

UNITED STATES DEPARTMENT OF COMMERCE  
WASHINGTON, D. C.

National Bureau of Standards  
Certificate of Analyses

Standard Sample 85 B  
Aluminum Alloy  
(Wrought)

ANALYST	COPPER Electrolytic	MAGNESIUM NaOH-Mg <sub>2</sub> P <sub>2</sub> O <sub>7</sub>	MANGANESE Photometric	IRON Photometric	CHROMIUM Photometric	SILICON	NICKEL Photometric	ZINC ZnS-ZnO	TITANIUM Photometric	LEAD Weighed as PbO <sub>2</sub>	GALLIUM	VANADIUM Photometric
1.....	<sup>a</sup> 4.00	1.50	<sup>b</sup> 0.61	<sup>c</sup> 0.23	<sup>d</sup> 0.208	<sup>e</sup> 0.17	<sup>f</sup> 0.087	0.030	0.023	0.023	<sup>g</sup> 0.018	0.006
2.....	3.98	1.49	<sup>h</sup> .61	<sup>i</sup> .24	<sup>j</sup> .205	<sup>k</sup> .18	.091	<sup>l</sup> .027	.020	.022	<sup>m</sup> .019	.006
3.....	<sup>n</sup> 4.00	<sup>o</sup> 1.50	<sup>h</sup> .61	{ <sup>i</sup> .23 <sup>o</sup> .24 }	<sup>p</sup> .206	<sup>q</sup> .18	{ <sup>r</sup> .077 <sup>r</sup> .078 }	{ <sup>s</sup> .032 <sup>t</sup> .033 }	.022	{ <sup>u</sup> .020 <sup>u</sup> .023 }		
4.....	{ <sup>n</sup> 3.97 4.03 }	<sup>u</sup> 1.48	<sup>h</sup> .61	<sup>i</sup> .23	<sup>j</sup> .232	<sup>e</sup> .17	.086	.026	.022	.017		
5.....	3.98	1.49	<sup>v</sup> .61	<sup>w</sup> .24	<sup>x</sup> .21	<sup>v</sup> .17	<sup>f</sup> .079	<sup>s</sup> .026	.024	<sup>s</sup> .016		
6.....	3.99	{ <sup>u</sup> 1.49 1.50 }	<sup>h</sup> .61	<sup>w</sup> .24	<sup>j</sup> .21	<sup>k</sup> .19	<sup>t</sup> .087	{ <sup>s</sup> .034 <sup>l</sup> .036 }	.022	.027	<sup>k</sup> .02	
7.....	3.99	<sup>u</sup> 1.49	<sup>b</sup> .63	<sup>z</sup> .24	<sup>j</sup> .208	<sup>q</sup> .19	.086	<sup>s</sup> .030	.022	<sup>s</sup> .022		
Average.....	3.99	1.49	0.61	0.24	0.211	0.18	0.084	0.030	0.022	0.021	0.019	0.006

<sup>a</sup> Three-gram sample dissolved in sulfuric-nitric-hydrochloric acids. Solution evaporated to fumes of sulfuric acid, diluted and filtered. Silica treated with H<sub>2</sub>SO<sub>4</sub>-HF and the residual solution combined with the first filtrate. First cathode deposit dissolved and replated.  
<sup>b</sup> Persulfate oxidation and potentiometric titration with sodium arsenite solution standardized with potassium permanganate.  
<sup>c</sup> SnCl<sub>2</sub>-K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> method.  
<sup>d</sup> Persulfate oxidation and potentiometric titration with ferrous ammonium sulfate solution standardized with potassium dichromate.  
<sup>e</sup> Solution in NaOH, double HClO<sub>4</sub> dehydration with intervening filtration.  
<sup>f</sup> Dimethylglyoxime-gravimetric method.

<sup>g</sup> Gallium extracted with ether, precipitated with cupferron, and ignited to the oxide. See NBS J. Research 16, 585 (1935) RP853.  
<sup>h</sup> Periodate method.  
<sup>i</sup> Ortho-phenanthroline method.  
<sup>j</sup> Diphenylcarbazide method.  
<sup>k</sup> Gravimetric method.  
<sup>l</sup> ZnHg(CNS)<sub>2</sub> method.  
<sup>m</sup> 8-hydroxyquinoline-photometric method.  
<sup>n</sup> Iodide-thiosulfate method.  
<sup>o</sup> Magnesium oxquinolate precipitation, and titration by the KBr-KBrO<sub>3</sub>-Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> method.  
<sup>p</sup> Persulfate oxidation and titration with ferrous sulfate and potassium dichromate with diphenylamine sulfonate as indicator.

<sup>q</sup> Molybdisilicic acid-photometric method.  
<sup>r</sup> Polarographic method.  
<sup>s</sup> Dithizone-photometric method.  
<sup>t</sup> Zinc oxyquinolate precipitation after H<sub>2</sub>S separation of copper, and titration by the KBr-KBrO<sub>3</sub>-Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> method.  
<sup>u</sup> Ethylenediaminetetraacetic acid (Versene) titration method.  
<sup>v</sup> Persulfate-arsenite method.  
<sup>w</sup> Iron reduced with H<sub>2</sub>S and titrated with KMnO<sub>4</sub>.  
<sup>x</sup> Persulfate oxidation and titration with ferrous sulfate-permanganate.  
<sup>y</sup> Tri-acid decomposition.  
<sup>z</sup> Bypyridine-photometric method.

List of Analysts

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| <p>Nonferrous Laboratory, National Bureau of Standards, R. K. Bell in charge. Analysis by E. E. Maczkowske and L. A. Machlan.</p> <p>2. J. R. Churchill, Aluminum Company of America, New Kensington, Pa.</p> <p>3. R. B. Beckett, Aluminum Laboratories, Ltd., Arvida, Quebec, Canada.</p> | <p>4. G. B. Wengert and P. F. Reigler, The Dow Chemical Co., Midland, Mich.</p> <p>5. C. J. Clausen, Jr., Kaiser Aluminum and Chemical Corporation, Spokane, Wash.</p> <p>6. R. L. Vitek and J. W. Mierzwa, Apex Smelting Co., Cleveland, Ohio.</p> <p>7. K. C. Braun, American Smelting and Refining Co., South Plainfield, N. J.</p> |
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The aluminum alloy for the preparation of this standard was furnished by the Aluminum Company of America.

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A. V. ASTIN, Director.