

## **4. CHEMICAL AND PHYSICAL INFORMATION**

### **4.1 CHEMICAL IDENTITY**

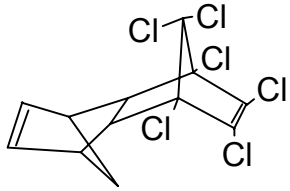
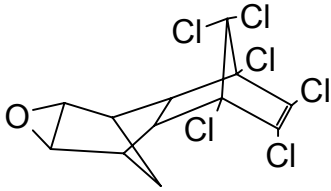
Information regarding the chemical identity of aldrin/dieldrin is located in Table 4-1.

### **4.2 PHYSICAL AND CHEMICAL PROPERTIES**

Information regarding the physical and chemical properties of aldrin/dieldrin is located in Table 4-2.

## 4. CHEMICAL AND PHYSICAL INFORMATION

**Table 4-1. Chemical Identity of Aldrin and Dieldrin<sup>a</sup>**

Characteristic	Aldrin	Dieldrin
Synonym(s)	1,2,3,4,10,10-Hexachloro-1,4,4 $\alpha$ 5,8,8 $\alpha$ -hexahydro-exo-1,4-endo-5,8-dimethano-naphthalene; HHDN <sup>b</sup>	1,2,3,4,10,10-Hexachloro-6,7-epoxy-1,4,4 $\alpha$ ,5,6,7,8,8 $\alpha$ -octahydro-1,4-endo,exo-5,8-dimethanonaphthalene; HEOD <sup>b</sup>
Registered trade name(s)	Aldrec; Aldrex; Drinox; Octalene; Seedrin; Compound 118	Alvit; Dieldrix; Octalox; Quintox; Red Shield <sup>c</sup>
Chemical formula	C <sub>12</sub> H <sub>8</sub> Cl <sub>6</sub>	C <sub>12</sub> H <sub>8</sub> Cl <sub>6</sub> O
Chemical structure <sup>d</sup>		
Identification numbers:		
CAS registry	309-00-2	60-57-1
NIOSH RTECS	IO2000000	IO1750000
EPA hazardous waste	P004	PO37
OHM/TADS	7215090 <sup>c</sup>	7216516 <sup>c</sup>
DOT/UN/NA/IMCO shipping	IM06.1 NA2762	UN2761
HSDB	199	322
NCI	C00044	C00124

<sup>a</sup>All information obtained from HSDB 2001a or 2001b unless otherwise noted.

<sup>b</sup>Tomlin 1997

<sup>c</sup>OHM/TADS 1990b

<sup>d</sup>Verschueren 2001

CAS = Chemical Abstracts Services; DOT/UN/NA/IMCO = Department of Transportation/United Nations/North America/International Maritime Dangerous/Goods Code; EPA = Environmental Protection Agency; HSDB = Hazardous Substances Data Bank; NCI = National Cancer Institute; NIOSH = National Institute for Occupational Safety and Health; OHM/TADS = Oil and Hazardous Materials/Technical Assistance Data System; RTECS = Registry of Toxic Effects of Chemical Substances

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**Table 4-2. Physical and Chemical Properties of Aldrin and Dieldrin<sup>a</sup>**

Property	Aldrin	Dieldrin
Molecular weight	364.91	380.91
Color	White (pure); tan to brown (technical grade)	White (pure); light brown (technical grade)
Physical state	Crystalline solid <sup>b</sup>	Crystalline solid <sup>b</sup>
Melting point	104–105.5 EC <sup>c</sup> ; 49–60 EC (technical grade) <sup>c</sup>	176–177 EC <sup>c</sup> ; 95 EC (technical grade) <sup>d</sup>
Boiling point	Decomposes <sup>e</sup>	Decomposes <sup>e</sup>
Density	1.6 g/L at 20 EC <sup>f</sup>	1.75 g/L at 25 EC <sup>f</sup>
Odor	Mild chemical odor <sup>e</sup>	Mild chemical odor <sup>e</sup>
Odor threshold:		
Water	No data	No data
Air	0.017 mg/kg <sup>c</sup>	0.041 mg/kg <sup>c</sup>
Solubility:		
Water at 20 EC	0.011 mg/L <sup>g</sup>	0.110 mg/L <sup>g</sup>
Organic solvents	Very soluble in most organic solvents <sup>b</sup>	Moderately soluble in common organic solvents except aliphatic petroleum solvents and methyl alcohol <sup>b</sup>
Partition coefficients:		
Log K <sub>ow</sub>	6.50 <sup>h</sup>	6.2 <sup>c</sup>
Log K <sub>oc</sub>	7.67 <sup>i</sup>	6.67 <sup>i</sup>
Vapor pressure:		
at 20 EC	7.5x10 <sup>-5</sup> mmHg <sup>b</sup>	3.1x10 <sup>-6</sup> mmHg <sup>b</sup>
at 25 EC	1.2x10 <sup>-4</sup> mmHg	5.89x10 <sup>-6</sup> mmHg <sup>j</sup>
Henry's law constant:		
at 25 EC	4.9x10 <sup>-5</sup> atm-m <sup>3</sup> /mol <sup>k</sup>	5.2x10 <sup>-6</sup> atm-m <sup>3</sup> /mol <sup>k</sup>
Autoignition temperature	No data	No data
Flashpoint	No data	No data
Flammability limits	Nonflammable <sup>f</sup>	Nonflammable <sup>f</sup>
Conversion factors	1 ppm=14.96 mg/m <sup>3</sup> at 25 EC, 1 atm	1 ppm=15.61 mg/m <sup>3</sup> at 25 EC, 1 atm <sup>l</sup>
Explosive limits	Stable <sup>f</sup>	Stable <sup>f</sup>

<sup>a</sup>All information obtained from HSDB 2001a or 2001b unless otherwise noted.

<sup>b</sup>Budavari et al. 2001

<sup>c</sup>Verschueren 2001

<sup>d</sup>Hayes 1982

<sup>e</sup>NIOSH 1997

<sup>f</sup>Weiss 1986

<sup>g</sup>Bus and Leber 2001

<sup>h</sup>Hansch et al. 1995

<sup>i</sup>Briggs 1981

<sup>j</sup>Grayson and Fosbraey 1982

<sup>k</sup>Guerin and Kennedy 1992

<sup>l</sup>EPA 1987a

