BENZIDINE in urine (SCREENING TEST)

8304

$C_{12}H_{12}N_{2}$	MW: 184.24	CAS: 92-87-5	RTECS: DC9625000
METHOD: 8304, Issue 2	EVALUATION: PARTIAL		lssue 1: 15 February 1984 Issue 2: 15 August 1993

BIOLOGICAL INDICATOR OF: exposures to benzidine-based azo dyes.

SYNONYMS: [1,1'-biphenyl]-4,4'-diamine

BIOLOGICAL SAMPLING		MEASUREMENT	
SPECIMEN:	2 urine samples, 150 mL each, before and after 6 h of exposure	TECHNIQUE:	VISIBLE ABSORPTION/THIN LAYER CHROMATOGRAPHY (TLC)
PRESERVATIVE:	none	ANALYTE:	2,4,5-trinitrobenzene sulfonic acid derivative of benzidine
SHIPMENT:with dry ice in insulated containerSTABILITY:stable for 2 months @ - 20 °CCONTROLS:collect urine from non-exposed workers	WAVELENGTH: TLC IDENTIFICATION: ESTIMATED LOD:	400 nm UV and visible $R_f = 0.41$ 0.1 µg/100 mL urine (visible	
		CALIBRATION.	absorption); 0.3 µg/100 mL urine (TLC)
		QUALITY CONTROL:	frozen pooled urine
		RANGE:	0.1 to 20 µg/100 mL urine
		PRECISION (Ŝ,):	0.12 @ 0.5 μg/100 mL urine
		ACCURACY:	± 53%

APPLICABILITY: This method is specific for aromatic amines and can be used to screen workers exposed to benzidine or benzidine-based azo dyes.

INTERFERENCES: In addition to false positives from other free aromatic amines, some drugs (e.g., antihistamines) contain free aromatic amines; however, these compounds do not produce R $_{\rm f}$ values corresponding to benzidine by TLC.

OTHER METHODS: This method replaces P&CAM 315 [1] with minor revisions. Method 8306 is a specific method for benzidine in urine by electron capture gas chromatography.

REAGENTS:

- 1. Methyl alcohol.
- 2. Benzidine, 99% (CAUTION: CARCINOGEN).
- Benzidine stock solution, 500 μg/mL. Weighed 50 mg benzidine. Dissolve in methyl alcohol to make 100 mL solution. Stable one month at -8°C.
- Calibration stock solution, 10 μg/mL. Dilute 1.00 mL benzidine stock solution to 50 mL with methyl alcohol. Prepare fresh daily.
- 5. Chloroform.
- 6. Hydrochloric acid.
 - a. 1.0 <u>N</u>. Dilute 83 mL conc. HCl to 100 mL with distilled water.
 - b. 0.1 <u>N</u>. Dilute 10 mL 1 <u>N</u> HCl to 100 mL with distilled water.
- 7. Sodium hydroxide, 1 <u>N</u>. Dissolve 40 g NaOH in water to make 1 L solution.
- 8. Sodium Chloride, crystals.
- Sodium acetate buffer, 2 <u>M</u>, pH 5.5 Titrate 2 <u>M</u> sodium acetate with 6 <u>N</u> HCI. Refrigerate.
- 2,4,6-Trinitrobenzene sulfonic acid (TNBS), 0.1 g/mL. Dissolve 2.5 g TNBS in 25 mL distilled water. Stable seven days when stored in the dark.
- 11. Acetone.
- 12. Formic acid.
- 13. Nitrogen, compressed.
- 14. Chloroform:formic acid, 90:10 (v/v). Prepare fresh daily.

EQUIPMENT:

- 1. Polyethylene bottles, 250-mL.
- 2. Spectrophotometer for measuring absorbance at 400 nm and 1-mL semi-microcuvettes.
- 3. Centrifuge, 400 rpm.
- 4. Rotator for mixing 25×200 mm test tubes.
- 5. pH meter.
- 6. TLC plates precoated with silica gel activated at 110 °C for 30 min., without flourescence indicator (0.5 mm thickness).
- 7. Chromatographic TLC tank.
- 8. UV source for reading TLC plates.
- 9. Glass bottles, 180-mL, with PTFE-lined caps.
- 10. Pipettes, glass, 2-, 5- and 100-mL, with pipet bulb.
- 11. Separatory funnels, 125-mL.
- 12. Volumetric flasks, 25- and 100-mL; 10-mL amber.
- Culture tubes, glass with PTFE-lined caps (16 x 125 mm and 25 x 200 mm).
- 14. Micropipettes, 0.01-, 0.1- and 0.7-mL.
- 15. Pasteur pipettes.
- 16. Plastic gloves.
- 17. Dessicator.
- 18. pH paper (pH 2).

* See Special Precautions.

SPECIAL PRECAUTIONS: Benzidine is a known human carcinogen. Appropriate precautions should be utilized to minimize exposures.

All wastes, including acetone rinses of dirty glassware should be collected and disposed of by approved methods.

SAMPLING:

- 1. Take pre- and post-shift urine samples (ca. 150 mL) in 250-mL polyethylene bottles.
- 2. Ship the samples with dry ice in an insulated container.

SAMPLE PREPARATION:

- 3. Defrost sample if frozen. Adjust the urine pH to between 5 and 6 with 1 <u>N</u> HCl or 1 <u>N</u> NaOH.
- 4. Pipet 100 mL urine into a 180-mL glass bottle. Start a blank control urine sample (100 mL) and two control urine samples (100 mL) spiked with benzidine (0.3 to 1.0 μg) at this point.
- 5. Add 0.2 g NaCl to the pH-adjusted urine.
- 6. Extract the urine twice more with 10 mL chloroform for 2 min. If an emulsion forms centrifuge to seperate the two phases. Collect and save the chloroform fraction.

- 7. Extract the urine twice more with 10-mL portions of chloroform. Combine the three chloroform fractions.
- 8. Re-extract the combined chloroform mixture with 2 mL 0.1 \underline{N} HCl for 30 min on a rotator.
- 9. Transfer the aqueous phase (ca. 2 mL) into a culture tube (16 × 125 mm) using a Pasteur pipet.
- Add 2 mL pH 5.5 sodium acetate buffer and 0.7 mL of TNBS reagent, mix well, and let stand for 15 min at room temperature. Start reagent blank (2 mL 0.1 <u>N</u> HCl, 2 mL pH 5.5 sodium acetate buffer, and 0.7 mL TNBS reagent).

CALIBRATION AND QUALITY CONTROL:

- 11. Prepare a series of working standards in the range 0 to 20 µg benzidine/100 mL urine by adding aliquots of calibration stock solution to 100-mL portions of control urine (urine pool previously shown to have <0.1 µg benzidine/100 mL urine).
- 12. Prepare and analyze the working standards (steps 3 through 10 and 14 through 16).
- 13. Prepare a calibration graph, absorbance at 400 nm vs. concentration of analyte (μg/110 mL urine).

MEASUREMENT:

- 14. Add 2 mL CHCl₃ to the extract in step 10 and shake for 1 min.
- Measure the absorbance vs. the reagent blank of the organic phase at 400 nm. Retain the organic phase for the benzidine-TLC confirmation (steps 16 through 19) if the sample contains more than 0.3 μg/100 mL.
- 16. Concentrate the organic phase containing the TNBS-amine derivative by evaporating with nitrigen to ca. 0.2 mL.
- 17. Spot 10 µL of the concentrated organic phase on an activated silica gel TLC plate.
- 18. Develop the plate in 90:10 chloroform:formic acid.
- 19. Compare the R_f of the unknown amine derivative with that of a benzidine-spiked derivative. Benzidine produces a spot with R_f = 0.41 which is yellow in visible light and dark under UV (254 nm) light.

CALCULATIONS:

20. Obtain the concentration of the analyte in the urine sample by comparing its absorbance with the calibration graph.

NOTE: Corrections for extraction efficiency are not needed since standards are prepared in urine and both standards and samples are treated the same way.

GUIDES TO INTERPRETATION:

In this laboratory, normal ranges of urine specimens from NIOSH employees not exposed to benzidine or aromatic amines where:

Number of Urine Specimens	<u>Aromatic_Amine_Conc. (µg/100_mL)</u>
10	<0.1
2	0.1 to 0.2
1	0.2
1	0.3

Benzidine was not detected by TLC in any of the 14 urine specimens (LOD = $0.3 \mu g/100 mL$).

EVALUATION METHOD:

Ten spiked urine specimens containing 0.5 μ g benzidine/100 mL urine each were analyzed. Precision, S_r, for the 10 specimens was 0.12.

REFERENCES:

[1] NIOSH Manual of Analytical Methods, 2nd. ed., V. 5, P&CAM 315, U.S. Department of Health Education, and Welfare, Publ. (NIOSH) 79-141 (1979).

METHOD WRITTEN BY:

William P. Tolos, NIOSH/DBBS.