

A Family of Cyanide-Bridged Molecular Squares: Structural and Magnetic Properties of $[\{M^{II}Cl_2\}_2\{Co^{II}(\text{triphos})(CN)_2\}_2] \cdot xCH_2Cl_2$, $M = Mn, Fe, Co, Ni, Zn$

F. Karadas (1), E. J. Schelter (1), M. Shatruk (1), A. V. Prosvirin (1), J. Bacsá (1), D. Smirnov (2), A. Ozarowski (2), J. Krzystek (2), J. Telser (3) and K. R. Dunbar (1)

(1) Department of Chemistry, Texas A&M University, College Station, (2) National High Magnetic Field Laboratory, Florida State University, Tallahassee, (3) Department of Chemical, Physical and Biological Sciences, Roosevelt University, Chicago, Illinois.



Single molecule magnets (SMMs), assembled from building blocks of simple paramagnetic metal-containing units have great potential for application in magnetic storage media and electronic devices. The key feature is a well-defined geometric relationship among the building blocks. This has been achieved in the synthesis of a novel family of “molecular squares” presented here.

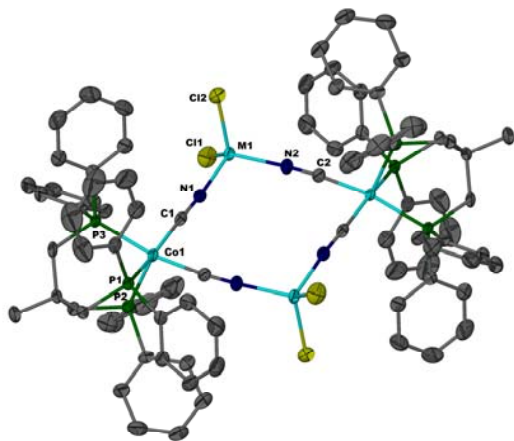


Figure 1. Structure of $[CoM]$ squares.

Of all the “molecular squares”, the one based on Co^{II} and Mn^{II} shows the most promise and best properties of being a SMM (see the brown trace in Fig. 2).

Magnetic properties of the “molecular squares” were also investigated using magnetic resonance at the NHMFL EMR and DC Field Facilities.

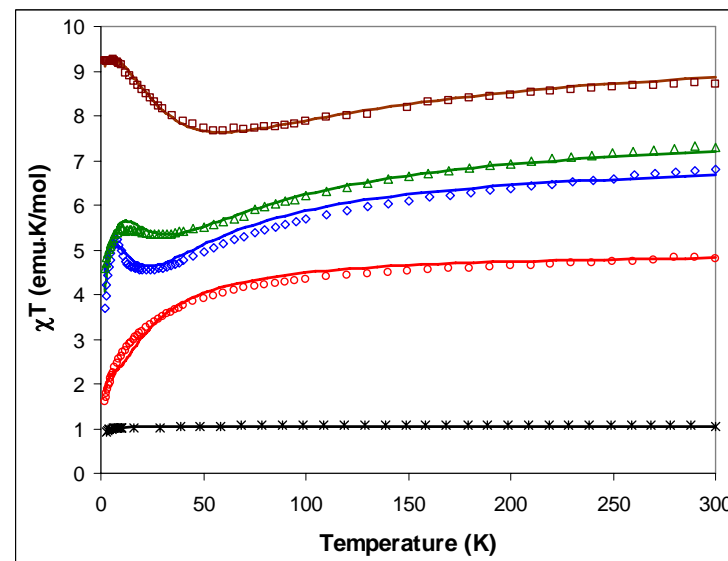


Figure 2. Magnetic properties of molecular squares: Temperature dependence of the χT product for compounds $[CoMn]$ (\square), $[CoFe]$ (Δ), $[CoCo]$ (\diamond), $[CoNi]$ (o), $[CoZn]$ (*).