3. CHEMICAL AND PHYSICAL INFORMATION

3.1 CHEMICAL IDENTITY

Data pertaining to the chemical identity of 1,3-but adiene are listed in Table 3-1.

3.2 PHYSICAL AND CHEMICAL PROPERTIES

The physical and chemical properties of 1,3-but adiene are presented in Table 3-2. 3. CHEMICAL AND PHYSICAL INFORMATION

Characteristic	Information	Reference
Chemical name	1,3-Butadiene	CAS 1989
Synonyms	<pre>Butadiene; buta-1,3-diene; biethylene; bivinyl; vinylethylene; erythrene; α,α-butadiene; trans-butadiene; divinyl; pyrrolylene</pre>	SANSS 1989; Chemline 1989; HSDB 1989; CAS 1989
Trade name(s)	No data	
Chemical formula	C ₄ H ₆	CAS 1989
Chemical structure		
Identification numbers:		
CAS registry NIOSH RTECS EPA hazardous waste OHM/TADS DOT/UN/NA/IMCO shipping	106-99-0 EI9275000 R0377-0754 No data No data	CAS 1989 SANSS 1989 Miller 1978
HSDB NCI	181 C50602	Chemline 1989 Chemline 1989

TABLE 3-1. Chemical Identity of 1,3-Butadiene

CAS = Chemical Abstracts Service NIOSH = National Institute for Occupational Safety and Health RTECS = Registry of Toxic Effects of Chemical Substances EPA = Environmental Protection Agency OHM/TADS = Oil and Hazardous Materials Technical Assistance Data Base DOT/UN/NA/IMCO = Department of Transportation/United Nations/North America/ International Maritime Consultive Organization HSDB = Hazardous Substance Data Bank NCI = National Cancer Institute SANSS = Structure and Nomenclature Search System

3. CHEMICAL AND PHYSICAL INFORMATION

TABLE 3-2. Physical and Chemical Properties of 1,3-Butadiene

Property	Information	Reference
Molecular weight	54.09	Weast et al. 1988
Color	Colorless	Sax and Lewis 1987
Physical state	Gas	Sax and Lewis 1987
Melting point	-108.9°C	Weast et al. 1988
Boiling point		
at 1 atm	-4.4°C	Weast et al. 1988
at 5 atm	47°C	Windholz et al. 1983
at 10 atm	76°C	Windholz et al. 1983
Density (liquid) at 20°C	0.6211 g/mL	Weast et al. 1988
Odor	Mildly aromatic	Sax and Lewis 1987
Odor threshold	5	
Water	0.0014 mg/L	Amoore and Hautula 1983
Air	1.0-1.6 ppm (recognition)	Amoore and Hautula 1983
	0.025 ppm (detection)	Verschueren 1983
Solubility		
Water at 25°C	735 ppm	McAuliffe 1966
Organic solvents	Alcohol, ether,	Weast et al. 1988
5	acetone, benzene,	Windholz et al. 1983
	polar and nonpolar	
	organic solvents	
Partition coefficient	0	
Log octanol/water	1.99	Hansch and Leo 1985
Log K _{oc} (calculated	2.46	Lyman et al. 1982
from K _{ow})		y
342	2.59	Verschueren 1983
Vapor pressure at 25°C	2100 mmHg	Daubert and Danner 1985
Henry's law constant	5	
at 25°C (calculated)	7.05x10 ⁻² atm-m ³ /mol	Hine and Mookerjee 1975
Autoignition temperature	414°C	Sax and Lewis 1987
Flashpoint	-76°C	Sax and Lewis 1987
Flammability limits		
in air	Extremely flammable	Miller 1978
Conversion factors		
ppm (v/v) to mg/m^3 in	2.21	IARC 1986
air (20°C)		
mg/m ³ to ppm (v/v) in	0.445	IARC 1986
air (20°C)		
Bioconcentration factor	19	Lyman et al. 1982
(calculated from K _{ow})		
Explosive limits	2-11.5%	Kirshenbaum 1978
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