

National Bureau of Standards

Certificate of Analysis

Standard Reference Material 2128

Spectrometric Standard Solutions

Gold, Mercury, Palladium, and Platinum

This Standard Reference Material (SRM) is intended for use in atomic absorption spectrometry, optical emission (plasma) spectrometry, spectrophotometry, or any other analytical technique that requires aqueous standard solutions for calibrating instruments. SRM 2128 consists of four single element solutions of Au, Hg, Pd, and Pt. Each solution contains 50 mL and was prepared gravimetrically at 22 °C to contain 10.00 ± 0.01 mg/mL of the metal ion in the percent (V/V) acid medium as shown in Table 1. The certified values (Table 1) are based on gravimetric procedures, i.e., weight per volume composition of high-purity metals dissolved in NBS high-purity reagents.

Table 1

<u>Solution</u>	<u>Metal</u>	<u>Concentration (mg/mL)</u>	<u>Source (Purity, %)</u>	<u>Acid Conc. (V/V) (Approximate)</u>
2128-1	Au	10.00 ± 0.01	SRM 685W (99.99)	HCl, 10%
2128-2	Hg	10.00 ± 0.01	SRM 743 (99.99)	HNO ₃ , 10%
2128-3	Pd	10.00 ± 0.01	Pd metal (99.99)*	HCl, 10%
2128-4	Pt	10.00 ± 0.01	SRM 680 (99.99)	HCl, 10%

*This high-purity material was analyzed by optical emission spectrometry and atomic absorption spectrometry and found to contain less than 50 µg/g total impurities.

SRM 2128 was prepared by T.C. Rains of the Inorganic Analytical Research Division. Atomic absorption and emission spectrometric analyses were made by T.A. Butler, T.A. Rush, T.C. Rains, and J.A. Norris.

The technical and support aspects involved in the preparation, certification, and issuance of this Standard Reference Material were coordinated through the Office of Standard Reference Materials by R.W. Seward.

Gaithersburg, MD 20899
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Stanley D. Rasberry, Chief
Office of Standard Reference Materials

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Procedures for Use

Stability:

This certificate is valid for one year from the shipping date provided the solutions are kept tightly capped and stored under proper laboratory conditions. NBS will monitor the stability of these solutions; if any changes occur that invalidate this certification, purchasers will be notified by NBS.

Preparation of Working Standard Solutions:

All solutions should be at 22 ± 1 °C and all glass or plastic surfaces coming into contact with the standard must have been previously cleaned. The working standard solution is prepared from the SRM solutions by serial dilution. The dilution should be made into certified volumetric class A flasks with 5 or 10 mL class A pipets. All volumetric transfers of solutions should be performed by a proven analytical technique. Each dilution should be acidified with an appropriate high-purity acid and diluted to calibrated volume using high-purity water. The stability of the working standard solution will depend upon the final acid concentration. To achieve the highest accuracy, it is recommended that the analyst prepare daily working solutions from 100 µg/mL dilutions of the original SRM stock solutions.