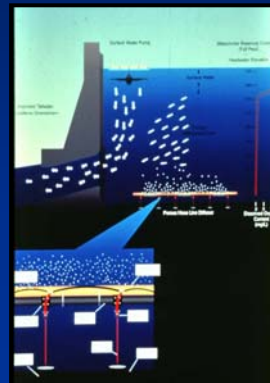
Recommended	0.15 mg L <sup>-1</sup> day <sup>-1</sup>
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# Recommendations

## Alternate Low Level Release



## Aeration System

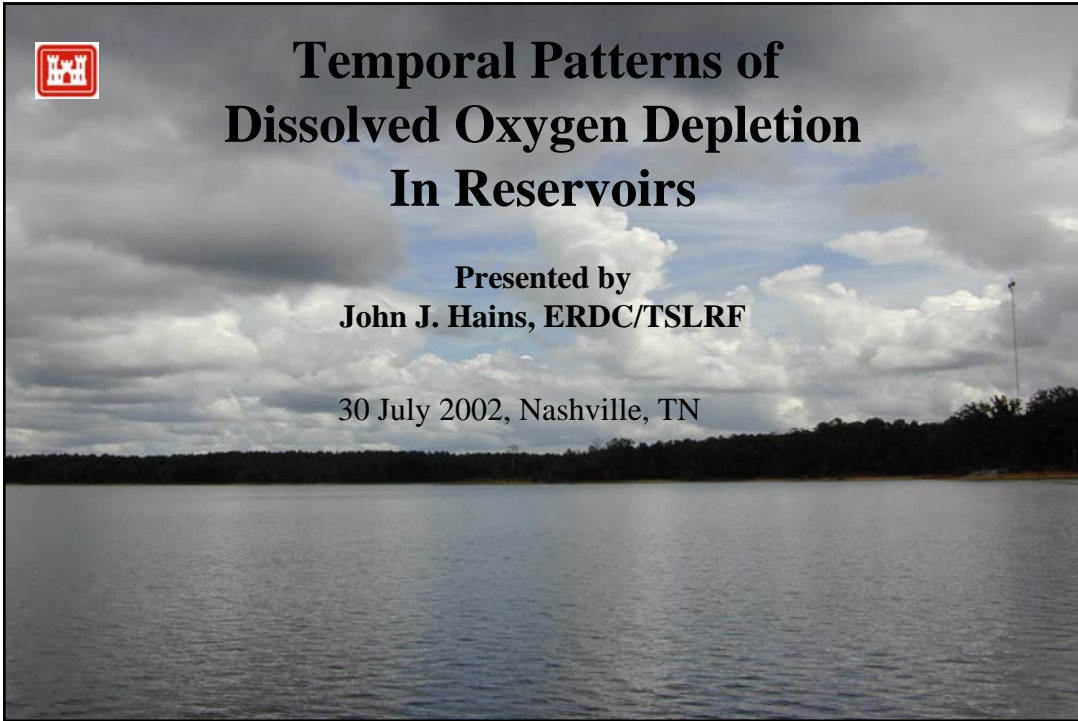




# Temporal Patterns of Dissolved Oxygen Depletion In Reservoirs

Presented by  
**John J. Hains, ERDC/TSLRF**

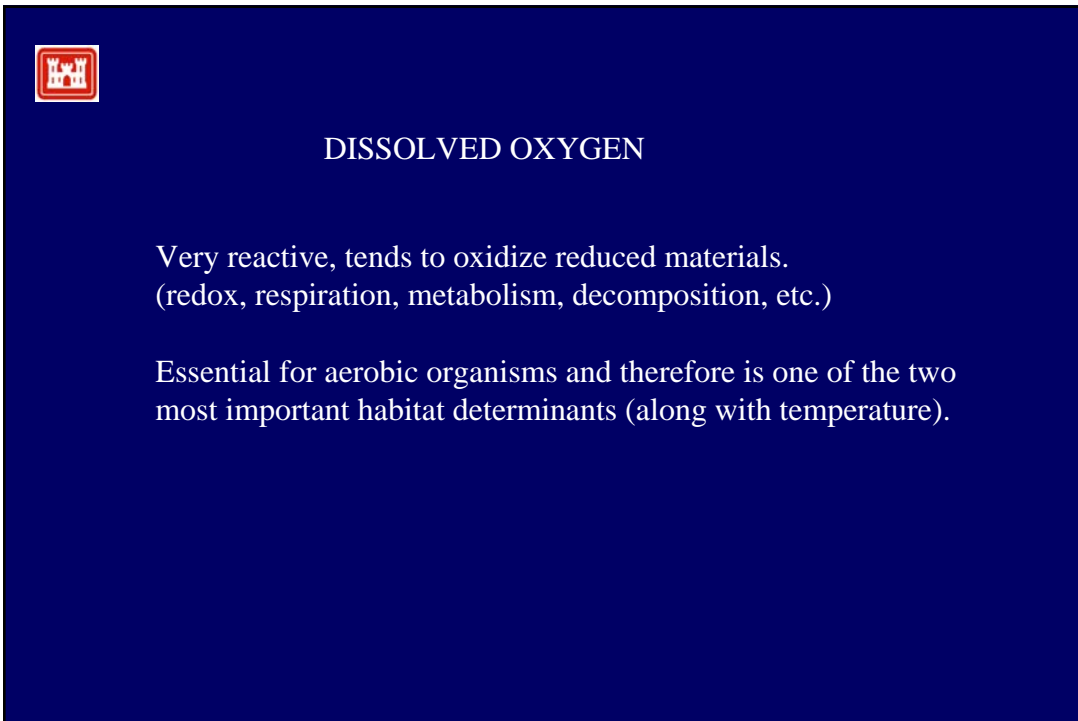
30 July 2002, Nashville, TN



## DISSOLVED OXYGEN

Very reactive, tends to oxidize reduced materials.  
(redox, respiration, metabolism, decomposition, etc.)

Essential for aerobic organisms and therefore is one of the two most important habitat determinants (along with temperature).







## DISSOLVED OXYGEN

Hutchinson's Master Variables

pH and pE

Of these, pE cannot be directly measured. Greatly affected by presence of oxygen, in aerobic environments oxygen can be used as a surrogate for pE.

Therefore, presence (or absence) of oxygen is critical for the biogeochemical state of the aquatic ecosystem.

Oxygen dynamics are therefore important for setting the limits of the environment for chemical reactions such as those involved with sulfides.



## Great Lies

Mark Twain – Statistics

Championship Wrestling – is real (sorry Steve)

Dams – don't inherently add anything to streams



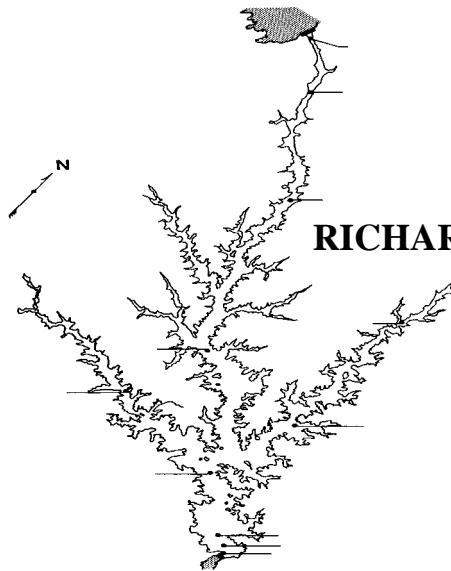
## DAMS DO ADD (REMOVE) FROM THE STREAM

Or, properly, they change the energetic status of the water during the conversion of lotic to lentic systems.

Kinetic energy is altered and converted to potential energy which is sometimes converted to electricity.



In a stream, this loss of kinetic energy results in a loss of aeration during stream flow (among other effects).



**RICHARD B. RUSSELL LAKE**



## J. STROM THURMOND LAKE (formerly Clarks Hill Lake)

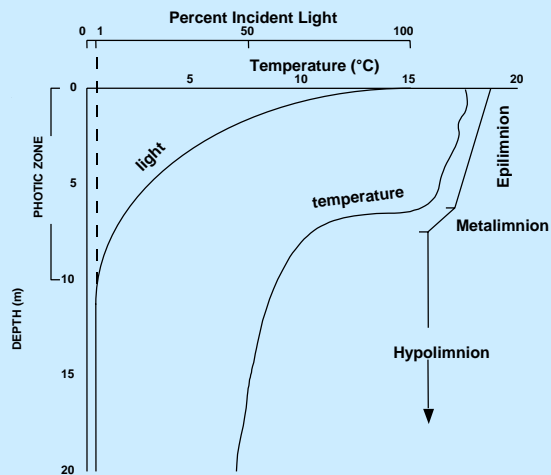


Diagram illustrating thermal patterns typical of a stratified monomictic lake.

What's a monomictic lake?



## Lake Classification Based on Mixing Processes

Monomictic = one period or season of mixing per year.

Common in south-temperate lakes. Example: Lake Lanier

Dimictic = two periods or seasons of mixing per year.

This is common in north-temperate lakes, especially those that have winter ice cover. Example: Lake Michigan

Meromictic = never mix completely, but always have an

unmixed deep density layer (the monimolimnion). Example: Carter's Lake.

Other types include polymixis, atelomixis, etc.





## Dissolved Oxygen Distributions in Lakes

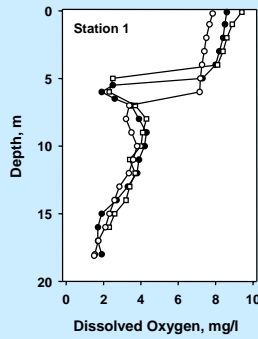
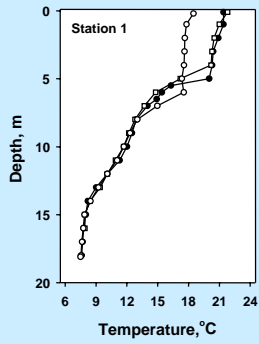
Orthograde – little, if any, gradient

Clinograde – gradient of declining concentration

Heterograde – gradient with discontinuity

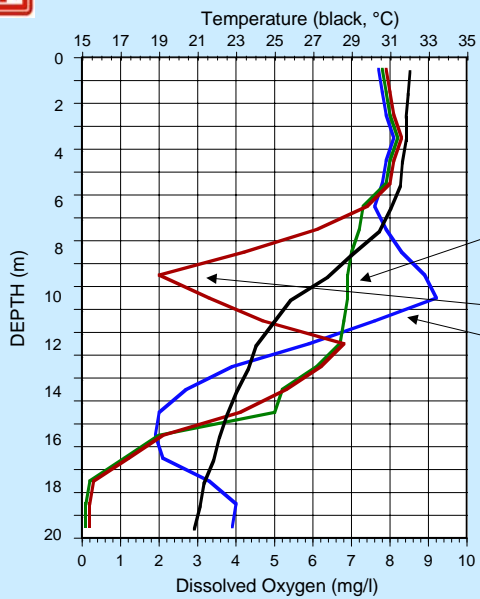
Negative – discontinuity is negative

Positive – discontinuity is positive



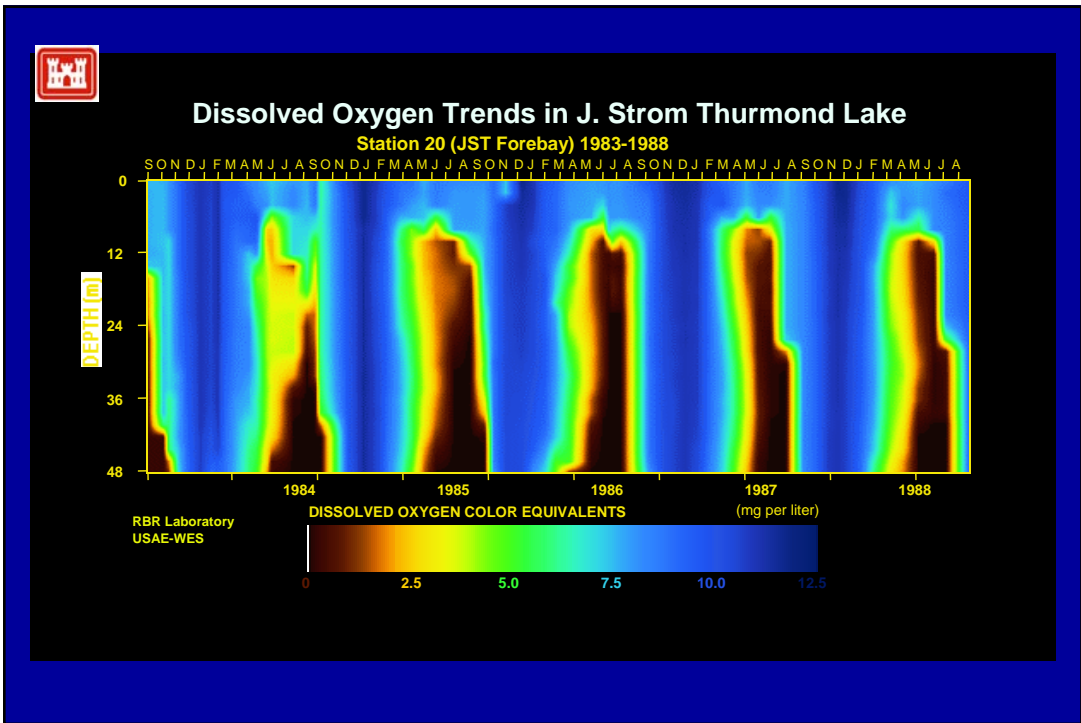
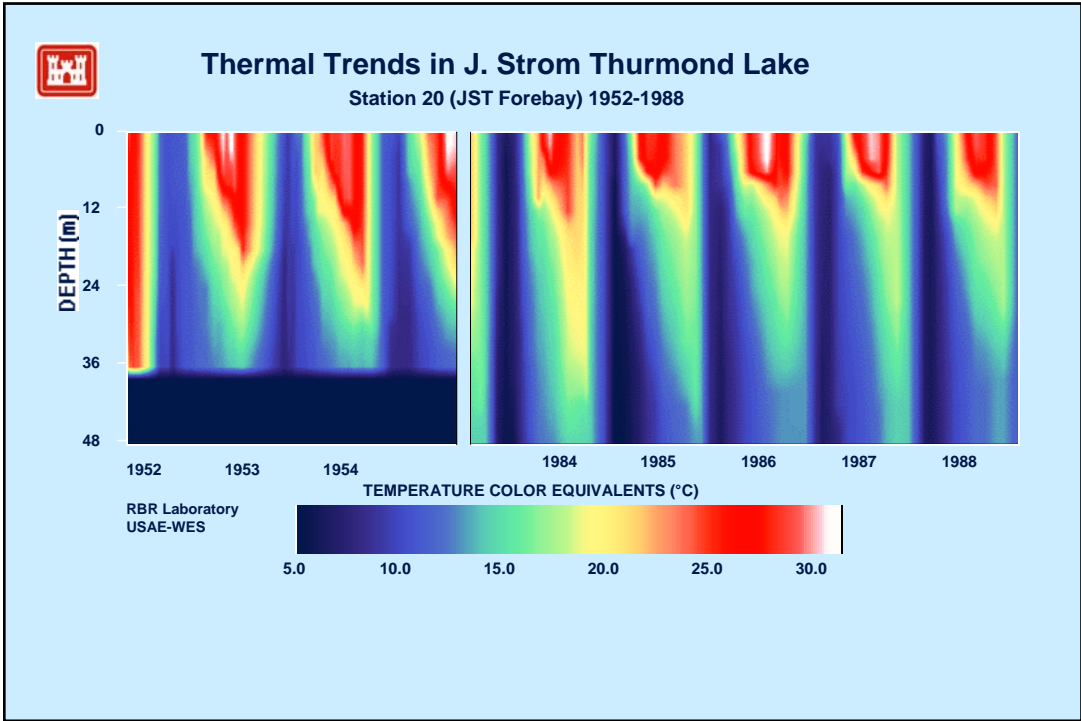
Symbol Legend

- 8/5/99
- 8/6/99
- 8/11/99

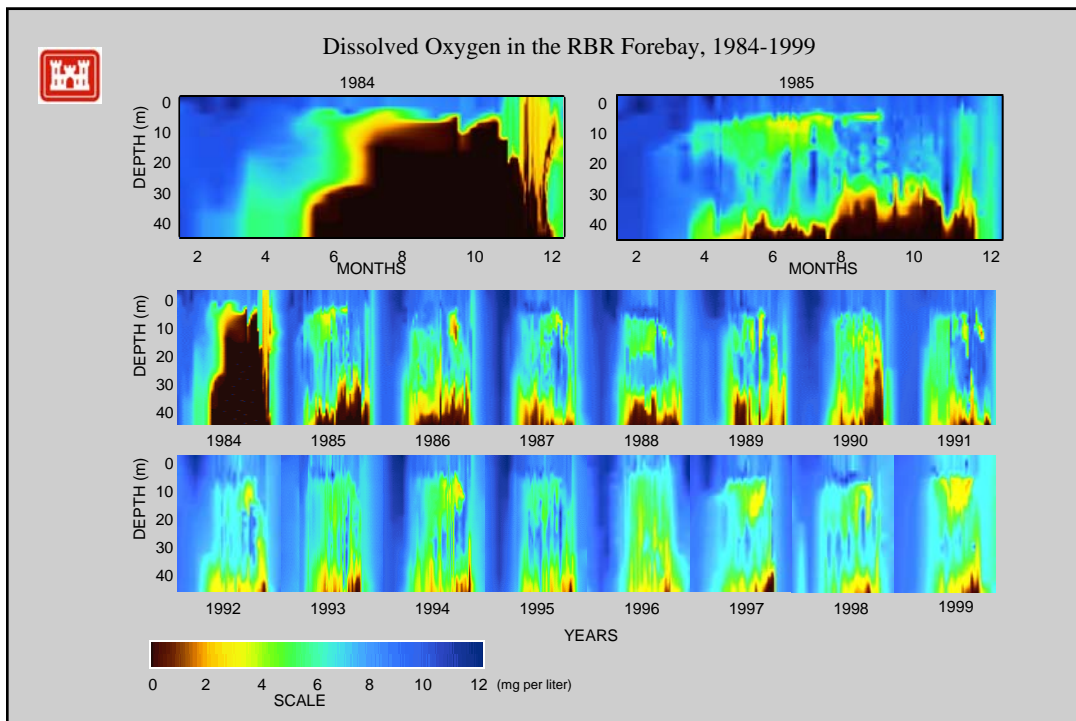
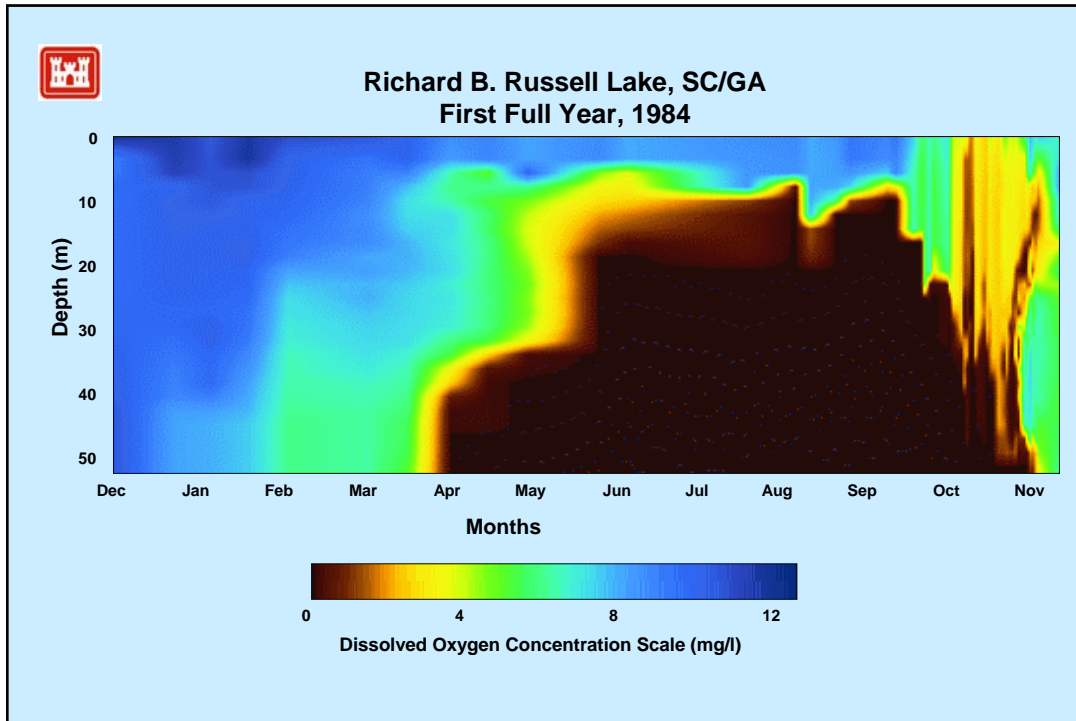


### Dissolved Oxygen Distributions in Lakes

- Orthograde – little, if any, gradient
- Clinograde – gradient of declining concentration
- Heterograde – gradient with discontinuity
  - Negative – discontinuity is negative
  - Positive – discontinuity is positive



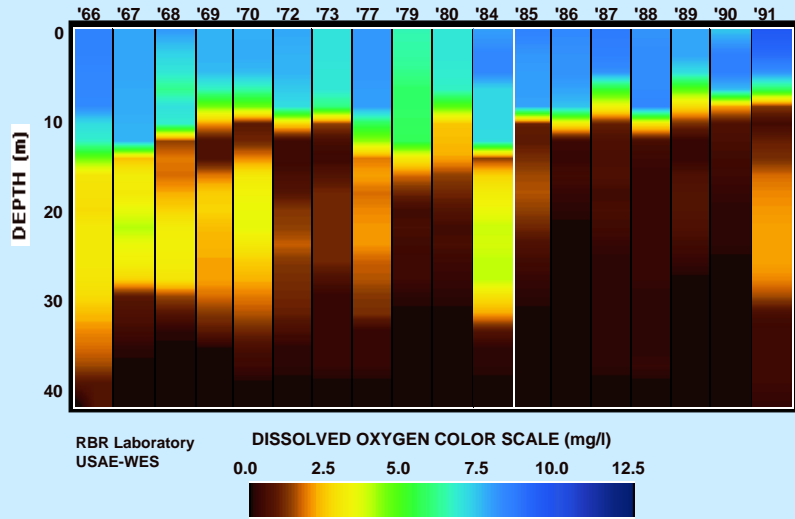




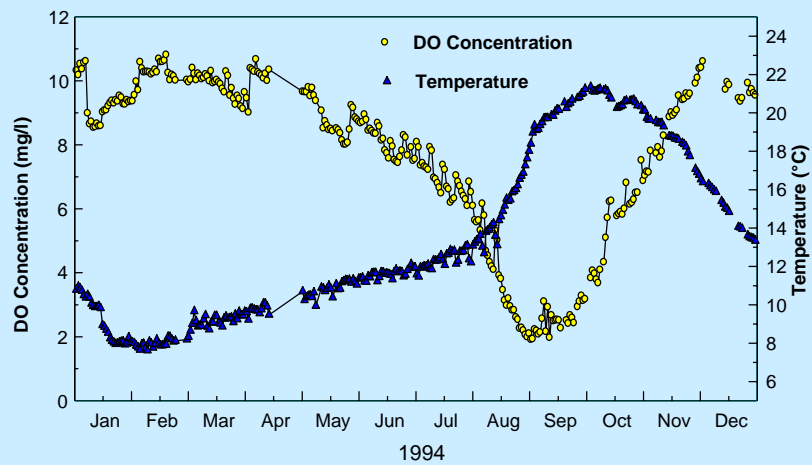


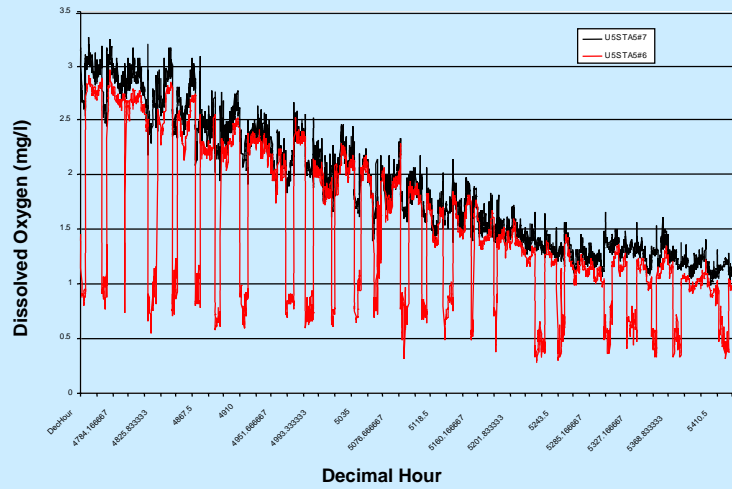


### HISTORICAL SEPTEMBER DISSOLVED OXYGEN CONCENTRATIONS J.STROM THURMOND FOREBAY



### 1994 Hartwell Dam Releases





Comparison of water quality from taps 6 and 7, penstock piezometer taps. Data collected during declining water quality conditions. Note periodic departures of tap #7 from #6 due to generation cycles.



## Summary and Questions